

# **De-Icing Impacts on the Danforth Campus**

Sustainability Exchange Spring 2020



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#### BACKGROUND

#### What is a De-icer?

- De-icers are salts that lower the freezing point of water (making it harder to freeze and easier to melt)
- · Landscaping companies add de-icers to roads and sidewalks to improve walking conditions during the winter

#### How do De-icers Affect Soil and Groundwater?

- Increases amount of conductive species in the pore water of the soil. De-icer salts are ionic - they can easily conduct (transmit) electric current through them.
- · Commonly used salts for de-icing are chloride salts (NaCl, MgCl\_) which dissociate into positively and negatively charged ions. Chloride ions (Cl<sup>-</sup>) present in these salts are documented by the EPA 303 D list as a high-priority pollutant in St. Louis
- The concentration of ions in the soil's pore water can be estimated by conductivity measurements - measure of ability to pass electrical flow

#### How Can We Use Conductivity Measurements?

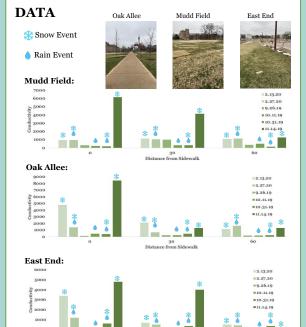
· They tell us how abundant ions are in the soil's pore water throughout the semester.



## **OBJECTIVES**

The overall goal is to understand the environmental impacts of de-icers on WashU's soil. We will accomplish this with the following objectives:

- 1) Expand on available data by sampling soil under different weather conditions and changes in conductivity
- 2) Improve the experimental design for soil sampling and testing
- 3) Develop application to help Focal Pointe track where has been de-iced



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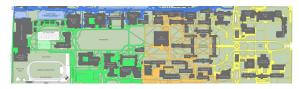
Distance from Sidewalk

### FINDINGS AND ANALYSIS

- · De-icers increase the conductivity in the soil pore water, due to an increase in the concentration of ionic species, e.g., sodium (Na<sup>+</sup>) & chloride ions ( Cl<sup>-</sup>)
- · Conductivity decreases with increasing distance from sidewalk. This suggests a decreasing concentration profile with the lowest de-icer concentrations farthest from the sidewalk
- · Conductivity decreases after rainfall events because the de-icer is diluted and pushed out of the pore water further into the subsurface
- · Overall implication: higher conductivity correlates to increasing ionic species in the soil which result in negative impacts to both plant growth and local water quality

## APP DEVELOPMENT

- · We are developing a beta version of an ArcGIS web app which will allow Focal Pointe to update progress on de-icing operations after a snow event
- We will be using Collector for ArcGIS in conjunction with ArcGIS Online



### RECOMMENDATIONS

- · Further study on how current de-icing practices impact soil and water quality in order to gain greater cross-seasonal understanding
- A pilot study of the ArcGIS app to inform further customization of the app to fit the specific practices and needs of Focal Pointe
- · Additional research on treatment technologies for in-situ reduction of salt adsorption by soil, e.g., biochar and gypsum mixture

## ACKNOWLEDGEMENTS

Dr. Avni Solanki, Cody Azotea, Chris Anderson, Focal Pointe, Dr. Bill Lowry, Bill Winston, Fall 2019 De-Icing Group, Washington University in St. Louis

## **METHODOLOGY**

#### Instrumentation:

Soil Testing Kit - measures pH levels at each of the 4 testing sites on Danforth campus

Pocket Pro+ - measures pH, temperature, TDS, salinity, and conductivity of all samples from each location

#### **Procedure:**

- 1. Sampling
- Use an auger to take 5 cm deep samples at 0 cm, 30 cm, and 60 cm away from the sidewalk at each location (Mudd Field, Oak Allee, and East End shown respectively in the map below)
- · Place Soil Testing Kit into soil once at each location and let stabilize for 10 minutes prior to recording data
- 2. Laboratory Analysis
- · Add 40 mL deionized water to 20 g of soil from each sample of all locations
- Stir with magnetic stir plate for 5 minutes then remove and let settle for 2 minutes

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· Use Pocket Pro+ to measure data from each sample and record



#### Spots lacking grass on Mudd Field (partially because of the conductive material from