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1 **Benefits of the use of sewage sludge over EDTA to remediate soils polluted with heavy**
2 **metals**

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13

14 **ABSTRACT**

15 Sewage sludges from urban waste water treatment plants are often used to remediate degraded
16 soils. However, the benefits of their use in metal-polluted soils remain unclear and need to be
17 assessed in terms of factors besides soil fertility. This study examines the use of thermal-
18 dried sewage sludge (TDS) as an amendment for heavy metal-polluted soil in terms of its
19 effects on soil chemical properties, leachate composition and the growth of native plant
20 communities. To assess the response of the soil and its plant community to an increase in
21 metal mobilization, the effects of TDS amendment were compared with those of the addition
22 of a chelating agent (EDTA). The experimental design was based on a real case scenario in
23 which soils from of an abandoned mine site were used in a greenhouse bioassay.
24 Two doses of TDS and EDTA were applied to a soil containing high Pb, Zn, Cu and Cd levels
25 (4925, 5675, 404 and 25 mg/kg respectively). Soil pH was 6.4 and its organic matter (OM)

26 content was 5.53 %. The factors examined after soil amendment were: soil fertility and heavy
27 metal contents, leachate element losses, the plant community arising from the seed bank
28 (plant cover, species richness and biodiversity, above/below ground biomass) and phytotoxic
29 effects (chemical contents of abundant species). TDS emerged as a good phytostabilizer of
30 Pb, Zn, Cu and Cd given its capacity to reduce the plant uptake of metals and achieve rapid
31 plant cover. This amendment also enhanced the retention of other elements in the plant root
32 system and overall showed a better capacity to remediate soils polluted with several heavy
33 metals. EDTA led to plant productivity losses and nutritional imbalances as it increased the
34 mobility of several elements in the soil and its leachates.

35

36 *Keywords:* polluted soils, ecological restoration, phytotoxicity, phytoremediation, plant
37 community

38

39 **INTRODUCTION**

40 When trying to ameliorate the impacts of old abandoned mines on land ecosystems, the main
41 problems faced are related to high concentrations of phytotoxic metals in top soil layers
42 (Gutiérrez-Ginés et al., 2013), along with low soil nutrient contents. Some researchers have
43 highlighted that soils containing heavy metals are among the most difficult to restore. This
44 has led to a constant search for plant species capable of accumulating heavy metals as well as
45 attempts to further our understanding of the effects of heavy metals on the plant species that
46 spontaneously grow at these old mine sites (Moreno-Jiménez et al., 2009; Pratas et al., 2013).
47 Another approach has been the use of soil amendments to help establish species targeted at
48 phytoremediating soils (Mench et al., 2007; Walker et al., 2003) and improving soil fertility
49 and structure.

50 Amendments composed of biowaste (unwanted material of biological origin, such as sewage
51 sludge, animal effluent, wood waste, green waste and crop residues) are attractive due to the
52 growing demand to incorporate waste in the ecological network while recovering waste
53 waters (Hall, 1995). Organic amendments can increase the amount of organic matter and plant
54 nutrients in soils, and act by retaining metals in the organic soil fraction (Park et al., 2011).
55 The better nutrient conditions of soils and the reduction of metal mobility and bioavailability
56 (thus toxicity) improve the growth of plants (Geebelen et al., 2002). However, these
57 amendments can also increase the leaching of nitrogen (Knowles et al., 2011) or upset the
58 balance of microflora and microfauna populations (Bright and Healey, 2003) with damaging
59 effects in particular on the microarthropods responsible for recycling soil organic matter
60 (OM) (Flores-Padarve et al., 2008; Flores-Padarvé and Hernández, 2012).

61 Other types of amendment, such as chelating agents, seek to increase the mobility of soil
62 metals and thus promote the absorption of micronutrients via a process known as assisted
63 phytoextraction. Plants grown in chelating agent-amended soils are able to accumulate higher
64 concentrations of heavy metals. However, this technique has possible detrimental effects such
65 as increasing the risk of leachates percolating to groundwater along with a loss of soil
66 nutrients (Jiang et al., 2003; Mench et al., 2007; Pastor et al., 2007; Thalayakumaran et al.,
67 2003). In spite of these drawbacks, some authors have continued to explore the use of
68 chelating agents to improve the phytoextraction capacity of some plants (Gheju and Stelescu,
69 2013; Hadi et al., 2014; Kambhampati and Vu, 2013; Markovska et al, 2013; Suthar et al,
70 2014).

71 The main goal of ecological soil remediation is to restore its multifunctionality and fertility.
72 This means that attention should be paid to the mobile heavy metal fraction both in terms of
73 its stabilization/immobilization and of extraction by its accumulating plant biomass.

74 This study was designed to examine the effects of sewage sludge amendment of a heavy
75 metal-polluted soil collected from an abandoned mine site. Apart from effects on soil
76 properties and leachate composition, we tried to assess the development of the plant
77 community that spontaneously thrives in the soil. We also compared the effects of this
78 amendment with likely metal immobilization actions with those of a chelating agent known to
79 increase the mobility and availability of metals.

80 To the best of our knowledge, this study is the first to address the issue of how soil
81 amendments affect native plant communities able to withstand incredibly high concentrations
82 of heavy metals in polluted environments.

83

84 **MATERIALS AND METHODS**

85 **Polluted soil and amendments used in the bioassay**

86 Soil was collected from an old silver mine in the province of Toledo (central Spain). This site
87 has been the objective of numerous studies by our research group (Hernández and Pastor,
88 2007; 2008). The mine was abandoned in the 1980s and now consists of a landfill of tailings
89 occupying some 20,000 m³. The site was left untreated causing the transfer of heavy metals to
90 neighbouring soils. The frequent use of this site for sheep grazing and as hunting grounds
91 determines the need to assess possible remediation strategies to mitigate pollution risks.

92 According to prior knowledge of the concentrations and distributions of pollutants at this site
93 (Gutiérrez-Ginés et al., 2013), we collected soil from the most polluted zone for our
94 experiment. The topsoil layer (0 – 20 cm) was carefully collected with the help of a hoe to
95 avoid disrupting its structure after cutting all vegetation back to the soil surface. A further soil
96 sample was obtained from the same zone for chemical determinations (see Table 1). The soil
97 samples collected mainly showed high Pb and Zn concentrations but Cu and Cd were also
98 present at concentrations higher than the permissible levels defined by Spanish law (BOE,

99 1990). Despite this, the soil sustains the growth of a plant community. The most common
100 species of this community are listed in Table 2.

101 As an organic amendment, we used thermal-dried sewage sludge (TDS) from an urban waste
102 water treatment plant. The chemical composition of the TDS used is detailed in Table 3. Its
103 heavy metal concentration is lower than the maximum permissible level stipulated for sewage
104 sludge used for agricultural purposes (BOE, 1990). The doses applied were 25 and 50 g of
105 TDS for each kg of soil, corresponding to field doses of 100 t/ha and 200 t/ha respectively.
106 Ethylenediaminetetraacetic acid (EDTA) was the chelating agent selected to assess
107 phytoextraction by the plant community. This type of amendment is especially efficient at
108 mobilizing bivalent ions and is used to increase the bioavailability of nutrients thus improving
109 soil fertility (Li and Shuman, 1996). Results of our prior work (Pastor et al., 2007) suggest
110 that EDTA produces fewer negative effects than DTPA on plant communities. EDTA is able
111 to extract Zn, Cd and Ni from soils (Li and Shuman, 1996). EDTA was used in its acid form.
112 The concentrations tested in this experiment were 0.5 g and 1 g of EDTA per kg of soil,
113 corresponding to field doses of 2 t/ha and 4 t/ha respectively.

114

115 **Experimental design and monitoring**

116 The bioassay was conducted in a greenhouse with controlled temperature (24°C) and humidity
117 (60%). Microcosms were set up in plastic pots (surface area 14 cm x 19 cm, depth 7.5 cm)
118 lined with black plastic to block the passage of light and thus avoid the growth of algae and
119 moss within the soil. In each pot, a grid was positioned 1 cm from the bottom to collect any
120 leached soil water. The pots were each filled with 1 kg of unsieved soil containing its
121 autochthonous seeds.

122 The TDS was mixed with the soil introduced in the corresponding pots from the first day of
123 the bioassay. EDTA was added 4 weeks later, after the first seeds had germinated to allow

124 sufficient time for a root system capable of absorbing mobilized metals to form. The chelating
125 agent was diluted in 100 mL of deionised water and added gradually over 8 days given that
126 the extracting effect of EDTA seems to improve when applied as several doses. This process
127 also minimized the risk of the solution percolating through the system before contacting the
128 soil particles containing the metals and trace elements. We are aware that the doses of both
129 amendments used were high but this was designed to ensure marked differences between
130 treatments and sufficient statistical power despite a fairly small number of replicates (three
131 per treatment).

132 Initially, the microcosms were watered daily with 100 mL of deionised water to keep them
133 wet until seedlings from the bank started to germinate. Once germinated, watering was
134 reduced to 50 mL every two days since this amount of water proved to be sufficient for plant
135 growth. When the microcosms became densely populated, the amount of water given was
136 doubled.

137 Following germination, numbers of seedlings in each pot were recorded weekly. When these
138 became so numerous they could not be counted without affecting the community, percentage
139 cover values were recorded. Cover was measured visually with the help of the method of
140 Emberger (1968). When the plants were mature, species were identified to estimate plant
141 diversity (as Shannon indices). The bioassay was run for 30 weeks when plants started to lose
142 their vigour.

143

144 **Sample preparation and chemical analyses**

145 Leachate collection started after the application of EDTA in the 4th week. Leachates were
146 collected in the 5th, 7th and 9th week of the trial. No leachates were produced after that period,
147 due to the development of the plant community. The leachate samples were filtered and stored

148 with a drop of HNO_3 at 4°C until analysis. Nutrient and trace element concentrations in the
149 leachates were determined by ICP-OES.

150 At the end of the bioassay, the soil was separated from the plants, which were split into above
151 and below surface biomass. Total, above-ground and below-ground biomass were expressed
152 in mg/cm^2 .

153 Original soil from the mine site, as well as soils from microcosms, were dried at room
154 temperature for one week and sieved through a 2 mm mesh. Tests conducted in each one
155 were: pH in slurry, organic matter by potassium dichromate reduction, total Kjeldahl nitrogen
156 (according to protocols described in Hernández and Pastor, 1989), metal concentration
157 determined by inductively coupled plasma-optical emission spectroscopy (ICP-OES, Perkin-
158 Elmer 4300DV). Pseudo-total metal concentrations were determined after HNO_3 and HClO_4
159 digestion (Walsh and SSSA, 1971) and bioavailable metal concentrations after extraction with
160 DTPA 0.005 M at pH 8.3 (Lindsay and Norwell, 1978). All plant samples (separating roots
161 from shoots) were washed with tap water and rinsed two times with deionised water, oven-
162 dried at 70°C for 48 h, weighed and ground in a IKA-WERKE Yellow Line A10 grinder. The
163 procedure for determining total metal concentrations was the same as used on the soil (Walsh
164 and SSSA, 1971).

165

166 **Data treatment**

167 All soil data were analysed by unifactorial ANOVA ($p < 0.05$) following their log and Box
168 Cox transformation to correct for deviations from normality and heteroscedasticity problems.
169 Leachate and biomass data were compared using the non-parametric Kruskal Wallis test at a
170 95% confidence level (indicated as * in the tables), given that the deviation from normality
171 and homoscedasticity hypotheses were too large for correction by transformation. All
172 statistical tests were performed using the software package STATISTICA 9.0.

173

174 **RESULTS**175 **Effects on soil variables**

176 Soil properties related to fertility (Table 4) were modified by the two types of amendment.

177 EDTA led to a slight decrease in pH, while TDS caused a marked drop in pH because it was

178 more acidic than the soil. The addition of TDS also gave rise to an increase in total amounts

179 of OM, N, Ca and P, but not their bioavailability. Only the bioavailability of Na was increased

180 by soil amendment with TDS. In contrast, the use of EDTA besides lowering the total Mg

181 concentration also affected the bioavailability of the remaining macronutrients.

182 Pseudo-total soil metal concentrations (Table 5) did not seem to be affected by either

183 amendment, despite the presence of metals in the TDS or the expected reduction in metals due

184 to their mobilization by EDTA. Only pseudo-total concentration of Al was lower in soils with

185 the highest rate of EDTA application, although the difference with control was not significant.

186 Neither was the difference produced in bioavailability very evident (Table 5). Only Zn

187 showed a reduced bioavailable fraction in response to the higher TDS dose, while

188 bioavailabilities of Fe and Cu increased, probably due to the drop in pH. Neither did EDTA

189 notably affect the bioavailable fraction, which only showed an increase in Cu concentration.

190

191 **Effects on the leachates produced**

192 Although no substantial effects were recorded in the macronutrient and metal concentrations

193 of the bioavailable fraction of the amended soils, their differing mobilities according to the

194 amendment used were reflected by the leachate data (Figure 1). The addition of EDTA to the

195 soil served to mobilize most of the elements examined (Ca, Mg, K, Na, Fe, Mn, Zn, Pb, Cu,

196 Cd and Ni). Their concentrations in the leachates were always significantly higher than for the

197 control treatment. Although the average concentration of Al in leachates was higher for the

198 highest EDTA dose, this difference was not significant. This could explain the lower pseudo-
199 total concentration of Al in soil treated with this amendment.

200 Although we expected the organic amendment to retain soil elements, their concentrations in
201 the leachates collected from the TDS treatments were somewhat higher than in controls yet
202 lower than in the case of EDTA. This difference with respect to the control was significant for
203 Ca, Mg, K, Na, Fe, Mn, Zn and Pb in the soils treated with the highest TDS dose, probably
204 due to the reduced soil pH and/or added metals in the TDS. The effect was more pronounced
205 in the leachates collected from the microcosms in Week 9.

206

207 **Effects on the plant community**

208 Changes produced during the growth of plants from the soil seed bank are shown in Figure 2.
209 The TDS amendment had a beneficial effect on the growth of species from the seed bank and
210 plant numbers were higher from the bioassay start in the microcosm soils treated with this
211 amendment than in untreated controls. This trend became more evident in the cover values
212 recorded after the third month of the bioassay. Thus, both TDS doses gave rise to cover values
213 of around 60% throughout the experiment compared to values of 20-30% observed for
214 controls. In contrast, the number of plants in the microcosms was unaffected by the addition
215 of EDTA. However, some two months after amendment, cover values gradually decreased
216 compared to controls with increasing concentrations of the chelating agent.

217 The remaining plant community factors measured at the end of the experiment (species
218 richness and diversity, soil cover, and above- and below ground phytobiomass) are provided
219 in Table 6. All factors except species richness were affected by the two soil amendments.
220 Despite no effects of TDS on plant number or biodiversity, plant growth was affected by this
221 treatment and both cover and biomass were significantly higher than in controls. Plants
222 growing in the EDTA-treated soils showed reduced growth in terms of significantly lower

223 cover, total phytobiomass, below-ground biomass and above-ground biomass values
224 compared to the values recorded in controls. Species richness and Shannon indices were also
225 lower. Finally, the most abundant species growing in the control microcosms were *Scirpoides*
226 *holoschoenus* (L.) Soják, *Bartsia trixago* L., *Rumex bucephalophorus* L. and *Galium*
227 *parisiense* L., reflecting the situation observed at the mine site. These species were less
228 represented in the microcosms amended with EDTA, while grasses thrived mostly in the TDS
229 treatments, especially *Agrostis castellana* Boiss. & Reut.

230

231 **Phytotoxicity-related effects**

232 The uptake of nutrients and metals from the soils in response to TDS or EDTA was assessed
233 by examining the root system obtained from the microcosms as a whole along with the above-
234 surface mass of species appearing in sufficient quantities for chemical analysis.

235 Despite expecting a reduction in the heavy metals absorbed by the plants in the TDS
236 treatments, due to their retention by the OM, heavy metal concentrations (Fe, Mn, Zn, Pb, Cd,
237 Ni) in the root biomass (Table 7) increased with TDS dose compared to the control
238 treatments. In contrast, macronutrients did show a decreasing tendency in response to this
239 treatment, especially in the case of P and K. EDTA significantly increased the entry of most
240 of the elements (Na, Al, Fe, Mn, Zn, Pb, Cu, Cd, Ni) in the roots with respect to controls.

241 Although both amendments led to greater concentrations of metals in the roots, levels
242 detected in the above-ground mass of the most abundant species (Table 8) indicated their
243 different translocation. Except in the case of *Vulpia myuros* (L.) C.C.Gmel., above-ground Zn
244 concentrations were reduced in the species examined (*Agrostis castellana*, *Lolium rigidum*
245 Gaudin, *Scirpoides holoschoenus*) when growing in soils amended with TDS. Lead levels also
246 fell in the above-ground parts of *Agrostis castellana*, *Lolium rigidum* and *Vulpia myuros*, but

247 increased in *Scirpoides holoschoenus*. In the EDTA treatments, metal concentrations were
248 higher in the majority of species.

249 An overview of the effects produced by the two soil amendments is provided in Table 9.

250

251 **DISCUSSION**

252 Given the low N and nutrient contents of the polluted soil site examined in this study, the
253 addition of TDS induced a considerable change to this ecosystem. Unexpectedly, TDS,
254 especially when used at high concentrations, increased the mobility of soil metals, as
255 indicated by both leachate and root metal levels. This effect may be attributed to the
256 considerable drop in soil pH induced by this amendment. Antoniadis et al. (2008) similarly
257 detected a one-unit reduction in the pH of a soil treated with sewage sludge, and this gave rise
258 to a marked increase in the metal concentrations of both the soil's bioavailable fraction and
259 of the raygrass that grew in it. Chiu et al. (2006) reported similar findings. Some of the
260 negative effects of adding sewage sludge to soil can be mitigated by blending this waste with
261 other soil amendments such as Ca-carbonate (Siebielec and Chaney, 2012) or sawdust
262 (Ammari et al., 2012).

263 The increase in metal mobility do not, however, seem to affect the concentrations of heavy
264 metals in the shoots and leaves of plants, which in most cases were lower than in controls.

265 This possible reduction in metal toxicity along with the nutrients supplied by the TDS induced
266 the rapid growth of plants from the soil seed bank, increasing plant cover and biomass.

267 Although plant community diversity was unaffected, a change in floristic composition was
268 observed in favour of Gramineae. Moreno-Peñaranda et al. (2004) and Newman et al. (2014)
269 also addressed plant community dynamics in non-metal polluted degraded soils amended with
270 sewage sludge over a 5- or 6 year period. The results of the study by Moreno-Peñaranda et al.
271 (2004) revealed the more rapid initial growth of plant covers attributable to the amendment,

272 though this difference was lost over the years and similar covers to controls which grew at a
273 slower rate were observed. This rapid expansion of cover does not necessary indicate an
274 improved diversity or a floristic composition corresponding to a more advanced succession
275 stage. Whichever the case, the large proportion of naked soil that exists at mine sites such as
276 that examined here means that the rapid formation of a plant cover that will impair the erosion
277 and transport of materials to adjacent zones is a priority of any remediation measure for
278 polluted soils. In this regard, TDS seems to be an effective amendment for the remediation of
279 soils such as the ones used here.

280 The application of EDTA to the soil gave rise to an impoverished soil due to a loss of
281 macronutrients, as also observed by Wasay et al. (2001) and by Barona et al. (2001). The
282 important loss of nutrients (Mg and K) using the higher EDTA dose suggests that, in the long
283 term, phytoremediation would be ever less efficient due to cumulative plant nutritional
284 deficiencies. EDTA persists in the environment (Adiloglu, 2002; Li and Shuman, 1996; Wu et
285 al., 2004), such that nutrient loss might continue into the long term. This reduction in soil
286 fertility means the soil cannot be considered recovered though heavy metal concentrations
287 dropped to acceptable levels. In addition, the decrease in pH could also give rise to the greater
288 mobility of metal ions and reduced fertility.

289 Changes produced in soil chemistry could produce effects on the biological activity of soil
290 microorganisms (Sastre et al., 2007; Tiller, 1989). Despite the increased mobility of metals
291 along with no marked change in soil chemical composition in the long term, this chelating
292 agent cannot be recommended for the phytoremediation of soils with high concentrations of
293 Pb, Zn, Cu, Cd. In effect, the levels of these heavy metals were practically identical to those
294 of the untreated soils. It therefore seems that EDTA may not be useful for the recovery of
295 some soils, since the loss of biomass of the more extracting species determines that the

296 quantity of metals removed from the system is similar in the long term to the amount present
297 if the agent had not been added (Madrid et al., 2003).

298 The greater toxicity induced by this chelating agent seemed to manifest as a decrease in plant
299 cover (though the number of seedlings germinating was unaffected). This indicates that this
300 variable may be a good indicator of toxicity due to soil metals as observed by Pastor et al.
301 (2003). In agreement with the findings of other authors (An, 2004; Landis and Yu, 1999;
302 Newman and Jagoe, 1996), germination did not emerge as a reliable indicator of ecotoxicity.
303 This amendment induced considerable reductions in plant cover and diversity along with a
304 drop in productivity. This observation is consistent with the arguments of Landis and Yu,
305 (1999) and highlights the idea that EDTA alone is not ideal for phytoextraction in a soil
306 polluted with several heavy metals.

307 The growth and presence of *Scirpoides holoschoenus* and *Agrostis castellana* in most of the
308 microcosms (control, TDS-amended and EDTA-amended) suggest the suitability of these
309 species for further heavy metal phytoremediation studies. Despite the substantial levels of
310 metals that can reach their above-ground parts, both species are considered metal excluders
311 (Meharg et al., 1999; Pastor et al, 2015). This fact, together with the rhizomatous
312 development of their below-ground phytomass, make them important constituents of plant
313 covers to stabilize heavy metals and prevent soil erosion.

314

315 **CONCLUSIONS**

316 The ecological restoration of a site whose soils are polluted with several heavy metals using
317 amendments cannot be assessed only in terms of effects produced on soil fertility. The
318 transfer of metal ions through leachates or the incorporation of these ions in the system's
319 autotrophic component is pivotal both for the health of the ecosystem being restored and to
320 prevent pollutants from spreading to adjacent ecosystems.

321 When applied to a soil of these characteristics, EDTA seems to lead to plant productivity
322 losses and nutritional imbalances, as it induces significant increases in the mobility of several
323 elements both in the soil and its leachates. This gives rise to a real increase in toxicity along
324 with soil impoverishment in the long term due to macronutrient loss.

325 TDS seems to be an interesting candidate for phytostabilizing soils polluted with Pb, Zn, Cu
326 and Cd given its capacity to achieve a rapid soil cover as well as enhancing the retention of
327 other elements in the plant root system, while avoiding their translocation to above-ground
328 parts. This could help minimize soil erosion and the real toxicity posed to the plants of the
329 ecosystem without a need to introduce allochthonous species. In future work, the optimal
330 dose of this type of amendment will need to be determined.

331 This bioassay conducted under controlled conditions but simulating a real scenario seems a
332 useful tool to assess the feasibility of strategies designed to remediate polluted soils.

333

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337

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- 467

468 Figure 1. Mean concentrations (mg/L) of macroelements and metals in leachates from
469 microcosms collected in weeks 5, 7 and 9 of the experiment. Values along the vertical axes in
470 the Zn and Pb graphs are expressed as logs. Values of 0 mg/L are represented as 0.01. TDS 1
471 and 2 represent the results of the two TDS treatments (25 g/kg and 50 g/kg respectively) and
472 EDTA 1 and 2 the results of the two EDTA treatments (0.5 g/kg and 1g/kg respectively).

473

474 Figure 2. Changes produced in the number of plants and plant cover throughout the bioassay.
475 TDS 1 and 2 represent the results of the two TDS treatments (25 g/kg and 50 g/kg
476 respectively) and EDTA 1 and 2 the results of the two EDTA treatments (0.5 g/kg and 1g/kg
477 respectively).

478 Table 1. pH, N (%), organic matter (OM, %), C (%), macroelements (mg/kg) and pseudo-
 479 total and available metals (mg/kg) recorded in the mine soil used in the bioassay.

| Parameter related to fertility | Metal concentration (mg/kg) | | | | |
|--------------------------------|-----------------------------|----|-----------|----|------|
| | Pseudo-total | | Available | | |
| pH | 6.4 | Al | 25100 | Al | 0.12 |
| N | 0.19 | Fe | 22600 | Fe | 6.4 |
| OM | 5.53 | Mn | 1190 | Mn | 7.4 |
| C | 3.21 | Zn | 5680 | Zn | 129 |
| Ca | 1930 | Pb | 4520 | Pb | 138 |
| Mg | 136 | Cu | 404 | Cu | 35 |
| K | 270 | Cd | 25 | Cd | 0.98 |
| Na | 10 | Ni | 30 | Ni | 0.36 |

480

481

482 Table 2. Herb species that thrive in the grassland community where the experimental soil was
 483 collected.

| GRAMINEAE | LEGUMINOSACEAE | OTHER | |
|----------------------------|------------------------------|--------------------------------|--------------------------------|
| <i>Agrostis castellana</i> | <i>Ornithopus compressus</i> | <i>Alyssum granatense</i> | <i>Sanguisorba minor</i> |
| <i>Arrhenaterum album</i> | <i>Trifolium arvense</i> | <i>Asparagus acutifolius</i> | <i>Scirpoides holoschoenus</i> |
| <i>Avena barbata</i> | <i>Trifolium gemellum</i> | <i>Bellardia trixago</i> | <i>Silene colorata</i> |
| <i>Bromus diandrus</i> | <i>Trifolium glomeratum</i> | <i>Capsella bursa-pastoris</i> | <i>Spergularia rubra</i> |
| <i>Bromus madritensis</i> | <i>Trifolium scabrum</i> | <i>Cerastium glomeratum</i> | <i>Thymus zizis</i> |
| <i>Bromus rubens</i> | <i>Trifolium suffocatum</i> | <i>Crassula tillaea</i> | <i>Verbascum pulverulentum</i> |
| <i>Bromus tectorum</i> | | <i>Diploxix catholica</i> | |
| <i>Corynephorus</i> | COMPOSITAE | <i>Echium plantagineum</i> | |
| <i>fasciculatus</i> | <i>Anacyclus clavatus</i> | <i>Erodium cicutarium</i> | |
| <i>Dactylis glomerata</i> | <i>Andryala integrifolia</i> | <i>Gallium parisiense</i> | |
| <i>Holcus setiglumis</i> | <i>Andryala laxiflora</i> | <i>Geranium dissectum</i> | |
| <i>Hordeum murinum</i> | <i>Carduus pynocephalus</i> | <i>Heliotropium</i> | |
| <i>Lolium rigidum</i> | <i>Carduus tenuiflorus</i> | <i>europaeum</i> | |
| <i>Melica ciliata</i> | <i>Centaurea melitensis</i> | <i>Hirschfeldia incana</i> | |
| <i>Mibora minima</i> | <i>Crepis vesicaria</i> | <i>Juncus bufonius</i> | |
| <i>Rostraria cristata</i> | <i>Chondrilla juncea</i> | <i>Lamium amplexicaule</i> | |
| <i>Stipa sp.</i> | <i>Filago minima</i> | <i>Mentha rotundifolia</i> | |
| <i>Vulpia bromoides</i> | <i>Filago pyramidata</i> | <i>Myosotis discolor</i> | |
| <i>Vulpia ciliata</i> | <i>Hypochoeris rostrata</i> | <i>Neatostema apulum</i> | |
| <i>Vulpia myuros</i> | <i>Leontodon saxatilis</i> | <i>Plantago afra</i> | |
| <i>Vulpia unilateralis</i> | <i>Logfia arvensis</i> | <i>Plantago coronopus</i> | |
| | <i>Picnomon acarna</i> | <i>Plantago lagopus</i> | |
| | <i>Tolpis barbata</i> | <i>Rumex bucephalophorus</i> | |

484

485

486 Table 3. pH, N (%), organic matter (OM, %), C (%), macroelements (mg/kg) and heavy
487 metals (mg/kg) recorded in the thermally-dried sewage sludge used in the bioassay.

| Fertility parameter | Value | Heavy metal | Value |
|---------------------|-------|-------------|-------|
| pH | 6.64 | Zn | 1120 |
| N | 3.94 | Pb | 398 |
| OM | 40.4 | Cu | 409 |
| P | 400 | Cd | 3.0 |
| Ca | 6360 | Ni | 175 |
| Mg | 1460 | | |
| K | 1560 | | |
| Na | 531 | | |

488

489

490 Table 4. Soil pH, N (%), organic matter (OM, %) and pseudo-total (T) and available (A)
 491 macroelements (mg/kg) recorded in the microcosms after the bioassay. *: significant
 492 difference (95%) versus control, **: significant difference (95%) versus control and versus
 493 the lower amendment dose.

| Parameter | Control | TDS | | | EDTA | |
|-----------|--------------|--------------|---------------|----------------|---------------|----|
| | | 25 g/kg | 50 g/kg | 0.5 g/kg | 1 g/kg | |
| pH | 7.45 ± 0.05 | 7.23 ± 0.14 | * 6.92 ± 0.03 | ** 7.27 ± 0.11 | * 7.21 ± 0.10 | * |
| N | 0.23 ± 0.02 | 0.29 ± 0.01 | * 0.32 ± 0.02 | ** 0.25 ± 0.03 | 0.25 ± 0.00 | |
| OM | 5.45 ± 0.07 | 6.08 ± 0.04 | * 6.86 ± 0.18 | ** 5.41 ± 0.40 | 5.45 ± 0.38 | |
| Ca T | 4860 ± 467 | 5110 ± 102 | 5780 ± 245 | * 4670 ± 123 | 4900 ± 307 | |
| Mg T | 2900 ± 316 | 2940 ± 235 | 2900 ± 290 | 2680 ± 264 | 2270 ± 154 | * |
| P T | 912 ± 167 | 1106 ± 91 | 1470 ± 90 | * 926 ± 51 | 1098 ± 140 | |
| K T | 10700 ± 2940 | 11200 ± 2030 | 10200 ± 3100 | 9940 ± 2320 | 5800 ± 1500 | |
| Na T | 225 ± 33 | 245 ± 14 | 219 ± 25 | 240 ± 14 | 249 ± 40 | |
| Ca A | 2220 ± 106 | 2140 ± 132 | 2100 ± 125 | 2560 ± 265 | * 2520 ± 226 | * |
| Mg A | 151 ± 5.0 | 145 ± 6.1 | 137 ± 8.1 | 160 ± 3.0 | 182 ± 32 | * |
| K A | 145 ± 5.0 | 137 ± 5.8 | 163 ± 7.6 | 127 ± 20 | * 116 ± 24 | ** |
| Na A | 23 ± 6.0 | 62 ± 2.9 | * 100 ± 18 | ** 30 ± 0.2 | * 32 ± 3.0 | * |

494

495

496 Table 5. Soil pseudo-total (T) and available (A) metal concentrations (mg/kg) recorded in the
 497 microcosms after the bioassay. *: significant difference (95%) versus control, **: significant
 498 difference (95%) versus control and versus the lower amendment dose.

| Element | Control | TDS | | | EDTA | |
|---------|---------------|--------------|--------------|--------------|--------------|----|
| | | 25 g/kg | 50 g/kg | 0.5 g/kg | 1 g/kg | |
| Al T | 25600 ± 5060 | 27200 ± 3920 | 24200 ± 5800 | 23100 ± 4450 | 14700 ± 2910 | |
| Fe T | 50700 ± 11700 | 50100 ± 6700 | 41800 ± 1750 | 45700 ± 2270 | 48600 ± 3760 | |
| Mn T | 1120 ± 79 | 1040 ± 58 | 984 ± 104 | 988 ± 60 | 973 ± 74 | |
| Zn T | 5780 ± 1840 | 5290 ± 264 | 5710 ± 848 | 5280 ± 487 | 5530 ± 733 | |
| Pb T | 4020 ± 622 | 4050 ± 564 | 4400 ± 423 | 4470 ± 368 | 4140 ± 428 | |
| Cu T | 117 ± 10 | 113 ± 3.7 | 114 ± 3.2 | 111 ± 5.2 | 109 ± 6.3 | |
| Cd T | 28 ± 3.6 | 25 ± 1.2 | 27 ± 2.4 | 27 ± 1.8 | 28 ± 4.3 | |
| Ni T | 29 ± 3.5 | 28 ± 1.7 | 28 ± 1.4 | 28 ± 0.8 | 27 ± 1.6 | |
| Al A | 0.007 ± 0.01 | 0.003 ± 0.01 | 0.030 ± 0.02 | 0.060 ± 0.04 | 0.010 ± 0.02 | |
| Fe A | 8.8 ± 5.6 | 18 ± 1.4 | * 17 ± 0.3 | * 12 ± 7.0 | 13 ± 3.1 | |
| Mn A | 6.5 ± 4.2 | 11 ± 7.6 | 15 ± 14 | 14 ± 9.9 | 14 ± 5.6 | |
| Zn A | 237 ± 11 | 224 ± 5.8 | 181 ± 14 | * 235 ± 17 | 246 ± 6.1 | |
| Pb A | 179 ± 8.5 | 180 ± 14 | 153 ± 11 | 157 ± 5.5 | * 155 ± 20 | |
| Cu A | 2.3 ± 0.4 | 3.4 ± 0.3 | * 3.1 ± 0.2 | * 3.2 ± 0.30 | * 3.7 ± 0.2 | ** |
| Cd A | 1.8 ± 1.0 | 1.8 ± 0.2 | 1.6 ± 0.1 | 1.9 ± 2.4 | 2.2 ± 0.1 | |
| Ni A | 0.9 ± 0.6 | 1.0 ± 0.1 | 0.9 ± 0.1 | 1.1 ± 1.8 | 1.3 ± 0.2 | |

499

500

501 Table 6. Plant community indicators at the end of the bioassay. *: significant difference (95%)
 502 versus control, **: significant difference (95%) versus control and versus the lower
 503 amendment dose. T. biomass: total biomass, A.G biomass: above-ground biomass, B.G.
 504 biomass: below-ground biomass.

| Parameter | TDS | | | | EDTA | |
|----------------------------------|-------------|-------------|-------------|-------------|-------------|----|
| | Control | 25 g/kg | 50 g/kg | 0.5 g/kg | 1 g/kg | |
| # seedlings | 112 ± 51 | 122 ± 48 | 126 ± 57 | 125 ± 50 | 115 ± 43 | |
| Cover (%) | 24 ± 4.9 | 61 ± 6.2 | * 62 ± 7.9 | * 24 ± 6.6 | 16 ± 6.9 | |
| Species richness | 9.3 ± 1.2 | 9.0 ± 2.6 | 8.3 ± 0.9 | 8.3 ± 0.6 | 5.0 ± 0.0 | * |
| Shannon index | 2.64 ± 0.21 | 2.30 ± 1.05 | 2.65 ± 0.16 | 2.53 ± 0.24 | 1.89 ± 0.22 | * |
| T. biomass (mg/cm ²) | 32 ± 9.0 | 41 ± 2.3 | * 42 ± 5.1 | * 19 ± 1.1 | * 15 ± 2.0 | * |
| A.G biomass | 21 ± 6.0 | 25 ± 5.2 | 27 ± 3.9 | 12 ± 1.0 | * 11 ± 1.9 | * |
| B.G biomass | 11 ± 1.0 | 16 ± 2.1 | * 14 ± 0.9 | * 7.1 ± 0.9 | * 4.2 ± 0.9 | ** |

505

506

507 Table 7. Macronutrient and metal concentrations (mg/kg) recorded in the below-ground
 508 phytomass of the microcosms. *: significant difference (95%) versus control.

| Element | Control | TDS | | EDTA | |
|---------|-------------|-------------|-------------|---------------|--------------|
| | | 25 g/kg | 50 g/kg | 0.5 g/kg | 1 g/kg |
| Ca | 4550 ± 2740 | 4250 ± 748 | 4230 ± 825 | 4650 ± 994 | 4090 ± 893 |
| Mg | 719 ± 58 | 538 ± 149 | 597 ± 82 | 772 ± 211 | 742 ± 354 |
| P | 1140 ± 117 | 936 ± 587 | 833 ± 7.1 | 1380 ± 738 | 1750 ± 1710 |
| K | 8050 ± 1070 | 4500 ± 2920 | 4720 ± 880 | 18700 ± 18500 | 7570 ± 3550 |
| Na | 605 ± 38 | 461 ± 74 | 533 ± 50 | 1131 ± 323 | * 1720 ± 707 |
| Al | 327 ± 194 | 1050 ± 726 | 495 ± 29 | 687 ± 374 | 1470 ± 547 |
| Fe | 913 ± 380 | 3370 ± 2380 | 8450 ± 3410 | * 3520 ± 1920 | 4630 ± 219 |
| Mn | 211 ± 127 | 435 ± 276 | 1180 ± 448 | * 604 ± 218 | 772 ± 351 |
| Zn | 342 ± 33 | 465 ± 188 | 715 ± 207 | * 567 ± 127 | 797 ± 196 |
| Pb | 251 ± 219 | 960 ± 587 | 1760 ± 1130 | * 1280 ± 559 | * 1720 ± 220 |
| Cu | 23 ± 1.3 | 19 ± 4.0 | 21 ± 8.8 | 35 ± 5.9 | * 41 ± 5.7 |
| Cd | 4.5 ± 1.4 | 9.4 ± 8.2 | 13 ± 5.3 | * 11 ± 6.7 | 13 ± 2.3 |
| Ni | 0.9 ± 1.0 | 3.0 ± 2.2 | 2.5 ± 1.2 | 5.1 ± 1.4 | * 7.9 ± 4.0 |

509

510

511 Table 8. Metal concentrations (mg/kg) recorded in the above-ground parts of some of the
 512 species growing in the microcosms.

| | | Zn | Pb | Cu | Cd |
|--------------------------------|----------|------|------|-----|------|
| <i>Agrostis castellana</i> | | | | | |
| Control | | 290 | 66 | 6.7 | 1.4 |
| TDS | 25 g/kg | 218 | 4.10 | 43 | 2.5 |
| | 50 g/kg | 203 | 3.30 | 6.6 | n.d. |
| EDTA | 0.5 g/kg | 953 | 125 | 2.5 | 2.0 |
| | 1 g/kg | 1170 | 145 | 5.0 | 2.0 |
| <i>Gallium parisiense</i> | | | | | |
| Control | | 400 | 43 | 2.0 | n.d. |
| EDTA | 0.5 g/kg | 483 | 60 | 5.1 | n.d. |
| <i>Lolium rigidum</i> | | | | | |
| Control | | 277 | 0.80 | 0.8 | n.d. |
| TDS | 25 g/kg | 108 | 0.70 | 1.0 | n.d. |
| | 50 g/kg | 8.5 | 0.70 | 0.9 | n.d. |
| <i>Plantago coronopus</i> | | | | | |
| Control | | 650 | 7.5 | 2.5 | n.d. |
| EDTA | 0.5 g/kg | 1250 | 118 | 255 | 3.0 |
| | 1 g/kg | 973 | 150 | 5.0 | 2.0 |
| <i>Scirpoides holoschoenus</i> | | | | | |
| Control | | 127 | 0.33 | 36 | n.d. |
| TDS | 25 g/kg | 71 | 12 | 1.0 | n.d. |
| | 50 g/kg | 69 | 7.2 | 5.4 | n.d. |
| EDTA | 0.5 g/kg | 216 | 25 | 14 | n.d. |
| <i>Vulpia myuros</i> | | | | | |
| Control | | 62 | 8.5 | 20 | n.d. |
| TDS | 25 g/kg | 140 | 2.3 | 7.5 | n.d. |
| | 50 g/kg | 165 | 2.5 | 6.2 | n.d. |

513

514

515 Table 9. Summary of the effects of the two amendments used in soils polluted
 516 with several metals

| Variable | TDS | EDTA |
|------------------------------|--|---------------------------------|
| Soil pH | Lowered | Lowered but less than with TDS |
| OM and N | Increased | Unchanged |
| Soil nutrients | Null or beneficial effects | Null or negative effects |
| Soil heavy metals | Unchanged | Unchanged |
| Nutrients in the 3 leachates | Increasing trend | Increased |
| Metals in the 3 leachates | Somewhat increased but less than with EDTA | Increased, especially Zn and Pb |
| Species richness | Unchanged | Lowered |
| Plant cover | Positive effects | Decreasing trend |
| Above-ground biomass | Unchanged | Negative effects |
| Root biomass | Positive effects | Negative effects |
| Root metals | Generally beneficial | Beneficial effects |

517

518

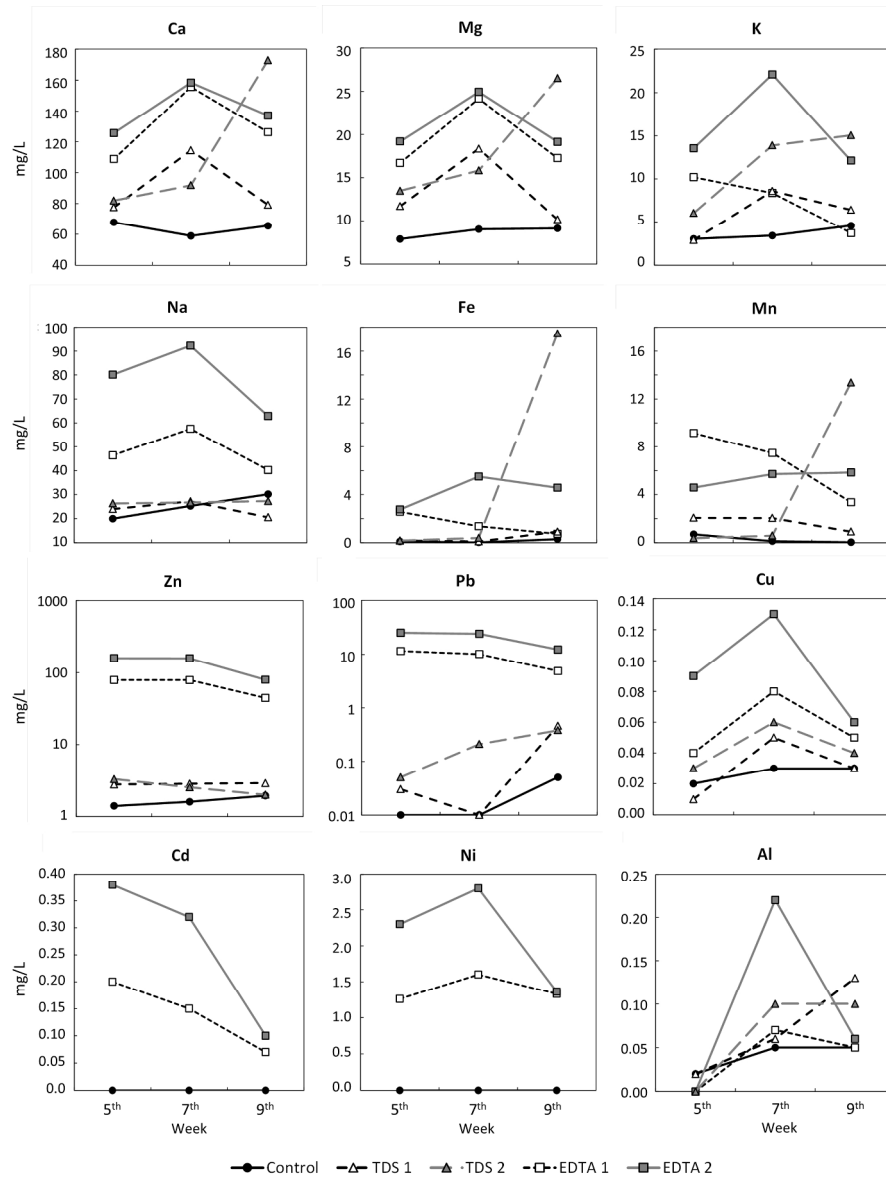


Figure 1. Mean concentrations (mg/L) of macroelements and metals in leachates from microcosms collected in weeks 5, 7 and 9 of the experiment. Values along the vertical axes in the Zn and Pb graphs are expressed as logs. Values of 0 mg/L are represented as 0.01. TDS 1 and 2 represent the results of the two TDS treatments (25 g/kg and 50 g/kg respectively) and EDTA 1 and 2 the results of the two EDTA treatments (0.5 g/kg and 1g/kg respectively).
190x254mm (300 x 300 DPI)

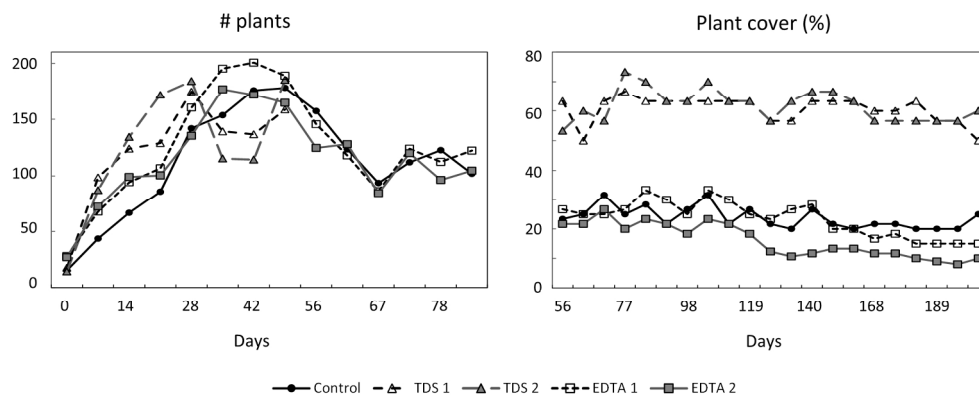


Figure 2. Changes produced in the number of plants and plant cover throughout the bioassay. TDS 1 and 2 represent the results of the two TDS treatments (25 g/kg and 50 g/kg respectively) and EDTA 1 and 2 the results of the two EDTA treatments (0.5 g/kg and 1g/kg respectively).
190x78mm (300 x 300 DPI)