Molecular Carbon Chains and Rings in TMC-1

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ABSTRACT

We present mapping results in several rotational transitions of HC₃N, C₆H, both cyclic and linear C₃H₂ and C₃H, towards the cyanopolyyne peak of the filamentary dense cloud TMC-1 using the IRAM 30m and MPIfR 100m telescopes. The spatial distribution of the cumulene carbon chain propadienylidene H₂CCC (hereafter l-C₃H₂) is found to deviate significantly from the distributions of the cyclic isomer c-C₃H₂, HC₃N, and C₆H which in turn look very similar. The cyclic over linear abundance ratio of C₃H₂ increases by a factor of 3 across the filament, with a value of 28 at the cyanopolyyne peak. This abundance ratio is an order of magnitude larger than the range (3 to 5) we observed in the diffuse interstellar medium. The cyclic over linear abundance ratio of C₃H also varies by ~ 2.5 in TMC-1, reaching a maximum value (13) close to the cyanopolyyne peak. These behaviors might be related to competitive processes between ion-neutral and neutral-neutral reactions for cyclic and linear species.

Subject headings: ISM: abundances — ISM: individual (TMC-1) — ISM: molecules — molecular processes

1. Introduction

Among the molecules discovered in the interstellar medium, C_3H_2 and C_3H are of peculiar interest for astrochemistry since both are observed in two isomeric forms: cyclic and linear. The ring molecule cyclopropenylidene (hereafter c- C_3H_2) was discovered in 1985 (Thaddeus, Vrtilek, & Gottlieb) and has immediatly retained attention due to its ubiquity in the galaxy (e.g., Matthews and Irvine 1985). One of its linear counterparts, the propadienylidene (l- C_3H_2), was discovered in TMC-1 by Cernicharo et al. (1991). The two isomeric shapes of C_3H , cyclopropynylidyne (c- C_3H) and propynylidyne (l- C_3H), were also detected in TMC-1 (Yamamoto et al. 1987; Thaddeus et al. 1985).

This variety of isomers for the same species raises the question of their formation. In particular, are rings and chains formed from the same progenitors and involved in the same reactions networks? Furthermore these isomers can be used to probe interstellar chemistry models which include heavy interstellar molecules such as cumulene carbenes (Bettens and Herbst 1996, 1997; see also Millar, Leung, and Herbst 1987). Previous observations by Cernicharo et al. (1999) have shown that the cyclic over linear abundance ratio of C_3H_2 (hereafter R_2) in the diffuse medium along the line of sight toward the continuum sources W51E1/E2, W51D and W49 is one order of magnitude smaller than its value in TMC-1. In a recent study, Turner, Herbst, & Terzieva (2000) have compared measurements in three translucent clouds and two dense clouds (TMC-1 and L183) of the C_3H cyclic over linear abundance ratio (R_1) and R_2 ratio - and column densities of 6 other hydrocarbons - to a modified version of the New Standard Model of chemistry (Lee et al. 1996). They found small variation of R_1 from source to source, and a systematic higher R_2 ratio with largest the values found in the two dark clouds. Also, it has been suggested that cyclic and linear isomers of a same compound could have different behavior in neutral-neutral and ion-neutral reactions (see, for example, Adams & Smith 1987; Maluendes, McLean, & Herbst 1993; Kaiser et al. 1997, 1999). The cyclic over linear abundance ratio of molecules like C_3H_2 and C_3H could then be used as a tool to investigate physical conditions in several media, from cold dark cloud to warmer and lower density media.

In order to investigate these questions, we mapped in detail the region around the cyanopolyyne peak (hereafter CP) in TMC-1 in both l-C₃H₂ and c-C₃H₂, and observed at selected positions the two isomeric forms of C₃H. C₆H and HC₃N have also been observed for comparison purpose.

2. Observations

The C_3H_2 and C_6H observations were made in 1990 and 1992 with the 100-meter telescope of the Max-Planck Institut für Radioastronomie at Effelsberg (Germany). We observed the $1_{10} - 1_{01}$ transitions of c- C_3H_2 at 18343.145 MHz, and simultaneously those of $C_6H^2\Pi_{3/2}$ J = 15/2 - 13/2 near 20794 MHz and l- C_3H_2 $1_{01} - 0_{00}$ at 20792.590 MHz. The antenna beamsize at 18.3 and 20.7 GHz are of 54" and 48", respectively. We used the 1024 channel autocorrelator to achieve a resolution of 0.05 km s⁻¹ at 18.3 GHz and 0.09 km s⁻¹ at 20.7 GHz, and the data were taken by position switching the telescope.

The HC₃N and C₃H observations were done in 1995 and 1999 with the IRAM 30-meter telescope at Pico Veleta (Spain). We observed the l-C₃H ${}^{2}\Pi_{3/2}$ J = 9/2 - 7/2 transition near 98 GHz and at the same time the transition of c-C₃H 2₁₂ - 1₁₁ near 91.5 GHz. The observations were made in frequency-switching mode. The autocorrelator was used as a spectral instrument with a velocity resolutions of 0.12 kms⁻¹. The half-power beamwidth (HPBW) and main-beam efficiency are 26" and 0.75 for l-C₃H and 27" and 0.78 for c-C₃H. System temperatures were in the range 90-130K. For HC₃N, observed at 90979.023 MHz (J = 10 - 9), the HPBW is 26". Pointing and calibration were monitored by regularly observing planets and quasars for both telescopes. Sample spectra are shown in Fig.1 and line parameters for selected positions are given in Table 1.

3. Results

3.1. C_3H_2 isomers and C_6H

The mapping results of this study are shown in Fig. 2. The distributions towards the CP in TMC-1 of c-C₃H₂ and l-C₃H₂ are compared to that of C₆H in Fig. 2a and b, respectively. Fig. 2c shows the distributions of C₆H and HC₃N. The spatial distributions of l-C₃H₂ and c-C₃H₂ (Fig. 2d) are clearly different, whereas the emissions of c-C₃H₂, C₆H and HC₃N have comparable distributions. The emission of l-C₃H₂ is shifted toward the west by approximatively 40" with respect to c-C₃H₂ or C₆H - note that, since C₆H and l-C₃H₂ were observed in the same bandwidth, this shift cannot be due to pointing errors during the observations.

In order to derive relative abundances, we first computed relations between the column-density and the observed line intensity valid for uniform physical conditions, i.e. $n(H_2) = 3 \times 10^4 \text{ cm}^{-3}$ and $T_K = 10 \text{K}$ in TMC-1 (Cernicharo & Guélin 1987). From a statistical equilibrium calculation using collisional excitation rates for c-C₃H₂ from Avery

& Green (1989), and assuming an ortho/para ratio of 3, we find: $N(c \cdot C_3 H_2)[cm^{-2}] = 2.2 \times 10^{13} \int T_{mb} dv [\text{K} \cdot \text{km} \cdot \text{s}^{-1}]$. This linear relationship indicates that the $1_{10} - 1_{01}$ transition of $c \cdot C_3 H_2$ is optically thin in TMC-1; it is in good agreement with the work of Cox et al. (1989). Madden et al. (1986), from observations of the isotopic $c \cdot C_2 ^{13} \text{CH}_2$ in TMC-1, derive an optical depth ranging from 5.6 to 6.8 for the $2_{12} - 1_{01}$ transition of $c \cdot C_3 H_2$ at 85 GHz. This result is in agreement with ours since the $2_{12} - 1_{01}$ line has a larger opacity than the $1_{10} - 1_{01}$ transition for dark cloud physical conditions. Indeed, with our code, we are able to reproduce the results of Madden et al. at 85 GHz, the $1_{10} - 1_{01}$ transition still being thin or marginally saturated ($\tau \leq 1.5$). For $l \cdot C_3 H_2$ and $C_6 H$ there are no collisional cross sections available. We have estimated cross sections from those of $H_2 \text{CO}$ (calculated by Green et al. 1978) and of $HC_3 N$ (Green & Chapman 1978) - see Cernicharo et al. (1999). This gives: $N(l \cdot C_3 H_2)[cm^{-2}] = 2.2 \times 10^{13} \int T_{mb} dv [\text{K} \cdot \text{km} \cdot \text{s}^{-1}]$ and $N(C_6 \text{H})[cm^{-2}] = 5.2 \times 10^{13} \int T_{mb} dv [\text{K} \cdot \text{km} \cdot \text{s}^{-1}]$. Note that for $C_6 \text{H}$, $\int T_{mb} dv$ is the total integrated intensity summed over the four components of the hyperfine structure. These relationships are in good agreement with previous works (Cernicharo et al. 1991; Bell et al. 1999).

Comparing two positions separated by 40" (which corresponds to 0.02 pc for the adopted distance of 100 pc to TMC-1 - Cernicharo & Guélin 1987), we derive the following column density: at the CP (0,0), $N(C_6H) = 8.3 \times 10^{12} \text{ cm}^{-2}$, $N(c\text{-}C_3H_2) = 5.8 \times 10^{13} \text{ cm}^{-2}$, $N(l\text{-}C_3H_2) = 2.1 \times 10^{12} \text{ cm}^{-2}$ (R₂=28±6); and at the edge of the TMC-1 filament (-40,0), $N(C_6H) = 4.7 \times 10^{12} \text{ cm}^{-2}$, $N(c\text{-}C_3H_2) = 2.8 \times 10^{13} \text{ cm}^{-2}$, $N(l\text{-}C_3H_2) = 3.2 \times 10^{12} \text{ cm}^{-2}$ (R₂=10±3). Between these two positions, the cyclic over linear abundance ratio changes by a factor of ~ 3. This variation cannot be an artefact caused by calibration errors. Indeed, while the calculated C₆H over $l\text{-}C_3H_2$ column density ratio varies from 1.5 to 4, the $c\text{-}C_3H_2$ over C₆H column density ratio remains constant ($\simeq 6.5$). Since the former ratio is unaffected by calibration and pointing errors (lines are observed in the same bandwidth), the observed variation of R₂ must be real. This result suggests that chemical gradients are present in TMC-1 on scales smaller than 0.02 pc.

To investigate further the spatial variations of molecular abundances across the TMC-1 filament, we averaged the spectra along the six different rows shown in Fig.2. Results are summarized in Fig.3: R_2 ranges from 12 to 37 and clearly increases from the west to the east of TMC-1. Note that a constant density of $n(H_2) = 3 \times 10^4$ cm⁻³ has been adopted. If we use instead H₂ densities derived by Pratap et al. (1997) from an analysis of the HC₃N transitions while keeping the same kinetic temperature (10 K), the values of R₂ are then lowered by 20 to 40 % but the west to east variation of R₂ remains. The rise of the cyclic over linear abundance ratio of the C₃H₂ isomers at a spatial scale of 2 arcmin (0.06 pc) is therefore a firm result.

3.2. C_3H isomers

As C_3H is thought to be formed by the same reaction as C_3H_2 , i.e. by the dissociative recombination of $C_3H_3^+$ (Adams & Smith 1987), it is interesting to compare $N(c-C_3H)/N(l-C_3H)$ (hereafter R_1) with R_2 . What is the value of R_1 and does it vary across the filament similarly to R_2 ? In order to answer these questions we observed $c-C_3H$ and $l-C_3H$ at 13 positions along two strips close to the CP. One strip crosses the filament from offset positions (60",60") to (-60",-60"), the other from (10,-70") to (-70",10"). We also observed the $c-C_3H_2$ and the $l-C_3H_2$ peaks. Where possible, we analyzed the data with the HFS method of CLASS (a software developed by the GILDAS working group). This method provides the total optical depth, the average linewidth and the brightness temperature of a line with hyperfine structure. A reliable estimate of the excitation temperature could be obtained for several points. We found that $3K \leq T_{ex} \leq 3.8K$ for $c-C_3H$ (in excellent agreement with Mangum & Wootten 1990), and $4.9K \leq T_{ex} \leq 6.7K$ for $l-C_3H$. In the following, we adopt $T_{ex} = 3.5K$ for $c-C_3H$ and $T_{ex} = 5.5K$ for the linear isomer.

Calculations of the column densities were done using the classical formula with $\mu = 2.4$ D (Yamamoto et al. 1987) for *c*-C₃H and $\mu = 3.1$ D (Green 1980) for *l*-C₃H. For each observed point, we computed the average of the total column densities by using each hyperfine transition weighted by the inverse of the square of the error. The resulting column densities and the values for R₁ are given in Table 2. Comparing with the values of R₂ in Fig.3, we see that: 1) With a value of 12 at the CP, R₁ is smaller than R₂; 2) As R₂ does, R₁ shows variations across the filament (by a factor of 2.5), mainly due to variations of $N(c-C_3H)$.

4. Discussion

The observations described above underscore the differences in the distribution and relative abundances of the cyclic and linear forms of C_3H_2 and C_3H in the molecular filament TMC-1. In the following, we study the origin of this behavior and suggest that it is driven by competitive processes between ion-neutral and neutral-neutral reactions.

4.1. Steady-state calculation

We assume here, as it has been proposed (see, for example, Adams & Smith 1987; Maluendes, McLean & Herbst 1993), that cyclic C_3H_2 and C_3H are both formed in the dissociative recombination of c- $C_3H_3^+$, while linear C_3H_2 and C_3H result from $H_2C_3H^+$. We also assume that the $C_3H^+ + H_2$ association reaction is the dominant formation mechanism of $C_3H_3^+$, which is produced in equal amounts in cyclic and linear forms.

At steady state, the chemical kinetic equation for the abundance of c-C₃H₂ - that is, x(c-C₃H₂) - is:

$$dx(c-C_{3}H_{2})/dt = {}^{2}b_{c}k_{c}^{r}x(c-C_{3}H_{3}^{+})x_{e} - {}^{2}K_{c}^{d}x(c-C_{3}H_{2}) = 0$$

where "b" is for "branching ratio", \mathbf{k}^r is for "recombination rate" $[\mathrm{cm}^3 \cdot \mathrm{s}^{-1}]$, \mathbf{K}^d is for "mean destruction rate" $[\mathrm{s}^{-1}]$ ($\mathbf{K}^d = \sum_i k_i^d \cdot x_i$, where \mathbf{x}_i is the second reactant). We use indices (l,c) and exponants (2,1) to distinguish between $\mathrm{C}_3\mathrm{H}_2$ and $\mathrm{C}_3\mathrm{H}$ in linear and cyclic forms (for example, ${}^2\mathrm{b}_c$ means "branching ratio for cyclic $\mathrm{C}_3\mathrm{H}_2$ "). Similar equations can be written for $k \cdot \mathrm{C}_3\mathrm{H}_2$, $c \cdot \mathrm{C}_3\mathrm{H}$ and $k \cdot \mathrm{C}_3\mathrm{H}$. Using the observed column densities for each species at the border of the filament (i.e. row number 2), where the electron density should be the highest in our data set, and writing $\mathbf{k}_c^r = \mathbf{k}_l^r$ and $\mathbf{X} \equiv \mathbf{x}(\mathbf{c}\cdot\mathrm{C}_3\mathrm{H}_3^+)/\mathbf{x}(\mathbf{l}\cdot\mathrm{C}_3\mathrm{H}_3^+)$, calculations lead to:

$$\begin{aligned} \mathbf{x}(c\text{-}\mathbf{C}_{3}\mathbf{H}_{2})/\mathbf{x}(c\text{-}\mathbf{C}_{3}\mathbf{H}) &= (^{2}\mathbf{b}_{c}/^{1}\mathbf{b}_{c})(^{1}\mathbf{K}_{c}^{d}/^{2}\mathbf{K}_{c}^{d}) \simeq 5 ;\\ \mathbf{x}(l\text{-}\mathbf{C}_{3}\mathbf{H}_{2})/\mathbf{x}(l\text{-}\mathbf{C}_{3}\mathbf{H}) &= (^{2}\mathbf{b}_{l}/^{1}\mathbf{b}_{l})(^{1}\mathbf{K}_{l}^{d}/^{2}\mathbf{K}_{l}^{d}) \simeq 2 ;\\ \mathbf{R}_{2} &= (^{2}\mathbf{b}_{c}/^{2}\mathbf{b}_{l})(^{2}\mathbf{K}_{l}^{d}/^{2}\mathbf{K}_{c}^{d})\mathbf{X} \simeq 16 ;\\ \mathbf{R}_{1} &= (^{1}\mathbf{b}_{c}/^{1}\mathbf{b}_{l})(^{1}\mathbf{K}_{l}^{d}/^{1}\mathbf{K}_{c}^{d})\mathbf{X} \simeq 6 .\end{aligned}$$

A simple and coherent set of solution for this system is ${}^{2}b_{l} \simeq {}^{1}b_{l} \simeq {}^{1}b_{c} \simeq {}^{2}b_{c}$; ${}^{1}K_{l}^{d} \simeq 2 {}^{2}K_{l}^{d}$; ${}^{1}K_{c}^{d} \simeq 5 {}^{2}K_{c}^{d}$; ${}^{1}K_{c}^{d} \simeq (1/6) {}^{1}K_{l}^{d} X$; ${}^{2}K_{c}^{d} \simeq (1/16) {}^{2}K_{l}^{d} X$.

Under the hypothesis of steady state and from the observed abundance ratio at the border of the TMC-1 filament, we find that $c-C_3H_2$ is destroyed approximatively 5 times slower than $c-C_3H$ and that the $l-C_3H$ mean destruction rate is twice the $l-C_3H_2$ one.

To our knowledge, no additional formation process of c-C₃H₂ in dark clouds can be invoked to explain an increase of its abundance with respect to l-C₃H₂. Variations of R₂ (and R₁) in TMC-1 could result from variations of the destruction rates of c-C₃H₂ and l-C₃H₂. Such an explanation has been proposed by Maluendes, McLean, & Herbst (1993): while c-C₃H₂ is inert with respect to most neutral-neutral reactions, l-C₃H₂, as l-C₃H and c-C₃H, are easily destroyed by these reactions. Hence, because neutral-neutral reactions proceed fast when a radical reacts with abundant atoms, it is possible in a dark cloud like TMC-1 that R₂ is one order of magnitude larger than in a more diffuse medium where the proportion of reactive ions (increasing with respect to reactive atoms) is sufficient to destroy c-C₃H₂ and l-C₃H₂ at the same rate. The same explanation holds for the progenitor ions c-C₃H₃⁺ and l-C₃H₃⁺ (see Cernicharo et al. 1999): while l-C₃H₃⁺ - assumed to be the progenitor of l-C₃H₂ - can be efficiently removed through ion-neutral reactions lowering the amount available to produce l-C₃H₂, c-C₃H₃⁺ is mainly affected by dissociative recombination to produce c-C₃H₂. R₂ thus depends on the ion-neutral reactions of l-C₃H₃⁺ that do not affect the cyclic ion: in dark clouds, where the abundance of reactive molecules can be hundred times larger than that of electrons, l-C₃H₃⁺ is removed faster by ion-neutral reactions than by the dissociative recombination lowering the production rate of l-C₃H₂, and increasing R₂. In a more diffuse medium, where the ion-neutral reactions are much less efficient, l-C₃H₃⁺ and c-C₃H₃⁺ are removed by the same process and R₂ is then closer to ${}^{2}K_{l}^{d}/{}^{2}K_{c}^{d}$ (whose value also decreases with decreasing density as suggested above). This explanation is strengthened by the lower R₂ values found in the diffuse medium from absorption measurements by Cernicharo et al. (1999) - 3 to 5 versus 10 to 40 in TMC-1 and by the observed R₁ variations across the filament. We test this hypothesis in the next section.

4.2. Chemical modelling

In order to study the impact of physical conditions on the R_2 ratio, we have run 25 models of gas phase chemistry with different densities and visual extinctions using a time dependent chemistry code solving the system of stiff ordinary differential equations with the Gear method. Although we have used a time dependent chemistry code - samples of the evolution chemistry are shown in Fig.4 - we discuss here the results at steady state. Indeed, we would like to study the basic processes of the isomeric differentiation, and to identify the dominant reactions leading to the observed large variations of R_2 in diffuse and dense clouds. We aim first at a qualitative description of the mechanism. A quantitative description would require a more accurate knowledge of the reaction rates (including branching ratios) important for c-C₃H₂ and l-C₃H₂. Fig.4 shows that c-C₃H₂/l-C₃H₂ and $c-C_3H_3^+/l-C_3H_3^+$ abundance ratios reach values close to the steady state value, just after they reach their maximum abundance. As a first step, it is thus reasonable to consider only the processes leading to variation in isometric abundances at steady state. We have used, as chemical network, the UMIST95 database (Millar, Farquhar, & Willacy 1997) in a pure gas-phase scheme. As recommanded by the authors, the species C_2H_5 , C_2H_6 and their ions have been excluded from the network. We have also excluded the species including the following atoms: P, Si, Cl, Na, Mg. On the other hand, we have included the grain surface formation of H₂ with a rate coefficient $1.5 \times 10^{-17} \text{ n}^2(\text{H}) \text{ cm}^{-3} \text{ s}^{-1}$. Moreover, we have updated the dissociative recombination rate of $c-C_3H_3^+$ at 300 K to $7.0\pm2.0 \times 10^{-7}$ $cm^3 s^{-1}$ (Abouelaziz et al. 1993) and assumed the same value for $l-C_3H_3^+$. Parameters of the models and initial elemental abundances are given in Table 3 and 4, respectively.

In Fig.4, we see that R_2 , and also X, increase with increasing visual extinction. This general trend of our models agrees with observations since R_2 is lower in the diffuse medium than in the TMC-1 dark cloud. The models also show that the increase is faster in higher density media. We have plotted in Fig.5 the electronic density with respect to R_2 for each model. The distribution of the points exhibit clearly a correlation between the ionization fraction and the cyclic over linear ratio which can be separated in two regimes: "low" $(1 \le A_V \le 2)$ and "high" $(5 \le A_V \le 10)$, $A_V = 3$ mag being an intermediate case. In the low extinction regime, R_2 strongly depends on the electronic abundance. This is an interesting result because it opens the possibility to use R_2 as a tool to probe the electronic abundance in low extinction regions where $H^{13}CO^+$ and DCO^+ cannot be detected. Why is the cyclic over linear ratio sensitive to the ionization fractionation? In the UMIST95 chemistry, C⁺ can destroy both l-C₃H₂ and c-C₃H₂, whereas C reacts only with the linear isomer. R₂ is then sensitive to the electronic abundance via the C^+/C ratio as shown in Fig.6. Note that the range 3 to 5 that Cernicharo et al. (1999) have observed in the diffuse medium corresponds, as expected, to high electronic abundances and to moderate visual extinctions. In the high extinction regime, R_2 is independent of the C^+/C ratio, while it is sensitive to the $(H_3^++C^+)/O$ ratio (Fig.7). Indeed, in the UMIST95 database the atomic oxygen reacts only with $l - C_3 H_2$ (O + $l - C_3 H_2 \rightarrow C_2 H_2 + CO$). This reaction is negligible with respect to other destruction reactions (in particular the reactions with C⁺ which affects both isomers) until all the carbon is locked into CO. It is thus unefficient in low visual extinction media where the abundances of C and C^+ remain high. On the other hand, it becomes the principal destruction channel of l-C₃H₂ at high visual extinctions: for A_V = 2 mag and a density of 10^3 cm⁻³ the O + l-C₃H₂ reaction is roughly 15% of the l-C₃H₂ destruction rate while it represents more than 80% of it at $A_V = 10$ mag. The variations of the cyclic over linear abundance ratio of C_3H_2 can thus be understood as a consequence of the competition between neutral-neutral and ion-neutral reactions in the interstellar medium.

The determination of electronic abundances from measurements of R_2 using Fig.5 may not be straightforward. Indeed, taking into account the large uncertainties in the reaction rates and branching ratios, it is difficult to infer quantitative results from this study on the C_3H_2 isomers. We can say however that low R_2 values are specific of high electronic abundance while high R_2 values indicate low electronic abundance. In a more general way, isomeric ratio can be used to probe physical conditions in different media from the diffuse gas to dark clouds. In the future, cyclic and linear species other than the isomers of C_3H_2 shall be included in chemical models. C_3H is a good candidate. Indeed, Kaiser et al. (1997, 1999) have shown that the neutral-neutral reaction $C + C_2H_2 \rightarrow C_3H + H$ produces linear and cyclic C_3H , with an increase of the linear over cyclic abundance ratio with rising collision energies. It would be interesting to make measurements of R_1 in absorption in the diffuse medium in order to compare this neutral-neutral formation path with the $C_3H_3^+$ dissociative recombination.

This study has been partly begun. Indeed, in a recent paper, Turner, Herbst, & Terzieva (2000) investigate the R_1 and R_2 ratios in three translucent clouds and in the dark clouds TMC-1 and L183. They compare observations to the predictions of a modified version of the New Standard Model of chemistry (Lee et al.1996) which includes cyclic and linear species for all precursor to C_3H and C_3H_2 . One of their conclusion is that the clear observed anticorrelation of R_2 with extinction suggests that the ratios are strongly affected by formation or destruction rates, and not just by branching ratios among the relevant chemical reactions.

5. Conclusion

In this paper, we have presented new results about the cyclic over linear abundance ratios of C_3H_2 and C_3H toward the CP of TMC-1. These ratios exhibit variations which are probably due to variations in the electronic abundance across the filament. Indeed, the chemical modelling of the reactions that affect c- C_3H_2 and l- C_3H_2 shows that their ratio is sensitive mainly to the behavior of the C⁺/C (at low visual extinction) and $(H_3^++C^+)/O$ (at high visual extinction) abundance ratios, which are both related to the electronic abundance. As C_3H_2 is observed in its cyclic and linear form in different physical conditions, it could be used as a tool to probe the electronic abundance in the ISM. However, to infer accurate values of the fractional ionization from cyclic and linear isomers abundances, the branching ratios and kinetic rates used in chemical networks must be more accurately known. In addition to laboratory studies, a systematic measurement of R_1 and R_2 in different astrophysical media should be useful for this work.

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Table 1: Line Parameters for C ₆ H, <i>l</i> -C ₃ H ₂ and <i>c</i> -C ₃ H ₂ at Selected	Positions in TMC-1
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	C ₆ H			l-C ₃ H ₂				c-C ₃ H ₂		
Position	T_{mb}	v_{LSR}	Δv	T_{mb}	VLSR	Δv	Position	T_{mb}	v_{LSR}	Δv
(",")	K	${\rm km~s^{-1}}$	${\rm km~s^{-1}}$	K	${\rm km~s^{-1}}$	${\rm km~s^{-1}}$	(",")	K	${\rm km \ s^{-1}}$	${\rm km~s^{-1}}$
0,0	$0.41 {\pm} 0.02$	5.72 ± 0.01	0.27 ± 0.02	0.27 ± 0.03	$5.83 {\pm} 0.02$	$0.33 {\pm} 0.03$	0,0	$3.69 {\pm} 0.59$	$5.54 {\pm} 0.01$	0.12 ± 0.02
	$0.13 {\pm} 0.02$	$6.02 {\pm} 0.05$	$0.29 {\pm} 0.07$					$4.44 {\pm} 0.59$	5.72 ± 0.01	$0.14 {\pm} 0.02$
								$5.46 {\pm} 0.59$	$5.89 {\pm} 0.02$	$0.20 {\pm} 0.02$
								$2.46 {\pm} 0.59$	$6.12 {\pm} 0.02$	$0.14 {\pm} 0.03$
-40,0	$0.31 {\pm} 0.03$	$5.74 {\pm} 0.01$	$0.25 {\pm} 0.00$	$0.31 {\pm} 0.07$	$5.78 {\pm} 0.03$	$0.44 {\pm} 0.08$	-50,0	2.20 ± 0.76	$5.48 {\pm} 0.05$	$0.20 {\pm} 0.00$
	$0.11 {\pm} 0.11$	$5.99 {\pm} 0.02$	$0.09 {\pm} 0.20$					$3.99 {\pm} 0.76$	$5.70 {\pm} 0.03$	$0.20 {\pm} 0.00$
0,40	$0.29 {\pm} 0.04$	$5.82 {\pm} 0.03$	$0.21 {\pm} 0.06$	$0.28 {\pm} 0.06$	$5.85 {\pm} 0.02$	$0.18 {\pm} 0.04$	0,50	$3.96 {\pm} 0.71$	$5.68 {\pm} 0.02$	$0.17 {\pm} 0.04$
	$0.25 {\pm} 0.04$	$6.07 {\pm} 0.04$	$0.21 {\pm} 0.09$	$0.26 {\pm} 0.06$	6.15 ± 0.02	$0.14 {\pm} 0.04$		$4.58 {\pm} 0.71$	$5.96 {\pm} 0.02$	$0.28 {\pm} 0.05$
40,-40	$0.48 {\pm} 0.03$	$5.78 {\pm} 0.01$	0.25 ± 0.00	$0.28 {\pm} 0.07$	$5.80 {\pm} 0.03$	$0.21 {\pm} 0.06$	50, -50	$2.83 {\pm} 0.78$	5.61 ± 0.03	$0.19 {\pm} 0.05$
	$0.22 {\pm} 0.03$	$6.03 {\pm} 0.03$	$0.27 {\pm} 0.05$	$0.23 {\pm} 0.07$	$6.08 {\pm} 0.03$	$0.13 {\pm} 0.06$		$4.88 {\pm} 0.78$	$5.88 {\pm} 0.03$	$0.34 {\pm} 0.06$

Notes – Positions are offsets relative to the Cyanopolypne Peak : $\alpha(1950) = 04^h 38^m 38.6^s$, $\delta(1950) = 25^\circ 35^\circ 45.0^\circ$ TMC-1 exhibits two main components at $v_{LSR} \simeq 5.3$ and 6.1 km s⁻¹ (Sume, Downes, & Wilson 1975) When possible, we have fitted several components along a line of sight

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Position	$N(c-C_3H)$	$N(l-C_3H)$	Nc/M
(",")	$10^{12} {\rm ~cm^{-2}}$	$10^{11} {\rm ~cm^{-2}}$	$(= R_1)$
-40,-40	$3.7 {\pm} 0.2$	≤ 4	≥ 9
-20,-20	$7.7 {\pm} 0.3$	$8.0 {\pm} 0.2$	$10{\pm}0.5$
$0,\!0$	$10.3{\pm}0.3$	$8.4 {\pm} 0.1$	12 ± 0.5
$20,\!20$	$10.0{\pm}0.3$	$7.6 {\pm} 0.4$	13 ± 1.2
40,40	$5.8{\pm}0.3$	$6.7 {\pm} 0.5$	$9{\pm}1.1$
$60,\!60$	$3.6{\pm}0.2$	$6.1{\pm}0.7$	$6{\pm}1.0$
10,-70	5.2 ± 0.2	$7.0 {\pm} 0.4$	$7{\pm}0.6$
-10,-50	$5.3 {\pm} 0.2$	$7.8 {\pm} 0.2$	7 ± 0.4
-30,-30	$5.1{\pm}0.2$	$8.0 {\pm} 0.1$	6 ± 0.4
-50,-10	$4.7 {\pm} 0.2$	$7.1 {\pm} 0.5$	7 ± 0.8
-70,10	$3.7{\pm}0.2$	$7.3 {\pm} 0.3$	5 ± 0.5
$30,-60^{1}$	$7.8{\pm}0.3$	$8.0 {\pm} 0.2$	$10{\pm}0.6$
$-30,0^{2}$	$10.0 {\pm} 0.4$	$8.2 {\pm} 0.2$	12 ± 0.7

Table 2: Derived Column Densities for c-C₃H and l-C₃H in TMC-1

Notes – Positions are offsets relative to the CP : $\alpha(1950) = 04^h 38^m 38.6^s, \ \delta(1950) = 25^\circ 35' 45.0"$

 1 c-C₃H₂ peak

 2 *l*-C₃H₂ peak

 Table 3: Parameters of the Chemical Models

Parameter	Value
$n(H_2) (10^3 \text{ cm}^{-3})$	1, 3, 10, 30, 100
T(K)	10
$A_V (mag)$	$1 \ , 2 \ , 3 \ , 5 \ , 10$
$\zeta~({ m s}^{-1})$	1.3×10^{-17}
$u_i \ (\mathrm{erg} \ \mathrm{cm}^{-3})$	2×10^{-15}

 ζ is the cosmic-ray ionization rate

 u_i is the density of ionizing radiation at $\mathbf{A}_V=\mathbf{0}$

	with respect to Hydrogen
He	1.4×10^{-1}
С	7.3×10^{-5}
Ν	2.14×10^{-5}
0	1.76×10^{-4}
\mathbf{S}	2.0×10^{-8}
Fe	3.0×10^{-9}

 Table 4: Initial Elemental Abundances

 with respect to Hydrogen

Figure Captions

Fig. 1.— Spectra of C₆H, cyclic and linear C₃H₂, and cyclic and linear C₃H towards the cyanopolyyne peak of TMC-1 ($v_{LSR} = 5.8 \text{ kms}^{-1}$). Note that the C₆H $^2\Pi_{3/2}$ J=15/2–13/2 quadruplet and the l-C₃H₂ 1₀₁-0₀₀ transition are observed in the same bandwidth. The J=3/2–1/2 F=2–1 and F=1–0 transitions of c-C₃H 2₁₂–1₁₁ are not shown here.

Fig. 2.— Contour levels of integrated intensity in TMC-1 for: **Panel (a):** C₆H ${}^{2}\Pi_{3/2}$ J=15/2-13/2 and *c*-C₃H₂ 1₁₀ - 1₀₁; **Panel (b):** C₆H ${}^{2}\Pi_{3/2}$ J=15/2-13/2 and *k*-C₃H₂ 1₀₁ - 0₀₀ (the crosses indicate the measured positions); **Panel (c):** C₆H ${}^{2}\Pi_{3/2}$ J=15/2-13/2 and HC₃N 10–9; **Panel (d):** *c*-C₃H₂ 1₁₀ - 1₀₁ and *k*-C₃H₂ 1₀₁ - 0₀₀ (numbers indicate the values of R₂, i.e. the ratio N(c-C₃H₂)/N(k-C₃H₂)). All contours go from 50 to 100 % of the peak value by steps of 10 %. The peak values are 0.16 K·kms⁻¹, 2.6 K·kms⁻¹, 0.15 K·kms⁻¹, and 1.75 K·kms⁻¹ for C₆H, *c*-C₃H₂, *k*-C₃H₂, and HC₃N respectively. The coordinates of the (0,0) position are: $\alpha = 04^{h}38^{m}38.6^{s}$, $\delta = 25^{\circ}35'45.0''$ (1950.0) corresponding to the cyanopolyyne peak. The boxes to the right compare the C₆H and *k*-C₃H₂ spectra averaged along the six directions indicated by the arrows. Note the significant emission of *k*-C₃H₂ toward the west side of the TMC-1 filament where the emission of C₆H decreases rapidly.

Fig. 3.— Variations of R_2 (i.e. the ratio $N(c-C_3H_2)/N(l-C_3H_2)$) across the TMC-1 filament using: **Lower panel:** a constant density of 3×10^4 cm⁻³; **Upper panel:** densities derived by Pratap et al. (1997). In order to increase the signal to noise ratio, the spectra have been averaged along six rows (separated by 28") parallel to the TMC-1 filament. Lower limits in row number 6 are derived from 3σ upper limits on the $l-C_3H_2$ emission. The cyanopolyyne peak (CP) is located within row number 4.

Fig. 4.— Chemical evolution for three different visual extinctions at $n(H_2) = 10^4 \text{ cm}^{-3}$ and T = 10 K. Upper boxes show abundances relative to H for C, l-C₃H₂, c-C₃H₂ and e^- . Lower boxes show the resulting cyclic over linear abundance ratio for C₃H₂ and C₃H₃⁺ isomers.

Fig. 5.— The predicted electronic abundance vs. R_2 at steady state. Two regimes can be distinguished according as the visual extinction is low or high.

Fig. 6.— The predicted dependence of R_2 with the C⁺/C ratio at steady state. A clear relationship is established at low visual extinction between R_2 and the ionization fraction of atomic carbon. Note that at high visual extinctions R_2 depends only weakly on C⁺/C.

Fig. 7.— Model prediction for the relation of R_2 vs. the $(H_3^++C^+)/O$ ratio at steady state. Note the good correlation between R_2 and $(H_3^++C^+)/O$ at high visual extinctions.













