

Elemental ratios in sediments as indicators of ecological processes in Spanish reservoirs

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ABSTRACT

Reservoir sediments are important archives of biogeochemical data, reflecting ecological processes that occur at the watershed system. In this paper, a preliminary study of the limnological significance of major element composition and element ratios in sediments of Spanish reservoirs is presented, by analysing two data-sets from different limnological regions. Reservoirs from Eastern and Western Spain present significant differences in the chemical composition of their sediments: higher average values for calcium and magnesium appear in the Eastern reservoirs, while, in Western Spain, higher mean values are found for alumino-silicate elements (i.e. silicon, aluminium, and potassium), iron and phosphorus. Meaningful ratios are Si/Al and Si/K, that are closely related to energy transport in sedimentary catchments, whereas in siliceous catchments Si/Al is an indicator of chemical weathering. Ca/Al ratio appears related to water mineralization, and, in the Eastern Region, the slope of the regression line between Ca/Al and conductivity reflects catchment mineralogy. In superficial sediments from both Eastern and Western regions, Fe/Al is linked to the authigenic precipitation of iron oxides and appears closely related to sedimentary phosphorus accumulation above background levels.

Keywords: Elemental composition, major elements, phosphorus, regional limnology.

RESUMEN

El sedimento de los embalses constituye un importante registro de datos biogeoquímicos, ya que refleja los procesos ecológicos que tienen lugar en el conjunto de la cuenca hidrográfica. A partir del análisis de dos conjuntos de datos obtenidos en distintas regiones limnológicas, se presenta una aproximación preliminar al significado limnológico de la composición química y los cocientes elementales en el sedimento de los embalses españoles. Las regiones Este y Oeste de la Península muestran diferencias significativas en cuanto a composición química del sedimento: los mayores valores promedio en la región Este corresponden a la concentración de calcio y magnesio, mientras que en la región Oeste, los mayores valores medios corresponden a los elementos asociados a los alumino-silicatos (aluminio, silicio y potasio), hierro y fósforo. Los cocientes elementales con mayor significado son: Si/K y Si/Al estrechamente relacionados con la energía de transporte en la región calcárea, mientras que en la región silícica Si/Al es un indicador del proceso de meteorización química. Ca/Al se encuentra asociado a la mineralización del agua, y en la región Este la pendiente de la recta de regresión entre Ca/Al y la conductividad del agua refleja la mineralogía de la cuenca. En ambas regiones, el cociente Fe/Al en el sedimento superficial indica la precipitación endógena de óxidos de hierro y aparece estrechamente asociado a la acumulación de fósforo sedimentario por encima del nivel basal.

Palabras clave: Composición elemental, elementos mayoritarios, fósforo, limnología regional.

INTRODUCTION

Reservoirs have been the major subject of study in Spanish Limnology over the last decades. Since the first work performed by Margalef and co-workers in the late seventies (Margalef *et al.*, 1976), a high number of studies have dealt with several aspects of reservoir limnology, both as a

regional study and as an intensive approach over a specific reservoir or a series of enchainned reservoirs. These studies have focussed on ecological features such as hydrodynamics (e.g., Vidal *et al.*, 2005, Rueda *et al.*, 2006), water chemistry (e.g. Armengol *et al.*, 1991, Riera *et al.*, 1992, Pozo *et al.*, 1997), eutrophication (e.g. Riera *et al.*, 1995, Fraile *et al.*, 1995, Boavida &

Marques, 1996, Palau, 2003), phytoplankton (e.g. Gomez *et al.*, 1995, Dasi *et al.*, 1998), zooplankton (e.g. Jaume, 1991) and bacterial communities (e.g. Simek *et al.*, 1998, 2000).

However, characterization and biogeochemistry of sediments in Spanish reservoirs has received relatively little attention. Issues focused by studies on reservoir sediments included quantification of sediment transport and accumulation (Diaz *et al.*, 1992, Vericat & Batalla, 2005, de Vente *et al.*, 2005, Verstraeten *et al.*, 2003), mineral composition (Valero-Garces *et al.*, 1999, Navas *et al.*, 2004), biological communities (e.g. Prat & Daroca, 1983, Real *et al.*, 1993, 2000), distribution of sedimentary phosphorus along vertical profiles (Armengol *et al.*, 1984, 1986) and at as a regional study (Lopez & Morgui, 1993), and distribution of grain-size (Morgui *et al.*, 1993); but few studies have reported data on geochemical composition (Avendaño & Cobo, 1998).

This lack in the information regarding Spanish reservoirs is especially unfortunate because lake and reservoir sediments are one of the most important archives of biogeochemical data. This information is very useful for reconstructions of ecological processes occurring across the whole watershed system. Changes in composition and structure along a vertical profile offer a time-series that covers an enhanced breadth of temporal variability, whilst even a thin layer of superficial sediment is the result of processes at an annual scale, so its chemical composition and grain-size structure reflect the system dynamics over a wider period than water column characteristics do.

The main inputs involved in the formation of reservoir sediments are similar to those involved in the formation of fine-grained facies in marine basins: 1) allochthonous detritus derived from weathering of catchments soils and parent rock, 2) biogenic components derived both from primary and secondary production and 3) authigenic material precipitated at or near the sediment-water interface as a consequence of E_H -pH controlled organic and inorganic precipitation (Sageman & Lyons, 2004). These biogeochemical processes may

be reconstructed using (as proxy data) the elemental concentration/accumulation values. Proxies employed in the study of sedimentary material should be based on a known relationship between a biogeochemical process and the corresponding flux of the related specific components. However, elemental concentration data suffer from the inherent limitation of reciprocal dilution (i.e. the constant sum effect) that may lead to spurious correlations and ambiguous determination of fluxes (Rollinson, 1993, Sageman & Lyons, 2004). To avoid this problem, the use of elemental ratios (Turekian & Wedepohl, 1961; Calvert & Pedersen, 1993, Lyons *et al.*, 2003) serves to normalize components to major dilutants. This allows for a better identification of the contributions to the flux variability, although normalization techniques must be applied carefully (Van de Weijden, 2002). Moreover, correct interpretation of elemental ratios depends on distinguishing elemental sources and transport modes. For example, in some cases an increase in the Ti/Al ratio has been related to increased eolian flux (Bertrand *et al.*, 1996), but in other cases the same signal has been related to increased fluvial inputs (Murphy *et al.*, 2000). Consequently, interpretation of elemental ratios may be different depending on limnological regions which are associated with different catchment mineralogy.

In order to give a preliminary approach to the geochemistry of sediments in Spanish reservoirs, the concentration of major elements and significant element ratios in two sets of data from different limnological regions are discussed in the next sections. The main purposes of this approach are to:

- A) Provide a reference data-set on chemical composition of sediments in Spanish reservoirs.
- B) Relate reservoirs sediment geochemistry with the limnological regions identified by previous studies.
- C) Establish the main relationships between elemental ratios and ecological processes in reservoirs and their dependence on the geological nature of the watershed at the regional scale.

Table 1.- Reservoirs and their hydrographical watershed included in the study. *Los embalses y su cuenca hidrográfica que se han muestreado para este estudio*

Reservoir	Hydrographical watershed
Set a: Catalanian reservoirs	
Santa Ana	Noguera Ribagorzana
Canelles	Noguera Ribagorzana
Escales	Noguera Ribagorzana
Talarn	Noguera Pallaresa
Terradets	Noguera Pallaresa
Camarasa	Noguera Pallaresa
St Llorenç Mongai	Segre
Oliana	Segre
Sau	Ter
Susqueda	Ter
Baells	Llobregat
St Pons	Llobregat
Riba roja	Ebro
Flix	Ebro
Siurana	Ebro
Foix	Foix
Boadella	Muga
Santa Fe	Tordera
Riudecanyes	Riudecanyes
Set b: Lozoya reservoirs	
Pinilla	Lozoya
Rio Sequillo	Lozoya
Puentes Viejas	Lozoya
Villar	Lozoya
Atazar	Lozoya

STUDY AREA

In this study, two data sets on sediment composition have been analysed. The first corresponds to sediments of 19 Catalanian reservoirs, which belong to ten hydrographical watersheds (Table 1). From their characteristics, these reservoirs (except Boadella) may be included in the limnological region East of the Iberian Peninsula as defined by Margalef *et al.* (1976). This region is characterized by sedimentary catchments areas and reservoirs present high values of conductivity, alkalinity and dissolved calcium (Margalef *et al.*, 1976, Armengol *et al.*, 1991). The second data set corresponds to sediments of five enchainned reservoirs in the Lozoya river, a siliceous watershed (Table 1). These may be included in the limnological region West of the Iberian Peninsula. Reservoirs in this region are characterized by low conductivity, alka-

linity and dissolved calcium concentrations (Margalef *et al.*, 1976, Armengol *et al.*, 1991).

MATERIAL AND METHODS

Data from Catalanian reservoirs were obtained on superficial samples (0-5cm) collected at a deep point, usually close to the dam. For the Lozoya reservoirs, data were obtained on superficial samples (0-2.5 cm), collected at three sampling points for each reservoir: upper, middle and dam.

In all cases, sediment samples were immediately frozen and kept at -20°C until processed. Then, samples were dried at 70°C and finely ground to perform the chemical analysis. Elemental composition was always analysed by X-ray fluorescence after fusion with lithium tetraborate. This method has several advantages: it is a non-destructive method that avoids loss of material and sample contamination during pre-treatment (Somogyi *et al.*, 1997), it avoids the use of hazardous chemicals (Tung, 2004), and it does not suffer from matrix effects (Lopez & Morgui, 1991, Somogyi *et al.*, 1997). The main disadvantage, however, is that XRF gives no information about the chemical form of the element. Even so, as will be described in the next sections, some of the mineral characteristics and chemical behaviour of the elements may be inferred from the analysis of the elemental ratios.

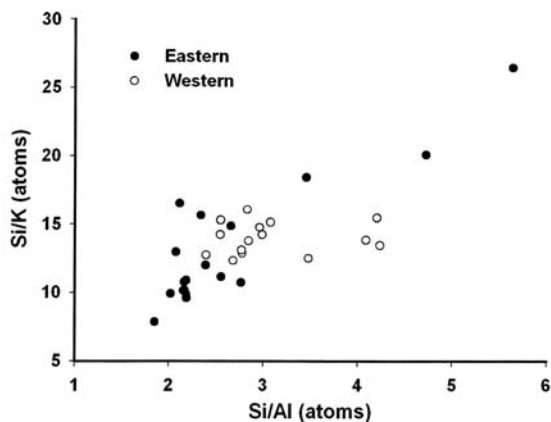


Figure 1.- Relationship between the sedimentary Si/Al and Si/K ratios expressed in atoms. *Relaciones entre las relaciones atómicas Si/Al y Si y Si/k del sedimento.*

RESULTS AND DISCUSSION

Alumino-silicate Elements

Silicon, aluminium and potassium are the main elements associated with the alumino-silicate fraction. These three elements are present in primary silicates as, for example, feldspars and in secondary minerals such as clays. Silicon is the most abundant element, reaching $310 \text{ mg}\cdot\text{g}^{-1}$ (equivalent to $11.5 \text{ mg}\cdot\text{at}\cdot\text{g}^{-1}$) in sediments of reservoirs from the Western region, but it is also abundant in reservoirs from non-siliceous catchments, where the mean value is $183 \text{ mg}\cdot\text{g}^{-1}$ ($6.53 \text{ mg}\cdot\text{at}\cdot\text{g}^{-1}$) (Table 2).

In general, silicon, aluminium and potassium are closely related, although several noticeable trends may be distinguished by analysing the behaviour of ratios between them in different conditions. The ratio K/Al is nearly constant and does not show significant differences between regions, reservoirs or across the longitudi-

nal axis in a reservoir (Table 3). The mean value for the Western area, 0.22, is similar to that reported for the continental crust (Condie, 1993), whereas the slightly lower mean observed in Eastern sediments, 0.20, approaches to that of carbonates (Gao *et al.*, 1998).

The ratios Si/Al and Si/K present a wider range in the Eastern reservoirs than in the Western ones (Table 3). Moreover, in non-siliceous catchments, Si/Al and Si/K varied simultaneously, but in the siliceous watershed Si/K remains fairly constant whereas Si/Al show greater variation (Fig. 1). In the Lozoya reservoirs Si/Al decreases along consecutive reservoirs, from 3.21 at Pinilla to 2.82 at El Atazar and also tends to decrease towards the dam within a single reservoir (Fig. 2A). A similar pattern may be observed in some Eastern reservoirs, e.g. those located in the Noguera Ribagorzana watershed, where Si/Al varies from 5.64 at Talarn to 2.08 at Camarasa (Fig. 2B).

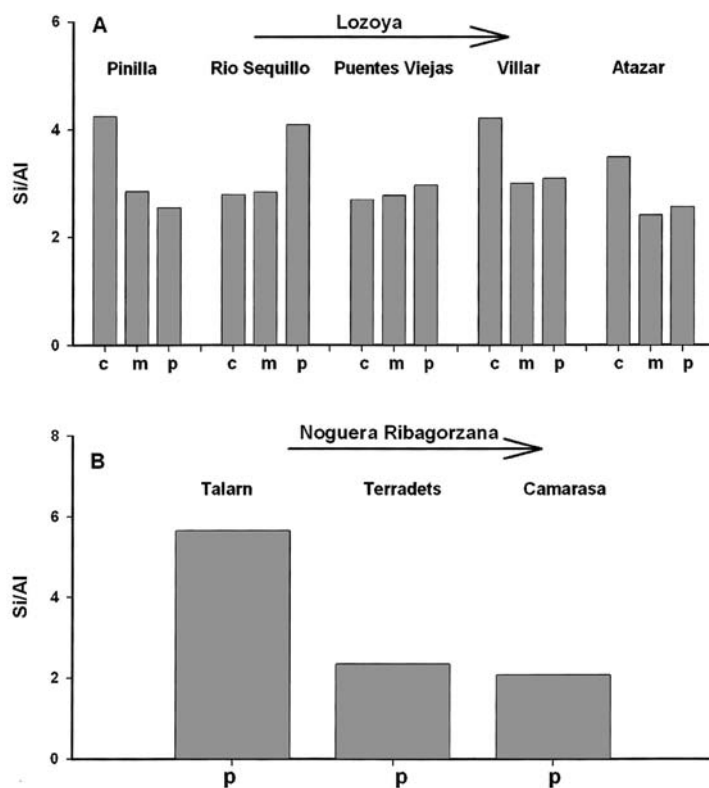


Figure 2.- Changes in the sedimentary Si/Al ratio in reservoirs along the river flow. A: Lozoya river. B: Noguera Ribagorzana river. Ratios are expressed in atoms. *Cambios en la relación Si/Al en el sedimento de los embalses según la dirección en el flujo de agua del río, A: río Lozoya. B: río Noguera Ribagorzana. La relaciones se expresan en átomos.*

Studies on the mineralogy of Eastern reservoir sediments have shown clays, as illite and kaolinite, and quartz as the main silicate minerals (Navas *et al.*, 2004). In this case, the ratios Si/Al and Si/K depends on the proportion of the coarse materials, mainly formed by quartz sands which contains only silicon, and fine clays which contain silicon, aluminium and potassium. When other processes are unlikely to have been comparatively significant, Si/Al and Si/K should vary simultaneously as observed in the studied reservoirs. Increase in Si/Al and Si/K may be related to enhanced input of quartz, which in turn may be due to eolian inputs (Werne *et al.*, 2002), but also to higher energy transport. In agreement with this, the higher values of Si/Al and Si/K observed in reservoirs from the upper catchments areas may be explained as consequence of the higher energy transport. As a whole, the results suggest that in the Eastern area, both ratios may be used as indicators of hydrodynamic and transport processes.

In the Western area, silicate minerals are present as aluminosilicates with different degrees of meteorization, from feldspars minerals to meteorized clays. Chemical weathering of silicates usually comes through incongruent dissolution, i.e. a new solid phase is formed from the solid background material. Globally this process involves a release of silicate and potassium to the

water, whereas aluminium remains in the solid phase (Tardy *et al.*, 2004). As a consequence, meteorization of silicate rocks determines the change in the Si/Al ratio in the solid phase, which decreases as the process continues, whereas Si/K is not modified. The pattern of variation observed in the Lozoya reservoirs (Fig. 2A) agrees with this behaviour. The lower values of Si/Al observed near the dams, and in the lower reservoirs (at watershed scale) indicates that the most meteorized materials with lower Si/Al ratio are accumulated down the river flow. So, in the Western region Si/Al ratios appear related to chemical weathering rather than to hydrodynamics.

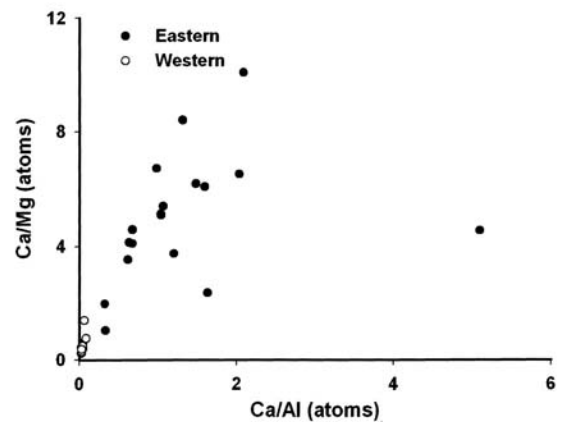


Figure 3.- Relationship between the sedimentary Ca/Al and Ca/Mg ratios expressed in atoms. *Relaciones entre las relaciones atómicas Ca/Al y Ca/Mg del sedimento.*

Table 2.- Range, mean, standard deviation and coefficient of variation for element concentrations in reservoir sediments grouped by limnological region. *Rango, media, desviación estandar y coeficiente variación en los sedimentos de los embalses estudiados agrupados por regiones limnológicas.*

	Al mg-at/g	K mg-at/g	Ca mg-at/g	Mg mg-at/g	Si mg-at/g	Fe mg-at/g	Mn µg-at/g	Ti µg-at/g	P µg-at/g
Eastern Region									
Max	3.83	0.90	5.53	1.40	9.41	0.87	27.03	108.75	65.47
Min	1.08	0.20	0.19	0.26	3.75	0.25	3.45	35.00	8.45
Med	2.67	0.54	2.71	0.64	6.53	0.60	10.06	67.19	27.61
Sd	0.76	0.16	1.24	0.31	1.19	0.17	5.63	16.83	16.17
CV	28.51	29.78	45.88	48.88	18.29	28.53	56.00	25.05	58.55
Western Region									
Max	3.71	0.86	0.21	0.44	11.52	0.98	73.89	101.25	79.55
Min	2.45	0.57	0.07	0.17	8.37	0.38	4.60	40.31	25.70
Med	3.14	0.69	0.12	0.34	9.54	0.73	16.69	81.67	49.80
Sd	0.33	0.08	0.04	0.07	0.93	0.19	17.55	16.20	16.05
CV	10.62	12.05	30.87	19.42	9.78	26.41	105.11	19.84	32.24

Alcaline-Earth Elements

Calcium and magnesium are the most abundant alkaline-earth elements in reservoir sediments. Calcium is mainly present as carbonate minerals, its concentration being more than twenty times higher in the Eastern reservoirs than found in the Western ones (Table 2).

In the Eastern region, sedimentary calcium may attain 220 mg.g⁻¹, which represents that more than 50 % of the dry weigh is found as carbonate mineral. Usually, calcium dominates over magnesium, the mean Ca/Mg ratio (around 4.7) being higher than the value for carbonate rocks (Gao *et al.*, 1998) and for sediments (Wedepohl, 1995). Ca/Mg ratio present in this region a wide range of variability with a coefficient of variation over 50 % (Table 3). Ca/Mg increased with Ca/Al (Fig. 3) suggesting that even in sedimentary catchments, magnesium is mainly associated with the aluminosilicate fraction.

The relative abundance of sedimentary calcium, expressed by the Ca/Al ratio is clearly related to water mineralization, as could be expected from the high solubility of calcite (Tardy *et al.*, 2004). However, there is not a single relationship between these two variables, because the abundance of other soluble minerals, for example gypsum or halite, also

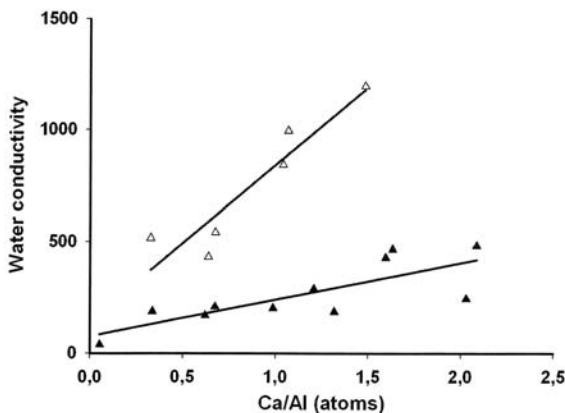


Figure 4.- Relationship between the sedimentary Ca/Al ratio and water conductivity. Values of Ca/Al are in atoms; values of conductivity are in $\mu\text{S cm}^{-1}$. *Relaciones entre el cociente Ca/Al del sedimento y la conductividad del agua. El cociente Ca/Al esta expresado como relación atómica y la conductividad eléctrica del agua en $\mu\text{S cm}^{-1}$.*

determine the conductivity associated with a specific Ca/Al value. Differences in mineralogy of the catchments are reflected in the different slopes of the regression lines between water conductivity and the Ca/Al ratio in sediments and are expressed in the following two equations (see also Fig. 4):

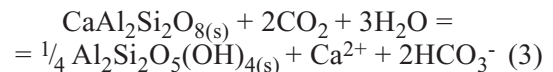
$$\text{Cond} = 164 * \text{Ca/Al} + 74.9; r^2 = 0.63; N=11 \quad (1)$$

for reservoirs in the Noguera Ribagorzana, Noguera Pallaresa, Segre, Llobregat, Tordera and Siurana catchments and,

$$\text{Cond} = 697 * \text{Ca/Al} + 144; r^2 = 0.87; N=6 \quad (2)$$

for Ter, Ebre and Foix catchments.

In the Western region, calcium is a minor component of sediments with a mean concentration below 0.5 mg.g⁻¹. In siliceous catchments a significant amount of the dissolved calcium comes from weathering of Ca-aluminosilicates through anorthite hydrolysis according to the equation (3) (Tardy *et al.*, 2004),



The stoichiometry of compounds released to water through the reaction (3) is the same as from calcite dissolution, but the concentration is considerably lower because silicates are poorly soluble, which explains the lower conductivity and dissolved calcium concentrations in the Western reservoirs.

Dissolution of anorthite involves a diminution of the Ca/Al ratio in the solid phase, as well as dissolution of calcite. In the Lozoya reservoirs, Ca/Al diminishes from an average value of 0.06 in Pinilla to 0.04 in El Atazar, an also decreases towards the dam inside each reservoir (Fig. 5). This pattern of variation could be related to calcite or silicate weathering. However, the value of Ca/Al in the end of Pinilla, Ca/Al = 0.09, is slightly lower than the observed in granites (Wedepohl, 1995), what suggests that weathering of Ca-silicates is the main process determining Ca/Al ratio in this watershed.

Table 3.- Range, mean, standard deviation and coefficient of variation for selected atomic ratios between elements in reservoir sediments grouped by limnological region. *Rango, media, desviación estandar y coeficiente de variación para los cocientes atómicos del sedimento de los embalses que se han utilizado en el estudio, agrupados por regiones limnológicas.*

	K/Al	Si/K	Si/Al	Ca/Al	Ca/Mg	Fe/Al	Mn/Fe * 1000	P/Ca * 1000	P/Fe * 1000	P/Al * 1000	Mn/Al * 1000
Eastern Region											
Max	0.26	26.41	5.64	5.09	10.07	0.26	31.15	197.75	87.30	22.49	7.11
Min	0.13	7.84	1.85	0.05	0.60	0.20	10.23	1.53	22.96	4.83	2.21
Mean	0.20	13.20	2.65	1.27	4.72	0.23	16.25	22.52	44.31	10.19	3.72
Sd	0.03	4.69	1.00	1.11	2.43	0.02	5.52	44.86	18.12	4.76	1.41
CV	16.11	35.54	37.88	87.75	51.47	6.82	33.97	199.20	40.90	46.66	38.05
Western Region											
Max	0.32	16.04	4.24	0.09	0.76	0.31	75.49	935.96	97.42	26.53	21.19
Min	0.17	12.32	2.40	0.03	0.23	0.15	11.68	137.85	42.72	8.34	1.87
Mean	0.22	13.96	3.09	0.04	0.37	0.23	20.83	465.75	69.13	15.87	5.09
Sd	0.05	1.17	0.62	0.02	0.14	0.05	16.48	247.80	16.39	5.09	4.89
CV	20.58	8.36	19.91	39.52	36.76	21.77	79.11	53.21	23.71	32.06	95.98

Metal Elements

Iron is, with aluminium, the main metallic element present in reservoir sediments. Iron geochemistry is, in some aspects, similar to that of aluminium. Both elements are present in primary rocks as alumino-silicates and they may form oxi-hidroxides after chemical weathering. Moreover, both present low concentrations in carbonate rocks (Gao *et al.*, 1998). Globally, these trends are reflected, in the lower iron concentrations observed in the Eastern reservoirs (Table 2), and in the constancy of the Fe/Al ratio that presents the same average value (0.23) in both areas (Table 3).

Nevertheless, there are significant differences in iron and aluminium biogeochemistry, one of the most important being a higher iron loss rate from primary silicates. In siliceous watersheds, river and lixiviation water may carry significant amounts of dissolved iron as a result of chemical weathering of soils. In fact, reservoirs in the Western region usually present high concentrations of dissolved iron (Margalef *et al.*, 1976). At pH values observed in reservoirs, usually up to 6, part of this dissolved iron may precipitate (Stumm & Morgan, 1981), increasing moderately the Fe/Al ratio in sediments. The presence of authigenic iron oxides explains the variation of Fe/Al ratio in the

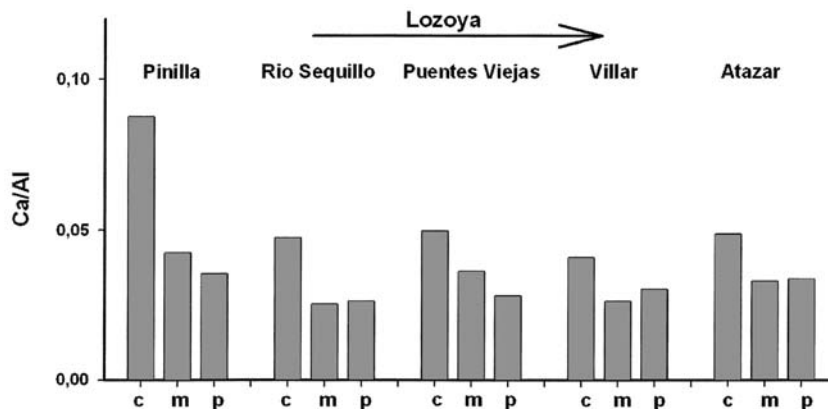


Figure 5.- Changes in the sedimentary Ca/Al ratio in reservoirs located along the Lozoya river. Values are in atoms. *Cambios en el cociente Ca/Al en el sedimento de los embalses situados a lo largo del río Lozoya. Valores expresados como relación atómica.*

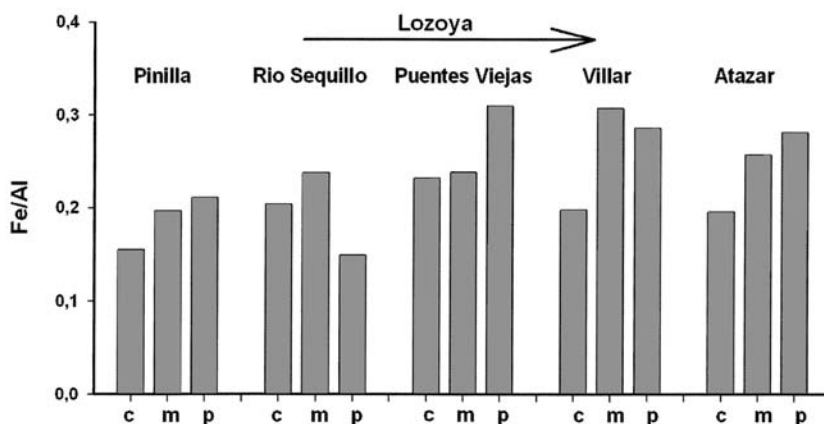


Figure 6.- Changes in the sedimentary Fe/Al ratio in reservoirs located along the Lozoya river. Values are in atoms. *Cambios en el cociente Fe/Al del sedimento de los embalses situados a lo largo del río Lozoya. Valores expresados como relación atómica.*

Lozoya reservoirs, where it increases from near 0.20 in Pinilla and Rio Sequillo to around 0.25 in Puentes Viejas, Villar and El Atazar (Fig. 6). Indeed, this variation causes the higher coefficient of variation observed for Fe/Al ratio in the Western region (Table 3).

The second important difference between iron and aluminium geochemistry is their different behaviour under reducing conditions. Iron's solubility increases with decrease of E_H (Stumm & Morgan, 1981), which determines the well-known negative correlation between dissolved iron and dissolved oxygen (Margalef *et al.*, 1976). Because aluminium oxides does not dissolve at the usual pH and redox conditions observed in reservoirs (Drever, 1982), a significant correlation between dissolved oxygen and Fe/Al ratio in sediment could be expected. However, at least in the Eastern reservoirs, for which data on dissolved components are available, this correlation is not observed (Fig. 7). In the Eastern reservoirs, sulphate is usually present at concentrations of up to 1 mM. (Margalef *et al.*, 1976), so sulphate-reduction is not limited by substrate availability. In this case, iron dissolved from oxi-hidroxi phases re-precipitates as iron sulphides preventing the loss of iron from the solid phase. In vertical profiles, the increase in the Fe/Al ratio associated with iron sulphide deposition clearly distinguishes the transition from oxic to anoxic zones or events (Van Capellen *et al.*, 1998). At a regional

level, however, Fe/Al ratio is not a useful indicator, as redox conditions may change iron distribution in the sediment, but without a significant modification of its total concentration. This may explain the low variability in the Fe/Al ratio in Eastern reservoirs in spite of their different redox characteristics.

Manganese, that can be present in sediment as carbonate phases and oxi-hidroxi phases, is usually a minor component of the sediments in reservoirs, with values below 0.5 % of dry weight. The mean value for the Western reservoirs, $16.7 \mu\text{g-at.g}^{-1}$, is higher than for the Eastern ones (Table 2), but this value may be an overestimation determined by the high level of manganese observed in El Atazar sediments, where it reaches up to $70 \mu\text{g-at.g}^{-1}$ in the dam. If samples from El Atazar are excluded from the data set, the mean value for Western reservoirs become $10.6 \mu\text{g-at.g}^{-1}$ (sd = 4.26; CV = 40.18) which is very close to the observed in the Eastern reservoirs.

Manganese concentration depends on redox conditions in a similar way to iron. Below E_H values around 500 mV manganese occurs as the soluble ion in water, whereas E_H values up to 500 mV favour insoluble manganese oxides (Hem, 1981). This explains the significant correlation between dissolved manganese and oxygen depletion in water (Margalef *et al.*, 1976). Consequently, Mn/Fe and Mn/Al ratios should be lower in sediment from reservoirs with bottom oxygen depletion and higher in

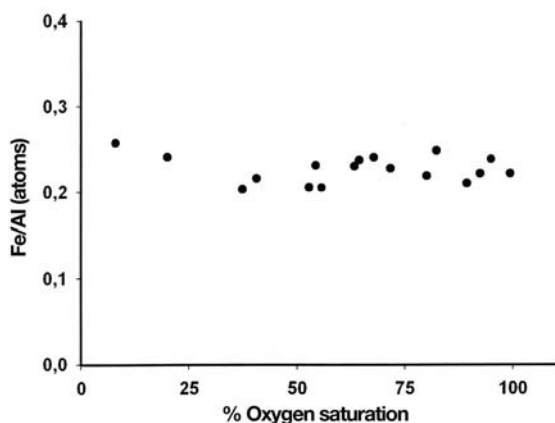


Figure 7.- Relationship between the sedimentary Fe/Al ratio and the annual average saturation of oxygen in bottom water. Values of Fe/Al are in atoms. *Relaciones entre el cociente Fe/Al del sedimento y el promedio anual de saturación de oxígeno en el agua del fondo de los embalses. Los cocientes se expresan como relaciones atómicas.*

sediments from reservoirs with well oxygenated hypolimnion. Some results at the Lozoya reservoirs are partially in agreement with this behaviour. El Atazar, with the highest Mn/Al ratio (>20 in atomMn/1000 atomAl), is an oligotrophic reservoir without summer oxygen depletion (Armengol, com. per.). The upper reservoirs, mainly Puentes Viejas and Villar are more eutrophic and suffer from hypolimnetic anoxia in some periods (Armengol, com. per.) and correspondingly they present lower values of the Mn/Al ratio (< 6 atomMn/1000 atomAl). The results observed suggest that manganese is released to the water in the upper reservoirs and then transported to the lower reservoirs where it precipitates due to the higher redox conditions.

In the Eastern reservoirs, no significant correlation is observed between bottom oxygen saturation and the Mn/Al ratio. In fact one of the highest values of the Mn/Al ratio (7.11 atomMn/1000 atomAl) appears at Susqueda with an average oxygen saturation in the bottom below 8 %. This suggests that concentrations and ratios of sedimentary manganese are also very sensitive to other factors (e.g. mineralogy of the catchments, carbonate equilibrium ...) in addition to the redox conditions. Consequently no generalizations of the significance of sedimentary manganese may be made at a global scale.

Phosphorus

Phosphorus is one of the elements that has been more studied in sediments of different bodies of water. Comparisons among data have to be done carefully because of the use of different methodologies, the results being, in many cases, affected by analytical procedures and by composition of the mineral matrix. Considering only the results for total phosphorus in superficial sediments, previous studies showed significant differences among Western and Eastern reservoirs, the first ones presenting higher concentrations of total phosphorus in sediments (Lopez & Morgui, 1993).

The results observed here confirm previous observation: the mean value in the Western reservoirs is approximately twice the observed in the Eastern reservoirs, the coefficient of variation being similar in both areas (Table 2).

Sedimentary phosphorus is present in several solid phases: phosphorus associated with calcium carbonates (detrital carbonate-fluorapatite, hidroxiapatite and phosphorus adsorbed onto calcium carbonate), phosphorus bound onto metal oxides (both iron or aluminium oxi-hydroxides) and refractory organic phosphorus are usually the most abundant forms (e.g. Jensen *et al.*, 1992; Baldwin, 1996; Lopez, 2004). The predominance of calcium bound phosphate in water bodies located in calcareous watersheds is a common feature and has been confirmed by several studies (Armengol *et al.*, 1984, 1986; Lopez & Morgui, 1993). In addition, precipitation of calcium carbonate phosphate phases has been proposed as the main chemical mechanism controlling dissolved phosphate in these areas and preventing hard-water bodies from eutrophication (Estrada, 1978). By contrast, in siliceous watersheds iron-bound phosphate and/or organic phosphates are the main sedimentary forms (Lopez & Morgui, 1993). Because iron-bound phosphate is formed through an adsorptive process and it is released to the water column under anaerobic conditions, water bodies located in siliceous catchments are expected to present higher dissolved phosphate, mainly if anoxic episodes occur. Actually, although a high

variability existed, Western reservoirs presented higher values of dissolved phosphate than Eastern reservoirs (Margalef *et al.*, 1976).

Taking into account that elements associated in the same mineral phase present a constant ratio (Rollinson 1993), when sedimentary phosphorus is mainly associated to calcium phases P/Ca should be fairly constant, whereas if sedimentary phosphorus is mainly seen as iron-bound phases the constant ratio should be P/Fe. Finally, variable P/Ca and P/Fe indicates an accumulation of organic phosphorus as the main mechanism controlling the amount of total phosphorus in sediments. So, from the previous considerations, we could expect a lower variability of the P/Ca ratio in Eastern reservoirs and a lower variability of the P/Fe ratio in Western ones. The values observed agree with those expected for the Western reservoirs. The coefficient of variation for P/Fe in this area is below 25 %, with an average value close to 70 atom P/1000 atom Fe (Table 3). However, in Eastern reservoirs the results do not match predicted ones. In this area, P/Ca present a coefficient of variation near 200 %, much greater than that of P/Fe or P/Al (around 50 %) (Table 3).

The clearest relationship between sedimentary phosphorus and other sedimentary elements appears between iron and phosphorus when concentrations are normalized relative to the alumi-

nium content (i.e. when P/Al and Fe/Al ratios are used instead of rough concentrations) (Fig. 8). The regression equations between these variables are significant for both areas, with a higher slope for the Eastern area (equation 4) than for the Western one (equation 5),

$$P/Al = Fe/Al * 203.31 - 35.89; r^2 = 0.44 \quad (4)$$

$$P/Al = Fe/Al * 78.37 - 2.30; r^2 = 0.64 \quad (5)$$

where P/Al expressed in atoms P/1000 atoms Al, and Fe/Al expressed in atom Fe/atom Al.

This relationship does not reflect dominance of any particular sedimentary phosphorus fraction. What it really reflects is that the increase of phosphate concentration over the background composition (which in turns may be dominated by calcium, iron or organic phosphorus) is linked with the increase in iron across the same background material. It is well known that authigenic iron oxihydroxides have a high capacity to adsorb phosphorus onto their surface (Likjlema, 1980; Lopez *et al.*, 1996). Thus, precipitation and accumulation of authigenic iron-oxides causes a corresponding increase in P/Al and Fe/Al and seem to be the main process explaining the global variability observed in Spanish reservoirs. Nevertheless, the high variability in the P/Fe ratio observed in the Eastern reservoirs also indicate that other processes have to be taken into account in order to explain the variability of sedimentary phosphorus within calcareous areas.

INFLUENCE OF THE REGIONAL LIMNOLOGY ON THE ELEMENTAL RATIOS

The determination of the elemental composition of sediments is relatively easy to do, being low-time consuming in terms of sampling and analytical procedures. For this reason, it may be widely used in monitoring and management studies of reservoirs. The variation in sediment composition detailed in previous sections indicate that interpretation of elemental ratios as

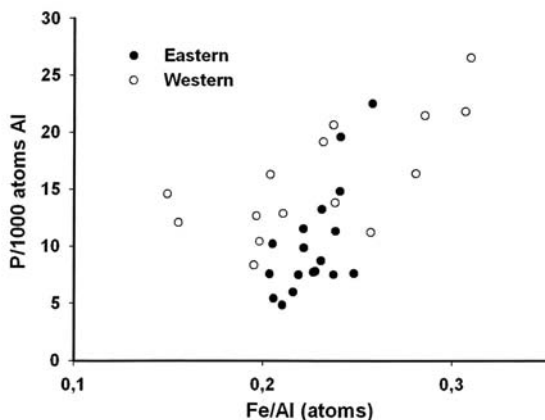


Figure 8. - Relationship between the sedimentary Fe/Al and P/Al ratios. Fe/Al are in atoms. P/Al are in atoms P/1000 atoms Al. *Relaciones entre los cocientes Fe/Al y P/Al del sedimento. El cociente Fe/Al esta expresado como una relación atómica y el P/Al esta en átomos de P / 1000 átomos de Al.*

indicators of ecological processes in reservoirs have to always be done in the context of the limnological region where reservoir is located. Certainly, much more data on chemical composition of sediment in Spanish reservoirs is needed to draw a more precise relationship between biogeochemical processes and element ratios. Even so, from the initial data available, some indications regarding the ecological significance of elemental ratios are given here as a resume of previous comments.

Si/Al, P/Al, P/Fe, Ca/Al and Ca/Mg present significantly different values in the two main limnological regions of the Iberian Peninsula. The first three ratios have higher average value in the siliceous region of West Spain, whereas the two last ratios are higher in the calcareous area of East Spain. In addition to the different values observed in the two regions, the ecological significance of the elemental ratios also depends on the area.

In East Spain, characterized by catchments with sedimentary mineralogy, Si/Al and Si/K ratios are closely related to energy transport, when both vary simultaneously. High values of these ratios are indicators of deposition of coarse particles, which in turns may be related to an increased eolian transport in deforested watersheds or with enhanced transport of eroding material. Ca/Al ratio is mainly related with water mineralization, higher values of conductivity being associated with higher Ca/Al and Ca/Mg ratios. In the Eastern area, the slope of the regression line between Ca/Al and conductivity reflects the presence of soluble minerals others than calcite, and, in some particular places, may be associated with contaminated inputs with a high concentration of dissolved salts. At a global scale, the presence of sedimentary minerals, which both have a high calcium content and are relatively soluble, causes some of the main characteristics of the reservoirs from the Eastern region: high concentrations of alkalinity and dissolved calcium and high Ca/Al ratios in the sediments.

In West Spain, characterized by catchments with poorly soluble silicate minerals, the ratios Si/Al and Ca/Al, are mainly related to chemical weathering. Both ratios decreased in the most meteorized materials, which, due to its lower

grain-size, tend to be easily transported and accumulated downriver. The low solubility of silicates determines the low conductivity, alkalinity and dissolved calcium of these reservoirs.

Other ratios as Si/K, K/Al and Fe/Al show no differences between limnological regions.

Even so, the ratio Fe/Al seems to be very useful to approach the eutrophication process at a global scale. In both areas, East and West Spain, Fe/Al is linked to the authigenic precipitation of iron oxides and it appears clearly related to the P/Al content in sediments. Because iron precipitation is one of the most efficient processes removing dissolved phosphate from water, considering together Fe/Al and P/Al in sediments may give a good estimation of the capacity of sediments to act as a phosphorus sink.

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