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# Elastic behaviour and radiation tolerance in Nb-based 211 MAX phases

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#### Abstract

MAX phase carbides are a set of materials that have attracted the research and industrial interest due to their unique combination of metallic and ceramic properties. In recent experimental studies it was determined that Nb-based MAX phases have good mechanical and thermal properties. In the present systematic density functional theory study we examine the elastic behaviour and radiation tolerance of a range of Nb<sub>2</sub>AC (A = Al, Ga, Ge, In, Sn, As, P, and S) MAX phases. It is found that the Nb-based 211 MAX phases studied here are mechanically stable and elastically anisotropic. Elastically, Nb<sub>2</sub>GeC possesses the highest level of anisotropy and Nb<sub>2</sub>InC, the lowest. The cross-slip pinning process is enhanced in Nb<sub>2</sub>GeC that is considerably reduced in Nb<sub>2</sub>InC. Nb<sub>2</sub>GeC, Nb<sub>2</sub>SnC, and Nb<sub>2</sub>SC are ductile, whereas the other Nb-based MAX phases considered here are brittle in nature. In particular, Nb<sub>2</sub>GeC is highly ductile and Nb<sub>2</sub>AlC is more brittle. Nb<sub>2</sub>PC and Nb<sub>2</sub>SnC are respectively, more stiff and flexible under tension or compression. Nb<sub>2</sub>SnC has the best thermal shock resistance among the Nb-based MAX phase carbides studied here. Regarding the radiation tolerance of these MAX phases it is anticipated that Nb<sub>2</sub>SnC will be the most resistant to radiation.

Keywords: MAX phases; First-principles calculations; Elastic properties; Defect processes

# **1. Introduction**

A large number of ternary compounds mainly carbides, some nitrides and a boride in hexagonal crystal system belong to the materials family known as "MAX phases" [1,2]. MAX phases are characterized with both metallic and ceramic properties. Chemically, they are defined as  $M_{n+1}AX_n$ , where M is an early transition metal, A is A-group element and X is carbon, nitrogen or boron and n is an integer commonly from 1–3 but with a highest value of 6 [3]. Depending on n, MAX phases are classified as 211, 312, and 413 phases for n = 1, 2, and 3, respectively. In essence, the MAX phases consist of  $M_{n+1}X_n$  ceramic layers inserted between the metallic A-layers of one-atom-thick (i.e. MAX phases are termed as nanolaminates). The layered structure of MAX phases is the key for their metallic and ceramic properties including damage tolerance, thermal and electrical conductivities, thermal shock resistance and machinability, commonly demonstrated by metals, together with light weight, high stiffness, resistance to oxidation and corrosion resistance that are characteristics of ceramics [4,5]. Furthermore, MAX phases are reported to show a good capability to annihilate radiation-induced defects at high temperatures [6–8]. MAX phase compounds are now being considered as promising candidate materials for application in accident tolerant fuel (ATF) claddings in 3rd generation (Gen-III) light-water reactors (LWRs) and future (Gen-IV) fission plants [9-11]. MAX phases, with Zr, Nb and/or Ti as an M element have recently been fabricated as potential fuel cladding coating materials either for ATF applications (LWRs), or for next-generation nuclear systems with corrosive primary coolants (e.g. Gen-IV lead- cooled fast reactors, LFRs) [12–18].

To date, about 80 MAX phases are synthesized with the majority being 211 phases, which comprise 58 members [19]. So far, different 11 transition metals are found to occupy the M-site in MAX phases. Nb has nine carbide phases in the 211 MAX family, namely Nb<sub>2</sub>AlC, Nb<sub>2</sub>PC, Nb<sub>2</sub>SC, Nb<sub>2</sub>CuC, Nb<sub>2</sub>GaC, Nb<sub>2</sub>GeC, Nb<sub>2</sub>AsC, Nb<sub>2</sub>InC, and Nb<sub>2</sub>SnC. Among those, Nb<sub>2</sub>SC, Nb<sub>2</sub>AsC, Nb<sub>2</sub>InC and Nb<sub>2</sub>SnC are superconductors [19]. Nb<sub>2</sub>AlC has attracted considerable interest as it possesses good mechanical and thermal properties, constituting it a promising material for high-temperature structural and functional applications [20–22]. Nb<sub>2</sub>PC, typically of phosphorus containing MAX phases, has larger elastic

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constants compared to most other related MAX phases [23]. Nb<sub>2</sub>GaC and Nb<sub>2</sub>InC are promising candidates for successful exfoliation into 2D MXene [24] systems. Nb<sub>2</sub>SnC is the first phase reported with an excellent electrochemical performance in Li-ion electrolyte. The Li-ions interact with Sn to form Li<sub>x</sub>Sn, which can progressively exfoliate some layers and single layers or Nb<sub>2</sub>C, remove Sn from the structure, and break the large MAX phase materials into minor and more electrochemically active units, assisting pseudocapacitive reaction and contributing to capacity improvement. Nb<sub>2</sub>SnC is highly conductive as it significantly increases its electrochemical performance. As anode in batteries it combines the advantages of layered materials and alloying elements and has a longer life cycle than most of the other Sn-containing nanomaterials [25,26]. Nb<sub>2</sub>SC is a better dielectric material [27]. Nb<sub>2</sub>CuC is an exceptional MAX phase whose A-site is also occupied by a transition metal Cu instead of A-group element [28]. Nb<sub>2</sub>AsC has the lowest superconducting transition temperature amongst all MAX phase superconductors [19].

Therefore, Nb-containing 211 MAX phase carbides have significant diversity in their properties. Understanding the elastic behaviour of materials is essential to decide on where they are most suitable, whereas the defect processes in crystals provides the information regarding the radiation tolerance of materials. The elastic properties of these compounds have been studied individually in previous studies with different methodologies and codes [19,27,29–37]. The defect processes in these crystals have never been investigated. In this study, we aim to investigate the elastic behaviour and defect processes in the Nb-based 211 MAX phase carbides excepting Nb<sub>2</sub>CuC as it has an additional transition metal Cu at A-site instead of A-group element that might be a barrier in understanding the general trend in typical Nb-based 211 MAX phases.

#### 2. Method of calculations

The present calculations were performed using density functional theory (DFT) as implemented in the CASTEP code [38–40]. Exchange-correlation interactions were described with the generalized gradient approximation (GGA) corrected by Perdew, Burke and Ernzerhof (PBE) [41,42] scheme. The interaction between electrons and ion cores are treated with ultrasoft pseudopotential [43]. To expand the eigenfunctions of the valence electrons and the nearly free electrons, a plane wave basis set with cutoff energy of 550 eV is employed. To optimize the geometry via minimizing the total energy and internal forces BFGS minimization scheme is used because of its advantage of ability to perform cell optimization, including optimization at fixed external stress [44]. Brillouin zone integrations are performed with a  $\Gamma$ -centered k-point mesh of  $10 \times 10 \times 2$  grid in the Monkhorst-Pack (MP) scheme [45]. All the crystal structures are relaxed until the residual forces on the atoms have declined to less than 0.01 eV/Å, maximum stress less than 0.02 GPa, maximum displacement less than  $5 \times 10^{-4}$ Å, and energy per atom less than  $5 \times 10^{-6}$  eV.

The CASTEP code is embodied with the finite-strain theory to calculate the elastic properties of materials [46]. In this theory, a specified set of identical strains (deformations) is applied to the conventional unit cell, allowing the relaxation of the atomic degrees of freedom. Then resultant external stresses are calculated. The stress tensor has six stress components  $\sigma_{ij}$  for each strain  $\delta_j$  applied to the conventional unit cell. Then elastic constants  $C_{ij}$  are calculated by solving a set of linear equations,  $\sigma_{ij} = C_{ij}\delta_j$ . This method is employed in the present study and it has already successfully predicted the elastic properties of a wide range of crystals [47–60].

To better understand the possible mechanisms of radiation tolerance in the Nb-based 211 MAX carbides, intrinsic defects (vacancies, antisite defects and interstitials) calculations are performed with DFT to determine the energetically favourable defect processes in Nb<sub>2</sub>AC (A = Al, Ga, Ge, In, Sn, As, P, and S). Vacancies and antisites are assumed as isolated defects on each symmetrically distinct lattice site. All possible interstitial sites are considered for each element within the 211 MAX phase structure. A  $3\times3\times1$  supercell of 72 lattice sites (36 Nb, 18A, and 18 C) are used for defects calculations with a cutoff energy of 450 eV and k-point mesh of  $3\times3\times2$  grid in the MP scheme. These calculations were performed under constant pressure conditions.

#### 3. Results and discussions

#### 3.1. Optimized structure

The optimized structure of Nb<sub>2</sub>AlC as a structural model of 211 MAX phases is shown in Fig. 1, indicating that MX slab is separated by one A-atomic layer. The calculated lattice parameters are listed in Table 1 along with reported measured values for comparison. Fig. 2 shows that the calculated lattice constants are consistent with the experimental ones. The lattice parameters of Nb<sub>2</sub>SnC exhibit relatively large deviation of 1.27% for *a*. In the most cases, the GGA trend ( $a_{calc}$ ,  $c_{calc.} > a_{expt}$ ,  $c_{expt.}$ ) is established. Violation is observed for the Ge-based phase, which is also observed for *c* in a previous study of Nb<sub>2</sub>GeC [61]. It is worth mentioning that Nb<sub>2</sub>GeC is synthesized only in the thin film form and during its fabrication some NbC<sub>x</sub> phases are obtained. The phase impurity me be the possible reason of violation of GGA trend for Nb<sub>2</sub>GeC. When the A-element moves from left to right across the periodic table in a period the lattice constant *a* increases, whereas the lattice constant *c* decreases followed by an increase for S.



Fig. 1. The optimized structure of Nb<sub>2</sub>AlC as a structural model of 211 MAX phases.



Fig. 2. Calculated versus experimental cell constants of Nb<sub>2</sub>AC MAX phases.

Phase	<i>a</i> (Å)	<i>c</i> (Å)	c/a	$V(Å^3)$	Remarks
Nb <sub>2</sub> GaC	3.143	13.640	4.340	116.66	This calc.
	3.13	13.56	4.332	115.05	Expt. [1]
Nb <sub>2</sub> PC	3.292	11.578	3.517	108.68	This calc.
	3.28	11.5	3.506	107.15	Expt. [1]
Nb <sub>2</sub> AlC	3.120	13.926	4.463	117.41	This calc.
	3.106	13.888	4.471	116.03	Expt. [20]
Nb <sub>2</sub> GeC	3.228	12.759	3.953	115.10	This calc.
	3.24	12.82	3.957	116.45	Expt. [61]
Nb <sub>2</sub> SC	3.295	11.690	3.548	109.90	This calc.
	3.294	11.553	3.507	108.56	Expt. [62]
Nb <sub>2</sub> AsC	3.325	11.977	3.602	114.67	This calc.
	3.317	11.90	3.588	113.39	Expt. [63]
Nb <sub>2</sub> InC	3.186	14.528	4.560	127.72	This calc.
	3.172	14.37	4.530	125.21	Expt. [64]
Nb <sub>2</sub> SnC	3.261	13.874	4.254	127.77	This calc.
	3.220	13.707	4.257	123.08	Expt. [65]

Table 1. Calculated and experimental lattice constants of Nb-based 211 MAX carbides.

#### 3.2. Elastic behaviour

Both elastic constants and moduli describe the elastic (i.e., mechanical) behaviour of materials. MAX phases have six different elastic constants  $C_{11}$ ,  $C_{33}$ ,  $C_{44}$ ,  $C_{66}$ ,  $C_{12}$ , and  $C_{13}$  as they crystallize in hexagonal space group  $P6_3/mmc$ . All elastic constants are independent excepting  $C_{66}$  since  $C_{66} = (C_{11} - C_{12})/2$ . Elastic constants ensure the mechanical stability of hexagonal crystals obeying the following conditions [66]:

$$C_{11} > 0, C_{33} > 0, C_{44} > 0, (C_{11} + C_{12})C_{33} > 2(C_{13})^2, \text{ and } (C_{11} - C_{12}) > 0$$
 (1)

The calculated elastic constants are listed in Table 2. These values are consistent with the ones reported in previous studies [19,27,29–37]. The present values meet the above conditions to be stable mechanically for the compounds studied here. The elastic constant  $C_{11}$  measures the elastic stiffness of the materials regarding (100)(100) strain. In this view, Nb<sub>2</sub>PC is the stiffest and Nb<sub>2</sub>SnC is the softest material in this group. Regarding (001)(001) strain, the elastic constant  $C_{33}$  also predict the same results, that is, Nb<sub>2</sub>PC is the stiffest and Nb<sub>2</sub>SnC is the softest one. The elastic constant  $C_{12}$  measures the resistance against deformation in the (110) plane along (100) direction. Therefore, Nb<sub>2</sub>GeC is most rigid and Nb<sub>2</sub>GaC is easily deformable materials among the compounds studied here. The low value of  $C_{12}$  and  $C_{13}$  imply that Nb<sub>2</sub>GaC, Nb<sub>2</sub>InC, Nb<sub>2</sub>AlC, and Nb<sub>2</sub>SnC are easier to shear along the b- and c-axis compared to other MAX phases under investigation when a stress is applied along the a-axis.

Elastic anisotropy is an inherent characteristic of solids. Primarily, it can be ensured from the elastic constants  $C_{11}$  and  $C_{33}$ . Either  $C_{11} > C_{33}$  or  $C_{33} > C_{11}$  is the indication of elastic anisotropy of crystals. For hexagonal crystals like MAX phases, a common anisotropy factor is  $A = 4C_{44}/(C_{11}+C_{33}-2C_{12})$ , which is calculated to give insight on the elastic shear anisotropy of Nb-based 211 MAX carbides and are listed in Table 2. Elastically, anisotropic crystals have A-value either greater or less than unity. Deviation from unity  $\Delta A$  quantifies the degree of elastic anisotropy, which is shown in Fig. 3. Thus, Nb<sub>2</sub>GeC is elastically highly anisotropic and Nb<sub>2</sub>InC possesses low level of elastic anisotropy. Additionally A-value provides information regarding screw dislocation and cross-slip pining process in crystals. The larger A-value leads to the driving force (tangential force) acting on screw dislocations to stimulate the cross-slip pinning process [67]. The comparatively high A-value is inclined to elastically enhance the cross-slip pinning process in Nb<sub>2</sub>GeC that is considerably reduced in Nb<sub>2</sub>InC.

There is another anisotropy factor for hexagonal crystals that can be named compressibility anisotropy factor and is expressed as  $k_c/k_a = (C_{11}+C_{12}-2C_{13})/(C_{33}-C_{13})$ . This describes linear compressibility of c-axis relative to the a-axis. The calculated value listed in Table 2 for compounds

studied here implies that the Nb<sub>2</sub>AlC is more compressible along c-axis than a-axis. For other phases, the compressibility is more profound along a-axis than c-axis, with most for Nb<sub>2</sub>PC. Therefore, the anisotropy in compression of Nb<sub>2</sub>AlC and Nb<sub>2</sub>PC is large along c- and a-axis, respectively.  $\Delta(k_c/k_a)$  defines the anisotropic level of the phases given in Fig. 3, which also exhibits the trend of elastic anisotropy level in Nb<sub>2</sub>AC MAX phases with the A-elements. The trends of  $\Delta A$  and  $A^U$  ( $A^U$  is discussed latter) are almost similar.  $\Delta(k_c/k_a)$  shows an increasing trend from Al to P when one goes through one group elements to the next group elements and then shows a decreasing trend.

Phase	$C_{11}$	$C_{33}$	$C_{44}$	$C_{66}$	$C_{12}$	$C_{13}$	Α	$k_{\rm c}/k_{\rm a}$
Nb <sub>2</sub> AlC	332.88	283.66	138.41	124.43	84.02	117.30	1.45	1.10
Nb <sub>2</sub> GaC	322.88	280.85	126.36	122.73	77.42	129.54	1.47	0.93
Nb <sub>2</sub> InC	280.36	266.05	103.67	100.35	79.63	112.73	1.29	0.88
Nb <sub>2</sub> GeC	283.60	275.05	151.66	73.58	136.44	160.77	2.56	0.86
Nb <sub>2</sub> SnC	263.65	260.44	109.52	87.53	88.59	121.38	1.56	0.79
Nb <sub>2</sub> PC	372.81	406.77	193.48	129.40	114.01	172.21	1.78	0.61
Nb <sub>2</sub> AsC	343.01	342.72	173.02	120.69	101.62	166.76	1.95	0.63
Nb <sub>2</sub> SC	323.69	333.35	125.69	103.84	116.01	145.76	1.38	0.79

Table 2. Elastic constants  $C_{ij}$  (in GPa) and anisotropy factors A and  $k_c/k_a$  for Nb-based 211 MAX carbides.



Fig. 3. Elastic anisotropic level in Nb<sub>2</sub>AC MAX phases.

Elastic constants  $C_{ij}$  can be used to calculate the bulk modulus B and shear modulus G. According to Voigt, Reuss, and Hill (VRH) approximations [68-70] they are identified by using V, R, and H as subscripts to B and G and their calculated values are listed in Table 3. It is evident that Nb<sub>2</sub>PC and  $Nb_2InC$  respectively, have highest and lowest value of B, indicating that they are respectively, most incompressible and compressible under pressure. They are also strongest and weakest materials in view of chemical bonding. The strength of chemical bonding should follow the order:  $Nb_2PC > Nb_2AsC > Nb_2SC$  $> Nb_2GeC > Nb_2GaC > Nb_2AlC > Nb_2SnC > Nb_2InC$ . For comparison, we have experimental value of B 165 and 208 GPa for Nb<sub>2</sub>AlC [71,72], 180±5 GPa for Nb<sub>2</sub>SnC [73], 224±2 GPa for Nb<sub>2</sub>AsC [74]. These values are consistent with the present values. Shear modulus for the compounds studied here is smaller in magnitude than the respective bulk modulus, indicating that the mechanical stability of these solids is controlled by shear modulus.  $Nb_2PC$  and  $Nb_2SnC$  respectively, have highest and lowest value of G, indicating that the shape change is easier in Nb<sub>2</sub>SnC and not in Nb<sub>2</sub>PC compared with other Nb<sub>2</sub>AC phases. Shear modulus is highly correlated with the materials' hardness. Accordingly, Nb<sub>2</sub>PC and Nb<sub>2</sub>SnC respectively, are the hardest and softest materials among Nb-based 211 MAX phase carbides.  $Nb_2GeC > Nb_2SnC.$ 

Bulk to shear modulus ratio  $B_{\rm H}/G_{\rm H}$  known as Pugh's ratio plays a vital role to characterize an important mechanical phenomena in materials [75]. To judge the mechanical failure mode of solids, Pugh's ratio with a threshold value of 1.75 serves as an indicator. Brittle failure occurs for a material whose Pugh's ratio is less than 1.75 and ductile failure happens for a material having a value greater than 1.75. Accordingly, Nb<sub>2</sub>GeC, Nb<sub>2</sub>SnC, and Nb<sub>2</sub>SC are ductile materials and the remaining majorities are brittle materials. Indeed, most of the MAX phases are brittle in nature [3,32,49–56]. Nb<sub>2</sub>GeC is highly ductile and Nb<sub>2</sub>AlC is more brittle in this group. In ductile materials, cracks progress sluggishly when plastic deformation occurs, while in brittle materials, cracks extend rapidly when stress is applied.

Young's modulus for polycrystalline aggregates are calculated from *B* and *G*, E = 9BG/(3B+G) and listed in Table 4. *E* is a measure of the ability of a material to resist changes in length when under lengthwise tension or compression. It is evident that Nb<sub>2</sub>PC and Nb<sub>2</sub>SnC, respectively have highest and lowest value of *E*, signifying that they are respectively, most stiff and flexible, under tension or compression compared with other Nb<sub>2</sub>AC phases. Young's modulus is also a measure of materials' hardness [19]. Based on *E*, if we rank the compounds studied here for hardness, we have the same order based on *G*, excepting an interchange of positions between Nb<sub>2</sub>InC and Nb<sub>2</sub>GeC. Young's modulus also has a good relation with the critical thermal shock resistance,  $R \propto 1/E$  [55], implying that the lower the *E* better the *R*. Thus Nb<sub>2</sub>SnC has better thermal shock resistance among all the Nb-based 211 MAX phase carbides studied here. Better thermal shock resistance is a precondition for a solid to be used as a thermal barrier coating (TBC) material. Fig. 4 shows a comparative picture of elastic moduli to understand at a glance.



Fig. 4. Elastic moduli of Nb<sub>2</sub>AC MAX phases.

Poisson's ratio for polycrystalline aggregates are calculated from *B* and *G*,  $\sigma = (3B-2G)/(6B+2G)$ , and are listed in Table 4 and shown in Fig. 5 along with  $B_H/G_H$ . Poisson's ratio is a decisive factor for many characterizing issues in solids. Poisson's ratio also justifies the failure modes of solids with a critical value of 0.26 [3]. Ductile and brittle materials have respectively, larger and smaller values than 0.26. This factor also finds the ductile/brittle materials as Pugh's ratio does. In this scale, Nb<sub>2</sub>GeC is again highly ductile and Nb<sub>2</sub>AlC is the most brittle in the groups. The A-group elements Al, Ga, and In come from group 13. The 211 MAX carbides with these elements show a decreasing trend of brittleness when we moves from up to down according to both Pugh's and Poisson's ratios. The In-based Nb<sub>2</sub>InC reaches near the brittle-ductile borderline. Ge is the first element of the next group 14 and the Ge-based Nb<sub>2</sub>GeC crosses the borderline and become a highly ductile material. Sn is the second element of the same group 14 and Sn containing Nb<sub>2</sub>SnC shows a decrease in its ductility and reaches near the brittle-ductile borderline. P is the first element of the group 15 and Nb<sub>2</sub>AsC show a decreasing trend of brittleness as of the element of group 13. S is the first element of the next group 16 and Nb<sub>2</sub>SC crosses the borderline and become a ductile material. It exhibits the same trend of Nb<sub>2</sub>GeC that contains the first element of group 14.

A crystalline solid is always stable under either central force or non-central force. A material will be stabilized by central force if its Poisson's ratio lies between 0.25 and 0.50, otherwise it will be stabilized by non-central force [3]. Accordingly, Nb<sub>2</sub>InC, Nb<sub>2</sub>GeC, Nb<sub>2</sub>SnC, and Nb<sub>2</sub>SC are stabilized with central force and the remaining phases with non-central force. It is observed that the ductile materials are mainly central force solids; while the brittle materials are non-central force ones. Poisson's ratio can predict atomic bonding nature in crystals by identifying the purely covalent crystals with a value of 0.10 and totally metallic crystals with a value of 0.33 [19]. It is expected that all the Nb-based 211 MAX carbides are characterized by partially metallic and covalent bonding as their Poisson's ratio lies between 0.10 and 0.33. However, the brittle phases are more covalent than the ductile ones. For comparison, we have experimental value of 0.21 for Nb<sub>2</sub>AlC [71], which is almost similar to the present value of 0.222.

Phase	$B_{ m V}$	$B_{ m R}$	$B_{ m H}$	$G_{ m V}$	$G_{ m R}$	$G_{ m H}$	$B_{ m H}/G_{ m H}$
Nb <sub>2</sub> AlC	176.29	176.18	176.24	122.30	118.54	120.42	1.46
Nb <sub>2</sub> GaC	177.74	177.69	177.71	114.43	109.54	111.98	1.59
Nb <sub>2</sub> InC	159.65	159.48	159.57	96.31	94.63	95.47	1.67
Nb <sub>2</sub> GeC	195.36	195.19	195.27	101.00	85.14	93.07	2.10
Nb <sub>2</sub> SnC	161.16	160.65	160.91	91.74	88.28	90.01	1.79
Nb <sub>2</sub> PC	229.92	226.83	228.37	149.54	139.91	144.72	1.58
Nb <sub>2</sub> AsC	211.00	208.98	209.99	132.92	121.32	127.12	1.65
Nb <sub>2</sub> SC	199.53	198.87	199.20	109.26	107.13	108.19	1.84

Table 3. Bulk and shear moduli (in GPa) and Pugh's ratio of Nb-based 211 MAX carbides.

Table 4. Young's modulus (in GPa), Poisson's ratio and universal anisotropy index of Nb-based 211 MAX carbides.

Phase	$E_{ m V}$	$E_{ m R}$	$E_{ m H}$	$\sigma_{ m V}$	$\sigma_{ m R}$	$\sigma_{ m H}$	$A^{\mathrm{U}}$
Nb <sub>2</sub> AlC	298.00	290.47	294.25	0.218	0.225	0.222	0.159
Nb <sub>2</sub> GaC	282.63	272.59	277.63	0.235	0.244	0.240	0.224
Nb <sub>2</sub> InC	240.57	237.01	238.79	0.249	0.252	0.251	0.090
Nb <sub>2</sub> GeC	258.45	222.99	240.93	0.280	0.310	0.294	0.932
Nb <sub>2</sub> SnC	231.33	223.84	227.59	0.261	0.268	0.264	0.199
Nb <sub>2</sub> PC	368.68	348.15	358.45	0.233	0.244	0.238	0.358
Nb <sub>2</sub> AsC	329.56	304.95	317.33	0.240	0.257	0.248	0.488
Nb <sub>2</sub> SC	277.18	272.46	274.82	0.268	0.272	0.270	0.103



Fig. 5. Pugh's and Poisson's ratio of Nb<sub>2</sub>AC MAX phases.

Depending on Voigt and Reuss limit of bulk and shear moduli, an anisotropy factor is defined as  $A^{U} = 5(G_{V}/G_{R}) + (B_{V}/B_{R}) - 6$ , which is known as universal anisotropy index as it is applicable for all kind of crystals [76] with different symmetries. The calculated value of  $A^{U}$  is listed in Table 4 and shown in Fig.

3, indicating that anisotropy level is high in Nb<sub>2</sub>GeC and low in Nb<sub>2</sub>InC. This prediction is same as predicted from shear anisotropy factor A.  $A^{U}$  directly measures the anisotropy level in crystals.

## 3.3 Defect processes

The reason for examining the point defect processes of materials is that they can determine their macroscopic properties for example their radiation tolerance. As it has been discussed in previous work this is particularly important as MAX phases are considered for nuclear applications [77,78]. From a physical viewpoint, the ability of a material to resist radiation will be dependent upon the ability of the material to form and accommodate point defects. In that respect, a high concentration of defects can result to the destabilization of the material [79-81].

To calculate the defect processes (refer to Table 5) we have considered all possible point defects and importantly all the possible interstitial defects (The lowest energy interstitial sites for the Nb<sub>2</sub>AC MAX phases are given in Table 6). In Table 5, we present all the defect processes considered here and the corresponding defect energies in Kröger–Vink notation [82]. In this notation, Nb<sub>i</sub> will denote a Nb interstitial defect, Nb<sub>A</sub> an antisite defect (i.e. a Nb atom in an A-site) and V<sub>Nb</sub> a vacant Nb site.

Table 5. The defect process reaction (Frenkel 1-3; antisite 4-6) energies for the Nb<sub>2</sub>AC MAX phase.

Defect Decetions				Defect en	ergy (eV)			
Defect Keactions	Nb <sub>2</sub> AlC	Nb <sub>2</sub> AsC	Nb <sub>2</sub> GaC	Nb <sub>2</sub> GeC	Nb <sub>2</sub> InC	Nb <sub>2</sub> PC	Nb <sub>2</sub> SC	Nb <sub>2</sub> SnC
(1) $Nb_{Nb} \rightarrow V_{Nb} + Nb_i$	7.8909	8.4693	5.4662	5.9219	7.7613	9.2965	6.1069	8.1782
(2) $A_A \rightarrow V_A + A_i$	4.5934	7.7893	3.7765	4.5134	6.6649	7.5774	7.3398	8.3452
(3) $C_C \rightarrow V_C + C_i$	3.0938	4.4458	3.9199	3.1356	4.9546	4.2133	3.7014	4.8415
(4) Nb <sub>Nb</sub> +A <sub>A</sub> $\rightarrow$ Nb <sub>A</sub> +A <sub>Nb</sub>	2.9800	8.3839	4.2095	5.8330	3.5978	10.0846	9.7703	5.1802
(5) $Nb_{Nb}+C_C \rightarrow Nb_C+C_{Nb}$	15.8236	12.4426	15.2602	13.5318	13.9492	13.0953	10.2005	13.4818
(6) $A_A + C_C \rightarrow A_C + C_A$	9.5503	4.0405	8.3706	5.5303	12.0852	2.8984	2.7874	10.1008
(7) $Nb_i+V_A \rightarrow Nb_A$	-5.7821	-3.7175	-3.4084	-2.9738	-6.1817	-2.2933	-0.5443	-6.4284
(8) $C_i + V_A \rightarrow C_A$	0.8008	-2.6554	-0.0104	-0.3059	0.6196	-3.2429	-3.1738	0.1920
(9) $Nb_i + A_A \rightarrow Nb_A + A_i$	-1.1887	4.0718	0.3681	1.5396	0.4831	5.2841	6.7956	1.9167
$(10)\;A_i + V_{Nb} \to A_{Nb}$	-3.7222	-4.1572	-1.6248	-1.6284	-4.6466	-4.4961	-3.1321	-4.9147
$(11)\ C_i + V_{Nb} \to C_{Nb}$	1.8521	-0.4272	1.0050	1.3226	0.1184	-1.0242	-0.3285	0.9071
(12) $Nb_i + V_C \rightarrow Nb_C$	2.9868	-0.0453	4.8691	3.1518	1.1148	0.6096	0.7208	-0.4449
$(13) \ A_i + V_C \rightarrow A_C$	1.0624	-5.5393	0.6846	-1.8128	-0.1539	-5.6494	-5.0800	-3.2778
$(14) \ Nb_i + C_C \rightarrow Nb_C + C_i$	6.0806	4.4005	8.7890	6.2874	6.0695	4.8229	4.4221	4.3965
$(15) \ A_i + Nb_{Nb} \rightarrow A_{Nb} + Nb_i$	4.1687	4.3122	3.8414	4.2934	3.1147	4.8005	2.9748	3.2635
$(16) \ A_i + C_C \rightarrow A_C + C_i$	4.1562	-1.0934	4.6045	1.3228	4.8008	-1.4361	-1.3787	1.5637
(17) $C_i + Nb_{Nb} \rightarrow C_{Nb} + Nb_i$	9.7430	8.0421	6.4712	7.2445	7.8797	8.2724	5.7784	9.0853
$(18) \ C_i + A_A \rightarrow C_A + A_i$	5.3941	5.1339	3.7661	4.2075	7.2844	4.3345	4.1660	8.5371

The most important defect processes indicating radiation tolerance are the Frenkel reactions (Table 5, relations 1-3) and the antisite reactions (Table 5, relations 4-6). Considering the materials with higher Frenkel and antisite energies will be more radiation persistent and in that respect Nb<sub>2</sub>SnC is the better material as it has higher defect energies than the other Nb-MAX phases considered (Table 5, relations 1-6). Relations 7-12 (refer to Table 5) reveal whether interstitial defects and vacancies are energetically

favourable to recombine and effectively to form antisite defects or if they rather remain as isolated defects. Finally, relations 13-18 (refer to Table 5) examine whether in a radiation environment where there is an over stoichiometry of self-interstitials there is the possibility to form antisite defects.

Phase	Atom	Energeticall	y preferable inter	stitials sites
Nb <sub>2</sub> AlC	Nb	0.0007	0.9993	0.7226
	Al	0.9979	0.0021	0.7509
	С	0.3335	0.6665	0.7500
Nb <sub>2</sub> AsC	Nb	0.9961	0.9984	0.7065
	As	0.9942	0.9974	0.7490
	С	0.3330	0.6646	0.7497
Nb <sub>2</sub> GaC	Nb	0.9015	0.2232	0.7511
	Ga	0.3829	0.2662	0.7497
	С	0.3331	0.6642	0.7499
Nb <sub>2</sub> GeC	Nb	0.3312	0.2400	0.7497
	Ge	0.9986	0.0010	0.7494
	С	0.3330	0.6648	0.7498
Nb <sub>2</sub> InC	Nb	0.3198	0.2298	0.7498
	In	0.3360	0.2453	0.7498
	С	0.9984	0.0012	0.6550
Nb <sub>2</sub> PC	Nb	0.9995	0.9998	0.7177
	Р	0.9999	0.9998	0.7490
	С	0.3329	0.6644	0.7497
Nb <sub>2</sub> SC	Nb	0.0001	0.0014	0.7513
	S	0.3269	0.2822	0.7503
	С	0.8618	0.9313	0.9472
Nb <sub>2</sub> SnC	Nb	0.3226	0.2176	0.7498
	Sn	0.3809	0.2889	0.7495
	С	0.3326	0.6638	0.7497

Table 6. The lowest energy interstitial sites for the Nb<sub>2</sub>AC MAX phases.

# 4. Conclusions

In the present study, DFT calculations were employed to investigate the elastic properties and defect processes of Nb-based 211 MAX phases. The lattice constant *a* exhibits the tendency to increase when the *A*-group element moves from left to right through the periodic table within a period and the lattice constant *c* decreases followed by an increase for S. All the compounds studied here are mechanically stable and elastically anisotropic. Nb<sub>2</sub>GeC is expected to be elastically high and Nb<sub>2</sub>InC to be low anisotropic. The cross-slip pinning process is enhanced in Nb<sub>2</sub>GeC and is significantly reduced in Nb<sub>2</sub>InC. Among the Nb-based 211 MAX phases, Nb<sub>2</sub>GeC, Nb<sub>2</sub>SnC, and Nb<sub>2</sub>SC are expected to be ductile i.e., damage tolerant and the rest to be brittle. Nb<sub>2</sub>GeC is predicted to be highly ductile and Nb<sub>2</sub>AlC to be more brittle. Nb<sub>2</sub>SnC has the best thermal shock resistance among the Nb-based 211 MAX phase carbides studied here. Finally, it is anticipated that Nb<sub>2</sub>SnC has also the highest radiation resistance.

# **CRediT** authorship contribution statement

**M.A. Hadi**: Conceptualization, Investigation, Data curation, Methodology, Formal analysis, Writing - original draft. **S.-R.G. Christopoulos:** Investigation, Data curation, Methodology. **A. Chroneos**: Formal analysis, Writing, Review & editing. **S.H. Naqib**: Formal analysis, Review & editing. **A.K.M.A. Islam**: Formal analysis, Review & editing.

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