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# **ULTRAVIOLET LINEAR DICHROISM STUDIES OF INDENE**

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# Abstract

Ultraviolet Linear Dichroism (LD) spectra of the indene molecule in stretched polyethylene (PE) film are reported. Differently polarized electronic transitions are resolved in the LD spectrum. The first electronic transition at  $\lambda_{max} = 285$  nm, which is hidden by the second one in solution, is clearly isolated in LD spectra and polarized along the long axis of the molecule. The second one which is located at about  $\lambda_{max} = 250$  nm is polarized along the short axis. The calculated spectrum of indene shows the same trends in complete agreement with experimental findings.

Keywords : Ultraviolet Absorption Spectra ; Linear Dichroism ; Streched polyethylene ; Electronic Transitions ; CNDO/M Semiempirical Calculations.

# 1. Introduction

Indene (Fig 1) has been the subject of numerous theoretical and experimental papers [1-14] because of its chemical and biological importance. This compound is a prototype of systems in which  $\pi$ electrons in the ethylenic bond and the phenyl ring have strong interactions. Moreover, the evolution of the hyperconjugative ratio of CH<sub>2</sub> group determines the main physicochemical properties of indene [15]. The study of the electronic spectrum and spectroscopic properties of this molecule provides valuable information on its photochemical reactivity. With the aim to characterize correctly the first electronic transitions, we have studied the absorption spectrum of indene with respect to the polarization directions of UV transitions using linear dichroism in an anisotropic host of stretched polyethylene. Such measurements give information about the number, position and polarization of the electronic transitions [16-17]. On the other side, the CNDO/M semiempirical calculations [18] are applied to elucidate the assignments of the UV absorption spectrum.

### 2. Materials and methods

#### 2.1 Chemicals

Indene was purchased from FLUKA and used without further purification. It was introduced into polyethylene (PE) sheet from solution in butylchloride (FLUKA; spectral quality). The film was stretched four times from its starting length.



Figure 1: Structure of indene.

#### 2.2 Polarized UV measurements

The spectra were performed on a CARY 118 spectrophotometer equipped with two rotatable Glan prism polarizers in the sample and reference beams.

## 2.3 Linear Dichroism (LD)

LD is defined as : LD =  $A_{//} - A_{\perp}$ 

Where  $A_{//}$  and  $A_{\perp}$  are the absorbances measured with plane polarized light parallel (//) and perpendicular ( $\perp$ ) to the unique sample axis (the stretching direction of the film).

The reduced linear dichroism is defined as :

$$LD^{r} = LD/A_{is}$$

where  $A_{iso}$  is the absorbance of a corresponding isotropic sample.

For an uniaxial sample such as the stretched polyethylene film, the isotropic absorbance can be calculated from the polarized components as [19] :  $A_{iso} = 1/3 (A_{//} + 2A_{\perp})$ 

the orientation  $K_i$  factor [20-23] which describes the middle alignment of the transition moment  $M_i$ is defined as :  $K_i = \langle \cos^2(M_i; u) \rangle$ 

Where  $(M_i; u)$  is the angle between the transition moment  $M_i$  and the streching direction u, the

angle brackets indicate averaging over all molecules under observation.

The determination of the orientation  $K_i$  factor is dependent on the nature of the spectra. In the UV absorption spectra where the bands are broad and overlap, the trial and error method (TEM) [20–23] is used in order to reduce the uncertainty of the  $K_i$ value. This method is based on the stepwise formation of linear combinations of polarized spectra;  $r_k (\lambda) [21, 24, 25]$  is written as :  $r_k (\lambda) = (1-K) A_{l/} (\lambda) - 2 KA_{\perp} (\lambda)$ 

The transition moment having the highest K value is the best aligned with the long molecular axis (L) which is normal to the smallest cross section of the molecule and the best aligned one with the stretching direction u [20]. The axis, which is the worst aligned with the stretching direction, coincides with one of the short axes of the molecule (i.e the in-plane axis or the axis perpendicular to the plane of chromophore). Furthermore, the transition moment with the lowest K value is, on the average, the worst aligned with the stretching direction.

#### 2.4 CNDO/ M Calculations

Theoretical semiempirical calculations using the CNDO/M method have been performed to characterize the UV transitions on indene at the molecular geometry reported in the literature [26]. Sixty singly excited configurations were determined making use of the configuration interaction treatment implemented in the CNDO/M procedure to explore excited states.

## 3. Results and discussion

In the UV observed spectra of indene in two solvents with different polarity, hexane and ethanol (Fig 2), a strong band located about 250 nm and a weak absorption between 280 and 290 nm have been observed in both solvents.



**Figure 2 :** Absorption spectra of indene in hexane (...) and ethanol (—) at room temperature,  $(C=10^{-4})$ 

The absorbance curves  $A_{//}(\lambda)$  and  $A_{\perp}(\lambda)$  (Fig3) are resolved better than those obtained in solutions. Furthermore, the LD<sup>r</sup> spectrum which permits the determination of the number of electronic transitions in the absorption spectrum [17-27], is resolved into two components in the spectral range of 230-300 nm.



**Figure 3 :** Absorbance curves  $A_{//}$ ;  $A_{\perp}$  and  $LD^r$  spectrum for indene aligned in stretched PE film at room temperature .

However, in this interval the LD<sup>r</sup> curve shows two main values : 0.425 and 0.362. Accordingly, the strong band observed in the UV spectra is the envelope of two pure  $\pi$ - $\pi$ \* electronic transitions with distinct polarization [27].

The transition moment directions relative to the molecular orientation axis are determined by the linear combination  $r_k$  ( $\lambda$ ) [20-22], where the K value is systematically varied from 0 to 1 in steps of 0.1 [24, 25] (Fig 4).



**Figure 4 :** Stepwise spectral reduction.  $r_k (\lambda) = (1-K) A_{//} (\lambda) - 2K A_{\perp}(\lambda)$  for indene. K is varied between 0 and 1 in steps of 0.1.

- These series of linear combinations are constructed in such a way that the contribution of the  $j^{th}$  transition at the moment  $M_i$  vanishes when

 $K = K_i$  in the  $r_k(\lambda)$  linear combination. The L<sub>b</sub> and L<sub>a</sub> electronic transitions which are located about 285 and 250 nm , have orientation factors  $K_1 =$ 0.51 and  $K_2 = 0.38$ , respectively. It can be concluded that the polarization directions of the transitions close to 285 and 250 nm form tow different angles with the orientation axis, one relatively small and the other relatively large. These LD measurements do not allow to obtain any precise experimental values of the angles  $\theta_i$ between the transition moments M<sub>i</sub> and the long axis of indene. Therefore, the observed K<sub>i</sub> values only provide qualitative information on the alignment of the transition moments M<sub>i</sub> with respect to the long axis. We report in figure 5 the most probable zone where the polarization direction of the two electronic transitions of indene might be situated.

The CNDO/M method predicts two  $\pi$ - $\pi$ \* electronic transitions in the low energy region for indene with the L<sub>b</sub> and L<sub>a</sub> transitions encountered at 280.35 and 252.47 nm, respectively, so that the observed transitions may be assigned to the calculated ones unambiguously. The theoretical values of the oscillator strengths seem to agree with the relative magnitudes of experimental data. Furthermore, the calculation predicts that the energy of the transition  $L_b (S_0 \rightarrow S_1)$  is low and its transition moment directed along the long axis, while the second transition  $L_a$  (S<sub>0</sub> $\rightarrow$ S<sub>2</sub>) has the transition moment directed along the short axis. The calculated angle  $\theta_i$  between the transition moment M<sub>i</sub> and the long axis of the molecule is consistent with the observed orientation factor K<sub>i</sub>. However, without additional information precise values for the transition moment angles  $\theta_i$  cannot be determined from the observed orientation factors [24, 25].



Figure 5: The most probable transition moment directions concluded from the present dichroic study with an inaccuracy of  $10^{\circ}$ . The lines (L) and (S) are respectively, the approximated long and in-plan short axis of indene.

The simulation of the experimental results was carried out by means of CNDO/M semiempirical calculations. In table 1 the theoretical results are summarized and listed along with the experimental data.

Experimental data			CNDOM			
λ(nm)	Log e	Ķ,	$\lambda(nm)$	f	θi	Nature
285	0.038	0.51	280.35	0.0082	3°09	л-л*
250	0.831	0.38	252.47	0.4363	98°57	π-π*
210	0.438	0.42	210.81	0.2000	75°28	л-л*

Table 1 : The observed and calculated spectroscopicparameters of singlet-singlet  $\pi$ - $\pi$ \* transitions of indene. $K_i$  : Calculated orientation factor from LinearDichroism Measurements.

f : Oscillator strength.

 $\theta_i$  : Angle between the transition moment  $M_i$  and the long axis of indene.

# 4. Conclusion

The strong absorption band of indene chromophore is confirmed, by the analysis of the linear dichroism spectra with the help of CNDO/M semiempirical calculations, to be a superposition of two  $\pi$ - $\pi$ \* electronic transitions with different polarization. However, the electronic transitions at 285 and 250 nm are polarized along the 'long' and 'short in-plane' axes of indene, respectively.

### References

[1] E. T. Harrigan, N. Hirota, Chem. Phys. Lett., 1 (22) 1 (1973) 29-32.

[2] F. S. Wettack, R. Klapthor, A. Shedd, M. Koeppe, K. Janda, P. Dwyer, K. Stratton, NBS Special publication (U.S), 526 (1978) 60 – 62.

[3] J. D. Quirk, D. Abstr. Int B., 30 (9) (1970) 4013.

[4] B. Brocklehurst, D. N. Tawn, Spectrochimica Acta, (30A) (1974) 1807-1815.

[5] G. Buemi, Z. Phys. Chem. Neue Folge, Bd 119 S (1980) 1 – 21.

[6] V. Ramamurthy, J. V. Caspar, D. F. Eaton,
E. W. Kuo, D. R. Corbin, J. Am . Chem. Soc., 10 (114) (1992) 3882-3892.

[7] J. Kao, J. Am. Chem. Soc. 13 (109) (1987) 3817-3828.

[8] J. Kao, N.L. Allinger, J. Am . Chem. Soc., 4 (99) (1977) 975-986. [9] J. E. Bloor, M. Eckert-Maksic, M. Hodoscek, Z. B. Maksic, K. Poljanec, New. J. Chem., 17 (1993) 157-160.

[10] M. J. S. Dewar, E. Haselbach,
 S. D. Worley, Proc. Roy. Soc. Lond. A., 315, 431-442 (1970).

[11] O. Ouamerali, J. Gayoso, Int. J. Quant. Chem., 29 (1986) 1599-1624.

[12] A. Artford, J. R. Lombardi, J. Mol. Spectroscopy., 34 (1970) 257-265.

[13] J. M. Hollas, Spectrochimica Acta, (19) (1963) 753-767.

[14] Y. Maouche, B. Maouche, J. Chim. Physico-Chimie Biologie, 84 (7-8) (1987) 883-888.

[15] B. Pullman, G. Berthier, Bull. Soc. Chim. Fr., (1948) 551-554.

[16] S. Ait-Lyazidi, M. Dkaki, M. Hnach, J. P. Aycard, C. Cazeau-Dubroca, Spectrochimica Acta, 53 A (1997) 605-610.

[17] B. Albinsson, B. Norden, J. Phys. Chem., 96 (1992) 6204.

[18] R. L. Ellis, G. Kuehnlenz, H. H. Jaffe, Theor. Chim. Acta (Berlin), 26 (1972) 131 [19] J. Michel, E. W. Thulstrup, "Spectroscopy with Polarized Light Solute alignment by Photo selection in Liquid Crystals, Polymers and Membranes", VCH Publshers New york (1986).

[20] E. W. Thulstrup, J. Michel, "Elementary Polarization Spectroscopy", VCH Publishers New york (1989).

[21] E. Madsen, I. Terpager, K. Olskaer, J. Spanget-Larsen, Chem. Phys., 165 (1992) 351.

[22] J. Michel, E. W. Thulstrup, Accounts. Chem. Res., 20 (1987) 192.

[23] E. W. Thulstrup, J. Michel, Spectrochim. Acta, 44A (1988) 767

[24] J. Spanget-Larsen, J. Waluk, E. W. Thulstrup, J. Phys. Chem., 94 (1990) 1800.

[25] J. Spanget-Larsen, J. Waluk, S. Eriksson,E. W. Thulstrup, J. Am. Chem. Soc., 114 (1992)1942.

[26] J. F. Southern, Diss. Abstr. Int. B., 33 (8) (1973) 3586.

[27] B. Albinsson, M. Kubista, K. Sandros,B. Norden, J. Phys. Chem., 94 (1990) 4006.