

Comparative assessment of innovative and conventional food preservation technologies: process energy performance and greenhouse gas emissions

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Abstract

This study aims to establish whether innovative food preservation technologies can offer significant reductions in energy consumption and corresponding greenhouse gas (GHG) emissions while delivering equivalent microbiological lethality, nutritional and organoleptic quality to conventional processes. The energy demand of high pressure processing, microwave, ohmic and conventional heating technologies, for achieving the same pasteurising effect in orange juice under commercially-representative processing conditions are measured and compared. The corresponding GHG emissions are evaluated using UK energy system emissions data, while the effect of equipment scale is explored empirically. The results show that for the same product quality, the innovative technologies are more energy- and non-renewable primary resource-efficient, with ohmic heating performing best, followed by high pressure

26 **processing at high fill-ratios.** More significant improvements are expected in future,
27 provided electricity grid decarbonisation is sustained. Energy performance improves
28 with equipment scale for the microwave and high-pressure systems, but remains
29 essentially constant for ohmic heating.

30

31 Keywords: food preservation, energy demand, carbon emissions, high pressure
32 processing, microwave volumetric heating, ohmic heating

33

34 *Industrial relevance:* Using orange juice pasteurisation as case study, this work shows
35 that for similar product quality, electrically-driven **innovative** food pasteurisation
36 technologies like **high pressure processing**, ohmic and microwave heating are more
37 beneficial than conventional techniques energy- and emission-wise, if a sufficient
38 portion of the electricity is renewable. This is the case given the current
39 decarbonisation level of the UK electricity grid and is expected to be more significant
40 as more electricity is sourced renewably in the future as currently projected. The result
41 should aid future industrial investment decisions.

42

43 **1. Introduction**

44

45 The demand for additive-free, extended shelf-life food products, with fresh-like tastes,
46 excellent nutritional quality and guaranteed microbial safety, has led to the
47 development of **innovative** processing and preservation technologies. A range of these
48 technologies is becoming commercially available to the food industry and there is
49 significant interest in their potentials. Among them are non-thermal processes such as
50 high pressure, pulsed electric field, pulsed light, ultrasonic processing, and volumetric
51 heating schemes such as microwave, radiofrequency and ohmic heating. Currently at

52 relatively low levels of industrial implementation, the potentials of these technologies
53 in contributing significantly to sustainably meeting the food intake needs of the growing
54 population of the world has long been recognised (Langelaan et al., 2013; Probst et
55 al., 2015). The food and beverage market is extremely competitive. Therefore,
56 improving quality alone is not sufficient to assure the success of a new technology.
57 The improvements must be achievable at a commercially viable process cost per
58 production unit. An understanding of the energy demand associated with the use of
59 novel preservation techniques is therefore commercially important, and is the subject
60 of the current contribution.

61

62 Although there are many studies and reviews on different aspects of quality and
63 microbial safety of various foods processed by these technologies (e.g. Atuonwu and
64 Tassou, 2018a, Barba et al., 2017 a,b), the same cannot be said for energy and
65 sustainability studies. A recent review on microwave food processing (Atuonwu and
66 Tassou, 2018b) for instance, concludes that little is available in the literature on studies
67 dedicated specifically to energy consumption analysis of microwave food processes.
68 An exception is in the area of microwave and microwave-assisted food drying
69 operations, which were extensively reviewed in that work. Similar conclusions could
70 be reached for most other innovative food preservation technologies, although some
71 energy studies at individual unit operation level, exist (e.g. Kurjak et al., 2012;
72 Cokgezme et al., 2017; Park et al., 2017; Atuonwu and Tassou, 2018c). Fewer
73 comparative studies across technologies are available in the literature. Toepfl et al.
74 (2006) explored the potentials of pulsed electric field PEF and high pressure
75 processing HPP technologies for energy-efficient and environmentally-friendly food
76 processing. Lung et al. (2006) estimated the potential energy savings of PEF orange

77 juice pasteurisation and radiofrequency RF cookie drying, relative to conventional
78 technologies. They reported a 100% savings in natural gas and 18% savings in
79 electricity for the PEF system, and a natural gas savings of 73.8-147.7 TJ per year,
80 but an increased electricity consumption for the RF system. Pereira and Vicente
81 (2010) highlighted the reduced environmental impact potentials of novel thermal and
82 non-thermal technologies in food processing. Sampedro et al. (2014) compared the
83 costs and environmental impacts of PEF, HPP and thermal pasteurisation
84 technologies based on commercial processing conditions validated for a 2-month shelf
85 life of orange juice under refrigeration conditions. The total electricity consumption of
86 the HPP and PEF systems per kg of juice was about 26 and 24 times respectively,
87 that consumed by the thermal system (extra energy from natural gas was also
88 consumed by the thermal system). Overall, there was a 7-8-fold increase in CO₂
89 emissions by the HPP and PEF systems compared to the thermal process. A
90 simulation-based sensitivity analysis of total production costs with respect to
91 equipment scale, showed that cost per unit reduced with scale. Details of the
92 simulation method were not provided. Rodriguez-Gonzalez et al. (2015) compared the
93 energy requirements of HPP, PEF, membrane filtration MF, ultraviolet UV radiation
94 and conventional HTST thermal pasteurisation using published information for the
95 inactivation of *Escherichia coli* in apple juice. They concluded that MF and UV have
96 the potential to consume less energy per unit mass, than HTST, PEF and HPP. Milani
97 et al. (2016) compared HPP and thermosonication, with conventional thermal
98 pasteurisation in terms of energy requirements and inactivation of *Saccharomyces*
99 *cerevisiae* ascospores. HPP was found to consume the least specific energy, followed
100 by thermosonication for the same level of inactivation under the experimental
101 conditions. For the same processing time, HPP also showed the highest level of

102 inactivation, followed by thermosonication. Aganovic et al. (2017) conducted
103 comparative energy and life cycle assessments of thermal, HPP and PEF
104 pasteurisation technologies for the preservation of tomato and watermelon juices, at
105 gate-to-gate (processing) and farm-to-gate (preparation, processing and waste
106 treatment) levels. Conventional thermal processing with a plate heat exchanger was
107 found to consume the least specific energy, followed by the PEF and HPP
108 technologies, respectively. There was no significant difference in the specific energy
109 consumption of each technology, for both products.

110

111 As most of the studies concerning **innovative** technologies in food processing have
112 been performed at lab-scale, the results obtained are difficult to scale-up and cannot
113 be generalized. Where scale-up studies have been done, few details have been
114 provided. The broad variety of equipment used, and the large range of different
115 processes, products, and recipes complicate comparison of energy use. Energy
116 consumption data of single production sites are rarely available due to nondisclosure,
117 and the same is true for single unit operations. In most studies, many of the results
118 are based on energy calculations, inferred from product temperature measurements.
119 For instance, in Rodriguez-Gonzalez (2015), Milani et al. (2016) and Aganovic et al.
120 (2017), the energy consumption of conventional thermal processes was calculated
121 based on product temperature measurements, rather than direct energy
122 measurements. Moreover, **only a few** studies have compared **innovative** and
123 conventional food preservation technologies on the basis of primary energy resource
124 efficiency and GHG emissions of the pasteurisation step.

125 The purpose of the current work is to measure and compare the energy performances
126 of HPP, microwave volumetric heating (MVH), ohmic heating (OH) and conventional

127 thermal treatment (UHT), whilst delivering equivalent microbiological lethality in
128 orange juice, under commercially-representative processing conditions. The GHG
129 emissions corresponding to the energy consumed across the technologies are
130 evaluated using the UK electricity grid and other energy system emissions data,
131 collected over many years while the effect of equipment scale is explored empirically.

132

133 **2. Materials and Methods**

134

135 A series of trials were conducted using continuous flow microwave processing,
136 conventional heat treatment, high pressure processing and ohmic heating. For the first
137 three processes, trials were conducted using orange juice produced at Campden BRI.
138 Fresh oranges of good and uniform quality (total mass, 300 kg) were purchased from
139 a local supplier (Drinkwater, Chipping Campden), delivered in 15 kg boxes the day
140 before processing, transferred into open crates and chill-stored at 5 °C. The oranges
141 were washed with water and juice, extracted using an FMC citrus reamer (designed
142 to mimic industrial extraction practices). The juice was collected in 10 L stainless steel
143 buckets, immediately wrapped in cling-film and covered in black bags. For the ohmic
144 heating process, fresh oranges were purchased from another supplier, the juices
145 extracted using a compact juicer (Philips HR1832/01) and processed immediately at
146 Brunel University London. In all cases, electrical energy consumption was measured
147 using energy meters (details in Section 2.5). Thermal energy delivery to the food was
148 determined from temperature measurements, mass measurements for batch
149 processes, flowrate measurements for continuous processes and thermophysical data
150 (e.g. density and specific heat capacity) from the literature.

151

152

2.1. Continuous microwave heating

Continuous microwave processing was conducted using a Dynowave – AMT 4 system (Advanced Microwave Technologies, Scotland), with a four-magnetron power source. As shown in the basic schematic of the microwave pasteurisation system (Fig. 1a), the orange juice flows through the microwave chamber, then through a holding tube, before being cooled via heat exchange with cold water. The 12-kW capacity unit uses microwave energy to quickly heat up a range of pumpable, non-flammable food products and biotechnical components. Cold juice (20 litres) at approximately 14°C, was added to the equipment feed tank, whilst external heat, using a water bath, was provided to the heat exchanger jacket (tubular heat exchanger), to keep the energy supplied stable and thus, minimise the temperature drop within the holding tube.

A progressive cavity pump (Seepex, UK) was used at 115 ± 5 L/h, for a target process of 75 °C for 26 s. The holding tube dimensions were calculated to be 22 mm diameter and 2.4 m length which yielded a process equivalent to 70 °C for 2 mins (based on the average flow rate with $z = 7.5$ C°). The come up time (CUT) was between 22 and 25 s. The inlet and outlet temperatures of the holding tube were continuously monitored using calibrated sensors. If the target temperature was not achieved, the product was manually diverted to the drain. Once the set initial temperature was reached, the liquid food was directed to the heat exchanger. The process was repeated three times.

Fig. 1

2.2. Ohmic heating

Here, 250 ml orange juice was fed into a 10-kW batch ohmic heater (CTech Innovation, Capenhurst, Chester, UK) operating at atmospheric pressure. The heater comprises:

178 • A polypropylene product container into which the juice was filled (with an electrode
179 at each end), located in a banded tray with an interlocked cover.

180 • A free-standing control panel (housing the power supply unit with a proportional-
181 integral-derivative PID controller for voltage control), attached to the heater.

182 The polypropylene product container internally measures 90mm wide x 95mm high,
183 with a variable length between 80mm and 300mm, adjustable between the two end
184 electrode housings via 80mm- and 220mm-long spacer sections, fitted with tie rods.

185 The maximum operating voltage between the electrodes is the mains supply voltage
186 (~240V). This electrode voltage is controllable via the PID controller between 0 and
187 maximum voltage to achieve desired product temperatures. In this work, only the
188 80mm option, corresponding to a maximum voltage gradient of ~30V/cm, is used. The
189 orange juice was heated from ambient to 76°C, with a holding time of 26s, similar to
190 the microwave system. The come-up time was in the range 50-70s. Electrical energy
191 consumption data was collected at two levels: from the mains (which determines the
192 total energy costs), and between the electrodes, via the control panel, by measuring
193 the electrode terminal voltages with a voltmeter, and the current through it, using a
194 clamp-on ammeter. This approach enables the actual electrode control voltage signal
195 and hence, voltage gradient dynamics to be determined. Fig. 1 b shows a schematic
196 of the ohmic heating system with the electrical energy instrumentation connections.

197

198 **2.3. HPP**

199 Energy use was monitored on a 700 ml laboratory-scale HPP system (EPSI, Belgium).
200 The pressure medium consisted of water with 3% (v/v) of MKU, an oil based corrosion
201 inhibitor. The pressure was recorded using an MMS3000 data logger (RiL Instruments,
202 Nottingham), logging at 1 second intervals. Fig. 1 c shows the operating principle of

203 the HPP system. The pressure medium is pumped via a pressure generator
204 (intensifier) to the already-filled pressure vessel containing the packaged juice, leading
205 to pressure build-up to a maximum of 600 MPa, which is maintained for a hold time of
206 3 mins, after which, rapid depressurisation occurs. The juice temperature during the
207 hold step is, 30 ± 2 °C and reduces to 12 ± 1 °C after depressurisation.

208

209 **2.4. UHT/HTST System**

210 Conventional thermal processing trials were conducted using an FT74XTS miniature-
211 scale UHT/HTST processing system (Armfield, UK). Fig. 1 d illustrates the principle of
212 the UHT system. Process water (PW) is heated by an electrical process heater (EH).
213 The resulting hot water exchanges heat with flowing cold orange juice. When the target
214 temperature is reached, the orange juice flows through the holding tube for the desired
215 residence time before subsequent cooling. The target process is set at 76.8°C for 15
216 seconds (equivalent of 70.0°C for 2 min based on an average flowrate of about 12 L/h
217 and $z = 7.5$ C°). The process was repeated 3 times.

218

219 **2.5. Energy measurements**

220 Electrical energy data for the HPP, MVH and UHT/HTST systems was recorded at 5
221 second intervals, using a Fluke 1730 energy logger. As each system was three-phase
222 power supply-driven, the logger monitored each phase voltage, while independently
223 monitoring the respective line currents using the induction current measuring principle
224 (clamping the jaws of the meter over each live phase conductor). For the OH process,
225 driven by a single-phase supply, phase voltage, current, power factor, power and
226 cumulative energy were each logged at 10 second intervals, via a Fluke 345 energy
227 logger. The per-phase voltage and current coil connections for all electrical energy

228 measurements are illustrated in Fig. 1 a- d. In each case as shown in the figures, the
229 current coil of the meter is in series with the phase-to-neutral circuit. This is achieved
230 by clamping the jaws of the meter around the live phase conductor L from the mains.
231 The voltage coil is connected in parallel (i.e. one terminal to the live phase conductor
232 L and the other to the neutral N). Note that for the three-phase power supply-driven
233 system, there are three live phase conductors (usually designated L1, L2 and L3) from
234 the mains, hence, three independent voltage and current coils are present in the meter
235 Fluke 1730. In all cases, the data is saved to internal memory, and the relevant Fluke
236 application software used to extract the individual components.

237

238 2.6. Comparative analysis methodology

239 2.6.1. Energy density and efficiency comparison by electrical measurements

240

241 To enable all processes under investigation to be logged using the same measuring
242 principle, the conventional technology is represented here by an electrically-powered
243 hot water-to-orange juice heat exchanger (UHT system). Energy comparisons are
244 initially made on the basis of energy density (electrical mains energy consumed per
245 litre of juice), and energy efficiency. Instantaneous energy density SPE_i for continuous
246 processes is calculated as the ratio of the instantaneous power consumption $P(t)$ to
247 the volumetric flowrate $\dot{V}(t)$

$$248 \quad SPE_i = P(t)/\dot{V}(t) \quad (1)$$

249 For processes with steady-states, the final value (SPE_f) of SPE_i is the required value.

250 For batch processes, the cumulative energy density SPE_f is the aggregated (or final)
251 mains energy consumed after processing time t_f , divided by the batch volume vol .

252

253 $SPE_f = \frac{1}{vol} \int_0^{t_f} P(t) dt$ (2)

254

255 The energy efficiency in each case is determined as the ratio of output heat energy
 256 Q_{Heat} to the input mains electrical energy Q_{Elect} . For the continuous and batch
 257 processes respectively, the instantaneous value of efficiency are calculated as

258

259 $\eta_{Elect(continuous)} = Q_{Heat}(t)/Q_{Elect}(t) = \dot{m}C(T_{out}(t) - T_{in}(t))/P(t)$ (3)

260

261 $\eta_{Elect(batch)} = Q_{Heat}(t)/Q_{Elect}(t) = mC(T(t) - T_{in})/\int_0^t P(t) dt$ (4)

262

263 where m is the hold up mass, \dot{m} , mass flowrate, C , specific heat capacity, T_{in} , inlet or
 264 initial temperature, T_{out} , outlet temperature and T , temperature within the batch vessel.

265 Equations (3 & 4) apply only to the electro-heating technologies (MVH, OH and UHT).

266 As HPP is non-thermal, evaluating the energy efficiency in an equivalent manner is
 267 not as straightforward. To facilitate energy efficiency comparison across the
 268 technologies, an equivalent HPP pasteurisation energy efficiency is proposed as

269

270 $\eta_{eq(HPP)} = Q_{Heat(T0=25)}/Q_{Elect(HPP)} = mC(T(t) - 25)/\int_0^t P(t) dt$ (5)

271

272 where, $\eta_{eq(HPP)}$ is the efficiency of the HPP system, referred to a thermal system, while
 273 $Q_{Heat(T0=25)}$ is the heat energy required to heat the orange juice from room temperature
 274 (25°C) to the pasteurising temperature of a conventional system (UHT), and $Q_{Elect(HPP)}$,
 275 the electrical energy consumed by the HPP equipment in achieving the same level of
 276 pasteurisation. P and m in (5) are based on HPP process data only.

277

278 **2.6.2. GHG emissions: innovative vs conventional gas-fired technologies**

279 Although the conventional technology (UHT) is represented by an electrically-powered
280 hot water **water-to-orange juice heat exchanger**, gas-fired hot water or steam heaters **are**
281 **most common** in industry (Masanet et al., 2008). Hence, the most effective way of
282 comparing the industrially-relevant gas-fired system with the **electrically-driven**
283 innovative technologies is to refer all energy consumption to the non-renewable
284 primary energy use. The GHG emissions corresponds to the amount of non-renewable
285 primary energy resource depletion due to processing. For comparison, this can be
286 used as an approximate measure to gauge the non-renewable primary energy
287 efficiency of the gas-fired and electrically-driven **innovative** process technologies.

288 First, the efficiency of the studied electrically-powered UHT is defined as

289

$$290 \eta_{UHT} = \eta_{Elect,J} = \eta_{Elect,HW} \eta_{HW,J} \quad (6)$$

291

292 where, $\eta_{Elect,J}$ is the efficiency of energy conversion from electricity to heat into the
293 juice, $\eta_{Elect,HW}$, the efficiency of energy conversion from electricity to hot water and
294 $\eta_{HW,J}$, the efficiency of energy conversion from hot water to heat into the juice.

295 Assuming the system is gas-fired, the efficiency $\eta_{G,J}$ (gas-to-juice) would be

296

$$297 \eta_{G,J} = \eta_{G,HW} \eta_{HW,J} = \eta_{G,HW} (\eta_{UHT} / \eta_{Elect,HW}) \quad (7)$$

298 where $\eta_{G,HW}$ is the efficiency of energy conversion from gas to hot water. By estimating
299 $\eta_{G,HW}$ and $\eta_{Elect,HW}$, $\eta_{G,J}$ can be roughly determined. $\eta_{G,HW}$, which corresponds to the
300 boiler efficiency is in the range $0.5 \leq \eta_{G,HW} \leq 0.8$ (Carbon Trust, 2012a, b), while $\eta_{Elect,HW}$,
301 the indirect resistance heating efficiency of the electrical heating element is estimated

302 as $\eta_{Elect,HW} = 0.8$ (CEATI, 2018). Based on the gas-fired system efficiency $\eta_{G,J}$, the
 303 energy density SPG_f with respect to the gas fired technology becomes

304

$$305 \quad SPG_f = \frac{1}{\eta_{G,J}} (SPE_f / \eta_{Elect}) \quad (8)$$

306

307 where the term in brackets is the specific thermal energy (kJ/L) delivered to the juice.

308

309 The GHG emissions corresponding to using the gas-fired system to supply thermal
 310 energy ${}^1Q_{Heat} = 1$ kWh to the juice is

311

$$312 \quad GHG \text{ Emissions}_G (kgCO_2e/kWh) = {}^1Q_G \zeta_G = {}^1Q_{Heat} \zeta_G / \eta_{G,J} \quad (9)$$

313

314 where $\zeta_G = 0.18416$ kgCO₂e/kWh is the 2017 GHG conversion factor (gross calorific
 315 value) for natural gas from the UK gas grid (UK GOV, 2018). Changes in the value of
 316 ζ_G over the years are minor, and hence, ζ_G is assumed constant for simplicity. 1Q_G is
 317 the gas energy in kWh that delivers ${}^1Q_{Heat} = 1$ kWh to the juice.

318

319 For the electrically-driven technologies with efficiencies η_{Elect} derived from
 320 experimental energy measurements, a similar expression may be written as

321

$$322 \quad GHG \text{ Emissions}_{Elect} (kgCO_2e/kWh) = {}^1Q_{Elect} \zeta_{Elect} = {}^1Q_{Heat} \zeta_{Elect} / \eta_{Elect} \quad (10)$$

323

324 where, ζ_{Elect} is the GHG conversion factor for the UK electricity grid. Estimates and
 325 projections of this, up until 2035 were collected from DBEIS (2018) and ICAX (2018),

326 and presented in Appendix A, Table A1. ${}^1Q_{Elect}$ is the electrical energy in kWh that
 327 delivers ${}^1Q_{Heat}=1$ kWh to the juice. ${}^1Q_{Heat}=1$ kWh is used as functional unit because,
 328 barring non-thermal effects, 1 kWh of heat delivered to the juice, will for the same initial
 329 conditions of the juice, achieve the same pasteurising effect across all technologies.

330

331 If 1 L of juice is chosen as functional unit instead of 1 kWh of juice-delivered thermal
 332 energy ${}^1Q_{Heat}$, the GHG emissions (kgCO₂e/L) can be determined by multiplying the
 333 respective kgCO₂e/kWh values (equations 9 & 10) by the corresponding total thermal
 334 energy density in kWh/L (i.e. $2.78e^{-4}SPE_f\eta_{Elect}$) as shown in equations (11) and (12).

335

$$336 \text{GHG Emissions}_G(\text{kgCO}_2\text{e/L}) = 2.78e^{-4}SPG_f\eta_{G,J}\text{GHG Emissions}_G(\text{kgCO}_2\text{e/kWh})$$

337 (11)

338

$$339 \frac{\text{GHG Emissions}_{Elect}(\text{kgCO}_2\text{e/L})}{(\text{kgCO}_2\text{e/kWh})} = \frac{2.78e^{-4}SPE_f\eta_{Elect}\text{GHG Emissions}_{Elect}}{(12)}$$

341

342

343 2.7. Equipment scale studies

344

345 To explore the effect of equipment scale on the specific energy consumption of the
 346 innovative technologies, some trials at different scales were performed, where
 347 possible (MVH and OH), while for the other case (HPP), experimental results at
 348 different scales from published works, and the current study were correlated.

349

350

351 **2.7.1. MVH System**

352

353 For the MVH system, trials were conducted on the same equipment (of cylindrical
354 cavity diameter 34mm, and length 550mm) using water at different flowrates (90-
355 240L/h), below and above that used in the original orange juice experiment (115L/h).
356 The energy input was scaled by different magnetron switching scenarios. Each of the
357 four magnetrons of the microwave heating system is switchable independently to
358 deliver 3kW of power at full-load. Hence, system performance for 3, 6, 9 and 12kW
359 microwave power delivery was evaluated, under the different product flow conditions.

360

361 **2.7.2. OH System**

362

363 For the batch OH system, the maximum voltage gradient was maintained at ~30V/cm
364 as the source voltage cannot exceed the mains value (~240V) and the electrode
365 separation was held constant at the minimum possible value (~8cm). Scaling was
366 achieved by changing the volume of orange juice from 250ml, through 375ml to 500ml.
367 For the batch ohmic heater with a rectangular section, at constant electrode separation
368 L , and constant plate length, W , this was achieved by varying the product level, x .

369

370 **2.7.3. HPP System**

371

372 For the HPP system, the power consumption-pressure characteristics of two systems
373 at 35L and 55L capacity from previously published studies (Rodriguez-Gonzalez et
374 al., 2015; Aganovic et al., 2017) are compared with that of the current study (a 700ml
375 system). The 55L system (Wave 6000/55 Hiperbaric, Burgos, Spain) operates at a

376 maximum pressure of 600 MPa, compression time of 3.5 min and holding time of 5
377 min. The 35L system (AVURE Technologies, Kent, WA, U.S.A.) has a maximum
378 pressure of 600 MPa, a compression time of 2 min and a holding time of 2 min.

379

380 **3. RESULTS AND DISCUSSION**

381 **3.1. Energy consumption analysis**

382 **3.1.1. MVH System**

383

384 The MVH results are shown in Fig. 2. The power consumption characteristics is
385 discontinuous as several subsections of the equipment (magnetron cooling, cabinet
386 cooling, stirrer operation and microwave power delivery) are controlled using on/off
387 control loops. Ordinarily, the MVH power consumption data would have been
388 considered once the product temperature had reached steady state (Fig. 2). However,
389 since the on-and-off switching times of the magnetron cooling system is somewhat
390 random, due to the variable initial state (temperature) of the magnetron within each
391 experimental period, the overall mains power consumption has no steady-state.
392 Further tests were performed to isolate the random magnetron cooling from the overall
393 energy consumption, and show that the magnetron cooling energy is $\sim 1.8\text{kW}$. Hence,
394 the steady-state MVH power consumption is evaluated as the sum of the minimum
395 value (with no magnetron cooling) and the time-averaged magnetron cooling energy.

396

397 Analysing the power consumption characteristics (Fig. 2 b), the maxima occurring in
398 the time region $0 \leq t < 8$, where t is in minutes, is most likely due to a combination of
399 two factors: the low product inlet temperature and high product outlet temperature
400 overshoot (maximum outlet temperature $\sim 85\text{-}95^\circ\text{C}$ as against the required 76°C) in

401 that region (Fig. 2 a). This maxima disappears in the region $8 \leq t < 20$ when the product
402 outlet temperature stabilises at the set-point. The minima in the region $11 \leq t < 13$ is
403 most likely due to the random magnetron cooling energy switch-off. When the energy-
404 density and efficiency (Fig. 2 c & d) are evaluated based on the time-averaged steady-
405 state values, the final energy density is 380kJ for 1 litre of orange juice, while the
406 energy efficiency is 45%. When the magnetron cooling energy is recovered, the
407 energy density reduces to 325 kJ/L, while the efficiency becomes 54%. It should be
408 noted that the specific energy consumption (and hence, energy density) depends
409 significantly on the product inlet temperature, which could be subject to environmental
410 variations (Fig. 2 a). A high product inlet temperature means less energy would be
411 spent to achieve the desired outlet temperature. For energy efficiency, these effects
412 tend to cancel out, making it a better index for comparing the performances of different
413 thermal technologies irrespective of inlet temperature variations.

414

415 **3.1.2. OH System**

416

417 Fig. 3 shows the various energy performance indicators of the OH system. As a batch
418 system, the comparison with continuous systems is based on the total energy use per
419 cycle of orange juice production. A cycle consists first of a transient stage, where the
420 juice is heated from its initial temperature to the required set-point of 76°C. Thereafter,
421 there is a steady-state stage, where the orange juice is maintained at this set-point for
422 26 seconds (the residence time in the holding tube of the microwave system). From
423 the experimental results the transient period is about 70s (Fig. 3 a), while the steady-
424 state period is shown (for clarity), extended to make a total time of 5 minutes. The
425 energy consumed in pasteurising a litre of orange juice is 208 kJ, while the cumulative

426 energy efficiency is 80%. At the beginning of the process, the energy efficiency is
427 observed to be 97%, rising to 99% towards the end of the transient stage (at 50s), the
428 onset of voltage control switching. The energy efficiency drops to 80% in the short
429 period of voltage control (50s to 70s), at the end of which, power supply to the
430 electrodes is switched off and the temperature maintained for 26s (during which time,
431 no heat losses are observed). An important implication of the voltage switching as
432 observed in Fig. 3 b is that while the mains voltage V_1 remains constant throughout
433 the process (as it should be, being independent of loading conditions I_1 and I_2), the
434 control voltage V_2 delivered to the electrode drops sharply. Since the currents, $I_1=I_2$
435 essentially, the power ($V_2 \times I_2 \times \cos \phi_2$), utilised for juice heating falls more sharply
436 than the power ($V_1 \times I_1 \times \cos \phi_1$) drawn from the mains. Therefore, the PID controller-
437 based voltage switching control wastes energy in much the same way as a throttling
438 valve does for fluid flow through pipes. This situation can be remedied in continuous-
439 flow ohmic heaters where product outlet temperature can be fixed for a constant
440 voltage gradient, by setting the juice flowrate and electrode length W , to achieve the
441 desired product outlet temperature in the resulting residence time. Unsteady
442 environmental conditions can then be compensated for by flow control. This way,
443 overall energy efficiencies above 95% are anticipated for the process. Note that the
444 entire ohmic system including the power supply system elements has a lagging power
445 factor $\cos \phi_1$ (Appendix A, Fig. A1), while within the ohmic cell bounded by electrodes,
446 the power factor $\cos \phi_2 \approx 1$ (as the food is essentially a pure resistor).

447

448

Fig. 3.

449

450

451 **3.1.3. HPP System**

452

453 Fig. 4 (a) shows the pressure-time and power-time characteristics of the HPP system.
454 By dynamically integrating the power curve, with respect to time, and dividing by the
455 hold-up volume of orange juice, the cumulative energy consumption per litre (energy
456 density) was obtained as shown in Fig. 4 (b). It is observed that to achieve the required
457 pasteurisation, 645 kJ of energy is required per L of orange juice. The seal of the HPP
458 equipment was less effective than desired. Short peaks of power were required to
459 maintain the holding pressure in the equipment due to small leaks around the seal as
460 can be observed in Fig. 4a. This leads to energy efficiency reductions.

461

462 **Fig. 4.**

463

464 **Fig. 5.**

465

466

467 **3.1.4. UHT/HTST System**

468

469 The UHT system of section 2.4, whose energy performance indicators are shown in
470 Fig. 5, has an energy density of 470 kJ/L, with an energy efficiency of 46%. As in all
471 continuous processes, whose input and output energies reach steady-states, the
472 steady-state values are considered. The abnormally high energy efficiency (about
473 200%) recorded in the transient period (Fig. 5d), occurs due the build-up of heat in the
474 indirect resistance heating element used in transferring electrically-generated heat to
475 the hot water. This coincides with the first temperature maxima in Fig. 5a. Hence,
476 although the control system reduces the electrical power input considerably after the

477 first maxima in Fig. 5b, the outlet temperature remains high, ensuring the output
478 thermal energy exceeds the input electrical energy for a short period. It is important to
479 note the very long transient period (~ 2 hours), which occurs due to the low heat
480 transfer rates.

481

482 **3.2. Energy consumption comparison**

483

484 As shown in Table 1, the batch OH process has the highest energy efficiency (80%)
485 of all the thermal processes, in spite of the energy lost by control voltage reductions.
486 It also has the least specific energy consumption. A continuous OH process, without
487 voltage switching temperature control would be expected to have the efficiency ($\eta >$
488 95%) before the onset of voltage switching is expected. The MVH and UHT/HTST
489 processes have comparable energy efficiencies (45 & 46%, respectively), when the
490 magnetron cooling is powered electrically from the cabinet. If this cooling energy is
491 discounted, the MVH system energy efficiency could rise to as high as 54%. The
492 specific energy consumption of the MVH system is considerably less than that of the
493 UHT/HTST system, counter-intuitively, due to the much higher temperature of the
494 orange juice fed into the MVH system (compare the inlet temperature values of Figs
495 2a & 5a). Hence, the MVH system requires much less energy to get one litre of orange
496 juice to a similar temperature set-point, even though the energy efficiencies are
497 approximately equal. The HPP system has the worst energy performance. One reason
498 for this is the low vessel filling ratio (36%) used. Previous studies (Atuonwu and
499 Tassou, 2018c; Rodriguez-Gonzalez et al., 2015) show that specific energy
500 consumption reduces significantly with vessel filling ratio. Using correlations from the
501 power-vessel filling ratio graphical relations of Rodriguez-Gonzalez et al. (2015) for

502 extrapolation, the efficiency of the HPP system is found to improve from its initial value
503 of 31% (at 36% filling ratio) to 51% at 60% filling ratio, and 78% at 85% filling ratio.
504 There also exists opportunities to improve HPP energy performance by recovering the
505 decompression via synchronised twin, semi-continuous systems (Toepfl et al., 2006).

506

507 Due to the low heat transfer rates between the hot water and juice in the UHT/HTST
508 system, its start-up performance is very poor, as seen in the long transient time. Over
509 this time (2 hours) of off-spec production, a total of 12.72MJ of energy is consumed.
510 Hence, for processes with frequent shut-downs and start-ups, this would be a major
511 problem. The HPP, MVH and OH processes do not have this problem as electric
512 switching has an almost instantaneous effect on thermal energy generation. When the
513 electrically-powered HTST system (UHT) is converted to an equivalent conventional
514 gas-fired system using equations (2) and (3), the energy efficiency is 29% for a 50%
515 boiler efficiency (UHT1), and 46% for an 80% boiler efficiency (UHT2). As the energy
516 efficiencies of UHT1 and UHT2 are with respect to gas, versus electricity for the other
517 technologies, a simple energy efficiency comparison will not suffice. Hence, in the next
518 section, GHG emissions are used to refer the energy performances of all the
519 technologies to the same basis: non-renewable primary energy resource use.

520

521 **3.3. Comparison of non-renewable primary energy use via GHG emissions**

522

523 Fig. 6 shows the GHG emissions (per kWh of thermal energy or equivalent) of all the
524 studied electrically-driven innovative and gas-fired technologies, based on their
525 various energy performances, over a 25-year period (2010-2035). With the rapid
526 decarbonisation of the UK electricity grid over time, the innovative technologies

527 become much more non-renewable primary energy resource-efficient and hence,
528 more environmentally-friendly. The hypothetical continuous OH process at 95%
529 efficiency is seen to have the best GHG performance, followed by the batch OH
530 process, which is then closely-followed by the HPP system at 95% filling ratio, and
531 then, the MVH1 system (MVH with recovered cooling energy). These are all currently
532 (2018 data), more non-renewable resource-efficient than the best gas-fired system.
533 The 60%-filling ratio HPP system, closely follows suit and is currently just as GHG-
534 efficient as the best gas-fired system. Using current projections, the improvements
535 attained by the innovative technologies are expected to be more significant with time.
536 By 2027, even the 36%-filling ratio HPP will become more energy resource-efficient
537 than the most-efficient gas-fired system, provided gas boilers do not become
538 significantly more efficient than 80%. It is important to note that the significant transient
539 energy losses (about 12.72 MJ) of the gas-fired system is ignored in this analysis. If
540 considered, the improvement due the innovative technologies becomes even more
541 significant. Similar conclusions as the foregoing can be reached by examining Table
542 2 where GHG emissions (in kgCO_{2e}/L of juice product), calculated for the years 2016-
543 2020 are presented. Clearly, the investigated innovative technologies are very
544 promising investments in terms of primary energy resource efficiency. Actual
545 investment decisions would however also consider capital costs, the applicability of
546 the specific technology to the product being processed, in-pack processing vs aseptic
547 filling-only possibilities, water consumption, packaging issues, amongst other factors.
548 The HPP process studied in this work (at 36% fill ratio) is a post-packaging process.
549 Hence, there is virtually no fouling of the processing vessel and consequent waste
550 water from cleaning. Moreover, pressurised water can be reused, thus, increasing
551 overall sustainability. The packaging however, must be capable of maintaining its

552 shape and form after the pressurisation and depressurisation processes. The
553 continuous-flow MVH, UHT and batch OH processes are pre-packaging processes
554 and so lead to fouling of hot surfaces over time, and subsequent cleaning operations
555 which generate wastewater. Moreover, they require aseptic filling with the attendant
556 energy and environmental costs. The same applies to the very high fill ratio (95%)
557 HPP, which is virtually unattainable in a post-packaging process. For the high-power
558 (12 kW) MVH system, the magnetrons require water cooling, which could constitute
559 waste water, or be continuously reused so its energy is recovered to preheat the juice.
560 All these aspects as well as factors such as shelf life, will figure in an overall
561 environmental analysis. The current study however focuses only on the energy
562 required for the actual pasteurisation process step and the associated GHG
563 emissions.

564 **Table 1**

565 **Table 2**

566 **Fig. 6.**

567

568 **3.4. Results of equipment scale studies**

569

570 Figs 7 – 9 show the magnitudes of the various energy performance indicators of the
571 MVH, OH and HPP processes, respectively, at different operating scales. For the MVH
572 system, it is observed (Fig. 7a) that irrespective of the flowrate, the electrical power
573 consumption is essentially a linear function of the number of magnetrons, switched on.
574 The heat delivery rate is a linear function of the number of switched-on magnetrons at
575 variable and constant product outlet temperature (Fig. 7b, d). The interplay of the
576 slopes and the vertical axes intercepts of the two functions makes the efficiency

577 (obtained by dividing the functions), a slowly increasing function of scale, represented
578 by the number of magnetrons (Fig. 7c). However, the overall economics of scale would
579 also consider factors other than energy. The experiments on which Fig. 7 were
580 obtained used the same equipment, with a single cavity size. A study (Wang et al.,
581 2015) with different effective cavity sizes (medium volumes), suggests that efficiency
582 increases marginally with cavity size, within limits.

583

584 For the batch OH system, the time-averaged electrical power consumption increases
585 with operating volume (Fig. 8a). The area under the power consumption-time graph is
586 observed over a large set of experiments (results not shown), to be approximately a
587 linear function of batch volume, provided the juice initial temperature is constant. The
588 heating power delivered to the juice however depends on the voltage switching regime
589 of the PID controller, which shows no consistent trend with volume over the
590 experiments conducted (Fig. 8b only shows one result from the set of experiments). A
591 consistent observation nonetheless, is the fact that prior to voltage switching, the ratio
592 of the heat delivered to the electrical energy consumed from the mains is somewhat
593 constant (~98%, see Fig. 8 b & c). The earlier in time the PID controller switching
594 occurs, the more energy is wasted leading to efficiency reductions (Fig. 8c) as the
595 corresponding area under the control voltage-time graph, which in a sense represents
596 the actual energy delivered to the fluid, drops (Fig. 8b). Over the large experimental
597 set (results not shown), this trend is found to be independent of batch volume. Hence,
598 the efficiency reduction in Fig. 8c is not due to batch volume, but to the voltage
599 switching regimes. For a continuous OH process where the temperature can be
600 controlled without voltage switching, the associated efficiency losses are not expected.
601 Hence, the efficiency is expected to remain essentially constant (~98%), irrespective

602 of scale. The energy density (or specific energy consumption in kJ/L) for the presented
603 case (Fig. 8d) is essentially constant, with volume as the areas under the power
604 consumption-time graphs (Fig. 8a) are essentially, linear functions of volume. This
605 behaviour is however only true if the juice initial temperature is the same across all the
606 experimental volumes. For the HPP system (Fig. 9), the energy consumed per litre of
607 product (assuming a 95% fill ratio) is calculated as 0.242 MJ for the 0.7L case, 0.084
608 MJ for the 35L case, and 0.095 MJ for the 55L case (without leakage) and 0.45 MJ
609 for the 55L case (with leakages considered). It therefore appears that larger-scale
610 systems have significantly better energy performances than small, lab-scale systems.
611 Many other factors contribute to the energy performance of an actual machine. These
612 include the equipment make (different for each studied case) – this significantly affects
613 start-up and shut-down-related energy indices, and machine conditions (e.g.
614 pressurising medium leakages), which greatly diminishes the performance of the 55L
615 system.

616 **Fig. 7.**

617
618 **Fig. 8.**

619
620 **Fig. 9.**

623 **3.5. Critical analysis of results with respect to previously published works**

624
625 **Previously published energy and sustainability studies on innovative mild food**
626 **preservation techniques (as investigated here), present conflicting results. Milani et al.**

627 (2016) report a HPP energy density of 77 kJ/L, lower than the values (645, 392 and
628 256 kJ/L) reported here. However, unlike the current study, the results of Milani et al.
629 (2016) were based purely on calculations of compressive work (based on pressure
630 measurements) and initial sensible heating (based on temperature measurements).
631 This underestimates the energy consumption as it does not consider many other
632 factors which also contribute to the energy consumption of typical HPP processes.
633 These include product loading in vessel, overall system movement from loading to
634 working positions and initial vessel filling with water (Atuonwu and Tassou, 2018c).
635 Others are, closing the HPP plug, moving a wedge to secure the plug, maintaining the
636 target pressure for the required time, where there are leakages and product unloading.
637 Furthermore, in Milani et al. (2016), the maximum operating pressure was 300 MPa in
638 27 s, as against the 600 MPa in 3 mins used here, and Atuonwu and Tassou (2018c)
639 show that energy consumption increases more than proportionately with operating
640 pressure. Using similar methods to Milani et al., Sulaiman (2015) realised an energy
641 density of 240 kJ/kg for strawberry puree processing at 600 MPa, 48 °C for 5 mins,
642 while Rodriguez-Gonzalez et al. (2015) recorded 338.1 kJ/kg, for processing apple
643 juice at 350 MPa, 40 °C and 482.8 kJ/kg at 500 MPa, 42 °C, both for 5 mins.
644 Processing details such as equipment volume and fill ratio were not stated, neither
645 was there any information on contributions of leakages, loading and other energy-
646 contributing factors. Aganovic et al. (2017), perhaps the only previous study in the
647 literature based on energy measurements, reports a much higher value (720 kJ/kg for
648 HPP at 600 MPa, 5 mins), for a 55 L system. This is most likely due to the low fill ratio
649 used (36%) and the significant energy losses during the hold period (see Fig. 9 c & f),
650 which could be attributed to leakages. Overall, the results presented in the current

651 study are within the range of all the reported previous literature values, while providing
652 more energy measurement, equipment parameter and scale details.

653

654 For conventional thermal treatment, Rodriguez-Gonzalez et al. (2015), using
655 calculations based on temperature measurement data from other studies, reported
656 specific energies in the range 166.9 - 228.5 kJ/kg, while Aganovic et al. (2017)
657 reported a value of 144 kJ/L. These again are underestimated values as in both cases,
658 the energy consumed was calculated as $mC\Delta T$ (see numerator of equations 3 & 4).

659 This is the energy delivered to the product which is usually much lower than the actual
660 energy consumed. Although Aganovic et al. (2017) adds a cooling energy arbitrarily
661 chosen as 33% of $mC\Delta T$ (which would increase the total energy value), they assume
662 70% heat recovery (which further reduces the total energy). It is therefore not
663 surprising that such low values of specific energy are calculated. The results of our
664 study (470 kJ/L) are however based on actual energy measurements for an electricity-
665 driven HTST system. When it is driven by a gas-fired boiler at 50% efficiency, the
666 energy density is 746 kJ/L (kJ of gas energy per litre of juice). From the foregoing
667 analysis, the method used in our study is a better representation of reality.

668

669 The MVH energy efficiency results presented in this work (45 – 54%) are well within
670 the range of values reported for microwave heating of different liquids at different batch
671 volumes (Wang et al., 2015). Other studies (e.g. Atuonwu and Tassou, 2018a) suggest
672 that possibilities exist, using improved power supplies, to achieve higher efficiencies
673 by microwave power amplitude, frequency and phase optimisation. For OH systems,
674 efficiencies (also termed “system performance coefficients”) in the range 57–86% were
675 reported for different voltage gradients in the range 20–60 V/cm, with the higher

676 efficiencies corresponding to the lower voltage gradients in orange juice (Icier and
677 Illicali, 2005). Similar results were reported for apple juice by Park et al. (2017) as long
678 as the sugar concentration was in the range 18–48 °Brix. Darvishi et al. (2015)
679 reported an efficiency range of 55–100%, but for tomato juice evaporation in the
680 voltage gradient range 6 to 16 V/cm. Our results are within the aforementioned ranges.
681 It should be noted that in all these ohmic heating studies, there was no temperature
682 control. The product temperature increased linearly with time, essentially, until the
683 process end point, where the system was switched off. In the tomato juice evaporation
684 process of Darvishi et al. (2015), the temperature was eventually limited at the boiling
685 point of the juice. In our current work however, temperature control was implemented
686 as it was essential to maintain a non-boiling temperature for a desired pasteurisation
687 time in a batch OH process. Our work demonstrates empirically for the first time, that
688 voltage modulation for temperature control in a batch OH system is an energy-wasting
689 exercise and should be avoided where possible. For this purpose, we recommend the
690 use of continuous-flow processes, where temperature control can simply be achieved
691 by appropriate selection of fixed voltage gradient and product residence times,
692 followed by a properly-insulated holding stage.

693

694 **4. Conclusions**

695

696 In this work, we have measured and compared the energy performances of innovative
697 technologies (high pressure processing HPP, continuous-flow microwave volumetric
698 heating MVH, ohmic heating OH) and conventional heat treatment UHT, whilst
699 achieving the same pasteurising effect in orange juice, under commercially-
700 representative processing conditions. We have also evaluated the sustainability of

701 each of these processes with respect to non-renewable energy resource use and
702 explored the effect of equipment scale using model-based and empirical approaches.
703
704 For similar product quality attributes, the OH process is observed to have the best
705 energy performance (80% efficiency) among all the processes. However, for the first
706 time, it is shown empirically for the OH process that voltage modulation, as commonly
707 used in batch operations for temperature control, wastes significant energy, in much
708 the same way as valve throttling does for fluid systems in pipes. It should therefore be
709 discouraged. For a continuous OH process which permits temperature control without
710 voltage switching, efficiencies higher than 95% may be obtainable. The MVH and
711 electrically-powered heat exchanger-based UHT processes have comparable energy
712 efficiencies (45 & 46%, respectively), when the MVH magnetron is cooled electrically.
713 If this cooling energy is supplied from available cooling water sources, the MVH
714 energy efficiency could rise to 54%. All the innovative technologies show significantly-
715 better start-up characteristics than the conventional heat exchanger-based process,
716 UHT, thus saving significant amounts of energy.

717

718 For a gas-fired system with equivalent heat exchanger behaviour as the UHT process,
719 the efficiency relative to primary energy is 29% (at 50% boiler efficiency), and 46% (at
720 80% boiler efficiency). Among the innovative technologies, the HPP system (with a
721 filling ratio of 36%) has the least energy efficiency 31%. By increasing the filling ratio
722 to 60% and 95% respectively, efficiencies of 51% and 78% are attained. In terms of
723 non-renewable primary energy usage and hence climate change (evaluated by the
724 greenhouse gas GHG emissions corresponding to equivalent pasteurising effects),
725 the OH, 95%-fill ratio HPP and microwave with magnetron heat recovery processes
726 are currently better than the best gas-fired systems, based on the UK electricity grid

727 conditions. With the current grid decarbonisation trend, all the electricity-driven
728 innovative technologies are anticipated within the next few years to far outperform the
729 best gas-based systems, thus justifying investment in these technologies. Actual
730 investment decisions would however also depend on the applicability of the specific
731 technology to the product being processed, in-pack processing versus aseptic filling-
732 only possibilities, capital costs amongst other factors. Energy performance improves
733 with equipment scale for the microwave and HPP systems, but remains essentially
734 constant for the ohmic heating system. In all these, no significant differences are
735 observed in the critical quality attributes of the raw and pasteurised orange juice
736 across all the technologies. The electricity-driven innovative technologies are thus
737 promising in terms of energy savings, environmental friendliness and product quality.

738

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740

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747

748

749

750

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752 **References**

753

754 Aganovic, K., Smetana, S., Grauwet, T., Stefan, T., Mathys, A., Loey, A. van., Heinz,
755 V. (2017). Pilot-scale thermal and alternative pasteurisation of tomato and
756 water melon juice: on energy comparison and life cycle assessment. *Journal of*
757 *Cleaner Production* 141, 514-525.

758 Atuonwu, J.C., Tassou, S.A. (2018a). Quality assurance in microwave food processing
759 and the enabling potentials of solid-state power delivery. *Journal of Food*
760 *Engineering* 234, 1-15.

761 Atuonwu, J.C., Tassou, S.A. (2018b). Energy issues in microwave food processing: a
762 review of developments and the enabling potentials of solid-state power
763 delivery. *Critical Reviews in Food Science and Nutrition*
764 <https://doi.org/10.1080/10408398.2017.1408564>

765 Atuonwu, J.C., Tassou, S.A. (2018c). Model-based energy performance analysis of
766 high pressure processing systems. *Innovative Food Science and Emerging*
767 *Technologies* 47, 214-224.

768 Barba, F.J., Koubaa, M., Prado-Silva, L. do., Orlie, V., Sant'Ana, A. de Souza. (2017).
769 Mild processing applied to the inactivation of the main foodborne bacterial
770 pathogens: A review. *Trends in Food Science and Technology* 66, 20-35.

771 Barba, F.J., Mariutti, L.R.B., Bragagnolo, N., Mercadante, A.Z., Barbosa-Canovas,
772 G.V., Orlie, V. (2017). Bioaccessibility of bioactive compounds from fruits and
773 vegetables after thermal and nonthermal processing. *Trends in Food Science*
774 *and Technology* 67, 195-206.

775 Carbon Trust (2018a). Low temperature hot water boilers, introducing energy saving
776 opportunities for business, p. 11.

777 Carbon Trust (2018b). Steam and high temperature hot water boilers, introducing
778 energy saving opportunities for business, p. 8.

779 CEATI (2018). Electrotechnologies, Energy efficiency reference guide for small to
780 medium industries. Accessed 28/05/2018 from
781 https://www.ceati.com/freepublications/7020_guide_web.pdf

782 Cokgezme, O.F., Sabanci, S., Cevik, M., Yildiz, H., Icier, F. (2017). Performance
783 analyses of evaporation of pomegranate juice in ohmic heating assisted
784 vacuum system. Journal of Food Engineering 207, 1-9.

785 Darvishi, H., Hosainpour, A., Nargesi, F., Fadavi, A. (2015). Exergy and energy
786 analyses of liquid food in an Ohmic heating process: A case study of tomato
787 production. Innovative Food Science and Emerging Technologies 31, 73-82.

788 DBEIS (2018). Collection, Provisional UK greenhouse gas emissions national
789 statistics, UK Department of Business, Energy and Industrial Strategy.
790 [https://www.gov.uk/government/collections/provisional-uk-greenhouse-gas-](https://www.gov.uk/government/collections/provisional-uk-greenhouse-gas-emissions-national-statistics#2018)
791 [emissions-national-statistics#2018](https://www.gov.uk/government/collections/provisional-uk-greenhouse-gas-emissions-national-statistics#2018), accessed 27/08/2018.

792 ICAX (2018). Grid carbon factors, Inter-seasonal Collection and Exchange.
793 https://www.icax.co.uk/Grid_Carbon_Factors.html, accessed 27/08/2018.

794 Icier, F., Ilicali, C. (2005). The effects of concentration on electrical conductivity of
795 orange juice concentrates during ohmic heating. European Food Resource
796 Technology 220,406-414.

797 Kurják, Z., Barhács, A., Beke, J. (2012). Energetic analysis of drying biological
798 materials with high moisture content by using microwave energy. Drying
799 Technology 30(3), 312-319.

800 Langelaan, H.C., Pereira da Silva, F., Thoden van Velzen, U., Broeze, J., Matser,
801 A.M., Vollebregt, M., Schroen, K. (2013). Technological options for feeding 10

802 billion people. Options for sustainable food processing. State-of-the-art report
803 K STOA 2013/122.

804 Lung, R., Masanet, E., McKane, A. (2006). The role of emerging technologies in
805 improving energy efficiency: Examples from the food processing industry. In
806 Proceedings of the Industrial Energy Technologies Conference, New Orleans,
807 Louisiana.

808 Masanet, E., Worrell, E., Graus, W., Galitsky, C. (2008). Energy efficiency
809 improvement and cost saving opportunities for the fruit and vegetable
810 processing industry, an Energy Star® guide for energy and plant managers.
811 Ernest Orlando Lawrence Berkeley National Laboratory LBNL-59289-Revision,
812 p. 18.

813 Milani, E.A., Ramsey, J.G., Silva, V.M. (2016). High pressure processing and
814 thermosonication of beer: Comparing the energy requirements and
815 *Saccharomyces cerevisiae* ascospores inactivation with thermal processing
816 and modelling. Journal of Food Engineering 181, 35-41.

817 Park, I., Ha, J., Kang, D. (2017). Investigation of optimum ohmic heating conditions for
818 inactivation of *Escherichia coli*O157:H7, *Salmonella enterica* serovar
819 Typhimurium, and *Listeria monocytogenes* in apple juice. BMC Microbiology
820 17:117 DOI 10.1186/s12866-017-1029-z.

821 Pereira, R.N., Vicente, A.A. (2010). Environmental impact of novel thermal and non-
822 thermal technologies in food processing. Food Research International 43,
823 1936-1943.

824 Probst, L., Frideres, L., Pedersen, B., Amato, F. (2015). Sustainable, Safe and
825 Nutritious Food, Food processing technologies. European Union, Business

826 Innovation Observatory, Contract No. 190/PP/ENT/CIP/12/C/N03C01, pp. 1-
827 16.

828 Rodriguez-Gonzalez, O., Buckow, R., Koutchama, T., Balasubramanian, V.M. (2015).
829 Energy requirements for alternative food processing technologies – principles,
830 assumptions and evaluation of efficiency. *Comprehensive Reviews in Food*
831 *Science and Food Safety* 14, 536-554.

832 Sampedro, F., McAloon, A., Yee, W., Fan, X., Geveke, D.J. (2014). Cost analysis and
833 environmental impact of pulsed electric fields and high pressure processing in
834 comparison with thermal pasteurization. *Food and Bioprocess Technology* 7,
835 1928-1937.

836 Sulaiman, A. (2015). Non-thermal and thermal processing of fruit products to control
837 enzymatic browning. PhD Thesis, Faculty of Engineering, University of
838 Auckland, Auckland, New Zealand.

839 Toepfl, S., Mathys, A., Heinz, V., Knorr, D. (2006). Review: potential of high hydrostatic
840 pressure and pulsed electric fields for energy-efficient and environmentally-
841 friendly food processing. *Food Reviews International*, 22(4), 405-423.

842 UK GOV (2018). Greenhouse gas reporting: conversion factors 2017. Accessed
843 28/05/2018 from [https://www.gov.uk/government/publications/greenhouse-](https://www.gov.uk/government/publications/greenhouse-gas-reporting-conversion-factors-2017)
844 [gas-reporting-conversion-factors-2017](https://www.gov.uk/government/publications/greenhouse-gas-reporting-conversion-factors-2017)

845 Wang, W., Zhao, C., Sun, J., Wang, X., Zhao, X., Mao, Y., Li, X., Song, Z. (2015).
846 Quantitative measurement of energy utilization efficiency and study of influence
847 factors in typical microwave heating process. *Energy* 87, 678-685.

848

Nomenclature

<i>C</i>	Specific heat capacity of juice in OH	J/kg/K
<i>I</i>	Current	A
<i>L</i>	Electrode separation in OH	m
<i>M</i>	Hold-up mass of juice in OH	kg
<i>P</i>	Power	W, kW
<i>Q</i>	Energy	J, kJ
<i>T</i>	Juice temperature	°C
<i>T</i>	Time	s, min
<i>V</i>	Voltage	V
<i>W</i>	Electrode length in OH	m
<i>X</i>	Juice height in OH system	m

Greek letters

η	Efficiency	
ρ	Juice density in OH	kg/m ³
ζ	GHG conversion factor	kgCO ₂ e/kWh

Subscripts

<i>Elect</i>	Electricity
<i>G</i>	Natural gas
<i>Heat</i>	Heat
<i>HW</i>	Hot water
<i>J</i>	Juice

850

Appendix A

851

852

Table A1.

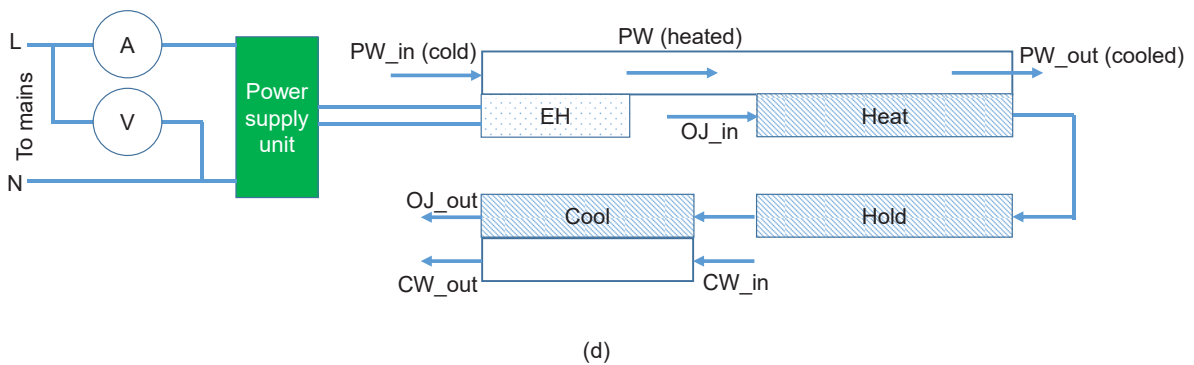
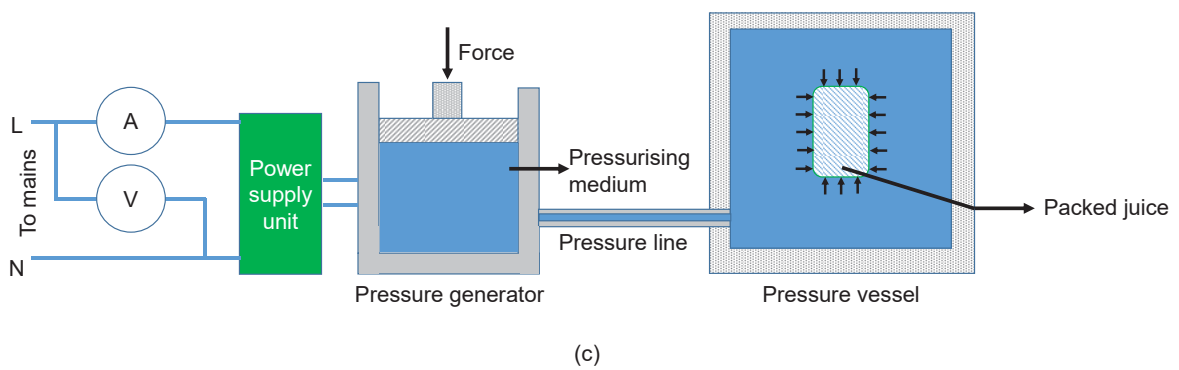
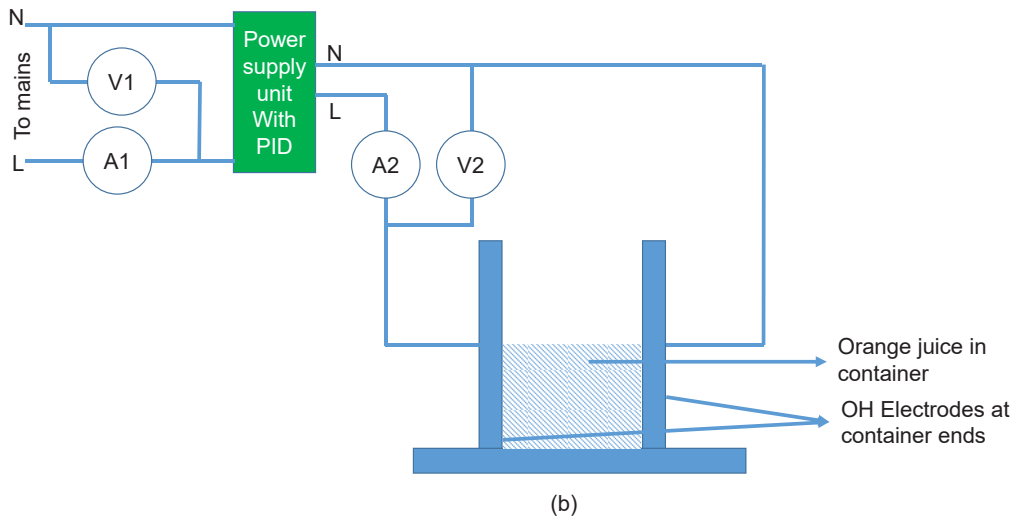
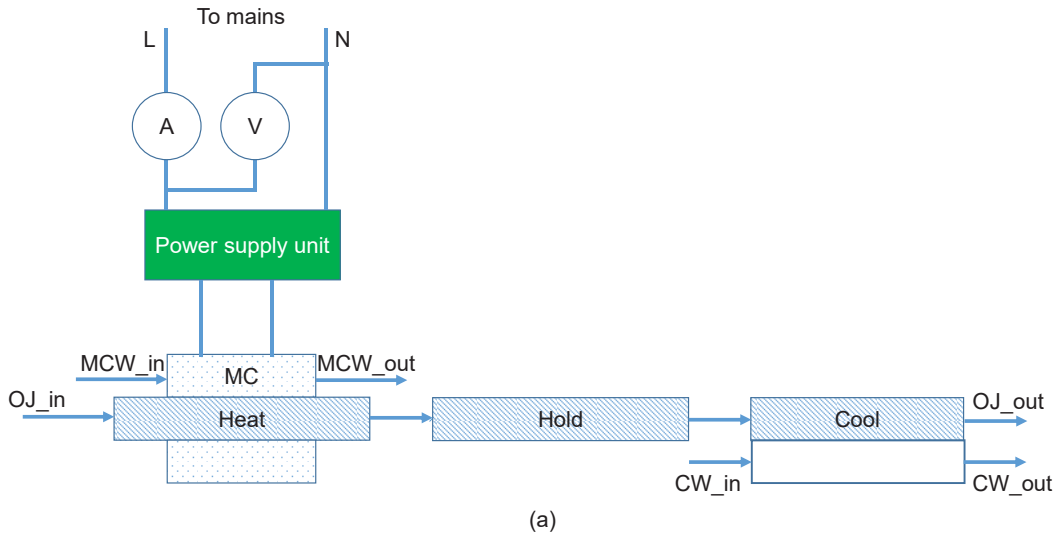
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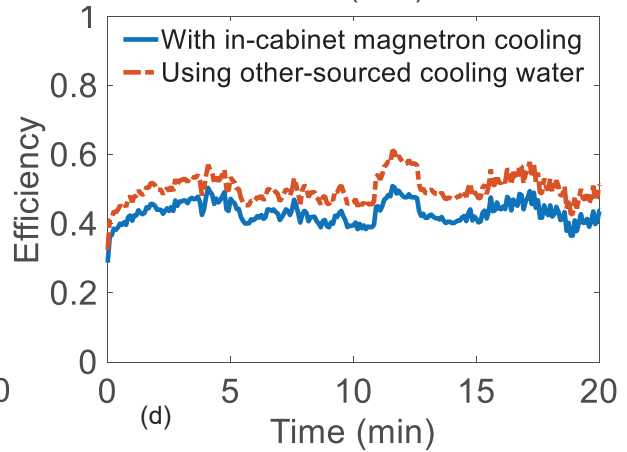
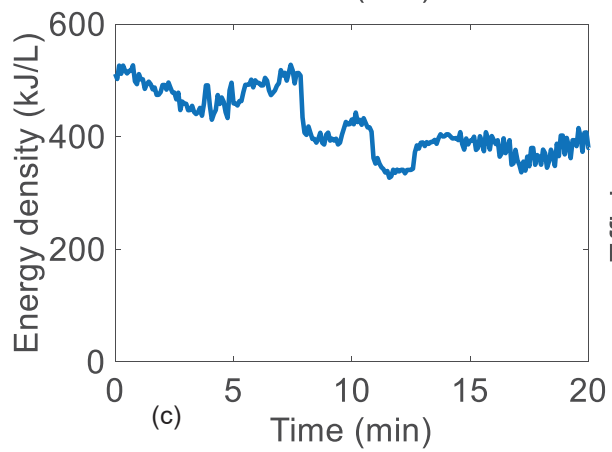
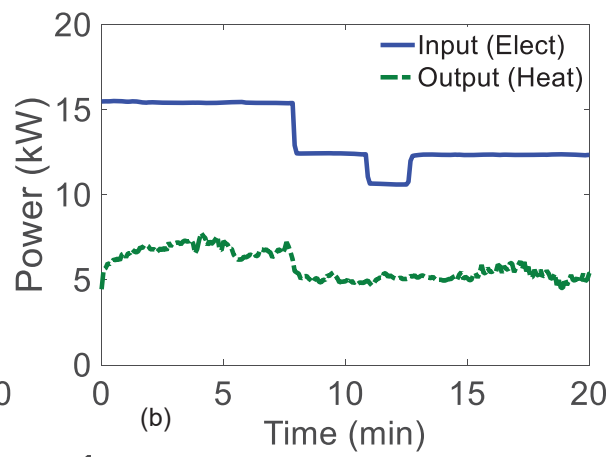
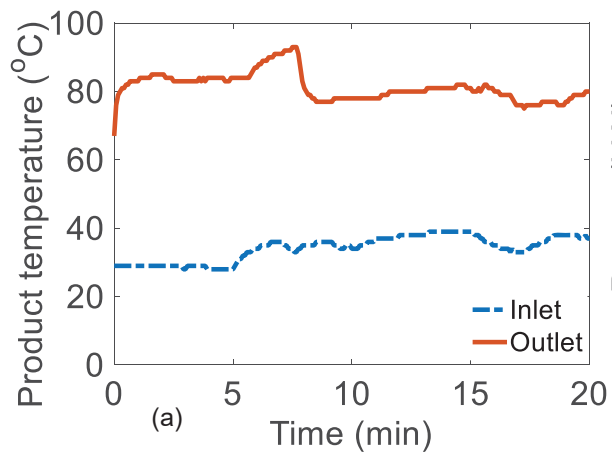
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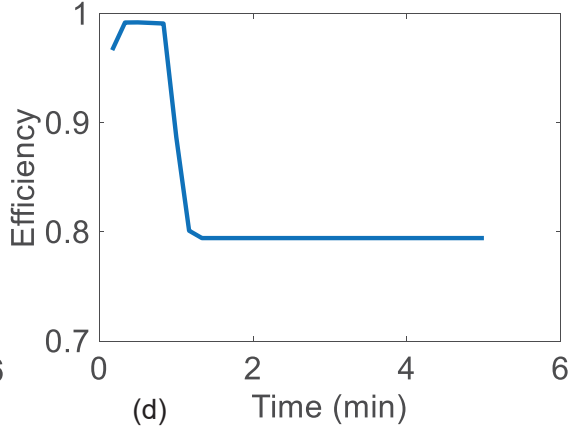
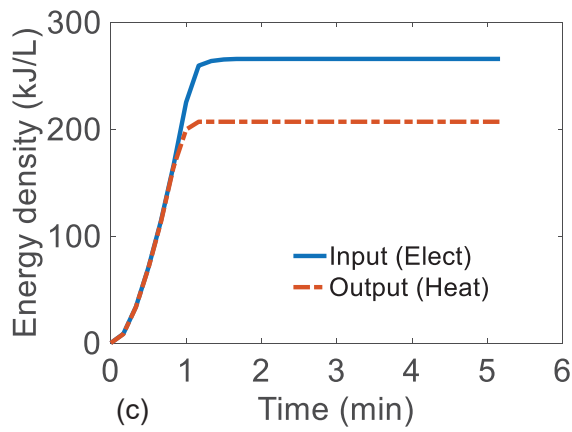
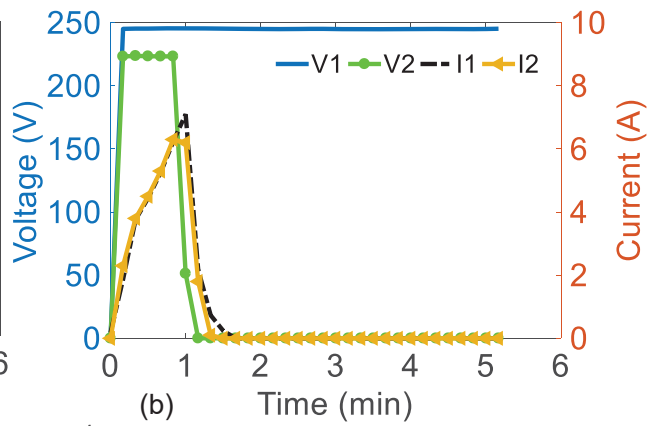
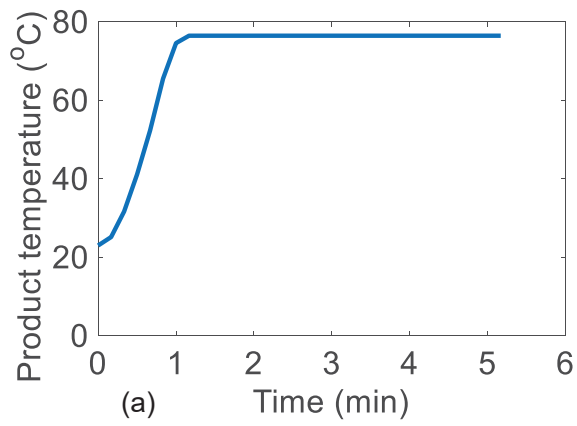
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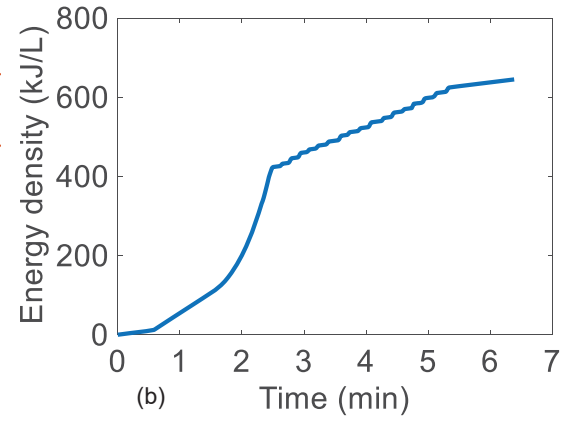
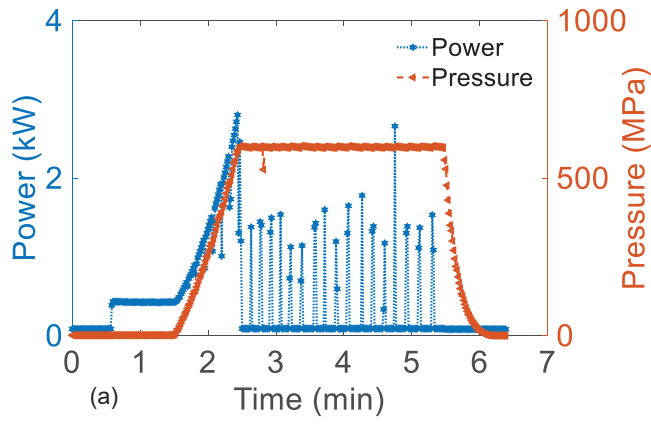
Fig. A1.

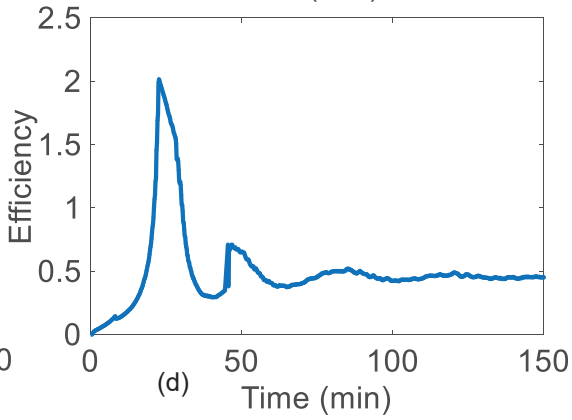
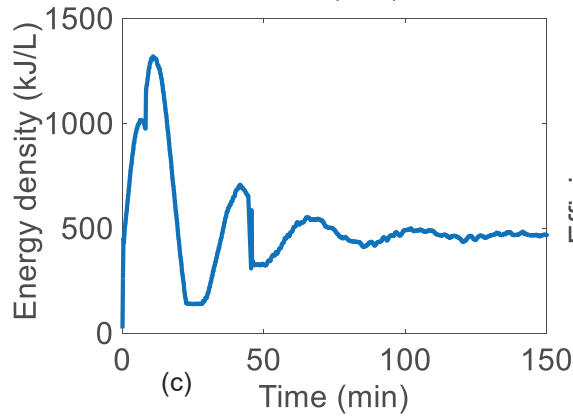
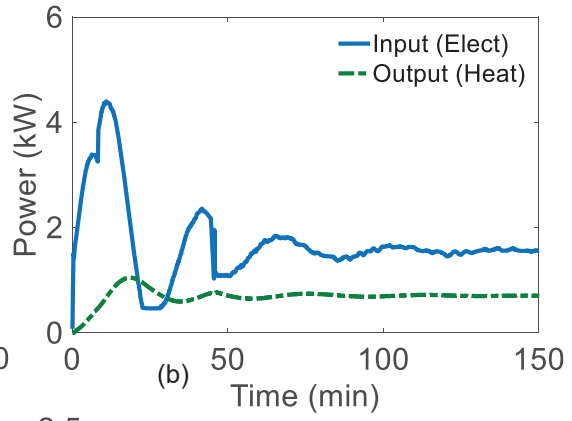
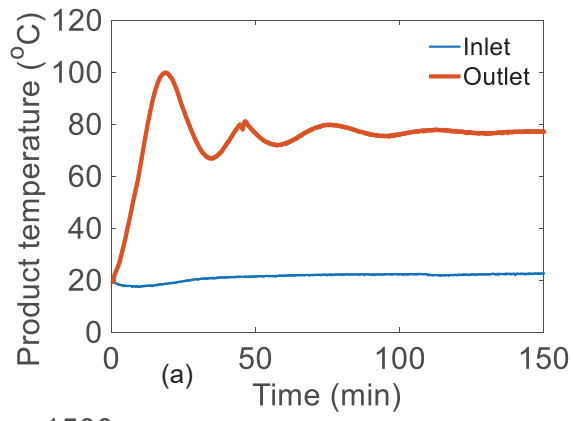
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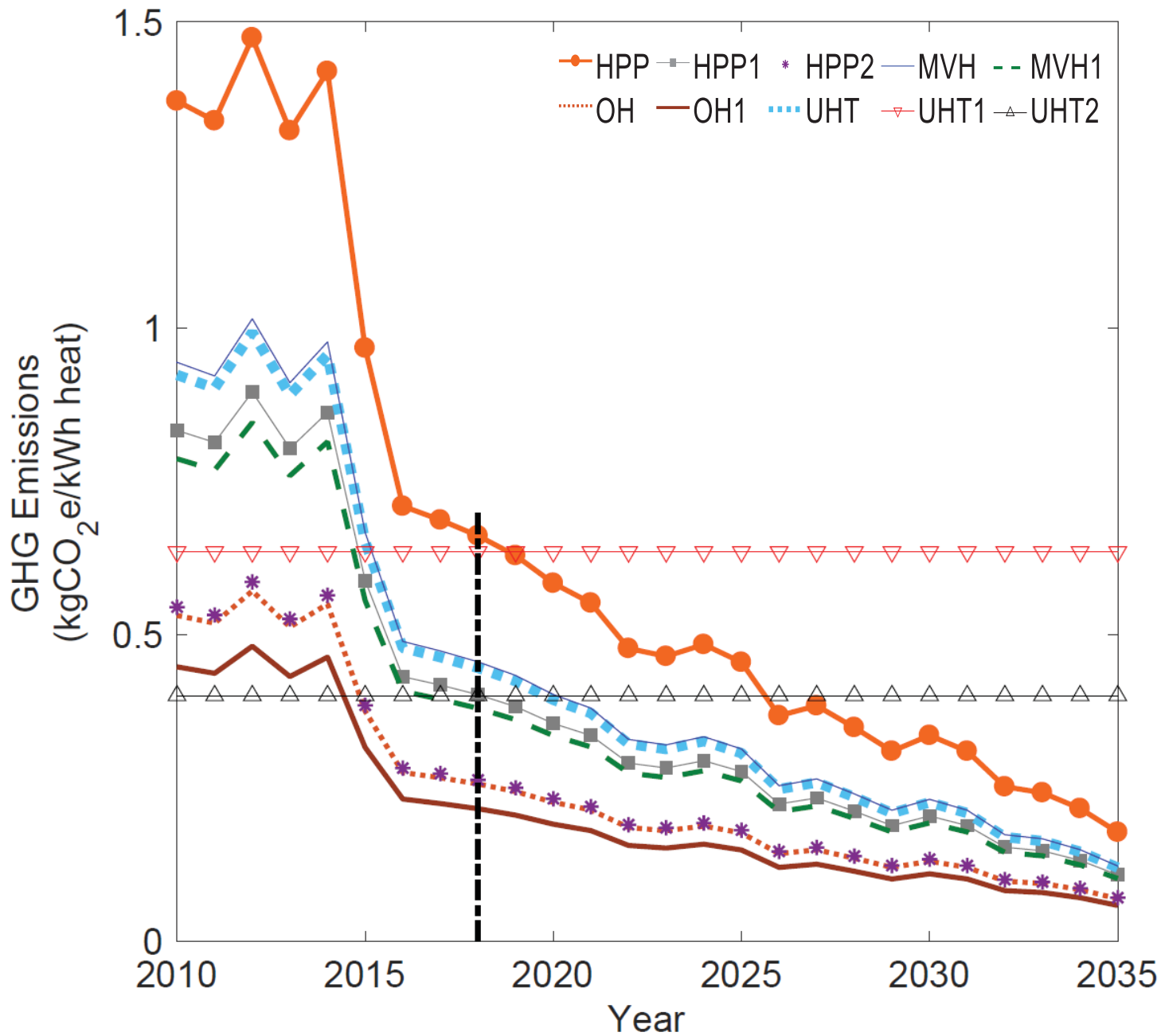


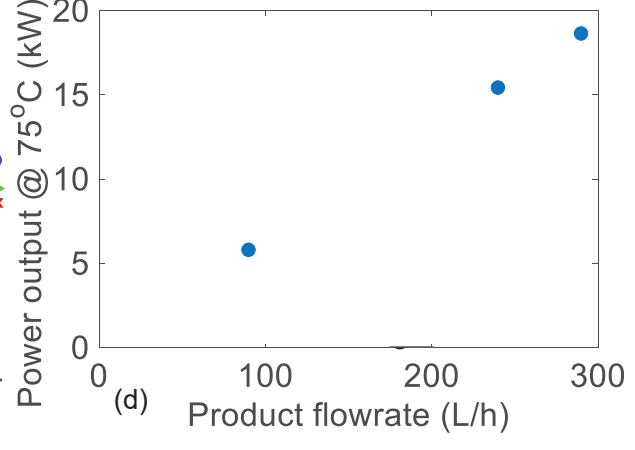
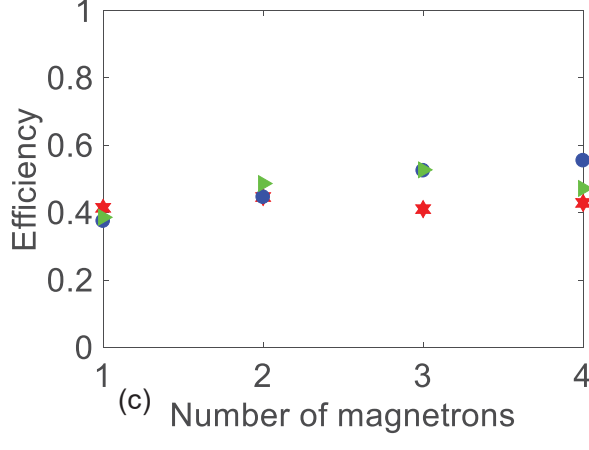
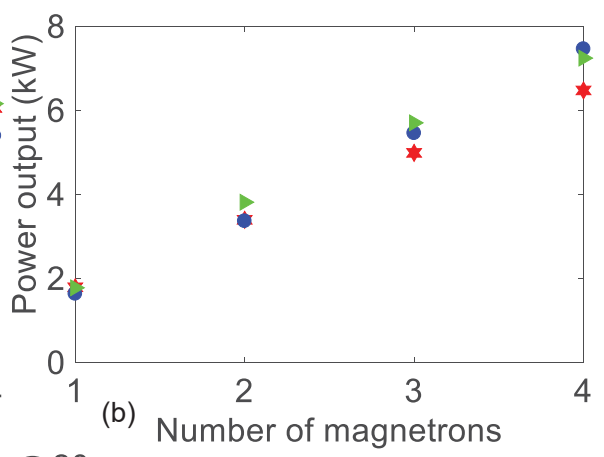
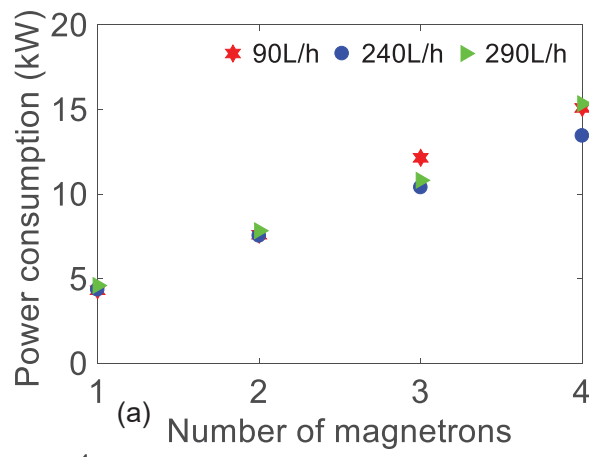


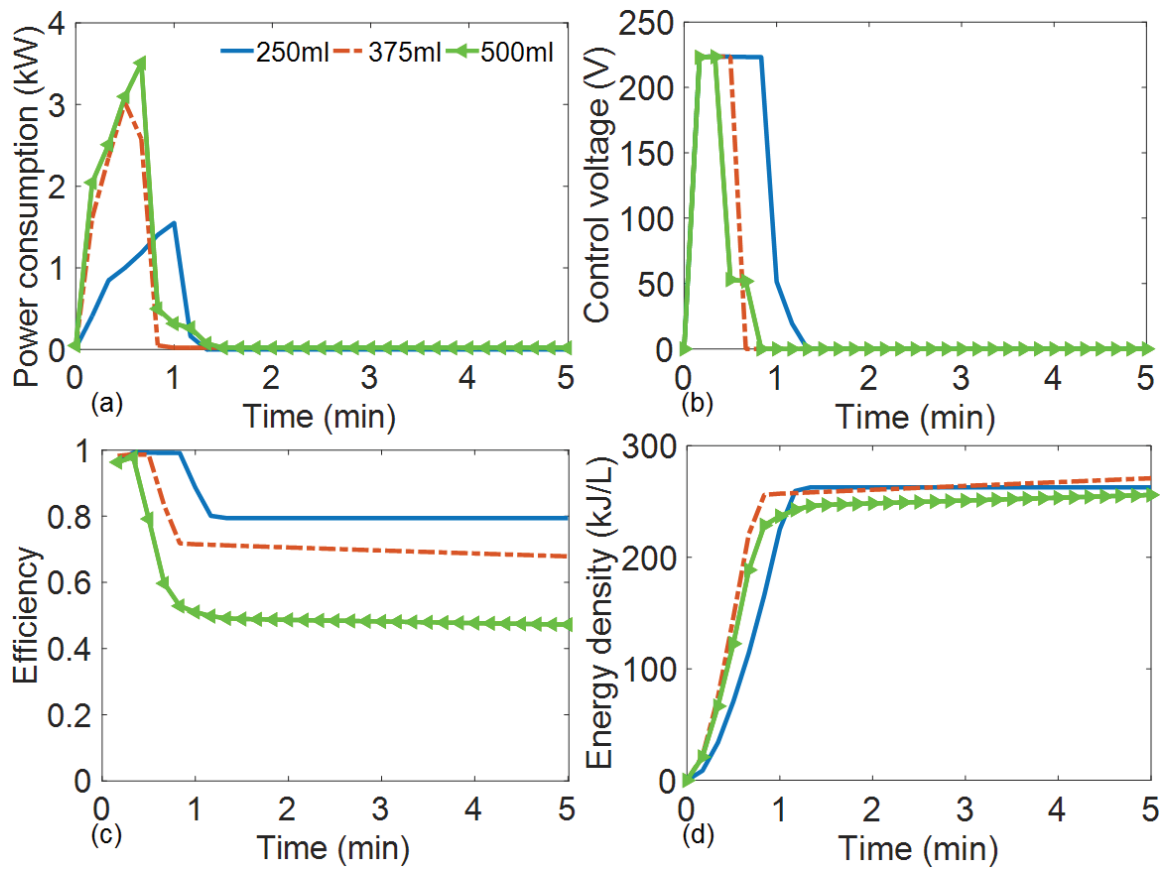


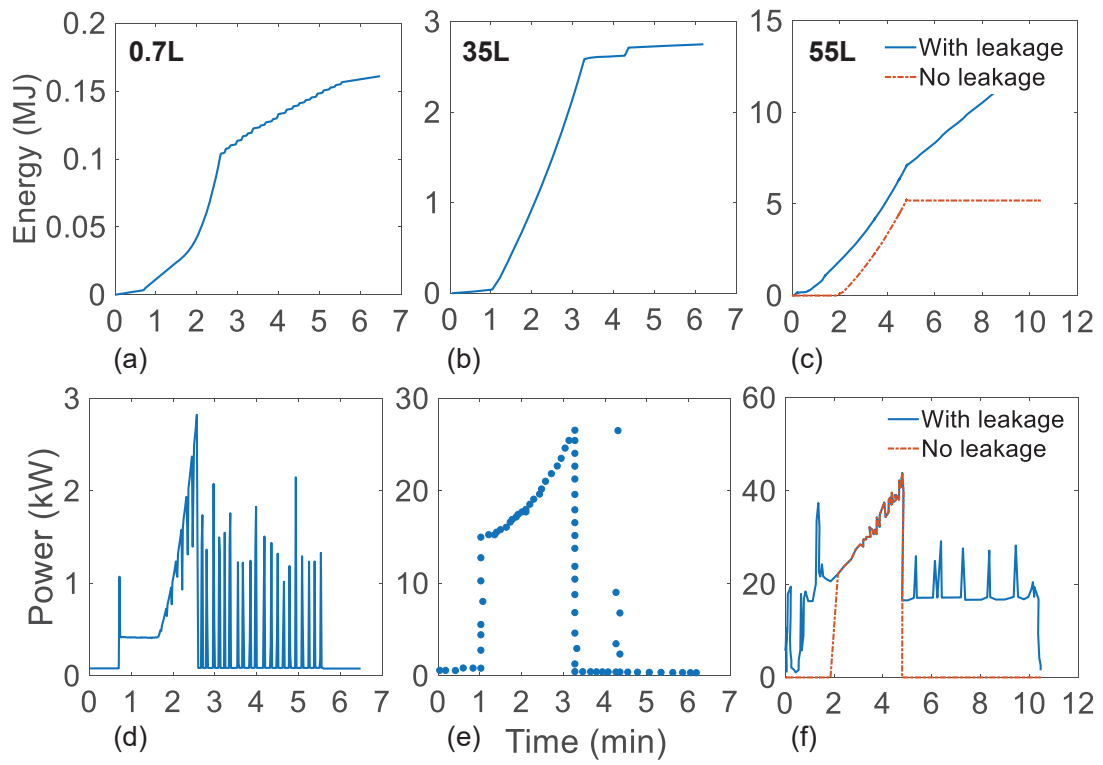












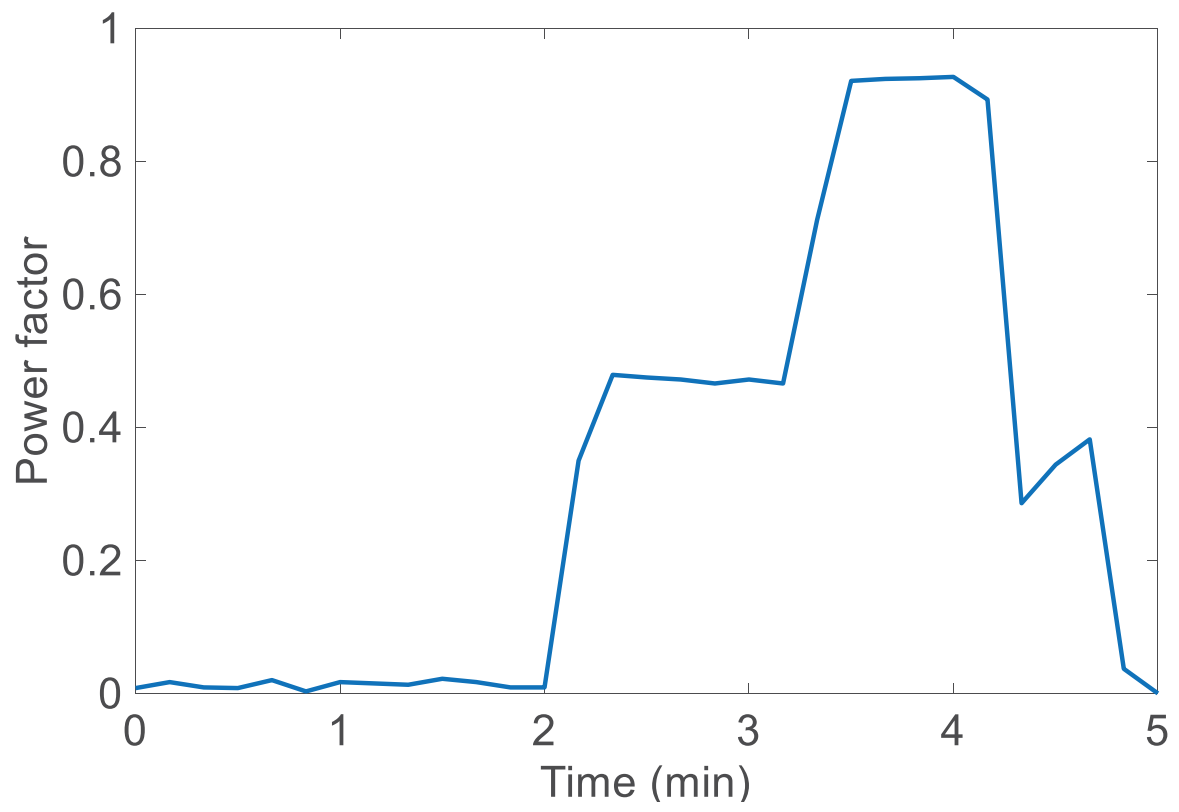


Fig. 1. (a) Continuous-flow microwave heating system (b) Batch ohmic heating system (c) High pressure processing system (d) Schematic of UHT/HTST system operating principle. Symbols used are OJ (orange juice), MC (microwave chamber), CW (cooling water), MCW (magnetron cooling water), $\text{_{in}}$ (inlet conditions), $\text{_{out}}$ (outlet conditions), V (voltage coil connections for energy measurement: two terminals connected to live L and neutral N power supply conductors, respectively), A (current coil connections for energy measurement, achieved by clamping the jaws of the meter over the live conductor L from the mains)

Fig. 2. MVH system energy performance-related characteristics

Fig. 3. OH system energy performance-related characteristics

Fig. 4. (a) Pressure-time & power-time characteristics (b) Energy density of HPP system

Fig. 5. UHT/HTST system energy performance-related characteristics

Fig. 6. GHG emission comparisons among processes 2010 – 2035, with 2018 vertical line

Fig. 7. Equipment scale results: continuous-flow microwave volumetric heating (MVH)

Fig. 8. Equipment scale results: batch ohmic heating (OH)

Fig. 9. Equipment scale results: high pressure processing (HPP)

Fig. A1. Power factor dynamics of total OH load (juice between electrodes and power supply system)

	HPP	HPP1	HPP2	MVH	MVH1	OH	OH1	UHT	UHT1	UHT2
Energy efficiency	0.31	0.51	0.78	0.45	0.54	0.8	0.95	0.46	0.29	0.46
Specific energy (kJ/L)	645	392	256	380	325	208	175	470	746	470
Transient time (min)	2.5			8	8	1		120		
KEY										
HPP1	HPP with 60% vessel filling ratio									
HPP2	HPP with 95% vessel filling ratio									
MVH1	MVH with cabinet cooling replaced by available cooling water									
OH1	Continuous OH process (without voltage switching control)									
UHT1	Gas-fired UHT/HTST system with 50% boiler efficiency									
UHT2	Gas-fired UHT/HTST system with 80% boiler efficiency									

Year	HPP	HPP1	HPP2	MVH	MVH1	OH	OH1	UHT	UHT1	UHT2
2016	0.0394	0.0240	0.0157	0.0232	0.0199	0.0127	0.0107	0.0287	0.0382	0.0241
2017	0.0382	0.0232	0.0152	0.0225	0.0192	0.0123	0.0104	0.0278	0.0382	0.0241
2018	0.0368	0.0223	0.0146	0.0217	0.0185	0.0119	0.0100	0.0268	0.0382	0.0241
2019	0.0350	0.0213	0.0139	0.0206	0.0176	0.0113	0.0095	0.0255	0.0382	0.0241
2020	0.0325	0.0197	0.0129	0.0191	0.0164	0.0105	0.0088	0.0236	0.0382	0.0241

Year	UK Electricity Grid Emissions factor (kgCO₂e/kWh)
2010	0.425
2011	0.415
2012	0.457
2013	0.41
2014	0.44
2015	0.3
2016	0.22
2017	0.213
2018	0.205
2019	0.195
2020	0.181
2021	0.171
2022	0.148
2023	0.144
2024	0.15
2025	0.141
2026	0.114
2027	0.119
2028	0.108
2029	0.096
2030	0.104
2031	0.096
2032	0.078
2033	0.075
2034	0.067
2035	0.055

Table 1. Energy performance comparison among processes. Note: UHT1 and UHT2 values are based on primary energy (kJ of natural gas) and the others on electrical energy (kJ)

Table 2. Pasteurisation process GHG emissions (kgCO₂e/L of juice) compared across technologies (2016 – 2020)

Table A1. UK electricity grid emissions factor, 2010-2035 (DBEIS, 2018)