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## The Chemical Constituents of *Ellipeia cuneifolia* and Their Antibacterial Activity (Komposisi Kimia *Ellipeia cuneifolia* dan Aktiviti Antibakterianya)

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### ABSTRACT

*Chromatographic purification of chloroform extract of the twigs of Ellipeia cuneifolia has led to the discovery of three compounds comprising of 2',4'-dihydroxy-4,6'-dimethoxychalcone; tepanone; and O-methylmoschatoline. Structures of the compounds were established by interpreting their spectral data and by comparing them with those of the literature. Two of them showed antibacterial activities.*

*Keywords: Annonaceae; antibacterial activity; chalcone; Ellipeia cuneifolia; oxoaporphine; retrochalcone; twig*

### ABSTRAK

*Penulenan kromatografi terhadap ekstrak kloroform ranting Ellipeia cuneifolia membawa kepada penemuan 3 sebatian iaitu 2',4'-dihidroksi-4,6'-dimetoksikalkon; tepanon; dan O-metilmoskatolina. Struktur sebatian ini telah ditentukan dengan mentafsirkan data spektrum dan membandingkannya dengan data spektrum daripada penerbitan. Dua daripadanya menunjukkan aktiviti antibakteria.*

*Kata kunci: Aktiviti antibakteria; Annonaceae; Ellipeia cuneifolia; kalkon; oksoaporfina; ranting; retrokalkon*

### INTRODUCTION

*Ellipeia* Hook & Thom. of the family Annonaceae is a very small genus of climbers with about merely five species and known to be distributed in Peninsular Malaysia, Borneo and Sumatra (Kessler 1993). In Peninsular Malaysia only one species of *E. cuneifolia* occurs, mostly in the north-eastern coast of peninsula (Ridley 1922; Sinclair 1955). Locally the plant is known as *tepan* and according to local herbalists, a decoction of the roots has been used for post-parturition. Previously, Colegate et al. (1992) reported the isolation of a new retrochalcone, tepanone. A reinvestigation of the plant collected from the same locality showed two chalcones and one alkaloid, besides tepanone ((*2E*)-1-phenyl-3-(2'-hydroxy-3',4',6'-trimethoxyphenyl) prop-2-enone). In this study, we report the results of a phytochemical investigation of the twigs. Compounds 1 showed inhibitory against *Bacillus subtilis*, *Enterobacter aerogenes* and *Escherichia coli*, whereas compound 3 exhibited very strong activities against *B. subtilis* and *Staphylococcus aureus*.

### MATERIALS AND METHODS

#### GENERAL

The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded with Avance III 600 MHz Bruker in  $\text{CDCl}_3$  and acetone- $d_6$  with the chemical shifts,  $\delta$ , in ppm and the values of coupling constants,  $J$  in Hz. LC-MSToF spectra were taken on the Dionex/Bruker Micro ToFQ, FTIR spectra were recorded

on the Perkin Elmer Spectrum 400 FT-IR/FT-NIR and UV spectrophotometer Shimadzu UV-160 (200–400 nm).

#### PLANT MATERIAL

Twigs of *Ellipeia cuneifolia* were collected in October 2011 from the coastal area of Kota Bharu, Kelantan. A voucher specimen (ALM 3142) was deposited at the herbarium of Universiti Kebangsaan Malaysia, Bangi.

#### EXTRACTION AND ISOLATION

Dried ground twigs of *Ellipeia cuneifolia* (0.95 kg) were steeped three times in chloroform at room temperature over three days each to give 7.86 g (0.83%) of a dark-green extract after solvent removal by a rotary evaporator. The  $\text{CHCl}_3$  extract was subjected to vacuum liquid chromatography (VLC) by using silica 7747 (Merck) eluted with increasing polarity of *n*-hexane and ethyl acetate. The eluates were combined based on their silica gel thin layer chromatography (TLC) (Merck 5554) profile to yield seven fractions (A–G). Fraction D (0.48 g) was purified by column chromatography (CC), preparative TLC and Sephadex LH-20 to give compounds 1 (3.4 mg). Fractions E (0.85 g) and G (0.56 g) were separately purified by CC to give compounds 2 (3.2 mg) and 3 (5.8 mg), respectively.

2',4'-Dihydroxy-4,6'-dimethoxychalcone 1 (3.4 mg): white needles;  $R_f = 0.78$  (4:6, *n*-hexane-ethyl acetate); ESI-MS ( $m/z$ ): 323.0891  $[\text{M}+\text{Na}]^+$ ,  $\text{C}_{17}\text{H}_{16}\text{O}_5$ ; UV (MeOH)  $\lambda_{\text{max}}$ : 226, 292, 357 nm; FTIR (ATR)  $\text{cm}^{-1}$ : 3241, 2940, 1623, 1609, 1511, 1196, 1109, 789;  $^1\text{H}$  NMR (acetone- $d_6$ , 600 MHz)  $\delta_{\text{H}}$ :

7.90 (1H, *d*, *J* = 15.3 Hz, H- $\alpha$ ), 7.74 (1H, *d*, *J* = 15.3 Hz, H- $\beta$ ), 7.68 (2H, *d*, *J* = 8.7 Hz, H-2, 6), 7.00 (2H, *d*, *J* = 8.7 Hz, H-3, 5), 6.06 (1H, *d*, *J* = 2.1 Hz, H-5'), 5.98 (1H, *d*, *J* = 2.1 Hz, H-3'), 3.96 (3H, *s*, 6'-OCH<sub>3</sub>), 3.86 (3H, *s*, 4-OCH<sub>3</sub>); <sup>13</sup>C-APT NMR (acetone-*d*<sub>6</sub>, 150 MHz)  $\delta_c$ : 192.2 (C-9), 168.1 (C-2'), 165.1 (C-4'), 163.4 (C-6'), 161.7 (C-4), 142.1 (C- $\beta$ ), 130.2 (C-2, 6), 128.1 (C-1), 125.0 (C- $\alpha$ ), 114.4 (C-3, 5), 105.4 (C-1'), 96.1 (C-3'), 91.3 (C-5'), 55.5 (6'-OCH<sub>3</sub>), 54.9 (4-OCH<sub>3</sub>).

Tepanone 2 (3.2 mg): yellow needles; *R*<sub>f</sub> = 0.69 (4:6, *n*-hexane-ethyl acetate); ESI-MS (*m/z*): 337.1005, C<sub>18</sub>H<sub>18</sub>O<sub>5</sub>; UV (MeOH)  $\lambda_{max}$ : 224, 257, 364 nm; FTIR (ATR) cm<sup>-1</sup>: 3306, 2847-2998, 1639, 1551, 1466, 1337, 1110, 1203, 847, 793; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta_H$ : 8.23 (1H, *d*, *J* = 16.2 Hz, H- $\beta$ ), 8.04 (2H, *d*, *J* = 7.2 Hz, H-2', 6'), 8.00 (1H, *d*, *J* = 16.2 Hz, H- $\alpha$ ), 7.55 (1H, *t*, *J* = 7.2 Hz, H-4'), 7.48 (2H, *t*, *J* = 7.5 Hz, H-3', 5'), 6.09 (1H, *s*, H-5), 6.61 (1H, *s*, 2-OH), 3.94 (3H, *s*, 4/6-OCH<sub>3</sub>), 3.90 (3H, *s*, 4/6-OCH<sub>3</sub>), 3.88 (3H, *s*, 3-OCH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta_c$ : 191.9 (C-9), 156.9 (C-6), 154.0 (C-4), 150.6 (C-2), 139.1 (C-1'), 135.8 (C- $\beta$ ), 132.2 (C-4'), 129.7 (C-3), 128.5 (C-2', 6'), 128.4 (C-3', 5'), 122.5 (C- $\alpha$ ), 105.2 (C-1), 88.2 (C-5), 61.3 (3-OCH<sub>3</sub>), 56.0 (4/6-OCH<sub>3</sub>), 55.9 (4/6-OCH<sub>3</sub>).

*o*-Methylmoschatoline 3 (5.8 mg): orange needles. *R*<sub>f</sub> = 0.39 (4:6, *n*-hexane-ethyl acetate). ESI-MS (*m/z*): 322.1098 [M+H]<sup>+</sup>, C<sub>19</sub>H<sub>15</sub>O<sub>4</sub>N. FTIR (ATR) cm<sup>-1</sup>: 3127, 3069, 2948, 2858, 1657, 1578, 1465, 1384, 1202, 1089, 935, 755. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta_H$ : 9.11 (1H, *d*, *J* = 8.4 Hz, H-11), 8.97 (1H, *d*, *J* = 5.4 Hz, H-5), 8.58 (1H, *dd*, *J* = 1.2, 7.8 Hz, H-8), 8.22 (1H, *d*, *J* = 5.4 Hz, H-4), 7.75 (1H, *dt*, *J* = 1.2, 8.4 Hz, H-10), 7.54 (1H, *t*, *J* = 7.8 Hz, H-9), 4.20 (3H, *s*, 3-OCH<sub>3</sub>), 4.11 (3H, *s*, 2-OCH<sub>3</sub>), 4.09 (3H, *s*, 1-OCH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta_c$ : 182.7 (C-7), 156.5 (C-1), 148.5 (C-3), 147.4 (C-2), 145.5 (C-6a), 144.6 (C-5), 134.6 (C-11a), 134.4 (C-10), 131.4 (C-7a), 131.1 (C-11c), 129.0 (C-8), 128.2 (C-9), 127.7 (C-11), 122.8 (C-3a), 119.2 (C-4), 115.7 (C-11b), 61.8 (3-OCH<sub>3</sub>), 61.5 (2-OCH<sub>3</sub>), 61.0 (1-OCH<sub>3</sub>).

#### BIOASSAY

The crude extract and pure compounds were assayed for antibacterial activity against the bacteria *Bacillus subtilis* (ATCC 11774), *Staphylococcus aureus* (ATCC 11632), *Enterobacter aerogenes* (ATCC 13048) and *Escherichia coli* (ATCC 10536). The crude extract was dissolved at a concentration of 2 mg/mL and pure compounds were prepared at 1 mg/mL in DMSO. The amounts of crude extract and pure compounds on the discs were 20 and 10  $\mu$ g, respectively, whereas the standard Chloramphenicol was 30  $\mu$ g.

#### RESULTS AND DISCUSSION

Compound 1 (3.4 mg) was isolated as white needles. The ESI-MS gave molecular ion at *m/z* 323.0891 [M+Na]<sup>+</sup>,

which corresponded to the molecular formula C<sub>17</sub>H<sub>16</sub>O<sub>5</sub>. The UV spectrum in methanol showed absorption at 226, 292 and 357 nm. The prominent and broad absorption at 3241 cm<sup>-1</sup> in the FTIR spectrum indicated the presence of hydroxyl (OH) group together with prominent band at 1623 cm<sup>-1</sup> for the existence of carbonyl (C=O) group. The <sup>1</sup>H NMR spectrum showed two two-proton doublets at  $\delta_H$  6.83 (H-3, H-5) and 7.18 (H-2, H-6) corresponding to four aromatic protons of ring B. Two doublet signals at  $\delta_H$  5.95 and 6.02 were assigned to H-3' and H-5' of ring A. Two distinct methoxy signals at  $\delta_H$  3.75 and 3.88, which were positioned at C-4 (ring B) and C-6' (ring A). A very downfield, signal at  $\delta_H$  13.91 was attributed to 2'-OH and another one singlet signal at  $\delta_H$  9.55 was assigned to 4'-OH. Two one-proton doublet signals at  $\delta_H$  7.74 and 7.90 (*J* = 15.3 Hz) were attributed to *trans* H- $\beta$  and H- $\alpha$ , respectively. Through comparison of the observed data with the literature, compound 1 is identified as 2',4'-dihydroxy-4,6'-dimethoxychalcone [10, 12]. This compound was also isolated from other genus of Annonaceae family i.e. *Goniothalamus gardneri* (Seidel et al. 2000).

Compound 2 (3.2 mg) appeared as yellow needles. The ESI-MS gave a molecular ion of 337.1005 [M+Na]<sup>+</sup> consistent with molecular formula of C<sub>18</sub>H<sub>18</sub>O<sub>5</sub>. Additionally, the FTIR spectrum showed the presence of OH at 3306 cm<sup>-1</sup> and C=O at 1639 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum showed distinct methoxy signals at  $\delta_H$  3.88 (C-3), 3.90 (C-4/C-6) and 3.94 (C-4/C-6) of ring B. Furthermore, two one-proton singlet signals at  $\delta_H$  6.09 and 6.61 were attributed to H-5 and 2-OH of ring B. The five aromatic signals at  $\delta_H$  7.48 (2H, *t*, *J* = 7.3 Hz, H-3', H-5'), 7.55 (1H, *t*, *J* = 7.3 Hz, H-4') and 8.04 (2H, *d*, *J* = 7.3 Hz, H-2', H-6') were corresponding to the protons of ring A. The two one-proton doublet signals (*J* = 16.2 Hz) at  $\delta_H$  8.00 and 8.23 were assigned to *trans* H- $\alpha$  and H- $\beta$ , respectively. Through comparison of the observed data with the literature, compound 2 is identified as tepanone (Colegate et al. 1992).

Compound 3 was isolated as orange needles. Its ESI-MS gave a molecular ion of *m/z* 322.1098 [M+H]<sup>+</sup>, which corresponds to the molecular formula C<sub>19</sub>H<sub>15</sub>O<sub>4</sub>N. The FTIR spectrum showed the presence of lactam carbonyl group at 1657 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum confirmed the presence of three methoxy groups at  $\delta_H$  4.09, 4.11 and 4.20. The aromatic protons of ring B showed two one-proton doublets signals at  $\delta_H$  8.22 and 8.79 (*J* = 5.4 Hz) due to H-4 and H-5. The four aromatic protons at  $\delta_H$  7.54 (*t*, *J* = 7.7 Hz, H-9), 7.75 (*dt*, *J* = 1.2, 7.7 Hz, H-10), 8.58 (*dd*, *J* = 1.2, 7.7 Hz, H-8) and 9.11 (*d*, *J* = 8.7 Hz, H-11), which were ascribed to the protons of the unsubstituted D ring of the aporphine nucleus [14]. The signal of H-11 was more downfield due to the presence of the three methoxy groups at ring A, which deshielded the proton. From the X-ray crystallography data, compound 3 was confirmed as *o*-methylmoschatoline. Figure 1 shows the structure of compound 3 by X-ray crystallography. This compound was isolated from genus *Xylopiia* (Annonaceae) i.e. *X. championii* (Wijeratne et al. 1996) and *X. ferruginea* (Zawawi et al. 2012).

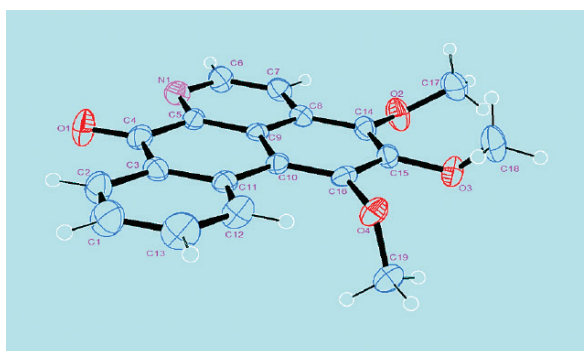
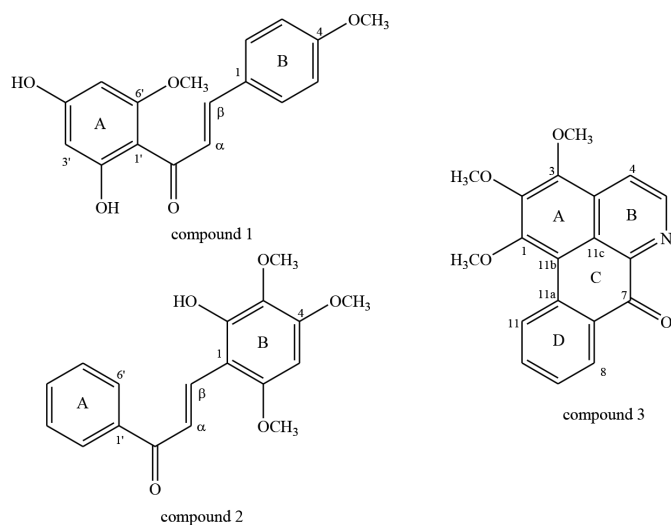
FIGURE 1. *o*-Methylmoschatoline structure by X-ray crystallography

TABLE 1. Antibacterial activity of the crude extract and compounds 1-3

Sample	Inhibitory zone (mm)			
	<i>B. subtilis</i>	<i>S. aureus</i>	<i>E. aerogenes</i>	<i>E. coli</i>
Crude extract	-	-	-	-
1	11.5±0.0	-	12.0±0.0	6.0±0.7
2	-	-	-	-
3	35±0.0	26.5±0.0	-	-
Chloramphenicol	23±0.0	20.5±0.2	26±0.7	26±0.3

The extract and compounds isolated from *E. cuneifolia* were evaluated for antibacterial activity. The crude extract did not have antibacterial activity. Compounds 1 showed inhibitory against *Bacillus subtilis*, *Enterobacter aerogenes* and *Escherichia coli*, whereas compound 3 exhibited very strong activities against *B. subtilis* and *Staphylococcus aureus*. The inhibitory values are shown in Table 1.

#### CONCLUSION

Chromatographic separation of chloroform extract of the twigs of *Ellipeia cuneifolia* yielded three compounds as 2',4'-dihydroxy-4,6'-dimethoxychalcone, tepanone and

*O*-methylmoschatoline. In addition, tepanone could be considered as a chemical marker for this species.

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