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4 **Influence of pine or oak wood on the degradation of alachlor and**
5 **metalaxyl in soil**

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15 **Abstract**

16 The objective of this work was to study the influence pine or oak wood added to
17 soil as an amendment (5% w/w) had on the degradation rate of two pesticides, alachlor
18 and metalaxyl, with different hydrophobic character. The formation of pesticide
19 metabolites and the soil dehydrogenase activity in non-amended and amended soil
20 samples were also monitored. The degradation of metalaxyl followed first-order
21 kinetics, while the degradation of alachlor followed first-order or biphasic kinetics in
22 the soil samples studied. The results indicated that the degradation rate was slower for
23 metalaxyl than for alachlor, and for both pesticides followed the order: pine amended
24 soil < oak amended soil < non-amended soil. The faster degradation rate in non-
25 amended soil was attributed to the higher sorption of pesticides by wood amended soils.
26 The alachlor ethane sulfonic acid (ESA), and two metalaxyl metabolites (2-[(2,6-
27 dimethylphenyl)-methoxyacetylamino]-propionic acid and N-(2,6-dimethyl-phenyl)-2-
28 methoxy-acetamide) were detected during the incubation period. Soil dehydrogenase
29 activity recorded close values in non-amended and amended soil treated with alachlor,
30 but it was higher in wood amended soil treated with metalaxyl. Pine and oak wood
31 increase the immobilization of the pesticides studied, but they also limit their
32 bioavailability in soil by decreasing their degradation rate in amended soil.

33 **Keywords:** pesticide, metabolite, degradation, soil, wood residue, remediation.

34

35 **Abbreviations:** S, soil; SS, sterilised soil; S+P, soil amended with pine wood; S+O, soil
36 amended with oak wood.

37 **1. Introduction**

38 Alachlor is a herbicide used in pre-emergence to control perennial grasses and
39 many broad-leaved weeds in corn and other crops. Given its widespread use, alachlor
40 has been found in soils and in both surface water and groundwater (Sánchez-Camazano
41 *et al.*, 2005; Vryzas *et al.*, 2009). Soil organic matter (OM) is the major factor for
42 alachlor adsorption, and it is sorbed to a lesser extent by clay colloids (Guo *et al.*,
43 1993). Biodegradation is the single most significant mechanism for controlling the
44 dissipation of alachlor in agricultural soils. Biological degradation by dechlorination in
45 soil leads to the formation of the ethane sulfonic acid (ESA) metabolite (Stamper *et al.*,
46 1998). Alachlor ESA is a more polar compound than alachlor and has been found in
47 groundwater and in surface water more often and in higher concentrations than the
48 parent herbicide.

49 Metalaxyl is a systemic acylanilide fungicide widely applied in different crops to
50 eliminate different fungal species. Metalaxyl is highly soluble in water and has a low
51 hydrophobicity, indicating low adsorption by soils and the possibility of leaching into
52 groundwater. Metalaxyl has recently been found in groundwater at concentrations of up
53 to $0.49 \mu\text{g L}^{-1}$, which exceeds the $0.1 \mu\text{g L}^{-1}$ EU limit (Hildebrandt *et al.*, 2008). Several
54 authors have indicated the importance of OM and clay content in the adsorption of
55 metalaxyl by soils (Andrades *et al.*, 2001). Its degradation in soil has been reported
56 mainly as biodegradation (Sukul and Spiteller, 2000). Metalaxyl is degraded in soil by
57 cleavage of the methyl ester group, forming the main acid metabolite (CGA-62826),
58 although biodegradation of metalaxyl could occur by benzylic hydroxylation of the
59 methyl chain or aromatic hydroxylation. A second metabolite (CGA 67868) is formed
60 either directly from metalaxyl or from the metabolite CGA 62826 by N dealkylation.
61 (Pesaro *et al.*, 2004).

62 Recent work has shown that the water pollution caused by pesticides from point
63 sources (spills, uncontrolled disposal, equipment washing water, etc.) can be more
64 serious than that due to agricultural practice (De Wilde *et al.*, 2007; Fait *et al.*, 2007).
65 The use of organic materials or wastes may prevent the mobility of pesticides from
66 these point sources of contamination and enhance their biodegradation (Rodríguez-Cruz
67 *et al.*, 2007a). In recent years, different low-cost adsorbent systems (biobed,
68 biomassbed, biofilter, etc.) have been developed to minimize point sources of pesticide
69 pollution by retaining and degrading pesticides (Castillo *et al.*, 2008; De Wilde *et al.*,
70 2007; Fait *et al.*, 2007). In particular, the use of wood residues as low-cost adsorbents
71 has recently been developed as a new technology for the immobilization of heavy
72 metals, dyes, pesticides, other organic compounds, etc., in soil (Gupta *et al.*, 2009;
73 Rodríguez-Cruz *et al.*, 2007b; Shukla *et al.*, 2002). In a recent study, Rodríguez-Cruz *et*
74 *al.* (2007b) found that oak and pine wood can be effectively used as adsorbents of
75 pesticides. The Freundlich constants (Kf) for alachlor and metalaxyl adsorption by oak
76 and pine were related to the lignin content of these woods. However, the influence the
77 addition of wood residues had on the degradation of pesticides in soils has been less
78 studied (Grenni *et al.*, 2009). The addition of an organic amendment to the soil affects
79 the biodegradation of pesticides because the OM and nutrients added can strongly affect
80 the structure and activity of bacterial and fungal populations as a result of the increased
81 metabolism of the readily available nutrients (Briceño *et al.*, 2007). Some organic
82 amendments may stimulate biodegradation, but others can reduce it (Moorman *et al.*,
83 2001; Rodríguez-Cruz and Lacorte, 2005).

84 The main objective of this work was to investigate the effects pine and oak wood
85 added to soil as amendments had on the rates of degradation of alachlor and metalaxyl.
86 These pesticides have different hydrophobic characters and water solubility and are

87 widely used in agriculture. Soil dehydrogenase activity was monitored in soil treated
88 withalachlor or metalaxyl and amended with either pine or oak residues or non-
89 amended in order to analyze the effect of the amendment and the pesticide on the
90 microbial activity in the soil. The findings of this study provide information regarding
91 the use of wood amendments for preventing soil and water contamination by pesticides
92 with different characteristics.

93

94 **2. Materials and methods**

95

96 *2.1. Soil and wood samples*

97 Soil samples were collected from the surface layer (0-15 cm depth) of an
98 agricultural field located in Aldearrubia (Salamanca, Spain). The soil was left overnight
99 at room temperature to reduce moisture content and then sieved (< 2 mm). The soil was
100 sandy-loam (11.8% clay, 13.6% silt and 74.5% sand), with 0.85% organic matter
101 content, a pH of 6.3, and a cation exchange capacity of 4.8 cmol kg⁻¹ (Rodríguez-Cruz
102 *et al.*, 2007a).

103 Pine and oak sawdust (< 1 mm) were obtained from a local industry in
104 Salamanca (Spain). They had different lignin contents: 18.2% for oak and 24.4% for
105 pine (Rodríguez -Cruz *et al.*, 2007b).

106 The amended soils were prepared by uniformly mixing soil with oak or pine
107 sawdust (5% w/w), similarly to other organic residues (Moorman *et al.*, 2001). Sub-
108 samples were analyzed to assess both the total organic carbon (TOC) and the soluble
109 carbon contents, as described elsewhere (Rodríguez-Cruz *et al.*, 2007b). The TOC
110 content in the soil amended with pine (S+P) or oak (S+O) was 2.89% and 2.79%,
111 respectively, and about fourfold greater than in non-amended soil (S) (0.72%).

112 Moreover, the soluble carbon content was higher in S+P (0.047%) and S+O (0.037%)
113 than in S (0.008%). Sample pH varied between 5.9 (S+O) and 6.6 (S+P).

114

115 *2.2. Chemicals*

116 Alachlor (99.5% purity) was supplied by Chem Service (West Chester, USA). It
117 is a herbicide with a water solubility of $240 \mu\text{g mL}^{-1}$ (pH 7, 20°C) and a log K_{ow} of 2.63.
118 The alachlor ESA was supplied by Monsanto Chemical Co. (St. Louis, MO, USA).
119 Metalaxyl (>98% purity) was supplied by Novartis Crop Protection AG (Basel,
120 Switzerland). It is a fungicide with a water solubility of $8400 \mu\text{g mL}^{-1}$ (22°C) and a log
121 K_{ow} of 1.75 (Tomlin, 2003). The two metalaxyl metabolites studied (> 99% purity), 2-
122 [(2,6-dimethylphenyl)-methoxyacetyl-amino]-propionic acid (CGA62826) and N-(2,6-
123 dimethylphenyl)-2-methoxy-acetamide (CGA92370), were supplied by Syngenta Crop
124 Protection AG (Münchwilen, Switzerland).

125

126 *2.3. Degradation experiments with amended and non-amended soil*

127 The pesticide degradation experiment was conducted in duplicate in accordance
128 with SETAC guidelines (Lynch, 1995). The standard compound was added to soil (200
129 g) to obtain a final pesticide concentration of 1 mg kg^{-1} . The final moisture content of
130 the soils was adjusted to 60% of their maximum water holding capacity. Some soil
131 samples were first sterilised by autoclaving at $120 \pm 2 \text{ }^\circ\text{C}$ for 20 min on two consecutive
132 days (SS), and then treated with alachlor or metalaxyl, other soil samples (S) were
133 treated only with alachlor or metalaxyl and others were treated with both pesticide and
134 pine (S+P) or oak (S+O) sawdust. The soils were maintained in Erlenmeyer flasks
135 plugged with sterilized cotton wrapped in gauze to allow air exchange. Soil moisture
136 was kept constant throughout the experiments by periodic weighing and the replacement

137 of any losses with sterile water. Samples were incubated at 20 ± 0.5 °C in the dark.
138 Solutions and instruments were sterilised and all steps were performed in a sterile
139 cabinet.

140

141 *2.4. Chemical analysis*

142 Alachlor and metalaxyl and their metabolites were measured immediately after
143 treatment and at different sampling times (0, 1, 2, 3, 6, 8, 10, 14, 20, 28, 51, 70, and 98
144 days). Two soil replicates (1 g) were taken from each microcosm and shaken with 5 mL
145 of methanol for 24 h at 20°C for residue analysis. Samples were centrifuged and 4 mL
146 of each supernatant were evaporated under an air stream using an Evaporator EVA-
147 EC2-L (VLMGmbH, Bielefeld, Germany) and re-dissolved in 0.5 mL of methanol for
148 analysis. The quantitative determination of alachlor, metalaxyl and their metabolites
149 was performed by HPLC-DAD-MS (Waters Assoc., Milford MA, USA). A Waters
150 Symmetry C18 (75 mm x 4.6 mm I.D., 3.5µm) column was used at ambient
151 temperature. The mobile phase was 70:30 acetonitrile/water for metalaxyl and its
152 metabolites and 80:20 acetonitrile/water for alachlor, and the flow rate of the mobile
153 phase was 0.4 mL min^{-1} . The mobile phase was 90:10 acetonitrile/water for alachlor
154 ESA with a flow rate of 0.5 mL min^{-1} . The sample injection volume was 10 µL.
155 Detection by HPLC-DAD was at 196 nm for alachlor, 205 nm for alachlor ESA, and
156 194 nm for metalaxyl and its metabolites, and detection by HPLC-MS to confirm the
157 identity of these compounds was carried out by monitoring the positive molecular ion
158 (m/z) 238.2 for alachlor, 280.3 for metalaxyl, 266.2 for CGA62826 and 194.2 for
159 CGA92370 and the negative molecular ion (m/z) 314 for alachlor ESA.

160

161 *2.5. Soil dehydrogenase activity*

162 Soil dehydrogenase activity considered as overall soil microbial activity was
163 measured using the Tabatabai method (Tabatabai, 1994).

164

165 *2.6. Statistical analysis of the data*

166 The data obtained were subjected to analysis of variance. Standard deviation
167 (SD) was used to indicate variability among replicates in the determination of pesticides
168 or their metabolites and the least significant difference (LSD), at a confidence level of
169 95%, was determined to evaluate the effects of different soil treatments on
170 dehydrogenase activity. The statistical software Statgraphics Plus version 5.1
171 (Statgraphics Plus Corp., Princeton, NJ, USA) was used.

172

173 **3. Results and discussion**

174

175 *3.1. Degradation kinetics of alachlor and metalaxyl and metabolite formation*

176 Figure 1 shows the degradation kinetics for alachlor and metalaxyl in non-
177 amended soil (S), non-amended and sterilised soil (SS) and soil amended with pine
178 (S+P) or oak (S+O). The data are plotted as residual concentrations of the pesticide
179 (percentage of pesticide initially applied) against the time of incubation of each soil.

180 The degradation of alachlor followed first-order kinetics in the SS, S+O and S+P
181 samples, but followed a biphasic pattern in the S sample with a rapid first phase and a
182 slow second phase of degradation (Figure 1). The half-life ($t_{1/2}$) of alachlor in non-
183 amended soil was calculated in the first phase because degradation was rapid (>90% in
184 this first phase). The $t_{1/2}$ was 3.2 days and increased to a lesser extension in oak
185 amended soil (16.7 days) and to a greater extension in pine amended soil (151 days)
186 (Table 1). The degradation of alachlor in S+P was slower (<50% at the end of the

187 experiment) and the $t_{1/2}$ value should be considered with caution given that only an r^2 of
188 0.53 was reached. A slight degradation was observed in the sterilised soil, which might
189 be caused by chemical and abiotic factors other than photodegradation, as the soil
190 samples were kept in the dark during the incubation period.

191 The higher $t_{1/2}$ observed in the amended soils could be due to the higher TOC
192 content of these soils compared to the non-amended soil. Given that alachlor is
193 adsorbed mainly by the soil OM, its degradation is expected to depend on the OC
194 content of the soil. Several authors have suggested that OM could play an important role
195 in enhancing alachlor sorption by an amended soil (Dorado *et al.*, 2005). Wood residues
196 could enhance herbicide adsorption, decreasing the bioavailability and increasing the $t_{1/2}$
197 of alachlor in amended soils. In addition, a higher adsorption of alachlor by soil
198 amended with pine occurs due to the lignin content of the wood, which influences the
199 sorption of pesticides by soils (Kf values were 22.4 and 41.4 for the adsorption of
200 alachlor by oak and pine wood, respectively), as reported previously (Rodriguez-Cruz *et*
201 *al.*, 2007b). Other authors have also indicated that the addition of amendment (sewage
202 sludge) to soil led to a decrease in alachlor degradation (Rodriguez-Cruz and Lacorte,
203 2005).

204 Concentrations of alachlor ESA were detected simultaneously to parent
205 compound in non-sterile conditions. The evolution of this metabolite during the
206 incubation period is included in Figure 2. The maximum amount detected was 36.7 μg
207 kg^{-1} dry soil in S after 14 days. The amount then decreased, possibly due to the
208 degradation of alachlor ESA and the low input of new metabolite in accordance with the
209 residual parent compound (less than 20% of the original pesticide applied). In the S+O
210 sample, a relative increase in alachlor ESA was recorded at 50 days of incubation after a
211 rapid degradation of the parent compound (close to 30%) occurred between 28 and 70

212 days of incubation. This compound was not subsequently degraded at the end of the
213 experiment (70 days). Lower amounts of this metabolite were found in the S+P sample
214 according to a lower degradation of alachlor in this soil (Figure 1).

215 The degradation of metalaxyl in the non-amended and amended non-sterilised
216 soil samples studied followed first-order kinetics (Table 1). The results indicated that no
217 metalaxyl degradation occurred in the non-amended and sterilised soil (Figure 1). This
218 suggests a microbial role in degrading this fungicide. The $t_{1/2}$ values indicated a slower
219 degradation of metalaxyl in the S+P (144 days) or S+O (68.6 days) samples compared
220 with the S (29.1 days) sample. As indicated for alachlor, an increase in metalaxyl
221 adsorption by the amended soils occurs, with the fungicide being less available to
222 degradation. The adsorption by pine amended soil could be higher than by oak amended
223 soil according to the K_f values of metalaxyl by pine (8.28) and oak (4.95) wood
224 (Rodriguez-Cruz *et al.*, 2007b). Similarly, Fernandes *et al.* (2006) reported an increase
225 in the $t_{1/2}$ of metalaxyl in a sandy soil with different organic amendments due to the
226 increase in sorption, with the fungicide being protected from degradation.

227 The two metalaxyl metabolites studied (CGA62826 and CGA92370) were
228 detected in the non-sterilised non-amended and amended soils (Figure 2).
229 Concentrations of CGA62826 (acid metabolite) were higher than those of CGA92370.
230 The maximum amounts of CGA62826 metabolite were 91.5 (S), 175 (S+O) and 240
231 (S+P) $\mu\text{g kg}^{-1}$ dry soil at 51, 70 and 98 days, respectively. For CGA92370 metabolite,
232 the maximum amounts of 23.5 (S), 2.25 (S+O) and 2.61 (S+P) $\mu\text{g kg}^{-1}$ dry soil were
233 recorded at 70, 51 and 6 days, respectively (Figure 2). The faster degradation of
234 metalaxyl in the non-amended soil gave rise to a higher production of CGA92370 in
235 comparison with the amended soils.

236

237 3.2. Soil dehydrogenase activity

238 The dehydrogenase activity of non-amended and amended soils treated with
239 alachlor or metalaxyl for different incubation times was monitored as an indicator of the
240 overall microbial activity of the soils to determine the pesticide's possible side effects
241 on microbiological activity in the soil (Sukul, 2006) (Figure 3).

242 Soil dehydrogenase activity recorded a similar behaviour in non-amended and
243 amended soil for soil samples treated with alachlor. The values were not significantly
244 different (LSD=36.0, $p < 0.1$). However, significant differences were noted for time
245 (LSD=55.0, $p < 0.05$). A maximum value of dehydrogenase activity was detected on day
246 14 for all soils and this was followed by a decrease from day 28 up to the end of the
247 experiment on day 70. The increase was sharper in wood amended soils than in non-
248 amended soil reaching values of 162, 227 and 277 $\mu\text{g TPF g}^{-1}$ dry soil for S, S+P and
249 S+O, respectively (Figure 3). The increase in the dehydrogenase activity observed at 14
250 days may correspond to a maximum activity of microorganisms to degrade alachlor in
251 non-amended soil. A peak in the amount of its metabolite alachlor ESA was detected at
252 this time (Figure 2), indicating that microbial growth was stimulated by the use of
253 alachlor as a source of energy. Sukul (2006) reported that soil dehydrogenase activity is
254 likely to be affected by pesticides. However the activity of microorganisms in amended
255 soils might also be affected by the wood amendment as an additional carbon source.
256 Some authors have noted the positive influence of organic amendments on the
257 dehydrogenase activity of the microbial community (Moorman *et al.*, 2001; Delgado-
258 Moreno and Peña, 2007; Grenni *et al.*, 2009).

259 The soil dehydrogenase activity in amended and non-amended soils treated with
260 metalaxyl increased in the order $S \leq S+P < S+O$ (LSD=14.8, $p < 0.001$). However, this
261 activity was not significantly different over the entire incubation period for either non-

262 amended or amended soils (Figure 3). In a previous study, Pesaro *et al.* (2004) reported
263 that changes in specific activities (e.g., pesticide degradation) are not necessarily
264 reflected by bulk microbial activities such as dehydrogenase activity.

265

266 **4. Conclusions**

267 The use of wood residues to immobilize pesticides in soils affects the
268 degradation kinetics of these compounds as shown in this study. The addition of a pine
269 or oak wood amendment to the soil caused a significant reduction in the degradation
270 rate of alachlor and metalaxyl. The specific activity of the microorganisms degrading
271 the pesticides was negatively affected by the wood amendments due to the increased
272 adsorption of pesticides by woods and the decreased bioavailability of the pesticide to
273 be degraded. The results indicate that the use of wood amendments could be effective
274 for limiting the leaching of pesticides in the soil, although the adsorption capacity of the
275 amended soils should be taken into account, since it could decrease the degradation rate
276 of these compounds. Further knowledge on the role wood residues play in influencing
277 the degradation of pesticides in the soil will provide a better understanding of the
278 bioavailability and potential toxicity of these contaminants and their metabolites.

279

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285

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354 **Table captions**

355

356 **Table 1.** Kinetic equation, correlation coefficient and half-life values of alachlor and
357 metalaxyl in non-amended (sterilised-SS and non-sterilised-S) and pine (S+P) or oak
358 (S+O) amended soils.

359

360

361

362 **Figure captions**

363

364 **Figure 1.** Degradation kinetics of alachlor and metalaxyl in sterilised soil (SS), non-
365 sterilised soil (S) and soil amended with pine (S+P) or oak (S+O) woods. Bars indicate
366 the standard deviation of the replicates (n = 4).

367

368 **Figure 2.** Formation of alachlor ESA and metalaxyl metabolites CGA 62826 and CGA
369 92370 in non-sterilised soil (S) and soil amended with pine (S+P) or oak (S+O) woods.
370 Bars indicate the standard deviation of the replicates (n = 4).

371

372 **Figure 3.** Dehydrogenase activity of non-amended (S) and amended soils (S+P and
373 S+O) treated with alachlor and with metalaxyl at different incubation times. Bars
374 indicate the standard error of the replicates (n = 2).

375

1

2

3 **Table 1.** Kinetic equation, correlation coefficient and half-life values of alachlor and
 4 metalaxyl in non-amended (sterilised-SS and non-sterilised-S) and pine (S+P) or oak
 5 (S+O) amended soils.

Pesticide/Treatment	Kinetic equation	r²	t_{1/2} (d)
Alachlor			
SS	$y = 98.935 e^{-0.0035x}$	0.78	198
S ^a	$y = 101.65 e^{-0.2187x}$	0.99	3.17
S+P	$y = 76.54 e^{-0.0046x}$	0.53	151
S+O	$y = 90.738 e^{-0.0414x}$	0.98	16.7
Metalaxyl			
SS	-	-	nd ^b
S	$y = 108.75 e^{-0.0238x}$	0.98	29.1
S+P	$y = 95.073 e^{-0.0048x}$	0.95	144
S+O	$y = 94.613 e^{-0.0101x}$	0.96	68.6

6 ^aFirst-order equation calculated for the first degradation phase (0-6 days); ^bnd, no
 7 degradation.

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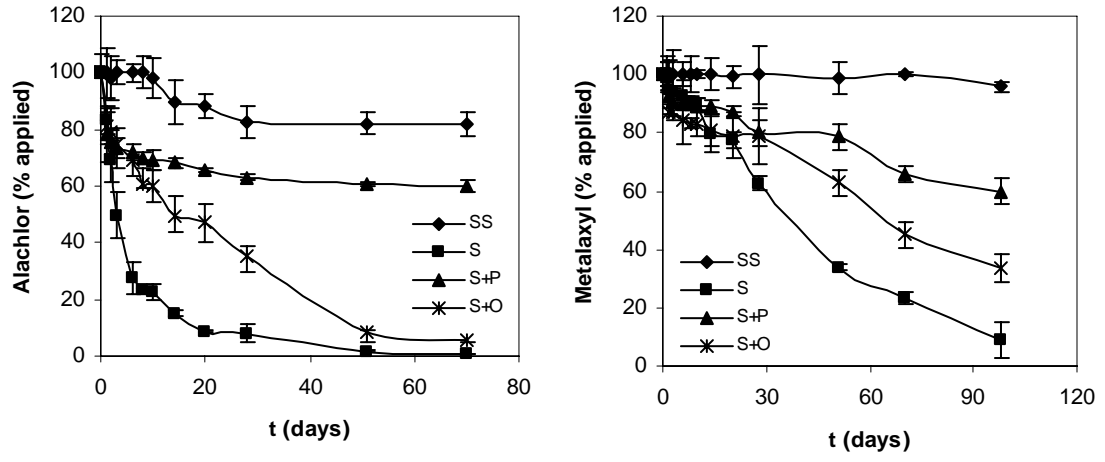


Figure 1

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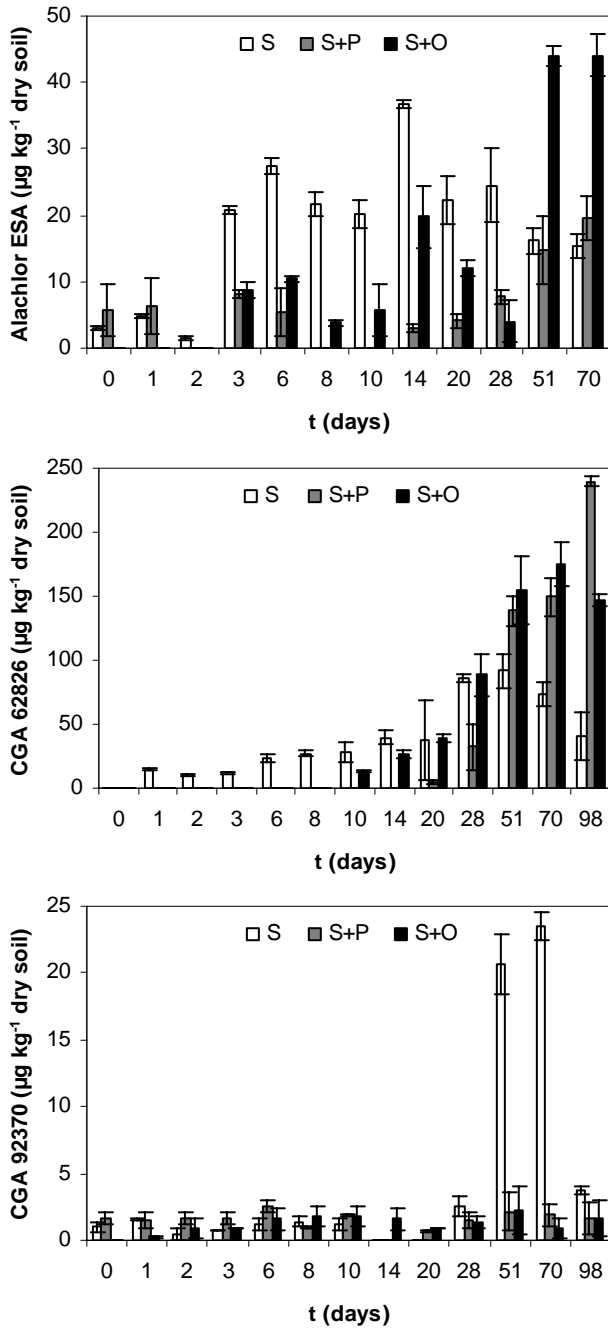


Figure 2.

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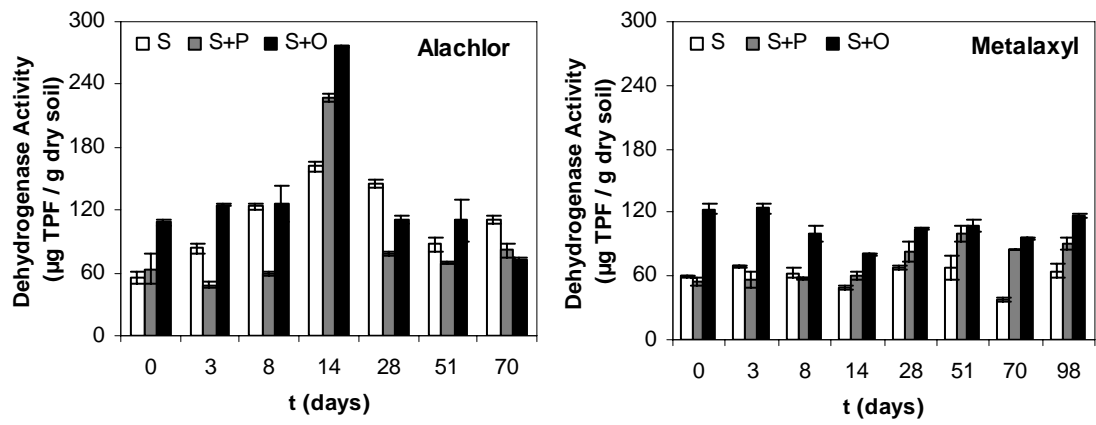


Fig. 3.