# Application of metabolomics to the analysis of ancient organic residues 

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#### Abstract

The grape is arguably one of the oldest cultivated products in human history and the analysis of its main product, wine, reveals clues to trade and associations of previous civilizations. In ancient times, wine was stored in clay amphorae, which, if not properly sealed with resin or pitch allowed the wine to wick into clay matrices, dry, and polymerize producing insoluble, intractable materials that may remain within the matrix for several thousand years. Presently, identification of wine residue is based upon the extraction of these polymeric materials from the ceramic matrix and analyzing/identifying the chemical fingerprints.

Two main biomarkers have historically been employed for the identification of wine residue: tartaric and syringic acids. In some cases, the presence of one of these biomarkers has been designated as the confirmatory signature of wine often leading to false positives as amphorae were re-used in antiquity. Herein, a novel approach utilizing metabolomics has been applied to archaeological objects in order to further mine possible biomarkers for a more accurate assessment of the original foodstuff. An untargeted metabolic profiling method was combined with a targeted analytical method resulting in the successful validation of eight representative biomarkers in two separate archaeological sites.


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## List of Abbreviations

| ANOVA | Analysis of Variance |
| :---: | :---: |
| B.C.E. | Before Christian era |
| B.P. | Before present |
| CV | Coefficient of variation |
| DAG | Diacylglycerol |
| DHB | Dihydrobenzoic |
| DIMS | Direct Infusion (Mass Spectrometry) |
| EIC | Extracted ion chromatogram |
| FT-ICR-MS | Fourier transform ion cyclotron resonance mass spectrometry |
| FTIR | Fourier transform infrared spectroscopy |
| GC-MS | Gas chromatography-mass spectrometry |
| g-log | Generalized - log |
| H | hour |
| HCl | Hydrochloric acid |
| HESI | Heated electrospray ionization source |
| HILIC | Hydrophilic Interaction liquid chromatography |
| HPLC | High Performance Liquid Chromatography |
| IT | Ion trap |
| KCl | Potassium chloride |
| KOH | Potassium hydroxide |
| LC-MS | Liquid chromatography-mass spectrometry |
| LTQ-FT-ICR-MS | Linear ion trap Fourier transform ion cyclotron resonance mass spectrometry |
| ml | Millilitre |


| mm | Millimetre |
| :---: | :---: |
| mM | Millimolar |
| MS | Mass spectrometry |
| MS/MS | Tandem Mass Spectrometry |
| $\mathrm{m} / \mathrm{z}$ | mass/charge |
| N | Normal |
| NCE | Normalized collision energy |
| PPM | Parts per million |
| PSI | Pound-force per square inch |
| QC | quality control |
| RPM | revolutions per minute |
| RSD | Relative standard deviation |
| RT | Retention time |
| S | seconds |
| SDS | Sodium dodecyl sulfate |
| SIM | Selected Ion Monitoring |
| SNR | Signal to Noise Ratio |
| SOP | Standard operating procedure |
| SPE | Solid phase extraction |
| TAG | triacylglycerol |
| TIC | Total ion chromatogram |
| $\mu \mathrm{l}$ | Microlitre |
| v/v | Volume to volume |

## 1. Introduction

In this thesis, a metabolomics approach to analytical testing was applied to the identification of organic residue in archaeological artifacts. The purpose was to develop a suite of biomarkers, in particular, for the identification of aged wine residue. The approach included a global untargeted analysis to model sherds in order to discern indicative biomarkers for materials aged under laboratory conditions and then apply those newly found biomarkers using a highly selective, targeted analytical method to archaeological artifacts. The extraction steps applied in this analysis produced three separate aliquots: polar, non-polar, and an alkaline fusion aliquot. However, an attempt was made to limit sample manipulation in order to increase the sensitivity for trace analyses of ancient organic residues.

The analysis of archaeological residues by chemical testing unofficially began in 1970 with the gas chromatography analysis of bog butter by Thornton, et al. (Thornton, et al., 1970). Bog butter was an ancient form of lard found in the British Isles; the waxy material that remained was analysed for its fatty acid constituents in order to identify the lipid origins. This was the first successful application of analytical chemistry to the study of archaeological materials. The hydrophobic lipid constituents proved a very useful material for analysis, as the fatty substituents were impervious to groundwater leaching and therefore lasted for several millennia, although in many circumstances the chemical fingerprint represented the degraded or altered version of the original. Modern day gas-chromatography in combination with mass spectrometry, as well as gas chromatography/combustion/isotope ratio monitoring mass spectrometry are currently utilised for the determination of the archeological fats in an attempt to identify the original plant or animal sources. The results of these types of analyses answer long standing archaeological questions which include the start of dairying in Britain, the effect of Roman rule on British diets during the Iron Age,
and the prevalence of pork production in the Neolithic era (Copley et al., 2005; Redfern et al., 2012; Mukherjee et al., 2008).

The analysis of fermented beverages in archaeological materials has not been as successful and suffers from the lack of a consistently applied analytical method. In 1993, an article in Analytical Chemistry detailed the analysis of amphorae from an early Mesopotamian society, dating from 3500 B.C. and 2900 B.C. and located in the western part of Iran (Michel et al., 1993). Sherds taken from representative amphorae were extracted in boiling solvent; the residue was analysed by Fourier transform infrared spectroscopy (FTIR) and identified as a calcium tartrate salt, purportedly the altered form of tartaric acid, a common acid found in grape products. The results are questionable since FTIR is a gross analytical technique often used for the initial determination of chemical classes such as resins, waxes, proteins, oils, and fats (Regert, 2011). Also, there is a great risk of confusing the infrared pattern of calcium tartrate with calcium oxalate, a common mineral deposition found on ancient pottery and stone. Therefore, the use of FTIR should be relegated to an initial survey of materials rather than as the definitive descriptor. However, FTIR was again used to determine a tartrate salt and ultimately the presence of wine in pottery dating c.6200-5800 B.C from the Hunan province in China (McGovern et al., 2004).

A substantial improvement in the analytical methodology applied to the analysis of archaeological wine within the past decade was the application of highly sensitive instruments including liquid and gas chromatography combined with mass spectrometry. In 2004, a group of researchers applied reversed phase liquid chromatography in combination with triple quadrupole mass spectrometry to a dried remnant of (presumably) wine from the 18th Dynasty of the Egyptian Kingdom (Guasch-Jane et al., 2004). Tartaric acid and syringic acid were both reported, strongly indicating the presence of wine. However, a separate research group at the University of California was unable to repeat the 2004 published methods and in 2010 published their own methodology for the analysis of
archaeological wine (Barnard et al., 2010). The researchers developed a new LC-MS/MS method using reversed phase liquid chromatography to analyse the samples from artifacts dating to 4100 B.C.E. and ultimately identified syringic acid. Recently, a new analytical method was developed that utilises gas chromatography-mass spectrometry. Using GC-MS, researchers have identified several characteristic acids including tartaric acid indicative of the presence of wine in archaeological plasters (Pecci et al., 2013). Although the method offers an improvement in the number of markers identified, the authors ascribe the origin of tartaric acid to a salt or to the free acid that has been preserved over time. This origin seems unlikely and the preservation of a small organic acid over an archaeological lifetime is more likely due to esterification within a polymeric network. A separate method employing GC-MS utilised a dual solvent extraction with heat on intact sherds to remove tartaric and syringic acids (Koh and Betancourt, 2010). The method was applied to a more recent group of artifacts by the same researcher and the results were nearly $100 \%$ positive for tartaric acid, and only slightly less for syringic acid (Koh et al., 2014). These unusually successful results certainly require further investigation.

Due to the chemical complexity of wine and its multiple constituents, section 1.3 of this introduction discusses the building blocks of wine (Cooper and Marshall, 2001). Based upon the enological studies on the polymerisation of wine over time, the discussion also includes how a polymeric network is formed during aging. It is assumed that a similar polymerisation has occurred in antiquity thereby allowing a polymeric network to reside within a clay matrix for an extended period of time. Sections 1.1 and 1.2 briefly describe the production of wine in ancient time and introduce the two archeological sites explored in this work. Section 1.4 introduces the triacylglycerol analysis of the VIndolanda samples. Section 1.5 briefly discusses the metabolite extraction procedures, with a focus on the alkaline attack utilised to break apart the polymeric network; sections 1.6 and 1.7 discuss the analytical and statistical procedures applied in this research. The final objectives are then described in section 1.8.

### 1.1 Introduction to Ancient Wine

The grape vine (Vitis vinifera) is one the oldest cultivated crops in human history having been domesticated from the original wild vine nearly 8000 years before present (This et al., 2006). Archaeological evidence and genetic determination suggest the domesticated western grape vine originated from the wild type (Vitis vinifera, subspecies sylvestris) in Transcaucasia, spreading to the Mediterranean through the Iberian peninsula and into Europe often transported by the major empires of the day, the Phonecians, the Carthaginians, and the Romans (McGovern, 2003: 39; Myles et al., 2011). Archaeobotanical evidence of a domesticated grape vine was excavated from a Neolithic site in Georgia, south of the modern day capital, Tblisi (McGovern, 2003: 23). Dated to 6000 B.C. by radiocarbon dating, this represents the earliest archaeological evidence of the domesticated grape vine. Later excavations unearthed the oldest (currently) large scale production of wine in neighboring Armenia dated to approximately 4100 B.C.E. (Barnard et al., 2010). At this site were the archaeobotanical remains of grapes including grape vines and grape seeds surrounding a press. This find pre-dates the next oldest site of wine production and consumption, 3150 B.C., from the tomb of Scorpion 1 one of the very earliest Egyptian kings. Grape seeds, skins, and dried pulp were identified in the tomb. The function of wine in these contexts is at present purely speculative and thought to support ceremonial or religious functions.

Due to the time required to nurture the young vines, ancient vineyards were developed in stable societal structures in order to provide the several years necessary for cultivation. The production of wine from those grapes was often organised around substantial infrastructure including immovable plaster bases or limestone monolith wine presses, as well as transport amphora (van Dommelen and Bellard, 2008). When the grapes were harvested, the vineyard workers would crush the grapes underfoot (tread). In the case of a two tier wine press (Figure 1.1), the grapes were crushed on the upper portion of the press allowing the juice to flow through an opening or gutter into the lower
portion in preparation for storage and transport in clay amphorae. Grapes are fruits that are high in sugar, with glucose and fructose accounting for approximately $20-25 \%$ of the berry's weight (Cheynier et al., 2010). Once crushed the uptake of glucose by the yeast Saccharomyces cerevisiae (naturally occurring on the surface of the grapes) initiates a cascade of reactions resulting in the production of ethanol (Ribereau-Gayon et al, 2006: 56).


Figure 1.1. An intact double wine press in Valencia, Spain, active between $5^{\text {th }}$ to $3^{\text {rd }}$ centuries B.C.E. (Pérez Jordà et al., 2011).

Besides the ethanol content of the wine, its organoleptic properties depend upon very reactive molecules originating from the flesh, skin, and seeds of the grape. The monomeric species, once extruded from the grape during crushing, interact chemically with one another to form polymeric compounds. This translates to a change in color, for example, from orange-red to reddish-brown, accompanied by a change in taste (Waterhouse and Kennedy, 2004:144). The resulting polymers develop into an intractable material, often precipitating out of the liquid. It is, therefore, not so
preposterous to theorise that ancient wine may still exist as a long lasting polymeric network stored within the interstices of the amphora's clay matrix. In order to determine the presence or absence of wine in an ancient artifact, the presumed polymer can be broken down chemically thereby releasing the original monomeric species.

### 1.2 Archaeological Samples

Two archaeological sites were chosen for the targeted biomarker analysis. The first site in Sardinia, Italy, experiences a dry climate with sandy soil in an area known for a history of wine production. The actual archaeological site is undisturbed and there were limited vegetal remains beyond carbonised grape pips. This site suggests limited interaction with other foodstuffs such as olives or cereals. The second site at Vindolanda, Northumberland, was a military/civilian encampment with a wet environment and heavy, peat soil. The materials gathered from this excavation included vegetal and animal remains suggesting a site that was inherently multi-use. Sardinia offered the optimum site for the application of this newly developed suite of biomarkers, whereas Vindolanda offered a further test of the robustness in the application of the same set of biomarkers due to the intrinsic nature of a multi-use environment.

### 1.2.1The History of Grape Cultivation and Production of Wine in Sardinia

Sardinia is an island in the Mediterranean well known for its Nuragic civilization, c. 3700 B.C.E. and famous for the prehistoric settlements containing nuraghi, large square towers that dot the island's landscape which remain to this day. The island was overrun several times in the ancient world, by the Phoenicians in the $8^{\text {th }}$ century B.C.E., the Carthaginians in 510 B.C.E., and then the Romans in 238 B.C.E. Sardinia was the 'breadbasket' for these ruling nations, its agricultural economy fueled by the Mediterranean climate with hot and dry summers and an average rainfall of 500 mm . Sandy soil in the southern part of the island is useful for growing grapes, many wild types of which are found on
the island. However, the exponential growth of vineyards determined from archaeobotanical research indicates the greatest evidence of Vitis domestication occurred from the middle Bronze Age to the start of Punic times, 3600-2300 B.C.E., a timeline that overlaps with the influx from Phoenicia (Di Rita and Melis, 2013).

### 1.2.1.1 The Punic Rural Settlement, Terralba, Sardinia



Figure 1.2. The location of the Terralba rural district circled on the map in the southern portion of Oristano Bay, Sardinia, Italy (reproduced from http://members.bib-arch.org/publication.asp?PubID=BSBA\&Volume=16\&Issue=1\&ArticleID=5).

In 2002, the Terralba rural research project began excavating in rural southwestern Sardinia, Figure 1.2, in order to understand the agricultural underpinnings of Punic Sardinia, particularly in relation to the larger Carthaginian economy. Whereas earlier archaeological teams focused on the urban centres created under Phoenician and Roman rule, these excavations were focused on the rural
villages. By the mid $6^{\text {th }}$ century B.C.E., there was an increase in Carthaginian settlements and the number of rural settlements in southwestern Sardinia increased measurably towards the end of the $5^{\text {th }}$ century B.C.E. (van Dommelen and Bellard, 2008). Over several years, the team studied ten rural sites in the western coast of Sardinia, south of the Bay of Oristano.

In the summer of 2007, an area known as Trunce e' Molas, a Punic farmstead active from the $5^{\text {th }}$ to the $2^{\text {nd }}$ century B.C.E. was excavated. The materials collected from the site numbered 14,500 and included iron tools (including sickles), coins, and pottery. Vegetal remains consisted mainly of carbonised grape pips, with some beet root; however, cereals were notably absent from the site. The archaeologists determined three phases of habitation within Trunc e' Molas. Phase I, from $2^{\text {nd }}$ century B.C.E. to the present day, was seriously disrupted by modern ploughing that overturned the soil and damaged stone walls from earlier inhabitation. Phase I also included trenches dug for modern vineyards that had since been pulled. No samples were taken from Phase I because of the disruption of the site and because of its intermingling with modern viticulture. Phase II represented the human inhabitation between the $4^{\text {th }}$ and $2^{\text {nd }}$ century B.C.E. Although there was some intermingling between phase I and II, the lower portions of phase II were considered secure, separated from modern materials. Phase III consisted of layers of sterile sand where no human habitation was found or preserved.

Excavated from Phase II were two large basins thought to be wine presses due to their sheer size as well as the numerous carbonised grape pips surrounding the area, Figure 1.3. The basins were well preserved and untouched by the modern ploughing. The two large ( $0.9 \mathrm{mx1.4} \mathrm{~m}$ ) basins were described as coeval, but with differing methods of construction. One basin was carved from a limestone monolith whereas the second was built with stones and finished with thick layers of plaster and a thinner upper coating of earthenware, referred to as the 'preparatory layer.' A lip around the edge suggests the basins were used to hold liquid. Based upon other archaeological
excavations, grape presses were rare in Punic settlements, particularly when compared to the proliferation of grape presses under Roman rule (van Dommelen and Bellard, 2008). Of the presses that were excavated in Punic establishments, there were three basic types: an exterior platform with a lime and sand mortar coating, lime coating over an interior platform, and a platform carved directly in the rock (Jordà et al., 2011).

The wine press excavated at Trunce e' Molas might represent the lower level of a two-tiered wine press; the upper level is believed not to have survived (personal communication: Prof. P. van Dommelen). Again, the fruit was placed on the upper level of the double wine press where it was crushed by treading. The extracted juiced flowed down into the lower basin, from which it would await transfer into clay amphorae for fermentation or immediate trade. Well-preserved double platforms for a wine press have been found in the Iberian coast at Alt de Benimaquia on Valencia also dating from between the $4^{\text {th }}$ and $2^{\text {nd }}$ century B.C.E. (Jordà et al., 2011).


Figure 1.3. The double collection basins excavated in 2007 by the Terralba rural project (van Dommelen, et al., 2007).

The excavated pottery found in Phase II included domestic ware, pieces of amphorae, and three nearly intact amphorae found in a well. Based upon macroscopic and microscopic examination including petrographic thin sections, two types of pottery fabric were excavated from this site (personal communication, Dr. Helen Loney). The sherds examined in this analysis were defined as Fabric B, a granular calcareous matrix with mica inclusions. The color of the fabric varied from pale brown to yellow to pink. The calcareous material was not local to Terralba, but thought to originate from western central Sardinia near the coast (van Dommelen and Trapichler, 2011).

### 1.2.2 The Vindolanda Settlement

The Vindolanda settlement located in modern day Northumberland, England, was settled in 85 A.D. by Roman soldiers stationed there to guard the Stanegate Road, the de facto frontier line prior to the establishment of Hadrian's Wall 37 years later (Figure 1.4). The settlement grew and receded in size over its 500 year history, with the greatest expansion to 8 acres during the years of 105-120 A.D. (Birley, 2010). Roughly nine Roman military forts have been excavated within the settlement, each fort representing a specific period. Archaeological excavations have also unearthed the remnants of a vibrant civilian encampment along the perimeter of the fort walls that included pottery, glass, metal workings, leather footwear and horse reins, jewelry, coins, and sundry objects representative of daily life.


Figure 1.4. A map of the England and Scotland border, with the location of Vindolanda settlement and Hadrian's wall.

The most famous archaeological finds are the Vindolanda Tablets, the earliest examples of handwritten documentation in Britain. The tablets are pieces of thin wood shavings similar to veneer, written in Latin with carbon based ink (Figure 1.5). Many of the tablets were excavated in 1973-1975 under the direction of Dr. Robin Birley, The Vindolanda Trust, and originate from the 1st and the 2 nd centuries A.D. Translation of the tablets reveal soldiers' letters home, invitations to social gatherings, dinner menus for high ranking officials, as well as general grocery lists.


Figure 1.5. Vindolanda Tablet 203 representing a menu, or grocery list: pork cutlet, bread, wine, olive oil. Reproduced from Vindolanda Tablets online: http://vindolanda.csad.ox.ac.uk/TVII-203.

The landscape around the fort (and then Hadrian's Wall) was open pasture land, suggested by pollen analysis from cored soil samples; the silver birch woodlands were already cleared by the 2500-1500 B.C.E. (early Bronze Age) and land use under the Roman occupation resulted in an increase in the cultivation of rye, Secale cereale (Dark, 2005). A multi-use product, the grain was often used for animal feed and stock supplies for the soldiers, whereas the plant was used for thatch work (Alcock, 2001:18). The main cereal for the Roman soldier was, however, wheat used for making bread (Triticum aestivum). Other foodstuffs included locally sourced meat and dairy from cattle, sheep, goat, and pig. Supplementing the soldiers' diet were provisions brought from Gaul, Italy and Spain, including walnuts, beer, fish sauce, olive oil, and wine (Alcock, 2001: 27-89). The civilians living and working around the fort were the probable providers of the local foodstuffs, through herding and farming on land leased to them by the military (Davies, 1971).

For this thesis, samples were gathered during an active excavation in late spring/early summer 2012 and represent amphorae, mortaria, and cooking vessels. All samples gathered represent Periods VII (212-280A.D.), VIII (300-367A.D.) and represent a mixture of military and civilian environs (Birley, 2010). Area A represents a military fort originally constructed in 213 A.D., rebuilt in the $4^{\text {th }}$ century,
and includes the centurion's apartment as well as general soldiers' barracks. Area B represents the civilian encampment active from 213-300 A.D. All contexts contained within each area that pertain to the samples analysed were described by Kate Sheehan-Finn, the Archaeological and Research Assistant, and are listed in Chapter 6.

### 1.3 Building Blocks of Wine and Polymer Formation

As mentioned earlier, wine is a chemically complex material and is composed of acids, flavonols, anthocyanidins, and tannins. The combinations of these materials maintain the polymeric network described in the following section.

### 1.3.1 Acids

Organic acids found in the must or the body of the grape gives wine its acidity and tartness; malic acid and tartaric acid account for the majority, contributing approximately $90 \%$ of the acidic content (Dinsmore-Webb, 1974:17). Citric, lactic, and succinic acids are also important acidic constituents of wine, with citric originating in the grape must and succinic and lactic acids originating as major byproducts from the fermentation process, Figure 1.6 (Thoukis et al.,1965; Kliewer, 1966).




Succinic Acid



Oxalic Acid
Ascorbic Acid


Glyoxylic Acid


Glutaric Acid $\quad \mathrm{OH}$

alpha Ketoglutaric Acid


Pyruvic Acid

Galacturonic Acid


Quinic Acid


Malonic Acid
(1.6. Representative organic acids found in wine; malic and tartaric acid constitute $90 \%$ of the wine's acidity.

Phenolic acids also originate from the flesh of the grape and are classified as either benzoic acids or cinnamic acids (Figure 1.7 and 1.8); there is a subset of cinnamic acids preferentially esterified with tartaric acid known as hydroxycinnamoyl tartrates (Ribereau-Gayon, 1972:87; Flamini, 2003; Ong and Nagel, 1978). In wine, these tartrates (Figure 1.9) are the most abundant phenolic group and are found in both red and white grapes (Ferrandino et al., 2012).

## Benzoic Acids



Figure 1.7. Carbon skeletons of benzoic acids.
$R=R^{\prime}=H$ : p-hydroxybenzoic acid
$\mathrm{R}=\mathrm{OH}, \mathrm{R}^{\prime}=\mathrm{H}$ : protocatechuic acid
$\mathrm{R}=\mathrm{OCH}_{3}, \mathrm{R}^{\prime}=\mathrm{H}$ : vanillic acid
$\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{OH}$ : gallic acid

$\mathrm{R}=\mathrm{H}$ : salicylic acid
$\mathrm{R}=\mathrm{OH}$ : gentisic acid
$R=R^{\prime}=\mathrm{OCH}_{3}$ : syringic acid

Cinnamic Acids


Figure 1.8. Carbon skeleton of the cinnamic acids.
$R=R^{\prime}=H$ : p-coumaric acid
$\mathrm{R}=\mathrm{OH}, \mathrm{R}^{\prime}=\mathrm{H}$ : caffeic acid
$\mathrm{R}=\mathrm{OCH}_{3}, \mathrm{R}^{\prime}=\mathrm{H}$ : ferulic acid
$\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{OCH}_{3}$ : sinapic acid


Figure 1.9. Example of an esterified cinnamic acid.
$\mathrm{R}=\mathrm{H}$ : mono-p-coumaroyltartaric acid
$\mathrm{R}=\mathrm{OH}$ : monocaffeoyltartaric acid
$\mathrm{R}=\mathrm{OCH}_{3}$ : monoferuloyltartaric acid

### 1.3.2 The Flavonoids

The flavonoids comprise the classes of flavonols, anthocyanidins, and tannins, built around the flavonoid skeleton seen in Figure 1.10. These compounds are found in the grape skin, stems, and seeds and are highly reactive molecules that account for the majority of the organoleptic properties in wine (Fulcrand et al., 1999). During polymerisation, the most reactive sites in the flavonoid backbone are the nucleophilic centres at carbon 6 (C-6) and carbon 8 (C-8), as well as reactive electrophilic centres at carbon 4 (C-4) (Figure 1.10).


Figure 1.10. The flavonoid skeleton, the backbone for the classes of compounds including flavonols, flavanols, anthocyanidins, and tannins. The carbon number is designated on the skeleton.

Flavonols, located in the grape skin, are often glycosylated on carbon 3 and include: kaempferol, quercitin, isohamnetin, myricetin, laricitrin, and syringetin (Figure 1.11). All six compounds are identified in the red grape, whereas only the first three compounds are found in the white grape (Ferrandino et al., 2012).


Figure 1.11. Backbone of a flavonol molecule.

Anthocyanidins are the main colorant for red grapes and are concentrated in grape skin. The five recognised compounds are: delphinidin, malvidin, petunidin, cyanidin, and paeonidin (Figure 1.12).

They are often glycosylated at C-3 and sometimes acylated with organic or cinnamic acids as seen in

Figure 1.13 (Saucier, 2010; Fong et al., 1971). Diglucoside (carbons 3 and 5) are found in modern Vitis hybrids (Dinsmore-Webb, 1974:69).

delphinidin, $3^{\prime}=5^{\prime}=\mathrm{OH}$
malvidin, $3^{\prime}=5^{\prime}=\mathrm{OCH} 3$
petunidin, $3^{\prime}=\mathrm{OCH} 3,5^{\prime}=\mathrm{OH}$
cyanidin, $3^{\prime}=\mathrm{OH}, 5^{\prime}=\mathrm{H}$
paeonidin, $3^{\prime}=\mathrm{OCH} 3,5^{\prime}=\mathrm{H}$

Figure 1.12. The carbon skeleton of the anthocyanidins the main colorants in the skin of red grapes, often glycosylated at carbon 3. The molecule has reactive nucleophilic centres at carbon 6 and at carbon 8.


Figure 1.13. Malvidin-3-monoglucoside-p-coumarate.

Tannins are highly reactive species originating from the grape skin, stems, and the seeds and are categorised as flavan-3-ols (catechins) and flavan-3,4-diols (leucoanthocyanidins), Figure 1.14
(Dinsmore-Webb, 1974:63).

## Flavan-3-ol


$\mathrm{R}=\mathrm{OH}, \mathrm{R}^{\prime}=\mathrm{H}$ : catechin
$\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{OH}$ : gallocatechin

Flavan-3,4-diol

$R=R^{\prime}=H$ : leucopelargonidin $\mathrm{R}=\mathrm{OH}, \mathrm{R}^{\prime}=\mathrm{H}$ : leucocyanidin $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{OH}$ : leucodelphindin

Figure 1.14. Two representative tannin structures.

### 1.3.3 Formation of Polymers in Wine

As shown, the building blocks of wine are highly reactive species that polymerise over time resulting in a change of flavor and color. One research group suggests this highly reactive nature is due to the lack of substitution found on the tannins and anthocyanidins, describing those molecules found in grapes to be the 'simplest encountered in higher plants' (Brouillard et al., 2003). For example, anthocyanins found in morning glory, Pharbitis nil, contain multiple glycosylated and cinnomylated residue side-chains resulting in fairly stable molecules. In contrast, the anthocyanins of the Vitis species are limited mainly to glycosylation, with rare exceptions of extending side chains, leaving open reactive nucleophilic and electrophilic centres to interact with similar centres in flavanol and tannin molecules.

The mechanisms of how wines age due to molecular polymerisation is an active branch of study in the field of enology. One polymerising mechanism was suggested by Timberlake and Bridle (1976) in which a model solution was prepared from a pure anthocyanin, a pure flavan-3-ol, and acetaldehyde (the acetaldehyde arising from at least two origins: fermentation and oxidation of ethanol). It was theorised that reactions formed a polymeric species from all three original components (Timberlake and Bridle, 1976). Since then, with the availability of more sensitive analytical techniques, researchers confirmed this mechanism, identifying dimers between catechins and anthocyanin, linked via an ethyl bridge (Es-Safi et al. 1999). Figure 1.15 is based upon the 1976 proposed mechanism. Here, the protonated acetaldehyde, which is prevalent in the acidic environment (A), attacks the C-8 nucleophilic centre of a flavan-3-ol (B). There is a loss of water, which provokes another attack on the C-8 of the anthocyanin (C), thereby joining the flavan-3-ol and the anthocyanin via an ethyl bridge.




Figure 1.15. The acetaldehyde, prevalent in the acidic environment, attacks the C-8 nucleophilic centre of a flavan-3-ol. A loss of water provokes another attack on the C-8 of the anthocyanin.

A second proposed mechanism involves the direct condensation between tannins and tannins, and tannins and anthocyanins (Jurd, 1969; Haslam, 1980; Singleton and Trousdale, 1992; Somers, 1971). The suggested initiator of this polymerisation is the electrophilic C-4; the result of a positive charge formed in acidic solution seen here in Figure 1.16.


Figure 1.16. Formation of carbonium ion on C-4 from flavan-3,4-ol in acidic solution.

Two types of polymeric formation are suggested: an A-T (anthocyanin-tannin) polymer formed from anthocyanin electrophilic C-4 linked to C-6 or C-8 nucleophilic centres of the condensed tannins, or a T-A polymer (tannin-anthocyanin) whereby the electrophilic C-4 from the tannin dimer links to the nucleophilic centre ( $\mathrm{C}-6$ or $\mathrm{C}-8$ ) of the anthocyanin. The first analytical proof of this polymeric condensation resulting from direct anthocyanin-tannin reactions was reported in 2000 (Remy, et al., 2000). Three years later an octamer formed by a proanthocyanidin heptamer and an anthocyanin was confirmed (Hayasaka and Kennedy, 2003). Figure 1.17 illustrates the formation of an A-T polymer. Initially two tannin molecules condense to form a dimer, followed by a linkage between the electrophilic C-4 on the malvidin molecule and the C-8 nucleophilic centre of the tannin dimer.



Figure 1.17. Formation of the A-T polymer.

Figure 1.18 was reproduced from the literature and illustrates a hypothetical wine polymer formed during aging; the illustration was based upon compounds identified after depolymerisation of red wine pomace (Wollmann and Hofmann, 2013). One of the compounds identified in the depolymerisation was tartaric acid. As one of the two main biomarkers in the analysis of archaeological wine, the origin of tartaric acid has been consistently attributed to either a salt, calcium tartrate, or to the free tartaric acid having remained due to an arid environment (Michel, et al., 1993; McGovern et al., 2004; 2009; 2013; Romanus et al., 2009; Guasch-Jane et al., 2004; Pecci et al., 2013). From the results presented in this thesis in Chapters 3, 5, and 6, it is far more likely that the tartaric acid was locked within a wine polymeric network over the course of an archaeological timeframe, and released only upon chemical attack of the sample.


Figure 1.18. Hypothesised wine polymer reproduced from Wollmann and Hofmann (2013).

### 1.4 Archaeological Triacylglycerols

There has been a voluminous amount of research devoted to the analysis of triacylglycerols (TAGs). The importance of these compounds lay in the fact that proper characterisation of the identified TAGs offers insight into the original animal or plant lipid used in the archaeological context. There have been multiple historical records written about cooking with animal fat and wine for Roman British soldiers and a portion of this research examined the presence of TAGs in the Vindolanda samples (Renfrew, 1985: 38-40; Alcock, 2010: 83). TAGs are abundant lipids consisting of three fatty acids of varying saturation esterified to a glycerol backbone, Figure 1.19. Prevalent in fats and oils, archaeological TAGs have been classified based upon their carbon distribution by high temperature
gas chromatography mass spectrometry resulting in specific profiles that distinguish between plant and animal fats, ruminant and non-ruminant fats, as well as the difference between adipose fat and dairy fats (Charters et al., 1995; Evershed et al., 1997; Dudd and Evershed, 1998). As an example, non-ruminant fats are classified based upon a narrow carbon distribution $\left(\mathrm{C}_{48}-\mathrm{C}_{54}\right)$ whereas ruminant fats and/or dairy have been classified with a broad range from $\mathrm{C}_{40} / 42-\mathrm{C}_{54}$ (Regert, 2011). Recently, increased sensitivity and greater structural information in the analysis of archaeological TAGs has been accomplished with nanoelectrospray combined with high resolution mass spectrometry (Garnier et al., 2009; Mirabaud et al., 2007). TAGs are labile compounds, however, and several studies have identified the modes of degradation within a laboratory context, ultimately assisting in the identification of the degraded archaeological TAGs (Dudd et al., 1998). Under oxic conditions, TAGs hydrolyse to their mono-, diacylglycerol and free acid components; however, under anoxic conditions, that hydrolysis slows considerably (Evershed, 2008a). It was presumed that TAG residue representative of foodstuffs within cook pots would still be present in the artifacts excavated from the anoxic waterlogged conditions of Vindolanda.


Figure 1.19. A general description for triacylglycerol; the black nomenclature denotes the glycerol backbone and the red nomenclature denotes the attached fatty acids.

In this thesis, lithiated TAGs were examined and fragmented for analysis by high resolution mass spectrometry. For the fatty acid descriptions of TAGs, the neutral losses were compared with published proposed fragmentation mechanisms (Hsu and Turk, 2010). Comparisons were also made
between experimental results and published results of lithiated archaeological TAGs (Garnier et al., 2009; Mirabaud et al., 2007).

### 1.5 Metabolite Extraction

Any good analytical scheme requires a uniform and reproducible extraction process so that accurate comparisons can be made between analytical runs as well as between experiments separated by time and different laboratories (Lin et al., 2007). For this research, samples were homogenised using the Bligh-Dyer method, a commonly used liquid-liquid extraction approach which utilises a methanol/water/chloroform mixture to separate polar and non-polar metabolites into separate phases (Bligh and Dyer, 1959). Those aliquots were removed and a strong alkaline solution was added to the remaining solid which contained the intractable polymer. Alkaline fusion followed by an acidic wash hydrolysed the ester bonds within the polymer releasing monomeric species which were then extracted for later analysis as seen in Figure 1.20. Alkaline fusion is a common method for the analysis of non-extractable phenolics in foodstuffs, and has also been applied to the analysis of ancient wine residue (Perez-Jimenez and Torres, 2011; Wollmann and Hofmann, 2013; GuaschJane et al., 2004; Pecci et al., 2013).

To concentrate low-level metabolites, a pre-concentration step was sometimes necessary prior to analysis. This was accomplished by solid phase extraction (SPE), using a silica or polymer based cartridge often functionalised with hydrophobic, anionic or cationic groups. The extracted sample was poured over the cartridge and the analytes of interest were bonded to the functionalised groups by a variety of interactions, or the analyte of interest washed through the cartridge whilst the interfering matrices remained bonded to the cartridge (Dettmer et al., 2007). SPE was used in several instances during this research. A strong cation exchange cartridge was used to clean up the alkaline fusion protocol in order to remove aluminate ions, a result of matrix interference; this protocol is described in more detail in Chapter 5. Also, an aminopropyl cartridge was used to focus
and concentrate low level TAGs found in the Vindolanda samples. This approach is described in more detail in Chapter 6.


Figure 1.20. Variable species of anthocyanidins based upon pH . The flavylium cation predominates at acidic $\mathrm{pH}<3$. Increasing the pH to 4-5 and the carbinol pseudobase predominates, with an increasing level of chalcone formation. At pH above 8, chalcone species convert to the unstable alpha-diketone via a base catalysed keto-enol tautomerism, which breaks down to form the subsequent aldehyde and carboxylic acid (reproduced from Brouillard et al., 1989).

### 1.6 Analytical Techniques

There are a number of analytical techniques utilised for metabolomics experiments (Dunn and Ellis, 2005). Rather than provide an overview of available instrumentation, only the techniques that were integral to this thesis are included and concern sample introduction, sample ionisation, and analyte detection: direct infusion and liquid chromatography, electrospray ionisation, and mass spectrometry, specifically the Fourier transform mass spectrometer and the triple quadrupole mass spectrometer. Initially, for an untargeted analysis, high throughput is required; this is accomplished with direct infusion nanoelectrospray. Combining direct infusion with high resolution mass spectrometry (MS), in this case Fourier transform MS, offers the most accurate masses in combination with an untargeted metabolite description. In order to increase sensitivity and focus on the most characteristic metabolites, a targeted sensitive approach is then utilised, typically by combining liquid chromatography with a triple quadrupole mass spectrometer.

### 1.6.1 Sample Introduction

### 1.6.1.1 Direct Infusion Nanoelectrospray

Samples introduced into the mass spectrometer via direct infusion electrospray (DIMS) are infused continuously without chromatographic separation. Coupling DIMS with high resolution instruments allows accurate mass to be determined for hundreds to thousands of analytes. However, the lack of chromatographic separation, combined with ion suppression due to salts and matrix interferences sometimes masks compounds of interest.

The first nanoelectrospray was described in the mid 1990's whereby analytes in solution were delivered via an electrically conductive capillary (1-2 $\mu \mathrm{m}$ diameter) under a very low flow rate of
$20 \mathrm{nl} / \mathrm{min}$ (Wilm and Mann, 1996). Each sample was loaded directly into a unique capillary tip which prevented (or certainly limited) cross contamination between samples. In 2000, a microchip electrospray device was developed whereby 400 nozzles at 20 micrometres in diameter were etched into a silicon wafer to produce a nanoelectrospray chip (Schultz et al., 2000). In this instance, an electrically conductive carbon coated tip picks up several microlitres of sample, connecting with one of the nozzles creating a current and thereby an ionised electrospray. The advent of this device allowed high-throughput analysis of hundreds of samples in only a few hours of analysis time and was utilised in this research.

### 1.6.1.2 Liquid Chromatography

Liquid chromatography (LC) is an analytical technique often utilised in biological applications whereby a complex mixture of compounds are separated based upon each compound's partitioning between a column stationary phase and the solvents (mobile phase) polarity, Figure 1.21. The mobile phase composition is either maintained (isocratic) or changed throughout the run (gradient elution). Because temperatures in the column compartment are often kept between $25-40^{\circ} \mathrm{C}, \mathrm{LC}$ is useful for compounds that easily degrade at high temperature or are difficult to volatilise without chemical modifications, thereby making them incompatible with gas chromatography.


Figure 1.21. A schematic of a liquid chromatograph: the sample is introduced via the injection valve and carried through to the column with a prescribed solvent mixture. The compounds in the sample separate and elute from the column to be detected.

The ability to resolve two adjacent compounds is critical for a successful LC experiment and from Equation 1.1, relies upon the interplay of three factors: capacity factor $k^{\prime}$, selectivity $\alpha$, and efficiency $N$. The capacity factor is a compound's retention time on the column in reference to the void volume, that volume of completely unretained compounds. A higher value represents a compound that is well retained on the column and a lower value represents a compound that elutes earlier. The greatest effect on the capacity factor is the strength of the mobile phase composition. The separation selectivity is described as the ratio of two capacity factors and is often maneuvered by changing the composition or polarity of the mobile phase.

$$
R=\left(\frac{1}{4}\right)(\alpha-1) \sqrt{N}\left[\frac{k^{\prime}}{1+k^{\prime}}\right]
$$

Equation 1.1. The parameters that are necesary for proper resolution in an LC experiment include: $\alpha$, the selectivity factor, the ability to separate two eluting analytes; $N$, the efficienicy of a column to successfully separate a chosen group of analytes; $\boldsymbol{k}^{\prime}$ the capacity factor, or an analytes' retention time in relation to an unretained material.

### 1.6.1.2.1 Hydrophilic Interaction Liquid Chromatography (HILIC)

Earlier analyses of archaeological wine residue relied upon reversed phase liquid chromatography for the separation of low molecular weight polar compounds (Guasch-Jane et al., 2004; Barnard et al. 2010; McGovern et al., 2009). Since reversed phase chromatography is used primarily for the separation of hydrophobic materials, polar compounds elute quickly off the column. The result is an analyte of interest eluting near the unretained void volume which may lead to irreproducible or questionable results. Chapter 4 of this thesis describes a HILIC elution protocol developed to increase the capacity factors and retention times of polar compounds on the column.

The term 'HILIC' (Hydrophilic Interaction Liquid Chromatography) was coined in 1990 to describe a type of chromatography primarily based on two factors: the elution of hydrophilic analytes from a
stationary polar phase into an increasingly polar mobile phase and the electrostatic effects between analytes and the stationary phase (Alpert, 1990). The initial mobile phase composition consists of a high percentage ( $>80 \%$ ) of organic solvent, which is preferred for electrospray ionisation (Nguyen and Schug, 2008). The chromatographic mechanism of HILIC is still not completely elucidated (Buszewsky and Noga, 2012; Dinh et al., 2011; Gama et al., 2012). However, the following parameters are considered integral to the elution procedure: partition of the analyte between the mainly organic mobile phase and an aqueous enriched stationary phase, adsorption between a charged analyte and the charged surface of the stationary phase, ion exchange between the charged analytes, the charged surface of the stationary phase, and the buffer salts, and size exclusion between analytes and the stationary phase (Jandera, 2011; Buszewsky and Noga, 2012).

### 1.6.2 Sample Ionisation

### 1.6.2.1 Electrospray Ionisation

In terms of sample ionisation, electrospray ionisation (ESI) describes a 'soft ionisation' technique whereby analytes are ionised with minimal fragmentation allowing detection of the deprotonated (in negative ion mode analyses) molecular ion. This is an advantage over the harsher methods such as electron impact ionisation where multiple mass fragments are produced from one ion. Originally designed by Dole in 1968 and later perfected by Fenn in the 1980's, ESI revolutionised biomolecular analysis and earned John Fenn a Nobel Prize in Chemistry in 2002 (Dole et al., 1968; Fenn et al.,1989; Fenn, 2003).


Figure 1.22. Reproduced from (Cech and Enke, 2001), image of a Taylor cone forming at the tip of a positively charged capillary needle. The smaller and highly charged 'offspring' droplets of smaller radius are attracted to the counter electrode and enter the mass spectrometer.

Figure 1.22 offers a schematic description of the ESI methodology. The analyte in solution travels down a metal capillary tube to which a voltage has been applied. The formation of a potential gradient at the capillary tip results in charge separation between the positive and the negative ions forming a Taylor cone; this misshapen liquid is a result of the opposing forces between the highly charged ions' attraction for the counter-electrode at the mass spectrometer entrance and the surface tension of the liquid.

$$
E_{c}=2\left(\frac{V_{C}}{r_{C}}\right) \ln \left(\frac{4 d}{r_{C}}\right)
$$

[^0] radius of the capillary; $d$, the distance from the tip of the capillary to the counter-electrode.

The electric field formed at the tip of the capillary $\left(E_{c}\right)$ is described by Equation 1.2 where $V_{C}$ is the voltage applied to the capillary, $r_{C}$ is the radius of the capillary, and $d$ is the distance to the counterelectrode. Smaller and more highly charged droplets break from the larger cone when the electrostatic repulsion exceeds the surface tension (the Comloumbic repulsion). This process continues while the droplets become smaller and smaller until the highly charged, smallest droplet enters the gas phase.

Two models describe this desolvation phenomenon. The charged residue model states that Coloumbic repulsion occurs continuously until each droplet represents one analyte ion (Dole et al., 1968). The second model, the Ion evaporation model, states that the fully desolvated ions evaporate from the highly charged surface, overcoming the surface tension of the liquid (Iribarne and Thomson, 1976). However, both references agree that the transfer from the charged droplet to a gas phase ion is more efficient with a small droplet radius as shown by equation 1.3:

$$
R \propto\left(\rho V_{f} 2 \gamma\right) 1 / 2
$$

Equation 1.3 A charged droplets radius is proportional to three conditions of the encasing solvent: the flow rate, the solvent's density and its surface tension.

Where $V_{f}$ is the flow rate, $\rho$ is the density of the solvent, and $\gamma$ is the solvent surface tension. The droplet formation and desolvation are aided by the addition of an inert gas such as nitrogen, known as sheath gas, as well as the use of an organic solvent with low surface tension and high volatility.

### 1.6.3 Analyte Detection

### 1.6.3.1 High Mass Accuracy Mass Spectrometry

Two of the most important parameters in mass spectrometry are mass resolution, the ability to resolve each peak in a spectrum and mass accuracy. Equation 1.4 is the definition of resolution based upon $m$ the mass of the peak, and $\Delta m$ the difference in $m / z$ at $50 \%$ of the peak's height, also known as full width at half maximum (FWHM).

$$
R=\frac{m}{\Delta m}
$$

Equation 1.1 One way to explain resolution of a peak in a mass spectrum is to take the mass of the interested peak and divide by the width of that peak at $50 \%$ height, where $\boldsymbol{m}$ is actually the mass to charge value

Mass resolution is critically linked to accurate mass measurement as is seen in Figure 1.23. In this illustration, the resolution is increased from 15,000 up to 80,000 thereby resolving the peaks such that the experimental mass to charge of the peak is within a 0.3 ppm mass accuracy. The mass accuracy is the difference between the actual mass and the experimental value; an accurate mass is critical in order to limit the number of possible empirical formulae which can be attributed to that compound.


Figure 1.23. High mass resolution is critical for the mass accuracy. The top spectrum taken at a resolution of 15,000 yielded a $\mathrm{m} / \mathrm{z}$ peak at mass $\mathbf{2 3 9 . 1 5 1 8 1}$. The bottom spectrum collected at a higher resolution at 80,000 yielded a more accurate mass of $\mathbf{2 3 9 . 1 5 0 3 3}$ at 0.3 ppm mass error (reproduced from Scigelova et al., 2011).

### 1.6.3.1.1 Fourier Transform Ion Cyclotron Resonance Mass Spectrometry (FT-ICR-MS)

Developed in 1974 by Alan Marshall and Melvin Comisarow, the FT-ICR-MS is currently the mass spectrometer of choice for high mass accuracy of less than 1ppm as well as extremely high resolution of $1 e^{6}$ thus allowing accurate description of extremely complex samples (Brown et al., 2005). The detection component of the FT-ICR-MS is the ion cyclotron resonance cell, seen in Figure 1.24. Kept at high vacuum ( $10^{-9} \mathrm{mbar}$ ), the cell is shielded and surrounded by a magnet of varying strength, 1-12 Tesla, based upon the instrument. The ions introduced into the cell are traveling at a specific velocity in a homogeneous magnetic field and subjected to the Lorentz force, a force acting perpendicular to the travel of the ion path. This force pushes the ions in a trajectory which begins a circular motion maintaining a constant distance from the centre of the cell whilst being electrostatically captured within the cell by opposing trapping plates (Comisarow, 1985; Marshall et al., 1988).


Figure 1.24. Schematic of an ICR cell. In this diagram, the ions traveling in a homogeneous magnetic field are subjected to the Lorentz force, a force acting perpendicular to the travel of the ion path. This force causes the ions to begin a circular trajectory around the centre of the cell, trapped within ICR cell by potentials from the trapping plates. Ions are detected once their radii have increased sufficiently to cause an 'imprint" on the trapping plates. Reproduced from http://www.chm.bris.ac.uk/ms/theory/fticr-massspec.html.

Equation 1.5 describes the cyclotron frequency of the ions:

$$
w_{c}=\frac{q B}{m}
$$

Equation 1.2 An ions cylcotron frequency $\left(w_{c}\right)$ is dependent upon: $q$, the charge of the ion; $B$, the strenfth of the magnetic field in Tesla; $m$, the mass of the ion.

Where $w_{c}$ is the ion cyclotron frequency, $q$ is the charge of the ion, $B$ is the strength of the magnetic field (Tesla), and $m$ is the mass of the ion. From this equation, the frequency of a moving ion must be calculated in order to determine its mass. For this, a pulse of energy is applied to the cell increasing the ions' trajectory radii until the ions are close enough to imprint an 'image current' or transients upon the detector plates. Since every one of the ions' cyclotron frequencies are detected at once, a Fourier transform is needed in order to deconvolute the transient signal from the time domain to the frequency domain, which is then used to produce a mass spectrum.

The high mass accuracy of the FT-ICR-MS limits the number of possible empirical formulae that can be attributed to a mass feature thus assisting in compound identification. One drawback with detection by FT-ICR-MS that may affect resolution and therefore mass accuracy is space charging effects caused by Coulombic interactions between the ion packets within the ICR cell (Dienes et al., 1996; Uechi and Dunbar, 1992). One way to prevent space charging is to limit the packet of ions delivered to the ICR cell. This can be achieved by combining the FT-ICR-MS with a separate mass spectrometer which gauges the amount of ions, such as the linear ion trap mass spectrometer.
1.6.3.1.2 Hybrid Instruments: linear ion trap Fourier transform ion cyclotron resonance mass spectrometry

The high resolution instrument used for this research was the hybrid linear ion trap Fourier transform ion cyclotron resonance mass spectrometer (LTQ-FT-ICR-MS), Figure 1.25. The LTQ is a linear ion trap which can act as a stand-alone instrument and is comprised of four rods with the two opposite rods having the same polarity, Figure 1.26. The linear ion trap operates by trapping ions radially within the quadrupole radiofrequency field and axially by the application of an electrostatic field in the lenses found at the front and back end of the trap. The trapped ions are focused in the centre of the trap by the addition of an inert gas such as helium. This effectively 'dampens' the kinetic energy of the ions and focuses the packet of ions within the $z$, or axial, direction.

Within this trap ions are isolated, fragmented (by collisional induced dissociation with the helium gas), and ejected for immediate detection or for transfer into the ICR cell. Ions are ejected either axially out the back end of the trap, or radially along the slits in the centre section allowing for greater ejection efficiency. One of the most important features is the automatic gain control (AGC) value (Finnigan ${ }^{\text {TM }}$ LTQ FT $^{\text {TM }}$ Hardware Manual, 2005). This user defined value determines the amount of ions ejected into the ICR cell, thereby preventing space charging and decreased mass resolution.


Figure 1.25. A schematic diagram of a hybrid LTQ-FT-ICR-MS. Ions are directed from the source (a) through the tube lens and skimmer (b), continuing through (c) Q0 (quadrupole) and (d) Q1 (octopole) into the (e) linear ion trap. A predetermined packet of ions are ejected from the linear ion trap through the octopoles (f) for detection in the ICR cell (g). The ICR cell is shielded on either side by a cryogenically cooled magnet. The ions travel through an atmosphere of increasing vacuum due to the process of differential pumping (h). Diagram reproduced from Finnegan LTQ FT hardware manual.


Figure 1.26. A representation of the linear ion trap's quadrupole assembly. Reproduced from the Finnigan ${ }^{\text {TM }}$ LTQ $_{\text {FT }}{ }^{\text {TM }}$ Hardware Manual.

### 1.6.3.2 Triple Quadrupole Mass Spectrometry

For targeted analysis of trace materials in this thesis, the mass detector of choice was the triple quadrupole mass spectrometer. The underlying principle of operation is based upon the four hyperbolic rods found within each quadrupole. The two rods opposite one another are held at the same potential. Alternating direct current and radio frequency ( $\mathrm{dc} / \mathrm{rf}$ ) voltages applied to the quadrupoles allow chosen $m / z$ to follow a stable trajectory to the exit, whereas all other $m / z$ will follow an unstable trajectory, careen into the sides of the poles and be neutralized, Figure 1.27. The instrument was initially described by Wolfgang Paul in 1953 for which he received the Nobel Prize in 1989 (Paul, 1990).


Figure 1.27. An image of four rods within a quadrupole; the red directional arrow represents a successful transmission of ions.

The triple quadrupole mass spectrometer, considered a low/medium resolution instrument, was introduced in 1978 for the isolation and fragmentation of chosen ions using tandem quadrupoles in sequence, Figure 1.28 (Yost and Enke, 1978). Quadrupoles 1 and 3 are mass filters, directing particular masses by varying dc/rf. Quadrupole 2 has two functions: a collisional cell used to fragment the precursor ion by bombardment with a stable gas such as argon, and as an ion guide that focuses and transmits the product ions to quadrupole 3. (Douglas, 2009).


Figure 1.28. A schematic of a triple quadrupole mass spectrometer where (a) ions enter the first quadrupole and mass selectively filtered for transfer into the collision cell (b). The fragmented ions are then filtered through the third quadrupole.

The triple quadrupole instrument can be run in several modes including:

- SIM/scan mode where a precursor ion is selected in Q1, fragmented in Q2 and Q3 scans for all product ions.
- Scan/SIM mode where Q1 is scanned across the whole mass range, all ions are fragmented in Q2 and only the chosen product ions are monitored and detected in Q3.
- neutral scan mode where Q1 is scanned across the chosen mass ranges, the precursor ions are fragmented in Q2, and the product ions are scanned across the chosen mass ranges in Q3 offset by the mass of a neutral loss.
- multiple reaction monitoring (MRM) where Q1 is set to a specific precursor ion that is fragmented in Q2 and Q3 is set to the specific product ions.

MRM mode was chosen for this research in order to achieve greater specificity and sensitivity for trace analysis of compounds in a complex sample matrix. This targeted method isolates and analyses only user defined specific ions, thus increasing sensitivity by isolating a chosen set of masses, as opposed to scanning the whole mass range. The MRM method was developed with pure standards dissolved in a solvent mixture most accurately representing the samples' elution profile off a column. In this way, the proper and most significant fragmentations were recorded and applied to the final method.

### 1.7 Data Analysis

### 1.7.1 SIMStitch Method

Since the examination of small molecules in a sample requires the identity of high as well as low abundance ions, the dynamic range of the analytical methodology is crucial. For LTQ-FT-ICR-MS, the manufacturers list a dynamic range of 5000 ; that is, at a maximum signal of $100 \%$, there is a minimum signal intensity of $.02 \%$ (LTQ FT hardware manual, 2004; Payne, 2011). In order to maximise this dynamic range, a methodological approach was utilised which allows identification of low abundance ions without increasing the amount of sample introduced into the ICR cell thus maintaining high resolution and mass accuracy, SIMstitch method (Payne et al., 2009; Southam et al., 2007). This approach to spectral processing was utilised for all untargeted MS analyses collected by DIMS-LTQ-FT-ICR-MS in this thesis.

For the SIMStitch method, a chosen mass range is analysed by collecting several small scan windows in a series of specified mass ranges. Only those masses of that particular window are analysed in the ICR cell at one time, thus preventing the deleterious effects of space charging. Transient files are then collected and averaged for each window; for example, for a window of $100 \mathrm{~m} / \mathrm{z}$ mass range, 24 transients were collected and averaged (Southam et al., 2007). A Fourier transformation is then applied to the averaged transients and the overlapped regions of all windows are "stitched" together with a unique algorithm written in the Matlab (The Mathworks, Natick, Massachusetts) environment resulting in a complete mass spectrum.

For the research undertaken in this thesis, each sample was run in triplicate or quadruplicate and in order to eliminate noise artifacts and false peaks, a three stage filtering step was applied. After a SNR was chosen, a $2 / 3$ or 2/4 replicate filter was chosen whereby the peak must be present in two out of the three (or four) replicate samples in order to be determined as 'real'. Finally, a sample
filtering step was applied whereby the percentage chosen was applied to a sample group such that a particular peak, (e.g., over a certain SNR and found in $2 / 3$ replicates) was also found in $x \%$ of samples analysed.

### 1.7.2 Matrix Processing

Data gathered in a metabolomics experiment are often treated as a matrix, with a row for each sample and a column for each of the features or variables such as $m / z$ values. The intensities of each variable are listed in the corresponding locations. The data stored within this matrix are then 'preprocessed' in preparation for statistical analysis including: normalisation, imputing missing values, and g-log transformation (a variable scaling function). Analysed along with individual samples are data from quality control samples, or QCs. These samples are a representative mixture of all individual samples to be analysed and are evenly distributed throughout the analyses in order to identify any technical drift during an analytical run (Dunn et al., 2012).

For this thesis, the data matrix was normalised in order to correct for the technical variation or sample dilution that may occur during analysis (Dieterie et al., 2006). For this, a reference spectrum was prepared from the QCs by calculating a median value of all descriptive features. Each sample spectrum was then normalised against the reference spectrum by dividing the features in the sample spectrum by the same features in the reference spectrum, resulting in a normalised data matrix. Due to the thousands of features that were often found in a DIMS experiment, there were invariably missing values in the original data matrix, that is, features that were identified in one sample set were not evident in a separate sample set. The missing values were imputed by several methods including K-Nearest Neighbors (KNN), or in some cases, by returning to the original mass spectrum to determine the noise peak (Hrydziuszko and Viant, 2011).

In order to limit systemic variability and to raise the relative importance of sample variability, the normalised data matrix was g-log transformed resulting in the attenuation of the most intense peaks while amplifying many of the less intense peaks. The result was a matrix with peaks that were of more similar mean intensity and variation. This transformation removed the biases that occur in multivariate statistics, including principal components analyses (which focus on the most varying peaks).

### 1.7.3 Statistical Analyses

Broadly speaking, the statistics applied to metabolomics data are defined as either univariate, determining the behavior of one variable throughout an experiment, or multivariate, following the changes of many variables using a single test. Univariate tests utilised for this thesis included the student's t-test and 1-way ANOVA. In order to determine significance amongst the tested variables when multiple variables were tested, the Benjamini-Hochberg correction was applied such that the user defined false discovery rate (FDR) was multiplied by the ( $p$-value/total number of features); the resultant adjusted p-value was a more robust descriptor for significance (Benjamini and Hochberg, 1995). Those compounds determined as significant were then examined for their fold change, that is, the difference in intensity amongst timed events.

Multivariate testing in this thesis consisted of principal components analysis (PCA), an unsupervised method of describing multivariate data in reduced dimensions by combining thousands of features (e.g., $m / z$ ) for each sample and reducing them into a few principal components (PC). The PCs are vectors in a ' $n$ ' dimensional space that collectively identify the greatest variation amongst samples. PC1 is plotted between the greatest variability; PC2 is orthogonal to PC1 and contains the next greatest amount of variation, with each subsequent PC orthogonal to its previous one until the total variability was described. A 'scores plot' was plotted between top ranking PCs, e.g., PC 1 and PC2, to visualise natural clustering in the data. These scores represent how well each sample spectrum was
represented or described by each PC. When patterns and clusters were observed in these reduced dimensionality plots, it was meaningful to look at the 'loadings' for each PC.

Other terms in statistics used in this thesis include the basic description of how well data points agree with one another in a particular measurement or experiment:

1. Mean: sum of measurements divided by the number of measurements.
2. Median: all measurements are collated in descending order and the middle value is chosen as the median value. In the case of an even number of values, the mean value of the two middle values is chosen as the median.
3. Standard Deviation: The standard deviation measures the spread of data from the measurements' mean,


Equation 1.3 Equation of standard deviation where: $\sigma$ represents the standard deviation, $i=$ total number of measurements, $x$ represents specific data measurements, and $\bar{x}$ is the mean value for all data measurements in experiment.
4. Coefficient of Variation: CV describes the variation of the data with regards to the mean of the total measurements,

$$
C V=\frac{\sigma}{\bar{x}}
$$

Equation 1.4 The coefficient of variation as described by the standard deviation, $\sigma$, divided by the mean of all data measurements, $\bar{x}$.
5. Relative Standard Deviation: RSD describes the coefficient of variation in terms of a percentage,

$$
R S D=100 * \frac{\sigma}{\bar{x}}
$$

Equation 1.8 The relative standard of deviation as described by the standard deviation, $\sigma$, divided by the mean of all data measurements, $\bar{x}$, multiplied by 100 .

### 1.7.4 Metabolite Identification

Oftentimes, the identification of a significant $m / z$ is the 'bottleneck' in a metabolomics study. In order to prepare a targeted method from the differentiated metabolites, the identity of those original metabolites ( $\mathrm{m} / \mathrm{z}$ ) must be confirmed. There are four levels of confidence (or criteria) in the identification of a metabolite starting at an unknown, to a putative identification of a compound class, to a putative identification of a compound, leading to a definitive identification. The definitive identification is based upon the comparison between the metabolite in question and an actual standard with at least two orthogonal properties such as retention time, fragmentation pattern, and $m / z$ (Dunn, et al. 2013).

### 1.8 Objectives

The aims of this research are to apply a metabolomics strategy to the analysis of organic residue in an archaeological object. Specific aims were addressed in the following chapters:

1. In Chapter 3, to utilise an untargeted metabolomics approach to laboratory aged wine and determine a suite of biomarkers that are most discerning for aged wine.
2. In Chapter 4, to develop a targeted LC-MS/MS method based upon those biomarkers in order to increase the sensitivity for trace level analysis.
3. In Chapter 5, to apply the targeted method to archaeological samples from Sardinia, Italy, in order to determine the presence/absence of ancient polymerised wine.
4. In Chapter 6, to apply that targeted method for wine residue to a multi-use site, in addition to utilise bioinformatics in the analysis and fragmentations of TAGs.

## 2. Materials and Methods

The methods presented here describe sample collection, sample extraction, and instrumental
analysis. Specific description of the wine aging study is discussed in Chapter 3. Descriptions of the ageing of foodstuffs are in Appendix, Chapter 6. Data processing is introduced in this chapter and more fully explained for each specific experiment in the following chapters.

### 2.1 Materials

### 2.1.1 Chemicals

Purchased from Fluka:
Vanillic acid 94770-10g
Ammonium acetate 17636-50g
Ammonium hydroxide >=25\% in water, 09857-100ml

## Purchased from Sigma-Aldrich:

Malic acid 240176-50g
2-hydroxycinnamic acid, predominantly trans H22809-5g
4-hydroxyphenylpyruvic acid 114286-1g
2,3-dihydroxybenzoic acid 126209-5g
Trans-ferulic acid $126708-5 \mathrm{~g}$
Citramalic acid 5415375-250mg
2-ketobutyric acid K401-5g
Malonic acid 01236-5g
3 -hydroxybenzoic acid 120000-5g
L tartaric acid 281530-5g
I-ascorbic acid A5960-25g
Syringic acid S6681-5g
Succinic acid S-7501-100g
Potassium hydroxide 221473-500g
p-coumaric acid C9008-1g
2-isopropyl malic acid $333115-10 \mathrm{mg}$
Oxaloacetic acid sigma 04126-1g
Ethyl acetate Sigma Aldrich 650528-1L
Hydrochloric acid, 37\% 320331-500ML

## Purchased from Acros organics:

m-hydroxybenzoic acid 14755-02-03(CAS)
Lithium chloride 199880050

## Purchased from Fischer Scientific:

Citric acid C/6200/53-500g
Chloroform C/4966/17-2.5L. 1149533.
Acetonitrile HPLC grade A/0627/17-2.5L

## Purchased from Scientific Laboratory Supplies:

Gallic acid 91215-100mg

## Purchased from J.T. Baker:

Water HPLC grade 4218-1L
Methanol LC-MS reagent 9822-1L

### 2.1.2 Miscellaneous

## SPE cartridges:

Phenomenex strata-x-c-33u polymeric strong cation. $30 \mathrm{mg} / 1 \mathrm{ml}$ part number: 8B-S029-TAK 100/box. Sorbent lot: S229-109
Supelco discovery DSC-NH2, 1ml tubes, part number: 52636-U, lot number: 2297701
Thermo Scientific C18 100/ml, part number: 60108-302

## Champagne vials:

Cronus, part\#: VZM-1508CC-100. Vial Champagne clear 8mm narrow mouth, screw top 1.5 ml . lot number 1314932Ca08.
Cronus, part\#: VCA-0804TB-100. Assembled black cap w/sil/PTFE septa B-425. lot number 130490-11. 100/pk

Poulten and Graf, disposable glass Pasteur pipettes 150 mm , 250/box.
1.8 ml glass vials with aluminum lined caps

Sarstedt 1000ul pipette tips lot 0220/3190011. 250/bag
Fisher brand $200 \mu \mathrm{l}$ pipette tips. 1000/bag.
Sarstedt 1.5 ml eppendrof tubes lot 0000/2069001
Whatman pH indicator strips catalog number 2613991. 100 strips
Glass wool Merck, K93007686
Precellys 7ml hard tissue homogenizing tubes. Product \#: CK28-7ml. Catalog number: 0904-01.
Precellys 2 ml hard grinding tubes. Product number MK28 R. Catalog number 03961-1-008.
Pyrex 15 ml glass centrifuge tubes, lot number: 07413036
96-well sample plate, Abgene, Epsom, UK
Fisher Scientific 500ml glass jar with cover, BTF-625-110K

### 2.1.3 Model Wine Samples

For the laboratory ageing experiments, model wines were taken from Three Choir Vineyards in Gloucestershire, UK. The wines were selected to meet the criteria that they were from a known source, were fermented, but had not been aged. The white wine was pressed from the Seyval Blanc grape. Seyval Blanc is a modern hybrid grape whereby the Vitis vinifera, European grape vine, is
crossed with a North American species in order to increase tolerance to cold weather and disease. Any of the North American Vitis species could be involved with its genetic description, including: riparia, labrusca, aestivalis, rupestris. This wine was fermented in a cold tank for three weeks. The fermentation was complete, based upon residual sugar analysis performed by an employee at Three Choirs Vineyards. Seven aliquots of this white wine were collected into clean glass bottles, each of 400 ml . The red grape varietal was Triomphe d'Alsace, another modern hybrid consisting of three Vitis species: vinifera, riparia, and rupestris. The grapes were pressed, left in a warm room at $25^{\circ} \mathrm{C}$ for 5 days before being transferred to a storage tank for a 4 week fermentation period. Three aliquots of this red wine were collected into clean glass bottles, each of 400 ml . All bottles were stored at $-80^{\circ} \mathrm{C}$ until further analysis.

### 2.1.4 Model Sherds

Model sherds were prepared by Graham Taylor, Experimental Archaeologist and Pottery Specialist in Northumberland, UK. The sherds were meant to mimic ancient amphorae and consisted of Devon ball clay (70\%), red Etruria marl (10\%), coquet river sand (15\%), and Ande site grit (5\%). The clay was fired electrically to a temperature of $1050^{\circ} \mathrm{C}$, estimating the temperature at which a Roman kiln was fired. The sherds were then hand sawn to a dimension of 3 cm in length by 1 cm in width.

### 2.1.5 Archaeological Samples from Sardinia, Italy

Two sections of the wine press, TM.003.3 and TM.027.3, as well as a separate portion of the preparatory layer from TM.003.3, were received from Professor Peter van Dommelen, University of Glasgow in 2012. Fifteen sherds representing transport amphorae were collected from the University of Leicester in January, 2013 and include representative pieces from the neck, the side wall, and the base. Sample identifications are given in Chapter 5. There were no soil samples accompanying the sherds. The wine press was chosen as a 'positive' control for archaeological wine.

Concerning the amphorae, the neck samples were originally chosen as the 'negative' control, whereas the base portions were considered separate 'positive' controls.

### 2.1.6 Archaeological Samples from Vindolanda, Northumberland, England

Seventy samples were chosen from those collected by volunteer excavators in June, 2012, during an active excavation under the direction of staff archaeologists. The samples represent mortaria, transport amphorae, cooking pots, soil, and storage jars, from various contexts within the site. All context descriptions were described by Kate Sheehan-Finn, Archaeological and Research Assistant of the Vindolanda Trust and are compiled in chapter 6.

### 2.1.7 Modern Foodstuffs

Representative food samples common to Roman British cooking were weighed out and aged for three months (lipids) or six months (bulk foodstuffs) at $40^{\circ} \mathrm{C}$ before extraction and analysis. The foodstuffs were purchased at local markets and include: clear honey, spelt flour, barley malt (barley originating from Australia), and three dried herbs: thyme, oregano, and basil. Foods purchased for the lipid aging experiment included: bacon back, goat cheese, sheep cheese, cows milk $4 \%$ fat, olive oil, and sardines packed in water. High temperature cooking experiments were not performed with these samples.

### 2.2 Methods

Attention was given to cleanliness in order to limit contamination. All glassware including Pyrex glass bottles holding the liquid chromatography mobile phase were autoclaved, solvent rinsed, and covered with methanol cleaned aluminum foil prior to use.

### 2.2.1 Laboratory Aging Study: Model Sherds

For the wine/water permeated model sherds, approximately 350 ml of HPLC grade water were added to a 500 ml glass jar containing 12 randomly selected model pottery sherds. A low flow of nitrogen gas was added for thirty seconds in order to produce an inert atmosphere. The jar was covered tightly with a threaded cap, secured with parafilm, and allowed to sit at room temperature for seven days. The procedure was repeated for white wine and for red wine.

After the seven day period, the model sherds were removed from the liquid and allowed to air dry on methanol-cleaned glass surfaces. There was heavy residue visible on the white wine sherds and the red wine sherds were heavily stained. After drying overnight, tweezers were used to transfer 4 randomly chosen sherds to clean glass jars that were either empty, or contained 2 cm of dry sand or 2 cm of wet peat. Within the 'dry sand' jars, a further 2 cm of sand were added to cover the sherds. Within the 'wet peat' jars, a further 2 cm of wet peat were added to cover the sherds. The final labeled jars were: water/empty, water/sand, water/peat, red/empty, red/sand, red/peat, white/empty, white/sand, and white/peat. The different environments imitated a dry sandy soil and a heavy, wet peat soil; the plain glass jars were utilised as baseline environments. All jars were placed in the ageing oven at $40^{\circ} \mathrm{C}$. After six months, the samples were removed from the oven and any heavy debris from sand or peat was removed. The samples were wrapped in aluminum foil (methanol rinsed) and stored in the $-80^{\circ} \mathrm{C}$ freezer until analysis.

### 2.2.2 Extraction Method: Model Sherds

A tile cutter was used to remove 100 mg of laboratory aged sherd; the tile cutter was rinsed with methanol between samplings. The sample was weighed on an analytical balance and the weight recorded; the sample was then transferred to a 2 ml Precellys hard grinding tissue tube. The amount of solvent to be added to the sample for a biphasic solution is based on the following ratio in full:
methanol:chloroform:water, 2:2:1.8 (v/v) (Bligh and Dyer, 1959). Four hundred microliters of methanol and $160 \mu$ l of water were added to the sample Precellys tube. The sample tube was homogenised in the Precellys 24 homogeniser at $2 \times 6800 \mathrm{rpm} / 30 \mathrm{~s}$. A glass pipette was used to transfer the slurry into a 1.8 ml glass vial; the Precellys tubes were then discarded. Four hundred microlitres of chloroform were added to the glass vial with a cleaned Hamilton syringe. Two hundred microlitres of water were then added with a 1 ml pipette. The sample was vortexed for 20 seconds and allowed to sit at room temperature for five min. The glass vials were then centrifuged for 10 min at 4000 rpm at $20^{\circ} \mathrm{C}$ and then transferred to a fume hood. Four hundred microlitres of the top polar layer were transferred to a 1.5 ml Eppendorf tube with a Hamilton syringe. The syringe was rinsed three times with methanol between samples. The polar aliquots were dried down in a Speedvac with no heat and stored at $-80^{\circ} \mathrm{C}$ until analysis. The non-polar fraction was transferred to a 1.8 ml glass vial and blown down to dryness under a stream of nitrogen, capped and stored at $-80^{\circ}$ C until analysis.

The original sample containing the pottery residue was air dried overnight in the fume hood. One hundred fifty microlitres of 4 M potassium hydroxide $(\mathrm{KOH})$ were then added to the glass vial which was vortexed for 30 s and then centrifuged at 6000 rpm for 2 min (Zsuga and Kiss, 1987; GuaschJane et al., 2004). The vial was then heated at $50^{\circ} \mathrm{C}$ for 10 min on a heating block and then cooled to room temperature. The sample was slowly acidified with $300 \mu \mathrm{l}$ of 2 N HCl , the pH was gauged with pH paper. Four hundred fifty microlitres of ethyl acetate were then added to the glass vial and the vial was vortexed for 30 s and then centrifuged at 4000 rpm for 10 min at $18^{\circ} \mathrm{C}$. Approximately $400 \mu \mathrm{l}$ of the organic extract was transferred into a 1.5 ml champagne vial and then blown down to dryness with nitrogen. The vials were capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

### 2.2.3 Extraction Method: Archaeological Sherds

With the side of the drill bit, the top 2 mm surface from the sherd was removed and discarded. A tile cutter was used to remove 1 g of sample sherd. The sample was weighed on an analytical balance, the weight recorded, and the sample transferred to a 7 ml Precellys hard tissue homogenising tube. The biphasic extraction was very similar to that which was described in Section 2.2.2, differing in the increased volume of solvents. Four millilitres of methanol and 1.6 ml of water in full were added to the sample Precellys tube. The sample tube was homogenised in the Precellys 24 homogeniser at $2 \times 6800 \mathrm{rpm} / 30 \mathrm{~s}$. A glass pipette was used to transfer the slurry and small particles into a 15 ml glass centrifuge tube. Two millilitres of water were added to the Precellys tube to dislodge any remaining particles and added to the centrifuge tube. The Precellys tube was then discarded. Four millilitres of chloroform were added with a graduated cylinder. The centrifuge tube was vortexed for 10 s and then centrifuged for one minute at 500 rpm , room temperature. The polar layer was removed and prepared for storage/analysis as above. The non-polar layer was removed and prepared for solid phase extraction. The remaining solid residue was allowed to air dry in the fume hood overnight.

For solid phase extraction, the non-polar layer was dried down to approximately $400 \mu \mathrm{l}$ under a stream of nitrogen. Otherwise, the non-polar fraction was dried down under a stream of nitrogen, capped and stored at $-80^{\circ} \mathrm{C}$. In preparation for SPE, the sample was removed from the freezer and reconstituted in $400 \mu$ l of chloroform. The aminopropyl cartridge was added to an SPE manifold and conditioned with 2 ml of hexane. The non-polar aliquot was added to the cartridge. One point four millilitres of ethyl acetate:hexane, 15:85 (v/v), were added to the cartridge to elute off the neutral TAGs which were collected, dried down and stored at $-80^{\circ} \mathrm{C}$ until analysis (Bodennec et al., 2000; Kaluzny et al., 1985).

For alkaline fusion, $600 \mu \mathrm{l}$ of 1 M KOH were added to the dried residue. The tube was vortexed for 10 s and heated at $50^{\circ} \mathrm{C}$ for 1 h in a water bath. The sample was allowed to cool to room temperature and the sample acidified by slowly adding 2.5 ml of 2 N HCl . The pH was gauged with pH paper. The sample was then allowed to sit for 24 h at room temperature. In preparation for solid phase extraction, the strong cation exchange cartridge was added to an SPE manifold and conditioned with 1 ml methanol followed by 1 ml HPLC water, per manufacturer's recommendation. A new 15 ml centrifuge tube was added to the SPE tank in order to catch the column runoff. The acidified sample was loaded onto the column and a vacuum drawn. The tube containing the run off was then extracted with 3 ml of ethyl acetate. The tube was vortexed for 10 s and centrifuged for 1 min at 500 rpm . The top organic layer was continuously transferred to a 1.5 ml champagne vial for nitrogen blow down. The vial was then capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

### 2.2.4 Sample Analysis: HILIC/LC-Triple quadrupole analysis

The final elution parameters for the liquid chromatography are shown in Table 2.1. The identity of each transition between parent and product ion with its collision energy was transferred to the targeted method, as shown in Table 2.2. These transitions between parent and product ions were determined by choosing the most intense fragment ions. The transitions were also limited to the elution time from the HILIC column, such that the final targeted method contained four segments: segment 1 from 0-17 min collected in full scan mode, segment 2 collected from 17-23 min SRM mode for 8 transitions of the early eluting acids, segment 3 from 23-33 min for the transitions of the later eluting acids, and segment 4 analysed at full scan from 33-43 min.

New mobile phases were prepared each time in clean, solvent rinsed 500 ml Pyrex glass bottles. Mobile phase B was prepared as 100 mM ammonium acetate in HPLC water. In a fume hood, several drops of ammonium hydroxide were added to the solution in order to raise the pH to above 8. The pH was gauged with pH paper and confirmed with a pH meter.

Table 2.1 Final elution parameters of the zic-pHILIC LC method used for the targeted analysis of the archaeological samples.

| Column | SeQuant zic-pHILIC column, 150x2.1, $5 \mu \mathrm{~m}$ |
| :---: | :---: |
| Mobile Phase A | Acetonitrile |
| Mobile Phase B | 100 mM ammonium acetate, $\mathrm{pH}=8.2$ |
| Mobile Phase C | Water |
| Temperature | $30^{\circ} \mathrm{C}$ |
| Sample Volume | $1 \mu \mathrm{l}$ |
| Step 1 | Time 0-7 minute 10\%B, 0\%C |
| Step 2 | Time: 7-21 minutes 10\%B, 50\%C |
| Step 3 | Time: 21-29 minutes 10\%B, 50\%C |
| Step 4 | Time 29-33 minutes 10\%B, 0\%C |
| Step 5 | Time: 33-35 minutes 10\%B, 0\%C |
| Flow Rate | $70 \mu \mathrm{l} / \mathrm{minute}$ |
| Detector | TSQ mass spectrometer |
| Electrospray voltage | -3600 |
| Vaporisation temperature | 0 |
| Sheath gas | 10 |
| Auxillary gas | 0 |
| Capillary temperature | 275 |

Table 2.2. The final parent-product ion transitions used for the analysis of the archaeological samples.

| Segment 1: | $\begin{array}{\|c\|} 0-17 \\ \text { minutes; full } \\ \text { scan mode } \end{array}$ |  |  | Segment 3: | 23-33 <br> minutes; <br> SRM mode |  |  | Segment 4: | 33-43 minutes; full scan mode |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Segment 2: | 17-23 <br> minutes; <br> SRM mode |  |  | Parent lon | Product Ion | Parent lon | Product Ion |  |  |
| Parent lon | Product Ion | Parent lon | Product Ion | isopropylmalic acid |  | tartaric acid |  |  |  |
| ferulic acid |  | meta-hydroxy acid |  | 175 | 85 | 149 | 73.1 |  |  |
| 193.1 | 134.1 | 162.9 | 91 | 175 | 113 | 149 | 87.1 |  |  |
| 193.1 | 149.1 | 162.9 | 93 | 175 | 115 | 149 | 103.1 |  |  |
| 193.1 | 178.1 | 162.9 | 119 | ascorbic acid |  | citric acid |  |  |  |
| syringic acid |  | para-coumaric acid |  | 175 | 70.8 | 191.1 | 85.1 |  |  |
| 197.1 | 95.1 | 162.9 | 93 | 175 | 86.9 | 191.1 | 87.1 |  |  |
| 197.1 | 123.1 | 162.9 | 117 | 175 | 114.9 | 191.1 | 111.1 |  |  |
| 197.1 | 167.1 | 162.9 | 119 |  |  | 191.1 | 173.1 |  |  |
| 197.1 | 182.1 |  |  | succinic acid |  | gallic acid |  |  |  |
| 2,3-DHB acid |  | vanillic acid |  | 117.1 | 73.1 | 168.9 | 78.9 |  |  |
| 152.8 | 80.9 | 167.1 | 108.1 | 117.1 | 99.1 | 168.9 | 125 |  |  |
| 152.8 | 90.5 | 167.1 | 123.1 | malonic acid |  | citramalic acid |  |  |  |
| 152.8 | 108 | 167.1 | 152.1 | 102.7 | 40.6 | 146.9 | 57 |  |  |
| 152.8 | 109 |  |  | 102.7 | 58.6 | 146.9 | 85 |  |  |
| gentisic acid |  | keto-butyric acid |  |  |  | 146.9 | 87 |  |  |
| 152.9 | 52.7 | 100.7 | 56.7 | malic acid |  | 146.9 | 129 |  |  |
| 152.9 | 80.9 | 100.7 | 73.1 | 133 | 71.1 |  |  |  |  |
| 152.9 | 108 |  |  | 133 | 73.1 |  |  |  |  |
|  |  |  |  | 133 | 115.1 |  |  |  |  |

### 2.2.5 Sample Analysis: DIMS-LTQ-FT-ICR-MS

Initially, an aliquot of 10 mM ammonium acetate in methanol/water ( $80 / 20$ ) ( $\mathrm{v} / \mathrm{v}$ ), was used to achieve a stable negative ion mode electrospray using a nanoelectrospray chip. Once a stable spray was achieved, the linear ion trap was tuned using a mixture of Thermo Scientific Pierce ESI Negative Ion Calibration Solution, Figure 2.1, epigallocatechin ( $\mathrm{m} / \mathrm{z} 305$ ), and epigallocatechin gallate ( $\mathrm{m} / \mathrm{z} 457$ ).


Figure 2.1. A mass spectrum of Thermo Scientific Pierce ESI Negative Ion Calibration Solution, a mixture of SDS ( $\mathbf{m} / \mathbf{z} \mathbf{2 6 5 \text { ), sodium }}$ taurocholate ( $\mathrm{m} / \mathrm{z} 514$ ) and Ultramark 1621 ( $\mathrm{m} / \mathrm{z} 980-1986$ ), reproduced from http://www.piercenet.com/product/esi-negative-ion-calibration-solution.

Five separate tune files were collected over the following $\mathrm{m} / \mathrm{z}$ ranges: 135-450, 420-730, 700-1010, $980-1030$, and 1400-2000, tuning on the largest mass within the middle of the window. Each tune file was then added to the instrument method, with one tune often utilised for several SIM windows.

### 2.2.5.1 Liquid Wine Aliquot Analysis

The liquid wine aliquots were prepared for analysis as described in Chapter 3 ; the $m / z$ range analysed was 1352000. The DIMS conditions for the sample delivery included a flow rate of approximately $200 \mathrm{nl} /$ minute with a 0.5 psi gas pressure. Samples were run in the negative ion mode at 1.7 kV ; sample volume was $5 \mu \mathrm{l}$. For the mass spectrometer, the resolution was maintained at 100,000 (measured from $\mathrm{m} / \mathrm{z} 400$ ) with an AGC target value of $1 \times 10^{6}$. Each SIM window acquisition was 0.3 s and the total run time was 6.5 min .

### 2.2.5.2 Sherd and Foodstuff Analysis; polar and alkaline fusion aliquot

Sherd samples were prepared for analysis as described in Chapter 3. For the polar sherds, the $m / z$ range analysed was 70-2000. The DIMS conditions for these samples were $5 \mu$ l sample volume, run with a gas pressure of 0.5 psi at a negative voltage of 1.75 kV . Resolution was maintained in the mass spectrometer at $100,000 \mathrm{at} \mathrm{m} / \mathrm{z} 400$ with an AGC target value of $1 \times 10^{6}$. Each SIM window acquisition was 0.3 s for a total run time of 6.5 min .

The alkaline fusion aliquots were analysed over a $m / z$ of $70-800$. The DIMS conditions for these samples were $5 \mu \mathrm{l}$ sample volume, run with a gas pressure of 0.5 psi at a negative voltage of 1.75 kV . Again, resolution was maintained in the mass spectrometer at 100,000 at $\mathrm{m} / \mathrm{z} 400$ with an AGC target value of $1 \times 10^{6}$. Ten SIM windows were collected. Each SIM window acquisition was 0.25 s for a total run time of 3.0 min .

### 2.2.5.3 Lithiated TAG Analysis

The non-polar aliquots were prepared for analysis as described in Section 2.2.3 and in Chapter 6. Briefly, the extracted samples were reconstituted in 2 mM lithium chloride ( LiCl ) in chloroform:methanol 1:4 (v/v). Ten microlitres of sample were added to the sample well; $8 \mu \mathrm{l}$ of sample were collected followed by $2 \mu \mathrm{l}$ of air aspirated into the nozzle. The sample was delivered for 15 min and 10 s at a gas pressure of 0.9 psi and a positive
voltage of 1.5 kV . The mass spectrometer was run in a data dependent mode for $\mathrm{MS}^{n}$ fragmentation. After collecting in full scan mode for 1 min , an initial scan was analysed in $\mathrm{MS}^{1}$. The most intense ion was then chosen for isolation, fragmentation $\left(\mathrm{MS}^{2}\right)$ and identification. The three most intense ions identified in the $\mathrm{MS}^{2}$ scan were then isolated, fragmented $\left(\mathrm{MS}^{3}\right)$ and identified in the linear ion trap. The ion initially fragmented in the full scan mode was put on an exclusion list for the remainder of the analysis thus preventing the ion from being refragmented during the analysis. This process cycled through the entire spectrum until the completion of the run. The scan event details are given in Table 2.3.

```
1: FTMS. res=100000 m/z (600.0-1000.0)
2: FTMS. res=100000 Dep MS/MS Most intense ion from scan even }
    Activation Type: CID
    Min.Signal Required: 500.0
    Isolation Width: 1.00 Da
    Normalised Coll. Energy: 35.0 eV
    Default Charge State: 1
    Activation Q: 0.250
    Activation Time: }\quad\mathbf{30.000 ms
3: ITMS. Dep MS 3}\mp@subsup{}{}{3}\mathrm{ Most intense ion from scan event 2
    Activation Type: CID
    Min. Signal Required: 500.0
    Isolation Width: 1.00 Da
    Normalised Coll. Energy: 35.0 eV
    Default Charge State: 1
    Activation Q: 0.250
    Activation Time: }\quad\mathbf{30.000 ms
4: ITMS. MS }\mp@subsup{}{}{\mathbf{3}}2\mathrm{ nd most intense ion from scan event 2
```

Activation Type: ..... CID
Min. Signal Required: ..... 500.0
Isolation Width: 1.00 Da

```Normalized Coll. Energy: \(\mathbf{3 5 . 0} \mathbf{~ e V}\)
    Default Charge State: 1
    Activation Q: 0.250
    Activation Time: }\quad\mathbf{30.000 ms
5: ITMS. MS 3rd most intense ion from }
```

Activation Type: ..... CID
Min. Signal Required: $\mathbf{5 0 0 . 0}$

```Isolation Width: 1.00 Da
```

Normalised Coll. Energy: 35.0 eV
Default Charge State: 1
Activation Q: 0.250

```Activation Time: \(\quad 30.000 \mathrm{~ms}\)
```


### 2.3 Instrument Settings

### 2.3.1 LTQ-FT-ICR-MS

Prior to analysis, a tune was performed on the front end of the instrument as mentioned above. If there was a noticeable drop in intensity, the transfer tube was removed and solvent cleaned.

### 2.3.1.1 Weekly External Calibration

A weekly calibration was performed by laboratory personnel. The automated procedure calibrates and corrects the voltages for ion transmission between the LTQ and the FT-ICR, as well as performs the mass calibration within the ICR cell determining the mass accuracy.

### 2.3.2 Triple Quadrupole Mass Spectrometer

Prior to analysis, a resolution check was performed on the instrument (described below). If the resolution failed due to mass shift, a calibration of the instrument was performed. Failure due to loss of intensity required cleaning of the S-lens, the main ion transmission device from the ion transfer tube into the first quadrupole, Figure 2.2.


Figure 2.1 A schematic drawing of the S-lens showing its spaced electrodes. Changing the S-lens voltages during tuning procedures allows for the maximum amount of ions to be transmitted to the first quadrupole (reproduced from TSQ Series Hardware Manual, Thermo Scientific, March 2009).

### 2.3.2.1 Resolution Check of the Mass Spectrometer

A portion of the Pierce Triple Quadrupole Calibration Solution (Thermo Scientific product number 88325) was directly infused into the instrument. Conditions including capillary temperature and sheath gas were varied in order to maintain an optimal spray stability, that is, the point where the RSD of the total ion current does not vary more than $2 \%$. For negative ion analysis, the peak at $\mathrm{m} / \mathrm{z} 506$ ak for polytyrosine was monitored. In profile mode, the peak at half maximum must lie between a certain defined criteria for quadrupole 1 and quadrupole 3. Mass accuracy was determined in centroid mode and should lie within $.05 \mathrm{~m} / \mathrm{z}$. A successful resolution check for all parameters in the negative ion mode was essential before attempting an analytical run.

### 2.4 Data-Pre-processing

All univariate statistical scripts, normalisation and g-log transformations protocols were written by Dr. Rob Davidson in the Matlab environment. The missing values script was written by Dr. Olga Hrydziuszko also written in the Matlab environment. Compound identification was searched using the MI-Pack program written by Dr. Ralf Weber in the Python environment. The TAG fragmentation algorithm was also written by Dr. Ralf Weber in the python environment.

### 2.4.1 Matrix Processing

Spectral requirements for DIMS data processing are described in Chapter 1 and specifically for each experiment in Chapter 3. In general, the collected mass spectra were 'binned' into SIM windows of $100 \mathrm{~m} / \mathrm{z}$. The edges of each window were 'stitched' together by the SIM-stitch program written in the MATlab environment. The SNR was set at 100 , using $2 / 3$ or $2 / 4$ replicates and the intragroup sample filter was based on the number of samples in each class/ total number of samples. The blank was subtracted from the samples in order to remove solvent peaks as well as contamination from the extraction procedures.

For DIMS analysis, any missing values were filled in by a separate code in Matlab, MVImpute version 05 which imputes the missing value with a feature from the original spectrum below the calculated SNR threshold. For LC analysis, missing values were removed by replacing a zero/not detected with half the value of the least intense feature for each particular peak (Dr. Warwick Dunn, personal communication). The final data matrix was then normalised by taking the ratio of the median of the intensity of the QCs and the corresponding peak intensity for each feature in a sample. The normalised matrix was then g-log transformed in order to limit intrasample variability (Parsons et al., 2007). Univariate statistics including ANOVA and the student's t-test were performed on the normalised matrix whereas the multivariate statistics including principal components analysis were performed on the normalised and g-log matrix.

### 2.4.2 MI-Pack and KEGG Database

As explained in Chapter 3, putative identifications were based upon a search of the Kyoto Encyclopedia of Genes and Genomes (KEGG) database using MI-Pack, a script written in the Python environment (Weber and Viant, 2010). The putative identifications utilised for this thesis included the 'single peak search' approach which compared the mass of the experimental compound with the mass of an identified compound found in the KEGG database within a user defined mass error. Calculated elemental composition includes adducts most commonly found in negative ion analyses, e.g., [M-H]. This approach results in a high number of false positives and suggested identifications should be confirmed with mass spectral fragmentation or orthogonal analytical approaches. In order to compare these results with the current published results, only the negative ions were analysed. The positive ion results were not examined for this preliminary research.

### 2.4.3 Triacylglycerol (TAG) Code

Lithiated TAGs were analysed by DIMS-FT-ICR-MS ${ }^{3}$ fragmentation and annotated by a code written in the python environment which compared the neutral losses from each level $\left(\mathrm{MS}^{2}\right.$ and $\left.M S^{3}\right)$ to a user generated library. The
user generated library was based upon published fragmentation mechanisms (Hsu and Turk, 2010). Robust statistical additions with user defined criteria were added to guard against spurious results. This included coverage of the times the identified peak is found in each scan. For example, $60 \%$ coverage was achieved when the chosen peak was identified three times in five scans. A second criterion determined the relative standard deviation of the peak's intensity. A variable intensity throughout the examined scans suggested noise and was not confirmed as a peak. The code is fully described in Chapter 6. The TAG fragmentation MS protocol is described in Section 2.2.5.3.

## 3. Laboratory Ageing of Wine and the Identification of Aged Wine Biomarkers

The objectives of the research described in this chapter were to monitor the ageing of wine in a controlled laboratory setting over several months using liquid wine samples as well as wine permeated sherds and to identify the extractable metabolites in order to develop a list of aged wine biomarkers. Initially, liquid aliquots of wine were aged over a four point time frame: zero, one, three, and six months in order to determine whether an ageing 'plateau' existed. The aging plateau as described in this research is the point at which the change in metabolic signature had largely ceased and the difference in two consecutive time points is insignificant based upon the PCA scores plot. The results suggested a six-month time frame was needed to reach this plateau. At that point it is hypothesised that a polymeric network had formed and that a chemical attack of the sample was necessary to release the monomeric species useful for a biomarker list. Using this six-month time frame, the laboratory aged sherds were extracted with a modified Bligh and Dyer biphasic extraction protocol resulting in a polar and a nonpolar aliquot (Bligh and Dyer, 1956). To the remaining pottery, a strong alkaline solution was added hydrolysing ester bonds and releasing monomeric species which were then extracted with ethyl acetate (Zsuga and Kiss, 1987). The extraction procedure resulted in three separate aliquots: polar, non-polar, and an alkaline fusion portion. The alkaline fusion portion, considered most useful for archaeological samples, was chosen to determine the relevant biomarkers.

### 3.1 Ageing of the Model Wine

### 3.1.1 Sample Preparation for Ageing Experiments

For the liquid wine aliquots, twenty-two individual 1 ml aliquots $(22 \times 1 \mathrm{ml})$ for each of the red wine and the white wine were pipetted into 1.8 ml glass vials for a total of 44 vials. The vials were then plugged with glass wool (Figure 3.1). For blanks, $8 \times 1 \mathrm{ml}$ aliquots of HPLC water were pipetted into 1.8 ml glass vials and plugged with glass wool (Figure 3.2 ). All vials were placed in a $40^{\circ} \mathrm{C}$ oven (Zimman and Waterhouse, 2004). Samples removed from the ageing oven after one, three, and six months of incubation included four vials of red wine and four vials of white wine along with associated water blanks. Collected samples were capped and stored in a $-80^{\circ}$ freezer prior to extraction and analysis. For time point zero, $\mathrm{t}_{0}, 100 \mu \mathrm{l}$ aliquots of red wine, white wine, and water were pipetted into 1.8 ml glass vials and allowed to dry in the ageing oven within two days, capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

| Liquid Aliquots: Red Wine; White Wine |  |  |
| :---: | :---: | :---: |
|  | $\qquad$ <br> Removed from oven: <br> 2 days | $\mathrm{n}=4$, representing zero month time point |
|  | $\xrightarrow[\text { Removed from oven: }]{ }$ <br> 1 month | $n=4$, representing one month time point |



Figure 3.1. Preparation of the ageing experiment for liquid red wine and white wine aliquots.


Figure 3.2. Preparation of the ageing experiment for liquid aliquots of water blank.

As stated in Chapter 2, for the wine/water permeated model sherds, approximately 350 ml of HPLC grade water were added to a 500 ml glass jar containing 12 randomly selected model pottery sherds (Figure 3.3). A low flow of nitrogen gas was added for 30 s in order to produce an inert atmosphere. The jar was covered tightly with a threaded cap, secured with parafilm, and allowed to sit at room temperature for seven days. The procedure was repeated for white wine and for red wine.

After the seven day period, the model sherds were removed from the liquid and allowed to air dry on methanol-cleaned glass surfaces. After drying overnight, tweezers were used to transfer 4 randomly chosen sherds to clean glass jars that were either empty, or contained 2 cm of dry sand or 2 cm of wet peat. Within the 'dry sand' jars, a further 2 cm of sand were added to cover the sherds. Within the 'wet peat' jars, a further 2 cm of wet peat were added to cover the sherds. The final labeled jars were: water/empty, water/sand, water/peat, red/empty, red/sand, red/peat, white/empty, white/sand, and white/peat. The different environments for this abiotic experiment imitated a dry sandy soil and a heavy, wet peat soil; the plain glass jars were utilised as baseline environments. All jars were placed in the ageing oven at $40^{\circ} \mathrm{C}$. After six months, the samples were removed from the oven and any heavy debris from sand or peat was removed. The samples were wrapped in aluminum foil (methanol rinsed) and stored in the $-80^{\circ} \mathrm{C}$ freezer until analysis.

## Modern Sherds



Figure 3.3. Preparation of the ageing experiment for the modern sherds permeated with water, red wine and white wine.

### 3.2 Analysis of Liquid Aged Wine/ Results and Discussion

All samples were run in triplicate in negative ion mode using a hybrid 7-Tesla Fourier transform ion cyclotron resonance mass spectrometer (LTQ-FT Ultra, Bremen, Germany) with a chip-based direct infusion nanoelectrospray ionisation assembly (Triversa, Advion Biosciences, Ithaca, NY). Quality control samples (QCs) were run with the samples and, unless otherwise noted, represent a mixture of all samples in that particular analysis. A full description of sample analysis is found in the Materials and Methods, Chapter 2. An explanation of the instrument and instrumental conditions can be found in the introduction, Chapter 1, as well as in the Materials and Methods, Chapter 2.

Unless otherwise noted, data collected from all samples were processed as follows. Red wine and white wine were processed as separately. The collected mass spectra were each 'binned' into windows of $100 \mathrm{~m} / \mathrm{z}$, whereby the edges of each window were 'stitched' using an in-house program, SIMstitch, written in the Matlab environment and more fully explained in Chapter 1.7.1 (Southam et al., 2007). The SNR was set at 100; replicate filtering was set at $2 / 3$. An intragroup sample filter was based upon the number of samples in each group/total number of samples and determined as $50 \%$. Blanks were run in a scattered fashion throughout the analyses and were subtracted from the samples in order to remove solvent peaks as well as contamination from the extraction procedures. Missing values were imputed as described earlier and the final data matrix was normalised and g-log transformed. Univariate statistics were performed on the normalised matrix whereas the multivariate statistics were performed on the normalised and g-log transformed matrix.

### 3.2.1 LTQ-FT-ICR-MS Analysis of Zero-, One-, Three-, and Six-Month Ageing Time-Points

A small subset of samples was analysed at the end of the three-month time period in order to refine the analytical methods and monitor the wine's ageing characteristics. Sample vials were removed from the $-80^{\circ} \mathrm{C}$ freezer and allowed to reach room temperature. One millilitre of methanol was added to the glass vials, $100 \mu \mathrm{l}$ added to the zero time point samples; the sides and the bottom of each vial were scraped with a glass pipette in order to disrupt the residue. Each sample was vortexed for 10 s and transferred to a 1.5 ml Eppendorf tube. Samples were centrifuged at 14,000 rpm for 5 min at $18^{\circ} \mathrm{C}$. The white wine samples were diluted 1 in 10 in 10 mM ammonium acetate solution (80/20, methanol/water); the red wine samples were diluted 1 in 5 in 10 mM ammonium acetate solution.

The final liquid ageing study was performed on the four time points: zero, one, three, and six months of the red and white wine, with accompanying blanks. Samples were removed from the $-80^{\circ} \mathrm{C}$ freezer and prepared as above. Again, the sides and the bottom of the vials were scraped with a glass pipette in order to disrupt the residue. Each sample was vortexed for 10 s and transferred to a 1.5 ml Eppendorf tube and centrifuged at $14,000 \mathrm{rpm}$ for 5 min at $18^{\circ} \mathrm{C}$. The samples analysed prior were removed from the $-80^{\circ} \mathrm{C}$ freezer, vortexed, and centrifuged, having undergone 2 freeze/thaw cycles. Based on the earlier analysis, the white wine samples were diluted 1 in 50 with 10 mM ammonium acetate solution in order to stabilise the electrospray current. The red wine samples were diluted 1 in 10 with 10 mM ammonium acetate. All samples were then vortexed for 10 seconds and centrifuged at $14,000 \mathrm{rpm}$ for 5 min at $18^{\circ} \mathrm{C}$. Eight microlitres of each sample were pipetted into the well plates for analysis in triplicate. All samples and QCs were analysed from $\mathrm{m} / \mathrm{z}$ 135-2000. The $m / z$ range was extended to 2000 in order to identify any polymer formation occurring in either of the wine systems. However, upon examination of the data, there was very
little information found in the upper $m / z$ range of the spectrum. Therefore, for statistical analyses of the data, $m / z 800$ was chosen as the upper mass limit. As stated earlier, in order to compare the results of this research with the published results of other examinations of archaeological wine, only the negative ion results were considered.


Figure 3.4 A typical negative polar spectrum collected from $m / z$ 135-2000. The overlapping SIM windows are shown as horizontal lines, in stepwise fashion indicating the overlap of $m / z 30$.

Figure 3.4 depicts a typical negative polar mass spectrum from $m / z 135-2000$ collected with the protocols referenced in Tables 1 and 2. Prior to analysis, five separate tune files were collected over the following $m / z$ ranges: 135-450, 420-730, 700-1010, 980-1030, and 1400-2000. The results were input to the file method and described more fully in Chapter 2.

Table 3.1 Conditions used for the collection of the liquid wine spectra in the LTQ-FT mass spectrometer.

|  |  |  |
| :---: | :---: | :---: |
| Scan Mode | Wide SIM |  |
|  |  |  |
| AGC target | $1 \times 10^{6}$ |  |
| Resolution |  |  |
| Window acquisition time | .30 seconds |  |
|  |  |  |
| Acquisition Time | 6.5 minutes |  |

Table 3.2 Conditions used for the electrospray ionisation via the Triversa nanoelectrospray.

|  |  |
| :---: | :---: |
| Sample Volume | $5 \boldsymbol{\mu} \mathrm{I}$ |
|  |  |
| Pressure | 0.5 psi |
|  |  |
| voltage | 1.7 kV |
|  |  |
| polarity | negative |

### 3.2.2 Univariate Statistics of Liquid Aged Wine

An ANOVA test was applied in order to determine those peaks which significantly decreased or increased in peak intensity across all four time points for red wine and for white wine datasets. The results are found in Appendix Tables 3.1 and 3.2 and were sorted on the largest decrease in fold change (a descriptive change in intensity between the initial value and the final value) utilising zeromonth time point as the control class, in conjunction with the adjusted p-value, indicative of significance. The largest decrease in fold change over six months indicated the greatest losses of metabolites from zero-month to the six-month time point, suggesting a loss of monomeric species
by (presumably) polymerisation. Appendix Tables 3.1 and 3.2 also contain putative identifications of the ions, the result of a MI-Pack search (described in Section 2.4.2).

The greatest loss over a six month period in both the red wine and the white wine was nominal mass 193, putatively identified as the acetate adduct of malic acid. Malic acid was one of the two main organic acids found in wine, tartaric being the other acid. Figures 3.5 a and 3.5 b represent the measurable drop off in signal from the initial time point and continuing throughout the remaining three time points for both malic acid and tartaric acid, putatively identified in the red wine aliquots. A t-test compared the three- and six-month time points for both the red wine and the white wine; the results indicated no significant change between the two time points suggesting $a$ halt in the metabolic change. Figures 3.6 and 3.7 are mass spectra of the red wine samples at zero month time point and six month time point. Figures 3.8 and 3.9 are mass spectra of the white wine samples at zero month time point and six month time point.

Other losses over the six month period include several putatively identified flavonols from the red wine aliquot including quercitin glucoside and kaempferol. The loss of flavonols could be due to the formation of a larger polymeric network which then precipitated out of solution. A second explanation is copigmentation, a synergistic combination between anthocyanins and flavonols (or other nonpigmented phenolic compounds) that change the hue and color of the wine (Boulton, 2001). The greatest loss of ion intensity over a six month time frame for the white wine aliquot include the organic acids putatively identified as malic, citric, and tartaric, as well as sugar acids.

As explained in the introduction, the organic components of wines polymerise over time by forming large polymers between reactive anthocyanidin and tannin species with other monomeric species such as organic and phenolic acids. As these reactions occur, the larger polymeric molecules form
and precipitate out of solution. The methanol extraction primarily targets the monomeric species whose availability presumably decreased over time. It was assumed that a similar loss of metabolites (as monomers) occurred within the laboratory aged sherds. But rather than precipitate out of solution, these polymers have lodged within the clay interstices and are thereby preserved within the clay matrix. Through chemical attack and extraction, the original building blocks of wine may be released and analysed.

### 3.2.2.1 Rationalising the Loss of Tartaric Acid over a Six-Month Timeframe

The intensity of the peak with $\mathrm{m} / \mathrm{z}$ 149, putatively identified as deprotonated tartaric acid, significantly decreased over the six-month time frame. However, there is a stable, low level of tartaric acid noticeable for the three- and six-month time points. This consistent low level may suggest the retainment of a tartrate salt, which is readily extracted with methanol/water. In a new wine, the precipitation of a potassium bitartrate salt is a common occurrence and the salt is often removed under controlled conditions by modern vintners (Dinsmore-Webb, 1974: 123; Berg and Keefer, 1958). As was mentioned earlier, tartaric acid identified in archaeological objects has often been attributed to the accumulation of a calcium tartrate salt, the substitution of calcium for two potassium cations due to the calcareous environment of the surrounding pottery (McGovern, 2003: 67). However, the formation of potassium tartrate in newly pressed wines, especially red wines, may be interrupted by the "complexation" of the tartaric acid with a phenolic compound (Balakion and Berg, 1968). A separate explanation for the loss of the acid is the condensation between the negatively charged acid and a nucleophilic site at C-4 flavan-3,4-ol forming an ester linkage. During alkaline hydrolysis, the ester bond is hydrolysed and the free acid is released.


Figure 3.5. The graphical loss of two compounds over a six month time frame from the red wine aliquot; intensity vs time in months. The bars represent the mean of four replicates, whiskers show error. (a) The loss of the ion at $\mathbf{m} / \mathbf{z}$ 193.03551 over a six month time frame. The ion was putatively identified as malic acid. (b) The loss of $\mathrm{m} / \mathrm{z} 149.00903$ over a six month time frame. The ion was putatively identified as tartaric acid.


Figure 3.6. A negative ion mass spectrum of a liquid red wine sample at zero month time point.


Figure 3.7. A negative ion mass spectrum of a liquid red wine sample at six month time point.


Figure 3.8. A negative ion mass spectrum of a liquid white wine sample at zero month time point.


Figure 3.9. A negative ion mass spectrum of a liquid white wine sample at six month time point.

### 3.2.3 Multivariate Statistics of Liquid Aged Wine

Multivariate statistics using principal components analysis (PCA) was applied to the finalized data matrices of all samples examined in this chapter. PCA was conducted using the PLS toolbox, from Eigenvector, in the Matlab environment.

### 3.2.3.1 Analysis of Zero-, One-, Three-, and Six-Month Ageing Time-points

Figure 3.10a shows the PCA scores plot for the red wine samples. The QCs in both the red wine analysis and the white wine analysis were clustered tightly together, confirming that the technical replication was relatively high with no perceived instrument drift over the whole analysis. In this scores plot, the greatest variability was along principal component 1(PC1). Zero-month was separated from the remaining three time points. The second greatest variability along PC2 separated the remaining three time points. Although statistically insignificant, a slight shift was obvious between three- and six-month time points. Figure 3.10 b shows the corresponding PCA scores plot of the white wine samples. The greatest separation amongst the white wine sample was again along PC1, explaining $79.2 \%$ of the variation in the dataset. The three- and six-month aliquots vary little and appear to cluster together along PC1. The results suggest a six month ageing plateau whereby the metabolic change ceases.

(a)

(b)

Figure 3.10. PCA plots of red and white liquid aliquots showing an ageing plateau around six months, a time where the metabolic signature of the wine is no longer significantly changing. (a) A six month time point ageing study for red wine; although small, there is still a slight change between three months and six months. (b) A six month time point ageing study for white wine; the metabolic signature of three months and six months appeared to cluster.

### 3.3 Analysis of Permeated Modern Sherds

### 3.3.1 Sample Extraction

Appendix Table 3.3 lists the laboratory-aged sherds that were sampled and extracted. The extraction method used was described earlier in Section 2.2.2. Briefly, there were four sherds for each of the nine categories including: water/peat, water/sand, water/empty jar, white/peat, white/sand, white/empty jar, red/peat, red/sand, red/empty jar. Two samples of the sherds (ca. 100 mg ) were taken with a tile cutter from separate areas of each sherd for a total of 72 samples. Samples were homogenised and extracted using a modified Bligh-Dyer extraction technique resulting in a polar and a non-polar extract (Bligh and Dyer, 1959). The remaining solids were allowed to dry overnight in a fume hood. After drying, $150 \mu \mathrm{l}$ of 4 M KOH were added to each sample and heated for 10 min on a heating block at $50^{\circ} \mathrm{C}$ (Zsuga and Kiss, 1987). The samples were first cooled to room temperature and then acidified with 400 $\mu \mathrm{l} 2 \mathrm{~N} \mathrm{HCl}$, and then extracted with $450 \mu \mathrm{l}$ ethyl acetate. The organic layer was transferred to a 1.8 ml champagne vial and blown down under a stream of nitrogen; the alkaline fusion aliquot (or KOH aliquot) was stored at $-80^{\circ} \mathrm{C}$.

At this point, the non-polar fraction was not considered for further analysis since the lipid fraction of wine was considered negligible. The polar aliquot and the alkaline fusion aliquot were considered for further mass spectrometric analysis.

All samples were run in either triplicate or quadruplicate in negative ion mode using the LTQ-FT Ultra with direct infusion nanoelectrospray. QCs were run with the samples and, unless otherwise noted, represent a mixture of all samples in that particular analysis.

### 3.3.2.1 Analysis of the Polar Aliquots

The dried polar aliquots were reconstituted in $30 \mu \mathrm{l}$ of 10 mM ammonium acetate for analysis in negative ion mode. Each sample was vortexed for 10 s and then centrifuged for 20 min at 14000 rpm at $18^{\circ} \mathrm{C}$. Five microlitres of each sample was pipetted into the well of a 384 well plate for quadruplicate analysis in negative ion mode from $m / z 70-2000$. The lower limit mass was shifted to $m / z 70$ in order to isolate and identify the majority of monomeric species.

### 3.3.2.2 Analysis of the Alkaline Fusion aliquot

Due to the high salt content produced from the alkaline fusion in the form of potassium chloride, the nanoelectrospray was highly unstable. A small sample set was analysed prior to determine the analytical protocols. The dried aliquots were reconstituted in $30 \mu \mathrm{l}$ of 10 mM ammonium acetate. The electrospray from the water sherds was inconsistent, resulting in only 3 of 9 usable spectra. Therefore, in order to limit the disruption of the electrospray during the total analyses, it was decided to limit the number of water sherds used in the final run. The final analysis included seven samples from the following groups: water/sand, white/glass, white/sand, red/glass, and red/sand. In an attempt to form a more stable electrospray, two samples of the red glass were sacrificed in order to determine the amount
of negative ion solution used to reconstitute the samples. The goal was to achieve a more stable electrospray, yet maintain the ion signal strength. However, the signal strength dropped considerably when the sample was reconstituted in $50 \mu \mathrm{l}$ of 10 mM ammonium acetate; therefore, $30 \mu \mathrm{l}$ volume was maintained.

For the final analysis of the alkaline fusions aliquots, the samples were reconstituted in $30 \mu \mathrm{l}$ of 10 mM ammonium acetate solution, vortexed for 10 s , and centrifuged for 6000 rpm for 20 min at $18^{\circ} \mathrm{C}$. QCs were prepared from an equal amount of all samples. Five microlitres of each sample were pipetted into the well plates for triplicate analysis in negative ion mode from $\mathrm{m} / \mathrm{z} 70-800$. The upper mass limit was chosen as $m / z 800$ based upon the successful release of monomeric species from the alkaline hydrolysis.

The peat samples were analysed separately from the glass and sand samples due to the overwhelming matrix effect of the wet peat. The final sample set included eight water/peat, seven red/peat and seven white/peat samples. All samples were prepared and analysed as above.

### 3.3.3 Multivariate Statistics of Wine Permeated Model Sherds

### 3.3.3.1 Analysis of Extracted Sherds: Polar Aliquot

Briefly, the data was processed using the SIMstitch method using a minimum SNR of 100, from a $m / z$ range from 70-2000, 2/4 replicate filtering, with a $30 \%$ sample filter. The final matrix processing included imputation of missing values, normalisation, and g-log transformation of the data.

The PCA scores plot of all samples analysed is presented in Figure 3.11a. The QCs are clustered together, indicating good technical replication with limited instrument drift throughout the run. The three remaining clusters are: 1. red/white/sand/glass, 2 . water/sand/glass, and 3. water/white/red/peat. The matrix effects of the dry sand do not appear to be significant in effecting the metabolic signature of the red wine or the white wine. However, all of the peat samples, water, white wine, and red wine, were clustered together, which suggest an overwhelming matrix effect of the peat in conjunction with a lack of signature molecules for the red and white wine. The most likely reason for this is the incomplete polymerisation of the red wine and the white wine, probably due to sample preparation.

Figure 3.11b depicts the PCA plot with the peat samples removed. The greatest separation is along PC1 between the water and the wine samples; the red and white wine samples separate along PC2. The top 50 loadings (those ions which are most characteristic for each representative sample type) separating the red wine and the white wine for the polar metabolites are listed in Appendix Table 3.4. In terms of archaeological samples, it is highly likely that metabolites extractable in a methanol/water environment do not represent what has remained over millennia due to the potential prolonged exposure to water. The use of polar aliquots for an archaeological sample may be limited to the environment from which the material was unearthed (soil, sand, or clay), certainly a useful description for a full-scale archaeological excavation.

(a)

(b)

Figure 3.11. PCA scores plot of the polar aliquot of laboratory aged sherds. The samples analysed were white wine, red wine, and water in three separate environments, empty jar, dry sand, and wet peat. (a) The QCs are clustered tightly together indicative of good technical repeatability with limited instrumental drift. The peat samples are separated from other water and wine samples, indicative of overwhelming matrix effects. (b) PCA scores plot with QCs and peat samples removed. The separation between water and both types of wine samples is along PC1; the red wine and the white wine separate along PC2. There appears to be little difference between the environments of the glass jar and the dry sand, the separation was based upon the metabolic signatures of the wine only. The results suggest the successful extraction of polar metabolites from the sherds.

### 3.3.3.2 Analysis of Extracted Sherds: Alkaline Fusion Aliquot

Briefly, the data were processed using the SIMstitch method using a minimum SNR of 100, from a $\mathrm{m} / \mathrm{z}$ range from 70-800, $2 / 3$ replicate filtering, with a $20 \%$ sample filter. The final matrix processing included imputation of missing values, normalisation, and g-log transformation of the data. A comparison between the top 50 loadings from the polar aliquot and the top 50 loadings from the KOH aliquots are given in Appendix Table 3.5. The differences in the top loadings indicate the success of the alkaline fusion on the pottery sherds.

The nanoelectrospray stability during this run was not continuously maintained and 2 out of 5 QCs, 3 out of 7 water/sand samples, an extraction blank, and one of the red samples were lost during the run due to an under fill error caused by the electrospray dropping out. Figures $3.12 a$ and $b$ represent the PCA results of the remaining samples. As can be seen in Figure 3.12a, the remaining QCs were clustered tightly together confirming good technical repeatability with limited instrument drift. Figure 3.12b represents the PCA plot with the QCs removed. The three samples that fell out of the $95 \%$ confidence limit were considered outliers and removed from further data analysis: red glass 3B, white glass 4A, and white sand 2B. Table 6 in the Appendix lists the top 200 loadings for white wine and the top loadings for red wine, based upon the separation PC2.

The greatest separation between the wine samples and the water samples is along PC1, as seen in Figure 3.12b. The top 150 loadings of the wine samples, the masses that most significantly characterise wine from the water group, were searched against the KEGG database. The top 100 putatively identified masses were used to compile a targeted biomarker list for application to archaeological
samples in an attempt to answer the question of presence/absence of aged wine remnants and are listed in Table 3.3. The complete list of the top 150 masses is listed in Appendix Table 3.7.

The information listed in Table 3.3 includes the suggested ion adduct, the empirical formula of the parent ion, the theoretical masses of the parent ion and of the adduct ion, and the mass error in ppm between the experimental mass and the theoretical mass. Putative identifications are listed and were based upon a search of the Kyoto Encyclopedia of Genes and Genomes (KEGG) database using MI-Pack, a script written in the Python environment (Weber and Viant, 2010). Putative metabolite identifications were chosen based upon a user defined parameter of 2 ppm mass error difference between the theoretical and experimental mass of proposed adduct formations in the negative ion mode. This mass error was based upon the weekly external calibration of the instrument. The adducts included: $[\mathrm{M}-\mathrm{H}]^{-}$, $[\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{]},[\mathrm{M}+\mathrm{K}-2 \mathrm{H}]^{]},[\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]^{-},[\mathrm{M}+\mathrm{Cl}]^{-}$, and $[\mathrm{M}+(37 \mathrm{Cl})]^{\prime}$. As an example, an experimental peak identified at $m / z 149.0090$ was putatively identified as the deprotonated form of tartaric acid having a theoretical $m / z$ of 149.0092 in the form [ $\mathrm{M}-\mathrm{H}]^{-}$. In certain cases, compounds were identified by several different adduct forms and thus one putative ID could be associated with three separate ions.

In nearly every case, the putative identifications offer several possible compounds based upon the experimentally determined mass. By comparing the resultant list from the KEGG results in Table 3.3 with the known components in wine described in the Introduction, Chapter 1, a putative list of compounds was compiled. Tartaric, malic, succinic, lactic, and citric acids are considered five of the major organic acids found in wine and their presence in the proposed biomarker list was expected. The presence of secondary organic acids on the list includes malonic acid, alpha-ketobutyric acid, as well as the sugar acids, including glucuronic acid. Phenolic acids are another major group of compounds in wine
and include the benzoic acids and the cinnamic acids. The benzoic acids are represented by vanillic acid, gallic acid, syringic acid and dihydroxybenzoic acid. The cinnamic acids are represented by coumaric acid, caffeic acid, and ferulic acid. Several of the remaining acids on the list are conjugated acids, either conjunctions of two separate acids, or possible artefacts of sample preparation including heating and/or methylation.

Under the action of alkaline fusion, ester bonds are hydrolysed releasing the acids for extraction and analysis. For the red wine sherds, the origin of the phenolic acids is mainly due to the breakdown products of the anthocyanidins including: syringic acid (malvidin), gallic acid (delphinidin), protocatechuic acid (cyanidin), vanillic acid (peonidin), and methylgallic acid (petunidin). The general breakdown of tannins such as catechin and gallocatechinn could release gallic acid and protocatechuic acid respectively. Free phenolic and organic acids esterified within a polymer are also released. In white wine, these include the following benzoic acids: protocatechuic, gallic and syringic acids (Tian et al., 2009; Betes-Saura et al., 1996). The organic acid, tartaric acid, could also originate from hydrolysis of the hydroxycinnamate tartrates (Figure 1.9), prevalent in both red and white wines. Since many of these acids are found in both red and white, it is risky to assign one acid as a biomarker for either red wine or white wine. A more conservative approach is to analyse a list of biomarkers to determine the absence/presence of wine. And comparative ratios of the chosen biomarkers found in aged wine may prove more reliable for the ultimate determination of red vs. white.

(a)

(b)

Figure 3.12. (a) The PCA scores plot of the alkaline fusion aliquot from the laboratory-aged sherds. The QCs are clustered together indicative of good technical repeatability with limited instrumental drift. (b) PCA scores plot of the alkaline fusion aliquot with QCs removed; three samples were also removed, having fallen outside of the $95 \%$ confidence limit: red glass $3 B$, white glass $4 A$, and white sand $2 B$. The wine samples separate from the water samples along PC1. Red wine and white wine samples separate along PC2 indicating distinct metabolic signature for both types of wine.

Table 3.3 The top putatively identified masses taken from the PCA presented in Figure 3.12b and sorted by PC1.

| m/z | Relative intensity | PC1 | Empirical formula (parent) | Empirical formula (peak) | Ion form | Theoretical mass (neutral) (Da) | Theoretical m/z (Da) | Mass error (ppm) | Putative annotation based on KEGG database | possible component in wine (after alkaline fusion) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 149.00912 | 12839920.3 | -0.102401503 | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{6}$ | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{6}$ | [M-H] | 150.01644 | 149.00916 | -0.29 | ['(R,R)-Tartaric acid', '(S,S)-Tartaric acid', 'meso-Tartaric acid'] | tartaric acid |
| 175.06112 | 4971425.5 | -0.094286393 | $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{O}_{3}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 116.04735 | 175.0612 | -0.45 | ['2-Oxopentanoic acid', '3-Methyl- <br> 2-oxobutanoic acid', '3- <br> Oxopentanoic acid', '5- <br> Oxopentanoate'] | oxopentanoic acid |
| 175.06112 | 4971425.5 | -0.094286393 | $\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{O}_{5}$ | $\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{O}_{5}$ | [M-H] | 176.06848 | 175.0612 | -0.45 | ['(2R,3S)-3-IsopropyImalate', '(R)-2-(n-Propyl)-malate', '2Propylmalate', '3-Propylmalate', 'alpha-IsopropyImalate'] | isopropylmalic acid |
| 133.01418 | 16494794.9 | -0.085369204 | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{5}$ | [M-H] | 134.02153 | 133.01425 | -0.52 | ['(R)-Malate', '(S)-Malate', '3-Dehydro-L-threonate', 'Malate'] | malic acid |
| 147.02996 | 1233057 | -0.075751737 | $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{O}_{3}$ | $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{O}_{3}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 88.016045 | 147.0299 | 0.42 | ['3-Hydroxypropenoate', '3Oxopropanoate', 'Pyruvate'] | pyruvic acid |


| 147.02996 | 1233057 | -0.075751737 | $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{O}_{5}$ | $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{O}_{5}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 148.03718 | 147.0299 | 0.42 | ['(R)-2-Hydroxyglutarate', '(R)-2Methylmalate', '(S)-2Hydroxyglutarate', '(S)-2- <br> Methylmalate', '2-Dehydro-3-deoxy-D-xylonate', '2-Dehydro-3-deoxy-L-arabinonate', '2- <br> Hydroxyglutarate', 'Citramalate', <br> 'D-Arabinono-1,4-lactone', 'D-Xylono-1,4-lactone', 'DXylonolactone', 'D-erythro-3Methylmalate', 'D-threo-3- <br> Methylmalate', 'L-Arabinono-1,4lactone', 'L-Arabinono-1,5lactone', 'L-Xylono-1,4-lactone', 'L-threo-3-Methylmalate'] | citramalic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 153.01939 | 1799686.6 | -0.070222273 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 154.02661 | 153.01933 | 0.37 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4Dihydroxybenzoate', 'Patulin'] | dihydroxybenzoic acid |
| 191.01982 | 860206.9 | -0.070063079 | $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{5}$ | $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{5}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 132.00588 | 191.01973 | 0.48 | ['2- <br> Hydroxyethylenedicarboxylate', <br> 'Oxaloacetate', 'trans-2,3- <br> Epoxysuccinate'] | oxaloacetic acid |
| 191.01982 | 860206.9 | -0.070063079 | $\mathrm{C}_{6} \mathrm{H}_{8} \mathrm{O}_{7}$ | $\mathrm{C}_{6} \mathrm{H}_{8} \mathrm{O}_{7}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 192.02701 | 191.01973 | 0.48 | ['(1R,2S)-1-Hydroxypropane-1,2,3tricarboxylate', '(1S,2S)-1- <br> Hydroxypropane-1,2,3- <br> tricarboxylate', '(4R,5S)-4,5,6- <br> Trihydroxy-2,3-dioxohexanoate', '2,5-Didehydro-D-gluconate', '2-Dehydro-3-deoxy-D-glucarate', '5-Dehydro-4-deoxy-D-glucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] | citric acid |
| 147.06634 | 1883832.2 | -0.068889369 | $\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}_{2}$ | $\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}_{2}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 88.05243 | 147.06628 | 0.38 | ['(R)-Acetoin', '1,4-Dioxane', '2Methylpropanoate', 'Acetoin', 'Butanoic acid', 'Ethyl acetate'] | butanoic acid |
| 147.06634 | 1883832.2 | -0.068889369 | $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{4}$ | $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{4}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 148.07356 | 147.06628 | 0.38 | ['(R)-2,3-Dihydroxy-3methylpentanoate', '(R)Mevalonate', '(R)-Pantoate', '(S)Mevalonate', '2,3-Dihydroxy-3methylpentanoate', '3,6-Dideoxy-L-galactose', 'Abequose'] | mevalonic acid |


| 133.05061 | 3301580.2 | -0.06607991 | $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}_{2}$ | $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}_{2}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 74.03678 | 133.05063 | -0.18 | ['(R)-Lactaldehyde', '(S)- <br> Lactaldehyde', '3- <br> Hydroxypropanal', 'Glycidol', <br> 'Hydroxyacetone', 'Lactaldehyde', <br> 'Methyl acetate', 'Propanoate'] |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 133.05061 | 3301580.2 | -0.06607991 | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{O}_{4}$ | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{O}_{4}$ | [M-H] ${ }^{-}$ | 134.05791 | 133.05063 | -0.18 | ['(R)-2,3-Dihydroxy-3methylbutanoate', '1-Deoxy-Dxylulose', '2,3-Dihydroxy-3methylbutanoate', '2-Deoxy-Larabinose', '2-Deoxy-alpha-Dribopyranose', 'Deoxyribose'] | sugar acid |
| 163.04002 | 1344736.2 | -0.050834612 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | [M-H] | 164.04735 | 163.04007 | -0.3 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2Hydroxycinnamate'] | coumaric acid/phenylpyruvic acid |
| 130.99865 | 298220 | -0.049814628 | $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{5}$ | $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{5}$ | [M-H] | 132.00588 | 130.9986 | 0.39 | ['2- <br> Hydroxyethylenedicarboxylate', <br> 'Oxaloacetate', 'trans-2,3Epoxysuccinate'] | oxaloacetic acid |
| 179.03508 | 313212.1 | -0.048992668 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{4}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{4}$ | [M-H] | 180.04226 | 179.03498 | 0.54 | ['2-Hydroxy-3-(4hydroxyphenyl)propenoate', '3-(4Hydroxyphenyl)pyruvate', 'Aspirin', 'Caffeate', 'trans-2,3Dihydroxycinnamate'] | caffeic acid |
| 211.02497 | 262104.9 | -0.046164544 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{6}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{6}$ | [M-H] | 212.03209 | 211.02481 | 0.74 | ['2-Hydroxy-6- <br> ketononatrienedioate', '3-(2- <br> Carboxyethenyl)-cis,cis- <br> muconate', '5-Carboxyvanillic acid'] | carboxyvanillic acid |
| 206.96793 | 191453.9 | -0.045614948 | $\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{O}_{7} \mathrm{P}$ | $\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{O}_{7} \mathrm{P}$ | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]^{-}$ | 185.99294 | 206.96761 | 1.54 | ['2-Phospho-D-glycerate', '3-Phospho-D-glycerate', '3-Phospho-DL-glycerate'] |  |
| 144.03033 | 141336.5 | -0.045401167 | $\mathrm{C}_{5} \mathrm{H}_{7} \mathrm{NO}_{4}$ | $\mathrm{C}_{5} \mathrm{H}_{7} \mathrm{NO}_{4}$ | [M-H] | 145.03751 | 144.03023 | 0.68 | ['2-Oxoglutaramate', '4Oxoglutaramate'] | 2-keto-glutaramic acid |
| 181.01434 | 180546.4 | -0.04498824 | $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{O}_{5}$ | [M-H] | 182.02153 | 181.01425 | 0.5 | ['2-Hydroxyisophthalic acid', '3,5Dihydroxyphenylglyoxylate', '4Hydroxyphthalate', 'Stipitatate'] | phthalic acid |


| 195.03004 | 323515.6 | -0.043538515 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{5}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{5}$ | [M-H] ${ }^{-}$ | 196.03718 | 195.0299 | 0.73 | ['3-(3,4- Dihydroxyphenyl)pyruvate'] | dihydroxyphenylpyruvic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 197.04542 | 6107928.3 | -0.041884423 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{3}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{3}$ | [M+Hac-H] | 138.0317 | 197.04555 | -0.65 | ['2-Hydroxy-5-methylquinone', '3,4-Dihydroxybenzaldehyde', '3Hydroxybenzoate', '4- <br> Hydroxybenzoate', 'Gentisate aldehyde', 'Salicylate', 'Sesamol'] | hydroxybenzoic acid |
| 197.04542 | 6107928.3 | -0.041884423 | $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{O}_{5}$ | $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{O}_{5}$ | [M-H] | 198.05283 | 197.04555 | -0.65 | ['3-(3,4-Dihydroxyphenyl)lactate', '3-Methoxy-4-hydroxymandelate', 'Syringic acid'] | syringic acid |
| 117.0193 | 2144058.9 | -0.041306416 | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{4}$ | [M-H] ${ }^{-1}$ | 118.02661 | 117.01933 | -0.29 | ['Methyl oxalate', 'Methylmalonate', 'Succinate'] | succinic acid |
| 103.00368 | 88796.1 | -0.037181229 | $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{O}_{4}$ | $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{O}_{4}$ | [M-H] | 104.01096 | 103.00368 | -0.03 | ['2-Hydroxy-3-oxopropanoate', 'Hydroxypyruvate', 'Malonate'] | malonic acid |
| 169.01433 | 182359 | -0.036673686 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{5}$ | [M-H] | 170.02153 | 169.01425 | 0.48 | ['Gallate'] | gallic acid |
| 197.00934 | 78631 | -0.035967678 | $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{O}_{6}$ | $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{O}_{6}$ | [M-H] | 198.01644 | 197.00916 | 0.9 | ['3,4-Dihydroxyphthalate', '4,5Dihydroxyphthalate'] | phthalic acid |
| 199.01683 | 1125705.9 | -0.03496134 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | [M+CI] | 164.04735 | 199.01675 | 0.42 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', <br> 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis- <br> 2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2Hydroxycinnamate'] | coumaric acid/phenylpyruvic acid |
| 183.03001 | 115160.7 | -0.033479499 | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{5}$ | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{5}$ | [M-H] | 184.03718 | 183.0299 | 0.61 | ['3,4-Dihydroxymandelate', '3-0Methylgallate'] | methyl gallic acid |
| 201.01393 | 345164.1 | -0.032280828 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | [ $\mathrm{M}+(37 \mathrm{Cl})]^{-}$ | 164.04735 | 201.0138 | 0.66 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2Hydroxycinnamate'] | coumaric acid/phenylpyruvic acid |


| 189.07696 | 334513.5 | -0.031522479 | $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}_{5}$ | $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}_{5}$ | [ $\mathrm{M}-\mathrm{H}$ ] | 190.08413 | 189.07685 | 0.59 | $\begin{gathered} {\left[{ }^{\prime}(R)-3-((\mathrm{R})-3-\right.} \\ \text { Hydroxybutanoyloxy)butanoate'] } \end{gathered}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 205.03551 | 30085.2 | -0.03117134 | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{5}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 146.02153 | 205.03538 | 0.64 | ['2-Oxoglutarate', '5-Hydroxy-2,4dioxopentanoate', 'Dehydro-D-arabinono-1,4-lactone', 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] | oxoglutaric acid |
| 205.03551 | 30085.2 | -0.03117134 | $\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{7}$ | $\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{7}$ | [M-H] | 206.04266 | 205.03538 | 0.64 | ['(2S,3R)-3-Hydroxybutane-1,2,3tricarboxylate', '(R)-2- <br> Hydroxybutane-1,2,4- <br> tricarboxylate', '2-Methylcitrate', 'Homoisocitrate'] | methylcitric acid |
| 161.04562 | 67109.1 | -0.030741974 | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{3}$ | $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}_{3}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 102.0317 | 161.04555 | 0.44 | ['(S)-Methylmalonate semialdehyde', '2-Methyl-3oxopropanoate', '2Oxobutanoate', 'Acetoacetate', 'Succinate semialdehyde'] | alpha ketobutyric acid |
| 161.04562 | 67109.1 | -0.030741974 | $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{5}$ | $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{5}$ | [ $\mathrm{M}-\mathrm{H}$ ] | 162.05283 | 161.04555 | 0.44 | ['(2R,3S)-2,3-DimethyImalate', <br> '(R)-2-Ethylmalate', '(R)-3,3Dimethylmalate', '(S)-2- <br> (Hydroxymethyl)glutarate', '1,5- <br> Anhydro-D-fructose', '2-Dehydro- <br> 3-deoxy-D-fuconate', '2-Dehydro- <br> 3-deoxy-L-fuconate', '2-Dehydro- <br> 3-deoxy-L-rhamnonate', '2-Deoxy- <br> scyllo-inosose', '2- <br> Hydroxyadipate', '3,6- <br> Anhydrogalactose', '3,6- <br> Anhydroglucose', '3-Ethylmalate', <br> '3-Hydroxy-3-methylglutarate', 'D-Fucono-1,4-lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5lactone', 'L-Rhamnono-1,4lactone', 'Lichenin'] | dimethyl malic acid/ ethyl malic acid/ sugar acids |
| 145.01434 | 43325.5 | -0.030439508 | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{5}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 146.02153 | 145.01425 | 0.63 | ['2-Oxoglutarate', '5-Hydroxy-2,4dioxopentanoate', 'Dehydro-D-arabinono-1,4-lactone', 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] | oxoglutaric acid |


| 181.05073 | 70728.7 | -0.029838377 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 122.03678 | 181.05063 | 0.53 | ['3-Hydroxybenzaldehyde', '4Hydroxybenzaldehyde', 'Benzoate', 'Salicylaldehyde', 'Tropolone'] | benzoic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 193.05078 | 194456.2 | -0.029083865 | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{4}$ | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{4}$ | [M-H] | 194.05791 | 193.05063 | 0.76 | ['2,4,8-Trihydroxy-1-tetralone', '5Hydroxyconiferaldehyde', '6Hydroxymellein', 'Dimethyl phthalate', 'Ferulate', 'Isoferulic acid', 'Kakuol', 'Methyl caffeate', 'Scytalone'] | ferulic acid |
| 129.01937 | 657155 | -0.028954418 | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}_{4}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 130.02661 | 129.01933 | 0.28 | ['(E)-Glutaconate', '2,5Dioxopentanoate', '2MethyImaleate', '4,5Dioxopentanoate', 'Acetylpyruvate', 'Itaconate', 'Mesaconate'] | metlymaleic acid |
| 255.05092 | 130108.2 | -0.028903558 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{5}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{5}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 196.03718 | 255.05103 | -0.43 | $\begin{gathered} \text { ['3-(3,4- } \\ \text { Dihydroxyphenyl)pyruvate'] } \end{gathered}$ | dihydroxyphenylpyruvic acid |
| 157.0507 | 47407.2 | -0.02795424 | $\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{4}$ | $\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{4}$ | [M-H] | 158.05791 | 157.05063 | 0.42 | ['(1S,4S)-4-Hydroxy-3-oxocyclohexane-1-carboxylate', '2-IsopropyImaleate', '5-D-(5/6)-5-C-(Hydroxymethyl)-2,6-dihydroxycyclohex-2-en-1-one', 'Dimethyl citraconate'] | isopropylmaleic acid |
| 188.99615 | 94910.2 | -0.027731671 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | [M+CI] | 154.02661 | 188.99601 | 0.73 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4Dihydroxybenzoate', 'Patulin'] | dihydroxybenzoic acid |
| 213.04063 | 51947.8 | -0.027101305 | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 154.02661 | 213.04046 | 0.78 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4Dihydroxybenzoate', 'Patulin'] | dihydroxybenzoic acid |


| 215.01744 | 167385.1 | -0.026803166 | $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{7}$ | $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{7}$ | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]^{-}$ | 194.04266 | 215.01732 | 0.54 | ['2-Dehydro-D-galactonate', '2-Keto-D-gluconic acid', '3-Dehydro- <br> L-gulonate', '5-Dehydro-Dgluconate', 'D-Fructuronate', 'DGalacturonate', 'D-Glucuronate', <br> 'D-Glucuronic acid', 'D- <br> Mannuronate', 'D-Tagaturonate', 'Galacturonic acid', 'L-Guluronic acid', 'L-Iduronic acid', 'beta-DGlucopyranuronic acid'] | sugar acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 253.07166 | 211403.7 | -0.026398465 | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{4}$ | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{4}$ | [M+Hac-H] | 194.05791 | 253.07176 | -0.41 | ['2,4,8-Trihydroxy-1-tetralone', '5Hydroxyconiferaldehyde', '6Hydroxymellein', 'Dimethyl phthalate', 'Ferulate', 'Isoferulic acid', 'Kakuol', 'Methyl caffeate', 'Scytalone'] | ferulic acid |
| 141.01942 | 49453.5 | -0.025952037 | $\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{O}_{4}$ | $\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{O}_{4}$ | [M-H] | 142.02661 | 141.01933 | 0.61 | ['(S)-5-Oxo-2,5-dihydrofuran-2acetate', '1,2,3,5- <br> Tetrahydroxybenzene', '2,5- <br> Dihydro-5-oxofuran-2-acetate', '2- <br> Hydroxymuconate semialdehyde', <br> '2-Oxo-2,3-dihydrofuran-5- <br> acetate', 'Kojic acid', 'cis,cis-4- <br> Hydroxymuconic semialdehyde', 'cis,cis-Muconate', 'cis,transHexadienedioate'] | kojic acid |
| 167.03504 | 350068.8 | -0.025797423 | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{4}$ | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{4}$ | [M-H] | 168.04226 | 167.03498 | 0.34 | ['(R)-4-Hydroxymandelate', '(S)-4- <br> Hydroxymandelate', '1,2- <br> Dihydrophthalic acid', '2,6- <br> Dihydroxyphenylacetate', '2,6- <br> Dimethoxybenzoquinone', '2- <br> Hydroxy-6-oxoocta-2,4,7- <br> trienoate', '3,4- <br> Dihydroxymandelaldehyde', '3,4- <br> Dihydroxyphenylacetate', '4- <br> Hydroxymandelate', '4- <br> Hydroxymethylsalicylate', '4Hydroxypheoxyacetate', <br> 'Homogentisate', 'Orsellinate', 'Vanillate'] | vanillic acid |


| 89.02438 | 57476.2 | -0.024994887 | $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}_{3}$ | $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}_{3}$ | [M-H] | 90.031695 | 89.024419 | -0.43 | ['(R)-Lactate', '(S)-Lactate', '3Hydroxypropanoate', 'DGlyceraldehyde', 'Glyceraldehyde', 'Glycerone', 'L-Glyceraldehyde', 'Lactate'] | lactic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

### 3.3.3.2.1. Alkaline Fusion Aliquot of the Peat Sherds

Due to the overwhelming matrix effect of the peat, the biomarker list chosen for the archaeological samples is listed in Table 3.3 and was developed from the results of the glass/sand matrix. However, a separate analysis of the peat samples was undertaken and may offer additional reference compounds for the analysis of objects from a heavy peat environment. Humic acid polymers often form in heavy peat soils; the application of alkaline hydrolysis to these polymers may release compounds that overlap with the proposed biomarkers for the aged wine.

### 3.3.3.2.1.1 Multivariate Statistics of Alkaline Fusion Peat Sherds

Figure 3.13a and 3.13b represent the PCA scores plots of the alkaline fusion results of the peat sherds immersed in water, red wine, and white wine and aged for six months. There is no obvious separation and univariate statistical analysis of the PC scores were then used to determine the significant differences between the three classes (water/red/white) along each principal component. Significance was determined between red and white wine along PC1 ( $\mathrm{p}=0.015$ ); significance was determined between water and wine along PC3 ( $\mathrm{p}=0.000369$ ). Table 3.4 represents the top 20 putatively annotated $\mathrm{m} / \mathrm{z}$ values that distinguish wine from water in a heavy peat environment. The results were taken from Figure 3.13b and sorted by PC3.

From the results in Table 3.4, phenylpyruvic/coumaric acids were the only putatively annotated compounds also found in the glass/sand samples. Free 12-hydroxydodecanoic acid was identified in the volatile components of wine (Borea Carnacini et al., 1980). Methylbezaldehyde was also identified in
the volatile portion of wine (Flamini and Traldi, 2010: 236). The lack of effective signature molecules suggests an incomplete polymerisation of the wine. The reason for this may be due to the matrix effect of the peat or to the preparation of the sherds. The sherds should be re-prepared as described in Chapter 7 in order to determine if there is substantial overlap between the biomarkers determined from the aged wine and the possible biomarkers from the humic acid polymers.

(a)

(b)

Figure 3.13. The PCA scores plot of the alkaline fusion aliquot from the laboratory-aged sherds from the peat matrix only. (a) The QCs are not clustered together suggesting some drift during the analysis. Red3A was removed, having fallen outside of the $95 \%$ confidence limit. (b) PCA scores plot of the alkaline fusion aliquot with QCs removed, showing PC1vsPC3; two samples were also removed, having fallen outside of the $95 \%$ confidence limit: white 1A and water 2B. Univariate analysis revealed that PC1 represents a significant difference between red and white wine; PC3 represents a significant difference between water and both types of wine.

Table 3.4 The top twenty putatively identified masses taken from PCA seen in Figure 3.13b and sorted by PC3.

| m/z | Relative intensity | PC3 | Empirical formula (parent) | Empirical formula (peak) | Ion form | Theoretical mass (neutral) (Da) | Theoretical m/z (Da) | Mass error (ppm) | Putative annotation based on KEGG database | possible components in wine (after alkaline fusion) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 590.44646 | 236712.5 | -0.1687862 |  | 0 |  |  |  |  | 0 |  |
| 618.47777 | 148087 | -0.1562345 |  | 0 |  |  |  |  | 0 |  |
| 634.47273 | 63584.6 | -0.1249341 |  | 0 |  |  |  |  | 0 |  |
| 119.05051 | 16281.1 | -0.11175 | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}$ | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}$ | $[\mathrm{M}+\mathrm{Cl}]$ | 84.081324 | 119.05073 | -1.81 | ['1-Methylpyrrolinium'] |  |
| 119.05051 | 16281.1 |  | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}$ | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}$ | [M-H] | 120.05752 | 119.05024 | 2.28 | ['2-Methylbenzaldehyde', '3Methylbenzaldehyde', '4Hydroxystyrene', 'Acetophenone', 'Phenylacetaldehyde', 'Styrene oxide', 'p-Tolualdehyde'] |  |
| 215.1655 | 339387.2 | -0.1113617 | $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{O}$ | $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{O}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 156.15142 | 215.16527 | 1.08 | ['(+)-Neomenthol', '(-)-Citronellol', <br> '(-)-Menthol', 'Decanal', 'betaCitronellol'] |  |
| 215.1655 | 339387.2 |  | $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{O}_{3}$ | $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{O}_{3}$ | [M-H] | 216.17255 | 215.16527 | 1.08 | ['12-Hydroxydodecanoic acid'] | 12Hydroxydodecanoic acid |
| 144.04541 | 216030.9 | -0.1066722 | $\mathrm{C}_{9} \mathrm{H}_{7} \mathrm{NO}$ | $\mathrm{C}_{9} \mathrm{H}_{7} \mathrm{NO}$ | [M-H] | 145.05276 | 144.04549 | -0.54 | ['1(2H)-Isoquinolinone', '3Methyleneoxindole', '8Hydroxyquinoline', 'Indole-3carboxaldehyde', 'Quinolin-2-ol', 'Quinolin-4-ol'] | possible fungicide |
| 205.01445 | 122744.4 | -0.1006064 | $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{O}_{5}$ | $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{O}_{5}$ | [M-H] | 206.02153 | 205.01425 | 0.98 | ['Flaviolin'] |  |
| 591.44997 | 76207.8 | -0.0933017 |  | 0 |  |  |  |  | 0 |  |
| 119.05017 | 202425.4 | -0.0829639 | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}$ | $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}$ | [ $\mathrm{M}+\mathrm{Cl}]^{-}$ | 84.081324 | 119.05073 | -4.67 | ['1-Methylpyrrolinium'] |  |


| 119.05017 | 202425.4 |  | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}$ | $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}$ | [M-H] | 120.05752 | 119.05024 | -0.58 | ['2-Methylbenzaldehyde', '3Methylbenzaldehyde', '4Hydroxystyrene', 'Acetophenone', 'Phenylacetaldehyde', 'Styrene oxide', 'p-Tolualdehyde'] | methylbenzaldehyde |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 199.01694 | 138861.2 | -0.0782624 | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{2}$ | $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{2}$ | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]^{-}$ | 162.06808 | 199.01669 | 1.27 | ['(1S,2S)-1,2-Dihydronaphthalene- <br> 1,2-diol', '1,2-Dihydronaphthalene-1,2-diol', '4- <br> Hydroxycinnamoylmethane', <br> 'Isosafrole', 'Methyl cinnamate', <br> 'Safrole', 'cis-1,2- <br> Dihydronaphthalene-1,2-diol', 'pMethoxycinnamaldehyde', 'trans-2Phenylcyclopropanecarboxylic acid'] |  |
| 199.01694 | 138861.2 |  | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | [ $\mathrm{M}+\mathrm{Cl}]^{-}$ | 164.04735 | 199.01675 | 0.97 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2Hydroxycinnamate', 'cis-pCoumarate', 'trans-2Hydroxycinnamate'] | phenylpyrivic acid/coumaric acid |
| 163.04019 | 289591.3 | -0.0760137 | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{O}_{3}$ | [M-H] | 164.04735 | 163.04007 | 0.74 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2Hydroxycinnamate', 'cis-pCoumarate', 'trans-2Hydroxycinnamate'] | phenylpyrivic acid/coumaric acid |
| 219.03014 | 163941 | -0.0753632 | $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{O}_{5}$ | $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{O}_{5}$ | [M-H] | 220.03718 | 219.0299 | 1.1 | ['(3E)-4-(2-Carboxyphenyl)-2-oxobut-3-enoate', '(3Z)-4-(2-Carboxyphenyl)-2-oxobut-3enoate', '3-[6- <br> (Carboxymethylene)cyclohexa-2,4-dien-1-ylidene]-2-oxopropanate', 'Purpurogallin'] |  |
| 138.03217 | 28433.3 | -0.0726176 | $\mathrm{C}_{3} \mathrm{H}_{10} \mathrm{NO}_{3} \mathrm{P}$ | $\mathrm{C}_{3} \mathrm{H}_{10} \mathrm{NO}_{3} \mathrm{P}$ | [M-H] | 139.03983 | 138.03256 | -2.79 | ['N-Monomethyl-2aminoethylphosphonate'] |  |


| 138.03217 | 28433.3 |  | $\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{NO}_{2}$ | $\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{NO}_{2}$ | [M+CI] | 103.06333 | 138.03273 | -4.06 | ['(R)-3-Amino-2-methylpropanoate', <br> '(S)-2-Aminobutanoate', '2-Amino-2-methylpropanoate', '3Aminoisobutyric acid', '4Aminobutanoate', 'D-2- <br> Aminobutyrate', 'HBA', 'L-3-Aminoisobutanoate', 'N,NDimethylglycine', ' N -Ethylglycine', 'N-Methyl-L-alanine', 'n-Propyl carbamate'] |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 138.03217 | 28433.3 |  | $\mathrm{C}_{5} \mathrm{H}_{11} \mathrm{NO}$ | $\mathrm{C}_{5} \mathrm{H}_{11} \mathrm{NO}$ | [M+K-2H] | 101.08406 | 138.03267 | -3.63 | ['2-Methylpropanal 0methyloxime', '2-methylbutanal oxime', '3-Methylbutyraldehyde oxime', '4-Methylaminobutanal', '5Aminopentanal', 'Pentanamide'] |  |
| 138.03217 | 28433.3 |  | $\mathrm{C}_{8} \mathrm{H}_{7} \mathrm{~N}$ | $\mathrm{C}_{8} \mathrm{H}_{7} \mathrm{~N}$ | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]^{-}$ | 117.05785 | 138.03252 | -2.52 | ['Indole', 'Phenylacetonitrile'] |  |
| 559.40093 | 207966 | -0.0706236 | $\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{4}$ | $\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{4}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{\text {] }}$ | 500.38656 | 559.40041 | 0.92 | ['3beta-Hydroxylanostane-7,11dione acetate'] |  |
| 241.21756 | 1376803 | -0.0700686 | $\mathrm{C}_{15} \mathrm{H}_{30} \mathrm{O}_{2}$ | $\mathrm{C}_{15} \mathrm{H}_{30} \mathrm{O}_{2}$ | [ $\mathrm{M}-\mathrm{H}]^{-}$ | 242.22458 | 241.2173 | 1.06 | ['12-Methyltetradecanoic acid', 'Pentadecanoic acid'] | pentadecanoic acid |
| 223.02509 | 112056.1 | -0.0686461 |  | 0 |  |  |  |  | 0 |  |
| 703.22247 | 92548.5 | -0.0681196 | $\mathrm{C}_{32} \mathrm{H}_{42} \mathrm{O}_{16}$ | $\mathrm{C}_{32} \mathrm{H}_{42} \mathrm{O}_{16}$ | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]^{-}$ | 682.24729 | 703.22196 | 0.73 | ['Bruceoside A'] |  |
| 221.0094 | 60414 | -0.0676636 |  | 0 |  |  |  |  | 0 |  |
| 531.3697 | 480083.2 | -0.06758 | $\mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O}_{4}$ | $\mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O}_{4}$ | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]^{-}$ | 472.35526 | 531.36911 | 1.1 | ['Alisol B', 'Alphitolic acid', 'Crataegolic acid', 'Echinocystic acid', 'Gratiogenin', 'Siaresinol', 'Sumaresinol'] |  |
| 151.00361 | 97380.4 | -0.0673335 |  | 0 |  |  |  |  | 0 |  |

### 3.3.4 Metabolites Selected for Further Analysis

The "single peak search" within MI-Pack software was used to search against the KEGG database, revealing putatively annotated matches of the observed peaks with a defined empirical formula within a 2 ppm mass error between the theoretical mass and experimental mass. Since one empirical formula could have multiple hits within that 2 ppm window, there are several putative identifications. In all cases, an attempt was made to determine which putative identifications were also possible chemical compounds of wine and are listed in the final column of Table 3.3, possible components in wine. The predominant adducts for the chosen compounds were the deprotonated ion forms as well as the acetate adducts. Greater assurance was placed on compounds identified by multiple adducts, such as the two stable isotopes of chlorine: the more intense $\mathrm{Cl}^{35}$ (75.8\%) and the lesser intense $\mathrm{Cl}^{37}$ (24.2\%). In certain cases, identification utilising one chloride isotope only $\left(\mathrm{Cl}^{37}\right)$ appeared nonsensical and those putative annotations were not included in the final list.

In several cases, the KEGG compound list included isomeric components. Hydroxybenzoic acid includes: 2- (ortho), 3- (meta), and 4- (para) hydoxybenzoic acids. Dihydroxybenzoic acids includes six compounds: 2,3- (2-pyrocatechuic acid), 2,4- ( $\beta$-resorcylic acid), 2,5- (gentisic acid), 2,6- ( $\gamma$-resorcylic acid), 3,4- (protocatechuic acid), and 3,5-dihydroxybenzoic acid ( $\alpha$-resorcylic acid). Coumaric acid includes: ortho-(2-hydroxycinnamic acid), meta- (3-hydroxycinnamic acid), and para-coumaric acid (4-hydroxycinnamic acid). An attempt was made to choose the best representative compounds for the targeted list. In some cases, the availability of the standard was the limiting factor.

The effect of alkaline fusion on the wine polymer was discussed in Chapter 1.5. The effect on anthocyanidin species was the release of a carboxylic acid and its corresponding aldehyde. The initial approach for the targeted list was to focus on the acids determined from the alkaline fusion of the laboratory aged sherds; therefore, the putative identification for aldehydes was not utilised in developing the targeted method. Also, the known volatility of the corresponding aldehydes may result in inconsistent results (Flamini and Traldi, 2010: 236). In terms of the success of the chemical attack on the samples, the results were overwhelmingly monomeric species further confirming the success of the alkaline fusion on the glass/sand laboratory aged sherds.

### 3.3.4.1 Fragmentation Confirmation of Selected Analytes

An attempt was made to confirm the majority of masses putatively identified in Table 3.3. The remaining alkaline fusion samples were pooled and samples were nanoelectrosprayed into the LTQ for 25-50 scans in negative ion mode. The chosen parent ions were focused within a 1 Dalton window in the ion trap and fragmented with a collision energy of $35 \%$ NCE with a gas pressure of 0.5 psi. The $\mathrm{Ms}^{2}$ fragment ions were then collected in the ion trap. The fragmentation procedure including isolation, fragmentation, and collection all occurred in the ion trap. Figure 3.14 shows the fragmentation spectrum of the peak with nominal mass 149, putatively identified as tartaric acid. The acids have common losses which include loss of water (18), carbon monoxide (28), and carbon dioxide (44).


Figure 3.14. The product ion spectrum of the peak with $\mathrm{m} / \mathrm{z} 149$, putatively identified as the deprotonated form of tartaric acid. The common losses are identified in the spectrum.

Table 3.5 The putatively identified acids were fragmented in order to confirm their identity. All acids have common losses. ***large loss of $32\left(\mathrm{O}_{2}\right)$

| Putative identification | Parent ion (m/z) | $-\mathrm{CH}_{3}(15)$ | $-\mathrm{H}_{2} \mathrm{O}$ <br> (18) | -CO (28) | $-\mathrm{CO}_{2}(44)$ | $-\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}(46)$ | - $\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}_{2}$ (62) | - $\mathrm{NH}_{3}(17)$ | $-\mathrm{C}_{2} \mathrm{H}_{7} \mathrm{NO}(61)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Succinic acid | 117 | xxx | 99 | xxx | 73 | xxx | xxx | xxx | xxx |
| Alpha ketobutyric acid | 101 | xxx | 73 | XXX | 57 | XXX | XXX | XXX | xXX |
| Tartaric acid | 149 | xxx | 131 | 103 | 87 | xxx | XXX | xxx | xxx |
| Malic acid | 133 | xxx | 115 | xxx | 89 | 87 | 71 | xxx | xxx |
| Citric acid | 191 | xxx | 173 | xxx | xxx | xxx | 129 | xxx | xxx |
| Dihydroxybenzoic acid <br> (possible mix with patulin) | 153 | 138 | 135 | 125 | 109 | 107 | $\boldsymbol{x} \boldsymbol{x} \boldsymbol{x}$ | $\boldsymbol{x X X}$ | $\boldsymbol{X X X}$ |
| Syringic acid | 197 | 182 | xxx | xxx | 153 | xxx | 129 | xxx | xxx |
| Isopropyl malic acid | 175 | xxx | 157 | xxx | 131 | 129 | 113 | xxx | xxx |
| Citramalic/mevalonic acid | 147 | xxx | 129 | xxx | xxx | 101 | 85 | xxx | xxx |
| Caffeic acid | 179 | 164 | xxx | 151 | 135 | xxx | xxx | xxx | xxx |
| Gallic acid | 169 | xxx | 151 | xxx | 125 | xxx | xxx | xxx | xxx |
| Coumaric acid | 163 | xxx | xxx | xxx | 119 | xxx | xxx | xxx | xxx |
| Vanillic acid | 167 | 152 | xxx | xxx | 123 | xxx | xxx | xxx | xxx |
| 2-keto-glutaramic acid | 144 | xxx | 126 | xxx | 100 | xxx | xxx | 127 | 83 |
| Carboxyvanillic acid | 211 | 196 | 193 | 183 | 167 | xxx | xxx | xxx | xxx |
| Hydroxybenzoic acid | 137 | xxx | 119 | 109 | 93 | xxx | xxx | xxx | xxx |
| Citraconic/methyl maleic acid | 129 | 114 | 111 | 101 | 85 | xxx | xxx | xxx | xxx |
| ***Methylcitric acid | 205 | xxx | 187 | xxx | 161 | 159 | xxx | xxx | xxx |

### 3.4 Conclusions

An untargeted metabolomics approach was taken in order to discover a suite of biomarkers that are predictive of the presence of aged red and white wine, and which could then be applied to archaeological samples to identify ancient wine. A model wine system was aged in the laboratory over six months in the form of liquid aliquots and as wine-permeated sherds. All samples were analysed by direct infusion FT-ICR mass spectrometry and the resultant spectra then 'stitched' together using the SIMstitch program thus producing a list of $m / z$ specific to each particular experiment. Multivariate and univariate statistical analyses were conducted on these mass spectral datasets to reveal any significant changes to the red and white wine as they age. Univariate statistics specifically monitored the loss of monomeric species as a presumed polymerisation occurred. The chemical fingerprints of the wine-permeated sherds were projected onto principal components scores plots which revealed the chemical similarities and differences between the samples, specifically the top PC loadings. In the case of the alkaline fusion aliquot, the top loadings were chosen for the next step in developing a targeted method for the analysis of archaeological samples.

The top loadings taken from the alkaline fusion aliquot (glass/sand) were searched against the KEGG database using the MI-Pack computer program. Putative identifications were based on comparison of theoretical mass values with experimental masses within a 2 ppm error. Nearly half of these ions were subsequently fragmented in an attempt to confirm their identities. Since many of these putatively identified compounds were acids, there were losses common to all of
them including loss of water, loss of carbon monoxide, and loss of carbon dioxide. Out of the 37 putatively identified acids, 20 of these acids were confirmed by MS/MS fragmentation.

It was decided that phthalic acids would not be pursued, since their origin was most probably from modern plasticizers. Also, sugar acids, such as glucuronic acid, were not pursued since the number of possible isomers for this type of compound was beyond the scope of this project. The low molecular weight of oxalic, butanoic, and lactic acid made these compounds impractical for the next phase of the research, developing a targeted LC-MS/MS method; therefore, these compounds were removed from further analysis. The remaining acids were selected for further studies and benefitted from being commercially available as standards.

In conclusion, using an untargeted high resolution mass spectrometry approach in a laboratory ageing situation, in conjunction with multivariate statistics, a suite of biomarkers was chosen based upon their characteristic significance in describing a particular sample. Those biomarkers were then used to prepare a targeted method to apply to archaeological samples and include the following acids: ketobutyric, ferulic, gentisic, 2,3-dihyhydroxybenzoic, syringic, p-coumaric, mhydroxycinnamic, vanillic, isopropyl malic, succinic, malonic, malic, tartaric, citric, gallic, and caffeic.

## 4. Development of a Targeted Analytical Method for Application to Archaeological Samples

The objectives of this chapter were to build a targeted method based upon the biomarkers discovered during the research presented in Chapter 3, as well as to refine the extraction procedures in order to determine the presence/absence of aged, polymerised wine in an archaeological sample. It was presumed that wine, if still present, was present in minute amounts. Therefore, qualitative sensitivity was paramount in this analytical approach. DIMS was discounted for sample introduction and a more sensitive chromatographic approach was chosen. DIMS is an appropriate choice for a global determination for all materials in a sample matrix, however, direct infusion of all materials concurrently creates a situation whereby salts, sample matrix, and analytes of interest vie for ionisation. Ion suppression due to a sample's high salt content was also an issue with DIMS analyses.

Chromatography utilises separation and focusing of each component, which results in enhancement of detected signals. The detector chosen for this analysis was the triple quadrupole mass spectrometer, a highly sensitive instrument that utilises two mass analysers for the identification of precursor-product ions and their relative intensities. This instrument has been used previously for similar analyses (Barnard et al. 2010; Guasch-Jane et al., 2004; Stern et al., 2008). Combining an LCtriple quadrupole approach offered unambiguous identification of analytes based upon: 1. retention time from the analytical column, 2. identification of the precursor-product ions, and 3. the comparison of the product ion spectra of archaeological samples with those obtained from commercial standards.

### 4.1 Development of a Targeted LC-MS/MS method

Reversed phase liquid chromatography (RPLC) has long been considered the most popular choice for analysis in liquid chromatography. In a recent review of chromatographic analyses of acids in wine and grape juice, 18 of 33 chromatographic techniques chosen for the identification of organic acids were RPLC (Mato et al., 2005). The remaining techniques used were either ion exclusion or ion exchange chromatography. Introduced in 1950 (Howard and Martin, 1950), separation by RPLC is based upon distribution of an analyte between the non-polar stationary phase and the mobile phase. Packing material for most reversed phase columns contain silica with C18 chains bonded to its surface thus creating an environment for hydrophobic interactions. Hydrophobic components will elute off the column only when its partition coefficient becomes more favorable towards the mobile phase which consists of an increasing amount of organic solvent. A hydrophilic component will not bind to the C18 particles and will not be retained on the column, eluting at or close to the dead volume; therefore, the reversed phase column initially chosen for the targeted analysis of organic and phenolic acids was the Synergi-Hydro column $150 \mathrm{~mm} \times 1 \mathrm{~mm}$ column with guard column. The Synergi-Hydro column attempts to create an environment for both hydrophobic and hydrophilic material by 'endcapping,' or chemically adhering a 'proprietary' polar compound to the end of the C18 chain, thereby creating an environment conducive to hydrophilic attachment.

The initial chromatographic method developed was based upon a published method for the elution of tartaric acid from a Synergi-Hydro column (Phenomenex Application Note: 14270). In this method, the mobile phase was 20 mM potassium phosphate. Since phosphates are powerful ionisation competitors and thus cannot be used in conjunction with a mass spectrometer detector, a mobile phase consisting of 10 mM ammonium acetate in HPLC grade water was prepared. Initially, the HPLC was connected to the LTQ-FT-MS using the following conditions, Table 4.1.

Table 4.1 The LC-FT-ICR-MS conditions for the elution of tartaric acid from the Synergi-Hydro column.

| Column | Synergi Hydro RP 150mmx1mm |
| :---: | :---: |
| Mobile Phase | 10 mM ammonium acetate, $\mathrm{pH}=6.8$ |
| Flow Rate | 33ul/minute |
| Detector | LTQ-FT-Mass spectrometer |
| Sheath Gas | 10 (psi) |
| Auxillary Gas | 5 (arbitrary units) |
| Spray Voltage | -3.5 (kV) |
| Capillary Temperature | $300^{\circ} \mathrm{C}$ |
| Capillary Voltage | -4.3 (kV) |

Ten milligrams of tartaric acid were weighed out and diluted in 1 ml of mobile phase and further diluted to 1:100. The dead time was monitored at approximately 3 min with the tartaric acid peak eluting at 3.80 min as seen in Figure 4.1. It was determined after several runs that the elution of the organic acid was too near the solvent front to maintain reproducibility and that reversed phase chromatography would not suit the purposes of this type of analysis of polar acids.


310712 _tatataic__cidi_3\#109-116 RT: 3.67-3.90 AV: 8 NL: 6.84E3
T: FTMS-pESF Fuli ms $70.00-800.001$


Figure 4.1. The elution of tartaric acid at $m / z 149$ using a Synergi-Hydro column $150 \mathrm{~mm} \times 1 \mathrm{~mm}$ column with guard column. The acid eluted at 3.8 min , a time close to the solvent front.

To maintain the organic and phenolic acids on an LC column and create an environment for longer retention times, another approach was required. As discussed in Chapter 1, hydrophilic interaction liquid chromatography (HILIC) is an orthogonal approach to reversed phase liquid chromatography whereby hydrophilic compounds are retained on the column and elute off into the increasingly aqueous mobile phase. Although the physical properties behind the separation in a HILIC experiment are currently under debate and not completely agreed upon (Dinh et al., 2011; Buszewski and Noga, 2012), there are known mechanisms which are considered critical for a HILIC separation.

One of the major separation mechanisms is described as the partition of the analyte between the 'interior' water-rich layer (on the surface of the polar stationary phase) and the 'exterior' organic
rich mobile phase. Aqueous buffer salts introduced into the mobile phase will naturally migrate to this aqueous rich layer thus creating a larger volume due to displacement (Guo and Gaiki, 2005). The larger volume enhances the partitioning effect of a compound between the aqueous layer and the mainly organic mobile phase (Creek et al., 2011).

A second common mechanism is described as the electrostatic interaction of the analyte with the solid phase. Depending upon the protonated/deprotonated status, elution may become dictated by an anion/cation exchange mechanism between the analyte and the stationary phase (Dinh et al. 2011; Guo et al., 2007). Based upon the type of HILIC column chosen, ions from buffer salts may bind to a charged stationary phase thereby suppressing the overall charge and limiting electrostatic interactions (Karatapanis et al., 2011). Originally bare silica columns, materials used in HILIC chromatography now include amino, amide, cyano, diol, or cyclodextrin groups on either silica or polymeric substrates (Ikegami et al., 2008; Karatapanis et al.,2011).

In order to develop a solvent gradient for the mixture of chosen acids, stock solutions were prepared by weighing out a standard acid at $10 \mathrm{mg} / 1 \mathrm{ml}$ of initial mobile phase. The final concentration was a diluted mixture based upon similar signals for all of the acids. The initial mobile phase concentrations were based on a published analysis of tartaric acid in fruit juice that used a Sequant 150x $2.1 \mathrm{~mm}(5 \mu \mathrm{~m})$ HILIC column (Ehling and Cole, 2011).

Table 4.2 Elution profiles for the separation for phenolic and organic acids, (a) published method from Ehling and Cole, 2011, (b) initial profile on capillary ZIC-HILIC column, (c) later elution profile with a change in mobile phase $A$, a shallower gradient, and a lower flow rate.

| Column | (a) ZIC-HILIC $150 \times 2.1 \mathrm{~mm}(5 \mu \mathrm{~m})$ <br> (Ehling and Cole, 2011) | (b) ZIC-HILLC $150 \mathrm{~mm} \times 300 \mu \mathrm{~m}, 3.5 \mu \mathrm{~m}$, 200Å | (c) ZIC-HILC $\underset{200 \AA}{150 \mathrm{~mm} \times 300} \mu \mathrm{~m}, 3.5 \mu \mathrm{~m}$, |
| :---: | :---: | :---: | :---: |
| Mobile Phase A | acetonitrile/water (90:10) containing $0.1 \%$ of ammonium acetate | acetonitrile/water (90:10) containing 100 mM ammonium acetate | acetonitrile |
| Mobile Phase <br> B | water containing $0.1 \%$ of ammonium acetate | 100 mM ammonium acetate | 100 mM ammonium acetate |
| Temperature | $30^{\circ} \mathrm{C}$ | $30^{\circ} \mathrm{C}$ | $30^{\circ} \mathrm{C}$ |
| Sample Volume | 10ヶl | 0.5/.8 $\mu \mathrm{l}$ | 0.5/.8 $\mu \mathrm{l}$ |
| Step 1 | 0-20 min, 0-55\% B at $0.2 \mathrm{~mL} / \mathrm{min}$ | $0-30 \mathrm{~min}, 0-50 \%$ B at $11 \mu \mathrm{~L} / \mathrm{min}$ (initially at $4 \mu \mathrm{l} / \mathrm{min}$ ) | 1-90 min, $10-60 \%$ B at $7 \mathrm{LL} / \mathrm{min}$ |
| Step 2 | $\mathbf{2 0 - 2 5 ~ m i n , ~ 5 5 \% ~ B ~ a t ~} 0.2 \mathrm{~mL} / \mathrm{min}$ | $30-33 \mathrm{~min}, 50 \% \mathrm{~B}$ at $11 \mu \mathrm{~L} / \mathrm{min}$ | 90-105 min, 60\% B at $7 \boldsymbol{\mu L} / \mathrm{min}$ |
| Step 3 | 25-38 min, $0 \%$ B at $0.6 \mathrm{~mL} / \mathrm{min}$ | 33-36 min, $10 \%$ B at $11 \mu \mathrm{~L} / \mathrm{min}$ | 105-115 min, $10 \%$ B at $7 \mu \mathrm{~L} / \mathrm{min}$ |
| Step 4 | 38-40 min, $0 \%$ B at $0.2 \mathrm{~mL} / \mathrm{min}$ | $36-41 \mathrm{~min}, 10 \%$ B at $11 \mu \mathrm{~L} / \mathrm{min}$ | 115-135 min, $10 \%$ B at $7 \mu \mathrm{~L} / \mathrm{min}$ |

Early attempts to separate and resolve a mixture of standard acids on a capillary ZIC-HILIC $150 \mathrm{~mm} \times 300$ $\mu \mathrm{m}, 3.5 \mu \mathrm{~m}, 200 \AA ̊$ column proved unsuccessful after multiple attempts at changing the flow rate, steepness of gradient, and mobile phase A (Table 4.2). However, during this time it was determined that increasing the concentration of mobile phase $B(100 \mathrm{mM}$ ammonium acetate) separated the acids into two classes: the early eluting phenolic acids and the later eluting organic acids. Later LC development of this initial mobile phase composition was based primarily on the increasing concentration of the strong eluting solvent, water.

Due to the difficulty in maintaining a constant pressure at capillary flow rates, a larger diameter analytical column was chosen for method development, an Agilent Zorbax column, $150 \times 2.1 \mathrm{~mm}$ column and guard column. The stationary phase was bare silica that was 'endcapped' with proprietary compounds in order to deactivate the majority of the silanol groups. The pKa of the silanol group is 4.9 and therefore imparts a negative charge at pH 6.8 , the pH of ammonium acetate (Agilent Zorbax column selection guide for HPLC, 2007). Unless attenuated by either the ammonium buffer cations or the water in the mobile phase (Hemström and Irgum, 2006), the anionic charged surface will repel the acidic analytes. If these conditions of deactivation are met (buffer salt and water), the chromatographic separation and elution of organic acids off the silica column would most likely be described by partition between the water-rich layer and the acetonitrile mobile phase with limited involvement of the silica solid phase.

In preparation for the next phase in method development, discussions with a colleague resulted in a ternary delivery system of solvents thereby maintaining a constant level of buffer salts throughout the
gradient (Dr. Ulf Sommer, personal communication). By maintaining this consistent concentration, the electrostatic effect between the buffer salts and the stationary phase did not change throughout the run resulting in a stable baseline.

The LC was connected to the mass spectrometer by a heated electrospray ionisation source (HESI). The conditions of the HESI source (voltage applied and nitrogen gas delivered) are critical in order to produce the charged droplets necessary for mass spectral detection. The conditions were chosen to increase the signal of several chosen ions and to maintain an excellent spray stability (no deviations beyond $2 \%$ ) when collected for two minutes. Ten microlitres of standard mix (gentisic, malic, and citric acids) were added to a sample well and $1 \mu$ l of the sample was injected onto the column in full injection mode.

Figure 4.2 shows a total ion chromatogram (TIC) with three extracted ion chromatograms (EIC) for gentisic, malic and citric acid. The first two acids are well-resolved. Citric acid, however, was problematic with poor focusing on the column. Two possible reasons for the broad peak of citric acid were: interference/chelation with iron in the stainless steel frits within the LC system (Alpert, 1990; Preinerstorfer et al., 2010) or that the pH of the mobile phase was below the compound's pKa resulting in a greater concentration of protonated species (vs deprotonated species). There were no obvious stainless steel components found on the liquid chromatograph, therefore, it was decided to raise the pH of the mobile phase to ensure full deprotonation (Aronson, 1983).
E:iTSQ_datalse ptember_2012 mbx_10mM 9/15/2012 11:45:16 AM
(a)

(b)

(d)


Figure 4.2. Gradient elution of a mixture of three acids on an Agilent Zorbax column $150 \times 2.1 \mathrm{~mm}$ (a) full TIC (b) extracted ion chromatogram (EIC) of base peak 153 representing gentisic acid (c) EIC of base peak 133, representing malic acid and (d) EIC of base peak 191, representing citric acid.

### 4.1.1 pKa of chosen metabolites

The measurement of pKa is the pH at which the deprotonated and fully protonated concentrations of a weak acid are in equilibrium. At pH higher than the pKa , the concentrations of the deprotonated compounds are greater, at pH below the pKa , the fully protonated acid prevails. Polyprotic acids such as citric acid have several pKas depending upon the number of protons available for donations.

### 4.1.2 ZIC-pHILIC column

The maximum pH range of the Agilent Zorbax column is 1-8. Therefore, in order to increase the mobile phase above pH 8, a separate column was chosen, the ZIC-pHILIC $150 \mathrm{~mm} \times 2.1 \mathrm{~mm}, 5 \mu \mathrm{~m}$ column from SeQuant. The polymeric stationary phase support is stable from pH 2-12. Attached to the polymer is a sulfoalkylbetaine, Figure 4.3. The quaternary ammonium and the sulfonic group are present in a 1:1 ratio for a total net charge of 0 ; however, due to the physical accessibility of the negative charge, and the physically hindered positive charge, the stationary phase yields a slightly negative charge (Jandera, 2011).


Figure 4.3. The zwitterionic stationary phase bound to the polymeric substrate on a ZIC-pHILIC column. The positively charged quarternary ammonium and the negatively charged sulfonate group together have a net charge of zero.

The sulfoalkylbetaine support provides a strong attraction for water resulting in a measurable layer of stagnant or slow moving water along the polymer layer (Jandera, 2011). Additional salts from the mobile phase are attracted to the cationic and anionic charges on the polymer support. The result is an attenuation of electronic attraction between solid support and analytes of interest as well as polymer swelling (Terayama et al., 2009). The increased polymer volume increases the volume of the aqueous layer which may increase the retention of the hydrophilic materials on the column (Melnikov et al., 2013).

### 4.1.3 Repeatability Study to Determine Retention Time Robustness

The gradient developed using the Zorbax column was applied to the ZIC-pHILIC column with several changes. The steepness of the gradient was lessened and the maximum percentage of water maintained at $50 \%(\mathrm{v} / \mathrm{v})$. Mobile phase B was prepared as 100 mM of ammonium acetate in 400 ml of HPLC grade water with the pH raised to 8.2 with the addition of a few drops of ammonium hydroxide. The pH was initially checked in the fume hood with a strip of pH paper and confirmed with a pH meter.

In Figure 4.4, two species were identified at $m / z 191$, eluting at 4 min and at 30 min . This suggests a dual citric acid species whereby the sharp peak eluting at 4 min probably represents an initially fully protonated species of citric acid partitioning into the acetonitrile and eluting early in the analysis. The species of interest eluting at 30 min is still broad and unfocused. Because there were inherent problems with the analysis of citric acid, the acid was not pursued in the analysis of the archaeological samples. The analysis of caffeic acid and gallic acid were also problematic; the acids were not properly focused on the column and therefore, were not pursued in the analysis of the archaeological samples. Although not
identified as a biomarker in Chapter 3, ascorbic acid was added to the mixture as an internal standard, to assess whether the retention times of the acids varied.

Table 4.3 Parameters of the two analytical ZIC-HILIC columns chosen for the analyses of the organic acids. The final column and parameters chosen for further analyses are for the SeQuant ZIC-pHILIC column.

| Column | Agilent Zorbax column, $150 \times 2.1$ mm | SeQuant ZIC-pHILIC column, $150 \times 2.1 \mathrm{~mm}, 5 \boldsymbol{\mu m}$ |
| :---: | :---: | :---: |
| Mobile Phase A | Acetonitrile | Acetonitrile |
| Mobile Phase B | 100 mM ammonium acetate, $\mathrm{pH}=6.8$ | 100 mM ammonium acetate, $\mathrm{pH}=8.2$ |
| Mobile Phase C | Water | Water |
| Temperature | $30^{\circ} \mathrm{C}$ | $30^{\circ} \mathrm{C}$ |
| Sample Volume | $1 \mu \mathrm{l}$ | $1 \mu \mathrm{l}$ |
| Step 1 | Time: 0-1 min 10\%B, 0\%C | Time 0-7 min 10\%B, 0\%C |
| Step 2 | Time: 1-15 min 10\%B, 60\%C | Time: 7-21 min 10\%B, 50\%C |
| Step 3 | Time: 15-23 min 10\% B, 60\% C | Time: 21-29 min 10\%B, 50\%C |
| Step 4 | Time: 23-25 min 10\%B, 0\% C | Time 29-33 min 10\%B, 0\%C |
| Step 5 | Time: 25-35 min 10\%B, 0\%C | Time: 33-35 min 10\%B, 0\%C |
| Flow Rate | 70رl/min (initially 100) | 70رl/min |
| Detector | TSQ mass spectrometer | TSQ mass spectrometer |
| Electrospray voltage (eV) | -3000 | -3600 |
| Vaporization temperature, ${ }^{\circ} \mathrm{C}$ | 0 | 0 |
| Sheath gas (psi) | 10 | 10 |
| Auxiliary gas (arbitrary units) | 0 | 0 |
| Capillary temperature, ${ }^{\circ} \mathrm{C}$ | 275 | 275 |
| Detection | Full scan in Q1, m/z 40-300 | Full scan in Q1, m/z 40-300 |

(a)

(b)

(c)

(d)

(e)

(f)

(g)

(h)

(i)

(j)

(k)

(I)


Citric acid

## Tartaric acid

## Malic acid

## Malonic acid

## Succinic acid

Isopropyl malic acid, *Ascorbic acid

## Vanillic acid

## M-coumaric acid, *p-coumaric acid

## Syringic acid

## 2,3-DHB acid, * 2,5-DHB (gentisic) acid

Ferulic acid

## Ketobutyric acid

Figure 4.4. Extracted ion chromatograms of 15 acids analyzed in a retention time study (a)EIC of base peak $m / z$ 191, citric acid (b) $\mathrm{m} / \mathrm{z} 149$, tartaric acid (c) $\mathrm{m} / \mathrm{z} 133$, malic acid (d) $\mathrm{m} / \mathrm{z} 103$, malonic acid (e) $\mathrm{m} / \mathrm{z} 117$, succinic acid (f) $\mathrm{m} / \mathrm{z} 175$, ascorbic and isopropyl malic acid (g) m/z 167, vanillic acid (h) m/z 163, p-coumaric acid and m-hydroxycinnamic acid (i) m/z 197, syringic acid (j) m/z 153, gentisic acid and 2,3-dihyhydroxybenzoic acid (k) m/z 193, ferulic acid (I) m/z 101, ketobutyric acid.

### 4.1.3.1 Relative Standard Deviation of Results

In order to determine the robustness of this method, a standard mix of acids was injected 31 times over the course of 36 hours. In order to develop a gradient run for the mixture of chosen acids, stock solutions were prepared by weighing out a standard acid at $10 \mathrm{mg} / 1 \mathrm{ml}$ of initial mobile phase. The final concentration was a diluted mixture based upon similar signals for all of the acids. Stock solutions were diluted in $90 / 10$ acetonitrile/ 100 mM ammonium acetate, pH 8.2 , with the addition of $20 \mu \mathrm{l}$ of potassium chloride ( $\mathrm{KCl}, 25 \mathrm{mg} / 1 \mathrm{ml} 100 \mathrm{mM}$ ammonium acetate, pH 8.2 ). KCl , a byproduct of the alkaline fusion reaction was added to in order to gauge its effect on retention time reproducibility.

The RSD calculated from the retention times are shown in Figure 4.5. The results represent a high repeatability of the retention times of the standards over the course of several days. Blanks were run periodically during the 36 hour period in order to determine the extent of carryover from the column. There was small carryover from the dihydroxybenzoic and succinic acids. There was also a 'ghosting' effect in malic acid and tartaric acid; that is, there was carryover present in the blanks eluting one minute later than the elution of the standard acid. Because of this effect, the identification of malic and tartaric acid in the archaeological samples was dependent upon narrow retention time windows.


Figure 4.5. The RSD of retention time with error bars included derived from 31 consecutive injections of standard acid mixtures onto the ZIC-pHILIC LC MS/MS over 36 h . The y axis is the retention time in minutes.

### 4.1.4 Determination of Transitions for Chosen Metabolites

After the elution parameters were finalised, the precursor-product ion transitions were determined.
The final targeted method is described in Chapter 2. Initially, the instrument was optimised for maximum signal and optimal electrospray stability. Each acid was then diluted in ACN/water (depending upon their elution from the column) and directly infused into the mass spectrometer. The automated procedures included the following steps:

1. The S-lens voltage was optimised for the particular parent ion.
2. The collision gas was stabilised in Q2 at 1.5 mTorr .
3. The product ions were collected through an 8 step collision energy starting from 5 ev to 50 eV

The final output of relative intensity vs. optimised collision energy for the product ions of each acid standard are in the Appendix, Chapter 4, Figures 1a-1p. The transitions between the precursor ion and the most intense product ions were transferred to the targeted method. These transitions were limited to the elution time from the ZIC-pHILIC column, such that the final targeted method contained four segments: segment 1 from 0-17 min collected in full scan mode, segment 2 collected from 17-23 min in selected reaction monitoring (SRM) mode for 8 transitions of the early eluting acids, segment 3 from 2333 min collected in SRM for the transitions of the later eluting acids, and segment 4 analysed at full scan from 33-43 min. The final identification of an acid in an archaeological sample was based on three criteria:

1. retention time
2. precursor-product ion transition
3. comparison between sample fragment intensities with standard fragment intensities

### 4.2 Extraction Procedures for Archaeological Samples

As introduced in Chapter 1, the original extraction method of 100 mg for laboratory sherds was increased to 1 g for archaeological samples due to the limited amount of analyte. The solvent volumes were also increased to take into account the larger sample. One hundred sixty microlitres of 4 M KOH were added to the laboratory sherds allowing enough liquid to cover the 100 mg (Zsuga and Kiss, 1987). This amount was diluted to $600 \mu \mathrm{l}$ of 1 M KOH , enough material to cover 1 g of archaeological samples. The samples chosen to optimise the extraction procedures were the wine press and its preparatory layer, as this object was considered the 'positive control' for the remnants of ancient wine and recorded in Appendix, Chapter 4, Table 1. In an attempt to reproduce several previously reported standard
operating procedures (SOPs) and define the most efficient and sensitive extraction for the analysis of archaeological wine, several methodological approaches were tried on the samples (Pecci et al., 2013; Stern et al., 2008; Barnard et al., 2010; Guasch-Jane et al., 2004; McGovern et al., 2009).

### 4.2.1 Initial extraction and Analysis: Alkaline Fusion

In order to determine if the length of alkaline fusion was critical in breaking down the presumably aged and oxidised polymers, samples of the wine press and its preparatory layer were extracted using a modified SOP. Briefly, 1 g of sample was homogenised in a 7 ml Precellys homogenisation tube with 5.6 ml of methanol/water. The slurry was transferred to a 15 ml glass centrifuge tube to which was added 4 ml of chloroform and 2 ml of water. The tube was vortexed and centrifuged. The top polar layer was removed as was the bottom non-polar layer; these extracts were not analysed as this simple solvent extraction was assumed not to be sufficient for isolating polymerised material. The remaining solid was allowed to dry overnight in a fume hood. Six hundred microlitres of 1 M KOH were added to the solid powder to break down the polymer; the sample was then vortexed for 103 -second bursts. A sample of the preparatory layer was then placed in a water bath at $50^{\circ} \mathrm{C}$ for I h . Separate samples of the preparatory layer and wine press were placed in a water bath at $50^{\circ} \mathrm{C}$ for I h followed by heating in a water bath in an oven at $50^{\circ} \mathrm{C}$ overnight. Afterward, each sample was then acidified with 2.5 ml 2 N HCl in order to drop the pH below 3 (checked with pH paper). Three millilitres of ethyl acetate were then added to the centrifuge tube to extract the monomeric species and the tube vortexed and centrifuged at 500 rpm for 3 min . The top organic layer was then continuously added to a 1.8 ml champagne vial and dried down under a stream of nitrogen. The sample vials were capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

For analysis by LC/MS/MS, the samples were removed from the freezer and warmed to room temperature. Thirty microlitres of mobile phase $\mathrm{A}, 90 / 10 \mathrm{ACN} / 100 \mathrm{mM}$ ammonium acetate, were added to the sample vials. Ten microlitres of each sample were added to the sample well; $1 \mu \mathrm{l}$ of sample was added to the LC column. All results are recorded in Appendix; Table 4.2. Figure 4.6 graphically represent the results based upon absence or presence of the acids. Five acids were identified in the prep layer following a 1 h KOH application. Three acids were identified in the preparatory layer after heating for 24 h in KOH ; the extended period of time for the alkaline fusion did not increase extraction efficiency, therefore, the alkaline fusion portion of the extraction was maintained for 1 h .

### 4.2.2 Continuation of Extraction and Analysis Procedures

In order to further define the best extraction procedure, four samples were prepared for analysis: wine press, wine press spiked with syringic and p-coumaric acids, preparatory layer, and extraction blank. Samples were extracted as before (Appendix, Table 4.1). After the samples were acidified with 2.5 ml of 2 N HCl , a clean-up step was added utilising a C18 solid phase extraction (SPE) cartridge. At the acidic pH , the majority of acids were fully protonated and more likely to adhere to the hydrophobic C18 cartridge.

Briefly, a C18 100 mg bed SPE cartridge was conditioned with 1 ml of methanol followed by 1 ml of HPLC grade water. One millilitre of acidified sample was added to the cartridge and the remaining sample was stored in the cold room at $4^{\circ} \mathrm{C}$. The flow through was collected in a 1.5 ml Eppendorf tube. The material remaining on the SPE cartridge was eluted off with $3 \times 1 \mathrm{ml}$ using solvents increasing in non-polarity: water, methanol, and ethyl acetate. All three eluates were collected for MS analysis. The water and methanol samples were dried in the Speed Vac overnight. The ethyl acetate portion was immediately
transferred to glass vials and dried down with a stream of nitrogen. All dried samples were stored at $80^{\circ} \mathrm{C}$ until analysis. The remaining samples stored in the cold room were extracted five days later with 1 ml of ethyl acetate, vortexed and centrifuged. The top organic layer was transferred to a glass vial and dried down under nitrogen. The dried samples were stored at $-80^{\circ} \mathrm{C}$ until analysis.

For analysis by LC/MS/MS, samples representing the three eluates from the SPE cartridge, as well as the ethyl acetate liquid extraction were removed from the freezer and warmed to room temperature. Thirty microlitres of mobile phase $90 / 10, \mathrm{ACN} / 100 \mathrm{mM}$ ammonium acetate were added to the sample vials. The vials were vortexed and centrifuged and $10 \mu \mathrm{l}$ of sample were added to the sample well; $1 \mu \mathrm{l}$ of sample was added to the LC column. All results are recorded in Appendix, Table 4.2.

Comparing the result of the extraction procedures, the methanol elution from the C18 SPE cartridge was moderately successful based on the six acids that were identified: syringic, vanillic, p-coumaric, succinic, and malonic acids, plus ferulic acid (wine press), and malic acid (preparatory layer), Figure 4.7. In contrast, Figure 4.8, eight acids were identified in the ethyl acetate liquid extraction of the preparatory layer: syringic, vanillic, p-coumaric, succinic, malonic, ferulic, malic, and tartaric acids. The results from the LC-MS/MS analysis of the preparatory layer/ethyl acetate extraction are seen in Figures 4.9 and 4.10. The supplementary data for this result is located in the Appendix, Chapter 4, Figure 2 (TIC), and Figure 3a-3h (EIC and product ions). Four of the identified compounds have multiple peaks within the chosen fragmentation window; the final identification was confirmed with the compounds' retention time and fragment intensities. There was contamination in both of the extraction blanks including succinic acid, as well as a low level of malonic acid. From these results, the most successful extraction procedure is the ethyl acetate liquid-liquid extraction that includes an extended period in acidic solution
suggesting a rate limiting step in the extraction procedure; that is, once the polymer is broken apart, a period of time is necessary for the liberated acids to migrate into the aqueous layer.

In an effort to duplicate this extraction procedure, a separate portion of the wine press was extracted and stored in the cold room for forty-eight hours at $4^{\circ} \mathrm{C}$ in its acidified aqueous layer. The samples were then extracted and analysed; six acids were successfully identified from this analysis. From these results, it was decided that a period of time should elapse between the acidification and the final extraction with ethyl acetate. Rather than store the samples in the cold room, a slight rise in temperature was accomplished by leaving the samples at room temperature for 24 h .

 results indicate that alkaline fusion for 1 hour and 24 hours depolymerise the wine residues to the same extent.


Figure 4.7. Presence or absence of acids from the methanol elution from the C18 SPE cartridge. Six acids were identified from the prep layer including: malic, vanillic, pcoumaric, syringic, and ferulic. Malonic acid was identified; however, a small amount of malonic acid was also identified in the two extraction blanks


Figure 4.8. Presence or absence of acids from the ethyl acetate liquid extraction. Eight acids were identified, including: tartaric, malic, malonic, succinic, vanillic, p-coumaric, syringic, and ferulic acids. Low levels of succinic acid and malonic acid were also identified in the extraction blanks.


Figure 4.9. Extracted ion chromatograms of $4 / 8$ acids identified in the preparatory layer of a Punic Sardinian wine press, (a) $\mathrm{m} / \mathrm{z} 197$, syringic acid (b) $\mathrm{m} / \mathrm{z}$ 193, ferulic acid (c) $\mathrm{m} / \mathrm{z} 167$, vanillic acid and (d) $\mathrm{m} / \mathrm{z} 163$, p -coumaric acid. The retention times for ferulic and vanillic acids are highlighted with an asterisk.
(a)

(b)

(c)

(d)
RT: $0.00-43.00$

Figure 4.10. Extracted ion chromatograms of $4 / 8$ acids identified in the preparatory layer of a Punic Sardinian wine press, (a) $\mathrm{m} / \mathrm{z} 149$, tartaric acid (b) $\mathrm{m} / \mathrm{z} 133$, malic acid (c) $\mathrm{m} / \mathrm{z} 117$, succinic acid and (d) $\mathrm{m} / \mathrm{z} 103$, malonic acid. The retention times for tartaric and succinic acids are highlighted with an asterisk.

### 4.3 Conclusions

Based upon the biomarkers determined from Chapter 3, a HILIC LC elution profile was successfully developed for the targeted analysis of fourteen phenolic and organic acids in an attempt to accurately describe the fingerprint of an aged, polymerised wine. These 14 acids are: ketobutyric, ferulic, gentisic, syringic, m-hydroxycinnamic, p-coumaric, 2,3-dihydroxybenzoic, vanillic, ascorbic, succinic, malonic, malic, tartaric, and isopropyl malic acid. Several extraction procedures in conjunction with the targeted LC/MS/MS method was applied to the Sardinian wine press and its preparatory layer, utilised as the archaeological 'positive control.' Eight acids were conclusively identified in the sample based upon each compound's retention time in combination with their fragmentation pattern. The most successful extraction procedure, based upon the number of acids identified, was the procedure which allowed the samples to sit for a period of time in an acidic medium prior to extraction with organic solvent. The final method was described in Chapter 2.2.1. This suggests a rate limiting step allowing for the depolymerised monomers to migrate into the aqueous environment. As described in Chapters 2 and 4, this new extraction method and targeted analytical approach was then applied to the analysis of transport amphorae and the wine press from the Sardinian archaeological site, as well as artifacts from the Roman British site of Vindolanda, in order to determine the presence of wine.

## 5. Applying a Targeted Approach to Archaeological Samples from Sardinia Italy

The objective of this chapter is to apply a targeted analytical approach that exploits the metabolomics-based discoveries from Chapter 3 to samples taken from the Punic farmstead, Trunce $e^{\prime}$ Molas, in Sardinia, Italy. This area has been actively growing grapes and preparing wine for several millennia. Samples include a double basin wine press as well as sherds representative of transport amphorae, identified by the archaeologists from the Terralba rural research project. This research will apply a predetermined suite of biomarkers that represent aged, polymerised wine to archaeological artifacts in order to describe the presence or absence of wine. In addition, the results will be compared with known standards in a further attempt to classify aged wine as more representative of white wine or red wine. A classification may assist archaeologists in their description of the objects excavated from this site.

### 5.1 Archaeological Sample Extraction

Table 5.1 in the Appendix represents a list of the archaeological sherds that were sampled and extracted from Trunce e' Molas in Sardinia, Italy. There were a total of 30 archaeological samples and one extraction blank. Sample identification was broken down as follows, Table 5.1.

Table 5.1 Identifications of the samples taken for analysis from the Punic farmstead, Trunce e' Molas, Sardinia, Italy.

| Sample Identification | Fabric | Location | Samples analyzed | lab identifier |
| :---: | :---: | :---: | :---: | :---: |
| TM.003.3 | cocciopesto /earthenware | Stone structure with plastered interior surfaces. Top earthenware preparatory layer. | 5 | wine press A |
| TM.027.3 | limestone /earthenware | Limestone monolith | 5 | wine press B |
| TM.003.3 | earthenware | Separated preparatory layer | 5 | preparatory layer |
| 1.TM.044.1.3 | B |  | 1 | neck |
| SAR.TM. 67 | B |  | 1 | neck |
| 1.TM.028.1.33 | B |  | 1 | neck |
| 1.TM.060.1.15 | B |  | 1 | neck |
| 1.TM.042.1.7 | B |  | 1 | neck |
| 1.TM.033.1.14 | B |  | 1 | side wall |
| SAR.TM. 4 | B |  | 1 | side wall |
| 1.TM.034.1.20 | B |  | 1 | side wall |
| 1.TM.033.1.15 | B |  | 1 | side wall |
| 1.TM.060.1.45 | B |  | 1 | side wall |
| 1.TM.060.1.56 | B |  | 1 | base |
| 1.TM.035.1.1 | B |  | 1 | base |
| 1.TM.036.1.2 | B |  | 1 | base |
| 1.TM.028.1.31 | B |  | 1 | base |
| 1.TM.033.1.20 | B |  | 1 | base |



Figure 5.1. Diagram illustrating the sample locations of the 15 sherds taken from transport amphorae


Figure 5.2. Photograph taken of the wine press TM.003.3 excavated from Trunce $e^{\prime}$ Molas, with its upper preparatory layer.

Thirty one samples were prepared for analysis: wine press (preparatory layer+mortar) $n=10$, preparatory layer only $n=5$, amphorae neck $n=5$, side wall of amphorae $n=5$, amphorae base $n=5$, and extraction blank n=1, Figures 5.1 and 5.2. The exact mass of each sample was recorded in the samples extraction log sheet located in the Appendix Table 5.1. As stated in the Chapter 1, the wine press and all sherds were excavated from Phase II stratigraphy by the archaeologists of the Trunc'e Molas rural research project and represent different amphorae. The distinction between neck, side wall, and base were categorised by the same archaeologists of the Terralba research project. Although there was mixing between Phase I and the upper portion of Phase II, the samples here examined were considered secure in the ancient timeline of $4^{\text {th }}$ and $2^{\text {nd }}$ century B.C.E. habitation. Also, the wine press which was discovered in situ, was well below the topsoil and modern plough lines; abrasions from modern agricultural work were also absent from the press (person communication: P. van Dommelen).

The archaeological sample extraction method is described in Chapters 1 and 2. Briefly, one gram of sample from each of the sherds was removed with a tile cutter and placed in a 7 ml Precellys homogenisation tube. Samples were homogenised and extracted using a modified Bligh-Dyer biphasic extraction procedure resulting in a polar and a non-polar extract (Bligh and Dyer, 1959). The remaining solids were allowed to dry overnight in a fume hood. After drying, $600 \mu \mathrm{l}$ of 1 M KOH were added to the sample and heated for one hour at $50^{\circ} \mathrm{C}$ in a water bath (Zsuga and Kiss, 1987). The samples were removed from the bath, cooled to room temperature, and then acidified to $\sim \mathrm{pH} 2$ with 2.5 ml of 2 N HCl . The pH was checked by strips of pH paper for randomly chosen samples. The samples vials were allowed to sit for 24 h at room temperature after which the samples were extracted with 3 ml of ethyl acetate. The samples were vortexed and centrifuged. The organic layer was then transferred to a 1.8
ml champagne vial with a glass pipette and blown down under a stream of nitrogen; the vials were refilled and dried down at least twice. All samples were capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

### 5.2 LC-MS/MS Analysis of the Archaeological Samples

The extracted samples were removed from the freezer and reconstituted in $30 \mu \mathrm{l}$ of mobile phase $90 \% \mathrm{ACN} / 10 \% 100 \mathrm{mM}$ ammonium acetate, $\mathrm{pH}=8.2$. The samples were vortexed for 10 s and then centrifuged for 20 min at $10^{\circ} \mathrm{C}$ at 6000 rpm . Ten microlitres of sample were added to the 96 well plate and $1 \mu$ l of sample was injected onto the column for analysis. Sample 1.TM.042.1.7 from the neck was diluted 1:5 (v/v) due to its viscous nature.

Table 5.2 The absolute intensities of the acids identified by LC-MS/MS from the initial targeted analysis on the Sardinian wine press, preparatory layer and 15 sherds representing transport amphorae.

| Sample Name | ketobutyric acid | ferulic acid | gentisic acid | syringic acid acid | coumaric acid | 2,3dihydroxybenzoic acid | vanillic acid | succinic acid | malonic acid | malic acid | tartaric acid | isopropyl malic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| standard mixture | 653221 | 1.00E+07 | $2.22 \mathrm{E}+09$ | 5889038 | 123447807 | 683912170 | 815101 | 2476835 | 653221 | 4.00E+06 | 1712784 | 650250 |
| blank | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 18325 | 3017 | 170716 | 41249 | 0 |
| extraction blank | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| QC1 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 9519 | 0 | 77648 | 0 | 0 |
| QC2 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 11896 | 0 | 56699 | 0 | 0 |
| QC3 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 36223 | 0 | 0 |
| SAR.TM. 67 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 4242 | 0 | 21638 | 0 | 0 |
| 1.TM.028.1.33 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 5349 | 0 | 9840 | 0 | 0 |


| Sample Name | ketobutyric acid | ferulic acid | gentisic acid | syringic acid | pcoumaric acid | 2,3dihydroxybenzoic acid | vanillic acid | succinic acid | malonic acid | malic acid | tartaric acid | isopropyl malic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1.TM.060.1.15 | 0 | 0 | 0 | 931 | 0 | 0 | 0 | 5165 | 0 | 5765 | 0 | 0 |
| $\begin{aligned} & \text { 1.TM.042.1.7 } \\ & (1: 5) \end{aligned}$ | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 36460 | 0 | 8033 | 0 | 0 |
| 1.TM.033.1.14 | 0 | 1276 | 0 | 1822 | 0 | 0 | 0 | 5605 | 0 | 4187 | 0 | 0 |
| SAR.TM. 4 | 0 | 2246 | 0 | 0 | 0 | 0 | 0 | 10481 | 0 | 7186 | 0 | 0 |
| 1.TM.033.1.15 | 0 | 1130 | 0 | 614 | 0 | 0 | 0 | 16276 | 0 | 7736 | 0 | 0 |
| 1.TM.060.1.45 | 0 | 956 | 0 | 1968 | 0 | 0 | 0 | 19654 | 4010 | 3397 | 0 | 0 |
| 1.TM.060.1.56 | 0 | 1532 | 0 | 842 | 0 | 0 | 0 | 100792 | 11109 | 2193 | 0 | 0 |
| 1.TM.035.1.1 | 0 | 1714 | 0 | 628 | 0 | 0 | 0 | 5286 | 0 | 0 | 0 | 0 |
| 1.TM.036.1.2 | 0 | 1469 | 0 | 0 | 0 | 0 | 0 | 4988 | 0 | 2039 | 0 | 0 |
| 1.TM.028.1.31 | 0 | 1123 | 0 | 1207 | 0 | 0 | 0 | 306697 | 4657 | 1892 | 0 | 0 |


| Sample Name | ketobutyric acid | ferulic acid | gentisic acid | syringic acid | pcoumaric acid | 2,3- <br> dihydroxybenzoic acid | vanillic <br> acid | succinic acid | malonic acid | malic acid | tartaric acid | isopropyl malic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1.TM.033.1.20 | 0 | 994 | 0 | 840 | 0 | 0 | 0 | 7257 | 0 | 2971 | 0 | 0 |
| blank | 0 | 1192 | 0 | 0 | 0 | 0 | 0 | 4535 | 0 | 2629 | 0 | 0 |
| QC4 | 0 | 2043 | 0 | 1055 | 0 | 0 | 0 | 17350 | 4981 | 1492 | 0 | 0 |
| QC5 | 0 | 1262 | 0 | 1012 | 0 | 0 | 0 | 18095 | 4656 | 2408 | 0 | 0 |
| WP2_TM. 003 | 0 | 2096 | 0 | 7197 | 8523954 | 0 | 5662 | 18897 | 0 | 1604 | 0 | 0 |
| WP3_TM. 003 | 0 | 5046 | 0 | 12103 | 12100482 | 0 | 8615 | 8246 | 1852 | 2055 | 0 | 0 |
| WP4_TM. 003 | 0 | 1770 | 0 | 3692 | 2085196 | 0 | 1922 | 80435 | 137095 | 0 | 0 | 0 |
| WP5_TM. 003 | 0 | 3464 | 0 | 11788 | 10440534 | 0 | 7335 | 229599 | 0 | 0 | 0 | 0 |
| WP6_TM. 027 | 0 | 0 | 0 | 4574 | 9070793 | 0 | 5121 | 7257 | 0 | 0 | 0 | 0 |
| WP7_TM. 027 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 4189 | 0 | 0 | 0 | 0 |


| Sample Name | ketobutyric acid | ferulic acid | gentisic acid | syringic acid | coumaric acid | 2,3- <br> dihydroxybenzoic acid | vanillic acid | succinic acid | malonic acid | malic acid | tartaric acid | isopropyl malic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| WP8_TM. 027 | 0 | 597 | 0 | 4292 | 9242684 | 0 | 5186 | 3627 | 0 | 0 | 0 | 0 |
| WP9_TM. 027 | 0 | 0 | 0 | 5756 | 14367540 | 0 | 10146 | 4760 | 0 | 0 | 0 | 0 |
| WP10_TM. 027 | 0 | 0 | 0 | 3132 | 13425446 | 0 | 5393 | 3588 | 0 | 0 | 0 | 0 |
| prep1 | 0 | 2558 | 0 | 19667 | 8379605 | 0 | 13461 | 225784 | 1764 | 0 | 0 | 0 |
| prep2 | 0 | 3524 | 0 | 20855 | 9404386 | 0 | 15741 | 2635324 | 5035 | 0 | 0 | 0 |
| prep3 | 0 | 4582 | 0 | 21594 | 5821124 | 0 | 14821 | 2565120 | 45942 | 0 | 0 | 0 |
| prep4 | 0 | 3845 | 0 | 14496 | 25262847 | 0 | 19413 | 242615 | 1634 | 0 | 0 | 0 |
| prep5 | 0 | 2693 | 0 | 9569 | 6032407 | 0 | 5045 | 33825 | 864 | 0 | 0 | 0 |
| Standard mix $(1: 2)$ | 5095282 | $9.00 \mathrm{E}+06$ | 98161000 | 2432432 | $1.005 \mathrm{E}+09$ | NA | 317863 | 306386 | 136119 | 0 | 0 | 0 |



Figure 5.3. This diagram compares the absolute intensities of the phenolic acids identified in the targeted analysis of 15 samples from a wine press and 15 samples of transport amphorae excavated from a Punic farmhouse in Sardinia, Italy.


- malic acid
- tartaric acid

■ isopropyl malic acid

Figure 5.4. This diagram illustrates the fate of three organic acids during this analyses. There is an obvious decline in malic acid as the runs progress. The reason for this loss of intensity is described in the chapter.

### 5.2.1 Results of Targeted Analysis

Table 5.2 shows the results of the absolute intensities of the LC-MS/MS results of the targeted analysis of the wine press, its preparatory layer and 15 amphorae sherds. Seven acids out of fourteen from the biomarker list were identified in the group of thirty archaeological samples, these included malic, malonic, succinic, vanillic, p-coumaric, syringic, and ferulic. Integration of the peaks was performed automatically by Xcalibur software; in certain cases, manual integration was necessary to maintain baseline to baseline continuity. Figure 5.3 compares the absolute intensities of the phenolic acids identified in the targeted analysis of samples. The highest intensities were identified in the wine press and its preparatory layer; acids were virtually absent from the sherds representing the neck, except for syringic acid identified in sample 1.TM.060.1.15. Ferulic and syringic acids were identified in low levels in several of the side walls, as well as the bases. The increased intensity of these six acids: ferulic, syringic, vanillic, p-coumaric, succinic, and malonic, in the wine press and in the preparatory layer, strongly suggest the presence of an aged, polymerised wine. In contrast, due to the lack of consistent grouping of biomarkers within each sample in conjunction with the low intensity of acids identified, indicate an absence of wine in the amphorae sherds. This result highlights the precarious nature of using 1 or 2 biomarkers for the identification of an organic residue within an archaeological object.

### 5.2.1.1 Sample Carryover

There was some carryover from the original standard evident in the first blank analysed. However, the following analysis of the extraction blank was devoid of acids. Identified in the blank analysed halfway through the run was a low level of ferulic acid, succinic acid, and malic acid, indicative of some carryover
within the column. This result calls into question the identification of these three acids in the samples with intensities lower than the intensities identified in the blank. Succinic acid was a recurring contaminant during this analysis, and its conclusive identification in archaeological samples was based upon the absolute intensity exceeding the blank readings. Malic acid was identified in sixteen of the samples; however, the results of six samples are considered suspect based upon the results of the blank discussed earlier. There was also a noticeable decay in the malic acid intensity throughout the analysis. This trend is discussed in the next section.

### 5.2.1.2 Loss of Certain Metabolites

Figure 5.4 illustrates three of the organic acids examined in this targeted analysis: malic, tartaric, and isopropyl malic acid. The QCs, listed from 1-5, are also illustrated in this chart. Malic acid was identified in all five of the QC samples; however, there was a 10 fold drop in the intensity from QC1 (77648) to QC5 (2408). There was also a consistent drop in the intensity of malic acid throughout the analysis. A separate standard mixture was prepared and analysed at the end of the analyses and neither malic nor tartaric acid were identified in this final mixture (both were added to the standard mixture). It was hypothesised at the time that the binding efficiency of the guard column or the analytical column was increasing over the course of the total analyses and that malic acid with three hydroxyl groups and tartaric acid with four hydroxyl groups were the most affected by this increased binding capacity. The sample matrix, either the pottery matrix or the salt residue from the alkaline fusion, were both considered as possible culprits.

Upon dry down with nitrogen, there was a white sediment in the base of the sample vials. Initially, the identity of the sediment was thought to be KCl salt, a byproduct of the alkaline fusion. In order to
confirm the identity of this crystalline salt, two representative samples of the white powder (found on the bottom of nearly all the samples) were taken from a neck sherd and a sample of the wine press for analysis by x-ray diffraction (XRD). X-ray diffraction is an analytical technique developed out of Bragg's law ( $\mathrm{n} \lambda=2 \mathrm{~d} \sin \theta$ ) which describes a unique crystalline structure based upon its ability to reflect incident radiation. The samples were analysed by Jackie Deans of the School of Chemistry, University of Birmingham. The samples contained a remnant of solvent (90/10 (v/v), ACN/100mM ammonium acetate) and were dried overnight in a glassware drying oven. Once dried, the samples were prepared by mixing a small portion of the powder with mineral oil and applying to the end of a diffraction pattern free glass rod. Each sample was scanned separately for 26 min . The sample was scanned between 5-90 ${ }^{\circ}$ with a 0.35 degree step. The results of the diffraction analyses are shown in Figures 5.5a and 5.5b and neither sample showed a distinct diffraction pattern; therefore, the white powder was not identified as a crystalline material, but rather an amorphous, poorly ordered material.


Figure 5.5. X-ray diffraction results of the white powder remaining after extraction of sherd (neck) and of a wine press sample. Each sample was scanned from 5 to $90^{\circ}$ for 26 min at a 0.35 degree step. The results indicate a poorly ordered, amorphous material.

In order to gain more information on this white powder, two separate samples were taken from two sherds (neck, side wall) for elemental analysis by scanning electron microscopy with energy dispersive Xray micro analysis (SEM/EDS). The two samples were adhered to an aluminum stub with double side carbon dots and then coated with carbon in order to provide a conductive surface. The samples were analysed by Paul Stanley at the Centre for Electron Microscopy in the School of Metallurgy and Materials, University of Birmingham. The surface of each sample was scanned manually across the surface to collect the elemental information. The elemental analyses of each sample were relatively
uniform across the surface of the sample and the graphs are shown in the Appendix, Figure 5.1. The common elements in the powder are aluminum and silicon, presumably leached out of the clay matrix during the alkaline fusion. Presumably the leaching of the silica/alumina matrix was chelating the most accessible compounds malic and tartaric acid (as well as isopropyl malic acid).

### 5.2.1.2.1 Pottery Science and the Bayer Method

Clay is produced from weathering of rocks and the elements most immune to such weathering are aluminium and silicon. Therefore, the main constituents of clay are silicon, surrounded by four oxygen atoms forming a tetrahedral, and aluminum, surrounded by six hydroxide (or oxygen in some cases) forming an octahedral. The electronic interaction amongst these two basic units form alternating sheets of silicate/aluminate/silicate separated by an interlayer often populated by other cations such as $\mathrm{Fe}, \mathrm{Mg}$, and K (Figure 5.6). Morphologically, these sheets appear as a lamellar-like structure and it is these sheets that provide the sliding movement of a clay. When water is added, the dipole attraction between the water molecules and the surface charged clays provide for ease of movement allowing for clays' most precious attribute: its plasticity (Rice, 1987: 41). This plasticity allows it to be molded into utilitarian forms and when heated and the added water is driven off, the form is maintained.


Figure 5.6. A diagram of the physical make-up of a clay body.
http://faculty.yc.edu/ycfaculty/ags105/week08/soil_colloids/soil_colloids_print.html

It is theorised that under strong alkaline solution, the aluminate (and silicate) building blocks leach out of the solution in a similar fashion to the preparation of aluminum ore from bauxite, the Bayer method (Loh et al., 2005, Caullett and Guth, 1989). The Bayer method was invented in 1887 for the purpose of extracting aluminates from ore at high pH followed by filtration, calcining and then collecting the product (Habashi, 2005). At alkaline $\mathrm{pH}, \mathrm{Al}(\mathrm{OH})_{4}{ }^{-}$is leached from aluminosilcates. As the pH is lowered, there is a transition in the oxidation state of aluminum from $\mathrm{Al}^{4}$ to $\mathrm{Al}^{6}$ forming $\mathrm{Al}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}{ }^{3+}$ the aluminate ion present in acidic medium $(\mathrm{pH}<3)$ (Franks and Gan, 2007). It is theorised that this ion might have built up on the guard column during the successive runs and complexed with malic and tartaric acids.
$\mathrm{Al}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}{ }^{3+}$ is gelatinous rather than crystalline which explains the lack of diffraction pattern. The origin of the silicon in the elemental analysis may be due to silanol groups which although uncharged, could form hydrogen bonds with the aluminate complex.

In terms of the analysis of ancient wine, one of the more problematic biomarkers to identify is tartaric acid. Depending upon the published method, the absence of this compound is often attributed to the assumption that a tartrate salt will not survive archaeological time. However, it is more likely that the original tartaric acid, combining within the wine polymer and then released via alkaline hydrolysis has combined with an aluminate ion leached from the pottery matrix. The current extraction method applied in this thesis probably introduces more of the matrix effect by allowing the sample to sit in acidified solution for 24 h , yet any caustic solution applied to a pottery matrix will leach $\mathrm{Al}(\mathrm{OH})_{4}{ }^{\text {a }}$ which upon acidification is converted to $\mathrm{Al}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}{ }^{3+}$. Alkaline fusion is the most effective method to break down the wine polymer, therefore, a clean-up procedure must be added to the extraction step.

### 5.2.1.3 Strong Cation Exchange Solid Phase Extraction (SPE)

In order to capture the positively charged aluminate complex, an SPE step was added to the sample preparation method. The cartridge chosen was a negatively charged phenyl sulfonate cartridge, Figure 5.7.


Figure 5.7. Phenyl sulfonate active group on the strong cation exchange SPE cartridge. At pH 2, the available negative charge binds the positively charged aluminate ion. The analytes of interest, either neutral or negatively charged at low pH will flow through the cartridge.

At acidic pH , the acids of interest are either fully protonated or negatively charged. In either case, the acids will not bind to the stationary phase and instead will flow through the SPE cartridge. In preparation for this added step, pottery samples were acidified for 24 h . After the allotted time, the SPE cartridges were conditioned per manufacturer recommendation with 1 ml methanol followed by 1 ml deionized water. The samples were centrifuged and the acidic aqueous layer was then added to the cartridge, the flow through captured and then extracted with ethyl acetate and dried down under a stream of nitrogen.

Prior to rerunning the samples, the guard column was regenerated with a wash of 0.5 M NaCl for at least 30 column volumes. This was followed with a water rinse of 30 column volumes.

### 5.2.1.3.1 Retention Time Shift Post-SPE

A mixture of 9 acids was spiked into extracts of the wine press and then eluted through the SPE cartridge in order to determine if there was any retention time shift due to the procedure. The retention time of eight of the nine acids were shifted slightly. A major shift occurred with ferulic acid, its retention time shifted by nearly two min. The change in retention times are given below. However,
none of the retention time shifts were large enough to warrant a change in the time span of the $\mathrm{MS} / \mathrm{MS}$ segments.

Table 5.3 The retention time for standard acids before and after sample clean-up with solid phase extraction.

| identification | RT pre SPE | RT post SPE |
| :---: | :---: | :---: |
| ferulic acid | 20.45 | 19.06 |
| syringic acid | 21.01 | 20.55 |
| p-coumaric acid | 21.46 | 21 |
| vanillic acid | 21.96 | 21.57 |
| malonic acid | 26.43 | 26.15 |
| malic acid | 26.8 | 27.11 |
| tartaric acid | 27.55 | 28 |
| succinic acid | 26.16 | 26.02 |
| isopropylmalic <br> acid | 24.73 | 25.25 |

### 5.2.2 Reanalysis of Archaeological Samples

Following discovery of the aluminate ion interference with the organic acids, re-analysis of the entire sample set was undertaken. The thirty one samples were re-extracted: wine press (prep layer+mortar) $n=10$, wine press (prep layer) $n=5$, amphorae neck $n=5$, side wall of amphorae $n=5$, amphorae base $n=5$, and extraction blank $n=1$. The exact mass of each archaeological sample was recorded in the samples extraction log sheet located in the Appendix, Chapter 5, Table 2. The finalised standard operating procedure was described in Chapter 2. The samples were extracted as before with the additional SPE step. All samples were capped and stored at $-80^{\circ} \mathrm{C}$ until analysis.

As before, prior to analysis by LC-MS/MS, the samples were removed from the freezer and reconstituted in $30 \mu$ l of mobile phase $90 \% \mathrm{ACN} / 10 \% 100 \mathrm{mM}$ ammonium acetate, $\mathrm{pH}=8.2$. Based upon the results of the earlier analysis, two acids were dropped from the targeted list: 2,3-dihydroxybenzoic acid and ketobutyric acid.

Table 5.4 The absolute intensities of the LC-MS/MS targeted analysis of the Sardinian wine press, its preparatory layer, and fifteen sherds representing transport amphorae.

| Sample name | ferulic acid | gentisic acid | syringic acid | p-coumaric acid | vanillic acid | succinic <br> acid | malonic acid | malic acid | tartaric <br> acid | isopropyl malic acid |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| QC1 | 3293 | 0 | 5426 | 8748 | 9408 | 667960 | 587254 | 0 | 0 | 0 |
| QC2 | 3728 | 0 | 5772 | 8948 | 8621 | 702961 | 663624 | 0 | 0 | 0 |
| blank | 0 | 0 | 0 | 0 | 0 | 39382 | 47344 | 0 | 0 | 0 |
| Ext_blnk_31 | 0 | 0 | 0 | 0 | 4337 | 39974 | 43318 | 0 | 0 | 0 |
| WP5_TM. 003 | 19855 | 0 | 14309 | 15642 | 9842 | 395223 | 743383 | 1390045 | 5885 | 182923 |
| WP9_TM. 027 | 9317 | 0 | 17258 | 26968 | 25393 | 934154 | 1077255 | 1338433 | 0 | 117354 |
| 1.TM.060.1.56 | 0 | 0 | 1894 | 6332 | 5001 | 92586 | 32532 | 0 | 0 | 0 |


| prep11 | 8681 | 0 | 25264 | 16010 | 29954 | 1868501 | 1105105 | 749561 | 0 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1.TM.033.1.14 | 1352 | 0 | 2735 | 3634 | 3401 | 34856 | 48847 | 0 | 0 | 0 |
| WP4_TM. 003 | 11607 | 0 | 8952 | 12413 | 9854 | 575125 | 545164 | 1232819 | 15952 | 619539 |
| QC3 | 0 | 0 | 3808 | 5208 | 5762 | 406642 | 0 | 0 | 0 | 0 |
| blank | 0 | 0 | 0 | 0 | 0 | 35731 | 0 | 0 | 0 | 0 |
| WP6_TM. 027 | 6977 | 0 | 13972 | 39168 | 27748 | 1314494 | 1328163 | 0 | 0 | 0 |
| prep15 | 7223 | 0 | 20221 | 30003 | 26246 | 1752745 | 1178465 | 0 | 0 | 0 |
| WP1_TM. 003 | 10323 | 0 | 16692 | 5963 | 8201 | 471943 | 757062 | 0 | 0 | 0 |
| prep12 | 0 | 0 | 503 | 0 | 0 | 553122 | 1001212 | 0 | 0 | 0 |
| SAR.TM. 4 | 0 | 0 | 0 | 8364 | 0 | 150143 | 183042 | 224330 | 0 | 0 |


| 1.TM.033.1.15 | 0 | 0 | 0 | 0 | 0 | 58572 | 0 | 0 | 0 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| prep14 | 10348 | 0 | 16674 | 13496 | 16828 | 2308857 | 1224998 | 0 | 0 | 0 |
| WP10_TM. 027 | 6025 | 0 | 5604 | 11881 | 7187 | 929745 | 1097694 | 0 | 0 | 0 |
| 1.TM.060.1.15 | 0 | 0 | 2115 | 2099 | 2683 | 76572 | 0 | 0 | 0 | 0 |
| 1.TM.035.1.1 | 0 | 0 | 241 | 0 | 0 | 56492 | 0 | 0 | 0 | 0 |
| 1.TM.028.1.33 | 0 | 0 | 792 | 0 | 2040 | 58731 | 0 | 0 | 0 | 0 |
| 1.TM.028.1.31 | 0 | 0 | 856 | 0 | 2064 | 401987 | 0 | 0 | 0 | 0 |
| SAR.TM. 67 | 0 | 0 | 991 | 2209 | 2942 | 270373 | 0 | 0 | 0 | 0 |
| QC 4 | 2750 | 0 | 4408 | 6250 | 5258 | 695976 | 587750 | 0 | 0 | 0 |
| blank | 0 | 0 | 0 | 0 | 1819 | 50906 | 0 | 0 | 0 | 0 |


| 1.TM.033.1.20 | 439 | 0 | 491 | 806 | 1806 | 53599 | 0 | 0 | 0 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| WP2_TM. 003 | 13510 | 0 | 21394 | 8252 | 10774 | 803938 | 1307686 | 0 | 0 | 0 |
| WP8_TM. 027 | 6546 | 0 | 15500 | 33502 | 32132 | 1561604 | 1232583 | 683616 | 0 | 0 |
| 1.TM.042.1.7 | 0 | 0 | 249 | 549 | 0 | 68190 | 24710 | 0 | 0 | 0 |
| 1.TM.060.1.45 | 0 | 0 | 706 | 0 | 2603 | 44701 | 0 | 0 | 0 | 0 |
| 1.TM.034.1.20 | 0 | 0 | 456 | 0 | 2147 | 52046 | 0 | 0 | 0 | 0 |
| 1.TM.036.1.2 | 0 | 0 | 228 | 0 | 1385 | 63625 | 0 | 0 | 0 | 0 |
| QC 5 | 3026 | 0 | 5258 | 8878 | 5962 | 762201 | 833825 | 0 | 0 | 0 |
| blank | 0 | 0 | 0 | 0 | 1124 | 49937 | 0 | 0 | 0 | 0 |
| WP7_TM. 027 | 2341 | 0 | 1905 | 0 | 4682 | 0 | 0 | 0 | 0 | 0 |


| WP3_TM. 003 | 0 | 0 | 407 | 0 | 0 | 79277 | 0 | 0 | 0 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| prep13 | 0 | 0 | 0 | 0 | 0 | 307473 | 0 | 0 | 0 | 0 |
| 1.TM.044.1.3 | 0 | 0 | 1742 | 0 | 0 | 98625 | 0 | 0 | 0 | 0 |



Figure 5.8. Graph illustrating the absolute intensities of the phenolic acids identified in the wine press and in the sherds from Trunc e Molas site, Sardinia, Italy. The greatest concentration of multiple biomarkers/sample is located in the wine press and in three of the prep layer samples. Prep layers 2 and 3 represent misinjections.

### 5.2.2.1.1 Sample Identification

Using the improved analytical method and based upon retention time, fragment ion transitions, and fragmentation intensities, two additional acids were identified in two samples of the wine press, bringing the total to nine acids detected. While a relative small increase, this should improve the specificity of the chemical fingerprint used to determine whether the archaeological sample was exposed to wine. Furthermore, between six and seven acids were identified in the majority of the remaining archaeological samples. Three acids were identified in the extraction blank: malonic, succinic, and vanillic acid. These levels identified in the extraction blank were considered baseline and any acid with an intensity below the extraction blanks intensity was considered not identified. For fragmentation intensities, absolute intensities were taken from each separate product ion and then relative intensities of each precursor product ions were determined; the relative intensities were then compared with the relative intensities of those same ions from the standards. In certain cases, inaccurate intensities could be explained by co-eluting peaks.

The results strongly corroborate the presence of wine in the press, particularly sample TM. 003 and its prepatory layer. Figures 5.9-5.17 represent the extracted ion chromatograms and the transitions of the identified acids in the targeted analysis of sample 4 from the wine press, TM.003.3. Nine of the acids were identified in this sample and in sample 5 (TM.003.3). Eight of the acids were identified in wine press sample 9 (TM.027.3), and in prep layer 11 . The majority of the remaining samples contained six acids: malonic, succinic, vanillic, syringic, p-coumaric, and ferulic acids. Comparisons of absolute intensities (Figure 5.8 ) show the greatest intensity and variety of acids were found in the wine press and its prep layer particularly when compared with the fifteen sherds taken from the transport amphorae.

This compares with the initial analysis of the archeological samples, however there was a difference in intensities. Intensities of the re-analysed samples were greater confirming the importance of the additional SPE step. Figure 5.18 compares the absolute intensities of four acids malonic, malic, vanillic, and syringic acids with the SPE step and without the SPE step. The intensities of malic and malonic acids are appreciably increased with the additional step of the SPE. For the phenolic acids, there was a general increase in intensity for the majority of samples although not as obvious most likely due to the fact that the phenolic acids have little interaction with the aluminate ion. However, there was at least one sample where the sample signal was lost suggesting a possible interaction between the phenolic acid and the ring structure on the SPE cartridge in the form of a pi-pi bond interaction.

The addition of an SPE step to the sample extraction protocol was done in order to capture a (possible and not yet confirmed) positively charged aluminate ion. However, by run 30 there was an increase in the column pressure suggesting a blockage of normal flow possibly caused by the accumulation of a compound adhering to the guard column. This affected the later sample runs particularly the final five analyses.

The results did not confirm the presence of wine in any of the sherds from the transport amphorae. For the results to suggest the presence of wine, a sample should contain a full biomarker complement including phenolic and organic acids. However, there was not even a complete list of phenolic acids identified in the sherds. Again, this highlights inconsistencies and spurious results that may occur when choosing only 1 or 2 biomarkers to identify archaeological residue. One theory for the lack of a biomarker signature is that these amphorae never held wine and instead were at the site in preparation for filling. Ancient texts on agriculture strongly recommend preparing amphorae by lining the interior
with pitch, to prevent seepage into the clay matrix (Columella, book 12: 197). Pitch was not identified on any of the interior surfaces of these sherds further suggesting that these amphorae were not ready to receive wine.

### 5.2.2.1.2 Insoluble or 'Bound' Lipids

It has been reported that insoluble (in chloroform), polymerised lipids were extracted under alkaline fusion; particularly prevalent were dicarboxylic acids (Evershed et al., 2002). Figure 5.19 is a typical output for the LC-MS/MS undertaken in this research. The early eluting peaks from 4.55-7.10 were tentatively identified as lipids, bound via ester bonds as a polymeric matrix and broken down under the alkaline attack. Tentative identifications were accomplished by comparing the experimental masses with the published masses in the LipidMaps database. The eluting peak at 4.55 minutes contains $\mathrm{m} / \mathrm{z}$ 283 and 255 , tentatively identified palmitic and stearic acids. This peak was also identified in the extraction blank and was considered a byproduct of the extraction procedure. The largest eluting peaks at retention time 5.6 and 6.03 include $m / z 201$ and 187 which were tentatively identified as decanedioic acid (sebacic) and nonanedioic acid (azelaic). Tentative identifications of other masses include: 3,12-dihydroxy-hexadecanoic acid ( $\mathrm{m} / \mathrm{z} 287$ ), 2-hydroxydecandioc acid ( $\mathrm{m} / \mathrm{z} 217$ ) , and 3-methyldodecanedioic acid ( $m / z 244$ ). At this point, the origin of these fatty acids is unknown. One explanation for these acids is the degradation over an archaeological span of an unsaturated fatty acid by oxidation and dehydration, producing smaller remnants of the original fatty acid, as seen in Figure 5.20 (Regert et al., 1998). Therefore, the azelaic and sebacic may represent the degraded form of oleic acid.


Figure 5.9. The identification of malonic acid from the Sardinian wine press based upon a) two product ions from $m / z 103$ and b) retention time (26.14).


Figure 5.10. The identification of succinic acid from the Sardinian wine press based upon a) two product ions from $m / z 117$ and b) retention time ( 25.94 ).


Figure 5.11. The identification of malic acid from the Sardinian wine press based upon a) three product ions from $m / z 133$ and b) retention time ( 26.84 ).


Figure 5.12. The identification of tartaric acid from the Sardinian wine press based upon a) three product ions from $m / z 149$ and b) retention time (27.73).


Figure 5.13. The identification of p-coumaric acid from the Sardinian wine press based upon a) three product ions from $m / z 163$ and $b$ ) retention time ( 20.93 ).


Figure 5.14. The identification of vanillic acid from the Sardinian wine press based upon a) three product ions from $\mathrm{m} / \mathrm{z} 167$ and b) retention time (21.56).


Figure 5.15. The identification of isopropyl malic acid from the Sardinian wine press based upon a) three product ions from $\mathbf{m} / \mathrm{z} 175$ and b) retention time ( 25.17 ).


Figure 5.16. The identification of ferulic acid from the Sardinian wine press based upon a) three product ions from $m / z 193$ and $b$ ) retention time ( 18.62 ).


Figure 5.17. The identification of syringic acid from the Sardinian wine press based upon a) four product ions from $m / z 197$ and b) retention time (20.48).


Figure 4.18. The graphs compare the absolute intensities from the LC results before and after SPE clean up on four acids: (a) malic acid (b) malonic acid (c) syringic acid and (d) vanillic acid, without SPE clean up and with SPE clean up. Overall, the absolute intensities are greater in the post SPE sample run. The phenolic acids, syringic and vanillic, are less affected by the SPE step.
(a)

(b)

(c) (200
(d)

(e)
(200
(f)


Figure 5.19. A representative total ion chromatogram from the LC-MS/MS analysis of the wine press and sherds taken from Trunce Molas, Sardinia, Italy. Early eluting peaks from 4.55-7.10 (a) represent insoluble lipid material released from the alkaline fusion. Tentative identifications include (masses rounded up to the nearest integer): 255: palmitic acid, 283: stearic acid, 201: sebacic acid, 187: azelaic acid, 287: 3,12-dihydroxy-hexadecanoic acid, 217: 2-hydroxydecandioc acid, and 243: 3-methyl-dodecanedioic acid.
(a)

(b)







Figure 5.20. Illustration of suggested formation of unsaturated acids from (a) 9-octadecenoic acid (b) oxidation at the site of the double bond and (c) dehydration, reproduced from Regert et al., 1998.

### 5.2.2.1.3 Multivariate Analysis of Archaeological Samples

Utilising the multiple biomarkers identified in the wine press and its preparatory layer offers a chance to categorise the results as red wine or white wine by comparing ratios of the intensities of all nine acids identified in the samples. Ratios were logged (base 10) and the results are tabulated in Appendix, Chapter 5, Table 3. PCA was chosen as an unsupervised method of comparing the archaeological results with standard laboratory aged results. In order to have at least $\mathrm{n}=10$ classification for the standard samples, ratios were prepared from the DIMS samples (alkaline fusion aliquot) from Chapter 3. Due to the inherent uncertainties in comparing two analytical platforms (high resolution DIMS with LC-triple quad analysis), several standard sherds were also analysed by LC-MS/MS as 'validation' samples. A relative standard deviation was calculated for the logged ratios from DIMS and LC-MS/MS of the standards only, in order to determine how variable the results were between the two platforms. An example is seen in Table 5.5.

Figure 5.21 illustrate the graphs of the RSD calculated from the logged ratios of red and white wine standard sherds analysed by two separate analytical platforms. 200 was identified as the cutoff point, based upon the start of a 'kneebend' in the graph, where the calculated RSD veers from the group. Only those ratios with the $200 \%$ cutoff were chosen: originally there were 36 ratios, 7 were removed, for a final total of 29 ratios used for classification.

Table 5.5 The table illustrates four representative logged ratios only from acids 103 (malonic), 117 (succinic), 133 (malic), and 149 (tartaric). Ten DIMS samples of the alkaline fusion aliquot of red wine were compared with 3 samples of the lab aged sherds analysed by LC-MS/MS. The standard deviation, mean, and relative standard deviation were calculated for the 13 samples, for all ratios examined.

|  |  | 103:117 | 103:133 | 117:133 | 103:149 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| DIMS | RG3A | -1.47106 | -1.50041 | -2.5077 | -1.48006 |
|  | RS3B | -1.45267 | -1.38201 | -2.03877 | -1.85403 |
|  | RS2B | -1.5742 | -1.69206 | -2.43994 | -2.10307 |
|  | RG4A | -1.28959 | -1.46066 | -2.25733 | -1.57074 |
|  | RG1A | -1.84215 | -1.642 | -2.06484 | -2.02399 |
|  | RS1A | -1.70278 | -1.54046 | -1.79581 | -1.50232 |
|  | RS4A | -1.43297 | -1.4787 | -1.88438 | -1.52619 |
|  | RS2A | -1.39828 | -1.53215 | -2.48595 | -1.39064 |
|  | RS4B | -1.54929 | -1.58637 | -2.16048 | -1.69141 |
|  | RS3A | -1.45338 | -1.4845 | -2.05726 | -1.66818 |
| LC- <br> MS/MS | $\begin{aligned} & \text { red 1, } \\ & \text { 1:10 } \end{aligned}$ | -0.44752 | -0.63316 | -0.18565 | -0.46627 |
|  | $\begin{aligned} & \text { red 2, } \\ & \text { 1:10 } \end{aligned}$ | -0.45113 | -0.58501 | -0.13388 | -0.44493 |
|  | $\begin{aligned} & \text { red 3, } \\ & \text { 1:10 } \end{aligned}$ | -0.38493 | -0.67971 | -0.29478 | -0.64963 |
|  | SD | 0.497168 | 0.401757 | 0.888646 | 0.551778 |
|  | MEAN | -1.26538 | -1.32286 | -1.71591 | -1.41319 |
|  | RSD | -39.29 | -30.3703 | -51.7887 | -39.0449 |



Figure 5.21. The relative standard deviation of the acids' logged ratios for the laboratory sherds (a) red and (b) white. 200 was chosen as the cutoff point for both red and white, based upon the start of a 'kneebend' in the graph at which point the ratios deviate from the group.

In preparation for multivariate analysis, all missing values in the wine press and the preparatory layer samples were assigned a value equal to half of the lowest intensity for each metabolite. This approach was applied to all acids. In the case of malonic, vanillic, and succinic acids, the missing values were half of the value in the extraction blank. Figure 5.22 represents a plot of the samples vs PC2 scores of laboratory aged sherds analysed by DIMS and LC/MS/MS, as well as the results from the wine press analysed by LC-MS/MS only. Preparatory layer 2 and 3 , and wine press sample 3 were removed from this classification due to the lack of original features. The greatest separation along PC1 (60.57\%) separated the samples based upon the analytical platform, DIMS vs. LC-MS/MS. The second greatest variation was identified by PC2 between the white DIMS and the red DIMS, with a smaller variation between the white and red LC samples. The archaeological samples from the wine press and the preparatory layer are most closely aligned with the white LC samples. A PCA scores test identified significant differences between all groups except the white LC and the archaeological samples; there was no significant difference between these two groups along PC2.

The multivariate analysis of the logged ratios of 9 acids suggests the biomarker fingerprint of the archaeological samples are more similar to laboratory aged white wine than to red wine. This very preliminary data suggests an application for classifying samples according to a suite of biomarkers, as opposed to a single marker as reported previously. The results of this classification support the archaeological descriptions of the wine press as a lower basin of a two tier wine press where the upper portion did not survive. Therefore, treading occurred on the upper platform. Whether the original grapes were red or white, the skins/seeds were not transferred to the lower vat, only the extracted juice resulting in a lower concentration of phenolic material from the grape must.


Figure 5.22. A plot of sample vs. PC2 scores taken from a PCA scores plot of laboratory aged sherds analysed by DIMS and LC/MS/MS, and an archaeological wine press analysed by LC/MS/MS. The plot aligns the archaeological samples more closely with LC-MS/MS analysis of laboratory aged white wine.

### 5.4 Conclusions

Taken together with Chapter 3 and 4, these results demonstrate for the first time the application of archaeological metabolomics to discover a suite of organic acid biomarkers and then their successful transfer to a targeted analytical method in an attempt to identify polymerised wine residues found in archaeological samples. The observations that the majority of the biomarkers were identified in the wine press and its preparatory layer strongly suggests the presence of an aged, polymerised wine consistent with the archaeological context. When compared with the biomarker fingerprint of the modern, laboratory aged wine sherds it was shown that the identified acids found in the wine press more closely aligned with white wine, which can be explained by the object's identification as a lowest tier basin. The presence of wine was not confirmed in the amphorae sherds. This result suggests that the amphorae may never have held wine and were instead at the site in preparation for filling and transport. The identity of 1 or 2 phenolic acids within several of the sherds highlights the fact that the use of 1 or 2 biomarkers for the identification of archaeological residue leads to spurious results and may ultimately produce incorrect conclusions about the objects usage within the site.

Also noted in this chapter was the detrimental effect of the matrix interference on several of the acids. The usage of alkaline solution with aluminum-silicate matrix (pottery) will ultimately leach out aluminum hydroxide. Upon acidification, it is hypothesised that the newly formed aluminate ion complexes with acids containing 3 or more hydroxyl groups, like malic acid and tartaric acid. The additional clean up step to remove the aluminate ion greatly increased the response of all acids, particularly the organic acids.

In order to characterise the remnants from an archaeological sample, it is critical to use multiple biomarkers for a responsible identification since an attempt to analyse liquid or food remnants in containers that were stored in the soil and were most likely reused multiple times results in a level of uncertainty. In the next chapter, these biomarkers will be applied to archaeological objects with considerable interference from food remnants.

# 6. Pilot Study: The Metabolomics Approach Applied to Archaeological Samples from Vindolanda, Northumberland, England 

In Chapter 3 of this thesis, laboratory aged sherds doped with water and wine were aged in different environments in order to determine a biomarker signature for aged wine, and then apply those biomarkers to a targeted analysis of archaeological artifacts. In order to test the applicability of the identified biomarker list, two contrasting archaeological sites were chosen. Chapter 5 applied the targeted analysis of the biomarker list to a dry, sandy site at a Punic farmstead in Sardinia, Italy. Analysis of the winepress was highly indicative of wine remnants, whereas the application of the biomarker list to a group of amphora sherds was indeterminate. The second archaeological site chosen to test the biomarker list was Vindolanda, a Roman British garrison active from circa 85-400
A.D.

### 6.1 Introduction

The environment in the Vindolanda settlement is a wet, anoxic environment with a heavy peat, organic rich soil. The site consists of transport amphorae, mortaria, and cooking pots from both military and civilian encampments. From the objects gathered in the early summer of 2012, two sample sets were analysed. Initially, a sample set consisting of transport amphorae were analysed for the remnants of wine utilising the targeted method developed in Chapter 4 and refined in Chapter 5. A second sample set was then analysed which included cooking pots and mortaria in order to identify the remnants of wine or vinegar used in food preparation. Since olive oil was a common ingredient in food preparation, several non-polar aliquots were also examined for the
presence of TAGs (Sheehan-Finn, 2012). Since the objects examined in this chapter represent multiple locations, the specific context information is given with each sample set listed in section 6.2 and for the remaining samples in the Appendix, Chapter 6.

Thirty two samples were extracted and analysed; the results of sixteen of those samples, as well as three soil samples are given here in the following sections. The remaining sample results are given in the Appendix, Chapter 6, Tables 5-14. Eight contexts represent area B, a civilian encampment active from 213-300A.D., and one context represents area A, a military fort originally constructed in 213 A.D., rebuilt in the $4^{\text {th }}$ century. Samples were extracted using the method described in Chapter 2.2.3. The alkaline fusion aliquot, and in certain cases the non-polar aliquot, were analysed. The alkaline fusion aliquot was analysed by the targeted LC-MS/MS method developed in Chapter 4; the non-polar aliquot was analysed by a data dependent fragmentation method explained in Chapter 2.2.5.3.

The samples represent mortaria, amphorae, and cook pots that were excavated from several contexts within specific areas of the Vindolanda settlement by volunteers working under the direction of staff archaeologists from the Vindolanda Trust. Soil samples were also collected by volunteer excavators on the same day; one from a known context and one sample from an unknown context. In order to determine the presence or absence of wine in these samples, nine acids were targeted in the LC-MS/MS analysis. The nine acids chosen were those acids which were identified in the wine press from Chapter 5 and are considered highly indicative for the presence of wine: malic, iso-propyl malic, tartaric, succinic, vanillic, ferulic, p-coumaric, syringic, and malonic acids.

Samples were designated as in the following example: V12-30B/4, whereby V12 represents the place and year of excavation, Vindolanda, 2012, 30B represents the context, and /4 represents sample 4 taken from that particular context.

### 6.1.1 Targeted LC-MS/MS Analysis for Wine Residue

Prior to analysis by LC-MS/MS, the extracted samples were removed from the $-80^{\circ} \mathrm{C}$ freezer and reconstituted in $30 \mu$ l of mobile phase $90 \% \mathrm{ACN} / 10 \% 100 \mathrm{mM}$ ammonium acetate, $\mathrm{pH}=8.2$. The samples were vortexed for 10 s and then centrifuged for 20 min at $10^{\circ} \mathrm{C}$ at 6000 rpm . Ten microlitres of each sample were added to a 96 well plate in a randomised order and $1 \mu \mathrm{l}$ of sample was injected onto the column for analysis.

Samples were analysed in two batches over several days. The first day of sample runs experienced a substantial increase in column pressure which necessitated halting the run and changing the inline filter frit. The increased pressure shifted the retention times for several of the acids, as well as resulted in a drop in intensity for those acids. The samples that were affected are listed in the Appendix, Chapter 6, Tables 5-14 and their results are considered suspect; therefore, those samples were not used for the results of Chapter 6. As noted in Chapter 5, the elution time between 4-7 min yielded 'bound' lipids released after alkaline fusion. Putative identifications are given in the tables below based upon the integer mass and comparison with literature sources. For absolute confirmation of the lipids identity, a more targeted approach is necessary.

### 6.1.2 TAG analysis

As stated in the introduction, TAGs are lipid fractions composed of three fatty acyl chains esterified to a glycerol backbone. When known, the position of the acyl chains are designated from top to bottom, based on a Fisher projection as $s n-1$, $s n-2$, and $s n-3$, where $s n=$ stereospecifically numbered; TAGs are identified by their acyl chains followed by the glycerol designation, triacyl-sn-glycerol, and are written as $\mathrm{C}_{\text {total number of acyl carbons: total number of double bonds }}$ ) (IUPAC-IUB Commission on Biochemical Nomenclature, 1977).

As described in Chapter 1, determination of TAG content in archaeological artifacts is useful to determine the origins of the lipid, e.g., plant vs. animal, ruminant vs. non-ruminant, ruminant dairy vs. ruminant adipose fat and is often undertaken with high temperature-gas chromatography-mass spectrometry (Evershed et al., 2002; Regert, 2011). In some cases, the identification of the acyl chain locations on the glycerol backbone will assist in the identification (Mottram et al., 1997; Lin and Arcinas, 2008). For samples analysed in this chapter, all samples were reconstituted in a lithium salt solution. The lithium cation-oxygen bond is stronger than that of other cations including sodium, potassium, as well as ammonium adducts, producing intense signals in the mass spectrometer down to the $\mathrm{MS}^{3}$ fragmentation pattern (Adams and Gross, 1986). Analysis of lithiated TAGs has been successfully applied to the study of archaeological artifacts in earlier studies (Garnier et al., 2009; Mirabaud et al., 2007).

The approach discussed here combines the mass spectral fragmentation data with an algorithm written in the Python environment by Dr. Ralf Weber, University of Birmingham, which identified the acyl chains on a particular TAG. Briefly, using collisionally induced dissociation, samples were fragmented from 600-1000 $\mathrm{m} / \mathrm{z}$ in a data dependent manner. The most intense ion from an initial
full scan is fragmented; the top three most intense ions from the resultant MS/MS fragmentation are then fragmented. In order to remove noise peaks and limit the number of falsely detected peaks, necessary criteria were added to the algorithm including: coverage within a certain number of scans (60\%), ppm mass difference between the experimental mass and the theoretical mass stored within the Viant laboratory's empirical formulae library ( 3 ppm for FT-ICR-MS, 60 ppm for IT), and RSD of peak intensities across the different scans (<30\%). The results of a data dependent collection were compared against a library of fatty acids determined from the expected fragmentation losses based on the published mechanism illustrated in the Appendix, Chapter 6, Figures 1 and 2 (Hsu and Turk, 2010).

As an example, Figure 6.1 represents the full scan mode of a lithiated TAG; the largest peak in the spectrum ( $m / z 867$ ) was then fragmented. Figure 6.2 illustrates the $\mathrm{MS}^{2}$ spectrum, the three resultant diacylglycerols (DAG) resulting from the fragmentation of $m / z 867$. There are two peaks associated with each DAG, one representing a loss of the neutral fatty acid from the original lithiated TAG, and the second peak resulting from the loss of the lithiated salt of the same neutral fatty acid. Figure 6.3 illustrates the $\mathrm{MS}^{3}$ level of fragmentation of the DAG at $m / z 611$. The predominant peak at $m / z 329$ represents the loss of 18:0 as an $\alpha, \beta$-unsaturated fatty acid. Its associated peak is $m / z$ 289, the lithiated 18:1 acid; the two peaks are separated by a mass of 40 , the dehydrated glycerol backbone.

In Hsu and Turk's 2010 paper, the fragmentation of $m / z 867$ yielded a peak at $m / z 585$ that was considerably smaller in intensity than either peak at $m / z 611$ or $m / z 583$. The rationale for the lower intensity was that the peak at $m / z 585$ was due to the loss of the fatty acyl chain at sn-2, the sterically hindered and energetically unfavorable position. Also, it was theorised that the highly labile $\alpha$-hydrogen on the sn-2 acyl chain functions to easily eliminate either the $s n-1$ or the sn-3 acyl
chain. However, under the experimental conditions for this thesis, the majority of the $\mathrm{MS}^{2}$ peaks are relatively similar in intensity, offering no obvious distinction between the acyl chain position on the glycerol backbone. The reason may be due to the multiple isomers formed within these TAGs, or the equal lability of any of the three $\alpha$-hydrogens found on the three acyl chains.


Figure 6.1. Direct infusion FT-ICR-MS of an archaeological sample, non-polar extract reconstituted in 2 mM LiCl (1:4 CHCl3:MeOH).

m/z

Figure 6.2. Direct infusion FT-ICR-MS ${ }^{2}$ from the fragmentation of $m / z 867$. The results identify the loss of three fatty acids: 16:0 ( $\mathrm{m} / \mathrm{z} 611$ ), 18:0 ( $\mathrm{m} / \mathrm{z} 583$ ), and 18:1 ( $\mathrm{m} / \mathrm{z} 585$ ),

| 867.8; MS2 |  |  |  |
| :---: | :---: | :---: | :---: |
| $\mathbf{m / z}$ | Intensity | Relative | explanation |
| 611.55858 | 64517.6 | 100 | neutral loss of FA (16:0) |
| 583.52726 | 59139.9 | 91.66 | neutral loss of FA (18:0) |
| 585.54299 | 57895.9 | 89.74 | neutral loss of FA (18:1) |
| 605.55042 | 15031.8 | 23.3 | neutral loss of FA (16:0); lithiated salt |
| 579.53487 | 14971.2 | 23.2 | neutral loss of FA (18:1); lithiated salt |
| 577.51917 | 10770.7 | 16.69 | neutral loss of FA (18:0); lithiated salt |



Figure 6.3. Direct infusion IT-MS ${ }^{3}$ from the fragmentation of $m / z 611$, the two resulting fatty acyl chains are separated by 40, the dehydrated glycerol backbone.

| 611.5; MS3 |  |  |  |
| :---: | :---: | :---: | :---: |
| $\mathbf{m} / \mathbf{z}$ | Intensity | Relative | explanation |
| 329.33557 | 5577.5 | 100 | loss of $\alpha, \beta$ unsaturated $18: 0$ |
| 289.28076 | 5197.1 | 93.18 | lithiated $18: 1$ associated with mz 329 |
| 331.35944 | 3625.1 | 64.99 | loss of $\alpha, \beta$ unsaturated $18: 1$ |
| 291.37537 | 2889.3 | 51.8 | lithiated $18: 0$ associated with mz 331 |
| 611.55225 | 1040.3 | 18.65 | base peak |
| 347.41809 | 829.2 | 14.87 | loss of the $18: 1$ as a ketene |
| 287.24274 | 757.6 | 13.58 | lithiated $18: 2$ |
| 345.30127 | 642.9 | 11.53 | loss of the $18: 0$ as a ketene |
| 327.27698 | 453.9 | 8.14 | loss of $18: 0$ as fatty acid |

### 6.1.2.1 TAG algorithm output

Table 6.1 illustrates the raw output from the algorithm. The TAG output includes the neutral fatty acid loss from $\mathrm{MS}^{2}$, as well as the $\alpha, \beta$-unsaturated fatty acid and the lithiated fatty acid losses from $\mathrm{MS}^{3}$ that are linked by a difference in mass of 40 Da. Illustrated in Table 6.1, row 1 represents the neutral loss from the full scan to level $2, \mathrm{MS}^{2}$; the fatty acyl component was identified as 16:0, such as palmitic acid. Row 2 identified the loss from $\mathrm{MS}^{2}$ to $\mathrm{MS}^{3}$; the fatty acyl component was identified as $18: 1$, such as oleic acid. Row 3 links the $m / z$ from row 2 (331) with a $m / z$ separated by -40 (291). Row 4 identifies the third fatty acyl component (291) as 18:0, such as stearic acid.

Table 6.1 The TAG algorithm output from the data dependent scans identifying a neutral loss at Ms ${ }^{2}$ (level 2), as well as the $\alpha, \beta$-unsaturated fatty acid and the lithiated fatty acid from $\mathrm{MS}^{3}$.

| m/z | m/z | Spectrum Header | Fatty acyl identification |
| :---: | :---: | :---: | :---: |
| 867.7994493 | 611.5578521 | ['FTMS + p ESI Full ms [600.00-1000.00]', 'FTMS + c ESI d Full ms2 867.80@cid35.00 [225.00-880.00]'] | $\begin{aligned} & \text { [256.2402304, 'FA(16:0)', } \\ & \text { 'e', '1to2'] } \end{aligned}$ |
| 611.5578521 | 331.3710066 | ['FTMS + p ESI Full ms [600.00-1000.00]', 'FTMS + c ESI d Full ms2 867.80@cid35.00 [225.00-880.00]'] | $\begin{aligned} & \text { [280.2413285, 'FA(18:1)- } \\ & \text { 2H', 'e', '2to3'] } \end{aligned}$ |
| 331.3710066 | 291.2958766 | ['FTMS + p ESI Full ms [600.00-1000.00]', 'FTMS + c ESI d Full ms2 867.80@cid35.00 [225.00-880.00]'] | ```[40.0, 'glycerol-2[H2O]', 'i', 3]``` |
|  | 291.2958766 | ['ITMS + c ESI d Full ms3 867.80@cid35.00 611.56@cid35.00 [155.00-625.00]'] | $\begin{aligned} & \text { [291.2869869, } \\ & \text { '[FA(18:0)+Li]+', 'p', 3] } \end{aligned}$ |

### 6.1.2.2 TAG analysis applied to aged lipid samples

Based upon literature sources for foodstuffs in Roman Britain several modern lipid samples were initially aged by sonicating a 6 g modern laboratory sherd in 50 ml of chloroform with 1 g of fat for 2 x 30 min; the fats examined were: goat cheese, sheep cheese, cows milk, olive oil, sardines, back bacon, and plain sherds (Dudd et al., 1998). The sherds were allowed to air dry and were then weighed over four days until the weight was maintained over a 24 hour period. On the fourth day, four sherds for each fat sample, plus four plain sherds were placed in chloroform rinsed glass vials and stored in an oven for three months at $40^{\circ} \mathrm{C}$. Four sherds of each fat plus plain were wrapped in aluminum foil and stored at $-80^{\circ} \mathrm{C}$ until analysis, as zero month time point. After three months, the sherds were removed from the oven, wrapped in foil and stored in the $-80^{\circ} \mathrm{C}$ freezer until analysis. All samples were extracted using the extraction protocol described in Chapter 2.2.1 and analysed by the method described in Chapter 2.2.5.2. The results of the ruminant dairy samples are given in the Appendix, Chapter 6 tables 1, 2, and 3.

### 6.1.3 Foodstuff Analyses

In order to examine certain foodstuffs that may be rich in phenolic content and therefore overlap with the chosen wine biomarker list, seven foodstuffs common to Roman Britain were aged for 6 months at $40^{\circ} \mathrm{C}$ in preparation for analysis. Samples were extracted and analysed as described in Chapter 2 for laboratory aged standards analysed by DIMS-FT-ICR-MS. For data pre-processing, the collected mass spectra were each 'binned' into windows of $100 \mathrm{~m} / \mathrm{z}$, whereby the edges of each window were 'stitched' together. The signal to noise ratio was set at 100; replicate filtering was set at $12 \%$. Missing values were imputed using an in-house code written in the Matlab environment, MVImpute version05 (Hrydziuszko and Viant, 2011). The final data matrix was normalised and then
g-log transformed in order to stabilise the technical variation across all peaks measured (Parsons et al., 2007).

Figure 6.4 illustrates the PCA scores plot of the alkaline aliquots of seven foodstuffs. The greatest separation is along PC1 between the (adulterated) olive oil and the remaining classes. There was a small separation between the spices (oregano, thyme, and basil) and honey, barley malt and spelt wheat. Table 4 in the Appendix, Chapter 6, lists the top 200 loadings of the samples, sorted by PC1 values. Three of the top loadings, 1,7 , and 13 , were putatively identified as 2,3 -dihydrobenzoic acid, syringic acid, and p-coumaric acid, respectively. Looking at each class individually, the $m / z$ at 197.0455, putatively identified as syringic acid, is one of the most intense peaks in the oregano and in the basil samples. The $m / z 153.0192$ (putatively identified as 2,3-dihydrobenzoic acid) is also one of the most intense peaks in the mass spectra of all three spices. This suggests that the phenolic acid loadings are due in part to the contribution of the three classes of dried spices: basil, thyme, and oregano. These results also align with the results published on the phenolic acid content of spices determined after chemical hydrolysis (Herrmann and Nagel, 1989). Comparing these results to the results from the following archaeological contexts may suggest a secondary origin for phenolic acids, such as those found in the mortaria samples.

This small experiment highlights the commonplace nature of certain phenolic acids. For example, syringic acid, commonly used as the identifying compound for the remnants of red wine is also found in the alkaline fusion of common foodstuffs. Again, this highlights the dangers of utilising only 1 or 2 biomarkers to identify ancient wine.


Figure 6.4. PCA scores plot comparing the KOH aliquot of aged foodstuffs including: oregano, thyme, basil, honey, spelt wheat, barley malt and olive oil. The separation of the spices (oregano, thyme, basil) is due to an overwhelming phenolic acid content

### 6.2 Results: Area A

In the following sections, many of the contexts results are described in two tables. The first table contains the sample ID, sample type, sample weight, as well as the intensity of the nine acids identified from the targeted analysis. The final column suggests tentative identifications of bound lipids which were identified in the LC-MS/MS result. The second table illustrates the results from the TAG output and includes the acyl carbon number of each TAG identified as well as the number and identification of the TAGs associated with that designation. In certain cases, manual identification was included for descriptive purposes only; manual identification was utilised in cases where the samples did not pass the stringent rules put forth in the TAG output.

Area A represents a military stone fort built in A.D. 213, rebuilt in the 4th century, and consists of the centurion's apartment as well as soldiers' barracks. The samples collected from area A, having been excavated in 2011, were collected from the pottery conservation laboratory.

### 6.2.1 Context V11-VL-108A

This context represents the clay floor of the apartment where minimal animal bone was excavated. Several pieces of cookpot and amphora were gathered from this context. Targeted analysis of the acid components in sample 1, Table 6.2, identified three phenolic acids: syringic, vanillic, and pcoumaric acid. Ferulic acid was not identified in this analysis. Succinic acid was also identified. These results do not support the identification of wine within the cook pot. Levels of bound lipids were visible in this sample and were tentatively identified as hexadecanoic, octadecanoic, and octadecenoic acids. TAG analysis of the cookware identified $\mathrm{C}_{50: 1}$ (by manual identification). TAG
analysis of sample 7, cook pot, identified only one TAG $\mathrm{C}_{52: 1}$. Due to the limited TAG results, no conclusive origin of the lipid is possible.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Cook pot | 1.02 | ND | 21166 | ND | ND | 1690 | 5187 | ND | ND | 6914 | hexadecanoic, octadecanoic, and octadecenoic acids |


|  | Acyl carbon number | Types of <br> TAGs | Fatty acid <br> constituents | Acyl carbon <br> number <br> MANUAL ID | Types of <br> TAGs <br> MANUAL ID | Fatty acid <br> constituents <br> MANUAL ID |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Cook pot, <br> base <br> V11-108A/8 | ND | ND | ND | C50:1 | 1 | $16: 0 / 16: 0 / 18: 1$ |
| Cook pot, <br> base <br> V11-108A/7 | C52:1 | 1 | $18: 1 / 18: 0 / 16: 0$ | ND | ND | ND |

### 6.3 Analysis of samples: Area B


#### Abstract

As stated in the Chapter 1 of this thesis, area B represents strip buildings used for civilian encampment located outside of the borders of the fort. The dates of habitation of the civilian encampment are circa 213-300 A.D., and are contemporaneous with the fort excavated in area A.


### 6.3.1 Context V11-VL-3B

This context was excavated from topsoil to the clay floor of one of the strip buildings, dated to after 213 A.D. based upon the coinage excavated from the area. These objects were excavated in 2011 and were housed in the pottery conservation laboratory.

Two sample classes were taken from this context, storage vessels and mortaria. Mortaria were tools used in meal preparation for grinding ingredients such as oil, wine, vinegar, spices, and plant and animal fats. They were often course grained, shallow bowls with thick side walls able to sustain the pressure of repetitive grinding (Alcock, 2001: 69). Lipid analyses on Iron Age mortaria sherds identified both animal as well as leafy plant products (Cramp, 2008; Cramp et al., 2011).

Sample 4 represents a mortarium base, gray/buff color. The results reported in Table 6.3 identified ferulic, syringic, p-coumaric, vanillic, succinic and malonic acids. Low levels of bound lipids were also tentatively identified. The identification of six out of the nine targeted acids is not highly indicative of wine, particularly in a multi-use environment. In this type of environment, the most highly indicative biomarker fingerprint for wine would be the identification of all nine acids.

Sample 6 was taken from a neck of a storage vessel. The results of the targeted analysis identified succinic and syringic acids, results not indicative of the presence of wine. The identification of syringic acid in both of these samples highlights the danger of a single biomarker approach for a complex foodstuff such as wine, particularly in an environment with other food contaminants. The need for a multi-biomarker approach is critical especially when there is overlap with several of the metabolites.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Base of mortarium V11-3B/4 | . 978 | 83721 | 42630 | ND | ND | 13164 | 21446 | ND | 5520 | 7053 | hexadecanoic, octadecanoic acids and unidentified at $m / z 237$ |
| Neck of storage vessel | 1.01 | ND | 24004 | ND | ND | ND | ND | ND | ND | 2649 | hexadecanoic, octadecanoic, and unidentified at $m / z 237$ |

### 6.3.2 Context V12-VL-30B

Context V12-30B is composed of the rubble found at the base of a separate context (V12-11B, backfill from an 1830s excavation). The date for this context is 213-270 A.D. Four samples were taken for analysis; the results of two of the samples are given below in Table 6.4.

The biomarker fingerprint for the base of a storage vessel sample 4, was not diagnostic for either wine or tentatively diagnostic for olive oil. In contrast, the biomarker fingerprint of sample 3 taken from the side of a cook pot, contained seven out of the nine targeted acids. The results suggest the presence of wine or vinegar, common ingredients in Roman British cooking (Alcock, 2001: 83). Translations from a surviving $2^{\text {nd }}$ century A.D. cookbook written by the Roman writer Apicus, offer a unique perspective as to the recipes of that time and the ingredients suggest a palate that relies on heavily spiced, salty dishes with strong flavors. Current literary sources have collated several of the historic recipes and wine in its original form, its soured form (vinegar), sweetened with honey (caroenum), or as a cooked syrup (defructum), are common ingredients (Renfrew, 1985: 35-44). Several of the recipes are given in the Appendix, Chapter 6, Figure 3.

Concerning the lipid fraction, TAGs were not identified in the total lipid extract of sample 3. Low levels of hexadecanoic, octadecanoic acid as well as a reoccurring deprotonated mass at 237 were tentatively identified in the bound lipid fraction.

|  | Sample weight (g) | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | $\begin{aligned} & \text { p-Coumaric } \\ & \text { acid } \end{aligned}$ | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Base of storage vessel <br> V12-30B/4 | 1.04 | ND | 10963 | 45661 | ND | 2347 | 2189 | ND | ND | 1692 | hexadecanoic octadecanoic acids and unidentified m/z 237 |
| Side of cook pot V12-30B/3 | . 994 | 88718 | 40630 | 161859 | ND | 9310 | 14019 | ND | 9716 | 11591 | hexadecanoic octadecanoic acids and unidentified m/z 237 |

No TAGs identified in VL-30B/3

### 6.3.3 Context V12-VL-21B

This context was excavated to the stone layer which sits above a Period V (circa 120-130A.D.) ditch. The context is secure and represents a civilian encampment from 213-270 A.D. Sample 2 represents the shoulder of a cookpot and the results are given in Table 6.5. The biomarker fingerprint suggests the presence of wine/vinegar as an ingredient with seven out of the nine targeted acids identified including: four phenolic acids, malonic, succinic, and malic acids. In the mass spectrum, the fragmentation profile of the ferulic acid peak was slightly askew, with a higher intensity of the 149 peak. This was explained by the surround co-eluting peaks whose fragmentation pattern was overwhelmingly $m / z$ 149. The tentatively bound lipids included hexadecanoic, octadecanoic, and octadecenoic acids.

TAG analysis of the sample identified a narrow range of acyl carbons from $\mathrm{C}_{48}-\mathrm{C}_{54}$. The identification of odd number fatty acids $\left(\mathrm{C}_{15}, \mathrm{C}_{17}\right.$, and $\left.\mathrm{C}_{19}\right)$ represents a contribution from a ruminant animal, resulting from the fatty acid synthesis by the bacteria within the rumen (Christie, W.W., 1978). However, due to the TAG hydrolytic degradation, it is not possible to determine whether the fats originate from adipose tissue or from dairy products, such as milk or cheese.



### 6.3.4 Context V12-VL-41B

The context is a layer of 'burnt laminate' suggesting the application of heat, either intentionally or an unintentional fire at some point in the past. The context was dated from 231-270 A.D. Sample V1241B/1 was taken from the side of a thick walled amphora that appeared to be fabricated by joined rings or coils (personal communication: Dr. Helen Loney). The results of the targeted LC-MS/MS analysis listed in Table 6.6 identified three phenolic acids as well as succinic acid; therefore, the biomarker fingerprint of four out of the nine targeted acids does not suggest the presence of wine or vinegar. Tentatively identified within the bound lipid portion of the spectrum was octadecenoic acid, $m / z$ 281. Its concentration overwhelms the spectrum and is obvious in Figure 6.5. A probable explanation for this $m / z$ is oleic acid, indicative of olive oil. The identification of olive oil in an archaeological artifact is rare due to the degradation of its composite unsaturated fatty acids, commonly occurring over an archaeological time span due to microbial induced hydrolysis of the double bonds (Eglinton and Logan, 1991). The remarkable preservation of the (tentatively identified) unsaturated fatty acid reveals the capability of anoxic, waterlogged environments to slow the rate of decomposition. As well as the $m / z$ 281, there are smaller contributions from $m / z 171$ (decanoic acid), $m / z 255$ (hexadecanoic acid), and $m / z 295$. The peak at $m / z 295$ may suggest any of the following compounds: a $\mathrm{C}_{19}$ cyclopropane fatty acid, originating from a bacterial lipid, a branched fatty acid, or an isoprenoid. This is conjecture, however, and targeted analysis is necessary for confirmatory evidence.

As stated above, the tentative identification of octadecenoic acid within the amphora suggests the presence of olive oil, a plant oil that is very high in oleic acid and whose TAG composition is predominantly comprised of trioleoylglycerol (OOO) (Rossell and Pritchard (ed.), 1991: 87; Kiritsakis
et al., 2002). Olive oil contains free phenolic acids, which are often used to classify geographical origin; however, the identification of syringic, ferulic, and p-coumaric acids (Table 6.6) probably originates from the alkaline fusion products of the phenolic backbones of oleuropein (CarrascoPancorbo et al., 2004). The identification of syringic acid in the alkaline fusion of an amphora sherd that most likely contained olive oil is especially concerning for those researchers who utilise a single biomarker approach for the identification of wine in amphora. As the main packaging unit for nearly two thousand years, amphora were consistently re-used to ship wine, olive oil, fish sauce, spices, as well as salted fish, resulting in an mixing of compounds (Twede, 2002; Foley et al., 2012).

The total lipid extract and lithiated TAG analysis of VL-41B/1 held limited information, with only three TAGs passing the algorithm filter, $\mathrm{C}_{50: 1} / \mathrm{C}_{51: 3} / \mathrm{C}_{52: 1}$. If in fact there was a fire in this context, the concentration of the total lipid extract appears to have been affected, whereas the olive oil was probably driven into the pottery sustaining its preservation.

|  | Sample weight (g) | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Side amphora V12-41B/1 | . 999 | ND | 6065 | ND | ND | 6970 | ND | ND | 4258 | 2045 | octadecenoic acid (oleic acid) |


|  | Acyl carbon number | Types of <br> TAGs | Fatty acid constituents | Acyl <br> carbon <br> number <br> MANUAL <br> ID | Types of <br> TAGs <br> MANUAL <br> ID | Fatty acid <br> constituents <br> MANUAL ID |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| amphora | C50:1 | 1 | $16 / 16 / 18: 1$ | ND | ND | ND |
| VL-41B/1 | C51:3 | 1 | $19: 3 / 16 / 16$ | $18: 1 / 18 / 16$ |  |  |



20130726615\#259-598 RT: 4.39-10.16 AV: 340 NL: 5.81E6
T: - p ESI Q1MS [40.000-300.000]


Figure 6.5. The TIC and extraction ion chromatogram of the LC-MS/MS analysis of sample V12-41B/1, a sherd from the side of an amphora. The spectrum is dominated by the elution at 4.7 minutes of a 'bound' lipid, tentatively identified as oleic acid based upon $\mathrm{m} / \mathrm{z} 281$.

### 6.3.5 Context V12-VL-47B

This context represents fill material located over an aqueduct consisting of rubble and clay over the supporting wall. The context is dated to the period 130-205 A.D. Five samples were taken from the area, four cook pots and one piece of domestic ware. Visually, the bulk of the samples have gray fabric, except for the visually dissimilar sample 2 representing the domestic ware (e.g., dinner plate), which consists of a red fabric (personal communication: Dr. Helen Loney).

Samples 3 and 5 are samples from a cook pot; sample 3 represents a side wall and sample 5 represents a rim of a cook pot. At this time, it has not yet been confirmed by an archaeologist whether the samples originate from the same object. However, their biomolecular fingerprints are very similar and strongly indicate the presence of wine or vinegar with the identification of eight of the nine targeted acids, Table 6.7. As was stated earlier, wine either as liquid, syrup, or vinegar, was consistently used in cooking. Comparing the two samples, the intensities are similar for five of the acids: malonic, succinic, malic, tartaric, and vanillic; however, the intensities of syringic, ferulic, and p-coumaric are considerably less in sample 3. One explanation for this drop in intensity is sample preparation or more specifically the interactions between the three phenolic acids in the sample with the ring on the phenyl sulfonate group, the active group on the strong cation exchange SPE cartridge. Upon re-examination of Figure 5.18 (c), there is a difference in the absolute intensities between the syringic acid, with and without SPE clean up. Compared with the gain in intensity for vanillic, malonic, and malic acids, the gain in intensity for the syringic acid after the SPE step is moderate, and at times, the intensity of syringic acid is slightly less after the clean-up procedure. Since all samples were collected as flow through without a wash step, these results suggest the
addition of a wash step is necessary. A dilute salt solution may work to break up any interaction between the ring structures yet not disrupt the aluminate ion binding to the sulfate group.

The biomolecular fingerprint of samples 1,2 , and 4 , Table 6.7, were negative for wine ( $3 / 9$ and 5/9 acids) and inconclusive for the identification of foodstuffs. The total lipid extract for sample 1 yielded no TAG information.

## Table 6.7 Context V12-VL-47B

|  | Sample weight (g) | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | pCoumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Base of cook pot V12-47B/1 | 0.989 | ND | 9390 | ND | ND | 462 | ND | ND | ND | 128 | hexadecanoic, octadecanoic acids |
| Domestic ware V12- 47B/2 | 0.998 | 70366 | 28865 | ND | ND | 1335 | 1935 | ND | ND | 3711 | hexadecanoic, octadecanoic acids |
| Side of cook pot V12-47B/3 | 1.04 | 74806 | 41103 | 74220 | 1576 | 2857 | 10093 | ND | 4487 | 5205 | hexadecanoic, octadecanoic acids |
| Shoulder of cook pot V12-47B/4 | 0.997 | 80400 | 46721 | ND | ND | 4644 | 7881 | ND | ND | 5788 | hexadecanoic, octadecanoic acids, unidentified $m / z 237$ |
| Rim of cook pot V1247B/5 | 1.02 | 69913 | 43884 | 117159 | 1151 | 8757 | 11889 | ND | 12791 | 16994 | hexadecanoic, octadecanoic acids, unidentified $m / z 237$ |

No TAGs identified in VL-47B/1

### 6.3.6 Context V12-VL-49B

Context 49B is a drainage ditch which contained animal bones, pottery, and cobble stones and dated to circa 213 A.D. The most informative sample taken from this context is sample 3, which represents the rim of a cook pot. The total lipid extract TAG analysis identified TAGs from $\mathrm{C}_{44^{-5} 5: 2}$, Table 6.8 and Figure 6.6. The odd number acyl carbons, $\mathrm{C}_{15}, \mathrm{C}_{17}$, and $\mathrm{C}_{19}$, identify a ruminant animal, whereas the appearance of short acyl chains such as $\mathrm{C}_{14}$ and $\mathrm{C}_{12}$, suggest a dairy origin. Cheese was a common foodstuff in Roman Britain and is found in several recipes (Alcock, 2001: 59-62; Renfrew, 1985: 36, 44). As a comparison, Appendix, Chapter 6, Table 2 illustrates the TAGs identified in a laboratory aged sample of cow's milk. The short chain fatty acids from $\mathrm{C}_{4}-\mathrm{C}_{14}$ are evident in the standard. It has been theorised, that the shorter acyl chain are more readily hydrolysed and therefore do not remain over the period of an archaeological lifetime (Evershed, 2002).


Figure 6.6. TAG acyl carbon distribution VL-49B/3.

Table 6.8 TAG results from the total lipid extract of VL-49B/3. The acyl carbon distribution ranges from $\mathrm{C}_{44}-\mathrm{C}_{54: 2}$. The fatty acid constituents identify a ruminant animal; the presence of short acyl chains ( $\mathrm{C}_{12}$ and $\mathrm{C}_{14}$ ) suggest a dairy origin, rather than adipose fat.

|  | Acyl carbon number | Types of <br> TAGs | Fatty acid constituents |  |  |  |  | Acyl carbon number MANUAL ID | Types of TAGs MANUALID | Fatty acid constituents MANUAL ID |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} \hline \text { cook pot, } \\ \text { rim } \\ \hline \end{gathered}$ | C44:0 | 2 | 141/14/16 | 16/12/16 |  |  |  | ND | ND | ND |
| VL-49B/3 | C46:0 | 2 | 16/16/14 | 16/12/18 |  |  |  |  |  |  |
|  | C48:0 | 3 | 16/18/14 | 16/16/16 | 18/12/18 |  |  |  |  |  |
|  | C48:1 | 2 | 18:1/14/16 | 16/16:1/16 |  |  |  |  |  |  |
|  | C49:1 | 3 | 18:1/15/16 | 18:1/14/17 | 16/16/17:1 |  |  |  |  |  |
|  | C49:3 | 1 | 18:3/15/16 |  |  |  |  |  |  |  |
|  | C50:1 | 4 | 18:1/16/16 | 18:1/14/18 | 16/16:1/18 | 16/15/19:1 |  |  |  |  |
|  | C50:2 | 3 | 18:1/14/18:1 | 18:1/16/16:1 | 18:2/16/16 |  |  |  |  |  |
|  | C51:4 | 2 | 18:4/16/17 | 17:3/18:1/16 |  |  |  |  |  |  |
|  | C52:1 | 5 | 16/16/20:1 | 16/17/19:1 | 16/18:1/18 | 18/16:1/18 | 18/15/19:1 |  |  |  |
|  | C52:2 | 4 | 16/18/18:2 | 16/17/19:2 | 16/18:1/18:1 | 18:1/18/16:1 |  |  |  |  |
|  | C54:1 | 2 | 18:1/18/18 | 18/16/20:1 |  |  |  |  |  |  |
|  | C54:2 | 4 | 18/18:1/18:1 | 18/18/18:2 | 18/16/20:2 | 18:1/16/20:1 |  |  |  |  |

### 6.3.7 Context V12-VL-51B

This context is a fabric of road which ran from the northwest- to the southeast, along the aqueduct next to the strip building. The date of the context is 213 A.D. Four samples were analysed, the results are given for three samples and are listed in Table 6.9. The biomolecular fingerprint of sample 1, an amphora base, was negative for the presence of wine or vinegar since only $1 / 9$ of the acids were identified. The results of the targeted analysis identified only malonic acid; there was no evidence of bound lipids in the spectrum. The results of sample 4, a side wall of a mortarium, were negative for wine or vinegar content identifying only four of the nine targeted acids. The four acids that were identified included malonic, succinic, vanillic, and syringic. The origin of the acids is unknown at this time.

Sample 2 represents the upper portion of a cook pot. The biomolecular fingerprint of sample 2 suggests the presence of wine or vinegar with the identification of seven of the nine targeted acids. Neither tartaric acid nor iso-propyl malic acid was identified; however, the phenolic acids were identified, as well as malonic, succinic, and malic acids. Only one TAG was identified (manually) in this sample, $\mathrm{C}_{51: 3}$ which does not offer enough information for possible origin.

Table 6.9 Context V12-VL-51B, fabric of 3 ${ }^{\text {rd }}$ century road, circa 213 A.D.

|  | Sample weight | Malonic acid | Succinic acid | Malic <br> Acid | Tartaric acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} \hline \text { Base of amphora V12- } \\ 51 \mathrm{~B} / 1 \\ \hline \end{gathered}$ | 0.999 | 101361 | ND | ND | ND | ND | ND | ND | ND | ND | ND |
| mortarium V12-51B/4 | 0.999 | 139670 | 78303 | ND | ND | ND | 6348 | ND | ND | 4535 | ND |
| Shoulder of cook pot V12-51B/2 | 1.02 | 105195 | 62454 | 82802 | ND | 27466 | 21863 | ND | 22416 | 26192 | hexadecanoic, octadecanoic acids, and unidentified $m / z 237$ and $m / z$ 225 |

$\left.\begin{array}{|c|c|c|c|c|c|c|}\hline & \begin{array}{c}\text { Acyl carbon } \\ \text { number }\end{array} & \begin{array}{c}\text { Types of } \\ \text { TAGs }\end{array} & \begin{array}{c}\text { Fatty acid } \\ \text { constituents }\end{array} & \begin{array}{c}\text { Acyl carbon } \\ \text { number } \\ \text { MANUAL ID }\end{array} & \begin{array}{c}\text { Types of } \\ \text { TAGs } \\ \text { MANUAL ID }\end{array} & \begin{array}{c}\text { Fatty acid } \\ \text { constituents } \\ \text { MANUAL ID }\end{array} \\ \hline \text { cook pot, shoulder } & \text { ND } & \text { ND } & \text { ND } & \text { C51:3 } & 1 & 16: 1 / 16: 1 / 16: 1 \\ \hline \text { VL-51B/2 } & \text { Acyl carbon } \\ \text { number }\end{array} \begin{array}{c}\text { Types of } \\ \text { TAGs }\end{array} \quad \begin{array}{c}\text { Fatty acid } \\ \text { constituents }\end{array} \quad \begin{array}{c}\text { Acyl carbon } \\ \text { number } \\ \text { MANUAL ID }\end{array} \begin{array}{c}\text { Types of } \\ \text { TAGs } \\ \text { MANUAL ID }\end{array} \begin{array}{c}\text { Fatty acid } \\ \text { constituents } \\ \text { MANUAL ID }\end{array}\right]$

### 6.3.8 Context V12-VL-5B

Two samples were analysed from this context which is comprised of unstratified soil separating two other contexts, and dated to 353-358 A.D. The results for sample 2 are given below in Table 6.10. Sample 2 is a side wall from a mortarium, its biomolecular fingerprint is indeterminate for wine/vinegar since only six of the nine targeted acids were identified.

No TAGs were identified in sample 1, cook pot.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} \text { mortarium } \\ \text { V12-5B/2 } \\ \hline \end{gathered}$ | . 978 | 33129 | 27074 | ND | ND | 9629 | 10354 | ND | 7318 | 3774 | ND |


|  | Acyl carbon <br> number | Types of <br> TAGs | Fatty acid <br> constituents | Acyl carbon <br> number <br> MANUAL ID | Types of <br> TAGs <br> MANUAL ID | Fatty acid <br> constituents <br> MANUAL ID |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| cook pot, side | ND | ND | ND | ND | ND | ND |
| VL-5B/1 | ND |  | ND |  |  |  |

### 6.4 Soil samples

Table 6.11 represents the results of the analysis of two soil samples collected on the same day by volunteer excavators. One of the soil samples was taken from context 52 which is comprised of yellowish clay that is bound on one side by rough cut stones and dated to 213 A.D; this sample was extracted and analysed twice. A separate soil sample was taken which did not have a context designation, yet was adhered to the base of an amphora or storage vessel.

Succinic acid was the only acid identified in the soil samples 1 a and 1 b taken from context 52 . In contrast, targeted analysis of soil sample 2 from the unknown context identified seven out of the nine targeted acids. This sample was also devoid of 'bound' lipids. Rather than suggest the presence of wine, soil sample 2 may represent the monomeric species extracted from alkaline fusion applied to a humic polymer. The complexity of a soil sample cannot be underestimated, particularly a soil as rich in organic matter as the soil surrounding the Vindolanda settlement. The soil and humic substances (organic decomposition) encasing objects could include leaf litter, humic acid, and/or lignin polymers and the application of an alkaline material applied to any of these materials may release phenolic acids including benzoic and cinnamic acids (Vanholme et al., 2010; Burges et al., 1964; Rawlins et al., 2006; Bourbonniere and Meyers, 1983).

Concerning the organic acids, citric and malic acid have been identified in the rhizosphere, transported from a plants' roots into soil via the changing electrical gradient across cellular membranes in response to the plants nutritional needs (Rozycki and Strzelczyk, 1986). The symbiotic microbial population that cluster around the surface of the roots, however, metabolise those acid exudates, with some research suggesting $60 \%$ of the acid being transformed to $\mathrm{CO}_{2}$, and
incorporating the remaining 40\% leaving trace levels of the acids within the soil (Jones, 1998). Therefore, the intense malic acid peak is unexpected. Unfortunately, there is limited information concerning the alkaline hydrolysis of soils accompanying archaeological sherds and the current hypothesis is based upon the research carried out in this thesis. One important note is that the absence of acid signals (besides succinic acid) in soil samples 1 a and 1 b, versus the identification of seven acids in soil sample 2 highlights the immediate surroundings of soil samples and how quickly the soils' composition can change, probably due to polymeric networks, either wine polymers, lipid polymers, or humic acid polymers lodged within the soil preventing mass migration over long distances. Due to the unidentified context of this sample, a thorough explanation is unavailable; however, it should be taken as a cautionary tale for further investigations, particularly in the identification of a wine polymer whose biomarker list may overlap with several compounds found in humic substances. The actual nature of the humic substances is widely varied and dependent upon the organic decomposition from past lived fauna and flora. Therefore, future analyses of archaeological objects particularly collected in a heavy peat environment, should prepare and extract the adhered soil samples in the same fashion as the examined object.

Figure 6.7 compares six acids across soil sample 2 and five samples of the cooking pots representing four contexts: 52B, 47B, 21B, and 30B. The compared acids are: syringic, ferulic, vanillic, p-coumaric, tartaric and malonic. The difference in malonic acid is evident between the soil sample and the five pottery samples; malonic is considerably higher in the pottery samples than in the soil sample. Also, the level of syringic acid appears to be higher in the soil sample when compared with four of the five pottery samples. Figure 6.8 compares succinic acid and malic acid across the same samples. The intensities of these two acids are similar and suggest that, at least in a wet peat environment, these two acids occur relatively frequently and may not be descriptive for wine.

Table 6.11 Three soil samples were analysed by targeted LC-MS/MS for phenolic and organic acid content.

|  | Malonic acid | Succinic acid | Malic <br> Acid | Tartaric <br> acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Soil sample 1a Context 52: 1.0g | ND | 8376 | ND | ND | ND | ND | ND | ND | ND | hexadecanoic, octadecanoic acids |
| Soil sample 1b <br> Context 52: .99g | ND | 5455 | ND | ND | ND | ND | ND | ND | ND | hexadecanoic, octadecanoic acids |
| Soil sample 2 <br> Unknown context: .99g | 13617 | 65743 | 116758 | ND | 4744 | 5216 | ND | 4151 | 28547 | ND |



Figure 6.7. A comparison between soil sample 2 and five samples of cooking pots with the following acids: syringic, ferulic, vanillic, p-coumaric, tartaric, and malonic.


Figure 6.8. A comparison between soil sample 2 and five samples of cooking pots with the following acids: succinic and malic.

### 6.5 Discussion

### 6.5.1 Cooking Pots

As stated earlier in the chapter, in order to determine the presence or absence of wine in these samples, nine acids were targeted in the LC-MS/MS analysis. The nine acids chosen were those acids identified in the wine press from Chapter 5 and are considered highly indicative for the presence of wine: malic, iso-propyl malic, tartaric, succinic, vanillic, ferulic, p-coumaric, syringic, and malonic acids. Two out of the eight samples of cooking pots had biomarker fingerprints that strongly indicated the presence of wine or vinegar since 8/9 targeted acids were identified. The two sherds $\mathrm{V} 12-47 \mathrm{~B} / 3$ and $\mathrm{V} 12-47 \mathrm{~B} / 5$ were taken from the same context and may represent the same cooking pot. Out of the remaining six analysed samples of the cooking pots, two of the sherds contained 7/9 targeted acids, suggesting the presence of wine and/or vinegar; one of these samples also contained TAGs indicative of ruminant animals. That cooking pots are rich in information is not unknown and it has been theorised in earlier investigations that the application of heat drives material into the pottery matrix (Heron and Evershed, 1993).

### 6.5.2 Mortaria

The three mortaria extracted had a similar biomarker fingerprint that included malonic, succinic, vanillic, and syringic acids, with two of the samples containing p-coumaric and ferulic acid. The results did not indicate the presence of wine or vinegar in the mortaria. Two of the samples had intense signals for vanillic acid suggesting a phenolic rich biomarker fingerprint left by the foodstuffs. A small aging study of foodstuffs identified phenolic rich spices; however, the fingerprint gathered
from the laboratory aging study does not correlate with the results from the mortaria. Therefore, more aging studies are required to identify a suite of biomarkers representative of Roman British foodstuffs

### 6.5.3 Amphora

None of the amphora or the storage vessels indicated the presence of wine based upon the results of the targeted LC-MS/MS analysis. This is consistent with archaeological research in Vindolanda that reveals the main use for amphorae was to transport olive oil from the Iberian peninsula to the Vindolanda settlement (Sheehan-Finn, 2012). Wine delivered to the settlement was often transported through Gaul in wooden barrels (Symonds and Mason, (ed.) 2009). Based upon documentation in comparison with excavations records spanning the years 80-140 A.D., the percentage of amphora that transported wine (851) compared with the number of wooden barrels that transported wine $(108,501)$ was $0.8 \%$ (Birley and Blake, 2005).

### 6.5.4 Soil

Two soil samples were analysed in this research. Succinic acid was the only acid identified in one sample; however, 7/9 targeted acids were identified in the second soil sample. The results suggest an overlap in the aged wine targeted acids and the alkaline fusion portion of an organic rich peat soil. Although the intensities of the acids identified in the soil varied when compared with the acids identified in the cook pots, the results highlight the possible effect of the archaeological environment on the chosen biomarkers. The results of the soil sample may reflect the breakdown of a humic acid polymer lodged within the peat matrix.

Concerning these results, it is theorised that an aged wine polymer may not move from the pottery and travel into the surrounding soil; similarly, it is also suggested that the polymeric materials found in humic rich soil may not travel distances through the soil causing interference with the archaeological pottery. However, at this time, this situation is only hypothetical and requires multiple testing of archaeological sherds and the adhered soil to answer the question of humic polymer/aged wine interaction, following the protocols of a previous study which examined lipids in the soil of surrounding pottery sherds (Heron et al., 1991).

### 6.6 Conclusions

The objectives put forth in the introduction to this thesis were to develop a diagnostic suite of biomarkers to apply to archaeological artifacts in order to determine the presence or absence of wine. The list of biomarkers developed in Chapters 3 and 4 was successfully applied to an ancient wine press from Sardinia, Italy. The biomarker list was then applied to samples excavated from Vindolanda, a wet anoxic environment with rich, organic soil. Out of the 32 samples analysed in Chapter 6, wine was strongly indicated in two of the samples: V12-47B/3 and V12-47B/5. The two sherds were taken from the same context, and eight of the acids were identified in each sample.

Comparisons between the levels of succinic acid and malic acid amongst a soil sample and five cook pots revealed that these two acids may not be diagnostic for wine under the scenario of this environment. In order to prove these acids uniqueness for wine within a peat rich environment, a successful laboratory aging study focusing on a wet peat environment should be undertaken in order to determine if these acids are still characteristic for laboratory aged wine.

## 7. Conclusions and Future Work

Four specific aims were identified in Chapter 1 of this thesis in an attempt to apply metabolomics to the analysis of ancient organic residue. The first aim set out in Chapter 3 was to age modern wine enriched sherds in a laboratory setting and then successfully extract those materials that represent the aged, polymerised wine. Once the materials were extracted, the samples were then analysed in a globally untargeted approach. Multi-variate statistics were applied to the final analysis and the top 150 loadings, most discerning for wine were chosen to prepare a targeted biomarker list.

In order to accomplish this aim, sample sherds were aged in several different environments in order to mimic archaeological environments: standard clean glass, dry sand, and wet peat. A successful metabolic signature for the aged wine was extracted from the clean jar and dry sand environments. However, there was no obvious signature from the wet peat. The reason may be that the wine never formed a polymeric network within the clay matrix, possibly due to the fact that water was continuously added to the jar in order to maintain moisture over the six month period. For future studies, a change in the methodology would be to age the wine permeated sherds for six months in glass jars and then introduce those sherds to the three specific environments (glass jar, dry sand, and wet peat) for further aging.

Concerning the liquid aging study and the identified six month plateau, the time point at which the metabolic signature had ceased to change, this plateau may be analogous to the initial polymerisation that occurs during grape treading, whereby the most reactive species are condensing to form larger polymers. There may be other plateaus which occur during the aging/polymerisation; however, due to time constraints, this thesis was limited to the identification of the first plateau. A recent article identified pinotin pigments forming after a year of aging indicating a continuation of
the aging/polymerisation process (Arapitsas et al., 2014). Whether the kinetics of polymerisation allows for a second plateau was untested in this initial research and should be pursued in further studies by monitoring the aging of liquid aliquots over at least a two year period.

The second aim of this research was to develop a sensitive targeted LC-MS/MS method based upon the biomarkers determined from Chapter 3 in order to detect trace level analytes. A HILIC method was successfully prepared for the trace analysis of thirteen acids for validation studies on objects excavated from two archaeological sites: ketobutyric, ferulic, gentisic, syringic, m-hydroxycinnamic, p-coumaric, 2,3-dihydroxybenzoic, vanillic, succinic, malonic, malic, tartaric, and isopropyl malic acid. Prior analyses utilised reversed phase liquid chromatography for the initial separation; however, using the orthogonal approach increased the capacity factor for the studied acids. The longer elution times allowed for more robust retention times and better reproducibility between samples and experiments. A short description of the acids on the biomarker list follows.

Gentisic acid and 2,3 dihydroxybenzoic acid were both chosen from the untargeted analysis discussed in chapter 3. However, the untargeted DIMS yielded only a $m / z$ and a suggested molecular formula, $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{4}$ with six possible choices for dihydroxybenzoic (DHB) acids. The final acids chosen were based upon literature searches on identified wine acids as well as on the final success of the LC elution profile. Neither 2,3-DHB or gentisic acid, identified as a wine acid, were identified in any of the archaeological samples. Another DHB found in wine is 3,4-DHB, protocatechuic acid. This acid is recommended as a replacement for the current acids on the biomarker list at $m / z 153$.

Meta-hydroxycinnamic and para-coumaric acids were two phenolic acids also chosen from a possible list of six isomers identified from one $m / z$. However, by comparing the fragmentation intensities between the standards and the archaeological results, p -coumaric was identified in the archaeological samples. Other phenolic acids include syringic, vanillic and ferulic acids. The initial
standard of ferulic acid analysed was trans-ferulic acid; however, the identification of ferulic acid in the archaeological samples revealed several extra peaks within the acid's retention time window which suggests an addition of cis-ferulic acid. It is unclear if the cis-ferulic acid is due to the aged wine polymer, or if it is an effect of extracting the samples in light. After SPE clean-up, however, the isomers appear to combine and elute in one larger peak.

Other acids include ketobutyric, malonic and succinic acids. Ketobutyric acid was not identified in the archaeological samples of Chapter 5 and was removed from the targeted biomarker list in Chapter 6. Malonic and succinic acids were consistently two of the most abundant acids identified in the archaeological samples.

Malic, tartaric, and isopropyl malic acid were the three organic acids used in this research to distinguish a list of acids identified in an archaeological sample from suggestive to highly indicative of wine. As stated in the introduction, earlier published results identified wine in archaeological amphora by the presence of tartaric acid as the sole biomarker due to its high concentration within grapes. However, this research has shown that identifying any foodstuff origin based upon only one or two biomarkers is risky and increases the chance of false positives. Since tartaric acid may also originate from microbial excretions, its identification should therefore be taken in combination with the remaining acids (Galkin et al., 1998).

The third aim of this thesis was to validate the targeted method on archaeological samples from a sandy soil environment of Sardinia, Italy, in an area with a rich viticulture history. For the first time, a metabolomics approach utilising a predetermined list of biomarkers was successfully applied to the analysis of an archaeological wine press, identifying nine of the thirteen targeted acids. The targeted analysis of amphorae sherds yielded indeterminate results since six or fewer of the acids were
identified, indicating the precarious nature of assigning only one or two biomarkers to the identification of ancient foodstuffs.

The fourth aim of this thesis was to apply that targeted method to the analysis of archaeological objects found in an environment with heavy food contamination in order to test the robustness of the biomarker list. As stated in Chapter 6, the original biomarker list containing thirteen acids was dropped to the nine acids identified in the Sardinian wine press; these nine acids are considered highly indicative for the presence of wine: malic, iso-propyl malic, tartaric, succinic, vanillic, ferulic, p-coumaric, syringic, and malonic acids. Two samples taken from cooking pots from the Vindolanda settlement yielded acid signatures highly indicative of wine based upon the identification of eight of the nine targeted acids. This result suggests a successful application to a multi-use environment and is the first time that the presence of wine was indicated in an ancient cook pot. In addition, the nonpolar extracts were analysed for TAGs, and in several cases, combining the results of the alkaline fusion extracts with the results from the non-polar ones resulted in a highly descriptive story for the object and foodstuff origin.

The soil and environs of Vindolanda is a wet environment with an organic rich soil containing humic acid polymers. Based upon the initial analyses of a limited number of soil samples, there may be overlap between the targeted biomarker list for aged wine and humic polymers from peat-rich soils. This result highlights the importance of analysing the soil surrounding the archaeological objects. In order to test the theory that the humic acid polymers are immovable in the soil and therefore are not responsible for the biomarker fingerprint found in the excavated pottery sherds, a separate experiment should be carried out whereby in a peat rich environment, alkaline fusion is applied to the excavated pottery sherds and to the adhering soil for at least $\mathrm{n}=10$, similar to a previous study on soil lipids (Heron et al., 1991).

As was identified in Chapter 5, applying an alkaline solution to pottery leaches out aluminate ions which may chelate the polyprotic acids such as malic and tartaric acids. The current method introduced a novel sample preparation step which removed the aluminate ion by cleaning up the sample on a strong cation exchange SPE column. In order to improve the method, the following procedures could be modified: decrease the 24 h time frame the samples spend in the acidified solvent prior to SPE clean-up, use a larger cartridge bed for SPE ( 500 mg instead of 100 mg ), and introduce a wash step consisting of a dilute salt solution to break up any pi-pi interactions on the cartridge.

The metabolomics approach of global untargeted analysis in combination with statistical analysis resulted in a targeted biomarker list for the identification of aged wine residue that was successfully applied to two separate archaeological sites. This approach could also be applied to model sherds enriched with multiple foodstuffs, allowed to age over a prescribed time point. Using any of the recipes found from Roman texts, model sherds enriched with those representative foodstuffs and aged over a prescribed time could offer more specific biomarker lists which could add to the robustness in an environment with heavy food residues. The results of this type of analyses may yield biomarker information for objects such as mortaria, which although only a limited number were analysed, had fairly consistent phenolic fingerprints. It would not be surprising that repetitive grinding of phenolic rich foods would leave a fingerprint within the pottery matrix.

## 8. References

Adams, J., Gross, M.L. (1986). Energy requirements for remote charge site ion decompositions and structural information from collisional activation of alkali metal cationized fatty alcohols. Journal of the American Chemical Society. 108(22): 6915-6921.

Agilent Zorbax column selection guide for HPLC, 2007.

Alcock, J.P. (2001). Food in Roman Britain. The History Press Ltd, Gloucestershire, United Kingdom.

Alpert, A.J. (1990). Hydrophilic-interaction chromatography for the separation of peptides, nucleic acids and other polar compounds. Journal of Chromatography A. 499(19): 177-196.

Arapitsas, P., Speri, G., Angeli, A., Perenzoni, D., and Mattivi, F. (2014) Kinetics and timing of the aging: The influence of storage on the "chemical age" of red wines. Metabolomics 10: 816-832.

Aronson, J.N. (1983). The Henderson-Hasselbalch Equation Revisited. Biochemical Education. 11(2): 68.

Balakian, S. and H.W. Berg, H.W. (1968). The role of polyphenols in the behavior of potassium bitartrate in red wines. American Journal Enology and Viticulture. 19(2): 91-100.

Barnard, H., Dooley, A.N., Areshian, G., Gasparyan, B. and Faull, K.Y. (2010). Chemical Evidence for Wine Production around 4000 BCE in the Late Chalcolithic Near Eastern Highlands. Journal of Archaeological Science. 38(5): 977-984.

Benjamini, Y. and Hochberg, Y. (1995). Controlling the False Discovery Rate: A Practical and Powerful Approach to Multiple Testing. Journal of the Royal Statistical Society. Series B. 57(1): 289-300.

Berg, H.W. and Keefer, R.M. (1958). Analytical determination of tartrate stability in wine I. Potassium bitartrate. American Journal Enology and Viticulture. 9(4): 180-193.

Bete's-Saura, C., Andre's-Lacueva, C., and Lamuela-Raventos, R.M. (1996). Phenolics in White Free Run Juices and Wines from Penede`s by High-Performance Liquid Chromatography: Changes during Vinification. Journal of Agricultural Food Chemistry. 1996, 44(10): 3040-3046.

Birley, A. and Blake, J. (2005) Vindolanda: The Excavations of 2003/2004. Vindolanda Trust, Chesterholm Museum, Bardon Mill.

Birley R. (2009). Vindolanda: A Roman Frontier Fort on Hadrian's Wall. Amberley Publishing, Stroud Gloucestershire.

Birley, A. (2010). The nature and significance of extramural settlement at Vindolanda and other selected sites on the Northern Frontier of Roman Britain. PhD Thesis, University of Leicester, School of Archaeology and Ancient History.

Bligh, E.G. and Dyer, W.J. (1959). A rapid method for total lipid extraction and purification. Canadian Journal of Biochemistry and Physiology. 37(8): 911-917.

Bodennec, J., Koul, O., Aguado, I., Brichon, G., Zwingelstein, G., and Portoukalian, J. (2000). A procedure for fractionation of sphingolipid classes by solid-phase extraction on aminopropyl cartridges. Journal of Lipid Research. 41(9): 1524-1531.

Boulton, R. (2001). The Copigmentation of Anthocyanins and Its Role in the Color of Red Wine: A Critical Review. American Journal of Enology and Viticulture. 52(2): 67-87.

Bourbonniere, R.A. and Meyers, P.A. (1983). Characterization of sedimentary humic matter by alkaline hydrolysis. Org. Geochem. 5(3): 131-142.

Brouillard, R., Mazza, G., Saad, Z., Albrecht-Gary, A.M. and Cheminattet, A. (1989). The Copigmentation Reaction of Anthocyanins: A Microprobe for the Structural Study of Aqueous Solutions. Journal of the American Chemical Society. 11 I (7): 2604-2610.

Brouillard, R., Chassaing, S., and Fougerousse, A. (2003). Why are grape/fresh wine anthocyanins so simple and why is it that red wine color lasts so long? Phytochemistry. 64(7): 1179-1186.

Brown, S.C., Kruppa, G. and Dasseux, J.L. (2005). Metabolomics applications of FT-ICR mass spectrometry. Mass Spectrometry Reviews. 24(2): 223-231.

Burges, N.A., Hurst, H.M. and Walkden, B. (1964). The phenolic constituents of humic acid and their relation to the lignin of the plant cover. Geochimiea et Cosmochimica Acta. 28(10-11): 1517-1554.

Buszewski, B. and Noga, S. (2012). Hydrophilic interaction liquid chromatography (HILIC)--a powerful separation technique. Analytical and Bioanalytical Chemistry. 402(1): 231-247.

Carnacini A.B., Capella P., and Amati A (1980). Volatile components of Albana wine. IV. Hydroxy compounds. American Journal of Enology and Viticulture. 31(4): 313-315.

Carrasco Pancorbo A., Cruces-Blanco, C., Carretero, A.S., and Gutiérrez, A.F. (2004). Sensitive Determination of Phenolic Acids in Extra-Virgin Olive Oil by Capillary Zone Electrophoresis. Journal of Agriculture and Food Chemistry. 52 (22): 6687-6693.

Caullet, P. and Guth. J.L. (1989). Observed and Calculated Silicate and Aluminosilicate Oligomer Concentrations in Alkaline Aqueous Solutions. Zeolite Synthesis; ACS Symposium Series; American Chemical Society: Washington, DC.

Cech, N.B. and Enke, C.G. (2001). Practical implications of some recent studies in electrospray ionization fundamentals. Mass spectrometry reviews. 20(6): 362-387.

Charters S., Evershed R.P., Blinkhorn P.W., Denham V. (1995). Evidence for the mixing of fats and waxes in archaeological ceramics. Archaeometry. 37: 113-127.

Chemistry of Wine (2010). Cheynier, V., Schneider, R., Salmon,J-M, and Fulcrand, H. Elsevier.

Christie, W.W. (1978). The composition, structure and function of lipids in the tissues of ruminant animals. Progress in Lipid Research. 17(2): 111-205.

Columella, L.J.M. (publ. 1955). On Agriculture X-XII on Trees. Cambridge, Massachusetts: Harvard University Press.

Comisarow, M.B. (1985). Fundamental aspects and applications of Fourier transform Ion-Cyclotron Resonance Mass Spectrometry. Analytica Chimica Acta. 178: 1-15.

Cooper, H.J., Marshall, A.G. (2001). Electrospray ionization Fourier transform mass spectrometric analysis of wine. Journal of Agriculture and Food Chemistry. 49(12): 5710-5718.

Copley, M.S., Berstan, R., Mukherjee, A.J., Dudd, S.N., Straker, V., Payne, S., and Evershed, R.P. (2005). Dairying in antiquity. III. Evidence from absorbed lipid residues dating to the British Neolithic. Journal of Archaeological Science. 32: 523-546.

Cramp, L. (2008). Foodways and Identity: Organic residue analysis of Roman mortaria and other pottery. Thesis, University of Reading, Department of Archaeology, School of Human and Environmental Science.

Cramp, L.E., Evershed, R.P. and Eckardt, H. (2011). 'What was a mortarium used for? Organic residues and cultural change in Iron Age and Roman Britain.'. Antiquity. 85: 1339-1352.

Creek, D., Jankevics, A., Breitling, R., Watson, D.G., Barrett, M.P. and Burgess, K.E.V. (2011). Toward Global Metabolomics Analysis with Hydrophilic Interaction Liquid Chromatography Mass Spectrometry: Improved Metabolite Identification by Retention Time Prediction. Analytical Chemistry. 83(22): 8703-8710.

Davies, R.W. (1971). The Roman Military Diet. Britannia 2: 122-142.

Dettmer, K., Aronov, P.A., and Hammock, B.D. (2007). Mass spectrometry-based metabolomics. Mass spectrometry reviews. 26(1): 51-78.

Di Rita, F. and Melis, R.T. (2013). The cultural landscape near the ancient city of Tharros (central West Sardinia): vegetation changes and human impact. Journal of Archaeological Science. 40(12): 4271-4282.

Dienes, T., Pastor, S.J., Schurch, S., Scott, J.R., Yao, J., Cui, S., and Wilkins, C.L. (1996). Fourier Transform Mass Spectrometry- Advancing Years (1992-mid 1996). Mass Spectrometry Reviews. 15(3): 163-211.

Dieterle, F., Ross, A., Schlotterbeck, G. and Senn, H. (2006). Probabilistic quotient normalization as robust method to account for dilution of complex biological mixtures. Application in H-1 NMR metabonomics. Analytical Chemistry. 78(13): 4281-4290.

Dinh, N.P., Jonsson, T., and Irgum, K. (2011). Probing the interaction mode in hydrophilic interaction chromatography. Journal of Chromatography A. 1218(35): 5880-5891.

Dinsmore-Webb, A. (ed.) (1974). Chemistry of Winemaking. Washington, D.C.: American Chemical Society.

Dole, M., Mack, L.L., Hines, R.L., Mobley, R.C., Ferguson, L.D. and Alice, M.B. (1968) Molecular Beams of Macroions. The Journal of Chemical Physics. 49(5): 2240-2249.

Douglas, D.J. (2009). Linear quadrupoles in mass spectrometry. Mass spectrometry reviews. 28(6): 937-960.

Dudd S.N., Evershed R.P. (1998). Direct demonstration of milk as element of archaeological economies. Science. 282: 1478-1481.

Dunn, W.B. and Ellis, D.I. (2005). Metabolomics: Current analytical platforms and methodologies. Trends in Analytical Chemistry. 24(4): 285-294.

Dunn, W.B., Erban, A., Weber, R.J.M., Creek, D.J., Brown, M., Breitling, R., Hankemeier, T. Goodacre, R., Neumann, S., Kopka, J., and Viant, M.R. (2013). Mass appeal: metabolite identification in mass spectrometry-focused untargeted metabolomics. Metabolomics. 9(supplement): 44-66.

Dunn, W.B., Wilson I.D., Nicholls A.W., and Broadhurst D. (2012). The importance of experimental design and QC samples in large-scale and MS-driven untargeted metabolomic studies of humans. Bioanalysis. 4(18): 2249-2264.

Eglinton, G. and Logan, G.A. (1991). Molecular preservation. Philosophical Transactions of the Royal Society B: Biological Sciences. 333(1268): 315-327; discussion 327-328.

Ehling, S. and Cole, S. (2011). Analysis of organic acids in fruit juices by liquid chromatography-mass spectrometry: an enhanced tool for authenticity testing. Journal of Agriculture and Food Chemistry. 59(6): 2229-2234.

Es-Safi, N., Fulcrand, H., Cheynier, V. and Moutounet, M. et. al. (1999) Studies on the AcetaldehydeInduced Condensation of (-)-Epicatechin and Malvidin 3-O-Glucoside in a Model Solution System. Journal of Agriculture and Food Chemistry. 47(5): 2096-2102.

Evershed R.P., Mottram H.R., Dudd S.N., Charters S., Stott A.W., Lawrence G.J., Gibson A.M., Conner A., Blinkhorn P.W., Reeves V. (1997). New criteria for the identification of animal fats preserved in archaeological pottery. Naturwissenschaften. 84: 402-406.

Evershed, R.P. (2008a). Organic Residue Analysis in Archaeology: the Archaeological Biomarker Revolution. Archaeometry. 50(6): 895-924.

Evershed, R.P. (2008b): Experimental Approaches to the Interpretation of Absorbed Organic Residues in Archaeological Ceramics. World Archaeology. 40(1): 26-47.

Evershed, R.P., Dudd, S.N., Copley, M.S., Berstan, R., Stott, A.W., Mottram, H., Buckley, S. A., and Crossman, Z. (2002). Chemistry of Archaeological Animal Fats. Acc. Chem. Res. 35(8): 660-668.

Fenn, J.B. (2003). Electrospray wings for molecular elephants (Nobel lecture). Angewandte Chemie. 42(33): 3871-3894.

Fenn, J.B., Mann, M., Meng, C.K., Wong, S.F. and Whitehouse, C.M. (1989). Electrospray Ionization for Mass Spectrometry of Large Biomolecules. Science 246(4926): 64-71.

Ferrandino, A., Carra A., Rolle L., Schneider A. and Schubert A. (2012). Profiling of Hydroxycinnamoyl Tartrates and Acylated Anthocyanins in the Skin of 34 Vitis vinifera Genotypes. Journal of Agriculture and Food Chemistry. 60(19): 4931-4945.

Finnigan ${ }^{\text {TM }}$ LTQ FT $^{\text {TM }}$ Hardware Manual 1153760 Revision B; 2004.

Finnigan ${ }^{\text {TM }}$ LTQ FT $^{\text {TM }}$ Hardware Manual. 1153760 Revision C - December 2005. Thermo Electron Corporation.

Flamini, R. (2003). Mass spectrometry in grape and wine chemistry. Part I: polyphenols. Mass Spectrometry Reviews. 22(4): 218-250.

Flamini, R. and Traldi, P. (2010). Mass Spectrometry in Grape and Wine Chemistry. John Wiley \& Sons, Inc., Hoboken, New Jersey.

Foley, B. P., Hansso, M.C., Kourkoumelis, D.P. and Theodoulou, T.A. (2012). Aspects of ancient Greek trade re-evaluated with amphora DNA evidence. Journal of Archaeological Science. 39(2): 389-398.

Fong, R., Kepner, R.E. and Dinsmoor Webb, A. (1971). Acetic-Acid-Acylated Anthocyanin Pigments in the Grape Skins of a Number of Varieties of Vitis Vinifera. American Journal of Enology and Viticulture. 22(3): 150-155.

Franks, G. V. and Gan, Y. (2007). Charging Behavior at the Alumina-Water Interface and Implications for Ceramic Processing. Journal of the American Ceramic Society. 90(11): 3373-3388.

Fulcrand, H., Remy, S., Souquet, J.M., Cheynier, V. and Moutounet, M. (1999). Study of Wine Tannin Oligomers by On-Line Liquid Chromatography Electrospray Ionization Mass Spectrometry. Journal of Agricultural and Food Chemistry. 47 (3): 1023-1028.

Galkin, S., Vares, T., Kalsi, M. and Hatakka, A. (1998). Production of organic acids by different whiterot fungi as detected using capillary zone electrophoresis. Biotechnology Techniques. 12(4): 267271.

Gama, M.R., da Costa Silva, R. G., Collins, C. H. and Bottoli, C. B. G. (2012). Hydrophilic interaction chromatography. Trends in Analytical Chemistry. 37: 48-60.

Garnier, N., Rolando, C., Høtje, J.M., and Tokarski, C. (2009). Analysis of archaeological triacylglycerols by high resolution nanoESI, FT-ICR MS and IRMPD MS/MS: Application to 5th century $B C-4$ th century AD oil lamps from Olbia (Ukraine). International Journal of Mass Spectrometry. 284(1-3): 47-56.

Grasset L., and Ambkeá, A (1989). Structure of humin and humic acid from an acid soil as revealed by phase transfer catalyzed hydrolysis. Organic Geochemistry. 29(4): 881-891.

Guasch-Jane, M. (2011). The meaning of wine in Egyptian tombs: the three amphorae from Tutankhamun's burial chamber. Antiquity. 85(2011): 851-858.

Guasch-Jane, M., Andres-lacueva, C., Jauregui, O. and Lamuela-Raventos, R.M. (2006). First evidence of white wine in ancient Egypt from Tutankhamun's tomb. Journal of Archaeological Science. 33(8): 1075-1080.

Guasch-Jane, M., Ibern-Gomez, M., Andres-Lacueva, C., Jauregui, O. and Lamuela-Raventos, R.M. (2004). Liquid Chromatography with Mass Spectrometry in Tandem Mode Applied for the Identification of Wine Markers in Residues from Ancient Egyptian Vessels. Analytical Chemistry. 76(6): 1672-1677.

Guo, Y. and Gaiki, S. (2005). Retention behavior of small polar compounds on polar stationary phases in hydrophilic interaction chromatography. Journal of Chromatography A. 1074 (1-2): 71-80.

Guo, Y., Srinivasan, S., and Gaiki, S. (2007). Investigating the Effect of Chromatographic Conditions on Retention of Organic Acids in Hydrophilic Interaction Chromatography Using a Design of Experiment. Chromatographia. 66(3-4): 223-229.

Habashi, F. (2005) A short history of hydrometallurgy. Hydrometallurgy.79(1-2): 15-22.

Haslam, E. (1980) In vino veritas: oligomeric procyanidins and the ageing of red wines. Phytochemistry. 19(12): 2577-2582.

Hayasaka, Y. and Kennedy, J.A. (2003). Mass spectrometric evidence for the formation of pigmented polymers in red wine. Australian Journal of Grape and Wine Research. 9(3): 210-220.

Hemström, P. and Irgum, K. (2006). Hydrophilic interaction chromatography. Journal of Separation Science. 29(12): 1784-1821.

Heron, C., Evershed, R.P. and Goad, J.L. (1991). Effects of Migration of Soil Lipids on Organic Residues Associated with Buried Potsherds. Journal of Archaeological Science. 18(6): 641-659.

Heron, C. and Evershed, R.P. (1993). The Analysis of Organic Residues and the Study of Pottery Use. Archaeological Method and Theory. 5: 247-284.

Herrmann, K. and Nagel, C.W. (1989). Occurrence and content of hydroxycinnamic and hydroxybenzoic acid compounds in foods. Critical Review of Food Science Nutrition. 1989; 28(4): 315-347.

Howard, G.A. and Martin, A.J. (1950). The separation of the C12-C18 fatty acids by reversed-phase partition chromatography. Journal of Biochemistry. 46(5): 532-538.

Hrydziuszko, O. and Viant, M.R. (2011). Missing values in mass spectrometry based metabolomics: an undervalued step in the data processing pipeline. Metabolomics. 8(S1): 161-174.

Hsu, F.F., Turk, J. (2010). Electrospray Ionization Multiple-Stage Linear Ion-trap Mass Spectrometry for Structural Elucidation of Triacylglycerols: Assignment of Fatty Acyl Groups on the Glycerol Backbone and Location of Double Bonds. Journal of the American Society of Mass Spectrometry. 21(4): 657-669.

Ikegami, T., Tomomatsu, K., Takubo, H., Horie, K., and Tanaka, N. (2008). Separation efficiencies in hydrophilic interaction chromatography. Journal of Chromatography A. 1184(1-2): 474-503.

Iribarne, J.V. and Thomson, B.A. (1976). On the evaporation of small ions from charged droplets. The Journal of Chemical Physics. 64(6): 2287-2294.

Jandera, P. (2011). Stationary and mobile phases in hydrophilic interaction chromatography: a review. Analytica chimica acta. 692(1-2): 1-25.

Jones, D.L. (1998). Organic acids in the rhizosphere - a critical review. Plant and Soil. 205(1): 25-44.

Jordà, G.P. Parreno, C.M., Martin, A.M., and Santos, D.Q. (2011). Stone wine presses and cellars in the Iberian Iron Age territory of Kelin (Utiel-Requena, València) ( 6 th-2nd centuries BC). Paisajes y Patrimonio Cultural del Vino y otras bebidas psicotrópicas. 149-158.

Jurd, L. (1969). Review of polyphenol condensation reactions and their possible occurrence in the aging of wines. American Journal of Enology and Viticulture. 20(3): 191-195.

Kaluzny, M.A., Duncan, L.A., Merritt, M.V. and Epps, D.E.(1985). Rapid separation of lipid classes in high yield and purity using bonded phase columns. Journal of Lipid Research. 26(1): 135-140.

Karatapanis, A.E., Fiamegos, Y. C., and Stalikas, C. D. (2011). A revisit to the retention mechanism of hydrophilic interaction liquid chromatography using model organic compounds. Journal of Chromatography A. 1218(20): 2871-2879.

Karbowiak, T., Gougeon, R. D., Alinc, J. B., Brachais, L., Debeaufort, F., Voilley, A., and Chassagne, D. (2009). Wine Oxidation and the Role of Cork. Critical Reviews in Food Science and Nutrition. 50(1): 20-52.

Kennedy, J.A. and Jones, G.P. (2001). Analysis of Proanthocyanidin Cleavage Products Following AcidCatalysis in the Presence of Excess Phloroglucinol. Journal of Agriculture and Food Chemistry. 49(4): 1740-1746.

Kind, T. and Fiehn, O. (2006). Metabolomic database annotations via query of elemental compositions: Mass accuracy is insufficient even at less than 1 ppm. BMC Bioinformatics. 7:234.

Kliewer, W.M. (1966). Sugars and Organic Acids of Vitis Vinifera. Plant Physiology. 41(6): 923-931.

Koh, A.J. and Betancourt, P.P. (2010). Wine and olive oil from an early Minoan hilltop fort. Mediterranean Archaeology and Archaeometry. 10(2): 15-23.

Koh, A.J., Yasur-Landau, A. and Cline, E.H. (2014). Characterizing a Middle Bronze Palatial Wine Cellar from Tel Kabri, Israel. Plos One. 9(8): 1-15.

Kvande, H. (ed.) 2005. Boehmite vs. Gibbsite precipitation: Light Metals 2005, The Minerals, Metals \& Materials Society.

Lin, C.Y., Wu, H., Tjeerdema, R.S. and Viant, M.R. (2007). Evaluation of metabolite extraction strategies from tissue samples using NMR metabolomics. Metabolomics. 3(1): 55-67.

Lin, J.T., Arcinas, A.J. (2008). Regiospecific Identification of 2-(12-Ricinoleoylricinoleoyl)-1,3-diricinoleoyl-glycerol in Castor (Ricinus communis L.) Oil by ESI-MS4. Journal of Agricultural and Food Chemistry. 56: 3616-3622.

Marshall, A., Hendrickson, C.L. and Jackson, G.S. (1988). Fourier Transform Ion Cyclotron Resonance Mass Spectrometry: A Primer. Mass Spectrometry Reviews. 17(1): 1-35.

Mato, I., Suarez-Luque,S., and Huidobro, J.F. (2005). A review of the analytical methods to determine organic acids in grape juices and wines. Food Research International. 38(10): 1175-1188.

McGovern, P.E. (2003). Ancient Wine: The Search for the Origins of Viniculture. Princeton, New Jersey, Princeton University press.

McGovern, P.E., Luley, B.P., Rovira, N., Mirzoian, A., Callahan, M.P., Smith, K.E., Hall, G.R., Davidson, T. and Henkin, J.M. (2013). Beginning of viniculture in France. Proceedings of the National Academy of Sciences. 110(25): 10147-10152.

McGovern, P.E., Mirzoian, A., Hall, G.R. (2009). Ancient Egyptian herbal wines. Proceedings of the National Academy of Sciences. 106(18): 7361-7366.

McGovern, P.E., Zhang, J., Tang, J., Zhang, Z., Hall, G.R., Moreau, R.A., Nuñez, A., Butrym, E.D., Richards, M.P., Wang, C., Cheng, G., Zhao, Z., Wang, C. (2004). Fermented beverages of pre- and proto-historic China. Proceedings of the National Academy of Sciences. 101(51): 7593-17598.

Melnikov, S. M., Holtzel, A., Seidel-Morgenstern, A. and Tallarek, U. (2013). How ternary mobile phases allow tuning of analyte retention in hydrophilic interaction liquid chromatography. Analytical Chemistry. 85(18): 8850-8856.

Michel, R.H., McGovern, P.E., and Badler, V.R. (1993). The First Wine \& Beer Chemical Detection of Ancient Fermented Beverages. Analytical Chemistry. 65(8): 408A-413A.

Miller, P.E. and Denton, M.B. (1986). The Quadrupole Mass Filter: Basic Operating Concepts. Journal of Chemical Education. 63(7): 617-622.

Mirabaud, S., Rolando, C., and Regert, M. (2007). Molecular Criteria for Discriminating Adipose Fat and Milk from Different Species by NanoESI MS and MS/MS of Their Triacylglycerols: Application to Archaeological Remains. Analytical Chemistry. 79(16): 6182-6192.

Mottram, H.R., Woodbury, S.E., and Evershed, R.P. (1997). Identification of triacylglycerol positional isomers present in vegetable oils by high performance liquid chromatography atmospheric pressure chemical ionization mass spectrometry, Rapid Communications in Mass Spectrometry. 11(12): 12401252.

Mukherjee, A.J., Gibson, A.M., and Evershed, R.P. (2007). Trends in pig product processing at British Neolithic Grooved Ware sites traced through organic residues in potsherds. Journal of Archaeological Science 35: 2059-2073.

Myles, S. Boyko, A.R., Owens, C.L., Brown, P.J., Grassi, F., Aradhya, M.K., Prins, B., Reynolds, A., Chia, J., Ware, D., Bustamante, C.D., and Buckler, E.S. (2011). Genetic structure and domestication history of the grape. Proceedings of the National Academy of Sciences. 108(9): 3530-3535.

Nardini, M. and Ghiselli, A. (2004). Determination of free and bound phenolic acids in beer. Food Chemistry. 84(1): 137-143.

Nguyen, H. P. and Schug, K.A. (2008). The advantages of ESI-MS detection in conjunction with HILIC mode separations: Fundamentals and applications. Journal of Separation Science. 31(9): 1465-1480.

Ong, B.Y. and Nagel, C.W. (1978). Hydroxycinnamic Acid-Tartaric Acid Ester Content in Mature Grapes and During the Maturation of White Riesling Grapes. America Journal of Enology and Viticulture. 29(4): 277-281.

Parsons, H.M. Ludwig, C., Günther, U.L. Viant, M.R. (2007). Improved classification accuracy in 1- and 2-dimensional NMR metabolomics data using the variance stabilising generalised logarithm transformation. BMC Bioinformatics. 8(234): 1-16.

Paul, W. (1990). Electromagnetic traps for charged and neutral particles. Review of Modern Physics. 62(3): 531-540.

Payne, T.G. (2011). Profiling the Metabolome using Fourier Transform Ion Cyclotron Resonance Mass Spectrometry, Optimised Signal Processing, Noise Filtering and Constraints Methods. PhD thesis, University of Birmingham, School of Electronic, Electrical and Computer Engineering.

Payne, T.G., Southam, A. Gg. Arvantis, T.N., and Viant, M.R. (2009). A Signal Filtering Method for Improved Quantification and Noise Discrimination in Fourier Transform Ion Cyclotron Resonance Mass Spectrometry-Based Metabolomics Data. Journal of the American Society for Mass Spectrometry. 20(6): 1087-1095.

Pecci, A., Giorgi, G., Salvini, L. and Ontiveros, M. (2013). Identifying wine markers in ceramics and plasters using gas chromatography-mass spectrometry. Experimental and archaeological materials. Journal of Archaeological Science. 40(1): 109-115.

Pérez-Jiménez, J. and Torres, J.L. (2011). Analysis of Nonextractable Phenolic Compounds in Foods: The Current State of the Art. Journal of Agricultural and Food Chemistry. 59(24): 12713-12724.

Poncet-Legrand, C., Cabane, B., Bautista-ortin, A.B., Carrillo, S., Fulcrand, H., Perez, J., and Vernhet, A. (2010). Tannin Oxidation: Intra- versus Intermolecular Reactions. Biomacromolecules. 11(9): 2376-2386.

Preinerstorfer, B., Schiesel, S., Lammerhofer, M., and Lindner, W. (2010). Metabolic profiling of intracellular metabolites in fermentation broths from beta-lactam antibiotics production by liquid chromatography-tandem mass spectrometry methods. Journal of Chromatography A. 1217(3): 312328.

Rawlins, A.J., Bull, I.D., Poirier, N., Ineson, P. and Evershed, R.P. (2005). The biochemical transformation of oak (Quercus robur) leaf litter consumed by the pill millipede (Glomeris marginata). Soil Biology \& Biochemistry. 38(5): 1063-1076.

Redfern, R.C., Millard, A.R., and Hamlin, C. (2012). A regional investigation of subadult dietary patterns and health in late Iron Age and Roman Dorset, England. Journal of Archaeological Science. 39: 1249-1259.

Regert, M., Bland, H. A., Dudd, S. N., Bergen, P. F. V. and Evershed, R. P. (1998). Free and bound fatty acid oxidation products in archaeological ceramic vessels. Proceedings of the Royal Society B:
Biological Sciences. 265(1409): 2027-2032.

Regert, M. (2011). Analytical Strategies for discriminating archeological fatty substances from animal origin. Mass Spectrometry Reviews. 30: 177-220.

Remy, S., Fulcrand, H., Labarbe, B., Cheynier, V. and Moutounet, M. (2000). First confirmation in red wine of products resulting from direct anthocyanin-tannin reactions. Journal of Science and Food Agriculture. 80(6): 745-751.

Renfrew, J. (1985) Food and Cooking in Roman Britain. English Heritage, Birmingham

Ribereau-Gayon, P. (1972). Plant Phenolics. New York, New York: Hafner Publishing Company.

Ribereau-Gayon, P., Dubourdieu, D., Don`eche, B., Lonvaud, A. (2006). Handbook of Enology, Volume 1, The Microbiology of Wine and Vinifications, 2nd Ed.

Rice, P. (1987). Pottery Analysis: A Sourcebook. Chicago, Illinois: The University of Chicago Press.

Romanus, K., Baeten, J., Poblome, J., Accardo, S., Degryse, P., Jacobs, P., De Vos, D. and Waelkens, M. (2009). Wine and olive oil permeation in pitched and non-pitched ceramics: relation with results from archaeological amphorae from Sagalassos, Turkey. Journal of Archaeological Science. 36 (3): 900-909.

Rossell, J. B. and Pritchard, J. L. R. (ed) (1991). Analysis of Oilseeds, Fats and Fatty Foods. Elsevier.

Rózycki, H. and Strzelczyk, E. (1986). Organic acids production by Streptomyces spp. isolated from soil, rhizosphere and mycorrhizosphere of pine (Pinus sylvestris L.). Plant and Soil. 96(3): 337-345.

Saucier, C. (2010). How do wine polyphenols evolve during wine ageing? Cerevisia 35(1): 11-15.

Schultz, G.A., Corso, T.N., Prosser, S.J. and Zhang, S. (2000) A Fully Integrated Monolithic Microchip Electrospray Device for Mass Spectrometry. Analytical Chemistry. 72(17): 4058-4063.

Scigelova, M., Hornshaw, M., Giannakopulos, A. and Makarov, A. (2011) Fourier Transform Mass Spectrometry. Molecular Cellular Proteomics. 10(7): 1-19.

Sheehan-Finn,K. (2012). Vindolanda's Amphora Stamps: 2007-2012. Report issued for the Vindolanda Trust.

Singleton, V.L. and Trousdale, E.K. (1992) Anthocyanin-Tannin Interactions Explaining Differences in Polymeric Phenols Between White and Red Wines. American Journal of Enology and Viticulture. 43(1): 63-70.

Somers, T.C. (1971). The polymeric nature of wine pigments. Phytochemistry. 10(9): 2175-2186.

Southam, A.D., Payne, T., Cooper, H.J., Arvantis, T.N., and Viant, M.R. (2007). Dynamic Range and Mass Accuracy of Wide-Scan Direct Infusion Nanoelectrospray Fourier Transform Ion Cyclotron Resonance Mass Spectrometry-Based Metabolomics Increased by the Spectral Stitching Method. Analytical Chemistry. 79(13): 4595-4602.

Stern, B., Heron, C., Tellefsen, T. and Serpico, M. (2008). New investigations into the Uluburun resin cargo. Journal of Archaeological Science. 35(8): 2188-2203.

Symonds, M.F.A. and Mason, D.J.P. (ed.) (2009). Frontiers of Knowledge: A Research Framework for Hadrian's Wall, Part of the Frontiers of the Roman Empire World Heritage Site, vol I. Durham County Council, Durham.

Terayama, Y., Kikuchi, M., Kobayashi, M., Hino, M. and Takahara, A. (2009). Influence of salt concentration on swelling states of poly(sulfobetaine) brush at aqueous solution interface. Journal of Physics: Conference Series. 184:012011

This, P., Lacombe, T., and Thomas, M.R. (2006). Historical origins and genetic diversity of wine grapes. TRENDS in Genetics. 22(9): 511-519.

Thornton, M. D., Morgan, E. D., and Celoria, F. (1970). The composition of bog butter. Science and Archaeology. 1(2/3): 20-25.

Thoukis, G., Ueda, M. and Wright, D. (1965). The Formation of Succinic Acid during Alcoholic Fermentation. American Journal of Enology Viticulture. 16(1): 1-8.

Tian, R., Pan, Q., Zhan, J., Li, J., Wan, S., Zhang, Q., and Huang, W. (2009). Comparison of Phenolic Acids and Flavan-3-ols During Wine Fermentation of Grapes with Different Harvest Times.
Molecules. 14(2): 827-838.

Timberlake, C.F. and Bridle, P. (1976). Interactions between anthocyanins, phenolic compounds and, and acetaldehyde and their significance in red wines. American Journal of Enology and Viticulture. 27(3): 97-105.

TSQ Series Hardware Manual 70111-97163 Revision C March 2009. Thermo Scientific.

Twede, Diana. (2002). The Packaging Technology and Science of Ancient Transport Amphoras. Journal of Packaging Technology and Science. 15(4): 181-195.

Uechi, G.T. and Dunbar, R.C. (1992). Space Charge Effects on Relative Peak Heights in Fourier Transform-lon Cyclotron Resonance Spectra. Journal of the American Society of Mass Spectrometry. 3(7): 734-741.

Uroz, S., Calvaruso, C., Turpault, M. P. and Frey-Klett, P. (2009). Mineral weathering by bacteria ecology, actors and mechanisms. Trends in microbiology. 17(8): 378-387.
van Dommelen, P. and Bellard, C.G. (2008). Rural Landscapes of the Punic World. Monographs in Mediterranean Archaeology. 11.
van Dommelen, P. and Sharpe, L. (2004). Surveying Punic rural settlement: the Terralba Rural Settlement Project, Sardinia. Antiquity. (299) March 2004.
van Dommelen, P. and Trapichler, M. (2011). "Fabrics of Western Sardinia". In FACEM (version 06/06/2011)
van Dommelen, P., Bellard, C.G. and Tronchetti, C. (2007). The Punic farmstead at Truncu'e Molas (Sardinia, Italy): excavations 2007. Antiquity. 82(315).

Vanholme, R. Demedts, B., Morreel, K., Ralph, J. and Boerjan, W. (2010). Lignin Biosynthesis and Structure. Plant physiology. 153(3): 895-905.

Verma, B., Hucl, P. and Chibbar, R. N. (2009). Phenolic acid composition and antioxidant capacity of acid and alkali hydrolysed wheat bran fractions. Food Chemistry. 116(4): 947-954.

Villas-Boas, S., Mas, S., Akesson, M., Smedsgaard, J. and Nielsen, J. (2004). Mass spectrometry in metabolome analysis. Mass Spectrometry Reviews. 24 (5): 613-646.

Waterhouse, A. and Kennedy, J.A. (eds.) 2004. Red Wine Color: Exploring the Mysteries. Washington D.C.: American Chemical Society.

Weber, R.J.M. and Viant, M.R. (2010). MI-Pack: Increased confidence of metabolite identification in mass spectra by integrating accurate masses and metabolic pathways. Chemometrics and Intelligent Laboratory Systems. 104(1): 75-82.

Wilm, M. and Mann, M. (1996). Analytical Properties of the Nanoelectrospray Ion Source. Analytical Chemistry. 68(1): 1-8.

Wollmann, N. and Hofmann, T. (2013). Compositional and Sensory Characterization of Red Wine Polymers. Journal of Agricultural and Food Chemistry. 61(9): 2045-2061.

Wu, H., Southam, A.D., Hines, A. and Viant, M.R. (2008). High-throughput tissue extraction protocol for NMR and MS-based metabolomics. Analytical Biochemistry. 372(2): 204-212.

Yost, R.A. and Enke, C.G. (1978). Selected Ion Fragmentation with a Tandem Quadrupole Mass Spectrometer. Journal of the American Chemical Society. 100(7): 2274-2275.

Zsuga, M. and Kiss, A. (1987). Alkaline degradation of parent chromonoid compounds (chromone, flavone, isoflavone). Acta Chimicia Hungarica. 124: 485-489.

## 9. Appendix

## Chapter 3

Table 1 A table of $\mathrm{m} / \mathrm{z}$ identified from the analysis of the red wine aliquots over four time points: zero, one, three, six months, which includes putative identifications. Univariate statistics were applied to this data; using the four time points, a fold change was determined using zero month as the reference time point, fold change 1 represents one month, fold change 2 represents three months, and fold change 3 represents the six month time point. The table is sorted on the smallest fold change 3 in conjunction with the smallest adjusted $p$-value, that is, the most significant peaks with the greatest change in intensities from zero month to six months.

| M/Z | adjusted p value | fold change 1 | fold change 2 | fold change 3 | Empirical formula (parent) | $\begin{aligned} & \text { Ion } \\ & \text { form } \end{aligned}$ | $\begin{aligned} & \text { Theoretical } \\ & \text { mass } \\ & \text { (neutral) (Da) } \end{aligned}$ | Theoretical m/z (Da) | $\begin{aligned} & \text { Mass } \\ & \text { error } \\ & \text { (ppm) } \end{aligned}$ | KEGG_COMPOUND |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 149.00847 | 8.30E-14 | 0.1315784 | 0.0150446 | 0.00379 |  |  |  |  |  | 0 |
| 193.03551 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C10H8N2 | [M+(37Cl)]- | 156.068748 | 193.0351996 | 1.61 | ['3-Indoleacetonitrile'] |
| 193.03551 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C4H6O5 | [M+Hac-H]- | 134.021525 | 193.0353786 | 0.68 | ['(R)-Malate', '(S)-Malate', '3-Dehydro-Lthreonate', 'Malate'] |
| 193.03551 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C6H1007 | [M-H]- | 194.042655 | 193.0353786 | 0.68 | ['2-Dehydro-D-galactonate', '2-Keto-D-gluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-Dgluconate', 'D-Fructuronate', 'D-Galacturonate', 'D-Glucuronate', 'D-Glucuronic acid', 'DMannuronate', 'D-Tagaturonate', 'Galacturonic acid', 'L-Guluronic acid', 'L-Iduronic acid', 'beta-D-Glucopyranuronic acid'] |
| 193.03525 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C10H8N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 156.068748 | 193.0351996 | 0.26 | ['3-Indoleacetonitrile'] |
| 193.03525 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C4H6O5 | [M+Hac-H]- | 134.021525 | 193.0353786 | -0.67 | ['(R)-Malate', '(S)-Malate', '3-Dehydro-Lthreonate', 'Malate'] |
| 193.03525 | $2.73 \mathrm{E}-09$ | 0.1090107 | 0.0241104 | 0.006121 | C6H1007 | [M-H]- | 194.042655 | 193.0353786 | -0.67 | ['2-Dehydro-D-galactonate', '2-Keto-D-gluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-Dgluconate', 'D-Fructuronate', 'D-Galacturonate', <br> 'D-Glucuronate', 'D-Glucuronic acid', 'D- <br> Mannuronate', 'D-Tagaturonate', 'Galacturonic acid', 'L-Guluronic acid', 'L-Iduronic acid', 'beta- <br> D-Glucopyranuronic acid'] |
| 349.1363 | 0 | 0.0470638 | 0.0076826 | 0.006479 | C10H18N4O6 | [M+Hac-H]- | 290.122636 | 349.1364896 | -0.54 | ['N-(L-Arginino)succinate'] |


| 349.1363 | 0 | 0.0470638 | 0.0076826 | 0.006479 | C19H24N2OS | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 328.160935 | 349.1356036 | 1.99 | ['Methotrimeprazine'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 349.1363 | 0 | 0.0470638 | 0.0076826 | 0.006479 | C24H24 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 312.1878 | 349.1364066 | -0.31 | ['1,3,5-Triphenylcyclohexane', '2,4,6-Triphenyl- 1-hexene'] |
| 194.03892 | 0 | 0.0898494 | 0.0234006 | 0.007464 |  |  |  |  |  | 0 |
| 193.03459 | 0.12046482 | 0.1090774 | 0.0232775 | 0.009106 |  |  |  |  |  | 0 |
| 321.09384 | 5.00E-15 | 0.0247784 | 0.0159402 | 0.010464 |  |  |  |  |  | 0 |
| 316.0439 | 5.16E-13 | 0.1790347 | 0.0196174 | 0.013673 | C8H16NO10P | [M-H]- | 317.051187 | 316.0439106 | -0.03 | ['N-Glycolyl-D-mannosamine 6-phosphate'] |
| 350.12036 | 0 | 0.0711192 | 0.0197312 | 0.014881 | C18H24O5S | [M-2H]- | 352.134447 | 350.1198942 | 1.33 | ['Estradiol-17beta 3-sulfate'] |
| 387.0923 | $4.40 \mathrm{E}-14$ | 0.0587978 | 0.0229128 | 0.0192 |  |  |  |  |  | 0 |
| 389.09357 | $3.00 \mathrm{E}-15$ | 0.0530667 | 0.0266131 | 0.020181 | C17H24N2O4S | [M+K-2H]- | 352.14568 | 389.0942866 | -1.84 | ['Mercaptoacetyl-Phe-Leu'] |
| 328.96802 | $9.00 \mathrm{E}-15$ | 0.0887608 | 0.02963 | 0.021356 |  |  |  |  |  | 0 |
| 147.02928 | $3.75 \mathrm{E}-10$ | 0.0397688 | 0.0166963 | 0.023502 |  |  |  |  |  | 0 |
| 427.04946 | 0 | 0.0835066 | 0.0351284 | 0.02537 |  |  |  |  |  | 0 |
| 327.05675 | 6.27E-12 | 0.086622 | 0.0351769 | 0.026424 | C14H14N2O5 | [ $\mathrm{M}+(37 \mathrm{C})$ ]- | 290.090273 | 327.0567246 | 0.08 | ['N2-Malonyl-D-tryptophan'] |
| 278.08791 | $8.81 \mathrm{E}-13$ | 0.1522343 | 0.031381 | 0.029503 | C14H15N3O | [ $\mathrm{M}+(37 \mathrm{C})$ ]- | 241.121512 | 278.0879636 | -0.19 | $\begin{aligned} & \text { ['4-(Dimethylamino)phenylazoxybenzene', 'CX- } \\ & 516 \text { '] } \end{aligned}$ |
| 278.08791 | $8.81 \mathrm{E}-13$ | 0.1522343 | 0.031381 | 0.029503 | C8H13NO6 | [ $\mathrm{M}+\mathrm{Hac}$ - H ] | 219.074289 | 278.0881426 | -0.84 | ['N-Acetyl-D-mannosaminolactone', 'O-Succinyl-L-homoserine'] |
| 322.07786 | 9.16E-13 | 0.0674221 | 0.0402546 | 0.042062 |  |  |  |  |  | 0 |
| 264.07227 | 3.59E-13 | 0.2367747 | 0.0521316 | 0.043858 |  |  |  |  |  | 0 |
| 145.01417 | $3.00 \mathrm{E}-15$ | 0.0796128 | 0.0497918 | 0.044018 | C5H605 | [M-H]- | 146.021525 | 145.0142486 | -0.54 | ['2-Oxoglutarate', '5-Hydroxy-2,4dioxopentanoate', 'Dehydro-D-arabinono-1,4lactone', 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] |
| 293.06828 | 8.11E-07 | 0.1143822 | 0.0732897 | 0.055831 |  |  |  |  |  | 0 |
| 149.00818 | 0.45950564 | 0.8392681 | 0.070108 | 0.056388 | C3H4N4O2 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 128.033426 | 149.0080946 | 0.57 | ['Ammelide'] |
| 535.15162 | $3.02 \mathrm{E}-13$ | 0.0690184 | 0.0630969 | 0.059946 |  |  |  |  |  |  |
| 465.10413 | 7.46E-13 | 0.2926402 | 0.1042087 | 0.063117 | C19H18O10 | [ $\mathrm{M}+\mathrm{Hac}$ - H ]- | 406.09 | 465.1038536 | 0.59 | ['Lancerin'] |
| 465.10413 | 7.46E-13 | 0.2926402 | 0.1042087 | 0.063117 | C21H22O12 | [M-H]- | 466.11113 | 465.1038536 | 0.59 | ['Plantagoside', 'Quercetin 7-O-glucoside'] |
| 149.00903 | 0 | 0.1701555 | 0.0809639 | 0.070141 | C2H2O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 89.99531 | 149.0091636 | -0.9 | ['Oxalate'] |
| 149.00903 | 0 | 0.1701555 | 0.0809639 | 0.070141 | C4H606 | [M-H]- | 150.01644 | 149.0091636 | -0.9 | $['(R, R)$-Tartaric acid', '(S,S)-Tartaric acid', 'meso- Tartaric acid'] |
| 321.02283 | 0 | 0.1384302 | 0.0809052 | 0.074236 |  |  |  |  |  | 0 |


| 329.06657 | 2.20E-14 | 0.1268503 | 0.1002789 | 0.087236 | C15H10O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 270.052825 | 329.0666786 | -0.33 | ['2-Hydroxydaidzein', "3',4',7- <br> Trihydroxyisoflavone", '3,6,4- <br> Trihydroxyflavone', "4',6,7- <br> Trihydroxyisoflavone", '5-Deoxykaempferol', 'Aloe-emodin', 'Apigenin', 'Baicalein', 'Emodin', 'Galangin', 'Genistein', 'Islandicin', 'Lucidin', 'Morindone', 'Norobtusifolin', 'Norwogonin', 'Purpurin 1-methyl ether', 'Sulphuretin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 329.06657 | 2.20E-14 | 0.1268503 | 0.1002789 | 0.087236 | C17H14O7 | [M-H]- | 330.073955 | 329.0666786 | -0.33 | ['(+)-Bisdechlorogeodin', '(-)- <br> Bisdechlorogeodin', "3',4',5-Trihydroxy-3,7dimethoxyflavone", 'Aflatoxin G2', 'Aurantioobtusin', 'Cirsiliol', 'Hildecarpin', 'Tricin'] |
| 329.06657 | 2.20E-14 | 0.1268503 | 0.1002789 | 0.087236 | C17H15O7 | [M-2H]- | 331.08178 | 329.0672272 | -2 | ['Malvidin'] |
| 425.03382 | 0 | 0.1277376 | 0.095168 | 0.0935 |  |  |  |  |  | 0 |
| 173.00927 | $2.05 \mathrm{E}-12$ | 0.1436403 | 0.1178257 | 0.096651 | C4H2O4 | [M+Hac-H]- | 113.99531 | 173.0091636 | 0.61 | ['Acetylenedicarboxylate'] |
| 173.00927 | 2.05E-12 | 0.1436403 | 0.1178257 | 0.096651 | C6H6O6 | [M-H]- | 174.01644 | 173.0091636 | 0.61 | ['Dehydroascorbate', 'cis-Aconitate', 'transAconitate'] |
| 371.08311 | 1.10E-12 | 0.3653067 | 0.138132 | 0.107381 |  |  |  |  |  | 0 |
| 265.07154 | 2.20E-14 | 0.2903006 | 0.1467446 | 0.107418 | C11H1004 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 206.05791 | 265.0717636 | -0.84 | ['2-Hydroxy-3-methylbenzalpyruvate', '2- <br> Hydroxy-8-methylchromene-2-carboxylate', 'Lathodoratin', 'Scoparone', 'cis-1,2-Dihydroxy-1,2-dihydro-8-carboxynaphthalene'] |
| 449.10911 | $2.01 \mathrm{E}-12$ | 0.2993693 | 0.1664052 | 0.110737 | C21H22O11 | [M-H]- | 450.116215 | 449.1089386 | 0.38 | ["2',3,4,4',6-Peptahydroxychalcone 4-0glucoside", 'Astilbin', 'Neoastilbin', 'Neocarthamin'] |
| 539.14094 | $1.92 \mathrm{E}-12$ | 0.2253417 | 0.1062973 | 0.114507 |  |  |  |  |  | 0 |
| 146.0458 | 2.80E-14 | 0.3609297 | 0.1505606 | 0.115217 | C3H5NO2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 87.032029 | 146.0458826 | -0.57 | ['2-Oxazolidinone', 'Dehydroalanine'] |
| 146.0458 | 2.80E-14 | 0.3609297 | 0.1505606 | 0.115217 | C5H9NO4 | [M-H]- | 147.053159 | 146.0458826 | -0.57 | ['2-Oxo-4-hydroxy-5-aminovalerate', 'DGlutamate', 'DL-Glutamate', 'Isoglutamate', 'L-4-Hydroxyglutamate semialdehyde', 'LGlutamate', 'L-threo-3-Methylaspartate', 'N-(Carboxymethyl)-D-alanine', 'N-Methyl-Daspartic acid', 'O-Acetyl-L-serine'] |
| 671.1828 | 7.32E-13 | 0.1521108 | 0.1206806 | 0.119309 |  |  |  |  |  | 0 |
| 271.0246 | 0 | 0.3164202 | 0.1591549 | 0.120488 |  |  |  |  |  | 0 |
| 661.18321 | 5.40E-12 | 0.212673 | 0.1614287 | 0.121094 |  |  |  |  |  | 0 |
| 267.03556 | $1.49 \mathrm{E}-09$ | 0.3541283 | 0.139664 | 0.126354 |  |  |  |  |  | 0 |
| 375.11443 | 3.81E-13 | 0.2938413 | 0.1424299 | 0.133768 | C16H22N2O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 338.147788 | 375.1142396 | 0.51 | ['Aziridyl benzoquinone'] |


| 201.03825 | 1.16E-08 | 0.279144 | 0.194838 | 0.140132 | C6H12O6 | [M+Na-2H]- | 180.06339 | 201.0380586 | 0.95 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'DAldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'D- <br> Fuconate', 'D-Galactose', 'D-Glucose', 'DGulose', 'D-Hamamelose', 'D-Hexose', 'D-Idose', <br> 'D-Mannose', 'D-Psicose', 'D-Sorbose', 'D- <br> Tagatose', 'D-Talose', 'Fructose(pyranose)', <br> 'Ketose', 'L--Fructose', 'L-Fuconate', 'L- <br> Galactose', 'L-Gulose', 'L-Rhamnonate', 'LSorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha-D-Glucose', 'alpha-D-Mannose', 'alpha-LSorbopyranose', 'beta-D-Fructose', 'beta-DGalactose', 'beta-D-Glucose', 'beta-DHamamelopyranose', 'beta-D-Mannose', 'muco-Inositol', 'myo-Inositol', 'scyllo-Inositol'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 189.05773 | 2.73E-11 | 0.2725396 | 0.1981769 | 0.14099 |  |  |  |  |  | 0 |
| 477.06757 | 1.81E-11 | 0.2689602 | 0.1891489 | 0.142129 |  |  |  |  |  | 0 |
| 214.04888 | 2.20E-14 | 0.3898431 | 0.2413598 | 0.14343 | C10H11NO3 | [M+Na-2H]- | 193.073894 | 214.0485626 | 1.48 | ['(3-Arylcarbonyl)-alanine', '3-Carbamoyl-2phenylpropionaldehyde', '4-Hydroxy-5-phenyltetrahydro-1,3-oxazin-2-one', '5,6-Dihydroxy-3-methyl-2-oxo-1,2,5,6tetrahydroquinoline', 'Cichorine', 'Gentioflavine', 'Phenylacetylglycine'] |
| 214.04888 | 2.20E-14 | 0.3898431 | 0.2413598 | 0.14343 | C3H10NO4P | [M+Hac-H]- | 155.034747 | 214.0486006 | 1.31 | ['D-1-Aminopropan-2-ol O-phosphate', 'L-1-Aminopropan-2-ol O-phosphate', 'NMethylethanolamine phosphate'] |
| 214.04888 | 2.20E-14 | 0.3898431 | 0.2413598 | 0.14343 | C5H14NO6P | [M-H]- | 215.055877 | 214.0486006 | 1.31 | ['sn-glycero-3-Phosphoethanolamine'] |
| 214.04888 | $2.20 \mathrm{E}-14$ | 0.3898431 | 0.2413598 | 0.14343 | C6H13NO5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 179.079374 | 214.0487756 | 0.49 | ['1-Amino-1-deoxy-scyllo-inositol', 'DGalactosamine', 'D-Glucosamine', 'DMannosamine', 'Kanosamine', 'Nojirimycin', 'beta-D-Glucosamine', 'neo-Inosamine-2'] |
| 214.04888 | 2.20E-14 | 0.3898431 | 0.2413598 | 0.14343 | C7H15NO2S | [ $\mathrm{M}+(37 \mathrm{Cl})]-$ | 177.082351 | 214.0488026 | 0.36 | ['Dihomomethionine'] |
| 214.04888 | 2.20E-14 | 0.3898431 | 0.2413598 | 0.14343 | C7H15NO4 | [M+K-2H]- | 177.100109 | 214.0487156 | 0.77 | ['Anthopleurine'] |
| 478.07088 | 4.18E-12 | 0.2753949 | 0.1944882 | 0.146733 |  |  |  |  |  | 0 |
| 285.04023 | 8.30E-14 | 0.3493056 | 0.203257 | 0.146914 | C15H1006 | [M-H]- | 286.04774 | 285.0404636 | -0.82 | ['2-Hydroxygenistein', '6- <br> Demethoxycapillarisin', 'Aureusidin', 'Citreorosein', 'Datiscetin', 'Fisetin', 'Isoscutellarein', 'Kaempferol', 'Luteolin', 'Maritimetin', 'Orobol', 'Scutellarein'] |


| 325.09278 | 5.26E-13 | 0.385688 | 0.2130926 | 0.147888 | C15H18O8 | [M-H]- | 326.10017 | 325.0928936 | -0.35 | ['4-O-beta-D-Glucosyl-4-hydroxycinnamate', '4- <br> O-beta-D-Glucosyl-cis-p-coumarate', <br> 'Bilobalide', 'cis-beta-D-Glucosyl-2- <br> hydroxycinnamate', 'p-Coumaroyl-D-glucose', <br> 'trans-beta-D-Glucosyl-2-hydroxycinnamate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 325.09278 | 5.26E-13 | 0.385688 | 0.2130926 | 0.147888 | C19H16N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 288.126263 | 325.0927146 | 0.2 | ['Difenpiramide'] |
| 205.05088 | 3.56E-12 | 0.3377731 | 0.2048182 | 0.154257 | C11H1004 | [M-H]- | 206.05791 | 205.0506336 | 1.2 | ['2-Hydroxy-3-methylbenzalpyruvate', '2-Hydroxy-8-methylchromene-2-carboxylate', 'Lathodoratin', 'Scoparone', 'cis-1,2-Dihydroxy-1,2-dihydro-8-carboxynaphthalene'] |
| 205.05088 | 3.56E-12 | 0.3377731 | 0.2048182 | 0.154257 | C9H6O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 146.03678 | 205.0506336 | 1.2 | ['Coumarin'] |
| 150.01241 | 2.20E-14 | 0.3531449 | 0.1830736 | 0.154631 |  |  |  |  |  | 0 |
| 507.11481 | 2.30E-11 | 0.3747309 | 0.2424242 | 0.162779 | C21H20011 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 448.100565 | 507.1144186 | 0.77 | ['Astragalin', 'Aureusidin 6-O-glucoside', 'Carthamone', 'Fisetin 8-C-glucoside', 'Isoorientin', 'Kaempferol 3-O-beta-Dgalactoside', 'Luteolin 7-O-beta-D-glucoside', 'Orientin', 'Plantaginin', 'Quercitrin'] |
| 214.99649 | 1.61E-06 | 0.4742 | 0.2587974 | 0.164592 | C2H5O6P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 155.982378 | 214.9962316 | 1.2 | ['2-Phosphoglycolate'] |
| 214.99649 | $1.61 \mathrm{E}-06$ | 0.4742 | 0.2587974 | 0.164592 | C4H908P | [M-H]- | 216.003508 | 214.9962316 | 1.2 | ['3-Phospho-D-erythronate', '4-Phospho-Derythronate'] |
| 214.99649 | 1.61E-06 | 0.4742 | 0.2587974 | 0.164592 | C6H1006 | [M+K-2H]- | 178.04774 | 214.9963466 | 0.67 | ['2,4,6/3,5-Pentahydroxycyclohexanone', '2-Dehydro-3-deoxy-D-galactonate', '2-Dehydro-3-deoxy-D-gluconate', '2-Dehydro-D-glucose', '2-Deoxy-5-keto-D-gluconic acid', '3-Dehydro-2-deoxy-D-gluconate', '3-Keto-beta-D-galactose', '5-Dehydro-2-deoxy-D-gluconate', '5-Dehydro- <br> D-fructose', '5-Deoxy glucuronic acid', 'D-Galactono-1,4-lactone', 'D-Galactono-1,5lactone', 'D-Glucono-1,4-lactone', 'D-Glucono-1,5-lactone', 'D-galacto-Hexodialdose', 'Gulono-1,4-lactone', 'Hexose-1,5-lactone', 'L-Galactono-1,4-lactone', 'L-Gulono-1,4-lactone', 'myo-Inosose-5'] |
| 151.00357 | 2.32E-11 | 0.3585314 | 0.2279074 | 0.172366 |  |  |  |  |  | 0 |
| 173.10452 | 1.82E-10 | 0.3732101 | 0.2357923 | 0.178295 | C6H14N4O2 | [M-H]- | 174.111676 | 173.1043996 | 0.7 | ['Amino acid(Arg-)', 'D-Arginine', 'L-Arginine'] |
| 148.03317 | 0 | 0.2078304 | 0.147258 | 0.18146 |  |  |  |  |  | 0 |
| 387.07804 | $3.81 \mathrm{E}-13$ | 0.260909 | 0.1899433 | 0.192606 | C16H20N2O5S | [ $\mathrm{M}+\mathrm{Cl}]$ - | 352.109295 | 387.0786966 | -1.7 | ['Benzylpenicilloic acid'] |
| 273.0016 | 4.01E-06 | 0.6481906 | 0.2919672 | 0.195089 | C4H7O8P | [M+Hac-H]- | 213.987858 | 273.0017116 | -0.41 | ['2-Oxo-3-hydroxy-4-phosphobutanoate'] |


| 273.0016 | 4.01E-06 | 0.6481906 | 0.2919672 | 0.195089 | C6H11010P | [M-H]- | 274.008988 | 273.0017116 | -0.41 | ['1-Phospho-alpha-D-galacturonate', '3-Dehydro-L-gulonate 6-phosphate', '6-Phospho-2-dehydro-D-gluconate', 'D-Glucuronate 1phosphate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 515.22071 | $1.92 \mathrm{E}-12$ | 0.3816218 | 0.2696937 | 0.209189 | C34H30N4 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 494.247046 | 515.2217146 | -1.95 | ['UCL 1684'] |
| 147.02979 | $3.37 \mathrm{E}-13$ | 0.2477154 | 0.1829392 | 0.212782 | C3H4O3 | [M+Hac-H]- | 88.016045 | 147.0298986 | -0.74 | ['3-Hydroxypropenoate', '3-Oxopropanoate', 'Pyruvate'] |
| 147.02979 | 3.37E-13 | 0.2477154 | 0.1829392 | 0.212782 | C5H8O5 | [M-H]- | 148.037175 | 147.0298986 | -0.74 | ['(R)-2-Hydroxyglutarate', '(R)-2-Methylmalate', '(S)-2-Hydroxyglutarate', '(S)-2-Methylmalate', '2-Dehydro-3-deoxy-D-xylonate', '2-Dehydro-3-deoxy-L-arabinonate', '2-Hydroxyglutarate', 'Citramalate', 'D-Arabinono-1,4-lactone', 'D-Xylono-1,4-lactone', 'D-Xylonolactone', 'D-erythro-3-Methylmalate', 'D-threo-3- <br> Methylmalate', 'L-Arabinono-1,4-lactone', 'L-Arabinono-1,5-lactone', 'L-Xylono-1,4-lactone', 'L-threo-3-Methylmalate'] |
| 311.06188 | 1.06E-10 | 0.6773053 | 0.2473923 | 0.216064 |  |  |  |  |  | 0 |
| 329.10878 | $1.92 \mathrm{E}-10$ | 0.2896865 | 0.1852571 | 0.216801 |  |  |  |  |  | 0 |
| 312.06524 | 9.86E-11 | 0.7095464 | 0.2538063 | 0.216911 |  |  |  |  |  | 0 |
| 384.96649 | $2.80 \mathrm{E}-14$ | 0.5007189 | 0.2910484 | 0.222273 |  |  |  |  |  | 0 |
| 325.04114 | $1.70 \mathrm{E}-08$ | 0.5731017 | 0.2791467 | 0.222945 |  |  |  |  |  | 0 |
| 475.13037 | $2.41 \mathrm{E}-12$ | 0.3853396 | 0.2556664 | 0.226803 |  |  |  |  |  | 0 |
| 316.09652 | $2.60 \mathrm{E}-09$ | 0.6020491 | 0.2853716 | 0.228 |  |  |  |  |  | 0 |
| 292.10352 | 6.06E-13 | 0.2935021 | 0.255741 | 0.234716 | C11H19NO8 | [M-H]- | 293.111069 | 292.1037926 | -0.93 | ['N-Acetylmuramate'] |
| 209.03056 | 2.20E-14 | 0.1693487 | 0.1832235 | 0.239391 | C4H6O6 | [M+Hac-H]- | 150.01644 | 209.0302936 | 1.27 | ['(R,R)-Tartaric acid', '(S,S)-Tartaric acid', 'mesoTartaric acid'] |
| 209.03056 | 2.20E-14 | 0.1693487 | 0.1832235 | 0.239391 | C6H1008 | [M-H]- | 210.03757 | 209.0302936 | 1.27 | ['Glucarate'] |
| 456.17233 | $4.73 \mathrm{E}-12$ | 0.4710542 | 0.3346852 | 0.242794 |  |  |  |  |  | 0 |
| 537.16722 | $1.38 \mathrm{E}-13$ | 0.2921718 | 0.2355661 | 0.243085 |  |  |  |  |  | 0 |
| 293.10281 | 9.62E-09 | 0.7850553 | 0.3984659 | 0.244741 | C13H14O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 234.08921 | 293.1030636 | -0.87 | ['1-Acetoxychavicol acetate', 'Coixinden B'] |
| 293.10281 | $9.62 \mathrm{E}-09$ | 0.7850553 | 0.3984659 | 0.244741 | C15H18O6 | [M-H]- | 294.11034 | 293.1030636 | -0.87 | ['Tutin'] |
| 329.0724 | $9.69 \mathrm{E}-10$ | 0.3280667 | 0.2220302 | 0.251395 |  |  |  |  |  | 0 |
| 224.04177 | 2.07E-12 | 0.4731896 | 0.3541985 | 0.252767 | C5H11NOS2 | [M+Hac-H]- | 165.028208 | 224.0420616 | -1.3 | ['4-Methylthiobutylthiohydroximate'] |
| 475.21842 | 2.55E-11 | 0.5255358 | 0.2845786 | 0.256265 |  |  |  |  |  | 0 |
| 344.07522 | 2.20E-14 | 0.21198 | 0.2884878 | 0.260398 | C11H19NO9 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 309.105984 | 344.0753856 | -0.48 | ['N-Acetylneuraminate', 'O-Acetylneuraminic acid'] |
| 487.14576 | 1.16E-09 | 0.3921999 | 0.3333083 | 0.262443 |  |  |  |  |  | 0 |
| 533.1358 | 4.01E-12 | 0.4278908 | 0.3314699 | 0.266055 |  |  |  |  |  | 0 |


| 260.07736 | $5.60 \mathrm{E}-11$ | 0.9000296 | 0.3623426 | 0.266392 | C10H15NO7 | [M-H]- | 261.084854 | 260.0775776 | -0.84 | ['Hymexazol O-glucoside', 'Hymexazol Nglucoside'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 260.07736 | $5.60 \mathrm{E}-11$ | 0.9000296 | 0.3623426 | 0.266392 | C14H13N3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 223.110947 | 260.0773986 | -0.15 | ['Mepanipyrim'] |
| 313.07752 | 8.76E-10 | 0.5628097 | 0.2762627 | 0.266728 | C7H16N8O4 | [M+K-2H]- | 276.129452 | 313.0780586 | -1.72 | ['Trimethylenetetraurea'] |
| 315.09315 | $2.58 \mathrm{E}-09$ | 0.6182748 | 0.2888085 | 0.268638 | C14H18N2O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 278.126658 | 315.0931096 | 0.13 | ['Oxadixyl', 'alpha-Ribazole'] |
| 459.1356 | $6.27 \mathrm{E}-12$ | 0.4374709 | 0.3097557 | 0.269108 |  |  |  |  |  | 0 |
| 439.08591 | $2.52 \mathrm{E}-13$ | 0.4148144 | 0.3119318 | 0.273358 |  |  |  |  |  | 0 |
| 145.02942 | $1.26 \mathrm{E}-09$ | 0.4025725 | 0.2996906 | 0.289749 | C9H6O2 | [M-H]- | 146.03678 | 145.0295036 | -0.58 | ['Coumarin'] |
| 327.09311 | $1.48 \mathrm{E}-07$ | 0.6676362 | 0.3603943 | 0.291584 | C9H16O9 | [M+Hac-H]- | 268.079435 | 327.0932886 | -0.55 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 349.01775 | $1.30 \mathrm{E}-10$ | 0.3011965 | 0.304668 | 0.293827 |  |  |  |  |  | 0 |
| 439.07717 | $1.01 \mathrm{E}-11$ | 0.4797876 | 0.3794624 | 0.303106 |  |  |  |  |  | 0 |
| 367.02843 | $2.04 \mathrm{E}-09$ | 0.4564197 | 0.2829375 | 0.306401 |  |  |  |  |  | 0 |
| 210.02611 | 6.70E-10 | 0.5602934 | 0.4275785 | 0.308395 |  |  |  |  |  | 0 |
| 299.0983 | 7.14E-09 | 0.9093184 | 0.4644473 | 0.310729 | C14H18N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 262.131743 | 299.0981946 | 0.35 | ['Methohexital', 'Physovenine'] |
| 224.95746 | 5.48E-12 | 0.5261936 | 0.344552 | 0.316431 |  |  |  |  |  | 0 |
| 340.97655 | $1.38 \mathrm{E}-13$ | 0.5153677 | 0.3534763 | 0.320481 |  |  |  |  |  | 0 |
| 206.95535 | 6.26E-09 | 0.5260582 | 0.4428468 | 0.326538 | C3H6O6S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 169.988512 | 206.9549636 | 1.87 | ['(2R)-3-Sulfolactate', '(R)-2-O-Sulfolactate', '(S)-2-O-Sulfolactate', '(S)-3-Sulfolactate', '3Sulfolactate', 'Glycerone sulfate'] |
| 343.08805 | $4.92 \mathrm{E}-12$ | 0.4509893 | 0.2986489 | 0.327658 | C15H18N2O5 | [M+(37Cl)]- | 306.121573 | 343.0880246 | 0.07 | ['Z-Gly-Pro'] |
| 431.14052 | $2.34 \mathrm{E}-13$ | 0.5721897 | 0.3520579 | 0.331072 |  |  |  |  |  | 0 |
| 175.06133 | $6.77 \mathrm{E}-08$ | 0.5192425 | 0.3936447 | 0.333649 | C5H8O3 | [M+Hac-H]- | 116.047345 | 175.0611986 | 0.75 | ['2-Oxopentanoic acid', '3-Methyl-2oxobutanoic acid', '3-Oxopentanoic acid', '5Oxopentanoate'] |
| 175.06133 | 6.77E-08 | 0.5192425 | 0.3936447 | 0.333649 | C7H12O5 | [M-H]- | 176.068475 | 175.0611986 | 0.75 | ['(2R,3S)-3-Isopropylmalate', '(R)-2-(n-Propyl)malate', '2-Propylmalate', '3-Propylmalate', 'alpha-IsopropyImalate'] |
| 195.04925 | $3.80 \mathrm{E}-11$ | 0.2445641 | 0.2146929 | 0.333874 |  |  |  |  |  | 0 |
| 253.05636 | $1.11 \mathrm{E}-11$ | 0.696028 | 0.4450446 | 0.341549 | C6H1007 | [M+Hac-H]- | 194.042655 | 253.0565086 | -0.59 | ['2-Dehydro-D-galactonate', '2-Keto-D-gluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-Dgluconate', 'D-Fructuronate', 'D-Galacturonate', <br> 'D-Glucuronate', 'D-Glucuronic acid', 'D- <br> Mannuronate', 'D-Tagaturonate', 'Galacturonic acid', 'L-Guluronic acid', 'L-Iduronic acid', 'beta- <br> D-Glucopyranuronic acid'] |
| 607.15187 | $2.76 \mathrm{E}-08$ | 0.6563634 | 0.4272613 | 0.343646 |  |  |  |  |  | 0 |
| 465.14614 | 0.00144975 | 0.5844204 | 0.302111 | 0.350071 |  |  |  |  |  | 0 |
| 422.9224 | $1.61 \mathrm{E}-10$ | 0.6941811 | 0.4202825 | 0.350521 |  |  |  |  |  | 0 |
| 168.99941 | 2.75E-09 | 0.3472652 | 0.3705017 | 0.350589 |  |  |  |  |  | 0 |


| 203.01997 | $3.53 \mathrm{E}-08$ | 0.6126981 | 0.4734441 | 0.354093 | C7H8O7 | [M-H]- | 204.027005 | 203.0197286 | 1.19 | ['Oxaloglutarate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 152.0114 | $4.44 \mathrm{E}-06$ | 0.5497884 | 0.4220416 | 0.35586 |  |  |  |  |  | 0 |
| 529.11784 | $1.48 \mathrm{E}-10$ | 0.5643161 | 0.3988349 | 0.356263 |  |  |  |  |  | 0 |
| 593.19349 | $4.40 \mathrm{E}-11$ | 0.4668984 | 0.3702708 | 0.363427 |  |  |  |  |  | 0 |
| 193.02419 | $1.97 \mathrm{E}-05$ | 0.7346498 | 0.4868844 | 0.366508 | C11H8O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 156.057515 | 193.0239666 | 1.16 | ['1-Naphthaldehyde', '2-Naphthaldehyde'] |
| 537.06298 | $3.41 \mathrm{E}-09$ | 0.5734473 | 0.4611645 | 0.372423 |  |  |  |  |  | 0 |
| 295.06701 | $5.09 \mathrm{E}-10$ | 0.7547813 | 0.4239837 | 0.37326 |  |  |  |  |  | 0 |
| 245.00673 | 3.04E-05 | 0.6392776 | 0.4628263 | 0.379509 | C3H707P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 185.992943 | 245.0067966 | -0.27 | ['2-Phospho-D-glycerate', '3-Phospho-Dglycerate', '3-Phospho-DL-glycerate'] |
| 245.00673 | $3.04 \mathrm{E}-05$ | 0.6392776 | 0.4628263 | 0.379509 | C6H1008 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 210.03757 | 245.0069716 | -0.99 | ['Glucarate'] |
| 245.00673 | 3.04E-05 | 0.6392776 | 0.4628263 | 0.379509 | C7H12O5S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 208.040547 | 245.0069986 | -1.1 | ['2-(2-Methylthio)ethylmalic acid', '3-(2Methylthio)ethylmalic acid'] |
| 208.97096 | 5.34E-07 | 0.531239 | 0.4312958 | 0.381597 |  |  |  |  |  | 0 |
| 197.0555 | 0 | 0.3916328 | 0.3419088 | 0.382883 |  |  |  |  |  | 0 |
| 252.96041 | $1.44 \mathrm{E}-06$ | 0.5565313 | 0.4809974 | 0.383241 |  |  |  |  |  | 0 |
| 587.12356 | $2.76 \mathrm{E}-08$ | 0.3867935 | 0.3859943 | 0.388177 |  |  |  |  |  | 0 |
| 217.01221 | $4.18 \mathrm{E}-10$ | 0.7520005 | 0.4177919 | 0.390371 | C6H12O4S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 180.045632 | 217.0120836 | 0.58 | ['5-Methylthio-D-ribose'] |
| 217.01221 | $4.18 \mathrm{E}-10$ | 0.7520005 | 0.4177919 | 0.390371 | C6H12O6 | [M+K-2H]- | 180.06339 | 217.0119966 | 0.98 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'DAldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'D- <br> Fuconate', 'D-Galactose', 'D-Glucose', 'D- <br> Gulose', 'D-Hamamelose', 'D-Hexose', 'D-Idose', <br> 'D-Mannose', 'D-Psicose', 'D-Sorbose', 'D- <br> Tagatose', 'D-Talose', 'Fructose(pyranose)', <br> 'Ketose', 'L-Fructose', 'L-Fuconate', 'L- <br> Galactose', 'L-Gulose', 'L-Rhamnonate', 'L- <br> Sorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha- <br> D-Glucose', 'alpha-D-Mannose', 'alpha-LSorbopyranose', 'beta-D-Fructose', 'beta-D- <br> Galactose', 'beta-D-Glucose', 'beta-D- <br> Hamamelopyranose', 'beta-D-Mannose', 'muco-Inositol', 'myo-Inositol', 'scyllo-Inositol'] |
| 217.01221 | $4.18 \mathrm{E}-10$ | 0.7520005 | 0.4177919 | 0.390371 | C9H8O5 | [M+Na-2H]- | 196.037175 | 217.0118436 | 1.69 | ['3-(3,4-Dihydroxyphenyl)pyruvate'] |
| 409.07524 | 2.68E-09 | 0.6037053 | 0.4767959 | 0.393868 |  |  |  |  |  | 0 |
| 475.19727 | 0.00669331 | 0.5471717 | 0.468483 | 0.399269 | C23H28O7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 416.183505 | 475.1973586 | -0.19 | ['Erioflorin methacrylate'] |
| 475.19727 | 0.00669331 | 0.5471717 | 0.468483 | 0.399269 | C25H32O9 | [M-H]- | 476.204635 | 475.1973586 | -0.19 | ['2-Methoxyestrone 3-glucuronide'] |
| 193.01448 | 5.68E-07 | 0.7531526 | 0.73541 | 0.400316 |  |  |  |  |  | 0 |
| 373.09873 | $1.57 \mathrm{E}-10$ | 0.6495727 | 0.3878782 | 0.401437 |  |  |  |  |  | 0 |
| 470.15153 | 7.93E-11 | 0.5459139 | 0.4192783 | 0.404812 |  |  |  |  |  | 0 |
| 219.05128 | $5.04 \mathrm{E}-11$ | 1.1371003 | 0.5687679 | 0.405348 | C12H10N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 182.084398 | 219.0508496 | 1.96 | ['Azobenzene', 'Harman'] |


| 219.05128 | 5.04E-11 | 1.1371003 | 0.5687679 | 0.405348 | C6H8O5 | [M+Hac-H]- | 160.037175 | 219.0510286 | 1.15 | ```['2-Formylglutarate', '2-Oxoadipate', '3- Oxoadipate', '3D-(3,5/4)- Trihydroxycyclohexane-1,2-dione', 'D-2,3- Diketo 4-deoxy-epi-inositol']``` |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 219.05128 | 5.04E-11 | 1.1371003 | 0.5687679 | 0.405348 | C8H12O7 | [M-H]- | 220.058305 | 219.0510286 | 1.15 | ['(R)-(Homo)2-citrate', '1-Hydroxypentane-1,2,5-tricarboxylate'] |
| 369.06747 | 3.57E-09 | 1.0115992 | 0.5088755 | 0.419746 | C12H18O13 | [M-H]- | 370.074745 | 369.0674686 | 0 | ['1,2-beta-D-Glucuronosyl-D-glucuronate', 'Digalacturonate'] |
| 369.06747 | 3.57E-09 | 1.0115992 | 0.5088755 | 0.419746 | C16H18N2O4S | [ $\mathrm{M}+\mathrm{Cl}]$ - | 334.09873 | 369.0681316 | -1.79 | ['Penicillin G'] |
| 491.17705 | $1.82 \mathrm{E}-08$ | 0.6139424 | 0.4146412 | 0.42592 | C17H34N4O10 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 454.227496 | 491.1761026 | 1.93 | ['Ribostamycin', 'Xylostasin'] |
| 491.17705 | $1.82 \mathrm{E}-08$ | 0.6139424 | 0.4146412 | 0.42592 | C19H28O11 | [M+Hac-H]- | 432.163165 | 491.1770186 | 0.06 | ['Zizybeoside I'] |
| 491.17705 | $1.82 \mathrm{E}-08$ | 0.6139424 | 0.4146412 | 0.42592 | C25H30N2O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 454.210388 | 491.1768396 | 0.43 | ['beta-Funaltrexamine'] |
| 196.05455 | 5.00E-15 | 0.4229404 | 0.3874496 | 0.430849 |  |  |  |  |  | 0 |
| 196.05371 | 0 | 0.4229405 | 0.0954447 | 0.430849 | C10H11NO | [ $\mathrm{M}+\mathrm{Cl}]$ - | 161.084064 | 196.0534656 | 1.25 | ['Boschniakine', 'Indole-3-ethanol'] |
| 196.05371 | 0 | 0.4229405 | 0.0954447 | 0.430849 | C11H13N | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 159.104799 | 196.0534056 | 1.55 | ['Pargyline'] |
| 196.05371 | 0 | 0.4229405 | 0.0954447 | 0.430849 | C13H10O2 | [M-2H]- | 198.06808 | 196.0535272 | 0.93 | ['1,2-Dihydroxyfluorene', '3,4- <br> Dihydroxyfluorene', '4-Hydroxybenzophenone', <br> 'Capillarin', 'Dehydrosafynol'] |
| 212.98937 | $2.92 \mathrm{E}-10$ | 0.5866853 | 0.5476482 | 0.433125 |  |  |  |  |  | 0 |
| 275.01718 | 2.57E-05 | 0.7985902 | 0.5232248 | 0.439891 | C4H908P | [M+Hac-H]- | 216.003508 | 275.0173616 | -0.66 | ['3-Phospho-D-erythronate', '4-Phospho-Derythronate'] |
| 275.01718 | $2.57 \mathrm{E}-05$ | 0.7985902 | 0.5232248 | 0.439891 | C6H13010P | [M-H]- | 276.024638 | 275.0173616 | -0.66 | ['2-Carboxy-D-arabinitol 1-phosphate', '6-Phospho-D-gluconate'] |
| 275.01718 | $2.57 \mathrm{E}-05$ | 0.7985902 | 0.5232248 | 0.439891 | C8H14O8 | [M+K-2H]- | 238.06887 | 275.0174766 | -1.08 | ['3-Deoxy-D-manno-octulosonate'] |
| 351.03339 | 2.77E-08 | 0.6460223 | 0.4617414 | 0.442043 |  |  |  |  |  | 0 |
| 149.04545 | 1.22E-11 | 0.3099331 | 0.2872372 | 0.445354 | C3H6O3 | [M+Hac-H]- | 90.031695 | 149.0455486 | -0.66 | ['(R)-Lactate', '(S)-Lactate', '3- <br> Hydroxypropanoate', 'D-Glyceraldehyde', <br> 'Glyceraldehyde', 'Glycerone', 'L- <br> Glyceraldehyde', 'Lactate'] |
| 149.04545 | 1.22E-11 | 0.3099331 | 0.2872372 | 0.445354 | C5H10O5 | [M-H]- | 150.052825 | 149.0455486 | -0.66 | ['D-Apiose', 'D-Arabinose', 'D-Lyxose', 'DRibose', 'D-Ribulose', 'D-Xylose', 'D-Xylulose', 'LArabinofuranose', 'L-Arabinose', 'L-Lyxose', 'LRibulose', 'L-Xylose', 'L-Xylulose', 'Ribulose', 'Xylose', 'alpha-D-Lyxose', 'alpha-D-Ribulose', 'alpha-L-Arabinose', 'beta-D-Apiose', 'beta-DRibofuranose', 'beta-D-Ribopyranose', 'beta-DXylose', 'beta-L-Arabinose'] |


| 212.03332 | 5.23E-08 | 0.6441335 | 0.621294 | 0.448378 | C10H9NO3 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 191.058244 | 212.0329126 | 1.92 | ['5,6-Dihydroxy-3-methyl-2-oxo-1,2- <br> dihydroquinoline', '5-Hydroxyindoleacetate', '5-Phenyl-1,3-oxazinane-2,4-dione'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 212.03332 | 5.23E-08 | 0.6441335 | 0.621294 | 0.448378 | C3H8NO4P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 153.019097 | 212.0329506 | 1.74 | ['Fosamine'] |
| 212.03332 | 5.23E-08 | 0.6441335 | 0.621294 | 0.448378 | C6H11NO5 | [M+Cl]- | 177.063724 | 212.0331256 | 0.92 | ['3-Keto-scyllo-inosamine', '4-Hydroxy-4methylglutamate'] |
| 212.03332 | 5.23E-08 | 0.6441335 | 0.621294 | 0.448378 | C7H13NO4 | [M+K-2H]- | 175.084459 | 212.0330656 | 1.2 | ['Calystegin B2'] |
| 212.03332 | 5.23E-08 | 0.6441335 | 0.621294 | 0.448378 | C9H10O6 | [M-2H]- | 214.04774 | 212.0331872 | 0.63 | ['2-Hydroxy-6-oxonona-2,4-diene-1,9-dioate'] |
| 665.21446 | 8.78E-09 | 0.5583893 | 0.4501172 | 0.453145 | C24H42O21 | [M-H]- | 666.221865 | 665.2145886 | -0.19 | ['1,3-alpha-D-Mannosyl-1,2-alpha-D-mannosyl- <br> 1,2-alpha-D-mannosyl-D-mannose', <br> 'Cellotetraose', 'Glycogen', 'Isolychnose', <br> 'Lychnose', 'Maltotetraose', 'Stachyose', 'alpha-D-Galactosyl-(1-6)-alpha-D-galactosyl-(1-6)-beta-D-fructosyl-(2-1)-alpha-D-glucoside'] |
| 270.04672 | 8.34E-10 | 0.8059132 | 0.6088738 | 0.455271 |  |  |  |  |  | 0 |
| 317.10878 | 2.17E-07 | 0.6176856 | 0.4482826 | 0.455382 | C18H2ON2S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 296.13472 | 317.1093886 | -1.92 | ['Methdilazine'] |
| 391.10931 | 1.92E-11 | 0.3035513 | 0.3293264 | 0.455431 | C13H21N2O4PS | [M+Hac-H]- | 332.095968 | 391.1098216 | -1.31 | ['Butamifos', 'O-Ethyl-O-(5-methyl-4- nitrophenyl)(1- methylpropyl)phosphoramidothioate'] |
| 391.10931 | 1.92E-11 | 0.3035513 | 0.3293264 | 0.455431 | C16H22N2O7 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 354.142703 | 391.1091546 | 0.4 | ['Dinocton 6'] |
| 391.10931 | $1.92 \mathrm{E}-11$ | 0.3035513 | 0.3293264 | 0.455431 | C17H26N2O4S | [M+K-2H]- | 354.16133 | 391.1099366 | -1.6 | ['Sultopride'] |
| 663.15659 | 9.52E-08 | 0.7729028 | 0.5854414 | 0.461372 |  |  |  |  |  | 0 |
| 348.92176 | 4.07E-09 | 0.7136336 | 0.5673939 | 0.464179 |  |  |  |  |  | 0 |
| 211.01008 | $4.73 \mathrm{E}-12$ | 0.5046896 | 0.4851478 | 0.464927 | C4H8O2S2 | [M+Hac-H]- | 151.996574 | 211.0104276 | -1.65 | ['3-Mercapto-2-mercaptomethylpropanoate', '4,5-cis-Dihydroxy-1,2-dithiacyclohexane', 'Oxidized dithiothreitol'] |
| 447.13564 | 9.59E-09 | 0.5641487 | 0.4920347 | 0.465324 | C20H30N2O5S | [M+K-2H]- | 410.187545 | 447.1361516 | -1.14 | ['Benfuracarb'] |
| 245.01517 | 6.42E-10 | 0.7267123 | 0.5308309 | 0.472251 |  |  |  |  |  | 0 |
| 161.04565 | 6.20E-07 | 0.8777122 | 0.7706932 | 0.472652 | C4H6O3 | [M+Hac-H]- | 102.031695 | 161.0455486 | 0.63 | ['(S)-Methylmalonate semialdehyde', '2-Methyl-3-oxopropanoate', '2-Oxobutanoate', 'Acetoacetate', 'Succinate semialdehyde'] |


| 161.04565 | 6.20E-07 | 0.8777122 | 0.7706932 | 0.472652 | C6H10O5 | [M-H]- | 162.052825 | 161.0455486 | 0.63 | ['(2R,3S)-2,3-Dimethylmalate', '(R)-2Ethylmalate', '(R)-3,3-Dimethylmalate', '(S)-2(Hydroxymethyl)glutarate', '1,5-Anhydro-Dfructose', '2-Dehydro-3-deoxy-D-fuconate', '2-Dehydro-3-deoxy-L-fuconate', '2-Dehydro-3-deoxy-L-rhamnonate', '2-Deoxy-scyllo-inosose', '2-Hydroxyadipate', '3,6-Anhydrogalactose', <br> '3,6-Anhydroglucose', '3-Ethylmalate', '3-Hydroxy-3-methylglutarate', 'D-Fucono-1,4lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5-lactone', 'L-Rhamnono-1,4-lactone', 'Lichenin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 355.0669 | 6.75E-11 | 0.5687875 | 0.539913 | 0.477677 | C13H12O8 | [M+Hac-H]- | 296.05322 | 355.0670736 | -0.49 | ['Phaseolic acid'] |
| 326.96093 | $2.01 \mathrm{E}-11$ | 0.8606344 | 0.5608697 | 0.483441 |  |  |  |  |  | 0 |
| 443.19235 | 8.03E-09 | 0.6073158 | 0.5391161 | 0.48371 |  |  |  |  |  | 0 |
| 195.05088 | $7.00 \mathrm{E}-15$ | 0.4970207 | 0.445317 | 0.490658 | C10H10N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 158.084398 | 195.0508496 | 0.16 | ['1,5-Naphthalenediamine', 'Nicotyrine'] |
| 195.05088 | $7.00 \mathrm{E}-15$ | 0.4970207 | 0.445317 | 0.490658 | C4H8O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 136.037175 | 195.0510286 | -0.76 | ['Threonate'] |
| 195.05088 | 7.00E-15 | 0.4970207 | 0.445317 | 0.490658 | C6H12O7 | [M-H]- | 196.058305 | 195.0510286 | -0.76 | ['2-Carboxy-D-arabinitol', 'D-Altronate', 'DGluconic acid', 'D-Mannonate'] |
| 310.96601 | 1.80E-08 | 0.922486 | 0.6498982 | 0.492097 | C6H10010S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 273.999472 | 310.9659236 | 0.28 | ['L-Iduronate 2-sulfate'] |
| 393.04403 | 3.86E-08 | 0.6641472 | 0.4616755 | 0.501341 | C15H16O11 | [M+Na-2H]- | 372.069265 | 393.0439336 | 0.25 | ['2-O-Caffeoylglucarate'] |
| 294.0828 | $2.05 \mathrm{E}-10$ | 1.1432976 | 0.7965046 | 0.503097 | C8H13NO7 | [M+Hac-H]- | 235.069204 | 294.0830576 | -0.88 | ['N-Acetylgalactosaminate', 'N-Glycolyl-Dmannosaminolactone'] |
| 379.02841 | 3.72E-09 | 0.5571035 | 0.4780298 | 0.508254 |  |  |  |  |  | 0 |
| 324.09347 | $1.84 \mathrm{E}-05$ | 1.0562043 | 0.6449995 | 0.510103 | C11H19NO10 | [M-H]- | 325.100899 | 324.0936226 | -0.47 | ['N-Glycoloyl-neuraminate'] |
| 324.09347 | $1.84 \mathrm{E}-05$ | 1.0562043 | 0.6449995 | 0.510103 | C15H17N3O3 | [ $\mathrm{M}+(37 \mathrm{Cl})]$ - | 287.126992 | 324.0934436 | 0.08 | ['SR95531'] |
| 338.10915 | 3.24E-06 | 0.9803515 | 0.6289546 | 0.515461 | C16H19N3O3 | [M+(37Cl)]- | 301.142642 | 338.1090936 | 0.17 | ['Febrifugine', 'Isofebrifugine'] |
| 273.01003 | $1.83 \mathrm{E}-08$ | 0.7188113 | 0.588413 | 0.51746 |  |  |  |  |  | 0 |
| 513.08652 | 7.79E-11 | 0.6123679 | 0.5315235 | 0.525245 |  |  |  |  |  | 0 |
| 209.06696 | 3.00E-15 | 0.4672038 | 0.4600067 | 0.526261 | C5H10O5 | [M+Hac-H]- | 150.052825 | 209.0666786 | 1.35 | ['D-Apiose', 'D-Arabinose', 'D-Lyxose', 'DRibose', 'D-Ribulose', 'D-Xylose', 'D-Xylulose', 'LArabinofuranose', 'L-Arabinose', 'L-Lyxose', 'LRibulose', 'L-Xylose', 'L-Xylulose', 'Ribulose', 'Xylose', 'alpha-D-Lyxose', 'alpha-D-Ribulose', 'alpha-L-Arabinose', 'beta-D-Apiose', 'beta-DRibofuranose', 'beta-D-Ribopyranose', 'beta-DXylose', 'beta-L-Arabinose'] |
| 209.06696 | 3.00E-15 | 0.4672038 | 0.4600067 | 0.526261 | C7H14O7 | [M-H]- | 210.073955 | 209.0666786 | 1.35 | ['Sedoheptulose', 'alpha-DMannoheptulopyranose', 'beta-DSedoheptulopyranose'] |


| 445.208 | $2.96 \mathrm{E}-07$ | 0.6891546 | 0.6165431 | 0.527769 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 194.94682 | 0.00013296 | 0.7407897 | 0.7130936 | 0.53079 |  |  |  |  |  | 0 |
| 377.13007 | $1.49 \mathrm{E}-09$ | 0.5346965 | 0.4547914 | 0.533645 | C13H23N2O3PS | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 318.116703 | 377.1305566 | -1.29 | ['Tebupirimfos'] |
| 377.13007 | $1.49 \mathrm{E}-09$ | 0.5346965 | 0.4547914 | 0.533645 | C16H26N2O4S | [M+Cl]- | 342.16133 | 377.1307316 | -1.75 | ['Penicillin K'] |
| 325.07752 | $1.59 \mathrm{E}-07$ | 0.7425966 | 0.564222 | 0.543721 |  |  |  |  |  | 0 |
| 195.05285 | 8.79E-08 | 0.572757 | 0.4372166 | 0.544333 |  |  |  |  |  | 0 |
| 461.13007 | $1.43 \mathrm{E}-07$ | 0.6244438 | 0.5951622 | 0.548495 |  |  |  |  |  | 0 |
| 214.89208 | $4.90 \mathrm{E}-07$ | 0.8120964 | 0.6357129 | 0.54898 | H4O7P2 | [M+K-2H]- | 177.943231 | 214.8918376 | 1.13 | ['Diphosphate'] |
| 527.10215 | $1.92 \mathrm{E}-11$ | 0.4235609 | 0.4954678 | 0.563015 |  |  |  |  |  | 0 |
| 675.17542 | 5.80E-06 | 0.5187501 | 0.5226454 | 0.580023 |  |  |  |  |  | 0 |
| 737.23617 | $2.98 \mathrm{E}-05$ | 0.5684854 | 0.5880734 | 0.582512 |  |  |  |  |  | 0 |
| 217.00253 | 0.00183037 | 0.7962293 | 0.922499 | 0.593591 | C7H6N2O4 | [M+Cl]- | 182.032758 | 217.0021596 | 1.71 | ['2,4-Dinitrotoluene', '2,6-Dinitrotoluene', '3,5Dinitrotoluene', 'Nitrofurylacrylamide'] |
| 217.00253 | 0.00183037 | 0.7962293 | 0.922499 | 0.593591 | C8H8N2O3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 180.053493 | 217.0020996 | 1.98 | ['Isonicotinylglycine', 'Nicotinurate'] |
| 195.05241 | $3.53 \mathrm{E}-11$ | 0.5803249 | 0.4787904 | 0.594097 | C5H4N4O | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 136.038511 | 195.0523646 | 0.23 | ['Hypoxanthine'] |
| 195.05241 | $3.53 \mathrm{E}-11$ | 0.5803249 | 0.4787904 | 0.594097 | C7H8N4O3 | [M-H]- | 196.059641 | 195.0523646 | 0.23 | ['1,7-Dimethyluric acid', '3,7-Dimethyluric acid'] |
| 337.07748 | $6.53 \mathrm{E}-13$ | 0.7228047 | 0.6649034 | 0.597697 | C12H18O11 | [M-H]- | 338.084915 | 337.0776386 | -0.47 | ['L-Ascorbic acid-2-glucoside'] |
| 337.07748 | $6.53 \mathrm{E}-13$ | 0.7228047 | 0.6649034 | 0.597697 | C16H16N2O4 | [ $\mathrm{M}+(37 \mathrm{Cl})]$ - | 300.111008 | 337.0774596 | 0.06 | ['5-Nitro-2-(3-phenylpropylamino)benzoic acid', 'Desmedipham', 'Phenmedipham'] |
| 199.01005 | $7.03 \mathrm{E}-08$ | 0.6273519 | 0.5078039 | 0.597884 | C5H12O4S2 | [M-H]- | 200.017704 | 199.0104276 | -1.9 | ['(R)-2-Hydroxypropyl-CoM', '(S)-2-Hydroxypropyl-CoM'] |
| 153.01922 | $2.68 \mathrm{E}-09$ | 0.7265566 | 0.6551012 | 0.610829 | C7H6O4 | [M-H]- | 154.02661 | 153.0193336 | -0.74 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 330.98369 | $1.10 \mathrm{E}-07$ | 0.5755378 | 0.5774092 | 0.625129 |  |  |  |  |  | 0 |
| 165.04061 | $1.24 \mathrm{E}-08$ | 0.3633088 | 0.4059095 | 0.626453 | C3H6O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 106.02661 | 165.0404636 | 0.89 | ['D-Glycerate'] |
| 165.04061 | $1.24 \mathrm{E}-08$ | 0.3633088 | 0.4059095 | 0.626453 | C5H1006 | [M-H]- | 166.04774 | 165.0404636 | 0.89 | ['D-Ribonate', 'D-Xylonate', 'L-Arabinonate'] |
| 445.12005 | $1.78 \mathrm{E}-07$ | 0.7989763 | 0.7363941 | 0.62941 | C20H28N2O5S | [M+K-2H]- | 408.171895 | 445.1205016 | -1.01 | ['Tamsulosin'] |
| 445.12005 | $1.78 \mathrm{E}-07$ | 0.7989763 | 0.7363941 | 0.62941 | C23H24N2O4S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 424.14568 | 445.1203486 | -0.67 | ['Eprosartan'] |
| 153.05561 | $1.96 \mathrm{E}-07$ | 0.6990111 | 0.715614 | 0.632227 | C6H6O | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 94.041865 | 153.0557186 | -0.71 | ['Arene oxide', 'Phenol'] |
| 153.05561 | 1.96E-07 | 0.6990111 | 0.715614 | 0.632227 | C8H10O3 | [M-H]- | 154.062995 | 153.0557186 | -0.71 | ['2,6-Dimethoxyphenol', '4-Hydroxy-3-methoxy-benzenemethanol'] |
| 403.10928 | 0.00022601 | 0.7527866 | 0.4895915 | 0.637497 |  |  |  |  |  | 0 |
| 405.12496 | $1.14 \mathrm{E}-07$ | 0.7235801 | 0.6191933 | 0.639371 |  |  |  |  |  | 0 |
| 401.09369 | $1.88 \mathrm{E}-07$ | 0.9598867 | 0.6885419 | 0.639894 |  |  |  |  |  | 0 |
| 237.06143 | 7.17E-11 | 0.5666956 | 0.5986669 | 0.64181 | C12H12N2O | [M+(37Cl)]- | 200.094963 | 237.0614146 | 0.06 | ['4,4-Diaminodiphenyl ether', 'Harmalol'] |


| 237.06143 | 7.17E-11 | 0.5666956 | 0.5986669 | 0.64181 | C6H1006 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 178.04774 | 237.0615936 | -0.69 | ['2,4,6/3,5-Pentahydroxycyclohexanone', '2-Dehydro-3-deoxy-D-galactonate', '2-Dehydro-3-deoxy-D-gluconate', '2-Dehydro-D-glucose', '2-Deoxy-5-keto-D-gluconic acid', '3-Dehydro-2-deoxy-D-gluconate', '3-Keto-beta-D-galactose', '5-Dehydro-2-deoxy-D-gluconate', '5-Dehydro-D-fructose', '5-Deoxy glucuronic acid', 'D-Galactono-1,4-lactone', 'D-Galactono-1,5lactone', 'D-Glucono-1,4-lactone', 'D-Glucono-1,5-lactone', 'D-galacto-Hexodialdose', 'Gulono-1,4-lactone', 'Hexose-1,5-lactone', 'L-Galactono-1,4-lactone', 'L-Gulono-1,4-lactone', 'myo-Inosose-5'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 237.06143 | 7.17E-11 | 0.5666956 | 0.5986669 | 0.64181 | C8H14O8 | [M-H]- | 238.06887 | 237.0615936 | -0.69 | ['3-Deoxy-D-manno-octulosonate'] |
| 257.00672 | $4.24 \mathrm{E}-07$ | 1.2958549 | 0.9201223 | 0.650848 | C11H8O6 | [M+Na-2H]- | 236.03209 | 257.0067586 | -0.15 | ['2-Carboxy-2-hydroxy-8-carboxychromene', '2-Hydroxy-3-carboxybenzalpyruvate'] |
| 257.00672 | 4.24E-07 | 1.2958549 | 0.9201223 | 0.650848 | C4H707P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 197.992943 | 257.0067966 | -0.3 | ['Phosphoenol-4-deoxy-3-tetrulosonate'] |
| 257.00672 | $4.24 \mathrm{E}-07$ | 1.2958549 | 0.9201223 | 0.650848 | C6H1109P | [M-H]- | 258.014073 | 257.0067966 | -0.3 | ['2-Dehydro-3-deoxy-6-phospho-D-gluconate', '2-Dehydro-3-deoxy-D-galactonate 6phosphate', '2-Deoxy-5-keto-D-gluconic acid 6phosphate', '6-Phospho-5-dehydro-2-deoxy-Dgluconate', 'D-Glucono-1,5-lactone 6phosphate'] |
| 257.00672 | $4.24 \mathrm{E}-07$ | 1.2958549 | 0.9201223 | 0.650848 | C8H12O7 | [M+K-2H]- | 220.058305 | 257.0069116 | -0.75 | ['(R)-(Homo)2-citrate', '1-Hydroxypentane-1,2,5-tricarboxylate'] |
| 256.02272 | $2.47 \mathrm{E}-08$ | 0.8633354 | 0.6618869 | 0.65132 | C11H9NO5 | [M+Na-2H]- | 235.048074 | 256.0227426 | -0.09 | ['3,4-Dihydro-7-methoxy-2-methylene-3-oxo$2 \mathrm{H}-1,4$-benzoxazine-5-carboxylic acid'] |
| 256.02272 | $2.47 \mathrm{E}-08$ | 0.8633354 | 0.6618869 | 0.65132 | C8H13NO6 | [M+K-2H]- | 219.074289 | 256.0228956 | -0.69 | ['N-Acetyl-D-mannosaminolactone', 'O-Succinyl-L-homoserine'] |
| 606.19681 | 4.84E-08 | 0.6722764 | 0.7017095 | 0.65223 | C30H35NO11 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 585.221014 | 606.1956826 | 1.86 | ['epsilon-Rhodomycin T'] |
| 317.05467 | $9.89 \mathrm{E}-11$ | 0.8820438 | 0.6932609 | 0.652948 | C12H14N2O6 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 282.085188 | 317.0545896 | 0.25 | ['Dinoseb acetate'] |
| 317.05467 | $9.89 \mathrm{E}-11$ | 0.8820438 | 0.6932609 | 0.652948 | C16H12N2O4 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 296.079708 | 317.0543766 | 0.93 | ['3,3-Dimethoxybenzidine-4,4-diisocyanate'] |
| 505.20813 | 0.00073329 | 0.750391 | 0.6625687 | 0.658016 | C24H3008 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 446.19407 | 505.2079236 | 0.41 | ['Estrone glucuronide', 'Yangambin'] |
| 505.20813 | 0.00073329 | 0.750391 | 0.6625687 | 0.658016 | C26H34O10 | [M-H]- | 506.2152 | 505.2079236 | 0.41 | ['Limonoate'] |
| 333.05897 | 3.66E-11 | 0.7599722 | 0.6696672 | 0.658688 | C17H14NO4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 296.092284 | 333.0587356 | 0.7 | ['2,3,9,10-Tetrahydroxyberberine'] |
| 333.05897 | $3.66 \mathrm{E}-11$ | 0.7599722 | 0.6696672 | 0.658688 | C7H1509P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 274.045373 | 333.0592266 | -0.77 | ['1-Deoxy-D-altro-heptulose 7-phosphate'] |
| 333.05897 | $3.66 \mathrm{E}-11$ | 0.7599722 | 0.6696672 | 0.658688 | C9H19O11P | [M-H]- | 334.066503 | 333.0592266 | -0.77 | ['2-(alpha-D-Galactosyl)-sn-glycerol 3phosphate', '2-(beta-D-Glucosyl)-sn-glycerol 3phosphate', "alpha-D-Galactosyl-(1,1')-snglycerol 3-phosphate", 'sn-glycero-3-Phospho-1-inositol'] |


| 334.06244 | $1.92 \mathrm{E}-11$ | 0.785739 | 0.6669554 | 0.659607 | C12H15N3O6 | [M+(37Cl)]- | 297.096087 | 334.0625386 | -0.3 | ['Musk xylene'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 308.09861 | 2.60E-09 | 0.9275886 | 0.6489633 | 0.662561 | C11H19NO9 | [M-H]- | 309.105984 | 308.0987076 | -0.32 | ['N-AcetyIneuraminate', 'O-Acetylneuraminic acid'] |
| 429.06512 | 3.27E-10 | 0.4298733 | 0.485579 | 0.663043 |  |  |  |  |  | 0 |
| 323.03847 | $1.94 \mathrm{E}-10$ | 0.6472829 | 0.5793439 | 0.669212 |  |  |  |  |  | 0 |
| 269.08758 | $1.62 \mathrm{E}-07$ | 0.9578489 | 0.7123637 | 0.675424 | C13H16N2O2 | [ $\mathrm{M}+(37 \mathrm{Cl})]$ - | 232.121178 | 269.0876296 | -0.18 | ['Aminoglutethimide', 'Melatonin'] |
| 269.08758 | $1.62 \mathrm{E}-07$ | 0.9578489 | 0.7123637 | 0.675424 | C7H14O7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 210.073955 | 269.0878086 | -0.85 | ['Sedoheptulose', 'alpha-DMannoheptulopyranose', 'beta-DSedoheptulopyranose'] |
| 517.12002 | 0.0003364 | 0.7834451 | 0.8329271 | 0.69354 |  |  |  |  |  | 0 |
| 518.12344 | 0.00485222 | 0.7791812 | 0.7104766 | 0.697284 |  |  |  |  |  | 0 |
| 224.00538 | 0.01093812 | 0.6600771 | 0.6581408 | 0.703178 | C9H7N3S | [ $\mathrm{M}+\mathrm{Cl}]-$ | 189.036069 | 224.0054706 | -0.4 | ['Tricyclazole'] |
| 205.03559 | $1.01 \mathrm{E}-06$ | 0.8743662 | 0.7089508 | 0.710468 | C5H6O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 146.021525 | 205.0353786 | 1.03 | ['2-Oxoglutarate', '5-Hydroxy-2,4dioxopentanoate', 'Dehydro-D-arabinono-1,4lactone', 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] |
| 205.03559 | $1.01 \mathrm{E}-06$ | 0.8743662 | 0.7089508 | 0.710468 | C7H1007 | [M-H]- | 206.042655 | 205.0353786 | 1.03 | ['(2S,3R)-3-Hydroxybutane-1,2,3tricarboxylate', '(R)-2-Hydroxybutane-1,2,4tricarboxylate', '2-Methylcitrate', 'Homoisocitrate'] |
| 245.04303 | $1.76 \mathrm{E}-06$ | 0.8515053 | 0.7554842 | 0.712671 | C11H12O5 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 224.068475 | 245.0431436 | -0.46 | ['Sinapate'] |
| 245.04303 | $1.76 \mathrm{E}-06$ | 0.8515053 | 0.7554842 | 0.712671 | C6H15O8P | [M-H]- | 246.050458 | 245.0431816 | -0.62 | ['Glycerophosphoglycerol'] |
| 245.04303 | $1.76 \mathrm{E}-06$ | 0.8515053 | 0.7554842 | 0.712671 | C7H14O7 | [M+Cl]- | 210.073955 | 245.0433566 | -1.33 | ['Sedoheptulose', 'alpha-DMannoheptulopyranose', 'beta-DSedoheptulopyranose'] |
| 315.12955 | 3.69E-06 | 0.9917881 | 0.6712976 | 0.716674 |  |  |  |  |  | 0 |
| 601.13889 | 6.51E-09 | 0.7139133 | 0.7063692 | 0.719651 |  |  |  |  |  | 0 |
| 268.94688 | 0.02927171 | 1.0342467 | 0.8159588 | 0.733822 |  |  |  |  |  | 0 |
| 216.00585 | 0.00806021 | 0.6833554 | 1.0015149 | 0.744351 |  |  |  |  |  | 0 |
| 357.10371 | $2.85 \mathrm{E}-10$ | 0.6460287 | 0.6819763 | 0.755495 |  |  |  |  |  | 0 |
| 221.0306 | $1.03 \mathrm{E}-10$ | 1.514048 | 0.9663753 | 0.755694 | C5H6O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 162.01644 | 221.0302936 | 1.39 | ['4-Hydroxy-2-oxoglutarate', 'D-4-Hydroxy-2oxoglutarate'] |
| 461.16637 | 0.00010206 | 0.7535825 | 0.8116996 | 0.757433 | C24H28N2O5 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 424.199823 | 461.1662746 | 0.21 | ['Benazepril'] |
| 209.09348 | $3.24 \mathrm{E}-09$ | 1.9581177 | 1.3310131 | 0.761755 | C10H14N2O3 | [M-H]- | 210.100443 | 209.0931666 | 1.5 | ['2,6-Dihydroxypseudooxynicotine', 'Aprobarbital'] |
| 209.09348 | $3.24 \mathrm{E}-09$ | 1.9581177 | 1.3310131 | 0.761755 | C8H10N2O | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 150.079313 | 209.0931666 | 1.5 | ['N-Methylanthranilamide'] |
| 217.03566 | 0.00303073 | 1.0980489 | 0.8153223 | 0.766704 | C6H6O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 158.021525 | 217.0353786 | 1.3 | ['2-Hydroxymuconate', '2-Maleylacetate', '3-Hydroxy-cis,cis-muconate', '4-Methylene-2oxoglutarate', 'gamma-Oxalocrotonate'] |


| 605.19342 | $2.02 \mathrm{E}-05$ | 0.7774883 | 0.8135175 | 0.770257 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 563.18298 | 0.00026707 | 0.9249743 | 0.7921744 | 0.770863 | C18H32O16 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 504.16904 | 563.1828936 | 0.15 | ['1F-beta-D-Fructosylsucrose', '6-alphaMaltosylglucose', '6F-alpha-D- <br> Galactosylsucrose', 'Cellotriose', 'D-Gal alpha 1->6D-Gal alpha 1->6D-Glucose', <br> 'Galactomannan', 'Isomaltotriose', <br> 'Maltotriose', 'Melezitose', 'Panose', <br> 'Raffinose', 'Umbelliferose', 'beta-D- <br> Fructofuranosyl O-beta-D-glucopyranosyl-(1-6)-alpha-D-glucopyranoside'] |
| 342.97422 | 0.01693952 | 0.9873538 | 1.035809 | 0.774457 | C9H11N2O8P | [M+K-2H]- | 306.025306 | 342.9739126 | 0.9 | ["2',3-Cyclic UMP"] |
| 251.07712 | 0.01071764 | 1.2745136 | 0.9861017 | 0.77455 | C13H14N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 214.110613 | 251.0770646 | 0.22 | ['Harmaline'] |
| 251.07712 | 0.01071764 | 1.2745136 | 0.9861017 | 0.77455 | C7H12O6 | [M+Hac-H]- | 192.06339 | 251.0772436 | -0.49 | ['2D-5-O-Methyl-2,3,5/4,6pentahydroxycyclohexanone', 'Quinate', 'Valiolone'] |
| 210.0411 | 3.47E-10 | 1.9695227 | 1.0548326 | 0.791598 | C7H5NO3 | [M+Hac-H]- | 151.026944 | 210.0407976 | 1.44 | ['6-Imino-5-oxocyclohexa-1,3dienecarboxylate'] |
| 210.0411 | 3.47E-10 | 1.9695227 | 1.0548326 | 0.791598 | C9H9NO5 | [M-H]- | 211.048074 | 210.0407976 | 1.44 | ['5-(2-Formylethyl)-4,6-dihydroxypicolinate', 'Betalamic acid', 'DIMBOA'] |
| 152.99572 | 4.23E-06 | 0.7694445 | 0.8265903 | 0.792313 | C3H7O5P | [M-H]- | 154.003113 | 152.9958366 | -0.76 | ['Glycerol 1,2-cyclic phosphate', 'Hydroxyacetone phosphate', 'Propanoyl phosphate'] |
| 152.99572 | 4.23E-06 | 0.7694445 | 0.8265903 | 0.792313 | C4H6O4 | [M+Cl]- | 118.02661 | 152.9960116 | -1.91 | ['Methyl oxalate', 'Methylmalonate', 'Succinate'] |
| 152.99572 | 4.23E-06 | 0.7694445 | 0.8265903 | 0.792313 | C5H8O3 | [M+K-2H]- | 116.047345 | 152.9959516 | -1.51 | ['2-Oxopentanoic acid', '3-Methyl-2oxobutanoic acid', '3-Oxopentanoic acid', '5Oxopentanoate'] |
| 310.11426 | $7.35 \mathrm{E}-07$ | 1.0094389 | 0.7094194 | 0.795665 | C15H19N3O2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 273.147727 | 310.1141786 | 0.26 | ['N-Desmethylzolmitriptan'] |
| 310.11426 | $7.35 \mathrm{E}-07$ | 1.0094389 | 0.7094194 | 0.795665 | C9H17NO7 | [M+Hac-H]- | 251.100504 | 310.1143576 | -0.31 | ['Muramic acid'] |
| 154.9751 | 3.08E-05 | 1.0407251 | 0.9465319 | 0.802833 | C2H5O6P | [M-H]- | 155.982378 | 154.9751016 | -0.01 | ['2-Phosphoglycolate'] |
| 154.9751 | 3.08E-05 | 1.0407251 | 0.9465319 | 0.802833 | C3H4O5 | [M+Cl]- | 120.005875 | 154.9752766 | -1.14 | ['Hydroxymalonate'] |
| 154.9751 | 3.08E-05 | 1.0407251 | 0.9465319 | 0.802833 | C4H6O4 | [M+K-2H]- | 118.02661 | 154.9752166 | -0.75 | ['Methyl oxalate', 'Methylmalonate', 'Succinate'] |
| 152.95945 | $4.43 \mathrm{E}-06$ | 0.8716356 | 0.8408769 | 0.803023 | C3H2O5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 117.990225 | 152.9596266 | -1.15 | ['Oxomalonate'] |
| 152.95945 | $4.43 \mathrm{E}-06$ | 0.8716356 | 0.8408769 | 0.803023 | C4H4O4 | [M+K-2H]- | 116.01096 | 152.9595666 | -0.76 | ['Formylpyruvate', 'Fumarate', 'Maleic acid'] |
| 398.13041 | 0.00489583 | 1.3180826 | 0.7692107 | 0.805601 |  |  |  |  |  | 0 |
| 489.1462 | $7.43 \mathrm{E}-09$ | 0.4196573 | 0.6723242 | 0.81228 |  |  |  |  |  | 0 |
| 151.02603 | $7.04 \mathrm{E}-07$ | 0.8978715 | 0.8281472 | 0.824827 | C5H4N4O2 | [M-H]- | 152.033426 | 151.0261496 | -0.79 | ['Alloxanthine', 'Xanthine'] |
| 195.04875 | $6.08 \mathrm{E}-05$ | 0.6236576 | 0.5474836 | 0.827591 |  |  |  |  |  | 0 |
| 603.1779 | $1.21 \mathrm{E}-07$ | 0.4801917 | 0.7331303 | 0.838032 |  |  |  |  |  | 0 |
| 271.10321 | $2.73 \mathrm{E}-05$ | 1.0601982 | 0.7937968 | 0.841613 | C13H18N2O2 | [M+(37Cl)]- | 234.136828 | 271.1032796 | -0.26 | ['Lenacil', 'p-Coumaroylputrescine'] |


| 271.10321 | $2.73 \mathrm{E}-05$ | 1.0601982 | 0.7937968 | 0.841613 | C7H16O7 | [M+Hac-H]- | 212.089605 | 271.1034586 | -0.92 | ['Perseitol', 'Volemitol'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 463.32768 | 0.49219606 | 0.9146847 | 1.0825694 | 0.848701 |  |  |  |  |  | 0 |
| 388.99778 | $1.73 \mathrm{E}-06$ | 0.7626183 | 0.7696675 | 0.85262 |  |  |  |  |  | 0 |
| 215.00499 | 0.04544988 | 0.9250156 | 1.0660222 | 0.868341 | C3H8O3S2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 155.991489 | 215.0053426 | -1.64 | ['2-(Methylthio)ethanesulfonate'] |
| 387.11431 | 3.20E-05 | 1.1642861 | 0.8388888 | 0.876146 | C17H24N2O4S | [M+Cl]- | 352.14568 | 387.1150816 | -1.99 | ['Mercaptoacetyl-Phe-Leu'] |
| 445.18686 | 0.45754487 | 0.7507914 | 1.0936735 | 0.900807 | C22H26O6 | [M+Hac-H]- | 386.17294 | 445.1867936 | 0.15 | ['(+)-Eudesmin', 'Burseran'] |
| 445.18686 | 0.45754487 | 0.7507914 | 1.0936735 | 0.900807 | C24H3008 | [M-H]- | 446.19407 | 445.1867936 | 0.15 | ['Estrone glucuronide', 'Yangambin'] |
| 224.05672 | 3.36E-08 | 1.5943842 | 1.0123749 | 0.916325 | C10H11NO5 | [M-H]- | 225.063724 | 224.0564476 | 1.22 | ['2-Amino-2-deoxyisochorismate', '4-Amino-4deoxychorismate'] |
| 224.05672 | 3.36E-08 | 1.5943842 | 1.0123749 | 0.916325 | C8H7NO3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 165.042594 | 224.0564476 | 1.22 | ['3,4-Dihydroxymandelonitrile', '4- <br> Nitroacetophenone', '4-Pyridoxolactone', '5- <br> Pyridoxolactone', 'Formylanthranilate', 'HBOA'] |
| 417.12494 | 0.00558841 | 1.0683865 | 0.89838 | 0.918479 | C18H26N2O5S | [ $\mathrm{M}+\mathrm{Cl}]-$ | 382.156245 | 417.1256466 | -1.69 | ['Furathiocarb'] |
| 475.16681 | 0.34839962 | 0.972892 | 0.9598871 | 0.921405 |  |  |  |  |  | 0 |
| 388.11774 | 0.13887748 | 1.0989958 | 0.8676693 | 0.925065 | C19H21N5O2 | [M+K-2H]- | 351.169525 | 388.1181316 | -1.01 | ['Pirenzepine'] |
| 272.90563 | 0.03504469 | 1.164847 | 0.9726916 | 0.94008 |  |  |  |  |  | 0 |
| 469.0965 | 0.00035833 | 0.7699898 | 0.9243663 | 0.942129 |  |  |  |  |  | 0 |
| 647.14429 | 0.02487663 | 1.3287418 | 1.0693405 | 0.943864 | C34H30N2O9 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 610.195133 | 647.1437396 | 0.85 | ['Atalanine'] |
| 201.02576 | 0.18876585 | 1.0835349 | 1.0919296 | 0.947852 | C8H10N2S | [M+Cl]- | 166.05647 | 201.0258716 | -0.56 | ['Ethionamide'] |
| 416.98428 | 0.8492882 | 1.015241 | 0.9777597 | 0.950534 |  |  |  |  |  | 0 |
| 472.16733 | 0.01149644 | 1.4006218 | 1.1706685 | 0.954364 | C21H31N3O5S | [M+Cl]- | 437.198444 | 472.1678456 | -1.09 | ['FMLP'] |
| 577.14134 | 0.0024369 | 0.8391029 | 1.1019944 | 0.965724 |  |  |  |  |  | 0 |
| 459.17202 | 0.26364783 | 0.9163207 | 0.9501999 | 0.969943 |  |  |  |  |  | 0 |
| 200.0567 | $3.76 \mathrm{E}-10$ | 1.7606944 | 1.2592771 | 0.975589 | C6H7NO3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 141.042594 | 200.0564476 | 1.26 | ['2-Aminomuconate semialdehyde', '6-Oxo-1,4,5,6-tetrahydronicotinate', 'Gentianaine'] |
| 517.14096 | 0.68986816 | 0.9324054 | 1.0185801 | 0.990394 |  |  |  |  |  | 0 |
| 313.11388 | 0.01942606 | 1.2527742 | 0.998339 | 0.994632 | C9H18O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 254.10017 | 313.1140236 | -0.46 | ['2-(beta-D-Glucosyl)-sn-glycerol', '3-beta-D-Galactosyl-sn-glycerol'] |


| 401.13 | 0.10417652 | 1.0856536 | 1.0801216 | 1.001097 | C12H22O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 342.116215 | 401.1300686 | -0.17 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-D-glucose', 'Cellobiose', 'D-Fructosyl-D-fructofuranose', 'D- <br> Glucosyl-D-mannose', 'Epimelibiose', <br> 'Gentiobiose', 'Inulobiose', 'Isomaltose', <br> 'Lactose', 'Lactulose', 'Laminaribiose', <br> 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alpha-Trehalose', 'alpha-Cellobiose', 'alpha-D-Aldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 311.09825 | 9.93E-09 | 1.518312 | 1.0126555 | 1.002915 | C11H20010 | [M-H]- | 312.10565 | 311.0983736 | -0.4 | ['6-O-(beta-D-Xylopyranosyl)-beta-Dglucopyranose', '6-O-beta-D-Xylopyranosyl-Dglucose', 'Arabino-galactose', 'Vicianose'] |
| 311.09825 | 9.93E-09 | 1.518312 | 1.0126555 | 1.002915 | C7H16N8O4 | [M+Cl]- | 276.129452 | 311.0988536 | -1.94 | ['Trimethylenetetraurea'] |
| 297.08266 | 0.03789948 | 1.20141 | 0.9868918 | 1.021982 | C14H16N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 260.116093 | 297.0825446 | 0.39 | ['Maculosin'] |
| 297.08266 | 0.03789948 | 1.20141 | 0.9868918 | 1.021982 | C8H14O8 | [M+Hac-H]- | 238.06887 | 297.0827236 | -0.21 | ['3-Deoxy-D-manno-octulosonate'] |
| 503.16175 | 0.05218903 | 1.283564 | 1.1582157 | 1.02389 | C18H32O16 | [M-H]- | 504.16904 | 503.1617636 | -0.03 | ['1F-beta-D-Fructosylsucrose', '6-alphaMaltosylglucose', '6F-alpha-D- <br> Galactosylsucrose', 'Cellotriose', 'D-Gal alpha 1->6D-Gal alpha 1->6D-Glucose', <br> 'Galactomannan', 'Isomaltotriose', <br> 'Maltotriose', 'Melezitose', 'Panose', <br> 'Raffinose', 'Umbelliferose', 'beta-D- <br> Fructofuranosyl O-beta-D-glucopyranosyl-(1-6)-alpha-D-glucopyranoside'] |
| 266.9397 | 0.61460998 | 1.0504721 | 0.9426151 | 1.038006 |  |  |  |  |  | 0 |
| 199.03795 | 0.64110124 | 0.9557178 | 1.0026336 | 1.040477 | C10H10O3 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 178.062995 | 199.0376636 | 1.44 | ['1,2-Dihydroxy-3,4-epoxy-1,2,3,4tetrahydronaphthalene', '3-Acetyl-6methoxybenzaldehyde', 'Coniferyl aldehyde', 'Vermelone'] |
| 199.03795 | 0.64110124 | 0.9557178 | 1.0026336 | 1.040477 | C3H9O4P | [M+Hac-H]- | 140.023848 | 199.0377016 | 1.25 | ['2-Hydroxypropylphosphonate'] |


| 199.03795 | 0.64110124 | 0.9557178 | 1.0026336 | 1.040477 | C6H12O5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 164.068475 | 199.0378766 | 0.37 | ['(+)-Quercitol', '(-)-Viburnitol', '1,5-Anhydro-Dglucitol', '1,5-Anhydro-D-mannitol', '2-Deoxy-Dgalactose', '2-Deoxy-D-glucose', '6-Deoxy-Dgalactose', '6-Deoxy-D-glucose', '6-Deoxy-Lgalactose', 'D-Rhamnose', 'L-Fuculose', 'LRhamnofuranose', 'L-Rhamnose', 'LRhamnulose', 'alpha-D-Quinovopyranose', 'alpha-L-Rhamnose', 'beta-D-Fucose', 'beta-LRhamnose'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 199.03795 | 0.64110124 | 0.9557178 | 1.0026336 | 1.040477 | C7H14O2S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 162.071452 | 199.0379036 | 0.23 | ['7-Mercaptoheptanoic acid'] |
| 199.03795 | 0.64110124 | 0.9557178 | 1.0026336 | 1.040477 | C7H14O4 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 162.08921 | 199.0378166 | 0.67 | ['beta-Cymaropyranose', 'beta-LOleandropyranose'] |
| 505.17759 | 0.97515814 | 1.0029903 | 0.9963443 | 1.044953 |  |  |  |  |  | 0 |
| 519.15676 | 0.32298745 | 1.0433581 | 1.0920165 | 1.048378 |  |  |  |  |  | 0 |
| 323.06186 | 6.79E-05 | 0.6143976 | 0.8321323 | 1.050635 |  |  |  |  |  | 0 |
| 263.96788 | 0.41478987 | 1.0945957 | 0.9677855 | 1.058849 | C5H10NO7P | [M+K-2H]- | 227.019492 | 263.9680986 | -0.83 | ['L-Glutamyl 1-phosphate', 'L-Glutamyl 5phosphate', 'alpha-D-Glutamyl phosphate'] |
| 621.18837 | 0.08257576 | 0.8644057 | 0.8812629 | 1.061483 |  |  |  |  |  | 0 |
| 306.08298 | 1.72E-07 | 1.5379363 | 1.0634273 | 1.066674 | C15H15N3O2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 269.116427 | 306.0828786 | 0.33 | ['Disperse Yellow 3', 'Methyl red'] |
| 306.08298 | 1.72E-07 | 1.5379363 | 1.0634273 | 1.066674 | C9H13NO7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 247.069204 | 306.0830576 | -0.25 | ['N-Succinyl-L-glutamate'] |
| 341.07243 | 0.0201002 | 1.0374981 | 0.9518966 | 1.077556 |  |  |  |  |  | 0 |
| 487.13046 | 0.01301508 | 0.7701902 | 0.9755343 | 1.080991 |  |  |  |  |  | 0 |
| 341.10879 | $2.64 \mathrm{E}-05$ | 1.071709 | 1.1837597 | 1.101231 | C10H1809 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 282.095085 | 341.1089386 | -0.44 | ['Xylobiose'] |
| 341.10879 | $2.64 \mathrm{E}-05$ | 1.071709 | 1.1837597 | 1.101231 | C12H22011 | [M-H]- | 342.116215 | 341.1089386 | -0.44 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-D-glucose', 'Cellobiose', 'D-Fructosyl-D-fructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', 'Gentiobiose', 'Inulobiose', 'Isomaltose', 'Lactose', 'Lactulose', 'Laminaribiose', 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alpha-Trehalose', 'alpha-Cellobiose', 'alpha-D-Aldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |
| 399.07804 | 2.36E-09 | 2.0735073 | 1.2200164 | 1.118709 | C14H17N2O4PS | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 340.064668 | 399.0785216 | -1.21 | ['Pyridafenthion'] |
| 521.17243 | 0.01721419 | 0.8776961 | 0.9999763 | 1.120026 |  |  |  |  |  | 0 |


| 473.11469 | 0.19656423 | 1.411616 | 1.2397144 | 1.128522 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 477.14613 | 0.27640069 | 0.9638726 | 0.9364469 | 1.13233 |  |  |  |  |  | 0 |
| 599.12333 | $2.44 \mathrm{E}-05$ | 0.8754605 | 1.0083366 | 1.139792 |  |  |  |  |  | 0 |
| 192.02328 | $1.14 \mathrm{E}-08$ | 2.2321375 | 1.6375073 | 1.142043 |  |  |  |  |  | 0 |
| 661.15987 | 0.00147709 | 1.2275803 | 1.145108 | 1.152819 |  |  |  |  |  | 0 |
| 383.08309 | $1.04 \mathrm{E}-09$ | 2.577215 | 1.3772077 | 1.157212 | C17H18N2O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 346.116488 | 383.0829396 | 0.39 | ['Miraxanthin-V', 'Nifedipine'] |
| 647.20385 | 0.00090076 | 0.8307284 | 0.8952926 | 1.176263 | C33H34N6O6 | [M+K-2H]- | 610.253984 | 647.2025906 | 1.95 | ['Candesartan cilexetil'] |
| 345.08261 | 0.00763764 | 0.978572 | 1.0938536 | 1.226003 | C18H16N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 308.116093 | 345.0825446 | 0.19 | ['Citrus Red No.2'] |
| 335.1571 | 0 | 3.9919919 | 2.6130833 | 1.270028 |  |  |  |  |  | 0 |
| 471.13555 | $4.41 \mathrm{E}-05$ | 1.0004019 | 1.1966979 | 1.276521 |  |  |  |  |  | 0 |
| 355.08805 | 6.53E-07 | 1.4448639 | 1.180762 | 1.281174 |  |  |  |  |  | 0 |
| 412.14603 | $4.73 \mathrm{E}-12$ | 2.5661531 | 1.7232769 | 1.302525 | C22H26N2O4S | [M-2H]- | 414.16133 | 412.1467772 | -1.81 | ['Diltiazem'] |
| 591.17792 | $1.06 \mathrm{E}-06$ | 1.0409529 | 1.3408306 | 1.303731 |  |  |  |  |  | 0 |
| 449.15129 | 0.00205325 | 0.9799767 | 1.2029758 | 1.305144 |  |  |  |  |  | 0 |
| 251.11484 | $4.27 \mathrm{E}-08$ | 2.9365328 | 1.6132863 | 1.323062 |  |  |  |  |  | 0 |
| 597.08416 | 0.00031838 | 1.2277685 | 1.2987713 | 1.356039 |  |  |  |  |  | 0 |
| 262.09301 | 0.01034049 | 1.1711377 | 1.0475362 | 1.361226 | C14H15N3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 225.126597 | 262.0930486 | -0.15 | ['Cyprodinil', 'ortho-Aminoazotoluene', 'para(Dimethylamino)azobenzene'] |
| 262.09301 | 0.01034049 | 1.1711377 | 1.0475362 | 1.361226 | C8H13NO5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 203.079374 | 262.0932276 | -0.83 | ['N2-Acetyl-L-aminoadipate'] |
| 221.06693 | 9.78E-09 | 1.0199648 | 1.2554341 | 1.400382 | C12H12N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 184.100048 | 221.0664996 | 1.95 | ['2,4-Diphenyldiamine', 'Benzidine', 'Diquat', 'Withasomnine'] |
| 221.06693 | $9.78 \mathrm{E}-09$ | 1.0199648 | 1.2554341 | 1.400382 | C6H10O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 162.052825 | 221.0666786 | 1.14 | ['(2R,3S)-2,3-Dimethylmalate', '(R)-2Ethylmalate', '(R)-3,3-Dimethylmalate', '(S)-2(Hydroxymethyl)glutarate', '1,5-Anhydro-Dfructose', '2-Dehydro-3-deoxy-D-fuconate', '2-Dehydro-3-deoxy-L-fuconate', '2-Dehydro-3-deoxy-L-rhamnonate', '2-Deoxy-scyllo-inosose', '2-Hydroxyadipate', '3,6-Anhydrogalactose', '3,6-Anhydroglucose', '3-EthyImalate', '3-Hydroxy-3-methylglutarate', 'D-Fucono-1,4lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5-lactone', 'L-Rhamnono-1,4-lactone', 'Lichenin'] |
| 221.06693 | $9.78 \mathrm{E}-09$ | 1.0199648 | 1.2554341 | 1.400382 | C8H14O7 | [M-H]- | 222.073955 | 221.0666786 | 1.14 | ['6-Acetyl-D-glucose'] |
| 152.0352 | $1.64 \mathrm{E}-11$ | 2.0538587 | 1.7262199 | 1.401555 | C7H7NO3 | [M-H]- | 153.042594 | 152.0353176 | -0.77 | ['1-Carbapen-2-em-3-carboxylic acid', '2Nitroanisole', '3-Amino-4-hydroxybenzoic acid', '3-Hydroxy-2-methylpyridine-5-carboxylate', '3Hydroxyanthranilate', '4-Aminosalicylate', '4Nitroanisole', 'AHBA', 'Salicylhydroxamic acid', 'o-Hydroxylaminobenzoate'] |


| 388.06509 | 0.03643281 | 1.4359156 | 1.3769155 | 1.403039 | C11H20NO12P | [M-H]- | 389.072317 | 388.0650406 | 0.13 | ['N-Acetylneuraminate 9-phosphate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 388.06509 | 0.03643281 | 1.4359156 | 1.3769155 | 1.403039 | C13H21NO10 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 351.116549 | 388.0651556 | -0.17 | ['N-Acetyl-4-O-acetylneuraminate', 'N-Acetyl-7-O-acetylneuraminate', 'N-Acetyl-9-OacetyIneuraminate'] |
| 223.10912 | 4.84E-10 | 2.0625306 | 1.1433096 | 1.424848 | C9H12N2O | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] - | 164.094963 | 223.1088166 | 1.36 | ['Fenuron'] |
| 210.07746 | 1.88E-07 | 1.8867485 | 1.7806103 | 1.444953 | C10H13NO4 | [M-H]- | 211.084459 | 210.0771826 | 1.32 | ['Enicoflavine', 'Methyldopa anhydrous'] |
| 210.07746 | $1.88 \mathrm{E}-07$ | 1.8867485 | 1.7806103 | 1.444953 | C8H9NO2 | [M+Hac-H]- | 151.063329 | 210.0771826 | 1.32 | ['(E)-4-Hydroxyphenylacetaldehyde-oxime', '(R)-Mandelamide', '(Z)-4- <br> Hydroxyphenylacetaldehyde-oxime', '2-Amino- <br> 3-methylbenzoate', '2-Descarboxy-cyclo-dopa', <br> 'Acetaminophen', 'Dopamine quinone', 'L- <br> Phenylglycine', 'N-(Acetyloxy)benzenamine', 'N-Methyl-4-aminobenzoate', ' N Methylanthranilate'] |
| 403.14566 | $2.72 \mathrm{E}-08$ | 2.0390648 | 1.3553072 | 1.446327 | C12H24O11 | [M+Hac-H]- | 344.131865 | 403.1457186 | -0.15 | ['Clusianose', 'Melibiitol'] |
| 343.12444 | $9.31 \mathrm{E}-06$ | 1.5187133 | 1.499538 | 1.452153 | C12H24O11 | [M-H]- | 344.131865 | 343.1245886 | -0.43 | ['Clusianose', 'Melibiitol'] |
| 343.12444 | $9.31 \mathrm{E}-06$ | 1.5187133 | 1.499538 | 1.452153 | C20H22N2S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 322.15037 | 343.1250386 | -1.74 | ['Mequitazine'] |
| 197.04576 | 0.0003572 | 1.289618 | 1.4587064 | 1.459957 | C7H14NO3 | [M+K-2H]- | 160.097369 | 197.0459756 | -1.09 | ['3-Dehydrocarnitine'] |
| 197.04576 | 0.0003572 | 1.289618 | 1.4587064 | 1.459957 | C7H6O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 138.031695 | 197.0455486 | 1.07 | ['2-Hydroxy-5-methylquinone', '3,4Dihydroxybenzaldehyde', '3-Hydroxybenzoate', '4-Hydroxybenzoate', 'Gentisate aldehyde', 'Salicylate', 'Sesamol'] |
| 197.04576 | 0.0003572 | 1.289618 | 1.4587064 | 1.459957 | C9H10O5 | [M-H]- | 198.052825 | 197.0455486 | 1.07 | ['3-(3,4-Dihydroxyphenyl)lactate', '3-Methoxy- <br> 4-hydroxymandelate', 'Syringic acid'] |
| 446.31246 | 0.57723831 | 1.3973134 | 2.0250206 | 1.466737 |  |  |  |  |  | 0 |
| 265.09267 | 5.96E-05 | 1.1027568 | 1.2751198 | 1.629895 |  |  |  |  |  | 0 |
| 443.17721 | 2.20E-07 | 1.4212885 | 1.649314 | 1.635068 | C17H26N8O5 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 422.202617 | 443.1772856 | -0.17 | ['Blasticidin S'] |
| 463.14555 | 3.87E-07 | 1.2273393 | 1.3925406 | 1.640055 | C17H24O11 | [M+Hac-H]- | 404.131865 | 463.1457186 | -0.36 | ['Gardenoside', 'Scandoside methyl ester'] |
| 438.16186 | 3.94E-10 | 3.6756697 | 2.7180226 | 1.653591 |  |  |  |  |  | 0 |
| 371.11945 | 3.03E-07 | 1.4700412 | 1.6568416 | 1.699604 | C11H20010 | [M+Hac-H]- | 312.10565 | 371.1195036 | -0.14 | ['6-O-(beta-D-Xylopyranosyl)-beta-Dglucopyranose', '6-O-beta-D-Xylopyranosyl-D- <br> glucose', 'Arabino-galactose', 'Vicianose'] |
| 513.12293 | 4.62E-06 | 1.0744056 | 1.6170062 | 1.719623 |  |  |  |  |  | 0 |
| 378.10416 | 3.00E-15 | 9.1166602 | 2.7338447 | 1.724065 | C14H21NO11 | [M-H]- | 379.111464 | 378.1041876 | -0.07 | ['3-(4-Deoxy-beta-D-gluc-4-enuronosyl)-N-acetyl-D-glucosamine', '4-Deoxy-beta-D-gluc-4-enuronosyl-(1,3)-N-acetyl-D-galactosamine'] |
| 382.13544 | $2.04 \mathrm{E}-08$ | 1.773105 | 1.6014548 | 1.728624 | C14H25NO11 | [M-H]- | 383.142764 | 382.1354876 | -0.12 | ['Lacto-N-biose', 'N-Acetyllactosamine', 'beta-D-Galactosyl-(1->3)-N-acetyl-D-galactosamine'] |
| 382.13544 | $2.04 \mathrm{E}-08$ | 1.773105 | 1.6014548 | 1.728624 | C18H23N3O4 | [M+(37Cl)]- | 345.168857 | 382.1353086 | 0.34 | ['Pefurazoate'] |


| 147.06619 | 0.0029495 | 1.4189799 | 1.6491337 | 1.729317 | C4H8O2 | [ $\mathrm{M}+\mathrm{Hac-H}$ ]- | 88.05243 | 147.0662836 | -0.64 | ['(R)-Acetoin', '1,4-Dioxane', '2- <br> Methylpropanoate', 'Acetoin', 'Butanoic acid', 'Ethyl acetate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 147.06619 | 0.0029495 | 1.4189799 | 1.6491337 | 1.729317 | C6H12O4 | [M-H]- | 148.07356 | 147.0662836 | -0.64 | ['(R)-2,3-Dihydroxy-3-methylpentanoate', '(R)Mevalonate', '(R)-Pantoate', '(S)-Mevalonate', '2,3-Dihydroxy-3-methylpentanoate', '3,6-Dideoxy-L-galactose', 'Abequose'] |
| 531.19307 | $2.73 \mathrm{E}-07$ | 1.3096052 | 1.6127909 | 1.732604 |  |  |  |  |  | 0 |
| 267.10832 | 0.10758083 | 1.3488933 | 1.5092538 | 1.773126 | C14H18N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 230.141913 | 267.1083646 | -0.17 | ['Camoensine'] |
| 212.05673 | $5.60 \mathrm{E}-11$ | 1.8444209 | 1.8460338 | 1.796894 | C7H7NO3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 153.042594 | 212.0564476 | 1.33 | ['1-Carbapen-2-em-3-carboxylic acid', '2Nitroanisole', '3-Amino-4-hydroxybenzoic acid', '3-Hydroxy-2-methylpyridine-5-carboxylate', '3Hydroxyanthranilate', '4-Aminosalicylate', '4Nitroanisole', 'AHBA', 'Salicylhydroxamic acid', 'o-Hydroxylaminobenzoate'] |
| 212.05673 | 5.60E-11 | 1.8444209 | 1.8460338 | 1.796894 | C9H11NO5 | [M-H]- | 213.063724 | 212.0564476 | 1.33 | ['N,N-Dihydroxy-L-tyrosine'] |
| 307.06697 | $4.83 \mathrm{E}-08$ | 2.1565645 | 1.7126419 | 1.800215 | C12H16NO7 | [M+Na-2H]- | 286.092679 | 307.0673476 | -1.23 | ['N-Glucosylnicotinate'] |
| 307.06697 | $4.83 \mathrm{E}-08$ | 2.1565645 | 1.7126419 | 1.800215 | C15H14N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 270.100443 | 307.0668946 | 0.25 | ['Dihydroxycarbazepine'] |
| 669.1052 | 0.0009287 | 1.6873112 | 1.555145 | 1.804639 |  |  |  |  |  | 0 |
| 309.08264 | 3.74E-07 | 2.3785596 | 2.0986587 | 1.823537 |  |  |  |  |  | 0 |
| 525.0864 | $2.32 \mathrm{E}-10$ | 1.0460035 | 1.6517741 | 1.825637 |  |  |  |  |  | 0 |
| 577.19863 | $4.00 \mathrm{E}-08$ | 1.3968763 | 1.7670732 | 1.837293 |  |  |  |  |  | 0 |
| 323.09824 | $2.51 \mathrm{E}-06$ | 1.4043843 | 1.6783233 | 1.851573 | C12H20O10 | [M-H]- | 324.10565 | 323.0983736 | -0.41 | ["Bis-D-fructose 2',1:2,1-dianhydride", "DFructofuranose 1,2':2,3-dianhydride"] |
| 323.09824 | $2.51 \mathrm{E}-06$ | 1.4043843 | 1.6783233 | 1.851573 | C16H18N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 286.131743 | 323.0981946 | 0.14 | ['Cromakalim', 'Difenoxuron', 'Levcromakalim', 'Pilosine'] |
| 323.09824 | $2.51 \mathrm{E}-06$ | 1.4043843 | 1.6783233 | 1.851573 | C8H24N4O3P2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 286.132367 | 323.0988186 | -1.79 | ['Schradan'] |
| 407.05968 | $1.78 \mathrm{E}-08$ | 1.7074064 | 1.8490362 | 1.858836 | C16H18O11 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 386.084915 | 407.0595836 | 0.24 | ['O-Feruloylgalactarate'] |
| 407.05968 | $1.78 \mathrm{E}-08$ | 1.7074064 | 1.8490362 | 1.858836 | C9H17O12P | [M+Hac-H]- | 348.045768 | 407.0596216 | 0.14 | ['2-(alpha-D-Mannosyl)-3-phosphoglycerate', '2-O-(6-Phospho-alpha-mannosyl)-D-glycerate'] |
| 212.09313 | 0.00029309 | 1.8727739 | 1.880618 | 1.865334 | C10H15NO4 | [M-H]- | 213.100109 | 212.0928326 | 1.4 | ['Kainic acid', 'N-(3-Oxohexanoyl)homoserine lactone', 'alpha-Allokainic acid'] |
| 212.09313 | 0.00029309 | 1.8727739 | 1.880618 | 1.865334 | C8H11NO2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 153.078979 | 212.0928326 | 1.4 | ['1-(4-Hydroxyphenyl)-2-aminoethanol', 'Dopamine', 'Vanillylamine'] |
| 269.12529 | 7.98E-09 | 2.4763026 | 2.13995 | 1.874463 |  |  |  |  |  | 0 |
| 677.21465 | $4.61 \mathrm{E}-10$ | 1.2683692 | 1.8861452 | 1.890582 |  |  |  |  |  | 0 |
| 247.04577 | 0.00013765 | 1.5782354 | 1.4470844 | 1.90651 | C13H10N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 210.079313 | 247.0457646 | 0.02 | ['2-Aminoacridone'] |


| 247.04577 | 0.00013765 | 1.5782354 | 1.4470844 | 1.90651 | C7H8O6 | [M+Hac-H]- | 188.03209 | 247.0459436 | -0.7 | ['(E)-2-(Methoxycarbonylmethyl)butenedioate', <br> '(E)-3-(Methoxycarbonyl)pent-2-enedioate', <br> '(Z)-But-1-ene-1,2,4-tricarboxylate', '(Z)-But-2-ene-1,2,3-tricarboxylate', 'transHomoaconitate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 151.06109 | $3.03 \mathrm{E}-07$ | 1.3707113 | 1.9608998 | 1.91479 | C3H8O3 | [M+Hac-H]- | 92.047345 | 151.0611986 | -0.72 | ['Glycerol'] |
| 151.06109 | 3.03E-07 | 1.3707113 | 1.9608998 | 1.91479 | C5H12O5 | [M-H]- | 152.068475 | 151.0611986 | -0.72 | ['D-Apiitol', 'L-Arabitol', 'Ribitol', 'Xylitol'] |
| 333.02285 | $1.97 \mathrm{E}-10$ | 1.0893804 | 1.6382619 | 1.959041 | C12H15N2O3PS | [ $\mathrm{M}+\mathrm{Cl}]$ - | 298.054103 | 333.0235046 | -1.97 | ['Phoxim', 'Quinalphos'] |
| 333.02285 | $1.97 \mathrm{E}-10$ | 1.0893804 | 1.6382619 | 1.959041 | C6H11010P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 274.008988 | 333.0228416 | 0.03 | ['1-Phospho-alpha-D-galacturonate', '3-Dehydro-L-gulonate 6-phosphate', '6-Phospho-2-dehydro-D-gluconate', 'D-Glucuronate 1phosphate'] |
| 395.05971 | 4.92E-06 | 1.7397759 | 1.6814715 | 1.969053 | C14H19N2O7P | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 358.092991 | 395.0594426 | 0.68 | ['N1-(5-Phospho-alpha-D-ribosyl)-5,6dimethylbenzimidazole'] |
| 561.16717 | 7.93E-06 | 1.0262043 | 1.4743198 | 1.987671 | C18H30O16 | [M+Hac-H]- | 502.15339 | 561.1672436 | -0.13 | ['alpha-L-Rhamnopyranosyl-(1->2)-beta-D-galactopyranosyl-(1->2)-beta-Dglucuronopyranoside'] |
| 207.08768 | 3.31E-09 | 1.2723167 | 2.0001001 | 2.004061 | C6H12O4 | [M+Hac-H]- | 148.07356 | 207.0874136 | 1.29 | ['(R)-2,3-Dihydroxy-3-methylpentanoate', '(R)Mevalonate', '(R)-Pantoate', '(S)-Mevalonate', '2,3-Dihydroxy-3-methylpentanoate', '3,6-Dideoxy-L-galactose', 'Abequose'] |
| 481.09658 | 0.00158748 | 1.8948495 | 2.2241491 | 2.015567 | C12H23O14P | [M+Hac-H]- | 422.082548 | 481.0964016 | 0.37 | ['6-Phospho-beta-D-glucosyl-(1,4)-D-glucose', 'Lactose 6-phosphate', 'Maltose 6-phosphate', 'Sucrose 6-phosphate', 'alpha,alpha-Trehalose 6-phosphate', 'beta-D-Fructofuranosyl-alpha-Dmannopyranoside 6F-phosphate'] |
| 267.0719 | $1.29 \mathrm{E}-08$ | 2.0563167 | 1.908346 | 2.01582 | C9H16O9 | [M-H]- | 268.079435 | 267.0721586 | -0.97 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 203.05641 | $1.84 \mathrm{E}-10$ | 1.8362368 | 2.172464 | 2.033347 | C6H8O4 | [M+Hac-H]- | 144.04226 | 203.0561136 | 1.46 | ['2,3-Dimethylmaleate', '2Methyleneglutarate', 'Methylitaconate', 'Triacetate'] |
| 203.05641 | $1.84 \mathrm{E}-10$ | 1.8362368 | 2.172464 | 2.033347 | C8H12O6 | [M-H]- | 204.06339 | 203.0561136 | 1.46 | ['3-o-Ethyl-L-ascorbic acid'] |
| 413.09384 | 0.00023455 | 2.853221 | 2.4425751 | 2.067906 |  |  |  |  |  | 0 |
| 268.07526 | 4.03E-09 | 2.1858686 | 1.9932921 | 2.075941 |  |  |  |  |  | 0 |
| 272.07734 | 4.09E-06 | 2.4971553 | 2.5479332 | 2.116802 | C12H15N2O4 | [M+Na-2H]- | 251.103183 | 272.0778516 | -1.88 | ['3-Oxohexobarbital'] |
| 272.07734 | $4.09 \mathrm{E}-06$ | 2.4971553 | 2.5479332 | 2.116802 | C9H11NO5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 213.063724 | 272.0775776 | -0.87 | ['N,N-Dihydroxy-L-tyrosine'] |
| 549.16732 | $3.69 \mathrm{E}-08$ | 2.0681477 | 2.1367664 | 2.129861 |  |  |  |  |  | 0 |


| 541.11776 | $1.41 \mathrm{E}-09$ | 1.5353098 | 1.9941368 | 2.146672 | C18H32O16 | [M+K-2H]- | 504.16904 | 541.1176466 | 0.21 | ['1F-beta-D-Fructosylsucrose', '6-alphaMaltosylglucose', '6F-alpha-D- <br> Galactosylsucrose', 'Cellotriose', 'D-Gal alpha 1->6D-Gal alpha 1->6D-Glucose', <br> 'Galactomannan', 'Isomaltotriose', <br> 'Maltotriose', 'Melezitose', 'Panose', <br> 'Raffinose', 'Umbelliferose', 'beta-D- <br> Fructofuranosyl O-beta-D-glucopyranosyl-(1-6)-alpha-D-glucopyranoside'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 515.1619 | $1.35 \mathrm{E}-09$ | 1.133866 | 1.8141361 | 2.218174 |  |  |  |  |  | 0 |
| 387.15071 | 0.00068105 | 1.7513409 | 1.8472211 | 2.251688 |  |  |  |  |  | 0 |
| 277.05629 | $5.56 \mathrm{E}-07$ | 1.8160409 | 2.2337566 | 2.263406 | C11H14NO6 | [M+Na-2H]- | 256.082114 | 277.0567826 | -1.78 | ['Nicotinate D-ribonucleoside'] |
| 154.01447 | 7.61E-10 | 1.3246129 | 1.8501949 | 2.332328 | C6H5NO4 | [M-H]- | 155.021859 | 154.0145826 | -0.73 | ['2,6-Dihydroxynicotinate', '4-Nitrocatechol'] |
| 208.09825 | $1.70 \mathrm{E}-10$ | 3.2286185 | 2.8991357 | 2.33911 | C11H15NO3 | [M-H]- | 209.105194 | 208.0979176 | 1.6 | ['Anhalamine', 'Propoxur', 'Tyr-OEt', 'pLactophenetide'] |
| 208.09825 | $1.70 \mathrm{E}-10$ | 3.2286185 | 2.8991357 | 2.33911 | C9H11NO | [M+Hac-H]- | 149.084064 | 208.0979176 | 1.6 | ['3-Phenylpropionaldoxim', 'D-Cathinone'] |
| 533.17238 | $7.48 \mathrm{E}-09$ | 1.8621381 | 2.3805793 | 2.391576 |  |  |  |  |  | 0 |
| 511.10736 | $2.99 \mathrm{E}-10$ | 1.7846116 | 2.2804387 | 2.4165 |  |  |  |  |  | 0 |
| 213.02579 | $5.44 \mathrm{E}-10$ | 1.4608246 | 2.3079008 | 2.435811 | C4H10O2S2 | [M+Hac-H]- | 154.012224 | 213.0260776 | -1.35 | ['Dithioerythritol', 'Dithiothreitol'] |
| 429.1249 | $1.64 \mathrm{E}-11$ | 1.5582525 | 1.9730319 | 2.439144 |  |  |  |  |  | 0 |
| 441.06537 | $2.77 \mathrm{E}-08$ | 1.393014 | 2.0746617 | 2.440903 |  |  |  |  |  | 0 |
| 400.99776 | $9.15 \mathrm{E}-13$ | 2.2376289 | 2.5501557 | 2.445803 |  |  |  |  |  | 0 |
| 349.05409 | $1.48 \mathrm{E}-07$ | 2.8888772 | 2.2251783 | 2.451883 | C11H20O10 | [M+K-2H]- | 312.10565 | 349.0542566 | -0.48 | ['6-O-(beta-D-Xylopyranosyl)-beta-D- <br> glucopyranose', '6-O-beta-D-Xylopyranosyl-D- <br> glucose', 'Arabino-galactose', 'Vicianose'] |
| 349.05409 | $1.48 \mathrm{E}-07$ | 2.8888772 | 2.2251783 | 2.451883 | C14H16O9 | [M+Na-2H]- | 328.079435 | 349.0541036 | -0.04 | ['2-Succinyl-5-enolpyruvyl-6-hydroxy-3-cyclohexene-1-carboxylate', 'Bergenin'] |
| 349.05409 | $1.48 \mathrm{E}-07$ | 2.8888772 | 2.2251783 | 2.451883 | C7H15O10P | [M+Hac-H]- | 290.040288 | 349.0541416 | -0.15 | ['D-glycero-D-manno-Heptose 1-phosphate', 'D-glycero-D-manno-Heptose 7-phosphate', 'Sedoheptulose 1-phosphate', 'Sedoheptulose 7-phosphate'] |
| 534.17616 | 9.04E-09 | 1.996231 | 2.5228577 | 2.488739 |  |  |  |  |  | 0 |
| 519.05264 | $1.44 \mathrm{E}-05$ | 1.6437678 | 2.550501 | 2.517986 |  |  |  |  |  | 0 |
| 517.17751 | 3.93E-09 | 1.4911424 | 2.2928153 | 2.546192 |  |  |  |  |  | 0 |
| 336.0935 | $7.75 \mathrm{E}-10$ | 2.3136231 | 2.3302293 | 2.546633 |  |  |  |  |  | 0 |
| 467.14068 | 8.85E-10 | 1.0857067 | 2.1548589 | 2.574758 |  |  |  |  |  | 0 |
| 468.17233 | 8.37E-09 | 2.0344133 | 2.6607498 | 2.625511 | C16H27NO11 | [M+Hac-H]- | 409.158414 | 468.1722676 | 0.13 | ['Linustatin'] |
| 328.08848 | 0.00333953 | 0.7314206 | 3.2645005 | 2.625638 |  |  |  |  |  | 0 |
| 543.1568 | $2.70 \mathrm{E}-08$ | 0.9632301 | 1.7431449 | 2.666492 |  |  |  |  |  | 0 |


| 214.07238 | $1.43 \mathrm{E}-10$ | 2.0024729 | 2.5473006 | 2.685338 | C7H9NO3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 155.058244 | 214.0720976 | 1.32 | ['(3S,5S)-Carbapenam-3-carboxylic acid'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 214.07238 | $1.43 \mathrm{E}-10$ | 2.0024729 | 2.5473006 | 2.685338 | C9H13NO5 | [M-H]- | 215.079374 | 214.0720976 | 1.32 | ['Succinylproline'] |
| 485.15129 | 2.34E-09 | 3.3119098 | 3.0212735 | 2.692304 |  |  |  |  |  | 0 |
| 392.11985 | $1.80 \mathrm{E}-07$ | 2.6529722 | 2.7752758 | 2.777083 |  |  |  |  |  | 0 |
| 541.17748 | 3.18E-09 | 1.9802531 | 3.5750077 | 2.884382 |  |  |  |  |  | 0 |
| 457.11992 | $1.93 \mathrm{E}-09$ | 2.2188947 | 2.9631352 | 2.896844 | C20H24N2O8 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 420.153268 | 457.1197196 | 0.44 | ['2-N,6-N-Bis(2,3-dihydroxy-N-benzoyl)-Lserine'] |
| 467.08096 | $9.33 \mathrm{E}-11$ | 2.3324016 | 2.7318179 | 2.899764 |  |  |  |  |  | 0 |
| 351.09314 | $1.58 \mathrm{E}-06$ | 1.9982095 | 2.0825511 | 2.915146 |  |  |  |  |  | 0 |
| 253.09278 | $1.12 \mathrm{E}-05$ | 2.564232 | 3.3736173 | 2.927675 | C13H16N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 216.126263 | 253.0927146 | 0.26 | ['Girgensonine', 'Tetrahydroharmine'] |
| 253.09278 | $1.12 \mathrm{E}-05$ | 2.564232 | 3.3736173 | 2.927675 | C7H14O6 | [M+Hac-H]- | 194.07904 | 253.0928936 | -0.45 | ['(-)-Quebrachitol', '1-O-Methyl-myo-inositol', '3-O-Methyl-myo-inositol', '4-O-Methyl-myoinositol', '5-O-Methyl-myo-inositol', '6-0-Methyl-myo-inositol', 'D-Pinitol', 'Methyl beta-D-galactoside', 'O-Methyl-scyllo-inositol'] |
| 253.09278 | $1.12 \mathrm{E}-05$ | 2.564232 | 3.3736173 | 2.927675 | C9H18O8 | [M-H]- | 254.10017 | 253.0928936 | -0.45 | ['2-(beta-D-Glucosyl)-sn-glycerol', '3-beta-D-Galactosyl-sn-glycerol'] |
| 453.06527 | 3.66E-12 | 1.7743048 | 2.6935394 | 2.948445 |  |  |  |  |  | 0 |
| 305.02793 | $2.36 \mathrm{E}-09$ | 2.7283129 | 3.1057682 | 2.948923 | C9H16O9 | [M+K-2H]- | 268.079435 | 305.0280416 | -0.37 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 535.04747 | $1.06 \mathrm{E}-09$ | 1.9830727 | 2.7997677 | 3.021548 |  |  |  |  |  | 0 |
| 340.09648 | $1.04 \mathrm{E}-09$ | 2.5537685 | 2.6293846 | 3.028684 | C17H15N5O | [M+Cl]- | 305.12766 | 340.0970616 | -1.71 | ['Zaleplon'] |
| 339.09311 | $7.69 \mathrm{E}-11$ | 2.6889599 | 2.7452994 | 3.102891 | C12H20O11 | [M-H]- | 340.100565 | 339.0932886 | -0.53 | ['3-Ketolactose', '3-Ketosucrose', 'Cellobiono-1,5-lactone'] |
| 531.1567 | 7.30E-10 | 2.2380584 | 2.9356673 | 3.104888 |  |  |  |  |  | 0 |
| 527.16189 | $1.52 \mathrm{E}-06$ | 1.8217221 | 2.5630108 | 3.168129 |  |  |  |  |  | 0 |
| 336.12988 | $2.09 \mathrm{E}-05$ | 2.4797513 | 2.6852022 | 3.206049 | C13H23NO9 | [M-H]- | 337.137284 | 336.1300076 | -0.38 | ['Streptobiosamine'] |
| 513.1463 | $1.94 \mathrm{E}-10$ | 0.8874944 | 2.3866428 | 3.221929 |  |  |  |  |  | 0 |
| 293.08754 | $3.21 \mathrm{E}-13$ | 2.6808233 | 3.3887599 | 3.250388 | C11H18O9 | [M-H]- | 294.095085 | 293.0878086 | -0.92 | ['Tuliposide B'] |
| 293.08754 | $3.21 \mathrm{E}-13$ | 2.6808233 | 3.3887599 | 3.250388 | C15H16N2O2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 256.121178 | 293.0876296 | -0.31 | ['Ancymidol'] |
| 293.08754 | 3.21E-13 | 2.6808233 | 3.3887599 | 3.250388 | C9H14O7 | [M+Hac-H]- | 234.073955 | 293.0878086 | -0.92 | ['(R)-(Homo)3-citrate', '1-Hydroxyhexane-1,2,6tricarboxylate'] |
| 454.15669 | 1.96E-09 | 3.0728973 | 3.5000202 | 3.263628 | C21H27N3O6 | [M+(37Cl)]- | 417.189987 | 454.1564386 | 0.55 | ['Casimiroedine'] |
| 445.15642 | $2.66 \mathrm{E}-10$ | 1.8409979 | 3.013828 | 3.275494 | C2OH3ON2O5S | [M+Cl]- | 410.187545 | 445.1569466 | -1.18 | ['Benfuracarb'] |
| 223.04466 | $1.14 \mathrm{E}-06$ | 1.2389683 | 3.0135906 | 3.308369 |  |  |  |  |  | 0 |
| 385.13516 | $2.32 \mathrm{E}-10$ | 3.5944554 | 3.6322963 | 3.324913 | C12H22O10 | [M+Hac-H]- | 326.1213 | 385.1351536 | 0.02 | ['2-O-alpha-L-Rhamnopyranosyl-Dglucopyranose', 'Robinobiose', 'Rutinose'] |
| 385.13516 | $2.32 \mathrm{E}-10$ | 3.5944554 | 3.6322963 | 3.324913 | C18H24N2O5 | [M+(37Cl)]- | 348.168523 | 385.1349746 | 0.48 | ['Enalaprilate', 'Funebrine'] |
| 384.12285 | $1.07 \mathrm{E}-07$ | 1.5472581 | 2.7685024 | 3.330142 |  |  |  |  |  | 0 |
| 547.18796 | $1.14 \mathrm{E}-07$ | 1.8781706 | 2.8428449 | 3.352564 |  |  |  |  |  | 0 |


| 475.03454 | $1.22 \mathrm{E}-07$ | 3.1271099 | 3.7228951 | 3.491515 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 347.0469 | $1.40 \mathrm{E}-13$ | 2.3468745 | 3.5464483 | 3.578353 | C13H16N2O5S | [M+Cl]- | 312.077995 | 347.0473966 | -1.43 | ['Epithienamycin B', 'Epithienamycin D'] |
| 452.14103 | $9.52 \mathrm{E}-11$ | 3.8248025 | 4.1553498 | 3.610093 |  |  |  |  |  | 0 |
| 415.14568 | $2.62 \mathrm{E}-10$ | 2.1150372 | 3.1898242 | 3.682884 |  |  |  |  |  | 0 |
| 335.0385 | $1.22 \mathrm{E}-11$ | 3.0827858 | 3.4219048 | 3.757381 | C6H13010P | [M+Hac-H]- | 276.024638 | 335.0384916 | 0.03 | ['2-Carboxy-D-arabinitol 1-phosphate', '6-Phospho-D-gluconate'] |
| 293.05118 | $1.20 \mathrm{E}-11$ | 4.2632068 | 4.0511461 | 3.787062 |  |  |  |  |  | 0 |
| 461.15116 | $1.39 \mathrm{E}-08$ | 2.47009 | 3.1928509 | 3.797204 |  |  |  |  |  | 0 |
| 191.0197 | $6.69 \mathrm{E}-12$ | 10.989506 | 6.4675903 | 3.799667 | C4H4O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 132.005875 | 191.0197286 | -0.15 | ['2-Hydroxyethylenedicarboxylate', 'Oxaloacetate', 'trans-2,3-Epoxysuccinate'] |
| 191.0197 | $6.69 \mathrm{E}-12$ | 10.989506 | 6.4675903 | 3.799667 | C6H8O7 | [M-H]- | 192.027005 | 191.0197286 | -0.15 | ['(1R,2S)-1-Hydroxypropane-1,2,3tricarboxylate', '(1S,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(4R,5S)-4,5,6-Trihydroxy- <br> 2,3-dioxohexanoate', '2,5-Didehydro-Dgluconate', '2-Dehydro-3-deoxy-D-glucarate', '5-Dehydro-4-deoxy-D-glucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] |
| 474.15463 | 1.30E-10 | 1.4102167 | 3.002475 | 3.885623 |  |  |  |  |  | 0 |
| 447.33288 | 0.00027997 | 3.1878961 | 5.3802663 | 3.895351 |  |  |  |  |  | 0 |
| 400.98934 | $1.08 \mathrm{E}-11$ | 3.9512866 | 4.1585047 | 3.919364 |  |  |  |  |  | 0 |
| 397.09874 | $4.60 \mathrm{E}-12$ | 1.973103 | 2.8698226 | 3.932107 | C12H18O11 | [M+Hac-H]- | 338.084915 | 397.0987686 | -0.07 | ['L-Ascorbic acid-2-glucoside'] |
| 397.09874 | $4.60 \mathrm{E}-12$ | 1.973103 | 2.8698226 | 3.932107 | C18H2ON2O6 | [M+(37Cl)]- | 360.132138 | 397.0985896 | 0.38 | ['3-Methoxytyramine-betaxanthin', 'Nitrendipine'] |
| 397.09874 | $4.60 \mathrm{E}-12$ | 1.973103 | 2.8698226 | 3.932107 | C19H24N2O3S | [M+K-2H]- | 360.150765 | 397.0993716 | -1.59 | ['LY395153'] |
| 373.13511 | 5.04E-11 | 2.7544142 | 3.4958803 | 3.961131 |  |  |  |  |  | 0 |
| 545.17245 | $2.77 \mathrm{E}-10$ | 1.227723 | 2.8300321 | 3.97262 |  |  |  |  |  | 0 |
| 487.1669 | $1.24 \mathrm{E}-09$ | 2.0400498 | 3.7584996 | 4.078818 |  |  |  |  |  | 0 |
| 443.14075 | 7.07E-11 | 2.2849398 | 3.9855075 | 4.129649 | C14H24O12 | [M+Hac-H]- | 384.12678 | 443.1406336 | 0.26 | ['Acetyl-maltose'] |
| 443.14075 | 7.07E-11 | 2.2849398 | 3.9855075 | 4.129649 | C20H26N2O7 | [M+(37Cl)]- | 406.174003 | 443.1404546 | 0.67 | ['Myxochlin A'] |
| 443.14075 | $7.07 \mathrm{E}-11$ | 2.2849398 | 3.9855075 | 4.129649 | C20H28N2O5S | [M+Cl]- | 408.171895 | 443.1412966 | -1.23 | ['Tamsulosin'] |
| 473.15117 | $4.32 \mathrm{E}-11$ | 1.5520364 | 3.1689033 | 4.159902 |  |  |  |  |  | 0 |
| 539.10222 | 7.93E-10 | 2.8657084 | 4.0723665 | 4.202114 | C18H30016 | [M+K-2H]- | 502.15339 | 539.1019966 | 0.41 | ['alpha-L-Rhamnopyranosyl-(1->2)-beta-D-galactopyranosyl-(1->2)-beta-Dglucuronopyranoside'] |
| 319.04353 | $2.72 \mathrm{E}-12$ | 3.658652 | 4.1866656 | 4.218418 | C10H18O9 | [M+K-2H]- | 282.095085 | 319.0436916 | -0.51 | ['Xylobiose'] |
| 319.04353 | $2.72 \mathrm{E}-12$ | 3.658652 | 4.1866656 | 4.218418 | C13H14O8 | [M+Na-2H]- | 298.06887 | 319.0435386 | -0.03 | ['D-Prephenyllactate'] |


| 319.04353 | $2.72 \mathrm{E}-12$ | 3.658652 | 4.1866656 | 4.218418 | C6H13O9P | [M+Hac-H]- | 260.029723 | 319.0435766 | -0.15 | ['(3S,4R)-Ketose 1-phosphate', '1D-myo-Inositol 3-phosphate', 'Aldohexose 6-phosphate', 'DAllose 6-phosphate', 'D-Allulose 6-phosphate', <br> 'D-Fructose 1-phosphate', 'D-Fructose 6phosphate', 'D-Galactose 1-phosphate', 'D- <br> Galactose 6-phosphate', 'D-Glucose 1phosphate', 'D-Glucose 6-phosphate', 'DHamamelose 2(1)-phosphate', 'D-Hexose 6phosphate', 'D-Mannose 1-phosphate', 'DMannose 6-phosphate', 'D-Tagatose 6phosphate', 'D-arabino-Hex-3-ulose 6phosphate', 'Fructose 1-phosphate(pyranose)', 'Hexose 1-phosphate', 'Inositol 1-phosphate', <br> 'L-Galactose 1-phosphate', 'L-Gulose 1phosphate', 'L-Tagatose 6-phosphate', 'L-myoInositol 5-phosphate', 'Sorbose 1-phosphate', 'alpha-D-Galactose 1-phosphate', 'alpha-DGlucose 3-phosphate', 'alpha-D-Glucose 6phosphate', 'alpha-D-Hexose 1-phosphate', 'alpha-D-Hexose 6-phosphate', 'beta-DFructose 2-phosphate', 'beta-D-Fructose 6phosphate', 'beta-D-Glucose 1-phosphate', 'beta-D-Glucose 6-phosphate', 'myo-Inositol 4phosphate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 469.15626 | $1.81 \mathrm{E}-11$ | 2.4123072 | 3.8251926 | 4.224103 |  |  |  |  |  | 0 |
| 224.0496 | $1.89 \mathrm{E}-11$ | 2.0733941 | 3.4173083 | 4.24793 |  |  |  |  |  | 0 |
| 509.09173 | $8.48 \mathrm{E}-13$ | 1.5762087 | 3.7008657 | 4.265369 |  |  |  |  |  | 0 |
| 529.14108 | $2.20 \mathrm{E}-07$ | 2.1867706 | 3.4903216 | 4.31744 |  |  |  |  |  | 0 |
| 205.07207 | 7.78E-08 | 3.3003648 | 4.4402087 | 4.32231 | C6H1004 | [M+Hac-H]- | 146.05791 | 205.0717636 | 1.49 | ['(R)-3-Hydroxy-3-methyl-2-oxopentanoate', <br> '(R)-4-Dehydropantoate', '(S)-2-Aceto-2- <br> hydroxybutanoate', '2-Aceto-2- <br> hydroxybutanoate', '2-Dehydropantoate', '3- <br> Hydroxy-3-methyl-2-oxopentanoic acid', '3- <br> Hydroxy-5-oxohexanoate', '4-Hydroxy-2- <br> oxohexanoic acid', 'Adipate', 'Mevaldate'] |
| 462.15483 | $9.42 \mathrm{E}-08$ | 2.583375 | 3.3275961 | 4.325948 |  |  |  |  |  | 0 |
| 589.19867 | 3.03E-08 | 1.8103313 | 3.3728601 | 4.456761 |  |  |  |  |  | 0 |
| 413.13 | 5.27E-11 | 4.2175306 | 4.4788266 | 4.471917 |  |  |  |  |  | 0 |
| 483.11216 | 2.80E-07 | 3.2598389 | 4.0644544 | 4.481632 |  |  |  |  |  | 0 |
| 339.12949 | $2.55 \mathrm{E}-06$ | 2.4059455 | 3.9326141 | 4.494842 | C13H24O10 | [M-H]- | 340.13695 | 339.1296736 | -0.54 | ['Methyl-2-alpha-L-fucopyranosyl-beta-Dgalactoside'] |


| 441.12505 | 3.57E-10 | 2.6334666 | 4.0778814 | 4.532993 | C20H24N2O7 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 404.158353 | 441.1248046 | 0.56 | ['Myxochelin A'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 421.07528 | $1.04 \mathrm{E}-11$ | 2.1936123 | 4.0857677 | 4.533534 | C12H23O14P | [M-H]- | 422.082548 | 421.0752716 | 0.02 | ['6-Phospho-beta-D-glucosyl-(1,4)-D-glucose', 'Lactose 6-phosphate', 'Maltose 6-phosphate', 'Sucrose 6-phosphate', 'alpha,alpha-Trehalose 6-phosphate', 'beta-D-Fructofuranosyl-alpha-Dmannopyranoside 6F-phosphate'] |
| 421.07528 | $1.04 \mathrm{E}-11$ | 2.1936123 | 4.0857677 | 4.533534 | C14H24O12 | [M+K-2H]- | 384.12678 | 421.0753866 | -0.25 | ['Acetyl-maltose'] |
| 415.10928 | $1.18 \mathrm{E}-10$ | 2.7089319 | 3.6646195 | 4.643629 |  |  |  |  |  | 0 |
| 353.10879 | $1.33 \mathrm{E}-10$ | 2.6055777 | 4.2951352 | 4.66696 | C11H1809 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 294.095085 | 353.1089386 | -0.42 | ['Tuliposide B'] |
| 417.02061 | 5.63E-08 | 2.781381 | 3.3414251 | 4.673092 |  |  |  |  |  | 0 |
| 207.04913 | $1.81 \mathrm{E}-08$ | 2.2139015 | 3.5047331 | 4.740241 |  |  |  |  |  | 0 |
| 325.11389 | 2.30E-06 | 2.9753195 | 4.4236702 | 4.902078 | C12H22O10 | [M-H]- | 326.1213 | 325.1140236 | -0.41 | ['2-O-alpha-L-Rhamnopyranosyl-Dglucopyranose', 'Robinobiose', 'Rutinose'] |
| 325.11389 | 2.30E-06 | 2.9753195 | 4.4236702 | 4.902078 | C16H2ON2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 288.147393 | 325.1138446 | 0.14 | ['Methyl 2-(4-isopropyl-4-methyl-5-oxo-2-imidazolin-2-yl)-p-toluate', 'Methyl 6-(4-isopropyl-4-methyl-5-oxo-2-imidazolin-2-yl)-mtoluate'] |
| 375.03352 | 6.37E-12 | 3.9828739 | 4.4704587 | 4.913412 | C12H18O11 | [M+K-2H]- | 338.084915 | 375.0335216 | 0 | ['L-Ascorbic acid-2-glucoside'] |
| 375.03352 | $6.37 \mathrm{E}-12$ | 3.9828739 | 4.4704587 | 4.913412 | C14H17N2O4PS | [ $\mathrm{M}+\mathrm{Cl}]$ - | 340.064668 | 375.0340696 | -1.47 | ['Pyridafenthion'] |
| 375.03352 | 6.37E-12 | 3.9828739 | 4.4704587 | 4.913412 | C15H14O10 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 354.0587 | 375.0333686 | 0.4 | ['2-Caffeoylisocitrate'] |
| 337.11387 | $2.44 \mathrm{E}-06$ | 3.0975974 | 4.4181921 | 4.987821 | C11H18O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 278.10017 | 337.1140236 | -0.46 | ['Tuliposide A'] |
| 337.11387 | $2.44 \mathrm{E}-06$ | 3.0975974 | 4.4181921 | 4.987821 | C17H2ON2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 300.147393 | 337.1138446 | 0.08 | ['Bifenazate'] |
| 400.11776 | $5.27 \mathrm{E}-11$ | 3.7647008 | 4.1045607 | 5.031193 |  |  |  |  |  | 0 |
| 191.01885 | 8.30E-14 | 78.900447 | 46.434825 | 5.038264 | C5H6N4O3 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 170.043991 | 191.0186596 | 1 | ['5-Ureido-4-imidazole carboxylate'] |
| 427.10926 | 5.20E-07 | 4.2059415 | 4.3800524 | 5.04602 |  |  |  |  |  | 0 |
| 386.10212 | 1.03E-11 | 4.4689105 | 4.2979373 | 5.152308 |  |  |  |  |  | 0 |
| 383.11943 | $1.46 \mathrm{E}-08$ | 2.446913 | 4.419375 | 5.174557 | C12H20O10 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 324.10565 | 383.1195036 | -0.19 | ["Bis-D-fructose 2',1:2,1-dianhydride", "DFructofuranose 1,2':2,3-dianhydride"] |
| 383.11943 | $1.46 \mathrm{E}-08$ | 2.446913 | 4.419375 | 5.174557 | C14H24O12 | [M-H]- | 384.12678 | 383.1195036 | -0.19 | ['Acetyl-maltose'] |
| 342.98365 | 5.65E-11 | 4.5011677 | 5.3427734 | 5.288079 |  |  |  |  |  | 0 |
| 385.00288 | 7.00E-15 | 3.4128219 | 5.0582649 | 5.397598 |  |  |  |  |  | 0 |
| 374.04946 | $1.64 \mathrm{E}-12$ | 3.4432618 | 4.3059785 | 5.427069 |  |  |  |  |  | 0 |
| 441.16142 | 4.00E-14 | 1.9092537 | 3.7776368 | 5.472952 |  |  |  |  |  | 0 |
| 425.07021 | 4.18E-13 | 3.0542767 | 4.6732769 | 5.502602 |  |  |  |  |  | 0 |
| 265.05625 | $1.21 \mathrm{E}-12$ | 3.7761599 | 4.1872425 | 5.55954 | C7H1007 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 206.042655 | 265.0565086 | -0.98 | ['(2S,3R)-3-Hydroxybutane-1,2,3tricarboxylate', '(R)-2-Hydroxybutane-1,2,4tricarboxylate', '2-Methylcitrate', 'Homoisocitrate'] |
| 265.05625 | $1.21 \mathrm{E}-12$ | 3.7761599 | 4.1872425 | 5.55954 | C9H16N4S2 | [M+Na-2H]- | 244.08164 | 265.0563086 | -0.22 | ['Metiamide'] |


| 575.18313 | 2.30E-11 | 3.0755518 | 5.2656521 | 5.616734 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 385.09872 | $2.62 \mathrm{E}-11$ | 4.8068869 | 4.7064377 | 5.636592 |  |  |  |  |  | 0 |
| 261.06137 | $3.45 \mathrm{E}-07$ | 3.5717731 | 4.932399 | 5.758635 |  |  |  |  |  | 0 |
| 399.11433 | $1.04 \mathrm{E}-10$ | 4.2011967 | 4.727204 | 5.764864 | C12H20O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 340.100565 | 399.1144186 | -0.22 | ['3-Ketolactose', '3-Ketosucrose', 'Cellobiono-1,5-lactone'] |
| 399.11433 | $1.04 \mathrm{E}-10$ | 4.2011967 | 4.727204 | 5.764864 | C19H26N2O3S | [M+K-2H]- | 362.166415 | 399.1150216 | -1.73 | ['DU 122290'] |
| 223.04612 | $1.33 \mathrm{E}-12$ | 2.6681982 | 4.6163182 | 5.765654 | C11H10N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 186.079313 | 223.0457646 | 1.59 | ['Credazine', 'Deoxyvasicinone'] |
| 223.04612 | $1.33 \mathrm{E}-12$ | 2.6681982 | 4.6163182 | 5.765654 | C5H8O6 | [M+Hac-H]- | 164.03209 | 223.0459436 | 0.79 | ['2-Dehydro-D-xylonate'] |
| 379.06471 | 3.81E-13 | 3.2647186 | 4.9133998 | 5.794453 | C12H22O11 | [M+K-2H]- | 342.116215 | 379.0648216 | -0.29 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-D-glucose', 'Cellobiose', 'D-Fructosyl-D-fructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', <br> 'Gentiobiose', 'Inulobiose', 'Isomaltose', 'Lactose', 'Lactulose', 'Laminaribiose', <br> 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alpha-Trehalose', 'alpha-Cellobiose', 'alpha-D-Aldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |
| 266.05964 | $1.03 \mathrm{E}-11$ | 3.9255859 | 4.2796763 | 5.840599 |  |  |  |  |  | 0 |
| 390.13344 | 3.85E-09 | 3.7589046 | 3.8146016 | 5.880415 |  |  |  |  |  | 0 |
| 424.14607 | $7.32 \mathrm{E}-13$ | 5.5127403 | 6.0223677 | 5.920582 |  |  |  |  |  | 0 |
| 520.31284 | 0.04122018 | 3.1424856 | 7.3491811 | 5.992662 |  |  |  |  |  | 0 |
| 209.0556 | 2.20E-14 | 4.4884595 | 5.8024095 | 6.318664 | C12H12O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 172.088815 | 209.0552666 | 1.59 | ['Capillone'] |
| 410.13045 | $1.14 \mathrm{E}-11$ | 3.527985 | 4.9790131 | 6.374387 | C13H21NO10 | [M+Hac-H]- | 351.116549 | 410.1304026 | 0.12 | ['N-Acetyl-4-O-acetylneuraminate', 'N-Acetyl-7-O-acetyIneuraminate', 'N-Acetyl-9-OacetyIneuraminate'] |
| 391.06473 | $1.30 \mathrm{E}-10$ | 5.0653109 | 6.7670898 | 6.582032 | C16H18O10 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 370.09 | 391.0646686 | 0.16 | ['Fraxin'] |
| 208.05464 | 6.62E-13 | 3.9877925 | 5.44874 | 6.790505 |  |  |  |  |  | 0 |
| 389.12999 | 8.46E-12 | 4.4942104 | 5.3223434 | 6.887534 | C17H26N2O4S | [M+Cl]- | 354.16133 | 389.1307316 | -1.91 | ['Sultopride'] |
| 482.15157 | $6.14 \mathrm{E}-06$ | 5.0081626 | 6.9659244 | 6.995297 |  |  |  |  |  | 0 |
| 369.10382 | 8.30E-14 | 3.619161 | 4.972374 | 7.281638 |  |  |  |  |  | 0 |
| 263.07701 | 6.29E-09 | 2.8046349 | 6.3091821 | 7.297703 | C14H14N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 226.110613 | 263.0770646 | -0.21 | ['Metyrapone'] |
| 263.07701 | 6.29E-09 | 2.8046349 | 6.3091821 | 7.297703 | C8H12O6 | [M+Hac-H]- | 204.06339 | 263.0772436 | -0.89 | ['3-o-Ethyl-L-ascorbic acid'] |
| 559.18811 | $1.84 \mathrm{E}-10$ | 2.9053877 | 6.573124 | 7.365639 |  |  |  |  |  | 0 |


| 321.0826 | 3.18E-07 | 4.7487993 | 6.9876311 | 7.528341 | C16H16N2O3 | [M+(37Cl)]- | 284.116093 | 321.0825446 | 0.17 | ['Metominostrobin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 351.15199 | $4.17 \mathrm{E}-13$ | 5.5493446 | 8.4484312 | 7.944238 |  |  |  |  |  | 0 |
| 397.13509 | $2.66 \mathrm{E}-07$ | 4.0239173 | 7.5415192 | 7.96425 | C19H26N2O3S | [M+Cl]- | 362.166415 | 397.1358166 | -1.83 | ['DU 122290'] |
| 345.02282 | 3.81E-13 | 3.2764926 | 6.9503998 | 8.258422 |  |  |  |  |  | 0 |
| 381.10381 | $2.59 \mathrm{E}-13$ | 2.7078004 | 5.2806353 | 8.366795 |  |  |  |  |  | 0 |
| 342.11931 | $1.80 \mathrm{E}-14$ | 10.791161 | 9.7813507 | 8.565065 |  |  |  |  |  | 0 |
| 458.15976 | $9.54 \mathrm{E}-09$ | 2.4369165 | 5.5648031 | 8.635193 |  |  |  |  |  | 0 |
| 437.07037 | $1.30 \mathrm{E}-14$ | 4.26284 | 7.7518897 | 8.735145 |  |  |  |  |  | 0 |
| 427.14567 | $2.04 \mathrm{E}-07$ | 2.1576393 | 4.7519551 | 8.815182 |  |  |  |  |  | 0 |
| 469.11975 | $1.27 \mathrm{E}-11$ | 3.3654411 | 7.3528669 | 9.158153 |  |  |  |  |  | 0 |
| 355.12441 | $1.92 \mathrm{E}-11$ | 1.7517238 | 4.7785364 | 9.405106 |  |  |  |  |  | 0 |
| 320.09856 | 4.65E-08 | 5.6995241 | 8.2086562 | 9.481797 | C10H15NO7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 261.084854 | 320.0987076 | -0.46 | ['Hymexazol O-glucoside', 'Hymexazol Nglucoside'] |
| 466.15666 | $2.47 \mathrm{E}-10$ | 4.7789002 | 7.7821343 | 9.490838 |  |  |  |  |  | 0 |
| 207.05271 | 2.20E-14 | 5.5565958 | 8.5239201 | 10.15512 |  |  |  |  |  | 0 |
| 457.15626 | $1.14 \mathrm{E}-11$ | 3.1909662 | 7.2016084 | 10.17518 |  |  |  |  |  | 0 |
| 326.98882 | 5.35E-13 | 7.0839207 | 10.490157 | 10.3027 | C7H15010P | [M+K-2H]- | 290.040288 | 326.9888946 | -0.23 | ['D-glycero-D-manno-Heptose 1-phosphate', 'D-glycero-D-manno-Heptose 7-phosphate', 'Sedoheptulose 1-phosphate', 'Sedoheptulose 7-phosphate'] |
| 451.08598 | 5.00E-15 | 5.8994555 | 9.5491668 | 10.36302 |  |  |  |  |  | 0 |
| 471.17194 | 7.57E-13 | 3.4207196 | 7.1570119 | 10.4715 |  |  |  |  |  | 0 |
| 453.10164 | $5.24 \mathrm{E}-13$ | 4.3988393 | 9.867378 | 11.2169 |  |  |  |  |  | 0 |
| 359.03843 | $9.15 \mathrm{E}-13$ | 6.2603968 | 10.267949 | 11.95718 |  |  |  |  |  | 0 |
| 358.05443 | $2.52 \mathrm{E}-13$ | 5.5208847 | 9.586293 | 12.21416 |  |  |  |  |  | 0 |
| 425.13003 | $1.08 \mathrm{E}-11$ | 9.7268731 | 12.59722 | 12.40165 | C20H24N2O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 388.163438 | 425.1298896 | 0.33 | ['Nisoldipine'] |
| 455.14071 | $1.33 \mathrm{E}-12$ | 3.8480884 | 9.624797 | 12.40364 | C21H26N2O7 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 418.174003 | 455.1404546 | 0.56 | ['Nimodipine'] |
| 405.08029 | 5.00E-15 | 10.845743 | 14.601419 | 13.39942 |  |  |  |  |  | 0 |
| 529.17759 | 4.90E-12 | 2.5174434 | 8.1351353 | 13.8723 |  |  |  |  |  | 0 |


| 207.05096 | $2.80 \mathrm{E}-14$ | 7.3718725 | 11.20727 | 13.87457 | C5H8O5 | [M+Hac-H]- | 148.037175 | 207.0510286 | -0.33 | ['(R)-2-Hydroxyglutarate', '(R)-2-MethyImalate', '(S)-2-Hydroxyglutarate', '(S)-2-Methylmalate', '2-Dehydro-3-deoxy-D-xylonate', '2-Dehydro-3-deoxy-L-arabinonate', '2-Hydroxyglutarate', 'Citramalate', 'D-Arabinono-1,4-lactone', 'D-Xylono-1,4-lactone', 'D-Xylonolactone', 'D-erythro-3-Methylmalate', 'D-threo-3- <br> Methylmalate', 'L-Arabinono-1,4-lactone', 'L-Arabinono-1,5-lactone', 'L-Xylono-1,4-lactone', 'L-threo-3-Methylmalate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 279.07193 | $9.74 \mathrm{E}-13$ | 6.498242 | 12.717197 | 15.25485 | C8H12O7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 220.058305 | 279.0721586 | -0.82 | ['(R)-(Homo)2-citrate', '1-Hydroxypentane-1,2,5-tricarboxylate'] |
| 411.11438 | $2.66 \mathrm{E}-12$ | 8.3099995 | 14.870155 | 17.82251 |  |  |  |  |  | 0 |
| 394.13546 | 9.05E-10 | 4.1087288 | 9.6447409 | 19.1882 |  |  |  |  |  | 0 |
| 396.12283 | 3.30E-09 | 5.387287 | 13.906043 | 23.19137 | C15H22N6O5S | [M-2H]- | 398.137241 | 396.1226882 | 0.36 | ['S-Adenosyl-L-methionine'] |
| 399.1507 | 5.06E-09 | 2.798875 | 11.774758 | 24.01297 | C13H24O10 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 340.13695 | 399.1508036 | -0.26 | ['Methyl-2-alpha-L-fucopyranosyl-beta-Dgalactoside'] |
| 365.10894 | 6.52E-11 | 5.4811247 | 16.860221 | 28.96936 | C18H22N2O2S | [ $\mathrm{M}+\mathrm{Cl}]$ - | 330.1402 | 365.1096016 | -1.81 | ['Pyributicarb'] |
| 365.10894 | 6.52E-11 | 5.4811247 | 16.860221 | 28.96936 | C19H24N2OS | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 328.160935 | 365.1095416 | -1.65 | ['Methotrimeprazine'] |
| 192.05968 | 1.88E-09 | 14.926316 | 35.449695 | 31.99183 |  |  |  |  |  | 0 |
| 453.12509 | $1.45 \mathrm{E}-09$ | 8.0064399 | 20.573735 | 32.87536 |  |  |  |  |  | 0 |
| 395.11944 | 8.15E-10 | 6.9488762 | 20.350613 | 34.46331 | C19H24N2O3S | [M+Cl]- | 360.150765 | 395.1201666 | -1.84 | ['LY395153'] |
| 208.05363 | 1.08E-10 | 6.9985704 | 10.95796 | 55.56405 | C11H11NO | [ $\mathrm{M}+\mathrm{Cl}]-$ | 173.084064 | 208.0534656 | 0.79 | ['1,3-Dimethyl-8-isoquinolinol', 'Pyroquilon'] |
| 208.05363 | 1.08E-10 | 6.9985704 | 10.95796 | 55.56405 | C14H10O2 | [M-2H]- | 210.06808 | 208.0535272 | 0.49 | ['1,2-Anthracenediol', '9,10- <br> Dihydroxyanthracene', 'Phenanthrene-3,4diol'] |
| 191.05527 | 7.78E-08 | 31.535158 | 66.471808 | 57.64901 | C6H10N4O2 | [M+Na-2H]- | 170.080376 | 191.0550446 | 1.18 | ['N-Isopropylammelide'] |
| 191.05623 | 4.01E-09 | 72.724748 | 144.77501 | 113.8161 | C5H8O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 132.04226 | 191.0561136 | 0.61 | ['(4S)-4,5-Dihydroxypentan-2,3-dione', '(S)-2Acetolactate', '2-(Hydroxymethyl)-4oxobutanoate', '2-Acetolactate', '3-Hydroxy-3-methyl-2-oxobutanoic acid', '4-Hydroxy-2oxopentanoate', 'Deoxyribonolactone', 'Glutarate'] |
| 191.05623 | 4.01E-09 | 72.724748 | 144.77501 | 113.8161 | C7H12O6 | [M-H]- | 192.06339 | 191.0561136 | 0.61 | ['2D-5-O-Methyl-2,3,5/4,6pentahydroxycyclohexanone', 'Quinate', 'Valiolone'] |

Table 2 A table of $\mathrm{m} / \mathrm{z}$ identified from the analysis of the white wine aliquots over four time points: zero, one, three, six months, which includes putative identifications. Univariate statistics were applied to this data; using the four time points, a fold change was determined using zero month as the reference time point, fold change 1 represents one month, fold change 2 represents three months, and fold change 3 represents the six month time point. The table is sorted on the smallest fold change 3 in conjunction with the smallest adjusted p-value, the peaks changing most significantly between zero month and six months, decreasing over time to increasing over time.

| M/Z | adjusted p value | fold change 1 | fold change 2 | fold change 3 | Empirical formula (parent) | Ion form | Theoretical mass (neutral) (Da) | Theoretical $\mathrm{m} / \mathrm{z}$ (Da) | $\begin{aligned} & \text { Mass } \\ & \text { error } \\ & \text { (ppm) } \end{aligned}$ | KEGG_COMPOUND |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 193.03569 | $2.37 \mathrm{E}-11$ | 0.012760957 | 0.003183353 | 0.002184046 | C4H6O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 134.021525 | 193.0353786 | 1.61 | ['(R)-Malate', '(S)-Malate', '3-Dehydro-Lthreonate', 'Malate'] |
| 193.03569 | $2.37 \mathrm{E}-11$ | 0.012760957 | 0.003183353 | 0.002184046 | C6H1007 | [M-H]- | 194.042655 | 193.0353786 | 1.61 | ['2-Dehydro-D-galactonate', '2-Keto-Dgluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-D-gluconate', 'D-Fructuronate', 'DGalacturonate', 'D-Glucuronate', 'DGlucuronic acid', 'D-Mannuronate', 'DTagaturonate', 'Galacturonic acid', 'LGuluronic acid', 'L-Iduronic acid', 'beta-DGlucopyranuronic acid'] |
| 529.03705 | 1.81E-10 | 0.024171317 | 0.003751459 | 0.00293527 |  |  |  |  |  | 0 |
| 268.03919 | 7.87E-11 | 0.0160741 | 0.005888001 | 0.005709178 |  |  |  |  |  | 0 |
| 251.04101 | $1.66 \mathrm{E}-12$ | 0.044394299 | 0.007859212 | 0.005711511 | C6H8O7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 192.027005 | 251.0408586 | 0.6 | ['(1R,2S)-1-Hydroxypropane-1,2,3tricarboxylate', '(1S,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(4R,5S)-4,5,6- <br> Trihydroxy-2,3-dioxohexanoate', '2,5-Didehydro-D-gluconate', '2-Dehydro-3-deoxy-D-glucarate', '5-Dehydro-4-deoxy-Dglucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] |
| 251.04101 | $1.66 \mathrm{E}-12$ | 0.044394299 | 0.007859212 | 0.005711511 | C8H14N4S2 | [M+Na-2H]- | 230.06599 | 251.0406586 | 1.4 | ['Thiaburimamide'] |
| 267.03583 | $1.61 \mathrm{E}-10$ | 0.017506945 | 0.006788292 | 0.006251924 |  |  |  |  |  | 0 |
| 513.05935 | $9.19 \mathrm{E}-11$ | 0.032665725 | 0.00850576 | 0.007227499 |  |  |  |  |  | 0 |
| 491.12608 | 3.49E-09 | 0.034628872 | 0.008805512 | 0.007302791 |  |  |  |  |  | 0 |
| 327.0572 | $2.17 \mathrm{E}-10$ | 0.027169891 | 0.008208649 | 0.007678674 | C14H14N2O5 | [M+(37CI)]- | 290.090273 | 327.0567246 | 1.45 | ['N2-Malonyl-D-tryptophan'] |
| 207.04994 | 0.002408805 | 0.018348179 | 0.009366546 | 0.008951516 |  |  |  |  |  | 0 |


| 215.01769 | $4.79 \mathrm{E}-10$ | 0.02181524 | 0.009308708 | 0.00911371 | C6H1007 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 194.042655 | 215.0173236 | 1.7 | ['2-Dehydro-D-galactonate', '2-Keto-Dgluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-D-gluconate', 'D-Fructuronate', 'DGalacturonate', 'D-Glucuronate', 'DGlucuronic acid', 'D-Mannuronate', 'DTagaturonate', 'Galacturonic acid', 'LGuluronic acid', 'L-Iduronic acid', 'beta-DGlucopyranuronic acid'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 273.01036 | 1.82E-11 | 0.079492172 | 0.019645468 | 0.01100845 |  |  |  |  |  | 0 |
| 461.18576 | $1.74 \mathrm{E}-05$ | 0.071125351 | 0.015758176 | 0.011052202 | C23H37O5P | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 424.237863 | 461.1864696 | -1.54 | ['3beta-Hydroxy-16-phosphonopregn-5-en-20-one monoethyl ester'] |
| 379.04991 | $1.51 \mathrm{E}-10$ | 0.056543304 | 0.014110242 | 0.0126704 |  |  |  |  |  | 0 |
| 475.13098 | 1.96E-11 | 0.039472899 | 0.014184307 | 0.012822007 | C28H24O5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 440.162375 | 475.1317766 | -1.68 | ['Marchantin A'] |
| 395.04485 | 3.32E-08 | 0.058882192 | 0.016239291 | 0.012903764 | C11H23N2O7PS | [M+K-2H]- | 358.096363 | 395.0449696 | -0.3 | ['Pantetheine 4-phosphate'] |
| 476.13452 | $1.67 \mathrm{E}-11$ | 0.039962607 | 0.014899046 | 0.013327292 |  |  |  |  |  | 0 |
| 395.02387 | 3.93E-08 | 0.06816137 | 0.01799405 | 0.013863973 |  |  |  |  |  | 0 |
| 445.0844 | $1.69 \mathrm{E}-11$ | 0.070504677 | 0.018624773 | 0.015133199 |  |  |  |  |  | 0 |
| 475.20139 | $6.81 \mathrm{E}-05$ | 0.079681231 | 0.020394274 | 0.017069455 | C17H34N4O10 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 454.227496 | 475.2021646 | -1.63 | ['Ribostamycin', 'Xylostasin'] |
| 149.00871 | $4.32 \mathrm{E}-11$ | 0.051111537 | 0.021029003 | 0.017442906 |  |  |  |  |  | 0 |
| 149.00929 | $2.19 \mathrm{E}-10$ | 0.069968361 | 0.021218781 | 0.018218955 | C2H2O4 | [M+Hac-H]- | 89.99531 | 149.0091636 | 0.85 | ['Oxalate'] |
| 149.00929 | $2.19 \mathrm{E}-10$ | 0.069968361 | 0.021218781 | 0.018218955 | C4H6O6 | [M-H]- | 150.01644 | 149.0091636 | 0.85 | ['(R,R)-Tartaric acid', '(S,S)-Tartaric acid', 'meso-Tartaric acid'] |
| 150.01266 | 1.58E-10 | 0.071815728 | 0.023493804 | 0.018979126 |  |  |  |  |  | 0 |
| 455.04508 | $2.86 \mathrm{E}-10$ | 0.092563172 | 0.02180407 | 0.019623426 |  |  |  |  |  | 0 |
| 235.04602 | $5.01 \mathrm{E}-11$ | 0.117763125 | 0.02964588 | 0.022862654 | C12H10N2O | [M+(37Cl)]- | 198.079313 | 235.0457646 | 1.09 | ['4-(2-Pyrazinylethenyl)phenol', '4Hydroxyazobenzene', ' N Nitrosodiphenylamine', 'paraNitrosodiphenylamine'] |
| 235.04602 | 5.01E-11 | 0.117763125 | 0.02964588 | 0.022862654 | C6H8O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 176.03209 | 235.0459436 | 0.33 | ['(4S)-4,6-Dihydroxy-2,5-dioxohexanoate', '(4S,5S)-4,5-Dihydroxy-2,6-dioxohexanoate', <br> '2-Dehydro-D-glucono-1,5-lactone', '2- <br> Hydroxy-3-oxoadipate', '5-Dehydro-4-deoxy- <br> D-glucuronate', 'Ascorbate', 'D- <br> Galacturonolactone', 'D-Glucurono-6,2- <br> lactone', 'D-Glucuronolactone', 'L-xylo- <br> Hexulonolactone', 'Parapyruvate'] |
| 235.04602 | 5.01E-11 | 0.117763125 | 0.02964588 | 0.022862654 | C7H15N3O2S2 | [M-2H]- | 237.060571 | 235.0460182 | 0.01 | ['Cartap'] |
| 159.03012 | $5.78 \mathrm{E}-13$ | 0.038298153 | 0.024040367 | 0.022962486 | C4H4O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 100.016045 | 159.0298986 | 1.39 | ['Succinic anhydride'] |
| 159.03012 | $5.78 \mathrm{E}-13$ | 0.038298153 | 0.024040367 | 0.022962486 | C6H8O5 | [M-H]- | 160.037175 | 159.0298986 | 1.39 | ['2-Formylglutarate', '2-Oxoadipate', '3Oxoadipate', '3D-(3,5/4)- <br> Trihydroxycyclohexane-1,2-dione', 'D-2,3Diketo 4-deoxy-epi-inositol'] |


| 667.30171 | 1.78E-05 | 0.094261171 | 0.033148825 | 0.023217509 | C37H4ON2O6 | [M+Hac-H]- | 608.288638 | 667.3024916 | -1.17 | ['Berbamine', 'Gyrocarpine', 'Oxyacanthine', 'Pycnamine', 'Thalmine'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 653.28604 | $1.90 \mathrm{E}-05$ | 0.104129058 | 0.03367471 | 0.024867149 | C36H38N2O6 | [M+Hac-H]- | 594.272988 | 653.2868416 | -1.23 | $\begin{gathered} \hline \text { ['(+)-Atherospermoline', '(+)-Bebeerine', } \\ \text { 'Aromoline', 'Daphnandrine', } \\ \text { 'Isochondrodendrine', 'Obamegine'] } \\ \hline \end{gathered}$ |
| 653.17932 | 8.30E-11 | 0.060729644 | 0.028385866 | 0.026915447 |  |  |  |  |  | 0 |
| 201.04086 | 5.64E-10 | 0.129743173 | 0.054020291 | 0.027090118 | C6H14NO2S | [M+(37Cl)]- | 164.074526 | 201.0409776 | -0.58 | ['S-Methyl-L-methionine'] |
| 201.04086 | $5.64 \mathrm{E}-10$ | 0.129743173 | 0.054020291 | 0.027090118 | C6H6O4 | [M+Hac-H]- | 142.02661 | 201.0404636 | 1.97 | ['(S)-5-Oxo-2,5-dihydrofuran-2-acetate', '1,2,3,5-Tetrahydroxybenzene', '2,5- <br> Dihydro-5-oxofuran-2-acetate', '2Hydroxymuconate semialdehyde', '2-Oxo-2,3-dihydrofuran-5-acetate', 'Kojic acid', 'cis,cis-4-Hydroxymuconic semialdehyde', 'cis,cis-Muconate', 'cis,transHexadienedioate'] |
| 681.31757 | 5.76E-05 | 0.109003786 | 0.035200065 | 0.027256159 | C38H42N2O6 | [M+Hac-H]- | 622.304288 | 681.3181416 | -0.84 | ['(+)-O-Methylthalicberine', '(+)- <br> Tetrandrine', 'Cycleanine', 'Isotetrandrine', 'Obaberine', 'Rodiasine'] |
| 681.31757 | 5.76E-05 | 0.109003786 | 0.035200065 | 0.027256159 | C4OH46N2O8 | [M-H]- | 682.325418 | 681.3181416 | -0.84 | ['Fetidine'] |
| 461.11554 | $2.08 \mathrm{E}-10$ | 0.073571178 | 0.030181809 | 0.027857301 |  |  |  |  |  | 0 |
| 193.07211 | $7.91 \mathrm{E}-11$ | 0.082547223 | 0.038425293 | 0.028959632 | C5H10O4 | [M+Hac-H]- | 134.05791 | 193.0717636 | 1.79 | ['(R)-2,3-Dihydroxy-3-methylbutanoate', '1-Deoxy-D-xylulose', '2,3-Dihydroxy-3methylbutanoate', '2-Deoxy-L-arabinose', '2-Deoxy-alpha-D-ribopyranose', 'Deoxyribose'] |
| 193.07211 | 7.91E-11 | 0.082547223 | 0.038425293 | 0.028959632 | C7H14O6 | [M-H]- | 194.07904 | 193.0717636 | 1.79 | ['(-)-Quebrachitol', '1-O-Methyl-myoinositol', '3-O-Methyl-myo-inositol', '4-O-Methyl-myo-inositol', '5-O-Methyl-myoinositol', '6-O-Methyl-myo-inositol', 'DPinitol', 'Methyl beta-D-galactoside', 'O-Methyl-scyllo-inositol'] |
| 313.07796 | 2.02E-09 | 0.060072078 | 0.031853652 | 0.029339709 | C7H16N8O4 | [M+K-2H]- | 276.129452 | 313.0780586 | -0.31 | ['Trimethylenetetraurea'] |
| 205.08926 | $4.49 \mathrm{E}-11$ | 0.031905477 | 0.027091918 | 0.029921495 |  |  |  |  |  | 0 |
| 603.0742 | 7.20E-10 | 0.207119834 | 0.043014147 | 0.029932791 |  |  |  |  |  | 0 |
| 647.14495 | $1.23 \mathrm{E}-10$ | 0.114785036 | 0.038741534 | 0.032369544 | C34H30N2O9 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 610.195133 | 647.1437396 | 1.87 | ['Atalanine'] |
| 265.02022 | $1.27 \mathrm{E}-11$ | 0.175690012 | 0.048598865 | 0.033842732 | C6H6O8 | [M+Hac-H]- | 206.00627 | 265.0201236 | 0.36 | ['3-Oxalomalate'] |
| 265.02022 | $1.27 \mathrm{E}-11$ | 0.175690012 | 0.048598865 | 0.033842732 | C8H12N4OS2 | [M+Na-2H]- | 244.045255 | 265.0199236 | 1.12 | ['CGP 52608'] |
| 789.51353 | 0.02313151 | 0.073035042 | 0.026354515 | 0.03519344 |  |  |  |  |  | 0 |
| 535.15252 | $5.69 \mathrm{E}-11$ | 0.061989824 | 0.035987139 | 0.03519659 |  |  |  |  |  | 0 |
| 209.06711 | $3.13 \mathrm{E}-10$ | 0.113343246 | 0.054017361 | 0.036216013 |  |  |  |  |  | 0 |
| 379.02002 | $6.77 \mathrm{E}-12$ | 0.133384005 | 0.04215413 | 0.036508971 |  |  |  |  |  | 0 |
| 607.13747 | $1.81 \mathrm{E}-10$ | 0.117756881 | 0.037599964 | 0.037555258 |  |  |  |  |  | 0 |


| 191.02007 | $1.45 \mathrm{E}-10$ | 0.088398859 | 0.037153199 | 0.03914626 | C4H4O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 132.005875 | 191.0197286 | 1.79 | ['2-Hydroxyethylenedicarboxylate', 'Oxaloacetate', 'trans-2,3-Epoxysuccinate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 191.02007 | $1.45 \mathrm{E}-10$ | 0.088398859 | 0.037153199 | 0.03914626 | C6H8O7 | [M-H]- | 192.027005 | 191.0197286 | 1.79 | ['(1R,2S)-1-Hydroxypropane-1,2,3tricarboxylate', '(1S,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(4R,5S)-4,5,6- <br> Trihydroxy-2,3-dioxohexanoate', '2,5-Didehydro-D-gluconate', '2-Dehydro-3-deoxy-D-glucarate', '5-Dehydro-4-deoxy-Dglucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] |
| 439.08654 | $9.28 \mathrm{E}-11$ | 0.087740147 | 0.042411723 | 0.039616762 |  |  |  |  |  | 0 |
| 195.08779 | $3.03 \mathrm{E}-10$ | 0.071849065 | 0.041338294 | 0.040918021 |  |  |  |  |  | 0 |
| 197.08232 | $2.73 \mathrm{E}-10$ | 0.078577557 | 0.044864742 | 0.041770362 | C10H14O4 | [M-H]- | 198.08921 | 197.0819336 | 1.96 | ['1,2-Dihydroxymint lactone', '1-(3,4-Dimethoxyphenyl)ethane-1,2-diol', 'Paeonilactone A', 'cis-2,3-Dihydroxy-2,3-dihydro-p-cumate'] |
| 197.08232 | $2.73 \mathrm{E}-10$ | 0.078577557 | 0.044864742 | 0.041770362 | C7H16NO3 | [M+Cl]- | 162.113019 | 197.0824206 | -0.51 | ['Carnitine'] |
| 197.08232 | $2.73 \mathrm{E}-10$ | 0.078577557 | 0.044864742 | 0.041770362 | C8H10O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 138.06808 | 197.0819336 | 1.96 | ['1-(4-Hydroxyphenyl)ethanol', '2-Methyl-6-oxohepta-2,4-dienal', '3-Ethylcatechol', '3Methoxybenzyl alcohol', '4Hydroxyphenylethanol', 'Styrene cis-glycol'] |
| 197.08232 | 2.73E-10 | 0.078577557 | 0.044864742 | 0.041770362 | C8H18NO2 | [M+K-2H]- | 160.133754 | 197.0823606 | -0.21 | ['Methacholine'] |
| 587.0794 | $4.81 \mathrm{E}-10$ | 0.181507725 | 0.047689598 | 0.042061405 |  |  |  |  |  | 0 |
| 395.01498 | 3.08E-09 | 0.177130432 | 0.055127267 | 0.042391402 |  |  |  |  |  | 0 |
| 489.21712 | 0.000205573 | 0.138175441 | 0.049835569 | 0.042679958 | C18H36N4O10 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 468.243146 | 489.2178146 | -1.42 | ['Gentamicin A'] |
| 489.21712 | 0.000205573 | 0.138175441 | 0.049835569 | 0.042679958 | C24H34N4O5S | [M-H]- | 490.224993 | 489.2177166 | -1.22 | ['Glimepiride'] |
| 401.13047 | $7.52 \mathrm{E}-12$ | 0.103102444 | 0.053257771 | 0.043887464 | C12H22O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 342.116215 | 401.1300686 | 1 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-Dglucose', 'Cellobiose', 'D-Fructosyl-Dfructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', 'Gentiobiose', 'Inulobiose', <br> 'Isomaltose', 'Lactose', 'Lactulose', 'Laminaribiose', 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alphaTrehalose', 'alpha-Cellobiose', 'alpha-DAldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |
| 402.13402 | $1.55 \mathrm{E}-11$ | 0.103932714 | 0.054329495 | 0.04490222 |  |  |  |  |  | 0 |


| 329.10925 | $2.53 \mathrm{E}-05$ | 0.131275838 | 0.038938896 | 0.04534704 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 459.20664 | 0.000119174 | 0.103195333 | 0.048958951 | 0.046068796 | C23H37O5P | [M+Cl]- | 424.237863 | 459.2072646 | -1.36 | ['3beta-Hydroxy-16-phosphonopregn-5-en-20-one monoethyl ester'] |
| 513.10296 | 8.65E-09 | 0.039976458 | 0.04428579 | 0.047288625 | C22H22N6O5S2 | [M-H]- | 514.109313 | 513.1020366 | 1.8 | ['Cefpirome'] |
| 597.35007 | $1.53 \mathrm{E}-08$ | 0.12314928 | 0.053009628 | 0.048117539 | C40H50O2 | [M+Cl]- | 562.38108 | 597.3504816 | -0.69 | ['Rhodoxanthin'] |
| 312.06563 | $2.86 \mathrm{E}-09$ | 0.09035123 | 0.049783729 | 0.04839714 | C10H11N3O3S | [M+Hac-H]- | 253.052114 | 312.0659676 | -1.08 | ['Sulfamethoxazole'] |
| 312.06563 | $2.86 \mathrm{E}-09$ | 0.09035123 | 0.049783729 | 0.04839714 | C12H15N3O5S | [M-H]- | 313.073244 | 312.0659676 | -1.08 | ['Albendazole-beta-hydroxysulphone', 'Albendazole-gamma-hydroxysulphone'] |
| 625.38174 | 7.78E-08 | 0.110120554 | 0.049408334 | 0.048464379 |  |  |  |  |  | 0 |
| 375.31211 | $4.32 \mathrm{E}-10$ | 0.097987968 | 0.046438035 | 0.048923044 | C19H40O3 | [M+Hac-H]- | 316.297745 | 375.3115986 | 1.36 | ['1-O-Hexadecyl-sn-glycerol'] |
| 663.13944 | 3.62E-09 | 0.176182896 | 0.107312562 | 0.049098936 |  |  |  |  |  | 0 |
| 151.06118 | 2.07E-11 | 0.102217169 | 0.057190991 | 0.049153532 | C3H8O3 | [M+Hac-H]- | 92.047345 | 151.0611986 | -0.12 | ['Glycerol'] |
| 151.06118 | $2.07 \mathrm{E}-11$ | 0.102217169 | 0.057190991 | 0.049153532 | C5H12O5 | [M-H]- | 152.068475 | 151.0611986 | -0.12 | ['D-Apiitol', 'L-Arabitol', 'Ribitol', 'Xylitol'] |
| 393.04451 | 7.03E-09 | 0.130718025 | 0.054776525 | 0.049267042 | C15H16O11 | [M+Na-2H]- | 372.069265 | 393.0439336 | 1.47 | ['2-O-Caffeoylglucarate'] |
| 637.18449 | $9.44 \mathrm{E}-11$ | 0.095505722 | 0.051681209 | 0.049780565 |  |  |  |  |  | 0 |
| 181.07209 | $1.69 \mathrm{E}-11$ | 0.093147239 | 0.057911699 | 0.049998257 | C4H1004 | [M+Hac-H]- | 122.05791 | 181.0717636 | 1.8 | ['D-Threitol', 'Erythritol'] |
| 181.07209 | $1.69 \mathrm{E}-11$ | 0.093147239 | 0.057911699 | 0.049998257 | C6H14O6 | [M-H]- | 182.07904 | 181.0717636 | 1.8 | ['D-Iditol', 'D-Sorbitol', 'Galactitol', 'Mannitol'] |
| 181.07209 | $1.69 \mathrm{E}-11$ | 0.093147239 | 0.057911699 | 0.049998257 | C7H14NO3 | [M+Na-2H]- | 160.097369 | 181.0720376 | 0.29 | ['3-Dehydrocarnitine'] |
| 421.04797 | 4.10E-09 | 0.242283728 | 0.072903508 | 0.050578967 | C15H18N2O8S | [M+Cl]- | 386.07839 | 421.0477916 | 0.42 | $['(7 \mathrm{R})-7-(4-$ Carboxybutanamido)cephalosporanate'] |
| 311.06233 | $1.93 \mathrm{E}-09$ | 0.10507567 | 0.052817971 | 0.050843259 |  |  |  |  |  | 0 |
| 376.31547 | $2.28 \mathrm{E}-10$ | 0.091829837 | 0.047481077 | 0.051567417 |  |  |  |  |  | 0 |
| 445.19259 | $5.60 \mathrm{E}-09$ | 0.086377245 | 0.05266785 | 0.052716877 |  |  |  |  |  | 0 |
| 587.09651 | $1.94 \mathrm{E}-10$ | 0.191425056 | 0.061405247 | 0.052764334 | C32H22O10 | [M+Na-2H]- | 566.1213 | 587.0959686 | 0.92 | ['Ginkgetin'] |
| 152.06466 | 8.23E-11 | 0.102989816 | 0.059588393 | 0.053174979 |  |  |  |  |  | 0 |
| 211.08275 | $3.75 \mathrm{E}-10$ | 0.097456614 | 0.063438294 | 0.054985309 | C5H12O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 152.068475 | 211.0823286 | 2 | ['D-Apiitol', 'L-Arabitol', 'Ribitol', 'Xylitol'] |
| 211.08275 | $3.75 \mathrm{E}-10$ | 0.097456614 | 0.063438294 | 0.054985309 | C7H16O7 | [M-H]- | 212.089605 | 211.0823286 | 2 | ['Perseitol', 'Volemitol'] |
| 269.08792 | $1.04 \mathrm{E}-11$ | 0.08182843 | 0.065848621 | 0.056184409 | C13H16N2O2 | [M+(37Cl)]- | 232.121178 | 269.0876296 | 1.08 | ['Aminoglutethimide', 'Melatonin'] |
| 269.08792 | $1.04 \mathrm{E}-11$ | 0.08182843 | 0.065848621 | 0.056184409 | C7H14O7 | [M+Hac-H]- | 210.073955 | 269.0878086 | 0.41 | ['Sedoheptulose', 'alpha-DMannoheptulopyranose', 'beta-DSedoheptulopyranose'] |
| 333.05959 | 5.77E-11 | 0.10261159 | 0.06062334 | 0.056873314 | C7H1509P | [M+Hac-H]- | 274.045373 | 333.0592266 | 1.09 | ['1-Deoxy-D-altro-heptulose 7-phosphate'] |
| 333.05959 | $5.77 \mathrm{E}-11$ | 0.10261159 | 0.06062334 | 0.056873314 | C9H19011P | [M-H]- | 334.066503 | 333.0592266 | 1.09 | ['2-(alpha-D-Galactosyl)-sn-glycerol 3phosphate', '2-(beta-D-Glucosyl)-sn-glycerol 3-phosphate', "alpha-D-Galactosyl-(1,1')-snglycerol 3-phosphate", 'sn-glycero-3-Phospho-1-inositol'] |
| 251.0774 | $1.31 \mathrm{E}-12$ | 0.195313221 | 0.064052991 | 0.056885716 | C13H14N2O | [M+(37C) $]$ - | 214.110613 | 251.0770646 | 1.34 | ['Harmaline'] |


| 251.0774 | 1.31E-12 | 0.195313221 | 0.064052991 | 0.056885716 | C7H12O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 192.06339 | 251.0772436 | 0.62 | ['2D-5-O-Methyl-2,3,5/4,6pentahydroxycyclohexanone', 'Quinate', 'Valiolone'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 216.00597 | 1.02E-10 | 0.07809873 | 0.054972799 | 0.057404172 |  |  |  |  |  | 0 |
| 309.08308 | $2.74 \mathrm{E}-10$ | 0.150225212 | 0.0673631 | 0.058162226 |  |  |  |  |  | 0 |
| 473.22211 | $1.17 \mathrm{E}-06$ | 0.115053539 | 0.062169826 | 0.05956796 |  |  |  |  |  | 0 |
| 373.06294 | $4.46 \mathrm{E}-11$ | 0.206818671 | 0.075142631 | 0.060415739 |  |  |  |  |  | 0 |
| 195.05139 | $4.28 \mathrm{E}-12$ | 0.056538268 | 0.051339729 | 0.061388003 | C4H8O5 | [M+Hac-H]- | 136.037175 | 195.0510286 | 1.85 | ['Threonate'] |
| 195.05139 | $4.28 \mathrm{E}-12$ | 0.056538268 | 0.051339729 | 0.061388003 | C6H12O7 | [M-H]- | 196.058305 | 195.0510286 | 1.85 | ['2-Carboxy-D-arabinitol', 'D-Altronate', 'DGluconic acid', 'D-Mannonate'] |
| 215.00519 | $2.79 \mathrm{E}-11$ | 0.082883135 | 0.06591649 | 0.063075443 | C3H8O3S2 | [M+Hac-H]- | 155.991489 | 215.0053426 | -0.71 | ['2-(Methylthio)ethanesulfonate'] |
| 799.23813 | $4.49 \mathrm{E}-11$ | 0.150298171 | 0.07226101 | 0.067589117 |  |  |  |  |  | 0 |
| 401.09429 | $2.92 \mathrm{E}-09$ | 0.124978605 | 0.065598927 | 0.068086103 |  |  |  |  |  | 0 |
| 403.10993 | 5.70E-07 | 0.116447098 | 0.067949052 | 0.070151245 |  |  |  |  |  | 0 |
| 549.16798 | $3.75 \mathrm{E}-10$ | 0.241225096 | 0.087366254 | 0.071603034 |  |  |  |  |  | 0 |
| 347.28066 | $1.33 \mathrm{E}-06$ | 0.150881041 | 0.06610394 | 0.071689294 |  |  |  |  |  | 0 |
| 550.17148 | $1.02 \mathrm{E}-09$ | 0.239304354 | 0.087924587 | 0.072525585 |  |  |  |  |  | 0 |
| 207.0494 | 0.008654954 | 0.160231481 | 0.076488629 | 0.074016551 |  |  |  |  |  | 0 |
| 153.06554 | $3.09 \mathrm{E}-05$ | 0.139289193 | 0.078667782 | 0.075339857 |  |  |  |  |  | 0 |
| 447.13628 | $3.50 \mathrm{E}-11$ | 0.119856297 | 0.073511781 | 0.075491426 | C2OH3ON2O5S | [M+K-2H]- | 410.187545 | 447.1361516 | 0.29 | ['Benfuracarb'] |
| 219.05151 | $5.01 \mathrm{E}-10$ | 0.182982357 | 0.094231059 | 0.076117718 |  |  |  |  |  | 0 |
| 267.07225 | $6.52 \mathrm{E}-12$ | 0.140726105 | 0.081730699 | 0.079198856 | C9H16O9 | [M-H]- | 268.079435 | 267.0721586 | 0.34 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 601.13955 | $1.02 \mathrm{E}-09$ | 0.181573233 | 0.093103371 | 0.081400273 |  |  |  |  |  | 0 |
| 293.10316 | $1.72 \mathrm{E}-09$ | 0.16210397 | 0.083880985 | 0.082945031 | C13H14O4 | [M+Hac-H]- | 234.08921 | 293.1030636 | 0.33 | ['1-Acetoxychavicol acetate', 'Coixinden B'] |
| 293.10316 | $1.72 \mathrm{E}-09$ | 0.16210397 | 0.083880985 | 0.082945031 | C15H18O6 | [M-H]- | 294.11034 | 293.1030636 | 0.33 | ['Tutin'] |
| 241.09297 | $7.14 \mathrm{E}-11$ | 0.180398462 | 0.102519536 | 0.084006749 | C12H16N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 204.126263 | 241.0927146 | 1.06 | ['Bufotenine', 'Caulophylline', 'Psilocin'] |
| 241.09297 | $7.14 \mathrm{E}-11$ | 0.180398462 | 0.102519536 | 0.084006749 | C6H14O6 | [M+Hac-H]- | 182.07904 | 241.0928936 | 0.32 | ['D-Iditol', 'D-Sorbitol', 'Galactitol', 'Mannitol'] |
| 253.09303 | $2.86 \mathrm{E}-11$ | 0.19835296 | 0.093804177 | 0.084356426 | C13H16N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 216.126263 | 253.0927146 | 1.25 | ['Girgensonine', 'Tetrahydroharmine'] |
| 253.09303 | $2.86 \mathrm{E}-11$ | 0.19835296 | 0.093804177 | 0.084356426 | C7H14O6 | [M+Hac-H]- | 194.07904 | 253.0928936 | 0.54 | ['(-)-Quebrachitol', '1-O-Methyl-myoinositol', '3-O-Methyl-myo-inositol', '4-0-Methyl-myo-inositol', '5-O-Methyl-myoinositol', '6-O-Methyl-myo-inositol', 'DPinitol', 'Methyl beta-D-galactoside', 'O-Methyl-scyllo-inositol'] |
| 253.09303 | $2.86 \mathrm{E}-11$ | 0.19835296 | 0.093804177 | 0.084356426 | C9H18O8 | [M-H]- | 254.10017 | 253.0928936 | 0.54 | ['2-(beta-D-Glucosyl)-sn-glycerol', '3-beta-D-Galactosyl-sn-glycerol'] |
| 313.11436 | $1.03 \mathrm{E}-09$ | 0.159974651 | 0.095005472 | 0.087081929 | C9H18O8 | [M+Hac-H]- | 254.10017 | 313.1140236 | 1.07 | ['2-(beta-D-Glucosyl)-sn-glycerol', '3-beta-D-Galactosyl-sn-glycerol'] |
| 471.14196 | 7.89E-06 | 0.013807634 | 0.078868163 | 0.088540887 | C26H26O7 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 450.167855 | 471.1425236 | -1.2 | ["4-Benzyloxy-2-hydroxy-3',4',5',6- |


|  |  |  |  |  |  |  |  |  |  | tetramethoxychalcone"] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 208.05476 | 3.87E-10 | 0.15850753 | 0.092174244 | 0.091720853 |  |  |  |  |  | 0 |
| 505.2121 | $1.32 \mathrm{E}-05$ | 0.293907487 | 0.111494174 | 0.092647365 | C18H36N4O11 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 484.238061 | 505.2127296 | -1.25 | ['Kanamycin A', 'Kanamycin C'] |
| 727.2162 | $1.84 \mathrm{E}-09$ | 0.217301758 | 0.100315415 | 0.09489121 |  |  |  |  |  | 0 |
| 177.04078 | $4.07 \mathrm{E}-11$ | 0.135734242 | 0.097132949 | 0.095329537 | C4H6O4 | [M+Hac-H]- | 118.02661 | 177.0404636 | 1.79 | ['Methyl oxalate', 'Methylmalonate', 'Succinate'] |
| 177.04078 | 4.07E-11 | 0.135734242 | 0.097132949 | 0.095329537 | C6H10O6 | [M-H]- | 178.04774 | 177.0404636 | 1.79 | ['2,4,6/3,5-Pentahydroxycyclohexanone', '2- <br> Dehydro-3-deoxy-D-galactonate', '2- <br> Dehydro-3-deoxy-D-gluconate', '2-Dehydro-D-glucose', '2-Deoxy-5-keto-D-gluconic acid', '3-Dehydro-2-deoxy-D-gluconate', '3-Keto-beta-D-galactose', '5-Dehydro-2-deoxy-Dgluconate', '5-Dehydro-D-fructose', '5-Deoxy glucuronic acid', 'D-Galactono-1,4-lactone', <br> 'D-Galactono-1,5-lactone', 'D-Glucono-1,4lactone', 'D-Glucono-1,5-lactone', 'D-galacto-Hexodialdose', 'Gulono-1,4-lactone', 'Hexose-1,5-lactone', 'L-Galactono-1,4lactone', 'L-Gulono-1,4-lactone', 'myo-Inosose-5'] |
| 196.01025 | $2.38 \mathrm{E}-07$ | 0.255151118 | 0.116332763 | 0.096308774 |  |  |  |  |  | 0 |
| 207.04887 | 0.192271166 | 0.203922085 | 0.100018597 | 0.096945125 |  |  |  |  |  | 0 |
| 565.16309 | $1.81 \mathrm{E}-09$ | 0.23452454 | 0.112702062 | 0.097771087 | C32H32O7 | [M+K-2H]- | 528.214805 | 565.1634116 | -0.57 | ['Karwinskione'] |
| 661.11705 | $3.02 \mathrm{E}-07$ | 0.388075599 | 0.110035477 | 0.098485985 | C27H30017 | [M+Cl]- | 626.148305 | 661.1177066 | -0.99 | ['Quercetin 3-O-beta-D-glucosyl-(1->2)-beta-D-glucoside'] |
| 325.07798 | 8.74E-10 | 0.291153964 | 0.120265491 | 0.099923288 |  |  |  |  |  | 0 |
| 175.06142 | $4.40 \mathrm{E}-09$ | 0.179155739 | 0.106633986 | 0.100043262 | C5H8O3 | [M+Hac-H]- | 116.047345 | 175.0611986 | 1.26 | ['2-Oxopentanoic acid', '3-Methyl-2oxobutanoic acid', '3-Oxopentanoic acid', '5Oxopentanoate'] |
| 175.06142 | $4.40 \mathrm{E}-09$ | 0.179155739 | 0.106633986 | 0.100043262 | C7H12O5 | [M-H]- | 176.068475 | 175.0611986 | 1.26 | ['(2R,3S)-3-Isopropylmalate', '(R)-2-(n-Propyl)-malate', '2-Propylmalate', '3PropyImalate', 'alpha-IsopropyImalate'] |
| 311.16633 | 0.000567328 | 0.274957849 | 0.124997908 | 0.100235903 |  |  |  |  |  | 0 |
| 176.06487 | 3.09E-09 | 0.169800627 | 0.104689951 | 0.10072411 |  |  |  |  |  | 0 |
| 255.23041 | $1.73 \mathrm{E}-05$ | 0.200331389 | 0.108090514 | 0.10202572 |  |  |  |  |  | 0 |
| 533.13687 | $1.18 \mathrm{E}-09$ | 0.168253996 | 0.109991227 | 0.103627661 |  |  |  |  |  | 0 |
| 445.12074 | $1.69 \mathrm{E}-11$ | 0.166852097 | 0.101146585 | 0.104227501 | C20H28N2O5S | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 408.171895 | 445.1205016 | 0.54 | ['Tamsulosin'] |
| 445.12074 | $1.69 \mathrm{E}-11$ | 0.166852097 | 0.101146585 | 0.104227501 | C23H24N2O4S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 424.14568 | 445.1203486 | 0.88 | ['Eprosartan'] |
| 207.0529 | $1.48 \mathrm{E}-09$ | 0.215060469 | 0.10778754 | 0.104289025 |  |  |  |  |  | 0 |
| 207.05244 | $3.49 \mathrm{E}-09$ | 0.3459751 | 0.107787547 | 0.104289032 |  |  |  |  |  | 0 |
| 326.08134 | $9.59 \mathrm{E}-10$ | 0.302100646 | 0.106775979 | 0.105423338 | C11H13N3O3S | [M+Hac-H]- | 267.067764 | 326.0816176 | -0.85 | ['Sulfisoxazole'] |


| 563.18385 | 7.53E-11 | 0.150184015 | 0.107552214 | 0.107471812 | C18H32O16 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 504.16904 | 563.1828936 | 1.7 | ['1F-beta-D-Fructosylsucrose', '6-alphaMaltosylglucose', '6F-alpha-D- <br> Galactosylsucrose', 'Cellotriose', 'D-Gal alpha 1->6D-Gal alpha 1->6D-Glucose', <br> 'Galactomannan', 'Isomaltotriose', <br> 'Maltotriose', 'Melezitose', 'Panose', <br> 'Raffinose', 'Umbelliferose', 'beta-D- <br> Fructofuranosyl O-beta-D-glucopyranosyl-(1-6)-alpha-D-glucopyranoside'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 563.18385 | 7.53E-11 | 0.150184015 | 0.107552214 | 0.107471812 | C32H32O7 | [M+Cl]- | 528.214805 | 563.1842066 | -0.63 | ['Karwinskione'] |
| 555.10601 | $1.72 \mathrm{E}-09$ | 0.212908622 | 0.117994652 | 0.109793106 |  |  |  |  |  | 0 |
| 463.3284 | $1.19 \mathrm{E}-07$ | 0.208102375 | 0.1062898 | 0.109934737 |  |  |  |  |  | 0 |
| 549.15949 | $2.24 \mathrm{E}-09$ | 0.218145255 | 0.119322984 | 0.111711353 | C24H32O13 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 528.184295 | 549.1589636 | 0.96 | ['Deoxyloganic acid tetraacetate'] |
| 179.05643 | $1.24 \mathrm{E}-12$ | 0.165783433 | 0.118732854 | 0.113267233 | C4H8O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 120.04226 | 179.0561136 | 1.77 | ['D-Erythrose', 'D-Erythrulose', 'D-Threose'] |
| 179.05643 | $1.24 \mathrm{E}-12$ | 0.165783433 | 0.118732854 | 0.113267233 | C6H12O6 | [M-H]- | 180.06339 | 179.0561136 | 1.77 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'DAldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'D-Fuconate', 'D-Galactose', 'D-Glucose', 'DGulose', 'D-Hamamelose', 'D-Hexose', 'DIdose', 'D-Mannose', 'D-Psicose', 'DSorbose', 'D-Tagatose', 'D-Talose', <br> 'Fructose(pyranose)', 'Ketose', 'L-Fructose', 'L-Fuconate', 'L-Galactose', 'L-Gulose', 'LRhamnonate', 'L-Sorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha-D-Glucose', 'alpha-DMannose', 'alpha-L-Sorbopyranose', 'beta-DFructose', 'beta-D-Galactose', 'beta-DGlucose', 'beta-D-Hamamelopyranose', 'beta-D-Mannose', 'muco-Inositol', 'myoInositol', 'scyllo-Inositol'] |
| 399.07867 | $2.54 \mathrm{E}-08$ | 0.234706605 | 0.11306597 | 0.117026159 | C14H17N2O4PS | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 340.064668 | 399.0785216 | 0.37 | ['Pyridafenthion'] |
| 464.15452 | $9.68 \mathrm{E}-07$ | 0.342749728 | 0.14025758 | 0.117727181 |  |  |  |  |  | 0 |
| 437.09158 | $3.99 \mathrm{E}-07$ | 0.623432324 | 0.142851868 | 0.118572821 | C23H18N2O5 | [M+Cl]- | 402.121573 | 437.0909746 | 1.39 | ['Saphenamycin'] |
| 725.23663 | $1.69 \mathrm{E}-11$ | 0.166322045 | 0.129258675 | 0.119182786 | C24H42O21 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 666.221865 | 725.2357186 | 1.26 | ['1,3-alpha-D-Mannosyl-1,2-alpha-D-mannosyl-1,2-alpha-D-mannosyl-Dmannose', 'Cellotetraose', 'Glycogen', 'Isolychnose', 'Lychnose', 'Maltotetraose', 'Stachyose', 'alpha-D-Galactosyl-(1-6)-alpha-D-galactosyl-(1-6)-beta-D-fructosyl-(2-1)-alpha-D-glucoside'] |
| 389.13058 | $5.54 \mathrm{E}-09$ | 0.282563766 | 0.127553855 | 0.119592558 | C17H26N2O4S | [M+Cl]- | 354.16133 | 389.1307316 | -0.39 | ['Sultopride'] |
| 679.20741 | $9.30 \mathrm{E}-06$ | 0.092614305 | 0.117734912 | 0.119795402 |  |  |  |  |  | 0 |
| 566.16667 | $2.96 \mathrm{E}-09$ | 0.231748516 | 0.131435186 | 0.12113899 |  |  |  |  |  | 0 |
| 445.18213 | 0.000373126 | 0.204927049 | 0.173077102 | 0.121550319 | C22H32O7 | [M+(37Cl)]- | 408.214805 | 445.1812566 | 1.96 | ['Cascarillin', 'Nigakilactone F'] |


| 606.19782 | 2.37E-11 | 0.169412792 | 0.129141592 | 0.122838168 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 571.10137 | $1.09 \mathrm{E}-08$ | 0.302679776 | 0.133642529 | 0.123057278 | C29H26O10 | [M+K-2H]- | 534.1526 | 571.1012066 | 0.29 | ['Cercosporin'] |
| 555.0979 | $5.16 \mathrm{E}-09$ | 0.284701706 | 0.136014702 | 0.123373931 |  |  |  |  |  | 0 |
| 473.07922 | $1.88 \mathrm{E}-09$ | 0.711138401 | 0.16425272 | 0.123785942 |  |  |  |  |  | 0 |
| 387.11491 | 2.72E-10 | 0.270685307 | 0.133857817 | 0.124387842 | C17H24N2O4S | [M+Cl]- | 352.14568 | 387.1150816 | -0.44 | ['Mercaptoacetyl-Phe-Leu'] |
| 295.06743 | $1.89 \mathrm{E}-10$ | 0.250105408 | 0.149402149 | 0.125557774 |  |  |  |  |  | 0 |
| 587.11108 | $1.12 \mathrm{E}-08$ | 0.106166699 | 0.130162595 | 0.125920409 | C24H26N2O13 | [M+(37C) $]$ ] | 550.143493 | 587.1099446 | 1.93 | ['Betanin', 'Gomphrenin-I', 'Isobetanin'] |
| 388.11829 | $2.52 \mathrm{E}-10$ | 0.271781525 | 0.134266105 | 0.12701935 | C19H21N5O2 | [M+K-2H]- | 351.169525 | 388.1181316 | 0.41 | ['Pirenzepine'] |
| 388.11829 | $2.52 \mathrm{E}-10$ | 0.271781525 | 0.134266105 | 0.12701935 | C23H19NO5 | [M-H]- | 389.126324 | 388.1190476 | -1.95 | ['(+/-)-6-Acetonyldihydrosanguinarine'] |
| 461.13077 | $1.40 \mathrm{E}-11$ | 0.219489486 | 0.132722132 | 0.128969413 |  |  |  |  |  | 0 |
| 462.27163 | 0.031506332 | 0.195423403 | 0.075487839 | 0.129270103 |  |  |  |  |  | 0 |
| 415.07357 | $9.42 \mathrm{E}-07$ | 0.185519912 | 0.123389127 | 0.129315866 | C17H2ON2O6S | [M+Cl]- | 380.10421 | 415.0736116 | -0.1 | ['Methicillin'] |
| 445.1793 | 0.000225892 | 0.141290623 | 0.145254425 | 0.130650729 | C25H30O5 | [M+Cl]- | 410.209325 | 445.1787266 | 1.29 | ['Vismione D'] |
| 207.04832 | $2.01 \mathrm{E}-07$ | 0.299470054 | 0.124961556 | 0.132826007 |  |  |  |  |  | 0 |
| 617.1219 | $5.59 \mathrm{E}-09$ | 0.174768161 | 0.1192817 | 0.134550179 | C34H28O9 | [M+K-2H]- | 580.173335 | 617.1219416 | -0.07 | ['Mulberrofuran C'] |
| 445.1873 | $2.06 \mathrm{E}-08$ | 0.164840874 | 0.138513395 | 0.135978849 | C22H26O6 | [M+Hac-H]- | 386.17294 | 445.1867936 | 1.14 | ['(+)-Eudesmin', 'Burseran'] |
| 445.1873 | $2.06 \mathrm{E}-08$ | 0.164840874 | 0.138513395 | 0.135978849 | C24H3008 | [M-H]- | 446.19407 | 445.1867936 | 1.14 | ['Estrone glucuronide', 'Yangambin'] |
| 321.02321 | $2.43 \mathrm{E}-08$ | 0.255480124 | 0.130266484 | 0.13726328 | C14H14NO4PS | [M-2H]- | 323.038119 | 321.0235662 | -1.11 | ['EPN'] |
| 321.02321 | $2.43 \mathrm{E}-08$ | 0.255480124 | 0.130266484 | 0.13726328 | C6H15O5PS2 | [M+Hac-H]- | 262.009857 | 321.0237106 | -1.56 | ['Demeton-S-methylsulphon'] |
| 529.07349 | 4.17E-09 | 0.228140774 | 0.137345263 | 0.140801083 |  |  |  |  |  | 0 |
| 453.06601 | $1.12 \mathrm{E}-08$ | 0.484691323 | 0.151698817 | 0.141761157 | C13H19N4O12P | [M-H]- | 454.073714 | 453.0664376 | -0.94 | ['1-(5-Phosphoribosyl)-5-amino-4-(N-succinocarboxamide)-imidazole'] |
| 453.06601 | $1.12 \mathrm{E}-08$ | 0.484691323 | 0.151698817 | 0.141761157 | C15H20N4O8S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 416.100188 | 453.0666396 | -1.39 | ['O-Carbamoyl-deacetylcephalosporin C'] |
| 483.0765 | $5.34 \mathrm{E}-07$ | 0.197216796 | 0.128115735 | 0.142011575 | C16H22N4O9S | [M+(37Cl)]- | 446.110753 | 483.0772046 | -1.46 | ['Cephamycin C'] |
| 268.11197 | $1.14 \mathrm{E}-10$ | 0.191525674 | 0.125557732 | 0.142207832 |  |  |  |  |  | 0 |
| 207.05344 | 0.003781863 | 0.298613004 | 0.14661007 | 0.142667309 |  |  |  |  |  | 0 |
| 459.13629 | $7.24 \mathrm{E}-11$ | 0.218579431 | 0.154048299 | 0.14421659 |  |  |  |  |  | 0 |
| 192.05982 | $1.20 \mathrm{E}-09$ | 0.313369854 | 0.168718416 | 0.14526497 |  |  |  |  |  | 0 |
| 605.19426 | $1.93 \mathrm{E}-13$ | 0.18394206 | 0.142639057 | 0.147809402 |  |  |  |  |  | 0 |
| 460.13967 | 3.05E-10 | 0.201778688 | 0.142789618 | 0.148145063 | C17H29NO11 | [M+(37C) $]$ - | 423.174064 | 460.1405156 | -1.84 | ['Neolinustatin'] |
| 223.04345 | $2.73 \mathrm{E}-07$ | 0.315771397 | 0.154850674 | 0.148914673 |  |  |  |  |  | 0 |
| 519.22784 | 0.000578357 | 0.49327588 | 0.181741693 | 0.149017224 |  |  |  |  |  | 0 |
| 453.08642 | $2.75 \mathrm{E}-05$ | 0.547482484 | 0.187459452 | 0.149309733 |  |  |  |  |  | 0 |
| 202.07254 | $2.81 \mathrm{E}-10$ | 0.288709663 | 0.173937785 | 0.149354257 | C9H13N2O2 | [M+Na-2H]- | 181.097703 | 202.0723716 | 0.83 | ['Pyridostigmine'] |
| 186.11252 | 0.008378706 | 0.112593668 | 0.042966029 | 0.152555408 | C7H16N4O2 | [M-2H]- | 188.127326 | 186.1127732 | -1.36 | ['Homoarginine', 'Ngamma-Monomethyl-Larginine'] |
| 533.17315 | $3.12 \mathrm{E}-10$ | 0.349179011 | 0.174261044 | 0.153522697 |  |  |  |  |  | 0 |
| 209.05572 | $3.66 \mathrm{E}-09$ | 0.281249754 | 0.164311259 | 0.154027393 |  |  |  |  |  | 0 |


| 520.23132 | 0.000857341 | 0.510975489 | 0.187072962 | 0.1551474 | C27H38010 | [M-2H]- | 522.2465 | 520.2319472 | -1.21 | ['Trilobolide'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 489.11074 | $2.22 \mathrm{E}-08$ | 0.309698628 | 0.161157528 | 0.15693472 | C20H26N2O8S | [M+Cl]- | 454.14099 | 489.1103916 | 0.71 | ['Sirodesmin H '] |
| 489.11074 | $2.22 \mathrm{E}-08$ | 0.309698628 | 0.161157528 | 0.15693472 | C28H2206 | [M+CI]- | 454.14164 | 489.1110416 | -0.62 | ['Gnetin A', 'epsilon-Viniferin'] |
| 475.16744 | $1.01 \mathrm{E}-09$ | 0.313605543 | 0.177772706 | 0.156937526 |  |  |  |  |  | 0 |
| 321.08301 | $3.92 \mathrm{E}-07$ | 0.289071576 | 0.160027408 | 0.158636224 | C16H16N2O3 | [M+(37Cl)]- | 284.116093 | 321.0825446 | 1.45 | ['Metominostrobin'] |
| 321.08301 | 3.92E-07 | 0.289071576 | 0.160027408 | 0.158636224 | C17H20N2S | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]$ - | 284.13472 | 321.0833266 | -0.99 | ['Promazine', 'Promethazine'] |
| 341.06958 | 8.39E-08 | 0.285603355 | 0.146178739 | 0.162200026 | C19H16N2O2 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 304.121178 | 341.0697846 | -0.6 | ['Ceceline'] |
| 446.27671 | 0.015210175 | 0.222029583 | 0.094452282 | 0.162659041 |  |  |  |  |  | 0 |
| 396.34887 | 0.001730809 | 0.31099978 | 0.159031123 | 0.164364921 |  |  |  |  |  | 0 |
| 460.25591 | 0.066961158 | 0.190450987 | 0.091570441 | 0.165205051 |  |  |  |  |  | 0 |
| 371.12001 | $1.69 \mathrm{E}-11$ | 0.230845377 | 0.163166294 | 0.165395505 | C11H20010 | [M+Hac-H]- | 312.10565 | 371.1195036 | 1.36 | ['6-O-(beta-D-Xylopyranosyl)-beta-Dglucopyranose', '6-O-beta-D-Xylopyranosyl-D-glucose', 'Arabino-galactose', 'Vicianose'] |
| 478.26654 | 0.047619382 | 0.171373535 | 0.089577751 | 0.165620973 |  |  |  |  |  | - |
| 323.09871 | 1.27E-10 | 0.262748958 | 0.161847721 | 0.167090986 | C12H20010 | [M-H]- | 324.10565 | 323.0983736 | 1.04 | ["Bis-D-fructose 2',1:2,1-dianhydride", "DFructofuranose 1,2':2,3-dianhydride"] |
| 323.09871 | 1.27E-10 | 0.262748958 | 0.161847721 | 0.167090986 | C16H18N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})]^{-}$ | 286.131743 | 323.0981946 | 1.6 | ['Cromakalim', 'Difenoxuron', 'Levcromakalim', 'Pilosine'] |
| 323.09871 | 1.27E-10 | 0.262748958 | 0.161847721 | 0.167090986 | C8H24N4O3P2 | [M+(37C) $]^{-}$ | 286.132367 | 323.0988186 | -0.34 | ['Schradan'] |
| 267.10862 | $1.47 \mathrm{E}-10$ | 0.212058175 | 0.161099801 | 0.167849609 | C14H18N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 230.141913 | 267.1083646 | 0.96 | ['Camoensine'] |
| 477.14687 | $2.35 \mathrm{E}-06$ | 0.360863868 | 0.158614948 | 0.169118014 |  |  |  |  |  | 0 |
| 339.19716 | 0.000336891 | 0.461390107 | 0.213496187 | 0.169133172 | C22H28O3 | [M-H]- | 340.203845 | 339.1965686 | 1.74 | ['17-Hydroxy-3-oxo-17alpha-pregna-1,4 diene-21-carboxylic acid, gamma-lactone', 'Norethindrone acetate'] |
| 305.02828 | 5.26E-08 | 0.402039626 | 0.170630851 | 0.169977117 | C6H1504PS2 | [ $\mathrm{M}+\mathrm{Hac-H}$ ] | 246.014942 | 305.0287956 | -1.69 | ['Oxydemeton-methyl'] |
| 305.02828 | 5.26E-08 | 0.402039626 | 0.170630851 | 0.169977117 | C9H1609 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 268.079435 | 305.0280416 | 0.78 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 417.12559 | $2.62 \mathrm{E}-06$ | 0.294769827 | 0.205368663 | 0.170276529 | C18H26N2O5S | [M+Cl]- | 382.156245 | 417.1256466 | -0.14 | ['Furathiocarb'] |
| 262.09333 | 1.66E-09 | 0.320489828 | 0.192801808 | 0.171392231 | C14H15N3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 225.126597 | 262.0930486 | 1.07 | ['Cyprodinil', 'ortho-Aminoazotoluene', 'para-(Dimethylamino)azobenzene'] |
| 262.09333 | 1.66E-09 | 0.320489828 | 0.192801808 | 0.171392231 | C 8 H 13 NO 5 | [M+Hac-H]- | 203.079374 | 262.0932276 | 0.39 | ['N2-Acetyl-L-aminoadipate'] |
| 240.08066 | $1.70 \mathrm{E}-10$ | 0.275261645 | 0.187774532 | 0.172443871 |  |  |  |  |  | 0 |
| 341.07285 | 2.51E-09 | 0.304861725 | 0.16694804 | 0.175350997 |  |  |  |  |  | 0 |
| 651.1639 | $1.35 \mathrm{E}-10$ | 0.278147953 | 0.176136077 | 0.175943429 |  |  |  |  |  | 0 |
| 695.31327 | $9.08 \mathrm{E}-09$ | 0.267078563 | 0.201753162 | 0.176307268 |  |  |  |  |  | 0 |
| 534.24688 | 0.000493546 | 0.536092585 | 0.211369194 | 0.17717667 |  |  |  |  |  | 0 |


| 239.07731 | $1.12 \mathrm{E}-10$ | 0.306502922 | 0.198685718 | 0.178200051 | C6H12O6 | [M+Hac-H]- | 180.06339 | 239.0772436 | 0.28 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'DAldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'D-Fuconate', 'D-Galactose', 'D-Glucose', 'DGulose', 'D-Hamamelose', 'D-Hexose', 'DIdose', 'D-Mannose', 'D-Psicose', 'DSorbose', 'D-Tagatose', 'D-Talose', <br> 'Fructose(pyranose)', 'Ketose', 'L-Fructose', 'L-Fuconate', 'L-Galactose', 'L-Gulose', 'LRhamnonate', 'L-Sorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha-D-Glucose', 'alpha-DMannose', 'alpha-L-Sorbopyranose', 'beta-DFructose', 'beta-D-Galactose', 'beta-DGlucose', 'beta-D-Hamamelopyranose', 'beta-D-Mannose', 'muco-Inositol', 'myoInositol', 'scyllo-Inositol'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 309.11947 | 7.91E-11 | 0.328766764 | 0.192061619 | 0.181556467 | C16H2ON2O2 | [M+(37Cl)]- | 272.152478 | 309.1189296 | 1.75 | ['4-(3-Methylbut-2-enyl)-L-tryptophan', 'Dimethylallyltryptophan'] |
| 533.24336 | 0.001538379 | 0.559270459 | 0.211596767 | 0.181799398 | C26H42NO7S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 512.268201 | 533.2428696 | 0.92 | ['Sulfoglycolithocholate'] |
| 200.12961 | 0.001322894 | 0.177780396 | 0.096442619 | 0.182861947 | C8H15NO | [M+Hac-H]- | 141.115364 | 200.1292176 | 1.96 | ['(-)-Hygrine', 'Hygrine', 'Pelletierine', 'Physoperuvine', 'Trachelanthamidine', 'Tropine'] |
| 314.27041 | $7.43 \mathrm{E}-05$ | 0.279156614 | 0.16416566 | 0.183215598 |  |  |  |  |  | 0 |
| 326.27041 | 0.000110993 | 0.348192373 | 0.202101311 | 0.18347147 |  |  |  |  |  | 0 |
| 535.11623 | $4.36 \mathrm{E}-05$ | 0.400722188 | 0.186811389 | 0.18368769 |  |  |  |  |  | 0 |
| 327.09361 | 6.65E-10 | 0.262299198 | 0.178058044 | 0.185606325 | C9H16O9 | [M+Hac-H]- | 268.079435 | 327.0932886 | 0.98 | ['2(alpha-D-Mannosyl)-D-glycerate'] |
| 591.14277 | $8.22 \mathrm{E}-10$ | 0.272142168 | 0.161281848 | 0.186492966 |  |  |  |  |  | 0 |
| 249.06173 | $2.31 \mathrm{E}-11$ | 0.401460938 | 0.192445669 | 0.186967812 | C13H12N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 212.094963 | 249.0614146 | 1.27 | ['Harmine', 'Pyocyanine'] |
| 249.06173 | $2.31 \mathrm{E}-11$ | 0.401460938 | 0.192445669 | 0.186967812 | C7H1006 | [M+Hac-H]- | 190.04774 | 249.0615936 | 0.55 | ['2,4-Dihydroxyhept-2-enedioate', '3- <br> Dehydroquinate', '4-Hydroxy-2-oxoheptanedioate', '4-Hydroxy-4-methyl-2oxoadipate'] |
| 649.14817 | $2.04 \mathrm{E}-08$ | 0.294668822 | 0.19068029 | 0.187763668 |  |  |  |  |  | 0 |
| 342.07628 | $1.20 \mathrm{E}-08$ | 0.311498272 | 0.177002835 | 0.187884503 | C17H15N5O | [M+K-2H]- | 305.12766 | 342.0762666 | 0.04 | ['Zaleplon'] |
| 457.0843 | $9.14 \mathrm{E}-10$ | 0.674923098 | 0.238430542 | 0.187890051 |  |  |  |  |  | 0 |
| 519.12124 | 3.32E-09 | 0.464728687 | 0.208792792 | 0.191370604 |  |  |  |  |  | 0 |
| 534.17673 | $6.23 \mathrm{E}-11$ | 0.402073673 | 0.211157844 | 0.191510455 | C20H35NO13 | [ $\mathrm{M}+(37 \mathrm{Cl})]-$ | 497.210844 | 534.1772956 | -1.06 | ['Validamycin A'] |
| 237.06168 | $9.39 \mathrm{E}-09$ | 0.317005072 | 0.202711873 | 0.192354019 | C12H12N2O | [M+(37C) $]$ - | 200.094963 | 237.0614146 | 1.12 | ['4,4-Diaminodiphenyl ether', 'Harmalol'] |


| 237.06168 | 9.39E-09 | 0.317005072 | 0.202711873 | 0.192354019 | C6H1006 | [M+Hac-H]- | 178.04774 | 237.0615936 | 0.36 | ['2,4,6/3,5-Pentahydroxycyclohexanone', '2- <br> Dehydro-3-deoxy-D-galactonate', '2- <br> Dehydro-3-deoxy-D-gluconate', '2-Dehydro-D-glucose', '2-Deoxy-5-keto-D-gluconic acid', '3-Dehydro-2-deoxy-D-gluconate', '3-Keto-beta-D-galactose', '5-Dehydro-2-deoxy-Dgluconate', '5-Dehydro-D-fructose', '5-Deoxy glucuronic acid', 'D-Galactono-1,4-lactone', 'D-Galactono-1,5-lactone', 'D-Glucono-1,4lactone', 'D-Glucono-1,5-lactone', 'D-galacto-Hexodialdose', 'Gulono-1,4-lactone', 'Hexose-1,5-lactone', 'L-Galactono-1,4lactone', 'L-Gulono-1,4-lactone', 'myo-Inosose-5'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 237.06168 | $9.39 \mathrm{E}-09$ | 0.317005072 | 0.202711873 | 0.192354019 | C8H14O8 | [M-H]- | 238.06887 | 237.0615936 | 0.36 | ['3-Deoxy-D-manno-octulosonate'] |
| 205.07218 | $8.38 \mathrm{E}-10$ | 0.473539503 | 0.23023341 | 0.19424863 |  |  |  |  |  | 0 |
| 203.05654 | $9.62 \mathrm{E}-09$ | 0.336299332 | 0.2319242 | 0.194425103 |  |  |  |  |  | 0 |
| 327.2657 | 0.000214538 | 0.363880153 | 0.210790591 | 0.195390142 |  |  |  |  |  | 0 |
| 312.25477 | 0.000689718 | 0.31385083 | 0.205744252 | 0.198699 | C18H35NO3 | [M-H]- | 313.261694 | 312.2544176 | 1.13 | ['(+)-Prosopinine'] |
| 223.04398 | $4.36 \mathrm{E}-10$ | 0.388627349 | 0.223620043 | 0.199294136 |  |  |  |  |  | 0 |
| 469.06081 | $2.01 \mathrm{E}-07$ | 0.455248885 | 0.229579778 | 0.200615483 | C15H20N4O9S | [M+(37Cl)]- | 432.095103 | 469.0615546 | -1.59 | ['7a-Hydroxy-O-carbamoyldeacetylcephalosporin $\mathrm{C}^{\prime}$ ] |
| 186.11385 | 0.008951729 | 0.070804779 | 0.051082966 | 0.201681397 | C7H13NO | [M+Hac-H]- | 127.099714 | 186.1135676 | 1.52 | ['N-Cyclohexylformamide'] |
| 186.11385 | 0.008951729 | 0.070804779 | 0.051082966 | 0.201681397 | C9H17NO3 | [M-H]- | 187.120844 | 186.1135676 | 1.52 | ['(E)-2-Butenyl-4-methyl-threonine', '8-Amino-7-oxononanoate', 'N2-Acetyl-Llysine'] |
| 520.16102 | 4.02E-09 | 0.404443637 | 0.210901388 | 0.202235327 |  |  |  |  |  | 0 |
| 592.1459 | 3.08E-10 | 0.256310781 | 0.181289639 | 0.203193972 |  |  |  |  |  | 0 |
| 205.03583 | $1.55 \mathrm{E}-11$ | 0.168867642 | 0.192812829 | 0.203933116 |  |  |  |  |  | 0 |
| 458.12395 | $4.53 \mathrm{E}-10$ | 0.474368176 | 0.236143756 | 0.204091187 |  |  |  |  |  | 0 |
| 457.12049 | $4.80 \mathrm{E}-10$ | 0.47451844 | 0.235765773 | 0.204470695 | C20H24N2O8 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 420.153268 | 457.1197196 | 1.69 | ['2-N,6-N-Bis(2,3-dihydroxy-N-benzoyl)-Lserine'] |
| 328.2861 | $7.79 \mathrm{E}-05$ | 0.30487013 | 0.194071981 | 0.206398976 | C17H35NO | [M+Hac-H]- | 269.271864 | 328.2857176 | 1.16 | ['Capsi-amide'] |
| 461.15191 | $3.65 \mathrm{E}-09$ | 0.611495276 | 0.24530262 | 0.207706102 |  |  |  |  |  | 0 |
| 529.08229 | $3.26 \mathrm{E}-08$ | 0.402992496 | 0.205526621 | 0.208902923 |  |  |  |  |  | 0 |
| 326.23405 | 0.034159797 | 0.274475177 | 0.140394139 | 0.209044445 |  |  |  |  |  | 0 |
| 547.25904 | 0.002668841 | 0.616855884 | 0.266308142 | 0.210165583 |  |  |  |  |  | 0 |
| 513.08727 | $1.34 \mathrm{E}-08$ | 0.622846283 | 0.225858875 | 0.212667645 | C13H19N4O12P | [M+Hac-H]- | 454.073714 | 513.0875676 | -0.58 | ['1-(5-Phosphoribosyl)-5-amino-4-(N-succinocarboxamide)-imidazole'] |
| 519.15753 | $1.86 \mathrm{E}-09$ | 0.400057042 | 0.221445581 | 0.212870217 |  |  |  |  |  | 0 |
| 205.03342 | $1.19 \mathrm{E}-05$ | 0.256621674 | 0.175401365 | 0.214065347 |  |  |  |  |  | 0 |


| 328.24968 | 0.003757131 | 0.288993959 | 0.162515853 | 0.214733185 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 371.26386 | 0.029536825 | 0.296039416 | 0.18499692 | 0.214873888 |  |  |  |  |  | 0 |
| 191.05638 | $3.89 \mathrm{E}-09$ | 0.440378898 | 0.231357274 | 0.217954356 | C5H8O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 132.04226 | 191.0561136 | 1.39 | ['(4S)-4,5-Dihydroxypentan-2,3-dione', '(S)-2-Acetolactate', '2-(Hydroxymethyl)-4oxobutanoate', '2-Acetolactate', '3-Hydroxy-3-methyl-2-oxobutanoic acid', '4-Hydroxy-2oxopentanoate', 'Deoxyribonolactone', 'Glutarate'] |
| 191.05638 | 3.89E-09 | 0.440378898 | 0.231357274 | 0.217954356 | C7H12O6 | [M-H]- | 192.06339 | 191.0561136 | 1.39 | ['2D-5-O-Methyl-2,3,5/4,6pentahydroxycyclohexanone', 'Quinate', 'Valiolone'] |
| 445.15717 | $1.04 \mathrm{E}-09$ | 0.462481132 | 0.18751109 | 0.219387826 | C20H30N2O5S | [M+Cl]- | 410.187545 | 445.1569466 | 0.5 | ['Benfuracarb'] |
| 439.07749 | $2.29 \mathrm{E}-07$ | 0.368242052 | 0.236323157 | 0.223250409 |  |  |  |  |  | 0 |
| 368.24479 | 0.025239854 | 0.259072143 | 0.164009026 | 0.227008966 |  |  |  |  |  | 0 |
| 399.25058 | $1.90 \mathrm{E}-05$ | 0.275264297 | 0.198171653 | 0.233209577 |  |  |  |  |  | 0 |
| 311.09872 | $3.39 \mathrm{E}-10$ | 0.292177966 | 0.22243845 | 0.233210718 | C11H20010 | [M-H]- | 312.10565 | 311.0983736 | 1.11 | ['6-O-(beta-D-Xylopyranosyl)-beta-Dglucopyranose', '6-O-beta-D-Xylopyranosyl-D-glucose', 'Arabino-galactose', 'Vicianose'] |
| 311.09872 | 3.39E-10 | 0.292177966 | 0.22243845 | 0.233210718 | C7H16N8O4 | [M+Cl]- | 276.129452 | 311.0988536 | -0.43 | ['Trimethylenetetraurea'] |
| 745.4872 | 0.080534177 | 0.290001891 | 0.162389862 | 0.234256507 |  |  |  |  |  | 0 |
| 607.46046 | 0.008345374 | 0.634567228 | 0.300117447 | 0.236728052 |  |  |  |  |  | 0 |
| 677.11147 | 0.000179672 | 0.792150361 | 0.292971073 | 0.240517172 |  |  |  |  |  | 0 |
| 374.2829 | 0.079392584 | 0.253073942 | 0.174797196 | 0.242924056 | C24H40O3 | [M-2H]- | 376.297745 | 374.2831922 | -0.78 | ['Isolithocholate', 'Lithocholic acid'] |
| 592.44168 | 0.016502644 | 0.683661941 | 0.334514602 | 0.243911949 |  |  |  |  |  | 0 |
| 666.18265 | $1.91 \mathrm{E}-09$ | 0.860676671 | 0.302690849 | 0.245031815 |  |  |  |  |  | 0 |
| 370.26045 | 0.04281981 | 0.328019651 | 0.185778953 | 0.24707317 |  |  |  |  |  | 0 |
| 665.21513 | $4.31 \mathrm{E}-12$ | 0.320724378 | 0.246276003 | 0.247584441 | C24H42O21 | [M-H]- | 666.221865 | 665.2145886 | 0.81 | ['1,3-alpha-D-Mannosyl-1,2-alpha-D-mannosyl-1,2-alpha-D-mannosyl-Dmannose', 'Cellotetraose', 'Glycogen', 'Isolychnose', 'Lychnose', 'Maltotetraose', 'Stachyose', 'alpha-D-Galactosyl-(1-6)-alpha-D-galactosyl-(1-6)-beta-D-fructosyl-(2-1)-alpha-D-glucoside'] |
| 665.21513 | 4.31E-12 | 0.320724378 | 0.246276003 | 0.247584441 | C37H4009 | [M+K-2H]- | 628.267235 | 665.2158416 | -1.07 | ['Resiniferatoxin'] |
| 606.45728 | 0.010683619 | 0.691024884 | 0.314762186 | 0.249805293 |  |  |  |  |  | 0 |
| 374.29173 | 0.029900715 | 0.272333208 | 0.178090486 | 0.250036235 |  |  |  |  |  | 0 |
| 535.22271 | 0.002583654 | 0.717689014 | 0.287057789 | 0.251143706 |  |  |  |  |  | 0 |
| 196.04663 | 0.000239887 | 0.455234815 | 0.290284274 | 0.252479219 |  |  |  |  |  | 0 |
| 262.1661 | 0.000462728 | 0.198524749 | 0.1202221 | 0.25311304 |  |  |  |  |  | 0 |
| 376.23465 | 0.131159736 | 0.242863813 | 0.259294368 | 0.254325834 |  |  |  |  |  | 0 |


| 207.05113 | 7.87E-08 | 0.488098892 | 0.260571104 | 0.254781529 | C5H8O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 148.037175 | 207.0510286 | 0.49 | ['(R)-2-Hydroxyglutarate', (R)-2- <br> Methylmalate', '(S)-2-Hydroxyglutarate', '(S)- <br> 2-Methylmalate', '2-Dehydro-3-deoxy-Dxylonate', '2-Dehydro-3-deoxy-Larabinonate', '2-Hydroxyglutarate', <br> 'Citramalate', 'D-Arabinono-1,4-lactone', 'D-Xylono-1,4-lactone', 'D-Xylonolactone', 'D-erythro-3-Methylmalate', 'D-threo-3- <br> Methylmalate', 'L-Arabinono-1,4-lactone', 'L-Arabinono-1,5-lactone', 'L-Xylono-1,4lactone', 'L-threo-3-Methylmalate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 729.49239 | 0.112175319 | 0.31973611 | 0.181039442 | 0.25562958 |  |  |  |  |  | 0 |
| 338.27038 | 0.000732058 | 0.468767068 | 0.322829018 | 0.255660135 |  |  |  |  |  | 0 |
| 342.22892 | 0.037461259 | 0.280391536 | 0.23189892 | 0.256535327 |  |  |  |  |  | 0 |
| 339.27379 | 0.000707686 | 0.465498683 | 0.320302293 | 0.256781331 |  |  |  |  |  | 0 |
| 343.12493 | 3.08E-10 | 0.251148028 | 0.26238229 | 0.256857184 | C12H24O11 | [M-H]- | 344.131865 | 343.1245886 | 0.99 | ['Clusianose', 'Melibiitol'] |
| 343.12493 | 3.08E-10 | 0.251148028 | 0.26238229 | 0.256857184 | C2OH22N2S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 322.15037 | 343.1250386 | -0.32 | ['Mequitazine'] |
| 593.44485 | 0.015265747 | 0.683420968 | 0.326270796 | 0.257102773 |  |  |  |  |  | 0 |
| 373.27942 | 0.108954381 | 0.30535211 | 0.170783103 | 0.259336649 |  |  |  |  |  | 0 |
| 385.13573 | $2.14 \mathrm{E}-10$ | 0.404606006 | 0.291269684 | 0.261309595 | C12H22O10 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 326.1213 | 385.1351536 | 1.5 | ['2-O-alpha-L-Rhamnopyranosyl-Dglucopyranose', 'Robinobiose', 'Rutinose'] |
| 385.13573 | 2.14E-10 | 0.404606006 | 0.291269684 | 0.261309595 | C13H18N4O6 | [M+Hac-H]- | 326.122636 | 385.1364896 | -1.97 | ['6,7-Dimethyl-8-(D-ribity)/lumazine'] |
| 385.13573 | $2.14 \mathrm{E}-10$ | 0.404606006 | 0.291269684 | 0.261309595 | C18H24N2O5 | [M+(37C) $]$ - | 348.168523 | 385.1349746 | 1.96 | ['Enalaprilate', 'Funebrine'] |
| 372.27583 | 0.110921536 | 0.305366805 | 0.171357939 | 0.261374344 | C18H35NO3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 313.261694 | 372.2755476 | 0.76 | ['(+)-Prosopinine'] |
| 537.50905 | 5.38E-05 | 0.446944979 | 0.271351303 | 0.262680345 |  |  |  |  |  | 0 |
| 536.50573 | 7.79E-05 | 0.453643844 | 0.266184825 | 0.264754937 |  |  |  |  |  | 0 |
| 611.08809 | 4.34E-07 | 0.39696814 | 0.273821408 | 0.267529968 |  |  |  |  |  | 0 |
| 517.14186 | 2.86E-11 | 0.297913281 | 0.25092289 | 0.2679909 |  |  |  |  |  | 0 |
| 593.22502 | 0.000290318 | 0.512659721 | 0.340211347 | 0.268167806 | C27H34011 | [M+Hac-H]- | 534.210115 | 593.2239686 | 1.77 | ['Arctiin', 'Forsythin', 'Undulatone'] |
| 340.29026 | 0.000181274 | 0.427631098 | 0.284727863 | 0.268759818 |  |  |  |  |  | 0 |
| 397.10331 | 0.057478696 | 0.294004452 | 0.263787352 | 0.271432353 |  |  |  |  |  | 0 |
| 453.07452 | 5.25E-09 | 0.263772386 | 0.221025449 | 0.273684489 |  |  |  |  |  | 0 |
| 340.28574 | 0.001470604 | 0.461237037 | 0.303866076 | 0.277258659 | C18H35NO | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 281.271864 | 340.2857176 | 0.07 | ['Dodemorph'] |
| 319.06741 | 8.27E-07 | 0.685500708 | 0.328302434 | 0.278263529 | C16H14N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})]-$ | 282.100443 | 319.0668946 | 1.62 | ['Saphenic acid methyl ester'] |
| 372.27008 | 0.090837552 | 0.340441328 | 0.224313989 | 0.278907501 |  |  |  |  |  | 0 |
| 585.06374 | 2.11E-05 | 0.624977637 | 0.326402137 | 0.281510574 |  |  |  |  |  | 0 |
| 795.18942 | 5.64E-08 | 0.181464031 | 0.213763525 | 0.282102217 |  |  |  |  |  | 0 |
| 719.13114 | 1.25E-06 | 0.668746972 | 0.373576178 | 0.282411198 |  |  |  |  |  | 0 |
| 655.49166 | 0.122839501 | 0.422297759 | 0.215248161 | 0.282915857 |  |  |  |  |  | 0 |
| 549.23833 | 0.006858522 | 0.718708207 | 0.33820218 | 0.283129628 | C24H34N4O5S | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 490.224993 | 549.2388466 | -0.94 | ['Glimepiride'] |


| 341.28932 | 0.001424711 | 0.455198361 | 0.311466972 | 0.284213915 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 390.28668 | 0.06278445 | 0.336148971 | 0.202848538 | 0.284665452 |  |  |  |  |  | 0 |
| 665.17925 | $1.92 \mathrm{E}-09$ | 0.860100752 | 0.337441788 | 0.286286868 | C36H36O10 | [M+K-2H]- | 628.23085 | 665.1794566 | -0.31 | ['Gnidicin'] |
| 613.12675 | $2.38 \mathrm{E}-07$ | 0.324113114 | 0.250199108 | 0.287446183 |  |  |  |  |  | 0 |
| 596.52701 | 0.042087911 | 0.421717046 | 0.191250568 | 0.288346152 |  |  |  |  |  | 0 |
| 223.04482 | 0.000625661 | 0.468731761 | 0.273959257 | 0.290003339 |  |  |  |  |  | 0 |
| 402.2503 | 0.072347946 | 0.319274782 | 0.193786662 | 0.290478787 |  |  |  |  |  | 0 |
| 294.24394 | 0.085324266 | 0.321995254 | 0.200933294 | 0.291257498 |  |  |  |  |  | 0 |
| 277.05661 | 5.23E-07 | 0.446974503 | 0.251592264 | 0.29261993 | C11H14NO6 | [M+Na-2H]- | 256.082114 | 277.0567826 | -0.62 | ['Nicotinate D-ribonucleoside'] |
| 449.07093 | 3.87E-06 | 0.641107767 | 0.315286305 | 0.295585034 |  |  |  |  |  | 0 |
| 308.22349 | 0.099221168 | 0.331524548 | 0.184975969 | 0.296188273 |  |  |  |  |  | 0 |
| 201.02593 | $2.90 \mathrm{E}-05$ | 0.160919725 | 0.22686788 | 0.297286404 | C8H1ON2S | [ $\mathrm{M}+\mathrm{Cl}]$ - | 166.05647 | 201.0258716 | 0.29 | ['Ethionamide'] |
| 356.28104 | 0.000165059 | 0.378841059 | 0.239975243 | 0.29990542 | C18H35NO2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 297.266779 | 356.2806326 | 1.14 | ['3-Ketosphingosine', 'Cassine', 'Spiroxamine'] |
| 400.23465 | 0.061844571 | 0.33455239 | 0.211460318 | 0.302126171 |  |  |  |  |  | 0 |
| 747.50281 | 0.08382317 | 0.326503115 | 0.199084378 | 0.30251208 |  |  |  |  |  | 0 |
| 342.29269 | 0.007001517 | 0.491951594 | 0.332808422 | 0.302766258 | C24H40O | [M-2H]- | 344.307915 | 342.2933622 | -1.96 | ['3beta-Cyclopentyl-5alpha-androstan-17beta-ol'] |
| 404.26595 | 0.116718132 | 0.298607347 | 0.185146331 | 0.30331513 |  |  |  |  |  | 0 |
| 386.25532 | 0.105091148 | 0.317317296 | 0.190393201 | 0.304900788 |  |  |  |  |  | 0 |
| 405.26934 | 0.112175319 | 0.303331971 | 0.187669928 | 0.305210561 |  |  |  |  |  | 0 |
| 449.2512 | 0.048279776 | 0.247415853 | 0.182563255 | 0.308199017 |  |  |  |  |  | 0 |
| 388.27101 | 0.036778574 | 0.325987744 | 0.192983024 | 0.308227434 |  |  |  |  |  | 0 |
| 731.50787 | 0.130750816 | 0.338412092 | 0.200210619 | 0.309417746 |  |  |  |  |  | 0 |
| 654.44203 | 0.025508357 | 0.199970008 | 0.075090903 | 0.309523089 |  |  |  |  |  | 0 |
| 603.12106 | $9.80 \mathrm{E}-08$ | 0.507680158 | 0.30841514 | 0.312038815 |  |  |  |  |  | 0 |
| 623.16875 | 3.49E-07 | 0.293221457 | 0.329812654 | 0.312293143 |  |  |  |  |  | 0 |
| 387.25875 | 0.103639754 | 0.302180658 | 0.196870744 | 0.31284468 |  |  |  |  |  | 0 |
| 483.13667 | 9.97E-07 | 0.379563259 | 0.336171663 | 0.31612154 |  |  |  |  |  | 0 |
| 192.01533 | $2.46 \mathrm{E}-05$ | 0.214341697 | 0.258681428 | 0.317899184 |  |  |  |  |  | 0 |
| 469.05194 | $9.61 \mathrm{E}-05$ | 0.580311814 | 0.358804957 | 0.319369477 |  |  |  |  |  | 0 |
| 659.1304 | $7.26 \mathrm{E}-08$ | 0.356374739 | 0.337477476 | 0.321129093 |  |  |  |  |  | 0 |
| 517.17826 | $1.03 \mathrm{E}-07$ | 0.42809667 | 0.304474581 | 0.322034878 |  |  |  |  |  | 0 |
| 689.14354 | 0.025239854 | 0.219643787 | 0.311084776 | 0.322597432 | C38H28N4O7 | [M+K-2H]- | 652.195801 | 689.1444076 | -1.26 | ['Esmeraldin B'] |
| 563.14753 | 3.65E-07 | 0.403601693 | 0.298971329 | 0.323864797 |  |  |  |  |  | 0 |
| 722.17273 | $4.83 \mathrm{E}-07$ | 0.501359068 | 0.361543518 | 0.326105966 |  |  |  |  |  | 0 |
| 716.5163 | 0.159459598 | 0.397651455 | 0.229000523 | 0.328226736 |  |  |  |  |  | 0 |
| 389.27442 | 0.06238205 | 0.338290702 | 0.205438545 | 0.330686006 |  |  |  |  |  | 0 |


| 293.05153 | $1.25 \mathrm{E}-05$ | 0.619793384 | 0.348980131 | 0.33157477 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 379.08482 | $2.45 \mathrm{E}-05$ | 0.163180452 | 0.291867384 | 0.332335762 |  |  |  |  |  | 0 |
| 667.52784 | 0.112175319 | 0.36815064 | 0.224088088 | 0.333204461 |  |  |  |  |  | 0 |
| 383.08366 | 5.00E-08 | 0.594078982 | 0.351638091 | 0.334841921 | C17H18N2O6 | [M+(37C) $]$ - | 346.116488 | 383.0829396 | 1.88 | ['Miraxanthin-V', 'Nifedipine'] |
| 715.51285 | 0.178263027 | 0.399881615 | 0.231331861 | 0.336557482 |  |  |  |  |  | 0 |
| 342.30169 | 0.015975666 | 0.516010743 | 0.304611547 | 0.338875443 | C18H37NO | [M+Hac-H]- | 283.287514 | 342.3013676 | 0.94 | ['Octadecanamide'] |
| 344.2446 | 0.146595516 | 0.412110227 | 0.239068542 | 0.339603337 |  |  |  |  |  | 0 |
| 683.52303 | 0.126278304 | 0.451300246 | 0.248368854 | 0.339904583 |  |  |  |  |  | 0 |
| 721.16973 | 3.10E-07 | 0.518393891 | 0.374858917 | 0.341540582 |  |  |  |  |  | 0 |
| 681.17457 | $2.41 \mathrm{E}-09$ | 0.937027422 | 0.401791737 | 0.341719475 |  |  |  |  |  | 0 |
| 671.48654 | 0.159013601 | 0.318223823 | 0.190111233 | 0.341949698 |  |  |  |  |  | 0 |
| 340.28284 | 0.00476544 | 0.556416444 | 0.375218904 | 0.343968587 |  |  |  |  |  | 0 |
| 645.13925 | $1.37 \mathrm{E}-07$ | 0.76416854 | 0.347532268 | 0.345802448 |  |  |  |  |  | 0 |
| 265.05659 | 3.64E-09 | 0.557516972 | 0.429634993 | 0.346323288 | C7H1007 | [M+Hac-H]- | 206.042655 | 265.0565086 | 0.31 | ['(2S,3R)-3-Hydroxybutane-1,2,3tricarboxylate', '(R)-2-Hydroxybutane-1,2,4tricarboxylate', '2-Methylcitrate', 'Homoisocitrate'] |
| 265.05659 | 3.64E-09 | 0.557516972 | 0.429634993 | 0.346323288 | C9H16N4S2 | [M+Na-2H]- | 244.08164 | 265.0563086 | 1.06 | ['Metiamide'] |
| 568.49558 | 0.061897456 | 0.471451261 | 0.22065709 | 0.347678881 | C32H63NO3 | [M+Hac-H]- | 509.480794 | 568.4946476 | 1.64 | ['N-(Tetradecanoyl)-sphing-4-enine'] |
| 410.23195 | 0.066961158 | 0.394160977 | 0.232638437 | 0.349995551 | C22H36O7 | [M-2H]- | 412.246105 | 410.2315522 | 0.97 | ['Grayanotoxin I'] |
| 266.05993 | 8.51E-09 | 0.580757008 | 0.440647657 | 0.352105468 |  |  |  |  |  | 0 |
| 649.25463 | 0.002187499 | 1.427248885 | 0.437110412 | 0.352820204 | C35H41NO11 | [M-2H]- | 651.267964 | 649.2534112 | 1.88 | ['Rifamycin Z'] |
| 386.10271 | $4.43 \mathrm{E}-07$ | 0.520292693 | 0.321133082 | 0.354590369 |  |  |  |  |  | 0 |
| 221.06716 | $1.32 \mathrm{E}-08$ | 0.441418363 | 0.346229548 | 0.358032977 |  |  |  |  |  | 0 |
| 354.26539 | 0.013313179 | 0.431104104 | 0.25557659 | 0.359387426 |  |  |  |  |  | 0 |
| 685.53844 | 0.101197722 | 0.30036119 | 0.252785041 | 0.360617255 |  |  |  |  |  | 0 |
| 383.12004 | $1.05 \mathrm{E}-09$ | 0.58858046 | 0.393053547 | 0.362659434 | C12H20O10 | [M+Hac-H]- | 324.10565 | 383.1195036 | 1.4 | ["Bis-D-fructose 2',1:2,1-dianhydride", "DFructofuranose 1,2':2,3-dianhydride"] |
| 383.12004 | $1.05 \mathrm{E}-09$ | 0.58858046 | 0.393053547 | 0.362659434 | C14H24O12 | [M-H]- | 384.12678 | 383.1195036 | 1.4 | ['Acetyl-maltose'] |
| 385.09929 | 6.15E-07 | 0.51311762 | 0.32559028 | 0.362905929 |  |  |  |  |  | 0 |
| 608.17741 | 9.92E-08 | 0.486489934 | 0.393941772 | 0.364094192 |  |  |  |  |  | 0 |
| 563.25412 | 0.020065624 | 0.760877883 | 0.413590484 | 0.368060518 | C24H42O7P2 | [M+Hac-H]- | 504.240581 | 563.2544346 | -0.56 | ['SR 12813'] |
| 607.17368 | $7.78 \mathrm{E}-08$ | 0.493540507 | 0.405805855 | 0.373710772 |  |  |  |  |  | 0 |
| 411.0787 | $9.93 \mathrm{E}-05$ | 0.695377644 | 0.421788433 | 0.373955206 | C12H16O12 | [M+Hac-H]- | 352.06418 | 411.0780336 | 1.62 | ['4-(4-Deoxy-alpha-D-gluc-4-enuronosyl)-Dgalacturonate', '4-(4-Deoxy-beta-D-gluc-4-enuronosyl)-D-galacturonate'] |
| 340.28119 | 0.115034403 | 0.633208601 | 0.410290731 | 0.377517034 |  |  |  |  |  | 0 |
| 147.06642 | 3.18E-06 | 0.387771775 | 0.248598999 | 0.37914359 | C4H8O2 | [M+Hac-H]- | 88.05243 | 147.0662836 | 0.93 | ['(R)-Acetoin', '1,4-Dioxane', '2Methylpropanoate', 'Acetoin', 'Butanoic |


|  |  |  |  |  |  |  |  |  |  | acid', 'Ethyl acetate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 147.06642 | 3.18E-06 | 0.387771775 | 0.248598999 | 0.37914359 | C6H12O4 | [M-H]- | 148.07356 | 147.0662836 | 0.93 | ['(R)-2,3-Dihydroxy-3-methylpentanoate', <br> '(R)-Mevalonate', '(R)-Pantoate', '(S)- <br> Mevalonate', '2,3-Dihydroxy-3- <br> methylpentanoate', '3,6-Dideoxy-Lgalactose', 'Abequose'] |
| 397.06315 | 0.000337831 | 0.829262075 | 0.475667461 | 0.38758541 |  |  |  |  |  | 0 |
| 337.07798 | $8.42 \mathrm{E}-08$ | 0.470652896 | 0.370366553 | 0.390855437 | C12H18O11 | [M-H]- | 338.084915 | 337.0776386 | 1.01 | ['L-Ascorbic acid-2-glucoside'] |
| 337.07798 | $8.42 \mathrm{E}-08$ | 0.470652896 | 0.370366553 | 0.390855437 | C16H16N2O4 | [M+(37Cl)]- | 300.111008 | 337.0774596 | 1.54 | ['5-Nitro-2-(3-phenylpropylamino)benzoic acid', 'Desmedipham', 'Phenmedipham'] |
| 593.19438 | $4.21 \mathrm{E}-08$ | 0.628146927 | 0.415285455 | 0.392866043 | C34H36O7 | [M+K-2H]- | 556.246105 | 593.1947116 | -0.56 | ['Ingenol 3,20-dibenzoate'] |
| 216.12462 | 0.02226835 | 0.353058249 | 0.230686487 | 0.405223416 |  |  |  |  |  | 0 |
| 323.13513 | $2.16 \mathrm{E}-08$ | 0.519594712 | 0.444451543 | 0.405911814 |  |  |  |  |  | 0 |
| 453.05712 | 0.000435195 | 1.029632534 | 0.480135913 | 0.411180946 |  |  |  |  |  | 0 |
| 543.40238 | 0.020936268 | 0.526139937 | 0.106312266 | 0.41304159 |  |  |  |  |  | 0 |
| 369.10439 | $4.21 \mathrm{E}-08$ | 0.640834543 | 0.449129609 | 0.416953223 |  |  |  |  |  | 0 |
| 473.1155 | $1.38 \mathrm{E}-07$ | 0.793922124 | 0.452791755 | 0.423861783 |  |  |  |  |  | 0 |
| 263.04096 | $2.63 \mathrm{E}-09$ | 0.266692662 | 0.355207168 | 0.424084341 | C6H15N2O6P | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 242.066776 | 263.0414446 | -1.84 | ['2-Deoxystreptamine 4-phosphate', '2Deoxystreptamine phospahte', '5-Phosphonooxy-L-lysine', 'Phosphoallohydroxy-L-lysine'] |
| 263.04096 | $2.63 \mathrm{E}-09$ | 0.266692662 | 0.355207168 | 0.424084341 | C7H8O7 | [M+Hac-H]- | 204.027005 | 263.0408586 | 0.39 | ['Oxaloglutarate'] |
| 793.12193 | $6.34 \mathrm{E}-05$ | 1.66740441 | 0.587537817 | 0.427418685 |  |  |  |  |  | 0 |
| 223.04816 | 3.29E-06 | 0.606056869 | 0.459515598 | 0.434321557 |  |  |  |  |  | 0 |
| 217.00355 | 0.000126961 | 0.414715257 | 0.44307043 | 0.435164111 |  |  |  |  |  | 0 |
| 225.09813 | $4.66 \mathrm{E}-05$ | 0.520444187 | 0.388220931 | 0.442264656 | C12H16N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 188.131348 | 225.0977996 | 1.47 | ['N,N-Dimethyltryptamine'] |
| 225.09813 | $4.66 \mathrm{E}-05$ | 0.520444187 | 0.388220931 | 0.442264656 | C9H18NO4 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 204.123584 | 225.0982526 | -0.54 | ['O-Acetylcarnitine'] |
| 765.23213 | $2.21 \mathrm{E}-09$ | 0.464917189 | 0.47998572 | 0.445445771 |  |  |  |  |  | 0 |
| 753.1961 | $5.64 \mathrm{E}-08$ | 0.466908781 | 0.426603582 | 0.44661319 |  |  |  |  |  | 0 |
| 223.08285 | 0.001384765 | 0.782212868 | 0.605114934 | 0.450900564 |  |  |  |  |  | 0 |
| 474.11889 | 0.000149693 | 0.855332926 | 0.489559446 | 0.456153986 | C16H21N3O8S | [M+Hac-H]- | 415.104939 | 474.1187926 | 0.21 | ['Cephalosporin C'] |
| 441.08941 | 7.91E-07 | 1.226247482 | 0.594275294 | 0.457915809 |  |  |  |  |  | 0 |
| 763.21671 | $1.21 \mathrm{E}-09$ | 0.332391416 | 0.442908973 | 0.460309032 |  |  |  |  |  | 0 |
| 756.21428 | $1.52 \mathrm{E}-05$ | 0.706603512 | 0.551205648 | 0.463408923 |  |  |  |  |  | 0 |
| 467.08156 | $1.40 \mathrm{E}-05$ | 0.727219545 | 0.473758643 | 0.465517689 | C16H22N4O8S | [M+(37Cl)]- | 430.115838 | 467.0822896 | -1.56 | ['N-EthyImaleimide-S-glutathione'] |
| 261.06167 | $1.27 \mathrm{E}-07$ | 0.769003848 | 0.542475576 | 0.466668699 |  |  |  |  |  | 0 |
| 629.12214 | $4.06 \mathrm{E}-06$ | 0.454560254 | 0.458737119 | 0.469792378 |  |  |  |  |  | 0 |
| 295.10382 | 7.46E-07 | 0.719090615 | 0.548842867 | 0.49512866 | C15H2ON2S | [M+Cl]- | 260.13472 | 295.1041216 | -1.02 | ['Methaphenilene'] |


| 621.18947 | 3.92E-07 | 0.67799198 | 0.473821327 | 0.505752417 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 443.10506 | $4.44 \mathrm{E}-09$ | 0.627801169 | 0.495788942 | 0.508422104 |  |  |  |  |  | 0 |
| 635.1689 | 7.16E-06 | 0.687636708 | 0.489560184 | 0.508663593 |  |  |  |  |  | 0 |
| 517.10557 | $3.76 \mathrm{E}-06$ | 0.762354481 | 0.472362585 | 0.514123447 | C30H24O6 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 480.15729 | 517.1058966 | -0.63 | ['Blestriarene $\mathrm{B}^{\prime}$ ] |
| 279.07226 | $1.81 \mathrm{E}-06$ | 1.070940162 | 0.645805286 | 0.518782685 | C8H12O7 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 220.058305 | 279.0721586 | 0.36 | ['(R)-(Homo)2-citrate', '1-Hydroxypentane-1,2,5-tricarboxylate'] |
| 659.1082 | 0.000566944 | 0.43285009 | 0.343851858 | 0.525494407 |  |  |  |  |  | 0 |
| 323.0623 | $1.72 \mathrm{E}-06$ | 1.389760074 | 0.557169017 | 0.52609037 |  |  |  |  |  | 0 |
| 797.18706 | $2.87 \mathrm{E}-06$ | 1.260493091 | 0.515535254 | 0.534503616 |  |  |  |  |  | 0 |
| 441.1257 | $9.48 \mathrm{E}-07$ | 0.85763559 | 0.570577307 | 0.535202872 |  |  |  |  |  | 0 |
| 203.05547 | 6.25E-05 | 1.037688591 | 0.655793226 | 0.545170182 |  |  |  |  |  | 0 |
| 421.03959 | 3.57E-06 | 1.50033872 | 0.615525964 | 0.545874144 |  |  |  |  |  | 0 |
| 681.21013 | $2.38 \mathrm{E}-07$ | 0.544704155 | 0.514389729 | 0.546986756 |  |  |  |  |  | 0 |
| 737.23702 | $5.19 \mathrm{E}-08$ | 0.420286896 | 0.510873371 | 0.553775483 |  |  |  |  |  | 0 |
| 437.03459 | $4.34 \mathrm{E}-06$ | 1.376614817 | 0.6686147 | 0.558929367 |  |  |  |  |  | 0 |
| 553.10326 | $9.96 \mathrm{E}-05$ | 1.50458501 | 0.548720854 | 0.561607836 |  |  |  |  |  | 0 |
| 448.10329 | $2.35 \mathrm{E}-06$ | 0.99900636 | 0.533069535 | 0.561740986 |  |  |  |  |  | 0 |
| 324.06569 | $1.81 \mathrm{E}-06$ | 1.448767016 | 0.600324694 | 0.563978536 |  |  |  |  |  | 0 |
| 677.14332 | 3.87E-05 | 0.502981331 | 0.493098646 | 0.569545294 |  |  |  |  |  | 0 |
| 415.1055 | $2.49 \mathrm{E}-05$ | 0.635153336 | 0.569207387 | 0.575888796 |  |  |  |  |  | 0 |
| 620.17735 | 0.001463923 | 0.740493144 | 0.596512377 | 0.576870458 |  |  |  |  |  | 0 |
| 725.20068 | 0.007483464 | 0.699870205 | 0.583861396 | 0.590973868 |  |  |  |  |  | 0 |
| 263.07562 | 0.046270723 | 0.965795821 | 0.521306924 | 0.591749063 |  |  |  |  |  | 0 |
| 409.13581 | 0.015605426 | 0.9036873 | 0.758824874 | 0.59452714 | C21H28N2S2 | [M+(37Cl)]- | 372.169392 | 409.1358436 | -0.08 | ['Buthiobate'] |
| 205.07115 | 0.00332185 | 1.332650983 | 0.728602215 | 0.604974925 |  |  |  |  |  | 0 |
| 196.00922 | 0.049745326 | 1.927514658 | 0.579182402 | 0.60748005 |  |  |  |  |  | 0 |
| 609.15306 | 0.000315518 | 0.97204515 | 0.606533789 | 0.608330383 |  |  |  |  |  | 0 |
| 755.21144 | $1.56 \mathrm{E}-05$ | 0.80105343 | 0.667269441 | 0.62262907 |  |  |  |  |  | 0 |
| 739.1809 | 0.023755073 | 0.850774363 | 0.811741574 | 0.629284667 |  |  |  |  |  | 0 |
| 239.11372 | 0.11284827 | 0.719741008 | 0.532207854 | 0.642332251 |  |  |  |  |  | 0 |
| 381.06805 | $1.42 \mathrm{E}-05$ | 1.570344041 | 0.799964689 | 0.647729096 |  |  |  |  |  | 0 |
| 595.17367 | $3.76 \mathrm{E}-06$ | 0.509146699 | 0.56211032 | 0.651192117 |  |  |  |  |  | 0 |
| 667.19495 | $1.81 \mathrm{E}-05$ | 0.885712895 | 0.621937823 | 0.655030135 |  |  |  |  |  | 0 |
| 761.18387 | 0.00011843 | 5.80095297 | 0.606159658 | 0.657457903 |  |  |  |  |  | 0 |
| 603.15248 | 0.000148349 | 0.323164509 | 0.543898948 | 0.659606284 |  |  |  |  |  | 0 |
| 335.09874 | 0.000176028 | 1.097953882 | 0.648527404 | 0.662115714 | C11H16O8 | [M+Hac-H]- | 276.08452 | 335.0983736 | 1.09 | ['Ranunculin'] |
| 335.09874 | 0.000176028 | 1.097953882 | 0.648527404 | 0.662115714 | C18H22N2S | [M+K-2H]- | 298.15037 | 335.0989766 | -0.71 | ['Trimeprazine'] |
| 224.04976 | 0.000337259 | 0.883859809 | 0.697390506 | 0.663684741 |  |  |  |  |  | 0 |


| 797.25898 | 0.001648919 | 0.685629917 | 0.682672522 | 0.666900272 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 669.12974 | 5.88E-06 | 0.390577547 | 0.590177469 | 0.668828247 |  |  |  |  |  | 0 |
| 447.09987 | $1.47 \mathrm{E}-06$ | 1.100633256 | 0.670682033 | 0.671610217 |  |  |  |  |  | 0 |
| 769.22739 | 0.000106507 | 0.590067729 | 0.651161197 | 0.673800216 |  |  |  |  |  | 0 |
| 604.18204 | $9.44 \mathrm{E}-08$ | 0.672404622 | 0.606174615 | 0.675522725 |  |  |  |  |  | 0 |
| 723.18532 | 0.000585119 | 0.990117272 | 0.753125954 | 0.679897896 |  |  |  |  |  | 0 |
| 521.31701 | 0.192271166 | 1.009149881 | 0.382618619 | 0.684283547 |  |  |  |  |  | 0 |
| 611.14565 | 0.122568765 | 0.795893006 | 0.671857234 | 0.68593956 | C20H32N6O12S2 | [M-H]- | 612.151968 | 611.1446916 | 1.57 | ['Glutathione disulfide'] |
| 603.17858 | $1.75 \mathrm{E}-08$ | 0.657431803 | 0.610382644 | 0.686044341 |  |  |  |  |  | 0 |
| 619.17385 | $2.53 \mathrm{E}-05$ | 0.85685503 | 0.676149635 | 0.686080998 |  |  |  |  |  | 0 |
| 415.10254 | 0.173785416 | 0.777313423 | 0.679511682 | 0.686634774 | C17H22N4O6 | [M+K-2H]- | 378.153936 | 415.1025426 | -0.01 | ['Reduced riboflavin'] |
| 353.07296 | 0.033162358 | 0.676096578 | 0.699801106 | 0.695529805 |  |  |  |  |  | 0 |
| 645.11689 | 0.008234153 | 0.689438468 | 0.759088206 | 0.697369728 |  |  |  |  |  | 0 |
| 307.06744 | $5.89 \mathrm{E}-07$ | 1.58527508 | 0.773777462 | 0.697748695 | C12H16NO7 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 286.092679 | 307.0673476 | 0.3 | ['N-Glucosylnicotinate'] |
| 307.06744 | $5.89 \mathrm{E}-07$ | 1.58527508 | 0.773777462 | 0.697748695 | C15H14N2O3 | [M+(37Cl)]- | 270.100443 | 307.0668946 | 1.78 | ['Dihydroxycarbazepine'] |
| 337.11439 | 0.028774901 | 1.005798509 | 0.768387984 | 0.698141256 | C11H18O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 278.10017 | 337.1140236 | 1.09 | ['Tuliposide A'] |
| 337.11439 | 0.028774901 | 1.005798509 | 0.768387984 | 0.698141256 | C17H20N2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 300.147393 | 337.1138446 | 1.62 | ['Bifenazate'] |
| 416.11324 | 3.24E-06 | 1.46429757 | 0.759630265 | 0.701545715 | C14H19N3O6S | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 357.099459 | 416.1133126 | -0.17 | ['Deacetoxycephalosporin C'] |
| 536.30854 | 0.298093796 | 0.98981838 | 0.396715089 | 0.704042002 |  |  |  |  |  | 0 |
| 681.20122 | 0.003315422 | 0.702817821 | 0.663703938 | 0.705762961 |  |  |  |  |  | 0 |
| 520.31359 | 0.258643198 | 1.060350567 | 0.450105061 | 0.706070624 |  |  |  |  |  | 0 |
| 534.29288 | 0.474744409 | 0.888687248 | 0.432272308 | 0.712280451 |  |  |  |  |  | 0 |
| 535.16781 | 0.002194322 | 0.723735552 | 0.612763222 | 0.713035546 |  |  |  |  |  | 0 |
| 417.11386 | $1.42 \mathrm{E}-05$ | 1.506073583 | 0.714965698 | 0.717698206 | C19H26O6S | [M+Cl]- | 382.145012 | 417.1144136 | -1.33 | ['2-Methoxyestradiol-17beta 3-sulfate'] |
| 561.13179 | 0.000959703 | 0.922915123 | 0.702725666 | 0.725480664 |  |  |  |  |  | 0 |
| 532.12448 | $2.08 \mathrm{E}-07$ | 1.364713024 | 0.785317718 | 0.725578981 |  |  |  |  |  | 0 |
| 749.16578 | 0.173785416 | 0.656669837 | 0.705102644 | 0.725646739 |  |  |  |  |  | 0 |
| 264.08068 | 0.001418363 | 0.76685749 | 0.732362523 | 0.726738148 |  |  |  |  |  | 0 |
| 263.0773 | 0.000978128 | 0.771773208 | 0.733257005 | 0.730188366 | C14H14N2O | [M+(37Cl)]- | 226.110613 | 263.0770646 | 0.89 | ['Metyrapone'] |
| 263.0773 | 0.000978128 | 0.771773208 | 0.733257005 | 0.730188366 | C8H12O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 204.06339 | 263.0772436 | 0.21 | ['3-o-Ethyl-L-ascorbic acid'] |
| 495.0678 | 0.004998252 | 0.922213548 | 0.712826928 | 0.734862387 |  |  |  |  |  | 0 |
| 587.12624 | $1.17 \mathrm{E}-05$ | 1.317081623 | 0.949364012 | 0.737292796 |  |  |  |  |  | 0 |
| 587.12465 | $1.17 \mathrm{E}-05$ | 1.317081632 | 0.949364127 | 0.737292842 |  |  |  |  |  | 0 |
| 785.2231 | $1.56 \mathrm{E}-06$ | 0.447973789 | 0.620333219 | 0.739474227 |  |  |  |  |  | 0 |
| 415.10971 | 4.35E-06 | 1.521172422 | 0.809203 | 0.749376432 |  |  |  |  |  | 0 |
| 194.031 | 0.366499125 | 0.636623235 | 0.682991132 | 0.7500321 |  |  |  |  |  | 0 |
| 735.14935 | $7.61 \mathrm{E}-05$ | 0.41750339 | 0.66874656 | 0.758049626 |  |  |  |  |  | 0 |
| 353.10934 | 0.001188747 | 1.176728823 | 0.896009101 | 0.761228663 | C11H18O9 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 294.095085 | 353.1089386 | 1.14 | ['Tuliposide B'] |


| 353.10934 | 0.001188747 | 1.176728823 | 0.896009101 | 0.761228663 | C2OH21NO3S | [M-2H]- | 355.124216 | 353.1096632 | -0.92 | ['AL-321'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 798.22553 | 0.031506332 | 0.649375224 | 0.790201899 | 0.768871518 |  |  |  |  |  | 0 |
| 501.14681 | $8.04 \mathrm{E}-05$ | 1.076107957 | 0.76465077 | 0.769569281 | C18H30016 | [M-H]- | 502.15339 | 501.1461136 | 1.39 | ['alpha-L-Rhamnopyranosyl-(1->2)-beta-D-galactopyranosyl-(1->2)-beta-Dglucuronopyranoside'] |
| 738.16786 | 0.001283603 | 1.214348369 | 0.913766399 | 0.771229415 |  |  |  |  |  | 0 |
| 339.09363 | 0.000373126 | 0.689254358 | 0.682814926 | 0.774338572 | C12H20O11 | [M-H]- | 340.100565 | 339.0932886 | 1.01 | ['3-Ketolactose', '3-Ketosucrose', 'Cellobiono-1,5-lactone'] |
| 516.12961 | 0.000121116 | 1.076511667 | 0.817920602 | 0.778044156 |  |  |  |  |  | 0 |
| 270.04677 | 0.510547104 | 1.295584696 | 1.047352758 | 0.780537073 |  |  |  |  |  | 0 |
| 601.12698 | 0.01665336 | 0.841783843 | 0.768556529 | 0.783020621 |  |  |  |  |  | 0 |
| 263.07453 | 0.018379776 | 0.869237119 | 0.494653499 | 0.78349077 |  |  |  |  |  | 0 |
| 555.12126 | 0.042973061 | 0.673048529 | 0.835566976 | 0.784780468 |  |  |  |  |  | 0 |
| 223.04614 | 0.00761868 | 1.061816871 | 0.844927251 | 0.789898118 | C11H10N2O | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 186.079313 | 223.0457646 | 1.68 | ['Credazine', 'Deoxyvasicinone'] |
| 223.04614 | 0.00761868 | 1.061816871 | 0.844927251 | 0.789898118 | C5H8O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 164.03209 | 223.0459436 | 0.88 | ['2-Dehydro-D-xylonate'] |
| 650.18786 | 7.92E-06 | 1.530812868 | 1.018816498 | 0.797115026 |  |  |  |  |  | 0 |
| 400.11839 | $7.35 \mathrm{E}-05$ | 1.257188965 | 0.849379633 | 0.7979741 |  |  |  |  |  | 0 |
| 615.14245 | 0.010558303 | 0.5976506 | 0.742246757 | 0.802056706 | C34H28O9 | [M+Cl]- | 580.173335 | 615.1427366 | -0.47 | ['Mulberrofuran $\mathrm{C}^{\prime}$ '] |
| 737.16479 | 0.001226956 | 1.197409544 | 0.89312352 | 0.802353488 |  |  |  |  |  | 0 |
| 511.06321 | 0.005638508 | 0.672795123 | 0.619668217 | 0.814860008 |  |  |  |  |  | 0 |
| 399.11491 | $3.87 \mathrm{E}-05$ | 1.295489367 | 0.875274963 | 0.823759844 | C12H20O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 340.100565 | 399.1144186 | 1.23 | ['3-Ketolactose', '3-Ketosucrose', 'Cellobiono-1,5-lactone'] |
| 399.11491 | 3.87E-05 | 1.295489367 | 0.875274963 | 0.823759844 | C19H26N2O3S | [M+K-2H]- | 362.166415 | 399.1150216 | -0.28 | ['DU 122290'] |
| 552.30357 | 0.410104488 | 1.005134787 | 0.44864007 | 0.830553577 | C24H39N5O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 493.290035 | 552.3038886 | -0.58 | ['Syringolin A'] |
| 392.24887 | 0.119537469 | 0.337696952 | 0.335318905 | 0.837867788 |  |  |  |  |  | 0 |
| 791.17651 | 0.058195886 | 1.035675441 | 0.810797837 | 0.858985585 |  |  |  |  |  | 0 |
| 266.05186 | 0.552331795 | 1.468229663 | 1.085869627 | 0.866086659 |  |  |  |  |  | 0 |
| 579.17902 | 0.00203753 | 0.749662709 | 0.759786205 | 0.872615098 |  |  |  |  |  | 0 |
| 395.06023 | 0.015669573 | 0.624034348 | 0.714021724 | 0.878351889 | C14H19N2O7P | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 358.092991 | 395.0594426 | 1.99 | ['N1-(5-Phospho-alpha-D-ribosyl)-5,6dimethylbenzimidazole'] |
| 446.31308 | 0.586533994 | 0.740834163 | 0.446103317 | 0.884270278 |  |  |  |  |  | 0 |
| 648.20852 | 0.133460873 | 0.745617619 | 0.96693484 | 0.912243504 |  |  |  |  |  | 0 |
| 323.05951 | 0.000776221 | 2.05833405 | 0.778885123 | 0.913872573 | C12H16NO7 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 286.092679 | 323.0591306 | 1.17 | ['N-Glucosylnicotinate'] |
| 323.05951 | 0.000776221 | 2.05833405 | 0.778885123 | 0.913872573 | C13H12O4S | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 264.045632 | 323.0594856 | 0.08 | ['(1R,2R)-3-[(1,2-Dihydro-2-hydroxy-1-naphthalenyl)thio]-2-oxopropanoic acid'] |
| 649.18434 | 7.89E-07 | 1.395167772 | 0.975879432 | 0.923512938 |  |  |  |  |  | 0 |
| 297.08309 | 0.063869114 | 0.647522665 | 0.801672782 | 0.927433066 | C14H16N2O3 | [M+(37Cl)]- | 260.116093 | 297.0825446 | 1.84 | ['Maculosin'] |
| 297.08309 | 0.063869114 | 0.647522665 | 0.801672782 | 0.927433066 | C15H2ON2S | [M+K-2H]- | 260.13472 | 297.0833266 | -0.8 | ['Methaphenilene'] |
| 297.08309 | 0.063869114 | 0.647522665 | 0.801672782 | 0.927433066 | C8H14O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 238.06887 | 297.0827236 | 1.23 | ['3-Deoxy-D-manno-octulosonate'] |


| 439.074 | 3.21E-07 | 1.814348935 | 0.991542963 | 0.936693669 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 673.14802 | 0.000493546 | 0.538566307 | 0.828961924 | 0.948778612 |  |  |  |  |  | 0 |
| 740.21959 | $5.42 \mathrm{E}-09$ | 1.70001067 | 1.128641628 | 0.949919581 |  |  |  |  |  | 0 |
| 739.21642 | $1.10 \mathrm{E}-08$ | 1.663162005 | 1.125989247 | 0.967573025 |  |  |  |  |  | 0 |
| 171.03023 | 0.125458505 | 0.913765106 | 1.04588089 | 0.967965841 | C5H4O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 112.016045 | 171.0298986 | 1.94 | ['2-Furoate'] |
| 171.03023 | 0.125458505 | 0.913765106 | 1.04588089 | 0.967965841 | C7H8O5 | [M-H]- | 172.037175 | 171.0298986 | 1.94 | ['2-Hydroxy-5-methyl-cis,cis-muconate', '2-Hydroxyhepta-2,4-dienedioate', '2-Oxo-5-methyl-cis-muconate', '2-Oxohept-3enedioate', '3-Dehydroshikimate', '5Dehydroshikimate', '5- <br> Methylmaleylacetate', 'cis,cis-2,4-Dihydroxy-5-methyl-6-oxo-2,4-hexadienoate'] |
| 797.22293 | 0.911166782 | 1.025392454 | 1.05201031 | 0.983366587 |  |  |  |  |  | 0 |
| 593.15816 | $1.61 \mathrm{E}-05$ | 1.470505754 | 1.006516628 | 0.98838187 |  |  |  |  |  | 0 |
| 210.02597 | 0.953174626 | 0.932018475 | 1.111555566 | 0.99297092 |  |  |  |  |  | 0 |
| 254.05189 | 0.100068085 | 2.190182284 | 1.28681286 | 0.997011189 |  |  |  |  |  | 0 |
| 657.1297 | 0.032076287 | 1.218997155 | 0.898026356 | 0.9996675 |  |  |  |  |  | 0 |
| 343.26057 | 0.859868907 | 1.228799917 | 0.934579331 | 1.002416067 |  |  |  |  |  | 0 |
| 589.12728 | $4.74 \mathrm{E}-10$ | 2.13036596 | 1.108469937 | 1.002433735 |  |  |  |  |  | 0 |
| 343.11365 | 0.919124952 | 0.940848474 | 0.983482444 | 1.003037953 | C17H22O5 | [M+(37Cl)]- | 306.146725 | 343.1131766 | 1.38 | ['Arnicolide A', 'Confertiflorin', 'Eupaformonin', 'Gaillardin', 'Isotenulin', 'Ligulatin B', 'Lipiferolide', 'Matricin', 'Ovatifolin', 'Pyrethrosin', 'Tenulin', 'Viscidulin B', 'Xanthinin', 'Xanthumin'] |
| 735.22122 | 0.000233714 | 0.873232319 | 1.102811441 | 1.030580114 |  |  |  |  |  | 0 |
| 341.10906 | 0.746287328 | 1.079290714 | 1.029591337 | 1.037100222 | C10H18O9 | [M+Hac-H]- | 282.095085 | 341.1089386 | 0.36 | ['Xylobiose'] |
| 341.10906 | 0.746287328 | 1.079290714 | 1.029591337 | 1.037100222 | C12H22O11 | [M-H]- | 342.116215 | 341.1089386 | 0.36 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-Dglucose', 'Cellobiose', 'D-Fructosyl-Dfructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', 'Gentiobiose', 'Inulobiose', <br> 'Isomaltose', 'Lactose', 'Lactulose', <br> 'Laminaribiose', 'Levanbiose', 'Maltose', <br> 'Mannobiose', 'Melibiose', 'Nigerose', <br> 'Palatinose', 'Sucrose', 'alpha,alpha- <br> Trehalose', 'alpha-Cellobiose', 'alpha-D- <br> Aldosyl beta-D-fructoside', 'alpha-D- <br> Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |


| 342.11264 | 0.778762398 | 1.076680357 | 1.036802489 | 1.041225971 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 791.24751 | 0.027723709 | 0.752559405 | 1.100706092 | 1.047192099 |  |  |  |  |  | 0 |
| 391.24553 | 0.133394441 | 0.348440606 | 0.225847465 | 1.051067099 |  |  |  |  |  | 0 |
| 594.16172 | $5.73 \mathrm{E}-06$ | 1.549653263 | 1.039818321 | 1.056045489 |  |  |  |  |  | 0 |
| 547.15248 | 0.099451368 | 0.938816606 | 0.966977974 | 1.070307171 |  |  |  |  |  | 0 |
| 459.11523 | 9.29E-06 | 0.719190034 | 0.954481093 | 1.077279656 | C23H24N2O4S | [M+Cl]- | 424.14568 | 459.1150816 | 0.32 | ['Eprosartan'] |
| 472.10334 | 0.675156518 | 1.073832871 | 1.082554719 | 1.081027113 |  |  |  |  |  | 0 |
| 733.16951 | 0.00138755 | 0.761442 | 1.041301726 | 1.085687875 |  |  |  |  |  | 0 |
| 751.18034 | 0.373256035 | 1.126545059 | 1.051012233 | 1.090092552 |  |  |  |  |  | 0 |
| 647.20483 | 0.017136506 | 0.774473458 | 1.049448566 | 1.09518145 |  |  |  |  |  | 0 |
| 379.05597 | 0.454594974 | 1.034859375 | 0.921585912 | 1.104858229 |  |  |  |  |  | 0 |
| 325.11147 | 0.351328837 | 1.463622615 | 1.393348125 | 1.112518638 |  |  |  |  |  | 0 |
| 240.03621 | 0.384159557 | 1.683118603 | 1.325376632 | 1.118962372 |  |  |  |  |  | 0 |
| 341.10419 | 0.877249114 | 0.868876765 | 1.115434007 | 1.13324231 | C11H20N4O6 | [M+(37Cl)]- | 304.138286 | 341.1047376 | -1.61 | ['Nopaline'] |
| 575.18428 | 6.67E-06 | 1.399240828 | 1.231556207 | 1.154952989 |  |  |  |  |  | 0 |
| 569.08575 | 0.027215533 | 1.62955519 | 1.087042613 | 1.165263975 |  |  |  |  |  | 0 |
| 585.1088 | $1.25 \mathrm{E}-06$ | 0.673609485 | 0.916524951 | 1.174179921 |  |  |  |  |  | 0 |
| 266.09632 | 0.002522976 | 0.81709569 | 1.111277421 | 1.183842394 |  |  |  |  |  | 0 |
| 265.09297 | 0.00178862 | 0.870955187 | 1.090116333 | 1.186481978 |  |  |  |  |  | 0 |
| 661.14781 | 0.001036454 | 0.886358364 | 1.010067757 | 1.19078204 |  |  |  |  |  | 0 |
| 443.14135 | $2.58 \mathrm{E}-05$ | 1.44975123 | 1.250524596 | 1.208725752 | C14H24O12 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 384.12678 | 443.1406336 | 1.62 | ['Acetyl-maltose'] |
| 443.14135 | $2.58 \mathrm{E}-05$ | 1.44975123 | 1.250524596 | 1.208725752 | C20H28N2O5S | [M+Cl]- | 408.171895 | 443.1412966 | 0.12 | ['Tamsulosin'] |
| 578.16707 | $2.56 \mathrm{E}-08$ | 1.95360876 | 1.294105186 | 1.210737111 | C23H34N3O10P | [M+Cl]- | 543.198185 | 578.1675866 | -0.89 | ['Phosphoramidon'] |
| 351.09367 | 0.000278057 | 1.706430659 | 1.339758346 | 1.218910871 |  |  |  |  |  | 0 |
| 413.13055 | 0.000122308 | 1.612868884 | 1.424759567 | 1.226149128 |  |  |  |  |  | 0 |
| 437.07952 | 0.045932976 | 1.174866234 | 1.123262758 | 1.236836684 |  |  |  |  |  | 0 |
| 545.17322 | 0.000417514 | 1.260379478 | 1.276442226 | 1.236877854 |  |  |  |  |  | 0 |
| 531.12113 | $1.34 \mathrm{E}-10$ | 2.4822952 | 1.387706536 | 1.269609079 |  |  |  |  |  | 0 |
| 723.22149 | $1.58 \mathrm{E}-07$ | 1.778411749 | 1.436651019 | 1.272880447 |  |  |  |  |  | 0 |
| 651.20007 | 3.64E-07 | 1.731467193 | 1.364605937 | 1.290951429 |  |  |  |  |  | 0 |
| 743.15 | 0.057682136 | 1.567118972 | 1.455660058 | 1.308078684 |  |  |  |  |  | 0 |
| 427.10995 | $5.74 \mathrm{E}-05$ | 1.726635114 | 1.35560492 | 1.309623645 |  |  |  |  |  | 0 |
| 325.11435 | 0.017136506 | 1.382520197 | 1.312772488 | 1.32299601 | C12H22O10 | [M-H]- | 326.1213 | 325.1140236 | 1 | ['2-O-alpha-L-Rhamnopyranosyl-Dglucopyranose', 'Robinobiose', 'Rutinose'] |
| 325.11435 | 0.017136506 | 1.382520197 | 1.312772488 | 1.32299601 | C16H2ON2O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 288.147393 | 325.1138446 | 1.55 | ['Methyl 2-(4-isopropyl-4-methyl-5-oxo-2-imidazolin-2-yl)-p-toluate', 'Methyl 6-(4-isopropyl-4-methyl-5-oxo-2-imidazolin-2-yl)-m-toluate'] |
| 479.1022 | 0.552473972 | 1.342508104 | 1.165269974 | 1.32687929 | C21H27N3O4S3 | [M-2H]- | 481.116373 | 479.1018202 | 0.79 | ['Yersiniabactin'] |


| 577.16337 | $1.07 \mathrm{E}-09$ | 2.185459224 | 1.44759664 | 1.327018447 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 749.23724 | 0.000566944 | 1.090265788 | 1.371592036 | 1.327712374 |  |  |  |  |  | 0 |
| 591.17887 | $1.00 \mathrm{E}-05$ | 1.495515193 | 1.378824324 | 1.341123565 |  |  |  |  |  | 0 |
| 367.08876 | $2.58 \mathrm{E}-05$ | 1.470951889 | 1.467728053 | 1.347338415 | C18H22N2O2S | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 330.1402 | 367.0888066 | -0.13 | ['Pyributicarb'] |
| 413.16699 | 0.157948056 | 1.020665607 | 1.2217286 | 1.351044384 | C20H28N2O5 | [M+(37Cl)]- | 376.199823 | 413.1662746 | 1.73 | ['Enalapril', 'Remifentanil'] |
| 729.21117 | $3.31 \mathrm{E}-09$ | 0.440619582 | 1.115682129 | 1.368964619 |  |  |  |  |  | 0 |
| 763.18065 | 0.000404022 | 1.1065813 | 1.35657262 | 1.381554439 |  |  |  |  |  | 0 |
| 498.11896 | $4.93 \mathrm{E}-05$ | 1.3677969 | 1.356928384 | 1.390292052 |  |  |  |  |  | 0 |
| 727.19535 | $1.00 \mathrm{E}-08$ | 0.49466379 | 1.056670674 | 1.390407943 | C40H36O11 | [M+Cl]- | 692.225765 | 727.1951666 | 0.25 | ['Kuwanone G'] |
| 719.22612 | 0.00135294 | 1.012156169 | 1.42514717 | 1.398736328 |  |  |  |  |  | 0 |
| 340.06064 | $5.58 \mathrm{E}-07$ | 2.597695685 | 1.501329299 | 1.408152781 |  |  |  |  |  | 0 |
| 556.1611 | $5.67 \mathrm{E}-09$ | 0.594517688 | 1.219793772 | 1.41705226 |  |  |  |  |  | 0 |
| 562.17165 | $7.31 \mathrm{E}-05$ | 1.76854529 | 1.554011068 | 1.420756575 |  |  |  |  |  | 0 |
| 326.11776 | 0.002444671 | 1.473961698 | 1.430913858 | 1.426326981 |  |  |  |  |  | 0 |
| 753.1603 | 0.003038714 | 1.621694851 | 1.654156527 | 1.427740239 |  |  |  |  |  | 0 |
| 339.05719 | $6.42 \mathrm{E}-07$ | 2.647515795 | 1.543462462 | 1.441392964 | C14H17N2O4PS | [M-H]- | 340.064668 | 339.0573916 | -0.59 | ['Pyridafenthion'] |
| 339.13003 | 8.87E-06 | 2.399543535 | 1.652842563 | 1.456481976 | C13H24O10 | [M-H]- | 340.13695 | 339.1296736 | 1.05 | ['Methyl-2-alpha-L-fucopyranosyl-beta-Dgalactoside'] |
| 592.18228 | 2.86E-06 | 1.646207354 | 1.497139216 | 1.45735462 |  |  |  |  |  | 0 |
| 745.16996 | $2.21 \mathrm{E}-05$ | 1.671139033 | 1.479648775 | 1.458716778 | C40H36O12 | [M+K-2H]- | 708.22068 | 745.1692866 | 0.9 | ['Sanggenon C', 'Sanggenon D'] |
| 692.16232 | 0.000432863 | 0.96868551 | 1.293997589 | 1.461079615 |  |  |  |  |  | 0 |
| 522.14029 | 0.082796135 | 1.282255768 | 1.404445281 | 1.464846228 |  |  |  |  |  | 0 |
| 416.14972 | 0.010975374 | 0.984858925 | 1.371299846 | 1.466142011 |  |  |  |  |  | 0 |
| 327.1089 | 0.022373545 | 1.142356399 | 1.340390355 | 1.467594706 | C15H20O8 | [M-H]- | 328.11582 | 327.1085436 | 1.09 | ['Anisatin', 'Paeonoside'] |
| 471.09993 | 0.001704966 | 1.339620185 | 1.358290999 | 1.483595418 |  |  |  |  |  | 0 |
| 569.0983 | 0.024112234 | 1.446486839 | 1.160147862 | 1.497422295 |  |  |  |  |  | 0 |
| 453.16212 | 0.003331989 | 1.154941462 | 1.544039641 | 1.523560684 |  |  |  |  |  | 0 |
| 652.20336 | 0.000180909 | 1.932361742 | 1.591398334 | 1.543007525 |  |  |  |  |  | 0 |
| 415.14624 | 0.000107725 | 1.013775528 | 1.463242332 | 1.56821849 |  |  |  |  |  | 0 |
| 797.329 | 0.23731873 | 1.735278591 | 1.914264303 | 1.569051918 |  |  |  |  |  | 0 |
| 738.20374 | $1.12 \mathrm{E}-05$ | 1.145796361 | 1.420546748 | 1.573020213 |  |  |  |  |  | 0 |
| 487.16755 | $4.14 \mathrm{E}-08$ | 2.39723522 | 1.779336289 | 1.615983931 |  |  |  |  |  | 0 |
| 473.15173 | $2.43 \mathrm{E}-08$ | 1.906222289 | 1.604532608 | 1.623113851 | C16H22N4O9 | [M+Hac-H]- | 414.138681 | 473.1525346 | -1.7 | ['Clavamycin $\mathrm{A}^{\prime}$ ] |
| 474.15522 | $6.10 \mathrm{E}-08$ | 1.917931414 | 1.621992638 | 1.634029321 |  |  |  |  |  | 0 |
| 721.20561 | $3.02 \mathrm{E}-07$ | 1.150321382 | 1.561423304 | 1.635708865 |  |  |  |  |  | 0 |
| 676.16725 | 3.63E-07 | 1.104281235 | 1.599061085 | 1.654278032 |  |  |  |  |  | 0 |
| 515.12622 | 7.21E-08 | 2.256321696 | 1.779001586 | 1.658552192 | C30H24O6 | [M+Cl]- | 480.15729 | 515.1266916 | -0.92 | ['Blestriarene B'] |
| 793.17225 | 0.003415801 | 1.586051873 | 1.368559758 | 1.659065855 |  |  |  |  |  | 0 |


| 691.15887 | 3.23E-06 | 1.111946803 | 1.492771012 | 1.661452154 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 799.20248 | 0.000214066 | 1.285821209 | 1.511843188 | 1.664838117 |  |  |  |  |  | 0 |
| 443.12029 | $2.28 \mathrm{E}-05$ | 1.201363151 | 1.421584387 | 1.676282129 |  |  |  |  |  | 0 |
| 617.19455 | 0.001657699 | 1.454433128 | 1.672971091 | 1.676585868 |  |  |  |  |  | 0 |
| 191.05518 | 0.441774569 | 1.216983651 | 0.501812835 | 1.684064394 | C6H10N4O2 | [M+Na-2H]- | 170.080376 | 191.0550446 | 0.71 | ['N-Isopropylammelide'] |
| 558.14013 | 0.004100429 | 1.43950663 | 1.600823654 | 1.692602873 |  |  |  |  |  | 0 |
| 675.16376 | 6.20E-08 | 1.151204669 | 1.643969294 | 1.694839654 |  |  |  |  |  | 0 |
| 599.14731 | $1.28 \mathrm{E}-06$ | 1.22780813 | 1.55583541 | 1.696676339 |  |  |  |  |  | 0 |
| 497.11547 | $1.91 \mathrm{E}-06$ | 1.693579873 | 1.677395286 | 1.702387394 | C22H24N2O9 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 460.148183 | 497.1146346 | 1.68 | ['Oxytetracycline'] |
| 553.09072 | $2.34 \mathrm{E}-07$ | 2.093897434 | 1.632471585 | 1.708030135 |  |  |  |  |  | 0 |
| 467.10524 | 4.92E-06 | 1.291361513 | 1.625852577 | 1.708189399 |  |  |  |  |  | 0 |
| 569.06841 | 0.000561386 | 3.674563864 | 1.867743595 | 1.709129937 |  |  |  |  |  | 0 |
| 722.20899 | $1.81 \mathrm{E}-07$ | 1.184054929 | 1.643549458 | 1.724539715 |  |  |  |  |  | 0 |
| 685.18435 | $1.95 \mathrm{E}-08$ | 1.003370536 | 1.540442833 | 1.737141115 |  |  |  |  |  | 0 |
| 547.11616 | $7.44 \mathrm{E}-11$ | 4.052528009 | 2.120818313 | 1.74575381 |  |  |  |  |  | 0 |
| 677.21569 | 9.38E-07 | 1.893343721 | 1.845321218 | 1.749142299 |  |  |  |  |  | 0 |
| 571.14347 | $3.83 \mathrm{E}-07$ | 0.568681018 | 1.43938307 | 1.78118855 | C17H35N4O13P | [M+K-2H]- | 534.193829 | 571.1424356 | 1.81 | ["5'-Phosphoribostamycin"] |
| 483.09993 | $8.32 \mathrm{E}-05$ | 1.565930442 | 1.621499609 | 1.782336229 |  |  |  |  |  | 0 |
| 397.13553 | $1.19 \mathrm{E}-07$ | 1.947547695 | 1.895291108 | 1.783600722 | C19H26N2O3S | [M+Cl]- | 362.166415 | 397.1358166 | -0.72 | ['DU 122290'] |
| 737.20067 | $1.52 \mathrm{E}-05$ | 1.330521894 | 1.671595929 | 1.800839499 |  |  |  |  |  | 0 |
| 678.21901 | 0.011608632 | 1.808478173 | 1.723195471 | 1.814991524 |  |  |  |  |  | 0 |
| 590.16658 | $2.73 \mathrm{E}-07$ | 2.064263078 | 1.844895156 | 1.842119631 | C26H37NO8S2 | [M+Cl]- | 555.196063 | 590.1654646 | 1.89 | ['Tiapamil'] |
| 689.17952 | $3.00 \mathrm{E}-07$ | 1.326684896 | 1.822824157 | 1.865609667 |  |  |  |  |  | 0 |
| 481.12065 | $1.82 \mathrm{E}-07$ | 1.938000271 | 1.942806952 | 1.866265626 | C22H24N2O8 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 444.153268 | 481.1197196 | 1.93 | ['Doxycycline', 'Tetracycline'] |
| 481.12065 | $1.82 \mathrm{E}-07$ | 1.938000271 | 1.942806952 | 1.866265626 | C31H24O3 | [M+K-2H]- | 444.172545 | 481.1211516 | -1.04 | ['Difenacoum'] |
| 249.09813 | 0.000213196 | 1.054540496 | 1.297373993 | 1.882258278 | C14H16N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 212.131348 | 249.0977996 | 1.33 | ['3,3-Dimethylbenzidine'] |
| 249.09813 | 0.000213196 | 1.054540496 | 1.297373993 | 1.882258278 | C8H14O5 | [M+Hac-H]- | 190.084125 | 249.0979786 | 0.61 | $['(R)-3-((R)-3-$ Hydroxybutanoyloxy)butanoate'] |
| 657.18961 | $4.26 \mathrm{E}-07$ | 1.440205612 | 1.921320786 | 1.911565465 |  |  |  |  |  | 0 |
| 499.13113 | $4.69 \mathrm{E}-06$ | 2.055712533 | 2.027425321 | 1.914222024 |  |  |  |  |  | 0 |
| 471.17249 | 3.19E-07 | 2.678904741 | 2.134031934 | 1.930069234 |  |  |  |  |  | 0 |
| 657.15279 | $1.16 \mathrm{E}-05$ | 0.987101537 | 1.705125569 | 1.94834811 |  |  |  |  |  | 0 |
| 633.15337 | 8.17E-07 | 0.925384434 | 1.650071598 | 1.948568748 |  |  |  |  |  | 0 |
| 485.15195 | 7.61E-10 | 1.603369316 | 1.907840331 | 1.949273132 |  |  |  |  |  | 0 |
| 455.06862 | $1.35 \mathrm{E}-10$ | 5.102849973 | 2.385196443 | 1.951052285 |  |  |  |  |  | 0 |
| 573.13188 | 4.71E-09 | 2.971710326 | 2.120191373 | 1.955393378 |  |  |  |  |  | 0 |
| 766.19904 | 8.45E-06 | 1.019826193 | 1.576595597 | 1.964293643 |  |  |  |  |  | 0 |


| 561.16808 | $2.89 \mathrm{E}-08$ | 2.571945638 | 2.196411911 | 1.98843462 | C18H30016 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 502.15339 | 561.1672436 | 1.49 | ['alpha-L-Rhamnopyranosyl-(1->2)-beta-D-galactopyranosyl-(1->2)-beta-Dglucuronopyranoside'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 471.12649 | 0.002122029 | 1.595998681 | 1.951038742 | 1.993593865 | C22H26O10 | [M+Na-2H]- | 450.1526 | 471.1272686 | -1.65 | ['Auriculoside'] |
| 795.24244 | 8.96E-09 | 1.839605213 | 1.904284845 | 2.007468249 |  |  |  |  |  | 0 |
| 576.15138 | $2.35 \mathrm{E}-06$ | 2.069670473 | 1.837611792 | 2.076123726 |  |  |  |  |  | 0 |
| 509.11598 | 3.70E-05 | 1.767738939 | 2.205251288 | 2.085232023 |  |  |  |  |  | 0 |
| 723.29187 | 0.019321607 | 4.762474697 | 2.590454708 | 2.087084261 |  |  |  |  |  | 0 |
| 622.15676 | 0.000576191 | 2.543152387 | 2.467181306 | 2.091936677 |  |  |  |  |  | 0 |
| 765.19588 | $2.94 \mathrm{E}-07$ | 1.089579886 | 1.747611498 | 2.098614299 |  |  |  |  |  | 0 |
| 747.22168 | $2.47 \mathrm{E}-08$ | 1.418387701 | 2.079814691 | 2.102885447 |  |  |  |  |  | 0 |
| 646.15677 | $5.14 \mathrm{E}-09$ | 1.195098356 | 1.869143777 | 2.141054771 |  |  |  |  |  | 0 |
| 753.23241 | $3.46 \mathrm{E}-09$ | 1.745783539 | 2.017352342 | 2.163992339 |  |  |  |  |  | 0 |
| 679.15947 | $4.41 \mathrm{E}-07$ | 2.291658105 | 2.071577909 | 2.176689427 |  |  |  |  |  | 0 |
| 521.1369 | 7.04E-07 | 1.6105832 | 1.878827097 | 2.178095638 | C25H28N2O7S | [M+Na-2H]- | 500.161725 | 521.1363936 | 0.97 | ['(S)-N-[3-(3,4-Methylenedioxyphenyl)-2-(acetylthio)methyl-1-oxoprolyl]-(S)-alanine benzyl ester'] |
| 796.24576 | 0.000587972 | 1.906271875 | 1.860585211 | 2.195978889 |  |  |  |  |  | 0 |
| 317.06485 | 0.151031231 | 2.066861541 | 2.409266644 | 2.198738946 | C10H18O9 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 282.095085 | 317.0644866 | 1.15 | ['Xylobiose'] |
| 317.06485 | 0.151031231 | 2.066861541 | 2.409266644 | 2.198738946 | C14H16O7 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 296.089605 | 317.0642736 | 1.82 | ['de-Hypoxanthine futalosine'] |
| 317.06485 | 0.151031231 | 2.066861541 | 2.409266644 | 2.198738946 | C8H19O3PS2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 258.051327 | 317.0651806 | -1.04 | ['Demeton-O', 'Demeton-S'] |
| 735.18471 | $3.58 \mathrm{E}-08$ | 1.61483484 | 1.900372878 | 2.204261187 |  |  |  |  |  | 0 |
| 737.30741 | 0.067320882 | 4.251774625 | 2.748931828 | 2.208736539 |  |  |  |  |  | 0 |
| 473.1407 | 0.003451522 | 0.849237672 | 1.782863812 | 2.20977229 |  |  |  |  |  | 0 |
| 789.19594 | $1.91 \mathrm{E}-07$ | 1.377441852 | 2.031487811 | 2.21417963 |  |  |  |  |  | 0 |
| 396.12351 | $2.26 \mathrm{E}-06$ | 3.040133982 | 1.964941629 | 2.219815292 |  |  |  |  |  | 0 |
| 399.15136 | 0.000249961 | 1.266750564 | 1.937066895 | 2.223604444 | C13H24O10 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 340.13695 | 399.1508036 | 1.39 | ['Methyl-2-alpha-L-fucopyranosyl-beta-Dgalactoside'] |
| 559.11603 | $2.75 \mathrm{E}-05$ | 1.882420629 | 2.085285745 | 2.233710531 |  |  |  |  |  | 0 |
| 451.11033 | 0.000243002 | 1.374515411 | 2.2107158 | 2.250427411 | C22H26N2O4S | [M+K-2H]- | 414.16133 | 451.1099366 | 0.87 | ['Diltiazem'] |
| 379.12512 | $7.59 \mathrm{E}-06$ | 2.21408217 | 2.133370911 | 2.254289256 | C19H22N2O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 342.157958 | 379.1244096 | 1.87 | ['Phenisopham'] |
| 499.09477 | 0.000344769 | 1.977897593 | 2.010523258 | 2.262174553 |  |  |  |  |  | 0 |
| 645.15287 | 3.78E-09 | 1.35444009 | 2.002630766 | 2.266078084 |  |  |  |  |  | 0 |
| 607.28034 | 0.059543669 | 3.887132 | 2.608288023 | 2.285746954 | C35H39N5O5 | [M-2H]- | 609.29512 | 607.2805672 | -0.37 | ['Ergocristine'] |
| 607.28034 | 0.059543669 | 3.887132 | 2.608288023 | 2.285746954 | C37H4ON2O6 | [M-H]- | 608.288638 | 607.2813616 | -1.68 | ['Berbamine', 'Gyrocarpine', 'Oxyacanthine', 'Pycnamine', 'Thalmine'] |
| 513.11034 | $2.16 \mathrm{E}-09$ | 2.39469831 | 2.24790236 | 2.319696374 |  |  |  |  |  | 0 |
| 593.26473 | 0.076528266 | 3.774954646 | 2.711357178 | 2.336083775 | C36H38N2O6 | [M-H]- | 594.272988 | 593.2657116 | -1.65 | $\begin{gathered} \hline \text { ['(+)-Atherospermoline', '(+)-Bebeerine', } \\ \text { 'Aromoline', 'Daphnandrine', } \\ \text { 'Isochondrodendrine', 'Obamegine'] } \\ \hline \end{gathered}$ |


| 575.1479 | $2.07 \mathrm{E}-06$ | 2.309996761 | 2.067221105 | 2.338468411 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 754.23508 | 0.000283793 | 1.86910588 | 2.304446828 | 2.34289936 |  |  |  |  |  | 0 |
| 469.15691 | $4.75 \mathrm{E}-10$ | 1.969059025 | 2.398758857 | 2.352981405 |  |  |  |  |  | 0 |
| 505.1347 | $5.60 \mathrm{E}-07$ | 1.884235347 | 2.228746618 | 2.357714557 | C22H22O10 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 446.1213 | 505.1351536 | -0.9 | ['Biochanin A-beta-D-glucoside', 'Dihydrogranaticin', 'Glycitin', 'Obtusifolin 2glucoside', 'Physcion 8-glucoside', 'Trifolirhizin'] |
| 473.14595 | 0.000209106 | 2.182828932 | 2.245634248 | 2.412200176 | C22H22O8 | [M+Hac-H]- | 414.13147 | 473.1453236 | 1.32 | ['Picropodophyllin', 'Podophyllotoxin'] |
| 441.16206 | 0.000323735 | 2.256964611 | 2.321373203 | 2.420042198 |  |  |  |  |  | 0 |
| 379.06528 | $3.29 \mathrm{E}-06$ | 2.241822327 | 2.10823736 | 2.431835114 | C12H22O11 | [M+K-2H]- | 342.116215 | 379.0648216 | 1.21 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-Dglucose', 'Cellobiose', 'D-Fructosyl-Dfructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', 'Gentiobiose', 'Inulobiose', <br> 'Isomaltose', 'Lactose', 'Lactulose', 'Laminaribiose', 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alphaTrehalose', 'alpha-Cellobiose', 'alpha-DAldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |
| 425.13075 | 4.87E-05 | 2.33522705 | 2.457835254 | 2.43314838 |  |  |  |  |  | 0 |
| 660.17223 | $8.05 \mathrm{E}-09$ | 1.519331442 | 2.588973733 | 2.474051479 |  |  |  |  |  | 0 |
| 395.12008 | $4.30 \mathrm{E}-10$ | 3.486362231 | 2.612580737 | 2.474393882 | C19H24N2O3S | [M+Cl]- | 360.150765 | 395.1201666 | -0.22 | ['LY395153'] |
| 545.13686 | $6.01 \mathrm{E}-10$ | 1.710424751 | 2.061547017 | 2.488508621 |  |  |  |  |  | 0 |
| 736.18838 | $1.33 \mathrm{E}-07$ | 1.823511365 | 2.072580622 | 2.492186492 |  |  |  |  |  | 0 |
| 514.11395 | $3.70 \mathrm{E}-10$ | 2.632729928 | 2.461372407 | 2.541736813 | C24H23N5O6 | [M+K-2H]- | 477.164835 | 514.1134416 | 0.99 | ['CB3717'] |
| 467.14145 | $1.90 \mathrm{E}-05$ | 3.233017686 | 3.048211671 | 2.568600283 |  |  |  |  |  | 0 |
| 412.1184 | $1.27 \mathrm{E}-09$ | 3.840928808 | 2.677916986 | 2.592684057 |  |  |  |  |  | 0 |
| 793.22687 | $3.37 \mathrm{E}-10$ | 1.837388508 | 2.295659194 | 2.595720801 |  |  |  |  |  | 0 |
| 687.20027 | $1.72 \mathrm{E}-09$ | 1.139776852 | 2.133661546 | 2.59575888 | C32H38N6O7S | [M+K-2H]- | 650.252271 | 687.2008776 | -0.88 | ['BQ 518'] |
| 715.19533 | $6.27 \mathrm{E}-11$ | 1.141351715 | 2.322803275 | 2.618126919 |  |  |  |  |  | 0 |
| 795.20651 | $3.64 \mathrm{E}-09$ | 2.233212048 | 2.622929941 | 2.656763345 |  |  |  |  |  | 0 |
| 621.29624 | 0.036778574 | 4.489798235 | 2.807770628 | 2.664134247 | C38H42N2O6 | [M-H]- | 622.304288 | 621.2970116 | -1.24 | ['(+)-O-Methylthalicberine', '(+)- <br> Tetrandrine', 'Cycleanine', 'Isotetrandrine', 'Obaberine', 'Rodiasine'] |
| 546.14022 | $1.43 \mathrm{E}-06$ | 1.874507461 | 2.116413172 | 2.685226728 |  |  |  |  |  | 0 |
| 409.09949 | $5.11 \mathrm{E}-07$ | 3.014227093 | 2.883945702 | 2.704049498 | C12H25N2O10P | [M+Na-2H]- | 388.124686 | 409.0993546 | 0.33 | ['Fructoselysine 6-phosphate'] |


| 747.18525 | 5.18E-09 | 1.465588842 | 2.554647319 | 2.706021125 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 489.13645 | $1.38 \mathrm{E}-10$ | 3.025073191 | 2.539911647 | 2.716173298 |  |  |  |  |  | 0 |
| 395.08378 | $4.96 \mathrm{E}-08$ | 1.529613064 | 2.375563256 | 2.71691426 |  |  |  |  |  | 0 |
| 341.14564 | 0.00042444 | 1.052538558 | 1.681749258 | 2.738476232 |  |  |  |  |  | 0 |
| 631.17405 | $4.79 \mathrm{E}-10$ | 2.356962113 | 2.610526813 | 2.743448503 |  |  |  |  |  | 0 |
| 724.22447 | $2.59 \mathrm{E}-12$ | 3.816078314 | 3.144001756 | 2.750426433 |  |  |  |  |  | 0 |
| 560.15603 | $9.44 \mathrm{E}-09$ | 2.30112041 | 2.637427444 | 2.80452897 |  |  |  |  |  | 0 |
| 621.15322 | $1.75 \mathrm{E}-08$ | 3.445730059 | 2.903824426 | 2.812619648 |  |  |  |  |  | 0 |
| 589.19943 | $1.10 \mathrm{E}-06$ | 1.713064548 | 2.584505227 | 2.855861904 |  |  |  |  |  | 0 |
| 506.14526 | 2.85E-10 | 2.484954515 | 2.642729932 | 2.866426713 |  |  |  |  |  | 0 |
| 673.18434 | $4.81 \mathrm{E}-10$ | 2.123418644 | 2.883815157 | 2.867385118 |  |  |  |  |  | 0 |
| 411.11493 | $1.40 \mathrm{E}-09$ | 4.196233778 | 2.947600094 | 2.867751546 |  |  |  |  |  | 0 |
| 532.16084 | $1.70 \mathrm{E}-10$ | 2.277724669 | 2.685335674 | 2.885867923 |  |  |  |  |  | 0 |
| 685.12461 | 5.67E-10 | 1.649456344 | 2.468970733 | 2.89264675 |  |  |  |  |  | 0 |
| 761.16625 | $2.68 \mathrm{E}-05$ | 2.584288052 | 2.76280654 | 2.901296589 |  |  |  |  |  | 0 |
| 505.14172 | $2.20 \mathrm{E}-10$ | 2.586391237 | 2.684769417 | 2.908809903 |  |  |  |  |  | 0 |
| 661.22013 | $4.35 \mathrm{E}-07$ | 2.605328534 | 2.989101633 | 2.922644584 |  |  |  |  |  | 0 |
| 529.17826 | 8.05E-09 | 2.790159579 | 2.976503144 | 2.94327752 |  |  |  |  |  | 0 |
| 796.20988 | 6.99E-09 | 2.320716048 | 2.855457929 | 2.954039555 |  |  |  |  |  | 0 |
| 491.15054 | 6.27E-07 | 3.816693141 | 2.711050452 | 3.004661885 |  |  |  |  |  | 0 |
| 659.16872 | $5.05 \mathrm{E}-11$ | 1.981138182 | 3.047365676 | 3.028470572 |  |  |  |  |  | 0 |
| 457.15692 | 3.68E-10 | 3.044761742 | 2.968636405 | 3.037934982 |  |  |  |  |  | 0 |
| 615.17892 | $9.13 \mathrm{E}-11$ | 2.506580327 | 3.019867201 | 3.08388978 |  |  |  |  |  | 0 |
| 531.1573 | $1.47 \mathrm{E}-10$ | 2.446465597 | 2.868559787 | 3.089806587 |  |  |  |  |  | 0 |
| 557.17319 | $2.19 \mathrm{E}-10$ | 2.857270634 | 3.101207764 | 3.101747719 |  |  |  |  |  | 0 |
| 717.17436 | $1.06 \mathrm{E}-08$ | 2.029264627 | 2.923261058 | 3.109210633 |  |  |  |  |  | 0 |
| 718.17774 | 0.000831748 | 2.333676794 | 3.204320325 | 3.120844195 |  |  |  |  |  | 0 |
| 743.16668 | $3.29 \mathrm{E}-05$ | 2.394438559 | 2.556545418 | 3.169699849 |  |  |  |  |  | 0 |
| 671.16884 | $1.70 \mathrm{E}-10$ | 1.685062573 | 2.915321133 | 3.182141128 |  |  |  |  |  | 0 |
| 511.07158 | 8.38E-10 | 2.770077293 | 2.787761769 | 3.196311449 |  |  |  |  |  | 0 |
| 775.18054 | 7.44E-11 | 1.25121271 | 2.740632765 | 3.259161324 |  |  |  |  |  | 0 |
| 196.04562 | 0.09578271 | 6.089523306 | 3.674827543 | 3.333372299 |  |  |  |  |  | 0 |
| 489.13978 | $1.45 \mathrm{E}-10$ | 4.325777326 | 3.245706741 | 3.34824049 | C20H22N8O5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 454.171317 | 489.1407186 | -1.92 | ['Methotrexate'] |
| 489.13978 | $1.45 \mathrm{E}-10$ | 4.325777326 | 3.245706741 | 3.34824049 | C22H22O9 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 430.126385 | 489.1402386 | -0.94 | ['Ononin'] |
| 768.21456 | 5.57E-09 | 2.668568924 | 3.526609864 | 3.350596702 |  |  |  |  |  | 0 |
| 643.17396 | 3.43E-11 | 2.137005211 | 3.547285347 | 3.370998864 |  |  |  |  |  | 0 |
| 689.21601 | 3.47E-07 | 1.868677943 | 2.739696299 | 3.379817282 | C38H38O10 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 654.2465 | 689.2159016 | 0.16 | ['Mezerein'] |
| 377.10957 | $2.06 \mathrm{E}-06$ | 3.418397389 | 3.180538363 | 3.449513949 | C20H24N2OS | [M+K-2H]- | 340.160935 | 377.1095416 | 0.08 | ['Lucanthone', 'Propiomazine'] |


| 544.1611 | 1.39E-09 | 2.5844992 | 3.481247384 | 3.467860049 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 759.16176 | 8.26E-12 | 1.062664438 | 2.681836965 | 3.471567284 |  |  |  |  |  | 0 |
| 439.11006 | $5.65 \mathrm{E}-12$ | 3.445480744 | 3.556550101 | 3.509476586 |  |  |  |  |  | 0 |
| 719.18951 | $1.15 \mathrm{E}-10$ | 1.671626023 | 2.764331886 | 3.512068512 |  |  |  |  |  | 0 |
| 748.18865 | 4.27E-06 | 2.382512062 | 4.387903773 | 3.573934551 |  |  |  |  |  | 0 |
| 767.21162 | $2.37 \mathrm{E}-10$ | 2.891256848 | 3.717247581 | 3.604218828 |  |  |  |  |  | 0 |
| 355.05219 | $9.62 \mathrm{E}-09$ | 4.551262282 | 3.735514557 | 3.648020463 |  |  |  |  |  | 0 |
| 383.10862 | 0.000159989 | 4.079802118 | 3.358448301 | 3.67710006 | C19H22O6 | [M+(37Cl)]- | 346.14164 | 383.1080916 | 1.38 | ['Alectrol', 'Antheridic acid', 'Cynaropicrin', 'Gibberellin A29-catabolite', 'Gibberellin A3', 'Gibberellin A34-catabolite', 'Gibberellin A6', 'Hallactone A', 'Molephantin', 'Ponalactone A', 'Saupirin', 'Strigol'] |
| 600.17963 | 0.000118916 | 3.298177268 | 3.459973412 | 3.678559556 |  |  |  |  |  | 0 |
| 489.14658 | $2.21 \mathrm{E}-10$ | 4.663141357 | 3.591516716 | 3.681681787 |  |  |  |  |  | 0 |
| 487.12436 | $6.44 \mathrm{E}-12$ | 2.910435489 | 3.318125875 | 3.688932956 | C22H2009 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 428.110735 | 487.1245886 | -0.47 | ['epsilon-Rhodomycinone'] |
| 487.12436 | $6.44 \mathrm{E}-12$ | 2.910435489 | 3.318125875 | 3.688932956 | C24H24O11 | [M-H]- | 488.131865 | 487.1245886 | -0.47 | ['Phrymarolin I'] |
| 517.15717 | $1.81 \mathrm{E}-07$ | 1.811057232 | 2.768895692 | 3.694152036 |  |  |  |  |  | 0 |
| 365.1095 | $3.70 \mathrm{E}-10$ | 3.80775443 | 4.053270793 | 3.698354567 | C18H22N2O2S | [M+Cl]- | 330.1402 | 365.1096016 | -0.28 | ['Pyributicarb'] |
| 365.1095 | $3.70 \mathrm{E}-10$ | 3.80775443 | 4.053270793 | 3.698354567 | C19H24N2OS | [M+K-2H]- | 328.160935 | 365.1095416 | -0.11 | ['Methotrimeprazine'] |
| 687.16372 | $1.55 \mathrm{E}-11$ | 1.916101313 | 3.39363725 | 3.710978783 |  |  |  |  |  | 0 |
| 691.1952 | $1.41 \mathrm{E}-10$ | 2.39679074 | 3.356274031 | 3.719176266 | C38H38O10 | [M+K-2H]- | 654.2465 | 691.1951066 | 0.14 | ['Mezerein'] |
| 490.15017 | $2.19 \mathrm{E}-10$ | 4.767629511 | 3.686731148 | 3.755967363 |  |  |  |  |  | 0 |
| 751.21638 | $2.86 \mathrm{E}-11$ | 3.155217176 | 4.020853515 | 3.761711386 |  |  |  |  |  | 0 |
| 586.17195 | $1.38 \mathrm{E}-07$ | 2.592526018 | 3.625954552 | 3.794897181 |  |  |  |  |  | 0 |
| 789.23242 | $4.30 \mathrm{E}-12$ | 2.05914133 | 3.517800342 | 3.797732497 |  |  |  |  |  | 0 |
| 486.11898 | $1.57 \mathrm{E}-10$ | 4.828384836 | 4.030195739 | 3.801812804 |  |  |  |  |  | 0 |
| 663.16388 | 2.26E-12 | 4.066635353 | 3.804725005 | 3.820365181 |  |  |  |  |  | 0 |
| 585.13212 | $1.35 \mathrm{E}-10$ | 1.384118149 | 2.984409006 | 3.858598157 |  |  |  |  |  | 0 |
| 664.16731 | 5.16E-12 | 4.328232559 | 3.889301964 | 3.873349896 |  |  |  |  |  | 0 |
| 627.1407 | $2.81 \mathrm{E}-05$ | 5.14700632 | 3.911303231 | 3.881421601 |  |  |  |  |  | 0 |
| 676.20358 | 3.68E-10 | 2.063054146 | 3.618453109 | 3.940564043 |  |  |  |  |  | 0 |
| 453.09037 | $9.15 \mathrm{E}-07$ | 2.031509707 | 3.623289144 | 3.953558304 |  |  |  |  |  | 0 |
| 485.11547 | 7.91E-11 | 5.109475395 | 4.244580656 | 3.986840378 |  |  |  |  |  | 0 |
| 794.23062 | $1.12 \mathrm{E}-10$ | 2.560875205 | 3.503580197 | 4.006155964 |  |  |  |  |  | 0 |
| 693.21074 | $1.28 \mathrm{E}-09$ | 4.826607976 | 4.174919451 | 4.015324693 |  |  |  |  |  | 0 |
| 759.18494 | 8.17E-11 | 1.422439356 | 3.167745588 | 4.019921413 |  |  |  |  |  | 0 |
| 645.18924 | 6.80E-13 | 0.961201366 | 3.157588204 | 4.054415418 |  |  |  |  |  | 0 |
| 606.16153 | $2.66 \mathrm{E}-10$ | 5.150983331 | 4.004153973 | 4.074932673 |  |  |  |  |  | 0 |
| 393.10448 | $2.38 \mathrm{E}-09$ | 5.040189041 | 4.817078678 | 4.077418647 |  |  |  |  |  | 0 |


| 554.14536 | 5.93E-09 | 0.821209529 | 2.815783405 | 4.092856167 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 471.13603 | $2.24 \mathrm{E}-12$ | 1.937360094 | 3.607159607 | 4.113393453 |  |  |  |  |  | 0 |
| 537.32888 | 0.000493546 | 4.023136094 | 3.561425247 | 4.139472814 | C3OH48N2O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 500.361408 | 537.3278596 | 1.9 | ['Vicenistatin'] |
| 719.1539 | $2.57 \mathrm{E}-10$ | 1.713619251 | 3.68176235 | 4.142581965 |  |  |  |  |  | 0 |
| 472.1396 | $7.26 \mathrm{E}-13$ | 1.954909628 | 3.643482755 | 4.177190147 |  |  |  |  |  | 0 |
| 693.17449 | $3.69 \mathrm{E}-11$ | 2.807454258 | 3.479730089 | 4.258179627 |  |  |  |  |  | 0 |
| 488.13465 | $5.71 \mathrm{E}-11$ | 2.654114556 | 3.680824629 | 4.374446636 | C23H25N5O5 | [M+K-2H]- | 451.18557 | 488.1341766 | 0.97 | ['Doxazosin'] |
| 523.10792 | $6.94 \mathrm{E}-08$ | 3.810701924 | 4.11745457 | 4.375291952 |  |  |  |  |  | 0 |
| 688.16705 | 3.08E-09 | 1.762696976 | 3.901104196 | 4.401586667 |  |  |  |  |  | 0 |
| 694.17777 | 3.81E-08 | 2.685321918 | 3.643761875 | 4.408753083 | C34H33NO15 | [M-H]- | 695.185024 | 694.1777476 | 0.03 | ['Dexylosylbenanomicin A'] |
| 539.16278 | $1.69 \mathrm{E}-11$ | 2.391329383 | 4.040883111 | 4.477816913 | C26H34N2O4S2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 502.196002 | 539.1624536 | 0.61 | ['Myxothiazol ${ }^{\text {' }}$ ] |
| 487.13115 | $1.67 \mathrm{E}-11$ | 2.798772214 | 3.836834226 | 4.588931619 |  |  |  |  |  | 0 |
| 727.37731 | 0.00017282 | 2.328257951 | 4.114939944 | 4.596725382 |  |  |  |  |  | 0 |
| 675.20007 | $5.30 \mathrm{E}-11$ | 2.525854239 | 4.28543379 | 4.608239213 |  |  |  |  |  | 0 |
| 643.13574 | 8.02E-06 | 4.52183037 | 4.413814459 | 4.639277788 |  |  |  |  |  | 0 |
| 527.16265 | $7.91 \mathrm{E}-11$ | 3.334364273 | 4.757275201 | 4.660225056 |  |  |  |  |  | 0 |
| 720.19337 | $4.99 \mathrm{E}-10$ | 1.651115881 | 3.598275344 | 4.662642827 |  |  |  |  |  | 0 |
| 743.18958 | $1.94 \mathrm{E}-10$ | 1.330544192 | 3.602324443 | 4.663686752 | C40H36O12 | [M+Cl]- | 708.22068 | 743.1900816 | -0.67 | ['Sanggenon C', 'Sanggenon D'] |
| 529.14176 | 4.97E-13 | 2.453160123 | 3.698118959 | 4.665763578 |  |  |  |  |  | 0 |
| 717.21085 | $2.84 \mathrm{E}-10$ | 1.994811971 | 4.39998019 | 4.668347503 |  |  |  |  |  | 0 |
| 749.20086 | $1.67 \mathrm{E}-11$ | 2.44720464 | 4.266731952 | 4.674140066 |  |  |  |  |  | 0 |
| 397.10526 | $2.38 \mathrm{E}-07$ | 4.314901928 | 4.720191358 | 4.70415438 | C19H22O7 | [M+Cl]- | 362.136555 | 397.1059566 | -1.75 | ['Gibberellin A8-catabolite', 'Nagilactone C', 'Vernolide'] |
| 397.10526 | $2.38 \mathrm{E}-07$ | 4.314901928 | 4.720191358 | 4.70415438 | C20H24O6 | [M+K-2H]- | 360.15729 | 397.1058966 | -1.6 | ['Dibenzo-18-crown-6', 'Euponin', 'Fastigilin C', 'Lariciresinol', 'Molephantinin', 'Multigilin', 'Triptolide'] |
| 502.11387 | 3.81E-06 | 4.006320702 | 4.520358566 | 4.724317912 |  |  |  |  |  | 0 |
| 514.15036 | 6.67E-12 | 3.101219268 | 3.983628503 | 4.726185651 |  |  |  |  |  | 0 |
| 613.16325 | 2.57E-08 | 2.289900793 | 4.292989227 | 4.731788282 |  |  |  |  |  | 0 |
| 565.36025 | 0.000406045 | 4.480787461 | 4.22304107 | 4.741376546 |  |  |  |  |  | 0 |
| 453.12573 | $1.33 \mathrm{E}-10$ | 5.757709379 | 5.371211588 | 4.747587113 |  |  |  |  |  | 0 |
| 646.19321 | $4.29 \mathrm{E}-13$ | 1.037904235 | 3.65176193 | 4.768280488 |  |  |  |  |  | 0 |
| 399.10706 | $2.06 \mathrm{E}-06$ | 5.090507696 | 4.285070575 | 4.786059523 |  |  |  |  |  | 0 |
| 694.21402 | $2.74 \mathrm{E}-10$ | 5.85191063 | 5.095542751 | 4.829639723 |  |  |  |  |  | 0 |
| 779.2121 | $1.33 \mathrm{E}-05$ | 2.17461212 | 3.921464007 | 4.89337338 |  |  |  |  |  | 0 |
| 733.20578 | $8.69 \mathrm{E}-12$ | 2.270805552 | 4.553617086 | 4.909655716 |  |  |  |  |  | 0 |
| 513.14684 | $7.10 \mathrm{E}-14$ | 3.310480507 | 4.160019218 | 4.915287422 |  |  |  |  |  | 0 |
| 413.0869 | $1.48 \mathrm{E}-07$ | 3.265293234 | 4.706068149 | 4.944208024 | C17H20N4O6 | [M+K-2H]- | 376.138286 | 413.0868926 | 0.02 | ['Riboflavin'] |
| 589.16306 | $8.46 \mathrm{E}-11$ | 5.445025692 | 4.957310126 | 5.017031836 |  |  |  |  |  | 0 |


| 659.20512 | 6.18E-11 | 2.389409434 | 4.619794724 | 5.032548668 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 573.16785 | $1.16 \mathrm{E}-07$ | 4.339183629 | 4.782558888 | 5.057011303 |  |  |  |  |  | 0 |
| 557.13686 | $2.05 \mathrm{E}-11$ | 3.730786299 | 4.847002005 | 5.079249156 |  |  |  |  |  | 0 |
| 731.19006 | $9.35 \mathrm{E}-12$ | 2.251535881 | 4.66490343 | 5.136309148 |  |  |  |  |  | 0 |
| 605.15788 | $1.47 \mathrm{E}-10$ | 6.241849846 | 5.082148194 | 5.153301963 |  |  |  |  |  | 0 |
| 701.17948 | 3.72E-08 | 2.290197733 | 4.59315865 | 5.18581835 | C42H32O9 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 680.204635 | 701.1793036 | 0.25 | ['Canaliculatol', 'Copalliferol B'] |
| 571.11639 | 1.70E-10 | 4.731136154 | 5.293297377 | 5.255212703 | C21H26N8O6S2 | [M+Na-2H]- | 550.141676 | 571.1163446 | 0.08 | ['Cefclidin'] |
| 345.08312 | 3.12E-08 | 2.360171021 | 4.649231416 | 5.288149231 | C18H16N2O3 | [M+(37Cl)]- | 308.116093 | 345.0825446 | 1.67 | ['Citrus Red No.2'] |
| 759.2216 | 3.15E-11 | 1.798810421 | 4.559938641 | 5.332079012 |  |  |  |  |  | 0 |
| 661.18403 | $1.67 \mathrm{E}-11$ | 1.612875242 | 4.237786577 | 5.341303733 |  |  |  |  |  | 0 |
| 381.09822 | $5.64 \mathrm{E}-06$ | 6.149024326 | 5.835607973 | 5.341414123 | C12H24O11 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 344.131865 | 381.0983166 | -0.25 | ['Clusianose', 'Melibiitol'] |
| 381.09822 | $5.64 \mathrm{E}-06$ | 6.149024326 | 5.835607973 | 5.341414123 | C19H14O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 322.084125 | 381.0979786 | 0.63 | ['11-Deoxylandomycinone', 'Decarboxytetracenomycin F1', 'Tetrangomycin'] |
| 381.09822 | $5.64 \mathrm{E}-06$ | 6.149024326 | 5.835607973 | 5.341414123 | C21H18O7 | [M-H]- | 382.105255 | 381.0979786 | 0.63 | ['12-Deoxyaklanonic acid', 'Austrobailignan 1'] |
| 530.14524 | 6.80E-14 | 2.828210418 | 4.207140139 | 5.348573076 |  |  |  |  |  | 0 |
| 618.16178 | $1.07 \mathrm{E}-12$ | 3.671466761 | 4.853278989 | 5.350494582 |  |  |  |  |  | 0 |
| 647.1686 | $2.12 \mathrm{E}-11$ | 5.743247251 | 5.536474969 | 5.360319657 |  |  |  |  |  | 0 |
| 617.15819 | $1.66 \mathrm{E}-12$ | 3.638403671 | 4.977546041 | 5.377949175 |  |  |  |  |  | 0 |
| 679.19472 | $2.06 \mathrm{E}-10$ | 3.525742234 | 5.015495206 | 5.476751731 |  |  |  |  |  | 0 |
| 752.21963 | $4.32 \mathrm{E}-11$ | 4.510922069 | 5.998024793 | 5.490722673 |  |  |  |  |  | 0 |
| 511.13154 | $2.87 \mathrm{E}-12$ | 4.68716125 | 4.808392506 | 5.519845922 |  |  |  |  |  | 0 |
| 515.16246 | $1.09 \mathrm{E}-11$ | 2.539111307 | 4.726693924 | 5.539315506 |  |  |  |  |  | 0 |
| 516.16594 | $1.02 \mathrm{E}-11$ | 2.553197267 | 4.733402147 | 5.595365477 |  |  |  |  |  | 0 |
| 587.18306 | $1.50 \mathrm{E}-11$ | 3.472716447 | 4.974032729 | 5.683575364 |  |  |  |  |  | 0 |
| 527.1263 | 7.92E-11 | 6.316120548 | 5.51824541 | 5.732339742 |  |  |  |  |  | 0 |
| 720.15742 | $1.45 \mathrm{E}-10$ | 2.291393695 | 5.098757958 | 5.807161968 |  |  |  |  |  | 0 |
| 456.1084 | $2.89 \mathrm{E}-12$ | 5.923451133 | 5.916196631 | 5.82801174 |  |  |  |  |  | 0 |
| 328.08893 | 0.011037115 | 4.509415588 | 5.841314214 | 5.835602371 |  |  |  |  |  | 0 |
| 695.19009 | $1.26 \mathrm{E}-10$ | 2.874788141 | 4.989185067 | 5.86197269 |  |  |  |  |  | 0 |
| 755.40875 | $4.77 \mathrm{E}-05$ | 2.730065205 | 4.870748274 | 5.881440332 | C37H56N8O9 | [M-H]- | 756.417027 | 755.4097506 | -1.32 | ['Nostocyclopeptide A1'] |
| 680.19838 | $1.49 \mathrm{E}-10$ | 3.751405815 | 5.399771219 | 5.926927215 |  |  |  |  |  | 0 |
| 703.19511 | $1.12 \mathrm{E}-11$ | 1.564937359 | 4.320594688 | 6.043981212 |  |  |  |  |  | 0 |
| 469.12049 | $4.29 \mathrm{E}-13$ | 7.662945467 | 7.089481604 | 6.06900641 |  |  |  |  |  | 0 |
| 696.19332 | $4.45 \mathrm{E}-11$ | 2.840880343 | 5.086165884 | 6.110786651 |  |  |  |  |  | 0 |
| 446.0997 | $3.53 \mathrm{E}-07$ | 1.636396885 | 5.837012437 | 6.156311658 |  |  |  |  |  | 0 |
| 602.16659 | $4.99 \mathrm{E}-13$ | 4.976848889 | 5.93675965 | 6.182487467 |  |  |  |  |  | 0 |
| 583.15292 | 5.60E-09 | 3.29784217 | 6.031533634 | 6.379546719 |  |  |  |  |  | 0 |


| 397.09234 | $4.62 \mathrm{E}-10$ | 6.185792347 | 6.160626416 | 6.3900215 | C19H14O6 | [M+Hac-H]- | 338.07904 | 397.0928936 | -1.39 | ['8-O-Methylsterigmatocystin', 'Rabelomycin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 397.09234 | 4.62E-10 | 6.185792347 | 6.160626416 | 6.3900215 | C19H17N3O7 | [M-2H]- | 399.106652 | 397.0920992 | 0.61 | ['Nocardicin E', 'Nocardicin F'] |
| 397.09234 | $4.62 \mathrm{E}-10$ | 6.185792347 | 6.160626416 | 6.3900215 | C21H18O8 | [M-H]- | 398.10017 | 397.0928936 | -1.39 | ['Auramycinone', 'Dihydro-NAME', 'Nogalavinone'] |
| 541.14197 | 7.26E-13 | 4.019850483 | 6.254775031 | 6.523959072 |  |  |  |  |  | 0 |
| 794.1944 | $1.67 \mathrm{E}-11$ | 1.333453701 | 5.027521396 | 6.546063813 |  |  |  |  |  | 0 |
| 531.14928 | $6.80 \mathrm{E}-13$ | 4.820450536 | 5.742780238 | 6.556680162 |  |  |  |  |  | 0 |
| 662.18789 | $1.47 \mathrm{E}-11$ | 1.879992222 | 5.132835076 | 6.560013106 |  |  |  |  |  | 0 |
| 470.12398 | $1.15 \mathrm{E}-11$ | 8.186232737 | 7.574028134 | 6.574617087 |  |  |  |  |  | 0 |
| 501.11042 | 5.05E-11 | 6.952788685 | 6.354813894 | 6.603745333 |  |  |  |  |  | 0 |
| 546.10383 | $1.82 \mathrm{E}-11$ | 4.046323324 | 5.880156886 | 6.606658837 |  |  |  |  |  | 0 |
| 543.15762 | $9.30 \mathrm{E}-13$ | 4.856716006 | 6.64790088 | 6.627444047 |  |  |  |  |  | 0 |
| 543.12127 | $3.34 \mathrm{E}-10$ | 6.579225407 | 6.593936079 | 6.738426122 |  |  |  |  |  | 0 |
| 585.16841 | $1.68 \mathrm{E}-12$ | 4.845749712 | 6.453400107 | 6.757625761 |  |  |  |  |  | 0 |
| 648.17241 | $2.68 \mathrm{E}-11$ | 7.325608775 | 6.945043704 | 6.768653526 |  |  |  |  |  | 0 |
| 793.19075 | $1.17 \mathrm{E}-11$ | 1.498040829 | 5.402887803 | 7.034997915 |  |  |  |  |  | 0 |
| 779.17569 | $1.53 \mathrm{E}-05$ | 5.908123284 | 7.632136046 | 7.107758118 |  |  |  |  |  | 0 |
| 455.10491 | $2.43 \mathrm{E}-13$ | 7.033367087 | 7.276403284 | 7.123615427 |  |  |  |  |  | 0 |
| 528.12973 | $4.49 \mathrm{E}-11$ | 8.346128362 | 7.019242918 | 7.172258695 |  |  |  |  |  | 0 |
| 599.18407 | $3.41 \mathrm{E}-09$ | 5.549457022 | 7.215140775 | 7.395128143 | C28H38N2O8S | [M+K-2H]- | 562.23489 | 599.1834966 | 0.96 | ['TMC 126'] |
| 537.14704 | $4.70 \mathrm{E}-14$ | 1.863116594 | 5.66984365 | 7.595079287 |  |  |  |  |  | 0 |
| 507.11318 | 0.001717682 | 4.878033816 | 7.686915993 | 7.682674913 |  |  |  |  |  | 0 |
| 413.08921 | 8.22E-08 | 4.97158766 | 7.196870679 | 7.705675037 | C14H18N6O7S | [M-H]- | 414.095771 | 413.0884946 | 1.73 | ['Pyrazosulfuron-ethyl'] |
| 750.20418 | $7.81 \mathrm{E}-10$ | 3.102416835 | 7.066703625 | 7.764498072 |  |  |  |  |  | 0 |
| 343.06747 | $3.51 \mathrm{E}-08$ | 3.925149487 | 7.19797476 | 7.935675357 | C14H16O10 | [M-H]- | 344.07435 | 343.0670736 | 1.16 | ['Theogallin'] |
| 570.14018 | $1.24 \mathrm{E}-09$ | 2.141984972 | 5.660455602 | 7.938923684 |  |  |  |  |  | 0 |
| 495.13628 | $1.40 \mathrm{E}-08$ | 5.799401034 | 7.294473683 | 7.972850947 |  |  |  |  |  | 0 |
| 525.14702 | $2.00 \mathrm{E}-12$ | 5.486476207 | 7.595411095 | 7.975444857 |  |  |  |  |  | 0 |
| 495.07638 | $3.65 \mathrm{E}-11$ | 10.08800853 | 7.851615168 | 8.006697883 |  |  |  |  |  | 0 |
| 677.17936 | $5.16 \mathrm{E}-12$ | 3.308652367 | 6.442091375 | 8.299973545 |  |  |  |  |  | 0 |
| 629.15839 | $1.19 \mathrm{E}-12$ | 3.844999016 | 7.271293606 | 8.306780078 |  |  |  |  |  | 0 |
| 415.0986 | $9.72 \mathrm{E}-10$ | 6.734471344 | 8.153393803 | 8.363372353 | C19H22O8 | [M+(37Cl)]- | 378.13147 | 415.0979216 | 1.63 | ['Hydroxyvernolide'] |
| 415.0986 | $9.72 \mathrm{E}-10$ | 6.734471344 | 8.153393803 | 8.363372353 | C19H24O6S | [M+Cl]- | 380.129362 | 415.0987636 | -0.39 | ['2-Methoxyestrone 3-sulfate'] |
| 497.15195 | $2.75 \mathrm{E}-12$ | 5.294737203 | 7.005104167 | 8.392643735 |  |  |  |  |  | 0 |
| 437.13083 | $1.38 \mathrm{E}-09$ | 7.542201299 | 8.697908411 | 8.416934744 |  |  |  |  |  | 0 |
| 779.20064 | 3.01E-11 | 2.06644701 | 6.785560799 | 8.803377161 |  |  |  |  |  | 0 |
| 663.19964 | $1.45 \mathrm{E}-10$ | 5.726016774 | 8.341556022 | 8.907918858 | C36H36O10 | [M+Cl]- | 628.23085 | 663.2002516 | -0.92 | ['Gnidicin'] |


| 479.08155 | $3.48 \mathrm{E}-11$ | 10.14688996 | 8.737774404 | 8.984282499 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 553.1185 | $1.09 \mathrm{E}-10$ | 3.387685263 | 7.478865509 | 9.294921505 |  |  |  |  |  | 0 |
| 344.08383 | 0.023783086 | 6.109645326 | 9.797794241 | 9.430234125 |  |  |  |  |  | 0 |
| 559.15252 | $2.09 \mathrm{E}-12$ | 7.734747016 | 8.920349818 | 9.441713357 |  |  |  |  |  | 0 |
| 746.20935 | $1.06 \mathrm{E}-13$ | 1.337498395 | 7.636541121 | 9.65844685 |  |  |  |  |  | 0 |
| 791.21131 | $3.31 \mathrm{E}-12$ | 4.231635902 | 8.841848106 | 9.768733626 |  |  |  |  |  | 0 |
| 745.20572 | $1.51 \mathrm{E}-13$ | 2.132874421 | 7.938524152 | 9.796124872 |  |  |  |  |  | 0 |
| 567.15767 | $1.91 \mathrm{E}-07$ | 4.812891169 | 8.936862843 | 10.33728904 |  |  |  |  |  | 0 |
| 503.12613 | $6.27 \mathrm{E}-11$ | 7.31499257 | 9.20656377 | 10.35513103 |  |  |  |  |  | 0 |
| 382.10777 | $1.46 \mathrm{E}-09$ | 9.457479385 | 10.19610114 | 10.39993222 | C14H20N6O5S | [M-2H]- | 384.121591 | 382.1070382 | 1.92 | ['S-Adenosyl-L-homocysteine'] |
| 775.15728 | $3.71 \mathrm{E}-10$ | 1.992654063 | 6.721157393 | 10.4980891 |  |  |  |  |  | 0 |
| 678.1828 | $3.31 \mathrm{E}-12$ | 4.263292996 | 8.523987458 | 11.06139991 |  |  |  |  |  | 0 |
| 398.10264 | $2.73 \mathrm{E}-10$ | 11.37416403 | 11.10563554 | 11.23351374 |  |  |  |  |  | 0 |
| 704.16233 | $4.28 \mathrm{E}-12$ | 4.666570414 | 9.601626103 | 11.36385756 |  |  |  |  |  | 0 |
| 414.09761 | $1.38 \mathrm{E}-10$ | 8.369751407 | 10.9452172 | 11.52017027 | C16H21N3O8S | [M-H]- | 415.104939 | 414.0976626 | -0.13 | ['Cephalosporin $\mathrm{C}^{\prime}$ '] |
| 380.09219 | $3.32 \mathrm{E}-09$ | 4.891612984 | 10.12447648 | 11.52081408 |  |  |  |  |  | 0 |
| 413.094 | $1.49 \mathrm{E}-10$ | 8.398399564 | 10.96287418 | 11.56203685 |  |  |  |  |  | 0 |
| 397.0989 | $4.99 \mathrm{E}-10$ | 11.79525442 | 11.61466435 | 11.72303025 | C12H18O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 338.084915 | 397.0987686 | 0.33 | ['L-Ascorbic acid-2-glucoside'] |
| 397.0989 | $4.99 \mathrm{E}-10$ | 11.79525442 | 11.61466435 | 11.72303025 | C18H2ON2O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 360.132138 | 397.0985896 | 0.78 | ['3-Methoxytyramine-betaxanthin', 'Nitrendipine'] |
| 397.0989 | 4.99E-10 | 11.79525442 | 11.61466435 | 11.72303025 | C19H24N2O3S | [M+K-2H]- | 360.150765 | 397.0993716 | -1.19 | ['LY395153'] |
| 792.21485 | $5.77 \mathrm{E}-11$ | 4.248025456 | 10.74688758 | 11.83059543 |  |  |  |  |  | 0 |
| 399.10352 | 8.16E-11 | 12.60420092 | 12.18091629 | 12.24257854 | C19H22O7 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 362.136555 | 399.1030066 | 1.29 | ['Gibberellin A8-catabolite', 'Nagilactone C', 'Vernolide'] |
| 573.15736 | $9.50 \mathrm{E}-14$ | 4.59255658 | 10.26025809 | 12.30024596 | C30H29N3O7S | [M-2H]- | 575.172624 | 573.1580712 | -1.24 | ['4,5-Dihydro-4-hydroxy-5-S-glutathionylbenzo[a]pyrene', '7,8-Dihydro-7-hydroxy-8-S-glutathionyl-benzo[a]pyrene'] |
| 381.10423 | 2.07E-09 | 11.07994873 | 11.81653602 | 12.33280422 |  |  |  |  |  | 0 |
| 734.20924 | $4.70 \mathrm{E}-12$ | 4.642351692 | 11.42484538 | 12.34357052 |  |  |  |  |  | 0 |
| 569.11344 | 9.50E-14 | 5.563398333 | 9.986484331 | 12.34709474 |  |  |  |  |  | 0 |
| 430.09258 | $1.22 \mathrm{E}-09$ | 6.112166455 | 10.91138786 | 12.54872348 |  |  |  |  |  | 0 |
| 553.14194 | $1.06 \mathrm{E}-13$ | 2.315831307 | 9.549998813 | 12.92588482 |  |  |  |  |  | 0 |
| 619.13753 | 7.26E-13 | 4.10143325 | 9.94844802 | 13.06346676 |  |  |  |  |  | 0 |
| 601.16308 | $9.50 \mathrm{E}-14$ | 10.47704161 | 12.67352134 | 13.36008347 |  |  |  |  |  | 0 |
| 664.20352 | $8.60 \mathrm{E}-11$ | 8.821724098 | 13.45644948 | 14.4357901 |  |  |  |  |  | 0 |
| 569.17332 | 8.69E-12 | 8.400578141 | 13.28365651 | 14.79146622 |  |  |  |  |  | 0 |
| 776.21997 | $2.14 \mathrm{E}-13$ | 5.19953184 | 14.16045389 | 15.90449665 |  |  |  |  |  | 0 |
| 439.14641 | 0 | 5.94810494 | 12.51984178 | 16.31109327 |  |  |  |  |  | 0 |
| 455.14121 | 6.30E-14 | 6.94786087 | 14.31089701 | 16.67183593 | C21H26N2O7 | [M+(37Cl)]- | 418.174003 | 455.1404546 | 1.66 | ['Nimodipine'] |


| 456.14476 | $1.51 \mathrm{E}-13$ | 7.286628259 | 15.24144452 | 17.74454803 |  |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 429.08921 | 7.64E-11 | 9.180547371 | 16.57856969 | 18.90899972 | C12H18O13 | [M+Hac-H]- | 370.074745 | 429.0885986 | 1.42 | ['1,2-beta-D-Glucuronosyl-D-glucuronate', 'Digalacturonate'] |
| 703.15874 | $9.11 \mathrm{E}-13$ | 8.020614878 | 16.38850632 | 19.00504855 |  |  |  |  |  | 0 |
| 705.17443 | 3.73E-08 | 7.95055842 | 18.07650811 | 19.2106298 |  |  |  |  |  | 0 |
| 379.0887 | $2.44 \mathrm{E}-09$ | 8.356312636 | 16.6953025 | 19.23470062 |  |  |  |  |  | 0 |
| 569.13689 | $4.99 \mathrm{E}-13$ | 4.909548844 | 15.19201731 | 20.14635314 |  |  |  |  |  | 0 |
| 530.10891 | 9.50E-14 | 16.86506367 | 19.70278726 | 20.63243104 |  |  |  |  |  | 0 |
| 555.15756 | $1.51 \mathrm{E}-13$ | 8.055690325 | 18.03518958 | 20.96759429 |  |  |  |  |  | 0 |
| 620.14098 | 1.16E-12 | 5.665917232 | 15.87263805 | 21.00711113 | C28H29N2O12 | [M+Cl]- | 585.172053 | 620.1414546 | -0.77 | ['SN38 glucuronide carboxylate form'] |
| 761.20045 | $1.97 \mathrm{E}-13$ | 4.768902641 | 19.84072456 | 24.81735701 |  |  |  |  |  | 0 |
| 603.142 | $1.93 \mathrm{E}-13$ | 9.191864188 | 20.96769137 | 25.54446228 |  |  |  |  |  | 0 |
| 545.10051 | $1.62 \mathrm{E}-13$ | 15.8675972 | 24.22854326 | 26.5812912 |  |  |  |  |  | 0 |
| 529.10542 | $1.51 \mathrm{E}-13$ | 22.10648039 | 25.40386507 | 26.69687751 |  |  |  |  |  | 0 |
| 604.14574 | 9.50E-14 | 10.3551639 | 23.7997447 | 29.07942892 |  |  |  |  |  | 0 |
| 775.21666 | 8.40E-14 | 10.89801389 | 28.49036441 | 31.50678289 |  |  |  |  |  | 0 |
| 778.19927 | 5.65E-13 | 5.595429408 | 25.57649614 | 33.1719229 |  |  |  |  |  | 0 |
| 762.20414 | $2.14 \mathrm{E}-13$ | 6.271769002 | 29.65539184 | 37.34547994 |  |  |  |  |  | 0 |
| 777.1954 | $2.43 \mathrm{E}-12$ | 7.386652415 | 32.14790835 | 41.26623729 |  |  |  |  |  | 0 |
| 587.13865 | $9.63 \mathrm{E}-08$ | 8.13682682 | 36.44604042 | 42.1136558 | C23H34O15 | [M+K-2H]- | 550.189775 | 587.1383816 | 0.46 | ['Genipin 1-beta-gentiobioside'] |
| 587.14674 | $1.97 \mathrm{E}-13$ | 18.1069711 | 36.44604146 | 42.11365732 |  |  |  |  |  | 0 |
| 571.15223 | 6.30E-14 | 18.12710165 | 41.73818736 | 49.28429459 |  |  |  |  |  | 0 |
| 588.15103 | 6.30E-14 | 22.70332001 | 45.03445326 | 52.08031408 |  |  |  |  |  | 0 |
| 572.15598 | 6.30E-14 | 21.58985272 | 49.29691508 | 58.00088032 |  |  |  |  |  | 0 |

Table 3 Extraction of the aged laboratory sherds

|  | 10-Apr | 11-Apr-12 |  |  |  |  | 12-Apr-12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample Identification | weight (mg) | volume of solution 1 <br> (ul) | volume of ChCl 3 <br> (ul) | volume of water II <br> (ul) | volume of polar layer removed (ul) | volume of nonpolar removed (ul) | amount of 4M KOH <br> (ul) | 2N <br> HCl <br> (ul) | volume of ethyl acetate <br> (ul) | Estimated volume of depolymerised sample removed (ul) |
| Water_glass_1A | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_glass_1B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_glass_2A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_glass_2B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 300 |
| Water_glass_3A | 100 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | xxx |
| Water_glass_3B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_glass_4A | 105 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_glass_4B | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_sand_1A | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_sand_1B | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_sand_2A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_sand_2B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 350 |
| Water_sand_3A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 200 |
| Water_sand_3B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 200 |
| Water_sand_4A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_sand_4B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_1A | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_1B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_2A | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_2B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_3 | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Water_peat_3B | 103 | 560 | 400 | 200 | 400 | xxx | 150 | 400 | 450 | 400 |
| Water_peat_4A | 103 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Water_peat_4B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_glass_1A | 104 | 560 | 400 | 200 | 400 | xxx | 150 | 400 | 450 | 400 |
| Red_glass_1B | 100 | 560 | 400 | 200 | 400 | xxx | 150 | 400 | 450 | 400 |
| Red_glass_2A | 101 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Red_glass_2B | 105 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_glass_3A | 102 | 560 | 400 | 200 | 400 | xxx | 150 | 400 | 450 | 400 |
| Red_glass_3B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 200 |
| Red_glass_4A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_glass_4B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |


| Red_sand_1A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Red_sand_1B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_sand_2A | 101 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Red_sand_2B | 103 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Red_sand_3A | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_sand_3B | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_sand_4A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_sand_4B | 103 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Red_peat_1A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_1B | 101 | 560 | 400 | 200 | 400 | <200 | 150 | 400 | 450 | 400 |
| Red_peat_2A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_2B | 105 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_3A | 105 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_3B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_4A | 1004 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| Red_peat_4B | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_1A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_1B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_2A | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_2B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_3A | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_3B | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_4A | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_glass_4B | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_1A | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_1B | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_2A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_2B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_3A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_3B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_4A | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_sand_4B | 101 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_1A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_1B | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_2A | 104 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_2B | 103 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_3A | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| White_peat_3B | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |


| White_peat_4A | 100 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| White_peat_4B | 102 | 560 | 400 | 200 | 400 | 200 | 150 | 400 | 450 | 400 |

Table 4 A list of the ions, and the value at PC2 gathered from the PCA model of polar aliquot laboratory aged sherds, red vs. white. The top loadings represent the most characteristic ions for each samples, white wine and red wine.

| RED |  |  | WHITE |  |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{m} / \mathrm{z}$ | PC2 |  | $\mathrm{m} / \mathrm{z}$ | PC2 |
| 527.03951 | -0.194197 |  | 215.00466 | 0.223288 |
| 560.98543 | -0.1637913 |  | 133.01425 | 0.1833406 |
| 129.03594 | -0.1581585 |  | 653.21532 | 0.1547995 |
| 271.10313 | -0.1449812 |  | 483.13312 | 0.1534142 |
| 247.01336 | -0.1370775 |  | 463.14577 | 0.1357549 |
| 405.00264 | -0.1315703 |  | 459.13557 | 0.1308625 |
| 517.00433 | -0.1313375 |  | 377.08555 | 0.1290473 |
| 463.00853 | -0.1220809 |  | 387.11429 | 0.1130887 |
| 667.02731 | -0.1188552 |  | 402.13333 | 0.1070036 |
| 171.01486 | -0.1156026 |  | 401.12986 | 0.1058682 |
| 189.05752 | -0.1124271 |  | 217.00433 | 0.1020022 |
| 605.02044 | -0.1067045 |  | 483.12071 | 0.0987837 |
| 129.03645 | -0.1029562 |  | 217.02974 | 0.0973852 |
| 501.02638 | -0.1018672 |  | 263.04085 | 0.0954415 |
| 545.0116 | -0.1005002 |  | 215.03271 | 0.0934456 |
| 329.01647 | -0.0993011 |  | 397.24447 | 0.092283 |
| 171.02743 | -0.0982178 |  | 127.01675 | 0.0853919 |
| 384.27546 | -0.0948836 |  | 143.035 | 0.0846085 |
| 343.03704 | -0.0936314 |  | 341.10872 | 0.0843529 |
| 348.98245 | -0.092944 |  | 321.08015 | 0.0832489 |
| 490.99788 | -0.0917557 |  | 425.27584 | 0.0827095 |
| 128.0401 | -0.0912604 |  | 138.04892 | 0.080884 |
| 687.02433 | -0.0879469 |  | 137.04553 | 0.0783973 |
|  |  |  |  |  |


| 297.11895 | -0.0871712 |  | 217.00158 | 0.0757216 |
| :--- | :--- | :--- | :--- | :--- |
| 567.05197 | -0.0856982 |  | 439.08597 | 0.0749727 |
| 161.06248 | -0.0825848 |  | 420.23502 | 0.0744596 |
| 162.06596 | -0.0821917 |  | 143.99149 | 0.0741534 |
| 415.1456 | -0.0758327 |  | 129.01381 | 0.0730377 |
| 160.0662 | -0.0752306 |  | 478.98659 | 0.0726884 |
| 269.08748 | -0.0717381 |  | 329.2081 | 0.0699409 |
| 344.03805 | -0.071203 |  | 130.08737 | 0.069071 |
| 329.23314 | -0.0705731 |  | 217.00475 | 0.0662399 |
| 337.07742 | -0.0702788 |  | 203.02304 | 0.064961 |
| 348.98584 | -0.0637976 |  | 165.97897 | 0.0640402 |
| 211.0822 | -0.0635616 |  | 145.05066 | 0.0635167 |
| 346.01155 | -0.0614021 |  | 135.0663 | 0.0633933 |
| 345.03464 | -0.0596478 |  | 559.11953 | 0.0631855 |
| 199.03766 | -0.0590428 |  | 125.0432 | 0.0629459 |
| 411.00161 | -0.0544935 |  | 315.22888 | 0.0621023 |
| 399.08197 | -0.0540084 |  | 238.99371 | 0.06153 |
| 488.02344 | -0.0537299 |  | 463.32769 | 0.0612237 |
| 425.04009 | -0.0483255 |  | 191.07316 | 0.0606247 |
| 1254.7698 | -0.047535 |  | 199.02233 | 0.059333 |
| 487.04645 | -0.0451493 |  | 202.02655 | 0.0583972 |
| 1238.7748 | -0.044271 |  | 202.02622 | 0.0583972 |
| 343.12436 | -0.0436145 |  | 201.02539 | 0.0571529 |
| 441.01793 | -0.0434836 |  | 202.02885 | 0.0567729 |
| 485.04887 | -0.0431113 |  | 343.26006 | 0.0566648 |
| 443.19233 | -0.041562 |  | 239.07699 | 0.0563589 |
| 412.00214 | -0.0411227 |  | 173.0626 | 0.0550716 |
|  |  |  |  |  |

Table 5 The top 50 loadings from the wine polar aliquots and wine KOH aliquots (red and white wine inclusive).

| POLAR |  | KOH |  |
| :---: | :---: | :---: | :---: |
| m/z | PC1 | m/z | PC1 |
| 137.05 | -0.125 | 149.01 | -0.102 |
| 133.05 | -0.111 | 175.06 | -0.094 |
| 205.09 | -0.108 | 133.01 | -0.085 |
| 134.05 | -0.102 | 321.01 | -0.082 |
| 147.07 | -0.099 | 336.98 | -0.076 |
| 269.09 | -0.098 | 147.03 | -0.076 |
| 204.09 | -0.096 | 133.05 | -0.071 |
| 173.06 | -0.095 | 153.02 | -0.07 |
| 193.04 | -0.092 | 191.02 | -0.07 |
| 271.1 | -0.091 | 147.07 | -0.069 |
| 313.11 | -0.089 | 133.05 | -0.066 |
| 135.03 | -0.089 | 150.01 | -0.063 |
| 147.03 | -0.088 | 508.98 | -0.056 |
| 215 | -0.086 | 176.06 | -0.055 |
| 167.06 | -0.086 | 133.01 | -0.055 |
| 195.05 | -0.085 | 133.01 | -0.054 |
| 133.01 | -0.084 | 175.06 | -0.054 |
| 241.09 | -0.083 | 74.505 | -0.054 |
| 206.1 | -0.081 | 493.01 | -0.053 |
| 149.05 | -0.078 | 163.04 | -0.051 |
| 206.09 | -0.077 | 486.99 | -0.051 |
| 219.07 | -0.077 | 131 | -0.05 |


| POLAR |  | KOH |  |  |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{m} / \mathrm{z}$ | PC1 |  | $\mathrm{m} / \mathrm{z}$ | $\mathrm{PC1}$ |
| 197.07 | -0.076 |  | 220.01 | -0.047 |
| 315.13 | -0.074 |  | 211.03 | -0.046 |
| 138.05 | -0.074 |  | 530.96 | -0.046 |
| 181.07 | -0.072 |  | 206.97 | -0.046 |
| 117.02 | -0.071 |  | 144.03 | -0.045 |
| 129.04 | -0.07 |  | 181.01 | -0.045 |
| 148.07 | -0.07 |  | 477.07 | -0.044 |
| 135.05 | -0.068 |  | 193 | -0.044 |
| 255.11 | -0.067 |  | 307.01 | -0.044 |
| 165.04 | -0.066 |  | 195.03 | -0.044 |
| 253.09 | -0.065 |  | 144.05 | -0.043 |
| 475.17 | -0.063 |  | 134.05 | -0.042 |
| 211.08 | -0.062 |  | 187.04 | -0.042 |
| 461.13 | -0.06 |  | 322.01 | -0.042 |
| 161.06 | -0.06 |  | 197.05 | -0.042 |
| 217.02 | -0.059 | 320.99 | -0.042 |  |
| 311.1 | -0.057 | 117.02 | -0.041 |  |
| 146.05 | -0.056 |  | 342.99 | -0.04 |
| 211.01 | -0.056 | 329.09 | -0.04 |  |
| 179.06 | -0.056 | 189.02 | -0.04 |  |
| 160.07 | -0.055 |  | 151.01 | -0.038 |
| 212.03 | -0.055 |  | 209.01 | -0.038 |
|  |  |  |  |  |


| 199.02 | -0.077 |  | 215.07 | -0.05 |
| :--- | :--- | :--- | :--- | ---: |
| 343.12 | -0.076 |  | 179.04 | -0.049 |
| 214.99 | -0.076 |  | 524.95 | -0.047 |$\quad$|  | 209.07 | -0.055 |  | 196.99 |
| :--- | :--- | ---: | ---: | ---: |
| 397.07 | -0.054 |  | 199.17 | -0.038 |
| 189.06 | -0.054 |  | 103 | -0.037 |

Table 6 A list of the ions, corresponding intensity, and the value at PC2 gathered from the PCA model Figure 12b of the alkaline fusion laboratory aged sherds, red vs. white. The top loadings represent the most characteristic ions for each samples, white wine and red wine.

| Top 200 negative loadings for white wine |  |  | Top 200 positive loadings for red wine |  |  |  |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| M/Z | INTENSITY | PC2 |  | M/Z | INTENSITY | PC2 |
| 133.0147 | 71903.8 | -0.10587 |  | 338.97 | 1150248 | 0.145774 |
| 133.0137 | 82074.4 | -0.09579 |  | 197.0454 | 6107928 | 0.132806 |
| 477.0109 | 199390.7 | -0.09325 |  | 477.0673 | 531727.4 | 0.118234 |
| 305.0125 | 462880.7 | -0.08926 |  | 198.0491 | 414474.2 | 0.11277 |
| 473.2122 | 11932.8 | -0.08823 |  | 478.0708 | 119597.8 | 0.089112 |
| 267.0357 | 117614.7 | -0.08745 |  | 204.9911 | 121569 | 0.087793 |
| 505.3357 | 174178.9 | -0.08511 |  | 381.0171 | 107147.9 | 0.087318 |
| 470.994 | 139276.9 | -0.08471 |  | 187.0421 | 313918.4 | 0.086387 |
| 492.9849 | 176803.1 | -0.08138 |  | 327.0871 | 337434.4 | 0.08605 |
| 320.9863 | 859571.1 | -0.07991 |  | 169.0143 | 182359 | 0.085592 |
| 664.9836 | 98735.9 | -0.07882 |  | 199.0168 | 1125706 | 0.085588 |
| 476.99 | 115577.7 | -0.07541 |  | 253.0717 | 211403.7 | 0.083339 |
| 133.0142 | 16494795 | -0.07426 |  | 339.9733 | 80539.8 | 0.082664 |
| 493.0057 | 447212.5 | -0.07259 |  | 201.0139 | 345164.1 | 0.082125 |
| 320.9774 | 81105.8 | -0.07116 |  | 337.0198 | 206399.5 | 0.080551 |
| 461.0161 | 69129.2 | -0.06989 |  | 220.0015 | 67679.4 | 0.080186 |


| 326.9943 | 94775.9 | -0.06855 | 167.035 | 350068.8 | 0.078243 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 215.0174 | 167385.1 | -0.06821 | 611.2497 | 75170.1 | 0.074704 |
| 508.9798 | 460347.1 | -0.06695 | 125.0011 | 127918.5 | 0.074097 |
| 135.0186 | 145992.9 | -0.06628 | 117.0189 | 80536.2 | 0.073741 |
| 648.9887 | 56978.3 | -0.06564 | 293.0301 | 176702.6 | 0.073049 |
| 508.9594 | 65995.5 | -0.06432 | 119.0502 | 379780.7 | 0.071845 |
| 133.0155 | 3609.1 | -0.06416 | 188.9962 | 94910.2 | 0.070899 |
| 486.9889 | 341052 | -0.06402 | 187.0977 | 399486.6 | 0.068436 |
| 321.9897 | 35761.6 | -0.0633 | 329.0664 | 505228.1 | 0.067755 |
| 137.0609 | 181053 | -0.06311 | 328.0905 | 31023.9 | 0.066742 |
| 514.967 | 51574.7 | -0.06299 | 89.02438 | 57476.2 | 0.065648 |
| 275.0215 | 47151.7 | -0.06239 | 363.2385 | 30639.7 | 0.065598 |
| 649.0095 | 47489.1 | -0.06144 | 205.0144 | 98961.8 | 0.065051 |
| 680.9782 | 135886.4 | -0.06075 | 247.0247 | 32656 | 0.064841 |
| 680.958 | 59923 | -0.06063 | 295.2641 | 33168.3 | 0.064624 |
| 506.3391 | 52447.9 | -0.0588 | 202.9989 | 86451.7 | 0.064011 |
| 309.0825 | 51537.9 | -0.05846 | 407.0046 | 151572.9 | 0.062924 |
| 664.9633 | 49303.8 | -0.05725 | 183.03 | 115160.7 | 0.062921 |
| 665.0043 | 83147.6 | -0.057 | 186.0457 | 58457.4 | 0.06265 |
| 134.0183 | 45599.9 | -0.05688 | 138.0323 | 42004.6 | 0.062579 |
| 658.9953 | 39855.5 | -0.05674 | 331.0456 | 644847.9 | 0.062488 |
| 658.9887 | 66661.7 | -0.05612 | 199.05 | 25216.1 | 0.06229 |
| 199.1703 | 2247787 | -0.0542 | 211.025 | 262104.9 | 0.062031 |
| 342.9682 | 85536.3 | -0.05419 | 129.0365 | 111334.9 | 0.061633 |
| 321.0072 | 4254193 | -0.05349 | 374.3009 | 7910.7 | 0.06163 |
| 530.941 | 55674.2 | -0.0534 | 200.0202 | 106458.6 | 0.060818 |
| 658.9676 | 32114 | -0.05323 | 331.082 | 44186.4 | 0.060172 |
| 200.1738 | 253765.3 | -0.0528 | 444.9607 | 43176.9 | 0.058668 |
| 393.3008 | 74922.2 | -0.05225 | 163.04 | 1344736 | 0.057309 |


| 411.2359 | 51327.8 | -0.05211 | 355.3215 | 27417.7 | 0.05694 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 454.9991 | 37614.9 | -0.05155 | 190.9932 | 38303.2 | 0.05692 |
| 514.9879 | 75940.4 | -0.05124 | 319.0821 | 27502.2 | 0.056117 |
| 492.9642 | 29224.2 | -0.05048 | 345.228 | 332434.9 | 0.056019 |
| 322.9845 | 21379.3 | -0.05041 | 126.9981 | 40944.7 | 0.055918 |
| 322.0106 | 344746 | -0.05028 | 203.0119 | 28667.7 | 0.055095 |
| 509.9833 | 51762.1 | -0.05027 | 385.0177 | 35125.4 | 0.054984 |
| 173.0264 | 102621.2 | -0.05014 | 206.9882 | 19023.9 | 0.054152 |
| 478.0141 | 36201 | -0.04971 | 398.9549 | 29105.4 | 0.053882 |
| 530.9618 | 190465.1 | -0.04963 | 164.0435 | 114668.3 | 0.05311 |
| 494.0093 | 50165.4 | -0.04951 | 507.1146 | 22029.8 | 0.05264 |
| 524.9537 | 223538.8 | -0.04917 | 374.2991 | 127362.3 | 0.051869 |
| 483.4419 | 93626.7 | -0.04917 | 329.2331 | 2201009 | 0.05175 |
| 476.9815 | 26358.7 | -0.04786 | 330.2364 | 369463 | 0.051703 |
| 479.0099 | 63125.6 | -0.04732 | 209.0628 | 220540.3 | 0.051293 |
| 508.9509 | 24105.2 | -0.04647 | 128.0401 | 40673.2 | 0.051188 |
| 494.9835 | 40093.7 | -0.04646 | 411.2517 | 21538.8 | 0.050913 |
| 130.9987 | 298220 | -0.0462 | 311.1134 | 27918.2 | 0.050614 |
| 483.167 | 109726.7 | -0.04574 | 333.0613 | 79861.1 | 0.050355 |
| 197.0821 | 183841.5 | -0.04553 | 217.0122 | 11987.2 | 0.050111 |
| 508.972 | 74290 | -0.04538 | 515.0235 | 22779.5 | 0.050109 |
| 299.0254 | 538505.8 | -0.04538 | 413.3078 | 29674.6 | 0.050048 |
| 521.3308 | 34191.2 | -0.04524 | 383.3529 | 35462.9 | 0.050029 |
| 613.3745 | 37591.6 | -0.04467 | 153.0194 | 1799687 | 0.049715 |
| 259.1914 | 96167.2 | -0.04463 | 154.0228 | 94899.7 | 0.0493 |
| 427.231 | 70474.6 | -0.04437 | 481.3171 | 33057.7 | 0.04915 |
| 497.1826 | 104167.3 | -0.0443 | 188.1011 | 27975.9 | 0.048762 |
| 255.2309 | 959173.6 | -0.0441 | 191.0351 | 94746.1 | 0.047728 |
| 493.9881 | 21235 | -0.04386 | 373.2958 | 625185.2 | 0.047393 |


| 510.9787 | 28825.8 | -0.04367 | 327.2175 | 420766 | 0.047387 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 492.9768 | 31281.2 | -0.04354 | 120.0536 | 26293.9 | 0.047363 |
| 336.9706 | 30259.9 | -0.04351 | 202.0174 | 11448.4 | 0.047058 |
| 674.9618 | 43466.9 | -0.04344 | 197.0466 | 31070.9 | 0.04655 |
| 648.968 | 16588.9 | -0.04283 | 199.1715 | 9108.8 | 0.046404 |
| 674.9663 | 5725.8 | -0.04256 | 612.2532 | 21074.2 | 0.046073 |
| 665.9869 | 19618.6 | -0.04254 | 395.0983 | 15226.9 | 0.045887 |
| 199.1686 | 16430.2 | -0.04247 | 148.0697 | 92661.8 | 0.045811 |
| 511.1985 | 51475.3 | -0.04238 | 199.0968 | 8431.8 | 0.0458 |
| 487.9924 | 38597 | -0.04204 | 152.0116 | 18312.8 | 0.045501 |
| 473.4 | 114888.6 | -0.04194 | 138.0278 | 323556.6 | 0.045222 |
| 478.9885 | 21955.3 | -0.04181 | 319.0457 | 46926.5 | 0.045004 |
| 455.4107 | 36661.2 | -0.04154 | 161.0628 | 84723.9 | 0.044973 |
| 460.9951 | 16945.5 | -0.04148 | 319.0302 | 58867.5 | 0.044962 |
| 323.0115 | 77952.7 | -0.04137 | 499.0492 | 19536.7 | 0.044909 |
| 337.3109 | 170746.6 | -0.04113 | 163.0389 | 25534.2 | 0.044849 |
| 524.9454 | 80054.9 | -0.04044 | 137.0244 | 5126490 | 0.044817 |
| 539.3414 | 22925.7 | -0.04035 | 223.025 | 78464.2 | 0.044485 |
| 516.9657 | 30108.4 | -0.04033 | 173.0012 | 20127.5 | 0.044462 |
| 585.3432 | 23731.6 | -0.04003 | 328.2208 | 31153.9 | 0.044406 |
| 526.9524 | 23553.3 | -0.0398 | 535.1095 | 12683.5 | 0.044313 |
| 475.2008 | 42903.9 | -0.03962 | 138.0273 | 3458.9 | 0.04428 |
| 471.0029 | 11777.3 | -0.03913 | 221.0093 | 15493.1 | 0.044261 |
| 471.9975 | 10844.7 | -0.03824 | 329.0876 | 214837.2 | 0.044233 |
| 666.9818 | 20078.8 | -0.03787 | 331.2043 | 37486.3 | 0.044118 |
| 481.1877 | 13688.6 | -0.03779 | 208.0665 | 37983.7 | 0.043901 |
| 199.1692 | 49615.9 | -0.03772 | 147.0663 | 1883832 | 0.043791 |
| 307.0113 | 300145.2 | -0.03759 | 343.2123 | 237771 | 0.043294 |
| 595.5419 | 136565.3 | -0.03745 | 369.2645 | 24987.2 | 0.04322 |


| 571.1994 | 53079.4 | -0.03733 |  | 387.1659 | 42851.7 | 0.042923 |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| 477.9933 | 10798 | -0.03705 | 181.0143 | 180546.4 | 0.042825 |  |
| 681.982 | 18580.2 | -0.03654 |  | 395.2775 | 14448 | 0.04117 |
| 530.9325 | 14962 | -0.03596 |  | 199.0249 | 70983.8 | 0.041154 |
| 467.1722 | 10946.9 | -0.03593 |  | 345.2643 | 159427.1 | 0.041043 |
| 555.2044 | 7387.8 | -0.03565 |  | 641.4635 | 19354.6 | 0.040428 |
| 165.9792 | 25547.4 | -0.03556 |  | 483.3327 | 29554.3 | 0.040348 |
| 463.015 | 17536.3 | -0.03544 |  | 205.161 | 3472.7 | 0.040255 |
| 525.9572 | 19640.1 | -0.0352 |  | 449.109 | 8430.7 | 0.039971 |
| 530.9536 | 37396.2 | -0.03504 |  | 195.0291 | 7881.1 | 0.039855 |
| 659.9987 | 7092.6 | -0.03475 |  | 493.0988 | 8212.7 | 0.039855 |
| 133.013 | 8655.3 | -0.03457 |  | 197.0093 | 78631 | 0.039801 |
| 650.9867 | 8322.2 | -0.03454 |  | 125.0244 | 45838.1 | 0.039212 |
| 439.2677 | 8179.1 | -0.03454 |  | 213.0406 | 51947.8 | 0.039157 |
| 172.03 | 12744 | -0.03444 |  | 581.2395 | 10222.3 | 0.038623 |
| 484.4453 | 23223.6 | -0.03438 |  | 447.3117 | 31808.3 | 0.038492 |
| 492.9557 | 10313.1 | -0.03428 |  | 190.952 | 83211 | 0.038238 |
| 512.4766 | 110661.2 | -0.03408 |  | 179.0351 | 313212.1 | 0.038172 |
| 536.949 | 8323.3 | -0.03393 |  | 147.0452 | 39486.2 | 0.038048 |
| 471.0239 | 17836.2 | -0.03383 |  | 207.0301 | 116377.9 | 0.038018 |
| 283.2594 | 62435.3 | -0.03381 |  | 466.9427 | 12645.4 | 0.037371 |
| 482.998 | 8126.3 | -0.03369 |  | 131.0714 | 494901.4 | 0.037072 |
| 474.4033 | 31372.3 | -0.03319 | 70728.7 | 0.036934 |  |  |
| 511.4732 | 324923 | -0.03309 |  | 181.0507 | 255.0509 | 130108.2 |


| 718.9345 | 20686.2 | -0.03224 | 209.0093 | 101160.1 | 0.035661 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 484.1703 | 14999.7 | -0.03166 | 271.0822 | 17185.8 | 0.035316 |
| 498.1859 | 14839.1 | -0.0316 | 482.9166 | 7630.1 | 0.035099 |
| 729.6408 | 17820.8 | -0.03154 | 195.03 | 323515.6 | 0.035098 |
| 532.9394 | 15306.1 | -0.03149 | 243.1237 | 49227.2 | 0.034765 |
| 337.9845 | 133588.5 | -0.03136 | 395.2747 | 144163.1 | 0.034657 |
| 462.0191 | 6040.5 | -0.03131 | 211.0614 | 76892.5 | 0.034576 |
| 220.1457 | 897647.3 | -0.03108 | 445.2959 | 17422.6 | 0.034392 |
| 546.9273 | 41210.8 | -0.03104 | 383.2438 | 19751.5 | 0.034229 |
| 666.9608 | 6494 | -0.03095 | 151.0402 | 191046.4 | 0.034143 |
| 546.9358 | 95855.4 | -0.03087 | 293.2121 | 413741.2 | 0.033659 |
| 514.9588 | 7747 | -0.03084 | 519.2731 | 6385 | 0.03349 |
| 336.9812 | 2099819 | -0.03043 | 213.1134 | 160215.6 | 0.03341 |
| 531.9653 | 17235.9 | -0.0303 | 217.1083 | 324345.5 | 0.033214 |
| 338.9793 | 89595.1 | -0.02989 | 260.8975 | 30841.4 | 0.033117 |
| 529.4626 | 244608.1 | -0.02982 | 313.0927 | 260684.2 | 0.033098 |
| 489.2165 | 13518.7 | -0.02963 | 139.0287 | 14041.7 | 0.032826 |
| 757.6717 | 23179.7 | -0.02932 | 148.0333 | 40194.2 | 0.032436 |
| 461.1851 | 25286.8 | -0.0293 | 427.234 | 25770.2 | 0.03238 |
| 679.3176 | 79936.9 | -0.0293 | 191.0198 | 860206.9 | 0.0317 |
| 484.9968 | 6046.4 | -0.02927 | 146.9831 | 4200.4 | 0.031593 |
| 533.4065 | 52927.3 | -0.02924 | 357.1965 | 229198.5 | 0.031495 |
| 342.9892 | 223427.3 | -0.02916 | 123.0452 | 9926.4 | 0.030545 |
| 336.9723 | 119011.1 | -0.02828 | 147.03 | 1233057 | 0.030543 |
| 666.0079 | 8297 | -0.02784 | 327.1082 | 26625.9 | 0.030443 |
| 193.1348 | 209198.8 | -0.02781 | 294.2154 | 26600.1 | 0.03032 |
| 132.002 | 5160.5 | -0.02765 | 639.4479 | 7313.5 | 0.030171 |
| 530.4661 | 82362.7 | -0.0276 | 743.1623 | 10860.9 | 0.030088 |
| 536.9699 | 14082.7 | -0.02751 | 383.2074 | 22294.2 | 0.030068 |


| 582.3496 | 8426.6 | -0.02741 |  | 325.2018 | 54031.8 | 0.029803 |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| 552.9441 | 37987.2 | -0.02731 |  | 73.53314 | 17336.7 | 0.029782 |
| 265.1478 | 173713.2 | -0.02731 |  | 205.009 | 1850.8 | 0.029757 |
| 396.9935 | 11606.8 | -0.02723 |  | 665.1516 | 9882.7 | 0.029719 |
| 650.0128 | 3838.6 | -0.02701 |  | 132.9464 | 291371.9 | 0.029676 |
| 527.1724 | 31993.2 | -0.02648 |  | 413.2543 | 10933 | 0.029663 |
| 671.1398 | 7420.8 | -0.02628 |  | 317.0665 | 22657.4 | 0.029496 |
| 198.0409 | 23883 | -0.02625 |  | 555.3539 | 5563.9 | 0.029231 |
| 476.9691 | 11858.9 | -0.02621 |  | 132.0748 | 18603.9 | 0.029085 |
| 561.4378 | 90041.5 | -0.02615 |  | 329.1653 | 162442.6 | 0.029007 |
| 597.5575 | 37226.8 | -0.02609 |  | 270.978 | 14639.2 | 0.028862 |
| 552.923 | 7037.5 | -0.0259 |  | 151.0038 | 55326.4 | 0.028731 |
| 513.1567 | 39052.7 | -0.02571 |  | 341.1969 | 25697.2 | 0.028413 |
| 670.9713 | 11941.6 | -0.02557 |  | 403.1657 | 16295.2 | 0.028097 |
| 499.1409 | 34451.8 | -0.02532 |  | 655.4436 | 9303.3 | 0.027996 |
| 227.2016 | 306456.6 | -0.02512 |  | 121.0295 | 1673497 | 0.027766 |
| 221.1491 | 66900.8 | -0.025 |  | 505.0622 | 4661 | 0.027596 |
| 614.3779 | 8730 | -0.02486 |  | 333.22 | 135.0452 | 49124.8 |


| 655.166 | 10449.8 | -0.02332 |  | 192.9492 | 23599.5 | 0.025412 |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| 473.2825 | 30373.5 | -0.02316 |  | 192.0232 | 33143.7 | 0.025314 |
| 685.1553 | 5685.1 | -0.0228 |  | 525.1035 | 8640.3 | 0.024654 |
| 133.0501 | 74438.4 | -0.02277 |  | 127.0037 | 16504.5 | 0.024618 |
| 572.2027 | 4288.4 | -0.02266 |  | 152.0354 | 14025.8 | 0.024386 |
| 397.2566 | 24519.9 | -0.02259 |  | 339.3266 | 192104.1 | 0.023777 |
| 727.2669 | 64700.5 | -0.02209 |  | 371.2802 | 26932.3 | 0.023713 |
| 304.9915 | 88300.3 | -0.02205 |  | 401.0877 | 125876.4 | 0.023196 |
| 403.3248 | 110704.7 | -0.02184 |  | 277.2172 | 333304 | 0.02306 |
| 455.2417 | 9024.5 | -0.0218 |  | 144.0456 | 256363.6 | 0.022653 |

Table 7 A list of the ions, corresponding intensity, and the value at PC1 gathered from figure 3b of water vs. wine, KOH aliquot. The table is sorted on the top 150 values of PC1 from the smallest to the largest value.

| M/Z | INTENSITY | PC1 | Empirical formula (parent) | Empirical formula (peak) | Ion form | Theoretical mass (neutral) (Da) | Theoretical m/z (Da) | Mass error (ppm) | KEGG_COMPOUND |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 149.00912 | 12839920.3 | -0.102401503 | C2H2O4 | C2H2O4 | [M+Hac-H]- | 89.99531 | 149.0091636 | -0.29 | ['Oxalate'] |
| 149.00912 | 12839920.3 | -0.102401503 | C4H6O6 | C4H6O6 | [M-H]- | 150.01644 | 149.0091636 | -0.29 | ['(R,R)-Tartaric acid', '(S,S)-Tartaric acid', 'meso-Tartaric acid'] |
| 175.06112 | 4971425.5 | -0.094286393 | C5H8O3 | C5H8O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 116.047345 | 175.0611986 | -0.45 | ['2-Oxopentanoic acid', '3-Methyl-2-oxobutanoic acid', '3-Oxopentanoic acid', '5-Oxopentanoate'] |
| 175.06112 | 4971425.5 | -0.094286393 | C7H12O5 | C7H12O5 | [M-H]- | 176.068475 | 175.0611986 | -0.45 | ['(2R,3S)-3-Isopropylmalate', '(R)-2-(n-Propyl)-malate', '2-PropyImalate', '3-PropyImalate', 'alphaIsopropyImalate'] |
| 133.01418 | 16494794.9 | -0.085369204 | C2H2O3 | C2H2O3 | [M+Hac-H]- | 74.000395 | 133.0142486 | -0.52 | ['Glyoxylate'] |
| 133.01418 | 16494794.9 | -0.085369204 | C4H6O5 | C4H6O5 | [M-H]- | 134.021525 | 133.0142486 | -0.52 | ['(R)-Malate', '(S)-Malate', '3-Dehydro-L-threonate', 'Malate'] |
| 321.00724 | 4254192.7 | -0.082271453 |  | 0 |  |  |  |  | 0 |
| 336.98117 | 2099819.3 | -0.075821148 |  | 0 |  |  |  |  | 0 |
| 147.02996 | 1233057 | -0.075751737 | C3H4O3 | C3H4O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 88.016045 | 147.0298986 | 0.42 | ['3-Hydroxypropenoate', '3-Oxopropanoate', 'Pyruvate'] |
| 147.02996 | 1233057 | -0.075751737 | C5H8O5 | C5H8O5 | [M-H]- | 148.037175 | 147.0298986 | 0.42 | ['(R)-2-Hydroxyglutarate', '(R)-2-Methylmalate', '(S)-2Hydroxyglutarate', '(S)-2-Methylmalate', '2-Dehydro-3-deoxy-D-xylonate', '2-Dehydro-3-deoxy-L-arabinonate', '2-Hydroxyglutarate', 'Citramalate', 'D-Arabinono-1,4lactone', 'D-Xylono-1,4-lactone', 'D-Xylonolactone', 'D-erythro-3-Methylmalate', 'D-threo-3-Methylmalate', 'L-Arabinono-1,4-lactone', 'L-Arabinono-1,5-lactone', 'L-Xylono-1,4-lactone', 'L-threo-3-Methylmalate'] |


| 133.05012 | 74438.4 | -0.070841319 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 153.01939 | 1799686.6 | -0.070222273 | C7H6O4 | C7H6O4 | [M-H]- | 154.02661 | 153.0193336 | 0.37 | ['2,3-Dihydroxybenzoate', '2,5-Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 191.01982 | 860206.9 | -0.070063079 | C4H4O5 | C4H4O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 132.005875 | 191.0197286 | 0.48 | ['2-Hydroxyethylenedicarboxylate', 'Oxaloacetate', 'trans-2,3-Epoxysuccinate'] |
| 191.01982 | 860206.9 | -0.070063079 | C6H8O7 | C6H8O7 | [M-H]- | 192.027005 | 191.0197286 | 0.48 | ['(1R,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(1S,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(4R,5S)-4,5,6-Trihydroxy-2,3-dioxohexanoate', '2,5-Didehydro-D-gluconate', '2-Dehydro-3-deoxy-Dglucarate', '5-Dehydro-4-deoxy-D-glucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] |
| 147.06634 | 1883832.2 | -0.068889369 | C4H8O2 | C4H8O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 88.05243 | 147.0662836 | 0.38 | ['(R)-Acetoin', '1,4-Dioxane', '2-Methylpropanoate', 'Acetoin', 'Butanoic acid', 'Ethyl acetate'] |
| 147.06634 | 1883832.2 | -0.068889369 | C6H12O4 | C6H12O4 | [M-H]- | 148.07356 | 147.0662836 | 0.38 | ['(R)-2,3-Dihydroxy-3-methylpentanoate', '(R)Mevalonate', '(R)-Pantoate', '(S)-Mevalonate', '2,3-Dihydroxy-3-methylpentanoate', '3,6-Dideoxy-Lgalactose', 'Abequose'] |
| 133.05061 | 3301580.2 | -0.06607991 | C3H6O2 | C3H6O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 74.03678 | 133.0506336 | -0.18 | ['(R)-Lactaldehyde', '(S)-Lactaldehyde', '3Hydroxypropanal', 'Glycidol', 'Hydroxyacetone', 'Lactaldehyde', 'Methyl acetate', 'Propanoate'] |
| 133.05061 | 3301580.2 | -0.06607991 | C5H1004 | C5H1004 | [M-H]- | 134.05791 | 133.0506336 | -0.18 | ['(R)-2,3-Dihydroxy-3-methylbutanoate', '1-Deoxy-Dxylulose', '2,3-Dihydroxy-3-methylbutanoate', '2-Deoxy-L-arabinose', '2-Deoxy-alpha-D-ribopyranose', 'Deoxyribose'] |
| 150.01262 | 395150 | -0.062934511 |  | 0 |  |  |  |  | 0 |
| 508.97983 | 460347.1 | -0.055860612 |  | 0 |  |  |  |  | 0 |
| 176.06465 | 321586.9 | -0.05546201 |  | 0 |  |  |  |  | 0 |
| 133.01365 | 82074.4 | -0.055236115 |  | 0 |  |  |  |  | 0 |


| 133.0147 | 71903.8 | -0.054447121 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 175.06207 | 7760.3 | -0.053993069 |  | 0 |  |  |  |  | 0 |
| 74.50455 | 214325.8 | -0.053743837 |  | 0 |  |  |  |  | 0 |
| 493.00574 | 447212.5 | -0.052869566 |  | 0 |  |  |  |  | 0 |
| 163.04002 | 1344736.2 | -0.050834612 | C9H8O3 | C9H8O3 | [M-H]- | 164.047345 | 163.0400686 | -0.3 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2-Hydroxycinnamate'] |
| 486.98886 | 341052 | -0.050511794 |  | 0 |  |  |  |  | 0 |
| 130.99865 | 298220 | -0.049814628 | C4H4O5 | C4H4O5 | [M-H]- | 132.005875 | 130.9985986 | 0.39 | ['2-Hydroxyethylenedicarboxylate', 'Oxaloacetate', 'trans-2,3-Epoxysuccinate'] |
| 215.07339 | 235640.8 | -0.049587507 |  | 0 |  |  |  |  | 0 |
| 179.03508 | 313212.1 | -0.048992668 | C9H8O4 | C9H8O4 | [M-H]- | 180.04226 | 179.0349836 | 0.54 | ['2-Hydroxy-3-(4-hydroxyphenyl)propenoate', '3-(4Hydroxyphenyl)pyruvate', 'Aspirin', 'Caffeate', 'trans-2,3-Dihydroxycinnamate'] |
| 524.95372 | 223538.8 | -0.047072946 |  | 0 |  |  |  |  | 0 |
| 220.00756 | 258793.6 | -0.046855575 |  | 0 |  |  |  |  | 0 |
| 211.02497 | 262104.9 | -0.046164544 | C9H8O6 | C9H8O6 | [M-H]- | 212.03209 | 211.0248136 | 0.74 | ['2-Hydroxy-6-ketononatrienedioate', '3-(2-Carboxyethenyl)-cis,cis-muconate', '5-Carboxyvanillic acid'] |
| 530.96179 | 190465.1 | -0.045647873 |  | 0 |  |  |  |  | 0 |
| 206.96793 | 191453.9 | -0.045614948 | C3H707P | C3H707P | [M+Na-2H]- | 185.992943 | 206.9676116 | 1.54 | ['2-Phospho-D-glycerate', '3-Phospho-D-glycerate', '3-Phospho-DL-glycerate'] |
| 144.03033 | 141336.5 | -0.045401167 | C5H7NO4 | C5H7NO4 | [M-H]- | 145.037509 | 144.0302326 | 0.68 | ['2-Oxoglutaramate', '4-Oxoglutaramate'] |

$\left.\begin{array}{|l|l|l|l|l|l|l|l|l|l|}\hline 181.01434 & 180546.4 & -0.04498824 & \text { C8H6O5 } & \text { C8H6O5 } & \text { [M-H]- } & 182.021525 & 181.0142486 & & 0.5 \\ \text { ['2-Hydroxyisophthalic acid', '3,5- } \\ \text { Dihydroxyphenylglyoxylate', '4-Hydroxyphthalate', } \\ \text { 'Stipitatate'] }\end{array}\right]$

| 117.0193 | 2144058.9 | -0.041306416 | C4H6O4 | C4H6O4 | [M-H]- | 118.02661 | 117.0193336 | -0.29 | ['Methyl oxalate', 'Methylmalonate', 'Succinate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 342.98921 | 223427.3 | -0.040309349 |  | 0 |  |  |  |  | 0 |
| 329.08755 | 214837.2 | -0.040272106 | C18H16N2O2 | C18H16N2O2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 292.121178 | 329.0876296 | -0.24 | ['INF271'] |
| 189.02132 | 99540 | -0.040084789 |  | 0 |  |  |  |  | 0 |
| 151.01354 | 79540.4 | -0.038475372 |  | 0 |  |  |  |  | 0 |
| 209.00931 | 101160.1 | -0.038331243 |  | 0 |  |  |  |  | 0 |
| 196.99156 | 94211.2 | -0.037744932 |  | 0 |  |  |  |  | 0 |
| 199.17149 | 9108.8 | -0.037652533 |  | 0 |  |  |  |  | 0 |
| 103.00368 | 88796.1 | -0.037181229 | C3H4O4 | C3H4O4 | [M-H]- | 104.01096 | 103.0036836 | -0.03 | ['2-Hydroxy-3-oxopropanoate', 'Hydroxypyruvate', 'Malonate'] |
| 103.00368 | 88796.1 | -0.037181229 | CO2 | CO2 | [ $\mathrm{M}+$ Hac- H$]$ - | 43.98983 | 103.0036836 | -0.03 | ['CO2'] |
| 169.01433 | 182359 | -0.036673686 | C7H6O5 | C7H6O5 | [M-H]- | 170.021525 | 169.0142486 | 0.48 | ['Gallate'] |
| 154.0228 | 94899.7 | -0.036359313 |  | 0 |  |  |  |  | 0 |
| 197.00934 | 78631 | -0.035967678 | C8H6O6 | C8H6O6 | [M-H]- | 198.01644 | 197.0091636 | 0.9 | ['3,4-Dihydroxyphthalate', '4,5-Dihydroxyphthalate'] |
| 148.06974 | 92661.8 | -0.035767219 |  | 0 |  |  |  |  | 0 |
| 546.93579 | 95855.4 | -0.035692706 |  | 0 |  |  |  |  | 0 |
| 524.9454 | 80054.9 | -0.035499922 |  | 0 |  |  |  |  | 0 |
| 200.09299 | 198539.2 | -0.035489295 | C7H11NO2 | C7H11NO2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 141.078979 | 200.0928326 | 0.79 | ['Arecaidine', 'Ethosuximide', 'Guvacoline', 'LHypoglycin'] |
| 305.01245 | 462880.7 | -0.035406426 |  | 0 |  |  |  |  | 0 |
| 103 | 133588.5 | -0.035353525 |  | 0 |  |  |  |  | 0 |


| 207.03006 | 116377.9 | -0.035199659 | C10H8O5 | C10H8O5 | [M-H]- | 208.037175 | 207.0298986 | 0.78 | ['Fraxetin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 199.01683 | 1125705.9 | -0.03496134 | C10H10O2 | C10H10O2 | [M+K-2H]- | 162.06808 | 199.0166866 | 0.72 | ['(1S,2S)-1,2-Dihydronaphthalene-1,2-diol', '1,2- <br> Dihydronaphthalene-1,2-diol', '4- <br> Hydroxycinnamoylmethane', 'Isosafrole', 'Methyl <br> cinnamate', 'Safrole', 'cis-1,2-Dihydronaphthalene-1,2- <br> diol', 'p-Methoxycinnamaldehyde', 'trans-2- <br> Phenylcyclopropanecarboxylic acid'] |
| 199.01683 | 1125705.9 | -0.03496134 | C9H8O3 | C9H8O3 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 164.047345 | 199.0167466 | 0.42 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2-Hydroxycinnamate'] |
| 299.02541 | 538505.8 | -0.034655317 | C12H1ON2O5 | C12H10N2O5 | [M+(37CI)]- | 262.058973 | 299.0254246 | -0.05 | ['Cinoxacin'] |
| 135.01857 | 145992.9 | -0.034305511 |  | 0 |  |  |  |  | 0 |
| 164.04351 | 114668.3 | -0.033715967 |  | 0 |  |  |  |  | 0 |
| 183.03001 | 115160.7 | -0.033479499 | C6H4O3 | C6H4O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 124.016045 | 183.0298986 | 0.61 | ['2-Hydroxy-1,4-benzoquinone'] |
| 183.03001 | 115160.7 | -0.033479499 | C8H8O5 | C8H8O5 | [M-H]- | 184.037175 | 183.0298986 | 0.61 | ['3,4-Dihydroxymandelate', '3-0-Methylgallate'] |
| 373.29583 | 625185.2 | -0.03328096 |  | 0 |  |  |  |  | 0 |
| 223.02498 | 78464.2 | -0.033183711 |  | 0 |  |  |  |  | 0 |
| 199.02492 | 70983.8 | -0.033044207 | C6H4O4 | C6H4O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 140.01096 | 199.0248136 | 0.53 | ['cis-4-Carboxymethylenebut-2-en-4-olide', 'trans-4-Carboxymethylenebut-2-en-4-olide'] |
| 199.02492 | 70983.8 | -0.033044207 | C8H8O6 | C8H8O6 | [M-H]- | 200.03209 | 199.0248136 | 0.53 | ['(3S,4R)-3,4-Dihydroxycyclohexa-1,5-diene-1,4dicarboxylate', '2-Hydroxy-5-carboxymethylmuconate semialdehyde', '3-Carboxymethylmuconate', '4Fumarylacetoacetate', '4-Maleylacetoacetate', 'Phthalate 3,4-cis-dihydrodiol', 'cis-4,5-Dihydroxycyclohexa-1(6),2-diene-1,2-dicarboxylate'] |
| 319.02824 | 54064.5 | -0.03293396 | C12H15N2O3PS | C12H15N2O3PS | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 298.054103 | 319.0287716 | -1.67 | ['Phoxim', 'Quinalphos'] |


| 149.00817 | 15207.7 | -0.03286484 | C3H4N4O2 | C3H4N4O2 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 128.033426 | 149.0080946 | 0.51 | ['Ammelide'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 119.05024 | 379780.7 | -0.032784001 | C8H8O | C8H8O | [M-H]- | 120.057515 | 119.0502386 | 0.01 | ['2-Methylbenzaldehyde', '3-Methylbenzaldehyde', '4- <br> Hydroxystyrene', 'Acetophenone', <br> 'Phenylacetaldehyde', 'Styrene oxide', 'p- <br> Tolualdehyde'] |
| 680.97824 | 135886.4 | -0.032728389 |  | 0 |  |  |  |  | 0 |
| 205.01439 | 98961.8 | -0.032722519 | C10H6O5 | C10H605 | [M-H]- | 206.021525 | 205.0142486 | 0.69 | ['Flaviolin'] |
| 173.02638 | 102621.2 | -0.032478461 |  | 0 |  |  |  |  | 0 |
| 198.04905 | 414474.2 | -0.032454592 |  | 0 |  |  |  |  | 0 |
| 201.01393 | 345164.1 | -0.032280828 | C9H8O3 | C9H8O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 164.047345 | 201.0137966 | 0.66 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2-Hydroxycinnamate', 'cis-pCoumarate', 'trans-2-Hydroxycinnamate'] |
| 189.07696 | 334513.5 | -0.031522479 | C6H1003 | C6H1003 | [M+Hac-H]- | 130.062995 | 189.0768486 | 0.59 | ['(3R)-3-Methyl-2-oxopentanoic acid', '(R)Pantolactone', '(S)-3-Methyl-2-oxopentanoic acid', '1-Oxa-2-oxo-3-hydroxycycloheptane', '2-Hydroxyethyl methacrylate', '2-Oxohexanoic acid', '3-Methyl-2oxopentanoate', '3-Oxohexanoic acid', '4-Methyl-2oxopentanoate', '5-Oxohexanoic acid', '6-Hydroxyhexan-6-olide', 'Adipate semialdehyde', 'Ethyl 3-oxobutanoate', 'beta-Ketoisocaproate'] |
| 189.07696 | 334513.5 | -0.031522479 | C8H14O5 | C8H1405 | [M-H]- | 190.084125 | 189.0768486 | 0.59 | ['(R)-3-((R)-3-Hydroxybutanoyloxy)butanoate'] |
| 329.06638 | 505228.1 | -0.031465144 | C15H1005 | C15H1005 | [M+Hac-H]- | 270.052825 | 329.0666786 | -0.91 | ['2-Hydroxydaidzein', "3',4',7-Trihydroxyisoflavone", '3,6,4-Trihydroxyflavone', "4',6,7- <br> Trihydroxyisoflavone", '5-Deoxykaempferol', 'Aloeemodin', 'Apigenin', 'Baicalein', 'Emodin', 'Galangin', 'Genistein', 'Islandicin', 'Lucidin', 'Morindone', 'Norobtusifolin', 'Norwogonin', 'Purpurin 1-methyl ether', 'Sulphuretin'] |


| 329.06638 | 505228.1 | -0.031465144 | C17H14O7 | C17H14O7 | [M-H]- | 330.073955 | 329.0666786 | -0.91 | ['(+)-Bisdechlorogeodin', '(-)-Bisdechlorogeodin', "3',4',5-Trihydroxy-3,7-dimethoxyflavone", 'Aflatoxin G2', 'Aurantio-obtusin', 'Cirsiliol', 'Hildecarpin', 'Tricin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 325.07742 | 78652.1 | -0.031399294 |  | 0 |  |  |  |  | 0 |
| 378.99191 | 64958.4 | -0.031343078 |  | 0 |  |  |  |  | 0 |
| 205.03551 | 30085.2 | -0.03117134 | C5H6O5 | C5H6O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 146.021525 | 205.0353786 | 0.64 | ['2-Oxoglutarate', '5-Hydroxy-2,4-dioxopentanoate', 'Dehydro-D-arabinono-1,4-lactone', <br> 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] |
| 205.03551 | 30085.2 | -0.03117134 | C7H1007 | C7H1007 | [M-H]- | 206.042655 | 205.0353786 | 0.64 | ['(2S,3R)-3-Hydroxybutane-1,2,3-tricarboxylate', '(R)-2-Hydroxybutane-1,2,4-tricarboxylate', '2-Methylcitrate', 'Homoisocitrate'] |
| 492.98493 | 176803.1 | -0.031145628 |  | 0 |  |  |  |  | 0 |
| 202.99892 | 86451.7 | -0.031129631 |  | 0 |  |  |  |  | 0 |
| 358.9631 | 134325.5 | -0.031096467 |  | 0 |  |  |  |  | 0 |
| 508.97201 | 74290 | -0.030919315 |  | 0 |  |  |  |  | 0 |
| 407.00459 | 151572.9 | -0.030868335 |  | 0 |  |  |  |  | 0 |
| 161.04562 | 67109.1 | -0.030741974 | C4H6O3 | C4H6O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 102.031695 | 161.0455486 | 0.44 | ['(S)-Methylmalonate semialdehyde', '2-Methyl-3oxopropanoate', '2-Oxobutanoate', 'Acetoacetate', 'Succinate semialdehyde'] |
| 161.04562 | 67109.1 | -0.030741974 | C6H1005 | C6H1005 | [M-H]- | 162.052825 | 161.0455486 | 0.44 | ['(2R,3S)-2,3-Dimethylmalate', '(R)-2-Ethylmalate', '(R)-3,3-Dimethylmalate', '(S)-2-(Hydroxymethyl)glutarate', '1,5-Anhydro-D-fructose', '2-Dehydro-3-deoxy-Dfuconate', '2-Dehydro-3-deoxy-L-fuconate', '2-Dehydro-3-deoxy-L-rhamnonate', '2-Deoxy-scylloinosose', '2-Hydroxyadipate', '3,6-Anhydrogalactose', '3,6-Anhydroglucose', '3-Ethylmalate', '3-Hydroxy-3methylglutarate', 'D-Fucono-1,4-lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5-lactone', 'L-Rhamnono- |


|  |  |  |  |  |  |  |  |  | 1,4-lactone', 'Lichenin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 145.01434 | 43325.5 | -0.030439508 | C5H6O5 | C5H6O5 | [M-H]- | 146.021525 | 145.0142486 | 0.63 | ['2-Oxoglutarate', '5-Hydroxy-2,4-dioxopentanoate', 'Dehydro-D-arabinono-1,4-lactone', <br> 'Methyloxaloacetate', 'Oxaloacetate 4-methyl ester'] |
| 514.98789 | 75940.4 | -0.03026179 |  | 0 |  |  |  |  | 0 |
| 118.0227 | 102759.2 | -0.03018279 |  | 0 |  |  |  |  | 0 |
| 477.01091 | 199390.7 | -0.030080016 |  | 0 |  |  |  |  | 0 |
| 181.05073 | 70728.7 | -0.029838377 | C7H6O2 | C7H6O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 122.03678 | 181.0506336 | 0.53 | ['3-Hydroxybenzaldehyde', '4-Hydroxybenzaldehyde', 'Benzoate', 'Salicylaldehyde', 'Tropolone'] |
| 181.05073 | 70728.7 | -0.029838377 | C9H10O4 | C9H10O4 | [M-H]- | 182.05791 | 181.0506336 | 0.53 | ['(R)-3-(4-Hydroxyphenyl)lactate', "2',6-Dihydroxy-4methoxyacetophenone", '3,4- <br> Dihydroxyphenylpropanoate', '3-(2,3- <br> Dihydroxyphenyl)propanoate', '3-(4- <br> Hydroxyphenyl)lactate', '3-Methoxy-4- <br> hydroxyphenylglycolaldehyde', 'Homovanillate', 'cis-3- <br> (3-Carboxyethenyl)-3,5-cyclohexadiene-1,2-diol'] |
| 277.2172 | 333304 | -0.029509292 | C18H30O2 | C18H30O2 | [M-H]- | 278.22458 | 277.2173036 | -0.37 | ['(6Z,9Z,12Z)-Octadecatrienoic acid', '(9Z,11E,13E)Octadecatrienoic acid', '(9Z,12Z,15Z)-Octadecatrienoic acid', '5beta-Estrane-3alpha,17beta-diol', 'Crepenynate', 'Punicic acid'] |
| 154.01471 | 48661.1 | -0.029363054 | C6H5NO4 | C6H5NO4 | [M-H]- | 155.021859 | 154.0145826 | 0.83 | ['2,6-Dihydroxynicotinate', '4-Nitrocatechol'] |
| 478.07077 | 119597.8 | -0.029229517 |  | 0 |  |  |  |  | 0 |


| 163.03888 | 25534.2 | -0.029194754 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 327.08714 | 337434.4 | -0.029188092 | C14H18N4O3 | C14H18N4O3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 290.137891 | 327.0864976 | 1.96 | ['Benomyl', 'Trimethoprim'] |
| 327.08714 | 337434.4 | -0.029188092 | C16H12O4 | C16H12O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 268.07356 | 327.0874136 | -0.84 | ['1-[6-Hydroxy-2-(4-hydroxyphenyl)-1-benzofuran-3- <br> yl]ethanone', '6-Hydroxy-2-methoxyflavone', <br> 'Dalbergin', 'Formononetin', 'Isoformononetin', 'Tectochrysin'] |
| 327.08714 | 337434.4 | -0.029188092 | C18H16O6 | C18H16O6 | [M-H]- | 328.09469 | 327.0874136 | -0.84 | ['2-(4-Hydroxyphenyl)-5,6,7-trimethoxy-4H-1-benzopyran-4-one', '6-Hydroxy-2-(4-methoxyphenyl)- <br> 5,7-dimethoxy-4H-1-benzopyran-4-one', "7-Hydroxy- <br> 2',4',5-trimethoxyisoflavone", '9- <br> Demethylmunduserone', 'Betagarin', <br> 'Ophiopogonanone A'] |
| 193.05078 | 194456.2 | -0.029083865 | C10H1004 | C10H1004 | [M-H]- | 194.05791 | 193.0506336 | 0.76 | ['2,4,8-Trihydroxy-1-tetralone', '5- <br> Hydroxyconiferaldehyde', '6-Hydroxymellein', <br> 'Dimethyl phthalate', 'Ferulate', 'Isoferulic acid', <br> 'Kakuol', 'Methyl caffeate', 'Scytalone'] |
| 129.01937 | 657155 | -0.028954418 | C3H2O2 | C3H2O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ] | 70.00548 | 129.0193336 | 0.28 | ['Propynoate'] |
| 129.01937 | 657155 | -0.028954418 | C5H6O4 | C5H6O4 | [M-H]- | 130.02661 | 129.0193336 | 0.28 | ['(E)-Glutaconate', '2,5-Dioxopentanoate', '2- <br> Methylmaleate', '4,5-Dioxopentanoate', <br> 'Acetylpyruvate', 'Itaconate', 'Mesaconate'] |
| 209.06283 | 220540.3 | -0.028917582 | C6H12N4O2 | C6H12N4O2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 172.096026 | 209.0624776 | 1.69 | ['L-Capreomycidine'] |
| 255.05092 | 130108.2 | -0.028903558 | C9H8O5 | C9H8O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 196.037175 | 255.0510286 | -0.43 | ['3-(3,4-Dihydroxyphenyl)pyruvate'] |
| 470.99402 | 139276.9 | -0.028876043 |  | 0 |  |  |  |  | 0 |
| 148.03333 | 40194.2 | -0.028582145 |  | 0 |  |  |  |  | 0 |
| 313.0563 | 28914.6 | -0.02852408 |  | 0 |  |  |  |  | 0 |
| 193.0144 | 24727.2 | -0.028402811 |  | 0 |  |  |  |  | 0 |
| 208.96501 | 25035.8 | -0.028231207 |  | 0 |  |  |  |  | 0 |


| 337.0198 | 206399.5 | -0.028186546 | C10H12N4O5S | C10H12N4O5S | [M+(37Cl)]- | 300.052843 | 337.0192946 | 1.5 | ['Tazobactam'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 337.0198 | 206399.5 | -0.028186546 | C10H12N4O7 | C10H12N4O7 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 300.070601 | 337.0192076 | 1.76 | ['Urate-3-ribonucleoside'] |
| 214.07704 | 13444.7 | -0.028051426 |  | 0 |  |  |  |  | 0 |
| 157.0507 | 47407.2 | -0.02795424 | C7H1004 | C7H1004 | [M-H]- | 158.05791 | 157.0506336 | 0.42 | ['(1S,4S)-4-Hydroxy-3-oxocyclohexane-1-carboxylate', '2-Isopropylmaleate', '5-D-(5/6)-5-C-(Hydroxymethyl)-2,6-dihydroxycyclohex-2-en-1-one', 'Dimethyl citraconate'] |
| 497.18261 | 104167.3 | -0.027923037 |  | 0 |  |  |  |  | 0 |
| 188.99615 | 94910.2 | -0.027731671 | C7H6O4 | C7H6O4 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 154.02661 | 188.9960116 | 0.73 | ['2,3-Dihydroxybenzoate', '2,5-Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 188.99615 | 94910.2 | -0.027731671 | C8H8O3 | C8H8O3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 152.047345 | 188.9959516 | 1.05 | ['(R)-Mandelate', '(S)-Mandelate', "2',4- <br> Dihydroxyacetophenone", '2-(Hydroxymethyl)benzoic acid', '2-Hydroxyphenylacetate', "3',4- <br> Dihydroxyacetophenone", '3,4- <br> Dihydroxyphenylacetaldehyde', '3- <br> Hydroxyphenylacetate', '3-Methoxytropolone', '3- <br> Methylsalicylate', '4-Hydroxy-3-methoxy- <br> benzaldehyde', '4-Hydroxymethylsalicylaldehyde', '4- <br> Hydroxyphenacyl alcohol', '4-Hydroxyphenyl acetate', <br> '4-Hydroxyphenylacetate', '4-Methoxybenzoate', '4- <br> Methylsalicylate', '6-Methylsalicylate', <br> 'Menisdaurilide', 'Methyl salicylate', 'Phenoxyacetate', <br> 'Resorcinol monoacetate'] |
| 151.00376 | 55326.4 | -0.027711898 |  | 0 |  |  |  |  | 0 |
| 192.02322 | 33143.7 | -0.027567359 |  | 0 |  |  |  |  | 0 |
| 213.04063 | 51947.8 | -0.027101305 | C7H6O4 | C7H6O4 | [M+Hac-H]- | 154.02661 | 213.0404636 | 0.78 | ['2,3-Dihydroxybenzoate', '2,5-Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 213.04063 | 51947.8 | -0.027101305 | C9H1006 | C9H1006 | [M-H]- | 214.04774 | 213.0404636 | 0.78 | ['2-Hydroxy-6-oxonona-2,4-diene-1,9-dioate'] |


| 200.02024 | 106458.6 | -0.026854654 | C5H3NO4 | C5H3NO4 | [M+Hac-H]- | 141.006209 | 200.0200626 | 0.89 | ['5-Nitrofurfural'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 206.04604 | 36618 | -0.026829802 | C10H9NO4 | C10H9NO4 | [M-H]- | 207.053159 | 206.0458826 | 0.76 | ['1-Nitro-5,6-dihydroxy-dihydronaphthalene', '2Formaminobenzoylacetate', '3-Amino-4,7-dihydroxy-8methylcoumarin', '4-(2-Aminophenyl)-2,4dioxobutanoate'] |
| 206.04604 | 36618 | -0.026829802 | C8H5NO2 | C8H5NO2 | [M+Hac-H]- | 147.032029 | 206.0458826 | 0.76 | ['Indole-5,6-quinone', 'Isatin'] |
| 215.01744 | 167385.1 | -0.026803166 | C12H8O2S | C12H8O2S | [M-H]- | 216.024502 | 215.0172256 | 1 | ['1,2-Dihydroxydibenzothiophene'] |
| 215.01744 | 167385.1 | -0.026803166 | C6H1007 | C6H1007 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 194.042655 | 215.0173236 | 0.54 | ['2-Dehydro-D-galactonate', '2-Keto-D-gluconic acid', '3-Dehydro-L-gulonate', '5-Dehydro-D-gluconate', 'DFructuronate', 'D-Galacturonate', 'D-Glucuronate', 'DGlucuronic acid', 'D-Mannuronate', 'D-Tagaturonate', 'Galacturonic acid', 'L-Guluronic acid', 'L-Iduronic acid', 'beta-D-Glucopyranuronic acid'] |
| 571.19938 | 53079.4 | -0.026680312 |  | 0 |  |  |  |  | 0 |
| 546.92729 | 41210.8 | -0.026633111 |  | 0 |  |  |  |  | 0 |
| 293.03014 | 176702.6 | -0.026481388 | C11H14NO6 | C11H14NO6 | [M+K-2H]- | 256.082114 | 293.0307206 | -1.98 | ['Nicotinate D-ribonucleoside'] |
| 293.03014 | 176702.6 | -0.026481388 | C13H1008 | C13H1008 | [M-H]- | 294.03757 | 293.0302936 | -0.52 | ['Tricrozarin A'] |
| 253.07166 | 211403.7 | -0.026398465 | C10H1004 | C10H1004 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]-$ | 194.05791 | 253.0717636 | -0.41 | ['2,4,8-Trihydroxy-1-tetralone', '5- <br> Hydroxyconiferaldehyde', '6-Hydroxymellein', <br> 'Dimethyl phthalate', 'Ferulate', 'Isoferulic acid', <br> 'Kakuol', 'Methyl caffeate', 'Scytalone'] |
| 479.00988 | 63125.6 | -0.026325852 |  | 0 |  |  |  |  | 0 |
| 338.97931 | 89595.1 | -0.026238656 |  | 0 |  |  |  |  | 0 |
| 191.03512 | 94746.1 | -0.026221889 | C10H8O4 | C10H8O4 | [M-H]- | 192.04226 | 191.0349836 | 0.71 | ['1,3,6,8-Naphthalenetetrol', '2-Hydroxychromene-2carboxylate', '3,4-Dehydro-6-hydroxymellein', <br> 'Acamelin', 'Anemonin', 'Isoscopoletin', 'Methylenedioxycinnamic acid', 'Naphthazarin', |


|  |  |  |  |  |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |


| 89.02438 | 57476.2 | -0.024994887 | C3H6O3 | C3H6O3 | [M-H]- | 90.031695 | 89.0244186 | -0.43 | ['(R)-Lactate', '(S)-Lactate', '3-Hydroxypropanoate', 'DGlyceraldehyde', 'Glyceraldehyde', 'Glycerone', 'LGlyceraldehyde', 'Lactate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 89.02438 | 57476.2 | -0.024994887 | CH2O | CH2O | [M+Hac-H]- | 30.010565 | 89.0244186 | -0.43 | ['Formaldehyde'] |
| 319.03016 | 58867.5 | -0.024993576 | C15H1ON2O4 | C15H10N2O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ] | 282.064058 | 319.0305096 | -1.1 | ['N-(p-Nitrobenzyl)phthalimide'] |
| 494.00931 | 50165.4 | -0.024876577 |  | 0 |  |  |  |  | 0 |
| 664.98363 | 98735.9 | -0.024656939 |  | 0 |  |  |  |  | 0 |
| 476.99003 | 115577.7 | -0.024618005 |  | 0 |  |  |  |  | 0 |
| 513.15673 | 39052.7 | -0.024602819 | C21H32O12 | C21H32O12 | [M+(37C) )]- | 476.18938 | 513.1558316 | 1.75 | ['Kanokoside A'] |
| 335.00195 | 44001.3 | -0.024371536 |  | 0 |  |  |  |  | 0 |
| 499.14089 | 34451.8 | -0.024270569 |  | 0 |  |  |  |  | 0 |
| 381.01709 | 107147.9 | -0.024114182 |  | 0 |  |  |  |  | 0 |
| 186.04569 | 58457.4 | -0.024106908 |  | 0 |  |  |  |  | 0 |
| 658.99532 | 39855.5 | -0.023732286 |  | 0 |  |  |  |  | 0 |
| 550.99066 | 19262 | -0.023583887 |  | 0 |  |  |  |  | 0 |
| 658.98866 | 66661.7 | -0.023415301 |  | 0 |  |  |  |  | 0 |
| 530.95356 | 37396.2 | -0.023403802 |  | 0 |  |  |  |  | 0 |
| 487.99237 | 38597 | -0.023236294 |  | 0 |  |  |  |  | 0 |
| 403.26996 | 79648 | -0.02323353 |  | 0 |  |  |  |  | 0 |
| 202.07224 | 7061.7 | -0.023170765 | C6H9NO3 | C6H9NO3 | [M+Hac-H]- | 143.058244 | 202.0720976 | 0.7 | ['2-Hydroxymethylclavam'] |
| 202.07224 | 7061.7 | -0.023170765 | C8H13NO5 | C8H13NO5 | [M-H]- | 203.079374 | 202.0720976 | 0.7 | ['N2-Acetyl-L-aminoadipate'] |


| 202.07224 | 7061.7 | -0.023170765 | C9H13N2O2 | C9H13N2O2 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 181.097703 | 202.0723716 | -0.65 | ['Pyridostigmine'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 220.00153 | 67679.4 | -0.023119295 | C7H7NO5 | C7H7NO5 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 185.032424 | 220.0018256 | -1.34 | ['2-Amino-3-carboxymuconate semialdehyde'] |
| 220.00153 | 67679.4 | -0.023119295 | C8H9NO2S | C8H9NO2S | [M+(37C) $]$ - | 183.035401 | 220.0018526 | -1.47 | ['p-Hydroxyphenylacetothiohydroximate'] |
| 220.00153 | 67679.4 | -0.023119295 | C8H9NO4 | C8H9NO4 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}]-$ | 183.053159 | 220.0017656 | -1.07 | ['3,5-Dihydroxy-phenylglycine', '3-Carboxy-4-methoxy-N-methyl-2-pyridone', '3-Hydroxy-4-hydroxymethyl-2-methylpyridine-5-carboxylate', '4-Pyridoxate', '5-Methoxy-3-hydroxyanthranilate'] |

Chapter 4

Breakdown Curve of Ion $191.0 \mathrm{~m} / \mathrm{z}$


Figure 1a A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z}$ 191, citric acid.

Breakdown Curve of Ion $103.0 \mathrm{~m} / \mathrm{z}$


Figure $\mathbf{2 b}$ A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathbf{m} / \mathbf{z} 103$, malonic acid.


Figure 3c A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 193$, ferulic acid.


Figure 4d A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 163, \mathrm{~m}$-coumaric acid.

## Breakdown Curve of Ion $169.0 \mathrm{~m} / \mathrm{z}$



Figure 5 e A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 169$, gallic acid.


Figure 6 f A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 163, \mathrm{p}$-coumaric acid.


Figure 7 g A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z}$ 197, syringic acid.
(h)


Figure 8 h A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 133$, malic acid.


Figure 9i A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z}$ 149, tartaric acid.


Figure 10 j A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 153,2,3$-DHB acid.


Figure 11k A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z}$ 101, ketobutyric acid.


Figure 12l A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 167$, vanillic acid.



Breakdown Curve of Ion $152.9 \mathrm{~m} / \mathrm{z}$
(o)


Figure 150 A graph of relative intensity vs optimum collision energy for the formation of the product ions from $\mathrm{m} / \mathrm{z} 153,2,5 \mathrm{DHB}$ (gentisic) acid.

## Breakdown Curve of Ion $175.0 \mathrm{~m} / \mathrm{z}$

(p)



Figure $\mathbf{2}$ The total ion chromatogram from the Sardinian wine press from the ethyl acetate liquid/liquid extraction.

## wine_press_ethylAcetate_09



Figure 3a The identification of syringic acid from the Sardinian wine press based upon a) four product ions from $m / z 197$ and b) retention time (20.99).
wine_press_ethylAcetate_09
11/16/12 02:40:56


Figure 3b The identification of ferulic acid from the Sardinian wine press based upon a) three product ions from $\mathrm{m} / \mathrm{z} 193$ and $\mathbf{b}$ ) retention time ( 20.32 ).
wine_press_ethylAcetate_09



Figure 3 c The identification of vanillic acid from the Sardinian wine press based upon a) four product ions from $m / z 167$ and b) retention time ( 21.89 )
wine_press_ethylAcetate_09


m/z


Figure 3d The identification of p-coumaric acid from the Sardinian wine press based upon a) four product ions from $m / z 163$ and b) retention time ( 21.40 ).
wine_press_ethylAcetate_09


Figure 3 e The identification of tartaric acid from the Sardinian wine press based upon a) three product ions from $m / z 149$ and b) retention time ( 27.68 ).


Figure 3 f The identification of malic acid from the Sardinian wine press based upon a) three product ions from $m / z 133$ and b) retention time ( 26.87 ).
wine_press_ethylAcetate_09
11/16/12 02:40:56


Figure 3 g The identification of succinic acid from the Sardinian wine press based upon a) two product ions from $m / z 117$ and b) retention time ( 26.14 ).
wine_press_ethylAcetate_09


Figure 3 h The identification of malonic acid from the Sardinian wine press based upon a) two product ions from $m / z 103$ and $b$ ) retention time ( 26.28 ).

Table 1 Sample preparation for the three extraction procedures described in 4.2.1 and 4.2.2.


| 1. TM.003.3/ prepatory layer | 0.974 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2. TM.003.3/ prepatory layer | 0.986 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 3. TM.003.3/ wine press A | 0.981 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 4. extraction blank | NA | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |

Table 2 The results of the targeted analysis for 16 acids on samples of Sardinian wine press, the prepatory layer of the wine press, and the extraction blanks. The results were from the following extraction procedures: (1) an extended KOH preparation (2) C18 SPE flow through collect, water eluate, methanol eluate, and ethyl acetate eluate, (3) ethyl acetate liquid liquid extraction, and (4) re-preparation of ethyl acetate extraction method. SPE spiked sample spiked with syringic and p-coumaric acids.

| chemical name | RT | KOH EXTRACTION BLANK one hour | KOH prep layer, 1 hour | KOH prep layer, 1 hour, rerun | KOH EXTRACTION BLANK 24 hours | KOH prep layer, 24 hours | KOH prep layer, 24 hours, rerun | KOH wine press, 24 hours |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ketobutyric acid | 19.66 | N | N | N | N | N | N | N |
| ferulic acid | 20.55 | N | N | N | N | N | N | N |
| gentisic acid | 20.94 | N | N | N | N | N | N | N |
| syringic acid | 21.11 | N | N | (20.93) | N | 20.93 | 20.93 | N |
| mhydroxycinna mic acid | 21.13 | N | N | N | N | N | N | N |
| p-coumaric acid | 21.43 | N | 21.34 | 21.34 | N | 21.34 | 21.34 | 21.34 |
| 2,3-DHB | 21.52 | N | N | N | N | N | N | N |
| vanillic acid | 21.92 | N | 21.83 | 21.83 | N | 21.83 | 21.84 | N |
| caffeic acid | 24.1 | N | N | N | N | N | N | N |
| ascorbic acid | 25.67 | N | N | N | N | N | N | N |
| succinic acid | 26.09 | N | 26.21 | 26.21 | N | N | N | N |
| malonic acid | 26.23 | N | 26.34 | 26.35 | N | N | N | N |
| malic acid | 26.79 | N | N | N | N | N | N | N |
| tartaric acid | 27.62 | N | N | N | N | N | N | N |


| citric acid | 28.05 | N | N | N | N | N | N | N |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| gallic acid | 28.32 | N | N | N | N | N | N | N |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| chemical name | RT | SPE extraction_blank Water elution | SPE <br> Spiked wine press Water elution | SPE extraction_blank Methanol elution | SPE <br> Spiked wine press Methanol elution |  |  |  |
| ketobutyric acid | 19.66 | N | N | N | N |  |  |  |
| ferulic acid | 20.55 | N | (20.57) | N | N |  |  |  |
| gentisic acid | 20.94 | N | N | N | N |  |  |  |
| syringic acid | 21.11 | N | 21.31 | N | 21.04 |  |  |  |
| mhydroxycinna mic acid | 21.13 | N | N(21.25) | N | N |  |  |  |
| p-coumaric acid | 21.43 | N | 21.65 | N | 21.45 |  |  |  |
| 2,3-DHB | 21.52 | N | N | N | N |  |  |  |
| vanillic acid | 21.92 | N | $\begin{gathered} \hline \mathrm{N} \\ (21.29) \\ \hline \end{gathered}$ | N | N (21.03) |  |  |  |
| caffeic acid | 24.1 | N | N | N | N |  |  |  |
| ascorbic acid | 25.67 | N | N | N | N |  |  |  |
| succinic acid | 26.09 | N | (26.56) | 26.44 | 26.31 |  |  |  |
| malonic acid | 26.23 | N | N | (26.69) | N |  |  |  |
| malic acid | 26.79 | N | N | N | N |  |  |  |
| tartaric acid | 27.62 | N | N | N | N |  |  |  |
| citric acid | 28.05 | N | N | N | N |  |  |  |
| gallic acid | 28.32 | N | N | N | N |  |  |  |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |


| chemical name | RT | SPE wine press Water elution | SPE prep layer Water elution | SPE wine press Methanol elution | SPE prep layer Methanol elution | 161112_blank 90/10 |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ketobutyric acid | 19.66 | N | N | N | N | N |  |  |
| ferulic acid | 20.55 | N | N | 20.32 | 21.10 | N |  |  |
| gentisic acid | 20.94 | N | N | N | N | N |  |  |
| syringic acid | 21.11 | N | N | 20.99 | 20.99 | N |  |  |
| mhydroxycinna mic acid | 21.13 | N | N | N | N | N |  |  |
| p-coumaric acid | 21.43 | N | N | 21.4 | 21.4 | N |  |  |
| 2,3-DHB | 21.52 | N | N | N | N | N |  |  |
| vanillic acid | 21.92 | N | N (19.74) | 21.89 | 21.89 | N |  |  |
| caffeic acid | 24.1 | N | N | N | N | N |  |  |
| ascorbic acid | 25.67 | N | N | N | N | N |  |  |
| succinic acid | 26.09 | 26.51 | 26.26 | 26.26 | 26.2 | 26.27 |  |  |
| malonic acid | 26.23 | N (26.70) | N (26.52) | 26.4 | 26.4 | 26.46, possible carryover |  |  |
| malic acid | 26.79 | N | N | N | 27.12 | N |  |  |
| tartaric acid | 27.62 | N | N | N | N | N |  |  |
| citric acid | 28.05 | N | N | N | N | N |  |  |
| gallic acid | 28.32 | N | N | N | N | N |  |  |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| chemical name | RT | SPE flow through extraction_blank | SPE flow through $\qquad$ Wine press | 161112_blank_90/10 | SPE flow through Spiked_wine_press |  |  |  |
| ketobutyric acid | 19.66 | N | N | N | N |  |  |  |
| ferulic acid | 20.55 | N | N | N | N |  |  |  |
| gentisic acid | 20.94 | N | N | N | N |  |  |  |
| syringic acid | 21.11 | N | N | N | 21.19 |  |  |  |


| mhydroxycinna mic acid | 21.13 | N | N | N | N (21.18) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| p-coumaric acid | 21.43 | N | N | N | 21.49 |  |  |  |
| 2,3-DHB | 21.52 | N | N | N | N |  |  |  |
| vanillic acid | 21.92 | N | N | N | N (21.17) |  |  |  |
| caffeic acid | 24.1 | N | N | N | N |  |  |  |
| ascorbic acid | 25.67 | N | N | N | N |  |  |  |
| succinic acid | 26.09 | 26.27 | 26.27 | 26.27 | 26.26 |  |  |  |
| malonic acid | 26.23 | N (26.34) | N (26.53) | N (26.58) | 26.52 |  |  |  |
| malic acid | 26.79 | N | N | N | N |  |  |  |
| tartaric acid | 27.62 | N | N | N | N |  |  |  |
| citric acid | 28.05 | N | N | N | N |  |  |  |
| gallic acid | 28.32 | N | N | N | N |  |  |  |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| chemical name | RT | $\underset{n k}{\text { 151112_90/10_bla }^{\text {nk }}}$ | Ethyl_acetate_elution_ext_ blank | Ethyl_acetate_elution_pre |  |  |  |  |
| ketobutyric acid | 19.66 | N | N | N |  |  |  |  |
| ferulic acid | 20.55 | N | N | N |  |  |  |  |
| gentisic acid | 20.94 | N | N | N |  |  |  |  |
| syringic acid | 21.11 | N | N | N |  |  |  |  |
| mhydroxycinna mic acid | 21.13 | N | N | $N$ (21.38) |  |  |  |  |
| p-coumaric acid | 21.43 | N | N | 21.4 |  |  |  |  |
| 2,3-DHB | 21.52 | N | N | N |  |  |  |  |
| vanillic acid | 21.92 | N | N | N |  |  |  |  |
| caffeic acid | 24.1 | N | N | N |  |  |  |  |
| ascorbic acid | 25.67 | N | N | N |  |  |  |  |


| succinic acid | 26.09 | 26.26 | 26.33 | 26.27 |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| malonic acid | 26.23 | (26.46) | 26.64 | (26.46) |  |  |  |  |
| malic acid | 26.79 | N | (27.49) | (27.61) |  |  |  |  |
| tartaric acid | 27.62 | N | N | N |  |  |  |  |
| citric acid | 28.05 | N | N | N |  |  |  |  |
| gallic acid | 28.32 | N | N | N |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| chemical name | RT | Ethyl_acetate elution spiked_press | Ethyl_acetate elution_prep | Ethyl_acetate_extraction extraction_blank | Ethyl_acetate extraction wine press | Ethyl_acetate extraction spiked_press | Ethyl_acetate extraction prep layer |  |
| ketobutyric acid | 19.66 | N | N | N | N | N | N |  |
| ferulic acid | 20.55 | N | N | N | 20.65 | N | (20.65) |  |
| gentisic acid | 20.94 | N | N | N | N | N | N |  |
| syringic acid | 21.11 | 20.99 | N | N | 20.99 | (20.99) | 20.99 |  |
| mhydroxycinna mic acid | 21.13 | N (21.39) | N (21.39) | N | N | N | N |  |
| p-coumaric acid | 21.43 | N | N | N | 21.4 | 21.4 | 21.4 |  |
| 2,3-DHB | 21.52 | N | N | N | N | N | N |  |
| vanillic acid | 21.92 | N (20.98) | N | N | 21.89 | (21.89) | 21.89 |  |
| caffeic acid | 24.1 | N | N | N | N | N | N |  |
| ascorbic acid | 25.67 | N | N | N | N | N | N |  |
| succinic acid | 26.09 | 26.26 | 26.27 | 26.26 | 26.26 | 26.27 | 26.14 |  |
| malonic acid | 26.23 | 26.64 | 26.58 | (26.52), possible carryover | (26.52) | (26.52) | 26.28 |  |
| malic acid | 26.79 | N | N | N | N | N | 26.87 |  |
| tartaric acid | 27.62 | N | N | N | N | N | 27.68 |  |
| citric acid | 28.05 | N | N | N | N | N | N |  |
| gallic acid | 28.32 | N | N | N | N | N | N |  |
|  |  |  |  |  |  |  |  |  |


| chemical name | RT | $\begin{gathered} \text { 211112_90/10_bla } \\ \text { nk } \end{gathered}$ | 211112_ethyl_acetate_extr action_prep_sample3 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ketobutyric acid | 19.66 | N | N |  |  |  |  |  |
| ferulic acid | 20.55 | N | 20.51 |  |  |  |  |  |
| gentisic acid | 20.94 | N | N |  |  |  |  |  |
| syringic acid | 21.11 | N | 20.66 |  |  |  |  |  |
| m- <br> hydroxycinna mic acid | 21.13 | N | N |  |  |  |  |  |
| p-coumaric acid | 21.43 | N | 21.19 |  |  |  |  |  |
| 2,3-DHB | 21.52 | N | N |  |  |  |  |  |
| vanillic acid | 21.92 | N | 21.82 |  |  |  |  |  |
| caffeic acid | 24.1 | N | N |  |  |  |  |  |
| ascorbic acid | 25.67 | N | N |  |  |  |  |  |
| succinic acid | 26.09 | N | 26.32 |  |  |  |  |  |
| malonic acid | 26.23 | N | N |  |  |  |  |  |
| malic acid | 26.79 | $\begin{aligned} & \mathrm{N}(27.84, \text { possible } \\ & \text { ghosting effect) } \\ & \hline \end{aligned}$ | (26.92) |  |  |  |  |  |
| tartaric acid | 27.62 | N | N |  |  |  |  |  |
| citric acid | 28.05 | N | N |  |  |  |  |  |
| gallic acid | 28.32 | N | N |  |  |  |  |  |

Chapter 5
Table 1 Sample identification for the Sardinian wine study analyzed 20130221

|  | 20130205 | 20130206 |  |  |  |  | 20130207 |  | 20130208 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample Identification | weight (g) | volume of solution 1 (ml) | volume of CHCl3 (ml) | volume of water II (ml) | volume of polar layer removed (ul) | volume of non-polar removed (ul) | amount of 1M KOH (ul) | $\begin{gathered} 2 \mathrm{~N} \mathrm{HCl} \\ (\mathrm{ml}) \\ \hline \end{gathered}$ | volume of ethyl acetate (ml) | volume of depolymerised sample removed (ul) |
| TM.003.3/ wine press A | 0.981 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ wine press A | 0.982 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ wine press | 0.983 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ wine press | 1.003 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ wine press A | 1.002 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.027.3/ wine press B | 0.989 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.027.3/ wine press | 1.013 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.027.3/ wine press B | 1.003 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.027.3/ wine press B | 0.983 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.027.3/ wine press <br> B | 0.985 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ prepatory layer | 0.998 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ prepatory layer | 1 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ prepatory layer | 1.008 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ prepatory layer | 0.988 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| TM.003.3/ prepatory layer | 0.996 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |


| 1.TM.044.1.3/ neck | 0.985 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| SAR.TM.67/ neck | 0.988 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| Sample Identification | weight (g) | volume of solution 1 (ml) | volume of ChCl3 (ml) | volume of water II (ml) | volume of polar layer removed (ul) | volume of non-polar removed (ul) | amount of 1M KOH (ul) | $\begin{gathered} \text { 2N HCl } \\ \text { (ml) } \\ \hline \end{gathered}$ | volume of ethyl acetate (ml) | volume of depolymerised sample removed (ul) |
| 1.TM.028.1.33/ neck | 0.996 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM..060.1.15/ neck | 1.002 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.042.1.7/ neck | 0.968 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.033.1.14/ side wall | 0.989 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| SAR.TM.4/side wall | 1 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.034.1.20/ side wall | 0.987 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.033.1.15/ side wall | 0.985 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.060.1.45/ side wall | 1.01 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM..060.1.56/ base | 0.986 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.035.1.1/ base | 1.004 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.036.1.2/ base | 0.991 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.028.1.31/ base | 1.004 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| 1.TM.033.1.20/ base | 1.006 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |
| extraction blank | NA | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3.0 | NA |




Figure 1 Elemental results of white powder remaining after extraction (neck, side wall). The common elements in both spectra are aluminum and silicon, presumably leached out of the clay matrix during the alkaline fusion.

Table 2 Sample identification for the Sardinian wine study analyzed 20130607

|  | 20130601 | 20130602 |  |  |  |  | 20130603 |  | 20130604 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample Identification | weight (g) | volume of solution 1 (ml) | volume <br> of CHCl 3 <br> (ml) | volume of water II (ml) | volume of polar layer removed (ul) | volume of non-polar removed (ul) | $\begin{gathered} \text { amount of 1M } \\ \mathrm{KOH}(\mathrm{ul}) \end{gathered}$ | 2N HCl (ml) | volume of ethyl acetate (ml) | volume of depolymerised sample removed (ul) |
| TM.003.3/ wine press A | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ wine press $A$ | 1.02 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ wine press A | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ wine press A | 1.02 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ wine press $A$ | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/ wine press $B$ | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/ wine press $B$ | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/ wine press $B$ | 1.03 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/ wine press $B$ | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/ wine press $B$ | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ prepatory layer | 1.02 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ prepatory layer | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ prepatory layer | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ prepatory layer | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.003.3/ prepatory layer | 1.01 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.044.1.3/ neck | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| SAR.TM.67/ neck | 1.03 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.028.1.33/ neck | 1.01 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM..060.1.15/ neck | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.042.1.7/ neck | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.033.1.14/ side wall | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| SAR.TM.4/side wall | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.034.1.20/ side wall | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |


| 1.TM.033.1.15/ side wall | 1.03 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1.TM.060.1.45/ side wall | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM..060.1.56/ base | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| Sample Identification | weight (mg) | volume of solution 1 (ml) | volume of ChCl 3 (ml) | volume of water II (ml) | volume of polar layer removed (ul) | volume of non-polar removed (ul) | $\begin{aligned} & \text { amount of 1M } \\ & \mathrm{KOH}(\mathrm{ul}) \end{aligned}$ | 2N HCl (ml) | volume of ethyl acetate (ml) | volume of depolymerised sample removed (ul) |
| 1.TM.035.1.1/ base | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.036.1.2/base | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.028.1.31/ base | 0.98 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| 1.TM.033.1.20/ base | 0.99 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| extraction blank | NA | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/SPE spike | 0.97 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/SPE spike | 1 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |
| TM.027.3/SPE spike | 1.03 | 5.6 | 4 | 2 | NA | NA | 600 | 2.5 | 3 | NA |

Table 3 Logged ratios of identified organic acids for laboratory aged sherds and for archaeological samples.

|  | 103:117 | 103:133 | 117:133 | 103:149 | 117:149 | 133:149 | 103:163 | 117:163 | 133:163 | 149:163 | 103:167 | 163:167 | 103:175 | 117:175 | 133:175 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| WHTG2B | -1.32318 | -2.42251 | -2.25186 | -0.78193 | -0.57027 | 0.35329 | -0.92868 | 0.54125 | 0.75291 | 1.67647 | 1.14247 | 2.77580 | 0.80303 | 2.24180 | 2.38177 |
| WHS4A | -1.21750 | -2.43764 | -1.92448 | -0.45863 | -0.40660 | 0.50679 | -0.70698 | 0.75886 | 0.81089 | 1.72429 | 1.02918 | 2.94443 | 0.91372 | 2.24932 | 2.50525 |
| WHS1A | -1.63107 | -2.49965 | -1.74284 | -0.89993 | -0.57133 | 0.32398 | -0.11177 | 0.73114 | 1.05974 | 1.95504 | 1.18234 | 2.82363 | 0.79298 | 2.05093 | 2.13513 |
| WHG4B | -1.21446 | -2.28964 | -2.11150 | -0.47690 | -0.41227 | 0.55404 | -0.89704 | 0.73757 | 0.80219 | 1.76850 | 1.07058 | 2.84368 | 0.78989 | 2.14575 | 2.45693 |
| WHS4B | -2.07219 | -2.85631 | -1.41691 | -0.84783 | -0.80196 | -0.00571 | 0.65527 | 1.22436 | 1.27022 | 2.06648 | 1.37953 | 2.85060 | 0.75173 | 2.16366 | 2.33174 |
| WHG2A | -1.25654 | -2.32411 | -1.98211 | -0.76019 | -0.39783 | 0.45086 | -0.72557 | 0.49635 | 0.85870 | 1.70740 | 1.00803 | 2.77497 | 0.69734 | 2.07560 | 2.39861 |
| WHS3B | -1.25140 | -2.64846 | -2.15343 | -0.69456 | -0.64269 | 0.35294 | -0.90203 | 0.55684 | 0.60871 | 1.60434 | 0.97454 | 3.00141 | 0.99874 | 2.37160 | 2.77904 |
| WHS2A | -1.25866 | -2.55381 | -2.14036 | -0.83669 | -0.58720 | 0.35377 | -0.88170 | 0.42197 | 0.67146 | 1.61242 | 1.01378 | 2.90758 | 0.98883 | 2.30894 | 2.57662 |
| WHG3A | -1.18554 | -2.33391 | -2.12742 | -0.63032 | -0.43523 | 0.42915 | -0.94188 | 0.55522 | 0.75031 | 1.61470 | 1.06533 | 2.76306 | 0.97789 | 2.21369 | 2.30416 |
| WHG1A | -1.34660 | -2.40739 | -2.28874 | -0.86790 | -0.53184 | 0.18789 | -0.94214 | 0.47870 | 0.81476 | 1.53449 | 1.05938 | 2.59528 | 0.83539 | 2.12016 | 2.37747 |
| WHS3A | -1.19076 | -2.26796 | -1.92298 | -0.37063 | -0.48240 | 0.62885 | -0.73223 | 0.82013 | 0.70836 | 1.81961 | 1.13053 | 2.89681 | 0.86389 | 2.20773 | 2.61733 |
| WHTG3B | -1.05903 | -2.62644 | -1.97712 | -0.65185 | -0.37422 | 0.52062 | -0.91809 | 0.40718 | 0.68481 | 1.57966 | 0.89085 | 3.14706 | 1.10501 | 2.45826 | 2.59990 |
| WHTS2B | -1.93945 | -2.40572 | -0.82731 | -0.76425 | -1.05996 | 0.17009 | 1.11214 | 1.17519 | 0.87948 | 2.10954 | 1.00436 | 2.57582 | 1.32899 | 1.47064 | 1.50318 |
| RG3A | -1.47106 | -1.50041 | -2.50770 | -1.48006 | -1.54159 | -0.97274 | -1.03664 | -0.00901 | -0.07053 | 0.49831 | 1.01143 | 0.52767 | -0.63431 | 1.04078 | -0.64000 |
| RS3B | -1.45267 | -1.38201 | -2.03877 | -1.85403 | -1.75648 | -1.20192 | -0.58611 | -0.40137 | -0.30382 | 0.25074 | 0.90832 | 0.18008 | -0.41522 | 0.83766 | -1.07503 |
| RS2B | -1.57420 | -1.69206 | -2.43994 | -2.10307 | -1.87293 | -1.31789 | -0.86574 | -0.52887 | -0.29873 | 0.25631 | 0.81086 | 0.37417 | -0.57817 | 0.92872 | -0.84941 |
| RG4A | -1.28959 | -1.46066 | -2.25733 | -1.57074 | -1.41886 | -0.86802 | -0.96774 | -0.28114 | -0.12927 | 0.42157 | 1.07172 | 0.59264 | -0.26615 | 1.24280 | -0.54547 |
| RG1A | -1.84215 | -1.64200 | -2.06484 | -2.02399 | -1.90313 | -1.32289 | -0.22269 | -0.18183 | -0.06097 | 0.51926 | 1.05150 | 0.31911 | -0.57759 | 0.85134 | -0.84976 |
| RS1A | -1.70278 | -1.54046 | -1.79581 | -1.50232 | -1.62963 | -1.06385 | -0.09303 | 0.20045 | 0.07315 | 0.63893 | 1.30077 | 0.47661 | -0.23818 | 1.13845 | -0.83577 |
| RS4A | -1.43297 | -1.47870 | -1.88438 | -1.52619 | -1.51338 | -0.88655 | -0.45141 | -0.09322 | -0.08042 | 0.54642 | 1.06407 | 0.59215 | -0.17098 | 1.10980 | -0.75453 |
| RS2A | -1.39828 | -1.53215 | -2.48595 | -1.39064 | -1.34250 | -0.83468 | -1.08767 | 0.00764 | 0.05578 | 0.56361 | 1.23974 | 0.69747 | -0.47507 | 1.37360 | -0.30989 |
| RS4B | -1.54929 | -1.58637 | -2.16048 | -1.69141 | -1.63200 | -1.06702 | -0.61119 | -0.14211 | -0.08271 | 0.48228 | 0.92434 | 0.51936 | -0.40120 | 0.96142 | -0.77586 |
| RS3A | -1.45338 | -1.48450 | -2.05726 | -1.66818 | -1.65280 | -1.05218 | -0.60388 | -0.21481 | -0.19943 | 0.40119 | 1.00450 | 0.43232 | -0.30367 | 1.03562 | -0.86101 |
| white 2, 1:2 | -0.93645 | -1.58864 | -0.65219 | -1.53749 | -0.60104 | 0.05114 | -0.41395 | 0.52249 | 1.17468 | 1.12354 | 1.92366 | 2.33761 | -0.69503 | 0.24142 | 0.89361 |
| white 3, 1:2 | -1.26799 | -1.97835 | -0.71036 | -1.81857 | -0.55058 | 0.15978 | -0.79386 | 0.47413 | 1.18449 | 1.02472 | 1.64109 | 2.43495 | -0.94435 | 0.32364 | 1.03400 |
| red 1, 1:10 | -0.44752 | -0.63316 | -0.18565 | -0.46627 | -0.01875 | 0.16689 | -0.77047 | -0.32296 | -0.13731 | -0.30420 | 0.91984 | 1.69032 | -0.13682 | 0.31069 | 0.49634 |
| red 2, 1:10 | -0.45113 | -0.58501 | -0.13388 | -0.44493 | 0.00620 | 0.14009 | -0.74217 | -0.29104 | -0.15716 | -0.29725 | 0.89049 | 1.63266 | -0.20173 | 0.24940 | 0.38328 |
| red 3, 1:10 | -0.38493 | -0.67971 | -0.29478 | -0.64963 | -0.26470 | 0.03009 | -0.82972 | -0.44479 | -0.15001 | -0.18009 | 0.78170 | 1.61142 | -0.22874 | 0.15619 | 0.45097 |
| WP1_TM. 003 | 0.20524 | 0.34535 | 0.14011 | 2.41034 | 2.20510 | 2.06499 | 2.10367 | 1.89842 | 1.75832 | -0.30667 | 1.96526 | -0.13840 | 1.11066 | 0.90542 | 0.76531 |
| WP2_TM. 003 | 0.21128 | 0.58272 | 0.37144 | 2.64771 | 2.43643 | 2.06499 | 2.19994 | 1.98866 | 1.61722 | -0.44777 | 2.08413 | -0.11582 | 1.34804 | 1.13675 | 0.76531 |
| WP3_TM. 003 | 0.53635 | -0.09829 | -0.63464 | 1.96671 | 1.43036 | 2.06499 | 1.96099 | 1.42464 | 2.05927 | -0.00572 | 2.40490 | 0.44391 | 0.66703 | 0.13068 | 0.76531 |
| WP4_TM. 003 | -0.02324 | -0.35437 | -0.33114 | 1.53371 | 1.55695 | 1.88808 | 1.64265 | 1.66589 | 1.99702 | 0.10894 | 1.74291 | 0.10026 | -0.05554 | -0.03231 | 0.29883 |
| WP5_TM. 003 | 0.27437 | -0.27182 | -0.54619 | 2.10147 | 1.82710 | 2.37328 | 1.67692 | 1.40255 | 1.94874 | -0.42455 | 1.87813 | 0.20121 | 0.60894 | 0.33457 | 0.88076 |
| WP6_TM. 027 | 0.00449 | 0.58947 | 0.58498 | 2.65446 | 2.64997 | 2.06499 | 1.53012 | 1.52563 | 0.94065 | -1.12434 | 1.68002 | 0.14990 | 1.35478 | 1.35029 | 0.76531 |
| WP7_TM. 027 | 0.83737 | -0.09829 | -0.93566 | 1.96671 | 1.12933 | 2.06499 | 1.96099 | 1.12362 | 2.05927 | -0.00572 | 1.76507 | -0.19592 | 0.66703 | -0.17035 | 0.76531 |
| WP8_TM. 027 | -0.10275 | 0.25600 | 0.35876 | 2.62203 | 2.72478 | 2.36602 | 1.56575 | 1.66850 | 1.30974 | -1.05628 | 1.58388 | 0.01813 | 1.32235 | 1.42510 | 1.06634 |
| WP9_TM. 027 | 0.06190 | -0.09428 | -0.15618 | 2.56353 | 2.50163 | 2.65781 | 1.60147 | 1.53957 | 1.69575 | -0.96206 | 1.62760 | 0.02613 | 0.96282 | 0.90092 | 1.05710 |


| WP10_TM. 027 | 0.07212 | 0.50670 | 0.43458 | 2.57169 | 2.49957 | 2.06499 | 1.96563 | 1.89351 | 1.45893 | -0.60606 | 2.18393 | 0.21831 | 1.27201 | 1.19990 | 0.76531 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| prep11 | -0.22809 | 0.16860 | 0.39669 | 2.57461 | 2.80270 | 2.40602 | 1.83901 | 2.06710 | 1.67042 | -0.73560 | 1.56695 | -0.27206 | 1.27494 | 1.50303 | 1.10634 |
| prep12 | 0.25771 | 0.46674 | 0.20904 | 2.53174 | 2.27403 | 2.06499 | 2.52602 | 2.26831 | 2.05927 | -0.00572 | 2.66890 | 0.14288 | 1.23206 | 0.97435 | 0.76531 |
| prep13 | -0.05231 | -0.09829 | -0.04598 | 1.96671 | 2.01902 | 2.06499 | 1.96099 | 2.01330 | 2.05927 | -0.00572 | 2.40490 | 0.44391 | 0.66703 | 0.71934 | 0.76531 |
| prep14 | -0.27526 | 0.55435 | 0.82961 | 2.61935 | 2.89461 | 2.06499 | 1.95793 | 2.23319 | 1.40358 | -0.66141 | 1.86210 | -0.09583 | 1.31967 | 1.59493 | 0.76531 |
| prep15 | -0.17240 | 0.53753 | 0.70994 | 2.60253 | 2.77493 | 2.06499 | 1.59415 | 1.76655 | 1.05662 | -1.00837 | 1.65225 | 0.05810 | 1.30285 | 1.47525 | 0.76531 |


|  | 149:175 | 163:175 | 103:193 | 133:193 | 149:193 | 163:193 | 175:193 | 103:197 | 117:197 | 133:197 | 149:197 | 167:197 | 175:197 | 193:197 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| WHTG2B | 1.46993 | 2.60515 | 2.07115 | 0.21166 | 1.13522 | 0.60122 | 0.92356 | -1.04921 | 0.38956 | 0.52953 | -1.97277 | 1.43877 | 1.57874 | 0.13997 |
| WHS4A | 1.46585 | 2.43127 | 1.73617 | 0.05203 | 0.96543 | 0.27032 | 0.91340 | -1.11731 | 0.21829 | 0.47422 | -2.03071 | 1.33560 | 1.59153 | 0.25593 |
| WHS1A | 0.84291 | 2.06682 | 1.29411 | 0.32860 | 1.22390 | 0.45120 | 0.89531 | -1.13534 | 0.12260 | 0.20680 | -2.03064 | 1.25794 | 1.34214 | 0.08420 |
| WHG4B | 1.63461 | 2.66555 | 1.96762 | 0.06463 | 1.03094 | 0.33301 | 0.96631 | -1.08747 | 0.26839 | 0.57957 | -2.05378 | 1.35586 | 1.66704 | 0.31118 |
| WHS4B | 0.56909 | 1.41120 | 0.72426 | 0.04586 | 0.84212 | 0.15517 | 0.79625 | -1.30262 | 0.10931 | 0.27739 | -2.09888 | 1.41194 | 1.58001 | 0.16808 |
| WHG2A | 1.22192 | 2.43297 | 1.73360 | 0.36236 | 1.21105 | 0.51168 | 0.84870 | -1.22893 | 0.14932 | 0.47234 | -2.07763 | 1.37825 | 1.70127 | 0.32302 |
| WHS3B | 1.45887 | 2.50638 | 1.87657 | 0.05187 | 1.04750 | 0.41770 | 0.99563 | -1.00703 | 0.36583 | 0.77327 | -2.00266 | 1.37286 | 1.78030 | 0.40744 |
| WHS2A | 1.30367 | 2.49413 | 1.89549 | 0.24949 | 1.19046 | 0.59182 | 0.94097 | -0.97779 | 0.34233 | 0.61001 | -1.91875 | 1.32011 | 1.58779 | 0.26768 |
| WHG3A | 1.49710 | 2.55657 | 2.00720 | 0.19509 | 1.05947 | 0.51010 | 0.86438 | -0.92079 | 0.31501 | 0.40548 | -1.78517 | 1.23580 | 1.32627 | 0.09047 |
| WHG1A | 1.42083 | 2.47663 | 2.00151 | 0.33606 | 1.05579 | 0.58068 | 0.71973 | -1.04016 | 0.24462 | 0.50192 | -1.75989 | 1.28478 | 1.54208 | 0.25730 |
| WHS3A | 1.55235 | 2.55184 | 1.86276 | -0.11177 | 0.99948 | 0.31040 | 1.11125 | -0.92167 | 0.42217 | 0.83177 | -2.03292 | 1.34384 | 1.75343 | 0.40959 |
| WHTG3B | 1.32527 | 2.49774 | 1.80894 | 0.27763 | 1.17247 | 0.48367 | 0.89484 | -1.14720 | 0.20604 | 0.34768 | -2.04205 | 1.35325 | 1.49488 | 0.14164 |
| WHTS2B | 0.06305 | 0.99740 | -0.10778 | -0.29571 | 0.93434 | -0.17083 | 1.23005 | -0.01678 | 0.12488 | 0.15742 | -1.24683 | 0.14166 | 0.17419 | 0.03254 |
| RG3A | 1.02763 | 1.53496 | 2.04807 | -0.06153 | 0.50732 | 1.02044 | 0.56885 | -0.59313 | 1.08196 | -0.59882 | -1.16198 | 1.67509 | -0.00569 | -1.68078 |
| RS3B | 0.18474 | 0.83685 | 1.49443 | 0.09755 | 0.65211 | 1.30969 | 0.55456 | -0.04075 | 1.21214 | -0.70055 | -0.59531 | 1.25289 | -0.65980 | -1.91269 |
| RS2B | 0.33687 | 1.12205 | 1.67661 | 0.23014 | 0.78518 | 1.33974 | 0.55504 | -0.39729 | 1.10960 | -0.66853 | -0.95234 | 1.50689 | -0.27124 | -1.77813 |
| RG4A | 0.68660 | 1.38931 | 2.03947 | 0.15188 | 0.70272 | 1.35287 | 0.55084 | -0.30796 | 1.20099 | -0.58727 | -0.85880 | 1.50895 | -0.27932 | -1.78826 |
| RG1A | 0.04085 | 0.74195 | 1.27419 | 0.12086 | 0.70110 | 1.23333 | 0.58023 | -0.31647 | 1.11247 | -0.58863 | -0.89670 | 1.42893 | -0.27216 | -1.70110 |
| RS1A | 0.29348 | 0.73196 | 1.39381 | -0.12730 | 0.43848 | 1.10032 | 0.56578 | -0.14901 | 1.22763 | -0.74660 | -0.71479 | 1.37663 | -0.59759 | -1.97422 |
| RS4A | 0.35818 | 0.99783 | 1.51547 | 0.01281 | 0.63965 | 1.15729 | 0.62684 | -0.13630 | 1.14448 | -0.71985 | -0.76314 | 1.28078 | -0.58355 | -1.86433 |
| RS2A | 1.09531 | 1.65128 | 2.32741 | 0.04813 | 0.55596 | 1.23209 | 0.50783 | -0.66472 | 1.18396 | -0.49953 | -1.17254 | 1.84867 | 0.16518 | -1.68349 |
| RS4B | 0.46907 | 1.09346 | 1.53552 | 0.05940 | 0.62439 | 1.06645 | 0.56499 | -0.35557 | 1.00705 | -0.73022 | -0.92056 | 1.36262 | -0.37465 | -1.73727 |
| RS3A | 0.38907 | 1.00507 | 1.60838 | 0.01538 | 0.61600 | 1.21931 | 0.60062 | -0.13537 | 1.20393 | -0.69270 | -0.73599 | 1.33930 | -0.55733 | -1.89663 |
| white 2, 1:2 | 0.84246 | -0.28108 | 0.29953 | 1.88817 | 1.83702 | 0.71348 | 0.99456 | 1.55121 | 2.48765 | 3.13984 | 3.08870 | -0.37245 | 2.24624 | 1.25167 |
| white 3, 1:2 | 0.87422 | -0.15049 | -0.14071 | 1.83764 | 1.67787 | 0.65315 | 0.80365 | 1.21655 | 2.48454 | 3.19490 | 3.03513 | -0.42454 | 2.16090 | 1.35726 |
| red 1, 1:10 | 0.32945 | 0.63365 | 1.02837 | 1.66154 | 1.49465 | 1.79885 | 1.16520 | -0.60076 | -0.15324 | 0.03240 | -0.13449 | -1.52061 | -0.46394 | -1.62914 |
| red 2, 1:10 | 0.24320 | 0.54044 | 1.05886 | 1.64387 | 1.50378 | 1.80103 | 1.26059 | -0.53419 | -0.08306 | 0.05082 | -0.08927 | -1.42468 | -0.33246 | -1.59305 |
| red 3, 1:10 | 0.42088 | 0.60098 | 0.96423 | 1.64395 | 1.61386 | 1.79396 | 1.19298 | -0.63851 | -0.25358 | 0.04121 | 0.01112 | -1.42021 | -0.40976 | -1.60274 |
| WP1_TM. 003 | -1.29968 | -0.99300 | 1.86533 | 1.51998 | -0.54502 | -0.23834 | 0.75466 | 1.65662 | 1.45138 | 1.31127 | -0.75372 | -0.44704 | 0.54596 | -0.20870 |
| WP2_TM. 003 | -1.29968 | -0.85191 | 1.98585 | 1.40313 | -0.66187 | -0.21410 | 0.63781 | 1.78621 | 1.57493 | 1.20349 | -0.86150 | -0.41373 | 0.43818 | -0.19964 |
| WP3_TM. 003 | -1.29968 | -1.29396 | 2.36694 | 2.46523 | 0.40023 | 0.40595 | 1.69991 | 2.82590 | 2.28955 | 2.92419 | 0.85920 | 0.86491 | 2.15887 | 0.45896 |
| WP4_TM. 003 | -1.58925 | -1.69819 | 1.67181 | 2.02618 | 0.13810 | 0.02916 | 1.72735 | 1.78461 | 1.80784 | 2.13898 | 0.25090 | 0.14196 | 1.84015 | 0.11280 |


|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| WP5_TM.003 | -1.49252 | -1.06798 | 1.57334 | 1.84516 | -0.52812 | -0.10358 | 0.96440 | 1.71560 | 1.44123 | 1.98742 | -0.38586 | 0.03868 | 1.10666 | 0.14226 |
| WP6_TM.027 | -1.29968 | -0.17534 | 2.27958 | 1.69011 | -0.37488 | 0.74946 | 0.92480 | 1.97799 | 1.97350 | 1.38852 | -0.67647 | 0.44787 | 0.62321 | -0.30159 |
| WP7_TM.027 | -1.29968 | -1.29396 | 2.06610 | 2.16438 | 0.09939 | 0.10511 | 1.39907 | 2.15560 | 1.31823 | 2.25389 | 0.18890 | 0.19461 | 1.48857 | 0.08951 |
| WP8_TM.027 | -1.29968 | -0.24340 | 2.27484 | 2.01884 | -0.34719 | 0.70909 | 0.95249 | 1.90048 | 2.00324 | 1.64448 | -0.72154 | 0.33474 | 0.57814 | -0.37436 |
| WP9_TM.027 | -1.60071 | -0.63865 | 2.06304 | 2.15732 | -0.50049 | 0.46157 | 1.10022 | 1.79533 | 1.73343 | 1.88961 | -0.76820 | 0.19386 | 0.83251 | -0.26771 |
| WP10_TM.027 | -1.29968 | -0.69361 | 2.26052 | 1.75383 | -0.31117 | 0.29490 | 0.98851 | 2.29198 | 2.21987 | 1.78528 | -0.27971 | 0.32635 | 1.01997 | 0.03146 |
| prep11 | -1.29968 | -0.56408 | 2.10483 | 1.93624 | -0.46978 | 0.26582 | 0.82990 | 1.64090 | 1.86899 | 1.47230 | -0.93371 | -0.19811 | 0.36597 | -0.46393 |
| prep12 | -1.29968 | -1.29396 | 2.93197 | 2.46523 | 0.40023 | 0.40595 | 1.69991 | 3.29896 | 3.04125 | 2.83221 | 0.76722 | 0.77294 | 2.06690 | 0.36699 |
| prep13 | -1.29968 | -1.29396 | 2.36694 | 2.46523 | 0.40023 | 0.40595 | 1.69991 | 3.12587 | 3.17818 | 3.22415 | 1.15916 | 1.16488 | 2.45884 | 0.75893 |
| prep14 | -1.29968 | -0.63826 | 2.07328 | 1.51893 | -0.54607 | 0.11535 | 0.75361 | 1.86610 | 2.14136 | 1.31174 | -0.75325 | -0.09183 | 0.54643 | -0.20718 |
| prep15 | -1.29968 | -0.29130 | 2.21260 | 1.67506 | -0.38993 | 0.61845 | 0.90975 | 1.76551 | 1.93792 | 1.22798 | -0.83701 | 0.17136 | 0.46267 | -0.44709 |

Chapter 6



Figure 1 Proposed fragmentation scheme of lithiated TAGs, Ms ${ }^{1}$ to $\mathrm{MS}^{2}$, reproduced from Hsu and Turk, 2010.



Figure 2 Proposed fragmentation scheme of lithiated TAGs, $\mathrm{MS}^{2}$ to $\mathrm{MS}^{3}$, reproduced from Hsu and Turk, 2010.

## Mussels with lentils

1. one pint of mussels
2. 350 g lentils
3. spices (peppercorn, cumin seeds, coriander seeds, dried mint, rue, pennyroyal)
4. 1 onion
5. 15 ml vinegar
6. 15 ml honey
7. 15 ml defructum
8. 5 ml anchovy essence and olive oil

## Lamb stew

1. 900 g lamb
2. 1 onion
3. 1 teaspoon coriander seeds
4. pinch of pepper, lovage, and cumin
5. 5 ml anchovy essence
6. 15 ml olive oil
7. 15 ml wine
8. 1 tablespoon cornflour

## Truffles

1. 12 large truffles
2. salt
3. 15 ml olive oil
4. 15 ml wine
5. 15 ml caroenum
6. 15 ml honey
7. pinch of pepper
8. 2 teaspoons of cornflour

Figure 3 Three recipes reproduced from (Renfrew, 1985: 35,40,41

Table 1 Results of the TAG analysis of laboratory aged sherd doped with goats' cheese aged three months



Table 3 Results of the TAG analysis of sherd doped with sheep cheese aged three months
sheep 3A

| acyl carbon number | types of TAGs | fatty acid constituents |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C34 | 8 | 4.10 .20 | 4.15.15 | 4.12.18 | 4.14 .16 | 14.10.10 | 14.6.14 | 16.8.10 | 16.4.14 |  |  |  |
| C34:1 | 1 | 4.12.18:1 |  |  |  |  |  |  |  |  |  |  |
| C36 | 7 | 4.12.20 | 4.14 .18 | 4.15 .17 | 4.16 .16 | 6.16 .14 | 16.8.12 | 16.10.10 |  |  |  |  |
| C36:1 | 1 | 4.14.18:1 |  |  |  |  |  |  |  |  |  |  |
| C38 | 9 | 4.16 .18 | 4.15.19 | 4.14 .20 | 6.16 .16 | 6.15 .17 | 6.12 .20 | 6.18.14 | 8.14.16 | 16.12.10 |  |  |
| C38:1 | 4 | 4.14.20:1 | 4.18.16:1 | 4.18:1.16 | 16.6.16:1 |  |  |  |  |  |  |  |
| C40 | 10 | 18.4.18 | 18.6.16 | 18.8.14 | 10.10.20 | 10.12.18 | 10.14.16 | 16.4.20 | 16.6.18 | 16.8.16 | 16.5.19 |  |
| C40:1 | 11 | 6.16.18:1 | 6.14.20:1 | 6.18.16:1 | 6.15.19:1 | 4.17:1.19 | 4.18:1.18 | 4.17.19:1 | 4.16.20:1 | 4.16:1.20 | 18:1.8.14 | 18:1.10.12 |
| C42 |  |  |  |  |  |  |  |  |  |  |  |  |
| C42:1 |  |  |  |  |  |  |  |  |  |  |  |  |
| C44 |  |  |  |  |  |  |  |  |  |  |  |  |
| C44:1 | 2 | 10.18:1.16 | 10.16.18:1 |  |  |  |  |  |  |  |  |  |



Table 4 The top 200 loadings based upon the PCA scores plot from Chapter 6, Figure 6.4. The table is sorted on PC1 values staring from lowest to highest. The three yellow masses are phenolic acids common to the wine biomarker list.

| M/Z | INTENSITY | PC1 | Empirical formula (parent) | Empirical formula (peak) | Ion form | Theoretical mass (neutral) (Da) | Theoretical mass (Da) | Mass error (ppm) | KEGG_Putative identificaiton |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 153.01922 | 5394464 | -0.0567179 | C7H6O4 | C7H6O4 | [M-H]- | 154.02661 | 153.0193336 | -0.74 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 187.04179 | 6531863.5 | -0.0502019 |  | 0 |  |  |  |  | 0 |
| 179.03482 | 14153511 | -0.0490905 | C9H8O4 | C9H8O4 | [M-H]- | 180.04226 | 179.0349836 | -0.91 | ['2-Hydroxy-3-(4-hydroxyphenyl)propenoate', '3-(4-Hydroxyphenyl)pyruvate', 'Aspirin', 'Caffeate', 'trans-2,3-Dihydroxycinnamate'] |
| 277.21703 | 15907124 | -0.0485573 | C18H30O2 | C18H30O2 | [M-H]- | 278.22458 | 277.2173036 | -0.99 | ['(6Z,9Z,12Z)-Octadecatrienoic acid', '(9Z,11E,13E)-Octadecatrienoic acid', '(9Z,12Z,15Z)-Octadecatrienoic acid', '5beta-Estrane-3alpha,17beta-diol', 'Crepenynate', 'Punicic acid'] |
| 197.04548 | 6917421.6 | -0.0477345 | C7H14NO3 | C7H14NO3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 160.097369 | 197.0459756 | -2.52 | ['3-Dehydrocarnitine'] |
| 197.04548 | 6917421.6 | -0.0477345 | C7H6O3 | C7H6O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 138.031695 | 197.0455486 | -0.35 | ['2-Hydroxy-5-methylquinone', '3,4- <br> Dihydroxybenzaldehyde', '3-Hydroxybenzoate', <br> '4-Hydroxybenzoate', 'Gentisate aldehyde', <br> 'Salicylate', 'Sesamol'] |
| 197.04548 | 6917421.6 | -0.0477345 | C9H1005 | C9H1005 | [M-H]- | 198.052825 | 197.0455486 | -0.35 | ['3-(3,4-Dihydroxyphenyl)lactate', '3-Methoxy-4-hydroxymandelate', 'Syringic acid'] |
| 129.03634 | 1362343.4 | -0.047612 |  | 0 |  |  |  |  | 0 |
| 161.06253 | 1633131.8 | -0.0469184 |  | 0 |  |  |  |  | 0 |
| 154.02263 | 351008.9 | -0.0464517 |  | 0 |  |  |  |  | 0 |
| 137.02428 | 6583738.3 | -0.0460582 | C7H6O3 | C7H6O3 | [M-H]- | 138.031695 | 137.0244186 | -1.01 | ['2-Hydroxy-5-methylquinone', '3,4- <br> Dihydroxybenzaldehyde', '3-Hydroxybenzoate', <br> '4-Hydroxybenzoate', 'Gentisate aldehyde', <br> 'Salicylate', 'Sesamol'] |
| 209.0627 | 814900.4 | -0.0453093 | C6H12N4O2 | C6H12N4O2 | [M+(37Cl)]- | 172.096026 | 209.0624776 | 1.06 | ['L-Capreomycidine'] |


| 163.04002 | 489449.1 | -0.0451876 | C9H8O3 | C9H8O3 | [M-H]- | 164.047345 | 163.0400686 | -0.3 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2- <br> Hydroxycinnamate', 'cis-p-Coumarate', 'trans-2Hydroxycinnamate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 221.04557 | 724963.2 | -0.045185 | C11H1005 | C11H1005 | [M-H]- | 222.052825 | 221.0455486 | 0.1 | ['2-Hydroxy-4-hydroxymethylbenzalpyruvate', '2-Hydroxy-7-hydroxymethylchromene-2carboxylate', '2-Succinylbenzoate', '6-(2-Methoxyvinyl)benzo[1,3]dioxole-5-carboxylic acid', 'Fraxidin', 'Isofraxidin', 'Leptodactylone', 'Sphagnum acid'] |
| 221.04557 | 724963.2 | -0.045185 | C8H13N2O2P | C8H13N2O2P | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 200.071466 | 221.0461346 | -2.55 | ['Diamidafos'] |
| 221.04557 | 724963.2 | -0.045185 | C9H6O3 | C9H6O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 162.031695 | 221.0455486 | 0.1 | ['Umbelliferone'] |
| 203.035 | 608807.9 | -0.0443507 | C11H8O4 | C11H8O4 | [M-H]- | 204.04226 | 203.0349836 | 0.08 | ['1,2-Dihydroxy-8-carboxynaphthalene', '1,4-Dihydroxy-2-naphthoate', '1,4-Dihydroxy-6naphthoate', 'Droserone', 'Spirodilactone'] |
| 277.2152 | 488837.7 | -0.0432642 | C16H32O2 | C16H32O2 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 256.24023 | 277.2148986 | 1.09 | ['Hexadecanoic acid'] |
| 195.0299 | 332449.2 | -0.0429359 | C9H8O5 | C9H8O5 | [M-H]- | 196.037175 | 195.0298986 | 0.01 | ['3-(3,4-Dihydroxyphenyl)pyruvate'] |
| 141.01925 | 330838.9 | -0.0429189 | C6H6O4 | C6H6O4 | [M-H]- | 142.02661 | 141.0193336 | -0.59 | ['(S)-5-Oxo-2,5-dihydrofuran-2-acetate', '1,2,3,5-Tetrahydroxybenzene', '2,5-Dihydro-5-oxofuran-2-acetate', '2-Hydroxymuconate semialdehyde', '2-Oxo-2,3-dihydrofuran-5acetate', 'Kojic acid', 'cis,cis-4-Hydroxymuconic semialdehyde', 'cis,cis-Muconate', 'cis,transHexadienedioate'] |
| 377.08762 | 2423971.5 | -0.0424825 | C16H14O7 | C16H14O7 | [M+Hac-H]- | 318.073955 | 377.0878086 | -0.5 | ['Demethylsulochrin', 'Orsellinate depside'] |
| 259.06281 | 197681.4 | -0.0421662 |  | 0 |  |  |  |  | 0 |
| 329.06644 | 1159814 | -0.0421168 | C15H1005 | C15H10O5 | [M+Hac-H]- | 270.052825 | 329.0666786 | -0.73 | ['2-Hydroxydaidzein', "3',4',7- <br> Trihydroxyisoflavone", '3,6,4- <br> Trihydroxyflavone', "4',6,7- <br> Trihydroxyisoflavone", '5-Deoxykaempferol', 'Aloe-emodin', 'Apigenin', 'Baicalein', 'Emodin', 'Galangin', 'Genistein', 'Islandicin', 'Lucidin', 'Morindone', 'Norobtusifolin', 'Norwogonin', 'Purpurin 1-methyl ether', 'Sulphuretin'] |
| 329.06644 | 1159814 | -0.0421168 | C17H14O7 | C17H14O7 | [M-H]- | 330.073955 | 329.0666786 | -0.73 | ['(+)-Bisdechlorogeodin', '(-)- <br> Bisdechlorogeodin', "3',4',5-Trihydroxy-3,7dimethoxyflavone", 'Aflatoxin G2', 'Aurantioobtusin', 'Cirsiliol', 'Hildecarpin', 'Tricin'] |


| 207.02993 | 190672.7 | -0.0419134 | C10H8O5 | C10H8O5 | [M-H]- | 208.037175 | 207.0298986 | 0.15 | ['Fraxetin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 139.00362 | 289490.2 | -0.0418592 | C6H4O4 | C6H4O4 | [M-H]- | 140.01096 | 139.0036836 | -0.46 | ['cis-4-Carboxymethylenebut-2-en-4-olide', 'trans-4-Carboxymethylenebut-2-en-4-olide'] |
| 138.0277 | 478185.6 | -0.0415011 |  | 0 |  |  |  |  | 0 |
| 331.08208 | 1462635.1 | -0.0413294 | C15H12O5 | C15H12O5 | [M+Hac-H]- | 272.068475 | 331.0823286 | -0.75 | ['(6aS,11aS)-3,6a,9-Trihydroxypterocarpan', <br> '2,7,4-Trihydroxyisoflavanone', '2- <br> Hydroxydihydrodaidzein', '6,7,4- <br> Trihydroxyflavanone', 'Butein', 'Butin', 'Dihydrogenistein', 'Garbanzol', 'Licodione', 'Naringenin', 'Naringenin chalcone', <br> 'Pinobanksin', 'Rubrofusarin', 'Toralactone', 'pCoumaroyltriacetic acid lactone'] |
| 331.08208 | 1462635.1 | -0.0413294 | C17H16O7 | C17H16O7 | [M-H]- | 332.089605 | 331.0823286 | -0.75 | ['Sulochrin'] |
| 128.03998 | 439849 | -0.0411237 |  | 0 |  |  |  |  | 0 |
| 279.2327 | 9181564.9 | -0.0411056 | C18H32O2 | C18H32O2 | [M-H]- | 280.24023 | 279.2329536 | -0.91 | ['9-cis,11-trans-Octadecadienoate', 'Chaulmoogric acid', 'Linoleate', 'Malvalic acid', 'Stearolic acid'] |
| 191.05607 | 659197.6 | -0.0408467 | C5H8O4 | C5H8O4 | [M+Hac-H]- | 132.04226 | 191.0561136 | -0.23 | ['(4S)-4,5-Dihydroxypentan-2,3-dione', '(S)-2Acetolactate', '2-(Hydroxymethyl)-4oxobutanoate', '2-Acetolactate', '3-Hydroxy-3-methyl-2-oxobutanoic acid', '4-Hydroxy-2oxopentanoate', 'Deoxyribonolactone', 'Glutarate'] |
| 191.05607 | 659197.6 | -0.0408467 | C7H12O6 | C7H12O6 | [M-H]- | 192.06339 | 191.0561136 | -0.23 | ['2D-5-O-Methyl-2,3,5/4,6- <br> pentahydroxycyclohexanone', 'Quinate', 'Valiolone'] |
| 76.5096 | 121821.6 | -0.0403706 |  | 0 |  |  |  |  | 0 |
| 278.22044 | 3167042.7 | -0.0398457 |  | 0 |  |  |  |  | 0 |
| 187.00554 | 318658.4 | -0.0397204 | C6H6N4S | C6H6N4S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 166.031318 | 187.0059866 | -2.39 | ['6-Methylmercaptopurine'] |
| 395.09817 | 1920252.2 | -0.0396321 | C16H16O8 | C16H16O8 | [M+Hac-H]- | 336.08452 | 395.0983736 | -0.52 | ['5-0-Caffeoylshikimic acid', 'Altersolanol A'] |
| 139.04 | 300639.2 | -0.0395079 | C7H8O3 | C7H8O3 | [M-H]- | 140.047345 | 139.0400686 | -0.49 | ['2,3,5-Trihydroxytoluene', '2,4,5- <br> Trihydroxytoluene', '2,4,6-Trihydroxytoluene', <br> '4-Hydroxymethylcatechol', 'Gentisyl alcohol'] |


| 199.01676 | 249908.8 | -0.0393805 | C10H1002 | C10H10O2 | [M+K-2H]- | 162.06808 | 199.0166866 | 0.37 | ['(1S,2S)-1,2-Dihydronaphthalene-1,2-diol', '1,2- <br> Dihydronaphthalene-1,2-diol', '4Hydroxycinnamoylmethane', 'Isosafrole', 'Methyl cinnamate', 'Safrole', 'cis-1,2-Dihydronaphthalene-1,2-diol', 'pMethoxycinnamaldehyde', 'trans-2Phenylcyclopropanecarboxylic acid'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 199.01676 | 249908.8 | -0.0393805 | C9H8O3 | C9H8O3 | [ $\mathrm{M}+\mathrm{Cl}$ ]- | 164.047345 | 199.0167466 | 0.07 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2- <br> Hydroxycinnamate', 'cis-p-Coumarate', 'trans-2Hydroxycinnamate'] |
| 201.03801 | 953816.9 | -0.0389213 | C6H12O6 | C6H12O6 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 180.06339 | 201.0380586 | -0.24 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'D- <br> Aldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'DFuconate', 'D-Galactose', 'D-Glucose', 'D- <br> Gulose', 'D-Hamamelose', 'D-Hexose', 'D-Idose', <br> 'D-Mannose', 'D-Psicose', 'D-Sorbose', 'D- <br> Tagatose', 'D-Talose', 'Fructose(pyranose)', <br> 'Ketose', 'L-Fructose', 'L-Fuconate', 'L- <br> Galactose', 'L-Gulose', 'L-Rhamnonate', 'L- <br> Sorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha- <br> D-Glucose', 'alpha-D-Mannose', 'alpha-L- <br> Sorbopyranose', 'beta-D-Fructose', 'beta-DGalactose', 'beta-D-Glucose', 'beta-D- <br> Hamamelopyranose', 'beta-D-Mannose', 'mucoInositol', 'myo-Inositol', 'scyllo-Inositol'] |
| 199.02483 | 240163.1 | -0.0389118 | C6H4O4 | C6H4O4 | [M+Hac-H]- | 140.01096 | 199.0248136 | 0.08 | ['cis-4-Carboxymethylenebut-2-en-4-olide', 'trans-4-Carboxymethylenebut-2-en-4-olide'] |
| 199.02483 | 240163.1 | -0.0389118 | C8H8O6 | C8H8O6 | [M-H]- | 200.03209 | 199.0248136 | 0.08 | ['(3S,4R)-3,4-Dihydroxycyclohexa-1,5-diene-1,4dicarboxylate', '2-Hydroxy-5carboxymethylmuconate semialdehyde', '3Carboxymethylmuconate', '4- <br> Fumarylacetoacetate', '4-Maleylacetoacetate', 'Phthalate 3,4-cis-dihydrodiol', 'cis-4,5-Dihydroxycyclohexa-1(6),2-diene-1,2dicarboxylate'] |
| 181.05062 | 234605.6 | -0.0387545 | C7H6O2 | C7H6O2 | [M+Hac-H]- | 122.03678 | 181.0506336 | -0.08 | ['3-Hydroxybenzaldehyde', '4Hydroxybenzaldehyde', 'Benzoate', 'Salicylaldehyde', 'Tropolone'] |


| 181.05062 | 234605.6 | -0.0387545 | C9H1004 | C9H1004 | [M-H]- | 182.05791 | 181.0506336 | -0.08 | ['(R)-3-(4-Hydroxyphenyl)lactate', "2',6-Dihydroxy-4-methoxyacetophenone", '3,4Dihydroxyphenylpropanoate', '3-(2,3Dihydroxyphenyl)propanoate', '3-(4Hydroxyphenyl)lactate', '3-Methoxy-4hydroxyphenylglycolaldehyde', 'Homovanillate', 'cis-3-(3-Carboxyethenyl)-3,5-cyclohexadiene-1,2-diol'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 317.06649 | 2037180 | -0.0387343 | C11H14N4O5 | C11H14N4O5 | [M+Cl]- | 282.096421 | 317.0658226 | 2.1 | ['8-Oxocoformycin'] |
| 317.06649 | 2037180 | -0.0387343 | C14H1005 | C14H10O5 | [M+Hac-H]- | 258.052825 | 317.0666786 | -0.59 | ['3-Carbethoxypsoralen', 'Alternariol', 'Gentisin', 'Isogentisin', 'Kigelinone', 'Mesuaxanthone A', 'Norlichexanthone', 'Norrubrofusarin'] |
| 317.06649 | 2037180 | -0.0387343 | C16H14O7 | C16H14O7 | [M-H]- | 318.073955 | 317.0666786 | -0.59 | ['Demethylsulochrin', 'Orsellinate depside'] |
| 317.06649 | 2037180 | -0.0387343 | C17H14NO4 | C17H14NO4 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}$ ]- | 296.092284 | 317.0669526 | -1.46 | ['2,3,9,10-Tetrahydroxyberberine'] |
| 223.06123 | 486261.6 | -0.0387005 | C11H12O5 | C11H12O5 | [M-H]- | 224.068475 | 223.0611986 | 0.14 | ['Sinapate'] |
| 223.06123 | 486261.6 | -0.0387005 | C9H8O3 | C9H8O3 | [M+Hac-H]- | 164.047345 | 223.0611986 | 0.14 | ['2-Hydroxy-3-phenylpropenoate', '3-Coumaric acid', '4-Coumarate', 'Benzoyl acetate', 'Caffeic aldehyde', 'Phenylpyruvate', 'cis-2- <br> Hydroxycinnamate', 'cis-p-Coumarate', 'trans-2Hydroxycinnamate'] |
| 333.09773 | 1300773.5 | -0.0385117 | C15H14O5 | C15H14O5 | [M+Hac-H]- | 274.084125 | 333.0979786 | -0.75 | ['(-)-Epiafzelechin', '5-Deoxyleucopelargonidin', <br> 'Afzelechin', 'Apiforol', 'Fisetinidol', 'H37', <br> 'Luteoliflavan', 'Methysticin', 'Phloretin', <br> 'Ptaerochromenol', 'alpha-Pyrufuran', 'betaPyrufuran'] |
| 333.09773 | 1300773.5 | -0.0385117 | C17H1807 | C17H1807 | [M-H]- | 334.105255 | 333.0979786 | -0.75 | ['Byakangelicin'] |
| 211.0248 | 910183.8 | -0.0384871 | C9H8O6 | C9H8O6 | [M-H]- | 212.03209 | 211.0248136 | -0.06 | ['2-Hydroxy-6-ketononatrienedioate', '3-(2-Carboxyethenyl)-cis,cis-muconate', '5Carboxyvanillic acid'] |
| 197.00918 | 236432 | -0.0384558 | C8H6O6 | C8H6O6 | [M-H]- | 198.01644 | 197.0091636 | 0.08 | ['3,4-Dihydroxyphthalate', '4,5Dihydroxyphthalate'] |
| 205.01426 | 83451.9 | -0.0383958 | C10H6O5 | C10H6O5 | [M-H]- | 206.021525 | 205.0142486 | 0.06 | ['Flaviolin'] |
| 280.23607 | 1802041.1 | -0.0383548 |  | 0 |  |  |  |  | 0 |
| 191.03497 | 122503.1 | -0.0382761 | C10H8O4 | C10H8O4 | [M-H]- | 192.04226 | 191.0349836 | -0.07 | ['1,3,6,8-Naphthalenetetrol', '2- <br> Hydroxychromene-2-carboxylate', '3,4- <br> Dehydro-6-hydroxymellein', 'Acamelin', <br> 'Anemonin', 'Isoscopoletin', <br> 'Methylenedioxycinnamic acid', 'Naphthazarin', 'Scopoletin', 'trans-O- |



| 186.0455 | 1001322.2 | -0.0375563 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 154.02205 | 12699.9 | -0.0375245 |  | 0 |  |  |  |  | 0 |
| 180.03832 | 1177706.9 | -0.0373633 |  | 0 |  |  |  |  | 0 |
| 259.06098 | 113359.9 | -0.0370877 | C10H14N4O2 | C10H14N4O2 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 222.111676 | 259.0602826 | 2.69 | ['IBMX'] |
| 259.06098 | 113359.9 | -0.0370877 | C10H14N4S | C10H14N4S | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 222.093918 | 259.0603696 | 2.36 | ['6-(Pentylthio)purine'] |
| 259.06098 | 113359.9 | -0.0370877 | C12H8O3 | C12H8O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 200.047345 | 259.0611986 | -0.84 | ['5-Methylangelicin'] |
| 259.06098 | 113359.9 | -0.0370877 | C14H12O5 | C14H12O5 | [M-H]- | 260.068475 | 259.0611986 | -0.84 | ['Khellin'] |
| 417.08261 | 1642861.2 | -0.0370123 | C14H19N4O9P | C14H19N4O9P | [M-H]- | 418.088969 | 417.0816926 | 2.2 | ['5-Butyrylphosphoinosine'] |
| 537.10396 | 1109936.2 | -0.0369292 | C23H24N4O9 | C23H24N4O9 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 500.154331 | 537.1029376 | 1.9 | ['Isonocardicin A', 'Nocardicin A', 'Nocardicin B'] |
| 537.10396 | 1109936.2 | -0.0369292 | C27H22O12 | C27H22O12 | [M-H]- | 538.11113 | 537.1038536 | 0.2 | ['Lithospermic acid'] |
| 215.01158 | 6512568.1 | -0.0369175 | C10H10O3 | C10H10O3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 178.062995 | 215.0116016 | -0.1 | ['1,2-Dihydroxy-3,4-epoxy-1,2,3,4- <br> tetrahydronaphthalene', '3-Acetyl-6methoxybenzaldehyde', 'Coniferyl aldehyde', 'Vermelone'] |
| 215.01158 | 6512568.1 | -0.0369175 | C9H8O4 | C9H8O4 | [ $\mathrm{M}+\mathrm{Cl}$ ]- | 180.04226 | 215.0116616 | -0.38 | ['2-Hydroxy-3-(4-hydroxyphenyl)propenoate', '3-(4-Hydroxyphenyl)pyruvate', 'Aspirin', 'Caffeate', 'trans-2,3-Dihydroxycinnamate'] |
| 181.01425 | 321909.5 | -0.0367707 | C8H6O5 | C8H6O5 | [M-H]- | 182.021525 | 181.0142486 | 0.01 | ['2-Hydroxyisophthalic acid', '3,5Dihydroxyphenylglyoxylate', '4Hydroxyphthalate', 'Stipitatate'] |
| 395.05379 | 446992.2 | -0.0366284 | C18H16O8 | C18H16O8 | [ $\mathrm{M}+\mathrm{Cl}$ ]- | 360.08452 | 395.0539216 | -0.33 | ["3',4',5-Trihydroxy-3,6,7-trimethoxyflavone", <br> '6-Methoxyaromadendrin 3-0-acetate', <br> 'Acerosin', 'Arcapillin', 'Chrysosplenol C', <br> 'Irigenin', 'Oxyayanin A', 'Oxyayanin B', <br> 'Rosmarinate', 'Thymonin'] |
| 395.05379 | 446992.2 | -0.0366284 | C19H1807 | C19H18O7 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 358.105255 | 395.0538616 | -0.18 | ['Chryso-obtusin', 'Gardenin B'] |
| 395.05379 | 446992.2 | -0.0366284 | C22H14O6 | C22H14O6 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 374.07904 | 395.0537086 | 0.21 | ['Diospyrin', 'Isodiospyrin'] |
| 224.06128 | 81715.1 | -0.0366262 |  | 0 |  |  |  |  | 0 |
| 395.05463 | 32736.4 | -0.0365081 | C18H16O8 | C18H16O8 | [M+Cl]- | 360.08452 | 395.0539216 | 1.79 | ["3',4',5-Trihydroxy-3,6,7-trimethoxyflavone", <br> '6-Methoxyaromadendrin 3-0-acetate', <br> 'Acerosin', 'Arcapillin', 'Chrysosplenol C', <br> 'Irigenin', 'Oxyayanin A', 'Oxyayanin B', <br> 'Rosmarinate', 'Thymonin'] |
| 395.05463 | 32736.4 | -0.0365081 | C19H18O7 | C19H1807 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 358.105255 | 395.0538616 | 1.95 | ['Chryso-obtusin', 'Gardenin B'] |


| 395.05463 | 32736.4 | -0.0365081 | C22H14O6 | C22H14O6 | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 374.07904 | 395.0537086 | 2.33 | ['Diospyrin', 'Isodiospyrin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 205.05065 | 87172.7 | -0.0364532 | C11H1004 | C11H1004 | [M-H]- | 206.05791 | 205.0506336 | 0.08 | ['2-Hydroxy-3-methylbenzalpyruvate', '2-Hydroxy-8-methylchromene-2-carboxylate', 'Lathodoratin', 'Scoparone', 'cis-1,2-Dihydroxy-1,2-dihydro-8-carboxynaphthalene'] |
| 205.05065 | 87172.7 | -0.0364532 | C9H6O2 | C9H6O2 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 146.03678 | 205.0506336 | 0.08 | ['Coumarin'] |
| 269.08172 | 1631837.3 | -0.036289 | C14H1002 | C14H1002 | [M+Hac-H]- | 210.06808 | 269.0819336 | -0.79 | ['1,2-Anthracenediol', '9,10- <br> Dihydroxyanthracene', 'Phenanthrene-3,4-diol'] |
| 269.08172 | 1631837.3 | -0.036289 | C16H14O4 | C16H14O4 | [M-H]- | 270.08921 | 269.0819336 | -0.79 | ['(+)-Medicarpin', '(-)-Medicarpin', '2-0Methylisoliquiritigenin', 'Alloimperatorin', 'Imperatorin', 'Isoimperatorin', 'Pinostrobin', 'Strobopinin', 'Vignafuran'] |
| 205.05247 | 558213 | -0.0361867 |  | 0 |  |  |  |  | 0 |
| 357.06134 | 2441756 | -0.035946 | C16H1006 | C16H1006 | [M+Hac-H]- | 298.04774 | 357.0615936 | -0.71 | ["2',7-Dihydroxy-4',5methylenedioxyisoflavone", 'Bowdichione', 'Irilone'] |
| 161.04546 | 466227.9 | -0.0357169 | C4H6O3 | C4H6O3 | [M+Hac-H]- | 102.031695 | 161.0455486 | -0.55 | ['(S)-Methylmalonate semialdehyde', '2-Methyl-3-oxopropanoate', '2-Oxobutanoate', 'Acetoacetate', 'Succinate semialdehyde'] |
| 161.04546 | 466227.9 | -0.0357169 | C6H1005 | C6H1005 | [M-H]- | 162.052825 | 161.0455486 | -0.55 | ['(2R,3S)-2,3-Dimethylmalate', '(R)-2Ethylmalate', '(R)-3,3-Dimethylmalate', '(S)-2(Hydroxymethyl)glutarate', '1,5-Anhydro-Dfructose', '2-Dehydro-3-deoxy-D-fuconate', '2-Dehydro-3-deoxy-L-fuconate', '2-Dehydro-3-deoxy-L-rhamnonate', '2-Deoxy-scyllo-inosose', '2-Hydroxyadipate', '3,6-Anhydrogalactose', '3,6-Anhydroglucose', '3-EthyImalate', '3-Hydroxy-3-methylglutarate', 'D-Fucono-1,4lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5-lactone', 'L-Rhamnono-1,4-lactone', 'Lichenin'] |
| 335.077 | 558972.3 | -0.0356401 | C16H16O8 | C16H1608 | [M-H]- | 336.08452 | 335.0772436 | -0.73 | ['5-O-Caffeoylshikimic acid', 'Altersolanol A'] |
| 385.09276 | 827361.7 | -0.0356303 | C18H14O6 | C18H14O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 326.07904 | 385.0928936 | -0.35 | ['Jacareubin', 'Ophiopogonone A'] |
| 385.09276 | 827361.7 | -0.0356303 | C20H18O8 | C20H18O8 | [M-H]- | 386.10017 | 385.0928936 | -0.35 | ['Cleomiscosin A', 'Daphneticin', 'Diferulic acid', 'Glucosyloxyanthraquinone', 'Irisflorentin'] |
| 493.11397 | 2028202.5 | -0.0355765 | C24H18O8 | C24H18O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 434.10017 | 493.1140236 | -0.11 | ['Knipholone'] |
| 493.11397 | 2028202.5 | -0.0355765 | C26H22O10 | C26H22O10 | [M-H]- | 494.1213 | 493.1140236 | -0.11 | ['5-Methoxyhydnocarpin-D', 'Salvianolic acid $\left.A^{\prime}\right]$ |


| 209.00921 | 171169.2 | -0.0355146 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 375.07198 | 2208432.8 | -0.0353964 | C16H12O7 | C16H12O7 | [M+Hac-H]- | 316.058305 | 375.0721586 | -0.48 | ["3',4',5,7-Tetrahydroxy-3-methoxyflavone", <br> 'Azaleatin', 'Capillarisin', 'Isorhamnetin', <br> 'Pedalitin', 'Pinoquercetin', 'Rhamnetin', <br> 'Sexangularetin', 'Tamarixetin'] |
| 160.06623 | 321288 | -0.0351747 |  | 0 |  |  |  |  | 0 |
| 188.99604 | 111766.4 | -0.0350723 | C7H6O4 | C7H6O4 | [M+Cl]- | 154.02661 | 188.9960116 | 0.15 | ['2,3-Dihydroxybenzoate', '2,5- <br> Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |
| 188.99604 | 111766.4 | -0.0350723 | C8H8O3 | C8H8O3 | [M+K-2H]- | 152.047345 | 188.9959516 | 0.47 | ['(R)-Mandelate', '(S)-Mandelate', "2',4- <br> Dihydroxyacetophenone", '2- <br> (Hydroxymethyl)benzoic acid', '2- <br> Hydroxyphenylacetate', "3',4- <br> Dihydroxyacetophenone", '3,4- <br> Dihydroxyphenylacetaldehyde', '3- <br> Hydroxyphenylacetate', '3-Methoxytropolone', <br> '3-Methylsalicylate', '4-Hydroxy-3-methoxybenzaldehyde', '4- <br> Hydroxymethylsalicylaldehyde', '4- <br> Hydroxyphenacyl alcohol', '4-Hydroxyphenyl acetate', '4-Hydroxyphenylacetate', '4- <br> Methoxybenzoate', '4-Methylsalicylate', '6Methylsalicylate', 'Menisdaurilide', 'Methyl salicylate', 'Phenoxyacetate', 'Resorcinol monoacetate'] |
| 207.05101 | 329111.7 | -0.0350042 | C5H8O5 | C5H8O5 | [M+Hac-H]- | 148.037175 | 207.0510286 | -0.09 | ['(R)-2-Hydroxyglutarate', '(R)-2-MethyImalate', '(S)-2-Hydroxyglutarate', '(S)-2-Methylmalate', '2-Dehydro-3-deoxy-D-xylonate', '2-Dehydro-3-deoxy-L-arabinonate', '2-Hydroxyglutarate', 'Citramalate', 'D-Arabinono-1,4-lactone', 'D-Xylono-1,4-lactone', 'D-Xylonolactone', 'D-erythro-3-Methylmalate', 'D-threo-3- <br> Methylmalate', 'L-Arabinono-1,4-lactone', 'L-Arabinono-1,5-lactone', 'L-Xylono-1,4-lactone', 'L-threo-3-Methylmalate'] |
| 211.04198 | 246769.6 | -0.0349502 |  | 0 |  |  |  |  | 0 |
| 391.06693 | 684237.4 | -0.0347623 | C10H13N4O7P | C10H13N4O7P | [M+Hac-H]- | 332.052189 | 391.0660426 | 2.27 | ['2-Deoxyinosine 5-phosphate', 'Purine mononucleotide'] |
| 391.06693 | 684237.4 | -0.0347623 | C16H12O8 | C16H12O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 332.05322 | 391.0670736 | -0.37 | ["3,3',4',5,7-Pentahydroxy-8-methoxyflavone", <br> 'Laricitrin', 'Patuletin'] |
| 333.06137 | 2087294 | -0.0347116 | C14H1006 | C14H1006 | [M+Hac-H]- | 274.04774 | 333.0615936 | -0.67 | ["2,2',4,4-Tetrahydroxybenzil", 'Athyriol', <br> 'Bellidifolin', 'Isoathyriol', 'Swertianin'] |


| 333.06137 | 2087294 | -0.0347116 | C16H14O8 | C16H14O8 | [M-H]- | 334.06887 | 333.0615936 | -0.67 | ['5,5-Dehydrodivanillate', '6-Methoxytaxifolin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 257.06642 | 99395.5 | -0.0346071 | C7H16N4O4 | C7H16N4O4 | [M+K-2H]- | 220.117156 | 257.0657626 | 2.56 | ['1D-1-Guanidino-3-amino-1,3-dideoxy-scylloinositol'] |
| 257.06642 | 99395.5 | -0.0346071 | C9H1005 | C9H1005 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 198.052825 | 257.0666786 | -1.01 | ['3-(3,4-Dihydroxyphenyl)lactate', '3-Methoxy-4-hydroxymandelate', 'Syringic acid'] |
| 183.02992 | 103910.5 | -0.0342626 | C6H4O3 | C6H4O3 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 124.016045 | 183.0298986 | 0.12 | ['2-Hydroxy-1,4-benzoquinone'] |
| 183.02992 | 103910.5 | -0.0342626 | C8H8O5 | C8H8O5 | [M-H]- | 184.037175 | 183.0298986 | 0.12 | ['3,4-Dihydroxymandelate', '3-0MethyIgallate'] |
| 255.05077 | 103293.7 | -0.0342531 | C9H8O5 | C9H8O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 196.037175 | 255.0510286 | -1.01 | ['3-(3,4-Dihydroxyphenyl)pyruvate'] |
| 119.05016 | 101690.3 | -0.0342277 | C8H8O | C8H8O | [M-H]- | 120.057515 | 119.0502386 | -0.66 | ['2-Methylbenzaldehyde', '3- <br> Methylbenzaldehyde', '4-Hydroxystyrene', 'Acetophenone', 'Phenylacetaldehyde', 'Styrene oxide', 'p-Tolualdehyde'] |
| 243.05082 | 126889 | -0.0342201 | C8H8O5 | C8H8O5 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 184.037175 | 243.0510286 | -0.86 | ['3,4-Dihydroxymandelate', '3-0Methylgallate'] |
| 125.00098 | 990024.9 | -0.0340907 | C2H704P | C2H704P | [M-H]- | 126.008198 | 125.0009216 | 0.47 | ['2-Hydroxyethylphosphonate'] |
| 125.00098 | 990024.9 | -0.0340907 | C3H6O3 | C3H6O3 | [ $\mathrm{M}+\mathrm{Cl}]$ - | 90.031695 | 125.0010966 | -0.93 | ['(R)-Lactate', '(S)-Lactate', '3Hydroxypropanoate', 'D-Glyceraldehyde', 'Glyceraldehyde', 'Glycerone', 'LGlyceraldehyde', 'Lactate'] |
| 125.00098 | 990024.9 | -0.0340907 | C4H8O2 | C4H8O2 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 88.05243 | 125.0010366 | -0.45 | ['(R)-Acetoin', '1,4-Dioxane', '2- <br> Methylpropanoate', 'Acetoin', 'Butanoic acid', 'Ethyl acetate'] |
| 125.00098 | 990024.9 | -0.0340907 | H302P | H302P | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 65.987068 | 125.0009216 | 0.47 | ['Phosphinate'] |
| 144.04543 | 334245.1 | -0.0340845 | C9H7NO | C9H7NO | [M-H]- | 145.052764 | 144.0454876 | -0.4 | ['1(2H)-Isoquinolinone', '3-Methyleneoxindole', '8-Hydroxyquinoline', 'Indole-3carboxaldehyde', 'Quinolin-2-ol', 'Quinolin-4ol'] |
| 179.05605 | 468250 | -0.0340037 | C4H8O4 | C4H8O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 120.04226 | 179.0561136 | -0.36 | ['D-Erythrose', 'D-Erythrulose', 'D-Threose'] |


| 179.05605 | 468250 | -0.0340037 | C6H12O6 | C6H12O6 | [M-H]- | 180.06339 | 179.0561136 | -0.36 | ['2-Deoxy-D-gluconate', 'Aldohexose', 'DAldose', 'D-Allose', 'D-Altrose', 'D-Fructose', 'D- <br> Fuconate', 'D-Galactose', 'D-Glucose', 'DGulose', 'D-Hamamelose', 'D-Hexose', 'D-Idose', <br> 'D-Mannose', 'D-Psicose', 'D-Sorbose', 'D- <br> Tagatose', 'D-Talose', 'Fructose(pyranose)', <br> 'Ketose', 'L-Fructose', 'L-Fuconate', 'L- <br> Galactose', 'L-Gulose', 'L-Rhamnonate', 'L- <br> Sorbose', 'Sorbose', 'alpha-D-Galactose', 'alpha- <br> D-Glucose', 'alpha-D-Mannose', 'alpha-L- <br> Sorbopyranose', 'beta-D-Fructose', 'beta-DGalactose', 'beta-D-Glucose', 'beta-D- <br> Hamamelopyranose', 'beta-D-Mannose', 'mucoInositol', 'myo-Inositol', 'scyllo-Inositol'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 194.05398 | 673956.5 | -0.0339748 |  | 0 |  |  |  |  | 0 |
| 147.04692 | 98274.6 | -0.0339481 |  | 0 |  |  |  |  | 0 |
| 191.01973 | 295009.9 | -0.0339398 | C4H4O5 | C4H4O5 | [M+Hac-H]- | 132.005875 | 191.0197286 | 0.01 | ['2-Hydroxyethylenedicarboxylate', 'Oxaloacetate', 'trans-2,3-Epoxysuccinate'] |
| 191.01973 | 295009.9 | -0.0339398 | C6H8O7 | C6H8O7 | [M-H]- | 192.027005 | 191.0197286 | 0.01 | ['(1R,2S)-1-Hydroxypropane-1,2,3tricarboxylate', '(1S,2S)-1-Hydroxypropane-1,2,3-tricarboxylate', '(4R,5S)-4,5,6-Trihydroxy-2,3-dioxohexanoate', '2,5-Didehydro-Dgluconate', '2-Dehydro-3-deoxy-D-glucarate', <br> '5-Dehydro-4-deoxy-D-glucarate', 'Carboxymethyloxysuccinate', 'Citrate', 'Isocitrate'] |
| 223.02489 | 84225.1 | -0.0336847 |  | 0 |  |  |  |  | 0 |
| 401.12993 | 78988 | -0.0335669 | C12H22O11 | C12H22O11 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 342.116215 | 401.1300686 | -0.35 | ['2-O-beta-D-Glucopyranosyl-beta-Dglucopyranose', '2-alpha-D-Glucosyl-D-glucose', 'Cellobiose', 'D-Fructosyl-D-fructofuranose', 'D-Glucosyl-D-mannose', 'Epimelibiose', 'Gentiobiose', 'Inulobiose', 'Isomaltose', 'Lactose', 'Lactulose', 'Laminaribiose', 'Levanbiose', 'Maltose', 'Mannobiose', 'Melibiose', 'Nigerose', 'Palatinose', 'Sucrose', 'alpha,alpha-Trehalose', 'alpha-Cellobiose', 'alpha-D-Aldosyl beta-D-fructoside', 'alpha-D-Galactosyl-(1->3)-1D-myo-inositol', 'alpha-D-Glucosyl-(1,3)-D-mannose', 'alpha-Maltose', 'beta-Cellobiose', 'beta-D-Fructofuranosyl-alpha-D-mannopyranoside', 'beta-Lactose', 'beta-Maltose'] |


| 133.0137 | 13169.7 | -0.0335331 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 275.0559 | 628045.8 | -0.0334329 | C12H8O4 | C12H8O4 | [M+Hac-H]- | 216.04226 | 275.0561136 | -0.78 | ['Bergapten', 'Isobergapten', 'Norvisnagin', 'Sphondin', 'Xanthotoxin'] |
| 401.12412 | 45559.9 | -0.0334004 | C19H18O6 | C19H18O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 342.11034 | 401.1241936 | -0.18 | ['3-(4-Methoxyphenyl)-5,6,7-trimethoxy-4H-1-benzopyran-4-one', "4',5,6,7- <br> Tetramethoxyflavone", 'UWM6'] |
| 401.12412 | 45559.9 | -0.0334004 | C21H22O8 | C21H22O8 | [M-H]- | 402.13147 | 401.1241936 | -0.18 | ["(3'R,4'R)-3-Epoxyangeloyloxy-4-acetoxy-3',4dihydroseselin", '2-(2,5-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-4H-1-benzopyran-4-one', <br> '2-(3,5-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-4H-1-benzopyran-4-one', 'Flavanone 7-O-beta-D-glucoside', 'Nobiletin'] |
| 253.0715 | 789261.7 | -0.033369 | C10H1004 | C10H1004 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 194.05791 | 253.0717636 | -1.04 | ['2,4,8-Trihydroxy-1-tetralone', '5- <br> Hydroxyconiferaldehyde', '6-Hydroxymellein', 'Dimethyl phthalate', 'Ferulate', 'Isoferulic acid', 'Kakuol', 'Methyl caffeate', 'Scytalone'] |
| 210.07715 | 126400.7 | -0.0333683 | C10H13NO4 | C1OH13NO4 | [M-H]- | 211.084459 | 210.0771826 | -0.16 | ['Enicoflavine', 'Methyldopa anhydrous'] |
| 210.07715 | 126400.7 | -0.0333683 | C8H9NO2 | C8H9NO2 | [M+Hac-H]- | 151.063329 | 210.0771826 | -0.16 | ['(E)-4-Hydroxyphenylacetaldehyde-oxime', <br> '(R)-Mandelamide', '(Z)-4- <br> Hydroxyphenylacetaldehyde-oxime', '2-Amino- <br> 3-methylbenzoate', '2-Descarboxy-cyclo-dopa', <br> 'Acetaminophen', 'Dopamine quinone', 'L- <br> Phenylglycine', ' N -(Acetyloxy)benzenamine', ' N -Methyl-4-aminobenzoate', ' N Methylanthranilate'] |
| 343.08212 | 1300331.1 | -0.0332226 | C16H12O5 | C16H12O5 | [M+Hac-H]- | 284.068475 | 343.0823286 | -0.61 | ['(+)-Maackiain', '(-)-Maackiain', '2Hydroxyformononetin', '3-Methylgalangin', <br> 'Acacetin', 'Biochanin A', 'Calycosin', <br> 'Cypripedin', 'Genkwanin', 'Geraldone', <br> 'Glycitein', 'Lucidin omega-methyl ether', <br> 'Melannin', 'Obtusifolin', 'Physcion', 'Prunetin', 'Questin', 'Texasin', 'Wogonin'] |
| 343.08212 | 1300331.1 | -0.0332226 | C18H16O7 | C18H16O7 | [M-H]- | 344.089605 | 343.0823286 | -0.61 | ['(-)-Usnic acid', "3',5-Dihydroxy-3,4',7trimethoxyflavone", '7-Hydroxy-(S)-usnate', 'Cirsilineol', 'Eupatilin', 'Nevadensin', 'Obtusin', 'Pachypodol', 'Santin', 'Tambulin', 'Xanthomicrol'] |
| 251.07303 | 281953 | -0.0330092 |  | 0 |  |  |  |  | 0 |
| 555.11465 | 1580125.2 | -0.0330042 |  | 0 |  |  |  |  | 0 |
| 190.99312 | 37269.5 | -0.0329637 | C7H6O4 | C7H6O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 154.02661 | 190.9930616 | 0.31 | ['2,3-Dihydroxybenzoate', '2,5Dihydroxybenzoate', '3,4-Dihydroxybenzoate', 'Patulin'] |


| 209.04552 | 166727.3 | -0.0328744 | C10H10O5 | C10H10O5 | [M-H]- | 210.052825 | 209.0455486 | -0.14 | ['5-Hydroxyferulate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 209.04552 | 166727.3 | -0.0328744 | C8H6O3 | C8H6O3 | [M+Hac-H]- | 150.031695 | 209.0455486 | -0.14 | ['2-Carboxybenzaldehyde', 'Piperonal', 'alpha-Oxo-benzeneacetic acid'] |
| 217.0143 | 134212.3 | -0.032866 |  | 0 |  |  |  |  | 0 |
| 403.10334 | 165903.3 | -0.0328139 | C18H16O7 | C18H16O7 | [M+Hac-H]- | 344.089605 | 403.1034586 | -0.29 | ['(-)-Usnic acid', "3',5-Dihydroxy-3,4',7trimethoxyflavone", '7-Hydroxy-(S)-usnate', 'Cirsilineol', 'Eupatilin', 'Nevadensin', 'Obtusin', 'Pachypodol', 'Santin', 'Tambulin', 'Xanthomicrol'] |
| 213.02119 | 221683.1 | -0.0326735 | C4H8N4O4 | C4H8N4O4 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 176.054556 | 213.0210076 | 0.86 | ['Allantoate'] |
| 213.02119 | 221683.1 | -0.0326735 | C8H8N4S | C8H8N4S | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]-$ | 192.046968 | 213.0216366 | -2.1 | ['6-(Allylthio)purine'] |
| 208.06637 | 148073.6 | -0.0326287 |  | 0 |  |  |  |  | 0 |
| 269.06646 | 108764.3 | -0.032617 | C10H1005 | C10H1005 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 210.052825 | 269.0666786 | -0.81 | ['5-Hydroxyferulate'] |
| 269.06646 | 108764.3 | -0.032617 | C16H12N2 | C16H12N2 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 232.100048 | 269.0664996 | -0.15 | ['Cryptolepine'] |
| 327.08719 | 301802.1 | -0.0325825 | C14H18N4O3 | C14H18N4O3 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 290.137891 | 327.0864976 | 2.12 | ['Benomyl', 'Trimethoprim'] |
| 327.08719 | 301802.1 | -0.0325825 | C16H12O4 | C16H12O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 268.07356 | 327.0874136 | -0.68 | ['1-[6-Hydroxy-2-(4-hydroxyphenyl)-1-benzofuran-3-yl]ethanone', '6-Hydroxy-2methoxyflavone', 'Dalbergin', 'Formononetin', 'Isoformononetin', 'Tectochrysin'] |
| 327.08719 | 301802.1 | -0.0325825 | C16H2ONO4 | C16H2ONO4 | [ $\mathrm{M}+\mathrm{K}-2 \mathrm{H}$ ]- | 290.139234 | 327.0878406 | -1.99 | ['Pteleatine', 'Ribalinium'] |
| 327.08719 | 301802.1 | -0.0325825 | C18H16O6 | C18H16O6 | [M-H]- | 328.09469 | 327.0874136 | -0.68 | ['2-(4-Hydroxyphenyl)-5,6,7-trimethoxy-4H-1-benzopyran-4-one', '6-Hydroxy-2-(4-methoxyphenyl)-5,7-dimethoxy-4H-1-benzopyran-4-one', "7-Hydroxy-2',4',5trimethoxyisoflavone", '9Demethylmunduserone', 'Betagarin', 'Ophiopogonanone A'] |
| 126.99804 | 307852.8 | -0.0324265 | C3H6O3 | C3H6O3 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 90.031695 | 126.9981466 | -0.84 | ['(R)-Lactate', '(S)-Lactate', '3- <br> Hydroxypropanoate', 'D-Glyceraldehyde', <br> 'Glyceraldehyde', 'Glycerone', 'LGlyceraldehyde', 'Lactate'] |
| 391.10332 | 87724.6 | -0.0322512 | C17H1607 | C17H16O7 | [ $\mathrm{H}+\mathrm{Hac}-\mathrm{H}$ ]- | 332.089605 | 391.1034586 | -0.35 | ['Sulochrin'] |
| 347.07702 | 461859.9 | -0.0321014 | C15H12O6 | C15H12O6 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 288.06339 | 347.0772436 | -0.64 | ['(+-)-Dalbergioidin', '2,6,7,4- <br> Tetrahydroxyisoflavanone', '2-Hydroxy-2,3dihydrogenistein', '2-0-Methylswertianin', '3,5-Dimethoxy-1,6-dihydroxyxanthone', 'Carthamidin', 'Dihydrokaempferol', <br> 'Eriodictyol', 'Eriodictyol chalcone', 'Fustin', |


|  |  |  |  |  |  |  |  |  | 'Gentiacaulein', 'Micromelin', 'Okanin', 'Swerchirin'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 216.01505 | 430701 | -0.0320882 |  | 0 |  |  |  |  | 0 |
| 401.08768 | 376286.1 | -0.0320434 | C18H14O7 | C18H14O7 | [M+Hac-H]- | 342.073955 | 401.0878086 | -0.32 | ['Dihydromethylsterigmatocystin'] |
| 401.08768 | 376286.1 | -0.0320434 | C20H18O9 | C20H18O9 | [M-H]- | 402.095085 | 401.0878086 | -0.32 | ['1-Hydroxy-2-(beta-D-glucosyloxy)-9,10anthraquinone', 'Frangulin B'] |
| 89.0243 | 226574 | -0.0319944 | C3H6O3 | C3H6O3 | [M-H]- | 90.031695 | 89.0244186 | -1.33 | ['(R)-Lactate', '(S)-Lactate', '3- <br> Hydroxypropanoate', 'D-Glyceraldehyde', <br> 'Glyceraldehyde', 'Glycerone', 'LGlyceraldehyde', 'Lactate'] |
| 89.0243 | 226574 | -0.0319944 | CH2O | CH2O | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 30.010565 | 89.0244186 | -1.33 | ['Formaldehyde'] |
| 419.09821 | 132487.6 | -0.0319864 | C15H2807P2 | C15H2807P2 | [M+(37Cl)]- | 382.131031 | 419.0974826 | 1.74 | ['2-cis,6-trans-Farnesyl diphosphate', 'Nerolidyl diphosphate', 'trans,trans-Farnesyl diphosphate'] |
| 419.09821 | 132487.6 | -0.0319864 | C18H16O8 | C18H16O8 | [M+Hac-H]- | 360.08452 | 419.0983736 | -0.39 | ["3',4',5-Trihydroxy-3,6,7-trimethoxyflavone", <br> '6-Methoxyaromadendrin 3-0-acetate', <br> 'Acerosin', 'Arcapillin', 'Chrysosplenol C', <br> 'Irigenin', 'Oxyayanin A', 'Oxyayanin B', <br> 'Rosmarinate', 'Thymonin'] |
| 202.94144 | 147951.1 | -0.0318394 |  | 0 |  |  |  |  | 0 |
| 403.0669 | 759660.2 | -0.0317418 | C17H12O8 | C17H12O8 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}]$ - | 344.05322 | 403.0670736 | -0.43 | ['3,4,3-Tri-0-methylellagic acid'] |
| 405.08255 | 190870.4 | -0.0316899 | C17H14O8 | C17H14O8 | [M+Hac-H]- | 346.06887 | 405.0827236 | -0.43 | ["3',4',5,6-Tetrahydroxy-3,7dimethoxyflavone", 'Axillarin', 'Eupatolitin', 'Syringetin', 'Taxifolin 3-O-acetate'] |
| 405.08255 | 190870.4 | -0.0316899 | C19H18O10 | C19H18O10 | [M-H]- | 406.09 | 405.0827236 | -0.43 | ['Lancerin'] |
| 217.00871 | 1221105.3 | -0.0316788 | C9H8O4 | C9H8O4 | [M+(37Cl)]- | 180.04226 | 217.0087116 | -0.01 | ['2-Hydroxy-3-(4-hydroxyphenyl)propenoate', '3-(4-Hydroxyphenyl)pyruvate', 'Aspirin', 'Caffeate', 'trans-2,3-Dihydroxycinnamate'] |
| 171.02744 | 241846.1 | -0.0315208 | C5H1005 | C5H10O5 | [M+Na-2H]- | 150.052825 | 171.0274936 | -0.31 | ['D-Apiose', 'D-Arabinose', 'D-Lyxose', 'DRibose', 'D-Ribulose', 'D-Xylose', 'D-Xylulose', 'LArabinofuranose', 'L-Arabinose', 'L-Lyxose', 'LRibulose', 'L-Xylose', 'L-Xylulose', 'Ribulose', 'Xylose', 'alpha-D-Lyxose', 'alpha-D-Ribulose', 'alpha-L-Arabinose', 'beta-D-Apiose', 'beta-DRibofuranose', 'beta-D-Ribopyranose', 'beta-DXylose', 'beta-L-Arabinose'] |


| 217.0525 | 163287.1 | -0.0314781 |  | 0 |  |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 193.01428 | 50836 | -0.0314647 |  | 0 |  |  |  |  | 0 |
| 123.04507 | 62203 | -0.0314383 | C7H8O2 | C7H8O2 | [M-H]- | 124.05243 | 123.0451536 | -0.68 | ['2,3-Dihydroxytoluene', '3-Hydroxybenzyl alcohol', '4-Methylcatechol', 'Orcinol', 'Salicyl alcohol', 'o-Methoxyphenol', 'p-Hydroxybenzyl alcohol'] |
| 403.08403 | 102555.7 | -0.0313495 |  | 0 |  |  |  |  | 0 |
| 313.07161 | 1442123.9 | -0.0312855 | C15H1004 | C15H10O4 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 254.05791 | 313.0717636 | -0.49 | ['1,4-Dihydroxy-2-methylanthraquinone', "4',6Dihydroxyflavone", '7,4-Dihydroxyflavone', 'Alizarin 2-methyl ether', 'Anhydroglycinol', 'Chrysin', 'Chrysophanol', 'Daidzein', 'Digiferrugineol', 'Hispidol', 'Primetin', 'Rubiadin'] |
| 313.07161 | 1442123.9 | -0.0312855 | C17H14O6 | C17H14O6 | [M-H]- | 314.07904 | 313.0717636 | -0.49 | ['4-Methylcapillarisin', '6-Hydroxy-2-(4-hydroxyphenyl)-5,7-dimethoxy-4H-1-benzopyran-4-one', '7-Methylcapillarisin', 'Aflatoxin B2', 'Irisolidone', 'Pinobanksin 3-0acetate', 'Pisatin', 'Ventinone A'] |
| 313.07161 | 1442123.9 | -0.0312855 | C7H14N2O6S | C7H14N2O6S | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 254.05726 | 313.0711136 | 1.59 | ['5-L-Glutamyl-taurine'] |
| 285.04026 | 781724.3 | -0.0311497 | C15H1006 | C15H10O6 | [M-H]- | 286.04774 | 285.0404636 | -0.71 | ['2-Hydroxygenistein', '6- <br> Demethoxycapillarisin', 'Aureusidin', <br> 'Citreorosein', 'Datiscetin', 'Fisetin', 'Isoscutellarein', 'Kaempferol', 'Luteolin', <br> 'Maritimetin', 'Orobol', 'Scutellarein'] |
| 214.10842 | 336206.9 | -0.0311417 | C10H17NO4 | C1OH17NO4 | [M-H]- | 215.115759 | 214.1084826 | -0.29 | ['2-Amino-9,10-epoxy-8-oxodecanoic acid'] |
| 214.10842 | 336206.9 | -0.0311417 | C8H13NO2 | C8H13NO2 | [M+Hac-H]- | 155.094629 | 214.1084826 | -0.29 | ['Arecoline', 'Heliotridine', 'Retronecine', 'Scopoline'] |
| 539.11975 | 506767.5 | -0.0310363 | C18H30016 | C18H30016 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 502.15339 | 539.1198416 | -0.17 | ['alpha-L-Rhamnopyranosyl-(1->2)-beta-D-galactopyranosyl-(1->2)-beta-Dglucuronopyranoside'] |
| 385.07351 | 159242.4 | -0.0310198 | C15H16N4O6 | C15H16N4O6 | [ $\mathrm{M}+(37 \mathrm{Cl})$ ]- | 348.106986 | 385.0734376 | 0.19 | ['Musca-aurin VII'] |
| 385.07351 | 159242.4 | -0.0310198 | C15H18N4O4S | C15H18N4O4S | [ $\mathrm{M}+\mathrm{Cl}]$ - | 350.104878 | 385.0742796 | -2 | ['Biapenem'] |
| 146.98293 | 114213.8 | -0.030881 | C2H704P | C2H704P | [ $\mathrm{M}+\mathrm{Na}-2 \mathrm{H}]$ - | 126.008198 | 146.9828666 | 0.43 | ['2-Hydroxyethylphosphonate'] |


| 273.07607 | 59213.3 | -0.0308776 | C13H1003 | C13H1003 | [M+Hac-H]- | 214.062995 | 273.0768486 | -2.85 | ['1-Hydro-1,1a-dihydroxy-9-fluorenone', '2,2Dihydroxybenzophenone', '2,4- <br> Dihydroxybenzophenone', '3,4-Dihydroxy-3,4-dihydro-9-fluorenone', '4,4- <br> Dihydroxybenzophenone', '4,4- <br> Dimethylangelicin', '4,5-Dimethylangelicin', 'Diphenyl carbonate', 'Mycosinol', 'Phenyl salicylate'] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 273.07607 | 59213.3 | -0.0308776 | C15H14O5 | C15H14O5 | [M-H]- | 274.084125 | 273.0768486 | -2.85 | ['(-)-Epiafzelechin', '5-Deoxyleucopelargonidin', <br> 'Afzelechin', 'Apiforol', 'Fisetinidol', 'H37', <br> 'Luteoliflavan', 'Methysticin', 'Phloretin', 'Ptaerochromenol', 'alpha-Pyrufuran', 'betaPyrufuran'] |
| 221.06667 | 220150.1 | -0.0307292 | C12H12N2 | C12H12N2 | [M+(37Cl)]- | 184.100048 | 221.0664996 | 0.77 | ['2,4-Diphenyldiamine', 'Benzidine', 'Diquat', 'Withasomnine'] |
| 221.06667 | 220150.1 | -0.0307292 | C6H1005 | C6H1005 | [ $\mathrm{M}+\mathrm{Hac}-\mathrm{H}$ ]- | 162.052825 | 221.0666786 | -0.04 | ['(2R,3S)-2,3-DimethyImalate', '(R)-2Ethylmalate', '(R)-3,3-Dimethylmalate', '(S)-2(Hydroxymethyl)glutarate', '1,5-Anhydro-Dfructose', '2-Dehydro-3-deoxy-D-fuconate', '2-Dehydro-3-deoxy-L-fuconate', '2-Dehydro-3-deoxy-L-rhamnonate', '2-Deoxy-scyllo-inosose', '2-Hydroxyadipate', '3,6-Anhydrogalactose', '3,6-Anhydroglucose', '3-EthyImalate', '3-Hydroxy-3-methylglutarate', 'D-Fucono-1,4lactone', 'Diethyl pyrocarbonate', 'L-Fucono-1,5-lactone', 'L-Rhamnono-1,4-lactone', 'Lichenin'] |
| 221.06667 | 220150.1 | -0.0307292 | C8H14O7 | C8H14O7 | [M-H]- | 222.073955 | 221.0666786 | -0.04 | ['6-Acetyl-D-glucose'] |
| 717.14663 | 1099370.2 | -0.0307193 |  | 0 |  |  |  |  | 0 |
| 147.04512 | 37473.7 | -0.0304938 | C9H8O2 | C9H8O2 | [M-H]- | 148.05243 | 147.0451536 | -0.23 | ['3-Hydroxy-1-indanone', '3-Isochromanone', '4- <br> Hydroxycinnamyl aldehyde', <br> 'Dihydrocoumarin', 'Pyruvophenone', 'transCinnamate'] |
| 128.03522 | 73192 | -0.030486 | C5H7NO3 | C5H7NO3 | [M-H]- | 129.042594 | 128.0353176 | -0.76 | ['1-Pyrroline-4-hydroxy-2-carboxylate', '4Oxoproline', '5-Oxo-D-proline', '5-Oxoproline', 'L-1-Pyrroline-3-hydroxy-5-carboxylate'] |
| 387.10842 | 197572.3 | -0.0303295 | C16H22N4O3S | C16H22N4O3S | [M+(37Cl)]- | 350.141263 | 387.1077146 | 1.82 | ['Cafenstrole'] |


| 387.10842 | 197572.3 | -0.0303295 | C18H1606 | C18H1606 | [M+Hac-H]- | 328.09469 | 387.1085436 | -0.32 | ['2-(4-Hydroxypheny)-5,6,7-trimethoxy-4H-1-benzopyran-4-one', '6-Hydroxy-2-(4-methoxyphenyl)-5,7-dimethoxy-4H-1-benzopyran-4-one', "7-Hydroxy-2',4',5trimethoxyisoflavone", '9Demethylmunduserone', 'Betagarin', 'Ophiopogonanone $\mathrm{A}^{\prime}$ ] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 389.08766 | 149844.8 | -0.0302431 | C17H1407 | C17H1407 | [M+Hac-H]- | 330.073955 | 389.0878086 | -0.38 | ['(+)-Bisdechlorogeodin', '(-)Bisdechlorogeodin', "3',4',5-Trihydroxy-3,7dimethoxyflavone", 'Aflatoxin G2', 'Aurantioobtusin', 'Cirsiliol', 'Hildecarpin', 'Tricin'] |

Table 5 Context V12-VL-53B; shallow ditch with organic material. Date of context is circa 130 A.D.

|  | Sample weight | Malonic acid | Succinic acid | Malci acid | Tartaric acid | $\begin{gathered} \text { Gentisic } \\ \text { acid } \end{gathered}$ | $\begin{gathered} \text { p-Coumaric } \\ \text { acid } \end{gathered}$ | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic | 'Bound' lipids |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphora V12-53B/1 | 1.00 | ND | ND | ND | ND | ND | 1788 | 515 | ND | ND | 467 | ND |


|  | Acyl carbon <br> number | Types of <br> TAGs | Fatty acid constituents |  | Acyl carbon <br> number <br> MANUAL ID | Types of <br> TAGs <br> MANUAL ID |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Fatty acid <br> constituents <br> MANUAL ID |  |  |  |  |  |  |
| side of coarse, thick <br> walled amphora | C48:2 | 2 | $16: 1 / 16 / 16: 1$ | $16 / 16 / 16: 2$ | ND | ND |
| VL-53B/1 |  |  | ND |  |  |  |

Table 6 Context V12-VL-103B; water channel of insecure context time line, probably $3^{\text {rd }}$ century

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | $\begin{aligned} & \text { p-Coumaric } \\ & \text { acid } \end{aligned}$ | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphora V11103A/2 | . 987 | ND | ND | ND | ND | ND | ND | 1404 | ND | ND | 182 | ND |
| *Base of mortarium V11-103A/3 | 1.01 | ND | ND | ND | ND | ND | ND | 2555 | ND | ND | 925 | ND |

Table 7 Context V12-VL-48B, insecure context next to the B7 road, no date given, probably $3^{\text {rd }}$ century

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphorae V1248B/1 | . 983 | ND | 17789 | ND | ND | ND | 2651 | 1450 | ND | 1905 | 4369 | oleic, palmitic, and stearic acids |

Table 8 Context V12-VL-30B

|  | Sample weight (g) | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p- <br> Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | $\begin{gathered} \text { Syringic } \\ \text { acid } \end{gathered}$ | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphora/storage vessel V12-30B/1 | . 991 | ND | 5437 | ND | ND | ND | 634 | 288 | ND | ND | 519 | hexadecanoic, octadecanoic, and octadecenoic acids |
| *Side of amphora/storage vessel V12-30B/2 | 1.01 | ND | 30244 | ND | ND | ND | 11394 | 8929 | ND | 2990 | 6076 | ND |

Table 9 Context V11-VL-3B

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Base of mortarium V11-3B/5 | 1.01 | ND | 38562 | ND | ND | ND | 4727 | 7701 | ND | 10485 | 5919 | low levels hexadecanoic, octadecanoic acids, as well as unidentified lipids including mass 237 |
| *Base of storage vessel V11-3B/3 | . 988 | 1598 | ND | ND | ND | ND | ND | 7943 | ND | ND | 2793 | decanedioic acid (sebacic acid), nonanedioic acid (azelaic acid), Decanoic acid (capric acid), nonanoic acid, 1,8 -octanedioic acid (suberic acid), 3-hydroxydecanedioic acid, 14-hydroxytetradecanoic acid ( $\omega$-hydroxy myristic acid) |

Table 10 Context V11-VL-108A

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphora V11-108A/4 | . 984 | ND | 14108 | ND | ND | ND | 730 | 940 | ND | 1141 | 1691 | ND |

Table 11 Context V11-VL-124A, from the floor of a barrack;s room. 213-300 A.D.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of amphora V11124A/1 | . 983 | ND | ND | ND | ND | ND | ND | 1136 | ND | ND | ND | ND |
| *Side of amphora V11124A/2 | 1.01 | ND | 2851 | ND | ND | ND | 1712 | 1203 | ND | 513 | 1896 | ND |

Table 12 Context V12-VL-41B

|  | Sample weight (g) | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of cook pot V12-41B/3 | . 978 | ND | 6033 | ND | ND | ND | 3747 | 1477 | ND | 4513 | 3089 | ND |

Table 13 Context V12-VL-5B, unstratified soil separating two other contexts, 353-358 A.D.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *cook pot <br> V12-5B/1 | 1.01 | ND | 34523 | ND | ND | ND | 8265 | 8811 | ND | 9233 | 8112 | low levels hexadecanoic, octadecanoic, and octadecenoic acids |

Table 14 Context V12-VL-51B, fabric of $3^{\text {rd }}$ century road, circa 213 A.D.

|  | Sample weight | Malonic acid | Succinic acid | Malic Acid | Tartaric acid | Gentisic acid | p-Coumaric acid | Vanillic acid | Iso-propyl malic acid | Ferulic acid | Syringic acid | 'Bound Lipids' |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| *Side of cook pot v12-51B/5 | . 984 | ND | 22562 | ND | ND | ND | 9702 | 5292 | ND | 1828 | 1968 | low levels of hexadecanoic, octadecanoicand unidentified at $m / z 237$ |


[^0]:    Equation 1.2 The charge formed at the tip of a capillary is described by: $V_{C}$, the voltage applied to the capillary; $r_{C}$, the

