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Short communication

Presence of mono-, di- and galactooligosaccharides in commercial lactose-free UHT dairy products

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11 Abstract

12 In this work a study of mono-, di- and galactooligosaccharide (GOS) content as well as 13 the determination of thermal indicators of different commercial lactose-free UHT (LF-UHT) milks and dairy drinks has been carried out. Moreover, GOS formation during 14 hydrolysis of lactose in a commercial UHT milk using Lactozym[®] pure 6500 was also 15 16 studied. Presence of GOS was detected in all analyzed samples ranging from 950–4350 17 mg/L in LF-UHT milks and from 600-2260 mg/L in LF-UHT dairy drinks. Their GOS 18 contents correlate roughly with the remaining lactose content of samples. During 19 enzymatic hydrolysis of lactose in commercial UHT milk, the GOS formed reached a 20 maximum about 10000 mg/L when 75-90% of lactose was hydrolyzed and then 21 gradually decreased to values considerably lower than 5000 mg/L when over 99% of the 22 lactose was hydrolyzed. Therefore, GOS formation during low lactose milk

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- 23 manufacture might be improved by controlling enzymatic lactose hydrolysis so that it
- 24 would be possible to elaborate low-lactose milk products able to satisfy the
- 25 physiological requirements of the majority of intolerant groups and with a GOS content
- 26 high enough to confer beneficial effect as prebiotic.
- 27 *Keywords:* Lactose-free UHT; Milks; Dairy drinks; Lactozym[®]; GOS; Prebiotics;
- 28 Furosine; Food composition; Food analysis

29 Abbreviations

- 30 GOS Galactooligosaccharides
- 31 LF-UHT Lactose-free ultra-high temperature
- 32 TMSO Trimethyl silylated oximes

33 **1** Introduction

34 In recent years lactose-free UHT (LF-UHT) milk has become popular in many countries 35 since there are a significant number of people who are affected by lactose intolerance, 36 because they lack the required lactase on the intestinal brush borders to hydrolyze 37 lactose to glucose and galactose and to be absorbed through the digestive tract. As a 38 result, lactose reaches the colon where it is fermented, producing symptoms such as gas, 39 bloating and diarrhea. The deficiency of lactase in varying degrees (hypolactasia) 40 affects nearly 70% of the world population being the most common enzyme deficiency 41 in humans (Lomer et al., 2008).

As the interest of consumers for lactose-free (LF) milk grows, the number of dairies
producing this type of milk also increases. However, despite the increasing consumption

44 of LF milk, few data are available on the characterization and quality assessment of

these commercial milks (Adhikari et al., 2010; Messia et al., 2007; Rada-Mendoza et
al., 2005).

47 LF-UHT milks present large amounts of reducing monosaccharides which makes them 48 more susceptible to deterioration than ordinary UHT milk, since monosaccharides are 49 more reactive than lactose when participating in the Maillard reaction that takes place 50 not only in the heating process but also during storage of milk (Braekman et al., 2001; 51 Burvall et al., 1978; Tossavainen & Kallioinen, 2008). Moreover, there are several 52 procedures involved in the manufacture of LF-UHT milks such as removal (or not) of 53 some lactose prior to enzymatic hydrolysis; hydrolysis performed before or after 54 heating; use of enzymes from different origins, in soluble or immobilized form, etc. 55 (Harju, 2003; Jelen & Tossavainen, 2003; Jokar & Karbassi, 2011). These different 56 processes used in the production of LF-UHT milks can lead to a wide variability of 57 hydrolysis and heating markers such as lactulose and furosine, found in different 58 commercial samples of LF-UHT milks (Messia et al., 2007). It has been described that 59 heating prior to hydrolysis of lactose is suggested to avoid a considerable loss of 60 available lysine, an essential amino acid, as a result of the Maillard reaction (Rada-61 Mendoza et al., 2005).

During manufacture of LF-UHT milks, formation of galactooligosaccharides (GOS) can
also take place by transglycosylation reactions. GOS are produced from lactose by
glycosyl transfer of one or more D-galactosyl units onto D-galactose moiety of lactose
catalyzed by β-galactosidases (Mahoney, 1998). The amounts and composition of the
GOS fraction formed can be influenced by a number of factors, including enzyme
source, pH and temperature (Boon et al., 1998).

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GOS are defined as non-digestible food ingredients that beneficially affect the host by 68 69 selectively stimulating the growth and/or activity of one or a limited number of bacteria 70 in the colon that can improve host health (Gibson & Roberfroid, 1995). Although there 71 are some studies that show GOS formation during enzymatic hydrolysis of milk (Indyk et al., 1996; Mahoney, 1998; Mozaffar et al., 1985; Rustom et al., 1998), data about 72 73 GOS composition and variability of content in commercial LF-UHT milks are not 74 available. It would be interesting to know the GOS composition of LF products in order 75 to develop products that satisfy the physiological requirements of intolerant groups, that 76 would contain enough prebiotics to be beneficial. Therefore, the aim of this work was to 77 study carbohydrate composition (mono- and disaccharides and GOS) and the thermal 78 damage of commercial LF-UHT dairy products, as well as to research GOS formation 79 during controlled hydrolysis of lactose in milk.

80 2 Materials and methods

81 **2.1 Standards**

82 Lactose was acquired from Scharlau (Barcelona, Spain), D-galactose and D-glucose

from Fluka (Steinheim, Germany). Fructose, tagatose, β -1,6-galactobiose and phenyl- β -

84 glucoside were purchased from Sigma (St. Louis, MO, USA). Allolactose, 6'-

85 galactosyl-lactose and 4'-galactosyl-lactose were standards previously synthesized in

86 our laboratory (Cardelle-Cobas et al., 2008; Cardelle-Cobas, 2009; Martínez-

87 Villaluenga et al., 2008).

88 **2.2 Samples**

89 Samples of UHT milk (n=1), LF-UHT milks (n=7) and LF-UHT dairy drinks (n=5),

90 from different manufacturers were purchased from different Spanish supermarkets,

91 stored at refrigeration temperature and analyzed before the sell-by date. Labeling of

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- 92 samples showed lactose content lower than 100 mg/L. Protein content, determined using
- the Kjeldahl method (AOAC, 1990) was in the ranging from 26–31 g/L product.

94 **2.3** Enzymatic hydrolysis of lactose from milk

- 95 Hydrolysis of lactose from a commercial UHT milk was performed by β -galactosidase
- 96 (EC 3.2.1.23) activity from *Kluyveromyces lactis* of the commercial enzymatic
- 97 preparation, Lactozym® pure 6500 (U/mL) (Novozymes, Dittingen, Switzerland). For
- 98 incubations, 0.3 U/mL of this preparation was added to an UHT milk sample at 30 °C in
- 99 aseptic conditions. Reactions (in duplicate) were carried out in individual Eppendorf
- 100 tubes and incubated in an orbital shaker at 600 rpm at different times for 30 h and
- 101 immediately immersed in boiled water for 5 min to inactivate the enzyme. The samples
- 102 were stored at -18 °C for subsequent analysis.

103 **2.4 Determination of carbohydrates**

Before analysis, samples were deproteinized using anhydrous methanol (1:10, v:v) for
24 h. Supernatant was collected, centrifuged at 10,000 rpm for 5 min and stored at 4 °C

106 for further analysis. All preparations were carried out in duplicate.

1072.4.1 High-performance anion exchange chromatography with pulsed108amperometric detection (HPAEC-PAD) analysis

109 Mono-, disaccharides and GOS were analyzed by HPAEC-PAD in an ICS2500 Dionex

- 110 system (Dionex Corporation, Sunnyvale, CA) consisting of a GP50 gradient pump, and
- 111 ED50 electrochemical detector with a gold working electrode and Ag/AgCl reference
- electrode. Data acquisition and processing was performed with Chromeleon version 6.7
- 113 software (Dionex Corporation). For preparation of eluents, MilliQ water (Milli-Q
- 114 Synthesis A10 system; Millipore, Billerica, Mass., USA), NaOH (50%, w/v) and

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- 115 NaOAc (Fluka, Germany) were used. All eluents were degassed by flushing with116 helium for 25 min.
- 117 Separations were performed following the method described in Splechtna et al. (2006).
- 118 Elution was at room temperature on a CarboPac PA-1 column (250×4mm) connected to
- 119 a CarboPac PA-1 (50×4mm) guard column. Eluent A (100 mmol/L NaOH) and eluent B
- 120 (100 mmol/L NaOH and 50 mmol/L NaOAc) were mixed to form the following
- 121 gradient: 100% A from 0–20 min and 0–100% B from 20–70 min. After each run, the
- 122 column was washed for 10 min with 100% of eluent C (100 mmol/L NaOH and 1 mol/L
- 123 NaOAc) and re-equilibrated for 15 min with the starting conditions of the employed
- 124 gradient.
- 125 Two milliliters of methanolic extracts were evaporated, reconstituted in 1mL of
- 126 bidistilled water and filtered through a nylon Millipore FH membrane (0.22 μm)
- 127 (Bedford, MA) before injection (20 µL). Separations were performed at a flow rate of 1
- 128 mL/min. Quantification of GOS was performed by external calibration using a standard
- 129 solution of raffinose being the regression coefficient of calibration curve higher than
- 130 0.99. All analyses were carried out in duplicate and data were expressed as mean \pm
- 131 standard deviation (SD).
- 132 2.4.2 Gas chromatographic (GC) analysis
- 133 Trimethyl silylated oximes (TMSO) of monosaccharides (glucose, galactose, fructose
- and tagatose) and disaccharides (allolactose, lactose, β -1,6-galactobiose and sucrose)
- 135 were determined by GC following the method of Cardelle-Cobas et al., (2009).
- 136 Chromatographic analysis was performed on Agilent Technologies gas chromatograph
- 137 (Mod 7890A) equipped with a flame ionization detector (FID). Separation was carried
- 138 out in a fused silica capillary column HP-5MS (5% phenyl methylsilicone, 25 m x 0.32

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139 mmol/L x 0.25 µm thickness; J & W Scientific, Folsom CA, USA). Nitrogen was used 140 as carrier gas at a flow rate of 1 mL/min. Injector and detector temperatures were 280 141 and 315 °C, respectively. The oven temperature was programmed from 180–315 °C at a 142 heating rate of 3 °C/min and held 20 minutes. Injections were made in the split mode 143 (1:20). Data acquisition and integration was done using Agilent ChemStations Reb. 4B. 144 03.01 software (Wilmington, DE, USA). 145 The oximes were formed following the method of Brobst & Lott (1996). First, 1 mL of 146 methanolic extract (4 mg of sugars) was added to 0.4 ml of internal standard (I.S.)

147 solution (0.5 mg/mL phenyl- β -glucoside). Afterwards, the mixture was dried at 38–40

148 °C in a rotary evaporator (Büchi Labortechnik AG, Flawil, Switzerland). Sugar oximes

149 were formed by adding 200 μ L hydroxylamine chloride (2.5%) in pyridine and heating

150 the mixture at 70 °C for 30 min and then silvlated with hexamethyldisilazane (200 μ L)

and trifluoroacetic acid (20 μ L) and kept at 50 °C for 30 min. Reaction mixtures were

152 centrifuged at 7000 x g for 2 min at room temperature. Supernatants were injected in the

153 GC or stored at 4 °C prior to analysis. All analyses were carried out in duplicate and

154 data were expressed as mean \pm standard deviation (SD).

Response factors were calculated after the triplicate analysis of 5 standard solutions
(galactose, glucose, fructose, tagatose and lactose) over the expected concentration
range in samples.

158 **2.5 F**u

2.5 Furosine determination

159 Determination of furosine in the LF-UHT milks and LF-UHT dairy drinks was

160 performed by ion-pair RP-HPLC following the method of Resmini et al. (1990). Before

- 161 analysis, samples (2 mL) were hydrolyzed with 6 mL of 10.6 N HCl under inert
- 162 conditions at 110 °C for 24 h in a Pyrex screw-cap vial with PTFE-faced septa. The

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163	hydrolyzate was filtered through Whatman N° 40 filter paper and 0.5 ml of filtrate was
164	applied to a previously activated (methanol and water) Sep-Pak C_{18} cartridge
165	(Millipore). Furosine was eluted with 3 mL of 3N HCl and 50 μ l were injected into the
166	chromatograph.
167	RP-HPLC analysis of furosine was carried out in a C_8 column (250 mm x 4.6 mm, 5
168	μ m) (Alltech furosine-dedicated Nicolasville, KY) maintained at 35 °C using a linear
169	binary gradient at a flow rate of 1.2 mL/min. Mobile phase was constituted by solvent
170	A, 0.4% acetic acid, and solvent B, 0.3% KCl in phase A. The elution program was as
171	follows: 100% A from 0–12 min, 50% A from 20–22.5 min, and 100% A from 24.5–30
172	min. Detection, was performed using a variable wavelength UV detector at 280 nm
173	(LDC Analytical, SM 4000 Salem, NH). Acquisition and processing of data were
174	achieved with a HPChem Station (Hewlett-Packard) software. Quantification was
175	performed by the external standard method, using a commercial standard of pure
176	furosine (Neosystem Laboratories, Strasbourg, France). All analyses were done in
177	duplicate, and the data are the mean values expressed as milligrams of furosine per kg
178	of protein.

179 **3**

3 Results and discussion

The HPAEC-PAD profile of carbohydrates from a commercial LF-UHT milk is shown in Fig. 1. As expected, glucose and galactose (peak 1) were the main components due to the action of β -galactosidase. It was possible to identify the peak 2 as the disaccharide β -1,6-galactobiose and lactose which co-eluted with allolactose (peak 3). The most abundant GOS present was 6'-galactosyl-lactose (peak 4). Small amounts of 4'galactosyl-lactose (peak 5) were also detected in all studied samples. Moreover, a series of unidentified galactooligosaccharides (GOS) (marked with an asterisk) were detected

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187	as a result of transglycosylation catalyzed by added β -galactosidase. Commercial LF-
188	UHT dairy drinks showed similar HPAEC-PAD profiles of carbohydrates.
189	Due to coelution of galactose and glucose as well as lactose and allolactose their
190	quantification was performed by GC and only 6'-galactosyl-lactose and 4'-galactosyl-
191	lactose as well as the unidentified GOS were quantified by HPAEC-PAD.
192	Fig. 2 shows a GC-FID chromatogram of TMSO derivatives of carbohydrates of a
193	commercial LF-UHT milk. After derivatization, the reducing carbohydrates give rise to
194	two different derivatives corresponding to the $syn(E)$ and the anti (Z) isomers (Molnár-
195	Perl & Horváth, 1997). The panel I shows the complete profile where it can be
196	observed: tagatose (1) and fructose (2) (see panel Ia), glucose (3), galactose (4), internal
197	standard (I.S.) and disaccharide region where it was detected, lactose (5), allolactose
198	(6), and β -1,6-galactobiose (7) (panel Ib). Commercial LF-UHT dairy drinks showed

199 similar GC-FID profiles of carbohydrates.

200 **3.1 Lactose-free UHT milks (LF-UHT milks)**

201 Table 1 shows the content of carbohydrates (mono-, disaccharides and GOS) found in

the studied samples. The glucose and galactose level varied between 19881 and 25050

203 mg/L, and 17297 and 23464 mg/L, respectively. The large amount of both

204 monosaccharides in all samples indicates that lactose was not removed from milks prior

205 to enzymatic hydrolysis. The lower content of galactose with respect to glucose in all

206 samples was due to the formation of GOS during the transgalactosylating action of β -

207 galactosidase, as it was previously reported for enzymatic hydrolysis of milk (Indyk et

al., 1996, Messia et al., 2007; Rada-Mendoza et al., 2005). The lactose content in the

- studied samples varied from 33 mg/L–291 mg/L and the average value was about 96
- 210 mg/L. These values corresponded to a degree of hydrolysis higher than 99%. Only one

sample (number 2) showed a lactose content higher than the amount stated on the label
(lactose content < 100 mg/L).

The total content in GOS varied among samples in the range from 956–4350 mg/L as can be observed in Table 1. The GOS content increased with the remaining lactose content so that sample 2 showed the highest levels of GOS and lactose whereas sample 6 showed the lowest levels of both.

217 Regarding heating markers, lactulose was not detected in any of the samples examined 218 in our study (LOD = 10 mg/L). Given that lactulose is formed from lactose by 219 isomerization during the heat treatment of milk (López-Fandiño & Olano, 1999; 220 Martinez-Castro & Olano, 1978), the absence of lactulose is understandable considering 221 the low content of residual lactose in samples. Moreover, the absence of lactulose is 222 independent whether the UHT-treatment is performed before or after the enzymatic 223 hydrolysis since lactulose is formed from isomerization of lactose during heating of 224 milk. Thus, if UHT treatment is given after lactose hydrolysis, lactulose cannot be 225 formed in appreciable amounts. On the other hand, if hydrolysis is performed after UHT 226 treatment, the lactulose formed from lactose is hydrolyzed by the action of the β -227 galactosidase. These results are in agreement with those obtained by Rada-Mendoza et 228 al. (2005) for lactose hydrolyzed UHT milks. On the contrary, Messia et al. (2007) 229 reported lactulose contents of up to 401 mg/L in commercial lactose hydrolyzed milks 230 but the lactose contents detected in that samples were approximately up to one hundred 231 times higher than those found in the present study.

Isomerization of glucose to fructose and galactose to tagatose during UHT treatment of
LF-UHT milks was also detected since appreciable amounts of fructose (from 76 mg/L–
185 mg/L) and tagatose (from 36 mg/L–98 mg/L) were found in all samples (Table 1).

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235 The presence of fructose is independent whether the UHT-treatment is performed before 236 or after the enzymatic hydrolysis since fructose may be originated either during 237 enzymatic hydrolysis of lactulose present in UHT-treated milk or from glucose 238 isomerization during UHT treatment of hydrolyzed milk. The presence of tagatose in 239 LF-UHT milks is other indication that lactose has been hydrolyzed before UHT 240 treatment since galactose is only present in very small quantities in milk and tagatose 241 has not been detected in UHT milk (Troyano et al., 1992). 242 Fig. 3 shows the level of furosine found in LF-UHT milks samples. The values ranging 243 from 2442–4109 mg/kg protein were considerably higher than those reported for UHT 244 milks (Corzo et al., 1994), as it was previously observed in lactose-hydrolyzed milks 245 (Marconi et al., 2002; Messia et al., 2007; Rada-Mendoza et al., 2005; Tossavainen & 246 Kallioinen, 2008). As was inferred from the presence of tagatose, high furosine contents 247 are also an indication that the UHT treatment was given after the lactose hydrolysis, 248 since lactose-hydrolyzed milk, with high content of reducing monosaccharides, can be 249 more susceptible to the Maillard reaction than natural (non-hydrolyzed) milk 250 (Evangelisti et al., 1999). 251 Therefore, in agreement with results obtained by Rada-Mendoza et al. (2005), it can be 252 proposed that in order to avoid an excessive formation of furosine and, consequently,

253 loss of available lysine and changes in the organoleptic characteristics, it may be

advisable to hydrolyze lactose under aseptic conditions after thermal treatment

conditions.

3.2 Lactose-free UHT dairy drinks (LF-UHT dairy drinks)

Table 2 shows the carbohydrate composition of commercial LF-UHT dairy drinks. As itcan be observed, with the exception of sample 2, the rest of samples showed lower

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259 monosaccharide, disaccharide and GOS content than LF-UHT milks Also, small260 amounts of sucrose were detected in all samples.

261 Furosine content (Fig. 3B) ranged from 1547–4477 mg/kg protein and the interval of 262 variation was higher than in LF-UHT milks. Samples 1 and 5 which showed a level of 263 furosine considerably higher than that of the rest of samples, also displayed the highest 264 tagatose content. Elevated values of furosine in LF-UHT dairy drinks may be due to 265 different causes, including excessive heating, prolonged storage or addition of milk 266 powder; however tagatose is only formed from galactose during heat treatment so that 267 the presence of elevated amounts of both compounds may be due to excessive heating 268 during the manufacture of LF-UHT dairy drinks.

3.3 Formation of GOS during enzymatic hydrolysis of lactose from
 milk

271 The consumption of GOS has unquestionably been shown to have significant health 272 benefits in humans (Macfarlane et al., 2008). GOS have been used to simulate human 273 milk in infant formulas, and a low level (2400 mg/L) can improve stool frequency, 274 decrease fecal pH and stimulate intestinal bifidobacteria and lactobacilli as in children 275 fed with human milk (Ben et al., 2008). Various health benefits of these carbohydrates 276 have been reported by different research groups, so that currently the main focus among 277 prebiotics is on their production and use as a component of functional foods (Sangwan 278 et al., 2011).

- 279 Previous studies on the formation of oligosaccharides during lactose hydrolysis in milk
- showed that maximum content of GOS was achieved within the range 40–95%
- 281 hydrolysis of lactose and then gradually decreased with the lactose content (Mahoney,
- 282 1998). Fig. 4 depicts the formation of oligosaccharides originated from lactose

283 hydrolysis of a commercial UHT milk. Formation of total GOS reached a maximum 284 about 10000 mg/L (about 20% of total carbohydrates present) when percentage of 285 hydrolyzed lactose ranged from 75–90% of lactose was hydrolyzed and then they 286 gradually decreased to values below 5000 mg/L when over 99% of the lactose was 287 hydrolyzed. According to these data, for a residual lactose content lower than 1000 288 mg/L (about 2% of total carbohydrates), milks with a GOS content around 7800 mg/L 289 can be obtained. This GOS content would be enough to exert a beneficial effect on the 290 consumer's health ...

291 **4** Conclusion

292 Although extreme intolerance may require an exclusively lactose-free milk, most people 293 with hypolactasia, if given appropriate advice, can tolerate some lactose-containing 294 foods without symptoms (Lomer et al., 2008). It is generally accepted that 50-80% 295 lactose reduced milk will satisfy the physiological requirements of the majority of 296 intolerant groups (Indyk et al., 1996) so that it would be possible to elaborate a low-297 lactose milk with a noticeable GOS content. Thus, the functional properties of low-298 lactose milk can be improved by controlling the final stages of the enzymatic hydrolysis 299 of lactose.

300

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411 Figure captions

- 412 Fig. 1. HPAEC-PAD chromatogram of carbohydrates present in a commercial lactose-free
- 413 UHT (LF-UHT) milk. 1: galactose + glucose; 2: β-1,6-galactobiose; 3: allolactose +
- 414 lactose; 4: 6'-galactosyl-lactose; 5: 4'-galactosyl-lactose; * unknown
- 415 galactooligosaccharides



- Fig. 2. GC-FID profile of TMSO derivatives of carbohydrates present in a commercial
 lactose-free UHT (LF-UHT) milk. I) Complete chromatogram: la (1) tagatose and (2)
- 428 fructose; (3): glucose; (4): galactose; I.S: internal standard (phenyl-β-glucoside); lb: (5)

429 lactose, (6) allolactose; (7) β -1,6-galactobiose.





467 Fig. 4. Galactooligosaccharide (GOS) formation during enzymatic hydrolysis for 30 h of a commercial UHT milk using Lactozym[®] pure 6500 (0.3
 468 U/mL) at 30 °C and aseptic conditions (6'-β-gal-lac: 6'-galactosyl-lactose, 4'-β-gal-lac: 4'-galactosyl-lactose, Unk GOS: unknown

469 galactooligosaccharides)



471 **Table 1**

472 Content of mono- and disaccharides and galactooligosaccharides (GOS) found in commercial lactose-free UHT milks (LF-UHT milks).

473

Commercial lactose-free UHT milks								
Carbohydrates								
(mg/L milk)	1	2	3	4	5	6	7	$\mathbf{Mean} \pm \mathbf{SD}$
Galactose	23464	17297	20848	21386	21493	20211	22115	20974 ± 2044
Glucose	25050	19881	22901	22979	23227	21522	23712	22753 ± 1777
Fructose	132	111	76	185	118	106	94	117 ± 33
Tagatose	88	79	65	75	98	94	36	76 ± 20
Lactose	66	291	101	69	74	33	38	96 ± 83
Allolactose	283	841	446	229	349	139	151	348 ± 226
β-1-6-Galactobiose	339	747	409	248	335	166	191	348 ± 183
6'-galactosyl-lactose	936	1703	1070	687	805	271	302	825 ± 460
4'-galactosyl-lactose	42	39	31	254	28	31	32	$65\pm~77$
Unidentified GOS	635	1020	669	546	487	349	355	580 ± 220
Total GOS ^a	2235	4350	2625	1735	2004	956	1031	2134 ± 1080

⁴⁷⁴ ^a Total GOS includes quantification of: allolactose, β -1-6-galactobiose, 6'-galactosyl-lactose, 4'-galactosyl-lactose and unidentified GOS

475 (Marked as * in figure 1). Samples were prepared and analyzed in duplicate.

476

478 Table 2

Content of mono- and disaccharides and galactooligosaccharides (GOS) found in commercial lactose-free UHT dairy drinks (LF-UHT dairy 479 480 drinks).

481

	Lactose-free UHT dairy drinks					
Carbohydrates						
(mg/L milk)	1	2	3	4	5	$Mean \pm SD$
Galactose	16254	22277	13519	16168	12669	16177 ± 3436
Glucose	17372	23478	14355	17455	13534	17238 ± 3575
Fructose	389	39	309	348	187	255 ± 128
Tagatose	192	Tr	72	68	188	131 ± 61
Lactose	38	60	34	21	25	36 ± 14
Sucrose	9	209	180	83	247	146 ± 88
Allolactose	97	142	74	75	68	91 ± 29
β-1-6-Galactobiose	110	172	83	87	135	117 ± 43
6'-galactosyl-lactose	320	445	141	59	135	220 ± 146
4'-galactosyl-lactose	14	88	20	16	141	56 ± 51
Unidentified GOS	272	1418	524	365	695	655 ± 410
Total GOS ^a	813	2265	842	602	1048	1114 ± 597

482

483 ^a Total GOS includes quantification of: allolactose, β-1-6-galactobiose, 6'-galactosyl-lactose, 4'-galactosyl-lactose and unidentified GOS

484 (Marked as * in figure 1). Samples were prepared and analyzed in duplicate. Tr: trace (LOD = 5 mg/L)

485