

1           **Quantitative determination of octylphenol, nonylphenol, alkylphenol**  
2           **ethoxylates and alcohol ethoxylates by pressurized liquid extraction and**  
3           **liquid chromatography-mass spectrometry in soils treated with sewage**  
4           **sludges**

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12  
13           **Abstract**

14  
15           Surfactants have one of the highest production rates of all organic chemicals. Non-ionic  
16           surfactants, especially alkylphenol ethoxylates, received most attention as precursors of  
17           estrogenic metabolic products generated during wastewater treatment. Alkylphenols (octyl  
18           and nonylphenol), alkylphenol polyethoxylates (APEOs), and alcohol ethoxylates (AEOs)  
19           have been determined in a Mediterranean forest soil (Mediterranean Rendzic Leptosol)  
20           amended with sludges from six waste water treatment plants (WWTPs) located in the  
21           Valencian Community. These compounds were isolated from soil by pressurized liquid  
22           extraction (PLE) using a mixture acetone-hexane (50:50 v/v), the extracts were cleaned up by  
23           solid-phase extraction (SPE) with C<sub>18</sub>, and determined by liquid chromatography atmospheric  
24           pressure chemical ionization-mass spectrometry (LC-APCI-MS) using analytical standards  
25           for quantification. The method enabled high-reliable identification by monitoring the  
26           corresponding ammonium adduct [M+NH<sub>3</sub>]<sup>+</sup> for AEOs and APEOs, and the deprotonated  
27           molecule [M-H]<sup>-</sup> for octyl and nonylphenol. Recoveries, determined spiking soil samples at  
28           different concentrations, ranged from 89 to 94 %, with limits of quantification from 1 to 100  
29            g kg<sup>-1</sup>. Data obtained from a soil sample mixed with biosolids in the laboratory showed that  
30           these compounds are present at concentrations ranging from 0.02 to 5 mg kg<sup>-1</sup>. According to  
31           these concentrations, levels of possible risk can be concluded for the presence of non-ionic  
32           surfactant in soil. However, further assessment will be necessary to establish the relationship  
33           between exposure and effect findings.

34  
35           *Keywords:* Soil contamination, Amended soils, Surfactants, Non-ionic detergents, LC-APCI-  
36           MS.

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## 38 **1. Introduction**

39

40 In the Mediterranean Europe one of the critical factors regarding soils is the decline of the  
41 organic matter content. It is mainly due to intensive agriculture and the increasing incidence  
42 of forest fires, under an extreme and changing climate. Most of the agricultural soils of this  
43 area are of the Cambisol, Calcisol and Regosol types (FAO-UNESCO, 1988). These soils,  
44 under the particular climatic characteristic of water stress of the Mediterranean, are usually  
45 poor in organic matter and show conditions, which are not very favourable for an acceptable  
46 productivity or profitability.

47 To mitigate this problem, the use of sewage sludge from wastewater treatment plants as  
48 organic amendment has become usual in Europe during the last decade. Sewage treatment  
49 plants in Europe produce each year eight tons of sludge. Applying sludge to soil as organic  
50 amendment is one of most common ways of getting rid of it. In 2005 up to 50%, 54%, 65%  
51 and 71% in Germany, Spain, France, and United Kingdom, respectively, was handled this  
52 way (Margoarou, 2000).

53 The composition of the sludge is, however, barely known, and it may contain chemical  
54 substances which are potentially toxic, like heavy metals and persistent organic chemicals  
55 such as various surfactants (Generalitat Valenciana, 1995; Generalitat Valenciana, 1998). At  
56 the European level, this fact increasingly raises concern (Directive EEC/86/278). These  
57 pollutants may have an adverse effect on the soil microbial communities, threaten ground  
58 water aquifers, and also alter the natural solubility equilibrium for clays by ion exchange and  
59 complexation mechanisms (Lee et al., 2002; Abu-Zreig et al., 1999; Adeel and Luthy, 1995;  
60 Rosen, 1989).

61 Surfactants are classified in different groups: anionics, non-ionics, cationics and  
62 zwitterionics. Non-ionic surfactants, such as nonylphenol ethoxilates (NPEOs) and AEOs, are

63 applied as detergents, emulsifiers, humidifiers, stabilizers, skimmers and intermediates in the  
64 synthesis of a great variety of industries. Recent data reports about an annual production of  
65  $800 \cdot 10^6$  kg of AEOs in Western Europe, indicating that they are still the most widely used  
66 non-ionic surfactants (Castillo et al., 2000).

67 The European Commission (EU, 2000), elaborated a draft "Working Document on  
68 Sludge" that proposes maximum levels for some micro-contaminants among which AEOs  
69 and APEOs are present (the maximum value for their concentration in sludges for agricultural  
70 use is of  $50 \text{ mg kg}^{-1}$  of dry matter, and includes the nonylphenol and nonylphenol ethoxylate  
71 with one or two groups ethoxy).

72 There is a need of reliable tools for environmental monitoring as regards a series of  
73 pollutants introduced by anthropogenic impacts. The chemical complexity of APEOs and  
74 AEOs, which are mixtures of numerous isomers and oligomers, needs a quite sophisticated  
75 analytical methodology for its isolation, identification and quantification in soils that,  
76 generally, is based on their determination by gas chromatography-mass spectrometry (GC-  
77 MS) or liquid chromatography- mass spectrometry (LC-MS) (la Guardia et al., 2001; Bruno  
78 et al., 2002; Krogh et al., 2002; Krogh et al., 2003). The degradation products are  
79 alkylphenols, alkylphenol monoethoxylates and alkylphenol diethoxylates, which are  
80 recognized as endocrine disruptors (Magoarou, 2000).

81 The objective of this study is to develop and tune an analytical method, such that it is  
82 sensitive enough for the simultaneous determination of octylphenol, nonylphenol,  
83 alkylphenol ethoxylates and alcohol ethoxylates in soils treated with organic amendments.  
84 Pressurized liquid extraction (PLE), also known as accelerated solvent extraction (ASE), will  
85 be used to identify unequivocally the compounds by means of LC-MS. Concentrations of  
86 these types of compounds in samples from forest soils treated with organic amendments  
87 coming from different WWTPs of the Valencian Community, are also compared.

88

## 89 **2. Material and Methods**

### 90 *2.1. Reagents*

91 Octylphenol (OP), nonylphenol (NP), Nonidet 40 (NP<sub>6-15</sub>EOs), Triton X-100 (OP<sub>6-</sub>  
92 <sub>15</sub>EOs), nonylphenol monoethoxylate (NP<sub>1</sub>EO), the technical mixtures of nonylphenols (NP<sub>1-</sub>  
93 <sub>5</sub>EOs) and octylphenols (OP<sub>1-5</sub>EOs) with low ethoxylate grade, hexaethylene glycol  
94 monooctadecyl ether (C<sub>8</sub>EO<sub>6</sub>), hexaethylene glycol monodecyl ether (C<sub>10</sub>EO<sub>6</sub>), hexaethylene  
95 glycol monododecyl ether (C<sub>12</sub>EO<sub>6</sub>), hexaethylene glycol monotetradecyl ether (C<sub>14</sub>EO<sub>6</sub>) and  
96 hexaethylene glycol monohexadecyl ether (C<sub>16</sub>EO<sub>6</sub>) were provided by Aldrich (Madrid,  
97 Spain) and Symta (Madrid, Spain). Methanol and dichloromethane were HPLC grade and the  
98 deionised water was obtained with a MilliQ system. The solid phase used, C<sub>18</sub>, was acquired  
99 from Análisis Vínicos (Tomelloso, Spain). Ammonium acetate and anhydrous sodium  
100 sulphate were of analytical grade.

### 101 *2.2. Soils and sludges*

102 Six samples of the superficial horizon (A) of a Mediterranean forest soil (Rendzic  
103 Leptosol) (FAO, 1988), developed on Jurassic limestone, were taken from a hillside degraded  
104 by forest fire and further erosion. The previous vegetation of the area belonged to a sparse  
105 Pine (*Pinus halepensis*) forest with a dense shrubland stratum characterized by *Rosmarinus*  
106 *officinalis*, *Ulex parviflorus*, *Quercus coccifera*, *Rhamnus lycioides*, *Stipa tenacissima*,  
107 *Globularia alypum*, *Cistus clusii* and *Thymus vulgaris* (Gimeno-García et al., 2000).

108 The soil had a total carbonate content of 45.2%, pH of 7.1 and a sandy-loam texture.  
109 Once in the laboratory the soil was left to dry at room temperature for 24 hours, passed  
110 through a 2 mm sieve, homogenised and stored in polyethylene boxes sealed until analysis.  
111 Laboratory standard analytical methods were applied for the determination of the most  
112 important physical and chemical characteristics of these samples.

113 Each soil sample was added 10% sewage sludge from a different WWTP and after 24  
114 hours, analyzed. The selected WWTPs were located in distinct areas of the Valencian  
115 Community and used biological treatment. Samples 1-3 were added of sludges from WWTPs  
116 that receive mainly industrial effluents and samples 4-6 of sludges from WWTPs treating at  
117 equal domestic and industrial waste water (mainly from tanneries and textile industry).

118

### 119 *2.3. Extraction*

120 Five grams of soil were homogenized with anhydrous sodium sulphate, placed in a  
121 cylindrical cell (22 ml), and extracted using a Dionex Accelerated Solvent Extractor (Model  
122 2000, Salt Lake City CA, USA). Each extraction began with 3 min preheating time, followed  
123 by 5 min static extraction with acetone-hexane (50:50 v/v). Static extraction was performed at  
124 constant temperature and pressure (60 °C and 550 psi). Two extraction cycles were used. The  
125 40 ml extract was evaporated to dryness using rotary evaporation, re-dissolved in 100 ml of  
126 water and extracted with a glass column that contained 500 mg of previously activated C<sub>18</sub>  
127 with 10 ml of methanol and 10 ml of water. The retained surfactants were eluted with 10 ml  
128 of methanol. The extract was evaporated to 1 ml.

### 129 *2.4. Liquid chromatography- mass spectrometry*

130 LC-MS analyses were performed using an Agilent (Palo Alto, CA, USA) HP-1000 Series  
131 LC system consisting of an autosampler (volume injected was 25 µL) and a binary solvent  
132 pump, and a single quadrupole mass selective detector.

133 The chromatographic separation was carried out with a column Luna C<sub>18</sub> (150 x 4.6 mm),  
134 5 µm Phenomenex (Madrid, Spain), using a methanol-water gradient (both solvents contain 5  
135 mM ammonium acetate to obtain the ammonium adduct) with a flow of mobile phase of 1 ml  
136 min<sup>-1</sup>. The gradient begins with 70% methanol in water during 16 min and next the methanol  
137 percentage increases to 95% up to 30 min, and remains for 15 min more. Detection was

138 carried out using a mass spectrometer equipped with an APCI source. The conditions of the  
139 APCI were for the alkylphenol ethoxylates and the alcohol ethoxylates: source in positive  
140 ionization mode, capillary voltage of 4000 V, corona current 4  $\mu$ A, fragmentor voltage 140  
141 V, temperature of the APCI 400 °C and, temperature and flow of the drying gas, 350 °C and 3  
142 l min<sup>-1</sup> respectively. Nonylphenol and octylphenol, were detected in negative ionization mode  
143 and with corona current of 25  $\mu$ A. The remainder conditions were identical.

144

### 145 **3. Results and Discussion**

#### 146 *3.1. Extraction procedure*

147 For sample preparation, PLE followed SPE clean-up procedures have been developed and  
148 optimised. Pressurized liquid extraction has already been successfully used for recovering  
149 OPEAs, NPEOs and AEOs from soils and sludges (Petrovic et al. 2002; Krogh et al. 2003;  
150 Loyo-Rosales et al. 2003). In this study, different extraction solvents (methanol,  
151 dichloromethane and mixtures of hexane:acetone) have been tested. The recoveries obtained  
152 varied from 65-126% for methanol; from 70-170% for dichloromethane, and from 89-102%  
153 for the hexane/acetone 50:50 mixture. Therefore, the hexane:acetone 50:50 mixture was  
154 chosen as extraction solvent.

155 Regarding the extraction conditions, Petrovic et al. (2002) reported about problems with  
156 degradation at temperatures above 60 °C, for APEOs and their degradation products. In  
157 contrast, AEOs are fairly stable even at temperatures up to 150 °C. For simultaneous  
158 determination of the APEO and AEO, 60 °C was used as extraction temperature. The  
159 extraction pressure have no influence when the samples are dry (Petrovic et al. 2002). Thus  
160 the pressure, the static time and the number of extraction cycles was chosen according to  
161 Krogh et al. (2003).

#### 162 *3.2. LC-MS analysis*

163 As can be observed in Figure 1, OP, NP, OPEOs, NPEOs and AEOs were well resolved  
164 using the chromatographic conditions previously described. Retention times were repeatable  
165 and reproducible. When the studied compounds were analyzed simultaneously, OP and  
166 OPEOs eluted earlier than the respective NP and NPEOs. However, the individual  
167 homologues of OPEOs and NPEOs are not resolved co-eluting in the same peak. In addition,  
168 co-elution of the AEO C<sub>10</sub>EO<sub>6</sub> with the OPEOs was observed. In these cases, the compounds  
169 were distinguished based on their mass spectra. The various masses of the NPEOs are shown  
170 in the insert of Figure 1. The resulting mass spectra display a series of masses with 44 (-  
171 CH<sub>2</sub>CH<sub>2</sub>O-) mass units of difference between each compound making it possible to know the  
172 length of the ethoxylate chain. The same pattern was observed for OPEOs. On the contrary,  
173 AEOs showed mass spectra with 28 mass units of difference between each homologue.

174 Preliminary experiments were performed to differentiate between positive and negative  
175 ionization modes (PI or NI). The studied AEOs and APEOs gave response in PI but not in NI.  
176 In contrast, the OP and NP gave response only in NI mode providing the deprotonated  
177 molecule [M-H]<sup>-</sup> at  $m/z=$  205.1 and 219.1, respectively.

178 The optimal ionization depends as well on the LC mobile phase composition. Addition of  
179 acetic acid and ammonium acetate to enhance the signal of the studied compounds was  
180 tested. Acetic acid in the mobile phase provided the highest intensities of the [M+H]<sup>+</sup> ion  
181 while ammonium acetate increased the intensity of the [M+NH<sub>3</sub>]<sup>+</sup>. Ammonium adducts are a  
182 good choice for quantitative analysis of APEOs (Cohen et al., 2001) and addition of low  
183 NH<sub>4</sub><sup>+</sup> concentrations to the mobile phase served to stabilize the adduct ion signals in PI, and  
184 did not adversely affect in NI. Therefore, the different compounds were identified by the  
185 presence of the adduct with ammonium [M+NH<sub>3</sub>]<sup>+</sup> (AEOs and APEOs) or the deprotonated  
186 molecule [M-H]<sup>-</sup> (OP and NP).

187 For AEOs and APEOs detection, some studies utilise electrospray ionisation (ESI)  
188 (Ferguson et al., 2000; Cohen et al., 2001) while others use atmospheric pressure chemical  
189 ionization (APCI) (Krogh et al., 2002). In this study, both ESI and APCI, in negative and  
190 positive ion mode, have been investigated. The different instrumental settings, such as  
191 capillary and vaporiser temperature, corona current, sheath and auxiliary gas for APCI and  
192 spray voltage, sheath and auxiliary gas for ESI, were optimised; APCI and ESI in the positive  
193 ion mode gave similar signals. However, lower matrix and more stable analyte response were  
194 observed using APCI, which was the preferred ionisation mode.

### 195 *3.3. Analytical parameters*

196 The analytical parameters of the method are listed in Table 1. For the chromatographic  
197 procedures a linear relation is observed. This linearity was evaluated by analyzing standard  
198 solutions in triplicate at six different concentration levels, in the range from the LOQ to  
199 hundred times the LOQ (although the linearity interval is wider, it is difficult to find real  
200 samples with higher content of surfactants). The correlation coefficient was in all cases  
201  $>0.994$ .

202 The limits of detection (LODs) were calculated by using a signal-to-noise ratio (S/N) of  
203 3. The LODs were in the range of 0.3 to 1  $\mu\text{g kg}^{-1}$  for the OPEOs and NPEOs; 3  $\mu\text{g kg}^{-1}$  for  
204 the AEOs and 30  $\mu\text{g kg}^{-1}$  for the OP and NP. These values are in agreement with those  
205 reported in the literature (Cohen et al., 2001). The LOD for each individual compounds in  
206 wastewater were typically 1  $\mu\text{g l}^{-1}$ , whereas the detection limits in sludge samples were  
207 typically 100  $\mu\text{g kg}^{-1}$ , because it is a more complicated matrix. The limits of quantification  
208 (LOQs) were estimated by using a S/N of 10. The analytical method was validated at LOQs  
209 level and 10 times LOQ level (data not shown).

210 In validating the analytical method, concerning recovery and precision three sets of three  
211 replicated samples were extracted at three different days and subsequently analysed at



212 different days. The results are shown in Table 1. Satisfactory recoveries were achieved for all  
213 target analytes ranging from 89 to 102% with RSDs  $\leq$  20%.

214 Figure 1, shows the chromatograms obtained injecting extracts of a soil that does not  
215 contain surfactants, and the same soil spiked at 0.1 mg kg<sup>-1</sup> with the different surfactants. Soil  
216 extracts are clean without interfering substances. In the same Figure, the mass spectrum  
217 corresponding to the studied nonylphenol ethoxylates (n=1-15) is inserted. The current  
218 method provides the highest available sensitivity for comprehensive AEOs and APEOs  
219 analysis in environmental samples. This sensitivity was more than sufficient for the  
220 determination of AEOs and APEOs in sewage amended soils.

#### 221 *3.4. Analysed surfactant concentrations in soil samples treated with sludge*

222 The method was applied to the determination of NP, OP, AEOs and APEOs in soil treated  
223 with sludges from six WWTPs of the Valencian Community. Results presented in Table 2  
224 demonstrate that these compounds are incorporated in the treated soils. In all of them, the  
225 presence of OP, NP and their ethoxylates was detected in significant amounts. The presence  
226 of AEOs is scarce, although it is also detected in two samples. The homologues found in the  
227 samples were all included in the present study. The most abundant compound present in soil  
228 was NP, which agrees with previous findings in previous studies (Loyo-Rosales et al., 2003).  
229 The octylphenol ethoxylate and nonyl ethoxylate homologues found in the samples have been  
230 mainly mono-, di, tri-, tetra-, and pentaethoxylates. However, in sample S3 the entire pattern  
231 of homologues selected for the method validation was found. These results are in agreement  
232 with the proposed degradation pathway of these surfactants, which is through the breakdown  
233 of the ethoxylate groups (Petrovic et al., 2002; Loyo-Rosales et al., 2003). AEOs and APEOs  
234 concentrations detected in the soil treated with the sludges are always lower than 1 mg kg<sup>-1</sup>,  
235 which evidence their low persistence in the environment.

236

237 **4. Conclusions**

238 The method presented in this work is capable of analysing a wide range of  
239 octylphenol polyethoxylates, nonylphenol polyethoxylates and alcohol polyethoxylates in  
240 samples of soils treated with sludges from WWTPs. The determination of neutral surfactants  
241 by PLE, SPE and LC-MS requires a small amount of sample, provides satisfactory  
242 recoveries, repeatability and reproducibility, and is a sensitive, selective and reliable  
243 analytical method.

244 The application of the reported method to soil treated with organic amendments  
245 demonstrates that the presence of these compounds may present a risk in agricultural and  
246 restoration zones. Nonylphenol, octylphenol, and their mono-, di-, tri-, tetra-, and  
247 pentaethoxylates were detected in significant amounts. Although less frequent, highly  
248 ethoxylate NPs and OPs as well as AEOs were also found in some samples. Further studies  
249 are needed to establish to which extent these surfactant residues pose a risk to soil functions  
250 and its macro and micro-fauna.

251  
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310

311 Table 1. Mean Recoveries (R, %), relative standard deviation (RSDs,%), linear concentration  
 312 range, correlation coefficient (*r*), and limit of detection (LODs) of the different alcohol  
 313 ethoxylates (AEOs) obtained for spiked soil samples (n=3).

AEOs	Recovery and precision			Linearity		LODs ( $\mu\text{g kg}^{-1}$ )
	Concentration ( $\text{mg kg}^{-1}$ )	R (%)	RSDs (%)	Range ( $\text{mg kg}^{-1}$ )	<i>r</i>	
C <sub>8</sub> EO <sub>6</sub>	0.010	90	16	0.01-1	0.997	3.0
C <sub>10</sub> EO <sub>6</sub>	0.010	94	14	0.01-7.4	0.999	3.0
C <sub>12</sub> EO <sub>6</sub>	0.010	91	14	0.01-1.0	0.999	3.0
C <sub>14</sub> EO <sub>6</sub>	0.010	96	12	0.01-1.0	0.998	3.0
C <sub>16</sub> EO <sub>6</sub>	0.010	96	12	0.01-1.0	0.998	3.0
NP	0.100	97	19	0.1-10	0.997	30.0
NP <sub>1</sub> EO	0.010	89	13	0.01-1.0	0.998	3.0
NP <sub>2</sub> EO	0.005	90	15	0.005-0.5	0.997	1.0
NP <sub>3</sub> EO	0.005	94	12	0.005-0.5	0.999	1.0
NP <sub>4</sub> EO	0.005	91	17	0.005-0.5	0.999	1.0
NP <sub>5</sub> EO	0.005	89	13	0.005-0.5	0.994	1.0
NP <sub>6</sub> EO	0.005	92	13	0.005-0.5	0.997	1.0
NP <sub>7</sub> EO	0.001	89	15	0.001-0.1	0.998	0.3
NP <sub>8</sub> EO	0.001	90	16	0.001-0.1	0.997	0.3
NP <sub>9</sub> EO	0.001	94	15	0.001-0.1	0.999	0.3
NP <sub>10</sub> EO	0.001	91	14	0.001-0.1	0.999	0.3
NP <sub>11</sub> EO	0.001	95	10	0.001-0.1	0.995	0.3
NP <sub>12</sub> EO	0.001	94	9	0.001-0.1	0.998	0.3
NP <sub>13</sub> EO	0.001	89	19	0.001-0.1	0.998	0.3
NP <sub>14</sub> EO	0.001	90	16	0.001-0.1	0.997	0.3
NP <sub>15</sub> EO	0.001	94	17	0.001-0.1	0.999	0.3
OP	0.100	91	15	0.1-10	0.999	30.0
OP <sub>1</sub> EO	0.010	89	18	0.01-1.0	0.998	3.0
OP <sub>2</sub> EO	0.005	90	17	0.005-0.5	0.997	1.0
OP <sub>3</sub> EO	0.005	94	12	0.005-0.5	0.999	1.0
OP <sub>4</sub> EO	0.005	91	12	0.005-0.5	0.999	1.0
OP <sub>5</sub> EO	0.005	97	11	0.005-0.5	0.994	1.0
OP <sub>6</sub> EO	0.005	99	12	0.005-0.5	0.996	1.0
OP <sub>7</sub> EO	0.001	89	19	0.001-0.1	0.998	0.3
OP <sub>8</sub> EO	0.001	90	16	0.001-0.1	0.997	0.3
OP <sub>9</sub> EO	0.001	94	14	0.001-0.1	0.999	0.3
OP <sub>10</sub> EO	0.001	91	14	0.001-0.1	0.999	0.3
OP <sub>11</sub> EO	0.001	102	11	0.001-0.1	0.997	0.3
OP <sub>12</sub> EO	0.001	99	10	0.001-0.1	0.996	0.3
OP <sub>13</sub> EO	0.001	89	19	0.001-0.1	0.998	0.3
OP <sub>14</sub> EO	0.001	90	16	0.001-0.1	0.997	0.3
OP <sub>15</sub> EO	0.001	94	14	0.001-0.1	0.999	0.3

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316 Table 2. Concentration ( $\mu\text{g kg}^{-1}$ ) of the studied compounds in soil amended with the different  
317 sewage sludges.

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Soil	OP	NP	OPEOs	NPEOs	AEOs
S1	238	500	369	200	n.d.
S2	215	200	232	149	152
S3	105	300	125	329	n.d.
S4	125	150	124	134	37
S5	135	142	232	210	n.d.
S6	210	225	87	92	n.d.

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320 OP: Octyl phenol. NP: Nonyl phenol. OPEOs: Octylphenol ethoxylates. NPEOs:  
321 Nonylphenol ethoxylates. AEOs: Alcohol ethoxylates. n.d.: not detected.

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Legend of Figure:

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326 Fig.1. Chromatograms of (A) soil in which surfactants were not detected and (B) the same  
327 soil fortified with a mixture of the different NPEOs, OPEOs and AEOs at 0.1 mg kg<sup>-1</sup>. The  
328 mass spectrum corresponding to the NPEOs is shown as an insert.

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