

1 *Does Vegetation Affect the Methane Oxidation Efficiency of Passive Biosystems?*

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12

### 13 **Abstract**

14 It is often reported in the technical literature that the presence of vegetation improves the  
15 methane oxidation efficiency of biosystems; however, the phenomena involved and biosystem  
16 performance results are still poorly documented, particularly in the field. This triggered a study  
17 to assess the importance of vegetation in methane oxidation efficiency (MOE). In this study, 4  
18 large scale columns, each filled with sand, topsoil and a mixture of compost and topsoil were  
19 tested under controlled conditions in the laboratory and partially controlled conditions in the  
20 field. Four series of laboratory tests and two series of field tests were performed. 4 different plant

21 covers were tested for each series: *Trifolium repens* L. (White clover), *Phleum pratense* L.  
22 (Timothy grass), a mixture of both, and bare soil as the control biosystem. The study results  
23 indicated that up to a loading equal to 100 g CH<sub>4</sub>/m<sup>2</sup>/d, the type of plant cover did not influence  
24 the oxidation rates, and the MOE was quite high ( $\geq 95\%$ ) in all columns. Beyond this point, the  
25 oxidation rate continued to increase, reaching 253 and 179 g CH<sub>4</sub>/m<sup>2</sup>/d in laboratory and field  
26 tests respectively. In the end, the bare soil achieved as high or higher MOEs than vegetated  
27 biosystems. Despite the fact that the findings of this study cannot be generalized to other types of  
28 biosystems and plants and that the vegetation types tested were not fully grown, it was shown  
29 that for the short-term tests performed and the types of substrates and plants used herein,  
30 vegetation does not seem to be a key factor for enhancing biosystem performance. This key  
31 conclusion does not corroborate the conclusion of the relatively few studies published in the  
32 technical literature assessing the importance of vegetation in MOE.

33

34 Keywords: Landfill final covers; Landfill gas emission abatement; Plant cover

35

## 36 **1. Introduction**

37 Methane (CH<sub>4</sub>) is the main greenhouse gas emitted by landfills and represents 18% of the global  
38 anthropogenic CH<sub>4</sub> emissions (Bogner et al., 2007). Passive methane oxidation biosystems  
39 (PMOBs) are considered one of the most cost-effective technologies for mitigation of CH<sub>4</sub>  
40 emissions from landfills (IPCC, 2007). In recent years, considerable research has focused on  
41 improving these biosystems. They are usually made up of a sequence of soil layers, including a  
42 gas distribution layer (GDL) and a methane oxidation layer (MOL). The performance of passive  
43 methane oxidation biosystems (PMOBs) depends on several environmental factors, including  
44 temperature, degree of water saturation within the soils, organic matter content of the MOL,  
45 several other soil properties and characteristics and, last but not least, the presence of vegetation.

46 According to Reay et al. (2005), vegetation is a key determinant of biotic CH<sub>4</sub> oxidation. Despite  
47 this finding, its importance on the efficiency of PMOBs is still relatively poorly documented,  
48 particularly in the field, where, to the authors' knowledge, no technical papers have been  
49 published so far. This paper focuses on evaluating the importance of vegetation on CH<sub>4</sub>  
50 oxidation efficiency within engineered biosystems tested on a landfill and in the laboratory.

51 Documentation on CH<sub>4</sub> oxidation and emissions from several vegetated mediums (wetland rice  
52 fields, freshwater marshes, forest soil, grassland or cultivated soil) is widely available (Ding et  
53 al., 2004; Jia et al., 2001; Keppler et al., 2006; Nouchi et al., 1990; Reay et al., 2005; Watanabe  
54 et al., 1997). The effect of vegetation has been related to several bio-physico-chemical processes  
55 such as: diffusion, biological controls and CH<sub>4</sub> transport through the root system. Indeed,  
56 transport through vascular plants has been identified as one of the main pathways of CH<sub>4</sub>  
57 exchange between the atmosphere and the soil (Chanton, 2005; Chanton et al., 1989; Ding et al.,

58 2004; Schütz et al., 1989). Chanton (2005) reported methane consumption by the roots of two  
59 aquatic plants: *Pontederia cordata* and *Sagittaria lancifolia*. Vascular plants can help promote the  
60 diffusion of O<sub>2</sub> from the atmosphere to the rhizosphere through the aerenchyma system. In  
61 addition, plant roots can produce exudates and release them to the rhizosphere, which may  
62 substantially influence chemical or biological soil properties in ways that can help the  
63 biogeochemical process of methane oxidation (Hilger et al., 2000; Stralis-Pavese et al., 2004;  
64 Tanthachoon et al., 2008). Moreover, in a study of the effect of 3 plant species on the CH<sub>4</sub>  
65 oxidation capacity in forest soils, Reay et al. (2005) suggested that the presence of nitrogen-  
66 fixing (N-fixing) plants, such as alder, may result in large reductions in potential CH<sub>4</sub> oxidation  
67 in soils. Popp et al. (2000) reported on the other hand that there were no significant differences  
68 between CH<sub>4</sub> oxidation measured for bulk peat of a non-vegetated site core and the control  
69 vegetated site cores for the same time period. Thereby, quantitative estimates of methane  
70 oxidation in several vegetated mediums may be related to differences in the systems studied such  
71 as the plant species present (King et al., 1990; Van der Nat et al., 1997).

72 Recent research on engineered biosystems for landfills suggests that plants would positively  
73 contribute to CH<sub>4</sub> oxidation (Bohn et al., 2010; Hilger et al., 2000; Reichenauer et al., 2011;  
74 Wang et al., 2008). It is reported that vegetation can improve the air-filled capacity of soils  
75 through the formation of secondary macro-pores by spreading roots. In addition, vegetation  
76 controls moisture infiltration by means of plant evapotranspiration. Finally, plant growth may  
77 also provide nutrients for methanotrophs by root exudates and debris of dead plants and thereby  
78 increase oxidation efficiency. Some potential negative effects of vegetation on CH<sub>4</sub> oxidation  
79 include the potential for plant roots to create preferential channels for CH<sub>4</sub> emissions, and their  
80 competition for O<sub>2</sub> due to root respiration (Wang et al., 2008). Plants debris may also lead to O<sub>2</sub>

81 competition with bacteria degrading the debris and thus decreasing methane oxidation efficiency.  
82 However, the actual integrated effect of plant species on the methane oxidation process in  
83 landfill covers remains poorly documented in the technical literature, particularly in relation to  
84 engineered biosystems for landfills tested under field conditions.

85 In order to verify the validity of the hypothesis that the type of vegetation may affect the CH<sub>4</sub>  
86 oxidation efficiency of a biosystem, the following 3 types of plants were tested under the  
87 controlled conditions prevailing in the laboratory and under the partially controlled conditions of  
88 the Saint-Nicéphore landfill, QC, Canada: 1) White clover (*Trifolium repens L.*), which is a  
89 leguminous plant; 2) Timothy grass (*Phleum pratense L.*); and 3) a mixture of White clover and  
90 Timothy grass. An unplanted biosystem served as control.

91 The research reported in this paper included a series of 4 laboratory tests performed in sequence  
92 and 2 field tests. Each test was comprised of 4 columns containing the same sequence of  
93 materials and one of the 4 vegetation covers presented above. The methane oxidation efficiencies  
94 (MOE) of the columns were determined for several methane loadings, while temperature and  
95 degree of water saturation profiles were obtained throughout the tests by a data acquisition  
96 system.

97

## 98 2. Materials and Methods

### 99 2.1. Experimental set-up

#### 100 2.1.1. Laboratory set-up

101 Four columns measuring 0.61 x 0.46 x 0.52 m were built for the laboratory-scale experiment. A  
102 schematic of their design is presented in Figure 1. The methane oxidation layer (MOL) of the  
103 biosystems was constituted of the following materials, from the bottom up: a 0.30-m layer of fine  
104 sand, a 0.075-m layer of topsoil, and a 0.075-m layer of topsoil enriched with compost (5% dry  
105 weight). The gas distribution layer (GDL) that is usually constructed under the MOL was  
106 substituted by an empty space. The GDL and MOL were separated by a 2-cm thick perforated  
107 plastic plate covered by a fine wire mesh. Seepage water was collected at the bottom of the  
108 column and evacuated through an outlet.

109 The columns were set in an explosion-proof laboratory. A lighting system was installed in order  
110 to foster plant growth. It consisted of 100-Watt fluorescent lamps controlled by an electronic unit  
111 that was set to provide the required number of hours of light per day (14 h/d) and a light intensity  
112 of approximately 8000 lx. The temperature of the laboratory was maintained at 19°C by a  
113 cooling system. An aeration unit allowed the renewal of air in the laboratory.

114

115 Figure 1: Experimental design of laboratory columns

116

117 2.1.2. Field set-up

118 At the Saint-Nicéphore landfill site, in Quebec, Canada, four experimental biosystems measuring  
119 0.9 m x 0.9 m were installed during spring 2013. As shown in Figure 2, their design was quite  
120 similar to that adopted for the laboratory columns. The MOL material was the same as for the  
121 laboratory. A 0.10-m thick gas distribution layer underlying the MOL was built with 12.7-mm  
122 gravel topped by a fine wire mesh to avoid clogging of the gravel pores by the MOL material.  
123 The sides of the four columns were thermally insulated by surrounding them with a 0.30-m thick  
124 layer of locally-available silt (Figure 2). This insulation helped prevent lateral migration of  
125 moisture within the columns due to thermal gradients.

126

127 Figure 2: Experimental design of field columns

128

129 The cover materials used in both the laboratory and field columns were those widely available  
130 on-site. Their characteristics, presented in Table 1, were fixed and the same for all columns.  
131 Sand was placed in three 0.10-m layers and compacted to obtain a dry density of 1690 Kg/m<sup>3</sup>.  
132 The 0.075-m layers of unenriched and enriched topsoil were compacted at a density of  
133 1200 Kg/m<sup>3</sup>. These degrees of compaction were chosen to reproduce an existing experimental  
134 biosystem at the Saint-Nicéphore landfill site.

135

136 Table 1: Characteristics of cover materials

137

## 138 2.2. Experimental procedure

139 For this study, four series of 4 column tests were performed under controlled laboratory  
140 conditions, and two series of 4 column tests were performed under partially controlled field  
141 conditions. Each laboratory test was conducted over approximately 5 months, including the  
142 acclimatization period and plant germination period. In order to observe the influence of climatic  
143 conditions on field results, two field tests were performed each over two different seasons. The  
144 first test started in May and ended in August (spring to summer), which corresponds to the best  
145 growth period for plants. The second one started in August, a less favourable period for plant  
146 growth, and ended in October (summer to fall).

147 Before the laboratory and field tests, an air tightness test was performed within the columns. A  
148 controlled CH<sub>4</sub> loading was introduced at the bottom of the previously emptied columns and the  
149 CH<sub>4</sub> flux out of the column was monitored to assess losses due to leaks. The field columns  
150 showed a loss of about 6% of the CH<sub>4</sub> loading, while in the laboratory, the loss was 3.5%.

151

### 152 2.2.1. Plant seed

153 Two different plant species were selected and used in this study, based on their abundance on  
154 site. The first one was the *Trifolium repens L.*, White clover, a leguminous and perennial plant.  
155 According to the USDA NRCS Plant Materials Program, it is considered to be a beneficial  
156 component of seed mixture because of its N-fixing property by converting atmospheric nitrogen  
157 - through its root system - into a form that is usable by other plants and microorganisms in the  
158 soil. The second one was the *Phleum pratense L.*, Timothy grass. It is a perennial plant that has a  
159 shallow, compact, and fibrous root system. Timothy grass has a relatively high demand for



160 nutrients, especially nitrogen, which is often the major limiting nutrient for Timothy growth.  
161 However, its competition for N is low at the beginning of its growth.

162 For the experiments, columns were seeded with Timothy grass (TG column), White clover (WC  
163 column) and a mixture of 67% Timothy grass and 33% White clover (MIX column). Column 4  
164 was the control column and was therefore not seeded (bare soil, BS column). Based on  
165 preliminary seeding tests, the seeding density for each column was 6g/m<sup>2</sup>.

166

### 167 2.2.2. Irrigation

168 Field columns were naturally irrigated with rainwater, while laboratory columns were watered  
169 manually following approximately the 30-year monthly average rainfall for Drummondville,  
170 Quebec. The Environment Canada database was used for this purpose. Due to the intense  
171 aeration inside the explosion-proof chamber (6 air renewals per hour) where the columns were  
172 installed, the material dried faster than in normal field conditions. As a consequence, the amount  
173 of water added to the columns was adjusted to compensate for this condition. Thereby, the  
174 protocol may diverge from actual field conditions. Daily irrigation was recorded during the  
175 experimental period. The average daily precipitation was 2.7-mm in the laboratory and 1.54-mm  
176 in the field.

177

### 178 2.2.3. Biogas loading

179 After the columns were filled with soil, synthetic biogas (50% CH<sub>4</sub>/50% CO<sub>2</sub>, v/v) was applied  
180 to the bottom of the laboratory columns, whereas raw landfill gas (LFG) was applied to the field

181 columns. Methane loadings in each column were controlled by flow meters (Gilmont  
182 Instruments, Inc. GF 1060). Before injection into the column, the synthetic biogas was  
183 moisturized by bubbling through a water-filled bottle to prevent desiccation of soil. The LFG  
184 was already sufficiently wet. Columns were kept at the residual landfill gas exposure of  
185  $8 \text{ g CH}_4/\text{m}^2/\text{d}$  for one month (two weeks before and after seeding) to allow plants to germinate  
186 and methanotrophs to grow as reported in Kightley et al. (1995). Subsequently, the loading was  
187 increased gradually from 8 to 270 and  $180 \text{ g CH}_4/\text{m}^2/\text{d}$  for laboratory and field tests,  
188 respectively, as presented in Figures 3 and 4.

189

#### 190 2.2.4. Column instrumentation

191 Each column was equipped with temperature and water content probes (ECTM-5, from Decagon  
192 Devices), placed at 10, 20 and 30 cm below the surface. During the experimental period, daily  
193 values were recorded manually in laboratory tests using a ProCheck Decagon Device and  
194 automatically in field tests with data loggers. To better visualize the results, the water content  
195 was converted into degree of saturation using the usual soil mechanic formulas.

196 In order to collect gas samples from the headspace and be able to estimate  $\text{CH}_4$  surface  
197 emissions, 110 L and 300 L PVC caps were constructed respectively for laboratory and field  
198 columns, and were installed only when  $\text{CH}_4$  surface emission measurements were taken.

199 In the laboratory, the top of the cap was perforated and four tubes were introduced at 80, 60, 40,  
200 and 20% of the total height of the cap to cover its entire surface and volume (Figure 1). The  $\text{CH}_4$   
201 concentration in the headspace was measured with a gas chromatograph (Micro GC 3000A,  
202 Agilent Technologies). Gas samples were collected from each tube of the headspace with a

203 syringe at a regular frequency, and immediately analyzed.

204 In the field columns, only one sampling point was placed at the center of the cap. CH<sub>4</sub> surface  
205 emission was measured using a portable flame ionization detector (TVA-1000B, Thermo  
206 Scientific) equipped with a data acquisition system.

207

#### 208 2.2.5. Mass balance calculation of CH<sub>4</sub> oxidation efficiencies

209 The CH<sub>4</sub> oxidation efficiency (MOE) was calculated using the mass balance method in the  
210 headspace. This method is based on the CH<sub>4</sub> loading and the CH<sub>4</sub> surface emission of the  
211 biosystem. The MOE was calculated as follows:

$$212 \quad \text{MOE} = \frac{\text{Flux}_{\text{in}} - \text{Flux}_{\text{out}}}{\text{Flux}_{\text{in}}} \times 100 \quad (1)$$

213 where MOE is expressed as the percentage of CH<sub>4</sub> loading oxidized, **Flux<sub>in</sub>** and **Flux<sub>out</sub>** are the  
214 CH<sub>4</sub> inlet and outlet fluxes respectively (g CH<sub>4</sub>/m<sup>2</sup>/d). CH<sub>4</sub> outlet flux of the column was  
215 determined from the linear regression analysis of the temporal increase in chamber CH<sub>4</sub>  
216 concentration. The oxidation rate was calculated by multiplying the MOE by the CH<sub>4</sub> loading.

217

#### 218 2.2.6. Statistical analysis

219 A two-factor analysis of variance (ANOVA) was used to assess the effect of plant type and CH<sub>4</sub>  
220 loading on the methane oxidation efficiency of biosystems. Because of the substantial database  
221 obtained in the laboratory (4 replicates of each treatment), the significance threshold was

222 accepted at a level of  $p < 0.05$ , this level was also maintained for field results. Knowing the  
223 significant effect of gas loading on efficiency, and to isolate the effect of plant cover on the  
224 latter, a quadratic model was used. This model was also the one suggested when maximizing the  
225 predicted and adjusted R-squared values. In order to evaluate the effects of plant cover on depth  
226 profiles of temperature and degree of saturation of biosystems, another two-way ANOVA was  
227 performed. The significance threshold was maintained at  $p < 0.05$ .

228

### 229 **3. Results and discussion**

#### 230 **3.1. Methane oxidation efficiencies under laboratory conditions**

231 The methane oxidation efficiency (MOE) and oxidation rate values for different CH<sub>4</sub> loadings  
232 under laboratory conditions are presented in Figure 3. The results presented herein represent the  
233 average of the values obtained from the four perforated tubes on the top of the PVC cap. Since  
234 the MOE values calculated - based on the data obtained from those four sampling points - did not  
235 show a significant difference ( $< 0.5\%$  at all times), it was concluded that the gas within the  
236 headspace was uniformly distributed.

237 Throughout the present study, MOEs were 100% for loadings up to 125 g CH<sub>4</sub>/m<sup>2</sup>/d. Differences  
238 in MOEs of the biosystems became appreciable above this value. In spite of that, the maximum  
239 difference in MOEs between biosystems did not exceed 8% for the higher loading values.

240 As can be observed in Figure 3A-D, the oxidation rates of the 4 columns continued to increase  
241 with increasing CH<sub>4</sub> loadings for the four tests and the oxidation rates of the WC and TG  
242 columns were quite similar. Furthermore, except for test 1, the highest efficiency was obtained

243 for the BS columns, whose average oxidation rate was  $240 \text{ g CH}_4/\text{m}^2/\text{d}$  ( $\pm 2.8 \text{ g CH}_4/\text{m}^2/\text{d}$ ). For  
244 test 1, the maximum oxidation rate ( $255 \text{ g CH}_4/\text{m}^2/\text{d}$ ) was obtained for the WC column. The  
245 MOE of the BS column remained close to 90% for all laboratory tests. The constant values  
246 obtained for this control test confirm the good reproducibility of the adopted protocol.

247 For the TG WC and MIX columns, when the loading became greater than  $125 \text{ g CH}_4/\text{m}^2/\text{d}$ , the  
248 oxidation rates (and associated MOEs) started to differ from one test to another. A definite  
249 explanation cannot be provided herein considering the limited test result database. Variations  
250 observed in plant root density from one test to another may partly explain why oxidation rates  
251 did not remain constant for all columns (Ndanga et al., 2013). In fact, it could be hypothesized  
252 that preferential pathways, usually associated with the root system (Bohn et al., 2010; Scheutz et  
253 al., 2009; Wang et al., 2008), led to the increasing differences in MOEs between the columns for  
254 loadings greater than  $125 \text{ g CH}_4/\text{m}^2/\text{d}$ .

255 Despite the differences observed for loadings greater than  $125 \text{ g CH}_4/\text{m}^2/\text{d}$ , the MOE remained  
256 greater than 80% up to a loading of  $225 \text{ g CH}_4/\text{m}^2/\text{d}$ , irrespective of plant cover. The lowest  
257 oxidation rate was  $180 \text{ g CH}_4/\text{m}^2/\text{d}$ , which was obtained for the WC column at the end of test 4.  
258 This oxidation rate is nonetheless very high considering that it far exceeds what is considered the  
259 average methane loading applied to cover systems, in several landfills with gas collection  
260 systems in the U.S. and Canada, i.e.,  $28 \text{ g CH}_4/\text{m}^2/\text{d}$  (Capanema and Cabral, 2012).

261 In order to assess possible variations of MOE within the same loading, 3 MOE measurements  
262 were taken for each loading during the third laboratory test. Figure 3C presents the mean  
263 oxidation rates and MOEs for each loading increment. The first measurement was taken at least 7  
264 days after increasing the loading. For all loadings and all columns, there was never a significant

265 variation in MOE between the 3 measurements. The standard error generally did not exceed 3%.  
266 Therefore, values obtained with one measurement in other laboratory tests were considered  
267 representative of the real efficiency of columns.

268

269 Figure 3: Methane oxidation efficiency and oxidation rates at different CH<sub>4</sub> loadings under  
270 laboratory conditions. A - Lab test 1; B - Lab test 2; C - Lab test 3; D - Lab test 4

271

### 272 **3.2. Methane oxidation efficiencies under field conditions**

273 Figure 4 presents the MOEs and oxidation rates of the field column tests performed during  
274 “spring to summer” (field test 2; Figure 4A) and “summer to fall” (field test 3; Figure 4B). The  
275 MOEs during field test 2 remained at ~100% up to a loading equal to 95 g CH<sub>4</sub>/m<sup>2</sup>/d, irrelevant  
276 of plant cover. At the end of this test, MOE values remained greater than 80%. Similarly to what  
277 was obtained for the laboratory tests, the BS column was the most efficient at the end of the test  
278 and the oxidation rate reached 179 g CH<sub>4</sub>/m<sup>2</sup>/d. During field test 3, the MOEs were greater than  
279 95% for all columns up to a loading equal to 95 g CH<sub>4</sub>/m<sup>2</sup>/d. Until near the end of this test MOEs  
280 remained greater than 80% but, at the last measurement, the efficiencies dropped drastically  
281 when the air temperature reached the freezing point ( $\leq 0^{\circ}\text{C}$ ) before this last measurement.  
282 Nevertheless, in spite of colder weather, oxidation rate values remained high (between 100 and  
283 115 g CH<sub>4</sub>/m<sup>2</sup>/d). Probably due to decreasing air temperatures (as fall approached), MOEs in  
284 field test 2 were generally greater than MOEs in field test 3.

285 Repeatability of field measurements was verified by performing at least 2 MOE measurements  
286 for each loading during the field tests. In this case, the disparity in MOE values was greater than  
287 the disparity obtained during the laboratory tests. Indeed, for WC it reached a peak of 16% for a  
288 loading equal to 180 g CH<sub>4</sub>/m<sup>2</sup>/d, and 12% at 95 and 125 g CH<sub>4</sub>/m<sup>2</sup>/d. For all the other loadings  
289 and types of plants, the disparity remained less than 5%. For sake of presentation in Figure 4,  
290 only the mean oxidation rates and MOEs for each loading increment were retained.

291

292 Figure 4: Methane oxidation efficiency and oxidation rates at different CH<sub>4</sub> loadings under field  
293 conditions. A - Field test 1; B - Field test 2

294

### 295 **3.3. Discussion about methane oxidation efficiencies**

296 A two-factor ANOVA of all the MOE values of all columns was performed to further assess the  
297 level of influence of vegetation on methane oxidation. The analyses were split into two: the first  
298 one for loadings up to 100 g CH<sub>4</sub>/m<sup>2</sup>/d; and the second for loadings greater than this value. For  
299 both laboratory and field tests submitted to loadings lower than 100 g CH<sub>4</sub>/m<sup>2</sup>/d, the difference in  
300 MOE values was not significant ( $p < 0.05$ ). This means that the type – or absence – of plant cover  
301 does not influence the performance of a PMOB up to 100 g CH<sub>4</sub>/m<sup>2</sup>/d, which, as mentioned  
302 before, is a much greater value than what can be expected as far as residual CH<sub>4</sub> emissions from  
303 landfills are concerned.

304 For loadings greater than 100 g CH<sub>4</sub>/m<sup>2</sup>/d, the difference between the MOEs was statistically  
305 significant ( $F_{3,101} = 4.67$ ,  $p < .01$  and  $F_{3,31} = 4.82$ ,  $p < .01$  respectively for laboratory and field

306 tests). As can be clearly observed in Figure 3 and Figure 4, the performance of the BS column  
307 during the test was the highest. Hypothetically, soil diffusivity and macro-pore formation  
308 associated with vegetation may have facilitated the escape of CH<sub>4</sub> to the atmosphere (Bohn et al.,  
309 2010; Scheutz et al., 2009). However, the average deviation between the MOE of the vegetated  
310 columns and the BS column was only 4%. This small difference could be attributed either to  
311 vegetation or to incomplete air tightness of the columns (Section 2.2).

312 It is relevant to note that the increase in CH<sub>4</sub> loading during the laboratory and field tests resulted  
313 in ever greater oxidation rates for all columns and the maximum oxidation capacity was probably  
314 never achieved. A drop in efficiency was observed only in field test 3, when the air temperature  
315 was below the freezing point. The other tests were stopped at relatively high loadings (270 and  
316 180 g CH<sub>4</sub>/m<sup>2</sup>/d for laboratory and field tests, respectively), and high CH<sub>4</sub> oxidation rates were  
317 obtained.

318 All results considered, the differences between observed methane oxidation rates obtained for  
319 each type of vegetation tested were not significant ( $p < 0.05$ ). Robertson et al. (2000) compared  
320 two perennial crops, alfalfa (*Medicago sativa*) and poplar (*Populus* sp.) trees in a field study.  
321 They also observed no difference in the rates of CH<sub>4</sub> oxidation among any of the cropped sites.

322 Therefore, one can conclude that the 3 biosystems with plant covers showed MOE values  
323 comparable to those obtained for the unplanted biosystem, both in the laboratory and in the field.  
324 The findings above clearly diverge from what has been often reported in the technical literature  
325 relating to the positive impact of vegetation on methane oxidation in landfill covers Table 2.

326 In a column study, Bohn (2010) compared the methane oxidation potential of one column with  
327 compost material planted with a mixture of different types of grasses and herbages and three



328 columns with a mixture of clayey silt and greencut compost, unplanted, planted with Canadian  
329 goldenrod (*Solidago canadensis* L.) and planted with a mixture of leguminous plants. Submitted  
330 to a loading equal to 90.0 g CH<sub>4</sub>/m<sup>2</sup>/d , Bohn (2010) observed high methane oxidation in grass  
331 (90.0 g CH<sub>4</sub>/m<sup>2</sup>/d), *S. canadensis* (63 g CH<sub>4</sub>/m<sup>2</sup>/d) and leguminous plants (37 g CH<sub>4</sub>/m<sup>2</sup>/d). Only  
332 the control column showed a negative oxidation rate, i.e. methane production or temporal  
333 methane storage due to a clogged surface.

334 According to Bohn (2010), vegetation improved the soil's diffusivity and physical properties,  
335 which led to a significant and positive effect of vegetation on methane oxidation. Moreover, all  
336 the studies in Table 2 found a positive effect of vegetation on methane oxidation regardless of  
337 soil material used. Several mechanisms were proposed to explain this positive effect, such as  
338 regulation of soil moisture through water uptake and evapotranspiration, and oxygenation of the  
339 soil by plant roots, which create macro-pores therefore enhancing gas diffusion. Another  
340 mechanism is related to root exudates, which serve as selective substrates and promote the  
341 growth of methanotrophs.

342 In comparison with the study herein, the studies presented in Table 2 used different soil materials  
343 to constitute the CH<sub>4</sub> oxidation biosystem. With the exception of the studies by Hilger et al.  
344 (2000) and Wang et al. (2008), who used sandy loam and a red soil, all the others cited in Table  
345 2 used mature compost – mixed or not with soils. Composts are considered by several as the  
346 most suitable material for methane oxidation and plant growth (e.g. Huber-Humer and Lechner,  
347 2003). In the present study, the biosystems tested were made up of sand dominated materials:  
348 fine sand, top soil and enriched top soil. In other words, the materials tested were not – in  
349 principle - as favorable as those used elsewhere to evaluate the impact of vegetation on methane  
350 oxidation in landfill covers.

351 Test duration was an important constraint of this experimental study. At the end of the testing  
352 period (5 months), plants were not fully grown and their root systems were not fully established.  
353 It can be presumed that the microbial community was not fully developed either. In the case of  
354 the studies in Table 2, tests lasted from 6 to 18 months. Test duration might therefore explain – at  
355 least in part - the differences in outcomes between this study and those in Table 2. However,  
356 according to Habekost et al. (2008), 18 months may not be long enough for the vegetation to  
357 fully develop thereby limiting its capacity to influence CH<sub>4</sub> oxidation.

358

359 Table 2: Comparison with results from other studies

360

### 361 **3.4. Temperature and degree of water saturation monitoring**

362 During each field and laboratory test, temperature and degree of water saturation (Sr) values  
363 were recorded periodically. Figure 5 shows the maximum and minimum temperature values  
364 within the columns, and the average temperatures during two arbitrary loading applications in  
365 field (95 and 125 g CH<sub>4</sub>/m<sup>2</sup>/d) and laboratory tests (70 and 225 g CH<sub>4</sub>/m<sup>2</sup>/d). Averages here were  
366 determined for each depth.

367

#### 368 3.4.1. Temperature variations

369 In the beginning of the laboratory test, the temperature within the profiles remained quite similar  
370 to the room temperature (~ 19°C), which was finally the minimum value observed within the  
371 columns (T min in Figure 5a-d). Generally, the temperature within the biosystem remained

372 higher than the air temperature. When the CH<sub>4</sub> loading was increased, the temperature also  
373 increased. This is in agreement with the observation made by Einola et al. (2007), Börjesson et  
374 al. (2004) and Gebert et al. (2003), who observed that temperature increases within biosystems  
375 were attributed to CH<sub>4</sub> oxidation activity. The typical profiles in Figure 5c and Figure 5e show  
376 maximum variations of 5°C of the average temperature values for CH<sub>4</sub> loadings ranging from  
377 70 g CH<sub>4</sub>/m<sup>2</sup>/d to 225 g CH<sub>4</sub>/m<sup>2</sup>/d in laboratory tests, and 13°C for loadings ranging from  
378 95 g CH<sub>4</sub>/m<sup>2</sup>/d to 125 g CH<sub>4</sub>/m<sup>2</sup>/d in field tests at 10-cm depth.

379 Field test 3 (summer to fall) showed lower temperatures at 10 and 20 cm than field test 2 (spring  
380 to summer) throughout the testing period. Despite the fact that near the end of field test 3 the air  
381 temperature dropped below 0°C during the night, the average temperature within the soil  
382 remained higher than 5°C. This suggests that there might still have been oxidation activity within  
383 the columns (Figure 5f) (Einola et al., 2007).

384 However, for laboratory and field studies, there were no significant differences between  
385 temperatures within the profiles, regardless of plant cover (or column). The temperature profiles  
386 were generally quite similar for all columns (SD < 1°C). Furthermore, although the temperatures  
387 were generally lower in the deepest layers (20 and 30 cm depth; Figure 5a-d) of the biosystems  
388 tested, there were noticeable increases in temperature as CH<sub>4</sub> loadings were raised, indicating  
389 that bacterial activity was also occurring at these depths.

390 The highest temperature increase is expected to occur at the oxidation front, which is where  
391 oxidation activity is optimal within a biosystem. In the case of the laboratory and field tests  
392 presented herein, the oxidation front was located between 0 and 10-cm from the surface. In  
393 particular for laboratory test 3, the temperature at 10-cm depth (topsoil layer enriched with

394 compost; within the root zone) reached 33°C, which represented a thermal amplitude of 14°C  
395 compared to the air temperature. The upper 7.5-cm of this 10-cm layer was the most oxygenated  
396 and nutrient-rich part of the biosystems tested compared to the deepest layers (top soil and sand  
397 layers). Jugnia et al. (2008) also observed, during a field test in an organic matter-rich layer of an  
398 experimental landfill cover, an oxidation zone situated between 0 – 10-cm. Therefore, the  
399 nutrient content as well as climatic conditions, O<sub>2</sub> supply, precipitation and physical property of  
400 the soil cover affect the depth of the oxidation front (Berger, 2005; Humer and Lechner, 2001;  
401 Jugnia et al., 2008).

402

403 Figure 5: Temperature profiles within the columns and air temperature in field and lab tests

404

#### 405 3.4.2. Degree of water saturation

406 The degrees of water saturation Sr in Table 3 represent the average and standard deviations of all  
407 recorded values for each test. Since the upper parts of the soil columns were probably the most  
408 affected by precipitation and evapotranspiration, Sr variations were greater at a depth of 10 cm.  
409 Under field conditions, the drier regions of the biosystems were located at a depth of 20 cm.  
410 Moisturized gas from the bottom provided humidity to the soil at 30 cm depth. Consequently Sr  
411 values were the highest where biogas was injected. Despite the fact that the raw biogas was very  
412 wet, Sr values never reached 85%, which is the critical value above which there is pore  
413 occlusion, leading to a substantial decrease in gas migration through a porous system (Aachib et  
414 al., 2004; Cabral et al., 2004; Nagaraj et al., 2006). In the laboratory, the synthetic biogas was

415 also moisturized by bubbling through water-filled bottles before injection into the column;  
416 however, Sr values also remained well below 85% at the bottom of the columns.

417 According to the data obtained, vegetation seems to have a significant effect on Sr values for all  
418 biosystems ( $p < 0.05$ ). Despite the fact that the columns were submitted to the same watering  
419 conditions, the upper layer of the BS column was generally wetter than the vegetated columns,  
420 for both laboratory and field tests. Although evapotranspiration was not monitored, it is probable  
421 that it was the main cause for the lower Sr values within the root zone (first 10 cm). Indeed,  
422 plants are important contributors to reduction of soil moisture by water uptake and transpiration  
423 (Bohn et al., 2010; Reichenauer et al., 2011). Drying of the pores may lead to the formation of  
424 macro pores that eventually facilitate gas migration. Accordingly, the greater efficiency of the  
425 BS column at high loadings might be at least partly attributed to the absence of drier pores.

426 As observed for the MOEs, the variations of Sr values within the BS column were generally less  
427 than those observed for the vegetated columns (Table 3). This confirmed the good  
428 reproducibility of the adopted protocol and is in agreement with the observed significant effect of  
429 the vegetation on Sr values.

430

431 Table 3: Average values of degree of saturation in % and standard deviation for lab and field  
432 tests

433

#### 434 **4. Limitations**

435 One of the limitations of the study presented herein concerns the duration of the tests, which had  
436 to be limited to approximately five months for each repetition. Some vegetated biotic systems,

437 may take several years to completely develop their root system. According to an experimental  
438 study by Habekost et al. (2008), the first effects of the aboveground plant community on the  
439 microbial community composition becomes detectable at least four years after establishing the  
440 grassland systems. According to Habekost et al. (2008), these differences would presumably  
441 increase with time. Considering this, it is probable (but not verified) that the tests in this study  
442 were terminated before full root growth; and, for that matter, the same might have happened in  
443 all the studies referred to in Table 2. Therefore, one cannot ascertain that different types of fully  
444 grown vegetation would not lead to greater differences in MOEs than those obtained herein. This  
445 study showed that there were no perceptible differences in MOE values for loadings lower than  
446  $100 \text{ g CH}_4/\text{m}^2/\text{d}$  and relatively minimal differences for loadings greater than  $100 \text{ g CH}_4/\text{m}^2/\text{d}$ .

447 A second limitation is that the present tests only compared 4 types of plant covers (including  
448 bare soil). A third limitation concerns the space available for the laboratory testing program,  
449 which required an explosion-proof chamber. The only such laboratory available could only hold  
450 1 replicate of each of the 4 columns containing a different plant cover. As a consequence, the  
451 same tests had to be repeated 4 times over time, as explained above. This sequence of testing  
452 allowed a minimal statistical analysis of the results and revealed a significance level of 0.05.

453 Finally, the present study did not examine microbial activity and other bio-chemical processes  
454 involving plants exposed to biogas fluxes, such as the study of the influence of root exudates  
455 (over time or otherwise), evapotranspiration, etc.

456

## 457 **5. Conclusion**

458 This study evaluated whether the use of different types of common plants, including a N-fixing  
459 plant (White clover; WC), a non N-fixing plant (Timothy grass; TG) and a mixture of both  
460 (MIX), would affect CH<sub>4</sub> oxidation within passive biosystems.

461 An important conclusion – and contribution - from this study was the observation that high  
462 oxidation rates were obtained regardless of plant cover. In fact, up to a loading equal to  
463 100 g CH<sub>4</sub>/m<sup>2</sup>/d, the type of plant cover did not influence performance, herein expressed by the  
464 methane oxidation efficiency (MOE). The MOEs in the laboratory and field columns remained  
465 greater than 95%.

466 Until the highest CH<sub>4</sub> loading was applied, the oxidation rate increased following increases in  
467 CH<sub>4</sub> loading. The continuous increase in oxidation rates suggests that the maximum oxidation  
468 capacity of the biosystems tested may have never been fully attained. The oxidation rates  
469 obtained in the laboratory for the high end loadings varied between 191 and 253 g CH<sub>4</sub>/m<sup>2</sup>/d  
470 (MOEs = 71% – 94%). In the field, these oxidation rates were 179 and 105 g CH<sub>4</sub>/m<sup>2</sup>/d (MOEs =  
471 99% and 84%, respectively).

472 For higher loadings (270 and 180 g CH<sub>4</sub>/m<sup>2</sup>/d for laboratory and field tests, respectively), the  
473 plant cover that had the least effect on MOEs and oxidation rates was the MIX column in the  
474 laboratory, whereas the WC column was the least effective in the field. MOEs in the field may  
475 also be affected by climatic conditions.

476 Another noteworthy result of the present study is that unplanted biosystems achieved as high (if  
477 not higher) MOEs as planted biosystems. In other words, for the short-term test results presented,

478 vegetation may not necessarily be a key factor in biosystem performance. Nevertheless, despite  
479 the important database generated, this study has its limitations; accordingly, one cannot  
480 generalize the results obtained to all other types of biosystems and plants. In fact, for every case,  
481 one must take into account several other parameters and – possibly – phenomena, such as  
482 climatic conditions, physical characteristics of the cover soil (texture, compaction etc.) plant  
483 species, influence of root exudates, plant growth stage and plant maturity, etc.

484 The results also indicated a significant effect of vegetation on the values of degree of water  
485 saturation (Sr), most probably due to water uptake by the plant root system. There was no  
486 noticeable effect of vegetation on soil temperature. As expected, temperature was affected by the  
487 biotic oxidation activity occurring within the biosystem. The effects were greater with increasing  
488 CH<sub>4</sub> loadings.

489

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497

498



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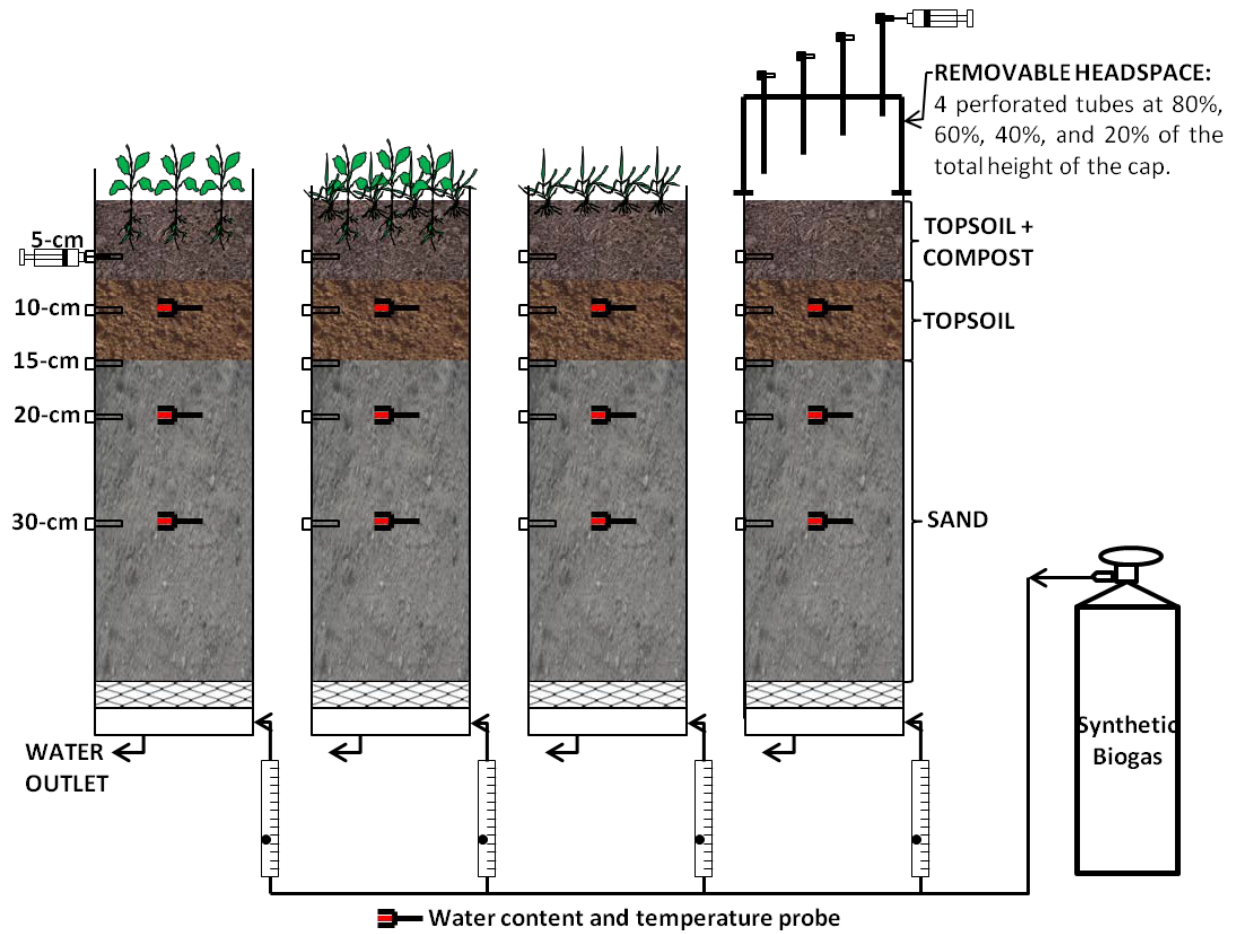


Figure 1: Experimental design of laboratory columns

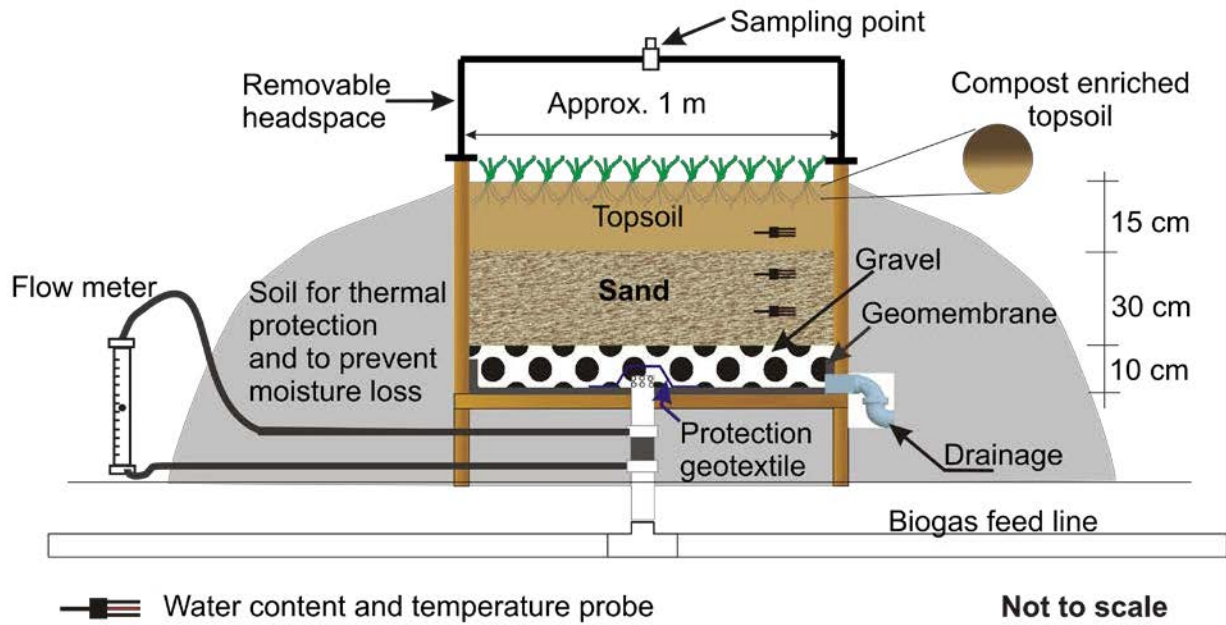
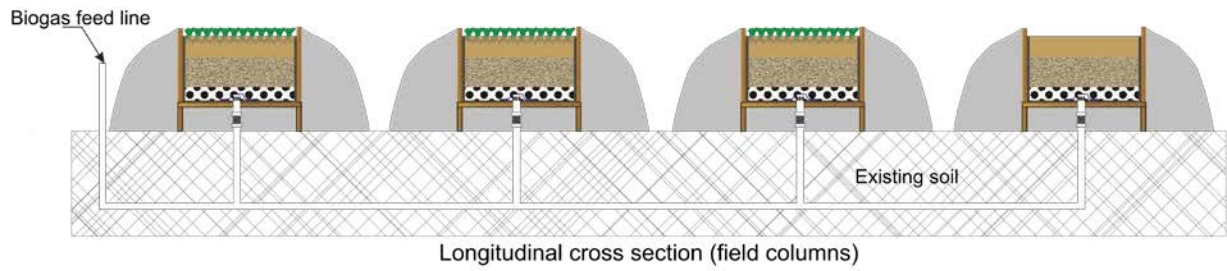


Figure 2: Experimental design of field columns



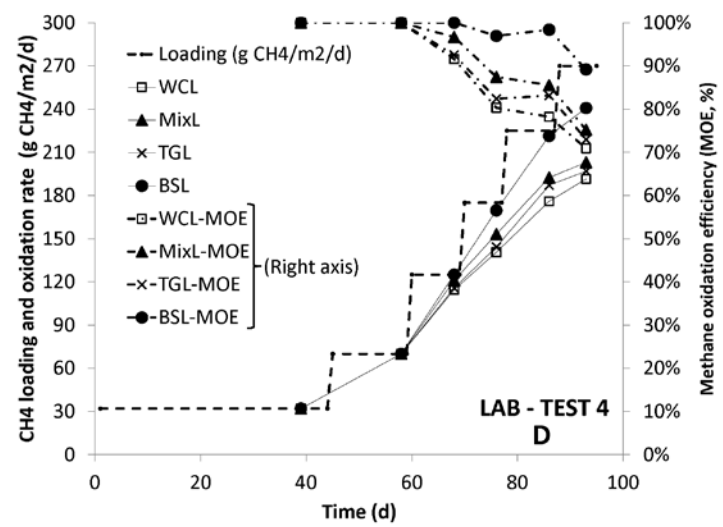
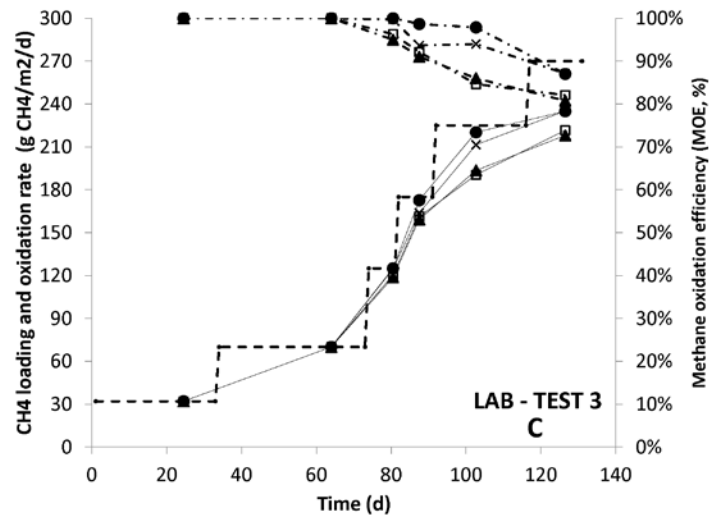
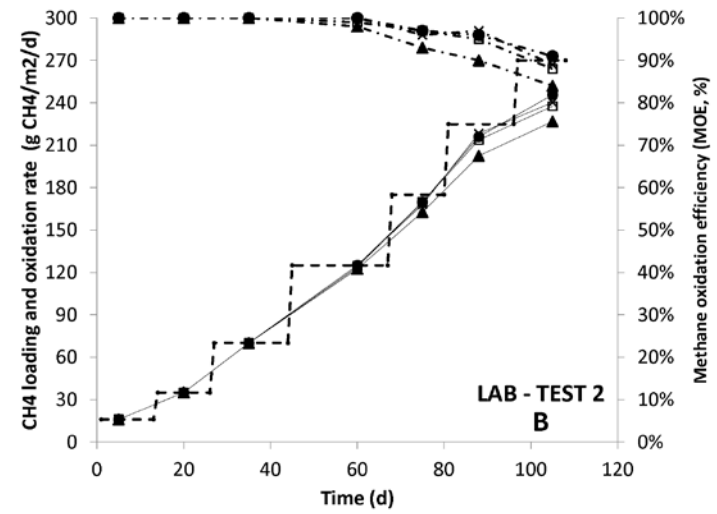
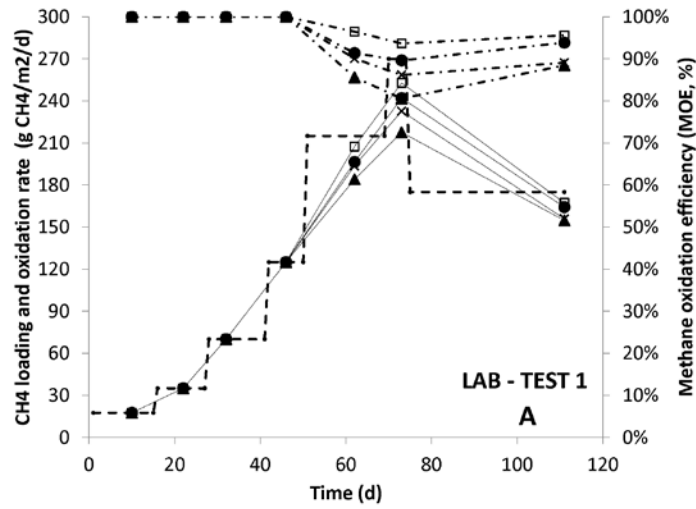


Figure 3: Methane oxidation efficiency and oxidation rates at different CH<sub>4</sub> loadings under laboratory conditions. A - Lab test 1; B - Lab test 2; C - Lab test 3; D - Lab test 4

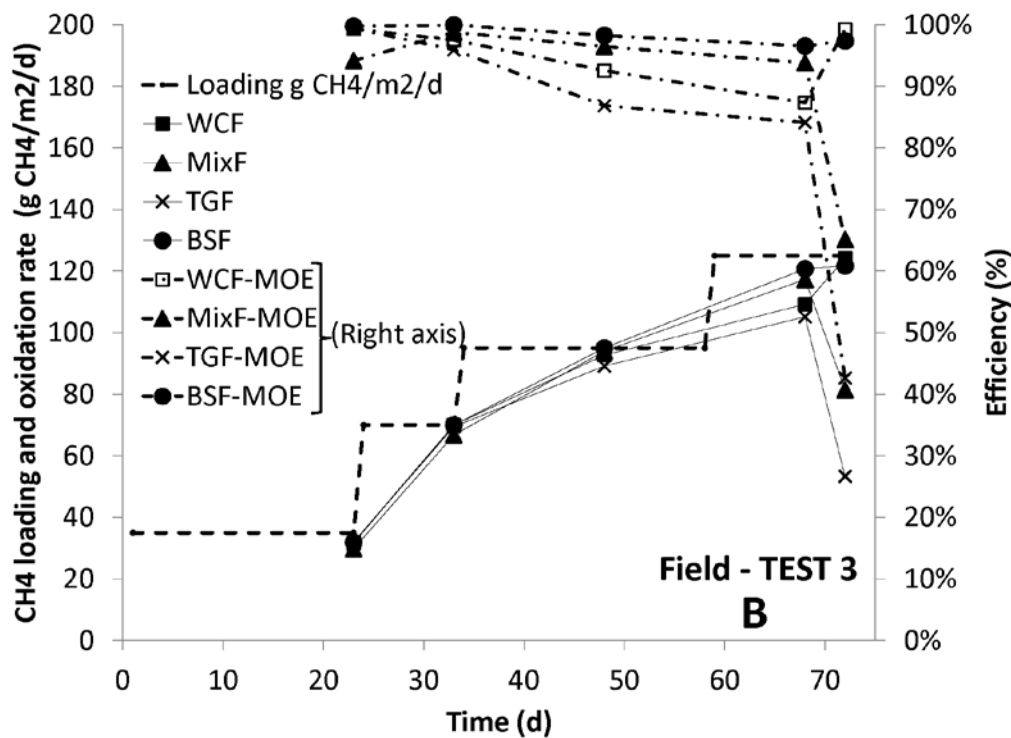
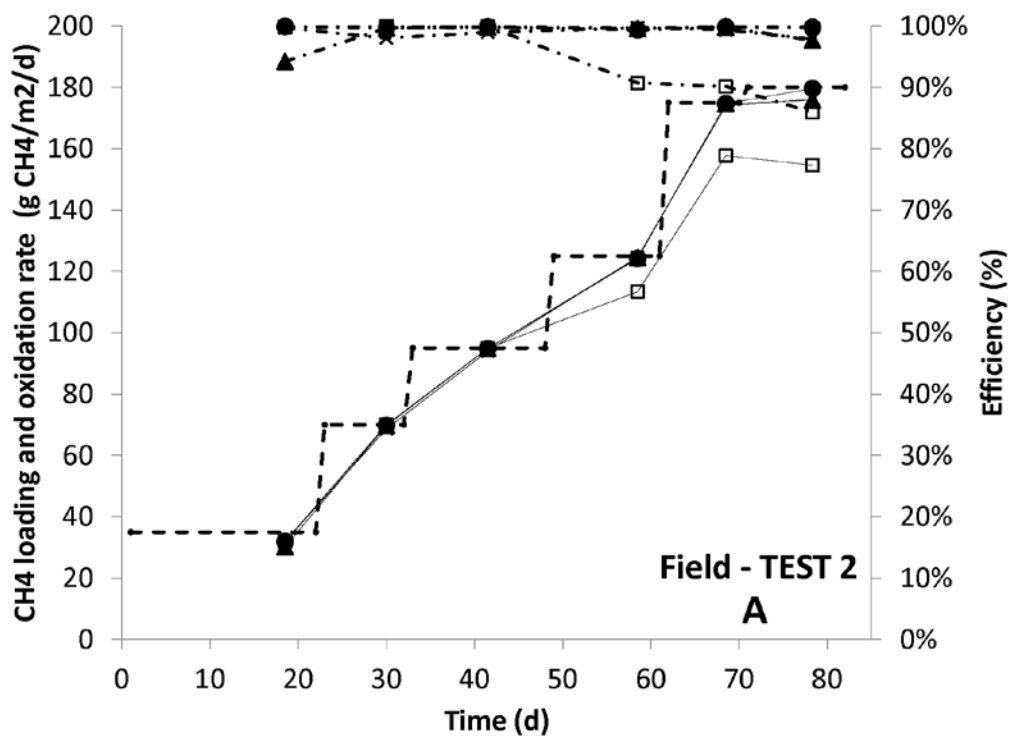


Figure 4: Methane oxidation efficiency and oxidation rates at different CH<sub>4</sub> loadings under field conditions. A - Field test 1; B - Field test 2

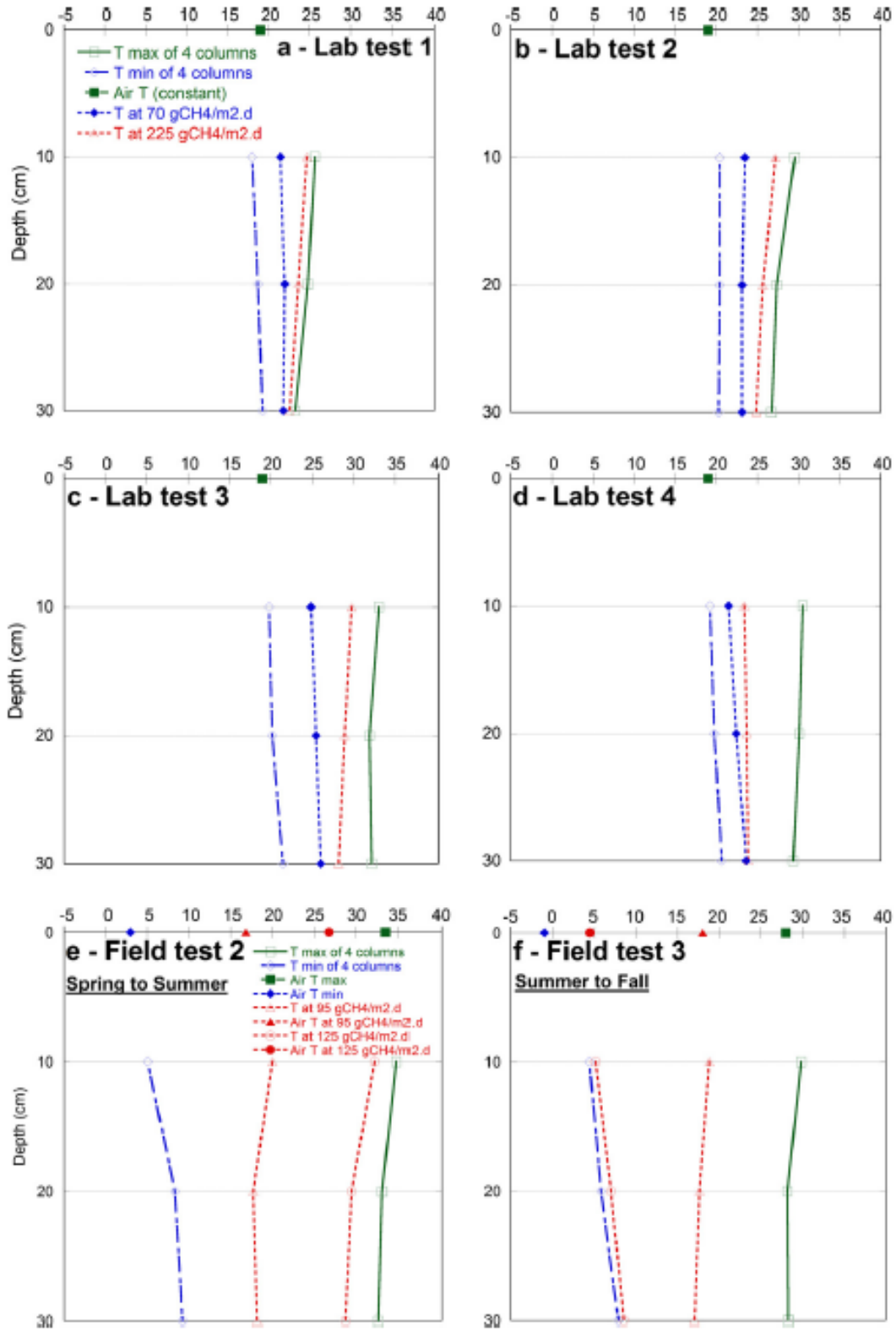


Fig. 5. Temperature profiles within the columns and air temperature in field and lab tests.

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