# Design and synthesis of 6,7-methylenedioxy-4-substituted phenylquinolin-2(1H)-one derivatives as novel anticancer agents that induce apoptosis with cell cycle arrest at G2/M phase 

Yi-Fong Chen ${ }^{\text {a }}$, Yi-Chien Lin ${ }^{\text {b }}$, Po-Kai Huang ${ }^{\text {b }}$, Hsu-Chin Chan ${ }^{\text {c }}$, Sheng-Chu Kuo ${ }^{\text {b }}$, KuoHsiung Lee ${ }^{\mathrm{d}, \mathrm{e},{ }^{*}}$, and Li-Jiau Huang ${ }^{\mathrm{a}, \mathrm{b},{ }^{*}}$<br>${ }^{\text {aTh }}$ The Ph.D. Program for Cancer Biology and Drug Discovery, China Medical University, No. 91 Hsueh-Shih Road, Taichung, Taiwan 40402, R.O.C.<br>${ }^{\text {b }}$ Graduate Institute of Pharmaceutical Chemistry, China Medical University, No.91, Hsueh-Shih Road, Taichung, Taiwan 40402, R.O.C.<br>${ }^{\text {c Department of Biochemistry, School of Medicine, China Medical University, No.91, Hsueh-Shih }}$ Road, Taichung, Taiwan 40402, R.O.C.<br>${ }^{\mathrm{d}}$ Natural Products Research Laboratories, UNC Eshelman School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina 27599-7568, USA<br>${ }^{\text {e }}$ Chinese Medicine Research and Development Center, China Medical University Hospital, Taichung, Taiwan 40402, R.O.C.


#### Abstract

Novel 6,7-methylenedioxy-4-substituted phenylquinolin-2-one derivatives 12a-n were designed and prepared through an intramolecular cyclization reaction and evaluated for in vitro anticancer activity. Among the synthesized compounds, 6,7-methylenedioxy-4-(2,4- dimethoxyphenyl)quinolin-2(1H)-one (12e) displayed potent cytotoxicity against several different tumor cell lines at a sub-micromolar level. Furthermore, results of fluorescence-activated cell sorting (FACS) analysis suggested that 12e induced cell cycle arrest in the G2/M phase accompanied by apoptosis in HL-60 and H460 cells. This action was confirmed by Hoechst staining and caspase-3 activation. Due to their easy synthesis and remarkable biological activities, 4 -phenylquinolin- $2(1 H)$-one analogs (4-PQs) are promising new anticancer leads based on the quinoline scaffold. Accordingly, compound 12e was identified as a new lead compound that merits further optimization and development as an anticancer candidate.


## Keywords

4-Phenylquinolin-2(1H)-one (4-PQ); Anticancer agent; Podophyllotoxin; Apoptosis; Structureactivity relationships (SAR)

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## 1. Introduction

Cancer is the leading cause of death in economically developed countries and the second leading cause of death in developing countries. ${ }^{1}$ Development of novel antitumor agents with cytotoxicity against cancer cell lines is an important focus of oncology. Natural products are often referred to as an inexhaustible source of lead compounds that are likely to exhibit multiple biological activities, including anticancer. ${ }^{2,3}$ Because of their intrinsic biorelevance, natural products represent a significant source of inspiration for the design of structural analogues with new pharmaceutical profiles.

The microtubule network is an essential component of the cytoskeleton of eukaryotic cells. Podophyllotoxin (PPT, 1) is a aryltetralin lignan isolated from the roots of Podophyllum peltatum L., P. emodi W., or P. pleianthum H.. ${ }^{4}$ PPT inhibits tubulin assembly and polymerization through interaction with the protein at the colchicine binding site, thus preventing the formation of the spindle, and arresting cell division in metaphase (G2/M stage). ${ }^{5-7}$ Numerous efforts have been made to improve PPT's safety profile while maintaining its potency. Thus, its use as a lead agent in the development of new anticancer drugs has led to the discovery of semisynthetic derivatives, such as etoposide (2) and teniposide (3), and a more soluble prodrug of etoposide, named etopophos (4) (Fig. 1). Semisynthetic derivatives of PPT are currently in clinical use for the treatment of various malignancies.

PPT's structure is complex due to the presence of four chiral carbons in ring C (Fig. 1), which creates challenges for SAR (structure-activity relationship) studies.Research efforts to develop new analogues with simpler structures are ongoing ot identify the critical substructures for desired biological activity, and almost all rings (ring A to E) and positions on the cyclolignan skeleton have been modified. ${ }^{8}$ Importantly, elimination of the stereogenic centers at C-2 and C-3 removes the possibility of epimerization at C-2. Epimerization of the natural lignan has plagued its clinical development, because the rapidly formed in vivo cislactone metabolite is inactive. ${ }^{9}$ Numerous attempts have been made to obtain potent synthetically feasible analogues of PPT with a heteroatom incorporated in ring C to prevent epimerization. 4-Aza-podophyllotoxin (5) (Fig. 2) and related analogues were identified as structurally simpler derivatives of PPT with reported antiproliferative activity. ${ }^{10-12}$ In addition, podophyllic aldehyde (6) was found to be a highly selective antitumor agent against HT-29 colon and A-549 lung carcinomas. Several aldehydes related to this compound but with different configurations have been synthesized and evaluated for cytotoxic activities in neoplastic cell lines. ${ }^{13,14}$ These podophyllic aldehyde-related compounds lacked the lactone ring, but maintained cytotoxicity at, or under, micromolar level.

Another naturally occurring cis-stilbene, combretastatin A-4 (CA-4, 7) (Fig. 2) was isolated from the bark of the south African tree Combretum caffrum (Combretaceae). ${ }^{15}$ CA-4 is a potent tubulin assembly inhibitor and vascular disrupting agent at low concentrations. ${ }^{16}$ Its relative molecular simplicity suggests numerous practical approaches to the design of new antineoplastic agents, and several active stilbene-based compounds have been identified. ${ }^{17}$ The main disadvantage of CA-4 is the ready isomerization of its cis-double bond to an inactive trans-form during storage and administration. ${ }^{18}$ Consequently, this double bond in the cis-stilbene motif has often been used as a modification site and replaced with various heterocyclic or carbonyl functional groups. ${ }^{19-21}$

According to previous SAR studies on PPT and related analogues, structurally modified compounds that lack the trans- $\gamma$-lactone ring (ring D ) can show potent and selective cytotoxicity against cancer cells. Thus, the $\gamma$-lactone ring is not an essential feature for
cytotoxicity of PPT analogues. These results encouraged us to select 4-aza-podophyllotoxin (5) as the template and design much simpler structures without a $\gamma$-lactone ring. In our previous investigation, 2-phenylquinolin-4(1H)-one analogs (2-PQs) (Fig. 2) were identified as antimitotic agents. ${ }^{22}$ Recently, we shifted the phenyl ring on the quinolinone from the $\mathrm{C}-2$ to $\mathrm{C}-4$ position. Hence, a series of 4 -phenylquinolin- $2(1 H)$-one analogs ( $4-\mathrm{PQs}$ ) was designed that mimics the structures of 5-7 (Fig. 3). In addition, the two aromatic rings (B and E ) of 4-phenylquinolin-2(1H)-one derivatives (4-PQs) adopt a conformation in which they are not coplanar, and the structural similarity between CA-4 and appropriately substituted 4-PQs might lead to interaction at the colchicine site to affect tubulin polymerization. Moreover, 4-PQ compounds would not be inactivated by cis to trans isomerization as with CA-4 derivatives.

In this article, we describe the synthesis and cytotoxicity evaluation of 6,7-methylenedioxy-4-substituted phenylquinolin-2(1H)-one derivatives 12a-n (Fig. 3). Also, additional biological studies have been performed to analyze how novel compounds of this class affect the cell cycle.

## 2. Chemistry

The synthetic route to 6,7 -methylenedioxy-4-substituted phenylquinolin $-2(1 H)$-one derivatives is illustrated in Scheme 1. The 4-phenylquinolin-2(1H)-one> derivatives 12a-n were prepared using the general Knorr quinoline synthesis (acid-catalysis). ${ }^{23,24}$ The synthesis was initiated with alkoxycarbonylation of substituted acetophenones $\mathbf{8 a}-\mathbf{n}$ with a diethyl carbonate (9) to provide the corresponding benzoylacetates 10a-n, ${ }^{25-27}$ which were condensed with 3,4-methylenedioxy aniline to give benzoylacetanilide intermediates 11a-n. The 4-phenylquinolin-2(1H)-one derivatives 12a-n were obtained by cyclization of key intermediates 11a-n with excess polyphosphoric acid (PPA) at $100-110^{\circ} \mathrm{C}$. The synthesized compounds $\mathbf{1 2 a} \mathbf{- n}$ are reported for the first time. All synthesized compounds were characterized by IR, ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR and mass spectrometry. As an example, the target compound 12a possessed a characteristic peak at 6.02 ppm , representative of the methylenedioxy protons. Three singlet peaks at $6.18,6.63$ and 6.87 ppm were assignable to $\mathrm{C}(3) \mathrm{H}, \mathrm{C}(5) \mathrm{H}$ and $\mathrm{C}(8) \mathrm{H}$ protons in the 2-quinolone core, and one broad singlet ( 11.78 ppm ) to an exchangeable NH group. The protons of the 4-phenyl ring appeared in the region 7.367.51 ppm . In the ${ }^{13} \mathrm{C}$ NMR spectrum, compound $\mathbf{1 2 a}$ possessed a characteristic absorption at 5161.70 ppm for the 2-quinolone amide carbon ( $\mathrm{NHC}=\mathrm{O}$ ). The IR spectrum of 12a showed an absorption at $1653.07 \mathrm{~cm}^{-1}$ for the amide $(\mathrm{NHC}=\mathrm{O})$ carbonyl group.

## 3. Biological evaluation

### 3.1.Invitro cytotoxicity assay

The newly synthesized analogues 12a-n were evaluated for cytotoxicity against Detroit 551 normal human cells and eight cancer cell lines: MCF-7 (breast adenocarcinoma), 2774 (ovarian carcinoma), SKOV-3 (ovarian carcinoma), HL-60 (leukemia), Hep 3B (hepatoma), H460 (non-small-cell-lung carcinoma), COLO205 (colorectal adenocarcinoma) and A498 (renal cell carcinoma). Etoposide and 5-FU was used as a positive control. The screening procedure was based on the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazoliun bromide (MTT) growth inhibition assay, and the results are summarized in Table 1.

Compound 12a, which contains an unsubstituted 4-phenyl E-ring, displayed poor cytotoxicity with $\mathrm{IC}_{50}$ values from 18.5 to greater than $50 \mu \mathrm{M}$. The steric parameters of hydrogen and fluorine are extremely similar (van der Waals radii of $1.2 \AA$ for hydrogen vs. $1.35 \AA$ for fluorine), and fluoro-substituted $\mathbf{1 2 1}, \mathbf{1 2 m}$, and $\mathbf{1 2 n}$ also displayed weak cytotoxicity against the cancer cell lines.

However, 12k, whose ring E bears the same 3,4,5-trimethoxy substitution found in PPT, exhibited significant antiproliferative activity with sub-micromolar $\mathrm{IC}_{50}(0.93-7.6 \mu \mathrm{M})$ against several cancer cell lines. However, potency dropped when the 3,4,5-trimethoxy group ( $\mathbf{1 2 k}$ ) was changed to a $2,4,6$-trimethoxy group ( $\mathbf{1 2 j}$ ). In addition, the effects of dimethoxy substitution on the phenyl E-ring is noteworthy. Generally, dimethoxy substitution on the 4 -phenyl ring ( $\mathbf{1 2 e}, \mathbf{1 2 f}, \mathbf{1 2 h}, \mathbf{1 2 i})$ led to more potent activity than monomethoxy substitution ( $\mathbf{1 2 b} \mathbf{- d}$ ). However, compound $\mathbf{1 2 g}$ with 2,6 -dimethoxy substitution was less potent than the other four dimethoxy substituted analogues. Among the four more active dimethoxy compounds, analogue 12e with 2,4-dimethoxy substitution exhibited the most potent anticancer activity with $\mathrm{IC}_{50}$ less than or equal to $1 \mu \mathrm{M}$ against five cell lines, $2774(0.4 \mu \mathrm{M})$, SKOV-3 $(0.4 \mu \mathrm{M})$, HL-60 $(0.4 \mu \mathrm{M})$, Hep 3B $(1.0 \mu \mathrm{M})$ and $\mathrm{H} 460(0.9 \mu \mathrm{M})$. Compound $\mathbf{1 2 e}$ was also generally more potent than the trimethoxy substituted 12k. Modification of the E-ring with different numbers and positions of methoxy substituents may affect its spatial rotation and the molecular orientation, which could increase binding to the biological target and anticancer activity of the optimized compounds. Finally, 12e showed only marginal toxicity against Detroit 551 normal human cells ( $\mathrm{IC}_{50}>$ $25 \mu \mathrm{M})$. It is also noteworthy that the cell lines more resistant to 12e, including MCF-7, COLO205, and A498, harbor wild-type p53.

It is generally accepted that PPT's cytotoxicity is attributable to inhibition of tubulin polymerization, while the target of etoposide is topoisomerase II. Recently, an alternative non-tubulin and non-antitopoisomerase mode of action has been proposed for some PPT derivatives. This mechanism is associated with cell death and sub-G1 apoptotic cell accumulation without any significant cell cycle arrest. ${ }^{13}$ Thus, mechanistic studies on 12e are underway in our laboratories.

### 3.2. Morphological changes and apoptosis

Apoptosis is one of the major pathways that lead to the process of cell death. Chromatin condensation, nuclear shrinking, and fragmented nuclei are known as classic characteristics of apoptosis. To examine the effect of $\mathbf{1 2 e}$ on apoptotic induction, the morphology changes of HL-60 and H460 cells were studied using Hoechst 33258 stain, which confirmed apoptosis as the cause of reduced cell viability. Compound 12e decreased cell viability and increased morphological changes in HL-60 and H460 cells. As shown in Figure 4, control cells with $0.1 \%$ DMSO treatment exhibited uniformly dispersed chromatin. Cells treated with compound 12e for 24 h demonstrated typical apoptotic characteristics, including condensation of chromatin and appearance of nuclear fragmentation and apoptotic bodies (the arrowhead indicates an apoptotic nucleus). These results clearly demonstrated that the compound $\mathbf{1 2 e}$ is effective in inducing cellular apoptosis.

### 3.3. Cell cycle analysis

Previous studies have indicated that PPT analogues induced DNA damage and caused G2/M arrest. ${ }^{28-30}$ To evaluate the effect of $\mathbf{1 2 e}$ on the cell cycle andgain further insight into the mode of action, we examined cell cycle accumulation at 24 h by propidium iodide staining and flow cytometry quantification in HL-60 and H460 cells treated with 12e. In HL-60 cells treated with $0.25 \mu \mathrm{M}, 0.5 \mu \mathrm{M}$, and $1.0 \mu \mathrm{M}$ concentrations of $\mathbf{1 2 e}, 11.87 \%, 25.02 \%$ and $40.54 \%$, respectively, of cells were in G2/M phase, $10.51 \%$ of control (untreated) cells were in this phase (Fig. 5A). The percentage of G2/M cells also increased in H 460 cells treated with 12e (Fig. 5B). As shown in Figure 5, the DNA cell cycle analysis revealed typical G2/ M arrest and apoptosis (a sub-G1 peak appeared) in response to treatment with 12e.

### 3.4. Mechanism of action

Cyclin B1 and CDK1 are intricately involved in cell cycle progression through the G2/M phase transition. ${ }^{31}$ In our previous studies, the exposure of HL-60 cells to $2-\mathrm{PQ}$ analogues led to $\mathrm{G} 2 / \mathrm{M}$ phase arrest. In the Western blot analysis, a marked decrease in the expressions of cyclin B1 and CDK1 were detected in HL-60 cells. ${ }^{32,33}$ As shown in Figure 6, compound 12e induced a transient increase followed by a decrease in cyclin B1 protein expression, whereas CDK1 protein expression level decreased in a concentration-dependent manner in HL-60 cells (Fig. 6A). However, CDK1 protein expression level increased in a concentration-dependent manner in H 460 cells (Fig. 6B). It was reported that treatment with deoxypodophyllotoxin (DPPT) in Hela cells induced G2/M phase arrest. In the Western blot analysis, cyclin B1 expression was observed to have a rapid increase within 3 h of DPPT treatment that remained elevated for up to 24 h . After 48 h , the expression of cyclin B1 returned to near control levels. ${ }^{34}$ Another study was reported that treatment of NTUB1 cells with higher concentration of 1-hydroxy-3-(3-alkylaminopropoxy)-9,10-anthraquinone (MHA) derivative for 24 h induced up-regulation of cyclinB1 and p21 expressions in NTUB1 cells while treatment with lower concentration of MHA for 24 h induced downregulation of cyclinB1 and up-regulation of p21 expressions in NTUB1 cells. ${ }^{35}$ These results suggest that cell cycle progression is well regulated by the timing of the expression of cell-specific cyclins and different concentrations of the compounds.

Apoptosis is associated with activation of caspases, an expanding family of cysteine proteases that play important roles in the execution phase of apoptosis triggered by various stimuli. ${ }^{36}$ Among the capsases, caspase 3 is the best understood and mediates several apoptotic pathways. As shown in Figure 6, compound 12e induced caspase 3 cleavage (17 kDa ) in HL-60 and H460 cells. The above preliminary investigation of the mechanism of action of compound 12e suggested that the anticancer effects of 12e against HL-60 and H460 cells are mediated via G2/M arrest and caspase-dependent apoptosis.

## 4. Conclusion

4-Phenylquinolin-2(1H)-one derivatives 12a-n were synthesized and tested for antiproliferative activity against several cancer cell lines. Compound 12e, containing 2,4dimethoxy substitution on the 4-phenyl ring, demonstrated potent antiproliferative activities with $\mathrm{IC}_{50}$ values of $0.4,0.4,0.4$ and $0.9 \mu \mathrm{M}$ against 2774 , SKOV-3, HL-60 and H 460 , respectively. The significant anticancer activity shown by 12e prompted us to evaluate cell viability of HL-60 and H460 cells, with the observation of morphological features of apoptotic cells in these cell lines. Furthermore, 12e could induce cell cycle arrest in the G2/ M phase and apoptosis by activation of caspase-3 in both HL-60 and H460 cells. In conclusion, among the newly synthesized 4-PQ derivatives, compound 12e was identified as a promising candidate; due to its excellent antiproliferative potency, together with remarkable apoptosis-inducing activity. Further structure optimization of 12e is ongoing.

## 5. Experimental section

### 5.1. Materials and physical measurements

All solvents and reagents were obtained commercially and used without further purification. The progress of all reactions was monitored by TLC on $2 \times 6 \mathrm{~cm}$ pre-coated silica gel 60 $\mathrm{F}_{254}$ plates of thickness 0.25 mm (Merck). The chromatograms were visualized under UV light at $254-366 \mathrm{~nm}$. The following adsorbent was used for column chromatography: silica gel 60 (Merck, particle size $0.063-0.200 \mathrm{~mm}$ ). Melting points ( mp ) were determined with a Yanaco MP-500D melting point apparatus and are uncorrected. IR spectra were recorded on Shimadzu IR-Prestige-21 spectrophotometers as KBr pellets. The NMR spectra were
obtained on a Bruker Avance DPX-200 FT-NMR spectrometer at room temperature, and chemical shifts are reported in ppm ( $\delta$ ). The following abbreviations are used: s, singlet; d , doublet; t , triplet; dd, double doublet; and m, multiplet. Low-resolution mass spectra were performed by Finnigan/Thermo Qust MAT95XL at National Chung Hsing University, Taichung, Taiwan.

### 5.2. Chemistry

5.2.1. General procedure for the synthesis of benzoylacetates 10a-n—The substrate $\beta$-ketoesters 10a-n were either purchased or synthesized following published procedures. Some benzoylacetates were commercially available. Ethyl 3-oxo-3-phenyl propanoate (10a) was purchased. The reaction of benzoylacetates $\mathbf{1 0 b}-\mathbf{n}$ was prepared as described in previous reports. ${ }^{25-27}$ A solution of a substituted acetophenone $\mathbf{8 a - n}(0.05 \mathrm{~mol})$ dissolved in toluene ( 50 mL ) was added dropwise to a solution containing diethyl carbonate (9) $(0.10 \mathrm{~mol})$ and sodium hydride ( $0.15 \mathrm{~mol} 60 \%$ dispersion in mineral oil). The mixture was stirred at room temperature, and then refluxed for 30 min . The mixture was poured into ice water, acidified with glacial acetic acid, and extracted with EtOAc $(3 \times 100 \mathrm{~mL})$. The EtOAc extract was then dried over anhydrous $\mathrm{MgSO}_{4}$. After removal of the solvent in vacuo, the crude products were purified by silica gel column chromatography eluting with dichloromethane to afford benzoylacetates 10b-n. All synthetic compounds were in agreement with ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, IR and mass spectroscopic data.
5.2.1.1. Ethyl 3-(2-methoxyphenyl)-3-oxopropanoate (10b): Yield $86 \%$ from $\mathbf{8 b}$ as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1735.93, $1674.21(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.14\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.83\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.91\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CO}-\underline{\mathrm{CH}}_{2} \underline{-}^{-}\right.$ CO-), 4.07 ( $\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 7.03 (t, $\left.J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right), 7.14$ (d, $J=8.2$ $\mathrm{Hz}, 1 \mathrm{H}, \operatorname{Ar}-\mathrm{H}), 7.56(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 7.70(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $\left.d_{6}\right): \delta 14.37,50.48,56.03,60.80,112.95,120.99,126.11,130.50,135.37$, 159.37, 168.20, 193.25; MS (EI, 70 eV ) m/z: $222.1\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ : 222.0892; found: 222.0884.
5.2.1.2. Ethyl 3-(3-methoxyphenyl)-3-oxopropanoate (10c): Yield $80 \%$ from 8c as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : $1743.65,1689.64(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \underline{\mathrm{H}}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.06-4.20(\mathrm{~m}, 4 \mathrm{H},-\mathrm{CO}-$ $\mathrm{CH}_{2}$-CO- and $\left.-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.19-7.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{ArH}), 7.40-7.57(\mathrm{~m}, 3 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 14.36,46.12,55.73,61.05,113.31,120.18,121.34,130.38,137.64$, 159.91, 168.08, 193.67; MS (EI, 70 eV ) m/z: $222.1\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ : 222.0892; found: 222.0898 .
5.2.1.3. Ethyl 3-(4-methoxyphenyl)-3-oxopropanoate (10d): Yield $87 \%$ from $8 d$ as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : $1735.93,1674.21(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $(200 \mathrm{MHz}$, DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}_{\left.-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.82\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.05-4.15(\mathrm{~m}, 4 \mathrm{H},-\mathrm{CO}-\mathrm{C}}\right.$ $\mathrm{CH}_{2}-\mathrm{CO}-$ and $\left.-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.03(\mathrm{~d}, J=\overline{8} .9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArH}), 7.93(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 2 \mathrm{H}$, $\mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 14.36,45.71,55.95,60.97,114.40$ (2C), 129.24, 131.24 (2C), 164.04, 168.25, 192.06; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z}: 222.1$ (M ${ }^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ : 222.0892; found: 222.0899.
5.2.1.4. Ethyl 3-(2,4-dimethoxyphenyl)-3-oxopropanoate (10e): Yield $70 \%$ from $\mathbf{8 e}$ as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1735.93, $1666.50(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.83-3.84\left(\mathrm{~m}, 8 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right.$ and $-\mathrm{CO}-\mathrm{CH}_{2} \underline{-}^{-}$ CO-), $4.08\left(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.58-6.62(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}), 7.75(\mathrm{~d}, J=9.4 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 14.39,50.42,56.03$ (2C), 60.64, 98.59,
$106.83,119.11,132.61,161.57,165.47,168.46,190.91$; MS (EI, 70 eV ) m/z: $252.1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{5}: 252.0998$; found: 252.0991 .
5.2.1.5. Ethyl 3-(2,5-dimethoxyphenyl)-3-oxopropanoate (10f): Yield $70 \%$ from $\mathbf{8 f}$ as a yellow liquid; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1735.93, $1674.21(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.73\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.91$ ( $\mathrm{s}, 2 \mathrm{H},-\mathrm{CO}-\mathrm{CH}_{2}$-CO-), 4.08 (q, $\left.J=7.1 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.12-7.15(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH})$, 7.22-7.24 (m, 1H, ArH); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-\bar{d}_{6}$ ): $\delta 14.38,50.38,55.90,56.42$, $60.75,113.94,114.51,121.47,126.35,153.40,153.81,168.12,192.75$; MS (EI, 70 eV ) m/ $z: 252.2\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{5}: 252.0998$; found: 252.0996 .
5.2.1.6. Ethyl 3-(2,6-dimethoxyphenyl)-3-oxopropanoate ( $\mathbf{1 0 g}$ ): Yield $78 \%$ from $\mathbf{8 g}$ as a colorless liquid; IR (KBr) v $\left(\mathrm{cm}^{-1}\right)$ : $1743.65,1705.07(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.14\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.75\left(\mathrm{~s}, 8 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right.$ and $\left.-\mathrm{CO}-\underline{\mathrm{CH}}_{2}-\mathrm{CO}-\right)$, $4.06\left(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\underline{C H}_{2} \mathrm{CH}_{3}\right), 6.70(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArH}), 7.35(\mathrm{t}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}$, ArH ); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 14.30,51.47,56.26$ (2C), 60.84, 104.71 (2C), 118.63, 132.07, 156.98 (2C), 166.82, 195.95; MS (EI, 70 eV ) m/z: 252.2 ( $\mathrm{M}^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{5}$ : 252.0998; found: 252.1006.
5.2.1.7. Ethyl 3-(3,4-dimethoxyphenyl)-3-oxopropanoate (10h): Yield $76 \%$ from $\mathbf{8 h}$ as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1735.93, $1674.21(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \underline{\mathrm{H}}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.83\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right)$, 4.04-4.15 (m, 4H, -CO-Cㅡ﹎ㅡㄴ $\left.-\mathrm{CO}-\mathrm{and}-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.02(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.43(\mathrm{~d}$, $J=1.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.59(\mathrm{dd}, J=8.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 14.34,45.62,55.85,56.10,60.98,110.78,111.20,123.85,129.19,149.11,154.03,168.29$, 192.09; MS (EI, 70 eV ) m/z: $252.2\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{5}$ : 252.0998; found: 252.1003.
5.2.1.8. Ethyl 3-(3,5-dimethoxyphenyl)-3-oxopropanoate (10i): Yield $81 \%$ from $\mathbf{8 i}$ as a yellow liquid; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1735.93, $1689.64(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.17\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.79\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right), 4.06-4.15(\mathrm{~m}, 4 \mathrm{H},-$ $\mathrm{CO}-\mathrm{CH}_{2}-\mathrm{CO}-$ and $\left.-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.77(\mathrm{~s}, 1 \mathrm{H}, \mathrm{ArH}), 7.07-7.08(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR (50 MHz, DMSO- $d_{6}$ ): $\delta 14.35,46.16,55.87$ (2C), 61.04, 105.95, 106.57 (2C), 138.21, 161.07 (2C), 168.05, 193.47; MS (EI, 70 eV ) m/z: $252.2\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{5}$ : 252.0998 ; found: 252.0994 .
5.2.1.9. Ethyl 3-oxo-3-(2,4,6-trimethoxyphenyl)propanoate (10j): Yield $\mathbf{6 0 \%}$ from $\mathbf{8 j}$ as a white solid; mp : 69-71 ${ }^{\circ} \mathrm{C}$; IR ( KBr ) v $\left(\mathrm{cm}^{-1}\right)$ : 1728.22, 1689.64 (C=O); ${ }^{1} \mathrm{H}$ NMR (200 Mhz, DMSO- $d_{6}$ ): $\delta 1.13\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \underline{\mathrm{H}}_{3}\right), 3.70\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CO}-\underline{C H}_{2}-\mathrm{CO}-\right), 3.74$ $\left(\mathrm{s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.05\left(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.25(\mathrm{~s}, 2 \mathrm{H}$, $\mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 14.41,51.47,55.95,56.32$ (2C), 60.72, 91.37 (2C), 111.65, 158.81 (2C), 163.17, 167.26, 194.47; MS (EI, 70 eV ) m/z: 282.2 (M+); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{6}: 282.1103$; found: 282.1106.
5.2.1.10. Ethyl 3-oxo-3-(3,4,5-trimethoxyphenyl)propanoate (10k): Yield $69 \%$ from $\mathbf{8 k}$ as a colorless liquid; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1743.65,1681.93(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H} \mathrm{NMR}(200 \mathrm{MHz}$, DMSO-d $d_{6}$ : $\delta 1.18\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}_{\left.-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.75\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.84(\mathrm{~s}, 6 \mathrm{H}, 2 \times-1 .}\right.$ $\left.\mathrm{OCH}_{3}\right), 4.07-4.19\left(\mathrm{~m}, 4 \mathrm{H},-\mathrm{CO}-\mathrm{CH}_{2}-\mathrm{CO}-\right.$ and $\left.-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.25(\mathrm{~s}, 2 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 14.36,45.8 \overline{8}, 56.41$ (2C), $60.52,61.01,106.46$ (2C), 131.51, 142.82, 153.24 (2C), 168.18, 192.68; MS (EI, 70 eV ) m/z: 282.2 (M ${ }^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{6}$ : 282.1103; found: 282.1102 .
5.2.1.11. Ethyl 3-(2-fluorophenyl)-3-oxopropanoate (101): Yield $73 \%$ from 81 as a yellow liquid; IR (KBr) v $\left(\mathrm{cm}^{-1}\right): 1743.65,1689.64(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 1.15$ $\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.04-4.15\left(\mathrm{~m}, 4 \mathrm{H},-\mathrm{CO}-\mathrm{CH}_{2}-\mathrm{CO}-\right.$ and $\left.-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.29-$ $7.39(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}), 7.64-7.72(\mathrm{~m}, 1 \mathrm{H}, \mathrm{ArH}), 7.82-7.87(\mathrm{~m}, 1 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz , $\left.D M S O-d_{6}\right): \delta 14.36,49.52\left(\mathrm{~d},{ }^{4} J_{\mathrm{CF}}=6.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 61.09,117.33\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23.0 \mathrm{~Hz}, 1 \mathrm{C}\right)$, $124.77\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=11.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 125.33,130.89,136.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=9.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 161.74\left(\mathrm{~d},{ }^{1}\right.$ $\left.J_{\text {cf }}=253.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 167.76,191.09$; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z} .210 .1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$. calculated for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{FO}_{3}$ : 210.0692; found: 210.0700.
5.2.1.12. Ethyl 3-(3-fluorophenyl)-3-oxopropanoate ( $\mathbf{1 0 m}$ ): Yield $69 \%$ from $\mathbf{8 m}$ as a yellow liquid; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1743.65, $1689.64(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.11\left(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.20(\mathrm{~s}$, $2 \mathrm{H},-\mathrm{CO}-\mathrm{CH}_{2}-\mathrm{CO}-$ ), $7.46-7.64(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}), 7.68-7.83(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz , DMSO- $\left.d_{6}\right): \bar{\delta} 14.36,46.08,61.13,115.35\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=22.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 121.12\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21.5\right.$ $\mathrm{Hz}, 1 \mathrm{C}), 125.06,131.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 138.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=6.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 162.64\left(\mathrm{~d},{ }^{1}\right.$ $\left.J_{\mathrm{CF}}=244.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 167.90,193.00 ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{eV}) \mathrm{m} / \mathrm{z}: 210.1\left(\mathrm{M}^{+}\right) ;$HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{FO}_{3}$ : 1210.0692; found: 210.0695.
5.2.1.13. Ethyl 3-(4-fluorophenyl)-3-oxopropanoate (10n): Yield $77 \%$ from $\mathbf{8 n}$ as a yellow liquid; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : $1743.65,1689.64(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO$\left.d_{6}\right): \delta 1.16\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.10\left(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.17(\mathrm{~s}$, $2 \mathrm{H},-\mathrm{CO}-\mathrm{CH}_{2}$-CO-), $7.30-7.39$ (m, 2H, ArH), 8.00-8.07 (m, 2H, ArH); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz , DMSO- $d_{6}$ ): $\bar{\delta} 14.36,45.94,61.08,116.28\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21.9 \mathrm{~Hz}, 2 \mathrm{C}\right), 131.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=9.6 \mathrm{~Hz}\right.$, 2C), 133.05, $165.80\left(\mathrm{~d},{ }^{1} J_{\mathrm{cF}}=251.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 168.01,192.53$; MS (EI, 70 eV$) \mathrm{m} / \mathrm{z}: 210.1$ $\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{FO}_{3}$ : 210.0692; found: 210.0701 .
5.2.2. General procedure for the synthesis of benzoylacetanilides 11a-n-A mixture of the substituted benzoyl acetate ( $\mathbf{1 0 a - n}, 1$ equiv) and 3,4-methylenedioxy aniline ( 1 equiv) was stirred in 150 mL toluene and then heated at reflux for $1-2 \mathrm{~h}$. The mixture was cooled to the room temperature and partitioned with $10 \% \mathrm{NaOH}(3 \times 50 \mathrm{~mL})$. The aqueous layer was acidified to $\mathrm{pH} 4-5$ with dropwise addition of glacial acetic acid. The resulting precipitate was isolated by suction filtration, washed with water and EtOH , and then airdried air to give the desired benzoylacetanilide (11a-n) of sufficient purity for the next reaction.
5.2.2.1. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-oxo-3-phenylpropanamide (11a): Yield 45\% from 10a as a light-yellow solid; mp: $149-151^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1654.03, 1684.89 $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 4.09\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2^{-}}\right), 5.97$ ( $\left.\mathrm{s}, 2 \mathrm{H},-\mathrm{O}-\underline{\mathrm{CH}}_{2}-\mathrm{O}-\right)$, $6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.94(\mathrm{dd}, J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.28(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}$, ArH), 7.51-7.58 (m, 2H, ArH), 7.63-7.74 (m, 1H, ArH), 8.00 (d, J=7.2 Hz, 1H, ArH), 10.11 (s, 1H, NH); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 48.41,101.43,101.67,108.53$, 112.33, 128.81 (2C), 129.25 (2C), 133.83, 134.03, 136.72, 143.45, 147.53, 165.39, 195.04; MS (EI, 70 eV ) m/z: $283.2\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{16} \mathrm{H}_{13} \mathrm{NO}_{4}: 283.0845$; found: 283.0836 .
5.2.2.2. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(2-methoxyphenyl)-3-oxopropanamide (11b): Yield $38 \%$ from 10b as a white solid; mp : $143-144{ }^{\circ} \mathrm{C}$; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1656.92 ,
 $\mathrm{CH}_{2} \underline{2}^{-}$), $5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\underline{\mathrm{H}}_{2}-\mathrm{O}-\right), 6.83(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.94(\mathrm{dd}, J=8.4,1.6 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}), 7.04(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.15(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.28(\mathrm{~d}, J=1.6 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}$ ), 7.56 (t, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 7.69 (d, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 10.01 ( $\mathrm{s}, 1 \mathrm{H}$, NH ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz , DMSO- $d_{6}$ ): $\delta 52.96,56.23,101.39,101.65,108.51,112.23$,
113.01, 120.94, 126.91, 130.45, 133.99, 134.96, 143.29, 147.49, 159.23, 165.82, 195.04; MS (EI, 70 eV ) m/z: $313.2\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{NO}_{5}: 313.0950$; found: 313.0959 .
5.2.2.3. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(3-methoxyphenyl)-3-oxopropanam ide (11c): Yield $37 \%$ from $10 c$ as a pale gray solid; $\mathrm{mp}: 141-142{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1656.92$, $1685.86(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 3.81\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.08(\mathrm{~s}, 2 \mathrm{H},-$ $\mathrm{CH}_{2}-$ ), 5.97 (s, 2H, $\left.-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.93$ (dd, $J=8.4,1.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}), 7.21-7.50(\mathrm{~m}, 4 \overline{\mathrm{H}}, \mathrm{ArH}), 7.59(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 10.12(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR (50 MHz, $\left.D M S O-d_{6}\right): ~ \delta 48.57,55.84,101.44,101.68,108.53,112.34,113.36,119.98$, 121.34, 130.43, 133.81, 138.11, 143.46, 147.54, 159.91, 165.35, 194.79; MS (EI, 70 eV ) m/ $z: 313.2\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 313.0950; found: 313.0958.
5.2.2.4. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(4-methoxyphenyl)-3-oxopropanamide (11d): Yield $38 \%$ from 10d as a light-brown solid; mp : $144-145^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1655.96$, $1674.28(\mathrm{C}=\mathrm{O})$; 1H NMR ( $200 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 3.84\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.02(\mathrm{~s}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \underline{2}^{-}$), 5.97 (s, 2H, $\left.-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.93$ (dd, $J=8.4,1.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}), 7.06(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArH}), 7.27(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{ArH}), 10.09(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 48.15,56.06,101.42$, $101.64,108.52,112.30,114.45$ (2C), 129.70, 131.23 (2C), 133.87, 143.40, 147.50, 163.89, 165.57, 193.34; MS (EI, 70 eV ) m/z: $313.1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 313.0950; found: 313.0944.
5.2.2.5. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(2,4-dimethoxyphenyl)-3-oxopropanami de (11e): Yield $44 \%$ from 10e as a white solid; mp : $164-166^{\circ} \mathrm{C}$; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1652.10, $1676.21(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 3.83\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.85(\mathrm{~s}, 3 \mathrm{H},-$ $\left.\mathrm{OCH}_{3}\right), 3.89\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2} \underline{-}^{-}\right), 5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.61-6.64(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}), 6.83(\mathrm{~d}, J=$ $8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 6.93 (dd, $J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 7.27 (d, $J=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 7.73 (d, $J=9.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), $9.96(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 52.90,56.13$, $56.29,98.76,101.36,101.60,106.78,108.50,112.16,119.73,132.62,134.09,143.21$, 147.48, 161.52, 165.20, 166.19, 192.68; MS (EI, 70 eV ) m/z: 343.2 (M+); HRMS (EI) $m / z$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{6}$ : 343.1056; found: 343.1049.
5.2.2.6. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(2,5-dimethoxyphenyl)-3-oxopropanami de (11f): Yield $48 \%$ from $10 f$ as a light-brown solid; mp: $144-146^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1646.32, $1668.50(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}-d_{1}$ ): $\delta 3.73\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.80(\mathrm{~s}$, $\left.3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.95\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}-\right), 5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.83(\mathrm{~d}, J=8.6 \mathrm{~Hz}, \overline{1} \mathrm{H}, \mathrm{ArH})$, 6.89-6.95 (m, 1H, ArH), 7.08-7.31 (m, 4H, ArH), $10.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz , $\left.\mathrm{CDCl}_{3}-d_{1}\right): \delta 52.88,56.00,56.68,101.38,101.61,108.53,112.19,114.04,114.61,121.01$, $127.11,133.98,143.27,147.49,153.34,153.65,165.80,194.58 ;$ MS (EI, 70 eV ) m/z: 343.2 $\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{6}: 343.1056$; found: 343.1051.
5.2.2.7. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(2,6-dimethoxyphenyl)-3-oxopropanami de (11g): Yield 59\% from 10 g as a white solid; mp: $131-133{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1654.03$, 1709.97 (C=O); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, D M S O-d_{6}$ ): $\delta 3.72\left(\mathrm{~s}, 6 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.74(\mathrm{~s}, 2 \mathrm{H},-$ $\left.\mathrm{CH}_{2}-\right), 5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\underline{C H}_{2}-\mathrm{O}-\right), 6.69(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArH}), 6.82(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}$, $\mathrm{ArH}), 6.92(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.25(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.33(\mathrm{t}, J=8.4 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}$ ), $9.94(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 53.89,56.36$ (2C), 101.40,101.76, 104.81 (2C), 108.46, 112.43, 119.26, 131.84, 133.89, 143.37, 147.44, 156.95 (2C), 164.02, 197.56; MS (EI, 70 eV ) m/z: 342.9 ( $\mathrm{M}^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{6}: 343.1056$; found: 343.1062.
5.2.2.8. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(3,4-dimethoxyphenyl)-3-oxopropanami de (11h): Yield $39 \%$ from $\mathbf{1 0 h}$ as a pale gray solid; mp : $191-192^{\circ} \mathrm{C}$; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1653.07, 1676.21 (C=O); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.84$ ( s , $\left.3 \mathrm{H},-\mathrm{OCH}_{3}\right), 4.03\left(\mathrm{~s}, 2 \mathrm{H},-\underline{\mathrm{C}}_{2}-\right), 5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH})$, $6.94(\mathrm{dd}, \bar{J}=8.3,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.27(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}$, ArH ), 7.47 (d, $J=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 7.67 (dd, $J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), 10.12 (s, 1H, NH ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 48.15,56.01,56.25,101.43,101.65,108.52$, $110.90,111.39,112.30,123.86,129.67,133.88,143.42,147.52,149.08,153.86,165.57$, 193.34; MS (EI, 70 eV ) m/z: $343.1\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{6}$ : 343.1056; found: 343.1050.
5.2.2.9. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(3,5-dimethoxyphenyl)-3-oxopropanami de (11i): Yield $61 \%$ from $10 i$ as a light-brown solid; mp: $158-160^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right)$ : 1653.07, $1695.50(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.79$ ( $\mathrm{s}, 6 \mathrm{H},-\mathrm{OCH}_{3}$ ), 4.06 ( s , $\left.2 \mathrm{H},-\underline{C H}_{2}-\right), 5.97$ (s, 2H, -O-C $\left.\underline{H}_{2}-\mathrm{O}-\right), 6.77-6.86(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArH}), 6.92$ (dd, $J=8.4,2.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ArH}$ ), 7.10 (d, $J=2.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArH}), 7.26(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 10.12(\mathrm{~s}, 1 \mathrm{H}$, $\mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 48.65,56.03$ (2C), 101.44, 101.64, 105.80, 106.60 (2C), 108.56, 112.31, 133.78, 138.67, 143.46, 147.53, 161.08 (2C), 165.30, 194.59; MS (EI, 70 eV ) m/z: $342.9\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}$ : 343.1056 ; found: 343.1059 .
5.2.2.10. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-oxo-3-(2,4,6-trimethoxyphenyl)propane mide (11j): Yield $51 \%$ from 10j as a white solid; mp: $124-126^{\circ} \mathrm{C}$; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1652.10$, 1706.11 (C=O); ${ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO- $d_{6}$ ): $\delta 3.69$ ( $\mathrm{s}, 2 \mathrm{H},-\mathrm{CH}_{2}-$ ), 3.72 (s, $6 \mathrm{H},-$ $\mathrm{OCH}_{3}$ ), $3.79\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 5.96\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.24(\mathrm{~s}, 2 \mathrm{H}, \mathrm{ArH}), 6.82(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.92(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.25(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 9.90(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{d}$ ): $\delta 53.90,55.95$, 56.36 (2C), 91.41 (2C), 101.38, 101.69, 108.46, 112.33 (2C), 133.96, 143.29, 147.42, 158.72 (2C), 162.91, 164.57, 197.17; MS (EI, 70 eV ) m/z: $372.6\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{NO}_{7}: 373.1162$; found: 373.1166 .
5.2.2.11. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-oxo-3-(3,4,5-trimethoxyphenyl)propane mide (11k): Yield $27 \%$ from 10 k as a white solid; mp : $186-188^{\circ} \mathrm{C} ;$ IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1649.21$, 1672.36 (C=O); ${ }^{1} \mathrm{H}$ NMR ( 200 MHz , DMSO- $d_{6}$ ): $\delta 3.74$ ( $\mathrm{s}, 3 \mathrm{H},-\mathrm{OCH}_{3}$ ), 3.83 ( $\mathrm{s}, 6 \mathrm{H}$, $\mathrm{OCH}_{3}$ ), $4.08\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}-\underline{-}^{-}, 5.97\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.93\right.$ (dd, $\bar{J}=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{A} r \mathrm{H}), 7.26-7.29(\mathrm{~m}, 3 \mathrm{H}, \mathrm{ArH}), 10.16(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz , DMSO- $d_{6}$ ): $\delta 48.42,56.59$ (2C), $60.65,101.45,101.67,106.48$ (2C), 108.56, 112.36, 131.98, 133.76, 142.65, 143.47, 147.53, 153.25 (2C), 165.46, 193.82; MS (EI, 70 eV ) m/z: $373.0\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{NO}_{7}$ : 373.1162; found: 373.1170.
5.2.2.12. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(2-fluorophenyl)-3-oxopropanamide (111): Yield $61 \%$ from 101 as a white solid; mp: 147-148 ${ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v}\left(\mathrm{cm}^{-1}\right)$ : 1653.07, 1691.64 (C=O); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}-d_{1}$ ): $\delta 4.10\left(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{CH}_{2}-{ }^{-}\right), 5.93(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-$ $\left.\mathrm{CH}_{2}-\mathrm{O}-\right), 6.73(d, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.86$ (dd, $J=8.2,2.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), $7.11-7.22$ (m, $2 \mathrm{H}, \mathrm{ArH}$ ), 7.26 (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), $7.53-7.60$ (m, 1H, ArH), $7.84-7.92$ (m, 1H, ArH), $9.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}-d_{1}$ ): $\delta 49.45\left(\mathrm{~d},{ }^{4} J_{\mathrm{CF}}=8.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 101.27$, 103.11, 108.05, 113.51, $117.01\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 124.56\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21.5 \mathrm{~Hz}, 1 \mathrm{C}\right)$, 124.77, 130.75, 131.79, $135.84\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=9.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 144.46,147.79,161.98\left(\mathrm{~d},{ }^{1} J_{\mathrm{cf}}=\right.$ $254.5 \mathrm{~Hz}, 1 \mathrm{C}$ ), 163 62, 194.75; MS (EI, 70 eV ) m/z: $301.0\left(\mathrm{M}^{+}\right.$); HRMS (EI) m/z: calculated for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{FNO}_{4}$ : 301.0750; found: 301.0760.
5.2.2.13. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(3-fluorophenyl)-3-oxopropanamide (11m): Yield $62 \%$ from 10 m as a pale gray solid; $\mathrm{mp}: 130-132^{\circ} \mathrm{C}$; $\mathrm{IR}(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1655.96$, $1695.50(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO- $d_{6}$ ): $\delta 4.10\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}{ }^{-}\right), 5.96(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-$ $\left.\mathrm{CH}_{2}-\mathrm{O}-\right), 6.84(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.92(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.25(\mathrm{~d}, J=1.8$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.51-7.86(\mathrm{~m}, 4 \mathrm{H}, \mathrm{ArH}), 10.13(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 48.51,101.41,101.71,108.56,112.35,115.31\left(\mathrm{~d},{ }^{2} J_{\mathrm{cf}}=22.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 120.94\left(\mathrm{~d},{ }^{2} J_{\mathrm{cf}}=\right.$ $21.0 \mathrm{~Hz}, 1 \mathrm{C}), 125.09,131.39,133.75,138.88,143.48,147.55,162.69\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=244.0 \mathrm{~Hz}\right.$, 1C), 165.13, 194.09; MS (EI, 70 eV ) m/z: $300.9\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{FNO}_{4}$ : 301.0750; found: 301.0747.
5.2.2.14. $N$-(Benzo[d][1,3]dioxol-5-yl)-3-(4-fluorophenyl)-3-oxopropanamide (11n): Yield $59 \%$ from 10 n as a white solid; mp : $164-165^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v}\left(\mathrm{cm}^{-1}\right): 1633.78$, 1690.68 (C=O); 1H NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 4.09$ ( $\mathrm{s}, 2 \mathrm{H},-\mathrm{CH}_{2}{ }^{-}$), 5.96 (s, 2H, -O-$\left.\mathrm{CH}_{2}-\mathrm{O}-\right), 6.83(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 6.93(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}), 7.27(\mathrm{~d}, J=1.8$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{ArH}$ ), $7.32-7.41$ (m, 2H, ArH), 8.04-8.11 (m, 2H, ArH), 10.11 (s, 1H, NH); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 48.36,101.44,101.68,108.53,112.35,116.29\left(\mathrm{~d},{ }^{2} J_{\mathrm{cf}}=21.5\right.$ $\mathrm{Hz}, 2 \mathrm{C}), 131.90\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=9.5 \mathrm{~Hz}, 2 \mathrm{C}\right), 133.47,133.77,143.47,147.53,165.30,165.69$ $\left(\mathrm{d},{ }^{1} J_{\mathrm{CF}}=250.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 193.71$; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z}: 300.9\left(\mathrm{M}^{+}\right)$; HRMS (EI) m/z: calculated for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{FNO}_{4}$ : 301.0750 ; found: 301.0755.

### 5.2.3. General procedure for the synthesis of 6,7-methylenedioxy-4substituted phenylquinolin-2-one derivatives 12a-n-A mixture of the

 benzoylacetanilide (11a-n) and PPA ( 10 g ) was heated at $100-110^{\circ} \mathrm{C}$ for 0.5 h to 1 h . The mixture was cooled after TLC monitoring indicated that the reaction was completed, and then the reaction mixture diluted with ice water, and extracted with $\mathrm{CHCl}_{3}(3 \times 50 \mathrm{~mL})$. The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, and the solvent was removed in vacuo. The crude products were purified by column chromatography (silica gel, chloroform/ethyl acetate $=2 / 1$ ) to give the corresponding pure 4-phenylquinolin-2-ones 12a-n.5.2.3.1. 6,7-Methylenedioxy-4-phenylquinolin-2(1H)-one (12a): Compound 12 a ( 0.16 g , $0.60 \mathrm{mmol})$ was obtained by cyclization of $11 \mathbf{a}(0.27 \mathrm{~g}, 0.95 \mathrm{mmol})$ with PPA $(10 \mathrm{~g})$; Yield: $63 \%$; white solid; mp: $274-275{ }^{\circ} \mathrm{C}$; IR ( KBr ) v $\left(\mathrm{cm}^{-1}\right)$ : 1653.07 ( $\mathrm{C}=\mathrm{O}$ ); ${ }^{1} \mathrm{H}$ NMR (200 MHz, DMSO- $d_{6}$ ): $\delta 6.02$ ( $\left.\mathrm{s}, 2 \mathrm{H},-\mathrm{O}-\underline{\mathrm{H}}_{2}-\mathrm{O}-\right), 6.18(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.87$ (s, $1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), $7.36-7.51$ (m, 5H, Ar-H), 11.78 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}-$ $\left.d_{1}\right): \delta 45.27,101.25,103.00,108.04,113.43,128.57$ (2C), 128.95 (2C), 131.81, 134.34, 136.05, 144.42, 147.76, 163.61, 196.51; MS (EI, 70 eV ) m/z: 265.1 (M ${ }^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{NO}_{3}$ : 265.0739; found: 265.0733.
5.2.3.2.6,7-Methylenedioxy-4-(2-methoxyphenyl)quinolin-2(1H)-one (12b): Compound $\mathbf{1 2 b}(0.19 \mathrm{~g}, 0.64 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 b}(0.29 \mathrm{~g}, 0.93 \mathrm{mmol})$ with PPA (10 g); Yield: $68 \%$; white solid; mp: $269-271^{\circ} \mathrm{C}$; IR (KBr) v $\left(\mathrm{cm}^{-1}\right): 1663.07(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.65\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 5.99,6.01$ (s, each $1 \mathrm{H},-\mathrm{O}_{\left.-\mathrm{CH}_{2}-\mathrm{O}-\right) \text {, }}$ 6.13 (s, 1H, Ar-H), 6.30 (s, 1H, Ar-H), 6.84 (s, 1H, Ar-H), 7.02 (t, J=7.4 Hz, 1H, Ar-H), 7.10-7.17 (m, 2H, Ar-H), 7.43 (t, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 11.71 (br. s, 1H, NH); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 55.89,95.79,102.26,103.90,112.00,113.69,119.44,121.17$, $126.27,130.37,130.77,136.09,143.71,149.69,150.35,156.41,161.85$; MS (EI, 70 eV ) m/ z: $295.1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{NO}_{4}$ : 295.0845; found: 295.0835.
5.2.3.3. 6,7-Methylenedioxy-4-(3-methoxyphenyl)quinolin-2(1H)-one (12c): Compound $\mathbf{1 2 c}(0.17 \mathrm{~g}, 0.58 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 c}(0.30 \mathrm{~g}, 0.96 \mathrm{mmol})$ with PPA $(10 \mathrm{~g})$; Yield: $60 \%$; white solid; mp : $278-279{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1653.07(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$

NMR (200 MHz, DMSO- $d_{6}$ ): $\delta 3.75\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 6.02\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.19(\mathrm{~s}, 1 \mathrm{H}$, Ar-H), 6.66 (s, 1H, Ar-H), 6.85 (s, 1H, Ar-H), 6.90-6.93 (m, 2H, Ar-H), 7.00 (dd, J= 8.2, $2.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 7.38 (t, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 11.73 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 55.65,96.01,102.38,103.75,112.87,114.37,114.75$ 118.75, 121.09, $130.23,136.79,138.98,143.78,150.49,151.57,159.80,161.62 ;$ MS (EI, 70 eV ) m/z: 295.2 $\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{NO}_{4}$ : 295.0845; found: 295.0851 .
5.2.3.4. 6,7-Methylenedioxy-4-(4-methoxyphenyl)quinolin-2(1H)-one (12d): Compound $\mathbf{1 2 d}(0.19 \mathrm{~g}, 0.64 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 d}(0.29 \mathrm{~g}, 0.93 \mathrm{mmol})$ with PPA (10 g); Yield: 68\%; white solid; mp: 269-271 ${ }^{\circ} \mathrm{C}$; IR (KBr) v $\left(\mathrm{cm}^{-1}\right): 1663.07(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO- $d_{6}$ ): $\delta 3.82\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 6.06\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\underline{\mathrm{CH}}_{2}-\mathrm{O}-\right), 6.19(\mathrm{~s}, 1 \mathrm{H}$, Ar-H), 6.75 (s, 1H, Ar-H), 6.90 (s, 1H, Ar-H), 7.07 (d, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 7.37$ (d, $J=$ $8.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 11.69 (br. s, 1H, NH); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz , DMSO- $d_{6}$ ): $\delta 55.70,96.06$, 102.37, 103.86, 113.13, 114.64 (2C), 118.60, 129.76, 130.32 (2C), 136.81, 143.82, 150,46, 151.53, 160.04, 161.73; MS (EI, 70 eV ) m/z: $295.2\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{NO}_{4}$ : 295.0845 ; found: 295.0851.
5.2.3.5. 6,7-Methylenedioxy-4-(2,4-dimethoxyphenyl)quinolin-2(1H)-one (12e):

Compound $\mathbf{1 2 e}(0.11 \mathrm{~g}, 0.34 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 e}(0.19 \mathrm{~g}, 0.55 \mathrm{mmol})$ with PPA ( 10 g ); Yield: $61 \%$; white solid; mp: $251-252{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1654.03$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO- $d_{6}$ ): $\delta 3.67\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right.$ ), $3.81\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 6.02$, 6.04 (s, each $\left.1 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.12(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.39(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.61-6.69(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ar}-$ H), $6.85(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 7.09(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 11.70$ (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 55.79,55.95,95.76,99.16,102.24,104.07,105.74,114.07,118.73$, 119.56, 131.07, 135.98, 143.69, 149.72, 150.28, 157.53, 161.57, 161.99; MS (EI, 70 eV ) m/ $z: 325.1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 325.0950; found: 325.0954.
5.2.3.6. 6,7-Methylenedioxy-4-(2,5-dimethoxyphenyl)quinolin-2(1H)-one (12f): Compound $12 \mathbf{f}(0.10 \mathrm{~g}, 0.31 \mathrm{mmol})$ was obtained by cyclization of $11 \mathbf{f}(0.17 \mathrm{~g}, 0.50 \mathrm{mmol})$ with PPA (10 g); Yield: $62 \%$; white solid; mp: $240-241^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1669.46$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.59\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.69\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 5.99$, 6.02 (s, each $1 \mathrm{H},-\mathrm{O}_{\left.-\mathrm{CH}_{2}-\mathrm{O}-\right), ~} 6.14(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.33(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.74(\mathrm{~d}, J=2.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{A} \mathrm{r}-\mathrm{H}), 6.98$ (dd, $J=8.9,2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 7.06(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}$, Ar-H), 11.71 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 55.94,56.38,95.77$, 102.27, 103.96, 113.22, 113.60, 115.31, 116.03, 119.38, 127.04, 136.06, 143.73, 149.50, 150.39 (2C), 153.61, 161.85; MS (EI, 70 eV ) m/z: $325.2\left(\mathrm{M}^{+}\right.$); HRMS (EI) m/z: calculated for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 325.0950; found: 325.0959 .

### 5.2.3.7. 6,7-Methylenedioxy-4-(2,6-dimethoxyphenyl)quinolin-2(1H)-one (12g):

 Compound $\mathbf{1 2 g}(0.49 \mathrm{~g}, 1.51 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 g}(0.75 \mathrm{~g}, 2.18 \mathrm{mmol})$ with PPA (10 g); Yield: 69\%; white solid; mp: 280-281 ${ }^{\circ} \mathrm{C}$; IR ( KBr ) v $\left(\mathrm{cm}^{-1}\right): 1654.03$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}-d_{1}\right): \delta 3.68\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right), 5.93\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2} \underline{2}^{-}\right.$ O-), 6.48 (s, 2H, Ar-H), 6.65 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 7.35(\mathrm{t}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 12.99$ (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}-d_{1}$ ): $\delta 55.91$ (2C), 96.48, 101.53, 103.72, 104.05 (2C), 114.71, 115.38, 119.94, 130.12, 135.68, 144.23, 146.92, $150.38,157.62(2 \mathrm{C}), 164.53$; MS (EI, 70 eV ) m/z: 325.3 ( $\mathrm{M}^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 325.0950; found: 325.0954.5.2.3.8. 6,7-Methylenedioxy-4-(3,4-dimethoxyphenyl)quinolin-2(1H)-one (12h): Compound $\mathbf{1 2 h}(2.29 \mathrm{~g}, 7.04 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 h}(3.00 \mathrm{~g}, 8.74 \mathrm{mmol})$ with PPA (10 g); Yield: $80 \%$; white solid; mp: 272-273 ${ }^{\circ} \mathrm{C}$; IR ( KBr ) $\vee\left(\mathrm{cm}^{-1}\right): 1647.28$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.77\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 6.05$
(s, 2H, -O-Cㅐㅡ른-O-), 6.23 (s, 1H, Ar-H), 6.81 (s, 1H, Ar-H), 6.88 (s, 1H, Ar-H), 6.93 (dd, $J=$ $8.2,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.98$ (d, $J=1.8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}$ ), 7.06 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 11.73 (br. $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 56.00(2 \mathrm{C}), 96.00,102.33,103.94,112.15$, $112.61,113.14,118.59,121.35,129.97,136.74,143.78,149.17,149.56,150.41,151.71$, 161.76; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z}: 325.1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 325.0950; found: 325.0954 .

### 5.2.3.9. 6,7-Methylenedioxy-4-(3,5-dimethoxyphenyl)quinolin-2(1H)-one (12i):

 Compound $\mathbf{1 2 i}(0.41 \mathrm{~g}, 1.26 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 i}(0.75 \mathrm{~g}, 2.18 \mathrm{mmol})$ with PPA (10 g); Yield: $57 \%$; white solid; mp: 283-284 ${ }^{\circ} \mathrm{C}$; IR ( KBr ) v $\left(\mathrm{cm}^{-1}\right)$ : 1654.03 $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO- $d_{6}$ ): $\delta 3.74\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right), 6.03\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}{ }^{-}\right.$ O-), 6.20 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 6.49-6.57 (m, 3H, Ar-H), 6.71 (s, 1H, Ar-H), 6.86 (s, 1H, Ar-H), 11.73 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 55.83$ (2C), 96.00, 100.87, 102.38, $103.79,106.96$ (2C), 112.87, 118.56, 136.73, 139.59, 143.82, 150.54, 151.70, 160.97 (2C), 161.64; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z}: 325.2\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{5}$ : 325.0950; found: 325.0954.5.2.3.10. 6,7-Methylenedioxy-4-(2,4,6-trimethoxyphenyl)quinolin-2(1H)-one (12j): Compound $\mathbf{1 2 j}$ ( $0.12 \mathrm{~g}, 0.34 \mathrm{mmol}$ ) was obtained by cyclization of $\mathbf{1 1 j}(0.22 \mathrm{~g}, 0.59 \mathrm{mmol})$ with PPA ( 10 g ); Yield: $57 \%$; white solid; mp: $287-288{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \vee\left(\mathrm{cm}^{-1}\right): 1653.07$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $\left(200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 3.60\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right), 3.80\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right)$, 5.99, 6.03 ( s , each $\left.1 \mathrm{H},-\mathrm{O}-\underline{C H}_{2}-\mathrm{O}-\right), 6.26(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.32(\mathrm{~s}, 2 \overline{\mathrm{H}}, \mathrm{Ar}-\mathrm{H}), 6.81(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-$ H), 8.26 (s, 1H, Ar-H), 11.62 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 55.81$, 56.17 (2C), 91.46 (2C), $95.73,102.14,103.52,106.78,114.72,121.07,136.01,143.64$, 146.02, 150.15, 158.19 (2C), 161.92, 162.10; MS (EI, 70 eV ) m/z: 355.3 ( ${ }^{+}$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{19} \mathrm{H}_{17} \mathrm{NO}_{6}$ : 355.1056; found: 355.1048.
5.2.3.11. 6,7-Methylenedioxy-4-(3,4,5-trimethoxyphenyl)quinolin-2(1H)-one (12k): Compound $\mathbf{1 2 k}(0.17 \mathrm{~g}, 0.48 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 k}(0.25 \mathrm{~g}, 0.67 \mathrm{mmol})$ with PPA ( 10 g ); Yield: $71 \%$; white solid; $\mathrm{mp}: 275-276{ }^{\circ} \mathrm{C}$; IR (KBr) v $\left(\mathrm{cm}^{-1}\right): 1647.28$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 3.68\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 3.75\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times-\mathrm{OCH}_{3}\right)$, 6.03 (s, 2H, -O-CH2 $\underline{H}_{2}-\mathrm{O}$ ), $6.23(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.65(\mathrm{~s}, 2 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.86(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 11.72$ (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 56.48$ (2C), 60.53, $96.00,102.37,103.97,106.36$ (2C), 113.07, 118.65, 133.12, 136.68, 138.03, 143.87, 150.52, 151.91, 153.39 (2C), 161.75; MS (EI, 70 eV ) m/z: $355.2\left(\mathrm{M}^{+}\right.$); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{19} \mathrm{H}_{17} \mathrm{NO}_{6}$ : 355.1056; found: 355.1050 .
5.2.3.12. 4-(2-Fluorophenyl)-6,7-methylenedioxyquinolin-2(1H)-one (121): Compound $121(0.11 \mathrm{~g}, 0.39 \mathrm{mmol})$ was obtained by cyclization of $111(0.21 \mathrm{~g}, 0.70 \mathrm{mmol})$ with PPA $(10 \mathrm{~g})$; Yield: $55 \%$; white solid; mp: $251-252{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v}\left(\mathrm{cm}^{-1}\right)$ : $1653.07(\mathrm{C}=0) ;{ }^{1} \mathrm{H}$
 $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.91$ (s, 1H, Ar-H), 7.33-7.47 (m, 3H, Ar-H), 7.51-7.62 (m, 1H, Ar-H), 11.87 (br. s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz, DMSO- $d_{6}$ ): $\delta 95.99,102.47,103.42,112.96$, $116.38\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 120.11,124.89\left(\mathrm{~d},{ }^{2} J_{\mathrm{cF}}=15.8 \mathrm{~Hz}, 1 \mathrm{C}\right), 125.52,131.39$, $131.67\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8.0 \mathrm{~Hz}, 1 \mathrm{C}\right), 136.46,144.04,146.18,150.73,159.02\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=245.0 \mathrm{~Hz}\right.$, 1C), 161.47; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z} .283 .1$ (M ); HRMS (EI) $\mathrm{m} / \mathrm{z}$. calculated for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{FNO}_{3}$ : 283.0645; found: 283.0641 .
5.2.3.13. 4-(3-Fluorophenyl)-6,7-methylenedioxyquinolin-2(1H)-one ( 12 m ): Compound $12 \mathbf{m}(0.13 \mathrm{~g}, 0.46 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 m}(0.21 \mathrm{~g}, 0.70 \mathrm{mmol})$ with PPA $(10 \mathrm{~g})$; Yield: $65 \%$; white solid; mp: $249-251^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v}\left(\mathrm{cm}^{-1}\right)$ : $1654.03(\mathrm{C}=0) ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz, DMSO- $d_{6}$ ): 56.07 (s, 2H,-o-CH2 $\underline{2}^{-0-}$ ), 6.25 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), 6.67 (s, 1H, Ar-
H), 6.90 (s, 1H, Ar-H), 7.24-7.37 (m, 3H, Ar-H), 7.51-7.62 (m, 1H, Ar-H) 11.81 (br. s, 1H, $\mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $596.08,102.44,103.64,112.62,115.97\left(\mathrm{~d},{ }^{2} J_{\mathrm{cf}}=\right.$ $23.0 \mathrm{~Hz}, 2 \mathrm{C}), 119.08,125.22,131.28\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8.5 \mathrm{~Hz}, 1 \mathrm{C}\right), 136.83,139.84\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8.0\right.$ $\mathrm{Hz}, 1 \mathrm{C}), 143.94,150.38,150.64,161.51,162.54\left(\mathrm{~d},{ }^{1} J_{\mathrm{f}}=243.5 \mathrm{~Hz}, 1 \mathrm{C}\right)$; MS (EI, 70 eV$) \mathrm{m} /$ z: 283.1 (M ); HRMS (EI) $\mathrm{m} / \mathrm{z}$ : calculated for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{FNO}_{3}$ : 283.0645; found: 283.0648.
5.2.3.14. 4-(4-Fluorophenyl)-6,7-methylenedioxyquinolin-2(1H)-one (12n): Compound 12b $(0.17 \mathrm{~g}, 0.60 \mathrm{mmol})$ was obtained by cyclization of $\mathbf{1 1 b}(0.21 \mathrm{~g}, 0.70 \mathrm{mmol})$ with PPA $(10 \mathrm{~g})$; Yield: $85 \%$; white solid; mp: $251-252{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr})$ v $\left(\mathrm{cm}^{-1}\right): 1668.50(\mathrm{C}=0) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): 56.02 ( $\left.\mathrm{s}, 2 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{O}-\right), 6.19(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 6.62(\mathrm{~s}, 1 \mathrm{H}$, Ar-H), 6.86 (s, 1H, Ar-H), 7.25-7.34 (m, 2H, Ar-H), 7.40-7.47 (m, 2H, Ar-H), 11.76 (br. s, $1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 96.06,102.41,103.67,112.90,116.11\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}\right.$ $=22.0 \mathrm{~Hz}, 2 \mathrm{C}), 119.02,131.15\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8.5 \mathrm{~Hz}, 2 \mathrm{C}\right), 133.94,136.79,143.88,150.56$, 150.74, 161.58, $162.70\left(d,{ }^{1} J_{\mathrm{CF}}=244.5 \mathrm{~Hz}, 1 \mathrm{C}\right)$; MS (EI, 70 eV ) $\mathrm{m} / \mathrm{z} .283 .1\left(\mathrm{M}^{+}\right)$; HRMS (EI) $\mathrm{m} / \mathrm{z}$. calculated for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{FNO}_{3}$ : 283.0645; found: 283.0636 .

### 5.3. Biological evaluation

5.3.1. Antiproliferative assay-Human tumor cell lines of the cancer screening panel were maintained in RPMI-1640 medium supplemented with $10 \%$ fetal bovine serum (GIBCO/BRL), penicillin ( $100 \mathrm{U} / \mathrm{mL}$ )/streptomycin ( $100 \mathrm{~g} / \mathrm{mL}$ ) ( $\mathrm{GIBCO} / \mathrm{BRL}$ ) and $1 \% \mathrm{~L}-$ glutamine (GIBCO/BRL) at $37{ }^{\circ} \mathrm{C}$ in a humidified atmosphere containing $5 \% \mathrm{CO}_{2}$. Human hepatoma Hep 3B and normal skin Detroit 551 cells were maintained in DMEM medium supplemented with $10 \%$ fetal bovine serum (GIBCO/BRL), penicillin ( $100 \mathrm{U} / \mathrm{mL}$ )/ streptomycin ( $100 \mathrm{~g} / \mathrm{mL}$ ) (GIBCO/BRL) and $1 \% \mathrm{~L}$-glutamine (GIBCO/BRL) at $37{ }^{\circ} \mathrm{C}$ in a humidified atmosphere containing $5 \% \mathrm{CO} 2$. Logarithmically growing cancer cells were used for all experiments. The human tumor cell lines were treated with vehicle or test compounds for 48 h . Cell growth rate was determined by MTT [3-(4,5-dimethylthiazol-2-yl)-2,5diphenyltetrazoliun bromide] reduction assay. ${ }^{37,38}$ After 48 h treatment, cell growth rate was measured by scanning with an ELISA reader with a 570 nm filter and the $\mathrm{IC}_{50}$ values of test compounds were calculated.
5.3.2. Hoechst 33258 staining-HL-60 cells were plated at a density of $1 \times 10^{5}$ cells per well in 24-well plates and then incubated with $0.25 \mu \mathrm{M}, 0.5 \mu \mathrm{M}$, and $1.0 \mu \mathrm{M}$, of compound 12e for 24 h . H460 cells were plated at a density of $5 \times 10^{4}$ cells per well in 24 -well plates and then incubated with $1.0 \mu \mathrm{M}, 2.5 \mu \mathrm{M}$, and $5.0 \mu \mathrm{M}$, of compound 12e for 24 h . Cells were examined directly and photographed under a phase contrast microscope. Nuclei were stained with Hoechst 33258 (bis-benzimide, Sigma) to detect chromatin condensation or nuclear fragmentation, morphological characteristics of apoptosis. Compound 12e treated cells were stained with $5 \mu \mathrm{~g} / \mathrm{mL}$ Hoechst 33258 for 10 min . After washing twice with PBS, cells were fixed with $4 \%$ paraformaldehyde (PFA) in PBS for 10 min at $25^{\circ} \mathrm{C}$. Fluorescence of the soluble DNA (apoptotic) fragments was measured in a Varian Fluorometer at an excitation wavelength of 365 nm and emission wavelength of $460 \mathrm{~nm} .{ }^{39}$
5.3.3. Cell cycle distribution analysis—Cell cycle analysis by FACS ${ }_{\circledR}$ was performed as described in the previous paper. ${ }^{40}$ HL-60 cells were co-treated with compound $\mathbf{1 2 e}(0.25$ $\mu \mathrm{M}, 0.5 \mu \mathrm{M}$, and $1.0 \mu \mathrm{M}$ ) for 24 h , and H 460 were also co-treated with compound $\mathbf{1 2 e}$ ( 1.0 $\mu \mathrm{M}, 2.5 \mu \mathrm{M}$, and $5.0 \mu \mathrm{M}$ ). After treatment, the cells were washed once with PBS and fixed with $70 \%$ ice-cold ethanol at $-20^{\circ} \mathrm{C}$ overnight. Then the cells were stained with a solution containing $0.1 \%$ Triton-X 100 (Sigma), $0.2 \mathrm{mg} / \mathrm{mL}$ RNase (Sigma) and $20 \mu \mathrm{~g} / \mathrm{mL}$ propidium iodide (PI, Sigma) in the dark for 30 min . Cell cycle distribution were measured using a FACScan flow cytometer (Becton Dickinson, San Jose, CA, USA) and all histograms were analyzed by ModFit software.
5.3.4. Western blot assay-The treated cells were collected and washed with PBS. After centrifugation, cells were lysed in a lysis buffer. The lysates were incubated on ice for 30 min and centrifuged at $12,000 g$ for 20 min . Supernatants were collected, and protein concentrations were then determined using the Bradford Assay. After adding a $5 \times$ sample loading buffer containing 625 mM Tris $-\mathrm{HCl}, \mathrm{pH}=6.8,500 \mathrm{mM}$ dithiothreitol, $10 \%$ SDS, $0.06 \%$ bromophenol blue, and $50 \%$ glycerol, protein samples were electrophoresed on $10 \%$ SDS-polyacrylamide gels and transferred to a nitrocellulose membrane. Immunoreactivity was detected using the Western blot chemiluminescence reagent system (PerkinElmer, Boston, MA). $\beta$-Actin was used as a loading control.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Podophyllotoxin (1)


Etoposide (2) $\quad \mathrm{R}_{1}=\mathrm{CH}_{3} \quad ; \mathrm{R}_{2}=\mathrm{H}$ Teniposide (3) $\quad \mathrm{R}_{1}=$

Etopophos (4) $\quad \mathrm{R}_{1}=\quad \mathrm{CH}$


Figure 1.
Structures of podophyllotoxin and related compounds: podophyllotoxin (1) and its semisynthetic derivatives in clinical use, etoposide (2), teniposide (3) and etopophos (4).


4-Aza-podophyllotoxin (5)


Podophyllic aldehyde (6)


Combretastatin A-4 (7)

2-Phenylquinolin-4-one analogs (2-PQs)
4-Phenylquinolin-2-one analogs (4-PQs; 12a-n)

Figure 2.
Stuctures of 4-aza-PPT analogues (5), podophyllic aldehyde (6), combretastatin A-4 (7), 2PQ derivatives and the target compounds (12a-n).


Podophyllic aldehyde (6)

Figure 3.
Rationally designed 4-PQ analogs mimic structures of PPT analogues and CA-4.


Figure 4.
Fluorescent images of Hoechst staining showing compound 12e induced cell death. (A)
Treatment of HL-60 cells with 12e for 24 h . The arrows indicate the formation of apoptotic bodies. (B) Treatment of H460 cells with 12e for 24 h . The arrows indicate the formation of apoptotic bodies. Scale bar $=50 \mu \mathrm{~m}$.
A




B





| Cell line | Conc. $(\mu \mathrm{M})$ | Cell cycle phase (\%) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Sub G1 | $\mathrm{G} 0 / \mathrm{G} 1$ | S | $\mathrm{G} 2 / \mathrm{M}$ |
| Control | 0 | 0.35 | 39.80 | 49.69 | 10.51 |
| HL-60 | 0.25 | 13.88 | 41.36 | 46.76 | 11.87 |
|  | 0.5 | 27.86 | 44.82 | 30.16 | 25.02 |
| Control | 0 | 19.65 | 12.46 | 46.99 | 40.54 |
|  | 1.0 | 10.80 | 1.57 | 44.06 | 54.37 |
| H460 | 2.5 | 11.65 | 9.60 | 18.82 | 71.58 |

Figure 5.
Compound 12e affected the cell cycle distribution. (A) HL-60 cells treated with 12e for 24 h. (B) H460 cells treated with 12e for 24 h .


Figure 6.
Regulation of mitotic phase- and apoptosis-associated proteins expression by compound 12e. (A) HL-60 cells treated with the indicated concentration of 12e for 24 h . (B) H 460 cells treated with the indicated concentration of 12e for 24 h . After treatment, the cells were harvested and subjected to Western blot. The relative amounts of cyclin B1, CDK1, and caspase 3 proteins were quantified and normalized to the corresponding $\beta$-actin protein amount. The quantitative data are shown under each protein, respectively.



|  | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | $\mathrm{R}^{4}$ | $\mathrm{R}^{5}$ | $\mathrm{R}^{6}$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{a}:$ | H | H | H | H | H |
| $\mathbf{b}:$ | $\mathrm{OCH}_{3}$ | H | H | H | H |
| $\mathbf{c}:$ | H | $\mathrm{OCH}_{3}$ | H | H | H |
| $\mathbf{d}:$ | H | H | $\mathrm{OCH}_{3}$ | H | H |
| $\mathbf{e}:$ | $\mathrm{OCH}_{3}$ | H | $\mathrm{OCH}_{3}$ | H | H |
| $\mathbf{f}:$ | $\mathrm{OCH}_{3}$ | H | H | $\mathrm{OCH}_{3}$ | H |
| $\mathbf{g}:$ | $\mathrm{OCH}_{3}$ | H | H | H | $\mathrm{OCH}_{3}$ |
| $\mathbf{h}:$ | H | $\mathrm{OCH}_{3}$ | $\mathrm{OCH}_{3}$ | H | H |
| $\mathbf{i}:$ | H | $\mathrm{OCH}_{3}$ | H | $\mathrm{OCH}_{3}$ | H |
| $\mathbf{j}:$ | $\mathrm{OCH}_{3}$ | H | $\mathrm{OCH}_{3}$ | H | $\mathrm{OCH}_{3}$ |
| $\mathbf{k}:$ | H | $\mathrm{OCH}_{3}$ | $\mathrm{OCH}_{3}$ | OCH | H |
| $\mathbf{l}:$ | F | H | H | H | H |
| $\mathbf{m}:$ | H | F | H | H | H |
| $\mathbf{n}:$ | H | H | F | H | H |

Scheme 1.
Reagents and conditions: (a) NaH , toluene, reflux; (b) 3,4-methylenedioxy aniline, toluene, reflux; (c) PPA, $100-110{ }^{\circ} \mathrm{C}$.


| Compd | $\mathbf{R}^{2}$ | $\mathbf{R}^{3}$ | $\mathbf{R}^{4}$ | $\mathrm{R}^{5}$ | $\mathrm{R}^{6}$ | $\mathrm{IC}_{50}(\mu \mathrm{M})^{\boldsymbol{a}}$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  | MCF-7 | 2774 | SKOV-3 | HL-60 | Hep 3B | H460 | COLO 205 | A498 | Detroit 551 |
| 12n | H | H | F | H | H | >30 | 17.6 | >30 | >60 | >20 | >20 | >50 | >50 | >20 |
| Etoposide |  |  |  |  |  |  |  |  | 5.48 |  | 1.0 |  |  |  |
| 5-FU |  |  |  |  |  |  |  |  | 22.3 |  | 26.7 |  |  |  |


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    *Corresponding authors. KHL: Tel: (919) 962-0066; fax: (919) 966-3893; khlee@unc.edu; LJH: Tel: +886 422053366 / Ext 1007; fax: +8864 22030760; ljhuang @ mail.cmu.edu.tw.
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