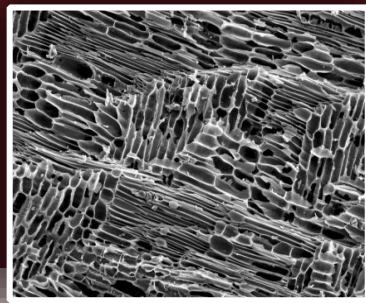


EFFECT OF BIOCHAR ON THE FATE AND BEHAVIOR OF ALLELOCHEMICALS IN SOIL

K. E. Hall, M. J. Calderon, K. A. Spokas, L. Cox, W. C.
Koskinen, J. Novak, K. Cantrell

Biochar

- ❑ Carbon-rich byproduct of biomass pyrolysis
- ❑ Use in carbon sequestration and as a soil amendment
- ❑ Benefits
 - ▣ Increased crop yield, plant growth, nutrient retention, water holding capacity, enhanced biological activity
- ❑ Neutral and negative effects
 - ▣ Plant growth suppression, decreased arbuscular mycorrhizal fungi



Biochar - diversity

Pyrolysis systems



- ❑ Remarkable variety in the chemical and physical properties of biochars
- ❑ Due to variation in:
 - ❑ Feedstock materials
 - ❑ Pyrolysis conditions
 - ❑ Post-production factors
- ❑ Mixing feedstock materials in different ratios prior to pyrolysis further enhances the diversity

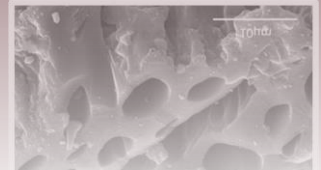
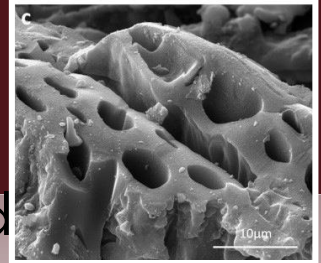
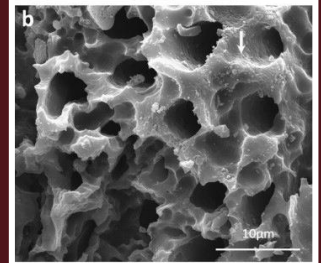
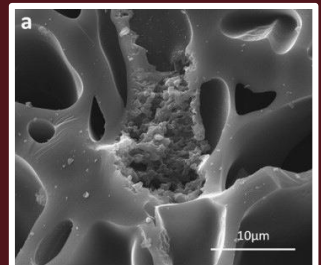
Biochar - diversity

- ❑ Biochars from feedstock mixtures offer potential customization of properties → “designer biochars”
 - ❑ Rebalance soil P concentrations
 - ❑ Improve soil moisture retention
- ❑ Combination biochars have chemistries non-predictable from their individual components
- ❑ Range of heterogeneous materials with non-uniform properties effects and behaviors



Biochar - sorption

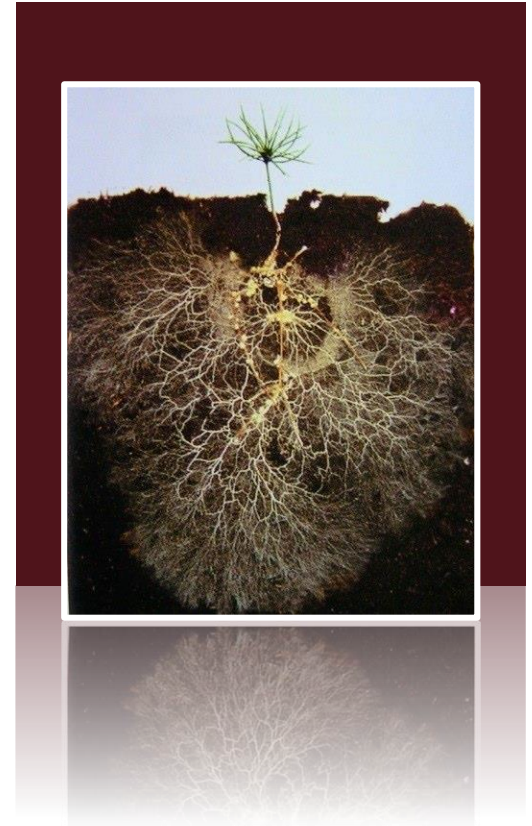
- ❑ Variability in surface properties affects sorption
 - ❑ specific surface area (SSA)
 - ❑ aromaticity
 - ❑ microporosity
- ❑ Research heavily focused on pesticides and environmental contaminants
- ❑ Naturally occurring aromatic acids (i.e. phenolic acids) from root exudates and vegetative materials may also be immobilized



Phenolic acids

Influence:

- Nutrient uptake
 - Protein synthesis
 - Humus formation
 - Plant signaling
 - Development of mutualistic relationships
 - **Allelopathy**
- Allelopathic effects → potential use in weed management in agroecosystems

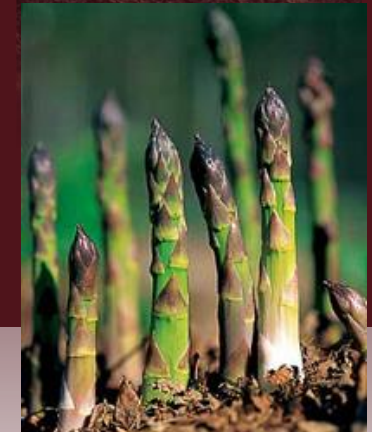


Phenolic acids - phytotoxicity

- ❑ Phytotoxicity of phenolic acids is affected by their bioavailability, persistence, and fate in soil
- ❑ Only effective in their free form (unbound)
- ❑ Sorption studies necessary to determine efficacy
 - ❑ Sorption to soil research available
 - Cecchi et al., 2004
 - Tharayil et al., 2006
 - ❑ Sorption to biochar research is lacking
 - Ni et al., 2011

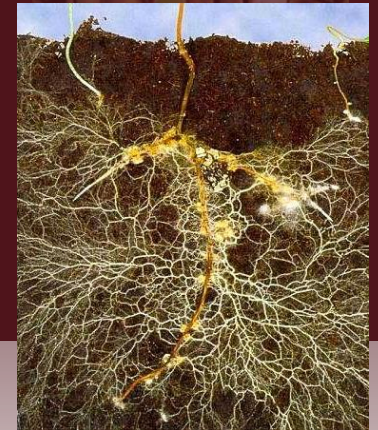
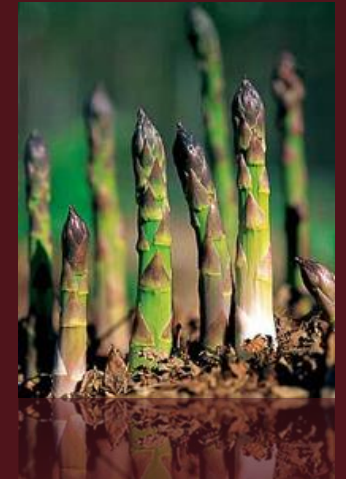
Biochar and allelochemicals

- ❑ Plant growth studies
- ❑ Biochars greatly differ in their ability to disrupt the function of allelopathic chemicals
- ❑ Reduce inhibitory effect of allelochemicals on corn seedling growth
- ❑ Asparagus- phenolic acid allelochemicals suppress seedling growth



Biochar and allelochemicals

- ❑ Allelochemicals negatively affect AM root colonization in asparagus (Elmer & Pignatello, 2011)
- ❑ Biochar can impact on mycorrhizal abundance/functioning
- ❑ One proposed mechanism: detox of allelochemicals or alteration of plant-fungus signaling



Objectives

- ❑ **Characterize the influence variation in biochar feedstock has on the sorption of 2 phenolic acids, ferulic acid and syringic acid, and dichlorocatechol**

Objectives

- ❑ **Characterize the influence variation in biochar feedstock has on the sorption of 2 phenolic acids, ferulic acid and syringic acid, and dichlorocatechol**
- ❑ **Determine how sorption by biochars prepared from mixtures of feedstock materials differs from those prepared from the pure feedstocks**

Biochar feedstock materials



Switchgrass (SG)



Swine solids (SS)



Poultry litter (PL)



Pine chips (PC)

Biochars



→ SG:SS (80:20)



→ PC:PL (90:10)

→ PC:PL (80:20)

→ PC:PL (50:50)

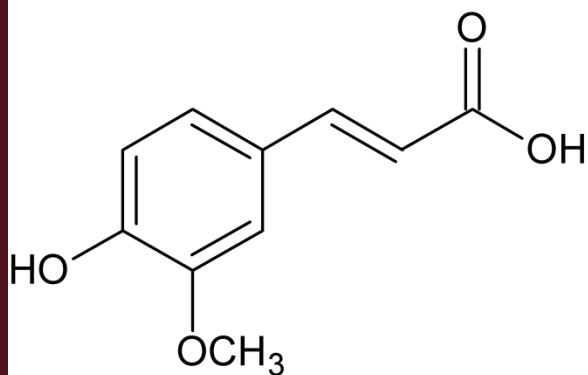
- ❑ Pyrolysis: 350°C, 2 hrs
- ❑ Waukegan silt loam soil (Rosemount, MN) included for comparison

Biochar properties

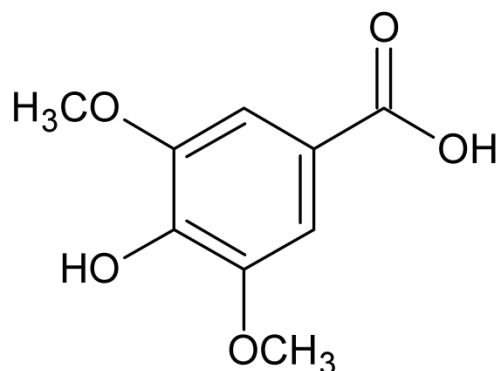
Feedstock	pH	% ash	% C	% H	% O	SSA (m²g⁻¹)
<i>Pure feedstock (100%)</i>						
SG	7.4	3.2	75.5	4.6	16.2	0.50
SS	6.5	35.0	51.0	3.7	3.2	1.01
PL	9.4	32.1	51.5	3.6	6.9	1.94
PC	7.1	1.8	78.7	4.9	14.3	<0.10
<i>Feedstock mixtures (w:w ratios)</i>						
SG:SS (80:20)	6.5	7.3	75.9	4.6	10.8	1.35
PC:PL (90:10)	6.4	4.4	78.1	4.8	11.7	1.11
PC:PL (80:20)	7.5	7.3	75.8	4.6	10.8	1.09
PC:PL (50:50)	7.4	18.5	63.7	3.8	10.3	1.14

Chemicals

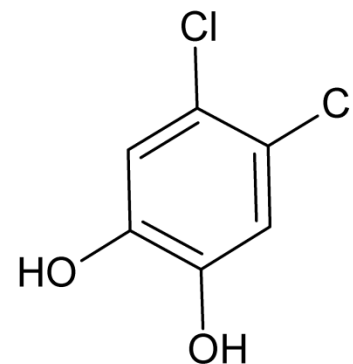
trans-ferulic acid



syringic acid



4,5-dichlorocatechol



- ❑ ¹⁴C labeled chemicals provided by Dr. Konrad Haider, Deisenhofen, Germany
- ❑ Purified by thin-layer chromatography

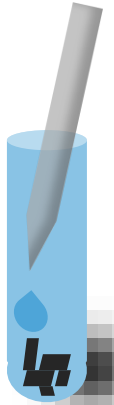
Sorption experiment

- Performed in duplicate using **batch equilibration method**

Biochar/soil (0.5 g)
added to 35 mL glass
centrifuge tubes with
Teflon lined caps



5 mL of a 1 $\mu\text{g mL}^{-1}$
phenolic acid ($>17 \text{ Bq mL}^{-1} \text{ }^{14}\text{C}$) in 0.005 M
 CaCl_2 solution added



Samples were shaken
horizontally approx.
18 h and centrifuged



Supernatant was analyzed for
 ^{14}C by liquid scintillation
counting with a Packard 1500
counter



Sorption calculations

- Sorption distribution coefficient, K_d (L kg⁻¹)

$$K_d = C_s / C_w$$

- Normalization to biochar OC content, K_{oc} (L kg⁻¹)

$$K_{oc} = (K_d / \%OC) \cdot 100$$

Pure feedstock biochar sorption

**Biochar
 K_d values
(L kg⁻¹)**



Switchgrass



Swine solids



Poultry litter



Pine chips

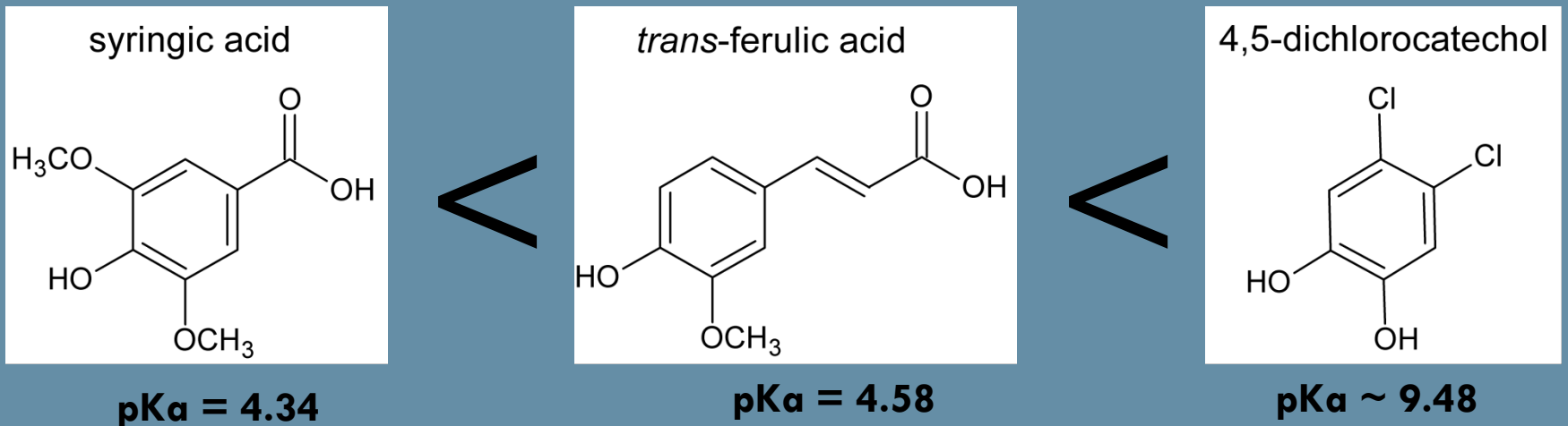
Ferulic acid	1.4 ± 0.18	1.6 ± 0.15	3.1 ± 0.40	75 ± 8.00
Syringic acid	0.07 ± 0.10	0.41 ± 0.02	0.43 ± 0.00	6.03 ± 0.06
Dichlorocatechol	*	*	25 ± 0.25	*

syringic acid < ferulic acid < dichlorocatechol

* Chemical was sorbed completely

Pure feedstock biochar sorption

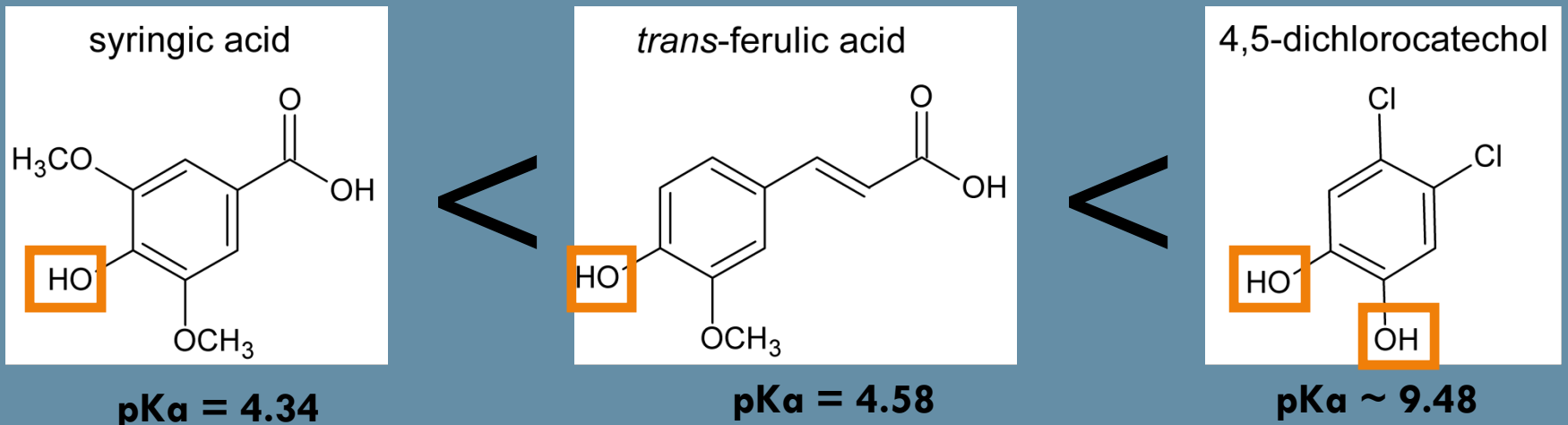
Sorption strength related to chemistries of compounds



- ❑ Cinnamic acid derivatives (ferulic) sorb more strongly than benzoic acid derivatives (syringic) to soil (Dalton et al., 1989)
- ❑ At pHs of biochars in this study (pH=6.4-9.4) syringic and ferulic acid are predominantly anionic; dichlorocatechol remains neutral

Pure feedstock biochar sorption

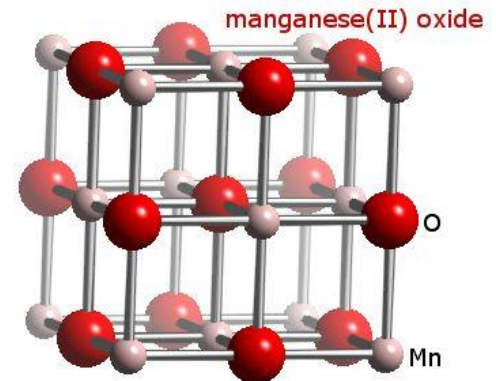
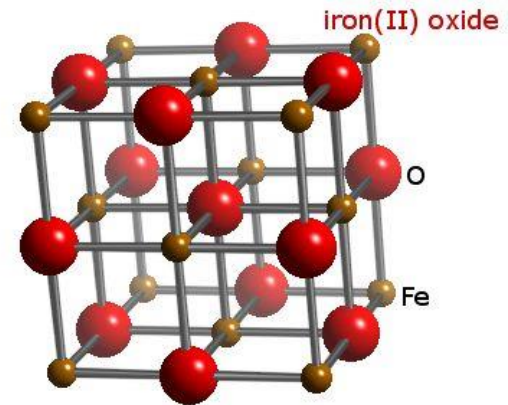
Sorption strength related to chemistries of compounds



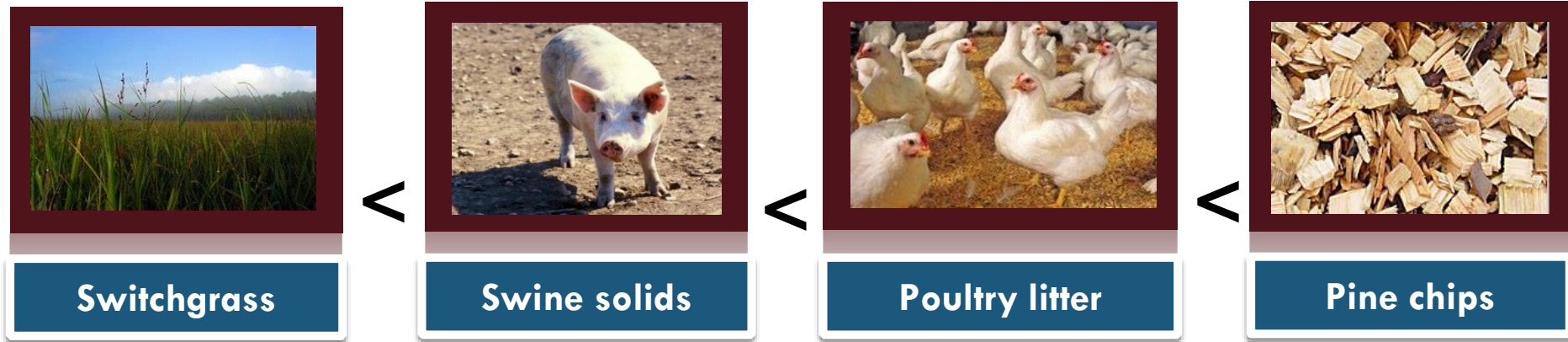
- Biochar has negative surface charge → molecular phenolic group more readily sorbed
- Free phenolic groups → greater sorption to soil (Cecchi et al., 2004)

Pure feedstock biochar sorption

- ❑ Ferulic and syringic acid may also rapidly react with iron and manganese oxides in biochar
- ❑ Biochars and soil can contain these metals- may provide pathways for abiotic interactions with metal oxides
- ❑ Removal of metal oxides from soils largely decreased sorption of phenolic acids
- ❑ Extent not analyzed



Pure feedstock biochar sorption



- ❑ Pure feedstock biochar K_d and K_{oc} values increased in same order for ferulic and syringic acid
- ❑ Certain biochar characteristics also influencing sorption
- ❑ No correlations between measured biochar properties (pH, OC, ash content) and sorption observed
- ❑ Correcting for OC did not reduce sorption variability

Pure feedstock biochar sorption

- ❑ Biochar characteristics and feedstock material
- ❑ Efforts to identify trends
- ❑ General grouping of feedstock:

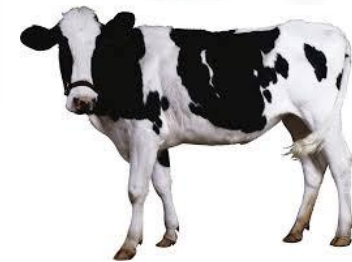


Wood vs Non-wood

Wood has...

PC

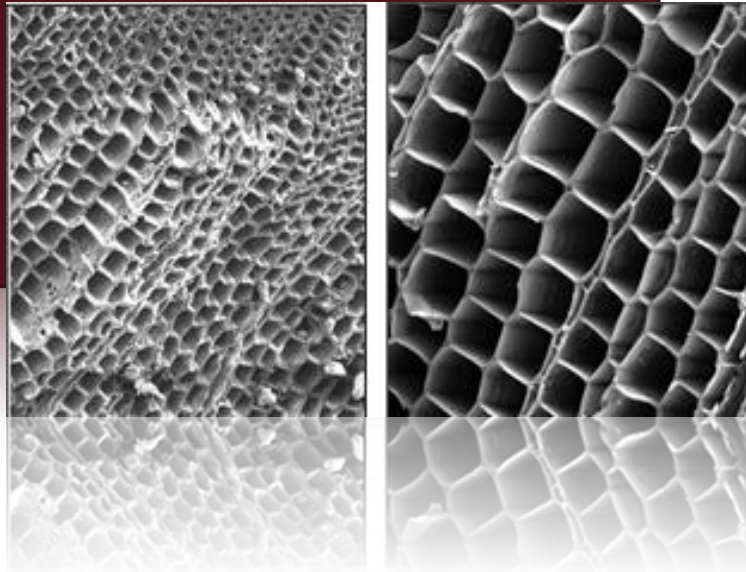
- lower ash content
- lower pH
- higher C/N ratio
- higher SSA



- subgroups: hard-wood, soft-wood, grass, manure

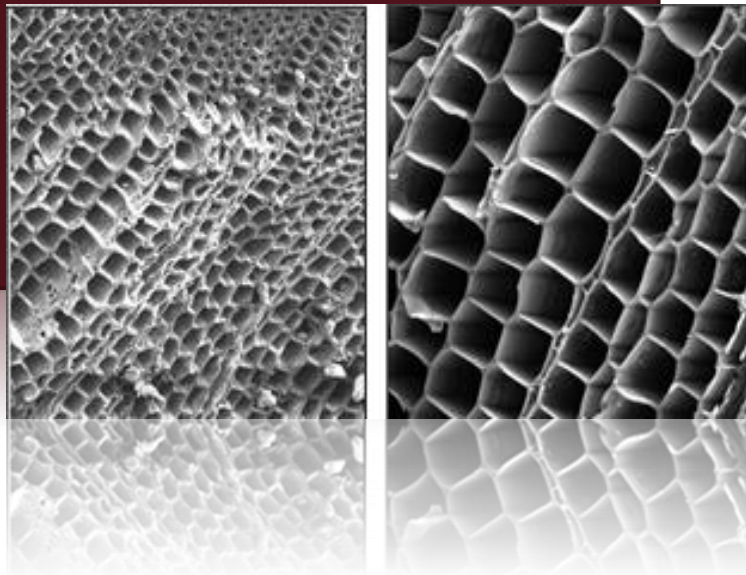
Pure feedstock biochar sorption

Higher SSA typically associated with higher sorption



- ❑ PC biochar had the *highest* sorption, but *lowest* SSA of all biochars studied
- ❑ Lower SSA may be due to resins, tars, or oils blocking pore space
- ❑ Resins in wood may alter surface properties of pores — alter sorption capacity

Pure feedstock biochar sorption



- ❑ Biochar maintains relic structure of parent material
- ❑ Pore distribution can vary
- ❑ Feedstock materials with large diameter cells → biochars with more macropores → adsorb large molecules →

- ❑ SSA measurements do not account for size/shape of pores
- ❑ Biochar SSA is temperature dependent

Pure feedstock biochar sorption

Effects of pyrolysis temperature on biochar properties

Low temp vs High temp

High temperature pyrolysis biochar has...

- Increased SSA
- Increased microporosity
- Decreased H/C ratio
(i.e. increased aromaticity)
- Decreased cation exchange capacity

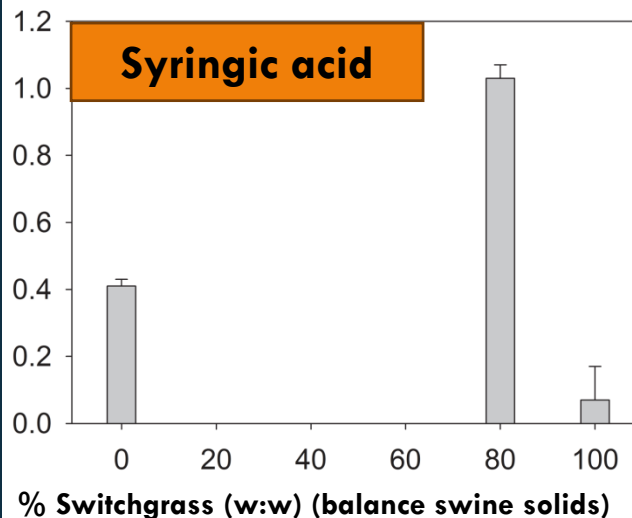
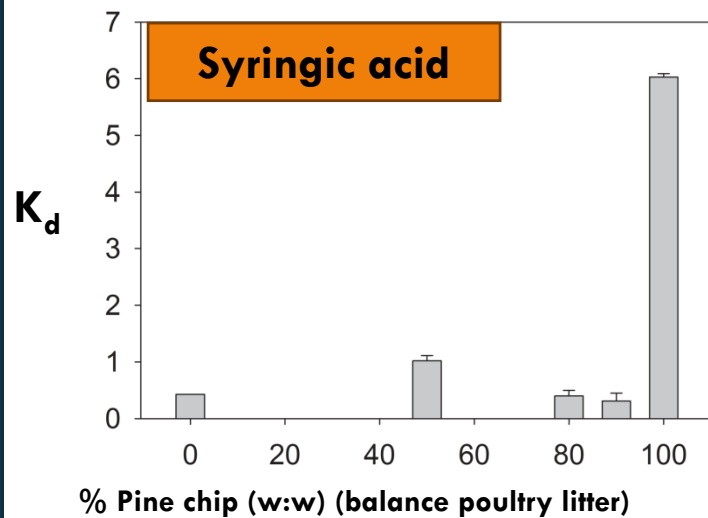
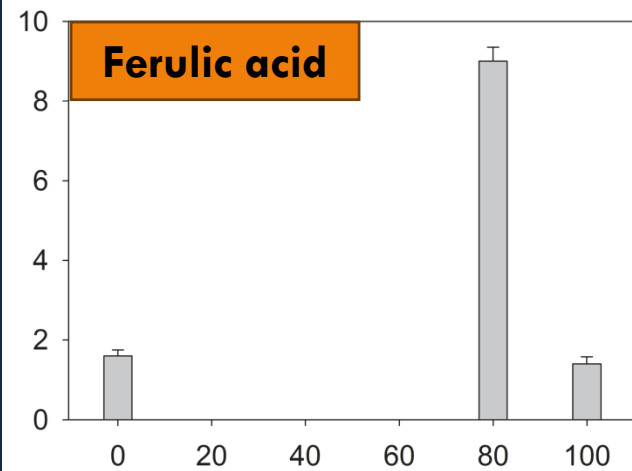
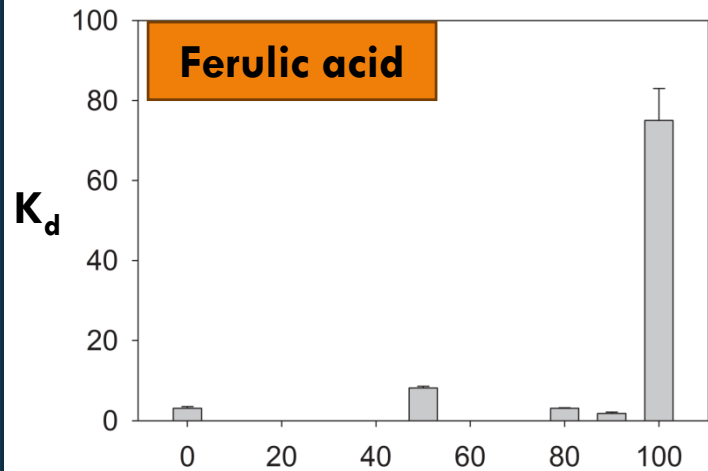
Pure feedstock biochar sorption

Effects of pyrolysis temperature on biochar properties



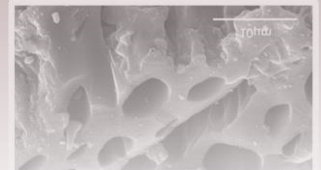
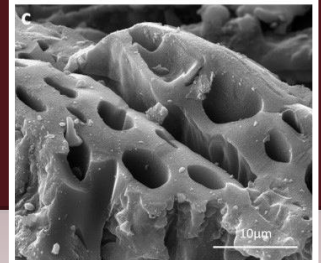
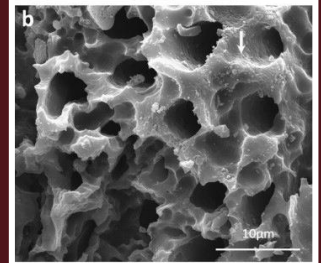
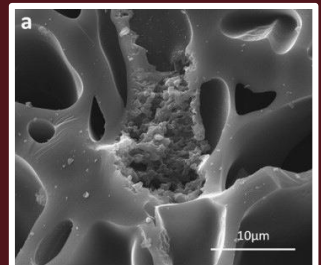
- ❑ High temperature (550°C) olive mill waste biochar
- ❑ Syringic acid $K_d = 14.6$
 - PC (350°C) $K_d = 6.0$
- ❑ Ferulic acid $K_d = 236$
 - PC (350°C) $K_d = 75$
- ❑ High SSA ($9.82 \text{ m}^2\text{g}^{-1}$)
 - PL (350°C) SSA = $1.94 \text{ m}^2\text{g}^{-1}$

Combination biochar sorption



Combination biochar sorption

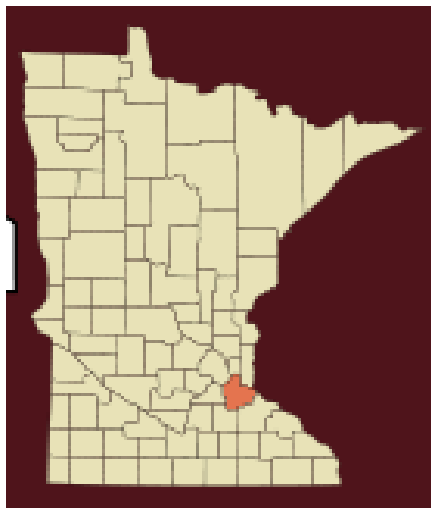
- ❑ Unable to predict sorption to combination biochars from amount sorbed to pure feedstock components
- ❑ Physicochemical alterations during pyrolysis
 - ❑ Trace metal constituents can act as catalysts → alter surface chemistries



Soil sorption



	SOIL	
	K_d (L kg ⁻¹)	K_{oc} (L kg ⁻¹)
Ferulic acid	29 ± 0.50	1160 ± 20
Syringic acid	12.0 ± 0.76	482 ± 30.4
Dichlorocatechol	56 ± 4.00	2240 ± 160



- Waukegan silt loam soil (Rosemount, MN)
- 6.0 pH in water
- 2.5% OC
- 15% clay
- 33% sand

Soil sorption



- ❑ **Syringic acid**

 - Sorption to soil > sorption to all biochars studied

- ❑ **Ferulic acid**

 - Sorption to soil > sorption to all biochars (except PC)

- ❑ Soil K_{oc} much higher than all biochar K_{oc} values

- ❑ Higher phenolic acid sorption to soil may also result from mineral interactions

Conclusion



- ❑ Biochars in this study (except PC) sorbed ferulic and syringic acid less than the reference soil
- ❑ Incorporation into this soil not likely to alter bioavailability of these phenolic acids to a large degree
- ❑ Biochar may have greater impacts on immobilization if...
 - ❑ Incorporated into soil with lower sorptive capacity
 - ❑ Presence of phenolic acids with different chemistries

Conclusion



- ❑ Phenolic acid structure, particularly the hydroxyl group, may impact its sorption to biochar
- ❑ Physicochemical variability among biochars affects their sorptive behavior
 - ❑ No observed correlation between sorption and biochar pH, OC, % ash, or SSA
- ❑ Combining feedstock materials unpredictably affects biochar sorption characteristics

THANK YOU

