

# Appendix A

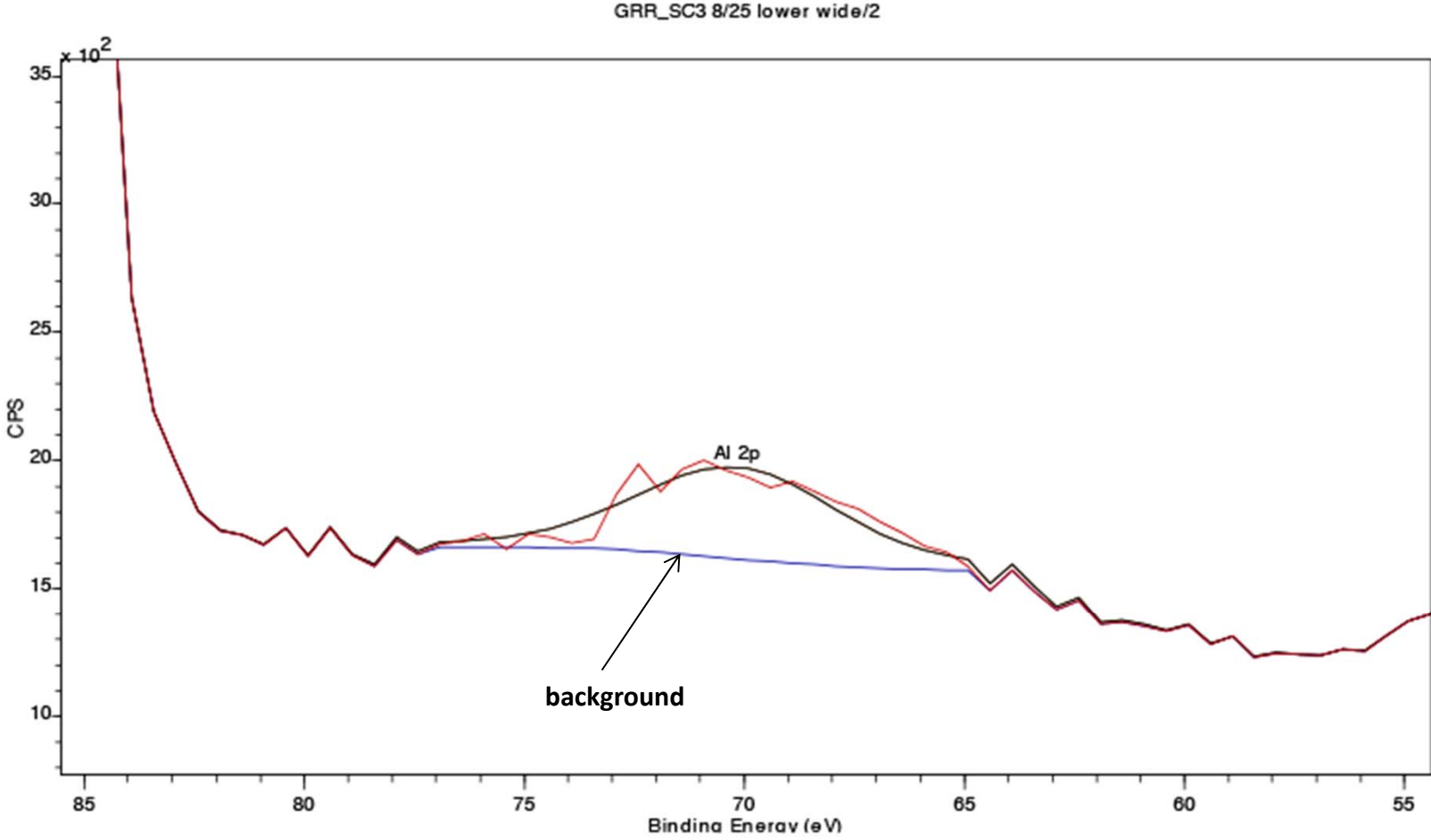
to

**ELECTRON MICROPROBE/SIMS ANALYSES OF Al IN OLIVINE: APPLICATIONS TO SOLAR WIND, PALLASITES, AND TRACE ELEMENT ANALYSES.**

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## XPS Modeling

Typical SC3 olivine Al 2p peak, Mg 2s peak to the left. The small Al 2s peak (not shown) is present which confirms identification. Data red; fit black. All 12 spots analyzed have measurable Al 2p peak with an average about 500 cps. For comparison, single crystal sapphire standard  $2 \times 10^4$  cps.



**Figure A1**

For sapphire standard:

$$\text{cps Al sapphire} = k n A \lambda \sin(\theta) \quad (\text{A1})$$

$k$  = cons;  $n$  = atoms Al/cc in sapphire;  $A$  = X-ray beam area;  $\lambda$  = photoelectron mean free path ( $\approx 16 \text{ \AA}$ );

$\theta$  = photoelectron take off angle (35 degrees).

For monolayer of Al contamination on olivine:

$$\text{cps Al ol} = k N A \quad (\text{S2})$$

$N$  = contamination atoms of Al/  $\text{cm}^2$

$$N = n [ I(\text{ol}) / I(\text{sap}) ] \lambda \sin(\theta) \approx 2e13 \text{ atoms/cm}^2$$

$I$ =intensity

If mean emp emission depth of Al  $K\alpha$  X-ray in olivine is 2 micron,  
for 100 ppm Al , about  $100 \times 10^{13}$  atoms/ $\text{cm}^2$ .

X-rays from surface contamination detected with more efficiency.

So SC3 Al contamination  $\geq 2\%$