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### **Paleoenvironment Reconstruction of the Eocene Southeastern Tethys Using Geochemistry of Sedimentary Rocks**

Liliana Calderón Convers

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## ABSTRACT

### PALEOENVIRONMENT RECONSTRUCTION OF THE EOCENE SOUTHEASTERN TETHYS USING GEOCHEMISTRY OF SEDIMENTARY ROCKS

By Liliana Calderón Convers

The Early Eocene hyperthermals are a series of transient global warming events that occurred between 54.09 and 52.63 Ma, and are considered as ancient analogues for future climate change. These hyperthermal events are well studied in the deep sea sites, but have not been investigated in the eastern Tethys Ocean. Here, we report new major and trace element geochemistry and stable carbon isotope data of the Early Eocene strata deposited in a shallow marine foreland basin prior to the main stage of India-Asian continent collision at Qumiba section in the Tingri County, southern Tibet. These geochemical data are used to reconstruct the paleoenvironment and past carbon cycle dynamics in the Early Eocene. The collision of India with Asia continent led to the formation of the Himalayas and the demise of the Tethys Ocean. This tectonic event played a significant role in the paleotopography, paleoceanography and paleoclimate of the southeastern Tethys. The youngest marine sediments (Enba Formation and Zhaguo Formation) represent the closure of the Tethys Ocean, and suggest an age of 53.67 to 52.63 Ma. Major element geochemistry and weathering indices indicate that the Eocene southeastern Tethys may have experienced periodic pulses of nutrient-rich detrital sediments, weathering supplies and the development of hypoxia in the water column, consistent with the warm and humid climate in the Early Eocene.

**Keywords:** Paleoenvironment, Eocene Thermal Maximum 2 (ETM 2), Qumiba section, geochemistry, Carbon, Nitrogen stable isotopes.

MONTCLAIR STATE UNIVERSITY

PALEOENVIRONMENT RECONSTRUCTION OF THE EOCENE SOUTHEASTERN  
TETHYS USING GEOCHEMISTRY OF SEDIMENTARY ROCKS

by

LILIANA CALDERON CONVERS

A Master's Thesis Submitted to the Faculty of

Montclair State University

In Partial Fulfillment of the Requirements

For the Degree of

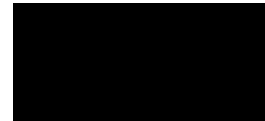
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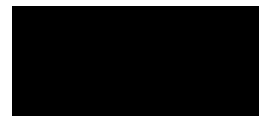
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TETHYS USING GEOCHEMISTRY OF SEDIMENTARY ROCKS

A THESIS

Submitted in partial fulfillment of the requirements

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LILIANA CALDERÓN-CONVERS

Montclair State University

Montclair, NJ

2020

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## 1 Introduction

The initial stage of the continental collision between India and Asia has complex consequences for the paleoenvironment and paleoclimate of the eastern Tethys ocean before its final closure (Aitchison et al., 2011; Hu et al., 2012; Jiang et al., 2016; Najman et al., 2010; Zhu et al., 2019). Numerous studies have investigated the age of the initial collision, but no consensus has been reached, with estimates ranging from ~65 to 34 million years ago (Ma) (Cai et al., 2011; Rowley, 1996; Searle et al., 1987; Singh, 2013; van Hinsbergen et al., 2012; Wang et al., 2012; Yin and Harrison, 2000). Furthermore, the growth of the Himalaya-Tibetan orogen is thought to have played a critical role in driving the long-term Cenozoic cooling through the negative silicate weathering feedback (Caves et al., 2016; Kump and Arthur, 1997; Raymo and Ruddiman, 1992). Despite these efforts, few studies have looked into the paleoenvironmental evolution of the eastern Tethys during the initial stage of the India-Asia continental collision (e.g., Kahsnitz, 2017; Wang et al., 2017; Zhang et al., 2012), possibly because of the sparse age constraint of this interval.

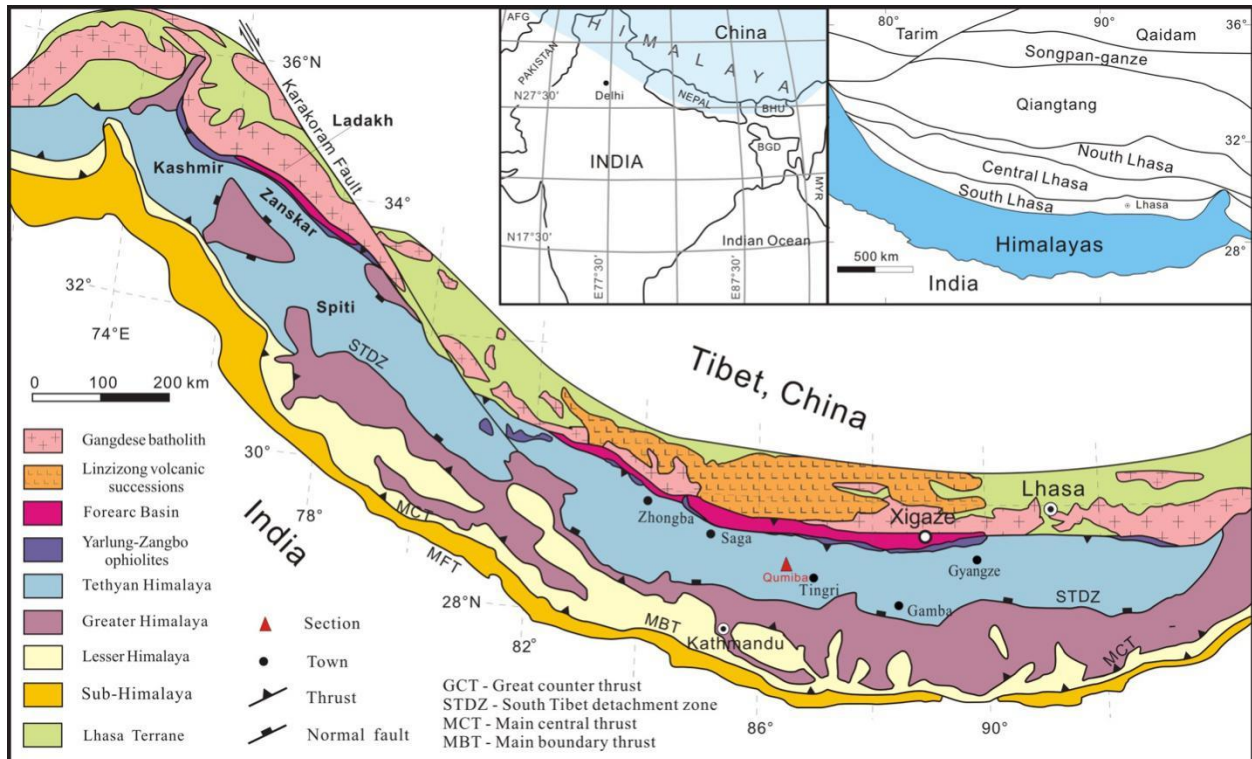
The Early Eocene Ypresian (ca. 56 to 47.8 Ma) is characterized by a hothouse climate with temperatures 15 °C warmer than the present day (Miller et al., 2020), and was punctuated by a series of rapid global warming events, known as Early Eocene hyperthermals (Nicolo et al., 2007; Slotnick et al., 2012). Large benthic foraminifera and negative carbon isotope excursion (CIE) allow for the recognition of the Paleocene-Eocene Thermal Maximum at Tingri in the Early Eocene Tethyan Himalaya (Zhang et al., 2013), but evidence for other Early Eocene hyperthermals is still lacking. Here, we present major and trace elements data and stable isotopes of carbon and nitrogen from the Qumiba section across the Early Eocene Enba and Zhaguo Formations in the hope to link the Early Eocene hyperthermal events to

geochemical characteristics of the siliciclastic sediments deposited in the foreland basin during the initial collision stage of the India-Asia continents.

## 2 Geological Background

Southern Tibet was located in the Tethyan Himalaya of the northern Greater Indian continental margin, formed by the collision of India and Asia in the Eocene epoch (~55 Ma) (Ding et al., 2016; Hu et al., 2016; Najman et al., 2017; Wang et al., 2008) (Figs 1 and 2).

Prior to the collision, this region was part of the interior Tethys oceans in the Cretaceous. The collision of Indian-Asian plates resulted in the closure of the Neo-Tethys Ocean and subsequent stages of tectonic uplift and the formation of the Tibetan Plateau (Spicer et al., 2020; Wang et al., 2008). Therefore, the sediments of the Eocene period record the final stages of sedimentation on the southern margin of the Tethys ocean (Najman et al., 2010).



**Figure 1. Geological map of the Himalayas modified from Hu et al. (2012) and Najman et al. (2017). The inset maps show the location of the Himalayas. The studied Qumiba section is shown in red triangle, which is located in the Tingri region in southern Tibet, China.**

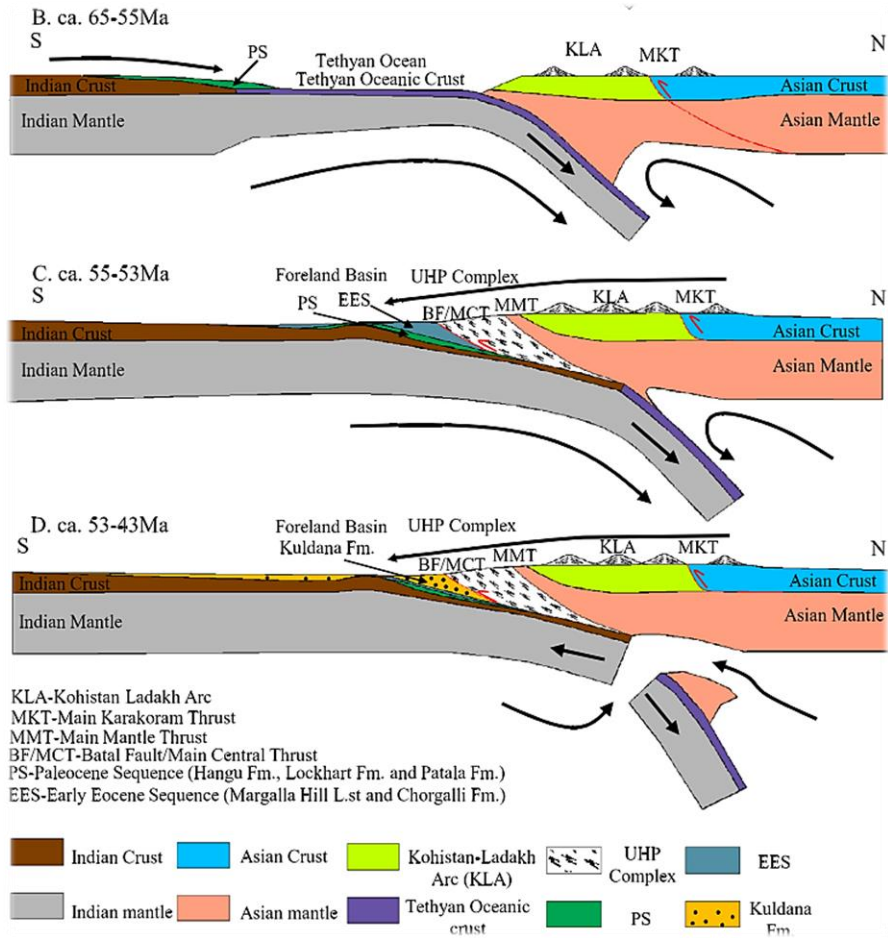
The study site, the Qumiba section, is located south of the Yarlung-Tsangpo structure in central Tethyan Himalayas in southern Tibet, west of the Zhepure Shan mountain in Tingri area (GPS N28° 41' 26", E86° 43' 37", elevation 4924 to 4970 m). Six stratigraphic units have been identified in the Zhepure Shan mountain (Willems et al., 1996), *i.e.* from oldest to youngest the Gamba Group (late Albian-early Santonian), the Zhepure Shanbei Formation (early Santonian-middle Maastrichtian), the Zhepure Shanpo Formation (middle Maastrichtian-early Paleocene), the Jidula Formation (Danian), the Zhepure Shan Formation (late Danian-Lutetian) and Zongpubei Formation (Lutetian or younger). The Gamba Group, Zhepure Shanbei Formation and Zhepure Shan Formation mainly consist of limestones and marls, while the Zhepure Shanpo Formation, Jidula Formation and Zongpubei Formation mainly consist of sandstones and mudstones. The Zongpubei Formation is composed of green-colored mudstones and sandstones overlain by red-colored mudstones, wackestones and sandstones, corresponding to Enba Fm. and Zhaguo Fm. respectively (Hu et al., 2012; Wang et al., 2002; Willems et al., 1996).

The studied successions consist of the upper part of Enba Formation and the lower part of the Zhaguo Formation (Najman et al., 2010). The Enba Formation consists of greenish-grey mudstones with thin interbedded sandstones and few thin limestones, with a thickness of approximately 105 m (Fig. 3). The quartzose sandstones are mature and lithic-rich, which are mainly volcanic fragments (Zhu et al., 2005). The sandstone beds become thicker and more abundant in the upper part of Enba Fm., with horizontal and wavy cross-lamination. Among them, hummocky cross-stratified beds can also be found, which implies the influence of storm waves (Figure 3; Zhu et al., 2005). It has been suggested that the Enba Fm. was deposited in an outer shelf environment, in which the mudstones were deposited by

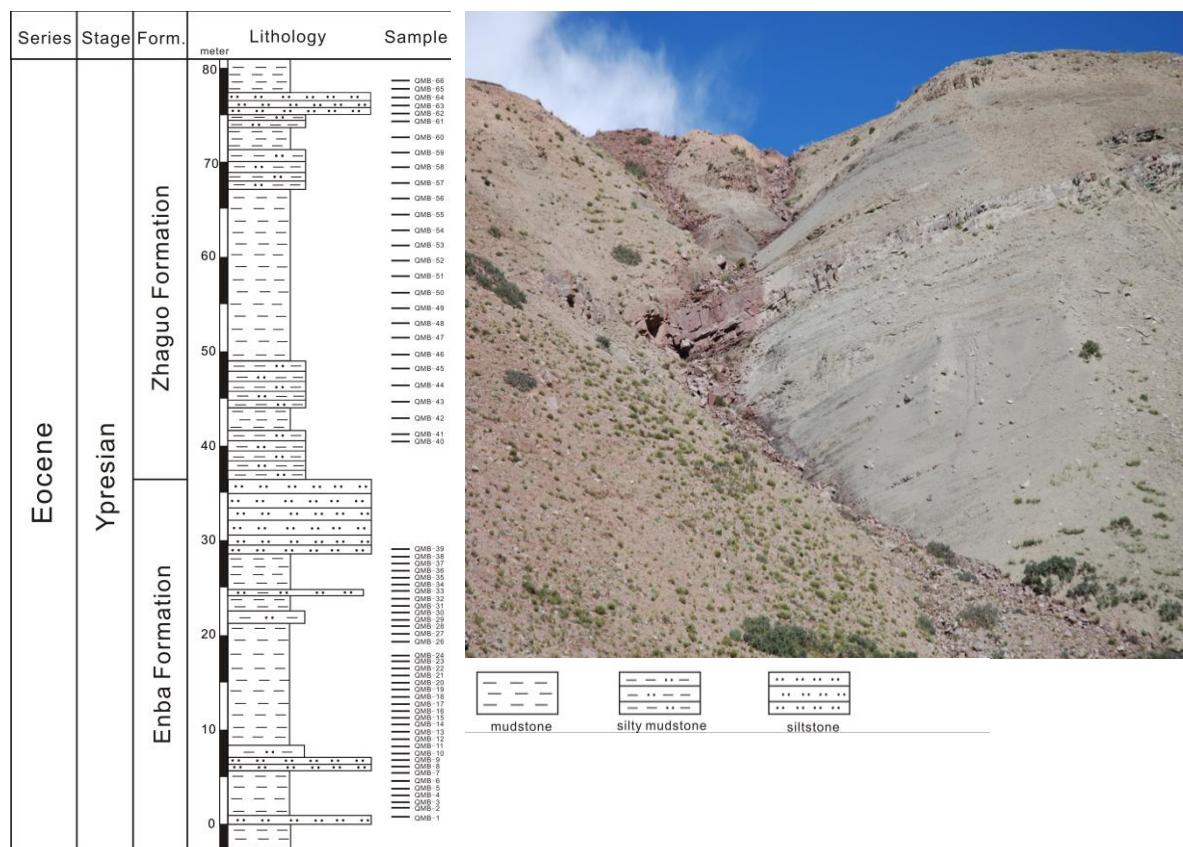
suspension settling and sandstones by turbidity currents or high-energy events (Hu et al., 2012; Zhu et al., 2005).

The overlying Zhaguo Fm. (~75m) is composed of red mudstones and interbedded sandstones, and has similar lithologies to those of the Enba Fm. (Zhu et al., 2005). The boundary between Enba Fm. and Zhaguo Fm. is marked by a thick sandstone bed (~8m) featured by poorly sorted, angular, and pebble/cobble-sized grains derived from the underlying unit, interpreted as paleoregolith (Zhu et al., 2005). The subtle angular variations are observed by the minor dip differences between the Enba Fm. and the Zhaguo Fm. ( $\approx 11^\circ$ ). But it is still controversial whether unconformities and depositional hiatus exist at the Qumiba section (Najman et al., 2010; Hu et al., 2012). The thin sandstone beds in the Zhaguo Formation are commonly characterized by fining upwards, scoured bases, trough cross-lamination, horizontal lamination, and wavy cross-lamination. The sedimentary environment was interpreted as the shallow marine shelf environment (Wang et al., 2002).

The ages of the Enba Fm. and Zhaguo Fm. are Early Eocene, evidenced by no difference of the stratigraphically mixed calcareous nannofossil assemblages preserved in the two formations (Najman et al., 2010). Since the youngest assemblage can represent the age of the sediments, the ages of the two formations correspond to Ypresian stage (50.6–53.5 Ma) according to the nannofossil zones NP11–12, which is also consistent with the foraminiferal zones P7–8 (ca. 50.4–52.3 Ma) (Zhu et al., 2005; Najman et al., 2010).



**Figure 2. Tectonic evolution model of India-Asia collision (Ding et al., 2016)**



**Figure 3.** The stratigraphic log of the Eocene strata of the Qumiba section in this study, sample positions are shown as horizontal bars. Also shown is the field photography of the Qumiba section, which consists of Lower Eocene Enba and Zhaguo Formations with the thick sandstone as their boundary.

### Motivation and Research Questions

Previous studies have shown that the nannofossil and foraminifera assemblages from the Lower Eocene Enba Formation and Zhaguo Formation are strongly influenced by the recycling and redeposition of Cretaceous and Paleocene age sediments from the Lhasa Terrain (Hu et al., 2012; Najman et al., 2010). This research aims to answer three research questions: 1) What are the paleoenvironmental and paleoclimatological responses in the eastern Tethyan ocean to tectonic forcings during India-Asia collision? 2) Are Early Eocene hyperthermal events recorded in the eastern Tethyan ocean? 3) What are the roles of recycled continental materials in the sediment provenance?

### **3 Materials and Methods**

A total of 66 freshly excavated outcrop samples were collected from the sandstone and mudstone bearing the Qumiba section for this study, and each sample was taken one meter apart from the Enba Formation to the Zhaguo Formation (Table 1). These samples were examined for major, trace and rare earth elements (REE) analysis using an Inductively Coupled Plasma – Mass Spectrometry (ICP-MS) housed at the Department of Earth and Environmental Studies at Montclair State University. Thirty out of 66 samples were analyzed for stable carbon and nitrogen isotopes via EA-IRMS (Elemental Analyzer - Isotope Ratio Mass Spectrometer) in order to understand the carbon and nitrogen cycle of the Early Eocene in response to the closure of the Tethys Ocean and Himalaya uplift.

#### **3.1 Major and Trace Element Analysis**

All the Qumiba samples were crushed using a SPEX zirconia ceramics ball mill. Around 100 mg powder was mixed with 400 mg of ultrahigh-purity lithium metatetraborate flux (<sup>(R)</sup>Spex-Certiprep) and fused in high-purity graphite crucibles at 1050 °C for 40 minutes in the Thermo Scientific™ Lindberg/Blue M™ Box Furnace. Afterwards, molten samples were immediately dissolved in 50 ml of 7% HNO<sub>3</sub>, then transferred for a second dilution of 6.5 mL of the sample plus HNO<sub>3</sub> (dilution factor of ~4000×) before ICP-MS analysis. The Milli-Q® Direct 8 Water Purification system was used for distilled and deionized experimental water; nitric acid from Fisher Scientific was of trace metal grade. Additional blanks and flux were prepared for analytical corrections to the analytical signal (<1.5% for all elements).



Instrument calibration was conducted using ten rock standards of the U.S. Geological Survey (USGS) (DNC-1, BHVO-2, G-2, W-2, BCR-2, GSP-2, QLO-1, AGV-2, BIR-1 and RGM-1). ICP-MS performed three analytical runs for all the samples, blanks and standards.

### **3.2 Stable carbon and nitrogen isotope analysis**

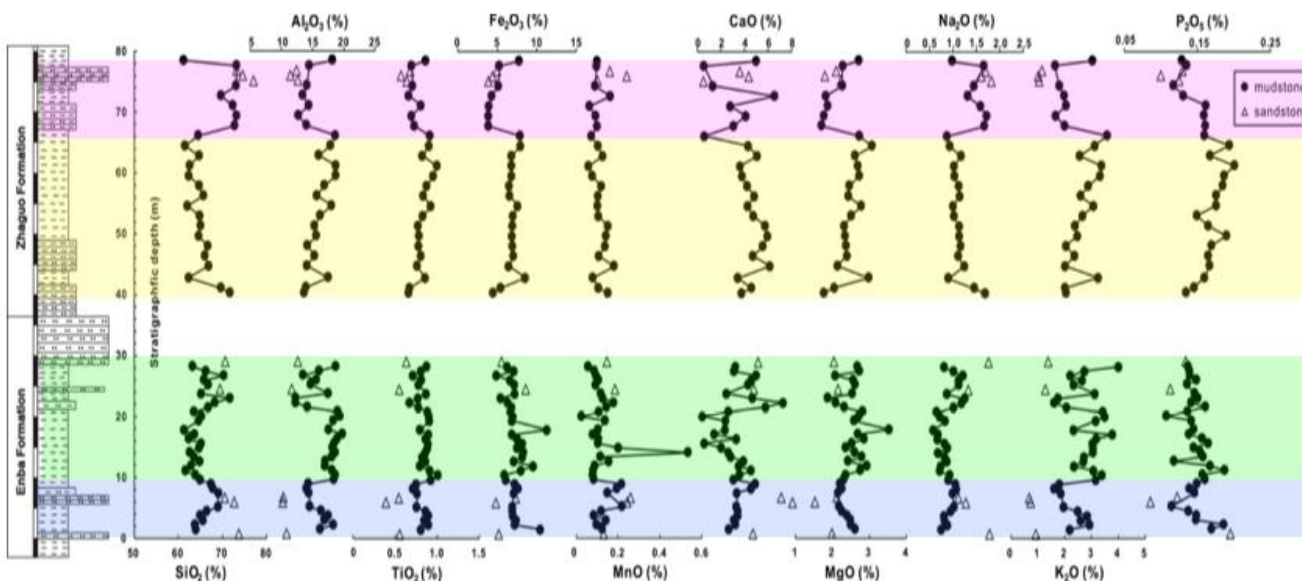
Thirty samples were analyzed for stable carbon and nitrogen isotopes, and samples were first crushed using a SPEX zirconia ceramics ball mill. Powers are then dissolved in 10% HCl and neutralized with deionized distilled water and centrifuged. Solutions were decanted, and powders were dried using the Labconco 4.5 Freezone freeze dryer before folding in a tin boat and loaded in a Thermo Elemental Analyzer coupled to Delta V Advantage IRMS at Montclair State University (USA) to measure stable carbon and nitrogen isotopes. Precision is better than 0.5‰ for carbon, and better than 0.1‰ for nitrogen isotopes.

## **4 Results**

Geochemical data for major, trace elements and REE concentrations, and stable carbon and nitrogen isotopes of the Qumiba section are shown in Table 1, 2, 3, 4 and 5. The major elements and trace elements are interpreted as weathering proxies and ocean redox proxies respectively. Trace elements and REE analysis are useful to reconstruct paleoenvironment and to trace sediment provenance. Based on the variations of major oxides contents, chemical weathering proxies, provenance proxies, redox proxies and nutrient proxies, the stratigraphic profile can be roughly divided into four parts (i.e. Part I: 0-10 m; Part II: 10-30 m; Part III: 40-66 m; Part IV: 66-80m) (Figure 4). In addition, Part II can be further subdivided to Part II-A (10-21m) and Part II-B (21-30m) because of the small variations inside.

#### 4.1 Major element geochemistry

The major oxides are dominated by SiO<sub>2</sub> (61.2 – 77.0%), Al<sub>2</sub>O<sub>3</sub> (10.3 – 19.8%), Fe<sub>2</sub>O<sub>3</sub> (3.9 – 11.3%) and CaO (0.4 – 8.1%) (Figure 4). Other major oxides such as MgO (1.5 – 3.5%), CaO (0.4 – 8.1%), K<sub>2</sub>O (0.7 – 4.0%), MnO (0.0 – 0.5%), Na<sub>2</sub>O (0.58 – 1.81%), TiO<sub>2</sub> (0.39 – 1.01%) and P<sub>2</sub>O<sub>5</sub> (0.10 – 0.19%) exhibit small variations. The mudstone samples commonly show higher Al<sub>2</sub>O<sub>3</sub> contents (12.3 – 19.8%, average 16.3%) and lower SiO<sub>2</sub> contents (61.2 – 73.2%, average 65.9%) than those of the sandstone samples (Al<sub>2</sub>O<sub>3</sub>: 11.5 – 12.7, average 10.3%; SiO<sub>2</sub>: 69.4 – 77.0%, average 72.7%). Both mudstone and sandstone samples show a linear trend on the SiO<sub>2</sub> vs. Al<sub>2</sub>O<sub>3</sub> (Figure 4), indicating they are similar to the siliciclastic rocks. Meanwhile, there is a strong positive correlation between Al<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> ( $R^2 = 0.86$ ;  $p$ -value < 0.001), suggesting there is no significant difference in provenance between the mudstones and sandstones. The stratigraphic variations of major oxides contents of the Qumiba section along with the lithology is shown in Figure 4. The samples of Part I and Part IV display higher SiO<sub>2</sub> contents and lower Al<sub>2</sub>O<sub>3</sub> contents than those of Part II and Part III. In addition, Part II shows small variations inside, in which the samples of Part II-A have lower SiO<sub>2</sub> content and higher Al<sub>2</sub>O<sub>3</sub> content than those of Part II-B.



**Figure 4.** Content of major oxides from the Qumiba section, including  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{TiO}_2$ ,  $\text{Fe}_2\text{O}_3$ ,  $\text{MnO}$ ,  $\text{CaO}$ ,  $\text{MgO}$ ,  $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ , and  $\text{P}_2\text{O}_5$ . Eight sandstone samples are plotted as triangles and 57 mudstone samples are plotted in closed circles.

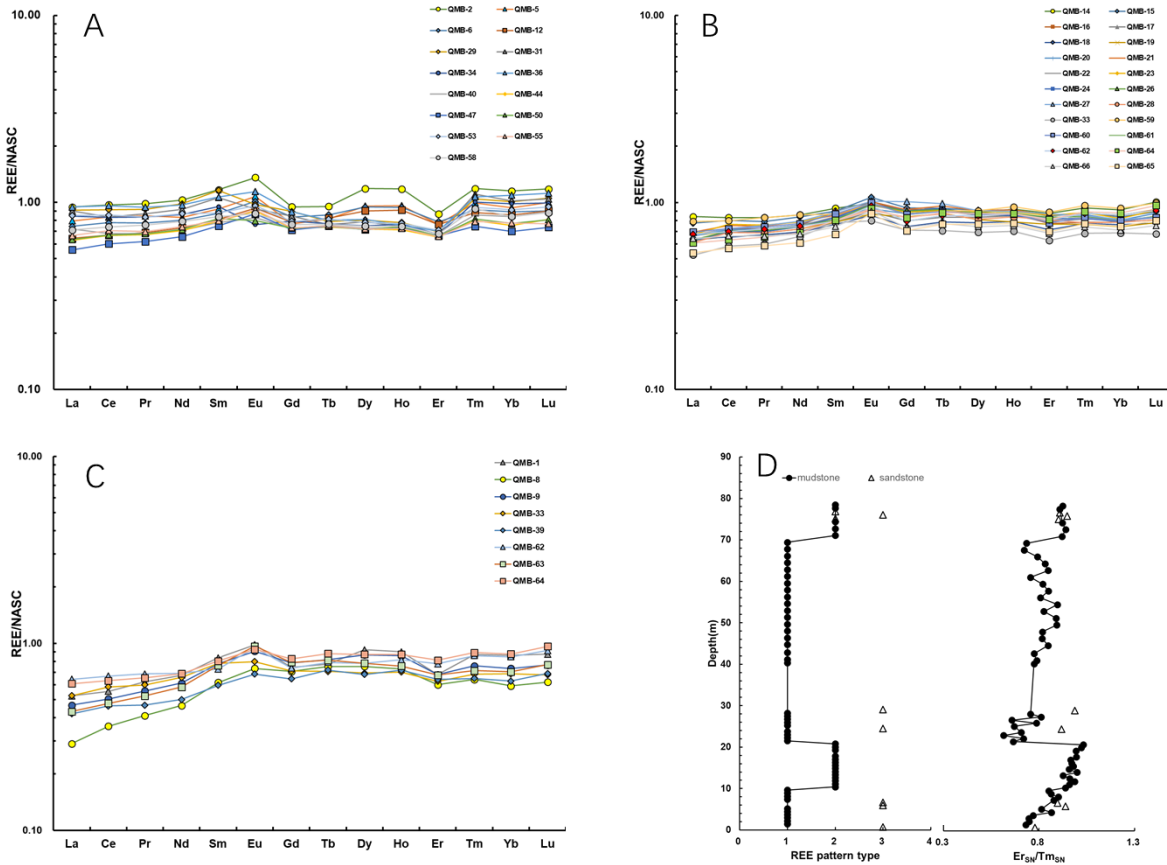
## 4.2 Trace element geochemistry

Trace elements can be used to analyze the material provenance, water column redox and nutrient conditions through the concentrations, enrichment factors and the ratios of different trace elements. In order to interpret and compare the results properly, it is common to use enrichment factors (EF) of trace elements via aluminum (Al) normalization:  $\text{EF}_{\text{element X}} = \text{X}/\text{Al}_{\text{sample}} / \text{X}/\text{Al}_{\text{average shale}}$  (Tribovillard et al., 2006). Europium anomaly ( $\text{Eu}/\text{Eu}^*$ ) (see below),  $\text{Y}/\text{Ho}$ ,  $\text{Th}/\text{Sc}$ ,  $\text{Zr}/\text{Sc}$ ,  $\text{La}/\text{Th}$  vs.  $\text{Hf}$ ,  $\text{La}-\text{Th}-\text{Sc}$ ,  $\text{Th}/\text{Co}$ , and  $\text{La}/\text{Sc}$  can be used as sediment provenance proxies (Figure 11; discussed below) (Floyd and Leveridge, 1987; Gu et al., 2002). In addition, cerium anomaly ( $\text{Ce}/\text{Ce}^*$ ),  $\text{U}/\text{Th}$  ratios, and EF U, are paleoredox indices to help reconstruct the water column oxygen conditions at the study site (Figure 13).  $\text{U}/\text{Th}$  ratios range from 0.15 to 0.29 with an average of 0.17. EF U values range from 0.48 to 1.24 with an average of 0.71. The elements, such as phosphorus (P), barium (Ba), cobalt

(Co), and nickel (Ni), are used as the nutrient proxies to explore the paleoproductivity in the ocean (Anagnostou et al., 2011; Large et al., 2014; Tribovillard et al., 2006). The enrichment factors of the nutrient elements (P, Ba, Co and Ni) show the similar variation trends in the stratigraphic profile of the Qumiba section, in which the values of Part II-B and Part IV are relatively higher than those of other parts.

### **4.3 Rare earth elements geochemistry**

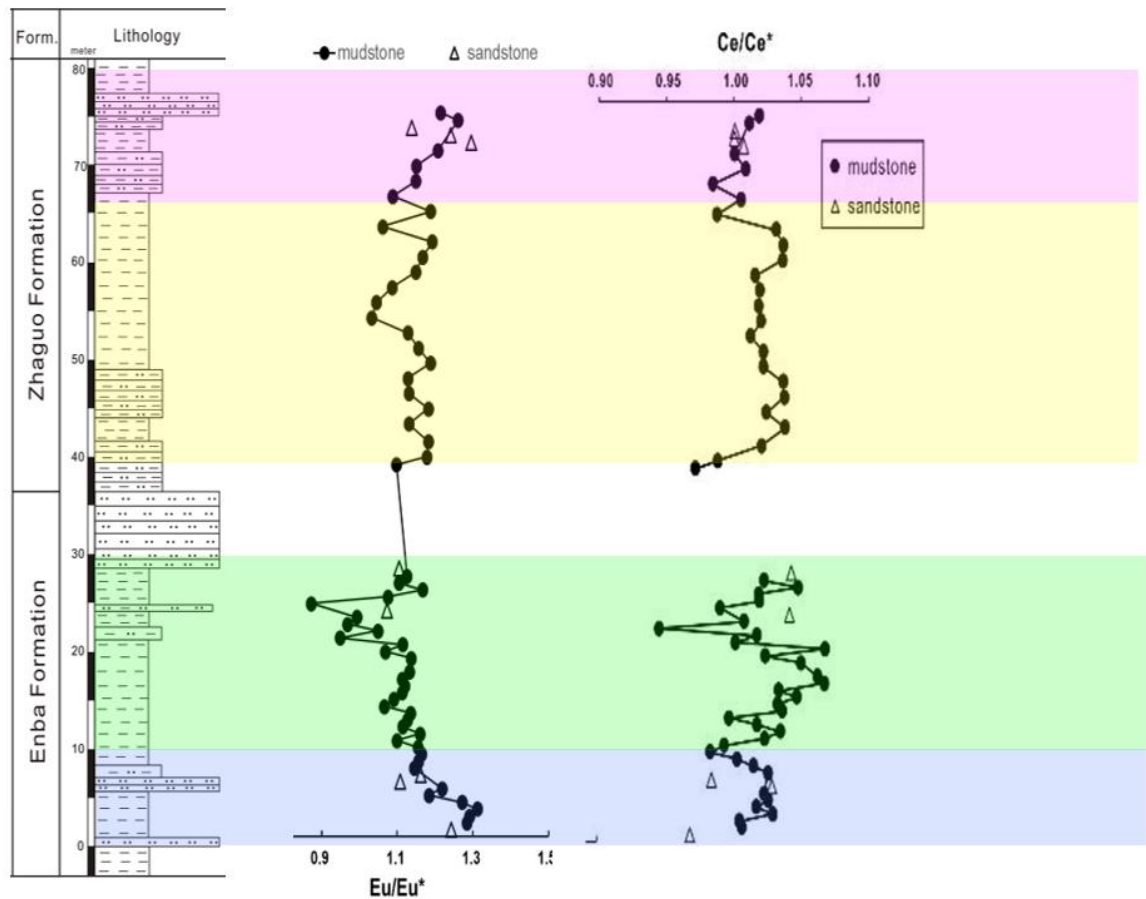
Rare earth elements (REEs) tend to be constant in abundance and steady in properties due to their low mobilities and solubilities during sedimentary process (Cullers et al., 1988; McLennan et al., 1993; McLennan, 1989). Thus, the REE geochemistry can be used to trace the signature of the source materials of sedimentary rocks and speculate changes in tectonics and paleoenvironment (McLennan et al., 1993; McLennan, 1989). All the measured REE concentrations of both mudstone and sandstone samples are normalized to the North American shale composition (NASC) (Gromet et al., 1984). The shale normalized trace elements (X) are presented as  $X_{SN}$ . The traditional formula  $Ce/Ce^* = Ce_{SN}/(0.5La_{SN} + 0.5Pr_{SN})$  and  $Eu/Eu^* = Eu_{SN}/(0.5Sm_{SN} + 0.5Tb_{SN})$  was used to calculate the cerium (Ce) anomalies and europium (Eu) anomalies respectively (Bau and Dulski, 1996). Negative  $Ce/Ce^*$  anomaly suggests low oxygen environment because Ce is depleted in oxic water due to its high redox sensitivity (German and Elderfield, 1990).



**Figure 5. REE patterns of the samples in the Qumiba section. (A) Type 1 REE pattern; (B) Type 2 REE pattern; (C) Type 3 REE pattern; (D) The stratigraphic variations of the REE pattern types and  $Er_{SN}/Tm_{SN}$  ratios**

The mudstone samples generally have higher total contents of REEs ( $\Sigma REE$ ) (126.7 – 214.99 ppm, average 161.37 ppm) than those of sandstone samples (89.11 – 151.08 ppm, average 123.93 ppm). The NASC-normalized REE abundance patterns of sandstone and mudstone samples can be divided into three types, of which mudstone samples display as Type 1 and Type 2 and sandstone samples display as Type 3 (Figure 5). Type 1 and Type 2 mudstones exhibit similar trends with flat REE patterns, except that Type 1 mudstones yield lower  $Er_{SN}/Tm_{SN}$  ratios than those of Type 2 mudstones. Stratigraphically, the Type 1 mudstones mostly correspond to Part I, Part II-B and Part III, while Type 2 mudstones correspond to Part II-A and Part IV. The Type 3 sandstones exhibit a left-inclined REE

pattern with LREE (La, Ce, Pr, Nd, Pm, Sm, Eu, Gd) depleted. The Eu anomalies of all the samples show a small range of ratios from 0.87 to 1.31 with an average value of 1.14. The  $\text{Eu}/\text{Eu}^*$  values decrease steadily from 1.3 to 0.87 in Part I and Part II from 0 to 30 m depth. Above 30 m, the  $\text{Eu}/\text{Eu}^*$  values remain stable with an average of 1.2 in Part III and then increase to 1.3 in Part IV. The Ce anomalies show small fluctuations, ranging from 0.94 to 1.07 with an average value of 1.02. In general, the  $\text{Ce}/\text{Ce}^*$  values of Part II-A and Part III are slightly higher than those of Part I, Part II-B and Part IV (Figure 6).



**Figure 6.  $\text{Eu}/\text{Eu}^*$  and  $\text{Ce}/\text{Ce}^*$  anomalies cross-plot showing variations.**

#### 4.4 Stable carbon and nitrogen isotopes

The stable carbon ( $\delta^{13}\text{C}$ ) and nitrogen isotopes ( $\delta^{15}\text{N}$ ) are useful stratigraphic and paleoenvironmental tools. There are two stable isotopes of carbon in the Earth's crust,  $^{13}\text{C}$  (1.11%) and  $^{12}\text{C}$  (98.89%) (Tyson, 1995). Similarly, there are two stable isotopes of nitrogen,  $^{15}\text{N}$  and  $^{14}\text{N}$  respectively (Talbot, 2002). The relative abundance of the two carbon and nitrogen isotopes in a sedimentary sample can be expressed as  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values in per mil (‰):

$$\delta^{13}\text{C} = [({}^{13}\text{C}/{}^{12}\text{C}_{\text{sample}})/({}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}) - 1] \times 1000 \quad (\text{Eq. 1})$$

$$\delta^{15}\text{N} = [({}^{15}\text{N}/{}^{14}\text{N}_{\text{sample}})/({}^{15}\text{N}/{}^{14}\text{N}_{\text{standard}}) - 1] \times 1000 \quad (\text{Eq. 2})$$

$\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  can be used as indicators of marine productivity and water column redox conditions. Additionally,  $\delta^{13}\text{C}$  gives information about carbon cycles and changes in the global climate.  $\delta^{15}\text{N}$  variations indicate the intensity of denitrification, in which high intensity of denitrification suggests oceanic dysoxic condition.  $\delta^{13}\text{C}_{\text{org}}$  and  $\delta^{15}\text{N}_{\text{org}}$  values are plotted versus sample depth (Figure 7), and low values of  $\delta^{13}\text{C}_{\text{org}}$  may be an indicator for Early Eocene hyperthermal 2 (ETM2 or H1; ca. 53.7 Ma).  $\delta^{13}\text{C}_{\text{org}}$  values range from -25.4‰ to -33.8‰ with an average value of -29.2‰ and  $\delta^{15}\text{N}_{\text{org}}$  values range from 4.6‰ to 7.0‰ with an average of 5.6‰. The most abrupt decrease in  $\delta^{13}\text{C}_{\text{org}}$  occurs at ~15m (-33.8‰), which is considered as a negative carbon isotope excursion (CIE), a possible sign of an Early Eocene hyperthermal (Figure 7).

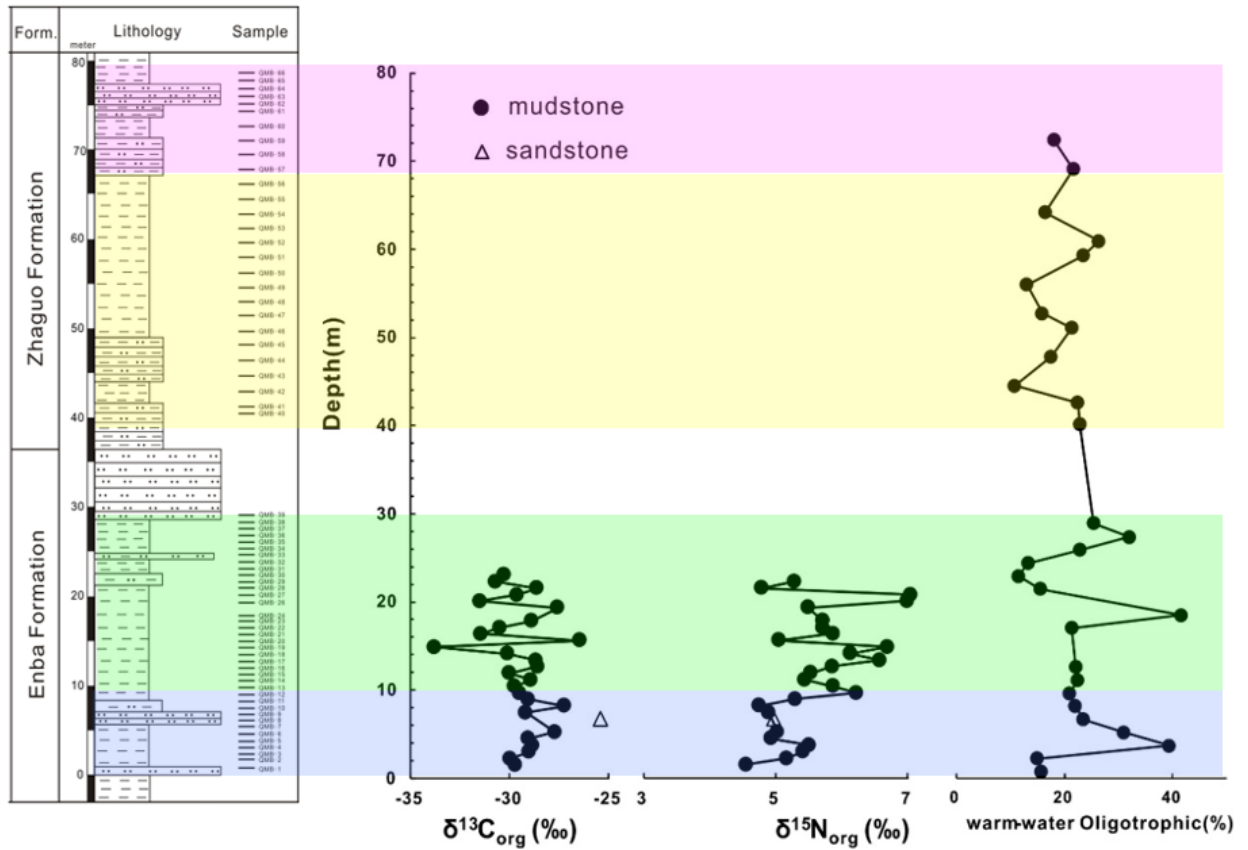


Figure 7. Stable carbon and nitrogen isotope profiles of the Enba Formation at the Qumiba section. Also shown is the warm-water oligotrophic *Neochiastozygus calcareus* nannofossils richness.

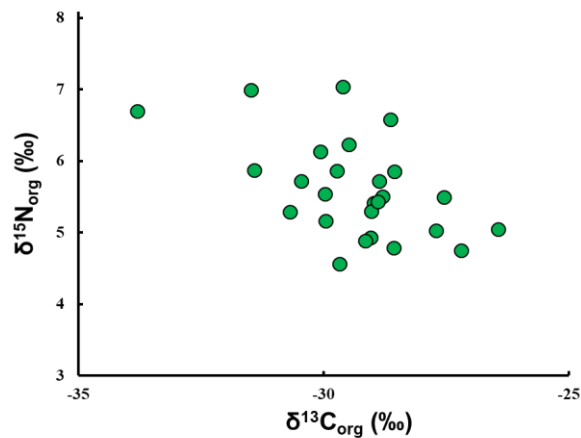


Figure 8. Cross-plot of  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  show statistically significant negative correlation ( $R^2 = 0.29$ ,  $p$  value  $< 0.005$ ), supporting warming ocean led to dysoxia condition.



In contrast,  $\delta^{15}\text{N}_{\text{org}}$  values range from 4.8‰ to 7.0‰ and exhibit three pulses of enrichment at approximately 10 m, 15 m and 20 m in the Enba Formation (Figure 7). In general,  $\delta^{15}\text{N}_{\text{org}}$  values exhibit an upward increasing trend. The highest  $\delta^{15}\text{N}_{\text{org}}$  values correspond to the lowest  $\delta^{13}\text{C}_{\text{org}}$ , which is an indicator of increased  $^{12}\text{C}$  input to the ocean-atmosphere system leading to warmer and more reduced water column conditions. The cross-plot of  $\delta^{13}\text{C}_{\text{org}}$  and  $\delta^{15}\text{N}_{\text{org}}$  from the Enba Formation shows a statistically significant negative correlation (Figure 8).  $\delta^{13}\text{C}_{\text{org}}$  is influenced by primary productivity and  $\delta^{15}\text{N}_{\text{org}}$  is affected by denitrification intensity through the reduction of nitrate by bacteria as a result of organic material input. Peaks of decreasing  $\delta^{13}\text{C}_{\text{org}}$  and increasing  $\delta^{15}\text{N}_{\text{org}}$  can suggest oversaturation of organic matter resulting in excessive nitrate causing restriction of oxygen transport related to dysoxic conditions.

## **5 Discussion**

### **5.1 Provenance proxies**

Several proxies have been suggested to infer the provenance of sedimentary rocks, including europium anomaly ( $\text{Eu}/\text{Eu}^*$ ), Y/Ho ratios, Th/Sc vs. Zr/Sc cross-plot, La/Th vs. Hf cross-plot (Floyd and Leveridge, 1987; Gu et al., 2002), La-Th-Sc and Th-Sc-Zr ternary diagrams (Bhatia and Crook, 1986) and Th/Co vs. La/Sc cross plot (Cullers, 2002). Here we take a multi-proxy approach to reconstruct the provenance of the sedimentary basin of the eastern Tethys.

The Qumiba section formed between 55 and 53 Ma in the Early Eocene in the subducting Paleo-Tethys foreland basin and was sourced mainly from the main mantle thrust (MMT), Kohistan-Ladakh, and main Karakoram thrust (MKT) according to Hu et al. (2012). The

La/Th vs. Hf cross-plot suggests these samples were derived from acidic arc source, with some influence of recycled sediment component (Figure 11), supporting the paleontological evidence of recycled Cretaceous and Paleocene fossils (Jiang et al., 2016). Both Th-Sc-Zr and La-Th-Sc ternary diagrams suggest the mudstone and sandstone samples from the Qumiba section are sourced from continental island arc (Figure 12), likely located in the Lhasa terrane as suggested by Hu et al. (2012).

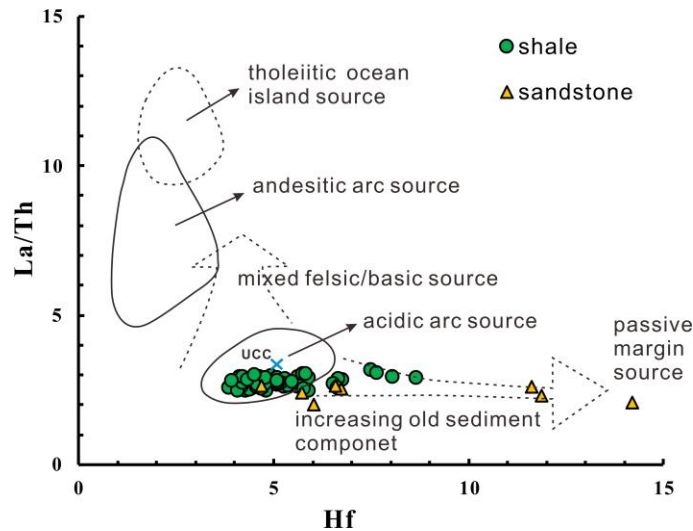


Figure 9. La/Th vs. Hf of both sandstone and mudstone samples modified from Floyd and Leveridge (1987).

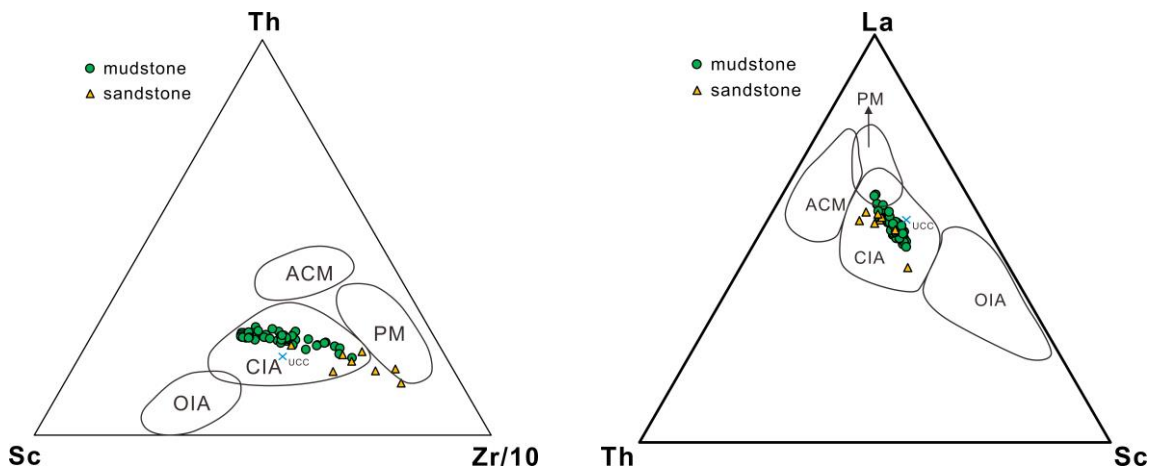
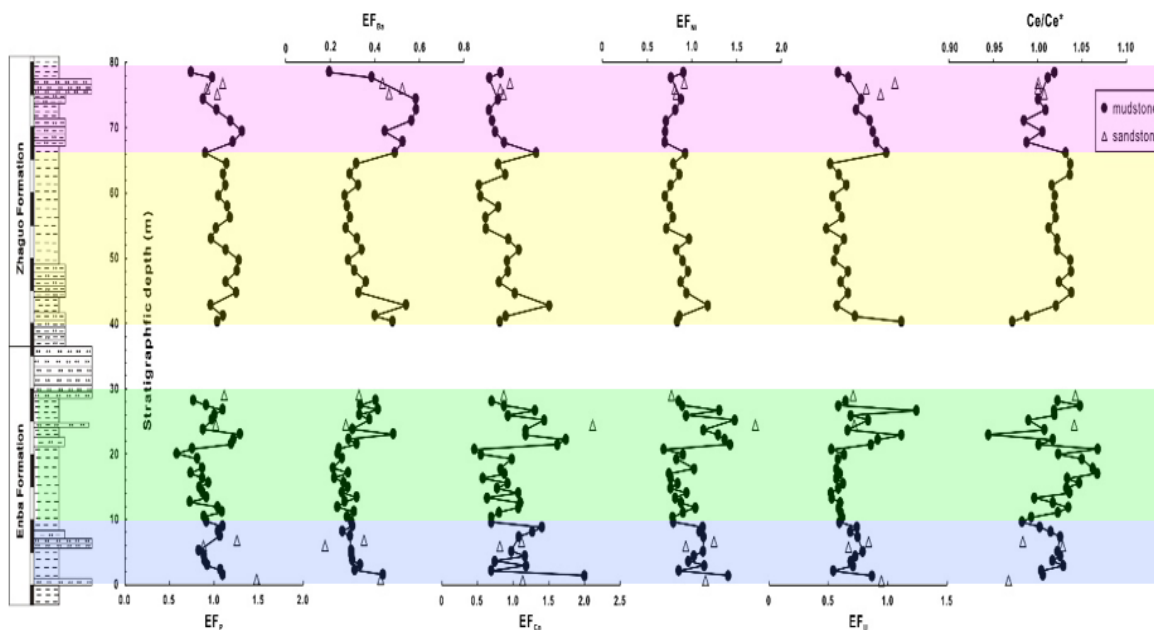


Figure 10. Th-Sc-Zr and La-Th-Sc ternary diagram showing tectonic discrimination (Bhatia and Crook, 1986); PM: passive margin; ACM: active continental margin; CIA: continental island arc; OIA: oceanic island arc.

## 5.2 Reconstructions of the Tethys ocean nutrient conditions

It has been suggested that marine surface productivity changes as a result of glacial-interglacial atmospheric CO<sub>2</sub> variations through the operation of biological pump (Berger et al., 1989; Paytan et al., 1996). The enrichment factor of phosphorus (P), barium (Ba), cobalt (Co), and nickel (Ni) have been suggested to be effective nutrient proxies (Anagnostou et al., 2011; Large et al., 2014; Tribovillard et al., 2006). For example, calibrations of deep-sea coral and sediments suggest that the enrichment of Ba element is a straightforward proxy for seawater dissolved Ba, which is an indicator of ocean paleoproductivity rate (Anagnostou et al., 2011; Dymond et al., 1992; Gingele et al., 1999). To interpret the elemental data as nutrient proxies, we follow Tribovillard et al. (2006) and calculate the enrichment factors (EF) of P, Ba, Co and Ni.  $EF_{\text{element X}} = X/AI_{\text{sample}}/X/AI_{\text{average shale}}$ , with values greater than 1 indicating element X is enriched compared to the average shale, and values less than 1 being depleted. We note that both  $EF_{\text{P}}$  and  $EF_{\text{Ni}}$  show strong enrichment at about 20 m (Part II), coincident with the maximum weathering indices, the lowest  $\delta^{13}\text{C}_{\text{org}}$ , and the maximum  $\delta^{15}\text{N}_{\text{org}}$ , consistent with warming-induced high weathering flux, causing high nutrient and low oxygen condition in the water column.



**Figure 11. Enrichment factor of phosphorous, barium, cobalt, nickel and uranium and Ce anomalies from the Enba and Zhaguo Formation at the Qumiba section.**

### 5.3 Reconstructions of the Tethys ocean redox conditions

Three trace element-based paleo-redox proxies are used in this study:  $Ce/Ce^*$ , and  $EF_U$  (Figure 13) (Tribovillard et al., 2006).  $Ce/Ce^*$  is a widely used paleoredox proxy for shallow water carbonates and mudstones (Murray et al., 1990; Webb and Kamber, 2000). The uranium (U) can always be precipitated by adsorption on or with organic matter in forms of +6 (i.e.  $UO_2^{2+}$ ) in reducing environment, which high U contents indicate less oxic to anoxic conditions (Webb and Kamber, 2000; Tribovillard et al., 2006). The negative anomaly of  $Ce/Ce^*$  and elevated  $EF_U$  both occur at ~20 m (top of Part IIA) and at ~66 m (top of Part III), consistent with increased chemical weathering driving low water column oxygen condition through elevated primary productivity (see nutrient proxies).

#### 5.4 Chemical weathering intensity near the eastern Tethys

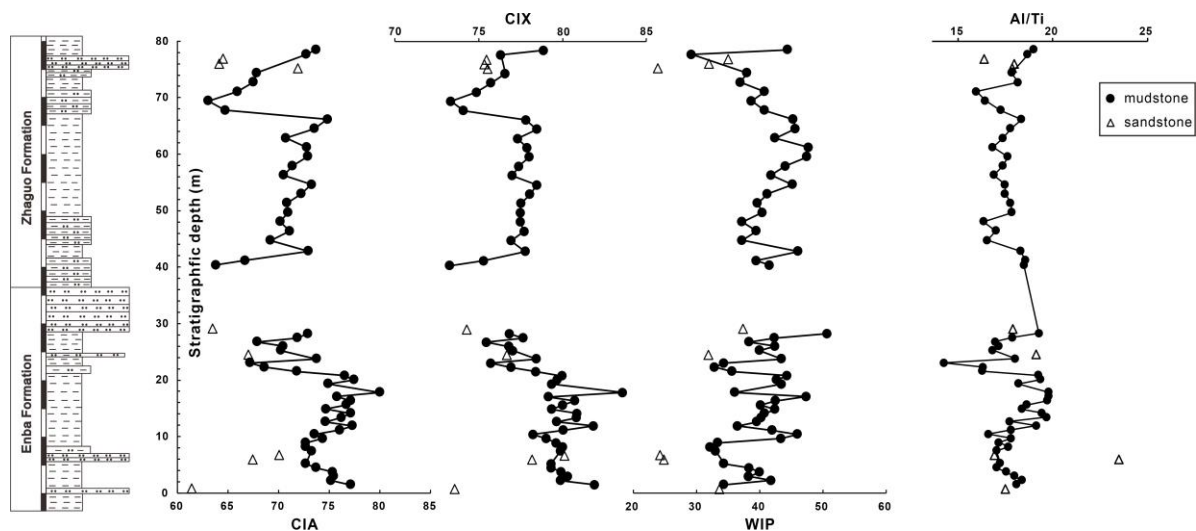
Chemical index of alteration (CIA) is a useful weathering proxy for mudstones generated in glaciated catchments (Nesbitt and Young, 1982; Wang et al., 2019), but controversies exist regarding its effectiveness in a wider range of global drainage basins (Li and Yang, 2010). Other chemical weathering proxies, such as chemical index of alteration without CaO (CIX) (Garzanti et al., 2014; Harnois, 1988), chemical index of weathering (CIW; Harnois, 1988), weathering index of Parker (WIP) (Parker, 1970), and weathering intensity scales (WIS) (Meunier et al., 2013) have also been proposed. Approaches that combine multiple weathering proxies are thought to be more reliable in fluvial drainage basins in China (Shao et al., 2012). We use three weathering proxies, namely CIA, CIX and WIP, to infer the weathering control on the composition of the siliciclastic sedimentary rocks in the Qumiba section.

$$\text{CIA} = \text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{Na}_2\text{O} + \text{K}_2\text{O} + \text{CaO}^*) \times 100 \quad (\text{Eq. 3})$$

$$\text{CIX} = \text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{Na}_2\text{O} + \text{K}_2\text{O}) \times 100 \quad (\text{Eq. 4})$$

$$\text{WIP} = (2\text{Na}_2\text{O} / 0.35 + \text{MgO} / 0.9 + 2\text{K}_2\text{O} / 0.25 + \text{CaO}^* / 0.7) \times 100 \quad (\text{Eq. 5})$$

In order to calculate CIA and CaO\*, the CaO incorporated in the silicate fraction needs to be known. An indirect method to quantify CaO\* has been suggested by McLennan et al. (1993), and reasonable Ca/Na ratios of silicate should be assumed. The molar proportion of P<sub>2</sub>O<sub>5</sub> needs to be subtracted from the molar proportion of total CaO. Then CaO\* can be approximated by phosphate-CaO corrected values and Na<sub>2</sub>O for samples with moles of CaO greater than Na<sub>2</sub>O after phosphate-CaO correction (Fedo et al., 1995; McLennan et al., 1993; Yang et al., 2020). Because CIX does not require the knowledge of CaO\*, we calculated CIX to compare the efficacy of this approach.



**Figure 12. Calculated weathering proxies CIA, CIX and WIP for the mudstone samples from the Enba and Zhaguo Formation at the Qumiba section.**

The major chemical weathering process in the continental upper crust is the formation of clay minerals through the degradation of feldspars, when Ca, Na and K ions are removed from the feldspars (Nesbitt and Young, 1982). In general, the CIA values for shales range around 70 to 75 due to the large amount of clay minerals, and the higher CIA indicates more intense chemical weathering (Bahlburg and Dobrzinski, 2011; Goldberg and Humayun, 2010). There appears to be three intervals where CIA, CIX and WIP show similar increasing trends: 10-20 m in the Enba Formation, ~40 m in the Zhaguo Formation, and 70-80 m in the Zhaguo Formation (Figure 9). These intervals also correspond to increased nutrient levels in the eastern Tethys ocean, as evidenced by the nutrient proxies shown below. The CIX vs. WIP cross-plot (Figure 10), shows the weathering trend of the mudstones from the four parts of the Qumiba section data plotted near the lowerleft corner suggest more sedimentary carbonate recycling.

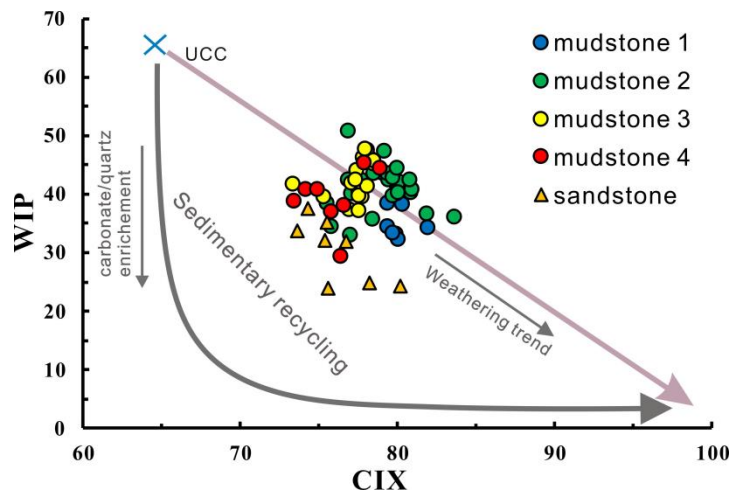


Figure 13. Weathering trend of the mudstones from the four parts of the Qumiba section (modified from Yang et al., 2020).

### 5.5 Carbon and nitrogen cycle during the Early Eocene Tethys

During carbon assimilation, marine phytoplankton preferentially take up  $^{12}\text{C}$  due to the kinetic effect (Galimov, 1985). Marine organic matter is composed of both marine and terrigenous particulate organic sources (Andrews et al., 1998; Dean et al., 1986). Studies have shown that modern terrestrial organic matter exhibits lower  $\delta^{13}\text{C}$  values (average  $-27\text{‰}$ ; Diefendorf et al., 2010) than marine organic matter ( $-22\text{‰}$ ; Hayes et al., 1999) as shown in the Global distribution of modern  $\delta^{13}\text{C}$  value ranges in carbon dioxide, methane, dissolved inorganic and organic carbon (DIC, DOC), particulate organic matter (OM) and carbonates (Figure 14). In the Cretaceous, however, marine organic matter is shown to have much lower values (average  $-28\text{‰}$ ) (Arthur et al., 1985), likely because the  $\text{CO}_2$  levels were much higher than the present day. The variation of marine  $\delta^{13}\text{C}_{\text{org}}$  is often interpreted as the relative proportions of marine versus terrestrial sources of the sedimentary organic matter (Hedges

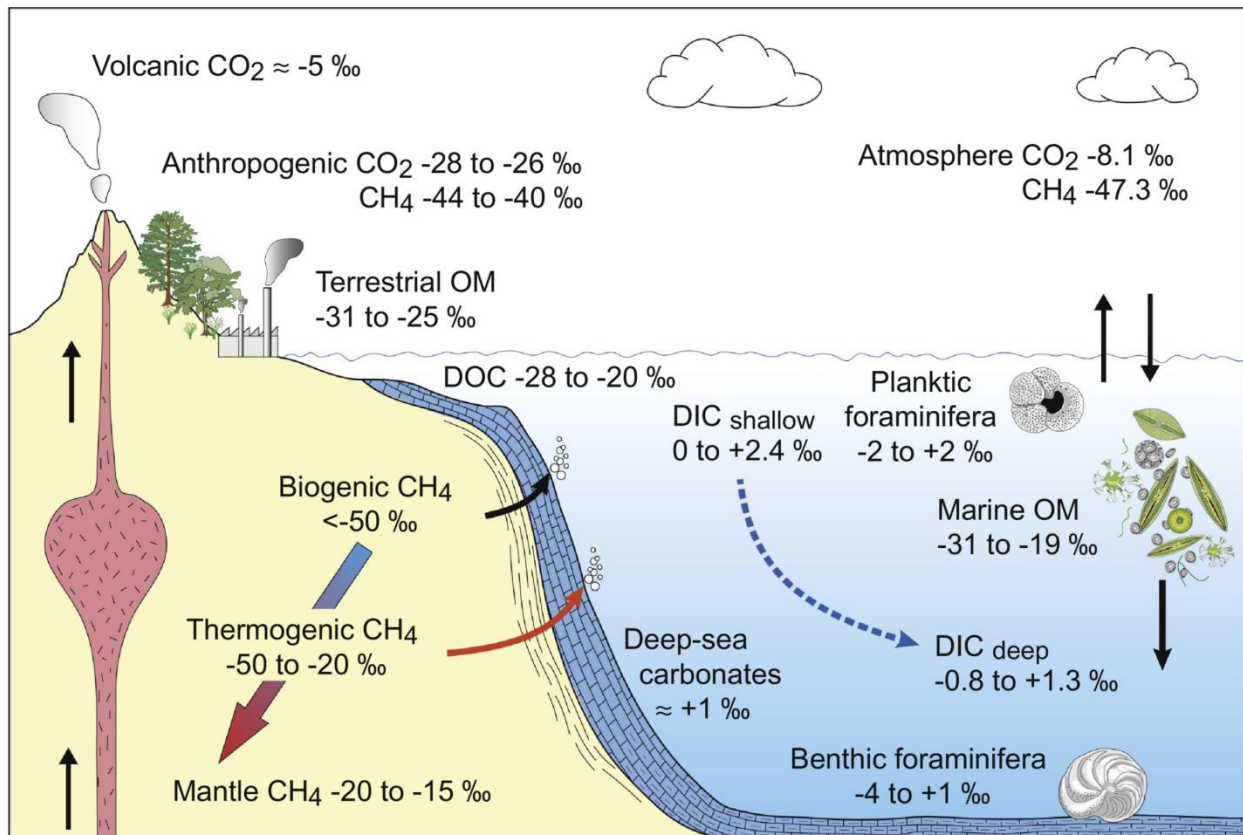
and Parker, 1976; Sackett, 1964). They can also be used as global stratigraphic correlation tools (chemostratigraphy) and paleoenvironmental tools (Ando et al., 2002; Erbacher et al., 2005; Mackensen and Schmiedl, 2019; Popp et al., 1997; Ruhl et al., 2009).

The Early Eocene eastern Tethys receives two primary sources: recycled organic matter from continental weathering and marine organic matter. The  $\delta^{15}\text{N}_{\text{org}}$  values ranging from -5‰ to 18‰ indicate terrigenous organic input (Peters et al., 1978; Schoeninger and DeNiro, 1984). The 4.6 to 7.0‰  $\delta^{15}\text{N}_{\text{org}}$  values in the Qumiba samples suggest terrigenous organic matter influence. Additionally, the negative correlation between  $\delta^{13}\text{C}_{\text{org}}$  and  $\delta^{15}\text{N}_{\text{org}}$  suggest increasing influence of nitrate reduction (Jacot Des Combes et al., 2008). To understand the relative contribution of recycled organic matter and marine organic matter, a simple two end-member mixing equation can be used, in which we assign mean  $\delta^{13}\text{C}$  values to recycled Cretaceous and Paleogene organic matter ( $\delta^{13}\text{C}_{\text{recycled}}$ ) and coeval marine organic matter ( $\delta^{13}\text{C}_{\text{MOM}}$ ):

$$F_{\text{recycled}} (\%) = (\delta^{13}\text{C}_{\text{org}} - \delta^{13}\text{C}_{\text{MOM}}) / (\delta^{13}\text{C}_{\text{recycled}} - \delta^{13}\text{C}_{\text{MOM}}) \times 100 \quad (\text{Eq. 6})$$

$\delta^{13}\text{C}_{\text{org}}$  is the measured carbon isotope value from the Qumiba section,  $\delta^{13}\text{C}_{\text{MOM}}$  is the carbon isotope value of coeval marine organic matter,  $\delta^{13}\text{C}_{\text{recycled}}$  is the recycled Cretaceous and Paleogene marine organic matter.





**Figure 14. Global distribution of modern  $\delta^{13}\text{C}$  value ranges in carbon dioxide, methane, dissolved inorganic and organic carbon (DIC, DOC), particulate organic matter (OM) and carbonates based on Mackensen and Schmiedl (2019).**

Compared to marine  $\delta^{13}\text{C}$ , nitrogen isotopes are used less frequently as stratigraphic correlation tool, but widely used as paleoenvironmental proxies (Altabet, 2006; Kashiyama et al., 2008; Rau et al., 1987; Sigman et al., 2009). Similar to carbon isotope fractionation, primary producers preferentially use  $^{14}\text{N}$  relative to  $^{15}\text{N}$  in nitrate during nitrate assimilation because of kinetic effects (Montoya, 1994; Waser et al., 1998). The degree of nitrogen fractionation depends on the amount of available nitrate and the rate of nitrate consumption (Deutsch et al., 2004; Sigman and Haug, 2003). The largest nitrogen fractionation incurs during denitrification (the microbial process of nitrate reduction and the ultimate production of  $\text{N}_2$ ) (Dähnke and Thamdrup, 2013; Kritee et al., 2012) as showing in the Global marine nitrogen cycle (Figure 15). Foraminifera bound nitrogen isotopes show that the early

Cenozoic is characterized by elevated  $\delta^{15}\text{N}$  values, an indicator of intensified denitrification facilitated by reduced oxygen concentration (Kast et al., 2019). Elevated  $\delta^{15}\text{N}$  values at the lower Enba Formation reflect stronger denitrification in the Qumiba section, supporting an episodic water column oxygen reduction.

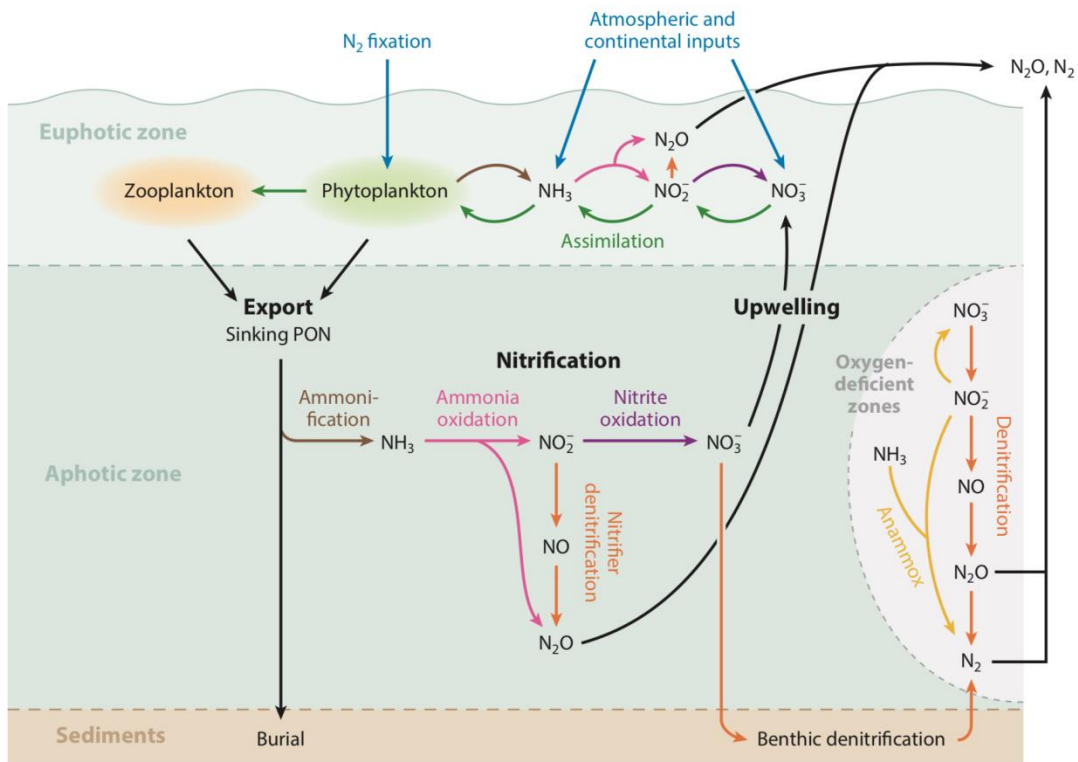


Figure 15. Global marine nitrogen cycle based on Casciotti et al. (2016).

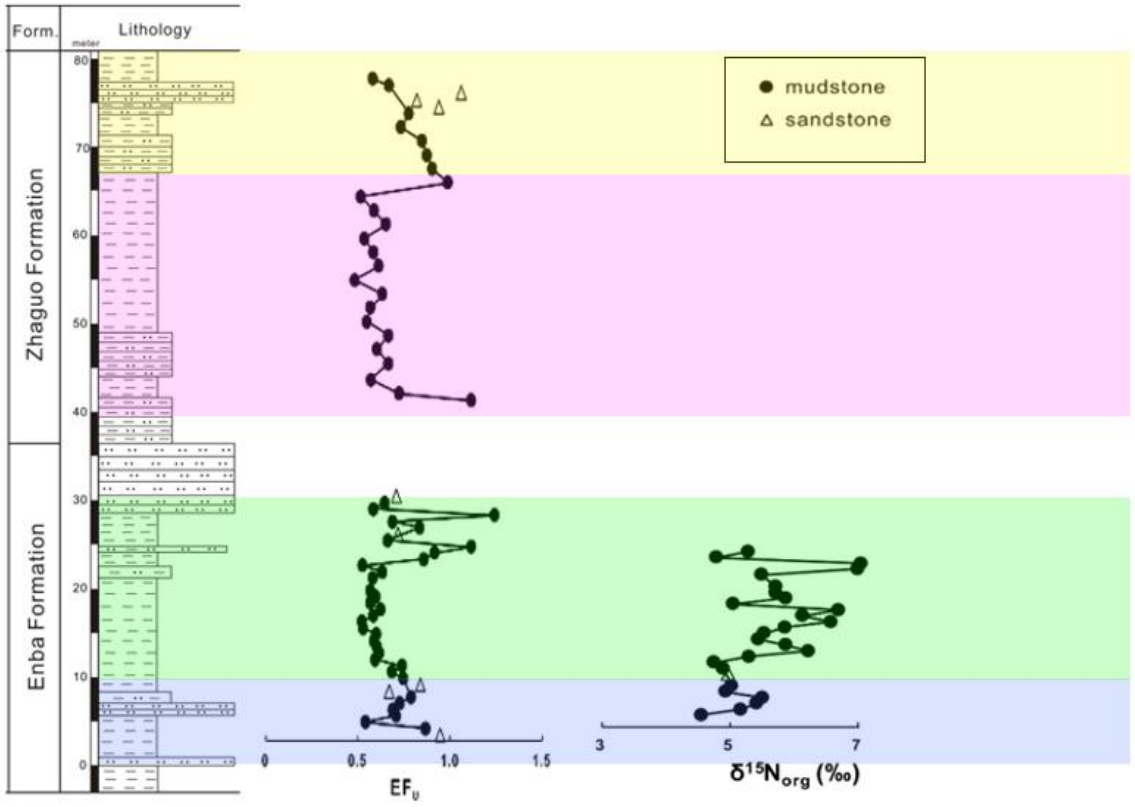


Figure 16.  $EF_u$ , and  $\delta^{15}N_{org}$  showing low oxygen, and denitrification for upper part of ~0 - 30m section.

## **6 Conclusion**

The initial stage of the India-Asia collision occurred during the early Cenozoic ice-free greenhouse world, which was punctuated by a series of transient global warming events, likely driven by volcanic eruptions and orbital forcings. To understand the paleoenvironmental conditions of the Early Eocene eastern Tethys ocean before its final closure, we study the youngest marine successions from the southern Tibet in their geochemical characteristics. We report new major, trace element geochemistry and stable carbon and nitrogen isotope data of the Early Eocene strata deposited in a shallow marine foreland basin prior to the main stage of India-Asian continent collision at Qumiba section in the Tingri County, southern Tibet. Major element geochemistry and weathering indices suggest that the Eocene southeastern Tethys may have experienced episodic input of nutrient-rich weathering supplies and the development of hypoxia in the water column, consistent with the warm and humid climate in the Early Eocene (Figure 16).

Table 1. Whole-rock compositions of the Qumiba section in the Tingri County, southern Tibet

Depth (m)	Sample	SiO <sub>2</sub>	TiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MnO	MgO	CaO	Na <sub>2</sub> O	K <sub>2</sub> O	P <sub>2</sub> O <sub>5</sub>	Total
0.74	QMB-1	73.69	0.56	10.86	5.23	0.13	1.98	4.72	1.77	0.91	0.15	100.00
1.49	QMB-2	64.00	0.81	16.23	10.45	0.12	2.61	2.70	0.75	2.17	0.17	100.00
2.23	QMB-3	63.73	0.90	18.32	7.24	0.10	2.48	3.24	0.88	2.92	0.19	100.00
2.97	QMB-4	65.53	0.85	16.93	7.40	0.14	2.47	3.12	0.83	2.58	0.15	100.00
3.71	QMB-5	64.86	0.90	17.55	6.97	0.08	2.39	3.45	0.82	2.83	0.15	100.00
4.46	QMB-6	66.42	0.86	16.38	6.93	0.12	2.25	3.46	0.96	2.49	0.14	100.00
5.20	QMB-7	68.95	0.76	14.56	6.90	0.22	2.17	3.35	1.02	1.96	0.11	100.00
5.94	QMB-8	72.62	0.39	10.25	4.82	0.24	1.52	8.09	1.27	0.71	0.09	100.00
6.68	QMB-9	70.46	0.54	10.25	7.35	0.26	2.10	7.13	1.11	0.66	0.12	100.00
7.43	QMB-10	69.05	0.76	14.43	7.10	0.15	2.14	3.38	0.99	1.85	0.15	100.00
8.17	QMB-11	67.79	0.71	14.01	7.65	0.20	2.25	4.58	1.07	1.60	0.14	100.00
8.91	QMB-12	67.35	0.75	14.35	7.21	0.22	2.19	4.94	1.05	1.80	0.15	100.00
9.65	QMB-13	64.89	0.93	18.41	6.08	0.08	2.30	3.10	0.90	3.15	0.16	100.00
10.40	QMB-14	63.87	1.01	18.68	5.92	0.08	2.35	3.61	0.93	3.38	0.16	100.00
11.14	QMB-15	61.59	0.91	18.12	7.99	0.08	2.74	4.58	0.73	3.07	0.19	100.00
11.88	QMB-16	62.91	0.80	16.94	9.58	0.09	2.92	3.51	0.74	2.34	0.17	100.00
12.62	QMB-17	64.85	0.86	17.04	7.08	0.16	2.42	3.91	0.88	2.69	0.12	100.00
13.37	QMB-18	63.49	0.83	18.09	8.17	0.11	2.78	2.84	0.82	2.71	0.16	100.00
14.11	QMB-19	62.57	0.86	18.64	8.28	0.54	2.59	2.65	0.68	3.04	0.15	100.00
14.85	QMB-20	64.68	0.89	18.22	7.53	0.20	2.35	2.06	0.88	3.04	0.15	100.00
15.60	QMB-21	65.14	0.89	18.54	8.18	0.10	2.52	0.61	0.82	3.03	0.16	100.00
16.34	QMB-22	62.35	0.87	18.99	7.53	0.11	2.85	3.31	0.67	3.16	0.16	100.00
17.08	QMB-23	63.60	0.90	19.79	6.90	0.08	2.68	1.46	0.69	3.77	0.14	100.00
17.82	QMB-24	61.31	0.80	17.60	11.27	0.10	3.52	2.36	0.58	2.32	0.15	100.00
19.31	QMB-26	64.57	0.91	18.38	6.90	0.14	2.58	2.39	0.84	3.15	0.14	100.00
20.05	QMB-27	65.24	0.90	19.42	6.97	0.03	2.71	0.44	0.72	3.48	0.11	100.00
20.79	QMB-28	63.57	0.89	19.05	6.72	0.11	2.81	2.64	0.66	3.41	0.14	100.00
21.54	QMB-29	66.70	0.78	14.18	6.82	0.14	2.31	5.82	1.02	2.07	0.16	100.00
22.28	QMB-30	68.27	0.68	12.26	6.34	0.18	2.07	7.29	1.18	1.59	0.14	100.00
23.02	QMB-31	71.63	0.78	12.30	5.46	0.13	1.87	4.69	1.25	1.74	0.15	100.00
23.76	QMB-32	65.09	0.87	17.52	7.22	0.12	2.53	2.51	0.87	3.12	0.15	100.00
24.51	QMB-33	69.37	0.55	11.65	8.60	0.19	2.14	4.81	1.33	1.26	0.11	100.00
25.25	QMB-34	66.77	0.78	14.68	7.17	0.09	2.61	4.31	1.13	2.33	0.14	100.00
25.99	QMB-35	65.76	0.82	15.58	6.64	0.11	2.56	4.62	1.12	2.64	0.15	100.00
26.74	QMB-36	70.21	0.71	13.45	4.91	0.10	2.06	5.02	1.21	2.20	0.14	100.00
27.48	QMB-37	66.22	0.81	16.05	7.10	0.09	2.72	3.14	1.01	2.72	0.14	100.00
28.22	QMB-38	63.15	0.87	18.76	6.27	0.06	2.67	3.27	0.81	4.00	0.14	100.00
29.04	QMB-39	70.62	0.63	12.62	5.54	0.15	2.03	5.16	1.75	1.37	0.13	100.00
40.33	QMB-40	71.67	0.66	13.67	4.48	0.15	1.75	3.75	1.69	2.04	0.13	100.00
41.16	QMB-41	69.63	0.67	13.91	5.44	0.11	2.04	4.59	1.45	2.01	0.15	100.00
42.80	QMB-42	62.32	0.86	17.48	8.55	0.08	2.97	3.45	0.91	3.24	0.16	100.00
44.74	QMB-43	66.79	0.76	14.06	6.47	0.18	2.13	6.19	1.24	2.01	0.17	100.00
46.39	QMB-44	65.98	0.81	15.31	7.03	0.11	2.38	4.73	1.11	2.37	0.16	100.00
48.03	QMB-45	66.67	0.78	14.17	6.93	0.14	2.36	5.59	1.16	2.04	0.17	100.00
49.68	QMB-46	64.54	0.78	15.60	6.86	0.14	2.33	5.96	1.13	2.46	0.19	100.00
51.33	QMB-47	65.04	0.77	15.32	6.93	0.15	2.31	5.78	1.13	2.38	0.17	100.00
52.97	QMB-48	64.76	0.83	16.21	6.99	0.11	2.50	4.77	1.02	2.66	0.15	100.00
54.62	QMB-49	62.02	0.93	18.06	7.59	0.10	2.77	4.29	1.00	3.06	0.18	100.00
56.26	QMB-50	65.61	0.83	15.67	6.60	0.10	2.42	4.86	1.14	2.58	0.18	100.00
57.91	QMB-51	64.73	0.88	16.93	4.49	0.12	2.45	4.23	1.12	2.87	0.19	100.00
59.55	QMB-52	62.35	0.95	18.75	6.83	0.08	2.71	3.80	1.04	3.31	0.19	100.00
61.19	QMB-53	62.51	1.00	18.73	6.79	0.06	2.68	3.65	1.02	3.36	0.20	100.00
62.84	QMB-54	64.67	0.83	16.01	6.79	0.13	2.61	5.07	1.17	2.55	0.17	100.00
64.49	QMB-55	61.51	0.90	17.88	7.97	0.10	3.06	4.34	0.93	3.11	0.19	100.00
66.14	QMB-56	64.53	0.91	18.66	7.90	0.07	2.72	0.59	0.88	3.58	0.16	100.00
67.78	QMB-57	72.69	0.73	14.00	3.88	0.10	1.69	3.09	1.67	1.99	0.16	100.00
69.43	QMB-58	73.19	0.69	12.72	3.90	0.09	1.76	4.12	1.72	1.66	0.16	100.00
71.07	QMB-59	72.30	0.81	14.37	3.98	0.06	1.87	2.82	1.58	2.05	0.16	100.00
72.72	QMB-60	69.65	0.66	13.42	4.31	0.16	1.82	6.56	1.32	1.97	0.13	100.00
74.36	QMB-61	73.05	0.71	14.09	5.12	0.09	2.25	1.35	1.44	1.79	0.12	100.00
75.19	QMB-62	76.96	0.64	12.68	3.91	0.10	2.22	0.52	1.81	1.04	0.13	100.00
76.01	QMB-63	74.50	0.57	11.42	4.40	0.24	1.78	4.38	1.61	1.00	0.10	100.00
76.83	QMB-64	73.11	0.68	12.40	4.96	0.16	2.10	3.62	1.70	1.15	0.13	100.00
77.65	QMB-65	73.18	0.70	14.50	5.26	0.10	2.27	0.57	1.66	1.63	0.14	100.00
78.48	QMB-66	61.20	0.86	18.23	7.75	0.10	2.70	5.02	0.98	3.02	0.13	100.00

\*Major elements expressed in weight percent % (wt. %)



Table 3. Rare Earth Elements REE of the Qumiba section

Depth (m)	Sample	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Y
0.74	QMB-1	0.52	0.55	0.62	0.67	0.84	0.98	0.74	0.77	0.92	0.90	0.68	0.87	0.86	0.88	1.02
1.49	QMB-2	0.94	0.97	0.99	1.03	1.17	1.36	0.95	0.95	1.19	1.18	0.87	1.19	1.15	1.18	1.34
2.23	QMB-3	0.68	0.70	0.72	0.74	0.77	0.89	0.61	0.67	0.82	0.81	0.63	0.84	0.81	0.86	0.92
2.97	QMB-4	0.81	0.85	0.85	0.87	0.91	1.07	0.73	0.75	0.97	0.96	0.70	0.94	0.94	1.00	1.06
3.71	QMB-5	0.80	0.84	0.85	0.83	0.92	1.08	0.77	0.82	0.96	0.96	0.76	1.00	0.94	1.00	1.08
4.46	QMB-6	0.75	0.78	0.78	0.80	0.87	1.01	0.84	0.86	0.94	0.94	0.79	0.91	0.89	0.90	1.07
5.20	QMB-7	0.72	0.75	0.75	0.78	0.85	0.98	0.76	0.81	0.93	0.95	0.74	0.91	0.90	0.93	1.05
5.94	QMB-8	0.29	0.36	0.41	0.46	0.62	0.73	0.71	0.75	0.75	0.73	0.60	0.64	0.59	0.62	0.85
6.68	QMB-9	0.47	0.50	0.56	0.61	0.78	0.91	0.79	0.82	0.87	0.86	0.68	0.76	0.74	0.77	0.97
7.43	QMB-10	0.69	0.73	0.73	0.77	0.85	0.95	0.81	0.85	0.90	0.91	0.75	0.85	0.82	0.92	1.02
8.17	QMB-11	0.55	0.60	0.62	0.66	0.73	0.86	0.75	0.81	0.86	0.85	0.72	0.80	0.75	0.86	0.98
8.91	QMB-12	0.64	0.67	0.69	0.73	0.81	0.93	0.79	0.83	0.90	0.91	0.76	0.88	0.86	0.90	1.03
9.65	QMB-13	0.82	0.78	0.77	0.78	0.80	0.91	0.77	0.81	0.85	0.87	0.78	0.91	0.82	0.90	0.97
10.40	QMB-14	0.84	0.83	0.83	0.86	0.93	0.99	0.87	0.90	0.88	0.92	0.88	0.93	0.91	1.01	1.02
11.14	QMB-15	0.77	0.80	0.79	0.84	0.90	1.07	0.93	0.93	0.91	0.91	0.85	0.88	0.84	0.89	1.01
11.88	QMB-16	0.67	0.72	0.72	0.76	0.90	1.01	0.91	0.92	0.86	0.87	0.77	0.78	0.78	0.85	0.92
12.62	QMB-17	0.63	0.66	0.67	0.70	0.79	0.89	0.80	0.83	0.83	0.85	0.79	0.82	0.80	0.84	0.94
13.37	QMB-18	0.64	0.65	0.67	0.69	0.77	0.86	0.74	0.79	0.78	0.79	0.72	0.77	0.76	0.78	0.88
14.11	QMB-19	0.67	0.70	0.69	0.72	0.79	0.87	0.84	0.88	0.81	0.79	0.76	0.77	0.74	0.78	0.87
14.85	QMB-20	0.66	0.71	0.72	0.76	0.83	0.93	0.87	0.90	0.86	0.87	0.83	0.86	0.80	0.87	0.92
15.60	QMB-21	0.70	0.76	0.75	0.79	0.85	0.98	0.92	0.96	0.90	0.91	0.86	0.88	0.85	0.92	0.97
16.34	QMB-22	0.62	0.69	0.72	0.75	0.84	0.97	0.89	0.91	0.85	0.86	0.80	0.83	0.78	0.81	0.91
17.08	QMB-23	0.68	0.76	0.76	0.79	0.87	0.98	0.89	0.91	0.89	0.89	0.85	0.88	0.82	0.91	0.96
17.82	QMB-24	0.62	0.70	0.70	0.74	0.83	0.98	0.91	0.92	0.87	0.85	0.82	0.82	0.79	0.84	0.94
19.31	QMB-26	0.63	0.70	0.71	0.74	0.81	0.97	0.90	0.92	0.88	0.87	0.83	0.84	0.80	0.84	0.94
20.05	QMB-27	0.66	0.72	0.75	0.79	0.89	1.02	1.01	0.99	0.91	0.90	0.87	0.85	0.81	0.83	0.98
20.79	QMB-28	0.61	0.69	0.69	0.72	0.80	0.91	0.83	0.87	0.81	0.84	0.81	0.78	0.78	0.81	0.93
21.54	QMB-29	0.91	0.92	0.92	0.98	1.16	0.97	0.89	0.79	0.81	0.78	0.70	1.05	1.01	1.05	1.22
22.28	QMB-30	0.67	0.72	0.75	0.82	0.95	0.92	0.81	0.75	0.76	0.74	0.67	0.93	0.91	0.93	1.14
23.02	QMB-31	0.90	0.83	0.87	0.92	1.06	0.92	0.84	0.76	0.79	0.75	0.68	1.11	1.02	1.04	1.21
23.76	QMB-32	0.88	0.89	0.89	0.95	1.01	0.91	0.83	0.78	0.77	0.76	0.68	0.96	0.95	0.94	1.05
24.51	QMB-33	0.52	0.58	0.60	0.66	0.78	0.80	0.71	0.71	0.69	0.70	0.63	0.68	0.69	0.68	0.86
25.25	QMB-34	0.85	0.83	0.83	0.86	0.95	0.76	0.80	0.77	0.75	0.77	0.67	1.01	0.99	0.99	1.06
25.99	QMB-35	0.73	0.77	0.78	0.80	0.89	0.90	0.79	0.79	0.77	0.77	0.70	0.89	0.88	0.88	0.97
26.74	QMB-36	0.95	0.96	0.94	0.97	1.14	0.89	0.80	0.81	0.75	0.70	1.07	1.09	1.12	1.20	1.20
27.48	QMB-37	0.69	0.74	0.72	0.77	0.83	0.87	0.74	0.77	0.74	0.74	0.68	0.83	0.83	0.84	0.92
28.22	QMB-38	0.73	0.74	0.72	0.74	0.80	0.86	0.74	0.77	0.74	0.73	0.67	0.89	0.89	0.92	0.94
29.04	QMB-39	0.42	0.46	0.47	0.50	0.60	0.69	0.65	0.72	0.68	0.72	0.64	0.65	0.63	0.69	0.76
40.33	QMB-40	0.72	0.68	0.68	0.71	0.78	0.83	0.73	0.75	0.73	0.73	0.66	0.86	0.84	0.89	0.90
41.16	QMB-41	0.64	0.64	0.66	0.70	0.77	0.88	0.72	0.71	0.70	0.69	0.63	0.81	0.80	0.82	0.90
42.80	QMB-42	0.69	0.70	0.69	0.72	0.78	0.89	0.73	0.75	0.73	0.73	0.67	0.86	0.81	0.83	0.90
44.74	QMB-43	0.58	0.64	0.65	0.68	0.81	0.88	0.74	0.75	0.73	0.73	0.67	0.79	0.77	0.78	0.89
46.39	QMB-44	0.63	0.67	0.67	0.70	0.80	0.90	0.73	0.74	0.72	0.71	0.65	0.80	0.75	0.81	0.88
48.03	QMB-45	0.61	0.67	0.68	0.71	0.82	0.88	0.75	0.75	0.73	0.73	0.66	0.81	0.77	0.79	0.91
49.68	QMB-46	0.56	0.62	0.64	0.66	0.78	0.84	0.71	0.74	0.71	0.72	0.66	0.74	0.70	0.75	0.83
51.33	QMB-47	0.56	0.60	0.62	0.66	0.75	0.87	0.71	0.75	0.72	0.73	0.66	0.75	0.70	0.74	0.85
52.97	QMB-48	0.66	0.70	0.71	0.76	0.83	0.91	0.75	0.76	0.74	0.74	0.67	0.82	0.83	0.84	0.92
54.62	QMB-49	0.63	0.64	0.63	0.68	0.71	0.79	0.70	0.75	0.71	0.73	0.66	0.73	0.69	0.72	0.80
56.26	QMB-50	0.63	0.67	0.68	0.72	0.82	0.80	0.73	0.75	0.72	0.74	0.66	0.82	0.77	0.81	0.84
57.91	QMB-51	0.66	0.68	0.68	0.70	0.77	0.77	0.70	0.75	0.71	0.73	0.66	0.78	0.71	0.75	0.82
59.55	QMB-52	0.66	0.68	0.67	0.70	0.78	0.81	0.70	0.75	0.71	0.73	0.66	0.81	0.74	0.81	0.80
61.19	QMB-53	0.85	0.86	0.84	0.87	0.88	0.96	0.79	0.81	0.78	0.77	0.71	0.94	0.91	0.95	1.00
62.84	QMB-54	0.62	0.66	0.65	0.69	0.79	0.88	0.72	0.75	0.72	0.73	0.66	0.78	0.79	0.81	0.89
64.49	QMB-55	0.67	0.71	0.70	0.74	0.80	0.91	0.73	0.75	0.72	0.73	0.66	0.79	0.78	0.77	0.89
66.14	QMB-56	0.77	0.79	0.76	0.83	0.87	0.86	0.76	0.77	0.74	0.75	0.67	0.85	0.80	0.83	0.90
67.78	QMB-57	0.84	0.81	0.81	0.83	0.80	0.93	0.77	0.78	0.76	0.74	0.69	0.95	0.89	0.96	1.05
69.43	QMB-58	0.71	0.74	0.76	0.79	0.84	0.87	0.76	0.77	0.75	0.74	0.68	0.93	0.84	0.88	1.09
71.07	QMB-59	0.79	0.80	0.83	0.86	0.86	0.99	0.86	0.90	0.90	0.95	0.89	0.96	0.93	0.99	1.08
72.72	QMB-60	0.70	0.72	0.74	0.77	0.87	1.00	0.86	0.90	0.87	0.87	0.78	0.83	0.80	0.83	1.01
74.36	QMB-61	0.68	0.70	0.72	0.75	0.80	0.96	0.79	0.85	0.85	0.87	0.81	0.87	0.86	0.91	0.99
75.19	QMB-62	0.64	0.67	0.69	0.69	0.73	0.95	0.74	0.79	0.78	0.82	0.78	0.86	0.85	0.91	1.04
76.01	QMB-63	0.43	0.48	0.52	0.58	0.76	0.97	0.79	0.81	0.78	0.76	0.68	0.71	0.70	0.77	1.07
76.83	QMB-64	0.61	0.63	0.65	0.69	0.80	0.93	0.83	0.88	0.87	0.87	0.81	0.89	0.88	0.96	1.13
77.65	QMB-65	0.54	0.57	0.59	0.61	0.67	0.87	0.71	0.77	0.77	0.77	0.70	0.77	0.74	0.80	0.89
78.48	QMB-66	0.64	0.66	0.66	0.68	0.74	0.90	0.74	0.77	0.74	0.75	0.68	0.74	0.71	0.75	0.82

**Table 4. Carbon Isotopes of Qumiba 1-30 samples**

Identifier 1	Depth (m)	Sample Number	d13C 2-pt w/ lin corr
QMB 2	1.49	2	-29.67
QMB 3	2.23	3	-29.96
QMB 4	2.97	4	-28.96
QMB 5	3.71	5	-28.80
QMB 6	4.46	6	-29.03
QMB 7	5.20	7	27.70
QMB 9	6.68	9	-25.39
QMB 10	7.43	10	-29.14
QMB 11	8.17	11	-27.19
QMB 12	8.91	12	-29.03
QMB 13	9.65	13	-29.48
QMB 14	10.40	14	-29.72
QMB 15	11.14	15	-28.89
QMB 16	11.88	16	-29.97
QMB 17	12.62	17	-28.55
QMB 18	13.37	18	-28.63
QMB 19	14.11	19	-30.06
QMB 20	14.85	20	-33.79
QMB 21	15.60	21	-26.44
QMB 22	16.34	22	-31.41
QMB 23	17.08	23	-30.45
QMB 24	17.82	24	-28.86
QMB 25	18.57	25	-27.55
QMB 26	19.31	26	-31.48
QMB 27	20.05	27	-29.61
QMB 28	20.79	28	-28.57
QMB 29	21.54	29	-30.68
QMB 30	22.28	30	-30.25

Excluded samples	Depth (m)	Sample number	d13C 2-pt w/ lin corr
QMB 1	0.74	1	-23.35
QMB 8	5.94	8	-22.15



**Table 5. Nitrogen Isotopes of Qumiba 1-30 samples**

Identifier 1	Depth (m)	Sample Number	d15N 2-pt w/ lin corr
QMB 2	1.49	2	4.56
QMB 3	2.23	3	5.17
QMB 4	2.97	4	5.41
QMB 5	3.71	5	5.51
QMB 6	4.46	6	4.93
QMB 7	5.20	7	5.03
QMB 10	7.43	10	4.97
QMB 11	8.17	11	4.89
QMB 12	8.91	12	4.76
QMB 13	9.65	13	5.29
QMB 14	10.40	14	6.23
QMB 15	11.14	15	5.87
QMB 16	11.88	16	5.44
QMB 17	12.62	17	5.54
QMB 18	13.37	18	5.86
QMB 19	14.11	19	6.58
QMB 20	14.85	20	6.13
QMB 21	15.60	21	6.70
QMB 22	16.34	22	5.05
QMB 23	17.08	23	5.87
QMB 24	17.82	24	5.72
QMB 25	18.57	25	5.72
QMB 26	19.31	26	5.49
QMB 27	20.05	27	6.99
QMB 28	20.79	28	7.04
QMB 29	21.54	29	4.79
QMB 30	22.28	30	5.29

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