

Controlling nanowire growth by light

G. Di Martino,^{1*} F. B. Michaelis,² A.R. Salmon,^{1,3} S. Hofmann,^{2*} J.J. Baumberg^{1*}

¹ NanoPhotonics Centre, Cavendish Laboratory, University of Cambridge, CB3 OHE, UK

² Department of Engineering, University of Cambridge, CB3 0FA, UK

³ Department of Chemistry, University of Cambridge, CB2 1EW, UK

Individual Au catalyst nanoparticles are used for selective laser-induced chemical vapour deposition of single germanium nanowires. Dark-field scattering reveals in real time the optical signatures of all key constituent growth processes. The process is initially triggered by plasmonic absorption in the Au catalyst, whilst once nucleated the growing Ge nanowire supports magnetic and electric resonances which then dominate the laser interactions. This spectroscopic understanding allows real-time laser feedback which is crucial towards realising the full potential of controlling nanomaterial growth by light.

KEYWORDS: Catalysis, laser induced chemical vapour deposition, germanium nanowires, plasmonics, dark field spectroscopy, magnetic resonances.

Bottom-up or self-assembly approaches to nanofabrication such as the catalytic growth of semiconductor nanowires (NWs) offer unique compositional and structural control and versatility, opening pathways to new functionalities in a wide range of applications including electronics, thermo-electrics, energy storage and conversion, and biomedical sensors^{1–7}. In the so-called vapour-liquid-solid (VLS) and vapour-solid-solid (VSS) growth mechanisms a liquid or solid catalyst such as a Au nanoparticle selectively dissociates a vapour phase precursor and mediates anisotropic NW crystal growth^{8,9}. This has been extensively explored for group IV^{10,11}, III-V^{12,13} and II-VI^{14,15} semiconductor materials and their device integration. Typically the reaction is triggered by globally heating the sample to temperatures in excess of 300°C, which not only limits the choice of substrate, and device integration pathways, but also acts on all catalyst particles with the same precursor and conditions, limiting the versatility of the process. Hence a range of local heating strategies have been explored, in various processes including laser irradiation^{16,17}. In this way not only the catalytic but also the plasmonic properties of Au nanoparticles can be exploited, opening pathways for selective

and effective local excitation at the sub-diffraction level. The potential of such approaches, often termed plasmon-assisted chemical vapour deposition (CVD), has been highlighted already by a number of reports^{18,19}, where the chemical growth reactions could be triggered selectively with high spatial resolution. However understanding of the complex light-matter interactions during such laser-induced growth remains rudimentary, and hence detailed structural growth control and the full potential of this approach could not be realised so far.

Here we study laser-driven catalytic nanowire CVD and show that the growth process can not only be optically powered but also monitored in real-time by optical spectroscopy at the individual nanoparticle level, which in combination allows a feedback process that is crucial to growth control. We find that the selectivity of the process is only initially given by the plasmonic absorption in the supported Au catalyst nanoparticle, whilst once nucleated the growing Ge nanowire supports magnetic and electric resonant modes which then dominate the laser interactions. This results in a process-dependent increase in absorption cross-section and to achieve NW growth selectivity we show that the laser power has to be reduced accordingly. Hence, if controlled, the growth process can be maintained at very low laser powers, i.e. is extremely efficient. This spectroscopic technique also enables us to follow the evolution of NW morphology during growth. We focus on Au catalysed Ge NW CVD for which the thermal growth mechanisms have been well studied,²⁰ and use it as model system to reveal the in-situ optical signatures that enable tracking of the key processes of laserdriven CVD. Key characteristics, such as diameter and change in length of the NWs, can be identified from the optical signatures in the scattering signal. Moreover VLS and VSS growth mechanisms become distinguishable by analysing such optical spectra. As we discuss, the spectroscopic feedback and insights gained are a key step towards realising the full potential of controlling nanomaterial growth by light.

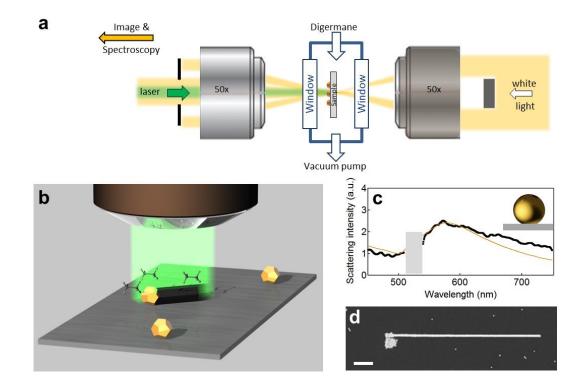


Figure 1 | Experimental laser-induced CVD setup. a, Configuration for optical growth monitored by dark field scattering, sample mounted in a vacuum chamber and back illuminated by high-angle white light with only low angle scattered light collected, pump 532nm laser focused to 1µm spot size. b, Sketch of PLiCVD triggering Ge NW growth on the illuminated Au NP. **c**, Typical initial scattering spectrum (black) of Au NP (points around laser line removed), with simulated spectrum (yellow) of 140nm faceted Au sphere on a 20nm-thick silicon nitride substrate. **d**, Long Ge nanowire grown, scale bar 2µm.

We use laser irradiation and selective absorption to locally heat and trigger reactions on individual supported catalyst nanoparticles in a gaseous precursor atmosphere (Figure 1a,b). We use diluted Ge₂H₆ as the precursor (0.2% in Ar, ~10 mbar total pressure during growth) and a customised stainless-steel vacuum chamber (base pressure ~10⁻⁷ mbar) in which we selectively illuminate well-separated (average distance > 5 μ m) 140 nm diameter Au nanoparticles (NPs) on a 20nm-thick transparent silicon nitride membrane with a continuous wave laser at λ =532 nm. This wavelength is close to the plasmonic resonance of each Au NP, centred around 580nm. The Au NPs are characterized by strong enhanced light absorption in this spectral region and are thus rapidly and locally heated by the laser. This leads to catalytic precursor dissociation and, upon Ge supersaturation of the Au NP, to Ge NW nucleation and growth. The reaction sequence and Ge growth lead in turn to an ever changing, process-dependent, optical absorption profile, which rapidly becomes no longer dominated by the Au plasmon and to which the laser has to be adapted, as we discuss below. Because the laser spot size is ~1 μ m and the collection region for spectroscopy is ~ 2 μ m, only one NP is illuminated and characterized at any time. Using different off-resonant

laser wavelengths (e.g. 780 nm) the initial power input required to trigger the catalytic Ge growth is much higher (>10x). This highlights the selective and effective local excitation of the process. Figure 1d highlights that Ge NW growth is catalysed only at the particle under the focal spot, leaving nearby NPs unreacted and the substrate uncoated by Ge. Smaller NPs have also been tested (~80nm), where the plasmon resonance remains spectrally close to the 532nm laser. Although the technique is indeed applicable to smaller NPs and hence smaller NW diameters, the reduced scattering intensities with these smaller sizes makes resolving growth details increasingly difficult. As a result for our studies here in the current optical set-up and to focus on understanding we mainly used 140 nm diameter Au NPs. We investigated extensive libraries with different growth parameters by translating different Au NPs under the laser focus, allowing rapid, facile combinatorial mapping and screening of the CVD parameter space which is arduous with conventional global back heating. We can check the Au NPs via their initial scattering spectra, automatically excluding dimers, agglomerates or oversized NPs that do not show the characteristic resonance expected for a 140nm Au NP (Figure 1c). Dark field spectra are collected by scattering unpolarised white light focussed from the back of the transparent membrane supporting the Au NPs, and using notch filters to remove the laser line. This ensures the same initial conditions in all cases and provides good reproducibility.

Upon laser illuminating each Au NP, it heats up and starts reacting with the Ge₂H₆ precursor. Figure 2 shows a post-growth scanning electron microscopy (SEM) image series representative of the initial Ge NW growth stages (Fig. 2a-d, see Methods), together with the corresponding optical darkfield scattering spectra recorded in-situ (Fig. 2e). The atomic number (Z) contrast of the backscatter electron detector allows clear differentiation between the growing Ge crystal and the Au-based catalyst in the SEM images, which we also confirm by energy-dispersive X-ray spectroscopy (EDX, see Figure S1). To verify our assignments we also record post-growth optical scattering spectra, which do not show any significant changes from the spectra we record during the final stage of our growth process (see Figure S2), corroborating our interpretation below. As soon as germanium nucleates from the Au NP, there is a red-shift and clear change in the scattering spectrum (Figure 2e). As the Ge crystal grows, the composite nanostructure starts to absorb more and more light as the optical cross-section increases, which can also be observed from the rise in scattering strength. If the laser power is not reduced accordingly to keep the temperature constant, this leads to detrimental overheating. The temperatures are then high enough to lose the selectivity of the catalytic NW growth process and instead result in non-selective pyrolytic Ge overgrowth, creating an agglomerate of Ge material and faceted Ge crystalline particles (Fig. 2k-I). The first symptom of such overheating

is promptly detected by our in-situ dark field spectroscopy, appearing as strong blue scattering (Fig. 2h), which fits well to modelled scattering from a dielectric alloyed sphere (blue line). A plausible explanation for this suggests the degeneration of the nanostructure into an agglomerate of interspersed Ge and Au clusters, arranged in a disordered matrix. This yields an overall dielectric-like (non-metallic) optical response to the structure. Interestingly, we find in this case different early stage Au-Ge morphologies (Fig. 2i,j), from the degeneration of the growth process. In order to avoid this overheating, we thus use the scattering intensity as a control signal to directly adjust the power of the laser (see Figure S3). By implementing this feedback we can maintain the catalytic process to be dominating and grow Ge NWs (Fig. 2f,g). It is therefore clear that real time monitoring of the scattering signal and adapting the laser power to the changing adsorption behaviour of the growing nanostructure is of paramount importance.

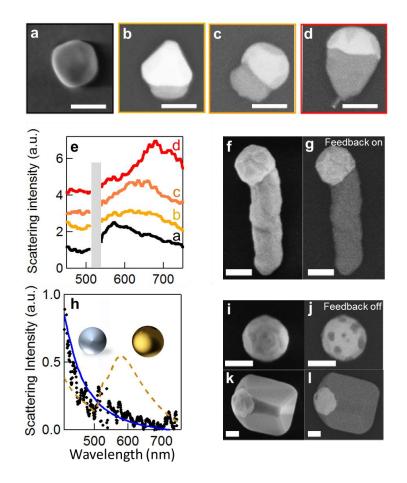


Figure 2 | Evolution of NW growth. a, SEM secondary electron (SE) image of Au NP before growth. **b-d**, High-angle backscattered electron (HA-BSE) SEM images of Ge NWs in early stage growth. **e**, Dark-field scattering spectra of (a-d), curves offset for clarity. **f**,**g**, SE and HA-BSE images of Ge NW grown implementing feedback, giving controlled NW growth. **h**, In-situ scattering spectrum (black) of overheated system shown in **i**, SE and **j**, HA-BSE images, with simulated scattering spectrum of 130nm SiO₂ sphere (blue) and 140nm faceted Au sphere (yellow). **k**, SE and **l**, HA-BSE images of a Au NP together with Ge agglomerate grown by pyrolytic deposition due to overheating. No feedback control is implemented in (i-I). Scale bars are all 100nm.

With the parameters used here, we generally observe two types of Ge NW growth dynamics, each characterized by a different evolution of the scattering spectra as highlighted by Figure 3. In type A (Fig. 3a-f), a continuous redshift of the scattering peak is seen, reaching a wavelength of 800nm by the end of the growth process (Figure 3e). The resulting Ge NW always shows an irregular, worm-like shape (Fig. 3a-c), with diameters on the order of 130 nm. The Au NP is at the tip of the Ge NW (Fig. 3d) with the catalyst-NW interface at post-growth stage being often very non-planar and irregular (Figure 3c-d). Moreover, the growth rate we extrapolate for this case is quite slow, \sim 0.04nm/sec (see Figure S4). For type B we observe a more complex evolution of the scattering spectra, with an initial redshift suddenly switching to a double peak (Fig. 3k). The resulting NWs in this case are comparatively much straighter (Fig. 3g,i), and their average diameter is always larger (of order 180 nm). The catalyst-NW interface now also appears more planar (Fig. 3g-j). Moreover for type B the growth rate is nearly a hundred-fold faster, ~2.7nm/sec (see Figure S4). Type B growth dynamics occur for slower laser feedback, i.e. when the laser power is reduced less rapidly (see Methods). Hence higher local growth temperatures occur for type B compared to type A and, as discussed below, the observed differences can be rationalised by assuming two different growth mechanisms: a VSS-type mechanism with a solid catalyst particle for type A, and a VLS growth mechanism for type B, where the catalyst forms a liquid alloy^{20,21}.

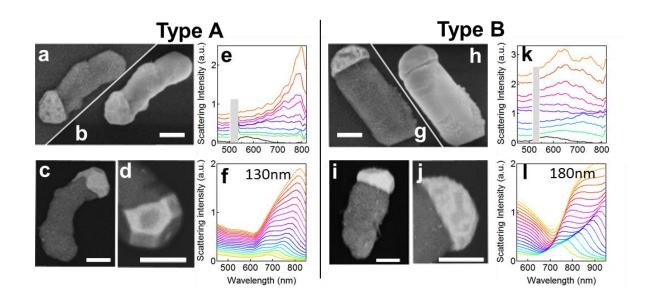


Figure 3 | **Dynamics of Type A,B growth. a,c,d,g,i,j**, High-angle backscattered SEM images (giving compositional contrast) and **b,h**, secondary electron SEM images (topology contrast). Scale bars are 100nm. **e,k**, In-situ dark-field scattering spectra of NW shown in (a-b), and (g-h) respectively. Offsets added for clarity. **f,l**, Finite-difference time-domain simulating scattering spectra for Ge cylinder with increasing length of 130nm and 180nm diameters, respectively.

We previously carried out a range of detailed studies on thermal catalytic NW CVD, highlighting how kinetically driven supersaturations determine the catalyst phase, and the important consequences for NW growth from solid vs liquid catalyst particles²⁰. For the first time, we connect here such levels of growth detail to the laser-driven process. It is well known that the simple Au-Ge eutectic phase diagram for the bulk equilibrium (eutectic temperature $T_e=361^{\circ}C$, at 28 at% Ge²²) does not fully capture the phase behaviour for catalytic growth on the nanoscale. Hence catalyst liquefaction can be observed well below T_e^{21} . Nonetheless, the higher the laser power, the higher the expected local temperature and the more likely catalyst liquefaction will occur. This can therefore account for all of the differences observed between type A and B nanowires. To first order, the NW diameter clearly relates to the catalyst particle diameter, but as shown previously in detail for thermal CVD, with catalyst liquefaction (VLS mechanism) the Ge solubility is much higher hence increasing the catalyst (alloy) particle volume¹¹. At the same time the wetting angle of the catalyst alloy at the tip of the growing NW is modified compared to a crystalline catalyst. This results here in larger NW diameters for the VLS case. For the VSS mechanism, the gas exposures here appear too high given the reduced diffusivities, which hence leads to defective, irregular NW growth.

Understanding the experimentally observed scattering spectra (Fig. 3e,k) requires further interpretation. We therefore model our system using the full geometry and material parameters of Ge and Au. In Figure 3, we show simulated scattering spectra of Ge nanocylinders of increasing length with different diameters placed on silicon nitride membranes. Scattering from the modelled 130nm and 180nm wide Ge nanocylinders (Fig. 3f,l) is able to reproduce the red-shifting resonance coming from the elongation of Ge NWs (Fig. 3e,k respectively) and the multiple peaks observed in the type B NWs (Fig. 3k), showing in general good agreement with the experimental data. Indeed our simulations show distinct multiple peaks start to be visible for diameters larger than 140nm, allowing as a result the differentiation of the spectral signatures of 130nm and 180nm wide Ge nanocylinders. The role of Ge in the scattering spectra results is of primary importance in providing key information about the growth process and hence is discussed further in the following section. This indicates that we are able to directly discern both the type of growth process, as well as the NW diameter directly from spectroscopy.

In contrast to metallic nanoparticles, for which the resonant scattering is dominated by electric resonances, dielectric nanoparticles can display both electric and magnetic dipole resonances simultaneously excited inside the particle. High-index dielectric or semiconductor nanostructures support (Mie-like) geometrical resonances in the visible spectral range^{23,24}, driven by

displacement currents²⁵. NWs support a limited number of transverse-electric (TE) and transversemagnetic (TM) resonances, which increase in number as their diameter is increased²⁶, and which can be spectrally tuned by varying the NW length and diameter²⁷. Sufficiently small diameter NWs (< 30nm) are predicted to support only the fundamental transverse-magnetic mode²⁶ while NWs possess a very strong polarization dependence, decreasing with increasing diameter²⁶. However as the in-situ scattering measured here is rather weak and from unpolarised light, higher brightness sources are required to exploit this information (see Figure S5). Since the scattering spectrum of high refractive index nanowires depends strongly on diameter, the presence of two distinct diameters from each type of growth observed here thus accounts for the distinct scattering spectra found (Fig. 3e,k).

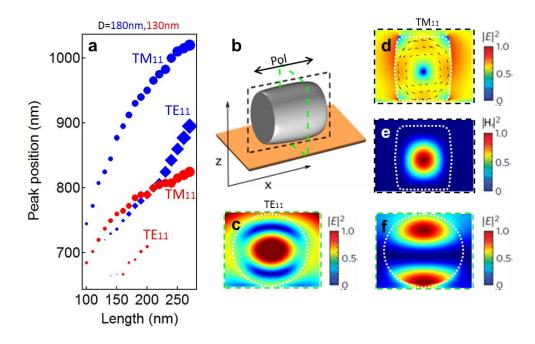


Figure 4 | Simulated resonant field modes for Ge nanocylinder. a, Peak positions of different resonant modes, Transverse Magnetic (TM, •) and Transverse Electric (TE, **■**), for increasing length of Ge nanocylinder, of diameters 130nm (red) and 180nm (blue). Marker size indicates intensity of scattering. **b**, Simulated geometry. **c-f**, Near-field distributions for Ge nanocylinder with diameter 130nm and length 270nm: **c**, E(yz) at resonant wavelength of 650nm, **d**, E(xy) at resonant wavelength of 840nm. In all simulations, incident Gaussian beam is polarised along x and propagating along -z.

We use numerical simulations to systematically study the resonant properties of Ge nanocylinders, and thus understand the spectral dynamics observed during Ge NW growth. For simplicity, we show here results for Ge nanocylinders with incident light polarised along their length and incident normally (Fig. 4b), but similar results are found for orthogonal polarisations (see Figure S5). The spectral positions of the resonances for different modes excited inside the NW show a general redshift with increasing length of the cylinder (Fig. 4a). This enables us to follow the elongation of NWs during growth. By analysing their near-field profile in the *xy* plane, the resonances can be

assigned,²⁶ revealing optical fields associated with both the dominant TE_{11} (Fig. 4c) and TM_{11} (Fig. 4f) modes. This confirms the excitation of both electric and magnetic resonances in our NWs. In the long wavelength region, the circulation of the electric field in the *xz* plane (Fig. 4d) produces a strong magnetic field component along *y* (Fig. 4e), responsible for the strong magnetic scattering of the TM₁₁ mode. Only in large diameter cylinders (Fig. 4a) does scattering from the TE₁₁ mode contribute significantly.

Our results show the capability to selectively grow single NWs by optically monitoring the catalytic NW growth and hence to be able to implement real-time feedback, adapting the laser power to the ever changing optical cross-section of the growing nanostructure^{28,29}. At the initial stage of NW PLiCVD we exploit the plasmonic absorption in an individual Au NP, which then selectively reacts with the Ge precursor following a VLS or VSS mechanism. Our results show that the nucleated and growing Ge crystal supports magnetic and electric resonant modes. The resulting scattering spectrum of the Ge NW strongly depends on its diameter as well as on its length with large resonant absorption enhancements at specific wavelengths²⁶. Here we locally heat single NPs using laser irradiation to harness the catalytic and plasmonic properties of Au. This technique operates at very low laser powers making it ideally suited to combine with other in-situ characterisation techniques, such as electron microscopy³⁰, to provide powerful future metrology and correlative imaging platforms. Exploiting this enhanced absorption with the thin support membrane allows us to drive the catalytic growth reaction at low laser power (0.2-0.6 mW), more than an order of magnitude lower than previous literature results, where powers >20mW were needed to trigger the reaction¹⁸.

Our assignments of the optical signatures also allow us to spectrally follow the NW evolution and in particular to distinguish two types of Ge NW growth dynamics. We propose that our laser feedback operates so as to either keep the Au NP solid or to locally increase the temperature enough to liquefy the catalyst. Such spectroscopic feedback coupled with optical irradiation forms a unique capability to direct nanomaterial growth down specific pathways. We focused here on Ge NW growth, but the approach of plasmonic-initiated catalytic growth is applicable to a wide range of material systems, such as Au catalysed group V, group III-V and group II-VI semiconductor NWs and their heterostructures, opening a wide range of applications.

METHODS

Sample preparation. The 138 \pm 5 nm diameter Au NPs were made in three seeded growth steps following the method of Ziegler *et al*³¹. The Au NPs were characterized by scanning transmission electron microscopy, with the mean diameter and standard deviation given from 50 measurements. The Au NPs were deposited onto silicon nitride 20 nm thick TEM grids (SiMPore Inc.) by drop casting. In a typical deposition, the as-prepared Au NP suspension was diluted by a factor of 10 (final concentration ~10⁸ ml⁻¹) and a portion was pipetted onto the grid. After 3 minutes deionised water was added and then the suspension was wicked away using lint-free paper. By the same wicking method the grid was twice rinsed with deionised water. The gas mixture used for these growth experiments is 0.2% digermane balanced in argon, with a gas pressure during growth of 10 mbar (no gas flow).

Optical setup. The sample is mounted in a custom-built low volume optical-accessible vacuum chamber, and is back illuminated by high-angle white light incident through a 50x/ 0.50NA dark-field objective (Olympus BD LMPLFLN) in an optical dark field configuration, with only low angle scattered light collected by a 50x/ 0.50NA (Mitutoyo BF G Plan Apo) objective, specifically designed for compensating transmission through the 3.5 mm BK7 glass window of the vacuum chamber. The dark field scattered light is then sent to a CCD camera (Infinity) and spectrometer (Ocean Optics QE65000) for spectral analysis. At the same time, a 532nm CW laser is focused onto the sample through the Plan Apo objective with a final spot size of 1µm. The power of the laser is modified by rotation of a computer-controlled half waveplate located just before a polarizer. Two notch filters screen the scattered laser light avoiding its detection on the CCD and spectrometer.

Power intensity and feedback control. Initial laser powers of 0.6mW are focussed onto the 1µm spot, which is the minimum power needed to trigger growth. The feedback control reduces the power of the laser when the scattering intensity, integrated over the 650-700nm region, exceeds set thresholds. We set a first intensity threshold at twice the initial scattering intensity, after which the laser power is reduced by 18% and 15% for type A and type B growth, respectively, whenever an increase of the scattering signal is detected. The integrated scattering intensity ontinues to rise until a second intensity threshold (at six times the initial scattering intensity) after which the laser power is now reduced by 28% and 25% for type A and type B growth, respectively. By this method we realise two different growth recipes resulting in the two different growth dynamics we have identified, characterised by a faster (type A) or slower (type B) laser power reduction. In these recipes, different local growth temperatures can be achieved. The overall result is an initial fast

decrease of the laser power until a stationary regime is reached, with eventual constant intensities of 0.1-0.2mW (see Figure S3).

Electron microscopy. A Hitachi S-5500 field emission scanning electron microscope (SEM) was used to characterise Au NPs and the grown Ge NWs directly on the 20 nm thick Si₃N₄ membrane TEM grid.

Numerical simulations. A commercial simulator based on the finite-difference time-domain method was used to perform the calculations of the far-field extinction spectra and near-field distributions (Lumerical Solutions Inc.). Perfectly matched layer (PML) absorbing boundaries conditions have been implemented. The simulated source is a broadband Gaussian beam polarised along x (or y, see Figure S5) and propagating along -z.

Acknowledgements

We acknowledge financial support from EPSRC grant EP/G060649/1, EP/L027151/1, EP/G037221/1, EPSRC NanoDTC, and ERC grant LINASS 320503. S.H. acknowledges funding from ERC grant InsituNANO 279342. We acknowledge valuable discussions with Angela Demetriadou and Federico Panciera during different stages of the research.

Author Contributions

Experiments were performed by G.D., with support for the chemical nanoassembly from A.S. and growth calibrations of B.M. The software and the experimental setup were developed by G.D. and B.M. Modelling and simulations were performed by G.D. S.H. and J.J.B. designed the experiments and coordinated the analysis. All authors contributed to the manuscript.

Corresponding Authors

- *E-mail: gd392@cam.ac.uk
- *E-mail: sh315@cam.ac.uk
- *E-mail: jjb12@cam.ac.uk

Supporting Information

Supporting Information Available: material analysis of a Ge NW using energy-dispersive X-ray (EDX) spectroscopy, analysis proving quenching of the system, details on laser power reduction due to feedback control, growth rate of the two different growth mechanisms, influence of polarization. This material is available free of charge via the Internet at <u>http://pubs.acs.org</u>.

Notes

The authors declare no competing financial interest.

References

- (1) Ionescu, A. M.; Riel, H. *Nature* **2011**, *479* (7373), 329–337.
- (2) Li, Y.; Qian, F.; Xiang, J.; Lieber, C. M. *Mater. Today* **2006**, *9* (10), 18–27.
- (3) Hu, J.; Odom, T. W.; Lieber, C. M. Acc. Chem. Res. **1999**, *32* (5), 435–445.
- (4) Fan, H. J.; Werner, P.; Zacharias, M. Small 2006, 2 (6), 700–717.
- (5) Chan, C. K.; Zhang, X. F.; Cui, Y. Nano Lett. 2008, 8 (1), 307–309.
- (6) Zheng, G.; Patolsky, F.; Cui, Y.; Wang, W. U.; Lieber, C. M. *Nat. Biotechnol.* **2005**, *23* (10), 1294–1301.
- Hochbaum, A. I.; Chen, R.; Delgado, R. D.; Liang, W.; Garnett, E. C.; Najarian, M.; Majumdar, A.; Yang, P. *Nature* 2008, 451 (7175), 163–167.
- (8) Wagner, R. S.; Ellis, W. C. Appl. Phys. Lett. 1964, 4 (5), 89–90.
- (9) Hofmann, S.; Sharma, R.; Wirth, C. T.; Cervantes-Sodi, F.; Ducati, C.; Kasama, T.; Dunin-Borkowski, R. E.; Drucker, J.; Bennett, P.; Robertson, J. *Nat. Mater.* **2008**, *7* (5), 372–375.
- (10) Cui, Y.; Lieber, C. M. Science 2001, 291 (5505), 851–853.
- (11) Kodambaka, S.; Tersoff, J.; Reuter, M. C.; Ross, F. M. Science 2007, 316 (5825), 729–732.
- (12) Hiruma, K.; Yazawa, M.; Katsuyama, T.; Ogawa, K.; Haraguchi, K.; Koguchi, M.; Kakibayashi, H. *J. Appl. Phys.* **1995**, *77* (2), 447–462.
- (13) Caroff, P.; Dick, K. A.; Johansson, J.; Messing, M. E.; Deppert, K.; Samuelson, L. *Nat. Nanotechnol.* **2009**, *4* (1), 50–55.
- (14) Kim, D. S.; Scholz, R.; Gösele, U.; Zacharias, M. *Small* **2008**, *4* (10), 1615–1619.
- (15) Colli, A.; Hofmann, S.; Ferrari, A. C.; Ducati, C.; Martelli, F.; Rubini, S.; Cabrini, S.; Franciosi, A.; Robertson, J. Appl. Phys. Lett. **2005**, *86* (15), 153103.
- (16) Lehmann, O.; Stuke, M. Science 1995, 270 (5242), 1644–1646.
- (17) Wanke, M. C.; Lehmann, O.; Müller, K.; Wen, Q.; Stuke, M. Science 1997, 275 (5304), 1284– 1286.
- (18) Cao, L.; Barsic, D. N.; Guichard, A. R.; Brongersma, M. L. Nano Lett. 2007, 7 (11), 3523–3527.
- (19) Boyd, D. A.; Greengard, L.; Brongersma, M.; El-Naggar, M. Y.; Goodwin, D. G. *Nano Lett.* **2006**, *6* (11), 2592–2597.
- (20) Kim, B. J.; Wen, C.-Y.; Tersoff, J.; Reuter, M. C.; Stach, E. A.; Ross, F. M. Nano Lett. 2012, 12 (11), 5867–5872.
- (21) Gamalski, A. D.; Tersoff, J.; Sharma, R.; Ducati, C.; Hofmann, S. *Nano Lett.* **2010**, *10* (8), 2972–2976.

- (22) Okamoto, H.; Massalski, T. B. Bull. Alloy Phase Diagr. 1984, 5 (6), 601–610.
- (23) Zhao, Q.; Zhou, J.; Zhang, F.; Lippens, D. Mater. Today 2009, 12 (12), 60–69.
- (24) Huang, L.; Yu, Y.; Cao, L. Nano Lett. 2013, 13 (8), 3559–3565.
- (25) Fu, Y. H.; Kuznetsov, A. I.; Miroshnichenko, A. E.; Yu, Y. F.; Luk'yanchuk, B. *Nat. Commun.* **2013**, *4*, 1527.
- (26) Cao, L.; White, J. S.; Park, J.-S.; Schuller, J. A.; Clemens, B. M.; Brongersma, M. L. *Nat. Mater.* **2009**, *8* (8), 643–647.
- (27) van de Groep, J.; Polman, A. *Opt. Express* **2013**, *21* (22), 26285–26302.
- (28) Michaelis, F. B.; Weatherup, R. S.; Bayer, B. C.; Bock, M. C. D.; Sugime, H.; Caneva, S.; Robertson, J.; Baumberg, J. J.; Hofmann, S. ACS Appl. Mater. Interfaces 2014, 6 (6), 4025– 4032.
- (29) Bock, M. C. D.; Denk, R.; Wirth, C. T.; Goldberg-Oppenheimer, P.; Hofmann, S.; Baumberg, J. J. *Appl. Phys. Lett.* **2012**, *100* (1), 013112.
- (30) Ferreira, P. j.; Mitsuishi, K.; Stach, E. a. *MRS Bull.* **2008**, *33* (02), 83–90.
- (31) Ziegler, C.; Eychmüller, A. J. Phys. Chem. C 2011, 115 (11), 4502–4506.

Table of Contents

