Short Asymmetric Synthesis of Phenantroindolizidines through Chiral Homoallylic sulfinamines.[†]

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[†]Electronic Supplementary Information (ESI) available: Copies of ¹H and ¹³C NMR spectra for compounds 2-8, 10-13, diastereomeric mixtures of 5 and 12, and (–)-tylophorine are provided. HPLC traces of compounds 5 and 12, with the corresponding diastereomeric mixtures, and chiral analysis for compounds 8 and *ent*-8 are included.

ABSTRACT: An efficient stereocontrolled preparation of chiral phenantroindolizidines is detailed. The synthesis relies on the stereoselective indium-mediated allylation of 2- (phenanthren-9-yl)acetaldehyde derivatives with chiral *tert*-butylsulfinamide. Chemoselective transformations from the corresponding homoallylic sulfinamine allow the synthesis of the phenantroindolizidines in only three synthetic operations, without any detectable racemization. Following this procedure, the synthesis of natural (–)-tylophorine was successfully accomplished.

Introduction

The use of plants of the Asclepiadaceae family, including Tylophora indica, Cryptocarya and Ficus species, in the traditional medicine led to the discovery of phenantroindolizidine alkaloids.¹ It has been shown that compounds of this family, from which (–)-Antofine, (–)-Tylophorine and (–)-Tylocrebine are representative members (Figure 1), display a wide range of biological activities, such as antibacterial,² antiasthmatic,³ antiviral⁴ and anti-inflamamatory⁵ activities. Moreover, their extremely high cytotoxicity (IC50 ~10 nM) against different cancer cell lines by a unique mode of action has especially attracted the attention of medicinal chemists.⁶

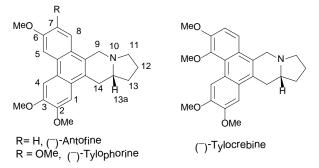
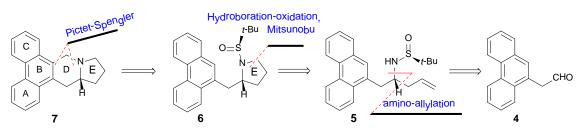


Figure 1. Naturally occurring phenantroindolizidines.

Due to the CNS toxicity of natural phenantroindolizidines,⁷ the search of potent nontoxic analogues with improved pharmacokinetic properties is a very active research area. Interestingly, the unnatural (*S*)-tylophorine was found to be even more potent in the inhibition of some specific cancer cell growth.⁸ In this context, the development of robust methods that allows the rapid construction of the required enantioenriched phenantroindolizidine scaffold is a need that has been addressed by a number of research groups. Some successful approaches reported in the last decade include the use of catalytic methods,⁹ chiral building blocks¹⁰ and chiral auxiliaries.¹¹ Importantly, access to both enantiomers, either *R*- or *S*-, phenantroindolizidine alkaloids is highly desirable to explore their biological profiles and we anticipated that the corresponding chiral homoallylc sulfinamines should provide a very convenient entry.¹² Herein, we report the realization of this strategy.¹³

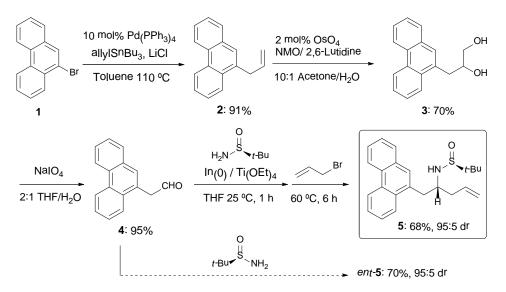
Results and discussion

In our retrosynthetic analysis of phenantroindolizidines we propose the formation of ring D in a late stage by a reliable Pictet-Spengler annulation.^{9a-9c} Importantly, the chiral *tert*-butyl sulfinyl auxiliary¹⁴ could be removed under the required acidic conditions for the cyclization, minimizing functional groups manipulation (Scheme 1). The formation of the pyrrolidine intermediate **7** was anticipated by straightforward synthetic elaboration of homoallyl sulfinamine **5**, which could be prepared from aldehyde **4** using a one-pot protocol developed in our group.¹⁵



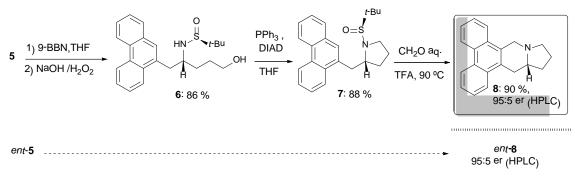
Scheme 1. Retrosynthetic Analysis of Phenantroindolizidine.

As outlined in Scheme 2, our synthesis began with the preparation of aldehyde 4. Stille coupling of commercially available 9-bromophenanthrene with allyltri-n-butyltin afforded compound 2 in excellent yield. After having explored different conditions for the oxidative cleavage of the olefinic double bond, including ozonolysis and Johnson-Lemieux oxidation, aldehyde 4 was efficiently prepared in a two-step procedure. The very low solubility of the intermediate diol 3 allowed its purification by recrystallization from ethyl acetate, before being oxidatively cleaved by NaIO₄. This method afforded aldehyde 4 which was pure enough to be used directly in the next step. The indiummediated aminoallylation took place with good yield and high diastereoselectivity,¹⁶ either with (S_S)-tert-butylsulfinamide to obtain compound 5 or with (R_S)-tert-butylsulfinamide to prepare *ent*-5.¹⁷



Scheme 2. Synthesis of Homoallylsulfinamines 5 and ent-5.

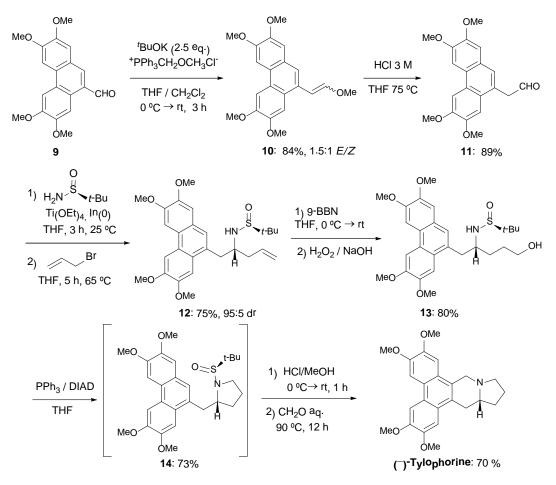
Crucial to our synthetic plan was the use chemoselective reactions to prepare pyrrolidine 7, while keeping the *tert*-butylsulfinyl as protecting group. We thus submitted intermediate 5 to a hydroboration-oxidation sequence, which took place without any detectable oxidation of the sulfur atom, and the obtained sulfinaminealcohol 6 was cyclized under Mitsunobu conditions to furnish the desired compound 7 in very good overall yield.¹⁸ Finally, a solution of this compound in TFA was treated with aqueous formalin at 90 °C for 12 h, and deprotection of the amino group was followed by the corresponding iminium formation and Pictet-Spengler annulations, affording compound **8** in excellent isolated yield. Physical and spectral data of compound **8** were in good agreement with those previously reported for a racemic sample.¹⁹ Following the same route (Scheme 3), but from *ent*-**5** we also prepared compound *ent*-**8**. Chiral HPLC analysis of both final products clearly evidenced that racemization did not occur in the reaction sequence used from each chiral homoallylic sulfinamine (**5** or *ent*-**5**), obtaining identical enantiomeric ratios in the final products (95:5 er). Despite the high occurrence of the phenantroindolizidine scaffold in prominent bioactive natural products, to our best knowledge this is the first report on the synthesis of the enantioenriched parent compounds.



Scheme 3. Synthesis of Phenantroindolizidines 7 and ent-7.

Having developed a convenient procedure for the synthesis of phenantroindolizidines, the stage was set to target naturally occurring family members and we illustrate the applicability of this protocol in the synthesis of (–)-Tylophorine (Scheme 4). The synthesis commenced by a two-step homologation of the readily available phenanthryl aldehyde 9^{20} A Wittig olefination with methoxymethyl phosphorous ylide allowed the preparation of a diastereomeric mixture of enol ethers $10 \ (E/Z \ 3:2, according to \ ^1\text{H-NMR})$, which were hydrolyzed to furnish the required aldehyde 11 in very good overall yield. The indium-mediated amino allylation of this aldehyde took place with 75% yield to afford homoallylic amine 12 with good diastereoselectivity [95:5 dr, determined by ¹H-NMR and HPLC, see the ESI].¹⁶ The hydroboration-oxidation sequence occurred with similar good efficiency to the one observed in the preparation of compound 6 and the obtained compound 13 was submitted to Mitsunobu cyclization conditions, which

also took place with 73% isolated yield. Although the pyrrolidine **14** could not be efficiently separated from the triphenylphosphine oxide byproduct, it was used in the next step. Deprotection of the amino group with HCl was followed by *in-situ* addition of aqueous formalin and after heating to 90 °C for 12 h, the desired (*R*)-(–)-tylophorine was obtained, which was isolated in good yield after chromatography purification. The spectroscopic data and the specific optical rotation value {[α]_D²⁰ -85.0 (c 0.5, CHCl₃)} obtained for our synthetic tylophorine are in accordance with those reported for the natural compound.^{9b, 10b, 11}



Scheme 4. Synthesis of (-)-Tylophorine.

Conclusions

We developed build enantioenriched have practical procedure to а phenantroindolizidines which the indium-mediated aminoallylation in of 2-(phenanthren-9-yl)acetaldehyde derivatives with chiral tert-butylsulfinamide is a key step. Importantly, in this synthetic approach the *tert*-butylsulfinyl group acted not only as a chiral director group, but also as a protecting group in the whole sequence, allowing the formation of target phenantroindolizidines in only three synthetic operations from the corresponding homoallylic amine. From each enantiomer of *tert*-butylsulfinamide, the corresponding enantioenriched parent phenantroindolizidine was prepared, the enantiomeric ratio of the final products being identical to the diastereomeric ratio of the corresponding chiral homoallylic sulfinamines. The usefulness of this protocol was illustrated by the synthesis of natural (–)-tylophorine and we hope it will find applications in the optimization of the biological profile of this family of bioactive compounds.

Experimental

General Remarks. TLC were performed on silica gel 60 F₂₅₄, using aluminum plates and visualized by exposure to ultraviolet light or with phosphomolybdic acid (PMA) stain. Flash chromatography was carried out on handpacked columns of silica gel 60 (230 – 400 mesh). Melting points are uncorrected. Optical rotations were measured using a polarimeter with a thermally jacketted 5 cm cell at approximately 20 °C and concentrations (c) are given in g/100 mL. Infrared analysis was performed with a spectrophotometer equipped with an ATR component; wavenumbers are given in cm⁻¹. Mass spectra (EI) were obtained at 70 eV; and fragment ions in m/z with relative intensities (%) in parentheses. HRMS analyses were carried out using the Electron Impact (EI) mode at 70 eV or by Q-TOF using Electro Spray Ionization (ESI) mode. GC analyses were obtained with an HP-5 column (30 m \times 0.25 mm, ID \times 0.25 μ m) and an EI (70 EV) detector. The temperature program: hold at 60 °C for 3 min, ramp from 60 °C to 270 °C at 15 °C/min, hold at 270 °C for 10 min. ¹H NMR spectra were recorded at 300 or 400 MHz for ¹H NMR and 75 or 100 MHz for ¹³C NMR, using CDCl₃ or DMSO- d_6 as the solvent and TMS as an internal Standard (0.00 ppm). The data is being reported as (s = singlet, d = doublet, t = triplet, m = multiplet or unresolved, br s = broadsignal, coupling constant(s) in Hz, integration). ¹³C NMR spectra were recorded with ¹H-decoupling at 100 MHz and referenced to CDCl₃ at 77.16 ppm or to DMSO- d_6 at 39.52 ppm. DEPT-135 experiments were performed to assign CH, CH₂ and CH₃.

9-Allylphenanthrene (2). To a solution of 9-bromophenanthrene (918 mg, 3.5 mmol) in dry toluene (50 mL), under argon atmosphere, was added allyltributylstannane (2.17 mL, 7 mmol), palladium tetrakis (464 mg, 0.35 mmol) and lithium chloride (178 mg, 4.2 mmol). The resulting mixture was heated to 110 °C for 2 h, when full conversion was verified by TLC. After cooling to room temperature, the mixture was diluted with EtOAc (50 mL), washed with brine (20 mL), dried over MgSO₄ and concentrated to dryness. The obtained residue was purified by flash chromatography (hexane) to obtain the desired product as a white amorphous solid (693 mg, 91%): R_f 0.37 (100% Hexane); GC t_R = 15.82 min. ¹H NMR, ¹³C NMR, IR and MS were in agreement with previously reported data.²¹

3-(Phenanthren-9-yl)propane-1,2-diol (3). To a solution of compound 2 (540 mg, 2.48 mmol) in acetone/water (35 mL/ 3.5 mL) were sequentially added 2,6-lutidine (0.90 mL, 7.80 mmol), NMO (704 mg, 5.8 mmol) and a solution of OsO₄ in t-BuOH (2.5 % wt in *t*-BuOH, 0.90 mL). The mixture was stirred under argon atmosphere for 3.5 h before being quenched with water (10 mL) and observing full consumption of the starting material by TLC. The mixture was extracted with EtOAc (3 x 15 mL) and the collected organic layers were washed with brine, dried over MgSO₄, filtered and concentrated under reduced pressure. The solid obtained was recrystallized from EtOAc (~14 mL) to provide the pure diol 3 as a grey solid (300 mg). The filtrate was concentrated under reduced pressure to dryness and purified by column chromatography (1:1 hexane/EtOAc to 100% EtOAc) to afford another 133 mg of product (70% total yield): R_f 0.12 (1:1 hexane/EtOAc); IR v 3377, 3290, 2884, 1385, 1305, 1112, 1033, 842, 722 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 8.90 – 8.83 (m, 1H), 8.81 – 8.75 (m, 1H), 8.24 - 8.17 (m, 1H), 7.94 - 7.88 (m, 1H), 7.74 - 7.58 (m, 5H), 4.76 - 4.69 (m, 2H), 3.92 - 3.82 (m, 1H), 3.53 - 3.37 (m, 3H), 2.95 (dd, J = 14.0, 7.9 Hz, 1H); ¹³C NMR (101 MHz, DMSO) δ 133.9 (C), 131.3 (C), 131.1 (C), 130.1 (C), 129.1 (C), 127.9 (CH), 127.6 (CH), 126.8 (CH), 126.6 (CH), 126.3 (CH), 126.1 (CH), 124.7 (CH), 123.3 (CH), 122.6 (CH), 71.5 (CH), 65.8 (CH₂), 37.6 (CH₂) ppm; LRMS (EI) *m/z* (%) 252(M+, 36), 207 (19), 191 (100), 178 (14), 165 (12), 151 (4); HRMS (EI) calcd for C₁₇H₁₆O₂ 252.1150, found 252.1156.

2-(Phenanthren-9-yl)acetaldehyde (4). To a suspension of diol **3** (350 mg, 1.39 mmol) in THF/H₂O (14.25 mL/6.75 mL) was added NaIO₄ (356 mg, 1.68 mmol) and the reaction mixture was stirred under argon atmosphere at 23 °C for 2.5 h (full consumption of starting material by TLC). The THF was removed under reduced

pressure and the aqueous phase was extracted with EtOAc (3 x15 mL). The collected organic layers were washed with brine, dried over MgSO₄ and concentrated to dryness to furnish pure aldehyde **4** as a brownish solid in almost excellent yield (290 mg, 95%): R_f 0.38 (9:1 Hexane/EtOAc); GC t_R = 16.66 min; IR v 3075, 2922, 2851, 2820, 1710, 739, 719 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.84 (t, J = 2.4 Hz, 1H), 8.79 – 8.74 (m, 1H), 8.69 (dd, J = 8.2, 0.5 Hz, 1H), 7.92 (dd, J = 8.1, 1.3 Hz, 1H), 7.87 (dd, J = 7.8, 1.5 Hz, 1H), 7.74 – 7.59 (m, 5H), 4.14 (dd, J = 2.4, 0.6 Hz, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 200.0 (CHO), 131.6 (C), 131.2 (C), 130.9 (C), 130.4 (C), 129.5 (C), 128.5 (C), 127.3 (CH), 127.1 (2CH + C), 126.9 (CH), 124.4 (CH), 123.5 (CH), 122.7(CH), 48.9 (CH₂); LRMS (EI) m/z (%) 220 (M⁺, 44), 191 (100), 165 (22), 94 (4), 82 (3); HRMS (EI) calcd for C₁₆H₁₂O 220.0888, found 220.0881.

(4R, S_S)-(N-tert-Butylsulfinyl)-5-[9-phenantryl]-pent-1-en-4-amine (5). To a dry flask were added (S_S)-N-tert-butylsulfinamide (161 mg, 1.33 mmol) followed by indium powder (191 mg, 1.66 mmol) under Ar. Then a solution of aldehyde 4 (306 mg, 1.39 mmol) in dry THF (2.7 ml) and Ti(OEt)₄ (0.6 mL, 2.66 mmol,) were added successively and the reaction mixture was stirred under argon for 1 h at 23 °C. At this time allyl bromide (170 µL, 2.00 mmol) was added to the mixture and the reaction was allowed to reach 60 °C and stirred at that temperature for 6 h. The mixture was allowed to reach room temperature and was carefully added over a stirring mixture of 4:1 EtOAc/brine (10 mL). The resulted white suspension was filtered through a short pad of Celite, washed with EtOAc and concentrated under reduced pressure. The resulted suspension was diluted in 4:1 EtOAc/hexane (50 mL) and filtered again through Celite. Organics were concentrated to dryness and the residue was purified by flash chromatography (hexane/EtOAc 7:3 to 6:4) to obtain the desired product as a white amorphous solid (328 mg, 68%, 95:5 according to ¹H-NMR and HPLC, see ESI): $[\alpha]_{D}^{20}$ 4.8 (c 0.68, MeOH); R_f 0.16 (7:3 hexane/EtOAc); HPLC analysis [Tracer Excel 120 column 15 cm x 0.46 cm, isocratic elution with 99:1 n-hexane/i-PrOH, 1.0 mL/min, UV detection at 254 nm, t_R (minor diastereoisomer) 45.05 min/ t_R (major diastereoisomer) 49.23 min]; IR v 3216, 3075, 2958, 2925, 1636, 1449, 1260, 1048,747, 726 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.79 – 8.71 (m, 1H), 8.71 – 8.63 (m, 1H), 8.15 – 8.06 (m, 1H), 7.88 – 7.79 (m, 1H), 7.74 - 7.54 (m, 5H), 5.96 - 5.77 (m, 1H), 5.23 (d, J = 12.9, 2H), 3.90 - 1003.76 (m, 1H), 3.57 - 3.40 (m, 2H), 3.20 (dd, J = 14.1, 7.3 Hz, 1H), 2.59 - 2.34 (m, 2H),1.10 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 134.3 (CH), 132.7131.6 (C), 131.2 (C),

130.9 (*C*), 130.0 (*C*), 128.9 (*C*H), 128.2 (*C*H), 127.0 (*C*H), 126.9 (*C*H), 126.5 (*C*H), 126.4 (*C*H), 124.5 (*C*H), 123.4 (*C*H), 122.6 (*C*H), 119.6 (*C*H), 55.9 (*C*), 54.7(*C*H), 40.3 (*C*H₂), 39.7 (*C*H₂), 22.6 (*C*H₃); LRMS (EI) m/z (%) 365 (M⁺, 0.2), 309 (55), 191 (100), 165 (20), 118 (50), 70 (47), 57 (26), 41 (9); HRMS (ESI) calcd for C₂₃H₂₈NOS 366.1892, found 366.1888.

(4S, $R_{\rm S}$)-(*N-tert*-Butylsulfinyl)-5-[9-phenantryl]-pent-1-en-4-amine (*ent*-5). It was prepared from aldehyde 4 (330 mg, 1.50 mmol) and ($R_{\rm S}$)-*N-tert*-butylsulfinamide (173 mg, 1.43 mmol) following the same procedure described above for compound 5. The expected compound was obtained as a white amorphous solid (365 mg, 70%) with identical spectroscopy and physical data than compound 5 except for $[\alpha]_{\rm D}^{20}$ –6.7 (*c* 1.00, MeOH).

(4R, S_S)-(*N-tert*-Butylsulfinyl)-5-[9-phenantryl]-pent-1-ol-4-amine (6). To a solution of compound 5 (803 mg, 2.20 mmol) in dry THF (3.3 mL), at 0 °C under argon atmosphere, was added a solution of 9-BBN in THF (13.2 mL, 0.5 M). The cooling bath was removed and the reaction mixture was stirred for 15 h at 25 °C. The reaction mixture was cooled down to 0 °C and a solution of 6M NaOH (5.9 mL) was carefully added, followed by H₂O₂ (5.1 mL, 33%). The reaction mixture was left stirring for 8 h, while the temperature increased to 25 °C. The mixture was extracted with EtOAc (3 x 20 mL), the collected organic layers were washed with water (10 x 5 mL), brine (1 x 5 mL), dried over MgSO₄ and concentrated to dryness. The crude product was purified by flash chromatography (EtOAc 100% to EtOAc/MeOH 98:2) to obtain the desired product as a white amorphous solid (725 mg, 86%): $\left[\alpha\right]_{D}^{20}$ -11.4 (c 0.7, MeOH); R_f 0.30 (98:2 EtOAc/MeOH); IR v 3233, 3075, 2925, 2866, 1449, 1038, 747, 726 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.78 – 8.71 (m, 1H), 8.70 – 8.62 (m, 1H), 8.11 – 8.03 (m, 1H), 7.86 - 7.78 (m, 1H), 7.71 - 7.54 (m, 5H), 3.84 - 3.56 (m, 4H), 3.35 (dd, J = 14.0, 7.1 Hz, 1H), 3.26 (dd, J = 14.1, 6.4 Hz, 1H), 2.02 - 1.86 (m, 1H), 1.86 - 1.63 (m, 4H), 1.04 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 132.9 (C), 131.6(C), 131.3(C), 130.9 (C), 129.9 (C), 128.8 (CH), 128.2 (CH), 126.9 (CH), 126.8 (CH), 126.5 (CH), 126.4 (CH), 124.4 (CH), 123.5 (CH), 122.6 (CH), 62.8 (CH₂), 57.5 (CH), 56.0 (C), 41.1 (CH₂), 32.8 (CH₂), 28.7 (CH₂), 22.7 (CH₃); LRMS (EI) m/z (%) 327 (M⁺-C₄H₈, 44), 245 (11), 191 (100), 165 (22), 118 (70), 88 (60), 71 (51), 57 (27), 41 (9); HRMS (ESI) calcd for C₂₃H₃₀NO₂S 384.1997, found 384.1980.

 $(4S,R_S)$ -(N-tert-Butylsulfinyl)-5-[9-phenantryl]-pent-1-ol-4-amine (ent-6). It was prepared following the same procedure described above for compound 6, showing identical spectroscopy and physical data except for $[\alpha]_D^{20}$ +9.4 (*c* 0.9, MeOH).

(2R,S_s)- (N-tert-Butylsulfinyl)-2-(phenanthren-9-ylmethyl)pyrrolidine (7). To a solution of compound 6 (617 mg, 1.61 mmol) in dry THF (16 mL) under argon atmosphere was added PPh₃ (512 mg, 1.95 mmol), followed by DIAD (400 µL, 2.03 mmol) at 25 °C. The reaction mixture turned yellow and was stirred for 16 h. The mixture was concentrated to dryness under reduced pressure and the residue was purified by flash chromatography (7:3 hexane/EtOAc). The desired product was obtained as a white amorphous solid (518 mg, 88%): $[\alpha]_D^{20}$ +29.6 (c 0.95, MeOH); R_f 0.24 (7:3 hexane/EtOAc); IR v 3733, 3632, 2963, 1450, 1360, 1065, 748 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 8.77 - 8.71 \text{ (m, 1H)}, 8.70 - 8.63 \text{ (m, 1H)}, 8.19 - 8.11 \text{ (m, 1H)},$ 7.86 - 7.79 (m, 1H), 7.73 - 7.65 (m, 2H), 7.63 - 7.55 (m, 3H), 4.18 - 4.06 (m, 1H), 3.87 -3.76 (m, 1H) 3.68 (dd, J = 13.5, 4.5 Hz, 1H), 2.97 - 2.84 (m, 2H), 1.98 - 1.83 (m, 1H), 1.78 - 1.57 (m, 3H), 1.26 (s, J = 4.2 Hz, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 133.3 (C), 131.8 (C), 131.1 (C), 130.8 (C), 130.0 (C), 128.2 (CH), 127.8 (CH), 127.0 (CH), 126.8 (CH), 126.5 (CH), 126.4 (CH), 124.7 (CH), 123.4 (CH), 122.6 (CH), 65.5 (CH), 57.4 (C), 41.8 (CH₂), 41.5 (CH₂), 31.8 (CH₂), 26.0 (CH₂), 24.1 (CH₃); LRMS (EI) m/z (%) 365 (M+, 0.1), 309 (M⁺-C₄H₈, 25), 191 (46), 118 (100), 70 (14), 57 (6), 41 (4); HRMS (EI) calcd for C₁₉H₁₉NOS 309.1187, found 309.1177.

(2*S*,*R*_S)- (*N*-*tert*-Butylsulfinyl)-2-(phenanthren-9-ylmethyl)pyrrolidine (*ent*-7). It was prepared following the same procedure described above for compound 7, showing identical spectroscopy and physical data except for $[\alpha]_D^{20}$ –27.7 (*c* 0.95, MeOH).

(*R*)-9,11,12,13,13a,14-Hexahydrodibenzo[*f*,*h*]pyrrolo[1,2-*b*]isoquinoline (8). To a solution of compound 7 (244 mg, 0.67 mmol) in TFA (6 mL) under argon atmosphere, was added 37% aqueous CH₂O (2.50 mL, 1.18 mL) and the reaction mixture was heated to 90 °C for 12 h. After cooling to room temperature, H₂O (15 mL) and 2 M NaOH (15 mL) were sequentially added to the mixture (pH > 12) and the product was extracted with EtOAc (3 x 40 mL). The collected organic layers were washed with brine, dried over MgSO₄ and concentrated under reduced pressure. The orange solid obtained was purified by flash chromatography (CH₂Cl₂/MeOH 98:2 to 90:10) to obtain a pale yellow solid (164 mg, 90%, 95:5 er according to chiral HPLC): $[\alpha]_D^{20}$ –146 (*c* 1.0, CHCl₃); mp 159-160 °C (MeOH) {lit.¹⁹ mp 153-154 °C for racemic mixture}; R_f 0.20 (95:5 CH₂Cl₂/MeOH); HPLC analysis [AD-H column 25 cm x 0.46

cm, isocratic elution with 75:25:0.1 *n*-hexane/*i*-PrOH/Et₃N, 1.0 mL/min, UV detection at 254 nm], t_R (minor) 9.54 min, t_R (major) 19.77 min. ¹H NMR, ¹³C NMR, IR and MS were in agreement with previously reported data.¹⁹

(S)-9,11,12,13,13a,14-Hexahydrodibenzo[f,h]pyrrolo[1,2-b]isoquinoline (ent-8).

It was prepared following the same procedure described above for compound **8**, showing identical spectroscopy and physical data except for $[\alpha]_D^{20}$ +138 (*c* 0.5, CHCl₃). The analysis by chiral HPLC also showed a 95:5 er (See ESI for HPLC traces).

2,3,6,7-Tetramethoxy-9-(2-methoxyvinyl)phenanthrene (10). To a suspension of t-BuOK (469 mg, 4.2 mmol) in dry THF (25 mL) was added phosphonium salt (1.420 g, 4.1 mmol) at 0 °C and stirred during 30 min under argon atmosphere. Then the suspension was stirred for 20 min at room temperature, cooled again to 0 °C and a solution of compound 9 (529 mg, 1.62 mmol) in CH₂Cl₂ (20 mL) was added. The reaction mixture was left stirring for 3 h while the temperature increased to 25 °C. At this point, the reaction was quenched by adding a saturated solution of NH₄Cl (25 mL). The mixture was extracted with EtOAc (3 x 25 mL) and the collected organic layers were washed with brine, dried over MgSO₄, filtered and concentrated under reduced pressure. The crude was purified by recrystallization from 1-propanol to give compound **10** (481 mg, 84%) as a yellow solid (diastereomeric mixture, E/Z ratio $\approx 1.5/1$): R_f 0.35 (6:4 hexane/EtOAc); IR v 2999, 2951, 2926, 2852, 1706, 1509, 1252, 1148, 773 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.03 (s, 0.43H, (Z)), 7.82 (s, 1H, (E)), 7.76 (s, 1H, (Z)), 7.48 (d, J = 12.9 Hz, 1H, (Z)), 7.43 (s, 0.56 H, (E)), 7.19 (d, J = 10.7 Hz, 1H, (E/Z)), 6.97 (d, J = 12.4 Hz, 0.57H, (E)), 6.41 (d, J = 7.1 Hz, 0.47H, (Z)), 6.38 (d, J = 7.9 Hz, C)0.54H, (E)), 5.78 (d, J = 7.1 Hz, 0.42H, (Z)), 4.12 (2 x s, 3H), 4.11 (2 x s, 3H), 4.05 (2 x s, 3H), 4.03 (bs, 3H), 3.82 (2 x s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 150.0 (CH), 149.14 (C), 149.10 (C), 149.05 (C), 149.01 (C), 148.9 (C), 148.84 (C), 148.7 (C), 148.6 (CH), 148.6 (C), 129.7 (C), 127.5 (C), 126.7 (C), 126.6 (C), 125.6 (C), 125.2 (CH), 125.1 (C), 124.8 (C), 123.9 (C), 123.7 (C), 121.8 (CH), 108.5 (CH), 108.2 (CH), 105.4 (CH), 105.3 (CH), 103.4 (CH), 103.3 (CH), 103.0 (CH), 102.9 (CH), 102.4 (CH), 60.8 (CH), 57.3 (CH), 56.3 (CH₃), 56.2 (CH₃), 56.17 (CH₃), 56.0 (CH₃), 55.96 (CH₃); LRMS (EI) m/z (%) 354 (M+, 100), 339 (15), 324 (10), 308 (17), 177 (7); HRMS (EI) calcd for C₂₁H₂₂O₅ 354.1467, found 354.1466.

2-(2,3,6,7-Tetramethoxyphenanthren-9-yl)acetaldehyde (11). To a suspension of compound 10 (531 mg, 1.5 mmol) in dry THF (1.85 mL) was added a solution of 3M HCl (0.65 mL). The mixture was stirred and heated to 75 °C under argon atmosphere for

2.5 h, before being quenched with aqueous solution of NaHCO₃ (6 mL, 10%). The mixture was extracted with CH_2Cl_2 (3 x 10 mL), and the collected organic layers were washed with brine (5 mL), dried over MgSO₄ and concentrated *in vacuo*. The residue was purified by flash chromatography (hexane/EtOAc 65:35 to 55:45) to obtain a yellow amorphous solid (454 mg, 89%): R_f 0.32 (1:1 hexane/EtOAc); ¹H NMR, ¹³C NMR and MS were in agreement with previously reported data.¹³

S_S)-(*N-tert*-Butylsulfinyl)-5-[2,3,6,7-tetramethoxyphenanthren-9-yl]-pent-1-(4R,en-4-amine (12). To a mixture of aldehyde 11 (622 mg, 1.83 mmol), (S_S) -N-tertbutylsulfinamide (210 mg, 1.74 mmol), indium powder (252 mg, 2.20 mmol) in dry THF (3.5 mL), was added Ti(OEt)₄ (0.8 mL, 3.48 mmol,) at 25 °C. After 3 h, allyl bromide (226 µL, 1.4 mmol) was added to the mixture and the reaction was allowed to reach 65 °C and stirred at that temperature for 6 h. The mixture was allowed to reach room temperature and was carefully added over a stirring mixture of 4:1 EtOAc/brine (25 mL). The resulted white suspension was filtered through a short pad of Celite, washed with EtOAc (5 x 20 mL) and concentrated under reduced pressure. The crude was purified by flash chromatography (hexane/EtOAc 3:7) to obtain the desired product as a pale yellow solid (672 mg, 75%, 95:5 according to ¹H-NMR and HPLC): $[\alpha]_D^{20}$ +34.8 (c 1.2, CHCl₃); { $lit^{13} [\alpha]_{D}^{25}$ +42.6 (c 1.0, CHCl₃)}; R_f 0.22 (3:7 hexane/EtOAc); HPLC analysis [Tracer Excel 120 column 15 cm x 0.46 cm, isocratic elution with 95:5 n-hexane/i-PrOH, 1.0 mL/min, UV detection at 254 nm], t_R (minor) 14.71 min / t_R (major) 17.16 min]; ¹H NMR, ¹³C NMR, IR and MS were in agreement with previously reported data.¹³

(4*R*,*S*_{*S*})-(*N*-*tert*-Butylsulfinyl)-5-[2,3,6,7-tetramethoxyphenanthren-9-yl]-pent-1-ol-4-amine (13). To a solution of compound 12 (485 mg, 1 mmol) in dry THF (1 mL) was added a solution of 9-BBN in THF (6 mL, 0.5M) at 0 °C, under argon. After being stirred for 15 h at room temperature, a solution of 6M NaOH (2.7 mL) followed by H_2O_2 (3.3 mL, 33%) were added to the mixture with ice cooling. The reaction mixture was left stirring for 8 h, while the temperature increased to 25 °C. The mixture was extracted with EtOAc (3 x 20 mL), the collected organic layers were washed with water (10 x 5 mL), brine (1 x 5 mL), dried over MgSO₄ and concentrated to dryness. The crude product was purified by flash chromatography (EtOAc/MeOH 98:2 to 95:5) to obtain the desired product as a white amorphous solid (403 mg, 80%): $[\alpha]_D^{20}$ -5.7 (c 1.03, MeOH); R_f 0.15 (98:2 EtOAc/MeOH); IR v 3298, 3179, 3003, 2930, 2830, 1614, 1509, 1470, 1252, 1146, 1029, 836, 774 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.82 (s, 1H), 7.76 (s, 1H), 7.40 (d, J = 11.3 Hz, 2H), 7.15 (s, 1H), 4.12 (s, 3H), 4.11 (s, 3H), 4.08 (s, 3H), 4.02 (s, 3H), 3.84 (d, J = 5.5 Hz, 1H), 3.81 – 3.71 (m, 1H), 3.71 – 3.63 (m, 1H), 3.63 – 3.53 (m, 1H), 3.36 (dd, J = 14.0, 6.2 Hz, 1H), 3.09 (dd, J = 14.0, 7.5 Hz, 1H), 2.51 (s, 1H), 1.98 – 1.81 (m, 1H), 1.81 – 1.64 (m, 3H), 1.10 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 149.1 (C), 149.0 (2C), 148.9 (C), 130.2 (C), 126.3 (CH), 126.2 (C), 125.4 (C), 125.1 (C), 123.9 (C), 108.0 (CH), 104.8 (CH), 103.6 (CH), 102.9 (CH), 62.69 (CH₂), 56.8 (CH), 56.3 (CH₃), 56.2 (CH₃), 56.1 (CH₃), 56.0 (CH₃), 55.9 (C), 41.5 (CH₂), 32.3 (CH₂), 28.7 (CH₂), 22.7 (CH₃); LRMS (EI) m/z (%) 503 (M+, 8.3), 447 (5), 382 (7), 311 (100), 118 (25), 88 (26), 71(13), 57 (13), 46 (9); HRMS (ESI) calcd for C₂₇H₃₈NO₆S [M + H]⁺ 504.2420, found 504.2411.

(*R*)-Tylophorine. To a solution of compound 13 (312 mg, 0.62 mmol) in dry THF (6.2 mL) under argon atmosphere was added PPh₃ (195 mg, 0.74 mmol), followed by DIAD (154 μ L, 0.78 mmol) at 25 °C. The reaction mixture turned yellow and was stirred for 16 h. The mixture was concentrated to dryness under reduced pressure and the residue was purified by flash chromatography (hexane/EtOAc 4:6 to 75% EtOAc). The obtained intermediate 14 [R_f 0.16 (3:7 hexane/EtOAc)] was contaminated with triphenylphosphine oxide according to ¹H NMR, and used in the next step without further purification.

To a mixture of compound **14** (155.2 mg, 0.52 mmol) in MeOH (5.2 mL) was added a solution of 4 M HCl in dioxane (3.25 mL) at 0 °C, under argon atmosphere. The mixture was stirred 1 h at 25 °C, then was added 37% formaldehyde (10.4 mL) and the solution was heated to 90 °C and stirred for 12 h in the dark. The cooled mixture was diluted with water (20 mL) and solution of 2 M NaOH (20 mL). The organics phases were extracted with CH₂Cl₂ (3 x 25 mL) and washed with water (10 mL) and brine (10 mL), dried over Na₂SO₄, and concentrated. The crude product was purified by flash column (100% CH₂Cl₂ to 95:5 CH₂Cl₂/MeOH), to obtain the desired product as a yellow solid (171 mg, 70%): $[\alpha]_D^{20}$ -85.0 (c 0.5, CHCl₃). {lit¹³ $[\alpha]_D^{20}$ -87 (c 1.0, CHCl₃)}; Rf 0.38 (95:5 CH₂Cl₂/MeOH); ¹H NMR, ¹³C NMR, IR and MS were in agreement with previously reported data. ^{9b, 10b, 11}

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Notes and references

- (1) (a) A. N. Rathnagiriswaran, K. Venkatachalam, Indian J. Med. Res. 1935, 22, 433.
- (b) T. R. Govindachari, M.V. Lakshmikantham, B. R. Pai, S. Rajappa, *Tetrahedron* 1960, **9**, 53.
- (2) R. B. Krishna, M. Balaji, R. P. Uma, G. Sailaja, K. Vaidyanath, G. Narasimha, *Afr. J. Biochem. Res.* 2009, **3**, 393-397.
- (3) M. Yadav, P. Dwivedi, P. Singh, V. K. Singh, *Indian J. Plant Physiol.* 2010, 15, 297-301.
- (4) K. Wang, B. Su, Z. Wang, M. Wu, Z. Li, Y. Hu, Z. Fan, N. Mi, Q. Wang, *J. Agric. Food Chem.* 2010, **58**, 2703-2709.
- (5) C. W. Yang, T. H. Chuang, P. L. Wu, W. H. Huang, S. Lee, *J. Biochem. Biophys. Res. Commun.* 2007, **354**, 942-948.
- (6) For an excellent review, see: (a) S. R. Chemler, *Curr. Bioact. Comp.* 2009, 5, 2-19.
 (b) The 60-cell line NCI data and in vivo data are available at: http://dtp.nci.nih.gov/dtpstandard/dwindex/index.jsp
- (7) M. Suffness, J. Douros, In: *Anticancer Agents Based on Natural Product Models;* Medicinal Chemistry Series, Vol. 16; Academic Press London, 1980; pp 465-487.
- (8) (a) D. Staerk, J. Christensen, E. Lemmich, J. O. Duus, C. E. Olsen, J. W. Jaroszewski, J. Nat. Prod. 2000, 63, 1584-1586. (b) W. Gao, W. Lam, S. Zhong, C. Kaczmarek, D. C. Baker, Y. C. Cheng, *Cancer Res.* 2004, 64, 678-688.
- (9) (a) W. Zeng, S. R. Chemler, J. Org. Chem. 2008, 73, 6045-6047. (b) D. N. Mai, J. P. Wolfe, J. Am. Chem. Soc. 2010, 132, 12157-12159. (c) M. B. Cui, H. J. Song, A. Z. Feng, Z. W. Wang, Q. M. Wang, J. Org. Chem. 2010, 75, 7018-7021. (d) S. V. Pansare, R. Lingampally, R. Dyapa, Eur. J. Org. Chem. 2011, 12, 2235-2238.
- (10) (a) X. Yang, Q. Shi, K. F. Bastow, K-H. Lee, *Org. Lett.* 2010, **12**, 1416-1419. (b)
 A. Stoye, T. Opatz, *Org. Lett.* 2010, **12**, 2140-2141. (c) M. J. Niphakis, G. I. Georg, *Org. Lett.* 2011, **13**, 196-199.
- (11) B. Su, F. Chen, Q. Wang, J. Org. Chem. 2013, **78**, 2775-2779 and references therein.
- (12) For recent synthetic applications of chiral homoallylic sulfinamines from our research group: (a) M. Medjahdi, J. C. Gonzalez-Gomez, F. Foubelo, M. Yus, *Eur. J. Org. Chem.* 2011, 2230-2234. (b) I. Bosque, J. C. Gonzalez-Gomez, F. Foubelo, M.

Yus, J. Org. Chem. 2012, 77, 780-784. (c) I. Bosque, Gonzalez-Gomez, J. C.; A. Guijarro, F. Foubelo, M. Yus, J. Org. Chem. 2012, 77, 10340-10346. (d) I. Bosque, J. C. Gonzalez-Gomez, M. I. Loza, J. Brea, J. Org. Chem. 2014, 79, 3982-3991.

(13) While this work was in progress, the use of chiral homoallylic sulfinamines in the synthesis of natural occurring phenantroindolizidine has been reported: Y. Zheng, Y. Liu, Q. Wang, *J. Org. Chem.* 2014, **79**, 3348–3357.

(14) For a recent comprehensive review on the synthetic applications of chiral *tert*butylsulfinamines, see: M. T. Robak, M. A. Herbage, J. A. Ellman, *Chem. Rev.* 2010, **110**, 3600-3740.

(15) (a) J. C. Gonzalez-Gomez, M. Medjahdi, F. Foubelo, M. Yus, *J. Org. Chem.* 2010,
75, 6308–6311. (b) J. C. Gonzalez-Gomez, F. Foubelo, M. Yus, *Org. Synth.* 2012, 89, 88–97.

(16) To determine the diastereoselectivity by ¹H-NMR and HPLC, an authentic diastereomeric mixture of homoallylic sulfinamines was prepared by epimerization of the sulfur atom, according to a reported procedure: K. Brak, K. T. Barret, J. A. Ellman, *J. Org. Chem.* 2009, **74**, 3606. This procedure allowed to unequivocally assign characteristic peaks of the minor diastereoisomer in the ¹H-NMR and the HPLC peaks corresponding to major and minor diastereoisomers.

(17) The absolute stereochemistry of compounds **5** and *ent*-**5** was predicted using our working model for this reaction (Reference 15), which has been also successful in former applications (Reference 12).

(18) Precedents for this pyrrolidine ring closure: (a) K. N. Hahn, O. O. Fadeyi, H. P. Cho, C. W. Lindsley, *Tetrahedron Lett.* 2012, 53, 3577-3580. (b) I. Bosque, E. Bagdatli, F. Foubelo, J. C. Gonzalez-Gomez, *J. Org. Chem.* 2014, 79, 1796-1804.

(19) L. M. Rossiter, M. L. Slater, R. E. Giessert, S. A. Sakwa, R. J. Herr, *J. Org. Chem.* 2009, **74**, 9554-9557.

(20) (a) The corresponding phenanthrene-9-carboxylic acid was prepared by Perkin condensation of homoveratric acid and veratraldehyde, followed by oxidative coupling mediated by FeCl₃, according to a reported procedure: K-L. Wang, M-Y. Lü, Q-M. Wang, R-Q. Huang, *Tetrahedron* 2008, **64**, 7504-7410. (b) Reduction of the acid to the corresponding alcohol was performed with LiAlH₄, see: Reference 4. (c) PCC oxidation afforded aldehyde **9**, according to: M. L. Bremmer, N. A. Khatri, S. M. Weinreb, *J. Org. Chem.* 1983, **48**, 3661–3666.

(21) M. Mayer, W. M. Czaplik, A. J. von Wangelin, Adv. Synth. Catal. 2010, 352, 2147–2152.