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Environmental forensic characterization of former rail yard soils located adjacent to the Statue of Liberty in the New York/New Jersey harbor

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Highlights

- We characterized organic and inorganic substances at a major urban brownfield site
- We detected abundant petroleum- and coal-derived biomarkers and PAHs in the soil
- Both barren low-functioning and forested soils contained similar organic compounds
- The barren site featured very high heavy metal loads and dormant microbial life
- Chemical and ecological studies are critical to brownfield restoration efforts

Graphical Abstract



1 Abstract

2 Identifying inorganic and organic soil contaminants in urban brownfields can give insights into 3 the adverse effects of industrial activities on soil function, ecological health, and environmental 4 quality. Liberty State Park in Jersey City (N.J., USA) once supported a major rail yard that had 5 dock facilities for both cargo and passenger service; it was later closed off to the public, and a 6 forest developed and spread in the area. The objectives of this study were to: 1) characterize the 7 organic and inorganic compounds in Liberty State Park soils and compare the findings to an 8 uncontaminated reference site (Hutcheson Memorial Forest); and 2) identify differences between 9 the barren low-functioning areas and the forested high-functioning areas of the brownfield. Soil 10 samples were solvent-extracted, fractionated, and analyzed by gas chromatography-mass 11 spectrometry and subjected to loss-on-ignition, pyrolysis-gas chromatography-mass 12 spectrometry, inductively-coupled-plasma mass spectrometry, and optical microscopy analyses. 13 Compared to soil from the reference site, the forested soils in Liberty State Park contained 14 elevated percentages of organic matter (30–45 %) and more contaminants, such as fossil-fuel-15 derived hydrocarbons and coal particles. Microscopy revealed bituminous and anthracite coal, 16 coke, tar/pitch, and ash particles. Barren and low-functioning site 25R had a similar organic 17 contaminant profile but contained a higher metal load than other Liberty State Park sites and also 18 lacked higher plant indicators. These can obscure the signatures of contaminants, and data from 19 adjacent barren and vegetated sites are valuable references for soils studies. A deeper 20 understanding of the chemistry, biochemistry, and ecology of barren soils can be leveraged to 21 prevent land degradation and to restore dysfunctional and phytotoxic soils.

23 Keywords: Brownfield, heavy metals, polycyclic aromatic hydrocarbons, soil forensics,

24 pyrolysis-GC-MS, petrography, industrial barrens

25

26 **1. Introduction**

27 Past industries have altered ecosystems by introducing inorganic and organic contaminants into

28 soils (Alker et al., 2000; Gallego et al., 2016; Qian et al., 2017; Thornton et al., 2008).

29 Specifically, former rail yard soils remain contaminated years after operations have ceased

30 (Jackson, 1997). Loading, unloading, and the train engines themselves can introduce both

31 inorganic compounds such as heavy metals and organic contaminants including polycyclic

32 aromatic hydrocarbons (PAHs) into the soil (Jackson, 1997; Lacey and Cole, 2003; Liu et al.,

33 2008; Malawska and Wilkomirski, 1999; Malawska and Wiłkomirski, 2000; Malawska and

34 Wiołkomirski, 2001; Wiłkomirski et al., 2011). Organic contaminants can originate from

35 lubricating oils, coal, oil, fertilizers and herbicides (Biache et al., 2017; Wiłkomirski et al.,

36 2011). PAHs, many of which are toxic or mutagenic to humans, plants, and animals, (Brooks,

37 2004; Smith et al., 2006) can originate from incomplete combustion of coal (Haritash and

38 Kaushik, 2009).

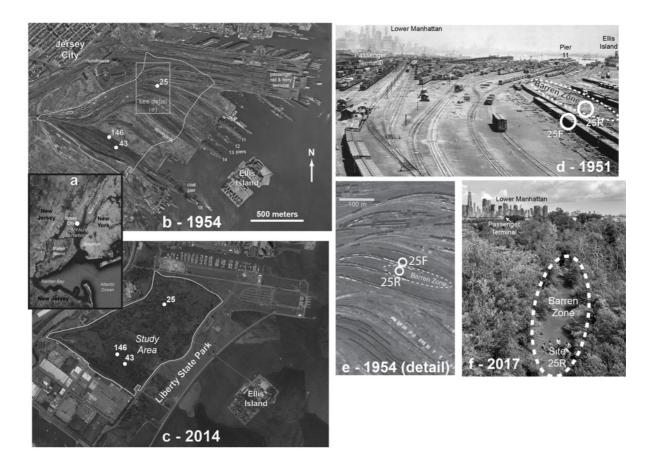
Liberty State Park (LSP) in Jersey City, New Jersey is located across the Hudson River from lower Manhattan and the Statue of Liberty and once supported a rail yard (Fig. 1A). Originally, LSP consisted of mudflats and salt marsh but was later filled in with New York City construction debris and municipal waste to prepare for the construction of the Central Railroad of New Jersey. The railroad was built between 1860 and 1919, operated until 1967, and was closed in 1970 (Gallagher et al., 2008a). Part of LSP was remediated but a 100-ha un-remediated site, our study area, was fenced to restrict public access. Despite its history of contamination, most areas within our un-remediated study site support a biodiverse and productive forest (Gallagher
et al., 2008a). Figure 1C shows the locations of our sites within the study area (43, 146, and 25).
The organic compounds at LSP had not been previously analysed but coal fragments are visible
throughout the site.

50 We previously discovered high enzyme activities at LSP site 146, which has high heavy 51 metal loads (Hagmann et al., 2015). This finding was surprising because heavy metal and 52 organic contaminants are often thought to decrease soil microbial functioning (Alisi et al., 2009; 53 Buettner and Valentine, 2011; Hamdi et al., 2007; Sprocati et al., 2012). Because organic 54 contaminants can play a role in soil functioning (Baran et al., 2004; Shen et al., 2005) and 55 display additive effects in combination with heavy metals (Maliszewska-Kordybach and 56 Smreczak, 2003; Shen et al., 2005), we identified organic contaminants at different sites within 57 LSP that might lead to differences in enzymatic activities. It is important to study both inorganic 58 and organic compounds to understand historic industrial activities (Ortiz et al., 2016a). 59 Figure 1 60 The use of pyrolysis gas chromatography – mass spectrometry (Py-GC-MS) is only 61 beginning to see wide application in characterization of brownfield contaminants (Kruge, 2015; 62 Kruge et al., 2018; Lara-Gonzalo et al., 2015). Here we studied LSP soils via Py-GC-MS and 63 GC-MS combined with organic petrography and determination of soil enzyme activities, major 64 and trace elements, total organic matter, and microbial counts. This unique combination of 65 methods, including the concurrent identification of organic and inorganic pollution with

66 measurements of enzymatic function and microbial counts, provides a multidimensional

67 understanding of the contaminants in LSP soils.

Fig. 1. Liberty State Park is located in Jersey City, New Jersey (a). Aerial images from: U. S. Geological Survey: 1954 (b, e) and 2014 (c). Study sites within LSP are indicated on the map (43, 146, 25F, 25R). A historical photograph of 25F and 25R from 1951 (d). Sites 25F and 25R are located next to each other; 25R is in a strip of land without vegetation (f). Photo Credits: d: Andrew Bologovsky f: Mike Peters Montclair State University. Photos used with permission.



68 Sites were chosen in LSP and at the reference site at Hutcheson Memorial Forest (HMF) 69 that would allow us to answer questions about the impact of industrial activities and vegetation 70 on organic contaminant profiles in the soils. Soils were characterized at four sites (43, 146, 25F, 71 and 25R) within the un-remediated and closed-off LSP area. Our first hypothesis was that there 72 will be more fossil fuel biomarkers at LSP compared to HMF. According to historic photographs 73 shown in Figure 1B, sites 43 and 146 lie along tracks to a coal pier while sites 25F and 25R are 74 located next to each other and along the route to "Pier 11" (Fig. 1D). All LSP sites have a 75 history of industrial contamination while the reference site HMF does not. Site 25R lacks plants 76 while the other three LSP sites (43, 146, and 25F) and reference site are vegetated. Our second 77 hypothesis was that the barren site 25R will have more or different contaminants compared to the 78 forested sites at LSP. Our two primary objectives were to: 1.) characterize the organic and 79 inorganic compounds at LSP and compare the findings to a reference site HMF; and 2.) identify 80 differences between the barren site 25R (area with no vegetation) and forested areas at LSP. 81

89

82 **2. Methods**

83 2.1 Site description

Soil samples were collected from four sites located within an un-remediated, restricted-access,
100-ha plot within LSP, Jersey City, NJ, USA (40° 42" 16 N, 74° 03' 06 W, Fig. 1A, B). LSP
was formerly a major rail yard and dock facility built on estuarine marshland and was abandoned
around 1969 (Gallagher et al., 2008a). In the intervening decades, dense forest cover took over
the restricted access plot without human intervention.

Soils for this study were collected from three different vegetated sites within LSP,

90 including 43, 146, and 25F (Fig. 1C). A fourth site—25R that is adjacent to 25F—is on a strip of

land that remains anomalously barren (Fig. 1F). As a reference, soil was also collected from a
location within HMF in Franklin Township, N.J—an abandoned agricultural field with no history
of industrial use or contamination about 60 km southwest of LSP—having the same natural
successional timeframe.

95

96 2.2 Soil collection and preparation

97 Soil was collected in July 2016 from LSP sites 25F, 25R, 43, and 146 as well as the reference 98 site HMF. For this study, five samples were collected along a transect at intervals of 4 m for this 99 study into separate bags (depth of 10 cm below the leaf-litter). The samples were stored in a 100 refrigerator (4 °C). At the lab the soils were sieved through a 2-mm sieve and equal amounts of 101 the five samples along a transect were combined into one bag labelled with the transect name and 102 site (e.g., 43A). At each site, five samples were collected along three parallel transects 10 m 103 apart. Thus, fifteen soil subsamples from a 16 by 20 m field grid were combined into three 104 composite samples (one for each transect), as seen in Figure 2. For each site, the three 105 composites samples were submitted separately for laboratory analysis. Figure 2 shows an 106 overview of the experimental procedures we used to determine the organic composition of the 107 soils as described in sections 2.4 and 2.5.

108

109 2.3 Percent organic matter

The percent organic matter was determined by loss on ignition (LOI). This procedure is described in Hagmann et al. 2015. Briefly, soil was dried at 70°C for 24 hours. The soil was ground using a mortar and pestle and samples were heated to 550°C for 4 hours. The percent

organic matter was determined gravimetrically. Results are presented as the mean values of thethree composite soil samples taken at each site.

115

116 2.4 Extraction, LC fractionation and GC-MS

For each site, only one analysis was performed, on a global sample combining each of the three transect composites (Section 2.2) in equal measure. Dry soil samples $< 2 \text{ mm} (7.8 - 11.0 \text{ g}_{dry})$

119 underwent solvent extraction as described in Lara-Gonzalo et al. 2015. Briefly, soils were

120 extracted using dichloromethane:methanol (3:1, v/v) in a Soxtherm system (Gerhardt analytical

121 systems: Königswinter, Germany). The extract was concentrated by rotary evaporation. Aliquots

122 of the Soxtherm extract were fractionated and gravimetrically quantified by open column liquid

123 chromatography (LC). First, maltenes and asphaltenes were separated by filtering through 0.45

124 µm filters using hexane and dichloromethane, respectively; then, maltenes were fractionated into

125 three fractions by LC in columns filled with silica gel and alumina. Fraction 1 (predominantly

126 aliphatic hydrocarbons) was eluted with hexane, Fraction 2 (predominantly aromatic

127 hydrocarbons) with a mix of dichloromethane:hexane (7:3, v/v) and finally, Fraction 3

128 (predominantly polar compounds) with dichloromethane:methanol (1:1, v/v).

129

Figure 2

130 The analysis of the LC fractions was carried out by GC/MS. The injection of the extracts was

131 performed on a GC/MS QP-2010 Plus (Shimadzu: Kyoto, Japan). A capillary column DB-5ms

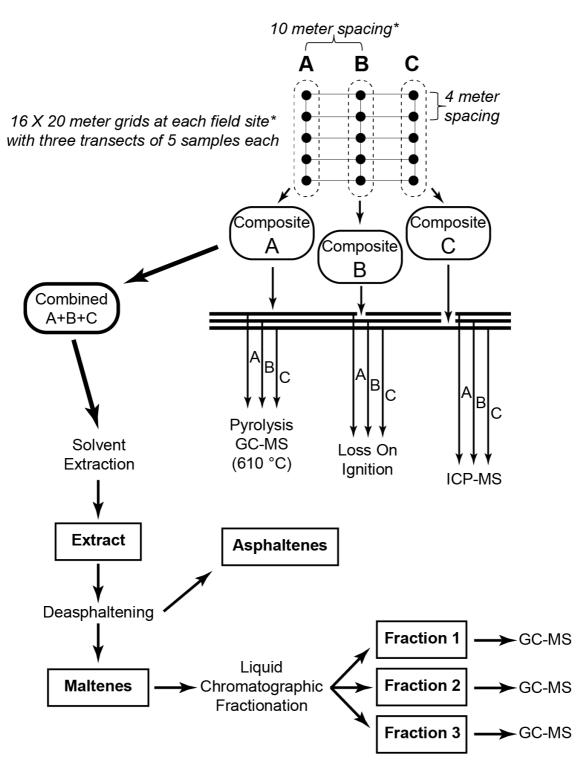
132 (5% phenyl 95% dimethylpolysiloxane; 60 m \times 0.25 mm i.d. \times 0.25 μ m film) from Agilent

133 Technologies was used with helium as carried gas at 1 mL/min. The initial oven temperature was

134 50 °C (held for 2 min) and ramped at 2.5 °C min⁻¹ up to 310 °C and held for 45 min. The mass

135 spectrometer was operated in electron ionization mode (EI) at 70 eV. It was calibrated daily by

Fig. 2. Flow chart illustrating the experimental design for the experiment.



autotuning with perfluorotributylamine (PFTBA) and the chromatograms were acquired in fullscan mode (mass range acquisition was performed from 45 to 500 m/z). Compounds were
identified using the NIST 2014 Mass Spectral Library (NIST 2014/EPA/NIH) and by reference
to the literature.

140

141 2.5 Pyrolysis-Gas Chromatography-Mass Spectrometry (Py-GC-MS)

142 For samples HMF, LSP 43, and LSP 146, Py-GC-MS was performed using a CDS Analytical

143 5000 Pyroprobe (Oxford, Pennsylvania, USA) coupled to a Thermo Electron DSQ GC/MS

144 (Madison, Wisconsin, USA) equipped with an Agilent DB-1MS column (30 m \times 0.25 mm i.d. \times

145 0.25 μ m film thickness). The GC oven temperature was programmed from 50 °C to 300 °C (at 5

¹⁴⁶ °C min⁻¹), with an initial hold of 5 min at 50 °C and a final hold of 15 min at 300 °C. Pyrolysis

147 was performed for 20 s at 610 °C. The MS was operated in full scan mode (50-500 Da, 1.08

scans s^{-1}). Quality control was performed by separately analyzing the three composited samples

from each site (Section 2.2). For each site, the three yielded very similar results; therefore, only

150 one was used to represent the site in this paper. Compounds were identified using the NIST MS

151 library and by reference to the literature.

152

149

153 2.6 Organic petrography

154 For petrographic analysis, soil samples were mixed with Lucite powder and prepared into pellets

155 in a Leco PR-15 mounting press. A GPX200 grinder/polisher was used to polish the samples

according to the standard sample preparation techniques (Taylor et al., 1998). A reflected light

157 microscope Leica DM 2500P linked to a TIDAS PMT IV photometric system was used to

158 determine petrographic composition of the samples by counting 500 points, and the results were

recalculated into volume percent. Representative photomicrographs were taken of each sample using the same microscope. Coal, coke, tar and pitch, fly ash/bottom ash, sediments, and other materials were the petrographic categories counted, and each category included several components identified based on their optical characteristics.

163

164 2.7 Elemental analysis

165 The concentrations of selected elements were determined using a Thermo iCAP Qc inductively 166 coupled plasma mass spectrometry (ICP-MS, Thermo Fisher Scientific, Bremen, Germany). The 167 soil was extracted following EPA method 3050B, which is described in Hagmann et al. (2015). 168 Briefly, 5 mL of 50% HNO₃ was added to a 0.5 g homogenized sample. The solution was heated 169 to 95 ± 5 °C for 15 minutes. After the sample cooled down, 2.5 mL of concentrated HNO₃ was added to the solution and the sample was heated 95 ± 5 °C for 30 minutes. This step was 170 171 repeated if brown fumes were present. Afterwards, the sample remained at 95 ± 5 °C until the 172 volume was reduced to approximately 2.5 mL. DI (1mL) water was then added to the solution 173 and drops of 30 % H₂O₂ (1.5 mL) were added. Further amounts of 30% H₂O₂ were added until 174 effervescence was minimal. The sample remained at 95 ± 5 °C until the volume dropped to 2.5 175 mL. The volume was brought up to 50 mL with DI water. The solution was then filtered 176 through a 1 µm filter. Each sample was further diluted to run on the ICP-MS. Samples were 177 diluted to at least 20 times to minimize the matrix effect. An aqueous standard curve was 178 prepared for Li, Mg, Al, P, K, Ca, Sc, V, Cr, Mn, Co, Ni, Cu, Zn, As, Mo, Ag, Cd, Ba and Pb 179 from a stock solution.

180

182 2.8 Acridine orange direct count (AODC)

183 Bacterial cell density was measured using epifluorescence microscopy after staining the soil 184 suspensions with acridine orange. Briefly, soil samples were suspended in phosphate buffered 185 saline (PBS) (0.0088 M Na₂HPO₄, 0.0013 M NaH₂PO₄ • H₂O, 0.15 M NaCl; pH 7.6) and fixed 186 in formalin. The fixed samples were then serially diluted in PBS with a final dilution of 10^{-3} . 187 The diluted samples were then stained with 0.1 % acridine orange and transferred on a black 188 polycarbonate 0.2 µm IsoporeTM membrane filter (Millipore, Waltham, M.A.) and observed 189 under an epifluorescence microscope with a 100X objective lens (Nikon eclipse Ti-S) (Hobbie et 190 al. 1977, Krumins et al. 2009). 191 192 2.9 Phosphatase assay 193 Phosphatase enzyme activity was measured for soil from each site as described in Hagmann et al.

2015. Briefly, moist soil (0.1 g) was added to 100 mL of 0.1 M MES buffer (pH 6.0) and then sonicated for 3 min at 25W. The slurry was stirred continuously, and each sample was added to the well (160 μ L). 4-MUB-phosphate was added (40 μ L) to three different wells (350 μ M in the well). The concentration was determined by preparing a standard curve using different concentrations of 4-MUB (0, 500, 1000, 1500 and 2500 pmols). Time points were collected on a microplate reader at 30 °C every 20 minutes for 6 hours to measure fluorescence intensity (320 nm ex./450 nm em.).

201

202

203

205 **3. Results and discussion**

206 3.1. Bulk organic matter content

207 The soil from the reference site HMF has an organic matter content of 7.4 % (by LOI), similar to

- 208 typical forested soils in New Jersey (Table 1) (Osman, 2013). It should be noted that the solvent
- 209 extraction were conducted on soil particles that were less than 2 mm in diameter. In contrast,
- forested LSP sites 43, 146, and 25F are unusually enriched in organic matter (30 45 %)
- 211 compared to typical New Jersey soils. Compared to the other LSP sites, the barren LSP site 25R
- has a lower percentage of organic matter (10.8 %). The reference soil from HMF and barren 25R
- soil have low solvent extract yields (1.3 and 1.2 g/kg_{soil}, respectively), compared to the forested
- LSP sites 43, 146 and 25F, which yielded 7.3 to 9.8 g/kg_{soil}. Asphaltenes, higher molecular
- weight compounds (500 1,000 or more Da) (Lewan et al., 2014), are dominant in the soils from
- HMF and LSP sites 43, 146 and 25F, composing > 70 % of the extracts. Fraction 3
- 217 (predominantly polar compounds) is also prevalent (14 22%) in these same soils (Table 1).
- 218 Site 25R is distinctive: in addition to the low extract yield, it contains only 47 % asphaltenes and
- 219 considerably more saturated hydrocarbons (Fraction 1, 17.6 %) than the other samples.
- 220

Table 1

The forested sites at LSP (43, 146 and 25F) contain both biological materials and contaminants. The additional organic matter and extract yield in soils from these sites is likely because of augmentation by fossil fuel materials. LOI measurements alone cannot distinguish between coal and natural soil organic matter (Ball, 1964). However, we do indeed commonly see chunks of coal in the LSP soils, which must obviously contribute to the elevated organic matter at LSP. Samples HMF and 25R contain less soil organic matter and extract, but HMF is forested, while 25R is barren (Fig. 1). The lower organic matter content and extract yield at 25R **Table 1.** Organic matter content (%, mean \pm SE; n = 3), extraction and liquid chromatographic results from HMF, 43, 146, 25F and 25R soils. Fraction 1 contains saturated hydrocarbons; fraction 2 – aromatic compounds and long chain (> C₂₇) normal alkanes; fraction 3 – polar compounds. The data show that soil from site 146 has the highest percent organic matter. The LC results show the percent of the total extract as well as yield per kg of dry soil.

Sample	Organic Matter (% of soil)	Extract yield (g/kg _{soil})	Fraction 1 mg/kg _{soil} (% of extract)	Fraction 2 mg/kg _{soil} (% of extract)	Fraction 3 mg/kg _{soil} (% of extract)	Asphaltenes mg/kg _{soil} (% of extract)
HMF	7.43 ± 0.63	1.28	43.0 (3.3)	52.0 (4.0)	287 (22.1)	918 (70.6)
LSP 43	29.8 ± 0.86	9.21	451 (4.9)	506 (5.5)	929 (10.1)	7,314 (79.5)
LSP 146	44.8 ± 2.18	9.83	549 (5.6)	451 (4.6)	1421 (14.5)	7,379 (75.3)
LSP 25F	31.6 ± 4.79	7.38	429 (5.8)	363 (4.9)	1,095 (14.8)	5,513 (74.5)
LSP 25R	10.8 ± 1.94	1.21	211 (17.6)	119 (9.9)	307 (25.6)	563 (46.9)

likely results from the lack of vegetation because plants provide constant organic inputs into the soil and obviously 25R soil does not receive these inputs. The organic matter content at HMF is normal for a forested soil and simply lacks the supplemental increment because of the contamination found at LSP. We note that the asphaltene contents of the forested LSP sites are high and further work must be done to characterize the asphaltenes, for example, by Py-GC-MS.

234 3.2. Py-GC-MS

235 Pyrolysis-GC-MS results (Table 2 and Fig. 3) showed that the reference site HMF contains 236 organic compounds found in typical forest soils including lignin [L1], polysaccharide [P1, P2, 237 P3, P4], and protein biomarkers [N2] (Fig. 3 A). Reference site HMF also contains low-weight 238 aromatic hydrocarbons [A1, A2, A4] and phenols [F1, F2, F3, F4] (Fig. 3 A). Vegetated LSP 239 sites 43, 146, and 25F contain the compounds present in HMF. These sites also contain PAHs 240 such as naphthalenes and phenanthrenes, where both parent and alkylated forms are present, 241 fluoranthene (Fig. 3 B), as well as dibenzofuran [DBF] and methyldibenzofuran [DBF1]. The 242 chromatogram of barren LSP site 25R shows a notable absence of organic compounds with a 243 biological derivation [e.g., L1, L2 etc.] and thus predominantly aromatic compounds are seen in 244 the trace. The aromatic compounds seen at site 25R are also present in similar proportions at the 245 other LSP sites (Fig. 3).

246

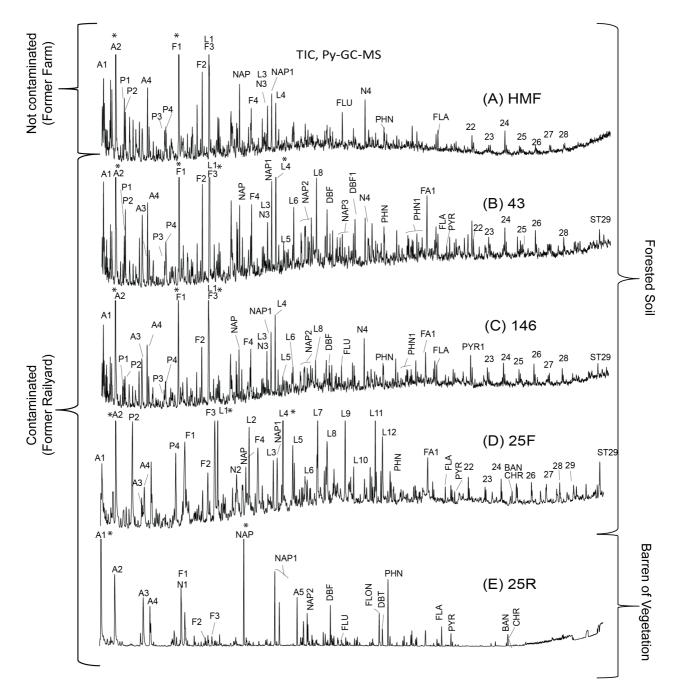
Table 2

247

Figure 3

The presence of polysaccharides, lignin, and proteins in HMF was expected because these compounds indicate plants and microorganisms in the soil (*e.g.* Hempfling and Schulten, 1990). These compounds are also present in vegetated LSP sites 43, 146, and 25F but not at

Fig. 3. Total ion current (TIC) for Py-GC-MS of soil samples from sites HMF (A), 43 (B), 146 (C), 25F (D) and 25R (E). The data for 25R show predominantly mono- and polycyclic aromatic hydrocarbons and aromatic hydrocarbons. See Table 2 for compound symbols.



Aliphatic Hydrocarbons	
Numerals	normal alkanes
Pr	pristane
Ph	phytane
Hx	hopanes (x is the carbon number)
Ts	18α(H)-22,29,30-trisnorhopane
Tm	17α (H)-22,29,30-trisnorhopane
TRx	tricyclic terpane
Aromatic Compounds	
Alkylated aromatic compounds, "x" is	degree of substitution
A1	benzene
A2	toluene
A3	m-xylene & p-xylene
A4	styrene
A5	biphenyl
NAP	naphthalene
NAPx	alkylnaphthalene isomers
PHN	phenanthrene
PHNx	alkylphenanthrene isomers
ANT	anthracene
FLA	fluoranthene
FLU	fluorene
FLON	9H-fluoren-9-one
PYR	pyrene
PYRx	alkypyrene isomers
CHR	chrysene
CHRx	alkylchrysene isomers
BAN	benzo[a]anthracene
BNT	benzonapthothiophene
BFLA	benzofluorene
DBF	dibenzofuran
DBFx	alkyldibenzofuran isomers
DBT	dibenzothiophene
Polysaccharide Marker Compounds	*
P1	cyclopentenone
P2	furfural
P3	3-methylcyclopentenone
P4	methylfurfual
P5	levoglucosan
Phenolic Compounds	
F1	phenol
F2 F3	2-methylphenol
F3 F4	3-methylphenol & 4-methylphenol vinylphenol
1 7	vinyipiiciioi

Table 2: Symbols for peak identification used in Figures 1-7.

Table 2. (continued)	
Lignin Marker Compounds	
Ll	guaiacol
L2	methylguaiacol
	ethylguaiacol
L4	vinylguaiacol
L5	syringol
L6	vanillin
L7	trans iso-eugenol
	acetovanillone
L9	vinylsyringol
L10	prop-1-enyl syringol
L11	prop-2-enyl syringol (trans)
L12	acetosyringone
Fatty Acid	dectosymigone
FA1	<i>n</i> -hexadecanoic acid
FA2	<i>n</i> -octadecanoic acid
FA3	methyl ester (E) 9-octadecanoic acid
FA4	octadecanoic acid, butyl ester
Nitrogen Compounds	
N1	benzonitrile
N2	benzeneacetonitrile
N3	indole
N4	diketodipyrole
Steroid	
S1	C29 steradiene
S2	cholest-5-en-3-ol
S2 S3	ergost-5-en-3-ol γ-stigmasterol
S5 S4	stigmast-5-en-3-ol
S5	stigmastanol
S6	stigmast-4-en-3-one
Plasticizer	
X1	diisobutyl phthalate
X2	1,2-benzenedicarboxylic acid, bis(2-ethylhexyl) ester
Other Compounds	
B1	abietane derivative
B2	β-amyrin derivative
B6	diploptene
B13	totarol
B14	1-heptacosanol
B15	3β-methoxy-D-friedoolean-14-ene
B16	olean-12-en-3-one
B17	lupeol
B18, B19	dammarane isomers
B20	cyclolaudenol
<u>B21</u>	lup-20(29)-ene-3,28-diol

251 barren site 25R. The pyrolyzates of the LSP sites 43, 146 and 25F soils have similar organic 252 compositions to each other, which is not surprising because they are all vegetated and located 253 within the same brownfield. The PAHs present at LSP sites indicate fossil fuel contamination 254 and include naphthalenes, fluoranthene, and fluorene, and heterocyclic compounds, such as 255 dibenzofuran. One likely source is coal, particles of which are evident to the naked eye in the 256 park soils, and it is known that the presence of coal in the soil can increase the amount of 257 aromatic compounds including PAHs (Kögel-Knabner, 2000; Laumann et al., 2011; Nádudvari 258 et al., 2018; Stout and Emsbo-Mattingly, 2008).

259 One example of coal indicators include dibenzofuran and methyldibenzofuran 260 (Nádudvari et al., 2018). In contrast to the other sites, 25R pyrolyzate shows a strong 261 predominance of aromatic compounds. These aromatic compounds include monoaromatic 262 hydrocarbons such as BTEX (benzene [A1], toluene [A2], ethylbenzene, xylene [A3]) and 263 polycyclic aromatic hydrocarbons (PAHs) [NAP, PHN, FLA, PYR, CHR, BAN]. Compared to 264 the other LSP sites, the different organic composition at site 25R can be largely attributed to the 265 absence of plants at 25R. Organic compounds at forested LSP soils come from vegetation or 266 industrial sources and compounds originating from plants can obscure signatures of 267 contaminants (Ortiz et al., 2016b). With many of the plant and microbial indicators missing, the 268 PAH indicators are more clearly visible in the 25R total ion current (TIC) trace. Our findings 269 indicate that the organic contaminant profile of site 25R is similar to the other LSP sites. This 270 suggests that the lack of plants at site 25R is not a result of differences in organic contaminant 271 composition.

272

274 3.3. Solvent extractable organic matter

275 3.3.1. Fraction 1 (saturated hydrocarbons)

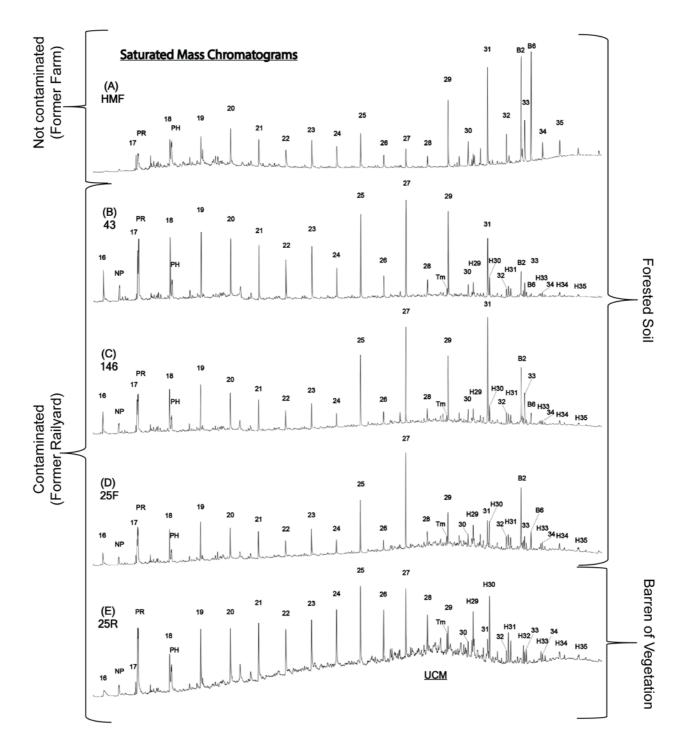
276 The saturated chromatograms (Fig. 4) contain two different sub-groups, normal isoprenoids (m/z 277 71) and triterpenoids (m/z 191). The distributions of saturated hydrocarbons indicate that the 278 soil of reference site HMF soil is distinctly different from soils from the four LSP sites (Fig. 4, 279 Supp. Fig. S1). The HMF samples show a predominance of two higher plant triterpenoid 280 biomarkers, diploptene [B2] and a β -amyrin derivative [B6] as well as the long-chain, odd 281 carbon number alkanes $n-C_{29}$ and $n-C_{31}$ (Figs. 4A and 5A). For soil samples 43, 146, and 25F, 282 we see a lower relative abundance of the two biomarkers [B2] and [B6] compared to HMF and a 283 broader range of odd-carbon number *n*-alkanes $(C_{24} - C_{31})$ (Fig. 4B-D, Fig. 5 B-D and Supp. Fig. 284 S1). The $C_{27} - C_{35}$ triterpenoid and isoprenoid alkanes are prominent in soils 43 and 146 (Fig. 4 285 B-D). The chromatogram for barren site 25R shows a distinctive "baseline hump", indicating an 286 unresolved complex mixture (UCM) of hydrocarbons (Fig. 4 E). Hopanes are relatively more 287 abundant in site 25R compared to the other sites, however, the biomarkers [B2] and [B6] are not 288 detected (Fig. 4 E).

289

Figure 4

There is notably more pristane [Pr] than phytane [Ph] at all LSP sites having pristane/phytane (Pr/Ph) ratios of *ca*. 3. In LSP soils, pristane was found more abundant than *n*-C₁₇ and phytane was found to be less abundant than *n*-C₁₈ (Fig. 4, Table 3, Supp. Fig. S1). Hopanes (m/z 191) are key compounds found in contaminated LSP sites (Fig. 5) and were not found in the reference site HMF (Fig. 5 A). Soils from 25R do not show the biomarkers [B2] and [B6], likely owing to the absence of plant biomass or low numbers of soil microbes.

Fig. 4. Total ion current (TIC) for fraction 1 of soil extracts from sites HMF (A), 43 (B), 146 (C), 25F (D) and 25R (E). The data show distributions of alkanes (normal isoprenoids (m/z 71) and triterpenoids (m/z 191). See Table 2 for compound symbols.



297

Table 3

298

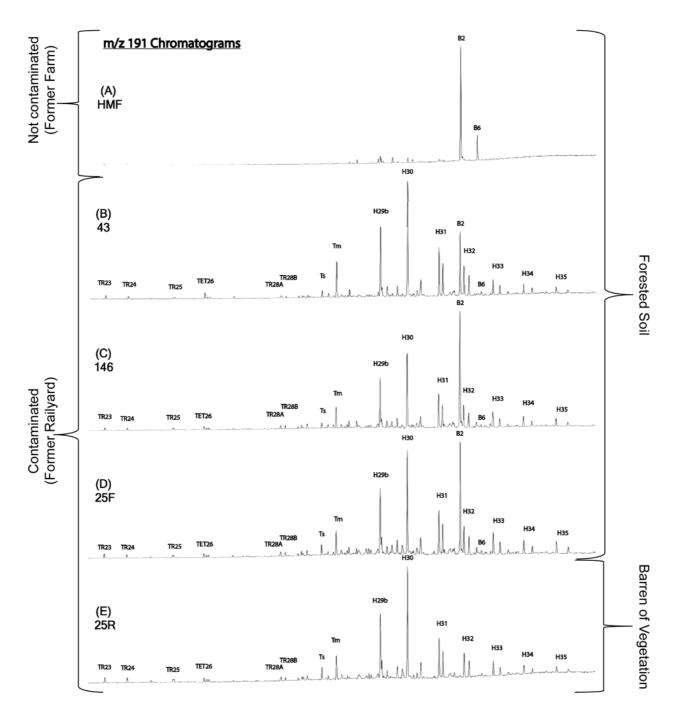
Figure 5

299	The presence of pristane, phytane, hopanes, tricylclic terpanes, sesquiterpanes and
300	steranes in LSP soils indicates fossil fuel contamination (Fig. 4, Supp. Fig. S2 and S3) (Killops
301	and Killops, 2005; Peters et al., 2005; Wang and Fingas, 2003; Wang and Stout, 2010; Wang et
302	al., 2005). The distribution of these compounds is similar in all samples analyzed, which
303	indicates a similar blend of contaminating hydrocarbons at the four LSP sites. Hopanes, clearly
304	visible in Figures 4 and 5, suggest the presence of heavy petroleum products such as fuel or
305	lubricating oil in addition to coal (Peters et al., 2005). The reference site HMF does not contain
306	hopanes, consistent with HMF soils containing less contamination compared to LSP. These
307	findings support our first hypothesis that we will find more fossil fuel contaminants in LSP
308	compared to HMF. The reference site HMF also yielded less of Fraction 1 (43.0 mg/kg _{soil})
309	compared to all LSP sites, where yields range from 211-549 mg/kg _{soil} . The high Pr/Ph ratios at
310	LSP sites (> 2.7) are consistent with the presence of bituminous coal (Table 3) (Peters et al.,
311	2005; Powell and McKirdy, 1973). The main railroad lines through sites 146 and 43 transported
312	coal from Pennsylvania to a coal pier to go to New York City (Fig. 1) (Caldes, 2010; French,
313	2002). Other railroad lines were directed to sites 25R and 25F and carried other cargo, however
314	all railroad tracks had locomotives that could spill coal. Our C27-C28-C29 sterane ternary diagram
315	pairs sites based on proximity to each other, for example, soils from sites 43 and 146 have higher
316	abundance of C ₂₉ compared to soils from 25R and 25F (Supp. Fig. S4) (Peters et al., 2005). The
317	GC-MS data on the saturated fraction reflects the presence of both coal and petroleum products
318	at LSP.

Table 3. Carbon preference index (CPI), odd even predominance (OEP), pristine/phytane (Pr/Ph), Pr/C17, Ph/C18, and weighted average carbon number of HMF, 43, 146, 25F and 25R are shown.

						weighted
						average
						carbon
Sample	CPI	OEP	Pr/Ph	Pr/C17	Ph/C18	number
HMF	2.573	2.782	0.634	2.464	2.137	26.33
43	4.726	4.480	3.533	2.667	0.539	23.82
146	6.770	6.689	2.720	2.430	0.635	25.23
25F	4.067	4.877	2.960	3.576	0.689	24.53
25R	1.533	1.225	3.148	11.853	1.264	23.80

Fig. 5. Mass chromatogram (m/z 191) showing the distribution of hopanes and tricyclic terpanes in the saturated fractions. The β -amyrin derivative is a soil microbe biomarker that was observed in all sites except for 25R. See Table 2 for compound symbols.



319 An odd-even predominance (OEP) is observed for HMF and the forested LSP soils. 320 especially in extracted m/z 71 (Fig. 4 and Supp. Fig. S1), reflecting the presence of higher plants, 321 specifically plant waxes. The triterpenoid biomarkers [B2] and [B6] are chemical signatures of 322 soil microbes and are seen in HMF and forested LSP soils. Not surprisingly, soils from barren 323 site 25R do not display OEP or biomarkers [B2] or [B6], consistent with the site's lack of 324 vegetation (Fig. 4). Within LSP, site 25R has higher isoprenoid to normal alkane ratios 325 (pristane/*n*-C17 and phytane/*n*-C18) compared to the other sites. This can be explained by 326 preferential biodegradation of *n*-alkanes compared to the isoprenoids at some point in the site's 327 history (Table 3) (Peters et al., 2005). Biodegradation at site 25R is also supported by the UCM 328 as seen in Figure 4E. This is surprising because site 25R supports no vegetation. It is possible 329 that microbial degradation of fossil fuels at site 25R took place at the time of contamination and 330 the signs we observe are from an earlier period of degradation.

331

332 3.3.2. Fraction 2 (aromatic hydrocarbons)

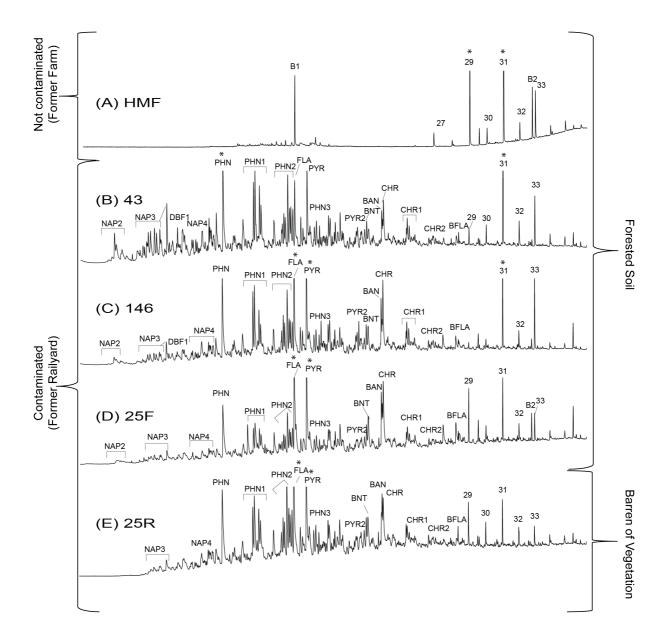
333 Soil extract from the reference site HMF had lower amounts of Fraction 2 (predominantly 334 aromatic hydrocarbons) (52.0 mg/kg_{soil}) than did LSP soil extracts (Table 1). For the forested 335 LSP sites, extract Fraction 2 values ranged from 363 to 506 mg/kg_{soil} while LSP soil 25R yielded 336 somewhat less than the other LSP sites, 119 mg/kg_{soil} of extract Fraction 2. We intended for the 337 LC separation to segregate saturated hydrocarbons into Fraction 1 and aromatic compounds into Fraction 2 but some long chain aliphatic hydrocarbons (> C_{27}) also eluted into Fraction 2 (Fig. 338 339 6). We observed only one major aromatic compound in the solvent extract from the 340 uncontaminated reference site HMF (Fig. 6 A), which is an abietane derivative [B1]. In the 341 extracts from LSP, the primary aromatic compounds are PAHs (Fig. 6 B - E) and include parent

342	and alkylated napthalenes, phenanthrenes, pyrenes, chrysenes and their isomers [e.g. NAPx,
343	PHN, PYR, CHR]. Also notable in LSP soils are heterocyclic compounds including
344	dibenzofurans and benzonaphthothiophene [DBF, BNT]. Sample 25R's PAH distributions
345	resemble those in the other LSP samples but there is also a noticeable UCM (Fig. 6 E).
346	Figure 6
347	The aromatic fraction data indicate that the chosen reference soil HMF is indeed relatively
348	uncontaminated while all LSP sites have significant PAH contamination. These data also
349	support our hypothesis that there are more fossil fuel biomarkers present at LSP compared to
350	HMF. The overall aromatic hydrocarbon distribution is similar for all LSP soils (Fig. 6 B-E),
351	reflecting similar mixtures of contaminants. Fraction 2 data for LSP 25R soil indicated a
352	significant UCM, which suggests weathering and biodegradation, in agreement with Fraction 1
353	data (Figs. 4 E, 6 E). If biodegradation is taking place, we might expect the concentration of low
354	weight molecular PAHs, like naphthalene, to decrease (Biache et al., 2017). We observe PAHs in
355	our data; perhaps these compounds are absorbed in coal particles and thus shielded from
356	biodegradation (Liu et al., 2008). We note that the PAH distributions in the pyrolyzates of the
357	LSP soils and extracted Fraction 2 show a greater predominance of parent PAHs than those of
358	the extract (Fraction 2), as seen by comparison of Figure 3E with 6E and specifically with PHN
359	in Supplemental Fig. S5 (Wang and Fingas, 2003). The greater predominance of parent
360	compounds indicates a pyrogenic source, which derives from incomplete combustion of fossil
361	fuels (Chen et al., 2004; Given, 1987; Micić et al., 2011; Stout and Emsbo-Mattingly, 2008).
362	
363	3.3.3. Fraction 3 (polar compounds)

364 The soil from the reference site HMF contains ergosterol and stigmasterol derivatives [S2, S3,

365 S4, S5, S6], a diterpenoid tentatively identified as totarol [B13], and octadecanoic acid butyl

Fig. 6. Total ion current (TIC) for fraction 2 of soil extracts from the HMF (A), 43 (B), 146 (C), 25F (D) and 25R (E) sites. The data show fewer polycyclic aromatic hydrocarbons in HMF. Distributions of aromatic compounds are similar for chromatograms B, C, D, and E. See Table 2 for compound symbols.



366	ester [FA4] (Fig. 7 A). This ester and the steroids are also present in the forested soils from LSP
367	(Fig. 7 B-D). The plasticizer, diisobutyl phthalate [X1], is present in all soils. Compounds
368	present in all LSP sites, but not in HMF, include 1-heptacosanol and 1,2-benzenedicarboxylic
369	acid, bis(2-ethylhexyl) ester [B14, X2]. The triterpenoids tentatively identified as 3β -methoxy-
370	D-friedoolean-14-ene and olean-12-en-3-one [B15, B16] were present in LSP 43 and 146 soils,
371	while lupeol and lup-20(29)-ene-3,28-diol [B17, B21] were found in LSP 146 and 25R soils
372	(Fig. 7 B-D). LSP site 25R has fewer polar compounds compared to the other sites but includes
373	the steroids stigmast-5-en-3-ol and stigmast-4-en-3-one [S4, S6] (Fig. 7 E).
374	Figure 7
375	The soil at HMF contains both ergosterol and stigmasterol derivatives. Ergosterols are
376	generally found in fungi, while stigmasterol typically occurs in higher plants (Peters et al., 2005).
377	HMF also contains totarol [B13], which is a cyclic diterpenoid that is present in resins from
378	higher plants (Tinoco et al., 2006), and a long chain alcohol, 1-heptacosanol, which is present in
379	terrestrial organic matter (Wang et al., 2009; Yunker et al., 1995). All of these compounds are
380	compatible with a normal, temperate forest soil. The vegetation present at HMF is
381	predominantly cedar. The plasticizer diisobutyl phthalate [X1] is present in all the studied soils,
382	possibly a result of having stored the soil samples in plastic bags. The origin of octadecanoic
383	acid butyl ester in soils is of unknown origin and could be another plasticizer introduced during
384	sample handling (Cahill et al., 2006). The dominant species present at LSP is Betula populifolia,
385	which is present at 43, 146 and 25F (Evans et al., 2015; Gallagher et al., 2008b). Lupeol [B8]
386	and lup-20(29)-ene-3,28-diol [B12] are biological indicators of birch (Peters et al., 2005). The
387	microbially and plant-derived steroids and triterpenoids are much less abundant at site 25R, not
388	unexpected since it is void of vegetation. The presence of life in 25R is also apparent in the

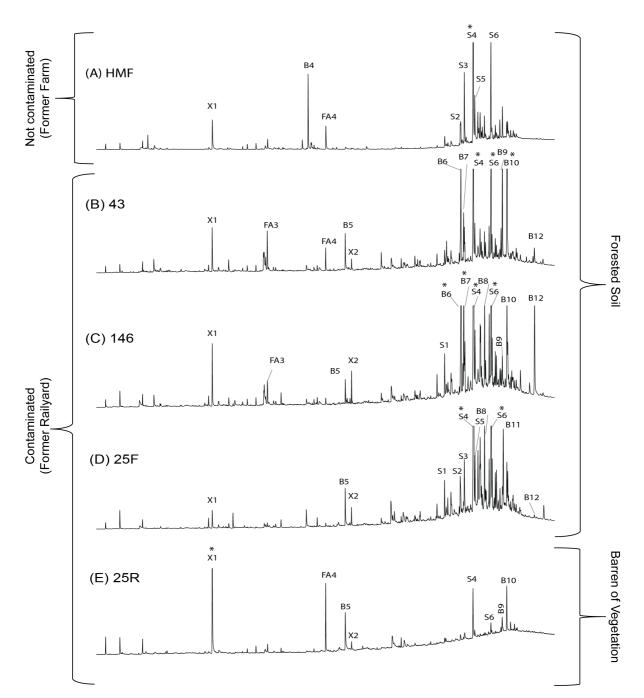


Fig. 7. Total ion current (TIC) for fraction 3 of soil samples from the HMF (A), 43 (B), 146 (C), 25F (D) and 25R (E). See Table 2 for compound symbols.

389 polar compounds, specifically sitosterol (stigmast-5-en-3-ol) and the degradation product of

390 sitosterol, stigmast-4-en-3-one (Mackenzie et al., 1982; Pautler et al., 2013).

391 3.4. Organic petrography

392 All samples (soil particles smaller than 2 mm) from LSP had high amount of organic matter and 393 it was represented dominantly by anthropogenic material. Coal particles were prominent in all 394 four samples, accounting for 14.3 to 30.0 vol. %, and they were especially abundant in samples 395 25R and 43. Coal particles were of various ranks from high volatile bituminous to anthracite, 396 with the latter being the dominant rank in all samples (Fig. 8). Carbonization-derived material 397 (coke) is also common and ranges from 6.6 to 12.0 vol. %, with samples 43 and 146 having the 398 largest proportions (Fig. 8 and 9). Tar and pitch-like material were prominent in samples 25R 399 and 25F (Figs. 8 and 9), and far less abundant in the other two samples. Combustion-derived 400 material accounted for 7.1 to 23.1%, being most common in sample 146 (Fig. 8). The 401 combustion-derived material was represented both by organic material (char) and mineral matter 402 (glass, spinel, etc.) (Fig. 9). In total, anthropogenic material accounted for more than 50 vol % of 403 all the samples with site 25R soil having the largest contribution (Fig. 8). Recent organic matter 404 (classified as "other" in Fig. 8) was rare and represented by wood fragments. 405 Figure 8 406 Figure 9 407 Organic petrography was useful in comparing LSP sites in terms of soil organic 408 components, including different ranks of coal, coke, tar and pitch, fly ash/bottom ash, sediments, 409 and other materials (Suárez-Ruiz et al., 2012). The organic petrography profiles of sites close to 410 each other at LSP share similarities; for example, soils from 43 and 146 that are located close in

411 proximity and are along train tracks to a coal pier based on historic aerial photographs (Fig. 1B)

Fig. 8. Organic petrographic composition (in volume %) of samples 43, 146, 25F and 25R. The data show that 25F and 25R have more tar/pitch compared to 43 and 146. HMF soil was not analyzed for organic petrography because coal particles were not visible to the naked eye.

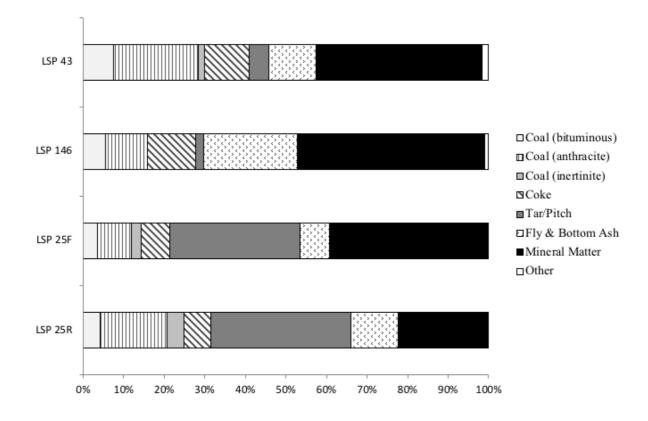
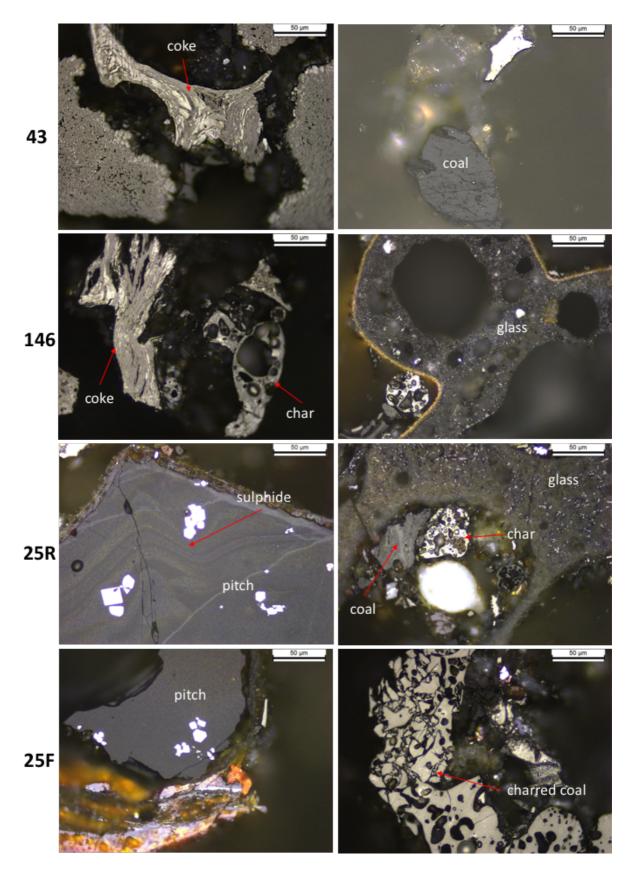


Fig. 9. Examples of photomicrographs of anthropogenic particles identified in the samples. Each sample has two photomicrographs. Reflected light, oil immersion.



412 have the highest abundance of coke. Soils from 25F and 25R, which are adjacent sites, were

413 both prominent in tar and pitch-like material. Site 25R has the largest proportion of non-mineral

414 material (Fig. 8), which could be a reason for or the result of a lack of vegetation at this site.

415

416 3.5. Inorganic elemental analysis

417 The concentrations of several inorganic compounds in HMF and LSP soils were also determined

418 (Supp. Table S1). As expected, we found the lowest concentrations of metals (V, Cr, Ni, Cu, Zn,

419 As, and Pb) at the reference site HMF; however, HMF soil showed a high Al concentration

420 ($20,527 \pm 249$ ppm), which was matched by only one of the LSP sites, 25R (Supp. Table S1).

421 There were elevated concentrations of Na, Co, Cu, Zn, As, and Pb at both 25F ([Na]: 873; [Co]:

422 256; [Cu]: 2256; [Zn] 14435; [As] 630; [Pb] 7145 mg/kg, triplicate measurements, see Supp.

423 Table S1 for errors) and 25R ([Na]: 4393; [Co]: 869; [Cu]: 7165; [Zn] 41271; [As]: 1162; [Pb]

424 20302 mg/kg) compared to the other LSP sites (Supp. Table S1).

425 Sanders (2003) in a survey of New Jersey soils, determined background levels of metals;

426 the HMF metal concentrations closely match his values from northeastern New Jersey ("urban

427 Piedmont") soils (Supp. Table S1) (Sanders, 2003), confirming that HMF is a good reference soil

428 for this study. The high concentrations of Al in HMF relative to LSP soils may suggest a higher

429 abundance of aluminosilicate clay minerals in HMF compared to LSP (Barton et al., 2002;

430 Buettner and Valentine, 2011). Figure 10 shows element concentration anomalies at LSP sites;

431 metal concentrations were normalized to the reference site HMF values. Specifically, we find

432 anomalously high levels of Na, Co, Cu, Zn, As, and Pb in LSP soils, especially at sites 25F and

433 25R (Fig. 10). For example, As concentrations in 25F and 25R were found to be 119 and 220

434 times higher compared to HMF, respectively. Although these sites are adjacent, As

435	concentrations are about two times higher at 25R compared to 25F. It is common to find As, Cr,
436	Cu, Pb and Zn in contaminated soil. In particular, As, Cu, Cr, and Zn are associated with
437	chromated copper arsenate, which was used in some rail yards for treating wood used for railroad
438	ties (Kumpiene et al., 2008). Another possible explanation for the anomalously high
439	concentrations is that bulk minerals, perhaps pyrites, sulfides, or oxides, were transported on the
440	railroad lines, possibly supplying not only Fe but also As, Cd, Co, Cu, and Ni (Finkelman, 1995).
441	It is also possible that coal and slag, which can include the trace elements in coal ashes, were
442	transported through LSP (Schweinfurth, 2003).
443	Figure 10
444	
445	3.6 Extracellular soil phosphatase activities and bacterial density
446	Site 146 has the highest bacterial density (cells/ $g_{dry \text{ soil}}$) (7.87 (± 0.82) × 10 ⁸) of the sites (Fig. 11
447	A). Soils from HMF (4.64 (± 0.93) × 10 ⁸), 43 (4.37 (± 0.35) × 10 ⁸), and 25F (4.97 (± 0.61) ×
448	10^8) are forested and have similar bacterial densities to each other. The barren site 25R has
449	lower but measurable bacterial density $(2.09 (\pm 0.76) \times 10^8 \text{ cells/g}_{dry \text{ soil}})$. The data in Figure 11 B
450	show that site 146 has the highest phosphatase activity $(2.60 (\pm 0.55) \times 10^6 \text{ pmol/g/h})$ of the soils
451	studied here. This result agrees with previous findings where we found that site 146 has higher
452	phosphatase, cellobiohydrolase and L-leucine-amino peptidase activities compared to three other
453	LSP sites and to HMF (Hagmann et al., 2015). Despite the measurable bacterial density, the
454	phosphatase activity at barren 25R is below the threshold of detection. It is surprising that site
455	25F, which is located next to site 25R, has high phosphatase activity (pmol/g/h) (1.38 (\pm 0.39) ×
456	10 ⁶). The phosphatase activities at HMF and 43 are (0.47 (\pm 0.38) \times 10 ⁶) and (0.67 (\pm 0.13) \times
457	10 ⁶), respectively.

Fig. 10 Concentrations elements listed in order of atomic weight normalized to HMF. The data is arranged from the lowest to the highest (43, 146, 25F and 25R) average metal concentration. The dotted line present in A. and B. is the background HMF.

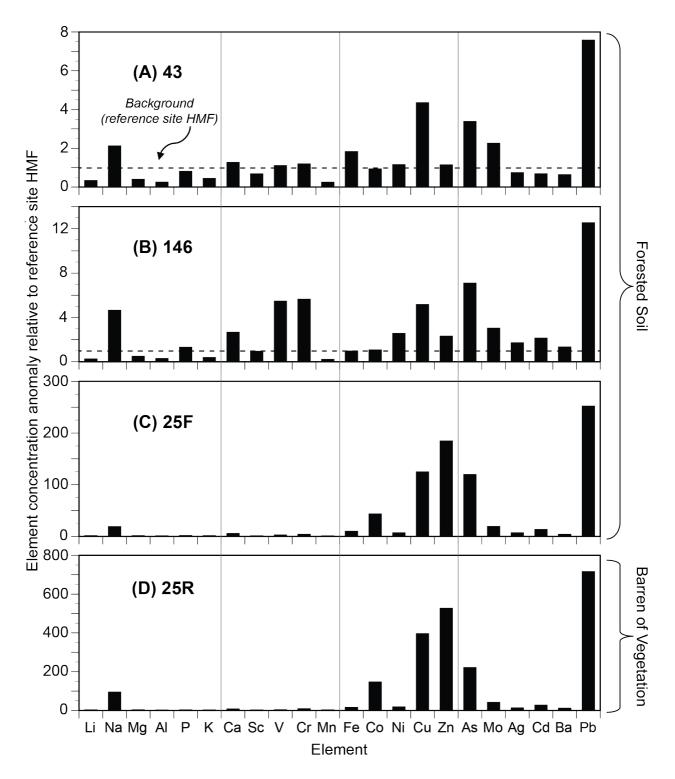
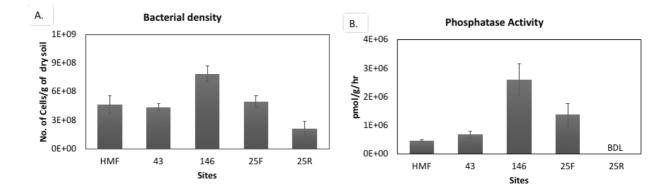


Figure 11

459	The low but measurable bacterial density we found at barren site 25R indicates the
460	presence of bacterial life. Even so, the phosphatase activity in the soil is below the threshold of
461	detection and it is possible the bacteria in 25R soil are dormant, perhaps owing to poor nutrient
462	cycling and availability. In a previous study, Singh and co-workers found via next-generations
463	sequencing that the bacterial community at site 25R is statistically distinct from the other LSP
464	sites (Singh et al., 2019b). The presence of bacteria in 25R soil is consistent with the
465	biodegradation at this site discussed in Section 3.3.1. It is possible that microbial degradation of
466	fossil fuels indicated by the UCM in the 25R GC-MS data took place at some time in the site's
467	history.
468	Site 146 has high bacterial density and phosphatase activity (Fig. 11) and also the highest
469	percent organic matter (44 %) (Table 1). We had originally hypothesized that the high function
470	at site 146 is due to differences in organic compound profiles, perhaps because of the absence of
471	specific contaminants that are present at other LSP sites. Conversely, we found that site 146 has
472	an organic composition similar to the other forested sites at LSP as seen in the pyrolysis and GC-
473	MS data (Figs. 3, 4, 5, and 6); however, organic composition may not explain the high activity at
474	this site. Site 25R is also distinct from the other LSP sites: it has no plants and no detectable
475	phosphatase activity. If this absence of function is a result of organic contaminants, we would
476	expect to observe differences in the organic contaminant profiles between site 25R and the other
477	LSP sites. Analogously to site 146, we found that the organic composition at site 25R is similar
478	to the other sites at LSP as seen in the pyrolysis and GC-MS data except that compounds
479	associated with higher plants are absent (Figs. 3, 4, 5, and 6).

Fig. 11. Bacterial density (cells/g dry soil, mean and error bars are SE; n = 3) and phosphatase activity (pmol*g⁻¹*hr⁻¹, mean and error bars are SE; n = 3) of HMF, 43, 146, 25F, and 25R soils are shown.



480 We considered the possibility that the low function at 25R is due not to differences in the 481 identity of organic contaminants but rather the concentration of specific compounds in the soil. 482 This is unlikely because the extract yields from 25R for soil particles less than 2 mm in diameter 483 are significantly lower than for the other LSP sites. These data together suggest that the organic 484 contaminant profile at site 25R is not responsible for the functional differences between this site 485 and other LSP sites. Previous findings by Singh et al. (2019a) showed that abiotic factors are 486 more important than biotic ones for enzymatic function in LSP soils. In light of these findings 487 and having now ruled out a different organic contaminant profile as the reason for the low 488 function at site 25R, our new hypothesis is that the higher concentrations of Na, Co, Cu, Zn, As, 489 and Pb at 25R (discussed in Section 3.5) are responsible for the poor function of the site (Singh 490 et al., 2019a). The concentrations of these metals are about 2 to 5 times higher than in the 491 forested and high functioning site 25F that is located adjacent to site 25R. We plan to further 492 investigate whether these elevated metal concentrations at 25R affect soil extracellular enzymatic 493 activity and plant production. For example, it is possible that the As concentrations at site 25R 494 are sufficiently high to inhibit phosphatase activity (Lee et al., 2011; Lorenz et al., 2006; Speir et 495 al., 1999).

496

497 3.7 Insights into functional variation between LSP sites

498 All LSP sites are contaminated with PAHs and other fossil fuel-derived components that are

499 absent or barely detectable at reference site HMF. The presence of the PAHs, hopanes, steranes,

500 and sesquiterpanes supports our first hypothesis that there are more fossil fuel biomarkers at LSP

501 compared to HMF. Vegetated sites at LSP have unnaturally high organic matter content (30 to

502 45 %), relative to pristine vegetated soils, reflecting large quantities of anthropogenic

503 contaminants in these soils. The distributions of the PAHs and fossil fuel biomarkers are similar 504 for all soils within LSP. However, the yield of extractable organic matter at 25R is lower and 505 thus we suspect that the extraordinarily high content of *inorganic* contaminants rather than 506 organic contaminants explains the barren site's lack of vegetation. We hypothesized that the 507 barren site 25R would have more or different contaminants compared to the forested sites at 508 LSP. This was found to be true for several inorganic contaminants but not for the organic 509 contaminant profiles within LSP.

510 LSP's history and the soils' chemical and petrographic fingerprints agree with the 511 location of the sites within the rail yard. Sites 43 and 146 are located on what was once a dense 512 array of railroad tracks leading to the coal pier. Their soils contain more coal and less tar and 513 pitch compared to sites 25F and 25R, which were on different railroad lines carrying other 514 unspecified cargo. We observe wider gaps between tracks in historical photographs of the 25F 515 and 25R area, in which we suspect that dumping activities occurred (Fig. 1 D and E). It is 516 possible that 25F is forested and has high phosphatase activity, because it was on an active 517 railroad track whereas 25R was on a gap, where more dumping occurred, and thus has low 518 phosphatase activity. While we did not find evidence that would explain these differences in the 519 nonbiologically-derived organic contaminant profiles, we note significantly higher metal 520 contaminant concentrations in 25R compared to 25F.

521

522 3.8 Implications for investigating contaminated ecosystems

523 In this study, we used several experimental methods in conjunction with pyrolysis GC-MS to

study soils from three types of sites: an uncontaminated vegetated reference site (HMF),

525 contaminated vegetated sites (LSP 43, 146, 25F) and a contaminated, industrial barren (25R).

526 Our data are consistent with the presence and absence of biologically derived organic matter in 527 the study sites: biological pyrolysis products are not present in the organic chemical fingerprint 528 of soil 25R, which instead shows a predominance of contaminant-derived aromatic compounds. 529 These data serve as a case study or reference set for the differences in data observed for 530 contaminated vegetated and contaminated barren soils, especially because indicators from 531 vegetation can chemically obscure the signatures of contaminants.

532 Site 25R in LSP is an example of an area that is contaminated and barren as a result of 533 past anthropogenic activities and, possibly, the subsequent environmental and physical 534 conditions. Such "industrial barrens" are present worldwide and are considered "extreme 535 environments" (Kozlov and Zvereva, 2007). Unfortunately, industrial barrens, and especially 536 those that have high heavy metal loads, are often unlikely to recover naturally but, even so, they 537 are studied less than other extreme environments. Many studies on industrial barrens focus on 538 evaluation of damage and development of rehabilitation measures while more fundamental 539 studies on soil function and detailed soil characterization are lacking, limiting our understanding 540 of the fundamental chemistry, biochemistry and ecology of these sites. Kozlov and Zvereva 541 (2007) suggested that researchers often choose the most contaminated but planted sites within an 542 area for their studies because they are more comparable with less disturbed sites.

543

544 4. Conclusion

545 Site 25R is an example of an industrial barren and here we have conducted a multifaceted 546 chemical and biochemical analysis of this soil and compared the findings to those in nearby 547 vegetated sites. The phosphatase activity at site 25R is below our limit of detection, yet direct 548 bacterial counts indicate the presence of bacteria. Interestingly, these findings suggest that

549 microbes in 25R soils are present but lying dormant, perhaps as a result of either inorganic 550 contaminant loads or physical abiotic restrictions in the soil. Perhaps fostering plant growth, 551 aerating the soil, or adding root exudates to the contaminated 25R soil will revive some of the 552 microbes and subsequently improve nutrient cycling and overall soil function. The inactive 553 microorganisms may be waiting for an environmental nudge or input to revitalize them and 554 unleash their functional potential in the currently contaminated, poorly functioning, and barren 555 25R soil. This same story of dormant and abiotically limited microbes, waiting to be revitalized, 556 may be playing out in other industrial barrens throughout the world. It behooves the scientific 557 community to more fully investigate these sites to obtain a deeper and more fundamental 558 understanding, which can then be leveraged to prevent land degradation and to restore 559 dysfunctional and phytotoxic soils.

560

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Liberty State Park.

Figure and Table Caption

568	Fig. 1. Liberty State Park is located in Jersey City, New Jersey (A). Aerial images from: U. S.
569	Geological Survey: 1954 (B, E) and 2014 (C). Study sites within LSP are indicated on the map
570	(43, 146, 25F, 25R). A historical photograph of 25F and 25R from 1951 (D). Sites 25F and 25R
571	are adjacent; 25R is in a strip of land without vegetation (F). Photo credits: D: Andrew
572	Bologovsky, F: Mike Peters (Montclair State University). Photos used with permission.
573	
574	Fig. 2. Flow chart illustrating the experimental design for the experiment.
575	
576	Fig. 3. Total ion current (TIC) for Py-GC-MS of soil samples from sites HMF (A), 43 (B), 146
577	(C), 25F (D) and 25R (E). The data for 25R show predominantly mono- and polycyclic aromatic
578	hydrocarbons and aromatic hydrocarbons. See Table 2 for compound symbols.
579	
580	Figure 4. Total ion current (TIC) for fraction 1 of soil extracts from sites HMF (A), 43 (B), 146
581	(C), 25F (D), and 25R (E). The data show distributions of alkanes (normal isoprenoids (m/z 71)
582	and triterpenoids (m/z 191). See Table 2 for compound symbols.
583	
584	Figure 5. Mass chromatogram (m/z 191) showing the distribution of hopanes and tricyclic
585	terpanes in the saturated fractions. The β -amyrin derivative is a soil microbe biomarker that was
586	observed in all sites except for 25R. See Table 2 for compound symbols.
587	
588	
589	

590	Fig. 6. Total ion current (TIC) for fraction 2 of soil extracts from the HMF (A), 43 (B), 146 (C),
591	25F (D), and 25R (E) sites. The data show fewer polycyclic aromatic hydrocarbons in HMF.
592	Distributions of aromatic compounds are similar for chromatograms B, C, D, and E. See Table 2
593	for compound symbols.
594	
595	Fig. 7. Total ion current (TIC) for fraction 3 of soil samples from the HMF (A), 43 (B), 146 (C),
596	25F (D) and 25R (E). See Table 2 for compound symbols.
597	
598	Fig. 8. Organic petrographic composition (in volume %) of samples 43, 146, 25F, and 25R. The
599	data show that 25F and 25R have more tar/pitch compared to 43 and 146. HMF soil was not
600	analyzed for organic petrography because coal particles were not visible to the naked eye.
601	
602	Fig. 9. Examples of photomicrographs of anthropogenic particles identified in the samples. Each
603	sample has two photomicrographs. Reflected light, oil immersion.
604	
605	Fig. 10. Element concentration anomalies relative to reference site HMF. Elements are
606	presented in order of atomic weight, with each normalized to its value in the HMF reference soil.
607	The plots are arranged from the lowest (site 43) to the highest (25R) overall metal concentration.
608	The dotted lines visible in A and B show the background HMF values.
609	
610	Fig. 11. Bacterial density (cells/g dry soil, mean and error bars are SE; $n = 3$) and phosphatase
611	activity (pmol*g ⁻¹ *hr ⁻¹ , mean and error bars are SE; n =3) of HMF, 43, 146, 25F, and 25R soils
612	are shown.

614	Table 1. Organic matter content (%, mean \pm SE; n = 3), extraction and liquid chromatographic
615	results from HMF, 43, 146, 25F and 25R soils. Fraction 1 contains saturated hydrocarbons;
616	fraction 2 – aromatic compounds and long chain (> C_{27}) normal alkanes; fraction 3 – polar
617	compounds. The data show that soil from site 146 has the highest percent organic matter. The
618	LC results show the percentage of the total extract as well as yield per kg of dry soil.
619	
620	Table 2 : Symbols for peak identification used in Figures 1-7.
621	
622	Table 3. Carbon preference index (CPI), odd-even predominance (OEP), pristine/phytane
623	(Pr/Ph), Pr/C17, Ph/C18, and weighted average carbon number of HMF, 43, 146, 25F and 25R.
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Supplementary Figures

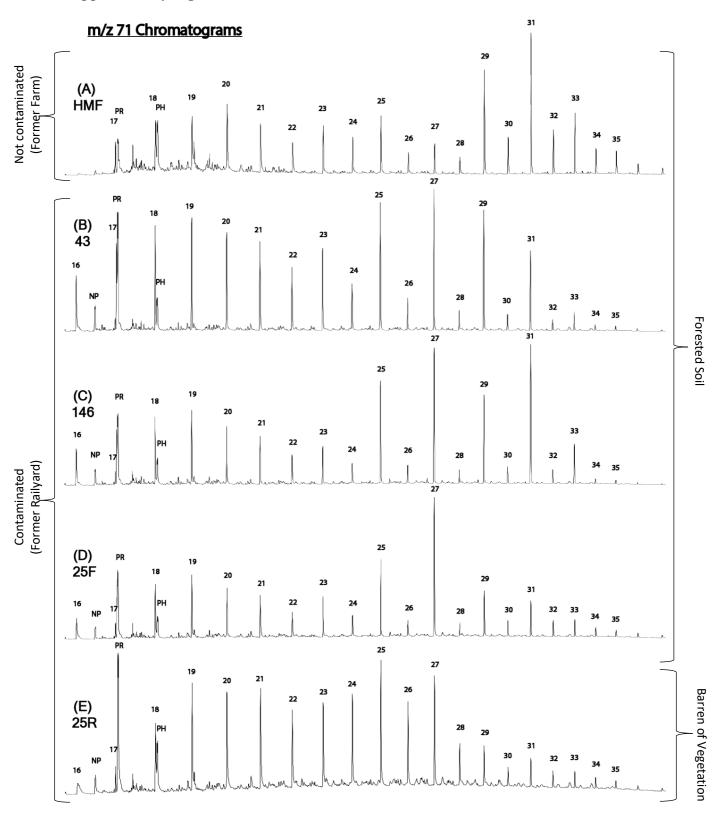


Fig. S1. Mass chromatograph (m/z 71) for all five samples (HMF, 146, 43, 25F, 25R), showing extracted normal alkanes with some isoprenoid compounds. See Table 1 for compound symbols.

m/z 123 Chromatograms

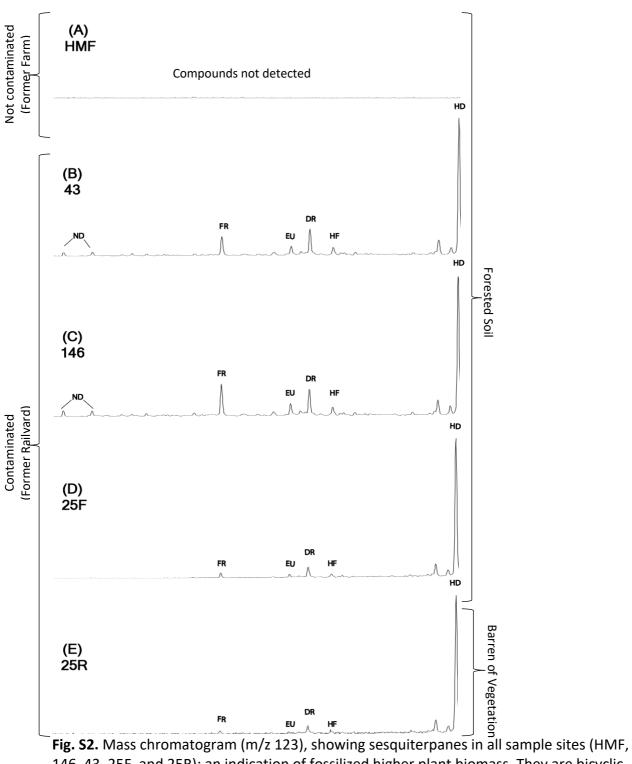


Fig. S2. Mass chromatogram (m/z 123), showing sesquiterpanes in all sample sites (HMF, 146, 43, 25F, and 25R); an indication of fossilized higher plant biomass. They are bicyclic compounds and the building components for triterpenoids. The compound symbols are as follows: nor-drimane (ND), farnesane (FR), 4A(H)-eudesmane (EU), 8B(H)-drimane (DR), homofarnesane (HF), 8B(H)-homodrimane (HD). Sesquiterpanes were not detected in HMF.

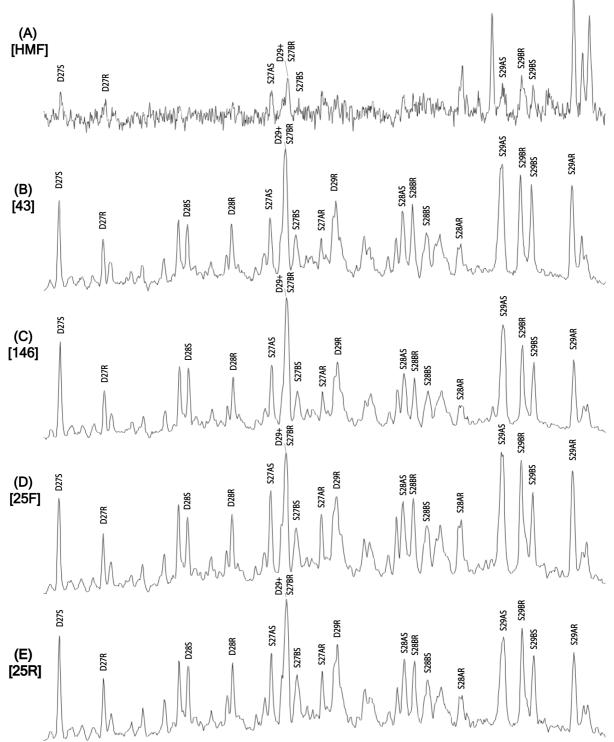


Fig. S3. Mass chromatogram (m/z 217), showing steranes in all sample sites (HMF, 146, 43, 25F, and 25R). The compound symbols are as follows:.

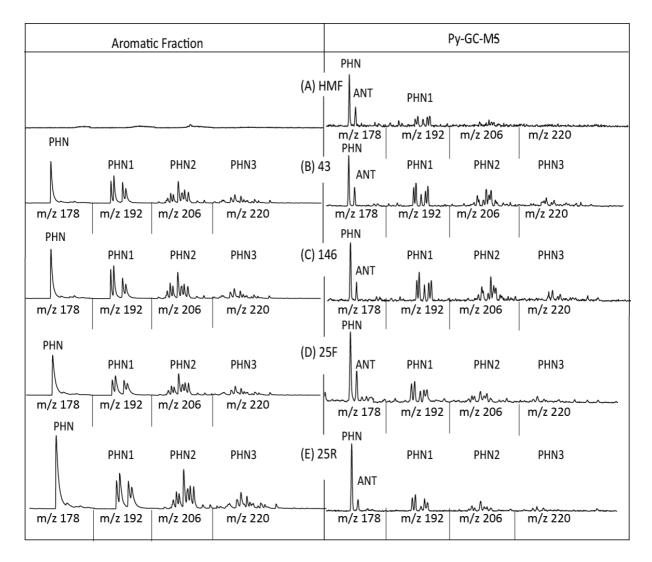


Fig. S4. Mass chromatograms (m/z 178, 192, 206, 220) showing the distribution of phenanthrene and methylated phenanthrenes in the aromatic fractions (on left) and Py-GC-MS (on right) of soil samples from HMF and LSP sites 43, 146, 25F and 25R. See Table 1 for compound symbols.

	43	146	25F	25R	HMF	Urban NJ Piedmont
mg/kg	Average ± standard error	Average ± standard error	Average ± standard error	Average ± standard error	Average ± standard error	Median
Li	7.7 ± 0.8	5.4 ± 0.7	24.1 ± 4.07	40.7 ± 12.4	23.8 ± 0.6	
Na	99 ± 8	218 ± 12	873 ± 129	4393 ± 669	47 ± 3	90.1
Mg	1240 ± 129	1477 ± 106	3132 ± 411	6213 ± 1819	3153 ± 60	2190
Al	4980 ± 325	5563± 1195	16132 ± 1519	20470 ± 4194	20527 ± 249	10500
Р	429 ± 17	690 ± 14	684 ± 14	789 ± 49	534 ±48	
Κ	839 ± 80	686 ± 16	1737 ± 113	2697 ± 404	1927 ± 78	693
Са	1948 ± 164	4065 ± 775	8228 ± 1402	10068 ± 1685	1542 ± 107	1425
Sc	2.06 ± 0.13	2.82 ± 0.25	2.40 ± 0.24	3.43 ± 0.48	3.06 ± 0.02	
V	32.9 ± 0.9	164 ± 8.3	70.7 ± 9.5	77.3 ± 11.8	30.1 ± 0.1	29.6
Cr	24.8 ± 0.7	118.0 ± 2.1	79.2 ± 3.2	165.2 ± 21.7	21.0 ± 0.3	18.5
Mn	159 ± 10	127 ± 4	460 ± 32	1042 ± 141	674 ± 56	311
Fe	34061±1819	17520 ± 1491	179062 ± 22211	266653 ± 59863	18722 ± 1540	14600
Со	5.5 ± 0.4	6.2 ± 0.3	256.5 ± 46.5	868.7 ± 178.8	6.0 ± 0.1	6.3
Ni	21.1 ± 1.4	46.7 ± 0.9	121.4 ± 13.8	317.4 ± 62.5	18.5 ± 0.3	12.4
Cu	79 ± 4	93 ± 3	2256 ± 320	7165 ± 1512	18 ± 2	29.5
Zn	89 ± 4	180 ± 24	14435 ± 2845	41271 ± 8526	78 ± 4	75.3
As	17.8 ± 1.0	37.4 ± 4.5	630 ± 187	1162 ± 207	5.3 ± 0.1	5.2
Мо	3.5 ± 0.1	4.7 ± 0.4	29.3 ± 1.6	63.1 ± 10.1	1.6 ± 0.1	
Ag	0.42 ± 0.02	0.97 ± 0.01	3.80 ± 0.07	7.04 ± 0.79	0.58 ± 0.01	< D.L.
Cd	0.20 ± 0.01	0.63 ± 0.06	3.88 ± 0.63	7.74 ±	0.30 ± 0.02	< D.L.
Ba	88 ± 3	181 ± 15	533 ± 53	1467 ± 267	139 ± 4	80.6
Pb	215 ± 9	355 ± 18	7145 ± 1351	$\begin{array}{c} 20302 \pm \\ 4203 \end{array}$	28 ± 1	111

Supplement Table S1. Concentrations ($\mu g/g$) of elements listed in order of atomic weight for the average of HMF, 43, 146, 25F and 25R of three replicates. (1 ± S.E.). Urban NJ Piedmont is from Sanders 2003, where values indicated are the median concentration (mg/kg). < D.L. is below detection limit, which is Ag = 0.2 and Cd = 0.4 mg/kg.