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Running head: MMME Comparison of Four Modeling Approaches

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**METAL MIXTURE MODELING EVALUATION PROJECT: 2. COMPARISON OF
FOUR MODELING APPROACHES**

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ABSTRACT

As part of the Metal Mixture Modeling Evaluation (MMME) project, models were developed by the National Institute of Advanced Industrial Science and Technology (Japan), the U.S. Geological Survey (USA), HDR|HydroQual, Inc. (USA), and the Centre for Ecology and Hydrology (UK) to address the effects of metal mixtures on biological responses of aquatic organisms. A comparison of the 4 models, as they were presented at the MMME Workshop in Brussels, Belgium (May 2012), is provided herein. Overall, the models were found to be similar in structure (free ion activities computed by WHAM; specific or non-specific binding of metals/cations in or on the organism; specification of metal potency factors and/or toxicity response functions to relate metal accumulation to biological response). Major differences in modeling approaches are attributed to various modeling assumptions (e.g., single versus multiple types of binding site on the organism) and specific calibration strategies that affected the selection of model parameters. The models provided a reasonable description of additive (or nearly additive) toxicity for a number of individual toxicity test results. Less-than-additive toxicity was more difficult to describe with the available models. Because of limitations in the available datasets and the strong inter-relationships among the model parameters (log K_M values, potency factors, toxicity response parameters), further evaluation of specific model assumptions and calibration strategies is needed.

Key words (<5 words): Biotic ligand model, Concentration addition, Metal bioavailability, Metal toxicity, WHAM-F_{TOX}

23 Note to the editor and reviewers: This is one of 11 manuscripts under consideration for an ET&C
24 Special Section on Metal Mixtures. The Section includes an introduction, a technical
25 background, a comparison of multiple modeling approaches, a lessons-learned manuscript, and
26 seven manuscripts on specific modeling and interpretation approaches. While each manuscript
27 should be able to stand alone, the individual manuscripts are interrelated and cross-reference
28 each other. If another cross-referenced, submitted manuscript is essential to complete the review
29 of the present manuscript, please request the other manuscript from the Corresponding Guest
30 Editor, copying the handling editor. The Corresponding Guest Editor for the series is Eric Van
31 Genderen (evangenderen@zinc.org). Any unpublished material provided to assist your review
32 must also be treated in confidence.

33

34

INTRODUCTION

35 In regulatory applications, metal-mixture toxicity has generally been modeled by Toxic Unit
36 (TU) or other additive approaches that are based on water-exposure concentrations [1]. Data
37 reviews [2,3] have shown that additive approaches based on dissolved-metal concentrations are
38 not always sufficient in predicting mixture toxicity. Rather, metal-mixture toxicity tests have
39 shown a wide range of organism responses with no clear patterns in additive and non-additive
40 behavior. As part of an effort to address metal-mixture toxicity, several quantitative models have
41 been developed to evaluate responses of aquatic organisms to metal mixtures and ultimately to
42 provide a priori predictions of toxicity. Two modeling frameworks have been considered for this
43 purpose: the Biotic Ligand Model (BLM) (as first presented by Di Toro et al. [4]) and WHAM-
44 F_{TOX} [5].

45 In the BLM framework, metal bioavailability is evaluated by considering competitive
46 interactions of metals and cations for binding to dissolved organic matter (DOM) and inorganic
47 ligands (e.g., HCO_3^- , CO_3^{2-} , Cl^-) using the Windermere Humic Aqueous Model (WHAM).
48 Competitive binding of metals and cations is also assumed to occur at binding sites on or in
49 biological organisms, which are referred to as the “biotic ligand(s).” The accumulation of metal
50 on the biotic ligand is then correlated to the toxic response of the organism (e.g., using a logit
51 response function). The BLM has been used by various investigators over the past decade to
52 develop predictive models for acute and chronic toxicity in single-metal exposures [6,7]. More
53 recently, several BLMs have been developed or revised for metal mixtures [8,9,10,11]. In these
54 models, metals have either been assumed to exhibit similar joint action (with toxicity expressed
55 in terms of concentration addition of metal accumulation on the biotic ligand) [8,9,10], or
56 independent joint action (with toxicity expressed in terms of a multiplicative function of the
57 responses to the individual metals) [11].

58 WHAM- F_{TOX} [5] was specifically developed to address the effects of metal mixtures on
59 aquatic organisms. This approach uses WHAM to evaluate competitive interactions of metals
60 and cations on DOM and inorganic ligands. In contrast to the BLM, WHAM- F_{TOX} does not
61 explicitly consider competitive binding of metals and cations to a biotic ligand. Rather, the
62 model assumes that non-specific accumulation of metabolically-active metals by the organism is
63 proportional to metal concentrations predicted to accumulate on humic acid (HA, when exposed
64 to the same exposure water (as calculated by WHAM). Accumulated metal is related to toxicity
65 using the F_{TOX} function, which is obtained by multiplying the calculated humic-bound metal and
66 proton concentrations (mmol g^{-1}) by cation-specific potency factors and then summing the results
67 over all cations. The resulting F_{TOX} value is correlated to the toxic response of the organism

68 using a linear-threshold response function, in which the effect thresholds depend only on the
69 organism (i.e., independent of water chemistry).

70 The Metal Mixture Modeling Evaluation (MMME) project was initiated to assess the current
71 capabilities of BLM- and WHAM-F_{TOX}-type models in predicting metal-mixture toxicity and to
72 promote the continued development of various modeling approaches [12]. As part of that
73 initiative, models were developed/refined and tested by researchers from (1) the National
74 Institute of Advanced Industrial Science and Technology (AIST) in Ibaraki, Japan; (2) the U.S.
75 Geological Survey (USGS) in Seattle, Washington and Boise, Idaho, USA; (3)
76 HDR|HydroQual, Inc. (HDR) in East Syracuse, New York, USA; and (4) the Centre for Ecology
77 and Hydrology (CEH) of the Natural Environment Research Council in Lancaster, UK. The
78 purpose of this paper is to summarize and compare the 4 models, as they were presented at the
79 MMME Workshop in Brussels, Belgium in May 2012. Revised versions of the initial AIST,
80 USGS, HDR and CEH models are presented in this issue [8,9,10,11,13]. Because the published
81 version of the USGS model differed substantially from the 2012 version evaluated in this article
82 and because the 2012 version of the USGS model is not otherwise available, it is included for
83 reference in Supporting Information File SI-2.

84

85 **MODEL DESCRIPTIONS**

86 The 4 models presented at the MMME Workshop were based on previous developments of
87 the BLM and WHAM-F_{TOX}. Although the models were based on differing frameworks, they
88 shared many similarities in their overall structure. These similarities included:

- 89 • A chemical speciation calculation to compute the free ion activities of metals and major
90 cations based on competitive binding to DOM and inorganic ligands using various
91 versions of WHAM.
- 92 • An evaluation of competing binding of metals and major cations to 1 or more binding
93 sites on an organism using conventional competitive equilibrium chemistry (by either
94 considering biotic ligand(s) or using metal binding to HA as a surrogate for non-specific
95 metal accumulation by the organism).
- 96 • A correlation of accumulated metal to toxicity using potency factors and/or toxicity-
97 response functions.

98 However, the 4 models differed in the details of their formulations and in calibration procedures
99 that were used by the different modeling groups in fitting the MMME project datasets. Those
100 datasets are listed in [Table 1](#); formulations of the 4 models are summarized in [Table 2](#), and
101 further details are provided below.

102 *AIST model*

103 The AIST model followed the BLM framework and considered a single biotic ligand as the
104 binding site for all metals on the organism [8,14]. Free ion activities of metals and other cations
105 were calculated using WHAM VII [15]. For this calculation, DOM was assumed to be 100%
106 fulvic acid (FA) and was set directly equal to the reported dissolved organic carbon (DOC)
107 concentration. Thus, conversion from DOC (mg L^{-1}) to the WHAM input for FA (g L^{-1}) was FA
108 = $0.001 \times \text{DOC}$. Carbonate species were not included in the calculations. WHAM-calculated
109 free ion activities were then used in computing competitive binding of metal(s) and major cations
110 on the biotic ligand. Initial estimates of binding constants ($\log K_M$ values) for metals and major

111 cations to the biotic ligand were obtained from previous studies (e.g., for trout [16,17]; for
 112 *Daphnia magna* [18,19]; for *D. pulex* [20,21]).

113 Toxicity was expressed as a function of the fractional coverage of accumulated metal on the
 114 biotic ligand. Following the premise that toxicity is not caused by a metal-induced response but
 115 rather by the role of the metal in blocking Ca uptake sites [14], all metals were assumed to elicit
 116 equally potent toxicological responses when bound to the biotic ligand. According to this
 117 assumption, metals exhibit similar joint action and metal-mixture toxicity can be described by a
 118 concentration-additive approach based on the accumulation of total metal on the biotic ligand.
 119 Response functions considered in the AIST model included a 2-parameter logit (mortality)
 120 function for rainbow trout (*Oncorhynchus mykiss*), cutthroat trout (*O. clarkii*) and daphnids, and
 121 a 2-parameter linear (growth reduction) function for a freshwater alga (*Pseudokirchneriella*
 122 *subcapitata*). For example, the 2-parameter logit function is given as:

$$123 \quad R = \frac{1}{1 + e^{-(a+b\theta_M)}} \quad (1)$$

124 where R is the biological response (e.g., fractional mortality or growth reduction), a and b are the
 125 logit parameters, and θ_M is the fractional coverage of accumulated metals on the biotic ligand.

126 The AIST model was calibrated using 4 of the 6 MMME calibration datasets (Table 1). For
 127 each dataset, the model was fit to observed mortality (or growth-reduction) responses by
 128 adjusting the logit response-function parameters. For *D. magna* and algae, log K_M values were
 129 also adjusted, with a different set of log K_M values used for each organism (Supporting
 130 Information File SI-1, Tables S1 and S2). Metal-mixture toxicity was predicted using the
 131 calibrated log K_M values and the response-function parameters derived from single-metal
 132 exposure studies.

133 *USGS model*

134 The 2012 version of the USGS model also followed the BLM framework and considered a
135 single biotic ligand as the binding site on the organism (see Supporting Information File SI-2).
136 Free ion activities of metals and cations were calculated using WHAM VII and the following
137 assumptions: DOM was specified as 2 times the reported DOC concentration, 65% of the DOM
138 was assumed to be active, and DOM was considered to be composed of 10% HA and 90% FA.
139 Thus, conversions from DOC (mg L^{-1}) to WHAM inputs for HA and FA (in g L^{-1}) were: HA =
140 $2 \times 0.65 \times 0.1 \times 0.001 \times \text{DOC}$ and FA = $2 \times 0.65 \times 0.9 \times 0.001 \times \text{DOC}$. In addition, carbonate species
141 were included in the calculations (by specifying pH and alkalinity). For field-collected water
142 samples, free ion activities of Al^{3+} and Fe^{3+} were assumed to be in equilibrium with amorphous
143 iron and aluminum hydroxides using solubility relationships [22,23]. WHAM-calculated free
144 ion activities were then used in computing competitive binding of metals and major cations on a
145 single-site biotic ligand. Binding constants ($\log K_M$ values) for metals and major cations to the
146 biotic ligand were determined from a re-evaluation of data from single-metal toxicity studies on
147 rainbow and cutthroat trout (Supporting Information File SI-1, Table S3). The $\log K_M$ values
148 remained constant and did not differ among biological species.

149 Toxicity was expressed as a function of the fractional coverage of accumulated metal on the
150 biotic ligand. In contrast to the AIST model, metals were assumed to have different potencies
151 when bound to the biotic ligand. This effect was included in the model by incorporating a *TOX*
152 function to account for apparent differences in toxicities of the various metals. In this approach,
153 the *TOX* function is conceptually similar to the F_{TOX} function [5]. However, 1 major difference
154 is that the *TOX* function is related to the fractional coverage of metal on the biotic ligand,
155 whereas F_{TOX} is expressed as a function of non-specific accumulation of metal on HA (in mmole
156 g^{-1}).

157 Toxic response was then determined in 2 steps. First, a potency factor (α_i) was defined to
 158 account for the relative toxicity of different metals when bound to the biotic ligand. This factor
 159 was multiplied by the fraction coverage of metal on the biotic ligand (θ_i) to calculate the toxic
 160 potency of a specified metal, as defined by the *TOX* function:

$$161 \quad TOX_i = \alpha_i \theta_i \quad (2)$$

162 The model was extended to metal mixtures using a concentration-addition type of approach.
 163 However, because each metal was considered to exhibit a different potency when bound to the
 164 biotic ligand, calculations were based on the summation of *TOX*_{*i*} values.

$$165 \quad TOX = \sum_{i=1}^n TOX_i = \sum_{i=1}^n \alpha_i \theta_i \quad (3)$$

166 where *n* is the number of metals in the mixture. Second, a 3-parameter logit function was used to
 167 calculate biological response as a function of *TOX*.

$$168 \quad R = \frac{1}{(1 + e^{-(\beta_1 + \beta_2 \log TOX)})^{\beta_3}} \quad (4)$$

169 where *R* is the biological response (e.g., fractional mortality or growth reduction), and β_1 , β_2 , and
 170 β_3 are the logit parameters. The model was calibrated using 5 of the 6 MMME calibration
 171 datasets (Table 1). For each dataset, the model was fit to observed mortality (or growth-
 172 reduction) responses from single-metal and metal-mixture exposures by adjusting potency
 173 factors (α_i) and the 3 logit parameters (β_1 , β_2 , β_3 ; Supporting Information File SI-1, Table S4).
 174 The potency factors were assumed to be dependent only on the metal, and the logit parameters
 175 were considered to be organism-specific in the initial calibration of the model. However, it was
 176 necessary to consider organism-specific potency factors in fitting datasets for *P. subcapitata*
 177 growth.

178 *HDR model*

179 The HDR model extended the BLM approach by considering a separate biotic ligand for each
180 metal (i.e., a Cd-specific, Cu-specific, Pb-specific, and Zn-specific biotic ligand). All metals
181 could compete for all binding sites, but only 1 metal was considered to be toxicologically active
182 at a given biotic ligand. The remainder of the model followed BLM calculations for single-metal
183 exposures [24]. Within the HDR model, metal and cation binding to DOM and inorganic ligands
184 were calculated based on WHAM V [25] and the following assumptions: DOM was specified as
185 2 times the reported DOC concentrations, 100% of the DOM was assumed to be active, and
186 DOM was considered to be composed of 10% HA and 90% FA. Thus, conversions from DOC
187 (mg L^{-1}) to WHAM inputs for HA and FA (in g L^{-1}) were: $\text{HA} = 2 \times 0.1 \times 0.001 \times \text{DOC}$ and $\text{FA} =$
188 $2 \times 0.9 \times 0.001 \times \text{DOC}$. In addition, carbonate species were included in the calculations (by
189 specifying pH and alkalinity). Because HDR did not explicitly consider the effects of Al or Fe
190 binding, Al^{3+} and Fe^{3+} were not included in the model.

191 In addition to metal and cation binding to DOM and inorganic ligands, the HDR model also
192 included simultaneous calculations for metal and cation binding to each of the metal-specific
193 biotic ligands. For each biotic ligand, $\log K_M$ values for the toxicologically-active metal and
194 competing cations were obtained from previously-calibrated, single-metal BLMs [24]. For the
195 remaining metals that were not toxicologically-active at a given biotic ligand but could compete
196 for binding to the site, $\log K_M$ values were initially set equal to their $\log K_M$ values on their
197 toxicologically-active biotic ligands. For example, the $\log K_M$ value for Cd on the Cu-specific
198 biotic ligand was set equal to the $\log K_M$ value for Cd on the Cd-specific biotic ligand. However,
199 adjustments in some of the $\log K_M$ values were made during model calibration. The final $\log K_M$
200 values for the HDR model are presented in Supporting Information File SI-1, Table S5.

201 Toxic response at each of the biotic ligands was correlated to the concentration of metal ‘*i*’
 202 on its toxicologically-active biotic ligand using a 2-parameter logit function:

$$203 \quad R_i = \frac{1}{1 + e^{-(a_i + b_i \log v_i)}} \quad (5)$$

204 where R_i represents the biological response (e.g., mortality) due to metal ‘*i*’; a_i and b_i are the
 205 metal-specific logit parameters; and v_i is the concentration of metal ‘*i*’ on its toxicologically-
 206 active biotic ligand (in nmol g⁻¹). For metal mixtures, the overall response (R) was determined
 207 by assuming independent joint action and expressing toxicity in terms of a multiplicative
 208 function of individual metal responses as:

$$209 \quad R = 1 - \prod_{i=1}^n (1 - R_i) \quad (6)$$

210 where n is the number of metals in the mixture. This approach is also referred to as response
 211 addition (see [1] for further discussion).

212 The HDR model was calibrated using 3 of the 6 MMME calibration datasets (Table 1). The
 213 model was calibrated separately for multiple data series within a given dataset by adjusting logit
 214 intercepts (a_i) and slopes (b_i) to fit observed mortalities for single-metal exposures [11]. These
 215 analyses ultimately provided a global fit of the logit slope and a distribution of logit intercepts
 216 that were used to quantify the unexplained variance or uncertainty associated with the single-
 217 metal exposure data (Supporting Information File SI-1, Table S6). Metal-mixture toxicity was
 218 predicted using the log K_M values and calibrated logit parameters from single-metal exposures.
 219 The HDR model also considered uncertainty in single-metal toxicity predictions to generate
 220 response envelopes for metal-mixture exposures (see [11] for details). Log K_M values for Cu and
 221 Zn on the Cd-specific biotic ligand were subsequently adjusted to provide a better calibration of
 222 the model to the *D. magna* mortality data (“Index 4”).

223 *CEH model*

224 The CEH model was based on WHAM-F_{TOX} [5,26]. In this calculation, WHAM VI [27] was
225 used to be consistent with previous applications of WHAM-F_{TOX} [5,28]. Dissolved organic
226 matter was specified as 2 times the reported DOC concentration, 65% of the DOM was assumed
227 to be FA, and the remaining 35% of the DOM was considered to be inert with respect to
228 metal/cation binding. Thus, the conversion from DOC (mg L⁻¹) to WHAM inputs for FA (in g L⁻¹)
229 was: $FA = 2 \times 0.65 \times 0.001 \times DOC$. Carbonate species were included in the calculations (by
230 assuming that total carbonate concentrations were equal to the reported alkalinity). For field-
231 collected water samples, free ion activities of Al³⁺ and Fe³⁺ were assumed to be in equilibrium
232 with amorphous iron and aluminum hydroxides using solubility relationships [22,29]. A small
233 concentration of WHAM HA (e.g., 10⁻¹⁰ g/L) was also included in the calculation. The resulting
234 concentrations of metals and protons on the HA were then used in toxicity calculations. In these
235 calculations, metal concentrations on HA included metals that were specifically-bound to HA
236 functional groups as well as metals that were nonspecifically-bound by electrostatic interactions
237 and held in close proximity to the HA (i.e., in the Donnan Layer). However, only protons that
238 were specifically-bound to HA functional groups were included in toxicity predictions. Because
239 the CEH model considered a distribution of binding sites in WHAM HA (see [27]),
240 representative composite binding constants were calculated for illustrative purposes by taking the
241 weighted-averages of the K_M values (see Supporting Information File SI-1, Table S7 for details).
242 (Note that log K_M values were also modified for electrostatic corrections in WHAM-F_{TOX}.)

243 Toxic response was determined by assuming that concentrations of metabolically-active
244 metals and protons on or in the organism were proportional to their predicted concentrations on
245 WHAM HA in the same exposure water. Toxicity was then determined in 2 steps. First, a

246 potency factor was defined to relate the amounts of accumulated metals and protons to toxic
 247 effect using the F_{TOX} function:

$$248 \quad F_{TOX} = \sum_{i=1}^{n+1} \alpha_i v_i \quad (7)$$

249 where α_i is the relative potency factor, v_i is the concentration of metal and protons on humic acid
 250 (in mmol g⁻¹), n is the number of metals in the mixture, and $n+1$ is considered to account for
 251 proton toxicity. Second, a 2 parameter linear response function was defined to relate toxic
 252 response to the F_{TOX} function

$$253 \quad \begin{aligned} R &= 0 && \text{for } F_{TOX} < F_{TOX-LT} \\ R &= \frac{F_{TOX} - F_{TOX-LT}}{F_{TOX-UT} - F_{TOX-LT}} && \text{for } F_{TOX-LT} \leq F_{TOX} \leq F_{TOX-UT} \\ R &= 1 && \text{for } F_{TOX} > F_{TOX-UT} \end{aligned} \quad (8)$$

254 where F_{TOX-LT} represents the lower threshold for toxicity and F_{TOX-UT} represents the threshold for
 255 the maximum toxic response.

256 The model was calibrated using all 6 MMME calibration datasets (Table 1). For each
 257 dataset, the model was fit to observed mortality (or growth-reduction) responses from single-
 258 metal and metal-mixture exposures by adjusting relative potency factors (α_i) and the linear
 259 response parameters (F_{TOX-LT} , F_{TOX-UT} ; Supporting Information File SI-1, Table S8). The relative
 260 potency factor for H⁺ (α_H) was set equal to 1.0 in the calibration, effectively normalizing the
 261 potency of metals relative to that of H⁺. Adjustments in F_{TOX-LT} and F_{TOX-UT} values were also
 262 examined in evaluations of the *D. magna* mortality data (“Index 4”).

263

264

METHODS

265 In the present study, model performance was examined by first re-computing results for the 4
266 models using specifications described in the ‘Model Descriptions’ section. This step served as
267 an independent check of results presented by the 4 modeling groups and also provided detailed
268 model outputs that were subsequently used in model-model comparisons. Three of the larger
269 calibration datasets were considered for this purpose, including:(i) “Index 8”, which consisted of
270 114 test results for *P. subcapitata* growth in field-collected water samples spiked with Cd, Cu,
271 Ni, and Zn (see Supporting Information File SI-3); (ii) “Index 4”, which consisted of 561 test
272 results for *D. magna* mortality in reconstituted laboratory water spiked with Cd, Cu, and Zn [30];
273 and (iii) “Index 6”, which consisted of 369 test results for cutthroat and rainbow trout mortality
274 in field-collected water samples spiked with Cd, Pb, and Zn [31]. Two additional datasets were
275 considered for model validation: (i) “Index V-1”, which consisted of 309 test results for *D.*
276 *magna* mortality in reconstituted laboratory water spiked with Cd and Zn [30]; and (ii) “Index V-
277 3”, which consisted of 96 test results for rainbow trout survival in reconstituted laboratory water
278 spiked with Cd, Cu, Zn [32].

279 Calculations for the 4 models were performed as follows: For the AIST model, free ion
280 activities of metals and cations were calculated using WHAM VII [15,33]. The remainder of the
281 calculation was performed in Excel, with fractional coverage of metal on the biotic ligand
282 computed from the WHAM-calculated free ion activities and AIST-chosen log K_M values.
283 Biological responses (e.g., mortality, growth reduction) were then determined as a function of
284 the fractional coverage of metal(s) using the AIST response functions (Eqn. 1; Supporting
285 Information File SI-1, Tables S1 and S2).

286 A similar procedure was followed for evaluation of the USGS model. Free ion activities of
287 metals and cations were calculated using WHAM VII. The fractional coverage of metal on the

288 biotic ligand, the TOX function, and toxicity were computed in Excel using the WHAM-
289 calculated free ion activities and the USGS-chosen log K_M values, potency factors (α_i), and
290 response-function parameters ($\beta_1, \beta_2, \beta_3$) (Eqns. 2-4; Supporting Information File SI-1, Tables
291 S3 and S4).

292 For evaluation of the HDR model, free ion activities were calculated using the TICKET
293 model [34] with the WHAM V database. Accumulation of metals and cations on each of the
294 biotic ligand sites (in nmol g(wet)^{-1}) were computed in Excel using the WHAM-calculated free
295 ion activities and HDR-chosen log K_M values (Supporting Information File SI-1, Table S5).
296 Toxic responses at each biotic ligand were determined using the HDR response-function
297 parameters (a_i, b_i) (Eqns. 5, 6; Supporting Information File SI-1, Table S6).

298 Finally, for evaluation of the CEH model, free ion activities and concentrations of metals and
299 cations on HA were computed using WHAM VI [27]. Because the concentration of specifically-
300 bound protons on HA was not included in the model output of the commercially-available
301 version of WHAM VI, concentrations of specifically-bound protons were estimated as the total
302 number of proton binding sites minus the surface charge (in equivalents per gram) minus 2 times
303 the summation of specifically-bound metals and cations. In this calculation, metals and divalent
304 cations are assumed to primarily occupy bidentate and tridentate binding sites on the HA.
305 Toxicity was then computed in Excel using the WHAM-calculated concentrations on HA and the
306 CEH-determined potency factors (α_i) and response-function parameters (F_{TOX-LT}, F_{TOX-UT}) (Eqns.
307 7, 8; Supporting Information File SI-1, Tables S7 and S8).

308 Results for the 4 models were then plotted in comparable formats with mortality (or growth
309 reduction) on the vertical axis and fractional coverage on the biotic ligand (θ_M) for the AIST
310 model, TOX for the USGS model, and F_{TOX} for the CEH model on the horizontal axis. This

311 allowed metals for a large number of single-metal and metal-mixture exposure tests to be plotted
 312 and visually compared in a concise and convenient format. Unfortunately, this graphical format
 313 is not directly applicable to the HDR model because (i) toxicity response functions for individual
 314 metals were not considered to have common logit slopes, and (ii) metal-mixture toxicity was
 315 described by independent joint action using a response-additive approach. However, a
 316 reasonable comparison was provided by converting the HDR model-predicted responses (R) to
 317 equivalent TOX (TOX_{equiv}) values by rearranging Equation 5:

$$318 \quad TOX_{equiv} = e^{-\left(\frac{2.303a}{b}\right)} \cdot \left[\frac{R}{1-R} \right]^{\left(\frac{2.303}{b}\right)} \quad (9)$$

319 In this equation, R is the fractional mortality (or growth reduction) for single-metal or metal-
 320 mixture exposures, a and b are the log-logit intercept and slope, and 2.303 is included because
 321 HDR used a mixed ln-log function in Equation 5 to describe toxicity. For subsequent
 322 calculations, TOX_{equiv} values were computed using response parameters (a , b) for Zn. This
 323 effectively normalized results for other metals to the toxicity of Zn. The TOX_{equiv} function is
 324 most appropriate for mortality (or growth reduction) responses near 50%. For responses near 0%
 325 or 100%, the TOX_{equiv} function is only a crude approximation for metals with log-logit response
 326 slopes that are different than Zn. Graphical comparisons of model coefficients and of model
 327 results for individual metal-mixture exposure tests were also prepared and analyzed.

328

329 RESULTS AND DISCUSSION

330 Observed mortality for *D. magna* and the corresponding AIST, USGS, HDR and CEH
 331 model-calibrated response curves are shown for single-metal (Figure 1A,C,E,G) and metal-
 332 mixture exposures (Figure 1B,D,F,H) following the graphical formats described above. Similar

333 comparisons for rainbow trout mortality are presented in **Figure 2**. In most cases, model-
334 calibrated response curves described the central tendency of observed mortalities for both single-
335 metal and metal-mixture exposures. However, scatter in the observed data around the model-
336 calibrated response curves varied from model to model, with the USGS *D. magna* results (Figure
337 1C,D) and the AIST rainbow trout results (Figure 2A,B) showing the largest variations of
338 observed mortality about the model-calibrated response curves. Observed mortality data was
339 also relatively widely scattered around the HDR model-calibrated response curves, particularly
340 for observations near 0% and 100% mortality (Figure 1E,F; Figure 2E,F). However, these
341 differences in part can be attributed to the TOX_{equiv} approximation (Eqn. 9) in which all metals
342 were assumed to have log-logit slopes similar to that of Zn. Finally, the observed mortality data
343 appeared most closely aligned to the CEH model-calibrated response curves for both *D. magna*
344 (Figure 1G,H) and rainbow trout (Figure 2G,H).

345 The graphical comparisons in Figures 1 and 2 provide an overview of how well the 4 models
346 were calibrated to observed mortality for 2 of the larger calibration datasets. Additional model-
347 model comparisons were made by examining model fits to individual-metal results. For
348 example, USGS, HDR and CEH model results for *D. magna* mortality in a Cd-only toxicity test
349 (Index 4, Cu-Cd #7-1, with 12.6 $\mu\text{g/L}$ dissolved Cd) are presented in **Figure 3A**. The 3 models,
350 which computed solution chemistry using 3 different versions of WHAM, predicted
351 approximately 35% of total dissolved Cd was free Cd. However, the fractional coverage of Cd
352 on the biotic ligand (or the HA surrogate in the CEH model) varied from 0.24% in the CEH
353 model to 11% in the USGS model. Despite these large differences, the associated calibration of
354 potency factors and/or response-function parameters resulted in similar predictions of mortality
355 by the 3 models. This finding demonstrates the strong inter-relationship of $\log K_M$ values,

356 potency factors and response-function parameters, and underscores the latitude that exists in
357 calibrating model parameters with toxicity datasets that include only measures of total dissolved
358 metal concentrations, water chemistry (pH, major ions, alkalinity, DOC) and a select
359 toxicological endpoint (e.g., mortality, growth reduction).

360 A similar example is given in Figure 3B for AIST, HDR, and CEH model results for *D.*
361 *magna* mortality in a Cu-only toxicity test (Index 4, Cd-Cu #5-2, with 83.8 µg/L of dissolved
362 Cu). In this case, model predictions of free Cu varied from 0.042% to 2.6% of the total dissolved
363 Cu concentration. These differences were due to the version of WHAM that was used in the
364 calculations and to assumptions for DOC composition and carbonate chemistry that were
365 employed by the 3 modeling groups. Differences in model predictions for free Cu were reflected
366 in differences for Cu accumulation on the biotic ligand (or the HA surrogate in the CEH model).
367 Despite these large differences, calibration of potency factors and/or response-function
368 parameters again resulted in similar predictions of mortality by the 3 models. This finding
369 demonstrates that the WHAM calculation can also have a large effect on the final calibration of
370 log K_M , potency factors, and response function parameters, but again the inter-relationships of
371 the multiple calibration parameters allows for compensation of those differences to produce
372 similar overall predictions among the different modeling approaches.

373 Based on results presented in Figure 3, it is difficult to evaluate calibration strategies in a
374 simple step-by-step procedure. Rather, a more holistic view of the calibration process is needed
375 (see comparison of model calibration parameters in [Table 3](#)). In all 4 models, log K_M values for
376 the initial calibration were fixed based on previous studies and were not considered as adjustable
377 parameters. Based on the remaining model parameters, the initial calibration of the AIST model
378 appeared to be most constrained, with response parameters allowed to be adjusted only as a

379 function of the organism (Table 3). This was followed by the USGS model which allowed
380 potency factors (α_i) to be adjusted as a function of only the metal, and response parameters (β_1 ,
381 β_2 , β_3) to be adjusted as a function of only the organism in its initial calibration. Additional
382 flexibility was considered in the initial calibration of the HDR model, which allowed toxicity
383 response parameters (a_i , b_i) to be adjusted as a function of both metal and organism. The CEH
384 model provided similar flexibility by allowing potency factors (α_i) to be adjusted as a function of
385 metal and organism, and by allowing small adjustments in response parameters (F_{TOX-LT} , F_{TOX-UT})
386 as a function of only the organism.

387 The use of a more constrained or a more flexible calibration strategy had a significant
388 effect on the ability of the 4 models to describe mortality (or growth reduction) data (see Figures
389 1 and 2). To illustrate this point, USGS and CEH model results for growth reductions of *P.*
390 *subcapitata* at pH 6 are given in Figure 4. As shown in Figure 4A, the growth in single-metal
391 and metal-mixture exposures were poorly described by the initial calibration of the USGS model,
392 which was based on potency factors (α_i) that were determined from global fits to the MMME
393 calibration datasets. As shown by the USGS modeling group (Supporting Information File SI-2),
394 specification of a separate set of potency factors (α_i) for *P. subcapitata* at pH 6 produced a much
395 closer correspondence of the model-calculated response curve and observed growth reductions
396 (Figure 4B). By comparison, the CEH model still appears to provide a better description of the
397 observed growth reductions (Figure 4C). The reason is in part due to the inclusion of
398 specifically-bound protons in the CEH toxicity evaluation. For example, in the calculated
399 growth reductions for *P. subcapitata* at pH 6, proton toxicity accounted for 1.7 ± 0.06 of the
400 computed F_{TOX} value (vertical gray bar in Figure 4C). This served to compress the effects of
401 metals to the right of the vertical gray bar. Subtracting the proton contribution from F_{TOX} and re-

402 plotting the model response curve and the observed growth reduction provided a different picture
403 of the variability that may be associated with metal accumulation in the organism (Figure 4D).
404 Therefore, excluding the extra factor of proton toxicity, which was included in the WHAM-F_{TOX}
405 calibration, would likely result in some added variability of observed responses around the
406 model-calculated response curve.

407 Next, the 4 models were further evaluated by comparing their final selection of model
408 parameters. Log K_M comparisons for Cd, Pb, and Zn were based on AIST values for rainbow
409 trout, USGS values that were previously determined from cutthroat trout and rainbow trout data,
410 HDR values for the Zn-specific biotic ligand, and average log K_M values for the distribution of
411 binding sites in the CEH model (Supporting Information File SI-1, Tables S1, S3, S5 and S7).
412 From a chemical perspective, the log K_M values for the HDR and CEH models followed an
413 expected increase in metal binding based on affinities of metals to oxygen donor groups on
414 organic acids (Cd < Zn < Pb; Figure 5A). In contrast, the AIST and USGS models had larger
415 binding constants for Cd that were similar to previously-reported log K_M values for Cd (e.g.,
416 [35,36,37]). This stronger binding of Cd to biological ligands was attributed to active Ca
417 transport and ionic mimicry in fish gills [35]. Another possible explanation for larger Cd binding
418 constants may be that Cd is binding to sulfur (and not oxygen) donor groups in the organism.

419 Toxicity parameters in the 4 models were also compared by combining potency factors (α_i)
420 and response function parameters into a single measure of the lethal accumulation at 50%
421 mortality (LA50) (see Supporting Information File SI-1, Tables S2, S4, S6, and S8). For the
422 comparison, LA50 values were expressed in terms of percent coverage on the biotic ligand or on
423 the surrogate HA binding sites for the WHAM-F_{TOX} model. The resulting LA50 values for
424 rainbow trout exposed to Cd, Zn, and Pb ranged from 2 to 3% of the binding sites in the AIST

425 and USGS models (Figure 5B). In contrast, the LA50 values for the HDR and CEH models
426 varied more (0.01% for Cd, 7% for Zn, and 2.9% for Pb for the HDR model; 0.05% for Cd, 9.8%
427 for Zn, and 22% for Pb for the CEH model). For comparison, experimentally-derived LA50
428 values for Cd, Zn and Pb in rainbow trout studies have ranged from 10% to 64% coverage of
429 strong binding sites on the gill [36,37,38,39]. These values represent the higher end of the
430 model-calculated LA50 values and are not supportive of the very low LA50 values for Cd in the
431 HDR model. However, there are some questions regarding the appropriateness of comparing
432 experimentally-derived LA50 values (which are based on estimates for the density of strong
433 binding sites that have been reported to vary as a function of water chemistry and the specific
434 metal being examined) and model-calculated LA50 values (which are generally based on a
435 binding site density that is considered to be constant across all water chemistries and metals).
436 An alternative interpretation of experimentally-derived LA50 values is provided by considering
437 the relative values of measured accumulations on a nmole per gram of fish gill basis. For
438 example, 24-h LA50 measurements for rainbow trout have been reported as 1.1 nmole g⁻¹ (ww)
439 for Cd and 32.8 nmole g⁻¹ (ww) for Pb [36]. This represents a difference of a factor of 30 in the
440 Cd and Pb accumulations on the gill that would elicit a 50% mortality response and is consistent
441 with the lower LA50 values for Cd that were computed by the HDR and CEH models. However,
442 a more appropriate comparison for the CEH model-calculated LA50 values would be body-
443 burden measurements for the various metals.

444 A final check on model calibration was performed by examining individual series of mixture
445 toxicity test results for *D. magna* mortality (Index 4). For the Cu-Zn test series #5-4, increases in
446 mortality were observed for *D. magna* exposed to a constant Cu concentration of 100 µg/L and
447 increasing Zn concentrations (Figure 6A). This behavior is similar to an additive response curve

448 (see Figure 1 in Meyer et al. [1]). Model-calculated response curves for the AIST, HDR and
449 CEH models were consistent with the observed trend, with the HDR model corresponding most
450 closely to the observed data. Differences in the AIST, HDR and CEH model-calculated curves
451 for the Cu-Zn mixture can be attributed to the calibration of the models and not to the differences
452 in their formulations.

453 The Cd-Cu test series #20-3 showed very different behavior (Figure 6B), with observed
454 mortalities for *D. magna* exposed to a constant Cd concentration of 19.5 µg/L and increasing Cu
455 concentrations following a less-than-additive (Case 2) response curve (see Figure 1 in Meyer et
456 al. [1]). The AIST model (which is based on concentration addition) did not predict the observed
457 decrease in mortality as Cu concentrations increased. However, both the HDR model (which is
458 based on independent joint action) and the CEH model (which is based on a F_{TOX} -additive
459 approach) predicted a decrease and then an increase in *D. magna* mortality as the Cu
460 concentration increased. Additional adjustments in the both HDR and CEH model calibrations
461 were required to fit the observed response. For the HDR model, this consisted of increasing the
462 log K_M value for Cu binding to the Cd-biotic ligand sites by 4 log units. For the CEH model,
463 global values of F_{TOX-LT} and F_{TOX-UT} that were reported for *D. magna* were adjusted by
464 optimizing the F_{TOX-LT} and F_{TOX-UT} to the test series #20 data. Although the reported adjustments
465 in F_{TOX-LT} and F_{TOX-UT} were not large (global fit: 1.88 and 2.95; test series #20 fit: 2.61 and 3.18),
466 model responses to the single metal and metal-mixture exposure tests were sensitive to the
467 changes (see Supporting Information File SI-1, Figures S25 and S26). Therefore, questions still
468 remain about how to formulate and calibrate models to reproduce observed responses exhibiting
469 less-than-additive (Case 2) behavior.

470 In addition to model calibration evaluations, the calibrated models were used in a blind
471 prediction of mortality for 2 validation studies (Index V-1; Index V-3). Comparisons of
472 observed mortality and model-calculated response curves for the *D. magna* validation study
473 (Index V-1) are presented in **Figure 7**. Model-data comparisons for single-metal exposures
474 (Figure 7A,C,E,G) were comparable to the *D. magna* calibration results presented in Figure 1.
475 Model-calculated response curves for metal mixtures tended to over-predict mortality by factors
476 of 1-2 on the θ_M scale for the AIST model (Figure 7B), factors of 2-5 on the *TOX* scale for the
477 USGS model (Figure 7D), approximately a factor of 2 on the *TOX_{equiv}* scale for the HDR model
478 (Figure 7F), and approximately a factor of 2 on the *F_{TOX}* scale for the CEH model (Figure 7H).

479 Similar results for the rainbow trout validation study (Index V-3) are shown in Supporting
480 Information File SI-1 (Figures S8, S15, S22 and S31). Model-data comparisons tended to show
481 more variability for single-metal exposures. For mixtures, model-calculated response curves
482 tended to over-predict trout mortality by a factor of 3 to 4 on the *TOX* scale for the USGS model.
483 Model-calculated response curves for the HDR and CEH models were in closer agreement to
484 observed mortality. The AIST model was not considered in the rainbow trout validation test
485 because a rainbow trout log K_M value for Cu was not provided for the AIST model.

486 The overall results of the present study highlighted similarities and differences in 4 models
487 that were developed to describe the effects of single-metal and metal-mixture exposures on
488 biological response (e.g., mortality, growth reduction). The 4 models were calibrated to
489 individual datasets that contained metal-exposure concentrations, water chemistry and biological
490 response data. Because measurements of metal accumulation on a representative biological site
491 were not available, independent calibration of log K_M , potency factors and response-function
492 parameters were not possible. Despite these limitations, calibration of models to single-metal

493 exposure data often provided a reasonable basis for predicting metal-mixture toxicity. This was
494 particularly true for metal-mixtures exhibiting additive (or near additive) behavior. The ability
495 of the models to reproduce less-than-additive behavior posed a greater challenge. This less-than-
496 additive toxicity was predicted by the HDR model (which considered independent joint action,
497 but required substantial adjustment of some $\log K_M$ values to describe the observed behavior)
498 and by the CEH model (which considered F_{TOX} addition, but required adjustment of F_{TOX}
499 parameters to individual datasets).

500 These findings indicate that competitive interactions among metals add a level of complexity
501 to toxicity evaluations that will in all likelihood only be appreciated through continued model
502 development. The application of more complex geochemical models (with multiple biotic ligand
503 sites or distributions of $\log K_M$ binding sites) may be needed for this purpose. This has led to
504 revisions in the 4 modeling approaches, particularly for the AIST and USGS models that were
505 considered in the present study. Revised versions of the AIST, USGS, HDR and CEH models
506 are described in various papers in this issue (see [8,9,10,11,13]). Model calibration remains a
507 key issue for the various modeling approaches. Therefore, a further evaluation of specific model
508 assumptions and calibration strategies that were used by the 4 modeling groups is considered in
509 the following paper [40].

510

511 **SUPPORTING INFORMATION**

512 File SI-1: Modeling parameters and additional analyses (Tables S1 to S8, and Fig. S1 to S31).

513 File SI-2: USGS model description (2012 version).

514 File SI-3: Description of “Index 7” and “Index 8” data sets.

515

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521

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Table 1. Datasets used in the Metal Mixture Modeling Evaluation project^a

Index	Species	Metal mixture / water type	Endpoint	Number of exposures ^b	Source	Datasets used for model calibration ^c			
						AIST	USGS	HDR	CEH
1	<i>Hyalella azteca</i> <i>Lampsilis siliquoidea</i>	Cd-Cu-Ni-Pb-Zn / Porewater	28-d survival	2/60 2/38	[41]		✓	✓	✓
4	<i>Daphnia magna</i>	Cd-Cu, Cu-Zn / Lab	48-h survival	387/174	[30]	✓	✓	✓	✓
6	<i>Oncorhynchus mykiss</i>	Cd-Pb-Zn / Field	96-h survival	298/71	[31]	✓	✓	✓	✓
7	<i>Pseudokirchneriella subcapitata</i>	Field mixture / Field	72-h growth	7/28	Present Study ^d	✓	✓		✓
8	<i>P. subcapitata</i>	Cd-Cu-Ni-Zn / Field	72-h growth	102/12	Present Study ^d	✓	✓		✓
9	<i>Lactuca sativa</i>	Ag-Cu, Cu-Zn / Hydroponic	4-d root growth	36/202	[42]				✓
V-1	<i>D. magna</i>	Cd-Zn / Lab	48-h survival	132/177	[30]				
V-2	<i>D. magna</i>	Cd-Cu-Zn Lab	48-h survival	3/12	[30]				
V-3	<i>O. mykiss</i>	Cd-Cu-Zn Lab	96-h survival	72/24	[32]				

^a See [12] for detailed descriptions.

^b Single-metal or reference exposures / mixture exposures

^c AIST = National Institute of Advanced Industrial Science and Technology, Japan; USGS = U.S. Geological Survey, USA; HDR = HDR|HydroQual, Inc., USA; CEH = Centre for Ecology and Hydrology, UK

^d See Supplemental Information, File SI-3

Table 2. Summary of formulations for the AIST, USGS, HDR, and CEH metal-mixture-toxicity models^a

	AIST	USGS	HDR	CEH
Solution chemistry	WHAM VII	WHAM VII	WHAM V	WHAM VI
Metal/cation binding to organisms	Competitive binding of metals/cations to a single BL site	Competitive binding of metals/cations to a single BL site	Competitive binding of metals/cations to multiple BL sites	Non-specific accumulation of metals / cations at a distribution of binding sites
Toxicity	Function of fractional coverage of metal on BL	Function of potency and coverage of each metal on BL ($TOX_i = \alpha_i \theta_i$) ^b	Function of potency and concentration of each metal on its toxicologically-relevant BL	Function of potency and concentration of protons and each metal on WHAM humic acid, assumed proportional to their binding on or in the organism ($F_{TOX} = \alpha_i \nu_i$) ^c
Toxic response	2-parameter logit (or linear) response function	3-parameter logit response function	2-parameter logit response function	2-parameter linear-threshold response function
Mixture response	Concentration additive	TOX additive ^b	Independent action	F_{TOX} additive ^c

^a AIST = National Institute of Advanced Industrial Science and Technology, Japan; USGS = U.S. Geological Survey, USA; HDR = HDR|HydroQual, Inc., USA; CEH = Centre for Ecology and Hydrology, UK; BL = biotic ligand; WHAM V, VI, and VII = versions of Windermere Humic Aqueous Model.

^b TOX = toxicity-response function in USGS model; α_i = potency factor for metal i ; θ_i = proportion of BL sites occupied by metal i (# of sites occupied / # total sites).

^c F_{TOX} = toxicity-response function in CEH model; α_i = potency factor for proton or metal i ; ν_i = concentration of BL sites occupied by protons or metal i (mmol g⁻¹ humic acid).

Table 3. Summary of parameters used in the AIST, USGS, HDR, and CEH metal-mixture-toxicity models^a

	AIST	USGS	HDR	CEH
Binding constants ($\log K_M$)	f(metal, organism) ^b	f(metal) ^c	f(metal) ^b	f(metal) ^c
Metal potency factors (α_i)	n/a	f(metal) ^d	n/a	f(metal, organism)
Proton potency factor (α_H)	n/a	n/a	n/a	f(organism)
Response parameters ($\beta_1, \beta_2, \beta_3, F_{TOX-LT},$ $F_{TOX-UT},$ etc.)	f(organism)	f(organism)	f(metal, organism)	f(organism)

^a AIST = National Institute of Advanced Industrial Science and Technology, Japan; USGS = U.S. Geological Survey, USA; HDR = HDR|HydroQual, Inc., USA; CEH = Centre for Ecology and Hydrology, UK.

^b $\log K_M$ values for the AIST and HDR models were taken from previously-calibrated, single-metal biotic ligand models. Additional adjustments of $\log K_M$ values were made during their studies.

^c $\log K_M$ values were held constant in the USGS and CEH models and were determined as follows:

USGS: from a re-evaluation of single-metal toxicity data for cutthroat and rainbow trout;
CEH: from previous calibration for WHAM VI (a version of the Windermere Humic Aqueous Model), using WHAM humic acid as a surrogate for non-specific accumulation of protons and metabolically-active metals by the organism.

^d In the initial calibration of the USGS model, metal potency factors were considered to be a function of only the metal. A separate set of potency factors was required for the final calibration of the algal dataset.

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642

643 **Figure Captions**

644 Figure 1. Model-data comparisons for the effects of single-metal and metal-mixture
645 exposures on mortality of *Daphnia magna* in laboratory water with spiked metals. Mortality
646 is shown as a function of fractional coverage of metal on the biotic ligand (θ_M) for the AIST
647 model (panels **A, B**); TOX for the USGS model (panels **C, D**); TOX_{equiv} for the HDR model
648 (panels **E, F**); and F_{TOX} for the CEH model (panels **G, H**). Observed responses (open
649 symbols) are compared to the model-calculated response curve for mortality (continuous
650 line). Dashed lines represent plus/minus a factor of 2 in the concentration at which a model-
651 calculated response occurs. See text for calibration procedures used in each model. Data
652 from Meyer et al. [30].

653 Figure 2. Model-data comparisons for the effects of single-metal and metal-mixture
654 exposures on mortality of rainbow trout (*Oncorhynchus mykiss*) in field-collected water with
655 spiked metals. Mortality is shown as a function of fractional coverage of metal on the biotic
656 ligand (θ_M) for the AIST model (panels **A, B**); TOX for the USGS model (panels **C, D**);
657 TOX_{equiv} for the HDR model (panels **E, F**); and F_{TOX} for the CEH model (panels **G, H**).
658 Observed responses (open symbols) are compared to the model-calculated response curve for
659 mortality (continuous line). Dashed lines represent plus/minus a factor of 2 in the
660 concentration at which a model-calculated response occurs. See text for calibration
661 procedures used in each model. Data from Mebane et al. [31].

662 Figure 3. Comparison of model-calculated responses for free metal as a percentage of the
663 total dissolved metal, percent accumulated metal on the biotic ligand (or on the WHAM
664 humic acid surrogate for generalized binding on or in organisms in the CEH model), and
665 percent mortality of *Daphnia magna* for: (A) 12.6 $\mu\text{g/L}$ total dissolved Cd (from Index 4, Cu-

666 Cd #7-1); and **(B)** 83.8 µg/L total dissolved Cu (from Index 4, Cd-Cu #5-2). Data from
667 Meyer et al. [30]; indexes are described in Van Genderen et al. [12].

668 Figure 4. Model-data comparisons for the effects of single-metal and metal-mixture
669 exposures on growth reduction of *Pseudokirchneriella subcapitata* at pH 6.0 in field-
670 collected water with spiked metals. Growth reduction is shown for **(A)** USGS model with
671 model parameters from a global calibration to all datasets; **(B)** USGS model with model
672 parameters from a calibration to *P. subcapitata* pH 6.0 data; **(C)** CEH model; and **(D)** CEH
673 model with the baseline effect of H⁺ removed from F_{TOX} . Observed responses for single-
674 metal (open symbols) and metal-mixture exposures (closed symbols) are compared to the
675 model-calculated response curve for mortality (continuous line). Dashed lines represent
676 plus/minus a factor of 2 in the concentration at which a model-calculated response occurs.
677 See text for calibration procedures used in each model

678 Figure 5. Comparison of **(A)** metal binding affinity to binding sites on or in the organism
679 ($\log K_M$); and **(B)** lethal accumulations for 50% mortality (LA50) for rainbow trout
680 (*Oncorhynchus mykiss*) based on the AIST, USGS, HDR and CEH model calibrations.
681 Average $\log K_M$ values are given for the CEH model for illustrative purposes. Metals are
682 arranged according to their expected affinity to bind to oxygen donor groups [43].

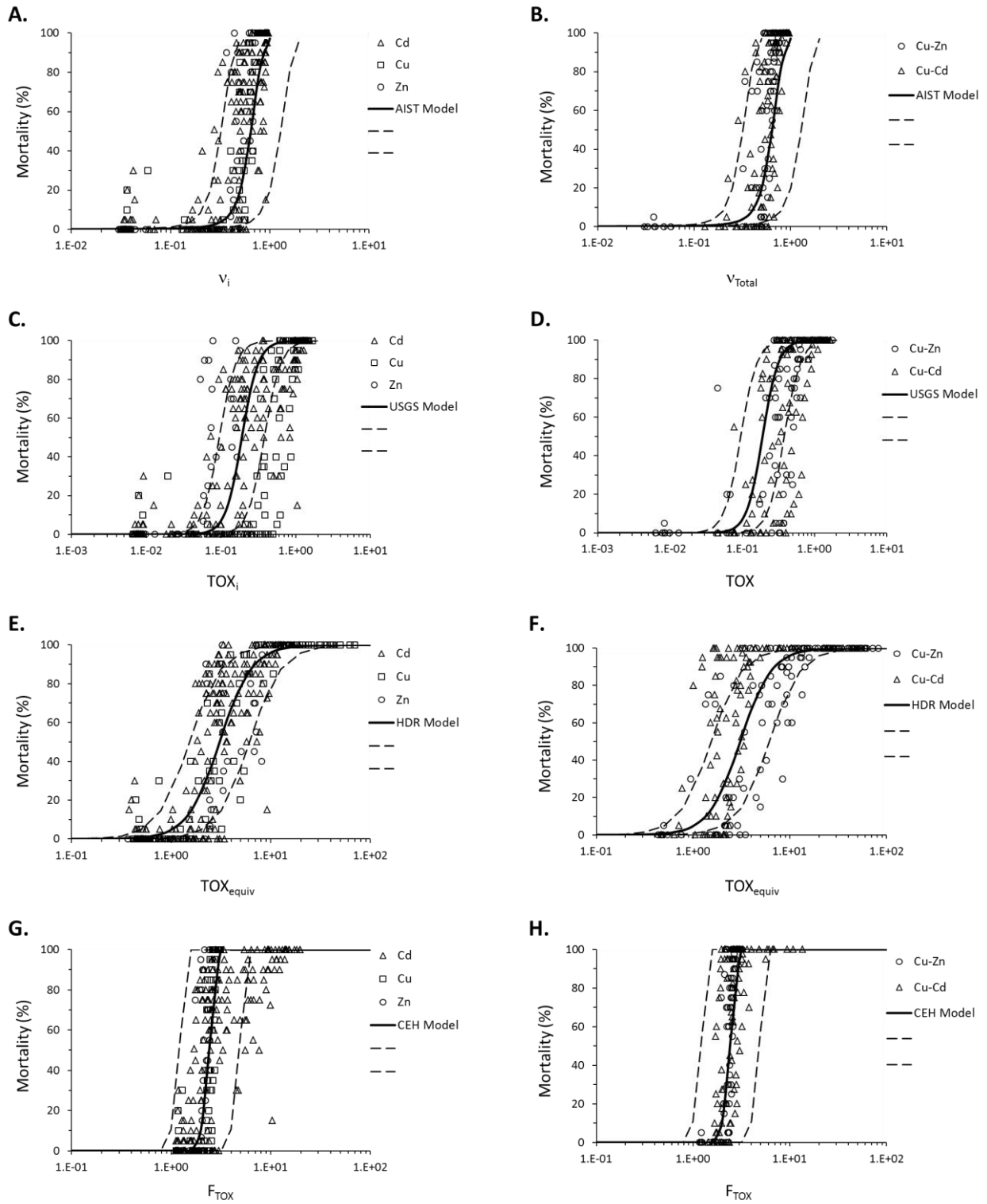
683 Figure 6. AIST, HDR and CEH model-data comparisons for the effects of metals on
684 mortality of *Daphnia magna* in laboratory water with spiked metal concentrations. Mortality
685 is shown: **(A)** as a function of total dissolved Zn with fixed total dissolved Cu concentrations
686 (test series #5-4); and **(B)** as a function of total dissolved Cu with fixed total dissolved Cd
687 concentrations (test series #20-3). CEH model-calculated response curves are based on
688 F_{TOX-LT} and F_{TOX-UT} values that were optimized to the individual test series ($F_{TOX-LT} = 2.30$,

689 $F_{TOX-UT} = 2.71$ for test series #5; $F_{TOX-LT} = 2.61$, $F_{TOX-UT} = 3.18$ for test series #20). Observed
690 responses for metal-mixture exposures (closed symbols) are compared to the model-
691 calculated response curve for mortality (dotted, short dashed and continuous lines). Data
692 from Meyer et al. [30]; indexes are described in Van Genderen et al. [12].

693 Figure 7. Model validation for the effect of single-metal and metal-mixture exposures on
694 mortality of *Daphnia magna* in laboratory water with spiked metals. Mortality is shown as a
695 function of fractional coverage of metal on the biotic ligand (θ_M) for the AIST model (panels
696 **A, B**); TOX for the USGS model (panels **C, D**); TOX_{equiv} for the HDR model (panels **E, F**);
697 and F_{TOX} for the CEH model (panels **G, H**). Observed responses (open symbols) are
698 compared to the model-calculated response curve for mortality (continuous line). Dashed
699 lines represent plus/minus a factor of 2 in the concentration at which a model-calculated
700 response occurs. Data from Meyer et al. [30].

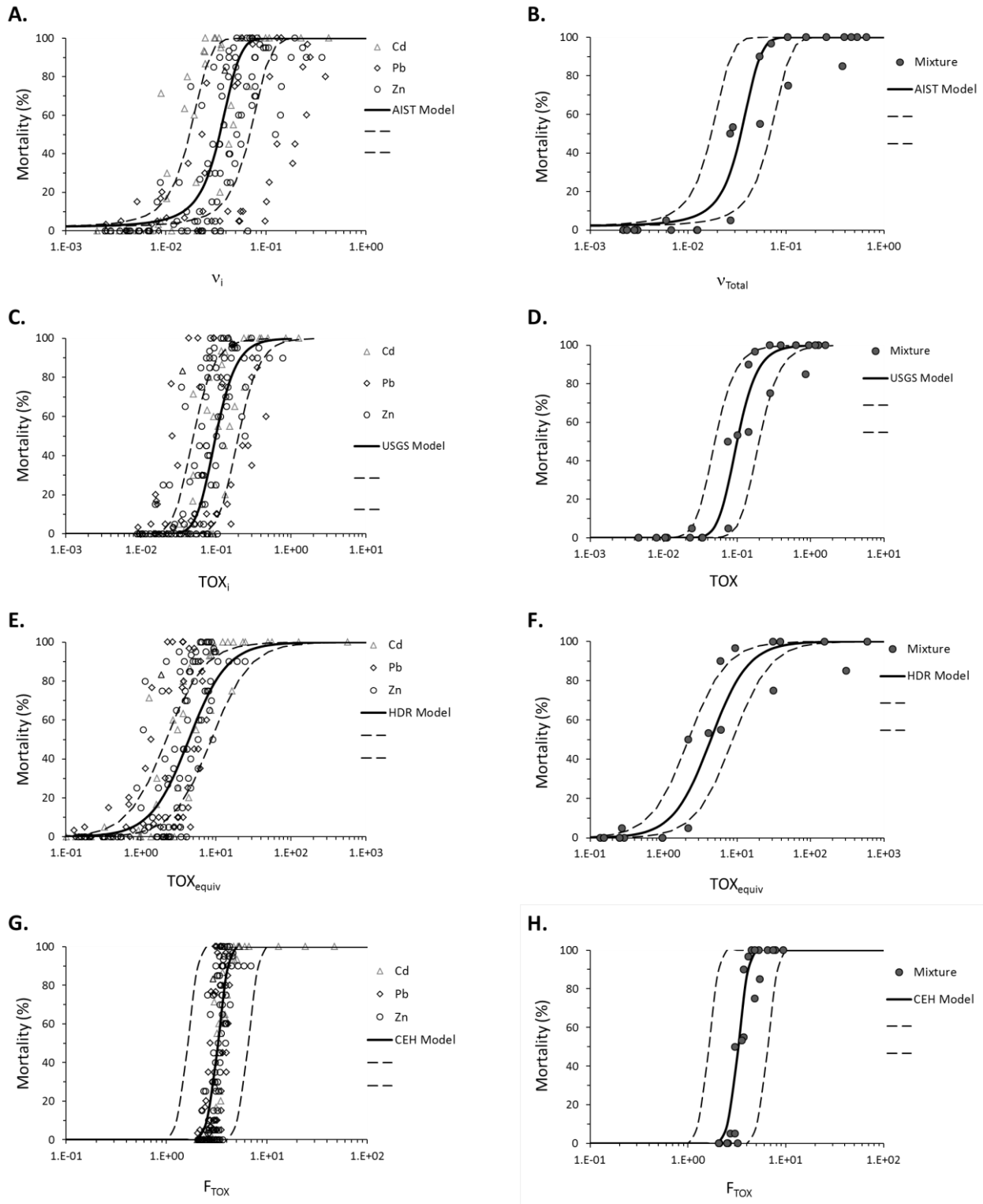
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704 Figure 1.



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706 Figure 2.

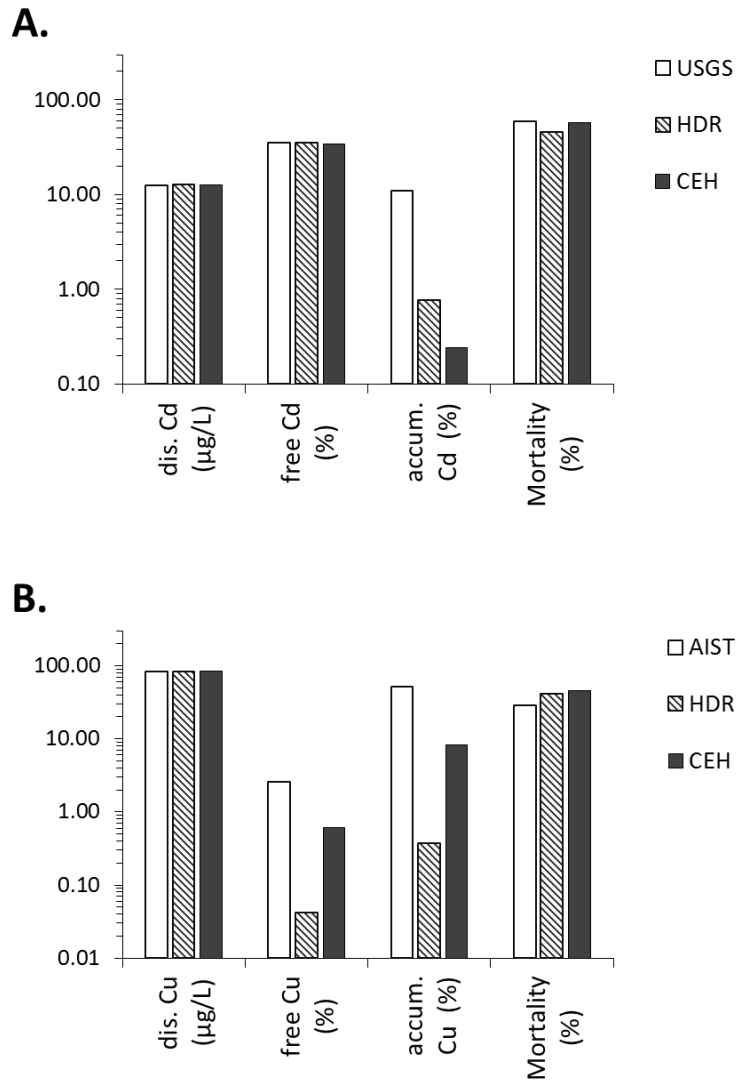


Figure 3.

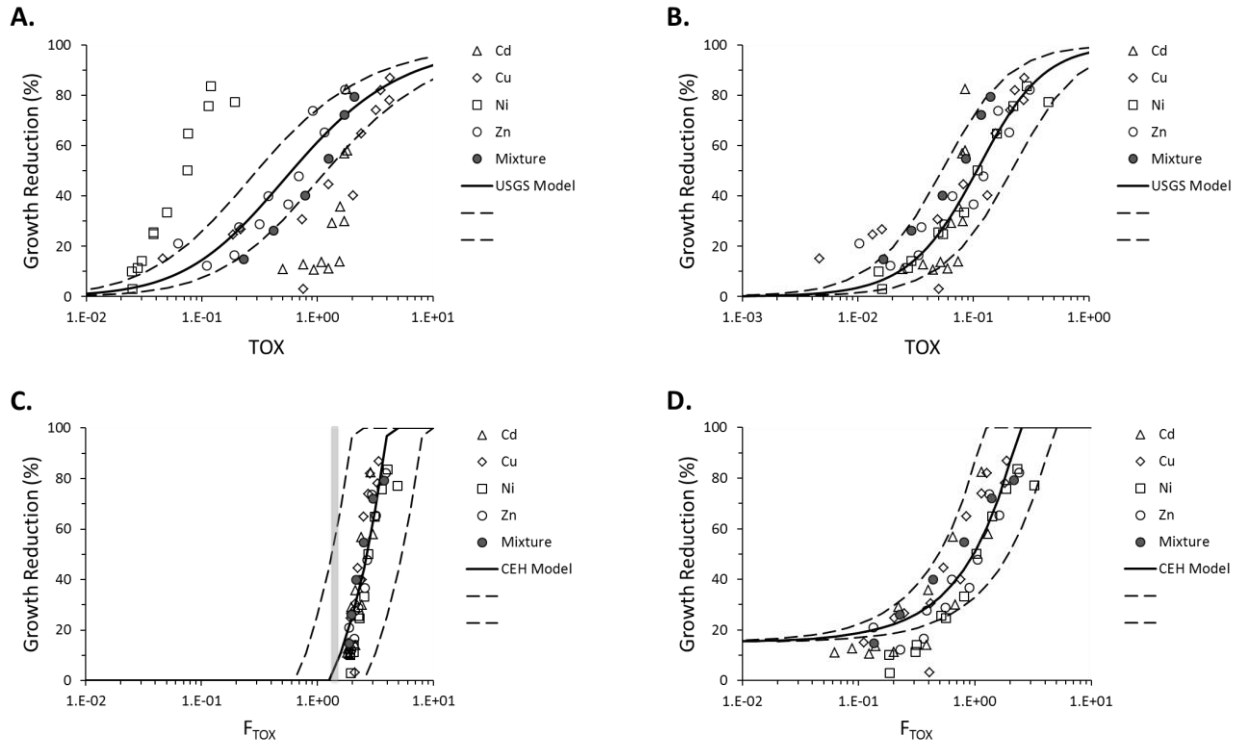


Figure 4.

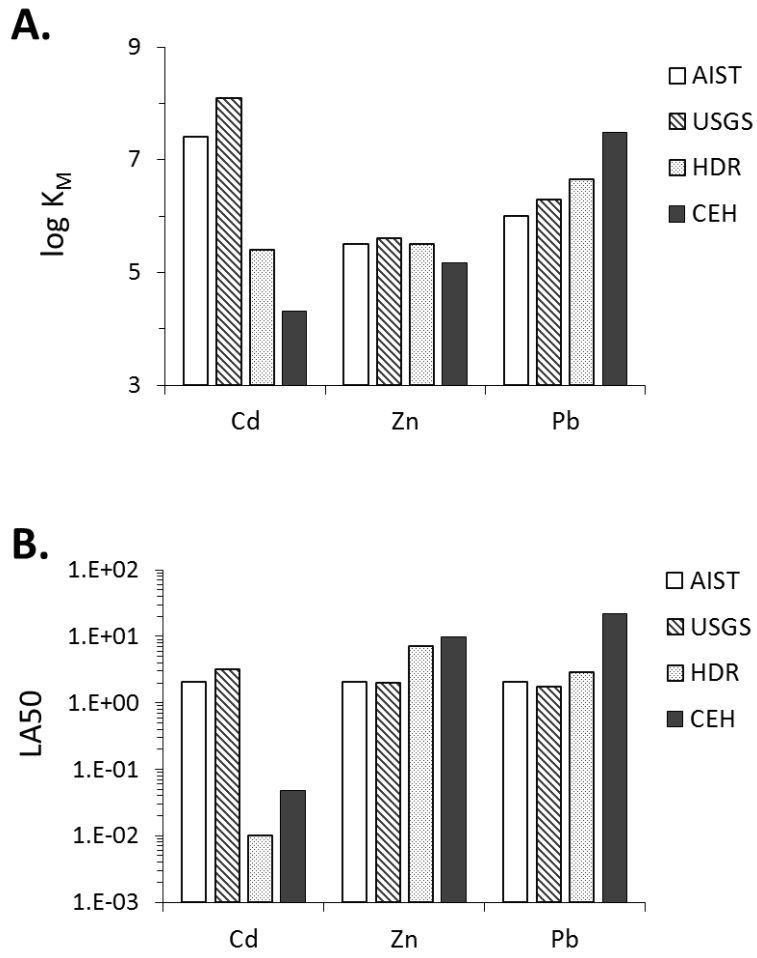


Figure 5.

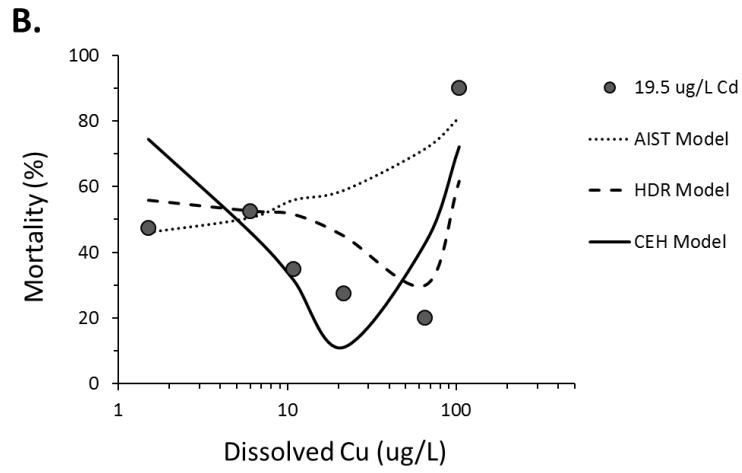
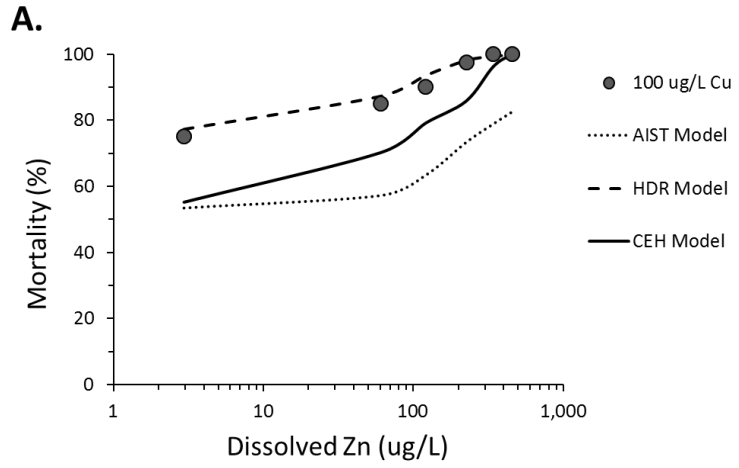


Figure 6.

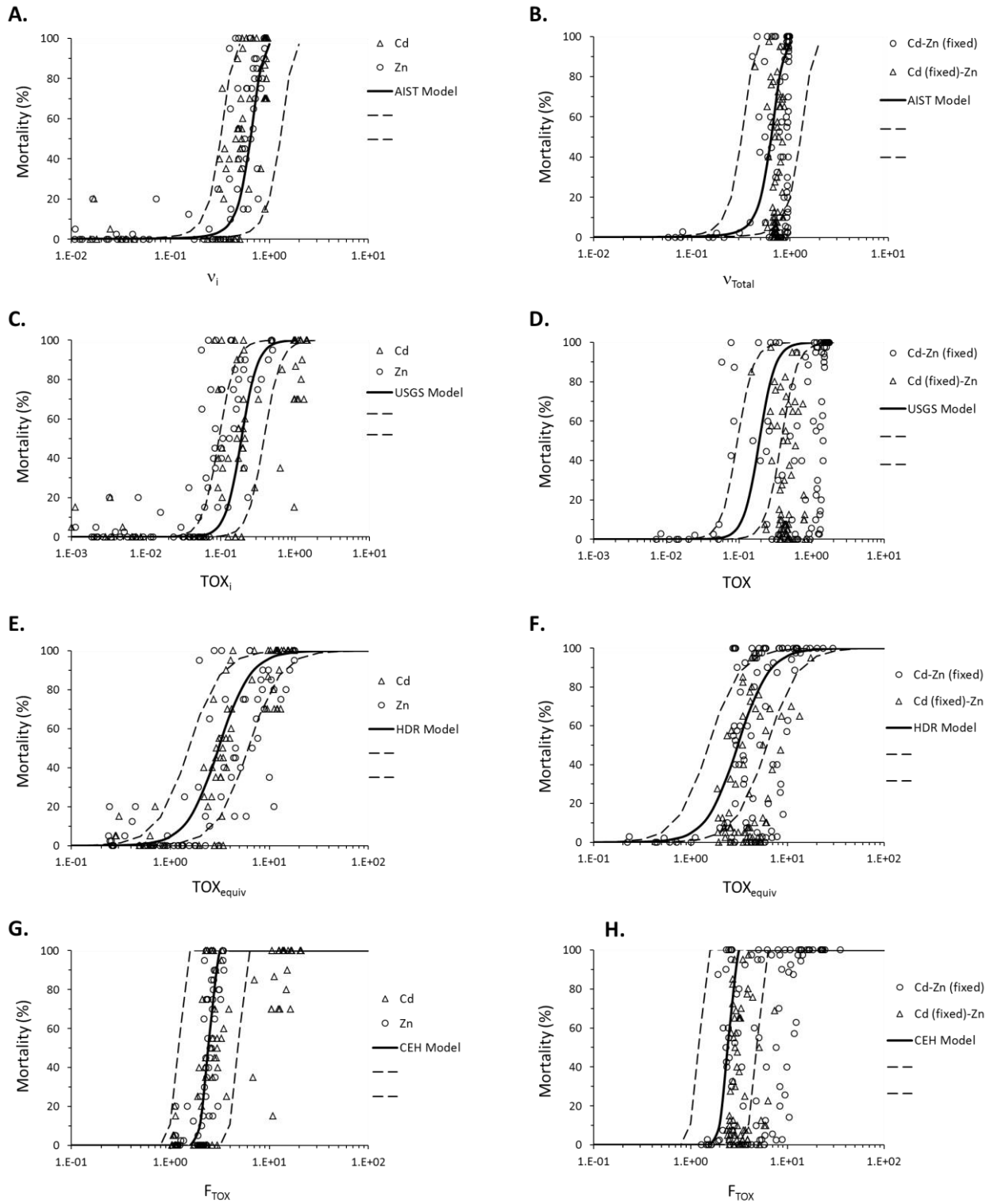


Figure 7.