

Delitpyrones: α -Pyrone Derivatives from a Freshwater *Delitschia* sp.

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Abstract:

In research focused on the discovery of new chemical diversity from freshwater fungi, a peak library was built and evaluated against a prostate cancer cell line, E006AA-hT, which was derived from an African American, as this population is disproportionately affected by prostate cancer. The chemical study of the bioactive sample accessioned as G858 (*Delitschia* sp.) led to the isolation of eight new α -pyrone derivatives (**1–7**, and **11**), as well as the new 3*S**,4*S**-7-ethyl-4,8-dihydroxy-3,6-dimethoxy-3,4-dihydronaphthalen-1(2*H*)-one (**15**). In addition, the known compounds 5-(3-*S*-hydroxybutyl)-4-methoxy-6-methyl-2*H*-pyran-2-one (**8**), 5-(3-oxobutyl)-4-methoxy-6-methyl-2*H*-pyran-2-one (**9**), pyrenocine I (**10**), 5-butyl-6-(hydroxymethyl)-4-methoxy-2*H*-pyran-2-one (**12**), sporidesmin A (**13**), 6-ethyl-2,7-dimethoxyjuglone (**14**), artrichitin (**16**), and lipopeptide 15G256 ϵ (**17**) were also obtained. The structures of the new compounds were elucidated using a set of spectroscopic (NMR) and spectrometric (HRMS) methods. The absolute configuration of the most abundant member of each subclass of compounds was assigned through a modified Mosher's ester method. For **15**, the relative configuration was assigned based on analysis of ³*J* values. Compounds **1**, **2**, **5–14**, **16**, and **17** were evaluated against the cancer cell line E006AA-hT under hypoxic conditions, where compound **13** inhibited cell proliferation at a concentration of 2.5 μ M.

Keywords: peak library | prostate cancer | cell proliferation | freshwater fungi | delitpyrones | *Delitschia* sp.

Article:

Introduction

The fungal kingdom, estimated at 2.2–3.8 million species, with just 3 to 8% identified (120000) [1], is the second most diverse on the planet [2]. Among those, the freshwater ascomycetes are underinvestigated, with only about 675 species described [3]. This diverse ecological group

relies upon freshwater for the whole or part of their life cycle [3],[4] and includes any species growing on substrates that are aquatic or semiaquatic [4]. Freshwater fungi play an important role in the degradation of organic materials [5]. Chemically, the ability of freshwater ascomycetes to produce bioactive metabolites merits investigation, as less than 250 metabolites have been described [6]. Thus, to contribute to the expansion of the chemical space of secondary metabolites from this ecological niche, our team has started a systematic study of bioactive compounds from freshwater ascomycetes [7]. Recent examples include five acetophenone derivatives [8] and 14 resorcylic acid lactones with cytotoxic activity [9], as well as isochromenones, isobenzofuranones, and tetrahydronaphthalenes [10].

According to Index Fungorum (<http://www.indexfungorum.org/names/names.asp>) and the Dictionary of Fungi [11], the genus *Delitschia* Auersw. (Delitschiaceae) comprises coprophilous and saprobic taxa with around 83 species; another resource (Genera of Ascomycota) lists a more conservative 52 species. Despite the diversity in this genus, only a handful have been investigated for bioactive secondary metabolites. Interestingly, the few chemical studies conducted of this genus have led to the isolation of compounds with new scaffolds, such as delitschiapyrone A [12], an α -pyrone-naphthalenone adduct, and a fimetarone derivative [13].

In the USA, African American men have the highest risk of both developing and dying of prostate cancer, relative to Caucasians [14],[15], and there are many social, economic, and biological factors that contribute to this health disparity [15]. Hence, with the goals of understanding the biology and ameliorating this disease, we have initiated a project focused on the discovery of research tools and drug leads for the treatment of prostate cancer in African Americans. Hypoxia (low oxygen conditions) in prostate tumors is an early event associated with an aggressive phenotype. Hypoxic conditions in the tumor microenvironment promote genetic, metabolic, and proteomic changes, which lead to increased glycolysis, angiogenesis, survival, stemness, invasiveness, and selection of resistant clones [16]. Levels of hypoxia in prostate tumors, and expression of hypoxia-related biomarkers (e.g., hypoxia-inducible factor HIF1 α and HIF2 α), are associated with poor prognosis and are the major reasons for treatment failure and disease relapse [16]. Therefore, we assessed the effect of potential drugs against African American prostate cancer cells under hypoxic conditions.

In this context, a study of the aquatic fungus *Delitschia* sp. (strain G858) led to the isolation and characterization of eight new structurally related α -pyrone derivatives (**1–7**, and **11**), as well as one new dihydronaphthalenone (**15**), along with eight known fungal metabolites (**8–10**, **12–14**, **16**, and **17**). The cytotoxic activities of compounds **1**, **2**, **5–14**, **16**, and **17** were evaluated against the E006AA-hT cell line, where compound **13** showed promising bioactivity.

Results and Discussion

A library of 64 freshwater fungal isolates from diverse habitats (Fig. 58S, Supporting Information), arrayed across 320 peaks, was assembled using an 8-step procedure. Starting with a mature fungal culture, the sample was first extracted and then fractionated into four peaks. These samples were used to generate mother plates (10 mg/mL), daughter plates (5 mg/mL), and granddaughter plates (5 μ g/well); each plate represented 16 fungi over 80 wells (16 extracts plus 4 fractions from each). The plates were evaluated in a panel of bioassays, and hits were

prioritized and dereplicated [17],[18]. The parent samples of active fractions were then used for isolation and elucidation of the hits, and this included fungal culture scale-up, when needed. Finally, pure compounds were evaluated for biological activities ([Fig. 1]).

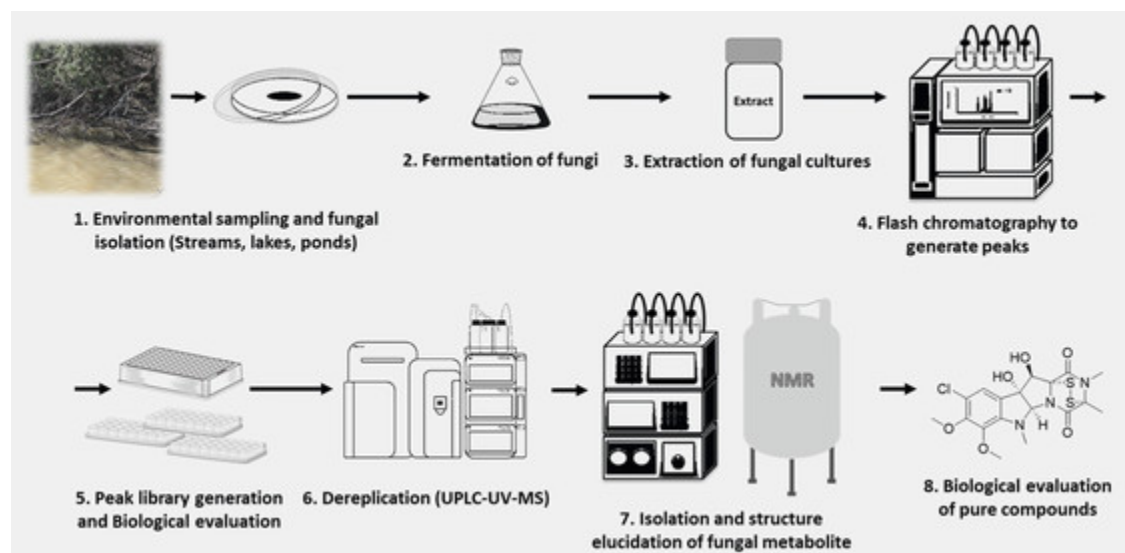


Figure 1. Strategy for uncovering new natural products from fungal peak libraries. The procedure starts with the sampling and isolation of axenic fungal strains, followed by fermentation, extraction, and fractionation into peaks. The peaks are used to generate mother plates, daughter plates, and granddaughter plates; each plate represents 16 fungal isolates over 80 wells (16 extracts plus 4 fractions from each). The plates are then evaluated in a panel of bioassays, and hits are prioritized and dereplicated. The parent samples of active fractions are used for isolation and elucidation of hits. Finally, pure compounds are evaluated for biological activities.

The library was tested against an African American prostate cancer cell line (E006AA-hT) under hypoxic conditions (1% O₂), where cancer cells are more drug resistant [19],[20]. Promising bioactivity was observed with peaks derived from strain G858 (20% viability when tested at a concentration of 2 µg/mL) (**Fig. 1S**, columns 1 and 3, Supporting Information). These samples displayed an HPLC-UV profile that indicated a series of related analogues (Supporting Information), which could not be identified using the dereplication protocols [17],[18].

A large-scale culture of G858 (*Delitschia* sp.) was grown on rice and extracted using protocols described previously [21]. The resultant extract was partitioned using organic solvents and fractionated using flash chromatography, yielding nine fractions. These were purified further using preparative and semipreparative HPLC to yield nine new natural products (**1–7**, **11**, and **15**), along with the known compounds 5-(3-S-hydroxybutyl)-4-methoxy-6-methyl-2H-pyran-2-one (**8**) [22], 5-(3-oxobutyl)-4-methoxy-6-methyl-2H-pyran-2-one (**9**) [23], pyrenocine I (**10**) [24], 5-butyl-6-(hydroxymethyl)-4-methoxy-2H-pyran-2-one (**12**) [23], sporidesmin A (**13**) [25], 6-ethyl-2,7-dimethoxyjuglone (**14**) [12],[26], artrichitin (**16**) [27], and lipopeptide 15G256ε (**17**) [28] ([Fig. 2]). The structures of the isolates were established using spectroscopic and spectrometric data, which for all known compounds compared favorably to the literature. The absolute configuration of an illustrative member of each subclass was assigned through a modified Mosher's ester method [29].

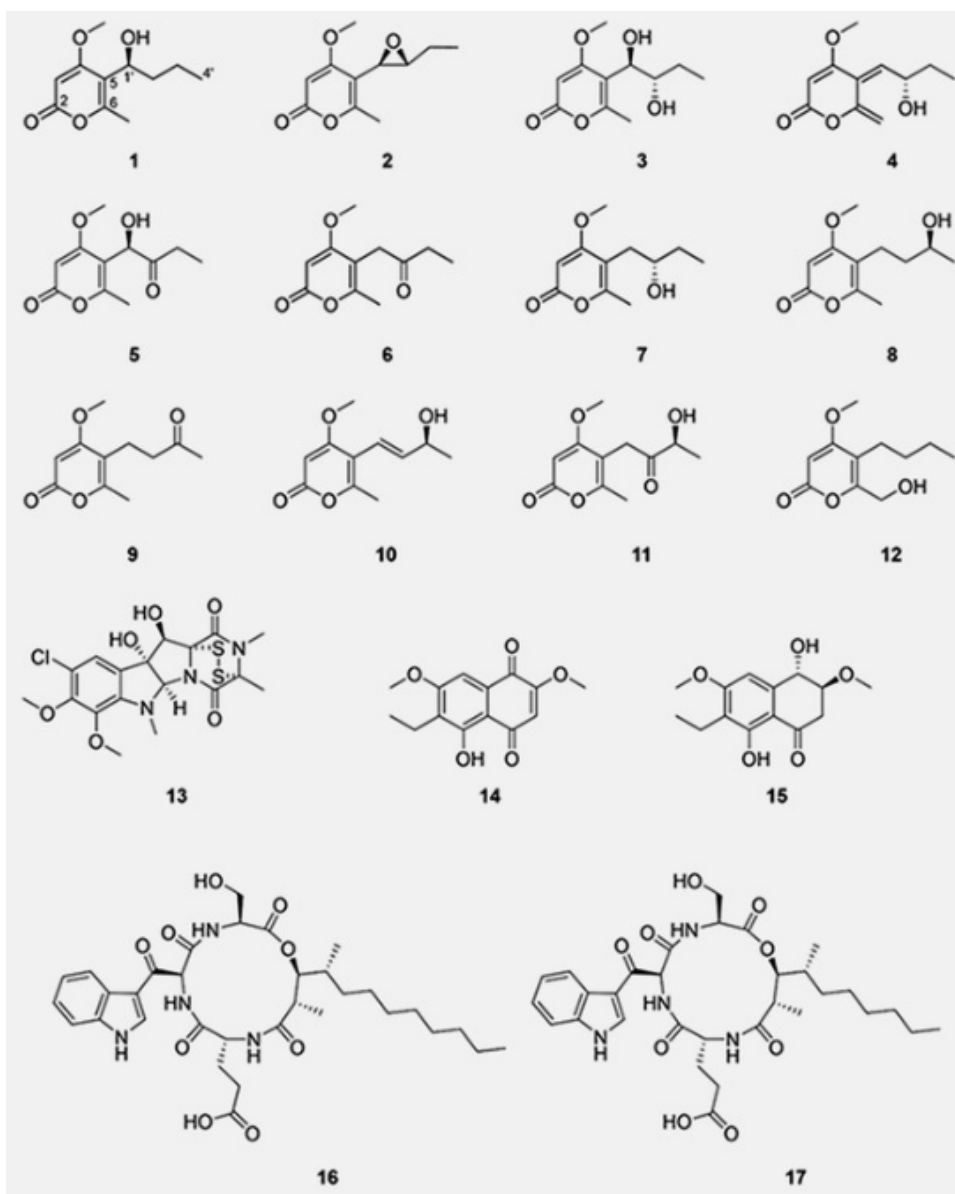


Figure 2. Structures of compounds 1–17.

Compound **1** was isolated as a colorless solid, and its molecular formula was $C_{11}H_{16}O_4$ on the basis of the HRESIMS data (m/z 213.1125), indicating an index of hydrogen deficiency of four. The 1H - and ^{13}C -NMR spectra ([Tables 1] and [2], **Figs. 5S** and **6S**, Supporting Information) in combination with HSQC data showed signals attributed to one olefinic proton (δ_H/δ_C 5.49/89.1), one methoxy group (δ_H/δ_C 3.86/56.3), one $-OH$ (δ_H 2.44), two CH_2 (δ_H/δ_C 1.64/1.86/39.4; δ_H/δ_C 1.28/1.42/19.5), and two CH_3 groups (δ_H/δ_C 0.93/14.0; δ_H/δ_C 2.29/17.9), as well as four quaternary carbons, including three olefinic (δ_C 170.6, 113.2, and 159.1) and one carbonyl (δ_C 164.0), attributed to an α,β unsaturated lactone. In general, these data showed notable similarity to that of (*R*)-5-(1-hydroxybutyl)-4-methoxy-6-methyl-2*H*-pyran-2-one, an α -pyrone derivative isolated from *Alternaria phragmospora* [23]. In our study, the absolute configuration at C-1' in **1** was established as *S* by comparison of its specific rotation ($[\alpha]_D^{27} = -15$) with that reported for its enantiomer (1'*R*)-

2',3'-dihydropyrenocine C ($[\alpha]_D^{27} = +15.7$) [12], and based on the outcomes obtained after applying a modified Mosher's method ([Fig. 3]) [29]. Thus, compound **1** was identified as (1'S)-2',3'-dihydropyrenocine C, and assigned the trivial name delityprone A (**1**).

Table 1. $^1\text{H-NMR}$ data for compounds **1–3**, **5–7**, and **11**.

Position	δ_{H} , multiplicity ($J = \text{Hz}$)						
	1	2	3	5	6	7	11
3	5.49, <i>s</i>	5.45, <i>s</i>	5.54, <i>s</i>	5.48, <i>s</i>	5.46, <i>s</i>	5.58, <i>s</i>	5.47, <i>s</i>
1'	4.63, <i>q</i> (6.5)	3.71, <i>d</i> (4.0)	5.15, <i>d</i> (6.2)	4.98, <i>d</i> (3.5)	3.42, <i>s</i>	2.41, <i>dd</i> (14.1, 8.6) 2.58, <i>dd</i> (14.1, 4.2)	3.48, <i>d</i> (17.9); 3.55, <i>d</i> (17.9)
2'	1.64, <i>ddt</i> (13.5, 10.2, 6.1) 1.86, <i>dddd</i> (13.4, 10.1, 8.0, 5.3)	3.10, <i>ddd</i> (6.9, 5.2, 4.1)	4.52, <i>q</i> (6.2)			3.55, <i>tt</i> (8.5, 4.4)	
3'	1.28, <i>m</i> ; 1.42, <i>m</i>	1.25, <i>dp</i> (14.4, 7.3) 1.35, <i>m</i>	1.77, <i>m</i> 1.86, <i>dp</i> (14.7, 7.4)	2.35, <i>q</i> (7.5); 2.35, <i>qd</i> (7.5, 0.5)	2.50, <i>q</i> (7.3)	1.46, <i>m</i> ; 1.53, <i>m</i>	4.38, <i>dq</i> (6.9, 5.0)
4'	0.93, <i>t</i> (7.4)	0.99, <i>t</i> (7.5)	1.06, <i>t</i> (7.4)	1.11, <i>t</i> (7.3)	1.09, <i>t</i> (7.3)	0.98, <i>t</i> (7.4)	1.47, <i>d</i> (7.1)
4-OCH ₃	3.86, <i>s</i>	3.82, <i>s</i>	3.89, <i>s</i>	3.78, <i>s</i>	3.78, <i>s</i>	3.88, <i>s</i>	3.77, <i>s</i>
6-CH ₃	2.29, <i>s</i>	2.34, <i>s</i>	2.34, <i>s</i>	2.33, <i>s</i>	2.17, <i>s</i>	2.28, <i>s</i>	2.19, <i>s</i>
1'-OH	2.44, <i>d</i> (7.7)			4.02, <i>d</i> (3.7)			
3'-OH							3.28, <i>d</i> (5.0)

For solvents and magnetic field used for each compound, see the Materials and Methods section

Table 2. $^{13}\text{C-NMR}$ data for compounds **1–3**, **5–7**, and **11**.

Position	δ_{C} , type													
	1		2		3		5		6		7		11	
2	164.0	C	164.3	C	162.5	C	163.3	C	164.3	C	167.4	C	169.5	C
3	89.1	CH	87.6	CH	89.4	CH	88.6	CH	88.0	CH	88.3	CH	88.2	CH
4	170.6	C	170.2	C	169.2	C	169.2	C	170.0	C	173.4	C	164.0	C
5	113.2	C	106.8	C	107.1	C	109.6	C	105.8	C	110.9	C	104.8	C
6	159.1	C	161.7	C	154.4	C	162.9	C	160.1	C	161.4	C	160.4	C
1'	68.7	CH	51.7	CH	76.7	CH	71.6	CH	37.8	CH ₂	33.1	CH ₂	33.6	CH ₂
2'	39.4	CH ₂	58.7	CH	81.7	CH	208.7	C	207.0	C	73.4	CH	208.5	C
3'	19.5	CH ₂	21.8	CH ₂	28.1	CH ₂	30.8	CH ₂	35.6	CH ₂	31.1	CH ₂	72.7	CH
4'	14.0	CH ₃	10.4	CH ₃	8.9	CH ₃	7.9	CH ₃	8.0	CH ₃	10.7	CH ₃	20.2	CH ₃
4-OCH ₃	56.3	CH ₃	56.2	CH ₃	56.9	CH ₃	56.4	CH ₃	56.4	CH ₃	57.1	CH ₃	56.5	CH ₃
6-CH ₃	17.9	CH ₃	18.2	CH ₃	17.8	CH ₃	17.7	CH ₃	17.6	CH ₃	17.9	CH ₃	17.7	CH ₃

For solvents and magnetic field used for each compound, see the Materials and Methods section

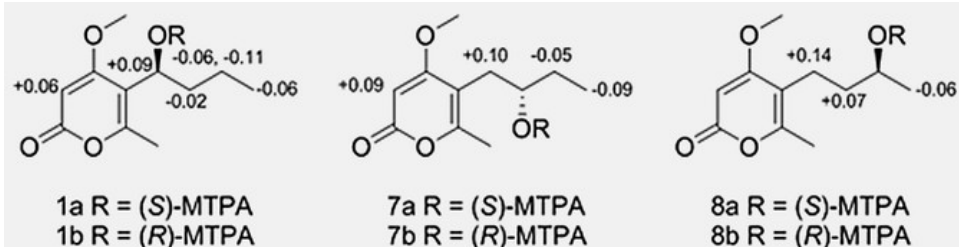


Figure 3. $\Delta\delta_{\text{H}}$ values [$\Delta\delta$ (in ppm) = $\delta_{\text{S}} - \delta_{\text{R}}$] obtained for compounds **1**, **7**, and **8**.

Compound **2** was obtained as a colorless solid. The molecular formula was $\text{C}_{11}\text{H}_{14}\text{O}_4$ based on the HRESIMS data (m/z 211.0955), indicating five degrees of unsaturation. The 1D-NMR spectra of **2** were similar to those of **1**, except for the chemical shift of the proton and carbon signals attributed to the oxymethine at C-1' in **1** ($\delta_{\text{H}}/\delta_{\text{C}}$ 4.63/68.7), which were both relatively more shielded in **2** ($\delta_{\text{H}}/\delta_{\text{C}}$ 3.71/51.7). Additionally, the replacement of a methylene group in **1** for an oxymethine group ($\delta_{\text{H}}/\delta_{\text{C}}$ 3.10/58.7) in **2**, as well as the increase in the unsaturation number, indicated **2** as 1',2'-epoxi-delitpyrone A (**2**). Key HMBC correlations ([Fig. 4]) confirmed the structure of **2**. The spatial orientation of the substituent at position 1' was considered to be the same as in compound **1** based on the specific rotation ($[\alpha]_{\text{D}}^{27} = -5$) and biogenetic considerations. The coupling constant ($J_{\text{H-1}'/\text{H-2}'} = 4.0$ Hz) did not permit a clear consideration of the orientation across the epoxide [30]. However, the isolation from this strain of the *E* oriented putative biosynthetic precursor of **2** (data not shown), as well as other analogues, supported the *trans* disposition of the substituents in the oxirane. Thus, the absolute configuration of **2** was presumed to be 1'*R*,2'*R*. While the absolute configuration at C-1' in **1** is *S*, it became *R* in **2**, based on the Cahn-Ingold-Prelog rules.

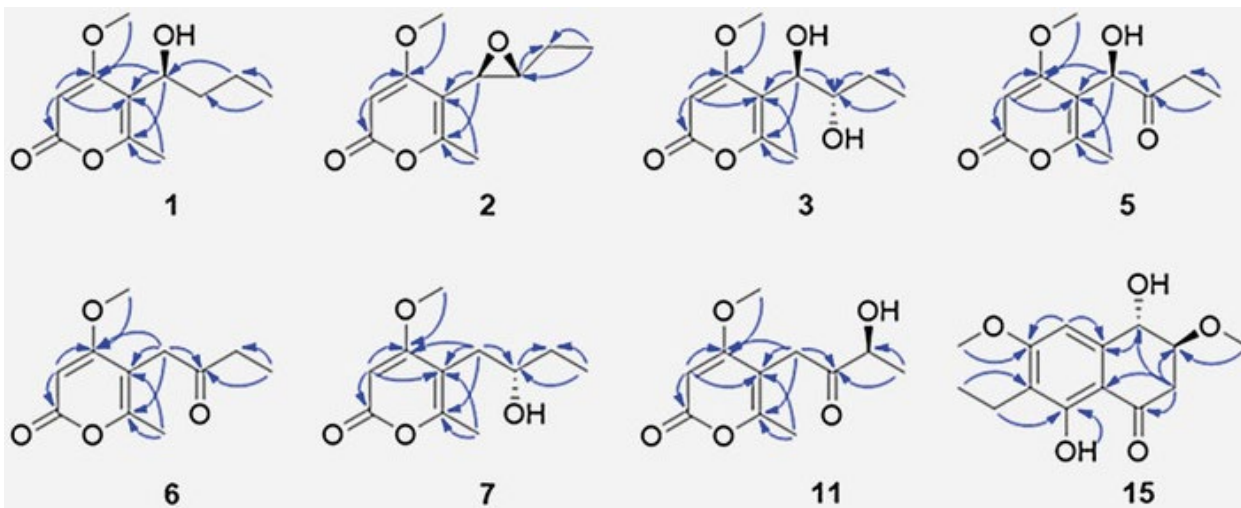


Figure 4. Key HMBC correlations of compounds **1–3**, **5–7**, **11**, and **15**.

Compound **3** (delitpyrone B) was isolated as a colorless solid, and its molecular formula was $\text{C}_{11}\text{H}_{16}\text{O}_5$ based on the HRESIMS data (m/z 229.1068). The NMR data for **3** resembled those for 1',2'-epoxi-delitpyrone A (**2**). Key differences were the deshielding of the methine groups at C-1' and C-2' ($\delta_{\text{H}}/\delta_{\text{C}}$ 5.15/76.7 and 4.52/81.7, respectively) ([Tables 1] and [2]), indicating the opening of the epoxide moiety by the nucleophilic substitution of water, yielding the diol

congener of **2**. HMBC correlations ([Fig. 4]) further confirmed the structure of **3**. The absolute configuration of compound **3** was proposed as 1'*R* and 2'*S* based on biogenic considerations.

Delitpyrone C (**4**) was isolated as a colorless solid, and its molecular formula was C₁₁H₁₄O₄ according to the HRESIMS data at *m/z* 211.0962, indicating five degrees of unsaturation. The ¹H-NMR for **4** was quite different than those recorded for **1–3**; for example, the characteristic methyl signal at δ_H ~ 2.3 ppm (6-CH₃) was replaced by two vinylic protons at δ_H 4.81 (*d*, *J* = 1.9) and 5.07 (*d*, *J* = 1.9). Additionally, comparison of the ¹H-NMR data of **4** with that of **3** indicated the replacement of the oxymethine group at C-1' with a vinylic proton δ_H 6.49 (*d*, *J* = 9.3) ([Table 3]). Altogether, this information indicated **4** as the dehydration product of **3**. Unfortunately, the 2D-NMR data for compound **4** were not collected due to the instability of the compound and the paucity of the sample. Interestingly, this compound was recently proposed as an intermediate in the biosynthesis of delitschiapyrone A, a pyrone-naphthalenone adduct isolated from the leaf-associated fungus *Delitschia* sp. FL1581 [12]. The isolation of **4** from a related species both supports the biosynthetic proposal [12] and suggests the conservation of biosynthetic machinery within this genus.

Table 3. NMR spectroscopic data for compound **4** (500 and 125 MHz, ¹H and ¹³C, respectively), recorded in acetone-*d*₆.

Position	δ _C ,	type	δ _H , multiplicity (<i>J</i> = Hz)
2	162.2	C	
3	78.7	CH	5.36, <i>s</i>
4	167.0	C	
5	123.5	C	
6	150.2	C	
1'	139.8	CH	6.49, <i>d</i> (9.3)
2'	69.3	CH	4.48, <i>m</i>
3'	31.2	CH ₂	1.60, <i>m</i>
			1.67, <i>m</i>
4'	10.0	CH ₃	0.99, <i>t</i> (7.4)
4-OCH ₃	57.1	CH ₃	3.92, <i>s</i>
6-CH _{2a}	90.6	CH ₂	4.81, <i>d</i> (1.9)
6-CH _{2b}			5.07, <i>d</i> (1.9)

Compound **5** (2'-oxodelitpyrone A), the 2'-oxo analogue of delitpyrone A, was obtained as a colorless solid. The molecular formula was C₁₁H₁₄O₅ according to the HRESIMS data (*m/z* 227.0916). Analysis of the ¹H- and ¹³C-NMR data ([Tables 1] and [2]) indicated that **5** shares the same core as **1**, with slight modifications in the side chain. The main difference was the replacement of the methylene group at the C-2' position in **1** by a carbonyl group (δ_C 208.7) in **5**. The structure of **5** was confirmed through the analysis of the HMBC spectrum ([Fig. 4]). The spatial orientation of the hydroxy group in **5** was presumed to be in a β orientation by comparison of its specific rotation ([α]_D²⁷ = -38) with that recorded for **1**. However, the absolute configuration at C-1' was assigned as *R*.

Delitpyrone D (**6**) was isolated as a colorless solid, and its molecular formula was C₁₁H₁₄O₄ according to the HRESIMS data (*m/z* 211.0967). The ¹H and ¹³C-NMR data ([Tables 1] and [2])

were found to be similar with those of **5**, with a difference observed in the chemical shift attributed to C-1'. Briefly, the oxymethine group in **5** ($\delta_{\text{H}}/\delta_{\text{C}}$ 4.98/71.6) was replaced by a methylene group in **6** ($\delta_{\text{H}}/\delta_{\text{C}}$ 3.42/37.8). The splitting pattern of H₃-4' (*t*) and H₂-3' (*q*), as well as key HMBC correlations from H₂-1' and H₂-3' to a ketone carbonyl group at C-2' ([Fig. 4]), confirmed the structure of **6** as 1'-dehydroxy-2'-oxodelitpyrone A.

Compounds **7** and **8** (a known compound) were isolated as structural isomers sharing the molecular formula C₁₁H₁₆O₄, as determined by the HRESIMS data (*m/z* 213.1122 and 213.1121, respectively), with an index of hydrogen deficiency of four. The ¹H- and ¹³C-NMR data for both compounds were similar to each other. In the case of **7**, signals for two methine groups, one vinylic proton ($\delta_{\text{H}}/\delta_{\text{C}}$ 5.58/88.3, *s* for H-3/C-3), one oxymethine ($\delta_{\text{H}}/\delta_{\text{C}}$ 3.55/73.4, C-2'), two methylene [$(\delta_{\text{H}}/\delta_{\text{C}}$ 2.41/2.58/33.1 for H-1'a/H-1'b/C-1') and ($\delta_{\text{H}}/\delta_{\text{C}}$ 1.46/1.53/31.1 for H-3'a/H-3'b/C-3')], and one methyl group ($\delta_{\text{H}}/\delta_{\text{C}}$ 0.98/10.7, *t*, *J* = 7.4 Hz for H₃-4'/C-4') were observed. Inspection of the splitting patterns in the ¹H-NMR spectra, together with HMBC correlations, established the structure of **7** as 5-(2-hydroxybutyl)-4-methoxy-6-methyl-2*H*-pyran-2-one, which was assigned the trivial name delitpyrone E. The absolute configurations at the chiral center in **7** (C-2') was established as *S* via the modified Mosher's ester method ([Fig. 3]).

Compound **11** was isolated as a colorless solid, and its molecular formula was C₁₁H₁₄O₅ on the basis of the HRESIMS (*m/z* 227.0914), indicating five degrees of unsaturation. The 1D-NMR spectra were similar with those of **6**, with the exception of the signals attributed to protons and carbons at positions 3' and 4'. Thus, the triplet and quartet in **6** ($\delta_{\text{H}}/\delta_{\text{C}}$ 2.50/35.6, *q*, *J* = 7.3 Hz for H-3'/C-3'; 1.09/8.0, *t*, *J* = 7.3 Hz for H₃-4'/C-4') were replaced by a doublet of quartets for an oxymethine and a methyl doublet in **11** ($\delta_{\text{H}}/\delta_{\text{C}}$ 4.38/72.7, *dq*, *J* = 6.9, 5.0 Hz for H-3'/C-3'; 1.47/20.2, *d*, *J* = 7.1 Hz for H₃-4'/C-4'). Thus, **11** was determined to be 3'-hydroxydelitpyrone D, and its structure was confirmed by analysis of the HMBC spectrum ([Fig. 4]). The absolute configuration at C-3' was presumed to be *S* based on biogenic considerations. The biosynthesis for compounds **1**–**11** is proposed (Fig. 56S, Supporting Information).

Compound **15** was isolated as a yellow solid, and its molecular formula was determined as C₁₄H₁₈O₅ by HRESIMS analysis (*m/z* 267.1224). The 1D- and 2D-NMR data for this compound suggested structural similarity with the known compound 6-ethyl-2,7-dimethoxyjuglone (**14**) [31], [32]. The key differences were the replacement of one *sp*² methine in **14** by a shielded methylene group at $\delta_{\text{H}}/\delta_{\text{C}}$ (2.56; 3.15/40.9, *dd*, *J* = 16.8, 10.5 Hz; *dd*, *J* = 16.8, 4.4 Hz for H-3a/H-3b/C-3) in **15**, as well as the ¹³C signals at C-4 and C-1 that were shielded from δ_{C} 179.8 to 72.1, and deshielded from δ_{C} 190.9 to 199.4, respectively, supporting the reduction of the *p*-naphthoquinone in **14** to a dihydronaphthalen-1(2*H*)-one moiety in **15** ([Table 4]). Inspection of the splitting patterns of signals, together with HMBC correlations ([Fig. 4]), established the structure of **15** as 7-ethyl-4,8-dihydroxy-3,6-dimethoxy-3,4-dihydronaphthalen-1(2*H*)-one. The relative configurations at positions C-3 and C-4 were established based on the analysis of coupling constant values and by comparison of the ¹H-NMR data collected for **15** with that reported for (3*S*,4*S*)-3,4,8-trihydroxy-3,4-dihydro-1(2*H*)-naphthalenone [33]. Briefly, a *trans*-pseudoaxial orientation between H-3 and H-4 was evident by the coupling constant ³*J*_{H-3-H-4} = 8.7 Hz, placing the hydroxy and methoxy substituents in an *anti*-orientation. Therefore, compound **15** was characterized as 3*S**,4*S**-7-ethyl-4,8-dihydroxy-3,6-dimethoxy-3,4-dihydronaphthalen-1(2*H*)-one.

Table 4. NMR data for compound **15** (700 and 175 MHz, ^1H and ^{13}C , respectively), recorded in CDCl_3 .

Position	δ_{C}	type	δ_{H} , multiplicity ($J = \text{Hz}$)
1	199.4	C	
2	40.9	CH_2	2.56, <i>dd</i> (16.8, 10.5) 3.15, <i>dd</i> (16.8, 4.4)
3	80.4	CH	3.60, <i>ddd</i> (10.6, 8.7, 4.4)
4	72.1	CH	4.71, <i>dd</i> (8.8, 2.1)
4a	142.6*	C	
5	100.3	CH	6.80, <i>s</i>
6	164.3	C	
7	118.8	C	
8	161.8	C	
8a	110.1	C	
1'	15.7	CH_2	2.65, <i>q</i> (7.5)
2'	13.3	CH_3	1.08, <i>t</i> (7.5)
3-OCH ₃	57.2	CH_3	3.49, <i>s</i>
6-OCH ₃	56.0	CH_3	3.94, <i>s</i>
4-OH			2.90, <i>s</i>
8-OH			12.62, <i>s</i>

* Signal obtained from HMBC spectrum

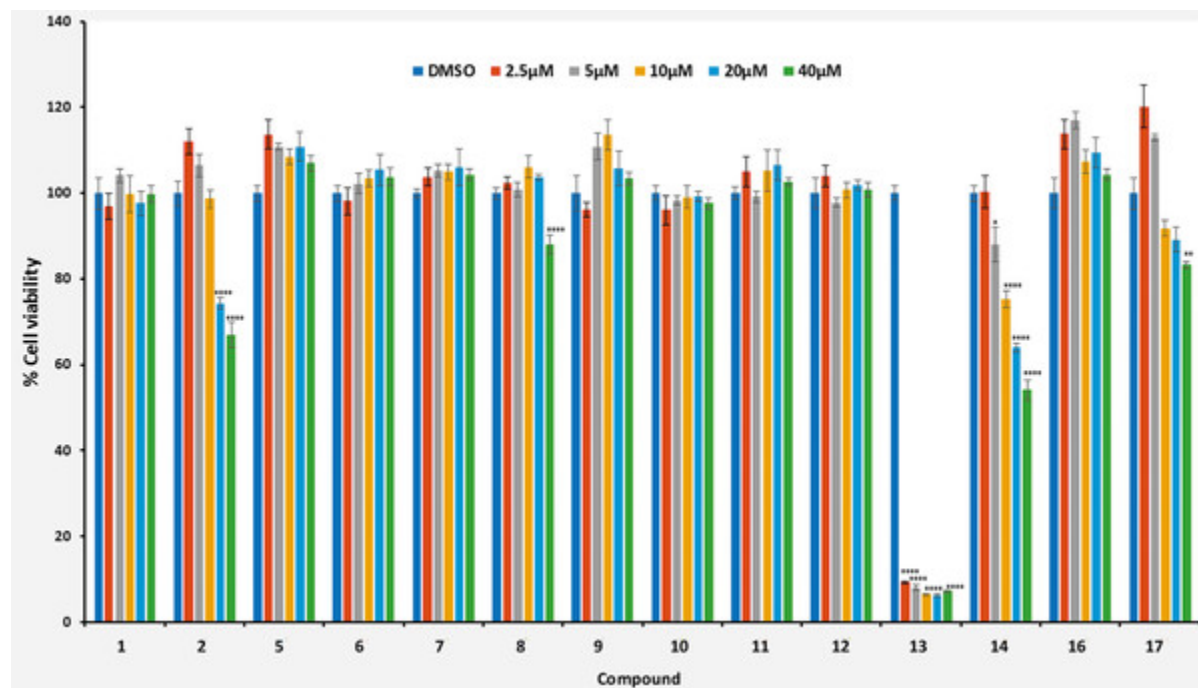


Figure 5. Cytotoxic activity of compounds **1**, **2**, **5–14**, **16**, and **17** against the African American prostate cancer cell line E006AA-hT. The percentage viability was calculated by comparing the absorbance of control and treated cells. Data represent the mean \pm SE of five replicates; * $p < 0.05$, ** $p < 0.01$, **** $p < 0.0001$, significant when compared to control.

The antiproliferative activities of compounds **1**, **2**, **5–14**, **16**, and **17** were evaluated against the African American prostate cancer cell line (E006AA-hT) under hypoxic conditions ([Fig. 5]). Of

these, sporidesmin A (**13**) strongly inhibited the viability of the E006AA-hT cells in a concentration-dependent manner and was effective even at the lowest concentration tested, i.e., 2.5 μM (**Fig. 2S**, Supporting Information). Compounds **2** and 6-ethyl-2,7-dimethoxyjuglone (**14**) also showed moderate growth inhibitory activity, especially at the highest concentration of 40 μM ([Fig. 5]).

In summary, the construction of a peak library derived from freshwater aquatic fungi ([Fig. 1] and **Fig. 58S**, Supporting Information), followed by the chemical study of the active extract obtained from fungal strain G858 (*Delitschia* sp.), led to the identification of eight new α -pyrone derivatives (**1–7**, and **11**), as well as the new 3*S**,4*S**-7-ethyl-4,8-dihydroxy-3,6-dimethoxy-3,4-dihydronaphthalen-1(2*H*)-one (**15**). The cytotoxicity of compounds **1**, **2**, **5–14**, **16**, and **17** was evaluated against an African American prostate cancer cell line (E006AA-hT), where **2**, **13**, and **14** showed activity. These results highlight the potential of fungi from underexplored environments as a source of new chemical entities, contributing to the expansion of the chemical space of natural products.

Materials and Methods

General experimental procedures

Optical rotation and UV experiments were conducted using a Rudolph Research Autopol III polarimeter (Rudolph Research Analytical) and a Varian Cary 100 Bio UV-vis spectrophotometer (Varian Inc.). NMR data were obtained using either a JOEL ECA-500 NMR spectrometer operating at 500 MHz for ^1H and 125 MHz for ^{13}C (JOEL Ltd.) or an Agilent 700 MHz NMR spectrometer equipped with a cryoprobe, operating at 700 MHz for ^1H and 175 MHz for ^{13}C (Agilent Technologies). HRESIMS data were collected with a Thermo QExactive Plus mass spectrometer coupled with an electrospray ionization source (Thermo Fisher Scientific). The purity of the compounds (> 95%) was assessed by ^1H -NMR (see Supporting Information) and using a Waters Acquity UPLC system paired with an Acquity BEH C_{18} column (1.7 μm ; 50 mm \times 2.1 mm) (Waters Corp.) and analyzed using the software XCalibur. Gemini-NX C_{18} analytical (5 μm ; 250 \times 4.6 mm), semipreparative (5 μm ; 250 \times 10.0 mm), preparative (5 μm ; 250 \times 21.2 mm), Luna PFP semipreparative (5 μm ; 250 \times 10.0 mm) and preparative (5 μm ; 250 \times 21.2 mm) columns (all from Phenomenex) along with Waters Atlantis T3 C_{18} semipreparative (5 μm ; 250 \times 10.0 mm) and preparative (5 μm ; 250 \times 19.0 mm) columns were used for chromatographic separations. A Varian ProStar HPLC system equipped with ProStar 210 pumps, a ProStar 335 photodiode array detector, and Galaxie Chromatography Workstation software (version 1.9.3.2, Varian Inc.) was used for data collection and analysis. Flash chromatography was conducted on a Teledyne Isco CombiFlash Rf 200 system using RediSep RF Gold HP silica columns (both from Teledyne Isco) that was coupled with UV and evaporative light-scattering detectors. All other reagents and solvents were obtained from Fisher Scientific.

Fungal strain isolation and identification

Specimen JF16003 was collected in Ariège, Rimont, France on submerged wood in LeBaup stream (September 2015) (**Fig. 58S**, Supporting Information) from which strain G858 was

isolated. Based on micromorphological examination of the fruiting bodies obtained from the incubated wood, the fungus was identified as *Delitschia* sp. (**Fig. 3S**, Supporting Information). This fungus develops on long-time submerged wood as a dense greenish black sterile subiculum, and ascomata occur when the stream dries out or when the wood is washed out on the banks. In addition, the partial large subunit (LSU; 28S) rDNA region was sequenced to corroborate morphological observations, since the 28S ribosomal gene region for most previously described *Delitschia* sp. is available in GenBank for phylogenetic comparison. The partial LSU region was amplified using primer combination LROR and LR6 [34],[35] and Sanger sequenced using primers LROR, LR3, and LR6. For phylogenetic analysis, all available *Delitschia* sp. sequences from GenBank were aligned using a previously published LSU alignment of members of the Pleosporomycetidae, Dothideomycetes [36]; these included reports of *Delitschia* sp. from which fungal metabolites have been isolated [12],[13],[26]. Protocols for DNA extraction, PCR, Sanger sequencing, and phylogenetic analysis have been reviewed [37]. Results of the Maximum Likelihood (ML) analysis showed that strain G858 was nested within the family Delitschiaceae, Pleosporales, Dothideomycetes, Ascomycota with 100% ML bootstrap and was a sister to *Delitschia* sp. (FL1581) [12] with high bootstrap support (**Fig. 3S**, Supporting Information). *Delitschia* sp. have been reported from herbivore dung [12],[38],[39] and submerged wood in freshwater [26],[40]. This study provides molecular evidence that *Delitschia* is not restricted to a coprophilous lifestyle but can also be lignicolous (**Fig. 4S**, Supporting Information). The partial LSU sequences were deposited in GenBank (MF684774, MF684775).

Fermentation, extraction, and isolation

A fresh culture of fungal strain G858 was grown in a petri dish containing potato dextrose agar (PDA; Difco) with an autoclaved piece of balsa wood. Once sufficient mycelial growth was observed on the PDA plates, a small agar plug with fungal mycelium was cut out from the leading edge of the colony and transferred aseptically to eight 10-mL liquid YESD (2% soy peptone, 2% dextrose, and 1% yeast extract) seed cultures. The cultures were shaken at room temperature at 125 rpm for 14 days. Afterwards, the seed cultures were used to inoculate eight 250-mL Erlenmeyer flasks with solid media (rice), prepared by autoclaving 10 g of rice with 20 mL of distilled H₂O; the flasks were incubated at room temperature for 28 days [8],[9].

A total of 60 mL of 1:1 MeOH-CHCl₃ was added to each of the flasks followed by chopping into small pieces with a spatula. The samples were shaken overnight (~ 16 h) using a rotary shaker at ~ 125 rpm at room temperature, filtered under vacuum, and the remaining solid portions were washed with small volumes of MeOH. The filtrates were combined, and 300 mL of both CHCl₃ and H₂O were added. This biphasic solution was stirred for 30 min and transferred to a separatory funnel in which a precipitate was collected prior to the organic layer being drawn off; both the organic phase and the precipitate were evaporated to dryness in vacuum. The organic extract was partitioned between 300 mL of 1:1 MeOH-CH₃CN and 300 mL of hexanes. The MeOH-CH₃CN layer was evaporated under reduced pressure to dryness [8],[21]. The defatted extract (~ 559 mg) was dissolved in CHCl₃, then adsorbed onto Celite 545. The sample was fractionated by flash chromatography via a 24 g RediSep Rf Gold HP silica column using a gradient solvent system of hexane-CHCl₃-MeOH at a 35 mL/min flow rate and 25.7 column volumes over 24.7 min to afford nine fractions (F1–F9). Fraction 3 (63.6 mg) was subjected to

preparative HPLC using a gradient system of 50:50 to 80:20 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 21.20 mL/min and a Gemini column to yield compound **14** (4.6 mg, t_R 18.0 min). Fraction 4 (73.8 mg) was subjected to preparative HPLC using a gradient system of 20:80 to 100:0 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 20.00 mL/min and a Gemini column to yield 10 subfractions (F_{4I}-F_{4X}). Subfractions F_{4I} and F_{4IV} afforded compounds **2** (1.3 mg) and **15** (2.0 mg), respectively. Fraction 5 (9.6 mg) was subjected to semipreparative HPLC using a gradient system of 40:60 to 70:30 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 4.60 mL/min and a Gemini column to yield 6 subfractions, where subfraction 3 yielded compound **13** (2.5 mg, t_R 13.7 min). Fraction 6 (136.7 mg) was subjected to preparative HPLC using a gradient system of 10:90 to 50:50 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 21.20 mL/min and a Gemini column to yield 18 subfractions (F_{6I}-F_{6XVIII}). Subfractions F_{6I}, F_{6IV}, F_{6VI}, F_{6VII}, and F_{6XII} afforded compounds **5** (10.3 mg, t_R 10.1 min), **9** (1.0 mg, t_R 14.9 min), **6** (8.7 mg, t_R 16.0 min), **1** (11.0 mg, t_R 17.2 min), and **12** (0.9 mg, t_R 22.1 min), respectively. Subfraction F_{6V} (17.3 mg) was further purified with preparative HPLC using an isocratic system 25:75 CH₃CN-H₂O (0.1% formic acid) over 20 min at a flow rate of 21.20 mL/min and a Luna PFP column to yield 6 subfractions, in which subfraction 1 afforded compound **10** (0.4 mg, t_R 7.5 min). Subfraction F_{6X} (6.8 mg) was further purified with semipreparative HPLC using a gradient system of 35:65 to 45:55 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 4.60 mL/min and a Gemini column to yield 4 subfractions (F_{6X1}-F_{6X4}), where F_{6X2}, and F_{6X3} afforded compounds **4** (1.6 mg, t_R 9.8 min) and **3** (0.8 mg, t_R 10.3 min), respectively. Fraction 7 (242.0 mg) was subjected to two different purification methods using preparative HPLC. Part of fraction 7 (100.0 mg) was subjected to preparative HPLC using a gradient system of 20:80 to 100:0 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 17.00 mL/min and an Atlantis T3 column to yield 9 subfractions (F_{7A-I}-F_{7A-IX}). Subfractions F_{7A-II} and F_{7A-III} afforded compounds **8** (3.7 mg, t_R 8.0 min) and **7** (9.0 mg, t_R 9.4 min), respectively. Part of fraction 7 (100.0 mg) was subjected to preparative HPLC using a gradient system of 10:90 to 50:50 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 21.20 mL/min and a Luna PFP column to yield 18 subfractions, where subfraction 4 afforded compound **11** (0.9 mg, t_R 10.6 min). The precipitate obtained from the CHCl₃-H₂O partition (80.0 mg) was subjected to preparative HPLC using a gradient system of 60:40 to 90:10 CH₃CN-H₂O (0.1% formic acid) over 30 min at a flow rate of 21.20 mL/min and a Gemini column to yield 8 subfractions. Subfraction 2 yielded compound **17** (1.3 mg), and subfractions 4–6 afforded compound **16** (47.1 mg). Compounds **3**, **7**, **8**, and **15** were further purified by analytical and semipreparative HPLC to generate materials of a higher purity, while all other isolated compounds were at a suitable purity for biological evaluation. Various columns and gradients, including isocratic, were used with a mobile phase of CH₃CN-H₂O (0.1% formic acid) to yield compounds **3** (0.1 mg), **7** (2.3 mg), **8** (1.3 mg), and **15** (0.4 mg).

Delitpyrone A (**1**): Colorless solid; $[\alpha]_D^{27} = -15$ (c 0.1, MeOH); UV (MeOH) λ_{\max} (log ϵ) 209 (4.17), 284 (3.82) nm; ¹H-NMR (CDCl₃, 700 MHz) and ¹³C-NMR (CDCl₃, 175 MHz), see [Tables 1] and [2]; HRESIMS m/z 213.1125 (calcd. for C₁₁H₁₇O₄, 213.1121).

1',2'-Epoxi-delitpyrone A (**2**): Colorless solid; $[\alpha]_D^{26} = -5$ (c 0.04, MeOH); UV (MeOH) λ_{\max} (log ϵ) 208 (3.78), 284 (3.22) nm; ¹H-NMR (CDCl₃, 500 MHz) and ¹³C-NMR

(CDCl₃, 125 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 211.0955 (calcd. for C₁₁H₁₅O₄, 211.0965).

Delitpyrone B (3): Colorless solid; ¹H-NMR (CDCl₃, 700 MHz) and ¹³C-NMR (CDCl₃, 175 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 229.1068 (calcd. for C₁₁H₁₇O₅, 229.1070).

Delitpyrone C (4): Colorless solid; ¹H-NMR (acetone-*d*₆, 500 MHz) and ¹³C-NMR (acetone-*d*₆, 125 MHz), see [Table 3]; HRESIMS *m/z* 211.0962 (calcd. for C₁₁H₁₅O₄, 211.0965).

2'-Oxodelitpyrone A (5): Colorless solid; [α]_D²⁷ = -38 (*c* 0.1, MeOH); UV (MeOH) λ_{\max} (log ϵ) 208 (4.11), 276 (3.74) nm; ¹H-NMR (CDCl₃, 500 MHz) and ¹³C-NMR (CDCl₃, 125 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 227.0916 (calcd. for C₁₁H₁₅O₅, 227.0914).

Delitpyrone D (6): Colorless solid; UV (MeOH) λ_{\max} (log ϵ) 208 (4.30), 283 (3.94) nm; ¹H-NMR (CDCl₃, 500 MHz) and ¹³C-NMR (CDCl₃, 125 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 211.0967 (calcd. for C₁₁H₁₅O₄, 211.0965).

Delitpyrone E (7): Colorless solid; [α]_D²⁷ = +17 (*c* 0.13, MeOH); UV (MeOH) λ_{\max} (log ϵ) 210 (3.92), 285 (3.65) nm; ¹H-NMR and ¹³C-NMR (MeOH-*d*₄, 500 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 213.1122 (calcd. for C₁₁H₁₇O₄, 213.1121).

5-(3-S-Hydroxybutyl)-4-methoxy-6-methyl-2H-pyran-2-one (8): Colorless solid; [α]_D²⁷ = +7 (*c* 0.09, MeOH); UV (MeOH) λ_{\max} (log ϵ) 210 (3.73), 284 (3.41) nm; ¹H-NMR and ¹³C-NMR (MeOH-*d*₄, 500 MHz), see Supporting Information; HRESIMS *m/z* 213.1121 (calcd. for C₁₁H₁₇O₄, 213.1121).

4-Methoxy-6-methyl-5-(3-oxobutyl)-2H-pyran-2-one (9): Colorless solid; UV (MeOH) λ_{\max} (log ϵ) 207 (3.67), 281 (3.12) nm; ¹H-NMR (MeOH-*d*₄, 500 MHz) and ¹³C-NMR (MeOH-*d*₄, 125 MHz), see Supporting Information; HRESIMS *m/z* 211.0968 (calcd. for C₁₁H₁₅O₄, 211.0965).

Pyrenocine I (10): Colorless solid; UV (MeOH) λ_{\max} (log ϵ) 232 (3.75), 284 (3.41) nm; ¹H-NMR and ¹³C-NMR (CDCl₃, 500 MHz), see Supporting Information; HRESIMS *m/z* 211.0965 (calcd. for C₁₁H₁₅O₄, 211.0965).

3'-Hydroxydelitpyrone D (11): Colorless solid; UV (MeOH) λ_{\max} (log ϵ) 211 (3.53), 2863 (3.43) nm; ¹H-NMR and ¹³C-NMR (CDCl₃, 500 MHz), see [Tables 1] and [2]; HRESIMS *m/z* 227.0914 (calcd. for C₁₁H₁₅O₅, 227.0914).

3S,4S*-7-Ethyl-4,8-dihydroxy-3,6-dimethoxy-3,4-dihydronaphthalen-1(2H)-one (15)*: Yellow solid; ¹H-NMR and ¹³C-NMR (CDCl₃, 700 MHz), see [Table 4]; HRESIMS *m/z* 267.1224 (calcd. for C₁₄H₁₉O₅, 267.1227).

Preparation of the (*R*)- and (*S*)-MTPA ester derivatives of compounds 1, 7, and 8

Independently, a total of 0.10 mg of compounds **1**, **7**, and **8** were mixed with 400 μL of pyridine-*d*₅ and transferred into NMR tubes. The reactions were initiated with the addition of 4 μL of *S*-(+)- α -methoxy- α -(trifluoromethyl)phenylacetyl (MTPA) chloride into each NMR tube. The tubes were carefully shaken and then immediately monitored by ¹H-NMR at 5, 10, and 15 min increments, with the reactions being completed by 10 min, yielding the mono (*R*)-MTPA ester derivatives of **1**, **7**, and **8** (**1b**, **7b**, and **8b**, respectively). The (*S*)-MTPA ester derivatives of compounds **1**, **7**, and **8** (**1a**, **7a**, and **8a**, respectively) were prepared in an analogous similar fashion [9].

Cell viability assay

African American cancer cells (E006AA-hT) were obtained from Dr. Koochekpour (Roswell Park Memorial Institute). Cells were treated with **1**, **2**, **5–14**, **16**, and **17** (2.5–40 μM , in DMSO) and cultured under hypoxia (1% O₂) for 48 h. An equal amount of DMSO was present in each treatment, including the control; the concentration of DMSO did not exceed 0.1% (v/v) in any treatment. Cell viability was measured using the MTT reagent, as detailed below. As a positive control, we used the HIF2 α inhibitor TC-S7009, which in our other completed studies inhibited E006AA-hT cell viability with an IC₅₀ value of \sim 20 μM ; HIF2 α inhibitors are important anticancer molecules [41]. TC-S7009 (\geq 99% purity) was procured from Tocris Bioscience (Avonmouth, Bristol).

For the MTT assay, 2000 E006AA-hT cells in 200 μL of media were dispensed in a 96-well plate on day 1, followed by exposure to **1**, **2**, **5–14**, **16**, and **17** at concentrations of 2.5–40 μM on day 2, and then cultured under hypoxic conditions (1% oxygen concentration). After 48 h of hypoxia treatment, 20 μL of MTT (5 mg/mL PBS) were added to each well. The plates were further incubated for 2 h. Following completion of the incubation time, the medium was decanted, the formazan formed was dissolved in 200 μL of DMSO, and, after 10 min, absorbance was measured at 550 and 660 nm on a microplate reader.

Conflict of Interest

The authors declare no conflict of interest.

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