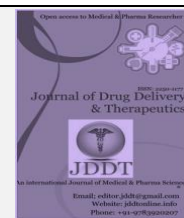
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Research Article

Formulation, Characterization and Antihelminthic Activity Testing of Nitazoxanide Superporous Hydrogel Tablets

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ABSTRACT

In the pharmaceutical field controlled release products have the ability to maintain desired medicament concentration or a longer period of time. Certain drugs are relatively insoluble in water and have high dose requirements that render unsuitable formulation difficulties in sustained release formulations. Nitazoxanide which is a high dose water insoluble antiprotozoal drug was formulated with the aim. To modulate gastro-retentive dosage form based on the superporous hydrogel composites. Foaming technique was used in the preparation of SPH composites. The superporous hydrogels were extremely sensitive to pH of swelling media and good porosity. Superporous hydrogels tablets of nitazoxanide showed good pre-compressional and post-compressional properties. Formulation X is the best formulation containing chitosan, polyvinyl alcohol, formaldehyde, exhibited good swelling ratio. The compatibility studies were performed by Fourier Transform Infrared (FT-IR) Spectroscopic Studies, Differential Scanning Calorimetry Studies (DSC). All formulations were evaluated for stability, drug content, and kinetic drug release & *in-vitro* drug release profile. It was concluded that the proposed gastro-retention drug delivery provides a different supply of nitazoxanide directly to the stomach.

Keywords: Nitazoxanide, Anti protozoal, foaming technique, Chitosan

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INTRODUCTION

Hydrogels are hydrophilic, cross linked polymers which are able to absorb large amounts of water and remain water in soluble all though slow swelling is beneficial their arises situations where a fast swelling polymer is more desirable therefore a new generation of hydrogels¹. Which swell and absorb water rapidly has been developed and is termed as superporous hydrogels which swell to an extent of uniform and equilibrium size in a short span of time. The synthesis involves either cross linking of monomers using multifunctional cross linking agent usually by polymerization and gas blowing technique or copolymerization by using chemical compounds/irradiation². For drug delivery applications conventional hydrogels is limited due to slow swelling kinetics. Improvement of swelling rate is possible by making

hydrogel porous or superporous. The word superporous means the pore structure of the hydrogel is open and connected. In this way the water will be absorbed into the hydrogel structure by means of diffusion and capillary actions. The draw back here is the more porous the structure is the weaker the hydrogel structure so to improve mechanical properties foaming and gelation process are used to improve compressive strength, elasticity³.

Nitazoxanide is an antihelminthic and antiprotozoal agent having broad spectrum of activity. It is chemically 2-acetyloxy (N-(5-nitro-2thiazolyl) benzamide. Nitazoxanide is a light yellow crystalline powder which is insoluble in water and poorly soluble in ethanol. It belongs to BCS class II drug in biopharmaceutical classification system i.e. low solubility and high permeability. It is used for treating diarrhea caused

by *Giardia lamblia* as well as for cryptosporidiosis in immune-compromised patient⁴.

The pharmaceutical composition of tablet dosage form comprises of a first portion of nitazoxanide in a controlled release manner and second portion with second quantity of nitazoxanide for immediate release effect here the control release property is assumed where in absorption and bioavailability of the active agent over extended period of time and also the immediate release portion of the active agent is made bio available without substantial delay. In the present research work 2³ full factorial design i.e. 3 factors and 2 levels was used to optimize the concentrations of chitosan, polyvinyl alcohol, microcrystalline cellulose and response is cumulative % drug release by using Design-Expert software.

MATERIALS AND METHODS

Materials

Chitosan, Poly vinyl alcohol (PVA), Formaldehyde and Sodium bicarbonate, microcrystalline cellulose, magnesium stearate, talc were obtained by Chalapathi institute of pharmaceutical sciences, lam, Guntur. Nitazoxanide is a kind gift sample from pellets pharma limited.

Methods

Preparation of nitazoxanide loaded Superporous Hydrogels

Chitosan is dissolved in 0.1M acetic acid solution to obtain 2%w/v. A 10%w/v aqueous solution of PVA was prepared. The chitosan solution was dispersed in PVA solution and homogenized for 15 minutes definite composition of formaldehyde was added as crosslinking agent. Calculated quantities of sodium bicarbonate was added and homogenized for a period of 4hours in ultraterax to induce uniform gelation, crosslinking and effervescent reactions. The swelled and formed hydrogels were left to stand overnight at room temperature. The hydrogels were freeze dried for 12 hours. The freeze dried samples were stored in well closed container. (Table no.1)

Formulation of SPH tablets of nitazoxanide

500mg drug equivalent superporous hydrogel, microcrystalline cellulose were weighed accurately were taken in a poly bag and blended by shaking for 10 minutes. To the powder blend magnesium stearate, talc, sodium bicarbonate were added and were mixed for 5 more minutes by shaking to ensure complete mixing. The powder blend is passed through 80 sieve and were compressed into tablets. (Table no.2)

Anthelmintic activity in *Pheretima posthuma*

Collection of *Pheretima posthuma*: *Pheretima posthuma* were collected from the swampy water along with amaravathi road, Chalapathi institute of Pharmaceutical Sciences, lam, Guntur, Andhra Pradesh, India. The average size of worms was 5-9cm.

Preparation of drug solutions: The standard drug metronidazole was received from chalapathi institute of pharmaceutical sciences and test samples were synthesized in chalapathi drug testing laboratory, chalapathi institute of pharmaceutical sciences. Metronidazole and test samples were prepared as Group-I is contains saline (0.9% Nacl in water), Group-II is containing metronidazole as a reference standard (400mg/ml), Group-III containing NSPH-I (nitazoxanide loaded superporous hydrogel) (200mg/ml), Group-IV containing NSPH-II (400mg/ml).

Optimization by 2³ Factorial Designs

The 2³ full factorial designs were carried out systematically with 3 factors at 2 levels to prepare the nitazoxanide loaded SPH tablets. A total of 8 experimental trials were done at all possible combinations. During optimization of excipient the amount of chitosan, Polyvinyl alcohol, Microcrystalline cellulose and response is cumulative % drug release were selected as the independent variables that were varied at 2 levels i.e. low and high. (Table 3)

RESULTS AND DISCUSSIONS

Characterization of superporous hydrogels

The following parameters are suitable for superporous hydrogels

Physical Appearance: Clarity, color, and presence of any particles were examined visually in the prepared hydrogels.

Percent Yield of SPH: The percent yield represents the amount of material produced after SPH freezing, compared with the amount retained for freeze drying. The formula can be calculated by,

$$\text{Percentage yield} = \frac{\text{Weight of sample after drying}}{\text{Weight of sample before drying}} \times 100$$

Drug content of SPH: Superporous hydrogel drug content was calculated to assess how much of a particular amount of superporous hydrogel is present. We taken in 100 ml volumetric flask, a superporous hydrogel required was taken. After that 10ml of buffer was added, mixed well and make upto volume. The above mixture was filtered, and drug content determined by using correct wavelength of a UV visible spectrophotometer⁵.

Gelatin Kinetics: The reaction to polymerization continues to increase the viscosity until the whole structure of the network gel is formed. Gelation time was calculated as the gel forming period and measured with a simple method of tilting with acetic acid after pH adjustment to 5.0. This parameter was taken as the time required for the reactant to avoid falling into the tilted tube position⁶.

Swelling Studies: Swelling studies included swelling time and swelling ratio.

Swelling Time: Swelling time is time taken by the hydrogel to reach its equilibrium swelling point where swelling is prevented. Swelling is usually measured gravimetrically and volumetrically a texture analyzer is used to assess swelling time. Dried SPH was allowed to hydrate more than swelling medium (25ml) at room temperature. At various time periods, the hydrogel was separated from the solution and measured after excess solvent on the surface was blotted.

Swelling Ratio: The swelling behavior of prepared SPH was carried out by the "T" bag weight method in which 0.1g of sample was added to a small bag made of nylon (50mm×90mm) 200mesh. The bag was completely immersed in the swelling media (200ml) simulated gastric fluid pH 1.2 at room temperature for 24 hrs to each the swelling equilibrium. Adhered liquid droplets on the surface of samples were removed by blotting with tissue paper. The swollen SPH & then dried in oven at 60°C for 6hrs until there was a no change in the weight of samples the equilibrium swelling was defined as follows. (Figure 1)

$$Q_s = \frac{W_s - W_d}{W_d}$$

Where, W_s is the weight of the swollen superporous hydrogel, W_d is the weight of the dried superporous hydrogel and Q_s is equilibrium swelling ratio⁷.

Density Measurement: Dried SPHC was used for density measurements, which actually showed an apparent density of SPHC. A sample of SPHC was taken and weighed to evaluate the mass of the piece. A sample of the polymer was dipped into a predetermined volume of hexane in a graduated cylinder and the rise in hexane volume was measured as the volume of the polymer. Density was calculated by

$$\text{Density} = \frac{M_{\text{SPHC}}}{V_{\text{SPHC}}}$$

Where, V_{SPHC} is the volume of the solvent displaced by SPHC, M_{SPHC} is the mass of the SPHC⁸

Porosity Measurement: The solvent replacement process has been used for porosity measurement. The dried hydrogels were absorbed overnight in absolute ethanol and weighed after unwanted ethanol on the surface was blotted. The porosity of the following equation was calculated.

$$\text{Porosity} = \frac{M_2 - M_1}{\rho V}$$

Where, M_1 and M_2 are the mass of the hydrogel before and after absorbed in absolute ethanol, ρ is the density of absolute ethanol and V is the volume of the hydrogel⁹.

Measurement of Void Fraction: By immersion of the hydrogels into HCl (pH 1.2) buffer upto equilibrium swelling, the void fraction was identified inside superporous hydrogels. The dimensions of the swollen hydrogels were measured and the sample volumes were determined as dimensional volumes by using these data. Meanwhile, the amount of buffer absorbed in the hydrogels was determined by removing the weight of the dry hydrogel from the weight of the hydrogel swollen, the resulting values were assigned as the total volume of pores in the hydrogels. Void fraction calculated by following formula¹⁰ (Table 4)

$$\text{Void fraction} = \frac{\text{Dimensional volume of the hydrogel}}{\text{Total volume of pores}}$$

Water Retention: Water retention capacity (W_{rt}) as a function of time at 37°C, was used to determine the following formula.

$$W_{rt} = \frac{W_p - W_d}{W_s - W_d}$$

Where, W_d is the weight of dried hydrogel, W_s is the weight of fully swollen hydrogel and W_p is weights of the hydrogel at various exposure times¹¹. (Table 5)

Characterization of tablets

The formulated tablets were evaluated for following parameters

Pre Compressional Parameter:

Angle of Repose: The angle of repose of the granules was determined by using the funnel method suggested by Neumann. The accurately weighed granules were taken in a funnel. The funnel height was raised to the point of the top of the granule heap. This funnel height was changed. The granules were allowed to flow freely to the surface through the funnel. The powder cone diameter has been determined, and the angle of repose was calculated by using the following formula,

$$\theta = \tan^{-1} h/r$$

Bulk Density: Bulk density is the ratio of mass and bulk volume. Accurately weighed 20 g granules were allowed to flow in a fine stream into a graduated cylinder and final volume was noted.

$$\text{Bulk density} = \frac{\text{Bulk mass}}{\text{Bulk volume}}$$

Tapped Density: The 20gm blend into a graduated cylinder of a mechanical tapping instrument was permitted in the fine stream. The measuring cylinder was tapped for 100 times, and final tapped volume was measured. Tapped density formula was calculated by,

$$\text{Tapped density} = \frac{\text{Bulk mass}}{\text{Tapped volume}}$$

Carr's Index: Carr's index evaluates interparticulate cohesive characteristics with the angle of repose measurements and studies the effects of bulk- tapped solid packing geometry.

$$\text{Carr's index} = \frac{\text{Tapped density} - \text{Bulk density}}{\text{Bulk density}} \times 100$$

Hausner's Ratio: Hausner's ratio is a simple method to evaluate the stability of the powder column and to estimate flow properties. Hausner's ratio was calculated using formula. (Table 6)

$$\text{Hausner's Ratio} = \frac{\text{Tapped density}}{\text{Bulk density}} \times 100$$

Post Compressional Parameter:

Tablet thickness: Thickness and diameter of tablets were important for uniformity of tablet size. Thickness and diameter were measured using Vernier caliper.

Hardness: The hardness test was calculated by taking 6 tablets of every formulation and adding the strength to the tablet by using Monsanto hardness tester.

Friability: Friability test was performed by using Roche friabilator. The 20 tablets were weighed separately and placed in the drum that revolves at 25 r/min dropping the tablets through a distance of six inches with each revolution. After 4 min the tablets were weighed, and the percentage loss in tablet weight was calculated by using formula,

$$\% \text{Loss} = \frac{W_0 - W_f}{W_f} \times 100$$

Where, W_0 is the initial weight of tablets and W_f is final weight of tablets.

Weight uniformity: 20 tablets were previously punched and measured there on a digital weighing balance. After that each tablet were weighed. The weight of a tablet was determined by our collective weight

Content of Uniformity:

The drug content is considered to check dose uniformity in the formulation. Total 20 tablets were weighed and powdered. The stock solution was prepared by dissolving a drug powder equivalent to 10 mg in 10 ml water. The stock solution was shaken on a sonicator for 20 minutes.

This resultant solution was further diluted with water to attain concentration upto 10µg/ml and the absorbance was measured at 286nm. (Table 7).

In-vitro Dissolution Study:

The in-vitro dissolution study was performed by using USP apparatus type II at 37°C ± 0.5°C using 900ml hydrochloric

acid buffer (pH-1.2) as a dissolution medium at 50 r/min. 5ml samples were withdrawn and replaced by new dissolution media at predetermined time intervals. The dissolution was continued the apparatus was allowed to the 12hours the graphs of time vs percentage release were plotted. Withdrawn samples was filtered through a 0.45µm membrane filter, diluted and assayed at 286nm using a UV-VIS spectrophotometer. Cumulative percentage drug release was calculated by using an equation obtained from a calibration curve. To ascertain the order and mechanism of drug release the in-vitro release data was subjected to various kinetic equations.

Kinetic Models

Treatment of dissolution data with different kinetic equations:

In order to consider the mechanism of release and release rate kinetics of the dosage form, the data attain were fitted into Zero order, First order, Higuchi matrix, and Korsmeyer and Peppas. Based on the r-value, the best-fit model selected.

Zero Order Kinetics:

Zero-order release would be concluded by the given formula:

$$\frac{dQ}{dt} = K_0$$

Where, Q = Drug released at time 't'

K_0 = Zero-order rate constant (h⁻¹).

When the data is plotted as cumulative percent drug released vs time, if the plot is linear then the data obeys zero-order release kinetics, with a slope equal to K_0 .

First Order Kinetics:

To study the first order release rate kinetics, the release rate data were fitted to the given formula:

$$\frac{dQ}{dt} = K_1 Q$$

Where, Q = Amount of drug remained at time 't'

K_1 = First-order rate constant (h⁻¹).

When the data is plotted as log cumulative percent drug remaining versus time; yields a straight line, indicating that the release follows first-order kinetics. The constant ' K_1 ' can be obtained by multiplying 2.303 with slope values.

Higuchi model:

Higuchi developed several hypothetical models to study the release of water soluble and low soluble drugs incorporated in semisolids and/or solid matrices. Mathematical expressions were obtained for drug particles dispersed in a uniform matrix behaving as the diffusion media. And the given equation is,

$$Q_t = KH t^{1/2}$$

Where, Q_t = amount of drug released in time t,

KH = Higuchi dissolution constant

Korsmeyer and Peppas model:

The release rate from sustained release polymeric matrices can be described by the equation proposed by korsmeyer et al.

$$Q = K_{kp} t^n$$

Where, Q = amount of drug released at time 't'

K_{kp} = Kinetic constant incorporating structural and geometric characteristics of the tablets

'n' = The diffusional exponent, indicative of the release mechanism.

The release exponent, n, is the slope of log fraction of drug release vs log time cure. (Table 8) (Figure 2)

Stability Studies:

Stability studies on promising formulation were carried out in compliance with ICH guidelines by storing tablets at 40°C /75% relative humidity for 3 months.

The sample was withdrawn at fixed time intervals of 0 (initial), 30, 60 and 90 days. At the end of three months, the time interval the tablets were tested for any hardness, changes in drug content, physical changes, and in-vitro dissolution studies. (Table 9) (figure 3).

Compatibility Studies:

Fourier Transform Infrared (FT-IR) Spectroscopic Studies

The compatibility between the pure nitazoxanide and SPH & combination of nitazoxanide + PVA and pure PVA was tested by using FT-IR spectroscopy. The chemical structure of the synthesized hydrogels was also studied. The FTIR spectrum was recorded and the range of 400-4000cm⁻¹ using KBr pellet method. (Figure 4)

Differential Scanning Calorimetry Studies (DSC)

The DSC was carried out for pure drug, tablet using a SHIMADZU DSC-60 plus differential scanning calorimeter. The system was calibrated with a high purity sample of nitazoxanide. Sample was scanned at the heating rate of 10°C/min over a temperature range of 50 to 250°C under nitrogen gas using aluminum pans. (Figure 5)

Evaluation of Anthelmintic activity using *Pheretima posthuma*:

Take 12 petridishes containing 10ml of sample in each petridish. Four groups were taken.

In group-I three petridishes were taken. In each petridish 10ml of saline solution was poured and one *Pheretima posthuma* was placed.

In group-II three petridishes were taken. In each petridish 10ml of metronidazole as a standard solution was poured and one *Pheretima posthuma* was placed.

In group-III three petridishes were taken. In each petridish 10ml of NSPH-I (nitazoxanide loaded superporous hydrogel) solution was poured and one *Pheretima posthuma* was placed.

In group IV three petridishes were taken. In each petridish 10ml of NSPH-II solution was poured and one *Pheretima posthuma* was placed. Observations were recored for (Table 10) the time taken for paralysis and time taken for death in minute. (Figure 6)

Optimized 2³ Factorial design

To classify the most important factors among all the factors, selection is done at beginning of the experimental procedure. From the experimental results, the effects of all studied variables and the variable interactions were graphically and statistically interpreted for all responses. The optimized nitazoxanide loaded SPH was used for the preparation of tablets. (Table 11)

The application of factorial design yielded the following regression equation.

Cumulative % drug release =

$$-113.50333+99.20833\times\text{Chitosan}+38.75000\times\text{PVA}+1254.5000\times\text{MCC}.$$

Where, negative values indicate a negative effect of a specific variable on the response factor and positive value indicates positive effect of a specific variable. The polynomial

regression result was expressed using 3D graph and contour plot. (Figure 7)

The ANOVA studies indicated that all models were significant ($p<0.05$) for all response parameters scrutinized. Model simplification was done by removing non-significant terms ($p>0.05$) in polynomial equations. (table.12)

Table 1: Formulation of Superporous Hydrogels

Ingredients (mg)	Formulation									
	I	II	III	IV	V	VI	VII	VIII	IX	X
Chitosan (2% W/V)	40	80	120	160	80	80	80	80	80	80
Polyvinyl alcohol (10% W/V)	400	400	400	400	200	200	400	400	400	800
Formaldehyde (10% V/V)	400	400	400	400	400	400	400	200	600	800
Sodium bicarbonate	80	80	80	80	80	80	80	80	80	80
Nitazoxanide	500	500	500	500	500	500	500	500	500	500

Table 2: Formulation of SPH tablets of Nitazoxanide

Ingredients (mg)	Formulation									
	I	II	III	IV	V	VI	VII	VIII	IX	X
Nitazoxanide	500	500	500	500	500	500	500	500	500	500
Microcrystalline cellulose	61.32	72.83	81.7	65.17	49.39	52.27	69.59	49.59	58.5	86.7
Sodium bicarbonate	16	16	16	16	16	16	16	16	16	16
Magnesium stearate	4	4	4	4	4	4	4	4	4	4
Talc	2	2	2	2	2	2	2	2	2	2

Table 3: Variables in 2³ Factorial Design

Independent Variable	Levels	
	Low	High
A: Chitosan	0.04%	0.16%
B: PVA	0.2%	0.8%
C: MCC	0.09%	0.12%
Dependent Variable		
Cumulative % drug release		

Table 4: Characterization of superporous hydrogels

Batch code	Parameters							
	Percent Yield (%)	Drug content (%)	Gelation time (sec)	Swelling time (min)	Swelling ratio	Density (g/cc)	Porosity (%)	Void fraction (%)
1	20.45±0.11	84.54±0.01	31.22±0.54	5.01±0.02	103.33±5.15	0.49±0.02	3.65±0.22	5.75±1.88
2	18.23±0.15	95.46±0.56	35.12±0.86	5.00±0.01	120.25±2.98	0.51±0.03	5.20±0.16	4.01±0.31
3	15.46±0.42	104.80±0.01	29.56±0.01	5.12±0.03	140.20±2.88	0.23±0.01	8.25±0.30	3.77±0.20
4	12.33±0.20	108.45±0.01	14.52±0.54	5.19±0.01	182.66±7.98	0.26±0.01	9.23±0.23	3.45±0.35
5	21.50±0.21	75.48±0.20	22.10±1.02	5.02±0.01	118.66±4.98	0.49±0.00	18.25±0.25	3.75±0.38
6	14.28±0.19	80.75±0.25	20.16±0.89	4.56±0.04	105.45±0.06	0.35±0.00	30.24±3.02	2.75±0.25
7	13.20±0.43	90.20±0.01	16.45±1.35	5.00±0.00	55.44±5.12	0.30±0.02	29.13±0.05	1.99±0.15
8	10.25±0.30	76.25±0.01	80.66±1.25	4.88±0.19	83.65±5.02	0.25±0.00	24.52±0.02	2.58±0.14
9	15.55±0.56	85.13±0.01	91.00±0.01	5.09±0.15	25.55±2.22	0.30±0.02	34.12±4.44	1.45±0.02
10	25.79±0.33	89.51±0.01	192.22±2.5	5.00±0.01	20.00±5.20	0.35±0.01	40.11±0.20	1.89±0.14

Table 5: Water Retention capacity

S.NO.	Time	Water retention capacity
1	0	0.9
2	3	0.75
3	6	0.82
4	9	0.85
5	12	0.92
6	15	0.87
7	18	0.84
8	21	0.87
9	24	0.84

Table 6: Precompressional parameters of SPH tablets

Batch code	Precompressional parameters				
	Bulk density (g/cm ³)	Tapped density (g/cm ³)	Carr's index (%)	Hauner's ratio	Angle of repose(°)
1	0.23±0.0012	0.25±0.0080	8.00±0.26	1.08±0.032	22.30±1.22
2	0.25±0.0093	0.30±0.00	16.66±2.25	1.2±0.035	26.40±5.24
3	0.26±0.0075	0.29±0.016	10.34±4.55	1.11±0.06	30.20±1.20
4	0.18±0.0042	0.25±0.042	25.00±2.24	1.38±0.05	28.31±1.60
5	0.59±0.014	0.80±0.089	26.25±6.25	1.35±0.08	20.15±3.25
6	0.48±0.026	0.59±0.040	18.64±8.45	1.22±0.10	19.00±1.25
7	0.45±0.021	0.50±0.58	10.02±0.32	1.11±0.05	22.14±2.20
8	0.40±0.020	0.45±0.25	11.11±7.45	1.12±0.04	30.25±1.46
9	0.20±0.001	0.23±0.0035	13.05±1.26	1.15±0.02	18.42±1.25
10	0.30±0.00	0.35±0.015	14.28±2.05	1.16±0.05	25.21±1.41

Table 7: Post compressional parameters of SPH tablets

Batch code	Post Compressional parameters						
	Weight variation (%)	Diameter (mm)	Thickness (mm)	Hardness (g/cm ²)	Friability (%)	Disintegration time (min)	Content uniformity
1	0.25±0.246	7.08±0.005	2.75±0.000	4.45±0.055	0.224	28.44±1.51	98.58±0.85
2	0.22±0.244	7.05±0.004	2.74±0.001	4.44±0.052	0.122	29.22±1.51	98.52±0.55
3	0.19±0.22	7.05±0.003	2.74±0.003	4.40±0.051	0.225	36.00±1.00	99.71±0.83
4	0.23±0.210	7.08±0.000	2.74±0.000	4.42±0.050	0.250	30.00±0.55	99.99±0.25
5	0.25±0.21	7.04±0.001	2.74±0.005	4.45±0.053	0.228	40.12±0.57	100.0±0.74
6	0.26±0.19	7.04±0.005	2.74±0.004	4.40±0.056	0.200	27.66±0.57	99.97±0.50
7	0.27±0.20	7.05±0.000	2.74±0.000	4.44±0.053	0.220	35.22±0.55	98.98±0.85
8	0.18±0.23	7.04±0.002	2.74±0.001	4.43±0.054	0.224	33.25±0.54	97.58±0.88
9	0.25±0.21	7.05±0.005	2.74±0.002	4.40±0.051	0.222	31.55±1.24	99.58±0.77
10	0.24±0.26	7.05±0.000	2.74±0.000	4.41±0.050	0.244	34.25±0.45	100.1±0.64

Table 8: Drug release kinetics of optimized formulations

Batch no	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10
Model name	R ²	R ²	R ²	R ²	R ²	R ²	R ²	R ²	R ²	R ²
Zero Order	0.7599	0.7044	0.8541	0.7872	0.776	0.8361	0.8377	0.8391	0.8189	0.8083
First Order	0.784	0.7296	0.8734	0.8185	0.8081	0.8657	0.8696	0.87	0.8523	0.8422
Higuchi mode	0.9289	0.8971	0.9254	0.9369	0.9255	0.9505	0.9685	0.9541	0.9628	0.9548
Kor's peppas	0.4759	0.4206	0.4944	0.4702	0.4477	0.5081	0.5499	0.5175	0.5315	0.512

Table 9: Drug content data of stability studies

Time	90days	60days	30days	0days
10000	89.26	89.3	89.43	89.72
20000	90	90.02	90.05	90.08
30000	94.16	94.27	94.38	94.48
40000	93.05	93.17	93.21	93.29
50000	97.81	97.88	98	98.03
60000	93.66	93.75	93.86	93.97
70000	95.55	95.65	95.79	95.87
80000	94.14	94.26	94.38	94.49
90000	98.33	98.42	98.53	98.66
100000	99.24	99.32	99.41	99.52

Table 10: Antihelminthic activity of Nitazoxanide loaded SPH

Name of the group	Treatment group	Sample concentration	Time taken for paralysis in minutes	Time taken for death in minutes
Group -I	Vehicle (Saline)	0.9% Nacl	-	-
Group -II	Metronidazole	400mg/ml	30	45
Group -III	NSPH -I	200mg/ml	75	80
Group -IV	NSPH -II	400mg/ml	55	66

Table 11: Observed response in 2³ factorial design for formulation f10

Std.	Run	Factor 1	Factor 2	Factor 3	Response
		A:Chitosan	B:PVA	C:MCC	Cumulative % drug release
6	1	0.16	0.2	0.12	54.14
3	2	0.04	0.8	0.09	29.51
7	3	0.04	0.8	0.12	74.67
1	4	0.04	0.2	0.09	25.07
4	5	0.16	0.8	0.09	32.86
2	6	0.16	0.2	0.09	27.35
5	7	0.04	0.2	0.12	37.00
8	8	0.16	0.8	0.12	99.52

Table 12: ANOVA for selected factorial model response of % cumulative drug release

Source	Sum of Squares	df	Mean square	F-value	P-value
Model	4197.37	3	1399.12	6.60	0.0499
A-chitosan	283.4	1	283.46	1.34	0.3120
B-PVA	1081.12	1	1081.12	5.10	0.0869
C-MCC	2832.79	1	2832.79	13.36	0.0217
Residual	848.35	4	212.09		

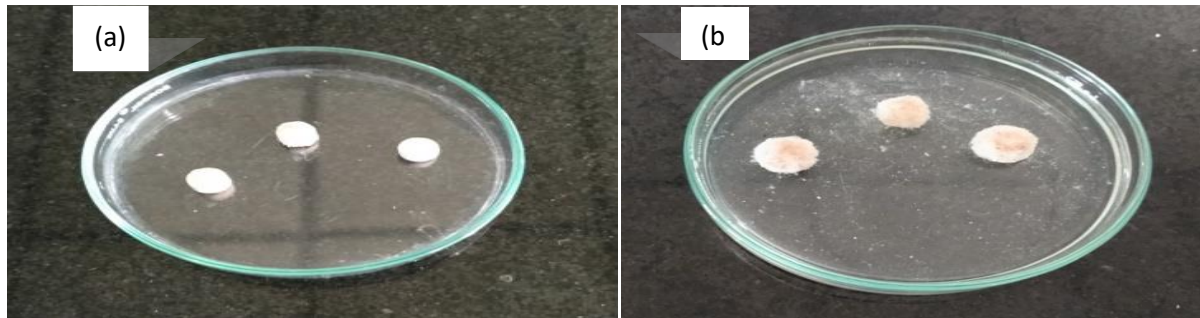


Figure 1: (a) Before swelling, (b) After swelling

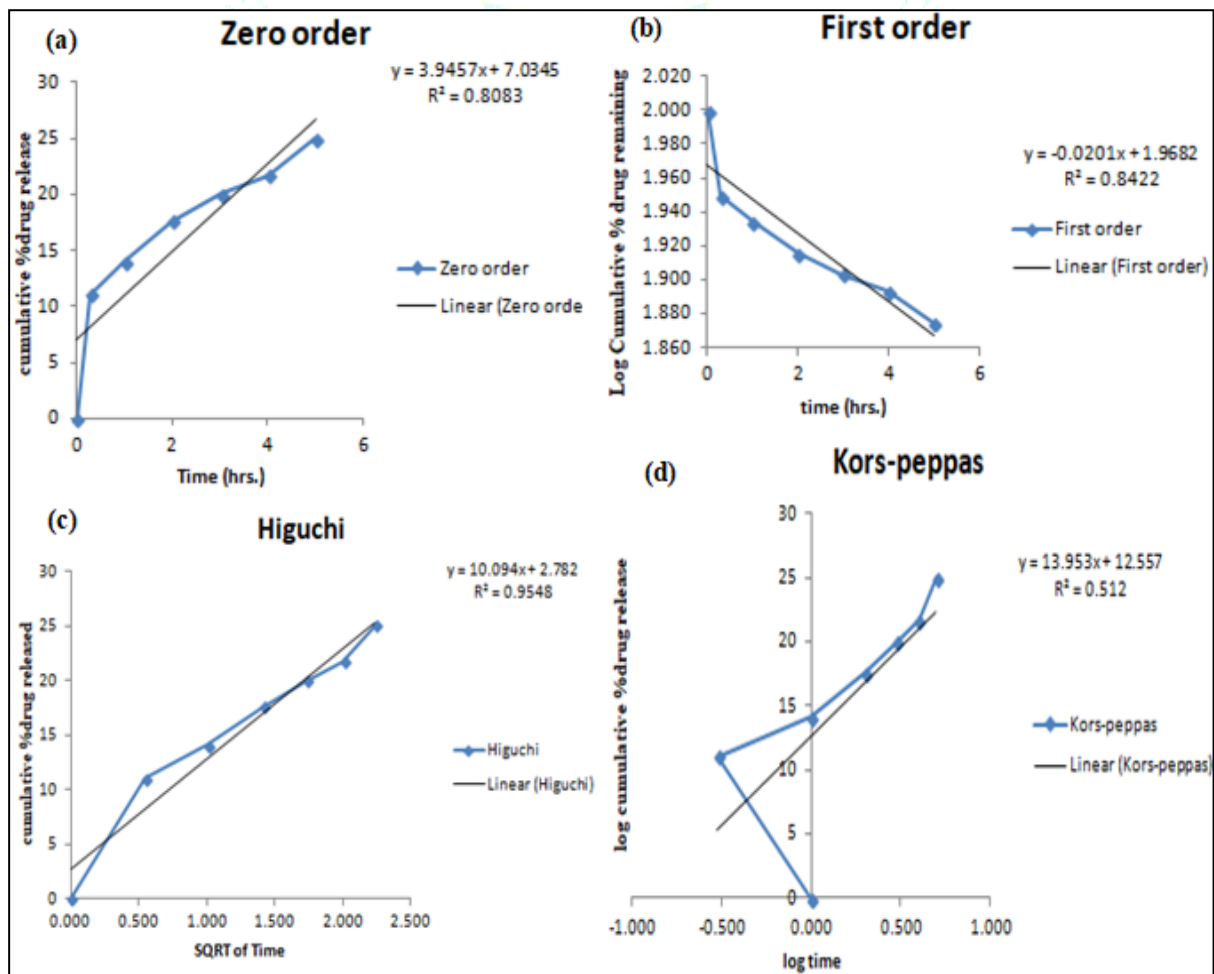


Figure 2: Drug release kinetic models for (a) zero order (b) first order (c) higuchi plot (d) kors-peppas

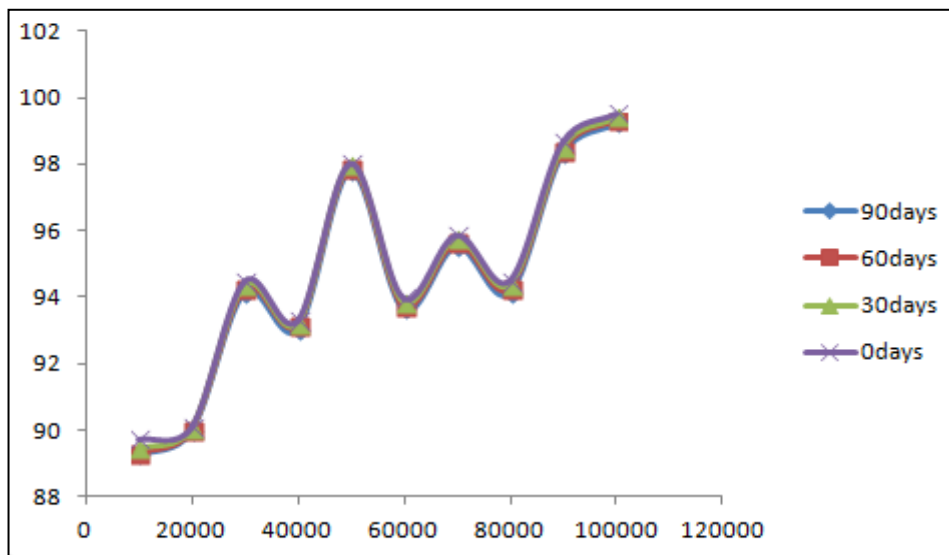


Figure 3: Dissolution profile of stability studies

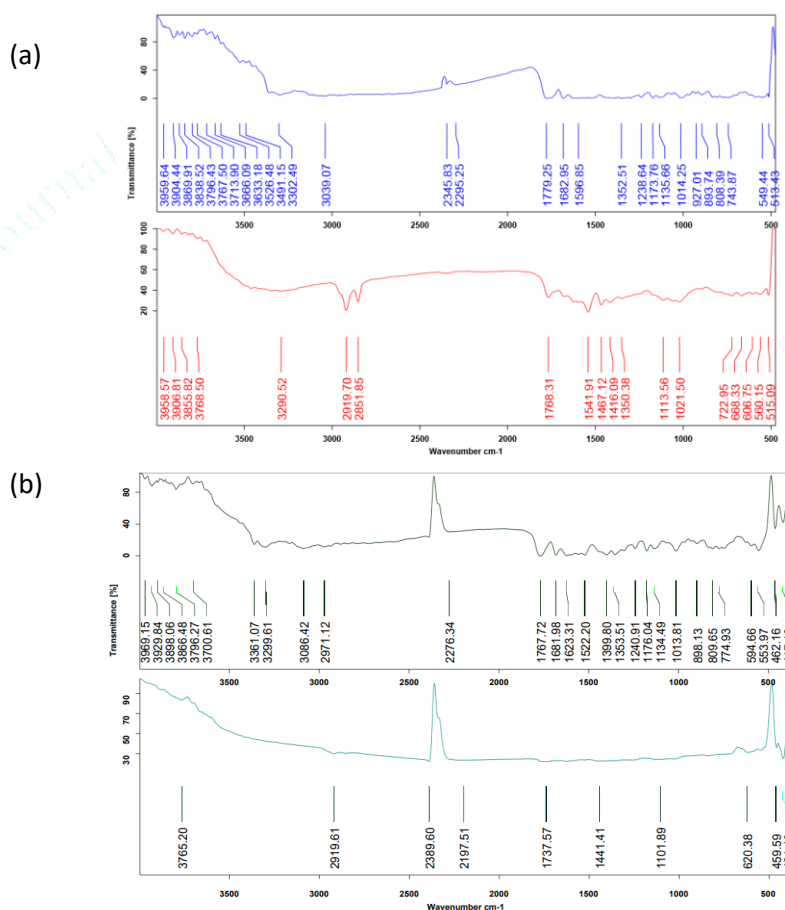


Figure 4: Comparison of FTIR study of nitazoxanide (a) Pure nitazoxanide and SPH,

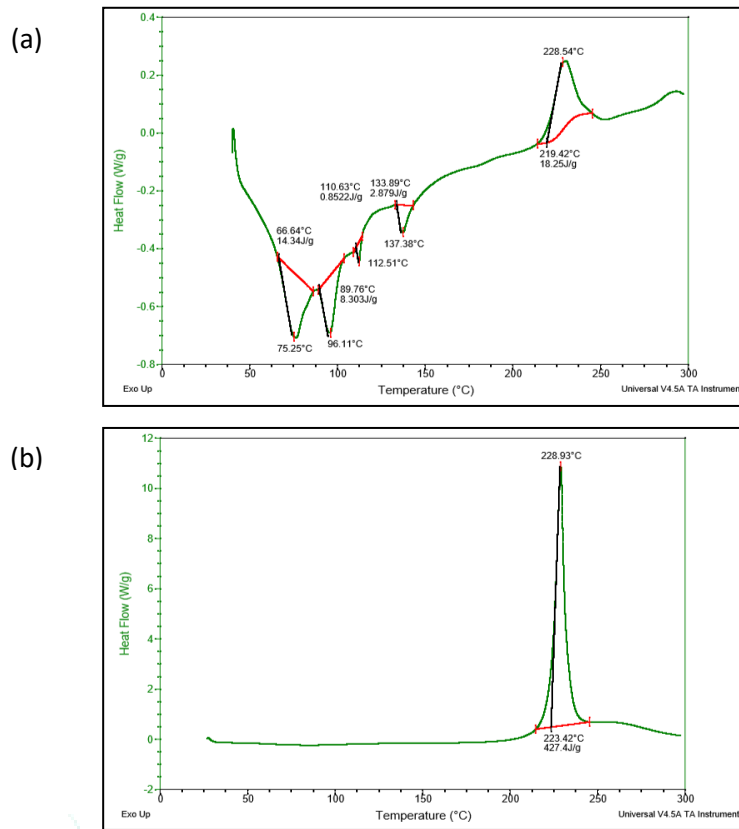


Figure 5: DSC graphs for (a) nitazoxanide loaded SPH tablet (b) Pure Nitazoxanide



Figure 6: Antihelminthic activity-time of paralysis testing of SPH tablets in *Pheretima posthuma*

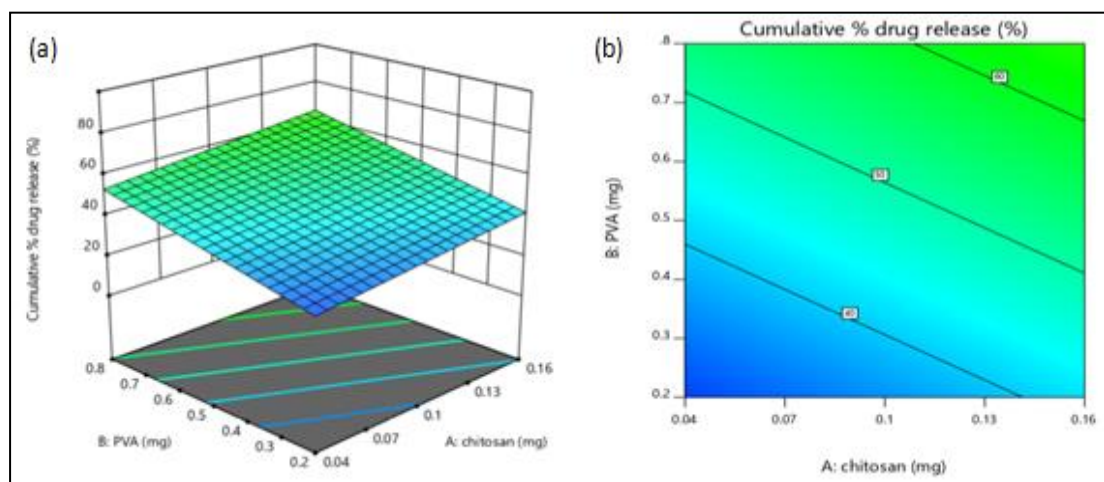


Figure 7: DOE Plots for (a) 3D Surface plot (b) Contour plot

CONCLUSION

This study discusses the preparation of nitazoxanide loaded superporous hydrogel tablets. The superporous hydrogel tablets of nitazoxanide were prepared by direct compression method. This study is mainly based on chitosan and used as a carrier. The SPH tablets of nitazoxanide showed good pre-compressional, post-compressional properties and better in-vitro drug release profiles are showed in this work. FI-IR and DSC studies showed that the drug was compatible with SPHC particles. The formulation F10 prepared by direct compression containing SPH tablets of nitazoxanide were optimized by using 2³ factorial design and prepared by cross linking technique exhibited good swelling index as well as maximum rate of drug release. So, finally all formulations were successfully formulated. This study also demonstrates that superporous hydrogel tablets may be suitable for use as a gastro-retentive drug delivery system.

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