Supporting Information

Synthesis, structure, spectroscopic studies and magnetic properties of Cu$_2$N$_2$O$_4$-, Cu$_2$N$_2$O$_2$(S$_2$)-, Cu$_2$N$_2$S$_4$-chromophores based on aminomethylene derivatives of pyrazole-5-one(thione)

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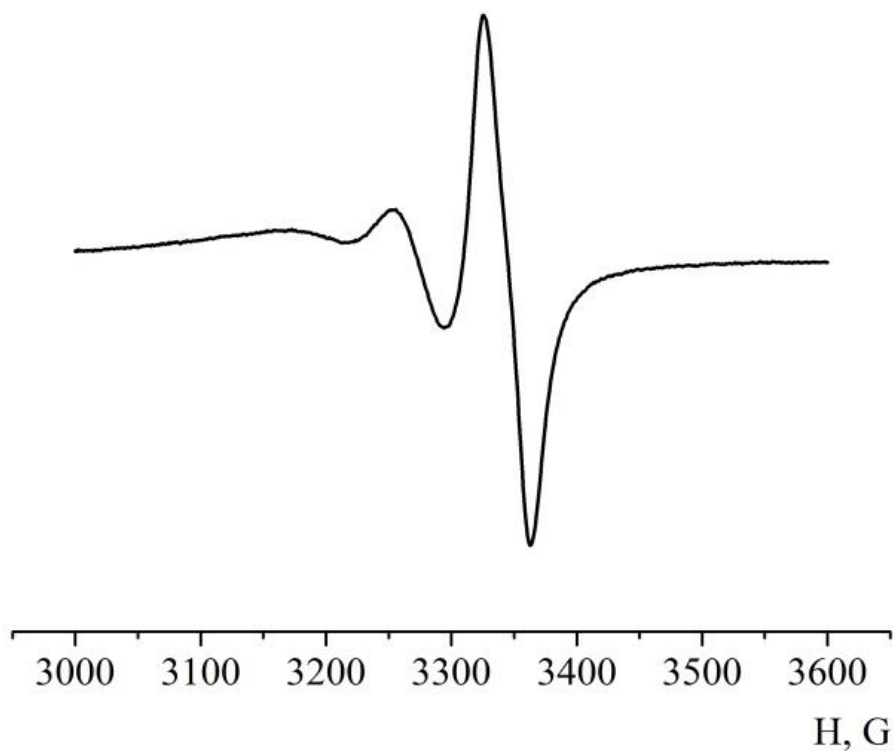
*1. EPR spectra of the complexes **10-13**.*

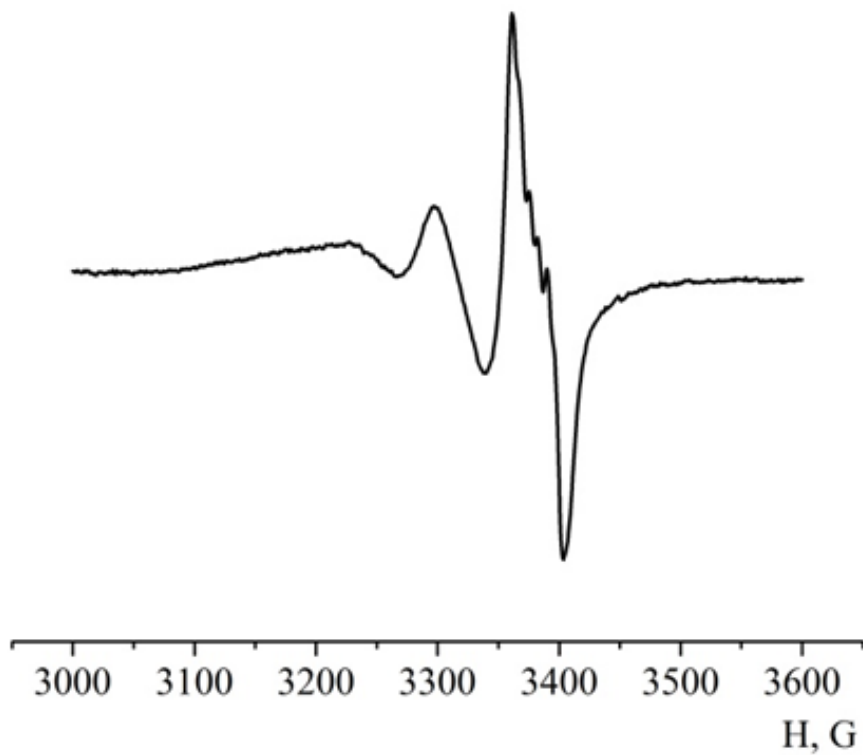**Fig. S1**. Experimental ESR spectrum of complex **10** (DMSO, 290 K, g = 2.13, a^{Cu} = 75 G).

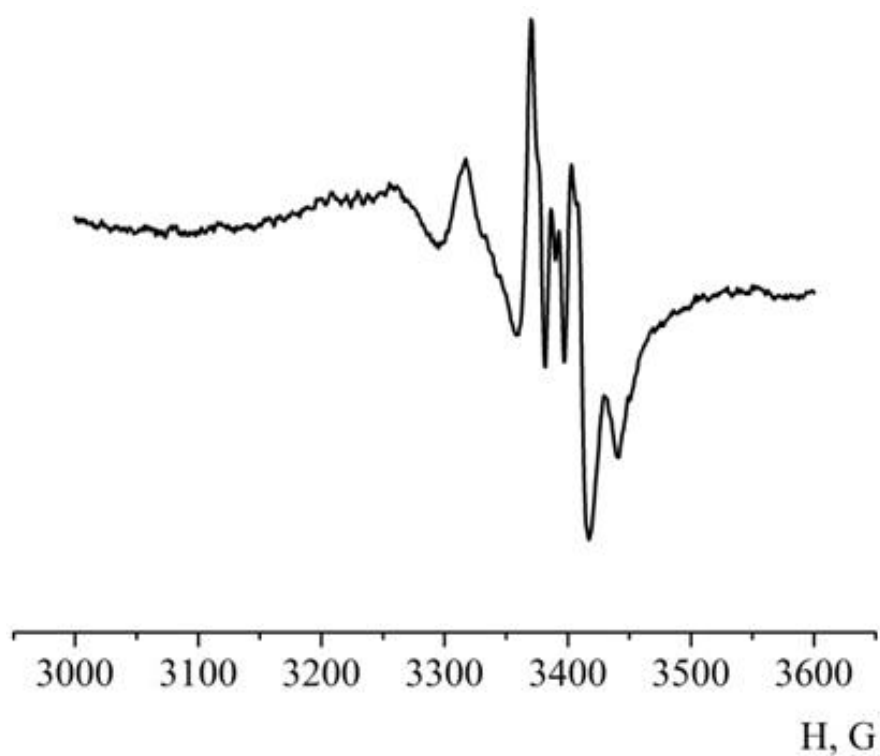**Fig. S2**. Experimental ESR spectrum of complex **11** (DMSO, 290 K, g = 2.10, a^{Cu} = 67 G).

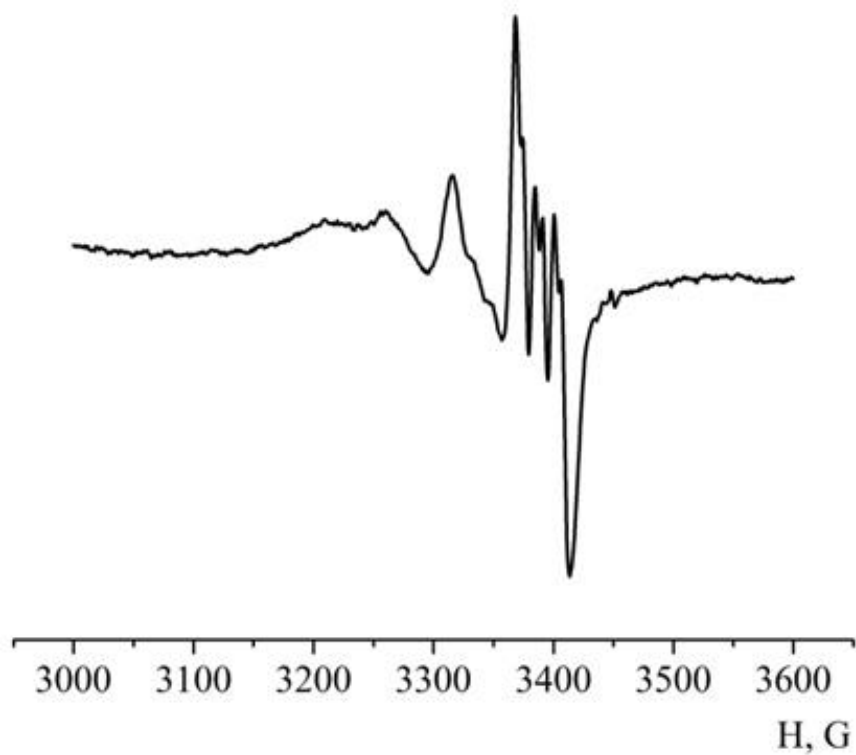**Fig. S3**. Experimental ESR spectrum of complex **12** (DMSO, 290 K, g = 2.08, a^{Cu} = 57 G).

Fig. S4. Experimental ESR spectrum of complex **13** (DMSO, 290 K, g = 2.08, a^{Cu} = 57 G).

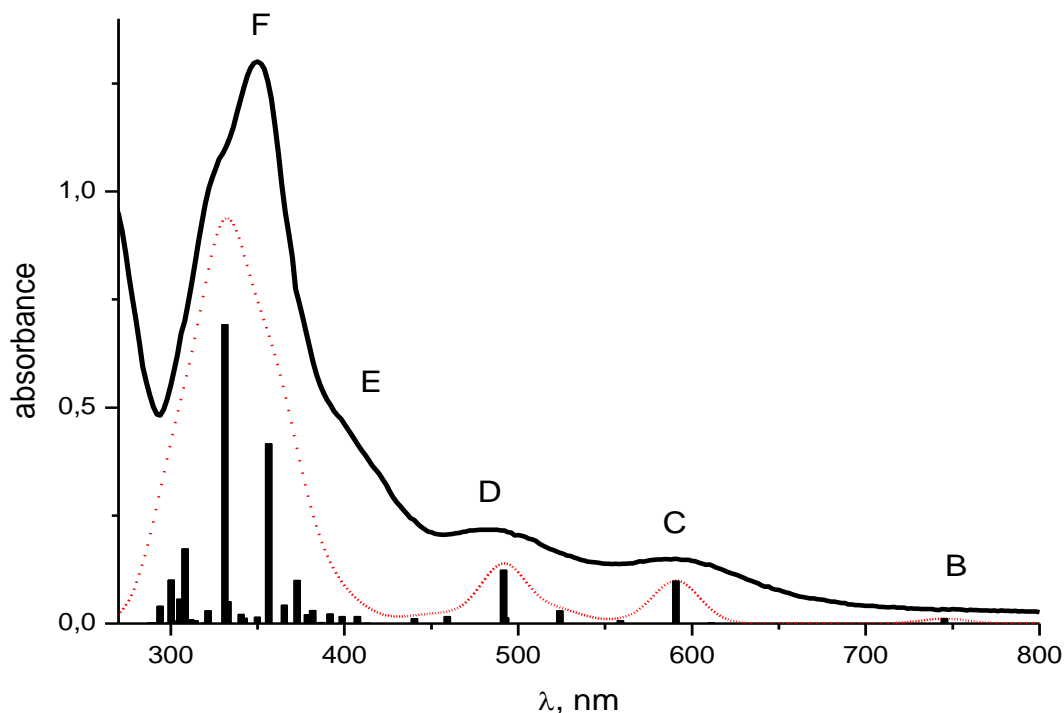## 2. Molecular orbital description and electronic spectra

Fig. S5. The experimental (solid black line) and calculated (dash red line) electronic absorption spectra of complex **12** in CH_2Cl_2 solution. The vertical lines show the wavelengths of electronic transitions, the ordinates of lines of equal values of the oscillator strength of the corresponding electronic transitions.

Table S1. Calculated wavelengths (λ), energies (E), oscillator strengths (f), involved molecular orbitals and their contributions for different electronic transitions in complex **12** according to TD-DFT calculations.

λ_{cal}, nm	E, eV	Electronic transitions, (contribution, %)	f	Character
1078.13	1.150	HOMO → LUMO (52%)	0.03	$\pi_L \to \pi^*_L$ LLCT
745.36	1.663	HOMO-2 → LUMO (92%)	0.01	$\pi_L \to d$ LMCT $n_{ter}(S) \to n^*_{br}(S)$
590.78	2.099	HOMO-3 → LUMO (78%)	0.10	$\pi_L \to d$ LMCT $\pi_L \to n^*_{br}(S)$

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491.63	2.522	HOMO-6 → LUMO (57%) HOMO-14 → LUMO (11%)	0.12	$\pi_L \to$ d LMCT $\pi_L \to n^*_{br}(S)$
407.42	3.043	HOMO-22 → LUMO (12%) HOMO-19 → LUMO (14%) HOMO-14 → LUMO (12%) HOMO-10 → LUMO (14%)	0.02	$\pi_L \to$ d LMCT $\pi_L \to n^*_{br}(S)$
356.45	3.478	HOMO-4 → LUMO+1 (56%)	0.42	$\pi_L \to n^*_{br}(S)$
331.21	3.743	HOMO-17 → LUMO (20%) HOMO-14 → LUMO (10%) HOMO-12 → LUMO (19%)	0.69	$\pi_L \to$ d LMCT

Table S2. The energies and fragment compositions of MOs for complex **12.**

N	MO (α)	Energy, eV	Cu1	Cu26	$S2_{ter}$	$S27_{ter}$	$S3_{br}$	$S28_{br}$	L1	L2
175	HOMO-22	-8.272	0.18	0.3	0.03	0.04	0.03	0.04	0.15	0.23
176	HOMO-21	-8.182	0.19	0.18	0.03	0.03	0.01	-	0.27	0.27
177	HOMO-20	-8.171	0.33	0.33	0.02	0.03	0.01	0.02	0.13	0.13
178	HOMO-19	-8.082	0.22	0.14	0.05	0.04	0.04	0.05	0.27	0.2
179	HOMO-18	-8.071	0.2	0.25	0.04	0.05	0.01	0.01	0.18	0.25
180	HOMO-17	-8.007	0.22	0.24	0.03	0.03	0.04	0.04	0.19	0.21
181	HOMO-16	-7.919	0.2	0.2	0.05	0.05	0.03	0.03	0.22	0.21
182	HOMO-15	-7.738	0.4	0.31	0.03	0.02	0.01	0.02	0.11	0.1
183	HOMO-14	-7.715	0.2	0.33	0.01	0.02	0.04	0.03	0.17	0.19
184	HOMO-13	-7.443	0.37	0.37	0.04	0.04	-	-	0.09	0.09
185	HOMO-12	-7.253	0.15	0.16	0.1	0.1	0.09	0.08	0.17	0.17
186	HOMO-11	-7.079	0.11	0.11	0.05	0.06	0.06	0.06	0.26	0.28
187	HOMO-10	-7.051	-	0.01	0.03	0.04	-	-	0.39	0.5
188	HOMO-9	-7.035	0.09	0.09	0.04	0.03	-	0.01	0.41	0.31

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189	HOMO-8	-6.955	0.03	0.03	0.02	0.02	0.02	0.03	0.43	0.42
190	HOMO-7	-6.906	0.05	0.05	0.07	0.07	0.04	0.04	0.34	0.34
191	HOMO-6	-6.594	0.10	0.13	0.03	0.03	0.06	0.1	0.22	0.33
192	HOMO-5	-6.588	0.12	0.1	0.01	-	0.16	0.12	0.29	0.18
193	HOMO-4	-6.113	0.06	0.06	0.02	-	0.02	-	0.42	0.43
194	HOMO-3	-6.021	0.05	0.05	0.07	0.06	0.03	0.03	0.36	0.35
195	HOMO-2	-5.780	0.03	0.02	0.24	0.13	-	-	0.34	0.23
196	HOMO-1	-5.758	0.05	0.06	0.16	0.27	-	-	0.17	0.28
197	HOMO	-5.021	0.18	0.18	0.07	0.07	0.13	0.13	0.12	0.12
198	LUMO	-3.355	0.19	0.19	0.05	0.05	0.15	0.15	0.11	0.11
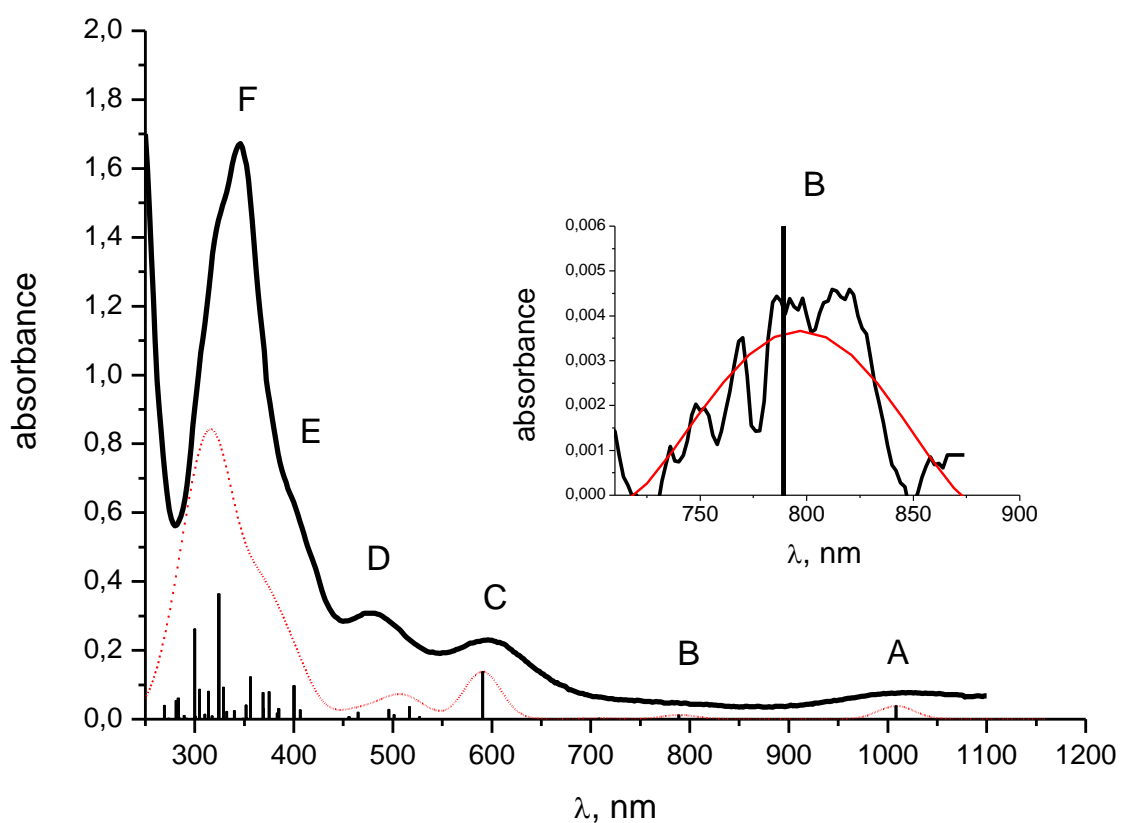
Fig. S6. The experimental (solid black line) and calculated (dash red line) electronic absorption spectra of complex **13** in CH$_2$Cl$_2$ solution. The vertical lines show the wavelengths of electronic transitions, the ordinates of lines of equal

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values of the oscillator strength of the corresponding electronic transitions. The large-scale band *B* is shown in the inset.

Table S3. Calculated wavelengths (λ), energies (E), oscillator strengths (f), involved molecular orbitals and their contributions for different electronic transitions in complex **13** according to TD-DFT calculations.

λ_{cal}, nm	E, eV	Electronic transitions, (contribution, %)	f	Character
1008.34	1.230	HOMO → LUMO (53%)	0.04	$\pi_L \to \pi^*_L$ LLCT
789.06	1.571	HOMO-2 → LUMO (85%)	0.01	$\pi_L \to d$ LMCT $n_{ter}(S) \to n^*_{br}(S)$
590.62	2.099	HOMO-3 → LUMO (72%)	0.14	$\pi_L \to d$ LMCT $\pi_L \to n^*_{br}(S)$
495.88	2.500	HOMO → LUMO+2 (94%)	0.03	$d \to \pi^*_L$ MLCT $n_{br}(S) \to \pi^*_L$
400.18	3.098	HOMO-1 → LUMO+1 (69%) HOMO-15 → LUMO (10%)	0.10	LMCT $\pi_L \to d$ $\pi_L \to n^*_{br}(S)$
375.11	3.305	HOMO-10 → LUMO (42%) HOMO-2 → LUMO+2 (35%)	0.08	$n_{ter}(S) \to \pi^*_L$ nLST
356.28	3.480	HOMO-4 → LUMO+1 (51%) HOMO-3 → LUMO+2 (18%)	0.12	$d \to \pi^*_L$ MLCT
324.15	3.825	HOMO-5 → LUMO+1 (66%)	0.36	$n_{br}(S) \to \pi^*_L$ nLST

Table S4. The energies and fragment compositions of MOs for complex **13**.

N	MO (α)	Energy, eV	Cu1	Cu36	S2$_{ter}$	S37$_{ter}$	S3$_{br}$	S38$_{br}$	L1	L2
166	HOMO-15	-7.936	0.30	0.34	0.03	0.04	0.01	-	0.12	0.15
167	HOMO-14	-7.917	0.33	0.32	0.05	0.05	0.01	0.02	0.12	0.10

168	HOMO-13	-7.790	0.32	0.30	0.01	-	0.05	0.05	0.13	0.13
169	HOMO-12	-7.659	0.21	0.20	0.02	0.02	0.04	0.04	0.24	0.24
170	HOMO-11	-7.578	0.34	0.33	0.05	0.05	-	-	0.11	0.11
171	HOMO-10	-7.374	0.06	0.05	0.03	0.03	0.01	0.01	0.40	0.41
172	HOMO-9	-7.274	0.28	0.29	0.04	0.04	-	-	0.17	0.17
173	HOMO-8	-7.171	0.18	0.16	0.11	0.11	0.08	0.08	0.15	0.14
174	HOMO-7	-6.842	0.12	0.12	0.06	0.06	0.07	0.07	0.25	0.25
175	HOMO-6	-6.616	0.10	0.10	0.02	0.02	0.06	0.06	0.32	0.32
176	HOMO-5	-6.499	0.12	0.12	0.04	0.04	0.13	0.13	0.21	0.21
177	HOMO-4	-6.157	0.10	0.10	0.05	0.05	0.07	0.07	0.28	0.29
178	HOMO-3	-6.013	0.07	0.07	0.13	0.13	0.05	0.05	0.26	0.25
179	HOMO-2	-5.784	0.04	0.04	0.12	0.14	0.03	0.03	0.29	0.31
180	HOMO-1	-5.727	0.06	0.06	0.21	0.19	-	-	0.25	0.22
181	HOMO	-5.086	0.17	0.17	0.09	0.09	0.11	0.11	0.13	0.13
182	LUMO	-3.377	0.20	0.20	0.06	0.06	0.13	0.13	0.11	0.11
183	LUMO+1	-2.050	0.02	0.02	0.03	0.03	-	-	0.43	0.45
184	LUMO+2	-1.888	0.05	0.05	0.02	0.02	-	-	0.44	0.41