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Contribution of a tannase from *Atopobium parvulum* DSM

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20469^T in the oral processing of food tannins

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21 ABSTRACT

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23 During oral passage, food tannins interact with the microbiota present in the oral
24 cavity. *Atopobium parvulum* strains are inhabitants of the human oral cavity. A gene
25 encoding a protein similar to bacterial tannases is present in *A. parvulum* strains. The
26 *tanA_{Ap}* (*apar_1020*) gene was cloned and expressed in *Escherichia coli* BL21 (DE3). The
27 overproduced TanA_{Ap} protein was purified to homogeneity. It exhibited optimal activity at
28 pH 6.0 and broad temperature range, being these properties compatible with its action
29 during food oral processing. However, purified TanA_{Ap} protein presented the lowest
30 specific activity among bacterial tannases (3.5 U/mg) and was unable to hydrolyze
31 complex tannin, such as tannic acid. These biochemical properties discard a main role of
32 TanA_{Ap} in the breakdown of complex food tannins during oral processing.

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37 *Keywords:*

38 Tannase, Food processing, Hydrolase, Gallic acid, Oral microbiome

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1. Introduction

Vegetable tannins are abundant in plants utilized as human food. Tannins occur widely on common foodstuffs, such as pomegranate, banana, strawberry, grape, cashew nut, and hazelnut. Drinks like wine and tea also contain these phenolic compounds (Shahidi & Naczki, 2003). The molar mass of tannins affects tannin characteristics directly, and it has been suggested that small molecule tannins have more antioxidant activity (Ordoudi & Tsimidou, 2006). To understand the biological effects of food tannins the insight on the metabolic fate and bioavailability of these metabolites in the human body is crucial, however this knowledge is currently scarce (Rechner, Kuhnle, Bremner, & Hubbard, 2002; Moco, Martin, & Rezzi, 2012). The interaction of tannins with human microbiota will determine in great extent the physiological effects of these polyphenols.

Although much work has been focused on factors that determine mechanical (e.g. rheological and fracture) and sensory properties of foods, far less attention has been paid to linking food transformations that occurs during oral processing with microbial action (Chen, 2009). The human microbiome is a dynamic community changing in response to natural perturbations such as diet (Turnbaugh, Hamady, Yatsunencko, Cantarel, Duncan, Ley, et al., 2009; Wu, Chen, Hoffmann, Bittinger, Chen, Keilbaugh, et al., 2011; Spencer, Hamp, Reid, Fishe, Zeisel, & Fodor, 2011; Zhang, Zhang, Wang, Han, Cao, Hua, et al., 2010). The oral cavity of humans hosts several hundred taxa, with remarkable diversity even among saliva, tongue, teeth, and other substrates (Segata, Haake, Mannon, Lemon, Waldron, Gevers, et al., 2012; Dewhirst, Chen, Izard, Paster, Tanner, Yu, et al., 2010). All of the surfaces of the mouth are covered in a bacterial biofilm (Wade, 2013). Studies on the tongue biofilm have been relatively few in number, compared with the significant number of investigations of dental plaque and the microbiota associated with periodontal

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65 disease and dental caries. The tongue is known to harbour a very diverse microbiota at
66 high cell density. Among the many bacteria present in the oral cavity, the species
67 *Atopobium parvulum* is of interest because its members are frequently isolated from the
68 human oral cavity, especially from the tongue (Riggio, Lennon, Rolph, Hodge, Donalson,
69 Maxwell, et al., 2008; Copeland, Sikorski, Lapidus, Nolan, del Rio, Lucas, et al., 2009).
70 The genome of *A. parvulum* type strain (IPP 1246^T) has been complete sequenced. An
71 ORF (*apar_1020*) **encoding a “putative uncharacterized protein” had 40% and 26%**
72 identity to TanA_{SI}, a tannase from *Staphylococcus lugdunensis*, and TanB_{Lp} (formerly
73 called TanLp1), a tannase from *Lactobacillus plantarum*, respectively.

74 The enzyme tannase (or tannin acyl hydrolase EC 3.1.1.20) belongs to the serine
75 esterases, catalyzing the hydrolysis of the ester bond (galloyl ester of an alcohol moiety)
76 and depside bond (galloyl ester of gallic acid) in tannins to release gallic acid (Aguilar,
77 Rodríguez, Gutierrez-Sánchez, Augur, Favela-Torres, Prado-Barragan, et al., 2007;
78 Chávez-González, Rodríguez-Durán, Balagurusamy, Prado-Barragán, Rodríguez,
79 Contreras, et al., 2012). Tannase activity contributes to the hydrolysis of natural tannins
80 present in the diet. Although bacteria possessing tannase activity, such as *L. plantarum*,
81 *Streptococcus gallolyticus*, and *Staphylococcus lugdunensis*, have been described in the
82 human gastrointestinal tract (Abdulmir, Hafidh, Mahdi, Al-Jeboori, & Abubaker, 2009;
83 Abdulmir, Hafidh, & Bakar, 2011; Noguchi, Ohashi, Shiratori, Narui, Hagiwara, Ko, et
84 al., 2007; Rusniok, Couvé, de Cunha, El Gana, Zidane, Bouchier, et al., 2010), there are
85 still many questions about the oral metabolism of food tannins. It will be interesting to
86 know if during the short period of oral processing, tannin hydrolysis began in the mouth.
87 The presence of a protein similar to bacterial tannases in *A. parvulum*, a species abundant
88 in the oral cavity, will be important for predicting their contribution to food tannin
89 breakdown. Therefore, the objective of this study was to find out the potential contribution

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90 of TanA_{Ap} (Apar_1020) protein from *A. parvulum* to tannin hydrolysis during food oral
91 processing.

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93 2. Materials and methods

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95 2.1. Strains, plasmids, and materials

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97 *A. parvulum* DSM 20469^T (IPP 1246^T, ATCC 33793^T) used through this study was
98 purchased from the DSM (German Collection of Microorganisms and Cell Cultures).

99 *Escherichia coli* DH10B was used as host strain for all DNA manipulations. *E. coli* BL21
100 (DE3), providing a T7 RNA polymerase, was used for heterologous expression in the
101 pURI3-Cter vector (Curiel, de las Rivas, Mancheño, & Muñoz, 2011). *E. coli* strains were
102 cultured in Luria-Bertani (LB) medium at 37 °C and shaking at 200 rpm. When required,
103 ampicillin was added to the medium at a concentration of **100 µg/mL**.

104 Plasmid DNA was extracted by a High Pure plasmid isolation kit (Roche). PCR
105 product was purified with a QIAquick gel extraction kit (Quiagen). Oligonucleotides were
106 purchased from Eurofins MWG Operon (Ebersberg, Germany). *DpnI* and Prime STAR HS
107 DNA polymerase were obtained from Takara. His-tagged protein was purified by a Talon
108 Superflow resin (Clontech). The compounds assayed in the study were methyl gallate
109 (Fluka), ethyl gallate (Aldrich), propyl gallate (Sigma), lauryl gallate (Aldrich), ethyl
110 protocatechuate (ethyl 3,4-dihydroxybenzoate) (Aldrich), and tannic acid (Sigma).

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112 2.2. Cloning of TanA_{Ap}

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114 The gene encoding for a putative tannase (*apar_1020*, or *tanA_{Ap}*) in *A. parvulum*
115 DSM20469^T (accession YP_003180040) was amplified by PCR by using the primers 1394
116 (5′–*ACTTTAAGAAGGAGATATACATatgtctgataatacgaatcaacctgca*) and 787 (5′–
117 *GCTATTAATGATGATGATGATGATGagcagacgcacacgagacaatcca*) (the nucleotides pairing
118 the expression vector sequence are indicated in italics, and the nucleotides pairing the
119 *tanA_{Ap}* gene sequence are written in lowercase letters). As a peptide signal was predicted
120 in the TanA_{Ap} sequence, oligonucleotides 1394 and 787 were used to clone TanA_{Ap}
121 lacking the 23-amino acid peptide signal sequence. Prime Star HS DNA polymerase
122 (TaKaRa) was used for the PCR amplification. The 1.7-kb purified PCR product was
123 inserted into the pURI3–Cter vector using a restriction enzyme–and ligation–free cloning
124 strategy (Curiel et al., 2011). The vector produces recombinant proteins having a six–
125 histidine affinity tag in their C–termini. *E. coli* DH10B cells were transformed,
126 recombinant plasmids were isolated, and those containing the correct insert were identified
127 by size, verified by DNA sequencing, and then transformed into *E. coli* BL21 (DE3) cells
128 for expression.

130 2.3. Enzyme production and purification

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132 *E. coli* BL21(DE3) harbouring the recombinant plasmid pURI3–Cter–TanA_{Ap} was
133 **grown in LB medium containing 100 µg/mL ampicillin on a rotary shaker (200 rpm) at 37**
134 °C until an optical density (OD) at 600 nm of 0.4 was reached. Isopropyl-β-D–
135 thiogalactopyranoside (IPTG) was added to a final concentration of 0.4 mM and protein
136 induction was continued at 22 °C during 18 h.

137 The induced cells were harvested by centrifugation (8,000 *g*, 15 min, 4 °C),
138 resuspended in phosphate buffer (50 mM, pH 6.5) and disrupted by French Press passages

139 (three times at 1,100 psi). The insoluble fraction of the lysate was removed by
140 centrifugation at 47,000 *g* for 30 min at 4 °C, and the supernatant was filtered through a
141 **0.2 μm** pore-size filter and then loaded onto a Talon Superflow resin (Clontech)
142 equilibrated in phosphate buffer (50 mM, pH 6.5) containing 300 mM NaCl and 10 mM
143 imidazole to improve the interaction specificity in the affinity chromatography step. The
144 bound enzyme was eluted using 150 mM imidazole in the same buffer. The purity of the
145 enzyme was determined by SDS-PAGE in Tris-glycine buffer. Fractions containing the
146 His6-tagged protein were pooled and analyzed for tannase activity.

148 *2.4. Enzyme activity assay*

150 Tannase activity was determined using a colorimetric assay using rhodanine, specific
151 for gallic acid (Inoue & Hagerman, 1988). Rhodanine reacts only with gallic acid and not
152 with galloyl esters or other phenolics. Gallic acid analysis in the reactions was determined
153 using the following colorimetric assay. Tannase enzyme (100 μg) in 700 μl of 50 mM
154 **phosphate buffer pH 6.5 was incubated with 40 μl of 25 mM methyl gallate (1 mM final**
155 **concentration) during 5 min at 37 °C. After this incubation, 150 μl of a methanolic**
156 rhodanine solution (0.667% w/v rhodanine in 100% methanol) was added to the mixture.
157 After 5 min incubation at 30 °C, **100 μl of 500 mM KOH was added. After an additional**
158 incubation of 5–10 min, the absorbance at 520 nm was measured on a spectrophotometer.
159 A standard curve using gallic acid concentration ranging from 0.125 to 1 mM was
160 prepared. One unit of tannase activity was defined as the amount of enzyme required to
161 **release 1 μmol** of gallic acid per minute under standard reaction condition.

163 *2.5. Determination of pH and temperature effects on tannase activity*

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2 165 The optimum pH value of TanA_{Ap} was determined by measuring its activity at
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4 166 different pH values (3.0–10.0). The following buffers all at 100 mM were used for the
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6 167 assay: acetic acid–sodium acetate (pH 3.0–5.0), citric acid–sodium citrate (pH 6), sodium
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8 168 phosphate (pH 7), Tris–HCl (pH 8), glycine–NaOH (pH 9), and sodium carbonate–
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10 169 bicarbonate (pH 10). The rhodanine assay was used for the optimal pH characterization of
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12 170 tannase. Since the rhodanine–gallic acid complex forms only in basic conditions, after the
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14 171 completion of the enzymatic degradation of methyl gallate, KOH was added to the
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16 172 reaction mixture to ensure that the same pH value (pH 11) was achieved in all samples
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18 173 assayed. Determinations were done in triplicate.

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23 174 The optimum temperature of TanA_{Ap} was assayed by incubating the purified
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25 175 protein in 25 mM phosphate buffer (pH 6.5) at seven different temperatures in the range of
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27 176 4–65 °C (4, 22, 30, 37, 45, 55 and 65 °C). To study the thermal stability of TanA_{Ap}, tannase
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29 177 was incubated at temperatures over the range of 22–65 °C (22, 30, 37, 45, 55 and 65 °C)
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31 178 for 30 min and 2, 4, 6, and 18 h. Aliquots were withdrawn in triplicate at these incubation
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33 179 times to test the remaining activity at standard conditions. The non-heated enzyme was
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35 180 considered as control (100%).

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42 43 182 *2.6. Effect of metal ions, reagents, and inhibitors on tannase activity*

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48 184 The effect of various metal ions (Mg²⁺, K⁺, Ca²⁺, Hg²⁺, and Zn²⁺), metal chelator
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50 185 EDTA, surfactants (SDS, Tween 80, and Triton–X–100), and other reagents (DMSO, and
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52 186 β–mercaptoethanol) on tannase activity was investigated by the rhodanine assay using
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54 187 methyl gallate as substrate. Purified TanA_{Ap} was incubated with additives (1 mM final
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56 188 concentration) at 30 °C for 1h. After incubation, the residual activity was measured in
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189 triplicate under the standard assay conditions. The relative activities were calculated with
190 respect to the control where the reaction was carried out in the absence of additives
191 (100%).

193 *2.7. HPLC-DAD analysis of substrate specificity*

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195 The substrate specificity of TanA_{Ap} was determined using five commercial
196 hydroxybenzoic esters (methyl gallate, ethyl gallate, propyl gallate, lauryl gallate, and
197 ethyl protocatechuate), and a hydrolyzable tannin (tannic acid). The standard enzyme
198 **assay was modified by using 200 µg of TanA_{Ap}**, and 1 mM substrate, in the reaction
199 mixture and incubated at 37 °C during 10 min. As controls, phosphate buffer containing
200 the reagents but the enzyme were incubated in the same conditions.

201 The reaction products were extracted twice with ethyl acetate (Lab-Scan, Ireland)
202 and analyzed by HPLC-DAD. A Thermo (Thermo Electron Corporation, Waltham,
203 Massachusetts, USA) chromatograph equipped with a P400 SpectraSystem pump, and
204 AS3000 autosampler, and a UV6000LP photodiode array detector were used. A gradient
205 of solvent A (water/acetic acid, 98:2, v/v) and solvent B (water/acetonitrile/acetic acid,
206 78:20:2, v/v/v) was applied to a reverse-phase Nova-pack C₁₈ (25 cm × 4.0 mm i.d.) 4.6
207 µm particle size, cartridge at room temperature as follows: 0–55 min, 80% B linear, 1.1
208 ml/min; 55–57 min, 90% B linear, 1.2 ml/min; 57–70 min, 90% B isocratic, 1.2 ml/min;
209 70–80 min, 95% B linear, 1.2 ml/min; 80–90 min, 100% linear, 1.2 ml/min; 100–120 min,
210 washing 1.0 ml/min, and reequilibration of the column under initial gradient conditions.
211 Detection was performed by scanning from 220 to 380 nm. Samples were injected onto the
212 cartridge after being **filtered through a 0.45 µm PVDF filter. The identification of**

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213 degradation compounds was carried out by comparing the retention times and spectral data
214 of each peak with those of standards from commercial suppliers
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216 3. Results and discussion
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218 3.1. Production and characterization of purified TanA_{Ap}
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220 Tannases are capable of hydrolyzing complex tannins, which represent an
221 important chemical group occurring in food plants. Studies related to tannase-producing
222 strain isolation have been conducted (Aguilar et al., 2007; Chávez-González et al., 2012).
223 Among the many bacteria present in the oral cavity, *Atopobium parvulum* strains are of
224 interest because they possess a gene putatively encoding a tannase. The *apar_1020* (*tanA_{Ap}*)
225 gene predicted to encode a 607 amino acid protein 39.5% identical to TanA_{SI}, a tannase
226 from *Staphylococcus lugdunensis*, and 26% identical to TanB_{Lp}, a tannase from
227 *Lactobacillus plantarum*, the only two bacterial tannases genetically described so far (data
228 not shown). Noteworthy, TanA_{SI} and TanB_{Lp} are only 27% identical among them
229 (Iwamoto et al., 2008). In TanA_{Ap} a signal peptide was predicted with a cleavage site at
230 residue 23. Therefore, processed mature TanA_{Ap} protein has 584 amino acid residues, with
231 a predicted molecular mass of 63.8 kDa, and an isoelectric point of 4.62. As TanA_{Ap}
232 showed high amino acid identity to bacterial tannases, the tannase activity of TanA_{Ap}
233 needs to be assayed.
234 The *tanA_{Ap}* gene was cloned into the pURI3-Cter expression vector by a ligation –
235 free cloning strategy described previously (Curiel et al., 2011). The vector incorporates the
236 DNA sequence encoding hexa-histidine to create a His-tagged fusion enzyme for further
237 purification step. The integrity of the construct was confirmed by DNA sequencing. The

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238 recombinant plasmid was transformed into *E. coli* BL21 (DE3) and expressed under the
239 control of an inducible IPTG promoter. Cell extracts were used to detect the presence of
240 overproduced proteins. SDS-PAGE analysis showed that there was one major band of
241 protein, approximately 66 kDa, in the intracellular soluble fraction of the pURI3-Cter-
242 TanA_{Ap} cells, which was absent in the control pURI3-Cter cells (Fig.1). The molecular
243 weight of the overproduced protein was consistent with the calculated molecular weight of
244 TanA_{Ap}. Since the cloning strategy would yield a His-tagged protein variant, *A. parvulum*
245 pURI3-Cter-TanA_{Ap} could be purified on an immobilized metal affinity chromatography
246 (IMAC) resin. The recombinant protein was observed as single band on SDS-PAGE (Fig.
247 1). Routinely about 12 mg of purified protein from 1-liter culture was obtained.

248 TanA_{Ap} protein purified by the affinity resin was biochemically characterized. A
249 method specific for the detection of gallic acid could be used for a reliable quantification
250 of tannase activity. Since tannase catalyzes the hydrolysis of the galloyl ester linkage
251 liberating gallic acid, the activity of tannase could be measured by estimating the gallic
252 acid formed due to enzyme action (Mueller-Harvey, 2001). Inoue & Hagerman (1988)
253 described a rhodanine assay specific for determining free gallic acid. Rhodanine reacts
254 with gallic acid to give a red complex with a maximum absorbance at 520 nm. Rhodanine
255 assay was used to determine the specific activity of TanA_{Ap}, simultaneously, the activity
256 of the previously described TanB_{Lp} tannase from *L. plantarum* was also determined as
257 reference. Using methyl gallate as substrate, the specific activity of TanA_{Ap} purified
258 enzyme was 3.5 U/mg, 116 times lower than that of TanB_{Lp} (408 U/mg). This low specific
259 activity could indicate that even though tannase action from *A. parvulum* could begin
260 almost immediately after food ingestion, its contribution to tannin breakdown might not be
261 relevant.

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262 In relation to the biochemical properties of the enzyme, TanA_{Ap} showed optimal
263 activity at pH 6 (Fig. 2A), slightly more acidic than the optimal pH for TanB_{Lp} (pH 7).
264 During food oral processing, saliva provides buffering effects. It was indicated that the pH
265 of saliva rises during the first 5 min after the intake of most foods, and falls to around 6, or
266 lower, approximately 15 min after food consumption (Humphrey & Williamson, 2001).
267 Therefore, TanA_{Ap} could find an adequate pH for activity during food oral processing.

268 Despite the optimum temperature of TanA_{Ap} was 55 °C, at 37 °C, the physiological
269 temperature for humans, 80% of the maximal activity was found. Similarly, more than
270 80% maximal activity was obtained at 20, 42, and 65 °C (Fig. 2B). Tannase TanB_{Lp} from
271 *L. plantarum* showed maximal activity at 40 °C, having only 50% of the maximal activity
272 at 30 or 60 °C. The thermal stability profile for TanA_{Ap} is shown in Fig. 2C. According to
273 the thermal stability profile, TanA_{Ap} was most stable at temperatures between 37 and 65
274 °C, and more than 60% enzyme activity remained after 18 h at 45 °C (Fig. 2C). Tannase
275 TanB_{Lp} from *L. plantarum* kept less than 20% of the maximal activity after incubation at
276 37 °C during 20 h. The data for TanA_{Ap} demonstrated that the enzyme exhibited high
277 thermal stability under prolonged incubation up to 45 °C. Despite the specific activity of
278 TanA_{Ap} is remarkably lower than the activity of TanB_{Lp}, TanA_{Ap} is more thermostable,
279 and therefore it is able to resist thermal unfolding in the absence of its substrate.

280 The effects of several ions and additives are shown in Table 1. Contrarily to
281 TanB_{Lp}, CaCl₂ did not activate TanA_{Ap} (Curiel, Rodríguez, Acebrón, Mancheño, de las
282 Rivas, & Muñoz, 2009). Tannase activity of TanA_{Ap} was not increased by any of the
283 additives assayed. Similarly to TanB_{Lp}, activity was greatly inhibited by β -
284 mercaptoethanol and by the metal ion Hg²⁺ (Curiel et al., 2009). ZnCl₂ significantly
285 inhibited TanA_{Ap} activity (relative activity 24%). The other metal ions and additives
286 assayed partially affected tannase activity (relative activity 74–96%). The different

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287 additive effect observed would suggest that there are notable structural differences among
288 both bacterial tannases, TanA_{Ap} from *A. parvulum*, and TanB_{Lp} from *L. plantarum*.

289 Despite the low specific activity showed by TanA_{Ap}, this protein posses
290 biochemical properties compatible with its action during food oral passage, since its pH
291 and temperature for activity are provided by the human saliva during food processing.
292 Oral processing occurs during a short time period, however, it has been described that
293 during the short period of oral processing, about 50% of bread and 25% of pasta starch are
294 hydrolyzed and transformed into smaller molecules by the amylase enzyme present in the
295 human saliva (Hoebler, Karinthe, Devaux, Guillon, Gallant, Bouchet, et al., 1998; Hoebler,
296 Devaux, Karinthe, Belleville, & Barry, 2000). The interaction of amylase enzyme with
297 starch ingredients produces almost an immediate effect on hydrolysis, and thus making the
298 food intake much easily mixable and digestible in the stomach. A similar situation could
299 be envisaged for the action of TanA_{Ap} on the tannins present on the diet.

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301 *3.2. Contribution of TanA_{Ap} to the hydrolysis of tannins from the diet*

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303 Tannins are natural polyphenolic compounds present in food plants. They are
304 characterized by their ability to form strong complexes with different minerals and
305 macromolecules, such as proteins, cellulose, starch, etc, causing astringency and
306 precipitation effects (Mingshu, Kai, Quiang, & Dongying, 2006; Serrano, Puupponen–
307 Pimiä, Dauer, Aura, & Saura-Calixto, 2009). As a result, tannins are considered
308 antinutritional. Tannases catalyzes the hydrolysis reaction of ester bonds present in the
309 gallotannins, complex tannins, and gallic acid and protocatechuic acid esters (Aguilar et
310 al., 2007; Curiel et al., 2009; Chávez-González et al., 2012). In order to know the substrate
311 specificity of TanA_{Ap}, several gallate and protocatechuate esters were assayed. As showed

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312 in Fig. 3, none of the esters assayed were significantly hydrolyzed. Methyl, ethyl, and
313 propyl gallate and ethyl protocatechuate were minimally hydrolyzed. Lauryl gallate,
314 possessing a long aliphatic alcohol chain, was not hydrolyzed at all (data not shown).
315 Contrarily to these results, *L. plantarum* tannase (TanB_{Lp}) was able to fully hydrolyze
316 gallic esters even those having an alcohol substituent as longer as lauryl (C12) (Curiel et
317 al, 2009). Structural differences among both bacterial proteins will be responsible of the
318 different spatial requirements observed for tannase activity.

319 It is noteworthy to mention that the colorimetric rhodanine assay used for the
320 detection and quantification of tannase activity is much more sensitive than the analysis of
321 the reaction products by HPLC. By using methyl gallate as substrate, the rhodanine assay
322 allowed to determine properly the biochemical properties of TanA_{Ap} assayed; however,
323 this characterization would not be possible by the HPLC analysis.

324 In order to evaluate the contribution of TanA_{Ap} action during oral processing of
325 diet tannins, a complex and natural tannin, tannic acid, was incubated in the presence of
326 TanA_{Ap}. Tannic acid is almost exclusively formed by poly-galloyl glucose derivatives
327 whose nature and complexity vary with the plant source. When TanA_{Ap} was incubated on
328 tannic acid, an hydrolysis profile identical to the control without enzyme was observed
329 (data not shown). This was an expected result considering the minimal degradation on
330 simple gallic acid esters observed after TanA_{Ap} action. As TanA_{Ap} did not show activity
331 on tannic acid, it could be possible that the natural tannin substrate for this enzyme will be
332 different and still remained unknown. In addition, specific reaction conditions or the
333 presence of an unknown cofactor will be required to increase TanA_{Ap} activity during food
334 processing. Further research will be needed to know the physiological role of TanA_{Ap} in *A.*
335 *parvulum* metabolism.

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336 The above results indicated that, even though *A. parvulum* tannase action could
337 begin almost immediately after food ingestion, its contribution to tannin breakdown would
338 not be relevant. Most of the tannin digestion could result from bacterial intestinal tannases
339 rather than from oral tannase. In the microbiome of the major site of food tannin
340 hydrolysis, the intestinal tract, at least three tannase-producing bacteria have been isolated,
341 *L. plantarum*, *S. lugdunensis* or *Streptococcus gallolyticus* (Iwamoto, Tsuruta, Nishitani,
342 & Osawa, 2008; Noguchi et al., 2007; Sly, Cahill, Osawa, & Fujisawa, 1997; Rusniok et
343 al., 2010). From these intestinal bacteria, only the biochemical properties of TanB_{LP} from
344 *L. plantarum* have been studied. TanB_{LP} possesses adequate properties for intestinal tannin
345 degradation. However, further testing would be required to define the metabolism of these
346 phenolic compounds comprehensively. In particular, the activity of the complex
347 communities of microorganisms present in all parts of the human digestive tract would
348 need to be examined.

349

350 4. Conclusions

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352 In the present study, a novel bacterial tannase namely TanA_{Ap} from *Atopobium*
353 *parvulum*, an inhabitant of the human oral cavity, was purified. TanA_{Ap} was biochemically
354 characterized by using a sensitive colorimetric method. Among bacterial tannases, TanA_{Ap}
355 possessed low specific activity and was unable to hydrolyze complex tannins. These
356 biochemical properties are not favourable for the breakdown of complex food tannins
357 during oral processing.

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359 Acknowledgements

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468 Figure captions

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470 Fig. 1. Purification of TanA_{Ap} tannase from *A. parvulum*. Analysis by SDS-PAGE of
471 soluble cell extracts of IPTG-induced *E. coli* BL21(DE3) (pURI3-Cter) (1) or *E. coli*
472 BL21(DE3) (pURI3-Cter-TanA_{Ap}) (2), flowthrough (3), or fractions eluted after His affinity
473 resin (4-6). The gel was stained with Coomassie blue. Molecular mass markers are located
474 at the left (SDS-PAGE Standards, Bio-Rad).

475
476 Fig. 2. Biochemical properties of TanA_{Ap} protein. (A) pH-activity profile of TanA_{Ap}. (B)
477 Temperature-activity profile of TanA_{Ap}. (C) Thermal stability profile for TanA_{Ap} after
478 preincubation at 22 °C (filled diamond), 30 °C (filled square), 37 °C (filled triangle), 45 °C
479 (cross), 55 °C (star), and 65 °C (filled circle) in phosphate buffer (50 mM, pH 6.5); at
480 indicated times, aliquots were withdrawn, and analyzed as described in the Materials and
481 Methods section. The experiments were done in triplicate. The mean value and the
482 standard error are showed. The percentage of residual activity was calculated by
483 comparing with unincubated enzyme.

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2 485 Fig. 3. Enzymatic activity of tannase from *A. parvulum* against gallic and protocatechuic
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4 486 acid esters. Hydrolase activity of purified TanA_{Ap} compared with control reactions on
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6 487 which the enzyme was omitted. HPLC chromatograms of TanA_{Ap} (200 µg) **incubated in**
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9 488 50 mM phosphate buffer pH 6, and 1 mM of methyl gallate (A), ethyl gallate (B), propyl
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11 489 gallate (C), and ethyl protocatechuate (D). The methyl gallate (MG), ethyl gallate (EG),
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14 490 propyl gallate (PG), ethyl protocatechuate (EP), gallic acid (GA), and protocatechuic acid
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16 491 (PA) detected are indicated. The chromatograms were recorded at 280 nm.
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Figure 1

Figure 1

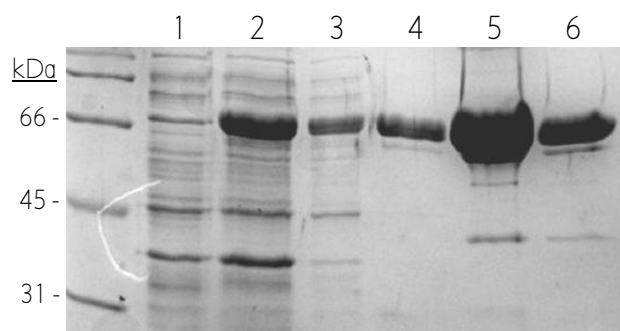


Figure 2

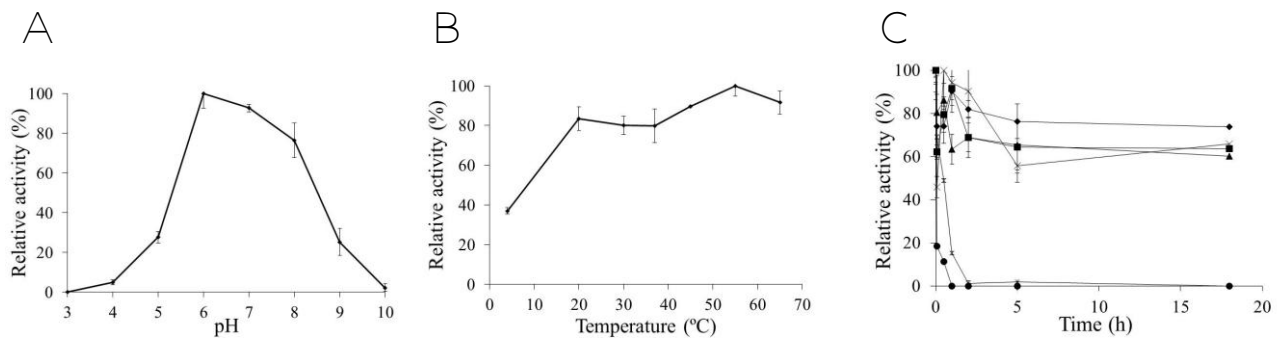


Figure 3

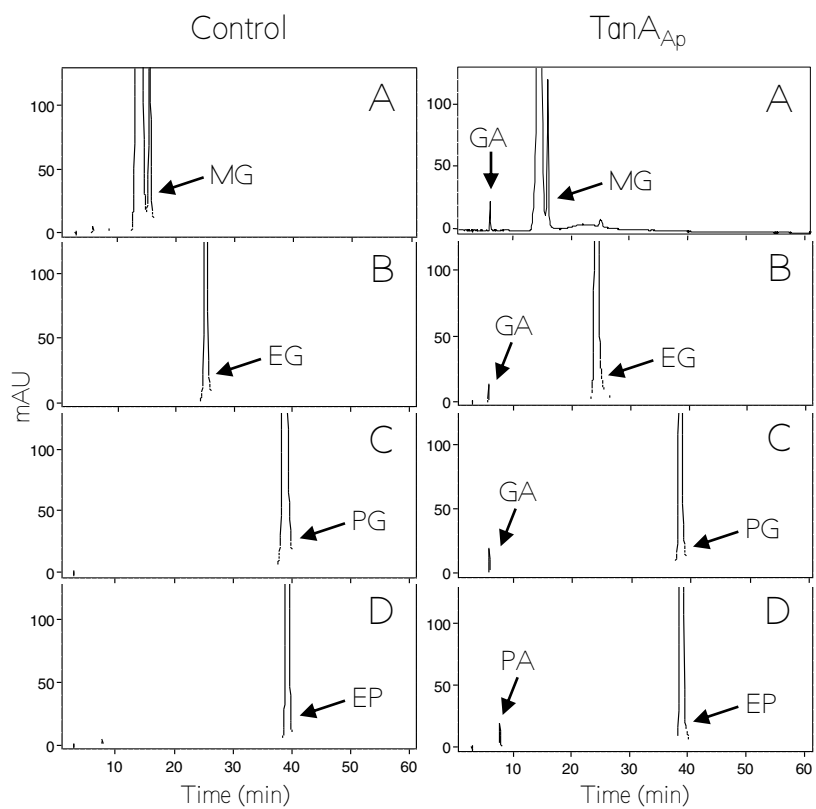


Table 1

Table 1. Effect of additives on *A. parvulum* tannase activity

Additions (1 mM)	Relative activity (%)
Control	100
EDTA	74
KCl	80
HgCl ₂	16
CaCl ₂	97
MgCl ₂	96
ZnCl ₂	24
Triton X 100	83
DMSO	82
Tween 80	94
Urea	83
β -mercaptoethanol	15