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telluride quantum dots to the earthworm <i>Fisenia fetida</i>
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25 Abstract

The bioaccumulation potential and toxic effects of engineered nanomaterials (ENMs) 26 27 on earthworms are poorly understood and there are concerns that the hazards will change with the type of coating and soil ageing. Two studies were conducted following 28 29 OECD TG 222 with minor modifications and additional endpoints to assess the effects of CdTe QDs with different coatings on earthworms (*Eisenia fetida*); and to determine 30 31 whether bioaccumulation or toxicity changed after soil ageing. Earthworms were 32 exposed to carboxylate (COOH), ammonium (NH_4^+), or polyethylene glycol (PEG) 33 coated CdTe QDs, or a micron scale (bulk) CdTe material, at nominal concentrations of 50, 500 and 2000 mg CdTe QD kg⁻¹ dry weight (dw) for 28 days in Lufa 2.2 soil. The 34 endpoints included survival, body weight, reproduction, metal accumulation, and 35 36 markers of oxidative or ionoregulatory stress. In the fresh soil study, earthworms accumulated similar amounts of Cd and Te in the CdTe-bulk exposures, while the 37 accumulation of Cd was higher than Te during the exposures to CdTe QDs. However, 38 neither the total Cd nor Te concentrations in the earthworms were easily explained by 39 the extractable metal fractions in the soil or particle dissolution, indicating other particle 40 metrics altered bioavailability. There were on effects on survival, but some retardation 41 42 of growth was observed at the higher doses. Inhibition of Na⁺K⁺-ATPase activity with disturbances to tissue electrolytes, as well as tissue Cu and Mn were observed, but 43 44 without depletion of total glutathione in the fresh soil experiment. Additionally, juvenile production was the most sensitive endpoint, with estimated nominal EC₅₀ of >2000, 45 108, 65, 96 mg CdTe kg⁻¹ for bulk, PEG-, COOH- and NH₄⁺-coated CdTe QDs, 46 respectively. In the aged soil study, the accumulation of Cd and Te was higher than in 47 48 the fresh soil study in all CdTe QD exposures. Survival of the adult worms was reduced in the top CdTe-COOH and $-NH_4^+$ QD exposures by 55 ± 5 and 60 ± 25 %, 49 50 respectively; and with decreases in growth. The nominal EC₅₀ values for juvenile 51 production in the aged soil were 165, 88, 78 and 63 mg CdTe kg⁻¹ for bulk, PEG-, COOH- and NH4⁺-coated CdTe QDs, respectively. In conclusion, exposure to 52 nanoscale CdTe QDs, regardless of coating, caused more severe toxic effects that the 53 CdTe bulk material and the toxicity increased after soil ageing. There was a coating-54 mediated effect in some cases that was not explained by the coating itself, but due to 55 subsequent differences in the metal content and behaviour of the materials. 56

58 **1. Introduction**

59 Quantum dot (QD) technology is a growing market within the nanotechnology industry. The applications include bio-imaging and medical diagnostics (Kairdolf et al., 60 2013), light-emitting diode (LED) technologies (Hardman, 2005), and photovoltaics 61 62 that offer alternative energy solutions (Sinha et al., 2012). Inevitably, the manufacture of QDs and their use in industrial processing or products will likely cause some 63 64 releases to the environment, as with other engineered nanomaterials [ENMS, (Lead 65 et al., 2018)]. The disposal of QDs in electronics and the subsequent burning of e-66 waste by developing countries is a particular concern for metal releases from CdTe 67 QDs (Luo et al., 2011). Once in the environment, CdTe QDs may also slowly degrade 68 and release their hazardous dissolved metals (Navarro et al., 2008).

69 Cadmium is a non-essential metal and known to bioaccumulate in earthworms 70 (Hopkin, 1989). The background concentrations of Cd in soil are usually around 1 mg 71 kg⁻¹ dry weight (dw) or less (Keshavarz Jamshidian et al., 2017), but contaminated 72 soils can have tens of mg of Cd (Spurgeon et al., 1994). Dissolved cadmium toxicity 73 is well established in earthworms; with a 28-day lethal concentration (LC₅₀) of 588 mg Cd kg⁻¹ dw in artificial soil (Van Gestel et al., 1991), and 56-day EC₅₀ for impairment 74 of reproduction of 46.3 mg Cd kg⁻¹ dw in artificial soil (Spurgeon et al., 1994). In 75 76 contrast to Cd, the ecotoxicity of tellurium is poorly understood. It has no known 77 biological functions in eukaryotic cells (Ba et al., 2010) and not much is known about 78 its toxicity to soil organisms. So far, the only known biological function of tellurium has been found in fungi, that can use tellurite instead of sulphur for production of amino 79 acids when sulphur is limited or absent (Ramadan et al., 1989). Due to its 80 81 biogeochemistry, Te is a difficult metalloid to measure in complex environmental matrices; nonetheless a few measurements of Te in soil and sediments are emerging, 82 suggesting that it is naturally present at low µg kg⁻¹ concentrations (Belzile and Chen, 83 2015). 84

Studies on the environmental toxicology of QDs have shown biological effects that relate to the reactive chemical properties of CdTe QD; such as their potential to generate reactive oxygen species, but also toxicity related to Cd dissolution from the materials in various organisms (Rocha et al., 2017). Earthworms, like some other soil invertebrates, have the ability to sequester metals such as Cd, and to form metal storage granules in the tissue (Brown, 1982). Earthworms also have the ability to

biosynthesise CdTe QDs from dissolved metals (Stürzenbaum et al., 2013). There are
some studies on neurotoxicity (Wu et al., 2015) and oxidative stress (Srivastava et al.,
2016) in the nematode worm, *Caenorhabditis elegans*, with CdTe QDs; and at least
one study on marine polychaete worms with CdS (Buffet et al., 2014). However, there
is sparse information on the toxicity of additions of CdTe QDs to the soil for
earthworms.

97 This experiment aimed to provide a 'baseline dataset' for the hazard of CdTe QDs in freshly prepared soil. The approach used the Organisation for Economic 98 99 Cooperation and Development (OECD) technical guidance (TG) 222 method (OECD, 2004), but with modification and additional endpoints to give a detailed mechanistic 100 101 understanding of the data. Recently, we also reported a coating related effect on the 102 toxicity of CuO ENMs to earthworms (Tatsi et al., 2018). However, coating-mediated 103 effects on the toxicity of CdTe QDs are unknown in earthworms, and exactly the same coatings were used here for the CdTe QDs ENMs. The study design therefore included 104 a control soil, commercially available micron-sized (bulk) CdTe powder, and CdTe 105 QDs made with organic surface coatings with terminal residues of carboxylate 106 107 (COOH), ammonium (NH₄⁺), or polyethylene glycol (PEG); to represent negative, 108 positive and neutral surface charges respectively. The potentially toxic metal component(s) of the composite QDs were explored by determining the extractable 109 110 metal fractions from soil compared to the observed accumulation of Cd and Te in the earthworm tissues. Also to enable some comparison with our previous experiments 111 112 on CuO ENMs (Tatsi et al., 2018), the survival and growth of the earthworms in freshly dosed soils were determined. Biochemical measurements were also made in the fresh 113 114 soil experiment to assess known mechanisms of metal toxicity including; effects on 115 ionic regulation (tissue metal concentrations, Na⁺/K⁺-ATPase activity) and oxidative 116 stress (total glutathione). Having established the response in fresh soil, a second experiment including similar endpoints was carried out with newly exposed 117 earthworms after a six-month period of ageing the soils in order to understand the 118 persistence of any hazard from CdTe QD materials in the soil. 119

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121 **2. Methodology**

122 Two experiments were carried out using a low (50), medium (500) and high 123 (2000 mg as CdTe kg⁻¹ dw) nominal concentration of CdTe QDs in a quadruplicate 124 test design. Unexposed controls and an equivalent CdTe micron-sized powder was

used as a bulk material control. The first experiment was conducted with these substances freshly spiked into the soil, and the second experiment used the same soils after six months of ageing. Hereafter referred to as 'fresh' and 'aged' soil experiments, respectively.

129

130 2.1 Engineered quantum dots and characterisation

131 The ENMs used in the experiments were provided by PlasmaChem as part of a European Commission Framework 7 project (www.nanosolutionsfp7.com). The 132 133 details of the coatings and their synthesis are commercially sensitive information. 134 However, the different coatings were polyethylene glycol (PEG), carboxylate (COOH) and ammonium (NH4⁺) to represent neutral, negative and positive coatings 135 136 respectively. For clarity, we use the term '-NH4+' to mean an -NH3 terminal ligand that 137 has been ionised with H⁺ ions to achieve positive charge. An appropriate micron-sized CdTe powder (size < 250 µm, Sigma-Aldrich, UK, CAS No. 1306-25-8) was used as a 138 139 bulk material control (refer to hereafter as 'CdTe-bulk'). The materials were extensively characterised as part of the NANOSOLUTIONS project [e.g., (Vassallo et al., 2018)] 140 141 and the characterisation of the batches of the materials used here is summarised in 142 Table 1. This includes the nominal primary particle size and chemical composition of each nanoscale QD based on the manufacturer's information, as well as 143 144 measurements of primary particle diameter, hydrodynamic diameter and dissolution of Cd and Te made in Plymouth (Table 1; Fig. S1). The QD powders supplied were 145 dark red (NH4+-coated) or bright red (PEG-, COOH-coated) in colour and no impurities 146 147 were identified by the supplier. The CdTe bulk powder was black in colour and no 148 impurities were noted by Sigma-Aldrich. All other chemicals used were analytical 149 grade and purchased from Sigma-Aldrich unless stated otherwise. The 150 thermogravimetric analysis (TGA) of the materials (Table 1) tentatively indicated that 151 the combustible organic components of the materials contributed roughly 8, 23 and 50 % of the mass of the CdTe–NH4⁺, -COOH and –PEG coated QDs, respectively. 152 However, this would include any 'unknown' organic impurities and potentially residual 153 water molecules in between the polymers of the coatings. At the time of the study (and 154 155 currently) it was not technically feasible to reliably quantify the molecular weight of the coatings *in situ* on the particles, or the precise stoichiometry or coverage of each 156 157 coating in the overall material in order to spike the soils solely based on Cd or Te concentrations in the core. Therefore, pragmatically, the same approach was taken as 158

we had done previously with CuO materials of the same coatings (Tatsi et al., 2018),
where the mass concentration of the whole material as CdTe QDs was used for dosing
and reported as mg CdTe kg⁻¹ dw for each soil treatment, and with measurements of
the metal concentrations in the soils and organisms to confirm the exposure.

163

164 2.2 Stock animals and test soil

165 Adult Eisenia fetida originating from a commercial supplier (Blades Biological, Kent, UK) were used from an internal age synchronised laboratory breeding culture 166 167 held at University of Plymouth for both experiments. The test species were kept in an artificial medium that comprised of bark chippings (1/3), Irish Moss Peat (1/3) and 168 169 loamy sand topsoil (1/3) with surplus horse manure (from un-medicated horses) as 170 feed at a temperature of 20 ± 1 °C. Adult earthworms between two to four months old, 171 and with visible clitellum, were hand-selected for the experiments. The earthworms were acclimatised to the test soil (Lufa 2.2) with feed one week prior to the experiment. 172 A standard sandy loam Lufa 2.2 (LUFA Speyer, Germany) soil was used with the 173 following composition (supplier's information, mean \pm SD, dry soil, n = not specified): 174 175 pH of 5.5 \pm 0.2 (measured in 0.01 M CaCl₂ solution); organic carbon, 1.8 \pm 0.2 %; 176 nitrogen content at 0.17 \pm 0.02 %; cation exchange capacity, 10.1 \pm 0.2 meg 100 g⁻¹. The water-holding capacity of the soil was measured in-house and was (mean \pm SD, 177 178 n = 3): 41.3 ± 3.0 g 100 g⁻¹ dw. The soil used in the experiments was sieved through a 2 mm mesh and was air dried at 25°C for 2 days. Soil pH was measured (see 179 180 Supplementary Information, Table S1) prior to the start and at the end of the experiment in a 1:1 soil: deionised water slurry, using a glass combination electrode 181 182 (Corning 420).

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184 2.3 Experimental designs and spiking of the test soil

185 2.3.1 Fresh soil experiment

The experiment was conducted using the standard OECD test guideline (TG) 222 for the earthworm reproduction test (OECD, 2004), but with a reduced number of earthworms and additional sub-lethal endpoints. The study design included an unexposed control soil, bulk-CdTe and the three variations of coated CdTe QDs: PEG, COOH, NH_4^+ at nominal concentrations of 50, 500 and 2000 mg CdTe kg⁻¹ dw. The 50 and 500 mg CdTe kg⁻¹ dw concentrations were selected on the basis of the known sub-lethal effects and 28-day lethal concentration for Cd toxicity to earthworms

193 respectively (Spurgeon et al., 1994; Van Gestel et al., 1991); and would also encompass the range of concentrations of Cd at contaminates sites such as metal 194 195 smelters [e.g., 0.2-102 mg Cd kg⁻¹ of soil, (Chlopecka et al., 1996)]. The TG 222 protocol allows a high dose of 1000 mg kg⁻¹ dw of the test substance in the soil as part 196 197 of a 'limit test.' Given that the CdTe QDs are a composite of Cd, Te, plus any coating; an upper concentration of 2000 mg kg⁻¹ dw was used for the QDs so that the individual 198 199 metals in the composite might approximate to the limit test values in TG 222. It was also intended as a high concentration to reveal any mechanistic aspects of the toxic 200 201 effects that might be observed. The QDs were mixed into the soil as powders, because 202 dry mixing has been found a suitable dosing method (Handy et al., 2012). The amount 203 of QD powder required to dose each of 4 replicates was weighed into 50 g dw of soil 204 and carefully mixed by hand for 10 minutes. Then, this aliquot of soil was added to the 205 remaining amount of soil (950 g) and further mixed by hand, followed by wetting of the soil to 50 – 60 % of the WHC with ultrapure Milli-Q water (18.2 Ω). The soil was left to 206 207 equilibrate with the moisture for one day to minimise the risk of the QDs changing 208 before the worms were added.

209 Adult *E. fetida* (n = 5 earthworms/250 g dw soil) with a mean weight of 1.86 ± 210 0.02 g for 5 earthworms (mean \pm SEM, n = 52, individual earthworm weight ~ 0.36 g) 211 were exposed in 4 replicates boxes of soil at 20 ± 1°C at 12:12 light:dark cycle. The 212 earthworms were fed dried horse manure (1 g/earthworm) wetted to 70 % of its water 213 holding capacity (WHC) with Milli-Q water. Life history traits such as survival and 214 weight of the earthworms were recorded at the beginning of the experiment and on days 14 and 28. Behavioural changes such as avoidance of burrowing into the soil, 215 216 were recorded following visual assessment in all treatments at the beginning of the 217 fresh soil experiment. Further endpoints such as total concentration of Cd, Te and 218 other elements in earthworm tissue and total glutathione and Na⁺/K⁺-ATPase activity, 219 were measured on day 28 in randomly selected earthworms (n = 8 earthworms for each analysis per treatment; methods described below). After 28 days, when all the 220 adults were removed from soils, additional feed was added (5 g dw, wetted to 70 % 221 222 WHC with Milli-Q water) and soils were left for a further 28 days to allow juveniles to hatch from cocoons. The juveniles were then counted following the OECD TG 222 223 (OECD, 2004). Briefly, each test vessel was placed in a water bath (60 °C) and the 224 juveniles that emerged to the surface were collected and counted. After counting, soils 225 226 were manually checked to ensure no juveniles or cocoons remained in the soil. The

juveniles collected from each treatment were washed in deionised water, blotted dry,and weighed to determine the treatment effect on juvenile growth.

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230 2.3.2 Aged soil experiment

231 The soil used in the aged soil experiment was the same as that used in the initial fresh soil experiment. At the end of the fresh soil experiment the test containers 232 233 were left for a total of 27 weeks (6 months) in the same experimental conditions, but 234 without any disturbance or watering. During the ageing period the containers were not 235 open, but airflow occurred via the pierced lids. Incidental plant growth had occurred 236 during the ageing period, from seeds likely present in the natural soils or the manure 237 used as feed. It was not an objective of the study to investigate plant growth in the 238 aged soils, but the opportunity was taken. The plant coverage in each exposure 239 container was quantified one day before starting the aged soil experiment. All the 240 containers were photographed, and their percent cover with plants was determined by 241 directly assessing the boxes (an acetate paper with grids was placed on each box to aid the estimation of percent cover). Once the above ground material was quantified 242 243 (see supplementary material, Figs. S2 and S3), it was removed by cutting. The roots 244 of the plants were left in the soil to increase the environmental realism of the aged soil study. Soil moisture content was then adjusted to 50 - 60 % and soil pH was measured 245 246 in a 1:1 soil:water slurry using a glass combination electrode (Corning instruments).

247 The same study design and methods were used in the aged soil experiment as 248 described in the fresh soil experiment. Five earthworms with a mean weight of 1.89 ± 0.03 g (mean \pm SEM, n = 52, individual weight ~0.38 g) were used in the experiment. 249 250 On days 0, 14 and 28, endpoints such as survival, biomass and appearance were 251 recorded. Reproduction was assessed after 56 days of exposure (described above). 252 In addition, the total concentration of Cd, Te and other electrolytes and trace metals in 253 earthworm tissues were measured on day 28. Due to a technical fault with a freezer, earthworm samples collected for biochemical analyses (total glutathione and Na⁺/K⁺-254 255 ATPase) were defrosted twice. These defrosted samples were analysed, but the 256 results were considered unreliable and therefore not used (data not shown). In addition, juvenile weight was not assessed in the aged soil study due to the difficulty 257 258 of obtaining the alive and intact juveniles from the soil. This was due to the presence 259 of plant roots in the soil that severely hampered the collection of juveniles. The roots were inevitably very tangled in the soil, making it difficult to manually find and remove 260

the juvenile worms that were extremely delicate. Several individuals stuck in the soil
for too long and, unfortunately, they did not survive the handling stress of the collection
procedure.

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5 2.4 Total and extractable metal analysis in samples

Total metal analysis in the soils and the earthworms was conducted to confirm 266 267 the exposure and the bioavailable fraction from the QDs, respectively, exactly as previously described (Tatsi et al., 2018). Total aqua regia extractable Cd and Te 268 269 concentrations were measured in triplicate at the beginning of both the fresh and aged 270 soil experiments (n = 12 per treatment). Additionally, to provide insight to the mobile fractions of Cd and Te, water (deionised water) and 0.1 M HCl extractable fractions 271 272 were determined in duplicate in the nominal 2000 mg CdTe kg⁻¹ dw exposures only in the fresh soil experiment (n = 8 per treatment), aged soils were not analysed. 273 Earthworms collected at the end of the fresh and aged soil experiments (n = 8 per 274 treatment) were allowed to depurate for 24 h on moist filter paper (the latter was 275 changed following 12 h to avoid coprophagy) according to (Arnold and Hodson, 2007). 276 277 Then the worms were washed, freeze-dried (for 48 h), weighed and digested in 1 ml 278 concentrated HNO₃. All acid digested samples were allowed to cool, diluted with 7 ml 279 ultrapure deionised water (18.2 Ω) and stored in the dark. The total concentration of 280 Cd and Te was measured in the soil and earthworm samples. In earthworms, also other essential elements (Ca, Fe, K, Mg, Mn, Na, Zn, Cu) were measured to assess 281 282 changes in the trace element and electrolyte composition of the earthworms by inductively coupled plasma optical emission spectrophotometry (ICP-OES, iCAP 700, 283 284 Thermo Fisher) or inductively coupled plasma mass spectrometry (ICP-MS, Thermo 285 Scientific X Series 2) as appropriate. Prior to analysis, the samples were sonicated for 286 15 minutes (at 0.05 kva, 30 kHz, Ultrawave Ltd), vortexed for 10 s, then hand shaken immediately prior to analysis to ensure good mixing. All samples were analysed 287 against matrix-matched standards where possible. The certified reference materials 288 for total Cd metal reported close to the expected values and were $100 \pm 4 \%$ (*n* = 3, 289 EnviroMAT contaminated soil, SS-1) and 102 \pm 7 % (*n* = 3, TORT-2, contaminated 290 lobster hepatopancreas). Spike recovery tests of earthworm and soil digests using 291 CdTe-NH4⁺ QDs were 78 and 73 %, respectively (values were calculated based on the 292 293 sum of [Cd] and [Te] and normalised to expected coating mass which was $8.8 \pm 0.5 \%$ 294 for NH₄⁺). Using CdCl₂ and Te standard for ICP-MS and -OES (TraceCERT by Sigma),

the recoveries for earthworm and soil digests were > 90 %. The calculated limit of detection (LOD) for Cd (analysed by ICP-OES) was 1 mg kg⁻¹ dw and 0.001 mg kg⁻¹ for Cd in soil and earthworm tissue, respectively. While the calculated LOD for Te (analysed by ICP-MS) was to around 0.01 mg kg⁻¹ dw and 0.1 mg kg⁻¹ dw in soil and earthworm tissue, respectively.

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301 2.5 Biochemical analyses

302 The biochemical methods had been used previously for ENMs and earthworms 303 in our laboratory (Tatsi et al., 2018). Biochemistry was performed on whole earthworm 304 tissues collected at the end of fresh soil experiments (*n* = 8 per treatment). Earthworms 305 from each test container were snap frozen in liquid nitrogen and stored at -80 °C until 306 homogenised in ice cold isotonic buffer (1:5 ratio, weight: volume) exactly as described in Tatsi et al. (2018). The homogenates were further diluted prior to analysis due to 307 high protein concentration (15 fold dilution of the original tissue, n = 8 per treatment) 308 and were then were assayed in triplicate for total protein, total glutathione (GSH) and 309 Na⁺/K⁺-ATPase activity exactly as described in Tatsi et al. (2018) and using a 310 311 VersaMax plate reader (Molecular Devices, UK). Briefly, GSH was quantified in 20 µl 312 of the diluted homogenate according to (Owens and Belcher, 1965). Na⁺/K⁺-ATPase activity was determined in 10 µl of the diluted homogenate based on a modification of 313 314 (McCormick, 1993). Total protein was determined in 25 µl of the diluted homogenate using the Pierce BCA kit (#RE232674, Thermo Scientific, UK). The concentrations of 315 316 GSH and Na⁺/K⁺-ATPase activity were normalised to total protein in the sample and data are expressed as nmol GSH per mg protein and µmol ADP per mg protein per 317 318 hour, respectively.

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320 2.6 Statistical analyses and data presentation

Statistical analyses were performed using R studio software (version 2.1) and 321 graphs were drawn using SigmaPlot (version 13.0 or 14.0). Data were checked for 322 323 normality (Shapiro-Wilk) and homogeneity of variance (Bartlett's test). Non-parametric data were transformed (log₁₀) and reanalysed as appropriate. The student's *t*-test (two-324 tailed, un-/paired) or Mann Whitney test were used for comparing two samples sets as 325 appropriate. Due to a few individual significant differences in pH between treatments 326 in the fresh or aged soil experiments, analysis of covariance (ANCOVA) was carried 327 out for the interactive effects of pH on all endpoints. When no interactions were found, 328

329 pH was omitted from the model to allow the performance of a post-hoc test. Treatment effects were assessed by one-way ANOVA followed by the Tukey-Kramer (due to 330 331 unequal sample size) post hoc test. Changes in biomass were analysed with repeated measures ANOVA followed by Tukey's honest significance difference (HSD) test to 332 333 identify the differences. The non-parametric Kruskal-Wallis test was used followed by a distribution free multiple comparisons test, Dunn's test, where data transformation 334 335 was unsuccessful. To evaluate the biological associations known for metal toxicity, the Spearman's rank, rs, correlations were carried out on all of the data, regardless of 336 337 treatment, for a specified endpoint within each experiment. The reproduction results were normalised to the control values and the 50 % effect concentration (EC₅₀) values 338 339 and its 95 % confidence intervals were estimated by logistic nonlinear regression analysis (sigmoid) on raw data using the log-transformed nominal concentrations in 340 SigmaPlot v 14.0. To estimate the nominal effect value, the log-transformed value was 341 reversed. The statistical significance level (α) for all tests was set at 0.05. 342

343

344 **3. Results**

345 *3.1* Soil pH

346 In the beginning of the fresh soil experiment the soil pH ranged from 5.36 to 5.89, and soil pH differed between QD treatments and the control soils. However, by 347 348 the end of the exposure (day 28) there were no differences in pH between the QDtreated soils and the control soils, with the pH values ranging from 5.39 to 5.78 (Table 349 350 S1). In the aged soil experiment, the pH ranged from 5.38 to 5.65, both at the beginning and the end of the experiment, and was not statistically different from the 351 352 corresponding pH measurements in the fresh soil experiment (P > 0.05, Student's t-353 test). Soil pH was included as a cofactor in all the endpoints analysed in both 354 experiments, but no significant interactions were found (P > 0.05, ANCOVA). Soil pH 355 was therefore not a factor in any of the biological measurments.

356

357 3.2 Total and extractable Cd and Te concentrations in soil

To enable the interpretation of the results, the total *aqua regia* extractable concentration of Cd and Te was measured at the beginning of both the fresh and aged soil experiments. In the former experiment, the control soil contained a background Cd concentration of 0.61 \pm 0.05 mg Cd kg⁻¹ dw (mean \pm SEM, *n* = 8), of this total Cd, 9.2 \pm 1.4 and 10 \pm 1.5 % was water and 0.1 M HCl-extractable respectively; indicating

363 that less than 10 % of the Cd in the soil was mobile. Te was not detected in the control soil, as the results were below the limit of detection, < 0.01 mg Te kg⁻¹ dw. Since the 364 CdTe QDs were dosed into the soil on the basis of the mass of the whole material, 365 including both metals present and any coating, it was expected that the measured total 366 367 Cd and Te concentrations (i.e., measured concentration of each metal) would be a 368 fraction of the total exposure concentration of the whole material. This was indeed the 369 case (Fig. 1). Nonetheless, exposure to CdTe bulk and the differently coated CdTe QDs caused the expected increase in total soil Cd concentrations in the fresh soil 370 371 experiment (Fig. 1). However, the measured total Cd in the soil did not differ significantly between the types of CdTe QD exposures within each nominal test 372 concentration (Fig. 1, P > 0.05, ANOVA). The measured total Te concentrations also 373 increased with the nominal exposure concentration of the whole material in the fresh 374 375 soil experiment. However, in contrast to Cd, the total tellurium concentration varied significantly between materials in each nominal test concentration (Fig. 1A and B, P < 376 0.05, ANOVA); with the amount of Te in the exposures followed the order of: COOH \leq 377 PEG < NH_4^+ < CdTe-bulk. This was not interpreted as a 'coating effect' per se, 378 379 because it did not follow the order of mass loss of the materials in the TGA 380 measurement of PEG > COOH > NH_4^+ (Table 1) and in any case, the effect was absent 381 in the Cd measurements. Instead, this was likely attributed to the ratio of Cd to Te in 382 the different materials as manufactured.

The water- and dilute acid-extractable Cd or Te was measured at the beginning 383 384 of the fresh soil experiment at the highest nominal exposure concentration and plotted against the total metal concentrations in the earthworms (Fig. 2). The amount of water-385 386 and 0.1M HCI-extractable Cd and Te varied between treatments. In the CdTe-bulk 387 exposures, less than 1 % of the Cd and Te were extractable and the Cd or Te metal 388 concentrations in the earthworms were also modest (Fig. 2); suggesting low availability of metal from the bulk material in the soil. For the CdTe QD treatments, regardless of 389 the extraction method or type of coating, only a small proportion of the Te was 390 391 extractable and with similarly low total Te concentrations in the earthworms. However, the situation was different for the extractable Cd from the QDs (Fig. 2), with a 392 considerable fraction of acid-extractable Cd in the soil. Overall, with the QD 393 treatments, the water available fraction of Cd and Te was the highest from CdTe-NH4⁺ 394 395 QD exposures, although this did not necessarily lead to a concomitant elevation of total Cd or Te in the earthworms according to the type of coating (Fig. 2). The total 396

397 concentration of Cd and Te in the soil were measured again prior to beginning of the 398 aged soil experiment. In general, the concentration of Cd and Te in the aged soil 399 followed a very similar pattern to the fresh soil experiment, but with a trend (not 400 statistically significant) of slightly lower values than in the fresh soil experiment. The 401 extractable fractions were not assessed in the aged soils.

402

403 3.3 Total Cd and Te in earthworm tissue

The total Cd or Te concentrations in the earthworms are shown in Fig. 3. The 404 405 control earthworms in the both the fresh and aged experiments had some background Cd in the tissues (4.23 \pm 0.34 mg Cd kg⁻¹ dw, mean \pm SEM, *n* = 8), while no Te was 406 407 detected. In the fresh soil experiment, an exposure concentration-dependent increase 408 in the total body burden of Cd was seen in all the CdTe-bulk and all the CdTe QD 409 exposures (Fig. 3). The accumulation of Cd was significantly lower in the CdTe-bulk when compared to the QD exposures in fresh soil at each exposure concentration. 410 411 However, within the CdTe QDs, there was no significant differences between the total concentrations of Cd in the earthworm tissues by the type of coating of the CdTe QDs 412 at each nominal test concentration (Fig. 3A, P > 0.05, ANOVA). 413 The total 414 concentration of Te in the earthworms during the fresh soil experiment was also measured (Fig. 3B). Here, the tissue Te concentration increased in a concentration-415 416 dependent manner in the CdTe-bulk exposures, but not in the CdTe QD exposures where the Te remained uniformly low. Within the QDs there was also no clear effect 417 418 on Te accumulation by the type of surface coating on the ENM in the fresh soil experiment (Fig. 3B). Together these observations suggest some Te accumulation for 419 420 the bulk material, but very limited Te accumulation for CdTe QDs in fresh soil. The 421 accumulation of Cd and Te were strongly correlated within each treatment group (rs= 422 0.9, 0.8, 0.8 and 0.9 for CdTe-bulk, -PEG, -COOH and -NH₄⁺ exposures, P < 0.05, 423 Spearman) and this is consistent with the notion of being exposed to a composite material containing both metals. The total Cd or Te in the soil was also significantly 424 425 correlated to the total Cd or Te in earthworms when analysing all the data together (rs = 0.8 and 0.7 for Cd and Te, P < 0.05, Spearman); and within the animals in each 426 treatment. However, this contrasts with the absence of any clear relationship between 427 428 the water- or acid-extractable fractions and accumulation of Cd or Te by the 429 earthworms noted in Fig. 2.

430 In the aged soil experiment, the accumulation of Cd and Te by earthworms was, overall, higher than in the fresh soil experiment. In the CdTe-bulk exposures the Cd 431 432 concentrations in the earthworms were significantly higher than those in the fresh soil study (Fig. 3). In the nominal 50 mg CdTe kg⁻¹ dw test concentration there were no 433 434 significant differences in the concentration of Cd in the earthworms by the type of coating on the CdTe QDs in the aged soil (Fig 3C). In the nominal 500 or 2000 mg 435 436 CdTe kg⁻¹ dw test concentrations in aged soil, there were no statistically significant 437 differences between the tissue concentrations of Cd between doses, but there was a 438 trend of higher tissue Cd concentration in earthworms exposed to the differently coated CdTe QDs when compared to the CdTe-bulk, although the trend was not 439 440 statistically significant. In the aged soil experiment, the concentration of Te in the 441 earthworms followed a similar pattern of accumulation to the fresh soil experiment, where the highest amount of Te was found in earthworms exposed to CdTe-bulk 442 material in all test concentrations. In the nominal 50 mg CdTe kg⁻¹ dw test 443 concentration, there were no statistically significant differences in Te accumulation by 444 the type of coating on the CdTe QDs. However, some differences in the total Te 445 accumulation emerged in the 500 mg CdTe kg⁻¹ dw test concentration, where the 446 447 concentration was highest in the CdTe-NH4⁺ QD exposure, compared to the other materials, and the Te accumulation was similar in the CdTe-PEG and -COOH QD 448 exposures. In the nominal 2000 mg CdTe kg⁻¹ dw test concentration, the accumulation 449 of Te was not statistically significantly different between the different CdTe QD 450 451 exposures, although there was a non-significant trend of higher Te in the CdTe-NH4⁺ 452 groups.

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454 *3.4 Survival, biomass and appearance of earthworms*

455 The survival and biomass (growth) of the earthworms are reported in Table 2 456 for both the fresh and aged soil experiments. First, consider the fresh soil experiment. The control animals were healthy and increased in biomass (19.9 \pm 5.1 %, Table 2). 457 In the fresh soil experiment, there were no statistically significant changes in the 458 459 survival in any of the treatments throughout the experiment, except some minor mortalities < 10 %. The biomass of earthworms in the CdTe-bulk exposures also 460 increased by as much as 22 % by day 28 (Table 2); overall indicating no effects on 461 462 growth or survival of earthworms exposed to the bulk material compared to the unexposed controls in the fresh soil experiment. An increase in biomass was also 463

464 observed in the nominal 50 mg CdTe kg⁻¹ dw for the QD exposures by day 28, but the 2000 mg CdTe kg⁻¹ dw exposure concentration of the QDs caused a uniform decrease 465 466 in biomass of around 50% by day 28, regardless of the type of coating on the ENMs 467 (Table 2). There were no obvious behavioural or changes in appearance of the 468 earthworms exposed to the CdTe-bulk or the nominal 50 and 500 mg CdTe kg⁻¹ dw of the CdTe QDs in the fresh soil experiment. However, earthworms from all the CdTe 469 470 QD exposures (excluding CdTe-bulk) at the nominal 2000 mg CdTe kg⁻¹ dw test concentration were sluggish and were not actively feeding (evidence of left-over 471 472 manure), although they were still moving in the soil. Furthermore, morphological changes were evident in earthworms from the 2000 mg CdTe kg⁻¹ dw CdTe QD ENM 473 exposures. The earthworms were visibly smaller and showed evidence of shedding of 474 475 the posterior segments (the last segments were clearly lighter coloured with signs of regeneration). A strong negative correlation was found between tissue Cd or Te and 476 477 biomass for the fresh soil experiment when analysing all the data from all treatments together ($r_s = 0.6$ and 0.5 for Cd and Te respectively, P < 0.05, Spearman). 478

479 In the aged soil study, the control earthworms were healthy in appearance and with good survival, however, a trend of biomass loss (8.1 ± 5.6 %, Table 2) was noted, 480 481 although the change was not statistically significant. The earthworms survived in all the CdTe-bulk exposures with only minor mortalities < 10 % at the nominal 2000 mg 482 CdTe kg⁻¹ dw exposures in the aged soil experiment. In the CdTe-bulk exposures, 483 earthworms lost biomass only at the nominal 2000 mg CdTe kg⁻¹ dw test concentration 484 485 (Table 2). At the nominal 2000 mg CdTe kg⁻¹ dw test concentration for the CdTe QDs, the CdTe-COOH QDs were the most toxic to earthworms in the aged soil experiment; 486 487 where survival was reduced to 55 \pm 5, 60 \pm 25, and 75 \pm 10% for the –COOH, -NH₄⁺ 488 and –PEG-coated QDs respectively by day 28 (Table 2). In addition, earthworms from 489 these treatments showed statistically significant weight loss compared to controls. 490 There were no obvious behavioural or appearance changes in the CdTe-bulk exposures or the nominal 50 mg CdTe kg⁻¹ dw CdTe QD exposures in the aged soil 491 experiment. Some earthworms appeared lethargic and non-responsive in the 500 and 492 493 2000 mg CdTe kg⁻¹ dw CdTe QD ENM exposures, and similar to the fresh soil study, there was evidence of shedding of the posterior segments and lack of feeding 494 495 (presence of surplus feed).

497 *3.5 Reproduction*

The reproductive success of the animals is reported in Table 3, and with dose-498 499 response curves for juvenile production shown in Fig. S4. The test results on 500 reproduction in the controls of both fresh and aged experiments met the validity criteria 501 of the OECD TG 222. In both the experiment with fresh and aged soil, the control 502 earthworms produced a healthy number of juveniles per earthworm (Table 3), and with 503 a coefficient of variation (CV %) of less than < 30 % in keeping with the OECD test. 504 Reproduction was not significantly reduced in any of the CdTe-bulk test 505 concentrations, or in the nominal 50 mg CdTe kg⁻¹ dw CdTe QD exposures in the fresh 506 soil experiment. However, it was significantly reduced in the CdTe QD treatments at the nominal 500 mg CdTe kg⁻¹ dw test concentration, and completely inhibited at the 507 508 2000 mg CdTe kg⁻¹ dw test concentration for the CdTe QD exposures, compared to controls in the fresh soil experiment. The 50% effect concentrations (EC₅₀) for 509 reproductive success (i.e., juvenile production) for the different materials are 510 511 presented in Table 3. The estimated EC₅₀ values were calculated using the nominal CdTe QD concentrations (response curves presented in Fig. S4). For the fresh soil 512 513 experiment, the EC₅₀ values decreased in the following order: CdTe-COOH < CdTe-514 NH₄⁺ < CdTe-PEG. For the CdTe bulk material in fresh soil there was no inhibition of juvenile production and so an EC50 value could not be estimated. The fresh wet weight 515 516 of the juveniles was also measured after they were rinsed in deionised water and padded dry (Table 3). There were some statistically significant differences between 517 518 treatments in the biomass of the juveniles in the fresh soil experiment, but the differences followed the same order as the total number of juveniles produced. That 519 520 is, where only a few juveniles were produced, those juveniles also tended to be small 521 (i.e., poor quality offspring).

522 In the aged soil experiment (Table 3), the controls produced more juveniles 523 than in the fresh soil experiment. Unlike the situation in fresh soil, the number of juveniles produced in the CdTe-bulk exposures declined, and with a complete 524 inhibition of reproduction at the nominal 2000 mg CdTe kg⁻¹ dw test concentration in 525 526 aged soil. The effects of the CdTe QD exposures on juvenile production were more pronounced in the aged soil experiment, with lower EC₅₀ values for the CdTe-PEG 527 and CdTe-NH₄⁺ treatments in the aged soil compared to those in fresh soil (Table 3). 528 529 The number of juveniles produced was significantly reduced or completely abolished in all the CdTe QD exposures at 500 and 2000 mg CdTe kg⁻¹ dw test concentrations 530

531 compared to the unexposed controls in aged soil. The EC₅₀ values for reproductive success decreased in the following order in aged soil: CdTe-NH4⁺ < CdTe-COOH < 532 533 CdTe-PEG < CdTe-bulk. However, while the ranking of the EC_{50} values are correct, all of the curve fits (Fig. S4), will have some uncertainty in the precise EC₅₀ value due 534 535 to the number of data points and their confidence intervals in the dose-response plot. 536

537 3.6 Sodium pump activity and tissue elemental composition

In the experiment with the fresh soil the control earthworms showed a normal 538 Na⁺/K⁺-ATPase (sodium pump) activity 6 – 8 μ mol ADP mg⁻¹ protein h⁻¹ (Fig. 4A). In 539 540 the fresh soil experiment, there was an overall trend of lower sodium pump activity in all the exposure concentrations of the CdTe bulk material, with significantly reduced 541 542 activities in the nominal 500 and 2000 mg CdTe kg⁻¹ dw test concentration (Fig. 4). There was also some inhibition of the Na⁺/K⁺-ATPase in the CdTe QD treatments in 543 the fresh soil experiment, but with no clear dose and coating-dependent trend in the 544 data (Fig. 4A). The substrates for the sodium pump include Na⁺ and K⁺ and the tissue 545 electrolytes of the earthworms are shown in Table S2. There was no clear effect on 546 the concentration of Na⁺ in earthworm tissues, however, the concentration of K⁺ was 547 significantly lower in the nominal 500 and 2000 mg CdTe kg⁻¹ dw CdTe-NH4⁺ 548 exposures compared to controls (P < 0.05, ANOVA; Table S2). In the aged soil study. 549 550 unfortunately failure of a deep freeze prevented any reliable determination of enzyme activity in those samples (data not shown). However, there was no depletion or 551 552 significant increase in the tissue concentration of Na⁺ or K⁺ between the test concentrations and the control for any treatment (P > 0.05, ANOVA; Table S2). 553

554 A range of other electrolytes and essential trace elements in the tissues of the 555 earthworms are shown in Table S2. Overall for the control earthworms, in both the 556 fresh and aged soil studies, the electrolytes and trace elements remained in the 557 expected range for a healthy earthworm population (Table S2). The earthworms in the fresh soil study exposed to the CdTe-bulk showed no significant treatment-dependent 558 changes in electrolytes or trace metals (Table S2). In contrast, the CdTe QD 559 560 exposures at 500 and 2000 mg CdTe kg⁻¹ dw test concentrations showed statistically significant decreases in essential and trace metals in the earthworms in fresh soil 561 (Table S2). The concentration of Ca, Mg and Fe were significantly reduced in the 562 nominal 500 and 2000 mg CdTe kg⁻¹ dw CdTe-NH4⁺ exposures compared to 563 unexposed controls in the fresh soil experiment. There was also a significant negative 564

565 correlation between the tissue concentration of Cd and the influenced metals (rs = -0.3, -0.4 and -0.3 for Ca, Mg and Fe respectively, P < 0.05, Spearman). Earthworm 566 567 tissue copper concentrations were significantly reduced, ranging from 5-6 mg Cu kg⁻ 568 ¹ in nominal 500 mg CdTe kg⁻¹ dw CdTe-COOH and NH₄⁺ and nominal 2000 mg CdTe 569 kg^{-1} dw CdTe-COOH exposures; compared to an expected concentration of 8 – 10 mg Cu kg⁻¹ dw in controls in the fresh soil study. Manganese (Mn) was significantly 570 571 depleted in earthworms from the nominal 500 mg CdTe kg⁻¹ dw CdTe-NH₄⁺ and 572 nominal 2000 mg CdTe kg⁻¹ dw CdTe-PEG, COOH and NH₄⁺ exposures, ranging from 14 - 15 mg Mn kg⁻¹ dw compared to controls at 43 mg Mn kg⁻¹ dw in the fresh soil 573 study. A significant strong negative correlation was found between the tissue Cd and 574 Cu or Mn ($r_s = -0.6$ for both, P < 0.05, Spearman). 575

576 In the experiment with the aged soil, there were no significant effects on the electrolytes in the earthworms, however trace metals, Cu and Mn, were affected 577 (Table S2). The concentration of Cu was significantly reduced compared to the control 578 in the nominal 50 and 500 mg CdTe kg⁻¹ dw CdTe-COOH and NH₄⁺ and in all the 2000 579 580 mg CdTe kg⁻¹ dw exposures (ranging from 2 - 5 mg Cu kg⁻¹ dw). The concentrations of Mn were significantly reduced compared to the control only in the nominal 2000 mg 581 CdTe kg⁻¹ dw CdTe-COOH and NH₄⁺ exposures (ranging from 18 – 25 mg Mn kg⁻¹ 582 dw). There was also a significantly strong negative correlation between the tissue 583 584 concentration of Cd and Cu or Mn ($r_s = -0.5$ for both, P < 0.05, Spearman) in the aged soil experiment. 585

586

587 3.7 Total glutathione as an oxidative stress marker

588 Total glutathione (GSH) concentrations were measured as an oxidative stress markers in the earthworms (Fig. 4B). Earthworms from the fresh soil experiment 589 590 sustained a normal total glutathione concentration (~16 nmol mg⁻¹ protein). In the 591 CdTe-bulk exposures there was no significant effect on the total GSH concentration. Also, there was no significant effects in the differently-coated CdTe QD exposures 592 when compared to the controls (P > 0.05, ANOVA). The results of the total GSH from 593 594 the aged soil study were not considered reliable due to thawing of the frozen samples (data not shown). 595

596

597 3.8 Quantification of plant material

598 It was not the aim of this study to assess the effects of CdTe QDs on plants, but there was some incidental plant growth in the soil during the six months of ageing. 599 600 Prior to the start of the aged soil exposures of the earthworms, the opportunity was 601 taken to measure the plant cover on the exposure soils. The unexposed control soils 602 had an overall cover of > 85 % of plants and ~ 80 % of moss. Overall, effects on plant 603 cover were evident only at the nominal 500 and 2000 mg CdTe kg⁻¹ dw exposures 604 (Fig. S2). There was a material size effect with the CdTe QDs being more toxic to plant growth than the CdTe QD bulk counterpart, and there was a coating-mediated effect 605 606 for the QDs (Fig. S2). At the highest concentration of the CdTe bulk material, the 607 vascular plant and moss cover was significantly reduced to around 20% of that in the 608 controls (Fig. S2). Clear effects on plant cover were also evident in the nominal 500 609 mg CdTe kg⁻¹ dw CdTe QD exposures, where the PEG- and COOH-coated QDs showed the least amount of plants and moss, while the CdTe-NH4⁺ exposure was 610 similar to the CdTe-bulk and the controls. However, at the highest QD ENM test 611 612 concentrations there were no vascular plants or moss growth (Figs. S2 and S3), except in the CdTe-PEG treated soils, which had a very limited (around 10 %) moss 613 614 cover. The highest CdTe QD treatments with no plant matter had a biofilm-like layer 615 and some mould, and the soil had dried in places (Fig. S4).

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- 617

618 4. Discussion

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This is the first study to investigate the effects of CdTe QDs with different surface 620 621 coatings on earthworms compared against a micron-sized (bulk) CdTe material in 622 fresh soil. The hazard to earthworms was also assessed after 6 months of ageing the 623 soils. In the fresh soil experiment, none of the CdTe materials (bulk or ENMs) altered 624 survival, but exposure to CdTe QDs caused a significant reduction in the number of juveniles produced and disturbances to essential metal concentrations in the 625 earthworm tissues. There was also evidence of a coating-mediated effect, which 626 627 altered the release of dissolved Cd and Te from the materials, rather than an effect 628 due to the type of chemical substance in the coating itself. In the aged soil experiment, unlike the fresh soil study, the CdTe-NH4⁺ QDs reduced the survival of the earthworms. 629 630 Both the CdTe-bulk and the CdTe QDs caused more pronounced effects on juvenile

production in the aged soil experiment compared to the fresh soil study, but the CdTeQDs remained more toxic than CdTe-bulk QDs.

633

4.1 Extractable metal fractions in the soil in relation to Cd and Te accumulation byearthworms

The exposures were confirmed by measuring the total metal concentrations of 636 637 Cd and Te in the soil (Fig. 1), and in the earthworms (Fig. 3). For the unexposed controls, the background concentration of Cd in Lufa 2.2 soils was low (0.6 ± 0.2 mg 638 639 Cd kg⁻¹ dw, Figure 1) and consistent with previous reports for the same soil [e.g., 0.38] 640 mg Cd kg⁻¹ dw, (Keshavarz Jamshidian et al., 2017)]. In the control soil, less than 10 % 641 of the total background Cd concentration was water and acid extractable, indicating 642 that the background Cd in the control soil was not mobile. There was no Te detected in the control soils, as expected for a rare metalloid with very low background 643 concentrations in unpolluted soil [e.g., values < 5 μ g kg⁻¹ dw, (Belzile and Chen, 644 2015)]. The background concentration of Cd in the earthworms was also in agreement 645 with previous studies using the same soils $[4.23 \pm 0.34 \text{ mg Cd kg}^{-1} \text{ dw this study}; 5.6$ 646 mg Cd kg⁻¹, (González et al., 2013)], while tissue Te concentration remained below 647 648 the detection limit.

649 The soils treated with CdTe QDs had elevated Cd and Te concentrations 650 compared to the unexposed control soils, and as expected, the measured metal concentrations increased with the nominal exposure concentration. This occurred in 651 652 both the fresh and aged soil experiments (Fig. 1). However, within each soil dose for each type of CdTe QD, there was also some slight variation in the measured 653 654 concentration of total Cd or Te according to the type of coating on the starting material. 655 This was likely due to the stoichiometry of the materials giving rise to slightly different 656 ratios of Cd or Te to the coating, by mass. The effect was small within each soil dose, 657 and only influenced the Te measurements in the soil for the CdTe-PEG and CdTe-COOH materials (Fig. 1). This was very similar to our previous findings on CuO ENMs 658 that had the same coatings; where the proportion of the overall mass due to the particle 659 660 core relative to the coating inevitably varied with the type of coating (Tatsi et al., 2018). Nonetheless, in the present study, within each type of coating on the CdTe QDs, the 661 ratios of the measured Cd and Te concentrations in the soil were consistent for that 662 663 material.

In the fresh soil experiment, attempts were made to determine the 664 bioaccessible fractions of Cd and Te in the soil. The water- and acid-extractable 665 666 fractions of metals from the soils were plotted against the measured metal concentrations in the earthworms (Fig. 2). For the unexposed controls, the Cd or Te 667 concentrations were very low or at the detection limit, as expected. The CdTe bulk 668 material also revealed very low water- or acid-extractable metal concentrations (e.g., 669 670 water-extractable of around 0.1 Cd kg⁻¹ dw or 0.035 mg Te kg⁻¹ dw of soil, Figs. 2C and D), which represented <1% of those metals in the bulk material exposures. The 671 672 CdTe bulk materials also showed negligible metal dissolution (Fig. S1). Together, this suggests the metals in the CdTe bulk material are not bioaccessible as dissolved 673 674 metals to the earthworms, and that the observed Cd and Te accumulation in the 675 earthworms was attributed to uptake of the intact CdTe bulk material. The 676 concentration of Cd compared to Te in the earthworm tissue (compare panels A and B in Fig. 3 for fresh soil) was consistent at around 1.7 which also indicates some 677 accumulation of the intact CdTe bulk material. 678

The situation was somewhat different for the ENMs in soil compared to the bulk 679 680 material (Fig. 2), with all the CdTe QDs showing a water- and acid-extractable fraction 681 of Cd in the soil, but not much extractable Te. The dialysis experiment also showed 682 more dissolution of Cd than Te from the ENMs (Fig. S1); which together implies the 683 Cd was more bioaccessible than the Te in CdTe QDs. Furthermore, the Cd accumulation in the earthworms was at least an order of magnitude higher than Te 684 685 (Figs. 2 and 3), despite no such differences in the exposure concentration of the metals in the soil from the ENMs (Fig. 1). This observation can only be explained by the 686 687 differential uptake and/or retention of Cd compared to Te in the tissues, and therefore 688 that the CdTe QDs do not remain intact. It has been suggested that CdTe QD likely 689 degrade in the soil to release dissolved Cd (Navarro et al., 2008). In aqueous media 690 for algae, the dissolution of Cd from CdTe-containing QDs and subsequent free metal ion uptake can explain most of the Cd accumulation by the organisms 691 [Chlamydomonas reinhardtii, (Domingos et al., 2011)]. Differential accumulation of Cd 692 693 and Se have been noted in earthworms exposed to CdSe QDs [9-20 nm, (Stewart et al., 2013)]. Fluorescence imaging has shown what appears to be intact CdTe QDs 694 695 (assumed from the fluorescence signal itself) in the gut lumen of the nematode, C. 696 elegans, (Qu et al., 2011) and the gastrointestinal tract of larvae of the leaf roller moth 697 (Al-Salim et al., 2011); although neither study demonstrated clear physiological uptake

698 inside the cells of the internal organs, it does suggest that CdTe QDs can be presented 699 to the gut of terrestrial organisms. Regardless, the contrasting behaviour of the CdTe 699 bulk material to remain intact with negligible dissolution, compared to the nano forms 691 which show some dissolved Cd, may simply arise from size-dependent effect on the 692 surface to volume ratio of the materials, although further experiments are needed to 693 verify this.

704 In contrast to knowledge on the biology of Cd, there are almost no studies on 705 Te accumulation in earthworms. The Te was not extractable from CdTe QDs (Fig. 2) 706 and so not very bioaccessible. Te was not accumulated as strongly as Cd (compare 707 Fig. 3A with 3B in the fresh soil experiment) and there was also no clear dose-708 dependent increase in the tissues. For the low concentrations of Te found in the 709 earthworms, this may suggest that either earthworms are effectively excreting Te, or the unidirectional uptake mechanism (currently unknown for Te) is much slower than 710 that for Cd. However, once inside the tissue, it has been suggested that dissolved Cd 711 712 and Te may together form biogenic CdTe QDs (Stürzenbaum et al., 2013). The mineralisation of biogenic particles in earthworms from dissolved metals is well known 713 714 (Brown, 1982), so the presence of particles in the tissue may not infer uptake of the 715 intact original material.

716

4.2 Nanomaterial coating-mediated effects on metal accumulation

718 A central concern for the environmental risk assessment of ENMs is whether or 719 not any particle coating-related effects would make the bioaccumulation potential of materials so different, that each form and type of coating would be required to be 720 721 treated as a new substance with respect to the regulation of chemicals. Similarly, there 722 are concerns that the transformation of particles in the environment would lead to 723 different hazards compared to the original 'pristine' ENM (Gardea-Torresdey et al., 724 2014; Lowry et al., 2012). First consider the issue of coating effects on bioaccumulation. The water- and acid-extractable fractions of Cd and Te did cluster 725 726 by the type of coating with respect to the metal accumulation in the earthworms (Fig. 727 2). However, we do not interpret that as a coating-effect *per se*; but simply as a 728 consequence of the effect of the coating on the dissolution of metal from the materials (Fig. S1). This 'coating-mediated effect' on the extractability of Cd and Te in the soil 729 730 from the CdTe QD exposures showed that more Cd was extractable from NH₄⁺-coated 731 QDs compared to the others (Fig. 2). In theory, the chemical bond anchoring the

732 coating (i.e., the surface ligand) to the core could add a mechanical stress to the metal atoms on the exterior of the core to promote dissolution [ligand-promoted dissolution, 733 734 (Louie et al., 2016)]. The exact mechanisms involved in this phenomena remain to be elucidated, but for example, the repulsion of ligands of the same charge might stiffen 735 the coating to add mechanical stress to the anchoring point on the core; hence the 736 coatings designed with surface charge tend to have more water- or acid-extractable 737 738 Cd in the soil (Fig. 2). Or it may relate to lipid solubility, with the PEG coating forming 739 a barrier to slow the dissolution of solutes.

740 Regardless of the mechanism(s), this coating-mediated effect on the dissolution (Fig. S1), and extractable metal fractions in the soil (Fig. 1), was not 741 742 observable in the mean values for total metal accumulation of Cd or Te in the 743 earthworms (Fig. 3). For example, the extractable Cd in soil from the CdTe-NH₄⁺ QDs was relatively high (Fig. 2), but this did not lead to the highest Cd accumulation in the 744 745 earthworms (Fig. 3). In short, the running order of any coating-mediated effect on the 746 physico-chemical properties of the QDs was not in the same order as the subsequent total metal accumulation in the earthworms. This was not resolved by normalising the 747 748 total metal bioaccumulation in the earthworms against the measured total metal 749 concentration in the soil (i.e., the transfer factor: the total metal concentration in the 750 earthworm divided by that in the soil, data not shown). Therefore, while there are some 751 coating-mediated effects on the accumulation of Cd and Te (Fig. 3), they are not explained by free metal ions released from the ENMs, although they are related to the 752 753 total metal in the soil during the exposure. From the viewpoint of environmental protection, threshold levels for metals in soil have been suggested for dissolved metals 754 755 on the basis of free ion activity (Lofts et al., 2004), and while ligand models have also 756 been applied to ENMs in soil (Judy and Bertsch, 2014), the assumption that free metal 757 ions lead to toxicity or accumulation is not proven for CdTe QDs here, and needs to 758 be verified for many other ENMs in soil.

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760 4.3 Effect of soil ageing on metal accumulation

Soil guidelines are not available for ENMs, and while the individual parameters in colloid theory such as soil pH and ionic strength have been considered (Cornelis et al., 2014), the sum effect of all the possible transformations of an ENM that constitute 'ageing' of the material in soil are not yet understood. In the present study, only the bulk material showed a soil ageing-dependent effect on total metal accumulation in 766 the earthworms; with much more metal accumulated in the aged soil experiment 767 (compare Fig 3A and 3B for Cd). This occurred, despite the total concentration of Cd 768 and Te in the soil being slightly lower in the aged compared to fresh soil experiments (Fig. 1); suggesting greater bioavailability and hazard of the CdTe bulk material over 769 770 time in soil. It was not possible to determine the water- and acid-extractable metals in 771 the aged soil at the start of the aged experiment, as this would have required 772 destructive sampling and disturbance of the soil boxes. However, the slightly lower 773 metal concentrations in the aged soil was somewhat expected, and likely due to 774 previous metal uptake by the earthworms in the fresh soil study, and metal uptake by 775 the plants that had subsequently grown in the exposure containers during soil ageing. 776 Thus the incidental plant growth in the soil (Figs. S2 and S3), may have partly 777 remediated the total metal in the soils. Notably, the ratio of Cd:Te in the tissue of 778 earthworms exposed to the bulk material was different in the aged soil experiment 779 compared to that in fresh soil; suggesting that the bulk material had at least partly 780 dissolved, with then separate uptake mechanisms for Cd and Te by the earthworms 781 (as discussed above). The bulk material had a maximum dissolution rate of $< 0.1 \, \mu g$ h⁻¹ for Cd (Table 1), and even at the highest soil dose with around 800 mg Cd kg⁻¹ of 782 783 soil (Fig.1), in a 250 g box of the soil (~ 200 mg Cd in total), this would equate to a 784 dissolution of 16.8 µg of Cd per week, or about 403 µg in six months. So, the material 785 form would still remain in excess after six months of ageing, and ignoring 786 transformation processes such as suphidation that might stabilise the materials (Lead 787 et al., 2018). The mechanism causing this soil ageing-dependent change in metal accumulation by earthworms requires further investigation. 788

789 In contrast to the bulk material, for all the CdTe QDs exposures in aged soil, 790 the accumulation of Cd was similar to that in the fresh soil study, although slightly 791 higher (Fig. 3C). There was also little difference in Te accumulation from the aged 792 compared to fresh soil study (compare Fig 3B and 3D) for the CdTe QDs. It is not clear why soil ageing failed to accelerate the metal accumulation in the earthworms, as it 793 794 did for the bulk material. The CdTe QDs are smaller and with a faster dissolution rate 795 than the bulk material (Fig. S1, Table 1), and so more uptake of free metal ions might be expected. But this was not the case, suggesting some other particle metric was 796 797 controlling the bioavailability of the CdTe QDs. Apart from size, the other key 798 difference was that CdTe QDs were all manufactured with a surface coating, when the 799 bulk material was not. So a lack of effect of soil ageing on metal accumulation may be 800 due to the presence of the coatings. It is possible that organic ligands in the soil, such as the polyanionic charges of humic and fulvic acids, could electrostatically attract the 801 positive coating of the CdTe-NH4⁺ QDs, repulsion in the case of CdTe-COOH QDs, or 802 show steric entanglement with the CdTe-PEG QDs. Similar ideas have been proposed 803 804 for the behaviour of QDs in natural water (Rocha et al., 2017). However, such phenomena would demand a clear and consistent coating-effect, but this was not 805 806 observed with different coating-mediated effects on bioaccumulation in the fresh 807 compared to aged soil experiments, and with exposure dose within each experiment 808 (Fig. 3). Regardless of the mechanisms, the bulk material presents a bioaccumulation 809 concern with soil ageing that is greater than the nano forms.

810

811 4.4 Effects on survival and body weight

812 In the fresh soil experiment, the control earthworms survived, gained weight, and did not show any abnormalities, indicating that the animals were healthy. 813 814 Furthermore, there were no effects on survival in any of CdTe treatments; and in the CdTe-bulk treatment the earthworms also gained weight (Table 2). However, despite 815 816 good survival, the animals in the CdTe QD treatments lost biomass (Table 2). There 817 was also a negative correlation between the metal concentrations in the tissue and biomass (Fig. S5). The loss of biomass was partly attributed to a lack of feeding 818 819 coupled with some shedding of the posterior segments. There were no consistent 820 effects of percentage change in body mass by the type of coating on the CdTe QDs in 821 the fresh soil experiment; and by day 28 all the earthworms in the 2000 mg kg⁻¹ CdTe QD treatments had lost around 50% of their body mass (Table 2). The latter suggests 822 823 the earthworms at the higher doses were just below the threshold for mortality. Indeed, 824 the shedding of posterior segments has been interpreted as a survival strategy (Sims 825 and Gerard, 1985) and possibly an 'excretion' mechanism for metals in earthworms. 826 The survival of the earthworms at the nominal 50 and 500 mg kg⁻¹ CdTe QD concentrations in the fresh soil experiment was expected with respect to Cd toxicity, 827 as earthworms are known to tolerate very high concentrations of dissolved Cd [28 day 828 mortality $LC_{50} = 588$ mg Cd kg⁻¹ dw, (Van Gestel et al., 1991)], which are partly 829 chelated by metallothionein in the tissue (Stürzenbaum et al., 2001). Furthermore, the 830 831 extractable Cd from the fresh soil experiment in most of the QD exposures was far 832 less than the expected lethal concentration of dissolved Cd, except in the nominal 2000 mg CdTe kg⁻¹ dw CdTe-NH₄⁺ QD exposures. In earthworms, a critical body 833

residue (CBR) of around 642 mg Cd kg⁻¹ dw is suggested to cause 50 % mortality 834 (Conder and Lanno, 2000). This concentration of Cd was generally exceeded in 835 836 earthworms exposed to the highest concentrations of CdTe QDs in the fresh (and aged) soil experiments (Fig. 3). Again, supporting the notion that the worms were close 837 838 to the threshold for mortality at the highest exposure concentration. The absence of effects of the CdTe bulk material on growth and survival (Table 2), and the relatively 839 840 modest Cd accumulation in those earthworms (Fig. 3), suggests the bulk material is 841 much less hazardous than the nano forms in fresh soil.

842 In the aged soil experiment, like in the fresh soil study, the control earthworms 843 survived and were healthy without a statistically significant loss of body mass. Similar to the fresh soil experiment, survival was also not affected by the CdTe-bulk 844 845 exposures. However, unlike the situation in fresh soil, the nominal 2000 mg CdTe kg⁻ ¹ dw test concentration of the CdTe bulk material caused a reduction in biomass (Table 846 2). Notably, this was accompanied by a much higher total Cd concentration in those 847 earthworms in the aged soil compared to fresh soil (Fig. 3). Together, this suggests 848 849 that the bulk material becomes more hazardous with soil ageing because the Cd in 850 the material becomes partly dissolved (discussed above) and therefore bioavailable 851 for uptake.

In contrast to the fresh soil experiment, exposure to CdTe QDs in aged soil 852 853 caused considerable mortality, even at the nominal concentration of 50 mg kg⁻¹ dw of CdTe QDs (Table 2). The CdTe QDs were, therefore, much more toxic after ageing of 854 855 the soil. In addition, a coating-mediated effect on survival was also apparent in the aged soil experiment, with the toxicity ranking CdTe-PEG < CdTe-NH4⁺ < CdTe-COOH 856 857 QDs at the highest exposure concentration (Table 2). In a study using the same coatings on CuO ENMs at a concentration of 1000 mg Cu kg⁻¹, the NH₄⁺-coated CuO 858 859 ENMs were also acutely toxic following a period of ageing, however, COOH-coated 860 CuO ENMs were not (Tatsi et al., 2018). This implies that, while coating-mediated effects can occur, the precise ranking of the effect also depends on the type of metal 861 core in the ENM. This might be expected, because the stability constants of metals 862 863 with organic ligands vary according to the charge density of the metal and the type of ligand (Albert, 1950). However, stability constants are not yet available for metals in 864 the solid or crystalline state as found in nanomaterials with ligands of different chain 865 866 lengths or terminal residues.

867 The increased toxicity of the CdTe QDs in aged soil was also reflected in the changes of body mass, with generally greater loss of body mass than that observed 868 869 in the fresh soil experiment (Table 2). A coating-mediated effect on growth was also 870 apparent, with the most prominent effects in the NH4⁺- and COOH-CdTe QD 871 exposures (Table 2). The same ammonium coating also significantly reduced earthworm biomass in exposures to 1000 mg Cu kg⁻¹ dw as CuO-NH₄⁺ (Tatsi et al., 872 873 2018). Changes in the appearance and feeding behaviour of the worms were also evident at lower concentrations in the aged soil experiment, and consistent with the 874 875 notion that the aged soil was more hazardous to the health of the earthworms.

876

877 4.5 Effects on reproduction

878 Reproduction is regarded as the most sensitive endpoint for metal exposure of earthworms. The control earthworms produced the expected number of juveniles in 879 both, fresh and aged soil experiments, meeting the validity criteria of the OECD TG 880 881 222 (OECD, 2004). In the fresh soil experiment, the number of juveniles produced was not affected in the CdTe-bulk exposures, while in the CdTe QD exposures the number 882 883 of juveniles decreased as the exposure concentration elevated (Table 2). A coating-884 mediated effect was evident, with the EC₅₀ values in the order: CdTe-COOH < CdTe-885 NH_4^+ < CdTe-PEG (Table 2); suggesting the PEG-coated material was the least 886 hazardous with respect to juvenile production. If the measured concentration of total 887 Cd in the exposure soils is considered in the fresh soil experiment (ranging from \sim 22 888 to 780 mg Cd kg⁻¹ dw, Fig. 1), the EC₅₀ values of between 65-108 (Table 3) are broadly comparable to the 56 day EC₅₀ for cocoon production of 46.3 mg Cd kg⁻¹ dw in 889 890 earthworms exposed to 0-300 mg Cd kg⁻¹ soil (Spurgeon et al., 1994). This suggests 891 the reproductive toxicity of the CdTe QDs in the fresh soil study could mostly be 892 explained by Cd exposure, and this was corroborated by a strong negative correlation 893 of juvenile production with the tissue Cd concentration, rather than tissue Te concentration, in the adult worms (Fig. S5). The decrease in reproductive success is 894 also consistent with bioenergetic theory, where the adult earthworms showing the least 895 growth (biomass) are expected to invest less energy in reproduction. Lower adult 896 biomass has been shown to be related to lower reproduction in regulatory tests 897 (OECD, 2004). Crucially, the juvenile earthworms also showed a decrease in biomass, 898 899 and this followed the same overall pattern as the effects on reproduction in the adults 900 (Table 3), suggesting the exposures also impaired off-spring quality. There are few

901 reports of juvenile production from adult earthworms exposed to ENMs, but like the
902 present study, TiO₂ ENMs also decrease juvenile production in earthworms (Schlich
903 et al., 2012).

904 In the aged soil experiment, the control earthworms produced more juveniles 905 than in the fresh soil study (Table 3). This is likely due to the higher organic matter 906 (i.e., plant roots) in the aged soil that is known to encourage earthworms to produce 907 more offspring (Lahive et al., 2017). Unlike the fresh soil experiment, in aged soil, the 908 CdTe bulk material inhibited juvenile production to the extent that no juveniles were 909 produced at the highest exposure concentration (Table 3) and the Cd concentration in 910 the adult earthworms strongly correlated with this reproductive failure (Fig. S5); again 911 suggesting that mobilisation of Cd from the bulk material during soil ageing caused the 912 toxicity. Furthermore, the reproductive toxicity of the CdTe QDs also increased in the aged soil experiment, but the difference between the coated QDs was less 913 pronounced, although CdTe-NH4⁺ QDs remained most hazardous (Table 3). The 914 further reduction in reproduction in the aged soil experiment may be explained by the 915 916 higher accumulation of Cd and Te in the earthworms (Fig. 3), which was also associated with lower biomass in the adult worms. So bioenergetically, like the fresh 917 918 soil experiment, poor growth or health in the adults likely gave rise to fewer offspring.

919

920 4.6 Effects on ion regulation

Cadmium is known to cause ion regulatory disturbances in organisms 921 922 (Stürzenbaum et al., 2004), therefore, the sodium pump activity was assessed. In the fresh soil study, earthworms in the controls sustained sodium pump activity (Fig. 4), 923 924 as well as an overall concentration of Na⁺ and K⁺ (Table S2), as expected in the 925 earthworms (Tatsi et al., 2018). The sodium pump activity was inhibited the most in 926 the CdTe-bulk treatment and at the highest exposure concentration (Fig. 4). However, 927 exposure to CdTe QDs in fresh soil also inhibited the sodium pump. Of the ENMs, the CdTe-PEG and -NH4⁺ QDs were most potent in inhibiting the sodium pump activity 928 929 (Fig. 4). In the nominal 500 and 2000 mg CdTe kg⁻¹ dw exposures to CdTe-NH₄⁺, 930 tissue K⁺ concentrations also decreased (Table S2), being consistent with inhibition of the sodium pump. There appears to be no other reports of the effects of CdTe QDs on 931 932 osmoregulation in earthworms, and whether the sodium pump inhibition is caused by 933 Cd alone, Te alone, or both, requires further investigation.

934 Several essential metals for earthworm health, were also measured (Table S2). 935 Of these, there were some transient changes in Zn and Fe in the fresh soil experiment, 936 but no clear dose effect, and no effects in the aged soil experiment. However, copper 937 and manganese showed some interesting changes. The control earthworms showed 938 the expected tissue Cu [8 – 10 mg Cu kg⁻¹ dw, (Streit, 1984; Tatsi et al., 2018)] and 939 Mn concentrations [43-54 mg Mn kg⁻¹ dw this study; ~ 50 mg Mn kg⁻¹ dw in *L. rubellus*, 940 (Oste et al., 2001)]. The manganese concentration was reduced by almost 3 times in 941 the CdTe QD ENM exposures, for both the fresh and aged soil experiments, and also 942 for the bulk material exposures in the aged soil experiment (Table S2). The 943 mechanistic cause of this Mn depletion in earthworms requires further investigation, but Cd-dependent inhibition of Mn uptake has been observed in fibroblast cell lines 944 945 (Yanagiya et al., 2000), possibly mediated by competition for uptake on divalent metal ion transporter 1 (DMT1). Copper depletion was correlated with higher tissue Cd (Fig. 946 S5) in the CdTe-COOH exposures (500 and 2000 mg CdTe kg⁻¹ dw) and CdTe-NH₄⁺ 947 exposure (500 mg CdTe kg⁻¹ dw), and with the effects evident in both fresh and aged 948 949 soil experiments (Table S2). The loss of Cu (and Mn) might arise from a passive non-950 specific electrolyte leak associated with tissue injury since other essential elements 951 such as K, Ca and Mg concentrations were significantly reduced compared to the controls in the fresh soil (Table S2). However, this seems unlikely in the absence of 952 953 glutathione depletion (Fig. 4, no clear difference in total GHS pool were observed) and the major electrolytes were undisturbed in the tissues in the aged soil experiment, 954 955 despite some tissue Cu depletion. Cu is a regulated metal in earthworms (Streit, 1984), and it could simply be that Cd²⁺ is displacing Cu from ligands such as metallothionein 956 957 and glutathione in the tissue. Regardless of the mechanism, both Cu and Mn were 958 decreased in the tissue of earthworms from both the fresh and aged soil studies, and 959 with the same coated QDs, COOH and NH₄⁺, being most potent for this effect (Table 960 S2).

962 *4.7 Plant growth during soil ageing*

It was not the purpose of this study to assess the effects of CdTe materials on 963 964 plant growth, but some incidental observation on plant coverage were made (Figs. S2 and S3). Similar to the earthworms, the QDs were more toxic to the plants and moss 965 966 than the bulk material and there was a coating-mediated effect on plant coverage; with CdTe-COOH and CdTe-NH4⁺ QDs abolishing growth at the highest 967 the 968 concentrations. In hydroponic media, plant cuttings from poplar trees exposed to 969 Cd/Se-containing QDs with cationic or anionic surface coatings, showed apparently 970 faster uptake of the cationic material; possible because it is electrostatically attracted 971 to the negatively charged surface of the plant (Wang et al., 2014). The effects of 972 organic coatings on CdS QDs have also been recently explored on soya bean seedling 973 (Majumdar et al., 2019); with the amino acid profiles in the plants clustering by 974 material, although the coating-effect was not consistent across a range of endpoints. Currently, there are too many data gaps to offer a clear mechanistic understanding of 975 976 ENM effects on plants (Schwab et al., 2016), and any apparent coating-mediated 977 effects on vascular plants requires further investigation.

978 The earthworms were inevitably feeding on the plant root material in the aged 979 soil experiment, and so this might be both a source of nutrition and metal exposure 980 from any nanomaterial on the surface of the roots. However, the plant growth or the 981 presence of ENMs in soil might be expected to alter the soil properties. Indeed, QDs 982 may change the porosity and the transport of solutes in soil with subsequent effects 983 on the plants and invertebrates (Al-Salim et al., 2011). In the present study, the 984 moisture content and pH was adjusted in the soils, but changes in the soil structure 985 with soil ageing cannot be excluded as a factor in the metal accumulation of the 986 earthworms in the aged soil experiment.

987

988 4.8 Conclusions and regulatory perspective

Taken together, the results here compared to the known toxicity of Cd in earthworms indicate the CdTe QD ENMs are of similar acute and reproductive toxicity to earthworms. When comparing the bulk vs nano-scale CdTe QDs, there may be a size-effect, since the nano-scale CdTe QDs were generally more toxic than the CdTebulk counterpart. There was also a coating-mediated effect, where the most toxic QD was generally the CdTe-COOH material, closely followed by CdTe-NH4⁺ QDs. However, this was not a direct coating effect *per se*, but likely a consequence of other

996 subtle changes in the physico-chemical properties of the material during manufacture, 997 such as the differences in the ratio of the apparent metal concentrations in each 998 material. In keeping with this notion, the toxic effects seen in the study were related to 999 accumulation of Cd in earthworm tissues, however, this was not easily explained by 1000 the measured dissolution rate of Cd from the QDs (Table 1) or the extractability of Cd 1001 from the soils (Fig. 2). The existing environmental risk assessment for dissolved Cd is 1002 likely to be protective of both bulk and nano forms of CdTe materials, although the type of coating and any subsequent changes in particle properties associated with 1003 1004 adding the coating, should be considered in the environmental risk analysis. Crucially, the toxicity of the CdTe QDs increased after a period of ageing of the soil, suggesting 1005 1006 that the hazard should be monitored, and any environmental risk assessment revisited 1007 in order to maintain protection of the soil ecosystem. Finally, it would be important to ascertain the mechanisms of toxicity for Te, since it is widely used in a QD form and 1008 currently there are no environmental risk assessments which give guidance on safe 1009 levels of Te in soils. 1010

1011

1012 Conflicts of interest

1013 The authors alone are responsible for the content and writing of the paper. The authors 1014 report no conflicts of interest.

1015

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1027 **References**

Al-Salim, N., et al., 2011. Quantum dot transport in soil, plants, and insects. Science
 of the Total Environment. 409, 3237-3248.

- Albert, A., 1950. Quantitative studies on the avidity of naturally occurring substances
 for trace metals. 1. Amino-acids having only two ionizing groups. Biochemical
 Journal. 47, 531-538.
- Arnold, R., Hodson, M., 2007. Effect of time and mode of depuration on tissue
 copper concentrations of the earthworms *Eisenia andrei*, *Lumbricus rubellus* and *Lumbricus terrestris*. Environmental Pollution. 148, 21-30.
- 1036Ba, L. A., et al., 2010. Tellurium: an element with great biological potency and1037potential. Organic & Biomolecular Chemistry. 8, 4203-4216.
- Belzile, N., Chen, Y.-W., 2015. Tellurium in the environment: A critical review
 focused on natural waters, soils, sediments and airborne particles. Applied
 Geochemistry. 63, 83-92.
- Besinis, A., et al., 2014. The antibacterial effects of silver, titanium dioxide and silica
 dioxide nanoparticles compared to the dental disinfectant chlorhexidine on
 Streptococcus mutans using a suite of bioassays. Nanotoxicology. 8, 1-16.
- Brown, B. E., 1982. The form and function of metal-containing 'granules'in invertebrate tissues. Biological Reviews. 57, 621-667.
- Buffet, P.-E., et al., 2014. Biochemical and behavioural responses of the marine
 polychaete *Hediste diversicolor* to cadmium sulfide quantum dots (CdS QDs):
 waterborne and dietary exposure. Chemosphere. 100, 63-70.
- 1049 Chlopecka, A., et al., 1996. Forms of cadmium, lead, and zinc in contaminated soils 1050 from southwest Poland. Journal of Environmental Quality. 25, 69-79.
- 1051 Conder, J. M., Lanno, R. P., 2000. Evaluation of surrogate measures of cadmium, 1052 lead, and zinc bioavailability to *Eisenia fetida*. Chemosphere. 41, 1659-1668.
- Cornelis, G., et al., 2014. Fate and bioavailability of engineered nanoparticles in
 soils: a review. Critical Reviews in Environmental Science and Technology.
 44, 2720-2764.
- Domingos, R. F., et al., 2011. Bioaccumulation and effects of CdTe/CdS quantum
 dots on *Chlamydomonas reinhardtii*–nanoparticles or the free ions?
 Environmental Science & Technology. 45, 7664-7669.
- Gardea-Torresdey, J. L., et al., 2014. Trophic transfer, transformation, and impact of
 engineered nanomaterials in terrestrial environments. Environmental Science
 & Technology. 48, 2526-2540.
- González, V., et al., 2013. Assessing the impact of organic and inorganic
 amendments on the toxicity and bioavailability of a metal-contaminated soil to
 the earthworm *Eisenia andrei*. Environmental Science and Pollution
 Research. 20, 8162-8171.
- Handy, R. D., et al., 2012. Practical considerations for conducting ecotoxicity test
 methods with manufactured nanomaterials: what have we learnt so far?
 Ecotoxicology. 21, 933-972.
- Hardman, R., 2005. A toxicologic review of quantum dots: toxicity depends on
 physicochemical and environmental factors. Environmental Health
 Perspectives. 114, 165-172.
- Hopkin, S. P., 1989. Ecophysiology of metals in terrestrial invertebrates. Elsevier
 Applied Science Publishers, Barking, UK.
- Judy, J. D., Bertsch, P. M., Bioavailability, toxicity, and fate of manufactured
 nanomaterials in terrestrial ecosystems. In: D. Sparks, (Ed.), Advances in
 Agronomy. Elsevier, Amsterdam, 2014, pp. 1-64.
- Kairdolf, B. A., et al., 2013. Semiconductor quantum dots for bioimaging and
 biodiagnostic applications. Annual Review of Analytical Chemistry. 6, 143 162.

- Keshavarz Jamshidian, M., et al., 2017. Toxicokinetics and time-variable toxicity of
 cadmium in *Oppia nitens* Koch (Acari: Oribatida). Environmental Toxicology
 and Chemistry. 36, 408-413.
- Lahive, E., et al., 2017. Sewage sludge treated with metal nanomaterials inhibits
 earthworm reproduction more strongly than sludge treated with metal metals
 in bulk/salt forms. Environmental Science: Nano. 4, 78-88.
- Lead, J. R., et al., 2018. Nanomaterials in the environment: behavior, fate,
 bioavailability, and effects—an updated review. Environmental Toxicology and
 Chemistry. 37, 2029-2063.
- Lofts, S., et al., 2004. Deriving soil critical limits for Cu, Zn, Cd, and Pb: a method
 based on free ion concentrations. Environmental Science & Technology. 38,
 3623-3631.
- Louie, S. M., et al., 2016. Critical review: impacts of macromolecular coatings on critical physicochemical processes controlling environmental fate of nanomaterials. Environmental Science: Nano. 3, 283-310.
- Lowry, G. V., et al., 2012. Transformations of nanomaterials in the environment. Environmental Science & Technology 46, 6893-6899.
- Luo, C., et al., 2011. Heavy metal contamination in soils and vegetables near an ewaste processing site, south China. Journal of Hazardous Materials. 186,
 481-490.
- 1100 Majumdar, S., et al., 2019. Surface coating determines the response of soybean 1101 plants to cadmium sulfide quantum dots. NanoImpact. 14, 100151.
- McCormick, S. D., 1993. Methods for nonlethal gill biopsy and measurement of Na⁺,
 K⁺-ATPase activity. Canadian Journal of Fisheries and Aquatic Sciences. 50,
 656-658.
- 1105 Navarro, E., et al., 2008. Environmental behavior and ecotoxicity of engineered 1106 nanoparticles to algae, plants, and fungi. Ecotoxicology. 17, 372-386.
- 1107 OECD, OECD guidelines for testing of chemicals: Earthworm reproduction test 1108 (*Eisenia fetida*/ *Eisenia andrei*). TG 222. Paris, France, 2004.
- Oste, L. A., et al., 2001. Cadmium uptake by earthworms as related to the availability
 in the soil and the intestine. Environmental Toxicology and Chemistry. 20,
 1785-1791.
- 1112 Owens, C., Belcher, R., 1965. A colorimetric micro-method for the determination of 1113 glutathione. Biochemical Journal. 94, 705.
- Qu, Y., et al., 2011. Full assessment of fate and physiological behavior of quantum
 dots utilizing Caenorhabditis elegans as a model organism. Nano Letters. 11,
 3174-3183.
- 1117 Ramadan, S. E., et al., 1989. Incorporation of tellurium into amino acids and proteins 1118 in a tellurium-tolerant fungi. Biological Trace Element Research. 20, 225.
- 1119Rocha, T. L., et al., 2017. Environmental behaviour and ecotoxicity of quantum dots1120at various trophic levels: A review. Environment International. 98, 1-17.
- Schlich, K., et al., 2012. Effect of TiO₂ nanoparticles in the earthworm reproduction
 test. Environmental Sciences Europe. 24, 5.
- Schwab, F., et al., 2016. Barriers, pathways and processes for uptake, translocation
 and accumulation of nanomaterials in plants–Critical review. Nanotoxicology.
 10, 257-278.
- Sims, R. W., Gerard, B. M., 1985. Earthworms: keys and notes for the identification and study of the species. The Pitman Press, London.

- Sinha, P., et al., 2012. Fate and transport evaluation of potential leaching risks from
 cadmium telluride photovoltaics. Environmental Toxicology and Chemistry.
 31, 1670-1675.
- Spurgeon, D., et al., 1994. Effects of cadmium, copper, lead and zinc on growth,
 reproduction and survival of the earthworm *Eisenia fetida* (Savigny):
 assessing the environmental impact of point-source metal contamination in
 terrestrial ecosystems. Environmental Pollution. 84, 123-130.
- Srivastava, S., et al., 2016. Curcumin and β-caryophellene attenuate cadmium
 quantum dots induced oxidative stress and lethality in *Caenorhabditis elegans* model system. Environmental Toxicology and Pharmacology. 42, 55-62.
- 1138 Stewart, D. T., et al., 2013. Quantum dots exhibit less bioaccumulation than free 1139 cadmium and selenium in the earthworm *Eisenia andrei*. Environmental 1140 Toxicology and Chemistry. 32, 1288-1294.
- 1141 Streit, B., 1984. Effects of high copper concentrations on soil invertebrates 1142 (earthworms and oribatid mites). Oecologia. 64, 381-388.
- 1143 Stürzenbaum, S., et al., 2013. Biosynthesis of luminescent quantum dots in an 1144 earthworm. Nature Nanotechnology. 8, 57.
- 1145 Stürzenbaum, S. R., et al., 2004. Cadmium detoxification in earthworms: from genes 1146 to cells. Environmental Science & Technology. 38, 6283-6289.
- Stürzenbaum, S. R., et al., 2001. Metal ion trafficking in earthworms Identification of
 a cadmium-specific metallothionein. Journal of Biological Chemistry. 276,
 34013-34018.
- Tatsi, K., et al., 2018. Copper accumulation and toxicity in earthworms exposed to
 CuO nanomaterials: Effects of particle coating and soil ageing. Ecotoxicology
 and Environmental Safety. 166, 462-473.
- 1153 Van Gestel, C., et al., 1991. Influence of cadmium, copper, and pentachlorophenol
 1154 on growth and sexual development of *Eisenia andrei* (Oligochaeta; Annelida).
 1155 Biology and Fertility of Soils. 12, 117-121.
- 1156 Vassallo, J., et al., 2018. The minimum inhibitory concentration (MIC) assay with 1157 *Escherichia coli*: An early tier in the environmental hazard assessment of 1158 nanomaterials? Ecotoxicology and Environmental Safety. 162, 633-646.
- Wang, J., et al., 2014. Uptake, translocation, and transformation of quantum dots
 with cationic versus anionic coatings by *Populus deltoides × nigra* cuttings.
 Environmental Science & Technology. 48, 6754-6762.
- 1162 Wu, T., et al., 2015. MPA-capped CdTe quantum dots exposure causes neurotoxic 1163 effects in nematode *Caenorhabditis elegans* by affecting the transporters and 1164 receptors of glutamate, serotonin and dopamine at the genetic level, or by 1165 increasing ROS, or both. Nanoscale. 7, 20460-20473.
- Yanagiya, T., et al., 2000. Suppression of a high-affinity transport system for
 manganese in cadmium-resistant metallothionein-null cells. Journal of
 Pharmacology and Experimental Therapeutics. 292, 1080-1086.
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1170 **Tables and figures**

1171 **Table 1** Characterisation of the CdTe QDs used in the experiments.

	QD variant	¹ Manufacturer's Information	² Estimated primary particle size, diameter (nm)	³ NTA, hydrodynamic diameter (nm)	⁴ TGA, (% weight loss)	⁵ Maximum rate of dissolution in Milli Q water (µg h ⁻¹)					
					,	Cd	Те				
	CdTe-Bulk	CASRN 1306-25-8 (Sigma-Aldrich 256544), Lot No. MKBK6448V, ≥99.99% purity trace metal basis	< 250 μm	172 ± 28	NA	<0.1	0.2				
	CdTe- Polyethylene Glycol	Lot No.YF140402, 99% purity, size 3-5 nm.	< 4	156 ± 72	50.4 ± 8.3	6.4	2.3				
	CdTe- Carboxylate	Lot No.YF140402, 99% purity, size 3-5 nm.	< 4	84 ± 58	23.4 ± 4.2	3.9	1.3				
	CdTe- Ammonium	Lot No. YF140402, 99% purity, size 3-5 nm.	< 4	75 ± 50	8.8 ± 0.5	29.0	14.9				
1172	¹ Supplied as dry	powders, spherical particles	for the Nanosolutions pr	oject via Alexei Antipo	ov, PlasmaC	Chem Gmb	oH.				
1173	² It was not possib	ole to detect the QDs using e	electron microscopy, ther	efore an estimate is g	jiven (Denm	ark Techn	ical University)				
1174	³ NTA- Nanopartio	³ NTA- Nanoparticle Tracking Analysis using NanoSight using 100 mg CdTe I ⁻¹ ENM stocks in Milli-Q water at University of									
1175	Plymouth. Data are mean ± S.D. <i>n</i> = 3 samples										
1176	⁴ TGA – thermogra	⁴ TGA – thermogravimetric analysis. Triplicate measurements (technical replicates) made on the same batch of the dry powders									
1177	using a TGA 4000 (Perkin Elmer) under an N ₂ flow of 20 ml min ⁻¹ from 25°C to 995°C at a heating rate of 10°C min ⁻¹ at the										

- 1178 University of Manchester. Note, this measurement would include the pyrolysis of the organic coating and any other combustible
- 1179 (presumably organic) impurities in/on the materials.
- ⁵Maximum slope from rectangular hyperbola function of curve fitting used to estimate the maximum rate of dissolution of Cd and Te
- 1181 from dialysis experiments conducted at University of Plymouth, conducted exactly according to (Besinis et al., 2014).

1182 NA – not applicable to the test item.

Soil [CdTe], nominal mg CdTe kg ⁻¹ dw		Control		CdTe-bulk		CdTe-PEG		CdTe-COOH		CdTe- NH₄⁺	
		Survival	Biomass	Survival	Biomass	Survival	Biomass	Survival	Biomass	Survival	Biomass
Fresh s	oil, day 14										
50	Total	5	2.1 ± 0.1ª	5	2 ± 0.1ª	5	2.3 ± 0.1ª	5	1.8 ± 0.1ª	5	2.1 ± 0.1ª
	%	100	[↑] 15.9 ± 6.4	100	[↑] 3.0 ± 3.6	100	[↑] 16.9 ± 3	100	[↑] 22.7 ± 10.4	100	[↑] 25.7 ± 0.9
500	Total			5	2.03 ±0.1ª	5	1.9 ± 0.06ª	4.8 ± 0.2	2 ± 0.1ª	5	1.9 ± 0.1ª
	%			100	[↑] 12.9 ± 2.6	100	10 ± 3.9	95 ± 5	[↑] 11.9 ± 2.4	100	[↓] 4.2 ± 1.5
2000	Total			5	1.9 ± 0.06ª	5	1.3 ± 0.1 ^b	5	1.4 ± 0.1 ^b	4.8 ± 0.2	1.23 ± 0.1^{b}
	%			100	[↑] 4.2 ± 1.8	100	[↓] 24.9 ± 3.9	100	[↓] 28 ± 2.4*	95 ± 5	[↓] 36.6 ± 4.3
Fresh s	oil, day 28										
50	Total	5	2.2 ± 0.1 ^a	4.8 ± 0.2	1.9 ± 0.1ª	5	2.4 ± 0.1 ^a	5	2.1 ± 0.1 ^a	5	2.2 ± 0.1 ^a
	%	100	[↑] 19.9 ± 5.1	95 ± 5	[↓] 0.4 ± 4.6	100	[↑] 21.7 ± 6.1	100	[↑] 28 ± 11.9	100	[↑] 31.6 ± 3.1
500	Total			5	2.2 ± 0.1ª	4.8 ± 0.2	1.9 ± 0.1 ^{b #}	5	1.9 ± 0.1 ^{b #}	5	1.66 ±0.1 ^b
	%			100	[↑] 22.7 ± 3.7	95 ± 5	[↑] 8.1 ± 3.3	100	[↑] 8.7 ± 4.2	100	[↓] 15 ± 1.9
2000	Total			4.8 ± 0.2	2.14 ± 0.1 ^{a #}	5	0.91 ± 0.1 ^{c#}	5	1.05 ± 0.1°#	4.8 + 0.2	0.95 ± 0.1 ^{c #}
	%			95 ± 5	10.9 ± 3.6 [↑]	100	[↓] 49.4 ± 3.8	100	[↓] 46.6 ± 2.9	95 ± 5	[↓] 50.9 ± 4.4
Aged so	oil, day 28										
50	Total	5 ± 0.3	1.7 ± 0.1 ^a *	5	1.8 ± 0.1ª	4 ± 0.5	1.4 ± 0.1 ^a *	5	1.7 ± 0.1 ^{a*}	5	1.8 ± 0.1ª *
	%	95 ± 5	[↓] 8.1 ± 5.6	100	[↓] 4.1 ± 7.7	85 ± 10	[↓] 22.4 ± 8	100	[↓] 17.2 ± 5.1	100	[↓] 9.4 ± 3.6
500	Total			5	1.7 ± 0.1ª	4.8 ± 0.2	1.8 ± 0.2 ^a *	5 ± 0.3	1.3 ± 0.1 ^b *	4 ± 0.4	1.2 ± 0.1 ^b *
	%			100	[↓] 2.9 ± 2.7	95 ± 5	[↓] 17 ± 9.2	90 ± 5	[↓] 28.2 ± 4.9	80 ± 8	[↓] 33.7 ± 3.7
2000	Total			4.8 ± 0.2	1.3 ± 0.1 ^b	4 ± 0.5 *	0.7 ± 0.1 ^c *	2.8 ± 0.2 *	0.6 ± 0.1 ^c *	3 ± 1.1	$0.8 \pm 0.2^{\circ}$
	%			95 ± 5	[↓] 31.4 ± 8.6	75 ± 10	[↓] 59.6 ± 7.4	55 ± 5	[↓] 64.7 ± 5.8	60 ± 25	[↓] 56.8 ± 7.2

Table 2 Survival and biomass of earthworms following 14 and 28 days exposure to CdTe QDs in fresh or 28 days in aged soils.

Survival is reported as the total number of earthworms per treatment (total) and as percent survival (%). Similarly, the biomass is reported as the total biomass of surviving earthworms per treatment (wet weight, g) and the percentage weight increase or decrease relative to the animals at the start of the experiment. Data presented as mean \pm SEM (*n* = 4 boxes of worms per

1190 treatment). Treatments that do not share a letter are statistically significantly different within an experiment at the specified time-

- point and the hash (#) denotes statistically significant differences between treatments on day 14 and 28 (*P* < 0.05 repeated
- 1192 measures ANOVA for biomass data or Kruskal-Wallis for survival data). Asterisk (*) denotes statistically significant differences
- 1193 between treatments in the fresh and aged soil experiments.
- 1194 Day 0 mean wet weight was 1.86 ± 0.02 g (mean \pm SEM, for a subsample of 5 of the initial earthworms, n = 52 treatments) in fresh
- 1195 soil experiment.
- 1196 Day 0 mean wet weight of 1.89 ± 0.03 g of per exposure replicate (mean ± SEM, for a subsample of 5 of the initial earthworms, *n* =
- 1197 52 treatments) in the aged soil experiment.
- 1198 ↑ Increase in wet weight relative to day 0
- 1199 \downarrow Decrease in wet weight relative to day 0
- 1200
- 1201
- 1202
- 1203
- 1204
- 1205 1206
- 1200
- 1207
- 1208

1209 **Table 3** Total number of juveniles produced and their total fresh weight in the fresh soil experiment and total number of juveniles

1210 produced in the aged soil experiment. The EC₅₀ and estimated NOEC values in each treatment in the fresh and aged soil

1211 experiment.

Soil [CdTe] nominal	Control		CdTe-bulk		CdTe-PEG		CdTe-COOH		CdTe-NH₄⁺	
mg CdTe kg⁻¹ dw	No of juveniles	Biomass (g)	No of juveniles	Biomass (g)	No of juveniles	Biomass (g)	No of juveniles	Biomass (g)	No of juveniles	Biomass (g)
Fresh soil										
50	25 ± 5ª	0.370 ± 0.06ª	22 ± 6^{a}	0.222 ± 0.09ª	18 ± 1ª	0.0267 ± 0.057ª	18.25 ± 4ª	0.220 ± 0.08ª	20 ± 3^{a}	0.307 ± 0.084ª
500			18 ± 2 ^a	0.250 ± 0.061ª	3 ± 2 ^b	0.033 ± 0.026 ^b	3 ± 1 ^b	0.005 ± 0.003 ^b	1 ± 0 ^b	0.006 ^b
2000			19 ± 3ª	0.142 ± 0.024 ^b	0ь	ND	0 ^b	ND	0 ^b	ND
EC ₅₀			NA		108 (72- 165)		65 (30- 142)		96 (20-610)	
NOEC			>2000		50 [´]		50 [′]		50	
Aged soil										
50	$41 \pm 5^{a} *$		30 ± 3^{a}		$30 \pm 2^{a} *$		32 ± 2 ^a *		$35 \pm 4^{a} *$	
500			10 ± 1 ^b *		2 ± 1°		0 ^{c *}		0°	
2000			0 ^c *		0 ^c		0°		0°	
EC ₅₀			165 (109- 243)		88 (62-127)		78 (NA)		63 (NA)	
NOEC			50		50		50		50	

Data presented as mean \pm SEM (*n* = 4). Different letters denote statistically significant differences between treatments within an experiment (*P* < 0.05, ANOVA). Asterisk (*) denote statistically significant differences between the treatment between fresh and aged soil experiment (*P* < 0.05 T-test, unpaired). The biomass is reported as the total biomass all the juveniles produced treatment 1215 (wet weight, g). EC₅₀ values for 50 % reduction in reproduction after 4 weeks of exposure in fresh and aged soils with CdTe QDs

- 1216 are presented with 95 % confidence intervals in the brackets, the concentration response curves are presented in Fig. S4. NA not
- 1217 possible to calculate EC₅₀ due to no effect; not possible to calculate confidence intervals due to high effect. Note, because of the
- difficulty in removing juveniles from amongst the plant roots that had grown in the soil during ageing, we were not confident in
- 1219 removing all of the small juveniles intact in order to report an accurate total biomass for juveniles in the aged experiment.



1221

Fig. 1 Total measured Cd (A, C) and Te (B, D) concentration in soil (mg kg⁻¹ dw) at

the beginning of the fresh (upper panels) and aged (lower panels) soil experiment.

1224 Data expressed as mean \pm SEM (n = 8). Different letters show statistically significant

differences between treatments within test concentration (P < 0.05, ANOVA). There

1226 were no statistically significant differences between treatments within a dose in the

1227 fresh compared to aged exposures (P > 0.05, *t*-test, unpaired).



Fig. 2 Relationship between the concentration of acid extractable Cd or Te (A or B) and water extractable Cd or Te (C or D) and measured total Cd or Te concentration in earthworm tissue in the fresh soil experiment. Data shown for nominal 2000 mg CdTe kg⁻¹ treatment for each replicate in the fresh soil only (n = 4). Data points represent mean values (n = 2 for soil samples, n = 4 for earthworm) for each replicate (n = 4), error bars excluded for clarity.



1236Fig. 3 Total measured Cd (A, C) and Te (B, D) concentration in earthworm tissue1237(mg kg⁻¹ dw) on day 28 of the fresh (upper panels) and aged (lower panels) soil1238experiments. Different letters denote the statistically significant differences between1239treatments within test concentration (P < 0.05, ANOVA). Asterisks denote statistically1240significant differences between treatments within a dose in the fresh compared to1241aged exposures (P < 0.05, t-test, unpaired).



Fig. 4 (A) Na⁺/K⁺-ATPase activity (sodium pump) expressed as ADP μ mol released per mg protein per hour in earthworms in the fresh soil experiment. (B) Total glutathione nmol per mg protein in the earthworms in the fresh soil experiment. Data is presented as mean ± SEM (*n* = 8). Different letters denote statistically significant differences between treatments (*P* < 0.05, ANOVA).