

## Angstrom-Resolved Interfacial Structure in Buried Organic-Inorganic Junctions

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Charge transport processes at interfaces play a crucial role in many processes. Here, the first soft x-ray second harmonic generation (SXR SHG) interfacial spectrum of a buried interface (boron–Parylene N) is reported. SXR SHG shows distinct spectral features that are not observed in x-ray absorption spectra, demonstrating its extraordinary interfacial sensitivity. Comparison to electronic structure calculations indicates a boron-organic separation distance of 1.9 Å, with changes of less than 1 Å resulting in easily detectable SXR SHG spectral shifts (ca. hundreds of milli-electron volts).

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Surfaces and interfaces play central roles in a variety of critical biological systems, electronics, batteries, and catalytic systems. Key chemical reactions and physical processes depend explicitly on the electronic structure of the interface and the dynamics across it. Experimentally, surfaces are often studied using a range of spectroscopic and imaging techniques, from grazing incidence x-ray scattering [1,2], to scanning probe [3,4] and total internal reflection [5–7] spectroscopies. These methods, however, are unsuited for the study of *buried* functional interfaces, which often govern critical chemical and physical processes. Interfaces between two bulk materials, or surfaces coated with macroscopically thick (hundreds of nanometers or greater) capping layers, cannot be specifically probed using these techniques. Absorption of photons by the bulk materials prevents optical probes from even reaching such interfaces, and the photoelectrons generated at the buried interface by high penetration x-ray probes cannot escape. Visible and IR second harmonic generation (SHG) and sum

frequency generation (SFG) [8,9] do not suffer from these limitations of studying buried interfaces. These techniques have been used for a variety of applications including characterization of semiconductors, solar cell devices, and biological structures [10–12]. While these techniques are quite powerful, they typically provide information about vibrational dynamics or valence transitions, and lack the elemental specificity of x-ray techniques.

Until recently, x-ray nonlinear spectroscopy was precluded by the lack of available light sources with sufficient coherence and flux, but the recent advent of x-ray free electron lasers (FELs) that generate femtosecond pulses with high peak powers and coherence has enabled such experiments [13–15]. Soft x-ray SHG (SXR SHG) offers powerful advantages compared to other surface-specific techniques [16,17]: it has high penetration depth and combines the element specificity of x-ray absorption spectroscopy with the interfacial specificity of second order nonlinear spectroscopies. Linear x-ray absorption

spectroscopy (XAS) is a general tool for studying compounds as it is element specific and sensitive to the chemical and molecular environment of a target atom. Element-specific measurements of core-to-valence transitions can resolve individual contributions to the electronic structure, which is not easily possible in optical spectroscopies that detect valence-to-valence or vibrational transitions. This is especially important for disentangling the contributions from hybridization at interfaces.

While x rays are highly penetrating, the use of different detection methods provides a range of depth sensitivities: transmission measurements are bulk sensitive, fluorescence detection is sensitive to approximately 1  $\mu\text{m}$  thick slabs based on the penetration of the photons [18], and photoelectron or total electron yield (TEY) detection provides sensitivity of a few nanometers due to the limited escape depth of photoelectrons [19–23]. Exploiting soft x-ray SHG has the potential for even higher surface sensitivity, since second harmonic (SH) photons are generated only from regions with broken centrosymmetry. Note that for hard x-ray radiation ( $>5$  keV;  $<0.25$  nm), the wavelength is small enough that the electric field is sensitive to inhomogeneities on an atomic scale, and the SH is generated throughout the material, rendering it bulk sensitive [24,25]. Instead, soft x-ray SH is generated only from 1–3 atomic layers at surfaces and interfaces of centrosymmetric media, as shown recently in the first demonstration of SXR SHG [16]. SXR SHG provides detailed electronic structure information analogous to that probed by x-ray absorption, with specificity for interfaces and no requirement for smooth surfaces. The elemental specificity of SXR SHG also reduces extraneous signals from interfaces containing other elements in a multilayer system. It is therefore ideally suited for the study of buried interfaces, but experimental proof is thus far lacking.

Here, we demonstrate the application of SXR SHG for probing the buried interface of a boron film with a support layer of Parylene N, a prototypical organic-inorganic interface. In the experiment, we compare the SXR SHG spectra of the boron-vacuum (B-V) and boron–Parylene N (B-PN) interfaces (Fig. 1), providing the first demonstration of probing the element-resolved electronic structure at a buried interface. To compare the interfacial sensitivity of SXR SHG to that of TEY XAS, we measured TEY XAS of similar samples (see Supplemental Materials for details [26] including Refs. [27–41]) using drain current for electron detection. The TEY XAS spectra of the B-V and B-PN samples are largely indistinguishable, because the probe depth of  $<10$  nm is too large. By contrast, the SXR SHG measurements show clear differences between B-V and B-PN interfaces, allowing a detailed determination of the bond characteristics. Accompanying detailed first principles calculations of SXR SHG spectra for both interfaces permits a detailed interpretation, showing that the observed experimental shift between the two spectra is

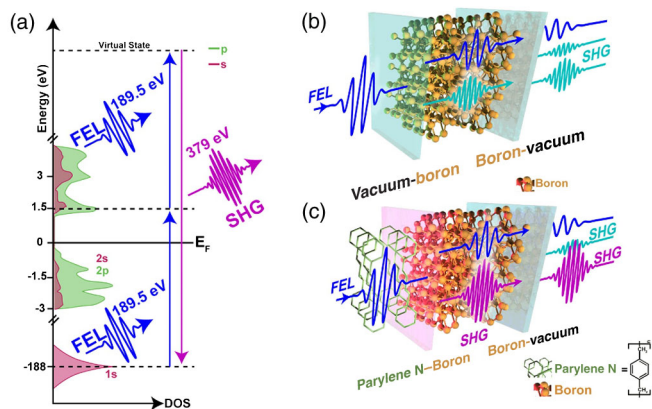


FIG. 1. Schematic of SHG from interfaces. In the energy level diagram (a), the density of states of boron ( $s$ -type red,  $p$ -type green) is resonantly pumped with an FEL pulse (blue). Because of selection rules, only the  $p$ -type states are probed. Two photons at this energy combine in the material and a second harmonic photon at twice the energy (purple) is emitted. The input energy is shown at 189.5 eV, generating a photon of 379 eV. During the experiment, the FEL energy was scanned from 184–196 eV. Two different interfaces were studied here, the (b) boron-vacuum interface and a (c) Parylene N–boron interface. The back boron-vacuum interface also generates some SHG signal (shown in light blue), but it will be less intense due to attenuation of the FEL pulse from transmission through the sample.

due to boron interactions with Parylene N. This indicates a strong surface spectral sensitivity to weak interactions, like London dispersion forces.

At the EIS-TIMEX beamline at the free-electron laser (FEL) FERMI [43], the SHG signal from the sample was detected using the same apparatus as in our recently reported study [16]. Nine different soft x-ray photon energies in the range from 184 to 200 eV were used. These fundamental input energies were just below, at, and just above the boron K edge [44]. The input intensity ( $I_0$ ) of the FEL was determined from the drain current of an ellipsoidal mirror upstream of the sample. The samples comprised an unsupported 200 nm boron film and a 200 nm boron film with a 100 nm Parylene N support layer, purchased from Lebow Corporation. TEY x-ray absorption spectra of the two materials were collected at Beamline 8-2 at SSRL with the Parylene N layer only 10 nm thick to enhance signal and the sample was mounted on silicon [45,46].

In our previous x-ray SHG study, it was found that the SHG signals were very sensitive to the quality of the FEL laser pulse [16]. Therefore, each FEL shot was filtered by the energy spectrum of the pulse collected before the sample [47]. The intensity of the generated SHG response is given by the relation

$$I_{\text{SHG}} \propto |\chi^{(2)}|^2 I_0^2,$$

where  $|\chi^{(2)}|$  is the second order nonlinear susceptibility of the interfacial layer of boron atoms. The SXR SHG

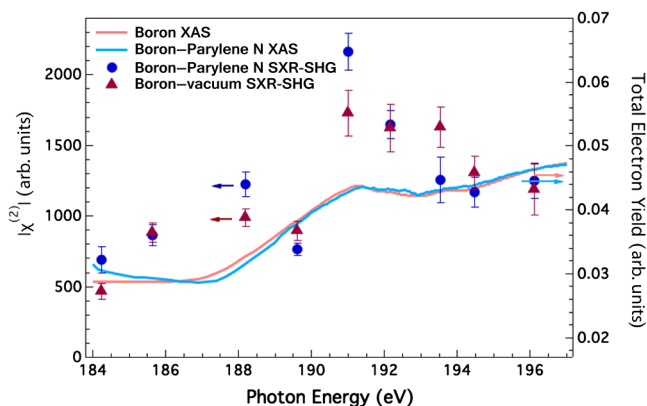


FIG. 2. Second harmonic generation spectra of the boron-vacuum and boron-Parylene N interfaces. The SHG spectrum of the B-V (dark red triangle) and B-PN (dark blue circle) interfaces, shown along with the linear x-ray absorption of the boron film (light red) and boron Parylene N multilayer film (light blue). The arrows indicate the appropriate y axis. The differences in the x-ray absorption spectra are believed to be due to the differences in the background and attempts to correct for it.  $|\chi^{(2)}|$ , determined from the linear regression slope of the SHG signal vs  $I_0^2$ .  $|\chi^{(2)}|$ , is significantly higher at the boron K edge for B-PN than for B-V. The error in FEL energy is smaller than the width of the marker size.

intensity was plotted proportionally to  $I_0^2$  (assuming constant pulse length and spot size) for each input energy and fit with a linear regression. As can be seen in the equation, the resulting slope is proportional to  $|\chi^{(2)}|^2$ . Finally, this slope was plotted as a function of photon energy to generate the nonlinear spectrum of the material properties at the surface or interface.

The measured SXR SHG spectra of the B-V and B-PN interfaces are shown in Fig. 2. Resonance effects can be seen in both spectra, as the SXR SHG intensity increases when the fundamental energy is at or above the boron K edge. There is an increase in cross section of B-PN compared to B-V at 188 eV, 191 eV, and a decrease at 193.5 eV. Most notably, we observe a substantial increase in the nonlinear response at energies slightly above the B K-absorption edge. We attribute this to dipole-allowed resonant transitions from  $1s$  to unoccupied states with B  $p$  character. The nonlinear response of this spectral region therefore becomes highly sensitive to the electronic valence structure of the interfacial bonds. The second photon absorption process, into a virtual state well above the conduction band to complete the SHG process, is non-resonant and thus less sensitive to the interfacial bonds. Well below and well above the edge, the SXR SHG spectra of B-V and B-PN are within error of each other.

In contrast to SXR SHG results, there are no major differences in the linear TEY spectra of the two materials (Fig. 2). The TEY spectra are essentially identical for both samples, indicating that this technique is insufficiently sensitive to the interface to capture the differences seen

in the present SXR SHG spectra. It should be noted that it is possible for some SXR SHG signal to be generated at the back B-V interface, but this contribution will be smaller than that of the front interface due to absorption of the fundamental by the boron layer. More specifically, and as we show, the SXR SHG signal is relevant for analysis at energies above the linear absorption edge, and we operate in direct resonance of the  $1s$  to  $2p$  transition (K edge). The attenuation length for x rays above the edge is approximately 50 nm. Therefore, the FEL beam is largely absorbed in the boron film and the remaining intensity on the rear side is expected to have a small contribution to the SXR SHG signal. In contrast, the generated SXR SHG signal (368 to 392 eV) from the front side is well above the B absorption edge and is therefore only weakly absorbed by the boron slab. Additionally, the back B-V interface is the same for both samples, and so will not affect any qualitative comparisons between the two interfaces. The Parylene N transmits over 90% in the photon energy ranges used here.

First principles electronic structure calculations via perturbation theory within density functional theory [16,48,49] were employed to simulate the SHG response function. Here, the B-V SHG calculation was performed using two layers of boron icosahedral unit cells. In order to understand the influence of the organic molecule on the electronic structure and resulting SXR SHG spectra at the interface, we use boron-ethane (B-E) as a proxy for B-PN for computational feasibility, since both B-E and B-PN have a similar calculated line shape (Supplemental Material, Fig. S7 [26]). Given resonant conditions, the second harmonic signal in this energy range is expected to arise primarily from the top boron layer [13], such that the difference between one layer and multiple layers of ethane in the simulation is negligible.

Our electronic structure calculations reveal a redshift of the SXR SHG spectrum at the boron K edge for B-E, as compared to B-V [Fig. 3(a)]. This is in general agreement with the experimental spectra for B-PN vs B-V, where we find a redshift of less than 2 eV of the main SXR SHG peak at 191 eV. Although for the larger calculation we were forced to use B-E for computational reasons, the differences were small as compared to B-PN [26]. We explored the effect of the interfacial bond length between boron and the organic layer on the simulated spectra in Fig. 3(b), where we find a monotonic blueshift in the 191 eV peak with increasing ethane separation, such that an increase in the separation distance from 0.9 to 1.9 Å resulted in a 2 eV shift, while a further increase to 2.9 Å lead to a further 1 eV shift. This shift is assigned to London dispersion forces as no other strong interactions are present. However, the exact spectral calculations are not quantitative because broadening, inaccuracies of the atomic positions, fluctuations in the molecular motions due to thermal energy, and the calculated spectral ringing are not

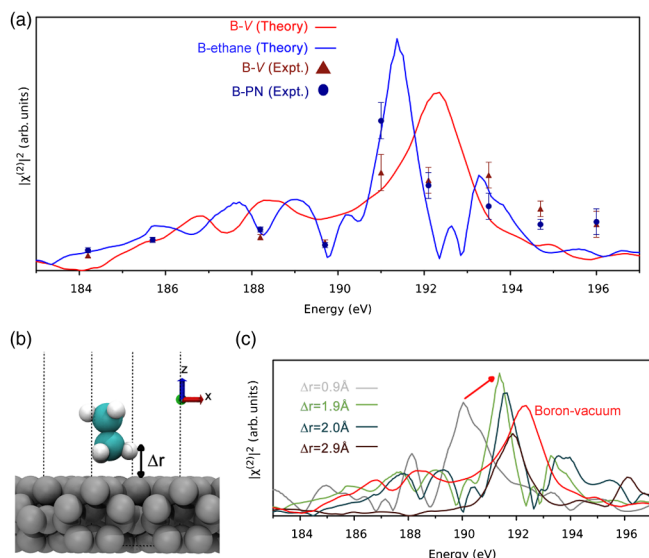


FIG. 3. (a) Comparison of theoretical calculations of SXR SHG boron-vacuum (red line) spectra and boron-ethane spectra (blue) at its equilibrium distance 1.9 Å with the experimental SXR SHG measurements of B-V (dark red triangles) and B-PN (dark blue circles). The calculation captures the enhancement of and the blueshift in the 191 eV feature in B-PN compared to B-V. (b) Schematic of boron-ethane computational unit cell. We vary the  $\Delta r$  distance between the hydrogen atom and the surface of boron. (c) Comparison of theoretical SXR SHG as a function of distance between boron and ethane, ranging from 0.9 Å (light gray), 1.9 Å (green), 2.0 Å (blue), to 2.9 Å (brown). As distance between the boron and ethane is increased, the primary peak at 191 eV blueshifts in energy (indicated by the red arrow).

accurately represented. A separation distance of 1.9 Å agrees with the minimum energy distance based upon a relaxation within DFT, and is a good match to the experimental SHG spectrum. Of note, an increase in the separation distance from 1.9 to 2.0 Å resulted in a 200 meV spectral shift. As 200 meV is easily resolvable at x-ray FELs, the ability to determine distances with a resolution of 0.1 Å using this computation method has been shown previously, particularly in the bulk [16,42]. This predicted high sensitivity of SXR SHG to interfacial bond length indicates a unique and general technique for elucidating interfacial structure, but due to the low spectral density such a high accuracy cannot be obtained here. Our calculations reveal that ethane induces a shift in the Hartree potential near the interface due to electronic screening, the magnitude of which is strongly dependent on the proximity of the ethane layer to the boron surface. Density of states calculations (Fig. S10 [26]) indicate that the boron core energy levels lie at lower energy in B-E compared to B-V as a result of this screening. Thus, the experimental spectral differences between B-V and B-PN can be attributed to electronic screening in the interfacial boron atoms, rather than to specific interactions, e.g., orbital hybridization.

In conclusion, linear XAS spectra of the B-V and B-PN samples exhibited no observable difference between the two samples, whereas SXR SHG reveals distinct differences due to its sensitivity to the interface. This is the first time that a buried interface has been resolved with atom-specific sensitivity. More generally, these experiments clearly demonstrate the sensitivity of SXR SHG to subtle changes in the interfacial electronic structure of the buried interface with sensitivity to a single atomic layer. The results show that SXR SHG is highly sensitive to interfacial bond lengths and to subangstrom bond length changes, resulting in measurable spectral shifts in the hundreds of milli-electron volt range. Under the assumption of comparability of an ethane-boron interfacial bond with a Parylene N–boron interfacial bond, it was possible to determine the bond length to be approximately 1.9 Å. However, the fact that ethane had to be used as a proxy for Parylene N in simulations of varying distance highlights the importance of developing numerical methods to enable computation of larger systems with high fidelity. While SXR SHG spectroscopy clearly is a unique and powerful new tool, profound understanding of interfacial electrostatics will require a carefully orchestrated duet of theory and experiment. Because of the ultrafast nature of the probe and its sensitivity to single atomic layers, the technique has great potential for future studies of dynamics of buried interfaces in electrochemical cells and catalysts. In the near future, SXR SHG spectroscopy can be used to probe the interfacial electronic structure in a variety of other systems of critical interest, including electronics, batteries, and photocatalytic systems that are difficult to study with other methods. Moreover, the newly revealed high sensitivity to interfacial bond lengths and symmetries will enable unique studies on interfacial strain and its influence on electronic transport properties.

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