

## Minireview

# The use of isothermal titration calorimetry to unravel chemotactic signalling mechanisms

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### Summary

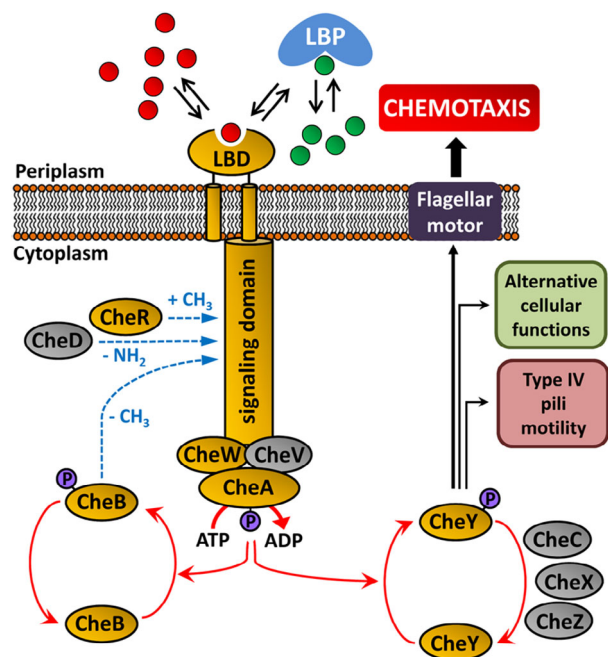
**Chemotaxis is based on the action of chemosensory pathways and is typically initiated by the recognition of chemoeffectors at chemoreceptor ligand-binding domains (LBD). Chemosensory signalling is highly complex; aspect that is not only reflected in the intricate interaction between many signalling proteins but also in the fact that bacteria frequently possess multiple chemosensory pathways and often a large number of chemoreceptors, which are mostly of unknown function. We review here the usefulness of isothermal titration calorimetry (ITC) to study this complexity. ITC is the gold standard for studying binding processes due to its precision and sensitivity, as well as its capability to determine simultaneously the association equilibrium constant, enthalpy change and stoichiometry of binding. There is now evidence that members of all major LBD families can be produced as individual recombinant proteins that maintain their ligand-binding properties. High-throughput screening of these proteins using thermal shift assays offer interesting initial information on chemoreceptor ligands, providing the basis for microcalorimetric analyses and microbiological experimentation. ITC has permitted the identification and characterization of many chemoreceptors with novel specificities. This ITC-based approach can also be used to identify signal molecules that stimulate members of other families of sensor proteins.**

### Introduction

Chemosensory pathways correspond to a major mechanism in bacterial signal transduction. Most chemosensory pathways mediate chemotaxis, whereas others are associated with type IV pili-based motility or the control of second messenger levels such as cAMP or c-di-GMP (Whitchurch *et al.*, 2004; Hickman *et al.*, 2005; Fulcher *et al.*, 2010; Wuichet and Zhulin, 2010). The key element of a chemosensory pathway is the ternary complex formed by chemoreceptors, the CheW coupling protein and the CheA autokinase (Parkinson *et al.*, 2015). Canonical chemoreceptors possess extracytoplasmic ligand-binding domains (LBD) and ligand recognition creates a molecular stimulus that is transduced across the membrane where it modulates CheA autophosphorylation, leading subsequently to changes in the transphosphorylation activity of the CheY response regulator (Fig. 1). The ratio of CheY to CheY-P defines the pathway output and, in the case of pathways that mediate chemotaxis, only CheY-P is able to bind to the flagellar motor altering its activity (Parkinson *et al.*, 2015; Bi and Sourjik, 2018). Chemotaxis pathways are able to sense compound gradients and this capacity is primarily due to temporal adaptation mechanisms that adjust the pathway sensitivity to the present signal concentration. The primary adaptation mechanism consists in the methylation and demethylation of several glutamyl residues at the chemoreceptor signalling domain by the CheR methyltransferase and the CheB methyl-esterase respectively (Parkinson *et al.*, 2015; Bi and Sourjik, 2018) (Fig. 1).

Chemosensory pathways represent complex signal transduction mechanisms and this complexity resides in different levels. First, a significant number of proteins participate in chemosensory signalling. Next to the six core proteins mentioned above, additional proteins were identified that are present in some but not all pathways, as exemplified by the CheD deamidase, the CheW-REC domain fusion protein CheV or the CheC, CheZ and CheX phosphatases (Wuichet and Zhulin, 2010). Furthermore, the scientific community keeps identifying additional proteins that participate in some signalling cascades (Lertsethtakarn *et al.*, 2015; Xu *et al.*, 2016;

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**Fig 1** Schematic representation of a chemosensory pathway. Core proteins, present in almost all chemosensory pathways, are shown in yellow, whereas auxiliary proteins, present in only some pathways, are coloured in grey. Phosphorylation processes are shown by red arrows and methylation, demethylation and deamidation by blue arrows. Signal molecules are shown by red and green spheres. LBD: ligand binding domain; LBP: ligand binding protein. [Color figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)]

Du *et al.*, 2018). The second level of complexity lies in the fact that bacteria frequently have multiple pathways that are composed of homologous signalling proteins. For example, *Myxococcus xanthus* and *Pseudomonas aeruginosa* possess eight and four pathways respectively (Zusman *et al.*, 2007; Ortega *et al.*, 2017b). However, it remains to be established to what degree these pathways are insulated or whether there is cross-talk among them. The third level of intricacy is the elevated number of chemoreceptors encoded in the genomes of many bacteria that, in some cases, can reach 80 (Bardy *et al.*, 2017; Gumerov *et al.*, 2020). The large majority of chemoreceptors are of unknown function and their identification represents a major research need (Martin-Mora *et al.*, 2018b). Furthermore, many chemoreceptor genes are not associated with signalling genes and a current challenge resides in identifying which receptor stimulates which pathway (Briegel *et al.*, 2016; Jones and Armitage, 2017; Ortega *et al.*, 2017b). There is also evidence that LBDs evolve rapidly causing a significant sequence divergence (Gavira *et al.*, 2020), which in turn hampers the functional annotation of chemoreceptors by homology. Therefore, experimental approaches are required for the identification of chemoreceptor ligands and function.

Isothermal titration calorimetry (ITC) is based on the detection of heat changes that arise from the molecular interaction of two molecules. In a typical experiment, a macromolecule solution is titrated with a ligand solution and the resulting thermogram is corrected using the heats that are generated by the injection of ligand into buffer. ITC has become the gold-standard technique for studying binding processes due to its high precision, sensitivity and its capability for the simultaneous determination of the association equilibrium constant as well as the enthalpy change and stoichiometry of binding (Krell, 2008; Ladbury, 2010; Vega *et al.*, 2015). In contrast to other techniques, both ligands are in solution and do not require modification or immobilization. A reliable determination of the  $K_D$  from a single ITC experiment is typically limited to the range 100  $\mu$ M to 1 nM (Krainer and Keller, 2015). ITC is not suited to study ultra-tight binding interactions. A requisite for the study of tight ligand binding is the presence of at least two points at the fast rising part of the titration curve, necessary to define the equilibrium. In case only a single point is observed, it can be concluded that binding occurs with ultra-tight affinity but 'equilibrium constants' provided by the analysis software are not valid. In those cases, the  $K_D$  can be assessed indirectly by competition or displacement assays, provided that a suitable competitive ligand is available, whose  $K_D$  falls within the directly accessible window (Velazquez-Campoy and Freire, 2006; Krainer and Keller, 2015). ITC stands out for its universality since there are almost no limitations as to the nature of ligands that can be analysed or the choice of the analysis buffer and temperature. Particularly over the last 10 years, ITC analyses have permitted to significantly advance our knowledge on chemoreceptors and chemosensory pathways. In this review, we aim at summarizing this work putting particular emphasis on the usefulness of ITC in the identification and characterization of novel chemoreceptors.

## Understanding CheR function

### *Product feedback inhibition and assessment of allostERICITY*

CheR is one of the core signalling proteins of a chemosensory pathway and is present in almost all pathways (Wuichet and Zhulin, 2010). The substrate and product of the CheR-catalysed reaction are S-adenosylmethionine (SAM) and S-adenosylhomocysteine (SAH) respectively. Early studies of *Salmonella typhimurium* CheR have shown that SAH binds tighter than SAM to the same site at CheR (Simms and Subbaramaiah, 1991), indicative of product feedback inhibition. Microcalorimetric studies of the three and, respectively, four CheR homologues of *Pseudomonas putida* KT2440

(Garcia-Fontana *et al.*, 2013) and *P. aeruginosa* PAO1 (Garcia-Fontana *et al.*, 2014) showed that in all cases SAH binding occurred with higher affinity than SAM (Fig. 2), suggesting that product feedback inhibition is a general characteristic of CheR enzymes. However, significant differences existed in the ratios of SAH/SAM binding constants, indicating different magnitudes of SAH-dependent inhibition among CheR homologues (Garcia-Fontana *et al.*, 2013; Garcia-Fontana *et al.*, 2014). So far, it has not been studied whether and to which degree alterations of the cellular SAH/SAM levels interfere with chemosensory signalling. However, data available do not preclude the possibility that the differential product feedback observed for the *Pseudomonas* CheR homologues may have a physiological significance.

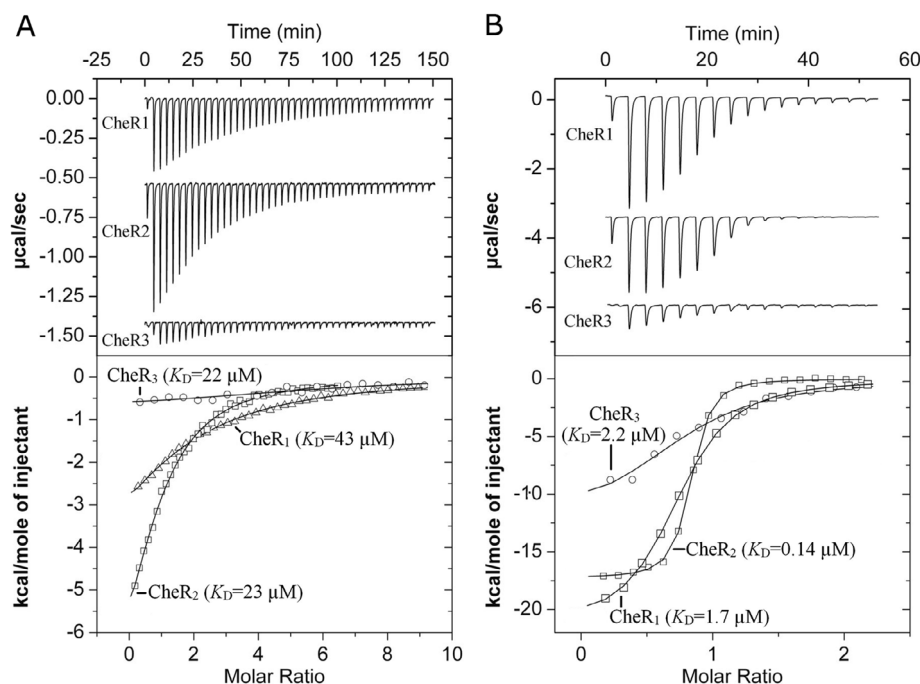
Approximately 10% of chemoreceptors, including the model receptors Tar and Tsr, contain a C-terminal pentapeptide that is tethered to the C-terminus via an unstructured linker (Perez and Stock, 2007; Bartelli and Hazelbauer, 2011). This pentapeptide forms an additional binding site for CheR and CheB (Wu *et al.*, 1996; Barnakov *et al.*, 1999). However, its relevance is poorly understood since the removal of this pentapeptide from Tar resulted in receptor inactivation (Li and Hazelbauer, 2006), whereas many other chemoreceptors do not require this pentapeptide for correct function (Matilla and Krell, 2017). ITC experiments showed that *Escherichia coli* and *S. typhimurium* CheR bound this pentapeptide (NWETF) with  $K_D$  values of approx. 2  $\mu\text{M}$  (Wu *et al.*, 1996) and 10  $\mu\text{M}$  (Yi and Weis, 2002). Similar microcalorimetric titrations using the pentapeptide from the McpB (synonym Aer2) chemoreceptor of

*P. aeruginosa* resulted in a  $K_D$  of 0.5  $\mu\text{M}$  (Garcia-Fontana *et al.*, 2014). Data thus indicate that CheR binds much tighter to the pentapeptide binding site as compared to the binding site covering the methylation sites for which a  $K_D$  of 120–200  $\mu\text{M}$  had been determined (Li and Hazelbauer, 2020).

One possibility to explain the relevance of this additional CheR binding site is a potential allostericity between the pentapeptide and substrate/product binding sites at CheR, a question that had been addressed using ITC. In initial experiments, CheR of *S. typhimurium* (Yi and Weis, 2002) and CheR<sub>2</sub> from *P. aeruginosa* (Garcia-Fontana *et al.*, 2014) were titrated with either pentapeptide, SAM or SAH. These experiments were then repeated in the presence of saturating concentrations of a ligand that binds to the other binding site. Both studies unequivocally concluded that there is no allostericity between both sites, providing further support to the view that pentapeptide binding mainly serves to localize CheR proteins close to the chemoreceptor methylation sites (Wu *et al.*, 1996).

#### Specificity of CheR homologues for chemoreceptors

*Pseudomonas aeruginosa* contains a single chemoreceptor with a C-terminal pentapeptide, McpB (or Aer2), and four CheR homologues (Garcia-Fontana *et al.*, 2014). The question is thus whether a single or multiple CheR homologues interact with this pentapeptide. The microcalorimetric titration of the four purified CheRs showed that exclusively CheR<sub>2</sub>, but not any of the remaining three



**Fig 2** Titration of the three CheR homologues of *P. putida* KT2440 with S-adenosylmethionine (A) and S-adenosylhomocysteine (B). Upper panel: Raw titration data; lower panel: integrated, concentration-normalized and dilution heat corrected titration data. The lines are best fits with a model for the binding of a single ligand to a macromolecule. The corresponding dissociation constants are indicated. Modified version of figure reproduced with permission from Garcia-Fontana *et al.* (2013).

CheR homologues, recognized this pentapeptide (Fig. 3) (Garcia-Fontana *et al.*, 2014). What is thus the sequence or structural property of CheR<sub>2</sub> that can be associated with pentapeptide recognition? An alignment of pentapeptide-dependent and independent CheR sequences revealed the presence of a three amino acid insert that was exclusively present in pentapeptide-dependent enzymes and that was located to the pentapeptide binding site (Perez and Stock, 2007; Garcia-Fontana *et al.*, 2014). Microcalorimetric titrations of a CheR mutant from which this insert had been removed showed an absence of pentapeptide binding and this three amino acid insert can thus be used to predict the capacity of CheR enzymes to bind to pentapeptides (Garcia-Fontana *et al.*, 2014). Further experiments revealed that

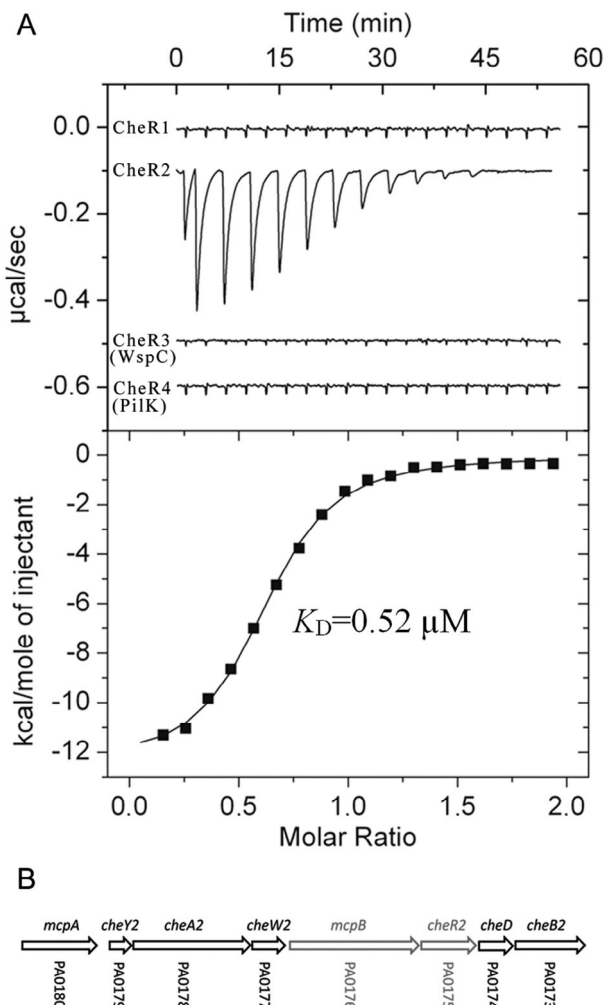
exclusively CheR<sub>2</sub> methylates the pentapeptide-containing receptor McpB and that the removal of the pentapeptide from the receptor abolished CheR<sub>2</sub> binding and methylation (Garcia-Fontana *et al.*, 2014). Microcalorimetry hence played a central role in establishing that the presence of this pentapeptide permits the targeting of a specific receptor by a specific methyltransferase. Interestingly, the presence of terminal pentapeptides is highly conserved in the family of McpB (Aer2) like chemoreceptors (Ortega *et al.*, 2020).

### Characterization of the c-di-GMP-mediated regulatory mechanisms

The c-di-GMP second messenger is a central signal molecule that controls motility and chemotaxis at the transcriptional and post-transcriptional level (Jenal *et al.*, 2017). The precise determination of c-di-GMP dissociation constants provide initial insight at which effector concentration responses are generated.

In *E. coli*, the PilZ domain-containing protein YcgR responds to c-di-GMP by reducing swimming speed and chemotaxis by acting as a flagellar brake through its direct interaction with different components of the flagellar motor (Paul *et al.*, 2010; Hou *et al.*, 2020). ITC studies showed that YcgR binds c-di-GMP with a  $K_D$  of 141 nM (Hou *et al.*, 2020) and fluorescence resonance transfer (FRET) assays revealed a strong interaction between YcgR and the flagellar stator protein MotA only in the presence of c-di-GMP (Boehm *et al.*, 2010). A recent high-throughput screening in *E. coli* identified the deacetylase CobB as a novel c-di-GMP binding protein. CobB was shown to modulate chemotaxis by deacetylating the response regulator CheY (Li *et al.*, 2010) and further studies demonstrated that c-di-GMP binding to CobB inhibits its deacetylation activity of the protein (Xu *et al.*, 2019). ITC studies showed that CobB binds this second messenger with much lower affinity ( $K_D$  of 4.7  $\mu$ M) (Xu *et al.*, 2019), indicating that its regulatory effects sets in at comparatively higher c-di-GMP levels.

In *P. aeruginosa*, the c-di-GMP responsive transcriptional regulator FleQ plays a key role in this regulatory mechanism and an initial measurements using labelled c-di-GMP resulted in an apparent  $K_D$  for FleQ binding in the range of 10–25  $\mu$ M (Hickman and Harwood, 2008). However, microcalorimetric titrations have permitted to precisely determine the  $K_D$  to be 7.2  $\mu$ M (Baraquet and Harwood, 2013). MapZ is another c-di-GMP responsive protein of *P. aeruginosa* that regulates chemotaxis at the post-transcriptional level. In its c-di-GMP bound form, it binds to CheR<sub>1</sub> inhibiting chemoreceptor methylation (Xu *et al.*, 2016). Microcalorimetric titrations determined the  $K_D$  of the c-di-GMP interaction with MapZ to be 8.8  $\mu$ M (Xu *et al.*, 2016). The proximities of the



**Fig 3** Specificity of recognition of CheR<sub>2</sub> by McpB. A. Microcalorimetric titration of the four CheR homologues from *P. aeruginosa* PAO1 with the C-terminal pentapeptide (GWEEF) from the McpB chemoreceptor. B. Organization of the *che2* gene cluster in *P. aeruginosa* PAO1. The genes concealing the proteins that specifically interact are shown in grey. Reproduced with permission from Garcia-Fontana *et al.* (2014).

dissociation constants for binding of c-di-GMP to FleQ and MapZ thus suggest that the cellular onset of c-di-GMP-mediated regulatory processes at the transcriptional and post-transcriptional level occur almost simultaneously. Furthermore, a  $K_D$  of c-di-GMP saturated MapZ for CheR<sub>1</sub> of 29 nM was reported by ITC (Yan *et al.*, 2018), indicating that c-di-GMP recognition by MapZ is the rate-limiting step in this regulatory process.

## Chemoreceptors

### *Ligand-binding studies to full-length receptors*

Very interesting insight was obtained from microcalorimetric titrations of full-length Tsr, either solubilized in detergent micelles or inserted into the membrane, with its ligand serine (Lin *et al.*, 1994). An  $n$ -value of 0.5 was obtained for the detergent solubilized sample, which is consistent with the binding of one ligand to the receptor dimer. In the membrane inserted Tsr, a condition under which chemoreceptors pack into arrays (Briegel *et al.*, 2012), this value was lower indicating that not all binding sites are accessible; probably due to the steric hindrance imposed by the proximity of receptors in the array. The authors determined a  $K_D$  of approximately 25  $\mu$ M for serine binding to the full-length receptor (Lin *et al.*, 1994) that was similar to the value of 35  $\mu$ M as reported by microcalorimetric titration of the individual Tsr LBD (Tajima *et al.*, 2011). These results thus demonstrated that the molecular features for ligand recognition are present in the individual LBD, validating in turn approaches using the individual domains that are discussed below.

### *Ligand-binding studies to individual ligand-binding domains*

The majority of chemoreceptors are of unknown function and their identification forms the basis for understanding the forces that have led to the evolution of chemoreceptors in a particular ecological habitat. The purification of significant amounts of full-length receptors is still a labour-intensive undertaking since it requires detergent-mediated protein extraction from the membrane. Data primarily available for Tar and Tsr show that the binding affinities of ligands to the full-length protein (Clarke and Koshland, 1979; Lin *et al.*, 1994; Bjorkman *et al.*, 2001) and to the individual LBD (Milligan and Koshland, 1993; Tajima *et al.*, 2011) are comparable. There is now evidence that LBDs of all major families can be produced as individual, soluble and folded protein, and Table 1 lists the articles that report microcalorimetric titrations of recombinant chemoreceptor LBDs with chemoeffectors. The individual LBDs can generally be produced in significant amounts in a one-step purification process using

affinity chromatography. In initial studies, the microcalorimetric titration of individual LBDs with ligands was used to characterize signal binding to chemoreceptors that were identified by a phenotypic analysis of mutants (Glekas *et al.*, 2010; Lacal *et al.*, 2010). However, the identification of chemoreceptor function by screening chemoreceptor mutants is not always successful, which is due to the fact that many bacteria possess multiple chemoreceptors with overlapping ligand specificities, to the masking of chemotaxis by energy tactic mechanisms or as a consequence of the insufficient receptor expression under the experimental conditions used (Alvarez-Ortega and Harwood, 2007; Ni *et al.*, 2013; Parales *et al.*, 2013; Ni *et al.*, 2015).

*Identification of novel chemoreceptors by the combined use of thermal shift assays and ITC.* A powerful approach that has accelerated the identification of novel chemoreceptors consists in the combined use of thermal shift assays (TSA) and isothermal titration calorimetry (McKellar *et al.*, 2015; Fernández *et al.*, 2016; Martin-Mora *et al.*, 2018b). Ligand binding to proteins does typically increase the  $T_m$  value or the midpoint of protein unfolding transition (Chiu and Prenner, 2011). The  $T_m$  value corresponds approximately to the temperature at which half the protein is in its native conformation, whereas the remaining half has undergone thermal unfolding. Commercially available ligand collections can thus be screened for compounds that increase the  $T_m$  of the individual LBDs. This assay can be conducted in high-throughput format using 96-well plates. Further technical detail on this assay can be found in the study by Ehrhardt *et al.* (2018) and Fernandez *et al.* (2018). Typically,  $T_m$  shifts superior to 2°C are considered significant and ITC studies are conducted to confirm ligand binding as well as to determine the thermodynamic parameters of the interaction. This combined TSA-ITC approach has resulted in the identification and characterization of a number of receptors with novel ligand specificities of which some are listed below.

- *Polyamines and histamine.* TSA-based ligand screening has led to the identification of two homologous receptors, TlpQ and McpU, in *P. aeruginosa* PAO1 and *P. putida* KT2440, respectively, that recognize the polyamines putrescine, cadaverine, spermidine, agmatine, ethylenediamine as well as histamine (Corral-Lugo *et al.*, 2016; Corral-Lugo *et al.*, 2018). The LBDs of both receptors share 62% sequence identity and, although both receptors possess the same ligand spectrum, ligands bind much tighter to TlpQ-LBD. The latter domain is so far the only LBD that binds five different ligands with nanomolar affinities and the  $K_D$  of 56 nM for spermidine is the highest affinity for

**Table 1** Dissociation constants for the binding of chemoeffectors to recombinant ligand binding domains of chemoreceptors as determined by ITC.

Receptor	LBD type	Species	Chemoeffector	$K_D$ ( $\mu$ M) range	References
McpH	dCACHE	<i>Pseudomonas putida</i> KT2440	Metabolizable purines	1.3–4.3	Fernandez <i>et al.</i> (2016)
McpG	dCACHE	<i>P. putida</i> KT2440	$\gamma$ -aminobutyrate (GABA)	0.175	Reyes-Darias <i>et al.</i> (2015b)
McpU	dCACHE	<i>P. putida</i> KT2440	Polyamines, histamine	0.48–39	Corral-Lugo <i>et al.</i> (2016) and Corral-Lugo <i>et al.</i> (2018)
McpA	dCACHE	<i>P. putida</i> KT2440	12 l-amino acids	0.6–373	Corral-Lugo <i>et al.</i> (2016)
TipQ	dCACHE	<i>P. aeruginosa</i> PAO1	Polyamines, histamine	0.056–1.7	Corral-Lugo <i>et al.</i> (2018)
PctA	dCACHE	<i>P. aeruginosa</i> PAO1	17 l-amino acids	0.28–116	Rico-Jimenez <i>et al.</i> (2013)
PctB	dCACHE	<i>P. aeruginosa</i> PAO1	5 l-amino acids	1.2–1096	Rico-Jimenez <i>et al.</i> (2013)
PctC	dCACHE	<i>P. aeruginosa</i> PAO1	GABA, 2 l-amino acids	1.2–80	Rico-Jimenez <i>et al.</i> (2013)
CtaA	dCACHE	<i>P. fluorescens</i> Pf0-1	7 l-amino acids	4.7–447	(Ud-Din <i>et al.</i> , 2020)
McpX	dCACHE	<i>Sinorhizobium meliloti</i> RU11/001	Quaternary ammonium compounds, L-Pro	0.138–2300	Webb <i>et al.</i> (2017b)
McpU	dCACHE	<i>S. meliloti</i> RU11/001	L-Pro <sup>aa</sup>	104	(Webb <i>et al.</i> , 2014)
McpC	dCACHE	<i>Bacillus subtilis</i> OI1085	12 l-amino acids	14–3700	Glekas <i>et al.</i> (2012)
McpB	dCACHE	<i>B. subtilis</i> OI1085	L-Asn <sup>bb</sup>	14	Glekas <i>et al.</i> (2010)
Mlp24	dCACHE	<i>Vibrio cholerae</i> O395N1	10 l-amino acids, calcium ions <sup>cc</sup>	4.7–452.5	Nishiyama <i>et al.</i> (2012) and Takahashi <i>et al.</i> (2019)
Mlp37	dCACHE	<i>V. cholerae</i> O395N1	Taurine, 3 l-amino acids <sup>dd</sup>	2.7–5.6	Nishiyama <i>et al.</i> (2016))
Psa-PscA	dCACHE	<i>P. syringae</i> pv. <i>actinidiae</i>	D/L-Asp, L-Glu	2.3–27	McKellar <i>et al.</i> (2015))
Psto-PscA	dCACHE	<i>P. syringae</i> pv. <i>tomato</i>	D/L-Asp, L-Glu	1.2–3.4	Cerna-Vargas <i>et al.</i> (2019)
McpA	dCACHE	<i>B. amyloliquefaciens</i> SQR9	Asp, citric acid <sup>ee</sup>	0.24–0.39	Feng <i>et al.</i> (2018)
McpB	dCACHE	<i>B. amyloliquefaciens</i> SQR9	Sodium decanoate <sup>ee</sup>	3.4	Feng <i>et al.</i> (2018)
McpC	dCACHE	<i>B. amyloliquefaciens</i> SQR9	Amino acids <sup>ee</sup>	3.62–3.64	Feng <i>et al.</i> (2018)
TipB	dCACHE	<i>B. amyloliquefaciens</i> SQR9	Phe, pentadecanoic acid <sup>ee</sup>	3.0–3.2	Feng <i>et al.</i> (2018)
Tip3	dCACHE	<i>Campylobacter jejuni</i> NCTC 11168	L-Ile <sup>ff</sup>	86	Liu <i>et al.</i> (2015)
TipC	dCACHE	<i>Helicobacter pylori</i> PMSS1	L-Lactate <sup>gg</sup>	155	Machuca <i>et al.</i> (2017)
McpP	sCACHE	<i>P. putida</i> KT2440	C2/C3-carboxylic acids	34–107	Garcia <i>et al.</i> (2015)
PA2652	sCACHE	<i>P. aeruginosa</i> PAO1	C4-dicarboxylic acids	23–1240	Martin-Mora <i>et al.</i> (2018a)
PscD	sCACHE	<i>P. syringae</i> pv. <i>actinidiae</i>	C2/C3-carboxylic acids	23–356	Brewster <i>et al.</i> (2016)
McpV	sCACHE	<i>S. meliloti</i> RU11/001	C2/C3/C4-carboxylic acids	3.4–280	Compton <i>et al.</i> (2018)
Tar	4HB	<i>Escherichia coli</i> K12	Asp and derivatives	0.54–268	Tajima <i>et al.</i> (2011) and Bi <i>et al.</i> (2013)
Tsr	4HB	<i>E. coli</i> K12	L-Ser	36	Takahashi <i>et al.</i> (2019)
PcaY_PP	4HB	<i>P. putida</i> KT2440	Cyclic carboxylic acids	3.7–138	Fernandez <i>et al.</i> (2017)
CtpH	4HB	<i>P. aeruginosa</i> PAO1	Pi, compounds with pyrophosphate moiety	22–103	Rico-Jimenez <i>et al.</i> (2016)
McpB	4HB	<i>Ralstonia pseudosolanacearum</i> Ps29	Boric acid	5.4	Hida <i>et al.</i> (2017)
MCP2201	4HB	<i>Comamonas testosteroni</i> CNB-1	TCA cycle intermediates	18–378	Ni <i>et al.</i> (2013)
MCP2983	4HB	<i>C. testosteroni</i> CNB-1	Cis-aconitate	6.6 and 152.7 <sup>hh</sup>	Ni <i>et al.</i> (2015)
MCP2901	4HB	<i>C. testosteroni</i> CNB-1	Citrate, aromatic hydrocarbons	19.6–1300	Huang <i>et al.</i> (2016)
McpR	not annotated	<i>B. amyloliquefaciens</i> SQR9	Arg	8.4 <sup>9g</sup>	Feng <i>et al.</i> (2018)
McpS	HBM	<i>P. putida</i> KT2440	TCA cycle intermediates, butyrate, acetate	8.5–574	Lacal <i>et al.</i> (2010), Lacal <i>et al.</i> (2011), and Pineda-Molina <i>et al.</i> (2012)
McpQ	HBM	<i>P. putida</i> KT2440	Citrate, citrate/metal ion complexes	14–39	Martin-Mora <i>et al.</i> (2016a)
McpK	HBM	<i>P. aeruginosa</i> PAO1	$\alpha$ -Ketoglutarate	301 and 81 <sup>hh</sup>	Martin-Mora <i>et al.</i> (2016b)
McpN	PilJ	<i>P. aeruginosa</i> PAO1	Nitrate	47	Martin-Mora <i>et al.</i> (2019)

<sup>a</sup>Fluorescence spectroscopy (Webb *et al.*, 2017a) revealed binding of different l-amino acids.

<sup>b</sup>McpB also mediates chemotaxis to Asp, Glu and His (Hanlon and Ordal, 1994), but ITC assays have not been conducted.

<sup>c</sup>Calcium ion binding enhances affinity for amino acids.

<sup>d</sup>Mlp37 also mediates taxis toward Cys, Asp, Thr, Lys, Val and Gly (Nishiyama *et al.*, 2016), but ITC assays have not been conducted.

<sup>e</sup>McpA, McpB, McpC and TipB mediate chemotaxis to a wide range of compounds (Feng *et al.*, 2018) and only some were used for ITC.

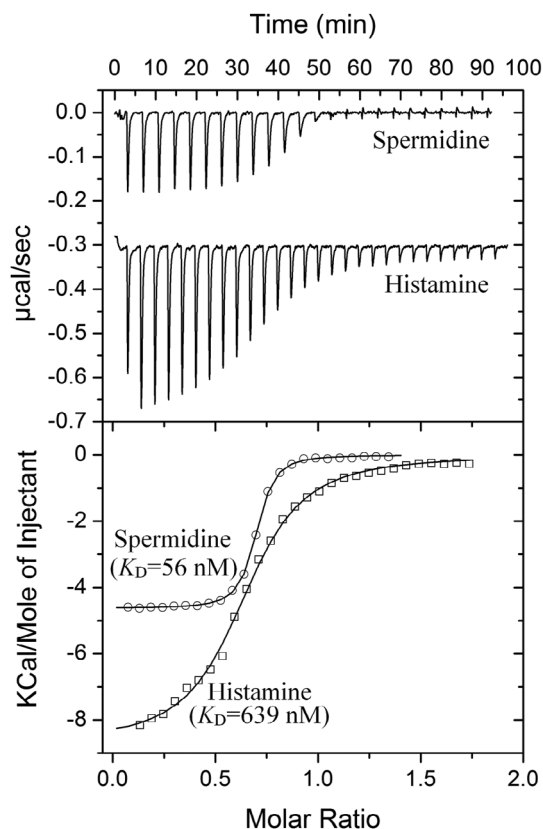
<sup>f</sup>Binding of nine additional compounds to Tip3 was confirmed by surface plasmon and nuclear magnetic resonance (Rahman *et al.*, 2014).

<sup>g</sup>Ligand range of the chemoreceptor has not been explored.

<sup>h</sup>Binding with cooperativity, given are  $K_{D1}$  and  $K_{D2}$ , respectively.

a chemoreceptor ligand so far measured (Fig. 4). Notably, in agreement with the high affinity for some of the TipQ ligands, the onset of chemotactic response was with 500 nM very low (Corral-Lugo *et al.*, 2018).

• **Nitrate.** The identification of the first chemoreceptor for nitrate illustrates well the usefulness of protein-based screening in the functional characterization of chemoreceptors. TSA and ITC revealed that McpN-LBD of



**Fig 4** Microcalorimetric titration of the LBD of the *P. aeruginosa* PAO1 TlpQ chemoreceptor with histamine and spermidine. Data were published by Corral-Lugo *et al.* (2018).

*P. aeruginosa* PAO1 binds specifically nitrate (Martin-Mora *et al.*, 2019), but chemotaxis experiments conducted under standard conditions did not provide any evidence for nitrate chemotaxis of the wild-type strain. However, the knowledge that there is a chemoreceptor that specifically binds nitrate and that was predicted to feed into the chemotaxis pathway (Ortega *et al.*, 2017b) encouraged experimentation to identify the growth conditions under which nitrate chemotaxis may be observed. This research revealed that nitrate taxis in PAO1 is only observed under nitrate starvation conditions, which is likely to be due to the fact that the presence of nitrate largely reduced *mcpN* transcript levels (Martin-Mora *et al.*, 2019).

- **Purines.** TSAs with the LBD of the *P. putida* KT2440 chemoreceptor McpH revealed  $T_m$  increases for adenine, guanine, xanthine and uric acid, which are all purine derivatives. In addition, subsequent ITC screening revealed binding to purine and the purine derivative hypoxanthine (Fernandez *et al.*, 2016). Remarkably, although these ligands differ in structure, their affinities for McpH-LBD were highly similar ( $K_D = 1.3\text{--}4.3\ \mu\text{M}$ ), indicative of a

plasticity in molecular recognition. These compounds form part of the purine degradation pathway that ultimately leads to urea, which serves as sole nitrogen source for KT2440 growth (Fernandez *et al.*, 2016). In contrast, naturally occurring purine derivatives that are of no apparent metabolic value, namely caffeine, theophylline and theobromine, showed no binding to McpH-LBD, suggesting that the metabolic value was a force that has shaped the evolution of this receptor (Fernandez *et al.*, 2016).

- **$\alpha$ -Ketoglutarate.** A number of compounds increased the  $T_m$  of McpK-LBD of *P. aeruginosa* PAO1, however, microcalorimetric studies showed that only  $\alpha$ -ketoglutarate ( $\alpha$ KG) showed binding (Martin-Mora *et al.*, 2016b). Interestingly, the binding curve was biphasic and data analysis revealed binding with positive cooperativity, that contrasts with the negative cooperativity with which aspartate is recognized by the Tar chemoreceptor (Milligan and Koshland, 1993). Further ITC experiments using 15 compounds that are structurally related to  $\alpha$ KG resulted in an absence of binding, indicating that McpK is a  $\alpha$ KG specific receptor (Martin-Mora *et al.*, 2016b). The existence of a  $\alpha$ KG-specific receptor may be related to the central metabolic role of this compound. Apart from being part of the tricarboxylic acid (TCA) cycle, it is also a branch point from which other pathways depart such as the GABA shunt or routes for the synthesis of purine, pyrimidines, coenzyme B or several amino acids.

- **D/L-Asp and L-Glu.** Chemoreceptors with a dCACHE LBD appear to be the primary family for chemotaxis to amino acids (Reyes-Darias *et al.*, 2015a). Characterized family members comprise, for example, broad range amino acid receptors in *Bacillus subtilis* (Glekas *et al.*, 2012), *Sinorhizobium (Ensifer) meliloti* (Webb *et al.*, 2017a), *P. putida* (Corral-Lugo *et al.*, 2016) and *P. aeruginosa* (Taguchi *et al.*, 1997; Rico-Jimenez *et al.*, 2013) or more specific receptors with preference for Gln (Rico-Jimenez *et al.*, 2013), Asn (Glekas *et al.*, 2010) or Arg (Feng *et al.*, 2018). TSA and ITC were central to characterize the PscA chemoreceptor of different pathovars of the plant pathogen *P. syringae*. PscA-LBD specifically binds D/L-Asp and L-Glu (McKellar *et al.*, 2015; Cerna-Vargas *et al.*, 2019) with affinities ranging between 1.2 and 3.4  $\mu\text{M}$  (Cerna-Vargas *et al.*, 2019). Interestingly, Asp and Glu are the most abundant proteinogenic amino acids in plants (Kumar *et al.*, 2017), a fact that may explain the evolution of an Asp- and Glu-specific chemoreceptor in plant pathogens. Chemotaxis to compounds released from the plant stomata is thought to facilitate bacterial plant entry (Matilla and Krell, 2018) and the presence of D-Asp, an enantiomer that cannot be

metabolized by *P. syringae*, was found to reduce virulence in this phytopathogen (Cerna-Vargas *et al.*, 2019).

- **Short-chain carboxylic acids.** The discovery of the homologous chemoreceptors for short-chain carboxylic acids in *P. putida*, *P. syringae* and *S. meliloti* is largely due to LBD-based ligand screening (Garcia *et al.*, 2015; Brewster *et al.*, 2016; Compton *et al.*, 2018). The common feature of these receptors is their sCACHE type LBD. The ligand profiles of these three chemoreceptors were different but overlapping and include a number of carboxylic acids that harbour primarily two or three carbon atoms. A common feature of all three receptors is the binding with significant affinity of acetate ( $K_D$  9.1–34  $\mu$ M) that also cause important chemotactic responses indicative of being the ligand with primary relevance.

- **Evidence for chemoreceptor signal antagonists.** Protein-based ligand screening has been central to the identification of signal antagonists, i.e., compounds that bind to the chemoreceptor LBD but do not cause downstream signalling. Malate is a strong chemoattractant for *P. aeruginosa* and the corresponding chemoreceptor has been identified (Alvarez-Ortega and Harwood, 2007). LBD-based ligand screening identified four additional ligands of this receptor, namely bromosuccinic, citramalic, methylsuccinic and citraconic acids (Martin-Mora *et al.*, 2018a). However, only the former two compounds triggered chemotaxis, whereas the latter two compounds, termed antagonists, were found to compete with the chemoattractants for binding but did not cause any significant response on its own. Importantly, the presence of increasing concentrations of antagonists reduced the magnitude of malate chemotaxis (Martin-Mora *et al.*, 2018a). Evidence for signal antagonists have also been obtained previously for the Tar chemoreceptor (Bi *et al.*, 2013), two-component systems (Busch *et al.*, 2007) and transcriptional regulators (Matilla *et al.*, 2018). Antagonists may thus correspond to a more general phenomenon in bacterial signal transduction.

*Use of microcalorimetry to characterize ligand binding at novel chemoreceptors.* In a number of cases, novel chemoreceptors have been identified by studying chemotaxis of chemoreceptor mutants. In subsequent studies, ITC has proven to be essential to determine whether ligand binding occurs directly or via a ligand-binding protein. Frequently, binding constants derived by ITC have identified ligands that are preferentially recognized providing interesting insight into forces that have led to chemoreceptor evolution in a particular habitat. Listed below are some representative examples.

- **Quaternary amines.** Analyses of alfalfa seed exudates led to the identification of quaternary ammonium compounds as abundant constituents and *S. meliloti* showed chemoattraction to these compounds (Webb *et al.*, 2017b). Chemotaxis assays with chemoreceptor mutants identified McpX as the corresponding chemoreceptor. TSAs have then been used to demonstrate that these compounds bind directly to the LBD and ITC experiments revealed that this receptor has a strong ligand preference for choline, which is the most abundant quaternary amine in seed exudates (Webb *et al.*, 2017b).

- **Taurine.** Taurine is a major bile component and abundant in the intestine of humans and other vertebrates. *Vibrio cholerae*, the etiological agent of cholera, was found to possess a chemoreceptor with a dCACHE domain, Mlp37, that mediates chemotaxis to taurine (Nishiyama *et al.*, 2016). In the presence of taurine, Mlp37 became methylated and ITC of the Mlp37-LBD was essential to demonstrate that this receptor binds taurine directly ( $K_D$  = 3.2  $\mu$ M), but also a number of amino acids with similar affinity.

- **Boric acid.** In analogy to the above studies, chemotaxis assays with chemoreceptor mutants of the phytopathogen *Ralstonia pseudosolanacearum* identified a chemoreceptor for boric acid, termed McpB. Boron serves as a micronutrient in prokaryotes and is required for the synthesis of a number of biologically active borate containing compounds (Rezanka and Sigler, 2008). Microcalorimetric titrations showed that boric acid binds directly to the individual 4HB LBD with a  $K_D$  of 5.4  $\mu$ M (Hida *et al.*, 2017). The derived n value suggested a one ligand per LBD dimer stoichiometry, similar to Tar (Milligan and Koshland, 1993); a finding that was verified by analytical ultracentrifugation (Hida *et al.*, 2017).

- **Inorganic phosphate.** *Pseudomonas aeruginosa* shows strong chemotaxis to inorganic phosphate (Pi), which is a central signal molecule controlling the expression of virulence related genes (Zaborin *et al.*, 2009; Bains *et al.*, 2012). Chemotaxis experiments revealed that a mutant defective in the Ctp<sub>H</sub> chemoreceptor failed to respond to high Pi concentrations, whereas a mutant in the Ctp<sub>L</sub> chemoreceptor did not respond to low Pi concentrations (Wu *et al.*, 2000). Microcalorimetric titrations showed that Ctp<sub>H</sub>-LBD bound preferentially Pi and in addition several pyrophosphate containing compounds with lower affinity (Rico-Jimenez *et al.*, 2016). In contrast, Ctp<sub>L</sub>-LBD failed to bind Pi, a finding that has motivated studies to identify potential periplasmic ligand-binding proteins. Pull-down experiments permitted the identification of Pi loaded PstS as the Ctp<sub>L</sub>-LBD ligand (Rico-Jimenez *et al.*, 2016). PstS is the periplasmic Pi



binding protein that provides the substrate to the PstABC transporter (Nikata *et al.*, 1996). Further ITC studies showed that PstS binds Pi with a  $K_D$  of 7 nM, representing ultratight binding that is related to the capacity of the receptor to respond to low Pi concentrations (Rico-Jimenez *et al.*, 2016).

So far, chemoreceptors for three important inorganic anions have been identified, namely nitrate, borate and inorganic phosphate. Common features of the corresponding chemoreceptors, McpN, McpB, CtpH and CtpL, include their high ligand specificity and the four helix bundle architecture of the corresponding LBDs (Rico-Jimenez *et al.*, 2016; Hida *et al.*, 2017; Martin-Mora *et al.*, 2019).

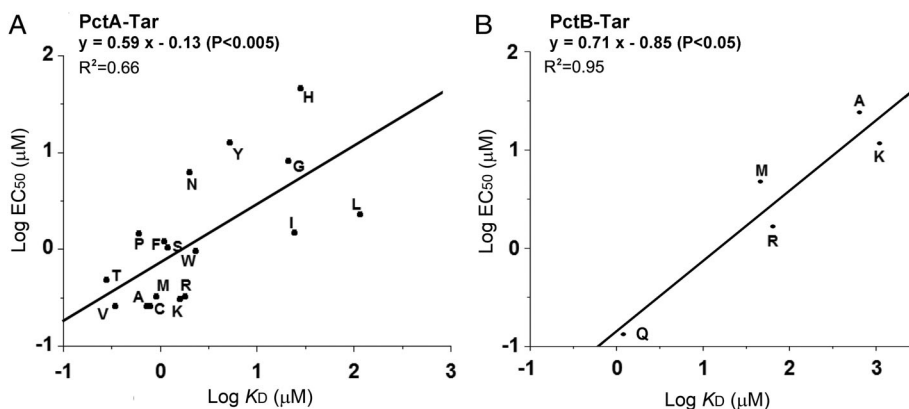
- **Citrate.** The saprophytic *P. putida* KT2440 was shown to possess with McpS a chemoreceptor for most TCA cycle intermediates (Lacal *et al.*, 2010; Pineda-Molina *et al.*, 2012). McpS is the first characterized member of the chemoreceptor family with a HBM domain (Ortega and Krell, 2014). Microcalorimetric titrations resulted in a stoichiometry of one ligand per LBD dimer (Pineda-Molina *et al.*, 2012) that corresponds thus to the same stoichiometry as the Tar chemoreceptor (Milligan and Koshland, 1993). Citrate is abundantly present in plant tissues and root exudates, where it is primarily complexed with bivalent cations like  $Mg^{2+}$ . However, ITC studies showed that McpS-LBD only recognized free citrate with low affinity and failed to bind citrate in complex with physiologically relevant metal ions (Lacal *et al.*, 2011). Considering the physiological relevance of citrate, it was hypothesized that there may be another receptor that mediates citrate chemotaxis. The research undertaken led to the identification of the HBM domain containing chemoreceptor McpQ that was shown by ITC to bind exclusively citrate either in its free or metal bound form (Martin-Mora *et al.*, 2016a). Interestingly, the affinity of metal-bound citrate was superior to that of free citrate (Martin-Mora *et al.*, 2016a), a finding that may

reflect the presence of primarily metal bound citrate in the environment.

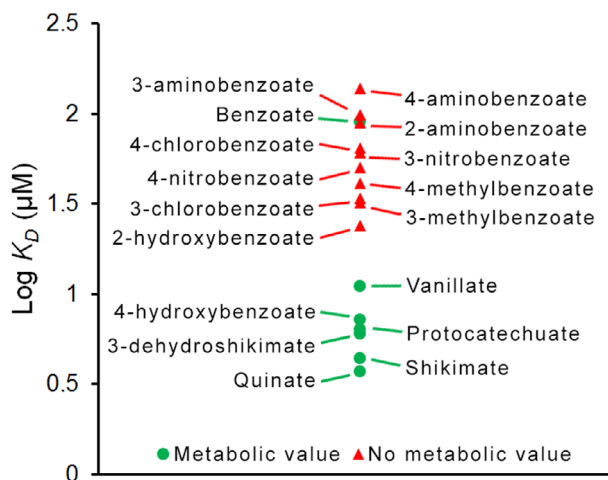
- **Cis-aconitate.** The intermediate following citrate in the TCA cycle is *cis*-aconitate. A mutant of the MCP2983 chemoreceptor from *Comamonas testosteroni* failed to mediate chemotaxis to a series of carboxylic acids and aromatic compounds (Ni *et al.*, 2015). However, microcalorimetric titrations of the MCP2983-LBD (4HB) showed that *cis*-aconitate is the only compound recognized. To explain the discrepancy between the specificity of ligand recognition and the broad range of chemotaxis mediated by this receptor, the authors argue that the metabolism of different chemoeffectors results in *cis*-aconitate that in turn is recognized by the chemoreceptor producing this broad chemotactic response (Ni *et al.*, 2015).

Taken together, microcalorimetry was essential to document the existence of chemoreceptors that recognize preferentially or exclusively a single TCA cycle intermediate, namely citrate (Martin-Mora *et al.*, 2016a), *cis*-aconitate (Ni *et al.*, 2015),  $\alpha$ -ketoglutarate (Martin-Mora *et al.*, 2016b) and malate (33,57), indicating the central physiological relevance of these compounds for a broad range of bacteria.

**Relating system input and output.** Signal transduction processes involve the conversion of an input into an output. The dissociation constant of a ligand to a chemoreceptor LBD is a measure of the signalling input. A remarkable characteristic of an ITC analysis is the precision with which  $K_D$  values can be determined. On the other hand, fluorescence resonance energy transfer (FRET) measurements using cells with fluorescent labelled CheY and CheZ is an approach that permits a very precise definition of the signalling output (Paulick and Sourjik, 2018). The combination of both approaches has provided interesting insight into chemoreceptor function.



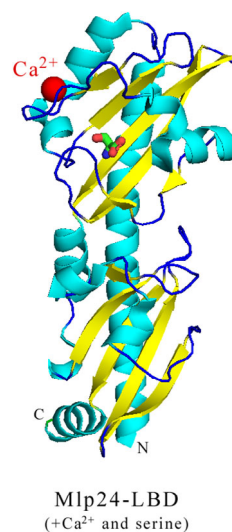
**Fig 5** Relationship between signalling input and output for the PctA (A) and PctB (B) chemoreceptors. Signalling input is represented by the  $K_D$  values determined by microcalorimetric titrations of individual LBDs. Signalling output corresponds to the  $EC_{50}$  values derived from FRET measurements of chemoreceptor chimera combining the PctA or PctB LBDs of *P. aeruginosa* PAO1 with the *E. coli* Tar signalling domain. Figure reproduced with permission from Reyes-Darias *et al.* (2015a).



**Fig 6** Relationship between chemoreceptor affinity and metabolic value of chemoattractants. Binding of ligands to the LBD of the PcaY\_PP chemoreceptor of *P. putida* KT2440. Shown are  $K_D$  values as derived by microcalorimetric titrations and the colour indicates the metabolic value. Data taken from Fernandez *et al.* (2017). [Color figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)]

dCACHE domains are extremely abundant in bacteria and present in all major prokaryotic signal transduction families that rely on extracytoplasmic sensing (Upadhyay *et al.*, 2016). In addition, almost one third of all chemoreceptors possess a CACHE domain (Ortega *et al.*, 2017a). The three paralogous chemoreceptors PctA, PctB and PctC of *P. aeruginosa* can be considered as models to study dCACHE-containing chemoreceptors (Taguchi *et al.*, 1997; Schmidt *et al.*, 2011; Rico-Jimenez *et al.*, 2013; McKellar *et al.*, 2015; Xu *et al.*, 2016; Corral-Lugo *et al.*, 2018; Gavira *et al.*, 2020). PctA and PctB recognize 17 and 5 amino acids respectively (Rico-Jimenez *et al.*, 2013). To determine the signalling output of both receptors, chimeric receptors were produced that contain the PctA or PctB LBD fused to the Tar signalling domain. These chimeras were introduced into a chemoreceptor-free *E. coli* strain and responses measured in function of different concentrations of each of the ligands, leading to the determination of  $EC_{50}$  values (Reyes-Darias *et al.*, 2015a). Interestingly, these  $EC_{50}$  values correlated with  $K_D$  values as determined by ITC, representing the signalling input (Fig. 5). There was thus a correlation between the magnitude of signal input and output, indicating that the onset of binding determines the onset of response (Reyes-Darias *et al.*, 2015a). Further studies are necessary to determine to what degree these findings apply to other chemoreceptors.

*Metabolic value has shaped chemoreceptor evolution.* Studies of *E. coli* have shown that chemotaxis occurs preferentially to amino acids that are rapidly consumed, indicating that the nutritional value of a ligand has shaped chemotaxis (Yang *et al.*, 2015). To elucidate potential



**Fig 7** Three dimensional structure of a chemoreceptor ligand binding domain with multiple bound ligands. LBD of Mlp24 chemoreceptor of *V. cholerae* O395N1 in complex with serine and  $\text{Ca}^{2+}$  (PDB ID: 6IOU). [Color figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)]

molecular mechanisms of this preference, we have conducted studies of the PcaY\_PP chemoreceptor of *P. putida* KT2440 (Fernandez *et al.*, 2017). This receptor and its homologue in *P. putida* F1 (Luu *et al.*, 2015) were found respond to a wide range of C6 ring containing carboxylic acids of which only some are of apparent metabolic value, as indicated by their capacity to support bacterial growth as sole carbon source.

Microcalorimetric titrations of PcaY\_PP-LBD revealed binding of 17 compounds, with  $K_D$  values between 3.7 and 138  $\mu\text{M}$  (Fernandez *et al.*, 2017). All compounds induced chemotaxis but only seven of these compounds were of metabolic value (Fernandez *et al.*, 2017). However, the affinities of metabolic value compounds were in general significantly higher than those of no metabolic value compounds (Fig. 6). These data suggest that the nutritional value of a compound has shaped molecular recognition during chemoreceptor evolution. The most tightly binding ligand was quinate which is a very abundant carbon storage compound in plants (Lehmann *et al.*, 2016) and chemotaxis of saprophytic *P. putida* KT2440 to this compound may represent a significant benefit.

*Binding of multiple ligands to chemoreceptor LBDs.* The Mlp24 chemoreceptor of *V. cholerae* was shown to mediate chemotaxis to a series of amino acids (Nishiyama *et al.*, 2012). When the 3D structures of Mlp24-LBD in complex with different amino acids were inspected, a  $\text{Ca}^{2+}$  ion was bound to a loop next to the amino acid binding pocket (Takahashi *et al.*, 2019) (Fig. 7). Chemotaxis experiments showed that  $\text{Ca}^{2+}$  binding does not trigger

chemotaxis and microcalorimetry played an essential role to determine the relevance of  $\text{Ca}^{2+}$  binding for the functioning of this receptor. Thus, when Mlp24-LBD was titrated with different amino acids in the absence and presence of saturating  $\text{Ca}^{2+}$  concentrations, the authors noted a significant increase in affinity for some but not all amino acids in the presence of  $\text{Ca}^{2+}$ ; a finding that was also reflected in an increase in the magnitude of chemotaxis to some of the amino acids. *V. cholerae* is present in seawaters and colonizes the intestine, niches that contain approximately 10 mM  $\text{Ca}^{2+}$ . As a result, the authors suggest that  $\text{Ca}^{2+}$  corresponds to a physiologically relevant co-signal that modulates the affinity for the primary amino acid signal (Takahashi *et al.*, 2019). Further studies will show whether the existence of co-signals is a more general phenomenon in chemoreceptors.

### Conclusions

Chemosensory systems are often highly complex since they involve many signalling proteins, which are frequently organized in multiple pathways that exert different cellular functions. Furthermore, a single bacterial genome can encode a large number of chemoreceptors that at times possess overlapping ligand specificities and the significant sequence divergence of LBDs hampers chemoreceptor annotation by homology with characterized receptors. These issues represent a significant challenge to the scientific community since the identification of the signal molecules recognized by specific chemoreceptors is key to understand their physiological role(s). ITC has been shown to be a very potent tool to tackle some of these challenges. Importantly, the fact that most LBDs can be produced in significant amounts as individual proteins that maintain ligand-binding properties has potentiated the role of ITC in the identification of many novel chemoreceptors. The major limitations of a traditional microcalorimetric analysis are the requirement of significant amounts of protein and its labour intensity. However, microcalorimetric technology is advancing rapidly and last generation instruments require much less protein. In addition, automated instruments that conduct experiments and analyse data in an almost autonomous manner are now available. This, and the fact that the approach used to study chemoreceptors can also be applied to other signal transduction systems will enhance the importance of microcalorimetry in laboratories studying prokaryotic signal transduction.

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