

SYNTHESIS OF SOME QUINOLINE THIOSEMICARBAZONE DERIVATIVES OF POTENTIAL ANTIMICROBIAL ACTIVITY

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مركبات الثيوسيميكرbazون تتميز بأن لها فاعليات بيولوجية مختلفة كمضادات للبكتريا والفطريات والسل وترياق ضد سموم المعادن وذلك اعتماداً على طبيعة المستبدلات على النتروجين رقم 1 ورقم 4 لجزئية الثيوسيميكرbazون إضافة إلى مشتقات 8-هيدروكسي كينولين وكما هو مدون بالتراث العلمي كمضادات للبكتريا حيث أن لها صفات مخيلية مستمدة من مجموعة 8-هيدروكسي وذرة النتروجين بحلقة الكينولين.

وتشمل تلك الدراسة تحضير مشتقات جديدة من 5-استيل (أو 5-بنزويل) 8-هيدروكسي كينولين -4- مستبدل-ثيوسيميكرbazون لاستيضاح تأثير تلك التحورات على النشاط المضاد للبكتريا ، كذلك تم حلقة بعض الثيوسيميكرbazونات للحصول على مشتقات من الثيازولدينون كوسيلة لادخال مجموعة البثول الخاصة بجزئية الثيوسيميكرbazون بداخل حلقة غير متجانسة والهدف من ذلك هو دراسة مدى ما يتأثر به النشاط البيولوجي من جراء أحتواء مجموعة الثيوسيميكرbazون في حلقة غير متجانسة. وقد تم اختيار المستبدلات R, R^1 لتعطي قاعة لدراسة نتائج الاختلاف في الخواص الالكترونية للمركبات المخلفة على النشاط الذي أعطته تلك المركبات. قد تم أثبات الترايب البنائية للمركبات باستخدام الأشعة تحت الحمراء والرنين النووي المغناطيسي كما تم استخدام طيف الكتلة لبعض المركبات. كما أجرى اختبار الفاعلية البيولوجية للمركبات المشيدة كمضادات للبكتريا والفطريات بالمقارنة مع عقار الاستربتومايسين كمضاد للبكتريا وعقار كلوتريمازول كمضاد للفطريات باستخدام طريقة *cup inhibition zone test*.

5-Acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemi-carbazones (IIa-m, IIIa-m respectively) have been prepared via the condensation of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline with the appropriate 4-substituted-3-thiosemicabazides (Ia-l). The thiosemicarbazones (IIa-l, IIIa-f) were subjected to cyclization into the corresponding thiazolidinones (IVa-l, Va-f) by the reaction with ethyl bromoacetate in the presence of anhydrous sodium acetate. The structures of the thiosemicarbazones as well as the corresponding thiazolidinones were assigned based on both elemental and spectroscopic evidences. The prepared compounds were also evaluated for antibacterial and antifungal activities.

INTRODUCTION

Thiosemicarbazones, a class of compounds possessing a wide spectrum of numerous pharmacological activities, have been studied for activity as antibacterial,¹⁻⁶ antifungal,⁷⁻⁹ antituberculous,¹⁰⁻¹³ anti-malarial,¹⁴⁻¹⁷ antiviral infection,¹⁸⁻²¹ as well as analgesic and antipyretic.²² In the past few years, thiosemicarbazones have been of great interest because of their reported antitumor activity.²³⁻³⁴

In addition, thiosemicarbazones were reported as antidotal for metals toxicity.³⁵⁻³⁶ In a search for new biologically active agents, many research workers have successfully synthesized a variety of different aromatic and heteroaromatic thiosemi-carbazone derivatives. This depending on the nature of the substituents at N¹ and N⁴ of the thiosemicarbazone moiety. In addition 8-hydroxyquinoline and its derivatives were reported as antimicrobial agents.³⁷ The antimicrobial activity has been

attributed to the chelating properties provided by the 8-hydroxy group and quinoline ring nitrogen.³⁸ In the present investigation, it was of interest to prepare new 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemi-carbazone derivatives, in which thiosemi-carbazone moiety incorporated with 8-hydroxyquinoline nucleus to explore this interesting modifications for the development of potential antimicrobial activity. Thiosemi-carbazones were then cyclized into thiazolidinone ring systems as a mean of trapping the SH function of thiosemicarbazone moiety within a heterocyclic ring. This was performed to study the effect on the activity of the product when the thiosemicarbazones are engaged in a rigid heterocyclic structure.

EXPERIMENTAL

Materials and equipments

Melting points were determined on an electrothermal melting point apparatus (Stuart Scientific Co.) and were uncorrected. Elemental microanalyses were performed on Perkin-Elmer, 240° Elemental Analyzer, at the Faculty of Science, Assiut University. ¹H-NMR spectra were run on Varian Em-360L NMR spectrophotometer (60 MHz) (Varian USA) at the Faculty of Pharmacy, Assiut University, and on Joel, Lambda, Oxford NMR YH (400MHz, Japan) at Assiut University Central Lab using tetramethylsilane (TMS) as an internal standard. The chemical shifts are expressed in δ (ppm). IR spectra were carried out as KBr disc on Shimadzu Infrared Spectrophotometer 200-91527 at the Faculty of Pharmacy, Assiut University. Mass spectra were performed with JEOL JMS600, Assiut University Central Lab, Assiut and at the Microanalytical center, Faculty of Science, Cairo University. The reported procedure for the synthesis of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline were utilized,³⁹ also 4-substituted-3-thiosemicarbazide compounds (Ia-l) were prepared according to reported method.¹⁴

Synthesis of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemi-carbazone compounds (IIa-m, and IIIa-m)

A mixture of thiosemicarbazide or appropriate 4-substituted-3-thiosemi-carbazide (5.3 mmol) and 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline (5.3 mmol) in 50 ml absolute ethanol containing 4 drops conc. HCl was heated under reflux for 2-8 hr. The precipitate formed directly or after addition of water for compounds IIb, IIc, IId, IIe, II f was filtered, dried and crystallized from suitable solvent. The yields, melting points and elemental microanalyses were listed in Tables 1, 3. The ¹H-NMR data were listed in Tables 2, 4.

Synthesis of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline 2-(3-substituted-4-oxo-thiazolidin-2-ylidene) hydrazone compounds (IVa-l&Va-f)

A solution of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4- substituted thiosemi-carbazones (1.5 mmol) in absolute ethanol (30ml) was treated with equimolar amount of ethyl bromoacetate (1.5 mmol, 0.166 ml) in presence of anhydrous sodium acetate (100 mg). The reaction mixture was heated under reflux for 4-6 h then concentrated and left over night. The formed crystals were filtered and recrystallized from absolute ethanol. The yield, melting point and elemental microanalyses were listed in Tables 5, 7. The ¹H-NMR data were listed in Tables 6, 8.

Antimicrobial activity (organisms and culture conditions)

Material and method

Antimicrobial activity of the synthesized compounds **IIa-m**, **IIIa-m**, **IVa-j** and **Va-f** were tested against:

a) Bacteria

Gram-positive bacteria: *Micrococcus luteus*, and *Staphylococcus aureus*. Gram-negative bacteria: *Pseudomonas aeruginosa* and *Serratia marscens*.

b) Fungi

Candida albicans, *Trichophyton rubrum*, *Geotrichum candidum*, and *Scopulariopsis brivicalis*.

Table 1: Physical data of 5-acetyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**II a-m**).

No.	R ¹	Yield %	M.P ^o Solvent of crys.	M.F/ M.Wt	Microanalysis Calculated/found			
					C %	H %	N %	S%
IIa	H	86	222-24	C ₁₂ H ₁₂ N ₄ OS	55.37	4.65	21.52	12.32
			E	260.32	54.33	4.76	21.26	12.27
IIb	CH ₃	81	163-65	C ₁₄ H ₁₅ N ₃ OS	56.91	5.14	20.42	11.69
			E/W	274.34	56.98	4.51	20.41	12.09
IIc	C ₂ H ₅	85	140-42	C ₁₄ H ₁₆ N ₄ OS	58.31	5.59	19.43	11.12
			E/W	288.37	58.15	5.88	19.54	11.29
II d	CH ₂ CH=CH ₂	86	142-44	C ₁₅ H ₁₆ N ₄ OS	59.98	5.37	18.6	10.68
			E/W	300.38	59.79	5.47	18.65	10.87
IIe	C ₄ H ₉ (n)	79	130-32	C ₁₆ H ₂₀ N ₄ OS	60.73	6.37	17.71	10.13
			E/W	316.42	60.72	5.85	17.67	9.83
II f	C ₆ H ₁₁ (c)	87	185-87	C ₁₈ H ₂₂ N ₄ OS	63.13	6.48	16.36	9.36
			E/W	342.46	62.81	6.98	16.23	9.37
II g	C ₆ H ₅	83	215-17	C ₁₈ H ₁₆ N ₄ OS	64.26	4.79	16.65	9.53
			E	336.41	63.99	4.90	16.73	9.76
II h	o-CH ₃ -C ₆ H ₄	75	210-12	C ₁₉ H ₁₈ N ₄ OS	65.12	5.18	15.99	9.15
			E	350.44	64.76	5.24	15.94	8.83
II i	m-CH ₃ -C ₆ H ₄	63	140-42	C ₁₉ H ₁₈ N ₄ OS	65.12	5.18	15.99	9.15
			E	350.44	64.74	4.66	15.96	9.01
II j	p-CH ₃ -C ₆ H ₄	65	220-22	C ₁₉ H ₁₈ N ₄ OS	65.12	5.18	15.99	9.15
			E	350.44	64.36	4.71	16.05	9.22
II k	p-OCH ₃ -C ₆ H ₄	77	158-60	C ₁₉ H ₁₈ N ₄ O ₂ S	62.28	4.95	15.29	8.75
			E	366.44	62.06	5.10	15.32	8.60
III	p-F-C ₆ H ₄	79	147-49	C ₁₈ H ₁₅ FN ₄ OS	61.00	4.27	15.81	9.05
			E	354.40	60.43	4.18	15.78	8.67
II m	o-Cl-C ₆ H ₄	75	195-97	C ₁₈ H ₁₅ ClN ₄ OS	56.92	3.98	14.75	8.44
			E	379.86*	57.14	4.10	14.64	7.42

* Contain 0.5 molecule of water

E: Ethanol

E/W: Ethanol/Water (2:1)

Table 2: $^1\text{H-NMR}$ data of 5-acetyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**II a-m**).

No.	R ¹	$^1\text{H NMR}$ (δ ppm in CDCl_3)*
IIa	H	10.23 (s, 1H, N ² HCS); 8.90 (d, 1H, H ₂ of quinoline); 8.80 (d, 1H, H ₄ of quinoline); 8.40-7.36 (m, 3H, OH, H _{3,6} of quinoline); 7.10 (d, 1H, H ₇ of quinoline); 4.13 (br. s, 2H, NH ₂); 2.43, 2.33 (2s, 3H, CH ₃); [80%, 20%]**
IIb	CH ₃	9.23 (br. s, 1H, N ² HCS); 9.13 (d, 1H, H ₂ of quinoline); 8.96 (d, 1H, H ₄ of quinoline); 8.63 (br. s, 1H, OH); 8.26-7.63 (m, 3H, N ⁴ HCH ₃ , H _{3,6} of quinoline); 7.45 (d, 1H, H ₇ of quinoline); 3.33 (d, 3H, NHCH ₃); 2.48, 2.45 (2s, 3H, CH ₃) [65%, 35%]
IIc	C ₂ H ₅ ***	9.20 (br. s, 1H, N ² HCS); 8.70 (d, 1H, H ₂ of quinoline); 8.56 (d, 1H, H ₄ of quinoline); 7.94 (s, 1H, OH); 7.59 (d, 1H, H ₆ of quinoline); 7.55-6.89 (m, 3H, N ⁴ HCH ₂ , H _{3,7} of quinoline); 3.42 (m, 2H, CH ₂ CH ₃); 2.19, 2.09 (2s, 3H, CH ₃) [60%, 40%]; 1.02, 0.94 (2t, 3H, CH ₂ CH ₃) [60%, 40%]
II d	CH ₂ CH=CH ₂	9.23 (br. s, 1H, N ² HCS); 9.10 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 8.60-7.70 (m, 4H, OH, N ⁴ HCH ₂ , H _{3,6} of quinoline); 7.34 (d, 1H, H ₇ of quinoline); 6.50-5.80 (m, 1H, CH=CH ₂); 5.46 (d, 1H, CH=CH ₂); 5.21 (d, 1H, CH=CH ₂); 4.50 (t, 2H, NHCH ₂); 2.53, 2.45 (2s, 3H, CH ₃); [60%, 40%]
Iie	C ₄ H ₉ (n)	9.25 (br. s, 1H, N ² HCS); 9.16-8.80(m, 2H, H _{2,4} of quinoline); 8.50 (br. s, 1H, OH); 8.10-7.25 (m, 4H, N ⁴ HCH ₂ , H _{3,6,7} of quinoline); 3.80 (q, 2H, NHCH ₂); 2.50, 2.41 (2s, 3H, CH ₃) [73%, 27%]; 2.00-1.16 (m, 4H, CH ₂ CH ₂ CH ₃); 1.10-0.70 (t, 3H, CH ₂ CH ₃)
II f	C ₆ H ₁₁ (c)	9.00-8.60 (m, 3H, N ² HCS, H _{2,4} of quinoline); 8.20 (s, 1H,OH); 8.06-7.10 (m, 4H, N ⁴ H, H _{3,6,7} of quinoline) 4.60-4.00 (m, 1H, NHCH of cyclohexyl); 2.40, 2.35 (2s, 3H, CH ₃) [60%, 40%]; 2.30-0.90 (m, 10H, (CH ₂) ₅ of cyclohexyl)
II g	C ₆ H ₅ ***	10.64, 10.18 (2s, 1H, N ² HCS) [83%, 17%]; 9.84, 9.24 (2s, 1H,N ⁴ Hphenyl), [83%, 17%]; 8.86 (d, 1H, H ₂ of quinoline); 8.68 (d, 1H, H ₄ of quinoline); 7.67-6.09(m, 9H, OH, H _{3,6,7} of quinoline, NHC ₆ H ₅); 2.49, 2.38 (2s, 3H, CH ₃) [83% 17%]**
II h	o-CH ₃ -C ₆ H ₄	10.95, 10.26 (2s, 1H, N ² HCS) [80%, 20%]; 9.90, 9.40 (2s, 1H,N ⁴ H-o.tolyl) [80%, 20%]; 9.30-8.90 (m, 2H, H _{2,4} of quinoline); 8.15-7.15 (m, 8H, OH, H _{3,6,7} of quinoline, NHC ₆ H ₄); 2.63 (s, 3H, CH ₃ of o.tolyl); 2.45, 2.30 (2s, 3H, CH ₃) [80% 20%]**
II i	m-CH ₃ -C ₆ H ₄	9.72 (br. S, 1H, N ² HCS); 9.60 (br. S, 1H, N ⁴ H- m.tolyl); 9.33-8.83 (m, 2H, H _{2,4} of quinoline); 8.60 (br. S, 1H, OH); 8.2 (d, 1H, H ₆ of quinoline); 7.96-7.06 (m, 6H, H _{3,7} of quinoline, NHC ₆ H ₄); 2.63 (s, 3H, CH ₃ of m.tolyl); 2.60, 2.50 (2s, 3H,CH ₃) [60%, 40%]
II j	p-CH ₃ -C ₆ H ₄	9.66, 9.54 (2s, 1H, NH ² CS) [60%, 40%]; 9.26 (br. S, 1H, N ⁴ H-p.tolyl); 9.10-8.80 (m, 2H, H _{2,4} of quinoline); 8.63 (br. S, 1H, OH); 8.17 (d, 1H, H ₆ of quinoline); 7.96-7.23 (m, 6H, H _{3,7} of quinoline, NHC ₆ H ₄); 2.6 (s, 3H, CH ₃ of p. tolyl); 2.50, 2.40 (2s, 3H,CH ₃) [60%, 40%]
II k	p-CH ₃ O-C ₆ H ₄	9.52, 9.42 (2s, 1H, NH ² CS) [75%, 25%]; 9.25 (br. S, 1H, N ⁴ H-p.methoxyphenyl); 9.12-8.70 (m, 2H, H _{2,4} of quinoline); 8.52 (br. S, 1H, OH); 8.10 (d, 1H, H ₆ of quinoline); 7.92-6.86 (m, 6H, H _{3,7} of quinoline, NHC ₆ H ₄); 3.86 (s, 3H, OCH ₃); 2.53, 2.43 (2s, 3H,CH ₃); [75%, 25%]
III	p-F-C ₆ H ₄	11.06, 10.53 (2s, 1H, N ² HCS) [80%, 20%]; 10.13 (s, 1H, N ⁴ H-p. fluorophenyl); 9.30-8.80 (m, 2H, H _{2,4} of quinoline); 8.36-7.06 (m, 8H, OH, H _{3,6,7} of quinoline, NHC ₆ H ₄); 2.66, 2.55 (2s, 3H, CH ₃) [80% 20%]**
II m	o-Cl-C ₆ H ₄	10.23, 10,10 (2s, 1H, N ² HCS) [60%, 40%]; 9.33-8.73 (m, 3H, N ⁴ H-o.chlorophenyl, H _{2,4} of quinoline); 8.62 (br. S, 1H, OH); 8.20 (d, 1H, H ₆ of quinoline); 7.96-7.03 (m, 6H, H _{3,7} of quinoline, NHC ₆ H ₄); 2.80, 2.70 (2s, 3H,CH ₃) [60%, 40%]**

*Protons of NH, NH₂ and OH groups are exchangeable by D₂O** d₆-DMSO: dimethylsulfoxide

*** 400 MHz

Table 3: Physical data of 5-benzoyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**III a-m**).

No.	R ¹	Yield %	M.P ^o Solvent of crys.	M.F/ M. Wt	Microanalysis Calculated/found			
					C %	H %	N %	S %
IIIa	H	81	245-47 E	C ₁₇ H ₁₄ N ₄ OS 322.39	63.33	4.38	17.38	9.95
					63.39	4.57	17.24	9.84
IIIb	CH ₃	82	263-65 E	C ₁₉ H ₁₇ N ₃ OS 345.41*	62.59	4.67	16.22	9.28
					63.08	3.87	16.65	9.79
IIIc	C ₂ H ₅	85	238-40 E	C ₁₉ H ₁₈ N ₄ OS 350.44	65.12	5.18	15.99	9.15
					65.02	5.44	16.05	8.88
III d	CH ₂ CH=CH ₂	72	218-20 E/W	C ₂₀ H ₁₈ N ₄ OS 362.45	66.28	5.01	15.46	8.85
					66.19	4.98	15.48	8.53
IIIe	C ₄ H ₉ (n)	79	165-67 E/W	C ₂₁ H ₂₂ N ₄ OS 378.49	66.64	5.86	14.80	8.47
					66.47	6.32	14.83	8.30
III f	C ₆ H ₁₁ (c)	77	240-42 E	C ₂₃ H ₂₄ N ₄ OS 404.53	68.29	5.98	13.85	7.93
					68.08	6.58	13.85	8.34
IIIg	C ₆ H ₅	71	185-87 E	C ₂₃ H ₁₈ N ₄ OS 398.48	69.32	4.55	14.06	8.05
					69.26	4.12	13.57	7.87
IIIh	o-CH ₃ -C ₆ H ₄	75	205-07 E	C ₂₄ H ₂₀ N ₄ OS 412.51	69.88	4.89	13.58	7.77
					69.62	4.18	13.58	7.90
IIIi	m-CH ₃ -C ₆ H ₄	77	180-82 E	C ₂₄ H ₂₀ N ₄ OS 412.51	69.88	4.89	13.58	7.77
					58.80	4.46	13.19	7.68
IIIj	p-CH ₃ -C ₆ H ₄	79	195-97 E	C ₂₄ H ₂₀ N ₄ OS 412.51	69.88	4.89	13.58	7.77
					69.12	5.47	13.51	7.75
IIIk	p-OCH ₃ -C ₆ H ₄	65	120-22 E	C ₂₄ H ₂₀ N ₄ O ₂ S 437.51	65.89	4.84	12.80	7.33
					66.00	5.03	12.65	7.17
III l	p-F-C ₆ H ₄	68	225-27 E	C ₂₃ H ₁₇ FN ₄ OS 416.47	66.33	4.11	13.45	7.70
					65.40	3.81	13.84	7.83
III m	o-Cl-C ₆ H ₄	67	182-84 E	C ₂₃ H ₁₇ ClN ₄ OS 450.93**	61.26	3.80	12.42	7.11
					59.57	3.65	14.39	7.69

* Contain 0.5 molecule of water

** Contain one molecule of water

E: Ethanol

E/W: Ethanol/Water (2:1)

Table 4: $^1\text{H-NMR}$ data of 5-benzoyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**III a-m**).

No	R ¹	$^1\text{H NMR}$ (δ ppm in CDCl_3)*
IIIa	H	9.13 (d, 1H, H ₂ of quinoline); 8.96 (br. s, 1H, N ² HCS); 8.5 (br. s, 1H, OH); 8.2-7.4 (m, 11H, H _{3,4,6,7} of quinoline, C-Ph, NH ₂)
IIIb	CH ₃	9.16 (d, 1H, H ₂ of quinoline); 8.6 (br. s, 1H, OH); 8.93-8.60 (br. m, 2H, OH, N ² HCS); 8.3-7.5 (m, 10H, H _{3,4,6,7} of quinoline, C-Ph, N ⁴ HCH ₃); 3.3 (d, 3H, NHCH ₃)
IIIc	C ₂ H ₅ ***	8.83 (dd, 1H, H ₂ of quinoline); 8.41 (br. s, 1H, OH); 7.75 (dd, 1H, H ₄ of quinoline); 7.70 (br. s, 1H, N ² HCS); 7.48-7.24 (m, 9H, H _{3,6,7} of quinoline, C-Ph, N ⁴ HCH ₂); 3.77 (m, 2H, CH ₂ CH ₃); 1.34 (t, 3H, CH ₂ CH ₃)
III d	CH ₂ CH=CH ₂	9.0 (d, 1H, H ₂ of quinoline); 8.6 (br. S, 1H, OH); 8.2-7.23 (m, 11H, N ² HCS, H _{3,4,6,7} of quinoline, C-Ph, N ⁴ HCH ₂); 6.56-5.83 (m, 1H, CH=CH ₂); 5.46 (t, 2H, CH=CH ₂); 4.5 (t, 2H, NHCH ₂)
IIIe	C ₄ H ₉ (n)	9.15 (d, 1H, H ₂ of quinoline); 8.76 (br. s, 1H, OH); 8.30-7.40 (m, 11H, N ² HCS, H _{3,4,6,7} of quinoline, C-Ph, N ⁴ HCH ₂); 3.90 (q, 2H, NHCH ₂); 2.10-1.30 (m, 4H, CH ₂ CH ₂ CH ₃); 1.10 (t, 3H, CH ₂ CH ₃)
III f	C ₆ H ₁₁ (c)	9.00 (d, 1H, H ₂ of quinoline); 8.56 (br. s, 1H, OH); 8.20-7.20 (m, 11H, N ² HCS, H _{3,4,6,7} of quinoline, C-Ph, N ⁴ H-cyclohexyl); 4.80-4.06 (m, 1H, NHCH of cyclohexyl); 2.56-1.00 (m, 10H, (CH ₂) ₅ of cyclohexyl)
IIIg	C ₆ H ₅	10.2 (br. s, 1H, N ² HCS); 9.15(d, 1H, H ₂ of quinoline); 8.96 (br. s, 1H, OH); 8.20-7.10 (m, 15H, N ⁴ H-phenyl, H _{3,4,6,7} of quinoline, C-Ph, NHC ₆ H ₅)
IIIh	o-CH ₃ -C ₆ H ₄	9.66 (br. s, 1H, N ² HCS); 9.15(d, 1H, H ₂ of quinoline); 8.95 (br. s, 1H, OH); 8.30-7.10 (m, 14H, N ⁴ H- o.tolyl, H _{3,4,6,7} of quinoline, C-Ph, NHC ₆ H ₄); 2.56 (s, 3H, CH ₃ of o.tolyl).
IIIi	m-CH ₃ -C ₆ H ₄	10.65 (br. s, 1H, N ² HCS); 9.50 (br. s, 1H, N ⁴ H-m.tolyl); 9.20 (d, 1H, H ₂ of quinoline); 8.20-6.90(m, 14H, OH, H _{3,4,6,7} of quinoline, C-Ph, NH C ₆ H ₄); 2.56 (s, 3H, CH ₃ of m.tolyl)**
IIIj	p-CH ₃ -C ₆ H ₄	9.7 (br. s, 1H, N ² HCS); 9.06 (d, 1H, H ₂ of quinoline); 8.83 (s, 1H, OH); 8.56-7.20 (m, 14H, N ⁴ H- p.tolyl, H _{3,4,6,7} of quinoline, C-Ph, NHC ₆ H ₄); 2.43 (s, 3H, CH ₃ of p. tolyl)
IIIk	p-CH ₃ O-C ₆ H ₄	9.93 (br. s, 1H, N ² HCS); 9.63 (br. s, 1H, N ⁴ H- p.methoxyphenyl); 9.16 (d, 1H, H ₂ of quinoline); 8.10-6.86 (m, 14H, OH, H _{3,4,6,7} of quinoline, C-Ph, NH C ₆ H ₄); 3.90 (s, 3H, OCH ₃)**
III l	p-F-C ₆ H ₄	10.15 (br. s, 1H, N ² HCS); 9.95(br. s, 1H, N ⁴ H- p.fluorophenyl); 9.20 (d, 1H, H ₂ of quinoline); 8.10-7.10 (m, 14H, OH, H _{3,4,6,7} of quinoline, C-Ph, NH C ₆ H ₄)**
III m	o-Cl-C ₆ H ₄	10.80 (br. s, 1H, N ² HCS); 10.13 (br. s, 1H, N ⁴ H- o. chlorophenyl); 9.65 (br. S, 1H, OH); 9.20 (d, 1H, H ₂ of quinoline); 8.60 (d, 1H, H ₄ of quinoline) 8.25-7.00 (m, 12H, OH, H _{3,6,7} of quinoline, C-Ph, NHC ₆ H ₄)**

* Protons of NH, NH₂ and OH groups are exchangeable by D₂O** d₆-DMSO: dimethylsulfoxide

*** 400 MHz

Table 5: Physical data of 5-acetyl-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene) hydrazone compounds (**IV a-l**).

No.	R ¹	Yield %	M.P ^o	M.F/ M.Wt	Microanalysis			
					Calculated/ found			
					C %	H %	N %	S %
IVa	H	58	243-45	C ₁₄ H ₁₂ N ₄ O ₂ S	54.40	3.91	18.12	10.60
				309.34*	53.54	3.41	17.85	10.37
IVb	CH ₃	61	270-72	C ₁₅ H ₁₄ N ₄ O ₂ S	57.31	4.49	17.82	10.20
				314.36	58.80	4.35	18.27	10.84
IVc	C ₂ H ₅	79	165-67	C ₁₆ H ₁₆ N ₄ O ₂ S	58.52	4.91	17.06	9.76
				328.39	58.32	5.04	17.03	9.58
IVd	CH ₂ CH=CH ₂	70	178-80	C ₁₇ H ₁₆ N ₄ O ₂ S	59.98	4.74	16.46	9.42
				340.40	59.41	4.72	16.44	9.43
IVe	C ₄ H ₉ (n)	71	165-67	C ₁₈ H ₂₀ N ₄ O ₂ S	60.65	5.66	15.72	9.00
				356.44	60.45	5.20	15.74	9.39
IVf	C ₆ H ₁₁ (c)	63	237-39	C ₂₀ H ₂₂ N ₄ O ₂ S	62.80	5.80	14.65	8.38
				382.48	62.50	5.36	14.59	8.53
IVg	C ₆ H ₅	65	247-49	C ₂₀ H ₁₆ N ₄ O ₂ S	63.81	4.28	14.88	8.52
				376.43	63.58	3.40	14.86	8.775
IVh	o-CH ₃ -C ₆ H ₄	63	230-32	C ₂₁ H ₁₈ N ₄ O ₂ S	64.60	4.65	14.35	8.21
				390.46	63.98	3.79	14.31	8.69
IVi	m-CH ₃ -C ₆ H ₄	58	210-12	C ₂₁ H ₁₈ N ₄ O ₂ S	64.60	4.65	14.35	8.21
				390.46	63.85	4.89	14.34	8.03
IVj	p-CH ₃ -C ₆ H ₄	60	250-52	C ₂₁ H ₁₈ N ₄ O ₂ S	64.60	4.65	14.35	8.21
				390.46	64.06	3.99	14.28	8.34
IVk	p-OCH ₃ -C ₆ H ₄	68	221-223	C ₂₁ H ₁₈ N ₄ O ₃ S	62.05	4.46	13.78	7.89
				406.46	61.26	4.55	13.75	7.90
IVl	p-F-C ₆ H ₄	66	205-210	C ₂₀ H ₁₅ FN ₄ O ₂ S	60.90	3.83	14.20	8.13
				394.42	59.98	3.72	14.06	8.23

* contain 0.5 molecule of water

Table 6: ¹H-NMR data of 5-acetyl-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene) hydrazone compounds (**IVa-l**).

No	R ¹	¹ H NMR (δ ppm in CDCl ₃)*
IVa	H	9.57 (d, 1H, H ₂ of quinoline); 9.10 (d, 1H, H ₄ of quinoline); 8.20-7.30 (m, 4H, OH, H _{3,6,7} of quinoline); 5.36 (br. s, 1H, NH); 3.90 (s, 2H, CH ₂ of thiazolidinone); 2.60 (s, 3H, CH ₃)
IVb	CH ₃	9.98 (d, 1H, H ₂ of quinoline); 9.30 (d, 1H, H ₄ of quinoline); 8.30-7.30 (m, 4H, OH, H _{3,6,7} of quinoline); 4.06 (s, 2H, CH ₂ of thiazolidinone); 3.33 (s, 3H, NCH ₃); 2.56 (s, 3H, CH ₃)
IVc	C ₂ H ₅	9.56 (d, 1H, H ₂ of quinoline); 9.00 (d, 1H, H ₄ of quinoline); 8.23 (br. S, 1H, OH); 7.99 (d, 1H, H ₆ of quinoline); 7.61 (dd, 1H, H ₃ of quinoline); 7.40 (d, 1H, H ₇ of quinoline); 4.00 (q, 2H, CH ₂ CH ₃); 3.90 (s, 2H, CH ₂ of thiazolidinone); 2.66 (s, 3H, CH ₃); 1.30 (t 3H, CH ₂ CH ₃)
IVd	CH ₂ CH=CH ₂	9.50 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 7.90-7.20 (m, 4H, OH, H _{3,6,7} of quinoline); 6.40-5.80 (m, 1H, CH=CH ₂) 5.46 (t, 2H, CH=CH ₂); 4.60 (d, 2H, NCH ₂); 2.86 (s, 2H, CH ₂ of thiazolidinone); 2.63 (s 3H, CH ₃)
IVe	C ₄ H ₉ (n)	9.50 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 7.90-7.25 (m, 4H, OH, H _{3,6,7} of quinoline); 4.00 (t, 2H, NCH ₂); 3.80 (s, 2H, CH ₂ of thiazolidinone); 2.60 (s, 3H, CH ₃); 2.10-1.20 (m, 4H, CH ₂ CH ₂ CH ₃) 1.00 (t, 3H, CH ₂ CH ₃)
IVf	C ₆ H ₁₁ (c)	9.70 (d, 1H, H ₂ of quinoline); 9.30 (d, 1H, H ₄ of quinoline); 8.20-7.40 (m, 4H, OH, H _{3,6,7} of quinoline); 4.50-4.16 (m, 1H, NCH of cyclohexyl); 4.00 (s, 2H, CH ₂ of thiazolidinone); 2.60 (s, 3H, CH ₃); 2.50–1.00 (m, 10H, (CH ₂) ₅ of cyclohexyl)
IVg	C ₆ H ₅	9.60 (d, 1H, H ₂ of quinoline); 9.10 (d, 1H, H ₄ of quinoline); 8.10-7.30 (m, 9H, OH, H _{3,6,7} of quinoline, NC ₆ H ₅); 4.10 (s, 2H, CH ₂ of thiazolidinone); 2.50 (s, 3H, CH ₃)
IVh	o-CH ₃ -C ₆ H ₄	9.40 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 8.30-7.00 (m, 8H, OH, H _{3,6,7} of quinoline, NC ₆ H ₄); 4.00 (s, 2H, CH ₂ of thiazolidinone); 2.33 (s, 3H, CH ₃); 2.26 (s, 3H, CH ₃ of o.tolyl)
IVi	m-CH ₃ -C ₆ H ₄	9.50 (d, 1H, H ₂ of quinoline); 9.00 (d, 1H, H ₄ of quinoline); 8.00-7.20 (m, 8H, OH, H _{3,6,7} of quinoline, NC ₆ H ₄); 4.10 (s, 2H, CH ₂ of thiazolidinone); 2.53 (s, 3H, CH ₃); 2.46 (s, 3H, CH ₃ of m.tolyl)
IVj	p-CH ₃ -C ₆ H ₄	9.43 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 7.96-7.16 (m, 8H, OH, H _{3,6,7} of quinoline, NC ₆ H ₄); 4.03 (s, 2H, CH ₂ of thiazolidinone); 2.50 (s, 3H, CH ₃); 2.43 (s, 3H, CH ₃ of p.tolyl)
IVk	p-OCH ₃ -C ₆ H ₄	9.33 (d, 1H, H ₂ of quinoline); 8.83 (d, 1H, H ₄ of quinoline); 8.13-6.96 (m, 8H, OH, H _{3,6,7} of quinoline, NC ₆ H ₄); 3.97 (s, 2H, CH ₂ of thiazolidinone); 3.90 (s, 3H, OCH ₃); 2.43 (s, 3H, CH ₃)
IVl	p-F-C ₆ H ₄	9.23 (d, 1H, H ₂ of quinoline); 8.90 (d, 1H, H ₄ of quinoline); 8.10-7.10 (m, 8H, OH, H _{3,6,7} of quinoline, NC ₆ H ₄); 4.30 (s, 2H, CH ₂ of thiazolidinone); 2.50 (s, 3H, CH ₃)**

* Protons of OH groups are exchangeable by D₂O** d₆-DMSO: dimethylsulfoxide

Table 7: Physical data of 5-benzoyl-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene) hydrazone compounds (**V a-f**).

No.	R ¹	Yield %	M.P ^o	M.F/ M.Wt	Microanalysis Calculated / found			
					C %	H %	N %	S %
Va	H	58	235-37	C ₁₉ H ₁₄ N ₄ O ₂ S 362.41	62.97	3.89	15.46	8.85
					62.77	3.39	15.40	9.16
Vb	CH ₃	66	177-79	C ₂₀ H ₁₆ N ₄ O ₂ S 376.43	63.81	4.28	14.88	8.52
					63.73	3.65	14.90	8.93
Vc	C ₂ H ₅	65	140-42	C ₂₁ H ₁₈ N ₄ O ₂ S 399.46*	63.14	4.54	14.03	8.03
					63.23	4.72	14.17	8.26
Vd	CH ₂ CH=CH ₂	69	197-99	C ₂₂ H ₁₈ N ₄ O ₂ S 402.47	65.65	4.51	13.92	7.97
					65.19	4.05	13.82	8.21
Ve	C ₄ H ₉ (n)	67	180-82	C ₂₃ H ₂₂ N ₄ O ₂ S 418.51	66.01	5.30	13.39	7.66
					65.87	4.89	13.38	8.00
Vf	C ₆ H ₁₁ (c)	62	250-52	C ₂₅ H ₂₄ N ₄ O ₂ S 444.55	67.54	5.44	12.60	7.21
					67.06	4.64	12.54	7.51

*contain 0.5 molecule of water

Table 8: ¹H-NMR data of 5-benzoyl-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene) hydrazone compounds (**V a-f**).

No	R ¹	¹ H NMR (δ ppm in CDCl ₃)*
Va	H	9.10 (d, 1H, H ₂ of quinoline); 8.10-7.20 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 4.00 (s, 2H, CH ₂ of thiazolidinone); 3.70 (s, 1H, NH)**
Vb	CH ₃	8.93 (d, 1H, H ₂ of quinoline); 8.20-7.20 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 3.86 (s, 2H, CH ₂ of thiazolidinone); 2.80 (s, 3H, CH ₃)
Vc	C ₂ H ₅	8.93 (d, 1H, H ₂ of quinoline); 8.20-7.20 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 3.89 (s, 2H, CH ₂ of thiazolidinone); 3.48 (q, 2H, CH ₂ CH ₃); 0.66 (t, 3H, CH ₂ CH ₃)
Vd	CH ₂ CH=CH ₂	9.00 (d, 1H, H ₂ of quinoline); 8.50-7.40 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 5.80-5.10 (m, 1H, CH=CH ₂); 4.80 (t, 2H, CH=CH ₂); 4.00 (d, 2H, NCH ₂); 3.90 (s, 2H, CH ₂ of thiazolidinone)
Ve	C ₄ H ₉ (n)	8.96 (d, 1H, H ₂ of quinoline); 8.50-7.40 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 3.80 (s, 2H, CH ₂ of thiazolidinone); 3.33 (t, 2H, NCH ₂); 1.50-0.70 (m, 4H, CH ₂ CH ₂ CH ₃); 0.50 (t, 3H, CH ₂ CH ₃)
Vf	C ₆ H ₁₁ (c)	9.10 (d, 1H, H ₂ of quinoline); 8.60-7.46 (m, 10H, OH, H _{3,4,6,7} of quinoline, C-Ph); 4.33-3.80 (m, 1H, NCH of cyclohexyl); 3.86 (s, 2H, CH ₂ of thiazolidinone); 1.90-0.50 (m, 10H, (CH ₂) ₅ of cyclohexyl)

* Protons of OH groups are exchangeable by D₂O

** d₆-DMSO: dimethylsulfoxide

Method: Agar cup diffusion method.^{40, 41}

(i) Preparation of the medium

Cultures were grown on nutrient agar medium of the following composition (g / L): Peptone 5 g, beef extract 3 g, NaCl 3 g, and agar agar 15 g while the tested fungal species were grown on sterilized sabouraud's dextrose agar of the following composition (g / L): Peptone 10 g, glucose 40 g, agar 20 g, and chloramphenicol 0.5 g (as a bacteriostatic agent). Streptomycin 1% solution and Cansten (Clotrimazol 1% solution) were used as positive controls for bacteria and fungi respectively. The media were inoculated at 121° and 1.5 atm. for 20 m, distributed in sterile plates (20 ml per plates) and allowed to solidify. The tested bacteria species were firstly grown in liquid culture for 48 h, and then 1 ml of each bacterial suspension was poured on the solidified agar medium and thoroughly distributed on the agar surface with a sterile L shape glass bar. Cups were made in the

solidified agar (6 / plate) with the aid of sterile cork borer, which were filled with 10 ul of the tested compounds. Five of these cups were devoted for the tested compounds, while the last one was left as control for the solvent.

(ii) Preparation of the solution of the tested compounds

The compounds were dissolved in DMSO and were tested at a concentration of 1% (w/v).

(iii) Procedure

An aliquot of 0.1 ml of each of the tested compound solution was pipetted into the appropriate cup; the last cup was used as control test for pure DMSO. The plates were left for one hour at room temperature to allow for prediffusion, then incubated at 37° for 48-96 hours and the inhibition zones around cavities were measured in mm. Results were recorded as the average of three readings in Tables 9-12.

Table 9: Antibacterial activity for 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**IIb-m**, **IIIa-m**) measured by inhibition zone test (mm).

No	R	R ¹	<i>M. luteus</i>	<i>S. aureus</i>	<i>P. aeruginosa</i>	<i>S. marscens</i>
IIb	CH ₃	CH ₃	-	-	-	8
IIc	CH ₃	C ₂ H ₅	-	-	-	10
IIf	CH ₃	C ₆ H ₁₁ (c)	-	-	-	7
IIg	CH ₃	C ₆ H ₅	-	-	-	7
IIh	CH ₃	o-CH ₃ -C ₆ H ₄	-	-	-	7
IIi	CH ₃	m-CH ₃ -C ₆ H ₄	-	-	-	10
IIj	CH ₃	p-CH ₃ -C ₆ H ₄	-	-	-	13
IIk	CH ₃	p-OCH ₃ -C ₆ H ₄	-	-	-	10
III	CH ₃	p-F-C ₆ H ₄	-	15	-	8
IIIm	CH ₃	o-Cl-C ₆ H ₄	-	15	-	10
IIIa	C ₆ H ₅	H	-	20	-	12
IIIb	C ₆ H ₅	CH ₃	-	-	-	12
IIIc	C ₆ H ₅	C ₂ H ₅	7	-	-	10
IIIg	C ₆ H ₅	C ₆ H ₅	-	-	-	8
IIIi	C ₆ H ₅	m-CH ₃ -C ₆ H ₄	-	-	-	8
IIIj	C ₆ H ₅	p-CH ₃ -C ₆ H ₄	-	-	-	8
IIIk	C ₆ H ₅	p-OCH ₃ -C ₆ H ₄	-	-	-	8
IIIl	C ₆ H ₅	p-F-C ₆ H ₄	10	-	-	8
IIIIm	C ₆ H ₅	o-Cl-C ₆ H ₄	7	-	-	8
Streptomycin			50	50	12	37

Table 10: Antibacterial activity for 5-acetyl (or 5-benzoyl)-8-hydroxy-quinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene)hydrazone compounds (**IVa-j**, **Va,c,d,f**) measured by inhibition zone test (mm).

No	R	R ¹	<i>M. luteus</i>	<i>S. aureus</i>	<i>P. aeruginosa</i>	<i>S. marcescens</i>
IVa	CH ₃	H	-	-	-	12
IVb	CH ₃	CH ₃	12	15	-	10
IVc	CH ₃	C ₂ H ₅	-	-	-	7
IVd	CH ₃	CH ₂ CH=CH ₂	-	-	-	7
IVe	CH ₃	C ₄ H ₉ (n)	-	-	-	10
IVf	CH ₃	C ₆ H ₁₁ (c)	10	10	-	-
IVg	CH ₃	C ₆ H ₅	8	-	-	8
IVh	CH ₃	o-CH ₃ -C ₆ H ₄	-	-	-	10
IVi	CH ₃	m-CH ₃ -C ₆ H ₄	-	-	-	8
IVj	CH ₃	p-CH ₃ -C ₆ H ₄	-	-	-	8
Va	C ₆ H ₅	H	20	20	-	-
Vc	C ₆ H ₅	C ₂ H ₅	-	-	-	8
Vd	C ₆ H ₅	CH ₂ CH=CH ₂	-	-	-	9
Vf	C ₆ H ₅	C ₆ H ₁₁ (c)	10	10	-	8
Streptomycin			50	50	12	37

Table 11: Antifungal activity for 5-acetyl (or 5-benzoyl)-8-hydroxy quinoline-4-substituted thiosemicarbazone compounds (**IIa-m**, **IIIa,g-j**) measured by inhibition zone test (mm).

No	R	R ¹	<i>C. albicans</i>	<i>T. rubrum</i>	<i>G. candidum</i>	<i>S. brevicaulis</i>
IIa	CH ₃	H	12	15	7	-
IIb	CH ₃	CH ₃	16	25	12	17
IIc	CH ₃	C ₂ H ₅	17	20	17	26
IId	CH ₃	CH ₂ CH=CH ₂	14	13	9	-
IIe	CH ₃	C ₄ H ₉ (n)	19	26	14	22
IIf	CH ₃	C ₆ H ₁₁ (c)	-	12	8	9
IIg	CH ₃	C ₆ H ₅	11	14	9	9
IIh	CH ₃	o-CH ₃ -C ₆ H ₄	9	15	8	-
IIi	CH ₃	m-CH ₃ -C ₆ H ₄	-	10	-	-
IIj	CH ₃	p-CH ₃ -C ₆ H ₄	10	17	-	-
IIk	CH ₃	p-OCH ₃ -C ₆ H ₄	10	20	-	-
III	CH ₃	p-F-C ₆ H ₄	12	21	7	6
IIIm	CH ₃	o-Cl-C ₆ H ₄	11	18	7	-
IIIa	C ₆ H ₅	H	9	13	9	-
IIIg	C ₆ H ₅	C ₆ H ₅	24	30	17	17
IIIh	C ₆ H ₅	o-CH ₃ -C ₆ H ₄	16	15	12	12
IIIi	C ₆ H ₅	m-CH ₃ -C ₆ H ₄	16	23	8	15
IIIj	C ₆ H ₅	p-CH ₃ -C ₆ H ₄	17	25	15	11
Clotrimazol			22	52	18	19

Table 12: Antifungal activity for 5-acetyl (or 5-benzoyl)-8-hydroxy-quinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene)-hydrazone compounds (**IVa,b,f** & **Va-e**) measured by inhibition zone test (mm).

No	R	R ¹	<i>C. albicans</i>	<i>T. rubrum</i>	<i>G. candidum</i>	<i>S. brevicaulis</i>
IVa	CH ₃	H	9	-	-	-
IVb	CH ₃	CH ₃	16	12	13	13
IVf	CH ₃	C ₆ H ₁₁ (c)	-	-	7	-
Va	C ₆ H ₅	H	9	-	8	-
Vb	C ₆ H ₅	CH ₃	-	-	6	-
Vc	C ₆ H ₅	C ₂ H ₅	-	-	8	-
Vd	C ₆ H ₅	CH ₂ CH=CH ₂	7	-	-	-
Ve	C ₆ H ₅	C ₄ H ₉ (n)	7	-	7	-
Clotrimazol			22	52	18	19

RESULTS AND DISCUSSION

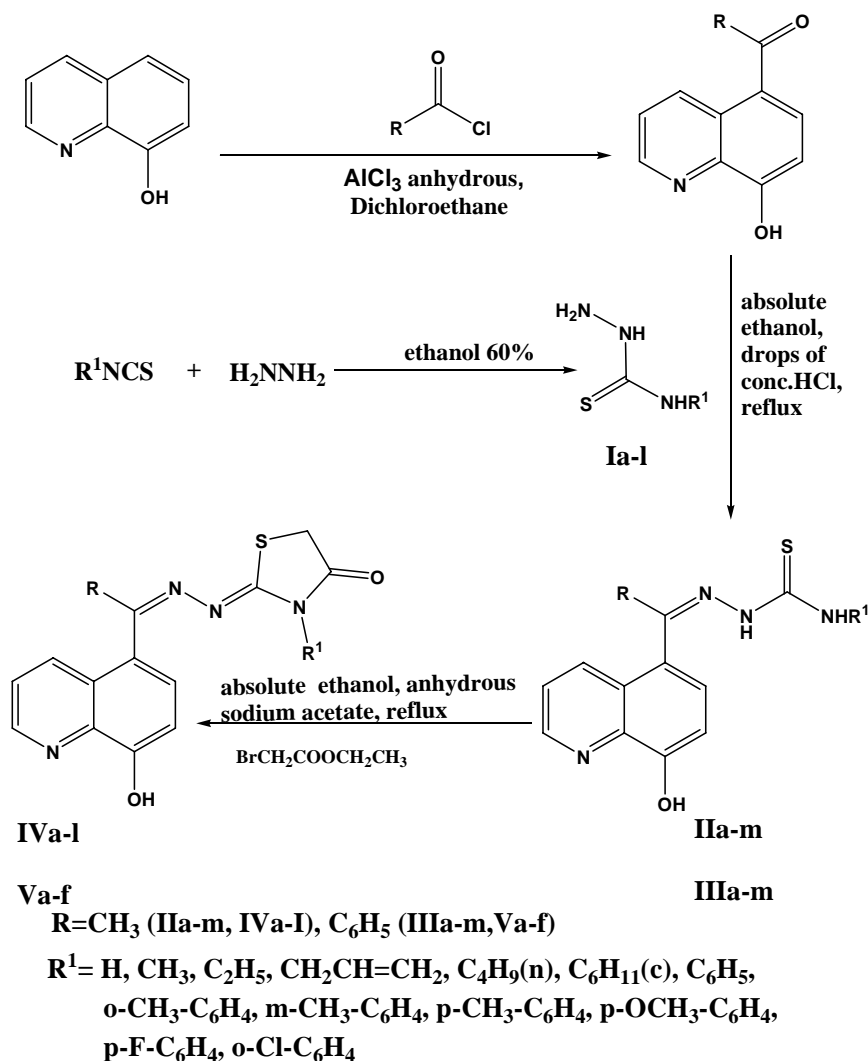
Chemistry

In the present investigation 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline were prepared by the reaction of 8-hydroxyquinoline with acetyl chloride or benzoyl chloride in the presence of anhydrous aluminum chloride as a catalyst using dichloroethane as a solvent under Friedel-Crafts acylation reaction condition.³⁹ The designed 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemi-carbazone compounds (**IIa-m**, **IIIa-m**) were prepared by the condensation of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline with an equimolar amount of thiosemicarbazide or the appropriate 4-substituted-3-thiosemicarbazides (**Ia-l**) in acidified ethanol under reflux for 2-8 hr. The IR spectra of such thiosemicarbazones lacked the band due to the carbonyl function of the starting 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline and showed bands due to NH functions at 3450-3340 cm⁻¹ and 3300-3200 cm⁻¹, the mixed vibrational coupling of the NCS moieties at 1540-1520 cm⁻¹, 1335-1320 cm⁻¹, 1180-1150 cm⁻¹, and 950-920 cm⁻¹, as well as a band at 1590-1580 cm⁻¹ characteristic for C=N and C=C function. In addition to a characteristic band at 3500-3300 cm⁻¹ for the stretching vibration of the OH group of 8-hydroxyquinoline. The ¹H-NMR data for 5-acetyl-8-hydroxyquinoline-4-substituted thiosemicarbazones (**IIa-m**) revealed the presence of E/Z geometric isomers although TLC showed that they turned out to be single isomer.

According to the ¹H-NMR spectra, compounds (**IIa-m**) appeared to be mixtures of unequal proportion of two isomers as predicted from the comparative measurements of the signal corresponding to CH₃ group of 5-acetyl-8-hydroxyquinoline. As a representative example, the mass spectrum of 5-acetyl-8-hydroxyquinoline-4-phenyl thiosemicarbazone (**IIg**) revealed the molecular ion peak M⁺ at m/z = 336, 5.7%.

5-Acetyl (or 5-Benzoyl)-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene) hydrazone compounds (**IVa-l**, **Va-f**) were prepared by the reaction of 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemicarbazones (**IIa-l**, **IIIa-f**) with an equimolar amount of ethyl bromoacetate in the presence of anhydrous sodium acetate and reflux in absolute ethanol. The IR spectra were characterized by some general features such as lack of the characteristic bands due to NH and NCS functions and exhibited the characteristic C=O band of the thiazolidinone ring at 1720-1700 cm⁻¹. In addition a band attributed to C=N stretching function at 1620-1590 cm⁻¹. Moreover, all compounds showed the characteristic band at 3500-3300 cm⁻¹ attributed to the OH stretching vibration of the 8-hydroxyquinoline.

The following scheme summarizes the sequences of the reactions involved for the preparation of the designed compounds.



Antimicrobial evaluation

In vitro antimicrobial screening: The prepared compounds as 1% (w/v) solution in DMSO were in vitro evaluated for antibacterial activity against Gram-positive bacteria (*Micrococcus luteus*, *Staphylococcus aureus*), Gram-negative bacteria (*Pseudomonas aeruginosa*, *Serratia marscens*) and for antifungal activity against *Candida albicans*, *Trichophyton rubrum*, *Geotrichum candidum*, and *Scopulariopsis brivicalis* using agar cup diffusion method.^{40,41} The zone of inhibition of the test compounds and the reference Streptomycin 1% (w/v) solution and Clotrimazole 1% (w/v) were measured. As a general feature the 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemicarbazones (**IIb-m**, **IIIa-m**) (Table 9) showed weak activities against *Serratia marscens* and without

significant effect against *Micrococcus luteus*, *Staphylococcus aureus* and *Pseudomonas aeruginosa* compared to Streptomycin and showed moderate to equal activity against fungi such as *Candida albicans*, *Trichophyton rubrum*, *Geotrichum candidum*, and *Scopulariopsis brivicalis* compared to Clotrimazole. As a general feature the 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-2-(3-substituted-4-oxothiazolidin-2-ylidene)hydrazones compounds (**IVa-j**, **Va,c,d,f**) (Table 10) were found to be displaying weak activities against *Serratia marscens* and without significant effect against *Micrococcus luteus*, *Staphylococcus aureus* and *Pseudomonas aeruginosa* compared to Streptomycin. On the other hand 5-acetyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**IIa-m**) were more

effective against fungi than 5-benzoyl-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**IIIa-m**) (Table 11). They showed scattered moderate activity against *Candida albicans* and *Geotrichum candidum*. In addition, 5-acetyl (or 5-benzoyl)-8-hydroxyquinoline-4-substituted thiosemicarbazone compounds (**IIa-l**, **IIIa-f**) were more effective against fungi than their corresponding thiazolidinones (compounds **IVa,b,f** & **Va-e**) (Table 12).

REFERENCES

- 1- R. Trave; Farmco Ed. Sci., 15, 468 (1960); through Chem. Abstr., 55, 2546b (1961).
- 2- Shah, J. K. Pravin and F. Dewhurst, Brit., 1, 314, 899 (1973); through Chem. Abstr., 79, 53062y (1973).
- 3- A. M. M. E. Omar, I. Chaaban, A. M. M. Hassan, K. A. A. Ismail and H. Abou-Shleib, Alex. J. Pharm. Sci., 1, 17 (1987).
- 4- A. M. M. E. Omar, I. C. Ahmed, O. M. Aboulwafa, A. M. Hassan, H. Abou-Shleib and K. A. Ismail; *ibid.*, 3, 211 (1989).
- 5- A. M. M. E. Omar, I. C. Ahmed, A. M. Hassan, O. M. Aboulwafa, H. Abou-Shleib and K. A. Ismail; *ibid.*, 3, 224 (1989).
- 6- A. M. M. E. Omar, I. C. Ahmed, A. M. Hassan, O. M. Aboulwafa, H. Abou-Shleib, and K. A. Ismail; *ibid.*, 4, 182 (1990).
- 7- T. Nishimura, H. Toku, K. Matsumoto, M. Iwata and T. Watanabe; Jpn. Kokai Tokkyo Koho; 79, 119, 029, 1979, through Chem. Abstr., 92, 53376h (1980).
- 8- T. Nishimura, Y. Miyamoto, K. Matsumoto and T. Watanabe; Jpn. Kokai Tokkyo Koho; 80, 15, 420, 1980, through Chem. Abstr., 93, 167685j (1980).
- 9- B.G. Benns, B.A. Gingras and C.H. Bayley, J. Appl. Microbiol, 8, 353, 1960, through Chem. Abstr., 55, 7548f (1961).
- 10- J. Bernstein, H. L. Yale, K. Losee, M. Holsing, J. Martins and W. A. Lott; J. Am. Chem. Soc., 73, 906 (1951).
- 11- G. Domagk, R. Behnisch, F. Mietzsch and H. Schmid, Naturwissenschaften, 33, 315, 1946, through, Goodman and Gilman's, the Pharmaceutical Basis of Therapeutics, 6th ed. P. Macmillan, New York 1980, p.1216.
- 12- W. J. Sydor, U. S. 3, 182, 082, 1965, through Chem. Abstr., 58, 1738h (1965).
- 13- F. Fujikawa, K. Hirai, T. Toyota, R. Tamada, S. Kijun, M. Naito and S. Tsukuma, Yakugaka Zasshi, 87, 844 (1967).
- 14- D. L. Klayman, J. F. Bartosevich, T. S. Griffin, C. J. Mason and J. P. Scovill, J. Med. Chem., 22, 855 (1979).
- 15- D. L. Klayman, J. P. Scovill, J. F. Bartosevich and C. J. Mason, Eur. J. Med. Chem., 16, 317 (1981).
- 16- D. L. Klayman, J. P. Scovill, J. F. Bartosevich and J. Bruce, J. Med. Chem., 26, 35 (1983).
- 17- J. P. Scovill, D.L. Klayman, C. Lambros, G. E. Childs and J. D. Notsch; *ibid.*, 27, 87 (1984).
- 18- V. K. Pandey and A. K. Agarwal, Acta Cienc. Indica, [Ser.] Chem., 6, 166, 1980, through Chem. Abstr., 94, 174983d (1981).
- 19- D. J. Bauer and P. W. Sadler, Brit., 975, 357, 1964, through Chem. Abstr., 62, 6462d (1965).
- 20- L. Heinisch, K. Kramarczyk, M. Tonew and G. Hesse, Pharmazie; 36, 259 (1981).
- 21- J. Easmon, G. Heinisch, W. Holzer and B. Rosenwirth, J. Med. Chem., 35, 3288 (1992).
- 22- Salwa M. H. Fahmy; Master Thesis in Pharmaceutical Science, Department of Pharmaceutical Chemistry, Faculty of pharmacy, University of Alexandria, (1988).
- 23- E. A. Coats, S. R. Milstein, M. A. Pleiss and J. A. Roesener, J. Med. Chem., 21, 804 (1978).
- 24- F.A. French and E.J. Blanz, J. Med. Chem., 9, 585 (1966).
- 25- E. J. Blanz, Jr. F.A. French, J.R. DoAmaral and D. A. French, *ibid.*, 13, 1124 (1970).
- 26- K. C. Agrawal, S. Clayman and A. C. Sartorelli, J. Pharm. Sci., 65, 297 (1976).
- 27- P. D. Mooney, B. Booth, E. C. Moore, K. C. Agrawal and A. C. Sartorelli, J. Med. Chem., 17, 1145 (1974).
- 28- K. C. Agrawal and A. C. Sartorelli, J. Pharm Sci., 57, 1948 (1968).
- 29- K. C. Agrawal, B. A. Booth, and A. C. Sartorelli, J. Med. Chem., 11, 700 (1968).
- 30- K. C. Agrawal, R. J. Cushley, W. J. McMurray and A. C. Sartorelli, *ibid.*, 13, 431 (1970).

- 31- F. A. French, E. J. Blanz, Jr., J. R. DoAmaral and D. A. French, *ibid.*, 13, 1117 (1970).
- 32- K. C. Agrawal, R. J. Cushley, S. R. Lipsky, J. R. Wheaton and A. C. Sartorelli; *ibid.*, 15, 192 (1972).
- 33- K. C. Agrawal, B. A. Booth, E. C. Moore and A. C. Sartorelli, *ibid.*, 15, 1154 (1972).
- 34- P. D. Mooney, B. A. Booth, E. C. Moore, K. C. Agrawal and A. C. Sartorelli, *ibid.*, 17, 1145 (1974).
- 35- Zeinab S. Farghaly; Master thesis, Assiut Univ., Faculty of Pharmacy, Assiut, Egypt (1985).
- 36- H. Y. Hassan, *Bull. Pharm. Sci., Assiut Univ.*, 22, Part 1, 97 (1999).
- 37- Sir Derek Barton and W. David Ollis, in "Comprehensive Organic Chemistry", P. G. Sammes (Ed.), Pergamon Press, 1979, Vol. 4, p.198.
- 38- W. Swhunack, K. Mayen and M. Haake, In *Arzneistoffe, Lehrbuch der Pharmazeutische Chemie* 12 Aufl. Vieweg Verlag, Braunscheig 533 (1984).
- 39- K. Matsumura, *J. Am. Chem. Soc.*, 52, 4433 (1930).
- 40- K. J. Kwon-Chung and J. E. Bennett, *Medical Mycology*, Lea and Febeiger Philadelphia, London, 1st ed., 1992, p. 866.
- 41- N. Iranpoor and F. Kazem, *Tetrahedron*, 54, 9475 (1998).