Improving the performance of calcium-containing spray formulations to limit the incidence of bitter pit in apple (*Malus x domestica* Borkh.)

Alvaro Blanco; Victoria Fernández^a, Jesús Val^{*}

Estación Experimental de Aula Dei (CSIC). Avda. Montañana 1005. 50059- Zaragoza. Spain

*corresponding autor. Tel: + 34 976 716130; FAX: +34 976 716145 E-mail address: <u>jval@eead.csic.es</u>

Abstract

Laboratory and field experiments were carried out with apples (*Malus x domestica* Bork) cv. 'Golden Reinders', to assess the efficacy of sodium salt of carboxymethyl ether of cellulose (0.5%, CMC) as an adjuvant for Ca spray formulations containing either Cachloride or Ca-propionate as active ingredient (120 or 250 mM Ca). This additive significantly increased the retention of Ca-containing solutions by the apple skin and prolonged the process of drying of the solution at room temperature. Four days after immersion of apples in 0.5% CMC plus CaCl₂ or Ca-propionate solutions (120 and 250 mM Ca) significant Ca increases were recorded in the peel and cortex of treated fruits. Application to apple trees of in-season sprays containing 250 mM CaCl₂ plus 0.05% Tween 20, Ca-propionate (120 and 250 mM Ca) plus 0.5% CMC or 250 mM CaCl₂ plus 0.5% CMC had no impact on fruit yield and quality, but significantly limited the rate of bitter pit incidence during the following 3-month cold-storage period. Evidence is provided that addition of appropriate adjuvants to Ca sprays can favour the distribution of Ca into the apple fruit and help to reduce the incidence of Ca-related disorders over the postharvest cold-storage period.

Keywords: adjuvants, apple, bitter pit, Ca-propionate, carbxymethylcellulose, calcium sprays, fruit quality, humectancy, spray retention.

1. Introduction.

Bitter pit remains as one of the main problems for apple growing industry around the world, particularly in areas where climatic conditions are generally dry. Such physiological disorder which develops during the period of fruit growth (Ferguson et al., 1999), has generally been related to calcium (Ca) deficiency in the fruit cortex (Ferguson and Watkins, 1989; Fallahi et al., 1997).

The low mobility of Ca in the plant poses serious problems to enhance the distribution of this element to the fruit via Ca application to the root system (Bangerth, 1979). Subsequently, treatment of aerial plant parts with Ca sprays, is recommended and applied in many fruit production areas of the world, either as routine applications to prevent the occurrence of localised Ca deficiency in the fruit or to improve fruit quality (Schlegel and Schönherr, 2002a,b; Lötze et al., 2008; Fernández et al., 2009). The efficacy of Ca treatments supplied to leaves and fruits may be highly variable and currently, many factors involved on the penetration and distribution of Ca within the fruit remain unclear (Saure, 2005; Bai et al., 2008; Val et al., 2008).

The effects of in-season spraying and/or post-harvest dipping of apple fruits in Ca solutions have been evaluated in various studies in terms of e.g., bitter pit development, Ca content increase and improved fruit firmness. However, inconsistent results have been often reported (van Goor, 1971; Lidster and Porritt, 1978; Hewett and Watkins, 1991; Neilsen et al., 2005; Lötze and Theron, 2006; Lötze et al., 2008; Val et al., 2008). Recently, Val et al. (2008) showed that in-season CaCl₂ sprays led to increased Ca concentrations in the skin, while no significant changes were measured in the cortex of apples.

Some studies estimated the permeability of apples to Ca solutions either with intact fruits (van Goor, 1973; Mason et al., 1974), fruit discs (Schlegel and Schönherr, 2002a,b) or cuticular membranes (Glenn and Poovaiah, 1985; Harker and Ferguson, 1988; Chamel, 1989; Harker and Ferguson, 1991). Regarding the penetrability of apples at different developmental stages, Schlegel and Schönherr (2002a,b) observed that apples were highly permeable to CaCl₂ solutions until June drop, the fruits turning significantly less penetrable after such date. The authors suggested that this may be due to the potential contribution of existing stomata and lenticels in the surface of young apple fruits, which would disappear after June drop (Schlegel and Schönherr, 2002a,b).

Penetration of the plant surface by a nutrient solution may occur via stomata, the cuticle, cuticular cracks and imperfections and through trichomes or specialised epidermal cells. Most research efforts in the last decades focused on investigating the diffusion of substances thorough the plant cuticle (Schönherr, 2006). To explain the mechanisms of cuticular penetration of apolar, lipophilic compounds the "diffusion-dissolution model" was proposed (Riederer and Friedmann, 2006). On the other hand, the penetration pathway of hydrophilic solutes through the cuticle is currently not fully understood (Fernández and Eichert, 2009) and the existence of "aqueous pores" as a parallel diffusion mechanism has been hypothesised (Schönherr, 2006). The occurrence of epidermal structures in plant surfaces such as lenticels, stomata or trichomes may significantly influence the rate of absorption of surface-applied agrochemicals. While the significance of the stomatal pathway on the absorption of foliar sprays has been a matter of controversy for many years, recent evidence shows that it can largely contribute to the uptake process (Eichert et al., 2008).

Using young apples and apple segments, Schlegel and Schönherr (2002a,b) suggested the major contribution of stomata and trichomes to the penetration of $CaCl_2$ solutions only at early developmental stages. On the other hand, Harker and Ferguson (1988, 1991) and Glenn and Poovaiah (1985) suggested that lenticels in mature apples were preferential sites for the uptake of Ca solutions through the fruit surface.

Several studies showed that surface treatment of cherry fruits with Ca compounds at different stages of development decreased the incidence of cracking (e.g., Glenn and Poovaiah 1987; Brown et al., 1995; Wermund et al., 2005). Furthermore, trials developed by Glenn and Poovaiah (1987) showed that the integrity of the cell wall structure was better preserved in apples treated with Ca, which presented a greater cell-to-cell contact as compared to non-treated fruits.

The effectiveness of Ca sprays is largely influenced by the prevailing environmental conditions, particularly relative air humidity (Schönherr, 2000, 2001). In this regard, Schönherr (2001) pointed out the relevance of the point of deliquescence (POD) of Ca compounds in relation to the rate of diffusion through the cuticle. Bai et al. (2008) showed that foliar treatment with Ca-chloride and Ca-hydroxide in combination with two non-ionic surfactants led to transient changes in the rate of photosynthesis and stomatal conductance of apple and bean leaves. Recently, Kramer et al. (2009a, b) assessed the permeability, drop and

deposit characteristics of $CaCl_2$ and Ca-acetate formulations applied to tomato fruit and adaxial apple leaf isolated cuticles. They found that surfactants increased the spreading of Ca within the droplet, and also that the rate of penetration of $CaCl_2$ was always higher than that of Ca-acetate.

Apart from surface-active agents, few studies tested the effect of employing adjuvants to improve the penetration rate and performance of Ca formulations (Mason et al., 1974; Schönherr, 2001; Fernández et al., 2009). Addition of suitable adjuvants into spray formulations can help increase the rate of retention, spreading, penetration and drying of the solution, thereby, improving the performance of fertilisers. Sodium salt of carboxymethyl ether of cellulose (CMC) is a food additive used by the agro-food industry to improve e.g., moisture retention, as a thickener or as an emulsion stabiliser (Ghannam and Esmail, 1998; Nie et al., 2004).

Calcium-propionate could potentially be a good Ca source since it is a small organic salt molecule, which may provide anti-fungal properties. This chemical is also used by the food industry, and has been successfully tested as a fungicide in peach for canker (Biggs et al., 1994) and brown rot (Biggs et al., 1997), and in apple to provide both protective and curative effects against infections caused by *Botrytis cinerea* (Droby et al., 2003).

The goal of this investigation was to develop laboratory and field studies intended to enhance the uptake of Ca in apple and thereby reduce bitter pit development and improve storage quality of fruit. The working hypothesis was to evaluate whether the addition of 0.5% CMC as formulation adjuvant to be used in combination with Ca-propionate and CaCl₂ as active ingredients, provided beneficial effects in terms of formulation properties, fruit tissue Ca increases and fruit quality and storability.

2. Materials and methods.

2.1 Physical-chemical properties of solutions.

In this experiment, the surface tension (IFT, mN m⁻¹) and the rate of retention of solutions containing the adjuvants 0.5% CMC (w/v) (ChemWorld S.A., Barcelona, Spain) and Tween[®]20 (Polyoxyethylene sorbitane monolaurate, Panreac, Barcelona, Spain) were determined in combination (solution) with CaCl₂ (Panreac, Barcelona, Spain) or Capropionate (Perstorp_Ltd., Perstorp, Sweden) as calcium sources (120 or 250 mM Ca) dissolved in distilled water. The solutions evaluated were: pure distilled water, 0.5% CMC, pure 120 and 250 mM Ca (CaCl₂ and Ca-propionate), 250 mM CaCl₂ plus 0.05% Tween 20 and finally, 120 and 250 mM CaCl₂ and Ca-propionate in combination with 0.5% CMC.

To record the surface tension, solutions were prepared immediately before measurement to avoid the occurrence of interactions among active ingredients and formulation adjuvants (Fernández et al., 2006). Data were recorded with a DSA 100 Drop Shape Analysis System (Krüss GmbH, Hamburg, Germany) with 10 repetitions per treatment.

The rate of retention of solutions by the apple skin was estimated after immersion of the outer surface of 23 cm² fruit discs, making sure that the fruit cortex was not wet during the process. Discs were weighed before and after immersion in the treatment solutions and the rate of retention per unit surface was determined as the weight difference after wetting (μ L cm²), with 5 repetitions. The time at room temperature (30-40% relative air humidity and 20-23 °C) until dryness of the solutions retained in apple discs was estimated for 2 h by the naked-eye observations.

2.2 Calcium penetration laboratory experiments.

A preliminary Ca application trial was carried out with harvested apples to test the effect of 0.5% CMC in combination with $CaCl_2$ or Ca-propionate. Cold stored, 'Golden Delicious' apples were randomly selected, carefully washed with soap, rinsed with tap water, subsequently washed in 0.1N HCl, and finally rinsed in distilled water to eliminate all residues from pre-harvest Ca-applications.

Seven treatments were applied to the fruits, namely: an untreated control, 250 mM $CaCl_2$ plus 0.05% Tween 20, 0.5% CMC alone, 0.5% CMC plus 120 mM or 250 mM $CaCl_2$ and 0.5% CMC in combination with either 120 or 250 mM Ca-propionate. The experimental unit was the single apple and treatments were replicated 5 times.

Formulations were applied by immersion of the entire fruit in the treatment solutions for 5 seconds. Thereafter, fruits were kept for 4 days under laboratory conditions (20-23 °C and 30-40% relative air humidity) before tissue mineral element analysis after thorough fruit washing as described above. Afterwards, fruits were processed for mineral element determination.

2.3 Field experiments.

A trial was performed during the summer of 2008 in a 'Golden Reinders'/PAJAM1 commercial orchard located in Alfamén (Middle Ebro Valley, Spain). Trees were selected as uniform as possible in terms of tree size and flowering, and had an average trunk girth of 17.9 \pm 2.1 cm. Based on the results obtained by Val et al (2008), trees were sprayed with 120 or 250 mM Ca-propionate in combination with 0.5% CMC, and were compared to trees sprayed with 250 mM CaCl₂ plus 0.5% CMC or 0.05% Tween 20, and also with untreated controls. In all, 5 treatments were applied. The experiment was designed as randomized blocks, with 4 replications, and single trees were the experimental unit.

Trees were sprayed to run-off, using in average 1.4 ± 0.2 L per tree of solution, paying special attention at targeting the fruits with the fertiliser solutions. Treatment application started on May 20th i.e., 44 days after full bloom (DAFB), and sprays were repeated at monthly intervals, until August 26th (approximately 1 month before harvest). A total of four in-season spray applications were applied to trees during the growing season of 2008.

Samples of five fruits per experimental unit at different dates along the growing season were randomly collected to record their weight and diameter and for mineral element analysis. A first sample was collected on May 8th prior to the 1st application of Ca-sprays. Afterwards, samples were routinely picked 7 days after each date of spray application. Twenty leaves per experimental unit were also randomly harvested around the tree canopy, from the middle of the shoots. Leaf samples were collected 51, 87, 116 and 151 DAFB, and used for biometry and mineral element analysis.

At harvest, (i.e., September18th 2008), yield and number of fruits per tree were recorded, and a randomly selected sample of fruits was used for fruit quality traits assessment (fruit length and diameter, flesh firmness (FF; using an Effegi penetrometer fitted with an 11.1 mm diameter tip), total soluble solids (TTS; using a PR-101 digital refractometer; ATAGO Co., Tokyo, Japan) and juice acidity (TA; g malic acid L⁻¹ juice). Besides, a larger sample (a box of around 12 kg in average) per experimental unit was also collected and stored under normal cold conditions (4° C) for bitter pit development evaluation. This was made by recording the number of fruits affected on a monthly basis, and expressing such proportion in relation to the total fruit number for statistical analysis.

2.4 Mineral element analysis.

For mineral element determination, fruits were carefully washed as previously described (Val et al., 2010), and a 1-cm transverse section of the fruit was cut at the equator, the skin was carefully separated from the cortex, and aliquots of each tissue were prepared for analysis. For tissue element determination, peels were finely cut (< 1 mm thick) and mixed. A 0.5 g aliquot was taken for mineral analyses. The cortex of each apple was cut into small pieces and mixed, and a 2-g aliquot was retained for analyses. Fruit tissues were wet digested using 10 mL HNO₃ and 2 mL H₂O₂ on a hot plate. Once the samples were dry, they were dissolved in 10 mL of HCl to which 15 mL of water was added.

For leaf mineral element analysis, 40 leaves from each experimental unit were randomly collected from the mid-section of growing shoots from around the tree canopy. After removing the petioles, the leaf tissue was carefully washed and scrubbed in a liquid soap solution (1%), rinsed in tap and deionised water to eliminate surface contamination, and finally dried in an oven at 60 °C for 2 days. The analyses were based on 1-g dry-weight aliquots. Dry-ashing was performed following the methods of C.I.I. (1969) and Pinta and DeWaele (1975) and Ca concentrations were subsequently determined by Atomic Absorption Spectroscopy (Thermo Scientific iCE 3300 AA Spectrophotometer, Cambridge, UK).

2.5 Statistical analysis.

Data were analysed by ANOVA and when significant, means were separated by Duncan's multiple range test. Angular transformation of data on bitter pit incidence was made prior to statistical analysis. For each treatment, the quadratic regression of the values was analysed against time, expressed as days of cold storage after harvest.

3. Results

3.1. Physical-chemical properties of solutions

The surface tension and rate of retention of the formulations used is shown in Table 1. Pure distilled water and CaCl₂ solutions had IFT values higher than 71 mN m⁻¹, while pure Capropionate solutions had significantly lower surface tension values ranging from 64.7 to 65.2 mN m⁻¹ for 120 and 250 mM Ca, respectively. Solutions containing 0.5% pure CMC had an IFT of 64 mN m⁻¹, all CMC combinations with Ca sources ranging between 60 to 63 mN m⁻¹. For both Ca compounds, solutions containing the highest Ca concentrations (i.e., 250 mM) had the lowest IFT values.

The lowest rates or retention were determined for 250 mM plus 0.05% Tween 20, pure 250 mM Ca-propionate and pure 120 mM Ca-propionate solutions (3.74, 3.94 and 5.1 μ L cm⁻¹, respectively). The values determined for unformulated 120 and 250 mM CaCl₂ solutions corresponded to 9.9 and 8.2 μ L cm⁻². Pure 0.5% CMC solutions had higher retention rates (11.4 μ L cm⁻²) as compared to the performance of pure water on the apple peel (7.5 μ L cm⁻²). Subsequently, the retention of Ca treatments by the surface of apples was very much improved by the addition of 0.5% CMC, especially in the case of Ca-propionate which showed poor retention rates when applied without formulants.

Drying of solutions at room temperature was prolonged in the presence of $CaCl_2$ (point of deliquescence of 32%, Schönherr, 2001) and also when 0.5% CMC was added to the formulations. Apple surfaces immersed in pure 0.5% CMC solutions dried after 1 h, in contrast to those dipped in pure distilled water, which dried approximately after 20 min. The external surface of apple discs treated with Ca-propionate-based solutions were totally dry

after 20-30 min, and the addition of 0.5% CMC to such Ca-formulations increased the drying period up to 60-70 min.

3.2 Calcium uptake laboratory experiments.

A set of preliminary trials, to test the performance of CMC as an adjuvant for Ca compounds was carried out by dipping apples in treatment solutions and analysing Ca increases in the peel and cortex of the fruits. Dipping apples in Ca-containing solutions led to increased Ca peel and cortex concentrations *versus* the values recorded for untreated fruits and those immersed in 0.5% CMC (Table 2). The highest peel Ca concentrations were determined after dipping apples in solutions containing 0.5% CMC in combination with 120 mM Ca-propionate, 120 mM CaCl₂ and 250 mM CaCl₂.Pulp Ca concentrations of Ca-treated apples were significantly higher than those measured for pure 0.5% CMC-treated ones.

3.3 Field experiment.

This trial was performed to assess the effect of Ca-propionate (120 and 250 mM Ca) plus 0.5% CMC as compared to 250 mM CaCl₂ plus either 0.05% Tween 20 or 0.5% CMC under field conditions. The analysis of Ca in the cortex of the fruits (Fig. 1) showed that the concentration of this element generally decreased as the growing season progressed, and that an increase was observed in CaCl₂-treated fruits during the first weeks following the start of the spraying program. By harvest, Ca concentration was greater in the cortex of fruits treated with 250 mM Ca plus CMC. In the peel (Fig. 1), Ca concentrations increased following the application of CaCl₂, although in these fruits, the concentration lowered to that of untreated fruits by the end of the growing season. On the contrary, when Ca-propionate was applied, values were close to normal during the growing season and for the 250 mM Ca treatment, an increase was recorded at the time of harvest.

At harvest, no differences in yield, mean fruit weight or productivity parameters were found. However, fruit quality traits showed differences among treatments (Table 3): an increase in flesh firmness was generally associated with Ca-treatments, which was significantly greater in those sprayed with 250 mM Ca-propionate plus CMC.

A large sample of fruits was cold-stored, where development of bitter pit was expected to occur. Data recorded and transformed along the following three months of cold storage showed a cuadratic increase in the proportion of affected apples (Fig. 2). The goodness of fit tests resulted into highly to very highly significance levels, and determination coefficients ranged between 57.10 and 73.28. The results show that apples from trees treated with Caformulations containing CMC as an adjuvant always presented a lower level of bitter pit incidence as compared to those collected from Tween20-treated, or untreated trees.

The evolution of leaf Ca concentrations along the growing season following the application of foliar Ca sprays was assessed as shown in Fig. 3. It was observed that leaf tissue Ca concentrations increased linearly during the season up to the time of harvest. Comparison of regression lines showed that they were parallel in all treatments, as slopes did not differ from each other, but those obtained from CaCl₂-treated trees had significantly greater intercepts than the rest of treatments.

4. Discussion.

In this study the effect of Ca formulations in increasing tissue Ca concentrations, reducing bitter pit and enhancing fruit quality traits was evaluated under laboratory and field conditions. The aim was to induce positive responses in fruits via the application of in-season Ca sprays and by adding the adjuvant CMC to the formulations. Such additive is commonly used in the food industry as thickener, water binder or as emulsion stabiliser (Ghannam and Esmail, 1998).

The efficiency of nutrient sprays is largely affected by the prevailing climatic conditions at the time of treatment, which may limit in time the process of penetration of the spray solution into the fruit (Fernández and Eichert, 2009). For instance, rapid drying of spray solutions can be expected under the warm and dry conditions observed in the summer in many arid and semi-arid areas of the world, such as in NE Spain. Thereby, addition of humectants and stickers, to the formulations such as CMC, may increase the retention of the active ingredient, prolong the penetration process and increase the rate of Ca distribution within the fruit (Fernández and Eichert, 2009), a hypothesis which was tested in this investigation.

The performance of CMC solutions in combination with $CaCl_2$ or Ca-propionate was evaluated and compared with the results obtained for $CaCl_2$ plus Tween 20 as reported previously (Val et al., 2008). Both Ca compounds had different points of deliquescence (32 % and >95% for CaCl₂ and Ca-propionate, respectively; Schönherr, 2001) and the organic salt was selected because it provides antifungal properties (Biggs et al., 1994; 1997), which could induce beneficial effects in terms of avoiding post-harvest disorders.

Initially the physical-chemical properties of the solutions were evaluated under laboratory conditions. Results indicated that addition of CMC increased significantly the rate of retention of Ca-containing solutions by the apple peel, in addition to the humectant effect conferred by this adjuvant. It was concluded that the sticking and water-binding properties of CMC will increase the humectancy and retention of the foliar spray formulation and favour the process of Ca penetration through the plant surface as observed in this investigation.

Development of laboratory dipping trials with 0.5% CMC and Ca-containing solutions (120 and 250 mM) provided evidence for the penetration and distribution of Ca in the apple fruit. Recently, Kraemer et al (2009a) showed that Ca penetration is directly related to the area covered by Ca within the spray droplet, which was found to vary in relation to the addition of a surfactant. Therefore, it could be also the case that CMC may affect the distribution of Ca in the drops after dipping apples into the solution or spraying the treatments under field conditions as suggested by Kraemer et al (2009a,b).

A decreasing trend in apple peel Ca concentrations was observed over the growing season regardless of the Ca-treatments applied to the trees. In contrast, cortex Ca concentrations fluctuated during the growing season and reached a maximum between 80 to 110 days after full bloom when fruits were treated with 250 mM CaCl₂ (plus 0.05% Tween 20 or 0.5% CMC). Similarly, maximal leaf Ca concentrations were recorded after application of both CaCl₂ containing formulations (i.e., with CMC or Tween 20) suggesting the higher permeability of apple fruits and leaves to CaCl₂ as compared to Ca-propionate as a Ca source. Increased Ca penetration rates in association with CaCl₂ versus other Ca-containing compounds such as e.g., Ca-acetate (Kraemer et al., 2009a,b), Ca-EDTA (Manganaris et al., 2005) or Ca-propionate (Schönherr, 2001) have been previously reported, which can be associated with the lower molecular size and lower point of deliquescence of the salt as compared to the organic compounds (Schönherr., 2001; Kraemer et al. 2009a).

Application of multiple Ca sprays containing 0.5% CMC was not detrimental for fruit quality and slowed the development of bitter pit during cold-storage. A reduction in the incidence of post-harvest disorders in apples by the application of in-season Ca sprays as been shown by some authors (e.g., Hewett and Watkins, 1991; Schmitz-Eiberger et al., 2002; Lötze et al., 2008).

Regarding the mechanisms of action of CMC, in addition to its potential to lower the point of deliquescence of the Ca formulation and capacity to increase the retention of the fertiliser by the plant surface as shown in this investigation, it should be added the fact that it may change the distribution of Ca within the spray drop and may also affect the rainfastness of Ca treatments, as suggested by Kraemer et al (2009b). However, more research efforts are required to clarify the mode of action of foliar formulation additives in combination with active ingredients as a strategy to improve the performance of nutrient sprays under field conditions.

5. Conclusion.

The use of 0.5% CMC in combination with $CaCl_2$ or Ca-propionate increased the rate of retention by the apple peel and solution humectancy, thereby favouring the process on penetration through the plant surface. Highest tissue Ca concentrations were detected when 250 mM Ca formulations were applied to the fruits. Foliar sprays containing 0.5% CMC plus chiefly 250 mM CaCl₂ or Ca-propionate led to significant decreases in the rate of bitter pit during cold storage. It is concluded that the adjuvant CMC lowered the point of deliquescence of the Ca-containing solutions and increased their rate of retention as assessed gravimetrically under laboratory conditions. However, more research efforts are required to optimise the use of CMC as a spray additive and to understand the mechanisms associated with the penetration and distribution of exogenous Ca solutions in plant tissues and the mode of action of adjuvants regarding such processes.

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Table 1.

Ca treatment	Adjuvant	IFT (mN m ⁻¹)	Retention rate* $(\mu L \text{ cm}^{-2})$	Time until dryness (min)
-	Pure water	73.2 ± 0.14 i	$7.47 \pm 1.02 \text{ ab}$	20
-	0.5% CMC	$64.0 \pm 0.08 \text{ ef}$	11.35 ± 1.05 de	90
120 mM CaCl ₂	-	72.4 ± 0.11 hi	9.99 ± 2.46 cde	60
250 mM CaCl ₂	-	$71.9\pm0.27~h$	8.82 ± 2.23 bcde	60
120 mM Ca-Propionate	-	65.2 ± 0.63 g	5.1 ± 0.45 a	28
250 mM Ca-Propionate	-	$64.7 \pm 0.41 \text{ fg}$	3.94 ± 0.27 a	20
250 mM CaCl ₂	0.05% Tween20	46.2 ± 0.18 a	3.74 ± 0.39 a	45
120 mM CaCl ₂	0.5 % CMC	63.1±0.5 de	10.5 ± 1.0 cde	> 90
250 mM CaCl ₂	0.5 % CMC	$60.1\pm0.1~b$	9.52 ± 2.25 cde	> 90
120 mM Ca-Propionate	0.5 % CMC	$62.9\pm~0.15~d$	$12.03 \pm 0.86 \text{ de}$	70
250 mM Ca-Propionate	0.5 % CMC	$61.2 \pm 0.10 \text{ c}$	12.79 ± 1.74 e	60

Surface tension (IFT; mN m⁻¹), rate of retention (μ L cm⁻²) and estimated drying time (min) at room temperature after dipping the outer surface of apple discs in treatment solutions. Data are means ± SE (N= 10, for IFT, and N= 5 for retention rate and time until dryness)

⁺: Within these columns, values followed by same letter indicate no significance differences according to Duncan's Multiple Range Test ($P \le 0.05$)

* the retention rate was determined gravimetrically after and prior to immersion of the surface of apple disks in the treatment solutions

Table 2

Calcium concentration (mg 100 g⁻¹ FW) in the cortex and skin of untreated apples versus the values recorded 4 days after immersion in: 0.5% CMC alone or in CaCl₂ and Ca-propionate (120 or 250 mM Ca) solutions in combination with 0.5% CMC. Post-harvest treatments were applied to 'Golden Delicious' apples. Data are means \pm SE (N= 5).

Treatments	Cortex	Skin
Untreated control	2.02±0.25 a	13.69±1.19 a
CMC alone	$2.12\pm0.03a$	13.73±1.29 a
120 mM CaCl ₂	4.61±1.83 b	29.99±3.50 c
250 mM CaCl ₂	3.65±0.45 ab	28.42±2.30c
120 mM Ca-propionate	5.60±0.69 b	32.89±0.66 c
250 mM Ca-propionate	4.41±0.50 b	21.35±0.66 b
significance	***	***

***: significant at $P \leq 0,001$.

Within columns, values followed by same letter are not significantly different according to Duncan's Multiple Range Test ($P \le 0.05$)

Table 3

Quality traits of 'Golden Reinders' fruits sampled at harvest and treated with multiple in-season Ca sprays during the growing season. The spray formulations applied to the trees were: 250 mM CaCl_2 plus 0.05% Tween 20, 250 mM CaCl₂ plus 0.5% CMC, 120 mM Ca-propionate plus 0.5% CMC and 250 mM Ca-propionate plus 0.5% CMC. Data are means \pm SE (N=10).

Treatments	Flesh firmness	TSS	Acidity
	(N)	(° Brix)	$(g L^{-1})$
Control	57.1±1.62 a	11.9±0.40	3.89±0.09
250 mM CaCl ₂ + Tween 20	61.8±2.11 ab	12.4±0.21	4.44±0.31
$250 \text{ mM CaCl}_2 + \text{CMC}$	63.4±2.21 ab	11.4±0.37	4.21±0.45
120 mM Ca-propionate + CMC	58.1±1.14 a	12.5±0.48	4.36±0.10
250 mM Ca-propionate + CMC	65.4±0.62 b	11.9±0.45	4.00±0.61
Significance	*	ns	ns

ns, *: non significant, or significant at $P \le 0.05$,

⁺: Within this column, values followed by same letter are not significantly different according to Duncan's Multiple Range Test ($P \le 0.05$)

Figures:

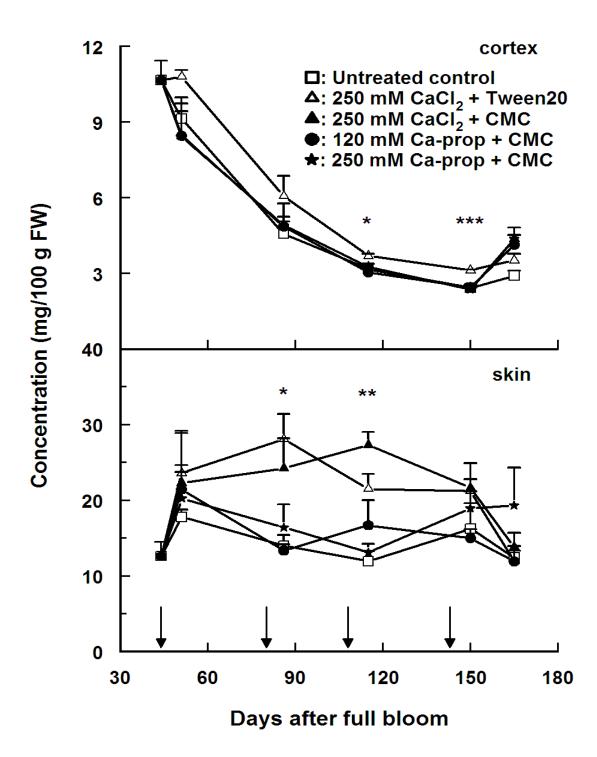


Fig. 1.- Changes in Ca concentration along the growing season in 'Golden Reinders' fruits treated with multiple pre-harvest sprays of 250 mM CaCl₂ (plus 0.05% Tween 20 or 0.5% CMC), and 120 or 250 mM Ca-propionate plus 0.5% CMC. Arrows indicate Ca-spray dates. (Mean \pm SE of 4 replicates).

*, **, ***: significant at $P \le 0.05$, $P \le 0.01$ or $P \le 0.001$ respectively, within each date of analysis.

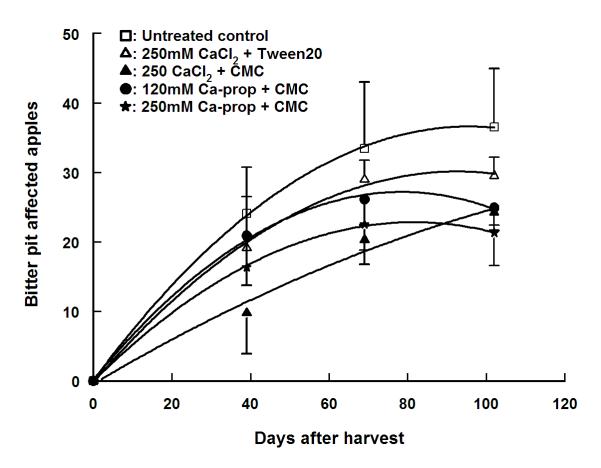


Fig. 2.- Development of bitter pit incidence during cold storage of 'Golden Reinders' apples sprayed with multiple pre-harvest treatments of 250 mM CaCl₂ (plus 0.05% Tween 20 or 0.5% CMC), and 120 or 250 mM Ca-propionate plus 0.5% CMC. (Mean \pm SE of 4 replicates).

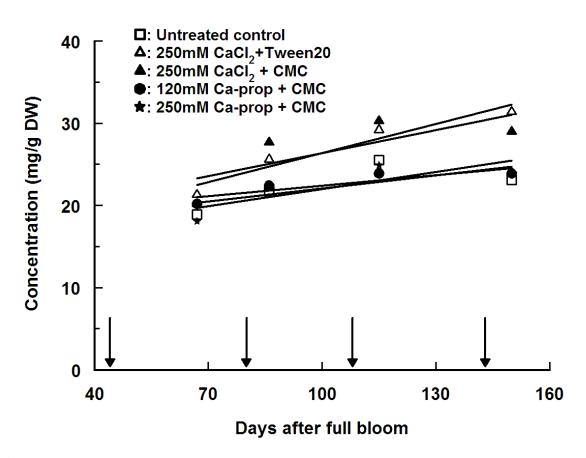


Fig. 3.- Leaf tissue Ca concentrations determined in leaves collected from apple trees sprayed with different Ca formulations during the growing season. Arrows indicate Ca-spraying dates. (Means of 4 replicates).