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48	Abstract	A decontamination been developed aromatic hydroca Agency priority p PAHs and NP we contamination le European Union in the total PAHs also present. Not by the European content in four of of 50 mg kg ⁻¹ for values below the sludge as agricu hydroxypropyl-β- decontamination content was rem whereas after thr	ion technique based in cyclodextrin extraction has to eliminate nonylphenol (NP) and 16 polycyclic arbons (PAHs; the US Environmental Protection pollutants list) from sewage sludge. In a first step, ere characterised in six sludges to determine evels according to limit values proposed by the Sludge Directive draft. There were few variations is content with levels of 1.88 to 3.05 mg kg ⁻¹ . Spredominated, but fluoranthene and pyrene were ne of the sludge exceeded the PAHs limit proposed in Union's draft Directive. On the contrary, NP of the six sludges was over the recommended limits NP ethoxylates. With the aim of obtaining NP e concentration limits proposed to use the sewage ltural amendments, a preliminary study using -cyclodextrin (HPBCD) extractions as a in technique was carried out. About 90 % of NP oved with only one extraction with HPBCD, ree sequential extractions using an aqueous HPBCD, the NP extraction percentage was less

		than 1 %. Simultaneously, PAHs extraction percentages obtained with HPBCD were also much higher than when aqueous solution was used, especially in the case of two- and three-ring PAHs. Finally, the potential environmental hazard of HPBCD leachates to aquatic organisms (<i>Daphnia magna</i>) was tested. These results indicate that the treatment of sewage sludge with cyclodextrin could allow their safe use as fertiliser in agriculture.
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RESEARCH ARTICLE

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4 Decontamination of polycyclic aromatic hydrocarbons

- ⁵ and nonylphenol from sewage sludge using
- 6 hydroxypropyl-β-cyclodextrin and evaluation
- 7 of the toxicity of leachates

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15Abstract A decontamination technique based in cyclodextrin extraction has been developed to eliminate nonylphenol 16(NP) and 16 polycyclic aromatic hydrocarbons (PAHs; the 17 18 US Environmental Protection Agency priority pollutants list) from sewage sludge. In a first step, PAHs and NP were 1920 characterised in six sludges to determine contamination 21levels according to limit values proposed by the European Union Sludge Directive draft. There were few variations in 22the total PAHs content with levels of 1.88 to 3.05 mg kg⁻¹. 23Three-ring PAHs predominated, but fluoranthene and pyrene 2425were also present. None of the sludge exceeded the PAHs limit proposed by the European Union's draft Directive. On 26the contrary, NP content in four of the six sludges was over 27the recommended limits of 50 mg kg^{-1} for NP ethoxylates. 28With the aim of obtaining NP values below the concentration 29limits proposed to use the sewage sludge as agricultural 30 31amendments, a preliminary study using hydroxypropyl-\beta-32cyclodextrin (HPBCD) extractions as a decontamination technique was carried out. About 90 % of NP content was 33 removed with only one extraction with HPBCD, whereas 34after three sequential extractions using an aqueous solution 3536 without HPBCD, the NP extraction percentage was less than 371 %. Simultaneously, PAHs extraction percentages obtained with HPBCD were also much higher than when aqueous 38

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solution was used, especially in the case of two- and three-39ring PAHs. Finally, the potential environmental hazard of40HPBCD leachates to aquatic organisms (*Daphnia magna*)41was tested. These results indicate that the treatment of sew-42age sludge with cyclodextrin could allow their safe use as43fertiliser in agriculture.44

Keywords Polycyclic aromatic hydrocarbons ·	45
Nonylphenol · Sewage sludge · Cyclodextrin · Toxicity	46

Introduction

After the progressive implementation of the Directives 48 91/271/EEC and 98/15/EEC concerning urban wastewater 49treatment, many countries are increasing the number of 50wastewater treatment plants (WWTP) to avoid the spread 51of disease, remove organic matter and some pollutants and 52preserve the quality of surface waters. Within a WWTP, large 53amounts of sewage sludge are generated which require dis-54posal. Final destination of sewage sludge used to be inciner-55ation or landfills (Pousada-Ferradas et al. 2012), but a useful 56and interesting option is the production of compost and the 57direct application of stabilised sludge to land (Laturnus et al. 582007). Although there are potential hazards in this practice, 59such as spread of pathogens, odours, heavy metals and 60 persistent toxic organics, sludge application to agricultural 61land have positive effects on soil, such as modification of soil 62 structure and addition of organic matter and nutrients for 63 plants, such as N, P and K (Morillo et al. 2002; Aparicio 64 et al. 2009). 65

The use of sewage sludge in agriculture is regulated by the 66 European Union's (1986) Council Directive 86/278/EEC. 67

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This Directive controls the quality of sludge, the amended soils, the loading rate and the crops that may be grown on treated land. This Directive requires that heavy metal contents (Cd, Cr, Cu, Ni, Pb and Zn) in both the sludge and soil comply with the established limits, but it does not yet establish maximum values for organic contaminants.

74To limit the concentration for certain organic pollutants in 75both sewage sludge and sludge-amended soils, the European Union (2000) launched a working document on sludge that 76although not being a formal regulation, the third draft of this 77document limits the values for some organic compounds, 78 79 including polycyclic aromatic hydrocarbons (PAHs; the sum of phenanthrene, fluorene, fluoranthene, pyrene, benzo(b) 80 fluoranthene, benzo(k)fluoranthene, benzo(a)pyrene, benzo 81 (ghi)pervlene, and indeno(1,2,3-c,d)pyrene) and the sum of 82 nonylphenol (NP) and nonylphenol mono (NP1EO) and di-83 ethoxylates (NP2EO; referred as NPE). Concentration limits 84 fixed in this Directive draft for land application of sludge are 85 50 mg kg⁻¹ dry matter for NPE and 6 mg kg⁻¹ for the sum of 86 87 these nine PAHs.

PAHs are formed and released into the environment as by-88 products of incomplete combustions through natural, e.g., by 89 90 forest fires, and man-made sources (Morillo et al. 2008). Some of them are highly toxic, present high persistence in 91the environment, low biodegradability and high lipophilicity. 9293 PAHs enter to wastewater-treatment plants through the sewerage and are almost completely removed from wastewater 94(up to 90 %), being concentrated in sludge because of their 95poor solubility in water and high adsorption capacity on solid 96 97 particles (Aparicio et al. 2009).

On the other hand, NP and NPE are used as nonionic 9899 surfactants in a large variety of industrial and domestic applications (Navarro et al. 2009). Although the use of 100 NPEs has been banned under Council Directive 2003/53/ 101 102EC (European Union 2003), their environmental presence 103 and risk is still high because of its historical and pervasive 104 widespread use. Industrial and urban wastewaters and agri-105cultural run-off discharge to WWTPs where the biodegradation of NPEs leads to the decrease in the length of ethoxy 106chain and an increase of hydrophobicity (Ifelebuegu 2011). 107 This is why NP is the main alkylphenol associated with 108 sewage sludge (90 %; Soares et al. 2008). Considering its 109endocrine disrupting properties, NP in sludge may impair 110111 environmental effects (Fernández-Sanjuán et al. 2009). The application of sewage sludge as soil fertiliser may produce 112effects to soil organisms or contaminate groundwater. To 113determine the content of PAHs and NP in sewage sludge will 114therefore be the first step to consider before their application 115to agricultural soils, in order to know the need of using 116technological treatments in the sludge management to reduce 117118 the concentration of these organic pollutants to levels that allow land application of sewage sludge according to the 119European Directive. 120

To assess the real hazard of these toxic chemicals in 121sewage sludge intended for agricultural land, it is therefore 122important to determine the available fraction. Traditionally, 123extraction techniques were aimed to determine the concen-124tration of the total organic contaminants in soils and sludge. 125The use of non-exhaustive extractions still gains importance 126to determine the bioavailable fraction of contaminants with 127the aim of assessing the real environmental risk (Alexander 1282000). This fraction needs to be reduced in sewage sludge to 129allow its land application. 130

Compounds with low water solubility and a high Kow 131coefficient show a high adsorption capacity and low avail-132ability to be desorbed from soils, sediments or sludge, de-133creasing the fraction that can really present a toxicological 134risk. Different extracting agents have been used to increase 135the solubility of organic contaminants but extracting only the 136really available fraction. Mild extraction procedures, such as 137extraction techniques with n-butanol or solvent mixtures 138(ethanol/water, methanol/ water, etc.), and extraction with 139surfactant agents like Triton X-100, have been used, but 140those molecules can represent also a health and environment 141hazard (Ying 2006). Other biodegradable complexing 142agents, such as cyclodextrins (CDs), which encapsulate 143poorly water soluble contaminants, have been used to en-144hance the water solubility of hydrophobic compounds and 145their removal from polluted sites, while minimising en-146vironmental impact (Petitgirard et al. 2009). CDs are 147polycyclic glucose oligosaccharides resulting from enzy-148matic degradation of starch by bacteria. CD molecules 149have a hydrophobic and non-polar cavity which permit 150the solubilisation of non-polar and low-polarity organic 151molecules of appropriate size and shape through the 152formation of water soluble inclusion complexes (Ginés 153et al. 1996; Villaverde et al. 2005a). 154

CDs in general are considered non-toxic, biodegradable 155molecules (Fenyvesi et al. 2005; Kiss et al. 2007; Wacker 156157has been reported to exhibit a low tendency to adsorb onto 158soil particles (Badr et al. 2004) and that it is well tolerated in 159humans (Gould and Scott 2005), being an alternative to the 160natural α -, β - and γ -cyclodextrins, with improved water 161 solubility and even more toxicologically benign. The chem-162ical properties of CDs combined with their non-toxic char-163acter to humans have led to their use in pharmaceuticals, as 164food additives, as well as in the environmental decontami-165nation procedures of wastewater (Olah et al. 1998) and soil 166(Fenyvesi et al. 2009; Hajdu et al. 2011; Villaverde et al. 1672012). Some CDs, in particular, have been shown to enhance 168 the solubility of several contaminants, such as PAHs, pesti-169cides or biphenyls (Villaverde et al. 2005b; Morillo et al. 1702012). In the case of hydroxypropyl- β -cyclodextrin 171(HPBCD), many studies have also demonstrated that a 172non-exhaustive aqueous extraction from soils was able to 173

at -2 °C until their characterisation for PAHs and NP concentration and all other experiments. 220

Analysis of PAHs content in sewage sludge

Sludge samples were spiked with the surrogate standards at a 223concentration of 2 mg kg⁻¹; 0.5 g of sludge were inserted in a 224glass tube with 10 mL hexane/dichloromethane (1:1, v/v) and 225placed in an ultrasonic bath for 10 min. The solution was 226 then centrifuged (10 min at 1,090×g; Beckman Coulter 227Avanti J-25I). The two last steps were repeated three times 228using fresh solvent. The sonicated extracts were evaporated 229in a Turbo Vap LV from Caliper Life Sciences to almost 230dryness (1 mL approximately) for further clean up. 231

The extracts were purified using solid-phase extrac-232tion (SPE) cartridges of neutral alumina of 5 g. The alu-233mina was solvated and conditioned with 40 mL hexane/ 234dichloromethane (1:1, v/v). Analyte elution was performed 235with 40 mL hexane–dichloromethane (1:1, v/v). The extract 236was concentrated to a volume of less than 1 mL by a Turbo 237Vap LV evaporator, transferred into an amber glass vial and 238reconstituted with hexane to a final volume of 1 mL. At this 239stage, anthracene-d₁₀ was added as an internal standard to give 240a concentration of 1 μ g mL⁻¹. Blanks were prepared at the 241same time under the same conditions. With these conditions, a 242good recovery was obtained for PAHs, but NP was poorly 243 recovered with alumina SPE. Therefore, a parallel extraction 244and clean-up was performed with florisil to enhance the 245recovery of NP. 246

Analysis of NP content in sewage sludge

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NP analysis was carried out according to the method 248proposed by Fernández-Sanjuán et al. (2009). Deuterated 249standard 4-n-NP-D8 was added to a sample aliquot of 0.1 g 250of the sieved sludge at 0.5 µg/g. Samples were homogenised 251and kept at room temperature overnight and subsequently 252extracted by sonication (10 min) with 5 mL of hexane/ 253dichloromethane (1:1, v/v) and repeated a second time. A third 254extraction was performed with 5 mL of dichloromethane/ 255acetone (1:1, v/v). After each extraction step, samples were 256centrifuged for 10 min and the extracts were combined and 257evaporated in a Turbo Vap LV to almost dryness. Extracts 258were subsequently cleaned up by SPE cartridges with 2595 g of florisil, conditioned with 20 mL of hexane/ 260dichloromethane (1:1, v/v) and 20 mL of dichloromethane/ 261acetone (1:1, v/v). The sample extract was eluted with 20 mL 262of hexane/dichloromethane (1:1, v/v) and 20 mL of 263dichloromethane/acetone (1:1, v/v). The eluent was evaporat-264ed to almost dryness and reconstituted with ethyl acetate to a 265final volume of 1 mL. NP was determined by gas chromatog-266 raphy-mass spectrometry (GC-MS) according to the method 267reported below. 268

174predict the bioaccessible fraction of different PAHs (Rhodes 175et al. 2010) or pesticides (Hartnik et al. 2008). In this study, three specific objectives were proposed: (1) 176177to analyse the content of 16 PAHs (included in the US 178Environmental Protection Agency (US EPA) priority pollutants list) and NP in six sewage sludge from WWTPs from 179180 NE Spain, (2) to carry out a non-exhaustive extraction of PAHs and NP from sewage sludge with HPBCD as a possi-181ble method for sludge decontamination and (3) to study the 182toxicological risk of the leachates obtained after HPBCD 183184extractions.

185 Experimental

186 Chemicals and reagents

Sixteen PAHs considered of primary environmental concern 187 according to the US EPA were analysed: naphthalene, 188 acenaphthylene, acenaphthene, fluorene, phenanthrene, an-189thracene, fluoranthene, pyrene, benzo[a]anthracene, chrys-190ene, benzo[b]fluoranthene, benzo[k]fluoranthene, benzo[a] 191 pyrene, indeno[1,2,3-cd]pyrene, dibenzo[a,h]anthracene and 192193 benzo[ghi]perylene. They were purchased from Supelco as mix solution of 200 mg L^{-1} in methanol. The surrogate 194 standard was a mixture containing naphthalene d-8, 195196 acenaphthene d-10, phenanthrene d-10, chrysene d-12 and pervlene d-12, purchased from Supelco as a solution of 197 200 mg L^{-1} in methanol. Anthracene d-10, from Supelco, 198 was also used as internal standard. 4-NP was supplied by Dr. 199 Ehrenstorfer (Cromlab, Spain). 2-HPBCD was supplied by 200Cyclolab (Budapest, Hungary). Solvents were supplied by 201202 Merck (Germany), alumina SPE cartridges of 5 g by the International Sorbent Technology (UK), florisil SPE from 203 Waters (USA) and nitrogen, for drying with 99.995 % of 204205purity, by Air Liquid (Spain).

206 Sample collection and preparation

Six sewage sludge samples were collected from different 207 WWTPs located in Northeast Spain and designated as A, 208B, C, D, E and F. Sludges A and B were from WWTPs 209210located in little urban areas and sludge was not digested. Sludges C-F were collected in WWTPs of industrial and 211highly urbanised areas and were anaerobic digested. One 212kilogramme of sample was collected from the sludge tanks 213214 using precleaned amber glass pots and were transported refrigerated to the laboratory. Sludge samples were placed 215in glass pots, frozen at -18 °C and lyophilised during 48 h 216217 $(10^{-2} \text{ mbar vacuum})$ in a freeze dryer (Lioalfa, Telstar, Spain). Afterwards, samples were sieved through 500 µm 218mesh to obtain a homogeneous material. Samples were kept 219

269 Instrumental analysis

Samples were analysed by a gas chromatographer (GC; 270271Agilent GC 6890N) coupled to a quadrupole mass spectrom-272eter (MS; Agilent MD 5975B). The system was operated in electron ionisation (70 eV). The separation was achieved 273with a 30×0.25-mm I.D. DB-5 MS column (J&W 274Scientific, Agilent Technologies) coated with 5 % phenyl-275methylpolysiloxane (film thickness, 0.25 µm). The oven 276temperature was programmed from 80 (holding time, 2771 min) to 175 °C at 6 °C/min (holding time, 4 min), to 278279235 °C at 3 °C/min and finally to 320 °C at 8 °C/min, keeping the final temperature for 5 min. The same GC-MS 280 program was used to determine PAHs and NP. Injection was 281performed in the splitless mode. Helium was the carrier gas 282(50 cm/s). Injector, transfer line and ion source temperatures 283 284were 280, 250 and 200 °C, respectively. For increased sen-285sitivity and specificity, quantification was performed in time-286scheduled selected ion monitoring using three ions per compound (Martínez et al. 2004). Internal standard quantifica-287tion was performed using the deuterated surrogate standards 288 present in each elution window. Peak detection and integra-289290 tion were carried out using Agilent ChemStation software.

Non-exhaustive extraction of PAHs and NP from sewagesludge using HPBCD solutions

Batch extraction experiments were carried out using the 293sludge E, and this same sludge spiked with the 16 EPA-294295PAHs (spiked E). Real and spiked sludge were treated with HPBCD solutions for its decontamination in comparison to 296297an aqueous extraction in the absence of HPBCD. For sludge spiking, 0.2 g of sludge E were inserted in a glass tube and 298spiked with 0.2 mL of 5 mg L^{-1} PAHs mix solution prepared 299 in acetone. During 1 h, the tube was shaken and vortexed for 300 10 s every 10 min. After complete acetone evaporation 301 302 (24 h), the tube was closed and extracted. The content of each PAH was increased in 5 mg kg $^{-1}$. 303

Batch extraction experiments Of the real and spiked sludge 304 305 E, 0.2 g was inserted in 25 mL glass centrifuge tubes (Corex) with 10 mL of an aqueous solution which contained 0.01 M 306 $Ca(NO_3)_2$ to maintain the ionic strength and 200 mg L⁻¹ 307 308 HgCl₂ to prevent the bacterial growth (named aqueous solution). In parallel, 0.2 g of sludge E (real and spiked) was 309 inserted in the Corex tubes with 10 mL of a solution con-310taining HPBCD (10 mM), Ca(NO₃)₂ (0.01 M) and HgCl₂ 311312 (200 mg L^{-1}) (named HPBCD solution). All batch extraction experiments were conducted in triplicate. The Corex tubes 313 were shaken on an orbital platform shaker at 200 rpm for 1 h 314315at a constant temperature of 20 °C. Blanks were prepared at 316 the same time under the same conditions. After centrifugation at 3,000 rpm for 10 min, target compounds were 317

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extracted from an aliquot of the aqueous supernatant with 318 hexane, and 1 mL of this organic phase was transferred into 319 amber glass vials. The rest of the supernatant was decanted 320 and then 10 mL of fresh aqueous or HPBCD solution were 321 added. The extractions were carried out three times in total, 322 and each extract was analysed individually. One millilitre of 323 the hexane extract was dosed in a vial, and at this stage, 324 anthracene-d₁₀ was added to a concentration of 1 μ g mL⁻¹ 325and analysed by GC-MS as previously described. 326 327

Toxicity evaluation of leachates

In order to study the toxicity of the leachates obtained after the 329 extraction of PAHs and NP from sewage sludge using aqueous 330 and HPBCD solutions, parallel extraction experiments were 331 332 performed but without Ca(NO₃)₂ and HgCl₂ since these compounds would affect the condition of Daphnia. The leachates 333 used in these experiments were obtained as described: 0.5 g of 334sludge E were inserted in 100 mL amber glass bottles with 335 25 mL of ASTM hard water (ASTM 1999). In parallel, 0.5 g 336 of sludge E were inserted in the 100 mL amber glass bottles 337 with 25 mL of 10 mM HPBCD solution prepared in hard 338 water. The suspensions were shaken for 1 h at a constant 339 temperature of 20 °C. After separation by decantation, the 340 leachates obtained were kept at 4 °C. 341

Bioassays with the grazer Daphnia magna were 342 conducted to evaluate sub-lethal feeding effects of filtered 343 fractions of leachates. Feeding responses to sludge leachates 344diluted 1 and 10 % in ASTM hard water in 24 h toxicity tests 345 were performed following Barata et al. (2008). Groups of 346 five neonates were exposed to 20 mL of test concentrations 347 in 30 mL borosilicate flasks in the presence of food. The alga 348 Chlorella vulgaris was added at a concentration of 5×10^5 349 cells/mL⁻¹ (equivalent to 1.5 µg C mL⁻¹). Treatments 350 consisted of an ASTM hard water control (W), 10 mM 351HPBCD solution controls and the selected leachates with 352 five replicates each. Each group of replicates consisted of 353 five vessels with animals and one blank. Blanks were used to 354assure that initial algal concentrations did not increase sig-355nificantly over the exposure period. Individual feeding rates 356(number of algal cells ingested per animal per hour) were 357determined as the change in cell density during 24 h 358according to the method described by Barata et al. (2008) 359and converted to proportional feeding rates relative to ASTM 360 W. Cell density was estimated from absorbance measure-361ments at λ =650 nm using standard calibration curves based 362 on at least 20 data points ($r^2 > 0.98$). Proportional responses 363 were arcsine transformed to meet ANOVA assumptions of 364 normality and variance homocedasticity (Zar 1996). 365 Differences in bioassay responses among treatments were 366 compared by one-way ANOVA following post hoc Tukey's 367 test at a p < 0.05 significance level. 368

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369 Results and discussion

370 Content of PAHs and NP in sewage sludge

371 PAHs

372 Table 1 shows the concentration of PAHs and NP in studied sludges. Sludge F had the highest concentrations of total 373 PAHs (3.05 mg kg⁻¹ dry weight (dw)) while sludge E had 374 the lowest (1.88 mg kg⁻¹dw). Among individual PAHs, the 375highest contribution was that of phenanthrene (15-30%) and 376 377 pyrene (10-24 %) followed by naphthalene (14-23 %). Anthracene and fluoranthene were found in fewer amounts 378 (9-21 and 9-13 %, respectively). Normally, when phenan-379 threne and pyrene are present at high levels in the sludges, it 380 indicates a domestic sewage water origin (Miégle et al. 2003; 381 382 Villar et al. 2006).

The limit concentration proposed for \sum_{PAHs} in EC 383 384Working Document on Sludge (European Union 2000) is 6 mg kg⁻¹dw indicating that none of the sludges exceeded 385this maximum level. Σ_{PAHs} was relatively low when com-386 pared with other studies. Dai et al. (2007) found concentra-387 tions from 2.5 to 26 mg kg⁻¹ of 16 PAHs in six sewage 388 sludge from a WWTP receiving steel and dyeing industry 389 wastewater from Beijing. Khadhar et al. (2010) found \sum_{PAHs} 390 levels from 0.1 to 7.8 mg kg^{-1} in sewage sludge samples 391from nine Tunisian WWTPs. Oleszczku (2009) found a total 392

content from 2.83 to 9.95 mg kg⁻¹ with pyrene and fluoranthene being the most abundant PAHs in the four sewage 394 sludges studied. Blanchard et al. (2004) found that sludges 395 from high traffic areas displayed high levels of pyrene, 396 fluoranthene and phenanthrene, contrasting with those from 397 residential zones which showed a more homogeneous PAHs 398 distribution. 399

In Spain, Abad et al. (2005) analysed 139 samples from 400 Catalonia and found a median concentration of 1.4 mg kg^{-1} , 401 but less than 3 % of the samples had PAHs concentrations 402 higher than the European Union-proposed limit. Pérez et al. 403 (2001) determinate the PAH content in six sewage sludge 404 samples and found that phenanthrene and pyrene were the 405most abundant PAHs. In 17 WWTP sludge from the 406 Valencian Community, Gómez-Rico et al. (2007) found that 407 only one sample exceeded the limit of 6 mg kg⁻¹ PAHs and 408 that phenanthrene was the major PAH followed by fluorene 409 and pyrene. 410

The different PAH distribution found among the sludge 411 studied indicated different sources. Table 2 shows that main-412 ly three-ring PAHs predominated (from 30 % in sludge B to 41354 % in sludge F). Four-ring PAHs were predominant (38 %) 414 only in sludge B. Shen et al. (2007) analysed 44 sludge 415 samples from WWTPs which received river waters, munic-416 ipal sewage or mixtures of industrial and domestic sewage in 417 the Yangtze River delta area and observed that the content of 418 two- and three-ring PAHs was much higher than of four-, 419

t1.1 **Table 1** Recoveries of nonylphenol and PAHs obtained from six sewage sludges (in milligrammes per kilogramme) and percentages of individual PAHs related to \sum_{PAHs}

Sludge	Sludge		А		В		С		D		Е		F	
		mg kg ⁻¹	%	mg kg ⁻¹	%	mg kg^{-1}	%	mg kg^{-1}	%	mg kg ⁻¹	%	mg kg ⁻¹	%	
Nonylphen	ol	44		4.5	_	72	_	199	_	72	_	53	_	
Naphtalene	;	0.55	23.51	0.27	13.59	0.37	13.65	0.30	14.34	0.40	21.32	0.47	15.26	
Acenaphty	lene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Acenaphter	ne	0.00	0.00	0.03	1.56	0.07	2.65	0.09	4.20	0.00	0.09	0.06	2.07	
Fluorene		0.00	0.00	0.08	3.74	0.18	6.55	0.22	10.28	0.00	0.20	0.11	3.48	
Phenanthre	ne	0.51	21.90	0.30	14.98	0.82	30.45	0.57	26.97	0.46	24.48	0.83	27.15	
Anthracene	•	0.36	15.28	0.20	9.90	0.25	9.19	0.23	11.15	0.24	12.87	0.65	21.19	
Fluoranthe	ne	0.29	12.50	0.22	10.77	0.24	8.82	0.24	11.38	0.20	10.64	0.37	12.21	
Pyrene		0.57	24.39	0.32	15.92	0.44	16.36	0.42	19.85	0.37	19.79	0.30	9.71	
Benzo(a)ar	thracene	0.00	0.00	0.11	5.55	0.06	2.20	0.04	1.85	0.04	2.18	0.07	2.31	
Chrysene		0.00	0.00	0.11	5.34	0.00	0.00	0.00	0.00	0.00	0.00	0.04	1.27	
Benzo(b)fl	uoranthene	0.00	0.00	0.23	11.15	0.10	3.57	0.00	0.00	0.07	3.61	0.03	1.13	
Benzo(k)fl	uoranthene	0.00	0.00	0.05	2.49	0.18	6.56	0.00	0.00	0.10	5.12	0.13	4.23	
Benzo(a)py	vrene	0.06	2.43	0.10	5.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Indeno(1,2	,3-cd)pyrene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Dibenzo(a,	h)anthracene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Benzo(g,h,	i)perylene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
\sum_{PAHs}		2.35	100	2.02	100	2.68	100	2.10	100	1.88	100	3.05	100	

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Sludge	А	В	С	D	Е	F
2-rings PA	Hs 23.51	13.59	13.65	14.34	21.32	15.2
3-rings PA	Hs 37.17	30.18	48.84	52.59	37.35	53.89
4-rings PA	Hs 36.89	37.58	27.38	33.08	32.60	25.4
5-rings PA	Hs 2.43	18.65	10.13	0.00	8.73	5.3
6-rings PA	Hs 0.00	0.00	0.00	0.00	0.00	0.00

five-, and sing-ring PAHs. They concluded that the source of
PAHs with ≥4 benzene rings is usually from combustion at
very high temperature, whereas PAHs with two and three
benzene rings originate from contamination of petroleum or
petroleum products. Studies carried out by Oleszczku (2007,
2008) show that mainly four-ring PAHs predominated in five
sewage sludges from Poland.

427 Nonylphenol

428 The concentration level of NP was highly variable in each WWTP, from 4.5 to 199 mg kg⁻¹ (Table 1). The highest 429concentrations of NP were found in sludge D followed by E 430431and C. The differences in NP content can be explained by the 432 high variability of influent wastewater composition and by the wide range of variables involved on each sludge treatment 433434 processes (González et al. 2010). Sludges C, D, E and F, which showed the highest NP content, are digested sludges, 435and it is known that during anaerobic digestion, the degrada-436tion of NPE to NP occurs, but NP persists (Minamiyama et al. 437 2006; Céspedes et al. 2008). The loss of weight achieved 438 during the sludge dehydration process can also explain the 439high NP levels found in such sludge. The concentrations were 440lower in raw sludge (sludges A and B) than in digested, as 441 observed also by Lian et al. (2009). Aparicio et al. (2009) 442443found a higher NP level in anaerobically digested dehydrated sludge than in primary and secondary sludge and explained it 444by NP1EO and NP2EO degradation to NP. 445

446 The European Union (2000) limits concentration proposed in the third draft of the Working Document on Sludge for NP 447 and NPE for land application of sewage sludge was 50 mg-448 kg^{-1} . Table 1 shows that by only considering the NP content, 449 four of the six sludges analysed exceed this maximum value, 450and, therefore, these sludges would not be apt to be used as 451fertilisers, and NP should be decreased for their further use in 452agriculture in order to preserve the environmental quality, 453considering the inherent toxicity of this contaminant. 454

Non-exhaustive PAHs and NP extraction using HPBCDsolutions

A non-exhaustive extraction of PAHs and NP from sewagesludge with HPBCD solutions has been used as a possible

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method for decontamination prior the sludge application to 459 agricultural land. 460

Figure 1 shows the percentages of individual PAHs extracted 462 after three sequential extractions by the aqueous and HPBCD 463 solutions from sludge E (Fig. 1a, b, respectively). The 464amount of PAHs removed by HPBCD extraction (Fig. 1b) 465 was much higher than by aqueous extraction in its absence 466 467 (Fig. 1a), although the percentages of PAHs removed varied depending on the individual PAH. When the aqueous solu-468 tion was used, only naphthalene and phenanthrene were 469removed, their total extraction percentages were in general 470 low (13 and 1.3 %, respectively), with little differences 471within the three extractions. Four- and five-ring PAHs were 472 not extracted. HPBCD was more efficient as an extraction 473technique for PAHs (Fig. 1b). Not only were naphthalene 474 and phenanthrene extracted in much higher percentages (33 475 and 15 %, respectively) but also anthracene (8 %) and some 476four-ring PAHs as well, such as fluoranthene and pyrene 477(although with extraction percentages lower than 2 %). 478

In order to confirm if only the type of PAH, and not their 479 amounts present, or the aging process influences on their 480individual availability, the sludge E was spiked with 16 EPA-481 PAHs and the extraction with the aqueous and HPBCD 482 solutions were also carried out on spiked E (Fig. 1c, d). 483 When the aqueous solution was used (Fig. 1c), two- and 484three-ring PAHs (naphtalene, acenaphtylene, acenaphtene, 485fluorene, phenanthrene and anthracene) were removed after 486 three sequential extractions but at different percentages, (57, 487 10, 11, 5, 2 and 0.6 %, respectively). The results achieved 488 were very different when HPBCD solution was applied 489 (Fig. 1d). Naphthalene, acenaphtene and fluorene reached 490 the highest extraction percentages (89, 39 and 48 %, respec-491tively), but fluorene, phenanthrene and anthracene were the 492PAHs with the highest increase in extraction percentages in 493comparison to the aqueous solution: from 5, 2 and 0.6 % to 49448, 32 and 36 % (that is 10, 16 and 60 times higher, respec-495tively). Four-ring PAHs, such as fluoranthene, pyrene, 496 benzo[a]anthracene and chrysene, which were not removed 497 by the aqueous solution, showed total extraction percentages 498about 1-3 % with HPBCD solution. Five- and six-ring PAHs 499were not removed at all. 500

It is worth to emphasise the importance of HPBCD first 501extraction. The extraction percentages obtained for naphtha-502lene, acenaphtylene and acenaphtene (47, 12 and 13 %, 503respectively) were similar or slightly higher than those 504obtained after the three sequential extractions using the 505aqueous solution (57, 10 and 11 %, respectively). In the case 506 of fluorene, phenanthrene and anthracene, these differences 507 were still higher: the extraction percentages obtained after 508the first extraction by HPBCD solution (19, 12 and 13 %, 509

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Fig. 1 Percentage of individual PAHs extracted after three sequential extractions with the aqueous and HPBCD solutions from sludge E (a, b, respectively) and from spiked sludge E (c, d, respectively)

510respectively) were much higher than the total extraction percentages obtained after three extractions with the aqueous 511solution (5.30, 1.62 and 0.56 %, respectively). In the case of 512four-ring PAHs (fluoranthene, pyrene, benzo[a]anthracene 513and chrysene), which were not removed by the aqueous 514solution, only with the first HPBCD extraction was it possi-515516ble to remove 50 % of the total amount extracted after the three sequential extractions. This behaviour was also ob-517served in the extraction of sludge E without spiking, 518commented previously. The first HPBCD extraction step 519importance has been observed by other authors (Villaverde 520 et al. 2005b). 521

522It is well known that PAHs present a low aqueous solubility. The higher the number of benzene rings, the higher 523their hydrophobicity and the lower their water solubility. For 524525this reason, the use of water as an extractant agent in PAHs removal from contaminated sludge is relatively ineffective. 526The increase observed in PAHs extraction percentages after 527528HPBCD in spiked and non-spiked sludge E is related to the fact that HPBCD molecules are able to form inclusion com-529plexes enhancing the water solubility of organic compounds, 530531such as PAHs, and therefore their extraction from polluted soils or sludge (Villaverde et al. 2005b; Gómez et al. 2010). The 532extraction percentages obtained were different for the PAHs 533studied depending on the appropriate size and shape of the 534535PAH with respect to HPBCD hydrophobic cavity dimensions. Although the extraction of naphthalene shows the highest 536

536 Although the extraction of naphthalene shows the highest 537 percentage obtained (Fig. 1c, d), the increase with HPBCD in relation to the aqueous solution was not very high (only 1.5 538times higher). On the contrary, phenanthrene and anthracene 539were the PAHs which showed the highest increase in extrac-540tion percentages (from 2 to 33 % and from 0.6 to 36 %, 541respectively). This is related with the appropriate size and 542shape of the phenanthrene molecule (0.58 nm width and 5430.78 nm length) with respect to HPBCD hydrophobic cavity 544dimensions (0.6 nm of diameter and 0.79 nm of depth). As a 545consequence, phenanthrene showed a best molecular accom-546 modation in the hydrophobic cavity of HPBCD molecules 547than other PAHs. Anthracene also showed a high increase in 548extraction percentage with HPBCD solution. Its molecule 549presents a higher length and a lower width than phenanthrene 550(0.5 nm width and 0.92 nm length) that permits its inclusion 551in the cavity of HPBCD, but the distance between the mol-552ecule of anthracene and the active points of HPBCD is not so 553close as in the case of phenanthrene. 554

The extremely lower extraction efficiency observed in the 555case of 4-rings PAHs (fluoranthene, pyrene, benzo[a]anthra-556cene and chrysene) may be attributable in part to a lower 557availability of the contaminants (which present a lower water 558solubility) for being transferred to the aqueous phase due to 559the high tendency to remain adsorbed on the hydrophobic 560surfaces of the sludge. Another influencing factor may be 561that they present a width higher than that of the HPBCD 562cavity; for this reason, only a small part of the molecules of 563 these compounds are included into the HPBCD cavity when 564the inclusion complex is formed, and a great part remains in 565

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contact with water, reducing their extraction percentage dueto stearic restraints (Semple et al. 2007).

The sequence of the extraction percentages obtained for 568 569six of the PAHs extracted (acenaphtene, fluorene, phenan-570 threne, anthracene, fluoranthene and pyrene) is in agreement with the sequence of solubility efficiency (Se) and the ap-571572parent formation constants (Kc) obtained by Morillo et al. (2012) who evaluated the influence of the three different 573CDs (including HPBCD) on the solubility of these PAHs 574selected among those 16 proposed by EPA as the most 575frequently occurring and/or dangerous in environmental 576577 samples. It was found that phenanthrene showed the highest Se and Kc values when they formed inclusion complexes 578with HPBCD, followed in decreasing order by anthracene, 579 fluorene, fluoranthene, acenaphthene and pyrene. They con-580cluded that these results were related with the molecular 581volume of the PAH and the dimensions of CD cavity, and 582583as consequence, the points of interaction between the PAHs 584and the CDs.

In spite of the different individual PAHs content of sludge 585E in comparison with spiked sludge E, the tendency in the 586extraction by HPBCD is the same: the two- and three-ring 587 588 PAHs are much favoured than those of the four-ring compounds. Taking into account that sludge from WWTPs con-589tain also fluoranthene and pyrene (four-ring PAHs), it would 590591be necessary to use a mix of HPBCD with another cyclodextrin with a high cavity dimensions, such as γ -CD or their 592derivates, in order to extract effectively such compounds. On 593594the other hand, the ageing of sludge E in comparison with spiked sludge E is another important factor. The PAHs pres-595ent in the aged sludge have become sequestered within the 596597structure of the sludge, in particular in the organic matter fraction, and therefore, their extractability is expected to be 598lower. Many research studies have shown that when the soil-599 pollutant contact time increases the biological and chemical 600 601 availability of contaminants decrease (Villaverde 2007; McAllister and Semple 2010). That is probably the reason 602 603 of the lower percentages extracted by HPBCD from sludge E in comparison to spiked sludge E: 33, 15 and 8 % instead of 604 89, 32 and 36 %, for naphthalene, phenanthrene and anthra-605 cene, respectively. However, in the case of 4-rings PAHs 606 the percentages obtained were very similar and very low 607 (< 3 %), indicating that HPBCD has a lower tendency to 608 609 extract such PAHs.

The bioavailable fraction of contaminants in soils sedi-610 ments and sludge (and not their total content) is the main 611 factor determining their fate in the environment and their 612 influence on living organisms. For this reason, several chem-613 ical techniques capable of reliably predicting the bioavail-614 ability of organic contaminants to biota have been studied. 615 616 Oleszczuk (2008, 2009) used three different extractants, Tenax-TA, n-butanol or HPBCD, to determine the content 617 of the bioavailable fraction of PAHs in sewage sludge, 618

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observing a significant positive relationship between Tenax-619 TA-extracted fraction and growth inhibition of Heterocypris 620 incongruens, whereas when using n-butanol or HPBCD, the 621 content of the extracted fraction was usually twice higher. 622 That is, higher amounts of PAHs were extracted with 623 HPBCD than those actually bioavailable. Then the extraction 624 of PAHs reached in this paper using HPBCD was more than 625 enough to obtain a decontaminated sludge since the rest of 626 PAHs not extracted are not bioavailable and, therefore, not 627 potentially toxic to the environment. 628

Nonylphenol

The total content of NP in sludge E was 72 mg kg⁻¹ 630 (Table 1), higher than the limit value proposed for the sum 631 of all the NP ethoxylates compounds by the Draft Document 632 of the European Union which was 50 mg kg⁻¹. This means 633 that before this sewage sludge can be used as fertiliser for 634 agricultural land, NP content has to be reduced. 635

The extraction percentage of NP from sludge E by the 636 aqueous solution after the three sequential extractions was 637 lower than 1 % (Fig. 2a), with no differences between the 638 amount of NP extracted after each of the three sequential 639 extractions. However, when HPBCD solution was used as 640 extraction technique, 105 ± 14 % of the NP was extracted. It 641 is important to emphasise that with only the first extraction, 642 89 % of NP was removed and the other two extractions 643 increased until 100 %. Hence, only one extraction, using 644 HPBCD solution, would be sufficient to reduce NP content 645 in a highly contaminated sludge under the limit value pro-646 posed by UE. 647



Fig. 2 Percentage of NP extracted after three sequential extractions by the aqueous (a) and HPBCD (b) solutions from sludge E

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648 The results obtained in the present study are in agreement with the research studies by Kawasaki et al. (2001) since 649 they showed that NP solubility (aqueous solubility 650 5.43 mg L^{-1}) increased more than 6.000 times when 651HPBCD was used in a particular range of concentrations 652 (0-60 mM). This increase is due to the fact that NP mole-653 654 cules, which have a smaller size than HPBCD cavity, can be included within such hydrophobic cavity and form inclusion 655 complexes. As result, NP hydrophilicity and water solubility 656 increase, and therefore its removal from contaminated sites 657 occurs. This is the reason why the extraction of the sludge 658 659 using HPBCD gives as such extremely high recovery result of NP in the leachates, and, therefore, a low final content of 660 NP is reached in the treated sludge. 661

662 Toxicity of leachates

The removal of PAHs and NP from sludge by HPBCD could 663 664 be a good alternative to existing sludge treatment methods which are basically aimed to reduce the water content. The 665 decontamination method we proposed in this study removes 666 the contamination load, especially for NP, and sludge may 667 668 easily obey with the limit values proposed by the European Union. In this sense, the procedure could be of great impor-669 tance for improving the sludge disposal alternatives in terms 670 671 of environmental safety and economic costs. It is also important to determine whether the leachates containing the 672 contaminants possess environmental hazard. Therefore, to 673 674 further test the potential environmental hazards of CDs leachates to aquatic organisms, the toxicity of sludge 675 HPBCD and water leachates were assessed using D. magna 676 677 feeding responses (Fig. 3), a model organism to test aquatic toxicity (Barata et al. 2008). 678

In all tests, Daphnia mortality was absent. HPBCD dilut-679 ed 1 and 10 % in ASTM water did not affect significantly 680 (P < 0.05) proportional feeding rates relative to controls 681 (Fig. 3). This result confirms that HPBCD does not impair 682 683 any effect on Daphnia. Sludge HPBCD and water leachates diluted 1 and 10 % in ASTM hard water inhibited signifi-684 cantly (P<0.05) Daphnia feeding rates relative to their re-685 686 spective controls (Fig. 3). It is important to emphasise that both, water and HPBCD sludge leachates had similar toxic-687 ity, despite the different amount of contaminants present in 688 689 each of them. Water leachates diluted at 10 % had higher toxicity than at 1 %, as expected. 690

According to Figs. 1 and 2 and Table 1, water leachates 691 diluted 1 and 10 % should contain 0.04–0.39 μ g L⁻¹ NP, and 692 those of HPBCD, 13 and 130 μ g L⁻¹. Estimated levels of 693 PAHs in both water and cyclodextrin leachates were low 694 $(0.17, 0.10, 0.57 \text{ and } 0.13 \text{ } \mu\text{g L}^{-1} \text{ for } 10 \% \text{ HPBCD leachates}$ 695 696 of pyrene, fluoranthene, phenanthrene or naphthalene, respectively). The combined presence of these compounds in 697 water and HPBCD leachates, however, caused similar effects 698



Fig. 3 Feeding responses of *D. magna* juveniles exposed to pure ASTM hard water (*W*) and HPBCD (*HPBCD*) and their sludge leachates (sludges W and HPBCD) diluted 1 and 10 % in ASTM hard water. Results are expressed as proportional responses (mean \pm SE) relative to W controls. *Different letters* indicate significant (*P*<0.05) differences across treatments after one-way ANOVA and post hoc Tukey's multiple comparison tests

on Daphnia feeding rates, as evidenced in Fig. 3. Previous 699 studies indicate that effects of naphthalene in Daphnia occur 700 only above a milligramme-per-litre levels, pyrene and fluo-701 ranthene above 10 μg L⁻¹ and phenanthrene above 702 40 μ g L⁻¹ (Olmstead and LeBlanc 2005). For NP, however, 703 sublethal detrimental effects on feeding and reproduction 704 occur at concentrations greater than $80 \ \mu g \ L^{-1}$ (Sun and 705 Gu 2005; Campos et al. 2012), being lethal above 300-706 400 μ g L⁻¹ (Comber et al. 1993). Accordingly, NP residues 707 present in HPBCD sludge leachates should impair Daphnia 708 feeding rates, whereas those measured in water sludge leach-709 ates did not. The fact that we found similar toxic effects for 710 both water and HPBCD leachates means that NP residues 711present in HPBCD leachates were not bioavailable to 712Daphnia. Indeed, it is likely that HPBCD encapsulated NP 713making it less bio-available to Daphnia. Such ameliorated 714 toxic effects of NP have been described for humic substances 715(Sun and Gu 2005; Burgess et al. 2005). Our results indicate 716further that observed toxicity was due to water soluble com-717 pounds not removed by HPBCD. However, although the 718leachates produced have been demonstrated not toxic due 719 to the contaminants extracted by HPBCD, they should be 720further treated within the WWTP in order to avoid the 721dispersion of such contaminants in the environment. For this 722 reason, HPBCD should be considered as a useful tool in 723 sewage sludge decontamination treatments to reduce for 724 example compounds such as NP and generate non toxic 725leachates. In this way, sludge highly contaminated with NP 726

727 or PAHs could be treated with HPBCD generating a728 contaminant-free sludge with added value, which may am-729 plify its use in agronomical practices.

730 Conclusions

731 Given the more restricted concentration limit values for some 732 organic compounds in sewage sludge included in the draft of the European Union regulations, a preliminary study using 733 734 HPBCD extractions has been proposed as a method for the decontamination of sludge. Among other contaminants, 735736 PAHs and NP are toxic compounds that are present in high concentrations in sludge and are prioritised by the European 737 Union since their presence can have serious environmental 738 implications because of their application to agricultural soils 739 at doses that may produce long-term contamination of soil 740 741 and groundwater. In this study, it has been demonstrated that these compounds are able to form inclusion complexes with 742 HPBCD and increase their solubility. It has been shown that 743 744 the application of HPBCD solutions to sludge can enhance the contaminants desorption and decontamination, at ex-745 penses of generating a leachate containing the contaminants 746released. We propose the use of HPBCD as a very effective 747 748 tool for the remediation of contaminated sewage sludge, increasing its safe usage for agricultural practices without 749 posing environmental or health risks. We also suggest that 750the leachate, although demonstrated not toxic because of 751those contaminants extracted by HPBCD, such as NP or 752PAHs, should not be released to receiving waters but rather 753be treated in the WWTP to avoid their dispersion in the 754755environment. Aspects related to the optimisation of HPBCD extractions for the sludge decontamination of 756 multiclass contaminants present in the WWTPs and evalua-757 tion of leachates in terms of contamination loads, toxicity 758 759 and possible decontamination treatments will be deeply investigated in further research. 760

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