



A holistic approach to interface stabilization for efficient perovskite solar modules with over 2,000-hour operational stability

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1 **A holistic approach to interface stabilization for efficient perovskite solar**
2 **modules with over 2000 hour operation stability**

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17 Upscaling of perovskite solar cells to module scale and affording long-term
18 stability have been recognized as the most important challenges for commercialization
19 of this emerging photovoltaic technology. In a perovskite solar module (PSM), each
20 interface within the device contributes to the efficiency and stability. Here, we employ
21 a holistic interface stabilization strategy by modifying all the relevant layers and
22 interfaces, namely the perovskite layer, charge transporting layers and the device
23 encapsulation to improve the efficiency and stability of PSMs. The treatments were
24 selected to be compatible with low-temperature scalable processing and the module
25 scribing steps. Our unencapsulated PSM achieved a reverse-scan efficiency of 16.6%
26 with a designated area of 22.4 cm². The encapsulated PSM retained approximately 86%

27 of the initial performance after continuous operation for 2000 h under AM 1.5G light
28 illumination, with translates into a T_{90} lifetime of 1570 h and an estimated T_{80} lifetime
29 of 2680 h.

30

31 **Introduction**

32 In the past decade, small-size perovskite solar cells (PSCs) with an active area of
33 $\sim 0.1 \text{ cm}^2$ have achieved outstanding power conversion efficiencies (PCE) over 25%.¹
34 For practical applications, PSCs must retain high efficiency when up-scaled. However,
35 when their size was increased to module scale with active area above 10 cm^2 the
36 efficiency decreased significantly. The efficiency of 10-cm^2 perovskite solar modules
37 (PSMs) is currently in the range of $10 \sim 17\%$.²⁻⁴ As the PCE is still much lower than
38 that of commercialized solar modules based on other photovoltaic technologies it is
39 desirable to further improve the PSMs efficiency.

40 Besides up-scalability and efficiency, the long-term stability of PSMs is another
41 important aspect for their practical application.^{5,6} In particular, the device operation
42 stability under continuous illumination is an important indicator of the long-term
43 stability of PSCs, as well as PSMs.^{5,7-9} To improve the continuous operation stability
44 of PSCs, the engineering of the individual functional layer and the relevant interfaces
45 including perovskite layer¹⁰⁻¹⁴, electron transport layer (ETL)¹⁵, hole transport layer
46 (HTL)¹⁶, counter electrode (CE)¹⁷ and encapsulation layer¹⁸ has been developed in
47 separate works. Although fine control of each interface has been proven to be of chief
48 importance for the photovoltaic performance of PSCs, similar interface engineering
49 strategies have not been widely considered and systematically investigated in the
50 context of PSMs. This has been proposed to be the main reason why the PSCs stability
51 reduce quickly when scaling up.^{2,4} It is thus desirable to transfer the accumulated
52 successful experiences in PSCs to PSMs.

53 Since each functional layer in a PSM contributes to its efficiency and stability, a
54 holistic approach to the optimization of the interfaces of each functional layer
55 throughout the entire device must be employed rather than the engineering of an

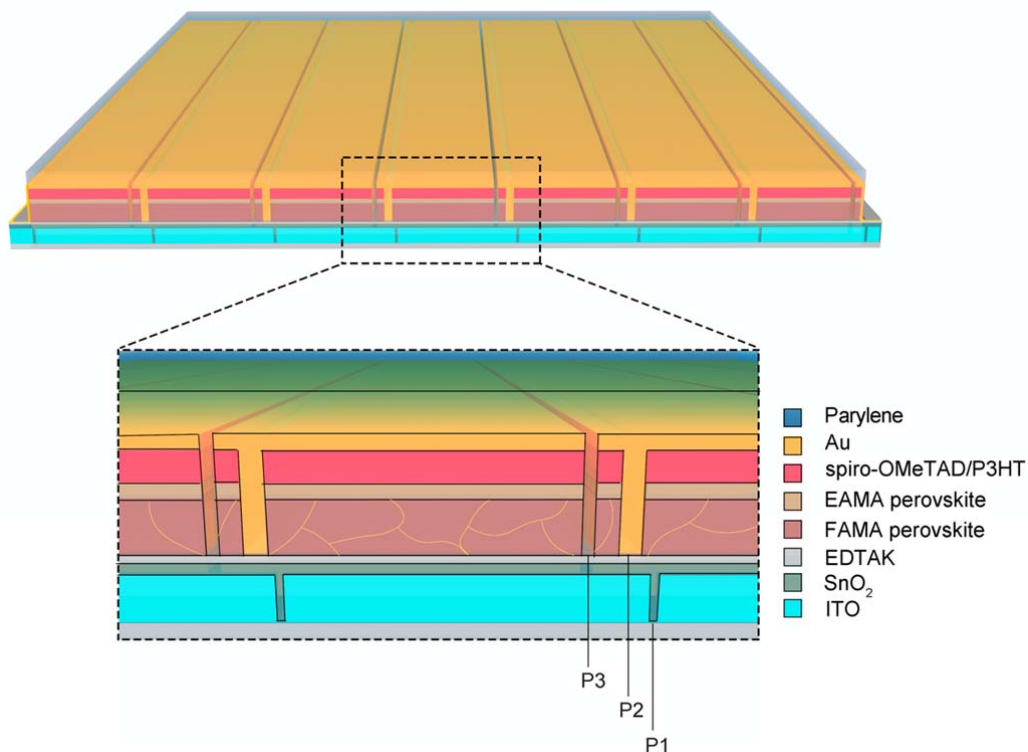
56 individual interface. However, the structure of PSMs is complex compared to
57 small-area cells due to the series connections of multiple sub-cells and P1-P2-P3
58 interconnection structures. The interface engineering strategies employed in lab-scale
59 PSCs that work for small-area cell are not always trivial to transfer to PSMs. For
60 example, TiO₂ ETL has played a pivotal role in high-efficiency PSCs, but if one does
61 not remove TiO₂ in the P2 patterning step for PSM, the relatively high resistance of
62 TiO₂ will cause a substantially lower FF.^{19,20} Besides, the exposure of P2 and P3
63 patterning lines also increase the possibility for external stressor induced degradation.²¹
64 Therefore, a holistic approach to PSM design that considers the process compatibility
65 with scalable production as well as the module structure and encapsulation is essential
66 for transferring the strategies of PSCs to PSMs.

67 Here, we report a holistic interface stabilization (HIS) strategy that takes into
68 consideration the engineering of all the relevant interfaces in a perovskite solar device
69 at the module scale. More specifically, we stabilize the SnO₂ ETL with
70 ethylenediaminetetraacetic acid dipotassium (EDTAK), reduce the defects in the
71 perovskite surface with an ethylammonium iodide (EAI) treatment, inhibit the
72 moisture ingress and Au migration into the
73 2,2',7,7'-tetrakis[N,N-bis(p-methoxyphenyl)amino]-9,9'-spirobifluorene
74 (spiro-OMeTAD) HTL with the incorporation of poly(3-hexylthiophene) (P3HT) and
75 ensure an effective device encapsulation with a parylene film. These treatments are
76 not only compatible with scalable processing but also with scribing steps. Our HIS
77 strategy leads to not only PSMs efficiency improvement, but also stability
78 enhancement. The PSMs without encapsulation achieved a reverse-scan efficiency of
79 16.6% with a designated area of 22.4 cm². The encapsulated PSM with the best
80 stability gave T₉₀ and T₈₀ operation stability lifetime of 1570 h and 2680 h,
81 respectively, under AM 1.5G light illumination test.

82

83 **Holistic interface stabilization approach**

84 We adopt an indium tin oxide (ITO)/SnO₂/Cs_{0.05}FA_{0.54}MA_{0.41}Pb(I_{0.98}Br_{0.02})₃ (FA is
 85 formamidinium, MA is methylammonium)/spiro-OMeTAD/Au configuration as the
 86 basis to fabricate PSMs.^{22,23} To improve both efficiency and stability, we devise an
 87 HIS strategy that optimize all the relevant layers and interfaces from the bottom ETL
 88 layer to the top encapsulation layer as shown in Figure 1. We consider the
 89 compatibility of each individual interface modification with the other modifications,
 90 with the PSM design and with the scalable manufacturing. Such integrated approach
 91 should facilitate the future transfer of the optimized device design to practical
 92 manufacturing processes.



93 **Figure 1 Holistic interface stabilization (HIS) strategy for PSMs.** The HIS
 94 strategy consists of four treatments for the main device layers and their interfaces: the
 95 use of EDTAK to modify SnO₂; the use of EAI/MAI to form the EAMA-based
 96 perovskite on the surface of Cs_{0.05}FA_{0.54}MA_{0.41}Pb(I_{0.98}Br_{0.02})₃ (FAMA based
 97 perovskite); the incorporation of P3HT into the spiro-OMeTAD layer; the
 98 employment of the parylene encapsulation.
 99

100

101 Regarding ETL and ETL/perovskite interface, we employ a commercial SnO₂ as
 102 ETL to take advantage of its efficient electron-selective properties²⁴, and its

103 compatibility with low-temperature and scalable coating process²². The use of
104 approximately 20 nm-thick SnO₂ layer also enables a low resistance of P2
105 interconnection to ensure low series resistance and high fill factor.^{19,25} Furthermore, we
106 use EDTAK to mitigate the reaction of OH⁻ in SnO₂ with perovskite and tune the
107 energy-level alignment at the ETL/perovskite interface. This modification also has
108 negligible influence on the conductivity of the SnO₂ layer, which is important to ensure
109 high-quality interconnection with low series resistance between each sub-cell in PSMs.

110 Regarding the perovskite/HTL interface, we use EAI/MAI surface treatment to
111 passivate the surface defects of Cs_{0.05}FA_{0.54}MA_{0.41}Pb(I_{0.98}Br_{0.02})₃ perovskite (PVSK)
112 and tune the energy-level alignment at this interface. The formed EAMA-perovskite
113 could also reduce the influences from moisture in both top surface region and the
114 regions related to wiring and encapsulation.

115 We then incorporate P3HT into the spiro-OMeTAD layer to induce a stable HTL
116 against moisture and Au migration during device operation and P2 cutting process.²¹

117 These steps, the coating of SnO₂ with EDTAK followed by the deposition of the
118 perovskite layer and the subsequent EAI/MAI treatment as well as the deposition of the
119 HTL are highly compatible with scalable coating processes developed for organic and
120 dye-sensitized solar modules.^{2,4}

121 On the top of metal electrode, parylene encapsulation was made via upscalable
122 chemical vapor deposition. Parylene encapsulation acts as a uniform and robust barrier
123 layer, which helps keep the whole module including the P2 and P3 patterning lines into
124 a closed space to prevent diffusion of external ambient gases such as moisture and
125 oxygen, and contain the internally generated volatile degradation products.²⁶

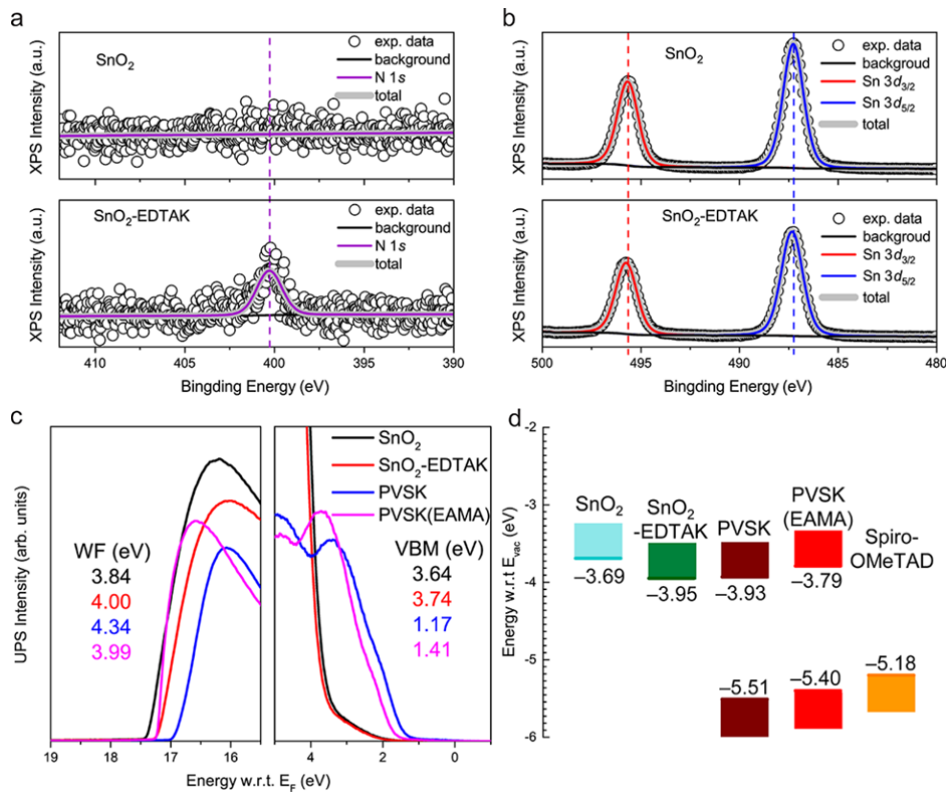
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127 **Stabilization of the ETL/perovskite interface**

128 A stable ETL/perovskite contact and suitable energy-level alignment at this
129 interface is essential for efficient and stable PSCs. The first step of our HIS strategy is
130 to mitigate the reaction of OH⁻ in SnO₂ with perovskite (Supplementary Figure 1 and
131 Supplementary Note 1) to form a stable ETL/perovskite interface by using EDTAK to

132 post-treat the SnO₂ layer on the basis of the acid-base neutralization reaction. The use
 133 of EDTAK also retained the positive passivation effect of K⁺.²² The treatment
 134 condition was optimized by changing the EDTAK concentration (Supplementary
 135 Table 1).

136 The observed N 1s peak in X-ray photoelectron spectroscopy (XPS) results in
 137 Figure 2a confirms the existence of EDTAK on the surface of the EDTAK treated
 138 SnO₂ (SnO₂-EDTAK) films. By immersing the SnO₂-EDTAK sample and pristine
 139 sample (SnO₂) into water and testing the pH value of the obtained aqueous solution,
 140 the pH value is slightly alkaline for the control sample, and neutral for the
 141 SnO₂-EDTAK sample. Based on the above observation, the reaction of KOH with
 142 perovskite is believed to be effectively mitigated by the EDTAK treatment.



143

144 **Figure 2 ETL/perovskite and perovskite/HTL interface stabilization and**
 145 **energy-level alignment. a-b,** The N 1s spectra (a) and Sn 3d spectra (b) of SnO₂ and
 146 SnO₂-EDTAK films measured by XPS measurements. In (a) and (b), black circles
 147 show the original XPS data, black curves represent the background, the grey bold
 148 curves show the total fitting results. In (a), the purple vertical dash line shows the N
 149 1s position with respect to SnO₂-EDTAK, the purple curves are the fitting results of

150 $N1s$ signal. In **(b)**, the red and blue vertical dash line show the Sn $3d_{3/2}$ and Sn $3d_{5/2}$
151 position, respectively, with respect to pristine SnO_2 , the red and blue curves are the
152 fitting results of Sn $3d_{3/2}$ and Sn $3d_{5/2}$ signal, respectively. **c**, UPS spectra (using the
153 He-I line with a photon energy of 21.22 eV) corresponding to the secondary electron
154 onset region and valence band maximum (VBM) of SnO_2 , SnO_2 -EDTAK, perovskite
155 (PVSK), EAI/MAI-treated perovskite (PVSK(EAMA)) with respect to the Fermi
156 energy (E_F). VBM for perovskites was determined from the semi-log plots
157 (Supplementary Figure 3). **d**, Diagram of the energy levels of the materials as
158 extracted from the UPS data.

159 Unlike a previous study²⁷ using EDTA as a pretreatment modifier to modify the
160 SnO_2 colloidal solution, in the current work EDTAK is used as an interface
161 post-treatment modifier to tune the energy-level alignment at the ETL/perovskite
162 interface.²⁸ To investigate this point, XPS and ultraviolet photoemission spectroscopy
163 (UPS) measurements were conducted. The observed slight shift of the Sn $3d$ peaks of
164 SnO_2 -EDTAK with respect to SnO_2 (Figure 2b) suggests a chemical interaction of
165 EDTAK with SnO_2 , which is likely to affect the energy-level alignment at the
166 ETL/perovskite interface. As shown in Figure 2c, the work function (WF) of pristine
167 SnO_2 and SnO_2 -EDTAK is determined to be 3.84 eV and 4.00 eV, respectively. On the
168 basis of our UPS and XPS analysis, EDTAK most likely act as interface modifier
169 leading to variations in the WF of SnO_2 -EDTAK, while minimum changes are
170 observed for VBM and XPS Sn $3d$ core levels (see Supplementary Figure 2 and
171 Supplementary Note 2).

172 The conduction band minimum (CBM) of the pristine SnO_2 was calculated to be
173 -3.69 eV (with respect to the vacuum level E_{vac}), in agreement with reported
174 values.^{24,25} The CBM of SnO_2 -EDTAK film shifts downward by about 0.26 eV with
175 respect to the pristine SnO_2 film, which was calculated to be -3.95 eV. The improved
176 energy-level alignment with CBM (-3.93 eV) of the perovskite (Figure 2d and
177 Supplementary Figure 3) is expected to be beneficial for charge extraction at
178 perovskite/ ETL interface.

179 To verify this, time resolved photoluminescence (TRPL) measurements were
180 conducted (Supplementary Figure 4). The faster quench of perovskite PL for

181 SnO₂-EDTAK versus SnO₂ indicates a faster charge transfer from perovskite to
182 SnO₂-EDTAK.

183 We also used secondary ion mass spectrometry (SIMS) measurements to
184 investigate whether K⁺ in EDTAK-SnO₂ can migrate into the perovskite film
185 (Supplementary Figure 5). The SIMS result confirms that the K⁺ is distributed in both
186 perovskite bulk film and in the EDTAK-SnO₂ film. The K⁺^{22,29} and Lewis base of the
187 alkylamine group³⁰ in EDTAK-SnO₂ could passivate perovskite defects, which is
188 supported by the longer PL lifetime of the glass/EDTAK/perovskite sample versus the
189 glass/perovskite sample (Supplementary Figure 4 and Supplementary Table 2).

190 To study the effect of the EDTAK treatment on device performance, small-size
191 PSCs were fabricated (Supplementary Figure 6). The current density-voltage (J-V)
192 curves of the devices based on pristine SnO₂ and EDTAK-SnO₂ are shown in Figure
193 3a. The pristine SnO₂ based PSCs (denoted as SnO₂/PVSK/spiro-OMeTAD/Au) show
194 a typical PCE of 19.2% with a short-circuit current density (J_{SC}) of 23.2 mA cm⁻², an
195 open-circuit voltage (V_{OC}) of 1.07 V, and a fill factor (FF) of 77.6%. EDTAK-SnO₂
196 based PSCs show a J_{SC} of 23.2 mA cm⁻², a V_{OC} of 1.10 V and a FF of 78.4%, leading
197 to an improved PCE of 20.1% with suppressed hysteresis (Supplementary Table 3). As
198 shown in Figure 3b, the representative external quantum efficiency (EQE) of
199 SnO₂-EDTAK device shows slightly higher values than the SnO₂ device at both the
200 short-wavelength and long-wavelength region, leading to a slightly higher integrated
201 J_{sc} for the former case (22.8 mA cm⁻² versus 22.4 mA cm⁻²). The improved J_{sc} as
202 well as improved V_{OC} and FF are likely due to dual function of the EDTAK treatment,
203 i.e., improvement of charge transport and suppression of charge recombination as
204 discussed above.

205 **Stabilization of the perovskite/HTL interface**

206 The second interface stabilization step is focused on the perovskite/HTL interface.
207 The higher defect density at the perovskite surface, especially at the surface grain
208 boundaries, with respect to the perovskite/ETL interface usually affects device
209 performance and stability significantly.^{14,31} Here, interface stabilization with

210 EAI/MAI modification is adopted on the surface of perovskite to passivate defects as
211 well as to enhance the device stability due to the robust stability of the EAMA based
212 perovskite³²⁻³⁴. With respect to long-alkyl or other large-size ammonium-based
213 cations, EA⁺ has a relative smaller size, which favors the formation of quasi-3D
214 perovskite even with a relative high EA content (Supplementary Note 3,
215 Supplementary Figure 7-9 and Supplementary Table 4).³⁵⁻³⁷

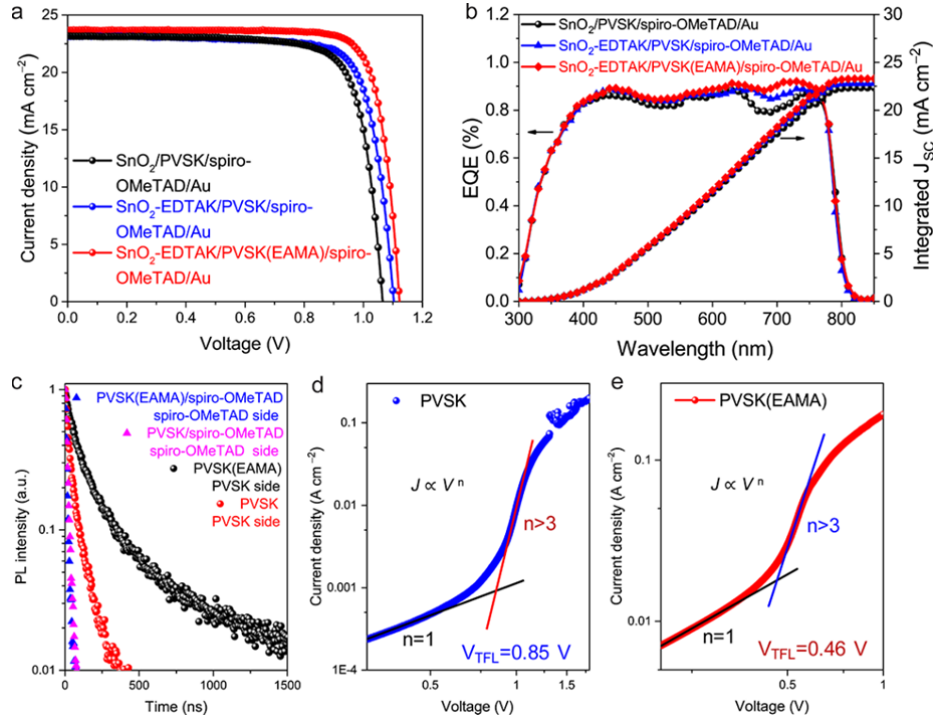
216 To check the effect of the EAI/MAI treatment on the device performance
217 (Supplementary Figure 10), small-size PSCs were fabricated, a typical J-V curve is
218 shown in Figure 3a. The devices with both EDTAK and EAI/MAI treatment (denoted
219 as SnO₂-EDTAK/PVSK(EAMA)/spiro-OMeTAD/Au, PVSK(EAMA) is
220 Cs_{0.05}FA_{0.54}MA_{0.41}Pb(I_{0.98}Br_{0.02})₃ perovskite with the EAI/MAI treatment) show a
221 typical PCE of 21.8% with a J_{SC} of 23.6 mA cm⁻², a V_{OC} of 1.12 V, and a FF of
222 82.3%. Based on EQE result in Figure 3b, the integrated current density is calculated
223 to be 23.3 mA cm⁻², which is consistent with J_{SC} obtained from the J-V curve. This
224 value is also higher than that of the device without treatment
225 (SnO₂/PVSK/spiro-OMeTAD/Au) and device with EDTAK treatment
226 (SnO₂-EDTAK/PVSK/spiro-OMeTAD/Au). Especially, the device with both EDTAK
227 and EAI/MAI treatment shows higher EQE values at the long-wavelength region than
228 that of device with EDTAK treatment. Considering that long-wavelength light is
229 mainly absorbed by the perovskite near the perovskite/HTL interface. The improved
230 current density is proposed attributed to the passivation effect of EAI/MAI treatment.
231 Besides, the improved FF and V_{OC} also contribute to significantly improved PCE
232 from 20.1% to 21.8% with EAI/MAI treatment.

233 To verify the passivation effect of EAI/MAI treatment, we used TRPL to
234 investigate the charge carrier properties of the perovskite with and without EAI/MAI
235 treatment (Figure 3c and Supplementary Table 5). The EAI/MAI treated perovskite
236 sample showed a much longer PL lifetime than that of the untreated perovskite sample
237 (122.4 ns versus 42.8 ns), which is likely the result of a reduction in nonradiative
238 recombination by defect passivation via the EAI/MAI treatment. The defect density

239 reduction from 0.86×10^{16} to $0.47 \times 10^{16} \text{ cm}^{-3}$ measured by space-charge-limited
240 current (SCLC) test further confirms the passivation effect of the EAI/MAI treatment
241 (Figure 3d and e and Supplementary Note 4).

242 In addition, the EAI/MAI treatment affects the energy-level alignment at the
243 perovskite/HTL interface. As shown in Figure 2c and d, the WF of EAI/MAI treated
244 perovskite and untreated perovskite is determined to be 3.99 eV and 4.34 eV,
245 respectively. The VBM is determined to be 1.41 eV and 1.17 eV, respectively,
246 yielding the ionization energy value of 5.40 eV and 5.51 eV for perovskite with and
247 without EAI/MAI treatment, respectively. The relative energy level position of the
248 Fermi level shift also indicates a more *n*-type nature for the top surface of perovskite
249 after the EAI/MAI treatment (Figure 2c, 2d, and Supplementary Figure 3), which is
250 likely to originate from the partial change of the lead halide rich surface to a more
251 organic halide rich surface induced by the EAI/MAI treatment.³⁸

252 The mismatch of the energy-level alignment between perovskite and
253 spiro-OMeTAD (ionization energy, 5.18 eV³⁹) is reduced from 0.33 eV to 0.22 eV
254 after the EAI/MAI treatment, which may benefit charge extraction at the
255 perovskite/HTL interface. A faster quenching of perovskite PL for the
256 PVSK(EAMA)/spiro-OMeTAD in comparison with PVSK/spiro-OMeTAD sample
257 measured by TRPL (Figure 3c and Supplementary Table 5), further suggests that
258 faster charge transfer occurs from EAI/MAI treated perovskite to spiro-OMeTAD.
259 This also contributes to the suppressed hysteresis for the PSCs with both EDTAK and
260 EAI/MAI treatment versus the PSCs with EDTAK treatment (Supplementary Table
261 3).



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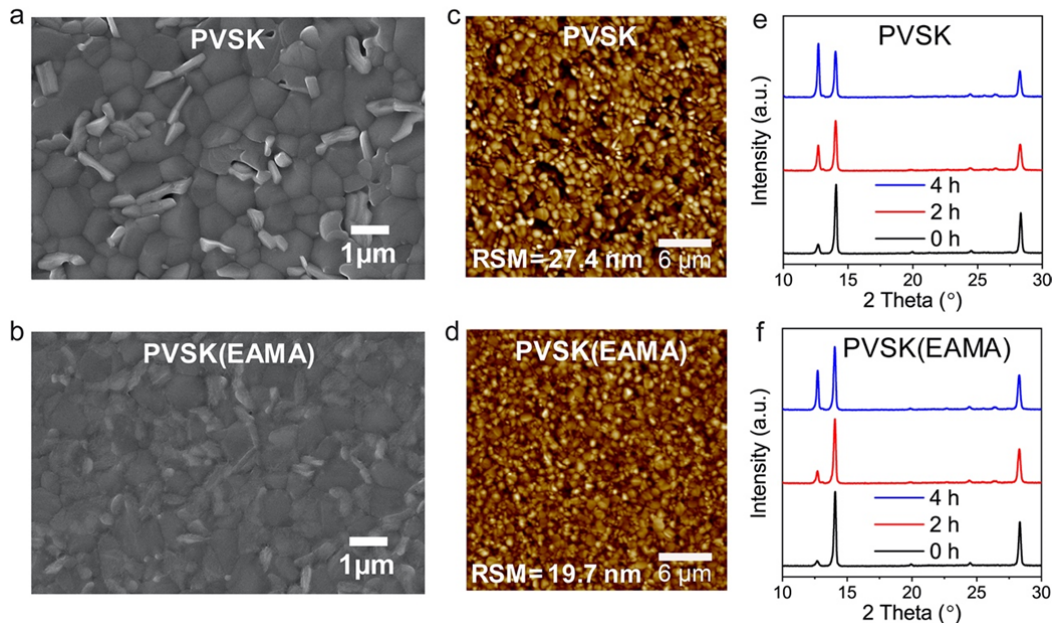
263 **Figure 3 PSCs performance with ETL/perovskite and perovskite/HTL interface**
 264 **stabilization treatments and passivation effect of EAI/MAI treatment. a, J-V**
 265 **curves and b, EQE spectra and integrated J_{sc} of the PSCs treated with both EDTAK**
 266 **and EAI/MAI (red curve), treated with EDTAK (blue curve) and without treatments**
 267 **(black curve). c, TRPL decay curves of the perovskite thin film with (blue and black**
 268 **curves) or without (pink and red curves) EAI/MAI treatment as bare film (black and**
 269 **red curves) or interfaced with spiro-OMeTAD (blue and pink curves). d-e, J-V**
 270 **characteristics of SCLC result. The J-V curve can be fitted according to $J \propto V^n$ with**
 271 **different values of the exponent n: $n = 1$ is the ohmic region (dark fitted lines in (d)**
 272 **and (e)); $n > 3$ is the trap-filled limited (TFL) region (red fitted line and blue fitted**
 273 **line in (d) and (e), respectively). The TFL voltage (V_{TFL}) is onset voltage of the TFL**
 274 **region.**

275 To further explore how the EAI/MAI treatment affects the perovskite film
 276 stability, the morphology of the perovskite with and without EAI/MAI treatment is
 277 measured by scanning electron microscopy (SEM) (Figure 4a-b). The grain
 278 boundaries are filled with additional structure for the EAI/MAI treated perovskite film
 279 with reduced roughness from 27.4 to 19.7 nm confirmed by atomic force microscopy
 280 (AFM) measurements (Figure 4c-d).

281 It has been reported that perovskite decomposition mainly starts from grain
 282 boundaries due to the relatively weak chemical binding and severe ion migration.^{14,31}

283 The newly formed EAMA perovskite is thus expected to improve the perovskite
 284 stability due to passivation of grain boundaries. To verify this, the perovskite films are
 285 further subjected to high-energy electron beam with a voltage of 20 kV under SEM
 286 measurements (Supplementary Figure 11). Obvious cracks are observed for the
 287 untreated perovskite film, while the EAI/MAI treated perovskite film remains densely
 288 packed grain, which indicates an enhanced stability of latter case.³⁰

289 To further study the perovskite stability, the perovskite films with and without
 290 EAI/MAI treatment are subject to a high relative humidity close to 100% at
 291 approximately 40 °C by monitoring the XRD evolution (Figure 4e-f). The PbI₂ peak
 292 in the untreated perovskite film shows much faster increase than that in the EAI/MAI
 293 treated perovskite film, which indicates slower degradation of EAI/MAI treated
 294 perovskite. The 85 °C thermal aging test of the films in a dry N₂ glove box also
 295 suggests the improved stability of EAI/MAI treated perovskite film (Supplementary
 296 Figure 12). These XRD observations together with the above morphology study
 297 coincidentally suggest that the stability of perovskite film can be improved via
 298 EAI/MAI treatment.³¹



299

300 **Figure 4 Perovskite film stability with perovskite/HTL interface stabilization.**

301 **a-b** Surface SEM images of the perovskite films without (PVSK) (**a**) and with
 302 (PVSK(EAMA)) (**b**) EAI/MAI treatment. **c-d**, AFM topography of the perovskite
 303 films without (**c**) and with (**d**) EAI/MAI treatment, RSM is root mean squared surface

304 roughness. **e-f**, XRD patterns evolution of the perovskite films without (**e**) and with
305 (**f**) EAI/MAI treatment under aging test in a high relative humidity close to 100%
306 condition at approximately 40 °C.

307 To apply the above studied interface stabilization strategy into large-scale device,
308 PSMs were fabricated based on mirror-like and uniform large-area perovskite films
309 (Supplementary Figure 13). Figure 5a shows the optical image of the PSM and Figure
310 5b shows the J-V curves of the PSMs with a designated area of 22.4 cm² obtained
311 from reverse scan.

312 Consistent with the PSCs performance, the PSM with both EDTAK and
313 EAI/MAI treatment (SnO₂-EDTAK/PVSK(EAMA)/spiro-OMeTAD/Au) showed
314 substantially better performance (Supplementary Table 6). A champion module with
315 both EDTAK and EAI/MAI treatment show a reverse-scan PCE of 16.6%, which
316 corresponds to an active-area-efficiency of 18.2% (Supplementary Figure 14). To
317 confirm our J-V results, we sent our module to National Institute of Advanced
318 Industrial Science and Technology (AIST) for J-V characterization according to their
319 standard protocol IEC60904-3 Ed. 3 (Supplementary Figure 15). The result is
320 relatively consistent with our in-house measurements (Supplementary Figure 16).

321 The PSMs with both EDTAK and EAI/MAI treatment also exhibit good
322 reproducibility (Supplementary Figure 17, Supplementary Table 7) and a small
323 hysteresis (Supplementary Figure 18). To study the operation stability of above PSMs
324 without encapsulation, we tracked the operation stability at a fixed bias that was set to
325 be the initial MPP voltage under continuous AM 1.5G light illumination in a N₂ box
326 with a relative humidity <5% (Figure 5c). The initial values of the PV parameters for
327 the devices tested for stability are reported in Supplementary Table 6. It was found
328 that the T₈₀ lifetime of the PSMs is estimated to be 24, 116 and 268 h, respectively.
329 The significantly improved device stability for the modules with both EDTAK and
330 EAI/MAI treatment is ascribed to the above-mentioned ETL/perovskite and
331 perovskite/HTL interface stabilization strategy.

332

333 **Stabilization of the HTL/electrode interface**

334 A stable HTL/electrode interface is also important for PSM stability.⁴¹ To further
335 improve the stability of HTL, we incorporated a small amount of P3HT into
336 spiro-OMeTAD.⁴² This mixed HTL (denoted as spiro-OMeTAD-P3HT) can deliver
337 comparable performance with the spiro-OMeTAD based PSM (Supplementary Figure
338 19 and 20, Supplementary Table 6). To study whether this strategy can improve the
339 stability, we measured the operation stability of PSM based on mixed HTL
340 (SnO_2 -EDTAK/PVSK(EAMA)/spiro-OMeTAD-P3HT/Au). It was found that the
341 module with both EDTAK and EAI/MAI treatment and mixed HTL shows a T_{80}
342 lifetime exceeding 1600 h, which is much longer than the pure spiro-OMeTAD case
343 (Figure 5c). We ascribe the improved stability to the HTL/CE interface engineering,
344 where the spiro-OMeTAD-P3HT layer may slow down the gold inward migration and
345 also enhance the stability of HTL against moisture. To verify this, we performed
346 SIMS measurements on the aged samples (Supplementary Figure 21). The SIMS
347 results coherently indicate that the slower Au migration into spiro-OMeTAD-P3HT
348 versus spiro-OMeTAD case, which slow down the irreversible degradation induced by
349 Au migration.⁴⁰ The improved shielding effect of the spiro-OMeTAD-P3HT is
350 assigned to the thiophene-like structures of P3HT that has a strong chemical
351 interaction with Au, which helps retard Au migration (Supplementary Figure 22 and
352 Supplementary Note 5).

353 In addition, moisture resistance of the HTLs was studied by measuring the
354 contact angle of a deionized water droplet. The contact angle is increased from 71.6°
355 for the spiro-OMeTAD sample to 107.0° for the spiro-OMeTAD-P3HT sample
356 (Supplementary Figure 23). This observation evidently demonstrates a change of HTL
357 from hydrophilic to hydrophobic through introducing additional conjugated polymers,
358 i.e., P3HT into spiro-OMeTAD. The improved moisture resistance is beneficial for the
359 device resistance against moisture during P2 cutting process and device operation.
360 Note that when water droplets were kept on the top of spiro-OMeTAD-P3HT layer for
361 several minutes during the contact angle test, the underneath perovskite still changed
362 its color from dark brown to yellow, which suggests the spiro-OMeTAD-P3HT is not

363 able to completely retard the moisture induced degradation. This observation inspired
364 us to make additional interface stabilization to further improve the PSMs stability.

365

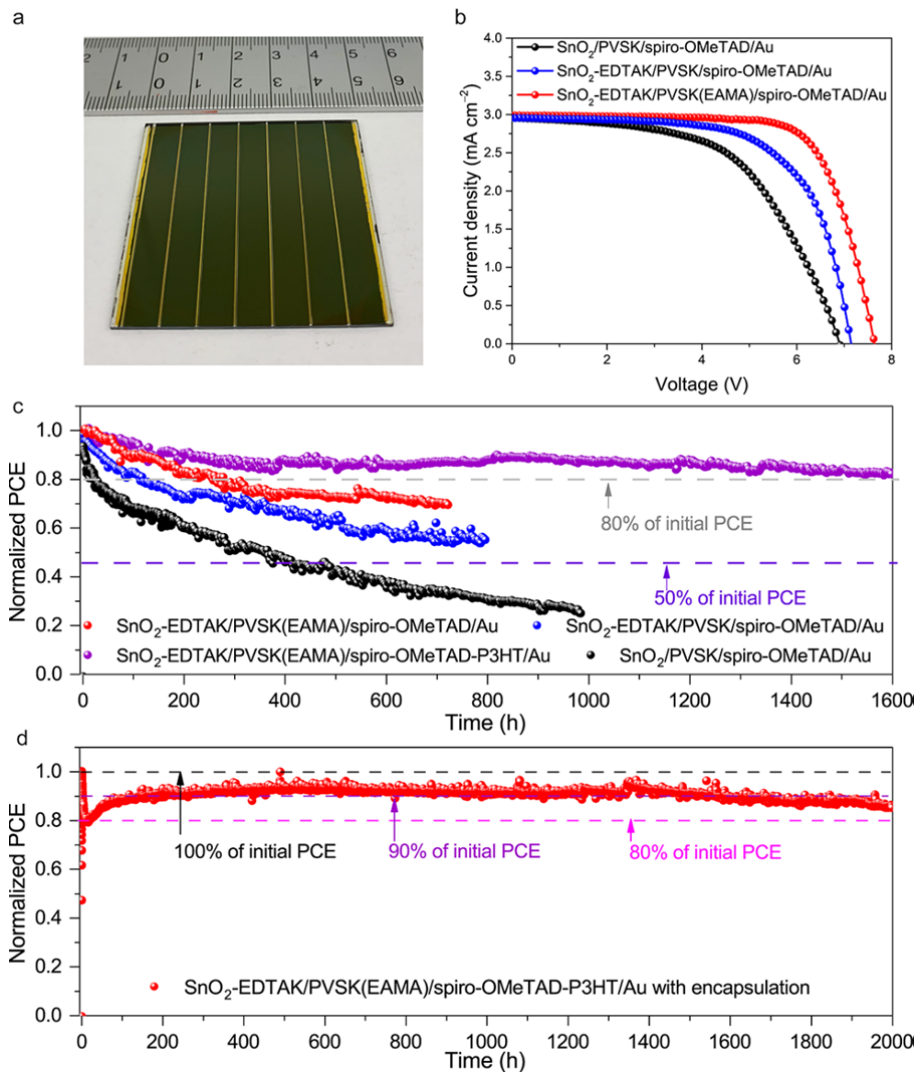
366 **Device encapsulation**

367 To further improve PSMs stability, parylene encapsulation⁴³ and cover glass
368 protection are employed on PSMs with both EDTAK and EAI/MAI treatment and
369 mixed HTL (Supplementary Figure 24-26). The encapsulated PSMs show a
370 comparable efficiency as PSMs without encapsulation (Supplementary Figure 27).

371 The operation stability of three encapsulated PSMs was investigated (Figure 5d
372 and Supplementary Figure 28), and the temperature on the surface of the module was
373 measured to be approximately 40 °C during stability measurements (Supplementary
374 Figure 29). All three modules show nearly identical stability evolution trend. At the
375 beginning, there is a very fast burn-in decay process. This might due to the thermal
376 effect, a faster perovskite degradation or spiro-OMeTAD conductivity variation
377 (Supplementary Note 6). We then observed a slightly increase of the performance
378 followed by a slow linear decay, where the slow increase after burn-in decay is likely
379 due to the perovskite recovery or spiro-OMeTAD conductivity recovery
380 (Supplementary Note 6). Figure 5d shows that the encapsulated PSM with the best
381 stability remained 86% of the initial performance after aging for 2000 h, and gave a
382 T_{90} lifetime of 1570 h and an estimated T_{80} lifetime of 2680 h.³¹ Moreover, all three
383 modules show a T_{80} lifetime over 2000 h with good stability reproducibility, which
384 give an average T_{80} lifetime of 2457 ± 274 h.

385 Here, the parylene encapsulation is demonstrated to act as a thin-film barrier
386 largely to enhance the long-term stability of PSMs (Supplementary Note 7). The
387 thermal stability (60 °C) of the PSMs was also studied. The PSMs retained over 80%
388 of their initial efficiency after 1000 h test (Supplementary Figure 30 and
389 Supplementary Table 8). When compared with the reported results, our modules
390 performance result is among the top reported efficiency and stability for PSMs
391 (Supplementary Table 9). Our PSMs stability study suggests the relatively good

392 stability of our solar module and verify the effectiveness of above studied HIS
 393 strategy.



394
 395 **Figure 5 PSMs photovoltaic performance and stability.** **a**, Photograph of a
 396 PSM. **b**, Representative J–V curves of the PSMs with varying structures. **c**, Operation
 397 stability of un-encapsulated PSMs with varying structures measured at a fixed bias
 398 near initial maximum power point (MPP) under continuous AM 1.5G light
 399 illumination in a N₂ box with a relative humidity <5%. The initial values of the PV
 400 parameters for the devices tested for stability are reported in Supplementary Table 6.
 401 **d**, Operation stability of encapsulated
 402 SnO₂-EDTAK/PVSK(EAMA)/spiro-OMeTAD-P3HT/Au PSM with parylene and a
 403 cover glass measured under the same conditions as in (b). The initial values of the PV
 404 parameters for the PSM tested for stability are reported in Supplementary Figure 27,
 405 PSM #1.

406

407 Conclusions

408 In summary, we demonstrated a holistic interface engineering strategy to
409 fabricated PSMs relying on treatments that are compatible with scalable processing
410 and scribing steps. First, we treated the SnO₂ ETL with EDTAK to mitigate the
411 reaction of OH⁻ with perovskite, passivate defects and afford good energy-level
412 alignment at the ETL/perovskite interface. Then, we passivated surface defects at the
413 perovskite/HTL surface with ethylammonium iodide as well as improved perovskite
414 stability and energy-level alignment at the perovskite/HTL interface. We then
415 improved the device stability by incorporating P3HT into the HTL and encapsulate the
416 device with a parylene film. The resultant PSMs without encapsulation achieved a
417 reverse-scan efficiency of 16.6% with a designated area 22.4 cm². Moreover, the
418 encapsulated PSM stability maintained about 86% of the initial performance after
419 continuous operation for 2000 h under AM 1.5G light illumination, which translates
420 into a T₉₀ lifetime of 1570 h and a T₈₀ lifetime of 2680 h. This demonstration is a step
421 forwards in the development and commercialization of large-scale perovskite
422 photovoltaics.

423

424 **Methods**

425 **Materials.** All reagents were used as received without further purification,
426 including PbI₂ (99.99%, TCI), formamidinium iodide (FAI, greatcell solar),
427 methylammonium iodide (MAI, greatcell solar), methylammonium bromide (MABr,
428 greatcell solar), methylammonium chloride (MACl, greatcell solar), ethylammonium
429 iodide (EAI, Sigma-Aldrich), cesium iodide (Sigma-Aldrich), 4-tert-butylpyridine
430 (99.9%, Sigma Aldrich), bis(trifluoromethane)sulfonimide lithium salt (Li-TFSI,
431 Sigma-Aldrich), acetonitrile (99.9%, Sigma-Aldrich), chlorobenzene (99.8%, Wako),
432 2,2',7,7'-tetrakis(N,N-di-p-methoxyphenylamine)-9,9'-spirobifluorene
433 (spiro-OMeTAD, Merck), poly(3-hexylthiophene) (P3HT, Luminescence Technology
434 Corp.), SnO₂ (Alfa Aesar, tin (IV) oxide, 15% in H₂O colloidal dispersion),
435 ethylenediaminetetraacetic acid dipotassium (Sigma-Aldrich), methanol (Wako),

436 isopropanol (Wako), N,N-dimethylformamide (DMF, Wako), dimethyl sulfoxide
437 (DMSO, Wako).

438 **PSCs fabrication.** The ITO substrate was sequentially washed with distilled
439 water, acetone, ethanol and isopropanol. Then treated using UV/O₃ for 30 min. The
440 SnO₂ layer was subsequently coated on ITO substrate with the SnO₂ nanocrystal
441 solution (diluted by H₂O to 2.5%) at 3000 rpm for 30 s, and annealed at 150 °C for 30
442 min in air. Then an ethylenediaminetetraacetic acid dipotassium methanol solution
443 was spin-coated on the SnO₂ film at 3000 rpm for 30 s, and annealed at 100 °C for 5
444 min in air. Then, a 1.35 M PbI₂ and 0.0675 M CsI solution (dissolved in mixed
445 DMF/DMSO (V:V = 19:1) solvent and stirred at 70 °C for 2 hours before filtered with
446 PTFE syringe filters) was spin-coated on substrate at 2000 rpm for 30 s. Then a mixed
447 organic cation solution (FAI 400 mg; MAI 200 mg; MABr 50 mg; MACl 50 mg,
448 dissolved in 10 mL isopropanol) was spin-coated at 1750 rpm. for 30 s and then
449 annealed at 150 °C for 15 min in air. Then a EAI/MAI solution dissolved in mixed
450 IPA/DMF (V:V = 200:1) was spin-coated on the top of perovskite layer at 6000 rpm
451 for 30 s and annealed at 70 °C for 5 min. Then the hole transport materials (HTM)
452 solution was deposited by spin-coating at 3000 rpm for 30 s. The spiro-OMeTAD
453 solution was prepared by dissolving spiro-OMeTAD in chlorobenzene (72.3 mg
454 mL⁻¹) with the addition of 17.5 μL Li-TFSI/acetonitrile (520 mg mL⁻¹), and 28.8 μL
455 4-tert-butylpyridine. A mixture of spiro-OMeTAD/P3HT (weight ratio, 9:1) was used
456 for mixed HTM solution preparation, similar amount of Li-TFSI and
457 4-tert-butylpyridine were added. Finally, a gold layer with a thickness of 100 nm was
458 deposited as the counter electrode on the top of HTL through shadow masks via
459 thermal evaporation under high vacuum (5×10^{-5} Torr).

460 **PSMs fabrication.** The perovskite solar module consists of seven perovskite
461 sub-cells connected in series on a 5 cm × 5 cm substrate. Between each cell there is a
462 line (P1) with a width of 50 μm patterned by laser-etching to separate ITO stripes,
463 with a dimension of 6.65 mm by 49 mm (Supplementary Figure 14). The ETL,
464 perovskite, and HTL were prepared using the same process as small-size PSCs. After

465 coating ETL, perovskite, and HTL, one more line (P2) with a width of about 250 μm
466 is patterned by CO_2 laser with a power of 5.6 W to expose the bottom ITO/ SnO_2
467 electrodes to form the series connections between the cells. A gold film with a
468 thickness of 120 nm was then thermally evaporated as the counter electrode and each
469 sub-cell was separated by mechanical scribing to form P3 patterning.

470 **PSMs encapsulation.** Parylene is used due to its robust water resistance
471 capability and the capability to provide compact enclosure to prevent leakage. In
472 addition, parylene encapsulation avoids the direct exposure of perovskite and HTL to
473 air at the scribed series interconnections region in a PSM, which favors the
474 enhancement of PSM stability. Before parylene encapsulation, metal wires were
475 connected to the PSM by ultrasonic soldering (Sunbonder USM-560). Then, the
476 parylene film was deposited by a chemical vapor deposition process using PDS 2010
477 LABCOATER (Supplementary Figure 24). The PSMs for encapsulation were first
478 loaded into the CVD chamber, which was pumped to a pressure below 1500 Pa. Then,
479 the precursor particles (Parylene C, SCS coatings, 2 g) were kept at a temperature of
480 175 $^\circ\text{C}$ for vaporization. The furnace tube is kept at a temperature of 690 $^\circ\text{C}$ to
481 pyrolyse the precursor into monomers. During the parylene deposition, the chamber is
482 kept at a pressure of approximately 3500 Pa. The substrate temperature during
483 deposition was approximately 40 $^\circ\text{C}$. The polymerization and deposition rate of
484 parylene is approximately 2 μm per hour. The thickness of the parylene encapsulation
485 was approximately 2 μm . A cover glass was further attached on the top of the PSMs to
486 provide mechanical protection for the underneath layers. For cover glass sealing, a
487 glass sheet with a thickness of 2 mm was carefully sealed on top of the parylene film
488 by kapton tape (Supplementary Figure 25). The space between glass and the parylene
489 film is approximately 0.2 mm, which is determined by the tape thickness.

490 **Photovoltaic characterization.** J–V characteristics of perovskite solar cells are
491 measured under one sun illumination (AM 1.5 G, 100 mW cm^{-2} , calibrated using a
492 KG3 reference Si-cell (Enlitech), Oriel Instruments Model Number 90026564, 2 \times 2
493 cm^2) using a solar simulator (Newport Oriel Sol 1A, Xenon-lamp, USHIO,

494 UXL-150SO) and a Keithley 2420 source meter in ambient air at about 25 °C and a
495 relative humidity of 40 ~ 60%. The small-size PSCs were measured using a metal
496 mask with the aperture area of 0.09 cm². The designated area of 22.4 cm² for modules
497 was defined by a corresponding metal mask. The J-V scan range is from -0.1 V to 1.2
498 V for small-size PSCs and from -0.2 V to 8.5 V for modules. All the J-V curves were
499 measured under reverse scan with a scan rate of 0.25 V s⁻¹ without preconditioning
500 unless otherwise specified. No preconditioning protocol was used before the
501 characterization. The EQE spectra of small-cells were characterized using Oriel IQE
502 200.

503 **Stability testing.** For continuous operation stability measurements, the PSMs
504 were loaded in our home-designed enclosure box with dry N₂ flow to maintain a
505 relative humidity below 5%. The PSMs were under continuous illumination by a solar
506 simulator (Peccell PEC-L01, AM1.5G without UV filter) and source meter (Keithley
507 2401), which was controlled by a LabView program to allow automatic measurements
508 on the PSMs. To simulate continuous PSM operation (i.e., operation stability), a fixed
509 bias voltage was continuously applied to the PSMs maintaining the PSM operation
510 near the MPP. The bias voltage value was set to be the initial MPP voltage during the
511 operation stability measurement. The photocurrent output of the modules was also
512 measured by Keithley 2401, and was recorded by the home-made LabView program.
513 For the thermal stress test under dry N₂ condition, the PSMs were loaded on a hotplate
514 at a temperature of approximately 60 °C for aging in a dry N₂ glove box with a
515 relative humidity below 5%.

516 **Characterization.** SEM measurements were carried out using a scanning
517 electron microscope (Helios NanoLab G3 UC, FEI). XRD measurements were carried
518 out in a Bruker D8 Discover instrument (Bruker AXS GmbH, Karlsruhe, Germany)
519 equipped with Cu wavelength $\lambda = 1.54 \text{ \AA}$ X-ray source operated at 1600 W and
520 Goebel mirror. The UPS and XPS spectra were recorded from an X-ray photoelectron
521 spectrometer (XPS-AXIS UltranHAS, Kratos) equipped with monochromatic Al-K α
522 = 1486.6 eV and nonmonochromatic He-I = 21.22 eV sources. XPS results were

523 fitted using Gaussian-Lorentzian functions after background subtraction
524 corresponding to the inelastic scattering processes. TRPL data were acquired using the
525 time-correlated, single-photon counting technique (Hamamatsu, C10627), and
526 excitation was provided by a femtosecond mode-locked Ti:sapphire laser
527 (Spectra-Physics, MAITAI XF-IMW) at 450 nm with an average power at 8 MHz of
528 0.74 mW. SIMS (Kratos Axis ULTRA) equipped with quadrupole mass spectrometer
529 (HAL 7, Hiden Analytical) was used to collect the elemental signal in positive ion
530 detection mode (PID). For sputtering in SIMS, 3 keV Ar⁺ primary beam with a current
531 of 10 mA and a diameter of 100 μm was utilized. To study the effect of mixed HTM
532 on retarding gold migration, we have prepared the samples according to a previous
533 study on a similar topic.⁴⁰ Our result was obtained from the samples with Au layer.
534 The samples were prepared on 1.5 cm × 1.5 cm ITO substrate according to the same
535 procedure as the fabrication procedure of our devices. Then, the samples were placed
536 on a hot plate which was kept at a temperature of 70 °C for 24 h in a N₂ glove box.
537 The aged samples were transferred into the SIMS chamber for measurements. The
538 beam was at an angle of 45 ° with respect to the sample surface normal. The
539 spectrometer was operated at a pressure of 10⁻⁸ torr. The SCLC data were collected
540 with a semiconductor characterization system in N₂ (4200-SCS, Keithley).
541 Absorbance was measured using a UV-Vis spectrometer (JASCO Inc., V-670).

542 **Reporting summary.** Further information on research design is available in the
543 Nature Research Reporting Summary linked to this article.

544 **Data availability**

545 All data generated or analysed during this study are included in the published
546 article and its Supplementary Information. The data that support the plots within this
547 article and other findings of this study are available from the corresponding author
548 upon reasonable request.

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558 **Author contributions**

559 Y.B.Q. supervised the project. Y.B.Q. and Z.L. conceived the ideas and designed
560 the experiments. Z.L. conducted the corresponding device fabrication and basic
561 characterization. Z.L. and L.Q. conducted module fabrication, encapsulation and
562 stability testing. L.Q. and L.K.O. helped with XPS, UPS and SIMS characterization
563 and analyses. S.H. helped with the module picture design and data analysis. Z.H.
564 helped with energy alignment analyses. M.J. helped with TRPL characterization. G.T.,
565 Z.W., Y.J., Y.D., and S.K. provided valuable suggestions for the manuscript. S.K.
566 contributed to the J-V characterization. Z.L. and Y.B.Q. participated in all the data
567 analysis. All authors contributed to writing the paper.

568 **Competing interests**

569 All the authors declare no competing interests.

570 **Additional information**

571 **Supplementary information** is available for this paper at <https://doi.org/>

572 **Correspondence and requests** for materials should be addressed to Y.B.Q.

573 **References**

- 574 1 <https://www.nrel.gov/pv/assets/pdfs/best-research-cell-efficiencies.20200218.pdf>.
- 575 2 Li, Z. *et al.* Scalable fabrication of perovskite solar cells. *Nat. Rev. Mater.* **3**,
576 18017 (2018).
- 577 3 Kim, D. H., Whitaker, J. B., Li, Z., van Hest, M. F. A. M. & Zhu, K. Outlook and
578 challenges of perovskite solar cells toward terawatt-scale photovoltaic module technology.
579 *Joule* **2**, 1437-1451 (2018).
- 580 4 Qiu, L., He, S., Ono, L. K., Liu, S. & Qi, Y. B. Scalable fabrication of metal halide

581 perovskite solar cells and modules. *ACS Energy Lett.* **4**, 2147-2167 (2019).

582 5 Ono, L. K., Qi, Y. B. & Liu, S. Progress toward stable lead halide perovskite solar
583 cells. *Joule* **2**, 1961-1990 (2018).

584 6 Correa-Baena, J.-P. *et al.* Promises and challenges of perovskite solar cells.
585 *Science* **358**, 739-744 (2017).

586 7 Domanski, K., Alharbi, E. A., Hagfeldt, A., Grätzel, M. & Tress, W. Systematic
587 investigation of the impact of operation conditions on the degradation behaviour of perovskite
588 solar cells. *Nat. Energy* **3**, 61-67 (2018).

589 8 Saliba, M., Stolterfoht, M., Wolff, C. M., Neher, D. & Abate, A. Measuring aging
590 stability of perovskite solar cells. *Joule* **2**, 1019-1024 (2018).

591 9 Khenkin, M. V. *et al.* Consensus statement for stability assessment and reporting
592 for perovskite photovoltaics based on ISOS procedures. *Nat. Energy* **5**, 35-49 (2020).

593 10 Saliba, M. *et al.* Incorporation of rubidium cations into perovskite solar cells
594 improves photovoltaic performance. *Science* **354**, 206-209 (2016).

595 11 Wang, Z. *et al.* Efficient ambient-air-stable solar cells with 2D–3D
596 heterostructured butylammonium-caesium-formamidinium lead halide perovskites. *Nat.*
597 *Energy* **6**, 17135, (2017).

598 12 Wang, L. *et al.* A Eu^{3+} - Eu^{2+} ion redox shuttle imparts operational durability to Pb-I
599 perovskite solar cells. *Science* **363**, 265-270 (2019).

600 13 Bai, S. *et al.* Planar perovskite solar cells with long-term stability using ionic
601 liquid additives. *Nature* **571**, 245-250 (2019).

602 14 Yang, S. *et al.* Stabilizing halide perovskite surfaces for solar cell operation with
603 wide-bandgap lead oxysalts. *Science* **365**, 473-478 (2019).

604 15 Fang, R. *et al.* [6,6]-Phenyl-C61-Butyric acid methyl ester/cerium oxide bilayer
605 structure as efficient and stable electron transport layer for inverted perovskite solar cells. *ACS*
606 *Nano* **12**, 2403-2414 (2018).

607 16 Jung, E. H. *et al.* Efficient, stable and scalable perovskite solar cells using
608 poly(3-hexylthiophene). *Nature* **567**, 511-515 (2019).

609 17 Wu, S. *et al.* A chemically inert bismuth interlayer enhances long-term stability of

610 inverted perovskite solar cells. *Nat. Commun.* **10**, 1161 (2019).

611 18 Cheacharoen, R. *et al.* Encapsulating perovskite solar cells to withstand damp heat
612 and thermal cycling. *Sustain. Energy Fuels* **2**, 2398-2406 (2018).

613 19 Yang, M. *et al.* Highly efficient perovskite solar modules by scalable fabrication
614 and interconnection optimization. *ACS Energy Lett.* **3**, 322-328 (2018).

615 20 Matteocci, F. *et al.* High efficiency photovoltaic module based on mesoscopic
616 organometal halide perovskite. *Prog. Photovolt: Res. Appl.* **24**, 436-445 (2016).

617 21 Bi, E. *et al.* Efficient perovskite solar cell modules with high stability enabled by
618 iodide diffusion barriers. *Joule* **3**, 1-13 (2019).

619 22 Bu, T. *et al.* Universal passivation strategy to slot-die printed SnO₂ for
620 hysteresis-free efficient flexible perovskite solar module. *Nat. Commun.* **9**, 4609 (2018).

621 23 Xu, Z. *et al.* A Thermodynamically favored crystal orientation in mixed
622 formamidinium/methylammonium perovskite for efficient solar cells. *Adv. Mater.* **31**, 1900390
623 (2019).

624 24 Jiang, Q. *et al.* Enhanced electron extraction using SnO₂ for high-efficiency
625 planar-structure HC(NH₂)₂PbI₃-based perovskite solar cells. *Nat. Energy* **1**, 16177 (2016).

626 25 Qiu, L. *et al.* Scalable fabrication of stable high efficiency perovskite solar cells
627 and modules utilizing room temperature sputtered SnO₂ electron transport layer. *Adv. Funct.*
628 *Mater.* **29**, 1806779 (2018).

629 26 Brinkmann, K. O. *et al.* **Suppressed** decomposition of organometal halide
630 perovskites by impermeable electron-extraction layers in inverted solar cells. *Nat. Commun.* **8**,
631 13938 (2017).

632 27 Yang, D. *et al.* High efficiency planar-type perovskite solar cells with negligible
633 hysteresis using EDTA-complexed SnO₂. *Nat. Commun.* **9**, 3239 (2018).

634 28 Li, X., Zhang, W., Wang, X., Gao, F. & Fang, J. Disodium edetate as a promising
635 interfacial material for inverted organic solar cells and the device performance optimization.
636 *ACS Appl. Mater. Interfaces* **6**, 20569-20573 (2014).

637 29 Son, D.-Y. *et al.* Universal approach toward hysteresis-free perovskite solar cell
638 via defect engineering. *J. Am. Chem. Soc.* **140**, 1358-1364 (2018).

- 639 30 Wu, W.-Q. *et al.* Bilateral alkylamine for suppressing charge recombination and
640 improving stability in blade-coated perovskite solar cells. *Sci. Adv.* **5**, eaav8925 (2019).
- 641 31 Liu, Z. *et al.* Gas-solid reaction based over one-micrometer thick stable perovskite
642 films for efficient solar cells and modules. *Nat. Commun.* **9**, 3880 (2018).
- 643 32 Im, J.-H., Chung, J., Kim, S.-J. & Park, N.-G. Synthesis, structure, and
644 photovoltaic property of a nanocrystalline 2H perovskite-type novel sensitizer
645 (CH₃CH₂NH₃)PbI₃. *Nanoscale Res. Lett.* **7**, 353 (2012).
- 646 33 Peng, W. *et al.* Engineering of CH₃NH₃PbI₃ perovskite crystals by alloying large
647 organic cations for enhanced thermal stability and transport properties. *Angew. Chem. Int. Ed.*
648 **55**, 10686-10690 (2016).
- 649 34 Zhou, J. *et al.* Efficient ambient-air-stable HTM-free carbon-based perovskite
650 solar cells with hybrid 2D–3D lead halide photoabsorbers. *J. Mater. Chem. A* **6**, 22626-22635
651 (2018).
- 652 35 Zhang, Y., Kim, S.-G., Lee, D., Shin, H. & Park, N.-G. Bifacial stamping for high
653 efficiency perovskite solar cells. *Energy Environ. Sci.* **12**, 308-321 (2019).
- 654 36 Gholipour, S. *et al.* Globularity-selected large molecules for a new generation of
655 multication perovskites. *Adv. Mater.* **29**, 1702005 (2017).
- 656 37 Wang, Y. *et al.* A mixed-cation lead iodide MA_{1-x}EA_xPbI₃ absorber for perovskite
657 solar cells. *J. Energy Chem.* **27**, 215-218 (2018).
- 658 38 Luo, D. *et al.* Enhanced photovoltage for inverted planar heterojunction perovskite
659 solar cells. *Science* **360**, 1442-1446 (2018).
- 660 39 Jiang, Y. *et al.* Combination of hybrid CVD and cation exchange for upscaling
661 Cs-substituted mixed cation perovskite solar cells with high efficiency and stability. *Adv.*
662 *Funct. Mater.* **28**, 1703835 (2017).
- 663 40 Domanski, K. *et al.* Not all that glitters is gold: metal-migration-induced
664 degradation in perovskite solar cells. *ACS Nano* **10**, 6306-6314 (2016).
- 665 41 Christians, J. A. *et al.* Tailored interfaces of unencapsulated perovskite solar cells
666 for >1,000 hour operational stability. *Nat. Energy* **3**, 68-74 (2018).
- 667 42 Kim, G.-W., Kang, G., Malekshahi Byranvand, M., Lee, G.-Y. & Park, T.

668 Gradated mixed hole transport layer in a perovskite solar cell: improving moisture stability
669 and efficiency. *ACS Appl. Mater. Interfaces* **9**, 27720-27726 (2017).
670 43 Jinno, H. *et al.* Stretchable and waterproof elastomer-coated organic photovoltaics
671 for washable electronic textile applications. *Nat. Energy* **2**, 780-785 (2017).
672