1	Effects of processing parameters on chemical and physical properties of enzymatically
2	interesterified beef tallow-corn oil blends
3	A. Burcu Aktas <sup>a</sup> , Banu Ozen <sup>b</sup> *, Cristina Alamprese <sup>c</sup>
4	
5	Running title: Enzymatic interesterification of tallow-corn oil
6	
7	<sup>a</sup> Cumhuriyet University, Food Engineering Department, Sivas, Turkey
8	<sup>b</sup> Izmir Institute of Technology, Food Engineering Department, Urla-Izmir, Turkey
9	<sup>c</sup> Università degli Studi di Milano, Department of Food, Environmental and Nutritional
10	Sciences (DeFENS), via Celoria 2, Milan, Italy
11	
12	* corresponding author: <u>banuozen@iyte.edu.tr</u> , Orcid ID: <u>https://orcid.org/0000-0002-0428-</u>
13	<u>320X</u>
14	
15	Authors' e-mails:
16	A.Burcu Aktas: <u>burcuaktas@cumhuriyet.edu.tr</u> Orcid ID: <u>https://orcid.org/0000-0003-</u>
17	2520-0976
18	Cristina Alamprese: <u>cristina.alamprese@unimi.it</u> Orcid ID: <u>https://orcid.org/0000-0002-</u>
19	<u>9119-6752</u>
20	

### 21 Abstract

Purpose of this study is to manufacture structured lipids by enzymatic interesterification of 22 beef tallow with corn oil and to investigate effects of process parameters on chemical and 23 physical properties of products. Full factorial design was constructed by considering blend 24 25 ratio and reaction time as process parameters. Enzymatic interesterification was catalyzed with sn-1,3 specific lipase. Structured lipids have higher free fatty acid content and lower 26 oxidative stability compared to initial blends. Interesterification did not cause trans fatty acid 27 28 formation and interesterified lipids mostly contained β crystals. Solid fat content and slip 29 melting point decreased up to 6 h of interesterification; however, longer reaction times have negative effects on these parameters. Statistical analyses results confirmed that reaction time 30 is highly important for enzymatic interesterification. 31

32 Practical applications: Some of interesterified lipids can be utilized as alternatives to
33 margarines or butterfat due to their lower trans fatty acid content and crystal morphology.

34

35 Keywords: tallow, corn oil, enzymatic interesterification, sn-1,3 specific lipase

36

37

38

39

- 40
- 41

42

# 44 **1 Introduction**

Beef tallow is considered as a by-product of meat industry. It has relatively a high melting 45 point range (40-60 °C); therefore, it is classified as a hard fat. Beef tallow contains low levels 46 of polyunsaturated fatty acids (PUFA), which limits its use in food industry as well as in 47 direct human consumption (Kowalski et al., 2004). Thus, tallow needs to be modified in order 48 to obtain a product with desirable properties. One of the possible methods for tallow 49 50 modification is interesterification with vegetable oils (Kowalski et al., 2005). Both chemical and enzymatic interesterification reactions are able to alter the physical and chemical 51 properties of fats. Chemical interesterification provides new properties to the modified lipids 52 53 by the random incorporation or the restructuring of acyl residues of triacylglycerols (TAG) by the help of a chemical catalyst. However, enzymatic interesterification (EI) leads to the 54 attachment of specific fatty acids to specific positions of TAG structure to produce novel 55 products by lipase enzyme (Martin et al., 2010). In an interesterification reaction by a 1,3-56 specific lipase, initially a mixture of TAGs, 1,2- and 2,3-diacylglycerols, and free fatty acids 57 is produced. Then, acyl migration takes place due to prolonged reaction periods that cause 58 the formation of 1,3-diacylglycerols and this reaction also allows some randomization of the 59 fatty acids existing at the sn-2 position of the TAGs (Xu, 2000; Rajendran, 2009). 60

There are several studies using this process in the modification of tallow. In a previous study, enzymatic interesterification of tallow with sunflower and soybean oils did not cause formation of trans fatty acids (Foglia et al., 1993). The enzymatic interesterification of beef tallow and rapeseed oil resulted in lower thermal resistant properties with respect to the nonesterified tallow (Kowalska et al., 2015). Another study of interesterification of tallow and sunflower oil showed that the physical properties of tallow could be improved as a result of 67 this process (Rodríguez et al., 2001). Lipids produced by enzymatic interesterification of beef 68 tallow with soybean and palm oils were classified as low trans-fat margarines due to their 69 desirable physico-chemical properties and polymorphs (Li et al., 2018). It was not 70 encountered any study in the literature regarding the production of structured lipids using 71 enzymatic interesterification of beef tallow with corn oil.

The aim of this research is to modify beef tallow by enzymatic interesterification with corn oil and to determine the effects of process parameters (i.e., blend ratio and reaction time) on various chemical (free fatty acids, fatty acid profile, and oxidative stability) and physical (slip melting point, solid fat content, and polymorphic behavior) properties of the structured lipids.

# 77 2 Materials and methods

#### 78 2.1 Fat samples and reagents

Two different breeds of 2-years old calves (Montafon and Holstein) were the sources of beef
tallow used in interesterification reactions and it was obtained immediately after slaughter
and stored at -20°C. Corn oil was obtained from a local market. Lipase enzyme from *Thermomyces lanuginosus* (Lipozyme TL IM) was obtained from Sigma–Aldrich (St. Louis,
MO). All other reagents and solvents are of analytical or chromatographic grade and were
obtained from Sigma (Sigma-Aldrich, Germany).

# 85 2.2 Enzymatic interesterification process

A full factorial experimental design was employed to evaluate the effects of reaction time (0, 3, 6, 9, and 12 h) and tallow to corn oil blend ratio (60:40, 70:30 and 80:20) on chemical and physical properties of structured lipids (Table 1-2), as a result 18 different blends including 3 central points were prepared. The liquefied tallow was mixed with corn oil at given ratios. The reaction was initiated by adding 10% (w/w) enzyme (Lipozyme TL IM) at 55 °C (Ronne et al., 2005). Enzymatic interesterification was performed in a shaking incubator with stirring at 120 rpm (Sartorious, Certomat B5-1, Germany). Reaction was stopped by denaturation of lipase enzyme by keeping the samples in a shaking water bath at 80 °C for 30 min. The denatured enzyme was removed by vacuum filtration.

### 95 **2.3 Chemical property analyses**

# 96 2.3.1 Free fatty acid content

- 97 Titrimetric method specified in AOCS standard official method Ca 5a-40 was used in free
  98 fatty acid (FFA) determination of the products (AOCS, 1989). The analyses were performed
- 99 twice. Acidity was expressed as percentage of oleic acid.

# 100 2.3.2 Mono, di and triacylglycerol content determination

101 Mono- (MAG), di- (DAG) and triacylglycerol (TAG) contents of structured lipids were

analyzed according to AOCS Cd11C-93 method by column chromatography (AOCS, 2002).

### 103 **2.3.3 Oxidative stability**

104 Rancimat apparatus was used in measurement of the oxidation induction time (873 Biodiesel,

105 Metrohm, Switzerland). Sample was placed inside the glass reaction vessel for the

- 106 measurement. Carrier medium was deionized water and reaction temperature was set to 120
- 107 °C for both columns with a constant 20 L/h air flow. Stability was expressed as the oxidation

108 induction time (h) (Uncu & Ozen, 2016).

#### 109 2.3.4 Fatty acid composition

110 Fatty acid composition of the samples was determined after converting them into the

111 corresponding fatty acid methyl esters (FAME). Chromatographic analyses were performed

with a GC (Agilent 6890) equipped with an auto-sampler, a split/splitless (1:50) injector and

a FID detector. An HP 88 capillary column (100 m x 0.25 mm ID x  $0.2 \mu$ m) was used in the

analyses. Conditions for GC analysis was described in a previous paper (Meng et al., 2010).

115 Supelco 37 Component Mix was used as standard (Sigma-Aldrich, Germany).

### 116 **2.4 Physical property analyses**

# 117 2.4.1 Determination of slip melting points

Slip melting point (SMP) was determined according to AOCS Official method Cc 3-25(AOCS, 1989). The analyses were replicated twice.

#### 120 **2.4.2 Determination of solid fat content**

121 Solid fat content (SFC) was determined by a nuclear magnetic resonance (NMR)

- spectrometer (Bruker, USA) according to AOCS Official Method Cd 16b-93 (AOCS, 2017).
- 123 Samples were melted at 80 °C and recrystallized at 0 °C for 30 min. Then, they were stabilized
- for 30 min at various temperatures (10, 20, 30 and 35 °C) before measuring the liquid signal.

# 125 **2.4.3** Crystal morphology

126 The polymorphic structures of structured lipids were determined by X-ray diffraction

127 (Philips, Holland) using Cu as anode material ( $k = 1.54056 \text{ A}^\circ$ , voltage 45 kV, tube current

40 mA, fixed 1.0-, 1.0-, and 0.76-mm divergence, anti-scatter and receiving slits). Samples

were scanned from 4 to 50° (2 $\theta$  scale) at a rate of 2.0°/min at ambient temperature. The

130 analyses were replicated twice.

#### 131 **2.5 Statistical analyses**

The analytical data were analyzed by the analysis of variance (ANOVA) in order to investigate the effect of blend ratio, reaction time and their interaction on chemical and physical properties of the structured lipids (MODDE 11 software, MKS Umetrics, Umea, Sweeden). To investigate the effects of processing parameters on interesterification process the principal component analysis (PCA) was also used (SIMCA 14.1, MKS Umetrics, Umea, Sweeden). Constructed models were defined in terms of number of principal components (PC), R<sup>2</sup> and Q<sup>2</sup> (a measure showing predictive ability of the model).

# **3 RESULTS AND DISCUSSION**

#### 140 **3.1 Chemical properties**

141 The fatty acid composition of the interesterified samples during the process is given in Table 142 1. The major fatty acids in all interesterified lipids are oleic, palmitic, stearic, and linoleic acids. In general, EI did not cause sharp changes in the amounts of fatty acids of the structured 143 144 lipids during reaction period and similar result was also observed in previous studies (Rønne 145 et al., 2005; Svensson & Adlercreutz 2008; Silva et al., 2009). As in the former researches, 146 enzymatic interesterification of tallow with corn oil did not cause formation of trans-fatty 147 acids (TFA) (Forssell et al., 1992; Foglia et al., 1993). Tallow-corn oil blends had 0.6-0.7% trans-fat content before interesterification. The range of trans fatty acids for lipids during 148 interesterification varied from 0.1 to 1.3%. Only three out of 15 interesterified products had 149 150 1% or higher concentrations of trans-fatty acids. These results indicate that produced structured lipids are mostly suitable for the manufacturing of low-trans containing 151 shortenings, margarines and frying fats. Saturated fatty acid content (SFA) of tallow (57.8%) 152

153 decreased with EI with corn oil. Moreover, there were small fluctuations in monounsaturated 154 fatty acid (MUFA) percentages of enzymatically interesterified lipids during the process. 155 These changes can be associated with the activity of the sn-1,3 lipase enzyme. In general, tallow has SFAs located in sn-2 position (Forssell et al., 1992). Therefore, while SFAs were 156 157 kept at sn-2 position, the MUFA and PUFA were presumably released from their positions 158 throughout the EI, thus causing increases in MUFA or PUFA of the samples. The Food and Agricultural Organization/World Health Organization (FAO/WHO) and the European Union 159 160 Committee advise that the minimum PUFA/SFA ratio should be 1 for controlling the saturated fat consumption and encouraging the intake of MUFA and PUFA. While the 161 PUFA/SFA ratio of the enzymatically interesterified lipids ranged between 0.3-0.9, tallow 162 had a ratio of 0.1. This result indicated that PUFA content of tallow was increased by both 163 blend ratio and EI reaction. ANOVA results (Table 3) indicated that while the models 164 constructed for PUFA% and SFA% were significant with significant lack of fit at 95% 165 confidence interval, the models for MUFA% and TFA% were found not significant. Reaction 166 time did not have any prominent effect on the fatty acid contents of the samples. The ANOVA 167 168 table reveals that blend ratio has important effect on SFA and PUFA content of the structured lipids (Table 3). While the increase in blend ratio of tallow led to a decrease in PUFA content 169 of the interesterified fats, SFA content of the enzymatically interesterified lipids increased as 170 171 expected.

Enzymatically interesterified samples had lower oxidative stabilities (0.6-3.9 h) in comparison to initial non-esterified blends. Oxidation induction times of lipids decreased regardless of blend ratio especially after 6 h reaction time (Table 2). In previous studies, the decrease in oxidative stability of interesterified fats compared to the initial mixture was also generally observed (Martin et al., 2010; Bryś et al., 2014; Kowalska et al., 2014). The 177 methods that are used in the production or purification of structured lipids, the oil sources, 178 and the presence of antioxidants during the manufacturing are among the main factors that 179 affect the oxidative stability of structured lipids. Moreover, the structure of the TAGs, including fatty acid composition and positional distribution on the glycerol backbone as well 180 181 as the interaction of these factors, has important impact on the oxidative stability of the 182 structured products. In the present study, since the enzyme used in interesterification reaction (Lipozyme TL IM) has regiospecifity on sn-1,3 bonds of TAG molecules, reaction products 183 184 would not be glycerol but sn-2 MAGs. Since, tallow has SFAs located in sn-2 position, 185 MUFAs and PUFAs were presumably the ones that are released from their positions throughout the enzymatic interesterification, causing decrease in oxidation induction time of 186 structured lipids. This decrease is more remarkable between 3-6 h of reaction. However, after 187 6 h of reaction, there were some increases in oxidation induction time of the samples, which 188 189 can be associated with the rearrangement of PUFAs in DAG and TAG molecules. It can be 190 suggested that 3 h reaction time could be more suitable for EI of tallow if sole consideration would be the oxidative stability of interesterified products. ANOVA results for oxidative 191 192 stability indicated that the constructed model was significant with significant lack of fit; however, it still reveals the significance of reaction time on the oxidative stabilities of the 193 samples and causes a decrease in oxidation induction time (Table 3). 194

Free fatty acid percentages (FFA%) of the samples are listed in Table 2. FFA% of tallow was 1.2% while blends without interesterification have a FFA range of 0.6-0.8%. Generally, FFA% of interesterified lipids (5.7-24.4%) increased sharply compared to starting blends. This indicates that neutralization should be applied to samples after enzymatic interesterification. There was a drastic increase in FFA% up to 6 h regardless of blend ratio; however, fluctuations for this value was observed after 6 h depending on the blend ratio. The fluctuations can be associated with the activity of the enzyme. Throughout interesterification reactions, enzyme acts on fatty acids of the TAG molecules and leads to formation of DAG and MAG molecules. Therefore, increases and fluctuations in FFA% could be observed during reaction time. Increasing trend was also observed in previous studies (Rønne et al., 2005; Kowalska et al. 2014). According to the ANOVA table, the model constructed for FFA% is not significant, even if reaction time resulted to have an important effect on FFA content (Table 3).

208 MAG, DAG and TAG contents of the structured lipids were determined to examine 209 changes in the glycerol backbone that occurred by the action of Lipozyme TL IM during interesterification. MAG, DAG and TAG contents are expressed in relative percentages of 210 211 the overall content (Table 2). The results are in accordance with previous studies that 212 observed a decrease in TAG% after interesterification (Ledóchowska & Wilczyńska, 1998; Kowalska et al., 2005; Kowalska et al., 2014). Generally, TAG% of enzymatically 213 214 interesterified lipids were lower than their starting blends. There was a drastic decrease in TAG% up to 6 h of EI process. After that point, fluctuations in TAG% were observed with 215 216 respect to the blend ratio. Although DAG content of the samples increased up to 9 h reaction time, later on DAG% decreased (Table 2). The same trend was also observed for MAG 217 content of the samples up to 6 h reaction time and MAG% decreased after that point (Table 218 219 2). These changes in TAG, DAG, and MAG contents of the interesterified lipids could be explained again with the activity of Lipozyme TL IM. The decrease in TAG content and the 220 increase of DAGs and MAGs up to 6 h of reaction confirms that the enzyme works 221 effectively, attacking the fatty acids located at sn1,3 positions of TAGs and providing the 222 223 formation of MAGs and DAGs. With the increase in reaction time, decrease in DAG and 224 MAG contents revealed that the fatty acids are snatched from their positions by the enzyme

and they participate in the production of new TAG molecules. Moreover, the fluctuations inTAGs after 6 h of reaction supports this explanation.

For a better evaluation of EI reaction, a correlation between FFA% and DAG+MAG 227 contents of interesterified lipids was also evaluated (Fig. 1). The Pearson correlation 228 coefficient r was 0.90, thus indicating a linearly increasing trend between FFA content and 229 DAG+MAG% of the samples during EI reaction. Increase in both FFA% and DAG+MAG% 230 231 is also an indication of the activity of the enzyme. The enzyme released the fatty acids from 232 their specific positions and caused an increase in both FFA% and DAG+MAG% (Fig. 1). A similar trend was also observed in another study (Kowalski et al., 2004) and the increase in 233 234 FFA and MAG+DAG contents was correlated with the reaction temperature and time, and the enzyme concentration. Moreover, it was commented that the decrease in TAG% was 235 inversely proportional with the same factors. ANOVA results indicated that while the models 236 constructed for TAG% and MAG% were significant, the model for DAG% was not 237 238 significant. The ANOVA table reveals that only reaction time was a significant factor for the models (Table 3). 239

## 240 **3.2 Physical properties**

Slip melting point, solid fat content and crystal morphology of initial blends and structured
lipids with various tallow to corn oil blend ratios and EI reaction times were determined.
Results are presented in Table 2. EI caused a decline in SMPs of structured lipids compared
to initial blends and tallow (Table 2). The same decline has also been observed in previous
researches (Bhattacharyya et al., 2000; Kowalska et al., 2014; Kowalska et al., 2015). While
SMP of tallow was 47.0 °C, SMP range of enzymatically interesterified samples was 33.145.9 °C and for the non-esterified blends this range was 43.2-45.9 °C. SMP of the

248 enzymatically interesterified samples decreased up to 6 h reaction time regardless of blend 249 ratio. After that point, there are some fluctuations in SMP of the samples. In the first 6 h of 250 interesterification, a decrease in TAG content and an increase in DAG+MAG content were observed along with a rise in SMP of the samples. The correlation between SMPs and TAGs 251 252 (r =0.87) was satisfactory. As TAG content of the interesterified samples decreased, SMP of 253 the samples decreased too (Fig. 2). Therefore, SMP of the samples could be associated with TAGs that were restructured during enzymatic interesterification reactions. ANOVA results 254 255 indicated a significant (p<0.05) model for SMP, with non-significant lack of fit and a 256 significant effect of reaction time (Table 3).

Solid fat content (SFC) is a measure of the percentage of fat in crystalline (solid) 257 phase to total fat across a temperature gradient and is an important parameter to decide on 258 259 the appropriateness of lipids for a possible application. SFC data of both interesterified lipids and non-interesterified blends over the temperature range of 10–35 °C are listed in Table 2. 260 261 As expected, raising temperature caused a marked decrease in the SFC values regardless of the reaction parameters. SFC profiles of non-interesterified blends had an increasing trend 262 263 with the increasing amounts of tallow in the blends. Interesterified lipids tended to have lower SFC values compared to their physical blends. Similar trends were also observed in previous 264 studies (Chang et al., 2005; Jin et al., 2008; Kowalska et al., 2015; Li et al., 2018). The 265 266 decrease in the SFC of interesterified lipids could be attributed to decreased proportion of the high-melting and medium chain TAGs in the structure of lipids. In addition, lower SFC of 267 structured lipids compared to both tallow and non-interesterified blends can be associated 268 with the alteration of TAG structure and the melting temperature of different crystals. SFC 269 270 of structured lipids slightly decreased throughout the EI process. However, there was a sharp 271 increase in SFC of structured lipids at all temperatures after 12 h of reaction time. ANOVA

results indicated that only the constructed model for SFC at 35 °C was significant even if
with a significant lack of fit. Although model is not that good it still indicates that reaction
time and the interaction of time x blend ratio were the significant factors meaning that time
highly affects the SFC of structured lipids at 35 °C (Table 3).

276 The polymorphic forms of the structured lipids and blends are provided in Table 2. 277 The same crystal types were also observed in previous studies (Li et al., 2018; Meng et al, 2011; Jin et al., 2008). Tallow contains mixtures of  $\beta$  and  $\beta$ ' forms dominated by  $\beta$ ' form.  $\alpha$ 278 279 forms were only observed in the enzymatically structured lipids E712 and E812, which are the samples having long reaction times. The non-interesterified blends also contain both  $\beta$ 280 and  $\beta$ ' forms, but dominated by  $\beta$  form. However, after EI only  $\beta$  form existed in most of the 281 282 samples. However, long reaction times resulted in different polymorphs: sample having 60:40 blend ratio after 12 h reaction time had  $\beta+\beta$ ' forms, while 70:30 and 80:20 blend ratios 283 with the same reaction time had  $\beta + \alpha$  and  $\alpha$  polymorphs, respectively. The  $\beta$ ' form of the 284 crystals had a high melting point, between 17 and 69 °C, whereas the melting point of the  $\beta$ 285 form was 32-78 °C depending on the chain length of the fatty acids (Akoh, 2017). Generally, 286 287 the  $\beta$  and  $\beta'$  crystal types were formed throughout the reaction and the SMPs of the samples were linked to the melting points of these crystal types. 288

### 289 **3.3 Principal Component Analysis**

Principal component analysis (PCA) was applied to whole data including all measured physical and chemical properties. The model was constructed with 6 PCs,  $R^2 = 0.96$ , and  $Q^2$ = 0.62. While first PC explains 44% of variation the second PC accounted for 24% of variation. According to the score plot, there is a rough separation of the samples according to the reaction time (Fig. 3a). While the non-esterified samples located at the left upper part

295 of ellipse, samples produced in 3, 6, and 9 h of reaction time are placed right bottom of the center and 12 h samples are towards non-esterified blends. It seems that there is a reverse 296 trend in the properties of structured lipids at 12 h reaction time and these samples are getting 297 closer to non-esterified blends instead of moving farther apart. Therefore, reaction times 298 299 longer than 6 h is not necessary for this application. As the loading plot shows, the structured lipids produced in 3, 6, 9 h reaction time are separated from non-esterified and 12 h samples 300 since they have higher MUFA, FFA, MAG and DAG contents (Fig. 3b). On the other hand, 301 302 TAG content, OS, SMP and SFC of initial blends are higher compared to interesterified samples with reaction times of 3, 6 and 9 h. The sample containing 70% tallow and 303 interesterified for 12 h (E712) is separately placed on the right side of ellipse in the score plot 304 305 due to its low trans-fat content (Fig. 3a). The multivariate analysis of the whole data also confirmed that reaction time is an important parameter for the enzymatic interesterification 306 307 reaction.

#### 308 CONCLUSION

309 Tallow and corn oil were used as substrates in the production of the enzymatically interesterified lipids. The enzymatic interesterification caused sharp decreases in oxidation 310 induction time of structured lipids. However, after 6 h of reaction, there were some increases 311 312 in oxidation induction time of the samples that can be associated with the rearrangement of PUFAs in DAG and TAG backbone. Generally, FFA content of the enzymatically 313 314 interesterified lipids increased significantly compared to starting blends. This means that 315 neutralization should be applied to the samples after the enzymatic interesterification. The 316 enzymatic interesterification of tallow with corn oil did not cause formation of trans-fatty acids. In general, reaction time longer than 6 h had a trend changing effect on the several 317

physical properties. The univariate and multivariate analyses of the results confirmed that reaction time is highly important for the EI reaction. It was observed that 12 h reaction time caused negative effect on the chemical and physical properties of the structured lipids. The structured lipids manufactured by the enzymatic interesterification of tallow and corn oil could be used in bakery industry since these lipids have desired  $\beta$  and  $\beta'$  polymorphic forms and low trans-fatty acid contents. Moreover, these structured lipids can be utilized as alternative products instead of margarines or butterfat.

# 325 Conflict of interest

326 The authors declare that they have no conflict of interest.

327

### 328 Acknowledgement

329 This study was supported by Izmir Institute of Technology Scientific Research Projects

330 (IYTE SRP) Program (Project No: 2017-IYTE-3)

331

# 332 **REFERENCES**

- Akoh, C. C. (2017). *Food lipids: chemistry, nutrition, and biotechnology*. CRC press.
- AOCS, (1989). AOCS Method Cc 3-25. Slip Melting Point, Standard Open Tube Melting
- Point. In Official Methods and Recommended Practices of the AOCS (5<sup>th</sup> ed.)
- 336 Champaign: *The American Oil Chemists' Society*.
- AOCS, (1989). AOCS Method Ca 5a-40. Determination of Free Fatty Acid Content, In
- 338 Official Methods and Recommended Practices of the AOCS (4<sup>th</sup> ed.) Champaign:

339 The American Oil Chemists' Society.

	340	AOCS, (1999).	. AOCS Method	Cd 16b-93.	Determination	of Solid Fat	Content, Ir	n Officia
--	-----	---------------	---------------	------------	---------------	--------------	-------------	-----------

- Methods and Recommended Practices of the AOCS (3<sup>rd</sup> ed.) Champaign: *The American Oil Chemists' Society*.
- AOCS, (2002). AOCS Method Cd 11bc-93. Determination of Mono-di-triacylglycerol
- 344 content, In Official Methods and Recommended Practices of the AOCS (3<sup>rd</sup> ed.)
  345 Champaign: *The American Oil Chemists' Society*.
- Bezerra, C. V., da Cruz Rodrigues, A. M., de Oliveira, P. D., da Silva, D. A., & da Silva, L.
- 347 H. M. (2017). Technological properties of amazonian oils and fats and their
  348 applications in the food industry. *Food Chemistry*, 221, 1466-1473.
- Bhattacharyya, S., Bhattacharyya, D. K., & De, B. K. (2000). DK Bhattacharyya,
  Modification of tallow fractions in the preparation of edible fat products. *European*

*Journal of Lipid Science and Technology*, *102(5)*, 323-328.

Bryś, J., Wirkowska, M., Górska, A., Ostrowska-Ligęza, E., & Bryś, A. (2014). Application
of the calorimetric and spectroscopic methods in analytical evaluation of the human
milk fat substitutes. *Journal of Thermal Analysis and Calorimetry*, *118(2)*, 841-848.

- 355 Chang, T., Lai, X., Zhang, H., Søndergaard, I., & Xu, X. (2005). Monitoring lipase-catalyzed
- interesterification for bulky fat modification with FT-IR/NIR spectroscopy. *Journal of Agricultural and Food Chemistry*, *53(26)*, 9841-9847.
- Fauzi, S. H. M., Rashid, N. A., & Omar, Z. (2013). Effects of chemical interesterification on
  the physicochemical, microstructural and thermal properties of palm stearin, palm
  kernel oil and soybean oil blends. *Food Chemistry*, *137(1-4)*, 8-17.
- Foglia, T. A., Petruso, K., & Feairheller, S. H. (1993). Enzymatic interesterification of
  tallow-sunflower oil mixtures. *Journal of the American Oil Chemists' Society*, *70(3)*,

**363 281-285**.

364	Forssell, P., Kervinen, R., Lappi, M., Linko, P., Suortti, T., & Poutanen, K. (1992). Effect of
365	enzymatic interesterification on the melting point of tallow-rapeseed oil (LEAR)
366	mixture. Journal of the American Oil Chemists' Society, 69(2), 126-129.

- Jin, Q., Zhang, T., Shan, L., Liu, Y., & Wang, X. (2008). Melting and solidification properties
- of palm kernel oil, tallow, and palm olein blends in the preparation of shortening. *Journal of the American Oil Chemists' Society*, *85(1)*, 23-28.
- Kowalska, D., Gruczynska, E., & Kowalska, M. (2015). The effect of enzymatic
  interesterification on the physico-chemical properties and thermo-oxidative stabilities
  of beef tallow stearin and rapeseed oil blends. *Journal of Thermal Analysis and Calorimetry*, *120*(1), 507-517.
- Kowalska, M., Żbikowska, A., & Kowalski, B. (2014). Enzymatically modified fats based
  on mutton tallow and rapeseed oil suitable for fatty emulsions. *Journal of the American Oil Chemists' Society*, *91(10)*, 1703-1710.
- Kowalski, B., Tarnowska, K., & Gruczynska, E. (2005). The properties of the mixture of beef
  tallow and rapeseed oil with a high content of tallow after chemical and enzymatic
  interesterification. *Grasas y Aceites*, *56(4)*, 267-275.
- Kowalski, B., Tarnowska, K., Gruczynska, E., & Bekas, W. (2004). Chemical and enzymatic
   interesterification of a beef tallow and rapeseed oil equal-weight blend. *European Journal of Lipid Science and Technology*, *106(10)*, 655-664.
- Ledóchowska, E., & Wilczyńska, E. (1998). Comparison of the oxidative stability of
  chemically and enzymatically interesterified fats. *Lipid/Fett, 100(8),* 343-348.
- Li, Y., Zhao, J., Xie, X., Zhang, Z., Zhang, N., & Wang, Y. (2018). A low trans margarine fat analog to beef tallow for healthier formulations: Optimization of enzymatic interesterification using soybean oil and fully hydrogenated palm oil. *Food Chemistry*,

*255*, 405-413.

- Martin, D., Reglero, G., & Señoráns, F. J. (2010). Oxidative stability of structured lipids.
   *European Food Research and Technology, 231(5)*, 635-653.
- 391 Meng, Z., Liu, Y. F., Jin, Q. Z., Huang, J. H., Song, Z. H., Wang, F. Y., & Wang, X. G.
- 392 (2011). Comparative analysis of lipid composition and thermal, polymorphic, and
- crystallization behaviors of granular crystals formed in beef tallow and palm oil. *Journal*of Agricultural and Food Chemistry, 59(4), 1432-1441.
- Meng, Z., Liu, Y., Shan, L., Jin, Q., & Wang, X. (2010). Reduction of graininess formation
- in beef tallow-based plastic fats by chemical interesterification of beef tallow and canola
  oil. *Journal of the American Oil Chemists' Society*, 87(12), 1435-1442.
- Oliveira, P. D., Rodrigues, A. M., Bezerra, C. V., & Silva, L. H. (2017). Chemical
  interesterification of blends with palm stearin and patawa oil. *Food Chemistry*, *215*, 369376.
- Rajendran, A., Palanisamy, A., & Thangavelu, V. (2009). Lipase catalyzed ester synthesis
  for food processing industries. *Brazilian Archives of Biology and Technology*, 52(1),
  207-219.
- 404 Rodríguez, A., Castro, E., Salinas, M. C., López, R., & Miranda, M. (2001).
- Interesterification of tallow and sunflower oil. *Journal of the American Oil Chemists' Society*, 78(4), 431-436.
- 407 Rønne, T. H., Yang, T., Mu, H., Jacobsen, C., & Xu, X. (2005). Enzymatic interesterification
- 408 of butterfat with rapeseed oil in a continuous packed bed reactor. *Journal of Agricultural*409 *and Food Chemistry*, 53(14), 5617-5624.
- 410 Silva, R. C., Cotting, L. N., Poltronieri, T. P., Balcão, V. M., de Almeida, D. B., Goncalves,
- 411 L. A., ... & Gioielli, L. A. (2009). The effects of enzymatic interesterification on the

- 412 physical-chemical properties of blends of lard and soybean oil. *LWT-Food Science and*413 *Technology*, 42(7), 1275-1282.
- 414 Svensson, J., & Adlercreutz, P. (2008). Identification of triacylglycerols in the enzymatic
  415 transesterification of rapeseed and butter oil. *European Journal of Lipid Science and*416 *Technology*, *110(11)*, 1007-1013.
- Uncu, O., & Ozen, B. (2015). Prediction of various chemical parameters of olive oils with
  Fourier transform infrared spectroscopy. *LWT-Food Science and Technology*, *63*(2),
  978-984.
- Xu, X. (2000). Production of specific-structured triacylglycerols by lipase-catalyzed
   reactions: a review. *European Journal of Lipid Science and Technology*, *102*(4), 287 303.

423

# 424 CAPTIONS TO FIGURES

Fig. 1. Correlation of free fatty acid content (FFA%) and mono and diacylglycerol content
(DAG+MAG%) of the structured lipids obtained by enzymatic interesterification of tallow
and corn oil

Fig. 2. Correlation of slip melting points (SMP) and triacylglycerol contents (TAG%) of the
structured lipids obtained by enzymatic interesterification of tallow and corn oil

430 Fig. 3. a) Score and b) loading plots of the PCA model constructed by using chemical and

431 physical parameters of enzymatically interesterified lipids obtained with tallow and corn oil

432 (in Fig. 3a samples are colored with respect to the blend ratio; see Tables 1-2 for sample

433 codes and abbreviations)









					Fatty A	Acid Re	lative Con	centratio	n (%)*				
Tallow to corn oil ratio (%)	Reaction Time (h)	Sample Code	MUFA	PUFA	SFA	TFA	C16:0	C18:0	C18:1n9c	C18:1n9t	C18:2n6t	C18:2n6c	PUFA/SFA
60	0	E60	35.6	29.9	34.5	0.6	16.5	16.2	33.5	0.5	0.0	29.7	0.9
70	0	E70	37.1	19.9	43.0	0.7	18.7	22.0	34.7	0.6	0.1	19.5	0.5
80	0	E80	38.0	13.9	48.1	0.7	19.9	25.5	35.5	0.6	0.1	13.5	0.3
60	3	E63	38.9	25.4	35.8	0.7	16.3	17.3	36.3	0.7	0.1	25.1	0.7
70	3	E73	41.4	20.4	38.2	1.3	16.6	18.9	38.1	1.3	0.0	20.2	0.5
80	3	E83	36.8	13.4	49.8	0.2	18.5	28.2	34.9	0.1	0.1	12.9	0.3
60	6	E66	39.4	25.3	35.3	0.7	15.8	17.3	36.9	0.7	0.1	25.0	0.7
70	6	E76	42.4	21.0	36.7	1.1	16.3	18.3	39.4	1.1	0.0	21.0	0.6
80	6	E86	42.6	15.7	41.8	0.1	17.3	21.6	40.6	0.1	0.1	15.3	0.4
60	9	E69	38.7	26.4	34.9	1.0	16.4	16.4	36.1	1.0	0.0	26.4	0.8
70	9	E79	36.9	18.2	44.8	0.6	17.3	24.9	34.6	0.6	0.0	18.0	0.4
80	9	E89	40.5	14.1	45.4	0.8	18.1	24.4	37.4	0.7	0.1	13.7	0.3
60	12	E612	34.3	25.2	40.5	0.7	18.6	18.8	31.1	1.4	0.1	25.0	0.6
70	12	E712	37.7	28.9	33.3	0.0	16.2	14.8	37.7	0.0	0.0	28.9	0.9
80	12	E812	35.8	14.5	49.8	0.9	20.4	25.3	32.8	0.8	0.2	14.0	0.3
70	6	ECP1**	39.9	20.3	39.8	0.1	16.9	20.3	37.7	0.0	0.1	20.1	0.5
70	6	ECP2**	40.9	21.6	37.5	0.2	16.4	18.5	38.6	0.2	0.0	21.4	0.6
70	6	ECP3**	40.7	20.3	39.0	0.2	16.7	19.8	38.3	0.1	0.0	20.1	0.5
		Tallow	38.9	3.3	57.8	0.5	22.5	32.1	30.6	0.5	0.1	3.0	0.1
		Corn oil	32.01	54.8	13.1	0.5	11.2	1.9	30.6	0.5	0.00	54.6	4.2

47 MUFA: monounsaturated fatty acid,; PUFA: polyunsaturated fatty acid; SFA: saturated fatty acid; TFA: trans fatty acid

\* Standard deviation values for MUFA=±0.43%; PUFA=±0.61%; SFA=±0.97%; TFA=±0.04%; C16:0=±0.2%, C18:0=±0.76%, C18:1n9c=±0.34%,

49 C18:1n9t= $\pm 0.05\%$ , C18:2n6t= $\pm 0.01\%$ , C18:2n6c= $\pm 0.61\%$  (calculated from ECPs)

50 \*\*ECP=central point

Tallow													
to corn													
oil ratio	Reaction	Sample	OS*	FFA*	TAG*	DAG*	MAG*	SFC*	SFC*	SFC*	SFC*	SMP*	Crystal
(%)	Time (h)	Code	(h)	(%)	(%)	(%)	(%)	(10°C)	(20°C)	(30°C)	(35°C)	(°C)	Morphology
60	0	E60	6.7	0.6	85.5	5.9	0.3	27.8	17.5	9.4	6.2	43.2	β+β'
70	0	E70	8.5	0.6	86.8	2.7	4.1	40.1	26.6	15.0	10.1	45.1	β+β'
80	0	E80	10.0	0.8	85.5	0.6	8.5	46.8	32.4	18.5	12.6	46.0	β+β'
60	3	E63	3.1	12.8	67.0	21.7	11.9	22.1	12.1	5.6	2.1	38.1	β
70	3	E73	1.8	17.8	69.0	23.0	7.8	25.0	16.5	5.2	2.8	38.2	β
80	3	E83	3.4	11.2	61.8	19.4	12.8	20.2	16.7	8.2	3.9	40.4	β'
60	6	E66	0.6	19.0	54.4	26.1	15.9	20.6	11.2	0.2	0.1	34.6	β
70	6	E76	0.8	25.6	49.8	30.4	22.3	25.7	16.4	0.9	1.2	33.1	β
80	6	E86	1.9	15.4	63.2	24.8	11.5	29.5	17.9	4.6	0.7	38.4	β
60	9	E69	1.4	5.7	68.4	14.0	12.1	21.0	12.7	6.8	3.8	40.0	β
70	9	E79	2.1	20.0	56.5	31.4	18.7	24.3	15.2	2.6	0.6	36.1	β
80	9	E89	1.8	20.5	48.0	21.1	17.7	35.7	23.3	7.4	2.1	38.7	β
60	12	E612	2.8	12.5	70.6	15.6	13.1	25.9	17.0	8.9	5.2	36.4	β+β'
70	12	E712	1.8	12.6	56.3	26.1	17.3	33.2	25.7	10.0	0.4	33.2	$\alpha + \beta$
80	12	E812	1.9	20.7	63.2	16.7	17.3	39.7	28.6	13.0	0.5	39.9	α
70	6	ECP1**	0.9	21.9	56.7	24.3	11.0	28.1	19.2	5.2	0.7	33.6	β
70	6	ECP2**	1.1	24.4	61.2	22.6	11.8	27.9	20.8	5.2	0.8	36.7	β
70	6	ECP3**	1.3	23.2	60.3	22.7	9.2	29.3	19.0	4.4	0.4	36.9	β
		Tallow	4.8	1.2	97.9	0.5	0.9	51.1	42.7	24.0	17.3	47.0	β+β'
		Corn oil	4.9	0.1	92.1	2.1	0.4						

Table 2 Results of chemical and physical analyses of the blends and enzymatically interesterified lipids obtained with tallow and corn oil

OS: oxidative stability; FFA: free fatty acid; TAG: triacylglycerol; DAG: diacylglycerol; MAG: monoacylglycerol; SFC: solid fat content; SMP: slip melting point

\* Standard deviation values for OS= $\pm 0.17$ ; FFA= $\pm 1.01$ ; TAG= $\pm 1.95$ ; DAG= $\pm 0.79$ ; MAG= $\pm 1.08$ , SFC10°C= $\pm 0.62$ , SFC20°C = $\pm 0.81$ , SFC30°C = $\pm 0.38$ , SFC35°C = $\pm 0.17$ , SMP= $\pm 1.5$  (calculated from ECPs)

57 \*\*ECP=central point

58

Table 3 ANOVA table showing the effect of process parameters on some chemical and physical properties of structured lipids produced by enzymatic

	ANOVA Table													
	OS	FFA	TAG	DAG	MAG	MUFA	PUFA	SFA	TFA	SFC (10°C)	SFC (20°C)	SFC (30°C)	SFC (35°C)	SMP
p value- model	0.02	0.14	0.05	0.20	0.01	0.81	0.00	0.00	0.84	0.12	0.06	0.40	0.02	0.05
p-value- lack of fit	0.00	0.01	0.13	0.08	0.93	0.06	0.02	0.06	0.72	0.01	0.05	0.08	0.00	0.21
R <sup>2</sup>	0.48	0.32	0.42	0.28	0.54	0.07	0.81	0.62	0.46	0.33	0.40	0.19	0.52	0.42
$\mathbf{R}_{\mathrm{adj}}^2$	0.37	0.17	0.30	0.12	0.44	-0.14	0.77	0.54	0.34	0.19	0.28	0.01	0.41	0.29
p value- factors														
blend ratio	0.53	0.46	0.45	0.98	0.27	0.41	0.00	0.00	0.54	0.02	0.01	0.18	0.78	0.29
reaction time	0.00	0.04	0.01	0.04	0.00	0.64	0.51	0.89	0.77	0.63	0.87	0.36	0.01	0.01
BR*RT	0.41	0.35	0.50	0.56	0.86	0.87	0.38	0.46	0.63	0.84	0.98	0.58	0.05	0.89

interesterification of beef tallow with corn oil

OS: Oxidative stability; FFA: free fatty acidity; TAG: triacylglycerol; DAG: diacylglycerol; MAG: monoacylglycerol; MUFA: monounsaturated fatty acids; PUFA: polyunsaturated fatty acids; SFA: saturated fatty acids; SFC: solid fat content; SMP: slip melting point

64

65

66