## Research paper

# Novel quinazolinone inhibitors of the Pseudomonas aeruginosa quorum sensing transcriptional regulator PqsR 

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#### Abstract

Rising numbers of cases of multidrug- and extensively drug-resistant Pseudomonas aeruginosa over recent years have created an urgent need for novel therapeutic approaches to cure potentially fatal infections. One such approach is virulence attenuation where anti-virulence compounds, designed to reduce pathogenicity without affording bactericidal effects, are employed to treat infections. P. aeruginosa uses the pqs quorum sensing (QS) system, to coordinate the expression of a large number of virulence determinants as well as bacterial-host interactions and hence represents an excellent antivirulence target.

We report the synthesis and identification of a new series of thiazole-containing quinazolinones capable of inhibiting PqsR, the transcriptional regulator of the pqs QS system. The compounds demonstrated high potency ( $\mathrm{IC}_{50}<300 \mathrm{nM}$ ) in a whole-cell assay, using a mCTX: $\mathrm{P}_{p q s A}$-lux-based bioreporter for the P. aeruginosa PAO1-L and PA14 strains. Structural evaluation defined the binding modes of four analogues in the ligand-binding domain of PqsR through X-ray crystallography. Further work showed the ability of 6-chloro-3((2-pentylthiazol-4-yl)methyl)quinazolin-4(3H)-one (18) and 6-chloro-3((2-hexylthiazol-4-yl)methyl)quinazolin-4(3H)-one (19) to attenuate production of the PqsR-regulated virulence factor pyocyanin. Compounds 18 and 19 showed a low cytotoxic profile in the A549 human epithelial lung cell line making them suitable candidates for further pre-clinical evaluation. © 2020 The Authors. Published by Elsevier Masson SAS. This is an open access article under the CC BY-


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## 1. Introduction

Pseudomonas aeruginosa (PA), a Gram-negative pathogenic bacterium found widely in nature, is a common cause of infection in immunocompromised patients. Chronic and recurring infections are widespread, and PA infection is frequently associated with respiratory failure, reduced pulmonary function and mortality within cystic fibrosis patients [1-3]. A rise in multidrug resistant cases of PA has raised this organism to a priority class pathogen of critical importance by the World Health Organisation [4].

In order to combat the threat of antimicrobial resistance, novel

[^0]strategies are required to provide long-lasting solutions, which drastically reduce the rate of emergence of resistance due to the high selective pressure posed by current antibiotic treatments. One such approach is the use of alternative treatments which can attenuate virulence within bacterial populations without directly killing the infectious organisms and/or sensitise these populations to the action of existing antibiotics [5,6]. The net result is a nonpathogenic population which can be cleared by the immune system or which may become sensitive to existing antimicrobials [6-9].

Extensive work associated with this approach has investigated inhibition of quorum sensing systems within bacterial populations. Quorum sensing (QS) is a cell-to-cell communication strategy used by microbes to coordinate the production of numerous traits including virulence factors in a population-dependent manner. QS relies on the production and sensing of small diffusible signal
molecules known as autoinducers (AIs) or QS signal molecules (QSSMs) [10-12]. Although most commonly observed as an intraspecies event, QS is also seen at the interspecies and inter-kingdom level through bacteria-bacteria communication systems as well as in bacteria-fungi mixed populations [13-15].

During infection QS in PA controls the production of virulence traits such as the phenazine, pyocyanin and the siderophore pyoverdine, which are capable of cytotoxic effects against mammalian cells and iron scavenging respectively, as well as those responsible for bacterial motility and biofilm formation [16-22].

PA utilises three closely interlinked QS systems to fully elicit its pathogenicity; the las and rhl systems which operate via $N$-acylhomoserine lactones as their $\operatorname{QSSMs}(\mathbf{1 , 2})$, whereas the pqs system uses alkylquinolones (AQ), namely 2-heptylquinolin- $4(1 \mathrm{H})$-one ( $\mathrm{HHQ}, 3$ ) and the Pseudomonas Quinolone Signal (2-heptyl-3-hydroxy-4(1H)quinolone, $\mathrm{PQS}, 4$ ) as their cognate signals (Fig. 1) [7,11,23-25].

The pqs QS system of $P A$ requires the $p q s A B C D E$ operon for the biosynthesis of HHQ 3, the precursor of PQS 4, though the pqsE gene product is also known to play both a biosynthetic and a key regulatory role in the regulation of multiple virulence factors [26-31]. HHQ is converted to PQS via the monooxygenase PqsH. The expression of the pqsABCDE operon is controlled by the LysR-type transcriptional regulator PqsR upon binding its cognate ligands, PQS and $\mathrm{HHQ}[32,33]$.

Inhibition of the pqs system can result in a reduction in the production of pyocyanin and alkylquinolones as well as many other virulence traits [32,34-37]. As such, PqsR has been validated as a target for the inhibition of the pqs system and hence virulence using both in vitro assays and in vivo in mouse infection models [37].

Herein we report the discovery of a new series of compounds with high potency against PqsR in two different PA strains (PAO1-L and PA14) that represent the two major genomic groups. Binding of compounds 6, 12, 18 and 19 to this regulatory protein is shown through co-crystallisation into the ligand-binding domain (LBD) of PqsR $[32,36]$. The lead compounds $(\mathbf{1 8}, \mathbf{1 9})$ were shown to attenuate pyocyanin production and demonstrated very low cytotoxicity against mammalian cells, supporting their suitability for further studies.

## 2. Results and discussion

### 2.1. SAR-based design and synthesis of PqsR inhibitors

Previous research has shown that quinazolinone scaffolds can provide the basis for PqsR antagonists, in part due to the similarity this core shares with the endogenous ligands PQS and HHQ [32]. Therefore, it was unsurprising that an in silico screen of the

University of Nottingham Managed Chemical Compound Collection (MCCC, a collection of 85,000 diverse compounds) gave several hit compounds bearing this structural motif. One such compound, 5 (Fig. 2), containing a thiazole group proved interesting for further structure activity relationship (SAR) exploration.

Initial testing of 5 showed an activity for PAO1-L of $13.2 \pm 2.73 \mu \mathrm{M}$ in a mCTX- $\mathrm{P}_{p q s A}$-lux luminescence based bioreporter assay. Further optimisation yielded 6, with an improved activity of $1.0 \pm 0.42 \mu \mathrm{M}$. Furthermore, homologous oxadiazole 7 was found to be inactive at $10 \mu \mathrm{M}$, providing evidence for the importance of the thiazole ring for inhibiting PqsR. As such, a SAR study was conducted around 6, primarily focussing on 2-substitution of the thiazole.

We envisaged that variation to the isopropyl group of compound 6 was the most promising area to explore with regards to improving potency. A range of alternate functionalities varying from alkyl chains to substituted amines and small aromatic groups provided a diverse set of compounds to explore the space available within the LBD of the PqsR protein where these ligands were believed to bind (Fig. 3).

A four-step synthetic procedure (Scheme 1) provided a robust route to the desired 2,4-disubstituted thiazoles bearing alkyl, amino and aryl functionalities. Initially, 2-amino-5-chlorobenzoic acid was condensed in formamide to give the corresponding 6-chloroquinazolin-4(3H)-one 36. Reaction with chloroacetone gave the intermediate 37 , which was brominated through refluxing in acetic acid with bromine to afford the $\alpha$-bromoketone $\mathbf{3 8}$. Thiazole formation in ethanol with a range of thioamides gave the 2,4disubstituted thiazoles 6, 8-26.

In addition to this series of 2,4-substituted thiazoles (6,8-26), a 2,5-substituted thiazole $\mathbf{3 5}$ analogous to $\mathbf{6}$ was synthesised to observe whether altering the orientation of the thiazole ring could improve efficacy, as was purported through a ligand docking screen in silico. Furthermore, in a bid to decrease lipophilicity, a range of 2-amino-5-substituted thiazoles were synthesised in a two-step route from 6 -chloroquinazolin-4(3H)-one (36).

Synthesis of the 5 -substituted thiazole 35 required an alternate four-step synthetic route (Scheme 2). Initially, bromomalonaldehyde was condensed with 2-methylpropanethioamide, to give 2-isopropylthiazole-5-carbaldehyde, which was immediately treated with hydroxylamine hydrochloride affording 2-isopropylthiazole-5-carbaldehyde oxime (39). Reduction of the oxime using zinc powder and hydrochloric acid gave the primary amine (40).

Separately, 2-amino-5-chlorobenzoic acid was treated with DMFDMA to give methyl 5-chloro-2-(((dimethylamino)methylene) amino)benzoate (41) [38]. Cyclisation under acidic conditions with

$N$-butanoyl-L-homoserine lactone 1


2-heptylquinolin-4(1H)-one 3 (HHQ)

$N$-(3-oxododecanoyl)-L-homoserine lactone 2


2-heptyl-3-hydroxyquinolin-4(1H)-one 4 (PQS)

Fig. 1. Known QSSMs in P. aeruginosa include $N$-acylhomoserine lactones which regulate $\mathbf{1}$ the $r h l$ and $\mathbf{2}$ las systems, and $\mathbf{3 , 4} \mathbf{4 H Q}$ and PQS which regulate the pqs system.


5


6


7

Fig. 2. Hit compound 5 found through in silico and subsequent in vitro testing of the MCCC compound library, was optimised to yield 6. Replacing the thiazole moiety with an oxadiazole, as in 7 , yielded an inactive compound when screened at $10 \mu \mathrm{M}$ ligand concentration.









31



33




Fig. 3. Structures of all synthesised compounds carried through to in vitro testing in PAO1-L.


Scheme 1. Synthetic route to compounds 6, 8-26: (i) formamide, $150^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (ii) $\mathrm{NaH}, \mathrm{NMP}, 0^{\circ} \mathrm{C}, 30 \mathrm{~min}$, then chloroacetone, $0^{\circ} \mathrm{C}, 1 \mathrm{~h}$; (iii) bromine, acetic acid, $65{ }^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (iv) thioamide or thiourea, ethanol, $80^{\circ} \mathrm{C}, 16 \mathrm{~h}$; for complete structure of the compounds, see Fig. 3 .


Scheme 2. Synthetic route to 35: (i) ethanol, rt, 4 h , then $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{TEA}, \mathrm{rt}, 2 \mathrm{~h}$; (ii) HCl , zinc powder, $60^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (iii) DMFDMA, $100^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (iv) acetic acid, ethanol, $100^{\circ} \mathrm{C}, 16 \mathrm{~h}$.
methylamino thiazole (40) gave the desired product 6-chloro-3-((2-isopropylthiazol-5-yl)methyl)quinazolin-4(3H)-one (35).

The 2 -amino-5-substituted thiazoles (27-31) were prepared from reacting 6 -chloroquinazolin- $4(3 \mathrm{H})$-one ( $\mathbf{3 6}$ ) with 2 -chloro-5(chloromethyl)thiazole, to give 6-chloro-3-((2-chlorothiazol-5-yl) methyl)quinazolin-4(3H)-one (34). Displacement of the chlorine at the 2 -position of the thiazole with five different amines under neutral conditions gave products $\mathbf{2 7} \mathbf{- 3 1}$ in yields of $21-84 \%$ (Scheme 3).

In addition to variations about the thiazole, changes to the halogen decorating the quinazolinone core could alter binding to PqsR significantly. It was believed that the 6-chloro functionality in 6 could bind into a deep pocket within the LBD of PqsR, and possibly form a hydrogen bond with Thr265, as reported before. However, compounds reported by Ilangovan et al. contained a chlorine at the 7-position of the bicyclic core [32]. Therefore, in order to fully investigate the binding mode of this series, a 7-chloroquinazolin- $4(3 \mathrm{H})$-one subset was synthesised.

Two compounds were synthesised bearing a 7-chloro substitution on the bicyclic ring, with variations on the thiazole: 32 contained a tert-butyl group at the thiazole 2-position, and 33 a pentyl chain. The synthetic route was similar to that by which the 6 -chloro series was synthesised (Scheme 4), though chlorination of quinazolinone $\mathbf{4 3}$ yielded the desired $\alpha$-haloketone 44 . Subsequent thiazole formation led to the synthesis of $\mathbf{3 2}$ and 33.

### 2.2. Inhibition of the pqs system by potential PqsR antagonists

A PqsR-dependent bioreporter assay was used to assess the degree to which the compounds synthesised in the SAR study could inhibit the pqs QS system. Introduction of a mCTX: $P_{p q s A}-l u x$ transcriptional into the chromosomal CTX sites of PA strains PAO1-L and PA14 results in the production of bioluminescence at high bacterial cell densities in cultures where PQS and HHQ have reached their threshold activation concentrations for PqsR activation and in turn
the $P_{p q s A}$ promoter [39]. Consequently, inhibitors of PqsR reduce bioluminescence levels in these strains. Initially 96 -well plates containing the PAO1-L mCTX: $\mathrm{P}_{p q s A}-$ lux bioreporter strain grown in lysogeny broth (LB) were treated with a single concentration of $10 \mu \mathrm{M}$ of each compound to determine which compounds were active. As a threshold level for compound selection, a $50 \%$ reduction of pqs activity compared to the DMSO control was set, where pqs activity was a ratio of luminescence over $\mathrm{OD}_{600}$. Only compounds above the $50 \%$ threshold progressed for $\mathrm{IC}_{50}$ determination.

Of the 29 compounds tested, 14 compounds showed $>50 \%$ inhibition at a $10 \mu \mathrm{M}$ concentration and their concentration-dose response curves were generated. Importantly, the active compounds were found to not inhibit growth compared to the DMSO control (Fig. S2).

From the SAR study, it was apparent that short, unbranched alkyl chains led to a loss of activity, (see compounds 8-11). Furthermore, the addition of heteroatoms into the alkyl chain at the 2-position of the thiazole appeared to reduce activity. Of the seven amino alkyl chain containing compounds, only $\mathbf{2 5}$ ( $\mathrm{IC}_{50}=3.5 \mu \mathrm{M}$ in PAO1-L) had inhibitory activity at a ligand concentration of $10 \mu \mathrm{M}$. The presence of a tertiary amine did not improve activity, and the inclusion of an additional heteroatom led to no inhibition (compound 24).

Compounds 23 and 26, both containing three heteroatoms in their side chains, were found to be inactive, further suggesting that increase in polarity in the side chain leads to inactivity. In the case of $\mathbf{2 3}$ this may be due to the polarity of the nitro group; its charged state may result in the loss of activity in the whole cell-based assay as a result of poor cell penetration. Compound 26 had the lowest ClogP value of all compounds in the series (1.31). Notably, no compound with a ClogP value under 3.00 showed any activity, which also suggests that difficulty in penetrating the lipid bilayer contributes significantly to the inactivity of these compounds.

For the given series, 6-chloro substitution on the quinazolinone ring demonstrated improved activity over a 7-chloro substituted

 compounds, see Fig. 3.



Scheme 4. Synthetic route to 32-34: (i) formamide, $150^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (i) $\mathrm{NaH}, \mathrm{NMP}, 0^{\circ} \mathrm{C}, 30 \mathrm{~min}$, then chloroacetone, $0^{\circ} \mathrm{C}, 1 \mathrm{~h}$; (iii) NCS, sulfuric acid, $\mathrm{DCM}, 40^{\circ} \mathrm{C}, 6 \mathrm{~h}$; (iv) thioamide, acetic acid, ethanol, $80^{\circ} \mathrm{C}, 16 \mathrm{~h}$; for complete structure of the compounds, see Fig. 3 .
scaffold. Compound $32\left(\mathrm{IC}_{50}=2.9 \mu \mathrm{M}\right)$ was only marginally more active than the lead compound 45 ( $\mathrm{IC}_{50}=5.0 \mu \mathrm{M}$, Fig. 4) of a previously described 7 -chloro substituted scaffold [32]. Furthermore, compound $\mathbf{3 3}$ was inactive when tested at a ligand concentration of $10 \mu \mathrm{M}$. Comparable 6-chloro substituted compounds 12 $\left(\mathrm{IC}_{50}=397 \mathrm{nM}\right)$ and $\mathbf{1 8}\left(\mathrm{IC}_{50}=313 \mathrm{nM}\right)$ both showed submicromolar activity, with $\mathbf{1 2}$ displaying seven-fold stronger activity in strain PAO1-L compared with the weakly active 32.

Moreover, inverting the heteroatoms within the thiazole ring reduced activity, as indicated by the difference in activities of $\mathbf{6}$ ( $\mathrm{IC}_{50}=1.0 \mu \mathrm{M}$ ) and 35; the 2,5-disubstituted thiazole $\mathbf{3 5}$ was inactive at a concentration of $10 \mu \mathrm{M}$, whereas the 2,4 -analogue $\mathbf{6}$ was previously shown to have an activity of $1.0 \mu \mathrm{M}$. This inversion of the thiazole ring may have further contributed to the inactivity of compounds 27-31 in addition to the increased polarity in the side chain.

One argument rationalising the intolerance of heteroatoms may be poor bacterial uptake, or high rates of transporter-based efflux [40]. Although a whole cell approach removes the ability to define a binding event in the context of the assay itself, in combination with X-ray crystallography this methodology allows for definitive selection of compounds which both bind to the desired molecular target, and also have a proven efficacy.

Dose response curves were obtained for all 14 compounds exhibiting $>50 \%$ inhibition in the $10 \mu \mathrm{M}$ spot test, with activities ranging from 244 nM to $3.55 \mu \mathrm{M}$ in the bioreporter strain PAO1-L. Active compounds were then assayed against the clinicallyrelevant strain PA14 (Table 1). In general, the compounds displayed similar activities in both strains PAO1-L and PA14.

The pharmacological evaluation shed further light on the importance of the 2 -substitution of the thiazole. Regarding the length and branching of the alkyl chain, two clear trends became
evident. It was apparent that an increased chain length improved potency, with the two most potent inhibitors bearing a hexyl (19, $\mathrm{IC}_{50}=298 \mathrm{nM}$ ) and pentyl chain ( $\mathbf{1 8}, \mathrm{IC}_{50}=313 \mathrm{nM}$ ) respectively. Moreover, increased branching at the 1-position of the alkyl chain greatly improved potency: compound $12\left(\mathrm{IC}_{50}=397 \mathrm{nM}\right)$ containing a tert-butyl side chain demonstrated potent activity in strain PAO1-L.

Strain PAO1-L is a moderately virulent laboratory adapted strain routinely used to screen for PqsR inhibitors [34,42]. However, given the genomic diversity of PA strains, it is essential to test isolates belonging to the two major genomic groups. Hence, we also screened the inhibitors against the highly virulent PA14 strain [43]. It was therefore encouraging to see that activities were comparable between these strains. Moreover, for a few select compounds ( $\mathbf{1 5} \mathbf{- 1 7}, \mathbf{1 9}, \mathbf{2 2}$ ) potencies were shown to be greater in PA14, suggesting broad activity across the two main PA clades.

Three compounds ( $\mathbf{1 8}, 19$ and $\mathbf{2 2}, \mathrm{IC}_{50}=244 \mathrm{nM}$ ), were selected for X-ray crystallography based on the high potency across both strains. Compound $\mathbf{1 2}$ was also chosen for further study, due to its excellent potency in PAO1-L. Furthermore, compound 6 was selected for X-ray crystallography studies to provide a comparison between the first optimised compound in the series, and the most potent compounds.

### 2.3. Structure of PqsR ligand-binding domain complexed with compounds 6, 12, 18 and 19

Compounds 6, 12, 18 and 19 were successfully soaked into a truncated form of PqsR ( $\mathrm{PqsR}^{94-309}$ ) containing the ligand-binding domain of the protein (see S3-6 for additional data including electron density plots and crystallographic table of data collection


45


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Fig. 4. A previously described PqsR antagonist (45) featured a quinazolinone scaffold with a 7 -chloro substituted ring. Removal of the chlorine atom led to a 10 -fold drop in activity, whilst a 6 -chlorine substituted ring was inactive. By contrast, the compound series described in this article found 6 -chlorine substitution to be optimal. Both 45 and an endogenous ligand, NHQ (46) were successfully crystallised in the PqsR ligand-binding domain [32].

Table 1
In vitro data for all compounds defined as active (inhibits pqs system to below $50 \%$ at $10 \mu \mathrm{M}$ ). NA denotes inactive compounds which showed a remaining activity above $50 \%$ at $10 \mu \mathrm{M}$ (remaining activity for not active compounds is shown in brackets). Reported values are mean $\pm$ SD of $n=2$ replicates. cLogP values were calculated using MarvinSketch based on the algorithm outlined by Viswanadhan et al. [41].

| Compound | $\mathrm{IC}_{50}$ value in PAO1-L/nM | $\mathrm{IC}_{50}$ value in PA14/nM | Predicted cLogP values |
| :---: | :---: | :---: | :---: |
| 6 | $1046 \pm 419$ | $1578 \pm 358.0$ | 3.38 |
| 8 | NA (83\%) | NA | 2.14 |
| 9 | NA (52\%) | NA | 3.03 |
| 10 | NA (79\%) | NA | 2.84 |
| 11 | NA (104\%) | NA | 2.09 |
| 12 | $397 \pm 141.5$ | $650 \pm 35.1$ | 3.94 |
| 13 | $1216 \pm 44.0$ | $1333 \pm 192.5$ | 3.57 |
| 14 | $1563 \pm 674.0$ | $1720 \pm 882.0$ | 3.73 |
| 15 | $1112 \pm 879.5$ | $646 \pm 90.2$ | 4.27 |
| 16 | $1572 \pm 1119$ | $639 \pm 73.5$ | 4.02 |
| 17 | $1360 \pm 501.0$ | $683 \pm 156.9$ | 4.02 |
| 18 | $313 \pm 156.2$ | $342 \pm 39.4$ | 4.17 |
| 19 | $298 \pm 182.0$ | $265 \pm 3.4$ | 4.62 |
| 20 | $1327 \pm 189.5$ | $1308 \pm 272.0$ | 3.10 |
| 21 | $2048 \pm 1241.0$ | $2916 \pm 1005.5$ | 3.64 |
| 22 | $244 \pm 49.6$ | $123 \pm 20.9$ | 4.04 |
| 23 | NA (89\%) | NA | 3.98 |
| 24 | NA (85\%) | NA | 2.13 |
| 25 | $3545 \pm 2934.0$ | NA | 3.50 |
| 26 | NA (108\%) | NA | 1.31 |
| 27 | NA (72\%) | NA | 4.00 |
| 28 | NA (83\%) | NA | 4.02 |
| 29 | NA (101\%) | NA | 2.17 |
| 30 | NA (67\%) | NA | 3.55 |
| 31 | NA (85\%) | NA | 2.18 |
| 32 | $2942 \pm 808.0$ | NA | 3.94 |
| 33 | NA (106\%) | NA | 4.17 |
| 34 | NA (95\%) | NA | 2.93 |
| 35 | NA (65\%) | NA | 3.43 |

and refinement). All four ligands occupied a similar space to that exhibited by a previously described quinazolinone-based inhibitor 45 and two endogenous ligands (HHQ 3, and NHQ 46) bearing quinolone scaffolds [32,44]. In each case, the quinazolinone core occupies the previously defined B pocket of the LBD [32], and the alkyl chain extends into the open A pocket (Fig. 5). In addition, no major conformational changes occurred to the protein backbone in each of the crystal soaking experiments, as well as the previously reported PqsR crystal structures.

As with 45, the electron density contours around the halogen atom in each crystal, suggesting that the chlorine has locked the ligand into its defined orientation with the quinazolinone core occupying the B pocket of the PqsR LBD. However, whilst in 45 the
carbonyl faces towards the B sub-pocket, the electron density maps suggest that the carbonyl groups of $\mathbf{6}, 12,18$ and 19 all face towards the opposite face of the LBD in the direction of Thr265. The key result of this is that it enables a hydrogen bond to form between the carbonyl of each of the four compounds with the hydroxyl group of Thr265 (Fig. 6). This is further supported by the observation that the hydroxyl component of Thr265 faces towards the carbonyl in all four structures, in contrast to $\mathbf{4 5}$ where it faces towards the chlorine atom and elicits a polar interaction with the halogen.

The importance of this hydrogen bond was further emphasised by the lower activities of compounds 32 and 33 , both featuring a 7 chloro substitution. Repositioning of the chlorine atom removes the possibility of forming a hydrogen bond between the carbonyl and


Fig. 5. (a) Overlaying the crystal structures of $\mathbf{1 2}$ (green), $\mathbf{1 8}$ (cyan) and HHQ (3, orange, PDB entry 6Q7U) shows that the core quinazolinone structures occupy almost identical spaces within the LBD, with the alkyl chains extending out to the LBD entrance. Importantly, the carbonyl groups of $\mathbf{1 2}$ and $\mathbf{1 8}$ face towards Thr265, enabling a H-bond to form, whereas the carbonyl of HHQ faces the opposite wall of the LBD in the direction of the B-sub pocket; (b) The same trends are observed between $\mathbf{6}$ (gold), $\mathbf{1 9}$ (magenta) and $\mathbf{4 5}$ (yellow, PDB entry 4 JVI ). The chlorine atoms of $\mathbf{6}, 19$ and $\mathbf{4 5}$ directly overlap, demonstrating its importance in locking the structure's conformation NHQ ( $\mathbf{4 6}$, PDB entry 4 JVD ) followed the above trends, but was omitted for clarity. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)


Fig. 6. The binding of $\mathbf{6}$ into the PqsR LBD as a surface representation (a), and the key binding interactions observed between the carbonyl of $\mathbf{6}$ and Thr 265 , and the sulfur atom of $\mathbf{6}$ and Tyr258 (b). Below these are analogous representations for $\mathbf{1 2}(\mathrm{c}-\mathrm{d}), \mathbf{1 8}(\mathrm{e}-\mathrm{f})$ and $\mathbf{1 9}(\mathrm{g}-\mathrm{h})$. In each surface representation the chlorine atom is buried deep into the B pocket, anchoring the ligand in place. Binding modes are seen to be similar in each example, with the hydroxyl of Thr265 turned to face the carbonyl of the quinazolinone, thereby eliciting a H-bond. Furthermore, the sulfur is angled towards the phenol of Tyr258 enabling a polar interaction.

Thr265, hence the significant observed loss of potency.
The thiazole moieties were shown to further contribute to the activity of the series through interactions with Tyr258. In all four crystal structures, the sulfur atom was shown to orient itself facing the tyrosine, likely due to the formation of a polar interaction between the sulfur atom and the phenol group. The inactivity of oxadiazole 7 suggests that the presence of the sulfur aids in binding to the LBD, further implicating the thiazole as an integral component of the series. The role of the sulfur is further highlighted by the inactivity of $\mathbf{3 5}$, whereby the 2,5-thiazole regioisomer is unable to form a polar interaction with Tyr258, leading to a lower potency when compared to the 2,4-thiazole analogue 6.

The majority of residues in the LBD of PqsR bear aliphatic side chains. As such, hydrophobic interactions are seen to drive activity in both endogenous ligands and synthetic inhibitors. The SAR study
indicated that increasing the length and branching of the aliphatic substitution at the 2-position of the thiazole dramatically improved potency. This is observed in the crystal structures, whereby the aliphatic chain forms hydrophobic interactions with the side chains of residues in the A pocket of the LBD. Compounds $\mathbf{6}\left(\mathrm{IC}_{50}=1.0 \mu \mathrm{M}\right)$ and $12\left(\mathrm{IC}_{50}=397 \mathrm{nM}\right)$ are weaker inhibitors than 18 $\left(\mathrm{IC}_{50}=313 \mathrm{nM}\right)$ and $19\left(\mathrm{IC}_{50}=298 \mathrm{nM}\right)$, as they have shorter alkyl chains unable to form as many hydrophobic interactions with the protein. In particular, 18 and 19 are able to reach Ile186 and Leu189 at the far edge of the A pocket, creating a larger network of hydrophobic interactions and improving the activity over shorter chain analogues such as $\mathbf{6}$ and 12 (Fig. S4).

### 2.4. Reduction of pyocyanin production by the PqsR inhibitors

To demonstrate a link between the inhibitory effects of $\mathbf{1 8}$ and 19 on PqsR and hence virulence gene expression, pyocyanin production in the PAO1-L was quantified after overnight incubation with either $\mathbf{1 8}$ or 19 at a concentration three times higher than their $\mathrm{IC}_{50}$, or a vehicle (DMSO) control. Both compounds reduced pyocyanin production to $23 \%$ and $36 \%$ respectively (Fig. 7), compared with the control. These results indicate that these compounds can attenuate the production of this virulence factor through inhibition of PqsR similar to that shown for other PqsR inhibitors [34,45,46].

### 2.5. Effect of PqsR inhibitors on cytotoxicity of lung epithelial cells

A cytotoxicity study using A549 lung epithelial cells was conducted with 18 and 19, to establish the suitability of these compounds for further studies. Compounds 18 and 19 tested at increasing concentrations ranging from 0.1 to $100 \mu \mathrm{M}$ showed no significant toxicity (Fig. 8). Testing at higher concentrations was not possible as neither compound was fully soluble above this concentration. However, the data showed that both $\mathbf{1 8}$ and 19 are not cytotoxic up to $100 \mu \mathrm{M}$. Therefore, the therapeutic index is $>292-$ fold in $\mathbf{1 8}$ and $>377$-fold in 19 against the A549 cell line relative to the activities calculated for PA14.

## 3. Conclusion

In summary, a new series of chloro-3-((2-substituted-thiazole) quinazolin- $4(3 \mathrm{H})$-one compounds was synthesised and tested in a whole cell bioreporter assay to determine their ability to inhibit PqsR, and subsequently reduce $P$. aeruginosa pyocyanin production. A SAR study showed key contributing functionalities, particularly a long or branched alkyl chain at the 2-position of the thiazole. These studies revealed a preference for chlorine substitution on the 6position of the quinazolinone ring.

X-ray crystallography into the LBD of PqsR confirmed a binding


Fig. 7. Compounds 18 and 19 were shown to significantly reduce pyocyanin production to $23 \%$ and $36 \%$ respectively against a control of $0.1 \%$ DMSO at $3 \times$ the $\mathrm{IC}_{50}$ value in P. aeruginosa strain PAO1-L.
mode similar to that seen in previously described quinazolinoneand quinolone-based inhibitor-bound crystal structures. Further assays confirmed that the most active compounds $\mathbf{1 8}$ and 19 were capable of significantly reducing production of the toxin pyocyanin, whilst remaining non-toxic to eukaryotic cells, providing a basis for further pre-clinical studies.

## 4. Experimental

### 4.1. Chemistry - general methods

Chemicals and solvents were provided by Fisher Scientific UK, Acros Organics, Sigma-Aldrich, Merck Millipore or Fluorochem. All reactions were monitored by TLC using Merck Silica Gel 60 Å F254 TLC plates or by LC-MS. Unless otherwise stated, all compounds were dried under high vacuum either at rt or within an oven at $40^{\circ} \mathrm{C}$. LC-MS data was collected on a Shimadzu UFLCXR HPLC system coupled to an Applied Biosystems API 2000 LC/MS/MS electrospray ionization (ESI). The column used was a Phenomenex Gemini-NX $3 \mu \mathrm{~m}-110 \mathrm{~A} \mathrm{C} 18,50 \times 2 \mathrm{~mm}$ at $40^{\circ} \mathrm{C}$. The flow rate was $0.5 \mathrm{~mL} / \mathrm{min}$, the UV detection was at 220 nm and 254 nm . The LCMS ran for 1 min at $5 \% \mathrm{~B} ; 5-98 \%$ B over $2 \mathrm{~min}, 98 \%$ B for $2 \mathrm{~min}, 98$ to $5 \%$ B over 0.5 min and then $5 \%$ for 1 min where solvent $A: 0.1 \%$ formic acid in water; solvent B: acetonitrile. Unless otherwise stated compounds reported had a purity $>95 \%$. NMR spectroscopy was performed using a Bruker AV(III) HD 400 NMR spectrometer equipped with a 5 mm BBFO ${ }^{+}$probe, recording ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR at 400.25 MHz and 100.66 MHz respectively; or a Bruker AV(III) 500 NMR spectrometer equipped with a 5 mm dual ${ }^{1} \mathrm{H} /{ }^{13} \mathrm{C}$ heliumcooled cryoprobe, recording ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR at 500.13 MHz and 125.77 MHz respectively. NMR data was processed using iNMR (version 5.5.7) referencing spectra to residual solvents. Chemical shifts are quoted as $\delta$ : values in ppm; coupling constants J are given in Hz and multiplicities are described as follows: s - singlet, d doublet, t - triplet, q - quartet, qi - quintet, sep - septet, m multiplet, app - apparent, br - broad.

### 4.1.1. General procedure for the preparation of thiazoles 6, 8-23,

 32, 33A solution of the appropriate thioamide or thiourea ( 0.13 mmol ) and $\alpha$-haloketone 38 or $44(0.06 \mathrm{mmol})$ in EtOH ( 4 mL ) was refluxed at $80^{\circ} \mathrm{C}$ for 16 h . ${ }^{a}$ After cooling to room temperature, the reaction mixture was concentrated in vacuo and purified through flash column chromatography. Chromatography was run with a gradient of ethyl acetate/petroleum ether (1:2) to pure ethyl acetate. ${ }^{b}$
a Where compound 44 was used as the $\alpha$-haloketone, $600 \mu \mathrm{~L}$ of acetic acid was added to aid in solubilizing the starting material.
b Excepting compounds $\mathbf{1 1}$ and $\mathbf{2 6}$ which were run in DCM/ methanol (19:1).
4.1.1.1. 6-Chloro-3-((2-isopropylthiazol-4-yl)methyl)quinazolin-4(3H)-one (6). Obtained $\mathbf{6}(18 \mathrm{mg}, 90 \%)$ as a white crystal: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.54(\mathrm{~s}, 1 \mathrm{H}), 8.05(\mathrm{~d}, J=1.68 \mathrm{~Hz}, 1 \mathrm{H}), 7.83$ (dd, $J=8.61,1.72 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{~d}, J=8.69 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~s}, 1 \mathrm{H}), 5.25(\mathrm{~s}$, $2 \mathrm{H}), 3.24-3.17(\mathrm{~m}, 1 \mathrm{H}), 1.26(\mathrm{~d}, J=6.84 \mathrm{~Hz}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (DMSO-d $d_{6} 101 \mathrm{MHz}$ ) $\delta 177.7,158.9,150.0,148.5,146.6,134.5,131.4$, 129.5, 125.1, 122.9, 115.7, 45.4, 32.4, 22.8 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{ClN}_{3} \mathrm{OS} m / z 320.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 320.1$ $(M+H)$.
4.1.1.2. 6-Chloro-3-((2-methylthiazol-4-yl)methyl)quinazolin-4(3H)one (8). Obtained $\mathbf{8}(17 \mathrm{mg}, 90 \%)$ as an orange crystal: ${ }^{1} \mathrm{H}$ NMR


Fig. 8. Cytotoxicity data for (left) 18 and (right) 19 indicated that neither shows significant toxicity up to $100 \mu \mathrm{M}$ in A549 lung epithelial cells. The lethal dose (LD $\mathrm{D}_{50}$ ) value could not be calculated, as cell viability was not reduced to $<50 \%$ (indicated by the red dashed line). It was not possible to test at higher compound concentrations due to insolubility in the test buffer. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)
$\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.34(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, \mathrm{~J}=1.24 \mathrm{~Hz}, 1 \mathrm{H}), 7.69-7.63$ (m, 2H), $7.20(\mathrm{~s}, 1 \mathrm{H}), 5.22(\mathrm{~s}, 2 \mathrm{H}), 2.66(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ 101 MHz ) $\delta 159.9,149.4,146.9,146.6,134.7,133.1,129.2,126.2,123.2$, 117.7, 45.4, 29.7, 19.1 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{ClN}_{3} \mathrm{OS} m / z 292.0(\mathrm{M}+\mathrm{H})$, found $m / z 292.1(\mathrm{M}+\mathrm{H})$.
4.1.1.3. 6-Chloro-3-((2-(trifluoromethyl)thiazol-4-yl)methyl)quina-zolin-4(3H)-one (9). Obtained 9 ( $3.5 \mathrm{mg}, 16 \%$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (CDCl $3,400 \mathrm{MHz}) \delta 8.35(\mathrm{~s}, 1 \mathrm{H}), 8.24(\mathrm{~d}, J=1.64 \mathrm{~Hz}, 1 \mathrm{H}), 7.71$ (s, 1H), 7.69-7.66 (m, 2H), $5.32(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$; LC-MS (+ESI) calculated for $\mathrm{C}_{13} \mathrm{H}_{7} \mathrm{ClF}_{3} \mathrm{~N}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z} 346.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 346.1$ $(M+H)$.
4.1.1.4. 6-Chloro-3-((2-ethylthiazol-4-yl)methyl)quinazolin-4(3H)one (10). Obtained $\mathbf{1 0}(14 \mathrm{mg}, 70 \%)$ as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}) \delta 8.28(\mathrm{~s}, 1 \mathrm{H}), 8.15(\mathrm{~s}, 1 \mathrm{H}), 7.61-7.55(\mathrm{~m}, 2 \mathrm{H}), 7.14(\mathrm{~s}, 1 \mathrm{H})$, $5.16(\mathrm{~s} 2 \mathrm{H}), 2.90(\mathrm{q}, J=7.52 \mathrm{~Hz}, 2 \mathrm{H}), 1.27(\mathrm{t}, J=7.56 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3} 101 \mathrm{MHz}\right) \delta$ 173.9, 159.7, 149.1, 147.0, 146.5, 134.6, 132.9, 129.1, 126.0, 123.1, 117.1, 45.3, 26.8, 14.0 ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{ClN}_{3} \mathrm{OS} m / z 306.0(\mathrm{M}+\mathrm{H})$, found $m / z 306.1(\mathrm{M}+\mathrm{H})$.
4.1.1.5. 2-(4-((6-Chloro-4-oxoquinazolin-3(4H)-yl)methyl)thiazol-2yl)acetonitrile (11). Obtained 11 ( $18 \mathrm{mg}, 90 \%$ ) as a red-orange solid: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.32(\mathrm{~s}, 1 \mathrm{H}), 8.23(\mathrm{~d}, J=1.80 \mathrm{~Hz}, 1 \mathrm{H})$, $7.70-7.64(\mathrm{~m}, 2 \mathrm{H}), 7.43(\mathrm{~s}, 1 \mathrm{H}), 5.24(\mathrm{~s}, 2 \mathrm{H}), 4.06(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3} 101 \mathrm{MHz}$ ) $\delta 159.8,158.4,150.6,146.6,146.5,134.8$, 133.3, 129.2, 126.1, 123.1, 119.8, 115.0, 45.3, 22.2 ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{ClN}_{4} \mathrm{OS} m / z 317.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 317.2$ ( $M+H$ ).
4.1.1.6. 3-((2-(Tert-butyl)thiazol-4-yl)methyl)-6-chloroquinazolin$4(3 \mathrm{H})$-one (12). Obtained 12 ( $9.0 \mathrm{mg}, 43 \%$ ) as a white crystal: ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.43(\mathrm{~s}, 1 \mathrm{H}), 8.27(\mathrm{~d}, J=1.80 \mathrm{~Hz}, 1 \mathrm{H})$, 7.70-7.65 (m, 2H), $7.19(\mathrm{~s}, 1 \mathrm{H}), 5.26(\mathrm{~s}, 2 \mathrm{H}), 1.40(\mathrm{~s}, 9 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3} 101 \mathrm{MHz}$ ) $\delta 182.4,159.9,148.9,147.1,146.6,134.7,133.1$, 129.2, 126.2, 123.2, 116.7, 45.3, 37.8, 30.9 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{ClN}_{3} \mathrm{OS} m / z 334.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 334.2$ ( $M+H$ ).
4.1.1.7. 6-Chloro-3-((2-isobutylthiazol-4-yl)methyl)quinazolin$4(3 \mathrm{H})$-one (13). Obtained 13 ( $11 \mathrm{mg}, 53 \%$ ) as a silver crystal: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.54(\mathrm{~s}, 1 \mathrm{H}), 8.07(\mathrm{~d}, J=2.36 \mathrm{~Hz}, 1 \mathrm{H})$,
7.86 (dd, $J=8.69,2.40 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.69 \mathrm{~Hz}, 1 \mathrm{H}), 7.42(\mathrm{~s}, 1 \mathrm{H})$, $5.25(\mathrm{~s}, 2 \mathrm{H}), 2.77(\mathrm{~d}, J=7.08 \mathrm{~Hz}, 2 \mathrm{H}), 1.95(\mathrm{qt}, J=6.72,6.72 \mathrm{~Hz}, 1 \mathrm{H})$, 0.88 (d, $J=6.65 \mathrm{~Hz}, 6 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR (DMSO-d $\left.{ }_{6}, 101 \mathrm{MHz}\right) \delta 170.1$, 158.9, 150.1, 148.6, 146.6, 134.5, 131.4, 129.5, 125.1, 122.9, 116.4, 45.3, 41.2, 29.1, 21.9 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{ClN}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z}$ $334.1(M+H)$, found $m / z 334.1(M+H)$.
4.1.1.8. 3-((2-Butylthiazol-4-yl)methyl)-6-chloroquinazolin-4(3H)one (14). Obtained 14 ( $29 \mathrm{mg}, 88 \%$ ) as a yellow crystal: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.34(\mathrm{~s}, 1 \mathrm{H}), 8.23(\mathrm{dd}, J=2.00,0.87 \mathrm{~Hz}, 1 \mathrm{H})$, $7.67-7.60(\mathrm{~m}, 2 \mathrm{H}), 7.19(\mathrm{~s}, 1 \mathrm{H}), 5.22(\mathrm{~s}, 2 \mathrm{H}), 2.92(\mathrm{t}, J=7.74 \mathrm{~Hz} 2 \mathrm{H})$, 1.71 (tt, $J=7.70,7.70 \mathrm{~Hz}, 2 \mathrm{H}), 1.38(\mathrm{tq}, J=7.45,7.45 \mathrm{~Hz}, 3 \mathrm{H}), 0.90(\mathrm{t}$, $J=7.38 \mathrm{~Hz}, 5 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO) $\delta 171.4,158.9$, $150.0,148.6,146.7,134.5,131.4,129.6,125.1,122.9,116.2,45.4,32.2$, 31.4, 21.5, 13.5 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{ClN}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z}$ $334.1(M+H)$, found $m / z 334.1(M+H)$.
4.1.1.9. 6-Chloro-3-((2-(pentan-2-yl)thiazol-4-yl)methyl)quinazolin-4(3H)-one (15). Obtained $\mathbf{1 5}(11 \mathrm{mg}, 45 \%)$ as a yellow oily solid: ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.37(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=1.52 \mathrm{~Hz}, 1 \mathrm{H})$, $7.69-7.63(\mathrm{~m}, 2 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 5.24(\mathrm{~s}, 2 \mathrm{H}), 3.14(\mathrm{qt}, J=6.96$, $6.96 \mathrm{~Hz}, 1 \mathrm{H}), 1.77-1.53(\mathrm{~m}, 2 \mathrm{H}), 1.33-1.25(\mathrm{~m}, 5 \mathrm{H}), 0.88(\mathrm{t}$, $J=7.32 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (CDCl ${ }_{3} 101 \mathrm{MHz}$ ) $\delta 178.6,159.8$, 148.8, 147.0, 146.5, 134.7, 133.1, 129.1, 126.2, 123.2, 116.7, 45.3, 39.9, 38.3, 21.4, 20.3, 13.9 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{OS} m / z 348.1(\mathrm{M}+\mathrm{H})$, found $m / z 348.2(\mathrm{M}+\mathrm{H})$.
4.1.1.10. 6-Chloro-3-((2-(2-methylbutyl)thiazol-4-yl)methyl)quina-zolin-4(3H)-one (16). Obtained 16 ( $20 \mathrm{mg}, 91 \%$ ) as a colourless oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.35(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=1.64 \mathrm{~Hz}, 1 \mathrm{H})$, $7.69-7.63(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{~s}, 1 \mathrm{H}), 5.24(\mathrm{~s}, 2 \mathrm{H}), 2.94(\mathrm{dd}, J=14.61$, $6.08 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{dd}, J=14.61,8.09,1 \mathrm{H}), 1.89-1.80(\mathrm{~m}, 1 \mathrm{H})$, $1.46-1.14(\mathrm{~m}, 2 \mathrm{H}), 0.92-0.87(\mathrm{~m}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ 101 MHz ) $\delta 171.6,159.7,149.2,146.9,146.6,134.7,129.2,126.2,123.0$, 117.3, 45.4, 40.3, 36.0, 29.7, 29.1, 19.0, 11.3 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{OS} m / z 348.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 348.2$ $(M+H)$.
4.1.1.11. 6-Chloro-3-((2-isopentylthiazol-4-yl)methyl)quinazolin$4(3 \mathrm{H})$-one (17). Obtained $17(13 \mathrm{mg}, 71 \%)$ as a white crystal: ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.36(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=1.48 \mathrm{~Hz}, 1 \mathrm{H})$, $7.69-7.63(\mathrm{~m}, 2 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 5.23(\mathrm{~s}, 2 \mathrm{H}), 2.95(\mathrm{t}, J=7.61 \mathrm{~Hz}, 2 \mathrm{H})$,
$1.65-1.62(\mathrm{~m}, 3 \mathrm{H}), 0.93(\mathrm{~d}, J=6.04 \mathrm{~Hz}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right.$ $101 \mathrm{MHz}) \delta 172.9,159.8,149.2,146.9,134.7,133.0,129.2,126.2$, 123.3, 117.2, 45.4, 38.9, 31.5, 29.7, 27.7, 23.3 ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{OS} m / z 348.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 348.2$ ( $M+H$ ).
4.1.1.12. 6-Chloro-3-((2-pentylthiazol-4-yl)methyl)quinazolin-4(3H)one (18). Obtained 18 ( $20 \mathrm{mg}, 90 \%$ ) as a white crystal: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.35(\mathrm{~s}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=1.92 \mathrm{~Hz}, 1 \mathrm{H}), 7.69-7.63$ (m, 2H), $7.20(\mathrm{~s}, 1 \mathrm{H}), 5.23(\mathrm{~s}, 2 \mathrm{H}), 2.93(\mathrm{t}, J=7.68 \mathrm{~Hz}, 2 \mathrm{H}), 1.77-1.69$ $(\mathrm{m}, 2 \mathrm{H}), 1.37-1.33(\mathrm{~m}, 4 \mathrm{H}), 0.90(\mathrm{t}, J=6.93 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3} 101 \mathrm{MHz}\right) \delta 172.8,159.9,149.1,147.0,146.6,134.7,133.1,129.2$, 126.2, 123.2, 117.2, 45.4, 33.4, 31.3, 29.6, 22.3, 13.9 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{OS} m / z 348.1(\mathrm{M}+\mathrm{H})$, found $m / z$ $348.2(\mathrm{M}+\mathrm{H})$.
4.1.1.13. 6-Chloro-3-((2-hexylthiazol-4-yl)methyl)quinazolin-4(3H)one (19). Obtained 19 ( $15 \mathrm{mg}, 66 \%$ ) as a white crystal: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.36(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=1.72 \mathrm{~Hz}, 1 \mathrm{H}), 7.69-7.64$ ( m, 2H), $7.21(\mathrm{~s}, 1 \mathrm{H}), 5.23(\mathrm{~s}, 2 \mathrm{H}), 2.94(\mathrm{t}, J=7.89 \mathrm{~Hz}, 2 \mathrm{H}), 1.74(\mathrm{tt}$, $J=7.64,7.64 \mathrm{~Hz}, 2 \mathrm{H}), 1.40-1.33(\mathrm{~m}, 2 \mathrm{H}), 1.30-1.27(\mathrm{~m}, 4 \mathrm{H}), 0.86(\mathrm{t}$, $J=6.93 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3} 101 \mathrm{MHz}\right) \delta 172.8,159.8$, 149.1, 146.9, 146.5, 134.7, 133.1, 129.1, 126.1, 123.2, 117.2, 45.4, 33.4, 31.4, 29.9, 28.7, 22.4, 14.0 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{ClN}_{3} \mathrm{OS} m / z 362.1(M+H)$, found $m / z 362.3(M+H)$.
4.1.1.14. 6-Chloro-3-((2-(furan-2-yl)thiazol-4-yl)methyl)quinazolin$4(3 \mathrm{H})$-one (20). Obtained $20(17 \mathrm{mg}, 80 \%)$ as a white crystal: ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.40(\mathrm{~s}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=1.88 \mathrm{~Hz}, 1 \mathrm{H}), 7.67$ $(\mathrm{d}, J=2.20 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.50(\mathrm{~d}, J=1.09 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{~s}, 1 \mathrm{H})$, 6.97 (d, $J=3.64 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{dd}, J=3.36,1.80 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~s}, 2 \mathrm{H})$ $\mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3} 101 \mathrm{MHz}\right) \delta$ 159.9, 158.9, 150.6, 148.5, 146.9, $146.6,143.9,134.8,133.2,129.2,126.2,123.2,117.3,112.3,109.5$, 45.4 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{ClN}_{3} \mathrm{O}_{2} \mathrm{~S} \mathrm{~m} / \mathrm{z} 344.0$ $(M+H)$, found $m / z 344.1(M+H)$.
4.1.1.15. 6-Chloro-3-((2-(2-(furan-2-yl)vinyl)thiazol-4-yl)methyl) quinazolin-4(3H)-one (21). Obtained $21(12 \mathrm{mg}, 51 \%)$ as a cream solid. A minor impurity remained, giving a final purity of $91.7 \%$ as determined by ${ }^{1} \mathrm{H}$ NMR: ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}, 400 \mathrm{MHz}$ ) $\delta 8.60$ (s, $1 \mathrm{H}), 8.09(\mathrm{~d}, J=2.32 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{dd}, J=8.73,2.36 \mathrm{~Hz}, 1 \mathrm{H})$, $7.77-7.73$ (m, 2H), 7.54 (s, 1H), 7.29 (d, $J=16.09 \mathrm{~Hz}, 1 \mathrm{H}), 7.08$ (d, $J=16.09 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=3.24 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{dd}, J=2.88,1.64 \mathrm{~Hz}$, 1H), 5.30 (s, 2H) ppm; ${ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO) $\delta$ 166.0, 159.0, 151.7, 151.1, 148.6, 146.7, 144.5, 134.6, 131.5, 129.6, 125.1, 122.9, 121.5, 118.3, 117.0, 112.7, 112.5, 45.5 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{ClN}_{3} \mathrm{O}_{2} \mathrm{~S} m / z 370.0(\mathrm{M}+\mathrm{H})$, found $m / z 370.2(\mathrm{M}+\mathrm{H})$.
4.1.1.16. 6-Chloro-3-((2-phenylthiazol-4-yl)methyl)quinazolin-4(3H)-one (22). Obtained 22 ( $20 \mathrm{mg}, 92 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.47(\mathrm{~s}, 1 \mathrm{H}), 8.27(\mathrm{~d}, J=1.72 \mathrm{~Hz}, 1 \mathrm{H}), 7.92-7.89$ $(\mathrm{m}, 2 \mathrm{H}), 7.70-7.64(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.41(\mathrm{~m}, 3 \mathrm{H}), 7.36(\mathrm{~s}, 1 \mathrm{H}), 5.32(\mathrm{~s}$, $2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3} 101 \mathrm{MHz}\right) \delta$ 169.1, 159.9, 150.7, 147.1, 146.6, 134.7, 133.1, 133.1, 130.4, 129.2, 129.0, 126.5, 126.2, 123.2, 117.9, 45.5 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{ClN}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z} 354.0$ $(M+H)$, found $m / z 354.2(M+H)$.
4.1.1.17. 6-Chloro-3-((2-(4-nitrophenyl)thiazol-4-yl)methyl)quinazo-lin-4(3H)-one (23). Obtained $\mathbf{2 3}(15 \mathrm{mg}, 48 \%)$ as a white crystal: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 8.66(\mathrm{~s}, 1 \mathrm{H}), 8.36-8.28(\mathrm{~m}, 2 \mathrm{H}), 8.15(\mathrm{~d}$, $J=2.08 \mathrm{~Hz}, 1 \mathrm{H}), 8.14-8.06(\mathrm{~m}, 2 \mathrm{H}), 7.91-7.83(\mathrm{~m}, 2 \mathrm{H}), 7.75(\mathrm{~d}$, $J=8.74 \mathrm{~Hz}, 1 \mathrm{H}), 5.39(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$; LC-MS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{ClN}_{4} \mathrm{O}_{3} \mathrm{~S} m / z 399.0(\mathrm{M}+\mathrm{H})$, found $m / z 399.1(M+H)$.
4.1.1.18. 6-Chloro-3-((2-((2-methoxyethyl)amino)thiazol-4-yl) methyl)quinazolin-4(3H)-one (24). Obtained $24(34 \mathrm{mg}, 51 \%)$ as a cream solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.47(\mathrm{~s}, 1 \mathrm{H}), 8.08(\mathrm{~d}$, $J=2.67 \mathrm{~Hz}, 1 \mathrm{H}), 7.85(\mathrm{dd}, J=9.06,2.70 \mathrm{~Hz}, 1 \mathrm{H}), 7.72-7.67(\mathrm{~m}, 2 \mathrm{H})$, $6.42(\mathrm{~s}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 2 \mathrm{H}), 3.41(\mathrm{t}, J=5.39 \mathrm{~Hz}, 2 \mathrm{H}), 3.32(\mathrm{t}, J=4.91 \mathrm{~Hz}$, 2H), 3.21 ( $\mathrm{s}, 3 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}, 101 \mathrm{MHz}$ ) $\delta 169.1,158.9$, 148.6, 146.6, 146.3, 134.5, 131.3, 129.5, 125.1, 122.9, 102.8, 70.1, 57.9, 45.5, 43.9 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S} \mathrm{~m} / \mathrm{z}$ $351.1(M+H)$, found $m / z 350.6(M+H)$.
4.1.1.19. 3-((2-(Butylamino)thiazol-4-yl)methyl)-6-chloroquinazolin$4(3 \mathrm{H})$-one (25). Obtained 25 ( $16 \mathrm{mg}, 24 \%$ ) as a white crystal: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.46(\mathrm{~s}, 1 \mathrm{H}), 8.08(\mathrm{~d}, J=2.65 \mathrm{~Hz}, 1 \mathrm{H})$, 7.86 (s, dd, $J=8.68,2.65 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.68 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{t}$, $J=5.67 \mathrm{~Hz}, 1 \mathrm{H}), 6.42(\mathrm{~s}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 2 \mathrm{H}), 3.11(\mathrm{dt}, J=6.87,5.20 \mathrm{~Hz}$, $2 \mathrm{H}), 1.46\left(\mathrm{tt}, J_{1}=J_{2}=7.07 \mathrm{~Hz}\right), 1.28(\mathrm{tq}, J=7.33,7.33 \mathrm{~Hz}, 2 \mathrm{H}), 0.83(\mathrm{t}$, $J=7.32 \mathrm{~Hz}, 3 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR (DMSO- $d_{6}, 101 \mathrm{MHz}$ ) $\delta 169.4,158.9$, $148.6,146.7,146.5,134.5,131.3,129.5,125.1,122.9,102.4,45.5,44.2$, 30.7, 19.5, 13.6 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{ClN}_{4} \mathrm{OS} \mathrm{m} / \mathrm{z}$ $349.1(M+H)$, found $m / z 348.9(M+H)$.
4.1.1.20. 2-(4-((6-Chloro-4-oxoquinazolin-3(4H)-yl)methyl)thiazol-2-yl)guanidine (26). Obtained 26 ( $13 \mathrm{mg}, 62 \%$ ) as a cream solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.55(\mathrm{~s}, 1 \mathrm{H}), 8.08(\mathrm{~d}, J=2.40 \mathrm{~Hz}, 1 \mathrm{H})$, $7.85-7.83(\mathrm{~m}, 1 \mathrm{H}), 7.70(\mathrm{dd}, J=8.77,2.36 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{br} \mathrm{s}, 4 \mathrm{H})$, 5.06 (s, 2H) ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{6} \mathrm{OS} \mathrm{m} / \mathrm{z}$ $335.0(M+H)$, found $m / z 335.2(M+H)$.
4.1.2. General procedure for the preparation of 3-((2-amino) thiazol-5-yl)methyl)quinazolin-4(3H)-one compounds 27-31 from 34

A solution of $34(0.32 \mathrm{mmol})$ and appropriate amine ( 0.96 mmol ) in DMSO ( 0.5 mL ) was refluxed at $100-130{ }^{\circ} \mathrm{C}$ (dependent on the boiling point of the amine) for 16 h . The reaction mixture was cooled to room temperature, diluted with water and extracted with ethyl acetate $(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over brine and concentrated in vacuo. Purification through flash column chromatography in ethyl acetate/petroleum ether (4:1) to ethyl acetate to methanol/ethyl acetate (19:1) yielded products $\mathbf{2 7 - 3 1}$ in yields of 21-84\%.
4.1.2.1. 6-Chloro-3-((2-(4-methylpiperidin-1-yl)thiazol-5-yl)methyl) quinazolin-4-(3H)-one (27). Obtained 27 ( $86 \mathrm{mg}, 72 \%$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.57(\mathrm{~s}, 1 \mathrm{H}), 8.12$ (d, $J=2.28 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{dd}, J=8.82,2.67 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.70 \mathrm{~Hz}$, 1 H ), $7.26(\mathrm{~s}, 1 \mathrm{H}), 5.20(\mathrm{~s}, 2 \mathrm{H}), 3.80-3.77(\mathrm{~m}, 2 \mathrm{H}), 2.92$ (ddd, $J=3.02$, $3.02,12.44 \mathrm{~Hz}, 2 \mathrm{H}), 1.65-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.58-1.51(\mathrm{~m}, 1 \mathrm{H}), 1.10$ (dddd, $J=4.14,4.14,4.14,11.93 \mathrm{~Hz}, 2 \mathrm{H}), 0.89(\mathrm{~d}, J=6.16 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 101 \mathrm{MHz}\right) \delta 171.7,159.0,147.8,146.6,140.2,134.6$, 131.6, 129.6, 125.0, 122.6, 119.8, 48.2, 42.2, 32.7, 30.0, 21.5 ppm; LCMS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{ClN}_{4} \mathrm{OS} m / z 375.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} /$ $z 374.6(\mathrm{M}+\mathrm{H})$.
4.1.2.2. 6-Chloro-3-((2-(cyclohexylamino)thiazol-5-yl)methyl)quina-zolin-4-(3H)-one (28). Obtained 28 ( $25 \mathrm{mg}, 21 \%$ ) as a brown solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.55(\mathrm{~s}, 1 \mathrm{H}), 8.11(\mathrm{~d}, J=2.52 \mathrm{~Hz}, 1 \mathrm{H})$, 8.67 (dd, $J=8.55,2.21 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.78 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~d}$, $J=7.48 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~s}, 1 \mathrm{H}), 5.15(\mathrm{~s}, 2 \mathrm{H}), 3.39-3.34(\mathrm{~m}, 1 \mathrm{H})$, $1.89-1.84(\mathrm{~m}, 2 \mathrm{H}), 1.69-1.64(\mathrm{~m}, 3 \mathrm{H}), 1.56-1.51(\mathrm{~m}, 1 \mathrm{H}), 1.31-1.11$ (m, 4H) ppm; ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}, 101 \mathrm{MHz}$ ) $\delta$ 169.0, 159.0, 147.9, 146.6, 139.5, 134.6, 131.6, 129.6, 125.0, 122.7, 117.9, 53.0, 42.3, 32.3, 25.3, 24.4 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{ClN}_{4} \mathrm{OS} \mathrm{m} / \mathrm{z}$ $375.1(\mathrm{M}+\mathrm{H})$, found $m / z 374.6(\mathrm{M}+\mathrm{H})$.
4.1.2.3. 6-Chloro-3-((2-(2-hydroxyethyl(methyl)amino)thiazol-5-yl) methyl)quinazolin-4-(3H)-one (29). Obtained 29 ( $94 \mathrm{mg}, 84 \%$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.57(\mathrm{~s}, 1 \mathrm{H}), 8.12(\mathrm{~d}$, $J=2.52 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{dd}, J=8.3,2.51 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.24(\mathrm{~s}, 1 \mathrm{H}), 5.20(\mathrm{~s}, 2 \mathrm{H}), 4.75(\mathrm{t}, J=5.74,1 \mathrm{H}), 3.55(\mathrm{dt}, J=6.31$, $6.31 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.43(\mathrm{t}, J=5.75,2 \mathrm{H}), 2.99(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}, 101 \mathrm{MHz}$ ) $\delta 171.1,159.0,147.8,146.6,140.3,134.6,131.6$, 129.5, 125.0, 122.6, 119.0, 58.1, 54.7, 42.3, 40.4 ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S} m / z 351.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 350.6$ $(M+H)$.
4.1.2.4. 3-((2-(Butylamino)thiazol-5-yl)methyl)-6-chloroquinazolin$4(3 \mathrm{H})$-one (30). Obtained $30(46 \mathrm{mg}, 41 \%)$ as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right) \delta 8.54(\mathrm{~s}, 1 \mathrm{H}), 8.11(\mathrm{~d}, J=2.22 \mathrm{~Hz}, 1 \mathrm{H})$, 7.85 (dd, $J=8.03,2.22 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.70 (d, $J=7.75 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.56 (t, $J=5.54 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H}), 5.15(\mathrm{~s}, 2 \mathrm{H}), 3.13(\mathrm{dt}, J=6.96,6.96 \mathrm{~Hz}$, $2 \mathrm{H}), 1.46(\mathrm{tt}, J=7.01,7.01 \mathrm{~Hz}, 2 \mathrm{H}), 1.29(\mathrm{tq}, J=7.36,7.36 \mathrm{~Hz}, 2 \mathrm{H})$, $0.85(\mathrm{t}, J=7.41 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 172.0$, 160.1, 146.7, 145.9, 139.8, 135.0, 133.5, 129.4, 126.3, 123.2, 118.8, 46.0, 43.2, 31.4, 20.1, 13.8 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{ClN}_{4} \mathrm{OS}$ $m / z 349.1(M+H)$, found $m / z 348.7(M+H)$.
4.1.2.5. 6-Chloro-3-((2-((2-methoxyethyl)amino)thiazol-5-yl) methyl)quinazolin-4-(3H)-one (31). Obtained $31(24 \mathrm{mg}, 22 \%$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 8.55(\mathrm{~s}, 1 \mathrm{H}), 8.11$ (d, $J=2.04 \mathrm{~Hz}, 1 \mathrm{H}), 8.08$ (s, br, 1 H ), 7.86 (dd, $J=8.85,2.85 \mathrm{~Hz}, 1 \mathrm{H}), 7.71$ $(\mathrm{d}, J=8.85 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{~s}, 1 \mathrm{H}), 5.16(\mathrm{~s}, 2 \mathrm{H}), 3.43(\mathrm{t}, J=5.37 \mathrm{~Hz}, 2 \mathrm{H})$, $3.38-3.34(\mathrm{~m}, 2 \mathrm{H}), 3.23(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right)$ $\delta 171.7,160.2,146.9,145.7,136.6,135.1,133.6,129.4,126.3,123.0$, 118.0, 70.5, 59.0, 45.9, 43.2 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S} \mathrm{~m} / \mathrm{z} 351.1(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 350.7(\mathrm{M}+\mathrm{H})$.
4.1.2.6. 3-((2-Tert-butyl)thiazol-4-yl)methyl)-7-chloroquinazolin-4(3H)-one (32). Obtained $\mathbf{3 2}(2.0 \mathrm{mg}, 11 \%)$ as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.47(\mathrm{~s}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=8.56 \mathrm{~Hz}, 1 \mathrm{H}), 7.74(\mathrm{~d}$, $J=1.99 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{dd}, J=8.55,2.01 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~s}, 1 \mathrm{H}), 5.28(\mathrm{~s}$, $2 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 182.9,160.3$, 148.7, 148.7, 148.3, 140.8, 128.4, 128.1, 127.0, 120.6, 117.1, 45.2, 38.0, 31.0 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{ClN}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z} 334.1$ $(M+H)$, found $m / z 334.2(M+H)$.
4.1.2.7. 7-Chloro-3-((2-pentylthiazol-4-yl)methyl)quinazolin-4(3H)one (33). Obtained 33 ( $2.0 \mathrm{mg}, 11 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.20(\mathrm{~d}, J=8.57 \mathrm{~Hz}, 1 \mathrm{H}), 7.99(\mathrm{~s}, 1 \mathrm{H}), 7.76(\mathrm{~d}$, $J=2.00 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{dd}, J=8.57,2.00 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~s}, 1 \mathrm{H}), 3.83(\mathrm{~s}$, $2 \mathrm{H}), 2.64(\mathrm{t}, \mathrm{J}=7.55 \mathrm{~Hz}, 2 \mathrm{H}), 1.73-1.67(\mathrm{~m} \mathrm{2H}), 1.36-1.28(\mathrm{~m}, 4 \mathrm{H})$, $0.94-0.86(\mathrm{~m}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 199.0, 197.2, 160.1, 148.3, 147.9, 141.3, 128.5, 128.5, 126.9, 120.2, 53.8, 43.6, 36.5, 31.2, 25.3, 22.4, 14.0 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{OS} m / z 348.1(\mathrm{M}+\mathrm{H})$, found $m / z 348.0(\mathrm{M}+\mathrm{H})$.
4.1.2.8. 6-Chloro-3-((2-chlorothiazol-5-yl)methyl)quinazolin-4(3H)one (34). A solution of $\mathbf{3 6}(1.00 \mathrm{~g}, 5.50 \mathrm{mmol})$ in toluene ( 15 mL ) was treated with $\mathrm{KOH}(620 \mathrm{mg}, 11.10 \mathrm{mmol})$ and TBAI ( 193 mg , 0.55 mmol ) and heated at $70^{\circ} \mathrm{C}$ for 10 min . To this was added 2-chloro-5-(chloromethyl)thiazole ( $740 \mu \mathrm{~L}, 6.60 \mathrm{mmol}$ ) dropwise and stirred for 1 h at $70^{\circ} \mathrm{C}$ forming a dark brown solution. The reaction mixture was cooled to room temperature, diluted with water and extracted with ethyl acetate ( $4 \times 20 \mathrm{~mL}$ ). The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The organic layer was concentrated in vacuo and triturated with diethyl ether ( $3 \times 20 \mathrm{~mL}$ ), yielding $34\left(1.53 \mathrm{~g}, 89 \%\right.$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.62(\mathrm{~s}, 1 \mathrm{H}), 8.13(\mathrm{~d}, J=2.57 \mathrm{~Hz}, 1 \mathrm{H}), 7.89$ (dd, $J=8.89,2.57 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=8.61 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{~s}$, 2H) ppm; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 160.1,154.2,146.6,145.3$,
141.3, 135.4, 134.2, 134.0, 129.6, 126.3, 123.0, 43.1 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{OS} m / z 312.0(\mathrm{M}+\mathrm{H})$, found $m / z$ $311.8(\mathrm{M}+\mathrm{H})$.
4.1.2.9. 6-Chloro-3-((2-isopropylthiazol-5-yl)methyl)quinazolin$4(3 \mathrm{H})$-one (35). To a solution of intermediate $41(140 \mathrm{mg}$, 0.58 mmol ) in acetonitrile ( 5 mL ) was added acetic acid ( $500 \mu \mathrm{~L}$ ) and $40(55 \mathrm{mg}, 0.29 \mathrm{mmol})$ and stirred at $100^{\circ} \mathrm{C}$ for 2 h . The reaction mixture was cooled to room temperature, neutralised with saturated $\mathrm{NaHCO}_{3}$, and extracted with ethyl acetate ( $3 \times 10 \mathrm{~mL}$ ). The combined organic layers were dried over brine, concentrated in vacuo and purified through HPLC, yielding $35(2 \mathrm{mg}, 2 \%)$ as a white solid: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d $\mathrm{d}_{6}$ ) $\delta 8.63(\mathrm{~s}, 1 \mathrm{H}), 8.12$ (d, $J=2.51 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{dd}, J=8.71,2.53 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{~s}, 1 \mathrm{H}), 7.72(\mathrm{~d}$, $J=8.69 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{~s}, 2 \mathrm{H}), 3.20(\mathrm{sept}, J=6.87 \mathrm{~Hz}, 1 \mathrm{H}), 1.27(\mathrm{~d}$, $J=6.85 \mathrm{~Hz}, 6 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO) $\delta 178.2,165.8$, 147.9, 146.6, 142.2, 134.8, 131.7, 131.6, 129.7, 125.1, 122.7, 41.8, 32.6, 22.7 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{ClN}_{3} \mathrm{OS} \mathrm{m} / \mathrm{z} 320.1$ $(M+H)$, found $m / z 320.2(M+H)$.
4.1.2.10. 6-Chloroquinazolin-4(3H)-one (36). A solution of 2-amino5 -chlorobenzoic acid ( $4.99 \mathrm{~g}, 29.00 \mathrm{mmol}$ ) and formamide ( 20 mL ) was refluxed at $150{ }^{\circ} \mathrm{C}$ for 16 h , forming a brown precipitate. The reaction mixture was cooled to room temperature and 20 g ice was added. The mixture was left to stand for 1 h , then the precipitate filtered and concentrated in vacuo, yielding 36 ( $5.13 \mathrm{~g}, 97 \%$ ) as brown microcrystals: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right) \delta 12.44$ (br s, $1 \mathrm{H}), 8.12(\mathrm{~s}, 1 \mathrm{H}), 8.06$ (d, $J=2.40 \mathrm{~Hz}, 1 \mathrm{H}), 7.85(\mathrm{dd}, J=8.65,2.52 \mathrm{~Hz}$, 1 H ), 7.70 (d, $J=8.77 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}, 101 \mathrm{MHz}$ ) $\delta 160.3,152.6,146.5,134.9,131.5,130.0,125.3,124.4 \mathrm{ppm}$; LC-MS $\left(+\right.$ ESI) calculated for $\mathrm{C}_{8} \mathrm{H}_{5} \mathrm{ClN}_{2} \mathrm{O} \mathrm{m} / \mathrm{z} 181.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z}$ $181.2(\mathrm{M}+\mathrm{H})$.

### 4.1.2.11. 6-Chloro-3-(2-oxopropyl)quinazolin-4(3H)-one

(37).

To a solution of $36(1.60 \mathrm{~g}, 6.76 \mathrm{mmol})$ in anhydrous NMP was added NaH ( $650 \mathrm{mg}, 27.08 \mathrm{mmol}$ ) in small portions, forming a white precipitate. After stirring for 30 min the solution returned to a clear brown state and chloroacetone ( $2.4 \mathrm{~mL}, 29.28 \mathrm{mmol}$ ) was added dropwise forming a deep red solution. The solution was concentrated in vacuo and recrystallised in 3:1 hexane/ethyl acetate, yielding 37 ( $1.80 \mathrm{~g}, 86 \%$ ) as white needle crystals: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.25(\mathrm{~s}, 1 \mathrm{H}), 8.07(\mathrm{~d}, J=2.40 \mathrm{~Hz}, 1 \mathrm{H}), 7.89$ (dd, $J=8.73,2.48 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.74 (d, $J=8.73 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.98 (s, 2H), 2.25 ( $\mathrm{s}, 3 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ 199.7, 159.87, 146.7, 146.4, 135.0, 133.3, 129.3, 126.1, 122.9, 54.7, 27.5 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{~m} / \mathrm{z} 237.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 237.1$ $(\mathrm{M}+\mathrm{H})$.
4.1.2.12. 3-(3-Bromo-2-oxopropyl)-6-chloroquinazolin-4(3H)-one (38). To a solution of $37(460 \mathrm{mg}, 1.94 \mathrm{mmol})$ in acetic acid ( 10 mL ) was added bromine ( $125 \mu \mathrm{~L}, 62.42 \mathrm{mmol}$ ) dropwise, forming a deep red solution. The reaction mixture was refluxed at $65^{\circ} \mathrm{C}$ for 16 h , forming a red precipitate. The reaction mixture was concentrated in vacuo, diluted with water and neutralised with 2 M NaOH . The organic layer was extracted with ethyl acetate ( $3 \times 20 \mathrm{~mL}$ ) and washed with water $(2 \times 20 \mathrm{~mL})$ and brine $(1 \times 20 \mathrm{~mL})$. The organic layers were combined and concentrated in vacuo and purified through flash column chromatography (1:1 ethyl acetate/petroleum ether), yielding 38 ( $790 \mathrm{mg}, 66 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 8.27(\mathrm{~s}, 1 \mathrm{H}), 8.08(\mathrm{~d}, J=2.36 \mathrm{~Hz}, 1 \mathrm{H}), 7.90$ (dd, $J=8.69,2.40 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.75 (d, $J=8.73 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.13 (s, 2H), 4.57 (s, 2H) ppm; ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ 194.6, 160.0, 146,7, 146.1, 135.2, 133.6, 129.5, 126.1, 122.8, 52.4, 31.4 ppm; LC-MS (+ESI) calculated for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{BrClN}_{2} \mathrm{O}_{2} \mathrm{~m} / \mathrm{z} 314.9(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 315.1$ $(M+H)$.
4.1.2.13. 2-Isopropylthiazole-5-carbaldehyde oxime (39). Bromomalonaldehyde ( $150 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) and 2-methylpro panethioamide ( $103 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) were dissolved in EtOH ( 5 mL ) and stirred at room temperature for 4 h . Hydroxylamine hydrochloride ( $90 \mathrm{mg}, 1.30 \mathrm{mmol}$ ) and TEA ( 1 mL ) was added and stirred for a further 2 h at room temperature. The reaction mixture was concentrated in vacuo and purified through flash column chromatography in petroleum ether/ethyl acetate ( $4: 1$ to $3: 2$ ) yielding 39 ( $70 \mathrm{mg}, 34 \%$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$, $400 \mathrm{MHz}) \delta 12.29(\mathrm{~s}, 1 \mathrm{H}), 8.28(\mathrm{~s}, 1 \mathrm{H}), 8.02-8.00(\mathrm{~m}, 3 \mathrm{H}), 7.54-7.50$ ( $\mathrm{m}, 3 \mathrm{H}$ ) ppm;
4.1.2.14. (2-Isopropylthiazol-5-yl)methanamine (40). To a solution of 39 ( $59 \mathrm{mg}, 0.29 \mathrm{mmol}$ ) in EtOH ( 4 mL ) was added zinc powder ( $47 \mathrm{mg}, 0.72 \mathrm{mmol}$ ) and hydrochloric acid ( $130 \mu \mathrm{~L}, 13 \mathrm{~N}, 1.72 \mathrm{mmol}$ ). The reaction mixture was stirred at $60{ }^{\circ} \mathrm{C}$ for 2 h , then cooled to room temperature. The mixture was diluted with water and neutralised with saturated $\mathrm{NaHCO}_{3}$, then extracted with ethyl acetate $(3 \times 15 \mathrm{~mL})$. The combined organic layers were dried over brine and concentrated in vacuo yielding $\mathbf{4 0}$ ( $55 \mathrm{mg}, \mathbf{1 0 0 \%}$ ) as a yellow solid: ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right) \delta 7.92-7.90(\mathrm{~m}, 2 \mathrm{H}), 7.79(1 \mathrm{H}, \mathrm{s})$, 7.53-7.47 (m, 3H), 5.02 (s, br, 2H), 4.10 (s, 2H) ppm;
4.1.2.15. Methyl 5-chloro-2-(((dimethylamino)methylene)amino) benzoate (41). A solution of 2-amino-5-chlorobenzoic acid ( 50 mg , 0.29 mmol ) in acetonitrile ( 5 mL ) was treated with 1,1-dimethoxy$\mathrm{N}, \mathrm{N}$ - dimethylmethanamine and stirred at $100^{\circ} \mathrm{C}$ for 2 h . The intermediate 41 was confirmed through LC-MS and carried through without purification: LC-MS (+ESI) calculated for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{~m} / \mathrm{z}$ $241.1(M+H)$, found $m / z 240.7(M+H)$.
4.1.2.16. 7-Chloroquinazolin-4(3H)-one (42).

4-chloroanthranilic acid ( $1.00 \mathrm{~g}, 5.82 \mathrm{mmol}$ ) was refluxed in formamide ( 10 mL ) at $150^{\circ} \mathrm{C}$ for 16 h , forming a brown precipitate. The reaction mixture was cooled to room temperature and 20 g ice was added. The mixture was left to stand for 1 h , then the precipitate filtered and concentrated in vacuo, yielding 42 ( $850 \mathrm{mg}, 81 \%$ ) as a brown solid: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.\mathrm{d}_{6}, 400 \mathrm{MHz}\right) \delta 12.39(\mathrm{~s}, 1 \mathrm{H}), 8.14$ ( $\mathrm{s}, 1 \mathrm{H}$ ), 8.11 (d, $J=8.56 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=2.09 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{dd}$, $J=8.52,2.11 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 101 \mathrm{MHz}\right) \delta 160.1$, 149.9, 146.9, 138.9, 128.0, 127.0, 126.4, 121.5 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{8} \mathrm{H}_{5} \mathrm{ClN}_{2} \mathrm{O} \mathrm{m} / \mathrm{z} 181.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 181.2$ $(M+H)$.
4.1.2.17. 7-Chloro-3-(2-oxopropyl)quinazolin-4(3H)-one
(43).

Compound 42 ( $850 \mathrm{mg}, 4.70 \mathrm{mmol}$ ) was dissolved in NMP ( 5 mL ) and treated with NaH ( $380 \mathrm{mg}, 9.40 \mathrm{mmol}$ ). The reaction mixture was stirred at room temperature for 30 min , then chloroacetone ( $1.1 \mathrm{~mL}, 14.10 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred for 30 min at room temperature, the diluted with water ( 20 mL ), and extracted with ethyl acetate $(4 \times 15 \mathrm{~mL})$. The combined organic layers were dried over brine, concentrated in vacuo and recrystallised in 2:1 hexane/ethyl acetate, yielding compound 43 ( 827 mg , $74 \%$ ) as a silver crystal: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.21$ (d, $J=8.52 \mathrm{~Hz}, 1 \mathrm{H}), 7.88(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=2.02 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{dd}$, $J=8.57,1.98 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~s}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$, 101 MHz ) $\delta 199.8,160.4,149.3,147.5,141.0,128.4,127.4,120.5,77.2$, 76.8, 54.8, 27.7 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{~m} / \mathrm{z}$ $237.0(\mathrm{M}+\mathrm{H})$, found $m / z 237.1(\mathrm{M}+\mathrm{H})$.
4.1.2.18. 7-Chloro-3-(3-chloro-2-oxopropyl)quinazolin-4(3H)-one (44). Compound 43 ( $235 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) and NCS ( 133 mg , $1.00 \mathrm{mmol})$ were stirred in $\mathrm{DCM}(5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{SO}_{4}(1 \mathrm{~mL})$ at $40^{\circ} \mathrm{C}$ for 6 h . The reaction mixture was cooled to room temperature, neutralised with NaOH and extracted with ethyl acetate
$(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over brine, concentrated in vacuo and purified through flash column chromatography in petroleum ether/ethyl acetate (7:3 to 3:7) yielding 44 as a white solid ( $30 \mathrm{mg}, 11 \%$ ): ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , Acetic Acid- $\mathrm{d}_{4}$ ) $\delta 8.43(\mathrm{~s}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=8.59 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{~d}, J=2.05 \mathrm{~Hz}, 1 \mathrm{H}), 7.59$ (dd, $J=8.63,2.01 \mathrm{~Hz}, 1 \mathrm{H}), 5.23(\mathrm{~s}, 2 \mathrm{H}), 4.52(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(101 \mathrm{MHz}\right.$, Acetic Acid- $d_{4}$ ) $\delta 195.7,160.4,149.8,147.7,141.2,128.5$, 128.4, 125.5, 119.8, 52.7, 46.4 ppm ; LC-MS (+ESI) calculated for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~m} / \mathrm{z} 271.0(\mathrm{M}+\mathrm{H})$, found $\mathrm{m} / \mathrm{z} 270.6(\mathrm{M}+\mathrm{H})$.

### 4.2. PqsR bioreporter assay

### 4.2.1. Bacterial growth conditions

P. aeruginosa strains PAO1-L and PA14 were grown in lysogeny broth (LB). All strains were incubated at $37^{\circ} \mathrm{C}$ for 16 h , prior to use when $\mathrm{OD}_{600}$ was approximately 2.5. Strains PAO1-L and PA14 contained a chromosomal mCTX: $P_{p q s A}-l u x$ fusion such that light is emitted following activation of the $p q s A$ promoter. Consequently, a reduction in bioluminescence is indicative of pqs system inhibition. Strain BL21 (DE3) contained a pET28:PqsR plasmid, allowing for overexpression of the protein PqsR, and containing a kanamycin resistance cassette. To these cultures was added $100 \mu \mathrm{~g} / \mathrm{mL}$ kanamycin for selection.

### 4.2.2. Bioluminescence reporter gene $I C_{50}$ and $10 \mu \mathrm{M}$ spot test assays

Two 1.5 mL Eppendorf tubes were filled with $500 \mu \mathrm{~L}$ of LB broth. To one was added $3.16 \mu \mathrm{~L}$ of a 10 mM stock of a given inhibitor, and to the other $1 \mu \mathrm{~L}$, resulting in overall concentrations of $63.2 \mu \mathrm{M}$ and $20 \mu \mathrm{M}$ respectively. A one in 10 serial dilution of both stock solutions into four further Eppendorf tubes each containing $450 \mu \mathrm{~L}$ of LB broth gave 10 concentrations of the compound from $63.2 \mu \mathrm{M}$ to 2 nM . Using only the central 60 wells of a Grenier 96 well flat black plate, $100 \mu \mathrm{~L}$ of each concentration was added to three wells for results in triplicate.

Concurrently, an overnight culture of either PAO1-L or PA14 with the relevant antibiotic, grown to an $\mathrm{OD}_{600}$ of between 2.0 and 3.0, was diluted in LB broth to $\mathrm{OD}_{600}=0.02$. To each of the wells containing a given concentration of compound in LB, was added $100 \mu \mathrm{~L}$ of the diluted P. aeruginosa, giving overall concentrations of $31.6 \mu \mathrm{M}$ to 1 nM . The outer wells were filled with $200 \mu \mathrm{~L}$ of LB broth, as well as any unfilled inner wells, and the plate placed into a luminometer-spectrometer (Tecan GENios Pro), running a script at $37^{\circ} \mathrm{C}$ over 24 h , with a kinetic cycle measuring $\mathrm{OD}_{600}$ and luminescence every 30 min .

A peak in luminescence was observed always between 8 and 9 h , after which deterioration of the growth curve led to a drop in bioluminescence. As such, readouts of the $\mathrm{OD}_{600}$ and bioluminescence were taken from the highpoint of the luminescence readout, and the results plotted in Graphpad Prism 8, measuring log [concentration] against luminescence (Relative Light Units)/ $\mathrm{OD}_{600}$.

A similar methodology was applied for the $10 \mu \mathrm{M}$ spot test assays, but all compounds were diluted to $10 \mu \mathrm{M}$ from the 10 mM stock by adding $1 \mu \mathrm{~L}$ of the stock to $500 \mu \mathrm{~L}$ of LB broth, yielding a $20 \mu \mathrm{M}$ solution. As with the $\mathrm{IC}_{50}$ assays, $100 \mu \mathrm{~L}$ was then added to at least three wells and diluted twofold by the addition of $100 \mu \mathrm{~L}$ of P. aeruginosa at $\mathrm{OD}_{600}=0.02$. As all compounds were tested at $10 \mu \mathrm{M}$, no serial dilutions of the stock solutions were performed. Data was displayed as a column graph, with all results normalised against the negative control containing $0.1 \%$ DMSO (100\%).

### 4.2.3. Crystallography and protein determination

Protein was prepared as previously described by Ilangovan et al. [32] Crystals were grown in 24 well sitting and hanging drop Cryschem plates. Crystallisation reservoir consisted of 100 mM
trisodium citrate, 200 mM ammonium acetate and 2-methyl-2,4pentanediol (MPD). Citrate $\mathrm{pH}(5.5-6.5)$ and MPD concentration $(3 \%-10 \%)$ were varied along the plate's $X$ and $Y$ axes respectively, and optimal conditions for crystal soaking of $6,12,18$ and 19 were found to be a pH of 6.25 containing $6 \%$ MPD. Ligands were dissolved in DMSO or a multi-component solvent mixture and allowed to incubate in the crystallisation drop for 24 h .

Diffraction images were integrated with DIALS [47] (via Xia2 or DUI) and scaled in the CCP4 suite. Structures were solved by Molecular Replacement with PHASER [48] using 6Q7W as a search model. Model was refined with REFAC5 [49] with Jellybody restrains and ligand fitted into the Fo-Fc density map using COOT [50]. Ligand restrains were generated in AceDrg [51]. Omit and POLDER maps were generated using Phenix [52]. Finished structures have been deposited in the PDB as $6 Z 17$ (6), $6 \mathrm{ZO7}$ (12), $6 \mathrm{Z5K}$ (18) and $6 \mathrm{YZ3}$ (19). Ligand description (CIF), is available in supplementary information.

The construct pET28a:pqsR is available upon request.

### 4.2.4. Pyocyanin quantification

A 5 mL liquid culture of PAO1-L wildtype in LB medium was grown overnight for over 16 h with shaking at 200 rpm . Flasks containing 15 mL overnight PAO1-L culture diluted in LB to an $\mathrm{OD}_{600}$ of 0.05 , and either an inhibitor compound at $3 \times \mathrm{IC}_{50}$ value, or an equivalent volume of DMSO control was added and incubated for a further 16 h with shaking at 200 rpm .

An $\mathrm{OD}_{600}$ measurement was taken for each culture after 16 h for normalisation, before each sample was centrifuged for 10 min at $10,000 \times g$ and $24^{\circ} \mathrm{C}$. The supernatants were filtered through a $0.22 \mu \mathrm{~m}$ filter, and 7.5 mL collected. To each 7.5 mL filtered supernatant was added 4.5 mL chloroform, and each sample vortexed at 3000 rpm for 10 s . The samples were centrifuged at $10,000 \mathrm{rcf}$ for 10 min at $4{ }^{\circ} \mathrm{C}$, and the aqueous layer discarded. To 3 mL of each sample was added to 1.5 mL 0.2 M HCl and each vortexed for 10 s , followed by centrifugation at $10,000 \times g$ for 2 min at $4^{\circ} \mathrm{C}$. Measurement of the $\mathrm{OD}_{520}$ of the aqueous layer of each sample against a 0.2 M HCl control provided the raw data which could subsequently be normalised against the $\mathrm{OD}_{600}$ readings taken after incubation. Adapted from Essar et al. [53].

### 4.2.5. Cytotoxicity

A549 lung epithelial cells were cultured in Gibco's DMEM media containing $10 \%$ FBS and $1 \%$ penicillin-streptomycin at $37^{\circ} \mathrm{C}$ with $5 \%$ $\mathrm{CO}_{2}$. When $80 \%$ confluent, the medium was removed and the cells washed with $2 \times 10 \mathrm{~mL}$ PBS and $1 \times 7 \mathrm{~mL}$ trypsin-EDTA. To the flask was added 7 mL trypsin-EDTA and incubated for 5 min at $37{ }^{\circ} \mathrm{C}, 5 \%$ $\mathrm{CO}_{2} .7 \mathrm{~mL}$ FBS was added to quench trypsinization, and the cells centrifuged at 300 g for 5 min .

The supernatant was removed and the cell pellet washed with DMEM-F12 medium and centrifuged again at 300 g for 5 min . The pellet was then resuspended in 3 mL DMEM-F12 media and diluted further to allow loading of $100 \mu \mathrm{~L}$ of suspension containing 10,000 cells per well into 96 well plates. Cells were incubated for 16 h , at which point $100 \mu \mathrm{~L}$ of each compound was added to the relevant wells to give a total volume of $200 \mu \mathrm{~L}$. Cells were incubated for a further 16 h , and $20 \mu \mathrm{~L}$ of Alamar blue added to each well. After 5 h incubation, fluorescence was measured with excitation at 510 nm and emission at 590 nm . Values were then normalised against the untreated cell control. Adapted from O'Brien et al. [54].

## Author contributions

SG wrote the manuscript with supervision from MS, MC and PW with input from all co-authors. The junior authors performed the experiments and analysed the data with supervision from MS, MC
and PW. FS, MS, MC, PW and JE contributed to the study and experimental design. (Synthesis: SG, FS; Biological Assays: SG and FS; X-ray crystallography WR with supervision from JE.)

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.ejmech.2020.112778.

## Abbreviations

| AI | auto inducer |
| :--- | :--- |
| AQ | alkyl quinolone |
| DCM | dichloromethane |
| DMEM | Dulbecco's modified eagle medium |
| DMFDMA | N,N-dimethylformamide dimethyl acetate |
| DMSO | dimethyl sulfoxide |
| EDTA | ethylenediaminetetraacetic acid |
| EtOH | ethanol |
| FBS | foetal bovine serum |
| HHQ | 2-heptyl-4-hydroxyquinoline |
| HPLC | high performance liquid chromatography |
| IC 50 | half maximal inhibitory concentration |
| IPTG | isopropyl $\beta$-D-1-thiogalactopyranoside |
| LB | lysogeny broth |
| LBD | ligand-binding domain |
| LC-MS | liquid chromatography-mass spectrometry |
| LD 50 | half maximal lethal dose |
| MCCC | managed chemical compound collection |
| MPD | 2-methyl-2,4-pentanediol |
| NBS | N-bromosuccinimide |
| NCS | N-chlorosuccinimide |
| NHQ | 2-nonyl-4-hydroxyquinoline |
| NMP | $N$-methyl pyrrolidinone |
| NMR | nuclear magnetic resonance |
| OD 600 | optical density at 600 nm |
| PA | Pseudomonas aeruginosa |
| PQS | 2-heptyl-3-hydroxy-4(1H)-quinolone |
| QS | quorum sensing |
| QSSM | quorum sensing signal molecule |
| rcf | relative centrifugal force |
| SAR | structure-activity relationship |
| SD | standard deviation |
| SDS-PAGE | sodium dodecyl sulfate-polyacrylamide gel |
|  | electrophoresis |
| TBAI | tetrabutylammonium iodide |

triethylamine
tetrahydrofuran
tris(hydroxymethyl)aminomethane
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