

ScIrP with ZrNiAl-type Structure

Ulrike Pfannenschmidt, Ute Ch. Rodewald,
and Rainer Pöttgen

Institut für Anorganische und Analytische Chemie, Universität Münster, Corrensstraße 30, 48149 Münster, Germany

Reprint requests to R. Pöttgen.
E-mail: pottgen@uni-muenster.de

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The phosphide ScIrP was synthesized from the elements in a bismuth flux and characterized by powder and single-crystal X-ray diffraction: ZrNiAl type, $P\bar{6}2m$, $Z = 3$, $a = 637.2(3)$, $c = 389.2(2)$ pm, $wR2 = 0.0280$, 250 F^2 values, 15 variables. The two crystallographically independent phosphorus sites have tricapped trigonal-prismatic metal coordination $P1Ir_3Sc_6$ and $P2Ir_6Sc_3$. The shortest interatomic distances occur for Ir–P (244–251 pm) within the 3D [IrP] network in which the scandium atoms fill cavities of coordination number 15 (4 Sc + 6 Ir + 5 P).

Key words: Phosphide, Scandium, Crystal Structure

Introduction

Ternary equiatomic phosphides $TT'P$ of the electron-poor (T) and electron-rich (T') transition metals have intensively been studied in the last 40 years with respect to their crystal structures [1–9] and physical properties [10–12]. Depending on the size and the electron count of the transition metal, these phosphides either crystallize with the orthorhombic TiNiSi [13] or the hexagonal ZrNiAl-type structure [14–16]. Very interesting magnetic properties have been reported for a series of compounds with chromium and manganese as transition metal components. Such phosphides and arsenides show magnetic ordering within the chromium or manganese substructure with the highest Curie temperature of 496 K for MnRuAs [12]. Another topic concerns superconductivity. The highest transition temperature of 13.3 K has been observed for ZrRuP [10, 11].

While many of such phosphides have been synthesized with the group IV and group V transition metals, with scandium so far only hexagonal ScCoP [5] and orthorhombic ScNiP [9] have been reported. During our recent flux-assisted synthesis of rare earth metal-

Table 1. Crystal data and structure refinement for ScIrP, space group $P\bar{6}2m$, $Z = 3$.

Refined composition	ScIrP
Formula weight, g mol ⁻¹	268.13
Unit cell dimensions, pm (Guinier data)	$a = 637.2(3)$ $c = 389.2(2)$
Cell volume, nm ³	0.1369
Calculated density, g cm ⁻³	9.76
Crystal size, μm^3	$30 \times 30 \times 40$
Transm. ratio (max/min)	1.86
Absorption coefficient, mm ⁻¹	76.9
Detector distance, mm	60
Exposure time, min	6
ω range; increment, deg	0–180; 1.0
Integr. param. A, B, EMS	13.2; 3.2; 0.012
$F(000)$, e	339
θ range for data collection, deg	3.6–34.8
Range in hkl	$\pm 10, \pm 10, \pm 6$
Total no. reflections	2189
Independent reflections / R_{int}	250 / 0.0511
Reflections with $I \geq 2 \sigma(I)$ / R_σ	246 / 0.0237
Data / ref. parameters	250 / 15
Goodness-of-fit on F^2	1.215
$R1$ / $wR2$ for $I \geq 2 \sigma(I)$	0.0135 / 0.0278
$R1$ / $wR2$ for all data	0.0142 / 0.0280
Flack parameter x	–0.01(2)
Extinction coefficient	0.0181(12)
Largest diff. peak / hole, e \AA^{-3}	1.49 / –1.76

iridium-phosphides [17, 18, and refs. therein], we obtained ScIrP. The synthesis and structure of this phosphide are reported herein.

Experimental Section

Synthesis

ScIrP was obtained from a bismuth flux [19]. Starting materials were pieces of scandium (Smart Elements, 99.999%), iridium powder (Heraeus, > 99.9%), red phosphorus (Hoechst, Knapsack, ultrapure), and bismuth shots (ABCR GmbH, > 99.99%). The elemental components in a molar ratio of 1 : 1 : 1 : 30 (Sc : Ir : P : Bi) were sealed in an evacuated silica tube. The ampoule was positioned in a muffle furnace, heated to 770 K at a rate of 50 K h⁻¹ and kept at that temperature for 24 h. Subsequently the temperature was raised to 1370 K at the same rate, and the sample was annealed at that temperature for 100 h followed by slow cooling to 970 K at a rate of 2 K h⁻¹ and further to 570 K at a rate of 4 K h⁻¹. The bismuth flux was slowly dissolved by a 1 : 1 molar mixture of H₂O₂ (ACROS, 35%) and glacial acetic acid (VWR International). The resulting sample was washed with demineralized water containing monoclinic IrP₂ as the main component and aggregates of pillar-shaped ScIrP crystals with metallic luster as reaction products. ScIrP can

Table 2. Atomic coordinates and anisotropic displacement parameters (pm²) for ScIrP. U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor. The anisotropic displacement factor exponent takes the form: $-2\pi^2[(ha^*)^2U_{11} + \dots + 2hka^*b^*U_{12}]$. $U_{23} = U_{13} = 0$.

Atom	Wyck. site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{eq}
Sc	3g	0.5773(3)	0	1/2	65(5)	52(6)	56(6)	26(3)	59(3)
Ir	3f	0.24809(5)	0	0	33(1)	38(1)	98(1)	19(1)	56(1)
P1	2c	2/3	1/3	0	46(6)	U_{11}	66(11)	23(3)	53(4)
P2	1b	0	0	1/2	92(9)	U_{11}	80(16)	46(5)	88(7)

also be obtained *via* flux-free synthesis from the elements, however, byproducts like IrP₂ and other yet unknown phases obtained thereby could not be removed so far.

EDX data

The single crystal investigated on the diffractometer was studied by EDX using a Zeiss EVO MA10 scanning electron microscope with Sc, Ir and GaP as standards for the semi-quantitative measurements. The analyses indicated Sc, Ir and P as the main components. Due to the significant overlap of the phosphorus K (2.1013 keV) and iridium M (1.977 keV) lines, a quantitative analysis was not possible. Other impurity elements (especially bismuth incorporation from the flux) were not observed.

X-Ray diffraction

The polycrystalline ScIrP sample was characterized by X-ray powder diffraction on a Guinier camera (equipped with a Fujifilm image plate system, BAS-1800) using $\text{CuK}\alpha_1$ radiation and α -quartz ($a = 491.30$, $c = 540.46$ pm) as an internal standard. The hexagonal lattice parameters were deduced from a least-squares refinement of the powder data. To ensure correct indexing, the experimental pattern was compared to a calculated one [20] using the positional parameters obtained from the structure refinement.

Pillar-shaped crystal fragments of ScIrP were separated from the flux-grown sample by mechanical fragmentation. The crystals were glued to quartz fibers using beeswax and were characterized by Laue photographs on a Buerger camera (white molybdenum radiation, image plate technique, Fujifilm, BAS-1800) in order to check their suitability for an intensity data collection. The data set was collected at room temperature by use of an IPDS II diffractometer (graphite-monochromatized $\text{MoK}\alpha$ radiation; oscillation mode). A numerical absorption correction was applied to the data set. All relevant crystallographic data and details of the data collection and evaluation are listed in Table 1.

Structure refinement

The structural relation of ScIrP with the ZrNiAl-type structure [14–16], space group $P\bar{6}2m$, was already evident from the X-ray powder diffraction data. Consequently we

Sc:	1	P2	269.4(2)
	4	P1	272.3(1)
	2	Ir	286.1(2)
	4	Ir	304.7(2)
	4	Sc	329.8(2)
Ir:	2	P1	244.1(1)
	2	P2	250.7(1)
	2	Ir	273.8(1)
	2	Sc	286.1(2)
	4	Sc	304.7(2)
P1:	3	Ir	244.1(1)
	6	Sc	272.3(1)
P2:	6	Ir	250.7(1)
	3	Sc	269.4(2)

Table 3. Interatomic distances (pm) in ScIrP. All distances within the first coordination spheres are listed.

used the structural parameters of isotypic ScAgSn [21] as starting values, and the structure was refined with anisotropic displacement parameters for all atoms with SHELXL-97 (full-matrix least-squares on F_o^2) [22]. All sites were fully occupied within two standard deviations. Refinement of the correct absolute structure was ensured through a calculation of the Flack parameter [23, 24]. The refinement smoothly converged to the residuals listed in Table 1. The atomic parameters and interatomic distances are listed in Tables 2 and 3.

Further details of the crystal structure investigation may be obtained from Fachinformationszentrum Karlsruhe, 76344 Eggenstein-Leopoldshafen, Germany (fax: +49-7247-808-666; e-mail: crysdata@fiz-karlsruhe.de, http://www.fiz-informationsdienste.de/en/DB/icsd/depot_anforderung.html) on quoting the deposition number CSD-422403.

Discussion

ScIrP crystallizes with the ZrNiAl [14–16] type structure. A projection of the structure is shown in Fig. 1. Striking structural motifs are two types of tri-capped trigonal prisms around the phosphorus atoms, *i. e.* P1Ir₃Sc₆ and P2Ir₆Sc₃. This coordination is typical for metal-rich phosphides. The shortest interatomic distances in the ScIrP structure occur for Ir–P (244–251 pm), slightly longer than the sum of the covalent radii for Ir + P of 236 pm [23]. Within the three-dimensional [IrP] network (Fig. 1) we observe also short Ir–Ir bonds (the triangles of the P1Ir₃Sc₆ prisms)

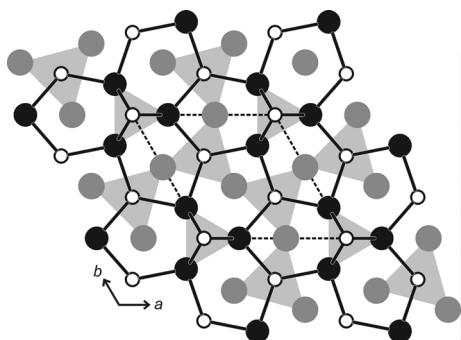


Fig. 1. Projection of the ScIrP structure onto the crystallographic xy plane. Scandium, iridium, and phosphorus atoms are drawn as medium-grey, black, and open circles, respectively. The three-dimensional [IrP] network and the trigonal-prismatic phosphorus coordination (light-grey shading) are emphasized.

of 274 pm, similar to those in *fcc* iridium (272 pm) [24]. As expected for a metal-rich phosphide, various Sc–Sc interactions also occur in the ScIrP structure. The shortest Sc–Sc distance of 330 pm compares well

with those in the structure of *hcp* scandium (6×325 and 6×331 pm) [24]. For further crystal chemical details on the broad family of ZrNiAl intermetallics we refer to review articles [25, 26, and refs. therein].

Finally we need to comment on the U_{33} displacement parameter of the iridium atoms, which is about three times higher than U_{11} . Such a displacement might point to superstructure formation, similar to ScAgSn [21] or ScPtSn [27]. We have carefully checked the reciprocal space with respect to weak superstructure reflections which would enlarge the unit cell. No additional reflections were detected. Magnetic susceptibility measurements revealed Pauli paramagnetism and gave no hint for superconductivity down to 2.5 K.

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