Effect of alkali/mixed alkali metal ions on the thermal and spectral characteristics of Dy³⁺:B₂O₃-PbO-Al₂O₃-ZnO glasses

ABSTRACT

Thermal and spectroscopic features of 50 B_2O_3 -10 PbO-10 Al₂O₃-10 ZnO-(x) Li₂O-(y) Na₂O-(z) K₂O-1.0 Dy₂O₃ (mol %) (x = 19, y = 0, and z = 0; x = 0, y = 19, and z = 0; x = 0, y = 0, and z = 19; x = 9.5, y = 9.5, and z = 0; x = 9.5, y = 0, and z = 9.5; x = 0, y = 9.5, and z = 9.5) glasses, that were fabricated by utilizing melt-quenching approach, are investigated by thermogravimetric analysis (TGA), differential scanning calorimetry (DSC), optical absorption, photoluminescence excitation (PLE), photoluminescence (PL), and PL decay lifetimes. PL spectra for all the Dy³⁺-doped samples show emission bands at 453 nm (blue), 482 nm (blue), 573 nm (yellow), 662 nm (red), and 752 nm (red) corresponding to the ${}^{4}I_{15/2} \rightarrow {}^{6}H_{15/2}$, ${}^{4}F_{9/2} \rightarrow {}^{6}H_{15/2}$, ${}^{4}F_{9/2} \rightarrow {}^{6}H_{13/2}$, ${}^{4}F_{9/2} \rightarrow {}^{6}H_{11/2}$, and ${}^{4}F_{9/2} \rightarrow {}^{6}H_{9/2}$ transitions, respectively, upon excitation at 350 nm. Here, Dy³⁺: Li–Na glass shows the highest PL intensity for all identified emissions. The yellow-to-blue (Y/B) emission intensity ratio (varied within the range 1.257–1.376), CIE chromaticity coordinates (x,y) (slight variation between (0.3410, 0.3802) and (0.3495, 0.3872), and correlated color temperatures (CCTs) (changed from 4953 K to 5212 K) are calculated following the PL spectra. Dy^{3+} : ${}^{4}F_{9/2}$ decay curves show non-exponential behavior and are fitted by the Inokuti-Hirayama (I-H) model, where S = 6 shows best fit, indicating dipole-dipole (d-d) interactions for Dy^{3+} excited (donor) and ground state (acceptor) ions.

Keyword: Dy³⁺: B₂O₃-PbO-Al₂O₃-ZnO glass; TGA/DSC; Judd-Ofelt theory; Luminescence property; Decay lifetime; Inokuti-Hirayama (I–H) model