

Geochemical processes controlling the groundwater chemistry and fluoride contamination in the aquifer systems on the eastern, western and northern flanks of Mount Meru, Tanzania



Introduction

Arusha volcanic region in northern Tanzania

- **Groundwater** – source of drinking water
 - ❖ High F^- conc.
- Dental and skeletal fluorosis



Location of Mount Meru in Arusha region, Tanzania.
(Source: modified after Wikimedia; commons.wikimedia.org)

Methods

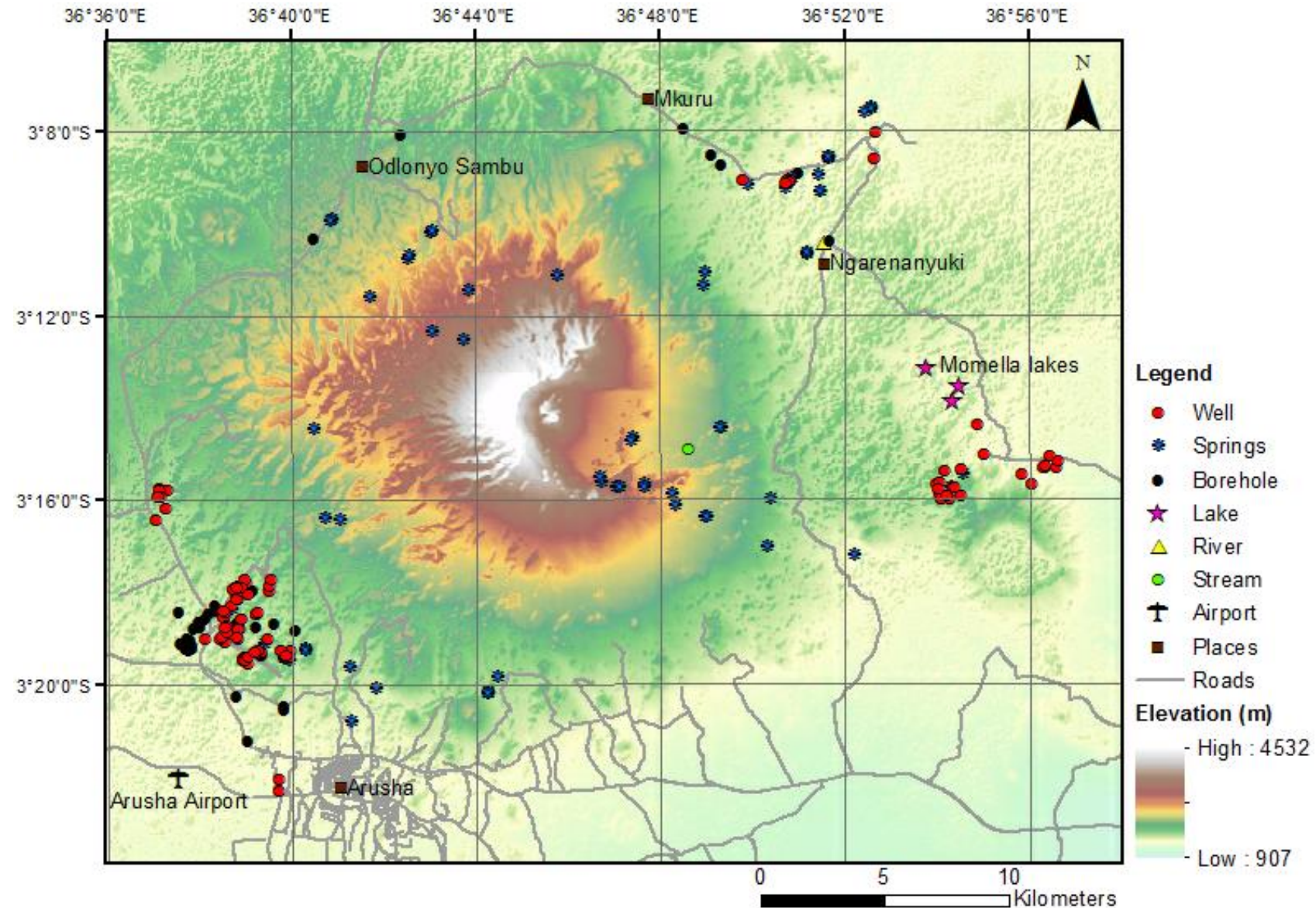
Two field campaigns;

- July – September 2017
- March – September 2018



Methods

181
groundwater points



Methods

158
water samples



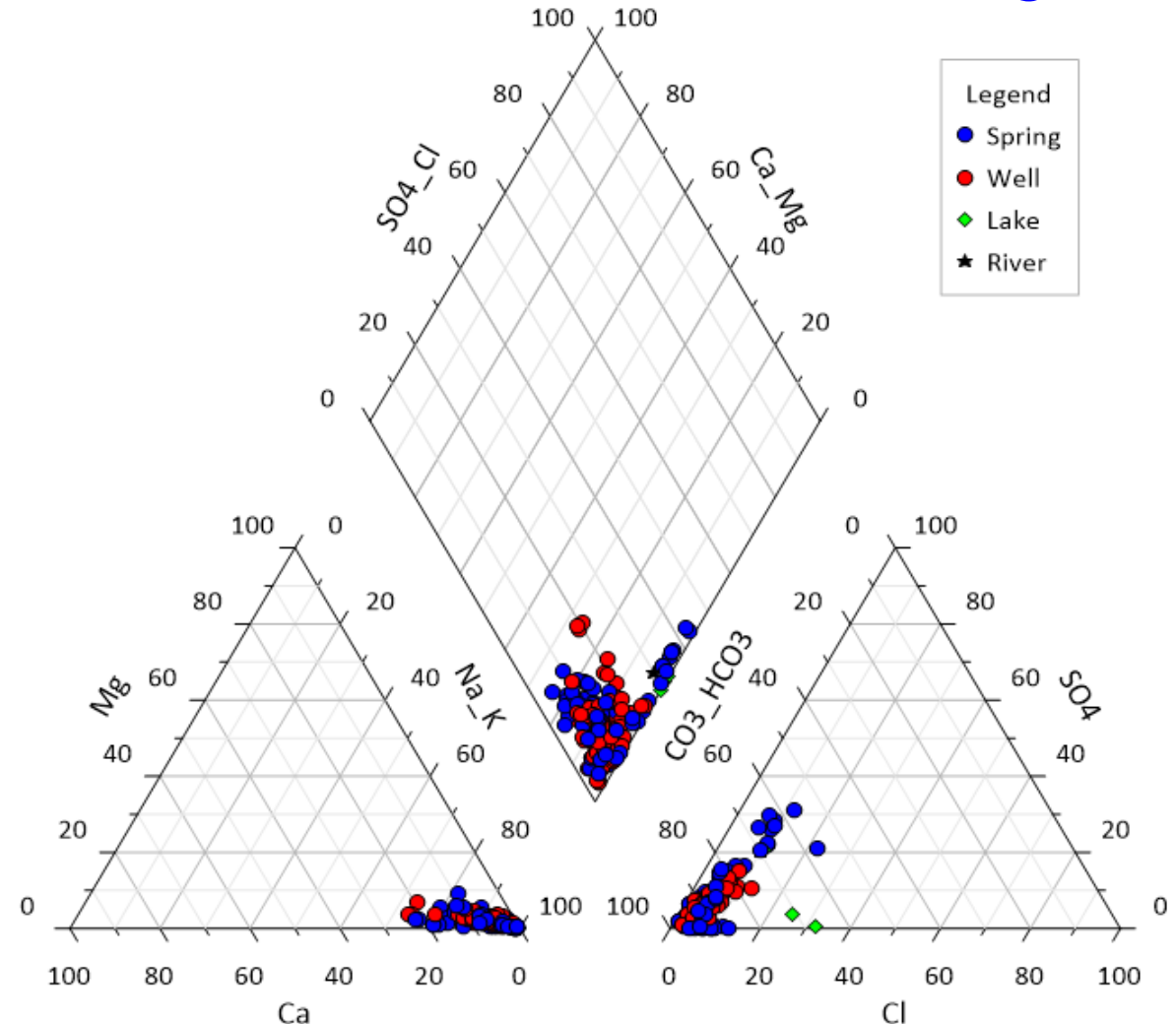
Results and Discussion

Dominant ions: **Sodium (Na^+)** and **bicarbonate (HCO_3^-)**

NaHCO_3 type water

Average pH value = **7.8**

Alkaline groundwater



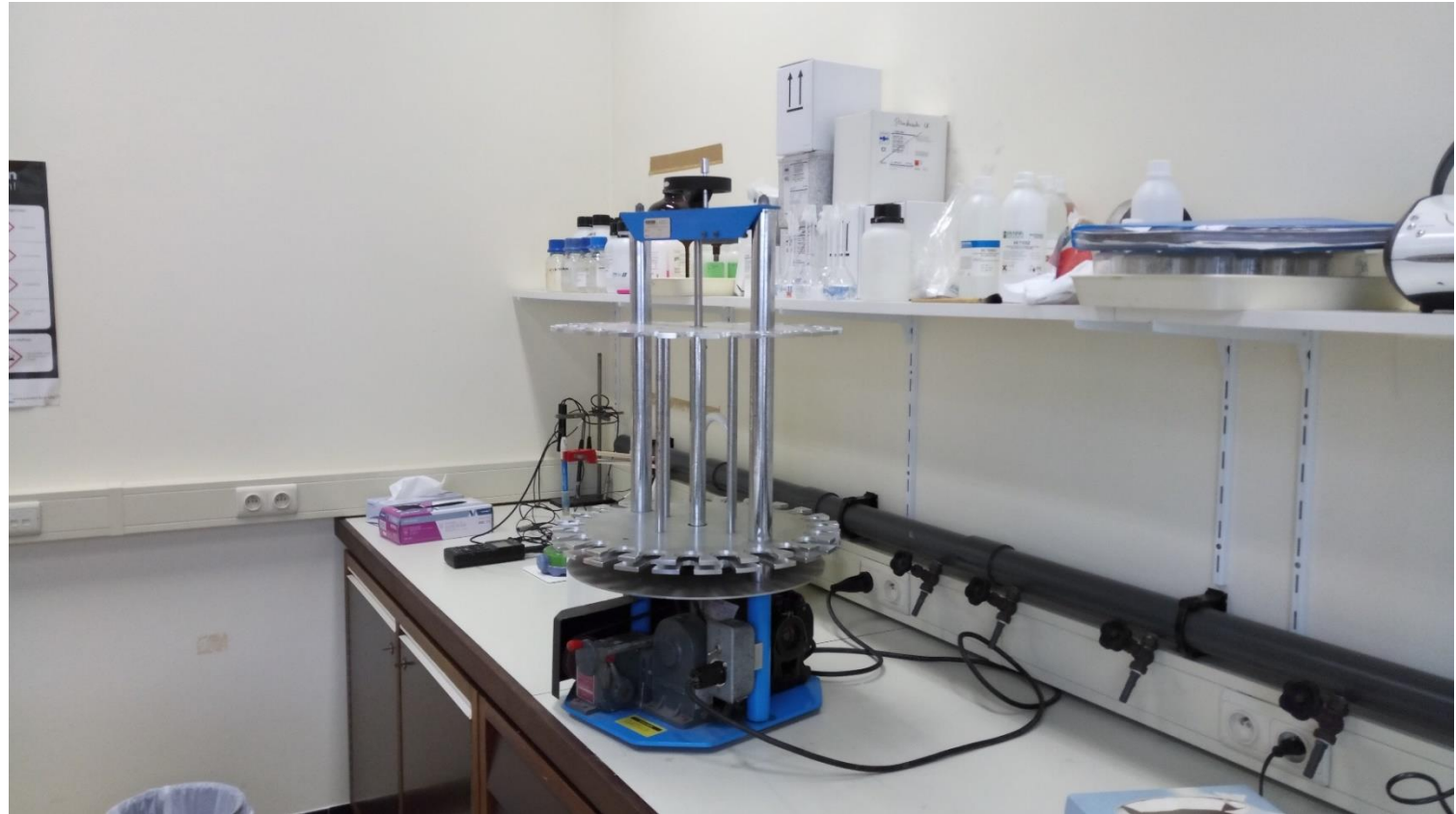
Results and Discussion

High values of F^- were recorded

Range: **0.15 – 301 mg/l**

Average value = **23 mg/l**

Median value = **10 mg/l**



Results and Discussion

F⁻ concentration

- **91%** of samples (**143** samples)
 - above 1.5 mg/l (WHO limit for drinking water)
- **9%** of samples (**15** samples)
 - below 1.5 mg/l
 - ❖ Springs at higher elevations on Mount Meru

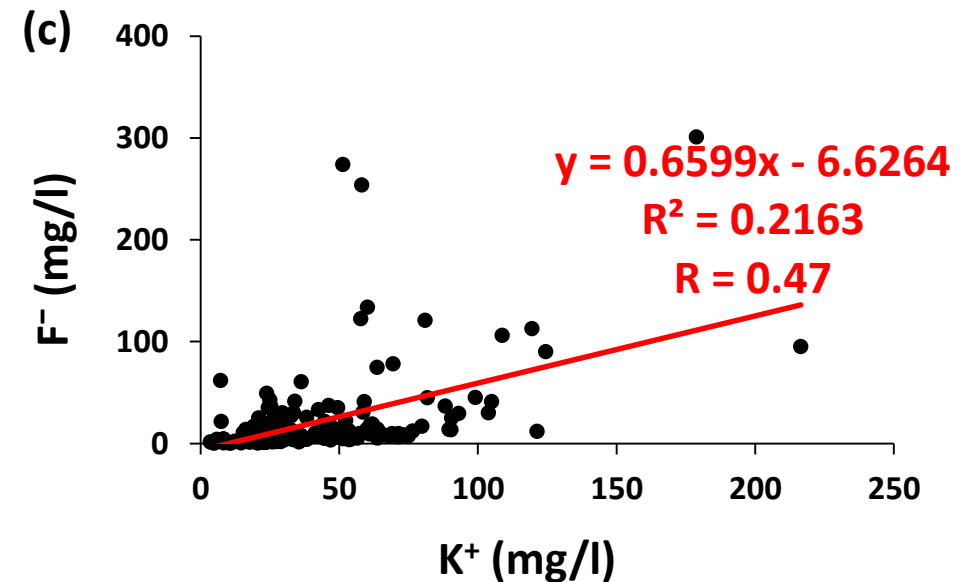
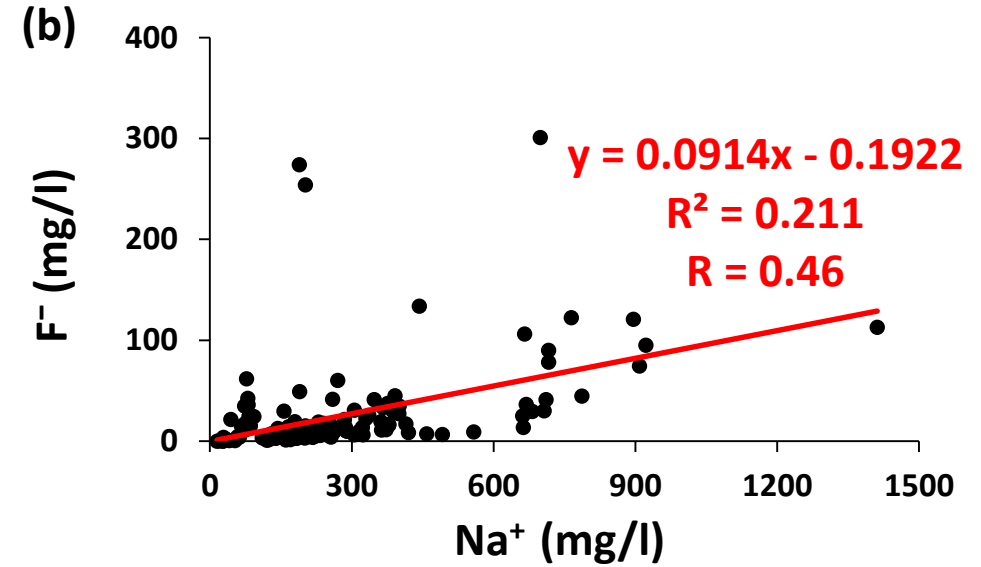
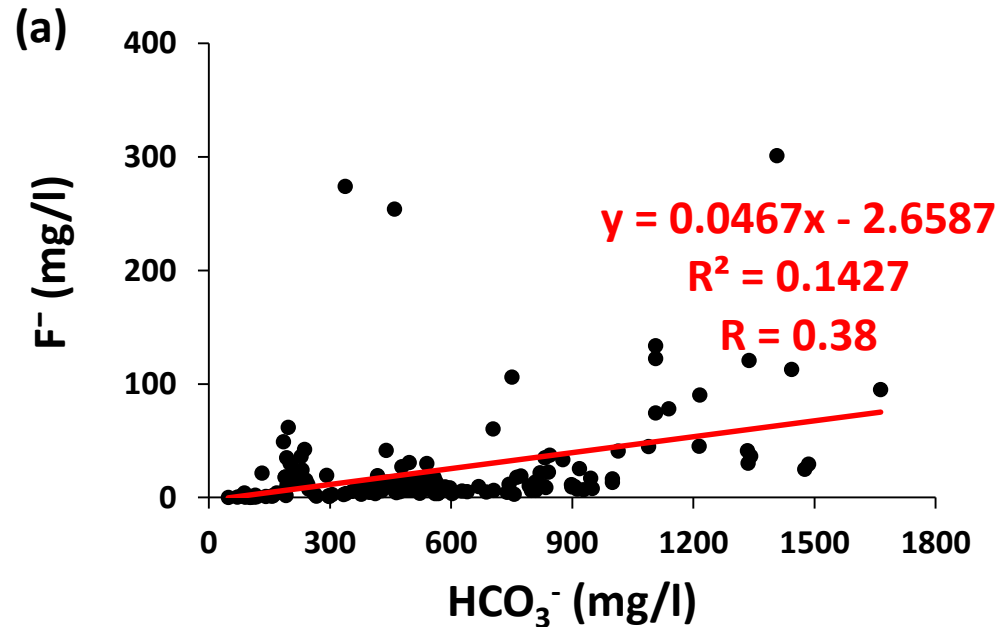
Results and Discussion

F:Cl ratios

- **99%** of samples (**156** samples) – F:Cl > 0.10
 - **1%** of samples (**2** samples) – F:Cl < 0.10
- **F⁻** is derived from **chemical weathering of rocks**

Results and Discussion

Correlation analysis



Results and Discussion

Significant **positive linear correlations** of F^- with HCO_3^- , Na^+ , K^+ and pH

- ***Weathering of silicate minerals***
 - ❖ ***Na-K-rich volcanic rocks***

$\alpha = 0.05$; *r*- correlation coefficients; $n=158$

	<i>r</i>	p-value
HCO_3^-	0.38	9.97E-07
Na^+	0.46	1.27E-09
K^+	0.47	7.45E-10
pH	0.33	2.00E-05

Results and Discussion

Significant **negative linear correlation** of F^- with Ca^{2+}

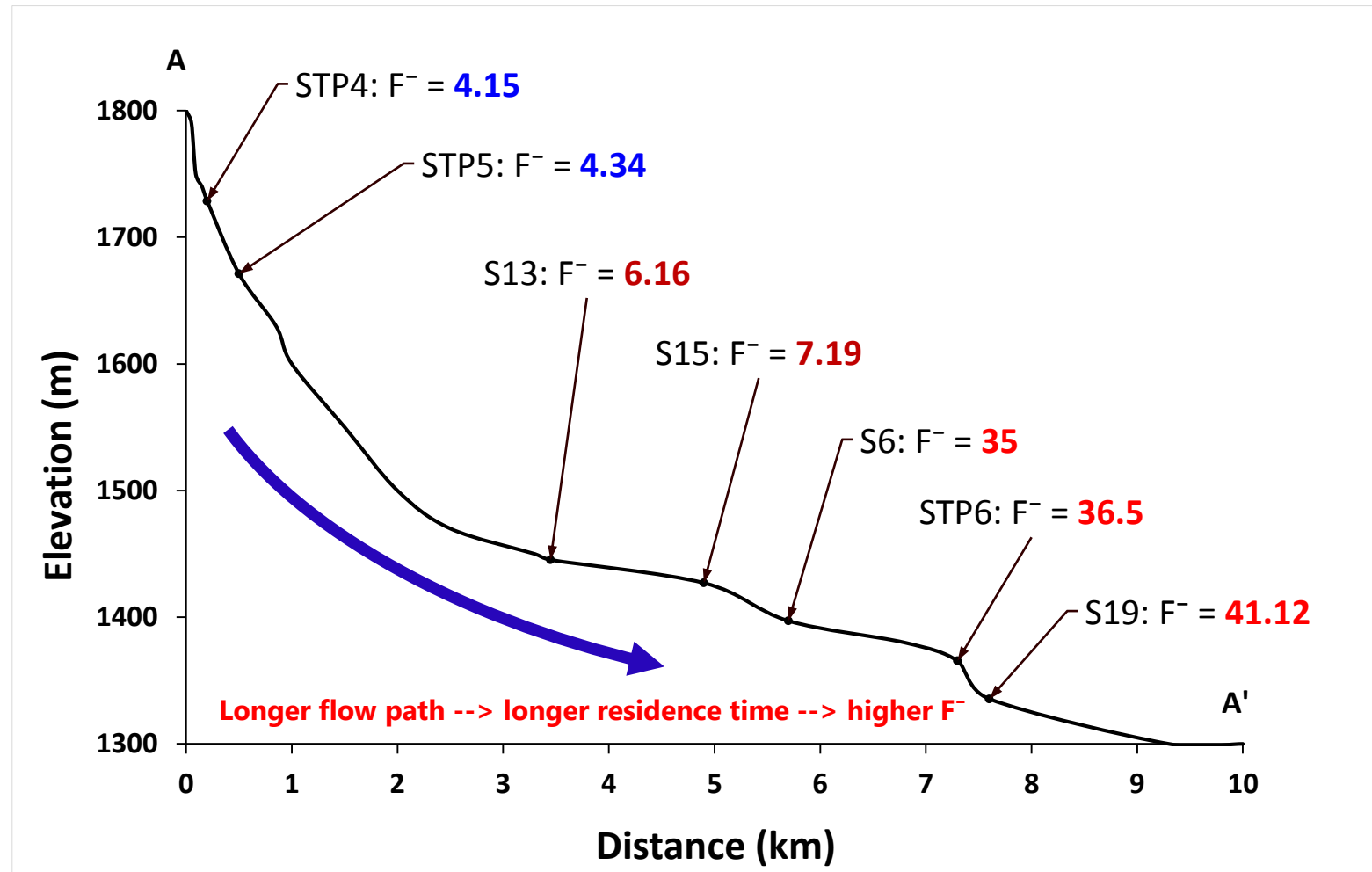
- ***Calcite precipitation***
- ***Dissolution of fluorite (CaF_2)***

$\alpha = 0.05$; r- correlation coefficients; $n=158$

	r	p-value
Ca^{2+}	- 0.21	6.84E-03

Results and Discussion

F^- (mg/l) **increase** with a **decrease** in elevation



Results and Discussion

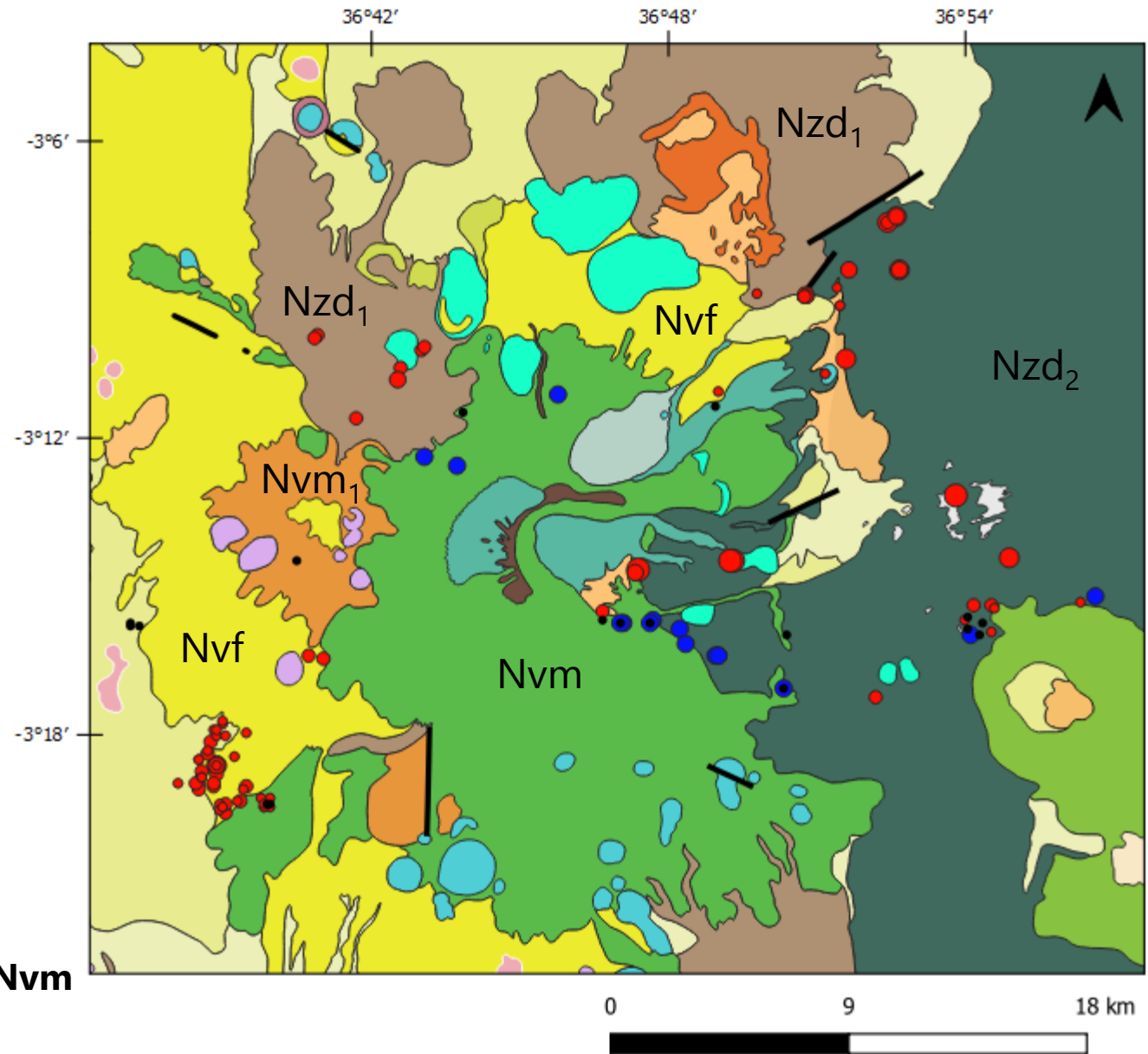
F⁻ conc. vs Geology

F⁻ (mg/l)

- F⁻ ≤ 1.5
- 1.5 < F⁻ ≤ 4.0
- 4.0 < F⁻ ≤ 10
- 10 < F⁻ ≤ 50
- 50 < F⁻ ≤ 100
- 100 < F⁻ ≤ 200
- 200 < F⁻ ≤ 300

Geological formations

- **Debris avalanche deposits** - Nzd₁, Nzd₂
- **Mantling ash** – Nvf
- **Pyroclastics with nephelinitic and phonolitic lavas** – Nvm
- **Nephelinite lavas and breccias** - Nvm₁



Results and Discussion

F⁻ conc. vs Geology

- **Lower F⁻ values**

- ❖ Pyroclastics with nephelinitic and phonolitic lavas
- ❖ Nephelinite lavas and breccias

- **Higher F⁻ values**

- ❖ Debris avalanche deposits
- ❖ Mantling ash deposits



Conclusion

Chemical evolution of groundwater;

- **Weathering and dissolution of silicate minerals**
 - ❖ **Chemical weathering of Na-K-feldspars**
- **Calcite precipitation and dissolution of fluorite (CaF₂)**

Conclusion

Factors controlling F^- concentrations in groundwater;

- **Nature of the geological formations**
- **Long residence time**

Recommendation

For safe drinking water

- Tapping water from the springs with lower F^- values





Thank You

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Ash cone, Mount Meru (Picture by George Bennett)