

# NIH Public Access

**Author Manuscript** 

Mol Cell. Author manuscript; available in PMC 2014 August 22.

Published in final edited form as:

Mol Cell. 2013 August 22; 51(4): 531–538. doi:10.1016/j.molcel.2013.07.012.

## Protons as second messenger regulators of G protein signaling

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## Summary

In response to environmental stress cells often generate pH signals that serve to protect vital cellular components and reprogram gene expression for survival. A major barrier to our understanding of this process has been the identification of signaling proteins that detect changes in intracellular pH. To identify candidate pH sensors we developed a computer algorithm that searches proteins for networks of proton-binding sidechains. This analysis indicates that G subunits, the principal transducers of G protein-coupled receptor signals, are pH sensors. Our structure-based calculations and biophysical investigations reveal that G subunits contain networks of pH-sensing sidechains buried between their Ras and helical domains. We show further that proton binding induces changes in conformation that promote G phosphorylation and suppress receptor-initiated signaling. Together, our computational, biophysical and cellular analyses reveal a new and unexpected function for G proteins as mediators of stress-response signaling.

## Introduction

Coordinated changes in intracellular pH are triggered by extracellular chemical signals (Hellwig et al., 2004), nutrient limitation (Young et al., 2010), ischemia (Schroeder et al., 2010), and other cell stresses (Webb et al., 2011). This ability to adjust cytoplasmic pH is an essential feature of cellular physiology that enables a rapid response to adverse growth conditions. However, we have limited knowledge of the specific cellular components responsible for detecting and promoting changes in intracellular proton abundance. The identification and characterization of new pH-sensing proteins, particularly those in established signaling pathways, would lend support to the emerging view that protons can act as second messenger regulators of cell signaling, survival, and stress responses (Dechant et al., 2010; Rubenstein and Schmidt, 2010).

pH sensors are specialized proteins that couple their biological function to physiological changes in cellular pH (6 < pH < 8) (Srivastava et al., 2007). Perhaps the best-known proton

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sensor is hemoglobin, which exhibits reduced affinity for oxygen as a consequence of respiratory acidosis in the peripheral vasculature (the so-called Bohr effect) (Perutz, 1978). Hemoglobin and other pH sensors detect changes in cellular pH using sidechains (Asp, Glu, His, Lys, Arg) or sidechain modifications (phosphorylation) that bind protons. In these sensors, proton-binding groups are organized in structural arrangements that shift their  $pK_a$  values into the physiological pH range ( $6 < pK_a < 8$ ). Prominent examples of these motifs include clusters of His residues and networks of surface-exposed basic and acidic sidechains. Lesser known, yet equally important, are examples of proton-binding motifs sequestered within protein cores or buried at the interface of protein-protein complexes (Tews et al., 2005). Although counterintuitive, pH-sensing by these core motifs exploits the effects of dehydration, which act to facilitate reversible proton binding to core sidechains by shifting their  $pK_a$  values into the physiological pH range (Isom et al., 2008; Isom et al., 2011a; Isom et al., 2010a).

In an effort to predict and validate new pH sensors, we have developed a new computer algorithm (pHinder) that searches protein structures for buried proton-sensing motifs. The strategy of limiting our structure-based analysis to core networks is based on several compelling observations. Buried ionizable sidechains in proteins are relatively rare (Bush and Makhatadze, 2011; Gunner et al., 2000; Kajander et al., 2000; Kim et al., 2005; Rashin and Honig, 1984), often have  $pK_a$  values within the physiological pH range (Isom et al., 2011a; Isom et al., 2010a), and are readily identifiable from protein structure. Furthermore, the large  $pK_a$  shifts exhibited by buried sidechains correspond to sizeable increments of Gibbs free energy (1.36 kcal/mol per  $pK_a$  unit) that can be harnessed by proteins to drive pH-dependent changes in structure, stability, and function. In contrast, surface-exposed ionizable residues are far more common, rarely exhibit highly perturbed  $pK_a$  values, and rarely induce large changes in protein stability and structure.

Here we report the use of our pHinder algorithm to identify a new pH sensor involved in cell signaling. This work reveals that G subunits, the principal transducers of G protein-coupled receptor signals (Neves et al., 2002; Rasmussen et al., 2011), use buried networks of acidic and basic sidechains to detect physiological changes in cellular pH. We show further that proton binding regulates G conformation and phosphorylation in a manner that suppresses GPCR signaling in cells. Taken together, our findings reveal a new and unexpected role for G proteins as integrators of hormone and metabolic stress-response pathways.

## Results

#### Ga subunits contain pH-sensing structural motifs

pHinder represents a structural proteomics calculation designed to infer pH-sensing capabilities from protein structure. This new algorithm predicts candidate pH sensors by identifying spatial networks of titratable sidechains using an automated, structure-based procedure (See Methods for details). Using pHinder to survey 11,890 non-redundant structures in the Protein Data Bank (PDB), we found that only 10% of proteins contained large (6 sidechains) proton-binding networks residing in their cores (Figures 1A and 1B). As expected, these calculations identified several documented pH sensors including phosphofructokinase (Trivedi and Danforth, 1966), gelsolin (Nag et al., 2009), and soluble-adenylyl cyclase (Tews et al., 2005). In addition, we made the unexpected observation that G subunits contained large core networks located at the interface of their Ras and helical domains (Figure 1C). When we expanded our pHinder analysis to all G structures (129 total structures) we found that 96% contained core networks (Figure 1A). In contrast, core networks were absent in small GTPases, which contain only the Ras-like domain (Figure 1B). These findings suggested that G subunits were pH sensors, and prompted us to

quantify the effect of pH on G structure and function, using biochemical and biophysical approaches, and the effect of pH on G signal transmission, using a model GPCR pathway in yeast (Figure 1D).

#### Proton binding regulates Ga stability and conformation

Within GPCR signaling pathways, G subunits function as signaling switches that are turned on by binding to GTP and turned off by GTP hydrolysis. As expected for a pH sensor, the sidechains responsible for these G functions are found in close proximity to the pH-sensing network identified by pHinder. However, these catalytic sidechains also reside exclusively in the Ras-like domain, which itself was not predicted to be pH sensitive (Figure 1B). As anticipated, the effects of pH on nucleotide binding (Figure S1) and GTP hydrolysis (Figure 2A and Figure S1) were negligible in representative G subunits from yeast (Gpa1) and mammalian (G i) sources. Given that the pH-sensing network was positioned between the Ras-like and helical domains, we next considered whether there may be pH-driven changes in G stability, dynamics, and conformation. Indeed, the Ras and helical domains undergo a dramatic reorientation in response to receptor-mediated activation of the G protein (Chung et al., 2011; Rasmussen et al., 2011; Van Eps et al., 2011; Wall et al., 1995; Westfield et al., 2011).

We started by examining the effect of pH on Gpa1 and G i thermostability. Cooperative, pH-dependent changes in protein stability are an indicator of pH sensing. These changes reflect the thermodynamic coupling of global protein stability to the ionization free energy of proton-binding sidechains (Isom et al., 2008; Isom et al., 2011a; Isom et al., 2010a). Our thermostability measurements (Figure 2B and Figure S1), quantified by the midpoint of thermal unfolding ( $T_m$ ) (Isom et al., 2011b; Isom et al., 2010b), indicated that both Gpa1 and G i were cooperatively sensitive ( $T_m$  of 8–10°C) to changes in pH when bound to GTP S, a GTP analog that is resistant to hydrolysis. The resulting thermal stability profiles, which report an apparent p $K_a$  value of 7, indicate that the core networks within G -GTP S contain ionizable (acidic and basic) sidechains having  $pK_a$  values shifted into the physiological pH range. It is unlikely that this pH-dependence originates from the titration of the -phosphate of GTP S, given the available evidence that the -phosphate is always charged above pH 6 due to a highly depressed pKa value (Schweins et al., 1995). As expected, mutating components of the core network dramatically altered the pH-dependent thermostability of GTP S-bound Gpa1 and G i (Figure S3), or resulted in G variants that failed to express (e.g. Lys270 and Lys 277 of G i). Based on these findings, we conclude that Gpa1 and G i are equipped to function as proton sensors.

In contrast to ]GTP-bound G, there was a minimal effect of pH on the thermostability of the GDP-bound protein (Figure 2B and Figure S1). The difference between GDP- and GTPyS-bound G can be attributed to a number of underlying factors that, in some cases, limit the utility of our thermostability assay. First, the conformation of the GDP-bound state, which is distinctly different from the GTP S-bound state, may have a smaller effective core network with fewer shifted  $pK_a$  values. This idea is supported by our pHinder calculations (Figure 1 A), which revealed that GDP-bound G conformations tend to have smaller, or fragmented core networks. Second, multiple  $pK_a$  shifts caused by the GDP-induced conformation may offset one another causing no net change in pH-dependent thermostability. Third, the presence of the negatively charged -phosphate on GTP S, which is missing in GDP, may alter the pKa values of the one or more sidechains exclusively in the GTP S-bound conformation. Fourth, the effects of pH sensing may be largely uncoupled from G thermal stability. Since our thermostability assay could not differentiate between these four scenarios, we investigated the relationship between pH, nucleotide occupancy, and G conformation using intrinsic tryptophan fluorescence, quantitative mass spectrometry, and NMR spectroscopy.

Intrinsic tryptophan fluorescence is often used to monitor the structural relationship between nucleotide binding and G protein activation (Higashijima et al., 1987). Here, we adapted the intrinsic fluorescence assay to interrogate the effect of pH on the conformation of GDP- and GTP S-bound G subunits. The titration curves (Figure 2C and Figure S1) derived from the intrinsic fluorescence spectra of Gpa1 and G i (see Figure S1) revealed that both nucleotide-bound states exhibit cooperative changes in conformation as a function of pH. As expected, the pH titration of GTP S-bound G corresponded to a higher level of intrinsic fluorescence than GDP-bound G (Figure 2C). This difference originates from a tryptophan residue (W211 in G i) within a regulatory motif (referred to as switch II) known to be sensitive to GTP binding. However, because G subunits contain multiple tryptophan residues, it is likely that the pH-dependent changes in intrinsic fluorescence we observe in the presence of both GDP and GTP S reflect additional contributions from tryptophan residues elsewhere in the Ras-like and helical domains. Remarkably, these changes were similar in magnitude to that which accompanies GTP S binding (Figure 2C), and G protein activation by receptors.

Next we corroborated our intrinsic fluorescence measurements by examining cysteine protection and backbone conformation as a function of pH. Using fast quantitative cysteine reactivity (fQCR) (Isom et al., 2011b; Isom et al., 2010b) and liquid chromatography-mass spectrometry (LC-MS) we identified specific cysteine sites that were exposed by pHdependent conformational changes in mammalian G i (Figures 2D-E). Two protected cysteines (Cys66 and Cys325) at the interface of the Ras-like and helical domains became less protected as pH was reduced from 7.0 to 6.0. However, the reactivity of two cysteines buried within the Ras (Cys254) and helical domains (Cys139) was unaffected by pH. This pattern of cysteine labeling corroborated our intrinsic fluorescence data, and was consistent with a pH-driven conformational change that displaces the helical domain from the Ras-like domain. We were also able to observe this pH-dependent change in G i conformation by NMR spectroscopy (Figure 2F and S2). Our NMR spectra revealed that G i sampled multiple conformations at pH 7.0, but sampled single conformations at pH 6.0 and 8.0. Once again these behaviors mimic those resulting from receptor-dependent activation of the G protein (Chung et al., 2011; Rasmussen et al., 2011; Van Eps et al., 2011; Wall et al., 1995; Westfield et al., 2011).

#### Proton binding regulates Gα phosphorylation and signaling

Taken together, our *in vitro* experiments established that a network of core ionizable sidechains regulate G conformation in response to changes in pH. Significantly, these pH-sensing residues lie within the active site of the protein and become exposed following GCPR activation (Rasmussen et al., 2011; Westfield et al., 2011). To assess the biological relevance of G proton sensing, we monitored GPCR signaling as a function of cellular pH. We did these experiments in yeast because it was possible to manipulate cellular pH genetically (by gene knockdown of proton pumps), metabolically (via glucose limitation) (Orij et al., 2011), and biochemically (using the metabolite acetic acid) (Orij et al., 2011), while simultaneously monitoring the activation of pheromone pathway components. To quantify cellular pH we used the geneticallyencoded pH sensor, pHluorin (See Figure S4) (Miesenbock et al., 1998; Orij et al., 2011; Orij et al., 2009).

Using all three perturbations we found that intracellular proton concentrations increased by up to 10-fold before recovering to baseline (Figures 3A–B). Notably, the speed, duration, and magnitude of these changes resembled the behaviors of other wellknown second messengers, such as cAMP and calcium. Our experiments also revealed that yeast G was rapidly phosphorylated as cellular pH was lowered, and reversibly dephosphorylated as cellular pH recovered to resting levels (Figures 3D–E). Thus, the dynamics of G phosphorylation mirrored the changes in cellular pH. Moreover, we found that pH-

dependent phosphorylation *in vivo* precisely matched the pH-dependent change in G conformation observed *in vitro* (Figures 4A–B). The kinase responsible for G phosphorylation (Elm1) (Hong et al., 2003; Torres et al., 2011) was itself not activated by pH, as another substrate for Elm1 (Snf1) was not phosphorylated under identical conditions (See Figure S3). Because Gpa1 phosphorylation was dependent on pH, and Elm1 activity was not, we conclude that proton binding within the G core network controls access to the kinase. Similar mechanisms of conformation-driven substrate phosphorylation have been reported for AMP-activated protein kinases and for G protein-coupled receptor kinases (Benovic et al., 1986; Rubenstein et al., 2008).

Last we assessed the effect of cellular pH on GPCR signaling at the level of the effector MAPK (Figures 4C-F). These experiments revealed that GPCR signaling in yeast is pHdependent, and maximal at pH 6.6 (Figures 4C–D). Moreover, although pheromone signaling could involve multiple proton sensors, there are several reasons to conclude that G is mostly responsible for the pH-dependent change in MAPK activity. First, pHdependent activation of two MAPKs, Fus3 (which requires the scaffold Ste5) and Kss1 (which does not), was identical (Figures 4C–D), indicating that the proton sensor acts upstream of the scaffolded MAPK cascade. Second, neither MAPK was activated in yeast lacking the G subunit (Ste4) (Figure 4E), establishing that protons were unlikely to regulate components downstream of the G protein heterotrimer. Third, deletion of the GTP aseactivating protein Sst2, which acts directly on G, led to increased MAPK activation (as expected), yet activation remained sensitive to changes in pH (Figures 4E–F). While several mammalian GPCRs are activated by binding to extracellular protons (Ludwig et al., 2003), this has not been observed in yeast. These genetic experiments indicate that pH sensing occurs at the level of the G protein. More generally, our computational, biophysical and cellular analysis reveals that the G protein functions as a pH sensor as well as a transducer of GPCR signaling.

## Discussion

Although every protein binds to and releases protons, few cellular proteins are bona fide pH sensors. Prominent examples include proteins involved in oxygen transport (hemoglobin), metabolism (phosphofructokinase), and signal transduction (soluble adenylyl cyclase). Each of these proteins contains proton-binding sidechains with  $pK_a$  values in the physiological range, and each is responsive to the transient increases in proton concentration caused by cellular stress conditions (6 < pH < 7.2). This situation is analogous to calcium sensors, such as calmodulin, that have binding affinities tuned to detect transient increases in calcium concentration elicited by extracellular chemical signals (0.1 to 1  $\mu$ M) (Clapham, 2007).

Although it has long been appreciated that changes in cellular pH are essential to proper cell function, the molecular details that underlie pH signaling and regulation remain poorly understood. In this context, identifying candidate pH sensors and predicting the  $pK_a$  values of their pH-sensing sidechains has been a major challenge (Alexov et al., 2011; Nielsen et al., 2011). Our pHinder software works to address this issue by identifying spatial arrangements of proton-binding sidechains that are likely to confer pH-sensing capabilities. Here we used the pHinder program to identify G subunits as pH sensors. We showed further that these proteins couple physiological changes in intracellular pH to large changes in cellular signaling.

Given that G proteins are conserved across different biological kingdoms, it is likely that pH sensing is a general feature of GPCR-mediated signaling in a variety of physiological contexts. For instance, in humans the ability to detect changes in cellular pH is important under conditions of ischemia and hypoxia (Bopassa, 2012), where diminished oxygen and

glucose lead to the anaerobic metabolism of glycogen and intracellular acidosis. Similarly, in plants and fungi, acidic pH signals are critical for nutrient acquisition and sensing (Gjetting et al., 2012; Orij et al., 2011), responding to osmotic stress (Kader and Lindberg, 2010), and protecting vital cellular machinery (Peters et al., 2013). Within these contexts, proton signals facilitate signal transduction in a manner analogous to other known second messengers.

In summary, our computational, biophysical and cellular analyses reveal that the G protein functions as a pH sensor as well as a transducer of GPCR signaling. Furthermore, our findings indicate that protons can disrupt ]G protein-mediated signaling to limit hormone or neurotransmitter signaling during brief periods of nutrient or oxygen deprivation. In this way, it appears that evolution has adapted pre-existing signaling proteins to detect proton second messengers and reprogram cellular physiology in response to cell stress.

## **Experimental Procedures**

#### pHinder calculation

The accession codes for the non-redundant PDB chain set used in this study (compiled on 12/26/2012) were downloaded from the National Center for Biotechnology Information (NCBI). The non-redundant database contained 12,486 unique PDB chain entries (out of a total of 81,418 chains). A subset of protein chains (596) that were much smaller (< 25) or larger (> 1000 residues) than typical proteins in the reference set was removed, bringing the total number of pHinder-analyzed protein chains to 11,890. The sets of 129 G proteins (PF00503) and 595 small GTPases (PF00071) were identified using the pFam database (Wellcome Trust Sanger Institute). A manuscript describing the pHinder algorithm, and web-based access to the pHinder software, are currently in preparation. Briefly, a Delaunay triangulation of the C atoms comprising the protein backbone was calculated using the 4D paraboloid method (O'Rourke, 1998). This triangulation was then used to calculate a surface (comprising triangular facets) that encapsulated the volume defined by the fold of protein backbone. Using a distance-to-plane calculation, the facets of the surface were used to classify the terminal atom of each sidechain as exposed (outside of the surface), marginal (intersecting or beneath the surface by  $< 3\text{\AA}$ ), or core (lying beneath the surface by  $= 3\text{\AA}$ ). Based on this classification, core ionizable networks were identified using a three-step procedure. First, the terminal atom of each core ionizable sidechain was triangulated with the terminal atoms of the other ionizable sidechains in its immediate microenvironment (defined by a 10 Å distance cutoff). These surrounding sidechains could be classified as marginal or exposed, thus some networks could extend from the core to the protein surface. Second, using a node-to-node distance cutoff of 10 Å, the triangulation was searched for contiguous networks of ionizable sidechains. Third, any two networks containing shared components were consolidated into a single network. This consolidation step was repeated iteratively until all incident networks were matched and combined. The total number of interconnected nodes within each unique network comprised the pHinder score. Although a protein chain may have contained multiple core networks, here we reported only the largest network per chain (Figures 1A-B).

#### Preparation of stock solutions

Concentrations of GDP (Sigma-Aldrich, G7127), GTP (Sigma-Aldrich, G8877), and GTP S (Roche Diagnostics, 11110349001) stock solutions were quantified spectroscopically using a molar extinction coefficient at 254 nm of 13,700  $M^{-1}$  cm<sup>-1</sup>. A 500 mM stock solution of the fluorogenic cysteine probe 4-(aminosulfonyl)-7-fluoro-2,1,3-benzoxadiazole (ABD) (TCI America, A5597) was made in DMSO, aliquoted, and stored at  $-20^{\circ}$ C. The concentration of all ABD solutions was quantified spectroscopically using a molar

extinction coefficient at 313 nm of 4200  $M^{-1}$  cm<sup>-1</sup>. Working stocks of ABD (26 mM) were made by combining 500 mM ABD stock in DMSO with 10  $\mu$ L of PBS (pH 7.0) and 179.6  $\mu$ L of water. It was necessary to store working ABD stocks at room temperature, instead of on ice, to prevent precipitation. All stock solutions of the reducing agent Tris(2-carboxyethyl)phosphine (TCEP) (TCI America, T1656) were made either fresh in water, or by adding TCEP crystals directly to premade buffer solutions.

#### **Protein production**

The proteins used in this study were overexpressed to high density using the auto-induction introduced by F.W. Studier (Studier, 2005). All purification steps were done in a PBS-GMT buffer (25 mM KPO<sub>4</sub>, 100 mM KCl, 50  $\mu$ M GDP, 50  $\mu$ M MgCl<sub>2</sub>, 1 mM TCEP, pH 7.0). The remaining details of our purification protocol can be found in Supplemental Experimental Procedures.

#### pH dependence of nucleotide hydrolysis

Steady-state GTP hydrolysis by yeast Gpa1 and mammalian G i was initiated by combining 800  $\mu$ L of 2X HOT-PBS-GMT (10  $\mu$ M GTP, 2 mM MgCl<sub>2</sub>, 2 mM TCEP, 1  $\mu$ L GTP <sup>32</sup>P, pH 7.0) with 800  $\mu$ L of protein sample (1  $\mu$ M protein in PBS) at 37°C. After each time interval, 90  $\mu$ L aliquots (in duplicate) were quenched in 750  $\mu$ L of icecold mixture of activated charcoal (5% w/v charcoal (Sigma-Aldrich, 260010), 50 mM phosphoric acid), vortex-mixed for 2 seconds, pulse-centrifuged to sequester protein and nucleotides in the pelleted charcoal fraction, and 200  $\mu$ L of supernatant (containing hydrolyzed <sup>32</sup>P<sub>i</sub>) was transferred to a clean PCR tube. The amount of <sup>32</sup>P<sub>i</sub> in each tube was quantified in a scintillation counter via Cerenkov counting. The data in Figure 2A and Figure S1 correspond to the mean (values) and standard deviation (error bars) of two independent experiments done at each of the indicated pH values.

#### pH dependence of Gα thermostability

Thermal unfolding of yeast Gpa1 and mammalian G i was quantified by fast quantitative cysteine reactivity (fQCR). The analysis of fQCR unfolding curves has been described previously (Isom et al., 2011b; Isom et al., 2010b). The details of the pH-dependent fQCR experiment can be found in the Supplemental Experimental Procedures.

#### pH dependence of Gα conformation

pH-dependent intrinsic fluorescence was measured using G protein stock (in PBS-GMT) diluted to 0.5 µM in PBS (pH 5.0, 5.5, 6.0, 6.5, 7.0, 7.5, or 8.0) with 5 µM GDP in a 1.5 mL tube. The samples were then vortex-mixed for 2 seconds, and equilibrated at ambient room temperature for 15 minutes. Because changes in chloride concentration affect the fluorescence measurement, the panel of PBS buffers used in this experiment was made as a self-consistent set by combining volumes of mono- and dibasic phosphate stock solutions composed of 25 mM potassium phosphate and KCl. Starting with pH 5.0, fluorescence spectra were collected in a quartz cuvette using an excitation wavelength of 280 nm, and matching excitation and emission bandwidths of 7.5 nm for G i, and 6.5 nm for Gpa1. The cuvette was then emptied and filled with the next protein sample (i.e. pH 5.5) without washing. Titration curves were assembled from the individual spectra using the fluorescence value at the emission wavelength of 345 nm. The data in Figure 2C and Figure S1 represent the mean (values) and standard deviation (error bars) of four experiments. In the case of G i, the pH-dependent fluorescence of the GTP S-bound state could not be measured because the amount of GTP S required for half-saturation (~50 µM) absorbed most of the excitation light. The detailed procedures for our pH-dependent mass spectrometry and NMR experiments can be found in the Supplemental Experimental Procedures.

#### In vivo quantification of intracellular pH and GPCR signaling

The details regarding our yeast strains and media, pHluorin calibration, pH manipulation procedures, and Western blotting can be found in the Supplemental Experimental Procedures.

## **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgments

H.G.D., D.G.I., V.S., R.B., and S.T.C. are supported by the National Institutes of Health (NIH) (grant GM059167 and GM073180). D.G.I. acknowledges the support of the UNC University Research Council and the James Moeser Award for Distinguished Research. V.S. acknowledges support from the National Science Foundation REU Site Award 1156840 to UNC-Chapel Hill. D.M.S acknowledges funding from the NIH (grant CA016086) for support of the UNC Proteomics Center. We thank Sharon Campbell for assistance with NMR, Ashutosh Tripathy and Gary Johnson for access to instrumentation, and UNC Research Computing for access to computing clusters.

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## Highlights

- Cellular pH is altered by nutritional and metabolic stress
- A novel structure-based algorithm predicts pH sensing by G proteins
- Proton binding alters G conformation, phosphorylation and signal transmission
- G proteins serve as integrators of receptor- and stress-mediated cell signaling

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#### Figure 1.

G subunits contain structural motifs capable of proton sensing. (**A**, **B**) pHinder results showing that core networks of ionizable sidechains identified in 129 G structures (purple bars in **A**) were much larger and more prevalent than core networks identified in a reference set of 11,890 non-redundant PDB chains (gray bars in **A** and **B**) and 595 small GTPases (green bars in **B**). (**C**) G subunits consist of a helical (yellow) and Ras-like (gray) domain, whereas small GTPases consist of only a Ras-like domain. In G proteins the core pH-sensing network identified by pHinder and the sidechains responsible for G function reside at the interface of the helical and Ras-like domains. Collectively, the core networks observed in structures of mammalian G i consistently included residues Lys46, Asp150, Asp200, Asp229, Arg242, Lys270, and Lys277. (**D**) The core GPCR signaling components of the yeast pheromone pathway.

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#### Figure 2.

G subunits are proton sensors *in vitro*. (A) Steady-state GTP hydrolysis by yeast and mammalian G (See Figure S1) was diminished at pH 6.0 relative to pH 7.0 and 8.0. (B) The thermostability of GTP S-bound yeast and mammalian G (See Figure S1) was cooperatively dependent on pH (Hill coefficient of 2). (C) Intrinsic tryptophan fluorescence indicated that yeast and mammalian G subunits (See Figure S1) undergo a large-scale, pH-triggered conformational change. (D, E) Cysteine labeling and mass spectrometry of mammalian G i confirmed a pH-dependent conformational change. The cysteine sites that were exposed by the pH-triggered conformational change (Cys66 and Cys325) reside at the interface of the helical and Ras-like domains (E), the same region that contains the core network of pH-sensing residues (Figure 1D). (F) Overlay of heteronuclear single quantum coherence spectra of G i at pH 6.0 and 7.0 (See Figure S2). Error bars represent mean  $\pm$  SEM.

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#### Figure 3.

G proton sensing leads to pH-dependent G phosphorylation. Changes in yeast cellular pH caused by diminished glucose (**A**), the addition of acetic acid (**B**), and genetic knockdown of the proton pump Pma1 (**C**) using a doxycycline(dox)/Tet-off(TetO<sub>7</sub>) repression system. In all three cases cellular pH was quantified using the genetically-encoded pH biosensor pHluorin (See Figure S4). Resting intracellular pH in yeast is ~0.2 pH units lower than in mammalian cells. (**D**) The amount of phosphorylated G (pGpa1) increased during a sustained reduction in cellular pH caused by lowering glucose concentration from 2% (H) to 0.05% (L), and by genetic knockdown of Pma1 (TetO<sub>7</sub>-Pma1 + dox). (**E**) Acetic acid treatment (45 mM) caused a transient change in cellular pH (refer to panel **B**) that triggered

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a reversible and pH-dependent change in G  $\,$  phosphorylation. Error bars represent mean  $\pm$  SEM.

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#### Figure 4.

pH-dependent changes in G conformation lead to concomitant changes in Gpa1 phosphorylation and MAPK activation. (A) Titration of yeast cellular pH using acetic acid resulted in the pH-dependent phosphorylation of Gpa1 (pGpa1). (B) The pattern of Gpa1 phosphorylation *in vivo* matched the pH-triggered Gpa1 conformational change observed by intrinsic tryptophan fluorescence *in vitro*. The *in vitro* data corresponds to the pH-dependent intrinsic fluorescence of the Gpa1-GDP state (normalized and inverted relative to Figure 2C). (C, D) Titration of yeast cellular pH using acetic acid revealed the pH-dependent activation of the pheromone-responsive MAP kinases (pFus3 and pKss1). (E, F) Comparison of pH-dependent MAP kinase activation in WT yeast with yeast strains lacking the G subunit (*ste4*), or the GTPase-activating protein Sst2 (*sst2*). Data were normalized to MAPK activation in WT yeast at resting pH. Each sample (in panels A, C, and E) was treated with the indicated dose of acetic acid and 0.3  $\mu$ M -factor for 10 minutes. The pH

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reported for each sample corresponded to the observed cellular pH after 10 minutes of acetic acid treatment (See Figure 3B). Error bars represent mean  $\pm$  SEM.