

1 **Thylakoid peroxidase activity responsible for oxidised chlorophyll accumulation**
2 **during ripening of olive fruits (*Olea europaea* L).**

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12 **Abstract**

13 Type III peroxidase (EC 1.11.1.7) (POX) is the enzyme direct responsible of the 13² OH
14 chlorophyll formation on oxidative catabolism of chlorophylls (chls). Despite the higher content
15 of oxidised derivatives of chlorophylls (ox-chls) in fruits of the Arbequina variety compared to
16 Hojiblanca, the evolution of total chlorophyll oxidative peroxidase activity (POX-chl) showed
17 that this activity levels were higher in fruits of Hojiblanca compared with Arbequina variety.
18 Subsequently, a deepened study on the subcellular distribution of POX-chl activity from
19 mesocarp and epicarp cells of olive fruit of both varieties was made, founding that the POX-
20 chl activity located in thylakoid fraction (the only fraction in direct contact with chls *in vivo*),
21 was in Arbequina fruits higher than in Hojiblanca ones and involved more than 50% of the
22 membranous POX-Chl activity. It has been demonstrated also, that the evolution of the POX-
23 chl activity in thylakoid membranes enriched fraction throughout the whole life cycle, was
24 parallel with the formation and accumulation of ox-chls in olive fruits. Data allowed to conclude
25 that the formation of ox-chls during the chl catabolism is mediated by a POX-chl activity
26 localized in thylakoid fraction and allow to hypothesize that the high percentage of POX
27 activity found in the soluble cell fraction, estimated at 99.8%, may be involved in the loss of
28 pigmentation by oxidation occurring during fruit processing for obtaining olive oil.

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31 **Keywords:** *Olea europaea* L.; Olive peroxidase; Chlorophyll oxidative peroxidase;
32 Subcellular distribution; Thylakoid fraction; Chlorophyll catabolism; Fruit ripening.

33

34 1. INTRODUCTION

35 Chlorophyll pigments are highly appreciated as functional components in fruits
36 and vegetables both for its green colouring properties as its health benefits for the human
37 consumption derived from their biological properties (Ferruzzi & Blakeslee, 2007). In
38 addition, the ripening process of fruits or technological treatment for food production is
39 associated with chemical and/or enzymatic specific transformations of these pigments
40 making them quality indicators of end products and demonstrating a potential applicability
41 as a tool for traceability the processing (Gandul-Rojas, Roca & Mínguez-Mosquera,
42 2000).

43 The fruit of the olive tree (*Olea europaea* L.) is mainly used to obtain olive oil, a
44 natural food that is obtained from the olive fruit solely by mechanical or physical
45 processes. The fundamental steps of the process are crushing, malaxation and phase
46 separation by solid/liquid centrifugation systems, being malaxation the step in oil
47 extraction that especially modifies their qualitative and quantitative composition. After the
48 crushing step, with the broken plant tissue, and released chloroplasts and thylakoid
49 membranes, the lipophylic chlorophyll pigments, will transfer to the oil phase during
50 malaxation, giving different colours to olive oil. In this process the chlorophylls undergo a
51 series of structural changes mainly due to breakage of cellular structures that allow the
52 release of acid as well as greater accessibility of their natural substrates for endogenous
53 enzymes. Different reactions can occur as pheophytinization, phytol hydrolysis via
54 chlorophyllase and chemical and enzymatic oxidative reactions that lead to the formation
55 of coloured oxidized intermediates and non coloured final products. Chlorophyllic pigment
56 content of olive fruits is cultivar dependent and negatively correlated to fruit ripening
57 (Mínguez-Mosquera & Garrido-Fernández, 1989; Roca & Mínguez-Mosquera, 2003) and
58 the occurrence of lipophylic chlorophyll compounds in virgin olive oil can be related to the
59 activities of several endogenous enzymes present in the olive fruit such as POX
60 promoting chlorophyll oxidation (Gandul-Rojas, Roca & Mínguez-Mosquera, 2004). So
61 that the study of catabolic processes involved in degradation of these pigments is
62 particularly relevant this field of food technology to know biochemical fundamentals that
63 define the profile of these intrinsic minor components of virgin olive oil, potentially
64 applicable as a tool for olive oil traceability. There currently exists a general consensus
65 regarding the succession of enzymatic reactions implicated in the catabolism of the
66 chlorophyll a (Chl) molecule which takes place during leaf senescence. First step is the

67 removal of Mg by Mg-dechelating substance (MCS) (Shioi, Tomita, Tsuchiya & Takamiya,
68 1996), producing pheophytin *a* followed by the removal of phytol and the formation of
69 pheophorbide *a* in a reaction catalyzed by pheophytin pheophorbide hydrolase (PPH)
70 (Schelbert, Aubry, Burla, Agne, Kessler, Krupinska & Hörtensteiner, 2009). Then
71 pheophorbide is degraded to fluorescent chl catabolites (FCCs), which are primarily
72 colorless catabolites, via the enzymatic system pheophorbide *a* oxygenase (PaO) and red Chl
73 catabolite reductase (RCCR). Finally, outside the chloroplast, the FCCs are exported to
74 the cytosol where they are modified with different functional groups until no coloured chl
75 catabolites (NCCs), considered as the final catabolites of chls (Hörtensteiner & Kräutler,
76 2011).

77 There is much less research regarding the degradation systems which are
78 involved in the chlorophyll catabolism of the fruit. It could not be demonstrated the
79 functionality of PPH, checking (Shemer, Harpaz-Saad, Belausov, Lovat, Krokhin, Spicer,
80 Standing, Goldschmidt & Eyal, 2008) that the pathway starts with sequential action of
81 chlorophyllase (CHLASE) that catalyzes the phytol hydrolysis in chl *a* resulting
82 chlorophyllide *a* (Amir-Shapira, Goldschmidt, & Altman, 1987) and MCS that becomes to
83 pheophorbide *a* (Suzuki & Shioi, 2002). Besides these main steps of the PaO pathway,
84 additional/alternative reactions of Chl breakdown have been described. These were
85 (mostly) inferred from the identification of different types of Chl degradation products,
86 such as pyropheophorbide *a* (Suzuki, Amano & Shioi, 2006), Chl-derived monopyrroles
87 (Suzuki and Shioi, 1999), urobilinogenoidic catabolites (Losey & Engel, 2001), and
88 various oxidized chl derivatives with an intact porphyrin ring, that they could not be
89 located within PaO route.

90 There is evidence (Maunder, Brown & Woolhouse, 1983; Yamauchi, Akiyama,
91 Kako, & Hashinaga, 1997; Takahashi, Adachi, Furuta, Yamamoto, Kurata, Azuma,
92 Miyoshi & Shimokawa, 2001; Roca, Gandul-Rojas & Mínguez-Mosquera, 2007) which
93 suggests the hypothesis of the coexistence of a specifically oxidative metabolism in
94 parallel to the general catabolic route (Janave, 1997), peroxidase (POX) being the
95 enzyme which has been most clearly linked with this metabolism (Gandul-Rojas, Roca &
96 Mínguez-Mosquera, 2004; Huff, 1982; Matile, 1980; Roca et al., 2007).

97 Aljuburi, Huff & Hsieh (1979) were the first to describe an enzyme which exhibits
98 an oxidising and decolourizing action on the chlorophyll in the flavedo of oranges in the
99 presence of H₂O₂ and a phenol. However, Matile (1980) was the first to suggest that POX

100 is responsible for that activity by observing an *in vitro* modification to chl in the presence
101 of H₂O₂, 2,4-dichlorophenol (2,4-DCP) and horseradish peroxidase (HRP). These results
102 were confirmed by Huff (1982), ascertaining that the chl *a* molecule was degraded by
103 POX extracted from oranges (*Citrus sinensis* L.), in presence of various phenols, even
104 determining the activity with various chl substrates and estimating the corresponding
105 kinetic parameters (K_m and V_{max}).

106 Until recently it was assumed that peroxidase mediates *in vitro* Chl degradation
107 only in the presence of phenolic compounds, being the highest enzyme activities in plant
108 systems found with *p*-coumaric acid, apigenin and naringenin, which have a hydroxyl
109 group at the *p*-position (Kato & Shimizu, 1985; Yamauchi & Eguchi, 2002; Yamauchi &
110 Minamide, 1985). Hynninen, Kaartinen, & Kolehmainen (2010) have defended the
111 opposing hypothesis demonstrating that, *in vitro*, HRP can oxidise chl *a* in the absence of
112 phenols, and for this reason propose a direct reaction between the chlorophyll molecule
113 and POX, in which chl *a* would act as a reducer substrate, rendering the presence of a
114 second electron donor substrate totally unnecessary. Recently, Vergara-Domínguez,
115 Roca & Gandul-Rojas, (2013) demonstrated that POX is capable of directly oxidising chl
116 *a* in the absence of another electron donor, however the presence of a cosubstrate donor
117 like 2,4-DCP accelerates the oxidation rate of the chlorophyllic substrate catalyzed by
118 POX.

119 Due to the multitude of isoforms of POX described in the bibliography (Akiyama &
120 Yamauchi, 2001; Funamoto, Yamauchi, & Shigyo, 2003; Gandul-Rojas et al. 2004;
121 Saraiva, Nunes & Coimbra, 2006), the study of the intracellular distribution of the POX-chl
122 activity was approached from different perspectives. Some papers demonstrate the
123 existence of this activity in thylakoid membranes (Martinoia, Dalling & Matile, 1982,
124 Johnson-Flanagan & Spencer, 1996; Gandul-Rojas et al., 2004); however, data exist to
125 demonstrate the presence of this activity in other subcellular fractions (Akiyama &
126 Yamauchi, 2001; Funamoto et al., 2003).

127 Akiyama & Yamauchi (2001), studying the distribution of different enzymes
128 involved in the degradation of chlorophylls and applying differential centrifugal techniques
129 demonstrated that, of all the POX-chl activity present in the cotyledon cells of the radish
130 (*Raphanus sativus* L.), only a percentage of less than 1 % were located in the
131 chloroplasts, whereas the rest was found in other cellular organelles. This fact was
132 understood as evidence that the majority of the POX-chl present in the cell is found on

133 the exterior of the chloroplast and that only a minority fraction is found in the interior of
134 this organelle, which would be the part potentially involved in the oxidative degradation of
135 chlorophyll that takes part *in vivo*. Similar results were obtained by Funamoto et al. (2003)
136 on measuring POX-chl activity in different subcellular fractions of broccoli florets
137 (*Brassica oleracea* L.). It was established that the microsomal and cytosolic fraction made
138 up approximately 95 % of the entire POX-chl activity present in the cell, whereas the
139 activity measured in the chloroplast only represented 1 %.

140 To date, there is no information regarding the distribution of POX-chl activity in
141 fruit. Nevertheless, the bibliography contains studies of the distribution of the POX activity
142 relating to the oxidation of phenols (POX-phe), although the data obtained are
143 contradictory. On the one hand, Civello, Martínez, Chaves & Añón, (1995), studying the
144 distribution of the POX activity against guaiacol in *Fragaria ananassa* (*Duch.*), found that
145 the specific POX-phe activity of the fraction linked to membranes was far higher than that
146 measured in the soluble fraction and was equivalent to approximately 95% of the total
147 POX-phe activity. On the other hand, Estrada, Bernal, Díaz, Pomar & Merino (2000),
148 studying the distribution of this activity in *Capsicum annuum*, (L.) from an organelle
149 homogenate, found that the POX activity measured over 4-methoxy-R-naphthol (4MN) in
150 the soluble fraction was always higher than that obtained in the rest of the subcellular
151 fractions (ionic bonding to membranes, ionic bonding to cellular wall and covalent
152 bonding to cellular wall) and that it increased from 77% to 96% with ripening.

153 An interesting model for the study of the physiological involvement of POX in chl
154 metabolism is provided by certain varieties of olive given that differences in the oxidised
155 chl profile for this fruit based on the variety have already been detected (Roca et al.,
156 2007). In general, in olive fruit varieties the qualitative composition of pigments is
157 basically the same, and is not subject to modification with the advancement of maturity in
158 the fruit (Roca & Mínguez-Mosquera, 2001). In contrast, in the variety Arbequina an
159 accumulation of 13² OH chl *a* has been detected during the transition from the periods of
160 growth through to maturity, reaching up to 15% of the chl fraction (Roca & Mínguez-
161 Mosquera, 2003), while in other varieties such as Hojiblanca and Picual the presence of
162 this compound represents 1% of the total chl fraction during the complete fruit life cycle.

163 The study of the involvement of POX-chl activity in the metabolism of chls in the
164 Hojiblanca variety has revealed that there is a correlation between the formation and/or
165 build-up of certain oxidized chl derivatives and the total POX-chl activity levels measured

166 from protein precipitated (Vergara-Domínguez, Gandul-Rojas & Roca, 2011). However,
167 in varieties with a high level of chlorophyllase (CHLASE) activity, such as Arbequina, it
168 was not possible to establish this correlation, and an *in vitro* competence between POX
169 and CHLASE was revealed towards the chl *a* substrate. Subsequently, (Vergara-
170 Domínguez et al., 2013), in an optimisation of the *in vitro* measurement of the POX-chl
171 activity established new measurement conditions that permitted the displacement of the
172 competence between CHLASE and POX towards the chl *a* substrate, in favour of the
173 latter.

174 These new measurement conditions permitted the study of the involvement of
175 POX-chl activity in the oxidation metabolism for chls in fruits with high CHLASE activity to
176 be studied, in order to establish distinctions between varieties with different oxidative chl
177 profiles (Arbequina vs. Hojiblanca). Consequently, in this paper a study has been made of
178 the subcellular distribution of the POX-chl activity in olives from both varieties and this
179 compared with the total measured for this activity, based on a protein precipitated.
180 Subsequently, the POX-chl activity specifically obtained in the enriched fraction in the
181 thylakoid membranes was monitored, to look for a correlation between the enzymatic
182 activity and the build-up of oxidized chl derivatives.

183 **2. MATERIAL AND METHODS**

184 **2.1 Plant material**

185 The work was carried out with two different varieties of fruit from the olive tree
186 (*Olea europaea* L.), Hojiblanca and Arbequina, from two plantation of young olive trees in
187 “Cortijo Carcahueso” (Aznalcóllar, Sevilla, Spain) and Hacienda Guzmán (Aceites del
188 Sur-Coosur S.A.). Six olive trees per variety were selected during the 2010–2011 and
189 2011-2012 seasons, and they were sampled every two weeks from mid-July (immature
190 fruit) to mid-December (mature fruit). The sampling was performed at the same time each
191 day, by randomly collecting 2 kg of fruit from branches within arm’s reach and around the
192 circumference of the tree, both the outer and the interior.

193 **2.2 Chemicals and standards.**

194 Tetrabutylammonium acetate and ammonium acetate were supplied by Fluka
195 (Zwijndrecht, TheNetherlands). HPLC reagent grade solvents were BDH-Prolabor (VWR
196 International Eurolab, Barcelona, Spain). For the preparation, isolation, and purification of
197 chl pigments, analytical grade solvents and reagents were used (Scharlau, Microdur,

198 Sevilla, Spain). The deionized water used was obtained from a Milli-Q 50 system
199 (Millipore Corp., Bedford, MA).

200 Chl *a* and *b* were isolated from fresh spinach leaves by means of acetone
201 extraction and transfer to diethyl ether, and subsequently were separated and isolated by
202 TLC on silicagel GF254 using the developing mixture light petroleum ether 65–
203 95 °C/acetone/diethylamine (10:4:1 v/v/v). Bands for chl *a* and *b* were scraped off at R_f
204 0.51 and 0.54 respectively, and eluted in acetone (Mínguez-Mosquera & Garrido-
205 Fernández, 1989). Pheophytin (Phy) *a* or *b* was prepared from pure chl *a* or *b* solution in
206 diethyl ether, by acidification with a few drops of HCl 10% (v/v) (Sievers & Hynninen,
207 1977). The resulting phy *a* or *b* was transferred to diethyl ether, washed with distilled
208 water, dried over anhydrous Na₂SO₄, evaporated and dissolved in acetone. The 13² OH
209 chl *a* and *b* were obtained by selenium dioxide oxidation of related chl (*a* or *b*) at reflux-
210 heating for 4 h in pyridine solution under argon (Laitalainen, Pitkänen & Hynninen, 1990).
211 15¹ OH lactone chl *a* and *b* were obtained by alkaline oxidation in aqueous medium. For
212 this purpose, solid and chromatographically pure chl (*a* or *b*) was dissolved in acetone
213 and mixed with 0.5% NaOH and exposed to atmospheric oxygen at room temperature for
214 10 min. The resulting oxidation products were transferred to diethyl ether by addition of
215 water saturated with NaCl, and 15¹ OH lactone chl (*a* or *b*) was isolated by NP-TLC and
216 semi-preparative HPLC (Mínguez-Mosquera, Gandul-Rojas & Garrido-Fernández, 1996).

217 **2.3 Pigment extraction**

218 Samples were taken from a homogenised triturate, prepared from 100 de-stoned
219 fruits (ca. 40 g) of the most representative size by accurately weighing from 4 to 15 g for
220 each analysis depending on the degree of ripeness of the fruits. Pigments were extracted
221 with N,N-dimethylformamide (DMF) saturated with MgCO₃ (Mínguez-Mosquera &
222 Garrido-Fernández, 1989). The solid residue was collected by vacuum filtration and the
223 extraction repeated until filtrates were colourless. The extracts combined in a funnel were
224 repeatedly treated with hexane. Chls, chl derivatives and xanthophylls were retained in
225 DMF phase. The hexane phase contained lipids and carotenes was discarded. The DMF
226 phase was treated with 10% (w/v) NaCl solution at 0 °C and the chls and xanthophylls
227 transferred to 100 ml of a mixture of diethyl ether/hexane (1:1 v/v). The aqueous layer
228 was washed with diethyl ether and finally discarded, eliminating polyphenols and other
229 water-soluble compounds. The combined organic phases were filtrated through
230 anhydrous Na₂SO₄ and evaporated to dryness under vacuum at a temperature below 30

231 °C. The dry residue was dissolved in 1.5 ml acetone prior to HPLC. Analysis was
232 immediate or followed storage at -20 °C not more 18 h. All analyses were performed
233 under green light.

234 **2.4 Obtaining the crude enzyme extract**

235 Acetone powders were obtained (Terpstra & Lambers, 1983) from 25 g pitted and
236 chopped fruit mixed with 20 volumes of acetone at -20 °C, which was ground with a
237 Polytron homogeniser on an ice bath. The solid residue was recovered by vacuum
238 filtration and treated similarly with 8 volumes of acetone (200 mL). This process was
239 repeated until colourless extracts were obtained (4 washes were usually sufficient).
240 Finally the precipitate was collected, dried at room temperature, weighed and stored at -
241 20 °C until used (Mínguez-Mosquera, Gandul-Rojas & Gallardo-Guerrero, 1994). The
242 crude enzyme extract was obtained by modifying the method developed by Johnson-
243 Flanagan & Thiagarajah (1990). The acetone powder (0.2 g) was mixed with 6 mL of
244 extraction buffer (5 mM Na-Pi pH7, 50 mM KCl and 0.24%(w/v) Triton X-100), and
245 magnetically stirred at room temperature for 1 h (Roca et al., 2007). The resulting mixture
246 was filtered through four layers of cotton gauze and centrifuged at 4500 x g for 5 min at 4
247 °C. The supernatant was the crude enzyme extract.

248 **2.5 Obtaining subcellular fractions**

249 The different subcellular fractions were obtained by cold (4°C) differential
250 centrifugation, similar to the specific methodology used for olives established by Salas,
251 William, Harwood & Sánchez, (1999) (Figure 1).

252 Between 8 and 10 g of pitted olives were immediately immersed in a 50 mM K-Pi
253 pH 7.0 buffer solution with 0.1% of ascorbate to prevent oxidation of the tissues. After
254 removing the excess buffer solution from the tissue, the sample was crushed four times in
255 the presence of 15 ml of a 50 mM Tricine-KOH pH 7.6 buffer solution, with 330 mM
256 sorbitol, 20 mM KCl, 2 mM MgCl₂, 5mM EDTA, 7 mM 2-mercaptoethanol, 3 mM
257 dithioerythritol, 2 mM metabisulphite, 0.1 % ascorbate and 10 % glycerol, to which 5 g of
258 polyvinyl polypyrrolidone washed with acid was added. Next it was filtered through 2
259 layers of Miracloth and centrifuged at 100 x g for 3 min. The resultant pellet, which
260 contained traces of cellular tissue and nuclei, was rejected and the supernatant subjected
261 to repeated centrifugations at 4200 x g for 10 min, 15,000 x g for 10 min (15 K) (enriched
262 fraction in low density membranes), 40,000 x g for 20 min (40 K) (enriched fraction in high
263 density membranes) and 150,000 x g for 80 min (150 K) (microsomal fraction). The

264 pellets resulting from the 15 K, 40 K and 150 K centrifugations were resuspended in 300
265 μL of incubation buffer solution, 100 mM Na-Pi pH 7, and quickly frozen in N_2 and stored
266 at $-80\text{ }^\circ\text{C}$. The supernatant obtained after the 150 K centrifugation was considered as the
267 soluble fraction.

268 The pellet from the centrifugation at $4200 \times g$ was resuspended with 1.5 ml of
269 resuspension buffer solution (50 mM Tricine-KOH pH 7.6, with 330 mM sorbitol,
270 containing 10 % glycerol) and centrifuged again at $6800 \times g$ for 4 min. The supernatant
271 was rejected and the pellet was considered as the chloroplast and thylakoid-enriched
272 fraction (7K). From this fraction, following the methodology described by Hörtensteiner,
273 Vicentini, & Matile, (1995), the thylakoid membranes-enriched fraction (7K*) was
274 obtained by resuspending the 7K fraction in a 100 mM Na-Pi pH 7 buffer solution and
275 then centrifuging at $20800 \times g$ for 5 min. The supernatant was rejected whereas the pellet
276 formed the thylakoid membrane enriched fraction (7K*) which was frozen in liquid N_2 and
277 stored at -80°C .

278 The entire procedure took place in a chamber at $4\text{ }^\circ\text{C}$.

279 **2.6 Determination of POX-chl enzymatic activity**

280 To measure the POX-chl activity, a standard reaction mixture (0.36 mL) was used
281 containing $80\text{ }\mu\text{L}$ crude enzyme extract, $239\text{ }\mu\text{L}$ incubation buffer (100 mM Na-Pi pH 7),
282 $15\text{ }\mu\text{L}$ chl a (4 mM) dissolved in acetone and $15\text{ }\mu\text{L}$ 2,4-dichlorophenol (DCP) (2%, w/v)
283 dissolved in acetone. The reaction was started by adding $11\text{ }\mu\text{L}$ H_2O_2 (0.2%, w/v).

284 To determine the subcellular distribution of the POX-Chl activity, the pellets from
285 the corresponding membrane protein fractions (fraction 7K*, 15K, 40K and 150K)
286 obtained by differential centrifugation, were resuspended in $239\text{ }\mu\text{L}$ of incubation buffer
287 solution whereas one aliquot of the soluble protein solution ($80\text{ }\mu\text{L}$) was mixed with $159\text{ }\mu\text{L}$
288 of the same buffer solution. In both cases, the reaction mixture was obtained by adding to
289 each corresponding protein fraction of $239\text{ }\mu\text{L}$, $80\text{ }\mu\text{L}$ of extraction buffer solution (5 mM
290 Na-Pi pH7, 50 mM KCl and 0.24% (w/v) Triton X-100) and the different substrates (chl a,
291 DCP and H_2O_2) in the aforementioned volumes (Table 1).

292 The reaction was carried out with strong stirring, at $30\text{ }^\circ\text{C}$ and in conditions of
293 darkness, taking aliquots at the initial time and after 50 min of reaction (for enzyme
294 solubilized from acetone powder) or 6 hours (for subcellular fractions). The reaction was
295 concluded with the addition of $300\text{ }\mu\text{L}$ of acetone to an aliquot portion of $75\text{ }\mu\text{L}$ of the

296 reaction mixture. After vigorous vortex stirring, it was immediately frozen at -30 °C until
297 carrying out of a quantitative analysis of the reaction products by HPLC.

298 **Table 1**

299 Composition of reaction mix prepared to quantify the POX-chl activity in different
300 subcellular fractions of olive fruits^a.

301

	Sample	Extraction buffer ^b (μ L)	Incubation buffer ^c (μ L)
7K* fraction ^d	pellet	80	239
15K, 40K, 150 K fractions	239 μ L	80	-
Soluble fraction	80 μ L	80	159

302 ^aAll reaction mixtures also contained 11 μ L of hydrogen peroxide at 0.2% (v/v), 15 μ L of
303 2,4-dichlorophenol at 2% in methanol (w/v) and 15 μ L of 4 mM chlorophyll *a* in acetone in
304 a total volume of 360 μ L. ^bExtraction buffer: 5 mM Na-Pi pH 7 with 50 mM KCl and 0,24
305 % (p/v) of Tritón X-100; ^cIncubation buffer: 100 mM Na-Pi pH 7. ^dThe chloroplast fraction
306 (7K) was resuspended in 100 mM Na-Pi pH 7 and centrifuged for 5min at 20,800xg to
307 obtain a enriched fraction in thylakoid membranes (7K*) (Hörtensteiner et al.1995).

308

309 Measurements from crude enzyme extracts were made in quadruplicate and
310 measurements from subcellular fraction were made in duplicate, both with their
311 corresponding blank tests with deactivated enzyme (15 min at 100 °C), in order to
312 evaluate the possible chemical autoxidation of the chl pigments.

313 The peroxidase activity was defined as the formation of oxidised chl derivatives
314 expressed in nkatal in relation to the amount of protein in the crude extract enzyme or
315 proteic fraction. The Katal unit of activity is defined as the quantity of enzyme necessary
316 to form a mole of oxidised chl derivatives per second.

317 **2.7 Determination of protein concentration**

318 Measurements of the amount of proteins were made in crude enzyme extracts
319 and subcellular fractions. The amount of protein was determined using a comercial
320 bicinchoninic acid kit for protein determination (Sigma-Aldrich) based on the Lowry's
321 method (Lowry, Rosebrough, Farr & Randall, 1951).

322 **2.8. Statistical data analysis**

323 Data were expressed as mean values \pm standard deviation (SD) and were
324 analyzed for differences between means using one-way analysis of variance (ANOVA).
325 The analyses were performed using Statistica for Windows (version 6, StatSoft, 2001)

326 and the t-Student test was used as a post hoc comparison of statistical significance (p
327 values < 0.05).

328 **3. RESULTS AND DISCUSSION**

329 **3.1. Evolution of the total POX-chl activity**

330 Difficulties in measuring POX-chl activity in plant tissues with a high level of
331 CHLASE activity (Vergara-Domínguez et al., 2011) resulted in the need to define a
332 methodology which, on using DCP as a cosubstrate reducer (Vergara-Domínguez et al.
333 2013), would permit the elimination of the competence between POX-chl and CHLASE.

334 The evolution of the POX-chl activity levels in the Arbequina variety with the new
335 methodology in the absence of CHLASE activity (Figure 2, continuous line), displayed a
336 relatively high period of activity in the months of August and September, coinciding with
337 the higher levels of CHLASE activity (Vergara-Domínguez et al., 2011), followed by a
338 period of declining levels of activity. The improvement in the determination of the POX-chl
339 activity was evident when compared with the evolution of POX-chl determined using the
340 previous methodology (Figure 2, discontinuous line), where for this period of ripening, the
341 POX-chl activity was minimum (approximately 1 nkat/g protein), due to the clear
342 interference of CHLASE and demonstrating the infra-valuation of the activity that was
343 taking place to date.

344 In the Hojiblanca variety, the evolution of the POX-chl activity, measured in the
345 absence of CHLASE activity (Figure 2), did not display any significant differences to the
346 evolution of the levels of said activity, previously measured in the presence of this
347 enzymatic competence (Vergara-Domínguez et al., 2011), as in the fruit from this variety
348 the levels of CHLASE activity are not significant in comparison with the levels of POX-chl.

349 Comparing varieties, the values of POX-chl activity in the Arbequina variety were
350 lower than those found in the Hojiblanca variety. The new measure of POX-chl activity
351 permitted a more exact quantification of this activity in the Arbequina variety, by
352 eliminating interference from the CHLASE activity (Vergara-Domínguez et al., 2013).
353 Nevertheless, the results obtained did not provide an explication for the higher build-up of
354 oxidized chl derivatives appearing in fruit from this variety (13%) with respect to fruit from
355 Hojiblanca (3%) (Vergara-Domínguez et al. 2011), making it necessary to adopt a new
356 approach to find an answer to this problem.

357 Given the wide distribution of POX in the different cellular compartments, the
 358 compartmentalisation of POX-chl activity in the different fractions from the mesocarp and
 359 epicarp cells of olives was studied in order to find, where applicable, possible differences
 360 between the Arbequina and Hojiblanca varieties, that would justify the differences existing
 361 in the profile of oxidized chl derivatives.

362 **3.2 Subcellular distribution of the POX-chl activity**

363 Using a homogenisation methodology especially defined for olive pulp by Salas et
 364 al., 1992 (Figure 1), followed by differential centrifugation, the following enriched
 365 subcellular fractions were obtained: thylakoid fraction (7K*), mitochondrial fraction (15K),
 366 high density membrane fraction (40K), low density membrane fraction (150K) and soluble
 367 fraction. Table 2 shows the distribution of the POX-chl activity measured in the two
 368 varieties of olive under study. It can be seen that the total of POX-chl activities measured
 369 in each of the subcellular fractions is significantly higher in the Hojiblanca variety with
 370 respect to the Arbequina variety, coinciding with the results described in section 3.1
 371 (Figure 2). In addition, the individualised study of the levels of specific POX-chl activity in
 372 each of the subcellular fractions also reveals significant differences between varieties. In
 373 the 15 K, 40 K, 150 K and soluble fractions, the values were always significantly higher in
 374 the Hojiblanca variety (Table 2), whereas specifically in the thylakoid membrane fraction
 375 this relation was inverted, and the Arbequina variety displayed a significantly higher level
 376 of POX-chl activity.

Table 2

POX-chl activity associated to subcellular fractions from *O. europea* L. fruits (cvs. Arbequina and Hojiblanca).

Material	POX-chl activity ^a	
	Arbequina variety	Hojiblanca variety
Membranous fractions ^b		
7K*	3.620 ± 0.217 ^a	2.880 ± 0.229 ^e
15 K	0.860 ± 0.005 ^b	2.430 ± 0.071 ^f
40 K	0.970 ± 0.038 ^c	2.820 ± 0.292 ^e
150 K	0.970 ± 0.023 ^c	1.750 ± 0.127 ^g
Soluble fraction	2632.590 ± 276.790 ^d	5450.700 ± 125.150 ^h
Total	2639.020 ± 277.073 ^d	5460.580 ± 125.750 ^h

^aData, expressed as nkat/kg of protein, are means of two determinations ± SD. Different letters represent significant differences among materials. Enriched fraction in thylakoid membranes (7K*) obtained as described in table 1.

377

378 Expressed as a percentage, in both varieties, the POX-chl activity found in the
379 soluble fraction was approximately 99.8 % of the total activity present in the cell and,
380 therefore, only the 0.2% remaining was distributed in the subcellular fractions associated
381 with the membranes (7K*, 15 K, 40 K and 150 K fractions). These results are similar to
382 those observed in the cotyledon cells of the radish (*Raphanus sativus* L.) (Akiyama &
383 Yamauchi, 2001) and of broccoli florets (*Brassica oleracea* L.) (Funamoto et al., 2003),
384 where it was established that of all the POX-chl activity present in the cells, only a
385 percentage of less than 1 % was located in the chloroplasts, whereas the rest was mainly
386 found in the soluble fraction.

387 In the processing of the olive fruit for oil extraction, the chl pigment transferring to
388 the oily phase is only partial despite its lipophilic nature. In the industrial processing is
389 estimated that around 60 % may be retained on the solid phase (pomace) and
390 approximately 20% is oxidized to non coloured products, and only 20 % is transferred to
391 the oil (Gallardo-Guerrero, Roca & Mínguez-Mosquera 2002). Thus the final
392 concentration of chl pigment in the virgin olive oil is also strongly affected by processing
393 parameters such as temperature, time or exposure to air, which could modulate the POX-
394 chl activity during the crushing and malaxation processes to obtain virgin olive oil causing
395 chl oxidation. In addition to the high percentage of POX-chl activity that is present in the
396 soluble cell fraction according to our results (99.8%), this oxidation process may also
397 involve the POX-chl from the seed because the fruit is not usually destoned for oil
398 extraction and in accordance with Luaces, Romero, Gutierrez, Sanz & Pérez (2007) olive
399 seed POX would represent more than 99% of total POX in the olive paste used in
400 process for olive oil.

401 Regarding varietal differences, although our results showed levels of total POX-
402 chl activity in Hojiblanca variety upper than in Arbequina, to date there are not
403 experimental data to estimate the percentage loss of chl pigments by oxidation during the
404 process of virgin olive oil obtaining, in operation comparable conditions for these two
405 varieties of olives.

406 Nevertheless, there are significant differences between varieties with respect to
407 the distribution of the POX-chl activity in the fractions associated with membranes, and
408 particularly the activity found in the thylakoid fraction (7K*) of both varieties. Whereas in
409 the Arbequina variety, the POX-chl activity located in this fraction was equivalent to
410 56.39 % of all the POD-chl activity associated with the membranes, in Hojiblanca this
411 percentage was significantly lower and only amounts to 29.14 %. The percentage of

412 POX-chl activity found in the mitochondrial fraction (15K) was of the same order as that
413 associated with the high density membranes (40K), representing a higher percentage in
414 fruit from the Hojiblanca variety in comparison to the same fractions from fruit from the
415 Arbequina variety. The distribution value corresponding to the microsomal fraction (150K)
416 did not display significant differences between the two varieties.

417 Given the thylakoid location of the chls, the most plausible explanation is that the
418 POX-chl activity associated with the soluble fraction is not responsible for the formation of
419 the oxidized derivatives of chlorophyll that appear *in vivo* in the mesocarp and epicarp
420 cells of the olive. It is the activity associated with the membranes, more specifically the
421 thylakoid membranes, that is potentially directly linked to the oxidation reactions of chl
422 that appear *in vivo* (Akiyama & Yamauchi, 2001). In this respect, it should be noted that
423 the level of POX-chl activity associated with the thylakoid fraction (7K*) in the Arbequina
424 variety was significantly higher than that found in the Hojiblanca variety (Table 2), in
425 addition to representing more than half of the total percentage of the POX-chl activity
426 associated with the membranes.

427 *In vivo*, the concentration of chls in the Hojiblanca fruit (297.05 ± 14.99 mg/kg dry
428 weight) was far higher than that of the Arbequina fruit (72.43 ± 7.01 mg/kg dry weight) in
429 line with previous results (Roca et al., 2001 & 2003). Given this factor, on expressing the
430 POX-chl activity in relation to the chl concentration present in the fruit, for the Arbequina
431 variety, the activity calculated in the thylakoid fraction was 2.01 nkat/g of chls, whereas
432 for the Hojiblanca variety it was 0,26 nkat/g of chl. Consequently, in relation to the
433 quantity of pigment present in the fruit, the Arbequina variety had 10 times more POX-chl
434 activity in the thylakoid fraction than the Hojiblanca variety.

435 To conclude, although the total level of POX-chl activity in fruit from the Hojiblanca
436 variety was greater than that of the Arbequina fruit, the study of the distribution of this
437 activity in the different subcellular fractions has shown that the thylakoid fraction of the
438 Arbequina fruit has a significantly higher level of POX-chl activity than that found in the
439 Hojiblanca variety. This led to the linking of the formation and build-up of oxidized
440 derivatives of chl that takes place during the ripening of the fruit of this variety to the high
441 levels of POX-chl activity observed in its thylakoid fraction.

442 **3.3 Evolution of the oxidized derivatives of chlorophyll and the POX-chl activity in**
443 **thylakoid membranes.**

444 The comparison of the POX-chl activity located in the thylakoid fraction (7K*) and
445 the build-up of oxidized derivatives during the lifecycle of the Arbequina fruit (Figure 3)
446 showed that the maximum presence of oxidized derivatives of chl, which occurred in
447 samples 5 and 7, coincided with a clear increase in the POX-chl activity measured in the
448 thylakoid fraction. Nevertheless, between samples 8 and 10, coinciding with very
449 advanced stages of ripeness of the fruit, high levels of POX-chl activity were observed,
450 not concomitant with a high presence of oxidized derivatives in the chl fraction. This may
451 be due to the drop in the chl concentration typical of this stage of ripeness (< 32 mg of
452 chl/kg dry weight), which would imply that although in the thylakoid fraction a high level of
453 POX-chl activity is potentially observed, the low availability of the chl substrate would
454 impede the formation of these oxidized derivatives of chl. The same kinetics is also
455 followed by another chl-degrading enzyme as CHLASE (Roca et al., 2003) in Arbequina
456 fruits. In the Hojiblanca variety, although the presence of oxidized derivatives is much
457 lower than that of the Arbequina variety and does not exceed 3%, the POX-chl activity
458 measured in the thylakoid fraction displayed good correlation to the percentage of these
459 compounds, both parameters exhibiting a maximum at the end of the ripening period
460 (sample 8).

461 In conclusion, the highest levels of the POX-chl activity measured in the thylakoid
462 fraction (7K*) of the Arbequina and Hojiblanca fruit were concomitant with the greatest
463 rates of formation and build-up of oxidized derivatives of chl. Therefore, these figures
464 suggest that a POX-chl activity located in the thylakoid fraction is directly implicated in the
465 formation of these chl catabolites.

466 Nevertheless, although the levels of POX-chl activity measured in the thylakoid
467 fraction (7K*) in the Arbequina fruit were higher than those measured in the Hojiblanca
468 fruit, these differences would appear to be insufficient for explaining the wide margin that
469 distinguishes the build-up of oxidized derivatives of chl in both varieties. These results
470 permit the hypothesis of the existence of a regulation at post-transcriptional level of the
471 POX-chl activity that might differentiate the two varieties. This regulation may be more
472 intense in the Hojiblanca variety, such that the increases of POX-chl activity measured *in*
473 *vitro* in the thylakoid fraction do not correspond to a significant build-up of oxidized
474 derivatives of chlorophyll *in vivo* as occurs in the Arbequina variety.

475 Information about the enzymes involved in the oxidation of the chlorophyllic
476 pigments in plant tissues *in vivo* is of huge importance in the food technology area for

477 preventing alterations to their profile during the production, harvest, storage and
478 processing of foods of plant origin. Regardless of varietal differences found, data of the
479 comparison of the POX-chl activity located in the thylakoid fraction and the build-up of
480 oxidized derivatives during the lifecycle of the olive fruits leads to conclude that POX
481 activity localized in chloroplasts is directly involved in the oxidative metabolism of
482 chlorophyll that occurs during fruit ripening. Also, the study of subcellular distribution
483 revealed that a very high percentage of the POX-Chl total activity (99.8%) was localized
484 in the soluble fraction and thus would not be involved in this chl catabolism, because
485 chlorophyll substrate and enzyme activity are in different subcellular compartments.
486 However, this potential activity POX-chl can have a high degree of involvement in the
487 oxidative degradation of these compounds when the loss of cellular compartmentalization
488 happen in the fruit and there is greater accessibility of chlorophyll substrate. This occurs
489 in the last stages of ripening, in senescence processes during storage of the fruit prior to
490 processing or in crushing and malaxation process in olives for olive oil extraction. The
491 importance of the control of these enzymatic oxidation reactions in order to preserve both
492 colouring and functional properties in raw material for olive oil is disclosed.

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500

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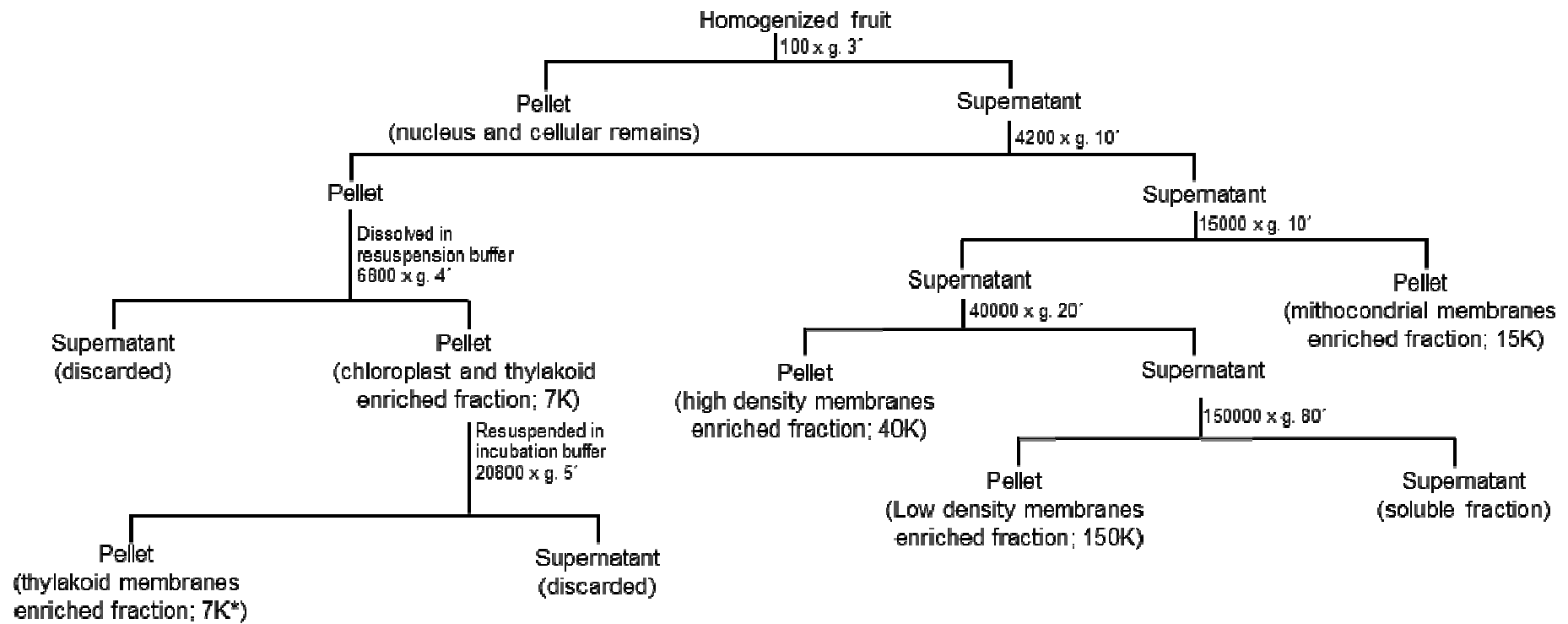
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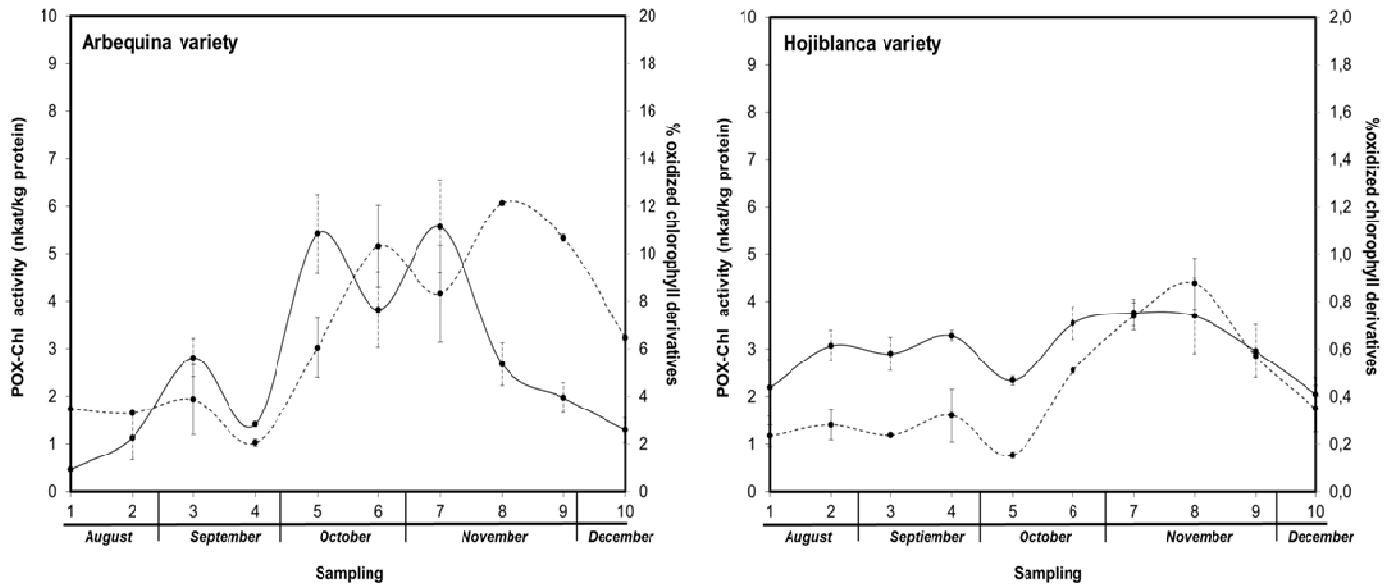
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638 **Figure 1.** Specific methodology to obtain different subcellular fractions from fruit of mesocarp *O. europaea* L. Based in
 639 Salas et al., (1999) and Hörstensteiner et al., (1995).



640

641 **Figure 2.** POD-chl activity evolution measured in solubilised enzyme extracts from
 642 acetone powder from *O. europaea* L. (cvs. Arbequina and Hojiblanca). In
 643 presence of competition with CHLASE activity (Vergara-Domínguez et al., 2011) (-
 644 -▲-) and in absence of competition with CHLASE activity (-■-). Mean values and
 645 standard deviations for n = 4.



646 **Figure 3.** POD-chl activity evolution located in thylakoid fraction (7K*) (-▲-) and
 647 percentage of oxidised chlorophyll derivatives(--■-) from *O. europaea* L. fruits
 648 (cvs. Arbequina and Hojiblanca). Mean values and standard deviations for n = 2.

