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Short-Term Response of Cytosolic NO₃ to Inorganic Carbon Increase in Posidonia oceanica Leaf Cells

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The concentration of CO₂ in the atmosphere has increased over the past 200 years and is expected to continue rising in the next 50 years at a rate of 3 ppm·year⁻¹. This increase has led to a decrease in seawater pH that has changed inorganic carbon chemical speciation, increasing the dissolved HCO₃. Posidonia oceanica is a marine angiosperm that uses HCO₃ as an inorganic carbon source for photosynthesis. An important side effect of the direct uptake of HCO₃ is the diminution of cytosolic Cl⁻ (Cl⁻c) in mesophyll leaf cells due to the efflux through anion channels and, probably, to intracellular compartmentalization. Since anion channels are also permeable to NO₃ we hypothesize that high HCO₃, or even CO₂, would also promote a decrease of cytosolic NO₃ (NO₃c). In this work we have used NO₃- and Cl⁻-selective microelectrodes for the continuous monitoring of the cytosolic concentration of both anions in P. oceanica leaf cells. Under light conditions, mesophyll leaf cells showed a NO_3 c of 5.7 \pm 0.2 mM, which rose up to 7.2 ± 0.6 mM after 30 min in the dark. The enrichment of natural seawater (NSW) with 3 mM NaHCO₃ caused both a NO₃ c decrease of 1 \pm 0.04 mM and a Cl_c decrease of 3.5 ± 0.1 mM. The saturation of NSW with 1000 ppm CO₂ also produced a diminution of the NO₃c, but lower (0.4 ± 0.07 mM). These results indicate that the rise of dissolved inorganic carbon (HCO₃ or CO₂) in NSW would have an effect on the cytosolic anion homeostasis mechanisms in P. oceanica leaf cells. In the presence of 0.1 mM ethoxyzolamide, the plasma membrane-permeable carbonic anhydrase inhibitor, the CO_2 -induced cytosolic NO_3^- diminution was much lower (0.1 \pm 0.08 mM), pointing to HCO₃ as the inorganic carbon species that causes the cytosolic NO₃ leak. The incubation of P. oceanica leaf pieces in 3 mM HCO₃-enriched NSW triggered a shortterm external NO₃ net concentration increase consistent with the NO₃c leak. As a consequence, the cytosolic NO₃ diminution induced in high inorganic carbon could result in both the decrease of metabolic N flux and the concomitant biomass N impoverishment in *P. oceanica* and, probably, in other aquatic plants.

Keywords: cytosolic NO₃-, elevated inorganic carbon, NO₃- efflux, anion channels, intracellular nitrate-selective microelectrodes, seagrasses

INTRODUCTION

Nitrate ion (NO₃) is the main source of inorganic nitrogen for plants in aerobic conditions. Compared with the concentrations in soils, the concentration of NO₃ in seawater is persistently very low, particularly in the oligotrophic Mediterranean Sea (Lepoint et al., 2002; Bethoux et al., 2005). Seagrasses are the unique angiosperms that evolved from land plants to live submerged in the sea, forming the basis of the most productive and widespread coastal ecosystems on the planet (Larkum et al., 2006). For vascular plants, colonizing the sea implicates losses and gains to effect structural and physiological adaptations to complete the submerged life cycle, achieved by a reverse evolutionary trajectory in the seagrass lineage (Williams, 2016). Thus, key land angiosperm innovations were lost in seagrasses including the entire collection of genes involved in stomata differentiation, genes related to the synthesis and sensing of terpenoids and other volatile substances, genes for ultraviolet protection, and phytochromes for far-red sensing (Olsen et al., 2016). However, to survive in the conditions of low NO₃ availability, seagrasses have evolved high affinity uptake systems to capture NO₃ through their leaves. These systems have been characterized for Zostera marina in which the uptake of NO3 and inorganic phosphate (Pi) are driven by the inwardly directed electrochemical gradient for Na+ (García-Sánchez et al., 2000; Rubio et al., 2005). In the Mediterranean, Posidonia oceanica is an endemic coastal species of huge ecological importance (Aires et al., 2011). Similar high-affinity and Na+-dependent uptake mechanisms also operate in P. oceanica for both nutrients and some amino acids (Rubio et al., 2018). In both cases, the low semi-saturation constants observed (2.3 and 8.7 μ M NO $_3^-$ for Z. marina and P. oceanica, respectively: García-Sánchez et al., 2000; Rubio et al., 2018) indicate that those systems are very efficient for NO₃ uptake at the very low concentrations of NO₃ in seagrass meadows (Touchette and Burkholder, 2000; Romero et al., 2006). Nevertheless, those systems are energetically expensive because seagrass leaf cells have to keep low homeostatic Na⁺ concentrations in the cytosol to maintain the Na⁺ motive force (Rubio et al., 2011; Rubio et al., 2018). In this scenario, namely low availability of NO₃ and the energetically expensive mechanism for its high-affinity uptake, the maintenance of NO₃ inside the cells appears critical. Therefore, as with other vascular plants, seagrasses must maintain intracellular NO₃ homeostasis to preserve the N metabolic flux.

In a previous work, besides cytosolic H⁺ and Na⁺, we also measured cytosolic Cl⁻ (using intracellular ion-selective microelectrodes) in the mesophyll cells of *P. oceanica* (Rubio et al., 2017). In that work, we demonstrated that this seagrass has a direct plasma membrane symporter to uptake HCO₃⁻ driven by the H⁺ electrochemical potential gradient. A significant increase of photosynthesis in natural seawater supplemented by 3 mM HCO₃⁻ supported the role of HCO₃⁻ uptake in the photosynthetic activity of this species (Rubio et al., 2017). Furthermore, the enrichment of seawater with 3 mM HCO₃⁻ also evoked a delayed, but significant, diminution of the cytosolic Cl⁻ concentration (Rubio et al., 2017). Similar cytosolic Cl⁻ efflux has been observed

in guard cells of the model land angiosperm Arabidopsis thaliana during stomatal closure in response to elevated CO_2 (Xue et al., 2011). This Cl^- efflux from guard cells is described as taking place through the plasma membrane S-type anion channels, whose activation responded to the cytosolic HCO_3^- concentration (Xue et al., 2011). Interestingly, in addition to Cl^- these anion channels have a high permeability to NO_3^- (Schmidt and Schroeder, 1994), which is also released from the guard cells during stomatal closure (Geiger et al., 2011; Demir et al., 2013; Maierhofer et al., 2014).

The release of Cl $^-$ from *P. oceanica* in response to HCO $_3^-$ is driven by the outwardly directed electrochemical potential gradient for Cl $^-$ (Rubio et al., 2017). Considering Cl $^-$ -selective microelectrodes are also partially sensitive to NO $_3^-$ (Miller and Zhen, 1991) we hypothesize that the increase of inorganic carbon in seawater could also lead to the decrease of cytosolic NO $_3^-$ in *P. oceanica* leaf cells. Such diminution would affect assimilation (Bloom et al., 2010) and could partially be responsible for the plant biomass nitrogen impoverishment expected under elevated inorganic carbon in vascular plants with non-saturated photosynthesis (Taub and Wang, 2008).

Therefore, the aim of this work was to measure the cytosolic NO₃ concentration in mesophyll leaf cells of P. oceanica to describe the responses to light and dark conditions and to the increase of dissolved inorganic carbon (CO₂ and HCO₃) in natural seawater. Ethoxyzolamide, the plasma membranepermeable carbonic anhydrase inhibitor (Sültemeyer et al., 1993), has been used in combination with CO2 to test for the effect of limiting any CO₂-dependent HCO₃ generation. Since the first work in Chara corallina (Miller and Zhen, 1991), NO₃-selective microelectrodes have been used in different plant species, such as barley (Zhen et al., 1991; Van Der Leij et al., 1998), A. thaliana (Cookson et al., 2005) and rice (Fan et al., 2007), plus the liverwort Conocephalum conicum (Trebacz et al., 1994). However, so far no direct measurement of cytosolic NO₃ has been reported for marine vascular plants. Furthermore, external NO₃ has been monitored, and the potential effects of elevated atmospheric CO2 on the hypothesized diminution on the biomass nitrogen content are also discussed.

MATERIALS AND METHODS

Plant Material

Posidonia oceanica (L.) Delile plants were sampled in Punta de Calaburras (36°30′23.4′′N 4°38′37.6′′W) Málaga, southern Spain, at 2 m depth. Plants with 6 to 12 leaves attached to a piece of the rhizome were collected and transported to the laboratory in a thermos container in less than 30 min. Then, plants were placed in an aquarium filled with continuously aerated natural seawater (NSW). The air used for this purpose was obtained from the compressed air supply of the Faculty of Sciences building. Concentration of CO_2 in the air was regularly monitored (390 ± 10 ppm) using an IRGA, LICOR LI-820, Li, Nebraska (USA). Temperature was held at 15°C, and illumination was at a light intensity of 150 μmol photons·m $^{-2}$ ·s $^{-1}$

with a photoperiod 16L/8D. Renewing the seawater every three days, plants were used for experiments within two weeks after sampling.

Cytosolic NO₃ and Cl⁻ Measurements

Cytosolic nitrate and chloride were measured by electrophysiological techniques using double-barreled intracellular microelectrodes. Glass capillary preparation details have been previously described (Fernández et al., 1999). In short, double-barreled capillaries with different diameter (1.5 and 0.75 mm o.d., respectively, Hilgenberg, Germany) were twisted before pulling using a Narishige PD-5 horizontal puller. Then, pulled double-barreled capillaries were heated for 30 min at 180°C and silanized by adding one drop of dichlorodimethylsilane dissolved in benzene (0.05% v/v) to the interior of the blunt end of the larger barrel, while the smaller barrel (that operates as voltage electrode) was not silanized. After that, the silanized double-barrelled capillaries were heated again for 60 min at 180°C. Once cool, the silanized barrel was backfilled with the appropriate chloride or nitrate sensor solution.

For Cl⁻-microelectrodes, the ionophore I (99408, Fluka), dissolved in a mixture of polyvinylchloride dissolved in tetrahydrofuran (PVC/THF, 4% w/v) was used and backfilled into the silanized barrel. Once THF evaporated, the remainder was filled with 0.5 M KCl (Planes et al., 2015). As described previously (Rubio et al., 2017), Cl⁻-microelectrodes were calibrated against NaCl solutions (1–100 mM) that contained 5 mM NaNO₃, the putative cytosolic NO₃ concentration (Miller and Smith, 2008), to minimize the interference between Cl⁻ and NO₃ in the cytosol (Felle, 1994). Calibration showed a linear relationship of 37 mV/pCl.

NO₃-microelectrodes were backfilled using a NO₃ sensor (Fluka 72549) based on the quaternary ammonium compound, methyltridodecylammonium nitrate (MTDDA.NO₃) containing the PVC/THF solution. These liquid ion-exchange based microelectrodes are highly selective for NO3, maintaining a nitrate detection limit of 0.5 mM in the presence of 100 mM Cl-, which means any interference will not be important in a physiological situation, as described in Miller and Zhen (1991). Once THF was evaporated from the nitrate-sensor cocktail and before use, the NO₃-selective barrel was backfilled with 0.1 M NaNO₃ and 0.1 M KCl. Then, NO₃-selective electrodes were calibrated against NO₃K solutions (0.1-20 mM) containing 10 mM KCl, to saturate the presumed interference using the Cl concentration reported previously in P. oceanica mesophyll leaf cells (Rubio et al., 2017), and KH₂PO₄ (from 15 to 50 mM), to give a constant background ionic strength in the calibration solutions (Miller and Zhen, 1991). NO₃ calibration showed a linear relationship of 53 mV/pNO₃.

For measurements, the microelectrode voltage barrel was backfilled with 0.5 M KCl (Fernández et al., 1999; Rubio et al., 2005). Then, the NO₃- or Cl⁻-selective microelectrode and the reference electrode (containing agar 0.03% (w/v) in 0.5 M KCl) were fixed to Ag/AgCl electrode holders and connected to a high-impedance differential amplifier (FD223a, WPI, Sarasota, Florida, USA). Amplifier signals were continuously monitored on a double pen chart recorder (Linseis L250E).

Measurements were performed on leaf pieces (≈ 1 cm long), longitudinally peeled to remove part of the epidermis and fixed with paraffin wax on a Plexiglas transparent chamber (1.1 ml volume). A gravity-based flow-through system permitted controlled changes of the assay medium at a rate of 10 ml·min $^{-1}$, which renewed chamber volume approximately 10 times every minute. This system kept the temperature, the ionic concentration, and gases constant during the experiments. Measurements were made under a microscope light of 150 μ mol photons·m $^{-2}\cdot s^{-1}$.

NO₃ Quantification in Assay Solutions

In order to monitor the net efflux and/or uptake of NO₃ from assay medium, plants were previously adapted to N-sufficiency by incubation in NSW enriched with 100 μM NaNO₃ during 2 days. Then, excised leaves (2-3 g fresh weight) were placed separately in 250 ml flasks and incubated in 100 ml NSW (control) or NSW supplemented with 3 mM NaHCO₃. The assay was carried out at 25°C with gentle and constant agitation. For each treatment, samples of assay medium were taken at 0, 1, 2, 3, 5, 7, 10, 15, 20, and 30 min. The highly sensitive method for NO₃ determination, based on vanadate NO3reduction and the subsequent spectrophotometric determination of NO₂⁻ (García-Robledo et al., 2014) was used to quantify NO₃ concentration in each sample. Net efflux and uptake rates were estimated as the slope of the linear phase of NO₃-concentration time course, as a function of fresh weight (FW). Six replicates were conducted for each assay.

Assay Solutions

Natural seawater (NSW) supplemented with 3 mM HCO₃⁻ was prepared by adding the appropriate volume of a 0.5 M NaHCO₃ stock solution (pH 8.2). CO₂-enriched NSW was obtained by bubbling with artificial air (Air Liquide, Spain, 1000 ppm CO₂-air). A stock solution of the plasma membrane-permeable carbonic anhydrase inhibitor ethoxyzolamide (EZ, 10 mM) was prepared in 0.05 M NaOH. The addition of equivalent volumes of NaOH without the inhibitor had no effect on measurements. All chemicals were analytical grade and were purchased from Sigma-Aldrich.

Data Presentation and Statistical Analyses

Time-course measurements are shown as single traces, representative of a number of equivalent experiments carried out under the same conditions, as stated in the figure legends. Data are presented as means, and error bars are standard deviations. Number of repetitions (n) is indicated in every experiment. Data were analyzed using SPSS Statistics, version 21. The significance level was set at P < 0.05.

RESULTS

Effect of Light-Dark Transitions on *P. oceanica* Cytosolic NO₃

In NSW, mesophyll leaf cells of *P. oceanica* showed a stable plasma membrane potential of -174 ± 8 mV (as in Rubio et al., 2017) and

a cytosolic NO_3^- concentration of 5.7 \pm 0.2 mM (n = 10). The transition from light to darkness evoked a fast depolarization of approximately 14 mV followed by a transient hyperpolarization of 22 mV and a subsequent, more prolonged, depolarization to level of at a lower membrane potential of -140 ± 5 mV (n = 12, P = 0.003, Student t test) after 25 min in the dark. With a delay of a few minutes, light-dark transition promoted the gradual increase of cytosolic NO_3^- concentration that stabilized at a higher value $(7.2 \pm 0.6 \text{ mM } NO_3^-; n = 10, P = 0.02$, Student t test) after 30 min in the dark (**Figure 1**).

Effect of HCO₃-Enriched NSW on Cytosolic NO₃

We have previously reported that in P. oceanica mesophyll leaf cells, incubated in light conditions, the addition of NSW enriched with 3 mM HCO₃ evoked an initial and transient plasma membrane depolarization that turned into a transient hyperpolarization to stabilize at a depolarized value (Rubio et al., 2017). The simultaneous measurement of cytosolic chloride showed a delayed but significant decrease of the cytosolic concentration of this anion concomitant with the extent of membrane depolarization (Rubio et al., 2017). The partial sensitivity of Cl--selective microelectrodes to NO₃ (Miller and Zhen, 1991; Felle, 1994) allows us to hypothesize that HCO₃ enrichment not only produces the cytosolic Cl decrease but could also evoke a cytosolic NO₃ shift. To test this hypothesis, cytosolic NO_3^- was measured in P. oceanica mesophyll leaf cells in the same conditions that we had reported for cytosolic chloride, the monitoring of which was used as a control in this work. Figure 2 shows the membrane potential response to the enrichment of NSW with 3 mM HCO₃ (black trace) and the simultaneous measurements of cytosolic Cl (green trace) or NO₃ (blue trace), values and stats are presented in **Table 1**. As

found previously (Rubio et al., 2017), the addition of 3 mM HCO₃ caused an initial and transient membrane depolarization (black trace) of approximately 5 mV, which turned into a transient hyperpolarization (reaching a minimum membrane potential of -181 ± 2 mV, n = 5, P = 0.012, Student t test), followed by a prolonged depolarization that stabilized within approximately 40 min at a membrane potential of -168 ± 3 (n = 5). Also in agreement with the previous study, the addition of 3 mM HCO₃ evoked a decrease of cytosolic Cl (after approximately 4 min) which continued progressively from 9.7 ± 0.2 mM to 6.3 ± 0.3 mM (green trace, n = 5; P = 0.004, Student t test) 20 min after HCO_3^- addition. Supporting the hypothesis of HCO₃ enrichment's producing a cytosolic NO₃ shift, P. oceanica mesophyll leaf cell cytosolic NO₃ also decreased in response to the enrichment of NSW with 3 mM HCO₃ (**Figure 2**, blue trace). In common with the response of cytosolic Cl-, the diminution of cytosolic NO₃ started approximately 4 min after HCO₃ treatment. Cytosolic NO₃ diminished gradually from 5.7 ± 0.2 mM to a steady lower value of 4.7 \pm 0.1 mM after 35 min of HCO₃ addition, resulting in a significant cytosolic NO₃ shift of 0.9 ± 0.06 mM (n = 5; P = 0.03, Student t test). In both cases, time courses of cytosolic NO $_3^-$ and cytosolic Cl⁻ decreases aligned with the recovery of the transient membrane hyperpolarization and profound membrane depolarization, supporting a cytosolic leak of negative charge induced by the HCO₃ addition.

Effect of CO₂ Increase on Cytosolic NO₃

As we have previously reported, in P. oceanica mesophyll leaf cells the direct uptake of HCO_3^- and the subsequent use of CO_2 for photosynthesis has consequences for anion homeostasis (Rubio et al., 2017). In this work, we show that HCO_3^- use in this plant has an effect not only on cytosolic Cl^- but also on

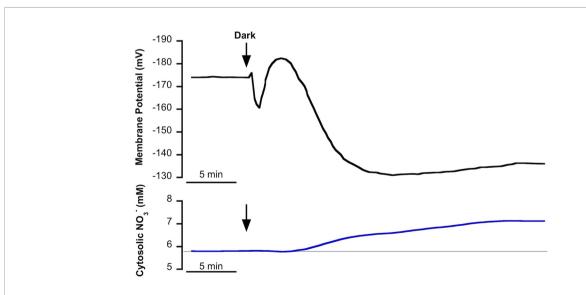


FIGURE 1 | Effect of light-dark transition on plasma membrane potential (Em, mV) and cytosolic NO $_3^-$ (mM) in *Posidonia oceanica* mesophyll leaf cells, incubated in natural seawater. Traces are representative examples of intracellular nitrate-selective microelectrode recordings from 10 independent experiments. Arrows indicate the onset of dark treatment. Auxiliary grey line shows the standard cytosolic NO $_3^-$ concentration under light conditions (150 μ mol photons m⁻² s⁻¹). Mean values and statistics are indicated in the text.

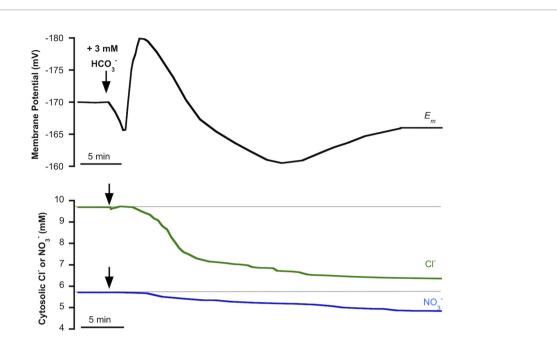


FIGURE 2 | Effect of the addition of 3 mM HCO $_3^-$ on the plasma membrane potential (*Em*, mV), cytosolic chloride (Cl $^-$, mM) or cytosolic NO $_3^-$ (NO $_3^-$, mM) measured in mesophyll leaf cells of *Posidonia oceanica*. Assay medium consisted of natural seawater and arrows indicate the addition of 3 mM HCO $_3^-$. Traces are representative examples from five independent recordings using intracellular Cl $^-$ -selective (green trace) or NO $_3^-$ -selective (blue trace) microelectrodes, respectively. Auxiliary grey lines indicate the normal cytosolic Cl $^-$ or NO $_3^-$ concentrations before HCO $_3^-$ addition. Mean values and statistics are indicated in the text and in **Table 1**.

TABLE 1 | Membrane potential (mV), cytosolic NO₃ (mM), and cytosolic CI⁻ (mM) measured in mesophyll leaf cells of *P. oceanica* incubated in natural seawater (NSW) supplemented with different inorganic carbon concentrations.

	Membrane Potential (mV)	Cytosolic NO ₃ (mM)	Cytosolic Cl ⁻ (mM)
NSW	-170 ± 8 initial value	5.7 ± 0.2 initial concentration	9.7 ± 0.2 initial concentration
+3 mM HCO ₃	$-181 \pm 2^*$ initial hyperpolarization -168 ± 3 final value	$4.7 \pm 0.1^*$ final concentration	$6.3 \pm 0.3^*$ final concentration
+1000 ppmCO ₂	-174 ± 3 initial hyperpolarization -165 ± 3 final value	5.3 ± 0.6 final concentration	
+1000 ppmCO ₂ (0.1 mM EZ)	-178 ± 2 final value	5.6 ± 0.1 final concentration	

EZ is the carbonic anhydrase inhibitor, ethoxyzolamide. Values are mean \pm SD of a number of experiments indicated in the text. Asterisks (*) denote significant differences with respect to control conditions (NSW), the statistical values are indicated in the text.

cytosolic NO₃ (Figure 2). The match of membrane depolarization and the onset of cytosolic Cl and NO₃ diminution suggests the involvement of plasma membrane Stype anion channels, which are permeable to both anions and whose activation takes place when cells depolarize (Schmidt et al., 1995; Roelfsema et al., 2004; Roberts, 2006). In terrestrial angiosperm guard cells, these channels are activated when the concentration of HCO₃ (not CO₂) increases and the cytosolic pH (pHc) is alkaline (Xue et al., 2011). These are similar conditions to those observed in P. oceanica in the presence of high HCO₃ (Rubio et al., 2017). In order to rule out the role of CO₂, cytosolic NO₃ was measured in mesophyll leaf cells of P. oceanica incubated in NSW supplemented with 1000 ppm CO2, in the absence or in the presence of the plasma membrane-permeable carbon anhydrase inhibitor ethoxyzolamide (0.1 mM EZ; Sültemeyer et al., 1993). As shown in Figure 3 (data are summarized in Table 1), the increase of CO2 in NSW evoked a gradual plasma membrane hyperpolarization, reaching a value of -174 ± 3 mV (n = 5) after 7 min of 1000 ppm CO₂ treatment. This maximum hyperpolarization was lower and almost 2 min delayed compared to that induced by the treatment with 3 mM HCO₃, included in **Figure 3** as control (black trace). Cytosolic NO₃ also decreased after the addition 1000 ppm CO₂; however, in these conditions cytosolic NO₃ diminution was lower than that induced by 3 mM HCO₃ treatment. After 15 min in the presence of 1000 ppm CO₂, cytosolic NO₃ stabilized at a concentration of 5.3 ± 0.6 mM, a statistically non significant shift (n = 5; P = 0.37, Student t test). Furthermore, in the presence of 0.1 mM EZ, the enrichment of NSW with 1000 ppm CO₂ evoked a marked plasma membrane hyperpolarization that reached a steady maximum value of -178 ± 2 mV after 12 min treatment. In the presence of EZ, cytosolic NO₃ showed a minimum, statistically non-significant, shift from 5.7 ± 0.2 to $5.6 \pm 0.1 \text{ mM NO}_{3}^{-}$ (n = 4; P = 0.53, Student t test). That neither

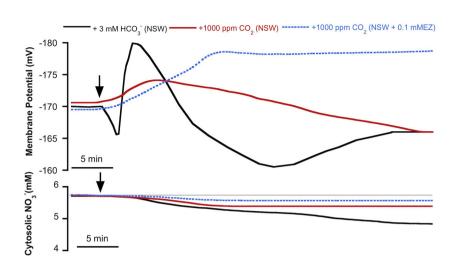


FIGURE 3 | Effect of inorganic carbon increase on the plasma membrane potential (*Em*, mV) and cytosolic NO₃ (mM) measured in *Posidonia oceanica* mesophyll leaf cells. Arrows indicate the onset of 3 mM HCO₃ addition to natural seawater (NSW, black traces, control conditions) and the addition of 1,000 ppm CO₂ to NSW (red traces) or to NSW containing 0.1 mM ethoxyzolamide (EZ, dashed blue traces). Traces are representative records of a minimum of four independent experiments using intracellular NO₃-selective microelectrodes. Mean values and statistics are indicated in the text and **Table 1**. Auxiliary grey line represents the normal cytosolic NO₃ concentration before the inorganic carbon additions.

the long-term depolarization nor the drop in cytosolic NO_3^- content evoked by HCO_3^- were evident with CO_2 in the presence of EZ (thus restricting HCO_3^- production from the CO_2 source) points to the need for substantial cytosolic HCO_3^- accumulation to effect ionic fluxes. Accordingly, these results strongly indicate that the cytosolic NO_3^- decrease is caused by the HCO_3^- enrichment and point to the S-type anion channel activation by the cytosolic HCO_3^- (not CO_2) increase.

Effect of HCO₃ Increase on External NO₃

The activation of S-type anion channels may allow the efflux of NO_3^- from mesophyll leaf cells; such a phenomenon of NO_3^- efflux should be higher in the case of NO_3^- -replete cells. Thus, in order to investigate if the HCO_3^- enrichment of NSW promoted NO_3^- efflux from *P. oceanica* leaves, the time course of external

NO₃⁻ concentration change was monitored in assay medium containing leaves from plants pre-incubated for 2 days in NSW containing 100 μM NO₃⁻. NO₃⁻-supplied leaves were then incubated in NSW containing the standard NO₃⁻ concentration (10 μM). After the addition of 3 mM HCO₃⁻, external NO₃⁻ concentration increased significantly (10.8 ± 0.04 μM NO₃⁻; n = 6; P < 0.001, Student t test) within the first minute of incubation. Maximum net efflux rate was 18 ± 2 nmol NO₃⁻·g_{FW}⁻¹·min⁻¹. Then, external NO₃⁻ decreased at a net rate of 8 ± 1 nmol NO₃⁻·g_{FW}⁻¹·min⁻¹ until 5 min of incubation to stabilize at a concentration of 9.2 ± 0.1 μM NO₃⁻ (**Figure 4**). Under control conditions (no HCO₃⁻ addition), no external NO₃⁻ increase was detected, but NO₃⁻ concentration progressively decreased at a net rate of 11 ± 2 nmol NO₃⁻·g_{FW}⁻¹·min⁻¹ during the first 10 min of incubation to reach a steady lower value of 7.6 ± 0.2 μM NO₃⁻ (n = 6;

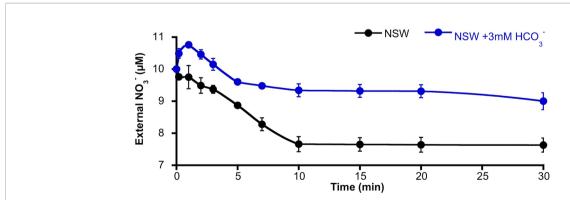


FIGURE 4 | Time course of the external nitrate concentration around *Posidonia oceanica* leaf pieces incubated in NSW. Plants were previously incubated in NSW containing 100 µM NO₃, then excised leaves were incubated in NSW (black trace, control condition) or NSW supplemented with 3 mM HCO₃ (blue trace). At different times within the first 30 min of incubation, samples were taken and used to determinate external NO₃ concentration. Data are mean ± SD of six independent assays. Mean values, net efflux, and influx NO₃ rates and statistics are indicated in the text.

P = 0.002, Student t test). External NO_3^- concentrations showed higher values in all samples from leaves incubated in NSW enriched with 3 mM HCO_3^- suggesting a net NO_3^- efflux from P. oceanica leaves under these conditions.

DISCUSSION

Cytosolic NO₃ Changes During Light–Dark Transitions in *P. oceanica* Leaf Cells

The concentration of NO₃ in the cytosol (NO₃c) depends on a series of highly regulated processes at the cellular level that includes sensing and transport at the plasma membrane (uptake and efflux), subcellular compartmentalization and metabolism. Depending on the technique and the plant system used, a wide range of cytosolic NO₃ values has been reported with also a large degree of variability. Therefore some authors suggest that NO₃c would not be subject to homeostasis, as would be the case for cytosolic pH or Ca²⁺ for example (for discussion see Siddigi and Glass, 2002 or Miller and Smith, 2008). After the development of intracellular NO₃-selective microelectrodes by Miller and Zhen (1991) it is possible to perform continuous measurements of the cytosolic free ion activity of NO₃ with the possibility to obtain instantaneous responses to changes in the experimental conditions. This approach has been used here to report the first cytosolic NO₃ values of a marine vascular plant.

Under light conditions the cytosolic NO_3^- measured using intracellular NO_3^- -selective microelectrodes in *P. oceanica* mesophyll leaf cells was 5.7 ± 0.2 mM (n = 10), almost double the value reported for both epidermal and mesophyll leaf cells of *A. thaliana* (2.2 and 2.8 mM NO_3^- c, respectively; Cookson et al., 2005). This cytosolic NO_3^- in mesophyll leaf cells of *P. oceanica* is also much higher than that reported for the intermodal cells of the freshwater alga *C. corallina* (1.6 mM NO_3^- ; Miller and Zhen, 1991) or in the liverwort *C. conicum* (0.63 mM NO_3^- ; Trębacz et al., 1994), using the same technique. However, the NO_3^- c value found in *P. oceanica* is similar to those observed in root cells of barley (5.4 mM NO_3^- ; Zhen et al., 1991) or maize (3.1 mM NO_3^- ; Miller and Smith, 1996).

The NO $_3^-c$ value in mesophyll leaf cells of *P. oceanica* is lower than the cytosolic Cl $^-$ concentration (Cl ^-c) of 9.7 mM Cl $^-$, yielding a NO $_3^-c$ /Cl ^-c of 0.6. This ratio is higher than that calculated for *C. conicum*, NO $_3^-c$ /Cl $^-c \approx 0.1$, taking 7.4 mM as the Cl ^-c (Trębacz et al., 1994), but it is in the range of the estimates for root cells of barley (NO $_3^-c$ /Cl $^-c \approx 0.9$) or *A. thaliana* mesophyll leaf cells (NO $_3^-c$ /Cl $^-c \approx 0.3$). Those values have been calculated considering Cl ^-c as 6 mM for barley root cells (Britto et al., 2004) and assuming that the reported cytosolic Cl $^-$ in *A. thaliana* root cells (8.7 mM Cl ^-c ; Planes et al., 2015) would be the same in mesophyll leaf cells.

Changes in environmental conditions have been related to NO₃⁻c alterations, supporting the idea that the cytosol operates a strong ion homeostasis not only for pH, Pi or Ca²⁺, but also for NO₃⁻ (reviewed by Miller and Smith, 2008). Light-dark transition triggers a slight increase of NO₃⁻c in the liverwort *C. conicum* (Trębacz et al., 1994) and a transient

increase, from 2 to 3.5 mM, which peaked after 7 min of darkness in the case of A. thaliana leaf cells (Cookson et al., 2005). A similar response has been found here in P. oceanica mesophyll leaf cells, in which cytosolic NO₃ reached a maximum value of 7.2 mM after 25 min of dark treatment. In the case of A. thaliana, since the effect was not observed when measurements were performed in the nitrate reductase (NR) mutant (nia1nia2) leaf cells, the NO₃c increase was explained because the shift to the dark inactivates NR, leading to a transient build-up of NO₃c due to a slower reduction rate (Cookson et al., 2005). A higher NO₃c in A. thaliana nia1nia2 mesophyll leaf cells and a similar time course of cytosolic increase to the rate of NR activity change in response to illumination transitions (half-life of 2 to 15 min in spinach; Huber et al., 1992; Kaiser et al., 1992; Riens and Heldt, 1992) support the evidence for the role of NR in regulating NO₃c (Cookson et al., 2005). In P. oceanica, the NO₃c increase observed in the dark could also explain the one-half diminution of the maximum high affinity NO₃ uptake observed previously in mesophyll leaf cells of this plant (Rubio et al., 2018), due to the apparent substrate inhibition of the transporter.

Inorganic Carbon Increase Triggers Cytosolic NO₃ Decrease

In a previous work we demonstrated that a plasma membrane nH^+/HCO_3^- symporter mediates the uptake of HCO_3^- in P. oceanica mesophyll leaf cells. Further, the direct uptake of HCO₃ followed by its internal dehydration renders CO₂ (used for photosynthesis) and hydroxyl anions, promoting membrane potential and cytosolic pH and Cl⁻ variations (Rubio et al., 2017). In the present work, we have also verified that uptake of HCO₃ promotes the decrease of NO₃c in P. oceanica mesophyll leaf cells. The increases of cytosolic HCO₃ and cytosolic pH have been related to promoting the opening of plasma membrane Stype anion channels in guard cells (Xue et al., 2011). A similar pathway seems to explain the NO₃c diminution observed in response to HCO₃ addition in *P. oceanica* mesophyll leaf cells, as was previously proposed in the case of Cl⁻c (Rubio et al., 2017), used as a control in this work. In both cases, the onset of NO₃c and Cl⁻c decreases matched that of the plasma membrane depolarization, and indeed depolarization is a required initial phase to activate S-type anion channels operating at the plasma membrane of guard cells (Schmidt et al., 1995; Roelfsema et al., 2004; Roberts, 2006).

S-type anion channels are encoded by the small *SLAC/SLAH* gene family, that share homology to transport systems found in different kingdoms (Dreyer et al., 2012). In *A. thaliana*, apart from *SLAC1* that is exclusively expressed in guard cells, four additional homologs (*SLAH1-4*) are present (Negi et al., 2008). SLAC1 and SLAH3 channels exhibit a permeability preference for NO₃ and Cl⁻ but not for malate (Schmidt and Schroeder, 1994; Geiger et al., 2009; Chen et al., 2010). The SLAH3 channel has the highest permeability for NO₃ (NO₃/Cl⁻ permeability ratio of 20) and in contrast to SLAC1, SLAH3 also requires extracellular NO₃ to induce its activity (Geiger et al., 2011).

Furthermore, a role in NO_3^- -dependent alleviation of ammonium toxicity in *A. thaliana* roots has been proposed for SLAH3 (Zheng et al., 2015).

Although seagrasses have lost stomata differentiation genes (Olsen et al., 2016) the presence of S-type anion channels cannot be ruled out in those plants since, presumably, these anion channels have evolved as emergency valves, rapidly releasing excess osmolytes under stress conditions (Roelfsema et al., 2012). The *Zostera marina* genome, the first one available from a seagrass (Olsen et al., 2016), contains two homologs for the *A. thaliana SLAC/SLAH* gene family (*Phytozome v12.1*, data base), one *SLAH1* homolog (*Zosma91g00860.1*) and one *SLAC1* homolog (*Zosma76g00610.1*). The presumed homologs of these channels in *P. oceanica* should be good candidates for the proposed channels for the leak of NO₃ and Cl⁻ ions from the cytosol, but further molecular analyses are needed to characterize the specific role of these anion channels to mediate anion efflux in response to high HCO₃.

In guard cells, stomatal closure in response to high CO2 is mediated by the activation of S-type (SLAC1/SLAH3) anion channels (reviewed by Hedrich and Geiger, 2017). Interestingly, guard cells do not sense CO2 themselves, but instead HCO3synthesized from CO₂ within the cytosol by carbonic anhydrases (Xue et al., 2011). The same case could be proposed in P. oceanica mesophyll leaf cells, since no significant NO₃c decrease was found in response to 1000 ppm CO₂ treatment in either the absence or the presence of the plasma membranepermeable carbon anhydrase inhibitor ethoxyzolamide, suggesting that HCO₃, not CO₂, is the inorganic carbon species that triggers NO₃c efflux from P. oceanica mesophyll leaf cells (Figure 5). Furthermore, probably due to the activation of the H⁺-pump in response to the cytosolic accumulation of CO₂, a weak acid, the 1000 ppm CO₂ treatment renders the hyperpolarization of the plasma membrane and the cytosolic acidification (see Rubio et al., 2017). Both responses are prolonged in the presence of ethoxyzolamide, that impairs HCO₃ production from the CO₂ source, leading to conditions that not support the activation of S-type anion channels described above and could explain the non-significant NO₃c diminution observed in response to CO₂ increase.

Cytosolic NO₃ Efflux Under High Inorganic Carbon Could Contribute to Biomass N-Impoverishment in Seagrasses

As we showed previously, direct HCO_3^- uptake has consequences for cytosolic ion homeostasis in *P. oceanica* mesophyll leaf cells (Rubio et al., 2017). The short-term net efflux of NO_3^- and the low NO_3^- net uptake rate observed in the present work also indicate that natural seawater HCO_3^- enrichment may be important not only for cytosolic ion homeostasis but for the metabolic flux of NO_3^- in this seagrass. In natural seawater, containing 500 mM Cl^- and $10~\mu M~NO_3^-$, the efflux of Cl^- and NO_3^- from *P. oceanica* mesophyll leaf cells is driven by the outwardly directed anion electrochemical potential gradients for both anions (**Figure 5**). This electrochemical potential gradient (in mV) is almost five-fold higher for NO_3^- (+330 mV) than for

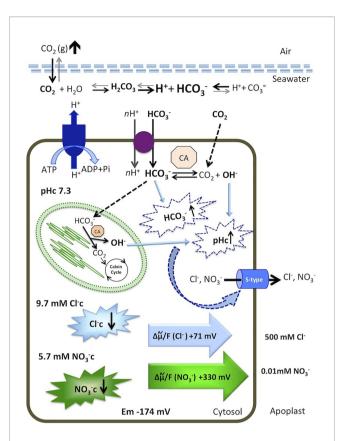


FIGURE 5 | Model for cytosolic NO $_3^-$ and Cl $^-$ responses to an increase of dissolved inorganic carbon in the marine angiosperm *Posidonia oceanica* leaf cells. Elevated atmospheric CO $_2$ level increases total dissolved inorganic carbon, shifting the equilibrium in favor of HCO $_3^-$ in the seawater. The HCO $_3^-$ is taken up by an H $^+$ -symport and dehydrated in the cytosol or transported to the chloroplast serving as inorganic carbon source for photosynthesis. Internal dehydration, catalyzed by the carbonic anhydrases (CA) renders CO $_2$, consumed by the Calvin Cycle, and OH $^-$ ions that increase cytosolic pH (pHc). Excess of cytosolic HCO $_3^-$ and elevated pHc activates S-type anion channels, through which Cl $^-$ and NO $_3^-$ leave the cells by an outwardly directed motive force of +71 mV and +330 mV, respectively. Consequently, the cytosolic concentrations of both anions decrease. Membrane potential, ion concentrations, calculations, and discussion are included in the text.

Cl $^-$ (+71 mV). However, in response to high HCO $_3^-$ the Cl $^-$ c decrease observed in *P. oceanica* mesophyll leaf cells (Δ Cl $^-$ c = 3.4 mM) was much higher than for NO $_3^-$ c (Δ NO $_3^-$ c = 1 mM). This lower NO $_3^-$ c leak than the expected from electrochemical potential gradient comparison with Cl $^-$ c could be explained by a different membrane permeability for both anions and/or different capability of compartmentalization. Nevertheless, even the observed decrease of NO $_3^-$ c, the short-term NO $_3^-$ efflux, and the low NO $_3^-$ uptake rate could impair N-assimilation in *P. oceanica* leaf cells in natural seawater containing high inorganic carbon.

An impaired metabolic flux of N should be relevant in the context of atmospheric CO_2 rise. Oceans have been the sink for 30% of CO_2 released during the industrial era, at a higher rate (3.8 GTons·year⁻¹) than the 1.8 GTons·year⁻¹ fixed by photosynthesis or the 2 GTons·year⁻¹ removed by abiotic absorption (Behrenfeld et al., 2002). Atmospheric CO_2 is

exchanged into aquatic environments rendering the dissolved inorganic carbon (DIC) equilibrium. Controlled by pH, this equilibrium generates the distribution of DIC species: dissolved CO₂, bicarbonate (HCO₃) and carbonate (CO₃²⁻) ions. Consequently, elevated atmospheric CO2 concentration increases the total DIC and lowers the pH, shifting the relative proportion of each DIC species. Under current ocean pH (~8.04) and atmospheric CO₂ (~ 410 ppm), the smallest pool of DIC is dissolved CO₂, but this will have the greatest increase (> 250%) among the DIC constituents as the pH drops (~ 0.3-0.4 pH units) under the predicted rise in atmospheric CO₂ (1000 ppm) for 2100. In contrast, the HCO₃ pool will only increase by 15% at that date (Koch et al., 2013). However, in terms of absolute concentration (mol·kg⁻¹) HCO₃ levels will rise more than dissolved CO2 (Raven et al., 2005). Using the predicted data for 2100, calculations render 2.05 mmol·kg⁻¹ HCO₃ and only 0.03 mmol·kg⁻¹ CO₂ (Koch et al., 2013). Considering such a HCO₃ rise (1.2 fold higher than actual concentration) and that the addition of 3 mM HCO₃ corresponds to an increase by 2.3 fold of the HCO₃ concentration in natural seawater, a NO₃ c leak of 0.5 mM would be expected in P. oceanica mesophyll leaf cells by 2100. However, long-term effect of elevated atmospheric CO₂ concentration on seagrasses N content needs further investigation, since N-deficiency induces high-affinity NO₃ uptake, which contributes to NO₃ homeostasis (reviewed by Rubio and Fernández, 2019).

Furthermore, the concomitant natural seawater pH decrease under elevated atmospheric CO2 could alter N availability in seawater and even the energy cost for nutrient uptake. A pH change from 8.1 to 7.8 evokes a decrease in the amount of NH₃ in the NH₄⁺/NH₃ ratio (Raven, 1986), whereas the amount of NO₃⁻ would not be affected (Zeebe and Wolf-Gladrow, 2001). Instead, the 0.3 external pH unit decrease renders a rise of the proton motive force at the plasma membrane of *P. oceanica* leaf cells of -18 mV, considering -174 mV as membrane potential, 7.3 as the cytosolic pH and constant temperature (Rubio et al., 2017). This amount represents, approximately, a 15% increase of the proton motive force (inwardly directed), which could prompt the activity of H⁺-dependent transport systems in this seagrass, including the plasma membrane H⁺/HCO₃ symporter (Rubio et al., 2017). As discussed by Rubio and Fernández (2019), ocean acidification could increase both the H⁺ and even the Na⁺ motive force due to changes in the activity of the plasma membrane Na⁺/ H⁺ antiporter found in seagrass species (Rubio et al., 2011). The activity of this antiporter generates a lower cytosolic Na+ concentration at acid external pH and, consequently, a rise of the inwardly directed Na⁺ motive force could be also expected, favoring high-affinity NO₃ uptake based on Na⁺-dependent transport systems in seagrasses (reviewed by Rubio and Fernández, 2019). In fact, a recent study in P. oceanica beds from the Gulf of Naples (Italy) shows that the long-term exposure (9 months) to acidified seawater (pH 7.82) under constant DIC conditions promotes the diminution of the C/N molar ratio, due to the increase of N content by 21% and 70% in leaves and rhizome, respectively, whereas C content of those organs is not affected by external pH acidification (Scartazza

et al., 2017). Those authors suggest that seawater acidification promoted a feed-forward long-term effect on N accumulation in P. oceanica, especially at the rhizome, although they recognize the need to specify the effect of acidification on the nutrient availability in their study (Scartazza et al., 2017). Interestingly, long-term nutrient enrichment seems to modulate the effects of ocean acidification on P. oceanica. Molecular analysis indicates that after 18 months at low pH (7.78) conditions the expression of nitrate transporter genes in *P. oceanica* leaves is altered; while NRT1_6.3 and NRT1_2.13 (involved in NO₃ sensing and low-affinity transport, respectively) are overexpressed the high-affinity NO₃ transporter gene NRT2 shows a downregulated expression (Ravaglioli et al., 2017). Thus, long-term overexpression of nitrogen transporter genes following nutrient additions at low pH suggests enhanced nutrient uptake and proposes that the effects of ocean acidification on P. oceanica depend upon local nutrient concentration (Ravaglioli et al., 2017).

Contrary to the acidification effect in P. oceanica meadows, several lines of evidence show that the most common effect of elevated CO₂ is a decrease in the dry mass concentration of N in plant tissue (Cotrufo et al., 1998; Curtis and Wang, 1998; Norby et al., 1999; Jablonski et al., 2002; Ainsworth and Long, 2005; Taub et al., 2008). This suggests that physiological changes leading to decreased biomass N under elevated CO₂ predominate in their effects over factors that would tend to increase N content (reviewed by Taub and Wang, 2008). Consequently, plants with non-saturated photosynthesis at actual atmospheric CO₂, show an ionomic imbalance, with N the main nutrient that decreased at high C assimilation (Loladze, 2014). The physiological mechanisms responsible for this phenomenon have not been well established yet, although different hypotheses are proposed to account for it. In terrestrial vascular plants, the best-supported are the decrease of specific N uptake and assimilation due to a diminution of the transpiration-driven mass flow of N by a decreased stomatal conductance at elevated CO₂ and the biomass dilution of N by increased photosynthetic assimilation of C (reviewed by Taub and Wang, 2008).

Seagrass meadows rank among the most productive ecosystems on Earth (Duarte and Chiscano, 1999), which largely contribute to C uptake in coastal waters and with most species capable of utilizing HCO₃ as a CO₂ source for photosynthesis (Raven et al., 2014). With the exception of Cymodocea nodosa (considered C4), seagrasses show C3 photosynthetic pathways and are not saturated at the current ocean DIC concentration (reviewed by Koch et al., 2013). As occurs in C3 terrestrial plants, a higher C assimilation occurs in seagrasses due to the rise of DIC discussed above (Borum et al., 2016). The unpredicted effect of HCO₃ enrichment on anion homeostasis in P. oceanica mesophyll leaf cells, leading to the short-term efflux of NO₃ from the cytosol, possibly through the activation of S-type anion channels, supports a new mechanism as a key consideration in understanding the expected biomass Nalteration in seagrasses under elevated DIC, and probably, for terrestrial plants growing in waterlogged alkaline environments.

DATA AVAILABILITY STATEMENT

The datasets generated for this study are available on request to the corresponding author.

AUTHOR CONTRIBUTIONS

LR, DG-P, and JF performed the experiments. DG-P and LR accomplished data analysis. LR, JD, and JF discussed the results, wrote and edited the manuscript. LR and JF conceived the project. All authors contributed to the article and approved the submitted version.

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- **Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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