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Silver substrates for surface enhanced Raman scattering: Correlation between nanostructure and Raman scattering enhancement

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The fabrication of substrates for Surface Enhanced Raman Scattering (SERS) applications matching the needs for high sensitive and reproducible sensors remains a major scientific and technological issue. We correlate the morphological parameters of silver (Ag) nanostructured thin films prepared by sputter deposition on flat silicon (Si) substrates with their SERS activity. A maximum enhancement of the SERS signal has been found at the Ag percolation threshold, leading to the detection of thiophenol, a non-resonant Raman probe, at concentrations as low as 10^{-10} M, which corresponds to enhancement factors higher than 7 orders of magnitude. To gain full control over the developed nanostructure, we employed the combination of *in-situ* time-resolved microfocus Grazing Incidence Small Angle X-ray Scattering with sputter deposition. This enables to achieve a deepened understanding of the different growth regimes of Ag. Thereby an improved tailoring of the thin film nanostructure for SERS applications can be realized. © 2014 AIP Publishing LLC.

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The tremendous enhancement of the electromagnetic field (EM) in the close vicinity of nanostructured noble metal surfaces is well accepted to constitute the basis of Surface-Enhanced Raman Scattering (SERS).¹ This enhancement is primarily due to localized surface plasmons on the metallic nanostructures and the maximum EM enhancement occurs in the gaps between the nanostructures, known as “hot-spots.” A chemical enhancement mechanism due to the electronic interaction between the molecule and the metal has also been identified, although this effect is usually smaller than the EM enhancement.²

The intensification of the Raman signal achieved through the resonant interaction between the optical fields and localized surface plasmons can lead to an outstanding sensitivity, allowing for the detection of single molecules.^{3,4} Thus, SERS substrates could be used for non-invasive biological assays,⁵ biosensing,⁶ or environmental analysis⁷ with high sensitivity.

Several methods have been commonly employed to fabricate SERS-active substrates, e.g., the immobilization of colloidal nanoparticles on solid surfaces^{8,9} or lithographic techniques.^{10,11} The use of patterned templates exploiting the self-assembly capabilities of block co-polymer thin films has also been explored.¹² Nevertheless, the fabrication of reliable SERS substrates on macroscopic areas is still an enormous technological challenge.

In order to fabricate homogeneous SERS substrates over macroscopic areas different approaches have recently been investigated such as the laser-direct-writing technique,¹³ the nanosphere lithography method,¹⁴ or the convective self-assembly of nanoparticles.¹⁵ However, all of them involve

several fabrication steps. On the other hand, sputter deposition is a widely employed method in industry that can cover large areas presenting some advantages over wet approaches,¹⁶ and the preparation of SERS substrates using vacuum deposition is a one-step process that can be applied on a variety of different kinds of substrates with excellent reproducibility.^{17,18} However, to fully exploit its fabrication capabilities for SERS applications, it is of utmost importance to achieve a deep understanding of the growth mechanism of the deposited nanostructured film. In this sense, Grazing Incidence Small Angle X-Ray Scattering (GISAXS)¹⁹ constitutes a very valuable technique for the morphological characterization of thin films.^{20,21} Recently, it has also been proven to be a very powerful tool for *in-situ* characterizing the growth of thin nanostructured metallic films, with time resolution in the order of tenths of milliseconds.^{22–24} Due to the shallow incidence angle used, GISAXS provides statistically relevant information over a large sample area.

In this article, we correlate the morphology of nanostructured Ag thin films deposited on flat silicon (Si) substrates and their SERS activity. We have employed radio-frequency (RF) sputter deposition and microfocus GISAXS (μ GISAXS) together with SERS measurements at selected film thicknesses. The sensitive, homogeneous over macroscopic areas and easily produced SERS substrates presented here match the needs for reproducible, low-cost sensors based on the SERS effect. In addition, we have followed *in-situ* and in real-time the morphology development of the deposited thin films, allowing for the extraction of important parameters concerning the growth kinetics and therefore for tuning the nanostructure for specific needs. This information permits to adjust the morphological parameters for further applications such as antimicrobial coatings²⁵ or plasmonics.²⁶

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The deposition of Ag (Kurt J. Lesker; Purity 99.99%) was performed using a portable RF sputter deposition chamber specially designed and manufactured for μ GISAXS.²² Acid-cleaned flat Si pieces ($1.8 \times 1.8 \text{ cm}^2$) were used as substrates.²⁷ The substrates were kept at room temperature during the entire process and a sputter rate of $0.93 \pm 0.04 \text{ nm/min}$ was employed (Ar pressure: 1.8×10^{-2} millibars; Power: 100 W).²⁷

μ GISAXS was performed at the P03/MiNaXS beamline^{28,29} of the PETRA III storage ring at DESY (Hamburg, Germany). An incident photon energy of 13 keV with a beam size of $(31 \times 24) \mu\text{m}^2$ at the sample position was used. The sample to detector distance was set at $3627 \pm 2 \text{ mm}$ and a PILATUS 1M (Dectris Ltd., Switzerland) with a pixel size of $(172 \times 172) \mu\text{m}^2$ was used as detector. During the sputter deposition, the scattering patterns were continuous and uninterruptedly recorded at a frame rate of 10 images per second. In order to achieve a good separation between the Si and Ag Yoneda peaks, an incident angle of $\alpha_i = 0.5^\circ$ was used. The μ GISAXS data were analysed using the DPDAK software package. After performing horizontal and vertical line cuts of the two-dimensional (2D) scattering patterns, the obtained curves were fitted using Lorentzian functions. The simulation of the μ GISAXS patterns was performed using the IsGISAXS software package.^{27,30}

For the SERS characterization, $3 \mu\text{l}$ droplets of thiophenol solution were deposited on the Ag thin films. The droplets were dried inside a fume hood. A Renishaw *InVia* Raman microscope (Renishaw plc., United Kingdom) employing an incident laser wavelength of $\lambda = 532 \text{ nm}$ and a $50\times$ objective with a numerical aperture, $\text{NA} = 0.75$, leading to a power density of $3.2 \text{ mW}/\mu\text{m}^2$ at the sample position, was used for the acquisition of the spectra. Each spectrum was acquired with an acquisition time of 10 s and 1 accumulation. The enhancement reproducibility of the substrates was tested by acquiring spectra at multiple positions.

To give an overview of the growth process, Fig. 1 presents selected 2D μ GISAXS patterns obtained during the sputter deposition of Ag. At the early stages of the sputter deposition, two side peaks (along q_y) appeared indicating the existence of laterally ordered nanostructures on the sample surface. As the deposition proceeded, the side peaks shifted towards smaller q_y values, being arrested at a fixed position at later sputtering times. During the entire process both side peaks remained symmetric with respect to the scattering

plane. At an effective film thickness around 3 nm peaks along q_z became clear. These peaks shifted towards lower q_z values as the sputter time increased and more modulations appeared and shifted towards lower q_z indicating the vertical growth of the nanostructured thin film.

In order to extract quantitative information from the scattering patterns, out-of-plane cuts (along q_y)²⁷ at the Yoneda peak³¹ position of Si ($q_{z,c}(\text{Si}) = 0.733 \text{ nm}^{-1}$) and off-detector cuts (along q_z)²⁷ at $q_y = 0.112 \text{ nm}^{-1}$ were performed. Figs. 2(a) and 2(b) show 2D maps of the out-of-plane cuts and of the off-detector cuts versus the effective thickness of the deposited film, respectively.

The side peak position as well as the full-width-at-half-maximum (FWHM) extracted from Lorentzian fittings are depicted in Fig. 2(c). A continuous shift of the side peak position towards lower q_y values was found indicating an increase in the average distance between Ag clusters due to coalescence effects.³² Furthermore, an overall decrease of the FWHM was observed, revealing an increase in the order of the nanoclusters since the width of the peak is related to the mean cluster distance distribution.

At the early stages of sputter deposition, a change in the slope of the FWHM was identified (inset in Fig. 2(c)), indicating a transition in the growth kinetics from a predominant nucleation process to coalescence of adjacent clusters.²³ The nucleation threshold was found to occur at an effective film thickness of $\delta = 0.28 \pm 0.05 \text{ nm}$. As the deposition continued, the FWHM passed a local maximum at $\delta = 5.7 \pm 0.1 \text{ nm}$, revealing a broadening of the cluster distance distribution that can be attributed to the growth of a second layer of Ag. This is in accordance with the observed growth behaviour of ultrathin Ag films deposited from the plasma phase on oxide layers.^{33,34} Below the percolation threshold, a predominantly island growth mechanism occurs whilst a layer-by-layer growth predominates after percolation.^{32,33} Therefore, we identified this point with the percolation threshold of the Ag cluster layer.

From the evolution of the Si and Ag Yoneda peaks (Fig. 2(d)) at $q_{z,c}(\text{Si}) = 0.733 \text{ nm}^{-1}$ and $q_{z,c}(\text{Ag}) = 0.883 \text{ nm}^{-1}$, respectively, the threshold between two different coalescence mechanisms was identified. At an effective thickness of $\delta = 1.1 \pm 0.1 \text{ nm}$, the intensity at the Ag Yoneda peak became dominant, revealing the transition from a coalescence process mainly due to diffusion phenomena to a predominantly adsorption-driven coalescence.²³ Above the diffusion

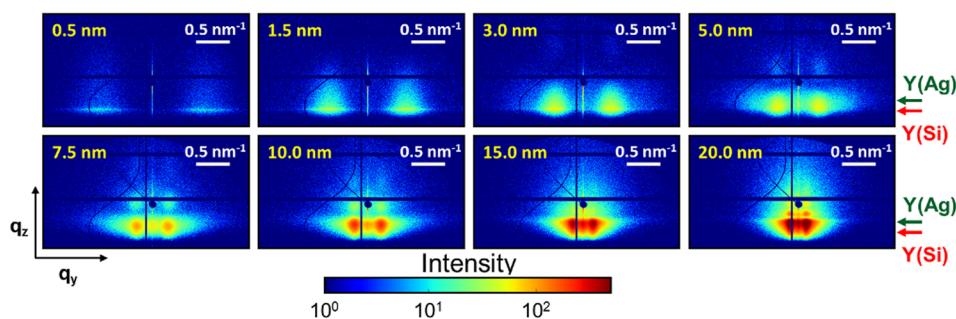


FIG. 1. Selected 2D μ GISAXS patterns during RF-sputter deposition of Ag. The corresponding effective film thickness is indicated in each image. The dark blue circle corresponds to the specular beam stop used to prevent the detector from being damaged due to the high X-ray intensity. The dark blue horizontal and vertical stripes correspond to the intermodule detector gaps. The positions of the Si Yoneda peak, $Y(\text{Si})$, and Ag Yoneda peak, $Y(\text{Ag})$, as well as the coordinate system (q_y, q_z) are indicated.

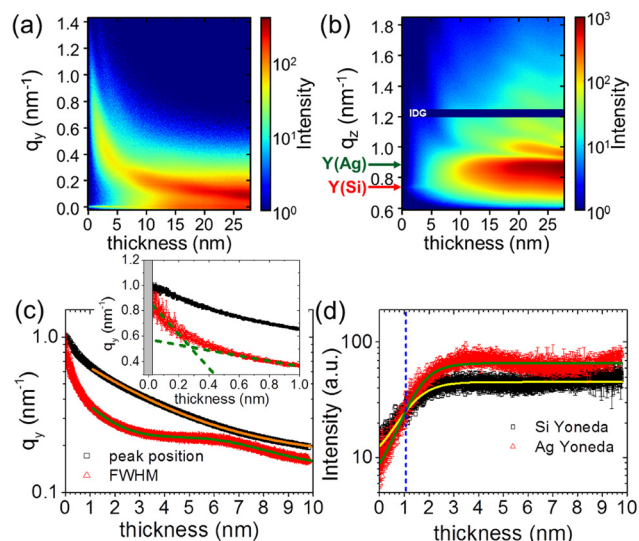


FIG. 2. (a) Out-of-plane (along q_y) line cuts versus sputtered thickness at the Si Yoneda peak ($q_{z,c}(\text{Si}) = 0.733 \text{ nm}^{-1}$). (b) Off-detector (along q_z) line cuts versus sputtered thickness at $q_y = 0.112 \text{ nm}^{-1}$. IDG: Intermodule Detector Gap. Y(Si): Si Yoneda peak. Y(Ag): Ag Yoneda peak. (c) Evolution of the peak position and FWHM of the side peak. The continuous orange line is the exponential fitting of the peak position and the continuous green line represents an exponential plus Gaussian fitting of the FWHM. The inset shows a zoom of the early stage of Ag deposition. (d) Si and Ag Yoneda peak intensities versus the effective sputtered thickness. The continuous lines are Boltzmann sigmoidal fittings of the data. The blue dashed line marks the crossing point of the sigmoidal fittings.

threshold, the shift in the side peak position is well described by an exponential decay $q_y(\delta) = q_0 + A \exp(-\delta/\delta_0)$, where $q_0 = 0.16 \text{ nm}^{-1}$, $A = 0.64 \text{ nm}^{-1}$, and $\delta_0 = 3.36 \text{ nm}$ (continuous line in Fig. 2(c)). This permits to extract the mean cluster distance throughout the sputter deposition process using the relation $D \approx 2\pi/q_y$.

The homogeneity of the deposited film was investigated by performing a lateral μGISAXS scan after the complete deposition. No variation in the scattering patterns was found indicating that the nanostructured thin film was homogeneous over the scanned area.²⁷ Furthermore, no influence of the X-ray beam on the Ag growth was observed.²⁷

To verify the SERS activity of the substrates, several droplets of a 10^{-4} M thiophenol solution were deposited on Ag layers with different effective thickness and nanostructure deduced from the *in-situ* μGISAXS study. We also deposited a droplet on a bare Si substrate for comparison. Thiophenol was employed as Raman probe instead of the more commonly used Rhodamine 6G and crystal violet since the latter ones exhibit Raman resonance effects under visible illumination, obscuring the SERS enhancement solely due to surface effects. Whilst no thiophenol signal was observed on

Si, the characteristic SERS bands of thiophenol adsorbed on Ag³⁵ became clear for all the substrates (Fig. 3(a)).

The Raman intensity was quantified using the integrated peak area of the 999 cm^{-1} , 1023 cm^{-1} , and 1074 cm^{-1} bands, corresponding to the benzene ring breathing mode, the in-plane CH bending mode and a combination of the benzene ring breathing and CS stretching modes, respectively.³⁶ A maximum intensification of the SERS signal was found for an effective film thickness of $5.6 \pm 0.1 \text{ nm}$ (Fig. 3(b)), which corresponds to the percolation threshold identified by the *in-situ* μGISAXS study. Below and above the percolation threshold, the SERS enhancement was found to decrease symmetrically. As the nanostructured film grows, the gap between clusters is reduced, and a higher intensification of the local electromagnetic field takes place. However, when the percolation threshold is passed, the film presents a high connectivity, and thus, the density of available “hot-spots” is significantly reduced. At the percolation threshold, the optical properties of the film changes drastically, and the maximum surface plasmon resonance takes place.^{33,37}

Fig. 3(c) shows the integrated intensity of the aforementioned SERS bands at different thiophenol concentrations for the substrate with an effective thickness of $5.6 \pm 0.1 \text{ nm}$. As the thiophenol solution was diluted, a decrease of the SERS intensity was observed up to a concentration of 10^{-7} M followed by a saturation of the Raman enhancement. This behaviour is found for molecules that adsorb to the surface of SERS substrates.³⁸ The SERS spectrum of thiophenol was still visible for a concentration as low as 10^{-10} M , and from the saturation effect, the SERS enhancement factor can be estimated to be higher than 7 orders of magnitude. This estimation is in good agreement with the calculated enhancement factors for several different SERS bands of thiophenol.²⁷

In order to extract relevant morphological information concerning the observed SERS activity of the substrates, we simulated the μGISAXS patterns assuming a hexagonal paracrystalline arrangement of hemispherical Ag nanoclusters.^{23,34} We also assumed that all the material deposited is distributed into clusters, i.e., the total volume of the clusters equals the volume of a homogenous layer with the corresponding effective thickness. The position and shape of the side peak (Fig. 4, middle row) show an excellent agreement with the experimental data (Fig. 4, upper row), confirming the hemispherical shape of the clusters for all thicknesses. Therefore, the mean radii and mean gap between Ag clusters have been extracted. The results are presented in Table I and depicted as sketches in the lower row of Fig. 4.

For the substrate presenting the maximum SERS enhancement ($\delta = 5.6 \pm 0.1 \text{ nm}$) a mean gap of around 1 nm was calculated. In the model we have used, effective film

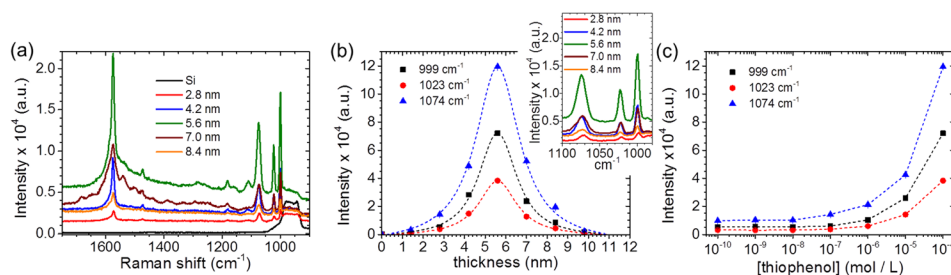


FIG. 3. (a) Surface-Enhanced Raman spectra of thiophenol on sputtered Ag samples at different film thicknesses. (b) Intensity of the characteristic SERS bands shown in the inset for different Ag thickness. The dashed lines are Voigt fittings to the data. (c) Intensity of the characteristic SERS bands at different thiophenol concentrations on the 5.6 nm Ag thickness film.

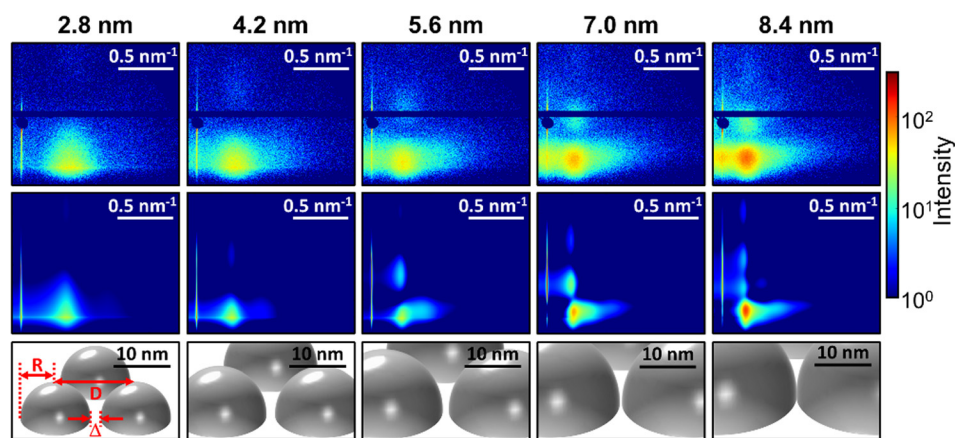


FIG. 4. Upper row: 2D- μ GISAXS patterns for the thicknesses at which the SERS activity of the Ag thin films was investigated. Middle row: Simulated 2D- μ GISAXS patterns. Lower row: Real space sketches of the morphology. D = mean cluster correlation distance; R = mean cluster radius; and Δ = mean gap between clusters.

TABLE I. Morphological parameters of the nanostructured Ag thin films extracted from the μ GISAXS data simulation.

δ^a (nm)	D^b (nm)	R^c (nm)	Δ^d (nm)
2.8 ± 0.1	14.2 ± 0.1	6.3 ± 0.1	1.6 ± 0.1
4.2 ± 0.1	18.1 ± 0.1	8.4 ± 0.1	1.3 ± 0.1
5.6 ± 0.1	21.9 ± 0.1	10.5 ± 0.1	0.9 ± 0.1
7.0 ± 0.1	25.8 ± 0.1	12.7 ± 0.1	0.4 ± 0.1
8.4 ± 0.1	28.8 ± 0.1	14.4 ± 0.1	0.0 ± 0.1

^aEffective film thickness.

^bMean cluster correlation distance.

^cMean cluster radius.

^dMean gap between clusters.

thicknesses above 5.6 nm have smaller mean gaps, what could lead to higher SERS enhancements.³⁹ Nevertheless, the high connectivity of the deposited clusters and the increased electron tunnelling probability between nanostructures⁴⁰ are responsible for the decrease of the observed SERS enhancement.

In summary, SERS substrates over macroscopic areas have been prepared by RF-sputter deposition of Ag finding enhancement factors higher than 7 orders of magnitude for an effective film thickness corresponding to the percolation threshold, extracted from the *in-situ* μ GISAXS measurements. At this effective thickness a mean gap between hemispherical Ag clusters of around 1 nm has been found. Moreover, thiophenol was still detected at concentrations as low as 10^{-10} M. The substrates presented homogeneous nanostructured morphology over macroscopic areas and exhibited high SERS sensitivity which, together with the easiness of fabrication, are the specific needs for SERS-based sensors. In addition, from the unique combination of *in-situ* time resolved μ GISAXS and sputter deposition, we have identified the main growth kinetic regimes as well as their thresholds, which is crucial for nanostructure tailoring. This information is of high relevance not only for the purpose of this article but also for adjusting the nanostructured Ag morphology to further different specific requirements.

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