# Identification and development of 2,3-dihydropyrrolo[1,2*a*]quinazolin-5(1*H*)-one inhibitors targeting bromodomains within the Switch/Sucrose Non-Fermenting complex

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ABSTRACT: Bromodomain containing proteins PB1, SMARCA4, and SMARCA2 are important components of SWI/SNF chromatin remodeling complexes. We identified bromodomain inhibitors targeting these proteins, which display unusual binding modes involving water displacement from the KAc binding site. The best compound binds the fifth bromodomain of PB1 with a  $K_D$  of 124 nM, SMARCA2B and SMARCA4 with  $K_{DS}$  of 262 and 417 nM respectively, and displays excellent selectivity over bromodomains other than PB1, SMARCA2, and SMARCA4.

#### INTRODUCTION

The structure of chromatin, the complex of proteins and DNA through which DNA is packaged, is regulated by the post-translational modification of histone tails and the action of chromatin remodeling complexes such as the Switch/Sucrose Non-Fermenting (SWI/SNF) complex. The multi-subunit SWI/SNF complex is based on the mutually exclusive helicase/ATPase proteins SMARCA2 and SMARCA4, with additional core subunits and regulatory or specificity proteins. Many components contain domains that recognize histone post-translational modifications, enabling cross-talk between the two mechanisms of regulation.

The SWI/SNF complex is subject to considerable research interest, since components are mutated in many cancers<sup>1-3</sup> and are implicated in developmental disorders.<sup>4</sup> In some tumor types, mutations within the SWI/SNF complex lead to context-specific vulnerabilities, such as dependence on SMARCA2/4 in SMARCB1 (Snf5) mutant rhabdoid tumors<sup>5</sup> or the essential presence of SMARCA2 in SMARCA4-mutant lung cancer.<sup>6,7</sup> The biological roles of the complexes has led to interest in generating chemical probes that target different domains with-in the multi-subunit complex, to assist the elucidation of their function and therapeutic relevance.<sup>8</sup> Bromodomain containing proteins (BCPs) are prevalent in SWI/SNF complexes and are present in SMARCA2/4 helicases, BRD7/9, and PB1. SMARCA2/4 and BRD7/9 contain one bromodomain each whilst PB1 has six distinct bromodomains. In yeast SWI/SNF

bromodomains are implicated in enhancing chromatin remodeling efficiency, loci targeting, and protein-protein interactions.<sup>9,10</sup> In eukaryotes the SMARCA4 bromodomain is reported to act in the DNA damage response.<sup>11,12</sup> Mutations in PB1's bromodomains have been found in clear cell renal carcinoma and linked to genome instability and aneuploidy.<sup>13,14</sup>

Bromodomains are protein modules of ~ 110 amino acids that recognize acetylated lysine (KAc) and share a common structure of four alpha helices linked by flexible loop regions and have been popular targets for chemical probe development.<sup>15,16</sup> 61 bromodomains are expressed in the human proteome, across 42 BCPs, which can be grouped into sub-families based on amino acid sequence and structural alignment.<sup>17</sup> Chemical probes have now been developed for multiple sub-families and have helped identify new biology.<sup>18,19</sup> In particular, compounds (+)JQ1 and IBET-151 have demonstrated the therapeutic potential of bromodomain and extra-terminal (BET) proteins in cancer, inflammation and heart failure.<sup>20–22</sup>

The chemical probes I-BRD9<sup>23</sup> and LP99,<sup>24</sup> targeting BRD9 and BRD7/BRD9 respectively, have been developed to study some bromodomains in the SWI/SNF complex. Recently a chemical probe, PFI-3, targeting the bromodomains in SMARCA2 and SMARCA4, as well as the fifth bromodomain of PB1 (PB1(5)), has been developed based on a salicyclic acid fragment hit, and been used to demonstrate a role for SMARCA4 in embryonic stem cell maintenance.<sup>25</sup> These results demonstrate that SWI/SNF BCPs are druggable, despite being predicted to be more challenging to target due to a featureless binding pocket lacking a clearly defined hydrophobic groove and ZA binding channel as in BRD4(1) (supporting information).<sup>17,26</sup> A short ZA loop and a wider entrance than BET bromodomains have been identified as factors that may alter ligand binding mode in the structurally related bromodomains PB1(5) and SMARCA2/4, whilst PB1(2-4) are defined by an aromatic residue in the pocket that is proposed to act as a 'lid'.<sup>26</sup> Given the varied structural nature of the six bromodomains in PB1, one single probe is unlikely to be able to target all simultaneously. Several compounds with different chemotypes may be required.

In the course of library screening, compound **1** and other related analogues were found to be weak PB1(5) inhibitors but with poor aqueous solubility. In this work we describe the optimization of **1**, leading to a family of compounds with increased potency, improved physicochemical properties, and varied selectivity within the SWI/SNF bromodomains, including the potential to target PB1 selectively over SMARCA4 and SMARCA2 bromodomains.

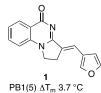


Figure 1: Initial hit from screening of a commercial library.

#### **RESULTS AND DISCUSSION**

The library of commercial analogues of **1** showed varied response by differential scanning fluorimetry (DSF) against PB1(5). DSF has been demonstrated to be an operationally simple assay, suitable for rapid assessment of potency and selectivity for bromodomains.<sup>19,24,27</sup> Limited SAR from the library indicated sensitivity to the size of the aromatic side chain and that the 2,3-dihydropyrrolo[1,2-*a*]quinazolin-5(1*H*)-one core bound within the KAc binding site (supporting information). We proposed that planarity of the compounds with conjugated aromatic ring substituents contributed to the poor solubility of the series and that analogues with greater  $sp^3$  character in their substituents, or the potential for ionization, would have improved aqueous solubility.<sup>28</sup>

Compounds were synthesized by the routes shown in Scheme 1. Amidation of 2-aminobenzamide 2 with 4-chlorobutanoyl chloride<sup>29</sup> gave amide 3, which could be cyclized to the 2,3dihydropyrrolo[1,2-a]quinazolin-5(1H)-one core 4 using potassium tert-butoxide. The isomer was confirmed on the basis of NOE correlations and subsequent crystallography. The weak nucleophilicity of the core was then exploited by reacting compound 4 with substituted N,N-dialkylformamides using Vilsmeier-Haack conditions, giving analogues incorporating nitrogen-based aliphatic side chains solely as the E regioisomer. This chemistry was unsuitable for piperazine-based formamides. For these side chains, analogues 12-15 were synthesized as single regioisomers by displacement of the dimethylamino group in compound 5 using cyclic secondary amines and DMAP catalysis. Due to the potential for hydrolysis of the enamine-like moiety, the aqueous stability of compound 5 was assessed by <sup>1</sup>H-NMR in a D<sub>2</sub>O time course experiment and was stable for >24 hours. All analogues appeared to have increased aqueous solubility compared to compound 1.

Their binding to PB1(5) was assessed by the operationally simple DSF assay (Table 1). Although the simple scaffold **4** itself showed no interaction, aliphatic analogues **5-15** generally showed improved binding to PB1(5). The shape of the side chain was a key factor in determining activity; inclusion of propyl side chains in **7** led to moderate activity but variation to *iso*-propyl in **8** caused a loss of binding. Six membered aliphatic rings (**10-14**) were better tolerated than analogues containing six-membered aromatic side chains (see Supporting Information), likely due to their greater flexibility. However, expansion to a seven membered ring reduced activity.

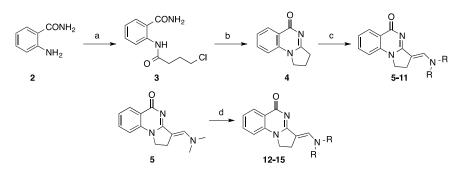
Table 1: Effect of aliphatic side chai	in substitution on bind-
ing to PB1(5) and SMARCA4 by DS	SF and AlphaScreen.

Í	N II	$DT_{m}(^{\circ}C) \qquad < 0  0-1$	1-3 3-5 5-7 7-9
	N N R R		
Cpd	Side Chain	PB1(5) $\Delta T_m$ (°C) <sup><i>a</i></sup>	SMARCA4 IC <sub>50</sub> (µM) <sup>b</sup>
5	N	4.2 ± 0.6	11.0 ± 2.6
6	~ <sup>-</sup> ~	4.3 ± 0.4	ND <sup>c</sup>
7	~N	4.4 ± 0.6	ND
8	N N	1.1 ± 0.7	ND
9	N	5.8 ± 0.1	9.6 ± 1.4
10	N	5.0 ± 0.5	6.2 ± 1.1
11	NO	4.7 ± 0.6	ND
12	N_N-	3.5 ± 0.2	7.3 ± 2.6
13	N_N_	4.3 ± 0.4	6.9 ± 1.5
14	NN-Boc	3.7 ± 0.4	ND
15	N_N~	2.5 ± 0.6	22.0 ± 3.6
		6.4	

<sup>*a*</sup> Values shown are the average of three replicates and standard deviation by DSF assay. <sup>*b*</sup> Values shown are the average of two replicates by AlphaScreen assay. <sup>*c*</sup> IC<sub>50</sub> not determined.

It was anticipated that compounds targeting PB1(5) might interact with the SMARCA2 and SMARCA4 bromodomains due to the similarity of their binding pockets. We confirmed the binding of these analogues to the SMARCA4 bromodomain in an AlphaScreen assay (Table 1), which established that the compounds had affinities for this bromodomain ranging from  $6.2-22.0 \mu$ M.

The improved solubility and activity of the compounds allowed us to conduct isothermal calorimetry (ITC) experiments with analogues **9** and **10**. These had  $K_D$  values of 2.3  $\mu$ M and 1.2  $\mu$ M respectively for the PB1(5) bromodomain. In both compounds **9** and **10**, the entropy change upon binding (T $\Delta$ S)



<sup>*a*</sup> Reagents and conditions: (*a*) 4-chlorobutanoyl chloride, Et<sub>3</sub>N, THF, 0 °C to rt, 12 h, 92%; (*b*) 'BuOK, THF, rt, 1 h, 77%; (*c*) N,N-dialkylformamide, POCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 70 °C, 16–74%; (*d*) H-NR<sub>2</sub>, DMAP, EtOH, 70 °C, 24–96 h, 39–81%. For R see Table 1.

contributed favorably to binding (T $\Delta$ S values of 4.7 and 5.3 kcal/mol respectively). In addition, interaction of these two inhibitors with PB1(5) was driven by enthalpic contributions ( $\Delta$ H -2.9 and -2.7 kcal/mol respectively). The underlying molecular mechanism for the observed thermodynamics was evident on analysis of the binding mode of compound **10**, revealed by a co-crystal with PB1(5) bromodomain (Figure 2).

This structure demonstrated that the rigid core of 10 sits deep in the hydrophobic KAc binding site of PB1(5), displacing the four waters observed in the apo-form and most BRD-ligand crystal structures. The carbonyl of the 5,6-dihydropyrimidin-4(1H)-one ring formed a direct hydrogen bond to Tyr696, whilst the amide hydrogen bonds to Asn739. This binding mode is similar to that adopted by PFI-3 and its salicylic acid precursor, which also displace conserved waters and form direct hydrogen bonds to Tyr696 and Asn739.26 These interactions are in marked contrast to those in most bromodomain inhibitors, in which a KAc mimetic, directly hydrogen bonding to the asparagine residue, forms a water-mediated hydrogen bond network to tyrosine using the structural conserved waters. The crystal structure highlighted the narrowness of the PB1(5) pocket, which the compound occupies well, and indicated that the side chain is key in orienting the core within the pocket. This helped explain the lack of activity of the unsubstituted core. The piperidine side chain projects out from the pocket forming a close interaction with the hydrophobic rim of the binding site, explaining the loss in activity in compounds 8 and 15, which contain more sterically demanding side chains.

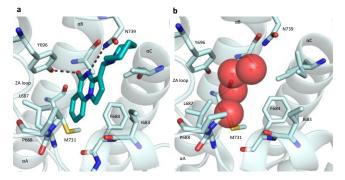


Figure 2: Binding modes of bromodomain inhibitors. (a) Cocrystal structure of **10** with PB1(5) at 2.3 Å (PDB 5FH6). Hydrogen bonds are shown by black dashed lines. (b) *Apo*-crystal structure of PB1(5) (PDB: 3G0J) showing conserved crystallographic waters (red spheres) which are displaced in (a).

The carbonyl of Met731 was observed to be in an unsolvated

cavity in the structure, and it was hypothesized that substitution on the aromatic ring *ortho* to the carbonyl group might enable an additional interaction in this region. Compounds **16**-**20** could be accessed by incorporating substituted 2aminobenzamides into the synthetic methodology previously developed. Compound **21** featuring hydroxyl substitution was obtained by deprotection of the methoxy analogues using BBr<sub>3</sub>. Detailed procedures are in the supporting information.

Screening these analogues against PB1(5) by DSF showed that substitution of chlorine at  $R_2$  in compound **18** doubled the observed  $\Delta T_m$  in comparison to unsubstituted **5**, whilst bromine substitution gave a smaller increase (Table 2). All other substitutions reduced binding compared with analogue **5**. This SAR provides valuable compounds suitable as negative controls for the exploration of the biological activity of these compounds.

Table 2: Effect of	aromatic	substitution	on	binding	to	the
PB1(5) bromodom	ain.					

$\begin{array}{c} R_2 & O \\ \downarrow & \downarrow \\ N \\ \downarrow & N \\ \end{pmatrix} \\ \end{pmatrix} \\ N \\ \end{pmatrix} \\ \\ N \\ \end{pmatrix} \\ \\ \\ \\ \\ \\$							
Cpd	R <sub>2</sub>	$\frac{\text{PB1(5)}}{\Delta \text{T}_{\text{m}}(^{\circ}\text{C})^{a}}$	Cpd	R <sub>2</sub>	$\frac{\text{PB1(5)}}{\Delta \text{T}_{\text{m}}(^{\circ}\text{C})^{a}}$		
5	Н	$4.2 \pm 0.6$	19	Br	6.9 ± 0.9		
16	Me	$1.4 \pm 0.4$	20	OMe	$-0.7 \pm 0.7$		
17	F	$1.4 \pm 0.8$	21	ОН	3.4 ± 0.2		
18	Cl	8.4 ± 0.6					

<sup>*a*</sup> Values shown are the average of three replicates and standard deviation.

A  $K_{\rm D}$  of 137 nM for PB1(5) was obtained for **18** by ITC, with a significant increase in the enthalpic contribution to binding ( $\Delta$ H -6.0 kcal/mol). Since Cl/Br inclusion also increased binding of the core scaffold ( $\Delta$ T<sub>m</sub> 2.5 °C and 3.1 °C respectively, see supporting information) we proposed that the improved interaction was due to formation of a halogen bond to the carbonyl of Met731. Whilst bromination would be predicted to give a stronger halogen bond,<sup>30</sup> the effect is countered by steric constraints within the cavity.



		$\Delta T_{\rm m}$ (°C)						
Cpd	Side Chain	PB1(5) <sup>a</sup>	PB1(2) <sup>a</sup>	PB1(3) <sup>b</sup>	PB1(4) <sup>a</sup>	SMAR- CA2A <sup>a</sup>	SMAR- CA2B <sup>a</sup>	SMAR- CA4 <sup>a</sup>
18	N _	8.4	2.2	1.0	0.4	4.4	2.7	3.6
22	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	7.3	2.9	0.2	0.6	3.6	2.5	2.8
23	N	6.9	1.2	0.2	0.4	2.6	1.3	1.4
24	N	9.8	4.7	0.8	0.9	6.3	4.3	5.3
25	NO	7.4	1.9	0.4	0.3	2.8	1.7	2.7
26	N_N-	10.2	4.3	1.3	1.1	6.6	4.2	5.4
27	N_N_	9.2	4.0	1.4	0.9	4.8	4.1	5.6

<sup>a</sup> Values shown are the average of three replicates. Standard deviation is < 0.6 °C. <sup>b</sup> Values are a single measurement.

This interaction was confirmed by a co-crystal structure of **18** with PB1(5), in which the chlorine occupies the cavity previously observed with a separation of 3.2 Å from the carbonyl oxygen of Met731. This is consistent with a halogen bond (Figure 3).<sup>30</sup> A slight twist in the scaffold, to maximize interactions with both Tyr696 and Asn739, can be observed. Comparison with PFI-3 shows that the interaction with Met731 has not been previously exploited (supporting information).

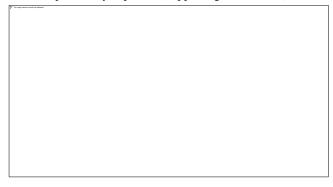


Figure 3: Co-crystal structure of **18** with PB1(5) at 1.5 Å (PDB 5FH7). (a) Cut-through illustrating positioning of chlorine in cavity. (b) Surface view illustrating fit in PB1(5) pocket.

To improve the aqueous solubility of the chlorinated derivatives, we varied the enamine substituent using the existing methodology to access analogues **22-27**, the majority of which retained their affinity for the target as assessed by DSF (Table 3). Compounds **24**, **26**, and **27** had greatly improved DMSO and aqueous solubility. The binding of these compounds to PB1(5) was assessed by ITC (Table 4) and all displayed similar  $K_{\rm D}$  values to compound **18**.

As shown in Table 3, the analogues were screened by DSF against other bromodomains within the SWI/SNF complex, including other structurally distinct bromodomains within

protein PB1, to determine their selectivity. When interpreting this data, it should be noted that different proteins do not behave identically in DSF assays, and  $\Delta T_m$  values should not be viewed as an absolute scale. For example, probe PFI-3 has  $K_D$  values by ITC of 89 nM and 48 nM for SMARCA4 and PB1(5) respectively, but  $\Delta T_m$  shifts of 5.1 °C and 7.5 °C.<sup>25</sup>

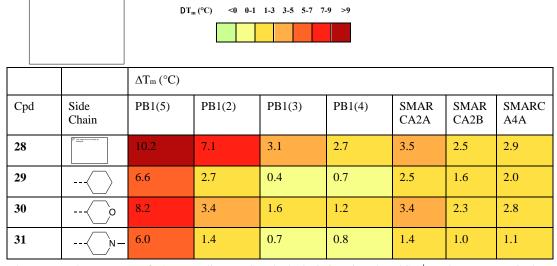
Table 4: Binding of key compounds to PB1(5) was assessed by isothermal calorimetry.

Cpd	K×1E6	$K_{\rm D}$ (nM)	$\Delta H$	$T\Delta S$
	(1/M)		(kcal/mol)	(kcal/mol)
18	$7.29 \pm 0.91$	137 (157-122)	-6.2	3.0
24	$7.86 \pm 0.89$	127 (143-114)	-5.2	4.0
26	$8.09\pm0.66$	124 (135-114)	-5.0	4.3
27	$7.05\pm0.68$	142 (157-129)	-4.4	4.7
28	$5.88 \pm 0.60$	170 (189-154)	-6.8	2.3



Figure 4: Representative ITC trace, measured using **26** and the bromodomain of PB1(5). 30 consecutive injections of PB1(5) into a solution of **26** in 20 mM HEPES, 150 mM NaCl, 0.05 mM TCEP. Raw heats (left) and normalized injection heats with a nonlinear least-squares fit for a single binding site model (right) are shown.

Table 5: Structure activity relationships of compounds bearing alkene based side chains against SWI/SNF bromodomains.



<sup>a</sup> Values shown are the average of three replicates. Standard deviation is <0.6 °C. <sup>b</sup> Values are a single measurement.

In our screening, compounds appeared to interact with the bromodomains in both SMARCA2 isoforms, A and B, and in SMARCA4 as well as PB1(5). ITC was used to further assess the binding of compound **26** to SMARCA2B and SMARCA4, giving  $K_D$  values of 262 nM and 417 nM respectively. Binding to the SMARCA bromodomains had not been observed prior to inclusion of the chlorine substitution, suggesting that the proposed halogen bond is again important for binding.

The bromodomain PB1(2) also demonstrated a moderate response, despite containing a broader KAc binding pocket than PB1(5). By contrast, minimal protein stabilization was observed with bromodomains PB1(3) and PB1(4). This may be due to the tyrosine residue in the ZA loop which acts as a 'lid' to the pocket in these bromodomains.<sup>26</sup>

We were interested to establish whether further side chain variation could alter selectivity within SWI/SNF bromodomains. Analogues 28-31 incorporating alkene side chains synthesized by reaction of 6-chloro-2,3were dihydropyrrolo[1,2-a]quinazolin-5(1H)-one (18c) with the appropriate aliphatic aldehydes under basic conditions (see supporting information for full methods). These were screened by DSF, which suggested reduced binding to PB1(5) for the majority of ligands except compound 28. When assessed by ITC, compound 28 had a  $K_D$  of 170 nM with PB1(5). DSF results suggested an intriguing change in selectivity amongst SWI/SNF bromodomains. In contrast to all previous compounds tested in this report, and the chemical probe PFI-3,<sup>26</sup> 28 showed a stronger interaction with PB1(2) than with the SMARCA2/4 bromodomains. The weak binding to SMARCA4 was confirmed by ITC, with a  $K_D$  of 2.03  $\mu$ M for the interaction with compound 28.

Understanding the reason for the change in affinity would help design of future chemical probes with selectivity for bromodomains not yet targeted such as PB1(2-4) and offers the intriguing possibility of targeting PB1 selectively over SMARCA2/4. We obtained a co-crystal structure of **28** with PB1(5) to understand its unique features. This revealed a shift in the ZA loop region of PB1(5), making the KAc pocket wider, and increasing its similarity to the PB1(2) KAc pocket. The ethyl side chains interact with the hydrophobic edges of the pocket, and begin to occupy the ill-defined peptide-binding channel in PB1(5). Although no structure was obtained to confirm this, we propose that in SMARCA4, Ile1543 blocks this channel preventing the ethyl side chains being accommodated in the smaller pocket.

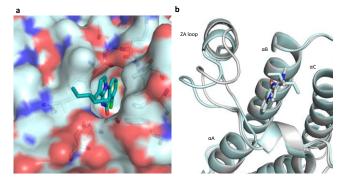


Figure 5: (a) Co-crystal structure of **28** with PB1(5) at 1.6 Å (PDB 5FH8). (b) Overlay of PDB 5FH7 (gray) of **18** and PDB 5FH8 (cyan) of **28** showing the shift in the ZA loop of the PB1(5) domain on binding of **28**.

We assessed the selectivity of the most potent pan-PB1, SMARCA2/4 analogue **26** against representative bromodomains (PCAF, BRD4(1), CREBBP, TRIM33B) from other sub-families using DSF. No significant binding was observed.

Finally, the ability of compound **26** to displace SMARCA2 from chromatin was assessed using a fluorescence recovery after photo-bleaching (FRAP) assay (Figure 7).<sup>31</sup> U2OS cells were transfected with a GFP-linked SMARCA2 construct. In order to increase global histone acetylation and thereby the assay window the broad-spectrum histone-deacetylase (HDAC) inhibitor, suberoylanilide hydroxamic acid (SAHA) was used. Cells transfected with a construct carrying the N1464F mutation, unable to bind to chromatin via the bromodomain were used as a positive control. Treatment of cells with compound **26** at 1 or 5  $\mu$ M reduced FRAP recovery times back to that of the N1464F mutant and unstimulated levels indicating that compound **26** was able to displace full length SMARCA2 from chromatin.

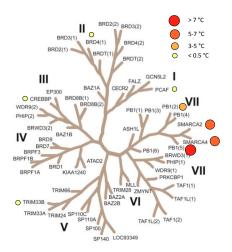


Figure 6: Selectivity of **26**. The inhibitor **26** was screened at 10  $\mu$ M against selected bromodomains by DSF assay. Temperature shifts for screened proteins are shown on the phylogenetic tree.

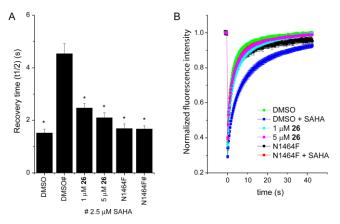


Figure 7: Compound **26** inhibits SMARCA2 association with chromatin in cells. (a) FRAP half recovery times of GFP-SMARCA2 are significantly decreased when treated with compound **26** at 1 or 5  $\mu$ M as indicated. Cells expressing the SMARCA2 bromodomain-inactivating mutant (N1464) were analyzed as comparison. Significant differences to cells treated with SAHA of p< 0.0001 are shown by \*; (b) Time dependence of fluorescence recovery in the bleached area of cells expressing wt or mutant GFP-SMARCA2 with the corresponding treatment.

## CONCLUSIONS

We describe the optimization of an inhibitor series targeting bromodomains found within the SWI/SNF complex from a weakly potent hit with poor physicochemical properties. Improvement of solubility has allowed co-crystal structures to be obtained demonstrating the important role of water displacement in the binding of these inhibitors. Chlorination of the series has demonstrated the potential for exploitation of previously unexplored interactions deep within the PB1(5) KAc binding pocket through halogen bonding. Side chain variation in 28 shows that the second and fifth bromodomains of PB1 can be targeted selectively over the SMARCA2/4 helicases, in contrast to the selectivity shown by the chemical probe PFI-3. Lead inhibitor 26 displays good affinity for PB1(5), SMARCA4, and SMARCA2 as assessed by ITC, excellent selectivity within the bromodomain family and the ability to displace SMARCA2 from chromatin in cells, making it suitable as a chemical probe with a distinct chemotype to PFI-3 and for further development of SWI/SNF bromodomain inhibitors.

### ASSOCIATED CONTENT

#### **Supporting Information**

Additional structural images and screening data, ITC traces, X-ray refinement statistics, additional text describing biological methods and synthetic procedures, characterization data, NMR. Supporting Information is available free of charge on the ACS Publications website. Additional data related to this publication is available at the xxxx data repository.

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## ABBREVIATIONS

ATP, adenosine triphosphate; BCP, bromodomain containing proteins; BET, bromodomain and extra-terminal domain; BRD7, bromodomain containing protein 7; BRD9, bromodomain containing protein 9; DMAP, *N*,*N*-dimethyl-4-aminopyridine; DSF, differential scanning fluorimetry; FRAP, fluorescence recovery after photobleaching; GFP, green fluorescent protein; HDAC, histone deacetylase; ITC, isothermal titration calorimetry; KAc, acetyllysine; NOE, nuclear Overhauser effect; PB1, polybromo-1; PB1(X), Xth bromodomain of PB1; PDB, protein databank; SAHA, suberoylanilide hydroxamic acid; SAR, structure activity relationship; SMARCA2/4, SWI/SNF related, matrix associated, actin dependent regulator of chromatin, subfamily A, member 2/4; SWI/SNF, Switch/Sucrose Non-Fermenting.

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