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Impacts of stream acidification on litter breakdown: implications for assessing ecosystem functioning

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Summary

1. Scientific understanding of acidification in aquatic ecosystems relies on effective assessment, which at present is mostly limited to chemical and sometimes structural biological variables. Effects on ecosystem functioning are, in contrast, largely neglected. Litter breakdown is a potentially useful, highly integrative and crucial process that could enhance such assessment programmes.

2. Breakdown rates of beech *Fagus sylvatica* leaves were determined in 25 woodland headwater streams along an acidification gradient in the Vosges Mountains, France. Additional data relating to micro-organisms (microbial respiration, fungal biomass and degree of conditioning measured as leaf palatability) and macroinvertebrates (shredder diversity, abundance and biomass) associated with decomposing leaves were collected to elucidate the mechanisms underlying leaf breakdown.

3. Breakdown rates varied more than 20-fold between the most acidified and circum-neutral sites ($k = 0.0002\text{--}0.0055\text{ day}^{-1}$). Stream water alkalinity and total Al concentration together accounted for 88% of the variation in litter breakdown rates among streams. Microbial factors associated with decaying leaves, particularly microbial respiration, declined with increasing stream acidity and were significantly related to Ca^{2+} and total Al concentrations.

4. Total abundance, biomass and richness of leaf-shredding invertebrates associated with decomposing leaves were not related to stream acidity. However, the abundance and biomass of the amphipod *Gammarus fossarum*, an acid-sensitive and particularly efficient leaf-shredder, showed a strong positive relationship with leaf breakdown rate. *Gammarus* abundance and microbial respiration together accounted for 85% of the variation in litter breakdown rates among streams.

5. *Synthesis and applications.* These results indicate that leaf-litter breakdown responds strongly to stream acidification, with both microbial decomposers and invertebrate detritivores markedly affected. Measuring leaf breakdown rate may be developed into a simple, powerful and low-cost tool for assessing a critical component of ecosystem functioning. We advocate further investigation of this approach for the routine bio-monitoring of freshwaters affected by, or recovering from, other anthropogenic stresses.

Key-words: bioassessment, environmental monitoring, fungi, leaf decomposition, leaf-shredding invertebrates, microbial respiration, water quality

Introduction

Long-range atmospheric transport and subsequent deposition of acidifying substances has degraded surface waters in many areas of the world (Schindler 1988; Driscoll *et al.* 2001). Since the 1980s, rates of acidic

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deposition have decreased across North America and Europe following national and international abatement protocols (e.g. the Clean Air Act in the USA and the Convention on Long-Range Transboundary Air Pollution in Europe; Likens, Butker & Buso 2001). However, both empirical results and dynamic modelling indicate that the expected recovery of alkalinity has not occurred in all areas where sulphate deposition has decreased (Stoddard *et al.* 1999). In addition, Ca^{2+} and other cations continue to be depleted in surface waters draining geologically sensitive areas, apparently as a result of the high rates of acidic deposition in the past and recent decreases in cation deposition (Lawrence *et al.* 1999; Stoddard *et al.* 1999). Further, the relative importance of elevated nitrogen deposition has recently increased in freshwater acidification (Sullivan *et al.* 1997; Lepori, Barbieri & Ormerod 2003a,b). Thus, it has become clear that acidification impacts will continue to deteriorate aquatic ecosystems for decades (Driscoll *et al.* 2001), perhaps exacerbated by the effect of global warming (Schindler *et al.* 1996).

Sound understanding of acidification effects in ecosystems relies on effective assessment. Most assessment programmes currently in place are largely limited to atmosphere-water or surface-water chemistry, whereas effects on biological communities and whole ecosystems are now rarely analysed (Lancaster *et al.* 1996; Bradley & Ormerod 2002). Episodic inputs of acidic rainwater, and especially meltwater, typically result in large temporal fluctuations in the chemical quality of acid-sensitive waters. Consequently, large sampling efforts are needed to derive reliable information on the ecological effects of acidification from chemical measures alone. In contrast, biological variables integrate acidification effects through time, thus circumventing the problem of temporal variability in water chemistry (Lancaster *et al.* 1996). Benthic invertebrates have been used successfully to this effect (Otto & Svensson 1983; Fjellheim & Raddum 1990; Larsen *et al.* 1996), although some concern has been raised that differences among regions in species' tolerance limits may weaken the use of general indicator models (Hämäläinen & Huttunen 1996).

Many goals in the conservation and restoration of aquatic ecosystems relate to ecosystem processes (Boon 2000). In addition to structural biological parameters (e.g. macroinvertebrate, fish and diatom community structure), it is desirable to determine which ecosystem processes respond to anthropogenic stresses, and which may be good indicators of functional ecosystem damage (Wallace, Grubaugh & Whiles 1996; Gessner & Chauvet 2002). Functional traits of both plants and animals can indicate ecosystem functioning (Statzner *et al.* 2001; Hooper *et al.* 2002) but the extent to which modified community structure reflects altered ecosystem processes in streams is poorly known (Wallace & Webster 1996). Assessment schemes targeted at ecosystem processes are few (Minshall 1996; Bunn & Davies 2000; Gessner & Chauvet 2002). This gap is a

fundamental limitation of current ecosystem assessment, particularly in view of the American Clean Water Act (33 USC Section 1251 *et seq.*; <http://www4.law.cornell.edu/uscode/33/1251.html>) and the European Water Framework Directive (EU 2000; http://europa.eu.int/comm/environment/water/water-framework/index_en.html), which both draw explicit attention to the functional integrity of aquatic ecosystems.

Examples of potentially useful ecosystem processes in stream assessment include primary production, sediment and community respiration, nutrient transformations and leaf-litter breakdown (Minshall 1996; Giller & Malmqvist 1998; Gessner & Chauvet 2002). Because forested headwater streams receive a large proportion of their energy inputs from allochthonous leaf litter from riparian vegetation (Wallace *et al.* 1997, 1999), leaf breakdown plays a pivotal functional role (Webster & Benfield 1986; Maltby 1996; Gessner, Chauvet & Dobson 1999; Graça 2001). Consequently, litter breakdown is a key process for assessing the possible impact of acidification on stream ecosystem functioning (Maltby *et al.* 1995; Wallace, Grubaugh & Whiles 1996; Bunn, Davies & Mosisch 1999; Nelson 2000; Niyogi, Lewis & McKnight 2001; Gessner & Chauvet 2002). Effects of stream acidification on leaf breakdown have been suggested, but so far from only a small number of sites (Burton, Stanford & Allan 1985; Chamier 1987; Groom & Hildrew 1989; Mulholland *et al.* 1992; Griffith & Perry 1993; Dangles & Guérol 1998; cf. Niyogi, Lewis & McKnight 2001).

The main objective of the present study was to determine whether leaf breakdown was systematically related to acidification across a range of forest streams. Leaf breakdown was studied along an acidification gradient at 25 sites differing in pH, alkalinity, Ca^{2+} and total Al concentration. A range of biological variables relating to the microbial and macroinvertebrate communities associated with decomposing leaves was analysed. Finally, interannual variability in litter breakdown was examined by repeating the experiment in three consecutive years in seven selected streams.

Methods

STUDY SITES

The Vosges Mountains, in north-eastern France, are sensitive to acidification due to the prevalence of quartz-enriched sandstone and granite bedrock (Dambrine *et al.* 1995). Atmospheric deposition during the last few decades (up to $1.0 \text{ kmol ha}^{-1} \text{ year}^{-1}$ of H^+ and $15 \text{ kg ha}^{-1} \text{ year}^{-1}$ of SO_4^{2-}) has resulted in surface water acidification (Probst *et al.* 1990a; Dambrine *et al.* 1998), particularly in catchments covered by coniferous forests (Probst *et al.* 1990b). An important characteristic of the area is that small-scale differences in mineral composition of the underlying bedrock produce marked differences in buffering capacity, ion content and acidification, sometimes even between adjacent streams.

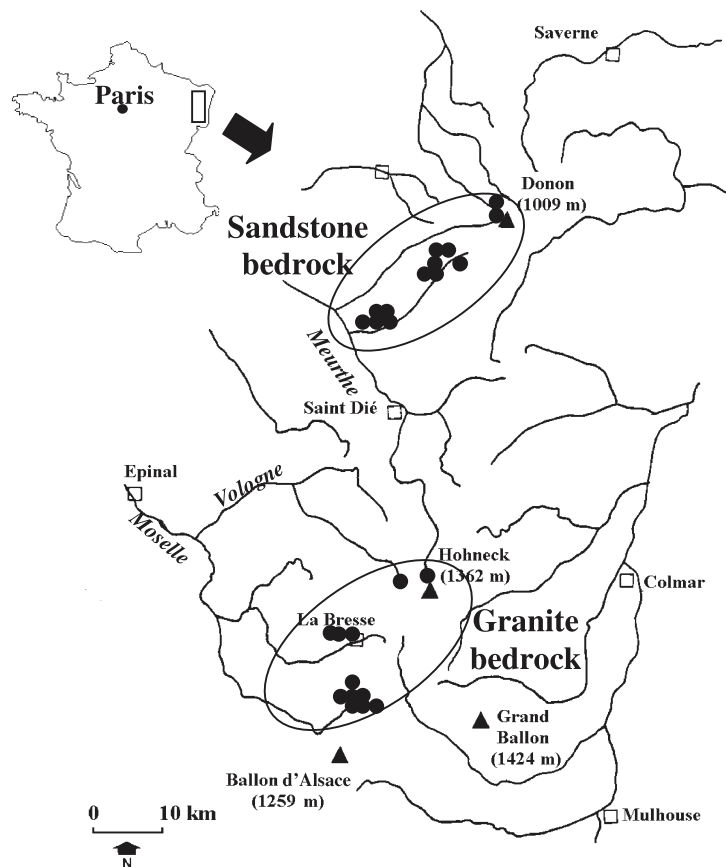


Fig. 1. Location of the study sites in the Vosges Mountains, north-eastern France.

Twenty-five first- and second-order streams varying in acidification status were selected (Table 1 and Fig. 1) based on previous data from 394 streams in the region (Guérol *et al.* 1997). Vegetation in the selected catchments was dominated by silver fir *Abies alba* Mill., Norway spruce *Picea abies* L. and beech *Fagus sylvatica* L. Beech is by far the most common deciduous tree species adjacent to streams and provides almost all leaf-litter inputs. All study sites were located upstream of direct influences such as housing, farming and industrial activities (except selective logging), and showed similar morphology and hydrology.

WATER CHEMISTRY

Temperature and pH were measured monthly from December 1998 to November 1999 *in situ* at all sites using a glass electrode for solutions of low ionic strength (combined pH electrode for deionized water, S7 head; Bioblock Scientific, Illkirch, France). Field and subsequent laboratory measurements of pH produced similar results. Water samples for chemical analyses were also collected monthly and returned to the laboratory in polyethylene bottles. Alkalinity was determined by Gran (1952) titration to pH 3.8. Conductivity was measured with a Metrohm Herisau Conductometer E518 (Herisau, Switzerland) at 20 °C. Concentrations of Ca^{2+} , Mg^{2+} , Na^+ , K^+ and total Al (after acidification with HNO_3) were determined by atomic absorption

spectrophotometry (Aanalyst 100; Perkin Elmer), and concentrations of Cl^- , SO_4^{2-} and NO_3^- by ion chromatography (Dionex 1500i with a AS 4 A SC column; Sunnyvale, USA).

LEAF BREAKDOWN

Beech leaves were collected from trees just before abscission in the autumn of 1998. Three grams of air-dried leaves were enclosed in 5-mm plastic mesh bags (15 × 10 cm), eight of which were submerged at each site on 10 December 1998. Air-drying can entail rapid initial mass loss of leaves through leaching (Gessner & Schwoerbel 1989), but this effect is small for beech leaves (4–5%; Dangles & Guérol 2001a) and was not specifically accounted for. The bags were secured to the bank with plastic-coated wire and tied to boulders on the stream bed. Four replicate bags were recovered after 100 and 200 days and returned to the laboratory. The leaves were rinsed individually with water from the corresponding stream to remove fine particulate matter and invertebrates. Thirty 2-cm diameter discs were cut from the leaves of three bags avoiding the central veins. The remaining leaf material was air-dried to constant mass and subsamples ashed for 4 h at 550 °C to relate air dry mass to ash-free dry mass (AFDM).

The same protocol was used to determine the breakdown of beech leaves at a smaller set of seven sites (GS, WA, RR, BR, MR, HR and LM) in the winter and

Table 1. Characteristics of the study sites. Mean pH values ($n = 7-11$) are from Guérold *et al.* (1997)

Stream	Site	Grid reference	Bedrock geology	Mean pH	Stream order	Altitude (m a.s.l.)	Distance from source (m)	Channel width (m)	Channel slope (‰)	Riparian vegetation
Gentil Sapin	GS	48°02'17"N 7°05'23"E	Sandstone	4.49	1	510	3100	2.30	10.0	<i>Abies alba</i> , <i>Fagus sylvatica</i> , <i>Picea abies</i> , <i>Acer pseudoplatanus</i>
Rabodeau	RB	48°27'35"N 7°03'21"E	Sandstone	4.66	2	520	5000	4.90	16.0	<i>A. alba</i> , <i>F. sylvatica</i> , <i>P. abies</i> , <i>A. pseudoplatanus</i>
Wassongoutte	WA	47°58'40"N 6°53'22"E	Granite	4.71	1	690	2600	2.00	13.8	<i>F. sylvatica</i> , <i>A. alba</i>
Bas-Rupts	BR	48°02'17"N 6°52'54"E	Granite	4.81	1	845	900	1.40	20.0	<i>A. alba</i> , <i>F. sylvatica</i>
Rouge-Rupt	RR	47°58'28"N 6°53'19"E	Granite	4.88	2	675	4700	5.50	8.4	<i>F. sylvatica</i> , <i>A. alba</i>
Malfosse	MA	48°25'18"N 6°56'07"E	Sandstone	4.96	2	350	3400	5.00	8.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
Longfoigneux	LG	47°58'15"N 6°52'17"E	Granite	4.99	1	600	1180	1.30	22.0	<i>F. sylvatica</i> , <i>A. alba</i>
Ventron	VE	47°58'18"N 6°55'22"E	Granite	5.01	1	820	2500	1.50	10.0	<i>F. sylvatica</i> , <i>A. alba</i>
Repafosse	RE	48°25'31"N 6°57'16"E	Sandstone	5.02	1	405	1800	2.10	6.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
Ravines	RV	48°25'47"N 6°57'08"E	Sandstone	5.04	1	410	3300	2.50	6.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
Madeleine	MD	48°27'26"N 7°07'50"E	Sandstone	5.18	1	510	2400	1.90	16.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
Echarges	EC	47°58'39"N 6°54'29"E	Granite	5.22	1	730	2000	2.30	14.0	<i>F. sylvatica</i> , <i>A. alba</i>
Chevrosgoutte	CH	48°24'21"N 6°55'01"E	Sandstone	5.31	1	390	1500	1.20	28.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
Moyens-Rupts	MR	48°02'24"N 6°52'55"E	Granite	5.76	1	850	1000	1.60	24.0	<i>A. alba</i> , <i>F. sylvatica</i>
Grands-Clos	GC	47°58'47"N 6°52'35"E	Granite	5.77	1	650	1750	2.30	21.1	<i>F. sylvatica</i> , <i>A. alba</i>
La Maix	LM1	48°29'03"N 7°05'40"E	Sandstone	5.85	1	516	2200	2.00	8.0	<i>F. sylvatica</i> , <i>A. alba</i> , <i>Alnus glutinosa</i>
Hauts-Rupts	HR	48°02'41"N 6°53'37"E	Granite	6.24	2	855	2550	3.30	13.9	<i>A. alba</i> , <i>F. sylvatica</i>
La Maix	LM3	48°29'34"N 7°03'36"E	Sandstone	6.38	1	514	30	1.30	3.0	<i>F. sylvatica</i> , <i>A. alba</i> , <i>A. glutinosa</i>
La Maix	LM2	48°28'06"N 7°05'07"E	Sandstone	6.44	2	512	2250	2.00	8.0	<i>F. sylvatica</i> , <i>A. alba</i> , <i>A. glutinosa</i>
Tihay	TI	47°58'50"N 6°52'32"E	Granite	6.47	1	660	1250	1.80	20.0	<i>A. alba</i> , <i>F. sylvatica</i> , pasture
Meurthe	ME	48°04'50"N 7°00'52"E	Granite	6.48	1	825	3300	2.90	6.8	<i>F. sylvatica</i> , <i>A. alba</i> , pasture
Vologne	VO	48°03'17"N 6°59'36"E	Granite	6.60	1	810	2100	3.10	19.5	<i>F. sylvatica</i> , <i>A. alba</i>
Plaine	PL	48°31'39"N 7°08'24"E	Sandstone	6.94	2	490	4900	4.90	3.3	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i> , <i>A. glutinosa</i>
Truite	TR	48°28'08"N 7°08'47"E	Sandstone	7.05	1	500	2500	2.10	12.0	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i>
La Maix	LM4	48°29'08"N 7°05'16"E	Sandstone	7.26	2	410	4900	2.20	9.5	<i>F. sylvatica</i> , <i>P. abies</i> , <i>A. alba</i> , <i>A. glutinosa</i>

spring of 1997–98 and 1999–2000, to test for interannual variability of leaf breakdown rates. These selected sites represented the full range of pH values between 4.5 and 7.3. As in 1998–99, leaves were placed in the streams in December and retrieved after 100 and 200 days.

MICROBIAL RESPIRATION

Microbial respiration was measured on leaves retrieved after 100 days. Three replicate sets of 30 leaf discs each were placed in 10-mL flasks filled with water collected at the same time in the corresponding site (base flow conditions) and incubated with stirring. Temperature was maintained at 4 ± 1 °C during incubations, close to ambient water temperature for all streams in early March. An additional flask containing only stream water was used as a control. Four oxygen probes (Cellox 325, Woburn, USA) connected to an oxymeter (WTW Multiline P4, Woburn, USA) measured oxygen concentrations every 16 s for 3–5 h, depending on rates of oxygen decline. Respiration rates were calculated from slopes of oxygen concentrations regressed against time during periods of linear decrease, and are expressed as mg O₂ g leaf AFDM⁻¹ day⁻¹.

FUNGAL BIOMASS

Ergosterol was used as an indicator of fungal biomass (Gessner & Newell 2002). Three replicate sets of 30 discs from the leaves retrieved after 100 days were blotted dry and preserved in pure methanol. Ergosterol was extracted by 30 min refluxing in alcoholic base, purified by solid-phase extraction and quantified by high-pressure liquid chromatography (Gessner & Schmitt 1996).

LEAF PALATABILITY

The palatability of leaves to detritivores was defined as leaf consumption by the shredding amphipod *Gammarus fossarum* (Koch), an abundant species in most circum-neutral streams in the study area. Animals were collected with a hand net from site LM4. They were acclimated to laboratory conditions (8 ± 1 °C, natural 12-h daylight period, water from LM4) for 24 h. Ten gammarids and the 30 leaf discs used previously for measuring microbial respiration were then placed in plastic cylinders (diameter 6 cm, height 5 cm) covered with a 0.3-mm mesh screen on both ends. The initial AFDM of the leaf material was determined on five additional discs cut from the leaves from each bag. Three replicate cylinders with animals, each corresponding to a different leaf bag, and one control without animals were used for each of the 25 streams. All cylinders were placed together in an aquarium filled with filtered water from LM4 and aerated with an aquarium pump. After 6 days the animals and the remaining leaf fragments were removed, oven-dried (105 °C, 24 h) and ashed (550 °C, 4 h). Consumed leaf mass was calculated as the difference between the initial and remaining AFDM.

SHREDDER ABUNDANCE AND BIOMASS

Invertebrate shredders (as classified by Tachet *et al.* 2000) from the leaf bags were preserved in 70% ethanol, identified to the lowest practicable taxonomic level (mostly species or genus) and counted. Biomass was determined by weighing dried individuals (105 °C, 24 h) to the nearest 0.01 mg.

DATA ANALYSIS

Principal components analysis (PCA) was carried out using the correlation matrix of $\log(x + 1)$ transformed data to ordinate streams with respect to all physico-chemical variables. Calculations and graphics were made using ADE 4.0 software, a data analysis package designed for multivariate statistics (Thioulouse *et al.* 1997; <http://pbil.univ-lyon1.fr/ADE-4>).

Leaf breakdown rates (k) were estimated by fitting the AFDM data to the exponential model, $Y_t = Y_0 e^{-kt}$, where Y_t is the AFDM remaining at time t in days, and Y_0 the AFDM at the beginning of the experiment (Petersen & Cummins 1974).

Partial least-squares (PLS) regression (Abdi 2003) was used to construct a descriptive model predicting litter breakdown from water chemistry. PLS regression analysis can be viewed as a generalized multiple regression that copes with collinearity, thus adequately taking into account the auto-correlations among the physico-chemical variables (e.g. pH, alkalinity, Ca²⁺ and Mg²⁺ concentrations). The generated variable importance in the projection (VIP) values reflect the importance of a given variable (e.g. mean pH, minimum alkalinity, minimum total Al concentration) in the model with respect to both the breakdown rate and the projected water chemical variables. PLS regressions were calculated with the software SIMCA (version 7.0; Umetri AB, Umeå, Sweden). Mean, median and stress values based on the 11 monthly measurements were used as input variables. Stress values refer to either the lowest (e.g. pH, alkalinity, Ca²⁺ concentration) or highest (e.g. total Al concentration) value in the data set, depending on whether high or low values were considered stressful.

Multiple linear regressions were used to detect effects of the three major stresses associated with acidified waters (i.e. low pH, low Ca²⁺ and high total Al concentrations) on eight leaf-associated biological variables: microbial respiration; fungal biomass; leaf palatability; abundance, biomass and richness of shredder taxa; and abundance and biomass of *Gammarus*. Simple linear regressions were computed using each of these biological variables as a predictor of leaf breakdown rate. Further, a PLS analysis followed by a stepwise multiple linear regression was conducted using all significant ($P < 0.05$) predictors from the simple linear regressions on the response variable; the PLS analysis was carried out first to select variables for the subsequent stepwise regressions that showed no auto-correlation. For these analyses, we considered breakdown rates calculated for the first 100 days only.

Finally, a two-way analysis of variance (ANOVA) was used on log-transformed data from the seven streams examined in three consecutive years to test for effects of 'year' and 'site' on leaf breakdown rate.

Results

WATER CHEMISTRY

The first axis of the PCA of physical and water chemical data was defined primarily by four key variables related to stream water acidity: pH, alkalinity and concentrations of Ca^{2+} and Mg^{2+} (Fig. 2). It explained 62% of the total variance. An additional 23% of the variance was explained by the second axis, which separated the streams in catchments underlain by sandstone vs. granite bedrock. Streams in granitic catchments (e.g. LM1 and GC) were characterized by lower ion contents at a

given pH, including lower concentration of sulphates but also lower concentrations of total Al, compared with streams draining sandstone. The dispersion of individual samples at a given site indicated seasonal and event-driven chemical changes. For example, at the sites ML, MD, CH and LM1, pH dropped by ≈ 1.8 units during a prolonged period of snow-melt in March 1999. Other streams (e.g. GS and RB), however, were permanently acidic, and water chemistry of the relatively well-buffered sites (LM4, LM3 and ME) also varied little during the study.

LEAF BREAKDOWN RATES

Leaf breakdown rates ranged from 0.0002 day^{-1} at the most acidic site (GS) to 0.0055 day^{-1} at the circumneutral site LM4. Alkalinity was the single best physicochemical predictor of breakdown rate, as inferred from the

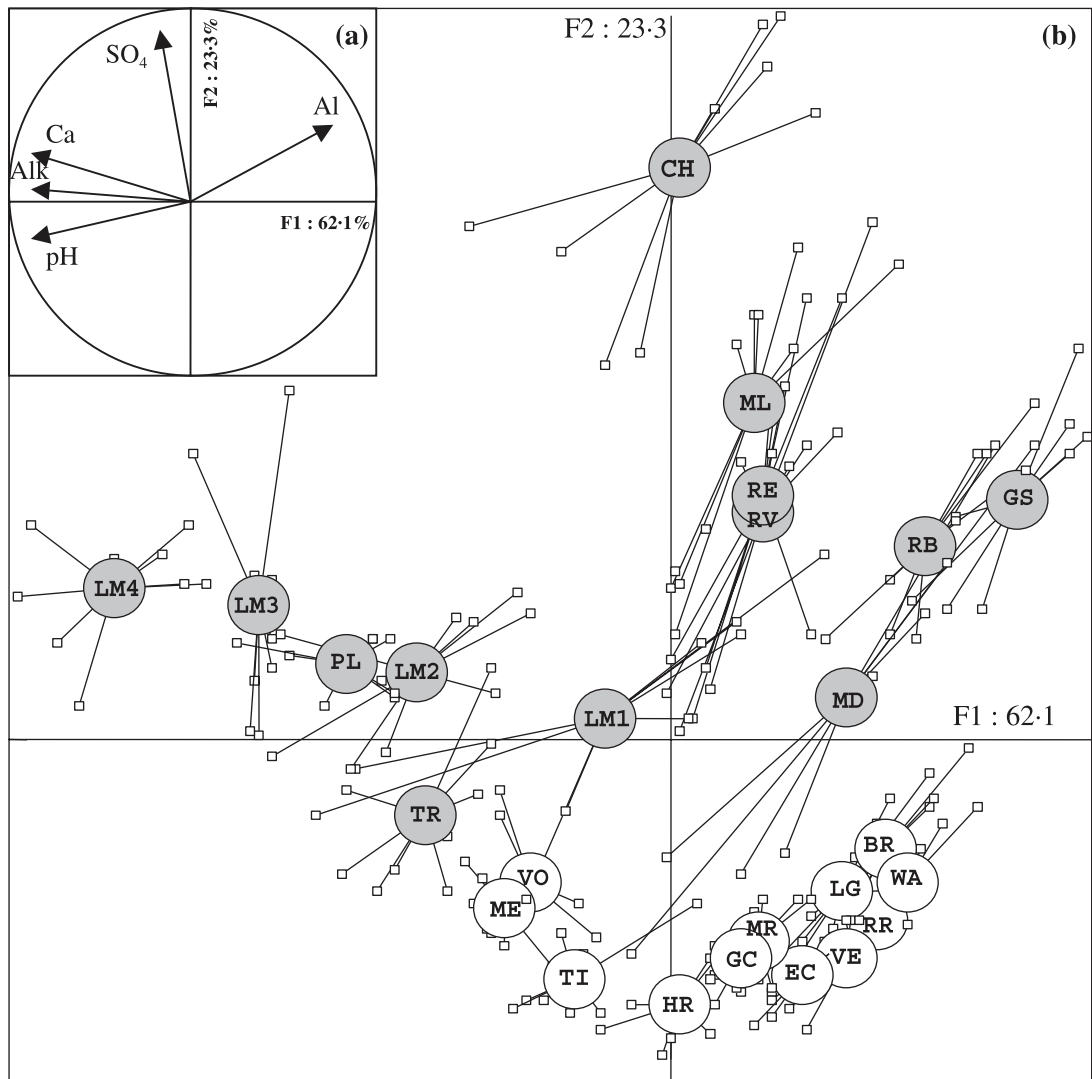


Fig. 2. PCA of physicochemical variables determined monthly between December 1998 and December 1999 at 25 stream sites in the Vosges Mountains. (a) Position of variables on the $F1 \times F2$ plane. (b) Ordination of the 25 selected sites on the $F1 \times F2$ plane. Small squares represent the sample position at each sampling date. Open circles denote the weighted average of all samples taken from a given stream. Grey circles, streams draining sandstone bedrock; open circles, streams draining granitic bedrock. Lines link samples to their weighted average.

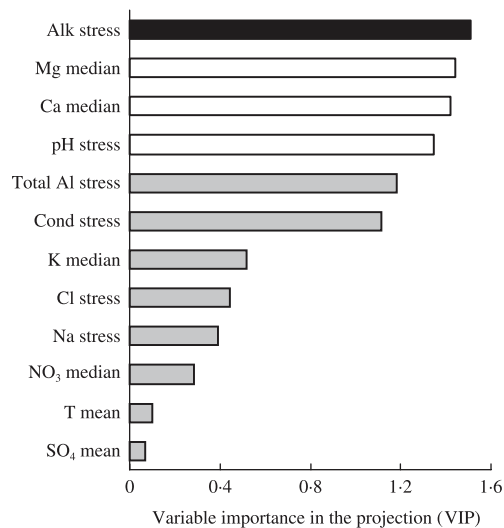


Fig. 3. Results of PLS regression analysis showing the importance of different physicochemical variables in the projection. Because VIP values of mean, median and stress values for each parameter were very similar, only the highest of these values is presented for each physicochemical variable. The black bar represents the most important variable (MIV); white bars show variables significantly correlated with MIV (d.f. = 23, $r^2 > 0.50$, $P < 0.05$, Pearson rank test); grey bars show parameters not significantly correlated with MIV. Alk, alkalinity; Cond, conductivity; T, temperature.

highest VIP (1.47) assigned to alkalinity in the PLS regression analysis (Fig. 3). Stream pH and concentrations of Ca^{2+} and Mg^{2+} were significantly correlated with alkalinity ($P < 0.05$), whereas total Al concentration (VIP = 1.18) was not ($P = 0.21$). Combining alkalinity and Al concentration in a single predictor, $\log(1 + \text{alkalinity}_{\text{stress}}/\text{total Al}_{\text{stress}})$, improved the prediction of leaf breakdown rate (VIP = 1.60). As a result, leaf breakdown rate showed a strong positive relation with this ratio, irrespective of whether data from streams with zero alkalinity were included in the regression

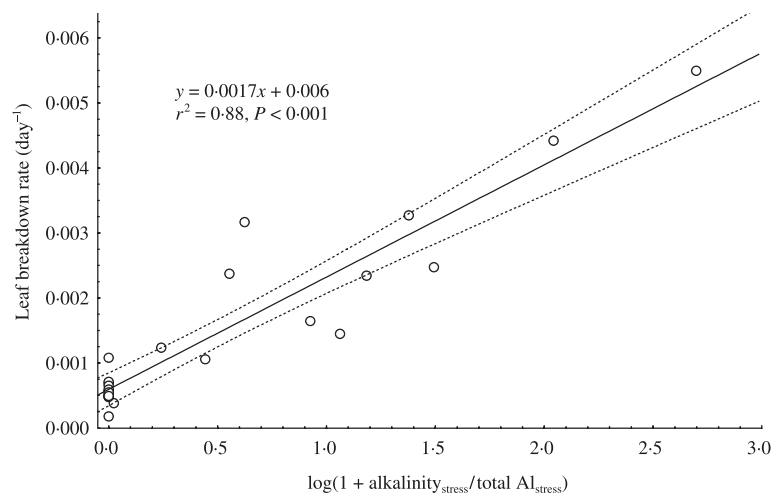


Fig. 4. Regression of leaf breakdown rate vs. $\log(1 + \text{alkalinity}_{\text{stress}}/\text{total Al}_{\text{stress}})$. The dashed lines represent 95% confidence limits of the regression.

analysis (Fig. 4) or excluded ($y = 0.0017x + 0.0007$, $r^2 = 0.78$). Leaf breakdown rates ($P > 0.88$) were not significantly related to physical variables such as channel width, slope and depth, distance from source and altitude. Similarly, the influence of discharge was likely to be insignificant because variations across streams were small (about 15 and 30 L s^{-1} at baseflow in first- and second-order, respectively).

BIOLOGICAL FACTORS ASSOCIATED WITH DECAYING LEAVES

Multiple linear regression analysis showed that five out of eight biological variables associated with decaying leaves were significantly predicted by pH, concentrations of Ca^{2+} or total Al (Table 2). In no case was the addition of a second variable significant. The highest coefficients of determination were between ergosterol content and pH ($r^2 = 0.38$), microbial respiration and total Al ($r^2 = 0.60$), leaf palatability and pH ($r^2 = 0.62$) and *Gammarus* abundance, biomass and Ca^{2+} concentration ($r^2 = 0.48$ and $r^2 = 0.52$, respectively). Non-linear models did not generally improve relationships.

The number and biomass of *Gammarus* associated with decaying leaves were more strongly related to leaf breakdown rate for each leaf pack than microbial variables (Fig. 5). Among the latter, leaf breakdown rate was more closely related to microbial respiration than to fungal biomass and leaf palatability. This relationship was positive, although several sites had high rates of microbial respiration but low rates of litter breakdown. Total shredder density and biomass were weakly correlated with leaf breakdown rate, as was shredder richness.

The six significant predictors from the single linear regressions (fungal biomass, microbial respiration, leaf palatability, shredder abundance, *Gammarus* abundance and biomass) were used in a PLS analysis followed by a stepwise multiple regression. Two significant predictors,

Table 2. Multiple linear regression analysis of various biological variables associated with decaying leaves after 100 days of exposure against the three major stress factors (pH, calcium (Ca²⁺), aluminium (Al)) in acidified water (mean, median and stress values of pH, Ca²⁺ and total Al) at the 25 study sites. Overall *r*² and *P*-values are based on multiple regressions with mean, median and stress values of pH, Ca²⁺ and Al. Only the two variables providing the best univariate regressions are shown

Dependent variable	Overall <i>r</i> ²	Overall <i>P</i> -value	Independent variable	Regression coefficient	<i>P</i> -value
Ergosterol content	0.38	0.006	pH (stress)	0.61	0.001
			Ca ²⁺ (median)	0.53	0.007
Microbial respiration	0.60	< 0.001	Total Al (median)	-0.73	< 0.001
			pH (stress)	0.70	< 0.001
Leaf palatability	0.62	< 0.001	pH (median)	0.76	< 0.001
			Al (mean)	-0.52	0.008
Shredder richness	0.13	0.213	Ca ²⁺ (median)	-0.29	0.154
			pH (median)	-0.08	0.692
Shredder abundance	0.10	0.310	pH (stress)	0.01	0.961
			Total Al (mean)	0.10	0.624
Shredder biomass	0.17	0.125	pH (mean)	0.30	0.144
			Ca ²⁺ (median)	0.06	0.757
<i>G. fossarum</i> abundance	0.48	0.001	Ca ²⁺ (mean)	0.68	< 0.001
			pH (stress)	0.62	0.001
<i>G. fossarum</i> biomass	0.52	< 0.001	Ca ²⁺ (median)	0.70	< 0.001
			pH (stress)	0.68	< 0.001

Gammarus abundance (VIP = 1.37) and microbial respiration (VIP = 0.89), emerged in the final regression model:

leaf breakdown rate (day⁻¹) = -1.5 + 0.65 × *Gammarus* abundance (individuals per bag) + 85 × microbial respiration (mg O₂ g AFDM⁻¹ day⁻¹).

This model explained 84.6% of the variation in leaf breakdown rate, of which *Gammarus* abundance explained 70.1% and microbial respiration the remaining 14.5%. None of the variables added to *Gammarus* abundance and microbial respiration was significant, and a multiple regression including all six variables explained only an additional 4.7% of variation in leaf breakdown rate.

AMONG-YEAR VARIABILITY OF LEAF BREAKDOWN

There were significant differences in breakdown rates of beech leaves among the 3 study years (Fig. 6; two-way ANOVA, *F*_{6,2} = 25.6, *P* < 0.001). This temporal variability was especially pronounced in the circumneutral streams, because of much lower breakdown rates in 1998–99. Qualitatively, however, the overall response pattern was constant among years. The only exception was in the changed rank order among the four acid streams during 1998–99, reflected in the significant interaction between stream and year on leaf breakdown rates (two-way ANOVA, *F*_{6,12} = 8.2, *P* < 0.001). However, these streams still had greatly reduced breakdown compared with the circumneutral streams (i.e. 8.1 ± 1.3 and 13.4 ± 1.5 times lower values in 1998–99 and 1997–98, respectively) and intermediate breakdown rates were consistently observed in MR.

Discussion

EFFECT OF ACIDIFICATION ON LEAF BREAKDOWN

Our results support the contention that stream acidification has a profound impact on litter breakdown, a key process in forested headwater streams. On average, breakdown was 10 times slower in the most acidified streams (pH < 5.0) compared with the sites of pH *c.* 7, which were similar (e.g. LM4 with *k* = 0.0055 day⁻¹) to breakdown rates of beech litter in other unimpacted streams throughout Europe (0.0020–0.0058 day⁻¹, Pattee, Bornard & Mourelatos 1986; 0.0024–0.0073 day⁻¹, Chauvet, Giani & Gessner 1993; 0.0045 day⁻¹, Gessner & Chauvet 1994; 0.0060 day⁻¹, Fleituch 2001).

Although water chemistry of the acidified streams in the Vosges Mountains was variable, PLS confirmed that pH, alkalinity and total Al concentrations were the major variables explaining biological responses to acidification in freshwaters (cf. Townsend, Hildrew & Francis 1983; Rutt, Weatherley & Ormerod 1990; Rosemond *et al.* 1992), including ecosystem processes such as leaf breakdown. The effective prediction of breakdown rates by our regression model with only two variables, and the lack of confounding relationships with other chemical and/or physical stream characteristics, suggests that the ultimate causes of reduced breakdown in acidified freshwaters might be simply related to stream water alkalinity and total Al concentration. Previous field observations suggested that differences in environmental factors other than pH-related variables, particularly temperature and physical abrasion related to stream flow, were small and contributed only marginally to variations in leaf breakdown rate among streams (Dangles & Guérol 2001b).

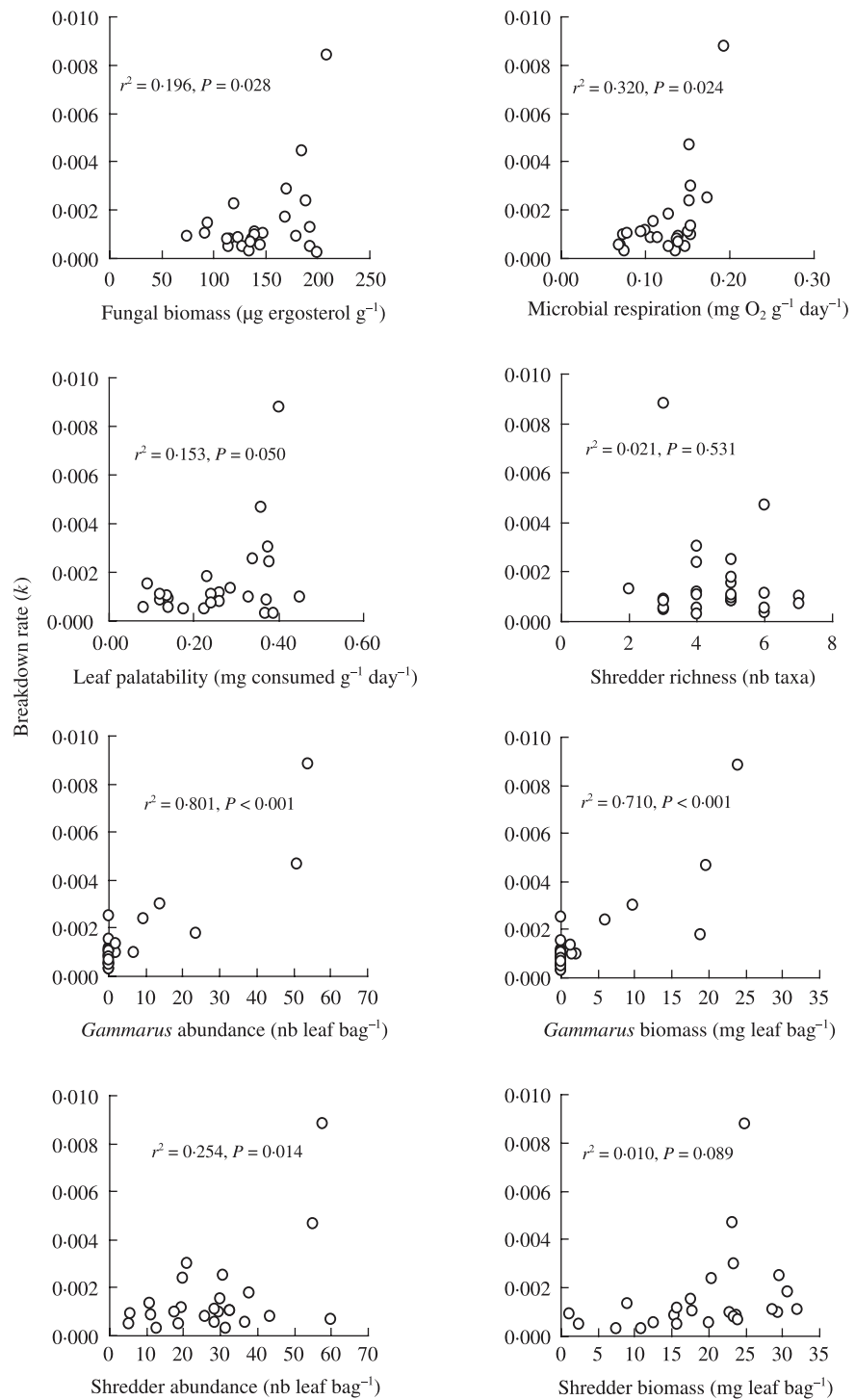


Fig. 5. Simple linear regressions of biological parameters vs. leaf breakdown rate calculated for the first 100 days of leaf breakdown in the streams. Biological variables likewise refer to values determined on leaves after 100 days of stream exposure. All values are means from three replicate leaf bags. nb = number.

In addition, our results indicate that the relation between acidification and leaf breakdown is incremental rather than showing an abrupt shift beyond a certain threshold. Although at variance with several studies conducted to establish critical limits of surface water acidification based on the response of invertebrates or fish (Fjellheim & Raddum 1990; Lien *et al.* 1996), this finding is consistent with the pattern of litter breakdown

found by Niyogi, Lewis & McKnight (2001) in a series of streams affected by acid mine drainage.

EVIDENCE FOR BIOLOGICAL MECHANISMS

Evidence from various studies has indicated that acidification reduces leaf breakdown rates in streams (Mulholland *et al.* 1992; Meegan, Perry & Perry 1996;

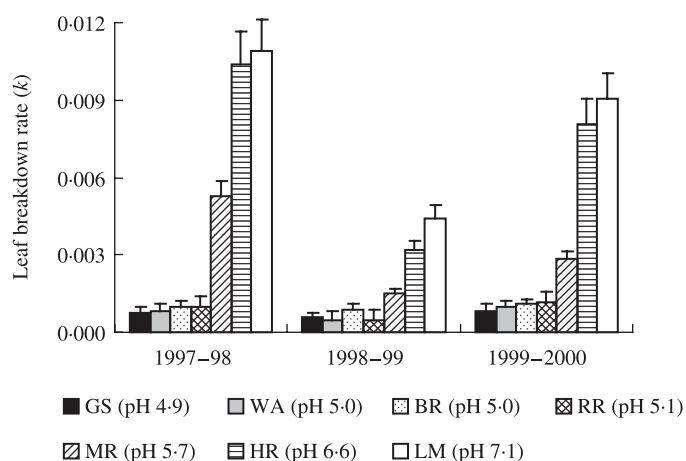


Fig. 6. Comparison of leaf breakdown rates at seven sites in the winter and spring of 1997–98, 1998–99 and 1999–2000. Bars indicate asymptotic standard errors. GS, WA, BR and RR are acidic streams; MR is slightly acidic; HR and LM are circumneutral sites. Values of stream pH are means of 28 measurements made over the 3-year study period.

Dangles & Guérol 1998). It has not been clear, however, whether the effect is mainly due to lower microbial activity or to changes in shredder communities (Maltby 1996). Our results indicate that the decrease in breakdown rates in acidified streams is mediated by both, as several microbial variables and leaf palatability were negatively related to stream acidity. Underlying mechanisms include elevated concentration of H and Al ions that affect ion regulation of shredders such as gammarids and trichopterans (Herrmann *et al.* 1993; Felten & Guérol 2001). Calcium deficiency adds to the pH and Al stress, especially for gastropods and crustaceans (Rosemond *et al.* 1992), and calcium is also instrumental in the activity of fungal pectinases, a key enzyme in leaf degradation in streams (Suberkropp & Klug 1980; Chamier & Dixon 1982).

Rather than showing a gradual response, there appeared to be a threshold of fungal biomass, microbial respiration and leaf palatability below which litter breakdown was consistently low, whereas both low and high values were apparent above the threshold. Similarly, the relationship between breakdown rate and the total abundance or biomass of shredders was only weakly significant, with a threshold effect also noticeable. This lack of a significant relationship may be partly explained by the fact that nemouroid stoneflies (*Leuctra*, *Protonemura* and *Nemoura*; Zwick 2000) dominated the shredder assemblages in the acid streams. Previous studies have indicated that their role in leaf breakdown may be reduced in acidic conditions compared with circumneutral streams (Hieber & Gessner 2002), especially when the shredders feed on recalcitrant leaves such as beech (Groom & Hildrew 1989; Dangles & Guérol 2001a,b; Dangles 2002).

A limited impact of nemouroid stoneflies on beech leaf breakdown in acidic conditions would also explain our observation that species richness of the shredder assemblages was unrelated to breakdown rate. This finding contrasts with laboratory feeding trials in which

deliberate experimental manipulations of shredder richness enhanced leaf breakdown significantly (Jonsson & Malmqvist 2000). Rather than species richness, the abundance and biomass of a single acid-sensitive species, *G. fossarum*, were good predictors of litter breakdown rate; *G. fossarum* was never observed in streams with a mean pH < 6 and their abundance and biomass explained 80% and 73%, respectively, of the variation in breakdown among streams. Thus, differences in breakdown rates among streams appeared to be primarily related to the taxonomic composition of the shredder assemblages, specifically to the presence or absence of a single highly efficient shredder. Tests for effects of large shredders (including mainly large caddisflies in addition to *G. fossarum*) vs. small shredders (comprising primarily nemouroid stoneflies) did not reveal strong relationships with breakdown rate (simple linear regressions $P = 0.33$ and $P = 0.26$, respectively). This result suggests that species identity (specifically the presence of *G. fossarum*) rather than a general species trait such as body size was the main determinant of litter breakdown rates across the acidity gradient. *Gammarus fossarum* may well be a key species in the functioning of the circumneutral beech forest streams we studied, not least because of its high abundance.

Even though biodiversity effects (Loreau *et al.* 2001) were not apparent in the present study, changes in shredder community structure through the loss of a key species caused by acidification had a profound effect on ecosystem functioning. Consequently, the specific sequence of species loss caused by environmental stressors must be taken into account when assessing the consequences of species losses resulting from environmental change (Jonsson *et al.* 2002).

Surface water acidification and recovery from acidification are temporal phenomena that may be only partly reflected by patterns observed across spatial gradients. In addition to the general quandary of basing causal inferences on correlations, this is a potential caveat

of our approach to compare leaf breakdown rates across an acidity gradient if the aim is to make causal inferences about the effects of anthropogenic surface water acidification or recovery. The effect of lowered pH and increased Al concentrations could be tested with acid or Al-addition experiments in a set of currently still sufficiently buffered streams. Experimental simulation of the simultaneous loss of cations, however, is extremely difficult under field conditions. Consequently, even ecosystem-level acidification experiments (Hall *et al.* 1980; Brezonik *et al.* 1986; Ormerod *et al.* 1987) would represent at best only part of the acidification caused by air pollution. This limitation severely restricts the scope for inferences from such experimental results. Taking advantage of differences in stream water chemistry resulting from slight differences in bedrock composition within a region would thus seem to be a valuable, although not conclusive, approach for gaining insight into the mechanisms underlying the effects of stream acidification. As acidified streams recover, it will be possible to test experimentally the prediction derived from the present results: that poorly buffered streams when released from acidification stress show breakdown rates akin to those in circumneutral softwater streams with a higher buffering capacity. The predicted recovery of streams from acidification makes such examinations a real possibility.

SYNTHESIS AND APPLICATIONS

Despite marked reductions in atmospheric S deposition over the past 30 years, many streams remain acidified in several parts of the world (Likens, Driscoll & Buso 1996; Arndt *et al.* 1997). With the naturally low buffering capacity of soils depleted of exchangeable cations, the expected recovery of stream water alkalinity following reduced acid loadings has been delayed in both North America and Europe (Probst *et al.* 1999; Stoddard *et al.* 1999). Liming might not offer a generally feasible symptomatic treatment because effective technical interventions are exceedingly costly (Bengtsson & Bogelius 1995; Dickson & Brodin 1995). Consequently, further reduction in acidifying emissions, especially NO_x, is urgently needed.

For effective and cost-efficient management, such efforts must go hand in hand with appropriate assessment. Networks measuring changes in air quality and atmospheric deposition, e.g. National Atmospheric Deposition Program (NADP) and Clean Air Status and Trends Network (CASTNet), are insufficient alone to assess acidification or recovery in aquatic ecosystems (Driscoll *et al.* 2001). Networks monitoring changes in water quality are currently rare, and networks assessing responses among biota are even more sparse (Driscoll *et al.* 2001). Exceptions are two long-term surveys conducted in Great Britain, which have shown only modest ecological response to both reduced acidifying emissions and liming (Lancaster *et al.* 1996; Bradley & Ormerod 2002). Clearly, integrative, simple functional

assessment methods, like that described here, would be an asset to such monitoring efforts.

Litter breakdown can clearly be a useful measure of ecosystem functioning in acid-sensitive headwaters. Despite notable among-year variation, possibly a large and rather constant reduction in breakdown rates (c. 90%) was observed at the acidic sites in all years. Determining leaf breakdown rate and/or related variables may thus provide an effective, straightforward and relatively low-cost approach to stream assessment. Litter breakdown also allows the simultaneous evaluation of associated processes (e.g. microbial respiration) and patterns (e.g. shredder and fungal biomass and community composition) (Gessner & Chauvet 2002). Such multiparameter approaches (Barbour, Stribling & Karr 1995; Barbour 1997; Karr 1999) not only would facilitate the identification of mechanisms underlying changes in ecosystem functioning, but could augment routine biomonitoring of freshwaters impacted by a range of anthropogenic stressors.

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