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磷氰酸根负离子对构建新型含磷化合物的研究

The Phosphaetynolate Anion as the Building Block for Construction of Novel Phosphorus Frameworks: An Experimental and Theoretical Study

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The Phosphaetynolate Anion as the Building Block for Construction of Novel Phosphorus Frameworks: An Experimental and Theoretical Study

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requirements for the degree of

Doctor of Philosophy

in

Chemistry

by

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DEDICATION

Dedicated to my parents and sister
For their endless love and all aspects of support

Dedicated to my girlfriend Sunny
For her endless love and understanding

致谢

I would like to thank all Zhao and Bertrand group members, both past and present, for helpful chemistry (and general) discussions and for making the lab a very enjoyable place to work.

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摘要

磷是一种非常重要的化学元素，并且涉及到化学科学的各个领域。磷化学的研究范围广泛，涉及从材料科学到生物化学。近年，对于磷氰酸根负离子（phosphaetynolate anion, PCO^- ）的研究表明，该离子可以作为“磷负离子”的供体，构建新型结构复杂的磷化合物。整篇博士论文以磷氰酸根负离子为出发点，以实验和理论相结合的研究方法，合成分离表征了一系列新型含磷化合物。论文共分为六章，在第一章绪论里，我们将介绍磷氰酸根负离子及其反应活性的背景。第二章，我们重点介绍以磷氰酸根负离子为原料，用紫外线照射的方法，制备第一例单线态的磷宾（phosphinidene）以及反应性研究。在第三章，我们将介绍第一例路易斯碱稳定的二配位磷正离子母体（parent phosphenium）的合成。第四章，我们将介绍磷氰酸根负离子过渡金属（Cu, Au, Co, Ir）化合物的合成及其反应活性研究。值得一提的是，利用 IrPCO 化合物特殊的反应性，合成制备了第一例双金属稳定的 P=P (dimetalladiphosphene) 结构。在第五章，我们将以现代量子化学的方法研究磷氰酸根负离子的环加成反应。解释阐明了电荷分布和轨道重叠对这类反应高选择性的调控作用。第六章是全文总结。

关键词： 磷氰酸根负离子，磷宾，二配位磷正离子，过渡金属，环加成，密度泛函理论

Abstract

Phosphorus is an essential element in all fields of chemical science, ranging from material science to biochemistry. Recently, phosphaetynolate anion (PCO^-), formally featuring a phosphorus of oxidation state -1, has emerged to be an excellent building block for construction of structurally sophisticated compounds. The thesis will focus on synthesis and isolation of a series of phosphorus-containing compounds prepared from PCO^- . In Chapter 1, we will briefly introduce the background of PCO^- chemistry, including its synthesis and reactivity. In Chapter 2, we will prepare an extremely reactive monocoordinated phosphorus species from PCO^- under photolysis, namely a singlet phosphinophosphinidene. In Chapter 3, we successfully isolate the first base-stabilized parent phosphenium cation and related species. In Chapter 4, we prepare several transition metal phosphaetynolate complexes MPCO ($\text{M} = \text{Cu, Au, Co, Ir}$) and study their reactivities. It is very important to note that IrPCO is unstable and rearranges into Ir-P=P-Ir , which is the first dimetalladiphosphene. Finally, in Chapter 5, a thorough theoretical study is carried out to understand the cycloaddition chemistry of PCO^- , showing that electronic properties of different unsaturated compounds play a crucial role in reactivity and regioselectivity.

Keywords: phosphaetynolate anion, phosphinidene, phosphenium, transition metal, cycloaddition, DFT

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第一章 绪论

1.1 工业制备含磷化合物

1669 年，德国商人 Henning Brandt 将砂石、木炭、石灰、动物尿液放一起加热提炼出了白磷(图 1.1, P_4)^[1]。这是人类首次发现磷元素。

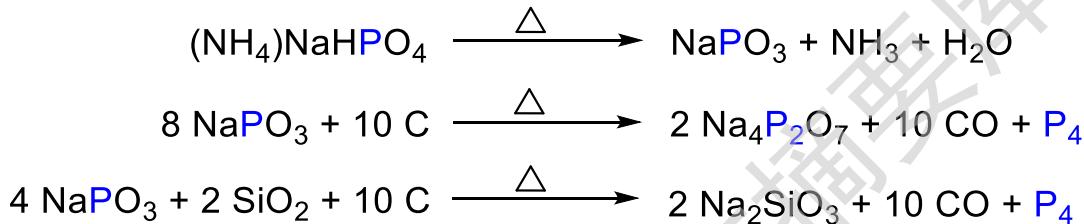


图 1.1 白磷 P_4 的制备

事实上，近一个世纪以来，大多数含磷化学品的制备仍然采取白磷为原料^[2]。如图 1.2 所示，天然磷矿中(phosphate rock)的磷元素多以+5 价的形式存在，经过湿法加工，可以直接转化为磷酸使用，但是此方法会产生大量的副产物石膏($CaSO_4$)。另外一种方式是将磷矿用碳粉还原为白磷。在工业上，大多数白磷会被氧化为三氯化磷(PCl_3)，再转化为人类常用的有机磷化合物。

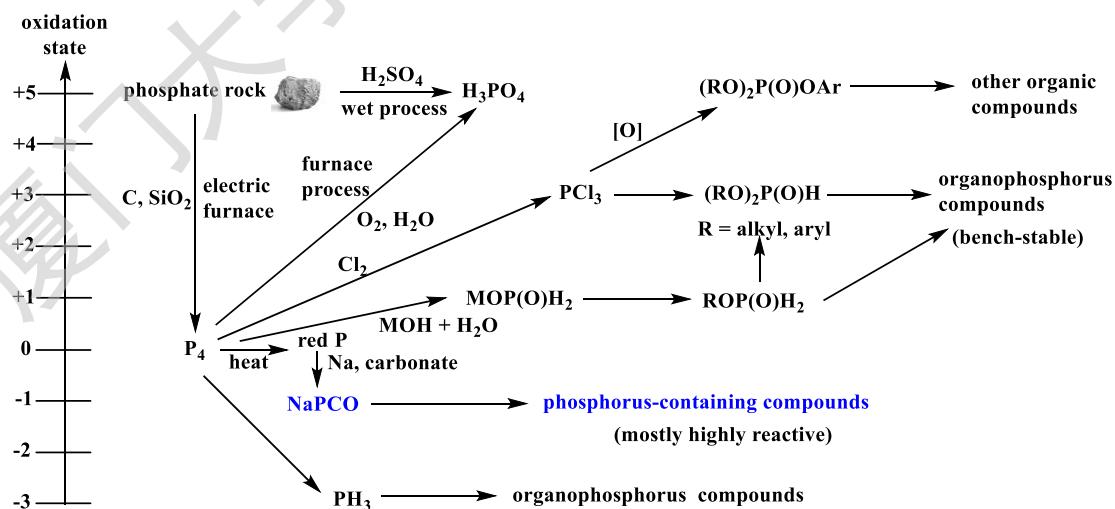


图 1.2 含磷化合物的制备

传统上, 氢亚磷酸酯(*H*-phosphonates)或氢次磷酸酯(*H*-phosphinates)是重要的磷工业中间体。自 1945 年, Todd^[3]发现了氢亚磷酸酯可以与亲核试剂的反应后, 以氢亚磷酸酯或二级膦氧(*H*-phosphine oxide)为磷源来制备含磷化合物的方法受到广泛重视并取得了许多进展^[4]。虽然磷烷(PH_3)是一种易自燃的剧毒气体, 但目前仍然有很多工业需要以其为原料^[2]。

1.2 低配位磷化合物的合成和反应性

1.2.1 磷氰酸类化合物的合成和反应

磷氰酸根负离子(phosphaethynolate anion, PCO^-)是氰酸根负离子(cyanate anion, NCO^-)的含磷类似物。在 1992 年, 磷氰酸根负离子的锂盐(LiPCO)^[5]首次被德国化学家 Becker 和 Westerhausen 成功合成。但是其合成方法苛刻, 并且 LiPCO 自身稳定性较差, 造成磷氰酸根负离子的研究并不多。在 2014 年, 瑞士化学家 Grützmacher 等人^[6]报道了一种高效简便合成磷氰酸钠(NaPCO)的新方法(图 1.3)。该方法以空气和水均稳定的红磷为起始原料, 用萘钠作为强还原剂, 并原位与叔丁醇、碳酸酯反应, 可以大量制备磷氰酸钠的二氧六环加合物。现代量子化学计算表明, 磷氰酸根负离子是以下两种主要共振式的杂合体(图 1.4)。有意思的是, 磷氰酸钠可以在水中存活数小时, 初步研究表明其对很多不饱和有机化合物有较高的活性^[7](图 1.5)。

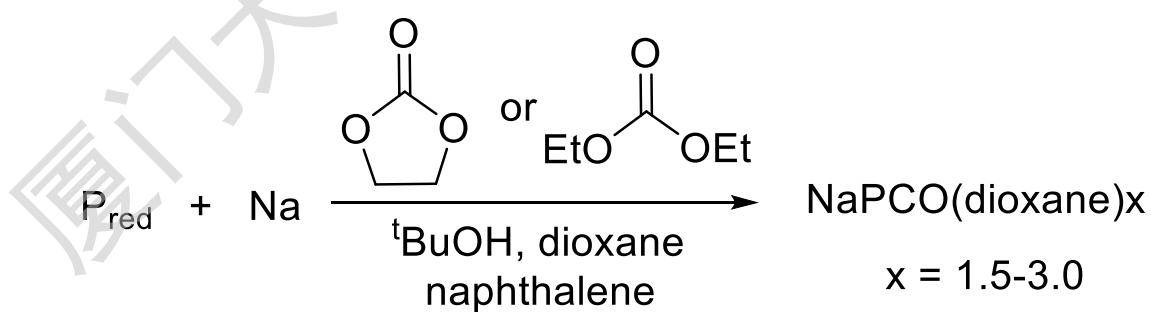


图 1.3 磷氰酸钠的合成

举例来说, 磷氰酸根负离子可以与烯酮(ketene)或者碳二亚胺化合物(carbodiimide)发生环加成反应, 生成四元环的阴离子产物^[7a](图 1.5A,C)。此外, 磷氰酸根负离子也可以与缺电子的炔烃发生环加成反应生成相应的磷杂苯^[7b](图

1.5B,E)。磷氰酸根负离子还可以与 2-羰基吡喃发生[2+4]环加成反应，脱去一分子的二氧化碳，形成磷杂苯酚产物^[7b](图 1.5D)。

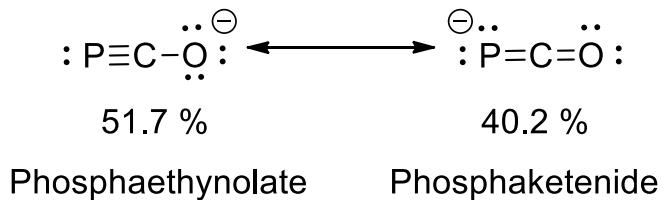


图 1.4 磷氰酸根负离子的自然共振分析

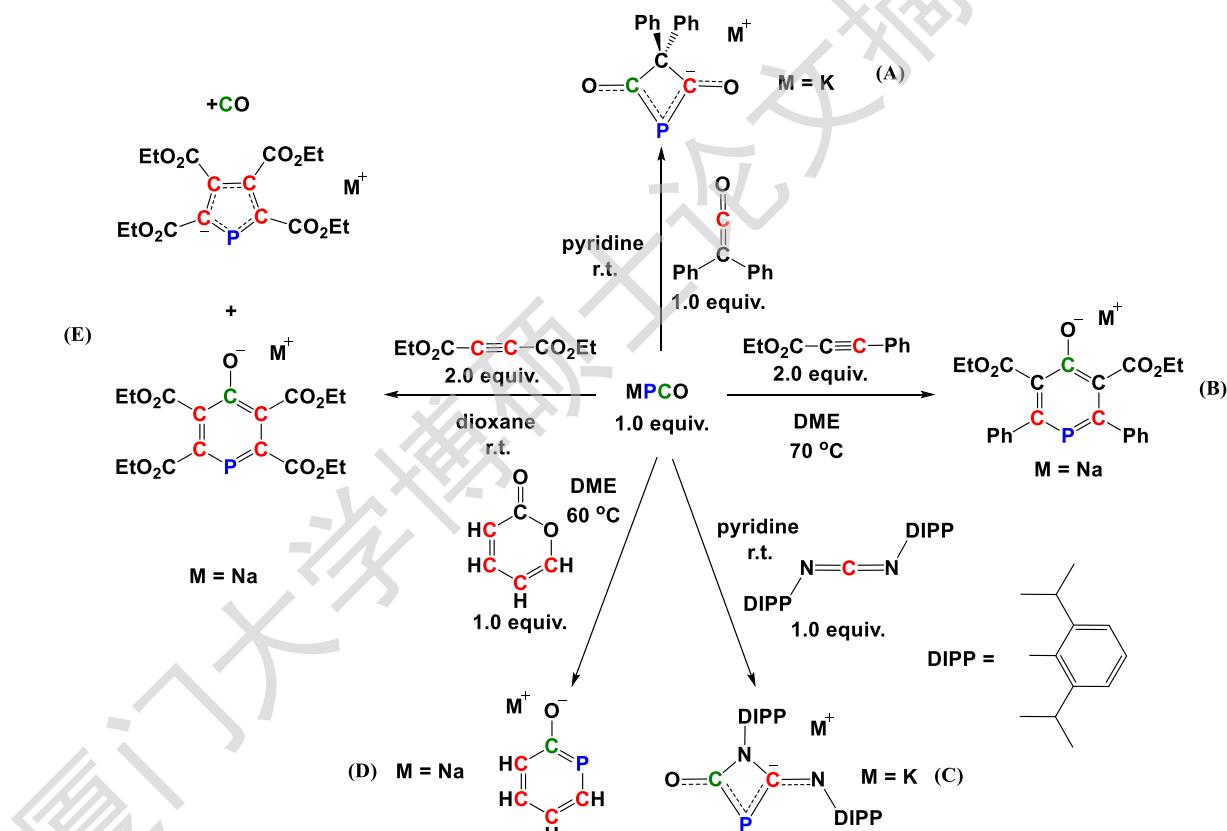


图 1.5 磷氰酸根负离子参与的环加成反应

由于磷氰酸根负离子有很强的还原性^[8]，所以可以被常用的氧化剂氧化(例如 I₂或者 SO₂)，生成四聚体双负离子^[9](图 1.6)。

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