Copyright

by

Seung Wook Kim

2020

The Dissertation Committee for Seung Wook Kim Certifies that this is the approved version of the following Dissertation:

Amphiphilic π -Allyliridium Catalyzed Nucleophilic and ElectrophilicAllylation

Committee:

Michael J. Krische, Supervisor

Hung-Wen (Ben) Liu

Kami L. Hull

Christian P. Whitman

Amphiphilic π -Allyliridium Catalyzed Nucleophilic and ElectrophilicAllylation

by

Seung Wook Kim

Dissertation

Presented to the Faculty of the Graduate School of

The University of Texas at Austin

in Partial Fulfillment

of the Requirements

for the Degree of

Doctor of Philosophy

The University of Texas at Austin May 2020

Acknowledgements

First of all, I would like to thank Prof. Krische for giving me great opportunities and support during my Ph.D. His mentorship and training have been valuable, and they enabled me to grow into a more mature synthetic chemist. All the passion you showed influenced how to think about chemistry as well as about life.

Many former and current post-docs and students in the Krische research group have helped and guided me in the right direction as well. It was unforgettable that we always encouraged, helped and cheered each other on, and it was, I strongly believe, a fantastic work atmosphere.

Finally, I would like to thank my family for their dedication and support during my entire education, and I want to thank my love Ahyeon. Without you, I wouldn't be able to finish this study. You are one of the best supporters and always give me your passion, energy and cheer all the way from South Korea. Last but not least, I would like to finalize this acknowledgement with praise to my God.

Abstract

Amphiphilic π-Allyliridium Catalyzed Nucleophilic and Electrophilic Allylation

Seung Wook Kim, Ph.D. The University of Texas at Austin, 2020

Supervisor: Michael J. Krische

Transition metal-catalyzed allylic substitution has emerged as a powerful method for stereoselective C-N bond formation. Chiral iridium-phosphoramidite complexes have proven especially effective as catalysts for regio- and enantioselective allylic amination, but are limited to aryl-substituted π -allyl electrophiles. With commercially available π allyliridium *C*,*O*-benzoates, which are stable to air, water and SiO₂ chromatography, and are well known to catalyze allylic acetate mediated carbonyl allylation, highly regio- and enantioselective electrophilic allylation of aliphatic amines, primary and secondary aromatic or heteroaromatic amines were demonstrated. Furthermore, indoles and related azoles can also undergo the amination and generate enantiomerically enriched *N*-allyl indoles with completely *N*-selective and exclusive branched regioselectivity, which are an unmet challenge in this field. Moreover, indoles and related azoles are prevalent structural motifs in clinical candidates and FDA approved drugs, so these results also show the utility and importance of asymmetric allylic allylation.

Table of Contents

List of Tablesx
List of Figures xii
List of Schemes xiv
Chapter 1: Intermolecular Late Transition Metal Catalyzed Enantioselevtive Allylic Amination
1.1 Introduction1
1.2 Intermolecular Iridium Catalyzed Enantioselective Allylic Amination2
1.2.1 Allylic Carbonate
1.2.2 Allylic Ester
1.2.3 Allylic Alcohol16
1.2.4. Allylic Phosphate
1.3 Intermolecular Rhodium Catalyzed Enantioselective Allylic Amination20
1.3.1 Allylic Derivatives
1.3.2 C-H Functionalization
1.4 Intermolecular Palladium Catalyzed Enantioselective Allylic Amination26
1.4.1 Allylic Epoxide, Aziridine, Cyclic Carbonate and Carbamate
1.4.2 Allylic Ester and Carbonate
1.4.3 Allylic Alcohol35
1.4.4. C-H Functionalization
1.5 Intermolecular Ruthenium Catalyzed Enantioselective Allylic Amination39
1.6 Conclusion41

Chapter 2: Asymmetric Allylation of Glycidols Mediated by Allyl Acetate via Iridium-Catalyzed Hydrogen Transfer	42
2.1 Introduction	42
2.2 Reaction Development and Scope	43
2.3 Discussion	46
2.4 Conclusion	47
2.5 Experimental Details	48
Chapter 3: Nickel-Catalyzed Cross-Coupling of Vinyl Dioxanones to Form Enantiomerically Enriched Cyclopropanes	96
3.1 Introduction	98
3.2 Reaction Development and Scope	100
3.3 Discussion	103
3.4 Conclusion	104
3.5 Experimental Details	104
Chapter 4: Amphiphilic π-Allyliridium C,O-Benzoates Enable Regio- and Enantioselective Amination of Branched Allylic Acetates Bearing Linear Alkyl Groups	244
4.1 Introduction	244
4.2 Reaction Development and Scope	246
4.3 Discussion	250
4.4 Conclusion	251
4.5 Experimental Details	252

Chapter 5: Regio- and Enantioselective Iridium-Catalyzed Amination of Racemic Branched Alkyl-Substituted Allylic Acetates with Primary and Secondary	
Aromatic and Heteroaromatic Amines	386
5.1 Introduction	386
5.2 Reaction Development and Scope	388
5.3 Discussion	395
5.4 Conclusion	397
5.5 Experimental Details	398
Chapter 6: Regio-and Enantioselective Iridium-Catalyzed N-Allylation of Indoles and Related Azoles with Racemic Branched Alkyl-Substituted Allylic Acetates	1d 530
6.1 Introduction	530
6.2 Reaction Development and Scope	531
6.3 Discussion	537
6.4 Conclusion	537
6.5 Experimental Details	538
Chapter 7: Hydroamination versus Allylic Amination in Iridium-Catalyzed Reaction of Allylic Acetates with Amines: 1,3-Aminoalcohols via Ester-Directed	15
Regioselectivity	651
7.1 Introduction	651
7.2 Reaction Development and Scope	653
7.3 Discussion	656
7.4 Conclusion	658
7.5 Experimental Details	658

Chapter 8: Inversion of Enantioselectivity in Allene Gas versus Allyl Aceta Reductive Aldehyde Allylation Guided by Metal-Centered Stereogenici	ite ity: An
Experimental and Computational Study	
8.1 Introduction	739
8.2 Reaction Development and Scope	741
8.3 Discussion	746
8.4 Conclusion	749
8.5 Experimental Details	751

List of Tables

Table 2.1	Iridium catalyzed C-H allylation of glycidols 2.1a-2.2f to form adducts 2.2a-
	2.2f and <i>epi</i> - 2.2a - 2.2f
Table 3.1	Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a-
	3.1i with tri(<i>p</i> -tolyl)boroxine 3.2a to form cyclopropanes 3.3a-3.3i 100
Table 3.2	Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a
	or 3.1h with boroxines 3.2b-3.2d to form cyclopropanes 3.3j-3.3o 101
Table 3.3	Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a,
	3.1b, 3.1d, 3.1h and 3.1f with B ₂ (pin) ₂ to form cyclopropanes 3.3p-3.3t 102
Table 4.1	Influence of Base in the Amination of α -Methyl Allyl Acetate 4.1a to
	Form Allylic Amine 4.3a 247
Table 4.2	Regio- and Enantioselective Iridium-catalyzed Amination of Branched
	Allylic Carboxylates
Table 5.1	Iridium-catalyzed amination of α -methyl allyl acetate 5.1a with primary
	aromatic and heteroaromatic amines 5.2a-5.2l to form enantiomerically
	enriched allylic amines 5.4a-5.4l
Table 5.2	Iridium-catalyzed amination of α -methyl allyl acetate 5.1a with
	secondary aromatic and heteroaromatic amines 5.3a-5.3l to form
	enantiomerically enriched allylic amines 5.5a-5.5l
Table 5.3	Iridium-catalyzed amination of α -substituted allyl acetates 5.1a-5.1g
	with secondary heteroaromatic amine 5.3m to form enantiomerically
	enriched allylic amines 5.6a-5.6g
Table 6.1	Regio- and enantioselective iridium-catalyzed allylic alkylation of
	indoles and related azoles 6.1a-6.1s using α -methyl allyl acetate 6.2a 533

Table 6.2	Regio- and enantioselective iridium-catalyzed allylic alkylation of indole	e
	6.1a using diverse α-substituted allyl acetates 6.2b-6.2j	.535
Table 7.1	Selected optimization experiments in the reaction of cinnamyl acetate	
	7.1a with benzyl amine 7.2a to form hydroamination product 7.3a	.654
Table 7.2	Regioselective iridium-catalyzed hydroamination of linear allylic	
	acetates	.655
Table 8.1	Selected optimization experiments in the enantioselective reductive	
	coupling of allene 8.1a with aldehyde 8.2a and divergent	
	enantioselectivity observed upon use of allyl acetate vs allene	
	pronucleophiles	.743
Table 8.2	Enantioselective iridium-catalyzed reductive coupling of gaseous allene	
	8.1a with aldehydes 8.2a-8.2s mediated by 2-propanol	745

List of Figures

Figure 1.1	Examples of Bioactive Molecules Bearing a-Chiral Amines	1
Figure 4.1	Phosphoramidite modified iridium complexes and amphiphilic π -	
	allyliridium C,O-benzoates for regio- and enantioselective allylic	
	amination	.245
Figure 4.2	Stereochemical models accounting for equivalent π -facial selectivity in	
	crotylation and amination	.251
Figure 5.1	Cationic vs neutral chiral iridium complexes for regio- and	
	enantioselective allylic amination	.387
Figure 6.1	Iridium catalysts for asymmetric allylic amination and chiral indole-	
	containing clinical candidates	.531
Figure 7.1	Cationic vs neutral iridium catalysts promote Tsuji-Trost allylic	
	amination and hydroamination, respectively	.652
Figure 8.1	Allene feedstocks formed in petroleum cracking	.739
Figure 8.2	Milestones in metal-catalyzed allene-carbonyl reductive coupling and	
	related hydrogen auto-transfer processes	.740
Figure 8.3	Diastereomeric composition of the (S) -Ir-V and calculated	
	thermodynamic stabilities	.744
Figure 8.4	Hypothesis for enantiodivergence in aldehyde allylations mediated by	
	gaseous allene vs allyl acetate	.746
Figure 8.5	Computed energy profiles of the kinetic pathways leading to	
	diastereomers D and C and, therefrom, (S) -and (R) -product enantiomers,	
	respectively	.748

List of Schemes

Scheme 1.1	First Enantioselective Iridium Catalyzed Allylic Amination4
Scheme 1.2	Proposed Catalytic Cycle for Iridium Catalyzed Allylic Amination4
Scheme 1.3	Iridium Catalyzed Enantioselective Allylic Amination to Form
	Highly Functionalized Chiral Branched Amines5
Scheme 1.4	Iridium Catalyzed Enantioselective Allylic Amination with
	Sulfonamide and Carboxamide
Scheme 1.5	Sequential Catalytic Isomerization and Enantioselective Allylic
	Amination6
Scheme 1.6	Iridium Catalyzed Enantioselective Allylic Aminations with
	Premetalated Nucleophiles7
Scheme 1.7	Iridium Catalyzed Enantioselective Monoallylation of Ammonia and
	Primary Carbamate
Scheme 1.8	Air Stable Iridium Catalysts Derived from Other Diene Ligands9
Scheme 1.9	Iridium Catalyzed Enantioselective Allylic Amination with Indoles
	and Azoles10
Scheme 1.10	Iridium Catalyzed Enantioselective Allylic Aminations to Form α -
	Chiral Indole Derivatives11
Scheme 1.11	Sequential Allylic C-H Functionalization and Enantioselective
	Allylic Amination
Scheme 1.12	Regio- and Enantioselective Synthesis of N-Substituted 2-Pyridones12
Scheme 1.13	Enantioselective Allylic Amination with N-Aryl Phosphoramidite
	Ligands13

Scheme 1.14	Enantioselective Iridium Catalyzed Allylic Aminations of	
	Diphenylsulfilimine	.14
Scheme 1.15	Enantioselective Iridium Catalyzed Electrophilic Allylations of	
	Alkyl-Substituted Allylic Acetates	.15
Scheme 1.16	Enantioselective Allylic Aminations of Allylic Alcohols Activated	
	by Lewis Acids	.16
Scheme 1.17	Enantioselective Allylic Aminations by Iridium Complex Modified	
	by Chiral Phosphine-Olefin Ligands	.17
Scheme 1.18	Enantioselective Allylic Aminations of Allylic Phosphate	
	Proelectrophiles with Iridium Pybox Complexes	.18
Scheme 1.19	Iridium Catalyzed Enantioselective Allylic Aminations with Allylic	
	Phosphates	.19
Scheme 1.20	Rhodium Catalyzed Formation of Enantiomerically Enriched Allylic	
	Amines Bearing α-Tetrasubstituted Tertiary Carbons	.20
Scheme 1.21	Enantioselective Synthesis of Branched Allylic Amines via Rhodium	
	Catalyzed Allylic Substitution	.21
Scheme 1.22	Enantioselective Synthesis of Branched Allylic Amines via Rhodium	
	Catalyzed Hydroamination	.22
Scheme 1.23	Rhodium Catalyzed Enantioselective Synthesis of Branched Allylic	
	Amines Using a Variety of Nucleophiles	.23
Scheme 1.24	Sequential Allylic C-H Functionalization and Fischer Indole	
	Synthesis	.24
Scheme 1.25	Rhodium Catalyzed Hydrofunctionalization to Form Chiral Allylic	
	Amines	.25

Scheme 1.26	Rhodium Catalyzed Hydrofunctionalization to Form Chiral Allylic
	Amines via Alkyne Isomerization20
Scheme 1.27	Palladium Catalyzed Dynamic Kinetic Asymmetric Transformation
	of Butadiene Monoxide and Vinyl Aziridine28
Scheme 1.28	Palladium Catalyzed Dynamic Kinetic Asymmetric Cycloadditions
	of Isocyanates
Scheme 1.29	Palladium Catalyzed Enantioselective Allylic Aminations of
	Divinylethylene Carbonate
Scheme 1.30	Palladium Catalyzed Enantioselective Allylic Aminations of Vinylic
	Cyclic Carbonate
Scheme 1.31	Palladium Catalyzed Enantioselective Allylic Amination of Vinyl
	Benzoxazinones
Scheme 1.32	Palladium Catalyzed Enantioselective Allylic Aminations of Allylic
	Proelectrophiles
Scheme 1.33	Palladium Catalyzed Enantioselective Allylic Aminations of
	Racemic Morita-Baylis-Hillman Adducts
Scheme 1.34	A Palladium Catalyzed Synthesis of Chiral α,β -Unsaturated γ -Amino
	Esters
Scheme 1.35	Palladium Catalyzed Regio- and Enantioselective Allylic
	Aminations of 2-Pyridones
Scheme 1.36	Cooperative Catalysis by Palladium and a Chiral Phosphoric Acid
	Promoted Regio- and Enantioselective Allylic Aminations
Scheme 1.37	Palladium Catalyzed Enantioselective Hydroamination of 1,3-Dienes37
Scheme 1.38	Palladium Catalyzed Regio- and Enantioselective Hydroaminations
	of Terminal Allenes
	•

Scheme 1.39	Ruthenium Catalyzed Enantioselective Allylic Aminations40
Scheme 3.1	Synthesis of enantiomerically enriched cyclopropanes from vinyl-
	dioxanones by way of transient (cyclopropyl-carbinyl)nickel species96
Scheme 3.2	General catalytic mechanism. Haptomeric equilibria are excluded for
	clarity
Scheme 4.1	Identical π -allyliridium <i>C</i> , <i>O</i> -benzoate complexes promote both
	nucleophilic and electrophilic allylation250
Scheme 5.1	Iridium-catalyzed amination of enantiomerically enriched deuterated
	allylic acetate 5.1h with the enantiomeric catalysts (S)-Ir-II and (R)-
	Ir-II
Scheme 5.2	General catalytic mechanism and stereochemical model for
	enantioselective iridium-catalyzed allylic amination
Scheme 6.1	Iridium-catalyzed amination of enantiomerically enriched deuterated
	allylic acetate 6.2k with the enantiomeric catalysts (S)-Ir-I and (R)-
	Ir-I
Scheme 7.1	Deuterium labelling experiments657
Scheme 7.2	General catalytic mechanism for iridium catalyzed hydroamination
	of linear allylic acetates, as corroborated by deuterium labeling657

Chapter 1: Intermolecular Late Transition Metal Catalyzed Enantioselective Allylic Amination

1.1 Introduction

Since the pioneering work of Tsuji and Trost on allylic substitution reactions catalyzed by late transition metals, diverse synthetic methodologies for enantioselective allylic substitution have emerged over the last decades.¹ With the use of suitable chiral ligands around the catalyst metal center, enantioselective allylic substitution allows for the construction of covalent bonds between carbon and various other elements in a stereocontrolled fashion. A wide range of carbon-carbon and carbon-heteroatom bonds have been constructed *via* enantioselective allylic substitutions catalyzed by late transition metals. In particular, enantioselective allylic aminations have been highlighted



Figure 1.1 Examples of Bioactive Molecules Bearing α -Chiral Amines.

as one of the most powerful methods to provide enantiomerically enriched allylic amine products. The efficient enantioselective construction of this synthetically useful motif can streamline the approaches for natural products, semisynthetic and synthetic pharmaceutical and agrochemical ingredients.

In this review, we summarize intermolecular enantioselective allylic aminations employing racemic allylic derivatives, dienes, allenes, and alkynes fragments as allyl donors catalyzed by late transition metals. Literature is organized by transition metal catalyst. Asymmetric hydrogenations, enantio- and diastereospecific allylic aminations are not discussed. Cyclic allylic proelectrophiles and symmetric allylic derivatives are not explicitly depicted.

1.2 Intermolecular Iridium Catalyzed Enantioselective Allylic Amination

The formation of branched chiral allylic amines has been greatly enabled by late transition metal catalyzed enantioselective allylic amination. Since the seminal work of Tsuji and Trost, iridium catalysts feature prominently in these transformations. In 1997, the first iridium catalyzed allylic amination was reported by Takeuchi.² Using [Ir(COD)Cl]₂ and triphenyl phosphite, high levels of regioselectivity for branched products were observed, which is distinct from that of palladium catalyzed reactions, which give linear products from nucleophilic attack of the amine on the less substituted terminus of the allylic group. Utilizing this complementary regioselectivity, the first iridium catalyzed enantioselective allylic alkylation was reported by Helmchen in 1997,³ opening a new avenue for the construction of synthetically useful chiral products utilizing catalysts modified by chiral ligands. A series of studies by Helmchen and Takeuchi aimed at better understanding iridium catalyzed allylic substitutions have initiated the

large expansion of this enantioselective allylic amination chemistry over the past decade to form chiral branched allylic amines.

1.2.1 Allylic Carbonate

In 2001, Takeuchi reported the very first iridium catalyzed allylic amination, generating racemic branched allylic amine products.⁴ Subsequent to this effort, in 2002, Hartwig reported the first iridium catalyzed enantioselective allylic amination (Scheme 1.1).⁵ Using the Feringa ligand **1.4**, which was used for enantioselective conjugate additions and known for its π -accepting properties,⁶ primary and secondary aliphatic amines reacted with allylic carbonates to form enantiomerically enriched allylic amines with high yields, and excellent regio- and enantioselectivities. Further mechanistic investigations revealed that the active species was the cyclometalated iridacycle formed from base mediated C-H activation of the initial iridium complex coordinated by the phosphoramidite ligand (Scheme 1.2).^{7,8} For this reason, weak nucleophiles, such as aniline, were not strong enough to enable this cyclometallation process to occur and therefore did not furnish the product of chiral allylic amines. To address this issue, Hartwig demonstrated that adding a catalytic amount of additional base such as DABCO plays a key role in this activation process and enforces the formation of branched allylic aryl amines with high regio- and enantioselectivity.⁹ Another approach to generate this cyclometalated complex is the *in situ* preparation where heating of [Ir(COD)Cl]₂, phosphoramidite ligand, and N-propylamine in THF at 50 °C, followed by vacuum evaporation, yields the active catalyst. Many of the following works have exploited these *in situ* conditions for the generation of catalytically competent catalyst.



Scheme 1.1 First Enantioselective Iridium Catalyzed Allylic Amination.



Scheme 1.2 Proposed Catalytic Cycle for Iridium Catalyzed Allylic Amination.



Scheme 1.3 Iridium Catalyzed Enantioselective Allylic Amination to Form Highly Functionalized Chiral Branched Amines.

In 2004, Helmchen reported the iridium catalyzed regio- and enantioselective allylic amination of linear dienyl carbonates (Scheme 1.3).¹⁰ Using iridium catalysts modified by a chiral phosphoramidite ligand, excellent levels of regio- and enantioselectivity of highly functionalized branched amines were observed. One year later, it was shown that the loading of the iridium catalyst could be decreased with the use of catalytic amounts of Pb(II) salts and tetrahydrothiophene (THT). These additives were effective in helping to vacate coordination sites at the iridium complex, allowing for oxidative addition of allylic proelectrophiles and significantly enhancing reaction rate.¹¹ In 2005, using an iridium pre-catalyst and the chiral phosphoramidite ligand 1.13 developed by Alexakis,¹² Helmchen reported an enantioselective allylic amination of sulfonamides in good yields with high regio- and enantioselectivity, although reactions with alkyl-substituted allylic proelectrophiles yielded incomplete regioselectivities (Scheme 1.4).¹³ Use of base to generate anionic *N*-nucleophiles resulted in slightly higher yields and enantioselectivities. A year later, the scope of these conditions was extended with the use of phthalimide and *ortho*-nosylamine as nucleophiles.¹⁴ The allylic amine products can be readily transformed into enantiomerically enriched primary amine hydrochloride salts. This method was later extended to other carboxamide for the asymmetric synthesis of amino alcohols (not shown).¹⁵



Scheme 1.4 Iridium Catalyzed Enantioselective Allylic Amination with Sulfonamide and Carboxamide.



Scheme 1.5 Sequential Catalytic Isomerization and Enantioselective Allylic Amination.

In 2006, Hartwig reported a sequential, two catalytic process using a palladium catalyst and then an iridium catalyst modified by the phosphoramidite ligand **1.17**¹⁶ which was developed to enforce both yield and selectivity (Scheme 1.5).¹⁷ Iridium catalyzed allylic amination with racemic branched allylic proelectrophiles represents a challenge due to low levels of enantioselectivity typically observed.¹⁸ To broaden the scope of electrophiles for this transformation, a sequence of palladium catalyzed isomerization followed by iridium catalyzed allylic amination was developed, allowing

the formation of branched allylic amines in good yields with high regio- and enantioselectivity, using aliphatic and aromatic amines.



Scheme 1.6 Iridium Catalyzed Enantioselective Allylic Aminations with Premetalated Nucleophiles.

In 2007, Hartwig developed an enantioselective allylic amination using di-tertbutyliminodicarboxylate and trifluoroacetamide as ammonia surrogates to form allylic amines in high yields with good regio- and enantioselectivity (Scheme 1.6).¹⁹ Although the generation of stoichiometric premetalated nucleophiles and metallic byproducts, from treatment of amide with stoichiometric potassium and lithium salts was inevitable, the resulting products could be easily transformed to the enantiomerically enriched allylic carbamates or primary amines. In an analogous manner, the enantioselective allylic amination of sodium benzotriazolide was reported by Zhao in 2012.²⁰ Using an iridium catalyst modified by a Feringa ligand, excellent yields and enantioselectivity were observed, although the branched N1 and N2 regioselectivities were modest.

A few years later, to streamline such stepwise processes, the direct allylation of ammonia and primary carbamates were disclosed (scheme 1.7).²¹ Enantioselective monoallylation of ammonia generally poses an intrinsic limitation for two reasons. First, an ammonia molecule bounded to a metal complex is no longer catalytically competent. These ammonia molecules can displace chiral ligands around the metal, resulting in the formation of an achiral complex. Second, the monoallylated product is more nucleophilic than ammonia, which promotes further allylation. Using ethylene ligated cyclometalated iridium catalysts modified by a chiral phosphoramidite ligand, which are stable to excess ammonia, both aryl and alkyl-substituted allylic carbonates react with ammonia to form monoallylation products in good yields with excellent levels of enantioselectivity. In an analogous manner, primary carbamates also participate in enantioselective iridium catalyzed allylic aminations.²²



Scheme 1.7 Iridium Catalyzed Enantioselective Monoallylation of Ammonia and Primary Carbamate.

Throughout extensive studies towards enantioselective allylic aminations, it has been found that the most active iridium catalyst species are those generated from [Ir(COD)CI]₂ and chiral phosphoramidite ligands with the use of bases (DABCO, TBD or "PrNH₂). However, formation of the cyclometalated iridacycle is very sensitive to the presence of either oxygen or water. To address these issues, air stable iridium catalysts derived from a chiral phosphoramidite ligand and dibenzo[a,e]cyclooctatetraene (dbcot), which displays strong binding affinity to iridium, were reported by Helmchen in 2008.²³ Taking advantage of this air stable catalyst, which avoids the need for Schlenk techniques and requirement for an inert atmosphere, Nelson and Marsdena reported iridium catalyzed allylic amination reactions of highly functionalized amines. The resulting products have the physicochemical properties, which can be useful drug-like space (Not shown).²⁴ In 2013, You reported an air stable iridium complex derived from dinaphthocyclooctatetraene (dncot) and a phosphoramidite ligand (Scheme 1.8).²⁵ Remarkably, using the more stable iridium catalysts derived from dbcot and dncot,



Scheme 1.8 Air Stable Iridium Catalysts Derived from Other Diene Ligands.

regioselectivity was considerably improved for substrates which showed incomplete regioselectivity from the cod complex.

In 2009, Hartwig reported an iridium catalyzed enantioselective *N*-allylation of imidazole, benzimidazole, and purine heterocycles (Scheme 1.9).²⁶ Using an ethylene ligated cyclometalated iridium complex that was proven effective for such transformations,⁸ high yields and good to excellent levels of regio- and enantioselectivities were observed. Notably, the process was deployed for the formal synthesis of a JNK3 inhibitor, which represents the utility of enantioselective allylic amination in synthesis applications. This method was later extended to the use of indoles.²⁷ In general, the C3 position of indole is the most reactive site for electrophilic aromatic substitution, which poses intrinsic challenges to undertake *N*-substitution due to largely weak acidity of N-H. In this process, installing electron withdrawing groups on the indole core increases the acidity of N-H, enabling enantioselective *N*-allylation of indoles with high regio- and enantioselectivity, although incomplete regioselectivity of alkyl-substituted allylic carbonates was observed.



Scheme 1.9 Iridium Catalyzed Enantioselective Allylic Amination with Indoles and Azoles.

In 2012, You developed the synthesis of *N*-allylindoles through a one-pot reaction involving an iridium catalyzed enantioselective allylic amination of indoline, followed by DDQ mediated oxidation (Scheme 1.10).²⁸ Using iridium catalysts, which were proven effective as more stable iridium species,²³ indoline reacts with allylic carbonates and the subsequent oxidation forms α -chiral indole derivatives with excellent levels of regio- and enantioselectivity, although incomplete regioselectivity upon use of alkyl-substituted allylic proelectrophiles was observed. Two years later, to broaden scope of indole derivatives, You disclosed another approach to the synthesis of enantiomerically enriched *N*-allylindoles.²⁹ *N*-allylation of 2-(phenylethynyl)aniline in the presence of an iridium catalyst modified by a chiral phosphoramidite ligand, followed by a gold catalyzed cyclization reaction, led to the enantiopure indole derivatives in good yields with high regio- and enantioselectivities.



Scheme 1.10 Iridium Catalyzed Enantioselective Allylic Aminations to Form α -Chiral Indole Derivatives.



Scheme 1.11 Sequential Allylic C-H Functionalization and Enantioselective Allylic Amination.

In 2013, Hartwig reported a sequence of two catalytic processes: allylic C–H functionalization and allylic amination (Scheme 1.11).³⁰ Using 4,5-diazafluorenone as a ligand, which reportedly promotes allylic C-H acetoxylation,³¹ and using *tert*-butyl perbenzoate as the oxidant and source of the benzoate group, the linear allylic ester was formed with a high linear-to-branched ratio. Subsequent allylic amination provides α -branched chiral allylic amines using both aliphatic and aromatic amines, and azoles.



Scheme 1.12 Regio- and Enantioselective Synthesis of *N*-Substituted 2-Pyridones.

In 2015, You reported an enantioselective allylic amination of 2-pyridones (Scheme 1.12).³² Despite competition between O- and N-alkylation when using iridium catalysts modified by an Alexakis ligand, high levels of regio- and enantioselectivity were achieved for N-allylated 2-pyridones.

In 2016, You developed iridium catalyzed enantioselective allylic aminations with a *N*-aryl phosphoramidite ligand (Scheme 1.13).³³ In general, iridium catalysts modified by a Feringa type ligand provide excellent regio- and enantioselectivity for a variety of substrates, although *ortho*-substituted phenyl allylic proelectrophiles were found to lead to a decrease in either yield or regio- and enantioselectivity in many cases. To address this issue, a series of *N*-aryl phosphoramidite ligands were identified that provided superior results to previously reported asymmetric allylic aminations of *ortho*-substituted cinnamyl carbonates. Presumably, the cyclometalated iridium complex formed through C-H (indicated in Scheme 1.13) activation of the ligand **1.50** is less sterically hindered in comparison with those derived from a Feringa type ligand, leading to high regio- and enantioselective control of diverse substrates containing sterically congested nucleophiles.



Scheme 1.13 Enantioselective Allylic Amination with *N*-Aryl Phosphoramidite Ligands.

1.2.2 Allylic Ester

In 2015, Evans reported a highly regio- and enantioselective allylic amination of the sulfur stabilized aza-ylide, diphenylsulfilimine (Scheme 1.14).³⁴ An iridium complex modified by ligand **1.53** catalyzed the formation of chiral allylic sulfilimines, which readily undergo cleavage with acid to afford enantiomerically enriched primary amine hydrochloride salts. Whereas carbonate leaving groups are generally used with aryl-substituted allylic proelectrophiles for high regio- and enantioselectivity, allylic benzoate electrophiles enforce both improved reactivity and selectivity for alkyl substituents.



Scheme 1.14 Enantioselective Iridium Catalyzed Allylic Aminations of Diphenylsulfilimine.

In 2018, the Krische group found that neutral π -allyliridium *C*,*O*-benzoates modified by SEGPHOS, which are well-known to catalyze nucleophilic carbonyl allylations,³⁵ promote highly regio- and enantioselective allylic aminations of branched allylic acetates bearing linear alkyl groups using primary aliphatic amines (Scheme 1.15).³⁶ Whereas the iridium phosphoramidite catalyzed allylic substitutions occur by way of cationic π -allylmetal intermediates, these π -allyliridium-*C*,*O*-benzoates react by



Scheme 1.15 Enantioselective Iridium Catalyzed Electrophilic Allylations of Alkyl-Substituted Allylic Acetates.

way of neutral π -allylmetal intermediates, which may account for their amphiphilic character. Notably, these aminations enable complete branched regioselectivity, overcoming a significant limitation associated with previously reported iridiumphosphoramidite catalysts, which display incomplete regioselectivity for alkyl-substituted allylic proelectrophiles. A year later, Krische showed that the corresponding tol-BINAPmodified iridium catalyst provides a significant expansion in scope, enabling highly enantioselective aminations of branched alkyl-substituted allylic acetates with electronically diverse primary and secondary aryl amines, including site-selective reactions of bis(amine) nucleophiles.³⁷ Mechanistic studies involving amination of the enantiomerically enriched, deuterium labeled acetate corroborates C-N bond formation *via* outer-sphere addition. Remarkably, the tol-BINAP-modified iridium *C,O*-benzoates catalyze highly enantioselective *N*-allylations of indoles and related azoles.³⁸ This reaction complements previously reported metal catalyzed indole allylations, enabling complete levels of *N*- *versus* C3-, and branched *versus* linear regioselectivity.

1.2.3 Allylic Alcohol

Whereas significant progress has been made in the formation of chiral allylic amines *via* iridium catalyzed allylic amination of allylic carbonate and ester, the use of allylic alcohols remains a challenge. Generally, these allylic proelectrophiles are made from the corresponding allylic alcohol; hence, direct use of allylic alcohols could streamline synthetic sequences. In 2007, Hartwig reported an enantioselective allylic amination with allylic alcohols (Scheme 1.16).³⁹ Using niobium ethoxide or triphenyl borane as an activator of the allylic alcohol, iridium catalysts modified by the ligand **1.17** formed branched allylic amines with high regio- and enantioselectivity, using primary and secondary aliphatic amines, and aryl amines.



Scheme 1.16 Enantioselective Allylic Aminations of Allylic Alcohols Activated by Lewis Acids.

A new approach to allylic amination was discovered in 2007 by the Carreira group.⁴⁰ Branched allylic alcohols have excellent reactivity under acidic conditions. With the use of iridium complex modified by phosphine-olefin ligands, branched allylic amines were formed regioselectively. Compared to iridium-phosphoramidite systems derived

from a Feringa type ligand, the η^3 -allyliridium complex derived from two phosphineolefin ligands is more electrophilic in character, which enables highly facile electrophilic allylic substitution. Subsequent to these efforts, an enantioselective allylic amination of racemic secondary allylic alcohols was achieved in 2012 by Carreira (Scheme 1.17).⁴¹ Chirally modified phosphine-olefin ligands were used, which allow the reaction to occur in a non-stereospecific fashion, probably due to displacement of the π -bond of (σ + π)-allyl (enyl) iridium intermediates by the tethered olefin of the phosphoramidite ligand. A diverse range of protected chiral allylic amines, all with high regio- and enantioselectivity, was observed. In 2019, Carreira demonstrated highly reactive or unstable nucleophiles such as hydrazines or *N*-hydroxylamines also participated in the enantioselective allylic amination reactions.⁴² While numerous transition metal catalyzed reactions require anhydrous systems to maintain their efficiency, utilizing commercially available aqueous solutions of these unstable molecules as nucleophiles is a significant advance.



Scheme 1.17 Enantioselective Allylic Aminations by Iridium Complex Modified by Chiral Phosphine-Olefin Ligands.

1.2.4 Allylic Phosphate

Whereas the pioneering reactions of enantioselective allylic substitution to form chiral branched allylic amines was catalyzed by an iridium-phosphoramidite complex, Takemoto applied a chiral Pybox ligand to the enantioselective allylic aminations of allylic phosphates in 2004 (Scheme 1.18).⁴³ Using an iridium catalyst modified by a chiral Pybox ligand, the chiral branched product was isolated with good yield and enantioselectivity. Notably, 1-naphthyl-substituted allylic proelectrophile could improve regioselectivity up to >95:5 branched to linear ratio. This method was later extended to the use of guanidines.⁴⁴ With electron withdrawing groups on the guanidines, products of monoallylation were observed in good yield with high enantioselectivity, albeit with lower levels of regiocontrol.



Scheme 1.18 Enantioselective Allylic Aminations of Allylic Phosphate Proelectrophiles with Iridium Pybox Complexes.

In 2011, You and Helmchen reported enantioselective allylic aminations of allylic phosphates with *ortho*-amino styrenes (Scheme 1.19).⁴⁵ Due to the highly electrophilic π -allyl iridium complex, sequential allylic vinylation and amination reactions were inevitable. To suppress the undesired pathway, the allylic phosphate was exploited, which led to exclusive *N*-allylation of amino styrenes with excellent levels of chemo-, regio-, and enantioselectivity. Notably, the amination product was transformed into 1,2-dihydroquinolines, which are a structural motif found in numerous bioactive molecules.



Scheme 1.19 Iridium Catalyzed Enantioselective Allylic Aminations with Allylic Phosphates.
1.3 Intermolecular Rhodium Catalyzed Enantioselective Allylic Amination

1.3.1 Allylic Derivatives

In 2012, Nguyen reported rhodium catalyzed regio- and enantioselective aminations of racemic tertiary allylic trichloroacetimidates with aryl amines *via* dynamic kinetic asymmetric transformation (DYKAT) (Scheme 1.20).⁴⁶ Whereas rhodium catalyzed allylic substitutions of allylic derivatives to form chiral allylic amines have largely focused on stereospecific processes⁴⁷ or kinetic resolution⁴⁸, a rhodium catalyst modified by a chiral diene ligand allows the formation of branched chiral allylic amines using racemic secondary allylic carbonates and a wide range of aniline nucleophiles.⁴⁹⁻⁵² DYKAT of tertiary allylic trichloroacetimidates promotes ionization of the allylic proelectrophile to form diasteromeric π -allylrhodium species. The matched π -allylrhodium intermediates enable facile substitution by anilines to form the preferred enantiomer of the product whereas a chirally modified rhodium complex could decrease the rate of nucleophilic attack by anilines, allowing rapid π - σ - π interconversion.



Scheme 1.20 Rhodium Catalyzed Formation of Enantiomerically Enriched Allylic Amines Bearing α -Tetrasubstituted Tertiary Carbons.

In 2016, Breit reported rhodium catalyzed dynamic kinetic asymmetric allylations of 2-hydroxypyridines with alkyl-substituted allylic carbonates (Scheme 1.21).⁵³ The combination of a rhodium catalyst with chiral DTBM-SEGPHOS ligand led to excellent levels of asymmetric induction for the *N*-allylation of 2-hydroxypyridines. The collective data corroborate a mechanism involving rhodium(I)-mediated allylic carbonate oxidative addition, which operates through a formal S_N2' mechanism to form a σ -allylrhodium complex. Releasing methanol and CO₂, the phenolate bound π -allylrhodium complex undergoes reductive elimination to form the *N*-selective allylation product. A year later, this was extended to the use of quinazolinones.⁵⁴ Using a rhodium catalyst modified by DTB-MeO-BIPHEP, quinazolinones reacted with both alkyl- and aryl-substituted allylic carbonates to form enantiomerically enriched allylic amines in good yields with high regio- and enantioselectivity, which was applied in the enantioselective formal synthesis of (-)-chaetominine.



Scheme 1.21 Enantioselective Synthesis of Branched Allylic Amines *via* Rhodium Catalyzed Allylic Substitution.

1.3.2 C-H Functionalization

Formation of α -chiral allylic amines *via* enantioselective hydroamination is a highly attractive and atom economical approach. In 2012, Breit reported an enantioselective rhodium catalyzed synthesis of chiral allylic amines (Scheme 1.22).⁵⁵ Using a rhodium catalyst modified the Josiphos ligand **1.94**, cyclohexylallene reacts with a variety of anilines to form chiral allylic amines. The proposed mechanism involves a rhodium(I)-mediated aniline oxidative addition followed by reversible hydrometallation of the terminal allene double bond to form a π -allylrhodium complex, which upon reductive elimination releases the branched allylic amine. This method was later extended to the use of imidazoles,⁵⁶ 2-pyridones,⁵⁷ 4-pyridones,⁵⁸ pyrazoles,⁵⁹ purines,⁶⁰ tetrazoles,⁶¹ 2-aminothiazoles,⁶² and pyridazineones,⁶³ leading to regio-, chemo- and enantioselective *N*-allylation of these heterocycles, which are all useful building blocks for natural products and pharmaceuticals (Scheme 1.23).



Scheme 1.22 Enantioselective Synthesis of Branched Allylic Amines *via* Rhodium Catalyzed Hydroamination.



Scheme 1.23 Rhodium Catalyzed Enantioselective Synthesis of Branched Allylic Amines Using a Variety of Nucleophiles.

In 2016, Breit developed a one-pot procedure for an enantioselective synthesis of indole derivatives (Scheme 1.24).⁶⁴ Using a rhodium catalyst modified by DTBM-SEGPHOS, various aryl hydrazines were coupled with terminal allenes, generating selective *N*-allylated hydrazines. The more acidic N-H bond at N1 compared to N2 promotes facile oxidative addition to the rhodium complex, which could differentiate nitrogen selectivity. Products through regio- and enantioselective allylation of aryl hydrazines were subjected to Fischer indolization, allowing the construction of α -chiral indole derivatives with high yields and enantioselectivity.



Scheme 1.24 Sequential Allylic C-H Functionalization and Fischer Indole Synthesis.

On the basis of Breit's prior observation that alkyne could serve as precursors to form π -allylrhodium species,⁶⁵ a rhodium catalyzed enantioselective coupling of indolines with internal alkyne using a rhodium catalyst modified by JoSPOphos was reported by Dong in 2015 (Scheme 1.25).⁶⁶ An allene intermediate was formed *via* rhodium hydride mediated alkyne isomerization, which upon hydroamination releases allylic amines instead of the enamine or imine product which are competing side products of alkyne mediated hydroamination. It was found that acidic additives play a key role to afford high regioselectivity of branched allylic amines in good yields with high enantioselectivity. Two years later, Dong disclosed a method for the hydrofunctionalization of 1,3-dienes

with indolines.⁶⁷ It was found that conjugated dienes could generate analogous π allylrhodium complexes which are typically generated by hydrometallation of an allene or alkyne. Using a rhodium catalyst modified by JoSPOphos, the enantioselective coupling of indolines with 1,3-dienes afforded *N*-allylated indolines with good yields along with high regio- and enantioselectivity was reported.



Scheme 1.25 Rhodium Catalyzed Hydrofunctionalization to Form Chiral Allylic Amines.

In 2016, Breit developed a regio- and enantioselective allylic amination of pyrazoles with internal alkynes in an atom-economic manner using a rhodium catalyst modified by JosPOphos (Scheme 1.26).⁶⁸ The reaction displays a broad substrate range of substituted pyrazoles and alkynes to provide chiral pyrazole derivatives in good yields along with good regio- and enantioselectivities. To note, both allenes and terminal alkynes also participate in the enantioselective synthesis of branched allylic amines which represents a highly flexible approach *via* an interconverting π -allyl and σ -allyl haptomers from allenes, internal and terminal alkynes. In a similar manner, Breit

expanded the scope of internal alkyne with triazoles while the synthesis of *N*-allylated trizoles in combination with the desired *N*-selectivity, high branched regio- and enantioselectivity remains uncommon. Using a modified rhodium catalyst with JoSPOphos ligand, good yields, and modest to good level of regio- and enantioselectivity were observed.⁶⁹



Scheme 1.26 Rhodium Catalyzed Hydrofunctionalization to Form Chiral Allylic Amines *via* Alkyne Isomerization.

1.4 Intermolecular Palladium Catalyzed Enantioselective Allylic Amination

While numerous iridium and rhodium catalyzed allylic aminations of allylic proelectrophiles with one terminal substituent to form branched products with excellent regio- and enantioselectivity have been documented, palladium catalyzed allylic amination forming chiral branched products remains less developed. Here, we mainly summarized examples of palladium catalyzed allylic amination forming branched products *via* unsymmetrical π -allyl palladium species.

1.4.1 Allylic Epoxide, Aziridine, Cyclic Carbonate and Carbamate

In 2001, Trost performed a dynamic kinetic asymmetric transformation (DYKAT) with commercially available butadiene monoxide (Scheme 1.27).⁷⁰ Using a palladium catalyst modified by a Trost type ligand, vinylglycinol products were formed in high yields with excellent regio- and enantioselectivity. Coordination of the pronucleophile to the oxygen leaving group via intramolecular hydrogen bond would help to deliver the nucleophile to the adjacent carbon. This allows high levels of the regioselective branched product despite the regio-control problem since nucleophilic attack in such systems are normally favored at the less substituted carbon.⁷¹ This method was later extended to use of a variety of nucleophiles including hydrazines, hydroxylamines, isatin derivatives, and purines (not shown).⁷²⁻⁷⁴ Taking advantage of ring strain to facilitate ring opening, the analogous vinyl aziridines were employed for hetereocycles bearing chiral 1,2-diamines using biologically active pyrroles and indoles.⁷⁵ In these processes, the amide anion leaving group in the π -allylpalladium intermediate was sufficiently basic enough to deprotonate the N-H from the nucleophiles and facilitate the transformation. N-allylated pyrroles and indoles were observed in good yields with high regio- and enantioselectivity, although a strong electron withdrawing group was necessary to give high levels of both reactivity and enantioselectivity.



Scheme 1.27 Palladium Catalyzed Dynamic Kinetic Asymmetric Transformation of Butadiene Monoxide and Vinyl Aziridine.

In 2003, Trost disclosed the asymmetric cycloaddition of isocyanates to vinyl aziridines (Scheme 1.28).⁷⁶ High yields and enantioselectivities were obtained for a broad array of imidazolidin-2-ones upon use of a palladium catalyst modified by a Trost ligand. To note, acidic additives are required to achieve high enantioselectivity since protonating the nitrogen upon opening of the aziridine which slows the cyclization rate. This allows π - σ - π interconversion of the diastereomeric π -allyl palladium intermediates, which can ultimately compete with product formation. In a similar manner, Zhang recently developed an efficient method for the enantioselective construction of vinylglycinol derivatives bearing α -tetrasubstituted tertiary carbons through decarboxylative cycloaddition of vinylethylene carbonates with isocyanates.⁷⁷ Using a palladium catalyst modified by SEGPHOS, good yields with good to excellent levels of enantioselectivity were observed.



Scheme 1.28 Palladium Catalyzed Dynamic Kinetic Asymmetric Cycloadditions of Isocyanates.

In 2006, Trost reported a palladium catalyzed asymmetric allylic alkylation of *meso-* and *dl*-divinylethylene carbonates (Scheme 1.29).⁷⁸ Using a palladium catalyst modified by a Trost ligand, only the *syn*-diastereomer was formed in good yields and high enantioselectivity, which was unexpected based on the well-established double inversion mechanism.⁷⁹ Due to the Curtin-Hammett type effects where the rapid and reversible matched ionization is unproductive, mismatched ionization followed by rapid π - σ - π interconversion to relieve steric repulsion between the substrate and asymmetric palladium complex effectively forms the unexpected diastereomer.



Scheme 1.29 Palladium Catalyzed Enantioselective Allylic Aminations of Divinylethylene Carbonate.

In 2013, Liu and Zhang found an efficient palladium catalyzed allylic amination of vinyl cyclic carbonates followed by the cleavage of phthalimide group, forming a series of chiral β -aryl- α , β -unsaturated amino alcohols (Scheme 1.30).⁸⁰ Using a palladium catalyst modified by a planar chiral ferrocene-based phosphinooxazoline (PHOX) ligand, products with good yields and high regio- and enantioselectivity were shown. In 2016, Kleij reported a palladium catalyzed regio- and enantioselective synthesis of allylic aryl amines bearing α -tetrasubstituted tertiary carbons using vinyl cyclic carbonates.⁸¹ The combination of a palladium catalyst and a Feringa chiral phosphoramidite ligand enabled DYKATs where the π - σ - π interconversion occurs faster than the subsequent nucleophilic attack. While it is generally known that palladium catalyzed allylic substitution undergoes through an outer sphere process which leads to formation of linear products, it is demonstrated density functional theory (DFT) calculations and mechanistic studies corroborate chelation assisted amine nucleophilic attack *via* inner sphere process which accounts for experimentally observed branched high regio- and enantioselectivity.⁸²



Scheme 1.30 Palladium Catalyzed Enantioselective Allylic Aminations of Vinylic Cyclic Carbonate.

In 2017, Lu reported a palladium catalyzed enantioselective allylic amination of vinyl benzoxazinones (Scheme 1.31).⁸³ While the branched products in such substrates were usually generated with limited enantiocontrol, presumably due to the steric repulsion between the ortho substituent and the metal complex,^{5,14,23,26,33} this protocol using the directing effect of the hydrogen bond presented chiral allylic amines with a variety of aliphatic amines in good yields with excellent regio- and enantioselectivty.



Scheme 1.31 Palladium Catalyzed Enantioselective Allylic Amination of Vinyl Benzoxazinones.

1.4.2 Allylic Ester and Carbonate

In 2001, Hou and Dai reported a palladium catalyzed allylic amination of monosubstituted allylic acetates (Scheme 1.32).⁸⁴ Employing a palladium catalyst modified by the ferrocene *P*,*N*-ligand **1.145**, good yields with high regio- and enantioselectivity were observed. In the amination reaction, a hydrogen bond between the amine and the free OH in the ligand might direct intramolecular nucleophilic attack, leading to high levels of regioselective ratio of branched to linear products. It was later found that the ferrocene modified palladium complex was highly efficient for enantioselective allylic aminations of both alkyl- and aryl-substituted dienyl esters.⁸⁵ In a similar manner, using a ferrocene based ligands SIOCPhos, Hou recently reported regio- and enantioselective allylic aminations of mono-substituted allylic proelectrophiles with *N*-Boc-*O*- methylhydroxylamine (not shown).⁸⁶ In 2017, Kleij developed a palladium catalyzed allylic amination of allylic proelectrophiles and aliphatic amines, providing allylic amines bearing α -tetrasubstituted tertiary carbons with good regio- and enantioselectivity.⁸⁷



Scheme 1.32 Palladium Catalyzed Enantioselective Allylic Aminations of Allylic Proelectrophiles.

While the formation of branched products at the sterically more hindered position of allylic substrates is challenging in palladium catalyzed transformations, a palladium catalyzed enantioselective allylic amination of acyclic Morita-Baylis-Hillman (MBH) adducts using a variety of anilines as nucleophiles was reported in 2012 by Liu, Wang and Ding (Scheme 1.33).⁸⁸ Using a palladium catalyst modified by a spiroketal-based bisphosphine ligand, optically active β -arylamino acid esters were formed in good yields with high regio- and enantioselectivities. Further mechanistic investigation revealed that the ligand plays a bifunctional role in the process: one P atom forms a carbon-phosphine σ -bond with the terminal carbon atom of the allyl moiety, and the other P atom coordinates to palladium, leading to an uncommon regioselective formation of the branched product.⁸⁹ This method was later extended to other *N*-heterocyclic nucleophiles, including imidazoles, triazoles and purines, with alkyl-substituted MBH propionate species.⁹⁰



Scheme 1.33 Palladium Catalyzed Enantioselective Allylic Aminations of Racemic Morita-Baylis-Hillman Adducts.

In 2017, Liu and Zhang reported a synthesis of chiral α,β -unsaturated γ -amino esters *via* enantioselective allylic amination reactions (Scheme 1.34).⁹¹ Using a palladium catalyst modified by DTBM-SEGPHOS, high yields with excellent regio- and

enantioselectivities were observed. Notably, control experiments found that the regioselectivity of aminated products might depend on steric hindrance on the right side of the allyl substrates.



Scheme 1.34 A Palladium Catalyzed Synthesis of Chiral α,β -Unsaturated γ -Amino Esters.

In 2016, Diaz and Castillon reported a palladium catalyzed asymmetric allylic amination of linear allylic carbonates (Scheme 1.35).⁹² Using a palladium catalyst modified by a Trost ligand, optically active vinylglycines were observed in good yields with high regio- and enantioselectivities. In this process, the excellent control of regioselectivity was shown to be due to hydrogen bonding interactions between the hydroxyl group in the substrate and the DACH-naphthyl ligand in the palladium complex, as absence of the hydroxyl group led to the opposite regioselectivity. In 2019, Zhang reported a method for the synthesis of *N*-substituted 2-pyridones *via* a palladium catalyzed regio- and enantioselective allylic amination of hydroxyl-containing allylic carbonates.⁹³ Complete levels of chemo-, regio- and enantioselectivities were observed upon use of a palladium complex modified by a chiral phosphoramidite ligand. Control experiments determined that the hydrogen bond interaction between a π -allylpalladium

intermediate and 2-hydroxypyridine played a crucial role in resultant high levels of branched regioselectivity accompanied by high yields.



Scheme 1.35 Palladium Catalyzed Regio- and Enantioselective Allylic Aminations of 2-Pyridones.

1.4.3 Allylic Alcohol

In 2014, Beller disclosed an enantioselective allylic amination of racemic allylic alcohols (Scheme 1.36).⁹⁴ A palladium catalyst modified by a phosphoramidite ligand in combination with a chiral phosphoric acid promoted formation of a variety of functionalized amines in high yields with complete levels of regio- and enantioselectivities, using racemic cyclic and acyclic alkyl-substituted allylic alcohols. Mechanistic studies involving amination of the deuterium labeled allylic alcohol found that the rate of σ - π - σ interconversion of allylic palladium intermediates is slower than the rate of a nucleophilic attack, which enables the highly regioselective formation of branched allylic amines.



Scheme 1.36 Cooperative Catalysis by Palladium and a Chiral Phosphoric Acid Promoted Regio- and Enantioselective Allylic Aminations.

1.4.4 C-H Functionalization

In 2001, Hartwig reported a palladium catalyzed hydroamination of cyclohexadiene with arylamines using a palladium catalyst modified by a Trost ligand (Scheme 1.37).⁹⁵ Good yields with high enantioselectivities were achieved although amine nucleophile scopes were limited to aryl amines. Mechanistic studies later showed that hydropalladation of the cyclohexadiene generates a π -allylpalladium complex followed by outer sphere amine addition to form the resulting ammonium salts which oxidatively protonate palladium to regenerate palladium hydride species.⁹⁶ In 2017. Malcolmson applied a PHOX ligand in palladium catalyzed allylic aminations of acyclic 1,3-dienes.⁹⁷ Using aliphatic amines, good yields of chiral allylic amines with excellent levels of regio- and enantioselectivity were reported. Interestingly, the electron deficient PHOX ligand was required for high regiomeric ratios of chiral branched to achiral linear products. Subsequent to these efforts, Malcolmson recently disclosed the development of palladium catalyzed hydroaminations of 1,4-disubstituted acyclic dienes with both amines.98 aliphatic and aromatic Whereas enantioselective intermolecular hydrofunctionalizations of dienes are largely limited to terminal 1,3-dienes, this protocol using palladium complexes with a noncoordinating BAr^F₄ counteranion promoted the

hydroamination of internal 1,3-dienes to form chiral allylic amines in good yields along with high levels of regio- and enantioselectivity.



Scheme 1.37 Palladium Catalyzed Enantioselective Hydroamination of 1,3-Dienes.

In 2012, Rhee reported a palladium catalyzed intermolecular hydroamination of alkoxyallenes (Scheme 1.38).⁹⁹ In an extension of their reported work,¹⁰⁰ using a palladium catalyst modified by a Trost ligand, hydroamination products were observed in excellent yield and enantioselectivity, which upon a subsequent ring-closing metathesis reaction formed allylic *N*,*O*-acetals.



Scheme 1.38 Palladium Catalyzed Regio- and Enantioselective Hydroaminations of Terminal Allenes.

Recently, Wang and Ding found an enantioselective alkoxycarbonylative amination of terminal allenes to form β -arylamino acid esters in good yields with high regio- and enantioselectivities.¹⁰¹ On the basis of their prior work (Scheme 1.33),⁸⁸ using a palladium catalyst modified by an aromatic spiroketal based diphosphine ligand, the corresponding phosphonium palladium intermediate **IV** was generated, presumably *via* methoxy carbonylpalladation of the allene and subsequent intramolecular rearrangement. The rest of the catalytic cycles proceeded in the presence of a copper salt as an oxidant. This protocol would effectively promote straightforward access to chiral α -methylene- β -arylamino acid esters, avoiding the tedious synthesis of MBH adducts.

1.5 Intermolecular Ruthenium Catalyzed Enantioselective Allylic Amination

Whereas transition metal catalyzed enantioselective allylic aminations have been developed largely by effective palladium and iridium catalysts, such transformations catalyzed by ruthenium catalysts remain uncommon. In 2001, Takahashi reported the first example of enantioselective allylic aminations catalyzed by the planar-chiral cyclopentadienyl ruthenium complex **1.185**, allowing the formation of branched products with high regioselectivity and up to 74% enantioselectivity (Scheme 1.39).¹⁰² Of note, it is the pioneering work toward asymmetric induction by planar-chiral complexes of late transition metals. Subsequent to this work, the same planar-chiral ruthenium complex was used for asymmetric auto tandem catalysis.¹⁰³ Regio- and enantioselective allylic substitution reactions of mono-substituted allylic halides with carboxamides produced enantiomerically enriched branched allylic amines in good yields with excellent levels of regio- and enantioselectivity. Further investigation found that the allylic substitution reaction followed by atom transfer radical cyclization formed optically active γ -lactams in a one pot process. In 2014, Kawatsura and Itoh reported regio- and enantioselective

allylic aminations of racemic mono-substituted allylic esters with cyclic secondary amines.¹⁰⁴ Using a ruthenium catalyst modified by (S,S)-*ip*-Pybox, enantiomerically enriched allylic amines with excellent regio- and enantioselectivities were formed.



Scheme 1.39 Ruthenium Catalyzed Enantioselective Allylic Aminations.

1.6 Conclusion

Enantioselective late transition metal catalyzed allylic aminations have become a widely applicable reaction for the synthesis of enantiomerically enriched allylic amines, which are versatile building blocks for natural products, semisynthetic and synthetic pharmaceutical and agrochemical ingredients. The improvement of ligand design in the catalyst has allowed a wide range of amine nucleophiles with mono- and di-substituted allylic electrophiles. As illustrated in this review, late transition metal catalyst has been proven efficient for the formation of highly regio- and enantioselective allylic amines with good functional group tolerance, which have been applied in asymmetric total synthesis of a variety of complex chiral molecules. Like any other synthetic methodology though, there are still some limitations to the use of late transition metal catalyzed allylic amination. For example, the use of tri-substituted allylic electrophiles and alkyl-substituted allylic derivatives remains challenging. Thus, to address these issues, future developments will be expected to provide more unforeseen aspects for late transition metal catalyzed enantioselective allylic aminations.

Chapter 2: Asymmetric Allylation of Glycidols Mediated by Allyl Acetate *via* Iridium Catalyzed Hydrogen Transfer*

2.1 Introduction

Epoxides are important building blocks in chemical synthesis, including the construction of polyketide natural products where they also appear as native structural motifs.¹ Accordingly, several methods have been reported for the asymmetric allylation² of glycidic aldehydes using reagents based on boron,^{3,4} tin,^{5,6} silicon,^{7,8} indium,⁹ and magnesium.¹⁰ These methods have proven effective in certain contexts;³⁻¹⁰ however, due to pronounced match-mismatch effects, only one diastereomer of the secondary homoallylic glycidol is generally accessible in highly diastereomerically enriched form.¹¹ Additionally, indirect formation of secondary homoallylic glycidols via enantioselective allylation of α,β -unsaturated aldehydes followed by Sharpless asymmetric epoxidation is problematic, as modest diastereoselectivities are evident in reactions of secondary (Z)allylic alcohols.¹² By harnessing the native reducing ability of alcohols, we have discovered a new, redox-economic class of C-C bond formations that merge the characteristics of carbonyl addition and transfer hydrogenation.¹³ These hydrogen auto transfer processes utilize alcohol oxidation to drive reductive generation of transient organometallic nucleophiles. The resulting carbonyl-organometal pair combines to furnish products of addition, directly converting lower alcohols to higher alcohols. Based on this pattern of reactivity, diverse enantioselective alcohol C-H functionalizations have been developed including the C-H allylation¹⁴ and crotylation^{15,16} of primary alcohols to

^{*}This chapter is based on the published work:

Kim, S. W.; Lee, W.; Krische, M. J. Org. Lett. 2017, 19, 1252.

form secondary homoallylic alcohols. In the present account, this allylation method is applied to the conversion of primary glycidols to secondary homoallylic glycidols.

2.2 Reaction Development and Scope

Initial studies were focused on the allylation of glycidol 2.1a, which is prepared through Sharpless asymmetric epoxidation of geraniol.¹⁷ For the sake of convenience, the cyclometalated π -allyliridium C,O-benzoate catalysts were generated in situ from commercial components. Thus, glycidol 2.1a was exposed to allyl acetate in the presence of [Ir(cod)Cl]₂, a series of 4-substituted 3-nitro-benzoic acids, assorted axially chiral chelating phosphine ligands, and various inorganic bases. A small set of optimization experiments quickly led to the identification of effective conditions. Thus, using the iridium catalysts (R)- or (S)-Ir-I modified by SEGPHOS, the diastereomeric secondary homoallylic glycidols 2.2a and epi-2.2a were obtained in excellent yields and diastereoselectivities, respectively (Table 2.1). Notably, unlike the corresponding allylborations,¹¹ the enantiomeric catalysts delivered 2.2a and epi-2.2a with roughly equivalent levels of catalyst directed diastereoselectivity (2.2a/epi-2.2a = 11:1 vs 1:12), suggesting the present iridium catalysts are insensitive to match-mismatch effects. Indeed, using an achiral iridium catalyst modified by dppf, a 1:1 mixture of diastereomeric secondary homoallylic glycidols 2.2b and epi-2.2b was obtained. The identification of favorable conditions for the asymmetric C-H allylation of glycidol 2.1a prompted a more detailed investigation into the scope of this process (Table 2.1). *cis*-Glycidols 2.1b-2.1e were specifically selected for study as the corresponding secondary homoallylic glycidols 2.2b-2.2e and epi-2.2b-epi-2.2e are inaccessible using conventional allylation methods²⁻¹⁰ due to the modest diastereoselectivities reported in Sharpless asymmetric epoxidations of secondary (Z)-allylic $alcohols^{12}$ along with the

stereochemically labile nature of (Z)- α,β -unsaturated aldehydes. In the event, *cis*-glycidols **2.1b–2.1e** were converted to the secondary homoallylic glycidols **2.2b–2.2e** and *epi-2.2b–epi-2.2e*, respectively, in good yields and good levels of catalyst-directed diastereoselectivity. In certain cases, the iridium catalysts (*R*)- or (*S*)-Ir-II modified by Cl,MeO-BIPHEP were found to enforce higher yields and diastereoselectivities than the iridium catalysts (*R*)- or (*S*)-Ir-I modified by SEGPHOS. To complete this study, the allylation of *trans*-glycidol **2.1f** was explored. Whereas secondary homoallylic glycidol **2.2f** was formed with excellent levels of catalyst-directed diastereoselectivity, the isomeric glycidol *epi-2.2f* was formed in low yield with less pronounced levels of stereocontrol. These data suggest match–mismatch effects may be more important in reactions of *trans*-glycidols. Finally, it is worth noting that due to the Horeau principle, all major reaction products derived from **2.1a**, **2.1b**, and **2.1f**, are obtained in >99% enantiomeric excess.¹⁸

Table 2.1Iridium catalyzed C-H allylation of glycidols 2.1a-2.2f to form adducts2.2a-2.2f and epi-2.2a-2.2f.a



^aCited yields are of material isolated by silica gel chromatography. ArCO₂H refers to 4-cyano-3nitrobenzoic acid. Diastereomeric ratios were determined by ¹H NMR of crude reaction mixtures. See Supporting Information for further experimental details. ^b1 mmol scale.of material isolated by silica gel chromatography. All reactions were conducted using enantiomerically enriched starting materials. See Supporting Information for further experimental details.



2.3 Discussion

The direct C–H allylation of primary glycidols **2.1a–2.1e** is both step-economic and redox-economic as it avoids discrete formation and isolation of less tractable glycidic aldehydes. Nevertheless, under certain circumstances it may be desirable to conduct the allylation from the aldehyde oxidation level. To assess the feasibility of utilizing glycidic aldehydes as reactants, the reductive coupling of allyl acetate with *dehydro-***2.1a** was performed using 2-propanol as the terminal reductant under otherwise standard conditions (eq 2.1). The respective secondary homoallylic glycidols **2.2a** and *epi-***2.2a** were formed in good yields with excellent levels of catalyst-directed diastereoselectivity. The efficiencies observed in the reactions of glycidic aldehyde *dehydro-***2.1a** were roughly equivalent to those observed using the corresponding glycidol **2.1a** (Table 2.1). To illustrate how the present method may be applied to polyketide construction, secondary homoallylic glycidol **2.2c** was subjected to conditions for regioselective epoxide ring opening using AlMe₃-*n*-BuLi (eq 2.2).¹⁹ The desired adduct **2.3c**, which was obtained in 77% yield, embodies a propionate-based stereotetrad spanning C17–C23 of dictyostatin, a marine macrolide that displays antimitotic activity against multidrugresistant cancer cell lines at nanomolar levels.^{20,21}



2.4 Conclusion

In summary, we report that glycidols prepared through Sharpless asymmetric epoxidation participate in direct carbinol C–H allylation with excellent levels of catalystdirected diastereoselectivity. This method is redox- and step-economic, as it avoids discrete formation of less tractable glycidic aldehydes. Further, this method overcomes limitations evident in corresponding allylations of glycidic aldehydes using allylboron reagents.^{3a,b,11} Finally, as Sharpless asymmetric epoxidations of secondary (*Z*)-allylic alcohols display low levels of diastereoselectivity,¹² indirect formation of the present secondary homoallylic glycidols through an asymmetric enal allylation–epoxidation sequence is not feasible. Future studies will focus on the use of α -olefins²² as pronucleophiles in alcohol-mediated carbonyl addition.

2.5 Experimental Details

General Information

All reactions were run under an atmosphere of argon. Sealed tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were flame dried followed by cooling in a desiccator. Tetrahydrofuran was distilled from sodium-benzophenone immediately prior to use. Dichloromethane was distilled from calcium hydride under a nitrogen atmosphere prior to use. Anhydrous solvents were transferred by oven-dried syringes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynanmic Absorbents F254). Visualization was accomplished with UV light followed by dipping in potassium permanganate or *p*-anisaldehyde stain solution and then heating. Purification of reactions was carried out by flash chromatography using Silacycle silica gel (40-63 µm, unless indicated specifically). Potassium carbonate was purchased through Fisher Chemical, flame dried prior to use, and stored in a desiccator.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer. Highresolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Specific optical rotations were recorded on an Atago AP-300 automatic polarimeter at the sodium line (589 nm) in CHCl₃. Solution concentrations are given in the units of 10^{-2} g mL⁻¹. Accurate masses are reported for the molecular ion (M-H, M, M+H, M+Na, or M+K), or a suitable fragment. ¹H nuclear magnetic resonance spectra were recorded on an Agilent MR (400 MHz). Chemical shifts are reported as parts per million (ppm) relative to residual CHCl₃ $\delta_{\rm H}$ (7.26 ppm). ¹³C nuclear magnetic resonance spectra were recorded on an Agilent MR (100 MHz) for CDCl₃ solutions, and chemical shifts are reported as parts per million (ppm) relative to residual CDCl₃ $\delta_{\rm C}$ (77.0 ppm).

Experimental Details and Spectral Data

Glycidols **2.1a**,^{1,2} *dehydro*-**2.1a**,³ **2.1b**,^{1,4} **2.1c**,⁵ **2.1d**^{4,6} and **2.1e**⁶ were prepared according to the published procedures and were identical in all respects to the reported materials.



To a flame dried round-bottomed flask charged with dry 4 Å molecular sieves (1 g) under an argon atmosphere was added CH₂Cl₂ (33 mL, 0.1 M with respect to allylic alcohol). The reaction vessel was placed in -40 °C bath and Ti(OⁱPr)₄ (1.93 mL, 6.54 mmol, 200 mol%) was added followed by D-(–)-diisopropyl tartrate (1.70 mL, 8.1 mmol, 250 mol%). The mixture was stirred vigorously for 30 minutes, at which point *tert*-butyl hydrogen peroxide (1.2 mL, 5.0-6.0 M in decane, 6.54 mmol, 200 mol%). After 30 minutes a solution of allylic alcohol⁷ (0.8 g, 3.27 mmol) in dry CH₂Cl₂ (17 mL, 0.2 M) was added slowly and the mixture was stirred for 40 h. The reaction vessel was transferred to an ice batch. Water (30 mL) and NaOH (30 mL, 10% aqueous solution) were added to the reaction mixture. The mixture was allowed to stir for 2 hours. The reaction mixture was filtered (celite) with the aid of CH₂Cl₂ (50 mL) and the filtrate was transferred to a separatory funnel. The organic layer was extracted with CH₂Cl₂ (50 mL × 2) and the combined organic layers were washed with brine (100 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-7:1) to furnish the title compound as a colorless oil (690 mg, 2.65 mmol) in 81% yield. The spectral data were identical to those reported for the corresponding racemate.⁷

<u>**TLC (SiO**</u>₂) $R_f = 0.29$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 3.94 (m, 2H), 3.77 (dd, *J* = 11.9, 4.1 Hz, 1H), 3.62 (dq, *J* = 12.2, 3.8 Hz, 1H), 3.13 (m, 2H), 1.05 (m, 21H).

¹³C NMR (100 MHz, CDCl₃): $\delta = 62.8, 61.3, 56.0, 55.7, 17.8, 11.9.$

<u>HRMS</u> (ESI) Calculated for $C_{13}H_{28}O_3Si [M+Na^+] = 283.1700$, Found 283.1701.

<u>FTIR</u> (neat) 3432, 2942, 2865, 1463, 1112, 1065, 995, 881, 779, 681 cm⁻¹.

 $[\alpha]_{D}^{28}$: +16.33 (*c* 1.0, CHCl₃).

HPLC: Enantiomeric excess was determined by HPLC analysis of the benzoate derivative of the product (Chiralcel OD-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 254 nm), $t_{major} = 4.7 \text{ min}, t_{minor} = 5.5 \text{ min}; ee = 93\%$.





Peak I #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
-						
1	4.719	VB	0.1416	1063.94421	115.38085	50.9285
2	5.452	BB	0.1640	1025.14819	96.45261	49.0715
Totals	з:			2089.09241	211.83346	



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	4.668	VB	0.1250	5977.96289	750.03583	96.6786
2	5.495	BB	0.1420	205.37094	22.60398	3.3214



<u>Procedures and Spectral Data for the Synthesis of Secondary Homoallylic Glycidols</u> 2.2a-2.2f, *epi*-2.2a-2.2f

(R)-1-((2R,3R)-3-methyl-3-(4-methylpent-3-en-1-yl)oxiran-2-yl)but-3-en-1-ol (2.2a)



Detailed Procedures

From alcohol oxidation level: An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **1a** (34 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (0.5 mL, 0.4 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–10:1) to furnish the title compound as a colorless oil (35.7 mg, 0.17 mmol, anti:syn = 11:1) in 85% yield. *From alcohol oxidation level:* An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (17 mg, 0.025 mmol, 2.5 mol%), (*R*)-SEGPHOS (30.5 mg, 0.05 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (19.0 mg, 0.1 mmol, 10 mol%), alcohol **1a** (170 mg, 1.0 mmol, 100 mol%) and K₂CO₃ (69.0 mg, 0.5 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (2.5 mL, 0.4 M) and allyl acetate (0.22 mL, 2.0 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1-10:1) to furnish the title compound as a colorless oil (171 mg, 0.17 mmol, anti:syn = 10:1) in 81% yield.

From aldehyde oxidation level: An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), aldehyde *dehydro-***1a** (0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.01 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (0.5 mL, 0.4 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–10:1) to furnish the title compound as a colorless oil (34.1 mg, 0.16 mmol, anti:syn = 14:1) in 81% yield.

Spectral data is reported for the major isomer.

<u>**TLC (SiO₂)**</u> $R_f = 0.62$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.86$ (m, 1H), 5.19 (m, 2H), 5.08 (m, 1H), 3.52 (td, J = 8.1, 3.9 Hz, 1H), 2.66 (d, J = 8.1 Hz, 1H), 2.5 (m, 1H), 2.30 (m, 1H), 2.08 (m, 2H), 1.90 (brs, 1H), 1.66 (m, 4H), 1.59 (m, 3H), 1.47 (m, 1H), 1.35 (s, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 133.6, 132.0, 123.4, 118.7, 68.8, 64.8, 61.4, 39.6, 38.5, 25.6, 23.7, 17.6, 16.5.$ **HRMS**(ESI) Calculated for C₁₃H₂₂O₂ [M+Na⁺] = 233.1512, Found 233.1506.

<u>FTIR</u> (neat) 3435, 2927, 2358, 1436, 1384, 1072, 986, 914, 819 cm⁻¹.

 $[\alpha]_{D}^{25}$: +21.17 (*c* 1.0, CHCl₃)


(S)-1-((2R,3R)-3-methyl-3-(4-methylpent-3-en-1-yl)oxiran-2-yl)but-3-en-1-ol (*epi-*2.2a)



Detailed Procedures

From alcohol oxidation level: An oven-dried pressure tube equipped with a magnetic stir bar was charged with [Ir(cod)Cl]₂ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **1a** (34 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (0.5 mL, 0.4 M) and allyl acetate (43 μ L, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–10:1) to furnish the title compound as a colorless oil (36.1 mg, 0.17 mmol, anti:syn = 1:12) in 86% yield.

From aldehyde oxidation level: An oven-dried pressure tube equipped with a magnetic stir bar was charged with [Ir(cod)Cl]₂ (3.4 mg, 0.005 mmol, 2.5 mol%), (S)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%),

aldehyde *dehydro*-**1a** (0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.01 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (0.5 mL, 0.4 M) and allyl acetate (43 μ L, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–10:1) to furnish the title compound as a colorless oil (34.9 mg, 0.17 mmol, anti:syn = 1:11) in 83% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO</u>₂) $R_f = 0.50$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.82$ (m, 1H), 5.23–5.10 (m, 2H), 5.06 (m, 1H), 3.55 (td, J = 7.5, 5.9 Hz, 1H), 2.74 (d, J = 7.9 Hz, 1H), 2.42–2.25 (m, 3H), 2.08 (m, 2H), 1.69 (m, 4H), 1.60 (m, 3H), 1.41 (m, 1H), 1.29 (s, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 133.4, 132.1, 123.4, 118.1, 69.6, 66.4, 61.9, 38.7, 38.6, 25.6, 23.7, 17.6, 17.3.

<u>**HRMS</u>** (ESI) Calculated for $C_{13}H_{22}O_2$ [M+Na⁺] = 233.1512, Found 233.1514. <u>**FTIR**</u> (neat) 3435, 2924, 2362, 1435, 1383, 1043, 997, 914, 822 cm⁻¹. [α]_D²⁵: -0.33 (*c* 1.0, CHCl₃).</u>



(R)-1-((2R,3S)-3-((benzyloxy)methyl)oxiran-2-yl)but-3-en-1-ol (2.2b)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **1b** (39 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1–3:1) to furnish the title compound as a colorless oil (35.6 mg, 0.15 mmol, anti:syn = 9:1) in 76% yield.

Spectral data is reported for the major isomer.

<u>**TLC (SiO₂)</u>** $R_f = 0.42$ (hexanes/ethyl acetate = 2:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.32 (m, 5H), 5.88 (m, 1H), 5.17 (m, 2H), 4.58 (m, 2H), 3.83 (dd, *J* = 10.9, 6.0 Hz, 1H), 3.67 (dd, *J* = 10.8, 5.7 Hz, 1H), 3.50 (td, *J* = 7.8, 4.8 Hz, 1H), 3.26 (td, *J* = 5.9, 4.3 Hz, 1H), 2.96 (dd, *J* = 8.0, 4.3 Hz, 1H), 2.60 (brs, 1H), 2.49 (m, 1H), 2.38 (m, 1H).

 $\frac{1^{3}C \text{ NMR}}{68.4, 57.9, 54.1, 39.3}$ (100 MHz, CDCl₃): $\delta = 137.2, 133.4, 128.5, 128.0, 127.9, 118.2, 73.6, 68.9, 68.4, 57.9, 54.1, 39.3.$

<u>HRMS</u> (ESI) Calculated for $C_{14}H_{18}O_3$ [M+Na⁺] =257.1148, Found 257.1151.

FTIR (neat) 3454, 2921, 1718, 1453, 1271, 1075, 1028, 918, 698 cm⁻¹.

[α]_D²⁵: -55.34 (*c* 1.0, CHCl₃).



(S)-1-((2R,3S)-3-((benzyloxy)methyl)oxiran-2-yl)but-3-en-1-ol (epi-2.2b)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1b** (39 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1–3:1) to furnish the title compound as a colorless oil (37.0 mg, 0.16 mmol, anti:syn = 1:8) in 79% yield.

Spectral data is reported for the major isomer.

<u>**TLC (SiO₂)</u>** $R_f = 0.35$ (hexanes/ethyl acetate = 2:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.32 (m, 5H), 5.80 (m, 1H), 5.15 (m, 2H), 4.58 (m, 2H), 3.72 (dd, *J* = 11.2, 4.1 Hz, 1H), 3.63–3.53 (m, 2H), 3.31 (dt, *J* = 6.5, 4.3 Hz, 1H), 3.02 (dd, *J* = 7.4, 4.5 Hz, 1H), 2.36 (m, 2H), 2.13 (brs, 1H).

 $\frac{13}{C \text{ NMR}} (100 \text{ MHz, CDCl}_3): \delta = 137.6, 133.1, 128.4, 127.8, 127.8, 118.5, 73.3, 69.0, 68.1, 59.2, 56.0, 38.7.$

<u>HRMS</u> (ESI) Calculated for $C_{14}H_{18}O_3$ [M+Na⁺] =257.1148, Found 257.1155.

<u>FTIR</u> (neat) 3421, 2926, 1737, 1365, 1229, 1094, 1027, 918, 700 cm⁻¹.

[α]_D²⁵: -86.67 (*c* 1.0, CHCl₃).



(*R*)-1-((2*R*,3*S*)-3-((*S*)-1-((4-methoxybenzyl)oxy)propan-2-yl)oxiran-2-yl)but-3-en-1-ol (2.2c)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1c** (51 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (49.1 mg, 0.17 mmol, anti:syn = 10:1) in 84% yield.

Spectral data is reported for the major isomer.

<u>**TLC (SiO₂**</u>) $R_f = 0.36$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.27 (m, 2H), 6.88 (m, 2H), 5.86 (m, 1H), 5.21 (m, 2H), 4.48 (s, 2H), 3.80 (s, 3H), 3.52 (m, 3H), 2.88 (m, 2H), 2.56 (m, 1H), 2.34 (m, 1H), 1.93 (brs, 1H), 1.79 (m, 1H), 1.10 (d, *J* = 6.9 Hz, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): δ = 159.1, 133.4, 130.5, 129.1, 119.1, 113.7, 73.0, 72.9,

68.0, 59.6, 58.1, 55.2, 40.0, 33.2, 13.8.

<u>HRMS</u> (ESI) Calculated for $C_{17}H_{24}O_4$ [M+Na⁺] = 315.1567, Found 315.1561.

<u>FTIR</u> (neat) 3438, 2962, 2362, 1612, 1513, 1247, 1088, 1034, 820 cm⁻¹.

 $[\alpha]_{D}^{25}$: +23.00 (*c* 1.0, CHCl₃).



(S)-1-((2R,3S)-3-((S)-1-((4-methoxybenzyl)oxy)propan-2-yl)oxiran-2-yl)but-3-en-1-ol (epi-2.2c)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1c** (51 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (48.5 mg, 0.17 mmol, anti:syn = 1:6) in 83% yield.

Spectral data is reported for the major isomer.

<u>TLC</u> (SiO₂) $R_f = 0.29$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.27 (m, 2H), 6.87 (m, 2H), 5.86 (m, 1H), 5.16 (m, 2H), 4.47 (s, 2H), 3.80 (s, 3H), 3.56 (m, 2H), 3.45 (dd, *J* = 9.0, 6.5 Hz, 1H), 2.97 (dd, *J* =

7.6, 4.3 Hz, 1H), 2.91 (dd, *J* = 9.5, 4.3 Hz, 1H), 2.37 (m, 2H), 2.21 (brs, 1H), 1.73 (m, 1H), 1.03 (d, *J* = 6.9 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{68.7, 60.2, 59.6, 55.2, 39.0, 33.2, 14.0.}$

<u>HRMS</u> (ESI) Calculated for $C_{17}H_{24}O_4$ [M+Na⁺] = 315.1567, Found 315.1556.

FTIR (neat) 3423, 2963, 2361, 1612, 1512, 1247, 1088, 1034, 820 cm⁻¹.

 $[\alpha]_{D}^{25}$: +25.00 (*c* 1.0, CHCl₃).



(R)-1-((2R,3S)-3-((R)-2,2-dimethyl-1,3-dioxolan-4-yl)oxiran-2-yl)but-3-en-1-ol (2.2d)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1d** (35 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (32.6 mg, 0.15 mmol, anti:syn = 10:1) in 76% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO₂)</u> $R_f = 0.42$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 5.84 (m, 1H), 5.22 (m, 2H), 4.18 (m, 2H), 3.85 (m, 1H), 3.63 (dt, *J* = 7.7, 3.9 Hz, 1H), 3.07 (m, 1H), 2.96 (dd, *J* = 7.3, 4.3 Hz, 1H), 2.55 (m, 1H), 2.31(m, 1H), 2.11 (d, *J* = 3.7 Hz, 1H), 1.48 (s, 3H), 1.38 (s, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 133.0, 119.3, 110.1, 74.7, 68.6, 66.5, 57.8, 57.6, 40.1, 26.5, 25.4.

<u>HRMS</u> (ESI) Calculated for $C_{11}H_{18}O_4$ [M+Na⁺] = 237.1097, Found 237.1095.

FTIR (neat) 3444, 2986, 1372, 1254, 1211,1154, 1058, 916, 840 cm⁻¹.

 $[\alpha]_{D}^{30}$: +89.00 (*c* 1.0, CHCl₃).



(S)-1-((2R,3S)-3-((R)-2,2-dimethyl-1,3-dioxolan-4-yl)oxiran-2-yl)but-3-en-1-ol (*epi-*2.2d)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1d** (35 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (30.9 mg, 0.14 mmol, anti:syn = 9:1) in 72% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO₂</u>) $R_f = 0.38$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.82$ (m, 1H), 5.17 (m, 2H), 4.11 (dt, J = 11.2, 6.3 Hz, 2H), 3.79 (dq, J = 7.1, 4.1 Hz, 1H), 3.69 (td, J = 7.0, 5.9 Hz, 1H), 3.11 (dd, J = 7.1, 4.4 Hz, 1H), 3.03 (dd, J = 6.9, 4.4 Hz, 1H), 2.35 (m, 3H), 1.47 (s, 3H), 1.37 (s, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 132.9$, 118.7, 110.1, 74.5, 68.4, 66.4, 59.3, 58.5, 39.0, 26.5, 25.4. **HRMS** (ESI) Calculated for C₁₁H₁₈O₄ [M+Na⁺] = 237.1097, Found 237.1094.

<u>FTIR</u> (neat) 3452, 2985, 1372, 1255, 1211,1154, 1057, 915, 843 cm⁻¹. $[\alpha]_D^{30}$: +127.33 (*c* 1.0, CHCl₃).



(R)-1-((2S,3R)-3-((R)-2,2-dimethyl-1,3-dioxolan-4-yl)oxiran-2-yl)but-3-en-1-ol (2.2e)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1e** (35 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (33.9 mg, 0.16 mmol, anti:syn = 1:10) in 79% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO₂</u>) $R_f = 0.52$ (hexanes/ethyl acetate = 2:1). <u>**1H NMR**</u> (400 MHz, CDCl₃): $\delta = 5.88$ (m, 1H), 5.16 (m, 2H), 4.14 (dd, J = 8.5, 6.1 Hz, 1H), 4.14 (dd, J = 8.5, 5.1 Hz, 1H), 3.92 (m, 1H), 3.65 (m, 1H), 3.04 (m, 2H), 2.47 (m, 1H), 2.37 (m, 2H), 1.45 (s, 3H), 1.35 (s, 3H). <u>1³C NMR</u> (100 MHz, CDCl₃): δ = 133.4, 118.4, 109.8, 73.4, 69.0, 68.2, 60.0, 57.4, 38.4, 26.8, 25.2.

<u>HRMS</u> (ESI) Calculated for $C_{11}H_{18}O_4$ [M+Na⁺] = 237.1097, Found 237.1094.

<u>FTIR</u> (neat) 3447, 2985, 2936, 1642, 1436, 1372, 1253, 1222, 1152, 1065, 918, 845 cm⁻¹.

[α]_D²⁷: -108.33 (*c* 1.0, CHCl₃).



(S)-1-((2S,3R)-3-((R)-2,2-dimethyl-1,3-dioxolan-4-yl)oxiran-2-yl)but-3-en-1-ol (*epi-*2.2e)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-SEGPHOS (6.1 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1e** (35 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1–3:1) to furnish the title compound as a colorless oil (31.7 mg, 0.15 mmol, anti:syn = 6:1) in 74% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO₂</u>) $R_f = 0.48$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 5.90 (m, 1H), 5.17 (m, 2H), 4.20 (m, 1H), 4.02 (m, 1H), 3.57 (m, 1H), 3.00 (m, 2H), 2.77 (dd, *J* = 2.4, 0.8 Hz, 1H), 2.44 (m, 2H), 1.45 (s, 3H), 1.34 (s, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 133.5, 118.0, 109.8, 74.5, 69.4, 68.4, 58.2, 56.3, 39.2, 26.7, 25.1.

<u>HRMS</u> (ESI) Calculated for $C_{11}H_{18}O_4$ [M+Na⁺] = 237.1097, Found 237.1095.

<u>FTIR</u> (neat) 3447, 2985, 2936, 1642, 1436, 1372, 1253, 1222, 1152, 1065, 918, 845 cm⁻¹.

[α]_D²⁶: -126.33 (*c* 1.0, CHCl₃).





(R)-1-((2R,3R)-3-(((triisopropylsilyl)oxy)methyl)oxiran-2-yl)but-3-en-1-ol (2.2f)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*R*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1f** (52 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–12:1) to furnish the title compound as a colorless oil (39.1 mg, 0.13 mmol, anti:syn = 17:1) in 65% yield.

Spectral data is reported for the major isomer.

<u>**TLC (SiO₂)</u>** $R_f = 0.58$ (hexanes/ethyl acetate = 4:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 5.87 (m, 1H), 5.14 (m, 2H), 3.94 (dd, J = 11.8, 3.0 Hz, 1H), 3.89 (m, 1H), 3.77 (dd, J = 11.8, 4.5 Hz, 1H), 3.20 (m, 1H), 3.02 (dd, J = 3.3, 2.3 Hz, 1H), 2.40 (m, 1H), 2.31 (m, 1H), 2.02 (brs, 1H), 1.05 (m, 21H).

¹³C NMR (100 MHz, CDCl₃): δ = 133.5, 118.1, 67.8, 63.0, 57.5, 55.4, 38.0, 17.9, 11.9.

<u>HRMS</u> (ESI) Calculated for C₁₆H₃₂O₃Si [M+Na⁺] =323.2013, Found 323.2013. **<u>FTIR</u>** (neat) 3438, 2942, 1463, 1114, 1067, 996, 882, 777, 682 cm⁻¹. $[\alpha]_D^{30}$: +106.00 (*c* 1.0, CHCl₃).



(S)-1-((2R,3R)-3-(((triisopropylsilyl)oxy)methyl)oxiran-2-yl)but-3-en-1-ol (epi-2.2f)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with $[Ir(cod)Cl]_2$ (3.4 mg, 0.005 mmol, 2.5 mol%), (*S*)-Cl-MeO-BIPHEP (6.5 mg, 0.01 mmol, 5 mol%), 4-CN,3-NO-benzoic acid (3.8 mg, 0.02 mmol, 10 mol%), alcohol **2.1f** (52 mg, 0.2 mmol, 100 mol%) and K₂CO₃ (13.8 mg, 0.1 mmol, 50 mol%). The vessel was purged with argon for 5 minutes. Anhydrous THF (1.0 mL, 0.2 M) and allyl acetate (43 µL, 0.4 mmol, 200 mol%) were sequentially added via syringe. The reaction vessel was sealed and the reaction mixture was allowed to stir at 100 °C for 48 h. After reaching ambient temperature, the reaction mixture was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–12:1) to furnish the title compound as a colorless oil (19.2 mg, 0.06 mmol, anti:syn = 1:7) in 32% yield.

Spectral data is reported for the major isomer.

<u>**TLC** (SiO₂</u>) $R_f = 0.54$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 5.85 (m, 1H), 5.14 (m, 2H), 3.92 (dd, J = 11.7, 2.9 Hz, 1H), 3.77 (dd, J = 11.8, 4.5 Hz, 1H), 3.60 (m, 1H), 3.11 (ddd, J = 4.6, 3.1, 2.3 Hz, 1H), 3.00 (dd, J = 4.7, 2.3 Hz, 1H), 2.39 (m, 2H), 2.03 (m, 1H), 1.06 (m, 21H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): δ = 133.4, 118.3, 70.1, 62.9, 58.2, 56.8, 39.2, 17.9, 11.9. **<u>HRMS</u>** (ESI) Calculated for C₁₆H₃₂O₃Si [M+Na⁺] =323.2013, Found 323.2014. **<u>FTIR</u>** (neat) 3426, 2942, 1463, 1112, 1067, 996, 882, 781, 682 cm⁻¹. $[\alpha]_D^{30}$: +107.33 (*c* 1.0, CHCl₃).



Procedures and Spectral Data for the Synthesis of 2.3c

(2S,3S,4S,5R)-1-((4-methoxybenzyl)oxy)-2,4-dimethyloct-7-ene-3,5-diol (2.3c)



Detailed Procedures

To a round-bottomed flask with **2.2c** (30.0 mg, 0.10 mmol) under an argon atmosphere was added 1,2-dichloroethane (3.0 mL, 0.03 M). The reaction vessel was placed in -40 °C bath and *n*-BuLi (80 μ L, 2.5 M in hexanes, 0.25 mmol, 200 mol%) was added. The mixture was stirred for 30 minutes, at which point trimethylaluminum (0.16 mL, 2.0 M in toluene, 0.31 mmol, 300 mol%) was added dropwise. The mixture was allowed to warm to -15 °C over 4 h. Water (3 mL) was added to the mixture and the mixture was transferred to a separatory funnel. The layers were separated and the aqueous layer was extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were washed with brine (10 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 7:1–4:1) to furnish the title compound as a colorless oil (23.7 mg, 0.08 mmol) in 77% yield.

<u>**TLC** (SiO₂)</u> $R_f = 0.40$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.24 (m, 2H), 6.89 (m, 2H), 5.82 (m, 1H), 5.09 (m, 2H), 4.46 (s, 2H), 4.36 (brs, 1H), 3.90 (ddd, *J* = 7.8, 6.3, 1.8 Hz, 1H), 3.81 (s, 3H), 3.75

(brs, 1H), 3.70 (m, 1H), 3.58 (dd, *J* = 9.1, 4.0 Hz, 1H), 3.44 (t, *J* = 9.2 Hz, 1H), 2.34 (m, 1H), 2.17 (m, 1H), 2.00 (m, 1H), 1.61 (m, 1H), 0.92 (d, *J* = 7.0 Hz, 3H), 0.74 (d, *J* = 6.9 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{76.0, 73.3, 55.3, 39.5, 37.4, 35.8, 13.0, 4.2.}$ (100 MHz, CDCl₃): $\delta = 159.4, 135.6, 129.4, 129.4, 116.8, 113.9, 82.6, 76.5, 76.0, 73.3, 55.3, 39.5, 37.4, 35.8, 13.0, 4.2.$

<u>HRMS</u> (ESI) Calculated for $C_{18}H_{28}O_4$ [M+K⁺] = 347.1619, Found 347.1611.

FTIR (neat) 3431, 2915, 1612, 1513, 1247, 1079, 1035, 970, 753 cm⁻¹.

[α]³⁰: +42.83 (*c* 1.0, CHCl₃).




(udd) tj



HPLC Data Establishing the Horeau Effect in the Allylation of Glycidols 2.2a

HPLC: Enantiomeric excess was determined by HPLC analysis of the benzoate derivative of the product (Chiralcel OD-H column, hexanes:*i*-PrOH = 90:10, 1.00 mL/min, 210 nm), ee = 99.5%.

Chapter 3: Nickel-Catalyzed Cross-Coupling of Vinyl Dioxanones to Form Enantiomerically Enriched Cyclopropanes*

3.1 Introduction

Cyclopropanes appear as substructures across diverse secondary metabolites¹ and are frequently found in commercial medicines, agrochemicals and fragrances.² Hence, the development of methods for cyclopropane formation represents a persistent challenge in chemical research.³ Among the most effective methods for the preparation of enantiomerically enriched cyclopropanes is the reaction of olefins with metal carbenoids.³ Here, we report a strategy for the asymmetric synthesis of cyclopropanes under the conditions of metal catalyzed cross-coupling. Specifically, nickel(0) catalysts^{4,5} react with enantiomerically enriched 4-aryl-5-vinyl-1,3-dioxanones to form (cyclopropyl-carbinyl)nickel(II) species, which, in the presence of organoboron reagents or B₂(pin)₂ deliver cyclopropanes in a stereospecific manner. Thus, the enantioselective synthesis of tetra-substituted cyclopropanes bearing all-carbon quaternary stereocenters is achieved (Scheme 3.1).



Scheme 3.1 Synthesis of enantiomerically enriched cyclopropanes from vinyldioxanones by way of transient (cyclopropyl-carbinyl)nickel species.

^{*}This chapter is based on the published work:

Guo, Y.-A.; Liang, T.; Kim, S. W.; Xiao, H.; Krische, M. J. J. Am. Chem. Soc. 2017, 139, 6847.



In connection with ongoing investigations into the formation of C-C bonds *via* hydrogenation and transfer hydrogenation,⁶ we recently reported an iridium catalyzed coupling of primary alcohols with isoprene oxide to form products of tert-(hydroxy)-prenylation – a byproduct-free transformation that occurs with exceptional control of *anti*-diastereo- and enantioselectivity.⁷ It was posited that cyclic carbonates derived from these reaction products should be predisposed toward cyclopropane formation under cross-coupling conditions, as geminal substitution of the neopentyl glycol precludes competing β -hydride elimination of the σ -benzylmetal intermediate and should conformationally bias the system toward olefin insertion⁸ *via* Thorpe-Ingold effect.⁹ However, the facility of conventional benzylic cross-coupling rendered the feasibility of the proposed cyclopropane formation uncertain.¹⁰

3.2 Reaction Development and Scope

In an initial experiment, vinyl-dioxanone 3.1a was exposed to the catalyst derived from Ni(cod)₂ (10 mol%) and PCy₃ (20 mol%) in the presence of tri(*p*-tolyl)boroxine **3.2a** and K₃PO₄ (200 mol%) in toluene (0.1 M) at 60 °C. To our delight, cyclopropane **3.3a** was formed in 36% yield as a single diastereomer. Conversion was found to be sensitive to concentration and temperature. At 45 °C under otherwise identical conditions, a 53% yield of cyclopropane 3.3a was obtained. Using the nickel catalyst modified by PCy₂Ph (20 mol%), cyclopropane **3.3a** was obtained in 77% yield. Finally, at slightly higher concentration (toluene, 0.2 M), an 85% yield of cyclopropane **3.3a** was achieved (eq. 3.1). Stereospecificity was corroborated by chiral stationary phase HPLC analysis of cyclopropane 3.3a. Relative stereochemistry of cyclopropane 3.3a was confirmed by single crystal X-ray diffraction analysis. p-Tolylboronic acid also delivers cyclopropane 3.3a (eq. 3.1), but in slightly lower yield. Application of these optimal conditions to unsubstituted methyl carbonate *model*-3.1a did not result in cyclopropane formation; rather, the indicated product obtained through β -hydride elimination of the σ -benzyl intermediate was formed (eq. 3.2). Cyclic carbonate 3.1a reacted more efficiently than related acyclic carbonates, suggesting the internal alkoxide generated upon ionizationdecarboxylation facilitates group transfer from boron to nickel through an internal boron ate-complex.

Optimal conditions utilizing tri(*p*-tolyl)boroxine **3.2a** were applied to a structurally diverse set of enantiomerically enriched vinyl-dioxanones **3.1a-3.1i** (Table 3.1). Vinyl dioxanones bearing a variety of substituted aromatic (**3.1a-3.1d**) and heteroaromatic (**3.1e-3.1i**) rings were converted to cyclopropanes **3.3a-3.3i** in good yield with complete levels of diastereoselectivity. Relative stereochemistry was assigned in analogy to that determined for **3.3a**. Although the preexisting non-epimerizable

quaternary stereocenter serves as an "internal standard," stereospecificity was spotchecked for compounds **3.3a**, **3.3b**, **3.3d** and **3.3h**. Notably, unlike prior work involving nickel catalyzed benzylic substitution, extended aromatic systems are not required.¹¹ Standard conditions also were applied to the coupling of vinyl-dioxanones **3.1a** and 1h with boroxines **3.2b-3.2d**, which incorporate *p*-CF₃-phenyl, *p*-methoxyphenyl and (*E*)styryl moieties, respectively (Table 3.2).

Table 3.1Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a-3.1i with tri(p-tolyl)boroxine 3.2a to form cyclopropanes 3.3a-3.3i.a



^aYields of material isolated by silica gel chromatography. All reactions were conducted using enantiomerically enriched starting materials. See Supporting Information for further experimental details.

Table 3.2Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a or
3.1h with boroxines 3.2b-3.2d to form cyclopropanes 3.3j-3.3o.^a



^aYields of material isolated by silica gel chromatography. All reactions were conducted using enantiomerically enriched starting materials. See Supporting Information for further experimental details.

The resulting cyclopropanes **3.3j-3.3o** were formed in good yield in a completely stereoselective fashion. The coupling of vinyl-dioxanones **3.1a**, **3.1b**, **3.1d**, **3.1h** and **3.1f** with $B_2(pin)_2$ under standard conditions delivers the cyclopropylcarbinyl boronates **3.3p-3.3t** in good yield with complete stereocontrol (Table 3.3).¹² To briefly illustrate the utility of coupling products, the neopentyl alcohol **3.3a** was subjected to Jones oxidation to provide the cyclopropyl carboxylic acid **3.4a** in good yield (eq. 3.3). Additionally, the cyclopropylcarbinyl alcohol **3.3h** was exposed to Mitsunobu conditions in the presence of phthalimide to furnish **3.4b** in excellent yield (eq. 3.4).

Table 3.3Stereospecific nickel-catalyzed cross coupling of vinyl-dioxanones 3.1a,
3.1b, 3.1d, 3.1h and 3.1f with B2(pin)2 to form cyclopropanes 3.3p-3.3t.ª



^aYields of material isolated by silica gel chromatography. All reactions were conducted using enantiomerically enriched starting materials. See Supporting Information for further experimental details.



3.3 Discussion

A general mechanism for stereospecific cyclopropane formation under the conditions of nickel catalyzed cross-coupling has been proposed (Scheme 3.2). Stereospecific oxidative addition of a nickel(0) species to the benzylic C-O bond occurs with inversion to furnish the indicated σ -benzylnickel(II) complex.¹⁰ Decarboxylation and transmetalation delivers the indicated alkene complex, which upon reversible migratory insertion⁸ provides a (cyclopropylcarbinyl)nickel(II) complex. Regardless of the kinetic diastereoselectivity of olefin insertion, reductive elimination occurs exclusively from a single stereoisomer of the (cyclopropylcarbinyl)nickel(II) species to release the cyclopropane and regenerate the zero-valent nickel catalyst.

Scheme 3.2 General catalytic mechanism. Haptomeric equilibria are excluded for clarity.



3.4 Conclusion

In summary, we report a new method for the preparation of enantiomerically enriched cyclopropanes *via* stereospecific nickel catalyzed cross-coupling of vinyldioxanones with boroxines or $B_2(pin)_2$. The collective data are consistent with a catalytic mechanism involving nickel(0)-mediated benzylic oxidative addition with inversion of stereochemistry followed by reversible olefin insertion to form a (cyclopropylcarbinyl) nickel complex, which upon reductive elimination delivers the cyclopropane. The novel reactivity embodied by this process should serve as the basis for the syntheses of diverse enantiomerically enriched cyclopropanes.

3.5 Experimental Details

General Information

All reactions were run under an atmosphere of argon. Sealed tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were ovendried followed by cooling in a desiccator. Tetrahydrofuran was distilled from sodiumbenzophenone immediately prior to use. Ethyl Acetate was dried over potassium carbonate and distilled immediately prior to use. Anhydrous solvents were transferred by oven-dried syringes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynanmic Absorbents F254). Visualization was accomplished with UV light followed by dipping in *p*-anisaldehyde stain solution then heating. Purification of reactions was carried out by flash chromatography using Silacycle silica gel (40-63 µm, unless indicated specifically). Potassium phosphate was purchased through Acros Organics, flame dried prior to use and stored in a desiccator. Ni(cod)₂ was purchased from Strem Chemicals. (*S*)-Ir-Tol-BINAP was synthesized according to literature procedures¹. *p*-Tolyl-boroxine $(\mathbf{3.2a})^2$, 4-(trifluoromethyl)phenylboroxine $(\mathbf{3.2b})^3$, 4-methoxyphenyl-boroxine $(\mathbf{3.2c})^3$ and (*E*)-styryl-boroxine $(\mathbf{3.2d})^3$ were synthesized according to literature procedures.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer. Highresolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M, M+H, or M-H), or a suitable fragment ion. ¹H Nuclear magnetic resonance spectra were recorded using a 400 MHz or a 500 MHz spectrometer. Coupling constants are reported in Hertz (Hz) for CDCl₃ solutions, and chemical shifts are reported as parts per million (ppm) relative to residual CHCl₃ $\delta_{\rm H}$ (7.26 ppm). ¹³C Nuclear magnetic resonance spectra were recorded using a 100 MHz or a 125 MHz spectrometer for CDCl₃ solutions, and chemical shifts are reported as parts per million (ppm) relative to residual CDCl₃ $\delta_{\rm C}$ (77.0 ppm). Fluorine-19 nuclear magnetic resonance (¹⁹F NMR) spectra were recorded with a Varian Gemini 400 (100 MHz) or a Bruker 500 (125 MHz) spectrometer. Melting points were taken on a Stuart SMP3 melting point apparatus.

Experimental Details and Spectral Data

Procedures and Spectral Data for the Synthesis of Vinyl-Dioxanones 3.1a-3.1i:

(1*R*,2*R*)-2-methyl-1-(*p*-tolyl)-2-vinylpropane-1,3-diol (SI-**3.1a**)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with K_3PO_4 (42.4 mg, 0.2 mmol, 5 mol%), (*S*)-Ir-Tol-BINAP (220 mg, 0.2 mmol, 5 mol%) and *p*-tolylmethanol (488 mg, 4.0 mmol, 100 mol%). Under an atmosphere of argon, anhydrous THF (8 mL, 0.5 M) and isoprene monoxide (1.18 mL, 12 mmol, 300 mol%) were sequentially added via syringe. After sealing the tube with cap, the reaction mixture was stirred at 45 °C for 24 h. The reaction was cooled to ambient temperature and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, methylene chloride: acetone = 30:1) to furnish the title compound as a yellow oil (626 mg, 3.0 mmol, *anti:syn* > 20:1) in 76% yield.

<u>**TLC**</u> (SiO₂) $R_f = 0.30$ (methylene chloride: acetone = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.20 – 7.07 (m, 4H), 6.03 (dd, J = 17.7, 11.0 Hz, 1H), 5.20 (dd, J = 11.0, 1.1 Hz, 1H), 5.01 (dd, J = 17.8, 1.2 Hz, 1H), 4.62 (s, 1H), 3.60 (d, J = 10.7 Hz, 1H), 3.51 (d, J = 10.7 Hz, 1H), 3.18 (brs, 1H), 2.90 (brs, 1H), 2.33 (s, 3H), 0.90 (s, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃) δ 139.6, 137.8, 137.2, 128.3, 127.6, 115.9, 79.9, 69.7, 46.2, 21.0, 17.8.

<u>HRMS</u> (ESI) Calcd. for $C_{13}H_{18}NaO_2^+$ [M+Na]⁺: 229.1199, Found: 229.1201.

<u>FTIR</u> (neat): 3377, 2966, 2919, 2977, 1637, 1515, 1460, 1415, 1378, 1201, 1039, 1018, 919, 821, 678 cm⁻¹.

 $[\alpha]_{D}^{33}$: -34.7 (*c* = 1.0, CHCl₃).

<u>**HPLC</u>** (two connected chiralcel OJ-H columns, hexanes:*i*-PrOH = 98:2, 0.80 mL/min, 230 nm), anti:syn = 35:1, ee = 93%.</u>





(4*R*,5*R*)-5-methyl-4-(*p*-tolyl)-5-vinyl-1,3-dioxan-2-one (3.1a)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with diol **SI-1a** (50 mg, 0.24 mmol, 100 mol%). Under argon atmosphere, acetonitrile (2.4 mL, 0.1 M) was added via syringe. CDI (77.8 mg, 0.48 mmol, 200 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a white solid (46.3 mg, 0.20 mmol) in 82 % yield.

<u>**TLC (SiO2)**</u> $R_f = 0.25$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.20 – 7.12 (m, 4H), 5.65 (ddd, J = 17.6, 11.1, 0.8 Hz, 1H), 5.30 (d, J = 11.1 Hz, 1H), 5.25 (s, 1H), 5.23 (d, J = 17.6 Hz, 1H), 4.42 (d, J = 11.0 Hz, 1H), 4.30 (dd, J = 11.0, 0.9 Hz, 1H), 2.36 (s, 3H), 1.04 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 148.4, 138.8, 134.4, 130.8, 128.7, 127.4, 117.8, 87.0,
75.1, 38.3, 21.2, 19.5.

<u>HRMS</u> (ESI) Calcd. for $C_{14}H_{16}NaO_3^+$ [M+Na]⁺: 255.0992, Found: 255.0994.

 $[\alpha]_{D}^{33}$: -73.7 (*c* = 1.0, CHCl₃).

m.p. : 85-86 °C

<u>FTIR</u> (neat): 2977, 1747, 1517, 1478, 1456, 1399, 1378, 1341, 1241, 1200, 1712, 1135, 1102, 1007, 931, 819, 764, 688 cm⁻¹.



(4*R*,5*R*)-4-(4-fluorophenyl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1b)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with (1R,2R)-1-(4fluorophenyl)-2-methyl-2-vinylpropane-1,3-diol⁴ (100 mg, 0.48 mmol, 100 mol%). Under argon atmosphere, acetonitrile (4.8 mL, 0.1 M) was added via syringe. CDI (154 mg, 0.95 mmol, 200 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated in vacuo. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a white solid (90.7 mg, 0.38 mmol) in 81 % yield. **TLC (SiO₂)** $R_f = 0.23$ (hexanes/ethyl acetate = 2:1).

¹**H NMR** (400 MHz, CDCl₃) δ 7.28 – 7.23 (m, 2H), 7.10 – 7.03 (m, 2H), 5.64 (ddd, J =17.6, 11.0, 0.7 Hz, 1H), 5.32 (d, J = 11.1 Hz, 1H), 5.28 (s, 1H), 5.22 (d, J = 17.6 Hz, 1H), 4.40 (d, *J* = 11.0 Hz, 1H), 4.33 (dd, *J* = 11.0, 0.8 Hz, 1H), 1.02 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 162.9 (d, J = 248.3 Hz), 148.1, 133.9, 129.6 (d, J = 3.2Hz), 129.3 (d, *J* = 8.3 Hz), 118.2, 115.1 (d, *J* = 21.7 Hz), 86.3, 75.2, 38.3, 19.2.

¹⁹**F** NMR (376 MHz, CDCl₃) δ -112.4 (tt, *J* = 8.6, 5.2 Hz).

HRMS (ESI) Calcd. for C₁₃H₁₃FNaO₃⁺ [M+Na]⁺: 259.0741, Found: 259.0744. $[\alpha]_{D}^{33}$: -46.0 (*c* = 1.0, CHCl₃). **m.p.** : 108-109 °C

<u>FTIR</u> (neat): 2977, 1748, 1608, 1512, 1480, 1456, 1378, 1228, 1203, 1174, 1134, 1098, 929, 832, 769 cm⁻¹.







(4*R*,5*R*)-4-(3,5-bis(trifluoromethyl)phenyl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1c)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with (1R,2R)-1-(3,5-bis(trifluoromethyl)phenyl)-2-methyl-2-vinylpropane-1,3-diol⁴ (164 mg, 0.5 mmol, 100 mol%). Under argon atmosphere, acetonitrile (5 mL, 0.1 M) was added via syringe. CDI (81 mg, 0.5 mmol, 100 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a white solid (115 mg, 0.32 mmol) in 65% yield.

<u>**TLC**</u> (SiO₂) $R_f = 0.23$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.92 (s, 1H), 7.74 (d, J = 1.5 Hz, 2H), 5.65 (dd, J = 17.5, 11.0 Hz, 1H), 5.52 – 5.34 (m, 2H), 5.22 (d, J = 17.5 Hz, 1H), 4.55 – 4.18 (m, 2H), 1.07 (s, 3H).

<u>1³C NMR</u> (125 MHz, CDCl₃) δ 147.3, 136.4, 132.6, 131.6 (q, J = 33.8 Hz), 127.6 (d, J = 4.1 Hz), 124.0, 123.3 - 122.7 (m), 121.8, 119.6, 85.4.

<u>HRMS</u> (ESI) Calcd. for $C_{15}H_{12}F_6NaO_3^+$ [M+Na]⁺: 377.0583, Found: 377.0590.

 $[\alpha]_{D}^{31}$: -44.0 (*c* = 1.0, CHCl₃).

m.p. : 61-62°C

<u>FTIR</u> (neat): 2977, 1744, 1467, 1402, 1276, 1225, 1167, 1126, 1104, 1000, 904, 759, 681 cm⁻¹.



(4R,5R)-4-(benzo[d][1,3]dioxol-5-yl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1d)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with (1R,2R)-1-(benzo[*d*][1,3]dioxol-5-yl)-2-methyl-2-vinylpropane-1,3-diol¹ (90 mg, 0.38 mmol, 100 mol%). Under argon atmosphere, acetonitrile (3.8 mL, 0.1 M) was added via syringe. CDI (124 mg, 0.76 mmol, 200 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 2:1) to furnish the title compound as a white solid (99.6 mg, 0.32 mmol) in 84 % yield.

<u>**TLC (SiO₂)**</u> $R_f = 0.20$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 6.80 – 6.75 (m, 2H), 6.71 (ddd, J = 8.0, 1.8, 0.5 Hz, 1H), 5.98 (s, 2H), 5.68 (ddd, J = 17.6, 11.1, 0.9 Hz, 1H), 5.31 (d, J = 11.0 Hz, 1H), 5.24 (d, J = 17.6 Hz, 1H), 5.19 (s, 1H), 4.39 (d, J = 11.0 Hz, 1H), 4.29 (dd, J = 11.0, 0.9 Hz, 1H), 1.01 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 148.2, 148.0, 147.5, 134.3, 127.5, 121.3, 117.9, 108.0, 107.6, 101.3, 86.8, 75.2, 38.4, 19.4.

<u>HRMS</u> (ESI) Calcd. for C₁₄H₁₄NaO₅⁺ [M+Na]⁺: 285.0733, Found: 285.0741. $[\alpha]_D^{33}$: -62.3 (c = 1.0, CHCl₃). **<u>FTIR</u>** (neat): 2970, 2360, 2341, 1748, 1505, 1491, 1447, 1399, 1377, 1253, 1205, 1102, 1038, 933, 815, 768, 669 cm⁻¹



(1S,2R)-1-(benzo[b]thiophen-2-yl)-2-methyl-2-vinylpropane-1,3-diol (SI-3.1e)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with K₃PO₄ (10.6 mg, 0.05 mmol, 5 mol%), (*S*)-Ir-Tol-BINAP (55.0 mg, 0.05 mmol, 5 mol%) and benzo[*b*]thiophen-2-ylmethanol (82.1 mg, 0.5 mmol, 100 mol%). Under an atmosphere of argon, anhydrous THF (1.0 mL, 0.5 M) and isoprene monoxide (147 μ L, 15 mmol, 300 mol%) were sequentially added via syringe. After sealing the tube with cap, the reaction mixture was stirred at 45 °C for 24 h. The reaction was cooled to ambient temperature and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, methylene chloride: acetone = 30:1) to furnish the title compound as a yellow oil (102 mg, 0.41 mmol, *anti:syn* > 20:1) in 82% yield.

<u>**TLC**</u> (SiO₂) $R_f = 0.32$ (dichloromethane/acetone = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.20 – 7.07 (m, 4H), 6.03 (dd, J = 17.7, 11.0 Hz, 1H), 5.20 (dd, J = 11.0, 1.1 Hz, 1H), 5.01 (dd, J = 17.8, 1.2 Hz, 1H), 4.62 (s, 1H), 3.60 (d, J = 10.7 Hz, 1H), 3.51 (d, J = 10.7 Hz, 1H), 3.18 (brs, 1H), 2.90 (brs, 1H), 2.33 (s, 3H), 0.90 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 145.2, 139.5, 139.1, 139.0, 124.1, 124.1, 123.3, 122.2, 122.1, 117.0, 69.7, 46.4, 18.0.

HRMS (ESI) Calcd. for C₁₄H₁₆NaO₂S⁺ [M+Na]⁺: 271.0763, Found: 271.0772.

<u>FTIR</u> (neat): 3345, 2964, 1636, 1457, 1415, 1124, 1014, 921. 832, 745, 725, 709 cm⁻¹.

 $[\alpha]_{D}^{33}$: -26.2 (*c* = 1.0, CHCl₃).

<u>**HPLC**</u> (one chiralcel OD-H columns, hexanes:*i*-PrOH = 97:3, 1.0 mL/min, 30 nm), anti:syn = 20:1, ee = 93%.





Signal 1: DAD1 D, Sig=230,16 Ref=360,100

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	52.648	MF	1.5267	553.52832	6.04287	4.6504
2	56.456	FM	1.7190	1.11971e4	108.56383	94.0700
3	66.709	MM	1.9341	152.31200	1.31253	1.2796
Totals :				1.19029e4	115.91923	

122

(4*S*,5*R*)-4-(benzo[b]thiophen-2-yl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1e)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with diol **SI-3.1e** (99.3 mg, 0.4 mmol, 100 mol%). Under argon atmosphere, acetonitrile (4 mL, 0.1 M) was added via syringe. CDI (65 mg, 0.4 mmol, 100 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a white solid (83.3 mg, 0.30 mmol) in 75% yield.

<u>**TLC** (SiO₂</u>) $R_f = 0.32$ (dichloromethane/acetone = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.86 – 7.80 (m, 1H), 7.79 – 7.76 (m, 1H), 7.40 – 7.35 (m, 2H), 7.32 (t, *J* = 0.7 Hz, 1H), 5.92 (ddd, *J* = 17.6, 11.1, 0.7 Hz, 1H), 5.59 (s, 1H), 5.38 (d, *J* = 11.1 Hz, 1H), 5.34 (d, *J* = 17.6 Hz, 1H), 4.46 (d, *J* = 11.1 Hz, 1H), 4.33 (dd, *J* = 11.0, 0.8 Hz, 1H), 1.21 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 147.4, 139.7, 138.5, 136.7, 134.2, 125.0, 124.6, 124.2, 123.9, 122.2, 118.6, 84.1, 75.3, 38.5, 19.4.

<u>HRMS</u> (ESI) Calcd. for $C_{15}H_{14}NaO_3S^+$ [M+Na]⁺: 297.0556, Found: 2970561.

 $[\alpha]_{D}^{32}$: -37.1 (*c* = 1.0, CHCl₃).

m.p. : 146-147 °C

<u>FTIR</u> (neat): 2979, 17332, 1488, 1404, 1332, 1222, 1180, 1127, 1091, 1046, 944, 840, 758, 728, 661 cm⁻¹.



tert-butyl 5-(hydroxymethyl)-1H-indole-1-carboxylate (SI-3.4)



Detailed Procedures

To a round-bottomed flask charged with *tert*-butyl 5-formyl-1*H*-indole-1-carboxylate⁵ (1.70 g, 6.88 mmol, 100 mol%) under an argon atmosphere was added EtOH (26.0 mL, 0.3 M). The reaction vessel was placed in an ice bath. After 10 minutes, sodium borohydride (390 mg, 10.32 mmol, 150 mol%) was added and the mixture was stirred for 1 h. Water (20 mL) was added to the reaction mixture and the mixture was extracted with CH₂Cl₂ (30 mL × 3). The combined organic layers were washed with brine (100 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-5:1) to furnish the title compound as a colorless oil (1.60 g, 6.5 mmol) in 94% yield. **TLC (SiO₂)** R_f = 0.39 (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.10 (d, J = 8.6 Hz, 1H), 7.59 (d, J = 3.7 Hz, 1H), 7.51 (d, J = 1.6 Hz, 1H), 7.28 (dd, J = 8.5, 1.7 Hz, 1H), 6.53 (d, J = 3.7 Hz, 1H), 4.72 (s, 2H), 2.35 (s, 1H), 1.67 (s, 9H).

¹³C NMR (125 MHz, CDCl₃) δ 149.8, 135.4, 134.7, 130.7, 126.3, 123.7, 119.5, 115.2, 107.3, 83.8, 65.5, 28.2.

HRMS (ESI) Calcd. for C₁₄H₁₇NO₃ [M+Na]⁺: 270.1101, Found: 270.1101.

<u>FTIR</u> (neat): 3357, 2978, 1729, 1472, 1369, 1218, 1158, 1081, 1021, 759, 723 cm⁻¹.


tert-butyl 5-((1R,2R)-1-hydroxy-2-(hydroxymethyl)-2-methylbut-3-en-1-yl)-1Hindole-1-carboxylate (SI-3.1f)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with K₃PO₄ (4.2 mg, 0.02 mmol, 5 mol%), (*S*)-Ir-Tol-BINAP (22 mg, 0.02 mmol, 5 mol%) and alcohol **SI-3.4** (100 mg, 0.40 mmol, 100 mol%). Under an atmosphere of argon, anhydrous THF (0.8 mL, 0.5 M) and isoprene monoxide (0.12 mL, 1.2 mmol, 300 mol%) were sequentially added via syringe. After sealing the tube with cap, the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1-4:1) to furnish the title compound as a yellow oil (98 mg, 0.30 mmol, *anti:syn* > 20:1) in 74% yield.

<u>TLC</u> (SiO₂) $R_f = 0.28$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.08 (d, *J* = 8.5 Hz, 1H), 7.61 (d, *J* = 3.7 Hz, 1H), 7.54 (d, *J* = 1.5 Hz, 1H), 7.31 – 7.27 (m, 1H), 6.57 (d, *J* = 3.7 Hz, 1H), 6.12 (dd, *J* = 17.8, 11.0 Hz, 1H), 5.27 (dd, *J* = 11.0, 1.2 Hz, 1H), 5.08 (dd, *J* = 17.8, 1.3 Hz, 1H), 4.85 (d, *J* = 2.5 Hz, 1H), 3.68 (dd, *J* = 10.7, 6.1 Hz, 1H), 3.62 (dd, *J* = 10.7, 5.1 Hz, 1H), 2.73 (d, *J* = 2.8 Hz, 1H), 2.34 (t, *J* = 5.8 Hz, 1H), 1.69 (s, 9H), 0.98 (s, 3H). 1³C NMR (125 MHz, CDCl₃) δ 149.7, 139.8, 135.2, 134.7, 130.1, 126.2, 124.1, 120.0, 116.2, 114.2, 107.4, 83.7, 80.1, 69.9, 46.7, 28.2, 17.7.

HRMS (ESI) Calcd. for C₁₉H₂₅NO₄ [M+Na]⁺: 354.1676, Found: 354.1680.

FTIR (neat): 3384, 2978, 1732, 1469, 1352, 1255, 1160, 1022, 754 cm⁻¹.

 $[\alpha]_{D}^{29}$: -24.0 (*c* = 1.0, CHCl₃).

<u>**HPLC</u>** (Chiralcel AS-H columns, hexanes:*i*-PrOH = 99:1 (100 minutes) – 98:2 (100 minutes), 1.00 mL/min, 230 nm), *anti:syn* = 85:1, ee = 88%.</u>





Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	010
1	128.845	BB	1.8847	3805.69751	25.50686	6.0828
2	162.928	BB	3.2629	5.87588e4	214.43245	93.9172

tert-butyl 5-((4*R*,5*R*)-5-methyl-2-oxo-5-vinyl-1,3-dioxan-4-yl)-1H-indole-1carboxylate (3.1f)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with diol **SI-3.1f** (160 mg, 0.48 mmol, 100 mol%). Under argon atmosphere, acetonitrile (4.8 mL, 0.1 M) was added via syringe. CDI (156 mg, 0.97 mmol, 200 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 2:1) to furnish the title compound as a white solid (147 mg, 0.41 mmol) in 85% yield.

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.13 (d, *J* = 8.8 Hz, 1H), 7.62 (d, *J* = 3.7 Hz, 1H), 7.49 (d, *J* = 1.7 Hz, 1H), 7.19 (dd, *J* = 8.6, 1.8 Hz, 1H), 6.57 (d, *J* = 3.8 Hz, 1H), 5.68 (dd, *J* = 17.6, 11.1 Hz, 1H), 5.38 (s, 1H), 5.30 (d, *J* = 11.1 Hz, 1H), 5.23 (d, *J* = 17.7 Hz, 1H), 4.44 (d, *J* = 11.0 Hz, 1H), 4.33 (d, *J* = 10.9 Hz, 1H), 1.67 (s, 9H), 1.05 (s, 3H). ¹³<u>C NMR</u> (125 MHz, CDCl₃) δ 149.5, 148.5, 135.3, 134.5, 130.2, 128.1, 126.7, 123.6, 120.1, 117.8, 114.5, 107.2, 87.3, 84.0, 75.1, 38.5, 28.1, 19.6.

<u>HRMS</u> (ESI) Calcd. for C₂₀H₂₃NO₅ [M+Na]⁺: 380.1468, Found: 380.1468. $[\alpha]_D^{29}$: -43.3 (*c* = 1.0, CHCl₃). **m.p.** : 158–162 °C

<u>FTIR</u> (neat): 2978, 1734, 1473, 1358, 1222, 1161, 1102, 1024, 760 cm⁻¹.



(1*R*,2*R*)-1-(2,3-dimethylquinoxalin-6-yl)-2-methyl-2-vinylpropane-1,3-diol (SI-3.1g)



Detailed Procedures

An oven-dried pressure tube equipped with a magnetic stir bar was charged with K_3PO_4 (4.2 mg, 0.02 mmol, 5 mol%), (*S*)-Ir-Tol-BINAP (22.0 mg, 0.02 mmol, 5 mol%) and (2,3-dimethylquinoxalin-6-yl)methanol⁶ (75.2 mg, 0.4 mmol, 100 mol%). Under an atmosphere of argon, anhydrous THF (0.8 mL, 0.5 M) and isoprene monoxide (0.16 mL, 1.6 mmol, 400 mol%) were sequentially added via syringe. After sealing the tube with cap, the reaction mixture was stirred at 60 °C for 24 h. The reaction was cooled to ambient temperature and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:10) to furnish the title compound as a yellow oil (92.0 mg, 0.34 mmol, *anti:syn* > 20:1) in 85% yield.

<u>**TLC (SiO₂**</u>) $R_f = 0.31$ (hexanes/ethyl acetate = 1:20).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.87 (d, *J* = 1.3 Hz, 1H), 7.84 (d, *J* = 8.6 Hz, 1H), 7.64 (dd, *J* = 8.6, 1.7 Hz, 1H), 6.12 (dd, J = 17.7, 11.0 Hz, 1H), 5.20 (dd, *J* = 11.0, 0.8 Hz, 1H), 4.98 (dd, *J* = 17.8, 0.9 Hz, 1H), 4.93 (s, 1H), 4.04 (brs, 1H), 3.71 (d, *J* = 10.6 Hz, 1H), 3.61 (d, *J* = 10.6 Hz, 1H), 3.24 (brs, 1H), 2.68 (s, 3H), 2.67 (s, 3H), 0.95 (s, 3H). ¹³<u>C NMR</u> (125 MHz, CDCl₃) δ 153.5, 153.4, 142.3, 140.4, 140.1, 139.3, 129.0, 127.1, 126.8, 116.3, 79.5, 69.7, 46.4, 23.0, 18.0.

<u>HRMS</u> (ESI) Calcd. for $C_{16}H_{21}N_2O_2^+$ [M+H]⁺: 273.1598, Found: 273.1597.

<u>FTIR</u> (neat): 3325, 2965, 2877, 1637, 1496, 1450, 1405, 1380, 1334, 1254, 1164, 1148, 1043, 021, 838, 809, 757, 666 cm⁻¹.

 $[\alpha]_{D}^{33}$: -23.8 (*c* = 1.0, CHCl₃).

HPLC (two connected chiralcel AD-H columns, hexanes:*i*-PrOH = 95:5, 0.80 mL/min,

230 nm), *anti:syn* = 30:1, ee = 90%.









(4*R*,5*R*)-4-(2,3-dimethylquinoxalin-6-yl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1g)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with diol **SI-3.1g** (65.0 mg, 0.24 mmol, 100 mol%). Under argon atmosphere, acetonitrile (2.4 mL, 0.1 M) was added via syringe. CDI (77.4 mg, 0.48 mmol, 200 mol%) was added in one portion at ambient temperature. The reaction mixture was stirred at 25 °C for 16 h. The reaction was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:10) to furnish the title compound as a yellow solid (51.0 mg, 0.17 mmol) in 72 % yield.

<u>TLC (SiO₂</u>) $R_f = 0.37$ (hexanes/ethyl acetate = 1:10).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.98 (d, J = 8.7 Hz, 1H), 7.88 (d, J = 1.4 Hz, 1H), 7.60 (dd, J = 8.7, 1.8 Hz, 1H), 5.69 (dd, J = 17.6, 11.1 Hz, 1H), 5.50 (s, 1H), 5.32 (d, J = 11.1 Hz, 1H), 5.22 (d, J = 17.6 Hz, 1H), 4.45 (d, J = 11.0 Hz, 1H), 4.39 (d, J = 11.1 Hz, 1H), 2.74 (s, 3H), 2.74 (s, 3H), 1.10 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 154.5, 154.4, 147.9, 141.0, 140.2, 134.6, 133.9, 128.2, 127.7, 127.4, 118.4, 86.6, 75.2, 38.5, 23.2, 23.2, 19.4.

HRMS (ESI) Calcd. for C₁₇H₁₉N₂O₃⁺ [M+Na]⁺: 299.1390, Found: 299.1392.

 $[\alpha]_{D}^{33}$: -70.8 (*c* = 1.0, CHCl₃).

m.p. : 218-220 °C (decomposed)

<u>FTIR</u> (neat): 2970, 1747, 1456, 1401, 1335, 1240, 1210, 1166, 1134, 1103, 999, 974, 841, 767, 670 cm⁻¹.



(*4R*,*5R*)-4-(6-methoxypyridin-3-yl)-5-methyl-5-vinyl-1,3-dioxan-2-one (3.1h)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with (1R,2R)-1-(6methoxypyridin-3-yl)-2-methyl-2-vinylpropane-1,3-diol⁴ (558 mg, 2.5 mmol, 100 mol%). Under argon atmosphere, acetonitrile (25 mL, 0.1 M) was added via syringe. CDI (810 mg, 5.0 mmol, 200 mol%) was added in one portion at ambient temperature. The mixture was stirred at 25 °C for 16 h. The reaction was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1) to furnish the title compounds as a white solid (418 mg, 1.68 mmol) in 82% yield.

<u>**TLC (SiO₂)</u>** $R_f = 0.31$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.07 (d, J = 2.6 Hz, 1H), 7.56 (dd, J = 8.7, 2.5 Hz, 1H), 6.78 (d, J = 8.7 Hz, 1H), 5.74 (dd, J = 17.6, 11.1 Hz, 1H), 5.38 (d, J = 11.1 Hz, 1H), 5.28 (s, 1H), 5.25 (d, J = 15.0 Hz, 1H), 4.42 (d, J = 11.0 Hz, 1H), 4.36 (d, J = 11.0 Hz, 1H), 3.97 (s, 3H), 1.04 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 164.7, 148.1, 146.0, 137.9, 133.7, 122.5, 118.7, 110.6, 85.0, 75.5, 53.6, 38.4, 19.0.

HRMS (ESI) Calcd. for C₁₃H₁₅NNaO₄⁺ [M+Na]⁺: 272.0894, Found: 272.0892

 $[\alpha]_{D}^{30}$: +110.9 (c = 0.61, CHCl₃)

m.p.: 102 – 104 °C

<u>FTIR</u> (neat): 1733, 1608, 1495, 1400, 1286, 1242, 1207, 1130, 1100, 1025, 940, 832, 768 cm⁻¹.



(4R,5R)-5-methyl-4-(2-phenylpyrimidin-5-yl)-5-vinyl-1,3-dioxan-2-one (3.1i)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with (1R,2R)-2methyl-1-(2-phenylpyrimidin-5-yl)-2-vinylpropane-1,3-diol⁴ (162 mg, 0.6 mmol, 100 mol%). Under argon atmosphere, acetonitrile (6 mL, 0.1 M) was added via syringe. CDI (194 mg, 1.2 mmol, 200 mol%) was added in one portion at ambient temperature. The mixture was stirred at 25 °C for 16 h. The reaction was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1) to furnish the title compounds as a white solid (128 mg, 0.43 mmol) in 72% yield.

<u>**TLC (SiO₂)</u>** $R_f = 0.25$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.75 (s, 2H), 8.52 – 8.40 (m, 2H), 7.57 – 7.48 (m, 3H), 5.82 (dd, J = 17.5, 11.0 Hz, 1H), 5.47 (d, J = 11.0 Hz, 1H), 5.39 (s, 1H), 5.29 (d, J = 17.5 Hz, 1H), 4.46 (d, J = 11.0 Hz, 1H), 4.43 (d, J = 11.1 Hz, 1H), 1.11 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 165.2, 156.1, 147.4, 136.7, 132.8, 131.3, 128.7, 128.4, 125.1, 119.9, 83.50, 75.7, 38.3, 18.6.

<u>HRMS</u> (ESI) Calcd. for $C_{17}H_{16}N_2NaO_3^+$ [M+Na]⁺: 319.1054, Found: 319.1059 [α]³⁰_D : +139.9 (c = 0.56, CHCl₃)

m.p.: 153 – 154 °C

<u>FTIR</u> (neat): 1736, 1722, 1588, 1544, 1434, 1398, 1242, 1211, 1133, 1102, 753, 730, 693 cm⁻¹.



Procedures and Spectral Data for the Model study of Cyclopropane Formation: methyl (1-(*p*-tolyl)but-3-en-1-yl) carbonate (*model*-3.1a)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with 1-(*p*-tolyl)but-3en-1-ol (324 mg, 2.0 mmol, 100 mol%) and 4-dimethylaminopyridine (439 mg, 3.6 mmol, 180 mol%). Under argon atmosphere, DCM (5 mL, 0.4 M) was added via syringe. Then, methyl chloroformate (0.23 mL, 3.0 mmol, 150 mol%) was added dropwise at 0 °C. The reaction mixture was stirred at ambient temperature for 16 h. Saturated aqueous ammonium chloride (15 mL) was added. The aqueous layer was extracted with ethyl acetate (30 mL x 2). The combined organic layers were washed with brine (20 mL), dried (Na₂SO₄), filtered and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1) to furnish the title compound as a colorless oil (356 mg, 1.7 mmol) in 86% yield.

<u>**TLC** (SiO₂</u>) $R_f = 0.35$ (hexanes/ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.25 (d, J = 8.0 Hz, 2H), 7.17 (d, J = 7.9 Hz, 2H), 5.72 (ddt, J = 17.2, 10.2, 7.0 Hz, 1H), 5.61 – 5.55 (m, 1H), 5.15 – 5.04 (m, 2H), 3.74 (s, 3H), 2.77 – 2.65 (m, 1H), 2.62 – 2.51 (m, 1H), 2.34 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 155.2, 138.0, 136.5, 133.0, 129.2, 126.5, 118.2, 79.4, 54.7, 40.6, 21.2.

<u>HRMS</u> (ESI) Calcd. for $C_{11}H_{13}^+$ [M-OCO₂Me]⁺: 145.1012, Found: 145.1011.

FTIR (neat): 2955, 1745, 1516, 1441, 1261, 1110, 1041, 939, 866, 791, 720 cm⁻¹.



(*E*)-4,4'-(but-1-ene-1,3-diyl)bis(methylbenzene) (SI-3.5)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with carbonate *model*-**3.1a** (22.0 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95% wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes) to furnish the known title compound⁷ as a colorless oil (7.3 mg, 0.03 mmol) in 31% yield.

<u>**TLC** (SiO</u>₂) $R_f = 0.28$ (hexanes).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.24 – 7.07 (m, 8H), 6.38 (d, J = 16.0 Hz, 1H), 6.31 (dd, J = 15.9, 6.2 Hz, 1H), 3.64 – 3.56 (m, 1H), 2.33 (s, 3H), 2.32 (s, 3H), 1.44 (d, J = 6.8 Hz, 3H).

<u>1³C NMR</u> (125 MHz, CDCl₃) δ 142.8, 136.7, 135.6, 134.8, 134.4, 129.1, 129.1, 128.1, 127.2, 126.0, 42.1, 21.3, 21.1, 21.0.

<u>**HRMS**</u> (CI) Calcd. for $C_{18}H_{20}^+$ [M]⁺: 236.1560, Found: 236.1570.

<u>FTIR</u> (neat): 3020, 2962, 2922, 2865, 1513, 1451, 1371, 1111, 1015, 967, 816, 798, 723 cm⁻¹



<u>Procedures and Spectral Data for the Synthesis of Enantiomerically Enriched</u> <u>Cyclopropanes 3.3a-3.3r:</u>

((1*S*,2*R*,3*S*)-1-methyl-2-(4-methylbenzyl)-3-(*p*-tolyl)cyclopropyl)methanol (3.3a)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (23.2 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1) to furnish the title compound as a white solid (23.8 mg, 0.08 mmol) in 85% yield. <u>**TLC (SiO₂)</u>** $R_f = 0.45$ (hexanes/ethyl acetate = 3:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.20 – 7.05 (m, 8H), 3.39 (dd, *J* = 11.6, 7.6 Hz, 1H), 3.24 (dd, *J* = 11.6, 4.4 Hz, 1H), 2.89 – 2.74 (m, 2H), 2.33 (s, 3H), 2.31 (s, 3H), 1.89 (d, *J* = 5.9 Hz, 1H), 1.52 (dd, *J* = 13.0, 7.1 Hz, 1H), 1.41 (s, 3H), 0.93 – 0.87 (m, 1H).

¹³C NMR (100 MHz, CDCl₃) δ 138.7, 135.7, 135.7, 135.3, 129.1, 129.1, 128.3, 128.0,
68.2, 35.1, 34.2, 29.6, 27.4, 21.0, 21.0, 17.2.

HRMS (ESI) Calcd. for C₂₀H₂₄NaO⁺ [M+Na]⁺: 303.1719, Found: 303.1720.

 $[\alpha]_{D}^{33}$: +24.0 (*c* = 1.0, CHCl₃).

m.p. : 59-60 °C

FTIR (neat): 3394, 2920, 2361, 2342, 1514, 1460, 1113, 1020, 825, 807, 759, 669 cm⁻¹

<u>**HPLC**</u> (two connected chiralcel OJ-H columns, hexanes:*i*-PrOH = 98:2, 1.0 mL/min, 230 nm), ee = 91%.







((1*S*,2*S*,3*R*)-2-(4-fluorophenyl)-1-methyl-3-(4-methylbenzyl)cyclopropyl)methanol (3.3b)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (23.6 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1) to furnish the title compound as a white solid (21.9 mg, 0.08 mmol) in 77% yield.

<u>TLC</u> (SiO₂) $R_f = 0.31$ (hexanes/ethyl acetate = 3:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.20 – 7.10 (m, 6H), 6.98 – 6.91 (m, 2H), 3.35 (d, J = 11.5 Hz, 1H), 3.22 (d, J = 11.5 Hz, 1H), 2.88 – 2.73 (m, 2H), 2.34 (s, 3H), 1.89 (d, J = 5.9 Hz, 1H), 1.50 – 1.44 (m, 1H), 1.41 (s, 3H).

¹³C NMR (100 MHz, CDCl₃) δ 161.4 (d, J = 244.5 Hz), 138.5, 135.4, 134.5 (d, J = 3.1 Hz), 130.0 (d, J = 7.8 Hz), 129.2, 128.0, 115.1 (d, J = 21.2 Hz), 68.1, 34.7, 34.1, 29.5, 27.9, 21.0, 17.1.

¹⁹F NMR (376 MHz, CDCl₃) δ -116.8 - -116.9 (m).

HRMS (CI) Calcd. for C₁₉H₂₀FO⁺ [M-H]⁺: 283.1493, Found: 283.1492.

 $[\alpha]_{D}^{33}$: +26.7 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3360, 2921, 1604, 1510, 1456, 1222, 1157, 1103, 1069, 1015, 838, 769 cm⁻¹

<u>HPLC</u> (chiralcel OJ-H columns, hexanes:*i*-PrOH = 95:5, 1.0 mL/min, 210 nm), ee = 93%



-116.86 -116.87 -116.88 -116.89 -116.91 -116.92 -116.93



30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)





((1S,2R,3S)-1-methyl-2-(4-methylbenzyl)-3-(p-tolyl)cyclopropyl)methanol (3.3c)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (35.4 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1) to furnish the title compound as a colorless oil (22.7 mg, 0.07 mmol) in 73% yield.

<u>**TLC** (SiO₂)</u> $R_f = 0.41$ (hexanes/ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl3) δ 7.68 (s, 1H), 7.59 (s, 2H), 7.23 – 7.04 (m, 4H), 3.36 (dd, J = 11.3, 4.9 Hz, 1H), 3.21 (dd, J = 11.3, 5.0 Hz, 1H), 2.84-2.86 (m, J = 7.2, 3.5 Hz, 2H), 2.33 (s, 3H), 1.97 (d, J = 6.0 Hz, 1H), 1.64 – 1.48 (m, 1H), 1.44 (s, 3H), 1.04 (t, J = 5.4 Hz, 1H).

1³C NMR (125 MHz, CDCl3) δ 141.8, 137.9, 135.8, 131.3 (q, J = 33.1 Hz), 129.3, 128.9 (d, J = 3.7 Hz), 127.9, 124.4, 122.2, 120.8 - 119.6 (m), 67.5, 34.9, 33.8, 30.5, 28.69, 20.9, 17.0.

<u>HRMS</u> (ESI) Calcd. for $C_{21}H_{20}F_6NaO^+$ [M+Na]⁺: 425.1311, Found: 425.1315.

 $[\alpha]_{D}^{32}$: +23.0 (*c* = 1.0, CHCl₃).

FTIR (neat): 3360, 2923, 1515, 1374, 1275, 1169, 1127, 1021, 894, 682 cm⁻¹.






((1S,2S,3R)-2-(benzo[d][1,3]dioxol-5-yl)-1-methyl-3-(4-

methylbenzyl)cyclopropyl)methanol (3.3d)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (26.2 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1) to furnish the title compound as a white solid (21.6 mg, 0.07 mmol) in 70% yield.

<u>TLC</u> (SiO₂) $R_f = 0.25$ (hexanes/ethyl acetate = 3:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.21 – 7.07 (m, 4H), 6.74 – 6.62 (m, 3H), 5.92 (s, 2H), 3.38 (dd, *J* = 11.5, 4.4 Hz, 1H), 3.25 (d, *J* = 11.5 Hz, 1H), 2.88 – 2.70 (m, 2H), 2.33 (s, 3H), 1.86 (d, *J* = 5.9 Hz, 1H), 1.46 – 1.40 (m, 1H), 1.39 (s, 3H), 0.98 (brs, 1H). <u>1³C NMR</u> (100 MHz, CDCl₃) δ 147.6, 145.9, 138.6, 135.4, 132.7, 129.2, 128.0, 121.4, 109.0, 108.1, 100.9, 68.2, 35.2, 34.1, 29.5, 27.8, 21.0, 17.1.

<u>HRMS</u> (CI) Calcd. for $C_{20}H_{22}O_3^+$ [M]⁺: 310.1563, Found: 310.1567.

 $[\alpha]_{D}^{33}$: +25.7 (*c* = 1.0, CHCl₃).

FTIR (neat): 3430, 2919, 1608 1503, 1490, 1441, 1234, 1189, 1039, 935, 808 cm⁻¹

<u>**HPLC</u>** (chiralcel AS-H columns, hexanes:*i*-PrOH = 95:5, 1.0 mL/min, 210 nm), ee = 91%</u>









((1*S*,2*R*,3*S*)-1-methyl-2-(4-methylbenzyl)-3-(*p*-tolyl)cyclopropyl)methanol (3.3e)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (27.4 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1) to furnish the title compound as a colorless oil (15.7 mg, 0.068 mmol) in 68% yield.

<u>**TLC (SiO**</u>₂) $R_f = 0.26$ (hexanes/ethyl acetate = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.75 – 7.69 (m, 1H), 7.66 – 7.60 (m, 1H), 7.36 – 7.27 (m, 2H), 7.20 (d, *J* = 8.0 Hz, 2H), 7.13 (d, *J* = 8.0 Hz, 2H), 6.96 (t, *J* = 1.0 Hz, 1H), 3.56 (d, *J* = 11.8 Hz, 1H), 3.40 (d, *J* = 11.8 Hz, 1H), 2.91 – 2.80 (m, 2H), 2.34 (s, 3H), 2.03 (d, *J* = 5.7 Hz, 1H), 1.61 (td, *J* = 7.1, 5.7 Hz, 1H), 1.44 (s, 3H).

<u>1³C NMR</u> (125 MHz, CDCl₃) δ 143.9, 140.0, 139.2, 138.0, 135.6, 129.2, 128.0, 124.3,

123.8, 122.9, 122.0, 121.5, 67.9, 33.8, 31.3, 30.4 (d, *J* = 7.9 Hz), 21.0, 16.6.

HRMS (ESI) Calcd. for C₂₁H₂₂NaOS⁺ [M+Na]⁺: 345.1284, Found: 345.1295.

 $[\alpha]_{D}^{33}$: +17.0 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 2285, 2922, 2359, 2340, 1514, 1457, 1436, 1068, 1020, 805, 746, 668 cm⁻¹







tert-butyl 5-((1*S*,2*S*,3*R*)-2-(hydroxymethyl)-2-methyl-3-(4methylbenzyl)cyclopropyl)-1H-indole-1-carboxylate (3.3f)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1e** (35.7 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxizne (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 70 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 12:1) to furnish the title compound as a colorless oil (28.4 mg, 0.07 mmol) in 70% yield. <u>**TLC (SiO₂)</u>** $R_f = 0.43$ (hexanes/ethyl acetate = 4:1).</u>

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.02 (d, *J* = 8.5 Hz, 1H), 7.56 (d, *J* = 3.7 Hz, 1H), 7.38 – 7.34 (m, 1H), 7.23 – 7.16 (m, 3H), 7.12 (d, *J* = 7.7 Hz, 2H), 6.48 (d, *J* = 3.7 Hz, 1H), 3.39 (dd, *J* = 11.6, 7.5 Hz, 1H), 3.24 (dd, *J* = 11.6, 4.2 Hz, 1H), 2.85 (d, *J* = 7.0 Hz, 2H), 2.33 (s, 3H), 2.03 (d, *J* = 5.8 Hz, 1H), tbs 1.66 (s, 9H), 1.60 (td, *J* = 7.1, 5.8 Hz, 1H), 1.43 (s, 3H), 0.87 (dd, *J* = 7.8, 5.0 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃) δ 149.7, 138.8, 135.4, 133.8, 133.1, 130.8, 129.2, 128.0, 126.2, 125.1, 120.2, 115.0, 107.1, 83.6, 68.3, 35.5, 34.2, 29.6, 28.2, 27.5, 21.0, 17.2. **HRMS** (ESI) Calcd. for C₂₆H₃₁NO₃⁺ [M+Na]⁺: 428.2196, Found: 428.2200. [α]_D²⁵: +33.2 (*c* = 1.0, CHCl₃).

FTIR (neat): 3394, 2976, 1731, 1473, 1369, 1251, 1163, 1132, 1022, 746 cm⁻¹





((1S,2S,3R)-2-(2,3-dimethylquinoxalin-6-yl)-1-methyl-3-(4-

methylbenzyl)cyclopropyl)methanol (3.3g)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1g** (29.8 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1) to furnish the title compound as a yellow solid (21.4 mg, 0.06 mmol) in 62% yield.

<u>TLC</u> (SiO₂) $R_f = 0.29$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.84 (d, J = 8.6 Hz, 1H), 7.69 (s, 1H), 7.56 (dd, J = 8.5, 1.3 Hz, 1H), 7.20 – 7.07 (m, 4H), 3.40 (d, J = 11.3 Hz, 1H), 3.29 (d, J = 11.5 Hz, 1H),

2.92 – 2.81 (m, 2H), 2.69 (s, 3H), 2.69 (s, 3H), 2.32 (s, 3H), 2.09 (d, *J* = 5.9 Hz, 1H), 1.75 – 1.71 (m, 1H), 1.47 (s, 3H), 1.13 (brs, 1H).

¹³C NMR (125 MHz, CDCl₃) δ 153.5, 152.8, 140.9, 140.4, 139.8, 138.3, 135.5, 130.7,

129.2, 128.0, 127.9, 126.1, 67.6, 35.7, 34.1, 31.0, 28.1, 23.1, 23.0, 21.0, 17.2.

<u>HRMS</u> (ESI) Calcd. for $C_{23}H_{27}N_2O^+$ [M+H]⁺: 347.2118, Found: 347.2120.

 $[\alpha]_{D}^{33}$: +25.0 (*c* = 1.0, CHCl₃).

m.p. : 152-153 °C

<u>FTIR</u> (neat): 3350, 2920, 2866, 2360, 2343, 1619, 1556, 1514, 1498, 1449, 1379, 1334, 1185, 1157, 1041, 1022, 837, 806, 669 cm⁻¹





f1 (ppm)

((1S,2S,3R)-2-(6-methoxypyridin-3-yl)-1-methyl-3-(4-

methylbenzyl)cyclopropyl)methanol (3.3h)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1h** (24.9 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1) to furnish the title compound as oil (27.2 mg, 0.92 mmol) in 92% yield.

<u>TLC</u> (SiO₂) $R_f = 0.10$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.95 (d, J = 2.5 Hz, 1H), 7.38 (ddd, J = 8.5, 2.5, 0.7 Hz, 1H), 7.17 (d, J = 8.0 Hz, 2H), 7.11 (d, J = 8.0 Hz, 2H), 6.63 (dd, J = 8.5, 0.7 Hz, 1H), 3.89 (s, 3H), 3.35 (dd, J = 11.6, 4.7 Hz, 1H), 3.22 (dd, J = 11.5, 3.8 Hz, 1H), 2.87 (dd, J = 15.0, 6.6 Hz, 1H), 2.73 (dd, J = 15.0, 7.6 Hz, 1H), 2.33 (s, 3H), 1.79 (d, J = 5.9 Hz, 1H), 1.46 – 1.41 (m, 1H), 1.40 (s, 3H), 1.13 (t, J = 5.4 Hz, 1H).

1³C NMR (100 MHz, CDCl₃) δ 162.8, 146.6, 139.4, 138.4, 135.5, 129.2, 128.0, 127.1, 110.2, 68.0, 53.3, 34.2, 32.0, 29.0, 27.8, 21.0, 17.0.

HRMS (ESI) Calcd. for C₁₉H₂₃NNaO₂⁺ [M+Na]⁺: 320.1621, Found: 320.1625

 $[\alpha]_{D}^{31}$: +100.6 (c = 1.3, CHCl₃)

FTIR (neat): 1607, 1494, 1408, 1373, 1316, 1283, 1258, 1109, 1022, 814, 726 cm⁻¹.

<u>**HPLC</u>** (chiralcel AS-H columns, hexanes:*i*-PrOH = 95:5, 1.0 mL/min, 280 nm), ee = 91%</u>





f1 (ppm)

186



((1S,2R,3S)-1-methyl-2-(4-methylbenzyl)-3-(2-phenylpyrimidin-5-

yl)cyclopropyl)methanol (3.3i)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1i** (29.6 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (30.1 mg, 0.085 mmol, 85 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1) to furnish the title compound (29.2 mg, 0.85 mmol) in 85% yield.

<u>**TLC (SiO**</u>₂) $R_f = 0.10$ (hexanes/ethyl acetate = 4:1).

<u>¹H NMR</u> (400 MHz, CDCl₃) δ 8.58 (s, 2H), 8.41 – 8.33 (m, 2H), 7.47 (s, 3H), 7.15 (s, 4H), 3.45 (d, *J* = 11.3 Hz, 1H), 3.22 (d, *J* = 11.3 Hz, 1H), 2.93 (dd, *J* = 14.9, 6.5 Hz, 1H), 2.75 (dd, *J* = 14.9, 7.8 Hz, 1H), 2.34 (s, 3H), 1.81 (d, *J* = 5.9 Hz, 1H), 1.54 (dt, *J* = 7.9, 6.3 Hz, 1H), 1.45 (s, 3H), 1.42 (s, 1H).

<u>1³C NMR</u> (100 MHz, CDCl₃) δ 162.5, 157.6, 138.0, 137.4, 135.7, 130.5, 130.2, 129.4, 128.6, 127.9, 127.9, 67.4, 34.0, 30.0, 29.8, 28.1, 21.0, 17.0.

HRMS (ESI) Calcd. for C₂₃H₂₄N₂NaO⁺ [M+Na]⁺: 367.1781, Found: 367.1784

 $[\alpha]_{D}^{31}$: +120.3 (c = 1.1, CHCl₃)

<u>FTIR</u> (neat): 2361, 2343, 2331, 1515, 1438, 1421, 1024, 748, 693, 669 cm⁻¹.







((*1S*,*2S*,*3R*)-1-methyl-2-(p-tolyl)-3-(4-(trifluoromethyl)benzyl)cyclopropyl)methanol (3.3j)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (23.3 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (43.9 mg, 0.085 mmol, 85 mol%) **3.2b**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1) to furnish the title compound as oil (24.7 mg, 0.74 mmol) in 74% yield. <u>**TLC (SiO₂)**</u> $R_f = 0.30$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.55 (d, J = 7.9 Hz, 2H), 7.41 (d, J = 7.9 Hz, 2H), 7.11 – 7.03 (m, 4H), 3.39 (dd, J = 11.9, 5.6 Hz, 1H), 3.26 (d, J = 11.4 Hz, 1H), 2.95 (dd, J = 15.4, 6.8 Hz, 1H), 2.87 (dd, J = 15.4, 7.5 Hz, 1H), 2.31 (s, 3H), 1.92 (d, J = 5.8 Hz, 1H), 1.55 (q, J = 6.8 Hz, 1H), 1.41 (s, 3H), 0.95 (s, 1H).

¹³C NMR (125 MHz, CDCl₃) δ 145.9, 136.0, 135.2, 129.2, 128.5, 128.5, 128.2, 125.5, 125.4, 125.4, 125.4, 123.3, 68.0, 35.1, 34.5, 29.6, 26.7, 21.0, 17.2.

HRMS (ESI) Calcd. for C₂₀H₂₁F₃NaO⁺ [M+Na]⁺: 357.1437, Found: 357.1442

 $[\alpha]_{D}^{29}$: +19.7 (c = 0.76, CHCl₃)

FTIR (neat): 1515, 1324, 1275, 1161, 1121, 1067, 1018, 913, 819, 749 cm⁻¹.







((1*S*,2*R*,3*S*)-2-(4-methoxybenzyl)-1-methyl-3-(*p*-tolyl)cyclopropyl)methanol (3.3k)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (29.8 mg, 0.1 mmol, 100 mol%), tri(*p*-methoxyphenyl)boroxine (34.2 mg, 0.085 mmol, 85 mol%) **3.2c**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95% wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 8:1) to furnish the title compound as a yellow solid (22.5 mg, 0.08 mmol) in 76% yield.

<u>**TLC (SiO₂)**</u> $R_f = 0.30$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.23 – 7.18 (m, 2H), 7.10 – 7.04 (m, 4H), 6.87 – 6.82 (m, 2H), 3.79 (s, 3H), 3.38 (d, *J* = 11.6 Hz, 1H), 3.24 (d, *J* = 11.6 Hz, 1H), 2.82 (dd, *J* = 15.1, 6.9 Hz, 1H), 2.76 (dd, *J* = 15.1, 7.3 Hz, 1H), 2.31 (s, 3H), 1.89 (d, *J* = 5.9 Hz, 1H), 1.51 (dd, *J* = 13.1, 7.1 Hz, 1H), 1.40 (s, 3H), 0.93 (brs, 1H).

<u>1³C NMR</u> (100 MHz, CDCl₃) δ 157.8, 135.7, 135.7, 133.9, 129.1, 129.0, 128.3, 113.9,
68.2, 55.2, 35.1, 33.7, 29.6, 27.5, 21.0, 17.2.

<u>HRMS</u> (CI) Calcd. for $C_{20}H_{24}O_2^+$ [M]⁺: 296.1771, Found: 296.1771.

 $[\alpha]_{D}^{33}$: +29.0 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3414, 2921, 1611, 1511, 1463, 1301, 1244, 1177, 1035, 824, 744 cm⁻¹




((1S,2R,3S)-2-cinnamyl-1-methyl-3-(p-tolyl)cyclopropyl)methanol (3.3l)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (23.2 mg, 0.1 mmol, 100 mol%), tri(phenylvinyl)boroxine **3.2d** (33.2 mg, 0.085 mmol, 85 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 65 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1) to furnish the title compound as a colorless oil (18.7 mg, 0.064 mmol) in 64% yield.

<u>**TLC (SiO**</u>₂) $R_f = 0.35$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.37 – 7.26 (m, 4H), 7.22 – 7.05 (m, 5H), 6.54 – 6.43 (m, 1H), 6.32 (dt, *J* = 15.8, 6.5 Hz, 1H), 3.40 (d, *J* = 11.6 Hz, 1H), 3.24 (d, *J* = 11.6 Hz, 1H), 2.41-2.39 (m, 2H), 2.31 (s, 3H), 1.80 (d, *J* = 5.9 Hz, 1H), 1.41-1.36(m, 1H), 1.37 (s, 3H) ¹³<u>C NMR</u> (100 MHz, CDCl₃) δ 137.6, 135.7 (d, *J* = 2.3 Hz), 130.1, 129.6, 129.1, 128.4, 128.2, 126.9, 126.0, 68.1, 34.7, 32.0, 29.6, 25.6, 20.9, 16.9.

HRMS (ESI) Calcd. for C₂₁H₂₄NaO⁺ [M+Na]⁺: 315.1719, Found: 315.1729.

 $[\alpha]_{D}^{33}$: +9.7 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3406, 2922, 1514, 1448, 1378, 1019, 963, 825, 803, 741, 692 cm⁻¹.





203

((1S,2S,3R)-2-(6-methoxypyridin-3-yl)-1-methyl-3-(4-

(trifluoromethyl)benzyl)cyclopropyl)methanol (3.3m)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1h** (24.9 mg, 0.1 mmol, 100 mol%), tri(p-tolyl)boroxine (43.9 mg, 0.085 mmol, 85 mol%) **3.2b**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated in *vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1,) to furnish the title compound as oil (25.2 mg, 0.72 mmol) in 72% yield.

<u>TLC</u> (SiO₂) $R_f = 0.10$ (hexanes/ethyl acetate = 3:1).

¹<u>H NMR</u> 1H NMR (500 MHz, CDCl₃) δ 7.94 (d, J = 2.4 Hz, 1H), 7.56 (d, J = 8.0 Hz, 2H), 7.40 (d, J = 8.6 Hz, 2H), 7.38 (dd, J = 6.4, 2.3 Hz, 1H), 6.64 (d, J = 8.4 Hz, 1H),

3.89 (s, 3H), 3.36 (d, *J* = 11.4 Hz, 1H), 3.25 (d, *J* = 11.4 Hz, 1H), 2.97 (dd, *J* = 15.2, 6.6 Hz, 1H), 2.84 (dd, *J* = 15.2, 7.7 Hz, 1H), 1.82 (d, *J* = 5.8 Hz, 1H), 1.46 (q, *J* = 6.5 Hz, 1H), 1.40 (s, 3H), 1.10 (s, 1H).

13C NMR (125 MHz, CDCl₃) δ 163.1, 146.7, 145.8, 139.4, 128.6 (q, J^{3}_{CF} = 32.5 Hz), 128.6, 126.7, 125.6 (q, J^{1}_{CF} = 3.8 Hz), 124.4 (q, J^{2}_{CF} = 271 Hz, 110.5, 67.9, 53.5, 34.7, 32.2, 29.1, 27.2, 17.1.

¹⁹**F NMR** (470 MHz, CDCl₃) δ -62.3.

HRMS (ESI) Calcd. for C₁₉H₂₀F₃NNaO₂⁺ [M+Na]⁺: 374.1339, Found: 374.1343

 $[\alpha]_{D}^{29}$: +5.7 (c = 0.82, CHCl₃)

<u>FTIR</u> (neat): 1607, 1495, 1375, 1324, 1286, 1259, 1161, 1120, 1067, 1018, 832, 732 cm⁻¹







((1S,2R,3S)-2-(4-methoxybenzyl)-3-(6-methoxypyridin-3-yl)-1-

methylcyclopropyl)methanol (3.3n)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1h** (29.8 mg, 0.1 mmol, 100 mol%), tri(*p*-tolyl)boroxine **3.2d** (34.2 mg, 0.085 mmol, 85 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a yellow solid (23.2 mg, 0.07 mmol) in 74% yield.

<u>TLC</u> (SiO₂) $R_f = 0.24$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.95 (d, J = 2.5 Hz, 1H), 7.38 (dd, J = 8.4, 2.3 Hz, 1H), 7.22 - 7.15 (m, 2H), 6.90 - 6.80 (m, 2H), 6.63 (d, J = 8.5 Hz, 1H), 3.89 (s, 3H), 3.79 (s, 3H), 3.34 (d, J = 11.4 Hz, 1H), 3.22 (d, J = 11.4 Hz, 1H), 2.85 (dd, J = 14.9, 6.6 Hz, 1H), 2.71 (dd, *J* = 15.0, 7.7 Hz, 1H), 1.78 (d, *J* = 5.8 Hz, 1H), 1.44 – 1.37 (m, 1H), 1.40 (s, 3H), 1.10 (brs, 1H).

1³C NMR (125 MHz, CDCl₃) δ 162.8, 157.9, 146.6, 139.4, 133.6, 129.0, 127.0, 114.0, 110.2, 68.0, 55.2, 53.3, 33.7, 32.0, 29.0, 28.0, 16.9.

<u>HRMS</u> (ESI) Calcd. for $C_{19}H_{24}NO_3^+$ [M+H]⁺: 314.1751, Found: 314.1754.

 $[\alpha]_{D}^{33}$: +9.0 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3351, 2949, 1606, 1511, 1495, 1374, 1284, 1245, 1177, 1031, 830 cm⁻¹







((1*S*,2*R*,3*S*)-2-cinnamyl-1-methyl-3-(*p*-tolyl)cyclopropyl)methanol (3.30)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1h** (23.2 mg, 0.1 mmol, 100 mol%), tri(phenylvinyl)boroxine **3.2d** (33.2 mg, 0.085 mmol, 85 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 60 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1) to furnish the title compound as a colorless oil (18.9 mg, 0.061 mmol) in 61% yield.

<u>TLC</u> (SiO₂) $R_f = 0.40$ (hexanes/ethyl acetate = 3:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.05 (d, J = 2.4 Hz, 1H), 7.45 (dd, J = 8.6, 2.5 Hz, 1H), 7.39 – 7.27 (m, 4H), 7.24 – 7.19 (m, 1H), 6.66 (d, J = 8.4 Hz, 1H), 6.52 – 6.46 (m, 1H), 6.32 (dt, J = 15.8, 6.5 Hz, 1H), 3.38 (d, J = 11.4 Hz, 3H), 3.24 (d, J = 11.5 Hz, 1H), 2.43-2.42 (m, 2H), 1.71 (d, J = 5.8 Hz, 1H), 1.37 (s, 3H), 1.31 (td, J = 7.2, 5.9 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 162.8, 146.6, 139.2, 137.5, 130.4, 129.2, 128.5, 127.0, 126.0, 110.2, 67.9, 53.3, 32.0, 31.7, 28.9, 25.8, 16.7.

<u>HRMS</u> (ESI) Calcd. for $C_{20}H_{23}NaNO_2^+$ [M+Na]⁺: 310.1802, Found: 310.1800.

 $[\alpha]_{D}^{33}$: +5.5 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3415, 2951, 1605, 1494, 1374, 1284, 1027, 964, 831, 742, 693 cm⁻¹.







((1*S*,2*R*,3*S*)-1-methyl-2-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)methyl)-3- (*p*-

tolyl)cyclopropyl)methanol (3.3p)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1a** (29.8 mg, 0.1 mmol, 100 mol%), B₂pin₂ (50.8 mg, 0.2 mmol, 200 mol%) **3.2a**, dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95% wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1) to furnish the title compound as a yellow solid (26.8 mg, 0.08 mmol) in 85% yield.

<u>**TLC (SiO**</u>₂) $R_f = 0.30$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃) δ 7.10 – 7.02 (m, 4H), 3.37 (dd, J = 10.9, 7.2 Hz, 1H), 3.01 (d, J = 10.9 Hz, 1H), 2.30 (s, 3H), 2.06 (brs, 1H), 1.35 – 1.31 (m, 1H), 1.29 – 1.25 (m, 16H), 0.78 – 0.71 (m, 1H).

¹³C NMR (100 MHz, CDCl₃) δ 136.1, 135.3, 128.8, 128.4, 83.5, 69.0, 36.6, 28.5, 25.0, 24.8, 24.8, 22.6, 21.0, 17.1.

<u>**HRMS**</u> (CI) Calcd. for $C_{19}H_{29}BO_3^+$ [M]⁺: 316.2204, Found: 316.2204.

 $[\alpha]_{D}^{33}$: +38.7 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3497, 2977, 2925, 1515, 1364, 1319, 1143, 1018, 967, 883, 848, 820, 748, 675 cm⁻¹





((1*S*,2*S*,3*R*)-2-(6-methoxypyridin-3-yl)-1-methyl-3-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)methyl)cyclopropyl)methanol (3.3q)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1h** (24.9 mg, 0.1 mmol, 100 mol%), bis(pinacolato)diboron (50.8 mg, 0.2 mmol, 200 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95% wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 45 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, CH₂Cl₂: acetone = 20:1–10:1) to furnish the title compound as a colorless oil (24.0 mg, 0.07 mmol) in 72% yield.

<u>**TLC** (SiO₂</u>) $R_f = 0.41$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.00 (d, J = 2.4 Hz, 1H), 7.39 (dd, J = 8.5, 2.5 Hz, 1H), 6.62 (d, J = 8.5 Hz, 1H), 3.90 (s, 3H), 3.39 (dd, J = 10.9, 8.0 Hz, 1H), 2.97 (dd, J = 10.9, 2.9 Hz, 1H), 2.09 (dd, J = 8.2, 3.1 Hz, 1H), 1.60 (d, J = 5.4 Hz, 1H), 1.34 – 1.29 (m, J = 5.9 Hz, 1H), 1.27 (d, J = 3.2 Hz, 12H), 1.23 (s, 3H), 0.81 – 0.70 (m, 1H). <u>1³C NMR</u> (125 MHz, CDCl₃) δ 162.7, 146.7, 139.2, 127.5, 110.0, 83.6, 68.9, 53.3, 33.5, 28.0, 24.8, 24.8, 22.9, 16.9.

HRMS (ESI) Calcd. for C₁₈H₂₈BNO₄ [M+H]⁺: 334.2187, Found: 334.2192.

 $[\alpha]_{D}^{25}$: +33.7 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3433, 2978, 1606, 1495, 1371, 1284, 1143, 1030, 967, 846, 755 cm⁻¹







223

tert-butyl 5-((1*S*,2*S*,3*R*)-2-(hydroxymethyl)-2-methyl-3-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)methyl)cyclopropyl)-1H-indole-1-carboxylate (3.3r)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1e** (35.7 mg, 0.1 mmol, 100 mol%), bis(pinacolato)diboron (50.8 mg, 0.2 mmol, 200 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 75 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1–10:1) to furnish the title compound as a colorless oil (33.1 mg, 0.08 mmol) in 75% yield.

<u>TLC</u> (SiO₂) $R_f = 0.43$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.99 (d, *J* = 8.5 Hz, 1H), 7.55 (d, *J* = 3.7 Hz, 1H), 7.38 (s, 1H), 7.17 (dd, *J* = 8.5, 1.6 Hz, 1H), 6.48 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.00 (d, *J* = 3.7 Hz, 1H), 3.43 – 3.32 (m, 1H), 3.43 – 3.32 (m, 1H), 3.43 – 3.41 (m, 1H), 3.4

J = 11.0 Hz, 1H), 2.04 (d, *J* = 5.3 Hz, 1H), 1.83 (d, *J* = 6.0 Hz, 1H), 1.66 (s, 9H), 1.45 – 1.33 (m, 2H), 1.30 (s, 3H), 1.28 (s, 12H), 1.24 (s, 1H), 0.79 (dd, *J* = 16.9, 9.8 Hz, 1H). <u>1³C NMR</u> (125 MHz, CDCl₃) δ 149.7, 133.6, 130.6, 126.0, 125.3, 120.4, 114.6, 107.1, 83.5, 69.1, 36.9, 29.7, 28.5, 28.2, 24.8, 24.8, 22.8, 17.1.

HRMS (ESI) Calcd. for C₂₅H₃₆BNO₅ [M+K]⁺: 480.2323, Found: 480.2339.

 $[\alpha]_{D}^{24}$: +31.7 (*c* = 1.0, CHCl₃).

<u>FTIR</u> (neat): 3486, 2978, 1732, 1474, 1369, 1264, 1216, 1133, 1023, 797 cm⁻¹





f1 (ppm)

((1S,2S,3R)-2-(4-fluorophenyl)-1-methyl-3-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-

2-yl)methyl)cyclopropyl)methanol (3.3s)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1b** (23.6 mg, 0.1 mmol, 100 mol%), bis(pinacolato)diboron (50.8 mg, 0.2 mmol, 200 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1) to furnish the title compound as a colorless oil (24.3 mg, 0.076 mmol) in 76% yield.

<u>**TLC** (SiO₂)</u> $R_f = 0.7$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 7.15 (dd, J = 8.4, 5.5 Hz, 2H), 6.92 (t, J = 8.7 Hz, 2H), 3.37 (dd, J = 11.0, 7.7 Hz, 1H), 2.96 (dd, J = 11.0, 2.7 Hz, 1H), 2.06 (dd, J = 8.0, 3.2 Hz, 1H), 1.68 (d, J = 5.8 Hz, 1H), 1.32 – 1.29 (m, 1H), 1.28 – 1.25 (m, 16H), 0.75 (dd, J = 18.2, 11.2 Hz, 1H).

 $\frac{^{13}\text{C NMR}}{^{13}\text{C}}$ (125 MHz, CDCl₃) δ 161.4 (d, J^{1}_{CF} = 243.9 Hz), 135.1 (d, J^{4}_{CF} = 3.1 Hz), 130.2 (d, J^{3}_{CF} = 7.8 Hz), 115.0 (d, J^{2}_{CF} = 21.1 Hz), 83.7, 69.1, 36.4, 28.5, 25.2, 25.0, 24.9, 23.2, 17.1.

HRMS (ESI) Calcd. for C₁₈H₂₆BFNaO₃ [M+Na]⁺: 343.1851, Found: 343.1857.

 $[\alpha]_{D}^{24}$: +25.6 (*c* = 1.2, CHCl₃).

FTIR (neat): 1510, 1363, 1319, 1215, 1142, 1125, 1017, 967, 848, 756 cm⁻¹





f1 (ppm)



((1*S*,2*S*,3*R*)-2-(benzo[d][1,3]dioxol-5-yl)-1-methyl-3-((4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)methyl)cyclopropyl)methanol (3.3t)



Detailed Procedures

An oven-dried sealed tube equipped with a magnetic stir bar was charged with vinyldioxanone **3.1d** (26.2 mg, 0.1 mmol, 100 mol%), bis(pinacolato)diboron (50.8 mg, 0.2 mmol, 200 mol%), dicyclohexylphenylphosphine (5.8 mg, 0.02 mmol, 20 mol%, 95%wt) and potassium phosphate (42.4 mg, 0.2 mmol, 200 mol%). This tube was transferred into an argon-atmosphere glovebox. Ni(cod)₂ (2.8 mg, 0.01 mmol, 10 mol%) and toluene (0.5 mL, 0.2 M) were added. After sealing the tube with cap, the tube was removed from the glovebox and the reaction mixture was stirred at 55 °C for 48 h. The reaction was cooled to ambient temperature, diluted with Et₂O (1 mL) and filtered through a plug of silica gel, which was rinsed with Et₂O (2 mL). The combined organic solution was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1) to furnish the title compound as a colorless oil (27.3 mg, 0.079 mmol) in 79% yield.

<u>**TLC** (SiO₂)</u> $R_f = 0.25$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 6.70 – 6.63 (m, 3H), 5.90 (s, 2H), 3.38 (dd, *J* = 10.9, 7.5 Hz, 1H), 3.00 (dd, *J* = 11.1, 2.8 Hz, 1H), 2.05 (t, *J* = 7.5, 3.0 Hz, 1H), 1.65 (d, *J* = 5.5 Hz, 1H), 1.28 – 1.23 (m, 17H), 0.73 (dd, *J* = 18.5, 11.2 Hz, 1H).

<u>1³C NMR</u> (125 MHz, CDCl₃) δ 147.5, 145.9, 133.4, 121.6, 109.4, 108.0, 100.9, 83.7,
69.2, 36.9, 28.5, 25.2, 25.0, 24.9, 23.2, 17.1.

HRMS (ESI) Calcd. for C₁₉H₂₇BNaO₅ [M+Na]⁺: 369.1844, Found: 369.1838.

 $[\alpha]_{D}^{24}$: +57.3 (*c* = 1.1, CHCl₃).

<u>FTIR</u> (neat): 1504, 1491, 1441, 1365, 1316, 1232, 1193, 1142, 1038, 936, 879, 810, 759, 733 cm⁻¹




Procedures and Spectral Data for the Synthesis of 3.4a-3.4b:

(1*S*,2*R*,3*S*)-1-methyl-2-(4-methylbenzyl)-3-(*p*-tolyl)cyclopropane-1-carboxylic acid (3.4a)



Detailed Procedures

A vial equipped with a magnetic stir bar was charged with **3.3a** (22.4 mg, 0.08 mmol, 100 mol%). Under argon atmosphere, acetone (0.8 mL, 0.1 M) was added via syringe. The mixture was cooled to 0 °C and freshly prepared H₂CrO₄ (0.16 mL, 2.5 M, 500 mol%) was added dropwise. The reaction mixture was stirred at ambient temperature for 4 h. 2-propanol (0.5 mL) was slowly added. The mixture was filtered through a plug of sodium sulfate, which was rinsed with ethyl acetate (2 mL). The filtrate was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 3:1) to furnish the title compound as a colorless oil (17.8 mg, 0.06 mmol) in 76% yield.

<u>TLC (SiO</u>) $R_f = 0.23$ (hexanes/ethyl acetate = 3:1).

<u>¹H NMR</u> (500 MHz, CDCl₃) δ 7.18 – 7.08 (m, 4H), 7.00 (s, 4H), 2.87 (dd, *J* = 15.1, 6.8 Hz, 1H), 2.79 (dd, *J* = 15.1, 7.7 Hz, 1H), 2.47 – 2.40 (m, 1H), 2.33 (s, 3H), 2.29 (s, 3H), 2.24 (d, *J* = 7.5 Hz, 1H), 1.49 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 177.4, 137.4, 136.0, 135.6, 133.6, 129.2, 128.8, 128.6, 128.1, 41.0, 34.2, 31.4, 30.8, 21.1, 21.0, 15.6.

HRMS (ESI) Calcd. for C₂₀H₂₂NaO₂⁺ [M+Na]⁺: 317.1512, Found: 317.1512.

 $[\alpha]_{D}^{34}$: -10.0 (*c* = 0.88, CHCl₃).

<u>FTIR</u> (neat): 2921, 2361, 1686, 1515, 1461, 1419, 1306, 1218, 1119, 1021, 915, 807 cm⁻¹



2-(((1S,2R,3R)-2-(6-methoxypyridin-3-yl)-1-methyl-3-(4-

methylbenzyl)cyclopropyl)methyl) isoindoline-1,3-dione (3.4b)



Detailed Procedures

An oven-dried vial equipped with a magnetic stir bar was charged with **3.3h** (29.7 mg, 0.1 mmol, 100 mol%), triphenylphosphine (39.3 mg, 0.15 mmol, 150 mol%), phthalimide (22.1 mg, 0.15 mmol, 150 mol%). Under argon atmosphere, THF (1 mL, 0.1 M) was added via syringe. DEAD (65.3 mg, 0.15 mmol, 150 mol%, 40% w/w in toluene) was added slowly at ambient temperature. The mixture was stirred at 25 °C for 2 h. Saturated NaHCO₃ was added and the mixture was extracted by EA (20 mL x 2), dried (Na₂SO₄), filtered and concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1) to furnish the title compounds as oil (37.4 mg, 0.88 mmol) in 88% yield.

<u>TLC</u> (SiO₂) $R_f = 0.20$ (hexanes/ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃) δ 8.08 (d, J = 2.5 Hz, 1H), 7.77 (s, 2H), 7.71 (s, 2H), 7.50 (dd, J = 8.5, 2.4 Hz, 1H), 7.05 (d, J = 7.7 Hz, 2H), 6.86 (d, J = 7.7 Hz, 2H), 6.69 (d, J = 8.5 Hz, 1H), 3.91 (s, 3H), 3.70 (d, J = 14.2 Hz, 1H), 2.92 (d, J = 14.2 Hz, 1H), 2.70 (d, J = 7.3 Hz, 1H), 2.17 – 2.14 (m, 4H), 2.00 (q, J = 7.1 Hz, 1H), 1.82 (d, J = 6.3 Hz, 1H), 1.32 (s, 3H).

¹³C NMR (125 MHz, CDCl₃) δ 168.7, 163.0, 147.3, 139.4, 138.5, 135.2, 133.8, 132.2, 129.0, 128.2, 126.8, 123.2, 110.4, 53.5, 44.4, 34.5, 33.1, 29.8, 27.2, 21.1, 19.0.
HRMS (ESI) Calcd. for C₂₇H₂₆N₂NaO₃⁺ [M+Na]⁺: 449.1836, Found: 449.1840

 $[\alpha]_{D}^{30}$: +97.0 (c = 1.0, CHCl₃)

<u>FTIR</u> (neat): 1709, 1493, 1396, 1383, 1349, 1285, 1259, 1060, 1024, 927, 833, 732, 711 cm⁻¹



Crystallographic Material for 3.3a:

Table 1. Crystal data and structure refinement for 1.			
Empirical formula	C20 H24 O		
Formula weight	280.39		
Temperature	100(2) K		
Wavelength	1.54184 Å		
Crystal system	monoclinic		
Space group	Ι2		
Unit cell dimensions	a = 34.187(2) Å	$\Box = 90^{\circ}$.	
	b = 5.8361(4) Å	$\Box = 91.395(7)^{\circ}.$	
	c = 33.294(3) Å	$\Box = 90^{\circ}.$	
Volume	6640.8(8) Å ³		
Z	16		
Density (calculated)	1.122 Mg/m ³		
Absorption coefficient	0.510 mm ⁻¹		
F(000)	2432		
Crystal size	0.350 x 0.038 x 0.032 mm ³		
Theta range for data collection	2.586 to 75.298°.		
Index ranges	-42<=h<=41, -3<=k<=7, -41<=l<=41		
Reflections collected	11258		
Independent reflections	8398 [R(int) = 0.0792]		
Completeness to theta = 67.684°	98.6 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	1.00 and 0.408 242		

Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	8398 / 505 / 773
Goodness-of-fit on F ²	1.122
Final R indices [I>2sigma(I)]	R1 = 0.0897, wR2 = 0.2422
R indices (all data)	R1 = 0.1131, wR2 = 0.2616
Absolute structure parameter	-0.3(6)
Extinction coefficient	n/a
Largest diff. peak and hole	0.403 and -0.404 e.Å ⁻³



Chapter 4: Amphiphilic π-Allyliridium *C,O*-Benzoates Enable Regioand Enantioselective Amination of Branched Allylic Acetates Bearing Linear Alkyl Groups*

4.1 Introduction

Since the discovery of the Tsuji-Trost reaction,¹ diverse catalytic systems for metal catalyzed allylic substitution have emerged.² Among the many useful transformations based on this pattern of reactivity, enantioselective iridium catalyzed aminations figure prominently.³⁻¹¹ Largely due to the pioneering work of Takeuchi,^{3,4} Helmchen,^{5,6} Hartwig,^{7,8} Carreira⁹ and You,^{10,11} a broad range of amine nucleophiles can now be accommodated in highly regio- and enantioselective reactions of linear or branched allylic acetates (and in certain cases allylic alcohols).³⁻¹¹ Two classes of iridium catalysts may be distinguished on the basis of their ability to promote regio- and enantioselective reactions of either linear or branched allylic acetates (Figure 1). For type I catalysts, linear allyl proelectrophiles are used under basic conditions. For type II catalysts, branched allyl proelectrophiles are used under acidic conditions. While type I catalysts are modified by one phosphoramidite and one external diene ligand, type II catalysts incorporate an internal mono-olefin ligand and two phosphoramidites, which may account for their exceptional electrophilicity. The internal olefin of type II catalysts may enhance enantioselectivity in reactions of branched allyl proelectrophiles by displacing the π -bond of (σ + π)-allyl (envl) iridium intermediates,¹² thus enabling rapid π facial interconversion in otherwise stereospecific substitutions.¹³

^{*}This chapter is based on the published work:

Meza, A. T.; Wurm, T.; Smith, L.; Kim, S. W.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. J. Am. Chem. Soc. 2018, 140, 1275.

Prior Work: Cationic Metal, Electrophilic π-Allyl



Our Work: Neutral Metal, Amphiphilic π-Allyl



Figure 4.1 Phosphoramidite modified iridium complexes and amphiphilic π -allyliridium *C*,*O*-benzoates for regio- and enantioselective allylic amination.

Both type I and II catalysts bear π -acidic ligands and a formal positive charge at the metal center. These features enforce electrophilic properties of the allyliridium intermediate. In contrast, we have developed π -allyliridium *C,O*-benzoates, which are neutral and incorporate relatively strong σ -donor ligands (Figure 4.1).^{14,15} As demonstrated by their ability to promote diverse allylative carbonyl additions, the supporting ligands of this catalyst confer nucleophilic character onto the allyl moiety.¹⁶ However, as observed in our initial studies,^{14b} carbonyl allylation is often accompanied by small quantities of *O*-allylation, suggesting amphiphilic character of these π -allyl complexes. Pursuant to an accumulation of similar observations by coauthors at Genentech, a systematic attempt to evoke electrophilic behavior was undertaken in the context of allylic amination. Here, we show that the commercially available π -allyliridium *C*,*O*-benzoate modified by SEGPHOS catalyzes asymmetric allylic amination. This catalyst overcomes a significant limitation in scope across all known catalytic systems – the ability to engage branched allylic acetates bearing linear alkyl groups with uniformly high levels of regio- and enantioselectivity.^{3-11,17,18}

4.2 Reaction Development and Scope

In an initial set of experiments, α -methyl allyl acetate **4.1a** (100 mol%) was exposed to benzyl amine **4.2a** (200 mol%) in the presence of the SEGPHOS modified π allyliridium *C*,*O*-benzoate (5 mol%) in THF (1 M) at 40 °C (Table 4.1). In the absence of base, products of amination were not observed. However, after screening various heterogeneous bases, it was found that reactions conducted in the presence of Cs₂CO₃ (120 mol%) provided the desired product of allylic amination **4.3a** exclusively as the branched regioisomer with high levels of enantiomeric enrichment. After further optimization, the allylic amine **4.3a** could be obtained in 90% isolated yield as a single regioisomer in 90% enantiomeric excess. Wet THF solvent is required, perhaps to partially solubilize Cs₂CO₃. Using distilled THF under otherwise optimal conditions, **4.3a** was obtained in only 61% yield. Optimal yields were re-stored using distilled THF in combination with water (250 mol%). The absolute stereochemistry of **4.3a** was determined by comparison of its optical rotation to that of an authentic sample reported in the literature (and x-ray analysis of **4.3j'**).¹⁹

Table 4.1Influence of Base in the Amination of α -Methyl Allyl Acetate 4.1a to
Form Allylic Amine 4.3a.^a



^aAll reactions were performed on a 0.44 mmol scale. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details. ^bConversion was determined by ¹H NMR. ^cCs₂CO₃ (200 mol%). ^dYield of material isolated by silica gel chromatography.

Under these optimized conditions, primary benzylic, allylic and aliphatic amines serve as nucleophilic partners in aminations of diverse branched allylic acetates (Table 4.2). Complete branched-regioselectivity was accompanied by high levels of enantioselectively (¹H NMR & HPLC). Additionally, the levels of chemoselectivity in favor of primary amine allylation were sufficiently high that *N*-benzyl ethylene diamine and tryptamine could be converted to the respective amination products **4.31** and **4.3a'** without competing allylation of the secondary amine. As demonstrated by the formation of **4.3z**, **4.3g'**, **4.3h'**, **4.3i'**, primary amines that are branched at the α -position are effective nucleophilic partners. Further, as illustrated by the formation of **4.3b** vs **4.3c**, **4.3e** vs **4.3f** and **4.3h** vs **4.3i**, excellent levels of catalyst-directed stereoinduction are observed. Perhaps the notable feature of this catalytic system, however, resides in the ability to engage branched allylic acetates bearing linear alkyl groups in highly regio- and enantioselective aminations – a capability that complements the scope of previously reported iridium catalysts for allylic amination.^{3-11,17,18}

Table 4.2Regio- and Enantioselective Iridium-catalyzed Amination of Branched
Allylic Carboxylates.^a



^aYields of material isolated by silica gel chromatography. Standard conditions: (*R*)-IrLn (5 mol%), amine (200 mol%), Cs₂CO₃ (200 mol%), "wet" THF (1 M), 50 °C. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details. ^bH₂NCH₂CH₂NHBn (400 mol%), ee% determined at the stage of **4.4l** (eq. 4.1). ^cThe cyclometallated iridium catalyst modified by the Roche ligand was used.¹⁸

To illustrate the utility of the reaction products a series of transformations were performed. The *N*-benzyl ethylene diamine adduct **4.31** was subjected to reductive amination with glyoxal to form the piperazine **4.41** (eq. 4.1).²⁰ Adducts **4.3x** and **4.3j'** were converted to 1-(*N*-benzyl-amino)-2-cyclohexene **4.4x** (eq. 4.2)²¹ and the cyclopropyl substituted lactam **4.4j'** through ring-closing metathesis (eq. 4.3).^{5g}



Scheme 4.1 Identical π -allyliridium *C*, *O*-benzoate complexes promote both nucleophilic and electrophilic allylation.^a



^aSee Supporting Information for further experimental details.

While numerous reports of "umpoled allylations" exist,²² the amphiphilic properties displayed by π -allyliridium *C*,*O*-benzoates appear quite unique.²³ The same catalyst, (*R*)-IrLn (5 mol%), will promote both nucleophilic and electrophilic allylation under very similar conditions. For example, under standard amination conditions, 4-aminomethyl-benzyl alcohol and α -methyl allyl acetate are directly converted to compound **4.7** in 36% yield. Alternatively, compound **4.5** can be converted to compound **4.7** in a step-wise manner with higher levels of stereoselectivity using the same iridium catalyst (Scheme 4.1).

4.3 Discussion

The nature of the C-N bond forming event merits discussion. For related enantioselective iridium catalyzed allylic aminations, an outer-sphere mechanism is postulated.³⁻¹¹ While it is likely such pathways are also operative in aminations catalyzed by π -allyliridium *C*,*O*-benzoates, an inner-sphere mechanism cannot be excluded. The

more stringent steric constraints of an inner-sphere mechanism may account for the fact that primary amines engage in allylation while secondary amines do not. Furthermore, enantiofacial selectivity for amination through an inner sphere mechanism matches that observed in nucleophilic allylations of α -substituted allylic acetates (Figure 4.2). Computational studies are underway to evaluate outer vs inner sphere pathways.

Figure 4.2 Stereochemical models accounting for equivalent π -facial selectivity in crotylation and amination.



4.4 Conclusion

In summary, highly tractable and commercially available π -allyliridium *C*,*O*-benzoates, which are well known to catalyze nucleophilic carbonyl allylation, are now shown to catalyze chemo-, regio- and enantioselective electrophilic aminations of branched allylic acetates bearing linear alkyl groups. These processes broaden access to chiral *N*-containing building blocks²⁴ and establish unique amphiphilic properties of π -allyliridium *C*,*O*-benzoates, which should inform the design of new catalytic process. More broadly, this work demonstrates how academic-industrial collaboration accelerates the discovery of robust, innovative methods for chemical synthesis.

4.5 Experimental Details

General Information

All reactions were carried out under inert gas atmosphere (nitrogen or argon) unless otherwise indicated. Resealable pressure tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were flame dried followed by cooling in a desiccator or under a stream of inter gas prior to use. All commercial reagents and anhydrous solvents were used as received from vendors (Fischer Scientific, Sigma Aldrich and Combi Blocks) without further purification. The used Iridium catalyst [(R)-(+)-5,5'-bis(diphenylphosphino)-4,4'-bi-1,3-benzodioxole][4-cyano-3-

nitrobenzenecarboxylato][1,2,3-eta-2-propenyl Ir(III)] and its enantiomer were either obtained from Strem Chemicals (catalog numbers: 77-5074 (R-catalyst) 77-5075 (Scatalyst)) or prepared according to literature known procedures.¹ The compounds (4-(pentafluoro- λ 6-sulfaneyl)phenyl)methanamine,² (E)-3-phenylprop-2-en-1-amine,³ geranylamine,⁴ 5-phenylpenten-3-yl acetate,⁵ α -cyclopropyl allyl acetate,⁶ and 1,7octadien-3-yl acetate⁷ were also prepared according to literature procedures. Preparative column chromatography employing Silicycle silica gel (40-63 µm) was performed according to the method of Still⁸ or on a Teledyne Isco Combiflash R_f utilizing Silicycle HP columns using a mobile phase composed of either heptane/isopropyl acetate, hexane/ethyl acetate or dichloromethane/methanol. Reactions were monitored by a Shimadzu LCMS/UV system with LC-30AD solvent pump, 2020 MS, Sil-30AC autosampler, SPD-M30A UV detector, CTO-20A column oven, using a 2-98% acetonitrile/0.1% formic acid (or 0.001% ammonia) gradient over 2.5 minutes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynamic Absorbents F). Visualization was accomplished with UV light followed by dipping in CAM, *p*-Anisaldehyde (PAA), or KMnO₄ stain solution followed by heating. Specific optical rotations were recorded on an Atago AP-300 automatic polarimeter at the sodium line (589.3 nm) in CHCl3 Solution concentrations are given in the units of 10-2 g ml⁻¹.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer. Highresolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M+H, M+Na), or a suitable fragment ion. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (400 MHz) or Varian INOVA (500 MHz) spectrometer equipped with a Bruker cryoprobe. Chemical shifts are reported in delta (δ) units, parts per million (ppm) downfield from tetramethylsilane or ppm relative to the center of the singlet at 7.26 ppm for deuteriochloroform. Data reported as multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet). Integration and coupling constants were reported in Hertz (Hz). Carbon-13 nuclear magnetic resonance (¹³C NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (100 MHz) or Varian INOVA (125 MHz) spectrometer and were routinely run with broadband decoupling. Chemical shifts are reported in delta (δ) units, ppm relative to the center of the triplet at 77.16 ppm for deuteriochloroform. Fluorine-19 nuclear magnetic resonance (¹⁹F NMR) spectra were recorded with a Varian Gemini (376 MHz) spectrometer.

Experimental Details and Spectral Data

Enantioselective iridium catalyzed allylic alkylation with primary amine nucleophiles-



General procedure

An oven-dried pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (200 mol%), (*R*)-Ir-SEGPHOS (5 mol%). The tube was purged with argon for 5 minutes. THF (1.0 M) was added followed by the allylic acetate (100 mol%) and the amine (200 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at the indicated temperature and stirred for the indicated period of time. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography.

(*R*)-N-benzylbut-3-en-2-amine (4.3a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 90% yield (63.2 mg, 0.39 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>TLC (SiO</u>₂) $R_f = 0.20$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.38 - 7.26$ (m, 4H), 7.26 – 7.17 (m, 1H), 5.72 (ddd, J = 17.6, 10.2, 7.7 Hz, 1H), 5.18 – 5.09 (m, 1H), 5.09 - 5.05 (m, 1H), 3.81 (d, J = 13.2 Hz, 1H), 3.69 (d, J = 13.3 Hz, 1H), 3.28 – 3.16 (m, 1H), 1.18 (d, J = 6.5 Hz, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 142.6, 140.7, 128.4, 128.1, 126.8, 114.7, 56.0, 51.4, 21.8.$

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{16}N[M+H^+] = 162.1283$, found 162.1274.

<u>FTIR</u> (neat): 3063, 3027, 2971, 1640, 1495, 1452, 1415, 1369, 1312, 1166, 1114, 1073, 1028, 993, 916, 731, 696 cm⁻¹.

 $[\alpha]_{\rm D}^{24} = -5.6 \ (c \ 0.35, \rm CHCl_3).$

<u>HPLC</u> (Chiralcel OD-H column, heptane:*i*-PrOH = 97.5:2.5, 1.00 mL/min, 220 nm), *ee* = 90%.





(R)-N-((R)-1-phenylethyl)but-3-en-2-amine (4.3b)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (107 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 93% yield, >20:1 dr (71.4 mg, 0.41 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>TLC (SiO2</u>) $R_f = 0.32$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.37 - 7.18$ (m, 5H), 5.63 (ddd, J = 17.1, 10.1, 8.0 Hz, 1H), 5.03 (dd, J = 10.2, 1.8 Hz, 1H), 4.92 (ddd, J = 17.1, 2.0, 0.9 Hz, 1H), 3.82 (q, J = 6.7 Hz, 1H), 2.98 – 2.86 (m, 1H), 1.32 (d, J = 6.7 Hz, 3H), 1.08 (d, J = 6.6 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 145.9$, 142.5, 128.4, 126.7, 126.6, 114.5, 54.9, 53.7, 25.0, 22.5.

<u>**HRMS**</u> (ESI): Calculated for C₁₂H₁₈N [M+H⁺] = 176.1439, found 176.1430 <u>**FTIR**</u> (neat): 2959, 2924, 1672, 1602, 1492, 1450, 1415, 1368, 1306, 1120, 1081, 1027, 993, 956, 916, 845, 761, 699 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{D}}^{\mathbf{24}} = +12.3 \ (c \ 0.97, \text{CHCl}_3).$



(*R*)-N-((*S*)-1-phenylethyl)but-3-en-2-amine (4.3c)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (107 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 92% yield, 14:1 dr (70.6 mg, 0.40 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>TLC</u> (SiO₂) $R_f = 0.32$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.35 - 7.70$ (m, 5H), 5.69 (ddd, J = 17.3, 10.2, 7.2 Hz, 1H), 5.09 - 5.02 (m, 1H), 5.01 - 4.96 (m, 1H), 3.88 (q, J = 6.6 Hz, 1H), 3.21 - 3.09 (m, 1H), 1.33 (d, J = 6.6 Hz, 3H), 1.13 (d, J = 6.4 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 146.0, 143.0, 128.4, 126.8, 126.5, 113.7, 54.7, 53.3, 23.8, 20.9.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{18}N[M+H^+] = 176.1439$, found 176.1430

<u>FTIR</u> (neat): 2966, 2926, 1638, 1492, 1451, 1416, 1368, 1326, 1209, 1127, 993, 915, 844, 760, 698 cm⁻¹.

 $[\alpha]_{D}^{24} = -78.8 \ (c \ 0.31, \text{CHCl}_3).$



(*R*)-N-(naphthalen-2-ylmethyl)but-3-en-2-amine (4.3d)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (138 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 99% yield (92.0 mg, 0.44 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>TLC (SiO</u>₂) $R_f = 0.20$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.86 - 7.72$ (m, 4H), 7.51 - 7.39 (m, 3H), 5.76 (ddd, J = 17.5, 10.2, 7.6 Hz, 1H), 5.19 - 5.06 (m, 2H), 3.98 (d, J = 13.2 Hz, 1H), 3.86 (d, J = 13.3 Hz, 1H), 3.34 - 3.21 (m, 1H), 1.21 (d, J = 6.4 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.6, 138.2, 133.5, 132.6, 128.0, 127.7, 127.6, 126.7, 126.4, 126.0, 125.5, 114.8, 56.0, 51.5, 21.8.

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{18}N[M+H^+] = 212.1439$, found 212.1429.

<u>FTIR</u> (neat): 3053, 2968, 2817, 2362, 1636, 1600, 1508, 1438, 1414, 1367, 1311, 1270, 1170, 1124, 993, 916, 854, 814, 746, 682 cm⁻¹.

 $[\alpha]_{D}^{24} = -8.1 \ (c \ 0.41, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-H column, heptane:*i*-PrOH = 99.0:1.0, 2.00 mL/min, 254 nm), *ee* = 90%.





(R)-N-((R)-1-(naphthalen-2-yl)ethyl)but-3-en-2-amine (4.3e)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (68.5 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 95% yield, >20:1 dr (43.0 mg, 0.19 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-3:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.34$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.82$ (ddt, J = 7.1, 5.4, 0.7 Hz, 3H), 7.68 (dd, J = 1.5, 0.9 Hz, 1H), 7.50 – 7.41 (m, 3H), 5.64 (ddd, J = 17.1, 10.2, 8.1 Hz, 1H), 5.06 (ddd, J = 10.1, 1.8, 0.6 Hz, 1H), 4.92 (ddd, J = 17.1, 1.8, 0.8 Hz, 1H), 4.00 (q, J = 6.7 Hz, 1H), 2.94 (ddt, J = 8.0, 7.2, 6.1 Hz, 1H), 1.40 (d, J = 6.7 Hz, 3H), 1.09 (d, J = 6.6 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 143.2$, 142.4, 133.4, 132.7, 128.2, 127.6, 127.6, 125.9, 125.3, 125.2, 124.8, 114.6, 55.0, 53.7, 24.9, 22.5. **HRMS** (ESI): Calculated for C₁₆H₁₉N [M+H⁺] = 226.1590, Found 226.1584.

<u>FTIR</u> (neat): 2960, 1449, 1368, 1126, 917, 856, 819, 746 cm⁻¹. $[\alpha]_D^{29} = +55.0 \ (c \ 1.0, \text{CHCl}_3).$



(R)-N-((S)-1-(naphthalen-2-yl)ethyl)but-3-en-2-amine (4.3f)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (151 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 97% yield, 16:1 dr (96.0 mg, 0.43 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>TLC</u> (SiO₂) $R_f = 0.30$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.87 - 7.76$ (m, 3H), 7.74 - 7.71 (m, 1H), 7.50 - 7.40 (m, 3H), 5.71 (ddd, J = 17.3, 10.2, 7.1 Hz, 1H), 5.07 (dt, J = 17.3, 1.4 Hz, 1H), 4.98 (dt, J = 10.2, 1.3 Hz, 1H), 4.06 (q, J = 6.5 Hz, 1H), 3.28 - 3.08 (m, 1H), 1.41 (d, J = 6.5 Hz, 4H), 1.16 (d, J = 6.3 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{125.4}, 125.1, 125.0, 113.8, 54.8, 53.4, 23.8, 20.9.}$

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}N[M+H^+] = 226.1596$, found 226.1586

<u>FTIR</u> (neat): 3055, 2965, 2923, 1637, 1600, 1506, 1451, 1415, 1367, 1315, 1269, 1128, 1019, 993, 947, 915, 891, 855, 817, 744, 659 cm⁻¹.

 $[\alpha]_{\rm D}^{24}$ = -66.2 (*c* 0.58, CHCl₃).



(R)-N-(naphthalen-1-ylmethyl)but-3-en-2-amine (4.3g)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (62.9 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 88% yield (37.2 mg, 0.18 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.52$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.08$ (d, J = 8.6 Hz, 1H), 7.84 (d, J = 8.6 Hz, 1H), 7.75 (d, J = 8.6 Hz, 1H), 7.54-7.44 (m, 3H), 7.43-7.38 (m, 1H), 5.79 (ddd, J = 17.2, 10.2, 7.8 Hz, 1H), 5.2 (d, J = 17.2 Hz, 1H), 5.13 (d, J = 10.2 Hz, 1H), 4.26 (d, J = 13.0 Hz, 1H), 4.26 (d, J = 13.0 Hz, 1H), 4.08 (d, J = 13.0 Hz, 1H), 3.36-3.28 (m, 1H), 1.62 (bs, 1H), 1.21 (d, J = 6.5 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{^{125}\text{ MHz}}$ (125 MHz, CDCl₃): $\delta = 142.7$, 136.3, 134.1, 132.0, 128.9, 127.9, 126.3(x2), 125.8, 125.6, 123.9, 115.3, 57.1, 49.3, 22.1.

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{17}N[M+H^+] = 212.1434$, Found 212.1429.

<u>FTIR</u> (neat): 3063, 3027, 2970, 2923, 1640, 1495, 1452, 1415, 1369, 1312, 1113, 1073, 1028, 993, 916, 713, 696 cm⁻¹.

 $[\alpha]_{D}^{24} = -42.5 \ (c \ 2.1, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 230 nm), ee = 88%.






(R)-N-((R)-1-(naphthalen-1-yl)ethyl)but-3-en-2-amine (4.3h)



Procedures

The allylic acetate (60.5 mg, 0.53 mmol, 100 mol%) and the primary amine (182 mg, 1.06 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 91% yield, 17:1 dr (108 mg, 0.48 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.43$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.22 - 8.14$ (m, 1H), 7.90 - 7.84 (m, 1H), 7.76 - 7.71 (m, 1H), 7.67 - 7.62 (m, 1H), 7.55 - 7.41 (m, 3H), 5.69 (ddd, J = 17.1, 10.1, 7.9 Hz, 1H), 5.04 - 4.84 (m, 1H), 4.81 (ddd, J = 17.2, 1.9, 0.9 Hz, 1H), 4.71 (q, J = 6.7 Hz, 1H), 3.06 - 2.94 (m, 1H), 1.44 (d, J = 6.7 Hz, 3H), 1.14 (d, J = 6.5 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{125.2}, 123.0, 122.6, 114.6, 54.0, 50.4, 24.6, 22.4.} (100 \text{ MHz}, \text{CDCl}_3): \delta = 142.7, 141.7, 134.0, 131.4, 128.9, 127.0, 125.7, 125.6, 125.2, 123.0, 122.6, 114.6, 54.0, 50.4, 24.6, 22.4.}$

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}N[M+H^+] = 226.1596$, found 226.1585.

<u>FTIR</u> (neat): 3064, 2969, 2923, 1595, 1509, 1448, 1417, 1393, 1367, 1323, 1230, 1171, 1125, 1087, 992, 918, 859, 798, 777, 735, 681 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{D}}^{\mathbf{24}} = -26.3 \ (c \ 0.37, \text{CHCl}_3).$



(R)-N-((S)-1-(naphthalen-1-yl)ethyl)but-3-en-2-amine (4.3i)



Procedures

The allylic acetate (60.5 mg, 0.53 mmol, 100 mol%) and the primary amine (182 mg, 1.06 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 95% yield, > 20:1 dr (113 mg, 0.50 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.43$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.20-8.16$ (m, 1H), 7.88-7.84 (m, 1H), 7.77 - 7.72 (m, 1H), 7.66 - 7.60 (m, 1H), 7.55 - 7.38 (m, 3H), 5.75 (ddd, J = 17.4, 10.2, 7.4 Hz, 1H), 5.20 - 5.10 (m, 1H), 5.08 - 5.02 (m, 1H), 4.74 (q, J = 6.5 Hz, 1H), 3.37 - 3.25 (m, 1H), 1.48 (d, J = 6.5 Hz, 3H), 1.16 (d, J = 6.4 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.9, 141.8, 134.0, 131.1, 128.9, 127.1, 125.8, 125.7, 125.3, 123.0, 122.9, 114.0, 53.6, 49.7, 22.9, 21.4.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}N[M+H^+] = 226.1596$, found 226.1586.

<u>FTIR</u> (neat): 3049, 2967, 1638, 1595, 1509, 1450, 1416, 1394, 1367, 1317, 1230, 1169, 1128, 995, 916, 859, 798, 776, 754, 684 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{D}}^{\mathbf{24}} = -56.7 \ (c \ 0.38, \text{CHCl}_3).$



(R)-N-cinnamylbut-3-en-2-amine (4.3j)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (53.3 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 85% yield (32.0 mg, 0.17 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-1:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.20$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.39 - 7.34$ (m, 2H), 7.33 - 7.27 (m, 2H), 7.24 - 7.19 (m, 1H), 6.51 (dt, J = 15.9, 1.5 Hz, 1H), 6.30 (ddd, J = 15.9, 6.6, 6.1 Hz, 1H), 5.70 (ddd, J = 17.2, 10.2, 7.7 Hz, 1H), 5.18 - 5.04 (m, 2H), 3.43 (ddd, J = 13.9, 6.1, 1.5 Hz, 1H), 3.33 (ddd, J = 13.9, 6.6, 1.3 Hz, 1H), 3.29 - 3.22 (m, 1H), 1.47 (br, 1H), 1.19 (d, J = 6.5 Hz, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 142.2, 137.1, 131.2, 128.5, 128.4, 127.3, 126.2, 114.8, 56.1, 49.3, 21.7.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N[M+H^+] = 188.1434$, Found 188.1427.

<u>FTIR</u> (neat): 2970, 1738, 1448, 1368, 1216, 1116, 966, 918, 747, 691 cm⁻¹.

 $[\alpha]_{D}^{30} = -8.00 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AS-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 96%.</u>





(*R*,*E*)-N-(but-3-en-2-yl)-2,7-dimethylocta-2,6-dien-1-amine (4.3k)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (61.3 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 90% yield (37.3 mg, 0.18 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-1:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.16$ (methanol/ethyl acetate = 1:9).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.69$ (ddd, J = 17.1, 10.2, 7.9 Hz, 1H), 5.28 – 5.20 (m, 1H), 5.13 – 5.02 (m, 3H), 3.19 (dtd, J = 25.9, 12.8, 6.8 Hz, 3H), 3.03 (br, 1H), 2.09 – 1.95 (m, 4H), 1.66 (d, J = 1.3 Hz, 3H), 1.60 (s, 3H), 1.58 (s, 3H), 1.17 (d, J = 6.5 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 141.7, 138.2, 131.5, 124.0, 121.9, 115.1, 56.0, 44.4, 39.6, 26.4, 25.7, 21.4, 17.6, 16.3.

HRMS (ESI): Calculated for C₁₄H₂₅N [M+H⁺] = 208.2060, Found 208.2054. **FTIR** (neat): 2924, 2359, 1629,1439, 1376, 1108, 993, 916, 753 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{D}}^{\mathbf{28}} = -6.38 \ (c \ 1.0, \text{CHCl}_3).$ <u>HPLC</u> Enantiomeric excess was determined by HPLC analysis of the *p*-tosyl derivative of the product (Chiralcel OJ-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 94%.





(R)- N^1 -benzyl- N^2 -(but-3-en-2-yl)ethane-1,2-diamine (4.3)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the amine (120 mg, 0.80 mmol, 400 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 710% yield (28.9 mg, 0.14 mmol) as a pale red oil after purification by flash column chromatography (SiO₂, ethyl acetate : MeOH = 10:1-3:1).

<u>**TLC (SiO**</u>₂) $R_f = 0.10$ (ethyl acetate).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.34-7.30 (m, 4H), 7.27-7.22 (m, 1H), 5.66 (ddd, *J* = 17.0, 10.2, 7.6 Hz, 1H), 5.08 (d, *J* = 17.0 Hz, 1H), 5.02 (d, *J* = 10.2 Hz, 1H), 3.79 (s, 2H), 3.16-3.09 (m, 1H), 2.78-2.69 (m, 3H), 2.68-2.61 (m, 1H), 1.56 (bs, 1H), 1.15 (d, *J* = 6.5 Hz, 3H)

 $\frac{^{13}C \text{ NMR}}{57.0, 54.1, 49.2, 47.1, 22.0.} \delta = 142.9, 140.7, 128.6 (2C), 128.3 (2C), 127.1, 114.6,$

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{20}N_2$ [M+H⁺] = 205.1699, Found 205.1693.

<u>FTIR</u> (neat): 3315, 3063, 3027, 2971, 2925, 2819, 2187, 1639, 1602, 1494, 1452, 1415, 1369, 1312, 1114 1073, 1027, 993, 911, 729, 697 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{p}}^{\mathbf{24}} = -5.4 \ (c \ 1.1, \text{CHCl}_3).$



(*R*)-N-(pyridin-2-ylmethyl)but-3-en-2-amineamine (4.3m)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (43.2 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 90% yield (29.0 mg, 0.18 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 2:1-1:2).

<u>**TLC (SiO**</u>₂) $R_f = 0.13$ (ethyl acetate).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.55$ (ddd, J = 4.9, 1.9, 0.9 Hz, 1H), 7.62 (td, J = 7.7, 1.8 Hz, 1H), 7.28 (m, 1H), 7.15 (ddd, J = 7.6, 4.9, 1.2 Hz, 1H), 5.73 (ddd, J = 17.2, 10.2, 7.8 Hz, 1H), 5.16 – 5.05 (m, 2H), 3.91 (d, J = 14.0 Hz, 1H), 3.81 (d, J = 14.0 Hz, 1H), 3.27 – 3.18 (m, 1H), 1.87 (br, 1H), 1.21 (d, J = 6.5 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 159.7, 149.2, 142.3, 136.3, 122.4, 121.8, 114.9, 56. 5, 52.6, 21.8.

<u>HRMS</u> (ESI): Calculated for $C_{10}H_{14}N_2$ [M+H⁺] = 163.1230, Found 163.1226.

<u>FTIR</u> (neat): 2962, 1592, 1434, 1216, 1118, 995, 921, 751 cm⁻¹.

 $[\alpha]_{D}^{30} = -19.50 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> Enantiomeric excess was determined by HPLC analysis of the product with (*S*)-Ir-SEGPHOS (Chiralcel AD column, hexanes:*i*-PrOH = 98:2, 0.80 mL/min, 210 nm), ee = 94%.





(R)-N-(3-bromobenzyl)but-3-en-2-amine (4.3n)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (74.4 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 86% yield (41.3 mg, 0.17 mmol) as a pale yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.64$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.49 (s, 1H), 7.37 (d, *J* = 7.8 Hz, 1H), 7.24 (d, *J* = 7.8 Hz, 1H), 7.18 (t, *J* = 7.8 Hz, 1H), 5.69 (ddd, *J* = 17.2, 10.2, 7.7 Hz, 1H), 5.16-5.06 (m, 2H), 3.77 (d, *J* = 13.4 Hz, 1H), 3.65 (d, *J* = 13.4 Hz, 1H), 3.23-3.16 (m, 1H), 1.36 (bs, 1H), 1.18 (d, *J* = 6.5 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 143.3, 142.5, 131.3, 130.1(x2), 126.9, 122.7, 115.2, 56.3, 50.9, 22.0.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{14}BrN[M+H^+] = 240.0382$, Found 240.0375.

<u>FTIR</u> (neat): 3311, 3073, 2971, 2923, 2818, 1775, 1640, 1594, 1568, 1470, 145, 1370, 1312, 1195, 1165, 1114, 1068, 1038, 994, 918, 883, 859, 829, 775, 683, 667 cm⁻¹. $[\boldsymbol{\alpha}]_{\mathbf{p}}^{\mathbf{24}} = -8.3 \ (c \ 3.1, \text{CHCl}_3).$

<u>**HPLC</u></u> (3x Chiralcel AD-H column, hexanes:***i***-PrOH = 98:2, 1.00 mL/min, 230 nm), ee = 92 %.</u>**





(R)-N-(benzo[d][1,3]dioxol-5-ylmethyl)but-3-en-2-amine (4.30)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (60.5 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 92% yield (37.7 mg, 0.18 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-1:1).

<u>**TLC** (SiO₂)</u> $R_f = 0.22$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 6.83$ (s, 1H), 6.75 (d, J = 1.1 Hz, 2H), 5.93 (s, 2H), 5.69 (ddd, J = 17.2, 10.2, 7.7 Hz, 1H), 5.16 – 5.03 (m, 2H), 3.70 (d, J = 13.1 Hz, 1H), 3.58 (d, J = 13.0 Hz, 1H), 3.24 – 3.15 (m, 1H), 1.16 (d, J = 6.5 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 147.6, 146.3, 142.4, 134.6, 121.1, 114.7, 108.7, 108.0, 100.8, 55.8, 51.1, 21.8.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{15}NO_2[M+H^+] = 206.1176$, Found 206.1169.

<u>FTIR</u> (neat): 2971, 1488, 1441, 1249, 1040, 923, 808, 754 cm⁻¹.

 $[\alpha]_{D}^{28} = -6.25 (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel AS-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 99%.</u>





(R)-N-(furan-2-ylmethyl)but-3-en-2-amine (4.3p)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (85.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 74% yield (48.9 mg, 0.32 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.23$ (hexanes:ethyl acetate 1:1).

¹<u>H</u> NMR (400 MHz, CDCl₃): $\delta = 7.37-7.34$ (m, 1H), 6.32-6.28 (m, 1H), 6.17-6.12 (m, 1H), 5.68 (ddd, J = 17.5, 10.2, 7.8 Hz, 1H), 5.17 – 5.04 (m, 2H), 3.79 (d, J = 14.3 Hz, 1H), 3.69 (d, J = 14.4 Hz, 1H), 3.24 - 3.15 (m, 1H), 1..17 (d, J = 6.5 Hz, 3H).

<u>¹³C NMR</u> (100 MHz, CDCl₃): δ = 154.1, 142.1, 141.7, 115.1, 110.0, 106.7, 55.7, 43.7, 21.7.

<u>HRMS</u> (ESI): Calculated for C₉H₁₄NO [M+H⁺] = 152.1075, found 152.1067.

<u>FTIR</u> (neat): 2973, 1672, 1600, 1505, 1439, 1416, 1370, 1315, 1251, 1147, 1110, 1074, 1011, 995, 918, 884, 804, 730 cm⁻¹.

 $[\alpha]_{D}^{24} = +18.5 \ (c \ 0.63, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-H column, heptane:*i*-PrOH = 99.0:1.0, 1.00 mL/min, 220 nm), *ee* = 92%.





(R)-N-(thiophen-2-ylmethyl)but-3-en-2-amine (4.3q)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (99.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 98% yield (71.5 mg, 0.43 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.40$ (hexanes:ethyl acetate 1:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.22 - 7.17$ (m, 1H), 6.96 – 6.92 (m, 1H), 6.92 - 6.89 (m, 1H), 5.69 (ddd, J = 17.6, 10.2, 7.7 Hz, 1H), 5.18 – 5.05 (m, 2H), 3.99 (dd, J = 14.1, 0.9 Hz, 1H), 3.90 (dd, J = 14.1, 0.8 Hz, 1H), 3.31 - 3.21 (m, 1H), 1.17 (d, J = 6.5 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 144.4, 142.2, 126.6, 124.7, 124.2, 115.0, 55.6, 45.7, 21.7.

<u>HRMS</u> (ESI): Calculated for C₉H₁₄NS [M+H⁺] = 168.0847, found 168.0838.

<u>FTIR</u> (neat): 3072, 2971, 1636, 1439, 1416, 1369, 1311, 1218, 1171, 1111, 1038, 993, 918, 852, 824, 693 cm⁻¹.

 $[\alpha]_{D}^{24} = +18.5 \ (c \ 0.63, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-H column, heptane:*i*-PrOH = 99.0:1.0, 0.50 mL/min, 220 nm), *ee* = 89%.





(R)-N-(thiazol-2-ylmethyl)but-3-en-2-amine (4.3r)



Procedures

The allylic acetate (22.8 mg, 0.20 mmol, 100 mol%) and the primary amine (45.6 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 18 hr). The title compound was obtained in 80% yield (26.8 mg, 0.16 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-1:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.21$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.71 (d, *J* = 3.3 Hz, 1H), 7.25 (d, *J* = 3.3 Hz, 1H), 5.69 (ddd, *J* = 17.2, 10.2, 7.6 Hz, 1H), 5.17 – 5.07 (m, 2H), 4.12 (d, *J* = 15.4 Hz, 1H), 4.06 (d, *J* = 15.4 Hz, 1H), 3.28 (ddt, *J* = 7.4, 6.5, 0.9 Hz, 1H), 1.94 (br, 1H), 1.21 (d, *J* = 6.5 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 172.0, 142.4, 141.6, 118.6, 115.3, 56.1, 48.2, 21.6.

<u>HRMS</u> (ESI): Calculated for $C_8H_{12}N_2S[M+H^+] = 169.0794$, Found 169.0789.

<u>FTIR</u> (neat): 2972, 1505, 1214, 1185, 1137, 993, 922, 750 cm⁻¹.

 $[\alpha]_{D}^{28} = -8.25 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OD-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 96%.</u>





(*R*)-*N*-benzyl-5-(benzyloxy)pent-1-en-3-amine (4.3s)



Procedures

The allylic acetate (103 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 84% yield (103 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0–0:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.26$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.45 - 7.19$ (m, 10H), 5.65 (ddd, J = 16.8, 10.5, 8.1 Hz, 1H), 5.18 - 5.07 (m, 2H), 4.48 (d, J = 11.8 Hz, 1H), 4.45 (d, J = 11.8 Hz, 1H) 3.82 (d, J = 13.3 Hz, 1H), 3.63 (d, J = 13.1 Hz, 1H), 3.62 - 3.46 (m, 2H), 3.27 - 3.19 (m, 1H), 1.90 - 1.80 (m, 1H), 1.78 - 1.67 (m, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{16.1}} (100 \text{ MHz}, \text{CDCl}_3): \delta = 140.5, 140.7, 138.5, 128.3, 128.2, 127.7, 127.5, 126.8, 116.1, 100.0, 73.0, 67.8, 59.0, 51.3, 35.7.$

HRMS (ESI): calcd for C₁₉H₂₄NO [M+H]⁺: 282.1858, found: 282.1846

<u>FTIR</u> (neat): 3028, 2856, 1639, 1494, 1453, 1362, 1203, 1098, 1027, 994, 918, 733, 695 cm⁻¹.

 $[\alpha]_D^{24} = -3.4 \ (c \ 0.84, \text{CHCl}_{3}).$

HPLC (Chiralpak OD-H; heptane/isopropanol 90.0:10.0, 0.9 mL/min, 220 nm) ee = 94%







(R)-5-(benzyloxy)-N-(cyclohexylmethyl)pent-1-en-3-amine (4.3t)



Procedures

The allylic acetate (103 mg, 0.44 mmol, 100 mol%) and the primary amine (99.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 85% yield (107 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0–0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.17$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.38 - 7.22$ (m, 5H), 5.59 (ddd, J = 17.0, 10.3, 8.1 Hz, 1H), 5.13 - 5.02 (m, 2H), 4.51 (d, J = 11.9 Hz, 1H), 4.47 (d, J = 12.0 Hz, 1H), 3.62 - 3.46 (m, 2H), 3.18 - 3.09 (m, 1H), 2.43 (dd, J = 11.5, 6.8 Hz, 1H), 2.29 (dd, J = 11.6, 6.6 Hz, 1H), 1.88 - 1.77 (m, 1H), 1.76 - 1.60 (m, 6H), 1.45 - 1.32 (m, 1H), 1.29 - 1.07 (m, 3H), 0.95 - 0.80 (m, 2H).

 $\frac{^{13}C \text{ NMR}}{59.7, 54.2, 38.1, 35.7, 31.6, 31.5, 26.7, 26.1(x2).} (100 \text{ MHz}, \text{CDCl}_3): \delta = 141.4, 128.3, 127.7, 127.5, 115.6, 100.0, 73.0, 67.8, 59.7, 54.2, 38.1, 35.7, 31.6, 31.5, 26.7, 26.1(x2).}$

HRMS (ESI): calcd for C₁₉H₃₀NO [M+H]⁺: 288.2327, found: 288.2316

<u>FTIR</u> (neat): 2920, 2850, 1639, 1495, 1450, 1415, 1363, 1261, 1206, 1100, 1027, 993, 916, 733, 696 cm⁻¹.

 $[\alpha]_D^{24} = -10.7 \ (c \ 0.30, \text{CHCl}_{3}).$

HPLC (Chiralpak OD-H, heptane/isopropanol 100:0, 0.9 mL/min, 220 nm) ee = 91%






(*R*)-5-(benzyloxy)-*N*-isobutylpent-1-en-3-amine (4.3u)



Procedures

The allylic acetate (103 mg, 0.44 mmol, 100 mol%) and the primary amine (64.4 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 78% yield (84.4 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.19$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.41 - 7.22$ (m, 5H), 5.60 (ddd, J = 17.0, 10.3, 8.1 Hz, 1H), 5.13 - 5.03 (m, 2H), 4.50 (d, J = 12.0 Hz, 1H), 4.47 (d, J = 11.9 Hz, 1H), 3.62 - 3.46 (m, 2H), 3.19 - 3.10 (m, 1H), 2.41 (dd, J = 11.5, 6.9 Hz, 1H), 2.28 (dd, J = 11.5, 6.6 Hz, 1H), 1.88 - 1.77 (m, 1H), 1.75 - 1.60 (m, 2H), 0.88 (d, J = 6.7, 3H), 0.87 (d, J = 6.7, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 141.4, 138.5, 128.3, 127.7, 127.5, 115.6, 73.0, 67.8, 59.6, 55.5, 35.7, 28.4, 20.8, 20.7.

HRMS (ESI): calcd for C₁₆H₂₆NO [M+H]⁺: 248.2014, found: 248.2004

<u>FTIR</u> (neat): 3031, 2952, 2867, 1639, 1495, 1453, 1415, 1385, 1363, 1246, 1203, 1101, 1027, 994, 916, 733, 696 cm⁻¹.

 $[\alpha]_D^{24} = -13.8 \ (c \ 0.72, \text{CHCl}_{3,}).$

HPLC (Chiralpak OD-H; heptane/isopropanol 100:0, 0.9 mL/min, 220 nm). ee = 93%







(R)-N-benzyl-5-((tert-butyldimethylsilyl)oxy)pent-1-en-3-amine (4.3v)



Procedures

The allylic acetate (114 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 80% yield (107 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0–0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.51$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.40 - 7.18$ (m, 5H), 5.66 (ddd, J = 16.4, 11.0, 8.2 Hz, 1H), 5.17 - 5.08 (m, 2H), 3.83 (d, J = 13.2 Hz, 1H), 3.75 - 3.61 (m, 2H), 3.63 (d, J = 13.2 Hz, 1H), 3.22 (q, J = 6.7 Hz, 1H), 1.82 - 1.68 (m, 1H), 1.68 - 1.58 (m, 1H), 0.86 (s, 9H), 0.03 (s, 6H).

¹³C NMR (100 MHz, CDCl₃): δ = 141.1, 140.8, 128.3, 128.2, 126.7, 115.9, 60.7, 58.9, 51.3, 38.6, 25.9, 18.2, -5.4.

HRMS (ESI): calcd for C₁₈H₃₂NOSi [M+H]⁺: 306.2253 found: 306.2244

<u>FTIR</u> (neat): 2952, 2927, 2856, 1494, 1471, 1462, 1415, 1387, 1360, 1254, 1091, 1028, 1004, 918, 833, 773, 730, 696, 662 cm⁻¹.

 $\left[\alpha\right]_{D}^{24} = -3.5 \ (c \ 0.51, \text{CHCl}_{3}).$

HPLC (Chiralpak AD-H, heptane/isopropanol 60:40, 1.0 mL/min, 254 nm) ee = 95% (NOTE: ee determined on *N*-benzyl-*N*-(5-((*tert*-butyldimethylsilyl)oxy)pent-1-en-3-yl)-4-nitrobenzamide)







2 5.874 MM 0.1626 20.21516 2.07224 2.7263

(*R*)-*N*-benzyloct-1-en-3-amine (4.3w)



Procedures

The allylic acetate (74.9 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 82% yield (79.0 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC (SiO2)**</u> $R_f = 0.54$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.36 - 7.28 (m, 4H), 7.28 - 7.18 (m, 1H), 5.62 (ddd, *J* = 17.1, 10.3, 8.2 Hz, 1H), 5.18 - 5.05 (m, 2H), 3.83 (d, *J* = 13.1 Hz, 1H), 3.64 (d, *J* = 13.2 Hz, 1H), 3.05 - 2.97 (m, 1H), 1.70 - 1.15 (m, 8H), 0.87 (t, *J* = 6.8 Hz, 3H). ¹³C NMP (100 MHz, CDCl₂): δ = 141.5, 140.8, 128.3, 128.2, 126.8, 115.9, 61.3, 51.3

 $\frac{^{13}C \text{ NMR}}{35.8, 31.9, 25.6, 22.6, 14.0.}$ (100 MHz, CDCl₃): δ = 141.5, 140.8, 128.3, 128.2, 126.8, 115.9, 61.3, 51.3, 35.8, 31.9, 25.6, 22.6, 14.0.

HRMS (ESI): calcd for C₁₅H₂₄N [M+H]⁺: 218.1909, found: 218.1899

<u>FTIR</u> (neat): 3063, 3027, 2955, 2927, 2856, 1639, 1494, 1453, 1414, 1377, 1200, 1111, 1028, 993, 916, 729, 697 cm⁻¹.

 $[\alpha]_D^{24} = +1.6 \ (c \ 0.77, \text{CHCl}_{3,}).$

HPLC (Chiralpak OD-H, heptane/isopropanol 99.0:1.0, 0.5 mL/min, 220 nm) ee = 93%





(*R*)-N-benzylocta-1,7-dien-3-amine (4.3x)



Procedures

The allylic acetate (33.6 mg, 0.20 mmol, 100 mol%) and the primary amine (42.9 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 85% yield (36.5 mg, 0.17 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-2:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.55$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.40 - 7.17$ (m, 5H), 5.78 (ddt, J = 16.9, 10.2, 6.7 Hz, 1H), 5.62 (ddd, J = 17.1, 10.3, 8.3 Hz, 1H), 5.28 - 5.06 (m, 2H), 5.03 - 4.85 (m, 2H), 3.83 (d, J = 13.2 Hz, 1H), 3.64 (d, J = 13.2 Hz, 1H), 3.02 (td, J = 7.7, 5.4 Hz, 1H), 2.03 (tdt, J = 7.1, 5.4, 1.4 Hz, 2H), 1.61 - 1.36 (m, 4H), 1.31 (br, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{61.1, 51.2, 35.1, 33.7, 25.2.} (100 \text{ MHz}, \text{CDCl}_3): \delta = 141.2, 140.7, 138.7, 128.3, 128.1, 126.8, 116.1, 114.5,$

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{21}N[M+H^+] = 216.1747$, Found 216.1743.

<u>FTIR</u> (neat): 2928, 1640, 1454, 1216, 993, 913, 752, 698 cm⁻¹.

 $[\alpha]_{D}^{31} = -51.0 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OD-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 95%.</u>





(*R*)-*N*-(4-(pentafluoro- λ^6 -sulfaneyl)benzyl)octa-1,7-dien-3-amine (4.3y)



Procedures

The allylic acetate (33.6 mg, 0.20 mmol, 100 mol%) and the primary amine (93.3 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 73% yield (49.8 mg, 0.15 mmol) as a pale yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.72$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.67 (d, *J* = 8.6 Hz, 2H), 7.39 (d, *J* = 8.6 Hz, 2H), 5.81-5.70 (m, 1H), 5.60-5.51 (m, 1H), 5.17-5.04 (m, 2H), 4.99-4.89 (m, 2H), 3.84 (d, *J* = 14.1 Hz, 1H), 3.68 (d, *J* = 14.1 Hz, 1H), 3.00-2.92 (m, 1H), 2.05-1.98 (m, 2H), 1.5z9-1.35 (m, 5H).

¹³C NMR (125 MHz, CDCl₃): δ = 152.7 (m, *J*_{CF} = 15.7 Hz), 145.1, 141.0, 138.8, 128.4, 126.2 (2C), (m, *J*_{CF} = 4.6 Hz, 2C), 116.7, 114.8, 61.3, 50.4, 35.4, 33.9, 25.4.

¹⁹**F** NMR (376 MHz, CDCl₃): $\delta = 87.2 - 83.4$ (m, 1F), 63.0 (d, J = 149.8 Hz, 4F).

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{20}F_5NS [M+H^+] = 342.1309$, Found 342.1302.

<u>FTIR</u> (neat): 2926, 1641, 1459, 1414, 1095, 993, 915, 830, 669 cm⁻¹.

 $[\alpha]_{D}^{23} = -37.2 \ (c \ 1.3, \text{CHCl}_3).$

<u>HPLC</u> (2x Chiralcel AS-H column, hexanes:i-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 92%.







benzyl (R)-4-(octa-1,7-dien-3-ylamino)piperidine-1-carboxylate (4.3z)



Procedures

The allylic acetate (33.6 mg, 0.20 mmol, 100 mol%) and the primary amine (93.7 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 70% yield (48.0 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-2:1).

<u>**TLC (SiO**</u>₂) $R_f = 0.28$ (ethyl acetate).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.36 - 7.25$ (m, 5H), 5.76 (ddt, J = 16.9, 10.2, 6.7 Hz, 1H), 5.51 (ddd, J = 17.0, 10.3, 8.3 Hz, 1H), 5.09 (s, 2H), 5.08 - 4.89 (m, 4H), 4.08 (br, 2H), 3.16 - 3.05 (m, 1H), 2.95 - 2.75 (m, 2H), 2.66 (tt, J = 10.2, 3.9 Hz, 1H), 2.09 - 1.96 (m, 2H), 1.87 (br, 1H), 1.79 - 1.66 (m, 1H), 1.47 - 1.01 (m, 8H).

¹³C NMR (100 MHz, CDCl₃): δ = 155.2, 141.6, 138.6, 136.8, 128.4, 127.9, 127.7, 115.5, 114.5, 66.9, 58.2, 51.3, 42.9, 42.6, 35.5, 33.6, 25.2.

<u>HRMS</u> (ESI): Calculated for $C_{21}H_{30}N_2O_2$ [M+H⁺] = 343.2380, Found 343.2376.

<u>FTIR</u> (neat): 2930, 1691, 1433, 1364, 1228, 996, 914, 752, 697cm⁻¹.

 $[\alpha]_{D}^{25} = +51.0 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (two connected chiralcel OD-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 91%.</u>





(*R*)-*N*-(2-(1*H*-indol-3-yl)ethyl)octa-1,7-dien-3-amine (4.3a')



Procedures

The allylic acetate (33.6 mg, 0.20 mmol, 100 mol%) and the primary amine (64.1 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 69% yield (37.0 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, ethyl acetate).

<u>**TLC** (SiO₂)</u> $R_f = 0.15$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.10$ (s, 1H), 7.62 (d, J = 7.9 Hz, 1H), 7.36 (d, J = 7.9 Hz, 1H), 7.16 (t, J = 7.6 Hz, 1H), 7.12 (t, J = 7.6 Hz, 1H), 7.03 (s, 1H), 5.80 - 5.70 (m, 1H), 5.55 (ddd, J = 16.7, 10.8, 8.4 Hz, 1H), 5.13-5.04 (m, 2H), 5.00-4.98 (m, 2H), 3.03-2.92 (m, 4H), 2.87-2.81 (m, 1H), 2.07-1.95 (m, 2H), 1.59-1.22 (m, 5H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{11}} (125 \text{ MHz, CDCl}_3): \delta = 141.5, 138.9, 136.6, 127.6, 122.2, 122.1, 119.4, 119.1, 116.2, 114.7, 114.3, 111.3, 62.2, 47.5, 35.3, 33.9, 26.0, 25.4.$

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{24}N_2$ [M+H⁺] = 269.2012, Found 269.2017.

<u>FTIR</u> (neat): 3296, 2969, 2928, 1457, 1414, 1378, 1339, 1306, 1231, 1160, 1127, 1104, 994, 951, 916, 816, 739 cm⁻¹.

 $[\alpha]_{D}^{24} = +48.5 \ (c \ 1.5, \text{CHCl}_3).$

<u>HPLC</u> (2x Chiralcel OD-H column , hexanes:i-PrOH = 90:10, 1.00 mL/min, 230 nm), ee = 91%.





(*R*)-*N*-benzyl-5-phenylpent-1-en-3-amine (4.3b')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 78% yield (85.3 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.30 – 7.12 (m, 10H), 5.68 (ddd, *J* = 17.3, 10.3, 8.2 Hz, 1H), 5.24 – 5.10 (m, 2H), 3.83 (d, *J* = 13.1 Hz, 1H), 3.64 (d, *J* = 13.1 Hz, 1H), 3.12 - 3.03 (m, 1H), 2.74 – 2.56 (m, 2H), 1.91 – 1.68 (m, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{125.7}, 116.4, 60.8, 51.2, 37.4, 32.2.} (100 \text{ MHz}, \text{CDCl}_3): \delta = 142.2, 141.0, 140.7, 128.4, 128.4, 128.3, 128.2, 126.8, 125.7, 116.4, 60.8, 51.2, 37.4, 32.2.}$

HRMS (ESI): calcd for C₁₈H₂₂N [M+H]⁺: 252.1752, found: 252.1743

<u>FTIR</u> (neat): 3061, 3025, 2920, 1639, 1602, 1495, 1452, 1108, 1072, 1028, 993, 918, 742, 696 cm⁻¹.

 $[\alpha]_D^{24} = +7.4 \ (c \ 0.48, \text{CHCl}_{3,}).$

HPLC (Chiralpak OD-H; heptane/isopropanol 99:1.0, 1.0 mL/min, 220 nm) ee = 94%







(R)-N-(cyclopropylmethyl)-5-phenylpent-1-en-3-amine (4.3c')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (62.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 75% yield (70.3 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane : methanol = 1:0-20:1).

<u>**TLC (SiO**</u>₂) $R_f = 0.45$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.32 - 7.12$ (m, 5H), 5.63 (ddd, J = 17.0, 10.3, 8.3 Hz, 1H), 5.19 – 5.06 (m, 2H), 3.09 - 2.99 (m, 1H), 2.74 - 2.54 (m, 2H), 2.46 (dd, J = 12.0, 7.0 Hz, 1H), 2.35 (dd, J = 12.0, 6.8 Hz, 1H), 1.92 - 1.80 (m, 1H), 1.80 - 1.67 (m, 1H), 1.00 – 0.85 (m, 1H), 0.52 – 0.39 (m, 2H), 0.16 – 0.03 (m, 2H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.1, 141.0, 128.4, 128.3, 125.7, 116.4, 61.4, 52.4, 37.1, 32.2, 11.3, 3.5, 3.3.

HRMS (ESI): calcd for C₁₅H₂₂N [M+H]⁺: 216.1752, found: 216.1742

<u>FTIR</u> (neat): 3077, 3002, 2921, 1603, 1496, 1453, 1318, 1113, 1045, 1016, 994, 917, 825, 748, 697 cm⁻¹.

 $[\alpha]_D^{24} = +4.2 \ (c \ 0.42, \text{CHCl}_{3,}).$

HPLC (Chiralpak OD-H, heptane/isopropanol 99.0:1.0, 1.0 mL/min, 220 nm) ee = 93%







(R)-5-phenyl-N-(3,3,3-trifluoropropyl)pent-1-en-3-amine (4.3d')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (99.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 80% yield (89.9 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.61$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.33 - 7.14$ (m, 5H), 5.59 (ddd, J = 17.1, 10.2, 8.2 Hz, 1H), 5.22 - 5.09 (m, 2H), 3.04 - 2.96 (m, 1H), 2.88 (dt, J = 12.3, 7.3 Hz, 1H), 2.78 - 2.55 (m, 3H), 2.33 - 2.16 (m, 2H), 1.87 - 1.65 (m, 2H).

 $\frac{^{13}C \text{ NMR}}{^{2}} (100 \text{ MHz, CDCl}_3): \delta = 141.9, 140.5, 128.4(x2), 125.8, 116.8, 61.3, 40.0 (q, J) = 3.7 \text{ Hz}, 37.2, 34.5 (d, J = 27.4 \text{ Hz}), 32.1.$

¹⁹**F** NMR (376 MHz, CDCl₃): $\delta = -65.1$ (t, J = 10.9 Hz, 3F).

HRMS (ESI): calcd for C₁₄H₁₉F₃N [M+H]⁺: 258.1470, found: 258.1460

<u>FTIR</u> (neat): 3028, 2926, 2859, 1641, 1603, 1496, 1454, 1440, 1384, 1339, 1252, 1139, 1030, 996, 921, 845, 748, 698, 654 cm⁻¹.

 $[\alpha]_D^{24} = +5.9 \ (c \ 0.47, \text{CHCl}_{3}).$

HPLC (Chiralpak OD-H, heptane/isopropanol 99.0:1.0, 1.0 mL/min, 220 nm) ee = 87%



€65.02 €65.05

10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -1; f1 (ppm)





(R)-N-((2-methyl-1,3-dioxolan-2-yl)methyl)-5-phenylpent-1-en-3-amine (4.3e')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (103 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 83% yield (94.7 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0-0:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.39$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.31 - 7.22$ (m, 2H), 7.21 - 7.12 (m, 3H), 5.62 (ddd, J = 17.1, 10.3, 8.1 Hz, 1H), 5.18 - 5.06 (m, 2H), 4.01 - 3.87 (m, 4H), 3.06 - 2.97 (m, 1H), 2.76 (d, J = 12.3 Hz, 1H), 2.73 - 2.59 (m, 2H), 2.56 (d, J = 12.2 Hz, 1H), 1.89 - 1.65 (m, 2H), 1.37 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.3, 141.2, 128.4, 128.3, 125.7, 116.1, 109.5, 64.9, 64.9, 61.5, 53.6, 37.3, 32.2, 22.9.

HRMS (ESI): calcd for C₁₆H₂₄NO₂ [M+H]⁺: 262.1807, found: 262.1795

<u>FTIR</u> (neat): 2980, 2879, 1603, 1496, 1454, 1376, 122, 1162, 1118, 1051, 994, 945, 918, 859, 749, 698 cm⁻¹.

 $[\alpha]_D^{24} = +1.7 \ (c \ 0.47, \text{CHCl}_{3,}).$

<u>HPLC</u> (Chiralpak OD-H (2x), heptane/isopropanol 99.6:0.4; 1.0 mL/min, 220 nm), ee = 93%






tert-butyl-(S)-3-((((R)-5-phenylpent-1-en-3-yl)amino)methyl)piperidine-1carboxylate (4.3f')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (189 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 83% yield, 17:1 dr (130 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: methanol = 1:0–20:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.39$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, DMSO- d_6): $\delta = 7.31 - 7.11$ (m, 5H), 5.59 (ddd, J = 17.0, 10.4, 7.9 Hz, 1H), 5.12 – 5.00 (m, 2H), 4.07 – 3.85 (m, 1H), 3.80 - 3.67 (m, 1H), 2.86 (q, J = 7.0 Hz, 1H), 2.77 (ddd, J = 13.6, 11.0, 3.1 Hz, 1H), 2.60 (t, J = 8.0 Hz, 2H), 2.56 – 2.43 (m, 1H), 2.33 (dd, J = 11.8, 8.1 Hz, 1H), 2.25 (dd, J = 11.9, 5.5 Hz, 1H), 1.78 – 1.20 (m, 15H), 1.14 - 1.00 (m, 1H).

 $\frac{^{13}C \text{ NMR}}{61.4, 50.0, 37.2, 36.1(x2), 32.2, 29.1, 28.4(x2), 24.4.}$

HRMS (ESI): calcd for C₂₂H₃₅N₂O₂ [M+H]⁺: 359.2699, found: 359.2686

<u>FTIR</u> (neat): 2975, 2928, 2853, 1685, 1496, 1420, 1390, 1364, 1265, 1240, 1174, 1147, 1030, 994, 967, 917, 881, 750, 698, 665 cm⁻¹. $[\alpha]_D^{24} = -14.3 \ (c \ 0.41, CHCl_3).$



(*R*)-*N*-(5-phenylpent-1-en-3-yl)cyclopropanamine (4.3g')



Procedures

The allylic acetate (40.9 mg, 0.20 mmol, 100 mol%) and the primary amine (22.8 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 79% yield (31.8 mg, 0.16 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.54$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.24-7.17$ (m, 2H), 7.14-7.07 (m, 3H), 5.67-5.56 (m, 1H), 5.11-5.08 (m, 1H), 5.07-5.04 (m, 1H), 3.10-3.02 (m, 1H), 2.63-2.47 (m, 2H), 2.07-2.00 (m, 1H), 1.83-1.72 (m, 1H), 1.68-1.56 (m, 1H), 0.42-0.26 (m, 3H), 0.24-0.16 (m, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 142.5, 141.5, 128.6 (2C), 128.5 (2C), 125.9, 116.0, 61.9, 37.1, 32.4, 28.6, 7.0, 6.5.

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{19}N[M+H^+] = 202.1590$, Found 202.1586.

<u>FTIR</u> (neat): 3084, 3026, 2924, 1738, 1639, 1603, 1496, 1453, 1369, 1237, 1014, 993, 916, 827, 746, 697 cm⁻¹.

 $[\alpha]_D^{24} = -11.3 \ (c \ 1.1, \text{CHCl}_{3}).$

<u>**HPLC</u>** (Chiralcel OD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 94%.</u>





tert-Butyl (R)-3-((5-phenylpent-1-en-3-yl)amino)azetidine-1-carboxylate (4.3h')



Procedures

The allylic acetate (89.9 mg, 0.44 mmol, 100 mol%) and the primary amine (152 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 74% yield (102 mg, 0.32 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptane: isopropyl acetate = 1:0–0:1).

<u>**TLC (SiO₂)</u>** $R_f = 0.32$ (hexanes/ethyl acetate = 1:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.32 - 7.22$ (m, 2H), 7.23 - 7.13 (m, 3H), 5.60 (ddd, *J* = 17.0, 10.2, 8.5 Hz, 1H), 5.18 - 5.06 (m, 2H), 4.09 - 3.97 (m, 2H), 3.64 - 3.52 (m, 3H), 3.04 - 2.94 (m, 1H), 2.73 - 2.53 (m, 2H), 1.87 - 1.65 (m, 2H), 1.43 (s, 9H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 156.1, 141.7, 140.6, 128.4, 128.3, 125.9, 116.9, 79.0,
60.1, 45.9, 37.2, 32.1, 28.4, 25.4.

HRMS (ESI): calcd for C₁₉H₂₉N₂O₂ [M+H]⁺: 317.2229, found: 317.2215

<u>FTIR</u> (neat): 2973, 2876, 1692, 1603, 1496, 1476, 1454, 1401, 1365, 1290, 1251, 1158, 1118, 1030, 995, 919, 861, 750, 698 cm⁻¹.

 $[\alpha]_D^{24} = +4.5 \ (c \ 0.40, \text{CHCl}_{3,}).$

HPLC (Chiralpak OD-H, heptane/isopropanol 70.0:30.0, 1.0 mL/min, 220 nm) ee = 91%





(R)-N-(5-phenylpent-1-en-3-yl)tetrahydro-2H-thiopyran-4-amine (4.3i')



Procedures

The allylic acetate (40.9 mg, 0.20 mmol, 100 mol%) and the primary amine (46.9 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 79% yield (41.8 mg, 0.16 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1).

<u>**TLC** (SiO</u>₂) $R_f = 0.50$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.30 – 7.25 (m, 2H), 7.20 - 7.15 (m, 3H), 5.58 (ddd, *J* = 17.2, 10.2, 8.3 Hz, 1H), 5.13 (d, *J* = 10.2 Hz, 1H), 5.08 (d, *J* = 17.2 Hz, 1H), 3.2-3.12 (m, 1H), 2.72-2.46 (m, 7H), 2.17-2.11 (m, 1H), 2.07-2.01 (m,1H), 1.83-1.65 (m, 2H), 1.57-1.48 (m, 1H), 1.44-1.34 (m, 1H), 1.09 (bs, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 142.3, 141.7, 128.6 (2C), 128.5 (2C), 126.0, 116.1, 57.9, 52.6, 37.8, 35.9, 34.3, 32.4, 28.0, 27.7.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{23}NS[M+H^+] = 262.1624$, Found 262.1620.

<u>FTIR</u> (neat): 3024, 2922, 2841, 1775, 1697, 1638, 1602, 1495, 1453, 1427, 1365, 1268, 1235, 1118, 1030, 994, 916, 747, 698, 659 cm⁻¹.

 $[\alpha]_D^{24} = -15.6 \ (c \ 1.4, \text{CHCl}_{3,}).$

<u>HPLC</u> (Chiralcel AS-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 95%.





(S)-N-benzyl-1-cyclopropylprop-2-en-1-amine (4.3j')



Procedures

The allylic acetate (28.0 mg, 0.20 mmol, 100 mol%) and the primary amine (42.9 mg, 0.40 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 76% yield (28.5 mg, 0.15 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 1:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.4$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.29-7.21$ (m, 4H), 7.21-7.15 (m, 1H), 5.74 (m, 1H), 5.09-5.02 (m, 2H), 3.78, (d, J = 13.3 Hz, 1H), 3.62, (d, J = 13.3 Hz, 1H), 2.24 (dd, J = 8.6, 7.8 Hz, 1H), 1.58 (s, 1H), 0.89 – 0.80 (m, 1H), 0.5-0.35 (m, 2H), 0.15-0.05.

 $\frac{^{13}C \text{ NMR}}{^{66.3}, 51.6, 16.8, 4.3, 2.5.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 140.8, 140.5, 128.6 (2C), 128.3 (2C), 127.0, 115.8, 166.3, 51.6, 16.8, 4.3, 2.5.$

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N[M+H^+] = 188.1434$, Found 188.1433.

<u>FTIR</u> (neat): 3339, 3076, 3025, 3002, 2815, 1640, 1603, 1494, 1453, 1428, 1413, 1358, 1302, 1172, 1101, 1074, 1046, 1018, 993, 917, 859, 822, 733, 697, 672 cm⁻¹. $[\alpha]_D^{24} = -14.6 \ (c \ 1.4, CHCl_3).$

<u>**HPLC</u>** (Chiralcel OD-H column, hexanes:*i*-PrOH = 97.5:2.5, 1.00 mL/min, 230 nm), ee = 90%.</u>





(R)-1-benzyl-4-(but-3-en-2-yl)piperazine (4.4l)



Procedures

To a round bottomed flask equipped with a magnetic stir bar charged with **4.31** (60 mg, 0.3 mmol, 100 mol%) dissolved in methanol (30 ml, 0.01M) was added glyoxal (4.5 μ l, 0.33 mmol, 110 mol%, of a 40 wt% aqueous solution). This mixture was cooled to 0 °C with an icewater bath, before sodium cyanoborohydride (37.0 mg, 0.63 mmol, 210 mol%) was added. The resulting reaction mixture was stirred for 17 hours. During this time the cooling bath was allowed to warm to room temperature. The reaction was quenched with NaHCO₃ (20 ml of a saturated aqueous solution) and further diluted with water (20 ml). The obtained aqueous solution was extracted with CH₂Cl₂ (3 x 30 ml). The combined organic extracts were dried over MgSO₄, filtered and the solvent was removed *in vacuo*. The resulting residue was subjected to flash column chromatography (SiO₂, hexanes, ethyl acetate 1:1) to furnish the title compound as a pale yellow oil in 69 % yield (47.5 mg, 0.21 mmol).

<u>TLC</u> (SiO₂) $R_f = 0.20$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.30-7.26 (m, 4H), 7.25-7.20 (m, 1H), 5.76 (ddd, *J* = 17.1, 10.4, 7.9 Hz, 1H), 5.10-5.02 (m, 2H), 3.49 (s, 2H), 2.91-2.83 (m, 1H), 2.49 (bs, 8H), 1.14 (d, *J* = 6.6 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{^{63.4}, 63.3, 53.5} (2C), 50.0 (2C), 17.7.$

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{22}N_2$ [M+H⁺] =231.1856, Found 231.1851.

<u>FTIR</u> (neat): 2932, 2807, 2764, 1494, 1453, 1372, 1318, 1265, 1139, 1073, 1048, 1028, 1011, 953, 916, 815, 736, 697 cm⁻¹. $[\alpha]_D^{24} = -18.3 \ (c \ 1.5, CHCl_3).$

<u>**HPLC</u></u> (Chiralcel AD-H column (2x), hexanes:***i***-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 96\%.</u>**







(R)-N-benzylcyclohex-2-en-1-amine (4.4x)



Detailed Procedures

To a round-bottomed flask with a magnetic stir bar was charged with **4.3x** (32.3 mg, 0.15 mmol, 100 mol%). Under argon atmosphere, toluene (6.5 mL, 0.02 M) was added via syringe, followed by the addition of a solution of Grubbs Catalyst II (19.1 mg, 0.023 mmol, 15 mol%) in toluene (1.0 mL). The reaction mixture was heated to 70 °C. After this mixture was cooled to ambient temperature, it was concentrated *in vacuo*. The residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1-2:1) to give 19.1 mg, 0.10 mmol (68%) as a light dark oil.

<u>**TLC** (SiO₂)</u> $R_f = 0.21$ (hexanes/ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.39 - 7.22$ (m, 5H), 5.87 - 5.64 (m, 2H), 3.88 (d, J = 13.0 Hz, 1H), 3.83 (d, J = 13.0 Hz, 1H), 3.22 (dp, J = 7.0, 2.3 Hz, 1H), 2.04 - 1.86 (m, 3H), 1.81 - 1.71 (m, 1H), 1.62 - 1.42 (m, 2H), 1.31 (br, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 140.7, 129.9, 128.9, 128.3, 128.1, 126.8, 52.3, 51.0, 29.4, 25.3, 20.2.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N[M+H^+] = 188.1434$, Found 188.1430.

<u>FTIR</u> (neat): 2929, 1452, 1215, 1103, 748, 689, 667 cm⁻¹.

 $[\alpha]_{D}^{26} = +67.5 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel AS-H column, hexanes:i-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 95%.





(S)-N-benzyl-N-(1-cyclopropylallyl)acrylamide (4.4j*)



Procedures

To a round bottomed flask equipped with a magnetic stir bar charged with **3j'** (90 mg, 0.48 mmol, 100 mol%) dissolved in absolute dichloromethane (10 ml, 0.05M) under an athmosphere of argon, acryloyl chloride (47.8 mg, 0.53 mmol, 110 mol%), triethylamine (97.3 mg, 0.96 mmol, 200 mol%) and Pyridine (2.0 mg, 0.024 mmol, 5 mol%) were added in this order. The resulting reaction mixture was stirred for 17 hours at room temperature. After this time the solvent was removed *in vacuo* and the resulting residue was directly subjected to flash column chromatography (SiO₂, hexanes, ethyl acetate 10:1 -> 5:1) to furnish the title compound as a colorless viscous oil in 89 % yield (104 mg, 0.48 mmol). According to the recorded NMR data the compound exists in solution as a 4:1 mixture of rotamers.

<u>**TLC** (SiO₂</u>) $R_f = 0.43$ (hexanes/ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.37-7.14 (m, 7.4H, major, minor), 6.60-6.49 (m, 0.4H, minor), 6.64-6.31 (m, 2.4H, major, minor), 5.87-5.75 (m, 1.4H, major, minor), 5.70-5.57 (m, 1.4 H, major, minor), 5.35-5.15 (m, 2.8H, major, minor), 4.92 (d, *J*=15.4 Hz, 0.4H, minor), 4.67 (d, *J*=18.4 Hz, 1.0H, major), 4.59-4.49 (m, 2.0H, major), 4.49 (d, *J*=15.4 Hz, 0.4H, minor), 3.66-3.56 (m, 0.3H, minor), 0.98-0.88 (m, 0.4 H, minor), 0.88-0.76 (m,

1.0H, major), 0.66-0.54 (m, 1.4H, major, minor), 0.47-0.38 (m, 1.0H, major), 0.33-0.17 (m, 2.8H, major, minor), 0.15-0.7 (m, 0.4H, minor).

¹³C NMR (125 MHz, CDCl₃): δ = 167.6 (minor), 167.2 (major), 139.2 (minor), 138.6

(major), 136.7 (major), 136.6 (minor), 129.1 (major, minor), 128.7 (major), 128.5

(major), 128.4 (minor), 128.1 (minor), 127.7 (minor), 127.3 (major), 126.9 (minor), 126.4

(major), 117.1 (major), 116.9 (minor), 65.5 (minor), 61.5 (major), 47.8 (major), 47.1

(minor), 14.3 (minor), 12.8 (major), 6.0 (minor), 5.5 (major), 4.6 (major), 4.5 (minor).

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{19}N[M+Na^+] = 264.1359$, Found 264.1355.

<u>FTIR</u> (neat): 3083, 3004, 1647, 1609, 1495, 1452, 1419, 1358, 1322, 1289, 1234, 1199, 1161, 1078, 1058, 1021, 978, 956, 923, 832, 794, 732, 697 cm⁻¹.

 $[\alpha]_D^{24} = +3.9 \ (c \ 1.4, \text{CHCl}_{3,}).$

<u>HPLC</u> (Chiralcel OD-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 89%.







(R)-1-benzyl-5-cyclopropyl-1,5-dihydro-2H-pyrrol-2-one (4.4j')



Procedures

To a round bottomed flask equipped with a magnetic stir bar and charged with Grubbs II (4.1 mg, 0.005 mmol, 5 mol%) 1.0 ml of a 0.1 M solution of $4j^*$ in toluene is added under an atmosphere of argon. (Caution: The substrate solution was degassed by bubbling argon through the solution for 3 minutes before usage.) The resulting reaction mixture is heated to 60 °C for 17 h. After this time the reaction mixture is directly subjected to flash column chromatography (SiO₂, hexanes, ethyl acetate 5:1–2:1) to furnish the compound as a pale violet oil in 65 % yield (15.6 mg, 0.065 mmol).

<u>**TLC (SiO₂**</u>) $R_f = 0.15$ (hexanes/ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.34-7.16 (m, 5H), 6.99 (dd, *J* = 1.8, 5.9 Hz, 1H), 6.19 (dd, *J* = 1.8, 5.9 Hz, 1H), 5.15 (d, *J* = 15.4 Hz, 1H), 4.31 (d, *J* = 15.4 Hz, 1H), 3.14 (td, *J* = 1.8, 9.1 Hz), 0.71-0.44 (m, 3H), 0.31-0.23 (m, 1H), 0.17-0.10 (m, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{43.9, 11.9, 5.1, 0.2, 0.1.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 171.3, 147.7, 138.1, 128.8, 127.8, 127.5, 127.1, 66.9,$

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{15}N[M+H^+] = 214.1226$, Found 214.1222.

<u>FTIR</u> (neat): 3004, 2922, 1683, 1495, 1433, 1403, 1358, 1338, 1270, 1223, 1078, 1026, 969, 867, 852, 807, 703 cm⁻¹.

 $[\alpha]_D^{24} = +4.5 \ (c \ 1.7, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OD-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 89%.</u>





tert-butyl (4-((1R,2R)-1-hydroxy-2-methylbut-3-en-1-yl)benzyl)carbamate (4.6)



Procedures

An oven-dried sealed tube under an atmosphere of N₂ was charged with *tert*-butyl (4-(hydroxymethyl)benzyl)carbamate (5) (150 mg, 0.63 mmol, 100 mol%), (*R*)-IrLn (32.5 mg, 0.03 mmol, 5 mol%), K₃PO₄ (67 mg, 0.32 mmol 50 mol%), THF (0.31 ml, 2.0 M) and H₂O (57 μ l, 3.2 mmol, 500 mol%). But-3-en-2-yl acetate (144mg, 1.26 mmol, 200 mol%) was added and the reaction mixture was allowed to stir at ambient temperature for 0.5 h. The reaction vessel was placed in an oil bath preheated to 60 °C. The reaction was stirred at this temperature for 48 h. After this time the reaction muxture was concentrated *in vacuo*. Purification of the remaining residue by column chromatography (SiO₂; hexanes: ethyl acetate 4:1) provided **6** as a colorless viscous oil in 83 % yield (152 mg, 0.52 mmol, 8:1 dr).

<u>TLC (SiO</u>₂) $R_f = 0.30$ (hexanes:ethyl acetate 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.32-7.21 (m, 4H), 5.77 (ddd, J= 17.2, 10.3, 8.3 Hz, 1H), 5.24-5.14 (m, 2H), 4.36-4.25 (m, 3H), 2.51-2.39 (m, 1H), 1.44 (s, 9H), 0.84 (d, J = 6.9 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{17}} (125 \text{ MHz}, \text{ CDCl}_3): \delta = 156.1, 141.7, 140.7, 138.5, 127.6 (2C), 127.3 (2C), 117.2, 79.7, 77.5, 46.6, 44.6, 28.6 (3C), 16.73.$

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{25}NO_3 [M+Na^+] = 314.1727$, Found 314.1720.

<u>FTIR</u> (neat) 3359, 2976, 2929, 1687, 1513, 1455, 1421, 1391, 1365, 1271, 1249, 1211, 1164, 1046, 1017, 915, 864, 82:5, 783, 676 cm⁻¹. $[\alpha]_D^{24} = -34.3 \ (c \ 1.1, \text{CHCl}_3).$

<u>**HPLC</u></u> (Chiralcel AS-H column (2x), hexanes:***i***-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 90\%.</u>**





(1R,2R)-1-(4-(aminomethyl)phenyl)-2-methylbut-3-en-1-ol (4.6')



Procedures

To a round bottomed flask equipped with a magnetic stir bar charged with **6** (140 mg, 0.48 mmol, 100 mol%) was added DCM (0.4 ml) and afterwards TFA (80 μ l). The resulting reaction mixture (0.1 M) was stirred for two hours at room temperature. After this time water (5 ml) and NaOH (20 ml of an 0.1 M aqueous solution) were added. The phases were separated and the aqueous phase was extracted with DCM (3x 20 ml). The combined organic extracts were dried over MgSO₄, filtered and the solvent was removed *in vacuo*. The resulting residue was subjected to flash column chromatography (SiO₂, hexanes, ethyl acetate 1:1) to furnish the title compound as a colorless oil in 84 % yield (77.0 mg, 0.40 mmol, 8:1 dr).

<u>TLC (SiO</u>₂) $R_f = 0.12$ (hexanes: ethyl acetate 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.31-7.25 (m, 4H), 5.77 (ddd, J= 17.1, 10.4, 8.3 Hz, 1H), 5.23-5.14 (m, 2H), 4.33 (d, J= 7.8 Hz, 1H), 3.84 (s, 2H), 2.51-2.41 (m, 1H), 1.64 (bs, 3H), 0.85 (d, J = 6.8 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{^{77.7}, 46.4, 46.2, 16.6}$ (125 MHz, CDCl₃): $\delta = 141.00, 140.7, 140.3, 127.1$ (2C), 127.0 (2C), 117.0,

HRMS (ESI): Calculated for C₁₂H₁₇NO [M+H⁺] =192.1383, Found 192.1385. **FTIR** (neat): 3293, 3076, 2971, 2928, 2868, 1737, 1638, 1511, 1455, 1438, 1415, 1371, 1262, 1228, 1175, 1097, 1034, 1017, 995, 914, 821, 735, 300 cm⁻¹. $[\alpha]_D^{24} = -19.8 \ (c \ 1.2, CHCl_3).$



(1*R*,2*R*)-1-(4-((((*R*)-but-3-en-2-yl)amino)methyl)phenyl)-2-methylbut-3-en-1-ol (4.7)



Procedures

Synthesis starting from compound 4.6'

Using General Procedure 1: the title compound was afforded after purification by flash column chromatography (ethyl acetate) to give 57.3 mg, 0.23 mmol (78%) as a pale orange oil. According to recorded NMR data the compound consisted of a 8:1 mixture of diastereomers.

Synthesis starting from (4-(aminomethyl)phenyl)methanol (4.5')

[(R)-(+)-5,5'-bis(diphenylphosphino)-4,4'-bi-1,3-benzodioxole][4-cyano-3-

nitrobenzenecarboxylato] [1,2,3-eta-2-propenyl Ir(III)] (20.7 mg, 0.02 mmol, 10 mol%), cesium carbonate (197 mg, 0.6 mmol, 300 mol%), and 5' (27.4 mg, 0.2 mmol, 100 mol%) were added to an oven-dried resealable pressure tube equipped with a magnetic stir bar. The tube was purged with argon for 5 minutes. Afterwards THF 0.2 ml, (1.0 M) was added followed by the addition of but-3-en-2-yl acetate (91.3 mg, 0.8 mmol, 400 mol%). The tube was sealed with a PTFE lined cap and heated at 80°C for 48 h. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, ethyl acetate) to give 18 mg, 0.073 mmol (36 %) of the
title compound as a pale orange oil. According to recorded NMR data the compound consisted of a 2:1 mixture of diastereomers.

<u>**TLC (SiO₂)**</u> $R_f = 0.18$ (ethyl acetate).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.25-7.19 (m, 4H), 5.79-5.59 (m, 2H), 5.17-4.99 (m, 4H), 4.27 (d, *J* = 7.9 Hz, 1H), 3.72 (d, *J* = 12.8 Hz, 1H), 3.60 (d, *J* = 12.8 Hz, 1H), 3.18-3.11 (m, 1H), 2.45-2.35 (m, 1H), 1.77 (bs, 2H), 1.11 (d, *J* = 6.5 Hz, 3H), 0.79 (d, *J* = 6.8 Hz, 3H).

<u>1³C NMR</u> (125 MHz, CDCl₃): δ = 142.6, 141.3, 140.9, 140.1, 128.3, 127.1, 117.1, 115.1, 77.8, 56.3, 51.2, 46.5, 22.0, 16.8.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{23}NO[M+H^+] = 246.1852$, Found 246.1842.

<u>FTIR</u> (neat): 3339, 3076, 3025, 3002, 2815, 1640, 1603, 1494, 1453, 1428, 1413, 1358, 1302, 1172, 1101, 1074, 1046, 1018, 993, 917, 859, 822, 733, 697, 672 cm⁻¹. $[\alpha]_D^{24} = -38.9 \ (c \ 1.1, CHCl_3).$



Single Crystal Diffraction Data for 4.3j'

Empirical formula	C13 H18 CI N	
Formula weight	223.73	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	monoclinic	
Space group	P 21	
Unit cell dimensions	a = 10.993(2) Å	α= 90°.
	b = 9.035(3) Å	$\beta = 109.866(7)^{\circ}.$
	c = 13.737(4) Å	$\gamma = 90^{\circ}.$
Volume	1283.2(6) Å ³	
Z	4	
Density (calculated)	1.158 Mg/m ³	
Absorption coefficient	0.268 mm ⁻¹	
F(000)	480	
Crystal size	0.510 x 0.130 x 0.030 mm ³	
Theta range for data collection	3.056 to 27.480°.	
Index ranges	-14<=h<=14, -11<=k<=11, -17<=l<=17	
Reflections collected	21792	
Independent reflections	5883 [R(int) = 0.0393]	
Completeness to theta = 25.242°	99.8 %	
Absorption correction	Semi-empirical from equivalents	

Max. and min. transmission	1.00 and 0.783	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	5883 / 1 / 288	
Goodness-of-fit on F ²	1.038	
Final R indices [I>2sigma(I)]	R1 = 0.0451, wR2 = 0.1108	
R indices (all data)	R1 = 0.0533, wR2 = 0.1160	
Absolute structure parameter	0.00(10)	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.529 and -0.213 e.Å ⁻³	

Figure S1. View of the cation 1 of **4.3j**' showing the atom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



Figure S2. View of the cation 2 of **4.3j**' showing the atom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



Chapter 5: Regio- and Enantioselective Iridium-Catalyzed Amination of Branched Alkyl-Substituted Allylic Acetates with Primary and Secondary Aromatic and Heteroaromatic Amines*

5.1 Introduction

Cyclometallated π -allyliridium *C*,*O*-benzoate complexes have been shown to catalyze diverse alcohol-mediated carbonyl allylations using allyl carboxylates as pronucleophiles.¹ In these umpoled allylations,² the *C*,*O*-benzoate moiety assists in maintaining neutrality and, hence, nucleophilicity of the π -allyliridium intermediate. Neutral iridium complexes that are not cyclometallated also display nucleophilic properties.³ In contrast, as illustrated by enantioselective Tsuji-Trost-type allylic aminations developed by Takeuchi,^{4,5} Helmchen,^{6,7} Hartwig,^{8,9} Carreira¹⁰ and You,^{11,12} cationic π -allyliridium species invariably serve as electrophiles. In this latter context, two distinct classes of iridium catalysts have emerged. Type I catalysts are used under basic conditions in combination with linear allyl proelectrophiles (as branched allyl proelectrophiles react stereospecifically).¹³ Type II catalysts are used under acidic conditions in combination with branched allyl proelectrophiles, which react in a non-stereospecific fashion, perhaps due to displacement of the π -bond of (σ + π)-allyl (enyl) iridium intermediates by the tethered olefin of the phosphoramidite ligand (Figure 5.1).¹⁴

The present authors recently found that neutral π -allyliridium *C*,*O*-benzoate complexes, which typically act as nucleophiles, can also display electrophilic behavior, representing the first examples of amphiphilic reactivity in the context of transition metal catalysis.

^{*}This chapter is based on the published work:

Kim, S. W.; Schwartz, L. A.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. J. Am. Chem. Soc. 2019, 141, 671.



Figure 5.1 Cationic vs neutral chiral iridium complexes for regio- and enantioselective allylic amination.

In the initial communication of these findings, enantioselective allylic aminations of branched allylic acetates bearing linear alkyl groups with primary aliphatic amines were disclosed.¹⁵ These aminations proceed with complete branched regioselectivity, overcoming a significant limitation associated with Type I and II catalysts, which display incomplete regioselectivity for π -allyl precursors bearing linear alkyl groups.^{16,17} With regard to the amine nucleophile, the Type III SEGPHOS-modified π -allyliridium complexes used in our initial study enforced high enantioselectivities in reactions of primary aliphatic amines (Figure 5.1). Here, we show that the corresponding tol-BINAP-

modified iridium catalyst avail a significant expansion in scope, enabling highly enantioselective aminations of branched alkyl substituted allylic acetates with electronically diverse primary and secondary aryl amines, including site-selective reactions of bis(amine) nucleophiles. Additionally, we report deuterium labelling studies that corroborate C-N bond formation through an outer-sphere mechanism.

5.2 Reaction Development and Scope

To develop highly regio- and enantioselective allylic aminations mediated by aryl amines, a series of π -allyliridium *C*,*O*-benzoate complexes were evaluated in reactions of α -methyl allyl acetate (100 mol%) and aniline (200 mol%) under conditions previously optimized for primary aliphatic amines.¹⁵ The iridium catalyst modified by tol-BINAP, *(S)*-Ir-II, delivered the product of allylic amination **5.4a** with significantly higher levels of enantioselectivity than the corresponding SEGPHOS-modified catalyst, *(S)*-Ir-I, but a lower isolated yield of **5.4a** was observed (eq. 5.1). Changing the solvent from THF to DME improved the isolated yield of **5.4a**, and by decreasing the reaction temperature from 80 °C to 70 °C **5.4a** could be formed in 82% yield and 89% enantiomeric excess (eq. 5.1).



Table 5.1Iridium-catalyzed amination of α -methyl allyl acetate **5.1a** with primary
aromatic and heteroaromatic amines **5.2a-5.2l** to form enantiomerically
enriched allylic amines **5.4a-5.4l**.^a



^aYields of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

Table 5.2Iridium-catalyzed amination of α -methyl allyl acetate 5.1a with secondary
aromatic and heteroaromatic amines 5.3a-5.3l to form enantiomerically
enriched allylic amines 5.5a-5.5l.^a



^aYields of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

Deviation from these reaction parameters did not avail further improvement, and given the low cost of tol-BINAP these conditions were adopted to explore the scope of primary aromatic and heteroaromatic amine nucleophiles **5.2a-5.2l** in aminations of α -methyl allyl acetate (Table 5.1). Amine nucleophiles containing a diverse array of functional groups were examined to mirror challenges faced in medicinal chemistry. In each case, the targeted products of allylic amination **5.4a-5.4l** were formed with complete branched regioselectivity and uniformly high levels of enantioselectivity. As illustrated in the formation of **5.4d**, which incorporates a pinacol boronate moiety, the reaction conditions tolerate rather sensitive functional groups. The tolerance of ortho-substituted anilines, as demonstrated by the formation of **5.4e**, is also noteworthy. Perhaps the most striking feature, however, is the compatibility of the catalyst with electronically diverse aryl amine partners and the tolerance of Lewis basic *N*-heterocycles, as illustrated by the formation **5.4a**, which has been prepared in enantiomerically enriched form in two separate reports.^{8d,18}

In a further exploration of scope, optimized conditions were applied to the amination of α -methyl allyl acetate using secondary aromatic and heteroaromatic amine nucleophiles **5.3a-5.3l** (Table 5.2). Indoline **5.3a**, 6- and 7-aza-indolines **5.3b** and **5.3c** and the 3,3-spirocyclic indoline **5.3d** each underwent asymmetric allylation with complete branched regioselectivity and high levels of enantioselectivity. Pronounced match-mismatched effects were observed in the conversion of chiral nonracemic (*S*)- and (*R*)-2-methyl indolines **5.3e** and **5.3f** to adducts **5.5e** and **5.5f**, respectively, suggesting the potential for kinetic resolution. The amination of **5.1a** using *N*-methyl aniline **5.3g** and related compounds **5.3h** and **5.3i** bearing electron withdrawing and donating groups at the para-position proceeded smoothly to form adducts **5.5g-5.5i**, respectively. Among these

three *N*-methyl aniline derivatives (**5.3g-5.3i**), amination using the more electron rich *N*methyl-para-anisidine **5.3i** occurred with notably higher levels of enantioselectivity. As illustrated by the formation of **5.5j** and **5.5k**, *N*-methyl anilines containing bromide (**5.3j**) and chloride (**5.3k**) functional groups are tolerated. Finally, amination of **5.1a** using *N*-Methyl-2-(methylamino)benzimidazole **5.3l** is remarkably efficient, providing adduct **5.5l** in 94% yield with complete selectivity for allylation of the extranuclear 2-(methylamino) moiety.

To assess how structural variation of the π -allyliridium intermediate impacts reactivity, regio- and stereoselectivity, a set of branched allylic acetates **5.1a-5.1g** were explored in aminations mediated by 2-(methylamino)benzoxazole 5.3m (Table 5.3). In addition to α -methyl allyl acetate **5.1a**, linear alkyl substituted allylic acetates **5.1b** and **5.1c** smoothly underwent amination to form adducts **5.6a-5.6c** as single regioisomers with high levels of enantiomeric enrichment. Allylic acetates **5.1d-5.1f**, which incorporate cycloalkyl-substituents, delivered adducts **5.6d-5.6f** as single regioisomers, although an erosion in enantioselectivity is observed using the larger cyclopentylsubstituted allyl acetate **5.1f**. Finally, using the enantiomeric iridium catalysts (*S*)-Ir-II and (*R*)-Ir-II, the (*S*)-citronellol-derived allylic acetate **5.1g** reacts with **5.3m** to form **5.6g** and iso-**5.6g**, respectively, with good levels of catalyst-directed diastereoselectivity.

Table 5.3Iridium-catalyzed amination of α -substituted allyl acetates 5.1a-5.1g with
secondary heteroaromatic amine 5.3m to form enantiomerically enriched
allylic amines 5.6a-5.6g.^a



^aYields of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

Having established the ability to functionalize both primary and secondary aromatic amines, the site-selective modification of reactants **5.3n-5.3p**, which incorporate both primary and secondary aromatic amines, was attempted using the branched allylic acetate **5.1a** (eq. 5.2-5.4). Upon exposure to standard conditions, 5-aminoindole **5.3n** undergoes completely chemoselective functionalization at the primary amine to form

adduct **5.7a** as a single constitutional isomer with excellent levels of enantioselectivity (eq. 5.2). Similarly, in the conversion of **5.3o** to adduct **5.7b**, complete control of regioand site-selectivity is accompanied by high levels of enantioselectivity (eq. 5.3). The structure of adduct **5.7b** was verified by single crystal X-ray diffraction analysis, further corroborating the absolute stereochemical assignment of adducts **5.4a-5.4l**, **5.5a-5.5l** and **5.6a-5.6g**. Finally, *N*-cyclohexyl-1,2-diaminobenzene **5.3p** reacts with **5.1a** to deliver adduct **5.7c**, which is modified exclusively at the primary amine (eq. 5.4). The ability to engage diamines in site-selective regio- and enantioselective amination enhances step economy by avoiding manipulations devoted to *N*-protection-deprotection.



Scheme 5.1 Iridium-catalyzed amination of enantiomerically enriched deuterated allylic acetate **5.1h** with the enantiomeric catalysts (*S*)-Ir-II and (*R*)-Ir-II.^a



^aYields of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

5.3 Discussion

To better understand the nature of the C-N bond forming event, asymmetric amination of the enantiomerically enriched (*Z*)-deuterated allylic acetate **5.1h** was conducted under standard conditions using (*S*)-Ir-II (Scheme 5.1, eq. 5.5).¹⁹ Compound **5.8a** is formed with complete alkene (*Z*)-stereoselectivity, as determined by ¹H NMR. Assuming formation of the π -allyliridium occurs with inversion of stereochemistry, as established in analogous iridium phosphoramidite catalyzed processes,⁴⁻¹² the stereochemistry of the amination product **5.8a** is consistent with outer-sphere addition of the nitrogen nucleophile. To corroborate the veracity of this experiment, the amination of allylic acetate **5.1h** was conducted using the enantiomeric iridium catalyst, (*R*)-Ir-II (Scheme 1, eq. 5.6). The amination product **5.8b** is formed with complete alkene (*E*)- stereoselectivity, as de-termined by ¹H NMR. The stereochemistry of **5.8a** is again consistent with outer-sphere C-N bond formation.

Based on the collective data, a general catalytic mechanism and stereochemical model were proposed (Scheme 5.2). The π -allyliridium(I) complex I is subject to outersphere amine addition amine to form the C-N bond and the zwitterionic iridium(I) olefin complex II. Deprotonation of ammonium moiety of complex II mediated by cesium carbonate generates the anionic iridium(I) species III. Alkene exchange with the allylic acetate releases the product of allylic amination and forms the olefin complex IV. Loss of acetate ion regenerates the π -allyliridium(I) complex I to close the catalytic cycle. The indicated stereochemical model accounts for the observed sense of absolute stereoinduction for outer sphere addition of a nucleophile to the neutral iridium π -allyl complex I. This model is based upon the coordination mode revealed in closely related crystal structures.²⁰ Orientation of the π -allyl is controlled through alleviation of steric clashes between the naphthyl and tolyl substituents of the phosphine ligand with the R-group of the resulting π -allyl, as illustrated in the disfavored mode of addition (Scheme 5.2).

Scheme 5.2 General catalytic mechanism and stereochemical model for enantioselective iridium-catalyzed allylic amination.



5.4 Conclusion

Previously reported enantioselective allylic aminations are largely restricted to chiral iridium-phosphoramidite catalysts.⁴⁻¹² We have shown that air and water stable π -allyliridium *C*,*O*-benzoates, which are well-known for their ability to catalyze nucleophilic carbonyl allylation,^{1,2} also promote highly regio- and enantioselective electrophilic allylation of aliphatic amines¹⁵ and, as demonstrated here, primary and secondary aromatic or heteroaromatic amines. These π -allyliridium *C*,*O*-benzoate catalyzed processes overcome a longstanding limitation associated with all known

catalytic systems for asymmetric allylic amination - the ability to promote highly enantioselective aminations of branched allylic acetates bearing n-alkyl groups with complete levels of regioselectivitity.^{4-12,16} Another notable feature of these catalysts involves the ability to promote site-selective *N*-allylations of reactants that incorporate both primary and secondary aromatic amines. Mechanistic studies establish an outersphere mechanism for C-N bond formation. This work, along with our initially communicated studies,¹⁵ significantly expands the scope of catalytic asymmetric allylic amination methodology, broadening access to chiral α -stereogenic amines.

5.5 Experimental Details

General Information

All reactions were carried out under inert gas atmosphere (nitrogen or argon) unless otherwise indicated. Resealable pressure tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were flame dried followed by cooling in a desiccator or under a stream of inter gas prior to use. All commercial reagents and anhydrous solvents were used as received from vendors (Fischer Scientific, Sigma Aldrich and Combi Blocks) without further purification. The used Iridium catalyst (*S*)-Ir-II and (*R*)-Ir-II was prepared according to literature known procedures.¹ Cesium carbonate was used as received from Rockwell Lithium. Preparative column chromatography employing Silicycle silica gel (40-63 µm) was performed according to the method of Still² or on a Teledyne Isco Combiflash Rf utilizing Silicycle HP columns using a mobile phase composed of either heptane/isopropyl acetate, heptanes/ethyl acetate or dichloromethane/methanol. Reactions were monitored by a Shimadzu LCMS/UV system with LC-30AD solvent pump, 2020 MS, Sil-30AC autosampler, SPD-

M30A UV detector, CTO-20A column oven, using a 2-98% acetonitrile/0.1% formic acid (or 0.001% ammonia) gradient over 2.5 minutes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynamic Absorbents F). Visualization was accomplished with UV light followed by dipping in CAM, *p*-Anisaldehyde (PAA), or KMnO₄ stain solution followed by heating.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer using a diamond ATR unit. High-resolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M+H, M+Na), or a suitable fragment ion. Nuclear magnetic resonance (¹H, ¹³C, ¹⁹F NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (400 MHz) or Varian INOVA (500 MHz) spectrometer equipped with a Bruker cryoprobe. The chemical shifts are given as parts per million (ppm) and were referenced to the residual solvent signal (CDCl₃: $\delta_{\rm H} = 7.26$ ppm, $\delta_{\rm C} = 77.16$ ppm). Specific optical rotations were recorded on an Atago AP-300 automatic polarimeter at the sodium line (589 nm) in CHCl3. Solution concentrations are given in the units of 10–2 g mL⁻¹.

Experimental Details and Spectral Data

Synthesis of Allylic Acetates 5.1e-5.1g

General procedure for the synthesis of allylic acetates. The allylic acetates **5.1e**, **5.1f**, and **5.1g** were prepared by the Grignard reaction and acetylation of the as shown below. The allylic acetates **5.1b**,³ **5.1c**,⁴ and **5.1d**⁵ were identical in all respects to the reported materials.



To a round-bottomed flask charged with the corresponding aldehyde under an argon atmosphere was added THF (0.2 M). The reaction flask was placed an ice batch. After 10 minutes, vinyl magnesium bromide solution (120 mol%, 1.0 M in THF) was added slowly and the mixture was stirred at room temperature for 1 hour, at which point acetic anhydride (150 mol%) and triethylamine (200 mol%) were added and the reaction was stirred vigorously overnight. After water was added, the mixture was transferred to a separatory funnel. The organic layer was extracted with diethyl ether and the combined organic layers were washed with 1 N HCl, dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography give the corresponding allylic acetate over 2 steps. 1-cyclobutylallyl acetate (5.1e)



The title compound was prepared by the general procedure.

<u>**TLC**</u> (SiO₂) $R_f = 0.61$ (heptanes: isopropyl acetate = 4:1).

<u>**H NMR**</u> (500 MHz, CDCl₃): δ = 5.70 (ddd, J = 17.2, 10.5, 6.3 Hz, 1H), 5.25 – 5.13 (m, 3H), 2.58 – 2.45 (m, 1H), 2.07 (s, 3H), 1.97 (dddd, J = 18.3, 9.9, 8.0, 4.7 Hz, 2H), 1.90 – 1.74 (m, 4H).

¹³C NMR (125 MHz, CDCl₃): δ = 170.6, 134.6, 116.9, 77.9, 38.3, 24.3, 24.0, 21.2, 18.0.

<u>LRMS</u> (CI): Calculated for C_7H_{11} [M–OAc]⁺ = 95, Found 95.

FTIR (neat): 2941, 1737, 1370, 1232, 1102, 1018, 972, 925 cm⁻¹.



1-cyclopentylallyl acetate (5.1f)



The title compound was prepared by the general procedure.

<u>**TLC**</u> (SiO₂) $R_f = 0.61$ (heptanes: isopropyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 5.70 (ddd, J = 17.2, 10.5, 6.7 Hz, 1H), 5.16 (dt, J = 17.2, 1.4 Hz, 1H), 5.09 (dt, J = 10.6, 1.3 Hz, 1H), 5.07 – 5.02 (m, 1H), 2.07 – 1.94 (m, 4H), 1.70 – 1.41 (m, 6H), 1.30 – 1.14 (m, 2H).

<u>**13C NMR**</u> (125 MHz, CDCl₃): δ = 170.4, 135.9, 116.9, 78.4, 43.5, 28.8, 28.6, 25.5, 25.3, 21.3.

<u>LRMS</u> (CI): Calculated for C_8H_{13} [M–OAc]⁺ = 109, Found 109.

FTIR (neat): 2954, 2869, 1738, 1370, 1232, 1018, 929, 893 cm⁻¹.



(5S)-5,9-dimethyldeca-1,8-dien-3-yl acetate (5.1g)



The title compound was prepared by the general procedure.

<u>**TLC**</u> (SiO₂) $R_f = 0.46$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.76$ (dddd, J = 17.2, 13.3, 10.5, 6.5 Hz, 1H), 5.38 – 5.27 (m, 1H), 5.23 (ddt, J = 17.2, 8.7, 1.3 Hz, 1H), 5.14 (tt, J = 10.5, 1.2 Hz, 1H), 5.08 (tdd, J = 8.4, 4.9, 3.5 Hz, 1H), 2.05 (d, J = 3.9 Hz, 3H), 2.02-1.87 (m, 2H), 1.67 (s, 3H), 1.60 (d, J = 3.1 Hz, 3H), 1.55-1.45 (m, 2H), 1.40-1.10(m, 3H), 0.91 (t, J = 6.4 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.3$ (d, J = 14.9 Hz), 136.9 (d, J = 48.9 Hz), 131.3 (d, J = 2.2 Hz), 124.6 (d, J = 5.1 Hz), 116.4 (d, J = 76.0 Hz), 73.2 (d, J = 68.0 Hz), 41.4 (d, J = 28.2 Hz), 37.0 (d, J = 48.7 Hz), 28.8 (d, J = 18.5 Hz), 25.7, 25.3 (d, J = 8.3 Hz), 21.3 (d, J = 8.2 Hz), 19.6 (d, J = 29.8 Hz), 17.7.

<u>HRMS</u> (CI): Calculated for $C_{14}H_{24}O_2 [M-OAc]^+ = 165.1643$, Found 165.1635. **<u>FTIR</u>** (neat): 2965, 2917, 2367, 1741, 1338, 1372, 1237, 1020, 988, 929, 668 cm⁻¹. $[\alpha]_D^{28} = -3.56 (c \ 0.2, CHCl_3).$



<u>Procedures and Spectral Data for Synthesis of Allylic Amines 5.4a-5.4l, 5.5a-5.5l,</u> <u>5.6a-5.6g, 5.7a-5.7c</u>

Enantioselective iridium catalyzed allylic alkylation with amine nucleophiles-



General procedure

An pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (200 mol%), (*S*)-Ir-II (5 mol%). The tube was purged with argon for 5 minutes. DME (1.0 M) was added followed by the allylic acetate (100 mol%) and the amine (200 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at the indicated temperature and stirred for the indicated period of time. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography.

(S)-N-(but-3-en-2-yl)aniline (5.4a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (81.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 82% yield (53.1 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC**</u> (SiO₂) $R_f = 0.40$ (heptanes: isopropyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.20 - 7.09$ (m, 2H), 6.68 (tt, J = 7.4, 1.1 Hz, 1H), 6.63 - 6.57 (m, 2H), 5.83 (ddd, J = 17.2, 10.3, 5.6 Hz, 1H), 5.21 (dt, J = 17.2, 1.4 Hz, 1H), 5.08 (dt, J = 10.4, 1.4 Hz, 1H), 3.98 (dddd, J = 9.5, 6.6, 4.8, 3.3 Hz, 1H), 3.59 (s, 1H), 1.31 (d, J = 6.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 147.4, 141.3, 129.1, 117.3, 114.1, 113.4, 51.1, 21.6. <u>HRMS</u> (ESI): Calculated for C₁₀H₁₃N [M+H⁺] = 148.1126, Found 148.1122. <u>FTIR</u> (neat): 3404, 2975, 1601, 1504, 1317, 1254, 992, 919, 748, 692 cm⁻¹. [α]_D²⁸ = -3.9 (*c* 0.2, CHCl₃).

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:*i*-PrOH = 97.5:2.5, 1.00 mL/min, 254 nm), *ee* = 89%.





(S)-N-(but-3-en-2-yl)-2,3-dihydrobenzo[b][1,4]dioxin-6-amine (5.4b)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (133.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 85% yield (76.5 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>TLC (SiO2</u>) $R_f = 0.23$ (heptanes: isopropyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 6.68$ (dd, J = 8.5, 0.4 Hz, 1H), 6.17 (d, J = 2.7 Hz, 1H), 6.14 (dd, J = 8.5, 2.7 Hz, 1H), 5.81 (ddd, J = 17.2, 10.3, 5.7 Hz, 1H), 5.20 (dt, J = 17.3, 1.4 Hz, 1H), 5.07 (dt, J = 10.3, 1.4 Hz, 1H), 4.34 – 4.18 (m, 2H), 4.21 – 4.13 (m, 2H), 4.00 – 3.77 (m, 1H), 3.33 (br, 1H), 1.28 (d, J = 6.6 Hz, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 143.9, 142.4, 141.5, 135.5, 117.5, 114.1, 107.4, 102.3, 64.8, 64.2, 51.9, 21.7.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{15}NO_2[M+H^+] = 206.1176$, Found 206.1183.

FTIR (neat): 3394, 2973, 1507, 1207, 1068, 915, 884, 794, 740 cm⁻¹.

 $[\alpha]_{D}^{28} = -0.87 (c \ 0.2, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:i-PrOH = 97.5:2.5, 1.00 mL/min, 254 nm), ee = 90%.







(S)-N-(but-3-en-2-yl)-4-fluoroaniline (5.4c)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (97.8 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 83% yield (60.3 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC (SiO₂)</u>** $R_f = 0.36$ (heptanes: isopropyl acetate = 4:1).</u>

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 1H NMR (400 MHz, Chloroform-d) δ 6.91 – 6.81 (m, 2H), 6.58 – 6.49 (m, 2H), 5.81 (ddd, *J* = 17.2, 10.3, 5.7 Hz, 1H), 5.20 (dt, *J* = 17.2, 1.4 Hz, 1H), 5.09 (dt, *J* = 10.4, 1.3 Hz, 1H), 4.01 – 3.83 (m, 1H), 3.48 (br, 1H), 1.31 (d, *J* = 6.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 156.9, 154.5, 143.7, 141.2, 115.6, 115.4, 114.3, 114.2, 51.8, 21.7.

¹⁹**F NMR** (376 MHz, CDCl₃): δ = -128.3.

<u>HRMS</u> (ESI): Calculated for $C_{10}H_{12}NF[M+H^+] = 166.1027$, Found 166.1028.

<u>FTIR</u> (neat): 2975, 1505, 1309, 1213, 1155, 991, 918, 815, 770, 506 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +1.78 (*c* 0.2, CHCl₃).

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:i-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.






(S)-N-(but-3-en-2-yl)-3-methyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2yl)aniline (5.4d)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (205.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 40 hr). The title compound was obtained in 85% yield (98.6 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC (SiO₂**</u>) $R_f = 0.58$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.61 (d, *J* = 8.7 Hz, 1H), 6.40 (d, *J* = 7.0 Hz, 2H), 5.83 (ddd, *J* = 17.2, 10.4, 5.3 Hz, 1H), 5.20 (dt, *J* = 17.2, 1.4 Hz, 1H), 5.08 (dt, *J* = 10.4, 1.3 Hz, 1H), 4.04 (p, *J* = 6.5 Hz, 1H), 3.74 (br, 1H), 2.47 (s, 3H), 1.33 – 1.29 (m, 15H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): δ = 149.5, 146.7, 140.8, 137.8, 114.5, 114.1, 109.6, 82.7, 50.4, 24.9, 24.9, 22.5, 21.5.

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{26}NO_2[M+H^+] = 288.2132$, Found 288.2138.

<u>FTIR</u> (neat): 2978, 2362, 1602, 1349, 1215, 1146, 754 cm⁻¹.

 $[\alpha]_{D}^{28} = -1.0 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:i-PrOH = 97:3, 1.00 mL/min, 280 nm), ee = 89%.





(S)-N-(but-3-en-2-yl)-2-fluoro-4-methoxyaniline (5.4e)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (124.2 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 78% yield (67.1 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC (SiO₂**</u>) $R_f = 0.58$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 6.69 - 6.60$ (m, 2H), 6.56 (dd, J = 8.9, 2.7 Hz, 1H), 5.82 (ddd, J = 16.7, 10.3, 5.8 Hz, 1H), 5.20 (dd, J = 17.3, 1.4 Hz, 1H), 5.08 (dd, J = 10.4, 1.4 Hz, 1H), 3.91 (p, J = 6.5 Hz, 1H), 3.73 (s, 3H), 3.52 (br, 1H), 1.33 (d, J = 6.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 153.08, 151.5 (d, *J* = 9.9 Hz), 150.71, 141.28, 129.7 (d, *J* = 12.1 Hz), 114.4 (d, *J* = 4.6 Hz), 114.19, 109.1 (d, *J* = 3.3 Hz), 102.38 (d, *J* = 22.7 Hz), 55.8, 51.9, 21.7.

¹⁹**F** NMR (471 MHz, CDCl₃): $\delta = -133.0$ (dd, J = 12.9, 9.8 Hz, 1F).

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{14}FNO[M+H^+] = 196.1132$, Found 196.1129.

<u>FTIR</u> (neat): 2964, 1515, 1279, 1214, 1152, 1035, 923, 755 cm⁻¹.

 $[\alpha]_{\mathbf{p}}^{\mathbf{28}} = +6.3 \ (c \ 1.0, \ \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel AS-H column, hexanes:i-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 88%.





12.5 12.9 13.0

20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 -22 f1 (ppm)



(S)-N1-(but-3-en-2-yl)-N4,N4-dimethylbenzene-1,4-diamine (5.4f)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (120.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 89% yield (74.5 mg, 0.39 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-4:1).

<u>**TLC** (SiO</u>₂) $R_f = 0.34$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 6.74 - 6.68$ (m, 2H), 6.64 - 6.57 (m, 2H), 5.84 (ddd, J = 17.2, 10.3, 5.7 Hz, 1H), 5.20 (dt, J = 17.2, 1.4 Hz, 1H), 5.06 (dt, J = 10.4, 1.4 Hz, 1H), 3.90 (ttd, J = 6.6, 5.3, 1.3 Hz, 1H), 3.25 (br, 1H), 2.81 (s, 6H), 1.28 (d, J = 6.6 Hz, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 144.1, 142.0, 139.9, 115.7, 115.2, 113.9, 52.2, 42.2, 21.7.$

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{18}N_2$ [M+H⁺] = 191.1543, Found 191.1535.

<u>FTIR</u> (neat): 2979, 2361, 1515, 1216, 814, 753 cm⁻¹.

 $[\alpha]_{D}^{28} = -26.3 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 254 nm), ee = 90%.</u>





(S)-N-(but-3-en-2-yl)-1-methyl-1H-indazol-6-amine (5.4g)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (129.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 30 hr). The title compound was obtained in 81% yield (71.7 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 30% over 20 min).

<u>TLC (SiO</u>₂) $R_f = 0.38$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.76$ (d, J = 1.0 Hz, 1H), 7.43 (dd, J = 8.6, 0.6 Hz, 1H), 6.49 (dd, J = 8.7, 1.9 Hz, 1H), 6.33 (dt, J = 1.7, 0.8 Hz, 1H), 5.88 (ddd, J = 17.2, 10.3, 5.5 Hz, 1H), 5.27 (dt, J = 17.2, 1.4 Hz, 1H), 5.13 (dt, J = 10.4, 1.3 Hz, 1H), 4.07 (dtd, J = 8.1, 6.7, 5.2 Hz, 1H), 3.93 (s, 3H), 3.91 – 3.82 (m, 1H), 1.37 (d, J = 6.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 146.6, 141.8, 140.8, 132.6, 121.5, 117.0, 114.3, 112.4, 88.6, 51.3, 35.2, 21.6.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{15}N_3$ [M+H⁺] = 202.1339, Found 202.1339.

<u>FTIR</u> (neat): 3312, 2973, 1624, 1492, 1254, 1098, 982, 919, 732, 620 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -60.3 \ (c \ 0.2, \ \rm CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:*i*-PrOH = 90:10, 1.00 mL/min, 210 nm), *ee* = 87%.





(S)-N-(but-3-en-2-yl)-2-morpholinopyrimidin-5-amine (5.4h)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (158.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 82% yield (84.5 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 80% over 20 min).

<u>TLC (SiO</u>) $R_f = 0.37$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.92 (s, 2H), 5.76 (ddd, *J* = 17.2, 10.3, 6.2 Hz, 1H), 5.18 (dt, *J* = 17.2, 1.3 Hz, 1H), 5.11 (dt, *J* = 10.3, 1.2 Hz, 1H), 3.87 – 3.80 (m, 1H), 3.79 – 3.76 (m, 4H), 3.63 (dd, *J* = 5.7, 4.0 Hz, 4H), 1.31 (d, *J* = 6.6 Hz, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 157.2, 145.2, 140.6, 133.4, 115.1, 66.9, 52.8, 45.2, 21.7.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{18}N_4O[M+H^+] = 235.1553$, Found 235.1555.

<u>FTIR</u> (neat): 2968, 1480, 1444, 1264, 1116, 954, 731 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +9.92 (*c* 1.0, CHCl₃).

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:*i*-PrOH = 85:15, 1.00 mL/min, 230 nm), *ee* = 89%.





(S)-N-(but-3-en-2-yl)benzo[d]thiazol-7-amine (5.4i)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (132.2 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 30 hr). The title compound was obtained in 81% yield (72.8 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 30% over 20 min).

<u>TLC (SiO2</u>) $R_f = 0.52$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 8.92 (s, 1H), 7.57 (dd, *J* = 8.1, 0.9 Hz, 1H), 7.38 (t, *J* = 8.0 Hz, 1H), 6.67 (d, *J* = 7.9 Hz, 1H), 5.90 (ddd, *J* = 17.2, 10.4, 5.6 Hz, 1H), 5.26 (dt, *J* = 17.3, 1.4 Hz, 1H), 5.13 (dt, *J* = 10.3, 1.3 Hz, 1H), 4.20 (dtd, *J* = 8.2, 6.7, 1.5 Hz, 1H), 3.60 (d, *J* = 6.7 Hz, 1H), 1.42 (d, *J* = 6.6 Hz, 3H).

¹³**C** NMR (100 MHz, CDCl₃): δ = 154.4, 151.9, 142.0, 140.5, 127.3, 120.5, 114.6, 113.3, 107.1, 51.6, 21.7.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{12}N_2S[M+H^+] = 205.0794$, Found 205.0802.

<u>FTIR</u> (neat): 3292, 2974, 1575, 1472, 1287, 1145, 1049, 920, 774, 717 cm⁻¹.

 $[\alpha]_{D}^{28} = +31.68 \ (c \ 0.2, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:i-PrOH = 97.5:2.5, 1.00 mL/min, 254 nm), ee = 90%.





(S)-N-(but-3-en-2-yl)-3-(methylsulfonyl)aniline (5.4j)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (150.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 40 hr). The title compound was obtained in 76% yield (75.3 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-4:1).

<u>TLC (SiO</u>) $R_f = 0.41$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.30 (t, *J* = 7.9 Hz, 1H), 7.18 (ddd, *J* = 7.7, 1.7, 0.9 Hz, 1H), 7.10 (t, *J* = 2.1 Hz, 1H), 6.84 – 6.74 (m, 1H), 5.79 (ddd, *J* = 17.0, 10.3, 5.4 Hz, 1H), 5.22 (dd, *J* = 17.3, 1.3 Hz, 1H), 5.12 (dt, *J* = 10.3, 1.2 Hz, 1H), 4.13 – 3.90 (m, 2H), 3.01 (s, 3H), 1.34 (d, *J* = 6.3 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 148.0, 141.3, 140.0, 130.1, 118.0, 115.3, 114.8, 111.0, 51.0, 44.4, 21.5.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{15}NO_2S[M+H^+] = 226.0896$, Found 226.0900.

FTIR (neat): 3379, 1599, 1487, 1296, 1141, 961, 757, 683 cm⁻¹.

 $[\alpha]_{D}^{28} = -33.3 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Two connected chiralcel AD-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 91%.</u>





(S)-N-(but-3-en-2-yl)pyridin-2-amine (5.4k)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (82.8 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (48.3 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-4:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.07$ (ddd, J = 5.0, 2.0, 0.9 Hz, 1H), 7.40 (ddd, J = 8.8, 7.1, 1.9 Hz, 1H), 6.56 (ddd, J = 7.1, 5.0, 1.0 Hz, 1H), 6.36 (dt, J = 8.4, 1.0 Hz, 1H), 5.87 (ddd, J = 17.2, 10.4, 5.3 Hz, 1H), 5.21 (dt, J = 17.2, 1.4 Hz, 1H), 5.08 (dt, J = 10.4, 1.4 Hz, 1H), 4.48 (br, 1H), 4.32 – 4.20 (m, 1H), 1.33 (d, J = 6.7 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{21.3.}} (125 \text{ MHz}, \text{CDCl}_3): \delta = 158.1, 148.2, 140.7, 137.4, 113.9, 112.9, 106.9, 49.4, 21.3.$

<u>HRMS</u> (ESI): Calculated for $C_9H_{12}N_2[M+H^+] = 149.1073$, Found 149.1073.

<u>FTIR</u> (neat): 3528, 2974, 1599, 1445, 1330, 1154, 987, 920, 751 cm⁻¹. $[\alpha]_{\mathbf{p}}^{\mathbf{28}} = +6.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 90:10, 1.00 mL/min, 210 nm), ee = 91%.





(S)-N-(but-3-en-2-yl)pyridin-3-amine (5.4l)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (82.8 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 30 hr). The title compound was obtained in 87% yield (56.7 mg, 0.38 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 50% over 20 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.22$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.02$ (dd, J = 2.9, 0.7 Hz, 1H), 7.93 (dd, J = 4.7, 1.4 Hz, 1H), 7.05 (ddd, J = 8.3, 4.7, 0.7 Hz, 1H), 6.86 (ddd, J = 8.3, 2.9, 1.4 Hz, 1H), 5.80 (ddd, J = 17.2, 10.3, 5.6 Hz, 1H), 5.21 (dt, J = 17.2, 1.3 Hz, 1H), 5.11 (dt, J = 10.4, 1.3 Hz, 1H), 3.97 (s, 1H), 3.69 (br, 1H), 1.34 (d, J = 6.7 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{21.6}} (100 \text{ MHz}, \text{CDCl}_3): \delta = 143.3, 140.3, 138.7, 136.6, 123.6, 119.1, 114.7, 51.0, 21.6.$

HRMS (ESI): Calculated for C₉H₁₂N₂ [M+H⁺] = 149.1073, Found 149.1075. **FTIR** (neat): 3253, 2973, 1578, 1481, 1414, 1241, 991, 917, 791, 706 cm⁻¹. $[\alpha]_{D}^{28} = -12.7 (c \ 0.2, CHCl_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:i-PrOH = 90:10, 1.00 mL/min, 210 nm), ee = 91%.</u>





(S)-1-(but-3-en-2-yl)indoline (5.5a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (104.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 92% yield (70.1 mg, 0.40 mmol) as a light purple oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 10 min).

<u>TLC (SiO2</u>) $R_f = 0.49$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.10-7.02 (m, 2H), 6.63 (t, *J* = 7.3 Hz, 1H), 6.49 (d, *J* = 7.8 Hz, 1H), 5.94 (ddd, *J* = 17.5, 10.5, 5.3 Hz, 1H), 5.26-5.15 (m, 2H), 4.22 (qd, *J* = 6.9, 5.1 Hz, 1H), 3.44-3.30 (m, 2H), 2.96 (t, *J* =18.5 Hz, 2H), 1.32 (d, *J* = 6.8 Hz, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): δ = 151.1, 138.4, 130.3, 127.2, 124.4, 117.2, 115.7, 107.6,

52.6, 47.1, 28.2, 15.5.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{15}N[M+H^+] = 174.1277$, Found 174.1227.

<u>FTIR</u> (neat): 3047, 3024, 2973, 2933, 2845, 1606, 1487, 1473, 1458, 1257, 1185, 1023, 919, 743 cm⁻¹.

 $[\alpha]_{D}^{28}$ = -551.36 (*c* 0.2, CHCl₃).

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:i-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 92%.





(S)-1-(but-3-en-2-yl)-2,3-dihydro-1H-pyrrolo[2,3-c]pyridine (5.5b)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (105.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 78% yield (59.8 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 20% over 20min).

<u>**TLC**</u> (SiO₂) $R_f = 0.25$ (heptane: isopropyl acetate =1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.89$ (d, J = 4.6 Hz, 1H), 7.80 (d, J = 0.9 Hz, 1H), 6.97 (dq, J = 4.6, 1.0 Hz, 1H), 5.88 (ddd, J = 17.4, 10.5, 5.5 Hz, 1H), 5.21 (dt, J = 12.2, 1.4 Hz, 1H), 5.18 (dt, J = 5.3, 1.4 Hz, 1H), 4.18 (qdt, J = 7.0, 5.5, 1.6 Hz, 1H), 3.45-3.32 (m, 2H), 2.98-2.91 (m, 2H), 1.31 (d, J = 6.9 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 147.7, 139.4, 139.2, 137.5, 129.5, 119.7, 116.2, 52.8, 46.9, 28.0, 15.7.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{14}N_2$ [M+H⁺] =175.1230, Found 175.1226.

<u>FTIR</u> (neat): 3037, 2974, 2360, 2341, 1597, 1493, 1426, 129.4, 1183, 923, 819, 772, 669, 567 cm⁻¹.

 $[\alpha]_{D}^{28} = -44.4 \ (c \ 0.2, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:i-PrOH = 85:15, 1.00 mL/min, 254 nm), ee = 93%.




(S)-1-(but-3-en-2-yl)-2,3-dihydro-1*H*-pyrrolo[2,3-*b*]pyridine (5.5c)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (105.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (56.8 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 10 min).

<u>TLC</u> (SiO₂) $R_f = 0.25$ (heptane: isopropyl acetate =9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.85-7.83 (m, 1H), 7.13 (dq, *J* = 6.9, 1.4 Hz, 1H), 6.37 (dd, *J* = 7.0, 5.3 Hz, 1H), 5.91 (ddd, *J* = 17.4, 10.5, 4.9 Hz, 1H), 5.20-5.14 (m, 2H), 4.86 (qdt, *J* = 6.8, 4.9, 1.8 Hz, 1H), 3.46-3.41 (m, 2H), 2.95-2.91 (m, 2H), 1.28 (d, *J* = 6.9Hz, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 162.5, 145.8, 138.4, 138.4, 130.7, 123.2, 115.4, 111.9, 49.5, 43.6, 25.7, 15.5.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{14}N_2$ [M+H⁺] = 175.1230, Found 175.1225.

<u>FTIR</u> (neat): 3709, 2972, 2360, 2341, 1610, 1577, 1491, 1463, 1443, 1391, 771, 669cm⁻¹.

 $[\alpha]_{D}^{28} = -31.7 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Two connected chiralcel OD-3 & OD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 94%.</u>







(S)-1-(but-3-en-2-yl)-2',3',5',6'-tetrahydrospiro[indoline-3,4'-pyran] (5.5d)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (166.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 85% yield (88.9 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 10%).

<u>**TLC**</u> (SiO₂) $R_f = 0.33$ (heptane: isopropyl acetate =9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.11-7.03 (m, 2H), 6.67 (td, *J* = 7.4, 1.0 Hz, 1H), 6.49 (dd, *J* = 7.8, 0.9 Hz, 1H), 5.92 (ddd, *J* = 17.4, 10.5, 5.1 Hz, 1H), 5.22 (dt, *J* = 10.4, 1.5 Hz, 1H), 5.18 (dt, *J* = 3.3, 1.5 Hz, 1H), 4.23 (qdt, *J* = 6.8, 5.1, 1.7 Hz, 1H), 3.97 (ddd, *J* = 11.8, 4.5, 2.2 Hz, 2H), 3.57 (tdd, *J* = 12.0, 7.7, 2.3 Hz, 2H), 3.39-3.29 (m, 2H), 1.97 (dtd, *J* = 13.9, 12.2, 4.7 Hz, 2H), 1.69-1.59 (m, 2H), 1.32 (d, *J* = 6.9 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 150.0, 138.3, 137.3, 127.9, 122.4, 117.3, 115.7, 107.5, 65.2, 56.4, 52.1, 41.7, 36.7, 36.6, 36.5, 15.5.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{22}NO[M+H^+] = 244.1696$, Found 244.1699.

<u>FTIR</u> (neat): 2936, 2848, 2360, 2341, 1660, 1482, 1462, 1251, 1104, 1027, 837, 750, 668, 547, 464 cm⁻¹.

 $[\alpha]_{D}^{28} = -44.7 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Two connected chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.</u>





(*S*)-1-((*S*)-but-3-en-2-yl)-2-methylindoline (5.5e)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (117.2 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 80% yield, >20:1 dr (69.2 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Heptane).

<u>**TLC** (SiO₂</u>) $R_f = 0.50$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.03-6.96 (m, 2H), 6.63-6.54 (m, 2H), 6.07 (ddd, *J* = 17.5, 10.6, 4.2 Hz, 1H), 5.26 (ddd, *J* = 17.5, 2.1, 1.5 Hz, 1H), 5.20, (ddd, *J* = 10.6, 2.2, 1.5 Hz, 1H), 4.01 (qdt, *J* = 6.8, 4.3, 2.2 Hz, 1H), 3.79 (tq, *J* = 8.9, 6.1 Hz, 1H), 3.18 (dd, *J* = 15.6, 9.1 Hz, 1H), 2.61 (ddt, *J* = 15.7, 8.8, 1.2 Hz, 1H), 1.33 (d, *J* = 5.3 Hz, 3H), 1.31 (d, *J* = 4.4 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 150.5, 140.5, 129.3, 126.8, 124.2, 117.0, 114.8, 108.9, 57.7, 53.0, 37.6, 21.6, 14.1.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N[M+H^+] = 188.1434$, Found 188.1427.

<u>FTIR</u> (neat): 3726, 2966, 2360, 2341, 1605, 1481, 1459, 1257, 772, 747, 720, 669, 656, 419 cm⁻¹.

 $[\alpha]_{D}^{28} = +57.9 \ (c \ 0.2, \ CHCl_3).$



(*R*)-1-((*S*)-but-3-en-2-yl)-2-methylindoline (5.5f)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (117.2mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 48% yield, 5:1 dr (39.8 mg, 0.21 mmol) as a light orange oil after purification by flash column chromatography (4g SiO₂, Heptane).

<u>**TLC** (SiO₂</u>) $R_f = 0.50$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.07-6.98 (m, 2H), 6.66-6.54 (m, 1H), 6.48-6.46 (m, 1H), 6.05 (ddd, *J* = 17.3, 10.5, 5.3 Hz, 1H), 5.19 (dt, *J* = 17.4, 1.7 Hz, 1H), 5.14 (dt, *J* = 10.5, 1.6 Hz, 1H), 4.10-4.03 (m, 1H), 3.91-3.84 (m, 1H), 3.18 (ddt, *J* = 15.6, 9.2, 0.9 Hz, 1H), 2.60 (ddt, *J* = 15.6, 7.9, 1.1 Hz, 1H), 1.42 (d, *J* = 7.0 Hz, 3H), 1.28 (d, *J* = 6.1 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 150.8, 140.3, 128.9, 127.1, 124.3, 116.8, 114.7, 107.7, 57.0, 53.4, 37.6, 22.0, 16.4.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N[M+H^+]=188.1434$, Found 188.1428.

<u>FTIR</u> (neat): 3706, 3627, 2971, 2360, 2341, 1606, 1483, 1459, 1258, 746, 720, 669, 409 cm⁻¹.

 $[\alpha]_{D}^{28} = -24.6 \ (c \ 0.2, \ CHCl_3).$



(S)-N-(but-3-en-2-yl)-N-methylaniline (5.5g)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (94.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 71% yield (50.3 mg, 0.31 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Heptane).

<u>**TLC**</u> (SiO₂) $R_f = 0.47$ (heptane: isopropyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.25-7.19 (m, 2H), 6.82-6.77 (m, 2H), 6.70 (tt, *J* = 7.3, 1.0 Hz, 1H), 5.91 (ddd, *J* = 17.2, 10.8, 4.2 Hz, 1H), 5.17 (p, *J* = 1.4 Hz, 1H), 5.13 (ddd, *J* = 9.3, 2.0, 1.4 Hz, 1H), 4.48 (qdt, *J* = 6.6, 4.1, 2.0 Hz, 1H), 2.73 (s, 3H), 1.25 (d, *J* = 6.8 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 150.0, 139.6, 129.1, 116.7, 115.0, 133.3, 55.1, 31.4, 15.5.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{15}N[M+H^+] = 162.1277$, Found 162.1278.

<u>FTIR</u> (neat): 3061, 2973, 2812, 2360, 2341, 1597, 1501, 1308, 1137, 1035, 990, 917, 746, 690, 518cm⁻¹.

 $[\alpha]_{D}^{28} = -185.6 (c \ 0.2, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 91%.</u>





(S)-4-(but-3-en-2-yl(methyl)amino)benzonitrile (5.5h)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (116.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 79% yield (64.7 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 10 min).

<u>TLC (SiO2</u>) $R_f = 0.25$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.48-7.43$ (m, 2H), 6.74-6.69 (m, 2H), 5.86 (ddd, J = 17.3, 10.6, 4.0 Hz, 1H), 5.21 (ddd, J = 10.6, 2.0, 1.1 Hz, 1H), 5.14 (ddd, J = 17.4, 2.0, 1.1, 1H), 4.53 (qdt, J = 6.7, 4.1, 2.0 Hz, 1H), 2.81 (s, 3H), 1.32 (d, J = 6.8 Hz, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 152.3$, 137.9, 133.5, 120.6, 115.8, 112.0, 97.6, 54.5, 31.5, 16.1.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{14}N_2$ [M+H⁺] = 187.1230, Found 187.1231.

<u>FTIR</u> (neat): 3726, 3627, 2360, 2341, 2111, 1602, 1517, 1383, 1179, 1111, 922, 816, 669, 543 cm⁻¹.

 $[\alpha]_{D}^{28} = -207.8 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Two connected chiralcel OD-3 & OD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.</u>





(S)-N-(but-3-en-2-yl)-4-methoxy-N-methylaniline (5.5i)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the secondary amine (120.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 89% yield (74.9 mg, 0.39 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-10:1).

<u>TLC</u> (SiO₂) $R_f = 0.38$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.86 - 6.77$ (m, 4H), 5.92 (ddd, J = 17.2, 10.7, 4.6 Hz, 1H), 5.18 - 5.09 (m, 2H), 4.28 (dtdd, J = 8.5, 6.6, 3.9, 1.8 Hz, 1H), 3.77 (s, 3H), 2.67 (s, 3H), 1.21 (d, J = 6.7 Hz, 3H).

<u>¹³C NMR</u> (100 MHz, CDCl₃): δ = 152.1, 144.7, 139.8, 116.4, 115.0, 114.5, 57.1, 55.7, 32.5, 15.2.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{17}NO[M+H^+] = 192.1383$, Found 192.1381.

<u>FTIR</u> (neat): 2975, 1509, 1464, 1242, 1110, 1039, 919, 815, 754 cm⁻¹.

 $[\alpha]_{D}^{28} = -112.0 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 93%.





(S)-3-bromo-N-(but-3-en-2-yl)-N-methylaniline (5.5j)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the secondary amine (163.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr). The title compound was obtained in 64% yield (67.6 mg, 0.28 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1–20:1).

<u>TLC (SiO</u>) $R_f = 0.59$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.06 (t, *J* = 8.1 Hz, 1H), 6.89 (t, *J* = 2.2 Hz, 1H), 6.81 (dd, *J* = 7.6, 1.7 Hz, 1H), 6.69 (dd, *J* = 8.5, 2.5 Hz, 1H), 5.88 (ddd, *J* = 17.4, 10.6, 4.1 Hz, 1H), 5.20 – 5.11 (m, 2H), 4.44 (qdt, *J* = 6.5, 4.0, 2.0 Hz, 1H), 2.72 (s, 3H), 1.27 (d, *J* = 6.8 Hz, 3H).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 151.1, 138.8, 130.2, 123.4, 119.2, 115.7, 115.3, 111.5, 54.9, 31.4, 15.7.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{14}BrN[M+H^+] = 240.0382$, Found 240.0382.

<u>FTIR</u> (neat): 1590, 1554, 1487, 1215, 1114, 981, 925, 752, 681 cm⁻¹.

 $[\alpha]_{D}^{28} = -95.8 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.







(S)-N-(but-3-en-2-yl)-3,4-dichloro-N-methylaniline (5.5k)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the secondary amine (154.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr). The title compound was obtained in 76% yield (76.9 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1-20:1).

<u>TLC</u> (SiO₂) $R_f = 0.59$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.22 (d, *J* = 9.0 Hz, 1H), 6.81 (d, *J* = 2.9 Hz, 1H), 6.60 (dd, *J* = 9.0, 3.0 Hz, 1H), 5.86 (ddd, *J* = 17.4, 10.6, 4.1 Hz, 1H), 5.18 (ddd, *J* = 10.6, 2.0, 1.2 Hz, 1H), 5.13 (ddd, *J* = 17.4, 2.0, 1.2 Hz, 1H), 4.39 (qdt, *J* = 6.5, 4.1, 2.1 Hz, 1H), 2.71 (s, 3H), 1.26 (d, *J* = 6.8 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 149.3, 138.4, 132.7, 130.3, 118.9, 115.5, 114.3, 112.5, 55.2, 31.5, 15.7.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{13}Cl_2N[M+H^+] = 230.0498$, Found 230.0496.

<u>FTIR</u> (neat): 2976, 1593, 1487, 1371, 1214, 1112, 997, 924 cm⁻¹.

 $[\alpha]_{D}^{28} = -100.0 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Two connected chiralcel OD-3 & OD-H column, hexanes:*i*-PrOH = 99.5:0.5, 1.00 mL/min, 254 nm), ee = 91%.</u>





(S)-N-(but-3-en-2-yl)-N,1-dimethyl-1H-benzo[d]imidazole-2-amine (5.5l)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the primary amine (141.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 94% yield (89.1 mg, 0.41 mmol) as a light orange oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 20%).

<u>**TLC** (SiO₂</u>) $R_f = 0.25$ (heptane: isopropyl acetate = 8:2).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.61-7.58 (m, 1H), 7.20-7.14 (m, 4H), 6.02 (ddd, *J* = 17.4, 10.5, 4.9 Hz, 1H), 5.29-5.25 (m, 1H), 5.25-5.23 (m, 1H), 4.16 (qdt, *J* = 6.8, 4.9, 1.8 Hz, 1H), 3.61 (d, 3H), 2.84 (s, 3H), 1.32 (d, *J* = 6.8 Hz, 3H).

<u>13</u>C NMR (100 MHz, CDCl₃): δ = 141.7, 138.6, 135.8, 121.5, 120.8, 117.8, 116.2, 108.3, 58.6, 33.1, 30.8, 15.7.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}N_3$ [M+H⁺] = 216.1495, Found 216.1498.

<u>FTIR</u> (neat): 2972, 2360, 2341, 1615, 1594, 1524, 1463, 1390, 1285, 1116, 922, 800, 741, 669 cm⁻¹.

 $[\alpha]_{D}^{28} = -129.9 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:*i*-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 87%.</u>







(S)-N-(but-3-en-2-yl)-N-methylbenzo[d]oxazol-2-amine (5.6a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 94% yield (83.7 mg, 0.41 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, heptanes: ethyl acetate = 20:1-10:1).

<u>TLC (SiO</u>₂) $R_f = 0.63$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.35 (d, *J* = 7.8 Hz, 1H), 7.25 (d, *J* = 7.7 Hz, 1H), 7.15 (t, *J* = 7.7 Hz, 1H), 7.03 – 6.96 (m, 1H), 5.91 (ddd, *J* = 17.3, 10.7, 4.5 Hz, 1H), 5.28 – 5.20 (m, 2H), 5.04 (dqd, *J* = 8.8, 6.4, 5.6, 2.0 Hz, 1H), 3.02 (s, 3H), 1.38 (d, *J* = 6.9 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 162.8, 148.8, 143.4, 137.2, 123.9, 120.2, 116.4, 116.0, 108.6, 54.1, 29.5, 15.9.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{14}N_2O[M+H^+] = 203.1179$, Found 203.1180.

<u>FTIR</u> (neat): 2975, 1632, 1575, 1459, 1424, 1246, 1127, 1001, 925, 793, 740 cm⁻¹.

 $[\alpha]_{D}^{28} = -94.5 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 91%.</u>





(S)-N-(5-(benzyloxy)pent-1-en-3-yl)-N-methylbenzo[d]oxazol-2-amine (5.6b)



Procedures

The allylic acetate (103.1 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 93% yield (131.9 mg, 0.41 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO</u>₂) $R_f = 0.44$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.43 - 7.37$ (m, 1H), 7.30 - 7.23 (m, 6H), 7.19 (td, J = 7.7, 1.1 Hz, 1H), 7.03 (td, J = 7.7, 1.2 Hz, 1H), 5.92 (ddd, J = 17.6, 10.4, 5.2 Hz, 1H), 5.30 - 5.27 (m, 1H), 5.25 (dt, J = 3.5, 1.3 Hz, 1H), 5.07 (ddd, J = 13.2, 7.0, 3.6 Hz, 1H), 4.44 (d, J = 2.8 Hz, 2H), 3.58 (dt, J = 9.5, 5.8 Hz, 1H), 3.52 (dt, J = 9.4, 6.7 Hz, 1H), 3.06 (s, 3H), 2.15 - 2.00 (m, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{125} \text{ MHz}, \text{ CDCl}_3): \delta = 163.0, 148.8, 143.4, 138.1, 135.8, 128.3, 127.7, 127.6, 123.9, 120.3, 117.0, 116.1, 108.7, 73.3, 66.9, 56.4, 31.0, 30.4.}$

<u>HRMS</u> (ESI): Calculated for $C_{20}H_{22}N_2O_2[M+H^+] = 323.1754$, Found 323.1757.

<u>FTIR</u> (neat): 2857, 1631, 1575, 1459, 1414, 1284, 1246, 1098, 1002, 907, 736, 697 cm⁻¹. $[\alpha]_{\mathbf{p}}^{\mathbf{28}} = -78.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel AS-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 92%.




(S)-N-methyl-N-(5-phenylpent-1-en-3-yl)benzo[d]oxazol-2-amine (5.6c)



Procedures

The allylic acetate (89.8 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 92% yield (118.3 mg, 0.40 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC (SiO**</u>₂) $R_f = 0.44$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.40 (dd, *J* = 7.8, 1.2 Hz, 1H), 7.30 – 7.25 (m, 3H), 7.24 – 7.14 (m, 4H), 7.04 (td, *J* = 7.8, 1.2 Hz, 1H), 5.91 (ddd, *J* = 17.6, 10.3, 5.2 Hz, 1H), 5.29 (t, *J* = 1.4 Hz, 1H), 5.26 (dt, *J* = 5.2, 1.3 Hz, 1H), 4.94 – 4.85 (m, 1H), 3.08 (s, 3H), 2.70 (t, *J* = 8.0 Hz, 2H), 2.15 – 2.00 (m, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{125} \text{ MHz}, \text{ CDCl}_3): \delta = 163.1, 148.8, 143.4, 141.2, 135.8, 128.5, 128.3, 126.1, 123.9, 120.3, 117.1, 116.1, 108.7, 58.7, 32.8, 32.5, 29.8.$

<u>HRMS</u> (ESI): Calculated for $C_{19}H_{20}N_2O[M+H^+] = 293.1648$, Found 293.1656.

<u>FTIR</u> (neat): 2941, 1630, 1574, 1496, 1458, 1245, 1125, 1000, 926, 738, 698 cm⁻¹.

 $[\alpha]_{D}^{28} = -29.5 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.</u>





(R)-N-(1-cyclopropylallyl)-N-methylbenzo[d]oxazol-2-amine (5.6d)



Procedures

The allylic acetate (61.7 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 81% yield (81.3 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1-15:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.56$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.34$ (dd, J = 7.8, 1.3 Hz, 1H), 7.23 (dd, J = 7.9, 1.1 Hz, 1H), 7.14 (td, J = 7.6, 1.2 Hz, 1H), 6.99 (td, J = 7.8, 1.3 Hz, 1H), 5.93 (ddd, J = 17.4, 10.5, 4.4 Hz, 1H), 5.36 (dt, J = 17.3, 1.6 Hz, 1H), 5.26 (dt, J = 10.5, 1.6 Hz, 1H), 4.16 – 4.04 (m, 1H), 3.16 (s, 3H), 1.14 (dtt, J = 9.6, 8.0, 4.9 Hz, 1H), 0.74 (dddd, J = 8.0, 6.7, 4.7, 3.3 Hz, 1H), 0.54 (tdd, J = 10.2, 4.3, 3.2 Hz, 1H), 0.48 – 0.35 (m, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{108.6, 64.2, 30.6, 12.2, 5.0, 3.0.} = 163.0, 148.7, 143.5, 135.5, 123.9, 120.1, 116.9, 115.9, 108.6, 64.2, 30.6, 12.2, 5.0, 3.0.$

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{16}N_2O[M+H^+] = 229.1335$, Found 229.1337.

<u>FTIR</u> (neat): 3007, 1629, 1573, 1458, 1423, 1245, 1124, 992, 908, 815, 738 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -35.3 \ (c \ 1.0, \ {\rm CHCl}_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.</u>







(R)-N-(1-cyclobutylallyl)-N-methylbenzo[d]oxazol-2-amine (5.6e)



Procedures

The allylic acetate (67.9 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 48 hr). The title compound was obtained in 71% yield (75.7 mg, 0.31 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.44$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.39 - 7.32$ (m, 1H), 7.25 (d, J = 5.9 Hz, 1H), 7.15 (td, J = 7.7, 1.2 Hz, 1H), 7.00 (td, J = 7.7, 1.3 Hz, 1H), 5.79 (ddd, J = 17.4, 10.7, 5.2 Hz, 1H), 5.26 - 5.14 (m, 2H), 4.82 - 4.69 (m, 1H), 2.99 (s, 3H), 2.71 (dq, J = 11.0, 7.7 Hz, 1H), 2.15 (dt, J = 12.6, 7.7 Hz, 1H), 2.05 - 1.77 (m, 5H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{108.6, 64.4, 35.9, 30.1, 26.3, 25.2, 17.7.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 163.3, 148.7, 143.5, 133.6, 123.9, 120.1, 117.0, 116.0, 108.6, 64.4, 35.9, 30.1, 26.3, 25.2, 17.7.$

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{18}N_2O[M+H^+] = 243.1492$, Found 243.1495.

<u>FTIR</u> (neat): 2938, 1629, 1573, 1458, 1416, 1245, 1122, 990, 917, 821, 738 cm⁻¹. $[\alpha]_D^{28} = -98.8 \ (c \ 1.0, CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 90%.</u>





(R)-N-(1-cyclopentylallyl)-N-methylbenzo[d]oxazol-2-amine (5.6f)



Procedures

The allylic acetate (74.0 mg, 0.44 mmol, 100 mol%) and the secondary amine (130.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 60 hr). The title compound was obtained in 81% yield (91.3 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.44$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.35 (dd, *J* = 7.8, 1.1 Hz, 1H), 7.25 (d, *J* = 7.7 Hz, 1H), 7.14 (td, *J* = 7.7, 1.2 Hz, 1H), 6.99 (td, *J* = 7.7, 1.2 Hz, 1H), 5.89 (ddd, *J* = 17.0, 10.5, 6.0 Hz, 1H), 5.30 – 5.18 (m, 2H), 4.56 – 4.45 (m, 1H), 3.06 (s, 3H), 2.25 (dp, *J* = 10.9, 8.0 Hz, 1H), 1.82 (dp, *J* = 12.5, 4.5, 3.8 Hz, 1H), 1.74 – 1.50 (m, 5H), 1.33 (dddd, *J* = 16.2, 10.9, 7.9, 2.4 Hz, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{108.6, 64.6, 41.0, 30.5, 30.3, 30.0, 25.6, 25.3.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 163.2, 148.7, 143.6, 135.1, 123.8, 120.1, 117.3, 116.0, 108.6, 64.6, 41.0, 30.5, 30.3, 30.0, 25.6, 25.3.$

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}N_2O[M+H^+] = 257.1648$, Found 257.1650.

<u>FTIR</u> (neat): 2951, 1629, 1573, 1458, 1245, 1123, 990, 921, 819, 738 cm⁻¹.

 $[\alpha]_{D}^{28} = -92.0 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 85%.</u>





N-((3S,5S)-5,9-dimethyldeca-1,8-dien-3-yl)-N-methylbenzo[d]oxazol-2-amine (5.6g)



Procedures

The allylic acetate (98.7 mg, 0.44 mmol, 100 mol%) and the primary amine (130.4 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr). The title compound was obtained in 80% yield, 10:1 dr (110.2 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 10 min).

<u>TLC (SiO2</u>) $R_f = 0.33$ (heptane: isopropyl acetate = 9:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.35$ (ddd, J = 7.8, 1.3, 0.6 Hz, 1H), 7.24 (ddd, J = 7.9, 1.2, 0.6 Hz, 1H), 7.14 (td, J = 7.7, 1.1 Hz, 1H), 6.99 (td, J = 7.7, 1.2 Hz, 1H), 5.85 (ddd, J = 172., 10.6, 5.4 Hz, 1H), 5.24 (dt, J = 7.5, 1.3 Hz, 1H), 5.20 (d, J = 1.5 Hz, 1H), 4.99 (ddq, J = 8.5, 5.6, 1.4 Hz, 1H), 4.93 (dddt, J = 8.4, 6.9, 5.4, 1.6 Hz, 1H), 3.02 (s, 3H), 2.03-1.93 (m, 1H), 1.89 (dt, J = 14.7, 7.1 Hz, 1H), 1.65-1.57 (m, 2H), 1.56 (d, J = 1.4 Hz, 3H), 1.52 (d, J = 1.3 Hz, 3H), 1.50-1.23 (m, 2H), 1.21-1.11 (m, 1H), 0.96 (d, J = 6.6 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{11}} (100 \text{ MHz, CDCl}_3): \delta = 163.0, 148.8, 143.6, 136.2, 131.4, 124.4, 123.8, 120.1, 116.8, 116.1, 108.6, 56.9, 38.5, 36.3, 29.9, 29.0, 25.5, 25.3, 20.0, 17.6.$

<u>HRMS</u> (ESI): Calculated for $C_{20}H_{28}N_2O[M+H^+] = 313.2274$, Found 313.2276.

<u>FTIR</u> (neat): 3726, 2961, 2914, 2360, 2341, 1632, 1575, 1459, 1246, 1125, 922, 754, 739, 669, 429 cm⁻¹.



N-((3*R*,5*S*)-5,9-dimethyldeca-1,8-dien-3-yl)-*N*-methylbenzo[*d*]oxazol-2-amine (iso-5.6g)



Procedures

The allylic acetate (98.7 mg, 0.44 mmol, 100 mol%) and the primary amine (130.4 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr) with (*R*)-Ir-II. The title compound was obtained in 76% yield, 20:1 dr (104.4 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 10 min).

<u>TLC (SiO</u>) $R_f = 0.33$ (heptane: isopropyl acetate = 9:1).

¹**H NMR** (400 MHz, CDCl₃): $\delta = 7.36$ (ddd, J = 7.8, 1.3, 0.6 Hz, 1H), 7.26-7.23 (m, 1H), 7.15 (td, J = 7.7, 1.2 Hz, 1H), 7.02-6.96 (m, 1H), 5.85 (ddd, J = 17.5, 10.4, 5.0 Hz, 1H), 5.22 (ddd, J = 5.3, 1.7, 1.2 Hz, 1H), 5.19 (td, J = 1.8, 1.2 Hz, 1H), 5.07 (ddq, J = 8.5, 5.7, 1.3 Hz, 1H), 4.97 (dtt, J = 11.1, 4.5, 1.7 Hz, 1H), 3.00 (s, 2H), 1.98 (p, J = 7.3 Hz, 2H), 1.83 (ddd, J = 13.6, 10.9, 2.9 Hz, 1H), 1.66 (d, J = 1.3 Hz, 3H), 1.60-1.56 (m, 3H), 1.47-1.24(m, 4H), 0.96 (d, J = 6.2 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{16.3}, 116.0, 108.6, 56.5, 37.9, 37.7, 29.6, 29.0, 25.7, 25.5, 19.4, 17.7.}$

<u>HRMS</u> (ESI): Calculated for $C_{20}H_{28}N_2O[M+H^+] = 313.2274$, Found 313.2278.

<u>FTIR</u> (neat): 2915, 2360, 2341, 1634, 1577, 1460, 1577, 1460, 1247, 924, 754, 740, 669, 650 cm⁻¹.



(S)-N-(but-3-en-2-yl)-1H-indol-5-amine (5.7a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the amine (116.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 91% yield (74.5 mg, 0.40 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 40% over 20 min).

<u>TLC</u> (SiO₂) $R_f = 0.32$ (heptanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.91$ (br, 1H), 7.19 (dt, J = 8.6, 0.8 Hz, 1H), 7.12 – 7.08 (m, 1H), 6.86 (d, J = 2.3 Hz, 1H), 6.63 (ddd, J = 8.7, 2.3, 0.4 Hz, 1H), 6.39 (ddd, J = 3.1, 2.0, 0.9 Hz, 1H), 5.90 (ddd, J = 17.3, 10.4, 5.7 Hz, 1H), 5.25 (dt, J = 17.2, 1.4 Hz, 1H), 5.08 (dt, J = 10.3, 1.4 Hz, 1H), 4.07 – 3.97 (m, 1H), 1.34 (d, J = 6.6 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 142.1$, 141.4, 130.1, 128.8, 124.4, 113.9, 112.8, 111.5, 103.7, 101.8, 52.5, 21.7.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{15}N_2$ [M+H⁺] = 187.1230, Found 187.1230.

<u>FTIR</u> (neat): 3404, 2975, 1626, 1581, 1469, 1231, 1167, 919, 797, 724, 602 cm⁻¹.

 $[\alpha]_{D}^{28} = -5.0 (c \ 0.2, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:*i*-PrOH = 80:20, 1.00 mL/min, 210 nm), ee = 91%.





(S)-8-bromo-N-(but-3-en-2-yl)-2,3,4,5-tetrahydrobenzo[b][1,4]oxazepin-6-amine (5.7b)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the amine (213.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 61% yield (79.8 mg, 0.27 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-5:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.24$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.53$ (d, J = 2.2 Hz, 1H), 6.46 (d, J = 2.2 Hz, 1H), 5.81 (ddd, J = 17.2, 10.3, 5.6 Hz, 1H), 5.17 (dt, J = 17.2, 1.3 Hz, 1H), 5.09 (dt, J = 10.3, 1.3 Hz, 1H), 4.50 (d, J = 6.3 Hz, 1H), 4.13 (t, J = 5.5 Hz, 2H), 3.85 (q, J = 6.3 Hz, 1H), 3.25 – 3.08 (m, 2H), 2.60 (br, 1H), 2.01 (ddd, J = 10.0, 6.5, 4.8 Hz, 2H), 1.31 (d, J = 6.6 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 156.5, 144.5, 140.8, 126.4, 116.5, 114.3, 113.5, 110.6, 71.9, 51.5, 45.6, 33.5, 21.6.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}BrN_2O[M+H^+] = 297.0597$, Found 297.0581.

<u>FTIR</u> (neat): 3314, 2918, 1576, 1498, 1422, 1377, 1256, 1214, 1081, 955, 918, 814, 684 cm⁻¹.

 $[\alpha]_{D}^{28} = -1.8 (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 90:10, 1.00 mL/min, 254 nm), *ee* = 90%.







(S)-N1-(but-3-en-2-yl)-N2-cyclohexylbenzene-1,2-diamine (5.7c)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the amine (167.4 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 86% yield (92.5 mg, 0.38 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 5% over 20 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.46$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.80 - 6.65$ (m, 4H), 6.01 - 5.76 (m, 1H), 5.21 (dd, J = 17.2, 1.7 Hz, 1H), 5.08 (dd, J = 10.4, 1.6 Hz, 1H), 3.93 (p, J = 6.5 Hz, 1H), 3.38 - 3.11 (m, 3H), 2.12 - 1.99 (m, 2H), 1.84 - 1.72 (m, 2H), 1.66 (dt, J = 13.1, 3.9 Hz, 1H), 1.39 (dd, J = 14.7, 11.5 Hz, 2H), 1.34 (d, J = 6.6 Hz, 3H), 1.31 - 1.14 (m, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 141.8, 136.7, 136.5, 119.2, 118.8, 114.0, 113.9, 113.4, 52.1, 51.5, 33.7 (d), 26.10, 25.0 (d), 21.87.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{24}N_2$ [M+H⁺] = 245.2012, Found 245.2016.

<u>FTIR</u> (neat): 3330, 2926, 1598, 1508, 1448, 1304, 1252, 1148, 916, 733 cm⁻¹.

 $[\alpha]_{D}^{28} = +30.0 \ (c \ 0.1, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, heptanes:*i*-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 92%.</u>





Procedures and Spectral Data for the deuterium labelling experiments

Preparation of (S,Z)-oct-1-en-3-yl-1-d acetate (5.1h). **5.1h** synthesized from commercially available (*S*)-(-)-1-Octyn-3-ol, >98% ee.

Procedures

To a round-bottomed flask charged with potassium carbonate (380.1 mg, 2.8 mmol, 110 mol%) was added deuterium oxide (6.8 mL, 0.37 M, 99.9 atom % D), followed by (*S*)-(-)-1-Octyn-3-ol (315 mg, 2.5 mmol, 100 mol%). The mixture was stirred at room temperature overnight. After anhydrous CH_2Cl_2 (10 mL) were added to the reaction mixture, the mixture was transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (5 mL × 2) and the combined organic layers were washed with H_2O (20 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

To a round-bottomed flask charged with the crude substrate under an argon atmosphere was added anhydrous CH_2Cl_2 (10 mL, 0.25 M with respect to propargylic alcohol). The reaction vessel was placed in an ice batch. After 5 minutes DIBAL (3 mL, 3.0 mmol, 1 M in hexanes) was added slowly and the solution was stirred at room temperature for 30 minutes. After the reaction vessel was placed in an ice batch, Schwartz's reagent (773.6 mg, 3.0 mmol, 120 mol%) was added to the reaction mixture. The mixture was stirred at room temperature for 2 hours, at which point saturated aqueous sodium bicarbonate (5 mL) were added and the reaction was stirred vigorously. After 2 hours, the reaction mixture was filtered (celite) with the aid of CH_2Cl_2 (10 mL) and the filtrate was

transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (10 mL × 2) and the combined organic layers were washed with H_2O (20 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

То round-bottomed flask charged with the crude substrate 4a and dimethylaminopyridine (15.2 mg, 0.13 mmol, 5 mol%) under an argon atmosphere was added CH₂Cl₂ (12.5 mL, 0.25 M with respect to propargylic alcohol), followed by acetic anhydride (0.28 mL, 3.0 mmol, 120 mol%) and triethylamine (0.41 mL, 3.0 mmol, 120 mol%). After 1 hour, saturated aqueous sodium bicarbonate (10 mL) was added and the mixture was transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (10 mL \times 2) and the combined organic layers were washed with 1 N HCl (10 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1) to furnish the title compound as a light yellow oil (240 mg, 1.40 mmol) in 56% yield over 3 steps.

<u>**TLC** (SiO</u>₂) $R_f = 0.72$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 5.76 (ddt, J = 10.4, 6.3, 2.6 Hz, 1H), 5.25 – 5.19 (m, 1H), 5.14 (dd, J = 10.5, 1.1 Hz, 1H), 2.06 (s, 3H), 1.66 – 1.52 (m, 2H), 1.34 – 1.25 (m, 6H), 0.91 – 0.84 (m, 3H).

²**H NMR** (92 MHz, CHCl₃): δ = 5.28 (s, 1D).

 $\frac{^{13}C \text{ NMR}}{^{21.3}, 14.0.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 170.4, 136.6, 116.2 (t), 74.9, 34.2, 31.6, 24.7, 22.5, 21.3, 14.0.$

LRMS (CI): Calculated for $C_8H_{14}D [M-OAc]^+ = 112$, Found 112.

<u>FTIR</u> (neat): 2932, 1731, 1372, 1247, 1021, 756, 667 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -74.5(c \ 1.0, \text{CHCl}_3).$





(S,Z)-N-methyl-N-(oct-1-en-3-yl-1-d)benzo[d]oxazol-2-amine (5.8a)



Procedures

An pressure tube equipped with a magnetic stir bar was charged with the amine **5.3m** (59.3 mg, 0.4 mmol, 200 mol%), cesium carbonate (130.3 mg, 0.4 mmol, 200 mol%) and (*S*)-Ir-II (11.1 mg, 0.01 mmol, 5 mol%). The tube was purged with argon for 5 minutes. DME (0.2 mL, 1.0 M) was added followed by the deuterated allylic acetate **5.1h** (34.3 mg, 0.2 mmol, 100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 70 °C for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-15:1). The title compound was obtained in 89% yield (46.2 mg, 0.18 mmol) as a colorless oil.

<u>TLC (SiO</u>₂) R_f

= 0.55 (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.35 (dd, J = 7.8, 1.1 Hz, 1H), 7.25 (d, J = 8.0 Hz, 1H), 7.15 (td, J = 7.7, 1.1 Hz, 1H), 6.99 (td, J = 7.7, 1.2 Hz, 1H), 5.85 (ddd, J = 10.5, 5.1, 2.4 Hz, 1H), 5.20 (dd, J = 10.6, 1.6 Hz, 1H), 4.86 – 4.77 (m, 1H), 3.02 (s, 3H), 1.74 – 1.66 (m, 2H), 1.36 – 1.27 (m, 6H), 0.86 (t, J = 6.9 Hz, 3H).

²**H NMR** (92 MHz, CHCl₃): δ = 5.28 (s, 1D).

 $\frac{^{13}C \text{ NMR}}{^{11}6.0, 108.6, 58.9, 31.6, 30.9, 29.7, 25.8, 22.5, 14.0.}$

<u>**HRMS**</u> (ESI): Calculated for $C_{16}H_{21}DN_2O[M+H^+] = 260.1868$, Found 260.1870.

<u>FTIR</u> (neat): 2930, 1636, 1577, 1460, 1246, 1216, 748 cm⁻¹.

 $[\alpha]_{D}^{28} = -71.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, heptanes:*i*-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 87%.</u>







(*R*,*E*)-*N*-methyl-*N*-(oct-1-en-3-yl-1-d)benzo[d]oxazol-2-amine (5.8b)



Procedures

An pressure tube equipped with a magnetic stir bar was charged with the amine **5.3m** (59.3 mg, 0.4 mmol, 200 mol%), cesium carbonate (130.3 mg, 0.4 mmol, 200 mol%) and (**R**)-Ir-II (11.1 mg, 0.01 mmol, 5 mol%). The tube was purged with argon for 5 minutes. DME (0.2 mL, 1.0 M) was added followed by the deuterated allylic acetate **5.1h** (34.3 mg, 0.2 mmol, 100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 70 °C for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-15:1). The title compound was obtained in 85% yield (44.1 mg, 0.17 mmol) as a colorless oil.

<u>**TLC** (SiO</u>₂) $R_f = 0.55$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.35 (dd, J = 7.8, 1.1 Hz, 1H), 7.26 – 7.22 (m, 1H), 7.15 (td, J = 7.7, 1.1 Hz, 1H), 6.99 (td, J = 7.8, 1.2 Hz, 1H), 5.86 (dd, J = 17.3, 5.3 Hz, 1H), 5.21 (dd, J = 17.4, 1.7 Hz, 1H), 4.82 (dddd, J = 8.6, 6.8, 5.3, 1.7 Hz, 1H), 3.02 (s, 3H), 1.76 – 1.63 (m, 2H), 1.41 – 1.21 (m, 6H), 0.87 (t, J = 7.0 Hz, 3H).

²H NMR (92 MHz, CHCl₃): δ = 5.28 (s, 1D).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{16.4}, 116.2, 116.0, 108.6, 58.9, 31.6, 30.9, 29.7, 25.8, 22.5, 14.0.}$

<u>**HRMS**</u> (ESI): Calculated for $C_{16}H_{21}DN_2O[M+H^+] = 260.1868$, Found 260.1872.

<u>FTIR</u> (neat): 2930, 1636, 1577, 1460, 1246, 1216, 753 cm⁻¹.

 $[\alpha]_{D}^{28} = +70.8 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, heptanes:*i*-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 96%.</u>






Single Crystal Diffraction Data for 5.7b

Empirical formula	C13 H18 Br Cl N2 O	
Formula weight	333.65	
Temperature	100(2) K	
Wavelength	1.54184 Å	
Crystal system	monoclinic	
Space group	P 21	
Unit cell dimensions	a = 7.1967(3) Å	α= 90°.
	b = 19.2597(4) Å	$\beta = 105.491(5)^{\circ}.$
	c = 10.4450(4) Å	$\gamma = 90^{\circ}.$
Volume	1395.15(9) Å ³	
Z	4	
Density (calculated)	1.588 Mg/m ³	
Absorption coefficient	5.697 mm ⁻¹	
F(000)	680	
Crystal size	0.160 x 0.070 x 0.030 mm ³	
Theta range for data collection	4.393 to 75.594°.	
Index ranges	-8<=h<=8, -23<=k<=23, -12<=l<=12	
Reflections collected	21291	
Independent reflections	5628 [R(int) = 0.0497]	
Completeness to theta = 67.684°	100.0 %	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	5628 / 2 / 351	
Goodness-of-fit on F ²	1.063 527	

Final R indices [I>2sigma(I)]	R1 = 0.0362, wR2 = 0.0947
R indices (all data)	R1 = 0.0369, wR2 = 0.0955
Absolute structure parameter	-0.009(15)
Extinction coefficient	n/a
Largest diff. peak and hole	0.701 and -0.579 e.Å ⁻³

Figure 1. View of cation 1 in **5.7b** showing the atom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



Figure 2. View of cation 2 in **5.7b** showing the atom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



Chapter 6: Regio- and Enantioselective Iridium-Catalyzed N-Allylation of Indoles and Related Azoles with Racemic Branched Alkyl-Substituted Allylic Acetates*

6.1 Introduction

Iridium-catalyzed allylic amination has emerged as an important method for enantioselective C-N bond formation.¹⁻⁹ Pursuant to the pioneering work of Takeuchi,^{1,2} studies from the laboratories of Helmchen,^{3,4} Hartwig,^{5,6} Carreira⁷ and You^{8,9} have shown that chiral phosphoramidite-modified iridium complexes are especially effective in this regard (Figure 6.1). Like the parent palladium-catalyzed Tsuji-Trost reactions,¹⁰ the iridium-phosphoramidite-catalyzed allylic alkylations occur by way of cationic π allylmetal intermediates. Recently, we found that π -allyliridium-C,O-benzoates, which are well-known to promote nucleophilic allylation of carbonyl compounds,^{11,12} are also effective catalysts for highly regio- and enantioselective electrophilic allylation of amines.¹³ These π -allyliridium-*C*,*O*-benzoates react by way of neutral π -allylmetal intermediates, which may account for their amphiphilic character. One advantage of the π -allyliridium-*C*, *O*-benzoate catalysts relates to their facile preparation and remarkable stability toward air, water and SiO₂.¹⁴ Using SEGPHOS-modified π -allyliridium-C,Obenzoate catalysts, allylic aminations mediated by aliphatic amine nucleophiles occur with high levels of regio- and enantioselectivity.^{13a} It was later found that the tol-BINAPmodified π -allyliridium-C,O-benzoate (S)-Ir-I is a superior catalyst, enabling highly regio- and enantioselective allulation of electronically diverse primary and secondary (hetero)aryl amines.^{13b}

^{*}This chapter is based on the published work:

Kim, S. W.; Schempp, T. T.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. Angew. Chem. Int. Ed. 2019, 58, 7762.

Figure 6.1 Iridium catalysts for asymmetric allylic amination and chiral indolecontaining clinical candidates.



6.2 Reaction Development and Scope

In the latter study, it was shown that 5-amino-indole (5-N)-**6.1a** undergoes allylic alkylation at the 5-amino moiety to furnish (5-N)-**6.3a** with complete site-selectivity (eq. 6.1), suggesting indoles may not be subject to π -allyliridium-*C*, *O*-benzoate-catalyzed *N*-allylation. To challenge this assumption, the parent indole **6.1a** was exposed to

essentially identical conditions for allylic alkylation (eq. 6.2). To our delight, the product 6.3a was obtained in excellent yield as a single regioisomer in 93% enantiomeric excess. These results were deemed significant, as indoles and related azoles are prevalent structural motifs in clinical candidates and FDA approved drugs (Figure 6.1),^{15,16} and allylation at C3 is the major product observed in the intermolecular allylation of structurally and electronically unbiased indoles using palladium,¹⁷ ruthenium¹⁸ or iridium¹⁹ catalysts. For example, to enforce N-allylation, use of electron deficient indoles^{17j,19c} or heteroatom substituted π -allylmetal species (from alkoxyallenes) is required.¹⁷⁰ These limitations have necessitated the development of indirect methods for the synthesis of enantiomerically enriched N-allyl indoles (Figure 6.1).²⁰ As the development of enantioselective methods for intermolecular indole allylation that are completely N-selective and display exclusive branched regioselectivity remains an unmet challenge,¹⁷⁻²⁰ an effort to assess the scope the π -allyliridium-C,O-benzoate-catalyzed indole allylation was undertaken. Here, we show that the tol-BINAP-modified iridium complex (S)-Ir-I catalyzes highly enantioselective allylation of diverse indoles and azoles with complete *N*-regioselectivity and complete branched regioselectivity.



Table 6.1Regio- and enantioselective iridium-catalyzed allylic alkylation of indoles
and related azoles **6.1a-6.1s** using α -methyl allyl acetate **6.2a**.^a



^aYields of material isolated by silica gel chromatography. Regio- and diastereoselectivities were determined by ¹H NMR analysis of crude reaction mixtures. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

To assess the generality of indole allylation catalyzed by (S)-Ir-I, the reaction conditions utilized in equation 6.2 were applied to structural diverse indoles and azoles **6.1a-6.1p** (Table 6.1). Beyond the parent indole **6.1a**, electronically diverse indoles substituted at the 2-position (**6.1b**), 4-position (**6.1e**), 5-position (**6.1f**), 6-position (**6.1g**, **6.1h**) and 7-position (**6.1i-6.1k**) each underwent efficient allylation to deliver the respective adducts (**6.3b**, **6.3e-6.3k**) with complete *N*-regioselectivity, complete branched

regioselectivity and uniformly high levels of enantioselectivity. The structurally related azoles tetrahydro-carbazole **6.1c**, the "homo-tetrahydro-carbazole" **6.1d**, 3-methyl-indazole **6.1l**, 2-ethyl- benzimidazole **6.1m**, 6-azaindole **6.1n**, 7-azaindole **6.1o**, carbazole **6.1p** and 2-methyl-3-carboxypyrrole ethyl ester **6.1q** also underwent efficient *N*-allylation to form adducts **6.3c**, **6.3d** and **6.3l-6.3q**, respectively, as single constitutional isomers with high levels of enantiomeric enrichment. Of relevance to medicinal chemistry applications, the formation of adduct **6.3o** demonstrates tolerance of a pinacol boronate moiety under the conditions of asymmetric *N*-allylation. Additionally, to further test the limits of our method, catalyst-directed diastereoselectivity was explored in the allylation of the structurally complex chiral azole **6.1r**, which is related to AZD-9496, a clinical stage non-steroidal oral estrogen receptor inhibitor.²¹ Remarkably, the enantiomeric iridium catalysts (*S*)-Ir-I and (*R*)-Ir-I deliver the diastereomeric adducts **6.3r** and **6.3s** in >95% yield with complete levels of catalyst-directed stereoinduction.

Using diverse α -substituted allyl acetates **6.2b-6.2j**, the scope of the electrophilic partner was subsequently evaluated in *N*-allylations of the parent indole **6.1a** (Table 6.2). Branched allylic acetates bearing linear alkyl groups **6.2b-6.2d**, which incorporate phenyl, benzyl ether and methyl sulfide moieties, respectively, were efficiently converted to adducts **6.4b-6.4d**. As illustrated by the formation adduct **6.4e**, branched alkyl groups are tolerated. Finally, the preparation of adducts **6.4f-6.4j** highlight the tolerance of cycloalkyl substituents. All adducts **6.4b-6.4j** were formed as single constitutional isomers and with high levels of enantiomeric enrichment. The absolute stereochemistry of adducts **6.3a-6.3s** and **6.4b-6.4j** was assigned in analogy to adduct **6.3a**, which was determined through comparison of its optical rotation to a sample reported in the literature.^{8b} **Table 6.2**Regio- and enantioselective iridium-catalyzed allylic alkylation of indole**6.1a** using diverse α -substituted allyl acetates**6.2b-6.2j**.^a



^aYields of material isolated by silica gel chromatography. Regio- and diastereoselectivities were determined by ¹H NMR analysis of crude reaction mixtures. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

To further assess the breadth of the present protocol for catalytic enantioselective N-allylation, the cyclization of racemic allylic acetate **6.5a** was explored (eq. 6.3). The cyclization of 5a provides an opportunity to test the limits of stereoinduction, as the rate of enantiotopic π -facial interconversion must exceed the rate of 5-membered ring formation.²² The cyclization of **6.5a** also provides another context to evaluate the fidelity of N- vs C3-regioselectivity. In the event, exposure of **6.5a** to (*S*)-Ir-I under standard

conditions at 50 °C provided the product of *N*-cycloallylation **6.6a** as single constitutional

isomer without competing C3-cycloallylation.



Scheme 6.1 Iridium-catalyzed amination of enantiomerically enriched deuterated allylic acetate 6.2k with the enantiomeric catalysts (*S*)-Ir-I and (*R*)-Ir-I.^a



^aYields of material isolated by silica gel chromatography. Regio- and diastereoselectivities were determined by ¹H NMR analysis of crude reaction mixtures. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.

6.3 Discussion

The basicity of the reaction conditions and the acidity of indoles suggest the active nucleophiles in the present protocol for iridium-catalyzed *N*-allylation are *N*-centered anions. In previously reported allylic aminations catalyzed by π -allyliridium-*C*, *O*-benzoates, significantly less acidic aliphatic and aromatic amines were utilized, which likely serve as neutral *N*-nucleophiles.¹³ Hence, although a mechanism involving outer-sphere addition was established in the previously reported allylic aminations of the less acidic aliphatic and aromatic amines, it was unclear whether inner- or outer-sphere addition pathways were operative in the present indole-mediated asymmetric aminations. To address this question, the enantiomerically enriched (*Z*)-deuterated allylic acetate **6.2k** was subjected to standard conditions for allylic amination using indole **6.1a** (Scheme 1).[23] Using the enantiomeric catalysts (*S*)-Ir-I and (*R*)-Ir-I the stereoisomeric products **6.4k** and iso-**6.4k** are generated, which corroborates C-N bond formation through outer-sphere addition.

6.4 Conclusion

In summary, using π -allyliridium-*C*,*O*-benzoate catalysts in combination with racemic branched alkyl-substituted allylic acetate proelectrophiles, we report the first highly enantioselective intermolecular Tsuji-Trost-type indole *N*-allylations wherein complete *N*-regioselectivity is accompanied by complete branched-regioselectivity. Future work will explore related electrophilic *N*- and C-allylations catalyzed by cyclometallated π -allyliridium *C*,*O*-benzoates. These efforts illustrate the effectiveness of

academic-industrial collaboration vis-á-vis development of asymmetric methods for unmet challenges in synthetic chemistry of relevance to the drug discovery enterprise.

6.5 Experimental Details

General Information

All reactions were carried out under inert gas atmosphere (nitrogen or argon) unless otherwise indicated. Resealable pressure tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were flame dried followed by cooling in a desiccator or under a stream of inter gas prior to use. All commercial reagents and anhydrous solvents were used as received from vendors (Fischer Scientific, Sigma Aldrich and Combi Blocks) without further purification. The used Iridium catalyst (S)-Ir-I and (R)-Ir-I was prepared according to literature known procedures.¹ Cesium carbonate was used as received from Rockwell Lithium. Preparative column chromatography employing Silicycle silica gel (40-63 µm) was performed according to the method of Still² or on a Teledyne Isco Combiflash R_f utilizing Silicycle HP columns using a mobile phase composed of either heptane/isopropyl acetate, heptanes/ethyl acetate or dichloromethane/methanol. Reactions were monitored by a Shimadzu LCMS/UV system with LC-30AD solvent pump, 2020 MS, Sil-30AC autosampler, SPD-M30A UV detector, CTO-20A column oven, using a 2-98% acetonitrile/0.1% formic acid (or 0.001% ammonia) gradient over 2.5 minutes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynamic Absorbents F). Visualization was accomplished with UV light followed by dipping in CAM, p-Anisaldehyde (PAA), or KMnO4 stain solution followed by heating.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perki*N*-Elmer 1600 spectrometer using a diamond ATR unit. High-resolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M+H, M+Na), or a suitable fragment ion. Nuclear magnetic resonance (¹H, ¹³C, ¹⁹F NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (400 MHz) or Varian INOVA (500 MHz) spectrometer equipped with a Bruker cryoprobe. The chemical shifts are given as parts per million (ppm) and were referenced to the residual solvent signal (CDCl₃: $\delta_{\rm H} = 7.26$ ppm, $\delta_{\rm C} = 77.16$ ppm). Specific optical rotations were recorded on an Atago AP-300 automatic polarimeter at the sodium line (589 nm) in CHCl3. Solution concentrations are given in the units of 10–2 g mL⁻¹.

Experimental Details and Spectral Data

Synthesis of Allylic Acetates 6.2d, 6.2g, and 6.2h

The allylic acetates **6.2d**, **6.2g**, and **6.2h** were prepared by the Grignard reaction and acetylation of the as shown below. The allylic acetates **6.2b**,³ **6.2c**,⁴ **6.2e**,⁵ **6.2f**,⁶ **6.2i**,⁷ **6.2j**,⁵ and **6.2k**⁷ were identical in all respects to the reported materials.



(S)-5-(methylthio)pent-1-en-3-yl acetate (6.2d)

Procedures

To a round-bottomed flask charged with the corresponding aldehyde (1.04 g, 10.0 mmol, 100 mol%) under an argon atmosphere was added THF (50.0 mL, 0.2 M). The reaction flask was placed an ice batch. After 10 minutes, vinyl magnesium bromide solution (12.0 mL, 12.0 mmol, 120 mol%, 1.0 M in THF) was added slowly and the mixture was stirred at room temperature for 1 hour, at which point acetic anhydride (1.42 mL, 15.0 mmol, 150 mol%) and triethylamine (2.79 mL, 20.0 mmol, 200 mol%) were added and the reaction was stirred vigorously overnight. After water (50 mL) was added, the mixture was transferred to a separatory funnel. The organic layer was extracted with diethyl ether (50 mL \times 2) and the combined organic layers were washed with 1 N HCl (50 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, pentanes: diethyl ether = 40:1–20:1) to furnish the title compound as a colorless oil (1.31 g, 4.58 mmol) in 75% yield over 2 steps.

<u>TLC</u> (SiO₂) $R_f = 0.40$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 5.77 (ddd, J = 17.0, 10.5, 6.4 Hz, 1H), 5.34 (dt, J = 7.3, 1.0 Hz, 1H), 5.27 (dt, J = 17.2, 1.3 Hz, 1H), 5.20 (dt, J = 10.5, 1.2 Hz, 1H), 2.49 (ddd, J = 8.4, 6.7, 1.9 Hz, 2H), 2.10 (s, 3H), 2.07 (s, 3H), 2.01 – 1.81 (m, 2H).

¹³C NMR (125 MHz, CDCl₃): δ 170.2, 135.8, 117.2, 73.6, 33.7, 29.6, 21.2, 15.5. HRMS (ESI): Calculated for C₈H₁₄O₂S [M+Na⁺] = 197.0607, Found 197.0607.







Procedures

To a round-bottomed flask charged with the corresponding aldehyde (0.90 g, 7.50 mmol, 100 mol%) under an argon atmosphere was added THF (37.5 mL, 0.2 M). The reaction flask was placed an ice batch. After 10 minutes, vinyl magnesium bromide solution (9.0 mL, 9.0 mmol, 120 mol%, 1.0 M in THF) was added slowly and the mixture was stirred at room temperature for 1 hour, at which point acetic anhydride (1.06 mL, 11.3 mmol, 150 mol%) and triethylamine (2.09 mL, 15.0 mmol, 200 mol%) were added and the

reaction was stirred vigorously overnight. After water (50 mL) was added, the mixture was transferred to a separatory funnel. The organic layer was extracted with diethyl ether (50 mL \times 2) and the combined organic layers were washed with 1 N HCl (50 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, pentanes: diethyl ether = 40:1–20:1) to furnish the title compound as a colorless oil (0.87 g, 4.58 mmol) in 61% yield over 2 steps.

<u>**TLC** (SiO₂</u>) $R_f = 0.39$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 5.73 (ddd, *J* = 17.1, 10.5, 6.6 Hz, 1H), 5.36 – 5.22 (m, 3H), 2.61 (dtdd, *J* = 16.0, 10.7, 5.1, 3.0 Hz, 2H), 2.55 – 2.27 (m, 3H), 2.12 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2$, 133.6, 118.3, 75.89 – 75.81 (m), 37.34 – 36.70 (m, 2C), 26.27 (dd, J = 13.7, 5.9 Hz), 21.01.

¹⁹**F** NMR (376 MHz, CDCl₃): δ = -83.1 - -83.8 (m, 1F), -95.6 - -96.3 (m, 1F).

<u>LRMS</u> (CI): Calculated for $C_7H_9F_2$ [M–OAc]⁺ = 131, Found 131.

<u>FTIR</u> (neat): 2958, 1738, 1413, 1372, 1297, 1229, 1170, 1020, 908 cm⁻¹.





tert-butyl 3-(1-acetoxyallyl)azetidine-1-carboxylate (6.2h)



Procedures

To a round-bottomed flask charged with the corresponding aldehyde (0.90 g, 7.50 mmol, 100 mol%) under an argon atmosphere was added THF (37.5 mL, 0.2 M). The reaction flask was placed an ice batch. After 10 minutes, vinyl magnesium bromide solution (9.0 mL, 9.0 mmol, 120 mol%, 1.0 M in THF) was added slowly and the mixture was stirred at room temperature for 1 hour, at which point acetic anhydride (1.06 mL, 11.3 mmol, 150 mol%) and triethylamine (2.09 mL, 15.0 mmol, 200 mol%) were added and the

reaction was stirred vigorously overnight. After water (50 mL) was added, the mixture was transferred to a separatory funnel. The organic layer was extracted with diethyl ether (50 mL \times 2) and the combined organic layers were washed with 1 N HCl (50 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, pentanes: diethyl ether = 40:1–20:1) to furnish the title compound as a colorless oil (1.09 g, 4.28 mmol) in 57% yield over 2 steps.

<u>**TLC (SiO₂**</u>) $R_f = 0.24$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 5.69$ (ddd, J = 17.1, 10.5, 6.4 Hz, 1H), 5.38 – 5.32 (m, 1H), 5.31 – 5.20 (m, 2H), 3.91 (td, J = 8.6, 2.5 Hz, 2H), 3.72 (dd, J = 8.8, 5.5 Hz, 1H), 3.62 (dd, J = 8.9, 5.6 Hz, 1H), 2.82 – 2.66 (m, 1H), 2.06 (s, 3H), 1.41 (s, 9H).

<u>1³C NMR</u> (125 MHz, CDCl₃): δ = 170.2, 156.2, 133.2, 118.7, 79.5, 75.2, 50.7 (d, J = 72.7 Hz), 31.6, 28.4, 21.1.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{21}NO_4$ [M+Na⁺] = 278.1363, Found 278.1371.

<u>FTIR</u> (neat): 2973, 1697, 1392, 1365, 1227, 1134, 1020, 942, 772, 735 cm⁻¹.



Procedures and Spectral Data for Synthesis of Allylic Amines 6.3a-6.3q, 6.4b-6.4j

Enantioselective Ir-catalyzed allylic alkylation with amine nucleophiles



General procedure

A pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (200 mol%), (*S*)-Ir-I (5 mol%). The tube was purged with argon for 5 minutes. DME (1.0 M) was added followed by the allylic acetate (100 mol%) and the amine (200 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at the indicated temperature and stirred for the indicated period of time. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography.

(S)-1-(but-3-en-2-yl)-1H-indole (6.3a)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (103.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 24 hr). The title compound was obtained in 78% yield (58.8 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.44$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.63 (dt, *J* = 7.9, 1.1 Hz, 1H), 7.36 (dd, *J* = 8.3, 0.9 Hz, 1H), 7.22 – 7.16 (m, 2H), 7.10 (ddd, *J* = 7.9, 7.0, 1.0 Hz, 1H), 6.52 (dd, *J* = 3.2, 0.8 Hz, 1H), 6.05 (ddd, *J* = 17.5, 10.4, 4.7 Hz, 1H), 5.18 (ddd, *J* = 10.4, 1.6, 0.9 Hz, 1H), 5.12 – 5.04 (m, 2H), 1.65 (d, *J* = 6.9 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 138.8, 135.7, 128.7, 124.7, 121.3, 121.0, 119.4, 115.5, 109.8, 101.4, 53.1, 19.7.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{13}N[M+H^+] = 172.1126$, Found 172.1122.

<u>FTIR</u> (neat): 2980, 1509, 1459, 1310, 1264, 1213, 926, 735 cm⁻¹.

 $[\alpha]_{D}^{28} = -20.6 \ (c \ 0.2, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 93%.







(S)-1-(but-3-en-2-yl)-2-phenyl-1H-indole (6.3b)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (170.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (80.5 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>TLC (SiO</u>) $R_f = 0.43$ (hexanes: ethyl acetate = 10:1).

¹H NMR (400 MHz, CDCl₃): δ = 7.67 – 7.62 (m, 1H), 7.52 – 7.39 (m, 6H), 7.14 (qdd, J = 8.1, 6.7, 1.2 Hz, 2H), 6.52 (d, J = 0.8 Hz, 1H), 6.22 (dddd, J = 17.3, 10.6, 3.7, 0.7 Hz, 1H), 5.29 (ddd, J = 10.6, 2.3, 1.0 Hz, 1H), 5.20 (ddt, J = 17.3, 2.0, 0.9 Hz, 1H), 5.17 – 5.10 (m, 1H), 1.62 (d, J = 7.1 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{121.0, 120.7, 119.7, 115.6, 113.0, 102.4, 52.9, 18.2.}$

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{17}N[M+H^+] = 248.1434$, Found 248.1435.

<u>FTIR</u> (neat): 2980, 1604, 1455, 1347, 1313, 1163, 921, 763, 700 cm⁻¹.

 $[\alpha]_{D}^{28} = -40.8 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:*i*-PrOH = 97.5:2.5, 1.00 mL/min, 254 nm), ee = 90%.</u>





(S)-9-(but-3-en-2-yl)-2,3,4,9-tetrahydro-1H-carbazole (6.3c)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (150.7 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 81% yield (80.3 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>TLC (SiO</u>₂) $R_f = 0.46$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.51 - 7.45$ (m, 1H), 7.39 - 7.34 (m, 1H), 7.14 - 7.03 (m, 2H), 6.19 (ddd, J = 17.3, 10.6, 4.0 Hz, 1H), 5.23 (ddd, J = 10.6, 2.3, 1.0 Hz, 1H), 5.15 (ddd, J = 17.4, 2.2, 1.0 Hz, 1H), 5.04 (dddd, J = 11.1, 7.1, 4.8, 2.0 Hz, 1H), 2.77 - 2.71 (m, 4H), 1.98 - 1.91 (m, 2H), 1.90 - 1.83 (m, 2H), 1.65 (d, J = 7.1 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 139.1, 135.2, 135.1, 127.9, 120.2, 118.4, 117.7, 115.2, 110.7, 109.8, 51.5, 23.5, 23.2, 23.1, 21.1, 18.4.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{19}N[M+H^+] = 226.1596$, Found 226.1605.

<u>FTIR</u> (neat): 2933, 1462, 1368, 1264, 1175, 922, 732, 703 cm⁻¹.

 $[\alpha]_{\mathbf{D}}^{\mathbf{28}} = -7.7 \ (c \ 0.2, \ \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.




(S)-5-(but-3-en-2-yl)-5,6,7,8,9,10-hexahydrocyclohepta[b]indole (6.3d)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (163.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 83% yield (87.4 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.46$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.53 - 7.46$ (m, 1H), 7.37 - 7.31 (m, 1H), 7.11 - 7.04 (m, 2H), 6.20 (ddd, J = 17.4, 10.5, 3.8 Hz, 1H), 5.24 (ddd, J = 10.7, 2.4, 0.9 Hz, 1H), 5.22 - 5.13 (m, 2H), 2.92 - 2.82 (m, 4H), 1.94 - 1.87 (m, 2H), 1.80 - 1.74 (m, 4H), 1.63 (d, J = 7.0 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{11}} (100 \text{ MHz}, \text{CDCl}_3): \delta = 139.7, 138.9, 134.5, 128.2, 119.9, 118.5, 117.5, 115.0, 114.5, 110.3, 51.2, 31.7, 28.1, 27.3, 26.9, 24.1, 18.9.$

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{21}N[M+H^+] = 240.1747$, Found 240.1747.

<u>FTIR</u> (neat): 2917, 2845, 1463, 1346, 1204, 1086, 919, 735 cm⁻¹.

 $[\alpha]_{D}^{28} = -8.8 \ (c \ 0.2, \ \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 90%.</u>





(S)-1-(but-3-en-2-yl)-1H-indole-4-carbonitrile (6.3e)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (125.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 24 hr). The title compound was obtained in 83% yield (72.0 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.26$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.58 (d, J = 8.3 Hz, 1H), 7.47 (dd, J = 7.3, 0.8 Hz, 1H), 7.36 (d, J = 3.3 Hz, 1H), 7.22 (dd, J = 8.3, 7.4 Hz, 1H), 6.77 – 6.70 (m, 1H), 6.03 (ddd, J = 17.1, 10.4, 5.1 Hz, 1H), 5.22 (dd, J = 10.5, 1.6 Hz, 1H), 5.14 – 5.03 (m, 2H), 1.68 (d, J = 6.9 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 138.0, 135.4, 130.0, 127.5, 125.0, 121.0, 118.8, 116.2, 114.6, 103.3, 100.6, 53.6, 19.8.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{12}N_2$ [M+H⁺] = 219.0893, Found 219.0900.

<u>FTIR (neat)</u>: 2983, 2360, 2342, 2222, 1435, 1274, 1260, 750 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -9.0 \ (c \ 1.0, \ {\rm CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 91%.





(S)-5-bromo-1-(but-3-en-2-yl)-1H-indole (6.3f)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (172.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (50 °C, 24 hr). The title compound was obtained in 79% yield (87.0 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.35$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.75 (dd, J = 1.9, 0.7 Hz, 1H), 7.27 (d, J = 9.0 Hz, 1H), 7.25 – 7.14 (m, 2H), 6.47 (dd, J = 3.3, 0.8 Hz, 1H), 6.02 (ddd, J = 17.3, 10.4, 4.9 Hz, 1H), 5.19 (ddd, J = 10.3, 1.6, 0.8 Hz, 1H), 5.09 – 4.98 (m, 2H), 1.64 (d, J = 6.9 Hz, 3H). ¹³<u>C NMR</u> (125 MHz, CDCl₃): δ 138.3, 134.4, 130.3, 125.9, 124.1, 123.4, 115.8, 112.7, 111.2, 101.0, 53.3, 19.7.

<u>HRMS</u> (ESI): Calculated for C₁₂H₁₂BrN [M+H⁺] = 250.0226, Found 250.0228. **<u>FTIR (neat)</u>**: 2982, 2370, 2342, 1462, 1275, 1262, 1208, 751 cm⁻¹. $[\alpha]_{D}^{28} = -18.5 (c \ 1.0, CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.





Methyl (S)-1-(but-3-en-2-yl)-1H-indole-6-carboxylate (6.3g)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (154.2 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 82% yield (82.0 mg, 0.36 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1).

<u>TLC (SiO</u>) $R_f = 0.38$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 8.14 (s, 1H), 7.80 (dd, J = 8.3, 1.5 Hz, 1H), 7.64 (d, J = 8.3 Hz, 1H), 7.34 (d, J = 3.2 Hz, 1H), 6.59 – 6.53 (m, 1H), 6.07 (ddd, J = 17.2, 10.4, 5.0 Hz, 1H), 5.19 (ddt, J = 13.9, 6.9, 1.6 Hz, 2H), 5.14 – 5.05 (m, 1H), 3.94 (s, 3H), 1.67 (d, J = 6.9 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 168.3, 138.3, 135.1, 132.3, 128.0, 123.0, 120.5, 120.4, 115.9, 112.1, 101.9, 53.1, 51.9, 19.9.

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{15}NO_2[M+H^+] = 252.0995$, Found 252.1003.

<u>FTIR (neat)</u> 2983, 2950, 2359, 2342, 1708, 1242, 750, cm⁻¹.

 $[\alpha]_{D}^{28} = -46.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), *ee* = 88%.





(S)-1-(but-3-en-2-yl)-7-fluoro-1*H*-indole (6.3h)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (118.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 92% yield (77.0 mg, 0.41 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.53$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.38 (d, J = 7.9 Hz, 1H), 7.19 (d, J = 3.2 Hz, 1H), 6.99 (td, J = 7.8, 4.4 Hz, 1H), 6.91 – 6.84 (m, 1H), 6.57 – 6.49 (m, 1H), 6.16 – 6.07 (m, 1H), 5.48 (tt, J = 5.3, 1.6 Hz, 1H), 5.18 (d, J = 10.4 Hz, 1H), 5.08 (d, J = 18.6 Hz, 1H), 1.64 (d, J = 7.9 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 151.1, 149.2, 139.2, 132.5, 125.7, 123.8, 119.6, 116.7, 115.4, 107.4, 102.4, 54.9, 54.9, 20.4.

¹⁹**F** NMR (376 MHz, CDCl₃): δ = -134.1.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{12}FN[M+H^+] = 190.1027$, Found 190.1026.

<u>FTIR (neat)</u> 2984, 2360, 2342, 1275, 1261, 1236, 750 cm⁻¹.

 $[\alpha]_{D}^{28} = -37.8 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.







(S)-7-bromo-1-(but-3-en-2-yl)-1*H*-indole (6.3i)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (177.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 77% yield (84.7 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.60$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.56 (dd, J = 7.8, 1.1 Hz, 1H), 7.36 (dd, J = 7.6, 1.1 Hz, 1H), 7.23 (d, J = 3.4 Hz, 1H), 6.93 (t, J = 7.7 Hz, 1H), 6.53 (d, J = 3.3 Hz, 1H), 6.36 (ddt, J = 6.6, 4.2, 2.0 Hz, 1H), 6.14 (ddd, J = 17.3, 10.5, 4.4 Hz, 1H), 5.22 (ddd, J = 10.5, 1.8, 0.9 Hz, 1H), 5.06 (ddd, J = 17.3, 2.0, 1.0 Hz, 1H), 1.61 (d, J = 6.8 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): δ 139.4, 132.1, 132.0, 127.3, 126.6, 120.5, 120.4, 115.4, 103.6, 102.4, 52.6, 20.5.

HRMS (ESI): Calculated for $C_{12}H_{12}BrN [M+H^+] = 250.0226$, Found 250.0223. **FTIR (neat):** 2980, 2360, 2342, 1333, 1301, 751, 718 cm⁻¹. $[\alpha]_D^{28} = -7.3 (c \ 1.0, CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), *ee* = 88%.





(S)-1-(but-3-en-2-yl)-3-methyl-1*H*-indazole (6.3j)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (116.3 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 67% yield (55.9 mg, 0.30 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1).

<u>TLC</u> (SiO₂) $R_f = 0.19$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.65 (dt, J = 8.1, 1.0 Hz, 1H), 7.38 (dt, J = 8.5, 1.0 Hz, 1H), 7.33 (ddd, J = 8.4, 6.7, 1.1 Hz, 1H), 7.11 (ddd, J = 7.9, 6.7, 1.0 Hz, 1H), 6.20 – 6.06 (m, 1H), 5.25 – 5.14 (m, 2H), 5.13 (dt, J = 17.2, 1.3 Hz, 1H), 2.59 (s, 3H), 1.72 (d, J = 6.9 Hz, 3H).

<u>1³C NMR</u> (125 MHz, CDCl₃): δ 141.4, 139.8, 138.6, 125.9, 123.7, 120.5, 119.7, 115.6, 109.4, 56.5, 19.3, 12.0.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{14}N_2$ [M+H⁺] = 187.1230, Found 187.1233.

<u>FTIR (neat)</u> 2360, 2342, 1276, 1261, 764, 750 cm⁻¹.

 $[\alpha]_{D}^{28} = -10.2 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 93%.





(S)-1-(but-3-en-2-yl)-2-ethyl-1H-benzo[d]imidazole (6.3k)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the benzimidazole (128.6 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 85% yield (74.9 mg, 0.37 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 2:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.26$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.73 (dt, *J* = 8.0, 0.9 Hz, 1H), 7.37 (dt, *J* = 7.9, 0.9 Hz, 1H), 7.18 (dddd, *J* = 22.6, 8.4, 7.2, 1.2 Hz, 2H), 6.14 (ddd, *J* = 17.3, 10.6, 4.0 Hz, 1H), 5.30 (ddd, *J* = 10.5, 2.2, 0.7 Hz, 1H), 5.18 (ddd, *J* = 17.3, 2.2, 0.7 Hz, 1H), 5.11 (ttd, *J* = 7.1, 4.7, 2.0 Hz, 1H), 2.92 (qd, *J* = 7.5, 1.3 Hz, 2H), 1.70 (d, *J* = 7.1 Hz, 3H), 1.45 (t, *J* = 7.5 Hz, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{111.6, 52.4, 21.5, 18.3, 12.2.} (100 \text{ MHz}, \text{CDCl}_3): \delta = 155.6, 143.0, 137.0, 133.7, 121.6, 121.5, 119.3, 116.6, 111.6, 52.4, 21.5, 18.3, 12.2.$

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{16}N_2$ [M+H⁺] = 201.1386, Found 201.1390.

<u>FTIR</u> (neat): 2978, 1517, 1457, 1402, 1277, 927, 745 cm⁻¹.

 $[\alpha]_{D}^{28} = +13.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 90%.





(S)-1-(but-3-en-2-yl)-1*H*-pyrrolo[2,3-*c*]pyridine (6.3l)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (104.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 71% yield (53.0 mg, 0.31 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.31$ (hexanes: ethyl acetate = 1:4).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 8.80 (s, 1H), 8.23 (d, J = 5.5 Hz, 1H), 7.51 (dd, J = 5.5, 1.1 Hz, 1H), 7.32 (d, J = 3.2 Hz, 1H), 6.52 (d, J = 3.1 Hz, 1H), 6.12 – 6.01 (m, 1H), 5.26 – 5.20 (m, 1H), 5.17 (dd, J = 7.0, 5.2 Hz, 1H), 5.11 (dt, J = 17.2, 1.2 Hz, 1H), 1.69 (d, J = 7.0 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 138.5, 138.1, 133.5, 133.3, 132.8, 128.5, 116.3, 115.4, 100.9, 53.9, 19.9.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{12}N_2$ [M+H⁺] = 173.1073, Found 173.1073.

<u>FTIR (neat)</u>: 3084, 3035, 3016, 2980, 2343, 1642, 1598, 1558, 1496, 817, 776, 734 cm⁻¹.

 $[\alpha]_{D}^{28} = -18.5 (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:i-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 90%.





(*S*)-1-(but-3-en-2-yl)-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*-pyrrolo[2,3*b*]pyridine (6.3m)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (213.9 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 76% yield (100 mg, 0.34 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.39$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): (d, J = 1.6 Hz, 1H), 8.35 (d, J = 1.7 Hz, 1H), 7.24 (d, J = 3.6 Hz, 1H), 6.48 (d, J = 3.6 Hz, 1H), 6.10 (d, J = 2.0 Hz, 1H), 5.72 (d, J = 2.0 Hz, 1H), 5.17 (dt, J = 10.5, 1.4 Hz, 1H), 5.11 – 4.98 (m, 1H), 1.60 (d, J = 7.0 Hz, 3H), 1.36 (s, 12H).

¹³C NMR (125 MHz, CDCl₃): δ 149.0, 148.6, 138.8, 135.9, 125.0, 120.2, 115.5, 100.4,
83.7, 50.4, 24.9, 19.8.

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{23}BN_2O_2[M+H^+] = 299.1928$, Found 299.1937.

<u>FTIR (neat)</u>: 2978, 2360, 2342, 1363, 1348, 1275, 1261, 764, 750, cm⁻¹.

 $[\alpha]_{D}^{28} = -24.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 92%.





(S)-9-(but-3-en-2-yl)-9H-carbazole (6.3n)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the carbazole (147.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 91% yield (88.6 mg, 0.40 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1).

<u>TLC (SiO</u>₂) $R_f = 0.43$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 8.15 (dt, *J* = 7.8, 1.0 Hz, 2H), 7.52 (dt, *J* = 8.3, 0.9 Hz, 2H), 7.46 (ddd, *J* = 8.2, 7.0, 1.2 Hz, 2H), 7.30 – 7.24 (m, 2H), 6.31 (ddd, *J* = 17.4, 10.6, 3.7 Hz, 1H), 5.46 (ttd, *J* = 7.0, 4.5, 1.7 Hz, 1H), 5.39 – 5.31 (m, 2H), 1.77 (d, *J* = 7.1 Hz, 3H).

<u>13C NMR</u> (125 MHz, CDCl₃): $\delta = \delta$ 139.6, 138.3, 125.4, 123.4, 120.3, 118.8, 116.0, 110.1, 51.5, 17.1.

HRMS (ESI): Calculated for $C_{16}H_{15}N[M+H^+] = 222.1277$, Found 222.1279.

<u>FTIR</u> (neat): 2982, 1595, 1481, 1452, 1328, 1223, 1155, 924, 747, 722 cm⁻¹.

 $[\alpha]_{D}^{28} = +2.0 (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Two connected chiralcel OJ-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 93%.</u>





Ethyl (S)-1-(but-3-en-2-yl)-2-methyl-1H-pyrrole-3-carboxylate (6.30)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (110.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (68 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1).

<u>TLC (SiO</u>) $R_f = 0.39$ (hexanes: ethyl acetate = 5:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 6.56 (s, 2H), 5.93 (ddd, J = 17.2, 10.5, 4.8 Hz, 1H), 5.15 (ddd, J = 10.5, 1.8, 0.9 Hz, 1H), 4.90 (ddd, J = 17.1, 1.9, 0.9 Hz, 1H), 4.73 (dd, J = 6.9, 4.9 Hz, 1H), 4.26 (q, J = 7.1 Hz, 2H), 2.52 (s, 3H), 1.53 (d, J = 6.9 Hz, 3H), 1.33 (t, J = 7.1 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 165.7, 138.5, 135.4, 116.4 115.5, 112.1, 109.5, 59.2, 53.0, 20.1, 14.6, 10.8.

HRMS (ESI): Calculated for $C_{12}H_{17}NO_2[M+H^+] = 208.1332$, Found 208.1331.

<u>FTIR (neat)</u>: 2982, 2360, 2342, 1696, 1223, 929, 764, 750 cm⁻¹.

 $[\alpha]_{D}^{28} = -9.2 (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.






(1*R*,3*R*)-9-((*S*)-but-3-en-2-yl)-1-(2,6-difluoro-4-iodophenyl)-3-methyl-2-(2,2,2-trifluoroethyl)-2,3,4,9-tetrahydro-1H-pyrido[3,4-b]indole (6.3p)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (445.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 95% yield, >20:1 dr (234.2 mg, 0.42 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.43$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.54 - 7.51$ (m, 1H), 7.36 - 7.32 (m, 1H), 7.25 - 7.16 (m, 2H), 7.16 - 7.07 (m, 2H), 5.87 (ddd, J = 17.4, 10.6, 4.0 Hz, 1H), 5.31 (s, 1H), 5.07 (ddd, J = 10.6, 2.3, 0.9 Hz, 1H), 4.86 (ddd, J = 17.4, 2.3, 0.9 Hz, 1H), 4.64 - 4.54 (m, 1H), 3.38 (dq, J = 12.3, 6.5 Hz, 1H), 3.25 (dq, J = 15.7, 9.9 Hz, 1H), 3.00 - 2.82 (m, 2H), 2.61 (ddd, J = 16.2, 9.7, 1.3 Hz, 1H), 1.49 (d, J = 7.1 Hz, 3H), 1.20 (d, J = 6.7 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{^{13}C \text{ NMR}} (100 \text{ MHz}, \text{CDCl}_3): \delta = 137.3, 135.6, 130.7, 127.5, 127.0, 124.3, 121.7 (d, J = 25.9 \text{ Hz}), 121.3, 119.05, 118.4, 116.8 (t, J = 14.5 \text{ Hz}), 115.9, 111.9, 109.6, 91.9 (t, J = 10.9 \text{ Hz}), 54.1, 52.4, 49.5, 48.7 (q, J = 32.3 \text{ Hz}), 25.5, 18.5, 18.4.$

¹⁹**F NMR** (376 MHz, CDCl₃): δ = -70.77, -110.47.

<u>HRMS</u> (ESI): Calculated for $C_{24}H_{22}N_2F_5I[M+H^+] = 561.0821$, Found 561.0821.

FTIR (neat): 2982, 1606, 1569, 1409, 1195, 1140, 1051, 1022, 843, 739 cm⁻¹.

 $[\alpha]_{\rm D}^{28}$ = -88.6 (*c* 0.2, CHCl₃).





20 -25 -30 -35 -40 -45 -50 -55 -60 -65 -70 -75 -80 -85 -90 -95 -100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -16 fl (ppm)

(1*R*,3*R*)-9-((*R*)-but-3-en-2-yl)-1-(2,6-difluoro-4-iodophenyl)-3-methyl-2-(2,2,2-trifluoroethyl)-2,3,4,9-tetrahydro-1H-pyrido[3,4-b]indole (6.3q)



Procedures

The allylic acetate (50.2 mg, 0.44 mmol, 100 mol%) and the indole (445.5 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions with (*R*)-Ir-I (70 °C, 24 hr). The title compound was obtained in 96% yield, >20:1 dr (236.7 mg, 0.42 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.43$ (hexanes: ethyl acetate = 10:1).

¹**H NMR** (400 MHz, CDCl₃): $\delta = 7.56 - 7.51$ (m, 1H), 7.37 - 7.31 (m, 1H), 7.26 - 7.18 (m, 2H), 7.12 (pd, J = 7.0, 1.4 Hz, 2H), 6.01 (ddd, J = 17.3, 10.6, 4.1 Hz, 1H), 5.27 (s, 1H), 5.13 (ddd, J = 10.5, 2.2, 0.8 Hz, 1H), 5.05 (ddd, J = 17.3, 2.2, 0.9 Hz, 1H), 4.53 (ttd, J = 7.0, 4.7, 2.0 Hz, 1H), 3.39 (dq, J = 12.2, 6.0 Hz, 1H), 3.24 (dq, J = 15.7, 9.8 Hz, 1H), 2.97 – 2.83 (m, 2H), 2.61 (ddd, J = 16.1, 9.5, 1.3 Hz, 1H), 1.34 (d, J = 7.1 Hz, 3H), 1.19 (d, J = 6.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 138.2, 135.4, 130.7, 127.6, 127.0, 124.2, 121.7 (d, *J* = 26.5 Hz), 121.2, 119.0, 118.4, 116.8 (t, *J* = 14.5 Hz), 115.7, 112.0, 109.4, 92.0 (t, *J* = 11.1 Hz), 54.1, 52.5, 50.0, 48.7 (q, *J* = 32.1 Hz), 25.5, 18.3, 17.7. ¹⁹F NMR (376 MHz, CDCl₃): δ = -70.9, -110.5.

<u>HRMS</u> (ESI): Calculated for $C_{24}H_{22}N_2F_5I[M+H^+] = 561.0821$, Found 561.0828.

<u>FTIR</u> (neat): 2981, 1606, 1569, 1408, 1264, 1139, 1049, 1021, 843, 735 cm⁻¹. $[\alpha]_D^{28} = -75.2 (c \ 0.2, CHCl_3).$





(S)-1-(5-phenylpent-1-en-3-yl)-1*H*-indole (6.4b)



Procedures

The allylic acetate (89.8 mg, 0.44 mmol, 100 mol%) and the indole (103.0 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 79% yield (91.0 mg, 0.35 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1).

<u>TLC (SiO</u>) $R_f = 0.36$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.65 (dq, J = 7.8, 1.0 Hz, 1H), 7.28 (dd, J = 6.9, 1.1 Hz, 3H), 7.24 – 7.15 (m, 3H), 7.15 – 7.01 (m, 3H), 6.58 (dt, J = 3.2, 0.9 Hz, 1H), 6.02 (dddd, J = 16.8, 10.5, 5.6, 0.7 Hz, 1H), 5.16 (ddt, J = 10.5, 1.7, 0.8 Hz, 1H), 5.02 (ddt, J = 17.1, 1.8, 0.9 Hz, 1H), 4.86 (d, J = 8.0 Hz, 1H), 2.56 (t, J = 7.7 Hz, 2H), 2.45 – 2.24 (m, 2H). ¹³C NMR (125 MHz, CDCl₃): δ 140.9, 137.6, 136.1, 128.7, 128.5, 126.2, 124.9, 121.4, 121.0, 119.5, 116.3, 109.9, 101.9, 57.3, 35.8, 32.3.

HRMS (ESI): Calculated for C₁₉H₁₉N [M+H⁺] = 262.1590, Found 262.1598. **FTIR (neat)**: 3026, 2928, 2360, 2342, 1610, 1476, 1459, 740 cm⁻¹. $[\alpha]_{D}^{28} = +32.1$ (*c* 1.0, CHCl₃).

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.





(S)-1-(5-(benzyloxy)pent-1-en-3-yl)-1H-indole (6.4c)



Procedures

The allylic acetate (103.1 mg, 0.44 mmol, 100 mol%) and the indole (103.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 65% yield (67.0 mg, 0.29 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.53$ (heptanes: isopropyl acetate = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.65 (dt, *J* = 7.8, 1.1 Hz, 1H), 7.41 (dd, *J* = 8.3, 1.0 Hz, 1H), 7.36 – 7.26 (m, 5H), 7.19 (ddd, *J* = 8.3, 7.0, 1.3 Hz, 1H), 7.15 – 7.09 (m, 2H), 6.55 (dd, *J* = 3.2, 0.8 Hz, 1H), 6.07 (ddd, *J* = 17.2, 10.4, 5.3 Hz, 1H), 5.26 – 5.15 (m, 2H), 5.09 (ddd, *J* = 17.3, 1.7, 1.0 Hz, 1H), 4.36 (s, 2H), 3.47 (dt, *J* = 9.8, 5.1 Hz, 1H), 3.16 (ddd, *J* = 9.5, 8.5, 4.8 Hz, 1H), 2.38 – 2.22 (m, 2H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{100} \text{ MHz}, \text{ CDCl}_3}: \delta = 138.2, 137.6, 136.1, 128.6, 128.4, 127.7, 127.6, 125.3, 126.4, 127.7, 127.6, 125.3, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 125.3, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 125.3, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 125.3, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 127.7, 127.6, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 126.4, 12$

121.3, 120.9, 119.4, 116.2, 110.1, 101.8, 73.1, 66.3, 54.7, 34.3.

<u>HRMS</u> (ESI): Calculated for $C_{20}H_{21}NO[M+H^+] = 292.1696$, Found 292.1698.

<u>FTIR</u> (neat): 2862, 1458, 1308, 1264, 1195, 1105, 734, 700 cm⁻¹.

 $[\alpha]_{D}^{28} = -20.7 \ (c \ 0.2, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 210 nm), ee = 91%.</u>







(S)-1-(5-(methylthio)pent-1-en-3-yl)-1H-indole (6.4d)



Procedures

The allylic acetate (34.9 mg, 0.2 mmol, 100 mol%) and the indole (46.9 mg, 0.84 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 71% yield (33.0 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1).

<u>TLC</u> (SiO₂) $R_f = 0.32$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ 7.64 (d, J = 7.8 Hz, 1H), 7.41 (d, J = 8.3 Hz, 1H), 7.23 – 7.17 (m, 1H), 7.16 (d, J = 3.3 Hz, 1H), 7.11 (td, J = 7.4, 7.0, 1.0 Hz, 1H), 6.56 (d, J = 3.2 Hz, 1H), 6.05 (ddd, J = 17.2, 10.5, 5.6 Hz, 1H), 5.21 (dt, J = 10.5, 1.3 Hz, 1H), 5.18 – 5.06 (m, 2H), 2.48 – 2.16 (m, 4H), 2.05 (s, 3H).

¹³C NMR (125 MHz, CDCl₃): δ 137.2, 136.1, 128.6, 125.0, 121.5, 121.0, 119.6, 116.6, 109.9, 102.0, 56.6, 33.3, 30.7, 15.6.

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{17}NS[M+H^+] = 232.1154$, Found 232.1156.

<u>FTIR (neat)</u>: 2916, 1611, 1511, 1476, 1459, 1424, 1309, 1216, 1014, 989, 927, 740, 667 cm⁻¹.

 $[\alpha]_{D}^{28} = -37.5 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 93%.





(S)-1-(5-methylhex-1-en-3-yl)-1H-indole (6.4e)



Procedures

The allylic acetate (31.2 mg, 0.2 mmol, 100 mol%) and the indole (46.9 mg, 0.4 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 48 hr). The title compound was obtained in 68% yield (29.0 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1).

<u>TLC (SiO</u>) $R_f = 0.63$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.64 (dt, *J* = 7.9, 1.0 Hz, 1H), 7.37 (d, *J* = 8.1 Hz, 1H), 7.23 – 7.15 (m, 2H), 7.10 (td, *J* = 7.4, 1.0 Hz, 1H), 6.55 (d, *J* = 3.1 Hz, 1H), 6.00 (ddd, *J* = 17.3, 10.4, 5.4 Hz, 1H), 5.16 – 5.09 (m, 1H), 5.02 – 4.95 (m, 2H), 2.00 (ddd, *J* = 13.8, 9.4, 5.7 Hz, 1H), 1.79 (ddd, *J* = 14.0, 8.2, 5.9 Hz, 1H), 1.53 – 1.42 (m, 1H), 0.94 (d, *J* = 6.6 Hz, 3H), 0.91 (d, *J* = 6.7 Hz, 3H).

¹³C NMR (125 MHz, CDCl₃): δ = 138.2, 136.0, 128.5, 124.9, 121.3, 120.9, 119.3, 115.6, 109.6, 101.6, 56.0, 43.2, 24.6, 22.9, 22.2.

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{19}N[M+H^+] = 214.1590$, Found 214.1592.

<u>FTIR</u> (neat): 2957, 1459, 1309, 1215, 923, 755, 739, 667 cm⁻¹.

 $[\alpha]_{D}^{28} = +10.3 \ (c \ 1.0, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.





(R)-1-(1-cyclopropylallyl)-1H-indole (6.4f)



Procedures

The allylic acetate (61.7 mg, 0.44 mmol, 100 mol%) and the indole (103.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (64.2 mg, 0.33 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC**</u> (SiO₂) $R_f = 0.66$ (heptanes: isopropyl acetate = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.65$ (ddd, J = 7.8, 1.3, 0.7 Hz, 1H), 7.36 (d, J = 3.2 Hz, 1H), 7.34 – 7.30 (m, 1H), 7.18 (ddd, J = 8.3, 7.0, 1.3 Hz, 1H), 7.11 (ddd, J = 8.0, 7.0, 1.1 Hz, 1H), 6.55 (dd, J = 3.2, 0.8 Hz, 1H), 6.03 (ddd, J = 17.2, 10.4, 5.3 Hz, 1H), 5.20 (ddd, J = 10.4, 1.2 Hz, 1H), 5.12 (ddd, J = 17.2, 1.7, 1.2 Hz, 1H), 4.29 (ddt, J = 8.5, 5.3, 1.7 Hz, 1H), 1.45 (qt, J = 8.2, 4.9 Hz, 1H), 0.86 – 0.75 (m, 1H), 0.62 (dddd, J = 9.1, 8.0, 5.7, 4.7 Hz, 1H), 0.49 (ddt, J = 9.4, 5.8, 4.8 Hz, 1H), 0.38 (ddt, J = 9.4, 5.8, 4.9 Hz, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{109.9, 101.2, 62.6, 14.9, 4.4, 3.6.} = 136.7, 136.0, 128.7, 125.6, 121.2, 120.9, 119.3, 116.4,$

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{15}N[M+H^+] = 198.1283$, Found 198.1279.

<u>FTIR</u> (neat): 3007, 1458, 1308, 1264, 1218, 927, 737 cm⁻¹.

 $[\alpha]_{D}^{28} = +25.5 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> (Two connected chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 92%.





(R)-1-(1-(3,3-difluorocyclobutyl)allyl)-1H-indole (6.4g)



Procedures

The allylic acetate (83.7 mg, 0.44 mmol, 100 mol%) and the indole (103.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 70% yield (76.2 mg, 0.31 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.60$ (heptanes: isopropyl acetate = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.63 (dt, *J* = 7.9, 1.0 Hz, 1H), 7.37 (dd, *J* = 8.3, 1.0 Hz, 1H), 7.21 (ddd, *J* = 8.3, 7.0, 1.3 Hz, 1H), 7.15 – 7.10 (m, 1H), 7.09 (d, *J* = 3.3 Hz, 1H), 6.55 (dd, *J* = 3.3, 0.8 Hz, 1H), 5.95 (ddd, *J* = 17.2, 10.4, 6.0 Hz, 1H), 5.24 (dt, *J* = 10.4, 1.1 Hz, 1H), 5.12 (ddd, *J* = 17.2, 1.5, 0.9 Hz, 1H), 4.84 – 4.73 (m, 1H), 2.89 – 2.79 (m, 2H), 2.59 – 2.42 (m, 2H), 2.29 – 2.15 (m, 1H).

 $\frac{^{13}C \text{ NMR}}{^{11}} (100 \text{ MHz}, \text{ CDCl}_3): \delta = 136.2, 134.6, 128.6, 124.6, 121.8, 121.1, 119.7, 119.4(m), 117.8, 109.6, 102.4, 62.7 (d,$ *J*= 4.1 Hz), 38.9 (dt,*J*= 49.0, 23.0 Hz), 26.8 (dd,*J*= 11.2, 7.9 Hz).

¹⁹**F** NMR (376 MHz, CDCl₃): δ = -83.08 - -84.39 (m, 1F), -92.95 - -94.40 (m, 1F).

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{15}NF_2$ [M+H⁺] = 248.1245, Found 248.1247.

<u>FTIR</u> (neat): 3052, 1458, 1298, 1265, 1171, 901, 737 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +14.0 (*c* 1.0, CHCl₃).

<u>**HPLC</u>** (Chiralcel AD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.</u>







tert-butyl (R)-3-(1-(1H-indol-1-yl)allyl)azetidine-1-carboxylate (6.4h)



Procedures

The allylic acetate (51.1 mg, 0.2 mmol, 100 mol%) and the indole (46.9 mg, 0.4 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 24 hr). The title compound was obtained in 74% yield (46.2 mg, 0.15 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 6:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.39$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.64 (d, *J* = 7.8 Hz, 1H), 7.40 (d, *J* = 8.2 Hz, 1H), 7.22 (ddd, *J* = 8.2, 6.9, 1.2 Hz, 1H), 7.16 – 7.08 (m, 1H), 7.05 (d, *J* = 3.3 Hz, 1H), 6.55 (d, *J* = 3.1 Hz, 1H), 5.93 (ddd, *J* = 16.8, 10.4, 6.0 Hz, 1H), 5.25 (dt, *J* = 10.5, 1.1 Hz, 1H), 5.12 (dt, *J* = 17.1, 1.1 Hz, 1H), 5.06 – 4.98 (m, 1H), 4.16 (t, *J* = 8.6 Hz, 1H), 3.90 (t, *J* = 8.6 Hz, 1H), 3.83 (dd, *J* = 8.9, 5.2 Hz, 1H), 3.55 (dd, *J* = 9.1, 5.3 Hz, 1H), 3.22 (dtt, *J* = 10.6, 8.1, 5.3 Hz, 1H), 1.43 (s, 9H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{17.9}, 109.7, 102.4, 79.7, 61.6, 32.0, 28.3.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 156.2, 136.1, 134.3, 128.6, 124.4, 121.7, 121.1, 119.7, 117.9, 109.7, 102.4, 79.7, 61.6, 32.0, 28.3.$

<u>HRMS</u> (ESI): Calculated for $C_{19}H_{24}N_2O_2$ [M+Na⁺] = 335.1730, Found 335.1738.

<u>FTIR</u> (neat): 2973, 1693, 1405, 1366, 1216, 1143, 931, 741 cm⁻¹.

 $[\alpha]_{D}^{28} = +25.5 \ (c \ 1.0, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 90%.</u>





(R)-1-(1-cyclopentylallyl)-1H-indole (6.4i)



Procedures

The allylic acetate (33.6 mg, 0.2 mmol, 100 mol%) and the indole (46.9 mg, 0.4 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr). The title compound was obtained in 64% yield (28.8 mg, 0.13 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 40:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.63$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.63 (dt, *J* = 7.9, 1.0 Hz, 1H), 7.36 (d, *J* = 8.3 Hz, 1H), 7.22 – 7.16 (m, 2H), 7.09 (ddd, *J* = 7.9, 6.9, 1.0 Hz, 1H), 6.54 (d, *J* = 3.5 Hz, 1H), 6.06 (ddd, *J* = 16.9, 10.4, 6.3 Hz, 1H), 5.12 (dt, *J* = 10.4, 1.3 Hz, 1H), 5.00 (dt, *J* = 17.1, 1.3 Hz, 1H), 4.62 – 4.52 (m, 1H), 2.64 – 2.48 (m, 1H), 1.94 (dtd, *J* = 12.5, 7.6, 4.5 Hz, 1H), 1.71 (dddd, *J* = 16.2, 10.2, 7.5, 3.0 Hz, 1H), 1.67 – 1.55 (m, 2H), 1.55 – 1.49 (m, 2H), 1.43 (dq, *J* = 12.8, 8.4 Hz, 1H), 1.18 – 1.08 (m, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{109.8, 101.5, 63.8, 44.2, 30.8, 30.3, 25.6, 25.1.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 137.1, 136.3, 128.4, 125.4, 121.2, 120.9, 119.3, 116.4, 109.8, 101.5, 63.8, 44.2, 30.8, 30.3, 25.6, 25.1.$

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{19}N[M+H^+] = 226.1590$, Found 226.1588.

<u>FTIR</u> (neat): 2955, 1458, 1215, 987, 924, 754, 738, 667 cm⁻¹.

 $[\alpha]_{D}^{28} = +23.0 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 92%.





(R)-1-(1-cyclohexylallyl)-1H-indole (6.4j)



Procedures

The allylic acetate (80.2 mg, 0.44 mmol, 100 mol%) and the indole (103.1 mg, 0.88 mmol, 200 mol%) were subject to standard reaction conditions (70 °C, 60 hr). The title compound was obtained in 60% yield (63.2 mg, 0.26 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 10% over 20 min).

<u>TLC (SiO</u>₂) $R_f = 0.60$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.63$ (ddd, J = 7.8, 1.3, 0.8 Hz, 1H), 7.37 (dd, J = 8.4, 1.0 Hz, 1H), 7.19 (ddd, J = 8.3, 7.0, 1.3 Hz, 1H), 7.16 (d, J = 3.2 Hz, 1H), 7.09 (ddd, J = 8.0, 7.0, 1.0 Hz, 1H), 6.54 (dd, J = 3.2, 0.9 Hz, 1H), 6.11 (ddd, J = 17.0, 10.3, 7.5 Hz, 1H), 5.23 – 5.08 (m, 2H), 4.56 – 4.47 (m, 1H), 1.99 – 1.89 (m, 2H), 1.80 (ddq, J = 13.2, 3.4, 1.8 Hz, 1H), 1.70 – 1.57 (m, 2H), 1.36 – 1.00 (m, 5H), 0.93 – 0.82 (m, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 136.2, 136.0, 128.3, 125.3, 121.2, 120.9, 119.2, 117.6, 109.8, 101.5, 64.4, 41.9, 30.6, 29.8, 26.3, 26.0, 25.8.

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{21}N[M+H^+] = 240.1747$, Found 240.1752.

FTIR (neat): 2922, 2850, 1509, 1457, 1304, 1263, 1203, 986, 924, 734 cm⁻¹.

 $[\alpha]_{D}^{28} = +40.9 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-3 column, heptanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 90%.</u>




Procedures and Spectral Data for Synthesis of 6.5a, 6.6a

7-(2-aminophenyl)hept-1-en-6-yn-3-ol (6.5a')



Procedures

To a round-bottomed flask charged with 2-Iodoaniline (527.8 mg, 2.41 mmol, 100 mol%), hept-1-en-6-yn-3-ol⁸ (318.6 mg, 2.89 mmol, 120 mol%), copper iodide (50.1 mg, 0.26 mmol, 10 mol%) and Tetrakis(triphenylphosphine)palladium (60.8 mg, 0.053 mmol, 2 mol%) under an argon atmosphere was added triethylamine (4.8 mL, 0.5 M). The reaction was placed in an oil bath at 40 °C for 24 hours, at which point saturated aqueous ammonium chloride (5 mL) were added and the mixture was transferred to a separatory funnel. The organic layer was extracted with ethyl acetate (10 mL × 3) and the combined organic layers were washed with H₂O (10 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1–3:1) to furnish the title compound as a light yellow oil (408.6 mg, 2.03 mmol) in 84% yield.

<u>TLC</u> (SiO₂) $R_f = 0.33$ (dichloromethane: diethyl ether = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.24 (dd, *J* = 7.6, 1.5 Hz, 1H), 7.08 (ddd, *J* = 8.7, 7.4, 1.6 Hz, 1H), 6.70 - 6.63 (m, 2H), 5.91 (ddd, *J* = 17.1, 10.4, 6.0 Hz, 1H), 5.29 (dt, *J* =

17.2, 1.4 Hz, 1H), 5.16 (dt, *J* = 10.5, 1.3 Hz, 1H), 4.37 – 4.31 (m, 1H), 4.18 (br, 1H), 2.68 – 2.52 (m, 2H), 1.84 (qd, *J* = 7.1, 1.9 Hz, 2H).

<u>1³C NMR</u> (125 MHz, CDCl₃): δ = 147.7, 140.4, 132.0, 129.0, 117.8, 115.1, 114.2, 108.6, 94.8, 77.6, 72.0, 35.8, 15.9.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{15}NO[M+H^+] = 202.1226$, Found 202.1233.

<u>FTIR</u> (neat): 3530, 2923, 1613, 1492, 1455, 1307, 1120, 1051, 991, 923, 748 cm⁻¹.



5-(1H-indol-2-yl)pent-1-en-3-yl acetate (6.5a)



Procedures

To a round-bottomed flask charged with **6.5a'** (408.6 mg, 2.03 mmol, 100 mol%) and zinc bromide (685.7 mg, 3.04 mmol, 150 mol%) under an argon atmosphere was added dimethylformamide (20.3 mL, 0.1 M). The reaction was placed in an oil bath at 150 °C for 24 hours. After reaching ambient temperature, the mixture was concentrated under reduced pressure, at which point ethyl acetate (10 mL) and H₂O (10 mL) were added and the mixture was transferred to a separatory funnel. The organic layer was extracted with ethyl acetate (10 mL × 3) and the combined organic layers were washed with H₂O (15 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

То round-bottomed charged with crude substrate a flask the and 4dimethylaminopyridine (12.4 mg, 0.10 mmol, 5 mol%) under an argon atmosphere was added CH₂Cl₂ (10.1 mL, 0.2 M with respect to 5a'), followed by acetic anhydride (0.23 mL, 2.4 mmol, 120 mol%) and triethylamine (0.42 mL, 3.0 mmol, 150 mol%). After 1 hour, saturated aqueous sodium bicarbonate (10 mL) was added and the mixture was transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (10 mL \times 2) and the combined organic layers were washed with 1 N HCl (10 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 8:1-4:1) to furnish the title compound as a light yellow oil (350.4 mg, 1.44 mmol) in 71% yield over 2 steps.

<u>TLC (SiO</u>) $R_f = 0.47$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.20$ (s, 1H), 7.54 (dd, J = 7.7, 1.1 Hz, 1H), 7.32 (dq, J = 8.1, 1.0 Hz, 1H), 7.13 (ddd, J = 8.1, 7.1, 1.3 Hz, 1H), 7.08 (ddd, J = 8.1, 7.1, 1.1 Hz, 1H), 6.26 (dd, J = 2.1, 1.0 Hz, 1H), 5.82 (ddd, J = 17.0, 10.5, 6.3 Hz, 1H), 5.43 – 5.36 (m, 1H), 5.29 (dt, J = 17.2, 1.3 Hz, 1H), 5.22 (dt, J = 10.5, 1.2 Hz, 1H), 2.87 – 2.80 (m, 1H), 2.80 – 2.72 (m, 1H), 2.15 – 2.06 (m, 4H), 2.00 (dddd, J = 13.9, 8.6, 7.0, 5.3 Hz, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{117.2}, 110.5, 100.0, 74.2, 33.9, 23.9, 21.3.} (125 \text{ MHz}, \text{CDCl}_3): \delta = 170.8, 138.5, 136.0 (2C), 128.8, 121.2, 119.8, 119.7, 117.2, 110.5, 100.0, 74.2, 33.9, 23.9, 21.3.}$

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{17}NO_2 [M+Na^+] = 266.1151$, Found 266.1160.

<u>FTIR</u> (neat): 3395, 1721, 1457, 1372, 1244, 1024, 962, 782, 749 cm⁻¹.



(S)-3-vinyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole (6.6a)



Procedures

A pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (130.3 mg, 0.4 mmol, 200 mol%) and (*S*)-Ir-I (11.1 mg, 0.01 mmol, 5 mol%). The tube was purged with argon for 5 minutes. DME (0.5 mL, 0.4 M) was added followed by the allylic acetate **5a** (48.7 mg, 0.2 mmol, 100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 70 °C for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-15:1). The title compound was obtained in 80% yield (29.3 mg, 0.16 mmol) as a colorless oil.

<u>**TLC** (SiO₂</u>) $R_f = 0.34$ (hexanes: ethyl acetate = 20:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.58 - 7.49$ (m, 1H), 7.32 - 7.27 (m, 1H), 7.12 - 7.02 (m, 2H), 6.16 (s, 1H), 6.02 - 5.90 (m, 1H), 5.26 (s, 1H), 5.23 (dd, J = 6.1, 1.1 Hz, 1H), 4.92 - 4.85 (m, 1H), 3.07 (dddd, J = 14.9, 8.6, 6.1, 1.1 Hz, 1H), 2.97 (dddd, J = 15.8, 8.5, 5.9, 1.1 Hz, 1H), 2.85 - 2.75 (m, 1H), 2.36 (ddt, J = 12.5, 8.7, 5.6 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 144.3, 137.6, 133.4, 132.7, 120.3, 120.1, 119.2, 116.8, 109.9, 92.5, 59.6, 35.7, 23.4.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{13}N[M+H^+] = 184.1121$, Found 184.1127.

<u>FTIR</u> (neat): 2952, 1454, 1264, 1217, 989, 924, 740, 702 cm⁻¹.

 $[\alpha]_{D}^{28} = -53.8 \ (c \ 1.0, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OD-3 column, hexanes:i-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 89%.





Procedures and Spectral Data for Deuterium Labelling Experiments

(*S*,*Z*)-1-(oct-1-en-3-yl-1-d)-1H-indole (6.4k)



Procedures

A pressure tube equipped with a magnetic stir bar was charged with the indole (46.9 mg, 0.4 mmol, 200 mol%), cesium carbonate (130.3 mg, 0.4 mmol, 200 mol%) and (*S*)-Ir-I (11.1 mg, 0.01 mmol, 5 mol%). The tube was purged with argon for 5 minutes. DME (0.2 mL, 1.0 M) was added followed by the deuterated allylic acetate $2k^7$ (34.3 mg, 0.2 mmol, 100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 70 °C for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1–15:1). The title compound was obtained in 75% yield (34.2 mg, 0.15 mmol) as a colorless oil.

<u>**TLC** (SiO₂</u>) $R_f = 0.63$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.63 (dt, *J* = 7.8, 1.0 Hz, 1H), 7.35 (d, *J* = 8.3 Hz, 1H), 7.22 - 7.14 (m, 2H), 7.10 (ddd, *J* = 7.9, 6.9, 1.0 Hz, 1H), 6.54 (d, *J* = 3.2 Hz, 1H), 6.01 (ddd, *J* = 10.4, 5.5, 2.6 Hz, 1H), 5.13 (dd, *J* = 10.5, 1.5 Hz, 1H), 4.91 - 4.82 (m, 1H), 2.08 - 1.92 (m, 2H), 1.32 - 1.21 (m, 6H), 0.84 (t, *J* = 6.9 Hz, 3H).

²**H NMR** (92 MHz, CHCl₃): δ = 5.07 (s, 1D).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 137.9, 136.1, 128.6, 124.9, 121.2, 120.9, 119.3, 115.9
- 115.3 (m), 109.8, 101.5, 58.2, 34.2, 31.5, 25.9, 22.5, 14.0.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}DN [M+H^+] = 229.1810$, Found 229.1813.

<u>FTIR</u> (neat): 2931, 1459, 1309, 1214, 750, 668 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +19.8 (*c* 1.0, CHCl₃).

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 88%.</u>







(*R*,*E*)-1-(oct-1-en-3-yl-1-d)-1H-indole (*iso*-6.4k)



Procedures

A pressure tube equipped with a magnetic stir bar was charged with the indole (46.9 mg, 0.4 mmol, 200 mol%), cesium carbonate (130.3 mg, 0.4 mmol, 200 mol%) and (*R*)-Ir-I (11.1 mg, 0.01 mmol, 5 mol%). The tube was purged with argon for 5 minutes. DME (0.2 mL, 1.0 M) was added followed by the deuterated allylic acetate 2k (34.3 mg, 0.2 mmol, 100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 70 °C for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1–15:1). The title compound was obtained in 76% yield (34.7 mg, 0.15 mmol) as a colorless oil.

<u>**TLC** (SiO₂</u>) $R_f = 0.63$ (hexanes: ethyl acetate = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.63 (dt, *J* = 7.8, 1.0 Hz, 1H), 7.35 (d, *J* = 8.2 Hz, 1H), 7.22 – 7.15 (m, 2H), 7.10 (ddd, *J* = 7.9, 6.9, 1.0 Hz, 1H), 6.54 (dd, *J* = 3.2, 0.8 Hz, 1H), 6.01 (dd, *J* = 17.1, 5.8 Hz, 1H), 5.01 (dd, *J* = 17.2, 1.6 Hz, 1H), 4.86 (dtd, *J* = 8.0, 6.1, 1.5 Hz, 1H), 2.10 – 1.91 (m, 2H), 1.32 – 1.23 (m, 6H), 0.84 (t, *J* = 6.9 Hz, 3H). ²**H NMR** (92 MHz, CHCl₃): δ = 5.19 (s, 1D).

<u>1³C NMR</u> (100 MHz, CDCl₃): δ = 137.9, 136.1, 128.5, 124.9, 121.2, 120.9, 119.3, 115.9
- 115.3 (m), 109.8, 101.5, 58.2, 34.2, 31.5, 25.9, 22.5, 14.0.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{20}DN [M+H^+] = 229.1810$, Found 229.1814.

<u>FTIR</u> (neat): 2932, 1459, 1309, 1215, 754, 668 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -22.8 \ (c \ 1.0, \ {\rm CHCl}_3).$

<u>HPLC</u> (Chiralcel AD-H column, hexanes:*i*-PrOH = 99:1, 1.00 mL/min, 254 nm), ee = 95%.







Single Crystal Diffraction Data for 6.3p

Empirical formula	C24 H24 Cl2 F5 I N2		
Formula weight	633.25		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	orthorhombic		
Space group	P 21 21 21		
Unit cell dimensions	a = 11.5320(11) Å	α= 90°.	
	b = 11.5452(11) Å	β= 90°.	
	c = 18.4648(19) Å	$\gamma = 90^{\circ}.$	
Volume	2458.4(4) Å ³		
Z	4		
Density (calculated)	1.711 Mg/m ³		
Absorption coefficient	1.574 mm ⁻¹		
F(000)	1256		
Crystal size	0.342 x 0.146 x 0.107 mm ³		
Theta range for data collection	2.082 to 30.656°.		
Index ranges	-16<=h<=16, -16<=k<=16, -26<=l<=26		
Reflections collected	45429		
Independent reflections	7552 [R(int) = 0.0574]		
Completeness to theta = 25.242°	100.0 %		
Absorption correction	Numerical		
Max. and min. transmission	0.9878 and 0.6732		
Refinement method	Full-matrix least-squares on F ² 650		

Data / restraints / parameters	7552 / 0 / 318
Goodness-of-fit on F ²	1.025
Final R indices [I>2sigma(I)]	R1 = 0.0336, wR2 = 0.0772
R indices (all data)	R1 = 0.0428, wR2 = 0.0796
Absolute structure parameter	-0.003(17)
Extinction coefficient	n/a
Largest diff. peak and hole	1.229 and -0.621 e.Å ⁻³

Figure 1. View of the cation in **1** showing the atom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



Figure 2. View of **1** showing the heteroatom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level. The Cl-H...Cl anion is shown.



Chapter 7: Hydroamination *versus* Allylic Amination in Iridium Catalyzed Reactions of Allylic Acetates with Amines: 1,3-Aminoalcohols *via* Ester-Directed Regioselectivity*

7.1 Introduction

The importance of nitrogen-containing compounds in pharmaceutical and agrochemical research continues to inspire efforts toward the development of metal catalysts for alkene hydroamination.^{1,2} Among late transition metal catalysts,^{1g} those based on rhodium^{3,4} and iridium^{5,6} have shown great promise, however, despite decades of research, significant challenges remain. Many hydroaminations are limited to intramolecular processes.^{4,6} Intermolecular variants often require highly activated alkenes,^{3b,5a-d} are accompanied by oxidative amination side-products^{3c-i,5f} or exploit specialized directing groups to control regioselectivity.^{3j,k} Further, the mechanisms of these processes are obfuscated by the fact that simple brønsted acids – even ammonium salts^{7b,f} – catalyze hydroamination with levels of efficiency equal to or greater than the corresponding metal catalyzed reactions.^{7,8} Here, we report the first hydroaminations of allylic acetates. Specifically, using a neutral dppf-modified iridium catalyst in the presence of Cs_2CO_3 , linear allylic acetates react with primary amines to form products of hydroamination with complete 1,3-regioselectivity. These results are remarkable, as iridium complexes are well-known to catalyze the substitution of allylic acetates with amines to form products of allylic amination (Figure 7.1). 9

^{*}This chapter is based on the published work:

Kim, S. W.; Wurm, T.; Brito, G. A.; Jung, W.-O.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. J. Am. Chem. Soc. 2018, 140, 9087.

Recently, in a collaborative endeavor, we reported that commercially available π allyliridium *C,O*-benzoates catalyze highly enantioselective substitutions of branched allylic acetates with primary amines.¹⁰ In the course of developing this process, significant quantities (30-40% yield) of hydroamination product were formed as single regioisomers in attempted substitutions of linear allylic acetates. Both iridium *C,O*-benzoates and noncyclometallated phosphine-modified iridium complexes derived *in situ* from [Ir(cod)Cl]₂ displayed roughly equivalent efficiencies. In both cases, chiral ligands did not induce asymmetry, so the simpler non-cyclometallated complexes were chosen for further optimization. From a survey of achiral phosphine ligands, the phosphine ligand dppf, 1,1'-bis(diphenylphosphino)ferrocene, was found to be most effective. Hydroamination products were not observed in the absence of metal, phosphine ligand or upon use of monophosphine ligands.



Figure 7.1 Cationic *vs* neutral iridium catalysts promote Tsuji-Trost allylic amination and hydroamination, respectively.

7.2 Reaction Development and Scope

The effect of base was systematically investigated (Table 7.1, entries 1-5). In the presence of $[Ir(cod)Cl]_2$ (2.5 mol%), dppf (5 mol%) and Cs₂CO₃ (200 mol%) at 90 °C in THF (1 M), cinnamyl acetate **7.1a** (100 mol%) and benzyl amine **7.2a** (200 mol%) were converted the hydroamination products **7.3a** and the deactylated compound **7.3a'** in a 2:1 ratio and a combined 66% yield along with trace amounts of cinnamyl alcohol. In an attempt to minimize formation of **7.3a**, lower loadings of Cs₂CO₃ were explored (Table 7.1, entries 1-3), however, although the reaction is heterogeneous, the combined yield hydroamination products **7.3a** and **7.3a'** declined.¹¹ Other bases were explored (Table 7.1, entries 4,5), but Cs₂CO₃ provided superior results. The use of cinnamic esters less prone to saponification were explored (Table 7.1, entries 6-10), but this also led to a decrease in the combined yield of hydroamination products **7.3a** and **7.3a'**. In all experiments, products of acyl transfer to benzyl amine were not observed.

Acute sensitivity to the steric features of the cinnamic ester (cf. Table 7.1, entries 1 and 7) along with the regioselectivity of hydroamination suggested chelation of iridium by the acetate carbonyl and olefin moieties might play an integral role in the mechanism. To perhaps favor chelation of iridium and minimize formation of **7.3a'** the stoichiometry of cinnamyl acetate **7.1a** and benzyl amine **7.2a** was altered (Table 7.1, entries 11-13). Using cinnamyl acetate **7.1a** (300 mol%) and benzyl amine **7.2a** (100 mol%), the hydroamination product **7.3a** was obtained in 76% yield accompanied by only trace quantities of the **7.3a'** (Table 7.1, entry 13). To mitigate solvent evaporation, a number of higher boiling solvents were evaluated and it was found that 1,2-dimethoxyethane (DME), provided **7.3a** with equivalent efficiencies (Table 7.1, entry 14). Finally, although this effect was not pronounced in reactions of cinnamyl acetate **7.1a**, it was also found that in certain cases milled Cs_2CO_3 led to higher and more reproducible yields.¹¹

Table 7.1Selected optimization experiments in the reaction of cinnamyl acetate 7.1awith benzyl amine 7.2a to form hydroamination product 7.3a.^a

		[lr(cod dp)Cl] ₂ (2.5 mol%) pf (5 mol%) B	nNH Ó ⊥ 」	O │ │ R BnNH OH │
Ph 7 1a	7 2a	Base	, Solvent (1 M) Ph	7 32	Ph 7 32
/.1a	1.24			7.Ja	7.54
Entry	7.1a / 7.2a (mol%)	R	Base (mol%)	Solvent	7.3a /7.3a' Yield (%)
1	100 / 200	Me	Cs ₂ CO ₃ (200 mol%)	THF	45 / 21
2	100 / 200	Me	Cs ₂ CO ₃ (50 mol%)	THF	15 / 6
3	100 / 200	Me	Cs ₂ CO ₃ (20 mol%)	THF	trace
4	100 / 200	Me	K ₂ CO ₃ (200 mol%)	THF	trace
5	100 / 200	Me	K ₃ PO ₄ (200 mol%)	THF	22 / 8
6	100 / 200	Et	Cs ₂ CO ₃ (200 mol%)	THF	28 / 9
7	100 / 200	′Pr	Cs ₂ CO ₃ (200 mol%)	THF	12 / trace
8	100 / 200	^t Bu	Cs ₂ CO ₃ (200 mol%)	THF	NR
9	100 / 200	ОМе	Cs ₂ CO ₃ (200 mol%)	THF	NR
10	100 / 200	NMe ₂	Cs ₂ CO ₃ (200 mol%)	THF	NR
11	100 / 100	Me	Cs ₂ CO ₃ (200 mol%)	THF	36 / trace
12	200 / 100	Me	Cs ₂ CO ₃ (200 mol%)	THF	65 / trace
13	300 / 100	Me	Cs ₂ CO ₃ (200 mol%)	THF	76 / trace
14	300 / 100	Me	Cs ₂ CO ₃ (200 mol%)	DME	77 / trace

^aAll reactions were performed on 0.2 mmol scale. All yields are of material isolated by silica gel chromatography. See Supporting In-formation for further experimental details.

To illustrate scope, optimal conditions for hydroamination were applied across a diverse set of reactants (Table 7.2). As demonstrated by the formation of hydroamination products **7.3a-7.3g**, primary benzylic amines are effective partners for hydroamination. Saturated primary amines provide moderate yields of hydroamination product (**7.3h**). A range of aryl-substituted linear allylic acetates also were evaluated (**7.3i-7.3p**). These experiments, which were all conducted using benzyl amine, reveal that both electron rich and electron deficient aryl groups, as well as heteroaryl groups, are tolerated. The formation of adduct **7.3p**, derived from AZD-9496 (a non-steroidal oral estrogen receptor inhibitor),¹² which incorporates an unprotected indole moiety, highlights the functional group tolerance of the present hydroamination method. As illustrated by formation of **7.3q** and **7.3r**, alkyl-substituted allylic acetates provide modest yields of hydroamination

product as single regioisomers. The regioselective formation **7.3r** is noteworthy, as it provides strong evidence of the directing influence of the acetate moiety. Indeed, as cinnamyl methyl ether, *trans-\beta*-methyl styrene and 1-octene do engage in hydroamination under these conditions, it would appear that the acetate moiety not only directs regioselectivity, but is required for hydroamination to proceed.

Table 7.2Regioselective iridium-catalyzed hydroamination of linear allylic
acetates.^a



^aYields of material isolated by silica gel chromatography. Standard conditions: [Ir(cod)Cl]₂ (2.5 mol%), dppf (5 mol%), Cs₂CO₃ (200 mol%), amine (100 mol%), DME (1 M), 90 °C. All yields are of material isolated by silica gel chromatography. See Supporting Information for further experimental details.

7.3 Discussion

To gain insight into the reaction mechanism, mono- and di-deuterated cinnamyl acetates d_1 -7.1a and d_2 -7.1a were exposed to standard hydroamination conditions (Scheme 7.1). To facilitate assignment of relative stereochemistry, the reaction products were converted to the cyclic carbamates d_1 -7.3a and d_2 -7.3a. Axial disposition of the phenyl moiety was corroborated by nOe experiments and is presumably due to nonbonded interactions with the N-benzyl moiety. As determined by ¹H NMR, ²H NMR and HRMS, both compounds d_1 -7.3a and d_2 -7.3a, completely retain deuterium and were each generated as 1.7:1 mixtures of diastereomers. Guided by the results of deuterium labeling, a general catalytic mechanism for iridium catalyzed hydroamination of linear allylic acetates was proposed (Scheme 7.2). Base-induced formation of amidoiridium species I enables entry into the catalytic cycle.¹¹ Association of d_1 -7.1a provides the chelated olefin complex II, which undergoes rapid and reversible inner- or outer-sphere alkene aminoiridation to form the σ -alkyliridium species *cis*-III and *trans*-III, respectively. Equilibration between cis-III and trans-III occurs in advance of turn-over limiting proto-demetalation mediated by benzyl amine to release the product d_1 -7.3a and regenerate amidoiridium species I to close the catalytic cycle. Consistent with this mechanistic interpretation, chiral iridium complexes do not deliver enantiomerically enriched hydroamination product. Additionally, using cis-cinnamyl acetate, low conversion to hydroamination product is observed (15% yield) and recovered cinnamyl acetate is partially isomerized to the trans-isomer (8:1, cis:trans). These data suggest the small quantity of hydroamination product observed in reactions of *cis*-cinnamyl acetate are likely formed by way of the *trans*-isomer and, for *cis*-cinnamyl acetate, allylic strain may prevent acetate-mediated chelation of the olefin by iridium. On the basis of these data, mechanisms involving alkene isomerization appear less plausible.

Scheme 7.1 Deuterium labelling experiments.^a



^aReactants and products characterized by ¹H NMR, ²H NMR and HRMS. See Supporting Information for further experimental details.

Scheme 7.2 General catalytic mechanism for iridium catalyzed hydroamination of linear allylic acetates, as corroborated by deuterium labeling.



7.4 Conclusion

To summarize, allylic acetates are well-known to undergo Tsuji-Trost amination upon exposure to amines in the presence of iridium catalysts with cationic character. Here, using neutral iridium catalysts under basic conditions, we report the first examples of allylic acetates hydroamination. The collective data, including deuterium labeling studies, corroborate a catalytic mechanism involving rapid, reversible acetate-directed aminoiridation with inner sphere/outer-sphere crossover in advance of turn-over limiting amine-mediated proto-demetalation. The present studies establish a new mechanistic pathway for alkene hydroamination under basic conditions, broadening access to 1,3aminoalcohols and related *N*-containing compounds.¹³ More broadly, this study and prior collaborative work from the present authors¹⁰ demonstrate the effectiveness of academicindustrial cooperation for the discovery of useful and robust methods for chemical synthesis.

7.5 Experimental Details

General Information

All reactions were carried out under inert gas atmosphere (nitrogen or argon) unless otherwise indicated. Resealable pressure tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were flame dried followed by cooling in a desiccator or under a stream of inter gas prior to use. All commercial reagents and anhydrous solvents were used as received from vendors (Fischer Scientific, Sigma Aldrich and Combi Blocks) without further purification. [Ir(cod)Cl]₂ and dppf were used as received from Strem Chemicals Inc. Cs₂CO₃ was used as received from Rockwell Lithium. The compounds (4-(pentafluoro- λ^6 -sulfaneyl)phenyl)methanamine¹

was prepared according to literature procedures. Preparative column chromatography employing Silicycle silica gel (40-63 μ m) was performed according to the method of Still² or on a Teledyne Isco Combiflash R_f utilizing Silicycle HP columns using a mobile phase composed of either heptane/isopropyl acetate, hexanes/ethyl acetate or dichloromethane/methanol. Reactions were monitored by a Shimadzu LCMS/UV system with LC-30AD solvent pump, 2020 MS, Sil-30AC autosampler, SPD-M30A UV detector, CTO-20A column oven, using a 2-98% acetonitrile/0.1% formic acid (or 0.001% ammonia) gradient over 2.5 minutes. Analytical thin-layer chromatography (TLC) was carried out using 0.25 mm commercial silica gel plates (Dynamic Absorbents F). Visualization was accomplished with UV light followed by dipping in CAM, *p*-Anisaldehyde (PAA), or KMnO₄ stain solution followed by heating. Specific optical rotations were recorded on an Atago AP-300 automatic polarimeter at the sodium line (589.3 nm) in CHCl₃ Solution concentrations are given in the units of 10-2 g ml⁻¹.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer. Highresolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M+H, M+Na), or a suitable fragment ion. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (400 MHz) or Varian INOVA (500 MHz) spectrometer equipped with a Bruker cryoprobe. Chemical shifts are reported in delta (δ) units, parts per million (ppm) downfield from tetramethylsilane or ppm relative to the center of the singlet at 7.26 ppm for deuteriochloroform. Data reported as multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet). Integration and coupling constants were reported in Hertz (Hz). Carbon-13 nuclear magnetic resonance (¹³C NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (100 MHz) or Varian INOVA (125 MHz) spectrometer and were routinely run with broadband decoupling. Chemical shifts are reported in delta (δ) units, ppm relative to the center of the triplet at 77.16 ppm for deuteriochloroform. Fluorine-19 nuclear magnetic resonance (¹⁹F NMR) spectra were recorded with a Varian Gemini (376 MHz) spectrometer..

Experimental Details and Spectral Data

General procedure for the synthesis of allylic acetates. The allylic acetate **7.1i**, **7.1j**, **7.1k**, **and 7.1o** were prepared by the Horner–Wadsworth–Emmons reaction, DIBAL reduction, and acetylation of the corresponding aldehyde as shown below. The allylic acetate **7.1l** and **7.1r** were prepared by the acetylation of the corresponding allylic alcohol as shown below. The allylic acetates **7.1m**,³ **7.1n**,⁴ and **7.1q**⁵ were prepared according to the published procedures and were identical in all respects to the reported materials.



To a round-bottomed flask charged with sodium hydride (140 mol%, 60% in mineral oil) under an argon atmosphere was added THF (0.3 M), followed by triethyl phosphonoacetate (150 mol%). After 10 minutes, the corresponding aldehyde was added and the mixture was stirred at room temperature overnight. After water was added, the

mixture was transferred to a separatory funnel. The organic layer was extracted with ethyl acetate and the combined organic layers were washed with brine, dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

To a round-bottomed flask charged with the crude substrate under an argon atmosphere was added CH₂Cl₂ (0.3 M). The reaction flask was placed an ice batch. After 10 minutes, diisobutylaluminum hydride (400 mol%, 1.0 M in hexanes) was added slowly and the mixture was stirred at room temperature for 1 hour, at which point saturated aqueous Rochelle salts were added and the reaction was stirred vigorously overnight. After water was added, the mixture was transferred to a separatory funnel. The organic layer was extracted with ethyl acetate and the combined organic layers were washed with brine, dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

То а round-bottomed flask charged with the crude substrate and 4dimethylaminopyridine (5 mol%) under an argon atmosphere was added CH₂Cl₂ (0.3 M), followed by acetic anhydride (150 mol%) and triethylamine (300 mol%). After 1 hour, saturated aqueous sodium bicarbonate was added and the mixture was transferred to a separatory funnel. The organic layer was extracted with CH₂Cl₂ and the combined organic layers were washed with 1 N HCl, dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography give the corresponding allylic acetate over 3 steps.

(E)-3-(4-methoxyphenyl)allyl acetate (7.1i)



The title compound was prepared by the general procedure (Horner–Wadsworth– Emmons reaction, DIBAL reduction, and acetylation of the corresponding aldehyde). The spectral data were consistent with those reported.⁶

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.36 – 7.29 (m, 2H), 6.88 – 6.83 (m, 2H), 6.60 (dd, *J* = 16.0, 1.4 Hz, 1H), 6.15 (dt, *J* = 15.8, 6.6 Hz, 1H), 4.70 (dd, *J* = 6.6, 1.3 Hz, 2H), 3.80 (s, 3H), 2.08 (s, 3H).

<u>**13C NMR**</u> (100 MHz, CDCl₃): δ = 170.9, 159.6, 134.0, 128.9, 127.9, 120.8, 114.0, 65.3, 55.3, 21.0.



(E)-3-(benzo[d][1,3]dioxol-5-yl)allyl acetate (7.1j)



The title compound was prepared by the general procedure (Horner–Wadsworth– Emmons reaction, DIBAL reduction, and acetylation of the corresponding aldehyde). The spectral data were consistent with those reported.⁷

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 6.90$ (d, J = 1.7 Hz, 1H), 6.79 (dd, J = 8.0, 1.7 Hz, 1H), 6.72 (d, J = 8.0 Hz, 1H), 6.52 (dt, J = 15.8, 1.4 Hz, 1H), 6.08 (dt, J = 15.8, 6.6 Hz, 1H), 5.91 (s, 2H), 4.66 (dd, J = 6.6, 1.4 Hz, 2H), 2.06 (s, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 170.6$, 147.9, 147.5, 133.8, 130.5, 121.3, 121.1, 108.1,

105.6, 101.0, 65.0, 20.8.


(E)-3-(3,4-dichlorophenyl)allyl acetate (7.1k)



The title compound was prepared by the general procedure (Horner–Wadsworth– Emmons reaction, DIBAL reduction, and acetylation of the corresponding aldehyde). The spectral data were consistent with those reported.⁸

¹<u>H NMR</u> (400 MHz, CDCl₃): δ = 7.47 (d, *J* = 2.0 Hz, 1H), 7.39 (d, *J* = 8.3 Hz, 1H), 7.20 (dd, *J* = 8.3, 2.1 Hz, 1H), 6.55 (dt, *J* = 16.1, 1.5 Hz, 1H), 6.28 (dt, *J* = 16.0, 6.2 Hz, 1H), 4.72 (dd, *J* = 6.2, 1.4 Hz, 2H), 2.11 (s, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): δ = 170.7, 136.4, 132.8, 131.8, 131.4, 130.5, 128.3, 125.7,

125.4, 64.5, 20.9.





(E)-3-(2,5-difluorophenyl)allyl acetate (7.1l)



The title compound was prepared by the general procedure (acetylation of the corresponding allylic alcohol). The spectral data were consistent with those reported.⁹

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.14$ (ddd, J = 8.9, 5.8, 3.1 Hz, 1H), 7.00 (td, J = 9.4, 4.6 Hz, 1H), 6.96 – 6.86 (m, 1H), 6.75 (dq, J = 16.1, 1.5 Hz, 1H), 6.35 (dt, J = 16.1, 6.1 Hz, 1H), 4.75 (dd, J = 6.2, 1.5 Hz, 2H), 2.12 (s, 3H). ¹³<u>C NMR</u> (100 MHz, CDCl₃): $\delta = 170.7, 158.7$ (dd, J = 242.0, 2.1 Hz), 156.3 (dd, J = 245.8, 2.3 Hz), 127.3 – 127.1 (m), 125.4 (dd, J = 14.8, 8.0 Hz), 125.2 (t, J = 2.6 Hz), 116.8 (dd, J = 25.1, 8.7 Hz), 115.7 (dd, J = 24.3, 8.8 Hz), 113.4 (dd, J = 24.6, 4.0 Hz),

¹⁹**F** NMR (376 MHz, CDCl₃): $\delta = -119.1$ (m, 1F), -123.9 (m, 1F).

64.6, 20.9.





(E)-3-(2-phenylpyrimidin-5-yl)allyl acetate (7.10)



The title compound was prepared by the general procedure (Horner-Wadsworth-

Emmons reaction, DIBAL reduction, and acetylation of the corresponding aldehyde).

<u>**TLC** (SiO</u>₂) $R_f = 0.56$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.80$ (s, 2H), 8.47 - 8.38 (m, 2H), 7.53 - 7.43 (m, 3H), 6.66 - 6.54 (m, 1H), 6.44 (dt, J = 16.1, 5.9 Hz, 1H), 4.77 (dd, J = 5.9, 1.4 Hz, 2H), 2.13 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 170.6, 163.7, 154.8, 137.2, 130.8, 128.6, 128.1, 127.3, 126.8, 126.7, 64.4, 20.9.

HRMS (ESI): Calculated for $C_{15}H_{14}N_2O_2$ [M+H⁺] = 255.1128, found 255.1128.

<u>FTIR</u> (neat): 3036, 2939, 1731,1662, 1582, 1540, 1450, 1432, 1385, 1366, 1328, 1316, 1297, 1250, 1228, 1170, 1077, 1065, 1021, 967, 942, 930, 906, 848, 820, 780, 747, 730, 697, 651 cm⁻¹.

Melting Point : 114–118 °C



(*E*)-3-(3,5-difluoro-4-((1R,3R)-2-(2-fluoro-2-methylpropyl)-3-methyl-2,3,4,9tetrahydro-1H-pyrido[3,4-b]indol-1-yl)phenyl)allyl acetate (7.1p)



Step 1: Borane-tetrahydrofuran complex (2.0 M in tetrahydrofuran, 16 mL, 33 mmol) was added to a solution of (E)-3-[3,5-difluoro-4-[(1R,3R)-2-(2-fluoro-2-methyl-propyl)-3-methyl-1,3,4,9-tetrahydropyrido[3,4-b]indol-1-yl]phenyl]prop-2-enoic acid (3.6 g, 8.1 mmol) in THF (16 mL) at rt. After 3 h, methanol was added and the reaction was allowed to stir for 20 minutes. The reaction was concentrated under reduced pressure, and the crude residue was purified by flash column chromatography (silica, 0% to 40% isopropyl acetate - heptane) to give (E)-3-[3,5-difluoro-4-[(1R,3R)-2-(2-fluoro-2-methyl-propyl)-3-methyl-1,3,4,9-tetrahydropyrido[3,4-b]indol-1-yl]phenyl]prop-2-en-1-ol (1.60 g, 3.73 mmol, 46% Yield).

<u>**TLC** (SiO₂</u>) $R_f = 0.48$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.58 - 7.50$ (m, 2H), 7.27 - 7.18 (m, 1H), 7.17 - 7.06 (m, 2H), 6.91 - 6.83 (m, 2H), 6.56 - 6.47 (m, 1H), 6.35 (dt, J = 15.9, 5.2 Hz, 1H), 5.28 (s, 1H), 4.34 - 4.28 (m, 2H), 3.74 - 3.62 (m, 1H), 3.11 (ddd, J = 15.2, 5.2, 2.0 Hz, 1H), 2.90 (dd, J = 20.0, 15.0 Hz, 1H), 2.64 (ddd, J = 15.1, 4.1, 1.5 Hz, 1H), 2.43 (dd, J = 25.0, 14.9 Hz, 1H), 1.23 (dd, J = 21.6, 18.1 Hz, 6H), 1.13 (d, J = 6.6 Hz, 3H).

 $\frac{13}{C \text{ NMR}} (100 \text{ MHz, CDCl}_3): \delta = 162.3 \text{ (dd, } J = 251.5, 8.3 \text{ Hz}), 139.2 \text{ (t, } J = 10.3 \text{ Hz}), 136.3, 131.9, 131.6, 128.2 \text{ (t, } J = 2.6 \text{ Hz}), 127.7, 121.4, 119.3, 118.2, 116.3 \text{ (t, } J = 15.0 \text{ Hz}), 110.8, 109.5 \text{ (d, } J = 22.1 \text{ Hz}), 108.7, 98.1, 96.5, 63.0, 57.0 \text{ (d, } J = 21.4 \text{ Hz}), 51.2 \text{ (d, } J = 3.7 \text{ Hz}), 50.9, 27.1, 25.2 \text{ (d, } J = 24.9 \text{ Hz}), 24.6 \text{ (d, } J = 24.6 \text{ Hz}).$

¹⁹**F NMR** (376 MHz, CDCl₃): $\delta = -139.4$ (m).

<u>HRMS</u> (ESI): Calculated for $C_{25}H_{28}ON_2F_3$ [M+H⁺] = 429.2148, found 429.2146.

<u>FTIR</u> (neat): 3413, 2979, 1629, 1449, 1372, 1202, 1132, 1021, 908, 731 cm⁻¹.





Step 2: 4-Dimethylaminopyridine (45.6 mg, 0.373 mmol) was added to a solution of (E)-3-[3,5-difluoro-4-[(1R,3R)-2-(2-fluoro-2-methyl-propyl)-3-methyl-1,3,4,9-

tetrahydropyrido[3,4-b]indol-1-yl]phenyl]prop-2-en-1-ol (1.60 g, 3.73 mmol) and acetic anhydride (0.420 mL, 4.48 mmol) in dichloromethane (75 mL) at rt. After 30 min, the reaction was concentrated under reduced pressure and the crude residue was purified by flash column chromatography (silica 0% to 30% isopropyl acetate - heptane) to give [(E)-3-[3,5-difluoro-4-[(1R,3R)-2-(2-fluoro-2-methyl-propyl)-3-methyl-1,3,4,9-

tetrahydropyrido[3,4-b]indol-1-yl]phenyl]allyl] acetate (1.64 g, 3.49 mmol, 93% Yield).

<u>**TLC** (SiO</u>₂) $R_f = 0.61$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.55 - 7.49$ (m, 1H), 7.43 (s, 1H), 7.23 - 7.17 (m, 1H), 7.14 - 7.05 (m, 2H), 6.86 (d, J = 9.9 Hz, 2H), 6.52 (d, J = 16.0 Hz, 1H), 6.28 (dt, J = 15.9, 6.1 Hz, 1H), 5.26 (s, 1H), 4.71 (dd, J = 6.1, 1.4 Hz, 2H), 3.72 - 3.63 (m, 1H), 3.09 (dd, J = 15.2, 5.0, 1.9 Hz, 1H), 2.88 (dd, J = 19.9, 15.0 Hz, 1H), 2.61 (ddd, J = 15.1, 4.1, 1.4 Hz, 1H), 2.39 (dd, J = 25.1, 15.0 Hz, 1H), 1.23 (d, J = 18.2 Hz, 3H), 1.17 (d, J = 18.4 Hz, 3H), 1.10 (d, J = 6.5 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{^{13}C \text{ NMR}} (100 \text{ MHz, CDCl}_3): \delta = 170.7 , 162.3 (dd, J = 251.8, 8.2 \text{ Hz}), 138.6 (t, J = 10.3 \text{ Hz}), 136.3, 131.7, 131.2 (d, J = 2.7 \text{ Hz}), 127.7 , 126.4 , 121.4 , 119.3 , 118.2 , 116.7 (t, J = 14.9 \text{ Hz}), 110.7 , 109.7 (dd, J = 23.8, 3.3 \text{ Hz}), 108.8, 98.0, 96.4, 64.3, 57.1 (d, J = 21.7 \text{ Hz}), 51.2 (d, J = 3.9 \text{ Hz}), 50.9, 27.1, 25.2 (d, J = 24.9 \text{ Hz}), 24.5 (d, J = 24.8 \text{ Hz}), 20.9.$

¹⁹**F NMR** (376 MHz, CDCl₃): $\delta = -139.3$ (m).

<u>HRMS</u> (ESI): Calculated for $C_{27}H_{29}O_2N_2F_3$ [M+H⁺] = 471.2254, found 471.2247.

<u>FTIR</u> (neat): 3401, 2974, 1732, 1629, 1570, 1428, 1231, 1022, 964, 866, 736, 703 cm⁻¹.





(E)-4-(benzyloxy)but-2-en-1-yl acetate (7.1r)



The title compound was prepared by the acetylation of the corresponding allylic alcohol.¹⁰

<u>TLC (SiO</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 4:1).

<u>¹H NMR</u> (400 MHz, CDCl₃): $\delta = 7.38 - 7.24$ (m, 5H), 5.94 - 5.78 (m, 2H), 4.60 - 4.53

(m, 2H), 4.51 (s, 2H), 4.02 (dt, *J* = 3.9, 1.1 Hz, 2H), 2.05 (s, 3H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{69.7}, 64.2, 20.9}$ (100 MHz, CDCl₃): $\delta = 170.6, 138.0, 130.9, 128.3, 127.7, 127.6, 126.5, 72.3, 69.7, 64.2, 20.9.$

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{16}O_3$ [M+NH₄⁺] = 238.1438, found 238.1442.

FTIR (neat): 2856, 1736, 1363, 1231, 1026, 908, 729, 669 cm⁻¹.



Procedures and Spectral Data for Synthesis of OAc-Amino Alcohols 7.3a-7.3r:

Regioselective Ir-catalyzed hydroamination with primary amine nucleophiles



General Procedure

An oven-dried pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (200 mol%), [Ir(cod)Cl]₂ (2.5 mol%), and dppf (5 mol%). The tube was purged with argon for 5 minutes. DME (1.0 M) was added followed by the allylic acetate (300 mol%) and the amine (100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 90 °C and stirred for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography under the noted conditions.

3-(benzylamino)-3-phenylpropyl acetate (7.3a)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 77% yield (43.6 mg, 0.15 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 20:1-10:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.22$ (dichloromethane: diethyl ether = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.40 - 7.19$ (m, 10H), 4.12 (dt, J = 11.1, 6.4 Hz, 1H), 3.97 (dt, J = 11.1, 6.6 Hz, 1H), 3.73 (t, J = 7.0 Hz, 1H), 3.65 (d, J = 13.1 Hz, 1H), 3.53 (d, J = 13.1 Hz, 1H), 2.05 (dq, J = 13.5, 6.7 Hz, 1H), 1.97 (s, 3H), 1.97 - 1.89 (m, 1H), 1.66 (br, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 170.9, 143.2, 140.4, 128.6, 128.3, 128.1, 127.3, 127.1, 126.8, 62.0, 59.4, 51.3, 36.8, 20.9.

<u>**HRMS**</u> (ESI): Calculated for $C_{18}H_{21}NO_2 [M+H^+] = 284.1654$, found 284.1651.

<u>FTIR</u> (neat): 3029, 1735, 1453, 1364, 1235, 1028, 737, 698 cm⁻¹.



3-((4-(pentafluoro-l6-sulfanyl)benzyl)amino)-3-phenylpropyl acetate (7.3b)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (46.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 63% yield (51.6 mg, 0.13 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 1:0–5:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.52$ (dichloromethane: diethyl ether = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.70 - 7.65$ (m, 2H), 7.40 - 7.34 (m, 4H), 7.31 - 7.26 (m, 3H), 4.17 (dt, J = 11.1, 6.5 Hz, 1H), 3.99 (dt, J = 11.1, 6.5 Hz, 1H), 3.73 - 3.64 (m, 2H), 3.58 (d, J = 13.9 Hz, 1H), 2.10 - 2.01 (m, 1H), 1.98 (s, 3H), 1.98 - 1.91 (m, 1H), 1.69 (br, 1H).

 $\frac{^{13}C \text{ NMR}}{^{125.87}(t), 61.8, 59.4, 50.3, 36.9, 20.9}$

¹⁹**F** NMR (376 MHz, CDCl₃): $\delta = 86.52 - 83.38$ (m, 1F), 63.03 (d, J = 149.7 Hz, 4F).

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{20}F_5NO_2S$ [M+H⁺] = 410.1208, found 410.1210.

<u>FTIR</u> (neat): 1736, 1366, 1239, 1039, 833, 701, 670 cm⁻¹.





3-((3-bromobenzyl)amino)-3-phenylpropyl acetate (7.3c)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (37.2 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 65% yield (51.2 mg, 0.13 mmol) as a yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 4:1).

<u>TLC</u> (SiO₂) $R_f = 0.20$ (hexanes: ethyl acetate = 3:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.45 - 7.43$ (m, 1H), 7.38 - 7.33 (m, 3H), 7.29 (dt, J = 7.8, 1.2 Hz, 3H), 7.20 - 7.14 (m, 2H), 4.14 (dt, J = 11.1, 6.4 Hz, 1H), 3.98 (ddd, J = 11.1, 7.0, 6.2 Hz, 1H), 3.71 (t, J = 7.0 Hz, 1H), 3.64 - 3.59 (m, 1H), 3.50 (d, J = 13.5 Hz, 1H), 2.06 (ddd, J = 14.0, 6.9, 6.2 Hz, 1H), 2.00 (s, 3H), 1.98 - 1.90 (m, 1H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 171.0, 142.9, 142.9, 131.1, 130.0, 129.9, 128.7, 128.7, 128.7, 128.7, 128.6, 127.5, 127.1, 126.7, 126.5, 122.5, 61.9, 59.5, 50.7, 36.9, 21.0.$ **HRMS**(ESI): Calculated for C₁₈H₂₀BrNO₂ [M+H⁺] = 362.0750, Found 362.0754.**FTIR**(neat): 3025, 2921, 1735, 1568, 1452, 1364, 1236, 1091, 1068, 1030, 996, 968, 763, 700, 669 cm⁻¹.



3-((benzo[d][1,3]dioxol-5-ylmethyl)amino)-3-phenylpropyl acetate (7.3d)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (30.2 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 74% yield (48.5 mg, 0.15 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 20:1-5:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.19$ (dichloromethane: diethyl ether = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.37 - 7.22$ (m, 5H), 6.78 (d, J = 1.6 Hz, 1H), 6.72 (d, J = 7.9 Hz, 1H), 6.67 (d, J = 7.9 Hz, 1H), 5.90 (s, 2H), 4.11 (dt, J = 11.1, 6.4 Hz, 1H), 3.96 (dt, J = 11.1, 6.7 Hz, 1H), 3.70 (t, J = 7.0 Hz, 1H), 3.54 (d, J = 13.0 Hz, 1H), 3.42 (d, J = 13.0 Hz, 1H), 2.04 (dt, J = 13.7, 6.8 Hz, 1H), 1.97 (s, 3H), 1.91 (dt, J = 13.7, 6.8 Hz, 1H), 1.63 (br, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{121.1, 108.6, 107.9, 100.8, 61.9, 59.2, 51.0, 36.8, 20.8.}$

<u>HRMS</u> (ESI): Calculated for $C_{19}H_{21}NO_4 [M+H^+] = 328.1543$, found 328.1551.

<u>FTIR</u> (neat): 2984, 1733, 1488, 1440, 1365, 1236, 1036, 929, 808, 734, 701 cm⁻¹.



3-((furan-2-ylmethyl)amino)-3-phenylpropyl acetate (7.3e)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (19.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 68% yield (37.1 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 30:1-10:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.33$ (dichloromethane: diethyl ether = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.36 - 7.21$ (m, 6H), 6.27 (dd, J = 3.2, 1.9 Hz, 1H), 6.12 - 6.02 (m, 1H), 4.05 (dt, J = 11.1, 6.2 Hz, 1H), 3.91 (dt, J = 11.1, 6.2 Hz, 1H), 3.70 (t, J = 6.8 Hz, 1H), 3.63 (d, J = 14.5 Hz, 1H), 3.50 (d, J = 14.5 Hz, 1H), 2.05 (td, J = 13.8, 6.3 Hz, 1H), 1.96 (s, 3H), 1.95 - 1.87 (m, 1H), 1.82 (br, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 170.9, 153.7, 142.6, 141.7, 128.5, 127.4, 127.2, 110.0, 106.8, 76.7, 61.9, 59.2, 43.7, 36.6, 20.8.

<u>**HRMS**</u> (ESI): Calculated for $C_{16}H_{19}NO_3 [M+H^+] = 274.1438$, found 274.1445. <u>**FTIR**</u> (neat): 2919, 1735, 1365, 1235, 1036, 918, 735, 700 cm⁻¹.



3-phenyl-3-((thiophen-2-ylmethyl)amino)propyl acetate (7.3f)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (22.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 71% yield (41.1 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 20:1-10:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.52$ (dichloromethane: diethyl ether = 5:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.38 - 7.25$ (m, 5H), 7.18 (dd, J = 5.1, 1.2 Hz, 1H), 6.91 (dd, J = 5.1, 3.4 Hz, 1H), 6.82 (dd, J = 3.4, 1.0 Hz, 1H), 4.11 (dt, J = 11.1, 6.4 Hz, 1H), 3.97 (dt, J = 11.1, 6.6 Hz, 1H), 3.86 – 3.68 (m, 3H), 2.04 (dq, J = 13.5, 6.9 Hz, 1H), 1.97 (s, 3H), 1.93 (dd, J = 13.9, 7.2 Hz, 1H), 1.70 (br, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 171.0, 144.3, 142.8, 128.6, 128.6, 127.4, 127.2, 126.5, 124.7, 124.3, 62.0, 59.1, 45.8, 36.8, 20.9.

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{19}NO_2S$ [M+H⁺] = 290.1209, found 290.1213.

<u>FTIR</u> (neat): 2919, 2849,1735, 1493, 1454, 1387, 1365, 1237, 1110, 1037, 852, 829, 761, 700 cm⁻¹.



3-phenyl-3-((pyridin-3-ylmethyl)amino)propyl acetate (7.3g)



The allylic acetate (105.7 mg, 0.6 mmol, 300 mol%) and the primary amine (21.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 67% yield (38 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 100% over 30 min).

<u>TLC (SiO2</u>) $R_f = 0.30$ (ethyl acetate: methanol = 15:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.52 - 8.45$ (m, 2H), 7.64 (dt, J = 7.7, 2.0 Hz, 1H), 7.42 - 7.33 (m, 2H), 7.33 - 7.24 (m, 3H), 7.23 (dd, J = 7.8, 4.7 Hz, 1H), 4.13 (dt, J =11.1, 6.4 Hz, 1H), 3.97 (dt, J = 11.2, 6.6 Hz, 1H), 3.72 (t, J = 7.0 Hz, 1H), 3.66 (d, J =13.5 Hz, 1H), 3.56 (d, J = 13.5 Hz, 1H), 2.08 (dt, J = 13.7, 6.7 Hz, 1H), 2.05 - 1.89 (m, 1H), 1.98 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 170.8, 149.5, 149.2, 134.8, 128.5, 128.3, 127.4, 123.8, 61.4, 57.2, 51.1, 36.3 20.8.

<u>**HRMS**</u> (ESI): Calculated for $C_{17}H_{21}N_2O_2[M+H^+]=285.1598$, found 285.1595.

<u>FTIR</u> (neat): 220, 1735, 1424, 13365, 1237, 1028, 762, 702 cm⁻¹.



3-((cyclohexylmethyl)amino)-3-phenylpropyl acetate (7.3h)



The cinnamyl acetate (105.7 mg, 0.6 mmol, 300 mol%) and the cyclohexylmethanamine (22.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions (90 °C, 24 hr). The title compound was obtained in 54% yield (31.2 mg, 0.11 mmol) as a yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethylacetate = 3:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.19$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.35 - 7.21$ (m, 7H), 4.10 (dt, J = 11.1, 6.3 Hz, 1H), 3.95 (dt, J = 11.1, 6.8 Hz, 1H), 3.66 (t, J = 6.9 Hz, 1H), 2.25 (qd, J = 11.5, 6.6 Hz, 2H), 2.01 (s, 4H), 1.97 - 1.86 (m, 1H), 1.71 - 1.61 (m, 6H), 1.38 (ddt, J = 11.1, 7.6, 3.4 Hz, 1H), 1.28 - 1.08 (m, 6H), 0.90 - 0.80 (m, 2H).

 $\frac{^{13}C \text{ NMR}}{^{38.2}, 36.9, 31.5, 26.7, 26.1, 20.9.}$ $\delta = 171.0, 143.7, 128.5, 127.1, 127.0, 62.1, 60.6, 54.3, 38.2, 36.9, 31.5, 26.7, 26.1, 20.9.$

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{27}NO_2 [M+H^+] = 290.2115$, Found 290.2119. **FTIR** (neat): 2920, 2850, 1739, 1450, 1356, 1237, 1037, 761, 700 cm⁻¹.



3-(benzylamino)-3-(4-methoxyphenyl)propyl acetate (7.3i)



The allylic acetate (123.7 mg, 0.6 mmol, 300 mol%) and the benzyl amine (21 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions (90 °C, 24 hr). The title compound was obtained in 64% yield (40.1 mg, 0.13 mmol) as a yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethylacetate = 2:1).

<u>TLC (SiO</u>) $R_f = 0.17$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.33 - 7.18$ (m, 11H), 6.90 - 6.85 (m, 2H), 4.08 (dt, *J* = 11.0, 6.3 Hz, 1H), 3.93 (dt, *J* = 11.1, 6.7 Hz, 1H), 3.80 (d, *J* = 0.5 Hz, 3H), 3.70 - 3.60 (m, 2H), 3.50 (d, *J* = 13.1 Hz, 1H), 2.08 - 1.95 (m, 4H), 1.90 (dt, *J* = 14.0, 6.9 Hz, 1H). ¹³<u>C NMR</u> (125 MHz, CDCl₃): $\delta = 171.0$, 158.8, 140.5, 135.1, 128.4, 128. 2, 128.2, 126.9, 114.0, 62.1, 58.8, 55.3, 51.3, 36.9, 21.0.

<u>HRMS</u> (ESI): Calculated for $C_{19}H_{23}NO_3[M+H^+] = 314.1751$, Found 313.1751.

<u>FTIR</u> (neat): 2931, 2836, 1735, 1584, 1511, 1454, 1302, 1243, 1176, 1033, 831, 735, 699 cm⁻¹.


3-(benzo[d][1,3]dioxol-5-yl)-3-(benzylamino)propyl acetate (7.3j)



The allylic acetate (132.1 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions for 48 h. The title compound was obtained in 67% yield (43.9 mg, 0.13 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 25:1-5:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.30$ (dichloromethane: diethyl ether = 3:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.34 - 7.15$ (m, 5H), 6.85 (d, J = 1.6 Hz, 1H), 6.76 (d, J = 7.8 Hz, 1H), 6.72 (dd, J = 7.9, 1.7 Hz, 1H), 5.94 (d, J = 0.8 Hz, 2H), 4.09 (dt, J = 11.1, 6.3 Hz, 1H), 3.93 (dt, J = 11.1, 6.3 Hz, 1H), 3.68 – 3.61 (m, 2H), 3.51 (d, J = 13.2 Hz, 1H), 2.05 – 1.95 (m, 1H), 1.98 (s, 3H), 1.87 (ddt, J = 13.8, 7.5, 6.3 Hz, 1H), 1.59 (br, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{120.6, 108.0, 107.0, 100.9, 61.9, 59.2, 51.2, 36.9, 20.9.}$

<u>HRMS</u> (ESI): Calculated for $C_{19}H_{21}NO_4 [M+H^+] = 328.1543$, found 328.1551.

<u>FTIR</u> (neat): 2917, 1732, 1486, 1365, 1236, 1036, 908, 728, 698 cm⁻¹.



3-(benzylamino)-3-(3,4-dichlorophenyl)propyl acetate (7.3k)



The allylic acetate (147 mg, 0.6 mmol, 300 mol%) and the primary amine (21.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 78% yield (55 mg, 0.16 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 100% over 30 min).

<u>TLC</u> (SiO₂) $R_f = 0.42$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.47 - 7.39$ (m, 2H), 7.34 - 7.28 (m, 2H), 7.24 (td, J = 4.8, 2.8 Hz, 3H), 7.16 (dd, J = 8.2, 2.0 Hz, 1H), 4.13 (dt, J = 11.2, 6.5 Hz, 1H), 3.95 (dt, J = 11.2, 6.5 Hz, 1H), 3.71 (t, J = 6.9 Hz, 1H), 3.64 (d, J = 13.2 Hz, 1H), 3.51 (d, J = 13.2 Hz, 1H), 2.05-1.95 (m, 4H), 1.95 - 1.84 (m, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{128.1, 127.1, 126.6, 61.5, 58.7, 51.4, 36.8, 20.9.}$

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{19}Cl_2NO_2[M+H^+] = 352.0866$, found 352.0858.

<u>FTIR</u> (neat): 1735, 1464, 1364, 1236, 1130, 1028, 824, 737, 699 cm⁻¹.



3-(benzylamino)-3-(2,5-difluorophenyl)propyl acetate (7.3l)



The allylic acetate (127 mg, 0.6 mmol, 300 mol%) and the primary amine (21.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 52% yield (33 mg, 0.10 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Isopropyl Acetate / Heptane = 0% - 100% over 30 min).

<u>TLC (SiO</u>₂) $R_f = 0.61$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.36 - 7.20$ (m, 5H), 7.12 (ddd, J = 8.8, 5.5, 3.1 Hz, 1H), 7.00 (td, J = 9.2, 4.4 Hz, 1H), 6.99 - 6.87 (m, 1H), 4.16 (dt, J = 11.1, 6.4 Hz, 1H), 4.11 - 3.99 (m, 2H), 3.67 (d, J = 13.0 Hz, 1H), 3.56 (d, J = 13.1 Hz, 1H), 2.14 - 2.02 (m, 1H), 1.97 (s, 3H), 2.05 - 1.90 (m, 1H).

 $\frac{^{13}C \text{ NMR}}{^{13}C \text{ NMR}} (100 \text{ MHz}, \text{CDCl}_3): \delta = 170.9, 159.3 \text{ (dd}, J = 210, 2.3 \text{ Hz}, 1\text{C}), 156.9 \text{ (dd}, J = 210, 2.3 \text{ Hz}, 1\text{C}) 140.0, 132.2 \text{ (dd}, J = 15.7, 6.5 \text{ Hz}, 1\text{C}), 128.4, 128.1, 127.1, 116.7 \text{ (dd}, J = 25.5, 8.4, 1\text{C}), 115.1-114.7 \text{ (m}, 2\text{C}), 61.7, 53.3, 51.5, 35.6, 20.8.$

¹⁹F NMR (376 MHz, CDCl₃): δ = -121.9 (dd, J = 2644, 18.5 Hz, 1F), -278.1 (dd, J=2244, 18.5 Hz, 1F).

<u>HRMS</u> (ESI): Calculated for $C_{18}H_{19}F_2NO_2[M+H^+] = 320.1457$, found 320.1452.

<u>FTIR</u> (neat): 2960, 1736, 1489, 1366, 1236, 1177, 1040, 874, 814, 737, 699 cm⁻¹.



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)



3-(benzylamino)-3-(pyridin-3-yl)propyl acetate (7.3m)



The allylic acetate (106 mg, 0.6 mmol, 300 mol%) and the primary amine (21.6 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 81% yield (46 mg, 0.16 mmol) as a light yellow oil after purification by flash column chromatography (4g SiO₂, Methanol / DCM = 0% - 10% over 30 min).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (ethyl acetate: methanol = 15:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.59 - 8.51$ (m, 2H), 7.76 (d, J = 7.8 Hz, 1H), 7.36 - 7.21 (m, 6H), 4.15 (dt, J = 11.3, 6.3 Hz, 1H), 3.94 (dt, J = 11.5, 6.4 Hz, 1H), 3.82 (t, J = 7.0 Hz, 1H), 3.67 (d, J = 13.2 Hz, 1H), 3.54 (d, J = 13.2 Hz, 1H), 2.14 (dt, J = 13.4, 6.9 Hz, 1H), 2.08 - 1.96 (m, 1H), 1.96 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ = 170.8, 149.5, 149.2, 134.8, 128.5, 128.3, 127.4, 123.8, 61.4, 57.2, 51.1, 36.3 20.8.

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{21}N_2O_2$ [M+H⁺] =285.1598, found 285.1598. **<u>FTIR</u>** (neat): 2924, 1735, 1426, 1365, 1237, 1026, 810, 737, 717, 700 cm⁻¹.



3-(benzylamino)-3-(thiophen-2-yl)propyl acetate (7.3n)



The allylic acetate (109.3 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions for 48 h. The title compound was obtained in 71% yield (41.1 mg, 0.14 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 10:1-5:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.40$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.34 - 7.19$ (m, 6H), 6.95 (dd, J = 5.1, 3.4 Hz, 1H), 6.91 (ddd, J = 3.4, 1.3, 0.5 Hz, 1H), 4.16 (dt, J = 11.1, 6.2 Hz, 1H), 4.07 – 3.96 (m, 2H), 3.77 (d, J = 13.2 Hz, 1H), 3.61 (d, J = 13.1 Hz, 1H), 2.11 (dtd, J = 14.0, 7.0, 6.1 Hz, 1H), 2.02 (dt, J = 7.5, 6.2 Hz, 1H), 1.98 (s, 3H), 1.63 (s, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{124.2}, 61.8, 54.8, 51.1, 37.4, 20.9}$

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{19}NO_2S[M+H^+] = 290.1209$, found 290.1211.

FTIR (neat): 2917, 1735, 1365, 1235, 1036, 907, 731, 697 cm⁻¹.



3-(benzylamino)-3-(2-phenylpyrimidin-5-yl)propyl acetate (7.30)



The allylic acetate (152.1 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions with Cs_2CO_3 (300 mol%). The title compound was obtained in 52% yield (37.6 mg, 0.10 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, toluene: ethyl acetate = 10:1–3:1).

<u>TLC (SiO</u>) $R_f = 0.55$ (toluene: ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 8.77$ (s, 2H), 8.49 – 8.40 (m, 3H), 7.50 (dd, J = 5.2, 2.1 Hz, 4H), 7.35 – 7.23 (m, 6H), 4.28 – 4.18 (m, 1H), 4.08 – 3.98 (m, 1H), 3.83 (t, J = 6.9 Hz, 1H), 3.71 (d, J = 13.2 Hz, 1H), 3.58 (d, J = 13.2 Hz, 1H), 2.19 – 2.08 (m, 1H), 2.03 (dd, J = 12.7, 7.0 Hz, 1H), 1.98 (s, 3H), 1.73 (br, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{128.6, 128.1, 128.1, 127.3, 61.3, 55.3, 51.4, 36.5, 20.9.}$

<u>HRMS</u> (ESI): Calculated for $C_{22}H_{23}N_3O_2$ [M+H⁺] = 362.1863, found 362.1867.

<u>FTIR</u> (neat): 3026, 2926, 1737, 1583, 1543, 1494, 1428, 1322, 1240, 1172, 1025, 927, 749, 694 cm⁻¹.



(benzylamino)-3-(3,5-difluoro-4-((1R,3R)-2-(2-fluoro-2-methylpropyl)-3-methyl-2,3,4,9-tetrahydro-1H-pyrido[3,4-b]indol-1-yl)phenyl)propyl acetate (7.3p)



The allylic acetate (282 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions with Cs_2CO_3 (300 mol%). The title compound was obtained in 53% yield (61 mg, 0.11 mmol) as a light yellow oil after purification by flash column chromatography (12g SiO₂, Isopropyl Acetate / Heptane = 0% - 100% over 30 min).

<u>TLC (SiO</u>₂) $R_f = 0.30$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.58 - 7.49$ (m, 2H), 7.37 - 7.30 (m, 2H), 7.30 - 7.21 (m, 3H), 7.17 - 7.05 (m, 2H), 6.91 - 6.83 (m, 2H), 5.31 (s, 1H), 4.19 - 4.06 (m, 1H), 3.98 - 3.86 (m, 1H), 3.76 - 3.60 (m, 3H), 3.59 - 3.51 (m, 1H), 3.14 - 3.04 (m, 1H), 2.95 - 2.79 (m, 1H), 2.68 - 2.57 (m, 1H), 2.52 - 2.35 (m, 1H), 2.08 - 1.89 (m, 5H), 1.32 - 1.06 (m, 9H).

13C NMR (100 MHz, CDCl₃): δ = 170.9, 162.4 (dd, *J* = 252.4, 7.8 Hz), 136.3, 131.9, 128.6, 128.3, 127.7, 127.4, 121.4, 119.3, 118.2, 110.8, 110.5, 108.8, 97.9, 96.2, 61.3, 58.8 (d, *J* = 8.0 Hz), 57.1 (d, *J* = 22.1 Hz), 51.6 – 50.8 (m, 2C), 36.2, 29.7, 27.0, 25.1 (d, *J* = 24.7 Hz), 24.7 (d, *J* = 24.9 Hz), 22.6, 21.8, 20.9.

<u>**19**F NMR</u> (376 MHz, CDCl₃): $\delta = -139.7$ (m).

<u>HRMS</u> (ESI): Calculated for $C_{34}H_{39}N_3O_2F_3[M+H^+] = 578.2989$, found 578.2984.

FTIR (neat): 2973, 2926, 1729, 1631, 1578, 1433, 1367, 1239, 1019, 908, 732 cm⁻¹.







3-(benzylamino)butyl acetate (7.3q)



The allylic acetate (68.5 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions. The title compound was obtained in 30% yield (13.3 mg, 0.06 mmol) as a light yellow oil after purification by flash column chrmmatography (SiO₂, hexanes: ethyl acetate = 5:1-1:1).

<u>TLC (SiO</u>) $R_f = 0.24$ (ethyl acetate: methanol = 10:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.33 - 7.18$ (m, 5H), 4.19 (dt, J = 11.1, 6.6 Hz, 1H), 4.11 (dt, J = 11.1, 6.7 Hz, 1H), 3.82 (d, J = 13.0 Hz, 1H), 3.72 (d, J = 13.0 Hz, 1H), 2.78 (h, J = 6.3 Hz, 1H), 1.99 (s, 3H), 1.83 - 1.72 (m, 1H), 1.71 - 1.61 (m, 1H), 1.40 (br, 1H), 1.11 (d, J = 6.3 Hz, 3H).

 $\frac{^{13}C \text{ NMR}}{35.7, 21.0, 20.4.} (100 \text{ MHz}, \text{CDCl}_3): \delta = 171.1, 140.6, 128.3, 128.1, 126.8, 62.1, 51.2, 49.7,$

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{19}NO_2[M+H^+] = 222.1489$, found 222.1495.

<u>FTIR</u> (neat): 2962, 1734, 1453, 1365, 1238, 1045, 731, 698 cm⁻¹.



3-(benzylamino)-4-(benzyloxy)butyl acetate (7.3r)



The allylic acetate (132.2 mg, 0.6 mmol, 300 mol%) and the primary amine (21.4 mg, 0.2 mmol, 100 mol%) were subject to standard reaction conditions with Cs_2CO_3 (300 mol%). The title compound was obtained in 53% yield (34.7 mg, 0.11 mmol) as a light yellow oil after purification by flash column chromatography (SiO₂, hexanes: ethyl acetate = 5:1–2:1).

<u>TLC (SiO</u>₂) $R_f = 0.30$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.37 - 7.19$ (m, 10H), 4.50 (s, 2H), 4.17 (qt, J = 11.0, 6.6 Hz, 2H), 3.77 (d, J = 1.5 Hz, 2H), 3.54 (dd, J = 9.4, 4.3 Hz, 1H), 3.41 (dd, J = 9.4, 5.8 Hz, 1H), 2.88 (qd, J = 6.2, 4.3 Hz, 1H), 1.98 (s, 3H), 1.80 (q, J = 6.6 Hz, 2H), 1.70 (s, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{12}} (100 \text{ MHz, CDCl}_3): \delta = 171.0, 140.5, 138.2, 128.4, 128.3, 128.1, 127.6, 127.6, 126.8, 73.2, 71.7, 62.0, 53.9, 51.1, 30.9, 21.0.$

<u>HRMS</u> (ESI): Calculated for $C_{20}H_{25}NO_3 [M+H^+] = 328.1907$, found 328.1907.

<u>FTIR</u> (neat): 1735, 1453, 1364, 1238, 1093, 1027, 734, 697 cm⁻¹.



Procedures and Spectral Data for Deuterium Labelling Experiments

(*E*)-3-phenylallyl-2-d acetate (d_1 -7.1a)



To a flame dried round-bottomed flask charged with LiAlD₄ (210 mg, 5 mmol, 100 mol%) under an argon atmosphere was added THF (15 mL, 0.33 M with respect to propargylic alcohol). The reaction vessel was placed in an ice batch. After 5 minutes a solution of 3-phenyl-2-propyn-1-ol (660 mg, 5 mmol, 100 mol%) in dry THF (5 mL, 1.0 M with respect to propargylic alcohol) was added slowly and the mixture was stirred at room temperature for 3 hours. After the reaction vessel was placed in an ice batch, water (1 mL), NaOH (1 mL, 10% aqueous solution) and water (3 mL) were added to the reaction mixture. After 10 minutes MgSO₄ was added, the reaction mixture was filtered (celite) with the aid of CH₂Cl₂ (10 mL) and the filtrate was concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

То a flask charged with 4round-bottomed the crude substrate and dimethylaminopyridine (30 mg, 0.25 mmol, 5 mol%) under an argon atmosphere was added CH₂Cl₂ (12.5 mL, 0.4 M), followed by acetic anhydride (0.52 mL, 5.5 mmol, 110 mol%) and triethylamine (1.0 mL, 7.5 mmol, 150 mol%). After 1 hour, saturated aqueous sodium bicarbonate (10 mL) was added and the mixture was transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (10 mL \times 2) and the combined organic layers were washed with 1 N HCl (30 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1) to furnish the title compound as a colorless oil (790 mg, 4.46 mmol) in 89% yield.

<u>TLC (SiO</u>₂) $R_f = 0.38$ (hexanes: ethyl acetate = 10:1).

<u>**H NMR**</u> (400 MHz, CDCl₃): $\delta = 7.40 - 7.22$ (m, 5H), 6.63 (t, J = 2.1 Hz, 1H), 4.72 (s,

2H), 2.09 (s, 3H).

²**H NMR** (92 MHz, CDCl₃) δ = 6.34 (s, 1D)

 $\frac{^{13}C \text{ NMR}}{64.9, 20.9}$ (100 MHz, CDCl₃): $\delta = 170.8, 136.1, 134.1, 128.5, 128.0, 126.5, 122.8$ (t),

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{11}DO_2[M+Na^+] = 200.0792$, found 200.0794.

FTIR (neat): 3025, 1733, 1379, 1225, 1021, 922, 764, 699 cm⁻¹.





(*E*)-3-phenylallyl-2,3-d₂ acetate (d_2 -7.1a)



To a flame dried round-bottomed flask charged with LiAlD₄ (210 mg, 5 mmol, 100 mol%) under an argon atmosphere was added THF (15 mL, 0.33 M with respect to propargylic alcohol). The reaction vessel was placed in an ice batch. After 5 minutes a solution of 3-phenyl-2-propyn-1-ol (660 mg, 5 mmol, 100 mol%) in dry THF (5 mL, 1.0 M with respect to propargylic alcohol) was added slowly and the mixture was stirred at room temperature for 3 hours. After the reaction vessel was placed in an ice batch, D₂O (1 mL), NaOH (1 mL, 10% aqueous solution) and water (3 mL) were added to the reaction mixture. After 10 minutes MgSO₄ was added, the reaction mixture was filtered (celite) with the aid of CH₂Cl₂ (10 mL) and the filtrate was concentrated under reduced pressure. The resulting oily residue was subjected to the next step without further purification.

То a round-bottomed flask charged with the crude substrate and 4dimethylaminopyridine (30 mg, 0.25 mmol, 5 mol%) under an argon atmosphere was added CH₂Cl₂ (12.5 mL, 0.4 M), followed by acetic anhydride (0.52 mL, 5.5 mmol, 110 mol%) and triethylamine (1.0 mL, 7.5 mmol, 150 mol%). After 1 hour, saturated aqueous sodium bicarbonate (10 mL) was added and the mixture was transferred to a separatory funnel. The organic layer was extracted with CH_2Cl_2 (10 mL \times 2) and the combined organic layers were washed with 1 N HCl (30 mL), dried (MgSO₄), filtered and concentrated under reduced pressure. The resulting oily residue was subjected to flash column chromatography (SiO₂, hexanes: ethyl acetate = 30:1) to furnish the title compound as a colorless oil (802 mg, 4.50 mmol) in 90% yield.

<u>**TLC** (SiO₂</u>) $R_f = 0.38$ (hexanes: ethyl acetate = 10:1).

¹**H NMR** (400 MHz, CDCl₃): δ = 7.40 – 7.22 (m, 5H), 4.71 (s, 2H), 2.09 (s, 3H).

²**H NMR** (92 MHz, CHCl₃) δ = 6.70 (s, 1D), 6.33 (s, 1D).

<u>**13C NMR**</u> (100 MHz, CDCl₃): $\delta = 170.8$, 136.0, 133.7 (m), 128.5, 128.0, 126.5, 122.7 (m), 64.9, 20.9.

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{10}D_2O_2$ [M+Na⁺] = 201.0855, found 201.0849.

<u>FTIR</u> (neat): 3025, 1733, 1379, 1225, 1021, 922, 751, 700 cm⁻¹.





General procedure for the deuterated cyclic carbamate.



An oven-dried pressure tube equipped with a magnetic stir bar was charged with cesium carbonate (200 mol%), [Ir(cod)Cl]₂ (2.5 mol%), and dppf (5 mol%). The tube was purged with argon for 5 minutes. DME (1.0 M) was added followed by the deuterated cinnamyl acetate (300 mol%) and the amine (100 mol%). The tube was sealed with a PTFE lined cap and was placed in an oil bath at 90 °C and stirred for 24 hours. After reaching ambient temperature, the crude reaction mixture was directly subjected to flash column chromatography.

To a round-bottomed flask charged with the purified substrate under an argon atmosphere was added MeOH (0.2 M), followed by potassium carbonate (150 mol%). After 3 hours, the reaction mixture was filtered (celite) with the aid of CH₂Cl₂ (5 mL) and the filtrate was concentrated under reduced pressure. The crude reaction mixture was subjected to the next step without further purification

To a round-bottomed flask charged with the crude substrate under an argon atmosphere was added toluene (1.0 M), followed by carbonyldiimidazole (150 mol%). The reaction flask was placed an oil bath at 40 °C and stirred for 24 hours. After reaching ambient temperature, the reaction mixture was concentrated under reduced pressure and the crude reaction mixture was directly subjected to flash column chromatography.

3-benzyl-4-phenyl-1,3-oxazinan-2-one (*d*₀**-7.3a**)



The title compound was prepared by the general procedure.

<u>**TLC** (SiO₂</u>) $R_f = 0.24$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.46 - 7.27$ (m, 6H), 7.24 (dt, J = 8.8, 3.1 Hz, 4H), 5.38 (d, J = 15.0 Hz, 1H), 4.50 (dd, J = 5.9, 4.0 Hz, 1H), 4.29 - 4.17 (m, 2H), 3.63 (d, J = 15.1 Hz, 1H), 2.37 (dddd, J = 14.5, 10.4, 6.0, 4.2 Hz, 1H), 1.96 (dtd, J = 14.1, 4.1, 2.9 Hz, 1H).

¹³C NMR (100 MHz, CDCl₃): δ 154.4, 139.8, 136.7, 129.0, 128.6, 128.2, 128.1, 127.6, 126.5, 62.9, 56.8, 50.2, 30.6.

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{17}NO_2 [M+H^+] = 268.1332$, found 268.1339.

<u>FTIR</u> (neat): 3028, 1683, 1420, 1302, 1215, 1120, 1077, 909, 729, 700 cm⁻¹.





(4*R*,5*S*)-3-benzyl-4-phenyl-1,3-oxazinan-2-one-5-d / (4*R*,5*R*)-3-benzyl-4-phenyl-1,3oxazinan-2-one-5-d (*d*₁-7.3a)



The title compound was prepared by the general procedure.

<u>**TLC** (SiO₂</u>) $R_f = 0.24$ (hexanes: ethyl acetate = 2:1).

¹**H NMR** (400 MHz, CDCl₃): $\delta = 7.42 - 7.36$ (m, 2H), 7.35 - 7.23 (m, 4H), 7.19 (ddd, J = 7.4, 4.0, 1.6 Hz, 4H), 5.34 (d, J = 15.1 Hz, 1H), 4.45 (d, J = 4.7 Hz, 1H), 4.26 - 4.10 (m, 2H), 3.59 (d, J = 15.1 Hz, 1H), 2.37 - 2.26 (m, 0.42H), 1.95 - 1.86 (m, 0.66H). ²**H NMR** (92 MHz, CHCl₃) $\delta = 2.35$ (s, 0.63H), 1.94 (s, 0.37H). ¹³**C NMR** (100 MHz, CDCl₃): $\delta = 154.4, 139.7, 136.7, 129.0, 128.6, 128.1, 128.0, 127.6, 126.4, 62.8 (d, <math>J = 2.7$ Hz), 56.7 (d, J = 3.3 Hz), 50.1 (d, J = 1.7 Hz), 30.2 (m). **HRMS** (ESI): Calculated for C₁₇H₁₆DNO₂ [M+H⁺] = 269.1395, found 269.1396. **FTIR** (neat): 3028, 1684, 1423, 1247, 1215, 1113, 1076, 729, 700 cm⁻¹.





(4*R*,5*S*)-3-benzyl-4-phenyl-1,3-oxazinan-2-one-4,5-d₂ / (4*R*,5*R*)-3-benzyl-4-phenyl-1,3-oxazinan-2-one-4,5-d₂ (*d*₂-7.3a)



The title compound was prepared by the general procedure.

<u>**TLC** (SiO₂</u>) $R_f = 0.24$ (hexanes: ethyl acetate = 2:1).

¹<u>H NMR</u> (400 MHz, CDCl₃): $\delta = 7.46 - 7.27$ (m, 6H), 7.21 (ddd, J = 7.2, 4.0, 1.6 Hz, 4H), 5.36 (d, J = 15.1 Hz, 1H), 4.27 - 4.14 (m, 2H), 3.60 (d, J = 15.1 Hz, 1H), 2.38 - 2.28 (m, 0.43H) 1.92 (ddd, J = 14.2, 4.3, 2.8 Hz, 0.66H). ²<u>H NMR</u> (92 MHz, CHCl₃) $\delta = 4.46$ (s, 1H), 2.34 (s, 0.57H), 1.92 (s, 0.33H)

¹³C NMR (100 MHz, CDCl₃): δ = 154.4, 139.7, 136.7, 129.0, 128.6, 128.1, 128.0, 127.6,

126.4, 62.8, 56.4 (m), 50.1 (d, *J* = 1.7 Hz), 30.2 (dd, *J* = 37.4, 17.2 Hz).

<u>HRMS</u> (ESI): Calculated for $C_{17}H_{15}D_2NO_2 [M+H^+] = 270.1459$, found 270.1458.

<u>FTIR</u> (neat): 3028, 1683, 1417, 1302, 1215, 1123, 1077, 909, 727, 700 cm⁻¹.




Chapter 8: Inversion of Enantioselectivity in Allene Gas *versus* Allyl Acetate Reductive Aldehyde Allylation Guided by Metal-Centered Stereogenicity: An Experimental and Computational Study*

8.1 Introduction

In connection with longstanding efforts to use abundant chemical feedstocks as pronucleophiles in metal-catalyzed carbonyl reductive coupling,^{1f} a diverse suite of enantioselective C-C couplings was developed in our laboratory.^{1,2} A distinguishing feature of these processes resides in the use of inexpensive terminal reductants (e.g. H₂, 2-PrOH) or, more ideally, dual use of alcohols as carbonyl proelectrophiles and reductants in carbonyl addition *via* hydrogen auto-transfer.¹ Given the high occurrence of allenes as constituents in the C3, C4 and C5 petroleum cracking fractions (Figure 8.1), these patterns of reactivity were applied to the first (2007) allene-carbonyl reductive couplings to furnish homoallylic alcohols (Figure 8.2).³



Figure 8.1 Allene feedstocks formed in petroleum cracking.

Kim, S. W.; Meyer, C. C.; Mai, B. K.; Liu, P.; Krische, M. J. ACS Catal. 2019, 9, 9158.

^{*}This chapter is based on the published work:

Seminal Report of Allenes as Allyl Donors in Carbonyl Reductive Coupling Krische 2007 (ref. 3)



Seminal Report of Enantioselective Allene-Carbonyl Reductive Coupling Krische 2009 (ref. 4a)



Figure 8.2 Milestones in metal-catalyzed allene-carbonyl reductive coupling and related hydrogen auto-transfer processes.

Using allene, methylallene and dimethylallene as pronucleophiles, products of carbonyl allylation, crotylation and *tert*-prenylation were obtained in reactions conducted from either the alcohol or aldehyde oxidation level.³ Shortly thereafter (2009), we reported the first enantioselective reactions of this type using dimethyl allene.^{4a}

Following this and other catalytic enantioselective allene-carbonyl additions developed in our laboratory,^{4,5} Buchwald reported an allene-ketone reductive coupling mediated by $(MeO)_2MeSiH.^6$ In this account, we report the first enantioselective allene-aldehyde reductive couplings.⁷ Additionally, we demonstrate that using the same antipode of chiral ligand, (*S*)-tol-BINAP, an inversion of enantioselectivity is observed for carbonyl allylations that employ gaseous allene vs allyl acetate as pronucleophile solely in response to stereogenicity at iridium.^{8,9} Experimental and computational studies corroborate intervention of diastereomeric π -allyliridium-*C*,*O*-benzoate complexes, which arise *via* allene hydrometalation (from a pentacoordinate iridium hydride) vs ionization of allyl acetate (from a square planar iridium species).

8.2 Reaction Development and Scope

An initial series of experiments were conducted in which a sealed reaction vessel back-filled with gaseous allene **8.1a** and charged with aldehyde **8.2a** (100 mol%), 2-propanol (200 mol%), THF (0.4 M) and an iridium catalyst (5 mol%) were heated to 60 °C (Table 1) for 24 hours. It was determined that the (*S*)-tol-BINAP-modified iridium catalyst, Ir-V, delivered the desired product **8.3a** with the highest levels of enantioselectivity as the (*R*)-enantiomer, which is the opposite enantiomer observed in corresponding carbonyl allylations mediated by allyl acetate **8.1b**.¹⁰ This surprising result compelled us to evaluate the diastereomeric composition of the iridium catalyst Ir-V, which is easily accomplished *via* LCMS due to the chromatographic stability of π -allyliridium-*C*, *O*-benzoates (Figure 8.3). The catalyst prepared from allyl acetate is enriched in diastereomer D, which, as indicated by experiment and computation, is the thermodynamically most stable isomer. In contrast, the catalyst recovered from the reaction mixture using allene or prepared from allene itself is enriched in diastereomer C.

It was posited that use of Ir-V derived from allene would improve enantioselectivity in the allene-mediated allylation of aldehyde 8.2a. Indeed, an increase from 86% to 92% ee in the formation of homoallylic alcohol **8.3a** was observed (Table 8.1). Using catalyst (S)-Ir-V derived from allene, the scope of the allene-mediated reductive aldehyde allylation mediated by 2-propanol was explored (Table 2). Diverse aryl aldehydes 8.2a-8.2i and heteroaryl aldehydes 8.2j-8.2p were converted to the corresponding homoallylic alcohols 8.3a-8.3p in good yield with uniformly high levels of enantioselectivity. As illustrated by the formation of adduct 8.3d, due to the mild reaction conditions, sensitive functional groups such as pinacol boronates are tolerated. The formation of 8.31, which incorporates an unprotected indole nitrogen, also is notable. The α,β -unsaturated aldehyde 8.2q, as well as linear and branched aliphatic aldehydes 8.2r and 8.2s also participate in highly enantioselective allylation. Allene-mediated allylation of 8.2a mediated by d_8 -2-propanol delivers deuterio-8.3a (eq. 8.1).¹¹ The pattern of deuterium incorporation corroborates reversible allene hydrometalation with incomplete regiocontrol. The relatively low levels of deuterium incorporation are attributed to reversibility of the hydrometalation event and H/D-exchange with adventitious water.¹²

Table 8.1Selected optimization experiments in the enantioselective reductive
coupling of allene 8.1a with aldehyde 8.2a and divergent
enantioselectivity observed upon use of allyl acetate vs allene
pronucleophiles.^a



^aYields are of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. ^bPhMe (0.4 M). ^cPhMe (0.1 M). ^d(*S*)-Ir-V derived from allene. See Supporting Information for experimental details.

Figure 8.3 Diastereomeric composition of the (*S*)-Ir-V and calculated thermodynamic stabilities.



Table 8.2Enantioselective iridium-catalyzed reductive coupling of gaseous allene8.1a with aldehydes 8.2a-8.2s mediated by 2-propanol.^a



^aYields of material isolated by silica gel chromatography. The catalyst (S)-Ir-V prepared from allene was used. Enantioselectivities were determined by chiral stationary phase HPLC analysis. See Supporting Information for further experimental details.



8.3 Discussion

The experimental data suggest that the observed divergence in enantioselectivity in reactions of allene vs allyl acetate is due to intervention of diastereomeric π allyliridium-*C*,*O*-benzoate complexes C and D, which arise *via* allene hydrometalation (from an pentacoordinate iridium hydride) vs ionization of allyl acetate (from a square planar iridium species), respectively (Figure 8.4).^{13,14}



Figure 8.4 Hypothesis for enantiodivergence in aldehyde allylations mediated by gaseous allene vs allyl acetate.

To challenge the veracity of our hypothesis, we then turned our efforts to density functional theory (DFT) calculations.¹⁵ The reaction pathways to form the π -allyliridium complexes under the different experimental conditions were first computed. When allyl acetate pronucleophile is used, the π -allyliridium is formed *via* coordination of allyl acetate to the *C*,*O*-benzoate complex 4 followed by ionization (TS1, Figure 8.5A). Although the common intermediate 4 can potentially lead to all four diastereomeric π -allyl complexes, formation of D is kinetically and thermodynamically favored (see Figure

S2 for less favorable pathways to A and B). The relatively low barrier of TS1D suggests a reversible ionization process. The thermodynamically more stable complex D is operative in the subsequent aldehyde addition step (vide infra). On the other hand, π allyliridium complexes derived from allene are formed via hydrometalation from a square pyramidal iridium hydride (6C or 6D, Figure 8.5B). DFT calculations suggest 6C is thermodynamically more stable than 6D, and subsequent allene coordination to form 7C and hydrometalation via TS3C are both very facile. Therefore, π -allyl complex C is formed preferentially from allene hydrometalation. The calculated transition states for allylation of aldehyde 8.2a by the diastereometric complexes C and D provide further insight into the origins of enantiodivergence. Complexes C and D are supported by the same antipode of the (S)-tol-BINAP ligand, but their stereogenicity at iridium is opposite. Therefore, examination of the transition states for carbonyl addition will reveal whether enantioselectivity is influenced more by the chirality of the metal center¹⁶ or the bisphosphine ligand. The aldehyde addition with C and D occurs by way of Zimmerman-Traxler-type transition states that place the Ar group of the aldehyde in a pseudoequatorial position (Figure 8.6).¹⁷ In the reaction with D, addition to the (Si)-face of the aldehyde (TS2D-S) to form (S)-8.3a is 2.2 kcal/mol more favorable than the (Re)-face addition (TS2D-R) to form (R)-8.3a. TS2D-R is destabilized by the 1,3-diaxial interactions between the aldehyde hydrogen and the benzene ring of the benzoate, which is co-planar with the aldehyde. In TS2D-S, the axial aldehyde hydrogen and the benzoate are on opposite faces of the chair, which relieves steric repulsion. In the allylation transition states with complex C (Figure 8.6B), because the stereogenicity at iridium is inverted, the (S)-selective transition state (TS2C-S) is now destabilized by the 1,3-diaxial interactions be-tween the aldehyde hydrogen and the benzoate. As such, the most favorable transition state from C is TS2C-R, which eventually leads to the (R)enantiomer of the homoallylic alcohol product **8.3a**.



Figure 8.5 Computed energy profiles of the kinetic pathways leading to diastereomers D and C and, therefrom, (*S*)-and (*R*)-product enantiomers, respectively.



Figure 8.6 Enantioselectivity determining aldehyde allylation transition states with π allyliridium complexes D and C. Activation free energies are with respect to D and C, respectively.

8.4 Conclusion

In summary, we report the first catalytic enantioselective aldehyde allylations mediated by gaseous allene. These processes exploit a feedstock pronucleophile (allene) in combination with a feedstock reductant (2-propanol) under non-cryogenic conditions with acetone as the sole stoichiometric byproduct. Remarkably, use of allene vs allyl acetate as pronucleophile results in an inversion of enantioselectivity using the same antipode of chiral ligand, (*S*)-tol-BINAP. The collective experimental and computational data corroborate intervention of diastereometric π -allyliridium-*C*, *O*-benzoate complexes, which arise *via* allene hydrometalation (from a pentacoordinate iridium hydride) vs allyl acetate ionization (from a square planar iridium species). These data should facilitate the

design of related chiral-at-metal complexes for enantioselective catalysis by providing insight into the structural and interactions features of the catalyst that influence enantioselectivity. More broadly, these studies and other work from our laboratory demonstrate how reactions that traditionally have employed organometallic reagents may now be conducted catalytically in the absence of premetalated reagents using abundant feedstocks.^{1f,18}

8.5 Experimental Details

General Information

All reactions were carried out under inert gas atmosphere (nitrogen or argon) unless otherwise indicated. Resealable pressure tubes (13x100 mm) were purchased from Fischer Scientific (catalog number 14-959-35C) and were oven dried followed by cooling in a desiccator or under a stream of inter gas prior to use. All commercial reagents and anhydrous solvents were used as received from vendors (Fischer Scientific, Sigma Aldrich, and Combi Blocks) without further purification. Preparative column chromatography employing Silicycle silica gel (40-63 µm) was performed according to the method of Still¹ or on a Teledyne Isco Combiflash Rf utilizing Silicycle HP column using a mobile phase composed of either hexanes/ethyl acetate, hexanes/acetone, or dichloromethane/methanol. Reactions were monitored by analytical thin-layer chromatography (TLC) using 0.25 mm commercial silica gel plates (Dynamic Absorbents F). Visualization was accomplished with UV light followed by dipping in CAM, *p*-Anisaldehyde (PAA), or KMnO₄ stain solution followed by heating.

Spectroscopy, Spectrometry and Data Collection

Infrared spectra were recorded on a Perkin-Elmer 1600 spectrometer using a diamond ATR unit. High-resolution mass spectra (HRMS) were obtained on a Karatos MS9 and are reported as m/z (relative intensity). Accurate masses are reported for the molecular ion (M+H, M+Na), or a suitable fragment ion. Nuclear magnetic resonance (¹H, ¹³C, ¹⁹F NMR) spectra were recorded with a Bruker BioSpin GmbG, Varian Gemini (400 MHz) or Varian INOVA (500 MHz) spectrometer equipped with a Bruker cryoprobe. The chemical shifts are given as parts per million (ppm) and were referenced

to the residual solvent signal (CDCl₃: $\delta_{\rm H} = 7.26$ ppm, $\delta_{\rm C} = 77.16$ ppm). Specific optical rotations were recorded on an Azzota Corp AP45 (589 nm) in CHCl3. Solution concentrations are given in the units of 10–2 g mL⁻¹.

Experimental Details and Spectral Data

Detailed Procedure for Preparation of Iridium Complexes

(S)-Ir-V Made from Allyl Acetate (D)



To a sealed tube equipped with a magnetic stir bar was added Cs_2CO_3 (977.5 mg, 3.0 mmol, 200 mol%), 3,4-dinitrobenzoic acid (636.4 mg, 3.0 mmol, 200 mol%), (*S*)-TolBINAP (1.02 g, 1.5 mmol, 100 mol%) and [Ir(cod)Cl]₂ (503.8 mg, 0.75 mmol, 50 mol%). The reaction vessel was purged with argon and THF (15.0 mL, 0.1 M) was added followed by allyl acetate (0.40 mL, 3.75 mmol, 250 mol%). The resulting mixture was stirred at room temperature for 30 min, and at 80 °C for another 90 min. After cooling to ambient temperature, the mixture was filtered through a celite plug with the aid of DCM (45 mL). The combined filtrate was concentrated *in vacuo* and subjected to flash column chromatography (DCM:THF = 10:1). The resulting gum-like residue was dissolved in THF (2.0 mL). Addition of HPLC grade hexanes (20 mL) led to formation of a precipitate. The product was filtered and washed with HPLC grade hexanes, followed by removal of trace amount of solvent *in vacuo*, to provide a light yellow powder (1.4 g, 1.25 mmol) in 83% yield as a mixture of stereoisomers. The yellow powder was

dissolved in toluene and hexanes was allowed to diffuse into the toluene solution at room temperature *via* slow vapor diffusion, resulting in formation of crystals.

Spectral data is reported for the major isomer.

 $\frac{^{31}P \text{ NMR}}{HRMS} (202 \text{ MHz, CDCl}_3): \delta = -13.78 \text{ (d, } J = 22.5 \text{ Hz}), -16.62 \text{ (d, } J = 22.5 \text{ Hz}).$ $\frac{HRMS}{HRMS} (ESI): \text{ Calculated for } C_{58}H_{47}\text{IrN}_2\text{O}_6\text{P}_2 \text{ [M+H^+]} = 1123.2616, \text{ Found } 1123.2624.$ $[\alpha]_{D}^{28} = -100.8 \text{ (c } 0.3, \text{ CHCl}_3).$

MP: 236-246 °C (decomposes)



(S)-Ir-V Made from Allene Gas (C)



To a sealed tube equipped with a magnetic stir bar was added K_2CO_3 (110.6 mg, 0.8 mmol, 200 mol%), 3,4-dinitrobenzoic acid (169.7 mg, 0.8 mmol, 200 mol%), (S)-TolBINAP (271.5 mg, 0.4 mmol, 100 mol%) and [Ir(cod)Cl]₂ (134.3 mg, 0.2 mmol, 50 mol%). The tube was fit with a rubber septum before being evacuated by piercing the septum with a needle connected to a Schlenk line. Immediately after evacuation, an allene balloon fitted with a needle was used to pierce the septum and refill the tube with allene gas. Under the allene balloon, THF (4.0 mL, 0.1 M) was added. The septum was quickly removed and the tube was sealed with a PTFE lined cap. The resulting mixture was stirred at room temperature for 30 min, and at 80 °C for another 90 min. After cooling to ambient temperature, the mixture was filtered through a celite plug with the aid of DCM (45 mL). The combined filtrate was concentrated in vacuo and subjected to flash column chromatography (DCM:THF = 10:1). The resulting gum-like residue was dissolved in THF (1.0 mL). Addition of HPLC grade hexanes (20 mL) led to formation of a precipitate. The product was filtered and washed with HPLC grade hexanes, followed by removal of trace amount of solvent in vacuo, to provide a light yellow powder (358.0 mg, 0.32 mmol) in 80% yield as a mixture of stereoisomers.

Spectral data is reported for the major isomer.

³¹**P** NMR (202 MHz, CDCl₃): $\delta = -4.26$ (d, J = 18.7 Hz), -13.67 (d, J = 19.8 Hz).

<u>HRMS</u> (ESI): Calculated for $C_{58}H_{47}IrN_2O_6P_2[M+H^+] = 1123.2616$, Found 1123.2624.

 $[\alpha]_{D}^{28} = -282.7 \ (c \ 0.3, \text{CHCl}_3).$

MP: 226-234 °C (decomposes)



(S)-Ir-V recovered from Allylation - It was prepared according to the following procedures and was identical in all respects to the (S)-Ir-V made from allene gas.

A pressure tube equipped with a magnetic stir bar was charged with aldehyde **2a** (37.0 mg, 0.2 mmol, 100 mol%) and (*S*)-**Ir-V made from allyl acetate** (11.2 mg, 0.01 mmol, 5 mol%). The tube was fit with a rubber septum and was evacuated by piercing the septum with a needle connected to a Schlenk line. Immediately after evacuation, an allene-filled balloon equipped with a needle was used to pierce the septum and refill the tube with allene gas. Under the allene balloon, toluene (2.0 mL, 0.1 M) was added followed by 2-propanol (31 μ L, 0.4 mmol, 200 mol%). The septum was removed quickly and the tube was sealed with a PTFE lined cap. The tube was placed at 60 °C for 24 hours. The solution was allowed to reach ambient temperature before being concentrated *in vacuo*. The residue was directly subjected to flash column chromatography. The title compound was obtained in 54% yield (6.1 mg, 5.4 µmol) as a light yellow solid after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1–10:1 then DCM:THF = 10:1).



Spectral data is reported for the major isomer.

<u>**31P NMR**</u> (202 MHz, CDCl₃): δ = -4.27 (d, J = 19.0 Hz), -13.68 (d, J = 18.7 Hz).

Procedures and Spectral Data for Synthesis of Secondary Alcohols 8.3a-8.3s





General procedure

A pressure tube equipped with a magnetic stir bar was charged with aldehyde (0.2 mmol, 100 mol%) and (*S*)-Ir-V (11.2 mg, 0.01 mmol, 5 mol%). The tube was fit with a rubber septum and was evacuated by piercing the septum with a needle connected to a Schlenk line. Immediately after evacuation, an allene-filled balloon equipped with a needle was used to pierce the septum and refill the tube with allene gas. Under the allene balloon, toluene (2.0 mL, 0.1 M) was added followed by 2-propanol (31 μ L, 0.4 mmol, 200 mol%). The septum was removed quickly and the tube was sealed with a PTFE lined cap. The tube was placed in an oil bath at the indicated temperature and stirred for the indicated time period. The solution was allowed to reach ambient temperature before being concentrated *in vacuo*. The residue was directly subjected to flash column chromatography.

(*R*)-1-(4-bromophenyl)but-3-en-1-ol (8.3a)



Procedures

The aldehyde (37.0 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 90% yield (41.0 mg, 0.18 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-10:1).

<u>**TLC (SiO₂)**</u> $R_f = 0.48$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.47 (d, *J* = 8.4 Hz, 2H), 7.24 (d, *J* = 8.3 Hz, 2H), 5.86 – 5.66 (m, 1H), 5.19 – 5.13 (m, 2H), 4.71 (ddd, *J* = 7.9, 4.8, 2.9 Hz, 1H), 2.54 – 2.41 (m, 2H), 2.06 (t, *J* = 2.8 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 142.8, 133.9, 131.5, 127.5, 121.2, 118.9, 72.5, 43.8.

<u>HRMS</u> (EI): Calculated for $C_{10}H_{11}BrO[M^+] = 225.9993$, Found 225.9990.

<u>FTIR</u> (neat): 3372, 3079, 2926, 1641, 1592, 1488, 1403, 1069, 1009, 917, 870, 821, 776, 717 cm⁻¹.

 $[\alpha]_{D}^{28} = +48.0 \ (c \ 0.25, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 92%.</u>





(R)-1-(4-cyanophenyl)but-3-en-1-ol (8.3b)



Procedures

The aldehyde (26.2 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 82% yield (28.4 mg, 0.16 mmol) as a yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.28$ (hexanes: ethyl acetate = 3:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.63 (d, *J* = 8.2 Hz, 2H), 7.47 (d, *J* = 8.3 Hz, 2H), 5.87 – 5.65 (m, 1H), 5.29 – 5.05 (m, 2H), 4.80 (dd, *J* = 8.0, 4.7 Hz, 1H), 2.57 – 2.49 (m, 1H), 2.43 (dt, *J* = 14.7, 7.9 Hz, 1H), 2.10 (s, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{43.9}}$ (125 MHz, CDCl₃): $\delta = 149.2$, 133.4, 132.2, 126.5, 119.5, 118.9, 111.2, 72.4,

<u>HRMS</u> (ESI): Calculated for $C_{11}H_{11}NO[M+Na^+] = 196.0733$, Found 196.0738.

<u>FTIR</u> (neat): 3405, 3077, 2979, 2922, 2228, 1641, 1609, 1604, 1412, 1198, 1055, 990, 919, 838 cm⁻¹.

 $[\alpha]_{D}^{28} = +46.2 \ (c \ 0.1, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:i-PrOH = 95:5, 1.00 mL/min, 230 nm), ee = 88%.





methyl (R)-4-(1-hydroxybut-3-en-1-yl)benzoate (8.3c)



Procedures

The aldehyde (32.8 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 88% yield (36.3 mg, 0.18 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-10:1).

<u>TLC (SiO2</u>) $R_f = 0.27$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.07 - 7.95$ (m, 2H), 7.43 (d, J = 8.3 Hz, 2H), 5.79 (dddd, J = 17.9, 9.8, 7.7, 6.5 Hz, 1H), 5.24 - 5.10 (m, 2H), 4.81 (dd, J = 8.0, 4.8 Hz, 1H), 3.91 (s, 3H), 2.59 - 2.51 (m, 1H), 2.51 - 2.42 (m, 1H), 2.22 - 2.09 (m, 1H).

 $\frac{^{13}C \text{ NMR}}{52.1, 43.8}$ (125 MHz, CDCl₃): δ = 166.9, 148.9, 133.8, 129.7, 129.3, 125.7, 119.1, 72.7, 52.1, 43.8.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{14}O_3$ [M+Na⁺] = 229.0835, Found 229.0844.

<u>FTIR</u> (neat): 3436, 2952, 1720, 1611, 1436, 1276, 1192, 1111, 1054, 1018, 917, 957, 768, 707 cm⁻¹.

 $[\alpha]_{D}^{28} = +45.0 \ (c \ 0.3, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel AD-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 92%.</u>





(*R*)-1-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)but-3-en-1-ol (8.3d)



Procedures

The aldehyde (46.4 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 85% yield (46.6 mg, 0.17 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-5:1).

<u>TLC (SiO</u>) $R_f = 0.28$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.85 - 7.76$ (m, 2H), 7.41 - 7.31 (m, 2H), 5.79 (ddt, *J* = 17.1, 10.0, 7.1 Hz, 1H), 5.21 - 5.09 (m, 2H), 4.76 (ddd, *J* = 8.0, 5.0, 3.2 Hz, 1H), 2.60 - 2.41 (m, 2H), 2.06 (d, *J* = 3.3 Hz, 1H), 1.34 (s, 12H).

 $\frac{^{13}C \text{ NMR}}{^{24.9}}$ (125 MHz, CDCl₃): $\delta = 147.0$, 134.9, 134.3, 125.1, 118.5, 83.8, 73.2, 43.8,

<u>HRMS</u> (ESI): Calculated for $C_{16}H_{23}BO_3$ [M+Na⁺] = 297.1625, Found 297.1643.

<u>FTIR</u> (neat): 3318, 2980, 2921, 1613, 1359, 1324, 1142, 1090, 1017, 859, 843, 660 cm⁻¹. $[\alpha]_{\mathbf{D}}^{\mathbf{28}} = +22.5 \ (c \ 0.4, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 230 nm), ee = 92%.</u>





(R)-1-(4-chloro-3-(trifluoromethyl)phenyl)but-3-en-1-ol (8.3e)



Procedures

The aldehyde (28.8 μ L, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 74% yield (37.1 mg, 0.15 mmol) as a pale yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1–10:1).

<u>**TLC (SiO₂**</u>) $R_f = 0.38$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.69 (s, 1H), 7.47 (d, J = 2.6 Hz, 2H), 5.86 – 5.68 (m, 1H), 5.24 – 5.13 (m, 2H), 4.77 (dd, J = 8.2, 4.6 Hz, 1H), 2.59 – 2.35 (m, 2H), 2.10 (s, 1H).

 $\frac{^{13}C \text{ NMR}}{^{12}} (125 \text{ MHz}, \text{CDCl}_3): \delta = 142.9, 133.3, 131.4, 131.1, 130.2, 125.0 (q, J = 4.0 \text{ Hz}), 124.0, 121.8, 119.6, 71.9, 43.9.$

¹⁹**F NMR** (471 MHz, CDCl₃): δ = -62.6

<u>HRMS</u> (EI): Calculated for $C_{11}H_{10}ClF_{3}O[M^+] = 250.0372$, Found 250.0368.

<u>FTIR</u> (neat): 3392, 2928, 1481, 1422, 1316, 1261, 1171, 1130, 1111, 1034, 921, 832, 664 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +46.0 (*c* 0.2, CHCl₃).

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 280 nm), ee = 90%.</u>






(R)-1-(benzo[d][1,3]dioxol-5-yl)but-3-en-1-ol (8.3f)



Procedures

The aldehyde (30.0 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 84% yield (32.7 mg, 0.17 mmol) as a brown oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.38$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.87$ (s, 1H), 6.78 (q, J = 8.0 Hz, 2H), 5.94 (d, J = 1.3 Hz, 2H), 5.78 (ddt, J = 14.2, 10.1, 7.2 Hz, 1H), 5.24 – 5.02 (m, 2H), 4.64 (t, J = 6.4 Hz, 1H), 2.47 (t, J = 6.6 Hz, 2H), 2.04 (s, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{101.0, 73.2, 43.8}$ (125 MHz, CDCl₃): $\delta = 147.7, 146.9, 138.0, 134.4, 119.2, 118.4, 108.1, 106.4, 101.0, 73.2, 43.8.$

<u>HRMS</u> (EI): Calculated for $C_{11}H_{12}O_3[M^+] = 192.0786$, Found 192.0791.

<u>FTIR</u> (neat): 3391, 2899, 1640, 1503, 1486, 1441, 1239, 1093, 1036, 919, 865, 810 cm⁻¹. $[\alpha]_{D}^{28} = +42.4 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 94%.</u>





(*R*)-1-(2,3-dichlorophenyl)but-3-en-1-ol (8.3g)



Procedures

The aldehyde (35.0 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 82% yield (35.7 mg, 0.16 mmol) as a pale brown solid after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.50 (dd, *J* = 7.8, 1.6 Hz, 1H), 7.39 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.24 (t, *J* = 7.8 Hz, 1H), 5.86 (dddd, *J* = 17.5, 9.6, 7.8, 6.4 Hz, 1H), 5.23 – 5.20 (m, 1H), 5.20 – 5.15 (m, 2H), 2.71 – 2.55 (m, 1H), 2.42 – 2.26 (m, 1H), 1.95 (s, 1H). ¹³<u>C NMR</u> (125 MHz, CDCl₃): δ = 143.8, 134.0, 133.1, 129.9, 129.3, 127.6, 125.4, 119.3, 70.1, 42.0.

<u>HRMS</u> (EI): Calculated for $C_{10}H_{10}OCl_2 [M^+] = 216.0109$, Found 216.0110.

<u>FTIR</u> (neat): 3288, 3074, 2942, 1640, 1421, 1181, 1103, 918, 779, 718 cm⁻¹.

 $[\alpha]_{D}^{28} = +92.9 \ (c \ 0.3, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 230 nm), ee = 91%.





(*R*)-1-(2,5-dimethoxyphenyl)but-3-en-1-ol (8.3h)



Procedures

The aldehyde (33.2 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 84% yield (35.0 mg, 0.17 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1-4:1).

<u>TLC (SiO2</u>) $R_f = 0.35$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.94$ (d, J = 3.1 Hz, 1H), 6.80 (d, J = 8.8 Hz, 1H), 6.76 (dd, J = 8.8, 3.0 Hz, 1H), 5.93 – 5.75 (m, 1H), 5.20 – 5.06 (m, 2H), 4.93 (dt, J = 8.1, 5.0 Hz, 1H), 3.81 (s, 3H), 3.77 (s, 3H), 2.63 – 2.54 (m, 2H), 2.53 – 2.43 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 153.7, 150.5, 135.1, 132.9, 117.7, 112.9, 112.5, 111.4, 132.9, 117.7, 112.9, 112.5, 111.4, 132.9, 117.7, 112.9, 112.5, 111.4, 132.9, 117.7, 112.9, 112.5, 111.4, 132.9, 112.5, 111.4, 143.143 (dt, <math>J = 8.1, 5.0$ Hz, 125 MHz, CDCl₃): $\delta = 153.7, 150.5, 135.1, 132.9, 117.7, 112.9, 112.5, 111.4, 143.143 (dt, <math>J = 8.1, 5.0$ Hz, 125 MHz, CDCl₃): $\delta = 153.7, 150.5, 135.1, 132.9, 117.7, 112.9, 112.5, 111.4, 143.143 (dt, <math>J = 8.1, 5.0$ Hz, 125 MHz, CDCl₃): $\delta = 153.7, 150.5, 135.1, 132.9, 117.7, 112.9, 112.5, 111.4, 143.143 (dt, <math>J = 8.1, 5.0$ Hz, 125 MHz, CDCl₃): $\delta = 153.7, 150.5, 135.1, 132.9, 117.7, 112.9, 112.5, 111.4, 143.143 (dt, <math>J = 8.1, 5.0$ Hz, 125 MHz, 125

69.7, 55.8, 55.7, 41.9.

HRMS (ESI): Calculated for $C_{12}H_{13}N[M+H^+] = 231.0992$, Found 231.0991.

<u>FTIR</u> (neat): 3440, 2941, 2834, 1494, 1464, 1429, 1276, 1213, 1178, 1157, 1043, 914, 803, 708 cm⁻¹.

 $[\alpha]_{D}^{28} = +36.7 \ (c \ 0.3, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 92%.</u>





(R)-1-(2-bromo-3,4,5-trimethoxyphenyl)but-3-en-1-ol (8.3i)



Procedures

The aldehyde (55.0 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 92% yield (58.1 mg, 0.18 mmol) as a pale yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>TLC (SiO</u>₂) $R_f = 0.28$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 6.97$ (s, 1H), 5.99 – 5.82 (m, 1H), 5.28 – 5.14 (m, 2H), 5.09 (dd, J = 8.7, 3.4 Hz, 1H), 3.89 (s, 3H), 3.88 (s, 3H), 3.87 (s, 3H), 2.63 (dt, J = 13.9, 4.6 Hz, 1H), 2.30 (dt, J = 15.2, 8.3 Hz, 1H), 1.99 (s, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 153.1, 150.6, 142.3, 138.6, 134.6, 118.8, 108.0, 105.9, 72.0, 61.2, 61.2, 56.3, 42.3.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{17}BrO_4 [M+Na^+] = 339.0202$, Found 339.0206.

<u>FTIR</u> (neat): 3471, 2937, 1569, 1480, 1393, 1323, 1195, 1161, 1103, 1038, 1006, 919, 851, 816 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +56.8 (*c* 0.4, CHCl₃).

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 230 nm), ee = 91%.







(*R*)-1-(6-methoxypyridin-3-yl)but-3-en-1-ol (8.3j)



Procedures

The aldehyde (27.4 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 91% yield (32.7 mg, 0.18 mmol) as a yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-5:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.23$ (hexanes: ethyl acetate = 3:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.19 - 7.99$ (m, 1H), 7.60 (dd, J = 8.5, 2.3 Hz, 1H), 6.72 (d, J = 8.5 Hz, 1H), 5.89 - 5.65 (m, 1H), 5.24 - 5.04 (m, 2H), 4.69 (t, J = 6.4 Hz, 1H), 3.91 (d, J = 2.1 Hz, 3H), 2.48 (t, J = 6.6 Hz, 2H), 2.33 (s, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 163.8, 144.5, 136.8, 133.9, 132.0, 118.8, 110.8, 70.8, 53.6, 43.5.

<u>HRMS</u> (ESI): Calculated for $C_{10}H_{13}NO_2 [M+Na^+] = 202.0838$, Found 202.0841.

<u>FTIR</u> (neat): 3337, 3077, 2979, 2946, 1608, 1574, 1493, 1395, 1331, 1256, 1125, 1026, 917, 831 cm⁻¹.

 $[\alpha]_{D}^{28} = +61.3 (c \ 0.4, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 254 nm), ee = 92%.</u>







(R)-1-(2-(dimethylamino)pyrimidin-5-yl)but-3-en-1-ol (8.3k)



Procedures

The aldehyde (30.2 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 76% yield (29.2 mg, 0.15 mmol) as a yellow oil after isolation by flash column chromatography (SiO₂, hexanes: acetone = 15:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.22$ (hexanes: ethyl acetate = 1:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.29$ (s, 2H), 5.86 – 5.69 (m, 1H), 5.20 – 5.06 (m, 2H),

4.60 (t, *J* = 6.6 Hz, 1H), 3.18 (s, 6H), 2.49 (h, *J* = 6.9 Hz, 2H), 2.26 (s, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 161.9, 156.2, 133.9, 123.5, 119.1, 69.7, 43.2, 37.4.

<u>HRMS</u> (ESI): Calculated for $C_{10}H_{15}N_3O[M+Na^+] = 194.1288$, Found 194.1285.

<u>FTIR</u> (neat): 3345, 2929, 1604, 1537, 1405, 1335, 1310, 1192, 1175, 1054, 973, 916, 799, 659 cm⁻¹.

 $[\alpha]_{D}^{28} = +39.1 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (two connected Chiralcel OD-H columns, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 254 nm), ee = 93%.</u>





(*R*)-1-(1H-indol-6-yl)but-3-en-1-ol (8.3l)



Procedures

The aldehyde (29.0 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (50 °C, 48 hr). The title compound was obtained in 79% yield (29.6 mg, 0.16 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, dichloromethane: diethyl ether = 1:0-10:1).

<u>**TLC**</u> (SiO₂) $R_f = 0.28$ (dichloromethane: diethyl ether = 10:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = \delta$ 8.24 (s, 1H), 7.62 (d, J = 8.1 Hz, 1H), 7.35 (s, 1H), 7.18 (t, J = 2.8 Hz, 1H), 7.11 (dd, J = 8.1, 1.4 Hz, 1H), 6.54 (t, J = 2.5 Hz, 1H), 5.84 (ddt, J = 17.2, 10.1, 7.1 Hz, 1H), 5.22 – 5.09 (m, 2H), 4.84 (t, J = 6.5 Hz, 1H), 2.59 (t, J = 6.6 Hz, 2H), 2.23 (s, 1H).

 $\frac{^{13}C \text{ NMR}}{^{118.0}, 108.3, 102.3, 74.0, 44.0.} = \delta 138.0, 135.8, 134.9, 127.3, 124.5, 120.6, 118.1, 118.0, 108.3, 102.3, 74.0, 44.0.$

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{13}NO[M+Na^+] = 210.0889$, Found 210.0884.

<u>FTIR</u> (neat): 3412, 2918, 1640, 1509, 1453, 1346, 1042, 989, 908, 867, 814, 767, 725 cm⁻¹.

 $[\alpha]_{D}^{28} = +39.0 \ (c \ 0.3, \ CHCl_3).$

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:*i*-PrOH = 85:15, 1.00 mL/min, 230 nm), ee = 87%.





(*R*)-1-(quinolin-2-yl)but-3-en-1-ol (8.3m)



Procedures

The aldehyde (31.4 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 88% yield (35.1 mg, 0.18 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-5:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.27$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 8.15$ (d, J = 8.5 Hz, 1H), 8.08 (d, J = 8.4 Hz, 1H), 7.83 (dd, J = 8.1, 1.4 Hz, 1H), 7.73 (ddd, J = 8.4, 6.9, 1.5 Hz, 1H), 7.54 (ddd, J = 8.1, 6.8, 1.2 Hz, 1H), 7.37 (d, J = 8.4 Hz, 1H), 5.89 (ddt, J = 17.2, 10.2, 7.0 Hz, 1H), 5.17 – 5.05 (m, 2H), 4.99 (t, J = 6.0 Hz, 1H), 4.89 (s, 1H), 2.75 (dddt, J = 14.2, 7.0, 4.6, 1.3 Hz, 1H), 2.56 (dtt, J = 14.3, 7.1, 1.3 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 161.3, 146.5, 136.8, 134.1, 129.8, 128.8, 127.6, 127.5, 126.4, 118.4, 117.9, 72.2, 42.6.

<u>HRMS</u> (ESI): Calculated for $C_{13}H_{13}NO[M+Na^+] = 222.0889$, Found 222.0896.

<u>FTIR</u> (neat): 3261, 2921, 2895, 1618, 1599, 1502, 1430, 1304, 1069, 919, 793, 757 cm⁻¹. $[\alpha]_{D}^{28} = -18.0 \ (c \ 0.3, CHCl_3).$

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 230 nm), ee = 87%.





(*R*)-1-(2-bromothiazol-4-yl)but-3-en-1-ol (8.3n)



Procedures

The aldehyde (38.4 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 71% yield (33.1 mg, 0.14 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-4:1).

<u>**TLC (SiO₂**</u>) $R_f = 0.32$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.15 (s, 1H), 5.80 (ddt, *J* = 17.2, 10.2, 7.1 Hz, 1H), 5.21 – 5.14 (m, 2H), 4.85 (dt, *J* = 7.8, 4.9 Hz, 1H), 2.71 (dddt, *J* = 14.1, 6.2, 4.6, 1.3 Hz, 1H), 2.61 – 2.50 (m, 1H), 2.45 (d, *J* = 5.3 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 159.2, 135.9, 133.5, 119.0, 117.6, 69.7, 41.6.

<u>**HRMS**</u> (ESI): Calculated for $C_7H_8BrNOS[M+Na^+] = 257.9381$, Found 257.9391.

<u>FTIR</u> (neat): 3350, 3076, 2907, 1640, 1519, 1416, 1317, 1243, 1012, 916, 758 cm⁻¹. $[\alpha]_D^{28} = +75.0 \ (c \ 1.0, \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 254 nm), ee = 90%.</u>







(R)-1-(5-(3-chloro-4-methoxyphenyl)furan-2-yl)but-3-en-1-ol (8.30)



Procedures

The aldehyde (47.3 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 79% yield (44.0 mg, 0.16 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: dichloromethane = 1:1-0:1).

<u>**TLC (SiO**</u>₂) $R_f = 0.55$ (dichloromethane).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.66 (d, J = 2.2 Hz, 1H), 7.50 (dd, J = 8.6, 2.2 Hz, 1H), 6.92 (d, J = 8.6 Hz, 1H), 6.47 (d, J = 3.3 Hz, 1H), 6.32 (d, J = 3.3 Hz, 1H), 5.85 (ddt, J = 17.1, 10.2, 7.0 Hz, 1H), 5.25 – 5.10 (m, 2H), 4.79 (dt, J = 9.6, 4.9 Hz, 1H), 3.92 (s, 3H), 2.68 (qt, J = 7.1, 1.3 Hz, 2H), 2.13 (d, J = 4.4 Hz, 1H).

 $\frac{^{13}\mathbf{C} \text{ NMR}}{^{11}} (125 \text{ MHz, CDCl}_3): \delta = 155.4, 154.2, 152.0, 133.6, 125.7, 124.6, 123.2, 122.8, 118.7, 112.1, 108.3, 104.9, 67.0, 56.2, 40.1.$

<u>HRMS</u> (ESI): Calculated for $C_{15}H_{15}ClO_3 [M+Na^+] = 301.0602$, Found 301.0601.

<u>FTIR</u> (neat): 3377, 2939, 1490, 1440, 1289, 1267, 1062, 1019, 909, 787, 730, 708 cm⁻¹. $[\alpha]_{D}^{28} = +30.0 \ (c \ 0.3, CHCl_3).$

<u>HPLC</u> (Chiralcel AS-H column, hexanes:i-PrOH = 95:5, 1.00 mL/min, 280 nm), ee = 88%.





(R)-1-(benzo[b]thiophen-3-yl)but-3-en-1-ol (8.3p)



Procedures

The aldehyde (32.4 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 81% yield (33.1 mg, 0.16 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1-8:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 4:1).

 $\frac{1}{H \text{ NMR}} (500 \text{ MHz, CDCl}_3): \delta = 7.88 (ddd, J = 7.4, 5.4, 1.4 \text{ Hz}, 2\text{H}), 7.45 - 7.30 (m, 3\text{H}), 5.96 - 5.83 (m, 1\text{H}), 5.26 - 5.17 (m, 2\text{H}), 5.15 (dt, J = 8.0, 3.9 \text{ Hz}, 1\text{H}), 2.77 (dddt, J = 14.0, 6.2, 4.6, 1.4 \text{ Hz}, 1\text{H}), 2.71 - 2.60 (m, 1\text{H}), 2.15 (d, J = 3.5 \text{ Hz}, 1\text{H}).$

¹³C NMR (125 MHz, CDCl₃): δ = 140.9, 138.8, 137.2, 134.2, 124.4, 124.0, 122.9, 122.2, 122.2, 118.7, 68.9, 41.8.

<u>HRMS</u> (ESI): Calculated for $C_{12}H_{12}OS[M+Na^+] = 227.0501$, Found 227.0507.

<u>FTIR</u> (neat): 3373, 3075, 2906, 1640, 1459, 1427, 1256, 1140, 1057, 1022, 915, 868, 834, 759, 732 cm⁻¹.

 $[\alpha]_{D}^{28} = +48.2 (c \ 0.3, \text{CHCl}_3).$

<u>HPLC</u> (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 210 nm), ee = 93%.







(*R*,*E*)-2-methyl-1-phenylhexa-1,5-dien-3-ol (8.3q)



Procedures

The aldehyde (29.2 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 24 hr). The title compound was obtained in 61% yield (23.1 mg, 0.12 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>TLC (SiO</u>₂) $R_f = 0.39$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): δ = 7.33 (t, *J* = 7.6 Hz, 2H), 7.30 – 7.26 (m, 2H), 7.24 – 7.19 (m, 1H), 6.53 (s, 1H), 5.84 (ddt, *J* = 17.2, 10.2, 7.1 Hz, 1H), 5.23 – 5.13 (m, 2H), 4.27 – 4.18 (m, 1H), 2.50 – 2.43 (m, 1H), 2.43 – 2.36 (m, 1H), 1.89 (d, *J* = 1.3 Hz, 3H), 1.80 (d, *J* = 3.0 Hz, 1H).

¹³C NMR (125 MHz, CDCl₃): δ = 139.5, 137.5, 134.5, 129.0, 128.1, 126.4, 125.7, 118.1, 76.5, 40.1, 13.7.

<u>HRMS</u> (EI): Calculated for $C_{13}H_{16}O[M^+] = 188.1201$, Found 188.1205.

<u>FTIR</u> (neat): 3366, 3079, 2919, 1641, 1442, 992, 915, 867, 747, 697 cm⁻¹.

 $[\alpha]_{D}^{28} = -4.5 \ (c \ 0.2, \ CHCl_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 254 nm), ee = 92%.</u>




(S)-1-phenylhex-5-en-3-ol (8.3r)



Procedures

The aldehyde (26.8 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 65% yield (22.9 mg, 0.13 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 20:1-10:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 7.29$ (t, J = 7.6 Hz, 2H), 7.23 - 7.17 (m, 3H), 5.96 - 5.66 (m, 1H), 5.19 - 5.10 (m, 2H), 3.68 (tt, J = 7.7, 4.6 Hz, 1H), 2.82 (ddd, J = 13.8, 8.9, 6.6 Hz, 1H), 2.70 (dt, J = 13.8, 7.8 Hz, 1H), 2.33 (dddt, J = 13.7, 5.8, 4.3, 1.3 Hz, 1H), 2.19 (dt, J = 13.9, 7.9 Hz, 1H), 1.84 - 1.73 (m, 2H).

 $\frac{^{13}C \text{ NMR}}{^{38.4}, 32.0}$ (125 MHz, CDCl₃): $\delta = 142.0, 134.6 \ 128.4, 128.4, 125.8, 118.3, 69.9, 42.1, 38.4, 32.0.$

HRMS (ESI): Calculated for $C_{12}H_{16}O[M+Na^+] = 199.1093$, Found 199.1096.

<u>FTIR</u> (neat): 3342, 3026, 2927, 1740, 1640, 1603, 1496, 1454, 1047, 994, 914, 746, 698 cm⁻¹.

 $[\alpha]_{D}^{28} = -6.0 \ (c \ 0.2, \ \text{CHCl}_3).$

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 98:2, 1.00 mL/min, 210 nm), ee = 90%.</u>





tert-butyl (R)-4-(1-hydroxybut-3-en-1-yl)piperidine-1-carboxylate (8.3s)



Procedures

The aldehyde (42.7 mg, 0.20 mmol, 100 mol%) was subjected to standard reaction conditions (60 °C, 48 hr). The title compound was obtained in 71% yield (36.3 mg, 0.14 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 15:1-4:1).

<u>**TLC** (SiO₂</u>) $R_f = 0.29$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = 5.76$ (dddd, J = 15.5, 11.5, 8.4, 6.1 Hz, 1H), 5.14 – 5.05 (m, 2H), 4.08 (s, 2H), 3.35 (ddd, J = 9.2, 6.1, 3.4 Hz, 1H), 2.59 (s, 2H), 2.29 (dddd, J = 14.1, 6.3, 3.2, 1.5 Hz, 1H), 2.06 (dt, J = 14.0, 8.4 Hz, 1H), 1.75 (dq, J = 11.3, 2.8 Hz, 1H), 1.54 (dt, J = 13.0, 3.0 Hz, 1H), 1.45 (ddq, J = 11.8, 5.7, 3.2 Hz, 1H), 1.39 (s, 9H), 1.26 – 1.14 (m, 2H).

¹³C NMR (125 MHz, CDCl₃): δ = 154.8, 134.8, 118.6, 79.3, 73.7, 43.8, 41.5, 38.8, 28.5, 28.1.

<u>HRMS</u> (ESI): Calculated for $C_{14}H_{25}NO_3 [M+Na^+] = 278.1727$, Found 2788.1726.

<u>FTIR</u> (neat): 3429, 2976, 2930, 2858, 1668, 1424, 1365, 1277, 1235, 1164, 982, 911, 866, 769 cm⁻¹.

 $[\alpha]_{\rm D}^{28} = -10.6 \ (c \ 0.2, \ \rm CHCl_3).$

<u>HPLC</u> Enantiomeric excess was determined by HPLC analysis of the *p*-nitrobenzoyl derivative of the product (Chiralcel OJ-H column, hexanes:*i*-PrOH = 97:3, 1.00 mL/min, 254 nm), ee = 90%.







Procedures and Spectral Data for Synthesis of deuterio-8.3a

(*R*)-1-(4-bromophenyl)but-3-en-3-d-1-ol (deuterio-8.3a)



Procedures

A pressure tube equipped with a magnetic stir bar was charged with aldehyde (32.4 mg, 0.20 mmol, 100 mol%) and (*S*)-**Ir-V** (11.2 mg, 0.01 mmol, 5 mol%). The tube was fit with a rubber septum. The tube was evacuated by piercing the septum with a needle connected to a Schlenk line. Immediately after evacuation, an allene balloon fitted with a needle was used to pierce the septum and refill the tube with allene gas. Under the allene balloon, toluene (2.0 mL, 0.1 M) was added followed by d_8 -isopropylalcohol (31 µL, 0.4 mmol, 200 mol%). The septum was quickly removed and the tube was sealed with a PTFE lined cap. The tube was placed in an oil bath at 60 °C for 24 hours. The reaction was allowed to reach ambient temperature and concentrated *in vacuo*. The title compound was obtained in 86% yield (39.2 mg, 0.17 mmol) as a light yellow oil after isolation by flash column chromatography (SiO₂, hexanes: ethyl acetate = 25:1–8:1).

<u>TLC (SiO</u>) $R_f = 0.45$ (hexanes: ethyl acetate = 4:1).

¹<u>H NMR</u> (500 MHz, CDCl₃): $\delta = \delta$ 7.50 – 7.44 (m, 2H), 7.25 – 7.20 (m, 2H), 5.91 – 5.64 (m, 1H), 5.23 – 5.08 (m, 2H), 4.70 (dd, J = 7.9, 5.0 Hz, 1H), 2.54 – 2.39 (m, 2H), 2.09 (s, 1H).

1³C NMR (125 MHz, CDCl₃): δ = 142.8, 133.9, 131.5, 127.5, 121.2, 118.8 (d, *J* = 17.0 Hz), 72.5, 43.8 (d, *J* = 15.1 Hz).

²**H NMR** (92 MHz, CHCl₃): δ = 5.82, 5.20, 2.48 (d, *J* = 3.9 Hz)

<u>HRMS</u> (EI): Calculated for $C_{10}H_{10}DBrO[M^+] = 227.0056$, Found 227.0063.

<u>FTIR</u> (neat): 3332, 3075, 2906, 1641, 1593, 1488, 1403, 1069, 1009, 917, 870, 821, 776, 717 cm⁻¹.

 $[\alpha]_{D}^{28}$ = +30.0 (*c* 0.2, CHCl₃).

<u>**HPLC</u>** (Chiralcel OJ-H column, hexanes:*i*-PrOH = 95:5, 1.00 mL/min, 210 nm), ee = 92%.</u>







Single Crystal Diffraction Data for (S)-Ir-V Made from Allyl Acetate

Empirical formula	C72 H63 Ir N2 O6 P2	
Formula weight	1306.38	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	orthorhombic	
Space group	P 21 21 21	
Unit cell dimensions	a = 15.155(2) Å	α= 90°.
	b = 17.531(3) Å	β= 90°.
	c = 22.268(3) Å	$\gamma = 90^{\circ}.$
Volume	5916.2(15) Å ³	
Z	4	
Density (calculated)	1.467 Mg/m ³	
Absorption coefficient	2.368 mm ⁻¹	
F(000)	2656	
Crystal size	0.410 x 0.250 x 0.140 mm ³	
Theta range for data collection	1.776 to 27.877°.	
Index ranges	-19<=h<=19, -23<=k<=22, -29<=l<=29	
Reflections collected	168850	
Independent reflections	14106 [R(int) = 0.0595]	
Completeness to theta = 25.242°	99.9 %	
Absorption correction	Numerical	
Max. and min. transmission	0.7723 and 0.5359	
Refinement method	Full-matrix least-squares on F ² 825	

Data / restraints / parameters	14106 / 777 / 819
Goodness-of-fit on F ²	1.045
Final R indices [I>2sigma(I)]	R1 = 0.0215, wR2 = 0.0353
R indices (all data)	R1 = 0.0261, wR2 = 0.0357
Absolute structure parameter	-0.0019(16)
Extinction coefficient	n/a
Largest diff. peak and hole	0.648 and -0.642 e.Å ⁻³

Figure S1. View of the Ir complex in **1** showing the heteroatom labeling scheme. Displacement ellipsoids are scaled to the 50% probability level.



References

- (1) For selected reviews on the allylic amination, see: (a) Trost, B. M.; Crawley, M. L. Chem. Rev. 2003, 103, 2921–2943. (b) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. Chem. Commun. 2007, 675–691. (c) Lu, Z.; Ma, S. Angew. Chem. Int. Ed. 2008, 47, 258–297. (d) Trost, B. M.; Zhang, T.; Sieber, J. D. Chem. Sci. 2010, 1, 427–440. (e) Hartwig, J. F.; Stanley, L. M. Acc. Chem. Res. 2010, 43, 1461–1475. (f) Tosatti, P.; Nelson, A.; Marsden, S. P. Org. Biomol. Chem. 2012, 10, 3147–3163. (g) Grange, R. L.; Clizbe, E. A.; Evans, P. A. Synth. 2016, 48, 2911–2968. (h) Qu, J.; Helmchen, G. Acc. Chem. Res. 2017, 50, 2539–2555. (i) Cheng, Q.; Tu, H. F.; Zheng, C.; Qu, J. P.; Helmchen, G.; You, S.-L. Chem. Rev. 2019, 119, 1855–1969. (j) Rössler, S. L.; Petrone, D. A.; Carreira, E. M. Acc. Chem. Res. 2019, 52, 2657–2672.
- (2) Takeuchi, R.; Kashio, M. Angew. Chem. Int. Ed. Engl. 1997, 36, 263–265.
- (3) Janssen, J. P.; Helmchen, G. *Tetrahedron Lett.* **1997**, *38*, 8025–8026.
- (4) Takeuchi, R.; Ue, N.; Tanabe, K.; Yamashita, K.; Shiga, N. J. Am. Chem. Soc.
 2001, 123, 9525–9534.
- (5) Ohmura, T.; Hartwig, J. F. J. Am. Chem. Soc. 2002, 124, 15164–15165.
- (6) (a) de Vries, A. H. M.; Meetsma, A.; Feringa, B. L. Angew. Chem. Int. Ed. Engl. 1996, 35, 2374–2376. (b) Feringa, B. L.; Pineschi, M.; Arnold, L. A.; Imbos, R.; de Vries, A. H. M. Angew. Chem. Int. Ed. Engl. 1997, 36, 2620–2623.
- Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. J. Am. Chem. Soc. 2003, 125, 14272–14273.

- (8) Marković, D.; Hartwig, J. F. J. Am. Chem. Soc. 2007, 129, 11680–11681.
- (9) Shu, C.; Leitner, A.; Hartwig, J. F. Angew. Chem. Int. Ed. 2004, 43, 4797–4800.
- (10) Lipowsky, G.; Helmchen, G. Chem. Commun. 2004, 4, 116–117.
- Welter, C.; Dahnz, A.; Brunner, B.; Streiff, S.; Dubon, P.; Helmchen, G. Org.
 Lett. 2005, 7, 1239–1242.
- (12) Tissot-Croset, K.; Polet, D.; Alexakis, A. Angew. Chem. Int. Ed. 2004, 43, 2426– 2428.
- (13) Weihofen, R.; Dahnz, A.; Tverskoy, O.; Helmchen, G. Chem. Commun. 2005, 3541–3543.
- Weihofen, R.; Tverskoy, O.; Helmchen, G. Angew. Chem. Int. Ed. 2006, 45, 5546–5549.
- (15) (a) Singh, O. V.; Han, H. *Tetrahedron Lett.* 2007, 48, 7094–7098. (b) Lee, J. S.;
 Kim, D.; Lozano, L.; Kong, S. Bin; Han, H. Org. Lett. 2013, 15, 554–557.
- (16) Leitner, A.; Shekhar, S.; Pouy, M. J.; Hartwig, J. F. J. Am. Chem. Soc. 2005, 127, 15506–15514.
- (17) Shekhar, S.; Trantow, B.; Leitner, A.; Hartwig, J. F. J. Am. Chem. Soc. 2006, 128, 11770–11771.
- (18) (a) Bartels, B.; Garcia-Yebra, C.; Rominger, F.; Helmchen, G. Eur. J. Inorg. Chem. 2002, 2569. (b) Polet, D.; Alexakis, A.; Tissot-Croset, K.; Corminboeuf, C.; Ditrich, K. Chem. Eur. J. 2006, 12, 3596–3609.
- (19) Pouy, M. J.; Leitner, A.; Weix, D. J.; Ueno, S.; Hartwig, J. F. Org. Lett. 2007, 9, 3949–3952.
- (20) Zhang, M.; Guo, X. W.; Zheng, S. C.; Zhao, X. M. *Tetrahedron Lett.* 2012, 53, 6995–6998.

- (21) Pouy, M. J.; Stanley, L. M.; Hartwig, J. F. J. Am. Chem. Soc. 2009, 131, 11312–11313.
- (22) Weix, D. J.; Markovic, D.; Ueda, M.; Hartwig, J. F. Org. Lett. 2009, 11, 2944–2947.
- (23) Spiess, S.; Welter, C.; Franck, G.; Taquet, J. P.; Helmchen, G. Angew. Chem. Int.
 Ed. 2008, 47, 7652–7655.
- (24) Tosatti, P.; Horn, J.; Campbell, A. J.; House, D.; Nelson, A.; Marsden, S. P. Adv.
 Synth. Catal. 2010, 352, 3153–3157.
- (25) Ye, K.-Y.; Zhao, Z.-A.; Lai, Z.-W.; Dai, L.-X.; You, S.-L. Synth. 2013, 45, 2109–2114.
- (26) Stanley, L. M.; Hartwig, J. F. J. Am. Chem. Soc. 2009, 131, 8971–8983.
- (27) Stanley, L. M.; Hartwig, J. F. Angew. Chem. Int. Ed. 2009, 48, 7841–7844.
- (28) Liu, W.-B.; Zhang, X.; Dai, L.-X.; You, S.-L. Angew. Chem. Int. Ed. 2012, 51, 5183–5187.
- (29) Ye, K.-Y.; Dai, L.-X.; You, S.-L. Chem. Eur. J. 2014, 20, 3040–3044.
- (30) Sharma, A.; Hartwig, J. F. J. Am. Chem. Soc. 2013, 135, 17983–17989.
- (31) Campbell, A. N.; White, P. B.; Guzei, I. A.; Stahl, S. S. J. Am. Chem. Soc. 2010, 132, 15116–15119.
- (32) Zhang, X.; Yang, Z.-P.; Huang, L.; You, S.-L. Angew. Chem. Int. Ed. 2015, 54, 1873–1876.
- (33) Zhang, X.; Liu, W.-B.; Cheng, Q.; You, S.-L. Organometallics 2016, 35, 2467–2472.
- (34) Grange, R. L.; Clizbe, E. A.; Counsell, E. J.; Evans, P. A. Chem. Sci. 2015, 6, 777–781.
- (35) Kim, S. W.; Zhang, W.; Krische, M. J. Acc. Chem. Res. 2017, 50, 2371–2380. 830

- Meza, A. T.; Wurm, T.; Smith, L.; Kim, S. W.; Zbieg, J. R.; Stivala, C. E.;
 Krische, M. J. J. Am. Chem. Soc. 2018, 140, 1275–1279.
- (37) Kim, S. W.; Schwartz, L. A.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. J. Am. Chem. Soc. 2019, 141, 671–676.
- (38) Kim, S. W.; Schempp, T. T.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. Angew. Chem. Int. Ed. 2019, 58, 7762–7766.
- (39) Yamashita, Y.; Gopalarathnam, A.; Hartwig, J. F. J. Am. Chem. Soc. 2007, 129, 7508–7509.
- (40) Defieber, C.; Ariger, M. A.; Moriel, P.; Carreira, E. M. Angew. Chem. Int. Ed. 2007, 46, 3139–3143.
- (41) Lafrance, M.; Roggen, M.; Carreira, E. M. Angew. Chem. Int. Ed. 2012, 51, 3470–3473.
- (42) Sandmeier, T.; Goetzke, F. W.; Krautwald, S.; Carreira, E. M. J. Am. Chem. Soc.
 2019, 141, 12212–12218.
- (43) Miyabe, H.; Matsumura, A.; Moriyama, K.; Takemoto, Y. Org. Lett. 2004, 6, 4631–4634.
- (44) Miyabe, H.; Yoshida, K.; Reddy, V. K.; Takemoto, Y. J. Org. Chem. 2009, 74, 305–311.
- (45) Ye, K.-Y.; He, H.; Liu, W.-B.; Dai, L.-X.; Helmchen, G.; You, S.-L. J. Am. Chem. Soc. 2011, 133, 19006–19014.
- (46) Arnold, J. S.; Nguyen, H. M. J. Am. Chem. Soc. 2012, 134, 8380–8383.
- (47) Turnbull, B. W. H.; Evans, P. A. J. Org. Chem. 2018, 83, 11463–11479.
- (48) Vrieze, D. C.; Hoge, G. S.; Hoerter, P. Z.; Van Haitsma, J. T.; Samas, B. M. Org. Lett. 2009, 11, 3140–3142.

- (49) Arnold, J. S.; Cizio, G. T.; Heitz, D. R.; Nguyen, H. M. Chem. Commun. 2012, 48, 11531–11533.
- (50) Arnold, J. S.; Nguyen, H. M. Synth. 2013, 45 (15), 2101–2108.
- (51) Arnold, J. S.; Mwenda, E. T.; Nguyen, H. M. Angew. Chem. Int. Ed. 2014, 53, 3688–3692.
- (52) Mwenda, E. T.; Nguyen, H. M. Org. Lett. 2017, 19, 4814–4817.
- (53) Li, C.; Breit, B. Chem. Eur. J. 2016, 22, 14655–14663.
- (54) Zhou, Y.; Breit, B. Chem. Eur. J. 2017, 23, 18156–18160.
- (55) Cooke, M. L.; Xu, K.; Breit, B. Angew. Chem. Int. Ed. 2012, 51, 10876–10879.
- (56) Xu, K.; Thieme, N.; Breit, B. Angew. Chem. Int. Ed. 2014, 53, 2162–2165.
- (57) Li, C.; Kähny, M.; Breit, B. Angew. Chem. Int. Ed. 2014, 53, 13780–13784.
- (58) Schmidt, J. P.; Li, C.; Breit, B. Chem. Eur. J. 2017, 23, 6531-6534.
- (59) Haydl, A. M.; Xu, K.; Breit, B. Angew. Chem. Int. Ed. 2015, 54, 7149–7153.
- (60) Thieme, N.; Breit, B. Angew. Chem. Int. Ed. 2017, 56, 1520–1524.
- (61) Xu, K.; Raimondi, W.; Bury, T.; Breit, B. Chem. Commun. 2015, 51, 10861– 10863.
- (62) Zheng, J.; Wörl, B.; Breit, B. Eur. J. Org. Chem. 2019, 5180–5182.
- (63) Parveen, S.; Li, C.; Hassan, A.; Breit, B. Org. Lett. 2017, 19, 2326–2329.
- (64) Xu, K.; Gilles, T.; Breit, B. Nat. Commun. 2015, 6, 1–7.
- (65) Lumbroso, A.; Koschker, P.; Vautravers, N. R.; Breit, B. J. Am. Chem. Soc. 2011, 133, 2386–2389.
- (66) Chen, Q. A.; Chen, Z.; Dong, V. M. J. Am. Chem. Soc. 2015, 137, 8392–8395.
- (67) Yang, X. H.; Dong, V. M. J. Am. Chem. Soc. 2017, 139, 1774–1777.
- (68) Haydl, A. M.; Hilpert, L. J.; Breit, B. Chem. Eur. J. 2016, 22, 6547–6551.

- (69) Berthold, D.; Breit, B. Org. Lett. 2018, 20, 598–601.
- (70) Trost, B. M.; Bunt, R. C.; Lemoine, R. C.; Calkins, T. L. J. Am. Chem. Soc. 2000, 122, 5968–5976.
- (71) Trost, B. M.; Toste, F. D. J. Am. Chem. Soc. 1999, 121, 4545–4554.
- Mangion, I.; Strotman, N.; Drahl, M.; Imbriglio, J.; Guidry, E. Org. Lett. 2009, 11, 3258–3260.
- (73) Li, G.; Feng, X.; Du, H. Org. Biomol. Chem. 2015, 13, 5826–5830.
- (74) Azzouz, M.; Soriano, S.; Escudero-Casao, M.; Matheu, M. I.; Castillón, S.; Díaz,
 Y. Org. Biomol. Chem. 2017, 15, 7227–7234.
- (75) (a) Trost, B. M.; Dong, G. Org. Lett. 2007, 9, 2357–2359. (b) Trost, B. M.;
 Osipov, M.; Dong, G. J. Am. Chem. Soc. 2010, 132 (44), 15800–15807
- (76) Trost, B. M.; Fandrick, D. R. J. Am. Chem. Soc. 2003, 125, 11836–11837.
- (77) Khan, A.; Xing, J.; Zhao, J.; Kan, Y.; Zhang, W.; Zhang, Y. J. Chem. Eur. J. 2015, 21, 120–124.
- (78) Trost, B. M.; Aponick, A. J. Am. Chem. Soc. 2006, 128, 3931–3933.
- (79) (a) Trost, B. M.; Weber, L. J. Am. Chem. Soc. 1975, 97, 1611. (b) Trost, B. M.;
 Verhoeven, T. R. J. Am. Chem. Soc. 1976, 98, 630. (c) Hayashi, T.; Hagihara, T.;
 Konishi, M.; Kumada, M. J. Am. Chem. Soc. 1983, 105, 7767.
- (80) Quan, M.; Butt, N.; Shen, J.; Shen, K.; Liu, D.; Zhang, W. Org. Biomol. Chem.
 2013, 11, 7412–7419.
- (81) Cai, A.; Guo, W.; Martínez-Rodríguez, L.; Kleij, A. W. J. Am. Chem. Soc. 2016, 138, 14194–14197.
- (82) Hu, L.; Cai, A.; Wu, Z.; Kleij, A. W.; Huang, G. A Angew. Chem. Int. Ed. 2019, 58, 14694–14702.

- (83) Wang, Y. N.; Wang, B. C.; Zhang, M. M.; Gao, X. W.; Li, T. R.; Lu, L. Q.; Xiao,
 W. J. Org. Lett. 2017, 19, 4094–4097.
- (84) You, S.-L.; Zhu, X. Z.; Luo, Y. M.; Hou, X. L.; Dai, L.-X. J. Am. Chem. Soc. 2001, 123, 7471–7472.
- (85) Zheng, W. H.; Sun, N.; Hou, X. L. Org. Lett. 2005, 7, 5151–5154.
- (86) Zheng, B. H.; Ding, C. H.; Hou, X. L. Synlett 2011, 2262–2264.
- (87) Guo, W.; Cai, A.; Xie, J.; Kleij, A. W. Angew. Chem. Int. Ed. 2017, 56, 11797– 11801.
- (88) Wang, X.; Meng, F.; Wang, Y.; Han, Z.; Chen, Y. J.; Liu, L.; Wang, Z.; Ding, K. Angew. Chem. Int. Ed. 2012, 51, 9276–9282.
- (89) Wang, X.; Guo, P.; Han, Z.; Wang, X.; Wang, Z.; Ding, K. J. Am. Chem. Soc.
 2014, 136, 405–411.
- (90) Wang, H.; Yu, L.; Xie, M.; Wu, J.; Qu, G.; Ding, K.; Guo, H. Chem. Eur. J. 2018, 24, 1425–1430.
- (91) Xia, C.; Shen, J.; Liu, D.; Zhang, W. Org. Lett. 2017, 19, 4251–4254.
- (92) Soriano, S.; Escudero-Casao, M.; Matheu, M. I.; Díaz, Y.; Castillón, S. Adv. Synth. Catal. 2016, 358, 4057–4066.
- (93) Khan, S.; Shah, B. H.; Khan, I.; Li, M.; Zhang, Y. J. Chem. Commun. 2019, 55, 13168–13171.
- (94) Banerjee, D.; Junge, K.; Beller, M. Angew. Chem. Int. Ed. 2014, 53, 13049–13053.
- (95) Löber, O.; Kawatsura, M.; Hartwig, J. F. J. Am. Chem. Soc. 2001, 123, 4366–4367.
- (96) Johns, A. M.; Utsunomiya, M.; Incarvito, C. D.; Hartwig, J. F. J. Am. Chem. Soc.
 2006, 128, 1828–1839.

- (97) Adamson, N. J.; Hull, E.; Malcolmson, S. J. J. Am. Chem. Soc. 2017, 139, 7180–7183.
- (98) Park, S.; Malcolmson, S. J. ACS Catal. 2018, 8, 8468–8476.
- (99) Kim, H.; Lim, W.; Im, D.; Kim, D. G.; Rhee, Y. H. Angew. Chem. Int. Ed. 2012, 51, 12055–12058.
- (100) Kim, H.; Rhee, Y. H. J. Am. Chem. Soc. 2012, 134, 4011-4014.
- (101) Liu, J.; Han, Z.; Wang, X.; Wang, Z.; Ding, K. J. Am. Chem. Soc. 2015, 137, 15346–15349.
- (102) Matsushima, Y.; Onitsuka, K.; Kondo, T.; Mitsudo, T.; Takahashi, S. J. Am. Chem. Soc. 2001, 123, 10405–10406.
- (103) Kanbayashi, N.; Takenaka, K.; Okamura, T. A.; Onitsuka, K. Angew. Chem. Int.
 Ed. 2013, 52, 4897–4901.
- (104) Kawatsura, M.; Uchida, K.; Terasaki, S.; Tsuji, H.; Minakawa, M.; Itoh, T. Org.
 Lett. 2014, 16, 1470–1473.

- For selected reviews on the use of glycidic alcohols and aldehydes in chemical synthesis, see: (a) Hanson, R. M. Chem. Rev. 1991, 91, 437. (b) Lauret, C. *Tetrahedron: Asymmetry* 2001, 12, 2359. (c) Riera, A.; Moreno, M. Molecules 2010, 15, 1041.
- (2) For selected reviews on enantioselective carbonyl allylation, crotylation and related processes, see: (a) Ramachandran, P. V. Aldrichim. Acta 2002, 35, 23. (b) Denmark, S. E.; Fu, J. Chem. Rev. 2003, 103, 2763. (c) Yu, C.-M.; Youn, J.; Jung, H.-K. Bull. Korean Chem. Soc. 2006, 27, 463. (d) Marek, I.; Sklute, G. Chem. Commun. 2007, 1683. (e) Hall, D. G. Synlett 2007, 1644. (f) Hargaden, G. C.; Guiry, P. J. Adv. Synth. Catal. 2007, 349, 2407. (g) Lachance, H.; Hall, D. G. Org. React. 2008, 73, 1. (h) Han, S. B.; Kim, I. S.; Krische, M. J. Chem. Commun. 2009, 7278. (i) Yus, M.; González-Gómez, J. C.; Foubelo, F. Chem. Rev. 2011, 111, 7774. (j) Moran, J.; Krische, M. J. In Asymmetric Synthesis The Essentials II; Christmann, M., Bräse, S. Eds. Wiley-VCH: Weinheim, 2012, p 187. (k) Shin, I.; Krische, M. J. Top. Curr. Chem. 2016, 372, 85. (l) Feng, J.; Kasun, Z. A.; Krische, M. J. J. Am. Chem. Soc. 2016, 138, 5467.
- (3) For methodological studies of the allylboron of glycidic aldehydes, see: (a) Roush, W. R.; Hoong, L. K.; Palmer, M. A. J.; Straub, J. A.; Palkowitz, A. D. J. Org. Chem. 1990, 55, 4117. (b) Roush, W. R.; Straub, J. A.; VanNieuwenhze, M. S. J. Org. Chem. 1991, 56, 1636. (c) Nowrouzi, F.; Janetzko, J.; Batey, R. A. Org. Lett. 2010, 12, 5490.
- (4) For applications of glycidic aldehyde allylboration in natural product synthesis, see: (a) Murga, J.; García-Fortanet, J.; Carda, M.; Marco, J. A. Synlett. 2004, 15, 836

2830. (b) García-Fortanet, J.; Murga, J.; Carda, M.; Marco, J. A.; Matesanz, R.;
Díaz, J. F.; Barasoain, I. *Chem. Eur. J.* 2007, *13*, 5060. (c) Ichige, T.; Okano, Y.;
Kanoh, N.; Nakata, M. *J. Org. Chem.* 2009, *74*, 230. (d) Murata, T.; Sano, M.;
Takamura, H.; Kadota, I.; Uemura, D. *J. Org. Chem.* 2009, *74*, 4797. (e) Yadav, J.
S.; Sukant Kishore Das, S. K.; Sabitha, G. *J. Org. Chem.* 2012, *77*, 11109.

- (5) For methodological studies of the allylstannylation of glycidic aldehydes, see: (a) Howe, G. P.; Wang, S.; Procter, G. *Tetrahedron Lett.* 1987, 28, 2629. (b) Pradilla, R. F.; Castellanos, A.; Fernández, J.; Lorenzo, M.; Manzano, P.; Méndez, P.; Priego, J.; Viso, A. *J. Org. Chem.* 2006, 71, 1569.
- (6) For applications of glycidic aldehyde allylstannylation in natural product synthesis, see: (a) Jørgensen, K. B.; Suenaga, T.; Nakata, T. *Tetrahedron Lett.* **1999**, *40*, 8855. (b) Bartlett, L.; Gross, L.; Péron, F.; Asby, D. J.; Selby, M. D.; Tavassoli, A.; Linclau, B. *Chem. Eur. J.* **2014**, *20*, 3306.
- (7) For methodological studies of the allylsilylation of glycidic aldehydes, see: Wang,
 S.; Howe, G. P.; Mahal, R. S.; Procter, G. *Tetrahedron Lett.* 1992, *33*, 3351.
- (8) For applications of glycidic aldehyde allylsilylation in natural product synthesis, see: Crimmins, M. T.; Caussanel, F. J. Am. Chem. Soc. 2006, 128, 3128.
- (9) For methodological studies of the allylindiation of glycidic aldehydes, see: (a) Paquette, L. A.; Mitzel, T. M. *Tetrahedron Lett.* 1995, *36*, 6863. (b) Paquette, L. A.; Mitzel, T. M. *J. Am. Chem. Soc.* 1996, *118*, 1931.
- (10) For methodological studies of the allylmagnesiation of glycidic aldehydes, see:
 (a) Takeda, Y.; Matsumoto, T.; Sato, F. *J. Org. Chem.* 1986, *51*, 4728. (b) Urabe, H.; Matsuka, T.; Sato, F. *Tetrahedron Lett.* 1992, *33*, 4179.
- (11) A systematic investigation of match-mismatch effects (references 3a and 3b) reveal a 3:1 diastereoselectivity in the mismatched case.

- Martin, V. S. Woodard, S. S; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K.
 B. *J. Am. Chem. Soc.* **1981**, *103*, 6237.
- (13) For selected reviews on catalytic carbonyl addition *via* alcohol-mediated hydrogen auto-transfer, see: (a) Ketcham, J. M.; Shin, I.; Montgomery, T. P.; Krische, M. J. *Angew. Chem. Int. Ed.* 2014, *53*, 9142. (b) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Gao, X.; Itoh, T.; Krische, M. J. *Nat. Prod. Rep.* 2014, *31*, 504. (c) Perez, F.; Oda, S.; Geary, L. M.; Krische, M. J. *Top. Curr. Chem.* 2016, *374*, 365.
- (14) For iridium catalyzed enantioselective C-H allylation of primary alcohols, see: (a) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. J. Am. Chem. Soc. 2008, 130, 6340. (b) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. J. Am. Chem. Soc. 2008, 130, 14891. (c) Lu, Y.; Kim, I. S.; Hassan, A.; Del Valle, D. J.; Krische, M. J. Angew. Chem. Int. Ed. 2009, 48, 5018. (d) Hassan, A.; Lu, Y.; Krische, M. J. Org. Lett. 2009, 11, 3112. (e) Schmitt, D. C.; Dechert-Schmitt, A.-M. R.; Krische, M. J. Org. Lett. 2012, 14, 6302. (f) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Krische, M. J. Angew. Chem. Int. Ed. 2013, 52, 3195. (g) Shin, I.; Wang, G.; Krische, M. J. Chem. Eur. J. 2014, 13382. (h) Liang, T.; Zhang, W.; Krische M. J. J. Am. Chem. Soc. 2015, 137, 16024.
- (15) For iridium catalyzed enantioselective C-H crotylation of primary alcohols, see:
 (a) Kim, I. S.; Han, S. B.; Krische, M. J. J. Am. Chem. Soc. 2009, 131, 2514. (b)
 Gao, X.; Townsend, I. A.; Krische, M. J. J. Org. Chem. 2011, 76, 2350. (c) Gao,
 X.; Han, H.; Krische, M. J. J. Am. Chem. Soc. 2011, 133, 12795.
- (16) For ruthenium catalyzed enantioselective C-H crotylation of primary alcohols, see: (a) Zbieg, J. R.; Moran, J.; Krische, M. J. J. Am. Chem. Soc. 2011, 133, 10582. (b) Zbieg, J. R.; Yamaguchi, E.; McInturff, E. L.; Krische, M. J. Science 838

2012, *336*, 324. (c) McInturff, E. L.; Yamaguchi, E.; Krische, M. J. J. Am. Chem. Soc. **2012**, *134*, 20628. (d) Liang, T.; Zhang, W.; Chen, T.-Y.; Nguyen, K. D.; Krische, M. J. J. Am. Chem. Soc. **2015**, *137*, 13066.

- (17) Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. 1980, 102, 5974.
- (18) (a) Vigneron, J. P.; Dhaenens, M.; Horeau, A. *Tetrahedron* 1973, 29, 1055. (b) For a historical review, see: Heller, D.; Drexler, H.-J.; Fischer, C.; Buschmann, H.; Baumann, W.; Heller, B. *Angew. Chem. Int. Ed.* 2000, 39, 495. Enantiomeric excess of glycidols 2.2b was 99.5%. See Supporting Information for further experimental details.
- (19) (a) Sasaki, M.; Tanino, K.; Miyashita, M. Org. Lett. 2001, 3, 1765. (b) Shang, S.;
 Iwadare, H.; Macks, D. E.; Ambrosini, L. M.; Tan, D. S. Org. Lett. 2007, 9, 1895.
- (20) (a) Pettit, G. R.; Cichacz, Z. A.; Gao, F.; Boyd, M. R.; Schmidt, J. M. J. Chem. Soc., Chem. Commun. 1994, 1111. (b) Pettit, G. R.; Cichacz, Z. A. WO 5430053, 1995; Chem. Abstr. 1995, 733500. (c) Isbrucker, R. A.; Cummins, J.; Pomponi, S. A.; Longley, R. E.; Wright, A. E. Biochem. Pharmacol. 2003, 66, 75.
- (21) (a) Pfeiffer, B.; Kuzniewski, C. N.; Wullschleger, C.; Altmann, K.-H. Top. Curr. Chem. 2009, 286, 1. (b) Dalby, S. M.; Paterson, I. Curr. Opin. Drug Disc. Devel. 2010, 13, 777.
- (22) For a review on the metal catalyzed reductive coupling of olefin-derived nucleophiles, see: Nguyen, K. D.; Park, B. Y.; Luong, T.; Sato, H.; Garza, V. J.; Krische, M. J. Science 2016, 354, 300.

Chapter 2 Supporting Information

- Gao, Y.; Hanson, R. M.; Klunder, J. M.; Ko, S. Y.; Masamune, H.; Sharpless, K.
 B. *J. Am. Chem. Soc.* **1987**, *109*, 5765.
- (2) Hanson, R. M.; Sharpless, K. B. J. Org. Chem. 1986, 51, 1922.
- (3) Nacro, K.; Baltas, M.; Escudier, J. M.; Gorrichon, L. *Tetrahedron* 1996, 52, 9047.
- (4) Schomaker, J. M.; Pulgam, V. R.; Borhan, B. J. Am. Chem. Soc. 2004, 126, 13600.
- (5) Kavitha, N.; Kumar, V. P.; Chandrasekhar, S. *Tetrahedron Lett.* **2013**, *54*, 2128.
- (6) Minami, N.; Ko, S. S.; Kishi, Y. J. Am. Chem. Soc. **1982**, 104, 1109.
- (7) Rodriguez-Berrios, R. R.; Torres, G.; Prieto, J. A. *Tetrahedron* **2011**, *67*, 830.

- For selected reviews on naturally occuring cyclopropane see: (a) Donaldson, W.
 A. *Tetrahedron* 2001, 57, 8589. (b) Chen, D. Y.-K.; Pouwer, R. H.; Richard, J.-A.
 Chem. Soc. Rev. 2012, 41, 4631. (c) Keglevich, P.; Keglevich, A.; Hazai, L.;
 Kalaus, G.; Szántay, C. *Curr. Org. Chem.* 2014, 18, 2037.
- (2) For selected reviews on cyclopropane chemistry in the context of pharmaceutical, agrochemical and fragrance research, see: (a) Salaün, J. Top. Curr. Chem. 2000, 207, 1. (b) Wessjohann, L. A.; Brandt, W.; Thiemann, T. Chem. Rev. 2003, 103, 1625. (c) Singh, A. K.; Prasad, J. S.; Delaney, E. J. In Asymmetric Catalysis on Industrial Scale: Challenges, Approaches and Solutions; Blaser, H.-U., Schmidt, E., Eds.; Wiley-VCH: Weinheim, Germany, 2004; pp 335-348. (d) Gagnon, A.; Duplessis, M.; Fader, L. Org. Prep. Proc. Int. 2010, 42, 1; (e) Marson, C. M. Chem. Soc. Rev. 2011, 40, 5514. (f) Meanwell, N. A. J. Med. Chem. 2011, 54, 2529. (f) Schröder, F. Chem. Biodiversity 2014, 11, 1734. (g) Talele, T. T. J. Med. Chem. 2016, 59, 8712.
- (3) For selected reviews on the synthesis of enantiomerically enriched cyclopropanes, see: (a) Charette, A. B.; Beauchemin, A. Org. React. 2001, 58, 1. (b) DelMonte, A. J.; Dowdy, E. D.; Watson, D. J. Top. Organomet. Chem. 2004, 6, 97. (c) Pellissier, H. Tetrahedron 2008, 64, 7041. (d) Goudreau, S. R.; Charette, A. B. Angew. Chem. Int. Ed. 2010, 49, 486. (e) Bartoli, G.; Bencivenni, G.; Dalpozzo, R. Synthesis 2014, 46, 979.
- (4) For selected reviews on nickel catalyzed cross-coupling, see: (a) Netherton, M. R.; Fu, G. C. Adv. Synth. Catal. 2004, 346, 1525. (b) Rosen, B. M.; Quasdorf, K. W.; Wilson, D. A.; Zhang, N.; Resmerita, A.-M.; Garg, N. K.; Percec, V. Chem. 841

Rev. 2011, *111*, 1346. (c) Joshi-Pangu, A.; Biscoe, M. R. Synlett 2012, 23, 1103.
(d) Yamaguchi, J.; Muto, K.; Itami, K. *Eur. J. Org. Chem.* 2013, 19. (e) Tasker, S. Z.; Sandley, E. A.; Jamison, T. F. *Nature* 2014, 509, 299. (f) Tobisu, M.; Chatani, N. Acc. Chem. Res. 2015, 48, 1717. (g) Weix, D. J. Acc. Chem. Res. 2015, 48, 1767. (h) Tollefson, E. J.; Hanna, L. E.; Jarvo, E. R. Acc. Chem. Res. 2015, 48, 2344. (i) Gu, J.; Wang, X.; Xue, W.; Gong, H. Org. Chem. Front. 2015, 2, 1411.
(j) Chen, T.; Han, L.-B. Angew. Chem. Int. Ed. 2015, 54, 8600. (k) Tellis, J. C.; Kelly, C. B.; Primer, D. N.; Jouffroy, M.; Patel, N. R.; Molander, G. A. Acc. Chem. Res. 2016, 49, 1429. (l) Cavalcanti L. N; Molander, G. A. Top. Curr. Chem. 2016, 374, 39.

- (5) For nickel-catalyzed cross-electrophile reductive coupling of 2-aryl-4-halotetrahydropyrans to form cyclopropanes, see: (a) Tollefson, E. J.; Erickson, L. W.; Jarvo, E. R. J. Am. Chem. Soc. 2015, 137, 9760. (b) Erickson, L. W.; Lucas, E. L.; Tollefson, E. J.; Jarvo, E. R. J. Am. Chem. Soc. 2016, 138, 14006.
- (6) For selected reviews on C-C bond forming transfer hydrogenation, see: (a) Ketcham, J. M.; Shin, I.; Montgomery, T. P.; Krische, M. J. Angew. Chem., Int. Ed. 2014, 53, 9142. (b) Feng, J.; Kasun, Z. A.; Krische, M. J. J. Am. Chem. Soc. 2016, 138, 5467. (c) Sam, B.; Breit, B.; Krische, M. J. Angew. Chem. Int. Ed. 2015, 54, 3267.
- (7) (a) Feng, J.; Garza, V. J.; Krische, M. J. J. Am. Chem. Soc. 2014, 136, 8911. (b)
 Feng, J.; Noack, F.; Krische, M. J. J. Am. Chem. Soc. 2016, 138, 12364. (c) Guo,
 Y.-A.; Lee, W.; Krische, M. J. Chem. Eur. J. 2017, 23, 2557.
- (8) Olefin insertions to form 5-membered rings have been realized in nickel catalyzed Heck reactions of secondary benzylic ethers: Harris, M. R.; Konev, M. O.; Jarvo, E. R. J. Am. Chem. Soc. 2014, 136, 7825.

- (9) For a review, see: Jung, M. E. Synlett **1999**, 843.
- (10) For stereospecific nickel catalyzed benzylic cross-coupling of organoboron reagents, see: (a) Maity, P.; Shacklady-McAtee, D. M.; Yap, G. P. A.; Sirianni, E. R.; Watson, M. P. J. Am. Chem. Soc. 2013, 135, 280. (b) Harris, M. R.; Hanna, L. E.; Greene, M. A.; Moore, C. E.; Jarvo, E. R. J. Am. Chem. Soc. 2013, 135, 3303.
 (c) Zhou, Q.; Srinivas, H. D.; Dasgupta, S.; Watson, M. P. J. Am. Chem. Soc. 2013, 135, 3307. (d) Zhou, Q.; Cobb, K. M.; Tan, T.; Watson, M. P. J. Am. Chem. Soc. 2016, 138, 12057.
- (11) The requirement of naphthyl or furyl substituents in prior nickel catalyzed benzylic substitutions was rationalized on the basis of an S_N2' mechanism for leaving group ionization (refs. 5a, 10b-10d). We propose for less reactive leaving groups it is necessary to extend the lifetime of the η^2 -nickel π -complex that precedes benzylic ionization and, for extended aromatic π -systems, lower LUMO energies strengthen π -backbonding. More reactive carbonate leaving groups compensate for a shorter lifetime of the η^2 -nickel π -complex. Furyl substituted benzyl donors have weaker benzylic C-O bonds due to $\pi \rightarrow \sigma^*$ hyperconjugation. For discussion in the context of the ionization of allylic leaving groups, see: Hassan, A.; Townsend, I. A.; Krische, M. J. *Chem. Comm.* **2011**, 10028.
- (12) For diastereoselective olefin borocyclopropanation, see: (a) Takai, K.; Toshikawa,
 S.; Inoue, A.; Kokumai, R.; Hirano, M. J. Organomet. Chem. 2007, 692, 520. (b)
 Benoit, G.; Charette, A. B. J. Am. Chem. Soc. 2017, 139, 1364

Chapter 3 Supporting Information

- (1) Feng, J.; Garza, V. J.; Krische, M. J. J. Am. Chem. Soc. 2014, 136, 8911.
- (2) Antoft-Finch, A.; Blackburn, T.; Snieckus, V. J. Am. Chem. Soc. 2009, 131, 17750.
- (3) Molander, G. A.; Ryu, D. Angew. Chem. Int. Ed. 2014, 53, 14181.
- (4) Guo, Y.-A.; Lee, W.; Krische, M. J. Chem. Eur. J. 2017, 23, 2557.
- Huang, H.; Li, X.; Yu, C.; Zhang, Y.; Mariano, P. S. Wang, W. Angew. Chem.
 Int. Ed. 2017, 56, 1500.
- (6) Perez-Melero, C.; Maya, A. B.; del Rey, B.; Pelaez, R.; Caballero, E.; Medarde, M. *Bioorg. Med. Chem. Lett.* 2004, *14*, 3771.
- (7) Liu, Z. Q.; Zhang, Y.; Zhao, L.; Li, Z.; Wang, J.; Li, H.; Wu, L. M. Org. lett.
 2011, 13, 2208.

- (1) (a) Tsuji, J.; Takahashi, H.; Morikawa, M. *Tetrahedron Lett.* 1965, *6*, 4387. (b)
 Trost, B. M.; Fullerton, T. J. J. Am. Chem. Soc. 1973, 95, 292.
- (2) For selected reviews on metal catalyzed allylic substitution, see: (a) Trost, B. M.; Van Vranken, D. L. Chem. Rev. 1996, 96, 395. (b) Trost, B. M.; Crawley, M. L. Chem. Rev. 2003, 103, 2921. (c) Graening, T.; Schmalz, H.-G. Angew. Chem., Int. Ed. 2003, 42, 2580. (d) Trost, B. M. J. Org. Chem. 2004, 69, 5813. (e) Lu, Z.; Ma, S. Angew. Chem., Int. Ed. 2008, 47, 258. (f) Trost, B. M.; Zhang, T.; Sieber, J. D. Chem. Sci. 2010, 1, 427. (g) Tosatti, P.; Nelson, A.; Marsden, S. P. Org. Biomol. Chem. 2012, 10, 3147. (h) Moberg, C. Top. Organomet. Chem. 2011, 38, 209. (i) Sundararaju, B.; Achard, M.; Bruneau, C. Chem. Soc. Rev. 2012, 41, 4467. (j) Oliver, S.; Evans, P. A. Synthesis 2013, 45, 3179. (k) Butt, N. A.; Zhang, W. Chem. Soc. Rev. 2015, 44, 7929. (l) Trost, B. M. Tetrahedron 2015, 71, 5708.
- (3) (a) Takeuchi, R.; Ue, N.; Tanabe, K.; Yamashita, K.; Shiga, N. J. Am. Chem. Soc.
 2001, 123, 9525. (b) Onodera, G.; Watabe, K.; Matsubara, M.; Oda, K.; Kezuka, S.; Takeuchi, R. Adv. Synth. Catal. 2008, 2725.
- (4) Reviews: (a) Takeuchi, R. Synlett 2002, 1954. (b) Takeuchi, R.; Kezuka, S. Synthesis 2006, 3349.
- (5) (a) Bartels, B.; García-Yebra, C.; Rominger, F.; Helmchen, G. *Eur. J. Inorg. Chem.* 2002, 2569. (b) Lipowsky, G.; Helmchen, G. *Chem. Commun.* 2004, 116.
 (c) Welter, C.; Koch, O.; Lipowsky, G.; Helmchen, G. *Chem. Commun.* 2004, 896. (d) Welter, C.; Dahnz, A.; Brunner, B.; Streiff, S.; Dübon, P.; Helmchen, G. *Org. Lett.* 2005, 7, 1239. (e) Weihofen, R.; Dahnz, A.; Tverskoy, O.; Helmchen, G. *Chem. Commun.* 2004, 3541. (f) Weihofen, R.; Tverskoy, O.; Helmchen, G. 845

Angew. Chem. Int. Ed. 2006, 45, 5546. (g) Speiss, S.; Berthold, C.; Weihofen, R.,
Helmchen, G. Org. Biomol. Chem. 2007, 5, 2357. (h) Speiss, S.; Raskatov, J. A.;
Gnamm, C.; Brödner, K.; Helmchen, G. Chem. Eur. J. 2009, 15, 11087. (i)
Gärtner, M.; Jäkel, M.; Achatz, M.; SonnenSchein, C.; Tverskoy, O.; Helmchen,
G. Org. Lett. 2011, 13, 2810.

- (6) Reviews: (a) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. *Chem. Commun.* 2007, 675. (b) Helmchen, G. In *Iridium Complexes in Organic Synthesis*; Oro, L. A., Claver, C., Eds.; Wiley-VCH: Weinheim, 2009; pp 211-250. (c) Qu, J.; Helmchen, G. *Acc. Chem. Res.* 2017, *50*, 2539.
- (7) (a) Ohmura, T.; Hartwig, J. F. J. Am. Chem. Soc. 2002, 124, 15164. (b) Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. J. Am. Chem. Soc. 2003, 125, 14273. (c) Leitner, A.; Shu, C.; Hartwig, J. F. Proc. Nat. Acad. Sci. 2004, 101, 5830. (d) Shu, C.; Leitner, A.; Hartwig, J. F. Angew. Chem. Int. Ed. 2004, 43, 4797. (e) Leitner, A.; Shekhar, S.; Pouy, M. J.; Hartwig, J. F. J. Am. Chem. Soc. 2005, 127, 15506. (f) Leitner, A.; Shu, C.; Hartwig, J. F. Org. Lett. 2005, 7, 1093. (g) Shekhar, S.; Trantow, B.; Leitner, A.; Hartwig, J. F. J. Am. Chem. Soc. 2006, 128, 11770. (h) Yamashita, Y.; Gopalarathnam, A.; Hartwig, J. F. J. Am. Chem. Soc. 2007, 129, 11680. (j) Pouy, M. J.; Leitner, A.; Weix, D. J.; Ueno, S.; Hartwig, J. F. Org. Lett. 2007, 9, 3949.
- (8) Reviews: (a) Hartwig, J. F.; Stanley, L. M. Acc. Chem. Res. 2010, 43, 1461. (b)
 Hartwig, J. F.; Pouy, M. J. Top. Organomet. Chem. 2011, 38, 169.
- (9) (a) Defieber, C.; Ariger, M. A.; Moriel, P.; Carreira, E. M. Angew. Chem. Int. Ed.
 2007, 46, 3139. (b) Roggen, M.; Carreira, E. M. J. Am. Chem. Soc. 2010, 132, 11917. (c) Lafrance, M.; Roggen, M.; Carreira, E. M. Angew. Chem. Int. Ed. 846

2012, *51*, 3470. (d) Rossler, S. L.; Krautwald, S.; Carreira, E. M. J. Am. Chem. Soc. **2017**, *139*, 3603.

- (10) (a) Ye, K.-Y.; Dai, L.-X.; You, S.-L. Org. Biomol. Chem. 2012, 10, 5932. (b) Liu, W.-B.; Zhang, X.; Dai, L.-X.; You, S.-L. Angew. Chem. Int. Ed. 2012, 51, 5183.
 (c) Ye, K.-Y.; Zhao, Z.-A.; Lai, Z.-W.; Dai, L.-X.; You, S.-L. Synthesis 2013, 2109. (d) Ye, K.-Y.; Dai, L.-X.; You, S.-L. Chem. Eur. J. 2014, 20, 3040. (e) Yang, Z.-P.; Wu, Q.-F.; You, S.-L. Angew. Chem. Int. Ed. 2014, 53, 6986. (f) Zhang, X.; Yang, Z.-P.; Huang, L.; You, S.-L. Angew. Chem. Int. Ed. 2015, 54, 1873. (g) Yang, Z.-P.; Wu, Q.-F.; Shao, W.; You, S.-L. J. Am. Chem. Soc. 2015, 137, 15899. (f) Ye, K.-Y.; Cheng, Q.; Zhuo, C.-X.; Dai, L.-X.; You, S.-L. Angew. Chem. Int. Ed. 2016, 55, 8113. (g) Zhang, X.; Liu, W.-B.; Cheng, Q.; You, S.-L. Organometallics 2016, 35, 2467. (h) Yang, Z.-P.; Zheng, C.; Huang, L.; Qian, C.; You, S.-L. Angew. Chem. Int. Ed. 2017, 56, 1530.
- (11) (a) Liu, W.-B.; Xia, J.-B.; You, S.-L. Top. Organomet. Chem. 2011, 38, 155. (b)
 Zhou, C.-X.; Zhang, W.; You, S.-L. Angew. Chem. Int. Ed. 2012, 51, 12662. (c)
 Wu, W.-T.; Zhang, L.; You, S.-L. Chem. Soc. Rev. 2016, 45, 1570.
- (12) For spectroscopic and crystallographic evidence of a stable rhodium enyl complex and its role in enabling regio- and stereospecific rhodium catalyzed allylic substitution, respectively, see: (a) Lawson, D. N.; Osborn, J. A.; Wilkinson, G. J. *Chem. Soc. A.* 1966, 1733. (b) Tanaka, I.; Jin-no, N.; Kushida, T.; Tsutsui, N.; Ashida, T.; Suzuki, H.; Sakurai, H.; Moro-oka, Y.; Ikawa, T. *Bull. Chem. Soc. Jpn.* 1984, *56*, 657. (c) Evans, P. A.; Nelson, J. D. J. Am. Chem. Soc. 1998, *120*, 5581.
- (13) Beyond the rather complex effects of Lewis basic additives (refs. 5a, 6a vs 6b), highly π -acidic ligands are required to preserve stereospecificity in iridium 847
catalyzed reactions of branched allyl proelectrophiles: Bartels, B.; Helmchen, G. *Chem. Commun.* **1999**, 741.

- (14) For initial reports, see: (a) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. J. Am. Chem. Soc. 2008, 130, 6340. (b) Kim, I. S.; Han, S. B.; Krische, M. J. J. Am. Chem. Soc. 2009, 131, 2514.
- (15) For preparation and chromatographic purification of the iridium catalyst in 85% yield, see: Gao, X.; Townsend, I. A.; Krische, M. J. J. Org. Chem. 2011, 76, 2350.
- (16) For a recent overview of enantioselective allylations and propargylations catalyzed by π-allyliridium *C*,*O*-benzoates, see: Kim, S. W.; Zhang, W.; Krische, M. J. Acc. Chem. Res. 2017, 50, 2371.
- (17) For other enantioselective metal catalyzed allylic aminations employing diverse allyl proelectrophiles, see: (a) Cooke, M. L.; Xu, K.; Breit, B. Angew. Chem., Int. Ed. 2012, 51, 10876. (b) Arnold, J. S.; Nguyen, H. M. J. Am. Chem. Soc. 2012, 134, 8380. (c) Chen, Q.-A.; Chen, Z.; Dong, V. M. J. Am. Chem. Soc. 2015, 137, 8392. (d) Xu, K.; Wang, Y.-H.; Khakyzadeh, V.; Breit, B. Chem. Sci. 2016, 7, 3313. (e) Mwenda, E. T.; Nguyen, H. M. Org. Lett. 2017, 19, 4814. (f) Guo, W.; Cai, A.; Xie, J.; Kleij, A. W. Angew. Chem. Int. Ed. 2017, 56, 11797.
- (18) For selected reviews on the catalytic enantioselective synthesis of allylic amines, see: (a) Cannon, J. S.; Overman, L. E. Acc. Chem. Res. 2016, 49, 2220. (b) Grange, R. L.; Clizbe, E. A.; Evans, P. A. Synthesis 2016, 2911. (c) Fernandes, Rodney A.; Kattanguru, Pullaiah; Gholap, Sachin P.; Chaudhari, Dipali A. Org. Biom. Chem. 2017, 15, 2672.
- (19) An enantiomeric enrichment of 82% ee was observed using the SEGPHOS-modified catalyst. For the Roche ligand, see: (a) Schmid, R.; Cereghetti, M.; Heiser, B.; Schönholzer, P.; Hansen, H.-J. *Helv. Chim. Acta* 1988, 71, 897. (b) 848

Schmid, R.; Foricher, J.; Cereghetti, M.; Schönholzer, P. *Helv. Chim. Acta* **1991**, 74, 370.

- (20) (a) Vrienze, D. C.; Hohe, G. S.; Hoerter, P. Z.; van Haistma, J. T.; Samas, B. M. Org. Lett. 2009, 11, 3140. (b) Höcker, J.; Rudolf, G.; Bächlein, F.; Fleischer, S.; Lindner, B. D.; Helmchen, G. Eur. J. Org. Chem. 2013, 5149.
- (21) Lee, B. K.; Kim, M. S.; Nahm, H. S.; Kim, D. S.; Lee, W. K.; Ha, H.-J. *Tetrahedron* 2006, 62, 8393.
- (22) For a related RCM, see: Edwards, A. S.; Wybrow, R. A. J.; Johnstone, C.; Adams, H.; Harrity, J. P. A. *Chem. Commun.*, 2002, 1542.
- (23) For selected reviews on carbonyl allylation via umpolung of π-allyls, see: (a) Masuyama, Y. In Advances in Metal-Organic Chemistry; Liebeskind, L. S., Ed.; JAI Press: Greenwich, 1994; Vol. 3, p. 255. (b) Tamaru, Y. In Handbook of Organopalladium Chemistry for Organic Synthesis; Negishi, E.-i., de Meijere, A., Eds.; Wiley: New York, 2002; Vol. 2, pp. 1917. (c) Tamaru, Y. In Perspectives in Organopalladium Chemistry for the XXI Century; Tsuji, J. Ed.; Elsevier: Amsterdam, 1999, pp. 215. (d) Kondo, T.; Mitsudo, T.-A. Curr. Org. Chem. 2002, 6, 1163. (e) Tamaru, Y. Eur. J. Org. Chem. 2005, 13, 2647. (f) Zanoni, G.; Pontiroli, A.; Marchetti, A.; Vidari, G. Eur. J. Org. Chem. 2007, 22, 3599.
- (24) Amphiphilic properties are postulated in palladium catalyzed conjugate allylation-allylic alkylations of alkylidene malononitriles and related compounds, however, both allyltributylstannane and allylic chloride were required: (a) Nakamura, H.; Shim, J.-G.; Yamamoto, Y. J. Am. Chem. Soc. 1997, 119, 8113. For related processes, see: (b) Nakamura, H.; Aoyagi, K.; Shim, J.-G.; Yamamoto, Y. J. Am. Chem. Soc. 2001, 123, 372. (c) Nakamura, H.; Shimizu, K. Tetrahedron

Lett. 2011, 52, 426. (d) Genady, A. R.; Nakamura, H. Org. Biomol. Chem. 2011,
9, 7180. (e) Masakazu, N.; Atsushi, W.; Itami, K. Chem. Sci. 2012, 3, 3474.

(25) For statistical analysis of the appearance of *N*-heterocycles in FDA approved drugs, see: Vitaku, E.; Smith, D.T; Njardarson, J. T. *J. Med. Chem.* 2014, 57, 10257.

Chapter 4 Supporting Information

- (1) Gao, X.; Townsend, I. A.; Krische, M. J. J. Org. Chem. 2011, 76, 2350–2354
- (2) Crowley, P. J.; Mitchell, G.; Salmon, R.; Worthington, P. A. *Chimia* **2004**, *58*, 138–142.
- (3) Lambert, K. M.; Bobbitt, J. M.; Eldirany, S. A.; Kissane, L. E.; Sheridan, R. K.;
 Stempel, Z. D.; Sternberg, F. H.; Bailey, W. F. *Chem. Eur. J.* 2016, 22, 5156–5159.
- (4) Mukherjee, S.; Poon, K. W. C.; Flynn, D. L.; Hanson, P. R. *Tetrahedron Lett.*2003, 44, 7187–7190.
- (5) Moragas, T.; Cornella, J.; Martin, R. J. Am. Chem. Soc. 2014, 136, 17702–17705.
- (6) El-Hellani, A.; Bour, C.; Gandon, V. Adv. Synth. Catal. 2011, 353, 1865–1870.
- (7) Kraffta, M. E.; Sugiuraa, M.; Abboud, K. A. J. Am. Chem. Soc. 2001, 123, 9174– 9175.
- (8) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923–2925.

Chapter 5

- For selected reviews on alcohol-mediated carbonyl allylation, see: (a) Han, S. B.; Kim, I. S.; Krische, M. J. Enantioselective iridium-catalyzed carbonyl allylation from the alcohol oxidation level via transfer hydrogenation: minimizing preactivation for synthetic efficiency. *Chem. Commun.* 2009, 7278. (b) Ketcham, J. M.; Shin, I.; Montgomery, T. P.; Krische, M. J. Catalytic Enantioselective C-H Functionalization of Alcohols by Redox-Triggered Carbonyl Addition: Borrowing Hydrogen, Returning Carbon. *Angew. Chem. Int. Ed.* 2014, *53*, 9142. (c) Kim, S. W.; Zhang, W.; Krische, M. J. Catalytic Enantioselective Carbonyl Allylation and Propargylation via Alcohol-Mediated Hydrogen Transfer: Merging the Chemistry of Grignard and Sabatier. *Acc. Chem. Res.* 2017, *50*, 2371.
- (2) For a recent review on "umpoled allylations," see: Spielmann, K.; Niel, G.; de Figueiredo, R. M.; Campagne, J.-M. Catalytic nucleophilic 'umpoled' π-allyl reagents. *Chem. Soc. Rev.* 2018, 47, 1159.
- (3) For selected examples of alcohol-mediated carbonyl allylation involving noncyclometallated iridium catalysts, see: (a) Bower, J. F.; Skucas, E.; Patman, R. L.; Krische, M. J. Catalytic C-C Coupling via Transfer Hydrogenation: Reverse Prenylation, Crotylation, and Allylation from the Alcohol or Aldehyde Oxidation Level. J. Am. Chem. Soc. 2007, 129, 15134. (b) Bower, J. F.; Patman, R. L.; Krische, M. J. Iridium-Catalyzed C-C Coupling via Transfer Hydrogenation: Carbonyl Addition from the Alcohol or Aldehyde Oxidation Level Employing 1,3-Cyclohexadiene. Org. Lett. 2008, 10, 1033. (c) Nguyen, K. D.; Herkommer, D.; Krische, M. J. Enantioselective Formation of All-Carbon Quaternary Centers C-H Functionalization of Methanol: Iridium-Catalyzed via Diene 851

Hydrohydroxymethylation. *J. Am. Chem. Soc.* **2016**, *138*, 14210. (d) Holmes, M. T.; Nguyen, K. D.; Schwartz, L. A.; Luong, T.; Krische, M. J. Enantioselective Formation of CF₃-Bearing All-Carbon Quaternary Stereocenters via C–H Functionalization of Methanol: Iridium Catalyzed Allene Hydrohydroxymethylation. *J. Am. Chem. Soc.* **2017**, *139*, 8114.

- (4) (a) Takeuchi, R.; Ue, N.; Tanabe, K.; Yamashita, K.; Shiga, N. Iridium Complex-Catalyzed Allylic Amination of Allylic Esters. *J. Am. Chem. Soc.* 2001, *123*, 9525. (b) Onodera, G.; Watabe, K.; Matsubara, M.; Oda, K.; Kezuka, S.; Takeuchi, R. Iridium-Catalyzed Enantioselective Allylic Alkylation using Chiral Phosphoramidite Ligand Bearing an Amide Moiety. *Adv. Synth. Catal.* 2008, 2725.
- (5) Reviews: (a) Takeuchi, R. Iridium Complex-Catalyzed Highly Selective Organic Synthesis. Synlett 2002, 1954. (b) Takeuchi, R.; Kezuka, S. Iridium-Catalyzed Formation of Carbon-Carbon and Carbon-Heteroatom Bonds. Synthesis 2006, 3349.
- (6) (a) Bartels, B.; García-Yebra, C.; Rominger, F.; Helmchen, G. Iridium-Catalysed Allylic Substitution: Stereochemical Aspects and Isolation of Ir^{III} Complexes Related to the Catalytic Cycle. *Eur. J. Inorg. Chem.* 2002, 2569. (b) Lipowsky, G.; Helmchen, G. Regio- and enantioselective iridium-catalysed allylic aminations and alkylations of dienyl esters. *Chem. Commun.* 2004, 116. (c) Welter, C.; Koch, O.; Lipowsky, G.; Helmchen, G. First intramolecular enantioselective iridium-catalysed allylic aminations. *Chem. Commun.* 2004, 896. (d) Welter, C.; Dahnz, A.; Brunner, B.; Streiff, S.; Dübon, P.; Helmchen, G. Highly Enantioselective Syntheses of Heterocycles via Intramolecular Ir-Catalyzed Allylic Amination and Etherification. *Org. Lett.* 2005, *7*, 1239. (e) 852

Weihofen, R.; Dahnz, A.; Tverskoy, O.; Helmchen, G. Highly enantioselective iridium-catalysed allylic aminations with anionic N-nucleophiles. *Chem. Commun.* **2005**, 3541. (f) Weihofen, R.; Tverskoy, O.; Helmchen, G. Salt-Free Iridium-Catalyzed Asymmetric Allylic Aminations with N,N-Diacylamines and ortho-Nosylamide as Ammonia Equivalents. *Angew. Chem. Int. Ed.* **2006**, *45*, 5546. (g) Speiss, S.; Berthold, C.; Weihofen, R., Helmchen, G. Synthesis of α , β unsaturated γ -lactams via asymmetric iridium-catalysed allylic substitution. *Org. Biomol. Chem.* **2007**, *5*, 2357. (h) Speiss, S.; Raskatov, J. A.; Gnamm, C.; Brödner, K.; Helmchen, G. Ir-Catalyzed Asymmetric Allylic Substitutions with (Phosphoramidite)Ir Complexes—Resting States, Synthesis, and Characterization of Catalytically Active (π -Allyl)Ir Complexes. *Chem. Eur. J.* **2009**, *15*, 11087. (i) Gärtner, M.; Jäkel, M.; Achatz, M.; SonnenSchein, C.; Tverskoy, O.; Helmchen, G. Enantioselective Iridium-Catalyzed Allylic Substitutions with Hydroxamic Acid Derivatives as N-Nucleophiles. *Org. Lett.* **2011**, *13*, 2810.

- (7) Reviews: (a) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. Iridium-catalysed asymmetricallylic substitutions. *Chem. Commun.* 2007, 675.
 (b) Helmchen, G. In *Iridium Complexes in Organic Synthesis*; Oro, L. A., Claver, C., Eds.; Wiley-VCH: Weinheim, 2009; pp 211-250. (c) Qu, J.; Helmchen, G. Applications of Iridium-Catalyzed Asymmetric Allylic Substitution Reactions in Target-Oriented Synthesis. *Acc. Chem. Res.* 2017, *50*, 2539.
- (8) (a) Ohmura, T.; Hartwig, J. F. Regio- and Enantioselective Allylic Amination of Achiral Allylic Esters Catalyzed by an Iridium–Phosphoramidite Complex. *J. Am. Chem. Soc.* 2002, *124*, 15164. (b) Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. Identification of an Activated Catalyst in the Iridium-Catalyzed Allylic Amination and Etherification. Increased Rates, Scope, and Selectivity. *J. Am.* 853

Chem. Soc. 2003, 125, 14272. (c) Leitner, A.; Shu, C.; Hartwig, J. F. Editing the stereochemical elements in an iridium catalyst for enantioselective allylic amination. Proc. Nat. Acad. Sci. 2004, 101, 5830. (d) Shu, C.; Leitner, A.; Hartwig, J. F. Enantioselective Allylation of Aromatic Amines after In Situ Generation of an Activated Cyclometalated Iridium Catalyst. Angew. Chem. Int. Ed. 2004, 43, 4797. (e) Leitner, A.; Shekhar, S.; Pouy, M. J.; Hartwig, J. F. A Simple Iridium Catalyst with a Single Resolved Stereocenter for Enantioselective Allylic Amination. Catalyst Selection from Mechanistic Analysis. J. Am. Chem. Soc. 2005, 127, 15506. (f) Leitner, A.; Shu, C.; Hartwig, J. F. Effects of Catalyst Activation and Ligand Steric Properties on the Enantioselective Allylation of Amines and Phenoxides. Org. Lett. 2005, 7, 1093. (g) Shekhar, S.; Trantow, B.; Leitner, A.; Hartwig, J. F. Sequential Catalytic Isomerization and Allylic Substitution. Conversion of Racemic Branched Allylic Carbonates to Enantioenriched Allylic Substitution Products. J. Am. Chem. Soc. 2006, 128, 11770. (h) Yamashita, Y.; Gopalarathnam, A.; Hartwig, J. F. Iridium-Catalyzed, Asymmetric Amination of Allylic Alcohols Activated by Lewis Acids. J. Am. Chem. Soc. 2007, 129, 7508. (i) Markovic, D.; Hartwig, J. F. Resting State and Kinetic Studies on the Asymmetric Allylic Substitutions Catalyzed by Iridium-Phosphoramidite Complexes. J. Am. Chem. Soc. 2007, 129, 11680. (j) Pouy, M. J.; Leitner, A.; Weix, D. J.; Ueno, S.; Hartwig, J. F. Enantioselective Iridium-Catalyzed Allylic Amination of Ammonia and Convenient Ammonia Surrogates. Org. Lett. 2007, 9, 3949.

(9) Reviews: (a) Hartwig, J. F.; Stanley, L. M. Mechanistically Driven Development of Iridium Catalysts for Asymmetric Allylic Substitution. *Acc. Chem. Res.* 2010, *43*, 1461. (b) Hartwig, J. F.; Pouy, M. J. *Top. Organomet. Chem.* 2011, *38*, 169. 854

- (10) (a) Defieber, C.; Ariger, M. A.; Moriel, P.; Carreira, E. M. Iridium-Catalyzed Synthesis of Primary Allylic Amines from Allylic Alcohols: Sulfamic Acid as Ammonia Equivalent. *Angew. Chem. Int. Ed.* 2007, *46*, 3139. (b) Roggen, M.; Carreira, E. M. Stereospecific Substitution of Allylic Alcohols To Give Optically Active Primary Allylic Amines: Unique Reactivity of a (P,alkene)Ir Complex Modulated by Iodide. *J. Am. Chem. Soc.* 2010, *132*, 11917. (c) Lafrance, M.; Roggen, M.; Carreira, E. M. Direct, Enantioselective Iridium-Catalyzed Allylic Amination of Racemic Allylic Alcohols. *Angew. Chem. Int. Ed.* 2012, *51*, 3470. (d) Rossler, S. L.; Krautwald, S.; Carreira, E. M. Study of Intermediates in Iridium–(Phosphoramidite,Olefin)-Catalyzed Enantioselective Allylic Substitution. *J. Am. Chem. Soc.* 2017, *139*, 3603.
- (11)(a) Ye, K.-Y.; Dai, L.-X.; You, S.-L. Enantioselective synthesis of 2,5dihydrobenzo[b]azepine derivatives via iridium-catalyzed asymmetric allylic amination with 2-allylanilines and ring-closing-metathesis reaction. Org. Biomol. Chem. 2012, 10, 5932. (b) Liu, W.-B.; Zhang, X.; Dai, L.-X.; You, S.-L. Asymmetric N-Allylation of Indoles Through the Iridium-Catalyzed Allylic Alkylation/Oxidation of Indolines. Angew. Chem. Int. Ed. 2012, 51, 5183. (c) Ye, K.-Y.; Zhao, Z.-A.; Lai, Z.-W.; Dai, L.-X.; You, S.-L. Highly Regioselective Allylic Substitution Reactions Catalyzed by an Air-Stable (π -Allyl)iridium Complex Derived from Dinaphthocyclooctatetraene and a Phosphoramidite Ligand. Synthesis 2013, 2109. (d) Ye, K.-Y.; Dai, L.-X.; You, S.-L. Regio- and Enantioselective Synthesis of N-Allylindoles by Iridium-Catalyzed Allylic Amination/Transition-Metal-Catalyzed Cyclization Reactions. Chem. Eur. J. 2014, 20, 3040. (e) Yang, Z.-P.; Wu, Q.-F.; You, S.-L. Direct Asymmetric Dearomatization of Pyridines and Pyrazines by Iridium-Catalyzed Allylic 855

Amination Reactions. Angew. Chem. Int. Ed. 2014, 53, 6986. (f) Zhang, X.; Yang, Z.-P.; Huang, L.; You, S.-L. Highly Regio- and Enantioselective Synthesis of N-Substituted 2-Pyridones: Iridium-Catalyzed Intermolecular Asymmetric Allylic Amination. Angew. Chem. Int. Ed. 2015, 54, 1873. (g) Yang, Z.-P.; Wu, Q.-F.; Shao, W.; You, S.-L. Iridium-Catalyzed Intramolecular Asymmetric Allylic Dearomatization Reaction of Pyridines, Pyrazines, Quinolines, and Isoquinolines. J. Am. Chem. Soc. 2015, 137, 15899. (h) Ye, K.-Y.; Cheng, Q.; Zhuo, C.-X.; Dai, L.-X.; You, S.-L. An Iridium(I) N-Heterocyclic Carbene Complex Catalyzes Asymmetric Intramolecular Allylic Amination Reactions. Angew. Chem. Int. Ed. 2016, 55, 8113. (i) Zhang, X.; Liu, W.-B.; Cheng, Q.; You, S.-L. Iridium-Amination Catalyzed Asymmetric with Allylic Reactions N-Aryl Phosphoramidite Ligands. Organometallics 2016, 35, 2467. (j) Yang, Z.-P.; Zheng, C.; Huang, L.; Qian, C.; You, S.-L. Iridium-Catalyzed Intramolecular Asymmetric Allylic Dearomatization Reaction of Benzoxazoles, Benzothiazoles, and Benzimidazoles. Angew. Chem. Int. Ed. 2017, 56, 1530.

- (12) (a) Liu, W.-B.; Xia, J.-B.; You, S.-L. Iridium-Catalyzed Asymmetric Allylic Substitutions. *Top. Organomet. Chem.* 2011, *38*, 155. (b) Zhou, C.-X.; Zhang, W.; You, S.-L. Catalytic Asymmetric Dearomatization Reactions. *Angew. Chem. Int. Ed.* 2012, *51*, 12662. (c) Wu, W.-T.; Zhang, L.; You, S.-L. Catalytic asymmetric dearomatization (CADA) reactions of phenol and aniline derivatives. *Chem. Soc. Rev.* 2016, *45*, 1570.
- (13) Bartels, B.; Helmchen, G. Ir-catalysed allylic substitution: mechanistic aspects and asymmetric synthesis with phosphorus amidites as ligands. *Chem. Commun.* 1999, 741.

- (14) For spectroscopic and crystallographic evidence of a stable rhodium enyl complex and its role in enabling regio- and stereospecific rhodium catalyzed allylic substitution, respectively, see: (a) Lawson, D. N.; Osborn, J. A.; Wilkinson, G. Interaction of tris(triphenylphosphine)chlororhodium(I) with iodomethane, methylallyl, and allyl chloride. *J. Chem. Soc. A.* **1966**, 1733. (b) Tanaka, I.; Jinno, N.; Kushida, T.; Tsutsui, N.; Ashida, T.; Suzuki, H.; Sakurai, H.; Moro-oka, Y.; Ikawa, T. Crystal Structures of (1,5-Cyclooctadiene)di-µ-methoxo-dirhodium(I) and Tetrakis(η³-allyl)di-µ-hydroxo-dirhodium(III). *Bull. Chem. Soc. Jpn.* **1983**, *56*, 657. (c) Evans, P. A.; Nelson, J. D. Conservation of Absolute Configuration in the Acyclic Rhodium-Catalyzed Allylic Alkylation Reaction: Evidence for an *Enyl* (σ + π) Organorhodium Intermediate. *J. Am. Chem. Soc.* **1998**, *120*, 5581.
- (15) Meza, A. T.; Wurm, T.; Smith, L.; Kim, S. W.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. Amphiphilic π-Allyliridium C,O-Benzoates Enable Regio- and Enantioselective Amination of Branched Allylic Acetates Bearing Linear Alkyl Groups. J. Am. Chem. Soc. 2018, 140, 1275.
- (16) For selected examples of catalytic enantioselective allylic aminations beyond iridium, see: (a) Cooke, M. L.; Xu, K.; Breit, B. Enantioselective Rhodium-Catalyzed Synthesis of Branched Allylic Amines by Intermolecular Hydroamination of Terminal Allenes. *Angew. Chem. Int. Ed.* 2012, *51*, 10876. (b) Arnold, J. S.; Nguyen, H. M. Rhodium-Catalyzed Dynamic Kinetic Asymmetric Transformations of Racemic Tertiary Allylic Trichloroacetimidates with Anilines. *J. Am. Chem. Soc.* 2012, *134*, 8380. (c) Chen, Q.-A.; Chen, Z.; Dong, V. M. Rhodium-Catalyzed Enantioselective Hydroamination of Alkynes with Indolines. *J. Am. Chem. Soc.* 2015, *137*, 8392. (d) Xu, K.; Wang, Y.-H.; Khakyzadeh, V.; 857

Breit, B. Asymmetric synthesis of allylic amines *via* hydroamination of allenes with benzophenone imine. *Chem. Sci.* **2016**, *7*, 3313. (e) Mwenda, E. T.; Nguyen, H. M. Enantioselective Synthesis of 1,2-Diamines Containing Tertiary and Quaternary Centers through Rhodium-Catalyzed DYKAT of Racemic Allylic Trichloroacetimidates. *Org. Lett.* **2017**, *19*, 4814. (f) Guo, W.; Cai, A.; Xie, J.; Kleij, A. W. Asymmetric Synthesis of α,α -Disubstituted Allylic Amines through Palladium-Catalyzed Allylic Substitution. *Angew. Chem. Int. Ed.* **2017**, *56*, 11797.

- (17) For selected reviews on the catalytic enantioselective synthesis of allylic amines, see: (a) Cannon, J. S.; Overman, L. E. Palladium(II)-Catalyzed Enantioselective Reactions Using COP Catalysts. *Acc. Chem. Res.* 2016, *49*, 2220. (b) Grange, R. L.; Clizbe, E. A.; Evans, P. A. Recent Developments in Asymmetric Allylic Amination Reactions. *Synthesis* 2016, 2911. (c) Mailyan, A. K.; Eickhoff, J. A.; Minakova, A. S.; Gu, Z.; Lu, P.; Zakarian, A. Cutting-Edge and Time-Honored Strategies for Stereoselective Construction of C–N Bonds in Total Synthesis. *Chem. Rev.* 2016, *116*, 4441. (d) Fernandes, Rodney A.; Kattanguru, Pullaiah; Gholap, Sachin P.; Chaudhari, Dipali A. Recent advances in the Overman rearrangement: synthesis of natural products and valuable compounds. *Org. Biomol. Chem.* 2017, *15*, 2672.
- (18) Madrahimov, S. T.; Markovic, D.; Hartwig, J. F. The Allyl Intermediate in Regioselective and Enantioselective Iridium-Catalyzed Asymmetric Allylic Substitution Reactions. J. Am. Chem. Soc. 2009, 131, 7228.
- (19) For related deuterium labelling experiments, see: (a) Zhang, P.; Brozek, L. A.;
 Morken, J. P. Pd-Catalyzed Enantioselective Allyl–Allyl Cross-Coupling. J. Am. *Chem. Soc.* 2010, 132, 10686. (b) Chen, J.-P.; Peng, Q.; Lei, B.-L.; Hou, X.-L.; 858

Wu, Y.-D. Chemo- and Regioselectivity-Tunable Pd-Catalyzed Allylic Alkylation of Imines. *J. Am. Chem. Soc.* **2011**, *133*, 14180.

(20) For a comparison of 4 closely related π -allyliridium *C,O*-benzoate complexes, see: Schmitt, D. C.; Dechert-Schmitt, A.-M. R.; Krische, M. J. Iridium-Catalyzed Allylation of Chiral β -Stereogenic Alcohols: Bypassing Discrete Formation of Epimerizable Aldehydes. *Org. Lett.* **2012**, *14*, 6302.

Chapter 5 Supporting Information

- (1) Feng, J.; Garza, V. J.; Krische, M. J. J. Am. Chem. Soc. 2014, 136, 8911.
- (2) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.
- (3) Tsutsumi, R.; Hong, S.; Krische, M. J. Chem. Eur. J. 2015, 21, 12903.
- (4) Chen, M. S.; Prabagaran, N.; Labenz, N. A.; White, M. C. J. Am. Chem. Soc. 2005, 127, 6970.
- (5) Moragas, T.; Cornella, J.; Martin, R. J. Am. Chem. Soc. 2014, 136, 17702.

Chapter 6

- a) R. Takeuchi, N. Ue, K. Tanabe, K. Yamashita, N. Shiga, J. Am. Chem. Soc.
 2001, 123, 9525-9534; b) G. Onodera, K. Watabe, M. Matsubara, K. Oda, S. Kezuka, R. Takeuchi, Adv. Synth. Catal. **2008**, 2725-2732.
- Reviews: a) R. Takeuchi, Synlett 2002, 1954-1965; b) R. Takeuchi, S. Kezuka, Synthesis 2006, 3349-3366.
- (3) a) B. Bartels, C. García-Yebra, F. Rominger, G. Helmchen, *Eur. J. Inorg. Chem.* 2002, 2569-2586; b) G. Lipowsky, G. Helmchen, *Chem. Commun.* 2004, 116-117; c) C. Welter, O. Koch, G. Lipowsky, G. Helmchen, *Chem. Commun.* 2004, 896-897; d) C. Welter, A. Dahnz, B. Brunner, S. Streiff, P. Dübon, G. Helmchen, *Org. Lett.* 2005, *7*, 1239-1242; e) R. Weihofen, A. Dahnz, O. Tverskoy, G. Helmchen, *Chem. Commun.* 2004, 3541-3543; f) R. Weihofen, O. Tverskoy, G. Helmchen, *Angew. Chem. Int. Ed.* 2006, *45*, 5546-5549; *Angew. Chem.* 2006, *118*, 5673-5676; g) S. Speiss, C. Berthold, R. Weihofen, G. Helmchen, *Org. Biomol. Chem.* 2007, *5*, 2357-2360; h) S. Speiss, J. A. Raskatov, C. Gnamm, K. Brödner, G. Helmchen, *Chem. Eur. J.* 2009, *15*, 11087-11090; i) M. Gärtner, M. Jäkel, M. Achatz, C. SonnenSchein, O. Tverskoy, G. Helmchen, *Org. Lett.* 2011, *13*, 2810-2813.
- (4) Reviews: a) G. Helmchen, A. Dahnz, P. Dübon, M. Schelwies, R. Weihofen, *Chem. Commun.* 2007, 675-691; b) G. Helmchen, In *Iridium Complexes in Organic Synthesis*; Oro, L. A., Claver, C., Eds.; Wiley-VCH: Weinheim, 2009; pp 211-250; c) J. Qu, G. Helmchen, *Acc. Chem. Res.* 2017, *50*, 2539-2555.
- (5) a) T. Ohmura, J. F. Hartwig, J. Am. Chem. Soc. 2002, 124, 15164-15165; b) C. A. Kiener, C. Shu, C. Incarvito, J. F. Hartwig, J. Am. Chem. Soc. 2003, 125, 14272-860

14273; c) A. Leitner, C. Shu, J. F. Hartwig, *Proc. Nat. Acad. Sci.* 2004, 101, 5830-5833; d) C. Shu, A. Leitner, J. F. Hartwig, *Angew. Chem. Int. Ed.* 2004, 43, 4797-4800; *Angew. Chem.* 2004, 116, 4901-4904; e) A. Leitner, S. Shekhar, M. J. Pouy, J. F. Hartwig, *J. Am. Chem. Soc.* 2005, 127, 15506-15514; f) A. Leitner, C. Shu, J. F. Hartwig, *Org. Lett.* 2005, 7, 1093-1096; g) S. Shekhar, B. Trantow, A. Leitner, J. F. Hartwig, *J. Am. Chem. Soc.* 2006, 128, 11770-11771; h) Y. Yamashita, A. Gopalarathnam, J. F. Hartwig, *J. Am. Chem. Soc.* 2007, 129, 7508-7509; i) D. Markovic, J. F. Hartwig, *J. Am. Chem. Soc.* 2007, 129, 11680-11681; j) M. J. Pouy, A. Leitner, D. J. Weix, S. Ueno, J. F. Hartwig, *Org. Lett.* 2007, 9, 3949-3952.

- (6) Reviews: a) J. F. Hartwig, L. M. Stanley, *Acc. Chem. Res.* 2010, *43*, 1461-1475;
 b) J. F. Hartwig, M. J. Pouy, *Top. Organomet. Chem.* 2011, *38*, 169-208.
- (7) a) C. Defieber, M. A. Ariger, P. Moriel, E. M. Carreira, *Angew. Chem. Int. Ed.* 2007, 46, 3139-3143; *Angew. Chem.* 2007, *119*, 3200-3204; b) M. Roggen, E. M. Carreira, *J. Am. Chem. Soc.* 2010, *132*, 11917-11919; c) M. Lafrance, M. Roggen, E. M. Carreira, *Angew. Chem. Int. Ed.* 2012, *51*, 3470-3473; *Angew. Chem.* 2012, *124*, 3527-3530; d) S. L. Rossler, S. Krautwald, E. M. Carreira, *J. Am. Chem. Soc.* 2017, *139*, 3603-3606.
- (8) a) K.-Y. Ye, L.-X. Dai, S.-L. You, Org. Biomol. Chem. 2012, 10, 5932-5939; b)
 W.-B. Liu, X. Zhang, L.-X. Dai, S.-L. You, Angew. Chem. Int. Ed. 2012, 51, 5183-5187; Angew. Chem. 2012, 124, 5273-5277; c) K.-Y. Ye, Z.-A. Zhao, Z.-W. Lai, L.-X. Dai, S.-L. You, Synthesis 2013, 2109-2114; d) K.-Y. Ye, L.-X. Dai, S.-L. You, Chem. Eur. J. 2014, 20, 3040-3044; e) Z.-P. Yang, Q.-F. Wu, S.-L. You, Angew. Chem. Int. Ed. 2014, 53, 6986-6989; Angew. Chem. 2014, 126, 7106-7109; f) X. Zhang, Z.-P. Yang, L. Huang, S.-L. You, Angew. Chem. Int. Ed. 2015, 861

54, 1873-1876; Angew. Chem. 2015, 127, 1893-1896; g) Z.-P. Yang, Q.-F. Wu,
W. Shao, S.-L. You, J. Am. Chem. Soc. 2015, 137, 15899-15906; h) K.-Y. Ye, Q.
Cheng, C.-X. Zhuo, L.-X. Dai, S.-L. You, Angew. Chem. Int. Ed. 2016, 55, 81138116; Angew. Chem. 2016, 128, 8245-8248; i) X. Zhang, W.-B. Liu, Q. Cheng,
S.-L. You, Organometallics 2016, 35, 2467-2472; j) Z.-P. Yang, C. Zheng, L.
Huang, C. Qian, S.-L. You, Angew. Chem. Int. Ed. 2017, 56, 1530-1534; Angew.
Chem. 2017, 129, 1552-1556.

- (9) a) W.-B. Liu, J.-B. Xia, S.-L. You, *Top. Organomet. Chem.* 2011, 38, 155-208; b)
 C.-X. Zhou, W. Zhang, S.-L. You, *Angew. Chem. Int. Ed.* 2012, 51, 12662-12686; *Angew. Chem.* 2012, 124, 12834-12858; c) W.-T. Wu, L. Zhang, S.-L. You, *Chem. Soc. Rev.* 2016, 45, 1570-1580.
- (10) For selected reviews on palladium catalyzed allylic alkylation, see: a) B. M. Trost,
 D. L. Van Vranken, *Chem. Rev.* **1996**, *96*, 395; b) B. M. Trost, M. L. Crawley, *Chem. Rev.* **2003**, *103*, 2921-2944; c) T. Graening, H.-G. Schmalz, *Angew. Chem. Int. Ed.* **2003**, *42*, 2580-2584; *Angew. Chem.* **2003**, *115*, 2684-2688; d) B. M.
 Trost, *J. Org. Chem.* **2004**, *69*, 5813-5837; e) B. M. Trost, T. Zhang, J. D. Sieber, *Chem. Sci.* **2010**, *1*, 427-440; f) B. M. Trost, *Tetrahedron* **2015**, *71*, 5708-5733.
- (11) For selected reviews on enantioselective alcohol-mediated carbonyl allylation, see: a) S. B. Han, I. S. Kim, M. J. Krische, *Chem. Commun.* 2009, 7278-7287; b) J. M. Ketcham, I. Shin, T. P. Montgomery, M. J. Krische, *Angew. Chem. Int. Ed.* 2014, *53*, 9142-9150; *Angew. Chem.* 2014, *126*, 9294-9302; c) S. W. Kim, W. Zhang, M. J. Krische, *Acc. Chem. Res.* 2017, *50*, 2371-2380.
- (12) For a recent review on "umpoled allylations," see: K. Spielmann, G. Niel, R. M. de Figueiredo, J.-M. Campagne, *Chem. Soc. Rev.* 2018, 47, 1159-1173.

- (13)a) A. T. Meza, T. Wurm, L. Smith, S. W. Kim, J. R. Zbieg, C. E. Stivala, M. J. Krische, J. Am. Chem. Soc. 2018, 140, 1275-1279; b) S. W. Kim, L. A. Schwartz, J. R. Zbieg, C. E. Stivala, M. J. Krische, J. Am. Chem. Soc. 2019, 141, 671-676.
- (14)For our initial report on the synthesis and chromatographic purification of π allyliridium-C,O-benzoate catalysts, see: X. Gao, I. A. Townsend, M. J. Krische, J. Org. Chem. 2011, 76, 2350-2354.
- (15)R. D. Taylor, M. MacCoss, A. D. G. Lawson, J. Med. Chem. 2014, 57, 5845-5859.
- (16)For WIN55212 and CPI-169, see: a) T. E. D'Ambra, K. G. Estep, M. R. Bell, M. A. Eissenstat, K. A. Josef, S. J. Ward, D. A. Haycock, E. R. Baizman, F. M. Casiano, N.C. Beglin, S. M. Chippari, J. D. Grego, R. K. Kullnig, G. T. Daley, J. Med. Chem. 1992, 35, 124-135; b) V. S. Gehling, R. G. Vaswani, C. G. Nasveschuk, M. Duplessis, P. Iyer, S. Balasubramanian, F. Zhao, A. C. Good, R. Campbell, C. Lee, L. A. Dakin, A. S. Cook, A. Gagnon, J.-C. Harmange, J. E. Audia, R. T. Cummings, E. Normant, P. Trojer, B. K. Albrecht, Bioorg. Med. Chem. Lett. 2015, 25, 3644-3649; c) R. G. Vaswani, V. S. Gehling, L. A. Dakin, A. S. Cook, C. G. Nasveschuk, M. Duplessis, P. Iyer, S. Balasubramanian, F. Zhao, A. C. Good, R. Campbell, C. Lee, N. Cantone, R. T. Cummings, E. Normant, S. F. Bellon, B. K. Albrecht, J.-C. Harmange, P. Trojer, J. E. Audia, Y. Zhang, N. Justin, S. Chen, J. R. Wilson, S. J. Gamblin, J. Med. Chem. 2016, 59, 9928-9941.
- (17)Intermolecular palladium-catalyzed indole allylation: a) B. M. Trost, M. J. Krische, V. Berl, E. M. Grenzer, Org. Lett. 2002, 4, 2005-2008; b) S. Ma, S. Yu, Tetrahedron Lett. 2004, 45, 8419-8422; c) M. Bandini, A. Melloni, A. Umani-Ronchi, Org. Lett. 2004, 6, 3199-3202; d) M. Kimura, M. Futamata, R. Mukai, Y.

- Tamaru, J. Am. Chem. Soc. 2005, 127, 4592-4593; e) B. M. Trost, J. Quancard, J. Am. Chem. Soc. 2006, 128, 6314-6315; f) H. Y. Cheung, W.-Y. Yu, F. L. Lam, T. T.-L. Au-Yeung, Z. Zhou, T. H. Chan, A. S. C. Chan, Org. Lett. 2007, 9, 4295-4298; g) I. Usui, S. Schmidt, M. Keller, B. Breit, Org. Lett. 2008, 10, 1207-1210; h) N. Kagawa, J. P. Malerich, V. H. Rawal, Org. Lett. 2008, 10, 2381-2384; i) M. R. Luzung, C. A. Lewis, P. S. Baran, Angew. Chem. Int. Ed. 2009, 48, 7025-7029; Angew. Chem. 2009, 121, 7159-7163; j) B. M. Trost, M. Osipov, G. Dong, J. Am. Chem. Soc. 2010, 132, 15800-15807; k) T. Hoshi, K. Sasaki, S. Sato, Y. Ishii, T. Suzuki, H. Hagiwara, Org. Lett. 2011, 13, 932-935; 1) Y. Liu, H. Du, Org. Lett. 2013, 15, 740-743; m) L. Y. Chen, X.-Y. Yu, J.-R. Chen, B. Feng, H. Zhang, Y.-H. Qi, W.-J. Xiao, Org. Lett. 2015, 17, 1381-1384; n) I. Bernar, B. Fiser, D. Blanco-Ania, E. Gómez-Bengoa, F. P. J. T. Rutjes, Org. Lett. 2017, 19, 4211-4214; o) S. H. Jang, H. W. Kim, W. Jeong, D. Moon, Y. H. Rhee, Org. Lett. 2018, 20, 1248-1251; p) T. Mino, D. Yamaguchi, C. Masuda, J. Youda, T. Ebisawa, Y. Yoshida, M. Sakamoto, Org. Biomol. Chem. 2019, 17, 1455-1465; q) C.-Y. Chang, Y.-H. Lin, Y.-K. Wu Chem. Commun. 2019, 1116-1119.
- (18)Intermolecular ruthenium-catalyzed indole allylation: a) A. B. Zaitsev, S. Gruber, P. A. Pluss, P. S. Pregosin, L. F. Veiros, M. Worle, J. Am. Chem. Soc. 2008, 130, 11604-11605; b) B. Sundararaju, M. Achard, B. Demerseman, L. Toupet, G. V. M. Sharma, C. Bruneau, Angew. Chem. Int. Ed. 2010, 49, 2782-2785; Angew. Chem. 2010, 122, 2842-2845; c) H. Lauwick, Y. Sun, H. Akdas-Kilig, S. Dérien, M. Achard, Chem. Eur. J. 2018, 24, 7964-7969.
- (19)Intermolecular iridium-catalyzed indole allylation: a) W.-B. Liu, H. He, L.-X. Dai, S.-L. You, Org. Lett. 2008, 10, 1815-1818; b) W.-B. Liu, H. He, L.-X. Dai, S.-L. You, Synthesis 2009, 2076-2082; c) L. M. Stanley, J. F. Hartwig, Angew.

Chem. Int. Ed. **2009**, *48*, 7841-7844; *Angew. Chem.* **2009**, *121*, 7981-7984; d) S.-J. Chen, G.-P. Lu, C. Cai, *Synthesis* **2014**, *46*, 1717-1724.

- (20) Indirect methods for the synthesis of *N*-allyl indoles: a) W.-B. Liu, X. Zhang, L.-X. Dai, S.-L. You, *Angew. Chem. Int. Ed.* 2012, *51*, 5183-5187; *Angew. Chem.* 2012, *124*, 5273-5277; b) K.-Y. Ye, L.-X. Dai, S.-L. You, *Chem. Eur. J.* 2014, *20*, 3040-3044; c) K. Xu, T. Gilles, B. Breit, *Nat. Commun.* 2015, *6*, 7616.
- H. M. Weir, R. H. Bradbury, M. Lawson, A. A. Rabow, D. Buttar, R. J. Callis, J. O. Curwen, C. de Almeida, P. Ballard, M. Hulse, C. S. Donald, L. J. L. Feron, G. Karoutchi, P. MacFaul, T. Moss, R. A. Norman, S. E. Pearson, M. Tonge, G. Davies, G. E. Walker, Z. Wilson, R. Rowlinson, S. Powell, C. Sadler, G. Richmond, B. Ladd, E. Pazolli, A. M. Mazzola, C.D'Cruz, C. De Savi, *Cancer Res.* 2016, *76*, 3307-3318.
- (22) For enantiotopic π-facial interconversion in palladium- and iridium-catalyzed Tsuji-Trost cyclizations to form 5- and 6-membered *N*-hetereocycles, see: a) B. M. Trost, M. J. Krische, R. Radinov, G. Zanoni, *J. Am. Chem. Soc.* 1996, *118*, 6297-6298. b) C. Welter, O. Koch, G. Lipowsky, G. Helmchen, *Chem. Commun.* 2004, 896-897. c) M. A. Schafroth, S. M. Rummelt, D. Sarlah, E. M. Carreira, *Org. Lett.* 2017, *19*, 3235-3238.
- (23) For related deuterium labelling experiments, see: (a) P. Zhang, L. A. Brozek, J. P. Morken, *J. Am. Chem. Soc.* 2010, *132*, 10686-10688. (b) J.-P. Chen, Q. Peng, B.-L. Lei, X.-L. Hou, Y.-D. Wu, *J. Am. Chem. Soc.* 2011, *133*, 14180-14183.

Chapter 6 Supporting Information

- (1) Feng, J.; Garza, V. J.; Krische, M. J. J. Am. Chem. Soc. 2014, 136, 8911.
- (2) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.
- (3) Moragas, T.; Cornella, J.; Martin, R. J. Am. Chem. Soc. 2014, 136, 17702.
- (4) Chen, M. S.; Prabagaran, N.; Labenz, N. A.; White, M. C. J. Am. Chem. Soc. 2005, 127, 6970.
- (5) Weiss, M.; Holz, J.; Peters. R. Eur. J. Org. Chem. 2016, 210.
- (6) Tsutsumi, R.; Hong, S.; Krische, M. J. Chem. Eur. J. 2015, 21, 12903.
- Kim, S. W.; Schwartz, L. A.; Zbieg, J. R.; Stivala, C. E.; Krische, M. J. J. Am. Chem. Soc. 2019, 141, 671.
- (8) Piers, E.; Tillyer, R. D. Can. J. Chem. 1996, 74, 2048.

Chapter 7

- For selected reviews on metal catalyzed hydroamination, see: (a) Müller, T. E.; Beller, M. Chem. Rev. 1998, 98, 675. (b) Hong, S.; Marks, T. J. Acc. Chem. Res. 2004, 37, 673. (c) Severin, R.; Doye, S. Chem. Soc. Rev. 2007, 36, 1407. (d) Müller, T. E.; Hultzsch, K. C.; Yus, M.; Foubelo, F.; Tada, M. Chem. Rev. 2008, 108, 3795. (e) Hesp, K. D.; Stradiotto, M. ChemCatChem 2010, 2, 1192. (f) Schafer, L. L.; Yim, J. C. H.; Yonson, N. In Metal-Catalyzed Cross Coupling Reactions and More; De Meijere, A., Bräse, S., Oestreich, M., Eds.; Wiley-VCH: Weinheim, Germany, 2014; Vol. 3, p 1135. (g) Huang, L.; Arndt, M.; Gooβen, K.; Heydt, H.; Gooβen, L. J. Chem. Rev. 2015, 115, 2596. (h) Gupta, A. K.; Hull, K. L. Synlett 2015, 26, 1779.
- (2) For related photoredox-mediated olefin hydroaminations, see: (a) Musacchio, A. J.; Nguyen, L. Q.; Beard, G. H.; Knowles, R. R. J. Am. Chem. Soc. 2014, 136, 12217. (b) Miller, D. C.; Choi, G. J.; Orbe, H. S.; Knowles, R. R. J. Am. Chem. Soc. 2015, 137, 13492. (c) Musacchio, A. J.; Lainhart, B. C.; Zhang, X.; Naguib, S. G.; Sherwood, T. C.; Knowles, R. R. Science 2017, 355, 727.
- (3) For intermolecular rhodium catalyzed olefin hydroamination, see: (a) Coulson, D. R. *Tetrahedron Lett.* 1971, *5*, 429. (b) Brunet, J.-J.; Neibecker, D.; Philippot, K. *J. Chem. Soc. Chem. Commun.* 1992, 1215. (c) Beller, M.; Eichberger, M.; Trauthwein, H. *Angew. Chem. Int. Ed. Engl.* 1997, *36*, 2225. (d) Beller, M.; Trauthwein, H.; Eichberger, M.; Breindl, C.; Herwig, J.; Müller, T. E.; Thiel, O. R. *Chem. Eur. J.* 1999, *5*, 1306. (e) Beller, M.; Trauthwein, H.; Eichberger, M.; Breindl, C.; Müller, T. E. *Eur. J. Inorg. Chem.* 1999, 1121. (f) Utsunomiya, M.; Kuwano, R.; Kawatsura, M.; Hartwig, J. F. *J. Am. Chem. Soc.* 2003, *125*, 5608. 867

(g) Baudequin, C.; Brunet, J.-J.; Rodriguez-Zubiri, M. *Organometallics* 2007, 26, 5264. (h) Jiménez, M. V.; Pérez-Torrente, J. J.; Bartolomé, M. I.; Lahoz, F. J.; Oro, L. A. *Chem. Commun.* 2010, 46, 5322. (i) Béthegnies, A.; Kirkina, V. A.; Filippov, O. A.; Daran, J. C.; Belkova, N. V.; Shubina, E.; Poli, R. *Inorg. Chem.* 2011, 50, 12539. (j) Ickes, A. R.; Ensign, S. C.; Gupta, A. K.; Hull, K. L. *J. Am. Chem. Soc.* 2014, *136*, 11256. (k) Ensign, S. C.; Vanable, E. P.; Kortman, G. D.; Weir, L. J.; Hull, K. L. *J. Am. Chem. Soc.* 2015, *137*, 13748.

- (4) For intramolecular rhodium catalyzed olefin hydroamination, see: (a) Diamond, S. E.; Szalkiewicz, A.; Mares, F. J. Am. Chem. Soc. 1979, 101, 490. (b) Takemiya, A.; Hartwig, J. F. J. Am. Chem. Soc. 2006, 128, 6042. (c) Liu, Z.; Hartwig, J. F. J. Am. Chem. Soc. 2008, 130, 1570. (d) Bauer, E. B.; Andavan, G. T. S.; Hollis, T. K.; Rubio, R. J.; Cho, J.; Kuchenbeiser, G. R.; Helgert, T. R.; Letko, C. S.; Tham, F. S. Org. Lett. 2008, 10, 1175. (e) Shen, X.; Buchwald, S. L. Angew. Chem. Int. Ed. 2010, 49, 564. (f) Julian, L. D.; Hartwig, J. F. J. Am. Chem. Soc. 2010, 132, 13813. (g) Liu, Z.; Yamamichi, H.; Madrahimov, S. T.; Hartwig, J. F. J. Am. Chem. Soc. 2011, 133, 2772. (h) Hua, C.; Vuong, K. Q.; Bhadbhade, M.; Messerle, B. A. Organometallics 2012, 31, 1790.
- (5) For intermolecular iridium catalyzed olefin hydroamination, see: (a) Casalnuovo,
 A. L.; Calabrese, J. C.; Milstein, D. J. Am. Chem. Soc. 1988, 110, 6738. (b) Dorta,
 R.; Egli, P.; Zurcher, F.; Togni, A. J. Am. Chem. Soc. 1997, 119, 10857. (c) Zhou,
 J.; Hartwig, J. F. J. Am. Chem. Soc. 2008, 130, 12220. (d) Sevov, C. S.; Zhou, J.;
 Hartwig, J. F. J. Am. Chem. Soc. 2012, 134, 11960. (e) Pan, S.; Endo, K.; Shibata,
 T. Org. Lett. 2012, 14, 780. (f) Sevov, C. S.; Zhou, J.; Hartwig, J. F. J. Am. Chem.
 Soc. 2014, 136, 3200.

- (6) For intramolecular iridium catalyzed olefin hydroamination, see: (a) Hesp, K. D.;
 Stradiotto, M. Org. Lett. 2009, 11, 1449. (b) Kashiwame, Y.; Kuwata, S.; Ikariya, T. Chem. Eur. J. 2010, 16, 766. (c) Hesp, K. D.; Tobisch, S.; Stradiotto, M. J. Am. Chem. Soc. 2010, 132, 413. (d) Kashiwame, Y.; Kuwata, S.; Ikariya, T. Organometallics 2012, 31, 8444.
- (7) For Brønsted acid catalyzed olefin hydroamination, see: (a) Schlummer, B.; Hartwig, J. F. Org. Lett. 2002, 4, 2002. (b) Anderson, L. L.; Arnold, J.; Bergman, R. G. J. Am. Chem. Soc. 2005, 127, 14542. (c) Li, Z.; Zhang, J.; Brouwer, C.; Yang, C.-G.; Reich, N. W.; He, C. Org. Lett. 2006, 8, 4175. (d) Marcseková, K.; Doye, S. Synthesis 2007, 145. (e) Rosenfeld, D. C.; Shekhar, S.; Takemiya, A.; Utsunomiya, M.; Hartwig, J. F. Org. Lett. 2006, 8, 4179. (f) Ackermann, L.; Kaspar, L. T.; Althammer, A. Org. Biomol. Chem. 2007, 5, 1975. (g) McBee, J. L.; Bell, A. T.; Tilley, T. D. J. Am. Chem. Soc. 2008, 130, 16562. (h) Griffiths-Jones, C. M.; Knight, D. W. Tetrahedron 2010, 66, 4150. (i) Henderson, L.; Knight, D. W.; Williams, A. C. Synlett 2012, 1667.
- (8) For selected Brønsted acid catalyzed variants of transformations that are typically conducted using metal catalysts, see: (a) Fürstner, A.; Szillat, H.; Gabor, B.; Mynott, R. J. Am. Chem. Soc. 1998, 120, 8305. (b) Wabnitz, T. C.; Yu, J.-Q.; Spencer, J. B. Chem. Eur. J. 2004, 10, 484. (c) Rhee, J. U.; Krische, M. J. Org. Lett. 2005, 7, 2493. (d) Mori, K.; Sueoka, S.; Akiyama, T. Chem. Lett. 2009, 38, 628. (e) Li, M.; Yang, T.; Dixon, D. J. Chem. Commun. 2010, 46, 2191. (f) Taylor, J. G.; Adrio, L. A.; Hii, K. K. Dalton Trans. 2010, 39, 1171. (g) Dang, T. T.; Boeck, F.; Hintermann, L. J. Org. Chem. 2011, 76, 9353.
- (9) For selected reviews on iridium catalyzed allylic amination, see: (a) Takeuchi, R.
 Synlett 2002, 1954. (b) Takeuchi, R.; Kezuka, S. Synthesis 2006, 3349. (c) 869

Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. Chem.
Commun. 2007, 675. (d) Helmchen, G. In Iridium Complexes in Organic
Synthesis; Oro, L. A., Claver, C., Eds.; Wiley-VCH: Weinheim, 2009; pp 211250. (e) Hartwig, J. F.; Stanley, L. M. Acc. Chem. Res. 2010, 43, 1461. (f)
Hartwig, J. F.; Pouy, M. J. Top. Organomet. Chem. 2011, 38, 169. (g) Liu, W.-B.;
Xia, J.-B.; You, S.-L. Top. Organomet. Chem. 2011, 38, 155. (h) Zhou, C.-X.;
Zhang, W.; You, S.-L. Angew. Chem. Int. Ed. 2012, 51, 12662. (i) Qu, J.;
Helmchen, G. Acc. Chem. Res. 2017, 50, 2539.

- Meza, A. T.; Wurm, T.; Smith, L.; Kim, S. W.; Zbieg, J. R.; Stivala, C. E.;
 Krische, M. J. J. Am. Chem. Soc. 2018, 140, 1275.
- (11) Despite heterogeneous conditions, in palladium catalyzed *N*-arylation excess milled cesium carbonate was required. This was attributed to rate-limiting deprotonation that converts an amine-palladium(II) species to an amidopalladium(II) species: Meyers, C.; Maes, B. U. W.; Loones, K. T. J. Bal, G.; Lemiére, G. L. F.; Dommisse, R. A. J. Org. Chem. **2004**, 69, 6010.
- Weir, H. M.; Bradbury, R. H.; Lawson, M.; Rabow, A. A.; Buttar, D.; Callis, R. J.; Curwen, J. O.; de Almeida, C.; Ballard, P.; Hulse M.; Donald, C. S.; Feron, L. J.; Karoutchi, G.; MacFaul, P.; Moss, T.; Norman, R. A.; Pearson, S. E.; Tonge, M.; Davies, G.; Walker, G. E.; Wilson, Z.; Rowlinson, R.; Powell, S.; Sadler, C.; Richmond, G.; Ladd, B.; Pazolli, E.; Mazzola, A. M.; D'Cruz, C.; De Savi, C. *Cancer Res.* 2016, *76*, 3307.
- (13) Mailyan, A. K.; Eickhoff, J. A.; Minakova, A. S.: Gu, Z.; Lu, P.; Zakarian, A. Chem. Rev. 2016, 116, 4441.

Chapter 7 Supporting Information

- Crowley, P. J.; Mitchell, G.; Salmon, R.; Worthington, P. A. *Chimia* 2004, 58, 138.
- (2) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.
- (3) Iwasaki, M.; Kobayashi, Y.; Li, J.-P.; Matsuzaka, H.; Ishii, Y.; Hidai, M. J. Org.
 Chem. 1991, 56, 1922.
- (4) Krätzschmar, F.; Kaßel, M.; Delony, D.; Breder, A. Chem. Eur. J. 2015, 21, 7030.
- (5) Dubovyk, I.; Watson, I. D. G.; Yudin, A. K. J. Am. Chem. Soc. 2007, 129, 14172.
- (6) Shang, X.; Xiong, Y.; Zhang, Y.; Zhang, L.; Liu, Z. Synlett **2012**, *23*, 259.
- Yao, B.; Liu, Y.; Wang, M.-K.; Li, J.-H.; Tang, R.-Y.; Zhang, X.-G.; Deng, C.-L.
 Adv. Synth. Catal. 2012, 354, 1069.
- (8) Gigant, N.; Bäckvall, J.-E. Org. Lett. 2014, 16, 1664.
- (9) Li, Z.; Zhang, Y.; Liu, Z.-Q. Org. Lett. 2012, 14, 74.
- (10) Bauer, J. M.; Frey, W.; Peters, R. Angew. Chem. Int. Ed. 2014, 53, 7634.

Chapter 8

- (1)For selected reviews on carbonyl reductive coupling of π -unsaturated feedstocks via hydrogenation, transfer hydrogenation and hydrogen auto-transfer, see: (a) Ngai, M.-Y.; Kong, J.-R.; Krische, M. J. Hydrogen-Mediated C-C Bond Formation – A Broad New Concept in Catalytic C-C Coupling. J. Org. Chem. 2007, 72, 1063-1072. (b) Iida, H.; Krische, M. J. Catalytic Reductive Coupling of Alkenes and Alkynes to Carbonyl Compounds and Imines Mediated by Hydrogen. Top. Curr. Chem. 2007, 279, 77-104. (c) Hassan, A.; Krische, M. J. Unlocking Hydrogenation for C-C Bond Formation: A Brief Overview of Enantioselective Methods. Org. Proc. Res. Devel. 2011, 15, 1236-1242. (d) Ketcham, J. M.; Shin, I.; Montgomery, T. P.; Krische, M. J. Catalytic Enantioselective C-H Functionalization of Alcohols by Redox-Triggered Carbonyl Addition: Borrowing Hydrogen, Returning Carbon. Angew. Chem. Int. Ed. 2014, 53, 9142-9150. (e) Kim, S. W.; Zhang, W.; Krische, M. J. Catalytic Enantioselective Carbonyl Allylation and Propargylation via Alcohol-Mediated Hydrogen Transfer: Merging the Chemistry of Grignard and Sabatier. Acc. Chem. Res. 2017, 50, 2371-2380. (f) Doerksen, R. S.; Meyer, C. C.; Krische, M. J. Feedstock Reagents in Metal-Catalyzed Carbonyl Reductive Coupling: Minimizing Preactivation for Efficiency in Target-Oriented Synthesis. Angew. Chem. Int. Ed. 2019, 58, In Press - DOI: 10.1002/anie.201905532.
- (2) For general reviews on metal-catalyzed carbonyl reductive coupling, see: (a) *Metal Catalyzed Reductive C-C Bond Formation*; Krische, M., Eds.; Topics in Current Chemistry 279; Springer-Verlag Berlin Heidelberg: Germany, 2007. (b) Moragas, T.; Correa, A.; Martin, R. Metal-Catalyzed Reductive Coupling 872

Reactions of Organic Halides with Carbonyl-Type Compounds. *Chem. - Eur. J.*2014, 20, 8242-8258. (c) Holmes, M.; Schwartz, L. A.; Krische, M. J.
Intermolecular Metal-Catalyzed Reductive Coupling of Dienes, Allenes, and
Enynes with Carbonyl Compounds and Imines. *Chem. Rev.* 2018, 118, 6026-6052. (d) Nguyen, K. D.; Park, B. Y.; Luong, T.; Sato, H.; Garza, V. J.; Krische,
M. J. Metal-Catalyzed Reductive Coupling of Olefin-Derived Nucleophiles:
Reinventing Carbonyl Addition. *Science* 2016, 354, aah5133.

- (3) (a) Skucas, E.; Bower, J. F.; Krische, M. J. Carbonyl Allylation in the Absence of Preformed Allyl Metal Reagents: Reverse Prenylation via Iridium-Catalyzed Hydrogenative Coupling of Dimethylallene. J. Am. Chem. Soc. 2007, 129, 12678-12679. (b) Bower, J. F.; Skucas, E.; Patman, R. L.; Krische, M. J. Catalytic C-C Coupling via Transfer Hydrogenation: Reverse Prenylation, Crotylation and Allylation from the Alcohol or Aldehyde Oxidation Level. J. Am. Chem. Soc. 2007, 129, 15134-15135.
- (a) Han, S. B.; Kim, I. S.; Han, H.; Krische, M. J. Enantioselective Carbonyl (4)Reverse Prenylation from the Alcohol or Aldehyde Oxidation Level Employing 1,1-Dimethylallene as the Prenyl Donor. J. Am. Chem. Soc. 2009, 131, 6916-6917. (b) Holmes, M.; Nguyen, K. D.; Schwartz, L. A.; Luong, T.; Krische, M. J. Enantioselective Formation of CF₃-Bearing All-Carbon Quaternary Stereocenters Functionalization of Methanol: Iridium via C-H Catalyzed Allene Hydrohydroxymethylation. J. Am. Chem. Soc. 2017, 139, 8114-8117. (b) Schwartz, L. A.; Holmes, M.; Brito, G. A.; Gonçalves, T. P.; Richardson, J.; Ruble, J. C.; Huang, K.-W.; Krische, M. J. Cyclometallated Iridium-PhanePhos Complexes Are Active Catalysts in Enantioselective Allene-Fluoral Reductive

Coupling and Related Alcohol-Mediated Carbonyl Additions that Form Acyclic Quaternary Carbon Stereocenters. *J. Am. Chem. Soc.* **2019**, *141*, 2087-2096.

- (5)For metal-catalyzed allene-carbonyl reductive couplings to form racemic homoallylic alcohols beyond those cited in reference 3, see: (a) Ngai, M.-Y.; Skucas, E.; Krische, M. J. Ruthenium Catalyzed C-C Bond Formation via Transfer Hydrogenation: Branch-Selective Reductive Coupling of Allenes to Paraformaldehyde and Higher Aldehydes. Org. Lett. 2008, 10, 2705-2708. (b) Skucas, E.; Zbieg, J. R.; Krische, M. J. anti-Aminoallylation of Aldehydes via Ruthenium Catalyzed Transfer Hydrogenative Coupling of Sulfonamido-Allenes: 1,2-Aminoalcohols. J. Am. Chem. Soc. 2009, 131, 5054-5055. (c) Zbieg, J. R.; McInturff, E. L.; Krische, M. J. Allenamide Hydro-Hydroxyalkylation: 1,2-Aminoalcohols via Ruthenium Catalyzed Carbonyl anti-Aminoallylation. Org. Lett. 2010, 12, 2514-2516. (d) Zbieg, J. R.; McInturff, E. L.; Leung, J. C.; Krische, M. J. Amplification of *anti*-Diastereoselectivity via Curtin-Hammett Effects in Ruthenium Catalyzed Hydrohydroxyalkylation of 1,1-Disubstituted Allenes: Diastereoselective Formation of All-Carbon Quaternary Centers. J. Am. Chem. Soc. 2011, 133, 1141-1144. (e) Sam, B.; Montgomery, T. P.; Krische, M. J. Ruthenium Coupling Catalyzed Reductive of Paraformaldehyde to Trifluoromethyl Allenes: CF₃-Bearing All-Carbon Quaternary Centers. Org. Lett. 15, Sam, 2013, 3790-3793. (f) Oda. S.; B.: Krische, M. J. Hydroaminomethylation Beyond Carbonylation: Allene-Imine Reductive Coupling by Ruthenium-Catalyzed Transfer Hydrogenation. Angew. Chem. Int. Ed. 2015, 54, 8525-8528.
- (6) (a) Liu, R. Y.; Zhou, Y.; Yang, Y.; Buchwald, S. L. Enantioselective Allylation Using Allene, a Petroleum Cracking Byproduct. J. Am. Chem. Soc. 2019, 141, 874

2251-2256. Also see, (b) Tsai, E. Y; Liu, R. Y; Yang, Y.; Buchwald, S. L. A Regio- and Enantioselective CuH-Catalyzed Ketone Allylation with Terminal Allenes. *J. Am. Chem. Soc.* **2018**, *140*, 2007-2011.

- (7) For selected reviews on enantioselective carbonyl allylation, see: (a) Denmark, S. E.; Fu, J. Catalytic Enantioselective Addition of Allylic Organometallic Reagents to Aldehydes and Ketones. *Chem. Rev.* 2003, *103*, 2763-2794. (b) Marek, I.; Sklute, G. Creation of Quaternary Stereocenters in Carbonyl Allylation Reactions. *Chem. Commun.* 2007, 1683-1691. (c) Yus, M.; González-Gómez, J. C.; Foubelo, F. Catalytic Enantioselective Allylation of Carbonyl Compounds and Imines. *Chem. Rev.* 2011, *111*, 7774-7854. (d) Spielmann, K.; Niel, G.; de Figueiredoa, R. M.; Campagne, J. M. Catalytic Nucleophilic 'Umpoled' *π*-Allyl Reagents. *Chem. Soc. Rev.* 2018, *47*, 1159-1173.
- (8) For selected reviews on enantioselective catalysis via chiral-at-metal complexes, see: (a) Knight, P. D.; Scott, P. Predetermination of chirality at octahedral centres with tetradentate ligands: prospects for enantioselective catalysis. *Coord. Chem. Rev.* 2003, 242, 125-143. (b) Bauer, E. B. Chiral-at-Metal Complexes and Their Catalytic Applications in Organic Synthesis. *Chem. Soc. Rev.* 2012, 41, 3153-3167. (c) Gong, L.; Chen, L.-A.; Meggers, E. Asymmetric Catalysis Mediated by the Ligand Sphere of Octahedral Chiral-at-Metal Complexes. *Angew. Chem., Int. Ed.* 2014, *53*, 10868-10874.
- (9) For a recent authoratitive review on diastereo- and enantiodivergent catalytic processes, see: Beletskaya, I. P.; Nájera, C.; Yus, M. Stereodivergent Catalysis. *Chem. Rev.* 2018, 118, 5080-5200.
- (10) Studies on the enantioselective π -allyliridium *C*,*O*-benzoate-catalyzed carbonyl allylations mediated by allyl acetate are longstanding and encompass a vast 875

number of examples, including applications in natural product total synthesis. In all cases, the enantiofacial selectivity of aldehyde addition using allyl acetate as a pronucleophile corresponds to that shown in the present study. For key primary literature, see: (a) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. Enantioselective Iridium Catalyzed Carbonyl Allylation from the Alcohol or Aldehyde Oxidation Level Using Allyl Acetate as an Allyl Metal Surrogate. J. Am. Chem. Soc. 2008, 130, 6340-6341. (b) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. Enantioselective Iridium Catalyzed Carbonyl Allylation from the Alcohol or Aldehyde Oxidation Level via Transfer Hydrogenative Coupling of Allyl Acetate: Departure from Chirally Modified Allyl Metal Reagents in Carbonyl Addition. J. Am. Chem. Soc. 2008, 130, 14891-14899. (c) Hassan, A.; Lu, Y.; Krische, M. J. Elongation of 1,3-Polyols via Iterative Catalyst-Directed Carbonyl Allylation from the Alcohol Oxidation Level. Org. Lett. 2009, 11, 3112-3115. (d) Han, S. B.; Hassan, A.; Kim, I.-S.; Krische, M. J. Total Synthesis of (+)-Roxaticin: A Departure from Stoichiometric Chiral Reagents, Auxiliaries, and Premetalated Nucleophiles in Polyketide Construction. J. Am. Chem. Soc. 2010, 132, 15559-15561. (e) Schmitt, D. C.; Dechert-Schmitt, A.-M. R.; Krische, M. J. Iridium-Catalyzed Allylation of Chiral β -Stereogenic Alcohols: Bypassing Discrete Formation of Epimerizable Aldehydes. Org. Lett. 2012, 14, 6302-6305. (f) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Krische, M. J. Protecting-group-Free Diastereoselective C-C Coupling of 1,3-Glycols and Allyl Acetate through Site-Selective Primary Alcohol Dehydrogenation. Angew. Chem. Int. Ed. 2013, 52, 3195-3198.

(11) For related deuterium labelling studies on corresponding allyl acetate mediated carbonyl allylations, see reference 10b.

- Klei, S. R.; Golden, J. T.; Tilley, T. D.; Bergman, R. G. Iridium-Catalyzed H/D Exchange into Organic Compounds in Water. *J. Am. Chem. Soc.* 2002, *124*, 2092-2093 and references cited therein.
- (13) The influence of remote electron donating or withdrawing groups on enantioselectivity can be understood on this basis. More Lewis acidic catalysts accelerate rate-determining carbonyl addition. For the less Lewis acidic catalyst Ir-IV where X = OMe (see Table 1), the kinetic catalyst diastereomer C slowly isomerizes to the thermodynamically preferred catalyst diastereomer D, which delivers the opposite enantiomer, resulting in an erosion in enantioselectivity.
- (14) An inversion of enantioselectivity is also observed in corresponding reactions of dimethyl allene vs allyl acetate: (a) Han, S. B.; Kim, I. S.; Han, H.; Krische, M. J. Enantioselective Carbonyl Reverse Prenylation from the Alcohol or Aldehyde Oxidation Level Employing 1,1-Dimethylallene as the Prenyl Donor. *J. Am. Chem. Soc.* 2009, *131*, 6916-6917. (b) (Addition/Correction) Han, S. B.; Kim, I. S.; Han, H.; Krische, M. J. Enantioselective Carbonyl Reverse Prenylation from the Alcohol or Aldehyde Oxidation Level Employing 1,1-Dimethylallene as the Prenyl Donor. *J. Am. Chem. Soc.* 2010 *132*, 12517-12517.
- (15) DFT calculations were performed at the ωB97X-D/6-311+G(d,p)-SDD/CPCM(PhMe)//B3LYP-D3/6-31G(d)-SDD level of theory. See Supporting Information for computational details.
- (16) (a) Chen, S.; Huang, X.; Meggers, E.; Houk, K. N. Origins of Enantioselectivity in Asymmetric Radical Additions to Octahedral Chiral-at-Rhodium Enolates: A Computational Study. *J. Am. Chem. Soc.* 2017, *139*, 17902-17907. (b) Tutkowski, B.; Meggers, E.; Wiest, O. Understanding Rate Acceleration and Stereoinduction of an Asymmetric Giese Reaction Mediated by a Chiral Rhodium Catalyst. *J. Am.* 877

Chem. Soc. **2017**, *139*, 8062-8065. (c) Chen, S.; Zheng, Y.; Cui, T.; Meggers, E.; Houk, K. N., Arylketone π -Conjugation Controls Enantioselectivity in Asymmetric Alkynylations Catalyzed by Centrochiral Ruthenium Complexes. *J. Am. Chem. Soc.* **2018**, *140*, 5146-5152.

- (17) Grayson, M. N.; Krische, M. J.; Houk, K. N., Ruthenium-Catalyzed Asymmetric Hydrohydroxyalkylation of Butadiene: The Role of the Formyl Hydrogen Bond in Stereochemical Control. J. Am. Chem. Soc. 2015, 137, 8838-8850.
- (18) For a practical perspective on green chemistry, see: Rossen, K. Greening Organic Chemistry with Process Chemistry. J. Org. Chem. 2019, 84, 4580-4582.

Chapter 8 Supporting Information

(1) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923-2925.