

Post Print

This article is a version after peer-review, with revisions having been made. In terms of appearance only this might not be the same as the published article.

3 4 5

1 2

Determination of steroid estrogens in wastewater by high performance liquid

6 chromatography-tandem mass spectrometry

7 Journal of Chromatography A, 1173 (2007) 81–87

8

10

Y.K.K. Koh¹, T.Y. Chiu², A. Boobis¹, E. Cartmell², J.N. Lester² and M.D. Scrimshaw^{3*}

11

- 12 ¹ Faculty of Medicine, Division of Experimental Medicine and Toxicology, Imperial College
- London, Hammersmith Campus, London, W12 ONN, United Kingdom.
- ² School of Applied Science, Centre for Water Science, Cranfield University, Bedfordshire,
- 15 MK43 0AL, United Kingdom.
- ³ Institute for the Environment, Brunel University, Uxbridge, Middlesex, UB8 3PH, United
- 17 Kingdom.

18

- 19 Keywords: estrogen; solid phase extraction; wastewater; liquid chromatography; mass
- 20 spectrometry

21

22

Abstract

- 23 This paper discusses the requirement for, and presents an analytical procedure for, the
- 24 determination of four unconjugated steroid hormones and a conjugated steroid
- 25 (estrone-3-sulfate) in wastewaters. The method quantifies the steroids by LC/MS/MS
- 26 following solid phase extraction and a two stage clean-up procedure. Samples were
- 27 extracted using C18 cartridges and eluates were then purified by gel permeation
- 28 chromatography, followed by a further clean-up step on an aminopropyl cartridge.
- 29 The limits of detection achieved were 0.2 ng l^{-1} for estriol, 17β-estradiol and 17α-
- 30 ethinylestradiol, and 0.1 ng l⁻¹ for estrone and the conjugate. The robustness of the
- 31 method was demonstrated by achieving recoveries of >83% for all steroids in settled
- sewage and final effluent samples with relative standard deviations of 0.5 12%. The
- method was used to analyse a range of samples from a wastewater treatment works in
- south east England which demonstrated a >80% removal for estrone, estradiol and
- estriol with little impact on concentrations of ethinylestradiol or the conjugate.

* corresponding author: <u>mark.scrimshaw@brunel.ac.uk</u>

1. Introduction

It has been estimated that over 99% of the estrogenic activity in sewage effluents and surface waters may be attributable to the presence of 17β -estradiol (E2) and 17α -ethynyl estradiol (EE2) at concentrations in the ng Γ^1 range [1]. Exposure studies indicate that fish exposed to such concentrations of these hormones exhibit changes in biomarkers for estrogenicity at environmentally relevant concentrations of 1 ng Γ^1 (E2) and as low as 0.1 ng Γ^1 for EE2 [2-7]. Excretion from humans is the major source of natural and synthetic steroid estrogens to sewage treatment works (STW). The predominant pathway of excretion is as conjugates, either glucuronides or sulfates in urine, with the sulfate conjugate of estrone (E1-3S) being the main urinary excretion product [8, 9]. Due to the activity of β -glucuronidase, gluconuride conjugates are broken down before reaching the STW, however, concentrations of the conjugated steroid, E1-3S, may be important when considering total load reaching STW [10].

Significant attention has, therefore, been given to the possibility of controlling the discharge of steroid estrogens from STW, and within the United Kingdom a research programme, the National Demonstration Programme, has been instigated to investigate the efficiency of a range of treatment processes at reducing concentrations of these compounds [11]. Overall, the work programme will cost up to £40M, with £5M being allocated for sampling and analysis costs. It is therefore important that robust methodology be available to underpin the research output from such studies.

The determination of free and conjugated steroid estrogens has been undertaken by gas chromatography mass spectrometry (GC/MS) or tandem MS [12-16] as well as by liquid chromatography (LC) MS/MS [10, 17-19]. An advantage of using LC/MS/MS is the ability to analyze the estrogens without derivatization, or the need to hydrolyse the conjugated form, which are limiting steps in determining both species [20, 21]. However, using LC/MSMS as an analytical tool is not without difficulties. It is known that electrospray ionisation (ESI) can experience effects related to matrix suppression and isobaric interference when analyzing estrogens as a result of co-eluting compounds during the chromatographic separation [22, 23]. Recent studies using ESI-LC/MS/MS have observed ion suppression that varied by a factor of 8–10 between and within runs for various analytes [24]. Therefore, the need to determine these compounds at trace concentrations in complex matrices such as wastewaters and

- effluents is likely to require intensive clean-up procedures and the use of appropriate
- 73 internal standards, which allow for correction due to losses during sample preparation
- and as a result of matrix effects.

- 76 Extraction of steroid hormones from wastewater is usually performed by off-line solid
- 77 phase extraction (SPE) using octadecyl (C_{18})-bonded silica adsorbent [3, 25-28].
- 78 Purification of wastewater extracts has been achieved by various means, including
- 79 liquid-liquid extraction, solid phase purification on C₁₈/NH₂ columns [29, 30], slica
- 80 gel column chromatography [31, 32], gel permeation on Biobeads SX-3 columns,
- 81 high performance liquid chromatography (HPLC) fractionation [25, 33], or
- combinations of all these methods [3, 34, 35].

83

- The objective of this work was to develop a sensitive and robust methodology for the
- 85 determination of four free steroid estrogens; estrone (E1), E2, estriol (E3), EE2 and
- the conjugated E1-3S applicable to a range of water and wastewater samples.

87

88 **2. Experimental**

- 89 *2.1 Reagents and chemicals*
- 90 All estrogen standards were purchased from Sigma Aldrich (Dorset, UK). Organic
- 91 solvents, dichloromethane (DCM), methanol (MeOH), acetonitrile (ACN), hexane
- 92 and ethyl acetate (EtOAc) were purchased from Rathburn Chemicals (Walkerburn,
- 93 UK). Deuterated $(d_{3/4/5})$ labelled internal standards of estrone-2,4,16,16- d_4 (E1- d_4),
- 94 17 β -estradiol-2,4,16,16,17- d_5 (E2- d_5), estriol-2,4,17- d_3 (E3- d_3), 17 α -ethynylestradiol-
- 95 2,4,16,16- d_4 (EE2- d_4) and sodium estrone-2,4,16,16- d_4 sulfate (E1-3S- d_4) were
- 96 obtained from C/D/N Isotopes (QMX Laboratories, UK). Stock solutions were
- 97 prepared in methanol. Two different solid phase extraction cartridges tC18
- 98 (500mg/6cc) and aminopropyl (NH₂) anion-exchange (500mg/6cc) were obtained
- 99 from Waters (Watford, UK) and Varian (Varian Inc, UK) respectively.

- 101 2.2 Analytical procedure
- Settled sewage and final effluent samples (1L) were filtered through GF/C (VWR
- 103 International, UK) filters prior to solid phase extraction (SPE). The samples were then
- loaded onto tC18 cartridges preconditioned with 5ml methanol followed by 5ml MQ
- water. The flow rate for sample extraction was kept constant between 5-10 ml min⁻¹

under vacuum using vacuum manifold. After the sample was loaded, the cartridge was washed with 3ml of Ultrapure water and then thoroughly dried for half an hour under vacuum prior to elution. The analytes were eluted using 10 ml MeOH followed by 10 ml DCM. A rotary evaporator (Heidolph Instruments, Germany) was employed to concentrate the extracts to 1 ml which was then evaporated to complete dryness under a gentle nitrogen stream. The dry sample was reconstituted with 0.2 ml DCM/MeOH (90:10 v/v). Gel permeation size exclusion chromatography was performed using a PLgel column, 5µm 50Å, 300 x 7.5 mm (Polymer Laboratories, UK). Conjugated and unconjugated steroids were detected at 280nm. A 6ml fraction was collected from the column using an isocratic elution of DCM/MeOH (90:10 v/v) running at 1 ml min⁻¹. All steroids eluted between 5.5 to 11.5 min, and a single fraction corresponding to this time window was collected. This fraction was dried by rotary evaporation to a final volume of approximately 0.2 ml. This was then reconstituted to 2 ml with hexane and loaded onto a conditioned (with 4ml 10% EtOAc/hexane and then 2 ml hexane) NH₂ SPE cartridge at a flow rate between 5-10 ml min⁻¹. The nonpolar steroids E1, E2 and EE2 were then eluted using 6ml EtOAc. The more polar conjugate (E1-3S) and E3 were subsequently eluted in a second fraction using 3% NH₄OH in methanol. The separate eluates were blown to dryness under a gentle stream of nitrogen, reconstituted with 0.2 ml MeOH/H₂O (10:90 v/v) and transferred to autosampler vials prior to analysis using LC/MS/MS.

126127

106

107

108

109

110

111

112

113

114

115

116

117

118

119

120

121

122

123

124

125

2.3 Instrumental analysis

128 Concentrations of steroid estrogens were determined using LC/ESI(^)/MS/MS 129 consisting of an HPLC (Waters Alliance HPLC system 2695) coupled to a Waters Ouattro Premier XE mass spectrometer with a Z-Spray ESI source (Micromass, UK). 130 131 The steroids were separated on a Gemini C18 column (3µm particle size, 100mm x 132 2mm i.d., Phenomenex, UK). The mass spectrometer was operated in the negative 133 electrospray ionisation mode using multiple reaction monitoring (MRM). The conditions for detection by the mass spectrometer were as follows: capillary voltage, 134 3.20kV; multiplier voltage, 650V; desolvation gas flow, 1000 1 h⁻¹; cone at -55V; RF 135 lens at 0.2V; cone gas flow at 49 l h⁻¹; desolvation temperature at 350°C and source 136 137 temperature at 120°C.

138

3. Results

140

- 141 3.1 Optimisation of LC conditions
- 142 LC optimisation was carried out on Gemini C18 column (Phenomenex, UK). A
- gradient separation was achieved using two solvents, water containing 0.1% NH₄OH
- 144 (A) and MeOH containing 0.1% NH₄OH (B). Gradient conditions were initiated with
- 145 20% B followed by an increase to 50% B (over 3.5 min). The proportion of solvent B
- was then increased to 60% maintained for 9 min before the column was returned to
- starting conditions 20% B (over 3 min) and held for 2.5 min to equilibration. The total
- run time was 18 min and a sample volume of 20µl of was injected into the HPLC
- 149 (Figure 1). Eight point calibration curves were made for each of the steroids within
- the linear range of the instrument $(1 100 \text{ ng ml}^{-1})$. The concentrations of the steroid
- estrogens in the samples were calculated relative to the deuterated standards using the
- 152 MassLynx software.

153

154 Please insert Figure 1.

155

- 156 3.2 Optimising MS/MS condition
- 157 The optimal MS/MS conditions for the analysis of conjugated and unconjugated
- steroid estrogens were examined. The optimisation was carried out in the negative
- mode using electrospray ionization. Single standard solutions were used to identify
- 160 W ions and peak retention times (Table 1). The optimisation of operating parameters
- affecting MS detection such as dwell time, cone voltages and collision energy on each
- ion were carried out by the direct infusion of 100 ng ml⁻¹ standard of each steroid at
- 163 flow rate of 10 µl min⁻¹. The optimum conditions were reached when the highest
- 164 intensities or superior signal-to-noise (S/N) resolution were achieved for each
- 165 conjugate and unconjugated steroid estrogens. For greater manipulation of MS
- settings and to improve sensitivity, chromatographic separation was divided into two
- acquisition periods. In the first period between 0 and 11 min, intensities of ions for
- 168 E1-3S, E1-3S- d_4 , E3 and E3- d_3 were monitored, while in the second acquisition
- period between 11 and 18 min, intensities of ions for E1, E1- d_4 , E2, E2- d_5 , EE2 and
- 170 EE2- d_4 were detected (Figure 2).

171

172 Please insert Table 1.

174 Please insert Figure 2.

176 3.3 Concentration/extraction and clean-up

Several SPE procedures were assessed using one litre samples spiked with the steroid estrogens at 15 ng l⁻¹ each. Utilising any selective elution for this first step, to begin a sample clean-up, resulted in poor recoveries, and the cartridges were therefore eluted with 10ml methanol followed by 10 ml DCM. The use of gel permeation as a subsequent preparation step was a challenge due to the relatively high polarity of E1-3S and E3. Therefore initial work focussed on finding a solvent system which would dissolve the range of estrogens, but which was also compatible with the PLgel column. The optimal compromise in adjusting polarity of the solvent mix and achieving desired swelling of the gel within the GPC column was achieved with 10% MeOH in

186 DCM.

The anion exchange SPE was used as a final clean-up step to remove interferences that may otherwise affect the LC/MS/MS analysis of the steroids. Employing a Varian NH₂ weak anion exchange cartridge, recoveries of more than 83% were achieved for all of the steroid estrogens including the conjugated steroid E1-3S. The scheme for the analytical procedure developed in this study is shown in Figure 3.

Please insert Figure 3.

3.4 Evaluation of method performance

The calibration curves for the determination of the analytes were obtained by performing a linear regression analysis on the standard solutions using the ratio of standard area to internal standard area. The calibrations were all linear with r^2 values greater than 0.998. The instrument detection limits (IDL) were 9-20 pg based on the extraction of reagent grade water samples spiked at 1 ng Γ^1 . The recoveries of the analytes were evaluated by spiking at both low (2 ng Γ^1) and high (15 ng Γ^1) concentrations in settled sewage and final effluent. The recoveries for three replicate samples spiked in samples of settled sewage and final effluent ranged from 83 - 100%, with relative standard deviations of 0.3 - 12% (Table 2). Recoveries obtained in this study were calculated by the subtraction of concentrations observed in unspiked samples. The method detection limit (MDL) is reported as concentrations

corresponding to a S/N ratio of 3 on the chromatogram of actual sample matrices. The

MDL were 0.2 - 0.1 ng l⁻¹ for settled sewage and sewage effluent samples spiked at 1

210 $\text{ng } 1^{-1} \text{ (Table 2)}.$

211

212 Please insert Table 2.

213

214

4. Discussion

- 215 *4.1 Optimisation of methodology*
- Widely used organic mobile phases in steroid analysis such as ACN and MeOH have
- been considered. Acetronitrile was previously reported to result in the co-elution
- between conjugates and free steroids [10]. Methanol gave superior chromatographic
- 219 resolution with regard to steroid conjugates as well as increased sensitivity compared
- 220 to ACN. This is in agreement with other studies [36-38]. Therefore methanol was
- 221 chosen as the organic mobile phase in this study. Sensitivity of LC/ESI(-)/MS/MS
- determination has been previously reported to improve by adding a strong base such
- as NH₄OH to the mobile phase [37]. It was thus necessary to investigate if this was
- applicable to this methodology. A concentration range of up to 0.1% NH₄OH was
- investigated in this study. An initial increase in the signal-to- noise (S/N) ratio (5%)
- was observed when concentrations of less than 0.01% NH₄OH were added. The
- 227 absolute abundance of ions for all compounds increased significantly when using
- 228 0.1% NH₄OH and this was incorporated into the mobile phase used for the analysis of
- the steroid estrogens.

230

- The steroid estrogen and conjugate steroids were analyzed by tandem MS/MS using
- 232 ESI interface in the negative ion MRM mode. Studies have shown that greater S/N
- ratio when operated in ESI thus leading to a lower LOD [39]. Single ion transitions
- were monitored for all the analytes which were characteristic of the parent compounds.
- A second transition was also monitored as confirmation (Table 2).

- 237 It was demonstrated that high recoveries could be obtained by employing a non-
- 238 specific SPE method to concentrate the analytes of interest from the complex
- 239 wastewater matrix. Automated high performance gel permeation chromatography
- 240 gave high reproducibility and high selectivity for the steroid compounds. The
- 241 physiochemical nature of the steroid estrogens (particularly the polar nature of E3 and

E1-3S) in this study proved to be a challenge when applied to the PLgel column. PLgel is compatible with an extensive range of organic solvents and over the pH range 7-14. However, in order to maintain the swelling of the resin, addition of polar solvents such as water at concentrations more than 10% by volume was cautioned by the manufacturer. In this study, we endeavoured to find a solvent mixture that dissolved both the nonpolar steroid and the polar conjugate and which also was compatible with the elution solvent used on PLgel column. The results from these experiments indicated that a small volume of MeOH was essential. Poor recoveries were obtained when MeOH was absent in the DCM mobile phase. When MeOH was present in a higher proportion than DCM, the packing material within the PLgel column changed and affected the column performance.

In the final (second) clean-up step, two commercially available SPE cartridges, tC18 and NH₂, were evaluated to assess their ability to remove interferences and provide a clean matrix for LC/MS/MS. Several solvent combinations were also tested. The tC18 was not effective as a clean-up step and hence high noise, ionisation suppression and poor recoveries were observed. Recoveries of less than 43% and 38% were achieved for the unconjugated steroids and conjugated E1-3S respectively (data not shown). In contrast, aminopropyl SPE gave good recoveries and little matrix interference, however, some interference due to ion suppression was observed when both the conjugated and unconjugated steroids were eluted simultaneously from the anion-exchange cartridge. The stepwise wash with 10% EtOAc/hexane and the separate elution of the hydrophilic conjugates and the hydrophobic unconjugated steroids reduced isobaric interferences and ion suppression thus resulting in an increase in *S/N* ratio. These observations and results concurred with those of the findings observed by others [38].

4.2 Method performance

The methodology described here obtained higher recovery of E1-3S compared to other studies; 10% influent and 49% for effluent [40]; 89% influent, 87% effluent and 93% for river water [37]; 87% laboratory water and 97% for surface water [30]. The method detection limit obtained was similar to that of Isobe et al. [19] (one step clean-up using Florisil) for E1, E1-3S and EE2 of 0.1 ng I⁻¹, 0.1 ng I⁻¹ and 0.2 ng I⁻¹ respectively. An advantage of the method described here, however, is an improved

276 MDL for E2 and E3 of 0.3 ng l⁻¹ and 1.5 ng l⁻¹ respectively. Other published works

277 have reported MDL for E1-3S at 0.16 ng l⁻¹ [38] and 0.2 ng l⁻¹ [37]. Although similar

278 MDL for E1-3S have been obtained with methodology employing two clean-up steps,

detection limits for non-conjugated steroids were compromised (E1, E2, E3 and EE2

at 0.8 ng 1^{-1} , 0.5 ng 1^{-1} , 1.4 ng 1^{-1} and 1.2 ng 1^{-1} respectively) [43]. The procedure

described here is thus more robust in comparison to other works that have included

conjugated steroids into their analysis with either a one or two step clean-up regime

283 [30, 37, 38, 40].

284

285

281

282

- 4.3 Application to wastewater samples
- The concentrations of the steroids detected in the wastewater are summarized in Table
- 3. All compounds were detected in the settled sewage. The treatment at the works was
- a trickling filter, and there was little or no removal of either the synthetic estrogen
- 289 (EE2) or of the conjugate, E1-3S. However, removal of the naturally occurring, free
- 290 steroids (E1, E2, and E3) was between 80 98%. Estrone was least efficiently
- removed, which may be a result of the biological transformation of E2 to E1 in the
- 292 filter. It is intended to apply the method developed here to more extensive studies of
- 293 biological treatment processes to understand the impact of process variables on
- 294 removal of estrogens from wastewaters. Data presented here corroborates with that
- 295 from other studies which have demonstrated the occurrence of E1-3S in wastewaters
- 296 [10, 37, 41] and receiving waters [30].

297298

Please insert Table 3.

299

300

5. Conclusions

- 301 A sensitive and selective analytical method based on SPE, GPC, aminopropyl anion
- 302 exchange and LC/MS/MS has been developed for the determination of polar
- 303 conjugated and nonpolar unconjugated steroid estrogens in complex wastewater
- samples at sub-ng 1⁻¹ concentrations. To analyse estrogens at such concentrations in
- wastewater is challenging due to matrix effects and the range of interferences likely to
- 306 be present. Therefore, a series of purifications steps have been optimised which result
- in the removal of many matrix interferences. The technique is relatively rapid, semi-
- 308 automated and hence not as time consuming as other extraction and cleanup
- approaches such as liquid-liquid extraction, or where derivatization is required, thus

allowing for rapid, selective and sensitive analysis of both conjugate and unconjugated steroids in wastewater.

Acknowledgement

One of the authors (Y.K.K. Koh) is grateful to the Public Utilities Board of Singapore for the award of a PhD scholarship. The authors would like to thank the following companies: Anglian Water Plc, Severn Trent Plc, Thames Water Plc, United Utilities Plc and Yorkshire Water (Kelda Group Plc) for providing their support and funding.

LIST OF TABLES Table 1. Optimised LC/MS/MS conditions for MRM chromatographic acquisition of steroid estrogens^a Table 2. Method recoveries (%) and relative standard deviations (RSD %) and method detection limit (MDL) from settled sewage and final effluent (n=3). Table 3. Concentrations of conjugated and unconjugated steroid estrogens in samples from a wastewater treatment works (January 2007). **LIST OF FIGURES** Figure 1. Total ion chromatogram of conjugated and unconjugated steroids standard solutions at 20 ng l⁻¹. Figure 2. (a) MRM chromatograms of E1-3S-d₄, E1-3S, E3-d₃ and E3. (b) MRM chromatograms of EE2-d₄, EE2, E2-d₅, E1-d₄, E2 and E1 (settled sewage spiked at 2 ng 1⁻¹). Figure 3. Diagram summarising the analytical procedure for the determination of the steroid estrogens.

References

- 1 2
- 3 [1] S. A. Snyder, D. L. Villeneuve, E. M. Snyder, J. P. Giesy, Environ. Sci. Technol.
- 4 (2001) 35 3620.
- 5 [2] E. J. Routledge, D. Sheahan, C. Desbrow, G. C. Brighty, M. Waldock, J. P.
- 6 Sumpter, Environ. Sci. Technol. (1998) 32 1559.
- 7 [3] C. Desbrow, E. J. Routledge, G. C. Brighty, J. P. Sumpter, M. Waldock, Environ.
- 8 Sci. Technol. (1998) 32 1549.
- 9 [4] A. C. Johnson, J. P. Sumpter, Environ. Sci. Technol. (2001) 35 4697.
- 10 [5] C. E. Purdom, P. A. Hardiman, V. J. Bye, N. C. Eno, C. R. Tyler, J. P. Sumpter,
- 11 Chem. Ecol. (1994) 8 275
- 12 [6] M. Cargouet, D. Perdiz, A. Mouatassim-Souali, S. Tamisier-Karolak, Y. Levi, Sci.
- 13 Total Environ. (2004) 324 55.
- 14 [7] V. J. Kramer, S. Miles-Richardson, S. L. Pierens, J. P. Giesy, Agua. Toxicol.
- 15 (1998) 40 335.
- 16 [8] T. A. Ternes, P. Kreckel, J. Mueller, Sci. Total Environ. (1999) 225 91.
- 17 [9] P. W. Tang, D. L. Crone, Anal. Biochem. (1989) 182 289.
- 18 [10] R. L. Gomes, J. W. Birkett, M. D. Scrimshaw, J. N. Lester, Int. J. Environ. An.
- 19 Ch. (2005) 85 1.
- 20 [11] UKWIR, UK Water Industry Research Limited, London (2006) 38 3.
- 21 [12] J. B. Quintana, J. Carpinteiro, I. Rodriguez, R. A. Lorenzo, A. M. Carro, R. Cela,
- 22 J. Chromatogr. A (2004) 1024 177.
- 23 [13] K. M. Lai, M. D. Scrimshaw, J. N. Lester, Appl. Environ. Microbiol. (2002) 68
- 24 859.
- 25 [14] H. B. Lee, T. E. Peart, M. L. Svoboda, J. Chromatogr. A (2005) 1094 122.
- 26 [15] M. Petrovic, E. Eljarrat, M. J. Lopez de Alda, D. Barcelo, J. Chromatogr. A
- 27 (2002) 974 23.
- 28 [16] H. B. Lee, T. E. Peart, J. AOAC Int. (1998) 81 1209.
- 29 [17] A. Lagana, G. Fago, A. Marino, D. Santarelli, Anal. Lett. (2001) 34 913.
- 30 [18] M. J. Lopez de Alda, S. Diaz-Cruz, M. Petrovic, D. Barcelo, J. Chromatogr. A
- 31 (2003) 1000 503.
- 32 [19] A. Lagana, A. Bacaloni, G. Fago, A. Marino, Rapid Commun. Mass Spectrom.
- 33 (2000) 14 401.
- 34 [20] C. H. Huang, D. L. Sedlak, Environ. Toxicol. Chem. (2001) 20 133.
- 35 [21] T. R. Croley, R. J. Hughes, B. G. Koenig, C. D. Metcalfe, R. E. March, Rapid
- 36 Commun. Mass Spectrom. (2000) 14 1087.
- 37 [22] T. Benijts, R. Dams, W. Gunther, W. Lambert, A. De Leenheer, Rapid Commun.
- 38 Mass Spectrom. (2002) 16 1358.
- 39 [23] T. Benijts, R. Dams, W. Lambert, A. De Leenheer, J. Chromatogr. A (2004)
- 40 1029 153.
- 41 [24] L. D. Bowers, Clin. Chem. (1998) 44 375.
- 42 [25] T. P. Rodgers-Gray, S. Jobling, S. Morris, C. Kelly, S. Kirby, A. Janbakhsh, J. E.
- 43 Harries, M. J. Waldock, J. P. Sumpter, C. R. Tyler, Environ. Sci. Technol. (2000) 34
- 44 1521.
- 45 [26] L. S. Shore, M. Gurevitz, M. Shemesh, Bull. Environ. Contam. Toxicol. (1993)
- 46 51 361.
- 47 [27] K. Shishida, S. Echigo, K. Kosaka, M. Tabasaki, T. Matsuda, H. Takigami, H.
- 48 Yamada, Y. Shimizu, S. Matsui, Environ. Technol. (2000) 21 553.
- 49 [28] M. J. Lopez de Alda, D. Barcelo, J. Chromatogr. A (2000) 892 391.
- 50 [29] A. C. Johnson, A. Belfroid, A. Di Corcia, Sci. Total Environ. (2000) 256 163.

- 1 [30] T. Isobe, H. Shiraishi, M. Yasuda, A. Shinoda, H. Suzuki, M. Morita, J.
- 2 Chromatogr. A (2003) 984 195.
- 3 [31] T. A. Ternes, M. Stumpf, J. Mueller, K. Haberer, R. D. Wilken, M. Servos, Sci.
- 4 Total Environ. (1999) 225 81.
- 5 [32] H. M. Kuch, K. Ballschmiter, Fresenius J. Anal. Chem. (2000) 366 392.
- 6 [33] S. A. Snyder, T. L. Keith, D. A. Verbrugge, E. M. Snyder, T. S. Gross, K.
- 7 Kannan, J. P. Giesy, Environ. Sci. Technol. (1999) 33 2814.
- 8 [34] D. G. J. Larsson, M. Adolfsson-Erici, J. Parkkonen, M. Pettersson, A. H. Berg,
- 9 P.-E. Olsson, L. Forlin, Aquat. Toxicol. (1999) 45 91.
- 10 [35] A. C. Belfroid, A. Van der Horst, A. D. Vethaak, A. J. Schafer, G. B. J. Rijs, J.
- 11 Wegener, W. P. Cofino, Sci. Total Environ. (1999) 225 101.
- 12 [36] H. H. Tabak, R. N. Bloomhuff, R. L. Bunch, Dev. Ind. Microbiol. (1981) 22 497.
- 13 [37] A. Gentili, D. Perret, S. Marchese, R. Mastropasqua, R. Curini, A. Di Corcia,
- 14 Chromatographia (2002) 56 25.
- 15 [38] S. Reddy, C. R. Iden, B. J. Brownawell, Anal. Chem. (2005) 77 7032.
- 16 [39] V. Ingrand, G. Herry, J. Beausse, M.-R. de Roubin, J. Chromatogr. A (2003)
- 17 1020 99.

- 18 [40] K. Komori, H. Tanaka, Y. Okayasu, M. Yasojima, C. Sato, Water Sci. Technol.
- 19 (2004) 50 93.
- 20 [41] G. D'Ascenzo, A. Di Corcia, A. Gentili, R. Mancini, R. Mastropasqua, M.
- 21 Nazzari, R. Samperi., Sci. Total Environ. (2003) 302 199.

1 List of figures

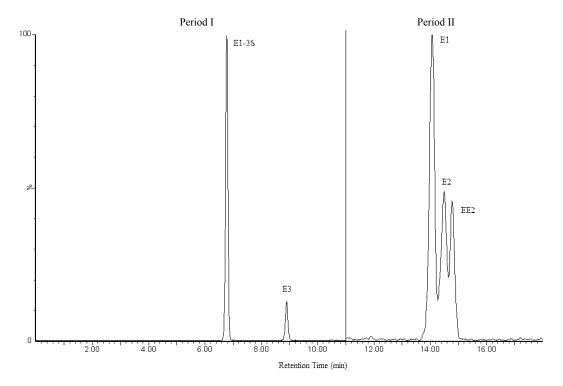


Figure 1. Total ion chromatogram of conjugated and unconjugated steroids standard solutions at $20~\rm ng~l^{\text{-}1}$.

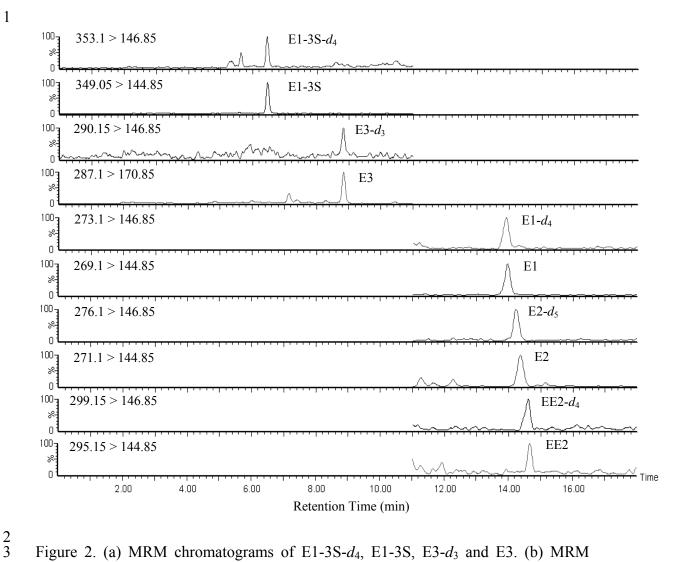


Figure 2. (a) MRM chromatograms of E1-3S- d_4 , E1-3S, E3- d_3 and E3. (b) MRM chromatograms of EE2- d_4 , EE2, E2- d_5 , E1- d_4 , E2 and E1 (settled sewage spiked at 2 ng l^{-1}).

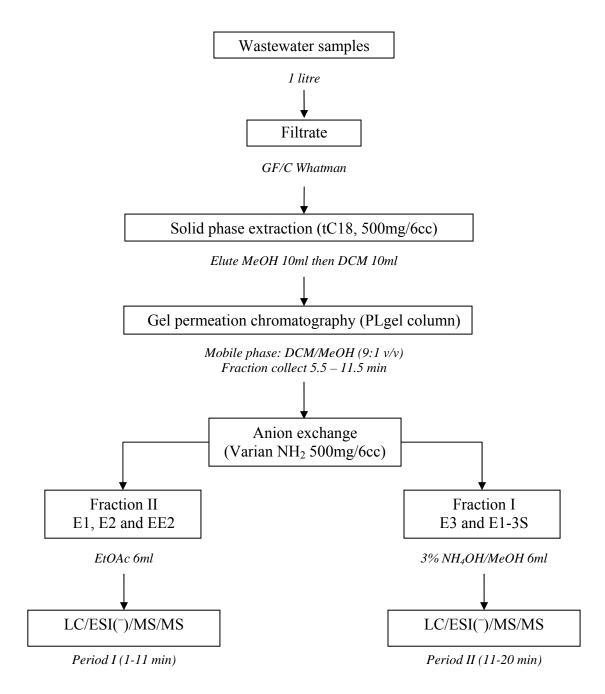


Figure 3. Diagram summarising the analytical procedure for the determination of the steroid estrogens.

Table 1. Optimised LC/MS/MS conditions for determination of the steroid estrogens^a

Estrogen	Period (min)	^a MRM <i>m/z</i>	Dwell time (msec)	Collision energy (V)	Cone (V)	Retention time (min)	IDL ^b (pg)
E1	II (11-20)	269.10>144.85	85	40	70	13.97	12
		269.10>158.80	85	45	70		
E2	II (11-20)	271.10>144.85	85	45	60	14.37	20
		271.10>158.80	85	40	60		
E3	I (0-11)	287.10>170.85	95	50	55	8.90	18
		287.10>144.85	95	50	55		
EE2	II (11-20)	295.15>144.85	85	40	60	14.67	16
		295.15>158.80	85	40	60		
E1-3S	I (0-11)	349.05>144.85	60	65	50	6.77	9
		349.05>269.00	60	40	50		
E1- d_4	II (11-20)	273.10>146.85	85	45	60	13.91	
E2- d_5	II (11-20)	276.10>146.85	85	50	55	14.23	
E3- d_3	I (0-11)	290.15>146.85	90	65	50	8.86	
EE2- <i>d</i> ₄	II (11-20)	299.15>146.85	85	50	60	14.60	
E1-3S-d ₄	I (0-11)	353.10>146.85	60	65	50	6.75	

^aMRM transitions: the first for quantification, the second for confirmation. ^b Instrument detection limit at 1 ng 1^{-1} replicate measurement (n=7).

4 5

Table 2. Method recoveries (%) and relative standard deviations (RSD %) and method detection limit (MDL) from settled sewage and final effluent (n=3).

Compound	Settled sewage (% RSD)		Final effluent (% RSD)		MDL ^c (ng l ⁻¹)	
·	Low	High	Low	High	Settled	Final
	spike ^a	spike ^b	spike ^a	spike ^b	sewage ^a	effluent ^a
E1	98 (1.5)	95 (4)	100(2)	88 (3)	0.1	0.1
E2	100 (0.6)	88 (1.6)	100(7)	88 (4)	0.2	0.2
E3	100 (0.7)	98 (0.3)	83 (12)	86 (6)	0.2	0.2
EE2	90 (10)	88 (5)	100(2)	83 (5)	0.2	0.2
E1-3S	97 (1)	95 (4)	96 (2)	99 (1)	0.1	0.1

^a2 ng of standard or ^b15 ng of standard was spiked to 1 litre of settled sewage/final effluent (15 ng l⁻¹ of deuterated internal standard).

Table 3. Concentrations of conjugated and unconjugated steroid estrogens in samples from a wastewater treatment works (January 2007).

2 3

Compound	Settled sewage	Final effluent	Removal (%)	
	$(ng l^{-1})$	$(ng l^{-1})$		
E1	15	3.0	80	
E2	5.0	0.7	86	
E3	50	1.0	98	
EE2	1.2	1.0	17	
E1-3S	10	12	=	