

# Phenylacetyl-Coenzyme A Is the True Inducer of the Phenylacetic Acid Catabolism Pathway in *Pseudomonas putida* U

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**Aerobic degradation of phenylacetic acid in *Pseudomonas putida* U is carried out by a central catabolism pathway (phenylacetyl-coenzyme A [CoA] catabolon core). Induction of this route was analyzed by using different mutants specifically designed for this objective. Our results revealed that the true inducer molecule is phenylacetyl-CoA and not other structurally or catabolically related aromatic compounds.**

Phenylacetyl-coenzyme A (CoA) catabolon (PhAc-CoACC) is a term used for a complex catabolic unit integrated by different degradative pathways involved in the assimilation of certain aromatic compounds (phenylacetic acid [PhAc], phenylethylamine, ethylbenzene, styrene, tropic acid, *trans*-styrylacetic acid, and *n*-phenylalkanoic acids [*n*-PhAs] containing an odd number of carbon atoms) which converge in a central route, namely, the PhAc-CoACC core (PhAc-CoACC) (2, 13), which converts these molecules into general metabolites (Fig. 1).

The genetic information required for the PhAc-CoACC in *Pseudomonas putida* U is contained in a piece of DNA (18 kb) organized in three consecutive operons (Fig. 2). We have previously reported that this route appeared when *P. putida* U was cultured in chemically defined media containing some of the aromatic compounds indicated above as carbon sources (4, 10, 11) and that other similar, structurally related molecules (2-OH-PhAc, 3-OH-PhAc, 4-OH-PhAc, benzoic acid, and *n*-PhAs with an even number of carbon atoms) did not have this effect (13, 14). Bearing in mind the diverse chemical structure of the aromatic compounds able to induce the pathway, it seems feasible to assume that a common intermediate (phenylacetyl-CoA) is the true inducer of the convergence pathway. Despite this, the possibility that other molecules generated from the aromatic compounds (PhAc or *n*-phenylalkanoyl-CoA or their  $\beta$ -oxidation derivatives) might also be involved in this induction cannot be ruled out.

In order to clarify this point, we designed, obtained, and characterized by different methods (6, 13–17, 19) several mutants of *P. putida* U affected in the catabolism of certain aromatic compounds (see below). Using these strains, we analyzed induction of the PhAc-CoACC by measuring two target enzymatic activities: the activity of phenylacetyl-CoA ligase (PCL) (encoded by the *paaE* gene, belonging to the first operon) (10, 11) and the activity of a permease (encoded by the *paaJ* gene [the third operon in Fig. 2]) involved in the uptake of [<sup>1-14</sup>C]PhAc (18).

*P. putida* U was maintained and cultured as previously reported (10). The medium used for growth of *P. putida* and different mutants was a chemically defined medium (10) containing PhAc, 4-OH-PhAc, 6-phenylhexanoic acid (PhH) (or 8-phenyloctanoic acid [PhO]), other aromatic compounds, or

combinations of these compounds as carbon sources (Table 1). In the experiments in which mutants were employed, 4-OH-PhAc and the carbon source used to test induction were added to the media together. The final concentration of each aromatic compound was 5 mM.

Since the PhAc-CoACC is required for degradation of PhAc, ethylbenzene, styrene, tropic acid, phenylethylamine, *trans*-styrylacetic acid, and *n*-PhAs with an odd number of carbon atoms, it is possible that any of these compounds could induce the PhAc-CoACC. However, although a priori this possibility cannot be ruled out, it seems more reasonable to assume that a single compound is the true inducer. If this is indeed the case, two different molecules, PhAc or PhAc-CoA, could be involved in this process.

PhAc is a catabolite generated from styrene (20), ethylbenzene (8) tropic acid (9), and phenylethylamine (3, 5), whereas PhAc-CoA is obtained from *trans*-styrylacetic acid, from *n*-PhAs containing an odd number of carbon atoms, from  $\beta$ -oxidation derivatives of these acids, or from polymer derivatives of these compounds (4). However, we have previously shown that *P. putida* U mutants that are defective in assimilation of all these compounds due to the existence of a mutation in some of the genes belonging to the second operon (the ring oxidation system) (Fig. 2) accumulates PhAc extracellularly when the bacteria are cultured in a chemically defined medium (4) containing molecules which could be catabolically converted into PhAc-CoA (13). These results suggested that this thioester is hydrolyzed to PhAc (probably by nonspecific thioesterases) before being released into the culture broth. It is thus difficult to establish which of these compounds (PhAc or PhAc-CoA) is the true inducer of the PhAc catabolic pathway, since the two molecules are interconvertible. In order to establish the nature of the molecule that acts as the true inducer of the PhAc-CoACC, three different types of mutants were obtained.

The first group (type I [Fig. 2]) includes mutants lacking PCL activity. These strains were obtained by disrupting the *paaE* gene, which encodes a protein having this enzymatic activity, with plasmid pK18:*mob* (17). These mutants are unable to grow in chemically defined media containing PhAc, ethylbenzene, styrene, or phenylethylamine as the sole carbon source, since all of them generate PhAc as a catabolic intermediate, which has to be activated to PhAc-CoA for further degradation (10, 11, 13). However, they grow well in a similar medium containing as carbon sources compounds that produce (through  $\beta$ -oxidation) PhAc-CoA as an intermediate (*trans*-styrylacetic acid or *n*-PhAs with an odd number of carbon

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TABLE 1. Evaluation of PCL activity, the [ $1\text{-}^{14}\text{C}$ ]PhAc transport system in *P. putida* U (wild type) and in different mutants (types I, II, and III), and PHPhA accumulation when the bacteria were cultured in a chemically defined medium

Strain	Carbon source(s) (5 mM)	PCL sp act (U/mg of protein) <sup>a</sup>	[ $1\text{-}^{14}\text{C}$ ]PhAc uptake (pmol/min) <sup>b</sup>	PHPhA content (%) <sup>c</sup>	Significance
<i>P. putida</i> U (wild type)	4-OH-PhAc	ND <sup>d</sup>	ND	ND	4-OH-PhAc does not induce PhAc-CoACC, 4-OH-PhAc is not polymerized
Wild type	4-OH-PhAc + PhAc	43	2,420	ND	PhAc-CoACC induction occurs, PhAc is not polymerized
Type I	4-OH-PhAc + PhAc	ND	ND	ND	PhAc-CoACC induction does not occur, PhAc is not activated to PhAc-CoA by other ligases, PhAc is not an inducer of PhAc-CoACC
Wild type	4-OH-PhAc + PhH (or PhO)	38	2,725	18–20	PhAc-CoACC induction occurs, PHPhAs accumulate
Type II	4-OH-PhAc + PhH (or PhO)	ND	ND	ND	PhAc-CoACC induction does not occur, PHPhAs do not accumulate, <i>n</i> -PhAs are not inducers of PhAc-CoACC
Type II	4-OH-PhAc + PhAc	37	2,740	ND	PhAc-CoACC induction occurs, PHPhAs not produced
Type III	4-OH-PhAc + PhH (or PhO)	ND	ND	50	PhAc-CoACC induction does not occur, <i>n</i> -PhAs and $\beta$ -oxidation derivatives are not inducers, strong PHPhA accumulation is observed
Type III	4-OH-PhAc + PhAc	30	2,300	ND	PhAc-CoACC induction occurs, PHPhAs are not produced

<sup>a</sup> See reference 10.

<sup>b</sup> [ $1\text{-}^{14}\text{C}$ ]PhAc uptake was assayed as reported in reference 18.

<sup>c</sup> PHPhA content is expressed as a percentage of the bacterial dry weight.

<sup>d</sup> ND, not detected.

to synthesize phenylacetyl-CoA from the phenylalkanoyl-CoA  $\beta$ -oxidation derivatives but were able to transport and activate these compounds since they, in contrast to the wild type, accumulate large amounts of poly(3-hydroxyphenylalkanoates) (PHPhAs) when they are cultured in a chemically defined medium containing 4-OH-PhAc and PhH or PhO as carbon sources (Table 1). The presence of PHPhAs as reserve material inside the bacteria was determined by direct microscopic observation (4). The amount of polymer that accumulated was quantified as previously reported (4, 7), and the PHPhA content was recorded as a percentage of the bacterial dry weight. The structures of the polymers synthesized were analyzed by nuclear magnetic resonance as indicated by García et al. (4). For this type of experiment, the wild type (*P. putida* U) or the different mutants were cultured in the same chemically defined media containing PhAc, 4-OH-PhAc, PhH, PhO, or combinations of these compounds as carbon sources (Table 1). In all cases the final concentration of each aromatic molecule was 5 mM. The data reported above indicate that all the enzymes required for transport, activation, and  $\beta$ -oxidation (except the 3-ketoacyl-CoA thiolase) (Fig. 3) are functional in these mutants. These mutants were used to analyze whether different

*n*-phenylalkanoyl-CoAs (or their  $\beta$ -oxidation derivatives) might be directly involved in induction of the PhAc-CoACC.

In all three types of mutants, PCL and PhAc permease activities (see above) were assayed. In these experiments, all the strains analyzed were cultured in the same medium containing the inducer molecule to be analyzed and a different carbon source (4-OH-PhAc, which does not induce the PhAc-CoACC [Table 1]) (14) to support bacterial growth.

The absence of [ $1\text{-}^{14}\text{C}$ ]PhAc uptake in the mutants belonging to the first group (those lacking a functional PCL) revealed that PhAc does not induce the PhAc-CoACC (Table 1). It is surprising that PhAc was unable to induce the PhAc-CoACC since PCL, the product of the *paaE* gene, is required for activation of PhAc to PhAc-CoA and for further catabolism of PhAc-CoA. It should therefore be assumed that a certain quantity of PCL or other acyl-CoA activating enzymes that nonspecifically activate PhAc to PhAc-CoA must be present in *P. putida* U and that the trace amounts of PhAc-CoA are sufficient to open the whole pathway. However, the results obtained with type I mutants (Table 1) indicated that no uptake of [ $1\text{-}^{14}\text{C}$ ]PhAc occurred, suggesting that PCL alone and not nonspecific acyl-CoA synthetases are involved in activation

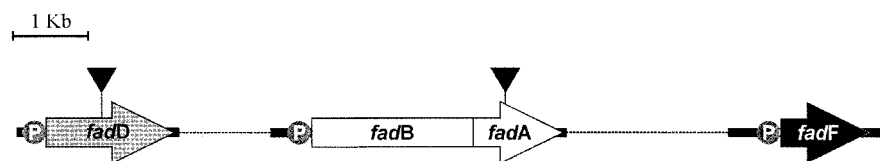


FIG. 3. Genetic organization of the  $\beta$ -oxidation genes in *P. putida* U. *fadD*, gene encoding an acyl-CoA synthetase; *fadF*, gene encoding an acyl-CoA dehydrogenase; *fadB*, two fused genes encoding several enzymatic activities including enoyl-CoA hydratase, 3-OH-acyl-CoA dehydrogenase, *cis*- $\Delta^3$ -*trans*- $\Delta^2$ -enoyl-CoA isomerase, and 3-OH-acyl-CoA epimerase; and *fadA*, gene encoding the 3-ketoacyl-CoA thiolase. Dashed lines indicate that some of the genes are not contiguous in the bacterial chromosome. The solid triangle indicates the site of insertion of the disruption element used to truncate the desired gene.

of PhAc. These results also reinforce our previous observations about the absence of PhAc passive diffusion in this bacterium (18) and show that this aromatic compound does not induce (at least in the absence of a functional PCL activity) other permeases which more or less nonspecifically could be involved in the uptake of PhAc.

Analysis of the results obtained with the mutants belonging to the second group (type II) (those lacking acyl-CoA synthetase activity which activates *n*-PhAs to their CoA thioesters) indicated that neither 6-phenylalkanoic acid nor 8-phenylalkanoic acid, two compounds that generated PhAc-CoA by  $\beta$ -oxidation, act as an inducer of PhAc-CoACC since neither PCL activity nor [ $1\text{-}^{14}\text{C}$ ]PhAc uptake was observed (Table 1). Furthermore, these mutants were unable to accumulate PHPhAs, suggesting that acyl-CoA synthetase is the only enzyme that activates *n*-PhAs to their CoA derivatives. The lack of *n*-phenylalkanoyl-CoAs implies that the monomers of PHPhAs (3-OH-*n*-phenylalkanoyl-CoAs) cannot be synthesized.

However, when these mutants were cultured in chemically defined medium containing 4-OH-PhAc and PhAc as carbon sources, both PCL activity and [ $1\text{-}^{14}\text{C}$ ]PhAc uptake were observed (Table 1), supporting the hypothesis that PhAc-CoA is the true inducer. The fact that the mutants in the third group (those lacking 3-ketoacyl-CoA thiolase activity) were also unable to induce either PCL activity or [ $1\text{-}^{14}\text{C}$ ]PhAc uptake further supports this hypothesis. Moreover, type III mutants accumulated PHPhAs (Table 1), suggesting that they are able to transport, activate to CoA thioesters, and synthesize the  $\beta$ -oxidation intermediates required for polymerization of these compounds (3-hydroxy-phenylalkanoyl-CoA). These data show that neither phenylalkanoyl-CoA nor its  $\beta$ -oxidative intermediates are able to induce the central pathway involved in degradation of PhAc.

The data in Table 1 indicate that 4-OH-PhAc does not induce the PhAc-CoACC and that, as previously reported by us (4), PhAc or 4-OH-PhAc cannot be polymerized by the enzymatic system involved in synthesis of PHPhAs because of the impossibility of production of the 3-OH-*n*-PhA monomers required for polymerization.

All the above observations allow us to conclude that the true inducer of the PhAc-CoACC is indeed phenylacetyl-CoA, the first common intermediate of all the pathways involved in degradation of ethylbenzene, styrene, phenylethylamine, tropic acid, *trans*-styrylacetic acid, and *n*-PhAs containing an odd number of carbon atoms. Recently, Ferrández et al. (1), using a different approach, have shown that binding of the repressor which controls the catabolic pathway involved in aerobic degradation of PhAc in *Escherichia coli* W to the DNA promoter region is avoided when PhAc-CoA is present, suggesting (as also shown here) that PhAc-CoA, an intermediate which until recently was considered a catabolic intermediate produced under anaerobic conditions (12), is the true inducer of the PhAc catabolic pathway.

It therefore seems reasonable to propose that each of the independent aromatic compounds indicated above induces its own partial degradative route and that once PhAc-CoA has been synthesized, this molecule opens the common catabolic pathway (the PhAc-CoACC).

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