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Crystal structure of 4,6-diamino-2-sulfanylidene-1,2-dihydropyridine-3-carbonitrile

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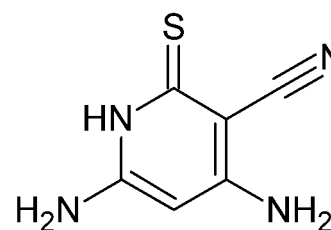
The title compound, C₆H₆N₄S, crystallizes with two independent molecules, *A* and *B*, in the asymmetric unit. Both independent molecules are almost planar [maximum deviations of 0.068 (6) Å in molecule *A* and 0.079 (6) Å in molecule *B*]. In the crystal, molecules *A* and *B* are linked by N—H...S, N—H...N and C—H...S hydrogen bonds, forming a three-dimensional network.

Keywords: crystal structure; polyfunctional pyridines; 3-cyanopyridine-2(1*H*)-thiones; hydrogen bonding.

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1. Related literature

For the synthesis of polyfunctional pyridines, see: Attaby *et al.* (1995); Asadov *et al.* (2003). For various biological properties of pyridine scaffold compounds, see: Abdel-Rahman *et al.* (2002); Rao *et al.* (2006). For the synthesis of 3-cyanopyridine-2(1*H*)-thiones, see: Fahmy & Mohareb (1986); Schmidt & Kubitzek (1960). For the use of 3-cyanopyridine-2(1*H*)-thiones in the synthesis of bio-active compounds, see: Taylor *et al.* (1983); Gangiee *et al.* (1991); Amr *et al.* (2003); Abu-Shanab *et al.* (2002); Awad *et al.* (1962); El-Gaby (2004); Miky & Zahkoug (1997); Guerrero *et al.* (1993); Krauze *et al.* (1999). For a similar crystal structure, see: Eyduran *et al.* (2007). For the synthesis of the title compound, see: Abu-Shanab (1999). For standard bond-length data, see: Allen *et al.* (1987).



2. Experimental

2.1. Crystal data

C₆H₆N₄S
M_r = 166.21
Orthorhombic, *Pca*2₁
a = 26.252 (8) Å
b = 4.3670 (14) Å
c = 12.523 (4) Å

V = 1435.7 (8) Å³
Z = 8
Mo *K*α radiation
μ = 0.38 mm⁻¹
T = 150 K
0.32 × 0.12 × 0.04 mm

2.2. Data collection

Bruker APEX 2000 CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2011)
T_{min} = 0.518, *T_{max}* = 0.928

11662 measured reflections
3412 independent reflections
2027 reflections with *I* > 2σ(*I*)
R_{int} = 0.137

2.3. Refinement

R[*F*² > 2σ(*F*²)] = 0.064
wR(*F*²) = 0.117
S = 0.87
3412 reflections
199 parameters
1 restraint
H-atom parameters constrained

Δρ_{max} = 0.42 e Å⁻³
Δρ_{min} = -0.34 e Å⁻³
Absolute structure: Flack (1983),
1573 Friedel pairs
Absolute structure parameter:
0.01 (13)

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...S1A ⁱ	0.88	2.44	3.293 (5)	163
N1A—H1A...S1 ⁱⁱ	0.88	2.80	3.579 (5)	149
N3—H31...N4A ⁱⁱⁱ	0.88	2.44	3.300 (8)	165
N3—H32...N2A ^{iv}	0.88	2.39	3.077 (8)	135
N3A—H33...S1A ⁱⁱ	0.88	2.53	3.392 (6)	168
N3A—H34...N2 ^v	0.88	2.20	2.981 (8)	148
N4—H41...S1A ⁱ	0.88	2.75	3.536 (6)	149
N4—H42...S1 ^{vi}	0.88	2.63	3.424 (6)	151
N4—H42...N2 ^{vi}	0.88	2.62	3.083 (9)	114
N4A—H44...S1 ⁱⁱ	0.88	2.53	3.353 (6)	157
C4—H4...S1 ^{vi}	0.95	2.74	3.551 (8)	143

Symmetry codes: (i) $-x + 1, -y + 1, z - \frac{1}{2}$; (ii) $-x + 1, -y + 2, z + \frac{1}{2}$; (iii) $x - \frac{1}{2}, -y + 1, z - 1$; (iv) $-x + \frac{1}{2}, y, z - \frac{1}{2}$; (v) $-x + \frac{1}{2}, y, z + \frac{1}{2}$; (vi) $-x + \frac{1}{2}, y - 1, z - \frac{1}{2}$.

Data collection: *SMART* (Bruker, 2011); cell refinement: *SAINTE* (Bruker, 2011); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HG5399).

References

- Abdel-Rahman, A., Bakhite, E. A. & Al-Laifi, E. A. (2002). *J. Chin. Chem. Soc.* **49**, 223–231.
- Abu-Shanab, F. A. (1999). *J. Chem. Res. (S)*, **7**, 430–431.
- Abu-Shanab, F. A., Elkholy, Y. M. & Elnagdi, M. H. (2002). *Synth. Commun.* **32**, 3493–3502.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Amr, A. E., Mohamed, A. M. & Ibrahim, A. A. (2003). *Z. Naturforsch. Teil B*, **58**, 861–868.
- Asadov, K. A., Burangulova, R. N., Guseninov, F. H., Gilmanov, R. Z. & Phaljachov, I. P. (2003). *Chem. Heterocycl. Compd*, **39**, 392–393.
- Attaby, F. A., Eldin, S. M. & Abdel-Razik, F. M. (1995). *Phosphorus Sulfur Silicon Relat. Elem.* **106**, 21–28.
- Awad, I. M. A., Abdel-rahman, A. E. & Bakhite, E. A. (1962). *Phosphorus Sulfur Silicon Relat. Elem.* **69**, 213–218.
- Bruker (2011). *SMART, SADABS and SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- El-Gaby, M. S. A. (2004). *J. Chin. Chem. Soc.* **51**, 125–134.
- Eyduran, F., Özyürek, C., Dilek, N., Ocak Iskeleli, N. & Şendil, K. (2007). *Acta Cryst.* **E63**, o2415–o2417.
- Fahmy, S. M. & Mohareb, R. M. (1986). *Tetrahedron*, **42**, 687–690.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Gangjee, A., Devraj, R. & Lin, F. (1991). *J. Heterocycl. Chem.* **28**, 1747–1751.
- Guerrera, F., Salerno, L. & Sarva, M. C. (1993). *Farmaco (Sci.)*, **48**, 1725–1733.
- Krauze, A., Germane, S. & Eberlins, O. (1999). *Eur. J. Med. Chem.* **34**, 301–310.
- Miky, J. A. A. & Zahkoug, S. A. (1997). *Nat. Prod. Sci.* **3**, 89–99.
- Rao, C. S., Venkaleswarlu, V. & Achaiah, G. (2006). *Bioorg. Med. Chem. Lett.* **16**, 2134–2138.
- Schmidt, U. & Kubitzek, H. (1960). *Chem. Ber.* **93**, 1559–1571.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Taylor, E. C., Palmer, D. C. & George, T. J. (1983). *J. Org. Chem.* **48**, 4852–4890.

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Crystal structure of 4,6-diamino-2-sulfanylidene-1,2-dihydropyridine-3-carbonitrile

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S1. Comment

Polyfunctional pyridines are highly reactive reagents that have been used extensively in heterocyclic synthesis (Attaby *et al.*, 1995; Asadov *et al.*, 2003). 3-Cyano-pyridine-2(1*H*)-thiones compounds (Fahmy & Mohareb 1986; Schmidt & Kubitzek, 1960) have also gained considerable interest due to their importance as intermediates for the synthesis of the biologically active deazafolic acid and deaza amino protein ring system (Taylor *et al.*, 1983; Gangiee *et al.*, 1991). In addition, 3-Cyano-2(1*H*)-pyridinethiones and their related compounds were found to be very reactive substances for the synthesis of many different heterocyclic systems which exhibited biological activities such as antibacterial, pesticidal, antifungal, acaricidal and neurotropic activities (Amr *et al.*, 2003; Abu-Shanab *et al.*, 2002; Awad *et al.*, 1962; El-Gaby, 2004; Miky & Zahkoug, 1997; Guerrero *et al.*, 1993; Krauze *et al.*, 1999; Abdel-Rahman *et al.*, 2002; Rao *et al.*, 2006). In light of these observations we report in this study the synthesis and crystal structure of the title compound.

In the title compound (Fig. 1), the asymmetric unit contains two independent molecules (A and B). Molecules A and B both are almost planar (Fig. 3), with the maximum deviations of -0.068 (6) Å for N4 in molecule A and 0.079 (6) Å for N2A in molecule B. The bond lengths of molecules A and B are comparable to those of the previously published structures (Eyduran *et al.*, 2007; Allen *et al.*, 1987).

In the crystal, the N—H \cdots S, N—H \cdots N and C—H \cdots S hydrogen bonds connect the molecules, forming a three dimensional network (Table 1, Fig. 2). In addition, C—H \cdots π interactions and π - π stacking interactions are not observed.

S2. Experimental

The title compound was prepared according to the reported method (Abu-Shanab, 1999). Crystals of the product were obtained in excellent yield (79%) and recrystallized from butanol to afford yellow needles (*M.p.* 583 K) in a sufficient quality for X-ray diffraction studies.

S3. Refinement

H-atoms were placed in calculated positions (C—H = 0.95 and N—H = 0.88 Å and were included as riding contributions with isotropic displacement parameters 1.2 times those of the attached atoms.

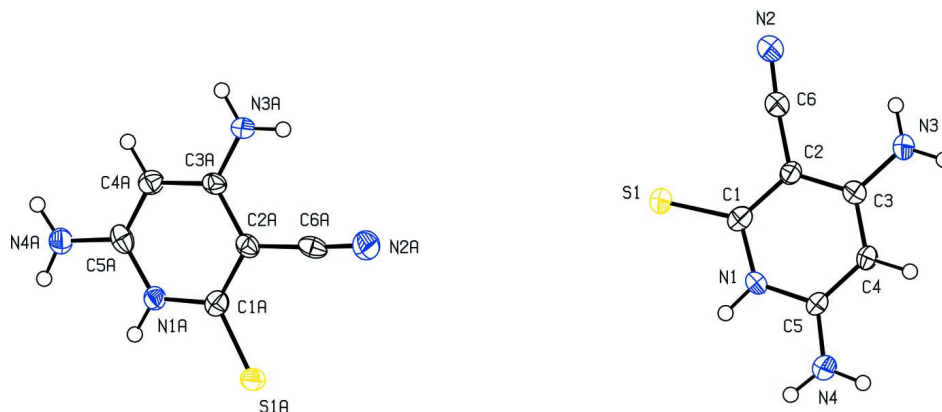


Figure 1

The title molecule showing the numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

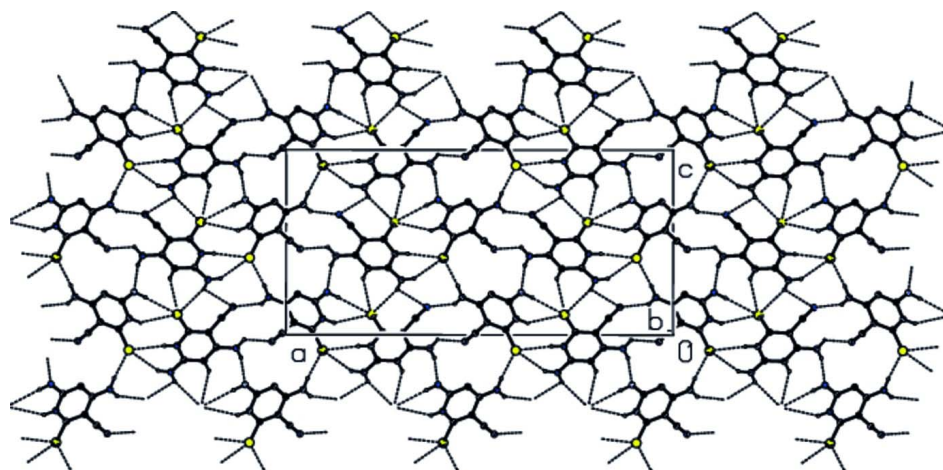


Figure 2

Packing viewed down the b axis showing the hydrogen bonding as dashed lines.

4,6-Diamino-2-sulfanylidene-1,2-dihydropyridine-3-carbonitrile

Crystal data

$C_6H_6N_4S$

$M_r = 166.21$

Orthorhombic, $Pca2_1$

Hall symbol: $P\ 2c\ -2ac$

$a = 26.252\ (8)\ \text{\AA}$

$b = 4.3670\ (14)\ \text{\AA}$

$c = 12.523\ (4)\ \text{\AA}$

$V = 1435.7\ (8)\ \text{\AA}^3$

$Z = 8$

$F(000) = 688$

$D_x = 1.538\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 748 reflections

$\theta = 2.3\text{--}23.4^\circ$

$\mu = 0.38\ \text{mm}^{-1}$

$T = 150\ \text{K}$

Needle, pale yellow

$0.32 \times 0.12 \times 0.04\ \text{mm}$

Data collection

Bruker APEX 2000 CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi and ω scans

Absorption correction: multi-scan
(*SADABS*: Bruker, 2011)

$T_{\min} = 0.518$, $T_{\max} = 0.928$

11662 measured reflections

3412 independent reflections

2027 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.137$
 $\theta_{\text{max}} = 28.7^\circ$, $\theta_{\text{min}} = 1.6^\circ$

$h = -34 \rightarrow 33$
 $k = -5 \rightarrow 5$
 $l = -16 \rightarrow 16$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.064$
 $wR(F^2) = 0.117$
 $S = 0.87$
 3412 reflections
 199 parameters
 1 restraint
 Hydrogen site location: inferred from
 neighbouring sites

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0282P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.42 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.34 \text{ e } \text{\AA}^{-3}$
 Absolute structure: Flack (1983), 1573 Friedel
 pairs
 Absolute structure parameter: 0.01 (13)

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.27967 (6)	0.5692 (5)	0.10895 (15)	0.0332 (6)
S1A	0.59399 (6)	0.7628 (4)	0.41599 (15)	0.0279 (5)
N1	0.28114 (18)	0.2124 (13)	-0.0607 (4)	0.027 (2)
N2	0.1365 (2)	0.4860 (15)	0.1504 (5)	0.039 (2)
N3	0.1288 (2)	0.0258 (14)	-0.0735 (5)	0.037 (2)
N4	0.2980 (2)	-0.1066 (15)	-0.2027 (5)	0.040 (2)
C1	0.2513 (3)	0.3403 (15)	0.0166 (5)	0.026 (2)
C2	0.1995 (2)	0.2719 (16)	0.0115 (5)	0.025 (2)
C3	0.1797 (2)	0.0835 (15)	-0.0684 (5)	0.025 (2)
C4	0.2122 (2)	-0.0471 (17)	-0.1416 (6)	0.030 (3)
C5	0.2637 (2)	0.0167 (17)	-0.1372 (6)	0.027 (3)
C6	0.1661 (2)	0.3946 (18)	0.0905 (6)	0.031 (3)
N1A	0.59425 (19)	1.1254 (12)	0.5874 (4)	0.0230 (17)
N2A	0.4628 (2)	0.4521 (15)	0.4637 (5)	0.038 (2)
N3A	0.45527 (19)	0.8585 (13)	0.7032 (4)	0.0327 (19)
N4A	0.6065 (2)	1.4593 (13)	0.7299 (4)	0.0277 (19)
C1A	0.5677 (3)	0.9133 (16)	0.5295 (5)	0.025 (2)
C2A	0.5201 (2)	0.8249 (16)	0.5670 (5)	0.023 (2)
C3A	0.5005 (3)	0.9504 (16)	0.6643 (5)	0.022 (2)
C4A	0.5300 (2)	1.1636 (15)	0.7195 (6)	0.026 (2)
C5A	0.5760 (2)	1.2555 (16)	0.6797 (5)	0.026 (2)
C6A	0.4896 (2)	0.6151 (16)	0.5089 (6)	0.025 (2)

H1	0.31370	0.25940	-0.06120	0.0320*
H4	0.19930	-0.18060	-0.19500	0.0360*
H31	0.11670	-0.09500	-0.12370	0.0450*
H32	0.10800	0.10930	-0.02670	0.0450*
H41	0.33050	-0.06070	-0.19550	0.0480*
H42	0.28820	-0.23400	-0.25310	0.0480*
H1A	0.62450	1.18140	0.56440	0.0270*
H4A	0.51800	1.24540	0.78520	0.0310*
H33	0.44400	0.93300	0.76400	0.0400*
H34	0.43690	0.72390	0.66770	0.0400*
H43	0.59690	1.53930	0.79120	0.0330*
H44	0.63580	1.51200	0.70130	0.0330*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0253 (9)	0.0437 (12)	0.0307 (10)	-0.0029 (9)	-0.0017 (8)	-0.0139 (10)
S1A	0.0223 (8)	0.0385 (10)	0.0229 (8)	0.0018 (9)	0.0020 (8)	0.0010 (9)
N1	0.016 (3)	0.037 (4)	0.027 (4)	-0.001 (3)	-0.001 (2)	-0.005 (3)
N2	0.026 (3)	0.054 (5)	0.037 (4)	0.002 (3)	-0.001 (3)	-0.011 (3)
N3	0.024 (3)	0.054 (4)	0.034 (4)	-0.004 (3)	-0.002 (3)	-0.020 (4)
N4	0.029 (3)	0.057 (5)	0.034 (4)	-0.009 (3)	0.004 (3)	-0.020 (3)
C1	0.031 (4)	0.025 (4)	0.021 (4)	0.003 (3)	0.000 (3)	-0.001 (3)
C2	0.023 (4)	0.031 (4)	0.021 (4)	0.005 (3)	-0.003 (3)	-0.006 (3)
C3	0.022 (3)	0.028 (4)	0.025 (4)	-0.002 (3)	0.001 (3)	-0.001 (4)
C4	0.025 (4)	0.034 (5)	0.030 (4)	-0.005 (3)	0.000 (3)	-0.017 (4)
C5	0.026 (4)	0.036 (5)	0.020 (4)	-0.003 (3)	0.001 (3)	-0.008 (3)
C6	0.023 (4)	0.040 (5)	0.031 (5)	0.001 (3)	-0.001 (3)	-0.007 (4)
N1A	0.022 (3)	0.027 (3)	0.020 (3)	-0.003 (3)	-0.001 (2)	0.002 (3)
N2A	0.035 (4)	0.042 (4)	0.036 (4)	-0.009 (3)	0.001 (3)	-0.003 (3)
N3A	0.021 (3)	0.049 (4)	0.028 (3)	-0.004 (3)	0.005 (3)	-0.012 (3)
N4A	0.026 (3)	0.032 (4)	0.025 (3)	-0.003 (3)	-0.002 (2)	0.000 (3)
C1A	0.028 (4)	0.023 (4)	0.023 (4)	0.004 (3)	-0.003 (3)	0.001 (3)
C2A	0.019 (3)	0.026 (4)	0.025 (4)	0.002 (3)	-0.002 (3)	0.004 (3)
C3A	0.018 (3)	0.026 (4)	0.023 (4)	0.006 (3)	0.000 (3)	0.003 (3)
C4A	0.027 (4)	0.025 (4)	0.026 (4)	0.002 (3)	0.001 (3)	-0.001 (3)
C5A	0.023 (4)	0.026 (4)	0.030 (4)	0.000 (3)	-0.008 (3)	0.009 (4)
C6A	0.020 (4)	0.024 (4)	0.032 (4)	0.006 (3)	0.005 (3)	0.007 (4)

Geometric parameters (Å, °)

S1—C1	1.700 (7)	N1A—C5A	1.374 (8)
S1A—C1A	1.711 (7)	N1A—C1A	1.367 (9)
N1—C1	1.365 (9)	N2A—C6A	1.150 (9)
N1—C5	1.363 (9)	N3A—C3A	1.345 (9)
N2—C6	1.151 (9)	C4—H4	0.9500
N3—C3	1.361 (7)	N4A—C5A	1.352 (8)
N4—C5	1.332 (9)	C1A—C2A	1.390 (9)

C1—C2	1.394 (9)	N1A—H1A	0.8800
N1—H1	0.8800	C2A—C3A	1.432 (9)
C2—C3	1.396 (9)	C2A—C6A	1.418 (9)
C2—C6	1.426 (9)	C3A—C4A	1.394 (10)
C3—C4	1.376 (9)	N3A—H34	0.8800
N3—H31	0.8800	N3A—H33	0.8800
N3—H32	0.8800	C4A—C5A	1.367 (8)
N4—H41	0.8800	N4A—H43	0.8800
N4—H42	0.8800	N4A—H44	0.8800
C4—C5	1.382 (8)	C4A—H4A	0.9500
C1—N1—C5	124.2 (5)	C5—C4—H4	120.00
S1—C1—N1	118.1 (6)	S1A—C1A—N1A	119.7 (5)
S1—C1—C2	125.8 (5)	S1A—C1A—C2A	122.5 (5)
N1—C1—C2	116.1 (6)	N1A—C1A—C2A	117.9 (6)
C1—N1—H1	118.00	C1A—N1A—H1A	118.00
C5—N1—H1	118.00	C5A—N1A—H1A	118.00
C1—C2—C3	121.5 (6)	C1A—C2A—C3A	120.3 (6)
C1—C2—C6	119.2 (6)	C1A—C2A—C6A	120.9 (6)
C3—C2—C6	119.3 (5)	C3A—C2A—C6A	118.8 (6)
N3—C3—C2	120.6 (5)	N3A—C3A—C2A	120.8 (6)
C3—N3—H31	120.00	C3A—N3A—H33	120.00
C3—N3—H32	120.00	C3A—N3A—H34	120.00
H31—N3—H32	120.00	H33—N3A—H34	120.00
C2—C3—C4	119.4 (5)	C2A—C3A—C4A	118.6 (6)
N3—C3—C4	120.0 (6)	N3A—C3A—C4A	120.7 (6)
C5—N4—H42	120.00	C5A—N4A—H44	120.00
C5—N4—H41	120.00	C5A—N4A—H43	120.00
C3—C4—C5	119.8 (6)	C3A—C4A—C5A	120.4 (7)
H41—N4—H42	120.00	H43—N4A—H44	120.00
N1—C5—N4	117.3 (5)	N1A—C5A—N4A	117.2 (5)
N4—C5—C4	123.8 (7)	N4A—C5A—C4A	123.2 (6)
N1—C5—C4	118.9 (6)	N1A—C5A—C4A	119.6 (6)
N2—C6—C2	175.5 (7)	N2A—C6A—C2A	176.7 (6)
C1A—N1A—C5A	123.3 (5)	C3A—C4A—H4A	120.00
C3—C4—H4	120.00	C5A—C4A—H4A	120.00
C5—N1—C1—S1	177.7 (5)	C5A—N1A—C1A—S1A	-179.5 (5)
C5—N1—C1—C2	-3.0 (10)	C5A—N1A—C1A—C2A	-0.5 (10)
C1—N1—C5—C4	3.5 (10)	C1A—N1A—C5A—C4A	2.6 (9)
C1—N1—C5—N4	-175.4 (6)	C1A—N1A—C5A—N4A	179.1 (6)
S1—C1—C2—C6	-1.5 (10)	S1A—C1A—C2A—C6A	-3.5 (10)
S1—C1—C2—C3	179.3 (5)	S1A—C1A—C2A—C3A	178.0 (5)
N1—C1—C2—C3	0.0 (10)	N1A—C1A—C2A—C3A	-1.1 (10)
N1—C1—C2—C6	179.2 (6)	N1A—C1A—C2A—C6A	177.5 (6)
C1—C2—C3—N3	-178.3 (6)	C1A—C2A—C3A—N3A	-177.7 (6)
C6—C2—C3—C4	-176.9 (7)	C6A—C2A—C3A—C4A	-178.1 (6)
C1—C2—C3—C4	2.4 (10)	C1A—C2A—C3A—C4A	0.5 (10)

C6—C2—C3—N3	2.4 (10)	C6A—C2A—C3A—N3A	3.7 (10)
N3—C3—C4—C5	178.8 (7)	N3A—C3A—C4A—C5A	179.8 (6)
C2—C3—C4—C5	-1.9 (10)	C2A—C3A—C4A—C5A	1.6 (10)
C3—C4—C5—N4	177.9 (7)	C3A—C4A—C5A—N4A	-179.4 (6)
C3—C4—C5—N1	-0.9 (11)	C3A—C4A—C5A—N1A	-3.2 (10)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...S1A ⁱ	0.88	2.44	3.293 (5)	163
N1A—H1A...S1 ⁱⁱ	0.88	2.80	3.579 (5)	149
N3—H31...N4A ⁱⁱⁱ	0.88	2.44	3.300 (8)	165
N3—H32...N2A ^{iv}	0.88	2.39	3.077 (8)	135
N3A—H33...S1A ⁱⁱ	0.88	2.53	3.392 (6)	168
N3A—H34...N2 ^v	0.88	2.20	2.981 (8)	148
N4—H41...S1A ⁱ	0.88	2.75	3.536 (6)	149
N4—H42...S1 ^{vi}	0.88	2.63	3.424 (6)	151
N4—H42...N2 ^{vi}	0.88	2.62	3.083 (9)	114
N4A—H44...S1 ⁱⁱ	0.88	2.53	3.353 (6)	157
C4—H4...S1 ^{vi}	0.95	2.74	3.551 (8)	143

Symmetry codes: (i) $-x+1, -y+1, z-1/2$; (ii) $-x+1, -y+2, z+1/2$; (iii) $x-1/2, -y+1, z-1$; (iv) $-x+1/2, y, z-1/2$; (v) $-x+1/2, y, z+1/2$; (vi) $-x+1/2, y-1, z-1/2$.