

1 **Environmental concentrations of antifouling paint particles are toxic to sediment-dwelling**  
2 **invertebrates**

3 Christina Muller-Karanassos<sup>1,2‡</sup>, William Arundel<sup>1,2‡</sup>, Penelope K Lindeque<sup>2</sup>, Tom Vance<sup>3</sup>, Andrew Turner<sup>4</sup>  
4 & Matthew Cole<sup>2\*</sup>

5  
6 <sup>1</sup>School of Biological and Marine Sciences, University of Plymouth, Drake Circus, Plymouth, PL4 8AA, UK

7 <sup>2</sup>Marine Ecology and Biodiversity Group, Plymouth Marine Laboratory, Prospect Place, Plymouth, PL1  
8 3DH, UK

9 <sup>3</sup>PML Applications, Prospect Place, Plymouth, PL1 3DH, UK

10 <sup>4</sup>School of Geography, Earth and Environmental Sciences, University of Plymouth, Drake Circus, Plymouth,  
11 PL4 8AA, UK

12 <sup>‡</sup>joint first authors

13 \* Corresponding author: MC; Email: mcol@pml.ac.uk; Telephone: +44 (0)1752 633100

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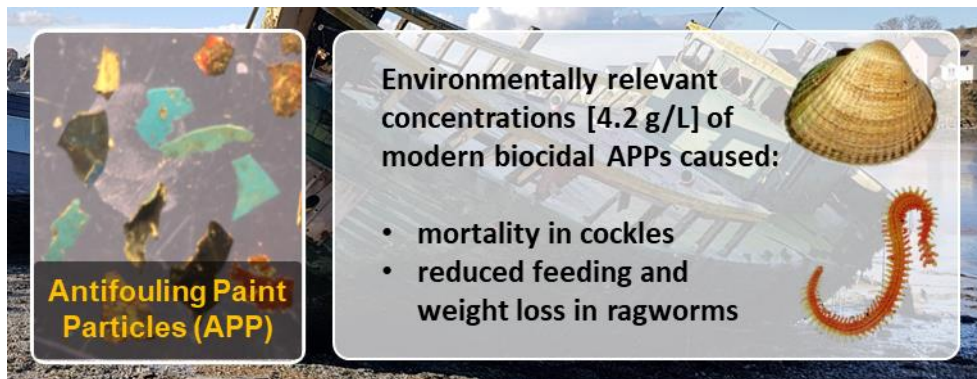
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16 **Abstract**

17 Antifouling paint particles (APPs) and associated metals have been identified in sediments around  
18 boatyards and marinas globally, but the effects of APPs on benthic organisms are largely unknown. Sub-  
19 lethal endpoints were measured following laboratory exposures of the harbour ragworm (*Hediste*  
20 *diversicolor*) and the common cockle (*Cerastoderma edule*) to environmentally relevant concentrations of  
21 biocidal ('modern' and 'historic') and biocide-free ('silicone') APPs added to clean estuarine sediment.  
22 Further, the 5-day median lethal concentrations (LC<sub>50</sub>) and effects concentrations (EC<sub>50</sub>) for modern  
23 biocidal APPs were calculated. For ragworms, significant decreases in weight (15.7%;  $p < 0.01$ ) and feeding  
24 rate (10.2%;  $p < 0.05$ ) were observed in the modern biocidal treatment; burrowing behaviour was also  
25 reduced by 29% in this treatment, but was not significant. For cockles, the modern biocidal treatment led  
26 to 100% mortality of all replicates before endpoints were measured. In cockles, there was elevated levels  
27 of metallothionein-like protein (MTLP) in response to both modern and historic biocidal treatments.  
28 Ragworms had a higher tolerance to modern APPs (5-day LC<sub>50</sub>: 19.9 APP g L<sup>-1</sup>; EC<sub>50</sub>: 14.6 g L<sup>-1</sup>) compared to  
29 cockles (5-day LC<sub>50</sub>: 2.3 g L<sup>-1</sup> and EC<sub>50</sub>: 1.4 g L<sup>-1</sup>). The results of this study indicate that modern biocidal  
30 APPs, containing high Cu concentrations, have the potential to adversely affect the health of benthic  
31 organisms at environmentally relevant concentrations. The findings highlight the need for stricter  
32 regulations on the disposal of APP waste originating from boatyards, marinas and abandoned boats.

33

34 **Graphical abstract**



35

36

37 **Keywords:** Microplastic, Fouling, Biocidal, Toxicity, Copper

38

39 **Highlights**

- 40 • Antifouling paint particles (APPs) are a type of microplastic debris
- 41 • Biocidal APPs containing copper proved toxic to sediment-dwelling biota
- 42 • Biocidal APPs were toxic at environmentally relevant concentrations
- 43 • Non-biocidal silicone-based APPs showed little toxicity
- 44 • Cockles were far more sensitive to APPs than ragworms

45

46 **Capsule**

47 In contrast to non-biocidal silicone formulations, copper-based biocidal antifouling paint particles (APPs)  
48 proved toxic to cockles and ragworms at environmentally relevant concentrations.

49

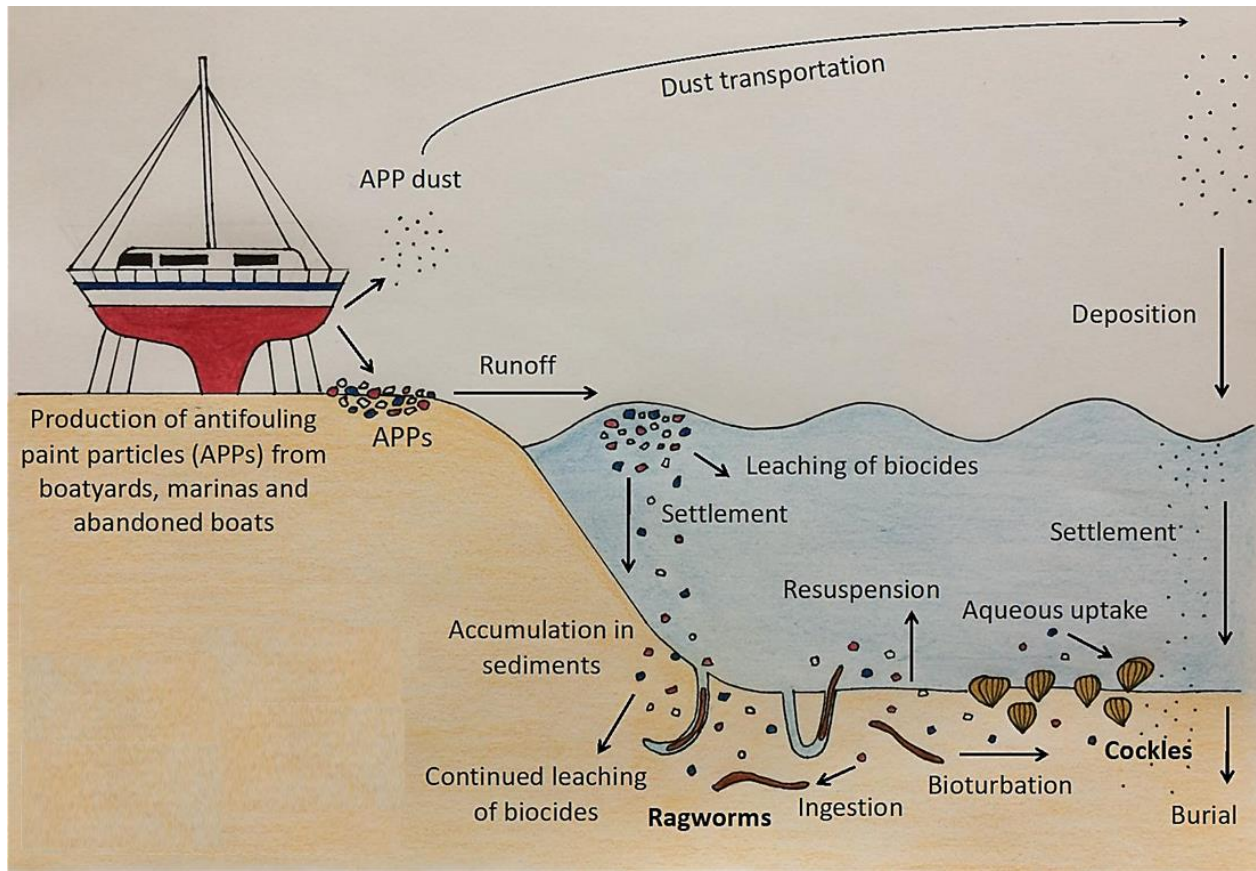
50 **1. Introduction**

51 Antifouling paint is widely applied to marine structures to reduce biofouling, with the antifouling  
52 paints and coatings market estimated to be worth US\$9.22 billion by 2021 ([Markets, 2016](#)). Biofouling of  
53 submerged marine structures can lead to increased frictional drag, reduced manoeuvrability of marine  
54 vessels, higher fuel consumption and increased cleaning and maintenance costs ([Chambers et al., 2006](#);  
55 [Yebra et al., 2004](#)). Antifouling coatings typically work by leaching biocides into the surrounding seawater  
56 and forming a protective microlayer ([Nurioglu et al., 2015](#); [Singh and Turner, 2009a](#)). Following the  
57 worldwide ban on organotin-based antifouling paints (e.g. tributyltin) in 2008, owing to toxic effects on

58 non-target organisms ([Evans et al., 1995](#); [IMO, 2018](#)), new tin-free antifouling paints were developed  
59 ([Yebera et al., 2004](#)). Most contemporary biocidal antifouling paints contain Cu (I) as the main biocide, in  
60 the form of cuprous oxide (Cu<sub>2</sub>O) or copper thiocyanate (CuSCN), in combination with Zn-based  
61 compounds such as zinc oxide (ZnO) ([Turner, 2010](#)); booster biocides such as zinc pyrithione (ZnPT), Irgarol  
62 1051 and diuron are also added to antifouling formulations to increase their effectiveness ([Turner, 2010](#)).  
63 Owing to their toxicity to marine life, a range of biocide-free antifouling formulations are also currently  
64 being developed. For example, silicone coatings work by reducing the adhesion of organisms to the  
65 surface of boats ([Almeida et al., 2007](#)) and have been found to be less toxic to marine organisms ([Karlsson  
66 and Eklund, 2004](#)).

67 Antifouling paint particles (APPs) are waste products, generated in boatyards and marinas during  
68 maintenance and cleaning of boat hulls and grounded ships and boats ([Turner, 2010](#)). The disposal of APP  
69 waste is largely unregulated in the recreational boating industry and APPs can be readily transported from  
70 hard-standings and slipways into the marine environment via run-off ([Connelly et al., 2001](#); [Thomas et al.,  
71 2003](#); [Turner, 2010](#)) (Figure 1). APPs also originate from weathering of old abandoned boats, which are  
72 often coated in numerous layers of historic antifouling paint and may contain banned or restricted  
73 compounds such as TBT and metals like Pb, historically used in antifouling and non-antifouling marine  
74 paints ([Rees et al., 2014](#); [Turner, 2010](#)). Fine APP particulates might also be aerially dispersed, as observed  
75 with microfibrils ([Liu et al., 2019](#)). APPs are highly heterogeneous in their chemical make-up, owing to the  
76 variety of antifouling formulations used over the past fifty years ([Sandberg et al., 2007](#); [Turner, 2010](#)).  
77 Once in the marine environment, APPs can accumulate in benthic sediments around marinas, boatyards  
78 and abandoned boats; in the Plym estuary (UK), sampling revealed APP concentrations of 430 particles L<sup>-1</sup>  
79 (0.2 g L<sup>-1</sup>) next to a boat maintenance facility and 400 particles L<sup>-1</sup> (4.2 g L<sup>-1</sup>) in an area containing  
80 abandoned boats ([Muller-Karanassos et al., 2019](#)). APPs continue to leach biocides into the surrounding  
81 environment, with several studies finding high metal concentrations, often exceeding environmental  
82 standards, in sediments contaminated with APPs ([Eklund et al., 2014](#); [Muller-Karanassos et al., 2019](#); [Rees  
83 et al., 2014](#); [Sapozhnikova et al., 2013](#); [Singh and Turner, 2009b](#); [Soroldoni et al., 2018a](#)). Biological activity  
84 (e.g. bioturbation, bioirrigation) has the capacity to redistribute and resuspend particles and metals  
85 present within intertidal habitats ([He et al., 2017](#); [Näkki et al., 2017](#)). Modern antifouling coatings are  
86 polymeric with an alkyd resin base ([Toben, 2017](#)) and as such, APPs can be considered as a type of  
87 microplastic (plastic debris, 1 µm - 1 mm in size) ([Boucher and Friot, 2017](#); [Hartmann et al., 2019](#)). Studies  
88 have shown that microplastic ingestion by marine organisms may lead to a reduction in feeding,  
89 reproduction ([Cole et al., 2015](#)) and energy reserves ([Wright et al., 2013](#)). Owing to their high metal

90 concentrations, including Cu, Zn, Sn and Pb, APPs pose an additional toxic threat to sediment-dwelling  
91 biota. Benthic organisms are essential for the functioning of marine coastal ecosystems and play an  
92 important role in energy transfer between pelagic and benthic ecosystems. Laboratory studies have  
93 shown that exposure to APPs can lead to an accumulation of biocidal metals (Cu and Zn) in the tissues of  
94 benthic marine organisms including the common mussel *Mytilus edulis* ([Turner et al., 2009](#)), the common  
95 periwinkle *Littorina littorea* ([Gammon et al., 2009](#)) and the lugworm *Arenicola marina* ([Turner et al., 2008](#)).  
96 Uptake of metals is thought to occur through both aqueous exposure to APP leachate and via direct  
97 ingestion of APPs, with a recent study finding evidence of APP ingestion by the harbour ragworm *Hediste*  
98 *diversicolor* collected from contaminated sediments in the Plym estuary (UK) ([Muller-Karanassos et al.,](#)  
99 [2019](#)). Exposure to APP leachate has been found to have sub-lethal effects on marine organisms including  
100 a reduction in growth, larval development and bioluminescence ([Ytreberg et al., 2010](#)). Only three  
101 publications – all focussed upon modern biocidal antifouling materials – have considered the direct  
102 toxicity of APPs these studies revealed exposure to increasing concentrations of APPs can cause: a  
103 decrease in fecundity and survival in the epibenthic copepod *Nitokra* sp. ([Soroldoni et al., 2017](#));  
104 decreased survival rates in pelagic copepods ([Molino et al., 2019](#)); and decreased survival in the benthic  
105 microcrustaceans *Monokalliapseudes schubarti* (a tanaid) and *Hyalella azteca* (an amphipod) ([Soroldoni](#)  
106 [et al., 2020](#)).  
107



108

109 **Figure 1.** Potential abiotic and biotic transport pathways for antifouling paint particles (APPs) in intertidal  
 110 habitats.

111

112 The aim of this study was to determine if exposure to both biocidal and non-biocidal APPs, at  
 113 environmentally relevant concentrations, negatively affects the health of benthic organisms with different  
 114 feeding modes. Two sediment-dwelling estuarine species were chosen for this study, the harbour  
 115 ragworm, *H. diversicolor*, and the edible cockle, *Cerastoderma edule*. *H. diversicolor* is a polychaete that  
 116 is widely distributed in estuaries within Northwest Europe (Budd, 2008), having an important role in  
 117 estuarine ecosystems as a bioturbator and as a food source for numerous species of wading birds (Goss-  
 118 Custard et al., 1989) and flatfish (Budd, 2008). *C. edule*, is a commercially important species eaten widely  
 119 throughout Europe and is also an important food source for wading birds and pelagic species in intertidal  
 120 mudflats, where it can make up a significant proportion of biomass (Romano et al., 2011). Two laboratory  
 121 studies were conducted: (1) an 18-day exposure to investigate sub-lethal health effects; and, (2) a 5-day  
 122 assay to calculate how APP concentrations affected mortality (i.e. LC50s). The findings of this study have  
 123 implications for the health of biodiverse intertidal ecosystems and the management of antifouling waste.

124

125 **2. Methods**

126

127 **2.1 Specimen collection and husbandry**

128           Adult *H. diversicolor* (ragworms; 0.29-1.01 g wet weight) and *C. edule* (cockles; 26-36 mm shell  
129 length, 10.5-25 g wet weight) were collected by hand from an intertidal mudflat at Saltram Park (N 50°  
130 22' 43.284" W 4° 6' 0.755") located within the Plym Estuary, UK (Figure 1) in June and July 2018 and  
131 transported to PML within 1 hour of sampling. There is no boating activity at Saltram Park and a recent  
132 field study showed minimal APP and metal contamination at this site ([Muller-Karanassos et al., 2019](#)). A  
133 salinity of 31.1 was measured at high water using an Oakton SALT 6+ handheld probe. On return to the  
134 laboratory, ragworms and cockles were allowed to acclimate for 4-7 days in polypropylene tanks (49 x 35  
135 x 15 cm) containing approximately 5 cm depth of sieved estuarine sediment (<1 mm) collected at Saltram  
136 Park and 5 cm depth of aerated filtered seawater (FSW, 0.2 µm Millipore filter diluted to 31.1 salinity with  
137 Milli-Q water).

138



139  
140 **Figure 2.** Animals and sediment were collected from Saltram Park, Plym Estuary (UK). Historic APPs were  
141 sampled from abandoned boats at Hooe Lake. Map data: Google Earth, Landsat / Copernicus.  
142

## 143 **2.2 APP generation and characterisation**

144 Three types of APPs were generated for use in the exposures, including two biocidal ('historic' and  
145 'modern') and one biocide-free ('silicone'). 'Historic' biocidal antifouling paint flakes were collected by  
146 hand from abandoned boats at Hooe Lake, in the Plym Estuary (N 50° 21' 22.464" W 4° 6' 28.943"; Figure  
147 2), and cleaned from any visible dirt and algae ([Singh and Turner, 2009a](#)). 'Modern' biocidal antifouling  
148 paint was scraped off rolled mild steel panels that had been painted with three commercially available  
149 biocidal paints in 2012, including an anti-corrosive layer, tie coat and top coat, and submerged in natural  
150 seawater off the coast of Orkney (Scotland) for 8 weeks in 2014. Non-biocidal 'silicone' antifouling paint

151 was scraped off rolled mild steel panels painted with commercially available silicone paint and submerged  
152 in natural seawater for 5-10 days at station L4 (off the coast of Plymouth, UK;  
153 [www.westernchannelobservatory.co.uk](http://www.westernchannelobservatory.co.uk)) in 2018. APPs were prepared by grinding down paint flakes using  
154 a pestle and mortar with the aid of liquid nitrogen; paint particles were passed through stainless steel  
155 sieves to collect the 100  $\mu\text{m}$ –1 mm fraction, a size range considered bioavailable to the target species and  
156 for which APP particles have been identified in estuarine sediments ([Muller-Karanassos et al., 2019](#)).  
157 Particle size distribution was evaluated by measuring a representative sub-sample of 100 APPs under a  
158 microscope. Metal analysis of the three APP types was carried out non-destructively using an energy-  
159 dispersive portable x-ray fluorescence (XRF) spectrometer (Niton XL3t He GOLDD+), with the focus on  
160 metals that are or have been commonly employed as biocides in antifouling paints (Cu, Hg, Pb, Sn and  
161 Zn). The instrument was operated in a low-density plastics mode with small-spot 3-mm collimation and  
162 thickness correction (between 0.5 and 3 mm) and performance was verified by analysis of Niton plastic  
163 reference discs containing known concentrations of various metals. APP fragments were characterised in  
164 clear polyethylene zip-bags for 60 s each at five different locations.

165

### 166 **2.3 Experimental set-ups**

167 Two laboratory-based exposure experiments, an 18-day and 5-day exposure, were carried out for  
168 each species in a temperature-controlled facility ( $13\pm 1^\circ\text{C}$ ) under a 12:12 hour light:dark cycle. Estuarine  
169 sediment was collected from Saltram Park, where our samples revealed no evidence of APPs ([Muller-  
170 Karanassos et al., 2019](#)); sediment was collected to a depth of approximately 20 cm at the same time as  
171 the biota, and sieved through a 1 mm stainless steel sieve with the aid of seawater to remove macrodebris.  
172 Sediment was stored in a plastic tub covered with aluminium foil until experiments were carried out. Prior  
173 to exposures, pre-weighed APPs were mixed into sediments in individual containers using a stainless-steel  
174 spatula and then allowed to settle for 24 h prior to the introduction of animals. Owing to the high-density  
175 of the APPs, we observed that particles remained within sediments during water changes. For the 18-day  
176 exposure, APP concentrations were based on environmental concentrations identified at Hooe Lake (0–  
177  $18.8 \text{ g L}^{-1}$ ; mean  $4.2 \text{ g L}^{-1}$ ) ([Muller-Karanassos et al., 2019](#)) adjusting for the density differences of the  
178 different APPs (SI, Table S1). Trial experiments carried out using the maximum environmental APP  
179 concentration ( $18.8 \text{ g L}^{-1}$ ) led to mortality of all *H. diversicolor* and *C. edule* in the modern treatment within  
180 6 days and, therefore, the mean environmental APP concentration was used in order to assess sub-lethal  
181 effects. Density-corrected APP concentrations were:  $4.2 \text{ g L}^{-1}$  for the historic biocidal treatment;  $3.0 \text{ g L}^{-1}$   
182 for the modern biocidal treatment; and  $2.1 \text{ g L}^{-1}$  for the non-biocidal silicone treatment. For the 5-day LC<sub>50</sub>



183 exposure, modern biocidal APPs (selected owing to their comparatively higher toxicity) were used at  
184 concentrations ranging from 0-30 g L<sup>-1</sup> (ragworms) and 0-6 g L<sup>-1</sup> (cockles).

185 After acclimation, individual organisms were weighed to ascertain pre-exposure wet weight.  
186 Ragworms were transferred into individual 100 mL polyethylene containers (1 ragworm per container)  
187 containing 50 mL of sieved sediment pre-mixed with APPs (silicone: 0.105 g, historic: 0.209 g, modern:  
188 0.152 g) or control sediment, and 50 mL of aerated FSW. Cockles were transferred into 1 L food-grade  
189 containers (1 cockle per vessel) containing 250 mL of control sediment or sediment pre-mixed with APPs  
190 (historic: 1.047 g; modern: 0.759 g; silicone: 0.525 g) and 700 mL of well-aerated, natural seawater filtered  
191 through a 0.2 µm glass fibre filter and diluted with ultrapure water to a salinity of 31.1 (matching estuarine  
192 conditions). Water was changed every 2-3 days and water quality parameters including temperature, pH,  
193 salinity, conductivity and dissolved oxygen were measured using YSI Pro 1030 and Pro 20 meters at every  
194 water change (SI, Table S2). For the 18-day exposure, ragworms (n=10 per treatment) were each fed 8 g  
195 of hatched *Artemia salina* nauplii (Instant Baby Brine Shrimp, Ocean Nutrition) every other day ([Moreira  
196 et al., 2005](#)), and cockles (n=10 per treatment) were fed every other day with an alternating diet of live  
197 *Thalassiosira rotula* (4000 cells mL<sup>-1</sup>) and freeze-dried multi-cell algal culture (5 Species Phytoplankton,  
198 Reefphyto). For the 5-day exposure, ragworms (n=5 per APP concentration) and cockles (n=5 per APP  
199 concentration) were not fed.

200

## 201 **2.4 18-day exposure**

### 202 **2.4.1 Feeding rate**

203 For ragworms, a feeding rate experiment was carried out following methods in [Moreira et al.  
204 \(2005\)](#). In summary, ragworms were transferred into individual Petri dishes containing 100 *A. salina*  
205 nauplii and 20 mL of diluted seawater and allowed to feed in darkness for 1 hour. Remaining *A. salina*  
206 nauplii were counted and feeding rate was determined as the number of nauplii consumed per hour.

207 For cockles, clearance rate was assessed using an algal feeding assay. A known concentration of  
208 *T. rotula* cells were introduced to experimental vessels and aliquots of water removed at the start of the  
209 experiment and after 1 hour. Five blank vessels were used to account for algal growth and settling. Algal  
210 concentration was assessed using the Sedgwick-Rafter counting method. Clearance rate (CR) was  
211 calculated as described in [Romano et al. \(2011\)](#), as follows:

$$212 \quad CR = V \times \frac{\log C_1 - \log C_2}{t} \times n$$

213 where  $V$  is the volume of water in the experimental vessel,  $C_1$  is the initial algal concentration,  $C_2$  is the  
214 final algal concentration,  $t$  is experimental duration, and  $n$  is the number of cockles per vessel.

#### 215 **2.4.2 Weight change**

216 Animals were re-weighed (post-exposure wet weight) and weight change was determined as the  
217 difference between pre- and post-exposure wet weight.

#### 218 **2.4.3 Burrowing**

219 For ragworms, burrowing behaviour was analysed following methods by [Buffet et al. \(2011\)](#).  
220 Ragworms were placed in 100 mL containers with 50 mL clean estuarine sediment from Saltram Park and  
221 50 mL of aerated filtered seawater. Burial state was recorded every 2 minutes for a total of 30 minutes.

222 For cockles, burrowing was analysed as in [Byrne and O'halloran \(2000\)](#) and [Møhlenberg and](#)  
223 [Kjørboe \(1983\)](#). Cockles were placed in individual containers set up in experimental chambers containing  
224 clean estuarine sediments, and burrowing state was recorded after 5, 10, 15, 20, 30, 40, 50, 60, 80, 100,  
225 120 and 180 minutes. Cockles were judged to be burrowed when approximately 50% of the shell was  
226 covered by sediment.

#### 227 **2.4.4 Metallothionein-like protein assay**

228 Metallothionein-like protein (MTLP) concentration in whole tissue was quantified for both species  
229 using the method described by [Viarengo et al. \(1997\)](#) and [UNEP \(1999\)](#) with a few alterations. Using frozen  
230 specimens, soft tissues were pooled in Petri dishes for each treatment (control, silicone, historic, modern)  
231 and defrosted in an oven for 10 minutes at 50 °C. Samples were weighed and volume of tissue measured  
232 before homogenisation of tissue in 2 volumes of homogenising buffer containing  $\beta$ -mercaptoethanol,  
233 phenylmethylsulphonylfuride and leupeptin. Homogenisation was performed using a 600 W Morphy  
234 Richards 402058 hand-blender in beakers until tissue was smooth enough to be removed using a 1 mL  
235 Gilson pipette. Two mL of sample was pipetted into 2 mL centrifuge tubes (6 replicates per treatment)  
236 and centrifuged at 2 °C, 14,000 x g for 45 minutes. One mL of supernatant was carefully removed by  
237 pipette and added to 1.05 mL of cold (-20 °C) absolute ethanol and 80  $\mu$ L of chloroform in 2 mL centrifuge  
238 tubes stored on ice. Tubes were inverted and vortexed before centrifugation at 0 °C and 6000 x g for 10  
239 minutes. Supernatant was carefully pipetted off into 15 mL tubes and volume recorded for each sample.  
240 Ten  $\mu$ L of a 100 mg mL<sup>-1</sup> stock of RNA, 40  $\mu$ L of 37% HCl and 3 volumes of cold absolute ethanol were  
241 added before vortexing and storing at -20 °C for 1 hour. Samples were then centrifuged at 0 °C and 4000  
242 x g for 15 minutes to form a pellet. Supernatant was carefully removed and the pellet washed with 5 mL  
243 of a *cleaner* solution (87:1:12 ethanol/chloroform/homogenising buffer, stored at -20 °C). Samples were  
244 re-centrifuged at 0 °C and 4000 x g for 5 minutes before removal of supernatant and drying. 150  $\mu$ L of

245 0.25 M NaCl solution and 150  $\mu$ L of 1N HCl containing 4 mM EDTA were added and then vortexed.  
246 Standards were made up using a 1 mg mL<sup>-1</sup> solution of glutathione in 0.25 M NaCl, and blanks were made  
247 up with 150  $\mu$ L of 0.25 M NaCl solution and 150  $\mu$ L of 1N HCl containing 4 mM EDTA. Just before analysis,  
248 0.43 mM (7.14 mg/42 mL) DTNB (Ellman's reagent) was dissolved in 0.2 M phosphate buffer pH 8  
249 containing 2 M NaCl and 4.2 mL of this solution was added to all samples, standards and blanks. All tubes  
250 were then centrifuged at 3,000 xg for 5 minutes at room temperature. Absorbance was measured at 412  
251 nm in a VWR UV-3100 PC spectrophotometer.

252

## 253 **2.5 5-day exposure**

### 254 **2.5.1 LC<sub>50</sub> and EC<sub>50</sub>**

255 Daily checks were carried out for mortality of both species. Healthy ragworms will naturally  
256 burrow to avoid predation ([Kalman et al., 2009](#)) and have a fast reaction time (personal observations).  
257 Ragworms were recorded as dead when they appeared to be motionless at the surface of the sediment  
258 and there was no response to touch, and adversely affected when they were found on the sediment  
259 surface and only reacted slightly to touch. Cockles were also deemed to be dead when no response was  
260 observed on gentle touching of valves or foot, and adversely affected when exhibiting gaping behaviour  
261 with minimal response to touch ([Thompson and Richardson, 1993](#)).

262

## 263 **2.6 Statistical analysis**

264 All ragworm and cockle data were checked for normality using Shapiro-Wilk's test. For the non-  
265 parametric ragworm data a Kruskal-Wallis test followed by Wilcoxon Rank Sum test was used, while a  
266 one-way analysis of variance (ANOVA) with Tukey's post-hoc test was used to compare differences in  
267 parametric cockle data. A generalised linear model (GLM) with binomial distribution was used to compare  
268 the number of animals burrowed after 30 minutes across treatments. The 5-day LC<sub>50</sub> and EC<sub>50</sub> values were  
269 calculated by probit analysis in SPSS. All other analyses were carried out using R statistical software v3.5.1  
270 ([R, 2019](#)). Biological data is presented as mean values  $\pm$  standard error, while metal chemistry data is  
271 presented as mean values  $\pm$  standard deviation; significant difference is attributed where  $p < 0.05$ .

272

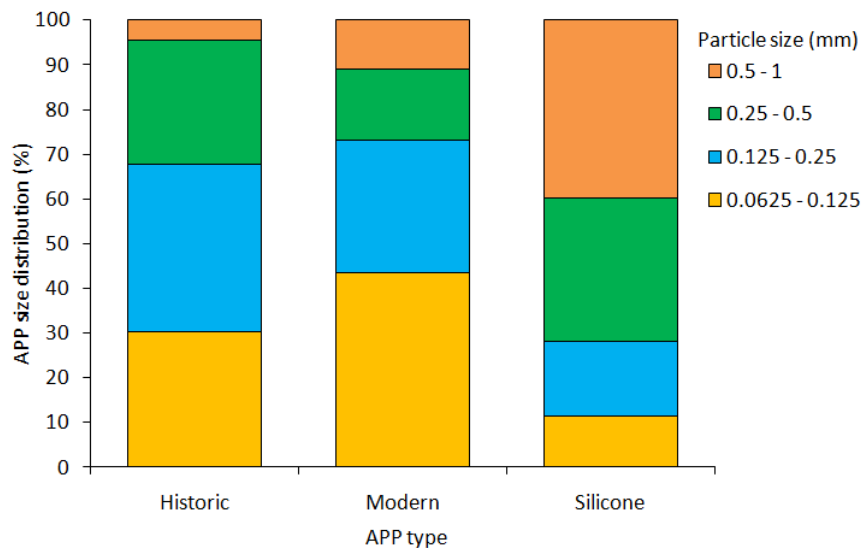
## 273 **3. Results**

### 274 **3.1 APP characterisation**

275 APPs ranged from 0.0625–1 mm in size, with particle size varying between the three APP types  
 276 (Figure 3). Overall, historic biocidal APPs had the highest proportion of particles in the size range 0.125–  
 277 0.25 mm (37.6%), modern biocidal APPs had the smallest particle size, with 43.6% of particles in the size  
 278 range 0.0625–0.125 mm. Non-biocidal silicone antifouling paint had a “rubber-like” consistency and was  
 279 much harder to breakdown into smaller particles via cryogenic grinding, and as a result these APPs had  
 280 the largest particle size, with 39.7% of particles in the size range 0.5–1 mm.

281 Metal concentrations varied considerably between APP types (Table 1). Historic biocidal APPs  
 282 had the highest concentrations of Zn, Pb and Hg, and relatively high Cu concentrations, whereas modern  
 283 biocidal APPs had the highest Cu concentrations with much lower concentrations of Zn. Copper was  
 284 detected in one reading taken from the non-biocidal silicone APPs but Zn was not detected; however,  
 285 there were detectable concentrations of Sn and Hg in the these APPs.

286



287  
 288 **Figure 3.** Particle size distribution of the different types of APPs used in exposure experiments.

289  
 290 **Table 1.** Metal concentrations (mg kg<sup>-1</sup>) in APPs used in 18-day and 5-day assays. The mean of 5  
 291 measurements is shown for each APP type ± standard error. Metal concentrations not detected were  
 292 replaced with measurement detection limits (values where < is shown) for the calculation of means.

APP type	Cu	Zn	Sn	Pb	Hg
Historic	148000	69000	412	1030	1620
	142000	64500	408	2010	1350
	176000	73200	326	565	1930

	181000	87100	576	503	2570
	168000	78200	510	1210	1590
Mean	<b>163000</b> ± 769	<b>74400</b> ± 3901	<b>446</b> ± 44	<b>1060</b> ± 272	<b>1810</b> ± 211
<b>Modern</b>	440000	4900	977	401	399
	427000	9180	853	<237	394
	428000	18000	941	<225	459
	420000	17200	782	<246	436
	421000	13500	750	<254	411
Mean	<b>427000</b> ± 357	<b>12600</b> ± 247	<b>861</b> ± 44	<b>272</b> ± 32	<b>420</b> ± 12
<b>Silicone</b>	<126	<81	838	<24	69
	<140	<89	913	<30	77
	150	<99	863	<42	69
	<130	<86	705	<38	74
	<145	<94	702	<34	78
Mean	<b>138</b> ± 4	<b>&lt;90</b> ± 3	<b>804</b> ± 43	<b>34</b> ± 3	<b>73</b> ± 2

293

### 294 3.2 18-day assay

295 Individual ragworms (*H. diversicolor*) were used for testing sub-lethal end-points across four  
 296 treatments (control n=9, non-biocidal silicone n=8, historic biocidal n=9 and modern biocidal n=10). Three  
 297 ragworms, from the control, non-biocidal silicone and historic biocidal treatments, spawned during the  
 298 exposure and were removed from further analysis to avoid bias. One ragworm from the non-biocidal  
 299 silicone treatment was lost during processing before end-point experiments were carried out. For cockles  
 300 (*C. edule*), 100% mortality was observed for the modern biocidal treatment after 10 days, so sub-lethal  
 301 endpoints could not be assessed; all cockles survived in control, non-biocidal silicone and historic biocidal  
 302 treatments (n=10).

#### 303 3.2.1 Feeding rate

304 The highest feeding rate occurred in the control treatment ( $17 \pm 4$  nauplii h<sup>-1</sup> ind.<sup>-1</sup>) and the lowest  
 305 was observed in the modern biocidal treatment ( $6 \pm 2$  nauplii h<sup>-1</sup> ind.<sup>-1</sup>). The feeding rate of ragworms  
 306 differed significantly between treatments (One-way ANOVA:  $F_{3,32}=3.131$ ,  $p<0.05$ ; Figure 4A), and the post-  
 307 hoc test showed a significant difference between the modern biocidal treatment and control (Tukey:  
 308  $p<0.05$ ).

309 No significant difference in clearance rates of cockles between treatments was observed (control:  
310 0.60 L h<sup>-1</sup> ind.<sup>-1</sup>; non-biocidal silicone: 0.63 L h<sup>-1</sup> ind.<sup>-1</sup>; historic biocidal: 0.82 L h<sup>-1</sup> ind.<sup>-1</sup>; ANOVA:  $F_{2,27}=2.797$ ,  
311  $p=0.08$ ; Figure 4B).

### 312 **3.2.2 Weight change**

313 Ragworms showed a significant difference in mean weight change between treatments (Kruskal-  
314 Wallis:  $p<0.01$ ; Figure 4C). Exposure to biocidal APPs resulted in a marked decrease in weight, however  
315 when compared with controls, significant differences were only observed in ragworms exposed to modern  
316 biocidal APPs (18.5 ± 4% weight loss; Wilcoxon:  $p<0.01$ ).

317 Weight change in cockles was not significantly affected by treatment (One-way ANOVA:  $F_{2,27}=3.30$ ,  
318  $p=0.052$ ; Figure 4D). However, compared with controls, cockles exposed to non-biocidal silicone APPs did  
319 show a significantly greater weight loss (Tukey:  $p<0.05$ ).

### 320 **3.2.3 Burrowing**

321 An average of 60±16% of ragworms exposed to modern biocidal APPs had buried after 30 minutes,  
322 while an average of 88-100% of individuals in the other treatments had successfully buried in this time  
323 period. However, there were no significant differences between treatments (GLM:  $p>0.05$ ; Figure 4E).

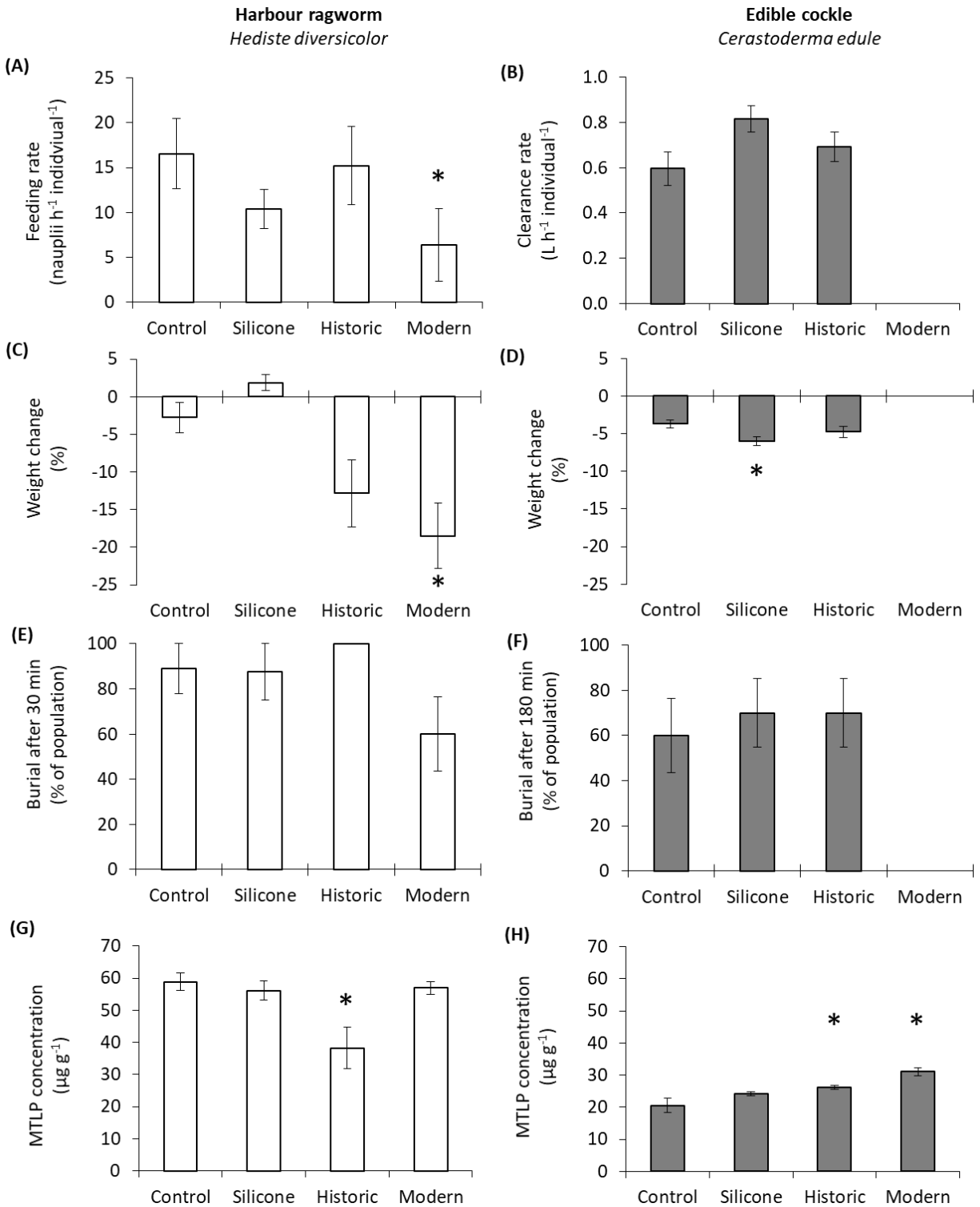
324 Approximately two-thirds of cockles had successfully buried after 180 minutes across all  
325 treatments, with no significant difference between treatments (GLM:  $p>0.05$ ; Figure 4F).

### 326 **3.2.4 Metallothionein-like protein**

327 MTLP concentrations in ragworms averaged 59 µg g<sup>-1</sup> in controls, with no significant difference  
328 between treatments (Kruskal-Wallis:  $p=0.07$ ); however, ragworms exposed to the historic biocidal APPs  
329 showed significantly reduced MTLP concentrations compared with the controls (38.3 µg g<sup>-1</sup>; Wilcoxon:  
330  $p<0.05$ ; Figure 4G).

331 Whole-tissue MTLP concentrations in cockles averaged 20.6 µg g<sup>-1</sup> in controls, which is  
332 comparable with other studies ([Aly et al., 2014](#)). MTLP concentrations were significantly affected by  
333 treatment (One-way ANOVA:  $F_{3,18}=13.49$ ;  $p<0.01$ ; Figure 4H), with MTLP levels significantly elevated in  
334 historic biocidal (26.2 µg g<sup>-1</sup>; Tukey:  $p<0.05$ ) and modern biocidal (31.1 µg g<sup>-1</sup>; Tukey:  $p<0.01$ ) treatments,  
335 as compared with controls.

336



337

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340

**Figure 4.** Sub-lethal health responses in ragworms (white bars) and cockles (grey bars) following an 18-day exposure to controls and silicone, historic or modern APP treatments: (A) Feeding rates (nauplii individual $^{-1}$  hour $^{-1}$ ); (B) Clearance rates ( $L h^{-1}$  individual $^{-1}$ ); (C-D) Weight change (%); (E-F) Percentage of

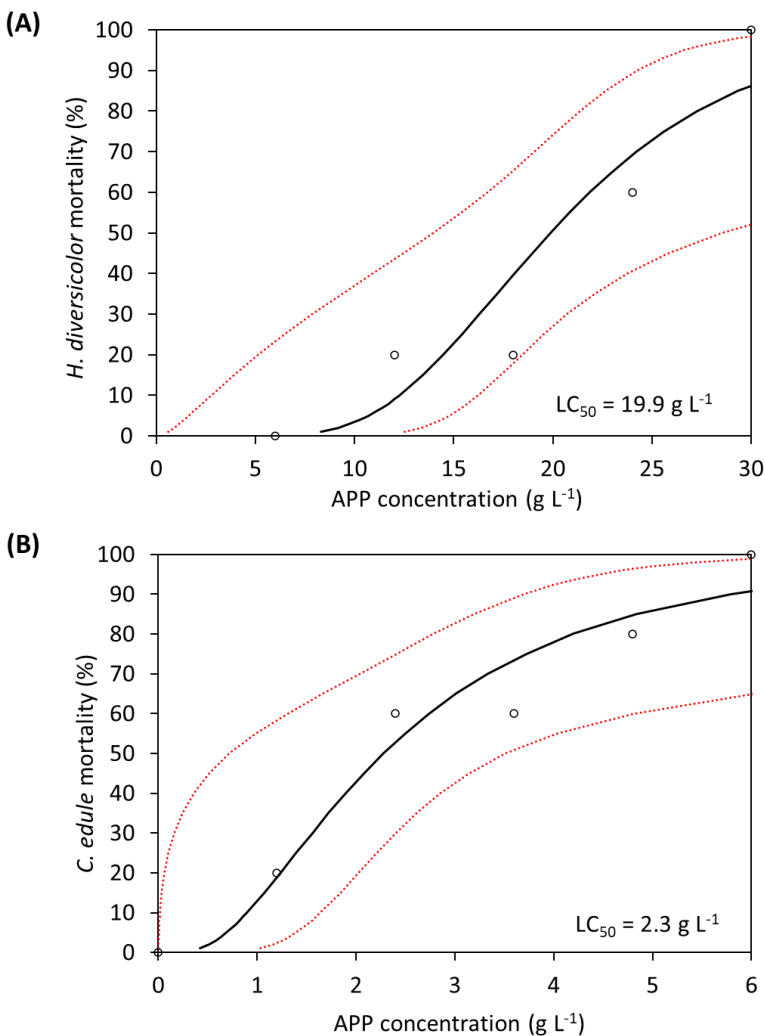
341 individuals burrowed after 30 minutes (ragworm) and 180 minutes (cockles); (G-H) Metallothionein-like  
342 protein (MTLP) concentration, noting that cockles in modern treatment died before the end of the 18-day  
343 exposure period. \* denotes significant difference from control treatment. Error bars indicate standard  
344 error.

345

### 346 3.3 5-day assay: LC50s

347 The 5-day LC<sub>50</sub> was 19.9 APP g L<sup>-1</sup> for ragworms (Figure 5A) and 2.3 g L<sup>-1</sup> for cockles (Figure 5B). In  
348 the control treatment, there were no mortalities for either species. The 5-day EC<sub>50</sub> values were calculated  
349 as 14.6 g L<sup>-1</sup> for ragworms and 1.4 g L<sup>-1</sup> for cockles (data not shown).

350



351

352 **Figure 5.** 5-day median lethal concentration (LC<sub>50</sub>) dose-response curves (black line) with 95% confidence

353 intervals (orange lines) for (A) ragworms and (B) cockles exposed to modern APPs.



354

## 355 4. Discussion

356

### 357 4.1 18-day exposure

358 Results from the 18-day study demonstrate that exposure to modern biocidal APPs at  
359 environmentally relevant concentrations leads to adverse health effects in the ragworm *H. diversicolor*  
360 and mortality in the cockle *C. edule*. As compared with controls, significant net decreases in weight  
361 ( $15.7 \pm 5\%$ ) and feeding rate ( $10.2 \pm 4\%$ ) were observed in ragworms exposed to modern biocidal APPs;  
362 furthermore modern biocidal APPs were associated with the highest percentage of un-burrowed  
363 ragworms, with  $29 \pm 20\%$  fewer ragworms burrowed after 30 minutes compared to the average control,  
364 although this difference was not statistically significant. Modern biocidal APPs were acutely toxic to  
365 cockles, resulting in mortality of all replicates in the first 10 days of exposure. Both historic and modern  
366 biocidal APPs caused significant increases in MTLPs in cockles. Largely, historic biocidal and non-biocidal  
367 silicone APPs did not cause substantial sub-lethal health effects in ragworms or cockles, although it was  
368 observed that exposure to historic biocidal APPs led to substantial weight loss in the ragworms.

369 A reduction in weight, feeding and burrowing activity of ragworms could lead to a range of  
370 consequences in the natural environment. Impairment of feeding activity in marine organisms can directly  
371 affect population parameters such as growth, reproduction ([Maltby et al., 2001](#)) and energy intake  
372 ([Kalman et al., 2009](#)). Previous studies have demonstrated that Cu exposure can have lethal and sub-lethal  
373 effects on ragworms, including reduced burrowing activity ([Thit et al., 2015](#)) and feeding ([Moreira et al.,](#)  
374 [2005](#)). Reduced burrowing efficiency in ragworms and cockles increases the likelihood of predation  
375 ([Kalman et al., 2009](#)), and reduces bioturbation activity – a vital process for sediment processing and  
376 deposition of organic matter in estuarine ecosystems ([Moreira et al., 2006](#)). Increased mortality of cockles  
377 caused by exposure to APPs could lead to a decrease in natural populations, with implications for  
378 ecosystem functionality (e.g. bioturbation and food webs). Soroldini et al. ([Soroldini et al., 2017](#);  
379 [Soroldini et al., 2020](#)) similarly found an increase in mortality in copepods and benthic invertebrates with  
380 increasing concentrations of APPs originating from a commonly-used contemporary commercial  
381 antifouling paint containing high concentrations of Cu and Zn (Cu:  $234 \pm 0.27 \text{ g kg}^{-1}$ , Zn:  $112 \pm 0.84 \text{ g kg}^{-1}$ ,  
382 Pb:  $0.51 \pm 0.01 \text{ g kg}^{-1}$ ). Another study found that leachate from sediment contaminated with APPs caused  
383 a reduction in gram-negative bacteria (*Vibrio fischeri*) bioluminescence, decreased growth rate of the red  
384 algae (*Ceramium tenuicorne*) and reduced larval development in the harpacticoid copepod *Nitocra*

385 *spinipes* ([Ytreberg et al., 2010](#)); while Cu was found to be more toxic to *V. fischeri* and *C. tenuicorne*, *N.*  
386 *spinipes* was more sensitive to Zn ([Ytreberg et al., 2010](#)).

387 In comparing the metal profiles of the biocidal APPs, it was evident that the modern paints  
388 contained far higher copper concentrations (Cu: modern:  $427,000 \pm 7,000 \text{ mg kg}^{-1}$ ; historic:  $163,000 \pm$   
389  $15,600 \text{ mg kg}^{-1}$ ). The lower Cu concentrations in the historic biocidal APPs could be attributed to the  
390 weathering of historic APP and a longer period of Cu leaching from aged historic paints ([Rees et al., 2014](#)),  
391 leading to more inert APPs and therefore a reduced toxicity. However, historic APPs contained the highest  
392 Zn concentrations ( $74,400 \pm 7,800 \text{ mg kg}^{-1}$ ) and also contain other toxic metals such as Pb ( $1,060 \pm 543 \text{ mg}$   
393  $\text{kg}^{-1}$ ), Hg ( $1,810 \pm 423 \text{ mg kg}^{-1}$ ) and Sn ( $446 \pm 87 \text{ mg kg}^{-1}$ ), which may still pose an ongoing risk to non-target  
394 organisms. Based on the heightened mortality and sub-lethal effects stemming from modern biocidal APP  
395 exposure, we surmise Cu is the most toxic biocidal metal present. [Soroldoni et al. \(2017\)](#) similarly  
396 identified that Cu was more toxic to copepods than Zn. Once metals have been taken up by an organism,  
397 they can be detoxified by accumulation in metal-rich granules or by binding with metallothionein (MT)  
398 and MTLP that regulate metal concentrations ([Rainbow, 2007](#)). The elevated MTLP in cockles from historic  
399 and modern biocidal treatments compared to the control suggests an upregulation of MTLP in response  
400 to metal exposure, particularly in the modern biocidal treatment (despite these cockles not surviving the  
401 full 18-day exposure). MTLP concentrations in ragworms were not affected by treatment. Previous studies  
402 also found no relationship between accumulated metal concentrations and MTLP levels in ragworms  
403 ([Poirier et al., 2006](#); [Solé et al., 2009](#)), and several studies have shown that exposure to Cu leads to an  
404 accumulation of this metal in *H. diversicolor* ([Berthet et al., 2003](#); [Geffard et al., 2005](#)). While ragworms  
405 have an intolerance to Cu, they have shown the capacity to regulate Zn body concentrations, thought to  
406 occur by reduced metal uptake rates, increased excretion rates and/or through storage of metals in  
407 granules or MTLPs ([Berthet et al., 2003](#); [Geffard et al., 2005](#)).

408 Other compounds found in antifouling paint such as booster biocides, solvents and binders can  
409 also have toxic effects ([Karlsson et al., 2010](#)) and the mixture of metals and other compounds found in  
410 APPs are likely to have synergistic effects ([Soroldoni et al., 2017](#)). It is therefore important to examine the  
411 toxicity of APPs as a whole, in addition to that of individual compounds found within antifouling paints.  
412 APPs were not tested for other compounds in this study, although they are likely to have contributed to  
413 the high toxicity observed with modern biocidal APPs. Although uptake routes were not investigated in  
414 the current study, exposure to biocides likely occurred via ingestion of APPs and uptake of dissolved  
415 metals leached into the water column and sediment porewater. By design, biocidal antifouling paints will  
416 constantly release metal ions into the surrounding water to prevent adherence and growth of fouling

417 organisms, so it is expected that biocidal APPs will continue to emit Cu and Zn throughout their lifespan,  
418 although whether the rate of dissipation changes over time remains unclear. A number of studies have  
419 exposed marine organisms to APP leachate solutions, which have been routinely demonstrated to be  
420 readily taken up by biota and cause toxicity ([Gammon et al., 2009](#); [Katranitsas et al., 2003](#); [Soroldoni et](#)  
421 [al., 2018b](#); [Tolhurst et al., 2007](#)). In the Plym estuary, the smallest APPs observed were ~0.5 mm in size  
422 ([Muller-Karanassos et al., 2019](#)); detecting smaller particles was hindered by sampling and analytical  
423 limitations, however abiotic and biotic processes can be expected to contribute to the proliferation of  
424 even smaller APPs. In this study APPs were prepared to 0.0625-1 mm in size, with differences in particle  
425 size profiles of biocidal and non-biocidal APPs observed: the majority of the “rubber-like” silicone APPs  
426 being >0.5 mm, and the majority of biocidal APPs being <0.5 mm diameter. We cannot say for certain  
427 whether the observed differences in particle size profiles might influence toxicity, however we consider  
428 all particles to be in the normal prey size range of cockles and ragworms. Toxicity studies using plastic  
429 particulates of distinct size (i.e. nano- vs micro) have revealed size dependent effects, with differences  
430 stemming from different modes of biological action; for example, 50 nm polystyrene nanoplastics  
431 triggered an immune response in mussels (likely owing to their capacity to readily translocate into  
432 circulatory fluids) while exposure to 20 µm polystyrene microplastics caused no observable impact ([Cole](#)  
433 [et al., 2020](#)). It is interesting to consider whether exposure to APPs of markedly different particle size (i.e.  
434 nano vs micro) would significantly effect toxicity; for example, might smaller APPs with larger surface  
435 areas increase metal leaching or penetrate deeper into tissues causing cellular toxicity ([Jeong et al., 2016](#))?  
436 Further work is needed to clarify the mechanisms responsible for toxicity of APPs to benthic organisms  
437 observed in the present study.

438 Non-biocidal silicone paints, absent of Cu and Zn, caused no adverse health effects on cockles or  
439 ragworms, supporting our hypothesis that Cu, likely in combination with other compounds, is the most  
440 toxic component of APPs. This is further supported by other studies: for example, [Watermann et al. \(2005\)](#)  
441 found that silicone coatings did not negatively affect barnacle cypris larvae settlement and luminescence  
442 of the bacteria *V. fischeri*; and [Karlsson and Eklund \(2004\)](#) observed no changes to growth of two  
443 macroalgae (*C. tenuicorne* and *Ceramium strictum*) and no mortality of the copepod *N. spinipes* when  
444 exposed to silicone paint. Biocide-free silicone coatings may be a suitable alternative to modern biocidal  
445 antifouling paints for high speed vessels, but further studies are needed to assess long-term effects.

446 Owing to their alkyd-resin base, APPs can be considered as microplastics ([Hartmann et al., 2019](#)).  
447 Microplastic debris can be ingested by a wide array of marine organisms, with evidence of sub-lethal harm  
448 in a number of studies e.g. [Cole et al. \(2015\)](#), [Wright et al. \(2013\)](#). In this study, it is evident that non-

449 biocidal silicone APPs (absent of metal additives) have a far lower toxicity than APPs containing biocides,  
450 highlighting the importance of considering additive and metal profiles when evaluating microplastic  
451 toxicity. While the physical properties of a microplastic (e.g. size, shape) have been shown to interfere  
452 with feeding and movement in marine biota (see [Galloway et al. \(2017\)](#) and [Setälä et al. \(2018\)](#)), their  
453 chemical composition is often overlooked. However, microplastics more generally should not be  
454 considered as a single polymeric compound, but a mixture of polymers, containing monomers and  
455 additives including metals, emollients, phthalates and flame retardants ([Rochman et al., 2019](#)). These  
456 additives can be highly toxic and have been associated with sub-lethal health effects and endocrine  
457 disruption in marine invertebrates ([Browne et al., 2013](#); [Cole et al., 2019](#)).

458

#### 459 **4.2 5-day assay**

460 Cockles were found to be much more sensitive to modern biocidal APPs when compared to  
461 ragworms. The LC<sub>50</sub> and EC<sub>50</sub> for cockles (2.3 g L<sup>-1</sup> and 1.4 g L<sup>-1</sup> respectively) were almost an order of  
462 magnitude lower than for ragworms (19.9 g L<sup>-1</sup> and 14.6 g L<sup>-1</sup> respectively). The maximum concentration  
463 of APPs found within the Plym Estuary at Hooe Lake ([Muller-Karanassos et al., 2019](#)), where boating  
464 activity is significant, was 18.8 g L<sup>-1</sup>, which is well above the LC<sub>50</sub> and EC<sub>50</sub> for cockles and above the EC<sub>50</sub>  
465 for ragworms. This suggests that APP concentrations found in the natural environment have the potential  
466 to cause mortality to cockles and adverse effects in ragworms.

467 Studies have shown that ragworms and cockles originating from metal-contaminated sites may  
468 have a higher tolerance to Cu and Zn compared to uncontaminated sites ([Durou et al., 2005](#); [Mouneyrac  
469 et al., 2003](#); [Naylor, 1987](#)). It is therefore not possible to generalise the findings of this study for all  
470 ragworm and cockle populations since the LC<sub>50</sub> for APPs will likely differ between sites. Indeed, ragworm  
471 and cockles could be identified in the vicinity of Hooe Lake, albeit in sparser numbers than elsewhere  
472 (personal observations). Metal tolerance has also been found to differ between taxa and more sensitive  
473 benthic organisms are expected to have lower APP LC<sub>50</sub> values. A study by [Buffet et al. \(2011\)](#) found that  
474 the bivalve mollusc *Scrobicularia plana* was less tolerant to Cu nanoparticles compared to *H. diversicolor*,  
475 supporting the findings of the current study. found a 4-day LC<sub>50</sub> value of 0.14 % of APPs by mass of dry  
476 sediment for the epibenthic copepod *Nitokra* sp. The current study showed a similar 5-day LC<sub>50</sub> (0.15 % of  
477 APPs by mass of wet sediment) for *C. edule* and a much higher 5-day LC<sub>50</sub> (1.42 % of APPs by mass of wet  
478 sediment) for *H. diversicolor*. These values are not directly comparable since the exposure periods differ  
479 and sediment wet weight was used instead of dry weight. However, it can be assumed that if water were  
480 removed from sediment in the current study this would produce higher percentages of APPs by mass of

481 dry sediment. This suggests that cockles are more tolerant than copepod species to APP-contaminated  
482 sediments, likely due to their larger body size and ability to accumulate metals.

483

#### 484 **Conclusions**

485 Given the current evidence, it can be concluded that biocidal APPs present a source of  
486 contaminant metals to estuarine and coastal sediments. Exposure to these anthropogenic particles pose  
487 both a physical and toxic risk to benthic species and the wider food web, necessitating stricter regulations  
488 for antifouling waste in marinas and boatyards. APPs derived from copper-based biocidal paints proved  
489 most toxic to cockles and ragworms. While exposure to silicone-based non-biocidal APPs resulted in  
490 increased weight-loss in cockles, current evidence indicates non-biocidal antifouling paints to be less toxic  
491 to non-target organisms.

492

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495

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