

Crossed Intramolecular Rauhut-Currier-Type Reactions via Dienamine Activation

Eugenia Marqués-López, Raquel P. Herrera, Timo Marks, Wiebke C. Jacobs, Daniel Könning, Renata M. de Figueiredo, and Mathias Christmann*

TU Dortmund University, Otto-Hahn-Str. 6, D-44227 Dortmund, Germany

mathias.christmann @tu-dortmund.de

Received Date (will be automatically inserted after manuscript is accepted)

ABSTRACT



The intramolecular Rauhut-Currier reaction creates a carbon-carbon bond between two tethered Michael-acceptors. Previous asymmetric versions have relied on 1,4-additions of chiral nucleophilic catalysts. Herein, we investigate a novel strategy that involves the formation of electron rich dienamines as key intermediates. Our methodology provides an efficient entry to the iridoid framework.

The use of functionalized chiral scaffolds as switches for biological signaling pathways has spurred the development of new reactions for their rapid assembly by means of asymmetric synthesis.¹ In this context, organocatalysis² complements existing transition metalbased catalysis allowing for the enantioselective synthesis of densely functionalized hetero- and carbocyclic motifs with often orthogonal chemoselectivity.³ While α , β -unsaturated aldehydes⁴ **1** are traditionally used as a^{1}/a^{3} -synthons (Scheme 1), organocatalysis has added new value to this important structural class. In addition to the long-known 1,4-addition of nucleophilic catalysts (**1** \rightarrow **2**) such as tertiary phosphines, 1,2-additions have recently allowed for new chemoselective transformations. *N*-heterocyclic carbenes have converted **1** into homoenolate equivalents⁵ **3** and secondary amines have been shown to convert **1** into electron-rich dienamines⁶ **5** via

 ⁽a) Carreira, E. M.; Kvaerno, L. Classics in Stereoselective Synthesis, Wiley-VCH: Weinheim, 2008.
 (b) Asymmetric Synthesis: The Essentials, Christmann, M.; Bräse, S., Eds., Wiley-VCH: Weinheim, 2007.

^{(2) (}a) Berkessel, A.; Gröger, H. Asymmetric Organocatalysis, Wiley-VCH: Weinheim, 2005. (b) Erkkilä, A.; Majander, I.; Pihko, P. M. Chem. Rev. 2007, 107, 5416. (c) Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. Chem. Rev. 2007, 107, 5471. (d) de Figueiredo, R. M.; Christmann, M. Eur. J. Org. Chem. 2007, 2575. (e) Dondoni, A.; Massi, A. Angew. Chem. Int. Ed. 2008, 47, 2. (f) Melchiorre, P.; Marigo, M.; Carlone, A.; Bartoli, G. Angew. Chem. Int. Ed. 2008, 3, 922. (h) Bertelsen, S.; Jørgensen, K. A. Chem. Soc. Rev. 2009, DOI: 10.1039/b903816g.

⁽³⁾ Shenvi, R. A.; O'Malley, D. P.; Baran, P. S. Acc. Chem. Res. 2009, 42, 530.

⁽⁴⁾ Escher, I.; Glorius, F. Science of Synthesis 2007, 25, 733.

 ^{(5) (}a) Burstein, C.; Glorius, F. Angew. Chem. Int. Ed. 2004, 43,
 (505. (b) Sohn, S. S.; Rosen, E. L.; Bode, J. W. J. Am. Chem. Soc. 2004,
 126, 14370. (c) Phillips, E. M.; Wadamoto, M.; Chan, A.; Scheidt, K. A.
 Angew. Chem. Int. Ed. 2007, 46, 3107.

^{(6) (}a) Chen, S.-H.; Hong, B.-C.; Su, C.-F.; Sarshar, S. *Tetrahedron Lett.* 2005, *46*, 8899. (b) Bench, B. J.; Liu, C.; Evett, C. R.; Watanabe, C. M. H. *J. Org. Chem.* 2006, *71*, 9458. (c) Bertelsen, S.; Marigo, M.; Brandes, S.; Diner, P.; Jørgensen, K. A. *J. Am. Chem. Soc.* 2006, *128*, 12973. (d) Hong, B.-C.; Wu, M.-F.; Tseng H.-C.; Liao, J.-H. Org. *Lett.* 2006, *8*, 2217. (e) Utsumi, N.; Zhang, H.; Tanaka, F.; Barbas III, C. F. Angew. Chem. Int. Ed. 2007, *46*, 1878 (f) Hong, B.-C.; Wu, M.-F.; Tseng, H.-C.; Huang, G.-F.; Su, C.-F.; Liao, J.-H. *J. Org. Chem.* 2007, *72*, 8459. (g) Hong, B.-C.; Tseng, H.-C.; Chen, S.-H. *Tetrahedron* 2007,

iminium ion 4, a key intermediate in iminium ion catalysis.⁷



Scheme 1. Activation of α,β -unsaturated aldehydes with different nucleophilic catalysts.

The intramolecular Rauhut-Currier (RC) reaction⁸ generates chiral cycloalkenes from acyclic precursors bearing two tethered Michael-acceptors. Surprisingly, a transannular version of this reaction by Moore⁹ preceded thorough investigations of Krische¹⁰ and Roush¹¹ of intramolecular RC reactions using alkyl phosphines as catalyst. In 2007, Miller¹² and shortly after Gladysz¹³ reported enantioselective intramolecular RC reactions of bis(enones). While Miller used cysteine derivatives as catalyst, Gladysz employed phosphines associated with a chiral rhenium-complex. Recently, Scheidt¹⁴ made progress in the intermolecular RC reaction using silyloxy-allenes. There is also evidence for the involvement of RC-type processes in biosynthesis of iridodials.¹⁵

Our previous work^{6h} in the area of dienamine catalysis was concerned with the d⁴-reactivity of α,β -unsaturated aldehydes (1 \rightarrow 5). Herein, we report the potential of 5 as a d²-synthon in a mechanistically distinct RC-type reaction. Since the Jørgensen-Hayashi catalyst (**IV**) is

- 63, 2840. (h) de Figueiredo, R. M.; Fröhlich, R.; Christmann, M. Angew. Chem. Int. Ed. 2008, 47, 1450. (i) Han, B.; He, Z.-Q.; Li, J.-L.; Li, R.; Jiang, K.; Liu, T.-Y.; Chen, Y.-C. Angew. Chem. Int. Ed. 2009, 48, 5474. (j) Hong, B.-C.; Nimje, R. Y.; Liao, J.-H. Org. Biomol. Chem. 2009, DOI: 10.1039/b906205j.
- (7) Ahrendt, K. A.; Borths, C. J.; MacMillan, D. W. C. J. Am. Chem. Soc. 2000, 122, 4243.
- (8) (a) Rauhut, M. M.; Currier, H. US Patent 3,074,999, **1963**. For an excellent review, see (b) Aroyan, C. E.; Dermenci, A.; Miller, S. J. *Tetrahedron* **2009**, *65*, 4069.
- (9) Erguden, J. K.; Moore, H. W. Org. Lett. 1999, 1, 375.
- (10) Wang, L.-C.; Luis, A. L.; Agapiou, K.; Jang, H.-Y.; Krische, M.
 J. Am. Chem. Soc. 2002, 124, 2402.
- (11) (a) Frank, S. A.; Mergott, D. J.; Roush, W. R. J. Am. Chem. Soc. **2002**, 124, 2404. (b) Thalji, R. K.; Roush, W. R. J. Am. Chem. Soc. **2005**, 127, 16778.
 - (12) Aroyan, C. E.; Miller, S. J. J. Am. Chem. Soc. 2007, 129, 256.
 - (13) Seidel, F.; Gladysz, J. A. Synlett 2007, 986.
- (14) Reynolds, T. E.; Binkley, M. S.; Scheidt, K. A. Org. Lett. 2008, 10, 2449.
- (15) Lorenz, M.; Boland, W.; Dettner, K. Angew. Chem. Int. Ed. 1993, 32, 912.

specific toward aldehydes, selective activation over other Michael acceptors such as α,β -unsaturated ketones might be attainable. More importantly, this mode of activation should allow the activation of β -disubstituted alkenes thereby providing an entry into the iridoid-class^{16,17} of natural products, found widely in plants and insects which possess important pharmacological activities.¹⁸ We started our investigations using compound **6a** as the test substrate in order to explore the viability of an enantioselective cyclization to give cyclopentene **7a**. An initial screening of catalysts **I-V** in the absence of any additives revealed catalyst **IV** to be best in terms of both reactivity and enantioselectivity (Table 1, entries 1-5).

Table 1. Catalyst and solvent screening.^a



entry	catalyst	additive	solvent	t [h]	yield [%] ^b	$ee [\%]^c$	
1	Ι	-	CH_2Cl_2	72	22	6	
2	II	-	CH_2Cl_2	72	n.d.	79	
3	III	-	CH_2Cl_2	72	-	-	
4	IV	-	CH_2Cl_2	72	60	88	
5	V	-	CH_2Cl_2	72	32	94	
6	IV	AcOH	CH_2Cl_2	3	57	89	
7^d	IV	AcOH	CH_2Cl_2	24	41	93	
8	IV	AcOH	PhMe	28	29	44	
9	IV	AcOH	MeCN	28	32	93	
10	IV	AcOH	CHCl ₃	3	57	87	
11^{e}	IV	AcOH	CH_2Cl_2	4	58	91	
12^{e}	VI	AcOH	CH_2Cl_2	4	58	76	
13^{e}	VII	AcOH	CH_2Cl_2	4	53	54	
$14^{e,f}$	IV	AcOH	CH_2Cl_2	5	63	91	

^{*a*} Experimental conditions: to a solution of aldehyde **6a** (0.19 mmol) and additive (26 mol%) in 0.5 mL of solvent, 26 mol% of catalyst **I-VII** in 1 mL of solvent was added. The product was purified by flash chromatography. ^{*b*} Yield of isolated product. ^{*c*} Determined HPLC methods. ^{*d*} Reaction was carried out at +5 °C. ^{*e*} Slow addition of aldehyde **6a** in CH₂Cl₂ (1.0 mL) to a solution of the catalyst (20 mol%) and the additive (20 mol%) in CH₂Cl₂ (0.5 mL). ^{*f*} 0.36 mmol-scale.

⁽¹⁶⁾ Unelius performed formal [4+2]-cycloadditions with 8-oxocitral and stoichiometric amounts of chiral secondary amines leading to iridoids: Santangelo, E. M.; Liblikas, I.; Mudalige, A.; Törnroos, K. W.; Norrby, P.-O.; Unelius, C. R. *Eur. J. Org. Chem.* **2008**, 5915. (17) (a) Tundis, R.; Loizzo, M. R.; Menichini, F.; Statti, G. A.;

^{(17) (}a) Tundis, R.; Loizzo, M. R.; Menichini, F.; Statti, G. A.; Menichini, F. *Mini-Reviews in Medicinal Chemistry* **2008**, *8*, 399. (b) Nangia, A.; Prasuna, G.; Rao, B. *Tetrahedron* **1997**, *53*, 14507.

^{(18) (}a) Pasteels, J. M.; Rowell-Rahier, M.; Braekman, J.-C.; Daloze, D. Biochem. Syst. Ecol. 1984, 12, 395. (b) Boros, C. A.; Stermitz, F. R. J. Nat. Prod. 1990, 53, 1055. (c) Boros, C. A.; Stermitz, R. R. J. Nat. Prod. 1991, 54, 1173. (d) Dawson, G. W.; Pickett, J. A.; Smiley, W. M. Bioorg. Med. Chem. 1996, 4, 351.

The use of acetic acid as an additive led to a considerable increase in the reaction rate without impairing the enantioselectivity (entry 6). Lowering the temperature to 5 °C improved the enantioselectivity at the expense of reactivity and yield (entry 7). Switching solvents to toluene or acetonitrile (entries 8, 9) again had a negative influence on the reactivity whereas the use of chloroform gave comparable results (entry 10). Slow addition of the substrate to a solution of the catalyst **IV** was the key to lowering the catalyst loading to 20 mol%, which was accompanied by a slight increase in the asymmetric induction (entry 11).



^{*a*} Experimental conditions: to a solution of **IV**·AcOH (20 mol%) in CH₂Cl₂ (1 mL) a solution of aldehydes **6b-h** (0.36 mmol) in CH₂Cl₂ (2 mL) was slowly added. The products (**7b-h**) were isolated by flash chromatography. ^{*b*} Isolated yield. ^{*c*} Determined by HPLC methods. ^{*d*} After one recrystallization.

However, employing catalysts **VI** and **VII** in this manner offered no improvements (entries 12 and 13). Finally, at 0.36 mmol-scale we were able to increase the yield to 63% and 91% *ee* (entry 14). In order to evaluate the scope of this new cyclization, various aldehydes (**6b-h**) were allowed to react under the optimized conditions. The desired iridoid aldehydes (**7b-h**) were obtained in good yields and enantioselectivities up to 96% *ee* as shown in Table 2.

The electronic environment at the acceptor strongly influences the reaction rate and the enantioselectivity. Electron-withdrawing groups in the aromatic ring increase the reaction rate (entries 2-4) whereas the electron-donating 4-methoxy-substituent reduced the reaction rate significantly (entry 5). Other Michael acceptors such as nitroalkenes can be employed but the reaction proceeds with somewhat lower enantioselectivity (entry 6). The presence of a methyl group in the β position of the aldehyde was crucial to the enantioselectivity and the reaction rate. We speculate that the methyl substituent renders the dienamine more nucleophilic and lowers the relative energy of the required 3Z-configuration (vide infra). In order to determine the stereochemical outcome of the cyclization, single crystals were obtained from 7e and the absolute configuration was determined to be R (Figure 1).



Figure 1. X-ray crystal structure of (*R*)-7e.

To further demonstrate the utility of our methodology, we identified (+)-rotundial (**7h**, entry 7), a powerful mosquito repellent from *Vitex rotundifolia*,¹⁹ as target molecule for the application of an asymmetric RC-type cyclization. The precursor dialdehyde **6h** was obtained in five steps from geranyl acetate. The yield of the cyclization step (36%, 86% *ee*) was hampered by the sensitivity of rotundial to degradation.²⁰ The overall yield (25%) compares favorably with previous syntheses²¹ and the sign of the optical reaction was in agreement with the assumed (*R*)-configuration.

Based on our previous experience, we propose the following catalytic cycle which is in agreement with our observations although further experiments will be necessary to support this working hypothesis (Scheme 2).

⁽¹⁹⁾ Watanabe, K.; Takada, Y.; Matsuo, N.; Nishimura, H. Biosci. Biotech. Biochem. 1996, 59, 1979.

⁽²⁰⁾ Another possible explanation is capture of the catalyst as described by Unelius (Ref. 16).

⁽²¹⁾ For the first synthesis of (R)- and (S)-rotundial, see: Takikawa, H.; Yamazaki, Y.; Mori, K. Eur. J. Org. Chem. **1998**, 229.

Scheme 2. Proposed reaction mechanism.



The first step is the condensation of the substrate 6a with the catalyst IV to form an iminium ion 8^{22} Jørgensen has shown using NMR studies that the iminium ion exists in an equilibrium with the electronrich dienamine 9.23 Previous work has focused on its reactivity as d⁴-synthon. In this instance, we were able to exploit the d²-reactivity with the aid of sterics. It is conceivable that acetic acid assists in the formation of the carbon-carbon bond by activating the acceptor. In our model system, the tentative intermediate is *cis*-10.²⁴ Hydrolysis of the iminium ion would lead to compound *cis*-11.²⁵ However, it should be noted that each step in the catalytic cycle is reversible and proton abstraction $(10\rightarrow 12)$ and reprotonation $(12\rightarrow 10)$ could easily lead to trans-10. Presumably, hydrolysis of 10 is slow, therefore protonation of 12 in the γ -position leads to iminium ion 13, which upon hydrolysis releases 7a to regenerate the catalyst.

In summary, we have developed an enantioselective organocatalytic RC-type cyclization of α , β -unsaturated aldehydes catalyzed by the commercially available Jørgensen-Hayashi catalyst **IV**. The iridoid products, cyclopentene derivatives bearing a tetra-substituted olefin, were obtained in moderate to good yields and with good enantioselectivity. Due to the unprecedented mode of activation in RC-processes (1,2- vs. 1,4-activation), complementary chemoselectivity can be achieved. The extremely simple operational procedure and the synthetic versatility of the products render this new approach highly appealing for the synthesis of optically active iriods, such as (+)-rotundial.

Acknowledgment Financial support was provided by the DFG (SPP1179), the Fonds der Chemischen Industrie (Dozentenstipendium to M.C.), the Alexander von Humboldt Foundation (fellowship to E.M.-L.) and the Aragón I+D Foundation and the Spanish Ministry of Science and Innovation (Programa Nacional de Movilidad de Recursos Humanos del Plan Nacional de I-D+I 2008-2011) and CSIC (PIE 200880I260) (R.P.H.). We thank Dr. Pablo J. Sanz Miguel for X-ray structure analysis.

Supporting Information Available Experimental procedures, NMR-spectra and characterization of all new compounds. This material is available free of charge via the internet at http://pubs.acs.org

^{(22) (}a) Seebach, D.; Grošelj, U.; Badine, D. M.; Schweizer, W. B.; Beck, A. K. *Helv. Chim. Acta.* **2008**, *91*, 1999. (b) Grošelj, U.; Schweizer, W. B.; Ebert, M.-O.; Seebach, D. *Helv. Chim. Acta.* **2009**, *92*, 1.

⁽²³⁾ For theoretical discussions concerning dienamine intermediates, see reference (6d) and Duarte, F. J. S.; Cabrita, E. J.; Frenking, G.; Santos, A. G. *Chem. Eur. J.* **2009**, *15*, 1734.

⁽²⁴⁾ The possibility of the enone to adopt an s-*cis* conformation and participate in a formal [4+2]-process can not be ruled out completely.

⁽²⁵⁾ EZ-Isomerization of the enol moiety of **10** could lead to an intramolecular attack of the iminium ion and to a dead end in the catalytic cycle.